

Article

Enhancing Biochemical Methane Potential and Enrichment of Specific Electroactive Communities from Nixtamalization Wastewater using Granular Activated Carbon as a Conductive Material

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Abstract: Nejayote (corn step liquor) production in Mexico is approximately $1.4 \times 10^{10} \text{ m}^3$ per year and anaerobic digestion is an effective process to transform this waste into green energy. The biochemical methane potential (BMP) test is one of the most important tests for evaluating the biodegradability and methane production capacity of any organic waste. Previous research confirms that the addition of conductive materials significantly enhances the methane production yield. This study concludes that the addition of granular activated carbon (GAC) increases methane yield by 34% in the first instance. Furthermore, results show that methane production is increased by 54% when a GAC biofilm is developed 10 days before undertaking the BMP test. In addition, the electroactive population was 30% higher when attached to the GAC than in control reactors. Moreover, results show that electroactive communities attached to the GAC increased by 38% when a GAC biofilm is developed 10 days before undertaking the BMP test, additionally only in these reactors *Geobacter* was identified. GAC has two main effects in anaerobic digestion; it promotes direct interspecies electron transfer (DIET) by developing an electro-active biofilm and simultaneously it reduces redox potential from -223 mV to -470 mV . These results suggest that the addition of GAC to biodigesters, improves the anaerobic digestion performance in industrial processed food waste.

Keywords: biochemical methane potential; redox potential reduction; direct interspecies electron transfer; electroactive biofilm; Nejayote; granular activated carbon

1. Introduction

The Mexican corn tortilla industry produces between $2.2\text{--}3.5 \text{ m}^3 \cdot \text{t}^{-1}$ of wastewater from processed corn [1,2]. The wastewater (Nejayote) presents an environmental problem for Mexican society due to its malodorous nature and its composition. It is imperative to have an explicit waste management strategy to deal with the unpleasant by-products, and it is also essential to ascertain the potential uses of the wastewater in order to harness and capitalise on the large quantities that are produced through nixtamalization. Nixtamalization is the ancient Aztec process of cooking corn with lime to produce corn masa [3]. Water, corn, and lime are cooked at $80 \text{ }^\circ\text{C}$ during three hours and is then

steeped for 15 hours [4]. Nowadays, the process was carried out on an industrial scale but based on the process carried out by the Aztecs known as nixtamalization. This complex waste has a pH in the 9–14 range, a chemical oxygen demand (COD) between 3.5 and 40 g·L⁻¹ and a phenol concentration of 4.2 g·L⁻¹ [3–6]. This wide range is a result of the proportions of water corn and lime that vary in different nixtamal plants [4,5,7]. Its organic matter content makes nejayote a suitable waste for anaerobic digestion process rather than any aerobic process due to the huge amount of energy that is required to degrade its high organic load [8–10]. A biochemical methane potential (BMP) test is the most efficient means of identifying the methane production capacity, in anaerobic digestion, for any organic waste [11–13]. Additionally, it can be useful to identify future inhibition and adaptation problems before scaling the process [14]. Given the lack of previous studies in relation to BMP testing of nejayote, this article is the first step in determining the biochemical methane potential via direct interspecies electron transfer (DIET) of nejayote in anaerobic digestion and it provides an opportunity to develop a perspective on its possible performance on a larger scale.

Anaerobic digestion process is based in four main steps: hydrolysis, acidogenesis, acetogenesis, and methanogenesis. In methanogenesis, step electrons are transported from acetoclastic bacteria to hydrogenotrophic archaea to reduce carbon dioxide to methane. This transport can be done by shuttle molecules as hydrogen or formate. Traditionally, interspecies hydrogen transfer (IHT) and interspecies formate transfer (IFT) have been intrinsically linked with interspecies electron transfer (IET). However, these can present problems in relation to H₂ partial pressure and formate concentration, respectively [15]. More recently, there has been a greater production of studies referring to DIET, which has become more widely accepted. DIET is another form of IET that can be produced through pili or employing conductive materials [16–19]. Exoelectrogenic bacteria and electrorophic methanogens are the partners that participate in DIET [20–22]. Short chain volatile fatty acids (VFA) and alcohols are the biodegradable compounds in the syntrophic association between exoelectrogenic bacteria and electrorophic methanogens [23,24]. For the first time in 2014, the DIET between exoelectrogenic bacteria and electrorophic methanogens was evidenced through pili in a *Geobacter-Methanosaeta* co-culture [18]. Not all exoelectrogenic bacteria are able to generate pili; however, the addition of conductive materials increases the genera of bacteria that are able to supply pili the genera of bacteria that are able to participate in DIET. In cases where the bacteria are not able to generate pili, conductive materials fulfil the role of pili in terms of facilitating electron exchange [23,25,26]. Granular activated carbon has shown high performance in promoting DIET in different studies [26,27]. Moreover, its low cost [28,29] and its high conductivity (3000 μS·cm⁻¹) [21] when compared to other conductive materials such as biochar (4 μS·cm⁻¹) [30], magnetite (160 μS·cm⁻¹) [31], or stainless steel (667 μS·cm⁻¹) [32], make it a preferential material to promote DIET in order to improve methane production yields [17,33]. Additionally, metallic materials, such as stainless steel, have corrosion problems [34].

Within the past three years, through the microorganism identification of biofilms developed in conductive materials, new exoelectrogenic bacteria have been discovered, such as *Thauera*, *Sporanaerobacter*, *Enterococcus*, *Pseudomonas*, *Anaerolinaceae*, *Bacteroides*, *Streptococcus*, *Syntrophomonas*, *Sulfurospirillum*, *Caloramator*, *Tepidoanaerobacter*, *Coprothermobacter*, *Clostridium sensu stricto*, *Peptococcaceae*, and *Bacillaceae*, also new electrorophic methanogen genera have been found able to participate in DIET, such as *Methanosarcina*, *Methanobacterium*, *Methanolinea*, *Methanothrix*, *Methanoregula*, and *Methanospirillum* [17,27,35–44]. However, the number of bacteria that are capable of participating in DIET through conductive materials is still unknown [45].

Industrial processed food is one of the most important industries that employ anaerobic digestion to degrade water pollution due to its high COD [10]. Most previous studies have used simple substrates such as ethanol or short chain VFA, or commercial waste, such as dog food [17,23,27]. At present, there are no studies that use a substrate of industrial food waste using conductive materials to promote DIET.

This study is the first step towards developing an understanding of the behaviour and growth of electroactive communities in a real wastewater, and thus providing an indication of the improvements in methane production and VFA and COD degradation, which can be made by implementing DIET through conductive materials on a large scale.

2. Materials and Methods

2.1. Nejayote and Inoculum

Nejayote was collected from an industrial corn flour plant in Yucatan State (Mexico). Nejayote was preserved in a cold box at 4 °C from the corn flour plant to the laboratory, which is in accordance with the Standard Test Method 1060 for the collection and preservation of samples [46]. Prior to undertaking the BMP test, the nejayote was characterized according to the following parameters: pH, total suspended solids (TSS), total solids (TS) and volatile solids (VS), COD, total nitrogen (TN), phosphates (PO_4^{-3}), ammonia nitrogen (NH_3N), sulfate (SO_4^{-2}), and total alkalinity (TA).

