

Guanidinium Thiocyanate Selective Ostwald Ripening Induced Large Grain For High Performance Perovskite Solar Cells

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ABSTRACT

Organic-inorganic lead halide perovskite has become one of the most attractive materials for future low-cost high-efficiency solar technology. However, the polycrystalline nature of perovskite thin-film often possesses an exceptional density of defects, especially at grain boundaries (GBs) and film surface, limiting further improvement in the power conversion efficiency (PCE) of the perovskite device. Here, we report a simple method to reduce GBs and to passivate the surface of a methylammonium lead triiodide (MAPbI₃) film by guanidinium thiocyanate (GUTS)-assisted Ostwald ripening post treatment. High-optoelectronic quality MAPbI₃ film consisting of micron-sized grains were synthesized by post-treating a MAPbI₃ film with GUTS/isopropanol solution (4 mg/mL, GUTS-4). Analysis of the electrochemical impedance spectra (EIS) of the solar cells showed that interfacial charge recombination resistance of the device based on a GUTS-4 post-treated MAPbI₃ absorber film was increased by a factor of 1.15 to 2.6, depending on light illumination intensity, compared to the control MAPbI₃ cell. This is consistent with results of the open-circuit voltage (V_{oc}) decay and the light intensity dependent photovoltage evolution which shows device with GUTS treatment had longer charge carrier lifetime and was more ideal (ideality factor=1.25). Further characterization by Kelvin probe force microscope indicated that GUTS-4 treatment shifted the energetics of the MAPbI₃ film by ~100 meV towards better energy level alignment with adjacent SnO₂ electron transport layer, leading to a more favorable charge extraction process at the MAPbI₃/SnO₂ interface. As a result, the PCE of PSCs was enhanced from 14.59% to 16.37% and the hysteresis effect was mitigated.

Keywords: Perovskite solar cells; Large grains; Improved fill factor; High Efficiency; Ostwald ripening.

INTRODUCTION

Recently, organic-inorganic lead halide perovskites have emerged as ideal materials for low-cost solution-processable, high-efficiency solar technologies.¹ Solar cells based on perovskites have shown sky-rocketing progress in power conversion efficiency (PCE) within seven years of development, from 3.8% in 2009 to over 22% in early 2016, making it the most commercially attractive solar cell technology at present.^{2,3} The nature of am-bipolar charge transport of perovskite (perovskite materials can transport both electron and hole between cell terminals)⁴⁻⁶ have led to the development of planar perovskite solar cell configurations in which perovskite functions as both the light absorber and charge transport layers. The structural simplicity of the planar configuration makes it even more economically viable and commercially attractive.

It is recognized that a delicate control of the optoelectronic quality of perovskite film is of vital importance to obtain a high-efficiency solar cell, especially for planar structure.^{6, 7} For example, pin-holes and incomplete surface coverage can result in low light absorption and low-resistance shunting paths while the high density of undesirable defects causes a high-rate of charge recombination and low charge collection efficiency. This in turn entails low photovoltaic performance of the corresponding devices. To solve this issue, methods for preparing pinhole-free, uniform and highly compact films have been developed such as the anti-solvent one-step dripping method.⁷⁻¹⁰ Fundamentally, the anti-solvent method involves in pouring of an anti-solvent onto a wet perovskite film to induce fast crystallization, resulting in formation of a smooth, fine-grained polycrystalline film which contains significant number of grain boundaries (GBs). Recent reports have suggested that, GBs in perovskite film are likely to function as: 1) centralized areas of electronic trap states and 2) pathways for ion migration.¹¹⁻¹³ The electronic trap states not only serve as barriers for charge transport, but they also enhance non-radiative recombination, severely reducing the charge carrier life time, and thus the overall photovoltaic device performance.¹⁴⁻¹⁶ Possibly, the high density of GBs or electronic defects in the perovskite films prepared by typical deposition techniques, including the antisolvent-dripping method, is one of the reasons why the best performing PSCs still lags behind the theoretical Schocley-Queisser efficiency limit for a single junction solar cell.¹⁷ In addition, ion migration

at GBs is suspected to cause current-voltage hysteresis of perovskite devices.¹⁸ As such, it is essential to reduce the number of GBs through increasing grain sizes and/or passivating the GBs of perovskite films to enable hysteresis-free high efficiency PSCs.

A variety of methods have been established to eliminate the GBs of perovskite films through the careful control of nucleation-growth rates by modification of underlying substrates^{11, 19} or altering the chemistry of precursor solutions.^{20, 21} Among them, methods based on tuning perovskite precursor composition seem to be the most successful. Yang *et al.* reported that methyammonium iodide (MAI) is an effective additive to tune the grain size of methyammonium lead tri-iodide (MAPbI₃) films from few hundred nanometer to few micrometer.²⁰ Pursuing a different direction, Carmona *et al.* stated that a moderate excess of PbI₂ in the MAPbI₃ precursor is beneficial for obtaining large homogeneous MAPbI₃ films.²² Crystal growth retardants, such as lead(II) chloride (PbCl₂) have also been shown to be an advantageous precursor additive for grain size enlargement of MAPbI₃ films.^{23, 24} Besides lead(II) halide based additives, non-halide lead source such as lead(II) thiocyanate (Pb(SCN)₂) were also reported to be effective for enhancing crystal domain sizes of perovskite films.²⁵⁻²⁹

Nevertheless, it is difficult to prepare high optoelectronic quality perovskite film with minimal defect densities, even with anti-solvent dripping method, and thus film post treatment is normally necessary.³⁰⁻³⁴ Among these, the method based on post treatment of MAPbI₃ perovskite film using methylammonium bromide (MABr)/isopropanol was reported to lead to secondary crystallization of MAPbI₃ films through an Ostwald ripening mechanism.³² The resulting MAPbI₃ films have enhanced grain size, crystallinity and yielded better PCE devices. The elegance of this finding is that post-treated perovskite films are pinhole-free, possessing large grain domain and low density of defects regardless of the quality of initial film. However, the MAPbI₃ films prepared by the MABr-assisted Ostwald ripening process often contain broad crystal size distributions even though the film has undergone an extreme heat treatment process (*i.e.* 150°C for 10 minutes). To date, only a few Ostwald ripening-assisted precursors have been utilized for the post-treatment of perovskite films due to strict requirements for efficient coarsening in Ostwald-type ripening

process.³² Due to this, there still remains considerable room to explore the application of Ostwald ripening-assisted for improving the quality of perovskite film for higher power conversion efficiency.

