

# Temperature Dependence of Steady-State and Presteady-State Kinetics of a Type IIb Na<sup>+</sup>/P<sub>i</sub> Cotransporter

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**Abstract** The temperature dependence of the transport kinetics of flounder Na<sup>+</sup>-coupled inorganic phosphate (P<sub>i</sub>) cotransporters (NaPi-IIb) expressed in *Xenopus* oocytes was investigated using radiotracer and electrophysiological assays. <sup>32</sup>P<sub>i</sub> uptake was strongly temperature-dependent and decreased by ~80% at a temperature change from 25°C to 5°C. The corresponding activation energy ( $E_a$ ) was ~14 kcal mol<sup>-1</sup> for the cotransport mode. The temperature dependence of the cotransport and leak modes was determined from electrogenic responses to 1 mM P<sub>i</sub> and phosphonoformic acid (PFA), respectively, under voltage clamp. The magnitude of the P<sub>i</sub>- and PFA-induced changes in holding current decreased with temperature.  $E_a$  at -100 mV for the cotransport and leak modes was ~16 kcal mol<sup>-1</sup> and ~11 kcal mol<sup>-1</sup>, respectively, which suggested that the leak is mediated by a carrier, rather than a channel, mechanism. Moreover,  $E_a$  for cotransport was voltage-independent, suggesting that a major conformational change in the transport cycle is electroneutral. To identify partial reactions that confer temperature dependence, we acquired presteady-state currents at different temperatures with 0 mM P<sub>i</sub> over a range of external Na<sup>+</sup>. The relaxation time constants increased, and the peak time constant shifted toward more positive potentials with decreasing temperature. Likewise, there was a depolarizing shift of the charge distribution, whereas the total available charge and apparent valency predicted from single Boltzmann fits were temperature-independent. These effects were ex-

plained by an increased temperature sensitivity of the Na<sup>+</sup>-debinding rate compared with the other voltage-dependent rate constants.

**Keywords** P<sub>i</sub> cotransport · Uncoupled leak · Temperature · Presteady-state · Activation energy

## Introduction

Secondary active transporters are solute carrier proteins that mediate the uphill movement of a specific solute by coupling this to the downhill movement of a specific ion (e.g., H<sup>+</sup>, K<sup>+</sup> or Na<sup>+</sup>) according to the free energy available from the driving ion's own electrochemical gradient. The coupling confers a strict stoichiometric ratio between the driven and driving species. This contrasts with the traditional view of ion channels, which, when in the conducting or open state, allow solute movement through their pore according to an electrodiffusive mechanism, thereby resulting in transport rates up to five orders of magnitude faster than for carriers. The kinetic and molecular models proposed to account for the functional data also reflect these differences between carriers and channels, and the activation energy for movement of an ion through a channel is much less than that associated with the transport cycle of a carrier (Van Winkle 1995). This underscores a thermodynamically based distinction between channels and carriers that is reflected in the greater apparent complexity of the conformational changes involved in the translocation process in the latter case, where ordered substrate binding/debinding and alternating accessibility of substrates to their respective binding sites are features of the currently postulated carrier mechanisms. Recently, the distinction between carriers and channels has become less clear as

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evidenced by a number of studies on cotransporters that have identified leak modes in the absence of substrate and which exhibit channel-like properties (DeFelice 2004; Sonders and Amara 1996). Moreover, studies on the serotonin transporter (SERT) indicate that at high substrate concentrations SERT exhibits transport behavior more consistent with a channel rather than a carrier (Adams and DeFelice 2002). Finally, functional studies on a bacterial homolog of the CIC channel have revealed that a single-amino acid substitution can switch the mode of operation from that of an  $H^+Cl^-$  antiporter with strict stoichiometric coupling to the characteristic CIC channel mode (Accardi and Miller 2004).

Electrogenic type II  $Na^+$ -coupled inorganic phosphate ( $P_i$ ) cotransporters (NaPi-II), encoded by members of the solute carrier *SLC34* gene family (<http://www.bioparadigms.org>), also possess the expected strict stoichiometrically coupled cotransport mode behavior of a carrier (Forster, Loo and Eskandari 1999; Virkki et al. 2005b) as well as leak-mode behavior (Forster et al. 2002). A  $Na^+$ -dependent leak mode for the rat NaPi-IIa cotransporter was proposed from the finding that the  $P_i$  transport inhibitor phosphonoformic acid (fosfocarnet or PFA), applied in the absence of  $P_i$  to NaPi-IIa-expressing *Xenopus* oocytes under voltage-clamp conditions, led to a decrease in the holding current (Forster et al. 1998). The magnitude of the PFA-induced change in holding current was proportional to the cotransport mode activity, which suggested a direct association of this pathway with functional expression of the cotransporter. Moreover, its reversal potential was reported to shift in a Nernstian manner according to the external  $Na^+$  concentration (Forster et al. 1998). The NaPi-IIa  $Na^+$ -leak mode was incorporated into a kinetic scheme as a “uniporter” pathway (Forster et al. 1998), similar to that proposed for other  $Na^+$ -driven cotransporters (e.g., the sodium glucose cotransporter [SGLT1] [Birnir, Loo and Wright 1991; Chen et al. 1997], the thyroid  $Na^+/I^-$  cotransporter [Eskandari et al. 1997] and the  $Na^+$ -dependent dicarboxylate cotransporter [Yao and Pajor 2000]). This scheme indicates that the leak and cotransport modes are mutually exclusive but share the same empty carrier and  $Na^+$  binding transitions. Studies involving both wild-type and mutant NaPi-IIa constructs (Kohler et al. 2002a, b; Virkki et al. 2005b) have also provided experimental evidence in support of this idea. Nevertheless, the reversal potential of  $I_{PFA}$  lies far to the left (typically  $-20$  mV with external 100 mM  $Na^+$ ) of what would be predicted if  $Na^+$  were the only ion involved, and this observation therefore suggests that  $I_{PFA}$  is more complex than originally proposed. Moreover, recent observations (A. Bacconi, L. V. Virkki and I. C. Forster, unpublished results) indicate that  $Cl^-$  ions may also be involved, which adds further weight to the concept, proposed for excitatory amino acid

cotransporters, that a transporter-associated, substrate-gated, channel-like pathway could also be present (e.g., Otis and Jahr 1998).

The distinction between carrier and leak modes can be experimentally demonstrated by determining their temperature dependence, based on the property that the activation energy for translocating solutes according to a carrier mechanism is significantly greater than that associated with ionic electrodiffusion for a channel (Van Winkle 1995). This hypothesis is supported by studies on a serotonin transporter (dSERT) (Beckman and Quick 2001) and a  $\gamma$ -aminobutyric acid (GABA) transporter isoform (GAT4) (Karakossian et al. 2005), in which significantly different temperature coefficients for the respective cotransport and leak modes were reported.

The aims of the present study were twofold. First, to gain further insight into the mechanism of the NaPi-II cotransport and leak modes, we determined the temperature dependence of the respective transport rates. We used the type IIb  $Na^+/P_i$  cotransporter (NaPi-IIb) cloned from flounder because of its 5- to 10-fold higher expression in *Xenopus* oocytes (Forster et al. 1997) compared with its mammalian cousins. We have previously shown that flounder NaPi-IIb exhibits leak and cotransport modes (Forster, Biber and Murer 2000) and displays similar kinetic features associated with the mammalian NaPi-IIa (Forster et al. 1997, 1999, 2000; Virkki, Murer and Forster 2006a, b). Therefore, our findings should be applicable to mammalian members of the SLC34 family. Second, to identify the contributions of resolvable partial reactions that comprise the overall cotransport cycle, we characterized the temperature dependence of presteady-state relaxations under different external superfusion conditions. Analysis of presteady-state relaxations allowed us to compare the temperature dependence of voltage-dependent partial reactions that precede  $P_i$  binding.

