





### On the biosynthetic origin of carminic acid

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Published in: Insect Biochemistry and Molecular Biology

*Link to article, DOI:* 10.1016/j.ibmb.2018.03.002

Publication date: 2018

Document Version Peer reviewed version

Link back to DTU Orbit

Citation (APA):

Rasmussen, S. A., Kongstad, K. T., Khorsand-Jamal, P., Kannangara, R. M., Nafisi, M., Van Dam, A., ... Frandsen, R. J. N. (2018). On the biosynthetic origin of carminic acid. Insect Biochemistry and Molecular Biology, 96, 51-61. DOI: 10.1016/j.ibmb.2018.03.002

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# Accepted Manuscript

On the biosynthetic origin of carminic acid



Silas A. Rasmussen, Kenneth T. Kongstad, Paiman Khorsand-Jamal, Rubini Maya Kannangara, Majse Nafisi, Alex Van Dam, Mads Bennedsen, Bjørn Madsen, Finn Okkels, Charlotte H. Gotfredsen, Dan Staerk, Ulf Thrane, Uffe H. Mortensen, Thomas O. Larsen, Rasmus J.N. Frandsen

PII: S0965-1748(18)30111-5

DOI: 10.1016/j.ibmb.2018.03.002

Reference: IB 3039

To appear in: Insect Biochemistry and Molecular Biology

Received Date: 8 July 2017

Revised Date: 9 March 2018

Accepted Date: 13 March 2018

Please cite this article as: Rasmussen, S.A., Kongstad, K.T., Khorsand-Jamal, P., Kannangara, R.M., Nafisi, M., Van Dam, A., Bennedsen, M., Madsen, Bjø., Okkels, F., Gotfredsen, C.H., Staerk, D., Thrane, U., Mortensen, U.H., Larsen, T.O., Frandsen, R.J.N., On the biosynthetic origin of carminic acid, *Insect Biochemistry and Molecular Biology* (2018), doi: 10.1016/j.ibmb.2018.03.002.

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#### 25 ABSTRACT

- 26 The chemical composition of the scale insect Dactylopius coccus was analyzed with the aim to discover new 27 possible intermediates in the biosynthesis of carminic acid. UPLC-DAD/HRMS analyses of fresh and dried 28 insects resulted in the identification of three novel carminic acid analogues and the verification of several 29 previously described intermediates. Structural elucidation revealed that the three novel compounds were 30 desoxyerythrolaccin-O-glucosyl (**DE-O-Glcp**), 5,6-didehydroxyerythrolaccin 3-O-β-D-glucopyranoside (**DDE-**31 **3-O-Glcp**), and flavokermesic acid anthrone (FKA). The finding of FKA in *D. coccus* provides solid evidence of 32 a polyketide, rather than a shikimate, origin of coccid pigments. Based on the newly identified compounds, we present a detailed biosynthetic scheme that accounts for the formation of carminic acid (CA) in D. 33 34 coccus and all described coccid pigments which share a flavokermesic acid (FK) core. Detection of coccid 35 pigment intermediates in members of the Planococcus (mealybugs) and Pseudaulacaspis genera shows that 36 the ability to form these pigments is taxonomically more widely spread than previously documented. The 37 shared core-FK-biosynthetic pathway and wider taxonomic distribution suggests a common evolutionary 38 origin for the trait in all coccid dye producing insect species. 39
- 40

Keywords: Carminic acid, carmine, anthraquinones, coccid pigment, polyketide, insects, Dactylopius coccus, 41

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#### 47 1. Introduction

48 Pigments derived from insects and especially coccids (scale insects) have been used by humans since 49 ancient times for dyeing textiles, in cosmetics and in paints, and for coloring foods (Donkin, 1977). The 50 most commonly used coccid dyes include kermesic acid (KA), laccaic acids (LA) and carminic acid (CA), 51 which share a red color hue due to a similar chromophore structure (Łagowska and Golan, 2009). The 52 compounds, or combinations of these, have been reported to be produced by several species of distantly 53 related scale insects (Hemiptera: Coccoidea). Mainly five species, namely Porphyrophora hamelii 54 (Armenian/Ararat cochineal), Kermes vermilio (kermes), Porphyrophora polonica (Polish cochineal), 55 Dactylopius coccus (Mexican cochineal) and Kerria lacca (Indian lac insect) have at various points in history, 56 and at different geographical localities, been utilized by humans for large scale production of coccid dyes 57 (Donkin, 1977). Carminic acid and its aluminum salt carmine (E120) is by many considered as the pinnacle 58 of coccid dyes, based on its hue, light, temperature, and oxidation stability, and the yields by which it can 59 be obtained from natural sources (Dapson, 2007). CA is known to be produced by P. hamelii (Asia Minor), P. 60 polonica (Europe), and D. coccus (Meso and South America), all of which have served as sources for the 61 compound (Wouters and Verhecken, 1989). Present day production is based on D. coccus due to its 62 exceptional high pigment content (16-22% of dry weight), low fat content, and the ease by which the insect 63 can be cultured and harvested from leaves of Opuntia cacti (Donkin, 1977; Downham and Collins, 2000). A 64 thorough introduction to the historical use and geopolitical role of carmine is given by Dapson (Dapson, 65 2007). 66 Although insect-derived pigments have been utilized by humans for millennia and remain of significant

67 value within the food colorant market, the underlying biochemistry for their production remains largely 68 unknown. The coccid dyes, such as CA, have by many authors been categorized as polyketides solely based 69 on their structure (Morgan, 2010; Cameron et al., 1978; Pankewitz and Hilker, 2008; Brown, 1975). The 70 biosynthesis mechanisms of formation of polyketides, via the successive condensation of acetyl-CoA and 71 malonyl-CoA units catalyzed by polyketide synthases (PKSs), is well described in bacteria, fungi, and plants 72 (Staunton and Weissman, 2001). However, no animal PKSs have yet been biochemically characterized, even 73 though many insect species are known to contain compounds that potentially may be synthesized via the 74 polyketide pathway. In most cases, the putative polyketides contents have been ascribed to the 75 sequestering of precursors, or the finished compounds, from the insects' diet (Pankewitz and Hilker, 2008). 76 This situation is seen in *Timarcha* spp. (leaf beetles) which accumulates anthraquinones from its host plant 77 Galium spp. (Rubiaceae) (Petitpierre, 1981), and in Laetilia coccidivora (pyralid moth) and Hyperaspis 78 trifurcate (coccinellid beetle) larva that accumulate CA by predating on Dactylopius spp. (Eisner et al., 79 1994). In other cases, the origins of the detected polyketides in insects have been linked to the activity of

