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1 **Cavity-Enhanced Absorption Detection of H₂S in the Near Infrared Using a**
2 **Gain-switched Frequency Comb Laser**

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11
12 **Abstract**

13 A custom-designed gain-switched frequency comb laser was passively coupled of to a medium
14 finesse cavity in the region between 6346 cm⁻¹ and 6354 cm⁻¹ for the development of a prototype
15 cavity enhanced absorption setup. The setup was applied to static gas detection of hydrogen sulfide
16 at the parts per thousand level in a laboratory environment. A Fourier transform spectrometer was
17 used for signal detection. The experimental performance of the setup was characterized in this
18 proof-of-principle investigation; advantages, drawbacks and future scope of the approach are
19 discussed in this article.

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29 **1. Introduction**

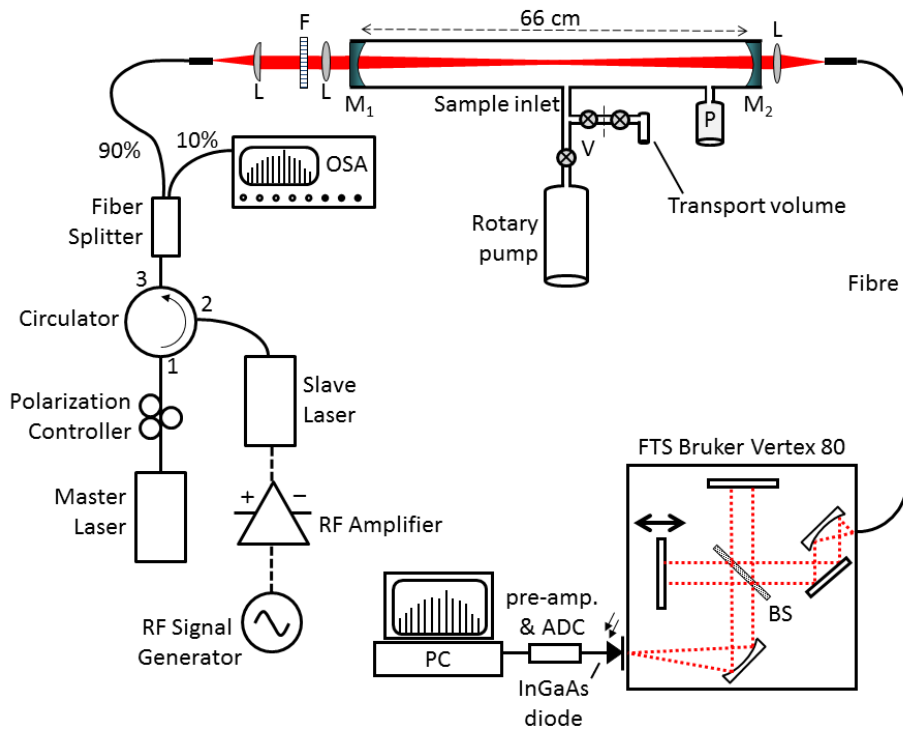
30 Among the many applications of frequency combs (FC) in metrology (e.g. optical clocks and low
31 noise frequency synthesis) [1-3], combs have long been recognized as superior light sources for
32 absorption spectroscopy [4-6]. FC lasers generate a large stable array of equally spaced, phase-
33 coherent, narrow-band spectral lines. In excess of 10^5 comb lines can be produced with mode-
34 locked short pulse lasers, with line spacing in the region of many 10s of MHz (depending on the
35 comb laser's repetition rate [1]). The comb lines' narrow bandwidth is determined by the
36 coherence time of the laser.

37 In many cases the spectral resolution in absorption experiments is however limited by the
38 spectrometer used for detection. In these cases FCs are essentially broadband lasers which act as
39 a "spectral ruler" [1]. To exploit the inherently high resolution of FCs, new detection schemes
40 using virtually imaged phase arrays spectrometers [7-9], comb-cavity vernier approaches [10,11]
41 and Fourier-transform based dual comb spectroscopy [12-17] have been established. Some of these
42 methods were recently reviewed [18]. To enhance the sensitivity in absorption experiments,
43 applications of FCs in conjunction with optically stable cavities covering a spectral range that is
44 limited by the high reflectivity range of the cavity mirrors have been developed [19,20]. The key
45 challenge of cavity-enhanced FC absorption spectroscopy is the coupling of the FC to the cavity
46 by matching the comb lines to an appropriately designed mode structure of a high finesse cavity –
47 different approaches to coupling and locking of FC lines to high finesse optical cavities have also
48 been reviewed [21].

49 In contrast to developing ultimate instrumental limits, the approach to cavity-enhanced absorption
50 detection in this work aims at experimental simplicity. The principle suitability of a custom-built
51 gain-switched frequency comb (GSFC) laser in conjunction with a very modest optical cavity was
52 examined without spectrally stabilizing or mode-matching the cavity. GSFCs are tunable FCs that
53 are simple in design and significantly cheaper than mode locked femtosecond FC lasers [22]. They,
54 however, do not deliver the finely spaced large number of comb lines and span a significantly
55 smaller wavelength region (~typically a few nm). GSFCs are generated by the direct modulation
56 of commercially available laser diodes, as a distributed feedback laser, with a radiofrequency (RF)
57 sine wave. External optical injection in a master-slave laser configuration can be employed to
58 generate narrow line widths and low intensity noise [23]. GSFCs can also be integrated as a

59 monolithic devices [24] (see conclusion & outlook). The approach is simple and cost-efficient,
 60 and enables wavelength tunability of the comb and the line spacing, but with a rather limited
 61 number of comb lines [22]. GSFC lasers are commonly used as benchtop units for optical
 62 telecommunication in the C-band [22]. Recently a dual comb GSFC setup was reported based on
 63 gain switching and optical injection locking [25], with potential applications of the dual comb
 64 architecture in the mid-IR or near UV. In Ref [25] the absorption feature of $H^{13}CN$ in the $1.5 \mu m$
 65 region were reported.

