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Functionalised silicon oxide nanoparticles for fingermark detection

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1 Abstract:

2 Over the past decade, the use of nanotechnology for fingermark detection has been attracting a lot of attention. A substantial number of nanoparticle types has thus been studied and 3 4 applied with varying success. However, despite all efforts, few publications present clear 5 supporting evidence of their superiority over standard and commonly used techniques. This 6 paper focuses on a rarely studied type of nanoparticles that regroups all desired properties for 7 effective fingermark detection: silicon oxide. These nanoparticles offer optical and surface 8 properties that can be tuned to provide optimal detection. This study explores their potential 9 as a new method for fingermark detection. Detection conditions, outer functionalisations and optical properties were optimised and a

Detection conditions, outer functionalisations and optical properties were optimised and a first evaluation of the technique is presented. Dye-doped silicon oxide nanoparticles were assessed against a one-step luminescent cyanoacrylate. Both techniques were compared on natural fingermarks from three donors collected on four different non-porous substrates. On average, the two techniques performed similarly but silicon oxide detected marks with a better homogeneity and was less affected by donor inter-variability. The technique remains to be further optimised and yet silicon oxide nanoparticles already show great promises for effective fingermark detection.

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Keywords: Nanotechnology, luminescence, sensitivity, selectivity, cyanoacrylate, non porous substrates.

22 **1. Introduction**

23 Nanoparticles applied for fingermark detection are known to offer several advantages over 24 traditional methods [1]. First, their small size may lead to detect marks with a high resolution 25 without risking over-development that can typically arise with conventional techniques such 26 as powder dusting or cyanoacrylate fuming. Second, some nanoparticles possess interesting 27 luminescent properties spread over a broad area of the electromagnetic spectrum ranging 28 from UV to infrared. Conventional luminescence properties, as well as up-conversion can be 29 used to mitigate substrate interferences. Infrared luminescence is also of great interest for 30 problematic surfaces since few materials are optically active in this range of wavelengths [2]. 31 Finally, and more importantly, the surface of nanoparticles can be precisely tuned to offer a 32 large panel of potential interaction with fingermarks. By grafting molecules or functional 33 groups onto their surfaces, it becomes possible to specifically target various components of 34 the fingermark residue, leading to an increased selectivity. Fingermarks left by smokers could 35 thus be distinguished from the marks left by non-smokers [3]. Altogether, these properties 36 show great promises for fingermark detection, and can lead to an overall increase of both 37 sensitivity and selectivity.

38 Over the past decade, these advantages attracted a lot of attention and a substantial number of 39 nanoparticle types have been studied and applied with varying success. They can be sorted 40 into three categories, i.e. metal, metal oxide and semi-conductors. Gold and silver 41 nanoparticles have been successfully used in techniques such as multi- or single-metal 42 deposition (gold) [4, 5] or physical developer (silver) [6]. Among metal oxides, we can cite 43 titanium dioxide (TiO₂) [7], aluminium oxide (Al₂O₃) [8, 9] or zinc oxide (ZnO) [10]. Metal 44 oxides are generally applied as dried powders, sometimes functionalised with aliphatic chains 45 to increase and favour hydrophobic interactions with greasy components of the fingermark 46 residue. Quantum dots are the most studied class of semiconductors. They attracted a lot of 47 attention mostly for their uncommon optical properties [11, 12]. This paper does not intend to 48 describe these nanoparticles in extensive details; thorough reviews are available elsewhere 49 [13, 14].

