

N2O and NO dynamics in AOB-enriched and mixed-culture biomass: experimental observations and model calibration

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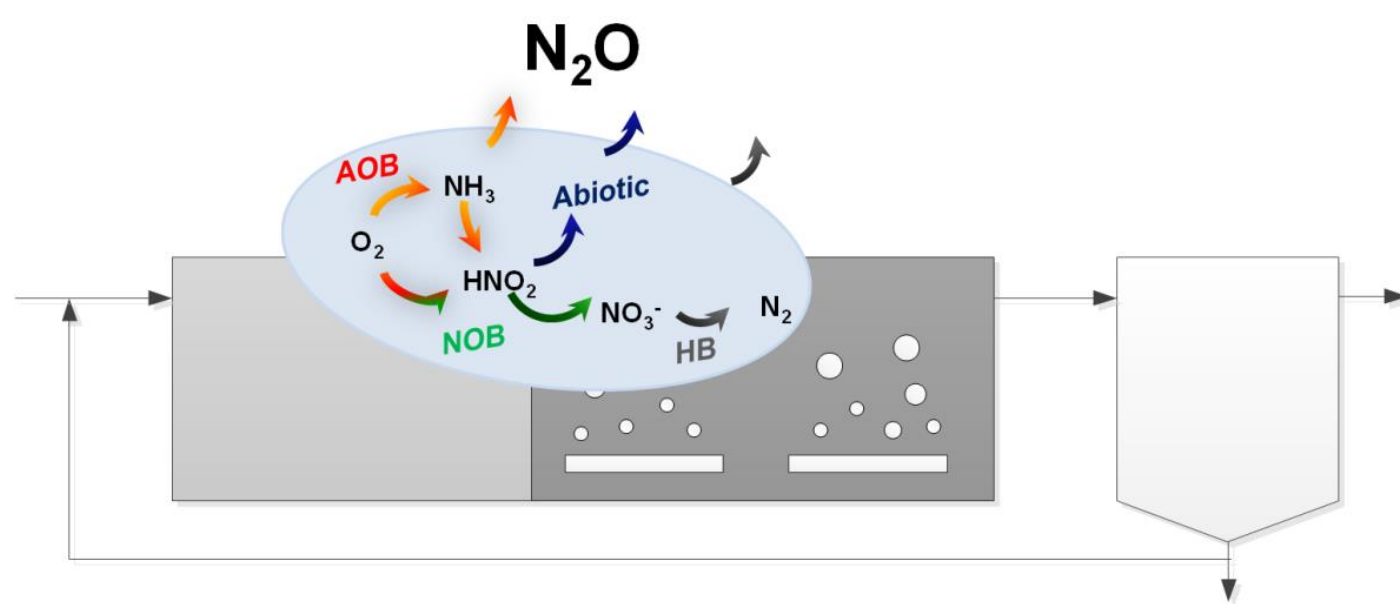
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Introduction

