

The Molecular Structure of Monofluorobenzaldehydes

Issiah Byen Lozada, Wenhao Sun, Jennifer van
Wijngaarden

Department of Chemistry, University of Manitoba
Winnipeg, MB, Canada

Previous Studies

J. Chem. Soc., Faraday Trans. 2, 1989, 85(2), 137–149

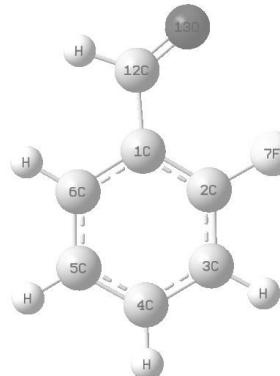
Rotational Isomerism in Monofluorobenzaldehydes

José L. Alonso* and Rosa M. Villamañán

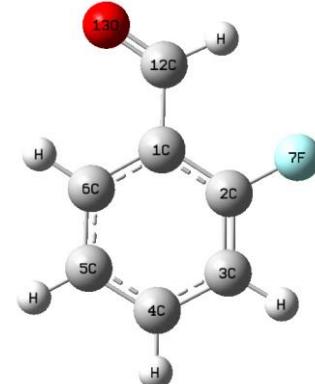
Departamento de Química-Física, Facultad de Ciencias, Universidad de Valladolid,
Valladolid 47005, Spain

Microwave spectra of 2- and 3-fluorobenzaldehydes have been investigated in the frequency region 26.5–40.0 GHz. Only the *O-trans* form of 2-fluorobenzaldehyde was detected in the gas phase. Analysis of 3-fluorobenzaldehyde confirmed the existence of two rotational isomers: *O-cis* and *O-trans*. From relative intensity measurements the *O-cis* form was shown to be more stable than the *O-trans* form by 300 ± 200 cal mol $^{-1}$. The inertia defect values and the variation of the rotational constants with the torsional quantum number demonstrate that the equilibrium geometry is planar for all rotamers. Molecular structures which are in very good agreement with experimental rotational constants have also been derived.

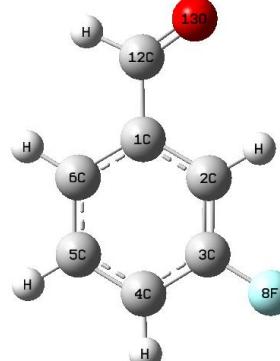
- RFMWDR technique (26.5 – 40.0 GHz)
- 3 vibrational states for *O-trans* 2FBA
- 4 vibrational states for each 3FBA conformer



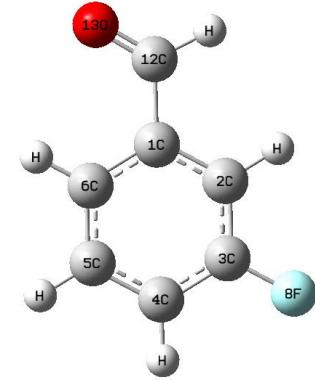
***O-cis* 2FBA**



***O-trans* 2FBA**

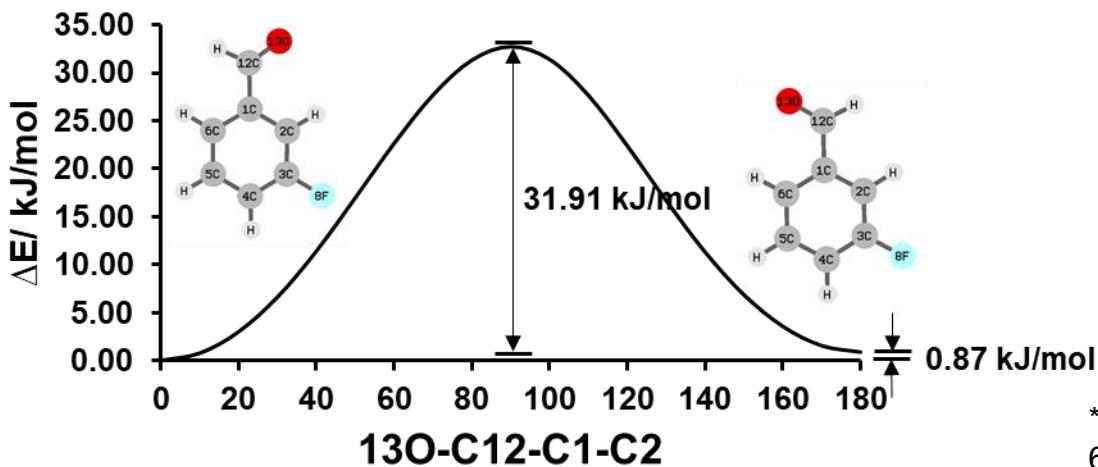
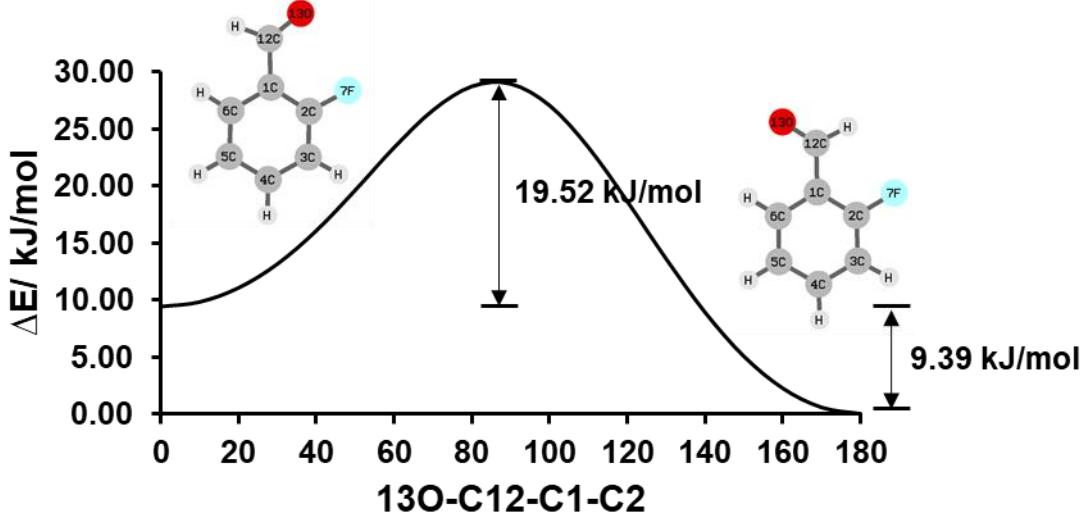