However, from a critical viewpoint, none of the nanoparticles studied until now and their subsequent applications entirely fulfil the criteria described above. Some have a size over 100 nm, and cannot really be classify as nanoparticles according to international [15] and national [16] organisations. Even if not everyone agrees on the 100 nm limit, sub-micron particles would be a more appropriate denomination for particles of a size above 200 nm. A similar comment can be made for nano-sized powder obtained after solvent evaporation. It is not 56 obvious that these nanoparticles are not agglomerated into larger clusters once dried. In that 57 case, properties arising from the nanometric-size such as luminescence may be retained, but 58 detection with a high resolution related with nanometric material will be lost. Moreover, 59 among nanoparticles used up-to-now few possess intrinsic luminescent properties. Most 60 metallic and metal-oxide nanoparticles present weak if no luminescence at all, restricting 61 their applications to light-coloured substrates only. Some attempts have been made to confer 62 luminescent properties to those nanoparticles types. For example, zinc oxide layer can be 63 added around gold nanoparticles to get luminescent fingermark [17]. However, this 64 application remains limited to non-porous substrates and implies a tedious protocol. Another 65 option is to coat nanoparticles with a luminescent dye [8, 9, 18]. Among intrinsic luminescent nanoparticles, quantum dots are the most commonly cited. These semi-conductor 66 67 nanoparticles are luminescent under UV excitation with a narrow emission peak. Their emission colour is directly related to the size of the particles. This particular property is due 68 69 to quantum confinement effect [19, 20] and has driven a lot of endeavour towards their use as 70 a new tool for fingermark detection [21-25, 26]. However, despite all efforts, few 71 publications present clear supporting evidence of their superiority over standard and 72 commonly used techniques. Cost, toxicity and tedious synthetic procedures set aside, the 73 main problem with quantum dots is the difficulty to properly functionalise their surface 74 without altering the structural properties and compromising the luminescence properties. 75 Very few occurrences of successfully and specifically functionalised particles for fingermark 76 detection can be found. Some quantum dots have been functionalised with carboxyl or amine 77 groups [27], but despite the presented results, the real effect of the surface modification on 78 the detection properties remains unclear. This emphasises the fact that even if particles are 79 nano-sized and possess interesting optical properties, their surface still has to offer extensive 80 functionalisation properties in order to be successfully used for fingermark detection.

81 Hence, despite all the nanoparticle types applied for fingermark detection so far, none of 82 them really benefit at the same time from the three advantages presented above (i.e. small 83 size, optical properties and surface modifications). One kind of nanoparticles - yet 84 underrepresented in the literature – could regroup all the properties in one single entity: the 85 silicon oxide nanoparticles (SiO₂). This paper focuses on them and explores how they can 86 offer potential as a method for fingermark detection. Silicon oxide nanoparticles consist in a 87 porous matrix of siloxane bonds, with an external layer of silanol groups that can further react 88 through hydrolysis and condensation with various alkoxysilanes, leading to a functionalised 89 layer covalently bound to the main matrix. As a consequence, SiO₂ nanoparticles could

90 present an almost unlimited range of functional groups. Dye molecules can be entrapped 91 within the siloxane matrix during the synthesis, providing a wide range of optical properties. 92 Various synthetic procedures exist, but two of them are most commonly considered: the 93 Stöber's synthesis and the reversed micro-emulsion. The Stöber's synthesis [28] leads to bulk 94 production, but the size and surface controls are limited. On the other end, reversed micro-95 emulsion [29] enables to accurately control size, optical properties and functionalisation all at 96 the same time, but with a lower yield.

- 97 SiO₂ have been studied and successfully applied for fingermark detection on very rare 98 occasions and with focus mostly put towards optical properties. Theaker et al. investigated 99 the use of hydrophobic micro- and nanoparticles containing sub-particles as well as rhodamine 6G among other dyes [30]. Applied as a dried powder or in suspension, these 100 101 particles were shown to detect fingermarks on several non-porous substrates such as glass or 102 stainless steel. Finely grounded xerogel containing either a europium complex [31] or 103 rhodamine B [32] was used as powder dusting to detect marks. More recently, 700 nm 104 amphiphilic SiO₂ particles were applied on fingermarks on glass microscopic slides but 105 without any particular dyes embedded in their matrix [33].
- In a recent study SiO₂ nanoparticles functionalised with various chemical groups and dispersed in aqueous solutions were used to specifically study the interaction occurring between fingermark residues and nanoparticles [34]. It was demonstrated that chemical interaction between carboxyl and amine groups could be promoted following three different procedures: by lowering the pH of the solution, by adding sodium chloride (NaCl) to decrease zeta potential intensity of the nanoparticles or by adding a diimide compound that activates the amide linkage formation.
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114 The study was focused on understanding the mechanism involved during the detection rather 115 than on the quality of the resulting detected mark. We think that gaining a better 116 understanding of the mechanism is a prerequisite to any work towards optimising the method. 117 This present study aims at exploring the possibilities offered by SiO₂ nanoparticles in terms of a new tool for fingermark detection, by optimising and comparing various detection 118 119 conditions, outer functionalisations and luminescent dyes. More work still needs to be 120 undertaken in order to provide a new fully operational technique, but the results obtained 121 during this study showed that SiO₂ nanoparticles are very promising and that research effort 122 should be further pursued.