80 endosymbiotic bacteria, such as the production of pederin, a polyketide-peptide hybrid, in *Paederus* spp. 81 (rove beetles), which depends on an endosymbiotic bacterium related to Pseudomonas aeruginosa (Piel, 82 2002; Kellner, 2002). A second example is the facultative endosymbiotic *Rickettsiella* spp. responsible for the production of the polyketide viridaphin A1 in various aphids (Acyrthosiphon pisum and Megoura 83 84 crassicauda) (Tsuchida et al., 2010; Horikawa et al., 2011). The biosynthetic origin of coccid pigments in 85 scale insects, however, remains a mystery. The hypothesis that coccid dyes are polyketides has solely been 86 based on their chemical structure, even though the polyketide class is characterized by a shared mode of 87 synthesis rather than shared structural features (Staunton and Weissman, 2001). In fact, for the 88 anthraquinone core of CA, one could envision that this is formed via the shikimate based chorismate/O-89 succinyl benzoic acid pathway, as described for lucidi, alizarin, and morindone in rubiaceous plants 90 (Leistner, 1973; Han et al., 2001). If CA is formed by this pathway it would not qualify as a polyketide and its 91 formation would not dependent on a PKS. However, it is possible to distinguish between the two 92 alternative pathways as the polyketide-based synthesis would include a unique anthrone intermediate, 93 which is not found in the shikimate-based pathway, where the anthraquinone is formed directly. 94 The present study aims at increasing our understanding of the CA origin and its biosynthesis in D. coccus. 95 We also aim to elaborate on the previously proposed links to other biosynthetic pathways responsible for 96 the production of coccid dyes within the Coccoidea superfamily. In our study, we report the presence of 97 flavokermesic acid anthrone (FKA) in D. coccus, which strengthens the hypothesis that coccid pigments are 98 formed via the polyketide pathway, as **FKA** is the first cyclic intermediate in a polyketide-based pathway. 99

### 100 **2. Materials and methods**

#### 101 2.1. Biological material

Adult *D. coccus* specimens were collected from *Opuntia* cacti pads on the Canary Islands, Lanzarote, near the village of Guatiza, June 2012. The insects were transported to Denmark either as live specimens on cacti pads or as dead specimens stored on dry ice. Additional *D. coccus* insects were collected from cacti pads near the city of Arequipa Peru, August 2012, flash frozen in liquid nitrogen, and shipped to Denmark on dry ice. Commercially available dried *D. coccus* insects were supplied by Chr. Hansen A/S.

Coccus hesperidium, Pseudococcus longispinus, Palmicultor browni, and Pseudaulacaspis pentagona were
 collected in the greenhouses of the Botanical Garden (Natural History Museum of Denmark, University of
 Copenhagen) in Copenhagen in June 2014, and identified using the latest available identification keys

- 110 (Dooley and Dones, 2015; Miller et al., 2014).
- 111

112 2.2. Instrumentation

113 Chemical analysis of D. coccus samples was performed using three different LC-MS setups. UPHLC-DAD-114 HRMS was performed on a maXis G3 QTOF mass spectrometer (Bruker Daltronics, Bremen, Germany) 115 equipped with an electrospray ionization source coupled to an Ultima 3000 UHPLC-DAD (Dionex). Separation was performed on a Kinetex C<sub>18</sub> column (150 × 2.1 mm, 2.6 µm, Phenomenex Inc., Torrance, CA, 116 USA) maintained at 40 °C using a linear H<sub>2</sub>O-acetonitrile gradient consisting of A: milliQ H<sub>2</sub>O containing 10 117 mM formic acid and B: acetonitrile containing 10 mM formic acid from 10 to 100% B in 10 min with a flow 118 119 rate of 400 µL min<sup>-1</sup>. The **FK** anthrone was detected on a HPLC-DAD-HRMS system consisting of an Agilent 1200 chromatograph comprising quaternary pump, degasser, thermostatted column compartment, 120 121 autosampler, and photodiode array detector (Agilent Technology, Santa Clara, CA, USA) and a Bruker 122 micrOTOF-Q II mass spectrometer (Bruker Daltonik, Bremen, Germany) equipped with an electrospray 123 ionization source and operated via a 1:99 flow splitter. Analyses were performed at 40 °C on a Luna C<sub>18</sub>(2) 124 reversed-phase column ( $150 \times 4.6$  mm, 3  $\mu$ m particle size, 100 Å pore size, Phenomenex Inc., Torrance, CA, USA) with a flow rate of 800  $\mu$ L min<sup>-1</sup>. HPLC solvent A consisted of H<sub>2</sub>O-acetonitrile 95:5 (v/v) with 0.1% 125 formic acid and solvent B consisted of acetonitrile- $H_2O$  95:5 (v/v) with 0.1 % formic acid. Separation was 126 127 obtained using a linear gradient from 0 to 100% B in 20 minutes. Mass spectra were acquired in negative 128 ionization mode. The search for coccid dye intermediates in the different scale insect species was 129 performed on a 6540 Ultra High Definition UHD Accurate Mass Quadrupole Q-TOF LC/MS system (Agilent 130 Technology, Santa Clara, CA, USA). Separation of the analytes was conducted on a Kinetex XB-C<sub>18</sub> (100 x 4.6 mm i.d. 2.6 µm, Phenomenex Inc., Torrance, CA, USA), column maintained at 35 °C. The analytes were 131 eluted with a flow rate of 400 400 µL min<sup>-1</sup>using a water-acetonitrile gradient consisting of the following 132 133 steps: 100% water for 50 s followed by an gradual increase to 18.6% over 60 s, to 37.8% over 60 s, to 52.2% 134 over 120 s, to 54.2% over 70 s, to 90% over 120 s, to 100% over 120 s followed by 60 s at 100% acetonitrile. 135 The column was reconstituted with 100% water for 110 s prior to injection of the subsequent sample. 136 NMR spectra of 5,6-didehydroxyerythrolaccin 3-O- $\beta$ -D-glucopyranoside (**DDE-3-O-Glcp**), 137 desoxyerythrolaccin O-glucopyranoside (DE-O-Glcp), and dcll were recorded on a Varian Inova 500 MHz 138 (Varian Inc., Palo Alto, California) using a 5-mm probe. Samples were dissolved in 500  $\mu$ l DMSO- $d_6$  and 139 referenced to  $\delta_{\rm H}$  at 2.50 ppm and  $\delta_{\rm c}$  at 39.5 ppm. The NMR spectrum of flavokermesic acid anthrone (**FKA**) 140 was recorded on a Bruker Avance III HD 600 MHz NMR spectrometer (<sup>1</sup>H operating frequency 600.13 MHz) 141 equipped with a cryogenically cooled 5-mm CPDCH probe-head (Bruker Biospin, Rheinstetten, Germany). 142 The sample was dissolved in acetone- $d_6$  and referenced to  $\delta_H$  2.05 ppm and  $\delta_C$  29.84 ppm. Following 143 structural elucidation of the described compounds, their presence in the original samples was verified using 144 targeted MS analysis.