Apart from grain size enlargement, approaches for passivation of GBs and interfaces/surface of perovskite film have also been investigated. Fullerene, an electron transport layer, was found to be an effective passivating agent which diffuses along GBs to passivate the defects.³⁵⁻³⁷ Several groups have noted that PbI_2 is able to passivate GBs and interfaces of perovskite films.^{21, 38, 39} These studies have shown that PbI_2 mainly occupies the space along GBs of perovskite film, forming an energy barrier that hinders leakage of both electrons and holes from perovskite film which reduces recombination. Furthermore, guanidinium iodide (GuI) was also found to successfully suppressing defects at GBs of MAPbI_3 films (more reference).⁴⁰ The use of (GuI) in MAPbI_3 precursor solutions has proven to extend the charge carrier lifetime by a factor of ten, and thus yielding PCE exceeding 17% for planar MAPbI_3 solar cell. The research showed that guanidinium ions do not incorporate in perovskite lattice, but instead, reside at the GBs of perovskite film, forming hydrogen bonds with under-coordinated iodine species and thus suppressing charge recombination. Although the benefits of GuI for defect mitigation in MAPbI_3 films is exceptional, the final structure of Gu-based products remain unclear.

Herein, we report a new method based on treatment of MAPbI_3 film using a guanidinium thiocyanate (GUTS) precursor solution. In this study, it was found that the GUTS-treatment effectively converts fined-grain nanometer-scale MAPbI_3 film into micron-sized MAPbI_3 perovskite film with a low density of GBs. We have also found that this GUTS-post treatment method successfully passivates the perovskite interface, which in turn significantly reduces surface charge recombination. The cooperation of these two effects enables improved performance of solar cells with less hysteresis.

1. EXPERIMENTAL SECTION

Materials preparation

All materials were purchased from Sigma-Aldrich and used as received without further purification unless otherwise stated. Methylammonium lead tri-iodide (MAPbI₃) perovskite films were prepared based on Lewis acid-base adduct approach, details of which are described in the previous reports.^{9, 10} In brief, a MAPbI₃ perovskite precursor solution was prepared by dissolving 461 mg of PbI₂, 159 mg of methylammonium iodide (MAI) (Dyesol), and 78 mg of dimethyl sulfoxide (DMSO) in 650 mg of dimethyl formamide (DMF) at room temperature, under rigorous stirring for four hours to obtain clear solution. The prepared MAPbI₃ precursor solution was filtered by a syringe filter (pore size: 0.22 μm) prior to use for deposition of films. Guanidinium thiocyanate (GUTS) precursors were prepared by dissolving different amounts of GUTS (2 mg, 4 mg, 6 mg, 8 mg, and 10 mg) in 1 mL anhydrous isopropanol (IPA). A solution for hole transport material was prepared by addition of 72.3 mg of 2,2',7,7'-Tetrakis-(N,N-di-4-methoxyphenylamino)-9,9'-spirobifluorene (Spiro-MeOTAD) (Borun New Material), 28.8 μL of 4-*tert*-butylpyridine, and 17.5 μL of Bis(trifluoromethane)sulfonimide lithium (Li-TFSI) solution (720 mg of Li-TFSI in acetonitrile) to 1 mL of chlorobenzene.

Device fabrication

Solar cells were fabricated using fluorine-doped tin oxide (FTO) coated glass (Nippon Electric Glass, 15 Ω/□) as substrate which was firstly patterned through partially removal of FTO via etching with 35.5 wt% HCl and zinc powder. The substrates were then cleaned for 20 mins in an ultrasonic bath with an aqueous solution of Decon-90 detergent (5%), followed by the same treatment in a mixture of acetone, isopropanol and ethanol (1:1:1, volume ratio). Prior to use, the substrates were treated with UV-Ozone for 30 mins to remove any remaining organic solvent residue. An electron transport layer based on SnO₂ (~40 nm) was deposited via spin-coating 0.1 M solution of tin(II) chloride (98%) in ethanol (96%) at 3000 rpm for 30 s in air. The film was then annealed in air at

185 °C for 1 hour before being cleaned with UV-Ozone for 20 mins and transferred to an Ar-filled glove box. A MAPbI₃ layer (~400 nm) was deposited onto the prepared SnO₂ layer at 4000 rpm for 25 s. During spin-coating, 0.5 mL of diethyl ether was dropped onto the center of the spinning substrate 18 s prior to the end of the program. The perovskite layer was then dried at 65 °C for 2 mins. For samples treated via the GUTS methodology, 100 µL of GUTS in IPA solution (2-10 mg/mL) was dropped on top of the perovskite film during a second spin coating cycle at 4000 rpm. The perovskite layer was then annealed at 100 °C for 2 mins. For convenience, the MAPbI₃ films with treatments of 2 mg/mL, 4 mg/mL, 6 mg/mL, 8 mg/mL and 10 mg/mL GUTS in IPA are abbreviated as GUTS-x-MAPbI₃, where x=2, 4, 6, 8 and 10. The hole-transport layer (~200 nm) was deposited from the prepared Spiro-OMeTAD solution onto the as-prepared perovskite layer at 4000 rpm for 30 s. The device fabrication was finished by deposition of a 100 nm layer of gold for use as a back contact on the prepared sample via an e-beam evaporation process operating at 10⁻⁶ torr pressure.

Characterization

The top-view and cross-sectional scanning electron microscopy (SEM) images of the samples were obtained using a field emission scanning electron microscope (FSEM JOEL 7001F) at an acceleration voltage of 5 kV. The UV-visible absorbance spectrum was measured with a UV-visible spectrometer (Cary 50). Crystal structure of the perovskite film deposited on FTO-glass substrate was determined by X-ray diffraction (Rigaku SmartLab) with a monochromatic CuK α ($\lambda=0.154$ nm) as a radiation source. A scan rate of 1.5°/minute and step size of 0.02° was used in the XRD measurement. The performance of perovskite devices was measured under irradiation of 100 mW/cm² (AM1.5, 1 sun) provided by a solar simulator (Oriel Sol3A, Newport) equipped with 450 W Xenon lamp. IPCE measurement was conducted by quantum efficiency system (IQE 200B, Newport) under AC mode. Electrochemical impedance spectroscopy (EIS) of the PSCs was performed in a frequency range from 1 MHz to 100 mHz using an electrical workstation (VSP BioLogic Science Instruments) under open-circuit condition with different light

illumination intensities. An AC voltage with a perturbation amplitude of 20 mV was applied on the device in the (EIS) measurement. Scanning Kelvin Probe Force Microscopy (KPFM) (Oxford instrument, Asylum Research) was performed on the prepared perovskite film in ambient conditions using NSG-03 Pt coated cantilever at room temperature. Work function of the cantilever was measured using HOPG standard sample.

The open-circuit voltage decay measurements were performed by monitoring the open-circuit voltage (V_{oc}) of PSC as a function of time in dark with an electrochemical workstation (BioLogic). The V_{oc} of the cell was generated using a white light LED with illumination intensity equivalent to 0.15 sun (illumination time: ~2 s). The open-circuit voltage decay (OCVD) was recorded when the white LED was turned off.

The light intensity dependence of open-circuit voltage (V_{oc}) measurement was carried out by measuring J - V plots of the solar cell under different illumination intensities generated by the solar simulator. The light intensities were controlled by using neutral density optical filters with transmittance of 1%, 11%, 19%, 30% and 55%.

The stability of the PSCs was tested by monitoring the efficiency of un-encapsulated PSCs which were stored in a desiccator (relative humidity ~33-35%) in dark. The performance of the device was measured every 3 days in ambient condition with relative humidity of 40-60%.

