1	Microplastics effects in Scrobicularia plana
2	
3 4	Francisca Ribeiro ¹ , Ana R. Garcia ^{2,3} , Beatriz P. Pereira ¹ , Maria Fonseca ¹ , Nélia C. Mestre ¹ , Tainá G. Fonseca ¹ , Laura M. Ilharco ² , Maria João Bebianno ^{1*}
5	¹ CIMA, University of Algarve, Campus de Gambelas, 8000-139 Faro, Portugal
6	2 Centro de Química-Física Molecular and IN – Institute of Nanoscience and Nanotechnology,
7	Complexo I, Instituto Superior Técnico, Universidade de Lisboa, Av. Rovisco Pais 1, 1049-001
8	Lisboa, Portugal
9 10	³ Departamento de Química e Farmácia, FCT, Universidade do Algarve, Campus de Gambelas, 8000-139 Faro, Portugal.
11	
12	*corresponding author: mbebian@ualg.pt
13	
14	
15	
16	
17	
18	
19	
20	
21	
22	
23	
24	
25	
26	

View metadata, citation and similar papers at <u>core.ac.uk</u>

brought to you by TCORE

33 Abstract

34

35 One of the most common plastics in the marine environment is polystyrene (PS) that can be broken down to micro sized particles. Marine organisms are vulnerable to the exposure 36 to microplastics. This study assesses the effects of PS microplastics in tissues of the clam 37 Scrobicularia plana. Clams were exposed to 1 mg L⁻¹ (20 µm) for 14 days, followed by 38 7 days of depuration. A qualitative analysis by infrared spectroscopy in diffuse reflectance 39 mode period detected the presence of microplastics in clam tissues upon exposure, which 40 41 were not eliminated after depuration. The effects of microplastics were assessed by a 42 battery of biomarkers and results revealed that microplastics induce effects on antioxidant 43 capacity, DNA damage, neurotoxicity and oxidative damage. S. plana is a significant target to assess the environmental risk of PS microplastics. 44 45 Key words: ecotoxicology, biomarkers, neurotoxicity, oxidative stress, genotoxicity, 46 bivalves, peppery furrow shell 47 48 49 **Main findings** 50 S. plana is an important target of PS microplastics ecotoxicity. S. plana can be a suitable biomonitor for assessing PS microplastics 51 52 environmental risk. 53 54 55 Introduction 56 57 Plastics are used in everyday life and in several items: cars, electronic equipment, 58 59 furniture, footwear, construction, food packages, among others. The largest plastics producers are the sectors of packaging (39%) and construction (21%), followed by 60 61 transportation, agriculture, household and electronics (Pinto, 2012). The annual production of plastics increased considerably from 1.5 million tons in 1950 (decade where 62 63 the commercial development of polyolefins, polypropylene and polyethylene started), to approximately 322 million tons in 2015 (PlasticsEurope, 2015; Wright et al., 2013b), 64

⁶⁵ representing an increase of 9% per year, approximately. This production volume, coupled

with their high durability, resistance to degradation, low weight and low recycled volume,
leads to the widespread and accumulation of discarded plastics in landfills and, as litter,
in terrestrial and aquatic habitats worldwide (Derraik, 2002; Moore, 2008; Thompson *et al.*, 2004). The consumption of plastics in many European countries indicates that the
plastic resins most used since 2007 are low density polyethylene (LDPE), high density
polyethylene (HDPE), polypropylene (PP), polyvinyl chloride (PVC), polyethylene
terephthalate (PET) and polystyrene (PS).

It is not possible to obtain reliable estimates of the amount of plastic debris 73 74 reaching the marine environment, but they are, however, quite substantial (Derraik, 2002). The major sources of plastic materials and debris in the sea are fishing fleet 75 76 (Cawthorn, 1989), and marine recreational activities (Pruter, 1987; Wilber, 1987). Plastic also reaches the sea as litter from land-based sources, carried by rivers and municipal 77 78 drainage systems (Derraik, 2002; Williams & Simmons, 1997). In the sea, these versatile 79 and non-biodegradable polymers are found in the form of larger items (macroplastics), 80 including hulls of boats and fishing nets many meters long, and tiny fragments (Browne et al., 2008; Canesi et al., 2015). When exposed to UV-B radiation, to the oxidative 81 82 properties of the atmosphere and to the hydrolytic properties of seawater, these plastics brittle and break into smaller pieces, until they reach micrometres in length 83 (microplastics), and potentially, also the nano-scale level (nanoplastics) (Browne et al., 84 85 2008; Canesi et al., 2015).

Microplastics are defined as particles with less than 5 mm in diameter, according 86 to the National Oceanic and Atmospheric Administration of the United States of America 87 (NOAA, 2015). Their presence in the ocean is distributed according to the currents 88 (Lusher, 2015). The distinction between primary and secondary microplastics is based on 89 whether these particles were originally manufactured to be that size (primary) or whether 90 91 they resulted from the breakdown of larger items (secondary) (Kershaw, 2015). The 92 primary source of microplastics includes different typology; polyethylene (PE), 93 polypropylene (PP) and polystyrene (PS) from cleaning products or cosmetics (Fendall & Sewell, 2009), or from industries or industrial effluents (Lusher, 2015). The secondary 94 95 source is the degradation of plastics under marine conditions, that dramatically reduces the molecular weight of these polymers (Andrady, 2011). Ideally, these particles may also 96 undergo further degradation by microbial action, releasing carbon (Andrady, 2011). 97 Although microplastics greatly exceed large plastic items in marine systems, they are still 98 99 only a small proportion of the total mass of plastics in the ocean. Therefore, microplastics became a growing issue in such a way that the Marine Strategy Framework Directive
(MSFD N° 2008/56/EC) highlights microplastics and their associated chemicals as one
of the major policy descriptors whose impact need to be assessed in the marine
environment (Zarfl *et al.*, 2011).

104 The presence of microplastics is documented in most habitats in the open ocean, seas and beaches, surface waters, the water column and in the deep ocean (Lusher, 2015), 105 and recently, in freshwater systems (Eerkes-Medrano et al., 2015). In Portugal, 62% of 106 microplastics were identified in the North Atlantic by trawling, with a density of 580 000 107 particles per km². 61% of the water samples collected in the Portuguese coast contained 108 microplastics and the concentration was higher in the Vicentina Coast and Lisbon (0.036 109 and 0.033 particles m⁻³, respectively) than in the Algarve and Aveiro areas (0.014 and 110 0.002 particles per m³, respectively) (Lusher, 2015). 111

112 The occurrence of plastic in the ocean and the potential impact to marine organisms are of growing concern (Canesi et al., 2015). The fact that microplastics have 113 114 such a small size and different shapes actively contributes to their bioavailability and 115 accumulation in organisms of lower trophic levels. As the particles interact with plankton 116 and sediments, both organisms that feed on suspended particles and the ones that feed on the bottom are at risk of, accidentally or selectively, ingesting plastic (Lusher, 2015). 117 Particles with less than 20 µm are likely to be ingested and egested (Lee et al., 2013) by 118 small organisms (Thompson et al., 2004; Wright et al., 2013b). Microplastics with size 119 120 between 1 and 5 mm can compromise the nutrition and digestion (Codina-García et al., 2013). The ingestion of plastics with a greater size can cause serious external and internal 121 injuries, ulcers, digestive tract blockage, false sense of fullness, loss of feeding capacity, 122 impairment and inability to avoid predators or death (Gall & Thompson, 2015). 123

Microplastics ingestion was documented for a wide range of marine vertebrates 124 125 and invertebrates for wild populations (Table 1). Whilst it is apparent that microplastics have become widespread and ubiquitous in the marine environment, the information on 126 127 accumulation, mode of action and biological impact of this emerging contaminant in marine organisms is still scarce (Wright et al., 2013b). Microparticles are inert and, after 128 129 ingestion, pass through the cell membrane and are incorporated into tissues, particularly in the gut cavity (Wright et al., 2013b). After passing to the circulatory system they are 130 131 transferred to other organs, via hemolymph, where they are retained for several weeks in several organs and cause adverse effects (Browne et al., 2008). Experiments with 132

different sizes (Mazurais *et al.*, 2015), shapes (Graham & Thompson, 2009) and types
(Green *et al.*, 2016) were conducted in order to mimic a more realistic scenario and truly
assess the effects of microplastics. Table 2 summarizes the information about the
ecotoxicological effects of microplastics in several bivalves that include weight loss,
reduced feeding activity, increased phagocytic activity, transference to the lysosomal
system, accumulation and inhibition of acetylcholinesterase (AChE) activity (e.g. Avio *et al.*, 2015a; von Moos *et al.*, 2012; Van Cauwenberghe *et al.*, 2015).

140 The aim of this study was to investigate the presence and mode of action of 141 polystyrene microparticles (one of the most largely used plastic worldwide, with a density of 1.09 g cm⁻³) in different tissues of the peppery furrow shell Scrobicularia plana and 142 assess the potential ecotoxicological risk of this emerging contaminant in this species. 143 144 The presence of microplastics in gills and digestive gland of S. plana was evaluated by 145 infrared spectroscopy in diffuse reflectance mode and their effects by using a battery of 146 biomarkers of oxidative stress (superoxide dismutase (SOD), catalase (CAT), glutathione peroxidases (GPx), biotransformation (glutathione-S-transferases (GST)), genotoxicity, 147 148 neurotoxicity and oxidative damage.

- 149
- 150

151 Materials and methods

152

153 Microplastics characterization

154 Monodisperse PS microplastics were obtained from Sigma-Aldrich (Germany) 155 with the particle size 20 μ m and density 1.05 g cm⁻³. Two stock solutions (100 mg L⁻¹) 156 were prepared: one in ultrapure water (18 MΩ/ cm) and another in natural seawater (S = 157 35), both maintained in constant aeration.

