

Thermal and moisture flux in soft fibreboards

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Abstract The thermal conductivity of several wet and dry processed industrial soft fibreboards was determined at temperatures between 10°C and 30°C. Additionally, the thermal conductivity was tested relative to the moisture content ranging from an oven-dry sample to a moisture content of 85% relative humidity. Furthermore, the water vapour diffusion was investigated to determine the resistance factor and the diffusion coefficient under ‘dry cup’ and ‘wet cup’ conditions and the sorptions were tested by means of the sorption isotherms.

The thermal conductivity increases with increasing temperature at about 0.45% per Kelvin and with increasing moisture at about 0.17×10^{-2} W/mK per percent volumetric moisture content. Furthermore, the thermal conductivity depends on the manufacture (dry or wet process) and is influenced by density. The water vapour resistance factor increases and the diffusion coefficient decreases with increasing density. Both factors decrease with increasing moisture content and they are dependant on board thickness, composition and manufacturing. The sorption isotherm is similar to solid wood below about 80% relative humidity but strongly increases for soft fibreboards with binding fibres and additives of ammonium phosphate or sodium borate and boric acid when above 80% relative humidity.

Wärme- und Feuchteverhalten poröser Holzfaserrplatten

Zusammenfassung An verschiedenen, nach dem Nass- oder nach dem Trockenverfahren, industriell hergestellten

Holzfaserrdämmplatten wurde die Wärmeleitfähigkeit in Abhängigkeit von der Temperatur zwischen 10°C und 30°C geprüft. An einem Teil der Platten wurde die Wärmeleitfähigkeit zusätzlich in Abhängigkeit von der Feuchte, von darrtrocken bis zu einer Feuchte bei 85 % relativer Luftfeuchte, bestimmt. Weiterhin wurden die Wasserdampfdiffusion unter Ermittlung der Diffusionswiderstandszahlen und der Diffusionskoeffizienten im Trocken- und im Feuchtbereich sowie das Sorptionsverhalten anhand der Sorptionsisothermen untersucht.

Die Wärmeleitfähigkeit steigt mit zunehmender Temperatur um 0,45 % pro Kelvin und mit zunehmender Feuchte um $0,17 \times 10^{-2}$ W/mK pro Prozent volumenbezogenen Feuchtegehalts. Zusätzlich besteht eine Abhängigkeit von der Dichte sowie vom Herstellungsverfahren (Nass- oder Trockenverfahren). Während mit zunehmender Dichte die Wasserdampfdiffusionswiderstandszahl steigt, verringert sich der Diffusionskoeffizient. Beide Kenngrößen nehmen mit zunehmender Feuchte ab und sind abhängig von der Probendicke, der Zusammensetzung und dem Herstellungsverfahren. Bis ca. 80 % relativer Luftfeuchte verhält sich die Sorptionsisotherme ähnlich wie diejenige von Vollholz, oberhalb davon steigt sie jedoch bei den Holzfaserrdämmplatten mit Bindefasern sowie Zusätzen von Ammoniumphosphat oder Borsäure und Borax steil an.

1 Introduction

Soft fibreboards constructed of wood fibres are used as insulating materials with specialized heat insulation and storage properties that induce a high phase lag from the external wall temperature to the interior room. Therefore, they are suitable for winter and summer heat insulation and have

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become more important in recent years. Certain characteristics of the materials, such as the thermal conductivity, water vapour diffusion and water sorption, are important for structural-physical calculations.

General mean values of thermal conductivity and water vapour resistance of insulating materials are available from industrial data sheets partially listed in reference books (e.g. Reyer et al. 2002). Thereby, the water vapour resistance factor is mostly indicated as $\mu = 5$ for soft fibreboards and $\mu = 1\text{--}2$ for mineral fibrous insulating materials. Further, natural insulating materials beside soft fibreboards like cellulose, flax and hemp, sheep wool and cotton have similar values to mineral fibrous insulating materials (Albrecht 1997). The thermal conductivity of these insulating materials mostly ranges between 0.040 and 0.055 W/mK, but is lower for polyurethane and polystyrene cellular plastics that have, in contrast, a clearly higher water vapour resistance than the above mentioned materials (Albrecht 1997; Reyer et al. 2002). The theoretical bottom value of the thermal conductivity of these materials (if their density tends to zero) is given by the thermal conductivity of air with 0.026 W/mK neglecting radiation and air flow (Kollmann 1951; Kollmann and Côté 1968; Scheiding 1998).

Therefore, for wood and wood-based materials, the thermal conductivity increases rather linearly with increasing density in the density range of 200 to 800 kg/m³ but decreases with decreasing particle size of the material (Kollmann 1951; Kollmann and Malmquist 1956; Kollmann and Côté 1968; Sonderegger and Niemz 2009). However, for soft fibreboards and other insulating materials with very low densities, the reduction of the thermal conductivity with decreasing density diminishes, eventually stops and then changes below about 40 kg/m³ and begins to increase due to cumulative gaseous flow and radiation exchange in the voids (Häupl 2008). This behaviour can also be found for wood-based loose-fill thermal insulation materials (Scheidung 1998; Schunk et al. 2009), hemp fibres (Grohe 2004) and textiles (Martin and Lamb 1987; Schardt et al. 1993).