***O-cis* 3FBA**



***O-trans* 3FBA**

Potential Energy Curve



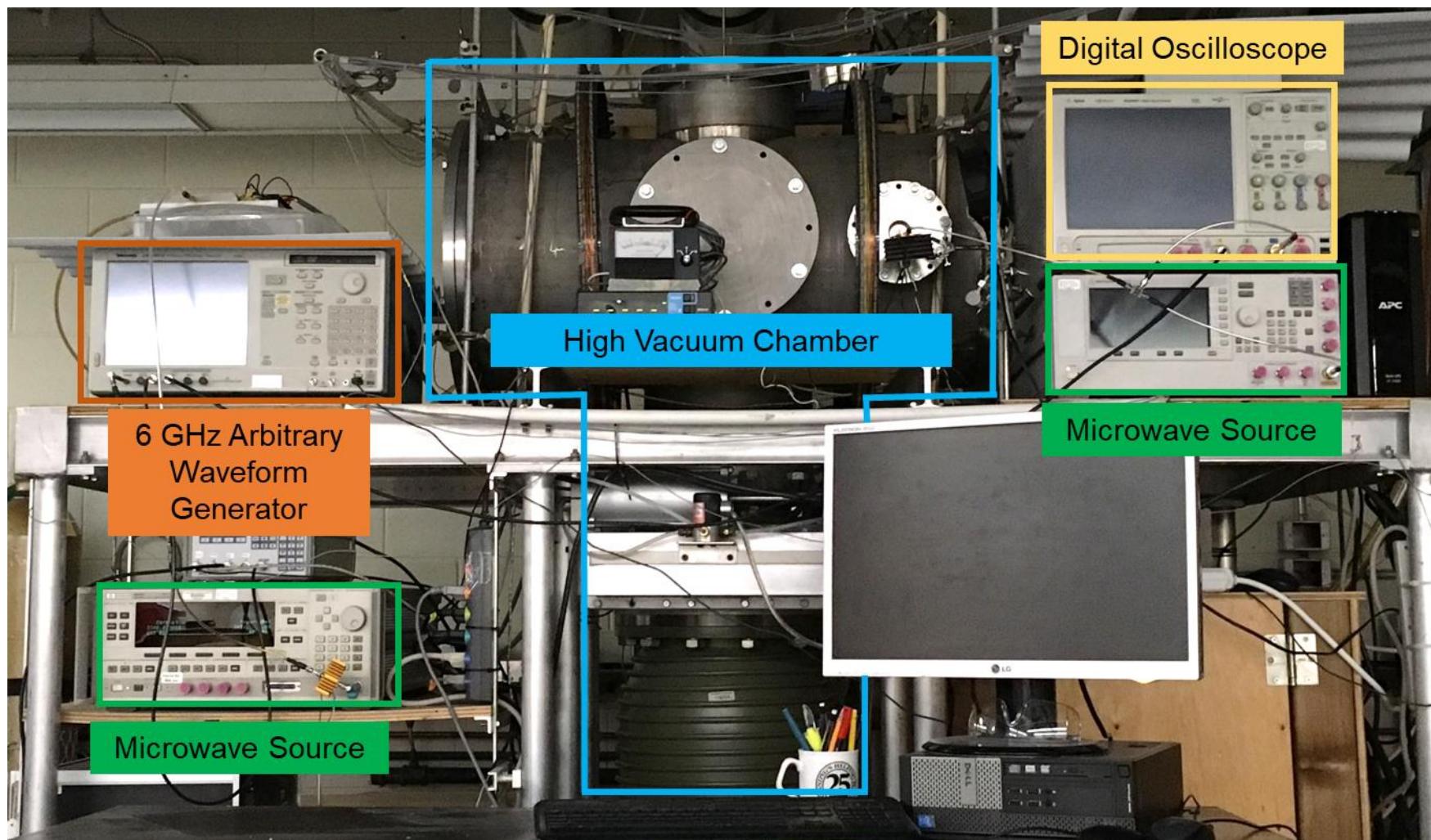
Previous study

	2-FBA	
/MHz	O-cis	O-trans
A		2567.609(3)
B		1560.8694(9)
C		970.954(1)

	3-FBA	
/MHz	O-cis	O-trans
A	2919.255(2)	3657.169(5)
B	1269.697(2)	1114.781(2)
C	884.964(2)	854.530(2)

* Potential energy curves are carried out at MP2/6-311++G(2d,2p)

