

The ammonia dimer revisited

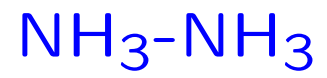
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Ad van der Avoird (Nijmegen)

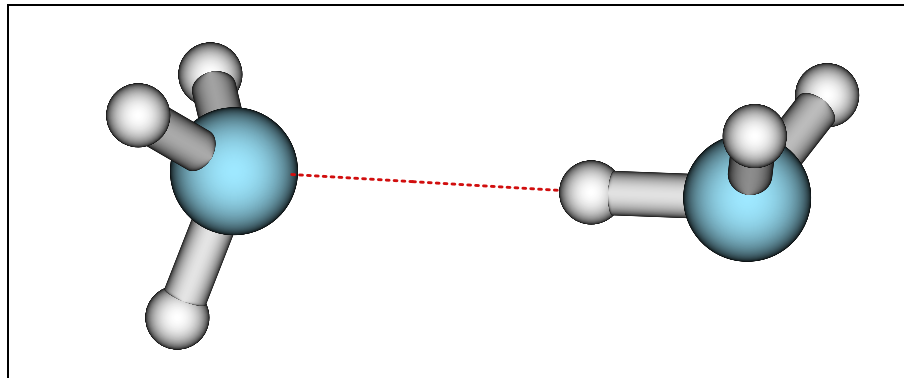
Ammonia dimer, a surprising structure

Nelson, Fraser, and Klemperer, *J. Chem. Phys.* **83**, 6201 (1985)

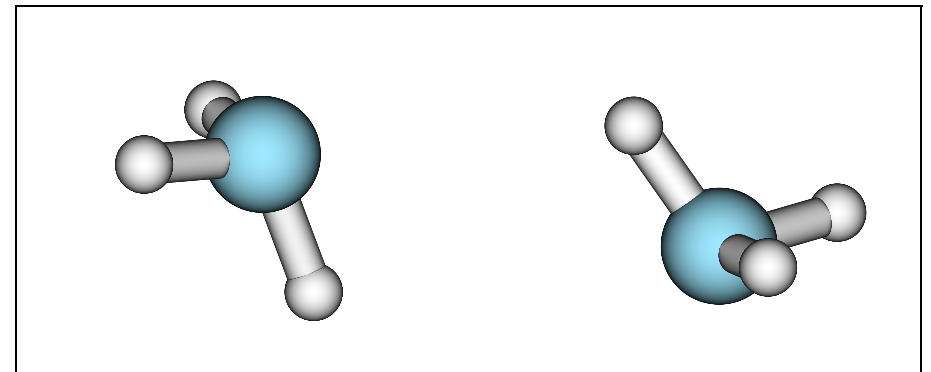
- MBER spectrum, ^{14}N quadrupole splittings
⇒ nearly cyclic structure
- Small dipole moment: 0.75 Debye (NH_3 monomer 1.47 D)
- *Ab initio* calculations had predicted a nearly linear $\text{N}-\text{H}\cdots\text{N}$ hydrogen-bonded structure
- Equivalent H-bonded structures with donor/acceptor interchanged vibrational averaging ?
- Structure similar for $^{14}\text{NH}_3-^{14}\text{NH}_3$, $^{14}\text{ND}_3-^{14}\text{ND}_3$, $^{15}\text{NH}_3-^{14}\text{NH}_3$
⇒ nearly rigid



Hydrogen-bonded structure



Structure of Klemperer *et al.*



Contradictory evidence

Havenith, Cohen, Busarow, Gwo, Lee, and Saykally

J. Chem. Phys. 94,4776 (1991)

Loeser, Schmuttenmaer, Cohen, Elrod, Steyert, Saykally, Bumgarner, and Blake

J. Chem. Phys. 97,4727 (1992)

- Large tunneling splittings observed in THz spectra
⇒ $\text{NH}_3\text{-NH}_3$ must be floppy