Results and discussion

Figure 1(a)-(f) illustrates the top-view scanning electron microscopy (SEM) images of GUTS- x -MAPbI₃ perovskite films with $x=0, 2, 4, 6, 8, 10$, respectively. In accordance to the previous reports,^{9, 10} MAPbI₃ film synthesized by our one-step Lewis acid-base adduct method is pinhole-free, highly compact with crystallite sizes ranging from 100 nm to 500 nm (Fig. 1a). When MAPbI₃ films were treated with GUTS-2 precursor (2 mg/mL GUTS in isopropanol (IPA)), larger grains with domain size up to 800 nm were observed while the MAPbI₃ film remains uniform and compact (Fig. 1b). Upon increasing the concentration of GUTS to 4 mg/mL (Fig. 1c), clear evidence of grain growth is seen with most grains exceeding 1 μ m. Compared to the pristine MAPbI₃ film, the lateral crystallite size of GUTS-4-MAPbI₃ film is five-folds larger, suggesting the effectiveness of this simple post processing procedure. As a general indication, the larger the perovskite film crystals, the higher the solar cell performance is expected because perovskite films with larger grains have lower numbers of GBs which act as barriers for charge transport between the cathode and anode in PCS.^{11, 12, 28} With further increment of the GUTS concentration to 6 mg/mL, 8 mg/mL and 10 mg/mL, even larger grain sizes are observed (~ 2 μ m in average with GuTS-10 treated film). However, pronounced contrast along GBs and surface of MAPbI₃ film are observed, which suggests formation of secondary phases. The amount of secondary phases increased as the concentration of GUTS increased, which is more clearly shown at lower magnification SEM images (Fig. S1). Furthermore surficial wrinkles are also found on the MAPbI₃ grains as the concentration of GUTS exceeded 6 mg/mL (Fig. 1d-f). The appearance of these wrinkled textures is likely induced by strain due to massive MAPbI₃ grain coarsening and/or competitive grain growth of MAPbI₃ crystals and secondary phases,¹⁰ which may lead to high surface roughness.

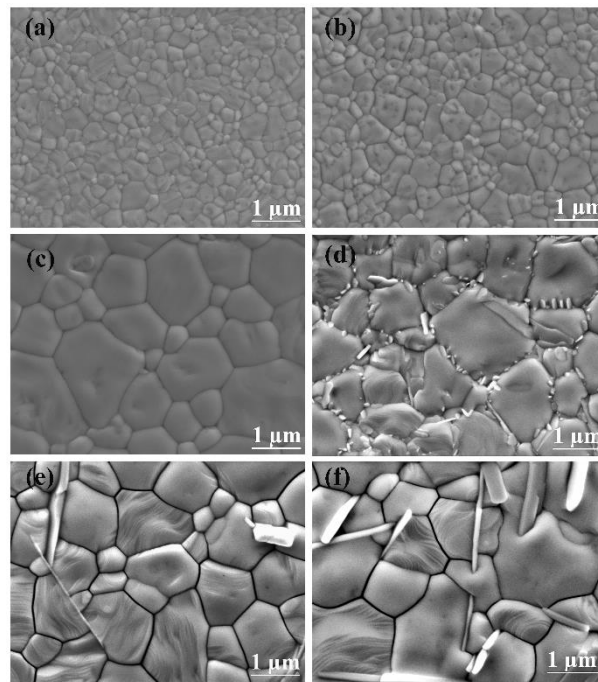
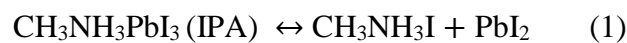


Fig. 1 Top-view SEM images of MAPbI₃ films. (a) Control MAPbI₃ film and the film treated with different concentrations of GUTS/IPA solution (b) 2 mg/mL, (c) 4 mg/mL, (d) 6 mg/mL, (e) 8 mg/mL, (f) 10 mg/mL.

Before further characterization of optoelectronic properties of the MAPbI₃ film, we first investigated the fundamental reason for the significant morphological change in the films. With the knowledge that IPA is capable of dissolving MAPbI₃ via MAI-extracting (1),^{41, 42} we presume that this extraction is the initial stages of Ostwald ripening which could be the potential origin for the above morphological change.



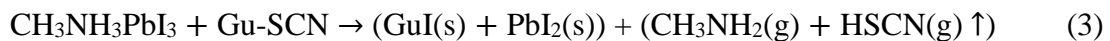
According to the Ostwald ripening process, once the GUTS/IPA solution is introduced to the surface of MAPbI₃ film, unstable small MAPbI₃ crystals with high surface energy will be partially dissolved into the IPA, giving up their mass so that large stable crystals can grow.^{32, 43} In this mechanism, the dissolution/de-nucleation rate of the unstable small MAPbI₃ crystals (depending on solubility capacity of solvent used) is extremely crucial because it determines the growth rate coefficient of stable large MAPbI₃ crystals.⁴⁴ In other words, the higher rate of de-nucleation/dissolution, the higher the rate of crystal growth, and thus larger crystal size could be obtained according to the following equation:

$$Z(D_N, G) = 1.1(D_N \times G)^{-0.5} \quad (2)$$

Where $Z(D_N, G)$ is the number of crystals per unit area, which is inversely proportional to the crystal size, D_N is de-nucleation rate of unstable small crystals and G is growth rate of stable large crystals.

To verify our hypothesis, we examined the effect of IPA-only post treatment on the morphology of MAPbI₃ film. We found that marginal grain size enhancement was observed with the MAPbI₃ film treated with IPA (Fig. S2) which is consistent with a recent study.³² We infer that the extremely low solubility of MAPbI₃ in IPA leads to a low rate of de-nucleation/dissolution of unstable small MAPbI₃ crystals, and thus inefficient coarsening process. It has been reported by Yang *et al.*, that methylammonium bromine (MABr) is able to tackle the problem as it favours the dissolution reaction (1).³² However, as stated, the intercalation of MABr and/or the I/Br exchange reaction could competitively take place, inhibiting the dissolution reaction, which in turn restricts effective Oswald ripening of MAPbI₃ film.