66 The objective of this study was to simply tune more than one laser comb line to absorption features
 67 of an appropriate target gas (hydrogen sulfide, H_2S) and to achieve sensitivities below the
 68 combustion limit of the hazardous gas in the parts per thousand (ppTv) regime using a low finesse
 69 ($f \sim 450$) optical cavity without a mode matching schemes.



70

71 **Fig. 1** Schematic of the experimental setup. Mirrors M_1 and M_2 are high-reflectivity dielectric mirrors ($R \sim 0.993$)
 72 forming a medium finesse optical cavity with a length of 66 cm. L: Lenses used for collimation and light collection.
 73 P: Absolute Pressure gauge. V: Valves. RF: Radio Frequency (~ 10 GHz). ADC: Analog-to-Digital Converter. OSA:
 74 Optical Spectrum Analyzer.

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78 2. Experiment

79 The schematic of the experimental setup is shown in Fig. 1. The main components of the GSFC
80 laser were a tunable master laser (HP-8168F), a distributed feedback slave laser (Thorlabs, ITC-
81 502), an radio frequency (RF) signal generator (HP-83751A), a biased RF amplifier (Aldetec,
82 ALP-0618S430), a polarization controller, and a circulator. The slave laser was kept at a stable
83 operating temperature by employing a thermoelectric cooler and a thermistor in the laser package.
84 The (tunable) frequency of the master laser was slightly detuned (to higher energies) from that of
85 the slave laser. Polarization controlled light from the master laser was injected via a circulator
86 (ports 1 & 2) into the slave laser. The optical injection power was ≈ 3.1 mW (5 dBm). For a DC
87 bias current of 45.1 mA the gain switching was achieved by continuously driving the slave laser
88 above and below its laser threshold using an amplified sinusoidal RF signal. This created a train
89 of pulses corresponding to an optical comb in the frequency domain. The (tunable) frequency of
90 the RF generator hence determined the free spectral range of the FC, which in the present
91 experiment was typically maintained at ~ 10 GHz (see Fig. 3). The absolute center frequency of
92 the GSFC was controlled through the frequency of the master laser, which was tunable external
93 cavity diode laser. In this experiment wavelength of the master laser was typically maintained at
94 ~ 1574.8 nm (~ 6349.4 cm^{-1}); cp. Fig. 3. Overall the GSFC produced twenty phase-coherent,
95 equally spaced comb lines with an individual spectral width of ~ 300 kHz [23] covering a range
96 from 6346 to 6354 cm^{-1} (1575.8...1573.8 nm) in the near infra-red with an overall quasi-
97 continuous integrated power close to ≈ 4 mW (6 dBm). For monitoring the GSFC emerging from
98 port 3 of the circulator (Fig. 1), the light was split into two beams (ratio 90:10) and the less intense
99 fraction of the beam was guided to an optical spectrum analyzer (OSA, Advantest Q8384). The
100 main beam (90%) was collimated with a plano-convex ($f \sim 25$ mm) and a biconvex lens ($f \sim 75$
101 mm). The collimated light was guided into a stable optical cavity consisting of two plano-concave
102 dielectric mirrors (diameter 25 mm, $r = -200$ cm, Layertec GmbH) at a distance of $d = 66.0 \pm 0.1$
103 cm. The modest reflectivity of the cavity mirrors of $R = 0.993 \pm 0.001$ was measured with a double-
104 beam UV/Vis/NIR spectrometer (PerkinElmer, Lambda-1050). The optical cavity was enclosed
105 by a stainless steel vacuum tube (diameter ~ 40 mm) which was fitted with an access port for
106 evacuation and sample injection as schematically shown in Fig. 1. Before experiments the static
107 gas cavity was evacuated by a rotary pump to ~ 0.1 mbar. The light exiting the cavity was collected
108 with an achromatic lens and focused into a multimode light-guide (diameter 1.5 mm) which was

109 connected to the entrance aperture of a Fourier transform spectrometer (FTS, Vertex 80 with CaF₂
110 beam splitter, Bruker Ltd.). In the spectrometer an InGaAs photodiode was used to detect the light
111 transmitted by the cavity. Using the Bruker OPUS software spectra were recorded with a spectral
112 resolution of 0.08 cm⁻¹ in an integration time of 120 s (corresponding to 4 scans per spectrum).
113 This integration time was based on the optimum measured signal-to-noise ratio of the FC,
114 established in a series of intensity stability measurements using the FTS and OSA.

