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**Astronomy
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Photodissociation of interstellar N₂[★]

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ABSTRACT

Context. Molecular nitrogen is one of the key species in the chemistry of interstellar clouds and protoplanetary disks, but its photodissociation under interstellar conditions has never been properly studied. The partitioning of nitrogen between N and N₂ controls the formation of more complex prebiotic nitrogen-containing species.

Aims. The aim of this work is to gain a better understanding of the interstellar N₂ photodissociation processes based on recent detailed theoretical and experimental work and to provide accurate rates for use in chemical models.

Methods. We used an approach similar to that adopted for CO in which we simulated the full high-resolution line-by-line absorption + dissociation spectrum of N₂ over the relevant 912–1000 Å wavelength range, by using a quantum-mechanical model which solves the coupled-channels Schrödinger equation. The simulated N₂ spectra were compared with the absorption spectra of H₂, H, CO, and dust to compute photodissociation rates in various radiation fields and shielding functions. The effects of the new rates in interstellar cloud models were illustrated for diffuse and translucent clouds, a dense photon dominated region and a protoplanetary disk.

Results. The unattenuated photodissociation rate in the Draine (1978, *ApJS*, 36, 595) radiation field assuming an N₂ excitation temperature of 50 K is $1.65 \times 10^{-10} \text{ s}^{-1}$, with an uncertainty of only 10%. Most of the photodissociation occurs through bands in the 957–980 Å range. The N₂ rate depends slightly on the temperature through the variation of predissociation probabilities with rotational quantum number for some bands. Shielding functions are provided for a range of H₂ and H column densities, with H₂ being much more effective than H in reducing the N₂ rate inside a cloud. Shielding by CO is not effective. The new rates are 28% lower than the previously recommended values. Nevertheless, diffuse cloud models still fail to reproduce the possible detection of interstellar N₂ except for unusually high densities and/or low incident UV radiation fields. The transition of N → N₂ occurs at nearly the same depth into a cloud as that of C⁺ → C → CO. The orders-of-magnitude lower N₂ photodissociation rates in clouds exposed to black-body radiation fields of only 4000 K can qualitatively explain the lack of active nitrogen chemistry observed in the inner disks around cool stars.

Conclusions. Accurate photodissociation rates for N₂ as a function of depth into a cloud are now available that can be applied to a wide variety of astrophysical environments.

Key words. astrochemistry – stars: formation – molecular processes – interplanetary medium – photon-dominated region (PDR) – ultraviolet: planetary systems

1. Introduction

Nitrogen is one of the most abundant elements in the universe and an essential ingredient for building prebiotic organic molecules. In interstellar clouds, its main gas-phase reservoirs are N and N₂, with the balance between these species determined by the balance of the chemical reactions that form and destroy N₂. If nitrogen is primarily in atomic form, a rich nitrogen chemistry can occur leading to ammonia, nitriles and other nitrogen compounds. On the other hand, little such chemistry ensues if nitrogen is locked up in the very stable N₂ molecule. The latter situation is similar to that of carbon with few carbon-chain molecules being produced when most of the volatile carbon is locked up in CO (Langer & Graedel 1989; Bettens et al. 1995).

Direct observation of extrasolar N₂ is difficult because, unlike CO, it lacks strong pure rotational or vibrational lines. N₂ is well studied at various locations within our solar system through its electronic transitions at ultraviolet wavelengths (e.g., Strobel 1982; Meier et al. 1991; Wayne 2000; Liang et al. 2007) and a detection in interstellar space has been claimed through UV absorption lines in a diffuse cloud toward the bright background star HD 124314 (Knauth et al. 2004). In dense clouds well shielded from UV radiation, most nitrogen is expected to exist as N₂ (e.g., Herbst & Klemperer 1973; Woodall et al. 2007) but can only be detected indirectly through the protonated ion N₂H⁺ (Turner 1974; Herbst et al. 1977) or its deuterated form N₂D⁺. N₂H⁺ emission is indeed widely observed in dense cores (e.g., Bergin et al. 2002; Crapsi et al. 2005), star-forming regions (Fontani et al. 2011; Tobin et al. 2012), protoplanetary disks (Dutrey et al. 2007; Öberg et al. 2010) and external galaxies

* Appendices are available in electronic form at <http://www.aanda.org>

(Mauersberger & Henkel 1991; Meier & Turner 2005; Muller et al. 2011).

Photodissociation is the primary destruction route of N_2 in any region where UV photons are present. Current models of diffuse and translucent interstellar clouds are unable to reproduce the possible detection of N_2 for one such cloud (Knauth et al. 2004). One possible explanation is that the adopted N_2 photodissociation rate is incorrect. Even in dense cores, not all nitrogen appears to have been transformed to molecular form (Maret et al. 2006; Daranlot et al. 2012). Observations of HCN in the surface layers of protoplanetary disks suggest that the nitrogen chemistry is strongly affected by whether or not a star has sufficiently hard UV radiation to photodissociate N_2 (Pascucci et al. 2009). Thus, not only the absolute photodissociation rate but also its wavelength dependence is relevant. All of these astronomical puzzles make a thorough study of the interstellar N_2 photodissociation very timely.

In contrast with many other simple diatomic molecules, the photodissociation of interstellar N_2 has never been properly studied (van Dishoeck 1988; van Dishoeck et al. 2006). The reason for this is that the photodissociation of N_2 , similarly to CO, is initiated by line absorptions at wavelengths below 1000 Å (1100 Å for CO), where high-resolution laboratory spectroscopy has been difficult. To compute the absolute rate and to treat the depth dependence of the photodissociation correctly, the full high-resolution spectrum of the dissociating transitions needs to be known. Because the absorbing lines become optically thick for modest N_2 column densities, the molecule can shield itself against the dissociating radiation deeper into the cloud. Moreover, these lines can be shielded by lines of more abundant species such as H, H_2 and CO. Until recently, accurate N_2 molecular data to simulate these processes were not available. Thanks to a concerted laboratory (e.g., Ajello et al. 1989; Helm et al. 1993; Sprengers et al. 2003, 2004, 2005; Stark et al. 2008; Lewis et al. 2008b; Heays et al. 2009, 2011) and theoretical (e.g., Spelsberg & Meyer 2001; Lewis et al. 2005a,b, 2008c,a; Haverd et al. 2005; Ndome et al. 2008) effort over the past two decades, this information is now available.

In this paper, we use a high resolution model spectrum of the absorption and dissociation of N_2 together with simulated spectra of H, H_2 and CO to determine the interstellar N_2 photodissociation rate and its variation with depth into a cloud. The effect of the new rates on interstellar N_2 abundances is illustrated through a few representative cloud models. In particular, the N_2 abundance in diffuse and translucent clouds is revisited to investigate whether the new rates alleviate the discrepancy between models and the possible detection of N_2 in one cloud (Knauth et al. 2004). The data presented here can be applied to a wide range of astrochemical models, including interstellar clouds in the local and high redshift universe, protoplanetary disks and exo-planetary atmospheres. The $^{14}N^{15}N$ photodissociation rate and isotope selective interstellar processes will be discussed in an upcoming paper (Heays et al. in prep.) and have been discussed in the context of the chemistry of Titan by Liang et al. (2007).

2. Photodissociation processes of N_2

2.1. Photoabsorption and photodissociation spectrum

The closed-shell diatomic molecule N_2 has a dissociation energy of $78\,715\text{ cm}^{-1}$ (9.76 eV, 1270 Å) (Huber & Herzberg 1979),

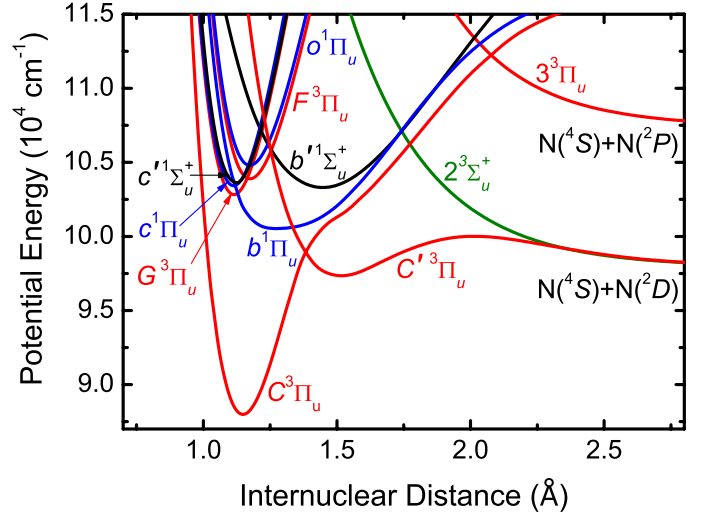


Fig. 1. Diabatic-basis potential-energy curves for electronic states of N_2 relevant to interstellar photodissociation. Blue curves: $^1\Pi_u$ states. Black curves: $^1\Sigma_u^+$ states. Red curves: $^3\Pi_u$ states. Green curve: $2^3\Sigma_u^+$ state. The energy scale is referenced to the $v = 0, J = 0$ level of the $X^1\Sigma_g^+$ ground state (not shown). The lowest dissociation limit, $N(^4S)+N(^2S)$ at $\sim 78\,715\text{ cm}^{-1}$ (9.76 eV), is beyond the scale of the figure. The H ionization potential of 13.6 eV provides an upper limit to the interstellar radiation field and corresponds to $109\,691\text{ cm}^{-1}$.

making it one of the most stable molecules in nature. Electric-dipole-allowed photoabsorption and predissociation in N_2 starts only in the extreme ultraviolet spectral region, at wavelengths shorter than 1000 Å. The molecular-orbital (MO) configuration of the $X^1\Sigma_g^+$ ground state of N_2 is

$$(1\sigma_g)^2(1\sigma_u)^2(2\sigma_g)^2(2\sigma_u)^2(1\pi_u)^4(3\sigma_g)^2. \quad (1)$$

Electric-dipole-allowed transitions from the ground state access only states of $^1\Pi_u$ and $^1\Sigma_u^+$ symmetry. In the region below the cutoff energy of the interstellar radiation field of $110\,000\text{ cm}^{-1}$ (13.6 eV, 912 Å), five such states are accessible: the c' and b' $^1\Sigma_u^+$ states, and the c , o , and b' $^1\Pi_u$ states. The c' , c , and o states (sometimes labeled c'_4 , c_3 , and o_3 ; respectively) have Rydberg character, the relevant transitions corresponding to single-electron excitations from the $3\sigma_g$ or $1\pi_u$ orbitals into a Rydberg orbital. On the other hand, the b' and b states are valence states of mixed MO configurations accessed by transitions in which one or two electrons are excited into antibonding orbitals. The relevant potential-energy curves (PECs) for these $^1\Pi_u$ and $^1\Sigma_u^+$ states are shown in Fig. 1, in blue and black, respectively. The c' and c states, whose PECs have the smallest equilibrium internuclear distance in Fig. 1, are the first members of Rydberg series converging on the ground state of the N_2^+ ion, $X^2\Sigma_g^+$, while the o state is the first member of the series converging on the first ionic excited state, $A^2\Pi_u$. In the case of the b' and b valence states, the extended widths of the corresponding PECs in Fig. 1 are due to the aforementioned configurational mixing. In addition, there are significant electrostatic interactions within the manifolds of a given symmetry, Rydberg-valence for $^1\Sigma_u^+$, and Rydberg-valence and Rydberg-Rydberg for $^1\Pi_u$, since the MO configurations of all of the isosymmetric states differ in exactly two of the occupied electron orbitals (Lefebvre-Brion & Field 2004). The PECs in Fig. 1 are shown in the diabatic (crossing) representation.

Most of the rovibrational levels of the singlet excited states are predissociated, i.e., the molecule is initially bound following photoabsorption, but then dissociates on timescales of a

nanosecond or less due to direct or indirect coupling to a dissociative continuum. For the $^1\Pi_u$ states considered here, spin-orbit coupling to the strongly-coupled and -predissociated $^3\Pi_u$ manifold (red PECs in Fig. 1), with ultimate dissociation via the C' state, provides the predissociation mechanism (Lewis et al. 2005a, 2008c), with a minor contribution from a crossing by the $2^3\Sigma_u^+$ state (green PEC in Fig. 1) at higher energies. For the $^1\Sigma_u^+$ states, two mechanisms are important (Heays 2011): first, a similar spin-orbit coupling to the $^3\Pi_u$ manifold, solely responsible for predissociation in the absence of rotation, and second, rotational coupling between the $^1\Sigma_u^+$ and $^1\Pi_u$ manifolds, followed by the $^1\Pi_u$ predissociation described above. For the wavelengths considered here, as implied by Fig. 1, these mechanisms result in primarily $N(^4S)+N(^2D)$ dissociation products, i.e., one of the nitrogen atoms is formed in an excited electronic state which decays on a timescale of 17 h into the ground state $N(^4S)$. This is consistent with the observations of Walter et al. (1993) who failed to detect direct $N(^4S)+N(^4S)$ dissociation products.