The inoculum composition employed was made according to Poggi Varaldo et al. [47], while using 30 g·L⁻¹ of deep soil, 300 g·L⁻¹ of cattle manure, 150 g·L⁻¹ of pig manure, 1.5 g·L⁻¹ of commercial Na_2CO_3 , and 1 L of tap water. The inoculum was characterized for TA, TS, and VS.

2.2. BMP Test

The trial was carried out in triplicate in 110 mL serum bottles that were capped with rubber septum sleeve stoppers. Three experimental conditions were assayed. In controls, the reactors (N) were filled with 60 g of inoculum and nejayote at a ratio of 2 based on VS [12]. In three reactors (N0) 3 g of GAC was added [33] at the beginning of the BMP test. In the other three reactors (N10), inoculum and 3 g of GAC were added to serum bottles ten days before undertaking the BMP test. After this time, nejayote was introduced in anoxic conditions. All of the tests were carried out for 30 days at 38 °C with an automatic agitation of 100 rpm [48]. All reactors were flushed with nitrogen in an anoxic chamber to avoid the presence of oxygen. Three blanks with 60 g of inoculum were tested to measure the inoculum methane potential. The methane production measurements are expressed at 0 °C and standard pressure of 760 mmHg (NCTP) in dry conditions.

Gas production measurement and numerical calculation was according to Valero et al. [49]. Headspace pressure was measured employing a digital pressure transducer (ifm Germany type PN78 up to 2 bars) with a syringe that was connected to pierce the septum. Statistical analysis was performed with the Analysis ToolPak in Microsoft Excel (Excel 2016 (v16.0)).

Once the BMP test was finished, COD, VFA, and redox potential were measured for all of the assays.

2.3. GAC Conductivity

GAC was introduced in PVC tubes, of specific lengths and known surface areas, to calculate its conductivity. A source meter (Keithley 2400, Cleveland, OH, USA) was used to measure the resistance of GAC. Once the resistance was determined, conductivity was calculated employing the following equations:

$$R = \rho \frac{A}{L} \quad (1)$$

$$K = \frac{1}{\rho} \quad (2)$$

where R is the resistance measured with Keithley 2400, A is the area of PVC tube, L is the tube length, ρ is the GAC resistivity, and K is the GAC conductivity. The process was undertaken in triplicate and the result was 3690 $\mu\text{S}\cdot\text{cm}^{-1}$ [50].

2.4. Analytical Methods

Biogas characterization was determined on a Molesieve column (30 m long, 0.53 mm internal diameter, and 0.25 μm film thickness) in a gas chromatograph (Clarus 500-Perkin Elmer, Waltham, MA, USA) with the thermal conductivity detector (TCD). Nitrogen was used as gas carrier and temperatures of 75, 30 and 200 °C were used for the injector, oven and detector, respectively. VFA were determined

by gas chromatography using (Clarus 500-Perkin Elmer, Waltham, MA, USA). The column employed was Agilent J&W (30 m long, 0.53 mm internal diameter) with a flame ionization detector (FID). Before determining VFA by chromatography, samples were filtered, acidified with phosphoric acid, and then centrifuged. Conductivity, temperature, pH, TSS, TS, and VS were analyzed, following standard methods [46]. Colorimetric methods (Hach Company DR-890, Loveland, CO, USA) were used to determine COD, TN, PO_4^{-3} , $\text{NH}_3\text{-N}$, and SO_4^{-2} . Redox potential was determined employing a redox potential sensor (Extech RE300, Nashua, NH, USA).

2.5. Microbial Community Analysis

After the BMP test, sludge from N, N0, and N10 and GAC from N0 and N10 was kept at $-80\text{ }^\circ\text{C}$ [27]. DNA was extracted from the five samples N sludge (S), N0 sludge (S0), N10 sludge (S10), GAC biofilm from N0 (C0), and GAC biofilm from N10 (C10). Three DNA extraction of each sample were undertaken, and then the DNA from each of samples were mixed before perform the sequencing. DNA from each sample was isolated with a DNA extraction kit ZymoBIOMICSTM DNA miniprep Kit, (Zymo Research, Irvine, CA, USA), in accordance with manufacturer's protocol. The DNA pellet was resuspended in 50 μL of TE buffer (10mM Tris-HCl (pH 8) and 1mM EDTA (pH 8)). DNA purity and DNA concentration were checked by employing a NanoDrop ND-1000 spectrophotometer (NanoDrop Technologies, Seattle, WA, USA). The A260/280 ratios were over 1.6 and DNA concentration in the range of 50–100 $\text{ng}\cdot\mu\text{L}^{-1}$. The primers employed for amplifying 16S rRNA genes (bacteria and archaea) were 341F and 805R. DNA was sent to Macrogen Inc. (Seoul, Korea) who performed sequencing on an Illumina MiSeq platform. MG-RAST software was used to analyze bacteria and archaeal communities through GREENGENES and RDP II databases [51].

2.6. Scanning Electron Microscopy (SEM)

GAC was analyzed by SEM (SEM, model JSM-6360LV, JEOL, Tokyo, Japan). In order to establish the existence of microbial communities attached to GAC, three samples of N10 were mounted on a metallic stub using double-sided adhesive tape, each being coated with a 15 nm gold layer and observed at 20 kV.

3. Results

3.1. Nejayote and Inoculum

Nejayote collected from industrial corn flour plant showed basic pH values of 10.2. The inoculum, developed in the laboratory, had a basic pH but with values that were lower than nejayote, in this case, the pH reached 8.1. The VS percentage obtained for nejayote was $0.85 \pm 0.01\%$, whereas inoculum values reached $2.69 \pm 0.01\%$. TS results were $1.22 \pm 0.01\%$ and $4.62 \pm 0.01\%$ for nejayote and inoculum, respectively. Nejayote and inoculum VS/TS ratios that were obtained were 0.69 and 0.55, respectively. TA concentration results were similar for nejayote and inoculum, reaching values of 1799 ± 116 and $1652 \pm 147\text{ mg CaCO}_3\text{ L}^{-1}$, respectively. COD concentration determined for nejayote was $15,433 \pm 827\text{ mg}\cdot\text{L}^{-1}$. TSS levels for nejayote of $2676 \pm 512\text{ mg}\cdot\text{L}^{-1}$ were recorded. Other parameters that were characterized for nejayote were $\text{NH}_3\text{-N}$ $4.6 \pm 0.1\text{ mg}\cdot\text{L}^{-1}$, TN $95 \pm 4\text{ mg}\cdot\text{L}^{-1}$, SO_4^{-2} $22 \pm 2\text{ mg}\cdot\text{L}^{-1}$, and PO_4^{-3} $59 \pm 1.2\text{ mg}\cdot\text{L}^{-1}$.