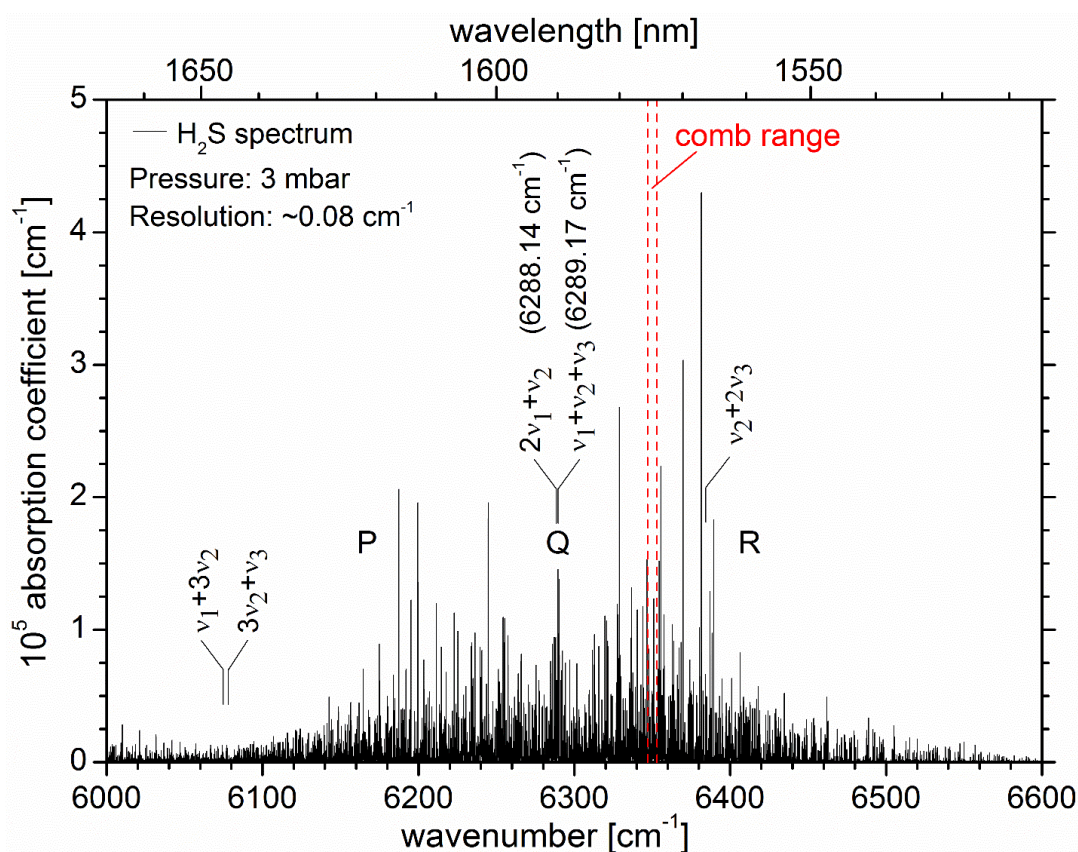
115 Hydrogen sulfide (H₂S) (BOC Industrial Gases, >99.9%) was used as target gas in this proof-of-
116 principle experiment. It is a poisonous and corrosive gas that can occur as a pollutant in industrial
117 settings, such as oil refineries, production facilities involving petro-chemistry, gas processing
118 plants and gas fields. Since it is heavier than air it can accumulate in badly ventilated areas and
119 constitute a fire or explosion hazard due to its flammable nature [26]. The main source of H₂S as
120 a natural atmospheric pollutant is from volcanic activity. The changes in the H₂S release rate from
121 dormant volcanos in comparison to other trace species is used to monitor volcanic activity before
122 potential eruptions [26]. Thus the spectroscopic detection of H₂S in industrial and environmental
123 applications is of high relevance and motivated its usage in this GSFC application. High detection
124 sensitivity was however not expected a priori, due to the comparatively small absorption cross-
125 sections of H₂S in the spectral target region at ~6350 cm⁻¹ and the modest mirror reflectivity.

126 **3. Results and Discussion**

127 **3.1 Near IR overview spectrum of H₂S using FT-IBBCEAS**

128 Before matching frequency comb lines with absorption features of H₂S, its near IR spectrum was
129 measured in the region between 6000 and 6600 cm⁻¹ using Fourier-transform incoherent broad-
130 band cavity enhanced absorption spectroscopy (FT-IBBCEAS) [27-29]. FT-IBBCEAS is an
131 approach to measure broadband cavity-enhanced spectra in the NIR region with high resolution
132 [30-32]. The spectrum was measured at a static H₂S pressure of ca. 3 mbar with an integration time
133 of 60 min and a spectral resolution of 0.08 cm⁻¹ (cavity length 644 cm, mirror reflectivity $R \approx$
134 0.9998); for more details concerning the experimental setup see Ref [30]. The line positions
135 obtained agreed with those in HITRAN [33] and the Pacific Northwest National Laboratory
136 (PNNL) data base [34] to within the resolution of the instrument. Fig. 2 shows the overview
137 spectrum of the 2.5 polyad of H₂S (asymmetric top, point group C_{2v}) containing features of 5
138 combination overtone bands $\nu_1+3\nu_2$, $3\nu_2+\nu_3$, $2\nu_1+\nu_2$, $\nu_1+\nu_2+\nu_3$, and $\nu_2+2\nu_3$. The spectrum is subject

139 to resonance interactions between all states in this polyad. The same region was spectroscopically
 140 studied previously by Ulenikov et al. [35]. The $\nu_1+\nu_2+\nu_3$ (asymmetric) and $2\nu_1+\nu_2$ (symmetric)
 141 bands, centered at $\sim 6289.17\text{ cm}^{-1}$ and $\sim 6288.14\text{ cm}^{-1}$ exhibits, are the strongest combination bands
 142 with clearly visible and overlapping P, Q and R ro-vibrational components; cf. Fig. 2. The bands
 143 $\nu_1+3\nu_2$ and $3\nu_2+\nu_3$, are significantly weaker and especially the very weak combination overtone
 144 $\nu_2+2\nu_3$ is negligible for detection applications. A transition in the R branch at $\sim 6339.868\text{ cm}^{-1}$ was
 145 used before by Modugno et al. [36] for detection of H_2S at the parts per million level ($\sim 4\text{ ppmv}$)
 146 using two-tone frequency modulation spectroscopy. More recently also Chen et al. [26] used R-
 147 branch lines of $\nu_1+\nu_2+\nu_3$ at $\sim 6362.88\text{ cm}^{-1}$ for off-axis integrated cavity output spectroscopy.



148
 149 **Fig. 2** The near IR absorption spectrum of H_2S at 3 mbar recorded using FT-IBBCEAS with a resolution 0.08 cm^{-1} .
 150 The region between 6000 and 6600 cm^{-1} is part of the $\nu = 2.5$ polyad. The two dashed vertical lines (red) in the R-
 151 branch of the $\nu_1+\nu_2+\nu_3$ and $2\nu_1+\nu_2$ combination bands of H_2S indicate the spectral region where the GSFC (20 lines)
 152 is located. Other band centers of weaker overtones in the region are also indicated.