## Materials and Methods

### Reagents and Solutions

All reagents were obtained from Sigma (Buchs, Switzerland) or Fluka (Buchs, Switzerland).

The compositions of the solutions used (in mM) were as follows: oocyte incubation (modified Barth's solution), NaCl (88), KCl (1),  $CaCl_2$  (0.41),  $MgSO_4$  (0.82),  $NaHCO_3$  (2.5),  $Ca(NO_3)_2$  (0.33) and 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES, 7.5), adjusted to pH 7.6 with tris(hydroxymethyl)aminomethane (TRIS) and supplemented with antibiotics (10 mg/liter gentamycin or streptomycin or 5 mg/liter doxycyclin); control superfusate (ND100), NaCl (100), KCl (2),  $CaCl_2$  (1.8),  $MgCl_2$  (1) and

HEPES (10), adjusted to pH 7.4 with TRIS at room temperature; control superfusate (ND0), as for ND100 but with *N*-methyl-D-glucamine or choline chloride used to replace  $\text{Na}^+$  (intermediate  $\text{Na}^+$  concentrations were achieved by proportionate mixing of the ND100 and ND0 solutions); substrate test solutions,  $\text{P}_i$  added to ND100 from 1 M  $\text{K}_2\text{HPO}_4$  and  $\text{KH}_2\text{PO}_4$  stocks that were mixed to give pH 7.4.

For a change in temperature from 25°C to 2°C, the superfusate solution pH increased by 0.25 pH units, due to the  $\log_{10}$  of acid dissociation constant (pKa) temperature dependence of the HEPES-TRIS buffer. Based on previous studies of the pH dependence of NaPi-IIa/b (Forster et al. 2000; Virkki et al. 2005a), this would result in a negligible change in electrogenic transport activity. Therefore, no pH adjustment was considered necessary in the present study.

### Molecular Biology and Expression in *Xenopus laevis* Oocytes

The cDNA encoding for flounder NaPi-IIb was subcloned into a KSM expression vector to improve its expression in *X. laevis* oocytes, as described previously (Virkki et al. 2003). The plasmid was linearized using *Xba*I (Promega, Madison, WI) and used as a template for the synthesis of capped cRNA using the Message Machine T3 kit (Ambion, Austin, TX). Stage V–VI defolliculated oocytes from *X. laevis* were isolated and maintained as described previously (Virkki et al. 2006b). Oocytes were injected with 50 nl of cRNA (0.2  $\mu\text{g}/\mu\text{l}$ ) encoding NaPi-IIb. Oocytes were incubated at 17°C in modified Barth's solution. Transport assays were performed 3–5 days after injection.

### Tracer Uptake

The temperature dependence of  $\text{P}_i$  transport was determined by performing  $^{32}\text{P}_i$  uptake assays at five different temperatures. Control oocytes and oocytes expressing NaPi-IIb (10–12 oocytes/group) were first allowed to equilibrate in ND100 solution without tracer at the appropriate temperature. After aspiration of this solution, we added 150  $\mu\text{l}$  of ND100 solution containing 1 mM  $\text{P}_i$  and  $^{32}\text{P}_i$  (specific activity 10 mCi/mmol  $\text{P}_i$ ). After 30 min, the uptake was stopped by washing the oocytes four times with 3 ml ice-cold ND0 solution containing 0.5 mM cold  $\text{P}_i$ . Oocytes were then placed individually in scintillation vials and lysed in 250  $\mu\text{l}$  10% sodium dodecyl sulfate.  $^{32}\text{P}_i$  activities of individual oocytes were measured using scintillation counting.

### Electrophysiology

Electrophysiology was performed using a two-electrode voltage clamp (Geneclamp 500; Molecular Devices, Sunnyvale, CA). Oocytes were mounted in a small recording

chamber (100  $\mu\text{l}$  volume) and continuously superfused (~0.2 ml/min) with test solutions supplied using a peristaltic pump. The recording chamber was machined from a block of thermally conductive epoxy (ER2074; Electrolube, UK), which was mounted on a Peltier thermoelement (CP1.4–127–045L; Melcor, NJ). The chamber temperature was controlled by a laboratory-built electronic regulator. Data were acquired online using Digidata 1200 hardware and compatible pClamp8 software (Molecular Devices). Currents were prefiltered at a bandwidth less than twice the sampling rate appropriate for the signals being acquired. Steady-state and presteady-state currents were acquired using voltage steps from a holding potential  $V_h = -60$  mV to test potentials in the range  $-180$  to  $+80$  mV. Presteady-state records were acquired with fourfold signal averaging. To account for the significant slowing of the presteady-state relaxations with temperature, the test pulse width was extended to ensure that the relaxations were complete before returning to the holding potential.

### Data Analysis

#### *Presteady Relaxations and Modeling*

Relaxations induced by voltage steps were quantified by fitting a decaying double exponential function using a fitting algorithm (Clampfit, v. 8.0; Molecular Devices). To resolve the charge movement associated with the exogenous protein, a component subtraction method (Hazama, Loo and Wright 1997) was employed: the fast component, assumed to represent the oocyte capacitive charging, was extrapolated to the step onset and subtracted from the total transient current. The slower component was numerically integrated to give the charge attributable to the exogenous protein.

For a voltage step from a holding potential ( $V_h$ ) to a test potential ( $V$ ) that induces the movement of a charge  $Q$ , the  $Q$ - $V$  data were fitted with a form of the Boltzmann equation:

$$Q = Q_{\text{hyp}} + Q_{\text{max}} / (1 + \exp(-ze(V - V_{0.5})/kT)) \quad (1)$$

The voltage dependence of the time constant ( $\tau$ ) of the slower relaxation induced by the voltage step was fitted as follows:

$$\tau = 1 / (k_f \exp(-zeV/2kT) + k_b \exp(zeV/2kT)) \quad (2)$$

Eqs. 1 and 2 are derived from a simple Eyring transition rate model for a two-state system that describes the distribution and rate of movement of mobile charged particles (of effective valence,  $z$ ) that move across the membrane over a symmetrical energy barrier, according to the direction and magnitude of the transmembrane electric field. For Eq. 1,  $Q_{\text{max}}$  is the maximum charge translocated,  $Q_{\text{hyp}}$  is

the steady-state charge at the hyperpolarizing limit and depends on  $V_h$ ,  $V_{0.5}$  is the voltage at which the charge is distributed equally between the two states,  $e$  is the elementary charge,  $k$  is Boltzmann's constant and  $T$  is the absolute temperature. For Eq. 2,  $k_f$  and  $k_b$  are the zero voltage forward and backward transition rates, respectively, for the relocation of the hypothetical particles according to a clockwise movement around the transport cycle. Based on this model, it can be shown that  $V_{0.5} = -kT/ze \ln(k_b/k_f)$ . Eqs. 1 and 2 were fit to the  $Q$ - $V$  and  $\tau$ - $V$  data, respectively, by nonlinear regression analysis with  $Q_{hyp}$ ,  $Q_{max}$ ,  $z$ ,  $V_{0.5}$ ,  $k_f$  and  $k_b$  as free parameters, for  $T = 278, 283, 288, 293$  K. All curve fitting using Eqs. 1 and 2 was performed using the nonlinear regression analysis algorithms in GraphPad (San Diego, CA) Prism version 3.02/4.02 for Windows (<http://www.graphpad.com>).

### Temperature Dependence

The Arrhenius equation (Segel 1976) was used for determining the temperature dependence of the transport kinetics:

$$k_r = A \exp(-E_a/RT) \quad (3a)$$

where  $k_r$  is the rate (uptake rate, substrate-induced current or partial reaction rate constant),  $A$  is a constant,  $R$  is the gas constant and  $E_a$  is the activation energy (in  $\text{kcal mol}^{-1}$ ). Fitting with a single exponential function yields an estimate for  $E_a$ . Commonly, Eq. 3a can be recast in a logarithmic form and linear regression analysis applied to the experimental data to obtain  $E_a$ :

$$\log k_r = -E_a/(2.303R)(1/T) + \log A \quad (3b)$$

Both methods yield similar estimates for  $E_a$ , and we used the logarithmic graphical representation for consistency with other studies.