124 **2. Materials and methods**

125 <u>2.1 Synthesis and functionalisation of SiO₂ nanoparticles</u>

SiO₂ nanoparticles were synthetised via reversed micro-emulsion where droplets of water containing dye act as microreactors, allowing nanoparticles growth. Even if this method is not the most cost-effective and has a low yield, it allowed synthetising nanoparticles with an acute size control, introducing a dye in the matrix and functionalising nanoparticles surface, all in a single synthetic process. Once nanoparticles were synthesised and functionalised, they were precipitated out of the emulsion, washed and dissolved in water.

- The same synthetic procedure described in Moret et al. was followed without any 132 modification [34]. Three luminescent dyes (rhodamine 6G, rhodamine B and tris(2,2'-133 bipyridyl)dichlororuthenium (II) hexahydrate (RuBpy)) were tested in order to compare 134 optical properties. The dye concentration of the solution added during the syntheses was 135 136 100 mM for both rhodamine 6G and rhodamine B, but for stability reasons it was lowered to 137 16.6 mM for RuBpy, as recommended by Bagwe et al. [35]. For the functionalisation step, two different silane coupling agents were used for comparison purposed (Table 1) 138 (carboxyethylsilanetriol sodium salt (CES) and 3-(triethoxysilyl)-propylsuccinic anhydride 139 (TES-PSA)). They both contained carboxyl groups. 140
- 141

Name	Abbreviation	Molecular structure		
carboxyethylsilanetriol sodium salt	CES	Na ⁺ -0 Na ⁺ -0 Na ⁺ -0 O ⁻ Na ⁺		
3-(triethoxysilyl)-propylsuccinic anhydride	TES-PSA			

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Table 1: Description of the two silane coupling agents used throughout the study.

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2.2 Characterisation of the nanoparticles

Hydrodynamic diameters and zeta potentials of the synthesized SiO₂ nanoparticles were
respectively measured by dynamic light scattering and laser Doppler micro-electrophoresis
using a Zetasizer Nano ZS (Malvern Instrument Ltd). The procedure described in Moret et al.

148 was followed (i.e. hydrodynamic diameter was measured five times and zeta potential was149 measure three time for each samples) [34].

150 An Hitachi F-2500 fluorescence spectrophotometer was used to measure photoluminescence

- 151 properties of the nanoparticles in solution. All measurements were performed at room
- 152 temperature on samples obtained after redispersing the nanoparticles in water.
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154 <u>2.3 Fingermark sampling</u>

155 In order to obtain realistic results, and as recommended in the guidelines published by the 156 International Fingerprint Research Group [36], only natural marks were collected. These 157 marks are said to be natural since no enrichment was made on purpose. The only restriction 158 was that the donors did not wash their hands half an hour prior deposition, but were asked to 159 behave normally otherwise. Three donors (one female, two males) were asked to deposit 160 series of marks on four different non-porous substrates (aluminium foils, black polyethylene 161 (PP), transparent polypropylene (PE) and glass). For the optimisation step, single appositions 162 were collected from one donor on aluminium foils. For the comparison step, sets of 20 163 successive depletive marks were collected for each donor and substrates. The samples were 164 left to age for one week, in an office drawer without specifically controlling the storage 165 conditions. To further test the technique, older marks on aluminium foils, transparent PP and 166 PE were processes as well (aged respectively of 18 months, 2 and 7 years). These marks 167 came from previous researches for which donors' identity, deposition and storage conditions 168 were unknown.

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170 <u>2.4 Fingermark detection</u>

171 Various parameters were successively assessed during this study. Three detection conditions
172 were evaluated and compared, as well as two functional groups containing carboxyl functions
173 and three luminescent dyes (Table 2).

Regardless of the chosen parameters, the overall detection protocol consists in a two-step immersion procedure. The items are first immersed in a bath containing the nanoparticles solution 60 min (first two detection condition) or 30 min (third detection condition), then rinsed with water. The rinsing step is necessary to remove any unwanted nanoparticles remaining on the substrate, which could lead to background noise. Samples are then left to air-dry.

