A centrifugal microfluidic platform for Chromium speciation on a lab-on-a-disc.

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The quantitative analysis of chromium species is referred to as chromium speciation. While chromium exists in oxidation states from 0 to VI, it is predominantly found in the (III) and (VI) states [1]. Industry effluent (e.g. textile/electroplating) is a common source of chromium pollution in the environment. Due to corrosion inhibitors used in pipes, and contamination leaching from sanitary landfills, drinking water supplies can become contaminated also [2]. The bioavailability and toxicity of chromium is largely dependent the oxidation state of the element [2]. Consumption of Cr (III) is an essential component in human diet, as it is responsible for maintaining glucose, lipid and protein metabolism [3]. In contrast, Cr (VI) is strongly oxidizing, exhibiting high toxicity, with carcinogenic and mutagenic properties [4]. It is recommended by the World Health Organisation (WHO) that the maximum allowable concentration of chromium (VI) in drinking water is 0.05 mg L⁻¹ [5].

Herein, describes the development of a centrifugal-based microfluidic device (CMD) optimised for the quantitative analysis of both chromium (III) and (VI) species in water. The CMD platform is inclusive of a centrifugally-driven lab-on-a-disc (LOAD) cartridge with a complementary 3d printed colourimetric sensor system with heating and results read-out capabilities. For analysis, the sample is loaded into a reservoir on the disposable microfluidic disc, along with reagents. Due to the low viscosity of the reagents, a microfluidic 'shield-shaped' reservoir was designed for the prevention of premature liquid flow. A centrifugal force is then obtained, through the spinning of the disc, for liquid pumping through microchannels, causing them to mix and react to form a coloured product into the optical-pathway detection zone, as previously reported [6]. In this case however, as two analytes were being assessed, a preferred path length of 50mm was selected to allow both samples to be run in duplicate. The incorporated optical detection system consisted of a light emitting diode (LED) and photodiode (PD) couple. Chromium (III) and (VI) were measured using 2,6-pyridine dicarboxylic acid and 1,5-diphenyl carbazide (DPC) respectively, forming complexes that were measured at 535 nm. The LOD for trivalent and hexavalent chromium using this device were 21 mg L⁻¹ and 4 µg L⁻¹, respectively. The linear range for quantitative analysis was found to be 69–1000 mg L⁻¹ for Cr(III) and 14–1000 µg L⁻¹ for Cr (VI). The device is simple to use, low in cost solution which could provide rapid on-site measurements comparable with standard benchtop spectrophotometers.

References:

- [1] F. A. Byrdy, L. K. Olson, N. P. Vela, and J. A. Caruso, "Chromium speciation by anion-exchange high-performance liquid chromatography with both inductively coupled plasma atomic emission spectroscopic and inductively coupled plasma mass spectrometric detection," *J. Chromatogr. A*, vol. 712, no. 2, pp. 311–320, 1995.
- [2] V. Gómez and M. P. Callao, "Chromium determination and speciation since 2000," *TrAC Trends Anal. Chem.*, vol. 25, no. 10, pp. 1006–1015, 2006.
- [3] R. Świetlik, "Speciation analysis of chromium in waters," *Polish J. Environ. Stud.*, vol. 7, no. 5, pp. 257–266, 1998.
- [4] K.-C. Hsu, C.-C. Sun, Y.-C. Ling, S.-J. Jiang, and Y.-L. Huang, "An on-line microfluidic device coupled with inductively coupled plasma mass spectrometry for chromium speciation," *J. Anal. At. Spectrom.*, vol. 28, no. 8, pp. 1320–1326, 2013.
- [5] F. Edition, "Guidelines for drinking-water quality," *WHO Chron.*, vol. 38, pp. 104–108, 2011.
- [6] G. Duffy, I. Maguire, B. Heery, C. Nwankire, J. Ducrée, and F. Regan, "PhosphaSense: A fully integrated, portable lab-on-a-disc device for phosphate determination in water," *Sensors Actuators B Chem.*, vol. 246, pp. 1085–1091, 2017.