



## Chemistry and insecticidal potential of bay leaf essential oil against stored grain pest of wheat

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**Abstract:** A laboratory experiment was conducted to study chemistry and insecticidal activity of bay leaf oil, its fractions and isolated compounds against stored grain pest of wheat i.e. *Tribolium castaneum* Herbst. Bay leaf essential oil extracted from dried and powdered bay leaves was subjected to column chromatography to have its fractions. Extensive column chromatography of polar fraction yielded Eugenol and 7, 7 Dimethyl-3-methylene bicyclo [2.2.1] heptan-4-ol which were identified by spectroscopic techniques. Bay leaf oil was tested for its insecticidal activity at five different concentrations in the range 4-12 mg g<sup>-1</sup> respectively against F<sub>1</sub> generation of red rust flour beetle adults. Maximum inhibition was observed at 12 mg g<sup>-1</sup> concentration. The activity was both time and concentration dependent. The fractions of bay leaf essential oil and the compounds isolated were tested at 4mg g<sup>-1</sup> concentration. Polar fraction was found to be more active as compared to nonpolar fraction as 100 and 53.1% mortality was obtained on 30<sup>th</sup> day for polar and nonpolar fractions, respectively. Comparison of eugenol and 7, 7 Dimethyl-3-methylene bicyclo [2.2.1] heptan-4-ol showed complete mortality on 33<sup>rd</sup> and 35<sup>th</sup> day respectively, which revealed that adults of *T. castaneum* were more susceptible to eugenol. The results indicated that bay leaf essential oil may have potential to control stored grain pest, *T. castaneum*.

**Keywords:** Essential oil, *Laurus nobilis*, Stored grain pest, *Tribolium castaneum*

### INTRODUCTION

Wheat is major food crop grown in Punjab. Substantial amount of wheat is stored in godowns by farmers, flourmills and government procurement agencies. Beetles are of great importance in grain storage systems as their activity on stored produce leads to losses both in quality and quantity. In India losses caused by deterioration under adverse storage conditions account for 6.5 percent of stored grains (Raju, 1984) and it ranged between 10-40% worldwide (Raja *et al.*, 2001). Indiscriminate use of large quantities of toxic and persistent pesticides (Agnihotri, 2000) can result in many problems in the biosphere leading to imbalance, damage to non-target organisms (Halder *et al.*, 2010). The chemical residues in grains pose health hazards to the consumers as most of our flourmills grind wheat without washing. In India, cereal consumption is quite high as compared to western countries and therefore even small amount of insecticide residues left on grains will result in a large intake. It is therefore imperative to find out non-insecticide control measures. One such possibility is the use of essential oils as pest control measure as they are non-toxic, non-hazardous for mammals as well as environment (Credland, 1992; Zhang *et al.*, 2000; Adedire and Akinkulore, 2005) and at the same time have sufficient persistence for the control of pests. Essential oils being volatile are known

to be effective against fungi (Stevic *et al.*, 2014; Soylu *et al.*, 2005), bacteria (Prabuseenivasan *et al.*, 2006), virus (Patel *et al.*, 2000) and insects (Geetha and Roy, 2014; Ayvaz *et al.*, 2010). Dried bay leaf (*L. nobilis*), a spice used in traditional culinary practices as a flavouring agent could be used as botanical biopesticide in postharvest crop protection (Kivçak and Mert, 2002). In this context, Papachristos and Stamopoulos (2002) reported that bay leaf essential oil has a repellent action against the bean weevil *Acanthoscelides obtectus* (Say). Bay leaf essential oil obtained from Tunisia, Algeria and Morocco were reported to possess repellent and fumigant toxicity against stored grain pests (Jemma *et al.*, 2012). Correspondingly, Cosimi *et al.* (2009) reported that the essential oil was repellent to *Sitophilus zeamais* (Motschulsky), *Cryptolestes ferrugineus* (Stephens) and *Tenebrio molitor* (L.). Andronikashvili and Reichmuth (2002) and Rozman *et al.* (2007) reported the fumigant activity of naturally occurring compounds of this essential oil and some other plants against three stored product beetles, *T. castaneum*, *S. oryzae* (L.) and *R. dominica* (F.). Besides various fumigant and repellent activities of bay leaf essential oil against stored grain pest (Cosimi *et al.*, 2009; Lee *et al.*, 2003), few reports are available on the insecticidal potential of bay leaf oil and its fractions against *T. castaneum*. So there exists tremendous scope for exploiting the activity of bay leaf essential