153

154

155

156 3.2 GSFC application to cavity-enhanced H₂S detection around 6350 cm⁻¹

157 Owing to the limited tunability of the GSFC, ro-vibrational R-branch transitions of the $\nu_1+\nu_2+\nu_3$
 158 combination band were used in this study to detect H₂S in the region around 6350 cm⁻¹, i.e. in the
 159 spectral region indicated by two vertical red lines in Fig. 2. Matching comb lines with H₂S
 160 absorption features turned out to be cumbersome as the FWHM of the absorption lines (as well as
 161 the FC line widths) are smaller than the spectral resolution of the detection setup.

162 The HITRAN database [33] gives the H₂S FWHM self-broadening and air-broadening coefficients
 163 at 296 K as ~ 0.316 cm⁻¹ atm⁻¹ and ~ 0.148 cm⁻¹ atm⁻¹, respectively (with negligible pressure shift).
 164 Since the overview spectrum in Fig. 2 was measured at ~ 3 mbar of H₂S at room temperature,
 165 pressure broadening was negligible (< 0.001 cm⁻¹) and the expected FWHM due to Doppler
 166 broadening at 296 K was $f_G \sim 0.013$ cm⁻¹. Thus the observed FWHM of the absorption lines in Fig.
 167 2 were limited by the resolution of the FTS of 0.08 cm⁻¹.

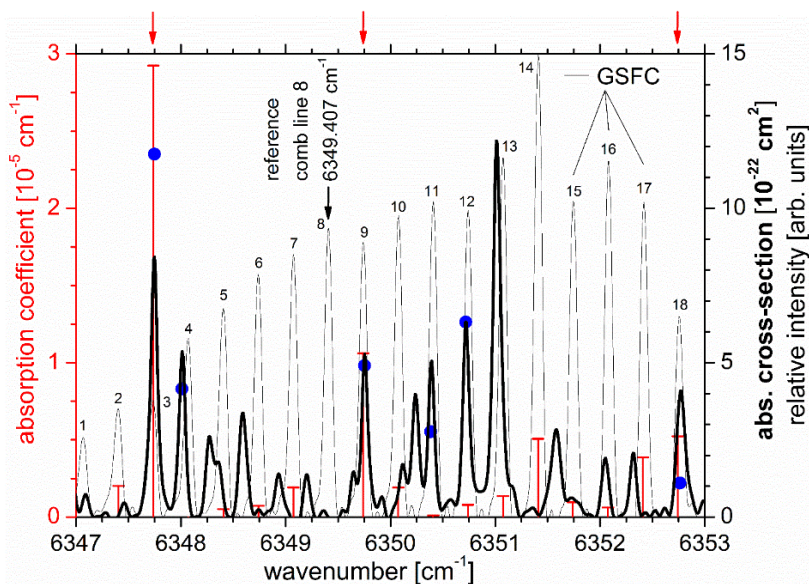
168 The situation was similar for H₂S detection with the GSFC in the static gas, where the same FTS
 169 was used in conjunction with the short optical cavity. The GSFC experiments were performed at
 170 ~ 82 mbar total pressure where the expected FWHM of absorption lines due to pressure broadening
 171 was $f_L \sim 0.012$ cm⁻¹ [33] (see also discussion below)^{#1}. The value of f_L is approximately the same
 172 size as the expected FWHM Doppler limit of 0.013 cm⁻¹ leading to an anticipated Voigt FWHM
 173 of $f_v \approx 0.5346 f_L + \sqrt{0.2166 f_L^2 + f_G^2} = 0.021$ cm⁻¹ [37]. Even though the comb lines possess a
 174 small band width of only $\approx 10^{-5}$ cm⁻¹ [23], and even though the absolute comb position could be
 175 finely tuned in relative steps of ~ 0.004 cm⁻¹, it was very cumbersome to overlap of comb lines
 176 with H₂S absorption lines, because all cavity transmission spectra were measured with the limited
 177 spectral resolution of the FTS of 0.08 cm⁻¹ (i.e. approximately 3-4 times the FWHM of the

^{#1}The true the mixing ratio in the cavity was 0.92 ppTv and hence the partial pressure of H₂S in the cavity was ca. 1 mbar (as will be shown below) and therefore the solely pressure broadened FWHM was expected to be

$$f_L(p) \approx \gamma_{\text{air}} (1 \text{ atm}) [p - p_{\text{self}}] + \gamma_{\text{self}} (1 \text{ atm}) p_{\text{self}} \approx 0.012 \text{ cm}^{-1}. \quad (2)$$

based on HITRAN broadening coefficients mentioned in the text.

178 expected absorption lines).^{#2} Therefore software was written that identifies the required FSR and
 179 optimum GSFC center position (arbitrarily chosen as the position of line 8 – see Fig. 3) to match
 180 as many comb lines as possible with absorption features in the relevant spectral region (see Fig.
 181 2). In the software the GSFC spectrum was treated as a series of delta functions, owing to the small
 182 bandwidth of $\approx 10^{-5}$ cm⁻¹ (300 kHz) per comb line [23]. Using the measured H₂S spectrum in Fig.
 183 2 the matching of up to three comb lines within experimental uncertainty was predicted for a center
 184 position of 6349.4 cm⁻¹ (position of comb line 8) and an FSR of 10 GHz (~ 0.334 cm⁻¹). The comb's
 185 center position (line 8) was first coarsely tuned several times between ~ 6349.15 and ~ 6349.45 cm⁻¹
 186 (this range corresponds approximately to the FSR of the GSFC). Where the largest intensity
 187 reduction on comb line 3 was observed in the coarse tuning, the comb's center position (line 8)
 188 was then finely tuned in steps of 0.004 cm⁻¹ (0.001 nm), i.e. from 6349.375 to 6349.415 cm⁻¹. For
 189 a comb center position at 6349.407 cm⁻¹ (line 8) the best overlap between comb line 3 (at
 190 6347.7487 cm⁻¹) and comb line 9 (at 6349.7532 cm⁻¹) with absorption features of H₂S was
 191 obtained;



192

193 **Fig. 3** The thick black trace is a small spectral segment of the H₂S reference spectrum in Fig. 2 which has been scaled
 194 at 6350.71 cm⁻¹ to the HITRAN absorption cross-sections (=blue solid dots) for a spectral resolution of 0.08 cm⁻¹ (bold
 195 axis title, right axis). The thin black trace shows 18 comb lines of the GSFC laser (relative intensity; second title on
 196 the right axis). The red trace (left axis) shows the absorption coefficient measured on the 18 comb lines for a center
 197 position of the GSFC at 6349.407 cm⁻¹ and a comb line spacing of 10 GHz. The red vertical arrows indicate comb
 198 lines which overlap significantly with H₂S absorption features at 6347.74, 6349.74, and 6352.74 cm⁻¹.