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Successive optimisations	Tested parameter				
1. Detection	pH 3 pH 6 Nat		Cl 0.5 M	EDC/NHS	
conditions	immersed 60 min	immersed 60 min		immersed 30 min	
2. Functionalisation	CES		TES-PSA		
3. Luminescent dye	rhodamine 6G	rhodamine B		RuBpy	

Table 2: Summary of the tested detection conditions and silicon oxide nanoparticles properties.

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186 <u>2.5 Results comparison and evaluation</u>

187 In order to determine if research on SiO₂ nanoparticles application was worth pursuing, the results were compared to a commonly used technique: a one-step luminescent cyanoacrylate 188 the LumicyanoTM [37]. Each fingermark was cut in half, one half being processed with the 189 nanoparticles solution and the other half fumed with LumicyanoTM (following the 190 manufacturer's instructions). For each fuming cycle, 1 g of LumicyanoTM was placed in 191 192 MVC1000 fuming cabinet (Foster & Freeman), at 80% of relative humidity. The samples 193 were exposed to cyanoacrylate fumes until the detection was considered as optimal (about 15 194 minutes). Since cyanoacrylate fuming results are generally dependent of the substrate type, 195 each surface was processed in a separate cycle.

196 Corresponding halves were then paired again before being photographed in luminescence 197 mode (Note: the luminescence of LumicyanoTM decreasing with time, the pictures were taken 198 the day of the fuming process). The imaging conditions were identical for LumicyanoTM and 199 SiO₂ nanoparticles, with an excitation at 495 nm (Minicrimescope MCS400) and an 200 observation at 590 nm (interferential filter with a band pass of 37 nm).

Results were assessed following the procedure described by Moret and Bécue [5]. Three independent evaluators were presented one half mark at a time, in a random order. Each half mark was thus assessed independently from its corresponding half. The average scores obtained from each technique and substrate were then compared.

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206 **3. Results and discussion**

- 207 <u>3.1 SiO₂ nanoparticles synthesis and characterisation</u>
- 208 Reversed micro-emulsion allowed obtaining stable solutions. DLS measurements gave an
- 209 average hydrodynamic diameter of 84.2 nm and a zeta potential of –36.7 mV at pH 6. These
- 210 values were consistent with those obtained previously [34].
- 211 Inclusion of dye molecules within the nanoparticles matrix helped obtaining various optical
- 212 properties (Figure 1). Rhodamine 6G, rhodamine B and RuBpy were selected because they

213 covered three distinct areas of the spectrum. They also have been used in previous studies







Figure 1: Excitation and emission spectra of silicon oxide nanoparticles solutions synthesised with three different luminescent dyes (RuBpy, rhodamine 6G and rhodamine B)

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219 <u>3.2 Detection mode</u>

Several parameters were compared to obtain better detection results. Three application modes were previously determined to detect mark with success: adjusting the pH of the nanoparticles solution to 3, adding sodium chloride (NaCl) to lower the zeta potential intensity and adding a diimide compound to activate the carboxyl groups and favour the interaction with the amine groups of the secretions [34].

These three detection modes were compared using the same nanoparticles batch containing rhodamine 6G and functionalised with CES on fingermarks deposited on aluminium foils (Figure 2). Luminescent results were successfully obtained in each case. It appeared however that the best results in terms of ridge clarity and luminescence intensity were obtained when NaCl was added in the solution to a concentration of 0.5 M. pH reduction and diimide addition (EDC/NHS) led to slightly inferior results. Addition of NaCl was consequently chosen as the best detection protocol and considered for the rest of the study.





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Figure 2: Comparison between three application modes of Rhodamine 6G-doped SiO₂ nanoparticles solution (adding EDC/NHS, lowering the pH to 3 and adding NaCl to a concentration of 0.5 M). Marks are one-week old, deposited on aluminium foil and visualised in luminescent mode (excitation at 495 nm and emission at 590 nm).

239 <u>3.3 Surface functionalisation</u>

240 The second optimisation consisted in selecting the most appropriate function containing 241 carboxyl group that can interact with fingermark residue. Two alkoxysilanes were grafted on 242 the surface of nanoparticles: CES and TES-PSA (Table 1). They both contain carboxyl group, 243 but the second one doubles the amount of functional groups present on the nanoparticle 244 surface, due to the opening of the succinic chain in water. Since carboxyl groups drive the 245 nanoparticles interaction with secretions, TES-PSA could lead to a better detection due to the larger amount of reactive groups. Comparisons were made between two solutions containing 246 247 nanoparticles functionalised respectively with CES and TES-PSA. As determined previously, 248 NaCl (0.5 M) was added to favour the interaction. A significantly higher background noise 249 appeared with TES-PSA functionalised nanoparticles and the obtained marks appeared 250 blurred (Figure 3). With CES, an obvious quality difference was observed. Sharp luminescent 251 ridges were obtained, with clear level 3 features. Since CES functionalised nanoparticles gave 252 the best results, they were selected for this study.



