

GAS-PHASE MOLECULAR STRUCTURE OF NOPINONE AND ITS WATER COMPLEXES STUDIED BY MICROWAVE FOURIER TRANSFORM SPECTROSCOPY AND QUANTUM CHEMICAL CALCULATIONS

ELIAS M. NEEMAN, JUAN-RAMON AVILES MORENO, T. R. HUET, *UMR 8523 CNRS - Universités des Sciences et Technologies de Lille, Laboratoire PhLAM, 59655 Villeneuve d'Ascq, France.*

Several monoterpenes and terpenoids are biogenic volatile organic compounds which are emitted in the atmosphere, where they react with OH, O₃ and NO_x etc. to give rise to several oxidation and degradation products.^a Their decomposition products are a major source of secondary organic aerosol (SOA).^b Spectroscopic information on these atmospheric species is still very scarce. The rotational spectrum of nopinone (C₉H₁₄O) one of the major oxidation products of β-pinene,^{c, d} and of its water complexes were recorded in a supersonic jet expansion with a Fourier transform microwave spectrometer over the range 2-20 GHz. The structure of the unique stable conformer of the nopinone was optimized using density functional theory and *ab initio* calculations. Signals from the parent species and from the ¹³C and ¹⁸O isotopomers were observed in natural abundance. A magnetic hyperfine structure associated with the pairs of hydrogen nuclei in the methylene groups was observed and modeled.

The structures of several conformers of the nopinone-water complexes with up to three molecules of water were optimized using density functional theory and *ab initio* calculations. The energetically most stable of calculated conformers were observed and analyzed. The rotational and centrifugal distortion parameters were fitted to a Watson's Hamiltonian in the A-reduction. The present work provides the first spectroscopic characterization of nopinone and its water complexes in the gas phase.

^aA. Calogirou, B.R. Larsen, and D. Kotzias, *Atmospheric Environment*, **33**, 1423-1439, (1999).

^bP. Paasonen et al., *Nat. Geosci.*, **6**, 438-442 (2013)

^cD. Zhang and R. Zhang *The Journal of Chemical Physics*, **122**, 114308, (2005).

^dR. Winterhalter et al. *Journal of Atmospheric Chemistry*, **35**, 165-197, (2000).