Similar to Br⁻ anion, SCN⁻ anion has stronger interaction with Pb²⁺ cation compared with I⁻ anion, thus one can expect that GU-SCN/IPA may follow the same grain coarsening mechanism as does by MABr/IPA.⁴⁵ However, as shown in Fig. S3, compared to MABr/IPA treatment, a MAPbI₃ film prepared by GUTS/IPA treatment is smoother with grain sizes two-folds larger on average and a narrower grain size distribution. In addition, micron-sized MAPbI₃ grains were easily formed by using the GUTS post treatment even without heat treatment (Fig. S4), indicating that the Ostwald ripening process enabled by the GUTS/IPA treatment is very effective. More interestingly, there is no concentration window for GUTS/IPA which is required to control the size of MAPbI₃ grains. Instead, the higher of the concentration of GUTS/IPA is used, the larger grain is observed (Fig 1 and Fig. S5). The above observations lead us to conclude that besides the dissolution-assisted effect of GUTS (as similar to MABr), there must be another source that aids the grain growth. There is a possibility that there is a reaction of GUTS/IPA and MAPbI₃ as follows:



The unstable HSCN (gas) is probably expelled from the film rapidly²⁸ since we could not detect any observable signal of sulfur by X-ray photoelectron microscopy (XPS), inductively coupled plasma spectroscopy (ICP-MS) and Fourier transform infrared spectroscopy (FTIR) (not shown here). The produced PbI_2 and GuI and/or a compound formed by them ($(\text{GuI})_x(\text{PbI}_2)_y$) accumulates at the surface and GBs of MAPbI_3 film as the secondary phases according to Fig. 1e-f. It is very likely that the as-formed CH_3NH_2 (MA) gas can dissolve MAPbI_3 crystal more efficiently than IPA, as suggested by previous reports.^{33, 46-49} We therefore propose that the formed MA (gas) enhances the rate of de-nucleation and increases the solubility of unstable small MAPbI_3 crystals which in turn facilitates the grain coarsening. Compared to conventional method of Ostwald ripening, the dissolution of small-sized MAPbI_3 crystals enabled by GUTS/IPA post treatment in this work is extremely fast and effective. As a result, the process is easier and more efficient to form large grain in the MAPbI_3 film. A comparison is outlined in Fig. 2.

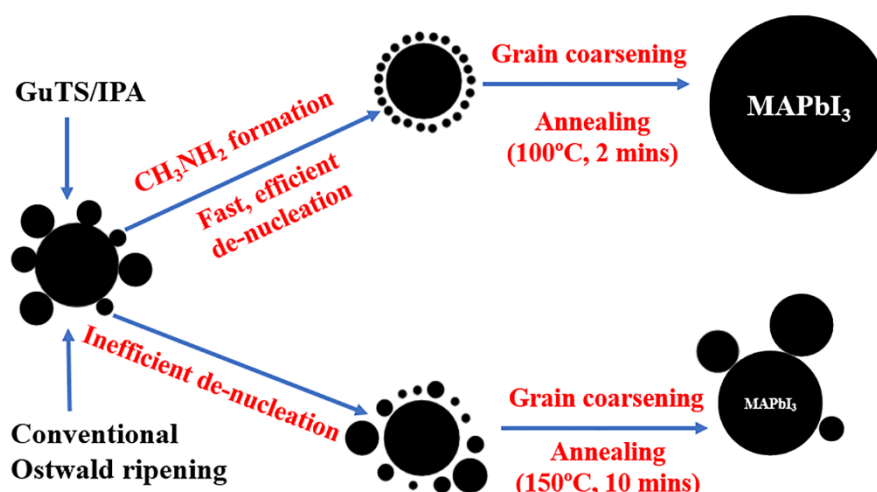


Fig. 2 Schematic illustration of Ostwald ripening process for the MAPbI_3 perovskite crystal growth assisted by GUTS in this work and by conventional method.

We have further found that incorporating GUTS into MAPbI_3 precursor solution does slightly enlarge domain size of MAPbI_3 films as illustrated in Fig. S6a-d. However, compared to GUTS post-treated MAPbI_3 film, those with GUTS additive in perovskite precursor are two-folds smaller in terms of domain size (Fig. S6-d versus Fig. 1c). Besides, adding GUTS additives to the MAPbI_3 precursor leads to the formation of a considerable impurities on surface of MAPbI_3 film as shown in Fig. S6(b, c, d), reducing the

optoelectronic quality of the films. We therefore propose that GUTS-assisted Ostwald ripening post-treatment could be the best procedure for fabricating MAPbI₃ film with the optimal morphology.

The absorption spectra of MAPbI₃ films prepared from different GUTS concentrations (Fig. S7) show characteristic absorption onset at around 770 nm regardless of GUTS concentration, indicating that the band gap of MAPbI₃ is not affected by GUTS treatment. Below 4 mg/mL GUTS in IPA, the MAPbI₃ film has slightly higher light absorption coefficient than the pristine one, suggesting an improved film crystallinity. Beyond that, the absorbance of MAPbI₃ films is reduced which can be ascribed to the formation of a substantial amount of secondary phases (as shown in Fig. 1d-f).

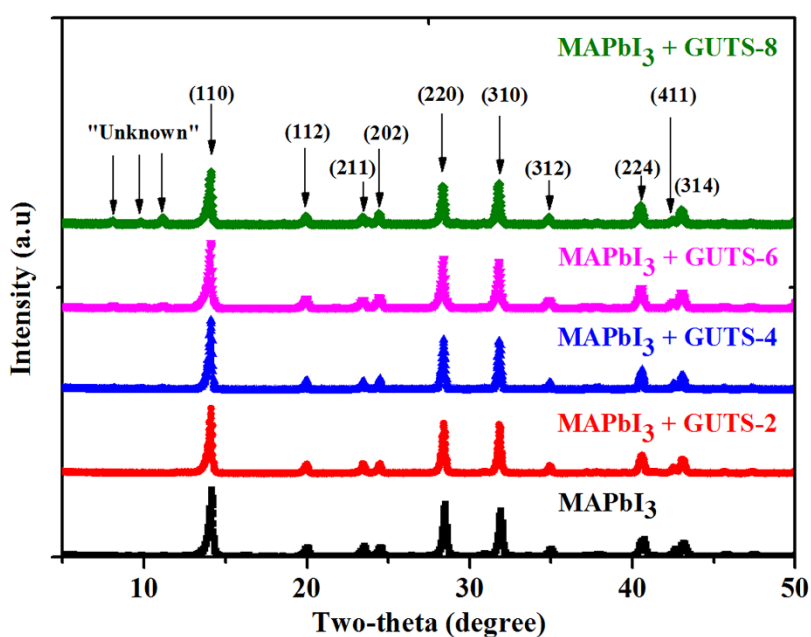


Fig. 3 XRD patterns of MAPbI₃ films with and without GUTS treatment.

X-ray diffraction (XRD) was used to structurally characterize the material phases in the MAPbI₃ films with and without GUTS-treatment (Fig. 3). As indexed in Fig. 3, the main peaks of all the films can be well assigned to the reflection of MAPbI₃. It should be noted that the peak intensity ratio between (220) and (310) plane changes as concentration of GUTS increases, indicating the change in growth rate of the MAPbI₃ crystals at different crystallographic directions. This change could also be associated with the formation of secondary phases, which might alter crystal growth rate at particular direction. In addition, weak diffraction peaks at 8.12°, 9.82°, and 11.2° are observed when the concentration of GUTS for post

treatment is over 6 mg/mL, which should be assigned to the observed secondary phases observed in top-view SEM images (Fig. 1d-f).