Many sorption theories exist (Skaar 1988 gives a broad overview) and sorption behaviour of the main wood species and wood-based materials are also very well known (recent investigations e.g. Sonderegger and Niemz 2006; Popper and Niemz 2009), however the sorption of soft fibreboards has scarcely been investigated. According to Scheiding (1998) the sorption behaviour of wood fibres and therefore soft fibreboards is comparable to solid wood. In recent years, investigations have been carried out on diverse natural fibres, not only for use as insulating material but also for polymer composite material (Alix et al. 2009; Bessadok et al. 2009; Hill et al. 2009). Diverse factors such as the manufacturing method (Seifert 1972), the treatment of the fibres (Bessadok et al. 2009) and the additives (Kollmann and Schneider 1959; Scheiding 1998; Lesar et al. 2009) influence the sorption of natural fibres and fibreboards. The

Hailwood-Horrobin model is often used to characterise the sorption isotherms of wood and adequately describes the sorption of other natural fibres (Hill et al. 2009), and is used for modelling moisture transport in wood (e.g. Frandsen et al. 2007).

In this study the properties of thermal conductivity, water vapour diffusion and water sorption were investigated for different types of soft fibreboards relative to the temperature, moisture content (MC) and density of the material.

2 Materials and methods

2.1 Thermal conductivity

28 industrially fabricated types of soft fibreboard varying in thickness and density from four different companies were investigated. 16 types were manufactured with the wet process and 12 types with the dry process. Table 1 gives an overview of the investigated boards.

The specimens used to determine the thermal conductivity were sized 500 mm × 500 mm × sample thickness. Before testing, the boards were conditioned at standard climatic conditions of 20°C and 65% relative humidity (RH) and 5 boards were additionally conditioned at 20°C and 80% RH as well as at 35% RH and at oven dry conditions for the purpose of determining the influence of moisture (Table 1). The measurements were carried out with the guarded hot plate apparatus λ -Meter EP500 (Lambda-Messtechnik GmbH, Dresden) according to ISO 8302:1991 at 10, 20, and 30°C. With a linear regression through the values at the three temperatures, the thermal conductivity at 10°C ($\lambda_{10,\text{reg}(T)}$) and the change of the thermal conductivity with increasing temperature per degree Celsius ($\Delta\lambda_{(T)/^\circ\text{C}}$) were determined according to (1):

$$\lambda_{(T)} = \lambda_{10,\text{reg}(T)} + \Delta\lambda_{(T)/^\circ\text{C}} \cdot (T - 10) \quad (1)$$

where T is the temperature [°C].

2.2 Water vapour diffusion

For measuring the water vapour resistance of the investigated soft fibreboards and the water vapour diffusion perpendicular to the plane, three cylindrical specimens of 140 mm diameter x board thickness per climate and type were tested (Table 1). The tests to determine the water vapour resistance were carried out according to ISO 12572:2001 under ‘dry cup’ (20°C–0/65% RH) and ‘wet cup’ conditions (20°C–100/65% RH). Figure 1 shows the test assembly. The specimens were first conditioned at standard climatic conditions (20°C/65% RH) and then put on top of a glass vessel filled up to approximately 20 mm under the brim with either a desiccant (silica gel) or totally

Table 1 Overview of the investigated types of soft fibreboard and the number of tested specimens
Tab. 1 Überblick über die untersuchten Holzfaserdämmplatten und die Anzahl der Proben

Class	Company	Additives	Manufacturing method	Thickness [mm]	Density [kg/m ³]	Number of tested specimens per type		
						Thermal conductivity ⁴	Diffusion	Sorption
1 (2 types)	1	Binding fibres (polyolefin); ammonium polyphosphate	Dry process	30, 60	55	3	2 × 3	3
2 (2 types)	1	Cellulose particles from newsprint (without wood fibres); binding fibres (polyolefin); sodium borate; boric acid	Dry process	30, 60	75	3	2 × 3	3
3 (3 types)	2	PUR-resin (4%)	Dry process	60	120, 135, 140	3	2 × 3	3
4 (2 types)	2	PUR-resin (4%); paraffin (2%)	Dry process	60	165, 240	3	2 × 3	3
5 (3 types) ¹	3	Binding fibres (copolyester); ammonium phosphate	Dry process	60, 100, 160	140, 105, 50	5	–	3
6 (2 types)	2	Saponified wood resin; wax emulsion	Wet process	18, 36	260	5	2 × 3	3
7 (8 types) ²	2	–	Wet process	20, 60 (50, 80)	150, 180, 185 (220)	5	2 × 3	3
8 (2 types) ³	4	Starch (5%); paraffin (0.5%)	Wet process	20	185, 215	5	2 × 3	3
9 (2 types) ¹	4	–	Wet process	19, 22	205, 130	5	2 × 3	3
10 (2 types)	4	Latex	Wet process	22	225, 245	5	2 × 3	3

¹Types attribution: Each thickness corresponds to only one density (same order) per sample type

²Types attribution: The thicknesses in brackets correspond to the density in brackets and the other thicknesses to the other densities

³One sample type (density 185 kg/m³) without paraffin

⁴Tests at standard climatic conditions and for 5 sample types (1 sample type of class 1, class 2 and class 7 (only 3 specimens per type) and both sample types of class 9) additionally at three further climates

desalinated water. The specimens were laterally sealed with a tight-fitting circular rubber band. After attaining the equilibrium moisture content (EMC), the vessels with the specimens were weighed seven times with an interval of 1 to 3 days between each weighing depending on the thickness and the permeability of the respective specimen. The water vapour resistance factor was calculated according to Annex G of ISO 12572 taking into account the thickness of the air layer in the test cup between the base of the specimen and the desiccant or the desalinated water, respectively (cf. Sonderegger and Niemz 2009):

$$\mu = \frac{\delta_a}{d} \cdot \left(\frac{A \cdot \Delta p_v}{G} - \frac{d_a}{\delta_a} \right) \quad (2)$$

where μ is the water vapour resistance factor [–], δ_a the water vapour permeability of air with respect to the partial vapour pressure [kg/(m s Pa)], d the mean thickness of the specimen [m], A the area of the specimen [m²], Δp_v the water vapour pressure difference across the specimen [Pa],

G the water vapour flow rate through the specimen [kg/s], and d_a the thickness of the air layer in the test cup between the base of the specimen and the desiccant or the desalinated water [m].

