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# *Clostridium carboxidivorans* Strain P7<sup>T</sup> Recombinant Formate Dehydrogenase Catalyzes Reduction of CO<sub>2</sub> to Formate

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**Recombinant formate dehydrogenase from the acetogen *Clostridium carboxidivorans* strain P7<sup>T</sup>, expressed in *Escherichia coli*, shows particular activity towards NADH-dependent carbon dioxide reduction to formate due to the relative binding affinities of the substrates and products. The enzyme retains activity over 2 days at 4°C under oxic conditions.**

Formate dehydrogenases (FDHs) catalyze the interconversion of CO<sub>2</sub> and formic acid through an oxidoreductive process (Fig. 1) (1). Consequently, when catalyzing CO<sub>2</sub> reduction, they are of interest for the sequestration of CO<sub>2</sub> and for the production of formic acid as a stabilized form of hydrogen fuel and as a source of commodity chemicals. In many bacteria and eukaryotes, FDHs catalyze the final step of catabolic processes in which formate is oxidized to CO<sub>2</sub> (2). The ability of certain members of this class, such as FDH from *Candida boidinii* in particular, to efficiently regenerate NADH in conjunction with formate oxidation has been a research focus (3).

Acetogens are known to possess a number of pathways distinct from those found in the other species. FDHs present in acetogens are known to take part in a carbon fixation metabolic pathway producing acetate (the Eastern branch of the Wood-Ljungdahl pathway), in which the first step involves reduction of CO<sub>2</sub> to formate (4). Several FDHs are known to catalyze CO<sub>2</sub> reduction under appropriate conditions (5–9). Those enzymes from acetogenic and related anaerobes, such as *Moorella thermoacetica* and *Clostridium pasteurianum*, are better than other FDHs as reduction catalysts but also show similar catalytic efficiency toward formate oxidation and are considered highly oxygen labile, requiring anaerobic expression and purification as well as anoxic assay conditions (10, 11). *Clostridium carboxidivorans* strain P7<sup>T</sup> (equivalent to ATCC BAA-624<sup>T</sup> and DSM 15243<sup>T</sup>) was isolated from the sediment of an agricultural settling lagoon after enrichment with CO as the substrate and is an obligate anaerobe that can grow autotrophically with H<sub>2</sub> and CO<sub>2</sub> or CO (fixing carbon via the Wood-Ljungdahl pathway) (12, 13). Therefore, when the gene of a selenocysteine-containing formate dehydrogenase H (FDH<sub>H</sub>) from the acetogen *Clostridium carboxidivorans* strain P7<sup>T</sup> was first identified, it was suggested that FDH<sub>H</sub> would catalyze the conversion of CO<sub>2</sub> to formate (14, 15). Here we report the first production of FDH<sub>H</sub> and its catalytic preference for CO<sub>2</sub> reduction, as well as its tolerance for oxic conditions.

**Cloning, expression, and purification of FDHs.** The overexpression and purification of recombinant FDH<sub>H</sub> from the *Clostridium carboxidivorans* strain P7<sup>T</sup> (FDH<sub>H</sub>-*CloCa*) was carried out, along with that of NAD<sup>+</sup>-dependent recombinant FDH from *Candida boidinii* (FDH-*CanBo*), in order to compare expression and activity of formate dehydrogenases that take part in distinct metabolic pathways. The DNA sequences for the FDH<sub>H</sub>-*CloCa* (UniProt E2IQB0) and FDH-*CanBo* (UniProt O13437) genes were codon optimized for expression in *Escherichia coli* (commercially synthesized by GeneArt, Germany), with the exception of

