

Fundamental frequencies of 2 – iodo- 5 – nitro toluene from FT – Raman and FT – IR spectra

Yogesh Kumar Agarwal and P K Verma*

Department of Physics, Aligarh Muslim University, Aligarh-202 002, Uttar Pradesh, India

K M Lal

Z. H. College of Engineering and Technology, Aligarh Muslim University, Aligarh-202 002, Uttai Pradesh, India

E-mail : yka123@yahoomail.com

Received 30 January 2003, accepted 24 June 2003

Abstract FT-Raman and FT-IR spectra of 2 - 10do-5-nitro toluene have been recorded in the region 50-500 cm⁻¹ and 400-4000 cm⁻¹ respectively. The vibrational analysis is carried out in terms of fundamentals. The probable mode of vibrations have been assigned assuming C symmetry.

keywords . FT-Raman, FT-IR spectrum, vibrational assignments, 2-lodo-5-nitro toluene

PACS Nos. 33 20.Ea, 33.20.Fb, 33 20.Tp

1. Introduction

In recent times, toluene and substituted toluene have become very important on account of their wide use in medicine and industry. Vibrational spectra of various substituted toluenes were reported earlier [1-5]. Among the halogenated nitro toluene, the vibrational spectra of chloronitro-, bromonitro-, and fluoronitro-toluenes, were reported earlier, but the vibrational spectra of iodonitro toluenes do not seem to have been investigated so far. This paper deals with the recording and analysis of the FT-Raman and FT-IR spectra of 2-iodo-5 mitro toluene (2-15-NT) in mid and far-infrared regions.

2. Experimental

Commercially available (2-15-NT) molecule of spec-pure grade was used without further purification for recording the FT-Raman and FT-IR spectra. The FT-Raman and FT-IR spectra of (2-15-NT) were recorded using Bruker IFS 66V FT-IR/FT-Raman spectrometer having a resolving power of 0.1 cm⁻¹. The FT-Raman spectrum of the powder sample over the

range $50-3500~\rm cm^{-1}$ are recorded using Nd YAG laser, at wavelength 1064 nm and power 200 MW and a germanium diode detector. The FT - IR spectra in far-infrared region (FIR) are obtained over the range of $50-500~\rm cm^{-1}$ using polyethylene technique at a resolution of $4~\rm cm^{-1}$. The FT - IR spectra in mid-infrared region (MIR) are recorded over the range of $400-4000~\rm cm^{-1}$ on a Nicolet Avtar - $360~\rm FT$ - IR spectrophotometer, using KBr pellet technique. The accuracy of the measurements was estimated to be with in $\pm 1~\rm cm^{-1}$ and accuracy in the assignment of the fundamental mode of the vibrations was estimated to be with in $\pm 20~\rm cm^{-1}$.

3. Discussion

Almost all the bands observed in the entire region, are analysed in terms of fundamental frequencies, overtones, combinations and differences. In (2-I 5-NT) molecule, parent ring is a benzene ring; therefore, the same nomenclature as that used in benzene is adopted to designate the various type of fundamental modes of vibrations. The replacement of the hydrogen by the substituent groups, no doubt, changes the value of the fundamental frequencies quite appreciably. Substitution of the

Corresponding Author

Table 1. Fundamental frequencies (cm⁻¹) and assignments for 2-iodo-5-nitro tolucne