The temperature coefficient for a  $10^\circ\text{C}$  change,  $Q_{10}$ , is related to  $E_a$  as follows:

$$\log Q_{10} = 10 E_a / (2.303RT_1T_2) \quad (4)$$

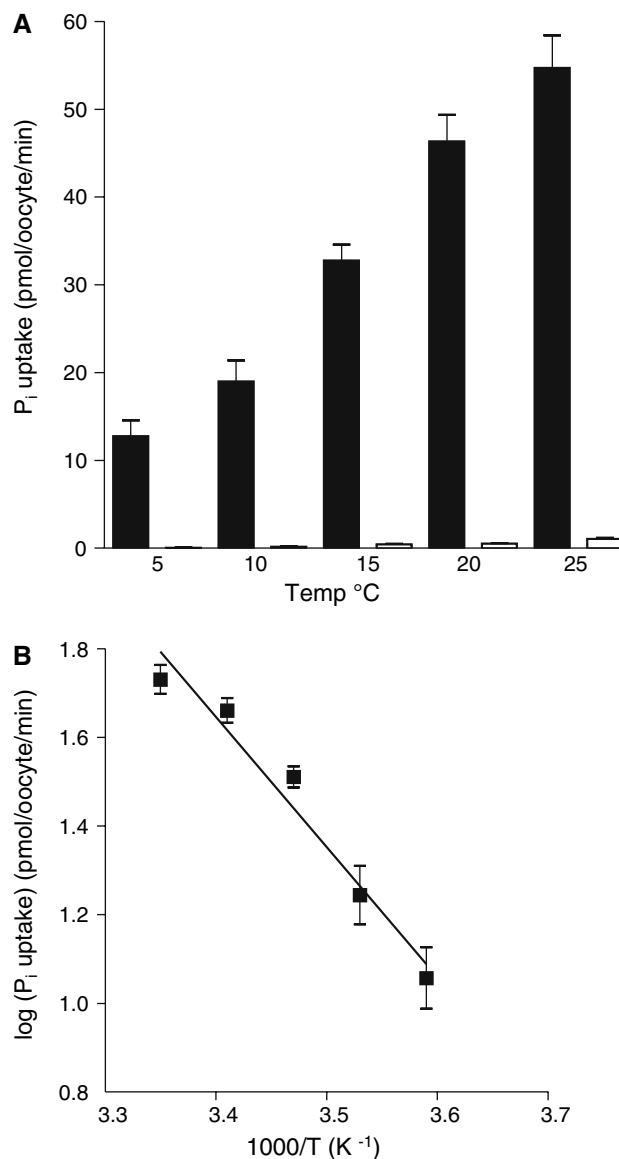
where  $T_1$  and  $T_2$  are the two temperatures (K).

## Results

### Temperature Dependence of Flounder NaPi-IIb in the Steady State

We first determined the temperature dependence of  $^{32}\text{P}_i$  uptake for oocytes injected with flounder NaPi-IIb cRNA. Figure 1A shows that  $^{32}\text{P}_i$  uptake exhibited a strong

temperature dependence with an approximately fivefold increase with increasing temperature over the  $20^\circ\text{C}$  range used in these experiments. The uptake levels for noninjected oocytes were at least 50–100 times smaller than those for NaPi-IIb-expressing cells and showed no systematic temperature dependence. We therefore ignored the contribution of any endogenous uptake in the subsequent analysis. Figure 1B shows the same data for the NaPi-IIb-expressing oocytes replotted in the form of an Arrhenius plot (Segel 1976). The dependence of uptake rate (scaled logarithmically) on the reciprocal of absolute temperature

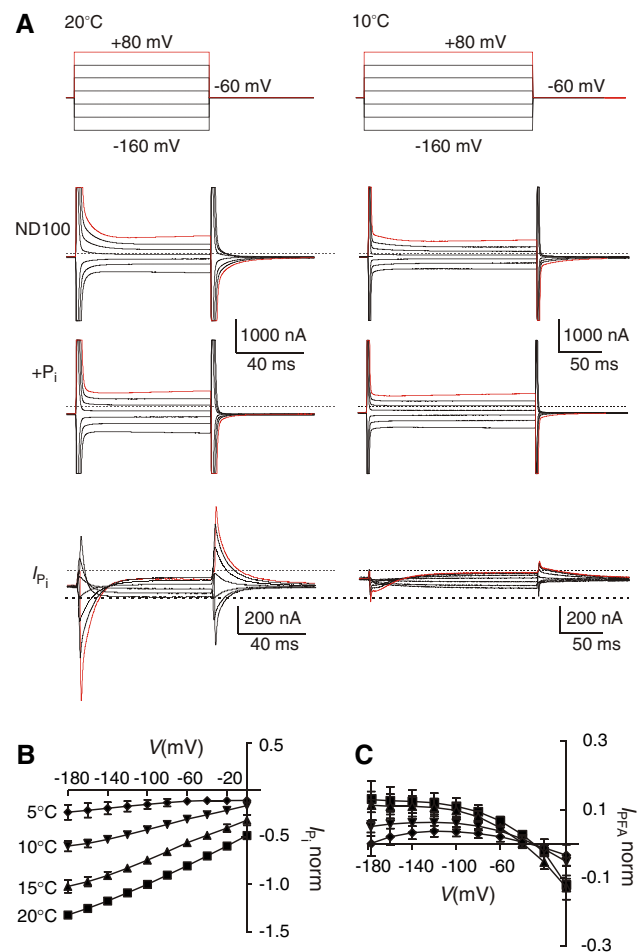


**Fig. 1** Temperature dependence of  $^{32}\text{P}_i$  uptake. (A) Uptake for NaPi-IIb-injected oocytes (filled columns) and noninjected oocytes (empty columns) from the same donor frog plotted as a function of incubation temperature. (B) Data for NaPi-IIb from A replotted as an Arrhenius plot (Eq. 3). The slope of the linear regression line was reported as  $-2,942 \pm 335.3 \text{ K}^{-1}$ , equivalent to  $E_a = 13.5 \pm 1.5 \text{ kcal mol}^{-1}$

showed a reasonably linear relationship. We fit these data using a linear regression algorithm ( $r^2 = 0.96$ ), and from the fitted slope we obtained an estimate of the apparent activation energy for the transport process,  $E_a = 13.5 \pm 1.5$  kcal mol<sup>-1</sup>, which corresponds to a 10°C temperature coefficient ( $Q_{10}$ ) of 2.3 (Eq. 4). These values are typical for an enzyme-mediated process and exceed those expected for a simple diffusion-mediated process (Van Winkle 1995).

The transport cycle for electrogenic NaPi-II cotransporters is proposed to involved transitions between a number of states (Forster et al. 2002), and the contributions of the respective partial reactions between states to the overall temperature dependence can therefore best be studied using electrophysiology. Figure 2A shows representative current recordings from the same oocytes expressing flounder NaPi-IIb at 20°C and 10°C in response to voltage steps for 80 ms in the range -160 to +80 mV from a holding potential of -60 mV. At 20°C, in the absence of external P<sub>i</sub> (with superfusion in ND100), we observed presteady-state relaxations superimposed on the oocyte capacitive charging transient, as we have previously reported for flounder NaPi-IIb (Forster et al. 1997, 2000; Virkki et al. 2006a, b). These relaxations were suppressed in the presence of 1 mM P<sub>i</sub>. At 10°C, the relaxations were clearly slower, and therefore we doubled the pulse width to ensure that a steady-state current was reached. Like the data at 20°C, the relaxations at 10°C were suppressed with 1 mM P<sub>i</sub>. The P<sub>i</sub>-dependent current was obtained by subtracting the records in ND100 from the corresponding records in ND100+P<sub>i</sub> (Fig. 2A, lower data set). The steady-state P<sub>i</sub>-dependent current ( $I_{P_i}$ ) was smaller at the lower temperature. Figure 2B shows pooled  $I$ - $V$  data ( $n = 4$ ) obtained from similar records as shown in Figure 2A, normalized to  $I_{P_i}$  at -100 mV at 20°C, for a larger range of temperatures and intermediate test voltages. For this steady-state analysis, we omitted the responses for  $V \geq 0$  mV because the data were contaminated by endogenous activating currents.