residue 139Sec, which was modified to 139Cys for FDH<sub>H</sub>-*CloCa* to avoid the need for use of a different expression system with selenoprotein-expressing elements. Plasmid DNAs were then reconstructed with the respective FDH genes inserted into the T7 promoter vector pETMCSIII for subsequent expression with an N-terminal His<sub>6</sub> tag (16). The successful reconstruction of the plasmid DNAs was confirmed by PCR coding region amplification and DNA sequencing. The recombinant plasmid was transformed into the *E. coli* BL21(DE3) strain for expression. The transformed cells for both FDH<sub>H</sub>-*CloCa* and FDH-*CanBo* were grown overnight in 50 ml of LBA medium (Luria-Bertani medium supplemented with 0.1 mg/ml ampicillin) at 37°C in a shaking incubator with full aeration. Overexpression of both FDH<sub>H</sub>-*CloCa* and FDH-*CanBo* was efficient under the conditions tested, with no need for induction. The cells were harvested by centrifugation (4,000 × g, 15 min), and approximately 0.5 g of cells were obtained. Cell pellets were resuspended in binding buffer (20 mM sodium phosphate [pH 7.4], 500 mM NaCl, and 20 mM imidazole). Resuspended cells were lysed using a high-pressure homogenizer (Emulsiflex-B15; Avestin, Canada), and the soluble fraction of the cell lysate was collected by centrifugation (20,000 × g, 1 h, 4°C). Proteins were purified by metal ion affinity chromatography (His GraviTrap; GE Healthcare). The supernatant of the cell lysate was loaded onto the column and washed with 10× the column volume of binding buffer, followed by washing with 20× the column volume of washing buffer (20 mM sodium phosphate [pH 7.4], 500 mM NaCl, and 40 mM imidazole). Proteins were eluted with 4 ml of elution buffer (20 mM sodium phosphate [pH 7.4], 500 mM NaCl, and 500 mM imidazole) and concentrated with a YM-10 centrifugation filter (Millipore). Final concentrated proteins were stored in 50 mM phosphate buffer (pH 7.0) at 4°C. The yields of FDH<sub>H</sub>-*CloCa* and FDH-*CanBo* were 0.4 mg and 3.3 mg per 50 ml of cell culture, respectively. The higher yield of the latter reflects its greater overexpression level and larger proportion in the soluble protein fraction. All purification procedures were carried out at 4°C under normal oxic conditions with no atmosphere control. The expression levels of the formate dehydrogenases were assessed by 20% SDS-PAGE (Fig. 2A).

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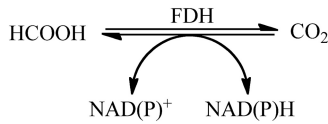


FIG 1 The reaction catalyzed by formate dehydrogenase, coupled to the redox of NAD.

The production of the formate dehydrogenases was also confirmed by liquid chromatography-mass spectrometry (LC/MS). Protein mass spectrometry was carried out by direct injection onto an Agilent 1100 series LC/MSD (mass spectrometer detector) time-of-flight (TOF) instrument running a mobile phase of 50:50 (vol/vol) 0.1% formic acid in MeCN–0.1% aqueous formic acid. The calculated molecular masses based on protein sequence for the N-terminally His<sub>6</sub>-tagged FDH<sub>CanBo</sub> and FDH<sub>H-CloCa</sub> were 41,324 and 80,739 Da, respectively. Mass spectrometry gave molecular masses of 41,324 (data not shown) and 80,735 Da (Fig. 2B) for the expressed FDH<sub>CanBo</sub> and FDH<sub>H-CloCa</sub>, respectively.

Several formate dehydrogenases, as well as many oxidoreductases, are known to contain Mo-pterin cofactors (17, 18); however, there have been cases where W replaces Mo in the cofactor (19). Inductively coupled plasma-optical emission spectrometry (ICP-OES) (Perkin-Elmer) revealed the presence of W in an FDH<sub>H-CloCa</sub> sample, which indicates that this enzyme, like the acetogenic *Moorella thermoacetica* formate dehydrogenase, contains a W cofactor.

**Catalytic activities of FDHs.** The purified FDHs were assessed for their catalytic activities. The initial velocities for enzyme catalysis in both directions were studied for these FDHs by monitoring NADH absorbance at 340 nm. It is impractical to define a reaction equilibrium position for the assay, since CO<sub>2</sub> is a substrate or product and the system is open. However, an excess of either formate or bicarbonate was present in order to ensure that the assays were initiated far from equilibrium. An initial assay was carried out with 1 μM enzyme in 0.1 M sodium phosphate buffer (pH 6.8) at 37°C with an excess

(0.1 M) of sodium formate or sodium bicarbonate (depending on the direction monitored) and 0.2 mM NAD<sup>+</sup> or NADH, respectively. Formate dehydrogenase activity was assessed by measuring the change in absorbance at 340 nm due to the reduction of NAD<sup>+</sup> or oxidation of NADH (Shimadzu UV 2450 spectrophotometer equipped with a thermostated cell compartment).