Infrared		Raman		Probable assignments		
Freq (cm ⁻¹)	Int	Freq (cm ⁻¹)	Int	Species	Mode No. (Wilson notation)	Description
3077	5 5	3085	0 5	a'	2	ν(C-H) stretching
3023	6.0	3067	07	a'	20a	v(C-H) stretching
3015	7 ()	3017	0.3	a'	20ь	v(C-II) stretching
2984	7 0	2973		a"	-	v, stretching in CH, group
2938	6.5	2922	0 4	a"	-	v _{ss} stretching in CH ₃ group
2892	6.0	~~		a'	-	v stretching in CH, group
1646	10	1639	0 2	a'	8b	v(C-C) stretching
1608	4 5	1603	0.4	a'	8a	v(C-C) stretching
1510	0.5	1516	0.4	a'	19b	v(C-C) stretching
1454	3 0	1475	0.3	a"	-	δ_{n_0} deformation in CH, grou
1423	5 0	1420	0.3	a'	19a	v(C-C) stretching
1375	3 0	1379	0.5	a'	-	δ , deformation in CH, grou
1340	0 0	1340	7 5	a'	_	v, stretching in NO2 group
1265	4 ()	1267	0.3	a'	3	δ (C-C) i p bending
1254	1.5	1255	0.3	a'	13	v (C-CH,) stretching
1192	5 0	1207	1 4	a'	18a	δ (C-C) ip bending
1130	6.0	1125	0.1	a'	18b	δ (C-C) ip bending
1090	5 0	1097	0.5	a'	-	rocking of CH, group
1008	3 0	1013	0.2	a'	-	rocking of CH, group
992	3 0	-	-	a"	17b	γ (C-H) op bending
946	8 0	-	-	a"	5	γ (C-H) op bending
923	8 0	920 _	0.1	a'	7b	ν(C··NO ₃) stretching
854	2 0		_	a'	-	δ_i deformation in NO, grou
810	1.5	807	0.7	a"	1 1	γ (C-H) o.p bending
746	1.0	-	-	a'	12	δ (C-C-C) ip bending
669	7 0	680	0.5	a'	1	γ (C-C) ring breathing
577	7.0	558	0.3	a"	16a	γ (C-C·C) o.p bending
515	6.0	534	0 4	a"	16b	γ (C-C-C) op bending
477	8 0	-	_	a'	6a	δ (C-C-C) i.p bending
431	8 0	_	_	a'	6b	δ (C-C-C) i.p. bending
380	6.4	375	0 2	a'	9a	δ (C-CH ₁) ip bending
295	7 0	282	0 4	a'	7a	v(C-I) stretching
273	6.1		_	a'	15	(C-NO ₂) ip bending
241	7.0	235	0 4	" a'	9b	δ (C-I) 1 p bending
207	7 2	211	0.8	a"	10ь	γ (C-CH ₁) o.p bending
187	7.5	185	0 2	и" а"	17a	γ (C-I) o p. bending
167	6.8			a"	17b	γ (C-NO ₃) o.p. bending
147	77	_	_	a"	-	torsion in CH, group
110	8 0	_	_	a"	_	torsion in NO, group

v = stretching, $\delta = \text{in-plane bending}$, $\gamma = \text{out-of-plane bending } v_{ni} = \text{asymmetric stretching}$, $\delta_{ni} = \text{symmetric deformation}$, $\delta_{ni} = \text{symmetric deformation}$.

hydrogen also changes the point group symmetry from Dec corresponding to benzene to C if NO2 and CH3 are regarded as point masses. (2-I 5-NT) will have in addition to 30 benzene like vibrations, six NO2 vibrations and nine CH₃ vibrations, making up a total of 45 fundamental vibrations. As the molecule belongs to C, point group, there is only one plane of symmetry i e plane of molecule. There will be only two types of vibrations that are a' (planar) and a'' (non The planar). obscrved fundamental frequencies and probable assignments \ are presented in Table 1. All the assignments are made by comparison of the experimental frequencies and assignments reported earlier [1-3] in similar molecules.

3.1 Stretching vibrations (v):

3.1.1. C-H and C-X stretching

Benzene has six modes of (C-H) stretching vibrations denoted by the numbers 2, 20a, 20b, 7a, 7b. and 13; in Wilson's notations. these have the frequencies 3062 $(a_{1\nu})$, 3080 $(e_{1\nu})$, 3046 $(e_{2\nu})$, and $3060 \text{ cm}^{-1} (b_{1u})$ respectively. In case of tri-substituted benzenes. out of six (C-H) stretching vibrations, three remain pure (C-H) stretching vibrations while the remaining three frequencies would be (C-X) stretching (X = substitution for H atom) and will be depending on the masses of the substituents. The three (C-H) stretching frequencies do not change due to substitution and lie in the range of 3000-3100 cm⁻¹. The (C-X) stretching decrease frequencies considerably [6] below 1200 cm-1. If the substituents, are light (atomic mass less than 25), these frequencies will lie above 1000 cm-1. Thus utilizing the mode of

Visual estimates of the intensities have been given in the visual scale of 1 to 10.