The line-by-line models previously used to compute the N₂ photodissociation rate require knowledge of the wavelengths, oscillator strengths, lifetimes, and predissociation probabilities of (transitions to) all rovibrational levels associated with the coupled excited singlet states. For the case of the isoelectronic molecule CO, molecular models have been built previously by specifying the term values, rotational and vibrational constants, oscillator strengths, Einstein A coefficients, and predissociation probabilities for each excited electronic state (e.g., van Dishoeck & Black 1988; Viala et al. 1988; Lee et al. 1996; Visser et al. 2009). These have allowed the rotationally-resolved absorption spectra of CO and its isotopologues to be constructed using simple scaling relations. Such models must be validated by a large quantity of laboratory data and have been shown to be incorrect when strong interactions occur between electronic states and their differing energetics. For the case of N₂, it is known that there are many wide-scale perturbations, together with rapid dependences of oscillator strengths and predissociation linewidths on rotational quantum number J , and strong, irregular isotopic effects. It is impossible to fully reproduce these effects using only a few spectroscopic constants.

The best way to simulate the N₂ spectrum, and the method employed here, is, at each energy, to solve the full radial diabatic coupled-channel Schrödinger equation (CSE) for the coupled electronic states described above, including all electrostatic, spin-orbit, and rotational couplings, using the quantum-mechanical methods of van Dishoeck et al. (1984). This is a *physically-based* technique, with great predictive powers which enables confidence in the computed spectrum in regions lacking experimental confirmation, even where perturbations are present. Furthermore, computations of isotopic spectra require only the change of a single parameter, i.e., the reduced molecular mass, in the molecular model: the results can be guaranteed since the underlying physics is the same for all isotopologues. The same cannot be said for the line-by-line models such as those employed for CO, which would also benefit from a CSE approach. The detailed CSE model for N₂ employed here has been described in Heays (2011)¹ incorporates earlier models of the $^1\Pi_u$ (Lewis et al. 2005a; Haverd et al. 2005) and $^3\Pi_u$ states (Lewis et al. 2008c) and has been tested extensively against laboratory data, including high-resolution spectra obtained at the SOLEIL synchrotron facility (Heays 2011; Heays et al. 2011). A complete discussion of the CSE model and a full listing of computed spectroscopic data is deferred to Heays et al. (in prep.).

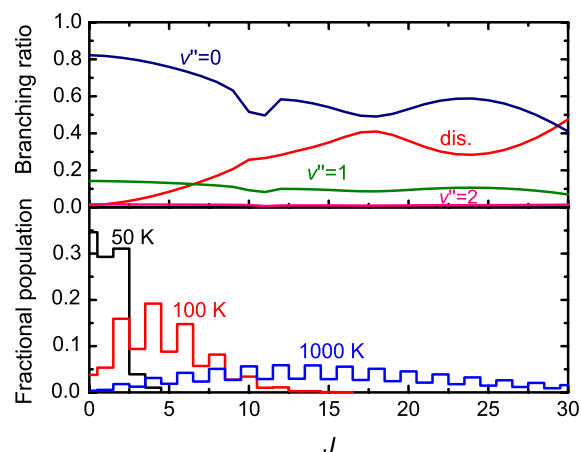


Fig. 2. *Top:* branching to various decay channels of the $c'(v' = 0)$ excited-state of N₂ as a function of total rotational quantum number, J . The figure includes spontaneous emission to several non-dissociative ground state vibrational levels ($v'' = 0, 1$ and 2) and decay due to predissociation (dis.). *Bottom:* fractional population of the N₂ ground state in its lowest vibrational level as a function of J and for several excitation temperatures. The 2:1 ratio of populations for even:odd J levels arises from the combined rotational and nuclear spin statistics.

For a given rotational-branch transition, combining the excited-state coupled-channel wavefunction with the X -state radial wavefunction and appropriate diabatic allowed transition-moment components yields the corresponding (continuous with wavelength) photoabsorption cross section, with the computed linewidths providing the required predissociation lifetime information. Total cross sections for a given temperature, assuming local thermodynamic equilibrium, are formed by summing the individual branch cross sections, weighted by appropriate Boltzmann and Hönl-London factors, and including rotational levels with J as high as 50.

CSE photoabsorption cross sections, σ_{abs} , are computed here over the wavelength range 912–1000 Å with a step size of 0.0001 Å, and for temperatures of 10, 50, 100, 500, and 1000 K. The Doppler broadening of the spectral lines is taken into account by convolution with a Gaussian profile having a thermal line width.

A 10% uncertainty is estimated for the total magnitude of the photoabsorption cross section and principally arises from the absolute uncertainty of the calibrating laboratory spectra (Haverd et al. 2005; Heays 2011). The laboratory measurements in question were recorded at 300 K or below, so the uncertainty may be somewhat larger for calculations employing an extrapolation to 1000 K. Additionally, 3% of the 1000 K ground state population will be in the first vibrational level, leading to a slight redistribution of the absorption cross section into hot bands. This is considered in the model calculations.

Photodissociation cross sections, $\sigma_{\text{pd}} = \eta \times \sigma_{\text{abs}}$, are obtained from the photoabsorption cross sections by comparing the predissociation and radiative lifetimes for each rovibrational level. The predissociation efficiency η is then given by $\eta = 1 - \tau_{\text{tot}}/\tau_{\text{rad}}$, where τ_{tot} is the inverse of the sum of the radiative and predissociation rates. For almost all transitions, $\eta \approx 1$: significant corrections for partial dissociation are needed only for the $b - X(1, 0)$ and $c' - X(0, 0)$ bands near 986 and 959 Å, respectively (Lewis et al. 2005b; Liu et al. 2008; Sprengers et al. 2004; Wu et al. 2012). For example, the top panel of Fig. 2 illustrates the CSE-computed branching ratio between spontaneous emission back to the ground state and dissociation as a function of

¹ Available on-line at <http://hdl.handle.net/1885/7360>

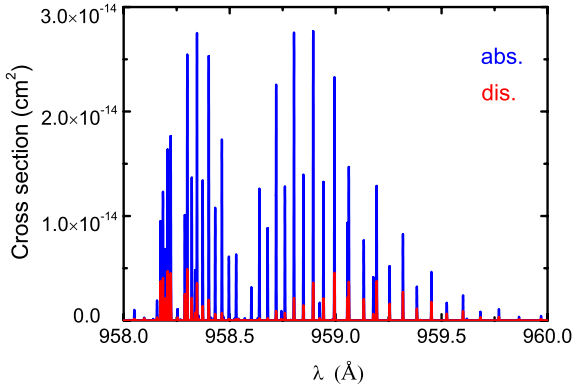


Fig. 3. CSE-calculated absorption cross section (blue) of the $c'(v' = 0)$ level of N_2 assuming an excitation temperature of 300 K. Also shown is a dissociation cross section (red) which has been corrected for the non-unity dissociation efficiency, η_J , of this band (see Fig. 2).

rotational level for $c' - X(0, 0)$. Such calculations were performed for all bands appearing between 955 and 991 Å. The difference between calculated absorption and dissociation cross sections for the very-strongly absorbing $c' - X(0, 0)$ band is demonstrated in Fig. 3, revealing a significant alteration of the band profile once the dissociation efficiency is considered.

The bottom panel of Fig. 2 shows the thermal population for various J levels of the ground vibrational state, assuming several temperatures. By comparing this with the top panel of Fig. 2 it can be seen that the dissociation fraction for this band will depend significantly on the temperature.

2.2. Photodissociation rates

The photodissociation rate, k_{pd} , of N_2 exposed to UV radiation can be calculated according to

$$k_{pd} = \int \sigma_{pd}(\lambda) I(\lambda) d\lambda \text{ s}^{-1}, \quad (2)$$

where the photodissociation cross section, σ_{pd} , is in units of cm^2 and I is the mean intensity of the radiation in photons $\text{cm}^{-2} \text{s}^{-1} \text{Å}^{-1}$ as a function of wavelength, λ , in units of Å. The unattenuated interstellar radiation field according to [Draine \(1978\)](#) is used in most of the following calculations and is given by

$$I(\lambda) = 3.2028 \times 10^{15} \lambda^{-3} - 5.1542 \times 10^{18} \lambda^{-4} + 2.0546 \times 10^{21} \lambda^{-5}. \quad (3)$$

Inside a cloud, self-shielding, shielding by H, H_2 , CO and other molecules, and continuum shielding by dust all reduce the photodissociation rate below its unattenuated value k^0 . The shielding function is defined to be

$$\Theta = k/k^0 \quad (4)$$

and can be split into a self-shielding,

$$\Theta_{SS} = \frac{\int I(\lambda) \exp[-N(N_2)\sigma_{abs}(\lambda)] \sigma_{pd}(\lambda) d\lambda}{\int I(\lambda) \sigma_{pd}(\lambda) d\lambda}, \quad (5)$$

and a mutual-shielding part,

$$\Theta_{MS} = \frac{\int I(\lambda) \exp[-N(X)\sigma_X(\lambda)] \sigma_{pd}(\lambda) d\lambda}{\int I(\lambda) \sigma_{pd}(\lambda) d\lambda}. \quad (6)$$

Here, $X = \text{H}, \text{H}_2$ or CO and N is the column density of the various species. A dust extinction term, $\exp(-\gamma A_V)$, can be written in place of the exponential term in Eq. (6) where A_V is the optical depth in magnitudes and γ depends on the assumed properties of the dust. This is further discussed in Sect. 3.3. In all cases, the integrals above are computed between 912 and 1000 Å.

3. Results

3.1. Unattenuated interstellar rate

Figure 4 shows model spectra of N_2 and $H_2 + H$ absorption for excitation temperatures of 50 and 1000 K. At 50 K the N_2 spectrum is made up of prominent well-separated bands. These represent excitation to a range of vibrational levels attributable to the five accessible electronic states. In contrast, the spectrum simulating a temperature of 1000 K includes the excitation of many more rotational levels and has few sizable windows between bands.

Unshielded photodissociation rates of N_2 immersed in a [Draine \(1978\)](#) field were calculated from the model photodissociation cross section using Eqs. (2) and (3), and assuming a range of excitation temperatures. These are plotted in Fig. 5 and listed in Table 1. The rate at 50 K is $1.65 \times 10^{-10} \text{ s}^{-1}$, where the uncertainty of 10% only reflects the uncertainty in the cross sections, not the radiation field (see below). This new value is 28% lower than the value of $2.30 \times 10^{-10} \text{ s}^{-1}$ recommended by [van Dishoeck \(1988\)](#). The latter estimate was based on the best available N_2 spectroscopy at the time, and has an order-of-magnitude uncertainty. For comparison, the unshielded photodissociation rate of N_2 at low T is around 35% smaller than that of CO computed by [Visser et al. \(2009\)](#).

Table 2 summarizes the contributions of individual bands to the total unattenuated dissociation rate. It is seen that the main contributions arise from bands 12, 21, 22, 23 and 24. Hence, the key wavelength ranges responsible for the photodissociation of N_2 are around 940 Å and between 957–980 Å.

The calculated unattenuated rate of N_2 increases with increasing temperature so that the value at 1000 K, $1.86 \times 10^{-10} \text{ s}^{-1}$, is 15% higher than for 10 K. This is largely due to a variable but overall increase with rotational quantum number J of the photodissociation branching ratios of the $c'(v = 0)$ and $b(v = 1)$ states. This can be seen in Fig. 2 for the $c'(v' = 0)$ state, where at 10 K all of the excited population is in levels with $J = 0-3$. These levels have a low predissociation probability and hardly contribute to the photodissociation rate. At higher temperatures, the excited population shifts to higher J , and at 1000 K the distribution maximum occurs around $J = 15-20$ for which the branching ratio to dissociation is much higher.