cp FTMW Spectrometer



Experimental Conditions

Spectral range 8-18 GHz

Bandwidth 2000 MHz

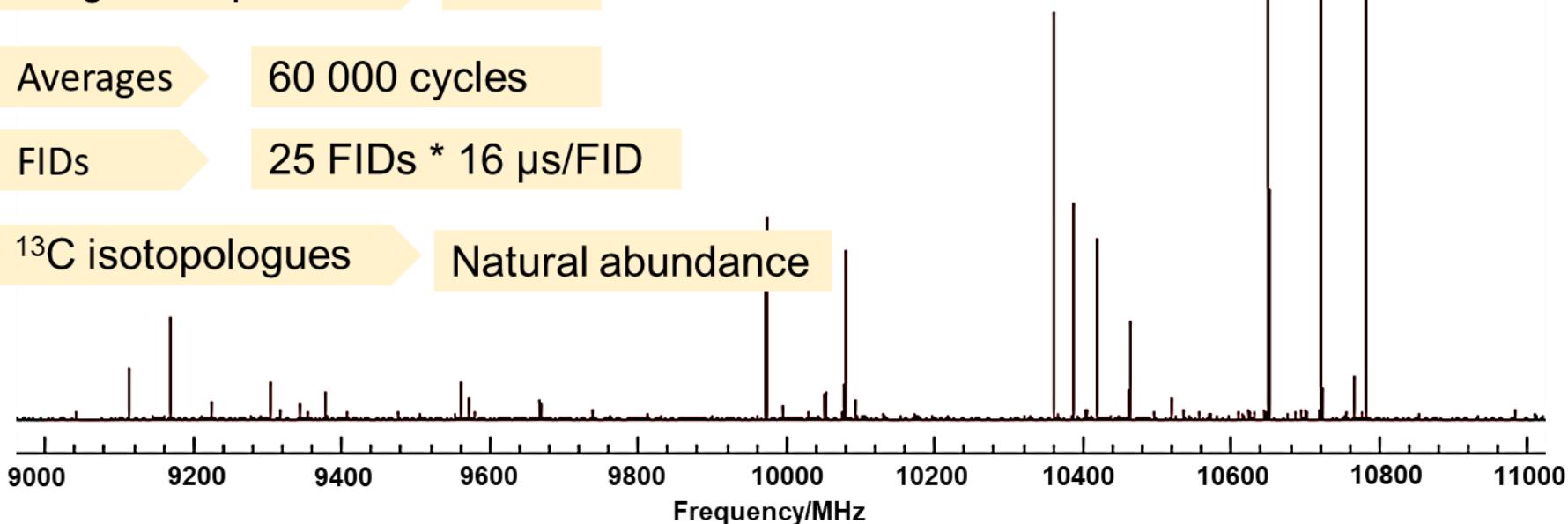
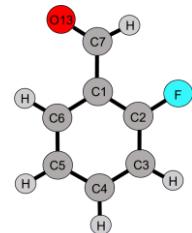
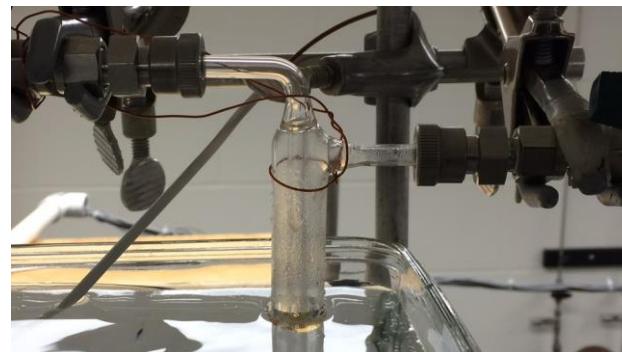
Carrier gas Neon