312. C-C stretching vibrations

According to Singh and Singh [6], there are six normal modes of benzene, which belong to the groups namely 8a, 8b, 19a, 19b, 14 and 1. The v(C-C) stretching vibrations of benzene come around 1600 cm^{-1} and correspond to a degenerate e_{2g} mode 8 (8a and 8b). In substituted benzenes, the degeneracy should be removed and the two components should appear near 1600 cm^{-1} giving tise to fairly strong bands in infrared and Raman spectra. Thus, the strong bands at (R1639, IR1646) and (R1603, IR1608) cm⁻¹ are assigned to modes 8a and 8b respectively. It could be seen that we have taken 8a > 8b. This is done in order to be in accordance with the results of normal mode analysis of trisubstituted benzenes [9, 10].

The pair of vibrations 19a and 19b arises from e_{1u} mode of benzene. They are usually strong in infrared and are expected in the range of 1400-1500 cm⁻¹. There are two bands at (R 1420, IR 1423) and (R 1510, IR 1516) cm⁻¹, whose origin can be attributed to modes 19a and 19b respectively.

The vibration mode 14, called the kekule mode, belongs to b_{S_n} species of benzene, expected near 1300 cm⁻¹, in disubstituted benzenes. A correct assignment of this band is complicated by the fact that the $\delta(C-C)$ in-plane-bending mode 3 also occurs in this region. The frequencies (R1267 IR 1265) cm⁻¹ observed in (2-15-NT) have been assigned to mode 3. Our assignment is in agreement with the assignments made in flouro nitro toluenes by Syam Sundar and Kanna Rao [1]. The frequencies (R 680, IR 669) cm⁻¹ have been assigned to the mode 1 in accordance to assignments made by earlier workers [1-5] and the frequency range agrees with that given by Shukla et al [11].

32. In-plane Bending vibrations:

321. (C-H) and (C-X) δ vibrations

The C-H in-plane bending modes lie in the region $1000-1300\,\mathrm{cm^{-1}}$ These arise due to a_{2g} (1340 cm⁻¹), e_{2g} (1178 cm⁻¹), b_{2u} (1152 cm⁻¹) and b_{1u} (1037 cm⁻¹) modes of benzene denoted by Wilson's numbers 3, 9a, 9b, 15, 18a and 18b, respectively. In trisubstituted benzene derivatives, three modes of vibrations remain almost unchanged and are called (C-H) δ vibrations, while three modes change considerably in frequency and are

called (C-X) δ vibrations. The frequencies for (C-H) δ modes have been assigned at (R 1267, IR 1265), (R 1207, IR 1192), and (R 1125, IR 1130) cm⁻¹ in agreement with the observations of several workers [1-3] and following the suggestions of Varsanyi [12]. These frequencies will originate from the mode 3, 18a, and 18b. The frequencies for (C-X) δ modes have been assigned at (R 375, IR 380), (IR 273) and (R 235, IR 241) cm⁻¹ as C-CH₃, C-NO₂ and C-I vibrations respectively corresponding to the mode 9a, 15 and 9b respectively. These assignments are in good agreement with those of Refs. [13-15] and the frequency range with that given by Rao [16], Varsanyi [12].

3.2.2. C-C-C in-plane bending

The normal modes 6a, 6b, and 12 are regarded as the (C-C-C) δ vibrations. The frequencies (IR 477), (IR 431), (IR 746) cm⁻¹, have been identified as (C-C-C) in-plane bending vibrations. The above assignments are within frequency range given by Varsanyi [12], for similar compounds and find support from the studies of other workers. [17, 18].