Chiral GC-MS was performed using 10 μg DDE-3-O-Glc that was hydrolyzed in 10% aqueous HCl for 90 min
at 90 °C, dried by a steam of N<sub>2</sub> and dissolved in 40 μL dry pyridine followed by 10 μL N-methyl-bistrifluoroacetamide (MBTFA) (GC-grade, 99%, Sigma-Aldrich) and heating to 65 °C for 40 min. The sample
was cooled to room temperature and subsequently analyzed on a CP-ChiraSil-L-Val GC column (25 m × 0.25
mm × 0.12 μm, Agilent Technology, Santa Clara, CA, USA) programmed to 70–150 °C at 4 °C min<sup>-1</sup> using an
HP 6890 series GC system and Agilent 5973 mass selective detector. The sample was compared to the
standards of D- and L-glucose (Sigma-Aldrich, St. Louis, MO, USA).

152

#### 153 2.3. Extraction and isolation

Desoxyerythrolaccin (DE) and DDE-3-O-Glcp were purified from 100 g of dried D. coccus. The insects were 154 155 grinded and extracted with MeOH: $H_2O$  (1:1) followed by liquid-liquid partitioning with EtOAc at pH 3 and 156 concentrated *in vacuo*. The crude extract was first separated by ion-exchange using an NH<sub>2</sub> flash column. Carboxylic acid containing compounds was retained when washed with 50% aqueous MeCN containing 10 157 158 mM ammonium formate. Finally, the acidic compounds (FK, KA, and CA) were eluted with 50% aqueous 159 MeCN adjusted to pH 11 with ammonium hydroxide. **DE** and **DDE-3-***O*-**Glc***p* were purified from the 50% 160 MeCN 10 mM ammonium formate eluate using reversed-phase semi-preparative chromatography on a 250 161 × 10 mm Luna2 C<sub>18</sub> column (Phenomenex, Torrance, CA, USA) using a Gilson HPLC system. Compounds were eluted with a gradient consisting of MilliQ H<sub>2</sub>O:MeCN, both containing 50 ppm TFA. FK and dcll were 162 163 recovered in the alkaline eluent of the NH2 column. The extract was then further purified on Isolute diol 164 material (Biotage, Uppsala, Sweden) on an Isolera auto flash purification system (Biotage, Uppsala, 165 Sweden) in a step-wise elution from dichloromethane to EtOAc to MeOH to afford dcll and FK.

166

### 167 2.4. Synthesis of flavokermesic acid anthrone

Synthesis of flavokermesic acid anthrone from flavokermesic acid was conducted according to a previously 168 169 published method by Schätzle, with slight modifications (Schätzle, 2012). In brief, 10 mg flavokermesic acid 170 was dissolved in 1mL glacial acetic acid and 0.2 mL hydriodic acid (57 wt. % in H<sub>2</sub>O) in a sealed microwave reactions vial and heated to 50 °C under stirring for 2 h in the dark. This adaption to the method reported 171 172 by Schätzle 2012 was done to ensure full conversion of the flavokermesic acid to the anthrone with only 173 limited decarboxylation. The hydriodic acid was quenched with 10 mL saturated Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and extracted 174 three times with diethyl ether. The ether phase was dried over MgSO<sub>4</sub> and lyophilized under reduced 175 pressure. The sample was at all times kept in the dark and chemical analyses were performed in amber vials 176 and NMR tubes to minimize the risk of dimerization and oxidation. The formation of the anthrone was 177 confirmed by NMR analysis (Supplementary data Table S1).

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#### 179 2.5. Biosynthetic models

The biosynthetic models for formation of CA and related compounds were formulated using the
retrosynthesis approach including commonly accepted enzymatic driven reactions, as described in the
BRENDA database (Schomburg et al., 2004), and the available structural data for coccid dyes (Morgan,
2010; Cameron et al., 1978, 1981; Brown, 1975, Peggie et al., 2007; Stathopoulou et al., 2013; Bhide et al.,
1969). The models were drawn using ChemDraw 15.9.9.106 (PerkinElmer Informatics, Inc., US).