3.2. BMP Test

Methane production did not peak for the three assays N, N0, and N10 until the third day of BMP testing. The rate at which methane was produced spiked during the third day and it resulted in 39, 46, and 45 $\text{L CH}_4\text{ kg}^{-1}\text{VS day}^{-1}$ for N, N0, and N10, respectively. As shown in Figure 1A, the first significant difference between N10 reactors and N and N0 reactors, was on the fifth day (t-student $p < 0.05$). N10 reactors resulted in higher methane production levels of $26\text{ L CH}_4\text{ kg}^{-1}\text{VS day}^{-1}$, whilst N and N0 reactors produced 17 and 10 $\text{L CH}_4\text{ kg}^{-1}\text{VS day}^{-1}$, respectively.

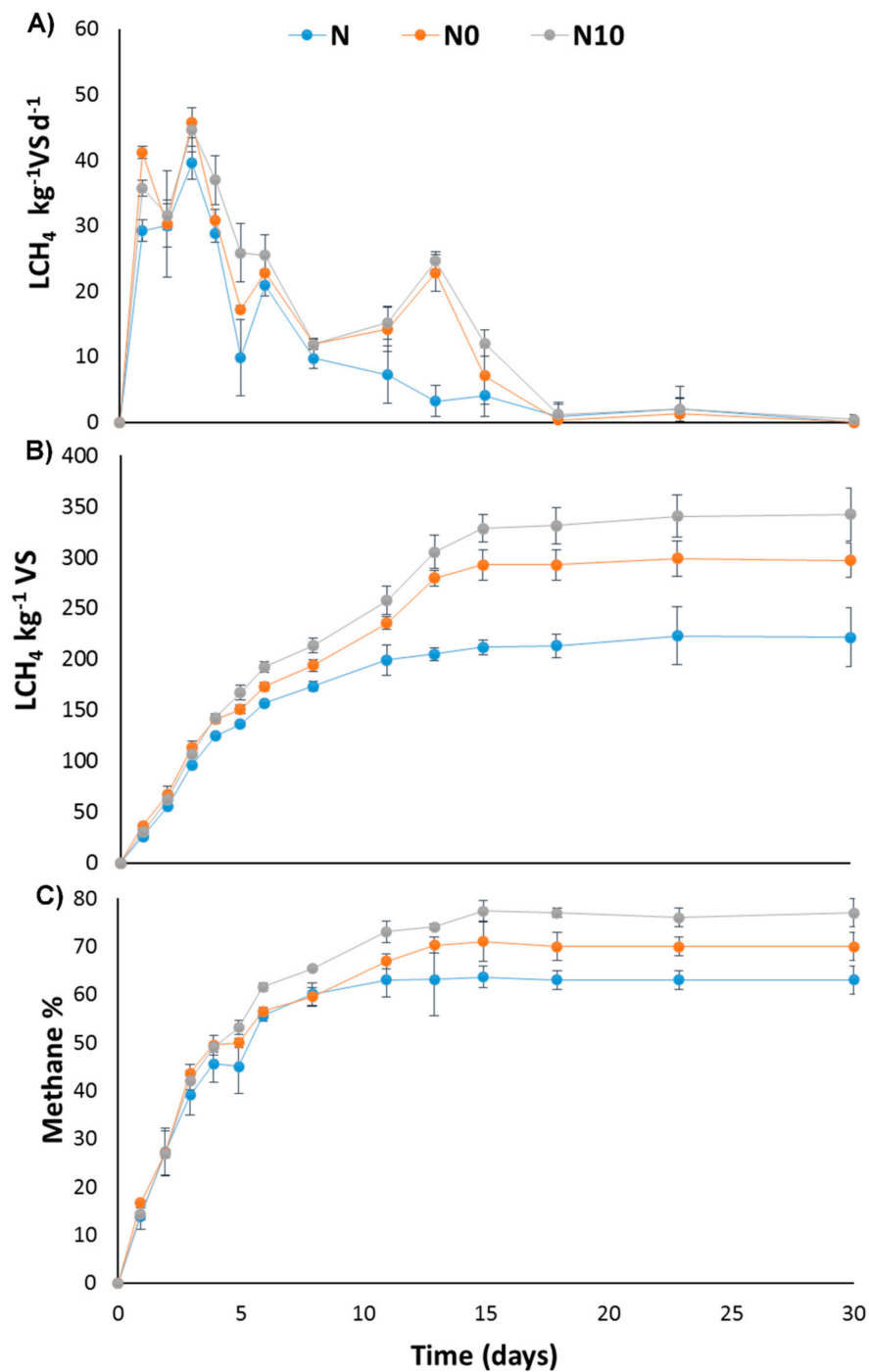


Figure 1. Biochemical Methane Potential (BMP) curves for control reactor (N), reactors with granular activated carbon (GAC) (N0) and reactors with biofilm GAC developed before undertaking the BMP test (N10). (A) Daily methane production (B) Cumulative methane production (C) methane biogas percentage. Mean values \pm SD from triplicate assays.

DIET or IHT occurs in the last step of the anaerobic digestion process, however adding GAC promotes DIET instead of IHT in N0 and N10 reactors [45]. The only compounds that can be oxidized by exoelectrogenic bacteria are alcohols and VFA [24,52]. When these compounds are available to stimulate the production of methane through DIET, a significant difference can be seen between DIET reactors through GAC and those reactors without conductive materials (t-student analyses $p < 0.05$). As shown in Figure 1A, from the tenth day until the fifteenth day of experimentation, N0 and N10

consistently produced methane over $10 \text{ L CH}_4 \text{ kg}^{-1}\text{VS day}^{-1}$, while N methane production was under $10 \text{ L CH}_4 \text{ kg}^{-1}\text{VS day}^{-1}$. Moreover, on one occasion during the same period, N0 and N10 actually reached more than $22 \text{ L CH}_4 \text{ kg}^{-1}\text{VS day}^{-1}$, while N did not produce values above $10 \text{ L CH}_4 \text{ kg}^{-1}\text{VS day}^{-1}$.

As highlighted in Figure 1B, the cumulative methane production results draws focus to the significant differences between N, N0, and N10 (t-student analysis $p < 0.05$). The final methane generation for N, N0 and N10 reached values of $222 \pm 23 \text{ L CH}_4 \text{ kg}^{-1}\text{VS}$, $297 \pm 10 \text{ L CH}_4 \text{ kg}^{-1}\text{VS}$ and $342 \pm 29 \text{ L CH}_4 \text{ kg}^{-1}\text{VS}$, respectively. N0 methane production was 34% higher than N, however the difference between N10 and N soared staggeringly to register an increase of 54%. Additionally, results show that the N10 methane volume generated was 15% higher than N0.

