

Vibrational spectra and force fields for 2,3-; 2,4-; 2,5- and 3,4-dihydroxybenzaldehydes

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Abstract Raman and IR spectra of 2,3-, 2,4-, 2,5- and 3,4-dihydroxy benzaldehydes have been recorded. The observed frequencies have been assigned to various normal modes of vibration in light of the normal coordinate calculation using Wilson's classical FG-matrix method. All the 42 normal modes have been assigned for all the molecules for the first time.

Keywords Vibrational spectra, force fields, normal modes

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1. Introduction

Phenol and its derivatives have been investigated extensively using vibrational spectroscopy [1-14]. Electronic and vibrational spectra of some tri-substituted benzenes have been studied and reported in literature [15-18]. Although several studies on the vibrational spectra of di-substituted phenols and benzaldehydes are available, complete interpretation of the entire spectra are available only for a very few of them.

The present work is in continuation of our vibrational studies on benzaldehyde derivatives [19] and deal with the vibrational spectral studies and force field calculations for 2,3-, 2,4-, 2,5- and 3,4-dihydroxy benzaldehydes (hereafter called DHBDs). To the best of our information, IR spectra of only 2,3-DHBD has so far been investigated. The purpose of this study is to (i) study the effect of the three substituents on the phenyl ring modes, (ii) study the effect of the OH group(s) on the internal modes of the CHO group and *vice-versa* and (iii) determine consistent force fields for these di-substituted benzaldehydes.

2. Experimental

All the four chemicals (purity > 97%) were purchased from the Aldrich Chemical Co., USA. These chemicals form

solids at room temperature and were used as such for recording the IR and the Raman spectra.

Raman spectra of these four compounds were recorded in the region 50–4000 cm^{-1} on a Spex model 1403 spectrometer equipped with a double monochromator and computer datamate. Samples were placed in a quartz cell and to excite these the 4880 Å line of an Ar^+ laser was used. The laser power on the samples was in the range 400–500 mW. The Raman frequencies are accurate within $\pm 1 \text{ cm}^{-1}$ and the resolution of the spectrometer under the present experimental conditions was of the order of 2 cm^{-1} .

IR spectra of the four compounds were recorded in KBr pellets in the region 180–4000 cm^{-1} on a Perkin Elmer 983 spectrometer equipped with a computer datamate.

3. Force field calculations

An approximate geometry has been used for these molecules to calculate *G*-matrix elements, as no structural studies for these are available. For the benzaldehyde part, structural parameters were same as in our earlier study and for the hydroxyl group, the parameters were taken from Ref. [11] and are as follows :

$$\begin{aligned} r(\text{C}-\text{OH}) &= 1.371 \text{ \AA}, & r(\text{O}-\text{H}) &= 0.81 \text{ \AA}, \\ \alpha(\text{C}-\text{O}-\text{H}) &= 108.5. \end{aligned}$$

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All the four molecules were assumed to belong to the C_1 point group symmetry. As these are 16 atomic molecules, the 42 normal modes of vibration are distributed between the two species of the C_1 point group as :

- (a) phenyl ring - $21a' + 9a''$,
- (b) CHO group - $4a' + 2a''$,
- (c) OH group - $2(2a' + 1a'')$.

The vibrational problems were set-up in terms of the internal coordinates. Using these internal coordinates, the symmetry coordinates were constructed in the way suggested in Refs. [20,21]. The F matrix was calculated assuming general valence force fields. The calculations were made using computer program of Schachtschneider [22]. The initial force constants were taken from our work on benzaldehyde [23] and the force constants related with the OH groups were taken by trial and error initially. In constructing the F matrix elements, total 68, 69, 70 and 69 planar and 26, 26, 27 and 27 non-planar force constants were taken for 2,3-, 2,4-, 2,5- and 3,4-DHBDs respectively. In constructing F -matrix, different numbers were assigned to the principal force constants for the two C–OH groups to see if there is any significant difference between the corresponding force constants of the two C–OH groups. After calculation, it was found that the force constants for the two C–OH groups in the case of the C–OH stretching and planar bending have nearly same values whereas for the C–OH non-planar bending the force constants for the two C–OH groups differ significantly. However, for the interaction force constants for the two C–OH groups no such calculations were made and so the same number was assigned to the interaction constants for the two C–OH groups. It was observed that some of the force constants had to be left either

due to their negligible magnitudes or due to their negligibly small contributions to the potential energy distributions (PEDs).

The force constants $\gamma(C-CHO)/\gamma(C-OH)$, $\nu(C-C)/\beta(C-OH)$, $\tau(C-CHO)/\tau(C-OH)$, $\phi(CCCC)/\tau(C-OH)$, $\tau(C-CHO)/\omega(C-CHO)$ and $\phi(CCCC)/\gamma(C-OH)$ were left for all the isomers due to drastic changes in their values and the disturbance of the entire set of the force constants during iteration. The force constants $\nu(O-H)/\nu(O-H)$ in 2,5-DHBD $\alpha(CHO)/\alpha(COH)$ in 2,4- and 2,5- DHBDs and $\beta(C-OH)/\beta(C-OH)$ in 2,5-DHBD were also omitted due to the same reasons. The force constants $\nu(C-CHO)/\nu(C-OH)^p$ and $\beta(C-CHO)/\beta(C-OH)^p$ in 2,4- and 3,4-DHBDs were kept fixed with their values as 0.0 due to their negligibly small contributions to the PEDs.

The remaining sets of the force constants were adjusted by trial and error method with the help of damping factor. The force constants and the fundamentals were computed at perturbation 1. The calculated sets of these force constants were used as the input and again the force constants and the fundamentals were computed. During the iteration process, all the planar principal force constants and some interaction force constants were re-adjusted one by one and the remaining force constants were kept fixed with values for benzaldehyde [23]. For non-planar force fields, all the force constants had to be re-adjusted one by one and finally, all the interaction force constants were kept fixed. After 3 to 6 such cycles, a good fit between the observed and the computed fundamentals were obtained. The final values of the force constants and their dispersions at perturbation 1 and their descriptions are collected in Table 1.

Table 1. Valence force constants for isomeric DHBDs.