The two FDHs displayed opposite catalytic behaviors under the conditions tested (Fig. 3A). The eluent from the purification of non-transformed *E. coli* BL21(DE3) cell lysate was used as a control and presented no conversion (data not shown). FDH<sub>CanBo</sub> showed high activity in NAD<sup>+</sup> reduction in the presence of formate, as expected. On the contrary, when presented with NADH with an excess of bicarbonate, it was inactive. Conversely, FDH<sub>H-CloCa</sub> showed minimal activity toward NAD<sup>+</sup> reduction but was active toward NADH oxidation. Under the conditions tested, the activities of these two enzymes seems to be consistent with what is known about their metabolic roles. The progress of the preferred reaction for each enzyme was more directly compared using the aforementioned assay conditions but with 0.1 μM FDH<sub>CanBo</sub> and 5.0 μM FDH<sub>H-CloCa</sub>, under which circumstances similar initial reaction rates were observed (Fig. 3B).

The FDHs were further characterized by monitoring their catalytic activities over a range of substrate concentrations. By maintaining an excess of one substrate and varying the concentration of the other, Michaelis-Menten kinetics were observed. Assays were performed for a range of NAD<sup>+</sup> (0.02 to 0.75 mM) or NADH (0.02 to 0.6 mM) concentrations in 0.1 M sodium phosphate (pH 6.8) at 37°C with 0.1 M sodium formate or sodium bicarbonate and 0.1 μM FDH<sub>CanBo</sub> or 1 μM FDH<sub>H-CloCa</sub>.

From the kinetic data presented in Table 1, it can be seen that both enzymes show similar binding affinities to their dinucleotide substrate; however, FDH<sub>CanBo</sub> exhibits a much higher turnover. This translates into an approximately 50-fold-higher efficiency of FDH<sub>CanBo</sub> than of FDH<sub>H-CloCa</sub> in their respective “preferred” directions.

In order to gain a better understanding of the differences in this

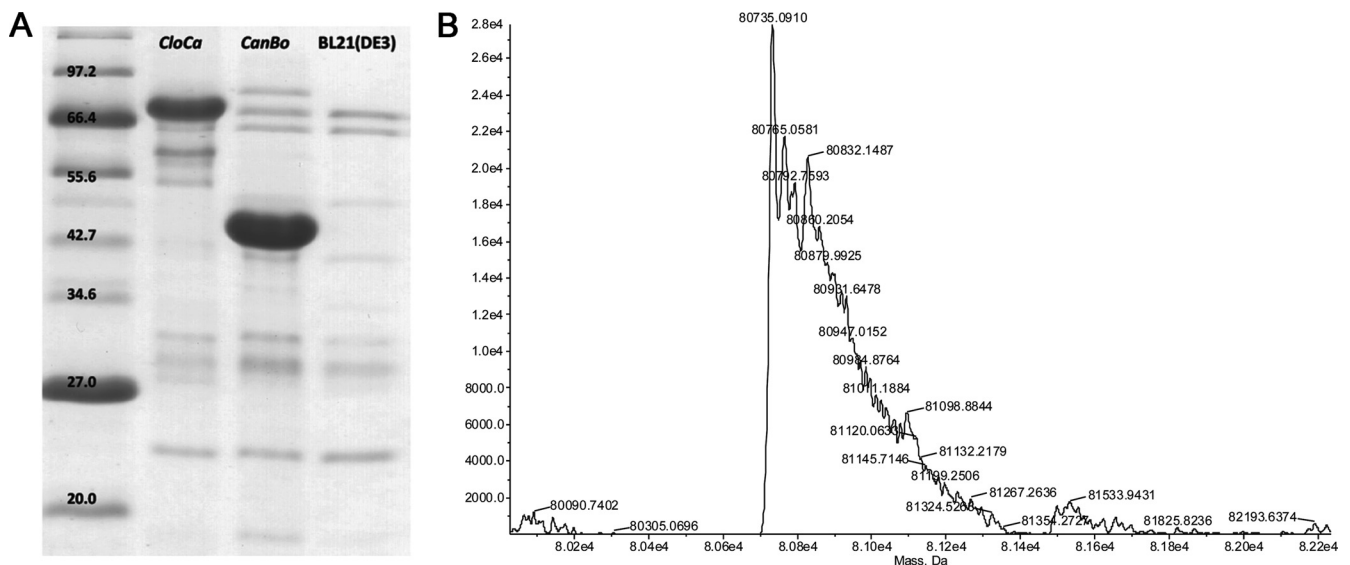


FIG 2 Twenty percent SDS-PAGE of purified FDH<sub>H-CloCa</sub> (80.7 kDa) and FDH<sub>CanBo</sub> (41.3 kDa), as well as the background from blank BL21(DE3) cells (A), and the mass spectrum of FDH<sub>H-CloCa</sub> (B).

