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Production of biogas from anaerobic digestion of poultry droppings and domestic waste using catalytic effect of silica gel

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ABSTRACT

This research work has been performed to produce biogas from poultry and household (kitchen) waste using silica gel as a catalyst. A fabricated laboratory scale digester was used to generate biogas from the locally available waste obtained from poultry farms and domestic kitchens. Two laboratory-scale digesters were prepared to digest the solid wastes with and without silica gel respectively. The operating temperatures of the digesters were maintained within 26°C-31°C. The water displacement method was used to investigate the volume of the produced gas. It was found that the production rate of biogas was increased while using silica gel as catalyst. The total gas production was found to be 7921 ml/kg of waste without silica gel whereas it was 10545 ml/kg with a maximum production rate of 1206 ml/kg in a day with silica gel as a catalyst and it was 33.12% higher than before. Finally, by using a k-type thermocouple, the flame temperature and total gas yield after adding the silica gel, the large scale set-ups can be a prospective source of clean energy.

Keywords: Biogas; Anaerobic digestion; Poultry waste; Silica gel; Flame temperature

INTRODUCTION

Biogas is a mixture of methane and carbon dioxide produced by the breakdown of organic waste by bacteria. Biogas is a by-product of the biological breakdown under the oxygenfree conditions of organic wastes such as plants, crop residues, fish residues, wood and bark residues, and human and animal manure [1]. Biogas which is produced when bacteria decompose organic material such as garbage and sewage is a mixture of about 60 percent methane and 40 percent carbon dioxide [2]. Methane is the main component of natural gas which is relatively clean burning, colourless, and odourless [3-6]. This gas can be captured and burned for cooking and heating that are being done on a large scale in some countries of the world. Gas production is enhanced with the use of methane starters. Under optimum conditions which are the pretreatment dilution, pH, agitation, light intensity and an addition of actizyme, maximum gas production can be obtained after 30 days of

anaerobic fermentation [7]. The energy scenario of Bangladesh is dominated by traditional energy sources that supply about 64 percent of the total energy consumption. The scarcity of the costly fossil fuels has already become a burning issue. Moreover, as a result of deforestation, there is a scarcity of fuel wood in some parts of our country. As most of the people largely depend on it, there is a need to look for an alternative fuel. In this prospect, biogas can be a great alternative source of energy in our country because it can be produced from the locally available waste materials such as cow dung, chicken droppings, fish waste, aquatic weed, vegetable waste, residues from domestic cooking etc. [8]. Thus, in this regard a biogas generating model has been fabricated to produce biogas from the anaerobic digestion of chicken droppings and kitchen wastes to analyse its feasibility for practical purpose. Silica gel was also used as a catalyst to increase the gas production [9]. Silica gel was chosen as a catalyst because of its cheapness and availability. According to a previous study [10], silica gel acts as a good catalyst for biogas production because the catalytic and absorbing effects of silica gel are used for the removal of undesired gases such as CO₂.

The anaerobic digestion is a renewable energy source because the process produces a methane and carbon dioxide rich biogas suitable for energy production which helps to replace fossil fuels. Also, the nutrient-rich solids left after digestion can be used as fertilizers [11]. Various research woks were performed to investigate the feasibility of biogas production and application. Along with personal, institutional and government projects all over the world, scholarly articles were also published based on projects and data collection. Abubakar and Ismail [12] performed an anaerobic digestion of cow dung. The average cumulative biogas yield and methane content observed was 0.15 L/kg of VS (volatile solid) and 47%, respectively. Their study established that cow dung is an effective feedstock for biogas production. Ghani and Idris [13] concluded a study on biogas production from municipal waste. They carried their experiment using the low and high strength chemical oxygen alongside municipal waste. From their study, they established that there is a potentiality of biogas production from the municipal waste leachate. The effects of loading rate, temperature, stirrer, both single and co-digestion of feedstock on biogas function and energy efficiency ratio were examined by utilizing an industrial-scale anaerobic digester with the retention time of 12 days [14]. They used cow and poultry manure for biogas production. Tests were performed in 24 various treatments. The results gained from feeding the digester with cow dung in 35°C through different loading rates with and without stirrer showed that the highest rate of biogas was developed in the loading rate of 1/4 along with a stirrer. Likewise, the results gained from various feeding processes with the loading rate of 1/4 showed that the highest biogas amount produced was by treatment I2 (36-40°C, with the stirrer) with about 950 lit/day, and most efficient energy rate was developed through treatment J2 (36-40°C, with the stirrer) with about 3.9 MJ.

