

**PYROLYTIC PRODUCTION OF CARBON
SELENIDE FROM CARBON
DISELENIDE**

BY

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TO PETE

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I. INTRODUCTION

This paper deals with the pyrolytic production of CSe from carbon diselenide,



Interest in the decomposition of CSe_2 stems from previous studies of molecules isoelectronic with CSe_2 (CO_2 , N_2O , CS_2 , OCS) [1-4].

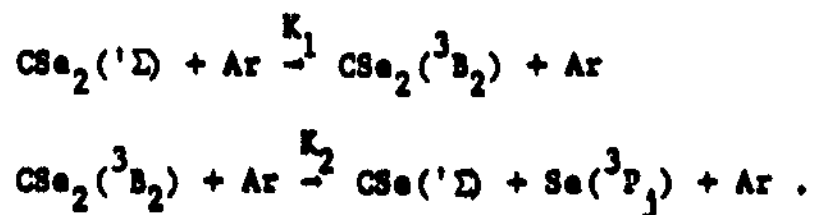
At this time little is known about CSe. In the past, investigations on CSe have been limited to collection of emission spectra and qualitative absorption data. For example, Callear and Tyerman produced CSe by flash photolysis of CSe_2 and observed its absorption spectrum [5]. However the spectrum did not show the relative intensities of the absorption peaks. Other studies have dealt with the emission spectra of CSe_2 and CSe [6-7] and the absorption spectrum of CSe_2 and Se [8-11]. To the author's knowledge this is the first quantitative study of CSe absorption.

An attempt was made first to obtain an absorption spectrum for CSe which included the relative intensities of the absorption peaks. Obtaining an absorption spectrum would have made a good starting point for an investigation of the properties of CSe. However, due to the lack of light intensity which resulted in a poor signal to noise ratio, poor resolution (broad spectral bandpass), the inability to reproduce shock conditions (the optical system could monitor only one wavelength per shock experiment) and the interference from CSe_2 hot bands, attainment of a spectrum proved impractical without modification of the experimental procedure.

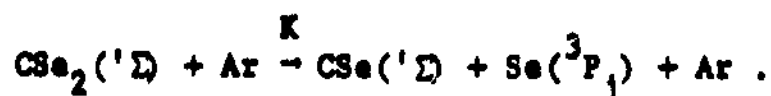
The focus of this paper is on the determination of the rate constant, $K(t)$, from a pseudo first order rate expression:

$$\frac{-\partial[\text{CSe}_2]}{\partial t} = k(t)[\text{Ar}][\text{CSe}] ,$$

where the concentration of argon is assumed to be time independent. The arrhenious parameters (pre-exponential factor and activation energy) are also evaluated. The results are compared with previous studies to help resolve the conflict arising from the two different theories which exist. These theories concern the mechanism by which the decomposition of CSe_2 proceeds. The first theory, developed by Graziano [11], states that the reaction proceeds via a two-step reaction:



The second theory, developed by R. L. Belford and J. R. Marquart, states that the reaction occurs via a one-step process:



Since CSe is very unstable and short-lived (millisecond life-times), it had to be produced immediately before its spectrum is obtained. For this reason we produced it by pyrolysis of CSe_2 in a shock tube.

II. EXPERIMENTATION

The two inch in diameter shock tube used to thermally dissociate CSe_2 gas has been described in detail elsewhere [11], thus only a brief summary will be given here. Light produced by a deuterium lamp is passed laterally through the shock tube. It is then analyzed using a McPherson 218 grating monochromator and a RCA IP28A photomultiplier. To simplify data collection

and analysis, a high speed transient recorder and an analog to digital converter is interfaced with a Vax/Vms-II computer. The system is triggered by a piezoelectric transducer which senses the approach of the shock wave. 2048 data points can be collected (at 0.5 microsecond intervals) for each shock. The experimental set up is shown in Figure 1. The shock tube also has interchangeable incident and reflected endplates. The latter instantly reflects the shock wave back past the optical window while the former delays the reflected shock wave passage past the optical window by approximately 200 microseconds. The delay in the passage of the reflected wave allows for the absorption of light at the lower temperature T_2 .^{*} This absorption occurs after the incident shock wave passes the window, but before passage of the reflected shock wave which increases the temperature to T_3 . This allows for the calculation of the absorbance of CSe_2 at temperatures where only a small amount of CSe_2 has been converted to products.

III. DATA ANALYSIS AND RESULTS

Light intensity data was collected and plotted against time for each shock run (see Figure 2). Figure 2a shows a shock run with the incident endplate and Figure 2b is with the reflected endplate. I_0 represents the incident light intensity. I_2 , visible distinctly in Figure 2b, represents the emerging light intensity after the incident shock wave has passed. This halt is only momentary in Figure 2a. I_3^{∞} represents the intensity of light

^{*}The subscript 0 refers to initial conditions, the subscript 2 refers to the conditions after the incident shock wave has passed, and subscript 3 refers to conditions after both shock waves have passed.

Figure 1. Experimental set up used for the collection of data in this work.

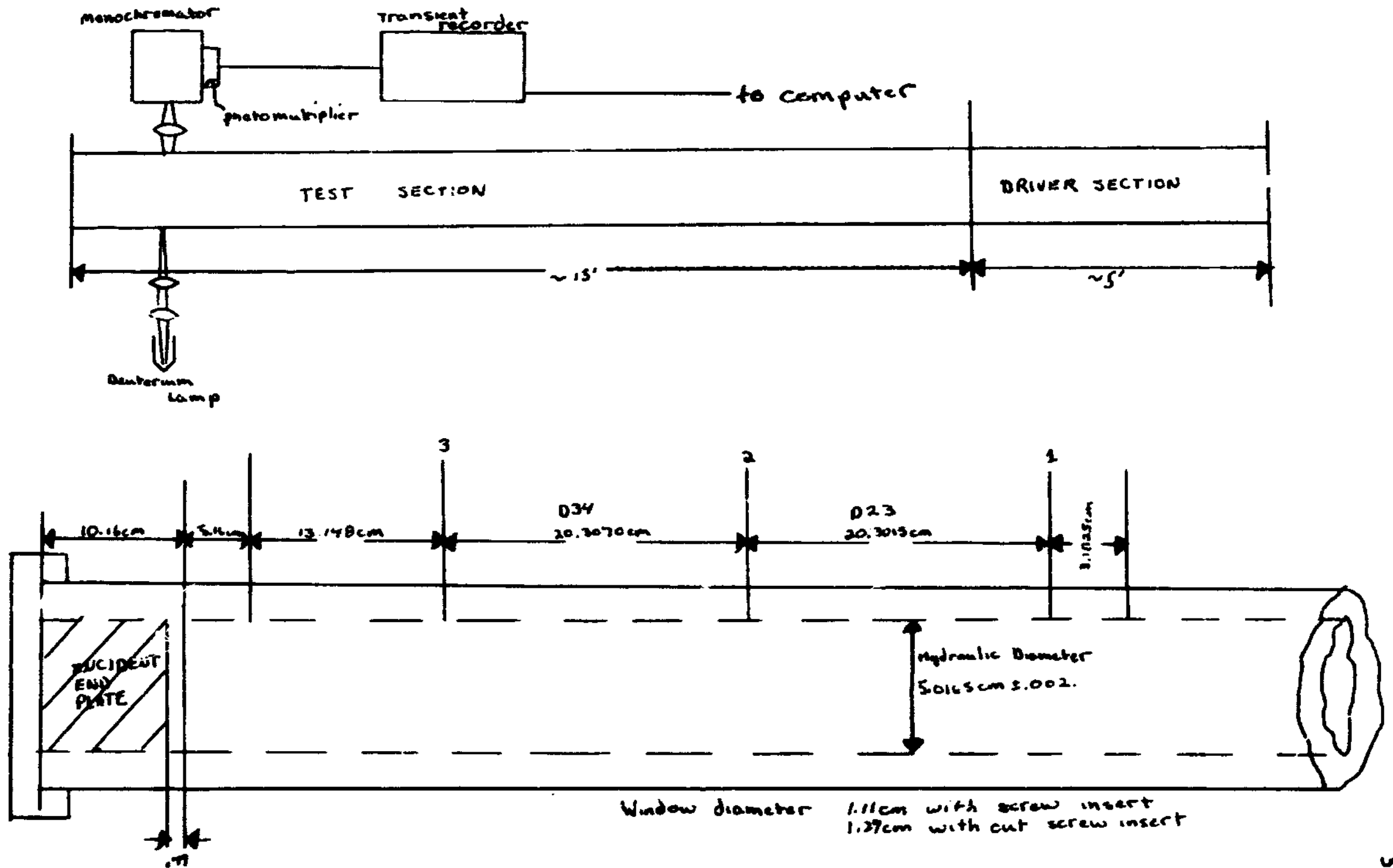
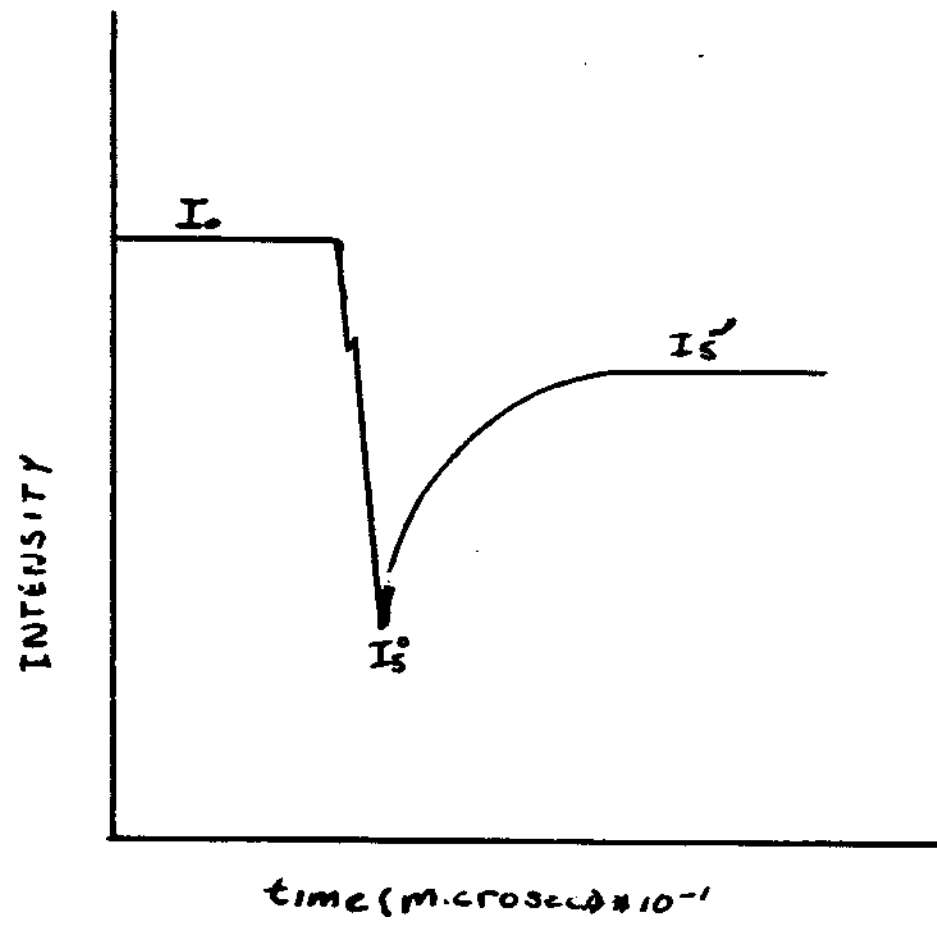
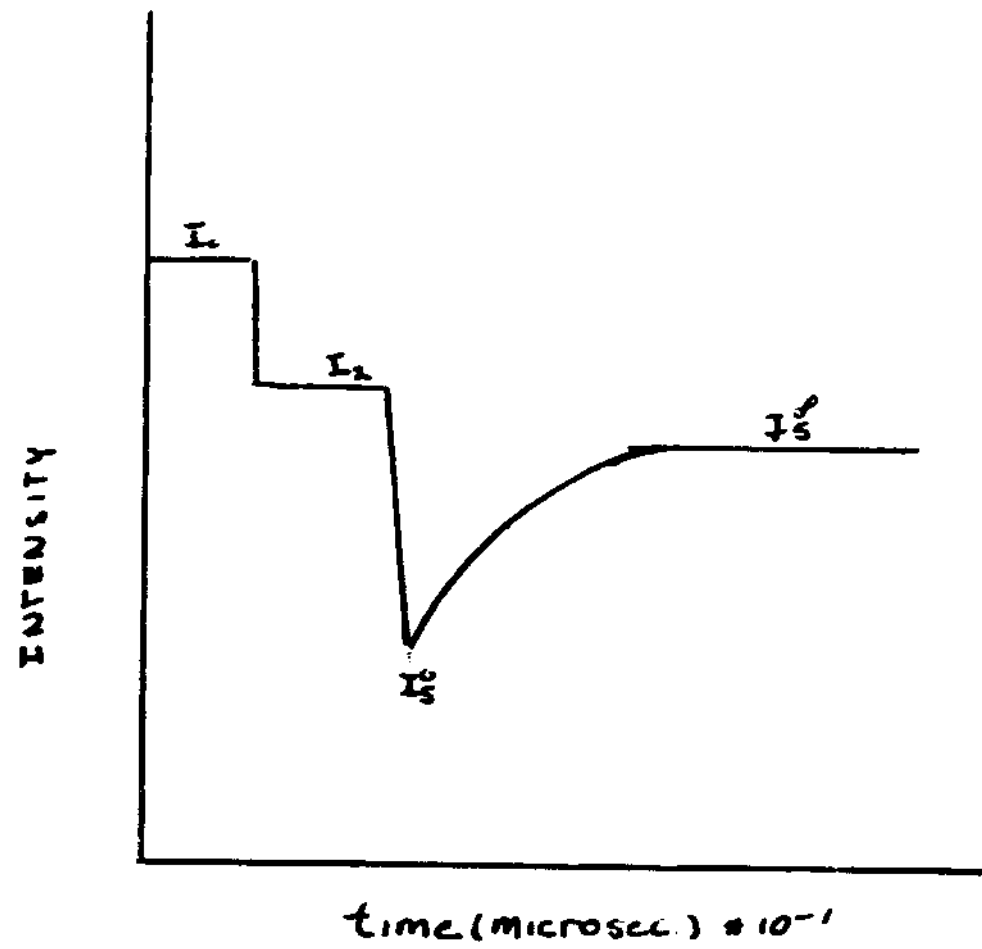


Figure 2. Graphs of intensity versus time. a) is with the reflected endplate and b) is with the incident endplate. I_0 is the incident intensity. I_2 (in b only) is the intensity after only the reflected wave has passed. I_5^0 is after both waves have passed but before any reaction has occurred and I_5^{∞} is after both waves have passed and at equilibrium. (Graph drawn without noise for simplicity.)



a



b

immediately after the incident and reflected waves have passed but before any CSe has been produced. And I_5^∞ is the light intensity after both waves have passed and the reaction has gone to completion or equilibrium. These graphs are drawn without noise for simplicity.