The microplastics size was determined by optical microscopy (OM) and dynamic light scattering (DLS), and the surface charge (zeta potential) by electrophoretic light scattering (ELS), for both PS solutions. The zeta potential of the microparticles was determined by electrophoresis mobility measurements at 25°C using a DLS particle sizer (ZetaSizer Nano ZS90, Malvern Inc.) in a disposable polycarbonate capillary cell (DTS1061). The sedimentation rate (SR) was measured by the change of turbidity with time (0-24 h), as described in Sousa and Teixeira (2013). The SR relates to the normalized microparticle turbidity C/C₀, where C is the turbidity at time t and C₀ the initial turbidity at time 0. Then, the SR is given by the expression δ (C/C0)/ δ t, estimated from the decrease in turbidity (C/C₀), which occurred within the first two hours for the fast sedimentation (fast SR) conditions and within 3-24 h for slow sedimentation (slow SR) conditions (Keller *et al.*, 2010).

- 171
- 172 Laboratory exposure assay

Scrobicularia plana (38 ± 5 mm shell length) were collected in Cabanas de Tavira, 173 Ribeira do Almargem (South of Portugal) (N 37°7'59.75" W 7 36'34.95") and transferred 174 175 to the laboratory, where they were acclimated for 7 days at constant aeration, with a photoperiod of 12h light and 12h darkness. Three replicate aquaria were used for each 176 control and exposed group to 1 mg L^{-1} of PS microplastics which corresponded to around 177 4 particles ml⁻¹. This value is below the concentrations used in previous laboratory 178 exposure experiments (summarized in Table 2) although higher than environmentally 179 relevant concentrations of microplastics found in seawater in regions highly 180 contaminated, e.g. ~0.5 particles mL⁻¹ in South Korea (Song et al., 2014) or 0.1 particles 181 mL⁻¹ (Norén, 2007). 182

Sixty clams were placed in each glass aquaria filled with 20 L of natural seawater 183 with constant aeration and no sediments were added. Glass Pasteur pipettes were used to 184 185 provide aeration and the use of plastic material was avoided during the experiment. Exposure ran for 14 days, followed by 7 days of depuration. The water was changed every 186 187 24 hours with subsequent addition of PS microplastics. Immediately before addition of PS microplastics, the 100 mg L⁻¹ stock solution in ultra-pure water was sonicated for 30 188 189 minutes (Ultrasonic bath VWR International, 230 V, 200 W, 45 kHz frequency). During the experiment abiotic parameters were checked in all tanks by measuring temperature 190 191 $(18.0 \pm 1 \text{ °C})$, salinity (35 ± 0.2) , percentage of oxygen saturation (93.0 %) and pH (7.8), with the multiparametric probe TRIPOD (from PONSEL). Clams were not fed to avoid 192 193 any interaction of microplastics and food.

Unexposed and exposed clams were collected after 0, 3, 7 and 14 days of exposure, and after the 7 days of depuration. The hemolymph was immediately collected by gently prying the shell open approximately 2 to 3 mm with a scalpel and the shell was

held open with forceps. Then, the posterior adductor muscle of the S. plana was gently 197 penetrated with a sterile hypodermic syringe 1 ml (12 mm x 12:33) and the hemolymph 198 easily collected using intermittent suction. To avoid potential contamination of the 199 haemolymph, the water inside the shell was drained from each clam prior to hemolymph 200 201 extraction, and a new syringe was used for each replicate of each treatment and each 202 sampling day. Gills and digestive gland were dissected and stored at -80 °C until analysis. No significant mortality was observed between treatments, during the accumulation and 203 depuration periods (p > 0.05). 204

205

206

207 Qualitative assessment of microplastics accumulation

208 A qualitative assessment of microplastics accumulation and transport to different 209 tissues was conducted by optical microscopy (OM) analysis of the hemolymph of control 210 and exposed clams at day 14. In addition, gills and digestive gland tissues collected at different times of exposure were lyophilized at -40 °C, during approximately 48 hours 211 with a Modulyo freeze dryer and analysed by infrared spectroscopy. This analysis was 212 performed in a Mattson RS1 Fourier transform infrared spectrophotometer, with a wide 213 band MCT (mercury cadmium telluride) detector, in the range 400-4000 cm⁻¹, at 4 cm⁻¹ 214 215 resolution. Since the samples were powders, in order to avoid compressing the tissues to the high pressures needed to prepare disks, the most convenient mode for obtaining the 216 spectra was in diffuse reflectance (DRIFT). Each lyophilized sample was diluted (~1:4) 217 in KBr (from Aldrich, FTIR grade) and finely grinded in an agate mortar, to reduce 218 particle size and thus decreases diffuse Fresnel reflectance. This mixture was placed in a 219 220 11 mm diameter sample cup and pressed to obtain a very smooth surface. The cup was filled, in order to attain a so-called infinite thickness (all the light is reflected or absorbed 221 222 by the sample) and mounted in a Graseby/Specac Selector accessory, to collect all the diffusely reflected radiation, excluding specular reflection. Each DRIFT spectrum 223 224 resulted from the ratio of 500 single-beam scans obtained for the sample to the same 225 number of background scans for pure KBr.

The spectra in reflectance (R_{∞}) were transformed into Kubelka-Munk units $[f(R_{\infty})]$ using the FIRST software, according to the Kubelka-Munk equation (Mitchell, 1993; Stuart, 2005)

- 229
- 9 $f(R_{\infty}) = \frac{(1-R_{\infty})^2}{2R_{\infty}}$

The spectra were baseline corrected, normalized to a typical band of the tissuesnot overlapped with a PS band, and the average of 10 samples was calculated.

- 232
- 233

234 Condition index

To assess the physiological status of control and PS exposed clams, soft tissues and shells were weighted, and the condition index (CI) determined as the percentage (%) of the ratio between drained weight of the soft tissues (g) and total weight (g), according to Gomes *et al.* (2013).

239

240 Biomarker analysis

241

242 Antioxidant enzymes

243

Prior to the analysis of the enzymatic activities, the tissues (gills and digestive 244 glands) of control and microplastic exposed clams (6 replicates of individual tissues) were 245 weighed and rapidly buffered in Tris-HCl buffer (50 mM Tris-HCl, 250 mM Sucrose, 246 5mM MgCl₂, 1mM DTT, pH=7.6) (the tissue-to-buffer ratio was 1:3 wet weight 247 tissue/volume of buffer), homogenized in an ice bath and centrifuged at 10000 g, for 10 248 minutes, at 4 °C, using a biofuge stratus 230 V centrifuge (Thermo scientific, Germany). 249 Both the soluble and pellet fractions were stored at -80 °C for future analysis. Enzyme 250 activities were measured in the cytosolic fraction. To determine SOD activity, the 251 252 reduction of cytochrome c by the system xanthine oxidase/hypoxanthine was measured at 550nm (McCord & Fridovich, 1969) and results expressed in U mg⁻¹ of total protein 253 concentration. CAT activity was determined by the decrease in absorbance at 240nm due 254 to H_2O_2 consumption, with a molar extinction coefficient of 40 M⁻¹ cm⁻¹ (Greenwald, 255 1987) and results expressed as μ mol min⁻¹ mg⁻¹ of total protein concentration. GP_X 256 activity was measured through NADPH oxidation in the presence of excess glutathione 257 reductase, reduced glutathione and hydroperoxide as substrate, at 340nm (Lawrence & 258 Burk, 1978) and results expressed as $mol min^{-1} mg^{-1}$ of total protein concentration. GST 259 activity was measured by the conjugation of 1-chloro-2,4-dinitrobenzene (CDNB) with 260

reduced glutathione (GSH) and the increase of absorbance measured at 340 nm (\mathcal{E} 340 (CDNB) = 9.6 mM⁻¹ cm⁻¹) (Habig *et al.*, 1974). The results are expressed in µmol CDNB min⁻¹ mg protein⁻¹.

- 264
- 265

266 **Oxidative damage**

267 Before the analysis of oxidative damage, gills and digestive gland (6 replicates of individual tissues per treatment) were weighed and rapidly buffered in 0.02M Tris-HCl 268 (0.1 M HCl, 0.2 M Tris, pH=8.6) (the tissue-to-buffer ratio was 1:3 wet weight 269 270 tissue/volume of buffer). Then, 10 µl of BHT (Butylated hydroxytoluene) was added, per 271 each ml of Tris-HCl [0.02M]. The samples were homogenized in an ice bath and centrifuged at 30000 g, for 45 minutes, at 4 °C, using a biofuge stratus 230 V centrifuge 272 273 (Thermo scientific, Germany). Both the cytosolic and mitochondrial fractions were stored 274 at -80 °C for future analysis. Lipid peroxidation (LPO) was quantified based on the method described by Erdelmeier et al. (1998) - determining malondialdehyde (MDA) and 275 4-hydroxyalkenals (4-HNE) concentrations upon the decomposition by polyunsaturated 276 fatty acid peroxides. The tissue supernatant (200 µL) was incubated at 45 °C, for 60 277 minutes, with 650 µL of 1-methyl-2-phenylindone diluted in methanol and 150 µL of 278 279 methanesulfonic acid. The absorbance was measured at 586 nm and LPO levels are expressed as nmol malondialdehyde (MDA) + 4-HNE g^{-1} per mg of protein. 280

281

282 AChE activity

Gills (6 replicates of individual tissues per treatment) were homogenized on ice in 283 284 five volumes of a Tris-HCl buffer (100 mM, pH 8.0) containing 10% Triton X-100 and centrifuged at 12000g, for 30 minutes, at 4° C. Anti-cholinesterase activity was measured 285 286 by the modified Ellman's colorimetric method (Ellman et al., 1961), assessed by the addition of Ellman's reagent – DTNB - using AChEI (acetylcholine) as substrate, for the 287 estimation of respective thiocholine (ChE). The absorbance is measured, at 405 nm 288 (coefficient of extinction of $\varepsilon = 13.6 \text{ mM}^{-1} \cdot \text{cm}^{-1}$), to estimate the amount of ChE liberated 289 by the reaction, which is proportional to the AChE activity (Colovic et al., 2013). The 290 results are expressed by nmol AChEI min⁻¹ mg protein⁻¹. 291