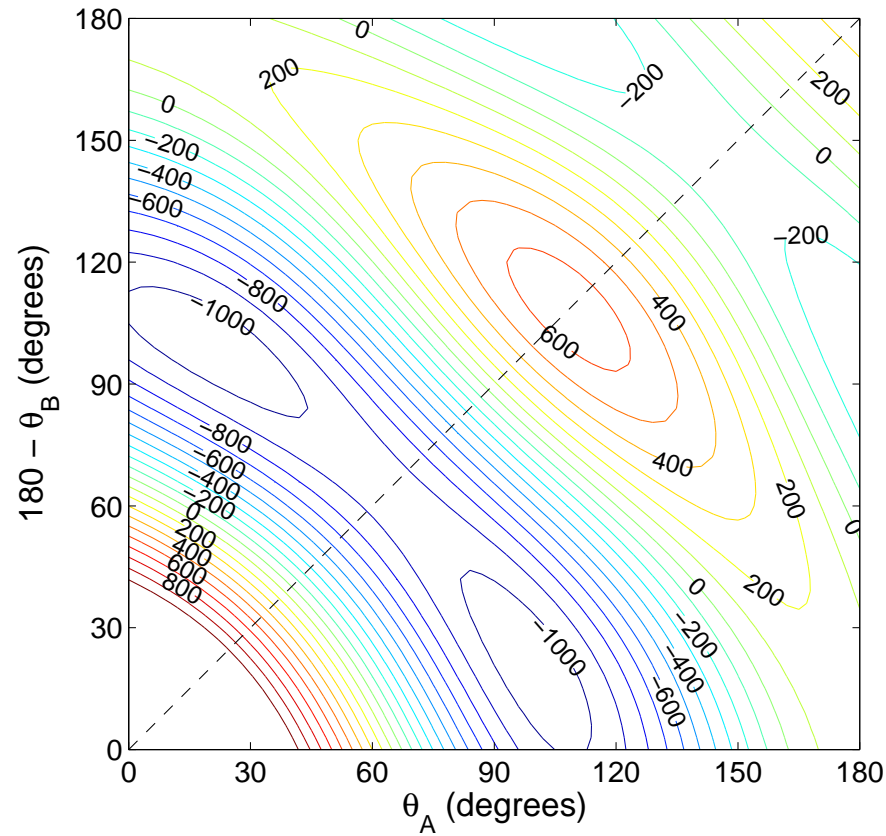
Our approach



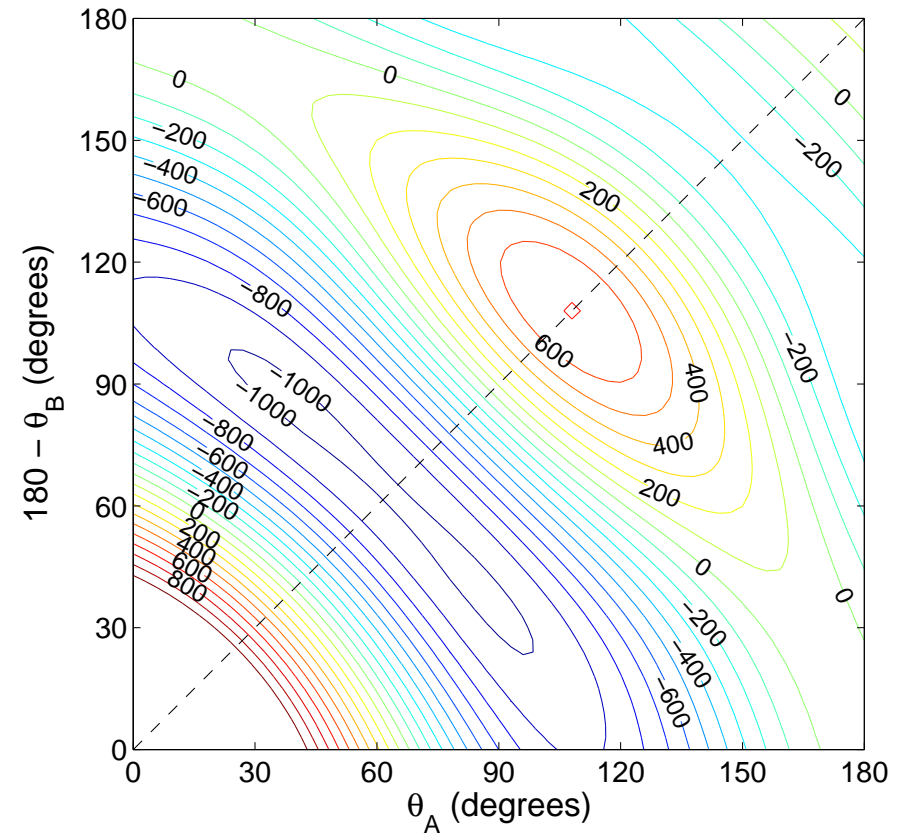
Compare with high-resolution THz and microwave spectra