Methane production was predicted relative to biochemical oxygen demand (BOD) concentration. Nejayote BOD is around $5870 \pm 1900 \text{ mg L}^{-1}$ [3,5,53,54]. Theoretically, 1 kg of BOD produces 350 L methane [55]. Thus, the methane production from nejayote expected through anaerobic digestion process without DIET promotion was 73.96 mL CH_4 ($\text{C}_6\text{H}_{12}\text{O}_6$ (DBO) $\rightarrow 3 \text{ CO}_2 + 3 \text{ CH}_4$) [55]. In this study, real methane production in N reactors was $69.12 \pm 8.96 \text{ mL CH}_4$. Likewise, through the DIET process an increase between 20–33% in the production of methane has been reported due to a greater recovery of electrons in the reduction of CO_2 to CH_4 through VFA (acetate, propionate, and butyrate) and alcohol (ethanol) degradation [18,25,38,56–58]. According to these increments, the methane production promoting DIET expected was in the range of 88.75–98.36 mL CH_4 . In this study, real methane production for N0 and N10 was 92.71 ± 5.38 and $106.69 \pm 8.15 \text{ mL CH}_4$, respectively. These values are in the expected range.

The first difference in the biogas composition came after the sixth day, when N10 methane percentage was recorded as 5% higher than N and N0, as illustrated in Figure 1C. This difference remained constant between N10 and N0 throughout the BMP test.

Between days eleven and eighteen, N0 methane composition was 5% higher than N. Thus, N10 was 10% higher than N. The maximum methane concentration was reached on the fifteenth day with values of 64%, 71%, and 77% for N, N0, and N10, respectively. After the eighteenth day, there was no further biogas production in all of the reactors, and therefore the biogas composition plateaued to show similar levels of methane concentration in all reactors.

Once the BMP test was completed, the contents of the different reactors were analyzed to determine pH, redox potential, VFA, and COD, as shown in Table 1. There were no significant differences in pH values between the three assays. A huge reduction of redox potential was produced in reactors with GAC, with the measured value for N being recorded as $-222 \pm 7 \text{ mV}$, when compared with $-466 \pm 1 \text{ mV}$ and $-471 \pm 2 \text{ mV}$ for N0 and N10, respectively. VFA composition was formed by acetic acid $203 \text{ mg}\cdot\text{L}^{-1}$ and butyric acid $90 \text{ mg}\cdot\text{L}^{-1}$ in N reactors, while in N0 and N10 assays the VFA composition was made up solely of acetic acid because butyric acid was only present in N reactors. COD concentration was higher in N reactors than in N0 and N10, with values of $532 \pm 10 \text{ mg}\cdot\text{L}^{-1}$, $307 \pm 21 \text{ mg}\cdot\text{L}^{-1}$, and $218 \pm 14 \text{ mg}\cdot\text{L}^{-1}$, respectively.

Table 1. Reactors analysis after BMP test.

| Parameter | N | N0 | N10 |
|--|----------------|----------------|----------------|
| pH | 7.3 ± 0.11 | 7.2 ± 0.03 | 7.3 ± 0.02 |
| Redox potential (mV) | -223 ± 5 | -467 ± 1 | -471 ± 2 |
| Acetic acid ($\text{mg}\cdot\text{L}^{-1}$) | 203 ± 12 | 139 ± 9 | 109 ± 8 |
| Butyric acid ($\text{mg}\cdot\text{L}^{-1}$) | 91 ± 5 | | |
| COD ($\text{mg}\cdot\text{L}^{-1}$) | 532 ± 10 | 307 ± 21 | 218 ± 14 |

Control reactor (N), Reactor with granular activated carbon (N0), reactor with GAC added 10 days before undertaking biochemical methane potential test.

3.3. Microbial Community Analysis

As shown in Figure 2, the four main bacteria phylum in all of the samples were *Actinobacteria*, *Bacteroidetes*, *Firmicutes*, and *Proteobacteria*. *Firmicutes* was the dominant phylum in S, comprising almost 50% of the total bacteria. In sludges S0 and S10, there was a lower percentage of *Firmicutes*, forming 26% and 25% of the total bacteria, respectively. Furthermore, the percentage of *Firmicutes* was even lower in C0 and C10, measuring 15% and 16%, respectively. In contrast, *Bacteroidetes* follows the opposite trend to *Firmicutes*. *Bacteroidetes* had its highest percentage in C0 and C10, attached to the carbon, reaching 48% in C0 and 43% in C10, while in S it accounted for 18% of the microbial community. The *Proteobacteria* phylum presented a difference between the samples: it was detected at 17% in the C10 biofilm; whilst, for the other samples, it never reached 10%.

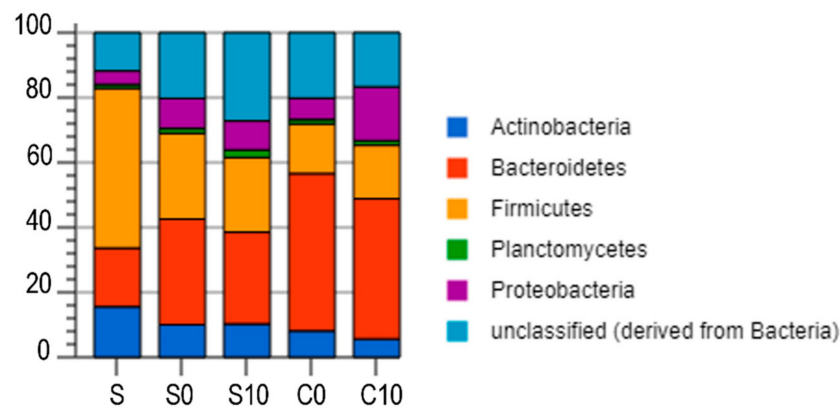


Figure 2. Bacteria phylum relative abundance. Phylum level with relative abundance lower than 1% were included in unclassified group. Sludge from control reactor (S), sludge from reactor with granular activated carbon (GAC) (S0), sludge from reactor with GAC biofilm developed before undertaking the BMP test (S10), GAC biofilm from reactor with GAC (C0), and GAC biofilm from reactor with GAC biofilm developed before undertaking the BMP test (C10).

Figure 3A illustrates the classification of bacteria by genus proportion. *Clostridium* anaerobic fermentative bacteria was highly developed in S, comprising 46% of the bacterial community, whereas it was less than 21% in S0 and S10, and interestingly, it accounted for 7% of the bacterial community attached to GAC in C0 and C10. *Parabacteroides* were developed in GAC biofilm and the sludge of GAC reactors with a relative abundance of 8%, 13%, 12%, and 14% for S0, S10, C0, and C10, respectively; however, there was no presence detected in S. *Parabacteroides* that is able to produce VFA as an end product of fermentation. This, in turn, can be oxidized by exoelectrogenic bacteria. Additionally, an enrichment of this genus in microbial fuel cells [31] and microbial electrolysis cells [32] has been observed.

Geobacter, an exoelectrogenic bacteria widely accepted as an exoelectrogen DIET partner [7,8], was only detected in the C10 sample, where the biofilm was developed 10 days before undertaking the BMP test. The amount of *Geobacter* in C10, whilst not dominant, remains a substantive 14% of the overall bacteria community and 82% of *Proteobacteria* phylum. The most dominant genus on the surface of both GAC samples C0 and C10 was *Prolixibacter*, with 31% and 21%, respectively. *Prolixibacter* had a relative abundance of 16%, 18%, and 11% in S, S0, and S10, respectively. It must be mentioned that *Prolixibacter* has been reported as a possible exoelectrogenic bacteria in microbial electrolysis cells [33] and in marine-sediment fuel cells [34].