Agrahari and Tiwari [15] investigated the production of biogas from kitchen wastes. They analysed different ratios of kitchen waste in a metal made portable floating type biogas plant. In their study, the temperature, solar radiation and relative humidity have been measured. They also analysed the constituent of biogas, pH, volume and rate of biogas production at different level of temperature observation on daily basis. Desai and Madamwar [16] performed an anaerobic digestion of a mixture of cheese whey, poultry waste and cattle dung. They used various adsorbents to improve the digester performance. The adsorbents appeared to improve the digester performance, for example about a two-fold enhancement in total gas production with 17% enriched methane content were achieved with the addition of 4 g litre (-1) of silica gel. An experiment was done

regarding the production of biogas from the municipal solid wastes with domestic sewage by using anaerobic digestion process [17]. The enhancement of the biogas production was done by adding domestic sewage to the municipal wastes. In that experiment, they succeeded to produce biogas having the quality of 68%-72%. Isci and Demirer [18] analysed the potentiality of biogas production from cotton wastes. They performed biochemical methane potential experiments for two different waste concentrations which are 30 g/l and 60 g/l. The results showed that cotton wastes can be treated anaerobically and an effective source of biogas. Karim, Hoffmann [19] showed that the unmixed and mixed digesters performed quite similarly when fed with 5% manure slurry and produced biogas at a rate of 0.84-0.94 L/d. The methane yield was found to be 0.26-0.28 L CH4/g loaded with volatile solid. They performed their experiment at a controlled temperature of 35°C and hydraulic retention time of 16.2 days, resulting in TS loadings of 3.1, 6.2, and 9.3 g/L d for 5%, 10%, and 15% manure slurry feeds, respectively. An experiment on biogas production from water hyacinth was conducted by Kunatsa [20]. From their experiment, they obtained the biogas yield of 1681.08 m³/day. It was also established that using the dry water hyacinth would produce more biogas in comparison to the fresh water hyacinth. Li, Chen [21] observed the anaerobic co-digestion for biogas production. Kitchen wastes were used with cattle manure for their research purpose. Biogas production using batch anaerobic digesters at the mesophilic and thermophilic temperatures was also studied [22]. Researchers performed the thermophilic digestion test with four different feeds to inoculum ratios and the mesophilic digestion was conducted at one feed to inoculum ratio (3:1). The results showed that the feed to inoculum ratio significantly affected the biogas production rate. In their experiment, 80% of the biogas production was obtained during the first 10 days of digestion. They obtained the biogas yields as 430, 372 and 358 mL/g VS (volatile solids), whereas the methane yields were 245, 206, and 185 mL/g VS. [23] studied on batch and continuous biogas production from animal wastes. Both processes were conducted within the mesophilic temperature ranges. In their setup, they found that maximum biogas yield were 3.603 and 2.685 litres in the continuous process and batch process at a temperature of 370°C and 400°C respectively. Biogas production using rice husk has also been performed [24]. It was studied on different parameters like water dilution, initial pH, heavy metals and nitrogen sources on digester performance. Biogas was produced at a rate of 30 mL/day and 69 mL/day for the control and poultry droppings, respectively, after two days while urea gave 8 mL/day on day four. They used poultry droppings (PD) as nitrogen supplements. In the study, they established that rice husk offers an alternative source of energy to agricultural farmers.