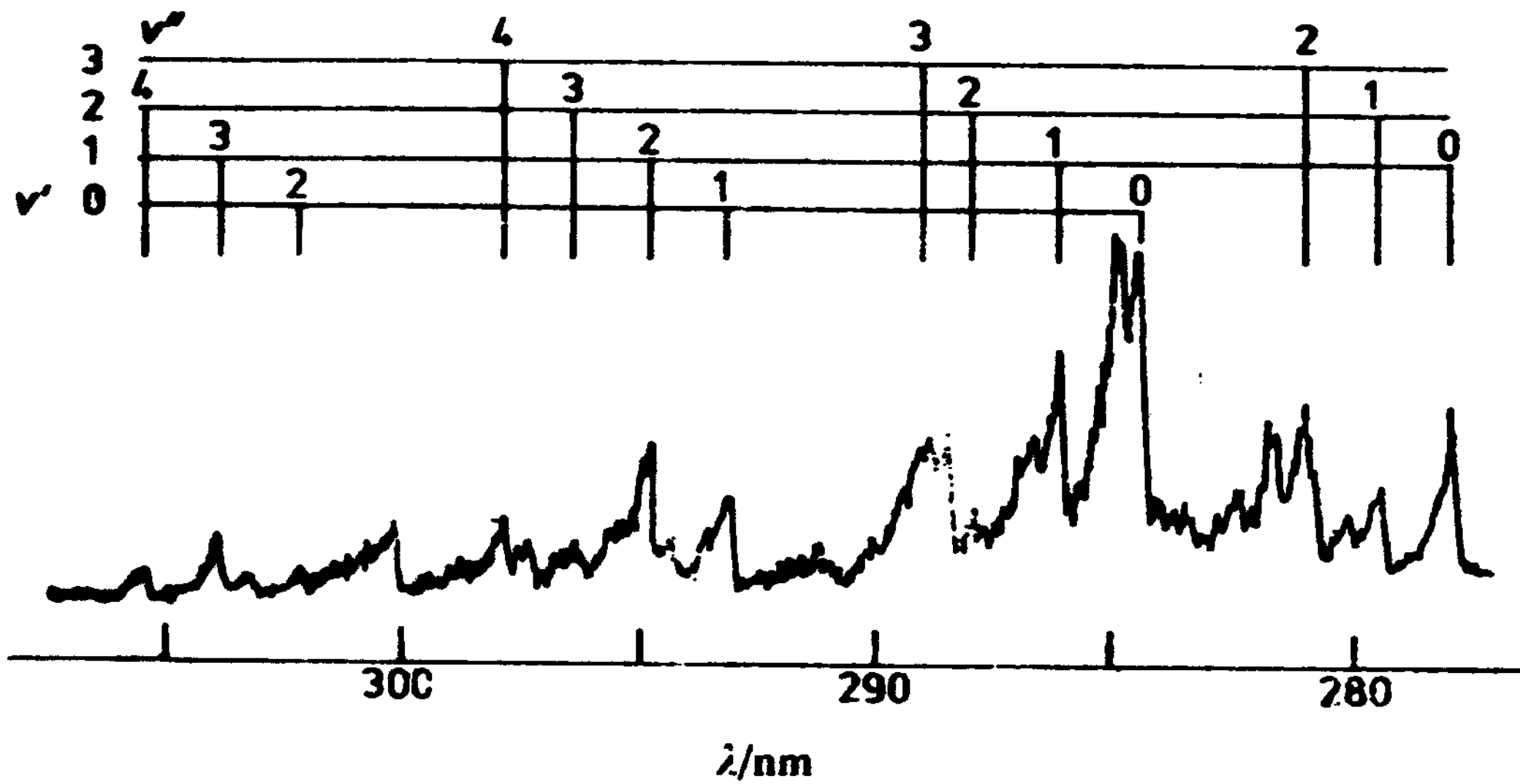
Alpha (see Appendix A) is a program written by J. R. Marquart to calculate the temperature, pressure, velocity, and density after the incident (T_2, P_2, V_2, ρ_2) and reflected (T_3, P_3, V_3, ρ_3) shock waves have each passed. These were calculated from the initial conditions (T_0, P_0, V_0, ρ_0) and the time required for the shock wave to pass two piezoelectric transducers spaced a known distance apart. The program was then modified to include calculations of the absorption coefficients ($\alpha_2, \alpha_3^0, \alpha_3^\infty$) and the fraction of CSe_2 converted to CSe, F , at infinite time (attainment of equilibrium) [8].

It should be mentioned that these calculations were done assuming ideal gas conditions. Boundary layer corrections or effects due to deceleration of the shock wave were not included. This does not effect data significantly when using the reflected endplate since deviations from ideal conditions are almost zero right after the incident wave has passed. The boundary corrections are significant when the incident endplate is in the shock tube since its reflected wave is delayed. The effect of these corrections on temperature is to increase T_2 by approximately eight degrees and to decrease T_3 by approximately two degrees.

Determination of the Rate Constant and Arrhenious Equation

The emission spectrum of CSe produced through photodissociation of CSe_2 at 147nm (see Figure 3) shows CSe's most intense transition is from the excited $V' = 0$ to the $V'' = 0$ ground state ($D \rightarrow X$) and occurs at

Figure 3. CSe (D \rightarrow X) emission spectrum excited through photodissociation of CSe₂ at 147.0nm sample pressure, 0.05 torr, resolution 0.025nm [7].



approximately 2840 ang. [7]. For this part of the experiment, absorption measurements were made at 2846 ang. and a temperature range from 1700 K to 2200 K.

At 2846 ang., the absorption coefficients, α_2 , α_5^0 , α_5^∞ given in cm^2/mole , were found using the Beer-Lambert relationship:

$$\alpha_a^b = \ln(I_0/I_a^b) / B \rho_a X_{\text{CSe}_2} \quad (1)$$

where $B = 5.0165$ cm is the path length, ρ_a is the total gas density in moles/cc and X_{CSe_2} is the mole fraction of CSe_2 in argon, the diluent gas. This equation would give the absorption coefficient of CSe if all CSe_2 were pyrolyzed or if CSe_2 is none were pyrolyzed. From studies on CSe_2 , Marquart and Belford have found that at room temperature CSe_2 absorbs mainly at 2308 ang. [8]. At high temperatures, the CSe_2 absorption peak at 2308 ang. is broadened due to the formation of hot bands. This results in significant absorption at higher wavelengths. Thus in the temperature range, 1700 K to 2200 K, the absorption coefficient is equal to:

$$\alpha_a^b = (F)\alpha_{\text{CSe}} + (1 - F)\alpha_{\text{CSe}_2} \quad (2)$$

where α_{CSe_2} and α_{CSe} are the absorption coefficients for CSe_2 and CSe respectively, and F is the fraction of CSe_2 converted to CSe .

After the incident shock wave has passed the temperature is around 1000 K. At temperatures less than 1200 K the amount of CSe produced is negligible (less than 0.1%), and the absorption coefficient calculated at T_2 is equal to the absorption coefficient of CSe_2

$$\alpha_2 = \alpha_{\text{CSe}_2} \quad T < 1200 \text{ K} .$$

After the incident and reflected waves have passed but before any reaction has occurred, the absorption coefficient is equal to the CSe_2 absorption coefficient at T_5 .

At temperatures greater than 2300 K only a negligible amount of CSe_2 exists (less than 0.8%) after a few microseconds. At these temperatures, the measured absorption coefficient is equal to the CSe absorption coefficient:

$$\alpha_5^{\infty} = \alpha_{\text{CSe}} \quad T > 2300 \text{ K} .$$

Since most of the data collected was at temperatures between 1700 K and 2200 K it was necessary to determine the fraction of CSe_2 converted as a function of temperature.

At any time during the dissociation process the amount of CSe plus CSe_2 is a constant. Thus after both waves have passed

$$\rho_5 \times \text{CSe}_2 = [\text{CSe}] + [\text{CSe}_2] \quad (3)$$

where ρ_5 is the density in moles/cc at T_5 . The fraction of CSe produced is

$$F = [\text{CSe}] / \rho_5 \times \text{CSe}_2 \quad (4)$$

The equilibrium constant (K_p) in $\text{cm}^3/\text{mole}/\text{sec}$ for the reaction is

$$K_p = [\text{CSe}]^2 RT_5 / [\text{CSe}_2] \quad (5)$$

where R is the gas constant. Solving (1) for the concentration of CSe_2 , substituting into (3), solving for [CSe], and then plugging into (2), a relationship between the fraction and the equilibrium constant and temperature can be obtained:

$$F = \frac{1}{2} \left\{ -\frac{K_p}{RT} \pm \left[\left(\frac{K_p}{RT} \right)^2 + \frac{4K_p}{RT} \rho_5 \alpha \text{CSe}_2 \right]^{1/2} / \rho_5 \alpha \text{CSe}_2 \right\} . \quad (6)$$

The equilibrium constant in atmospheres as a function of temperature (Kelvin) has previously been determined by Marquart and Belford [8],

$$\log_{10} K_p = -.705 \log_{10} (T_5) + 7.64 \times 10^{-5} (T_5) \\ + \frac{1.19 \times 10^4}{(T_5)^2} - \frac{1.85 \times 10^4}{T_5} + 9.68 .$$

Therefore, the fraction converted can now be determined as a function of temperature. The results of such calculations for each temperature is given in Table 1.

A plot of αCSe_2 versus temperature (see Figure 4) shows an exponential dependence of the CSe_2 absorption coefficient with temperature up to approximately 2200 K. Above this temperature the response of our electronic system was too slow to keep up with the fast change in intensity. This resulted in an experimental absorption coefficient that is probably lower than the true value. In the temperature range between 1700 K and 2200 K the exponential dependence can be approximated by a linear dependence without introducing too much error. The equation found is:

$$\alpha \text{CSe}_2 = 7143 T - 6.875 \times 10^6 . \quad (7)$$

Substituting (7) and (6) into (2) we can determine αCSe . The results are shown in Table 1. One can see that some of the values for the absorption coefficient were negative. This was due to some unexplained light emission found at high temperatures and at the larger of the two CSe_2 mole fractions used. As a result, these values were ignored. Upon inspecting the other