185

#### 186 **3. Results**

### 187 3.1. Detection of compounds in Dactylopius coccus extracts

HPLC-HRMS analysis of the raw extracts from fresh D. coccus showed that the main extractable pigment 188 components were CA, FK, KA and dcll as previously reported (Wouters, J., Verhecken, A., 1989; Peggie et 189 190 al., 2007; Méndez et al., 2004) (Figure 1A). These compounds were putatively identified based on relative 191 retention time, high-resolution mass, UV/VIS spectra, and MS/MS fragmentation patterns. Spiking with 192 authentic samples of CA and KA confirmed these compounds identity. FK was isolated and subjected to 193 structural elucidation by 2D NMR (Supplementary data Table S1), and the data were in agreement with and 194 confirmed the previously reported structure of FK (Wouters and Verhecken, 1987). CA and dcll were not 195 easily separated using reversed-phase HPLC, and isolation of the two compounds relied on normal-phase 196 flash chromatography on diol substituted silica. The putative dcll was further purified using semi-197 preparative HPLC and characterized by HR-MS and 2D NMR. Structural elucidation of the compound by 2D NMR experiments (Supplementary data Table S1) showed that dcll was flavokermesic acid 2-C-β-D-198 199 glucopyranoside, and the NMR data were in agreement with those reported for dcll (Stathopoulou et al., 200 2013). The reversed-phase HPLC-based analysis also revealed a previously undescribed major peak, eluting 201 at 13.40 minutes (Figure 1A and 1C). The mass of the corresponding compound equaled the theoretical 202 mass of flavokermesic acid anthrone (FKA), and this identity was confirmed by comparison with a FKA 203 standard, semi-synthesized from authentic FK. In addition, to the metabolites detected using reversed-204 phase chromatography, strong anion exchange SPE (SAX SPE) were used to identify three non-acidic 205 metabolites (DE, DDE-3-O-Glcp and DE-O-Glcp,) that all displayed UV/VIS spectra with similarities to those 206 reported for FK, KA, dcII and CA (Figure 1B, 1D, 1E and 1F). For unambiguously structure elucidation of DE 207 and DDE-3-O-Glcp, the compounds were purified in amounts sufficient for structural elucidation by HRMS 208 and NMR spectroscopy. Purification of the **DE-O-Glcp** compound ufortionatly did not yield sufficient 209 quantities for full structural elucidation by NMR.

- 3.2. Structural elucidation of DE, and the novel compounds DE-O-glucosyl, DDE-O-glucosyl and FKA
- 212 **DE** displayed UV/VIS spectrum (Figure 1E) similar to that of **FK**, indicating a similar core skeleton. The
- compound was not retained on a SAX column, suggesting that it lacked the carboxylic acid group found at
- 214 C-7 in **FK**. This conclusion was supported by HRMS (m/z 271.0600 [M+H]<sup>+</sup>, calcd 271.0600, ΔM 0.0 ppm),
- suggesting a molecular formula of  $C_{15}H_{10}O_5$ , i.e., **DE** lacking  $CO_2$  as compared to **FK**. Structural elucidation
- 216 was carried out by <sup>1</sup>H NMR and 2D NMR spectroscopy (Table S1). The <sup>1</sup>H NMR spectrum showed a signal for
- the OH-group positioned *peri* to the carbonyl group (δ 13.30, 1-OH), two sets of meta-coupled protons H-5
- 218 and H-7 ( $\delta$  6.54 and 7.04, respectively,  ${}^{3}J_{H5-H7}$  = 1.9 Hz) and H-2 and H-4 ( $\delta$  7.43 and 7.01, respectively,  ${}^{3}J_{H2-H4}$
- = 2.5 Hz), and a *peri*-positioned methyl group ( $\delta$  2.81, s, 11-CH<sub>3</sub>). The meta-coupling between H-5 and H-7
- clearly proves the lack of the carboxylic acid in position 7, and thus HRMS and NMR data supported the
- compound to be **DE**, also known as 3-hydroxy-aloesaponarin II (Mehandale et al., 1968), and previously
- observed in air-dried *D. coccus* (Sugimoto et al., 1998).
- 223 Structural analysis of the first novel compound **DDE-3-O-Glc** revealed that the compound was a hitherto
- undescribed *O*-glucoside of 5,6-didehydroxyerythrolaccin (**DDE**) (Figure 2). The high-resolution mass
- spectrum of **DDE-3-***O***-Glc***p* suggested a molecular formula of  $C_{21}H_{20}O_9$  ([M+H]<sup>+</sup> m/z 417.1180, calcd.
- 226 C<sub>21</sub>H<sub>21</sub>O<sub>9</sub><sup>+</sup> 417.1180, ΔM 0.0 ppm; [M-H]<sup>-</sup> *m*/z 415.1034, calcd. C<sub>21</sub>H<sub>19</sub>O<sub>9</sub><sup>-</sup> 415.1029, ΔM 1.2 ppm. In addition,
- the compound exhibited a loss of m/z 162.0528, which is likely due to the loss of a labile O-linked hexose
- 228 unit. The structural elucidation was carried out based on <sup>1</sup>H NMR and 2D NMR spectroscopy
- 229 (Supplementary data Table S1). The <sup>1</sup>H spectrum showed resemblance to that of **DE**, but instead of the two
- 230 doublets observed for the meta-coupled H-5 and H-7 in **DE**, signals for H-5 ( $\delta$  8.12, dd, 7.5, 1.2 Hz), H-6 ( $\delta$
- 231 7.78, t, 7.5 Hz), and H-7 (δ 7.75 dd, 7.5, 1.3 Hz) showed the absence of a hydroxyl group at C-6 in **DDE-3-***O*-
- **Glcp**. Furthermore, a doublet at  $\delta$  5.26 ( ${}^{3}J_{H1'-H2'}$  = 7.6 Hz) for a  $\beta$ -configuration of the anomeric proton H-1'
- 233 (as well as the  $^{13}$ C value of 101.1 ppm for C-1) and the remaining  $^{1}$ H and  $^{13}$ C signals for H-2' to H-6' and C-2'
- to C-6' (Supplementary data Table S1), are in agreement with a  $\beta$ -D-glucose unit (Bock and Pedersen, 1983).
- The *O*-linkage of the sugar was evident from a more deshielded anomeric proton (101.1 ppm) compared to
- that of *C*-glucosyl linkages reported for related compounds (Stathopoulou et al., 2013). In addition, a HMBC
- correlation from H-1' to C-3 (164.6 ppm) further confirmed the *O*-glucosyl linkage to C-3. Thus, to establish
- 238 the D- or L-configuration of the glucose moiety, an aliquot was hydrolyzed and analyzed by GC-MS and
- chiral GC-MS. This confirmed that the hexose moiety was D-glucose (Figure S2 and S3). Thus, the compound
  was identified as 5,6-didehydroxyerythrolaccin 3-*O*-β-D-glucopyranoside.
- 241 Several attempts to purify the second novel compound **DE-O-Glcp** did not yield sufficient quantities to
- allow acquisition of NMR data. However, the high-resolution mass spectrum of **DE-O-Glcp** suggested a
- 243 molecular formula of  $C_{21}H_{20}O_{10}$  ([M+H]<sup>+</sup> m/z 433.1129, calcd.  $C_{21}H_{21}O_{10}^{+}$  433.1129,  $\Delta M$  0.0 ppm;  $C_{21}H_{20}O_{10}$