The rate obviously depends on the choice of radiation field. For the alternative formulations of [Habing \(1968\)](#), [Gondhalekar et al. \(1980\)](#) and [Mathis et al. \(1983\)](#), the unattenuated rates are 1.45, 1.34 and $1.51 \times 10^{-10} \text{ s}^{-1}$ at 50 K, respectively. Additionally, Table 1 considers the unattenuated rates of N_2 assuming different blackbody radiation fields. In these calculations, the intensities have been normalized such that the integrated values from 912–2050 Å are the same as those of the [Draine \(1978\)](#) field. The adopted dilution factors are 1.9×10^{-9} , 3.4×10^{-12} , 1.2×10^{-13} , 1.6×10^{-14} , and 1.6×10^{-16} for blackbody temperatures of 4000, 6000, 8000, 10 000 and 20 000 K, respectively. The value of the unattenuated rate of N_2 at 4000 K (cool star) is 6 orders of magnitude smaller than that at 20 000 K (hot star), and increases steeply with stellar effective temperature. The photodissociation rate of N_2 at 20 000 K is comparable to that in the Draine interstellar field, $1.65 \times 10^{-10} \text{ s}^{-1}$. The

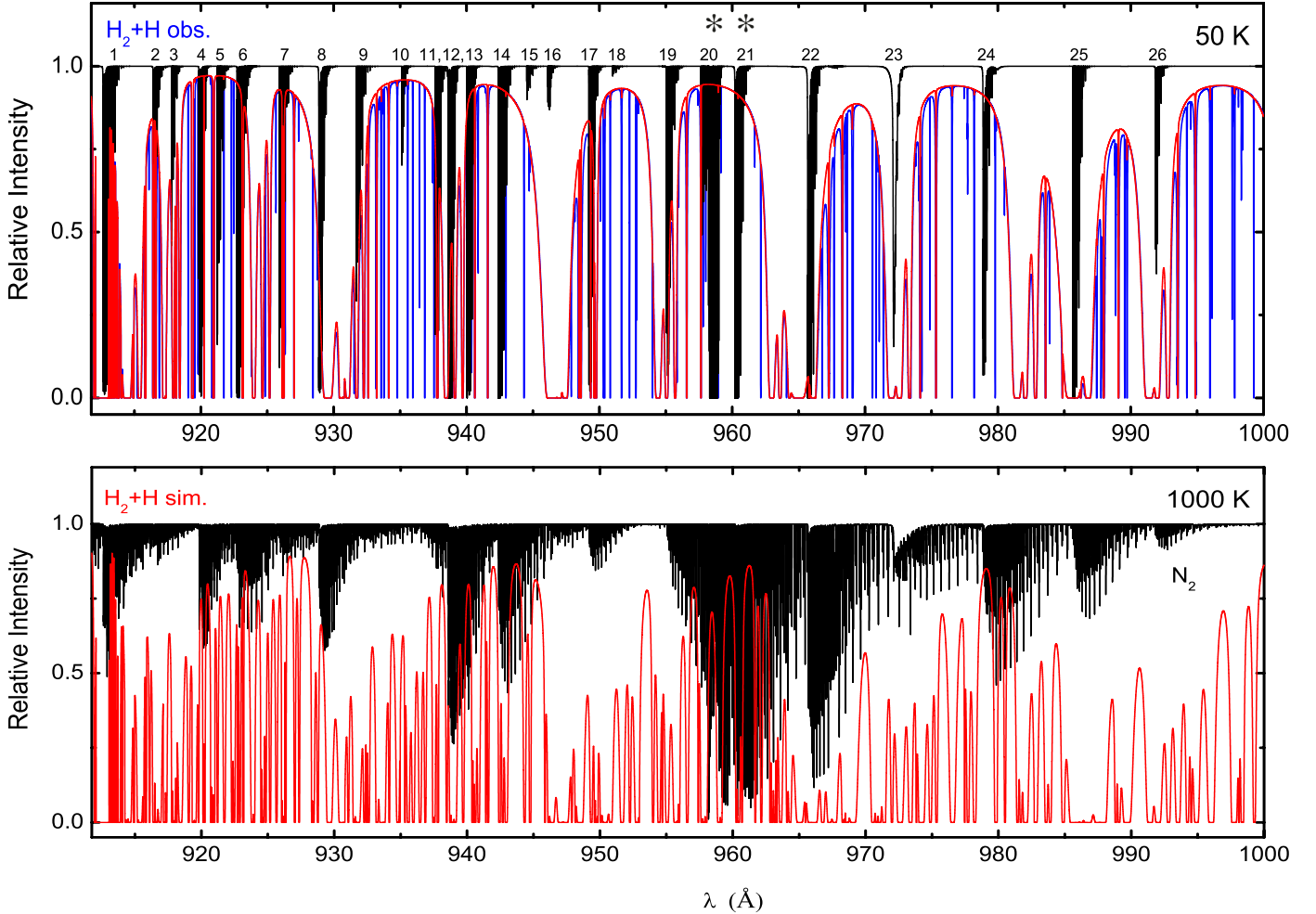


Fig. 4. Simulated absorption spectra for N₂ (black) and H₂+H (red) in the wavelength range 912–1000 Å assuming thermal excitation temperatures of 50 (*top*) and 1000 K (*bottom*). The column density of N₂ is 10¹⁵ cm⁻² and values for H₂ and H are taken to be half of the observed column densities in the well-studied diffuse cloud toward ζ Oph, as is appropriate for the center of the cloud: $N(\text{H}_2) = 2.1 \times 10^{20}$ and $N(\text{H}) = 2.6 \times 10^{20}$ cm⁻². The model H₂ Doppler width is 3 km s⁻¹. Also shown is the H₂ + H absorption spectrum (blue) toward ζ Oph using the observed column densities for individual J levels, showing enhanced non-thermal excitation of H₂ in the higher J levels. The asterisks indicate the $c'(0)$ (Band 20) and $c(0)$ (Band 21) bands, respectively, detected in absorption toward HD 124314.

calculated photodissociation rates for temperatures of 4000 and 10 000 K are close to those recommended by van Dishoeck et al. (2006).

3.2. Self-shielding

Although self-shielding is generally less important than mutual shielding for the case of N₂ (see Sect. 3.3), it is potentially important in protoplanetary disks and has been proposed to be responsible for the enrichment of ¹⁵N in bulk chondrites and terrestrial planets (Lyons 2009, 2010). In this work, we compute the self-shielding functions of N₂ at excitation temperatures of 10, 100 and 1000 K using the model absorption spectrum. Since this spectrum is constructed using thermal line widths and no turbulent broadening, it provides the maximum amount of shielding. For reference, the thermal widths of N₂ at 10, 100 and 1000 K correspond to full widths at half maximum of 0.1, 0.3 and 1.0 km s⁻¹.

As can be seen in Fig. 6, the photodissociation of N₂ is free of self-shielding up to a column density of around 10¹² cm⁻², but is fully shielded by 10¹⁸ cm⁻². For intermediate N₂ column densities the self-shielding function increases with increasing

excitation temperature. There are two reasons for this (Visser et al. 2009). First, the optical depth of each line increases linearly with the thermal population of its corresponding lower-state rotational level, but the self-shielding increases nonlinearly according to Eq. (5). Then, because the ground state population at higher temperatures is distributed over more levels (see Fig. 2) there is an overall decrease in the effectiveness of self-shielding. The second effect arises from the individual line profiles, which are constructed to have thermal broadening. Those lines appearing in the 1000 K spectrum are then 10 times broader than those at 10 K, leading to decreased peak optical depth at the line center and less effective self-shielding over the whole line profile.

3.3. Shielding by H₂, H and CO

The wavelength range over which N₂ can be photodissociated is exactly the same range over which H₂, H and CO absorb strongly. The amount by which N₂ is shielded depends on the column densities of each of these species and is characterized by the shielding function of Eq. (6).

Figure 4 overlays absorption spectra for N₂ and H+H₂ combined. Two forms of the latter are included: a representative

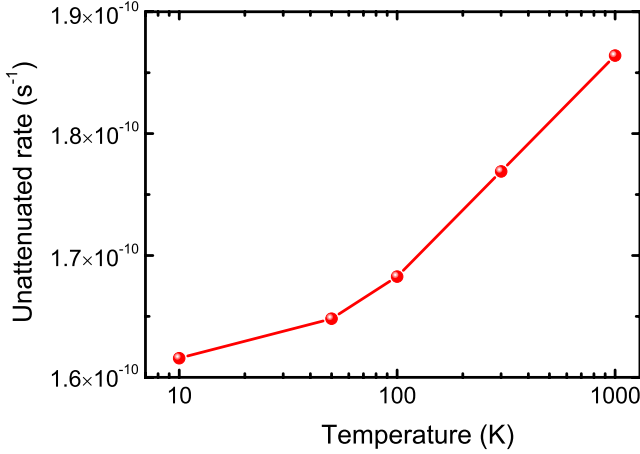


Fig. 5. Unattenuated photodissociation rates of N_2 immersed in a Draine (1978) field at various excitation temperatures.

Table 1. Unattenuated photodissociation rates of N_2 (excitation temperature 50 K) in a blackbody radiation field at various temperatures, T_{BB} .

T_{BB} (K)	k_{pd}^0 (s ⁻¹)	Previous
4000	2.18(-16)	3.0(-16) ^b
6000	9.96(-14)	–
8000	1.90(-12)	–
10 000	1.03(-11)	1.4(-11) ^b
20 000	1.97(-10)	–
Draine	1.65(-10)	2.3(-10) ^c

Notes. ^(a) All radiation fields have been normalized to a Draine (1978) field over the interval 912–2050 Å. ^(b) van Dishoeck et al. (2006) ^(c) van Dishoeck (1988).

example spectrum deduced from observations of $H_2(J)$ and H column densities of the well-studied and commonly-referenced diffuse cloud toward ζ Oph; and a simulated spectrum using column densities of 2.1×10^{20} and $2.6 \times 10^{20} \text{ cm}^{-2}$ for H_2 and H, respectively, and assuming purely thermal excitation of H_2 . The H_2 molecular data adopted for the synthetic spectra are those of Abgrall et al. (1993a,b) and were obtained from the Meudon PDR code website (Le Petit et al. 2006). The assumed column densities were taken to be half those of the observed ζ Oph cloud, as is appropriate for radiation penetrating to its center, and an excitation temperature of 50 K was used for the H_2+H and N_2 thermal models. The principal difference between observed and thermal H_2+H spectra is the appearance of additional lines in the observed spectrum from non-thermally populated higher- J levels. Thermal excitation H_2 spectra are used throughout the following mutual-shielding calculations, and do not include extrathermal excitations such as UV pumping. This negligence leads to a slight (approximately 3%) underestimate of shielding for the case of the ζ Oph cloud. A magnified version of the spectra in Fig. 4 is included in the appendix, and it is apparent that the ranges containing significant N_2 absorption and minimal shielding by H and H_2 are 919.8–920.2, 921.2–921.6, 922.6–923.1, 925.8–926.1, 935.1–935.4, 939.9–940.3, 942.3–942.8, 958.1–958.9, 959.0–959.1, 960.1–960.8 and 978.8–979.5 Å.

The calculated N_2 photodissociation rate at the center of the ζ Oph cloud is $6.96 \times 10^{-11} \text{ s}^{-1}$, corresponding to 58% shielding by H_2+H . Table 2 summarizes the contributions to the photodissociation rate of individual bands at the edge of the cloud

Table 2. Contributions of different bands to N_2 photodissociation at 50 K at the edge (unattenuated photodissociation) and in the center of the ζ Oph diffuse cloud.

Band	Excited state ^a	λ (Å)	Edge (%)	Center (%)	Shielding
1	$o(2)$	911.7–915.1	2.41	0.07	0.01
2	$b(11)$	915.1–917.4	0.58	0.47	0.34
3	$b'(7)$	917.4–919.0	0.09	0.06	0.30
4	$c(2)$	919.0–920.8	1.63	3.65	0.95
5	$c'(2)$	920.8–922.4	0.22	0.51	0.97
6	$b(10)$	922.4–924.7	1.74	3.73	0.91
7	$b'(6)$	924.7–927.7	0.33	0.66	0.84
8	$o(1) + b(9)$	927.7–930.8	3.65	2.31	0.27
9	$b'(5)$	930.8–933.9	0.23	0.10	0.18
10	$b(8)$	933.9–936.6	0.10	0.22	0.96
11	$b'(4)$	936.6–938.5	0.47	0.52	0.47
12	$c(1)$	938.5–939.6	9.31	5.28	0.24
13	$c'(1)$	939.6–941.5	1.67	3.47	0.88
14	$b(7)$	941.5–944.1	5.01	11.04	0.93
15	$b'(3)$	944.1–945.7	0.03	0.05	0.77
16	$o(0)$	945.7–947.9	0.05	0.00	0.00
17	$b(6)$	947.9–950.6	1.23	2.00	0.69
18	$b'(2)$	950.6–953.1	0.01	0.02	0.93
19	$b(5)$	953.1–956.8	1.04	0.41	0.17
20	$c'(0) + b'(1)$	956.8–959.5	2.79	6.21	0.94
21	$c(0)$	959.5–962.9	17.00	36.80	0.92
22	$b(4) + b'(0)$	962.9–969.4	22.51	2.02	0.04
23	$b(3)$	969.4–976.2	16.05	0.51	0.01
24	$b(2)$	976.2–983.1	9.28	19.83	0.91
25	$b(1)$	983.1–988.9	1.34	0.00	0.00
26	$b(0)$	988.9–1000.0	1.24	0.05	0.02

Notes. ^(a) The b , c , and o levels have $^1\Pi_u$ symmetry; b' and c'_4 have $^1\Sigma_u^+$ symmetry.

(unshielded) and at its center. The pattern of increasing and decreasing significance of individual N_2 bands under the influence of shielding is easily matched to the occurrence of overlapping features in Fig. 4a. The heavy shielding of bands 22 and 23 has a particularly large effect on the total photodissociation rate, the relative importance of the lightly shielded band 14 increases significantly in the center, and the 957–980 Å wavelength range remains particularly important for photodissociation throughout the cloud.

A similar investigation was performed considering the shielding of N_2 by CO. Simulated absorption spectra for both molecules are shown in Fig. 7, where the CO spectrum was generated by the photoabsorption model of Visser et al. (2009) assuming a column density of 10^{15} cm^{-2} , close to half of the observed ζ Oph value. Both spectra exhibit a complex pattern of bands so that overlaps are infrequent and do not occur at all in the most important photodissociation range, 957–980 Å. In this range CO hardly affects N_2 and, in general, shielding by H_2 and H is sufficiently dominant that the additional influence of CO can be neglected.