3.3. Out-of-plane bending:

3.3.1 (C-H) and (C-X) out-of-plane bending vibrations

The modes 5, 17a, 17b, 10a, 10b and 11 with frequencies (958) $b_{2g}, (975)\ e_{2u}, (849)\ e_{1g}, (673)\ a_{2u}$, represent (C-H) γ . These vibrations have been observed at (IR 946), (IR 992), (R 807, IR 810) cm $^{-1}$ in (2-I 5NT) corresponding to modes 5, 17b, and 11 respectively. The frequencies (R 185, IR 187), (R 211, IR 207) cm $^{-1}$ have been identified as C-I and C-CH $_3$ out-of-plane bending vibrations corresponding to modes 17b, and 10b respectively. Our assignments find support from the work of Green et al [17] and are within frequency range given by Varsanyi. [12].

3.3.2. C-C-C out of plane bending vibrations

The normal modes 4, 16a and 16b are regarded as the (C-C-C) γ vibrations. The frequencies (R 558, IR 577), (R 534, IR 515) cm⁻¹ have been identified as C-C-C out-of-plane bending vibrations. Our assignments find support from the work of several workers. [1, 19, 20].

3.4. Internal vibrations of methyl group (CH, group):

There must be three C-H stretching modes (two asymmetric and one symmetric type). The frequencies of asymmetric vibrations are higher than that of symmetric one. According to Colthup et al [21], these frequencies appear around fairly constant regions of 2960 cm⁻¹ and 2870 cm⁻¹ respectively. Singh et al [22] have assigned symmetric and asymmetric C-H stretching modes in CH₃ group in o-m-p- methoxy benzaldehydes within the frequency range (2829–2863) cm⁻¹ and (2945–2965) cm⁻¹ respectively. In the present case, the frequencies (R 2973, IR 2984) and (R2922, IR 2938) cm⁻¹ are taken as asymmetric C-H stretching modes and (IR 2892 cm⁻¹ as symmetric C-H stretching mode in CH₃. These assignments find support from the work of Fusion et al [23] and Rangacharyalu and Premaswarup [3].

The rocking vibrations of CH₃ group interact with other modes. As a result of this interaction, the rocking mode frequencies do not appear consistently. Several workers [24, 25] have assigned these vibrations at about $1170 \pm 40 \text{ cm}^{-1}$. Rangacharyulu and Premaswarup [3] have assigned out-of-plane rocking mode of CH₃ group at 1050 cm^{-1} in 2-bromo-5-nitro toluene, at 1090 cm^{-1} in 2-bromo-4- nitro toluene and 1080 cm^{-1} in 4-chloro-3-nitro toluene, respectively. In the present case, the frequencies at (R 1090, IR 1097), (R 1008, IR 1013) cm⁻¹, are assigned to this mode. Our assignments find support from the work of several workers. [3,11,17].

A literature survey [26-28] reveals that is xylene, the CH_3 torsional mode appears at about 180 cm⁻¹. We have observed tentatively this vibration at about (IR 147) cm⁻¹.

3.5. Internal mode of vibration of NO, group:

The symmetric stretching of the substituent NO₂ group have been assigned in the region (1338-1351) cm⁻¹ by various workers. In the present case, the frequencies (R1340, IR 1340) cm⁻¹ have been assigned to the symmetric stretching mode in NO₂ group, in accordance with the assignments made by previous workers. [29-32].

NO₂ deformation vibration has been assigned to the band observed at (IR 854) cm⁻¹, these vibrations have been reported in the region 824 – 890 cm⁻¹ by many workers [29-33]. The band observed at (IR 167) cm⁻¹ has been assigned to C-NO₂ out-of-plane bending vibration in (2-I 5-NT). Rai *et al* [2] have identified this vibration at 150 cm⁻¹ in 4-chloro 2-nitro toluene. Torsion mode vibration in NO₂ group has been observed at 110cm⁻¹ in (2-I 5-NT), Syam Sundar and Kanna Rao [1] have suggested this vibration in the region 111-120 cm⁻¹ in fluoro nitrotoluenes.