Stagnation pressure 1 bar

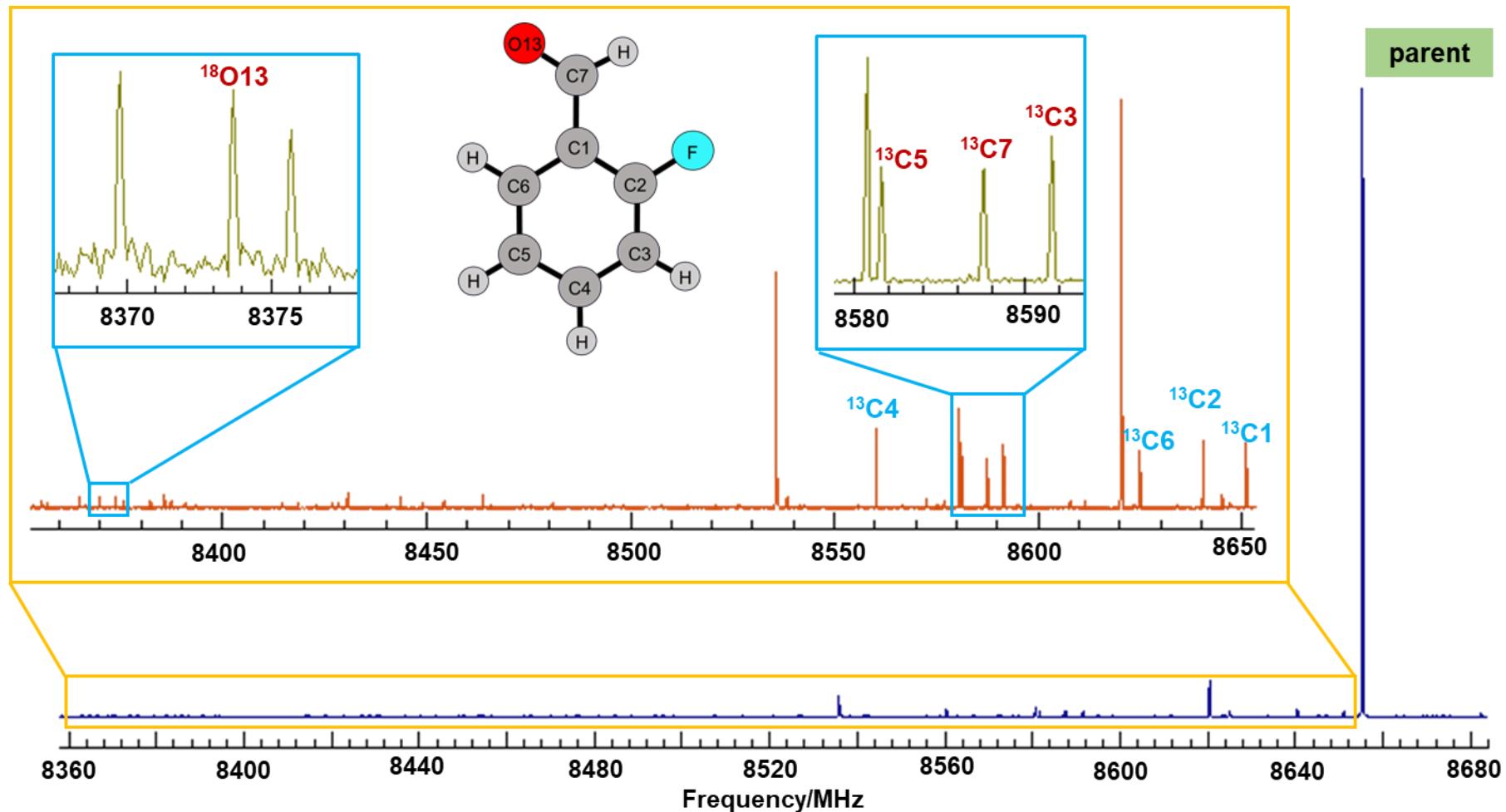
Averages 60 000 cycles

FIDs 25 FIDs * 16 μ s/FID

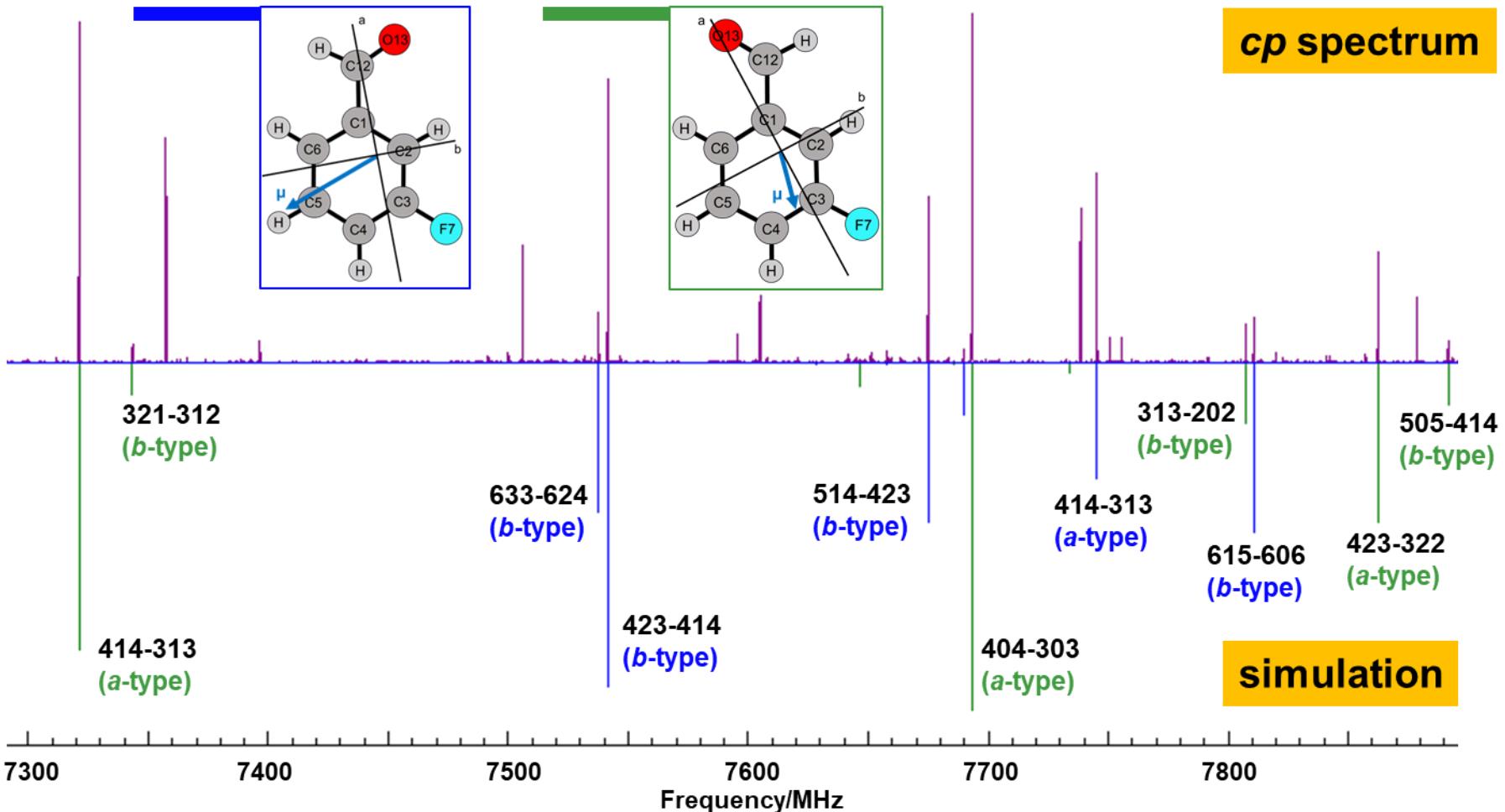
^{13}C isotopologues Natural abundance



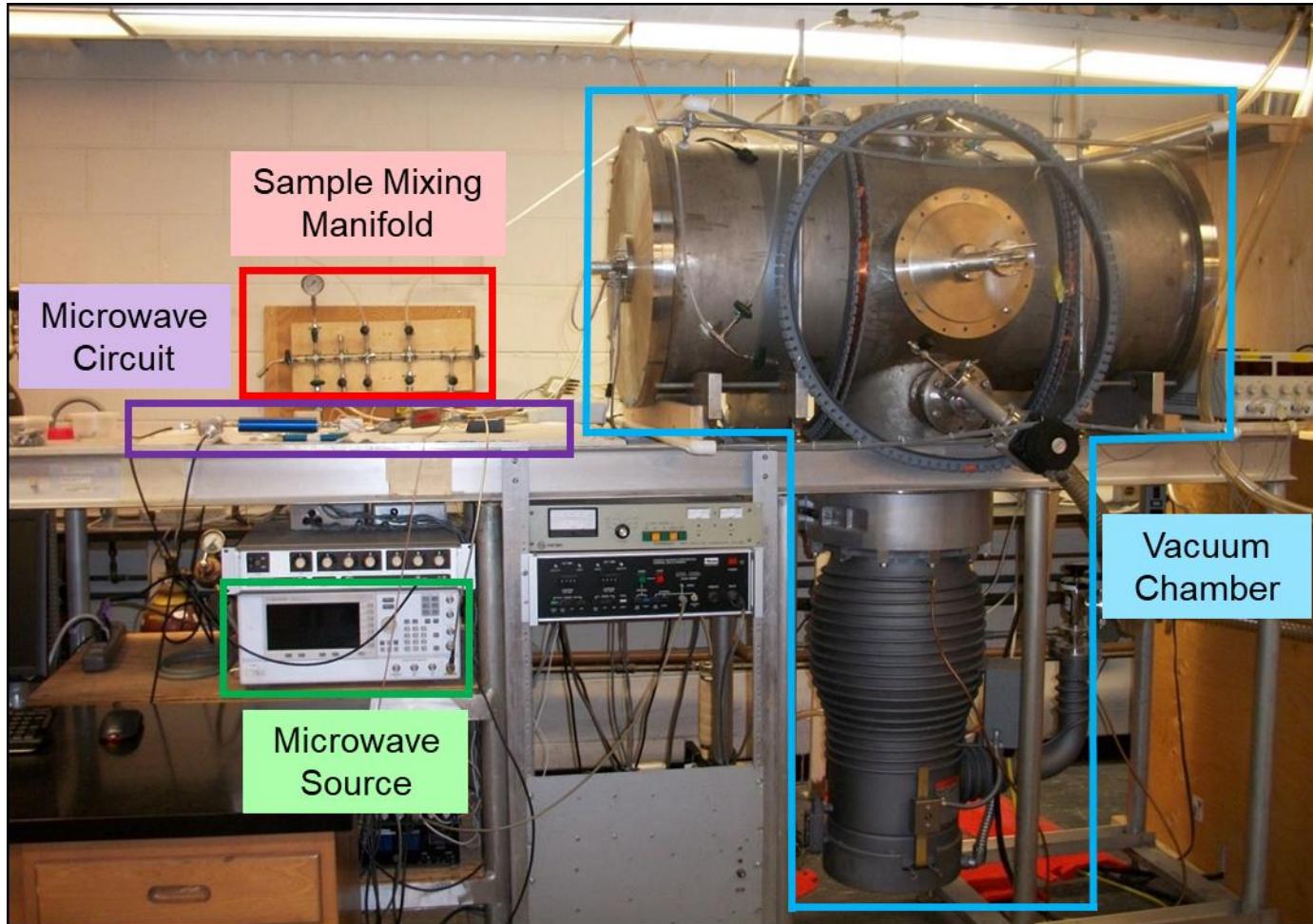
O-trans 2FBA: 414 - 313



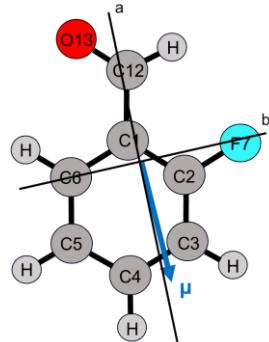
3FBA: O-cis & O-trans



Balle Flygare FTMW Spectrometer

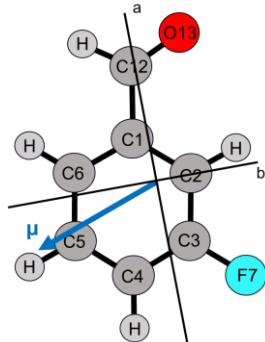


Line collection: 4-26 GHz



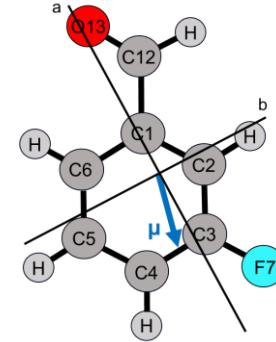
	Dipole/D	#line
a-type	3.81	124
b-type	0.17	0

Isotope	#line
¹³ C-1	25
¹³ C-2	25
¹³ C-3	28
¹³ C-4	28
¹³ C-5	25
¹³ C-6	29
¹³ C-12	28
¹⁸ O-13	12



	Dipole/D	#line
a-type	1.58	24