244  $([M+H]^+ m/z 431.0981, calcd. C_{21}H_{19}O_{10}^- 431.0984, \Delta M 0.7 ppm)$ ; and the loss of a hexose moiety (m/z)162.0530) generated a fragment with the same mass as DE. These results indicate that this compound is an 245 246 O-glucosylated form of **DE** (Figure 2), but the exact position of glucosylation could not be established. 247 The third novel compound was only detected in extracts from fresh and frozen D. coccus and identified as 248 flavokermesic acid anthrone (FKA) (Figure 2). The high-resolution mass spectrum of FKA detected in the 249 insect suggested a molecular formula of  $C_{16}H_{12}O_6$  ([M-H] m/z 299.0559, calcd.  $C_{16}H_{11}O_6$  299.0561,  $\Delta M$  1.0 250 ppm). Positive identification of this compound was achieved by comparison of retention time, high-251 resolution MS, MS/MS fragmentation pattern, and UV/VIS spectrum for FKA that had been prepared by 252 chemical semi-synthesis. See Table S1 for NMR data. Chemical semi-synthesis of FKA demonstrated that 253 the pure compound is prone to dimerization as well as oxidation to FK in the presence of oxidants under in 254 vitro conditions. The observed spontaneous oxidation to FK likely explains why only fresh and frozen D. 255 coccus was found to contain FKA while the compound was not detected in dried insects which have been 256 exposed to light, oxygen and other oxidative agents for longer periods of time.

257

258 3.3. LC-DAD/MS-based screening of coccid dye production in selected members of the superfamily

259 Coccoidea

260 To analyze the taxonomic distribution of the ability to produce coccid dyes, we collected representatives of

261 four different *Coccoidea* families found in Denmark. The collected species included nymphal states of

262 Coccus hesperidium, Pseudococcus longispinus, Palmicultor browni and Pseudaulacaspis pentagona.

263 Metabolites from approximately 1 g of each of the individual species were extracted and analyzed by LC-

264 MS/DAD. Extracted ion chromatograms (Figure 3) for masses equivalent to the known coccid dye

intermediates showed that *C. hesperidium* contained KA, FK, dcll and CA, while *P. longispinus* and *P. browni* 

266 both contained **FK** and **dcll**. Positive identification was based on accurate mass, retention time, UV/VIS

spectra and authentic standards. Analysis of the *P. pentagona* material did not reveal the presence of any
coccid dye intermediates in this species.

269

### 270 **4. Discussion**

Assyrian cuneiform texts document that the utilization of scale insects for the production of pigments and
dyes dates back to at least 3000 BC (Dapson 2007). Different civilizations separated by time and geography
have relied on a limited number of scale insect species and the pigments they produce (Morgan, 2010;
Cameron et al., 1978, 1981; Brown, 1975, Petitpierre, 1981: Peggie et al., 2007; Stathopoulou et al., 2013;
Bhide et al., 1969). The long history and traditions associated with coccid dye utilization have resulted in a
situation where scientific literature has focused on a very limited number of producing species (*Dactylopius*)

277 coccus, Kerria lacca, and Kermes vermilio) indicating a discontinuous taxonomic distribution of the trait and 278 unrelated biosynthetic origins for the various pigments. However, the number of shared metabolites and 279 the common FK core structure of the pigments suggest that all coccid dyes are formed via a similar 280 biosynthetic scheme with a common evolutionary origin. If so, then many more scale insect species, i.e. 281 those forming a monophyletic clade with known producers, would also be expected to be able to produce 282 coccid dyes or related metabolites. To this end, we have demonstrated that members of the Planococcus (mealybugs) and Pseudaulacaspis genera also produce coccid dyes. Hence, the ability to form these 283 284 pigments is indeed more widely taxonomically spread than previously believed. This would support the 285 most parsimonious explanation, being a common evolutionary origin within Coccoidea. Of evolutionary 286 importance here is that Porphyrophora and Dactylopius belong to two vastly different clades of Coccoidea. 287 Porphyrophora belongs to the more primitive Margarodidae and Dactylopius to the Dactylopiidae within the more highly derived Neococcoidea clade. The Margarodidae and Dactylopiidae share a most recent 288 289 common ancestor 250 million years ago (Vea et al., 2016).

290

#### 291 *4.1.* The biosynthetic origin of coccid dyes

The biosynthetic origin of coccid dyes such as **KA**, **FK**, **CA**, and **LA** has long been debated (Brown, 1975; Joshi and Lambdin, 1996; Ramirez-Puebla et al., 2010). Several studies have rejected the hypothesis that host plants supply the insects with any of the known coccid dye intermediates. Similarly, we were unable to detect any of these intermediates in *Opuntia* cacti pads in the case of *D. coccus* (data not shown). To say nothing about the wide variety of different hosts utilized by many *Coccoidea*, e.g. over 400 plant species are described as host for *Kerria lacca* (Sharma et al., 1997).

Accordingly, it seems clear that Coccoidea must be able to synthesize the coccid dyes de novo from simple 298 299 metabolites, e.g. glucose, present in the phloem sap of a wide range of plant species. Several studies have 300 suggested that endosymbiotic bacteria may be responsible for the formation of the coccid dyes, and 301 members of the Wolbachia and Azoarcus bacteria genera have been identified in the scale insects (Brown, 302 1975; Ramirez-Puebla et al., 2010; Pankewitz et al., 2007). However, none of these studies have proven a 303 direct link between the presence, or activity, of these endosymbionts with the formation of coccid dyes. An 304 alternative hypothesis is that the biosynthetic apparatus is encoded in the insects' nuclear genome. Several 305 examples exist where complex secondary metabolite's biosynthetic pathways are encoded by genes in the 306 genome of the producing insect, such as the Drosophila eye pigments drosopterin and ommochromes 307 (Chovnick et al., 1990; Nijhout, 1997). Though no one has yet succeeded in identifying or describing the 308 enzymological- or genetic basis for polyketide biosynthesis in insects, strong evidence does exist in support 309 of nuclear encodement of the enzymes required for producing coccid dyes. However, this has largely been