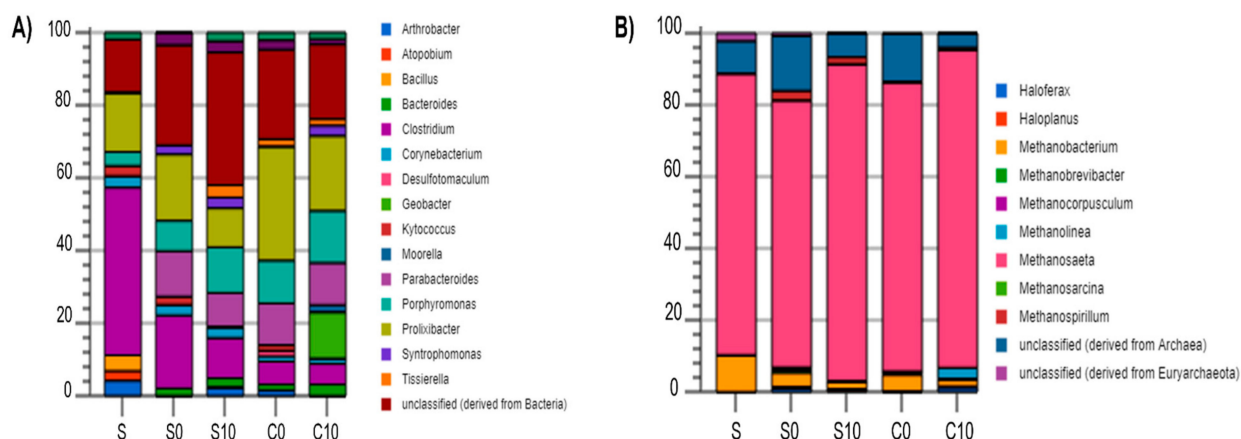


Figure 3. Bacteria (A) and Archaea community structure (B) at genus level. Genus level with relative abundance lower than 1% were included in unclassified groups. Sludge from control reactor (S), sludge from reactor with granular activated carbon (GAC) (S0), sludge from reactor with GAC biofilm developed before undertaking the BMP test (S10), GAC biofilm from reactor with GAC (C0) and GAC biofilm from reactor with GAC biofilm developed before undertaking the BMP test (C10).

A low enrichment for *Syntrophomonas* and *Bacteroides* was detected in the microbial community analysis. Both showed a relative abundance of 3% in C0 and C10, but contrastingly they were not detected in S. Both genera have only been reported as exoelectrogen DIET partners in studies with complex waste [19], as similar to nejayote.

In this study, the total amount of electroactive bacteria (*Geobacter*, *Proxilibacter*, *Bacteroides*, and *Syntrophomonas*) was 41%, 37%, 17%, 22%, and 16% for C10, C0, S10, S0, and S, respectively. This indicates an exoelectrogenic bacteria enrichment of 24% in C10 when compared with the control reactor, and an 18% higher electroactive bacteria community in C0 than S.

The archaeal community was dominated by *Methanosaeta* in all of the reactors, attached to the carbons and suspended in all sludges. It was most prevalent in C10, reaching 89% of the archaeal community, while in S, its percentage was 78%. *Methanosaeta* has been established as one of the main methanogens able to accept electrons in DIET [7]. *Methanolinea*, another methanogen associated with electrotrophic capacity in DIET [35], only appeared on the carbon surface of C10 with a relative abundance of less than 5% of the archaeal community.

The similarities of the microbial community structures in the five samples were analyzed at the genus level by hierarchically clustered heat analysis (Figure 4). The heatmap includes 17 genera. C10 is the sample with the highest concentration of different genera. The communities attached to GAC, C10, and C0 clustered together finally. These samples share a high similitude when compared with sludge samples (S, S0, S10).

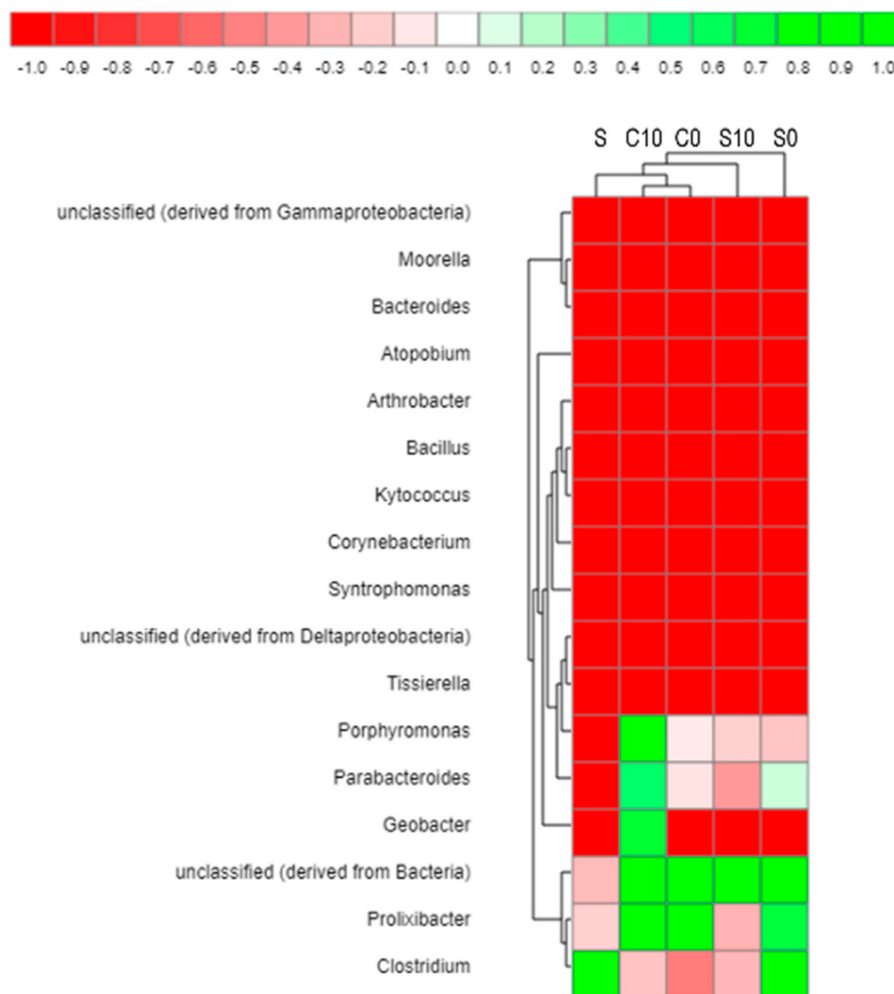


Figure 4. Heatmap of bacteria at genus level. Genus level with relative abundance lower than 1% were included in unclassified groups. Sludge from control reactor (S), sludge from reactor with granular activated carbon (GAC) (S0), sludge from reactor with GAC biofilm developed before undertaking the BMP test (S10), GAC biofilm from reactor with GAC (C0), and GAC biofilm from reactor with GAC biofilm developed before undertaking the BMP test (C10). Heatmap showing the 17 genera with significant difference of relative abundances among five samples. Heatmap is color-coded based scale (from red (−1) less abundance to green (+1) more abundance).