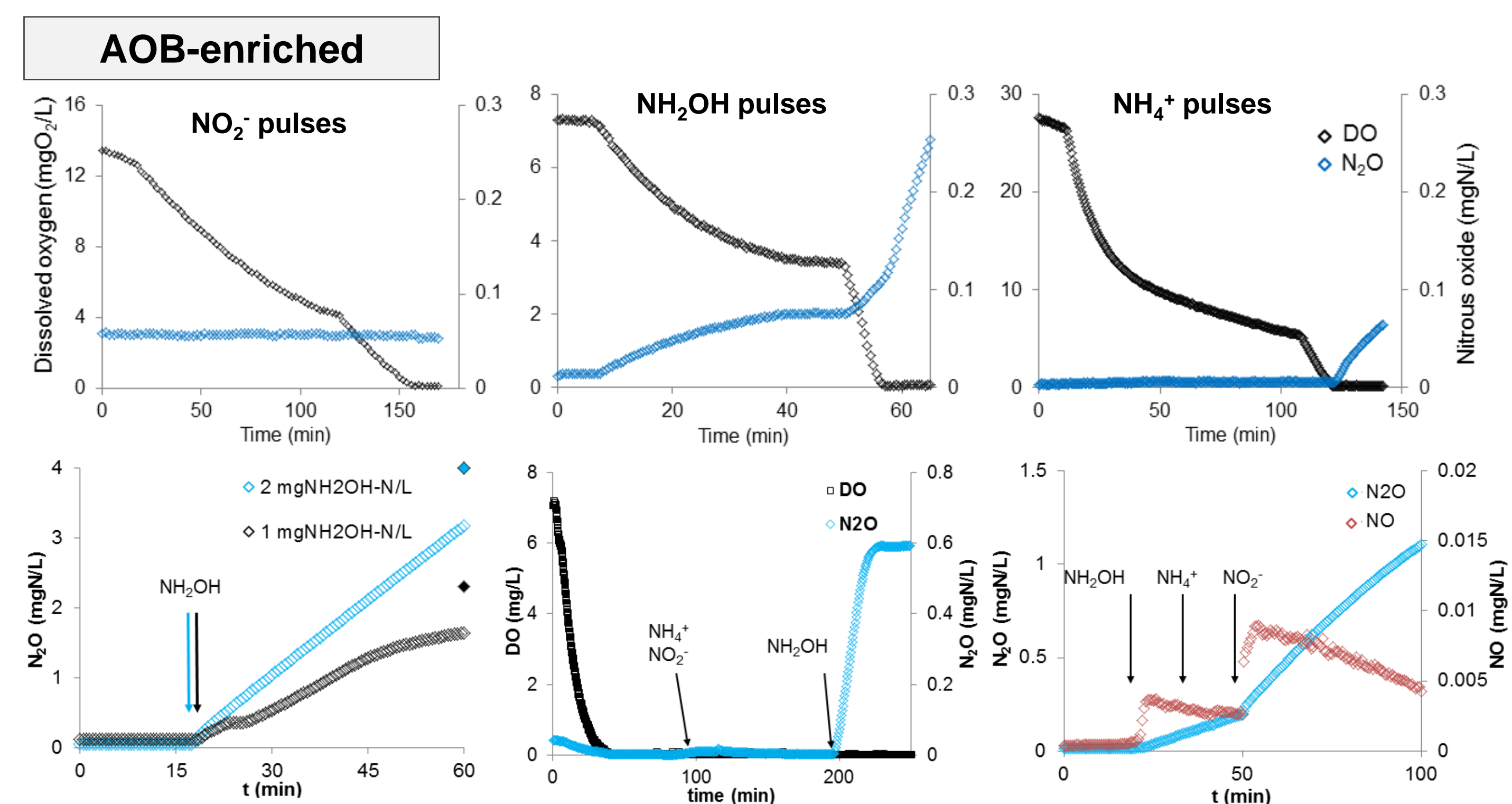
- Nitrous oxide (N₂O) emissions during nitrogen removal in wastewater treatment operations can compromise the environmental impact of new energy-saving technologies.
- Current process modelling efforts aim to reproduce N₂O experimental data with mathematical equations, structuring our understanding of the system.
- A mathematical model structure that describes N₂O production during biological nitrogen removal^[1] is calibrated for two biomasses representative of wastewater treatment operations: AOB-enriched and mixed culture.
- Extant respirometric assays are used to monitor N₂O, NO and DO dynamics.



Obj_1 → Quantify N₂O dynamics via extant respirometric assays from two biomasses: AOB-enriched and mixed liquor.

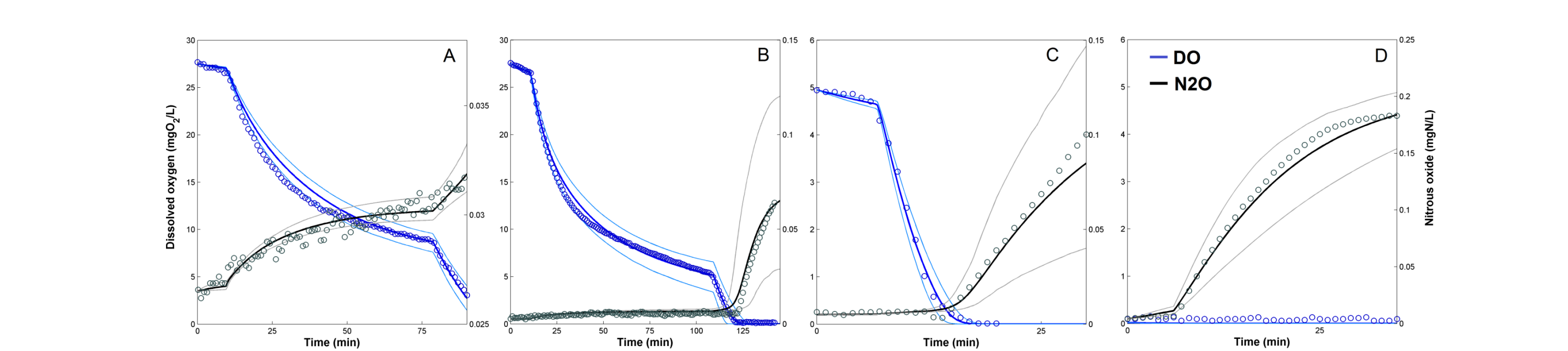
Obj_2 → Calibrate the NDHA model^[1] to describe N-removing processes and N₂O production for wastewater treatment operations.

Experimental and Modelling results



Top – Aerobic experiments. Dissolved oxygen and N₂O concentrations during NO₂⁻ (left), NH₂OH (middle) and NH₄⁺ (right) pulses (1-4 mgN/L).