We have previously reported that the type II Na<sup>+</sup>/P<sub>i</sub> cotransporters exhibit an uncoupled leak mode (Forster et al. 1998, 2000). This can be observed using the P<sub>i</sub> transport blocker PFA, which specifically inhibits the leak current. To determine the temperature dependence of the leak-mode activity, we recorded the response to PFA (1 mM) (*not shown*) at different temperatures and normalized the data to  $I_{P_i}$  for each cell in response to 1 mM P<sub>i</sub> at -100 mV and 20°C. The steady-state PFA-induced changes in holding current ( $I_{PFA}$ ) for the flounder isoform were typically  $\leq 10\%$  of  $I_{P_i}$  (Forster et al. 2000) and, like  $I_{P_i}$ , strongly temperature-dependent as shown for the data sets pooled and normalized to the corresponding  $I_{P_i}$  for each cell in Figure 2C. Note that each data point is the difference between the holding current in ND100 + PFA and ND100,



**Fig. 2** Steady-state temperature dependence of substrate-induced currents. (A) Representative current recordings from an oocyte that expressed NaPi-IIb, made at 20°C (*left*) and 10°C (*right*). Currents were recorded in response to the voltage jump protocol from a holding potential of -60 mV to potentials in the range -160 to 80 mV at 40-mV intervals, in the absence (ND100) and presence of P<sub>i</sub> (1 mM, total). *Dotted lines* indicate the steady-state zero current level. The P<sub>i</sub>-dependent current ( $I_{P_i}$ ) was obtained from the difference between corresponding traces in ND100+P<sub>i</sub> and ND100. Note that a longer pulse was used at 10°C to ensure that a steady state was reached. For each data set, traces at +80 mV are shown in *red*. *Dashed line* indicates  $I_{P_i}$  at -160 mV, 20°C to aid comparison. (B)  $I$ - $V$  plots of  $I_{P_i}$  at the indicated temperatures. Data are shown as means  $\pm$  SEM ( $n = 4$ ), normalized to  $I_{P_i}$  at 20°C and -100 mV. (C)  $I$ - $V$  plots of the PFA-induced change in holding current ( $I_{PFA}$ ). Data were obtained using the same protocols as in A except that 1 mM PFA was substituted for 1 mM P<sub>i</sub>. For each oocyte, the  $I_{PFA}$  data were first normalized to  $I_{P_i}$  for each cell at -100 mV before pooling. Same symbols as in B

and the NaPi-IIb PFA-inhibited leak current would be given by  $-I_{PFA}$ , under the assumption that PFA fully blocks the leak current. Unlike  $I_{P_i}$ , these currents showed a current reversal in the range -30 to -20 mV that appeared to be temperature-independent. Determination of exact reversal potential was not possible because of the relatively small magnitude of  $I_{PFA}$  (current magnitudes in the range 10–15



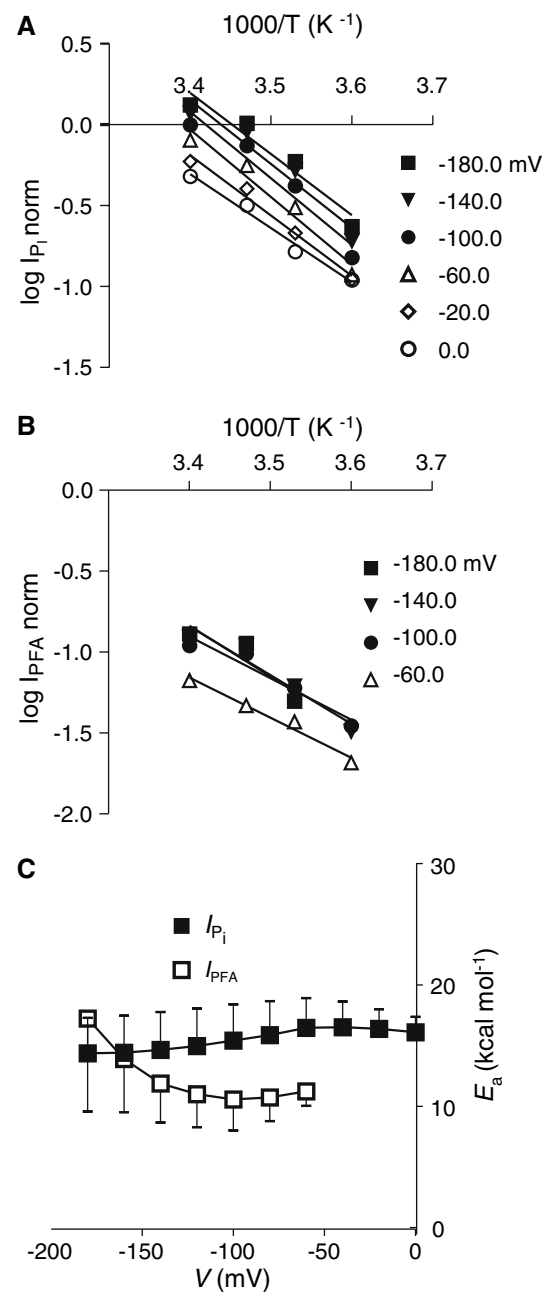
nA at  $-100$  mV) and uncertainties regarding the contribution of endogenous PFA response at potential extremes (Forster et al. 2000).

We compared the temperature dependence of the currents corresponding to the cotransport and leak modes by replotting the normalized  $I$ - $V$  data in the form of Arrhenius plots (Fig. 3). For  $I_{P_i}$ , the data were well described by the fit of a linear regression line over a range of potentials from  $-180$  to  $0$  mV (Fig. 3A). For  $I_{PFA}$ , the data were less reliable, particularly near the reversal potential. From the slopes of the Arrhenius plots, we obtained estimates of the activation energy for  $I_{P_i}$  and  $I_{PFA}$ , as shown in Figure 3C (see Eq. 3b). The  $E_a$  for  $I_{P_i}$  showed little variation with membrane potential and was larger than the  $E_a$  for  $I_{PFA}$  for all potentials. The mean  $E_a$  for  $I_{P_i}$  of  $\sim 15$  kcal mol $^{-1}$  corresponds to  $Q_{10} \sim 2.5$  (Eq. 4), which is typical for values reported in other studies on electrogenic cotransporter membrane proteins (Beckman and Quick 2001; Binda et al. 2002; Wadiche and Kavanaugh 1998); and for  $I_{PFA}$ , the  $E_a$  corresponded to a  $Q_{10} \sim 2.0$ .

#### Temperature Dependence of Presteady-State Relaxations

We next investigated the effect of temperature on the presteady-state relaxations to determine if specific transitions might dominate the temperature dependence of the cotransport function. Figure 4A shows representative presteady-state currents for the ON transition from the holding potential to the step potential, recorded from the same oocyte, superfused with ND100 at four temperatures. In each case, we applied baseline correction to remove the steady-state currents. As we reduced the temperature, the relaxations superimposed on the capacitive charging transient became slower and the initial amplitude decreased. There also appeared to be less charge movement for the same voltage change in the hyperpolarizing direction compared with depolarizing steps, which suggested an alteration in the voltage dependence of the charge distribution. The effect of temperature on the presteady-state relaxations was reversible. This was established for representative experiments ( $n = 3$ ) in which temperature was reduced progressively from  $20^\circ\text{C}$  to  $5^\circ\text{C}$  and then the test protocol was repeated at  $20^\circ\text{C}$ , whereupon the time course and amount of charge were unchanged (*data not shown*).