310 overlooked in past studies. Such evidence includes the report of a stable yellow color mutant of the 311 normally red Kerria lacca (Indian Lac-insect) (Negi, 1954), and a white color mutant incapable of producing 312 LAs (Chauhan and Teotia, 1973). Dissection of the genetic basis for these two mutations, by classical 313 genetic crossing experiments, showed that the two traits are non-allelic and that they follow simple 314 recessive inheritance (Chaucun, 1977; Chaucun and Mishra, 1977). If endosymbiotic bacteria were involved 315 in catalyzing steps in the formation of the core structure of the coccid dyes, the mutant trait would be expected to be maternally inherited as a result of transfer via eggs (Ferrari and Vavre, 2011). Any 316 317 conclusions on this matter must await demonstration of the ability of the color mutants to host endosymbiotic bacteria. Independently, support of an insect rather than bacterial origin of CA has been 318 319 provided by a series of microscopy studies of the hemolymph from various Dactylopius spp. These studies 320 described the existence of special granulocytes with a high concentration of secretory (M-) granules containing red pigments and small corpuscles of CA floating freely in the hemolymph (Joshi and Lambdin, 321 322 1996; Caselin-Castro et al., 2008, 2010).

323

#### 324 *4.2. Model for the biosynthesis of carminic acid*

Based on its structure, the biosynthesis of CA may be hypothesized to proceed by two different routes 325 326 (Figure 4). One envisioned biosynthetic scheme involves the polyketide pathway, also known as the 327 acetate/malonate pathway (Figure 4A). A second possible biosynthetic route involves the shikimate based chorismate/O-succinyl benzoic acid pathways (Figure 4B). Both biosynthetic schemes ultimately result in 328 329 the formation of anthraquinones; however they would be predicted to include different intermediates, e.g. 330 a unique anthrone in the case of a polyketide-based pathway, which can be used to distinguish between the two. Detection of the FK anthrone (FKA) in fresh and frozen D. coccus material (Figure 1) provides 331 332 support for a polyketide rather than shikimate origin of compounds with an FK core. That the anthrone is 333 detected in the current study can likely be attributed to the milder extraction conditions, and the use of 334 fresh material rather than dried insects or dyed textiles that have been used in previous reports on the subject. The anthrone is abundant in the fresh material as evident by the HPLC-HRMS-DAD analysis (Figure 335 336 1), but is not detected in dried insects, which is in good agreement with our previous observation that the 337 purified anthrone spontaneous oxidizes. Microbial based reduction of the FK anthraquinone to yield the 338 FKA anthrone is a possible alternative explanation for detection of the anthrone. de Witte and co-workers have previously shown that bacteria isolated from mammalian fecal material are capable of catalyzing the 339 340 reduction of the anthraquinone rhein to the corresponding anthrone. The currently available data do not 341 allow us to rule out this explanation, and further experiments e.g. feeding experiments are hence required 342 (de Witte et al. 1992).

343 Based on the detection of FKA and the novel CA related compounds identified in D. coccus, and under the 344 assumption that its formation is not the result of microbial reduction, we here propose a biosynthetic 345 pathway for the formation of CA in D. coccus (Figure 5). This biosynthetic pathway is an elaboration of the 346 models previous proposed by Brown (Brown, 1975) and Morgan (Morgan, 2010), and differs by including 347 additional intermediates and predictions for the required enzymatic activities and co-factors. The enzymatic machinery, responsible for the formation of polyketides in animals, remains unknown, and 348 349 several competing hypotheses exists. One possible explanation could be that the involved PKS has been 350 introduced into the genome of scale insects by horizontal gene transfer (HGT) from fungi or bacteria. 351 Several examples of HGT from fungi to insects have previously been documented such as the carotenoid 352 forming pathway in Acyrthosiphon pisum (pea aphid) (Moran and Jarvik, 2010). Synthesis of the FK core 353 requires the formation of a C7-C12 intermolecular bond in the octaketide backbone (Figure 5). Since fungal type I iterative PKSs have only been described to form either C2-C7 or C6-C11 bonds, fungi are an unlikely 354 355 donor (Li et al., 2010). Engineered bacterial type II PKS systems have previously been shown to be able to 356 produce FK, known as TMAC in the bacterial literature (Tang et al., 2004). Specifically, TMAC is formed by 357 the combined actions of the minimal actinorhodin PKS (act-KS $\alpha$ , act-KS $\beta$ , act-ACP) from *Streptomyces* coelicolor and the two cyclases (Zhul and ZhuJ) from Streptomyces sp. No. 1128 (Tang et al., 2004). To settle 358 359 whether HGT has formed the basis for CA production requires that the responsible genes are identified and 360 analyzed in the context of a high-quality scale insect genome sequence, which is not presently available. 361 A competing hypothesis for the origin of insect PKSs, and the one we favor, is that the putative PKSs may 362 have evolved from the insect's endogenous type I fatty acid synthase (FAS). Animal type I FAS and fungal type I iterative PKS are thought to have evolved from a common bacterial type I PKS ancestor (Hopwood 363 364 and Sherman, 1990; Kroken et al., 2003). The main product of FASs in animals is palmitic acid, a fully 365 reduced C-16 chain. This chain length is equivalent to an octaketide, which is the intermediate required for 366 FKA formation. Converting a FAS to a non-reducing PKS capable of producing a non-reduced linear 367 octaketide would require inactivation of the FAS's  $\beta$ -ketoreductase (KR) domain combined with a relaxation 368 of the substrate specificity of its  $\beta$ -ketosynthase domain (KS) to allow for non-reduced products to form. Non-reduced linear polyketides are highly reactive, due to the presence of carbonyl groups on every second 369 370 carbon atom, and they spontaneously fold into heterocyclic and aromatic structures via the formation of 371 intramolecular C-C bonds. Several studies have shown that non-reduced octaketides spontaneously form 372 the aromatic compounds SEK4 and SEK4b, which contain intermolecular bond configurations that differ 373 significantly from that of FKA (Figure 5) (Fu et al., 1994; Mizuuchi et al., 2009). The fact that we find only 374 FKA, and not SEK4 or SEK4b, in D. coccus extracts (data not shown) suggests that folding of the polyketide 375 chain does not proceed as a spontaneous reaction. Folding control of non-reduced polyketide backbones in