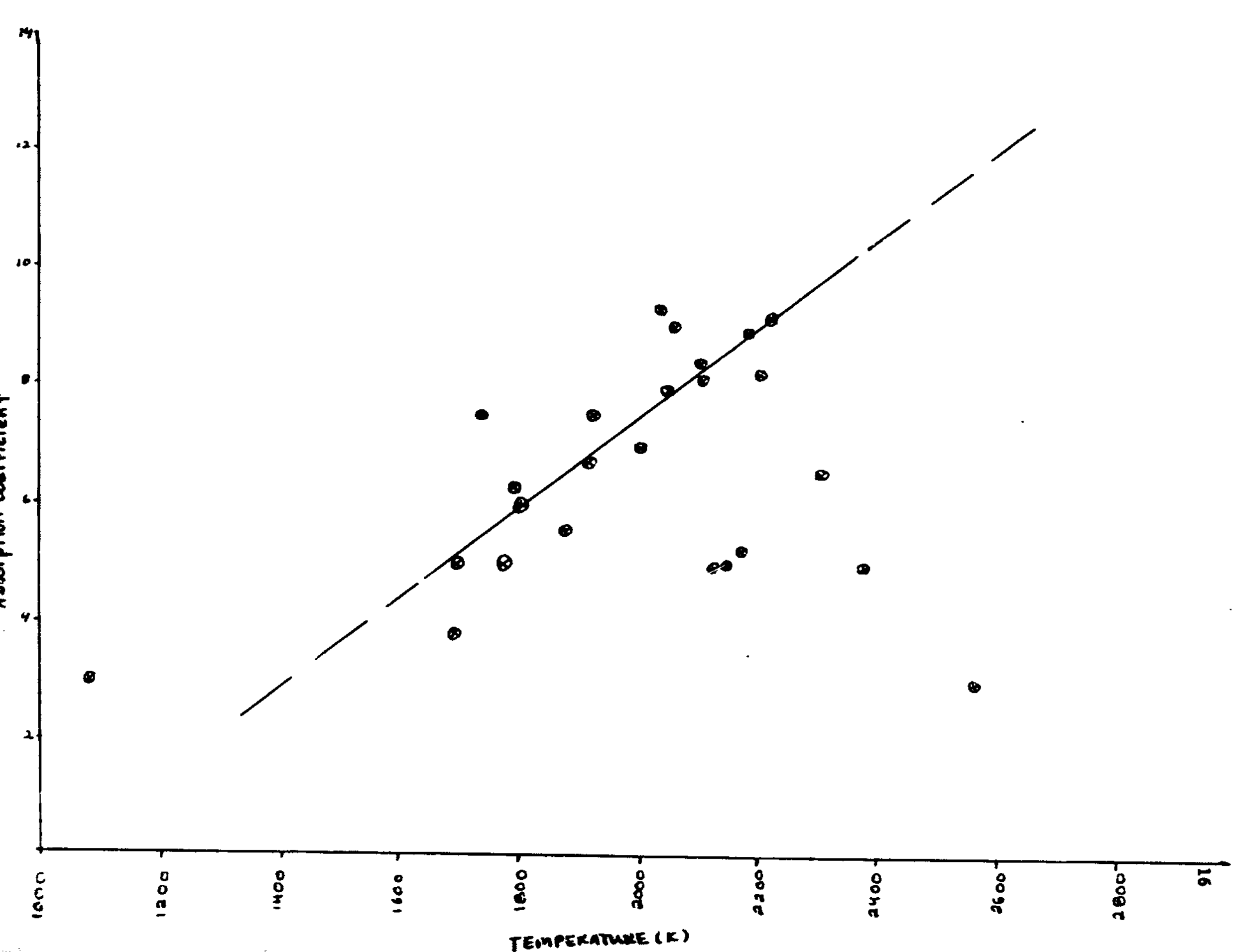
Table 1. Absorption Data

Endplate	SLIT	T ₂	T ₅	F	α ₂	α ₅ [∞]	α ₅ ⁰ = αCSe ₂
Reflected	1/2	1187.8	2379.1	0.978	0	4.6837 × 10 ⁵	5.0567 × 10 ⁵
Reflected	1/2	1123.3	2224.4	0.936	0	7.5656 × 10 ⁵	9.0133 × 10 ⁵
Reflected	1/2	1051.6	2053.0	0.799	0	2.2774 × 10 ⁶	9.1349 × 10 ⁵
Reflected	1/2	1075.2	2109.7	0.860	0	1.6604 × 10 ⁶	8.4085 × 10 ⁵
Reflected	1/2	1059.2	2072.3	0.822	0	2.4253 × 10 ⁶	9.3516 × 10 ⁵
Reflected	1/2	1113.1	2201.1	0.925	0	1.5098 × 10 ⁶	8.3503 × 10 ⁵
Reflected	1/2	1083.1	2129.2	0.877	0	2.1576 × 10 ⁶	8.6432 × 10 ⁵
Reflected	1/2	1059.1	2069.6	0.820	0	3.3486 × 10 ⁶	8.1036 × 10 ⁵
Reflected	1/2	1108.7	2188.2	0.918	0	2.0587 × 10 ⁶	9.1921 × 10 ⁵
Reflected	1/2	976.2	1871.3	0.502	0	3.5159 × 10 ⁶	5.5943 × 10 ⁵
Reflected	1/2	936.8	1780.9	0.254	0	7.6358 × 10 ⁵	6.2899 × 10 ⁵
Reflected	1/2	962.4	1842.0	0.343	0	4.7627 × 10 ⁵	5.9997 × 10 ⁵
Reflected	1/2	1119.9	2218.4	0.905	0	9.2699 × 10 ⁴	3.3747 × 10 ⁵
Reflected	1/2	900.9	1695.1	0.139	0	1.0601 × 10 ⁶	3.8493 × 10 ⁵
Reflected	1/2	1092.4	2148.9	0.872	0	1.9672 × 10 ⁵	4.7105 × 10 ⁵
Reflected	1/2	1079.4	2118.0	0.785	0	2.3008 × 10 ⁵	4.6812 × 10 ⁵
Reflected	1/2	1142.4	2268.6	0.953	0	1.8994 × 10 ⁶	6.4389 × 10 ⁵
Incident	1	930.5	1764.3	0.582	4.8811 × 10 ⁵	8.7897 × 10 ⁶	5.1155 × 10 ⁵
Incident	1	989.7	1905.7	0.835	1.2709 × 10 ⁶	4.0683 × 10 ⁶	7.9786 × 10 ⁵
Incident	1	917.2	1734.8	0.516	1.0308 × 10 ⁶	7.3282 × 10 ⁶	7.5238 × 10 ⁵
Incident	1	1050.1	2052.4	0.950	1.5052 × 10 ⁶	3.7631 × 10 ⁶	7.9507 × 10 ⁵
Incident	1	1028.9	2001.9	0.924	2.9680 × 10 ⁵	2.3250 × 10 ⁶	7.1048 × 10 ⁵
Incident	1/2	647.1	1089.9	.001	3.1306 × 10 ⁵	9.6848 × 10 ⁸	3.0694 × 10 ⁵
Incident	1/2	1266.2	2567.5	0.999	3.1306 × 10 ⁶	4.2262 × 10 ⁵	3.1843 × 10 ⁵
Reflected	1/2	1206.3	2425.6	0.997	0	2.4148 × 10 ⁶	0
Reflected	1/2	1089.9	2146.8	0.977	0	2.7861 × 10 ⁶	4.8205 × 10 ⁵
Reflected	1	989.1	1903.8	0.832	0	2.8081 × 10 ⁶	6.6902 × 10 ⁵

Table 1. Absorption Data

SLIT	T_2	T_5	F	α_2	α_5^∞	$\alpha_5^0 = \alpha CSe_2$	αCSe
1/2	1187.8	2379.1	0.978	0	4.6837×10^5	5.0567×10^6	2.353×10^5
1/2	1123.3	2224.4	0.936	0	7.5656×10^5	9.0133×10^6	1.399×10^5
1/2	1051.6	2053.0	0.799	0	2.2774×10^6	9.1349×10^6	3.17×10^5
1/2	1075.2	2109.7	0.860	0	1.6604×10^6	8.4085×10^6	3.27×10^5
1/2	1059.2	2072.3	0.822	0	2.4253×10^6	9.3516×10^6	7.01×10^5
1/2	1113.1	2201.1	0.925	0	1.5098×10^6	8.3503×10^6	0.788×10^6
1/2	1083.1	2129.2	0.877	0	2.1576×10^6	8.6432×10^6	9.876×10^5
1/2	1059.1	2069.6	0.820	0	3.3486×10^6	8.1036×10^6	1.60×10^6
1/2	1108.7	2188.2	0.918	0	2.0587×10^6	9.1921×10^6	1.277×10^6
1/2	976.2	1871.3	0.502	0	3.5159×10^6	5.5943×10^6	-2.953×10^6
1/2	936.8	1780.9	0.254	0	7.6358×10^5	6.2899×10^5	-1.65×10^7
1/2	962.4	1842.0	0.343	0	4.7627×10^5	5.9997×10^5	0.491×10^6
1/2	1119.9	2218.4	0.905	0	9.2699×10^4	3.3747×10^5	-8.46×10^5
1/2	900.9	1695.1	0.139	0	1.0601×10^6	3.8493×10^5	-3.1369×10^7
1/2	1092.4	2148.9	0.872	0	1.9672×10^5	4.7105×10^5	-1.045×10^6
1/2	1079.4	2118.0	0.785	0	2.3008×10^5	4.6812×10^5	-2.037×10^6
1/2	1142.4	2268.6	0.953	0	1.8994×10^6	6.4389×10^6	1.438×10^6
1	930.5	1764.3	0.582	4.8811×10^5	8.7897×10^6	5.1155×10^6	4.66×10^6
1	989.7	1905.7	0.835	1.2709×10^6	4.0683×10^6	7.9786×10^6	2.73×10^6
1	917.2	1734.8	0.516	1.0308×10^6	7.3282×10^6	7.5238×10^6	2.14×10^6
1	1050.1	2052.4	0.950	1.5052×10^6	3.7631×10^6	7.9507×10^6	3.35×10^6
1	1028.9	2001.9	0.924	2.9680×10^5	2.3250×10^6	7.1048×10^6	1.71×10^6
1/2	647.1	1089.9	.001	3.1306×10^5	9.6848×10^8	3.0694×10^6	-1.27×10^8
1/2	1266.2	2567.5	0.999	3.1306×10^6	4.2262×10^5	3.1843×10^6	4.08×10^5
1/2	1206.3	2425.6	0.997	0	2.4148×10^6	0	2.38×10^6
1/2	1089.9	2146.8	0.977	0	2.7861×10^6	4.8205×10^6	2.588×10^6
1	989.1	1903.8	0.832	0	2.8081×10^6	6.6902×10^6	1.45×10^6

Figure 4. Absorption coefficient of CSe_2 versus temperature.



values obtained, CSe doesn't show any distinct temperature dependence in this temperature range. Therefore a statistical average for the absorption coefficient was calculated, giving a mean = $1.76 \times 10^{+6} \text{ cm}^2/\text{mol}$ and a standard deviation of $1.24 \times 10^{+6} \text{ cm}^2/\text{mol}$.

Using the calculation of the absorption coefficient and other necessary data, the rate constant was evaluated.

The disappearance of $[\text{CSe}_2]$ can be described by the following differential equation:

$$\frac{\partial[\text{CSe}_2]}{\partial t} = -K(t)[\text{Ar}]\{[\text{CSe}_2] - [\text{CSe}_2]_{\text{eq}}\} \quad (8)$$

where $[\text{CSe}_2]_{\text{eq}}$ is the equilibrium concentration of CSe_2 . The fraction of CSe_2 not converted to CSe is:

$$1 - F = [\text{CSe}_2]_{\text{eq}}/\rho_5 \times \text{CSe}_2 \quad (9)$$

The solution to the differential equation in (8) is:

$$[\text{CSe}_2] - (1 - F)\rho_5 \times \text{CSe}_2 = B \exp(-K[\text{Ar}]t) \quad (11)$$

At $T = 0$,

$$B = [\text{CSe}_2]_0 - [\text{CSe}_2]_{\text{eq}} \quad (12)$$

Substituting (12) into (11) and solving for $[\text{CSe}_2]$ gives:

$$[\text{CSe}_2] = F[\text{CSe}_2]_0 \exp(-K[\text{Ar}]t) + (1 - F)[\text{CSe}_2]_0 \quad (13)$$

From Beer's law the absorption from the gas is

$$A = [[\text{CSe}] \times \text{CSe} + [\text{CSe}_2] \times \text{CSe}_2]B \quad (14)$$

and

$$[CSe] = [CSe_2]_0 - [CSe_2] . \quad (15)$$

Substituting (15) into (14), the absorpance is given as:

$$A = B[CSe_2]_0 \alpha CSe + (\alpha CSe_2 - \alpha CSe) B [CSe_2] . \quad (16)$$

Substituting the concentration of CSe_2 from (13) into (16) and setting all terms not dependent of time equal to A_5^{∞} (16) becomes:

$$A - A^{\infty} = (\alpha CSe_2 - \alpha CSe) BF [CSe_2]_0 \exp(-K[Ar]t) .$$

Since the concentration of argon is constant and approximately equal to the density at T_5 , $K(t)$ becomes a psuedo-first order rate constant. Solving for $K(t)$ in the above equation gives:

$$K(t) = \frac{-\ln(A - A^{\infty}) - \ln[(\alpha CSe_2 - \alpha CSe) BF \rho_5 \chi CSe_2]}{\rho_5 \cdot t} . \quad (17)$$

Equation (17) and six of the graphs of intensity versus time were used to determine $K(t)$ as a function of time during the reaction. The results are shown in Table 2. The values observed show that the rate constant remains constant within experimental error for the entire reaction. This supports the theory that the reaction proceeds via the Belford and Marquart one step mechanism [9] and not the Graziano two step mechanism [11]. If the latter were true, as the reaction proceeded, a decrease in the rate constant would be observed.

The value obtained for the rate constant should be considered a lower limit to the true value, since the data was not deconvoluted for time delay effects and it was not corrected for the boundary layer effects. A true

Table 2. Rate Constants (mole/cm³/sec)

Time μsec	Temp Kelvin	Run 1	Run 2	Run 3	Run 7	Run 5	Run 6
		1905.7	2001.9	2002.4	2053.0*	2072.3*	2109.7*
38.46					3.11×10^8	3.96×10^8	8.33×10^8
64		2.92×10^8	2.37×10^8	3.42×10^8			
76.7					5.58×10^9	5.02×10^2	8.83×10^8
96		2.69×10^8	2.06×10^8	3.01×10^8			
115.4					12.07×10^2	6.75×10^8	7.78×10^8
128		2.95×10^8	2.01×10^8	2.76×10^8			
160		2.93×10^8	2.08×10^8	2.68×10^8			
192		--	2.37×10^8	2.71×10^8			
224		--	2.29×10^8				
Average K(t)		2.87×10^8	2.20×10^8	2.92×10^8	6.92×10^8	5.24×10^8	8.31×10^8

*These runs were done at a higher concentration of CSe₂ and the light intensity data resulted in some unexplained emission which increased the observed K value making it too high.

graph of intensity versus time (see Figure 5) would probably descend deeper after the reflected shock has passed, and then as the reaction proceeds, it should probably rise faster. Equation (17) shows that both these con-
volution effects should cause a decrease in the value of the rate constant calculated, especially at higher temperatures. Boundary layer corrections would decrease αCSe_2 but not αCSe since it is apparently independent of small temperature changes. Again by (17) the resulting rate constant would decrease. These effects could be partially offset by the rather poor signal to noise ratio (approximately 5 for I_2) and a large monochromator slit width. Since the intensity of the dip was measured at the lowest possible point on the graph, any noise in the dip would cause the intensity to be smaller than the true value. Thus the calculated rate constant would be slightly larger than the actual value. A large slit width would also effect the rate constant. At 2846 ang., as at 2308 ang. in Figure 6, the absorption coefficient would be lowered by an increase in slit width. However since CSe_2 absorbance is due to hot bands, it varies slowly with wavelength, while αCSe probably varies more with wavelength in this range, so the lowering of the αCSe would be greater than that of αCSe_2 . As a consequence the rate constant would decrease slightly.