oil against stored grain pest of wheat. In the present study, bay leaf oil, its fractions were evaluated against stored grain pest of wheat *T. castaneum* under laboratory conditions. Moreover the isolation of pure compounds from the most bioactive fraction and their activity against *T. castaneum* was also undertaken.

## MATERIALS AND METHODS

Bay leaves were purchased from local market. The shade dried leaves were powdered and used for extraction of essential oil. The powdered leaves were taken in one litre flat-bottomed flask and 400 ml of distilled water was added to it. The contents were thoroughly mixed and flask was kept overnight. The flask was placed on hot plate and fitted with reflux condenser attached to Dean and Stark water apparatus. The contents were refluxed on a hot plate at 100°C for about five hours. The oil extracted using hydrodistillation method. The essential oil layer containing little water was collected in a conical flask. The same procedure was repeated to process 13 batches. The essential oil layers were pooled in a conical flask and partitioned using diethyl ether (3×100 ml). The ether layer was dried over sodium sulfate and excess solvent was evaporated using vacuum and finally oil was stored in a stoppered tube. The bay leaf essential oil obtained was pale yellow liquid with sweet spicy clove like odour having pH and refractive index of 6.9 and 1.51 respectively. The essential oil was insoluble in water and soluble in alcohol, diethyl ether and glacial acetic acid.

Bay leaf essential oil (12.0 g) was subjected to column chromatography to fractionate it into hexane and dichloromethane to have non-polar and polar fractions respectively using silica gel (60-120 mesh) as an adsorbent. The polar fraction (4.0 g) of essential oil was chromatographed over a silica gel by gradient dilution with hexane, hexane-dichloromethane and then pure dichloromethane (Table 1). Fractions (100 ml) were collected and concentrated over water bath, and similar fractions were combined according to thin layer chromatographic profiling. Two compounds eugenol and 7, 7 dimethyl-3methylene bicyclo [2.2.1] heptan-4-ol were isolated. The structural elucidation based on IR, <sup>1</sup>HNMR and <sup>13</sup>CNMR was carried out (Table 2). FT-IR spectra were measured in CHCl<sub>3</sub> solution or nujol on a Perkin Elmer, Model RX-1 FT-IR spectrophotometer. <sup>1</sup>HNMR and <sup>13</sup>CNMR spectra were recorded with Bruker AC (400 MHz) or mentioned otherwise as solutions (in CDCl<sub>3</sub>) using TMS as an internal reference. FTIR, <sup>13</sup>CNMR and <sup>1</sup>HNMR spectroscopic analysis was taken at Central Instrumentation Laboratories (CIL), Panjab University, Chandigarh. The chemical shifts are expressed in δ (ppm) values and the abbreviations, d, t and m stand for singlet, doublet, triplet and multiplet respectively.

**Insecticidal activity:** The wheat grains were sterilized at 60±5°C for 8 hours in order to eliminate both appar-

ent and hidden infestation of insects if any. These grains were conditioned at least for a week in an environmental chamber maintained at 30±2°C and 75 per cent relative humidity to raise their moisture content. The red flour beetle (*T. castaneum*) was obtained from laboratory cultures maintained in the Department of Entomology, PAU Ludhiana, maintained at 28±1°C and 70% relative humidity. The insects were reared in glass containers containing wheat flour and yeast (10:1, w/w). From these jars adults of known age (1-2 weeks) i.e. F<sub>1</sub> generation was obtained for the experimental purpose. The insecticidal activity of bay leaf essential oil against *T. castaneum* adults was tested at five different concentrations ranging from 4-12 mg g<sup>-1</sup> (Table 3). Stock solutions were prepared. For the experiment, wheat (20.0g) was spiked with different concentrations of the essential oil. The observations of mortality were taken after every 24 hrs till complete or constant mortality was obtained. The percent mortality was calculated using Abbott's formula (Abbott, 1925). The statistical analysis was carried out to determine LC<sub>50</sub> values of the bay leaf oil against *T. castaneum* adults (Table 4). Bay leaf essential oil was fractioned into non-polar and polar fractions respectively for insecticidal activity by column chromatography over silica gel (60-120 mesh size). The column chromatography of polar fraction of bay leaf essential gave two compounds eugenol (1) and 7, 7 dimethyl-3 methylene bicyclo [2.2.1] heptan-4-ol (2) which were characterized by spectral techniques and used for insecticidal activity at 4 mg g<sup>-1</sup>(Table 2). There were three replications for each treatment and for control only wheat and acetone were used.