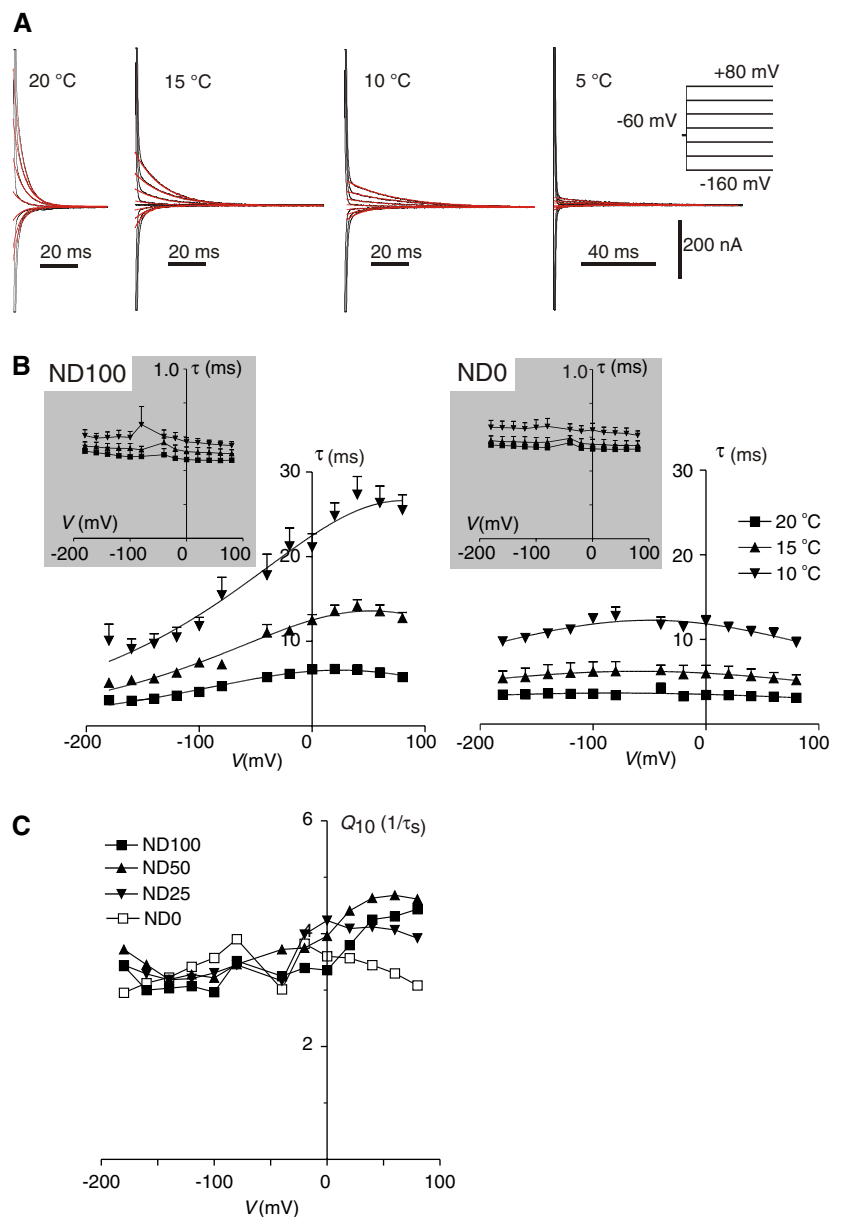
We quantified these data by fitting the total relaxation with a double exponential function to yield two time constants, the slower of which ( $\tau_s$ ) is shown superimposed on the total current (Fig. 4A). The voltage dependence of the time constants is shown in Figure 4B for ND100 and in the absence of external  $\text{Na}^+$  ions (ND0) and for three temperatures ( $20^\circ\text{C}$ ,  $15^\circ\text{C}$  and  $10^\circ\text{C}$ ) ( $n = 6$ ). Fits to the data at  $5^\circ\text{C}$  proved too unreliable because of noise contamination



**Fig. 3** Temperature dependence of  $I_{P_i}$  and  $I_{PFA}$ . (A) Arrhenius plots of the normalized magnitude of  $I_{P_i}$  replotted from the data in Figure 2B for the membrane potentials indicated. *Straight lines* represent linear regression fits to the data points to yield the apparent  $E_a$  given by Eq. 3. (B) Arrhenius plots for normalized  $I_{PFA}$  replotted from the data in Figure 2C for the membrane potentials indicated. *Straight lines* represent linear regression fits to the data points to yield  $E_a$ . (C)  $E_a$  plotted as a function of membrane potential for  $P_i^-$  ( $I_{P_i}$ , filled squares) and PFA-induced change in holding current ( $I_{PFA}$ , open squares). Note that for  $I_{PFA}$  we were unable to determine  $E_a$  for potentials close to the  $I_{PFA}$  reversal potential because of data unreliability

of the signal at low amplitudes. The faster time constant ( $\tau_f$ ) (Fig. 4B, inset) became slower with decreasing temperature and showed little voltage dependence. As in

**Fig. 4** Temperature effects on the kinetics of presteady-state currents. (A) Presteady-state relaxations recorded from the same oocyte in response to voltage steps from a holding potential of  $-60$  mV to voltages in the range  $-180$  to  $+80$  mV at the four temperatures indicated. The steady-state holding currents were subtracted for each data set. *Red traces* indicate the slower fitted component derived from the two-exponential fit to the complete trace, commencing approximately 1 ms after the voltage step onset. (B) Voltage dependence of the main relaxation time constant ( $\tau$ -V) reported by a two-exponential fit to the data in A for the three temperatures indicated, corresponding to the fitted curves in A. *Continuous lines* are fits with Eq. 2 to the data points. *Left*: Superfusion in ND100. *Right*: Superfusion in ND0. *Inset*: Temperature and voltage dependence of the fast component. Data at  $5^\circ\text{C}$  were too unreliable to determine the slow relaxation time constant. (C) Voltage dependence of  $Q_{10}$  of the reciprocal of the main relaxation time constant ( $1/\tau_s$ ) for the external indicated  $\text{Na}^+$  concentrations. Data points are joined by lines for visualization.  $Q_{10}$  was calculated from the ratio of  $1/\tau_s$  at  $20^\circ\text{C}$  to  $1/\tau_s$  at  $10^\circ\text{C}$  ( $n = 6$ )



previous studies, we assumed that this component represented the capacitive charging of the oocyte. Its temperature dependence may result from a temperature dependence of the membrane capacitance (White 1970) that would result in slower membrane charging. The small voltage dependence of this component most likely results from a systematic error in the fits because it was observed at all  $\text{Na}^+$  concentrations and temperatures.