Since so few points were obtained for the rate constant at different temperatures we could not accurately determine the arrhenius equation and thus the activation energy. Graziano determined the activation energy from an arrhenius plot of $\ln(K)$ vs. $1/T$ and it is shown in Figure 7 [11]. We added our data points to these (the circled X's in Figure 7). These points show good agreement with his low temperature points.

Figure 5. Graph of light intensity versus time. The dotted line shows the possible true value and the solid line shows the experimental values.

ØC288208

TEMP. = 1905.7

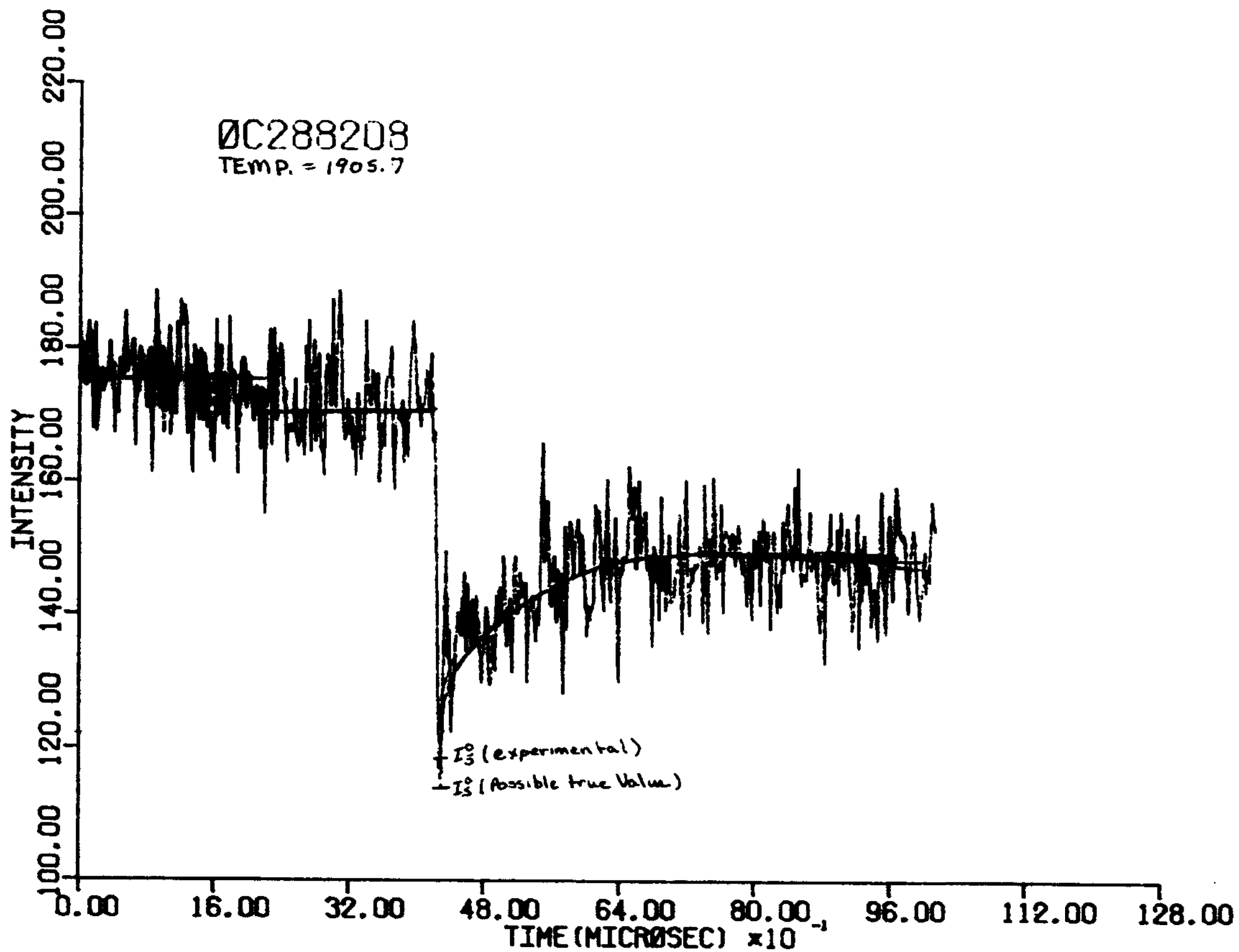


Figure 6. Absorption coefficient of CSe_2 versus slit width at room temperature and 2308 ang.

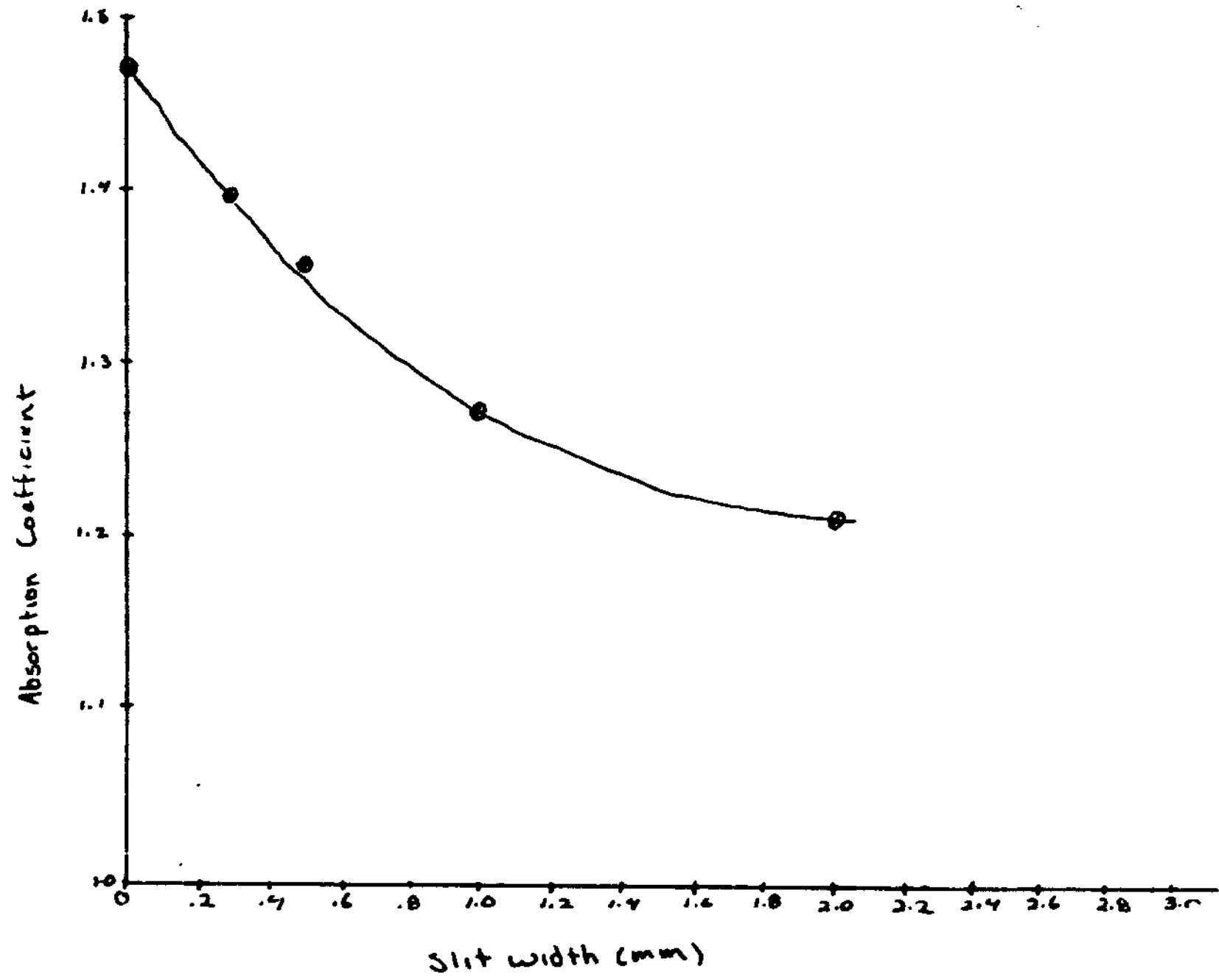
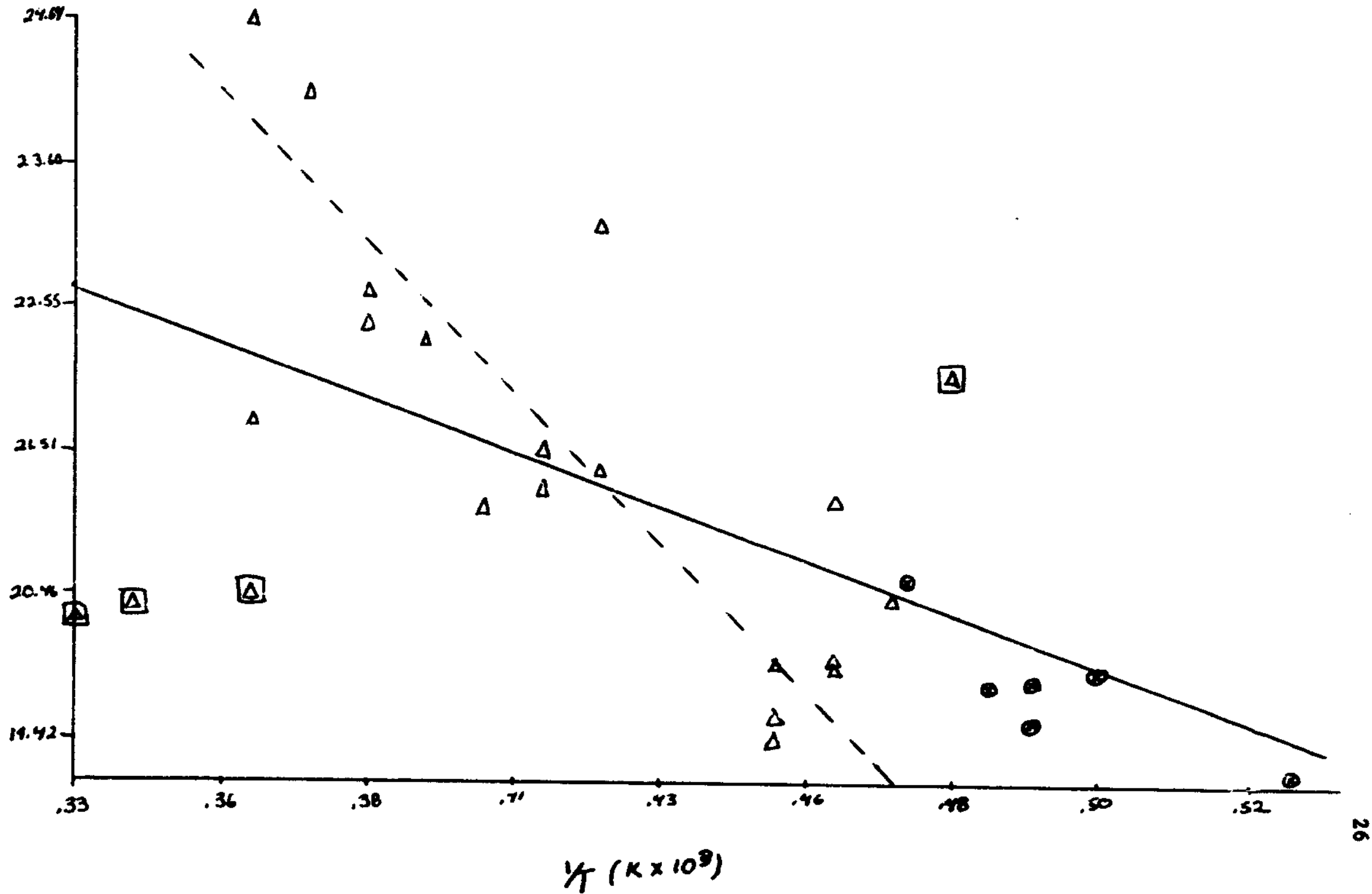


Figure 7. Natural log of the rate constant versus inverse temperature.
 Δ are Graziano's points, \odot are our added points and \square are
Graziano's points excluded from our least squares fit.