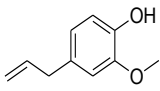
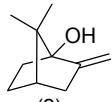
## RESULTS AND DISCUSSION

Essential oil of was extracted from dried and powdered bay leaves by hydrodistillation method in 3.0 percent yield which was more as compared to the yields of hydrodistilled bay leaf oil obtained from Tunisia, Algeria and Morocco showing yields of 0.584%, 0.46% and 0.655% respectively (Jemma, *et al.*, 2012) The bay leaf essential oil was pale yellow liquid having clove oil like odour with pH and refractive index of 6.9 and 1.51 respectively. It was insoluble in water and soluble in alcohol, diethyl ether and acetic acid. Thin

**Table 1.** Column chromatography of the polar fraction to separate the mixture of compounds.

S.N.	Solvent used	Weight (g)	TLC based remarks
1.	Hexane (5×100 ml)	Traces	-
2.	Hexane: Dichloromethane 5% (10×100 ml)	3.4	Pure compound(1), liquid
3.	Hexane: Dichloromethane 5% (10 × 100ml)	0.007	Mixture
4.	Hexane: Dichloromethane 5% (8×100 ml)	1.2	Pure compound (2), solid

**Table 2.** Spectral data of the isolated compounds.

Compound	IR (cm <sup>-1</sup> )	<sup>1</sup> HNMR (δ ppm)	<sup>13</sup> CNMR (δ ppm)
 (1)	3514, 3076, 1638, 1610, 1149, 1112, 1034, 995, 794, 746.	3.19 and 3.21 (d, 2H, J=6.7 Hz, C <sub>7</sub> -Hs), 3.7 (s, 3H, C <sub>10</sub> -Hs), 4.95 (m, 2H, C <sub>9</sub> -Hs), 5.57 (s, 1H, OH), 5.84 (m, 1H, C <sub>8</sub> -H), 6.56 (m, 2H, C <sub>5</sub> and C <sub>6</sub> -Hs), 6.73 (d, 1H, J=3.6 Hz, C <sub>3</sub> -H)	39.83(C <sub>7</sub> ), 5.56(C <sub>10</sub> ), 111.10(C <sub>3</sub> ), 114.29(C <sub>6</sub> ), 115.45(C <sub>9</sub> ), 121.11(C <sub>5</sub> ), 131.85(C <sub>4</sub> ), 137.81(C <sub>8</sub> ), 143.83(C <sub>1</sub> ), 146.43(C <sub>2</sub> )
 (2)	3376, 3081, 1635, 1455, 1375, 888.	0.97 (d, 6H, J=6.2 Hz, C <sub>7</sub> and C <sub>9</sub> -Hs), 4.61 (d, 2H, J=10.2 Hz, C <sub>10</sub> -Hs)	24.8 (C <sub>6</sub> ), 26.1 (C <sub>8</sub> ) <sup>a</sup> , 26.7 (C <sub>9</sub> ) <sup>b</sup> , 29.9 (C <sub>2</sub> ), 38.9 (C <sub>5</sub> ), 41.8 (C <sub>1</sub> ), 53.4 (C <sub>7</sub> ), 82.6 (C <sub>4</sub> ), 106.3 (C <sub>10</sub> ), 153.5 (C <sub>3</sub> ), (a) and (b) are interchangeable.