The slower time constant ( $\tau_s$ ) that we assumed to be associated with NaPi-IIb showed strong temperature and voltage dependence (Fig. 4B, C). For ND100 and ND0 superfusion,  $\tau_s$  increased approximately threefold for a decrease in temperature from  $20^\circ\text{C}$  to  $10^\circ\text{C}$ . For ND100 superfusion, the voltage at which  $\tau_s$  peaked shifted toward depolarizing potentials as the temperature decreased,

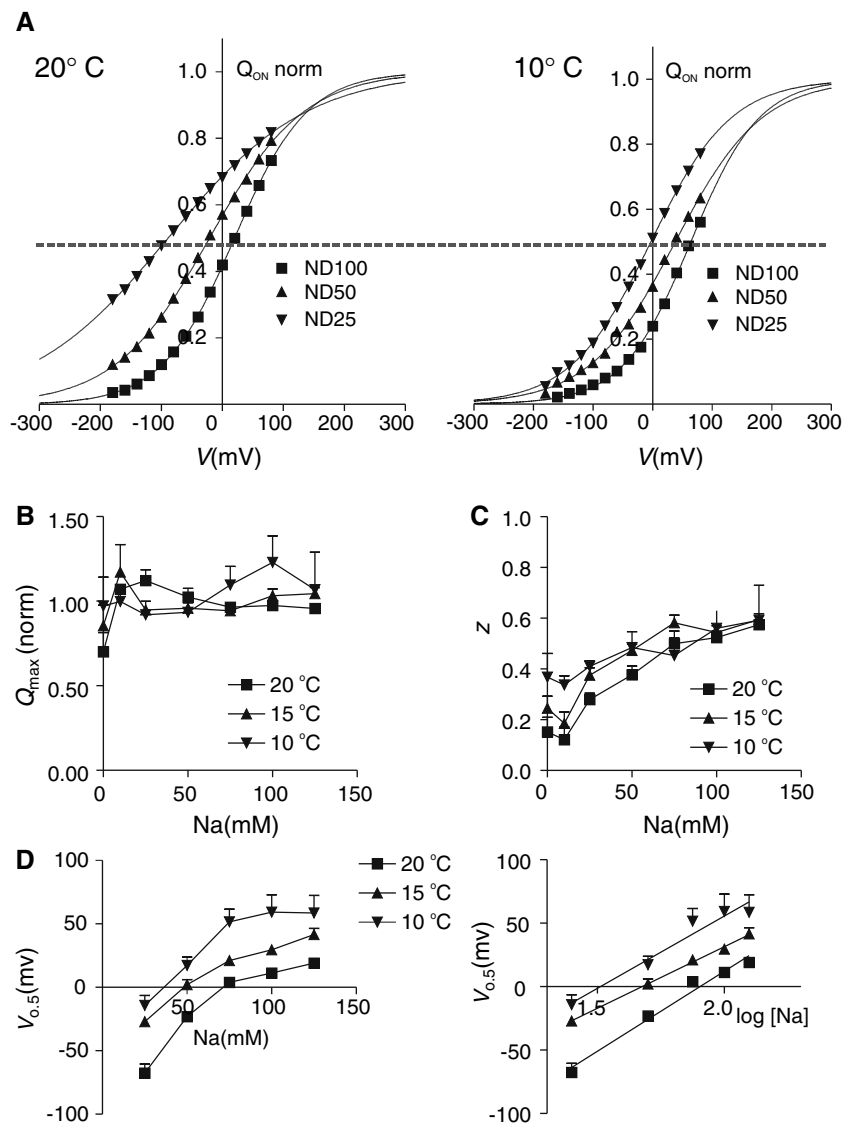
whereas for ND0, a shift of the peak  $\tau_s$  was less obvious. When expressed as  $Q_{10}$  (Fig. 4C), the temperature dependence of the relaxation rate constant ( $1/\tau_s$ ) showed little dependence on  $\text{Na}^+$  for  $V < 0$ . In contrast, for depolarizing potentials,  $Q_{10}$  increased for nonzero  $\text{Na}^+$  concentrations. This behavior suggested that, in addition to the overall slowing of the relaxation rates with decreasing temperature, the voltage dependence of the  $\text{Na}^+$  interaction with the NaPi-IIb protein was more temperature-dependent than that of the empty carrier alone.

To study the underlying mechanism in more detail, we determined the dependence of charge movement on external  $\text{Na}^+$  ranging from 0 (ND0) to 125 (ND125) mM. Numerical integration of the slower relaxations yielded an estimate of the charge ( $Q_{ON}$ ) moved for each voltage step from  $V_h$  to the

corresponding target potential as shown in Figure 5A at two temperatures. To facilitate comparison of data sets, the data were fit with Eq. 1, shifted by the predicted  $Q_{\text{hyp}}$  and normalized to  $Q_{\text{max}}$  for ND100 at 20°C. At 20°C (Fig. 5A, left), a decrease in external  $\text{Na}^+$  resulted in a hyperpolarizing shift in the charge distribution similar to the shift reported in previous studies (Forster et al. 1997; Virkki et al. 2006a, b). At 10°C, a similar hyperpolarizing shift with decreasing  $\text{Na}^+$

concentration was documented. For a given  $\text{Na}^+$  concentration, there was also a relative hyperpolarizing shift of the charge distribution at the lower temperature. For superfusion in ND100, the  $Q$ - $V$  behavior with respect to temperature corresponded qualitatively to that observed for the peak of the  $\tau_s$ - $V$  at ND100 (Fig. 4B, left).

The temperature dependence of the Boltzmann fit parameters with varying external  $\text{Na}^+$  is summarized in



**Fig. 5** Temperature dependence of fit parameters from  $Q$ - $V$  data. (A) Voltage dependence of the charge ( $Q$ - $V$ ) associated with the slow relaxation component for the same cell as in Fig. 4A A at 20°C (left) and 10°C (right) at three external  $\text{Na}^+$  concentrations (ND100, 100 mM; ND50, 50 mM; ND25, 25 mM). Continuous lines are fits with Eq. 1 to the data points. Data for each  $\text{Na}^+$  concentration were normalized to the predicted  $Q_{\text{max}}$  from the curve fit and offset by  $Q_{\text{hyp}}$ . The fit parameters were as follows: at 20°C,  $V_{0.5} = 20 \pm 2$  mV (ND100),  $-21 \pm 4$  mV (ND50),  $-90 \pm 9$  mV (ND25) and  $z = 0.42 \pm 0.01$  (ND100),  $0.32 \pm 0.02$  (ND50),  $0.22 \pm 0.02$  (ND25); at 10°C,  $V_{0.5} = 65 \pm 17$  mV (ND100),  $38 \pm 19$  mV (ND50),  $-2.8 \pm 4.2$  mV (ND25)

and  $z = 0.44 \pm 0.05$  (ND100),  $0.35 \pm 0.05$  (ND50),  $0.37 \pm 0.02$  (ND25). The absolute  $Q_{\text{max}}$  values for individual cells at 20°C and 100 mM  $\text{Na}^+$  ranged 11.5–29.6 nC. (B) Maximum available charge,  $Q_{\text{max}}$ , reported by Boltzmann fit plotted as a function of  $\text{Na}^+$  pooled from six cells and normalized to the  $Q_{\text{max}}$  reported from fit to ND100 data at 20°C. (C) Apparent valence,  $z$ , reported by Boltzmann fit plotted as a function of  $\text{Na}^+$  ( $n = 6$ ). (D) Midpoint voltage ( $V_{0.5}$ ) reported by Boltzmann fit plotted as a function of  $\text{Na}^+$  ( $n = 6$ ) and the same data replotted on a  $\log[\text{Na}]$  scale and fitted with a linear regression line with slopes  $126 \pm 11$  mV (20°C),  $97 \pm 3$  mV (15°C) and  $114 \pm 15$  mV (10°C)