376 fungal type I iterative non-reducing PKS systems and bacterial type III PKS systems are achieved by a 377 'Product Template' (PT) domain in the PKS (evolved from DH domain) or by trans-acting cyclases and 378 aromatases, respectively (Shen et al., 1995; Bringmann et al., 2006). In the case of coccid pigments, 379 controlled folding of the linear octaketide to form FKA could depend on similar mechanisms that would 380 require additional mutations in the FAS or trans-acting enzymes as presented in Figure 5. It has not escaped 381 our notice that the mutated FAS hypothesis potentially also can explain the formation of other polyketides 382 found in insects, such as 5-hydroxy-7-methyl-6-acetylpurpurin from *Ericoccus* spp. (Coccoidea: Ericoccidae) 383 (Banks and Cameron, 1970), chrysophanol in Galeruca tanaceti (Coleoptera: Chrysomelidae) (leaf beetle) 384 (Bringmann et al., 2006), and the predicted monomeric precursors of protoaphins in aphids (Brown, 1975). 385 The listed compounds are all likely also formed from non-reduced octaketide precursors but display 386 alternative backbone folds and would hence dependent on other cyclases than those involved in coccid 387 dyes biosynthesis.

388 In the case of CA formation, the enzymatic steps following formation of FKA are predicted to include two 389 oxidations and a C-glucosylation. Based on the metabolites detected in D. coccus, monooxygenation of the 390 central aromatic ring (position C-10), from FKA to FK, likely occur before oxidation of the outer ring 391 (position C-4) and before C-glucosylation (position C-2), as neither of the detected metabolites contain a C-392 4 oxidation without a C-10 oxidation and as all known glucosylated intermediates (e.g. dcll and CA) have 393 the C-10 oxidation. Several studies of bacterial and fungal systems have shown that efficient in vivo 394 anthrone oxidation is dependent on specific anthrone oxidases (Chung et al., 2002; Erhlich et al. 2010), 395 making it likely that FKA to FK conversion is an enzyme dependent reaction in vivo, rather than a 396 spontaneous reaction. The order of the two subsequent reactions (monooxygenation of C-4 and C-397 glucosylation at C-2) that ultimately yield CA is unclear as both KA and dcll accumulates. Hence, it is 398 impossible to determine whether both pathways are active *in vivo* or whether one represents a shunt. 399 Monooxygenation of the C-4 position that converts FK to KA is likely catalyzed by either a cytochrome P-400 450 or flavin-dependent monooxygenases. The accumulation of several intermediates suggests that the 401 natural CA biosynthetic pathway is imbalanced; a situation that likely is caused by insufficient flux through 402 downstream enzymatic steps resulting in the buildup of intermediates.

403

404 4.3. Decarboxylation of the FKA core

We also demonstrated the presence of **DE**, **DE**-*O*-**Glc***p*, and **DDE**-**3**-*O*-**Glc***p* in freshly collected insects. **DE** has previously been observed in air-dried *D. coccus* (Sugimoto et al., 1998). Dehydroxy- and *O*-glucosylated forms of **DE** had not previously been reported in *D. coccus*. The reason why we detect the *O*-glucosylated forms may be attributed to the mild extraction conditions and moderate pH compared to previously

409 reported extraction protocols, which may result in hydrolysis of O-glucosides. Kerria laccas (lac insect) and 410 Austrotachardia acacia (Maskell) are also known to accumulate DEL, erythrolaccin (EL) and iso-411 erythrolaccin (IEL) in their resin deposits (Chauhan, 1977; Caselin-Castro et al., 2010). The structural 412 similarity and co-occurrence in multiple species suggest a common biosynthetic origin for the FK (C16) and 413 EL (C15) compound families. This is further supported by the observation made by Chauhan and Mishra (Chauhan and Mishra, 1977) who noted that a single mutation in white K. lacca strains affected both the 414 body color (primarily caused by LA) and the resin color (primarily caused by EL) (Bhide et al., 1969). Based 415 416 on this, we propose that the FK (C16) and EL (C15) compound families are products of the same biosynthetic mechanism in scale insects and that the difference in carbon number is due to decarboxylation 417 418 of FKA, as presented in Figure 5. A highly similar anthraquinone decarboxylation step has previously been 419 documented in the chrysophanol biosynthetic pathway in G. tanaceti, though the responsible mechanism 420 and timing of the decarboxylation step is unknown (Bringman et al., 2006).

421

422 4.4. Evolution of the FK biosynthetic pathways / Variations to the FK biosynthetic pathway

423 CA has been shown to act as a chemo deterrent that protects the immobile scale insects from predatory 424 ants (Eisner et al., 1980). However, García Gil de Muñoz and co-workers recently extended CA's biological 425 function by showing that it may contribute to the innate immune system of the scale insect to protect 426 against invading microorganisms (García-Gil De Muñoz et al, 2002, 2005; De La Cruz Hernandez-Hernandez, 427 2003; García-Gil De Muñoz, 2007). Specifically, this system depends on encapsulation of the invaders by 428 melanization via the rapid formation of eumelanin by polymerization of tyrosine and L-DOPA (Satyavathi et 429 al., 2014; Charles and Killian, 2015). Phenoloxidase (PO) is responsible for catalyzing multiple steps of the melanization cascade: tyrosine to L-DOPA, L-DOPA to dopaquinone, dopamine to dopaminequinone, and 430 431 dopamine to N-arachidonoyl dopamine. Reactions that generate reactive radicals, including reactive oxygen 432 species (ROS) and DOPA semi-quinones, which in addition to encapsulation may harm intruders (González-433 Santoyo and Córdoba-Aguilar, 2012). As shown by Garcia et al., PO can also act directly on CA resulting in the formation of insoluble polymers of CA (García-Gil De Muñoz et al, 2005; García-Gil De Muñoz, 2007). 434 435 This process is attributed to the quinone nature of **CA** allowing it to participate in redox cycling with ROSs. 436 Relocation of the radical within the conjugated system of **CA** or **FK** would allow for activation of multiple 437 positions in the **FK** core (C4, C8, C10, and C9-OH) as described for other naphthoquinones (Frandsen et al., 438 2006). This is a very interesting observation as the formation of **FK** radicals potentially can explain how the 439 tyrosine-derived groups found in LAs are added to the FK core by radical activation of the C10 positions and 440 oxidative coupling with tyrosine or one of its derivates (tyrosol, N-acetyltyramine, tyramine or 2-(4-441 hydroxyphenyl)ethyl acetate) (Figure 6). Figure 6 summarizes the chemical diversity and the required