**Table 3.** Corrected percent mortality of *T. castaneum* adults by bay leaf oil at indicated time intervals.

No. of Days	Concentration (mg g <sup>-1</sup> )				
	12	10	8	6	4
1	83.3	73.3	61.6	60.3	53.3
2	88.3	76.6	73.3	73.3	60.0
4	95.0	81.6	76.6	76.6	65.0
5	100	81.6	77.5	77.5	65.0
7	-	98.0	83.9	79.2	69.7
8	-	100	83.9	79.2	69.7
10	-	-	88.2	82.0	71.9
11	-	-	100	88.6	71.9
15	-	-	-	88.6	75.0
27	-	-	-	91.6	81.2
33	-	-	-	100	87.5
37	-	-	-	-	100

**Table 4.** Statistical analysis of bay leaf oil for toxicity studies.

S. N.	Time (hrs)	LC <sub>50</sub> (mg g <sup>-1</sup> )	Heterogeneity	
			x <sup>2</sup>	d.f.
1	24	3.6	3.58	3
2	48	2.7	2.13	3
3	72	2.9	3.6	3
4	96	2.7	5.44	3
5	120	3.0	1.27	3

**Table 6.** Comparison of percent mortality of *T. castaneum* using isolated pure compounds at indicated time intervals after treatment at 4 mg g<sup>-1</sup>.

No. of Days	Corrected percent mortality	
	Eugenol	7,7 dimethyl-3methylene bicyclo[2.2.1]heptan-4-ol
1	50.0	47.5
2	50.0	50.0
7	60.0	52.5
14	60.7	56.0
20	64.1	59.8
25	64.5	65.6
28	79.1	78.1
30	87.5	84.4
32	95.8	90.6
33	100	90.6
34	-	93.8
35	-	100

**Table 5.** Comparison of corrected percent mortality of *T. castaneum* using non-polar and polar fractions of bay leaf oil at indicated time intervals after treatment at 4 mg g<sup>-1</sup>.

No. of Days	Corrected percent mortality	
	Non-polar fraction	Polar fraction
1	22.5	58.3
2	22.5	60.0
5	32.5	61.6
10	38.4	62.0
14	38.4	66.1
25	40.6	83.3
27	46.8	95.8
29	50.0	97.8
30	53.1	100

layer chromatography of bay leaf essential oil showed three different colored spots having R<sub>F</sub> values of 0.41, 0.59 and 0.76 when visualized by using sulfuric acid: methanol (9: 1) spray reagent.

**Identification and characterization of pure compounds:** Polar fraction upon column chromatography yielded two pure compounds eugenol (1) and 7, 7 dimethyl-3methylene bicyclo [2.2.1] heptan-4-ol(2). IR spectrum of compound (1) showed band at 3514 cm<sup>-1</sup> indicating the presence of -OH group. The bands at

3076, 1638 and 995  $\text{cm}^{-1}$  were due to  $-\text{CH}_2$  (stretching),  $\text{C}=\text{C}$  (stretching) and  $-\text{CH}_2$  (bending) respectively indicate the presence of vinyl group. Bands at 1610, 794 and 746  $\text{cm}^{-1}$  supported the presence of aromatic ring. Also the bands at 1149, 1122 and 1034  $\text{cm}^{-1}$  were attributed to  $-\text{C}-\text{O}$  stretching.  $^1\text{H}$  NMR spectrum of compound showed a 3H singlet at  $\delta$  3.7, which was typical of a methoxy group. The presence of two hydrogen atoms was confirmed by doublets at  $\delta$  3.21 and 3.19 ( $J=6.7$  Hz). A broad singlet due to hydroxyl group was obtained at  $\delta$  5.57 that was exchangeable with deuterium ( $\text{D}_2\text{O}$ ) and hence may be attributed to hydroxyl group ( $-\text{OH}$ ) which was supported by IR spectrum. In addition to this a three hydrogen multiplet system typical of a vinyl group together with another multiplet for three hydrogen atoms in the aromatic ring suggests this compound to be aromatic system very near to eugenol. An INEPT study of the compound showed the presence of two methylenes. Spectral data together with biogenetic considerations showed the compound to be eugenol. A survey of literature confirmed this compound to be eugenol as it showed superimposable IR spectrum with that reported in the literature (Bhagat *et al.*, 1982). This compound was also reported to be present in fresh bay leaf extract (Killic *et al.*, 2004).