In contrast to the water vapour resistance factor, which was determined on the basis of a water vapour pressure difference, the diffusion coefficient was determined on the basis of a water concentration difference within the specimen according to Siau (1995, p. 111) following Fick's first law:

$$D = \frac{G}{A} \cdot \frac{d}{\Delta c} \quad (3)$$

where D is the diffusion coefficient [m²/s] and Δc the water concentration difference [kg/m³].

Thereby, for determining the water concentration difference in the specimen, the gradient of the MC through the specimen was assumed to be linear. Thus, with the values of the EMC at standard climatic conditions (20°C/65% RH corresponding to the upper side of the specimen) and the

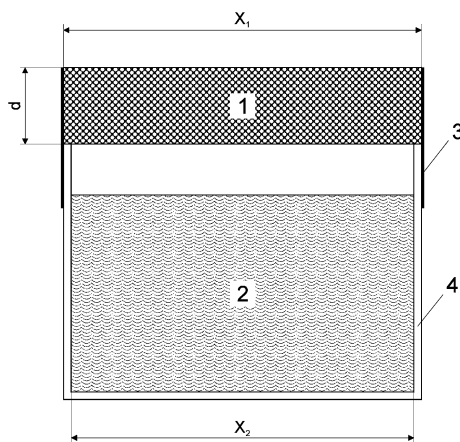


Fig. 1 Water vapour transmission properties: Test assembly for the ‘dry cup’ and ‘wet cup’ tests. 1—test specimen with X_1 and X_2 as the upper and lower exposed area and with the thickness d ; 2—desiccant/desalinated water; 3—rubber band; 4—glass vessel

Abb. 1 Wasserdampfdurchlässigkeit: Prüfanordnung unter ‘dry cup’ und ‘wet cup’ Bedingungen. 1—Probekörper mit X_1 und X_2 als oberer und unterer Prüffläche und mit der Dicke d ; 2—Trockenmittel/entsalztes Wasser; 3—Gummiband; 4—Glasgefäß

mean moisture of the specimen during the diffusion experiment, the water concentration difference could be calculated as follows:

$$\Delta c = 2 \cdot |MC_{20/65} - MC_{\text{mean}}| \cdot \frac{V}{m_{\text{dry}}} \quad (4)$$

where $MC_{20/65}$ is the moisture content at 20°C/65% RH [–], MC_{mean} the mean moisture content of the specimen during the experiment [–], V the volume of the specimen [m^3], and m_{dry} the oven-dry weight [kg].

2.3 Sorption

The sorption was tested on three specimens of 10 cm × 15 cm × board thickness per type measuring the EMC for the adsorption at 20°C and 35%, 50%, 65%, 80% and 93% RH and for the desorption (only for the dry processed boards) at 20°C and 93%, 90%, 80%, 65%, 50% and 35% RH (Table 1). The specimens were dried at 60°C before the adsorption tests. The single hydrate Hailwood-Horrobin model was used for determining the sorption isotherms and is comparable to the Dent-model. Therefore, according to Skaar (1988), the parameters A , B and C were calculated for all board types from the experimental data according to (5). From this, the complete hydration or complete monolayer coverage of all available sorption sites (6) and the total sorption isotherm (7) as well as the hydrated and dissolved water of the Hailwood-Horrobin model (8) and (9) and the primary and secondary sorbed water of the Dent model (10) and (11) were determined:

$$\frac{h}{m} = A + B \cdot h - C \cdot h^2 \quad (5)$$

$$M_0 = \frac{100}{\sqrt{(B^2 + 4 \cdot A \cdot C)}} \quad (6)$$

$$M = M_h + M_d = M_1 + M_2 \quad (7)$$

$$M_h = M_0 \cdot \frac{K_h \cdot K_d \cdot h}{1 + K_h \cdot K_d \cdot h} \quad (8)$$

$$M_d = M_0 \cdot \frac{K_d \cdot h}{1 - K_d \cdot h} \quad (9)$$

$$M_1 = M_0 \cdot \frac{b_1 \cdot h}{1 - b_2 \cdot h + b_1 \cdot h} \quad (10)$$

$$M_2 = M_0 \cdot \frac{b_1 \cdot b_2 \cdot h^2}{(1 - b_2 \cdot h) \cdot (1 - b_2 \cdot h + b_1 \cdot h)} \quad (11)$$

where

$$K_d = b_2 = \frac{-B + \sqrt{B^2 + 4 \cdot A \cdot C}}{2 \cdot A} \quad (12)$$

$$K_h = \frac{B}{A \cdot K_d} + 1 \quad (13)$$

$$b_1 = \frac{B}{A} + 2 \cdot b_2 \quad (14)$$

h Relative humidity [–]

m Moisture content [–]

A, B, C Parameter [–]

M_0 MC corresponding to the complete hydration (Hailwood-Horrobin model) or complete monolayer coverage of all available sorption sites (Dent model) [%]

M Total MC [%]

M_h MC of the hydrated water [%]

M_d MC of the dissolved water [%]

M_1 MC of the primary sorbed water [%]

M_2 MC of the secondary sorbed water [%]

K_d, K_h Coefficients of the Hailwood-Horrobin model [–]

b_1, b_2 Coefficients of the Dent model [–]

To ascertain the curve given by (5) in the lower humidity range, in addition to the values of the EMCs at the above-mentioned relative humidities, the MC after drying at 60°C was also used as an auxiliary data point under the assumption that the humidity at 60°C corresponds to a RH of 4.7% (equal to 40% RH at 20°C for a similar vapour pressure). Following isohydro curves on this humidity range (e.g. for Sitka spruce in Niemz 1993) the MC of a defined RH at 60°C corresponds to about two-thirds of this RH at 20°C. Thus for this auxiliary data point a RH of 3.13% was used.