Sl No.	Description [#]	2,3-DHBD [*]		2,4-DHBD [*]		2,5-DHBD [*]		3,4-DHBD [*]	
Planar principal force constants									
1.	$\nu(C-C)$ (R)	6.636	0.000	6.725	0.000	6.706	0.000	6.521	0.000
2.	$\nu(C-CHO)$ (t)	5.002	0.0	4.937	0.0	4.968	0.0	4.955	0.0
3.	$\nu(C-OH)$ (t_1)	3.946	0.221	3.863	0.186	4.124	0.139	4.094	0.149
4.	$\nu(C-H)$ (r)	5.087	0.0	5.042	0.0	4.996	0.0	5.042	0.0
5.	$\nu(C=O)$ (u)	9.144	0.0	9.179	0.0	9.074	0.0	9.113	0.0
6.	$\nu(C'-H)$ (v)	4.030	0.0	4.045	0.0	4.045	0.0	4.335	0.0
7.	$\alpha(CCC)$ (α)	1.409	0.0	1.179	0.0	1.416	0.0	1.168	0.0
8.	$\beta(C-CHO)$ (β_1)	1.567	0.0	1.658	0.0	1.725	0.0	1.308	0.0
9.	$\beta(C-OH)$ (β_2)	1.984	0.136	2.015	0.964	1.975	0.879	2.047	0.147
10.	$\beta(C-H)$ (β)	0.999	0.0	0.999	0.0	1.023	0.0	0.999	0.0
11.	$\beta(C=O)$ (θ_1)	1.964	0.0	2.022	0.0	1.977	0.0	1.938	0.0
12.	$\beta(C'-H)$ (θ_2)	1.239	0.0	1.138	0.0	1.170	0.0	1.188	0.0
13.	$\nu(O-H)$ (s_1, s_2)	5.838	0.042	5.551	0.047	6.001	0.069	5.941	0.053
14.	$\alpha(C-O-H)$ (θ_3)	0.351	0.123	0.360	0.021	0.365	0.035	0.361	0.099
15.	$\alpha(C-O-H)$ (θ_4)	0.231	0.224	0.235	0.021	0.232	0.035	0.234	0.071
16.	$\beta(C-OH)$ (β_3)	1.998	0.117	2.028	0.809	2.047	0.792	2.002	0.109
17.	$\nu(C-OH)$ (t_2)	3.889	0.264	3.804	0.164	4.199	0.136	3.961	0.122

Table 1. (Cont'd.)

Sl. No.	Description [#]	2,3-DHBD [*]		2,4-DHBD [*]		2,5-DHBD [*]		3,4-DHBD [*]	
Planar interaction force constants									
18.	$(RR)^o$	0.777	0.0	0.820	0.0	0.777	0.0	0.777	0.0
19.	$(RR)^m$	-0.349	0.0	-0.394	0.0	-0.349	0.0	-0.349	0.0
20.	$(RR)^p$	0.272	0.0	0.237	0.0	0.272	0.0	0.272	0.0
21.	$(rr)^o$	0.091	0.0	0.124	0.0	0.147	0.0	0.038	0.0
22.	$(rr)^m$	-0.003	0.0	-0.001	0.0	-0.032	0.0	-0.005	0.0
23.	$(\beta\beta)^o$	0.001	0.0	0.001	0.0	0.001	0.0	0.001	0.0
24.	$(\beta\beta)^m$	-0.054	0.0	0.026	0.0	-0.026	0.0	-0.026	0.0
25.	$(\alpha\alpha)^o$	-0.017	0.0	-0.017	0.0	-0.017	0.0	-0.017	0.0
26.	$(Rt)^o$	0.615	0.0	0.615	0.0	0.615	0.0	0.615	0.0
27.	$(Rt)^m$	0.208	0.0	0.208	0.0	0.208	0.0	0.208	0.0
28.	$(Rt)^p$	0.229	0.0	0.229	0.0	0.229	0.0	0.229	0.0
29.	$(R\alpha)^o$	0.404	0.0	0.404	0.0	0.404	0.0	0.404	0.0
30.	$(R\beta)^{o+}$	-0.264	0.0	-0.264	0.0	-0.264	0.0	-0.264	0.0
31.	$(R\beta)^{m+}$	-0.039	0.0	-0.039	0.0	-0.039	0.0	-0.039	0.0
32.	$(R\beta)^{p+}$	0.015	0.0	0.015	0.0	0.015	0.0	0.015	0.0
33.	$\beta_1\theta_2$	0.364	0.0	-0.364	0.0	-0.364	0.0	-0.364	0.0
34.	$\theta_1\theta_2$	0.475	0.0	0.475	0.0	0.475	0.0	0.475	0.0
35.	$\beta_1\theta_1$	-0.687	0.0	-0.687	0.0	-0.687	0.0	-0.687	0.0
36.	$t\theta_2$	-0.068	0.0	-0.068	0.0	-0.068	0.0	-0.068	0.0
37.	$(t\beta)^{o+}$	0.065	0.0	0.065	0.0	0.065	0.0	0.065	0.0
38.	$(t\beta)^{m+}$	0.058	0.0	0.058	0.0	0.058	0.0	0.058	0.0
39.	$(rr)^p$	-	-	-0.072	0.0	-0.072	0.0	-0.072	0.0
40.	ut_2	0.289	0.0	0.289	0.0	0.289	0.0	0.289	0.0
41.	$(\alpha\beta)^{o+}$	0.132	0.0	0.132	0.0	0.132	0.0	0.132	0.0
42.	$(Rt_1)^o = (Rt_2)^o$	0.274	0.241	0.432	0.399	0.505	0.249	0.139	0.275
43.	$(Rt_1)^m = (Rt_2)^m$	-0.415	0.251	-0.101	0.331	0.499	0.206	-0.041	0.235
44.	$(Rt_1)^p = (Rt_2)^p$	-0.026	0.022	-0.053	0.332	-0.239	0.274	-0.398	0.354
45.	$t_1\theta_3 = t_2\theta_4$	0.110	0.060	-0.013	0.354	0.207	0.396	0.137	0.901
46.	s_1s_2	-0.337	0.042	0.069	0.047	0.0	0.0	-0.229	0.053
47.	t_1t_2	0.060	0.072	-0.130	0.959	-0.279	0.468	0.064	0.358
48.	tt_1	0.226	0.096	0.225	0.107	0.233	0.115	-0.155	0.625
49.	tt_2	-0.249	0.115	0.0	0.0	-0.123	0.552	0.0	0.0
50.	$\beta_1\beta_2$	-0.093	0.049	0.402	0.307	-0.036	0.328	-0.244	0.444
51.	$\beta_1\beta_3$	0.054	0.056	0.0	0.0	-0.096	0.326	0.0	0.0
52.	$\theta_3\theta_4$	-0.048	0.038	-	-	-	-	-0.022	0.0
53.	$\beta_2\beta_3$	-0.208	0.341	-0.279	0.272	0.0	0.0	0.0	0.0
54.	$(\beta\beta)^p$	-	-	0.006	0.0	0.0	0.0	0.006	0.0
Non-planar principal force constants									
1	$\phi(\text{C}-\text{C}-\text{C}-\text{C}) (\phi)$	0.692	0.010	0.073	0.288	0.069	0.029	0.064	0.022
2	$\gamma(\text{C}-\text{OH}) (\delta_2)$	0.250	0.103	0.241	0.832	0.238	0.349	0.252	0.920
3	$\gamma(\text{C}-\text{H}) (\delta)$	0.318	0.014	0.314	0.029	0.329	0.032	0.315	0.199
4	$\tau(\text{C}-\text{CHO}) (\tau)$	0.009	0.0	0.009	0.0	0.009	0.0	0.009	0.0
5	$\gamma(\text{C}-\text{CHO}) (\delta_1)$	0.285	0.298	0.282	0.125	0.283	0.402	0.294	0.329
6	$\omega(\text{CHO}) (\omega)$	0.417	0.024	0.379	0.014	0.395	0.049	0.398	0.141
7	$\tau(\text{C}-\text{OH}) (\tau_1)$	0.029	0.002	0.027	0.001	0.027	0.002	0.022	0.003
8	$\tau(\text{C}-\text{OH}) (\tau_2)$	0.025	0.002	0.019	0.001	0.024	0.002	0.026	0.008
9	$(\text{C}-\text{OH}) (\delta_3)$	0.203	0.290	0.204	0.099	0.205	0.290	0.205	0.330