(—) Least squares fit including four pts with squares around them.
 (---) Least squares fit without four points



Graziano did a least squares fit on his data points to extract the activation energy and obtained an $E_a = 30.9$ kcal/mole. After eliminating four of his points (Δ) we obtained an activation energy of $E_a = 62.2$ kcal/mole. This value is in better agreement with a linear plot of the enthalpy of dissociation versus activation energy for CO_2 , OCS , NO_2 , CS_2 and CSe_2 (see Figure 7). This plot was obtained by Belford and Marquart [9].

Absorption Spectrum for CSe

This section of the paper covers the technique used to find the absorption spectrum for CSe , the problems encountered, and possible alterations in experimental technique to alleviate these problems.

The data are obtained from 2700 ang. to 2900 ang. at temperatures ranging from 1600 K to 2500 K. The absorption coefficient was obtained as before using Beer's law for each wavelength.

DATAFIT (see Appendix B), a nonlinear least squares program, was used to fit a trial equation with an exponential dependence of the CSe_2 absorption coefficient on wavelength and $1/T_2$. The resulting equation is

$$\alpha_{\text{CSe}_2} = (.253 \times 10^9 - .170 \times 10^6 T_2) \exp(-1.36 \lambda/T) .$$

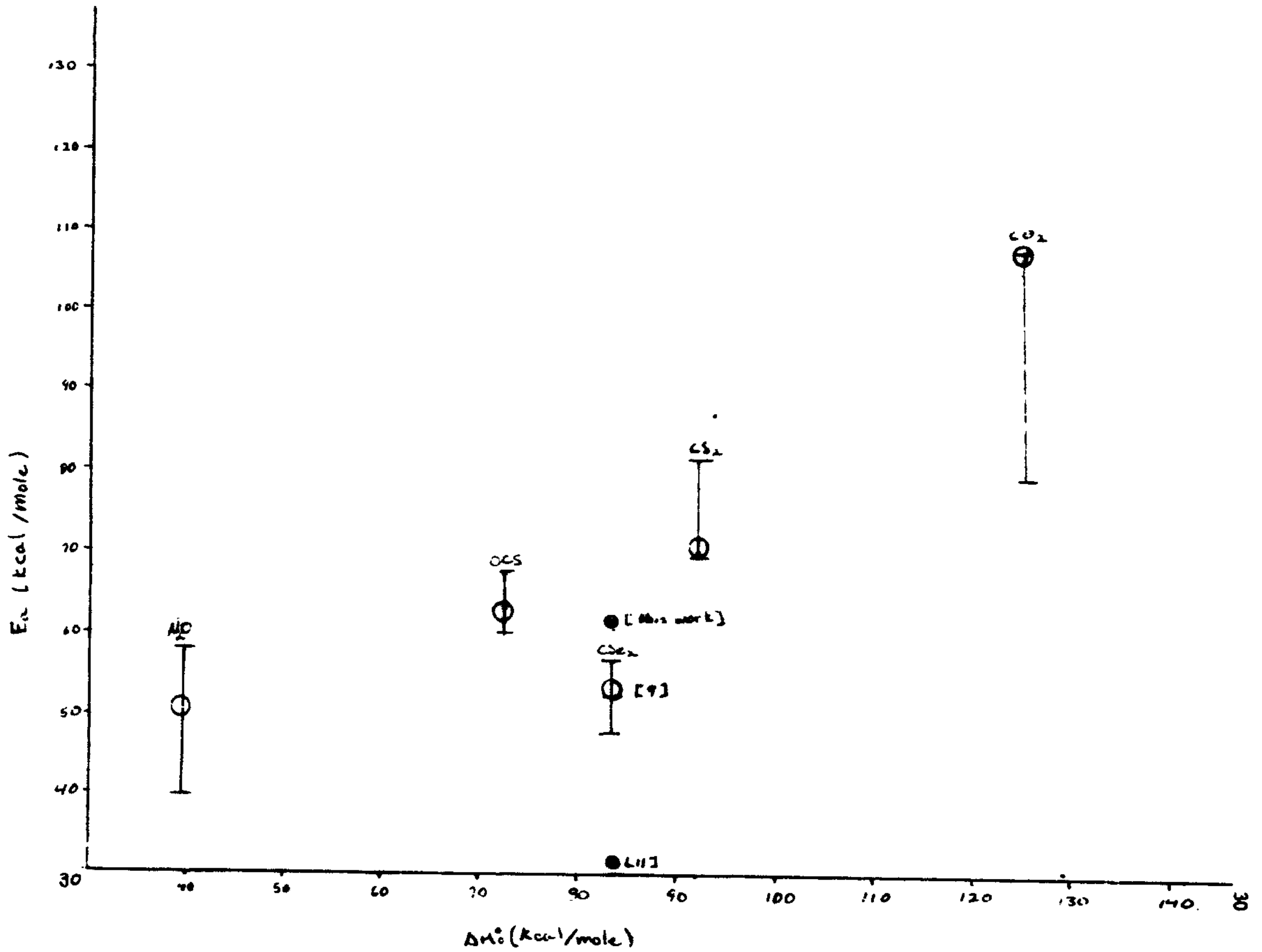
This equation was used to calculate the absorption coefficient of CSe_2 at T_3 , however at 2000 K and 2846 ang., the calculated absorption coefficient is $-1.25\text{E}07$, an absurd negative result. It was concluded that this was not a good way of determining the absorption coefficient at T_3 . It was later found in the calculation of the rate constant that the absorption coefficient for CSe_2 is a function of $\exp(T)$ and not $\exp(-1/T)$.

Another problem arose from the light source. Since the experiment was over in a few microseconds, it was necessary to open the slits to 1/2 or even 1 millimeter width. This resulted in poor resolution. Figure 8 shows the dependence of slit width on the absorption coefficient at 2308 ang. In the future the deuterium lamp should be replaced with a more intense light source to improve the signal to noise and allow for the use of narrower slits. (A DCL lamp produces light of greater intensity.)

Finally to determine the spectrum it would be best to be able to scan a number of wavelengths at a constant temperature. This is a problem which has been partially eliminated as we have become more accustomed to the shock tube and better able to reproduce shock conditions.

The modifications suggested above should make it possible to obtain an absorption spectrum in the future.

Figure 8. Enthalpy of dissociation versus activation energy for CO_2 , CS_2 , OCS , N_2O and Cl_2 . The vertical bars represent the reported activation energies. The circles are typical of recently reported activation energies [9].



APPENDIX A
PROGRAM ALPHA

000000

70

700

705

71

72

73

C

110

105

200

210

```

PROGRAM WRITTEN JANUARY, 1983 BY J. R. MARSHART FOR THE
2 INCH SHOCK TUBE AT UNIV. OF ILLINOIS.
PROGRAM CALCULATES P2, P2, U2, T2, P2, U2 FOR A REAL GAS
CONSISTING OF NEARLY PURE ARGON. IT MAKES CORRECTIONS FOR
BOUNDARY LAYER EFFECTS.
IN THIS VERSION X2 = 1.67E-07
REAL KP, ID, I2, T5, I25
CHARACTER*40 CHAR
OPEN(UNIT = 1, TYPE = 'OLD', FORM = 'FORMATTED')
OPEN(UNIT = 2, TYPE = 'NEW', FORM = 'FORMATTED')
D23=20.3015
D34=20.3070
D4W=10.300
D4R=0.29
D4T=10.14
D1A = 2.0165
R1=8.5143E7
R2=8.0575
READ(1,70) NPIS
FORMAT(12)
DO 321 I = 1, NPIS
READ(1,704) CHAR
FORMAT(A30)
WRITE(2,705) CHAR
FORMAT(X,10(' '),A30,' ',10(' '))
READ(1,71) T1, P1
FORMAT(F8.1,X,F5.1)
T1=T1+273.2
P1=P1/760
READ(1,72) T23, T34
FORMAT(2(F5.1,X))
READ(1,73) TEP
FORMAT(11)
THIS PART CALCULATES SHOCK VELOCITY AT END PLATE AND MACH
NUMBER
D4FPE=0.1
IF(IFPE.EQ.1) D4FPE=D4R
U23=UEA*D23/T23
U34=UEA*D34/T34
DECEL=(U23-U34)/(D23+D34)/2
UE=U34-UECFL*(D34/2+D4W+D4FPE)
U1=UE
MH=39.948
AE=SQRT((1.6067*K1*T1)/39.948)
AMACH=UE/A
WRITE(2,110) AMACH
FORMAT(2X,'MACH = ',F7.4)
D1=P1/(R2*T1)
WRITE(2,105) I1, P1, U1
FORMAT(2X,'I1=',F5.1,4X,'P1=',F9.5,4X,'U1=',1PF12.0)
THIS PART SEARCHES FOR P2, P2, U2
T2=T1+50.0
CONTINUE
H1=5.2033F6*I1-1.5515F4
CONTINUE
H2=5.2033F6*I2-1.5515F4
H2=SQRT((H1-H2)*2+U1**2)
T2=CALC=I1*U2/U1+39.948*U2*H1/K1-39.948*U2**2/K1
IF((T2-CALC-T2).LT.0.1)GO TO 250
T2=T2+0.1
GO TO 210

```



```

250 CONTINUE
    D2=U1*111/UP
    V2=U1-112
    P2=U2+R2+12
    WRITE(2,255)
    FORMAT(2X, 'CONTINUUM BOUNDARY LAYER CORRECTIONS:')
255 WRITE(2,257)
    FORMAT(2X, '*****')
257 WRITE(2,260) 12,P2,UP
    FORMAT(2X, '12= ',F6.1,3X, 'P2= ',F10.5,4X, 'D2= ',1PE12.4)
260 WRITE(2,270) U1,112,V2
    FORMAT(2X, 'U1= ',1PE10.4,4X, 'UP= ',1PE10.4,4X, 'V2= ',1PE10.4)
270 C THIS PART SEARCHES FOR I5, P5, U5
    TS=12+50.0
300 CONTINUE
    H15=5.2073F6+15-1.5515F9
    I15=(H15-H12)/V2-V2/2
    U5=U5+V2
    TS=CALC=12*I15/W2+39.940*W2*I15/K1-39.940*U5**2/W1
    IF (ABS(CALC-TS).LE.0.1) GO TO 350
    TS=TS+0.1
    GO TO 300
350 CONTINUE
    DS=U2*42/U5
    P5=U5+R2+15
    WRITE(2,400) 15,P5,U5
    FORMAT(2X, '15= ',F6.1,3X, 'P5= ',F10.5,4X, 'DS= ',1PE12.4)
400 WRITE(2,450) U5,W2
    FORMAT(2X, 'U5= ',1PE10.4,4X, 'W2= ',1PE10.4)
450 C THIS PART OF THE PROGRAM CALC. ROUGHLY WHERE THE
    REF. WAVES SHOULD APPEAR.
    IF (1EP.EQ.0) THEN
    TIME1 = 10.300/U1
    TIME2 = TIME1 + DWT/I15 + DWI/U1
    WRITE(2,5)
    FORMAT(1, ' THE REF. AND INC. WAVES PASS THE WINDOW AT:'
5 WRITE(2,6) TIME1, TIME2
    FORMAT(1, ' TIME1= ',F9.4, ' TIME2= ', F9.4)
6 ENDIF
    C THIS PART OF THE PROGRAM IS DESIGNED TO CALCULATE THE
    C ABSORPTION COEF. FOR USE AND CSE2 USING BFER'S LAW AND
    C DATA ABOVE W/OUT BOUNDARY LAYER CONDITIONS. BASE F
    X2 = 1.67E-03
    RCX = 5.0165
    R = R2.06
    C2 = U2*X2
    READ(1,70) T0, T2, IS
70 C CALCULATE THE FRACT. OF CSE2 CONVERTED TO CSF (F)
    AN = -0.705*ALOG10(T5) + 7.64E-05*15 + 1.14E+04/15**2
    RN = -1.25E+04/15 + 9.60
    AN = AN + RN
    KP = 10**AN
    G = KP/(R*T5)
    CS = 0.5*(-R + SQRT(G**2 + 4*G*DS*X2))
    CS = X2*DS
    F = CS/(U5*X2)
    WRITE(2,8) F
    FORMAT(2X, 'THE FRACTION OF CSE2 CONVERTED TO CSF IS = ',F1.
8 C CALC. THE ABSORB. COEF. (A2 AND A5)
    READ(1,70) WAVE NU