3 Results and discussion

3.1 Thermal conductivity

Figure 2 shows the thermal conductivity dependant on density. Overall, the thermal conductivity increases with increasing density. As one can see, the results of the wet and

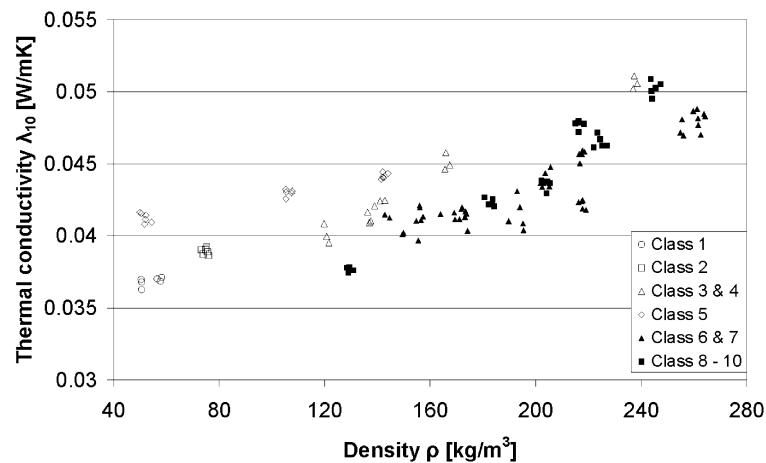


Fig. 2 Thermal conductivity depending on the density and grouped by classes. *Empty symbols* represent dry processed and *solid symbols* represent wet processed materials. λ_{10} = Thermal conductivity at 10°C/65% RH

Abb. 2 Wärmeleitfähigkeit in Abhängigkeit von der Dichte unterteilt nach Klassen. *Leere Symbole* bezeichnen nach dem Trockenverfahren und *ausgefüllte Symbole* nach dem Nassverfahren hergestellte Materialien. λ_{10} = Wärmeleitfähigkeit bei 10°C/65% relativer Luftfeuchte

the dry processed boards sit distinctly apart. Thereby, two points attract special attention: Firstly, the lowest density of the wet processed materials is located at about 130 kg/m³ due to the process conditions, whereas a density of 50 kg/m³ for the dry processed boards can be manufactured. Secondly, the thermal conductivity at the same density is lower for the wet processed than for the dry processed boards. This can be partially attributed to the glue fraction and the porosity distribution but is predominantly a result of the orientation of the fibres in the board. The thermal conductivity parallel to the grain is about 2.25 to 2.75 times higher than perpendicular to the grain (Kollmann and Côté 1968). This signifies that the orientation of the grain in the boards impacts the extent of the thermal conductivity. Due to the high water content of the wet process and the lower thickness of a single board, the fibres are highly oriented in the plane of the board, clearly more so than by the dry process. As an example, the higher insulating effect of multi-layered solid wood panels with soft fibreboards as middle layers totally disappears if the middle layers were cut into strips and turned 90° and glued again so that the originally in plane orientation was now perpendicular to the plane (Schulz and Tobisch 2006).

With $\lambda < 0.040$ W/mK below a density of 140 kg/m³, Müller (1998) determined similar values for wet processed starch bonded fibreboards on wet processed materials, as was measured in this study.

A phenomenon already observed for insulating material by Häupl (2008), the decrease of the thermal conductivity in Fig. 2, diminishes with decreasing density for very low densities.

The mean increase of the thermal conductivity with increasing temperature ($\Delta\lambda_{(T)}/^{\circ}\text{C}$ according to (1)) in the

range of 10°C to 30°C amounts (over all types of soft fibreboards) to $0.19 \pm 0.05 \times 10^{-3}$ W/mK or $0.45 \pm 0.13\%$ per Kelvin. The absolute increase is lower but the percentage increase is similar or higher compared to values determined on wood-based boards of medium or high density (OSB, MDF and others) in Sonderegger and Niemz (2009).

Table 2 shows the thermal conductivity of five board types depending on MC and the corresponding regression parameters. For one of these boards (type 2 of class 9), the increase of the thermal conductivity with increasing MC at the three measured temperatures (10°C, 20°C, 30°C) is shown in Fig. 3.

Furthermore, the increase of the thermal conductivity with increasing MC depends on density if the MC is dry-mass-referred as it is usual for wood and wood-based materials (cf. Table 2). In contrast, according to Cammerer (1987) the increase of the thermal conductivity depending on volumetric MC gives similar values for similar materials independent of the density. Figure 4 illustrates this dependency for the investigated boards. The increase in thermal conductivity is similar for all boards. The mean increase per percent volumetric MC amounts to 0.17×10^{-2} W/mK and is about 30% lower than given by Cammerer for wood and cork. Another approach towards analyzing the moisture dependent increase of the thermal conductivity is by determining the percentage increase of the thermal conductivity per MC (dry mass referred) specified in Künzel (1986). This alternative approach is also independent of the density (for materials with higher densities and a high correlation between thermal conductivity and density) and has the advantage that the hygroscopic equilibrium of the moisture is taken into account. However, for the investigated soft fibreboards, the influence of the density with this calculation method is still high. Thus, the percentage increase of

Table 2 Thermal conductivity (λ) dependant on the moisture content (ω) determined on the regression $\lambda = A + B \cdot \omega$, where $A = \lambda_{10,\text{dry}}$ (λ at dry condition and 10°C) and $B = \Delta\lambda_{\omega}$ (change of λ per percent MC in W/(m K))