Table 1. (Cont'd)

Sl No.	Description [#]	2,3-DHBD [*]		2,4-DHBD [*]		2,5-DHBD [*]		3,4-DHBD [*]	
Non-planar interaction force constants									
10	$(\phi\phi)''$	-0.012	0.0	-0.012	0.0	0.012	0.0	-0.012	0.0
11	$(\delta\delta)''$	0.001	0.0	0.001	0.0	0.001	0.0	0.001	0.0
12	$(\delta\delta)'''$	-0.001	0.0	-0.001	0.0	0.001	0.0	-0.001	0.0
13	$(\phi\delta)''$	0.019	0.0	0.019	0.0	0.019	0.0	0.019	0.0
14	$\phi_1\delta_1 = -\phi_6\delta_1$	0.010	0.0	0.010	0.0	0.010	0.0	0.010	0.0
15	$(\delta\delta_1)''$	0.001	0.0	0.001	0.0	0.001	0.0	0.001	0.0
16	$(\delta\delta_1)'''$	-0.001	0.0	-0.001	0.0	0.001	0.0	-0.001	0.0
17	$(\delta\delta_1)''$	-0.001	0.0	-0.001	0.0	0.001	0.0	-0.001	0.0
18	$\delta_1\omega$	0.001	0.0	0.001	0.0	0.001	0.0	0.001	0.0
19	$(\phi\tau)''$	0.001	0.0	0.001	0.0	0.001	0.0	0.001	0.0
20	$\phi_1\omega = -\phi_6\omega$	0.001	0.0	0.001	0.0	0.001	0.0	0.001	0.0
21	$(\phi\tau_1)'' = (\phi\tau_2)''$	0.007	0.0	0.007	0.0	0.007	0.0	0.007	0.0
22	$\delta_2\delta_3$	-0.027	0.077	-0.015	0.04	0.025	0.032	-0.032	0.083
23	$(\delta\delta)''$					0.001	0.0	-	-

ν : Stretching, α : Angle bending, β : In-plane bending, ϕ : Ring-torsion, γ : Out-of-plane bending, τ : Torsion of a top, ω : Wagging; o, m, p stand for ortho, meta and para respectively

† $(R/\beta)''$ means $R_i\beta_i = -R_{i+1}\beta_i$. Similar convention is adopted for all interaction terms bearing symbol †

* Units of force constants are as follows: mdyne/Å for stretch and stretch-stretch interaction, mdyne/rad for stretch-band interaction and mdyne Å/rad² for the other

4. Results and discussion

All the frequencies observed in the IR and Raman spectra together with their relative intensities, the calculated frequencies corresponding to the fundamentals along with the PEDs and the proposed mode assignments are presented in Tables 2–5 for the 2,3-, 2,4-, 2,5- and 3,4-DHBDs molecules respectively. The presently proposed vibrational assignments are based on the intensities and magnitudes of the observed frequencies and PEDs for the calculated fundamentals. Assistance has also been taken from the vibrational assignments proposed for benzene derivatives containing the CHO [24–34] and the OH [1,10,12 and 15] groups. The discussion of the vibrational mode assignments has been divided into the following three sections: (i) the phenyl ring modes, (ii) the CHO group modes and (iii) the OH group modes.

(i) Phenyl ring modes

As the phenyl ring modes are mostly well discussed in literature and can be easily assigned, here only controversial modes are discussed in the following.

The ring-breathing the trigonal planar ring-bending the Kekule C=C stretching and the umbrella C–H non-planar bending modes are some of the controversial and substituent sensitive ring modes.

In di-substituted and symmetrically tri-substituted (1,3,5-C₆H₃X₃) benzenes the ring-breathing and the trigonal planar ring-bending modes usually give rise to Raman lines with

characteristic features. However, in non-symmetrically tri-substituted benzenes no such lines are observed. For 1,3-dibromo-5-fluorobenzene the ring breathing and the trigonal bending are assigned at 519 and 996 cm⁻¹ whereas for 2,4-di-bromo-1-fluorobenzene these modes are assigned at 1018 and 596 cm⁻¹ by Belgaum *et al* [15]. Rastogi *et al* [16] have assigned the frequencies 1078 and 570 cm⁻¹ respectively to these modes. Varsanyi [35] has argued that for the 1,2,3-tri-substituted benzenes the trigonal bending mode has higher magnitude than the ring breathing mode whereas reverse is the case for the other tri-substituted benzenes. He has further argued that, owing to strong mixing of different symmetrically permissible modes, the two frequencies cannot be uniquely labelled as a C–C stretching mode and a C–C–C angle bending modes.

In the present case, in light of the force field calculations the frequencies 638, 758, 780 and 766 cm⁻¹ could be assigned to the ring breathing mode for the 2,3-, 2,4-, 2,5- and 3,4-DHBDs respectively. Similarly, the trigonal ring bending mode is assigned at 840, 646, 713, 643 cm⁻¹ respectively for the isomers 2,3-, 2,4-, 2,5- and 3,4-DHBDs. It is to be noted here that the presently assigned frequencies for the ring breathing and trigonal bending modes fall in the spectral ranges suggested by Varsanyi [35] for all the four isomeric HBDs.

The Kekule C=C stretching mode appears in the region 1200–1400 cm⁻¹ in substituted benzenes. The force field calculations suggest that this is a pure C=C stretching mode.

Table 2. Observed and calculated frequencies for 2,3-DHBD[#].