Figure 3: Comparison between Rhodamine 6G-doped silicon oxide nanoparticles
functionalised with TES-PSA ((triethoxysilyl)-propylsuccinic anhydride – left half) and CES
(carboxyethylsilanetriol – right half), and used to detect fingermarks. The illustrated
fingermark is one-week old, deposited on aluminium foil and visualised in luminescent mode
(excitation at 495 nm and emission at 590 nm).

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261 <u>3.4 Luminescence properties</u>

262 The last optimisation step was about the nanoparticles' optical properties. As described 263 above, three dyes were considered: rhodamine 6G used as a reference, rhodamine B and 264 RuBpy. Each half mark was observed under the luminescent conditions that lead to the best result. RuBpy was illuminated at 495 nm, whereas both rhodamine 6G and rhodamine B were 265 266 excited at 515 nm. All marks were observed at 590 nm. These conditions differed from the optimal conditions determined by spectrofluorimetry. If the theoretical conditions were to be 267 268 respected, the small Stokes shift of rhodamine 6G and B would lead to a background noise 269 due to the excitation wavelength. Based on the spectrofluorometry results (Figure 1), RuBpy 270 appeared to be the best choice since it has the larger Stokes shift (100 nm) compared to the 271 other two dyes (~20nm). These conditions were the one resulting in the best luminescence 272 intensity with the lowest background. Rhodamine 6G and rhodamine B showed equivalent 273 results in terms of luminescence intensity (Figure 4). RuBpy presented a much more intense 274 luminescence, compared to the two other dyes. It led to a light background staining, but since 275 RuBpy had a larger Stokes shift and the intensity of the final results was stronger, it was 276 chosen as the most suitable dye for this study. The observed background noise obtained after immersing the samples was not judged detrimental to fingermark visualisation. 277





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Figure 4: Comparison between SiO₂ nanoparticles synthetised with three different luminescent dyes (rhodamine 6G, rhodamine B and RuBpy). Marks are one-week old, deposited on aluminium foil and visualised in luminescent mode (excitation at 495 nm and emission at 590 nm).

These optimisation experiments led to select the parameters leading to the best detection quality. Optimal detection was thus obtained with silicon oxide nanoparticles containing RuBpy, functionalised with CES and applied in a solution at pH 6 containing 0.5 M of sodium chloride for 60 min. Should this technique be applied on a larger scale, further improvement would be needed, in order to reduce the immersion time for example. However, results obtained so far constitute a valid base to conduct a comparison with a benchmark technique.

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293 <u>3.5 Comparison with cyanoacrylate fuming</u>

294 Conventional cyanoacrylate fuming is one of the most commonly used techniques for the 295 detection of fingermarks on non-porous substrates. However, it cannot be directly compared 296 to SiO₂ nanoparticles since the obtained results are not luminescent. In order to avoid 297 overestimating results and introducing a bias toward one technique or the other, a 298 luminescent technique should therefore be used to offer a valid comparison. Luminescent 299 marks can be obtained using conventional cyanoacrylate by following a two-step procedure 300 consisting in fumigating the samples first, then dye-staining them with a luminescent dye 301 dissolved in a solvent [38]. However, to avoid tedious staining procedures and to obtain more 302 homogeneous results throughout the sample sets, a one-step luminescent cyanoacrylate was privileged during this study: the LumicyanoTM [37]. The obtained marks offer luminescent 303

properties either under UV (326 nm) or at 511 nm and the results can be observed with an emission band centred at 562 nm. It however appeared that good luminescent marks could be observed under the same conditions used for RuBpy (i.e. excitation at 495 nm and emission filter at 590 nm). Using the same visualisation conditions help to image both sides at the same time and to keep constant parameters for comparison.