Two-dimensional shielding functions for a range of H_2 and H column densities have been calculated. These are tabulated in Table 3 and shown graphically in Fig. 8. For these calculations an excitation temperature of 50 K was assumed for both N_2 and H_2 , and $b(H_2)$ (the nonthermal broadening) was set to 3 km s^{-1} . Obviously, the shielding function decreases with increasing $N(H_2)$ and $N(H)$, but H_2 plays the more important role (as is the case for the shielding of CO; Visser et al. 2009). Specifically, N_2 is close to fully shielded ($\Theta < 2\%$)

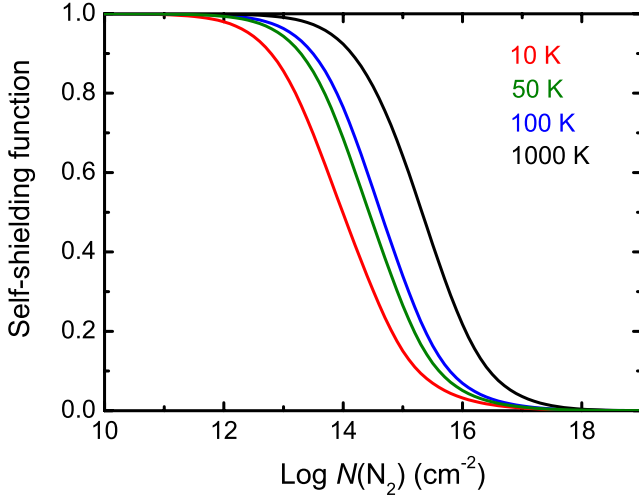


Fig. 6. N₂ self-shielding as a function of column density, $N(\text{N}_2)$, for excitation temperatures of 10, 50, 100 and 1000 K.

when $N(\text{H}_2) = 10^{22} \text{ cm}^{-2}$, and totally shielded by 10^{23} cm^{-2} . Electronic tables of the calculated shielding functions can be obtained from <http://www.strw.leidenuniv.nl/~ewine/photo>.

Since N₂ does not possess a permanent dipole moment, radiative decay from excited rotational levels of its electronic-vibrational ground state is slow. Then, the excitation temperature of N₂ is likely to be higher than that of CO and other molecules, and closer to the kinetic temperature. The effect of temperature on shielding by H₂ + H was investigated and is illustrated in Fig. 9. The same excitation temperatures are adopted for H₂ and N₂ because both are zero-dipole-moment molecules. The calculated shielding functions are somewhat erratic, and even show a peculiar non-monotonic temperature dependence at low H₂ column density. This arises from the small degree of overlap occurring between atomic H lines and N₂ bands. The distribution of N₂ lines over additional rotational transitions at higher temperatures leads to the variability of Fig. 9 and illustrates the need for high-resolution reference spectra in these kinds of applications. For significant H₂ column densities, and in contrast with N₂ self-shielding, the amount of shielding increases with increasing temperature. This results from a H₂ population that is spread over more rotational levels at higher temperatures, leading to an absorption spectrum featuring more lines available to shield N₂. This is clearly evident when comparing the various curves in Fig. 4.

Table 4 compares the H+H₂ shielding of N₂ with the CO shielding calculations of Visser et al. (2009). The two molecules follow a similar pattern, within 50%, up to $N(\text{H}_2) = 10^{22} \text{ cm}^{-2}$. This difference becomes more significant when $N(\text{H}_2) = 10^{23} \text{ cm}^{-2}$, but photodissociation has long ceased to be important as an N₂ destruction mechanism by then.

3.4. Shielding by dust

Dust grains compete with molecules in the cloud by also absorbing UV photons. For the 912–1000 Å wavelength range, the attenuation by dust is largely independent of wavelength and can be taken into account by an additional shielding term $\exp(-\gamma A_V)$ (van Dishoeck et al. 2006). For the wavelength range appropriate for N₂, a value of $\gamma = 3.53$ is found, using the method of Roberge et al. (1981) and standard diffuse-cloud grain properties

of Roberge et al. (1991). For larger dust grains of a few μm in size, such as is appropriate for protoplanetary disks, $\gamma \approx 0.6$ (van Dishoeck et al. 2006). The visual extinction A_V is computed from the total hydrogen column $N_{\text{H}} = N(\text{H}) + 2N(\text{H}_2)$ through the relation $A_V = N_{\text{H}}/1.6 \times 10^{21}$, based on Savage et al. (1977).

For diffuse clouds with total visual extinctions around 1 mag, radiation from the other side of the cloud may result in a shallower depth dependence than given by the above single exponential form. In such cases, a bi-exponential form $\exp(-\alpha A_V + \beta A_V^2)$ may be more appropriate (van Dishoeck & Dalgarno 1984; van Dishoeck 1988). For $A_V^{\text{tot}} = 1$ mag, $\alpha = 7.25$ and $\beta = 6.92$ are found.

The shielding of N₂ by dust under various conditions is listed in Table 4. This shows that shielding by normal interstellar dust is larger than that by H₂ and H for any A_V and implies that the “smoke screen” by dust also plays a significant role in diffuse and translucent clouds and photon-dominated regions. However, in protoplanetary disks where the larger dust particles absorb and scatter less efficiently, the effects of H₂ and H shielding become comparable, or even dominant, at large A_V .

4. Chemical models

As an example of how to apply the new photodissociation rates, we ran chemical models for a set of diffuse and translucent clouds, a photon-dominated region (PDR), and a vertical cut through a circumstellar disk. The models use the UMIST06 chemical network (Woodall et al. 2007), stripped down to species containing only H, He, C, N and O. Species containing more than two C, N or O atoms are also removed since they are not relevant for our purposes. Freeze-out and thermal evaporation are added for all neutral species, but no grain-surface reactions are included other than H₂ formation according to Black & van Dishoeck (1987). Self-shielding of CO is computed using the shielding functions of Visser et al. (2009); for N₂, we use the self-shielding functions calculated here at 50 K. The elemental abundances relative to H are 0.0975 for He, 7.86×10^{-5} for C, 2.47×10^{-5} for N and 1.80×10^{-4} for O (Aikawa et al. 2008). Enhanced formation of CH⁺ (and thus also CO) at low A_V is included following Visser et al. (2009) by supra-thermal chemistry, boosting the rate of ion-neutral reactions by setting the Alfvén speed to 3.3 km s^{-1} for column densities less than $4 \times 10^{20} \text{ cm}^{-2}$. Unless stated otherwise, the model of impinging UV flux is the Draine field of Eq. (3) modified by a scaling factor, χ . In all cases, the abundances of N, N₂ and CO reach steady state after ~ 1 Myr, regardless of whether the gas starts in atomic or molecular form.

4.1. Diffuse and translucent clouds

A set of diffuse and translucent cloud models was run for central densities $n_{\text{H}} = n(\text{H}) + 2n(\text{H}_2) = 100, 300$ and 10^3 cm^{-3} , at a temperature of 30 K, and assuming scaling factors of the UV flux of $\chi = 0.5, 1$ and 5. Figure 10 shows the abundances of N, N₂ and CO and the photodissociation rates of N₂ and CO as functions of depth into the cloud (measured in A_V) for the $n_{\text{H}} = 10^3$ and $\chi = 1$ model. Both CO and N₂ are rapidly photodissociated in the limit of low extinction and carbon and nitrogen are primarily in atomic form. Some CO is formed in a series of (supra-thermal) ion-molecule reactions starting with C⁺ at the edge (Visser et al. 2009). Since the ionization potential of atomic N lies just above that of H, preventing the formation of N⁺, N₂ can only form through slower neutral-neutral reactions. As a result,

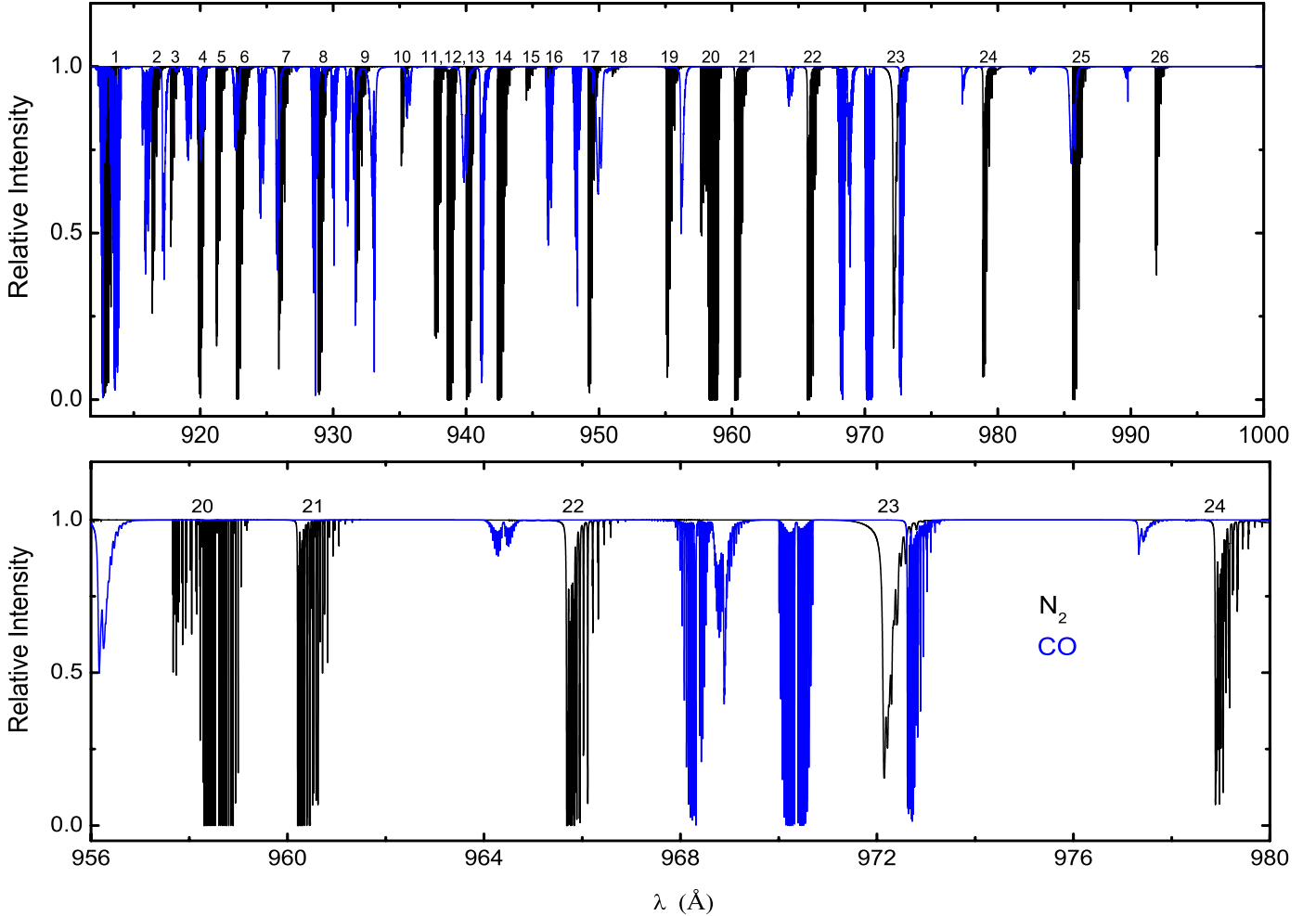


Fig. 7. *Top:* comparison of N₂ (black) and CO (blue) model absorption spectra between 912 and 1000 Å assuming an excitation temperature of 50 K for both molecules. The N₂ and CO column densities are both 10¹⁵ cm⁻². *Bottom:* blow-up of the above spectra for the wavelength region 956–980 Å.

the abundance of N₂ is three orders of magnitude lower than that of CO at the edge of the cloud. The conversion from N to N₂ occurs at an A_V of 1.5 mag, at which point CO has become the main form of carbon. The bottom panel of Fig. 10 illustrates that self-shielding and mutual shielding by H and H₂ significantly reduce the photodissociation rate relative to dust alone. The column densities of N₂ and CO at $A_V = 1.5$ mag are 1.5×10^{15} and 1.3×10^{16} cm⁻². At high A_V , atomic N is maintained at an abundance of 3×10^{-6} by the dissociative recombination of N₂H⁺, which in turn is formed from the reaction between N₂ and cosmic-ray-produced H₃⁺.

To investigate the role of turbulence or non-thermal motions on the results, a model has been run in which the Doppler width of the N₂ lines in the self-shielding calculation was increased to 3 km s⁻¹ rather than the thermal width at low temperatures. The resulting N₂ abundance as a function of depth is nearly identical to that presented in Fig. 10.