Observed				Calculated		Proposed assignments	Species
Raman		Infrared		cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel. Int.	cm ⁻¹	Rel. Int.	cm ⁻¹			
—		3327	(9)	3327	13(94)	O-H stretch	a'
3140	(1)	—		3140	13(106)	O-H stretch	a'
3101	(1)	—		3100	4(97)	C-H stretch	a'
3060	(5)	3062	(9)	3062	4(99)	C-H stretch	a'
3018	(1)	3024	(9)	3025	4(101)	C-H stretch	a'
—		2978	(8)			1628 + 1355	A'
—		2878	(8)			1650 + 1234	A'
—		2725	(8)	2725	6(99)	C-H stretch CHO group	a'
1774	(1)	—				1628 + 148	A''
1754	(1)	1746	(3)			1420 + 330	A'
1696	(2)	—		1695	5(31), 1(22), 2(24), 1(22), 11(10), 1(7)	C=O stretch (II)	a'
1657	(7)	1650	(10)				
1628	(2)	1611	(9)	1618	1(64), 5(13), 12(12), 7(9), 10(7)	C=C stretch	a'
1597	(2)	1586	(9)	1586	1(81), 7(14), 10(14)	FR between 1585.	a'
1574	(3)					C-C stretch	
						and 1243 + 330	
1481	(1)	1481	(9)	1475	1(47), 10(32)	C=C stretch	a'
1465	(1)	—		1461	10(45), 1(43)	C=C stretch	a'
1417	(3)	1397	(8)	1394	12(45), 5(36), 11(34)	C-H i.p.b. CHO group	a'
1355	(2)	1355	(8)	1344	1(142), 10(16)	C-C stretch	a'
1292	(3)	1280	(9)	1289	7(34), 3(16), 17(14)	C-OH stretch	a'
		1261	(8)			783 + 482	A'
1243	(5)	1235	(10)	1240	10(35), 1(28), 3(13), 17(11)	C-OH stretch	a'
1204	(1)	—		1207	10(51), 1(25)	C-H i.p.b.	a'
1176	(5)	1161	(8)	1170	10(55), 1(23)	C-H i.p.b.	a'
1143	sh					790 + 350	A'
1129	sh					638 + 494	A'
1080	(5)	1076	(5)	1065	1(46), 10(26)	C-H i.p.b.	a'
		1063	(5)			733 + 330	A''
1031	(1)	—		1035	14(55), 1(24)	C-O-H angle bending	a'
		1021	(4)	1022	3(73), 1(20), 6(18)	C-H o.p.b.	a''
988	(0)	—		994	6(70), 3(21), 1(8)	CHO wagging	a''
976	(1)	969	(8)	972	17(15), 1(14), 2(14), 7(10)	C-CHO stretch	a'
902	(1)	909	(4)	904	3(84), 1(26)	C-H o.p.b.	a''
840	(2)	840	(6)	825	7(58), 3(13)	C-C-C i.p.b.	a'
800	(1)	795	(5)	780	3(82), 1(14)	C-H o.p.b.	a''
790	(1)	783	(8)	781	15(82), 14(14)	C-O-H angle bending	a'
751	(0)	754	(6)			550 + 199	A'
724	(1)	733	(8)	700	1(26), 2(21), 5(13), 3(12)	C-C-C-C torsion	a''
649	(1)					450 + 199	A'
638	(3)	632	(7)	638	1(26), 3(17), 7(13), 17(11)	C=C stretch	a'
620	(1)	622	(6)	620	7(92)	C-OH torsion	a''
584	(1)	590	(5)	583	8(85), 1(6)	C-OH torsion	a''
560	(3)	550	(7)	546	7(34), 11(11), 1(13), 8(10), 9(10)	C-C-C i.p.b.	a'

Table 2. (Cont'd)

Observed				Calculated		Proposed assignments	Species
Raman		Infrared		cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel Int	cm ⁻¹	Rel Int	cm ⁻¹	Potential energy distribution*		
500	(2)	497	(4)	490	2(29), 1(24), 9(12), 3(8)	C-OH o.p.b.	
494	(10)	482	(4)	480	16(34), 9(23), 1(15), 7(10)	C-OH i.p.b.	
450	(9)	-		434	7(48), 17(19), 3(14), 1(8), 8(8)	C-C-C i.p.b.	
		442	(6)	447	5(40), 9(29), 1(23)	C-OH o.p.b.	
350	(0)	351	(3)	335	11(44), 8(18), 2(15), 7(15), 12(7)	C=O i.p.b.	
330	(1)	328	(3)	315	9(35), 16(33), 1(13)	C-OH i.p.b.	
253	(1)			256	1(45), 4(30), 1(7)	C-C-C-C torsion	
234	(1)			233	2(36), 9(35), 1(18), 5(14)	C-CHO o.p.b.	
199	(2)			180	8(64), 1(8)	C-CHO i.p.b.	
148	(2)			172	1(52), 2(29)	C-C-C-C torsion	
136	(0)					Lattice mode	
110	sh			108	1(42), 4(34), 5(11), 7(9)	C-CHO torsion	

* Abbreviations used . Rel. int. = Relative intensity; sh. = shoulder; i.p.b. = in-plane bend; o.p.b. = out-of-plane bend; FR = Fermi resonance.

* The numbers outside the brackets are the force constants number defined in Table 1 and those within the brackets are the corresponding contributions

Table 3. Observed and calculated frequencies for 2,4-DHBD[#]

Observed				Calculated		Proposed assignments	Species
Raman		Infrared		cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel Int	cm ⁻¹	Rel Int	cm ⁻¹	Potential energy distribution*		
3174	(2)			3174	13(99)	O-H stretch	
3135	(1)	3123	(9)	3125	13(101)	O-H stretch	
3088	(2)			3091	4(96)	C-H stretch	
3053	sh	3055	(9)	3050	4(99)	C-H stretch	
3001	(1)	3028	(9)	3004	4(102)	C-H stretch	
-		2730	(7)	2730	6(99)	C-H stretch CHO group	
1643	(4)	1630	(10)	1676	5(36), 2(25), 1(20), 12(18)	C=O stretch (II) } C=O stretch (I) }	
1627	(6)	1613	(10)				
1590	(6)	1597	(9)	1614	1(82), 7(10)	C=C stretch	a'
		1576	(9)	1588	1(69), 10(19), 7(10)	C=C stretch	a'
1545	(0)	-				1130 + 410	A''
1511	(3)	1494	(10)	1493	10(43), 1(38), 5(9), 12(5)	C=C stretch	a'
1477	(1)					1398 + 79	A'
1465	(1)					1428 + 136	A'
1442	sh.	1441	(9)	1444	1(55), 10(22), 3(6), 7(6)	C=C stretch	a'
-		1428	(9)			1129 + 301	A'
1398	(5)	1393	(9)	1375	12(48), 11(40), 5(30)	C-H i.p.b. CHO group	a'
1340	(6)	1328	(10)	1332	1(142), 10(16)	C=C stretch	a'
1239	(5)	1230	(10)	1230	10(75), 1(17)	C-H i.p.b.	a'
1223	(5)			1220	7(33), 3(21), 17(17), 1(15), 10(6)	C-OH stretch	a'
1185	(3)					FR between 1165, C-CHO stretch and 758 + 410	a'
1145	(5)	1164	(10)	1159	1(39), 2(27), 10(14)		
1130	(0)	1129	(10)	1139	10(54), 1(23), 7(6)	C-H i.p.b.	a'
1110	(0)					758 + 361	A'
1088	(0)			1106	10(32), 1(19), 17(17), 3(10)	C-H i.p.b.	a'
1035	(0)			1031	14(64), 1(16), 7(13)	C-O-H angle bending	a'
1000	(0)	1010	(5)	995	3(76), 1(22), 6(13)	C-H o.p.b.	a''