293 Genotoxicity

294 Genotoxicity was estimated using the comet assay in a slightly modified version of Singh et al. (1988) and described in Almeida et al. (2011). Microscopic slides were 295 coated with 0.65% normal melting point agarose (NMA), in Tris-acetate EDTA. After 296 297 collection, hemolymph cells were centrifuged at 3000 rpm for 3 min (4 °C), and the pellets with isolated cells suspended in 0.65% low melting point agarose (LMA) in Kenny's salt 298 299 solution, and casted on the microscope slides. Afterwards, the slides with the embedded cells were immersed in a lysis buffer (2.5 M NaCl, 100 mM EDTA, 10 mM Tris, 1% 300 Triton X-100, 10% Dimethylsulfoxide, 1% Sarcosil, pH 10, 4 °C), for the diffusion of 301 302 cellular components and DNA immobilization in agarose. Following the lysis step, slides 303 were placed in an electrophoresis chamber containing electrophoresis buffer (300 mM NaOH, 1 mM EDTA, adjusted at pH 13, 4 °C), gently submerged and left in this solution 304 305 for 15 minutes, to permit DNA unwinding. The electrophoresis was carried out and, once 306 concluded, the slides were removed and immersed in a neutralizing solution (0.4 mM Tris, pH 7.5) and rinsed with ultrapure water. Then, the slides were allowed to dry for 48 307 hours, from which the analysis in fluorescence microscope was possible. The slides were 308 stained with 4,6-diamidino-2- phenylindole (DAPI, 1 mg mL⁻¹) and the presence of 309 comets analysed using an optical fluorescence microscope (Axiovert S100) coupled to a 310 camera (Sony). Fifty randomly chosen cells for each slide (25 in each gel from each 311 312 organism) of nine organisms were scored with the Komet 5.5 image analysis system (Kinetic Imaging Ltd) at a total magnification of x400. Different parameters of the comet, 313 including the olive tail moment (OTM), comet tail length and amount of DNA in the 314 comet tail (Tail DNA %) were measured, and results are expressed as mean \pm STD. 315

316

317 Statistical analysis

318

Statistical differences between treatments, time of exposure/depuration and tissues were assessed using parametric tests (two-way ANOVA, followed by the Tukey's test) and non-parametric tests (Kruskal-Wallis, followed by multiple comparisons), after normality (Shapiro-Wilk W test) and homogeneity of variances verified (Levene's test).
Principal component analysis (PCA) for gills and digestive gland was used to evaluate the influence of PS microplastics in the determined biomarkers, in exposed clams, along the period of exposure and assess the overall results. Statistical significance was defined at p < 0.05.

327

328 Results

329

330 PS microplastics characterization

331

PS microplastics characterization is summarized in Table 3. DLS measurements show that d_h (hydrodynamic diameter) of PS microplastics is $18.4 \pm 1.33 \mu$ m, which is similar to the size specified by the manufacturer. ζ - potential measurements show that PS microplastics have highest negative surface charge in seawater (-12.4 ± 2.36 mV) than in Milli-Q water (-52.6 ± 2.34 mV) (p < 0.05), showing the tendency of these PS microplastics to aggregate in seawater.

Turbidity (C/C_0) of PS microplastics suspension was measured to assess the 338 339 sedimentation rate (SR) (Figure 1). The SR is faster during the first two hours, estimated from the initial 5% decrease in normalised particle turbidity and slows down after this 340 341 period of time. No significant differences in turbidity decrease were observed between Mili-Q water (97.94%) and seawater (94.65%) over time (24 h) (p > 0.05). However, a 342 343 sharp decrease in turbidity is observed in MQ water during the first 2 hours (1.68×10^{-1}) h⁻¹) although higher when compared to seawater $(1.04 \times 10^{-1} \text{ h}^{-1})$ (p < 0.05), while for the 344 last 22 hours there is a higher decrease in SR for MO water $(3.05 \times 10^{-4} h^{-1})$ compared to 345 seawater (1.16 x 10⁻³ h⁻¹) indicating that, after the initial time, PS microplastics tend to 346 sediment faster in natural seawater (p < 0.05). 347

348

349 **Condition index**

The condition index at the beginning of the experiment (time 0) was 36.01 ± 4.04 %. No significant changes were observed between unexposed and exposed organisms after 14 days (unexposed: 33.05 ± 4.76 %; exposed: 31.53 ± 5.30 %; p>0.05) and in the elimination period (unexposed: 31.31 ± 4.58 % exposed: 31.83 ± 4.72 %; p>0.05), indicating that the organisms were in good health throughout the duration of the experiment.

356

357 Qualitative assessment for microplastics accumulation

In order to corroborate the presence of PS in the clams exposed to the aquatic environment containing microplastic, optical microscopy and infrared data were analysed.

The PS microplastics present in the hemolymph observed under the OM are in Figure 2. The presence of a small aggregate of PS microplastics is observed in image A and two polystyrene particles in image B (highlighted by circles) indicating that, like for seawater, there is a tendency for the PS microplastics to form small aggregates/agglomerates in the hemolymph.

The diversity of the tissue samples implies variations of the infrared spectra from individual specimens in the same conditions. Moreover, due to the amount of PS microplastics expected to accumulate during a 14 days exposure, the spectral analysis is not intended as quantitative, but only as a proof of detection of PS in the gills and digestive gland of the clams. It consisted in a comparison of the average spectra of unexposed clams with clams exposed to PS microplastics.

372 In Figure 3, the DRIFT spectra of digestive gland and gills from clams not exposed to PS microplastics (DG T0 and Gills T0, respectively) are compared to that of digestive 373 374 gland from specimens into which microplastics were directly added in vitro, in a known 375 amount (50 µl). The spectrum of the same PS microplastics is also included in Figure 3 for comparison. In order to allow the comparison of the relative intensities of the bands, 376 the spectra were normalized to the amide II mode (C-N stretching), at 1543 cm⁻¹ 377 (Matthäus et al., 2008), common to all of them, and not overlapped with any PS band or 378 379 the water deformation mode. The average spectra of the digestive gland and gills are similar, with some features specific to the gills: a small band at 3060 cm⁻¹, a clear splitting 380 of the bands at 1082/1047 cm⁻¹ and at 609/588 cm⁻¹. On the other hand, the region below 381 1200 cm⁻¹ is much stronger (relatively to the amide II band) for the digestive gland. The 382 383 average spectrum of digestive gland with added microplastics (DG Added PS) shows a general increase of relative intensities in the regions overlapped with PS bands; new small 384 bands appear at 696 and 737 cm⁻¹ that may be assigned to the strongest bands of the 385 polymer, slightly shifted due to interactions with the specimen tissues. In the spectrum of 386 PS microplastic these bands appear at 700 and 756 cm⁻¹, and are related to out of plane 387 C-H deformation modes of the benzene rings (Holland-Moritz & Siesler, 1976; Koenig, 388 1999; Liang & Krimm, 1958). 389

The average DRIFT spectra of samples obtained before exposure to PS microplastics (T0) were compared to those exposed for 14 days (T14) and after 7 depuration days (T21) (Figures 4A and C for gills and digestive gland, respectively). Both
for gills and digestive gland the spectral region where PS absorbs mostly is stronger after
the exposure period and decreases during depuration, not achieving the initial relative
intensities.

396 The spectral subtractions (T14-T0) and (T21-T0) were calculated and compared 397 to the spectrum of PS microplasetics (Figures 4B and D for gills and digestive gland, respectively). The differences (T14-T0) are generally positive for both tissues, evidencing 398 the accumulation of PS microplastics during exposure. For the gills, the region where PS 399 absorbs more strongly (below 1000 cm⁻¹) corresponds to larger differences between T14 400 and T0, and some relative maxima observed in the high wavenumber region are 401 coincident with (or slightly shifted from) PS bands. For the digestive gland, clear maxima 402 in T14-T0 differences are coincident with PS bands, at 1725, 1590, 1435, 1360 and 1275 403 cm⁻¹. In the low wavenumber region, the spectra at 0 and 14 days are both strong, 404 405 rendering the definition of eventual PS bands more difficult. A partial elimination of PS 406 during the depuration period is unambiguous from the spectral differences (T21-T0) that 407 remain positive for both tissues, although smaller than (T14-T0).

408

409

410 Enzymatic activity

The antioxidant enzymes (SOD, CAT, GPx) and GST activities in the gills and 411 digestive gland of S. plana are presented in Figure 5. Only CAT shows a tissue specific 412 413 response with higher activity in the digestive gland when compared to the gills (p < 0.05). In the gills, the exposure to PS microplastics induces a significant increase in SOD 414 activity after 7 days of exposure that continues in the depuration period (p < 0.05) (Figure 415 416 5A), while for CAT activity there was a significant increase only after three days of exposure (p < 0.05) (Figure 5B). Similarly, in PS exposed clams, GPx activity increases in 417 clams exposed to PS microplastics after the 3^{rd} day of exposure (p < 0.05) and similarly 418 to SOD, GPx continues to increase at the end of the depuration period (p < 0.05) (Figure 419 5C). Moreover, exposure to PS microplastics induces an enhancement of GST activity at 420 the end of the exposure period (p < 0.05), but it decreases in the depuration period (p421 <0.05) (Figure 5D). 422

In the digestive gland, the SOD activity, like in the gills, increases with time of 423 exposure, but this increase was only significant at the end of the exposure period (day 14, 424 p < 0.05). Similarly to the gills, SOD activity in the digestive gland also increases at the 425 end of the depuration period (p < 0.05) (Figure 5E). On the other hand, CAT activity in 426 PS contaminated clams decreased (p < 0.05) with no differences in the depuration period 427 428 (p > 0.05) (Figure 5F). GPx activity, however, showed a bell shape behaviour with a significant increase after the 3rd day of exposure and a decrease in the remaining period 429 (Figure 5G). In clams exposed to PS microplastics, GST activity only significantly 430 decreased after the 3^{rd} day of exposure (p < 0.05). This decrease was similar at the end of 431 the depuration period (p < 0.05) (Figure 5H). 432

433

434 Comet assay

Genotoxic effects for both treatments (CTR and PS exposed clams), analysed by the comet assay and expressed as % of tail DNA and Olive Tail Moment are in Figure 6A-B, respectively. No significant changes were observed for the % of tail and OTM between controls and exposed clams except for day 7 where a significant increase occurred for OTM in exposed clams (p < 0.05). In the depuration period, significant differences were detected between control and clams previously exposed to PS microplastics (p < 0.05) for both parameters.