Empirical model potential

Electrostatic interactions favor H-bonded structures

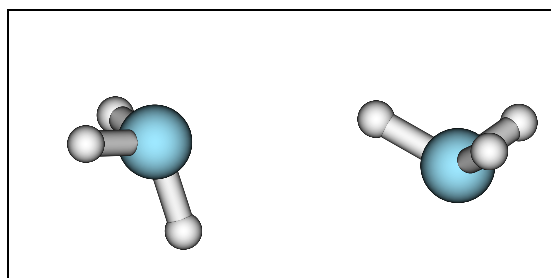


Dispersion and exchange interactions added

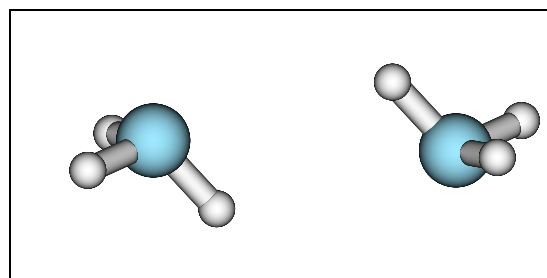


Donor-acceptor interchange pathway

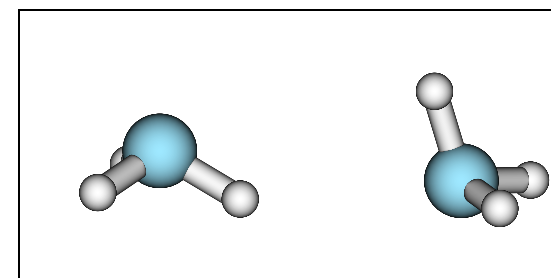
minimum



saddle (cyclic)



minimum



Barrier $\approx 6 \text{ cm}^{-1}$ (after fit to THz spectra)

Calculation of VRT states

Hamiltonian

$$H = \left[A j_{A_a}^2 + B j_{A_b}^2 + C j_{A_c}^2 \right] + \left[A j_{B_a}^2 + B j_{B_b}^2 + C j_{B_c}^2 \right] \\ - \frac{\hbar^2}{2\mu_{AB}R} \frac{\partial^2}{\partial R^2} R + \frac{1}{2\mu_{AB}R^2} \left[J^2 + j_{AB}^2 - 2\mathbf{j}_{AB} \cdot \mathbf{J} \right] + V(R, \boldsymbol{\omega}_A, \boldsymbol{\omega}_B)$$

Analytic basis

$$\chi_n(R) \sum_{m_A m_B} D_{m_A k_A}^{(j_A)}(\boldsymbol{\omega}_A)^* D_{m_B k_B}^{(j_B)}(\boldsymbol{\omega}_B)^* \langle j_A m_A; j_B m_B | j_{AB} K \rangle D_{MK}^{(J)}(\alpha, \beta, 0)^*$$