The peculiar shape of band at 3376  $\text{cm}^{-1}$  in IR spectrum of compound (2) suggested the presence of  $-\text{OH}$  group. Bands at 3081  $\text{cm}^{-1}$  ( $\text{C}-\text{H}$  bending) indicated the presence of exo-methylene double bond. Presence of gem dimethyl group was also supported by band at 1375  $\text{cm}^{-1}$ .  $^1\text{H}$  NMR spectrum showed a doublet at  $\delta$  0.97 ( $J=6.2$  Hz) arising due to presence of 6H indicating the presence of gem dimethyl group. A doublet was obtained at  $\delta$  4.61 ( $J=10.2$  Hz) for the two H-atoms indicated the presence of exomethylene double bond. Absence of signals in the region  $\delta$  3-4 due to H-atom(s) present on C having OH group indicated the tertiary nature of hydroxyl group. All these data suggested structure (2) for this compound (7,7 dimethyl-3 methylene bicyclo [2.2.1] heptan-4-ol). This structure was further confirmed by  $^{13}\text{C}$  NMR spectral data. It showed the presence of two olefinic carbons at  $\delta$  153.5 and 106.3, while the signal at  $\delta$  82.6 as singlet showed the presence of tertiary hydroxyl group. The data further supported structure (2) for this compound as it showed three methylenes  $\delta$  (38.9, 29.9 and 24.8 as triplet), two methyl groups  $\delta$  (26.7 and 26.1 as quintets), one methyne  $\delta$  (47.8 as doublet) and two quaternary carbons  $\delta$  (82.6 and 53.4 as singlets).

**Insecticidal activity:** Literature revealed that a variety of essential oils and their constituents possessed varying degrees of pest control properties. Their low mammalian toxicity and persistence in the environment made plant based volatile oils attractive candidates for the protection of stored food products against insect pests. Monoterpenoids are widely distributed in the

essential oils of aromatic plants (Roger, 1999; Dahmane *et al.*, 2016) and more than thousand different naturally occurring monoterpenoids were isolated from many higher plants including mint, pine, cedar, citrus and eucalyptus (Charlwood and Charlwood, 1991). Monoterpenes being lipophilic in nature may interfere with basic metabolite, biochemical, physiological and behavioral functions of insects (Brattsten, 1983).

The results pertaining to the treatment of bay leaf oil against *T. castaneum* are summarized in Table 3. The perusal of Table showed 83.3, 73.3, 61.6, 60.3 and 53.3 per cent mortality at concentration of 12, 10, 8, 6 and 4  $\text{mg g}^{-1}$  respectively one day after treatment. It showed more than 80 per cent mortality of adults at 12  $\text{mg ml}^{-1}$  after one day of exposure. After 2 days of treatment 88.3, 76.6, 73.3, 73.3 and 60.0 per cent mortality was observed at 12, 10, 8, 6 and 4  $\text{mg g}^{-1}$  respectively. The data revealed that more than 70 per cent mortality was observed at all the four concentrations except at lowest concentration of 4  $\text{mg g}^{-1}$ . Ninety five per cent mortality was observed on fourth day of exposure at 12  $\text{mg g}^{-1}$ . Complete mortality was observed after 5 and 8 days of application at the rate of 12 and 10  $\text{mg g}^{-1}$  respectively. The adult mortality of 83.9, 79.2 and 69.7 per cent was observed at treatments level of 8, 6 and 4  $\text{mg g}^{-1}$  respectively after 8 days of treatment. The data was in consonance with the earlier findings which showed that mortality increased with increase in concentration of lemongrass, eucalyptus, bottle brush and garlic oils (Chahal *et al.*, 2004, 2005 and 2006). The  $\text{LC}_{50}$  values of bay leaf essential oil ranges between 2.7-3.6  $\text{mg g}^{-1}$ , showing it to be moderately effective against *T. castaneum* (Table 4). In order to study interactions between monoterpenoid and insects pests attempts have been made to isolate pure compounds from these essential oils and evaluate their bio efficacy against *T. castaneum*.