- 442 enzymatic steps in the form of a meta-biosynthetic pathway accounting for all known coccoid pigments.
- 443 The model includes five different monooxygenases, acting on C4, C6, C8, C16, two dehydrogenases, a
- 444 decarboxylase and a *C*-glucosyltransferase.
- The extensive chemical diversity and existence of multiple alternative decoration patterns support the
- 446 hypothesis that the **FK** forming biosynthetic pathway has a long evolutionary history within the *Coccoidea*
- 447 superfamily.
- In summary, we propose that formation of **CA** depends on the activity of a modified fatty acid synthase or
- 449 polyketide synthase, possibly one or more cyclases/aromatases, one anthrone oxidases, a 'cytochrome
- 450 P450 monooxygenases'/'flavin-dependent monooxygenases', and a C-glucosyl transferase. Validation of the
- 451 proposed hypothetical biosynthetic schemes and the involved enzyme types naturally depends on future
- 452 biochemical evidence and mapping of their genetic basis in *D. coccus* or an endosymbiont organism.
- 453

### 454 Abbreviations

- 455 Flavokermesic acid anthrone (FKA); flavokermesic acid (FK); kermesic acid (KA); carminic acid (CA); C-
- 456 glucosylated flavokermesic acid (dcll); laccaic acid (LA); desoxyerythrolaccin (DE), 5,6-
- 457 didehydroxyerythrolaccin (**DDE**); didehydroxyerythrolaccin 3-*O*-β-D-glucopyranoside (**DDE-3-***O*-**Glc***p*);
- 458 erythrolaccin (EL); iso-erythrolaccin (IEL)
- 459

### 460 Acknowledgements

- The Danish Innovation Foundation is acknowledged for funding the work via grant no. XXXXXXXX. Center
  for Synthetic Biology funded by the University of Copenhagen's Excellence Program for Interdisciplinary
  Research is acknowledged for technical and financial support. The US NSF for funding AVD's involvement in
  the project via grant no. DBI-1306489. We would like to thank Cultivo de la cochinilla en Mala y Guatiza for
- their assistance with collection the live *Dactylopius coccus* specimens on Lanzarote. We are also thankful
- 466 for Jimmy Oluf Olsen, from the Royal Botanical Garden, the Natural History Museum of Denmark,
- 467 University of Copenhagen, for his assistance with collecting.
- 468

### 469 **Conflicts of interest**

- 470 Authors SAR, KTK, DS, CHG, UT, UHM, TOL, and RJNF declare no financial nor any competing financial
- 471 interests. The authors, PKJ, MB and BM are or were employed by the private company Chr. Hansen A/S that
- 472 produces and sells *D. coccus* derived carmine as a food pigment in a business-to-business setup.
- 473
- 474

### 475 Appendix A. Supplementary data

- 476 Supplementary data related to this article can be found at http://dx.doi.org/ToBeInserted
- 477

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### 635 TABLES AND ARTWORK





Figure 1. HPLC-HRMS analysis of CA related compounds found in *Dactylopius coccus* with annotation of
major peaks and UV/VIS spectra for the novel compounds. A: Annotated EIC chromatogram for the CA
related compounds identified in the raw extract of *D. coccus*, coloring scheme for the EIC is shown below
the chromatogram. B: Annotated EIC chromatogram for CA related compounds identified in the SAX
purified *D. coccus* extract, C-F: UV/Vis spectra for DE and the new compounds DDE-3-O-Glcp, DE-O-Glcp,
and FKA.



5,6-dihydrohydroxyerythrolaccin 3-O- $\beta$ -D-glucopyranoside **DDE-3-O-Glc***p* 



Flavokermesic acid anthrone **FKA** 

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648 Figure 2. Structures of the new compounds DDE-3-O-Glcp, and FKA

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653 *Hemiberlesia*, and *Palmicultor* genera. Extracted ion chromatograms at m/z 313.0351 ± 0.01 equal to the

654 [M-H]<sup>-</sup> ion of **FK** (green), at 329.0246 ± 0.01 for the [M-H]<sup>-</sup> ion of **KA** (purple), at 475.0875 ± 0.01 for the [M-

655 H]<sup>-</sup> ion of **dcll** (red), and at 491.0832 ± 0.01 for the [M-H]<sup>-</sup> ion of **CA** (green).



- 660 Figure 5. The two theoretical biosynthetic schemes that can lead to the formation of carminic acid with the first stable
  - 661 tricyclic intermediates shown in boxes. Panel A shows the polyketide based pathway while B shows the O-succinyl 662 benzoic bases mechanism.



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666 Figure 5. Model for carminic acid biosynthesis in Dactylopius coccus, including predicted enzyme types, substrates, 667 and co-factors. Compounds in brackets represent hypothetical intermediates that have not been detected. Signatures: 668 ● compound identified based on HRMS data; ●● compound identified by HRMS and NMR data; ●●● Compound 669 identified based on authentic standard and HRMS data. The numbering of carbon atoms refers to their position in the 670 polyketide backbone, counting from the carbon closest to the enzyme upon completion of the catalytic program. 671 Enzymes: OxidoR = oxidoreductase; MO = monooxygenase (three different a, b and f); UGT = UDP-glucose dependent 672 membrane bound glucosyltransferase; PKS/FAS = polyketide/'mutated fatty acid' synthase; Cyclase/aromatase = 673 small-molecule-foldases as found in bacterial type II iterative PKS systems.



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**Figure 6.** A summative meta-model for the biosynthesis of coccid dyes in scale insects, including predicted enzyme types, substrates, and cofactors. Compounds in brackets represent hypothetical intermediates. The meta-model includes one decarboxylation step and six different monooxygenases (MOa-f) reactions, which are likely catalyzed by different enzymes, based on different positions in the **FK** core that is modified. The oxidative coupling reactions, branching out from the two hypothetical intermediates, are likely unspecific reactions that occur via highly reactive radicals generated by the insect's innate immune system.

- Two novel coccid pigment intermediates isolated from *Dactylopius coccus*, including flavokermesic acid anthrone.
- Detection of the anthrone for the first time provides solid evidence for biosynthesis via the polyketide biosynthetic pathway.
- Coccid pigments is much wider spread within the Coccoidea superfamily than previously believed, suggesting a common genetic basis and evolutionary origin.