Tab. 2 Wärmeleitfähigkeit (λ) in Abhängigkeit vom Feuchtegehalt (ω), ermittelt anhand der Regressionsgleichung $\lambda = A + B \cdot \omega$, wobei $A = \lambda_{10,\text{dry}}$ (λ von darrtrockenen Platten bei 10°C) und $B = \Delta\lambda_{\omega}$ (Änderung von λ pro Prozent Feuchteänderung in W/(m K))

Material	Thickness ¹ [mm]	Density ¹ [kg/m ³]	ω [%] at climates			Regressionparameter		
			20/35	20/65	20/85	A (= $\lambda_{10,\text{dry}}$)	B (= $\Delta\lambda_{\omega}$)	R ²
Class 1, type 1	29	55	7.6	10.1	14.6	0.0360	0.90×10^{-4}	0.96
Class 2, type 1	31	75	8.1	11.1	15.2	0.0371	1.38×10^{-4}	0.95
Class 7, type 1	22	150	8.1	10.3	15.1	0.0384	2.06×10^{-4}	0.87
Class 9, type 1	21	205	8.2	11.3	15.2	0.0411	3.25×10^{-4}	0.89
Class 9, type 2	23	130	7.8	10.3	14.2	0.0360	2.04×10^{-4}	0.91

¹At climatic condition of 20°C and 65% RH

Fig. 3 Thermal conductivity depending on MC of type 2 of class 9 (density 130 kg/m³)

Abb. 3 Wärmeleitfähigkeit in Abhängigkeit von der Feuchte (Variante 2 der Klasse 9: Dichte 130 kg/m³)

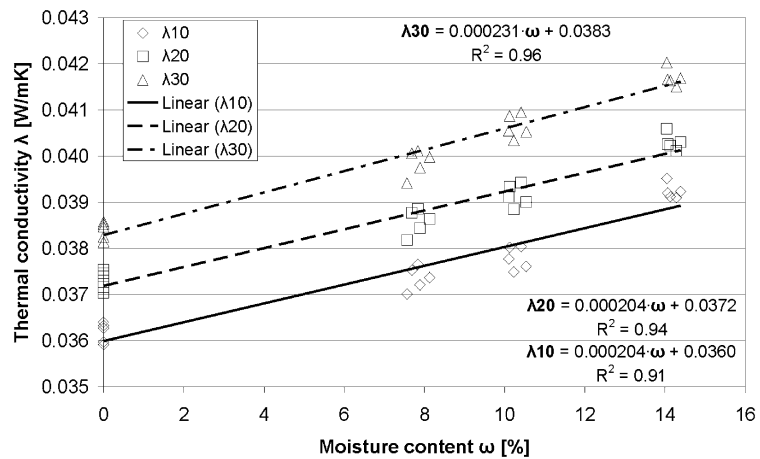
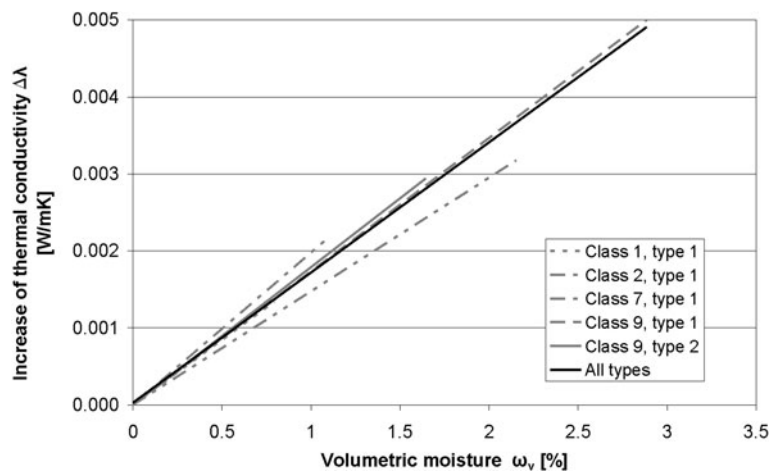


Fig. 4 Increase of thermal conductivity depending on volumetric moisture

Abb. 4 Zunahme der Wärmeleitfähigkeit in Abhängigkeit vom volumenbezogenen Feuchtegehalt



the thermal conductivity per 1% change of MC lies between 0.25% and 0.79% and increases with increasing density. The mean value corresponds to tests on cellulose flakes insulation by Kehrer et al. (2001) and sodium and potassium silicate bonded soft fibreboards in the lower humidity range by Scheiding (1998).

3.2 Water vapour diffusion

Figure 5 shows the water vapour resistance factor μ of the ‘wet cup’ and the ‘dry cup’ tests depending on density. The mean values are listed in Table 3. All types of soft fibreboard have a low resistance factor but a clear difference between

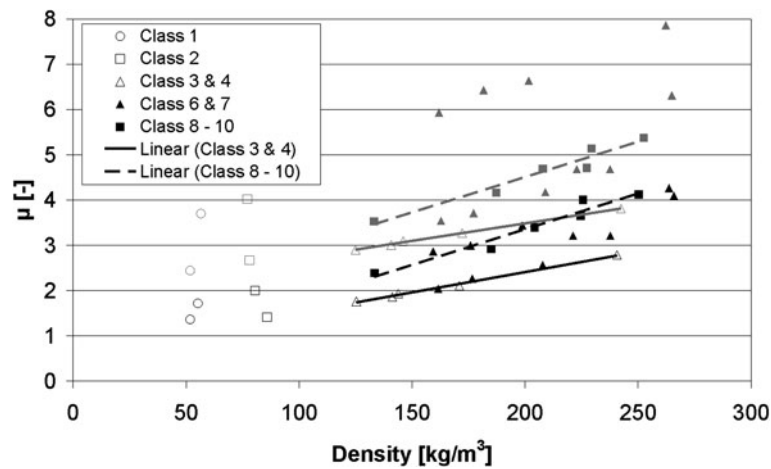


Fig. 5 Water vapour resistance factor (μ) of the ‘wet cup’ tests (black symbols) and the ‘dry cup’ tests (grey symbols) depending on the density (mean values). Empty symbols represent dry processed and solid symbols represent wet processed materials