To identify these unknown peaks, we collected XRD patterns of Gu-SCN (powder), GuI (powder) and “GuPbI₃” films (Fig. S9) because they are most likely the by-products of the reaction (3). A yellow GuPbI₃ film was prepared by spin-coating a precursor of GuI:PbI₂:DMSO (1:1:1, molar ratio) in DMF followed by annealing at 100°C for 2 mins. It is worth highlighting that no peak from Gu-SCN was found in the diffraction patterns of GUTS-treated MAPbI₃, indicating that Gu-SCN was completely consumed in the reaction (3) with MAPbI₃. In addition, the unknown peaks do not match any diffraction patterns of GuI or “GuPbI₃” phases either, suggesting that the by-products induced by reaction (3) (probably GuI and PbI₂) combined with each other to form intermediate phases (GuI)_x(PbI₂)_y, located at the GBs or/and the surface of MAPbI₃ film, in accordance with previous observation.⁴⁰ According to Ref 40, a tiny amount of the intermediate phases are not harmful to the resulting device performance, instead they could passivate the GBs/surface defects. We therefore have not intentionally excluded the subtle amount of the intermediate phases in the MAPbI₃.

Having shown that micron-sized MAPbI₃ films could be easily obtained through manipulation of the concentration of GUTS in the post treatment process, we proceed to integrate these films into solar cells. The performance of these MAPbI₃ devices is compared and listed in Table 1.

Table 1: Characteristic photovoltaic parameters of MAPbI₃ solar cells with different concentration of GUTS in the post treatment process.

Cell	PCE (%)	J_{sc} (mA/cm ²)	V_{oc} (V)	FF (%)	R_s (Ω /cm ²)	R_{sh} (k Ω /cm ²)
Reference	12.7±1.8	20.7±0.8	1.02±0.03	60±5	53±5	1.5±0.7
GUTS-2	13.5±1.5	20.7±0.5	1.04±0.03	63±4	44.0±3.3	3.4±0.9
GUTS-4	15.2±0.9	21.1±0.3	1.06±0.02	68±3	33±2.9	9.1±1.1
GUTS-6	13.9±1.1	20.6±0.38	1.04±0.03	65±2	38.1±3.8	5.2±1.4
GUTS-8	10.9±2	19.5±0.45	0.97±0.05	58±6	62.8±6.6	2.1±1.7

When the MAPbI₃ absorber layer was post treated with GUTS/IPA with concentration up to 4 mg/mL, the average power conversion efficiency (PCE) of the solar cells increases by nearly 20%, from 12.7±1.8% to 15.2±0.9%, mainly due to the increase of open-circuit voltage (V_{oc}), from average 1.02±0.03 V to average 1.06±0.02 V, and fill-factor (FF), from 60±5% to 68±3% (Table 1). These results are consistent with the observed reduction of the series resistance (R_s) by ~37%, and the significant increase of shunt resistance (R_{sh}) by six-folds of the devices (Table 1). The lower R_s implies the perovskite film has lower inter-particle contact resistance, higher internal film conductivity and better contacts with selective charge transport layers, while the larger R_{sh} indicates the greater surface and GBs passivation.^{10, 50} In addition, slightly higher current densities are also witnessed as the GUTS precursor concentration increases, which is consistent with the UV-visible light absorption spectrum (Fig. S7 and Fig. S8). However, further increase of the GUTS content up to 8 mg/mL leads to a dramatic reduction of all photovoltaic parameters even though grain size of the MAPbI₃ film was increased (Table 1). This is probably related to the formation of large quantities of secondary phases in the film and the increase in the roughness of top surface of the perovskite film as shown in Fig. 1d-e), which are believed to not only reduce the light absorption

of MAPbI₃ film (Fig. S7), but also cause current leakage, resulting in low solar cell performance. The above SEM, XRD, UV-vis absorbance and PCE data suggest that the best-performing device (GUTS-4-MAPbI₃ based solar cell, 15.2±0.8%) is only enabled by a synergetic effect of MAPbI₃ film morphology, crystallinity and the amount of incorporated secondary phases. In the next section, we will provide more insights into the effect of GUTS-4 treatment on the photovoltaic performance and electronic properties of the resultant solar cells.

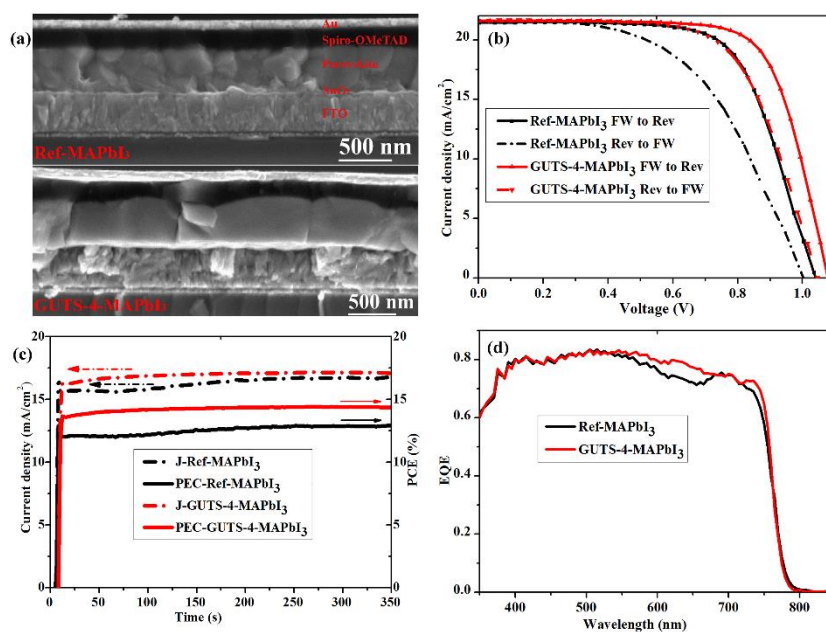


Fig. 4 (a) Cross-sectional SEM images of the completed pristine MAPbI₃ and GUTS-4-MAPbI₃ solar cells; (b) *J-V* curves of MAPbI₃ solar cells with and without GUTS-4 treatment, under reverse and forward voltage scan; (c) stabilized photocurrent measurement at a bias voltage (0.78 V for the pristine MAPbI₃ device and 0.85 V for the GUTS-4-MAPbI₃ device) at maximum power point and stabilized power output under 1 sun light illumination; and (d) corresponding IPCE spectra.

Fig. 4a shows the detailed cross-sectional SEM of n-i-p planar solar cell with Ref-MAPbI₃ (top) and with GUTS-4-MAPbI₃ (bottom) absorber layer. In general, both cells consist of a MAPbI₃ absorber (~400 nm) inserted between a SnO₂ electron transport layer (~40 nm) and a Spiro-OMeTAD based hole transport layer (~200 nm with current collector of FTO glass and gold. As

illustrated, unlike the Ref-MAPbI₃ layer which comprises of grains with variable sizes, the GUTS-4-MAPbI₃ layer is consisted of a monolithic MAPbI₃ layer, attesting the low density of GBs/defects, and thus superior solar cell performance.