442

443 AChE activity

The activity of AChE in the gills is in Figure 7. In PS exposed group the AChE activity significantly decreased (p < 0.05) on day 3 when compared to T0. AChE activity was also significantly lower on day 3 and 14 of exposure and after the depuration period, when compared to non-exposed clams (p < 0.05).

448

449 **Oxidative damage**

450 LPO levels were significantly higher in the digestive gland then in the gills ((p451 <0.05; Figure 8A-B). In the gills of PS exposed clams, although LPO levels remained 452 unchanged through time, they significantly decreased compared to those at the beginning 453 of the experiment (day 0) and to the other sampling days (p <0.05) while in the digestive 454 gland, in PS exposed clams, LPO levels significantly increased after 7 days of exposure 455 compared to controls (p < 0.05). In the elimination period, in the digestive gland of PS 456 exposed clams LPO levels significantly decreased when compared to controls (p < 0.05).

- 457
- 458

459 Principal Component Analysis

PCA was applied to all the data for the gills and digestive gland to explain the 460 effects of PS microplastics on biomarkers responses (Figure 9). Regarding the gills, the 461 two principal components represent 72.7 % of total variance, with PC1 representing 53.4 462 % and PC2 19.3 % (Figure 9 A). PCA indicates a clear separation between the initial time 463 464 of the experiment (T0) and the remaining days, between exposed clams and controls and 465 also a clear separation of the depuration period in exposed clams (T21) comparing to 466 others. SOD, CAT, GPx and the genotoxic parameters are in the positive part of PC1, 467 closely related with PS exposed clams, principally after 7 days of exposure (MICR 7). LPO and AChE are negatively related to the other biomarkers in PC1 and are more 468 influenced by non-exposed clams. In PC2, SOD, LPO, AChE and genotoxicity are in the 469 negative part, with SOD and genotoxic parameters being more influenced by exposed 470 471 clams on day 7 and after depuration (MICR 21). CAT, GPx and GST are in the positive part of PC2, with particularly GST being more influenced by exposed organisms at days 472 3 and 14. 473

474 In the digestive gland, the two principal components represent 75.2 % of total variance, with PC1 representing 53.4 % and PC2 21.8 % (Figure 9 B). In this case, there is a clear 475 separation between unexposed and exposed clams, where all the non-exposed clams are 476 477 in the positive part of the PC1. SOD and the genotoxic parameters are in the negative part of PC1 and more related to exposed clams at days 7 and 14 together with the MICR 21, 478 479 while CAT, GPx, GST and LPO are on the positive side and more related to non-exposed 480 clams. PC2 clearly isolates exposed clams at day 3 (MICR 3) from the rest, although 481 exposed clams from day 7 also fall in the negative part of PC2. Also in the negative part 482 of PC2 are LPO and GPx, with GST, CAT and SOD and Tail DNA in the positive side.

483

```
484 Discussion
```

485

The presence and mode of action of polystyrene microparticles in the gills and digestive gland of the peppery furrow shell *S. plana* and the potential ecotoxicological risk of this emerging contaminant was evaluated by characterizing the type of microplastics used and identifying their presence using infrared spectroscopy in diffuse reflectance mode, and by assessing their effects using a battery of biomarkers. These biomarkers were chosen to evaluate the possible effects that come from the physical damage caused by the microparticles, namely: oxidative stress and oxidative damage (through enzymatic activity and LPO), which in turn can lead to genotoxicity and/or neurotoxicity (analysed through the comet assay and the AChE activity).

One of the questions that arise from laboratory experiments is whether they are 495 able to really mimic the natural environment (Phuong et al., 2016) in terms of 496 environmental relevant exposure concentrations. Higher concentrations of microplastics 497 have been found in sediments, 3.3 particles g⁻¹ (Rhine estuary; Leslie et al., 2013) or 62 498 particles g⁻¹ (Wadden sea islands; Liebezeit & Dubaish, 2012). A major concern is the 499 lack of information on the environmental concentrations for plastic particles smaller than 500 50 µm, with only a few exceptions reported and where it was noted that the smaller 501 502 particles were much more abundant than the larger ones (Song et al., 2014). For these 503 reasons, it was considered that the concentration of 1 mg L⁻¹ (4 particles ml⁻¹) was a good 504 compromise between measured environmental concentrations taking into account, both 505 the water and sediment compartments, acknowledging also the lack of data on the 506 concentration for smaller sized particles and their potential to exist in the environment in higher concentrations than those currently detected for larger particles. 507

508 The combination of multiple analytical techniques (OM, DLS, ELS) to characterize PS microplastics using both natural seawater and ultrapure water (Milli-Q) 509 510 (Table 3 and Figure 1) provide an insight about the hazard and risk of these microparticles in the aquatic environment. Size and density are important proxies for microplastics 511 512 bioavailability that can also be enhanced by biological factors (Wright et al., 2013b). The PS microplastics used had a density of 1.09 g/cm³ and took 24 hours to sediment. Data 513 indicated that the particles start reaching the bottom after 2 hours of exposure and thus 514 are available to the organisms present therein. The PS microparticles used in this assay 515 516 are spherical in shape and of the same size (20 µm) and tend to form small aggregates in seawater (Table 3). Similarly, the ingestion of PS spheres (100 nm) in suspension-feeders 517 bivalve molluscs increased when they were fed with microplastics aggregates generated 518 manually in the laboratory (Wright et al., 2013a). Therefore it is hypothesised that the 519 effects of PS microplastics might be caused by the formation of aggregates when 520 accumulated in clam tissues, although it could be different if the microparticles were 521

heterogeneous and not of the same shape and size (Frias, 2015). Most of the microplastics found in the marine environment are secondary microplastics, resulting from the degradation of larger pieces, and thus, with different sizes, shape and sharp edges, that may contribute to injuries in the digestive tract (Browne *et al.*, 2008; von Moos *et al.*, 2012).

S. plana is able to accumulate PS microplastics from seawater in the gills and 527 digestive gland (Figures 3 and 4). Bivalves are able to select particles before ingestion 528 but not after ingestion (Wright et al., 2013b). The presence of microplastics in the gills 529 530 was noted after 14 days of exposure, with a partial recovery during the depuration period. This suggests that PS microplastics are trapped in this organ, the first in contact with the 531 532 polymer. The microparticles are also ingested through the inhalant siphon and subsequently transported to the mouth and to the digestive gland for intracellular 533 534 digestion (Hughes, 1969). This was noted by the presence of PS microparticles in S. plana digestive gland (Figures 4 C-D), where they are likely not digested, and some of them 535 536 were eliminated. The presence of small aggregates of microplastics in the haemolymph (Figure 2) indicates that PS microplastics were transported into the circulatory system, 537 538 indicating a possible translocation, where they can be retained for several weeks and then 539 transported to several tissues where they can cause harm. But, the question is if the 540 translocation really occurred or was it a contamination by microplastic while sampling haemolymph? Lambert et al. (2014) discuss whether the presence of microplastics in the 541 542 circulatory system is due to translocation or contamination due to sampling technique. PS microplastics in *M. edulis* persisted in the circulatory system for 48 days (Browne et al., 543 544 2008), giving an evidence of microplastic translocation. The accumulation, upon ingestion, of 2 μ m and 4-16 μ m PS fluorescently labelled microplastics (0.51 μ g L⁻¹) in 545 the gut cavity and digestive tubules of *Mytilus edulis* was also observed by Browne et al. 546 547 (2008), after 12 hours of exposure. The same author identified PS microparticles in the 548 haemolymph and haemocytes of the same mussel species although no toxicity effects 549 were observed. Conversely, in mussels exposed for 48h to microplastics (1-80 µm) an 550 increase in haemocytes and a decrease in lysosomal membrane stability was observed in 551 the same mussel species, indicating the presence of an inflammatory process (Bowmer & 552 Kershaw, 2010). Also, von Moos et al. (2012) shows that HDPE particles (0-80 µm) were 553 transported to the digestive gland where they accumulated in the lysosomal system of M. 554 edulis. However, in oysters C. gigas exposed to fluorescent microplastic beads (of 2 and 555 6 μm), microplastics were only present in the digestive gland (Sussarellu et al., 2016),

existing no evidence of transfer from the digestive track to the circulatory system. 556 Although there is some controversy about the translocation of microplastic in the 557 558 haemolymph of bivalve molluscs, the data reported is only on filter-feeders (Browne et 559 al., 2008; von Moos et al., 2012). It might occur that in the case of suspension-feeders 560 like S. plana, microplastics tend to translocate in the circulatory system but this needs to be confirmed in future studies. In S. plana exposed to gold nanoparticles (Au NPs), 561 nanoparticles were accumulated almost exclusively in the digestive gland, although they 562 were also present in gills. These results highlight the accumulation of both nano and 563 564 microparticles essentially in the digestive tract of this clam species. The accumulation of microplastics in this tissue might impair the digestive system with a consequent decrease 565 of feeding behaviour. There is very limited information regarding the capacity of aquatic 566 567 organisms to eliminate microplastics. The present results indicated that after a week of 568 depuration, microplastics were still present in both tissues (Figure 4). Future studies that quantify the amount of microplastics accumulated in clam tissues and assess whether 569 570 microplastics are eliminated in the pseudo-faeces of S. plana are necessary, in order to increase the knowledge about the accumulation, metabolism and elimination of PS 571 572 microplastics in this species.