Table 3. (Cont'd.).

Observed				Calculated		Proposed assignments	Species
Raman		Infrared		cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel. Int.	cm ⁻¹	Rel. Int.	cm ⁻¹	Potential energy distribution*		
994	(2)	973	(7)	963	1(26), 17(14), 7(13), 3(10), 15(10)	C-OH stretch	a'
936	(0)	-		950	6(77), 3(14), 1(6)	CHO wagging	a''
907	(0)	-		920	3(86), 1(22)	C-H o.p.b	a''
860	(0)	856	(7)			699 + 160	A'
838	(3)	823	(7)	826	15(72), 7(13)	C-O-H angle bending	a'
817	(0)	805	(8)	815	3(85), 1(20)	C-H o.p.b.	a''
758	(5)	756	(6)	780	1(28), 7(16), 3(13), 1(11)	C=C stretch	a'
738	(0)	726	(6)			425 + 301	A'
699	(0)	693	(7)	670	1(35), 2(17), 5(20)	C-C-C torsion	a''
666	(0)					566 + 99	A''
646	(8)	640	(9)	660	7(49), 17(14), 2(7), 1(6)	C-C-C i.p.b	a'
599	(0)	597	(7)	599	7(87), 1(6)	C-OH torsion	a''
566	(1)	551	(6)	540	16(31), 9(23), 8(19), 1(15)	C-OH i.p.b	a'
509	(1)	-		507	8(90)	C-OH torsion	a''
497	(1)	497	(7)	480	7(50), 3(23), 17(12), 1(7)	C-C-C i.p.b	a'
474	(0)	458	(5)	480	2(27), 9(18), 1(18), 5(12), 7(6)	C-OH o.p.b	a''
439	(6)	425	(6)	445	7(40), 11(22), 12(9), 17(9), 16(8), 9(6)	C-C-C i.p.b	a'
410	(1)	-		404	1(37), 9(16), 2(13), 5(13), 3(6)	C-OH o.p.b	a''
383	(0)	383	(4)			248 + 136	A''
361	(0)	367	(5)	346	16(36), 9(24), 1(23)	C-OH i.p.b	a'
		344	(4)			248 + 99	A'
301	(0)	-		307	11(34), 7(31), 2(14), 8(13), 12(7)	C=O i.p.b.	a'
		280	(4)	309	5(32), 1(34), 9(19), 2(5), 3(5)	C-CHO o.p.b	a''
248	(1)	-		218	1(45), 4(18), 2(10), 9(5)	C-C-C-C torsion	a''
160	(2)	-		173	8(75), 9(27), 1(7)	C-CHO i.p.b.	a'
-		-		193	4(36), 1(20), 2(14), 9(14), 3(8)	C-C-C-C torsion	a''
136	(1)					Lattice mode	
122	(1)	-		99	1(44), 4(28), 5(11), 3(5)	C-CHO torsion	a''
99	(10)					Lattice mode	
79	(9)					Lattice mode	

Notations as explained in Table 2

Table 4. Observed and calculated frequencies for 2,5-DHBD[#].

Observed				Calculated		Proposed assignments	Species
Raman		Infrared		cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel. Int.	cm ⁻¹	Rel. Int.	cm ⁻¹	Potential energy distribution*		
-		3861	(5)			3090 + 772	A'
-		3710	(4)			2990 + 719	A'
-		3280	(9)	3280	13(100)	O-H stretch	a'
-		-		3280	13(100)	O-H stretch	a'
3090	(2)	-		3088	4(96)	C-H stretch	a'
3027	(3)	-		3029	4(99)	C-H stretch	a'
2991	(3)	-		2990	4(102)	C-H stretch	a'
		2876	(6)			1486 + 1380	A'
		2772	(5)			1486 + 1281	A'
		2730	(7)	2730	6(99)	C-H stretch CHO group	a'
1749	(1)	1756	(3)			1430 + 320	A'
		1722	(4)			952 + 772	A'

Table 4. (Cont'd)