The significant reduction of GBs/defects within the MAPbI₃ film leads to a 12.2% improvement in the performance of the champion devices (Fig. 4b). In particular, the Ref-MAPbI₃ device shows a maximum PCE of 14.59% (11.5%), with a V_{oc} of 1.041 V (1.0 V), a J_{sc} of 21.46 mA/cm² (21.5 mA/cm²) and FF of 65% (53%) under reverse (forward) scan respectively. In contrast, the GUTS-4-MAPbI₃ solar cell produced a PCE 16.37% (14.21%), with a V_{oc} of 1.077 V (1.043 V), a J_{sc} of 21.6 mA/cm² (21.63 mA/cm²) and FF of 70.4% (65%) under a similar measurement protocol. The performance enhancement extracted from J - V curves is also in line with the stabilized PCE shown in Fig. 4c, where the stabilized PCE of the MAPbI₃ cell is raised by 11.6% (from ~12.9% to ~14.5%) with GUTS-4 post treatment. More interestingly, along with device performance enhancement, the degree of current voltage hysteresis is also suppressed with value decreases from 21% of Ref-MAPbI₃ solar cell to only 13% of GUTS-4-MAPbI₃ counter-part.

The Ref-MAPbI₃ cell exhibits a high external quantum efficiency (EQE) (over 70%) across a broad wavelength range from 400 nm to 750 nm, with the maximum EQE value of ~83% at ~500 nm and a depression around 650 nm (Fig. 4d, black curve). The depression in EQE spectrum is common for planar MAPbI₃ thin-film solar cell, possibly due to the light-field interference associating with short charge carrier lifetime of the material.⁵¹ Remarkably, this depressed region is greatly filled up when the MAPbI₃ absorber was treated with GUTS-4 as shown in Fig. 4d (red curve), indicating an improvement in charge carrier lifetime. As a result, the integrated current density of GUTS-4-MAPbI₃ cell is greater than the Ref-MAPbI₃ cell (19.3 mA/cm² vs 18.8 mA/cm², respectively), which is consistent with the J_{sc} obtained in the J - V measurement.

Since the main enhancement of device performance are V_{oc} and FF , owing to the plunge of the series resistance and the soar of shunt resistance of the device (Table 1), we speculate that the GUTS

post treatment probably enhances the electronic properties of the perovskite film towards accelerating the charge transport rate and reducing charge recombination rate. Aiming to interpret the dynamics of charge transfer and recombination in the solar cell devices and to verify this hypothesis, electrochemical impedance spectroscopy (EIS) measurements were conducted at open-circuit condition ($\sim V_{oc}$ bias) with various light illumination intensities.⁵²

The EIS spectrum shows two semi-circles in the Nyquist plots. The EIS spectrum can be well-fitted using the equivalent circuit shown in Fig. 5a (bottom right). In the equivalent circuit, the capacitive element, C_g , at high-frequency region, represents the dielectric properties of the perovskite absorber layer while the resistive element, R_3 , is associated with the transport resistance of electron along the perovskite layer interface. C_s and R_I obtained at low-frequency region correspond to the properties of perovskite interfaces. More specifically, C_s serves as ionic accumulation capacitance (in dark) or electronic accumulation capacitance (under light illumination) at electrode interfaces, while R_I coupled with R_3 determines surface recombination resistance.⁵²⁻⁵⁴ The extracted information from the EIS fitting for both types of devices shows that the bulk capacitance, C_g , is unchanged regardless of light illumination intensities but the interfacial charge accumulation capacitance, C_s , increases linearly with the illumination intensity (Fig. 5b), while the resistive parameters follow the inverse trend with the illumination intensity (Fig. 5c).

It is found that, compared to the Ref-MAPbI₃ solar cell, the interfacial capacitance, C_s , of the GUTS-4-MAPbI₃ is slightly higher under light illumination, indicating a slightly higher level of electronic accumulation or higher carrier density at electrode interface. It has been reported that V_{oc} of perovskite solar cell follows the trend with the C_s because under light illumination, the C_s can be proportionally associated with the density of minority carriers at perovskite interface.⁵⁴ The EIS result observed in the darkness shows that both devices have similar C_s , implying the same level of ionic accumulation at perovskite interface (Fig.S10 and Table 1, supplementary). This result also suggests that the higher carrier density at SnO₂/perovskite interface (the perovskite/Spiro-OMeTAD interface can be regarded as ohmic contact) of the GUTS-4-MAPbI₃ cell could be due to the longer carrier lifetime. This interpretation of the C_s is in good agreement with the slight increase of open-circuit voltage observed with the GUTS-4-MAPbI₃ samples.