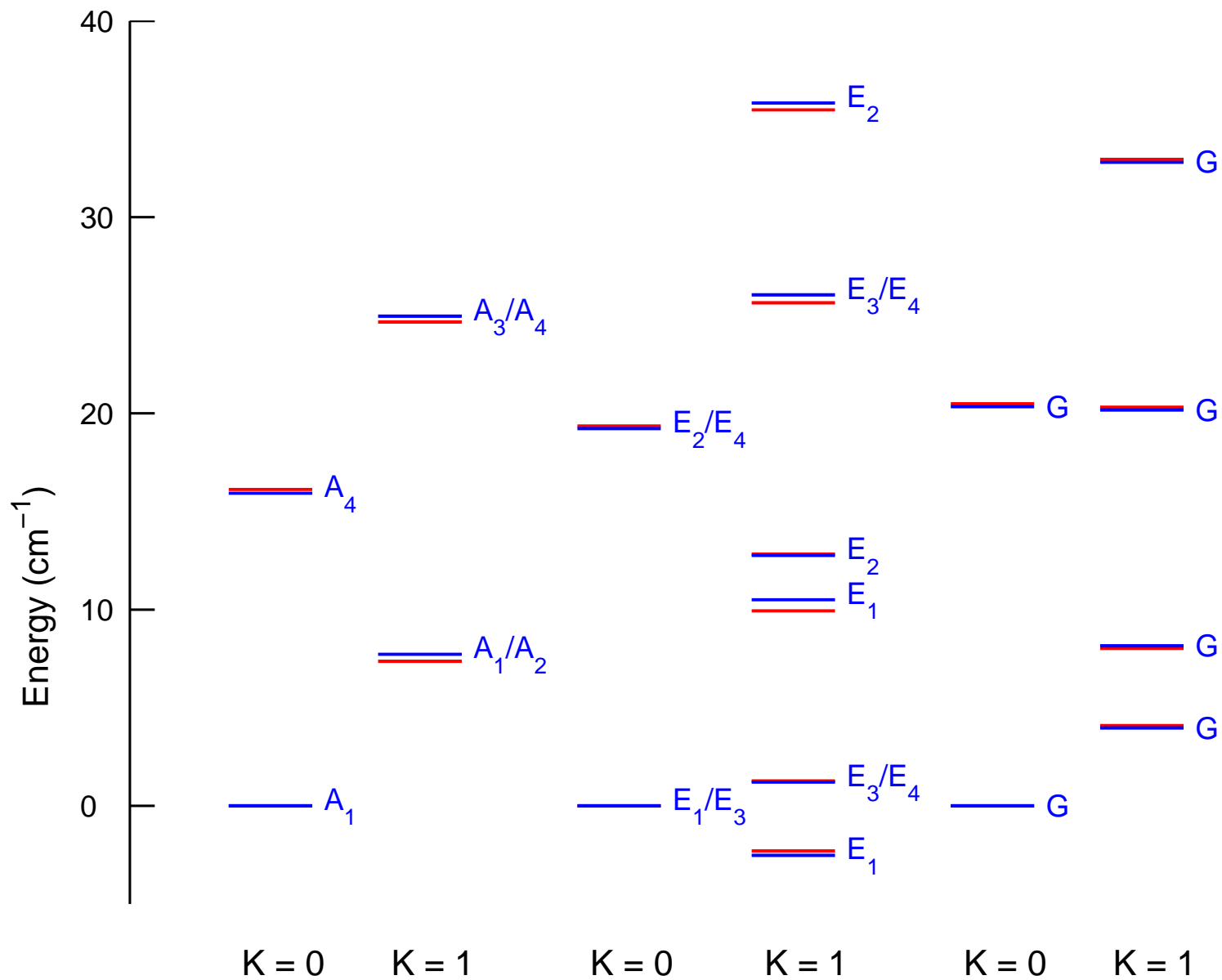
contracted DVR for $\chi_n(R)$

Basis adapted to irreps of PI group G_{36}

Permutation-inversion symmetry group G_{36}

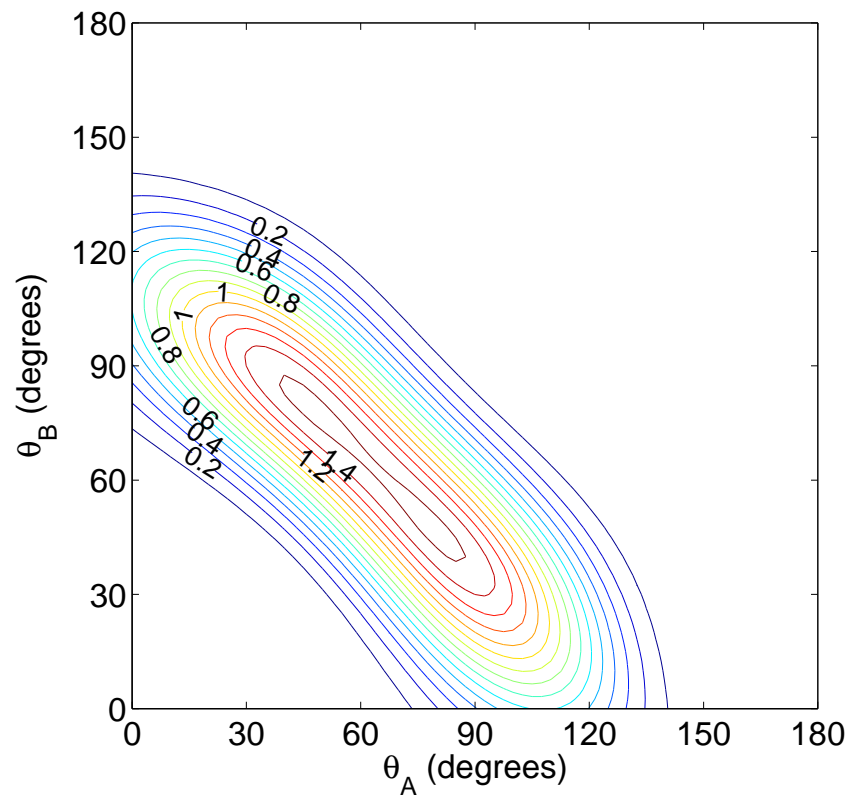
G_{36} irrep	dim.	statistical weight	
		$(\text{NH}_3)_2$	$(\text{ND}_3)_2$
A_1	1	66	561
A_2	1	78	528
A_3	1	66	561
A_4	1	78	528
E_1	2	36	576
E_2	2	36	576
E_3	2	30	600
E_4	2	42	552
G	4	144	1584

Calculated and measured energy levels ($J = 0, 1$)