Otaraku and Ogedengbe [25] conducted an experiment to produce biogas from sawdust, co-digested with cow dung and water hyacinth. They produced biogas at a rate of 0.045 litres/s fed when about 11.48 gm of sawdust waste was digested in a fixed amount of cow dung and water hyacinth of 7 gm. The optimization of biogas production from chicken droppings with cymbopogon citratus was also performed [26]. Chicken droppings were carried out for a period of 30 days at an average ambient temperature of $33.1 \pm 2^{\circ}$ C. They experimented these in identical reactors (A-C). The result suggested that chicken droppings produced on the average 1.8 L/kg/day of biogas, co-digestion of chicken droppings and C. citratus produced 1.3 L/kg/day of biogas while C. citratus alone produced 1.0 L/kg/day with estimated average methane content of 41.71%, 66.20% and 71.95% for reactors A–C respectively. Their result suggested that despite the higher biogas volumetric yield from chicken droppings digested alone, the co-digestion of chicken droppings with C. citratus had better gas quality. Quiroga, Castrillón [27] experimented on the effect of ultrasound pre-treatment in the anaerobic co-digestion of

cattle manure with food waste and sludge. A series of experiments were carried out under the mesophilic and thermophilic conditions in continuously stirred-tank reactors containing 70% cattle manure, 20% food waste and 10% sewage sludge. The ultrasound pre-treatment allows operating at lower HRT, achieving higher volumetric methane yields: 0.85 L CH4/L day at 36°C and 0.82 CH4/L day at 55°C, when cattle manure and sewage sludge were sonicated. With respect to the non-sonicated waste, these values represent the increases of up to 31% and 67% for the mesophilic and thermophilic digestion, respectively.

A study on a model biogas unit of a breeding farm was made [28]. The anaerobically fermented poultry manure and bovine animal manure were kept in the tank. From the setup, 6.33 m³ and 0.83 m³ biogas was produced from the fermentation of bovine and poultry animal manure respectively. Salam, Biswas [9] used catalyst (silica gel) to produce biogas from the mesophilic anaerobic digestion. They investigated the production ability of biogas from the mesophilic anaerobic digestion of cow dung (CD). The total gas yield was obtained about 27.3 L/kg CD for digestion without catalyst and about 30.5 L/kg of CD for digestion with catalyst. Tripathi, Kumari [29] generated biogas using pine needles as substrate. They reduced the pine needles to very fine sizes (1-2mm) before co-digesting these with the sewage waste water. It was observed that biogas production peaked from 1.4 liter/day to 1.9 liter/day during winter month, whereas it was 7.3 liter/day during months of March and April. The reduction in the volatile solids was also noticed during the months of March and April which was higher close to 64% during April compared to its value in winters and cassava peeling using 45 litres biogas digesters. Ukpai and Nnabuchi [30] did a study on the comparative biogas production. They used cow dung, cow pea and cassava peeling in their research with the assistance of 45 litres biogas digesters. The acid property and catalytic activity of silica gel treated with ammonium salts were investigated [31]. The study showed the effect of added ammonium carbonate or sulfate on the acid property of two commercial and a laboratory prepared silica gel. It was found that the treated gels of the high acid strength were catalytically active for the de-polymerization of paraldehyde and the first order rate constant increases with the increase of the acid amount at $H_0 \le -3$ of the gel catalysts. The catalytic activity of the treated and untreated silica gels for the depolymerization of paraldehyde was studied in connection with the acid property.

METHODS AND MATERIALS

This study is related to the anaerobic digestion of biogas from chicken droppings and kitchen waste. Besides, the effect of silica gel as a catalyst in this experiment has been observed. The raw materials and other relates materials is listed in Table 1.

Item	Description
Raw materials	Chicken droppings, kitchen waste
Fermentation	Continuous anaerobic digestion
Digester	Conical flask
Gas collection	Water displacement method
Catalyst	Silica gel

Table 1. Raw materials and others related data.

For producing biogas from chicken droppings and kitchen wastes, the anaerobic method was chosen. In this method, the fermentation process takes place about fifteen days. As the research work was done in a small setup laboratory, a conical flask was required. The conical flask contains the required amount of chicken droppings, kitchen wastes, and water. After digestion, the gas was collected by the water displacement method. The whole process was kept air tight. The schematic diagram of the experimental setup is given in Figure 1.