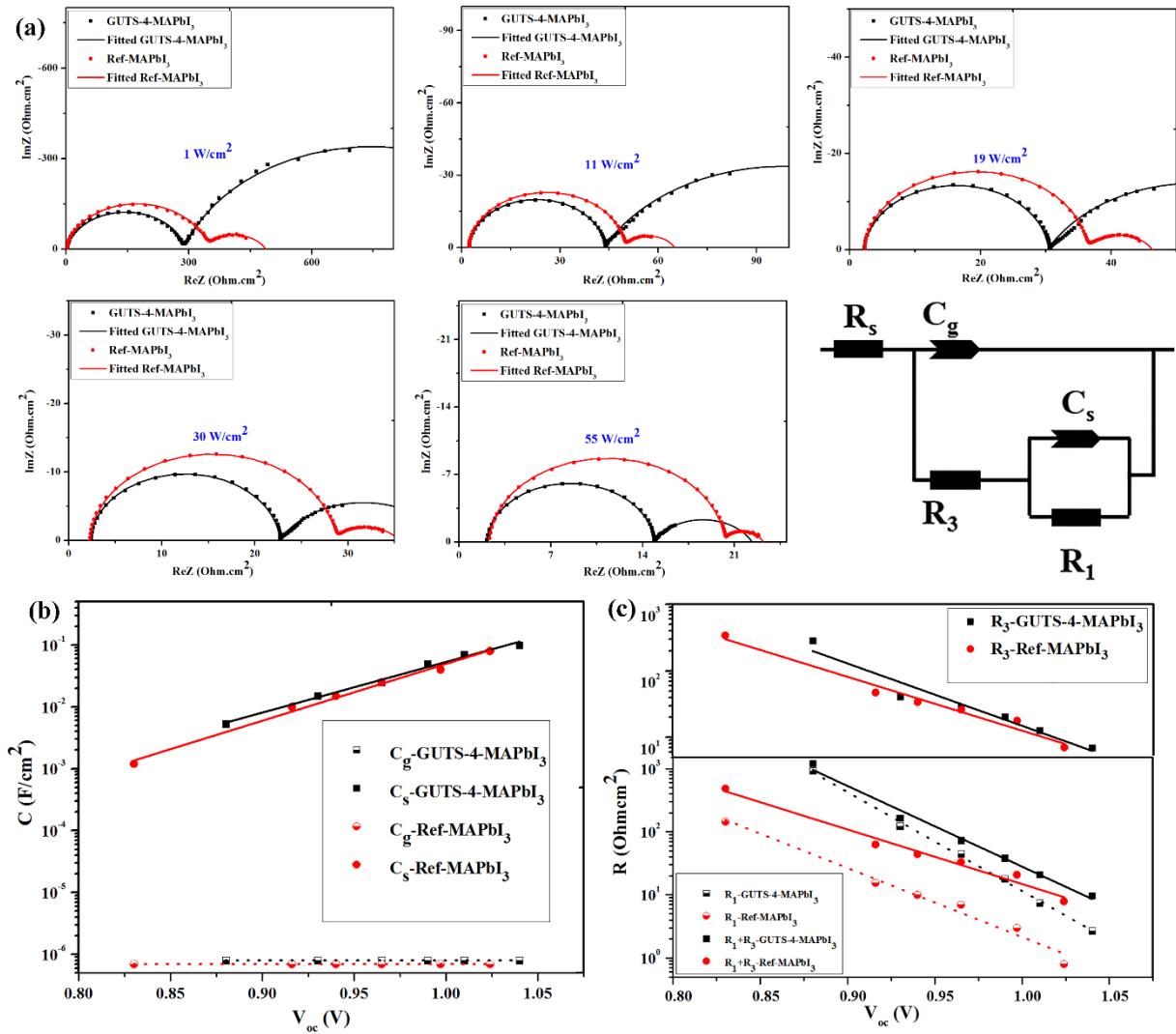


Fig. 5 (a) Nyquist plots of Ref-MAPbI₃ solar cell (circle red) and fitting (red line), and GUTS-4-MAPbI₃ (square black) and fitting (black line) under various light illumination intensities; and equivalent circuit (bottom right). (b) Extracted capacitance and (c) resistance.

Fig. 5c illustrates that the resistance, R_3 , of the GUTS-4-MAPbI₃ solar cell is always smaller than that of the reference cell and the ratio of the R_3 of the latter to the former is in the range of 1.01 and 1.29. For example, at 1 mW/cm², the ratio is ~1.2 while at 11 mW/cm², 19 mW/cm², 30 mW/cm², 55 mW/cm² and 100 mW/cm² illumination intensities, this value is 1.15, 1.21, 1.29, 1.4 and 1.01, respectively (Fig. 5c). The results in turn indicate that the GUTS-4-MAPbI₃ film has lower bulk resistance and better contact with SnO₂ layer than the pristine MAPbI₃ film. Combining this to the fact that the MAPbI₃ film with GUTS-4 treatment has much less GBs compared to the reference MAPbI₃ film (Fig. 1a, c), we conclude that, the GBs probably serve as barriers for charge transport (in both the bulk and the perovskite interfaces). Once

these barriers are eliminated, charge carrier can transport to SnO₂ more efficiently. However, compared to the reference cell, the GUTS-4-MAPbI₃ cell has higher low frequency resistance, R_1 , (Fig. 5c). Interestingly, the surface recombination resistance which is the sum of R_1 and R_3 , $R_1 + R_3$, of the GUTS-4-MAPbI₃ device is larger than the pristine sample, being consistent with the enhanced device performance (Table 1). For instance, compared to the Ref-MAPbI₃ device, the surface recombination resistance of the GUTS-4-MAPbI₃ counter-part is ~2.5-folds, 2.6-folds, 1.66-folds, 1.15-folds and 1.23-folds larger as light illumination intensity increased from 1 mW/cm² to 100 mW/cm², respectively (Fig. 5c). The above EIS results indicate that the trap-related recombination is significantly suppressed in GUTS-4-MAPbI₃ film, and its surface is efficiently passivated compared to the Ref- MAPbI₃.

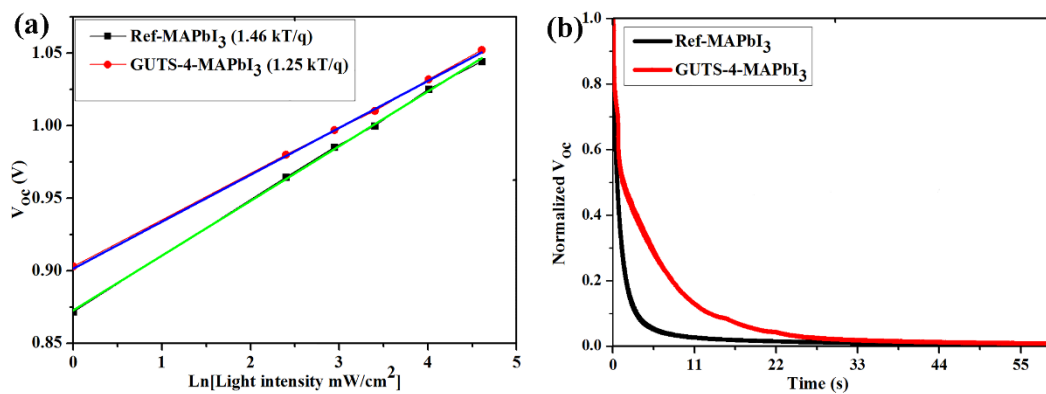


Fig. 6 Open-circuit voltage (V_{oc}) as a function of light illumination intensity (a); and V_{oc} decay of the reference and GUTS-4-MAPbI₃ cells (b).

By analysing the open-circuit voltage as a function of light illumination intensity (Fig. 6a), we confirm that besides direct/radiative recombination, trap-assisted recombination exists in both the GUTS-4-MAPbI₃ and the reference solar cells since the ideality factor, m , of both cells ($m_1=1.46$ for the reference cell and $m_2=1.25$ for GUTS-4-MAPbI₃ cell) are greater than 1.^{55, 56} However, compared to the reference cell, the contribution of trap-assisted recombination is smaller in the GUTS-4-MAPbI₃ cell since $m_2 < m_1$, which is consistent with the value of enhanced shunt resistance and surface charge recombination resistance of the device as shown above.

Photovoltage decay is another effective method to investigate the recombination in perovskite solar cells. Fig. 6b illustrates the decay of V_{oc} as a function of time upon removal of light illumination. Basically, a

longer decay time of V_{oc} is indicative of longer charge carrier life time, as can be found by the following equation:⁵⁷

$$\tau_{ir} = \left(-\frac{1}{V} \frac{dV}{dt} \right)^{-1} \quad (4)$$

Where τ_{ir} is instantaneous relaxation time and dV/dt is the decay rate of V_{oc} . The V_{oc} of both types of cells drops to nearly zero within 25 seconds (Fig. 6). However, the GUTS-4-MAPbI₃ cell shows a much slower decay pattern (Fig. 6b, red curve), indicative of a prolonged recombination period or longer carrier relaxation time, in accordance with the larger surface capacitance, C_s observed in EIS (Fig. 5b). This result again confirms that the perovskite interface contacts have been improved with GUTS-4 post treatment. We are aware that light soaking prior to V_{oc} decay measurement can give an electrostatic contribution to the photovoltage of PSCs as reported previously.⁵⁸ In our case, an illumination pulse with duration of ~2 s provided by a white light LED with low intensity equivalent to 0.15 sun was used. Therefore, we can rule out the electrostatic-assisted photovoltage decay effect. As suggested by Gottesman *et al.*, that the interface of perovskite and selective contacts play a key role in the V_{oc} decay pattern,⁵⁸ we correlate this prolonged duration of voltage decay to the improvement of perovskite interfaces besides the decrease of defect in the bulk of the perovskite layer.

