

INFRARED IDENTIFICATION OF THE CRIEGEE INTERMEDIATE $(\text{CH}_3)_2\text{COO}$

YI-YING WANG, *Applied Chemistry, National Chiao Tung University, Hsinchu, Taiwan*;
YUAN-PERN LEE, *Applied Chemistry, National Chiao Tung University, Hsinchu, Taiwan, Institute
of Atomic and Molecular Sciences, Academia Sinica, Taipei, Taiwan.*

The Criegee intermediates are carbonyl oxides that play critical roles in ozonolysis of alkenes in the atmosphere. We reported previously the mid-infrared spectra of the simplest Criegee intermediate CH_2OO ,^{a, b} and the methyl-substituted intermediate CH_3CHOO .^c Here we report the transient infrared spectrum of $(\text{CH}_3)_2\text{COO}$, produced on UV photolysis of a mixture of $(\text{CH}_3)_2\text{Cl}_2$, N_2 , and O_2 in a flow reactor, using a step-scan Fourier-transform spectrometer. Guided by results of quantum-chemical calculations, rotational contours of the four observed bands are simulated successfully and provide definitive identification of $(\text{CH}_3)_2\text{COO}$. Although all observed bands of $(\text{CH}_3)_2\text{COO}$ contain hot bands from four vibrational modes of low energy, we were able to simulate the spectra satisfactorily. Observed bands with origins near 887, 1040, 1368, and 1422 cm^{-1} agree satisfactorily with corresponding anharmonic vibrational wavenumbers at 903, 1061, 1364, and 1422 cm^{-1} predicted with the B3LYP/aug-cc-pVTZ method. Furthermore, we could also estimate the rate coefficient of the self-reaction of $(\text{CH}_3)_2\text{COO}$. The direct infrared detection of $(\text{CH}_3)_2\text{COO}$ should prove useful for future field measurements and laboratory investigations of this Criegee intermediate.

^aY.-T. Su, Y.-H. Huang, H. A. Witek, Y.-P. Lee, *Science* **340**, 174 (2013).

^bY.-H. Huang, J. Li, H. Guo, Y.-P. Lee, *J. Chem. Phys.* **142**, 214301 (2015).

^cH.-Y. Lin, Y.-H. Huang, X. Wang, J. M. Bowman, Y. Nishimura, H. A. Witek, Y.-P. Lee, *Nat. Comm.* **6**, 7012 (2015).