b-type	3.90	55

Isotope	#line
¹³ C-1	7
¹³ C-2	9
¹³ C-3	7
¹³ C-4	13
¹³ C-5	9
¹³ C-6	17
¹³ C-12	18



	Dipole/D	#line
a-type	2.18	39
b-type	0.08	10

Isotope	#line
¹³ C-1	6
¹³ C-2	9
¹³ C-3	8
¹³ C-4	9
¹³ C-5	8
¹³ C-6	10
¹³ C-12	9

* All dipole moments are calculated at MP2/ aug-cc-pVTZ

SPFIT: Watson - A reduction

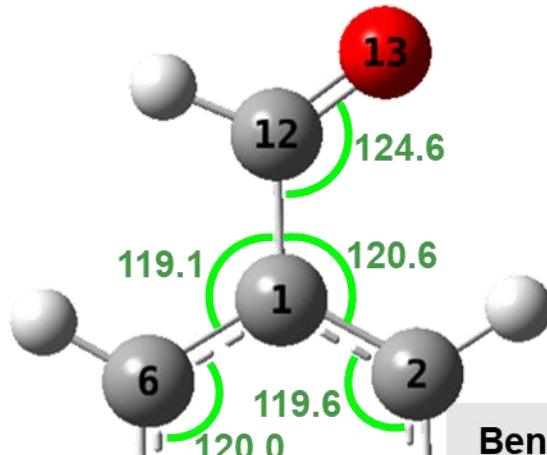
2FBA	O-trans	
	This work	Previous work*
A/MHz	2567.599481(46)	2567.609(3)
B/MHz	1560.865217(16)	1560.8694(9)
C/MHz	970.950256(10)	970.954(1)
Δ_J /kHz	0.059225(94)	0.072(2)
Δ_{JK} /kHz	0.13286(34)	0.128(4)
Δ_K /kHz	0.1297(16)	0.19(2)
δ_J /kHz	0.021890(48)	0.0221(5)
δ_K /kHz	0.15851(33)	0.151(4)
RMS/kHz	0.484	46