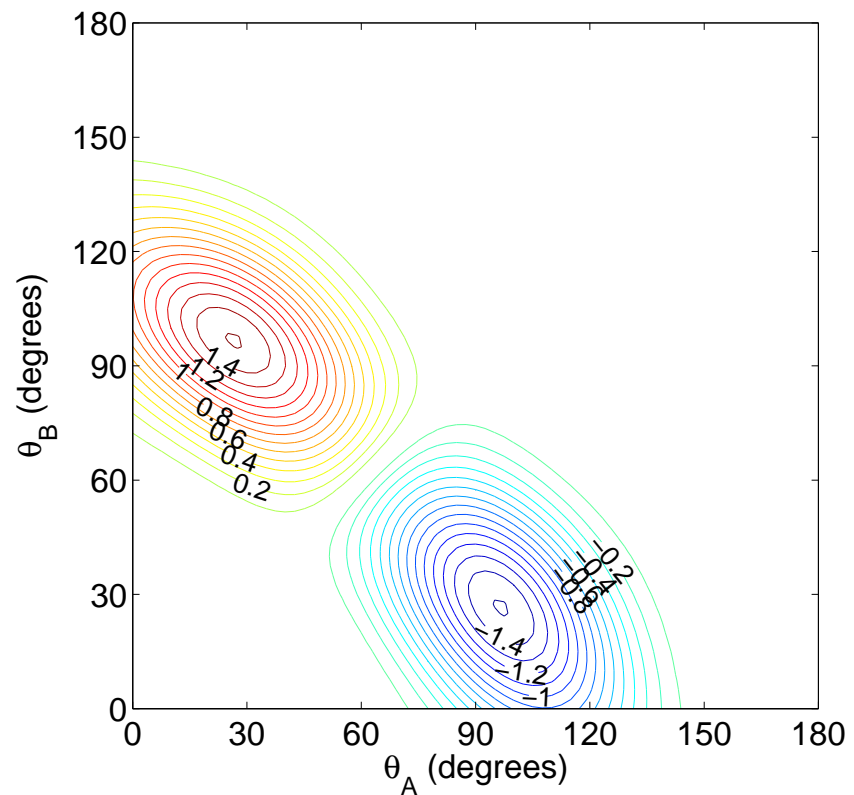


Wave functions of oNH₃-oNH₃ ($J = 0$)