^{#2} Note that using the same FT-spectrometer for FT-IBBCEAS and comb experiments has however the advantage potential wavenumber calibration discrepancies are essentially eliminated.

199

200 Fig. 3 illustrates the situation. The thick black trace in Fig. 3 is a small spectral segment of the H₂S
 201 reference spectrum in Fig. 2, which has been scaled at 6350.71 cm^{-1} ($\nu_1+\nu_2+\nu_3$; $4_{22}-3_{21}$) to the
 202 HITRAN absorption cross-sections represented by the blue solid dots. The HITRAN cross-section
 203 have been evaluated for a spectral resolution of 0.08 cm^{-1} and are only shown for the strongest
 204 lines in this spectral region. The cross-section values are shown on the right axis (bold axis title).
 205 The thin black trace shows the relative intensity of 18 comb lines of the GSFC, only the position
 206 of the comb lines is of relevance here. The red trace (left axis) shows the absorption coefficient
 207 measured on the 18 comb lines (discussed below) when the comb line spacing (FSR) was 10 GHz
 208 and the center position of the GSFC (line 8) was at 6349.407 cm^{-1} . As mentioned above, for these
 209 conditions comb lines 3 and 9 overlap well with H₂S absorption features at $6347.7487 \text{ cm}^{-1}$
 210 ($\nu_1+\nu_2+\nu_3$; $5_{24}-4_{23}$) and $6349.7532 \text{ cm}^{-1}$ ($\nu_1+\nu_2+\nu_3$; $7_{17}-6_{06}$), respectively. Comb line 18 also
 211 overlaps significantly with an absorption feature at 6352.74 cm^{-1} ($\nu_1+\nu_2+\nu_3$; $8_{27}-8_{08}$); these three
 212 lines are marked with red vertical arrows in Fig. 3.

213 The spectral separation between the peak absorptions at comb line 3 (at $6347.7487 \text{ cm}^{-1}$) and comb
 214 line 9 (at $6349.7532 \text{ cm}^{-1}$) is 2.00451 cm^{-1} , which is slightly larger than $6 \cdot \text{FSR} = 2.00133 \text{ cm}^{-1}$ of
 215 the comb. This leaves a potential discrepancy of 0.0032 cm^{-1} , which is smaller than the smallest
 216 tuning step size of 0.004 cm^{-1} of the comb's center and ca. 3 times smaller than the expected
 217 FWHM of the H₂S absorption lines at the corresponding wavelengths. The discrepancy was within
 218 the measurement accuracy of the setup. Nevertheless, assuming the HITRAN line positions to be
 219 correct within 0.001 cm^{-1} , a comb FSR of 0.334085 cm^{-1} (15.87 MHz larger than the 10 GHz used)
 220 would have been optimum for matching comb lines 3 and 9 with H₂S absorption lines.

221 In comparison, the spectral separation between the peak absorptions at comb line 3 (at 6347.7487
 222 cm^{-1}) and comb line 18 (at $6352.7623 \text{ cm}^{-1}$) is 5.0136 cm^{-1} , which is slightly larger than $15 \cdot \text{FSR}$
 223 $= 5.0033 \text{ cm}^{-1}$ of the comb. The discrepancy of 0.0103 cm^{-1} is more significant than that for lines
 224 3 and 9. It corresponds to approximately the HWHM of the expected Voigt profile and is thus in
 225 agreement with the observation in Fig. 3, i.e. comb line 18 was not exactly positioned in the
 226 absorption line center. In conclusion, the position match of comb line 3, 9 and 18 with H₂S
 227 absorption lines will never reach the theoretical optimum. The strongest absorption line at

228 $\sim 6351.08 \text{ cm}^{-1}$ was unfortunately not overlapping well with comb line 13 (see Fig. 3) for the given
 229 center position of the comb and other multiple matching were also not easily attained.

230 We arbitrarily used line 8 of the comb to define the comb's center position, since this line
 231 also served as a good reference point for the intensity measurement, I_0 , without the sample. In the
 232 optimum comb position (at 6349.407 cm^{-1} , see Fig. 3) line 8 does not overlap with any H_2S
 233 absorption feature, and its peak intensity can be used as "reference intensity", I_0^{ref} , for all other
 234 comb lines. Prior to measurements with H_2S the spectrum of the comb transmitted by an empty
 235 cavity (evacuated at $\sim 0.1 \text{ mbar}$) was extensively measured and averaged. From this reference
 236 measurement intensity ratios r_k for each comb line k and center comb line 8 can be calculated;
 237 $r_k = I_{0,k}^{\text{ref}} / I_{0,8}^{\text{ref}}$. For a measurement with H_2S at first the cavity transmission intensities I_k were
 238 stability corrected by dividing through the intensities recorded simultaneously by the OSA. Then
 239 the ratio r_k was used to evaluate I_0 on a line that matches a H_2S absorption by multiplying r_k by
 240 the peak intensity of comb line 8; $I_{0,k} = r_k I_{0,8}$. The absorption coefficient α_k on a matching comb
 241 line k can then be calculated from [27, 29]:

$$242 \quad \alpha_k = \frac{1-R}{d} \left(\frac{r_k I_{0,8}}{I_k} - 1 \right) \quad (1)$$