Both techniques were applied on sets of 20 depletive marks from 3 donors on 4 different substrates (a total of 240 marks) as described in the materials and methods section. The samples were processed and imaged the same day in order to avoid any age inconsistency, or luminescence intensity loss. Both techniques have been applied according to the established procedure and led to satisfactory results. Under white light, the results obtained with LumicyanoTM were similar to those that can be obtained with common cyanoacrylate.

Before discussing the sensibility obtained on depletive marks, the results obtained on the first 315 316 finger apposition are presented (Figure 5). Since results quality is related to the nature of the 317 substrate, they will be described separately. Aluminium foil is generally considered as an 318 ideal surface allowing good results to be obtained. Luminescent results were obtained for both techniques, with a stronger intensity for LumicyanoTM (Figure 5 – first column). 319 However, when ridge details were considered, it can be seen that cyanoacrylate-fumed marks 320 321 tended to give dotted marks with ridge-continuity disruption. This was especially the case for 322 donors A and B, but not for donor C for which very homogenous results were obtained. This phenomenon was however not observed for the fingermarks detected by SiO₂ nanoparticles. 323 324 Even if marks from donor B were not fully detected, nanoparticles developed marks with a 325 higher homogeneity for the three donors. An affinity of nanoparticles for the aluminium foil 326 was observed, leading to a slight background staining. This phenomenon, already noticed 327 during the optimisation phase, did not prevent fingermarks observation. The background was 328 not due to a lack of rinsing since it appears uniformly on the entire surface. It is most likely 329 due to unwanted interactions with the substrate. This issue could potentially be reduced by a 330 shorter immersion time in the nanoparticles solutions, and should be further studied. By 331 comparison, no background apparition was noticed with cyanoacrylate fuming.

Regarding black PE, the previous observations were still valid (Figure 5 – second column). Cyanoacrylate luminescence was less intense and led again to doted marks. For donor B, the marks were barely visible. The background noise observed with SiO_2 on aluminium foil is much more pronounced on PE. Sometimes it concealed the marks almost entirely (donors A and B). It was however not homogeneous and when absent it became possible to observe luminescent marks of high quality with clear level 3 features (donor C). Nanoparticles were thus effectively attracted on the secretions, but the ridges were partially covered by a nonselective deposition onto the surface. That reinforces the needs of further studies on the interaction with substrates.

Transparent PP provided significantly different results compared to the two previous substrates (Figure 5 – third column). For the cyanoacrylate, luminescent marks were obtained for both donor A and C, but the detection was not homogeneous towards the edge of the donor A's mark and almost no ridges were visible for the donor B. The situation was quite different for the SiO₂ nanoparticles; the detection itself was very homogeneous for all three donors in terms of mark quality, with no background staining observed.

On glass, the nanoparticles solution has largely washed away the marks (Figure 5 – second last). Only a weak luminescence and low quality marks were observed. On the other halves, cyanoacrylate performed well for the three donors, even if a light background staining appeared. Considering SiO_2 nanoparticles, the detection protocol seems not appropriate for this type of substrate and should thus be further adjusted. Reduction of immersion time or finding another carrier solvent could be key factors that could lead to better results.

353 In summary, for the first marks of the depletion series, cyanoacrylate succeeded in detecting 354 marks on all tested surfaces. However, differences in quality were consistently observed 355 between donors. These variations can be attributed to the quantity and quality variation 356 among donors' secretions. However, this trend was less pronounced with SiO₂ nanoparticles 357 that react not with specific compounds of the secretion, but with specific functional groups. The results quality was however more tightly related to the substrate type, which indicates 358 359 that the detection conditions can be further adjusted. Glass appeared to be a challenging 360 surface since nanoparticles solution tended to wash the secretion, and thus failing in detecting 361 any mark. Moreover, background staining appeared on black PE, impinging the detection of 362 the marks. Future optimisations need to be carried on, especially to decrease the unwanted 363 background staining and to obtain a more versatile technique that could be applied on a wider 364 range of substrates. SiO₂ nanoparticles application remains promising since the obtained 365 results are very homogeneous and appeared to be less donor-dependent. This, in itself, 366 represents a considerable advantage over cyanoacrylate fuming.





Figure 5: Comparison of fingermarks detected with RuBpy-doped silicon oxide nanoparticles
 (left halves) and LumicyanoTM (right halves) from three donors on four substrates.