Abb. 5 Wasserdampfdiffusionswiderstandszahlen (μ) im Feuchtbereich (schwarze Symbole) und im Trockenbereich (graue Symbole) in Abhängigkeit von der Dichte (Mittelwerte). Leere Symbole bezeichnen nach dem Trockenverfahren und ausgefüllte Symbole nach dem Nassverfahren hergestellte Materialien

Table 3 Mean water vapour resistance factors (μ) of the dry and wet processed soft fibreboards

Tab. 3 Gemittelte Wasserdampfdiffusionswiderstandszahlen (μ) der im Trocken- und Nassverfahren hergestellten Holzfaserdämmplatten

	$\mu_{\text{wet-cup}}$	$\mu_{\text{dry-cup}}$	μ_{mean}
Dry processed	1.89 ± 0.42	3.22 ± 0.53	2.55 ± 0.44
Wet processed	3.22 ± 0.70	5.10 ± 1.24	4.16 ± 0.90
All types	2.74 ± 0.89	4.42 ± 1.38	3.58 ± 1.09

dry and wet processed boards can be observed. Further, the water vapour resistance factor is, up to factor 2, higher in the dry climate than in the wet climate and increases in both climates with increasing density but only with a low correlation due to other different factors influencing the water vapour resistance such as composition, thickness and manufacturing (Fig. 5). Therefore, a linear regression curve is plotted for only two groups of boards with similar properties and thicknesses within the group.

Figure 6 shows the diffusion coefficients calculated according to (3) depending on density. In both humidity ranges, the dry processed boards show higher diffusion coefficients than the wet processed boards. This may be due to the reduced percentage of in plane orientation fibres of the dry processed boards.

The diffusion coefficient in the dry climate is 1.5 to 3.1 times higher than in the wet climate, although the water vapour resistance factor has similar tendencies. This is due to the fact that the MC at high RH increases exponentially and therefore, at a low RH, the water concentration difference at a given RH interval is much smaller than at a high RH. On the other hand the hygroscopic behaviour

of the boards induces a swelling of the fibres and thus the voids decrease with increasing MC. Since water transport in the hygroscopic range generally takes place in the gaseous phase in boards with low density, a decrease of the diffusion coefficient with increasing MC is estimated.

3.3 Sorption

Table 4 gives an overview of the parameters describing the sorption isotherms according to the Hailwood-Horrobin and Dent model. On account of the boards’ similar values, classes 3 and 4, 6 and 7 and 8 to 10 were each summarised as a single group.

Figure 7 shows the mean adsorption isotherm of classes 3 plus 4 and 8 to 10 and the mean adsorption and desorption isotherms of classes 6 and 7 according to the single hydrate Hailwood-Horrobin and the Dent model. Thereby, the primary water of the Dent model (M_1) is slightly higher than the hydrated water of the Hailwood-Horrobin model (M_h), and the secondary water of the Dent model (M_2) is slightly lower than the dissolved water of the Hailwood-Horrobin model (M_d). In contrast, the complete hydration (Hailwood-Horrobin) or complete monolayer coverage (Dent) of all available sorption sites (M_0) and the total sorption isotherm (M) are equal in both models (cf. Skaar 1988).

The values of M_0 fall in the range 4.08% to 5.80% for adsorption and 5.03% to 8.80% for desorption (Table 4 and Fig. 7). Thereby, boards with binding fibres (e.g. polyolefin) have lower values than boards with only wood fibres, which have similar values to those calculated by Spalt in Skaar (1988) for spruce wood with 6.47% for adsorption and 8.11% for desorption.

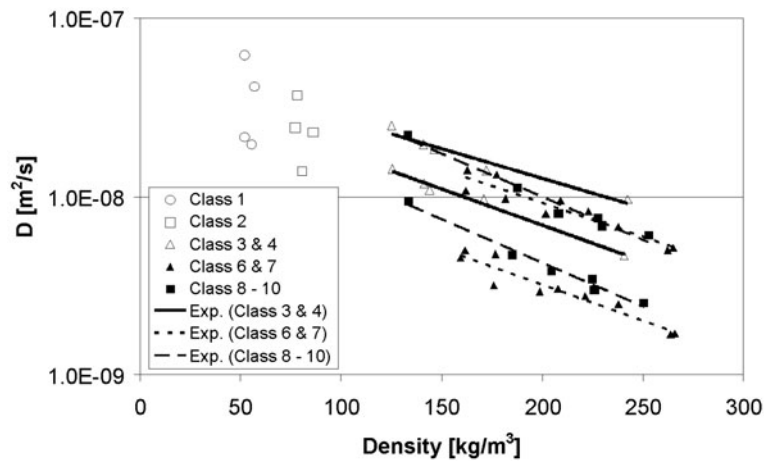


Fig. 6 Diffusion coefficients (D) of the ‘wet cup’ tests (*black symbols*) and the ‘dry cup’ tests (*grey symbols*) depending on the density (mean values). *Empty symbols* represent dry processed and *solid symbols* represent wet processed materials

Abb. 6 Diffusionskoeffizienten (D) im Feuchtbereich (*schwarze Symbole*) und im Trockenbereich (*graue Symbole*) in Abhängigkeit von der Dichte (Mittelwerte). *Leere Symbole* bezeichnen nach dem Trockenverfahren und *ausgefüllte Symbole* nach dem Nassverfahren hergestellte Materialien