Observed		Observed		Calculated		Proposed assignments	Species
Raman	Infrared	Raman	Infrared	cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel. Int.	cm ⁻¹	Rel. Int.	cm ⁻¹	Potential energy distribution*		
1661 (2)	1660 (7)	1687	1(32), 5(24), 2(24), 12(14), 7(10)	1687	1(32), 5(24), 2(24), 12(14), 7(10)	C=O stretch (II)	a'
1642 (2)	1646 (9)					C=O stretch (I)	
1631 (2)	1623 (9)	1632	1(63), 5(15), 12(10), 7(9)	1632	1(63), 5(15), 12(10), 7(9)	C=C stretch	a'
1579 (10)	1577 (10)	1584	1(69), 10(22), 7(13), 5(8)	1584	1(69), 10(22), 7(13), 5(8)	C=C stretch	a'
1485 (3)	1486 (10)	1489	10(43), 1(38), 5(9), 12(5)	1489	10(43), 1(38), 5(9), 12(5)	C=C stretch	a'
1422 (1)	1430 (9)	1449	1(55), 10(16), 17(9), 7(7)	1449	1(55), 10(16), 17(9), 7(7)	C=C stretch	a'
	1397 (6)					FR between 1389,	a'
1385 (2)	1380 (6)	1376	12(48), 11(38), 5(33)	1376	12(48), 11(38), 5(33)	C-H i.p.b. CHO group and 1281 + 109	
1365 (2)	1359 (7)	1350	1(133)	1350	1(133)	C=C stretch	a'
1302 (2)	1300 (10)	1310	7(48), 17(16), 10(14), 3(6)	1310	7(48), 17(16), 10(14), 3(6)	C-OH stretch	a'
1285 (2)	1281 (10)	1278	10(54), 1(17), 3(5), 7(5)	1278	10(54), 1(17), 3(5), 7(5)	C-H i.p.b.	a'
1261 (4)	1251 (8)	1257	1(38), 3(24), 10(11), 17(10)	1257	1(38), 3(24), 10(11), 17(10)	C-OH stretch	a'
1190 (1)						FR between 1179,	a'
1167 (2)	1160 (9)	1146	10(54), 1(29)	1146	10(54), 1(29)	C-H i.p.b. and 682 + 499	
1118 (2)	-	1111	10(39), 1(23), 2(12), 17(13)	1111	10(39), 1(23), 2(12), 17(13)	C-H i.p.b.	a'
1030 (2)	-	1010	3(89), 1(22)	1010	3(89), 1(22)	C-H o.p.b.	a''
1017 (2)	-	1008	14(92), 1(8)	1008	14(92), 1(8)	C-H-O angle bending	a'
-	1000 (4)	985	6(70), 3(20), 1(5)	985	6(70), 3(20), 1(5)	CHO wagging	a''
954 (2)	952 (5)	952	1(34), 7(21), 2(18), 10(7)	952	1(34), 7(21), 2(18), 10(7)	C-CHO stretch	a'
895 (1)	-	920	3(70), 6(21), 1(17)	920	3(70), 6(21), 1(17)	C-H o.p.b.	a''
-	881 (5)					487 + 393	A'
-	865 (5)					682 + 179	A''
-	832 (7)	832	3(88), 1(20)	832	3(88), 1(20)	C-H o.p.b.	a''
809 (1)	803 (9)	809	15(91)	809	15(91)	C-O-H angle bending	a'
780 (5)	772 (7)	780	1(31), 7(30), 3(14), 15(9)	780	1(31), 7(30), 3(14), 15(9)	C=C stretch	a'
713 (1)	719 (6)	699	7(33), 17(24), 3(21)	699	7(33), 17(24), 3(21)	C-C-C i.p.b.	a'
670 (1)	682 (8)	674	1(37), 2(18), 5(16)	674	1(37), 2(18), 5(16)	C-C-C-C torsion	a''
-	651 (8)					499 + 152	A''
-	-	604	7(32), 11(17), 12(11), 1(10), 8(10), 9(9)	604	7(32), 11(17), 12(11), 1(10), 8(10), 9(9)	C-C-C i.p.b.	a'
-	597 (5)	597	7(89), 1(5)	597	7(89), 1(5)	C-OH torsion	a''
-	568 (5)	568	8(94)	568	8(94)	C-OH torsion	a''
499 (4)	-	503	16(38), 9(26), 7(14), 1(5)	503	16(38), 9(26), 7(14), 1(5)	C-OH i.p.b.	a'
-	487 (7)	462	5(33), 9(20), 1(18), 2(12)	462	5(33), 9(20), 1(18), 2(12)	C-OH o.p.b.	a''
444 (10)	439 (5)	432	7(46), 17(22), 3(21), 1(16)	432	7(46), 17(22), 3(21), 1(16)	C-C-C i.p.b.	a'
-	393 (3)	402	1(39), 2(20), 9(19), 3(5)	402	1(39), 2(20), 9(19), 3(5)	C-OH o.p.b.	a''
380 (2)	373 (3)	356	9(31), 11(26), 16(15), 1(11), 12(6)	356	9(31), 11(26), 16(15), 1(11), 12(6)	C-OH i.p.b.	a'
-	346 (3)					234 + 109	A'
320 (2)	-	309	16(24), 11(21), 1(15), 8(13), 2(10), 9(7)	309	16(24), 11(21), 1(15), 8(13), 2(10), 9(7)	C=O i.p.b.	a'
290 (1)	-	315	1(33), 2(24), 9(22), 3(5)	315	1(33), 2(24), 9(22), 3(5)	C-C-C-C torsion	a''
244 (4)	-	234	4(43), 1(20), 2(11), 9(11), 5(10)	234	4(43), 1(20), 2(11), 9(11), 5(10)	C-CHO o.p.b.	a''
179 (1)	-	191	8(62), 9(9), 1(7)	191	8(62), 9(9), 1(7)	C-CHO i.p.b.	a'
152 (0)	-	151	1(59), 3(9), 2(9)	151	1(59), 3(9), 2(9)	C-C-C-C torsion	a''
109 (6)	-	112	4(46), 1(28), 5(13)	112	4(46), 1(28), 5(13)	C-CHO torsion	a''
79 (2)						Lattice mode	

*Notations as explained in Table 2.

Table 5. Observed and calculated frequencies for 3,4-DHBD[#].