573 A battery of biomarkers was used to assess the biological effects and toxicity of 574 PS microplastics in the gills and digestive gland of S. plana. Results indicate an oxidative stress response in gills and digestive gland cells that are tissue and exposure time 575 576 dependent (Figures 5A-E). SOD is the first defence line to protect these tissues against oxidative stress probably caused by injuries of PS microplastics in the tissues. SOD 577 578 activity enhanced in both tissues in the presence of PS microplastics reflects the need to balance the excess of superoxide radical $(O_2^{\bullet-})$ into the less damaging hydrogen peroxide 579 580 (H₂O₂) and thus, contribute to prevent cellular oxidative damage (Jo et al., 2008). The same response occurred in marine mussels Mytilus spp. exposed to 32 µg L⁻¹ of PS 581 582 microplastics (2 and 6 µm) after 14 days (Paul-Pont et al., 2016). CAT is involved in the removal of H₂O₂ - the main precursor of hydroxyl radical in aquatic organisms - and acts 583 584 as a defence mechanism toward the exogenous source of H₂O₂ (Regoli & Giuliani, 2014). However, CAT activity was only enhanced in the gills after three days of exposure and 585 inhibited in the digestive gland after 7 days of exposure (Figure 5B, F) and is apparently 586 not the antioxidant defence mechanism used by S. plana to respond to PS microplastics 587 toxicity. Avio et al. (2015) also noted an inhibition of CAT activity in the digestive tissue 588 589 of the marine mussel M. galloprovincialis exposed to microplastics. CAT and GPx are

both involved in the removal of H₂O₂. Regarding GPx, an increase in activity after 3 days 590 of exposure, suggests a defence mechanism, but the posterior reduction in GPx activity, 591 592 particularly in the digestive gland (Figure 5G), may be an indication of an inhibition 593 triggered by excess of ROS and the incapacity to deal with the negative effect of this 594 stressor. Glutathione peroxidases are known to be particularly sensitive in revealing the 595 early onset of a pro-oxidant challenge, even at low levels of environmental disturbance (Regoli & Giuliani, 2014). In S. plana exposed to mercury, a decrease in GPx activity in 596 the whole soft tissues occurred due to the higher toxicity of mercury (Ahmad et al., 2011). 597 598 Such ROS perturbations were also observed in mussels (M. galloprovincialis) exposed to PS and PE microplastics alone or in combination with pyrene (Avio et al., 2015). The 599 600 pro-oxidant challenge induced by microplastics on mussels was supported by the lack of 601 significant variation of malondialdehyde, lipofuscin and neutral lipids in digestive tissues. 602 The ROS production led to an inhibition of CAT and Se-GPx in M. galloprovincialis 603 exposed to PS and PE microplastics (Avio et al., 2015). Browne et al. (2013) showed that 604 PVC microparticles induced an oxyradical production in Arenicola marina, and 605 lugworms that ingested sediments with PVC reduce the capacity by more than 30% to 606 deal with oxidative stress. PS microbeads also led to an increase in ROS production in 607 haemocytes and to the enhancement of anti-oxidant and glutathione-related enzymes in 608 mussel tissues (Paul-Pont et al., 2016).

GST is usually associated with phase II biotransformation, involved in the 609 610 metabolism of lipophilic organic compounds by catalysing the conjugation of the reduced form of glutathione (GSH) to xenobiotic substrates, and has also a protective role against 611 612 oxidative stress (Lesser, 2006). S. plana may be using this detoxification mechanism to deal with the exposure to PS microplastics, by catalysing the conjugation of the reduced 613 614 form of glutathione (GSH) to xenobiotic substrates, playing a significant role in the 615 detoxification of the reactive products from lipid peroxidation (Lesser, 2006). Gills showed an increase in GST activity after 14 days of exposure (Figure 5D). Similar results 616 617 were found in gills of *M. galloprovincialis* after exposure to organic persistent pollutants, 618 such as pp'DDE (2,2-bis-(p-chlorophenyl)-1,1-dichlorethylene) (Hoarau et al., 2002; Khessiba et al., 2001). In the digestive gland, GST activity decreased (Figure 5H), as was 619 previously observed in mussels after PS-exposure for 14 days (Avio et al., 2015). 620

In the depuration period, SOD activity continued to increase in both organs (Figure 5A, E). The increase in enzymatic activity at the end of the depuration period can be due to an impairment of the filtration activity or to the presence of remaining

microparticles in the gills and digestive gland, that still induce an enzymatic response 624 (Paul-Pont et al., 2016), suggesting the inability of S. plana to eliminate microplastics, 625 626 by the mechanical damage caused by the release of the particles from the tissues, or even 627 the lack of capacity to recover after this elimination period. Nevertheless, 7 days of 628 depuration might not be enough for S. plana to completely recover from the exposure of 629 PS microplastics as suggested by the present results (Figure 5). In conclusion, it can be hypothesized that ROS are produced as a result of PS microplastics uptake, and possibly 630 cause injury of gills and/or internalization in the digestive gland cells, since it is known 631 632 that these microparticles are able to cross cell membranes, leading to cell damage (Browne et al., 2008; Rosenkranz et al., 2009; Van Cauwenberghe et al., 2015). 633

634 The levels of oxidative damage to lipids decreased in the gills in the PS exposed clams (Figure 8A). This could be linked to antioxidant defences, which consequently limit 635 636 the attack of ROS to membrane lipids. In the digestive gland, there is a tendency of LPO levels to increase (from day 0 to day 7), but no significant differences were noted (Figure 637 638 8B). The enzymatic activities measured in this tissue were always lower on day 7, with a recovery after one week of depuration, with the exception of GPx. A significant 639 640 enhancement of ROS in digestive gland haemocytes of Mytilus spp. after 7 days of micro-641 PS exposure (mixture of 2 and 6 µm) was previously reported but no antioxidant markers 642 were activated and no sign of lipid peroxidation was observed (Paul-Pont et al., 2016). Ahmad et al. (2011) showed an increase in LPO levels in both gills and digestive gland 643 644 of S. plana exposed to mercury. When enzymatic defences do not actively respond to the presence of PS microplastics this may result in an inflammation response and a lysosomal 645 646 membrane destabilization, as a cellular response observed in mussels (M. edulis) exposed to 2.5 g L⁻¹ of a high-density polyethylene (HDPE) particles (> $0-80 \mu m$) for 96 hours 647 (von Moos et al., 2012). Future exposure experiments with S. plana should look at this 648 649 aspect.

The Comet assay is a sensitive, rapid and economic technique for the detection of 650 DNA strand breaks and can be regarded as a good method to assess genotoxicity in 651 652 aquatic species (Jha, 2008). Moreover, Petridis et al. (2009) demonstrated that the blood cells of S. plana are suitable for screening genotoxic effects, using this method. DNA 653 654 strand breaks induced in PS exposed haemocytes of S. plana (Figure 6 A,B) are similar to mussels treated with PE microplastics where a significant enhancement of DNA strand 655 656 breaks was detected (Avio et al., 2015). Regarding the elimination period, there was an increase of both Tail DNA (%) and OTM, in PS exposed organisms. The mechanism of 657

genotoxicity of PS microplastics remains unknown, but it is suggested that it can be
related to ROS production and oxidative stress, not handled by the antioxidant defence
mechanism, as occurs with nanoparticles (Rocha *et al.*, 2014). Hence, causing damage by
covalently binding to DNA (Hossain & Huq, 2002) or by inhibiting DNA synthesis
(Hidalgo & Dominguez, 1998) and thus preventing cell division and DNA replication
(Singh *et al.*, 1988).

PS microplastics inhibit the AChE activity in clam gills at different times of exposure with this effect remaining even after 7 days of depuration (Figure 7). The ability of microplastics to inhibit AChE activity was previously described in juveniles of the common goby *Pomatoschistus microps* exposed to 18.4 and 184 μ g L⁻¹ of PE microspheres (1-5 μ m) for 96 hours, alone or in combination with pyrene (Oliveira *et al.*, 2013) or in combination with chromium (Luís *et al.*, 2015).

670 PCA for the gills and digestive gland show different responses that reflect the distinct physiological and metabolic functions of the two tissues (Figure 9A-B). Exposure 671 672 to PS microplastics in the gills was positively related to an increase in the activity of 673 oxidative stress enzymes and DNA damage, although negatively related to the oxidative 674 damage and to the activity of AChE (inhibition), meaning that there is an inflammation process and a neurotoxic effect and validating the hypothesis that gills had a more 675 676 effective response against oxidative stress than digestive gland (Figure 9A). In the case of the digestive gland, SOD is the biomarker that best relates to exposure to PS 677 microplastics. As gills are the main tissue involved in filtration, they are in direct contact 678 with the PS microplastics, being more susceptible to oxidative stress than the digestive 679 680 gland (Figure 9B). Despite existing data about the increasing occurrence of PS microplastics in the marine environment (Andrady, 2011; Cole et al., 2011; Wright et al., 681 682 2013b), there is still much to understand about their biological effects. Further studies are 683 needed to investigate the capacity to recover from the potential effects of PS and other 684 microplastics in marine organisms beyond 7 days of depuration. The evaluation of the biological effects of microplastics also requires a molecular-level to understand how they 685 interact with cells in a physiological environment, but up to date the functional 686 implications at cellular level still remains to be elucidated. 687

- 688
- 689

690 Conclusions

PS microplastics were taken up by clams, mainly by the gills where they tend to accumulate, but they were also present in the digestive gland where they seemed to be stored. Tissue-specific sensibility is involved in the clams response to PS exposure by inducing oxidative stress, with the gills providing a more effective response than digestive gland. The genotoxicity of PS microplastics increased with time. Furthermore, the detoxification process of PS microplastics in clams tissues was inefficient for the 7 days duration tested, indicating their potential trophic transfer.