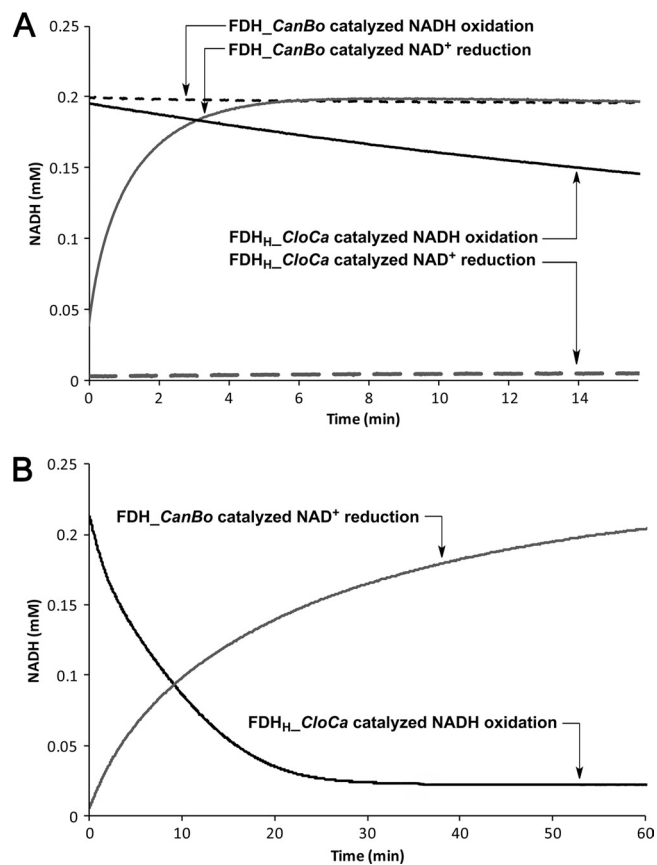


FIG 3 Catalytic activities of FDHs: all with 1  $\mu\text{M}$  enzyme (A) or FDH<sub>CanBo</sub>-catalyzed NAD<sup>+</sup> reduction with 0.1  $\mu\text{M}$  enzyme and FDH<sub>H\_CloCa</sub>-catalyzed NADH oxidation with 5  $\mu\text{M}$  enzyme (B).

preference of the two FDHs, the binding constants of the dinucleotide products were also calculated (Table 2). If a product is added to an assay mixture at initiation, it will compete with a substrate in binding to the enzyme. Since catalysis of the reverse reactions in each case is negligible, the Michaelis constant of the product reduces to the binding constant, which has the significance of an inhibition constant ( $K_p$ ).

$$v_i = \frac{V_{\max} [S]}{K_m \left( 1 + \frac{[P]}{K_p} \right) + [S]}$$

[S] is the substrate concentration, [P] is the product concentration,  $v_i$  is the initial velocity,  $V_{\max}$  is the maximum initial velocity for the substrate,  $K_m$  is the Michaelis constant for the substrate, and  $K_p$  is the Michaelis constant for the product.

TABLE 1 FDH kinetic parameters

Parameter	Value for indicated substrate	
	FDH <sub>H_CloCa</sub> for NADH	FDH <sub>CanBo</sub> for NAD <sup>+</sup>
$K_m$ (mM)	0.05	0.07
$V_{\max}$ ( $\mu\text{M min}^{-1}$ )	5.0	38
$k_{\text{cat}}$ ( $\text{s}^{-1}$ )	0.08	6.3
$k_{\text{cat}}/K_m$ ( $\text{s}^{-1} \text{mM}^{-1}$ )	1.6	90

TABLE 2 FDH formate and nicotinamide binding affinities

Substrate/product	Binding affinity, mM	
	FDH <sub>CanBo</sub>	FDH <sub>H_CloCa</sub>
NAD <sup>+</sup>	0.07 ( $K_m$ )	0.75 ( $K_p$ )
NADH	0.01 ( $K_p$ )	0.05 ( $K_m$ )
Formate	3.7 ( $K_m$ )	>100 ( $K_p$ )

For a fixed concentration of NADH (0.1 mM), the initial velocity of FDH<sub>CanBo</sub>-catalyzed NAD<sup>+</sup> reduction was measured for a range of NAD<sup>+</sup> concentrations (0.1 M sodium phosphate buffer [pH 6.8] and 0.1 M sodium formate). The saturation kinetics observed allowed for calculation of an apparent  $K_m$  ( $K_{m,\text{app}}$ ) (0.74, Eadie-Hofstee), which was used to calculate the binding constant of NADH:  $K_{m,\text{app}} = K_m \{1 + ([P]/K_p)\}$ .