243 The absorption coefficients for all α_k in Fig. 3 (red trace, left axis) were calculated using eq. (1).
 244 For the two most relevant lines 3 and 9 absorption coefficients
 245 $\alpha_3(6347.7487 \text{ cm}^{-1}) = 2.93 \times 10^{-5} \text{ cm}^{-1}$ and $\alpha_9(6349.7532 \text{ cm}^{-1}) = 1.06 \times 10^{-5} \text{ cm}^{-1}$ were
 246 determined (using $R = 0.993 \pm 0.001$ and $d = 66 \text{ cm}$). Since the H_2S absorption cross-section,
 247 $\sigma_3(6347.7487 \text{ cm}^{-1}) = 1.17 \times 10^{-21} \text{ cm}^2$ and $\sigma_9(6349.7532 \text{ cm}^{-1}) = 4.91 \times 10^{-22} \text{ cm}^2$ are known
 248 from the HITRAN database (converted to instrument resolution, blue dots in Fig. 3), the number
 249 density, $n = \alpha_k(\tilde{\nu}) / \sigma_k(\tilde{\nu})$, can be calculated. For absorption features on comb lines 3 and 9 we
 250 found $n_3 = 2.50 \times 10^{16} \text{ cm}^{-3}$ and $n_9 = 2.16 \times 10^{16} \text{ cm}^{-3}$, corresponding to a discrepancy of the number
 251 density of $\sim 13.6\%$. There are several reasons for this discrepancy: (i) The H_2S cross-sections in
 252 the HITRAN database have a reported uncertainty of 10% [30] which on two different lines could
 253 lead to a max error of 20%. (ii) Even though for this measurement the matching of the comb lines
 254 to the maximum of the absorption features on line 3 and 9 was satisfactory, tuning instabilities of

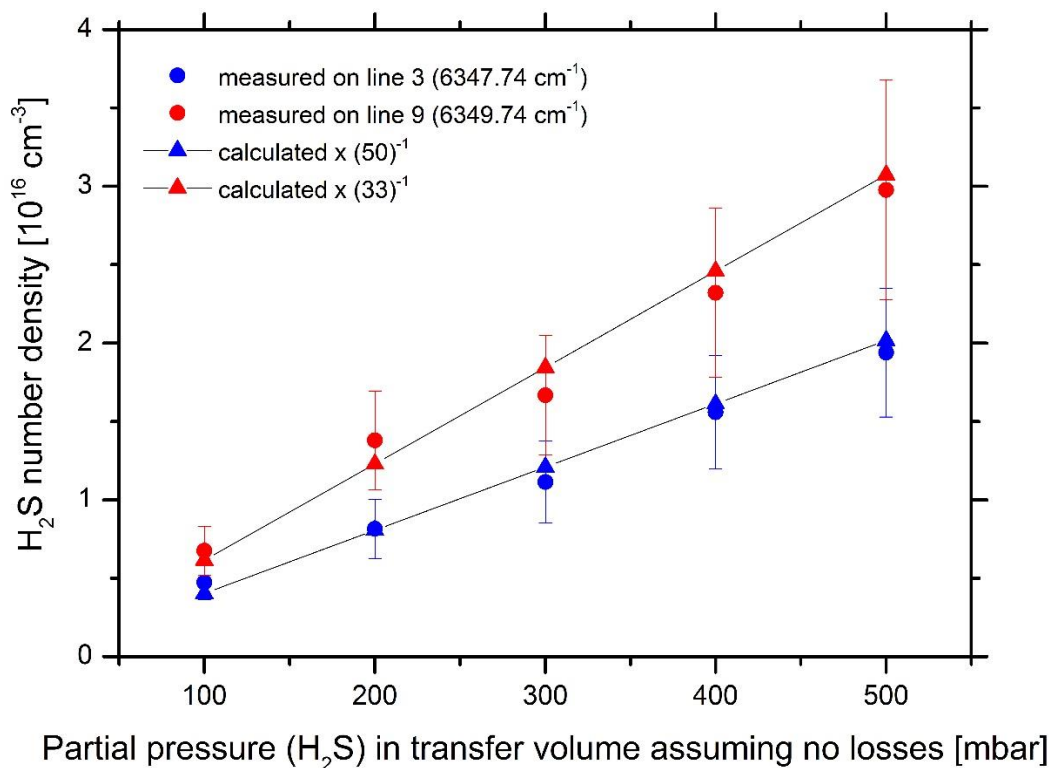
255 the GSFC may cause the comb lines to be not exactly at the maximum of the absorption feature
256 leading to an uncertainty in the expected relative absorption signals. When the GSFC was tuned
257 in between measurements the reproducibility of the number densities on the different absorption
258 lines is decreased due to this effect. The reason is that signal optimization is done at a spectral
259 resolution of only 0.08 cm^{-1} (limited by the FTS). This impinges on the overall error of the
260 measured number densities as further discussed in section 3.2. (iii) The comb intensity was
261 recorded for ~ 4 hours continuously and the absolute upper limit of the observed intensity
262 fluctuations was found to be $\approx 3.3\%$. Systematic uncertainties that may impinge on the absolute
263 value of the number density are the measured value of the mirror reflectivity (0.1%) and the length
264 of the cavity (0.3%), which are considered negligible in comparison to the uncertainties listed
265 above.