Table 4 Parameters (A , B , C) for the calculation of the sorption isotherms according to (5) and the fractions of the different moistures (as MC in percent) at 100% RH according to the Hailwood-Horrobin and the Dent theory. M_0 = Complete hydration (Hailwood-Horrobin) or complete monolayer coverage (Dent) of all available sorption sites; M = Total sorption; M_1 and M_2 = Primary and secondary water (Dent); M_h and M_d = Hydrated and dissolved water (Hailwood-Horrobin)

Tab. 4 Parameter (A , B , C) zur Berechnung der Sorptionsisotherme gemäß Gleichung (5) und die Anteile der verschiedenen Feuchten (in Prozent Holzfeuchte) bei 100% relativer Luftfeuchte nach den Modellen von Hailwood-Horrobin und Dent. M_0 = Vollständige Hydratisierung (Hailwood-Horrobin) bzw. einschichtige Bedeckung (Dent) aller verfügbaren Sorptionsstellen; M = Gesamte Sorption; M_1 und M_2 = Primäres und sekundäres Wasser (Dent); M_h und M_d = Hydratisiertes und gelöstes Wasser (Hailwood-Horrobin)

Type	A	B	C	M_0	M_h	M_d	M_1	M_2	M
Adsorption									
Class 1, type 1 (30 mm)	0.485	19.84	17.51	4.84	4.72	30.8	4.82	30.7	35.5
Class 1, type 2 (60 mm)	0.214	21.22	19.60	4.63	4.58	50.2	4.62	50.1	54.7
Class 2, type 1 (30 mm)	0.388	19.34	16.87	5.00	4.90	29.9	4.98	29.9	34.8
Class 2, type 2 (60 mm)	0.439	23.75	20.78	4.08	4.01	25.4	4.07	25.3	29.4
Classes 3 + 4 (mean)	1.484	15.01	12.09	5.80	5.31	17.4	5.68	17.0	22.7
Class 5, types 1 + 2 (mean)	0.316	23.58	20.93	4.14	4.09	29.7	4.14	29.6	33.8
Class 5, type 3 (160 mm)	0.105	19.67	17.35	5.04	5.01	36.3	5.03	36.2	41.3
Classes 6 + 7 (mean)	1.232	16.29	13.67	5.48	5.11	20.8	5.41	20.5	26.0
Classes 8 – 10 (mean)	1.400	15.37	12.50	5.71	5.27	18.2	5.61	17.8	23.4
Desorption									
Class 1, type 1 (30 mm)	0.751	13.70	11.34	6.71	6.38	25.8	6.65	25.5	32.2
Class 1, type 2 (60 mm)	0.491	15.29	13.78	6.19	6.00	44.0	6.17	43.8	50.0
Class 2, type 1 (30 mm)	0.496	15.17	12.38	6.26	6.07	24.3	6.22	24.1	30.3
Class 2, type 2 (60 mm)	0.497	19.10	15.51	5.03	4.90	19.6	5.00	19.5	24.5
Classes 3 + 4 (mean)	1.689	9.28	6.37	8.80	7.56	14.2	8.32	13.4	21.7
Class 5, types 1 + 2 (mean)	0.635	16.19	13.62	5.81	5.59	25.6	5.77	25.5	31.2
Class 5, type 3 (160 mm)	0.384	14.43	12.34	6.63	6.47	33.9	6.61	33.7	40.3

Figure 8 shows the total adsorption and desorption curves of classes 1, 2 and 5 at each type. Thereby, the values at 100% RH were lower for desorption than for adsorption due to the fact that the measurements begin or end at 93%

and not at 100% RH. Therefore, the true values of the desorption curves are higher in the upper hygroscopic range than the calculated ones. Although boards with binding fibres showed a lower M_0 than the other types with only