3FTBA	O-cis		O-trans	
	This work	Previous work*	This work	Previous work*
A/MHz	2919.250865(64)	2919.255(2)	3657.20290(52)	3657.169(5)
B/MHz	1269.700848(41)	1269.697(2)	1114.785459(66)	1114.781(2)
C/MHz	884.961618(28)	884.964(2)	854.522877(50)	854.530(2)
Δ_J /kHz	0.06489(31)	0.062(4)	0.02374(33)	0.024(4)
Δ_{JK} /kHz	-0.1789(15)	-0.17(2)	0.1237(17)	0.13(1)
Δ_K /kHz	0.6396(37)	0.67(5)	0.75(51)	\
δ_J /kHz	0.02357(15)	0.018(2)	0.00601(21)	\
δ_K /kHz	0.1043(17)	0.08(2)	0.115(18)	\
RMS/kHz	0.806	47	0.491	66

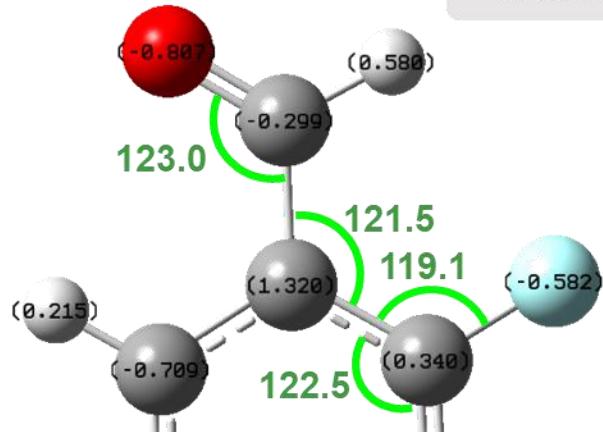
Effective Structure - r_0

	2FBA O-trans		O-cis		3FBA O-trans	
	r_e	r_0	r_e	r_0	r_e	r_0
R(C1-C2)	1.393	1.385(11)	1.398	1.407(9)	1.397	1.379(9)
R(C3-C2)	1.385	1.388(3)	1.382	1.375(11)	1.385	1.396(6)
R(C4-C3)	1.392	1.396(4)	1.391	1.392(4)	1.387	1.388(3)
R(C5-C4)	1.397	1.399(6)	1.393	1.395(4)	1.396	1.396(5)
R(C6-C5)	1.388	1.402(10)	1.393	1.405(10)	1.390	1.398(6)
R(C1-C6)	1.400	1.402(13)	1.396	1.385(7)	1.397	1.410(12)
R(C7-C1)	1.480	1.491(6)	1.479	1.491(5)	1.479	1.488(6)
R(O13-C7)	1.220	1.218(10)				
\angle (C3-C2-C1)	122.5	122.8(5)	117.9	117.3(5)	118.3	118.0(5)
\angle (C4-C3-C2)	118.5	118.1(3)	122.5	123.2(3)	122.2	122.5(2)
\angle (C5-C4-C3)	120.5	120.6(1)	118.8	118.5(2)	118.7	118.5(2)
\angle (C6-C5-C4)	120.0	120.0(2)	120.1	120.0(2)	120.5	120.5(2)
\angle (C2-C1-C6)	120.6	119.8(5)	119.8	119.5(5)	119.5	119.0(3)
\angle (C1-C6-C5)	118.1	118.7(6)	120.8	121.5(6)	120.7	121.3(5)
\angle (C7-C1-C2)	121.5	121.5(10)	119.9	119.4(6)	118.6	119.1(9)
\angle (O13-C7-C1)	123.0	122.5(11)				