```

```

76  FORMAT(F6.1)
    IF (WAVENU .EQ. 2500.0) THEN
    C1 = X2 * D1
    T5 = (1-F)*U5*Y2
    T1 = 10
    READ (1,*) SLIT
      IF (SLIT .EQ. 1.0) THEN
        A1 = 1.27E 04
      ELSEIF (SLIT .EQ. 0.5) THEN
        A1 = 1.355E 04
      ELSE IF (SLIT .EQ. 0.3) THEN
        A1 = 1.395E 06
      ELSE
        WRITE(2,6)
6      FORMAT(' VALUES FOR THAT SLIT WIDTH HAVEN'T BEEN CALC.')
        ENDIF
    A2 = (ALOG(I1/T2)/RCM + A1*C1)/U2
    A5 = (ALOG(I1/T5)/RCM + A1*C1)/U5
    C  THE CALC FOR THE OLD VALUE OF A2 AT T2
    AOLD = 2.230E 06 + 7.167E 10/T2 - 1.083E 13/T2**2
    7  WRITE(2,7) AOLD
    FORMAT(' ', 'THE OLD VALUE FOR A2 AT T2 IS ', 1PE12.4)
    ELSE
    A2 = (ALOG(I0/T2))/(RCM*C2)
    A5 = (ALOG(I0/T5))/(RCM*C5)
    ENDIF
    IF (F .LT. 10) THEN
    AOLD5 = 2.230E 06 + 7.167E 10/T5 - 1.083E 13/T5**2
    8  WRITE(2,8) AOLD5
    FORMAT(' ', 'THE OLD VALUE FOR A2 AT T5 IS ', 1PE12.4)
    END IF
    FACTOR = 1
    IF (SLIT .EQ. .3) FACTOR = .867
    IF (SLIT .EQ. .5) FACTOR = .693
    IF (SLIT .EQ. 1.0) FACTOR = .453
    A2 = A2 + FACTOR
    A5 = A5 + FACTOR
    9  WRITE(2,9)
    FORMAT(' ', 'THE ARSNKR. COEF. IS GIVEN WITH A 2MM SLIT IF A1 230)
    WRITE(2,11) A2
    WRITE(2,11) A5
    11  FORMAT(2X, 'THE ARSNKR. COEF. FOR USE? IS ', 1PE12.4, '/')
    12  FORMAT(2X, 'THE ARSNKR. COEF. DIV. BY 15', 1PE12.4, '/')
    77  READ(1,77) I25
    FORMAT(F4.0)
    C25 = D5 * X2
    A25 = (ALOG(I0/I25))/(RCM*C25)
    444  WRITE(2,444)
    FORMAT(' ', A1, 15)
    WRITE(2,11) A25
    RAS = A5 - (7105*15 - 605/000)*(1-F)/F
    454  WRITE(2,454) RAS, T5
    FORMAT(' ', 'THE ARSNKR. COEF OF USE IS ', 1PE12.4, ' AT ', 1PE12.4)
    RAS5 = A5 - A25*(1-F)/F
    457  WRITE(2,457) RAS5, 15
    321  FORMAT(' ', 'THE ARSNKR COEF OF USE IS ', 1PE12.4, ' AT ', 1PE12.4)
    CONTINUE
    500  WRITE(2,500)
    FORMAT(2X, ' WITH BOUNDARY LAYER CORRECTIONS')
    WRITE(2,550)

```

```

550  FORMAT(2X, '*****')
      TL=DWEP*DS/(D1*U1)+DWEF/US
      TH=4.7A36*N(A**2+P1*700/((ANALH**2.07)+A)
      R=SQRT(TL/TH)
      T2=I2*(1+0.2*K)
      D2=UP*(1.0+0.2*N/(1.66A7-1.0))
      P2=UP*R2+I2
600  WRITE(2,600) I2,P2,UP
      FORMAT(2X, 'I2=',F0.1,3X, 'P2=',F10.5,4X, 'D2=',1PE12.4)
      TS=I2+50.0
650  CONTINUE
      H1R=5.2033F0+I5-1.5515F4
      H5=(H1R-HI2)/V2-V2/2
      W2=UR+V2
      TSCALC=I2*H5/A2+39.948*A2*H5/K1-39.948*U5**2/K1
      IF (ABS(TSCALC-I5).LE.0.1) GO TO 700
      TS=I5+0.1
      GO TO 650
700  CONTINUE
      D5=UP*R2/UR
      P5=UR*R2+I5
750  WRITE(2,750) I5,P5,U5
      FORMAT(2X, 'I5=',F0.1,3X, 'P5=',F10.5,4X, 'D5=',1PE12.4)
      WRITE(2,800) UR,W2
800  FORMAT(2X, 'UR=',1PE10.0,4X, 'W2=',1PE10.4)
      STOP
      END

```

APPENDIX B
PROGRAM DATA FIT

PROGRAM DATAFIT
 PROGRAM DATAFIT IS FROM RONALD FUSNER'S THESIS, 1961
 AND WAS FITTED FOR VAX BY SUSAN SORLIF, 1987
 DATAFIT IS A NONLINEAR LEAST SQUARES ALGORITHM THAT CAN
 BE USED TO SOLVE ANY DIFFERENTIABLE NON-LINEAR EQUATION WITH 70
 OR LESS PARAMETERS AND 1000 OR LESS DATA SETS. DATAFIT USES THE
 MARQUARDT ALGORITHM TO OBTAIN A COMPROMISE BETWEEN THE GRADIENT
 METHOD AND THE LINEARIZATION METHOD.

THE USER HAS THE OPTION OF CHOOSING ADDITIONAL OUTPUT TO ACCOMPANY
 THE LISTING OF LEAST SQUARES VALUES FOR THE MODEL PARAMETERS
 AND THEIR ESTIMATED STANDARD ERRORS. THIS ADDITIONAL OUTPUT IS
 INTENDED TO HELP THE USER DECIDE TO THE LOCATED MINIMUM OF THE
 LEAST SQUARES HYPERSURFACE THAT THE PROGRAM HAS LOCATED IS IN
 FACT THE LOCATION OF THE DESIRED GLOBAL MINIMUM.

THE USER MUST SUPPLY A FUNCTION SUBROUTINE ENTITLED -FUNCTN-
 THAT, WHEN SUPPLIED WITH THE CORRECT VALUE OF THE INDEPENDENT
 VARIABLES AND CURRENT MODEL PARAMETERS WILL GIVE THE VALUE
 OF THE DEPENDANT VARIABLE.
 SEE THE SAMPLE FUNCTN SUBROUTINE SUPPLIED AT THE END OF THIS
 PROGRAM.

THE USER MUST ALSO SUPPLY THE PROGRAM WITH A DATA FILE THAT
 CONTAINS THE DATA, INITIAL PARAMETER GUESSES, DELTA B VALUES,
 PARAMETER NAMES, ECT. THAT ARE USED FOR COMPUTATION AND OUTPUT.

THE INPUT FILE MUST CONTAIN THE FOLLOWING DATA IN THIS ORDER:

1. A TITLE OF THE RUN 12 CHAR MAX
2. # OF PARAMETERS, # OF INDEPENDENT VARIABLES, # OF DATA SETS
 10 INDEPENDENT VARIABLES MAX.
3. EACH PARAMETER NAME FOLLOWED BY THE INITIAL GUESS OF THAT
 PARAMETER AND THE VALUE OF DELTA B TO BE ADDED TO COMPUTE
 THE DERIVATIVE BY FINITE DIFFERENCES. THEREFORE 2 CARDS
 FOR EACH PARAMETER. (E11.5, 2X, F6.1)
4. A CARD CONTAINING THE DEPENDENT VARIABLE NAME 12 CHAR MAX
5. CARDS CONTAINING EACH OF THE DEPENDENT VARIABLE'S NAMES
 12 CHAR MAX
6. THE # OF DATA POINTS TO BE DELETED BETWEEN 0 AND 30
7. A CARD FOR THE DATA POINTS TO BE DELETED WHICH CONTAINS THE #
 CHAR RUN NAME OF THAT DATA POINT. (IN INPUT ORDER)
8. THE OPTIONS WHICH CAN BE:
 - SEF -- TO SEF EACH ITERATION OF THE PROGRAM
 - SEF+ -- OPTION SEF PLUS SHOWING EACH MATRIX MANIPULATION.
 - LRFS -- PRINT RESIDUALS IN A LIST
 - PRFS -- PLOT OUT THE RESIDUALS IN PLOTS
 - LAMB -- RESFT THE VALUE OF LAMBDA: A VALUE BETWEEN 1000 AND
 .001 MUST FOLLOW THIS CARD. (F8.3)
 - LIMI -- RESFT THE # OF DECIMAL OF NO CHANGE THAT THE SUM
 OF SQUARES IS TO REACH IN ORDER TO TERMINATE THE
 PROGRAM: 5 IS DEFAULT AND A VALUE BETWEEN 1 AND
 14 MUST FOLLOW. (I4)
 - LOOP -- THE MAX ITERATIONS: DEFAULT 30 (I4)
 - REFT -- RESFT VALUE OF LAMBDA IF NO CONVERGENCE AFTER MAX ITER
 DEFAULT IS 100,000. (F8.3)
 - SMRP -- UNCOMPLETED DO NOT USE
 - STOP -- THIS MUST BE THE LAST OPTION AND MUST BE INCLUDED
 EVEN IF NO OTHER OPTIONS ARE CHOSEN.
9. THE DATA SET CARDS:
 EACH CARD CONTAINS THE 4 CHARACTER RUN LIST THAT IS ANY

ARBITRARY TAG THAT THE USER CARES TO PLACE ON EACH DATA SET.
 THE VALUE OF THE DEPENDENT VARIABLE FOLLOWED BY EACH OF THE
 DEPENDENT VARIABLES IN ORDER FROM $Y(N,1)$ TO $Y(N,I)$. IF THERE
 ARE I INDEPENDENT VARIABLES AND N DATA SETS.
 THERE MUST BE A CARD FOR EACH SET OF POINTS IN THE DATA SET.

A SAMPLE INPUT FILE IS SHOWN BELOW:

GAUSSIAN AND QUADRATIC PLUS RANDOM ERROR

2X 1Y 30
 MULTIPLIER
 21000E+03XX 2.0
 36500E+03XX 1.0
 DEPENDANT
 COUNT

0
 PRES
 LIMIT
 10
 STOP

'D 1'XXX.1979E+03XX.5000
 'U 2' A NEW FORMAT STATEMENT IS NEEDED
 'D 4' FOR MOST DATA SETS (SEE LABEL 16 IN MAIN PROGRAM)

NOTE: IF THE DATA SET IS LARGE, USE ABOUT 100 POINTS TO GET
 AN INITIAL GUESS FOR THE PARAMETERS FOR THE FULL SET.
 THIS WILL SPEED THINGS UP.

```

COMMON/NAME/RNAME
COMMON/INSTUFF/ NIV, NTERMS, NPTS, NDEL, B, DELTA, X, Y
COMMON/RESULT/ YCALC, SUMSQ, SIGMA, ITER, LAMBDA
COMMON/OPTIONS/ CLAM, SLIM, SUMP, RLAM
COMMON/OPT/OPT
CHARACTER*4 OPTION(10), 0
REAL YCALC(1000), SUMSQ, SIGMA(30), LAMBDA
INTEGER TITLE(9), HEAD(9)
CHARACTER*4 RNAME(30,3), XNAME(10,3), YNAME(3), UNAME(1000)
CHARACTER*4 NDEL(10)
INTEGER SUMP, SLIM
REAL Y(1000), X(1000,10), R(30), DELTA(30), CLAM
INTEGER ORDER(1000), OPT(10)
DATA HEAD/ 'NON', 'LINE', 'PAR 1', 'EAST', 'SQU', 'ARE',
1 'DAT', 'A FT', 'V 4'
DATA OPTUM/ 'SEF', 'LRES', 'PRES', 'REST', 'LAMB',
1 'LIMT', 'LOUP', 'SMNP', 'SFE+', 'STOP'
DATA OPT/10*0/
OPEN(UNIT = 22, TYPE = 'OLD', FORM = 'FORMATTED')
OPEN(UNIT = 19, TYPE = 'NEW', FORM = 'FORMATTED')

```

HEAD IN DATA AND OPTIONS

```

READ (22, 1050), (TITLE(J), J = 1, 9)
READ (22, 15), NTERMS, NIV, NPTS
FORMAT (I2, X, I2, X, T4)
DO 31 K = 1, NTERMS
READ (22, 1050), (BNAME(K, I), I = 1, 3)
READ (22, 14), R(K), DELTA R(K)
FORMAT (2X, F5.2, X, F5.0)

```


REPEAT

SUBROUTINE REPEAT IS THE SUBROUTINE THAT LOOPS THROUGH MARQUARDT'S ALGORITHM AND CHECK TO SEE IF THE SUM OF SQUARES HAS DECREASED FOR THIS PARTICULAR LOOP. IF IT HAS DECREASED THEN THE NEW PARAMETERS REPLACE THE OLD ONES AND LAMBDA IS REDUCED BY A FACTOR OF 10. IF THE SUM OF SQUARES IS LARGER IF THE SUM OF THE SQUARES IS LARGER THEN THE OLD PARAMETERS ARE RETAINED AND LAMBDA IS INCREASED BY A FACTOR OF 10.

```

SUBROUTINE REPEAT (X,Y,NPTS,NTERMS,B,DELTA, SIGMA, LAMBDA,
1          YCALC, SUMSQ)
DOUBLE PRECISION INVERSE(30,30), BETA(30), ALPHA(30,30)
REAL X(1000,10), Y(1000), DERIV(30), R(30)
REAL SIGMA(30), YCALC(1000)
REAL DELTA(30)
REAL LAMBDA, TEMPB(30)

```

SUMSQ INITIAL

```

DO 60 I = 1, NPTS
YCALC(I) = FUNK(X, Y, B)
SUMSQ = FUNKOFF(Y, NP, YCALC)

```

EVALUATE MATRICES

```

DO 34 J = 1, NTERMS
BETA(J) = 0.00
DO 34 K = 1, J
ALPHA(J, K) = 0.00

DO 50 I = 1, NPTS
CALL FPRIME (X, Y, B, DELTA, NTERMS, DERIV)
DO 46 J = 1, NTERMS
BETA(J) = BETA(J) + 1.00*(Y(I) - YCALC(I))*DERIV(J)
DO 46 K = 1, J
ALPHA(J, K) = ALPHA(J, K) + 1.00*DERIV(J)*DERIV(K)
CONTINUE
DO 53 J = 1, NTERMS
DO 53 K = 1, J
ALPHA(K, J) = ALPHA(J, K)