Results obtained with the depletive series are presented below. On average, by taking into account every mark from each donor and substrate, comparable scores were obtained for both techniques (1.3 for LumicyanoTM and 1.5 for SiO₂ nanoparticles) (Figure 6). When only the first five depletions were considered, a general increase of scores was noticed (1.8 for LumicyanoTM and 2.1 for SiO₂ nanoparticles), showing that nanoparticles gave slightly better results on average (Figure 6).



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Figure 6: Comparison of the average results obtained with LumicyanoTM (CA) and RuBpy doped SiO₂ nanoparticles.

383 By looking at each substrate separately and considering only the first five depletions, results 384 can be described in more details (Figure 7). On aluminium foil, equivalent results were 385 obtained showing that both techniques performed well, but on glass, as described above, SiO₂ 386 solution tends to wash the marks, leading to low quality results, cyanoacrylate fuming shows 387 superior scores. For the black PE, even if background staining was observed, the average 388 quality was still superior for SiO₂ nanoparticles compared to cyanoacrylate. The biggest 389 difference appeared on transparent PP where cyanoacrylate stopped being effective after just 390 a few depletions, while SiO₂ nanoparticles kept detecting marks. This was not only true for 391 the first five depletions, good quality results were obtained until the twentieth depletion 392 (Figure 8).





Figure 7: Comparison of the average results obtained with LumicyanoTM (CA) and RuBpy doped SiO₂ nanoparticles on four different substrates.









401 When average scores were calculated considering each donor individually, SiO_2 402 nanoparticles were less donor dependent that cyanoacrylate (Figure 9). The calculated 403 variance between donors for cyanoacrylate is 0.41, whereas the one for SiO_2 nanoparticles is 404 only 0.02.

405 To compare the efficiency of the two techniques over the entire depletion series, marks of a 406 score above or equal to 2 were added (Figure 10). Cyanoacrylate on aluminium detected the 407 greatest amount of marks. SiO_2 nanoparticles solution, as detailed above, was not efficient on 408 glass, but detected a comparable amount of marks on the three other substrates.









Figure 10: Comparison between the amount of mark with a score equal or above 2 detected
 per substrate for each technique.

To conclude the comparisons, aged marks on aluminium foils, transparent PP and PE were process following the same established protocol. Both cyanoacrylate fuming and SiO₂ nanoparticles solution were able to detect marks as old as 7 years. On aluminium foils, SiO₂ nanoparticles tended to give better ridge details. The reversed situation was observed on transparent PP. For the 7 years marks, LumicyanoTM was superior on transparent PP, with much better ridge quality. On PE, more completed marks were detected with SiO₂ nanoparticles, despite missing information in the centre, whereas cyanoacrylate presented good details on the upper part and very few details on the lower part.



Figure 11: Comparison of old fingermarks (18 months to 7 years) detected with RuBpy doped silicon oxide nanoparticles (left halves) and LumicyanoTM (right halves) on three
 substrates.

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433 **4.** Conclusions

This paper investigated an application of luminescent functionalised SiO₂ nanoparticles for 434 435 fingermark detection. Several optimisations of the detection protocol for considering dyedoped SiO₂ nanoparticles as efficient fingermark reagents were presented. Appropriate 436 detection conditions, outer functionalisations and luminescent dyes were successively studied 437 and selected, leading to an optimised protocol. A comparison to LumicvanoTM, a one-step 438 luminescent cyanoacrylate (CA), was then conducted to assess the efficiency of the method. 439 440 Both techniques were compared on series of 20 depletive marks from three donors on four 441 different non-porous substrates.

442 On average, both techniques performed similarly. However, for LumicyanoTM, quality 443 variations between donors were observed. Interestingly, that was not observed (or only to a 444 limited extend) for SiO₂ nanoparticles. This might be due to the fact that the technique targets 445 a specific functional group instead of particular compounds of the secretion. The SiO₂ results 446 quality was instead more related to the substrate type.

447 Heterogeneous detection results proved that the technique remained to be further optimised.

448 More work still needs to be undertaken in order to provide a fully operational technique, but

the reported results in this study showed that SiO_2 nanoparticles are very promising and that

450 research effort should be further pursued.

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Wavelength [nm]

Rhodamine B - em



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Figure 4 Click here to download high resolution image





Figure 6 Click here to download high resolution image



Figure 7 Click here to download high resolution image











Figure 11 Click here to download high resolution image