Absorption bands of N₂ have possibly been detected in observations of the diffuse cloud toward HD 124314 (Knauth et al. 2004). The two relevant bands, indicated in Fig. 4, are particularly strongly absorbing, and are relatively unshielded by hydrogen. The depth of the observed absorption indicates a total N₂ column density of $(4.6 \pm 0.8) \times 10^{13}$ cm⁻² and the stellar reddening of HD 124314 provides an estimate of the cloud's extinction, $A_V = 1.5$ mag. Figure 11 shows the cumulative N₂ column density calculated for a range of radiation field intensities

and n_H densities as a function of A_V . For comparison with the model, which considers only half of the cloud from edge to center and is irradiated from one side only, the observed A_V and N₂ column density must be halved. These models use the single exponential dust continuum shielding function; if the bi-exponential formulation were used, the model N₂ column densities would be even lower for small A_V . The maximum calculated column density occurs where the radiation field is weakest ($\chi = 0.5$) and for the highest density ($n_H = 10^3$ cm⁻³). These are extreme physical conditions for a cloud like HD 124314 and inconsistent with its relatively high H/H₂ column density ratio (André et al. 2003) and low CO column density (Sheffer et al. 2008). An independent confirmation of the N₂ detection is warranted. Observed upper limits toward other diffuse clouds with lower A_V are a few $\times 10^{12}$ cm⁻² (Lutz et al. 1979), which are consistent with the current models for typical densities of a few hundred cm⁻³ and $\chi \geq 1$.

4.2. Photon-dominated region

The PDR model is run assuming an n_H density of 10⁵ cm⁻³, a temperature of 100 K, and a UV flux of $\chi = 10^3$. Figure 12 shows the resulting abundances of N, N₂, C, C⁺ and CO and the relevant photodissociation rates as functions of A_V .

The calculated abundances of both N₂ and CO at low A_V are lower in the PDR model compared with the diffuse and

Table 3. Two-dimensional shielding functions $\Theta[N(\text{H}), N(\text{H}_2)]$ assuming an excitation temperature of 50 K.

log $N(\text{H}_2)$ (cm^{-2})	log $N(\text{H})$ (cm^{-2})		
	14	20	22
14.0	9.979(-1)	9.307(-1)	7.621(-1)
14.2	9.972(-1)	9.301(-1)	7.617(-1)
14.4	9.964(-1)	9.293(-1)	7.610(-1)
14.6	9.955(-1)	9.284(-1)	7.603(-1)
14.8	9.945(-1)	9.274(-1)	7.595(-1)
15.0	9.934(-1)	9.264(-1)	7.586(-1)
15.2	9.923(-1)	9.254(-1)	7.577(-1)
15.4	9.913(-1)	9.245(-1)	7.569(-1)
15.6	9.904(-1)	9.237(-1)	7.562(-1)
15.8	9.896(-1)	9.228(-1)	7.555(-1)
16.0	9.885(-1)	9.219(-1)	7.546(-1)
16.2	9.873(-1)	9.207(-1)	7.536(-1)
16.4	9.859(-1)	9.193(-1)	7.524(-1)
16.6	9.841(-1)	9.177(-1)	7.509(-1)
16.8	9.821(-1)	9.157(-1)	7.492(-1)
17.0	9.797(-1)	9.134(-1)	7.472(-1)
17.2	9.767(-1)	9.105(-1)	7.447(-1)
17.4	9.729(-1)	9.069(-1)	7.418(-1)
17.6	9.681(-1)	9.023(-1)	7.380(-1)
17.8	9.619(-1)	8.964(-1)	7.335(-1)
18.0	9.541(-1)	8.890(-1)	7.280(-1)
18.2	9.443(-1)	8.800(-1)	7.217(-1)
18.4	9.319(-1)	8.685(-1)	7.141(-1)
18.6	9.154(-1)	8.534(-1)	7.046(-1)
18.8	8.927(-1)	8.326(-1)	6.916(-1)
19.0	8.615(-1)	8.040(-1)	6.738(-1)
19.2	8.198(-1)	7.658(-1)	6.496(-1)
19.4	7.663(-1)	7.164(-1)	6.175(-1)
19.6	7.012(-1)	6.563(-1)	5.766(-1)
19.8	6.281(-1)	5.887(-1)	5.280(-1)
20.0	5.540(-1)	5.201(-1)	4.758(-1)
20.2	4.875(-1)	4.583(-1)	4.259(-1)
20.4	4.333(-1)	4.078(-1)	3.830(-1)
20.6	3.886(-1)	3.660(-1)	3.457(-1)
20.8	3.446(-1)	3.251(-1)	3.081(-1)
21.0	2.940(-1)	2.781(-1)	2.645(-1)
21.2	2.327(-1)	2.213(-1)	2.113(-1)
21.4	1.630(-1)	1.561(-1)	1.497(-1)
21.6	9.495(-2)	9.178(-2)	8.843(-2)
21.8	4.260(-2)	4.164(-2)	4.030(-2)
22.0	1.375(-2)	1.359(-2)	1.320(-2)
22.2	3.210(-3)	3.200(-3)	3.120(-3)
22.4	5.737(-4)	5.734(-4)	5.600(-4)
22.6	6.110(-5)	6.108(-5)	5.969(-5)
22.8	2.146(-6)	2.146(-6)	2.094(-6)
23.0	1.151(-8)	1.151(-8)	1.118(-8)

Notes. The notation $x(-y)$ indicates $x \times 10^{-y}$.

translucent clouds, because of the stronger UV field. For all models, the abundance of N₂ is several orders of magnitude lower than that of CO. Also, because of the increased radiation, the transition from N to N₂ occurs deeper into the PDR: at an A_V of about 3 mag. The column densities of N₂ and CO at this point are 6×10^{15} and $3 \times 10^{16} \text{ cm}^{-2}$. The minor wiggle in the atomic N abundance profiles at $A_V = 2-3$ mag is due to the abundance patterns of CH and OH. CH is the main destroyer of N at $A_V = 2$ mag but its abundance drops going into the cloud because its main precursor, C⁺, disappears. The simultaneously increasing extinction allows for an increase in the OH abundance, so that this becomes the main destroyer of N for $A_V > 3$ mag. Interestingly, the transition from N → N₂ occurs at nearly the same depth into the cloud than that of C⁺ → C → CO.

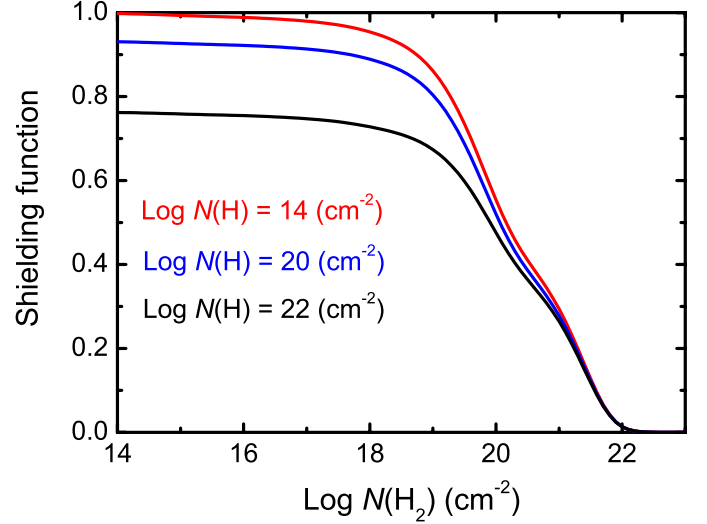


Fig. 8. Shielding of N₂ by H₂ + H as a function of H₂ column density, $N(\text{H}_2)$, for three different values of $N(\text{H})$. An excitation temperature of 50 K is adopted for both N₂ and H₂.

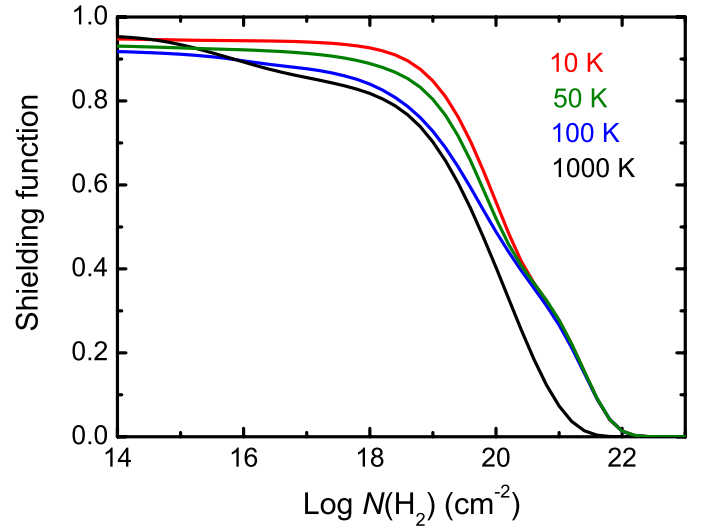


Fig. 9. Shielding of N₂ by H₂ + H as a function of H₂ column density $N(\text{H}_2)$, for N₂ and H₂ excitation temperatures of 10, 50, 100 and 1000 K. The column-density of H is set to 10^{20} cm^{-2} in all cases.

4.3. Circumstellar disk

The fourth model simulates a vertical slice through a circumstellar disk and its setup is identical to that of Visser et al. (2009). The slice is located at a radius of 105 AU in the standard model of D'Alessio et al. (1999), which supposes a disk of $0.07 M_{\odot}$ and 400 AU radius surrounding a T Tauri star of $0.5 M_{\odot}$ and $2 R_{\odot}$ radius. The surface of the slice, at a height of 120 AU, is illuminated by the Draine (1978) field with $\chi = 516$. We ran the model assuming dust grain sizes of 0.1 and $1 \mu\text{m}$. The results are plotted in Fig. 13.

Starting from the disk surface (high z) and moving inwards, the abundance profiles of N, N₂ and CO show the same qualitative trends as they do for the translucent cloud and PDR models. The main difference arises from freeze-out of N₂ and CO for z below 25 AU, at which point the dust temperature drops below ~ 20 K. The depletion of N₂ also drives down the abundance of N₂H⁺ which, in turn, restricts the abundance of

Table 4. Comparison of the shielding of $^{14}\text{N}_2$ and ^{12}CO by $\text{H}_2 + \text{H}$ and dust for a range of extinction, A_V , at 10 K and taking $N(\text{H}) = 5 \times 10^{20} \text{ cm}^{-2}$.

A_V	Shielding due to $\text{H}_2 + \text{H}$			Shielding due to dust	
	$\log N(\text{H}_2) (\text{cm}^{-2})$	$^{14}\text{N}_2$	$^{12}\text{CO}^a$	Interstellar	Protoplanetary
0.31	0	8.916(-1)	0.9	3.318(-1)	8.290(-1)
0.32	19	8.098(-1)	8.176(-1)	3.175(-1)	8.228(-1)
0.44	20	5.543(-1)	7.223(-1)	2.134(-1)	7.691(-1)
1.56	21	2.759(-1)	3.260(-1)	4.023(-3)	3.916(-1)
12.8	22	1.346(-2)	1.108(-2)	0	4.585(-4)
125.3	23	2.343(-8)	3.938(-7)	0	0

Notes. The unattenuated photodissociation rates of $^{14}\text{N}_2$ and ^{12}CO are 1.65×10^{-10} and $2.59 \times 10^{-10} \text{ s}^{-1}$, respectively. ^(a) The CO values are from Visser et al. (2009).

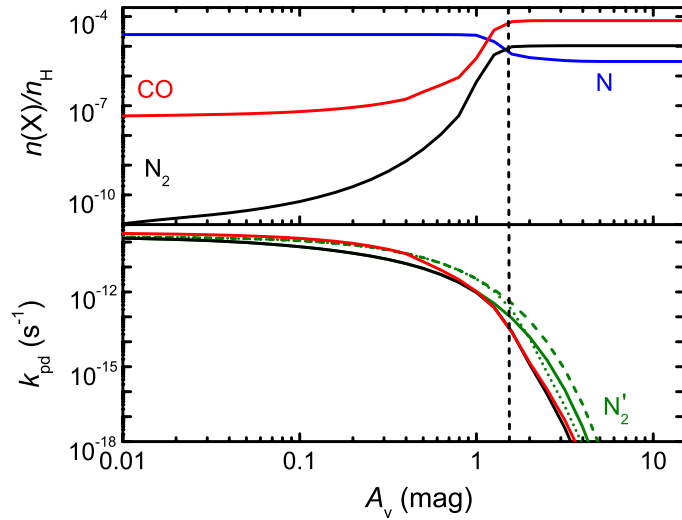


Fig. 10. *Top:* relative abundances of N, N_2 and CO as a function of depth into a translucent cloud with $\chi = 1$, $T = 30 \text{ K}$ and $n_{\text{H}} = 10^3 \text{ cm}^{-3}$. *Bottom:* photodissociation rates of N_2 (black) and CO (red) as functions of depth. Alternative photodissociation rate curves (N_2 , green) consider shielding by dust alone (dashed), dust + self-shielding (dotted), and dust + $\text{H} + \text{H}_2$ (solid). The conversion from N to N_2 occurs at $A_V \approx 1.5 \text{ mag}$, as shown by the vertical dashed line.