```

INVERT

```

WRITE (10, 55)
FORMAT(10, 116, 'INITIAL MATRIX/VECTOR VALUES')
WRITE (10, 56)
FORMAT(10, 1)
DO 57 J = 1, NTERMS
WRITE (10, 58) J, BETA(J), J, DERIV(J)
FORMAT(10, 1, 'BETA(', I2, ') = ', G20.12, 'X, 'DERIV(', I2, ') = ', G20.12)
WRITE (19, 56)
DO 59 J = 1, NTERMS
WRITE (19, 61) ((I, 1, 1, ALPHA(I, J)), I = 1, J)
FORMAT(10, 10 (A1, 'ALPHA(', I2, ', ', I2, ') = ', G20.12, 'X))

DO 70 J = 1, NTERMS
DO 70 K = 1, NTERMS
INVERSE(I, K) = ALPHA(J, K)

```



```

74 INVERSE(J,J) = (1.00 + 1.00*LAMBDA)*ALPHA(I,J)
80 CALL IMATRIX(INVERSE, NTERMS, UFI)
90
21 WRITE(19, 21)
   FORMAT(10I, T16, 'INVERSE ALPHA MATRIX')
   WRITE(19, 20)
   DO 22 J = 1, NTERMS
22   WRITE(19, 21) ((I, 1, J, INVERSE(I,J)), I = 1, J)
   WRITE(19, 23)
23   FORMAT(10I, T5, 'PARAMETER CHANGES')
   DO 240 I = 1, NTERMS
240  TEMPR(I) = 0.00
   DO 241 T = 1, NTERMS
   DO 242 J = 1, NTERMS
24   TEMPR(T) = BETA(J)*INVERSE(I,J) + TEMPR(T)
   WRITE(19, 30)
   DO 243 J = 1, NTERMS
748  WRITE(19, 29) J, R(J), TEMPR(J), FMPB(J) + R(J)
798  FORMAT(10I, T10, G17.6, ' + ' G17.6, ' = ' G17.6)
800  FORMAT(10I, T10, 'ORIGINAL B', T30, 'PLUS CHANGE', T50, 'GIVES NEW B')
81   DO 82 J = 1, NTERMS
   TEMPR(J) = B(J)
82   DO 83 K = 1, NTERMS
83   TEMPR(I) = FMPB(J) + BETA(K) + INVERSE(J,K)
84   DO 92 T = 1, NTERMS
84   YCALC(T) = FUNCTIN(X, T, TEMPR)
85   SUMSOR = FUNKOFF(Y, NPTS, YCALC)
86   SUMIR = SUMSOR1 - SUMSOR
87   IF (SUMIR .LE. 0.0) GOTO 95
88   IF (SUMIR .GT. 0.0) GOTO 101
95   LAMBDA = LAMBDA + 10.0
96   GO TO 71
101  DO 103 J = 1, NTERMS
101  R(J) = FMPB(J)
103  RICHAB(J) = 1.00*SQRT(DABS(INVERSE(J,J)))
103  LAMBDA = 0.10*LAMBDA
110  RETURN
   END

```

FPRIME

FPRIME IS THE SUBROUTINE THAT CALCULATES THE FUNCTION DERIVATIVES BY DIFFERENCE METHODS. DELTA U IS ADDED TO THE PARAMETERS AND THIS USED WITH THE YCALC VALUES TO OBTAIN A VALUE FOR THE SLOPE WHICH IS RETAINED AS A CLOSE APPROXIMATION OF THE FUNCTION DERIVATIVE.

```

SUBROUTINE FPRIME(X, I, R, DELTA, NTERMS, UFRIV)
REAL X(1000,10), R(30), DELTAB(30), DERIV(30)
DO 18 J = 1, NTERMS
AJ = B(J)
DELTA = DELTAB(J)
R(I) = AJ + DELTA
YFTI = FUNCTIN(X, T, R)
R(I) = AJ - DELTA
DERIV(J) = (YFTI - FUNCTIN(X, T, R))/(2*DELTA)

```

16

```

R(J) = A.I
RETURN
END

```

```

*****

```

```

FUNKOFF

```

```

SUBROUTINE FUNKOFF CALCULATES THE SUM OF THE SQUARES OF THE FUNCTION
BY CALCULATING THE VALUES OF THE INDEPENDANT VARIABLES AND SUBTRACT
THESE FROM THE EXPERIMENTAL VALUES. THESE VALUES ARE THEN SQUARED
AND THEN SUMMED TOGETHER.

```

```

FUNCTION FUNKOFF(Y, NPTS, YCALC)
REAL Y(200), YCALC(200)
FUNKOFF = 0.0
DO 1 N = 1, NPTS
FUNKOFF = FUNKOFF + (Y(N) - YCALC(N))*(Y(N) - YCALC(N))
RETURN
END

```

```

*****

```

```

IMATRIX

```

```

IMATRIX IS A MATRIX INVERSION SUBROUTINE THAT INVERTS THE ALPHA
MATRIX TO BE USED IN THE SOLUTION OF THE PARAMETER CORRECTION
VALUES. TAKEN MOSTLY FROM BEVINGTON, DATA REDUCTION AND ERROR
ANALYSIS FOR THE PHYSICAL SCIENCES.

```

```

SUBROUTINE IMATRIX(INVERSE, NORDER, DET)
DOUBLE PRECISION INVERSE, MAXIMUM, SAVE
DIMENSION INVERSE(30,30), IK(10), JK(10)
DET = 1.0
DO 100 K = 1, NORDER

```

```

FIND LARGEST ELEMENT OF MATRIX

```

```

MAXIMUM = 0.00
DO 30 I = K, NORDER
DO 30 J = K, NORDER
IF (DABS(MAXIMUM) - DABS(INVERSE(I,J))) > 24, 24, 30
MAXIMUM = INVERSE(I,J)
IK(K) = I
JK(K) = J
CONTINUE

```

```

INTERCHANGE ROWS AND COLUMNS IN ORDER TO INCREASE PRECISION

```

```

IF (MAXIMUM .EQ. 0.00) CALL ERROR(-4)
I = IK(K)
IF (I-K) 21, 51, 43
DO 50 J=1, NORDER
SAVE = INVERSE(K, J)
INVERSE(K, J) = INVERSE(I, J)

```

```

50 INVERSE(T,J) = SAVF*(-1.00)
51 I = JK(K)
52 IF (J-K) 21, 51, 53
53 DO 60 T = 1, NORDER
54 SAVE = INVERSE(I,K)
55 INVERSE(T,K) = INVERSE(I,J)
56 INVERSE(T,J) = SAVF*(-1.00)

```

ACCUMULATE ELEMENTS OF INVERSE MATRIX

```

60 DO 70 T = 1, NORDER
61 IF (T-K) 63, 70, 63
62 INVERSE(T,K) = (-1.00)*INVERSE(I,K)/MAXIMUM
63 CONTINUE
64 DO 80 T = 1, NORDER
65 DO 80 J = 1, NORDER
66 IF (T-K) 74, 80, 74
67 IF (J-K) 75, 80, 75
68 INVERSE(T,J) = INVERSE(I,J) + INVERSE(T,K)*INVERSE(K,J)
69 CONTINUE
70 DO 90 J = 1, NORDER
71 IF (J-K) 83, 90, 83
72 INVERSE(K,J) = INVERSE(K,J)/MAXIMUM
73 CONTINUE
74 INVERSE(K,K) = 1.00/MAXIMUM
75 DET = DET*MAXIMUM

```

RESTORE ORDERING OF MATRIX

```

101 DO 130 L = 1, NORDER
102 K = NORDER - L + 1
103 I = IK(K)
104 IF (J-K) 111, 111, 105
105 DO 110 I = 1, NORDER
106 SAVE = INVERSE(I,K)
107 INVERSE(T,K) = (-1.00)*INVERSE(I,J)
108 INVERSE(T,J) = SAVE
109 I = JK(K)
110 IF (I-K) 120, 130, 113
111 DO 120 J = 1, NORDER
112 SAVE = INVERSE(K,J)
113 INVERSE(K,J) = (-1.00)*INVERSE(I,J)
114 INVERSE(T,J) = SAVE
115 CONTINUE
116 RETURN
117 END

```

FITTER

SUBROUTINE FITTER DOES THE ACTUAL LOOPING FOR THE PROGRAM.

```

SUBROUTINE FITTER
COMMON/NAME/RNAME
COMMON/INSTUFF/ NIV, NTERMS, NPIS, NDFL, B, DELTA, X, Y
COMMON/RESULT/YCALC, SUMSOR, SIGMAR, ITER, LAMDA
COMMON/OPTIONS/CLAM, SLTM, SOUP, RLAM

```

```

COMMON/OPT/OPT
REAL YCALC(1000), SIIMSUR, LAMBDA, LAM
INTEGER NPT(30), SOUP, SLIM
CHARACTER*4 TITLE(9), HEAD(9), ONAME(1000), OPTUN(30)
CHARACTER*8 RNAME(30,3), XNAME(10,3), YNAME(3)
INTEGER NUFLTF(30)
REAL Y(1000), X(1000,10), B(30), DELTA B(30), ROLD(10)
REAL BSAVE(10), SIGMAH(30)
LAMBDA = 10.
LOOPS = 30
LIMIT = 5
IF(OPT(5) .EQ. 1) LAMBDA = CLAM
IF(OPT(6) .EQ. 1) LIMIT = SLIM
IF(OPT(7) .EQ. 1) LOOPS = SOUP
FLAG = 0
ITER = 0
DO 10 N=1, NTERMS
  BSAVE(N) = B(N)
  ROLD(N) = 0.0
  DFRFE = NPTS - NTERMS - 1.0
  DO 61 N=1, NPTS
    YCALC(N) = FUNCIN(X,N,R)
    ROLD = FUNKOFF(Y,NPTS,YCALC)
  IF (OPT(1) .EQ. 1) WRITE(19,93) ROLD
  FORMAT(19,93) 'THE INITIAL SUM OF SQUARES IS ', G18.10)
  DO 11 I = 1, LOOPS
    ITER = ITER + 1
    CALL REPEAT(X,Y,NPTS,NTERMS,DELTA B,SIGMAH,LAMBDA,YCALC,SIIMSUR)
    A1 = ROLD - SUMSOR
    X11 = ALOG10(SOLD)
    T1 = INT(X11)
    R1 = 10.0*(X11-T1)
    SIGFIG = -1.0*(ALOG10(A1/B1))
    IF (OPT(1) .EQ. 0) GO TO 67
    WRITE(19,62) ITER
    FORMAT(19,62) 'RESULTS OF ITERATION #', I3)
    WRITE(19,63) SIIMSUR, SORT(SUMSOR/DFRFE)
    FORMAT(19,63) 'THE SUM OF SQUARES IS ', G18.10,6X,
    'THE AVERAGE RESIDUAL IS ', G18.10)
    WRITE(19,64) B, 'PARAMETER', I4Y, 'VALUE', I7X, 'STANDARD ERROR')
    FORMAT(19,64)
    WRITE(19,65)
    FORMAT(19,65)
    DO 65 N = 1, NTERMS
      WRITE(19,65) N, (RNAME(N,L), L=1,3), R(N), SIGMAH(N)
      FORMAT(19,65) ' ', I3, 'A', 3(A4), 6X, G18.10, 6X, G18.10)
    CONTINUE
    WRITE(19,70)
    CALL LINE(1)
    WRITE(19,70)
    ROLD = SIIMSUR
    IF (SIGFIG .GE. LIMIT) WRITE(19,100)
    FORMAT(19,100) 'NORMAL CONVERGENCE')
    IF (SIGFIG .GE. LIMIT) GO TO 2
    CONTINUE
    IF(FLAG .EQ. 1) CALL ERROR(-5)
    LAM = 100000.
    CALL LINE(1)
    IF (OPT(2) .EQ. 1) LAM = RLAM
    WRITE(19,68) ITER, LAM
    FORMAT(19,68) 'ITERATIONS AND NO CONVERGENCE. ',

```

```

1  INITIAL PARAMETERS RESTORED NO LAMBDA SET TO 1.0F10.3 )
  CALL LINE(1)
  LAMBDA = LAM
  DO 69 N = 1, NIFRMS
  R(N) = BSAVE(N)
  FLAG = 1
  GOTO 60
  RETURN
  END

```

HEADING

SUBROUTINE PRINTS INFORMATION ON RUN

```

SUBROUTINE HEADING (TITLE, HEAD)
  INTEGER TITLE(9), HEAD(9)
  WRITE(19,6) (HEAD(I), I=1,9)
  WRITE(19,8) (TITLE(J), J=1,9)
  FORMAT ('0', I38, '0A8')
  FORMAT ('0', I40, '0A8')
  RETURN
  END

```

LINE

SUBROUTINE LINE SIMPLY PRINTS OUT THE REQUESTED NUMBER OF HEAVILY OVERPRINTED LINES USED IN BOTH THE HEADING SUBROUTINE AND THE ERROR SUBROUTINE.

```

SUBROUTINE LINE(N)
  DO 1 M = 1, N
  WRITE(19,2)
  WRITE(19,3)
  WRITE(19,4)
  WRITE(19,5)
  WRITE(19,6)
  FORMAT('0', I5, 'MM')
  FORMAT('0', I5, 'WW')
  FORMAT('0', I5, 'XX')
  FORMAT('0', I5, 'Y')
  FORMAT('0', I5, 'Z')
  RETURN
  END

```

INPUT

SUBROUTINE INPUT PRINTS OUT ALL OF THE INPUT DATA EXCEPT THE ACTUAL DATA SET ITSELF THAT WOULD BE OF INTEREST AND THAT COULD CHANGE BETWEEN RUNS.