The perusal of Table 5 indicates that at 4  $\text{mg g}^{-1}$  concentration the non polar and polar fractions of bay leaf oil showed 22.5 and 58.3 percent kill respectively after 24 hrs of exposure period. More than 50 percent insect mortality was observed in case of polar fraction. The nonpolar fraction showed 32.5, 38.4, 38.4, 40.6, 46.8, 50.0 and 53.1 percent mortality after 5,10,14,25,27,29 and 30 days of application. However the polar fraction showed 61.6,62.0, 66.1,83.3,95.8 and 97.8 percent mortality at the end of 5,10,14,25,27 and 29 days of treatment respectively. The complete mortality was observed after 30 days of treatment in case of polar fraction whereas 53.1 percent mortality was observed on same day with non polar fraction and it remained constant upto 45 days after application. It was also observed that rate of mortality of adult insect in nonpolar fraction was less than the polar fraction. Moreover the percent mortality in case nonpolar fraction remained constant after 30 days of exposure period showing that nonpolar fraction of oil was not exerting

any further toxic action on the insects. The difference in percent mortality between two fractions may be due to different toxic effect of compounds present in these two fractions. The above data revealed that polar fraction was more toxic than nonpolar fraction. These results were contradictory with the earlier findings which showed the non-polar fractions to be more bioactive as compared to polar fraction (Chahal *et al.*, 2014; Sharma and Chahal, 2012; Ray *et al.*, 2010; Sharma *et al.*, 2007). The difference may be due to presence of more bioactive compounds in polar fraction than nonpolar fraction.

In order to explore the potential of bioactive compounds present in the bay leaf oil, two major compounds eugenol (1) and 7,7 dimethyl-3-methylene bicyclo [2.2.1] heptan-4-ol (2) isolated from polar fraction were tested for their insecticidal activity against *T. castaneum*. The perusal of the Table 6 depicted the influence of both compounds on mortality rate of adults of test insect. After 24 hrs of exposure 50.0 and 47.5 per cent adult mortality was observed in case of compound (1) and (2) respectively. Eugenol (1) showed 60.0, 60.7 and 64.5 percent kill whereas compound (2) showed 52.5, 56.0 and 65.6 percent mortality at the end of 7, 14 and 25 days of treatment respectively which revealed that both the compounds were equally toxic towards stored grain pest. After 28 days of exposure more than 78 percent adult mortality was observed for both the compounds tested. After 30 days of treatment 87.5 (1) and 84.2 (2) percent mortality was attained. Complete mortality for compounds (1) and (2) was observed after 33 and 35 days of treatment. Although there was increase in percent mortality with increase in exposure period but percent kill was more by compound (1) than compound (2). Thus eugenol was more active as insecticide than 7,7 dimethyl-3-methylene bicyclo [2.2.1] heptan-4-ol. This data was in agreement with findings of Obeng and Reichmuth (2010) who investigated the toxicity and protectant potential of eugenol (1) against *Sitophilus granarius*, *S. zeamais*, *T. castaneum* and *Prostephanus truncates*. Eugenol applied topically, impregnated on filter papers was highly toxic to all four species. Beetle mortality was dose dependent. Eugenol was highly repellent to the four beetle species tested with overall repellency in the range of 80-100 percent. Development of eggs and immature stages inside grain kernels was completely inhibited by eugenol treatment. In another study, Huang *et al.* (2002) reported the toxic effect of eugenol (1) against the adults of *S. zeamais* and *T. castaneum*, eugenol (1) that significantly reduced the food consumption in case of *T. castaneum* in flour disc bioassay with no choice test.

## Conclusion

The present study revealed that the polar fraction of bay leaf essential oil was found to be most effective as compared to its fractions and compounds tested. The

essential oil at 10 and 12 mg ml<sup>-1</sup> was found to be effective in controlling infestation. The more toxic effect of polar fraction could be due to synergistic effect of the compounds present in it. Therefore plant oils may be the best alternative of insecticides because these are safe to environment and human beings.

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