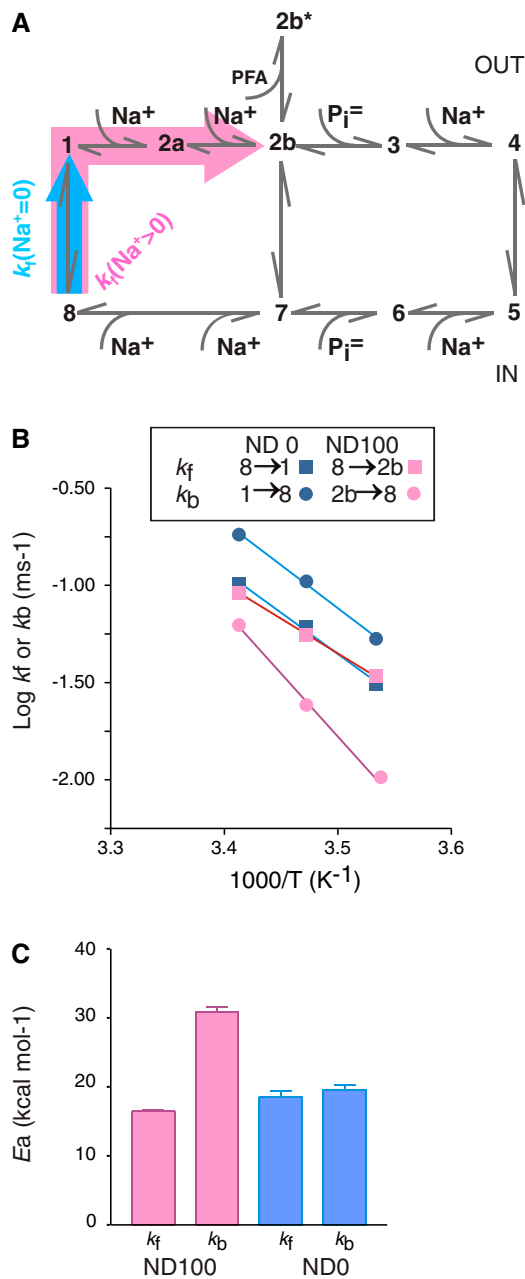


Figure 5B–D ( $n = 6$ ). As expected,  $Q_{\max}$ , the total mobile charge, normalized to that at ND100 and 20°C for each cell (to account for differences in expression level) was relatively insensitive to temperature (Fig. 5B). At 10°C, this parameter was subject to more uncertainty, where the signal-to-noise ratio was worse. Moreover, the normalized  $Q_{\max}$  remained reasonably constant for nonzero Na<sup>+</sup>. The apparent valence ( $z$ ), derived from the maximum slope of the Boltzmann fits, decreased with decreasing Na<sup>+</sup>; however, there was no statistical significance with respect to temperature (one-way analysis of variance,  $P < 0.05$ ) (Fig. 5C). Finally, the voltage at which 50% of the charge

**Fig. 6** Temperature dependence of partial reaction rates. (A) A kinetic scheme of electrogenic type II Na<sup>+</sup>/P<sub>i</sub> cotransport. Two Na<sup>+</sup> ions are proposed to bind before P<sub>i</sub> binding can occur, and transition 2a→2b is voltage-dependent (Virkki et al. 2006b). The PFA-blockable leak current is, in part, due to the translocation of Na<sup>+</sup> ions between states 2a or 2b and 7. In the presence of PFA, the transporter occupies an inactive state (2b\*). With saturating P<sub>i</sub>, the transport cycle proceeds through states 3 and 4 to the fully loaded carrier. Pink block arrow indicates the partial reactions that contribute to presteady-state charge movement with Na<sup>+</sup> in the superfusion medium. Blue block arrow indicates the empty carrier partial reaction (ND0 superfusion). The direction of the forward reaction ( $k_f$ ) is indicated for each. (B) Arrhenius plots of the zero-voltage rate constants ( $k_f$ ,  $k_b$ ) derived from fitting the time constants of the presteady-state relaxations (Fig. 4B) to Eq. 2. Inset identifies the rate constants for the two experimental superfusion conditions: ND100 (pink symbols) and ND0 (blue symbols). For ND100 superfusion, three partial reactions are lumped into one. Lines are linear regression fits to data points. (C)  $E_a$  for  $k_f$  and  $k_b$  derived from the slopes of the linear regression fits to the data in B. Error bars indicate error reported by fit

was moved ( $V_{0.5}$ ) decreased with diminishing Na<sup>+</sup> (Fig. 5D), and there was a clear temperature-dependent shift of this parameter. When the  $V_{0.5}$  data were replotted on a log [Na<sup>+</sup>] scale, the linear regression fits gave slopes for Na<sup>+</sup> varying from 125 to 25 mM of 126 ± 11 mV (20°C), 97 ± 3 mV (15°C) and 114 ± 15 mV (10°C). These slopes were not significantly different over the 10°C range of temperature ( $P < 0.05$ ).

### Discussion

This report describes the effect of temperature on the kinetics of a type II Na<sup>+</sup>/P<sub>i</sub> cotransporter expressed in *Xenopus* oocytes. Our estimates of the temperature dependence for the cotransport mode behavior obtained from uptake and electrophysiology assays, expressed as  $E_a$ , are in good agreement with one another. This is to be expected because we have shown that there is a tight correlation between charge translocation and P<sub>i</sub> cotransport (Forster et al. 1999; Virkki et al. 2005b). Moreover, the corresponding  $Q_{10}$  of ~2.3–2.5 for the P<sub>i</sub>-dependent current agrees with estimates reported for other cation-coupled electrogenic cotransporters, such as SGLT1 (Hazama et al. 1997; Parent and Wright 1993), rGAT1 (Binda et al. 2002) and GAT4 (Karakossian et al. 2005). For the cotransport mode of NaPi-IIb, the  $E_a$  of the P<sub>i</sub>-dependent current ( $I_{Pi}$ ) was essentially independent of membrane potential in the range where the transport rate is strongly voltage-dependent, i.e., for  $V < 0$  mV. If we assume that 1 mM P<sub>i</sub> is sufficient to ensure “ $V_{\max}$ ” conditions for all temperatures investigated, then our finding also accords with the behavior of SGLT1, for which the temperature dependence of the predicted maximum electrogenic sugar-induced

current was also reported to be independent of  $V$  (Parent and Wright 1993). A  $V_{\max}$  effect on the electrogenic activity of the human NaPi-IIa isoform expressed in *Xenopus* oocytes was also documented, with a  $Q_{10}$  of  $\sim 3$  (Wagner 1998), and this study also established that the apparent affinity of  $P_i$  was not significantly altered by temperature.

For the leak mode, quantified as  $I_{\text{PFA}}$ ,  $E_a$  was smaller than that for the cotransport mode. The uncertainty in the fit, especially for hyperpolarizing potentials, did not allow us to make a definite statement about the voltage dependence of  $E_a$  for the leak mode. The difference in  $E_a$  for these two modes suggests distinct transport mechanisms. Certainly, compared to two other cotransport systems, the temperature dependence of the NaPi-IIb leak mode is too large for it to result only from a simple channel-like conductance. For the glutamate transporter EAAT1 (Beckman and Quick 2001; Wadiche and Kavanaugh 1998) and the GABA transporter GAT4 (Karakossian et al. 2005),  $Q_{10}$  values close to unity were found for their respective leak modes, consistent with an electrodiffusive mechanism. In contrast, for the sodium-coupled glucose transporter SGLT1, an  $E_a = 21 \pm 3 \text{ kcal mol}^{-1}$  was documented (Loo et al. 1999). Like NaPi-IIb in the present study, this finding suggests that the leak mode in SGLT1 also involves large conformational changes. On the other hand, from the steady-state data alone, we are unable to exclude the possibility that there is an additional “channel” component of the leak pathway, which may also be PFA-sensitive and conductive for other ions, including  $\text{Cl}^-$  (I. C. Forster, L. V. Virkki, A. Bacconi, unpublished observations). Caution should also be observed with respect to overinterpreting the apparent differences in  $E_a$  for the two modes, given the uncertainties reported by the fitting algorithm and the fact that for flounder NaPi-IIb  $I_{\text{PFA}}$  is typically  $\leq 10\%$  of  $I_{\text{Pi}}$  and therefore more prone to contamination from endogenous PFA-dependent currents (Forster et al. 2000). It should also be noted that  $E_a$  determined from the Arrhenius plot represents an effective estimate in which the transport process (leak or cotransport mode) is lumped into a single reaction that obeys first-order kinetics. This is an oversimplification because a number of partial reactions have already been identified as contributors to the overall cotransport cycle (Forster et al. 1998; Virkki et al. 2005b, 2006b), any of which might dominate the overall temperature dependence (Fig. 6A). From our data, we can conservatively conclude that the estimates of  $E_a$  for the leak mode strongly indicate that conformational changes take place that are inconsistent with an exclusively electrodiffusive transport process.