3.4. Scanning Electron Microscopy (SEM)

Through SEM, it was possible to examine the carbon surface and its porosity free of microbial communities before adding GAC to the reactors (Figure 5A). Additionally, microbial communities attached to GAC are clearly visible in Figure 5B. The exoelectrogen partner *Geobacter* was identified in the microbiological analysis attached to the GAC in sample C10 with a relative abundance of 14% in the bacterial community. *Geobacter* has been described as rod-shaped bacteria with a length between 2–4 μm [59]. In Figure 5C, bacteria can be visualized with the same morphological characteristics described above. This, coupled with the high percentage of *Geobacter* attached to GAC, makes it plausible to identify them in the Figure 5C. *Methanosaeta* is a rod-shaped archaea with average dimensions between 2 and 6 μm in length by single cell. Cells are enclosed inside an annular, striated sheathed structure, and they are separated by partitions forming a filamentous structure. *Methanosaeta* filaments can reach 100 μm of length [60]. In Figure 5D, it is possible to see *Methanosaeta* communities. Besides *Methanosaeta* was the dominant archaea in the microbial communities analysis, making its identification easy by SEM.

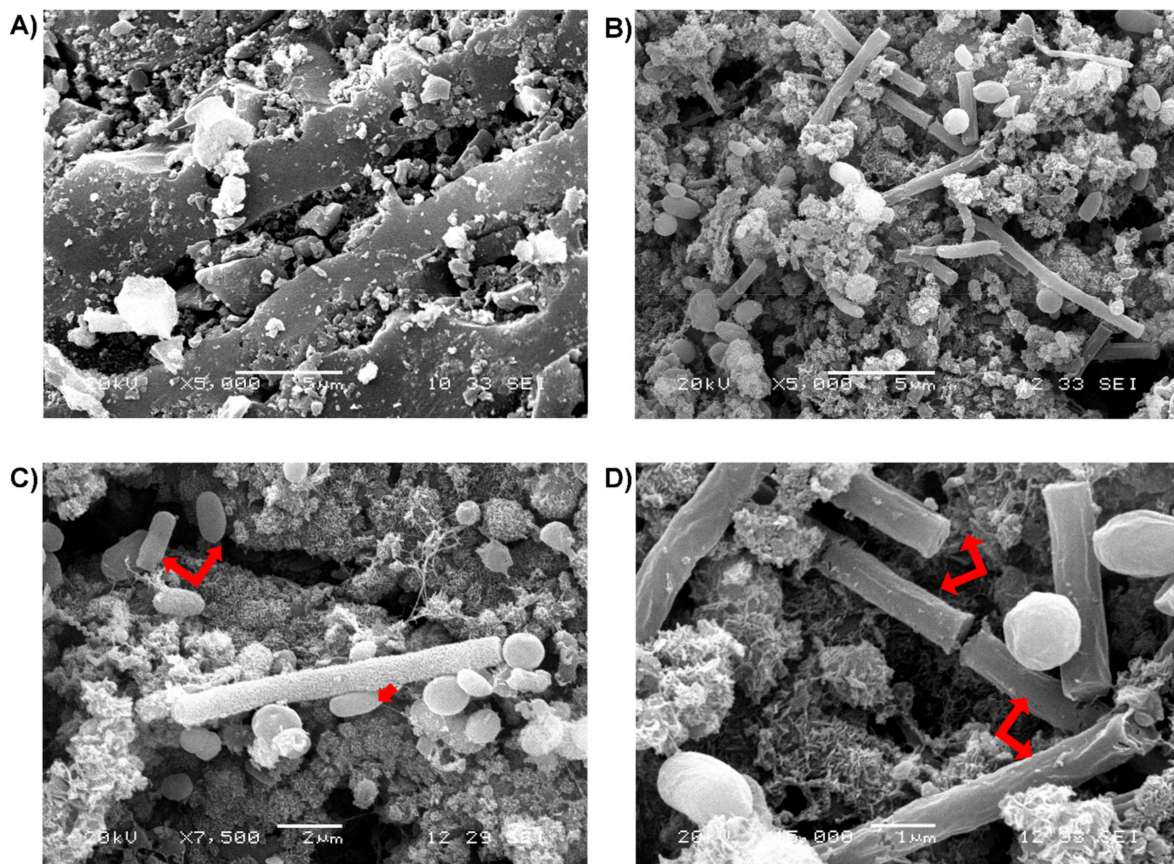


Figure 5. Scanning electron micrographs of the granular activated carbon surface (A), biofilm attached granular activated carbon (B), *Geobacter* communities attached to granular activated carbon (C), and *Methanosaeta* communities attached to granular activated carbon (D) after biochemical methane potential test (30 days).

4. Discussion

4.1. Nejayote and Inoculum

Table 2 shows the results of the physicochemical characterization of nejayote, as compared with the results that were obtained in other studies. As it is a wastewater that may come from an industrial process or from a more traditional process, which is not always done in the same way, the nejayote presents a high level of heterogeneity in the results of its composition, as seen in the wide range that is given for the majority of parameters. According to Ibarra-Mendivil et al. [61], different lime concentrations of 0%, 0.5%, and 1% were used for nixtamalization and pH values of 4.75, 7.72, and 11.01 were obtained, respectively. Small modifications in proportions of lime, cooking, and rest times, and different varieties of corn can give rise to notable variations in the composition of nejayote. The presence of lime in the process leads to the nejayote having a basic pH. This study is in the range expected with a value recorded of 10.2.

Table 2. Nejayote characterization compared with other studies.

| Parameter | Results | Other Studies |
|---|--------------|------------------------|
| pH | 10.2 | 6.3–11.6 [1,3–5,62,63] |
| COD (mg·L ⁻¹) | 15,433 ± 826 | 3430–40,058 [1,4,5,63] |
| NH ₃ -N (mg·L ⁻¹) | 4.65 ± 0.05 | 2 [1] |
| TN (mg·L ⁻¹) | 95.33 ± 3.7 | 209–428 [1,5] |
| SO ₄ ⁻² (mg·L ⁻¹) | 22.5 ± 2.5 | 13 [1] |
| PO ₄ ⁻³ (mg·L ⁻¹) | 58.75 ± 1.2 | 7.6–1321 [1,5] |
| TA (mgCaCO ₃ ·L ⁻¹) | 1799 ± 116 | 5768 [4] |
| TSS (mg·L ⁻¹) | 2676 ± 512 | 1810–8340 [3,4] |
| TS% | 1.22 ± 0.01 | 0.34–2.5 [1,4,63] |
| VS% | 0.84 ± 0.01 | 0.24–1.55 [1,63] |
| VS/TS | 0.69 ± 0.01 | 0.70 [63] |