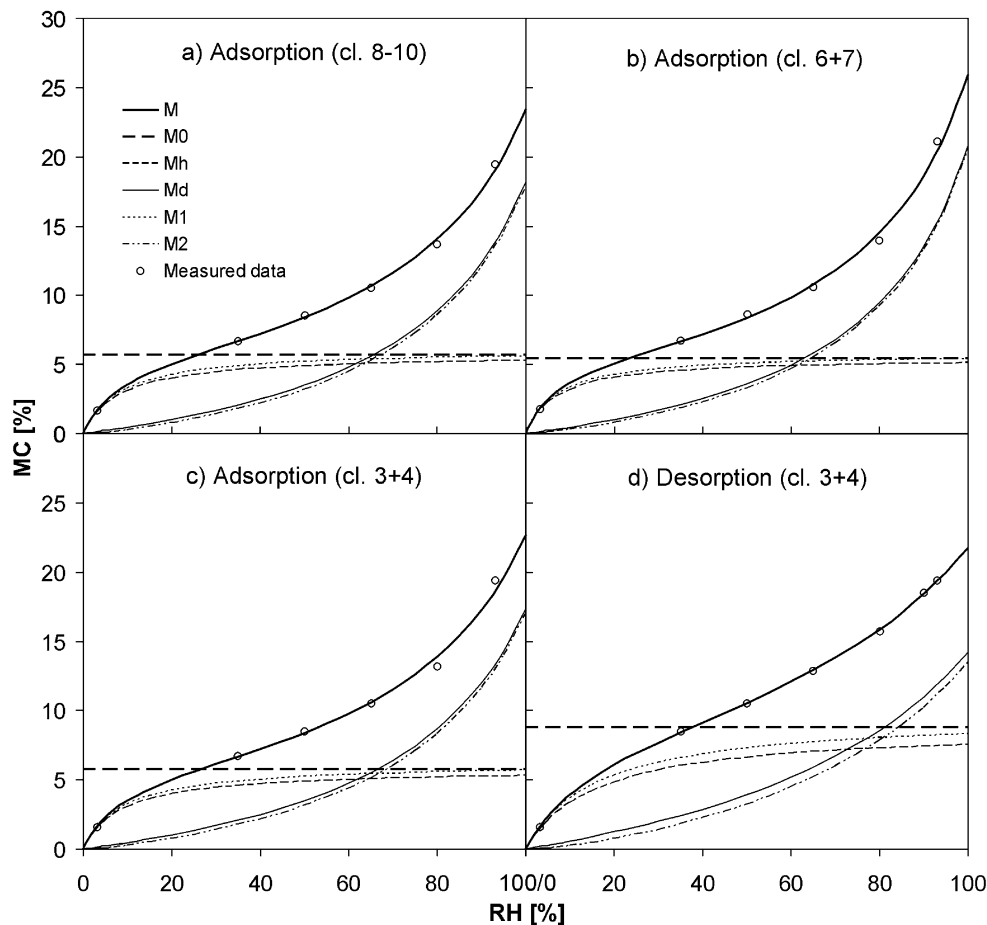


Fig. 7 Mean sorption isotherm of classes 8 to 10 (a), classes 6 plus 7 (b) and classes 3 plus 4 ((c) adsorption, (d) desorption) according to the Hailwood-Horrobin and the Dent model. M = Isotherm of the total sorption; M_0 = Complete hydration (Hailwood-Horrobin) and complete monolayer coverage (Dent) of all available sorption sites; M_h , M_d = Hydrated and dissolved water of the single hydrate Hailwood-Horrobin model; M_1 , M_2 = Primary and secondary water of the Dent model

Abb. 7 Gemittelte Sorptionsisotherme der Klassen 8–10 (a), 6 + 7 (b) und 3 + 4 ((c) Adsorption, (d) Desorption) gemäß dem Hailwood-Horrobin- und dem Dent-Modell. M = Isotherme der gesamten Sorption; M_0 = Vollständige Hydratisierung (Hailwood-Horrobin) bzw. einschichtige Bedeckung (Dent) aller verfügbaren Sorptionsstellen; M_h , M_d = Hydratisiertes und gelöstes Wasser des einfach hydratisierten Hailwood-Horrobin-Modells; M_1 , M_2 = Primäres und sekundäres Wasser des Dent-Modells

wood fibres (and therefore the sorption isotherm is lower in the predominant hygroscopic range as well), the sorption isotherm strongly increases above about 80% RH and exceeds the sorption isotherms of the other boards (cf. also the M -values of Table 4 at 100% RH). This effect is probably due to the addition of additives (ammonium polyphosphate, borate and boric acid) that changed the saturation of water vapour in the air, and not due to the addition of binding fibres (cf. Kollmann and Schneider 1959; Seifert 1972; Lesar et al. 2009).

4 Conclusion

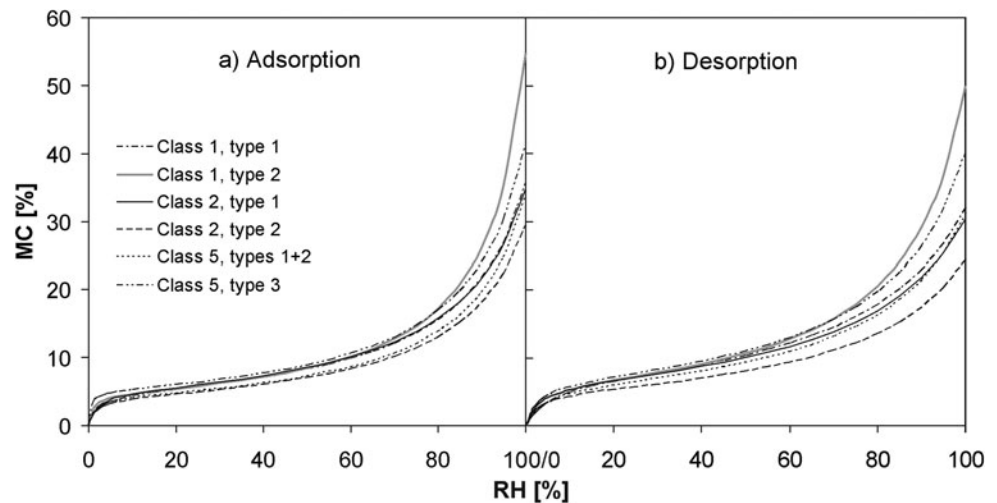
Both, thermal conductivity and water vapour diffusion of the investigated soft fibreboards depend on density but there

are other properties such as temperature, MC and manufacturing method that also have a large impact. The thermal conductivity increases with increasing temperature at about 0.45% per Kelvin and with increasing moisture at about 0.17×10^{-2} W/mK per percent volumetric MC. In contrast, the influence of the dry mass referred MC depends strongly on density but is clearly lower than 1% per change of 1% MC. Furthermore, the thermal conductivity depends on the manufacturing method of the boards and therefore the different orientation of the fibres in the boards.

The water vapour resistance factor is, up to factor 2, higher in the dry climate than in the wet climate. This tendency could be detected in the same order for the diffusion coefficient what primarily can be derived from the sigmoid sorption isotherm. Further, the diffusion coefficient is influ-

Fig. 8 Adsorption (a) and desorption isotherms (b) of the complete water according to the Hailwood-Horrobin and Dent model of the classes 1, 2 and 5 separated per types

Abb. 8 Adsorptions- (a) und Desorptionsisothermen (b) des gesamten Wassers nach dem Hailwood-Horrobin- und Dent-Modell der Klassen 1, 2 und 5 aufgeteilt nach Typen



enced by the manufacturing method resulting in lower values for wet processed boards. For soft fibreboards with binding fibres and additives of ammonium phosphate or sodium borate and boric acid, the sorption isotherm above 80% RH clearly differs from untreated boards due to a greater increase in MC.

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