symmetry A_1

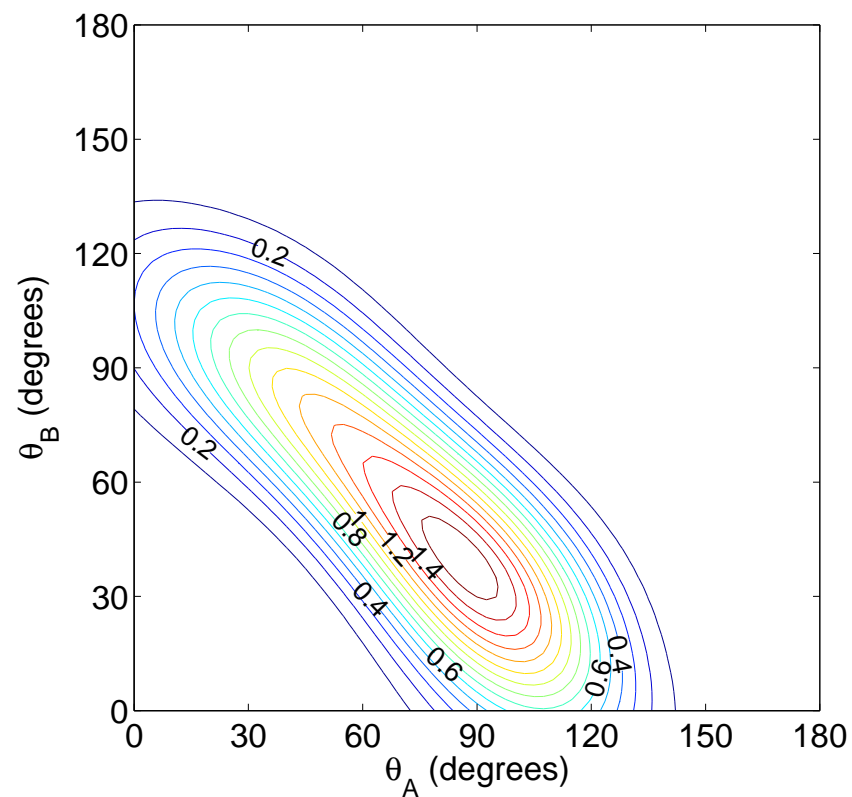


symmetry A_4

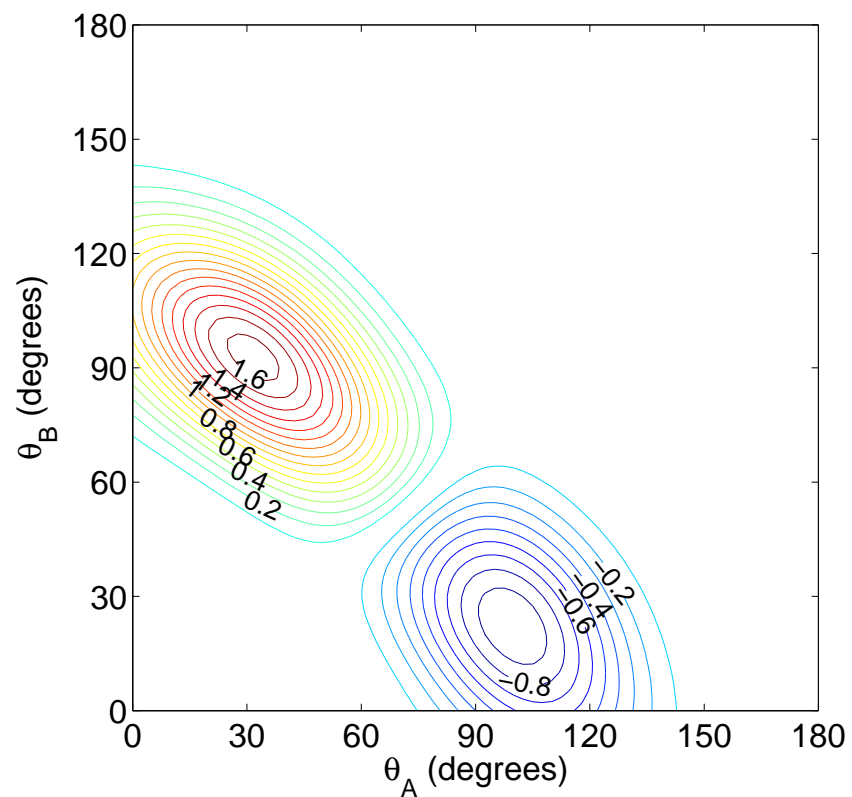


Wave functions of pNH₃-oNH₃ ($J = 0$)

symmetry G , lowest state



symmetry G , second state



Conclusions

- Good agreement with both MBER and THz spectra
- Dipole moment also agrees with experiment
- Ammonia dimer is very floppy
- Isotope substitution effects explained

Large amplitude internal motions

Why does the vibrationally averaged structure not change by isotope substitution?

Explanation

Only states of G symmetry observed by Klemperer *et al.*
mixed ortho NH_3 - para NH_3 , non-zero dipole moment

Two effects working in opposite direction

1. ND_3 orientations more localized in well region than NH_3
 \Rightarrow larger dipole moment
2. ortho-para difference smaller for ND_3 than for NH_3
 \Rightarrow more symmetric behavior of mixed ortho-para dimer
 \Rightarrow smaller dipole moment

Result: near cancelation of isotope effects on geometry

Revisiting

- Accurate potential surface calculated *ab initio* by Richard Dawes
- CCSD(T)-F12b(AE)/VTZ-F12 method;
more than 22 000 geometries for $R = 2.5 - 20$ Å
- D_0 measured by Crim *et al.*: 660 ± 20 cm⁻¹

Ab initio:

Minimum -1111.9 cm^{-1} ($\theta_A = 43^\circ$, $180^\circ - \theta_B = 87^\circ$)
(-1018 cm^{-1} in model potential, at nearly the same geometry)

Cyclic saddle point -1107.5 cm^{-1} ($\theta_A = 180^\circ - \theta_B = 67^\circ$)

Interchange barrier: 4.4 cm^{-1} , very similar to model potential

Current status:

- VRT levels calculated from fitted potential surface
- First results: $D_0 = 665 \text{ cm}^{-1}$ (for oNH₃-oNH₃)
experiment $660 \pm 20 \text{ cm}^{-1}$
- VRT level pattern OK, tunneling splittings too small
- Splittings depend very sensitively on the potential;
role of NH₃ umbrella mode dependence?
- Global fit of the *ab initio* data to accurate site-site model
cf. water dimer CCpol-8s potential, which produced accurate
tunneling splittings and intermolecular vibrational frequencies

Acknowledgement

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