Observed				Calculated		Proposed assignments	Species
Raman		Infrared		cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel. Int.	cm ⁻¹	Rel. Int.	cm ⁻¹			
3349	(2)	3326	(9)	3326	13(96)	O-H stretch	a'
3200	(2)	-		3200	13(103)	O-H stretch	a'
-		-		3072	4(97)	C-H stretch	a'
3054	(2)	-		3047	4(99)	C-H stretch	a'
3017	(1)	-		3023	4(100)	C-H stretch	a'
-		2825	(5)	2825	6(99)	C-H stretch CHO group	a'
1660	(10)	1656	(9)	1676	5(37), 2(25), 12(22), 4(15)	C=O stretch	a'
1608	(10)	-		1607	1(79), 7(10)	C=C stretch	a'
1595	(10)	1595	(10)	1587	1(69), 10(16), 7(11)	C=C stretch	a'
1540	(2)	1530	(6)	1511	1(38), 10(37), 5(13)	C=C stretch	a'
1450	(9)	1442	(10)	1430	1(54), 10(18), 3(8), 7(7)	C=C stretch	a'
-		1420	(6)	-	-	1302 + 114	A''
1394	(1)	1388	(7)	1375	12(50), 11(37), 5(32)	C-H i.p.b. CHO group	a'
1355	(1)	-		-	-	1202 + 155	A'
1302	(3)	1298	(10)	1321	1(146), 10(15)	C=C stretch	a'
1284	sh	-		-	-	1175 + 106	A'
1276	(1)	-		1265	10(79), 1(9)	C-H i.p.b.	a'
1202	(3)	1192	(8)	1212	7(33), 10(21), 1(10), 3(10), 17(10), 2(7)	C-H i.p.b.	a'
1175	(7)	1167	(10)	1178	10(35), 1(28), 3(15), 17(17)	C-OH stretch	a'
1130	(3)	1119	(6)	1134	1(32), 10(32), 2(13), 7(9)	C-CHO stretch	a'
1065	(3)	-		1072	10(28), 1(24), 2(19), 7(14), 17(11), 3(5)	C-H i.p.b.	a'
1040	(3)	-		1023	3(77), 1(14), 6(6)	C-H o.p.b.	a''
1018	(3)	-		1018	14(88), 1(10)	C-H-O angle bending	a'
986	(3)	974	(3)	950	1(44), 7(16), 3(11), 2(6), 10(6)	C-OH stretch	a'
966	(1)	-		979	6(48), 3(44), 1(12)	CHO wagging	a''
944	(1)	-		936	3(65), 6(27), 1(12)	C-H o.p.b.	a''
-		877	(5)	-	-	766 + 114	A''
847	(1)	-		843	3(89), 1(16)	C-H o.p.b.	a''
823	(2)	813	(6)	813	15(93)	C-O-H angle bending	a'
-		776	(3)	-	-	670 + 106	A''
766	(3)	755	(5)	735	1(27), 7(32), 3(17), 17(9)	C=C stretch	a'
724	(0)	-		-	-	643 + 74	A'
674	(0)	670	(4)	663	1(38), 2(18), 9(14), 5(8)	C-C-C torsion	a''
643	(4)	631	(6)	653	7(35), 17(23), 2(12), 1(8), 12(6), 11(5)	C-C-C i.p.b.	a'
606	(1)	592	(4)	592	8(83), 1(7)	C-OH torsion	a''
543	(1)	-		543	7(94)	C-OH torsion	a''
515	(2)	-		524	16(32), 7(22), 9(16), 1(12), 3(11)	C-OH i.p.b.	a'
488	(1)	-		478	7(33), 11(13), 1(10), 12(6), 17(5)	C-C-C i.p.b.	a'
-		-		503	2(33), 9(15), 1(10), 8(10)	C-OH o.p.b.	a''
465	(1)	452	(3)	-	-	2 * 235	A'
419	(3)	410	(3)	429	7(38), 8(18), 11(18), 3(11), 9(7)	C-C-C i.p.b.	a'
402	(1)	390	(3)	401	1(46), 2(12), 5(11), 3(6), 9(5)	C-OH o.p.b.	a''
-		-		318	7(30), 11(20), 9(15), 16(9), 1(6)	C=O i.p.b.	a'
-		362	(2)	-	-	290 + 74	A'

Table 5. (Cont'd.)

Observed				Calculated		Proposed assignments	Species
Raman		Infrared		cm ⁻¹	Potential energy distribution*		
cm ⁻¹	Rel. Int	cm ⁻¹	Rel. Int	cm ⁻¹			
290	(0)	—	—	297	16(27), 9(27), 11(18), 1(13), 7(10), 8(10)	C—OH i.p.b.	a'
270	(0)	—	—	280	9(28), 1(33), 5(13), 2(10)	C—C—C torsion	a''
253	(1)	—	—			Lattice mode	
235	(1)	—	—	214	1(46), 2(18), 5(11), 3(7), 9(7)	C—CHO o.p.b.	a''
—	—	—	—	192	4(54), 1(25), 9(7)	C—C—C torsion	a''
155	(3)	—	—	186	8(75), 1(10)	C—CHO i.p.b.	a'
114	(3)	—	—	102	1(47), 4(26), 5(10)	C—CHO torsion	a''
106	(3)	—	—			Lattice mode	
74	(5)	—	—			Lattice mode	

Notations as explained in Table 2

For the three isomeric methoxy benzaldehydes [MBDs] this mode is assigned in the range 1280–1310 cm⁻¹ [20]. In the present case, the frequencies 1355, 1340, 1365 and 1302 cm⁻¹ are assigned to the Kekule mode for the 2,3-; 2,4-; 2,5- and 3,4-DHBDs respectively. It is to be noted that the PED for this mode has only contribution from the ring C—C stretching force constant for the 2,5-isomer whereas for the rest three there is a very small contribution from the planar ring C—H bending force constants, in addition to large contribution from the ring C—C stretching force constant.

The umbrella mode has been observed as a strong IR band in benzene at 670 cm⁻¹ [36]. As mentioned earlier its magnitude is substituent sensitive. For the isomeric MBDs this mode is assigned in the range 740–775 cm⁻¹. In the present case this mode is assigned in the range 850–780 cm⁻¹ for all the four isomers.

In addition to the above four phenyl ring modes the two C—OH and the C—CHO stretching modes also deserve some discussion. For the isomeric MBDs [20], the C—CHO stretching mode is assigned in a very narrow frequency range 1150–1170 cm⁻¹. However, in the present case the C—CHO stretching mode appears to be affected drastically for the 2,3- and 2,5-DHBDs for which it appears in the range 950–980 cm⁻¹ whereas for the 2,4- and 3,4-DHBDs this mode has magnitude very close [1130–1170 cm⁻¹] to that for the isomeric MBDs [20]. It is interesting to note that in case of the isomers which have the C—CHO stretching mode in the region 950–980 cm⁻¹ [2,3- and 2,5-isomers], the two C—OH stretching modes appear in the region 1240–1305 cm⁻¹ whereas in case of those isomers which have the C—CHO stretching mode in the region 1130–1170 cm⁻¹ [2,4- and 3,4-isomers], one of the two C—OH stretching frequency appears in the region 1170–1240 cm⁻¹ and the other C—OH stretching mode appears in the region 980–1000 cm⁻¹.

(ii) CHO group modes :

Out of the six CHO internal modes, the CH stretching mode has its characteristic magnitude in the range 2700–2850 cm⁻¹. In the present case, this mode is assigned at 2725, 2730 and 2730 cm⁻¹ for the 2,3-; 2,4- and 2,5-DHBDs whereas for the 3,4-DHBD it is observed at 2825 cm⁻¹. For 2,3-; 2,4- and 2,5-DHBDs there appear two frequencies whereas for 3,4-DHBD only one frequency is observed in the C=O stretching region. All these four molecules may have two conformations (Figures 1 and 2). It can be seen that for the 3,4-DHBD, the CHO group modes are expected to

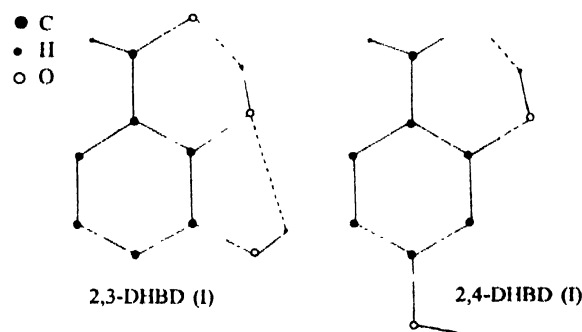


Figure 1a. 2,3-DHBD (I); 2,4-DHBD (I)