The COD concentration of nejayote is within the range observed in the different studies (Table 2), with a value of $15,433 \pm 826 \text{ mg}\cdot\text{L}^{-1}$, coming mainly from the pericarp tissue of corn. The content in TN was $95.33 \pm 3.7 \text{ mg}\cdot\text{L}^{-1}$, lower than the composition analyzed in other studies, although this concentration is not sufficient to cause inhibition in anaerobic digestion. Concentrations between 1.7 and $14 \text{ g}\cdot\text{L}^{-1}$ cause a 50% reduction in methane production [64]. Both nitrogen and phosphorus are necessary for the good performance of an anaerobic reactor, since they are indispensable nutrients for bacteria. Concentrations lower than $0.3 \text{ mg}\cdot\text{L}^{-1}$ of phosphorus prevent the formation of the microbial community [65]. The nejayote that is characterized in this study has a concentration of $58.75 \pm 1.25 \text{ mg}\cdot\text{L}^{-1}$, which will aid a good performance in the BMP test. The concentration of SO₄ is $22.5 \pm 2.5 \text{ mg}\cdot\text{L}^{-1}$, which is a little higher than the previously reported $13 \text{ mg}\cdot\text{L}^{-1}$ (Table 2). The COD/SO₄ ratio is 686, so the presence of sulfur-reducing products and possible inhibition effects in the anaerobic digestion process due to the byproducts generated by them will be non-existent [66]. The percentages in both VS and TS are within the range referenced in Table 2. The values are $0.85 \pm 0.01\%$ VS and $1.22 \pm 0.01\%$ TS.

The inoculum has an alkalinity of $1652.5 \pm 147 \text{ mg CaCO}_3\cdot\text{L}^{-1}$. This contribution of alkalinity to the medium will absorb possible acidification processes that occur during anaerobic digestion due to the formation of VFAs [67]. The VS percentage of the inoculum is $2.67\% \pm 0.1$. This percentage aids the good performance in the anaerobic digestion process [68].

4.2. BMP Test

The results of this study clearly demonstrate that generating a previous GAC biofilm before starting the wastewater anaerobic digestion enhances methane biogas production and results in an overall improved anaerobic digestion performance. The syntrophic relationship between exoelectrogenic bacteria and electrotrophic archaea is essential to enable DIET. Recently, it has been widely recognised that exoelectrogenic bacteria can only oxidize alcohols and VFA to send electrons to reduce carbon dioxide to methane within DIET [24,52]. The advantages of promoting DIET instead of IHT in methane production yield do not become obviously apparent until there are increased alcohols and VFA concentrations and they are available to be oxidized by exoelectrogenic bacteria. While examining the results of the study, it is important to state that the first difference between control reactors and GAC reactors in methane production did not appear until the seventh day in N0 and N10 (Figure 1A), due to the insufficient concentration of VFA and alcohols, which are available in the medium to be oxidized. In previous studies undertaken in batch assays with conductive materials, the difference in methane production emerges earlier due to the substrate employed. In these studies, the carbon sources used were synthetic wastewater, which has alcohols and VFA readily available from the beginning of experimentation [36,38]. Nejayote, on the other hand, is a complex waste that requires hydrolysis and acidogenesis steps before VFA and alcohols are available, which explains the delay in DIET.

After the BMP test was concluded, the COD concentration was higher in N reactors than N0 and N10 reactors. This fact explains part of the difference in accumulated methane production. Butyrate was detected in N medium, whilst there was no presence of butyrate in N0 and N10 reactors. The total VFA concentration of N was more than double that of N0 and N10, which can, in part, go some way towards explaining the difference between the methane potentials in the tests; however, it can not be held accountable for all of the differences. These differences in methane yield could also be due to some remnant organic matter that has not been yet converted in VFA in N reactors [49].

In Table 3, the results are compared with studies that were developed with conductive materials. It is noted in Table 3 that the percentage in the increase of methane production is within plausible values. An increase of 54% when the biofilm is previously generated in GAC and 34% without the previous formation of GAC biofilm are reasonable results.

Table 3. Methane production increase with conductive materials.

| Reactor Volume (mL) | Substrate | Conductive Material | Methane Production Increase (%) | Time (Days) | Reference |
|---------------------|------------------------|---------------------|---------------------------------|-------------|------------|
| 120 | Nejayote | GAC | 54 | 30 | This study |
| 120 | Nejayote | GAC | 34 | 30 | This study |
| 250 | Ethanol | Graphene | 25 | 12 | [38] |
| 500 | Synthetic wastewater | GAC | 86 | 43 | [17] |
| 120 | Synthetic wastewater | Magnetite | 32 | 20 | [56] |
| 250 | Glucose | Graphene | 51 | 15 | [36] |
| 250 | Sludge treatment plant | GAC | 17 | 20 | [33] |

4.3. GAC Conductivity

DIET was stimulated by the addition of GAC in the reactors. GAC works as a support for microorganism communities and as electron conductor between exoelectrogenic bacteria and electrotrophic archaea. DIET using stainless steel as conductive support had a kinetic advantage that is 108 times greater than IHT to compete for the electron donor [69]. Stainless steel conductivity is $667 \mu\text{S}\cdot\text{cm}^{-1}$, while the GAC conductivity employed in this study was $3600 \mu\text{S}\cdot\text{cm}^{-1}$. Similar conductive materials employed before, such as magnetite [56] and biochar [30], possessed conductivities of $160 \mu\text{S}\cdot\text{cm}^{-1}$ and $5 \mu\text{S}\cdot\text{cm}^{-1}$, respectively. GAC has substantially higher values in terms of conductivity; making it an ideal material to use given its surface area, its high conductivity, and it is economically low cost [28,29].

4.4. Redox Potential

One of the most important effects, which GAC causes in the improvement of the methanogenic performance, is the change that it produces in redox potential. Redox potential must be under -200 mV for methanogenic activity to be possible. When redox potential is higher than -200 mV , this activity is virtually negligible [70]. All of the reactors in this study were under this barrier. As shown in the results, redox potential changed from $-222 \pm 7 \text{ mV}$ to $-471 \pm 2 \text{ mV}$ from control reactors N to N0 and N10 reactors. GAC had a direct effect on reducing redox potential by generating a reductive atmosphere, which is supported by evidence in other studies [71] where carbon nanotubes caused a reduction in redox potential in an abiotic environment. The potential energy difference (ΔE^0) between the electron donor and electron acceptor is directly proportional to the energy (ATP) that a microbe could employ from the metabolic process [72]. Consequently, a high reducing micro-environment, produced by the GAC presence in the anaerobic reactor, expedites electron donation to provide a thermodynamic driving force for accomplishing the electrophilic attack in terms of reduction process, increasing carbon dioxide reduction to methane, which could be one of the key points for improving the reactor performance [71].