Figure 1. Schematic diagram of the experimental setup.

During the anaerobic digestion, carbohydrates, fats and proteins followed a step by step degradation and finally converted into methane and carbon di-oxide. It involved a four stage process including hydrolysis, acidogenesis, acetogenesis and methanogenesis. The overall process can be described by the chemical reaction where organic material such as glucose is biochemically digested into carbon dioxide (CO_2) and methane (CH_4) by the anaerobic microorganisms.



Figure 2. Stages of anaerobic digestion.

The digester was used for processing the slurry. The second and third sections were gas measuring section. Gas production was calculated by using the water displacement method. Many experiments [9, 32, 33] proved that this method is more effective to measure the volume of the produced gas. The first container contained the waste, which went through the anaerobic digestion process, and the second container containing water

was displaced by the gas pressure. Figure 3 represents the schematic setup model of the plant.



Figure 3. Photograph of the experimental setup.

Thus these two containers were connected by a plastic tube connecting the upper portion of the containers. There was also a pipe connecting the third container with a second one by a plastic tube from the bottom of a water filled container to the upper portion of the third empty container. The total system was completely gas sealed.

Experimental Procedure

Chicken droppings and kitchen wastes were collected from the local poultry farms and kitchens of student dormitories respectively. Then the total solid contents for the chicken droppings and kitchen wastes were calculated by drying those raw materials. Finally, according to the calculation, the experimental facility was set. Two experimental set-ups were prepared to investigate the production of biogas from the anaerobic digestion of chicken droppings and kitchen wastes, where one set-up was used for the anaerobic digestion with silica gel as catalyst while the other was used for the anaerobic digestion without silica gel. The conical flask contained the waste where the anaerobic digestion process took place and the first container contained water which was displaced by the gas pressure. After setting up the total facility, data collection was initiated. A total amount of 8% solid for the optimum gas production was chosen. In the slurry, the total solid content (TS) was maintained 8% (by wt.) for all the observations. It was found that the chicken droppings have a total solid of 32% and kitchen wastes have a total solid of 18%. Let,

The amount of

Chicken droppings or Kitchen waste = x gmThe amount of water = y gmSlurry = 2x + ySlurry = 700 gm

Therefore,

$$700 = 2x + y$$

Total solid = (x × 0.32 + x × 0.18) (1)

 $Total \ solid = 8\%$ Therefore,

$$8/100 = (x \times 0.32 + x \times 0.18) \tag{2}$$

Solving Equations (1) and (2)

$$x = 160 gm$$
$$y = 380 gm$$

Therefore, the amount of chicken droppings and kitchen wastes were 160 gm (each of them respectively) and the amount of water was 380 gm in the first and second setup. The total solid content of chicken droppings and kitchen wastes was determined by heating them in an oven for 120°C for 40 hours. A simple way of evaluating effects on gas production is by comparing the arithmetic means of biogas production in the digester unites with the disintegrated and non-disintegrated substrate [34]. The statistically proper way of comparing means is by applying a statistical test. The daily measured biogas production was considered as a random statistical sample where the independence between the measured data points can be assumed due to the experimental realization. The assumptions regarding normal distribution and homogeneity of variances need to be checked for all experimental data in order to select the suitable statistical test. They can be checked either graphically, numerically with distribution parameters, or with formal preliminary significance tests [34]. All statistical analyses were performed using the oneway analysis of variance (ANOVA). The significance level of 5% was used throughout the whole analysis and the standard deviation was kept within $\pm 1\%$. The standard deviations of the total solid and volatile solid determinations were added to the corrected values of the total and volatile solid according to the standard statistical rule [35, 36] to provide a standard deviation of the final results.