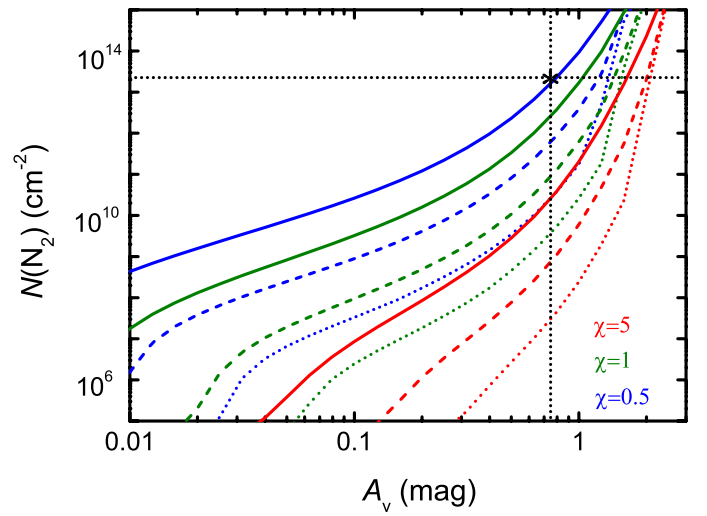


Fig. 11. Cumulative column density of N_2 as a function of extinction, A_V , at an excitation temperature of 30 K. Curves are shown for several different models: dotted, dashed, and solid lines indicate $n_{\text{H}} = 100$, 300 and 1000 cm^{-3} , respectively; blue, green and red lines scale the radiation field by $\chi = 0.5$, 1 and 5, respectively. The asterisk indicates half the N_2 column density and half the extinction observed by Knauth et al. (2004) in the diffuse cloud toward HD 124314.

atomic N. In the models presented here, nitrogen is fully converted into N_2 at heights where no freeze-out occurs.

Increasing the grain size from 0.1 to 1 μm allows the UV field to penetrate to heights of about 35 instead of 45 AU. The abundances of N_2 and CO are then a factor of 10–100 lower in the intervening zone, while that of atomic N is a factor of a few higher. Since atomic N is a prerequisite for an active nitrogen chemistry leading to species like HCN, this result illustrates that larger column densities of nitrogen-containing molecules can be expected in disks with grain growth.

The column densities integrated from surface to midplane are $2 \times 10^{17} \text{ cm}^{-2}$ for N_2 and $1.2 \times 10^{18} \text{ cm}^{-2}$ for CO, regardless of the grain size, since the bulk of the N_2 and CO are at high densities where UV photodissociation is negligible. The total column of N is $5 \times 10^{16} \text{ cm}^{-2}$ for 0.1 μm grains and $8 \times 10^{16} \text{ cm}^{-2}$ for 1 μm grains.

The models discussed above use a scaled Draine (1978) radiation field. A much cooler radiation field was also considered by assuming a 4000 K blackbody source scaled to the same flux between 912 and 2050 \AA . Then, both carbon and nitrogen are fully molecular at the disk surface since photodissociation of both CO and N_2 is negligible. A smaller amount of atomic N is maintained by chemical reactions for active nitrogen chemistry. This

result is qualitatively consistent with the observation of Pascucci et al. (2009) that the color of the radiation field affects the nitrogen chemistry, although that study applied to the inner rather than the outer disk.

5. Conclusions

In this work, we compute accurate N_2 photodissociation rates in the interstellar medium for the first time by employing new molecular data on its electronic transitions. The calculated N_2 photodissociation rate in an unattenuated interstellar radiation field is $1.65 \times 10^{-10} \text{ s}^{-1}$ (50 K), $\sim 28\%$ lower than the previously recommended value. This rate increases somewhat with temperature due to J -dependent predissociation rates.

The simulated spectra reveal that the most important range for photodissociation is 957–980 \AA , where H_2 and H absorption significantly overlap with N_2 absorption. In contrast, CO only weakly shields N_2 . Self-shielding and mutual shielding functions have been computed for a range of N_2 , H_2 and H column densities. For interstellar grains, shielding by dust is also effective. In protoplanetary disks, where dust particles have grown to μm size, the dust shielding becomes less than that of H_2 .

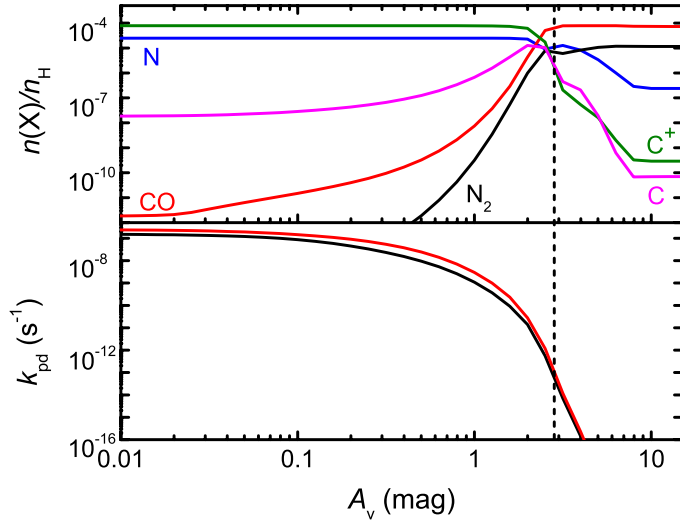


Fig. 12. *Top:* relative abundances of N (blue), N₂ (black), C (magenta), C⁺ (green) and CO (red) as functions of depth into a dense PDR. *Bottom:* photodissociation rates of N₂ and CO as functions of depth into a dense PDR. The vertical dashed line indicates where He⁺ and H₃⁺ take over from photodissociation as the main destruction mechanism for N₂.

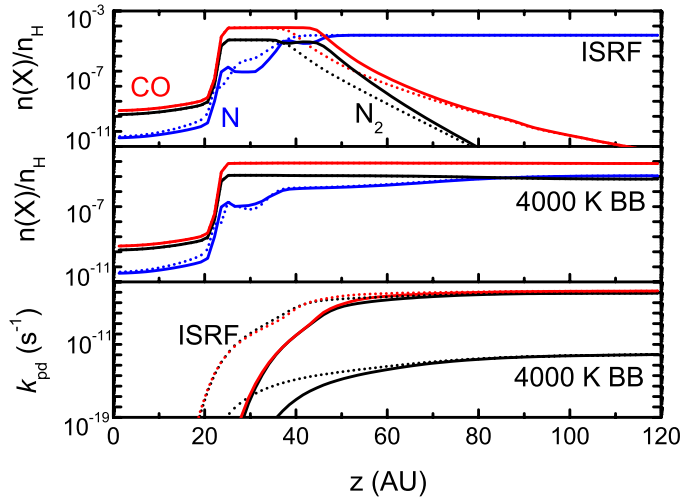


Fig. 13. *Top and middle:* relative abundances of N (blue), N₂ (black) and CO (red) as a function of height in a slice through a circumstellar disk exposed to the interstellar radiation field (ISRF) scaled by a factor $\chi = 516$ and a 4000 K black body field. *Bottom:* the relevant photodissociation rates of CO and N₂. Solid lines are for a grain size of 0.1 μm , dotted lines for 1 μm .

The new rates have been incorporated into models of diffuse and translucent clouds, of a dense PDR, and a protoplanetary disk. The translucent cloud models show that the observed column of interstellar N₂ in a translucent cloud with $A_V = 1.5$ mag can only be reproduced if the density n_H is higher than 1000 cm^{-3} and the radiation field has an intensity of less than half of the [Draine \(1978\)](#) field. For dense PDRs, the N₂ abundance only becomes significant at extinctions of more than 3 mag into the cloud but the transition of N \rightarrow N₂ occurs at nearly the same depth as that of C⁺ \rightarrow C \rightarrow CO. Disk models show that nitrogen is fully converted into N₂ at heights before freeze-out occurs, irrespective of grain size. However, an active nitrogen chemistry can take place in the upper layers of a disk where not all nitrogen is locked up in N₂, except for very cool radiation fields. Altogether, data are now available to accurately

model N₂ photodissociation in a wide variety of interstellar and circumstellar media.

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References

- Abgrall, H., Roueff, E., Launay, F., Roncin, J. Y., & Subtil, J. L. 1993a, *A&AS*, 101, 273
 Abgrall, H., Roueff, E., Launay, F., Roncin, J. Y., & Subtil, J. L. 1993b, *A&AS*, 101, 323
 Aikawa, Y., Wakelam, V., Garrod, R. T., & Herbst, E. 2008, *ApJ*, 674, 984
 Ajello, J. M., James, G. K., Franklin, B. O., & Shemansky, D. E. 1989, *Phys. Rev. A*, 40, 3524
 André, M. K., Oliveira, C. M., Howk, J. C., et al. 2003, *ApJ*, 591, 1000
 Bergin, E. A., Alves, J., Huard, T., & Lada, C. J. 2002, *ApJ*, 570, L101
 Bettens, R. P. A., Lee, H.-H., & Herbst, E. 1995, *ApJ*, 443, 664
 Black, J. H., & van Dishoeck, E. F. 1987, *ApJ*, 322, 412
 Crapsi, A., Caselli, P., Walmsley, C. M., et al. 2005, *ApJ*, 619, 379
 D'Alessio, P., Calvet, N., Hartmann, L., Lizano, S., & Cantó, J. 1999, *ApJ*, 527, 893
 Daranlot, J., Hincelin, U., Bergeat, A., et al. 2012, *Proc. Nat. Acad. Sci.*, 109, 10233
 Draine, B. T. 1978, *ApJS*, 36, 595
 Dutrey, A., Henning, T., Guilloteau, S., et al. 2007, *A&A*, 464, 615
 Fontani, F., Palau, A., Caselli, P., et al. 2011, *A&A*, 529, L7
 Gondhalekar, P. M., Phillips, A. P., & Wilson, R. 1980, *A&A*, 85, 272
 Habing, H. J. 1968, *Bull. Astron. Inst. Netherlands*, 19, 421
 Haverd, V. E., Lewis, B. R., Gibson, S. T., & Stark, G. 2005, *J. Chem. Phys.*, 123, 4304
 Heays, A. N. 2011, Ph.D. Thesis, The Australian National University
 Heays, A. N., Lewis, B. R., Stark, G., et al. 2009, *J. Chem. Phys.*, 131, 194308
 Heays, A. N., Dickenson, G. D., Salumbides, E. J., et al. 2011, *J. Chem. Phys.*, 135, 244301
 Helm, H., Hazell, I., & Bjerre, N. 1993, *Phys. Rev. A*, 48, 2762
 Herbst, E., & Klemperer, W. 1973, *ApJ*, 185, 505
 Herbst, E., Green, S., Thaddeus, P., & Klemperer, W. 1977, *ApJ*, 215, 503
 Huber, K. P., & Herzberg, G. 1979, *Molecular spectra and molecular structure IV: Constants of diatomic molecules* (New York: Van Nostrand)
 Knauth, D. C., Andersson, B.-G., McCandliss, S. R., & Warren Moos, H. 2004, *Nature*, 429, 636
 Langer, W. D., & Graedel, T. E. 1989, *ApJS*, 69, 241
 Le Petit, F., Nehmé, C., Le Bourlot, J., & Roueff, E. 2006, *ApJS*, 164, 506
 Lee, H.-H., Herbst, E., Pineau des Forets, G., Roueff, E., & Le Bourlot, J. 1996, *A&A*, 311, 690
 Lefebvre-Brion, H., & Field, R. W. 2004, *The spectra and dynamics of diatomic molecules* (Elsevier)
 Lewis, B. R., Gibson, S. T., Sprengers, J. P., et al. 2005a, *J. Chem. Phys.*, 123, 236101
 Lewis, B. R., Gibson, S. T., Zhang, W., Lefebvre-Brion, H., & Robbe, J.-M. 2005b, *J. Chem. Phys.*, 122, 144302
 Lewis, B. R., Baldwin, K. G. H., Heays, A. N., et al. 2008a, *J. Chem. Phys.*, 129, 204303
 Lewis, B. R., Baldwin, K. G. H., Sprengers, J. P., et al. 2008b, *J. Chem. Phys.*, 129, 164305
 Lewis, B. R., Heays, A. N., Gibson, S. T., Lefebvre-Brion, H., & Lefebvre, R. 2008c, *J. Chem. Phys.*, 129, 164306
 Liang, M.-C., Heays, A. N., Lewis, B. R., Gibson, S. T., & Yung, Y. L. 2007, *ApJ*, 664, L115
 Liu, X., Shemansky, D. E., Malone, C. P., et al. 2008, *J. Geophys. Res.*, 113, A02304
 Lutz, B. L., Snow, Jr., T. P., & Owen, T. 1979, *ApJ*, 227, 159
 Lyons, J. R. 2009, *Meteor. Planet. Sci. Suppl.*, 72, 5437
 Lyons, J. R. 2010, *Meteor. Planet. Sci. Suppl.*, 73, 5424
 Maret, S., Bergin, E. A., & Lada, C. J. 2006, *Nature*, 442, 425
 Mathis, J. S., Mezger, P. G., & Panagia, N. 1983, *A&A*, 128, 212
 Mauersberger, R., & Henkel, C. 1991, *A&A*, 245, 457