Based on the above equation, the  $K_p$  of FDH<sub>CanBo</sub> for NADH was calculated at 0.01 mM (Table 2). In the case of FDH<sub>H\_CloCa</sub>, it was found that a larger concentration of product (NAD<sup>+</sup>) was required to significantly reduce the initial velocity, which suggested that the affinity of the enzyme for the product dinucleotide is very low. This was confirmed with saturation kinetics observed for various NADH concentrations when 1.5 mM NAD<sup>+</sup> was present in the assay (0.1 M sodium phosphate buffer [pH 6.8] and 0.1 M sodium bicarbonate). The  $K_p$  was calculated at 0.75 mM ( $K_{m,\text{app}} = 0.15$  mM, Eadie-Hofstee), which is approximately 15-fold higher than the  $K_m$  for NADH with this enzyme. The  $K_m$  of FDH<sub>CanBo</sub> for formate was measured at 3.7 mM (0.6 to 100 mM sodium formate, 0.1 M sodium phosphate [pH 6.8], 37°C, and 0.5 mM NAD<sup>+</sup>). In the case of FDH<sub>H\_CloCa</sub>, however, when formate at a concentration of 100 mM was added to the assay mixture (0.1 M sodium phosphate [pH 6.8], 37°C, 0.1 M sodium bicarbonate, and 0.2 mM NADH), the initial reaction velocity remained the same. This suggests that the binding constant of FDH<sub>H\_CloCa</sub> for formate ( $K_{p,\text{HCOO}^-}$ ) has to be greater than 100 mM.

Despite the fact that both enzymes show a preference to bind NADH, it is noteworthy that for FDH<sub>CanBo</sub>, the preference is less ( $K_{m,\text{NAD}^+}$  is 7-fold greater than  $K_{p,\text{NADH}}$ ). This is combined with a relatively high binding affinity for formate, making the enzyme an efficient catalyst of formate oxidation. For FDH<sub>H\_CloCa</sub>, the preference for NADH is greater ( $K_{p,\text{NAD}^+}$  is 15-fold greater than  $K_{m,\text{NADH}}$ ), which, combined with a low binding affinity for formate, makes this enzyme a more efficient catalyst for formate production through CO<sub>2</sub> reduction.

**Concluding remarks.** For the first time, recombinant formate dehydrogenase from the *Clostridium carboxidivorans* strain P7<sup>T</sup> was expressed and purified using an *E. coli* host cell. FDH<sub>H\_CloCa</sub> and FDH<sub>CanBo</sub> prepared in the same way displayed different catalytic behaviors. While FDH<sub>CanBo</sub> was more efficient in NAD<sup>+</sup> reduction, FDH<sub>H\_CloCa</sub> displayed a strong preference for NADH oxidation. Relative to FDH<sub>CanBo</sub>, FDH<sub>H\_CloCa</sub> shows a 10-fold-lower binding affinity for NAD<sup>+</sup> and a binding affinity at least 30-fold lower for formate. These lower affinities for both products of NADH-dependent CO<sub>2</sub> reduction make it a much better catalyst for formate production. Expression was carried out aerobically, and the enzyme retained catalytic activity under normal oxic assay conditions, thus presenting relative robustness in comparison to related enzymes that are reported to be oxygen sensitive. Cur-



rently, we are investigating the activity of FDH<sub>H<sub>2</sub></sub>-CloCa in *E. coli* host cells, where formate production could serve in a useful role in alleviating fossil fuel dependence.