4. Conclusion

Vibrational assignments of such complicated molecules have been performed by taking the help of similar vibrations observed in similar molecules. These studies will be helpful to identify ground state vibrations of the above benzene derivative, and to enable us to propose a consistent assignment for C-X modes and the internal modes of the substituent groups and to study the effects of these substituents on the phenyl ring modes. These frequencies observed in FT-Raman and FT-IR spectra can be used as ground state vibrations, to confirm excited state vibrations of (2-15-NT) with the help of the electronic spectra.

Acknowledgments

We are thankful to Chairman, Department of Physics, Aligarh Muslim University, Aligarh, India for providing the necessary facilities to carry out this work, and encouragement.

Refereneces

- N Syam Sundar and R Kanna Rao Spectrochim. Acta. 49A 1691 (1993)
- [2] D K Rai, N P Singh and R B Singh Indian J Phys. 56B 62 (1985)
- [3] M Rangacharyulu and D Premaswarup Indian J Pure Appl Phys 16 37 (1978)
- [4] N Syam Sundar Spectrochim Acta. 41A 905 (1985)
- [5] N Syam Sundar Can. J Chem 62 2238 (1984)
- [6] R P Singh and R N Singh Indian J Pure Appl. Phys 26 644 (1988)
- [7] J R Scherer Spectrochim Acta (GB) 19 601 (1963)
- [8] J.R. Scherer Spectrochim. Acta. 21 321 (1965)
- [9] G Ramana Rao and A Anjaneyulu Asian Chem Lett 3 135 (1999)
- [10] G Ramana Rao and A Anjaneyulu Asian Chem Lett. 3 128 (1999)
- [11] A R Shukla, C M Pathak, N G Dongre and B P Asthana J. Raman Spectrose 17 299 (1986)
- [12] Varsanyi Assignments for Vibrational Spectra of Seven Hundred Benzene Derivatives, Vol I and II (London: Adam Hilger) (1974)
- [13] D K Rai and R B Singh Indian J Phys. 53B 144 (1979)
- [14] J K Wilmshurst and H T Bernstein Can J Chem. 35 911 \1957.
- [15] M A Shashidhar and K Suryanarayana Indian J. Pure Appl Phys 8 483 (1970)
- [16] C. N. R. Rao Chemical Applications of Infrared Spectroscopy (New York : Academic) (New York, London) (1963)
- [17] J S H Green, D J Harrison and W Kynaston Spectrochum Acta 27A 807 (1971), 28 33 (1972)
- [18] S K Singh and R N Singh Indian J Pure Appl Phys 21 165 (1983)
- [19] G Nonnenmacher and R Meche Spectrochim Acta (GB) 17 1049 (1961)
- [20] Tiers G V D J Chem Phys. 19 1072 (1951)
- [21] N B Colthup, L H Daly and S E Wiberley Introduction to Infrared and Raman Spectroscopy (New York: Academic) p370 (1975)
- [22] D N Singh, 1 D Singh and R A Yadav Indian J Pure Appl Phy. 76B 307 (2002)
- [23] N Fusion, C G Lagrange and M L Josien Spectrochim. Acta 16 106 (1960)
- [24] N L Owen and R E Hester Spectrochum Acta 25A 343 (1969)
- [25] M Harak, E R Lippincott and R K Khanna Spectrochum Acta 23A 1111 (1967)
- [26] J.J. Rush J. Chem. Phys. 47 3936 (1967)
- [27] K C Ingham and S J Strickler J Chem. Phys 53 4313 (1970)
- [28] J S H Green, W Kynaston and H A Gebbie Spectrochim Acta 19 807 (1963)
- [29] A P Upadhyay and K N Upadhyay Indian J. Phys 55B 215 (1981)
- [30] Y Kishore, S N Sharma and C P D Dwivedi Indian J Pure Appl Phys. 48 412 (1974)
- [31] J S H Green Spectrochim. Acta. 26A, 1925 (1970)
- [32] Shabbir Ahmad and P K Verma Indian J. Phys 62B 509 (1988)
- [33] D K Rai and R B Singh Indian J Phys. 60B 404 (1986)