- Meier, D. S., & Turner, J. L. 2005, *ApJ*, 618, 259
- Meier, R. R., Samson, J. A. R., Chung, Y., Lee, E.-M., & He, Z.-X. 1991, *Planet. Space Sci.*, 39, 1197
- Muller, S., Beelen, A., Guélin, M., et al. 2011, *A&A*, 535, A103
- Ndome, H., Hochlaf, M., Lewis, B. R., et al. 2008, *J. Chem. Phys.*, 129, 164307
- Öberg, K. I., Qi, C., Fogel, J. K. J., et al. 2010, *ApJ*, 720, 480
- Pascucci, I., Apai, D., Luhman, K., et al. 2009, *ApJ*, 696, 143
- Roberge, W. G., Dalgarno, A., & Flannery, B. P. 1981, *ApJ*, 243, 817
- Roberge, W. G., Jones, D., Lepp, S., & Dalgarno, A. 1991, *ApJS*, 77, 287
- Savage, B. D., Bohlin, R. C., Drake, J. F., & Budich, W. 1977, *ApJ*, 216, 291
- Sheffer, Y., Rogers, M., Federman, S. R., et al. 2008, *ApJ*, 687, 1075
- Spelsberg, D., & Meyer, W. 2001, *J. Chem. Phys.*, 115, 6438
- Sprengers, J. P., Ubachs, W., Baldwin, K. G. H., Lewis, B. R., & Tchang-Brillet, W.-Ü. L. 2003, *J. Chem. Phys.*, 119, 3160
- Sprengers, J. P., Ubachs, W., Johansson, A., et al. 2004, *J. Chem. Phys.*, 120, 8973
- Sprengers, J. P., Reinhold, E., Ubachs, W., Baldwin, K. G. H., & Lewis, B. R. 2005, *J. Chem. Phys.*, 123, 144315
- Stark, G., Lewis, B. R., Heays, A. N., et al. 2008, *J. Chem. Phys.*, 128, 114302
- Strobel, D. F. 1982, *Origins of Life*, 12, 244
- Tobin, J. J., Hartmann, L., Bergin, E., et al. 2012, *ApJ*, 748, 16
- Turner, B. E. 1974, *Ap&SS*, 29, 247
- van Dishoeck, E. F. 1988, in *Rate Coefficients in Astrochemistry*, eds. T. J. Millar, & D. A. Williams (Dordrecht: Kluwer Academic Publishers), 49
- van Dishoeck, E. F., & Black, J. H. 1988, *ApJ*, 334, 771
- van Dishoeck, E. F., & Dalgarno, A. 1984, *ApJ*, 277, 576
- van Dishoeck, E. F., van Hemert, M. C., Allison, A. C., & Dalgarno, A. 1984, *ApJ*, 81, 5709
- van Dishoeck, E. F., Jonkheid, B., & van Hemert, M. C. 2006, *Faraday Discussions*, 303, 231
- Viala, Y. P., Letzelter, C., Eidelsberg, M., & Rostas, F. 1988, *A&A*, 193, 265
- Visser, R., van Dishoeck, E. F., & Black, J. H. 2009, *A&A*, 503, 323
- Walter, C. W., Cosby, P., & Helm, H. 1993, *J. Chem. Phys.*, 99, 3553
- Wayne, R. P. 2000, *Chemistry of Atmospheres* (Oxford University Press)
- Woodall, J., Agúndez, M., Markwick-Kemper, A. J., & Millar, T. J. 2007, *A&A*, 466, 1197
- Wu, C. Y. R., Judge, D. L., Tsai, M.-H., et al. 2012, *J. Chem. Phys.*, 136, 044301

Appendix A: self-shielding functions

In this section, we provide a table containing the self-shielding functions of N₂ at 10, 100 and 1000 K (data for Fig. 6).

Table A.1. N₂ self-shielding as a function of column density, $N(N_2)$, for excitation temperatures of 10, 100 and 1000 K.

log $N(N_2)$ (cm ⁻²)	Temperature (K)		
	10	100	1000
10.0	9.998(-1)	1.000(0)	1.000(0)
10.2	9.997(-1)	9.999(-1)	1.000(0)
10.4	9.995(-1)	9.999(-1)	1.000(0)
10.6	9.991(-1)	9.998(-1)	1.000(0)
10.8	9.986(-1)	9.997(-1)	9.999(-1)
11.0	9.979(-1)	9.996(-1)	9.999(-1)
11.2	9.966(-1)	9.994(-1)	9.999(-1)
11.4	9.947(-1)	9.990(-1)	9.998(-1)
11.6	9.917(-1)	9.984(-1)	9.996(-1)
11.8	9.870(-1)	9.974(-1)	9.994(-1)
12.0	9.798(-1)	9.960(-1)	9.991(-1)
12.2	9.691(-1)	9.936(-1)	9.986(-1)
12.4	9.533(-1)	9.900(-1)	9.977(-1)
12.6	9.305(-1)	9.844(-1)	9.964(-1)
12.8	8.985(-1)	9.757(-1)	9.943(-1)
13.0	8.554(-1)	9.627(-1)	9.910(-1)
13.2	8.001(-1)	9.436(-1)	9.859(-1)
13.4	7.333(-1)	9.163(-1)	9.781(-1)
13.6	6.579(-1)	8.785(-1)	9.662(-1)
13.8	5.778(-1)	8.282(-1)	9.486(-1)
14.0	4.967(-1)	7.643(-1)	9.235(-1)
14.2	4.169(-1)	6.877(-1)	8.891(-1)
14.4	3.400(-1)	6.022(-1)	8.440(-1)
14.6	2.680(-1)	5.130(-1)	7.875(-1)
14.8	2.037(-1)	4.244(-1)	7.197(-1)
15.0	1.505(-1)	3.394(-1)	6.412(-1)
15.2	1.100(-1)	2.609(-1)	5.546(-1)
15.4	8.073(-2)	1.928(-1)	4.641(-1)
15.6	5.945(-2)	1.384(-1)	3.747(-1)
15.8	4.374(-2)	9.802(-2)	2.911(-1)
16.0	3.214(-2)	6.920(-2)	2.175(-1)
16.2	2.361(-2)	4.886(-2)	1.567(-1)
16.4	1.731(-2)	3.454(-2)	1.100(-1)
16.6	1.267(-2)	2.441(-2)	7.613(-2)
16.8	9.250(-3)	1.717(-2)	5.229(-2)
17.0	6.720(-3)	1.199(-2)	3.557(-2)
17.2	4.870(-3)	8.280(-3)	2.387(-2)
17.4	3.510(-3)	5.640(-3)	1.574(-2)
17.6	2.520(-3)	3.810(-3)	1.013(-2)
17.8	1.790(-3)	2.570(-3)	6.340(-3)
18.0	1.270(-3)	1.740(-3)	3.850(-3)
18.2	8.906(-4)	1.190(-3)	2.280(-3)
18.4	6.247(-4)	8.066(-4)	1.310(-3)
18.6	4.359(-4)	5.425(-4)	7.183(-4)
18.8	2.998(-4)	3.613(-4)	3.737(-4)
19.0	2.031(-4)	2.386(-4)	1.820(-4)

Notes. The notation $x(-y)$ indicates $x \times 10^{-y}$.

Appendix B: High resolution spectra

In this section, we provide the high resolution spectra of N_2 at 50 K (zoom in for Fig. 4). The H_2 and H column densities are the same as those used in Fig. 4.

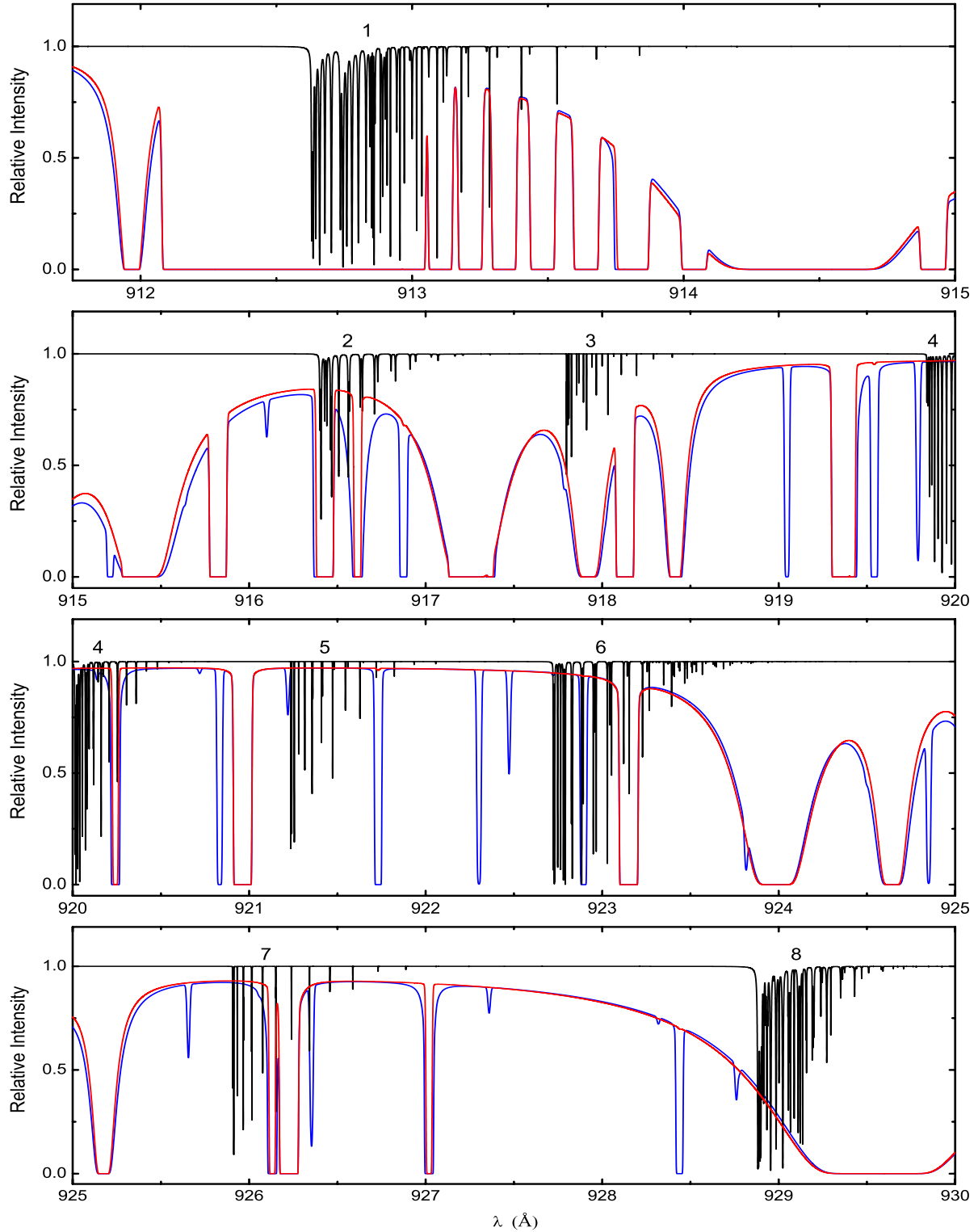


Fig. B.1. Zoom-in of the high-resolution spectra of N_2 (black line) and H plus H_2 (red line) in the wavelength range of 911.75–930 Å for a thermal excitation temperature of 50 K. The column density of N_2 is 10^{15} cm^{-2} and values for H_2 and H are taken to be half of the observed column densities in the well-studied diffuse cloud toward ζ Oph, as is appropriate for the center of the cloud: $N(H_2) = 2.1 \times 10^{20}$ and $N(H) = 2.6 \times 10^{20} \text{ cm}^{-2}$. The model H_2 Doppler width is 3 km s^{-1} . Also shown is the $H_2 + H$ absorption spectrum (blue) toward ζ Oph using the observed column densities for individual J levels, showing non-thermal excitation of H_2 .