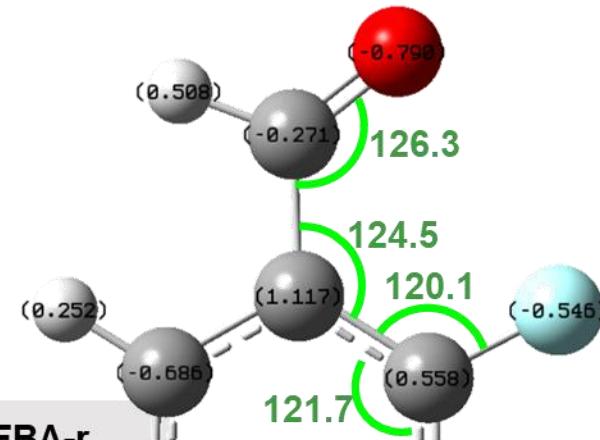
Structural Analysis: 2FBA



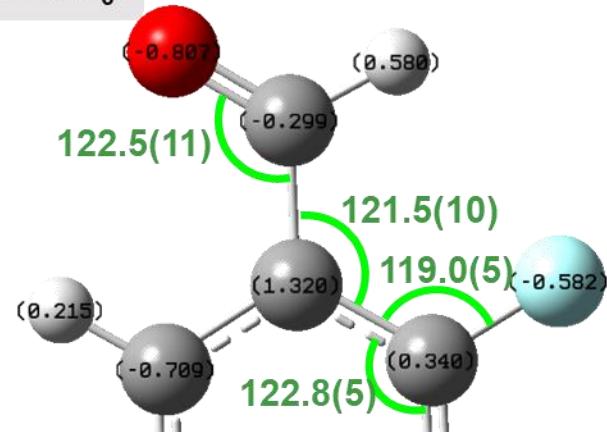
Benzaldehyde-r_e



O-trans 2FBA-r_e



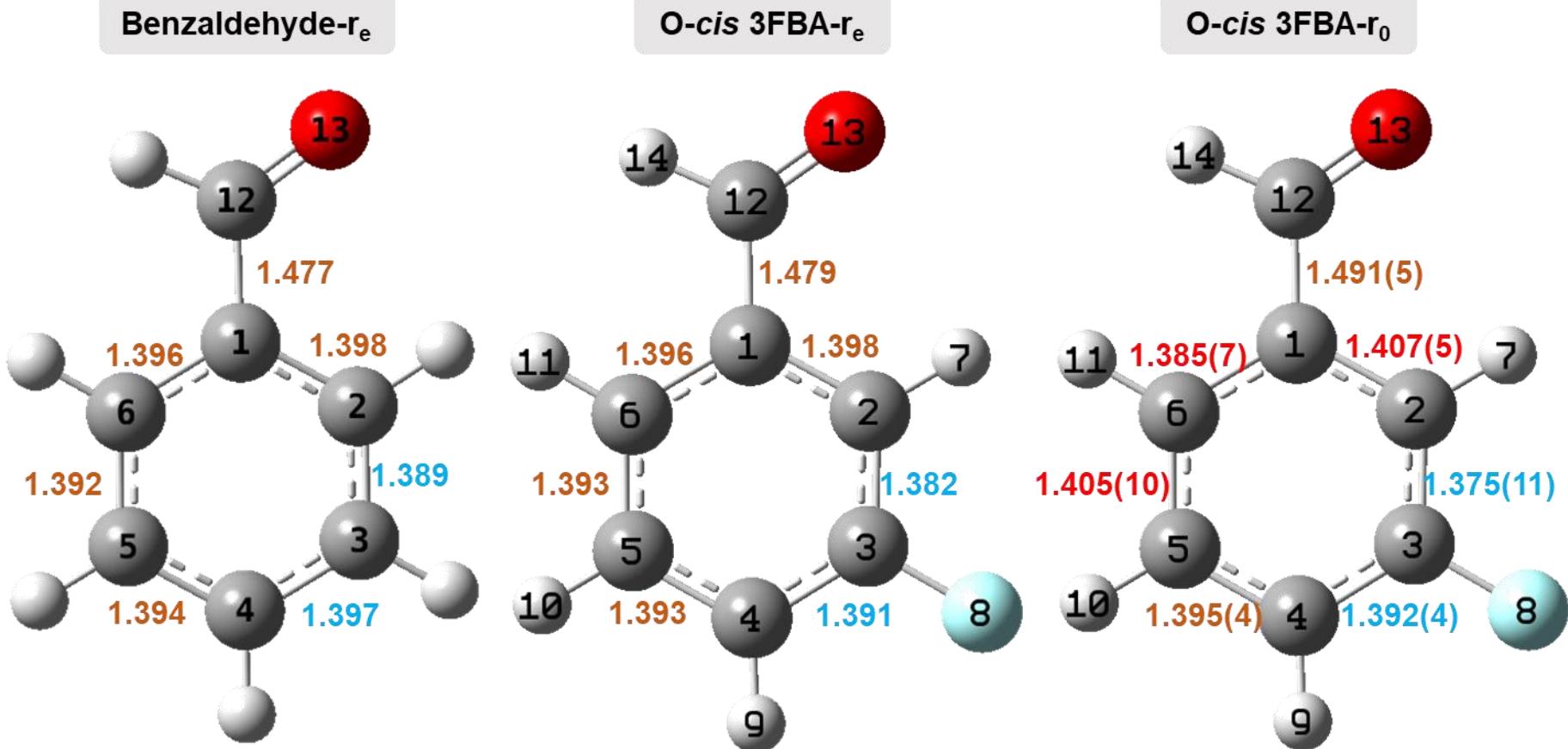
O-cis 2FBA-r_e



O-trans 2FBA-r₀

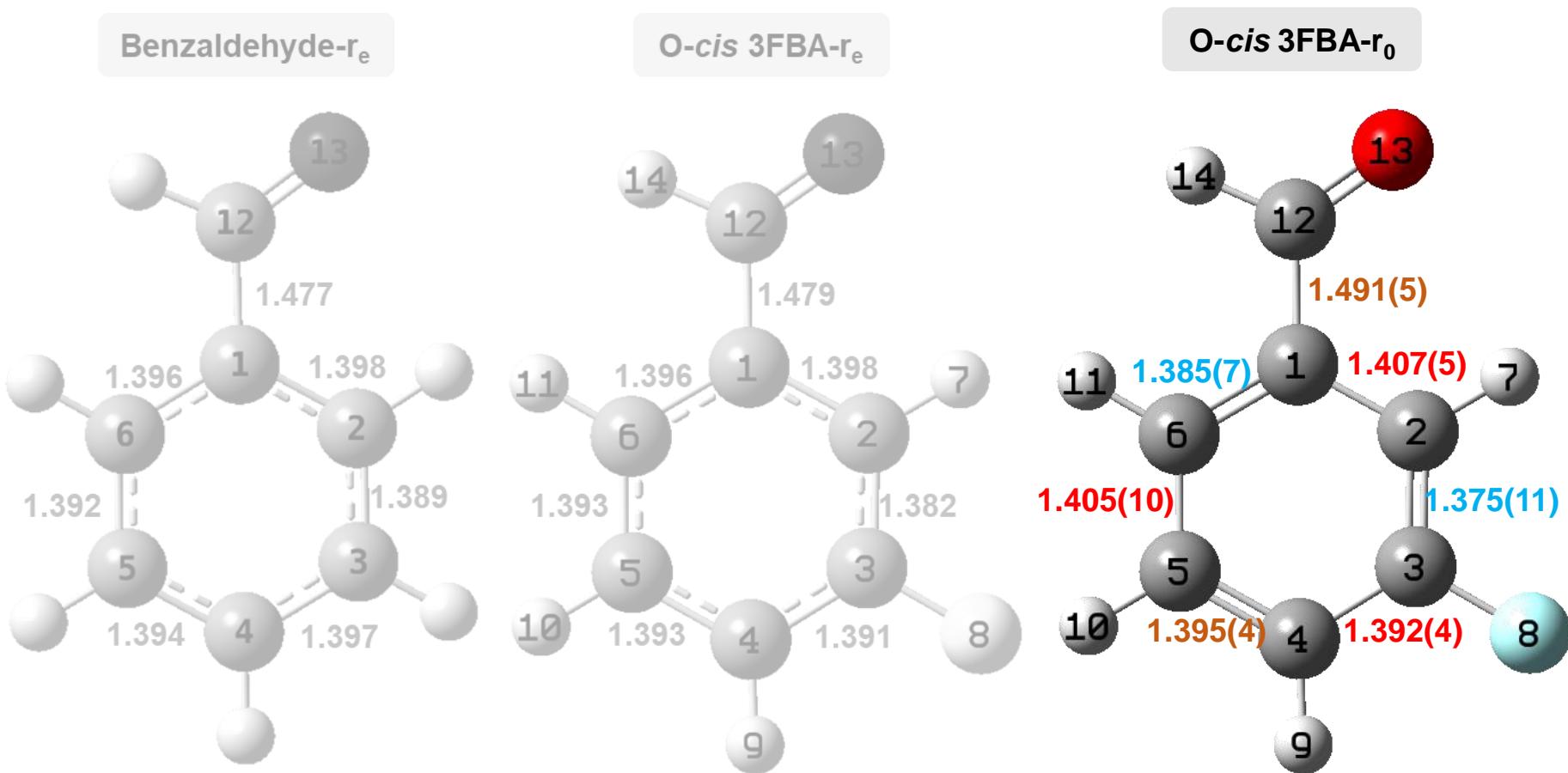
Equilibrium structures are calculated at MP2/ aug-cc-pVTZ, and atomic charges shown are Mulliken charges.