For linear combinations, the standard deviations were combined according to Equations (3) and (4).

$$y = k + K_a a + K_b b + K_c c + \cdots$$
(3)

$$\sigma_{y} = \sqrt{\{(k_{a}\sigma_{a})^{2} + (k_{b}\sigma_{b})^{2} + (k_{c}\sigma_{c})^{2} + \cdots ..\}}$$
(4)

For multiplicative applications, the standard deviations are formulated according to Equations (5) and (6).

$$y = \frac{kab}{cd} \tag{5}$$

$$\frac{\sigma_y}{y} = \sqrt{\left\{\left(\frac{\sigma_a}{a}\right)^2 + \left(\frac{\sigma_b}{b}\right)^2 + \left(\frac{\sigma_c}{c}\right)^2 + \cdots\right\}} \tag{6}$$

where, σ is the standard deviation and a, b, c, and d, are the independently measured quantities and k is a constant.

RESULTS AND DISCUSSION

As the data was taken to find the performance in gas production by using the mixture of kitchen waste and poultry droppings, the collected values were used to investigate the outcome of the research project. During the first observation, no silica gel was added and gas production was dependent only on the kitchen wastes and chicken droppings whereas the second setup/observation included additional silica gel along with the original raw waste.

The experimentally obtained data has been used to observe the relation between gas production and number of days. From the first setup, gas production was available only from the raw solid waste without silica gel. The second setup involved the production of

gas including silica gel. The number of observation days was 32 and during this time span the ambient temperatures varied between 26°C and 33°C.



(b) With silica gel

Figure 4. Total gas produced vs. Observed days for the mesophilic digestion of chicken droppings and kitchen waste.

Figure 4(a) shows that the gas production rate increased sharply between the 7th and 24th days. This is because during the period between the setup and initial gas production, the microorganisms responsible for the process were completely inactive [37]. The aerobic bacteria present were using up all the oxygen presents in the digester during this period. After all oxygen has been used up, the acid forming bacteria became active and gas production began. Wangliang, Zhikai [38] also experienced the same rise during their experiment. From Figure 4(b), it can be seen that the gas production was increased between the 3rd and 18th day of observation. The total amount of gas produced was about 8000 ml/kg using chicken droppings and kitchen wastes. The increase in gas production

was swift as the number of days increased. However, during the last observation days, the increase in gas production became steady and did not show further increase. The total amount of produced gas was 10545 ml/kg when the silica gel was used along with the chicken droppings and kitchen wastes. It was 33% higher compared to the first setup. When the gas production started, the anaerobes present in the organic material became activated and increased in numbers. At the maximum point of production, the anaerobes were acting on the maximum possible amount of the organic materials. After that point, the gas production declined as the excess substrates were converted into methane. There was also a decrease in either carbon or nitrogen available for use [37]. When one comes to an exhaust, the process becomes slower. As it continued, the condition became more appropriate for the methane forming bacteria. Thus, it led to an increase in the percentage composition of methane until the maximum level was reached. As the fermentation approached to the end, the burning characteristics of the gas were improved.



Figure 5. Variation in the daily gas production with respect to the number of observed days during the mesophilic digestion.

Figure 5 (a) shows the production rate of gas increased up to 628 ml/kg at first and then started to increase and decrease without maintaining any order. The maximum gas production rate was 628 ml/kg on the 4th day. The erratic response from gas production was found during the mesophilic disgestion. During the 18th, 20th and 21st observation days, there was no gas production from the system. On the other hand, Figure 5 (b) depicts that gas production rate that increased between 3rd and 17th day. The total amount of produced gas was 10545 ml/kg when silica gel was used along with the chicken droppings and kitchen wastes. Figure 5 (b) represents that the production rate was increased up to 1026 ml/kg. Salam, Biswas [9] also reported an increase in the total biogas production rate in his research. They found a daily gas yield of 1244 ml/kg cow dung with silica gel on the 34th day. The standard as well as expected values vary according to the setup, ambient condition and raw materials. Then the production of gas was increased and decreased without following any order. The maximum gas production rate was found to be 1026 ml/kg on the 3rd day which was 63.37% greater than the highest gas production in the 3rd observation day during observation without silica gel.



Figure 6. Variation of temperatures (°C) with observed days for the mesophilic digestion.