SUBROUTINE INPUT

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```

COMMON/NAMF/RNAMF
COMMON/INSTUFF/ NIV, NTERMS, NPIS, NDEL, R, DELTAB, X, Y
REAL X(1000,10), Y(1000), DELTAB(30), B(30)
INTEGER NIV, NTERMS, NPIS, NDEL
CHARACTER*4 RNAMF(30,3)
WRITE(19,1)
FORMAT(10,4R(' '), ' DATA INPUT ',4R(' '))
WRITE(19,7)
WRITE(19,2) NIV, NTERMS
FORMAT(1,12X,12, ' INDEPENDENT VARIABLES',7X,12,
1 EQUATION PARAMETERS')
WRITE(19,4)
FORMAT(10,4R(' '), ' INITIAL DATA PARAMETERS ',4R(' '))
WRITE(19,7)
WRITE(19,5)
FORMAT(1,27X,'NUMBER',RX,' R NAME',RX,' R VALUE',9Y,' DELTA B')
WRITE(19,6)
FORMAT(1,27X,6(' '),RX,7(' '),7X,7(' '),9X,7(' '))
WRITE(19,7)
L = 0
DO 6 M = 1, NTERMS
L = L + 1
IF(L .LT. 6 .AND. M-2 .LT. NTERMS) GOTO 8
L = 0
WRITE(19,7)
FORMAT(1,1)
WRITE(19,4) M, (RNAMF(M,J), J=1,3), B(M), DELTAB(M)
FORMAT(1,2RX,1B,3A8,4X,5I2.5,4X,6I2.5)
RETURN
END

```

RESID

SUBROUTINE RESID PRINTS OUT A LISTING OF THE RESIDUALS IN BOTH INITIAL - READ IN - ORDER AND IN ORDER OF THE ASCENDING RESIDUAL VALUES IN ORDER FOR PROBLEMS WITH THE DATA SET TO COME TO THE ATTENTION OF THE USER MORE RAPIDLY.

```

SUBROUTINE RESID(Y, YCALC, DNAME, TITLF, NPIS)
REAL Y(1000), YCALC(1000), RES(1000)
INTEGER ORDER(1000), TITLF(9)
CHARACTER*4 DNAME(1000)
WRITE(19,5)
WRITE(19,4) (TITLF(J), J=1,9)
WRITE(19,6)
WRITE(19,5)
WRITE(19,6)
WRITE(19,11)
WRITE(19,6)
WRITE(19,7)
WRITE(19,6)
WRITE(19,6)
WRITE(19,10)
DO 1 N = 1, NPIS
RES(N) = Y(N) - YCALC(N)
CALL SORT(RES, ORDER, NPIS)

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```

T = 0
DO 2 N = 1, NPTS
  T = T + 1
  M = ORDER(N)
  WRITE(19,9) N, DNAME(N), Y(N), YCALC(N), RFS(N), M, DNAME(M), Y(M),
1 YCALC(M), RFS(M)
  IF (T .LT. 6) GOTO 2
  I = 0
  WRITE(14,10)
  CUNITNIE
  WRITE(19,10)
  WRITE(19,6)
  FORMAT(11, T30, 'NON-LINEAR LEAST SQUARES DATA FIT V3.0')
  FORMAT(10, I40, 9A4)
  FORMAT(11, I21, 55X, 'RESIDUALS', 55X, '*')
  FORMAT(11, I21, 121(1X))
  FORMAT(11, I2(1X, 'DATA', 7X, 'DATA', 8X, 'CALCULATED', 6X,
1 'RESIDUAL', 4X), 1X)
  FORMAT(11, I2(1X, 'ORDER NAME', 7X, 'VALUE', 9X, 'VALUE', 21X),
1 '*')
  FORMAT(11, I2(1X, I14, 2X, A8, 2X, I3(613.6, 2X)), '*')
  FORMAT(11, I2(1X, I50(11), 1X, I50(11), 1X, I50(11), 1X, I50(11)), '*')
  FORMAT(11, I2(1X, I21X, 'ORDERED BY INPUT', 22X, '*'),
1 19X, 'ORDERED BY RESIDUALS', 20X, '*')
  RETURN
END

```

SORT

SUBROUTINE SORT IS A SIMPLE BUBBLE SORT SUBROUTINE THAT RETURNS THE ASCENDING ORDER IN THE ARRAY LABELED ORDER.

```

SUBROUTINE SORT(X, ORDER, NUMBER)
  REAL X(1000), Y(1000), Y1
  INTEGER ORDER(1000), NUMBER, NI

```

INITIALIZE ORDER ARRAY AND PLACE ALL X VALUES INTO THE CORRESPONDING

```

DO 1 N = 1, NUMBER
  Y(N) = X(N)
  ORDER(N) = N

```

SET FLAG AND LOOP PARAMETERS

```

M = NUMBER - 1
FLAG = 0

```

RUN THROUGH LOOP UNTIL A LOOP OCCURS THAT NO ACTION HAS TAKEN OR ELSE THE END OF THE LOOP HAS BEEN REACHED.

```

DO 3 N = 1, M
  IF (Y(N+1) .GE. Y(N)) GOTO 3
  NI = ORDER(N)

```

```

ORDER(N) = ORDER(N+1)
ORDER(N+1) = N1
Y1 = Y(N)
Y(N) = Y(N+1)
Y(N+1) = Y1
FLAG = 1
CONTINUE
M = M - 1
IF (FLAG .EQ. 0 .OR. M .EQ. 0) RETURN
FLAG = 0
GOTO 2
END

```

4
 00000000
 9
 10
 00000000

OUTPUT

SUBROUTINE OUTPUT PRINTS OUT THE RESULTS OF THE MARQUARDT'S METHOD

```

SUBROUTINE OUTPUT
COMMON/NAME/RNAME
COMMON/INSTUFF/ NIV, NITERMS, NPTS, NUCL, B, DELTA, X, Y
COMMON/RESULT/YCALC, SUMSQ, STGMAR, ITER, LAMDA
REAL YCALC(1000), SUMSQ, SIGMAB(30), LAMDA
REAL Y(1000), X(1000,10), R(30), DELTA(30)
INTEGER ITER
CHARACTER*4 RNAME(30,3)
WRITE(14,1)
WRITE(19,2)
WRITE(14,3)
DO 6 J = 1, ITER-45
WRITE(19,9) J, (RNAME(I,J), J=1,3), R(I), SIGMAB(I)
WRITE(19,10) ITER, SUMSQ, SURT(SUMSQ/NPTS)
WRITE(19,2)
FORMAT(10T)
FORMAT('0',45(' '), 'PROGRAM RESULTS', 46(' '))
FORMAT('0',40(' '), 'FITTED PARAMETER INFORMATION', 36(' '))
FORMAT('0', 20X, 14, 6X, 34X, 4X, 612.5, 2X, 612.5)
FORMAT('0', 'A TOTAL OF',
1 12, ' ITERATIONS, THE RESIDUAL SUM OF SQUARES IS ',
1 612.10, ' THE STANDARD DEVIATION IS ', 612.10)
RETURN
END

```

PLOT

SUBROUTINE PLOT IS USED TO GENERATE PLOTS OF FUNCTIONS AND RESIDUAL THAT ARE CALLED FROM ELSEWHERE IN THE PROGRAM.

```

SUBROUTINE PLOT(X,Y,NUMBER,OPT)
REAL X(1000),Y(1000)
INTEGER OPT, NUMBER
CHARACTER*1 C(11), A(111)
DATA C/' ', '1', '2', '3', '4', '5', '6', '7', '8', '9', 'x',
XMIN = X(1)
XMAX = X(1)

```


SUBROUTINE RESIDP PRINTS OUT PLOTS OF THE RESIDUALS IN ALMOST EVERY CONCEIVABLE ORDER IN ORDER FOR THE USER TO DISCOVER ANY PATTERNS IN THE RESIDUALS. THIS IS PROBABLY THE MOST USEFUL PROBLEM DETECTING OPTION IN THE WHOLE PROGRAM.

```

SUBROUTINE RESIDP(Y,X,NIV,YCALC,NPTS,SMOD,XNAME,YNAME)
REAL XX(1000), RES(1000), Y(1000), YCALC(1000), X(1000,10)
REAL XP(1000)
INTEGER ORDER(1000),SMOD
CHARACTER*4 XNAME(10,3),YNAME(3)
DO 1 N = 1,NPTS
  XX(N) = X(N)
  ORDER(N) = N
  RES(N) = Y(N) - YCALC(N)
CONTINUE
IF(NIV.EQ.1) THEN
  WRITE(19,34) (YNAME(M), M=1,3), (XNAME(I,1), I=1,3)
  FORMAT('11',2(1),T41,3(A4), ' AGAINST ',3(A4))
DO 35 K=1,NPTS
  XP(K) = X(K,1)
  CALL PLOT(XP,Y,NPTS,1)
ENDIF
WRITE(19,10)
FORMAT('11',2(1),T50, ' RESIDUALS PLOT')
WRITE(19,11)
FORMAT('11',T42, 'RESIDUALS AGAINST INPUT ORDER')
CALL PLOT(XX,RES,NPTS,1)
WRITE(19,10)
WRITE(19,31) (YNAME(M),M=1,3)
FORMAT('11',T42, 'RESIDUALS AGAINST ',3(A4))
CALL PLOT(Y,RES,NPTS,1)
DO 32 N = 1,NIV
  WRITE(19,10)
  WRITE(19,33) (YNAME(N,I),I=1,3)
  FORMAT('11',T42, 'RESIDUALS AGAINST ',3(A4))
DO 38 M = 1,NPTS
  XX(M) = X(M,N)
  CALL PLOT(XX,RES,NPTS,1)
CONTINUE
CALL SORT(RES,ORDER,NPTS)
DO 2 N = 1,NPTS
  XX(N) = RES(ORDER(N))
DO 60 I = 1,NPTS
  RES(N) = XX(I)
  XX(N) = N
  WRITE(19,10)
  WRITE(19,22)
  FORMAT('10',T42, 'RESIDUALS PLOTTED IN ORDER')
  CALL PLOT(XX,RES,NPTS,1)
RETURN
END

```

ERROR

SUBROUTINE ERRUP IS USED TO GENERATE SELFEXPLANATORY OUTPUT TO TELL THE USER WHAT WENT WRONG IN THE PROGRAM.

```

SUBROUTINE ERROR(NF)
CHARACTER*15 TELL1(10), TELL2(10)
DATA TELL1(1)/ 'DEGRADES OF' /
DATA TELL2(1)/ 'FREEDOM' /
DATA TELL1(2)/ 'PARAMETER' /
DATA TELL2(2)/ 'INFUR READ' /
DATA TELL1(3)/ 'DATA POINT' /
DATA TELL2(3)/ 'READ IN' /
DATA TELL1(4)/ 'A MATRIX S' /
DATA TELL2(4)/ 'INGULARITY' /
DATA TELL1(5)/ 'EXCEED MAX' /
DATA TELL2(5)/ 'ITER #' /
DATA TELL1(6)/ 'ILLEGAL OP' /
DATA TELL2(6)/ 'TION VALUE' /
DO 4 M = 1, 4
WRITE(19,1)
FORMAT(' ', 1)
CALL LINE(1)
IF (M.EQ.3) WRITE(19,2) TARS(NF)
IF (M.EQ.3 AND NF.LT.0) WRITE(19,3)
FORMAT('0', 'ERROR NUMBER', T2, ' DETECTED BY DATAFIT')
FORMAT('0', 'THIS IS A FATAL ERROR')
CONTINUE
WRITE(19,5)
FORMAT('0', ' AN ERROR HAS OCCURED IN YOUR DATAFIT PROGRAM' /
1 ' IF THE ERROR IS NOT FATAL AN ATTEMPT WILL BE MADE TO RECOVER' /
WRITE(19,6) TELL1(TARS(NF)), TELL2(1ABS(NF))
FORMAT('0', 'THE ERROR THAT OCCURED WAS IN ', 2(A10))
DO 7 M = 1, 3
WRITE(19,1)
CALL LINE(1)
IF (NE.LI.0) STOP
RETURN
END

```

FUNCTION

THIS IS WHERE THE USER SUPPLIES SUBROUTINE IS TO GO. IT MUST HAVE THE EXACT SAME FIRST TWO LINES AS IN THE EXAMPLE SUBROUTINE LISTED BELOW AND LIST THE RESULT INTO THE VARIABLE FUNCTION.

```

FUNCTION FUNCTN (X, T, B)
REAL X(1000,10), B(30)
F = X(1,1)
T = X(T,2)
FUNCTN = B(1)*ALPHA**2 - B(2)*ALPHA + R(T)
FUNCTN = B(1) + R(2)*1*EXP(-(R(T)/(1+A))
FUNCTN = B(1) + R(2)*(1-F)/F
RETURN
END

```

```

SUBROUTINE STGPAR (SHMSUR, X, NPTS)
REAL X(1000,10)

```

```

SUMX2 = 0
SUMX = 0
DO 1 I = 1, NPIS
SUMX2 = SUMX2 + X(T,1)*X(I,1)
SUMX = SUMX + X(T,1)
F = NPIS*SUMX2 - SUMX**2
SIGA = SUMX2*SUMSUR/(NPIS-2)
SIGA = SQRT(SIGA)
SIGB = SUMSUR*NPIS/(F*(NPIS-2))
SIGB = SQRT(SIGB)
WRITE(19,200) SIGA, SIGB
FORMAT(' 1', ' UNCERTIANTY IN PARAMETER A = ',1PE11.4,2('/),
' UNCERTIANTY IN PARAMETER B = ',1PE11.4)
RETURN
END

```

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