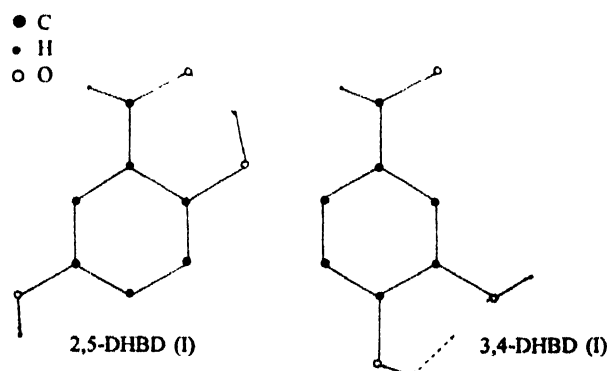


Figure 1b. 2,5-DHBD (I); 3,4-DHBD (I)

have the same magnitudes in the two conformations, as the CHO group is not expected to be involved in intramolecular hydrogen bonding. However, for the other three isomers the CHO group is expected to be involved in the intramolecular

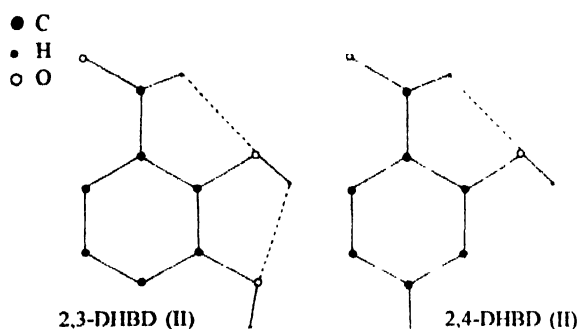


Figure 2a. 2,3-DHBD (II); 2,4-DHBD (II)

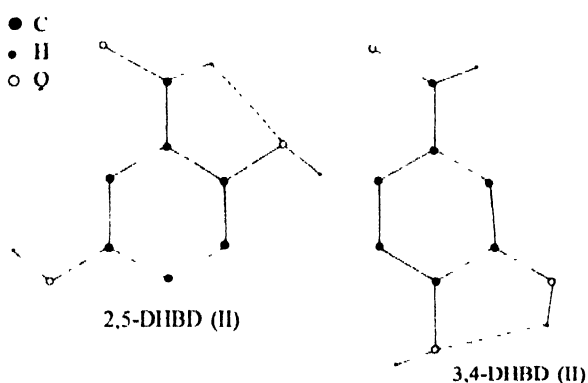


Figure 2b. 2,5-DHBD (II); 3,4-DHBD (II)

hydrogen bonding and the magnitude of the (C=O) mode in the configuration II is expected to have higher magnitude compared to that in the configuration I because the O atom of the CHO group in the configuration II is non-bonded whereas that in configuration I is bonded with the H atom of the OH group. Hence, we assign the frequencies 1696, 1643 and 1660 cm^{-1} to the (C=O) mode corresponding to the configuration II and the frequencies 1657, 1627 and 1646 cm^{-1} corresponding to the configuration I for the 2,3-, 2,4- and 2,5-DHBDs respectively. One could expect similar behaviour for the CHO group CH stretching mode. However, the region of the CH stretching mode is masked by the presence of strong $\nu(\text{O-H})$ modes of the OH groups. The other CHO group modes fall in the regions crowded by the other modes and so behaviour similar to the $\nu(\text{C=O})$ mode is not expected to be obvious for these modes.

The C-H planar bending mode is assigned in the region 1370–1420 cm^{-1} . It is a pure group mode as all the force constants involved in the PED of this mode are the CHO group force constants only. The CH non-planar bending mode of the CHO group appears in the region 900–1010 cm^{-1} . In the present work, this mode is assigned in the region

930–1100 cm^{-1} . This shows some mixing with other ring non-planar modes.

The C=O planar bending is observed in the region 550–625 cm^{-1} for the isomeric MBDs [20]. It has been observed in the neighbourhood of 650 cm^{-1} for the isomeric trifluoromethyl benzaldehydes [23]. In the present case, this mode is assigned in the region 300–350 cm^{-1} for all the isomers. However, due to strong mixing of this mode with the other planar modes unambiguous labelling is not possible. Hence, no conclusion could be drawn from the assignment of this mode regarding the hydrogen bonding. As expected, the CHO torsional mode has lowest magnitude of all the modes. It is observed in the region 100–120 cm^{-1} for all the four isomers. It shows mixing with the ring torsion and CHO wagging mode for all the molecules.

(iii) OH group modes :

An OH group has three normal modes of vibration, namely, the OH stretching, the C–O–H angle bending and the C–OH torsion. For substituted phenols, a very sharp line like band appears in the neighbourhood of 3600 cm^{-1} in very dilute solution prepared using non-polar solvents if intramolecular hydrogen bonding is absent [13]. In bonded form, a broad and intense band appears in the region 2500–3500 cm^{-1} . Presently, the region 2000–3000 cm^{-1} is masked by a broad and intense band for all the four isomers. This band is obviously due to the $\nu(\text{O-H})$ mode in bonded form. The magnitudes of the two $\nu(\text{O-H})$ modes could be ascertained on the basis of calculation.

It is also to be mentioned that for an OH containing benzene derivatives, the C–OH stretching and C–O–H angle bending modes interact strongly to give rise to two frequencies in the range 1100–1350 cm^{-1} . However, in the present case, the calculations do not indicate the presence of any such interaction for all the four isomers. However, it may be noted that the two C–O–H angle bending modes appear to interact for the 2,3-isomer only, as for this molecule one of the two C–O–H angle bending frequencies involves constants of both the C–O–H angle bending motions. For the remaining three isomers the two C–O–H angle bending modes do not interact. It appears that the interaction between the two C–O–H angle bending modes is induced by the presence of the CHO group in the juxtaposition of one of the two OH groups. It is further to be noted that the C–O–H angle bending mode of the OH group at the position 2 has higher magnitude than that at the position 3 (for 2,3-isomer) or 4 (for 2,4-isomer) or 5 (for 2,5-isomer) and for the 3,4-isomer the mode due to the OH group at the position 3 has a magnitude higher than that due to the OH group at the position 4. For the 2,3-, 2,4- and 2,5-DHBDs it could be due to intramolecular hydrogen bonding between the hydrogen of the OH group at the position 2 and the oxygen atom of the CHO group.

However, the above observation for the 3,4-isomer is not understandable from the present study.

The torsional mode of the OH group appears around 500 cm^{-1} in bonded form. In the present case, it has been assigned in the region $500\text{--}625\text{ cm}^{-1}$. Surprisingly enough, these modes appear to be pure OH torsional modes for all the isomers. Moreover, the two OH torsional modes do not appear to interact with each other even for 2,3- and 3,4-DHBDs

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