4.5. Relationship Between Redox Potential, Methane Production and Archaea Communities

There was evidence that low redox potential stimulated the syntrophic relationship between exoelectrogenic bacteria and electrotrophic methanogens, resulting in elevated levels in methane concentration, which can be observed in Figure 1C. A previous study observed that a pure methanogenic archaeal culture with carbon materials had higher methane yield than a control reactor without carbon material, due to lower redox potential [71]. That means that, without an exoelectrogenic partner to send electrons through DIET to electrotrophic methanogens, methane yield is improved due to redox potential reduction. In a pure culture of hydrogenotrophic methanogens, the redox potential was under -450 mV, causing an increase in its population [70]. This change may be a result of the fact that it requires less thermodynamic energy to undertake an electrophilic attack by these methanogens to reduce carbon dioxide to methane. The low redox potential that was obtained in this study (-471 mV), by adding GAC, caused an increase of 10% in the acetoclastic methanogen (*Methanosaeta*) population.

4.6. Microbial Analysis

Only *Methanosaeta* and *Methanosarcina* are able to generate methane from acetate [73]. Additionally, *Methanosaeta* is recognised as an electron acceptor partner in DIET, reducing carbon dioxide to methane, hence in the control reactor, *Methanosaeta* played a vital role as acetate consumer to produce methane. However, in N0 and N10 reactors, due to its attachment to the GAC surface, *Methanosaeta* worked as an electron acceptor to reduce carbon dioxide to methane. *Methanobacterium* is a hydrogenotrophic archaea that tended to be in suspended solution instead of being attached to the carbon surface [74], although *Methanobacterium* has been detected in conductive support as an electron acceptor in DIET [44]. *Methanobacterium* had a relative abundance over 10% in the N reactor, whereas when detected in the sludges of N0 and N10, it was consistently under 5%. In this study, *Methanobacterium* was responsible for reducing hydrogen partial pressure through IHT in control reactor N. This is because *Methanobacterium* needs to be attached to conductive material to accept electrons. *Methanolinea*, which is a hydrogenotrophic archaea, was only detected in C10 samples with a proportion of 3%. *Methanolinea* and *Methanobacterium* can accept electrons when they are attached to a conductive support; therefore, they are able to work as electrotrophic archaea in DIET [17]. In this study, the two different hydrogenotrophic archaea played distinct roles: one (*Methanobacterium*) was suspended in the bulk and participated in IHT by controlling hydrogen partial pressure, and other (*Methanolinea*) was attached to the GAC surface and accepted electrons to reduce carbon dioxide to methane.

A variety of electroactive bacteria were enriched in the GAC surface; *Geobacter*, *Bacteroides*, and *Syntrophomonas*. All of these bacteria genera have been acknowledged as participants in DIET as electron donors [40,58]. *Syntrophomonas* and *Bacteroides* were detected with a relative abundance of 2% and 3% for C0 and C10, respectively. However, there was no presence of *Syntrophomonas* and *Bacteroides* in N reactors. Previous studies have shown that *Bacteroides* have the potential to grow significantly on the anode of a bioelectrochemical system and on the surface of carbon cloth in anaerobic reactors. This genus is able to extracellularly transfer electrons to ferric iron and donate electrons to electrotrophic methanogens via DIET [40,75]. There is no evidence that *Syntrophomonas* are able to transport electrons extracellularly, but notable growth has been reported when acting as a DIET partner in anaerobic systems [16].

Geobacter were only detected in C10 with a relative abundance of 13%. It must be concluded that this was because the biofilm was developed before introducing nejayote into the reactor and *Geobacter* could then grow under favorable conditions. The absence of *Geobacter* in C0 can be explained by the fact that *Geobacter* cannot efficiently degrade complex organic waste [76]. An additional supposition is that *Geobacter* is not viable in complex waste under hard conditions, as relative high salinity environment [27,40]. *Geobacter* enrichment was produced during the ten days before undertaking the BMP test in N10 reactors. This enrichment was not possible in N0 reactors due to the fact that GAC

was added to the reactors at the same time as nejayote, and other exoelectrogenic bacteria coming from nejayote or inoculum, overtaking *Geobacter* and adapting better to the new environmental conditions.

Other enrichment produced in the BMP test involved *Prolixibacter*. This genus has not been reported before in any study that is related with DIET in anaerobic digestion as exoelectrogenic bacteria. The capacity of *Prolixibacter* to transport electrons extracellularly has not yet been concluded, but it has been reported as a fermentative bacterium in sediment fuel cells that could be able to transfer electrons extracellularly [77]. In addition, its enrichment in microbial electrolysis cells has been reported [78].

The GAC biofilm developed previously in N10 reactors had two positive effects in the exoelectrogenic bacteria community. C10 exoelectrogenic community was 4% higher than the C0 exoelectrogenic community and the *Geobacter* genus only appeared in C10 samples.

Microbial analysis has shown a relative abundance of 12% and 14% of *Parabacteroides* attached to N0 GAC and N10 GAC, respectively, while it was not detected in the N reactor. This genus is known for its capacity to degrade polysaccharides to acetate [79]. The main function of these fermentative bacteria was to convert organic matter to compounds that were available to be oxidized by exoelectrogenic bacteria. *Parabacteroides* played an important role in DIET, producing the essential VFA required by the exoelectrogenic bacteria to degrade and donate electrons to electrotrophic methanogens. *Parabacteroidetes* growth has been reported in microbial fuel cells [80], assuming the same role as in this study, however it has not been reported in studies that are related with the promotion of DIET in anaerobic reactors. This is a result of the fact that most of these studies employ synthetic wastewater that is easily converted to methane. One of the difficulties previously detected in DIET was the hydrolysis and acidogenesis limit of VFA and alcohol concentration that was required to take advantage of DIET instead of IHT [52]. *Parabacteroidetes* growth in the conductive material greatly assists in resolving this issue, by generating the crucial carbon source for exoelectrogenic bacteria to oxidize. The electroactive community in C10 was 8% higher than in C0 and 38% higher than S.

5. Conclusions

GAC addition had a positive impact on the BMP test. GAC had four main resulting effects on the anaerobic digestion process; increased methane production, redox potential reduction, DIET promotion, and electroactive biofilm development. A low redox potential made an electrotrophic attack to reduce carbon dioxide to methane easier. Additionally, DIET was promoted instead of IHT resulting in higher methane concentration and greater methane production. Furthermore, the electroactive community population intensified when the biofilm was developed 10 days before the assay, and the reactors with previous biofilm obtained a better yield than those without. This study was the first step to a better understanding of the relationship between DIET promotion and a complex wastewater, such as nejayote. Undertaking a study using bigger reactors is necessary to further support these promising results, in order to fully realise the potential of testing these conclusions on a large scale.

Author Contributions: D.V. design of experiments, performed physico-chemical parameter analysis and BMP tests, participated in the evaluation of microorganisms communities and drafted the manuscript. C.R. helped co-ordinate the project and revised the manuscript. B.C.-C. developed the DNA extraction protocol, participated in the evaluation of microorganisms communities and revised the manuscript. J.A.D.-M. performed VFA analysis and participated in physico-chemical parameter analysis and BMP test. R.T.-T. developed the DNA extraction protocol and performed MG RAST analysis. A.C.-V. participated in developing the DNA extraction protocol and performed DNA extraction. L.A.-G. designed and coordinated this work alongside helping to draft the manuscript. All authors read and approved the final manuscript.

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