266 The averaged number density of $2.33 \times 10^{16}\text{ cm}^{-3}$ in this measurement corresponds to 0.92 ppTv.
267 This mixing ratio, calculated from the absorption of sample inside the cavity, was a factor of ~ 35
268 lower than that initially expected from the partial pressure measured upon sample preparation. For
269 the preparation of H_2S /air gas mixtures a vacuum line with a mixing volume of $\sim 10\text{ dm}^3$ made of
270 Pyrex glass was set up inside a fume hood because of the hazardous nature of H_2S . For the α_k
271 measurement in Fig. 3 mixtures of H_2S in air were prepared at a ratio of 6:4 at room temperature
272 (assuming no H_2S losses, see below), i.e. the mixing volume was first filled with H_2S at $\sim 400\text{ mbar}$
273 and then air at a partial pressure of $\sim 600\text{ mbar}$ was added. The mixture was collected for transfer
274 to the cavity setup (Fig. 1) in a small detachable Pyrex glass tube which was connected on one side
275 of the vacuum line via a Teflon valve. The volume of this “transport tube” was $\sim 82\text{ cm}^3$. As the
276 cavity volume was 998 cm^3 the partial pressures of air and H_2S in the cavity were expected to be
277 49.3 mbar and 32.9 mbar , respectively (provided there are no losses of H_2S when filling the transfer
278 volume). The latter pressure value corresponds to a mixing ratio of 32.4 ppTv under standard
279 conditions (1 atm, 296 K), which is ca. 35 times larger than the amount of H_2S found in the
280 absorption experiment (see above). The reason for this difference is the significant sample loss *in*
281 *the mixing chamber* and transfer volume, as both consisted of Pyrex glass causing inherently large
282 wall losses. Due to efficient chemisorption at metal ion sites in the glass the wall losses in the
283 mixing and transport volume are far from negligible. The fractional throughput of H_2S through a
284 Pyrex glass enclosure is known to be approximately 3% [38], corresponding to losses of a factor
285 ~ 33 in agreement with the loss factor found in the absorption experiment. Although the data in

286 Ref. [38] are not universal for arbitrary experimental conditions, the low fractional throughput of
287 Pyrex indicates that much less H₂S was collected in the transfer volume. Note that the filling
288 pressure was not measured in the transfer volume itself, but at a distance from it in the larger
289 mixing volume, which was essentially filled with a large excess of air. After the transfer of the
290 sample to the GSFC setup the pressure was measure in the cavity again.

291 To systematically study the effect of losses in the transfer of gas mixtures from the mixing volume
292 to the cavity, the absorption coefficients were determined on laser lines 3 and 9 for five different
293 H₂S concentrations. For these measurements the mixing volume was initially filled with different
294 partial pressures of H₂S between 100 and 500 mbar, then filled up with air to 1000 mbar, and
295 finally transferred to the cavity in the transfer volume. The absorption coefficients on line 3 and 9
296 had an overall uncertainty of $\Delta_{\max} < 25\%$ (see Table 1; this error is not to be confused with the
297 difference (discrepancy) in the number densities based on measurements on comb line 3 and 9).
298 The maximal uncertainty of 25% is associated with the reproducibility of the measured absorption
299 coefficient per comb line. For each partial pressure of H₂S the absorption coefficient was measured
300 3-4 times. Between each measurement the GSFC was finely tuned to optimize the signal and to
301 obtain an uncertainty in the reproducibility for a given H₂S concentration, which was evaluated
302 based on a student's t-distribution. Apart from contributions to Δ_{\max} that are associated with the
303 uncertainty of the H₂S cross-sections in HITRAN (10%), the optimum position of the comb lines
304 with respect to the absorption maximum was difficult to establish from measurement to
305 measurement as a detection resolution of 0.08 cm⁻¹ was used for signal optimization. This
306 resolution is larger than the FWHM of the absorption line at the measurement conditions (~0.021
307 cm⁻¹) and much larger than the comb line width (~10⁻⁵ cm⁻¹). The accuracy therefore relied strongly
308 the tuning stability of the GSFC, causing a fluctuation in the absorption coefficient and a
309 discrepancy in the relative absorption coefficients measured on line 3 and 9. The 3 σ standard
310 deviation of the absorption coefficient on lines that were not matched perfectly with H₂S
311 absorption features was $\approx 3.6 \times 10^{-6}$ cm⁻¹. This value of the minimum absorption coefficient in an
312 integration time of 120 s corresponds to ≈ 680 ppmv, which can be taken as a conservative (upper)
313 limit of detection (LOD) for this approach, because some of the laser lines always overlap to a
314 certain extent with H₂S absorptions, all of which were taken into account in the above stated 3 σ
315 LOD. From the H₂S absorption measurements the number densities in the cavity was again

316 estimated based on HITRAN cross-sections, which were systematically (expectedly) smaller than
 317 those estimated from the pressure in the mixing volume due to the losses of H₂S in the transfer
 318 process. An expected loss factor of ca. 33 (averaged) was found for comb line 9, while the loss
 319 factor appeared to be larger (average ~50) for line 3. This difference corroborates that the comb
 320 lines 3 and 9 were not equally well matched to the maximum of a H₂S absorption line. While line
 321 9 was probably near the maximum of the absorption feature (owing to the smaller loss factor), line
 322 3 was positioned off the center of the H₂S absorption line. Within our spectral resolution of only
 323 0.08 cm⁻¹ the exact position cannot be independently verified. The fact that the ratio of loss factors
 324 is approximately constant for all concentrations indicates that the approach is reproducible within
 325 the above stated limits, i.e. the laser position and mode structure enabling the light to be transmitted
 326 by the cavity is sufficiently stable in the current approach. The data are summarized in Table 1 and
 327 graphically represented in Fig. 4.



328
 329 **Fig. 4** H₂S number density in the cavity as measured from the absorption coefficient on comb line 3 (blue circles) and
 330 line 9 (red circles) vs the alleged partial pressure of H₂S in the mixing volume. The triangular symbols represent the
 331 number densities derived from the partial pressures on the horizontal axis, divided by a loss factor of 50 (blue) and 33
 332 (red). They can then be compared with the number densities from the absorption measurement. Error bars refer to the
 333 uncertainties stated in column 4 of Table 1.