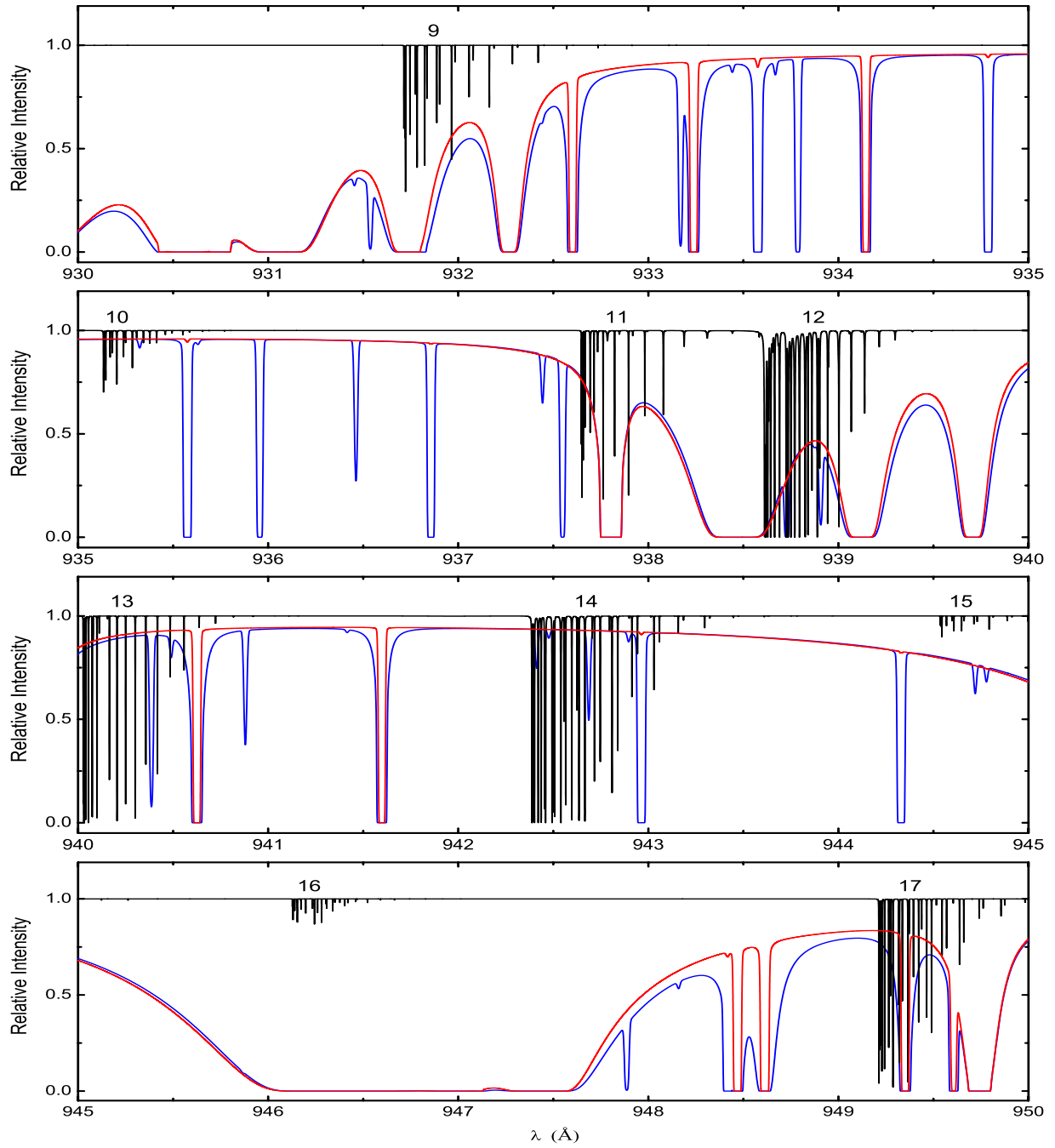


Fig. B.2. As Fig. B.1, but for 930–950 Å.

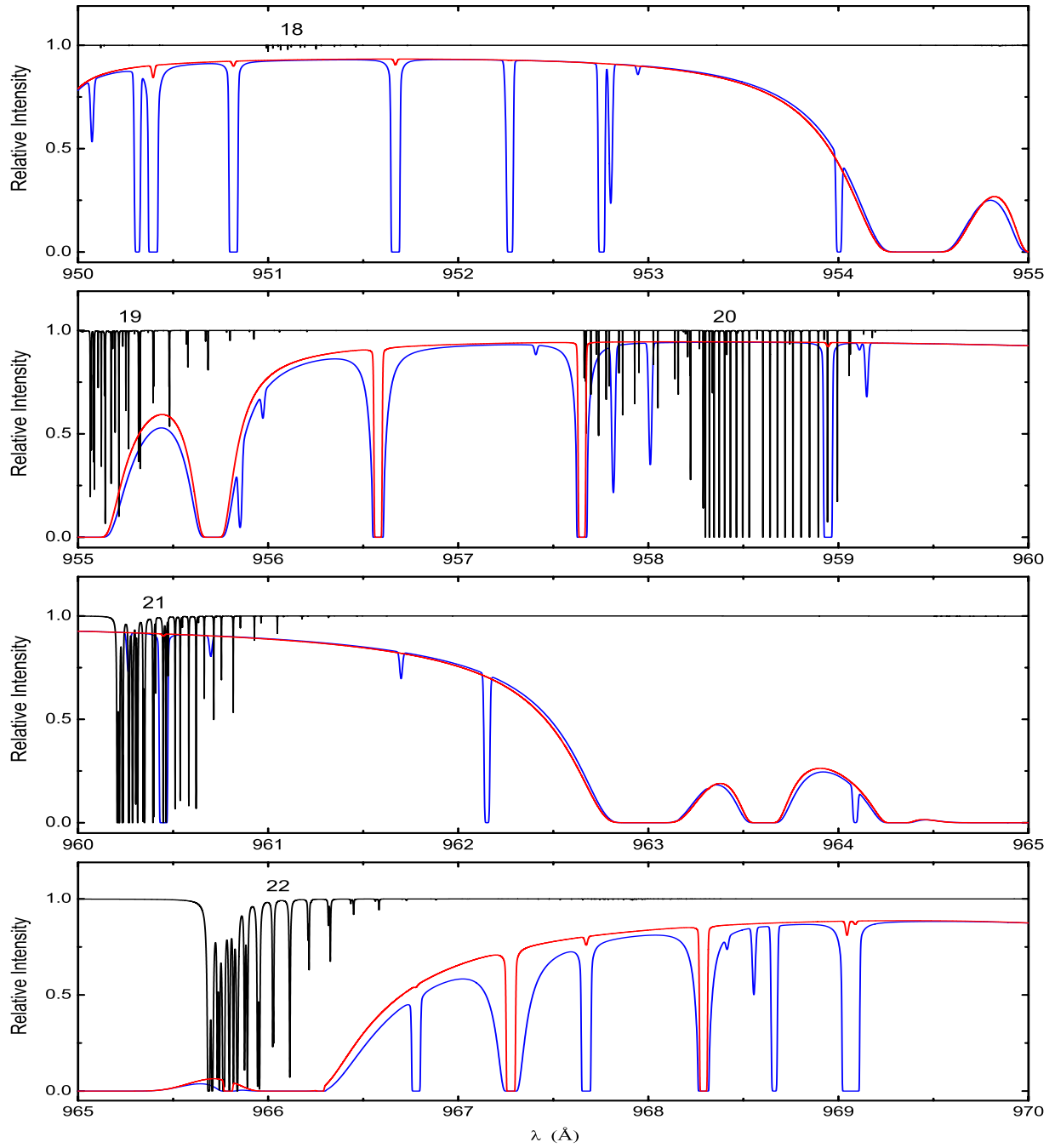


Fig. B.3. As Fig. B.1, but for 950–970 \AA .

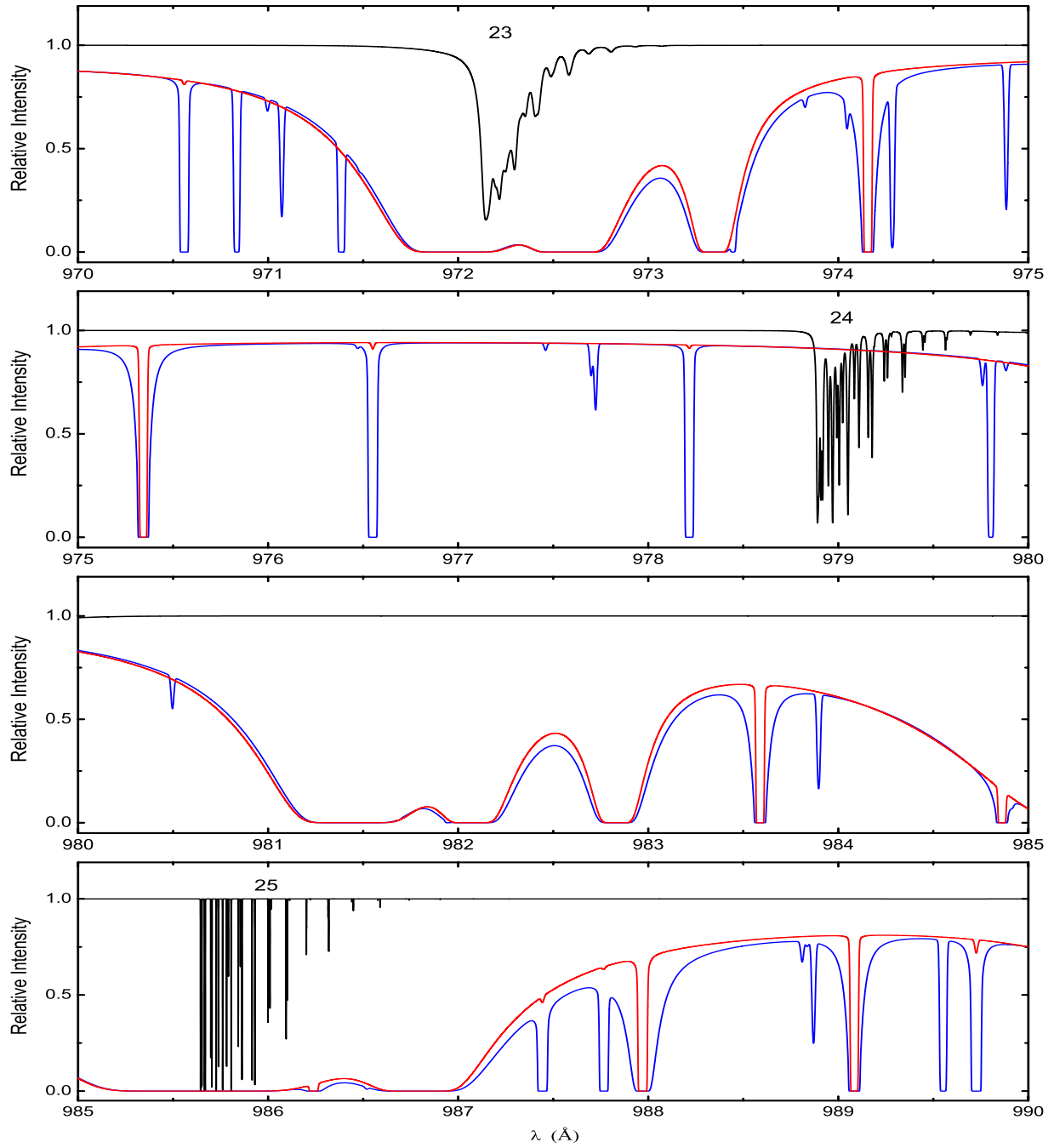


Fig. B.4. As Fig. B.1, but for 970–990 Å.

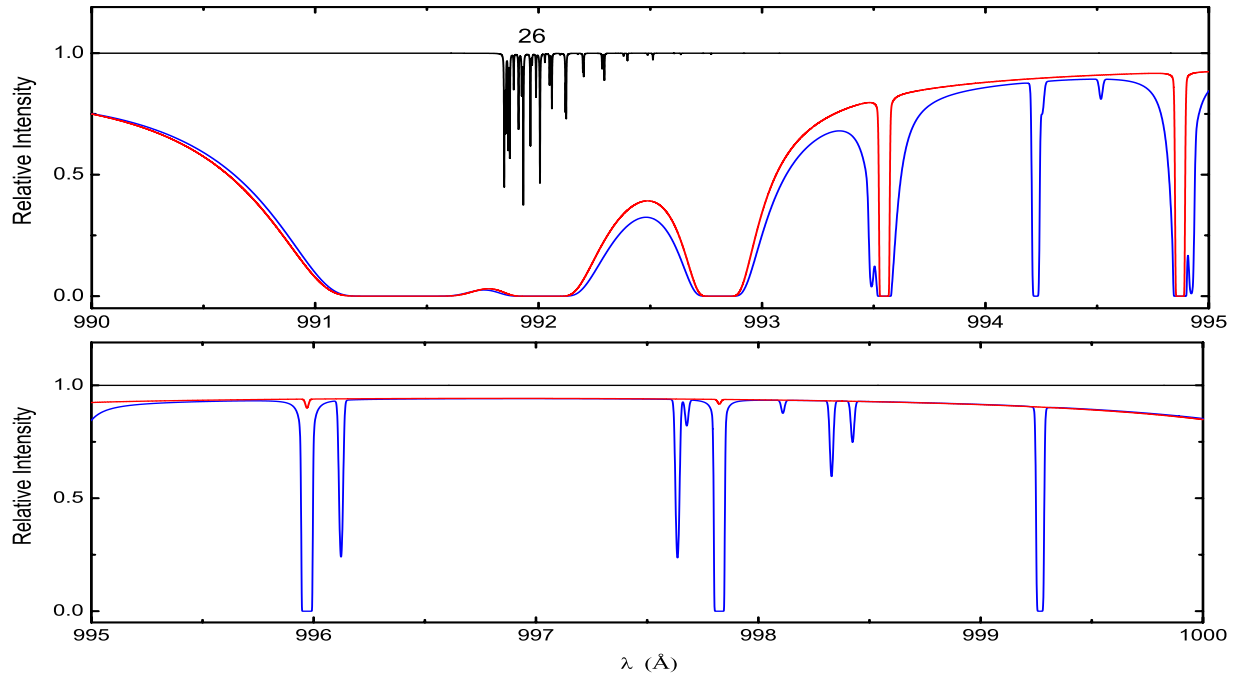


Fig. B.5. As Fig. B.1, but for 990–1000 Å.