ASSIGNMENT OF THE PERFLUOROPROPIONIC ACID-FORMIC ACID COMPLEX AND THE DIFFICULTIES OF INCLUDING HIGH K_a TRANSITIONS.

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We recently began an investigation into the perfluoropropionic $\operatorname{acid}\cdots$ formic acid complex using broadband microwave spectroscopy. This study aims to examine the possible double proton transfer between the two interacting carboxcyclic acid groups. The spectrum presented as a doubled set of lines, with spacing between transitions of < 1 MHz. Transitions appeared to be *a*-type, R branch transitions for an asymmetric top. Assignment of all $K_a = 1, 0$ transitions yields decent fits to a standard rotational Hamiltonian. Treatment of the doubling as either a two state system (presumably with a double proton transfer) or as two distinct, but nearly identical conformations of the complex produce fits of similar quality. Including higher K_a transitions for the *a*-type, R-branch lines greatly increases the error of these fits. A previous study involving the trifluoroacetic $\operatorname{acid}\cdots$ formic acid complex published observed similar high K_a transitions, but did not include them in the published fit.^a We hope to shed more light on this conundrum. Similarities to other double-well potential minimum systems will be discussed.

^aMartinache, L.; Kresa, W.; Wegener, M.;, Vonmont, U.; and Bauder, A. Chem. Phys. 148 (1990) 129-140.