From the presteady-state data analysis, we found that the relaxations in the presence and absence of external  $\text{Na}^+$  showed significant temperature dependence (Fig. 4). As expected, the predicted  $Q_{\max}$  from the Boltzmann data was

insensitive to temperature for  $\text{Na}^+ > 25 \text{ mM}$ . This result confirmed that we were able to detect all the charge over the temperature range used, except at low  $\text{Na}^+$  concentrations, where the curve fitting and numerical integration were less reliable. It also confirmed that the number of active transporters in the membrane did not change with temperature. Moreover, we observed no significant temperature dependence of the apparent valence ( $z$ ) for the Boltzmann fit.<sup>1</sup> In contrast, the midpoint voltage ( $V_{0.5}$ ) of the  $Q$ - $V$  distribution for a given  $\text{Na}^+$  concentration was shifted toward depolarizing potentials as the temperature was lowered (Fig. 5C), whereas the slopes of the  $V_{0.5}$  vs.  $\log[\text{Na}^+]$  plots were not statistically altered by temperature. The linear dependence of  $V_{0.5}$  on  $\log[\text{Na}^+]$  is the behavior predicted, e.g., for a two-state “ion-well” model for voltage-dependent  $\text{Na}^+$  interaction with a carrier (Mager, Cao and Lester 1998; Mager et al. 1996; Virkki et al. 2006b). The  $Q$ - $V$  relation for this model is a modified version of Eq. 1 to take account of  $\text{Na}^+$  dependence of the binding reaction:

$$Q = Q_{\max} / [(1 + \{[\text{Na}^+] / K^{\text{Na}} \exp(-Ve\delta/kT)\} n_{\text{Na}})] + Q_{\text{hyp}} \quad (5)$$

where  $K^{\text{Na}}$  is a zero-voltage  $\text{Na}^+$  binding constant that corresponds to  $[\text{Na}^+]$  when half the charge has moved and  $n_{\text{Na}}$  is the number of  $\text{Na}^+$  ions that bind at a hypothetical site located a fraction  $\delta$  through the membrane field.<sup>2</sup> For NaPi-IIb, we have shown that  $n_{\text{Na}} = 1$  (Virkki et al. 2006b) and the slope of  $V_{0.5}$  vs.  $\log[\text{Na}^+] = 2.303 \text{ kT}/(e\delta)$ , from which we can estimate  $\delta$ . Our data (Fig. 5C) suggest that  $\delta$  is not a function of temperature, which implies that  $\text{Na}^+$  ions move through the same fraction of the transmembrane field, independent of temperature. The mean of the slopes at the three test temperatures gave  $\delta = 0.51$ , which is consistent with our previous analyses of the NaPi-IIb presteady-state kinetics (Virkki et al. 2006a, b). On the other hand, the intercept with the  $\log[\text{Na}^+]$  axis shifted to the left with decreasing temperature, which indicated that  $K^{\text{Na}}$  decreased with decreasing temperature. In terms of the simple two-state model for charge translocation (see Materials and Methods),  $K^{\text{Na}}$  can also be expressed as the ratio of the zero-voltage rates ( $k_f$ ,  $k_b$ ), and therefore the observed dependence of  $K^{\text{Na}}$  on temperature could result from a change in either rate constant or both.

<sup>1</sup> The dependence of  $z$  on  $[\text{Na}^+]$  is a consequence of fitting a single Boltzmann function to the  $Q$ - $V$  data. The valence estimated from the single fit is a weighted value that depends on the relative contributions from the empty carrier and  $\text{Na}^+$  binding partial reactions. Similar behavior was observed by fitting a single Boltzmann to presteady-state data generated by a three-state model that simulates the empty carrier and one voltage-dependent  $\text{Na}^+$  binding step (I. C. Forster, unpublished data).

**Table 1** Temperature dependence of fit parameters for  $\tau$ - $V$  data

| Temp (°C) | ND100                     |                           |             | ND0                       |                           |             |
|-----------|---------------------------|---------------------------|-------------|---------------------------|---------------------------|-------------|
|           | $k_f$ (ms <sup>-1</sup> ) | $k_b$ (ms <sup>-1</sup> ) | $z$         | $k_f$ (ms <sup>-1</sup> ) | $k_b$ (ms <sup>-1</sup> ) | $z$         |
| 20        | 0.092 ± 0.004             | 0.063 ± 0.003             | 0.40 ± 0.02 | 0.102 ± 0.009             | 0.183 ± 0.011             | 0.18 ± 0.05 |
| 15        | 0.056 ± 0.003             | 0.024 ± 0.003             | 0.40 ± 0.03 | 0.061 ± 0.001             | 0.105 ± 0.002             | 0.23 ± 0.01 |
| 10        | 0.034 ± 0.002             | 0.010 ± 0.002             | 0.37 ± 0.04 | 0.031 ± 0.001             | 0.053 ± 0.002             | 0.27 ± 0.02 |

Fit parameters are given by Eq. 3. Errors are standard errors reported by fit algorithm

To distinguish between these possibilities, we fitted the  $\tau_s$ - $V$  data (Fig. 4B) for superfusion in ND100 with Eq. 2 to obtain estimates for  $k_f$  and  $k_b$ , as summarized in Table 1. These data confirmed that both rates became slower with decreasing<sup>2</sup> temperature. To further quantify the temperature dependence of the fit-derived reaction rates, we plotted these data in the Arrhenius form (Fig. 6B). For ND100 superfusion, the data show that the temperature dependence of the effective backward rate ( $k_b$ ) of the lumped Na<sup>+</sup> interaction (i.e., by considering the transitions  $8 \Leftrightarrow 1 \Leftrightarrow 2a \Leftrightarrow 2b$  as a single transition [Fig. 6A]) was greater than that of the effective forward reaction ( $k_f$ ). In terms of a simple ion-well model, we interpret this as indicating that the apparent affinity for Na<sup>+</sup> ions increases with decreasing temperature. We also fitted the  $\tau_s$ - $V$  data for ND0 superfusion with Eq. 2, to investigate the temperature dependence of the empty carrier (transition  $8 \Leftrightarrow 1$ , Fig. 6A). Here, the Arrhenius analysis showed that the temperature dependence for the effective  $k_b$  and  $k_f$  was similar and close to that of  $k_f$  with ND100 superfusion (Fig. 6B). Figure 6C summarizes these findings in terms of the activation energy for each rate constant.

Our finding that  $E_a$  determined under voltage clamp for the cotransport mode is essentially independent of membrane potential suggests that a major electroneutral conformational change occurs during the transport cycle. One likely candidate for the corresponding transition, according to the scheme of Figure 6A, is the translocation of the fully loaded carrier (transition  $4 \Leftrightarrow 5$ ), which we have previously proposed to be electroneutral (Forster et al. 1998) but without experimental evidence. In the present study, this proposal is supported by our finding that  $E_a$  for the empty

carrier zero-voltage rate constants corresponds closely to the  $E_a$  estimated from the steady-state analysis for the cotransport mode. Thus, in the absence of membrane potential, the partial reactions that describe the reorientation of the empty and fully loaded carrier are energetically similar. On the other hand, given that we have no information about the voltage dependence of the partial reactions between intracellularly oriented states, we cannot exclude that conformational changes associated with other transitions between states 5 and 8 also contribute to the voltage-independent activation energy observed for cotransport.

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<sup>2</sup> In applying this model to NaPi-IIb, we have lumped the empty carrier transition (1–8, Fig. 6A) and Na<sup>+</sup> binding transitions (1–2a, 2a–2b) into a single transition. A more accurate description of the presteady-state kinetics would require fitting the presteady-state relaxations to >2 exponentials and using at least a two Boltzmann fit to the data. We were unable to extract more than one component of nonendogenous presteady-state charge movement by exponential curve fitting with confidence. Moreover, attempts to fit a double Boltzmann function to our  $Q$ - $V$  data (e.g., as undertaken by Krofchick, Huntley and Silverman [2004] for SGLT1) were equally unsuccessful because of fit uncertainties with the larger number of free parameters.

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