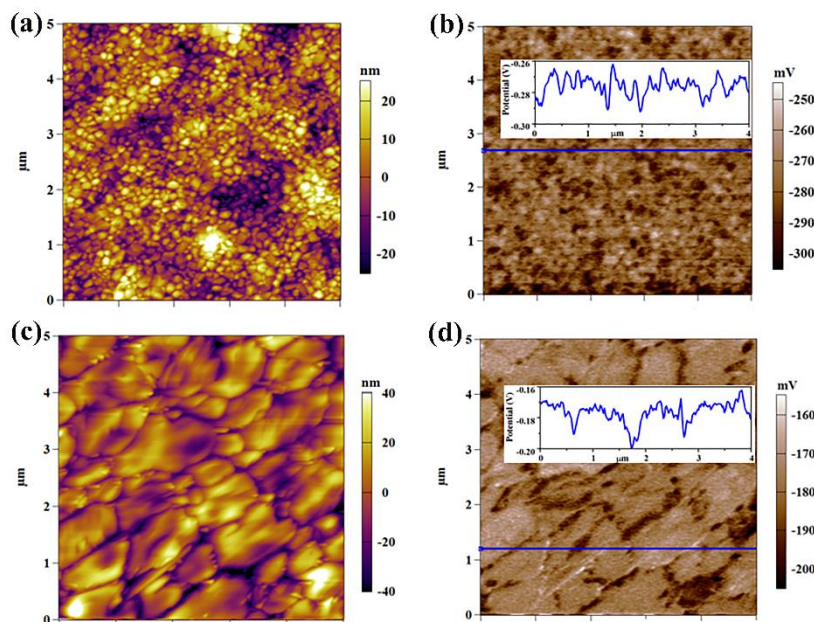


Fig. 7 KPFM topography of (a) Ref-MAPbI₃/SnO₂/FTO-glass film and (c) GUTS-4-MAPbI₃/SnO₂/FTO-glass film; (b) and (d) are the corresponding contact potential difference (CPD) images.

We carried out Kelvin probe force microscopy (KPFM) measurements to study the electronic properties of MAPbI₃ surface. KPFM topography of these films showing in Fig. 7a, c is in excellent agreement with the top-view SEM shown in Fig. 1a, c. The micrographs show that although the grain coarsening slightly increases the surface roughness of the MAPbI₃ film from 8.87 nm to 12.91 nm, it effectively enlarges the grain size of GUTS-4-MAPbI₃ while maintaining film uniformity over a large area (5×5 μm²) (Fig. 7a, c).

The mean value of contact potential differences (CPD) of the GUTS-4-MAPbI₃ film is 0.1 V higher than the reference sample, indicating a down-shifted of the electron quasi-Fermi level of the perovskite film (Fig. 7b, d). It has been reported that the conduction band of SnO₂ layer is ~170 meV below that of the MAPbI₃ perovskite film.⁵⁹ Therefore, the increase of the electron quasi-Fermi level of the GUTS-4-MAPbI₃ leads to improved energy band alignment with adjacent with SnO₂, which is likely to facilitate the electron extraction along the MAPbI₃/SnO₂ interface, and thus higher V_{oc} and FF of the resultant device.

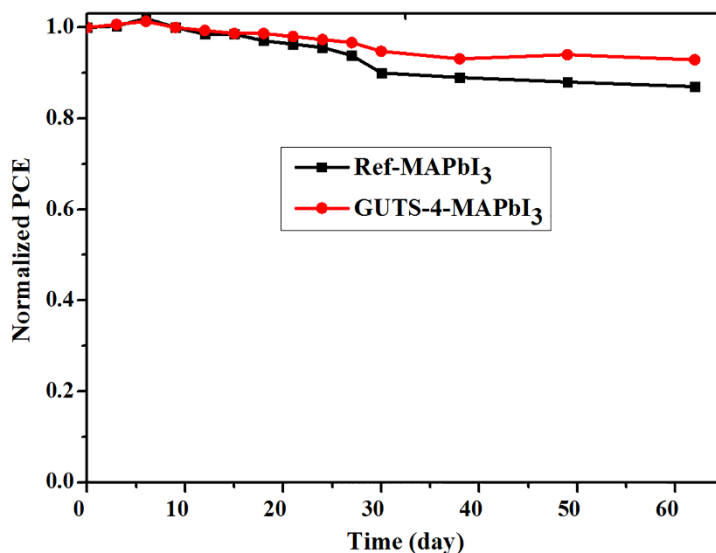


Fig. 8 Moisture stability of the ref-MAPbI₃ cell (black curve) and the GUTS-4-MAPbI₃ cell (red curve).

The solar cells were stored in a desiccator with a relative humidity of ~33% in dark.

The stability of the reference MAPbI₃ device and the GUTS-4-MAPbI₃ based device were monitored by measuring their performance every three days (Fig. 8). Both devices show similar stability patterns, where

the performance increases in the first few days due to the hole-conductivity enhancement of Spiro-OMeTAD in air. After this, the performance slowly decreases, which is correlated to the degradation of MAPbI₃ layer.⁴ After 60 days storage in dry air (*RH* 33%), the reference cell retained 87% of PCE while the GUTS-4-MAPbI₃ cell preserved 93% of PCE (Fig. 8). The better moisture stability of the GUTS-4-MAPbI₃ cell, as compared to the reference cell could be associated with the better perovskite film quality with less defects, which reduces the probability of water penetration and degradation of perovskites.^{46, 50}

Conclusions

We have here demonstrated the effect of GUTS-assisted Ostwald ripening post treatment on the morphology and charge recombination of MAPbI₃ film and its interfaces. It was discovered that the concentration of GUTS precursor plays a key role on crystal structure, optical properties, morphology and surface potential of perovskite film. An optimal content of GUTS precursor (GUTS-4, 4 mg/mL) induced micron-sized MAPbI₃ film, effectively passivates GBs of MAPbI₃ film from recombination and positively shifted the film surface potential as confirmed by KPFM. Further investigation of device recombination kinetics showed that the GUTS-4-MAPbI₃ solar cells have higher interfacial charge recombination resistance and lower charge extraction resistance compared to that of the reference MAPbI₃ cell, which is in excellent agreement with the results of V_{oc} decay studies. In addition, compared to the pristine MAPbI₃, the GUTS-4-MAPbI₃ device had lower ideality factor, suggesting less non-radiative recombination. The resultant performance of planar PSCs was improved by 20%, from average PCE of 12.7% for the reference MAPbI₃ to 15.2% for GUTS-4-MAPbI₃ cells, mainly due to the enhancement of the V_{oc} and FF . In addition, devices with GUTS-4-MAPbI₃ absorber showed less-pronounced hysteresis in the J - V curve and were more inert to moisture than untreated films. This work provides new insight into the mechanism that governs the morphology and optoelectronic properties of MAPbI₃ film for high energy conversion efficiency.

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