Asymmetric Synthesis by Chromium(II) Amino Acid Complexes

Csongor Hajdu

Supervisor: Dr. Károly Micskei

University of Debrecen, Faculty of Sciences
Department of Inorganic & Analytical Chemistry
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Chiral molecules play an essential role in the chemistry of biological processes. In the biosynthesis of natural products amino acids are used as chiral inductors as they build up enzymes, the catalysts of these processes.

Here we describe a system consisting of chromium(II) as reducing agent and amino acid ligands as chiral inductors which proved to be an effective reagent in enantiomeric reduction of prochiral functional groups in aqueous medium.

In our earlier work we found that acetophenone was reduced by chromium(II) amino acid complexes with excellent chemoselectivity (>95%) and enantioselectivities up to 75% e.e. In order to study the effect of the structure of various prochiral substrates, aryl alkyl ketones and benzo(hetera)cyclanones were treated with chromium(II) amino acid complexes in a DMF-water solvent mixture at room temperature. We found that that both yields and enantioselectivities strongly depend on the substrates and ligands. L-alanine, L-histidine and L-aspartic acid gave high conversions while L-valine and L-leucine reacted more sluggishly. The highest enantioselectivities were achieved using L-histidine (up to 55% e.e.). Data show a difference in the action of bi- and tridentate ligands: L-alanine and L-valine generated *R* alcohols while L-histidine and L-aspartic acid lead mostly to *S* products. The structure of the substrates also affects the obtained optical purity.

Enantioselective synthesis of ferrocenyl alcohols is questionable in many cases. Formerly it was found that by reduction of ferrocenyl ketones with chromium(II) complexes stable ferrocenylketyl radicals were obtained instead of the alcohols. By modification of the conditions, we could bring the reaction towards the formation of the alcohol. Thus, using amino acid complexes, it was possible to reduce monobenzoyl ferrocene with 2-25% e.e. The low stereochemical outcome can be explained by the formation of the stable radical that allows racemization processes.

Selective transformation of α -diketones plays an important role in synthesis of natural products. We reduced benzil with chromium(II) amino acid complexes in order to obtain the corresponding products selectively. In these reactions the enantioselectivities remained under 30%.

Some recent works support that the studied ketone reductions occur via organochromium(III) complex intermediates. We monitored the reaction of several ketones with chromium(II) amino acid complexes using UV-VIS spectrophotometry, detecting the absorption band in the 250-290 nm region, characteristic for C–Cr bonds. We found that the intermediates form with all studied substrates and complexes and proposed a mechanism for enantioselective ketone reduction.

The reduction of oximes has a more complicated mechanism. In reduction of acetophenone oxime with chromium(II) complexes, our first experiments lead to unwanted side products like acetophenone, 1-phenyl-ethanol and several coupling products. After optimization of the reaction it was possible to obtain the 1-phenyl-ethyl-amine with >95% chemoselectivity. The application of several amino acids as ligands the amine formed with enantiomeric excesses up to 50%.

Considering the essential role of amino acids in life phenomena, the asymmetric synthesis of amino acids using also amino acids as chiral inductors, might have special importance. We performed reduction of the C=N double bond of oxime precursors of α-amino acids in aqueous medium by chromium(II) complexes of amino acids, using the reaction conditions developed formerly for the aromatic oximes. The reduction of oximes of α -ketophenylacetic, α -keto- β -phenylpropionic and α -ketopropionic acids proceeded up to 90% conversion and 2-30% enantiomeric excess. Our UV-VIS of spectrophotometric demonstrated measurements the presence the organochromium(III) intermediate. This means that the reaction undergoes mostly through the basic steps unveiled formerly for the ketones.

We performed the diastereoselective pinacol coupling of several aromatic aldehydes. The most effective ligand was L-histidine that gave much better chemo-and enantioselectivity (e.e. = 67%) than other amino acids. We investigated the behavior of other aromatic aldehydes with chromium(II) L-histidine complex. In most cases we obtained conversions above 90% and enantiomeric excesses between 30-62%. We suppose that in an enantioselective coupling reaction the formation of the carbon-carbon bond occurs inside the common (chiral) coordination sphere of two associated organometallic intermediates.

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