Structural Analysis: O-cis 3FBA



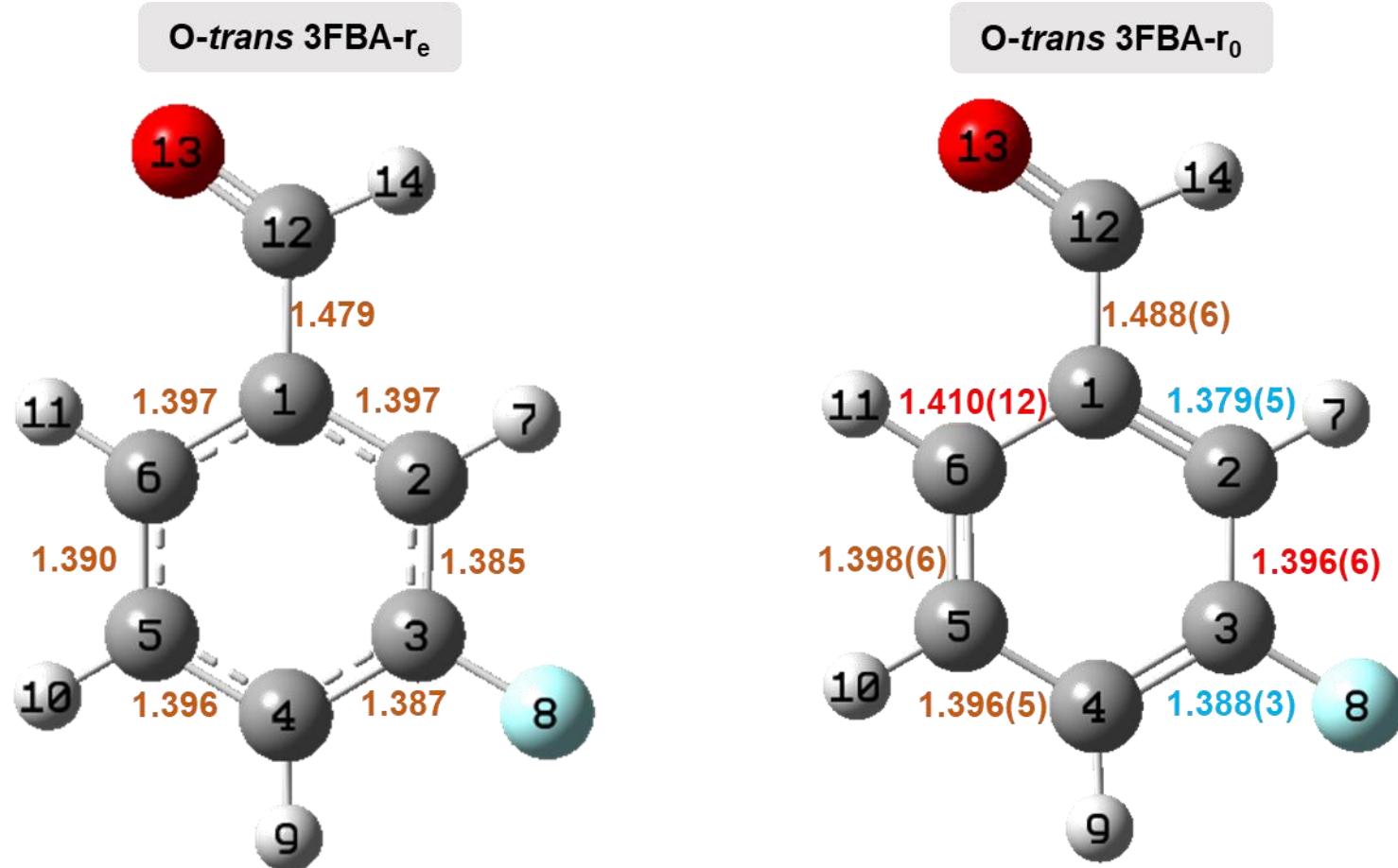
Equilibrium structures are calculated at MP2/ aug-cc-pVTZ.

Structural Analysis: O-*cis* 3FBA



Equilibrium structures are calculated at MP2/ aug-cc-pVTZ.

Structural Analysis: O-trans 3FBA



Equilibrium structure is calculated at MP2/ aug-cc-pVTZ.

Acknowledgements



UNIVERSITY
OF MANITOBA