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## REFERENCES

1. Kato N, Sahn H, Wagner F. 1979. Steady-state kinetics of formaldehyde dehydrogenase and formate dehydrogenase from a methanol-utilizing yeast, *Candida boidinii*. *Biochim. Biophys. Acta Enzymol.* **566**:12–20.
2. Overkamp KM, Kötter P, van der Hoek R, Schoondermark-Stolk S, Luttik MAH, van Dijken JP, Pronk JT. 2002. Functional analysis of structural genes for NAD<sup>+</sup>-dependent formate dehydrogenase in *Saccharomyces cerevisiae*. *Yeast* **19**:509–520.
3. San K-Y, Berrios-Rivera SJ, Bennett GN. May 2010. Recycling system for manipulation of intracellular NADH availability. US patent 7,709,261.
4. Ljungdahl LG, Wood HG. 1969. Total synthesis of acetate from CO<sub>2</sub> by heterotrophic bacteria. *Annu. Rev. Microbiol.* **23**:515–538.
5. Lu Y, Jiang ZY, Xu SW, Wu H. 2006. Efficient conversion of CO<sub>2</sub> to formic acid by formate dehydrogenase immobilized in a novel alginate-silica hybrid gel. *Catal. Today* **115**:263–268.
6. Miyatani R, Amao Y. 2002. Bio-CO<sub>2</sub> fixation with formate dehydrogenase from *Saccharomyces cerevisiae* and water-soluble zinc porphyrin by visible light. *Biotechnol. Lett.* **24**:1931–1934.
7. Parkinson BA, Weaver PF. 1984. Photoelectrochemical pumping of enzymatic CO<sub>2</sub> reduction. *Nature* **309**:148–149.
8. Reda T, Plugge CM, Abram NJ, Hirst J. 2008. Reversible interconversion of carbon dioxide and formate by an electroactive enzyme. *Proc. Natl. Acad. Sci. U. S. A.* **105**:10654–10658.
9. Ruschig U, Müller U, Willnow P, Höpner T. 1976. CO<sub>2</sub> reduction to formate by NADH catalyzed by formate dehydrogenase from *Pseudomonas oxalaticus*. *Eur. J. Biochem.* **70**:325–330.
10. Scherer PA, Thauer RK. 1978. Purification and properties of reduced ferredoxin-CO<sub>2</sub> oxidoreductase from *Clostridium pasteurianum*, a molybdenum iron-sulfur protein. *Eur. J. Biochem.* **85**:125–135.
11. Thauer RK. 1972. CO<sub>2</sub>-reduction to formate by NADPH. The initial step in the total synthesis of acetate from CO<sub>2</sub> in *Clostridium thermoaceticum*. *FEBS Lett.* **27**:111–115.
12. Liou JS-C, Balkwill DL, Drake GR, Tanner RS. 2005. *Clostridium carboxidivorans* sp. nov., a solvent-producing clostridium isolated from an agricultural settling lagoon, and reclassification of the acetogen *Clostridium scatologenes* strain SL1 as *Clostridium drakei* sp. nov. *Int. J. Syst. Evol. Microbiol.* **55**:2085–2091.
13. Paul D, Austin FW, Arick T, Bridges SM, Burgess SC, Dandass YS, Lawrence ML. 2010. Genome sequence of the solvent-producing bacterium *Clostridium carboxidivorans* strain P7<sup>T</sup>. *J. Bacteriol.* **192**:5554–5555.
14. Bruant G, Levesque M-J, Peter C, Guiot SR, Masson L. 2010. Genomic analysis of carbon monoxide utilization and butanol production by *Clostridium carboxidivorans* strain P7<sup>T</sup>. *PLoS One* **5**:e13033. doi:10.1371/journal.pone.0013033.
15. Ragsdale SW, Pierce E. 2008. Acetogenesis and the Wood-Ljungdahl pathway of CO<sub>2</sub> fixation. *Biochim. Biophys. Acta* **1784**:1873–1898.
16. Neylon C, Brown SE, Kralicek AV, Miles CS, Love CA, Dixon NE. 2000. Interaction of the *Escherichia coli* replication terminator protein (Tus) with DNA: a model derived from DNA-binding studies of mutant proteins by surface plasmon resonance. *Biochemistry* **39**:11989–11999.
17. Gladyshev VN, Khangulov SV, Axley MJ, Stadtman TC. 1994. Coordination of selenium to molybdenum in formate dehydrogenase H from *Escherichia coli*. *Proc. Natl. Acad. Sci. U. S. A.* **91**:7708–7711.
18. Schwarz G, Mendel RR. 2006. Molybdenum cofactor biosynthesis and molybdenum enzymes. *Annu. Rev. Plant Biol.* **57**:623–647.
19. Ljungdahl LG, Andreessen JR. 1978. Formate dehydrogenase, a selenium-tungsten enzyme from *Clostridium thermoaceticum*. *Methods Enzymol.* **53**:360–372.