While tabulating different values, the temperature variation was also observed and the relation with a number of days is plotted below in Figure 6. From Figure 6, it can be seen that the temperatures during this project varies between 29 and 35°C. Between the 15th and 20th day, the temperature was found to be the lowest. The total gas productions at a lower range of ambient temperatures were found to be significantly lower. At 18° to 28°C ambient temperature range, the total gas productions from two digesters were reported to be 10447 and 13139 ml/kg cow dung [9][35]. Ukpai and Nnabuchi [30] reported the total gas generation of 7312 ml/kg CD from anaerobic digestion of cow dung while slurry temperature was maintained between 22°C and 36°C. In this project, the daily gas yield was found higher when temperature was also high. For example, during the 3rd day of the observation period, the daily gas yield found to be relatively lower in a low temperature condition. Garba [39] also found the similar result when conducting

experiment. He found that the digester kept at 60°C produced more gas followed by the digester which was maintained at 40°C. This could be due to the fact that the methane producing bacteria are not active at this temperature. He also found at 70°C temperature, biogas production was very negligible. Thus, an optimum temperature range is always desired and very important for effective biogas generation.

A numerical and graphical comparison was made between two observations. After using the silica gel as a catalyst in the second setup, the gas production rate was found to be higher than the first. The excess amount of gas production was 2624 ml/kg which is higher than the first observation when only chicken droppings and kitchen waste were used. Figure 7 represents the total gas production rate for both experimental set-ups and it is clearly seen that the gas production was higher for the setup with silica gel than that of without silica gel. It has been reported that the addition of powdered activated carbon results in an increase in the total gas production with high methane content and stabilizes anaerobic process of performance [40]. Similar results were obtained with other adsorbents [41] and [42] studied the effect of silica gel as one of the adsorbents in order to improve the anaerobic codigestion of water hyacinth and cattle dung. Based on the review of literature, it is evident that the adsorbents are responsible for improved digestion. The surface of the adsorbent provides sites where substrate can accumulate thereby providing high localized substrate concentrations. These areas of adsorption provide a more favourable growth environment for the bacterial substrate system. A significant increase of about 33.12% in the total gas production was observed while using silica gel as a catalyst than the setup without silica gel. Though in the beginning, the difference was not very significant, it became more visible with the increase in a number of days.



Figure 7. Comparison between two set-ups and effect of catalyst (silica gel) on biogas production.

Determination of flame temperature

To determine the flame temperature, a k-type thermocouple was used. The type k thermocouple has a Chromel (Ni-Cr) positive leg and an Alumel (Ni-Al) negative leg. The temperature of the flame was measured around 619°C. Mandal, Kiran [43]

determined the quality of biogas by measuring the flame temperature. The highest flame temperature was reached at a value of 685°C. It was found that the flame temperature increased gradually with the increase in the percentage of methane. In their research work, the flame temperature varied between 600°C and 685°C for 50% and 59% methane content respectively. The feasibility of biogas in combustion has been a key research interest for the researchers and Noor, Wandel [44] performed simulation to study the airfuel ratio for the mixture of biogas and hydrogen for mild combustion. Besides, the effect of air-fuel ratio on temperature distribution and pollutants for biogas mild combustion was also studied by [45].

CONCLUSIONS

While using poultry droppings and kitchen waste the total gas production was 7921 ml/kg and the maximum rate was 628 ml/kg during the mesophilic digestion. For the second set-up which included 5 gm of catalyst (silica gel), the total gas production was found to be 10545 ml/kg with a maximum production rate of 1026 ml/kg. The total gas production increased up to 33.12% while using silica gel as catalyst. The effect of temperature was also observed and it was found that the lower temperature of the setup causes a reduction. The flame temperature by using the produced gas was measured and found to be 619°C. Gas production is not much satisfactory when compared to the optimum values of other organic waste as the volume occupied by the gas is too much. Gas production also depends on continuous feed into the process with a high retention time. From the experimental data, it was found that biogas production was not uniform. From above analysis, it can be said that by this composition it is possible to get the optimum amount of gas production by using a specific catalyst which is silica gel in this case.

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