699

700

701 Acknowledgements

This work was developed under the project EPHEMARE, JPIOCEANS programme of the EU, funded by the Portuguese Foundation for Science and Technology (JPIOCEANS/0005/2015). This work was also supported by the Portuguese Science Foundation (FCT) through the grant UID/MAR/00350/2013 attributed to CIMA of the University of Algarve. The authors would like to thank V. Sousa and M. R. Teixeira for their help in the characterization of PS microplastics and also T.L. Rocha and C. Cardoso for their helpful discussions and laboratory contributions.

709

710 **References**

- Ahmad, I., Mohmood, I., Mieiro, C.L., Coelho, J.P., Pacheco, M., Santos, M.A., Duarte, A.C., &
 Pereira, E. (2011). Lipid peroxidation vs. antioxidant modulation in the bivalve *Scrobicularia plana* in response to environmental mercury—organ specificities and age
 effect. Aquatic Toxicology, 103(3), 150-158.
- Almeida, C., Pereira, C., Gomes, T., Bebianno, M.J., & Cravo, A. (2011). DNA damage as a biomarker of genotoxic contamination in *Mytilus galloprovincialis* from the south coast of Portugal. *Journal of Environmental Monitoring*, 13(9), 2559-2567.
- Andrady, A.L. (2011). Microplastics in the marine environment. *Marine Pollution Bulletin*, 62(8),
 1596-1605.
- Avio, C.G., Gorbi, S., Milan, M., Benedetti, M., Fattorini, D., d'Errico, G., Pauletto, M.,
 Bargelloni, L., & Regoli, F. (2015). Pollutants bioavailability and toxicological risk from
 microplastics to marine mussels. *Environmental Pollution*, **198**, 211-222.
- Bowmer, T., & Kershaw, P. (2010). Proceedings of the GESAMP International Workshop on
 microplastic particles as a vector in transporting persistent, bioaccumulating and toxic
 substances in the oceans. 28–30th June 2010, UNESCO-IOC, Paris. *GESAMP Reports and Studies*, 82.
- Brillant, M., & MacDonald, B. (2002). Postingestive selection in the sea scallop (*Placopecten magellanicus*) on the basis of chemical properties of particles. *Marine Biology*, 141(3), 457-465.

- Browne, M.A., Dissanayake, A., Galloway, T.S., Lowe, D.M., & Thompson, R.C. (2008).
 Ingested microscopic plastic translocates to the circulatory system of the mussel, *Mytilus edulis (L.). Environmental Science & Technology*, 42(13), 5026-5031.
- Browne, M.A., Niven, S.J., Galloway, T.S., Rowland, S.J., & Thompson, R.C. (2013).
 Microplastic moves pollutants and additives to worms, reducing functions linked to health and biodiversity. *Current Biology*, 23(23), 2388-2392.
- Canesi, L., Ciacci, C., Bergami, E., Monopoli, M., Dawson, K., Papa, S., Canonico, B., & Corsi,
 I. (2015). Evidence for immunomodulation and apoptotic processes induced by cationic
 polystyrene nanoparticles in the hemocytes of the marine bivalve *Mytilus*. *Marine Environmental Research*, 111, 34-40.
- Carpenter, E.J., & Smith, K. (1972). Plastics on the Sargasso Sea surface. *Science*, 175(4027),
 1240-1241.
- Cawthorn, M. (1989). Impacts of marine debris on wildlife in New Zealand coastal waters.
 Proceedings of Marine Debris in New Zealand's Coastal Waters Workshop, 9, 5-6.
- Claessens, M., Van Cauwenberghe, L., Vandegehuchte, M.B., & Janssen, C.R. (2013). New techniques for the detection of microplastics in sediments and field collected organisms.
 Marine Pollution Bulletin, **70**(1), 227-233.
- Codina-García, M., Militão, T., Moreno, J., & González-Solís, J. (2013). Plastic debris in
 Mediterranean seabirds. *Marine Pollution Bulletin*, 77(1), 220-226.
- Cole, M., Lindeque, P., Halsband, C., & Galloway, T.S. (2011). Microplastics as contaminants in
 the marine environment: a review. *Marine pollution bulletin*, 62(12), 2588-2597.
- Colovic, M.B., Krstic, D.Z., Lazarevic-Pasti, T.D., Bondzic, A.M., & Vasic, V.M. (2013).
 Acetylcholinesterase inhibitors: pharmacology and toxicology. *Current Neuropharmacology*, 11(3), 315.
- De Witte, B., Devriese, L., Bekaert, K., Hoffman, S., Vandermeersch, G., Cooreman, K., &
 Robbens, J. (2014). Quality assessment of the blue mussel (*Mytilus edulis*): comparison
 between commercial and wild types. *Marine Pollution Bulletin*, **85**(1), 146-155.
- Derraik, J.G. (2002). The pollution of the marine environment by plastic debris: a review. *Marine Pollution Bulletin*, 44(9), 842-852.
- Devriese, L.I., van der Meulen, M.D., Maes, T., Bekaert, K., Paul-Pont, I., Frère, L., Robbens, J.,
 & Vethaak, A.D. (2015). Microplastic contamination in brown shrimp (*Crangon crangon*, Linnaeus 1758) from coastal waters of the Southern North Sea and Channel area. *Marine Pollution Bulletin*, 98(1), 179-187.
- Eerkes-Medrano, D., Thompson, R.C., & Aldridge, D.C. (2015). Microplastics in freshwater
 systems: a review of the emerging threats, identification of knowledge gaps and
 prioritisation of research needs. *Water Research*, **75**, 63-82.
- Fellman, G.L., Courtney, K.D., Andres, V., & Featherstone, R.M. (1961). A new and rapid colorimetric determination of acetylcholinesterase activity. *Biochemical Pharmacology*, 7(2), 88-95.
- Frdelmeier, I., Gérard-Monnier, D., Yadan, J.-C., & Chaudiere, J. (1998). Reactions of N-methyl2-phenylindole with malondialdehyde and 4-hydroxyalkenals. Mechanistic aspects of the
 colorimetric assay of lipid peroxidation. *Chemical Research in Toxicology*, 11(10), 11841194.
- Farrell, P., & Nelson, K. (2013). Trophic level transfer of microplastic: *Mytilus edulis* (L.) to
 Carcinus maenas (L.). *Environmental Pollution*, 177, 1-3.
- Fendall, L.S., & Sewell, M.A. (2009). Contributing to marine pollution by washing your face:
 Microplastics in facial cleansers. *Marine Pollution Bulletin*, 58(8), 1225-1228.
- Frias, J.P.G.L.d. (2015). Effects of the presence of microplastic particles in portuguese coastal
 waters and marine mussels. (Doctor of Phylosophy in Environmental Sciences),
 Universidade Nova de Lisboa, Monte de Caparica.
- Gall, S., & Thompson, R. (2015). The impact of debris on marine life. *Marine Pollution Bulletin*,
 92(1), 170-179.
- Goldstein, M.C., & Goodwin, D.S. (2013). Gooseneck barnacles (*Lepas spp.*) ingest microplastic
 debris in the North Pacific Subtropical Gyre. *PeerJ*, 1, e184
 https://doi.org/110.7717/peerj.7184.