334 **Table 1** Summary of measurement parameter concerning concentration dependent measurements. Values in Column
 335 4 vs column 1 are shown in Fig. 4 (circles). Col 1: Presumed H₂S partial pressure in 10 dm³ mixing volume without
 336 losses, in [mbar], measured with absolute pressure gauge. Col 2: Anticipated number density in the 1 dm³ cavity based
 337 on pressure in Col 1 and a 82 cm³ transfer volume, assuming no losses. Col 3: Measured absorption coefficient on line
 338 3 and 9. The reproducibility error is 25%. Col 4: Corresponding abs. cross-section from HITRAN database on line 3
 339 and 9. Col 5: Number densities following from values in Col 3 and 4. The error in col 3 is explicitly stated here per
 340 value (<25%). Col 6: Ratio of the anticipated number density (if there are no losses, Col 2) and the measured number
 341 density from the absorption (Col 5). The values corresponds to the approximate loss factor of H₂S in the transfer
 342 process from the mixing volume to the cavity. As a factor of ca. 33 is expected from ref [38]; the higher value on line
 343 3 indicates that the comb line was not exactly matched to the center (i.e. the maximum) of the H₂S absorption feature.

Pressure [mbar]	Number density [10 ¹⁷ cm ⁻³]	Absorption coefficient [10 ⁻⁶ cm ⁻¹]		Absorption cross-section [10 ⁻²² cm ²]		Number density [10 ¹³ cm ⁻³] (ppmv)		Number density ratio – loss factor –	
		Line 3	Line 9	Line 3	Line 9	Line 3	Line 9	Line 3	Line 9
in mixing volume	calculated from pressure	$\Delta_{\max} < 25\%$		From HITRAN (uncertainty 10%)		Measured ($\Delta_{\max} < 25\%$)		Avg: 50	Avg: 33
100	2.01	5.6	3.3	11.76	4.91	470 ± 100	680 ± 160	43	30
200	4.02	9.6	6.8	11.76	4.91	820 ± 190	1380 ± 320	49	29
300	6.03	13.1	8.2	11.76	4.91	1110 ± 260	1670 ± 380	54	36
400	8.04	18.3	11.4	11.76	4.91	1560 ± 360	2320 ± 530	52	35
500	10.05	22.8	14.6	11.76	4.91	1940 ± 450	2980 ± 680	52	34

344
 345 Fig. 4 shows, on the horizontal axis, the partial pressure of H₂S as measured in the mixing volume.
 346 On the vertical axis the number densities inside the cavity are shown (circular symbols). They
 347 were determined from the absorption coefficient measured on line 3 (blue data points) and line 9
 348 (red data points) for the respective mixing partial pressures. The error bars refer to the circular
 349 symbols. Additionally displayed are the number densities (triangular symbols, straight black line,
 350 no error bars) as they would be expected from the partial pressures, divided by the average loss
 351 factor for H₂S for the corresponding line (50 for line 3 and 33 for line 9; cf. Table 1). Fig. 4
 352 illustrates the consistency of Table 1 for measurement at different concentrations.

353 The first value in Fig. 4 (100 mbar) is essentially the detection limit. The lower explosion limit
 354 (LEL) of H₂S in air is ~4% while the upper explosion limit (UEL) is ~44% [39]. Since the LOD
 355 of ~0.7 per mille by volume is significantly smaller than the lower explosion limit (LEL) [39], the
 356 GSFC measurements presented here are well sufficient to detect H₂S at its LEL even in the near
 357 infrared.

358 Finally, an attempt was made to corroborate our results in Figs. 3 & 4 through the measurement of
 359 a high resolution line spectrum, by recording absorption coefficients on line 3 while stepping the

360 center of the GSFC over the optimum position at 6349.407 cm^{-1} (line 8 in Fig. 3) in steps of 0.004
361 cm^{-1} . The problem with this experimental line width test was that all absorption measurements
362 were taken with a resolution of 0.08 cm^{-1} , and the true position of the laser with respect to the line
363 center of the absorption line at 6347.74 cm^{-1} (line 3) could not be determined with sufficient
364 accuracy. The test measurement relied solely on the tuning stability of master laser. Even though
365 the spectrum of the line at 6347.74 cm^{-1} could be recorded by tuning the GSFC, the FWHM was
366 ca. half the size of the anticipated Voigt FWHM (0.021 cm^{-1}) according to the estimated pressure
367 and Doppler broadening, see above. This discrepancy is likely to be a consequence of the electronic
368 tuning stability and potential small drifts of the GSFC. The stability was investigated by
369 independently tuning to the GSFC center position to the maximum absorption signal many times
370 in the context of the I_0 measurements. The absolute laser position, as solely determined from the
371 tuning voltage, could be reproduced within 0.017 cm^{-1} , which is 20% smaller than the anticipated
372 FWHM of the expected H_2S absorption line. The tuning to the maximum position is uncertain
373 within that limit, explaining the different loss factors found for the laser comb lines 3 and 9.

374 **4. Conclusions and outlook**

375 Two comb lines (3 and 9) were matched to absorption lines of H_2S in the near infrared L-band
376 without locking the laser to the cavity mode structure. Without locking the laser line, the overlap
377 of the laser with the absorption features was accurate and stable to within $\approx 80\%$ of the FWHM of
378 the absorption line. From the non-resonant comb lines, an rms noise equivalent absorption for this
379 approach of approximately $3.9 \times 10^{-5}\text{ cm}^{-1}\text{ Hz}^{-1/2}$ was found, which corresponds to a detection limit
380 of ca. 700 ppmv of H_2S in 120 s. This sensitivity is well below the lower explosion limit of H_2S
381 of 4% and was achieved with a very modest finesse (~ 450) cavity.

382 Future work will focus on making the setup more compact by using custom-designed integrated
383 combs and improving the detection resolution through a dual comb approach. Alternative coupling
384 scheme of the laser to the cavity will be looked into and the detection cell is to be developed into
385 a flow cell. Other spectral regions are to be targeted.

386

387

388

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396

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