- Gomes, T., Araújo, O., Pereira, R., Almeida, A.C., Cravo, A., & Bebianno, M.J. (2013).
 Genotoxicity of copper oxide and silver nanoparticles in the mussel *Mytilus* galloprovincialis. Marine Environmental Research, 84, 51-59.
- Graham, E.R., & Thompson, J.T. (2009). Deposit-and suspension-feeding sea cucumbers
 (Echinodermata) ingest plastic fragments. *Journal of Experimental Marine Biology and Ecology*, 368(1), 22-29.
- Green, D.S., Boots, B., Sigwart, J., Jiang, S., & Rocha, C. (2016). Effects of conventional and
 biodegradable microplastics on a marine ecosystem engineer (Arenicola marina) and
 sediment nutrient cycling. *Environmental Pollution*, 208, 426-434.
- Greenwald, R.A. (1987). Handbook of methods for oxygen radical research. *Free Radical Biology and Medicine*, 3(2), 161.
- Habig, W.H., Pabst, M.J., & Jakoby, W.B. (1974). Glutathione S-transferases the first enzymatic
 step in mercapturic acid formation. *Journal of Biological Chemistry*, 249(22), 7130-7139.
- Hidalgo, E., & Dominguez, C. (1998). Study of cytotoxicity mechanisms of silver nitrate in human dermal fibroblasts. *Toxicology Letters*, 98(3), 169-179.
- Hoarau, P., Garello, G., Gnassia-Barelli, M., Romeo, M., & Girard, J.P. (2002). Purification and
 partial characterization of seven glutathione S-transferase isoforms from the clam
 Ruditapes decussatus. European Journal of Biochemistry, 269(17), 4359-4366.
- Holland-Moritz, K., & Siesler, H. (1976). Infrared spectroscopy of polymers. *Applied Spectroscopy Reviews*, 11(1), 1-55.
- Hossain, Z., & Huq, F. (2002). Studies on the interaction between Ag+ and DNA. Journal of Inorganic Biochemistry, 91(2), 398-404.
- Hughes, R.N. (1969). A study of feeding in *Scrobicularia plana*. Journal of the Marine Biological
 Association of the United Kingdom, 49(03), 805-823.
- Imhof, H.K., Ivleva, N.P., Schmid, J., Niessner, R., & Laforsch, C. (2013). Contamination of
 beach sediments of a subalpine lake with microplastic particles. *Current Biology*, 23(19),
 R867-R868.
- Jha, A.N. (2008). Ecotoxicological applications and significance of the comet assay. *Mutagenesis*,
 23(3), 207-221.
- Jo, P.G., Choi, Y.K., & Choi, C.Y. (2008). Cloning and mRNA expression of antioxidant enzymes
 in the Pacific oyster, *Crassostrea gigas* in response to cadmium exposure. *Comparative Biochemistry and Physiology Part C: Toxicology & Pharmacology*, 147(4), 460-469.
- Keller, A.A., Wang, H., Zhou, D., Lenihan, H.S., Cherr, G., Cardinale, B.J., Miller, R., & Ji, Z.
 (2010). Stability and aggregation of metal oxide nanoparticles in natural aqueous matrices. *Environmental Science & Technology*, 44(6), 1962-1967.
- Kershaw, P. (2015). Sources, fate and effects of microplastics in the marine environment: a global
 assessment. *Reports and Studies GESAMP*, **90**, 96.
- Khessiba, A., Hoarau, P., Gnassia-Barelli, M., Aissa, P., & Roméo, M. (2001). Biochemical
 response of the mussel *Mytilus galloprovincialis* from Bizerta (Tunisia) to chemical
 pollutant exposure. *Archives of Environmental Contamination and Toxicology*, 40(2),
 222-229.
- 827 Koenig, J.L. (1999). Spectroscopy of polymers: Elsevier.
- Lambert, C., Sussarell, R., Fabioux, C., Goic, N.L., Mingant, C., Huvet, A., Maes, T., Vethaak,
 D., Robbens, J., & Soudant, P. (2014). Translocation of microplastic in the circulatory
 system of the oyster *Crassostrea gigas* : Evidences and uncertainties : investigation to
 load., 49-50.
- Lawrence, R.A., & Burk, R.F. (1978). Species, tissue and subcellular distribution of non Se dependent glutathione peroxidase activity. *The Journal of Nutrition*, 108(2), 211-215.
- Lee, K.-W., Shim, W.J., Kwon, O.Y., & Kang, J.-H. (2013). Size-dependent effects of micro
 polystyrene particles in the marine copepod *Tigriopus japonicus*. *Environmental Science & Technology*, 47(19), 11278-11283.
- Leslie, H., Van Velzen, M., & Vethaak, A. (2013). Microplastic survey of the Dutch environment.
 Novel data set of microplastics in North Sea sediments, treated wastewater effluents and marine biota. Amsterdam: Institute for Environmental Studies, VU University Amsterdam.

- Lesser, M.P. (2006). Oxidative stress in marine environments: biochemistry and physiological
 ecology. *Annual Review of Physiology*, 68, 253-278.
- Liang, C., & Krimm, S. (1958). Infrared spectra of high polymers. VI. Polystyrene. *Journal of Polymer Science*, 27(115), 241-254.
- Liebezeit, G., & Dubaish, F. (2012). Microplastics in beaches of the East Frisian islands
 Spiekeroog and Kachelotplate. *Bulletin of environmental contamination and toxicology*,
 847 89(1), 213-217.
- Luís, L.G., Ferreira, P., Fonte, E., Oliveira, M., & Guilhermino, L. (2015). Does the presence of
 microplastics influence the acute toxicity of chromium (VI) to early juveniles of the
 common goby (*Pomatoschistus microps*)? A study with juveniles from two wild estuarine
 populations. *Aquatic Toxicology*, 164, 163-174.
- Lusher, A. (2015). Microplastics in the marine environment: distribution, interactions and effects
 Marine anthropogenic litter (pp. 245-307): Springer.
- Matthäus, C., Bird, B., Miljković, M., Chernenko, T., Romeo, M., & Diem, M. (2008). Infrared
 and Raman microscopy in cell biology. *Methods in cell biology*, 89, 275-308.
- Mazurais, D., Ernande, B., Quazuguel, P., Severe, A., Huelvan, C., Madec, L., Mouchel, O.,
 Soudant, P., Robbens, J., & Huvet, A. (2015). Evaluation of the impact of polyethylene
 microbeads ingestion in European sea bass (*Dicentrarchus labrax*) larvae. *Marine Environmental Research*, 112, 78-85.
- McCord, J.M., & Fridovich, I. (1969). Superoxide dismutase an enzymic function for
 erythrocuprein (hemocuprein). *Journal of Biological Chemistry*, 244(22), 6049-6055.
- Mitchell, M.B. (1993). Fundamentals and Applications of Diffuse Reflectance Infrared Fourier
 Transform (DRIFT) Spectroscopy, Structure Property Relations in Polymers (pp. 351 375): Ed Marek W. Urban, Clara D. Craver, ACS.
- Moore, C.J. (2008). Synthetic polymers in the marine environment: a rapidly increasing, long-term threat. *Environmental Research*, 108(2), 131-139.
- Murray, F., & Cowie, P.R. (2011). Plastic contamination in the decapod crustacean *Nephrops norvegicus* (Linnaeus, 1758). *Marine Pollution Bulletin*, 62(6), 1207-1217.
- NOAA. (2015). National Oceanographic Administration Service. Retrieved May, 2016, from
 http://www.noaa.gov
- 871 Norén, F. (2007). Small plastic particles in coastal Swedish waters. *KIMO Sweden*.
- Oliveira, M., Ribeiro, A., Hylland, K., & Guilhermino, L. (2013). Single and combined effects of
 microplastics and pyrene on juveniles (0+ group) of the common goby *Pomatoschistus microps* (Teleostei, Gobiidae). *Ecological Indicators*, 34, 641-647.
- Paul-Pont, I., Lacroix, C., Fernández, C.G., Hégaret, H., Lambert, C., Le Goïc, N., Frère, L.,
 Cassone, A.-L., Sussarellu, R., & Fabioux, C. (2016). Exposure of marine mussels
 Mytilus spp. to polystyrene microplastics: Toxicity and influence on fluoranthene
 bioaccumulation. *Environmental Pollution*, 216, 724-737.
- Petridis, P., Jha, A.N., & Langston, W.J. (2009). Measurements of the genotoxic potential of
 (xeno-) oestrogens in the bivalve molluse *Scrobicularia plana*, using the comet assay. *Aquatic Toxicology*, 94(1), 8-15.
- Phuong, N.N., Zalouk-Vergnoux, A., Poirier, L., Kamari, A., Châtel, A., Mouneyrac, C., &
 Lagarde, F. (2016). Is there any consistency between the microplastics found in the field
 and those used in laboratory experiments? *Environmental Pollution*, 211, 111-123.
- Pinto, J.C. (2012). Impactos ambientais causados pelos plásticos: uma discussão abrangente sobre os mitos e os dados científicos (2ª ed.): Editora E-papers.
- PlasticsEurope. (2015). Plastics-the Facts 2014/ 2015: an analysis of European plastics
 production, demand and waste data. Retrieved May, 2016, from
 <u>http://www.plasticseurope.org/documents/document/20150227150049-</u>
 final plastics the facts 2014 2015 260215.pdf
- Pruter, A. (1987). Sources, quantities and distribution of persistent plastics in the marine
 environment. *Marine Pollution Bulletin*, 18(6), 305-310.
- Regoli, F., & Giuliani, M.E. (2014). Oxidative pathways of chemical toxicity and oxidative stress
 biomarkers in marine organisms. *Marine Environmental Research*, 93, 106-117.

- Rocha, T.L., Gomes, T., Cardoso, C., Letendre, J., Pinheiro, J.P., Sousa, V.S., Teixeira, M.R., &
 Bebianno, M.J. (2014). Immunocytotoxicity, cytogenotoxicity and genotoxicity of
 cadmium-based quantum dots in the marine mussel *Mytilus galloprovincialis*. *Marine Environmental Research*, 101, 29-37.
- Rosenkranz, P., Chaudhry, Q., Stone, V., & Fernandes, T.F. (2009). A comparison of nanoparticle
 and fine particle uptake by *Daphnia magna*. *Environmental Toxicology and Chemistry*,
 28(10), 2142-2149.
- Singh, N.P., McCoy, M.T., Tice, R.R., & Schneider, E.L. (1988). A simple technique for
 quantitation of low levels of DNA damage in individual cells. *Experimental Cell Research*, 175(1), 184-191.
- Song, Y.K., Hong, S.H., Jang, M., Kang, J.-H., Kwon, O.Y., Han, G.M., & Shim, W.J. (2014).
 Large accumulation of micro-sized synthetic polymer particles in the sea surface microlayer. *Environmental science & technology*, 48(16), 9014-9021.
- Sousa, V.S., & Teixeira, M.R. (2013). Aggregation kinetics and surface charge of CuO
 nanoparticles: the influence of pH, ionic strength and humic acids. *Environmental Chemistry*, 10(4), 313-322.
- Stuart, B.H. (2005). Infrared Spectroscopy: Fundamentals and Applications (pp. 36): John Wiley
 & Sons, Ltd.
- 913 Sussarellu, R., Suquet, M., Thomas, Y., Lambert, C., Fabioux, C., Pernet, M.E.J., Le Goïc, N.,
 914 Quillien, V., Mingant, C., & Epelboin, Y. (2016). Oyster reproduction is affected by
 915 exposure to polystyrene microplastics. *Proceedings of the National Academy of Sciences*,
 916 113(9), 2430-2435.
- Thompson, R.C., Olsen, Y., Mitchell, R.P., Davis, A., Rowland, S.J., John, A.W., McGonigle,
 D., & Russell, A.E. (2004). Lost at sea: where is all the plastic? *Science*, **304**(5672), 838838.
- 920 Van Cauwenberghe, L., Claessens, M., Vandegehuchte, M.B., & Janssen, C.R. (2015).
 921 Microplastics are taken up by mussels (*Mytilus edulis*) and lugworms (*Arenicola marina*)
 922 living in natural habitats. *Environmental Pollution*, **199**, 10-17.
- Van Cauwenberghe, L., & Janssen, C.R. (2014). Microplastics in bivalves cultured for human
 consumption. *Environmental Pollution*, 193, 65-70.
- von Moos, N., Burkhardt-Holm, P., & Köhler, A. (2012). Uptake and effects of microplastics on
 cells and tissue of the blue mussel *Mytilus edulis L*. after an experimental exposure.
 Environmental Science & Technology, 46(20), 11327-11335.
- Ward, J., & Targett, N. (1989). Influence of marine microalgal metabolites on the feeding
 behavior of the blue mussel *Mytilus edulis*. *Marine Biology*, **101**(3), 313-321.
- Ward, J.E., & Kach, D.J. (2009). Marine aggregates facilitate ingestion of nanoparticles by
 suspension-feeding bivalves. *Marine Environmental Research*, 68(3), 137-142.
- Ward, J.E., Levinton, J.S., & Shumway, S.E. (2003). Influence of diet on pre-ingestive particle
 processing in bivalves: I: transport velocities on the ctenidium. *Journal of Experimental Marine Biology and Ecology*, 293(2), 129-149.
- Wilber, R.J. (1987). Plastic in the North Atlantic. *Oceanus*, **30**(3), 61-68.
- Williams, A., & Simmons, S. (1997). Estuarine litter at the river/beach interface in the Bristol
 Channel, United Kingdom. *Journal of Coastal Research*, 13(4), 1159-1165.
- Wright, S.L., Rowe, D., Thompson, R.C., & Galloway, T.S. (2013a). Microplastic ingestion
 decreases energy reserves in marine worms. *Current Biology*, 23(23), R1031-R1033.
- Wright, S.L., Thompson, R.C., & Galloway, T.S. (2013b). The physical impacts of microplastics
 on marine organisms: a review. *Environmental Pollution*, **178**, 483-492.
- 242 Zarfl, C., Fleet, D., Fries, E., Galgani, F., Gerdts, G., Hanke, G., & Matthies, M. (2011).
 243 Microplastics in oceans. *Marine Pollution Bulletin*, 62, 1589-1591.
- 944
- 945





Figure 3. Average DRIFT spectra of: unexposed digestive gland (DG T0) and gills (Gills T0) and digestive gland with
 added polystyrene (DG Added PS). The spectra are normalized to the band at 1543 cm⁻¹. The spectrum of polystyrene
 (PS) microparticles is included for comparison.



Figure 4. Comparison of the average DRIFT spectra in the 1800-450 cm⁻¹ region for gills (A) and digestive gland (C) of *S. plana*, taken at the beginning (T0) and end of the exposure period (T14) and after the depuration (T21); Spectral subtraction for gills (B) and digestive gland (D) as indicated compared with the PS microparticles spectrum



Figure 5. SOD, CAT, GPx and GST activities (mean \pm SD) in the gills (A, B, C and D) and digestive gland (E, F, G and H) of *S. plana* for control (CTR) and PS microplastics (MICR 1 mg L⁻¹) during exposure and depuration. Statistically significant differences between treatments at each time of exposure are indicated with #; different lower case letters indicate significant differences between controls over time and capital letters between microplastic exposed (MICR) treatment over time (p <0.05).



Figure 6. DNA damage (average \pm SEM) in the haemocytes of *S. plana* expressed as tail DNA % (A) and OTM (a.u.) (B) for control (CTR) and PS microplastics (MICR). Statistically significant differences between treatments at each time of exposure are indicated with #; different lower case letters indicate significant differences between controls over time and capital letters between microplastic exposed (MICR) treatment over time (p <0.05).





Figure 8. LPO (mean \pm SD) in gills (A) and digestive gland (B) of *S. plana* for control (CTR) and microplastics (MICR). Statistically significant differences between treatments at each time of exposure are indicated with #; different lower case letters indicate significant differences between controls over time and capital letters between microplastic exposed (MICR) treatment over time (p <0.05).



Figure 9. Principal component analysis (PCA) of a battery of biomarkers in the gills (A) and digestive gland (B) of *S. plana* unexposed \square) and exposed to PS microplastics \square)

Table 1. Evidence of microplastics ingestion in marine organisms

Species	Microplastics (%)	Mean (± SD) number of particles/ individual	Type and size (μm)	Reference
Phylum Arthropoda				
Gammarus pulex	_	_	Acrilic 29.5 ± 26	Imhof <i>et al.</i> (2013)
Notodromas monacha	_	-	Acrilic 29.5 ± 26	Imhof <i>et al.</i> (2013)
Phyllum Annelida				
Lumbriculus variegatus	-	-	Acrilic 29.5 ± 26	Imhof <i>et al.</i> (2013)
Phylum Mollusca				
Mytilus edulis	_	3.7 particles per 10 g tissue	Fibres 300-1000	De Witte <i>et al.</i> (2014)
Mytilus edulis	_	$0.36 (\pm 0.07)$ particles g ⁻¹	5 - 25	Van Cauwenberghe and Janssen (2014)
Cassostrea gigas	_	$0.47 (\pm 0.16)$ particles g ⁻¹	5 - 25	Van Cauwenberghe and Janssen (2014)
Phylum Crustacea				
Lepas spp.	33.5	1-30 particles/ individual	1.41	Goldstein and Goodwin (2013)
Nephrops norvegicus	83	_	_	Murray and Cowie (2011)
Crangon crangon	_	11.5 fibres per 10 g shrimp	300-1000	Devriese et al. (2015)
Phylum Chaetognatha				
Parasagitta elegans	100	_	0.1-3 PS	Carpenter and Smith (1972)
PS (Polystyrene)				
111				
112				
113				
114				
115				
116				

Table 2. Effects of microplastics to aquatic organisms

Species	Microplastics		Exposure		Effects	Reference
	Туре	Size (µm)	Concentration	Duration		
Phylum Mollusca						
Class Bivalvia						
Mytilus galloprovincialis	PE,PS, PE- PYR e PS_PYR	<100	1.5 g L ⁻¹	7 d	Adsorption of pyrene not differ between PS and PE; bioaccumulation in digestive tissues and gills	Avio <i>et al.</i> (2015)
Mytilus edulis	PS fluorescently labeled	2	0.51 μg L ⁻¹	12 h	Uptake accumulation in gut; and	Browne <i>et al.</i> (2008)
		4-16	_	3 d	hemolymph after 3 d	
		3 and 9.6				
	PS	10	50 particles ml ⁻¹	14 d	Greater accumulation of smaller particles; no	Van
		30	50 particles ml ⁻¹	-	significant effects on metabolism	Cauwenberghe et al. (2015)
		90	10 particles ml ⁻¹	-		
			110 particles ml ⁻¹ (Total concentration)			
	HDPE	0-80	2.5 g L ⁻¹	3, 6, 12, 24, 48 and 96 h	Uptake; retention in gut and transfer into the lymphatic system; immune response	von Moos <i>et al.</i> (2012)
	PS Microspheres fluorescently labeled	0.5	2 x 10 ⁷ particles ml ⁻¹	1 h	Uptake; trophic transfer to <i>Carcinus</i> <i>maenas</i>	Farrell and Nelson (2013)
	PS	10	2×10^4 particles ml ⁻¹	30 min	Intake	Ward and Targett (1989)
	PS	10, 30	3.10×10^5 particles ml ⁻¹	_	Intake	Claessens <i>et al.</i> (2013)
Mytilus trossulus	PS	10	1000 particles ml ⁻¹	_	Intake	Ward <i>et al.</i> (2003)

Crassostrea virginica	PS	10	1000 particles ml ⁻¹	45 min	Intake and egestion	Ward and Kach (2009)
Crassostrea gigas	PS	2	2.06 ± 170	2 months	Decreases in oocyte number, diameter,	Sussarellu <i>et al.</i> (2016)
		6	118 ± 15		and sperm velocity; decrease of larval	
			particles ml ⁻¹		development; endocrine disruption	
Placopecten	PS	15, 10,	$5 \ge 10^3$ particles	1 h	Intake, retention	Brillant and
magellanicus		16, 18, 20	m^{-1}		and egestion	MacDonald (2002)

PS (Polystyrene) PE (Polyethylene) PP (Polypropylene) LDPE (Low Density Polyethylene) HDPE (High-density Polyethylene) UPVC (Polyvinyl chloride) PE-PYR (Pyrene Treated Polyethylene) PS_PYR (Pyrene Treated Polystyrene)

1120

1121

.

Table 3. Characterization of PS microplastics using different techniques 1122

Particle characterization	Method	PS microplastics
Particle size $(\mu m)^{a}$	OM	20 ± 0.02
Density (g cm ⁻³) ^a	_	1.05
Mean particle diameter (μm) ^c	DLS	18.4 ± 1.33
Zeta (ζ) potential (mV) ^{b c}	ELS	Sea water: -12.4 ± 2.36 Mili-Q water: -52.6 ± 2.34

Original solution of PS microplastics from Sigma Aldrich (4.3 x 10⁶ particles ml⁻¹) a.

100 mg L^{-1} of microplastics dispersed in natural seawater 100 mg L^{-1} of microplastics dispersed in ultrapure water b.

c.

OM. Optical microscope

DLS. Dynamic light scattering

ELS. Electrophoretic light scattering

- 1124
- 1125
- 1126
- 1127
- 1128
- 1129