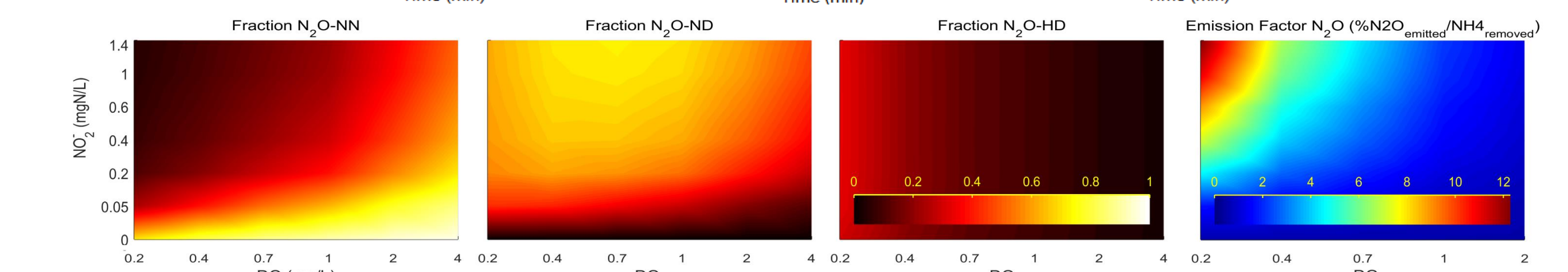
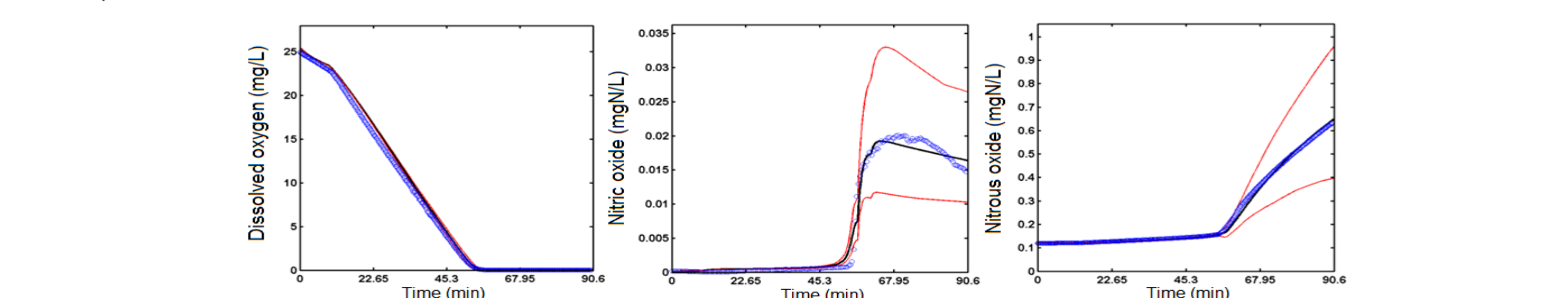
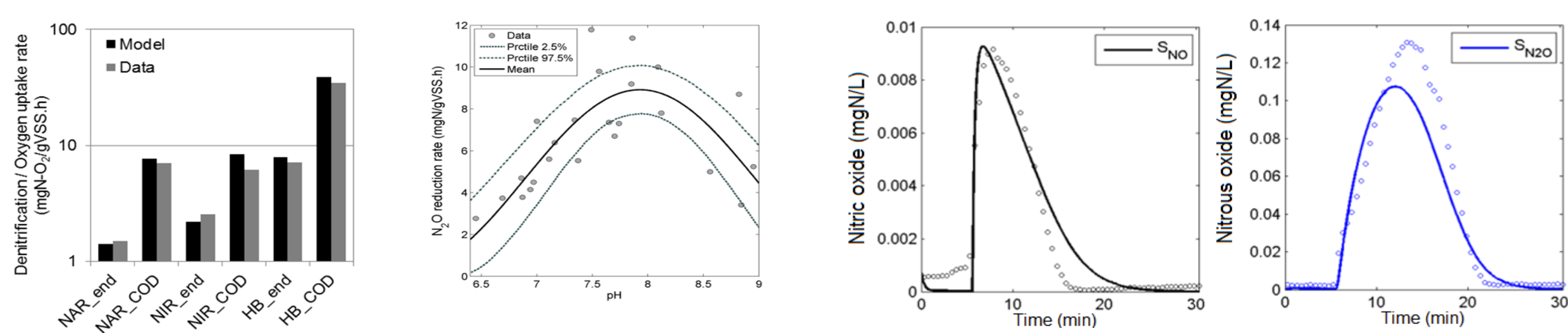
Bottom – Anoxic experiments. NH₂OH oxidation at excess NO₂⁻ (left), effect of NO₂⁻ and NH₄⁺ on N₂O production (middle), effect of NH₂OH and NO₂⁻ oxidation on N₂O and NO production (right).



Experimental and modelling results for DO and N₂O during NH₄⁺ oxidation at high DO (A), from high to low DO (B, C) and anoxic NH₂OH oxidation (D).^[2]

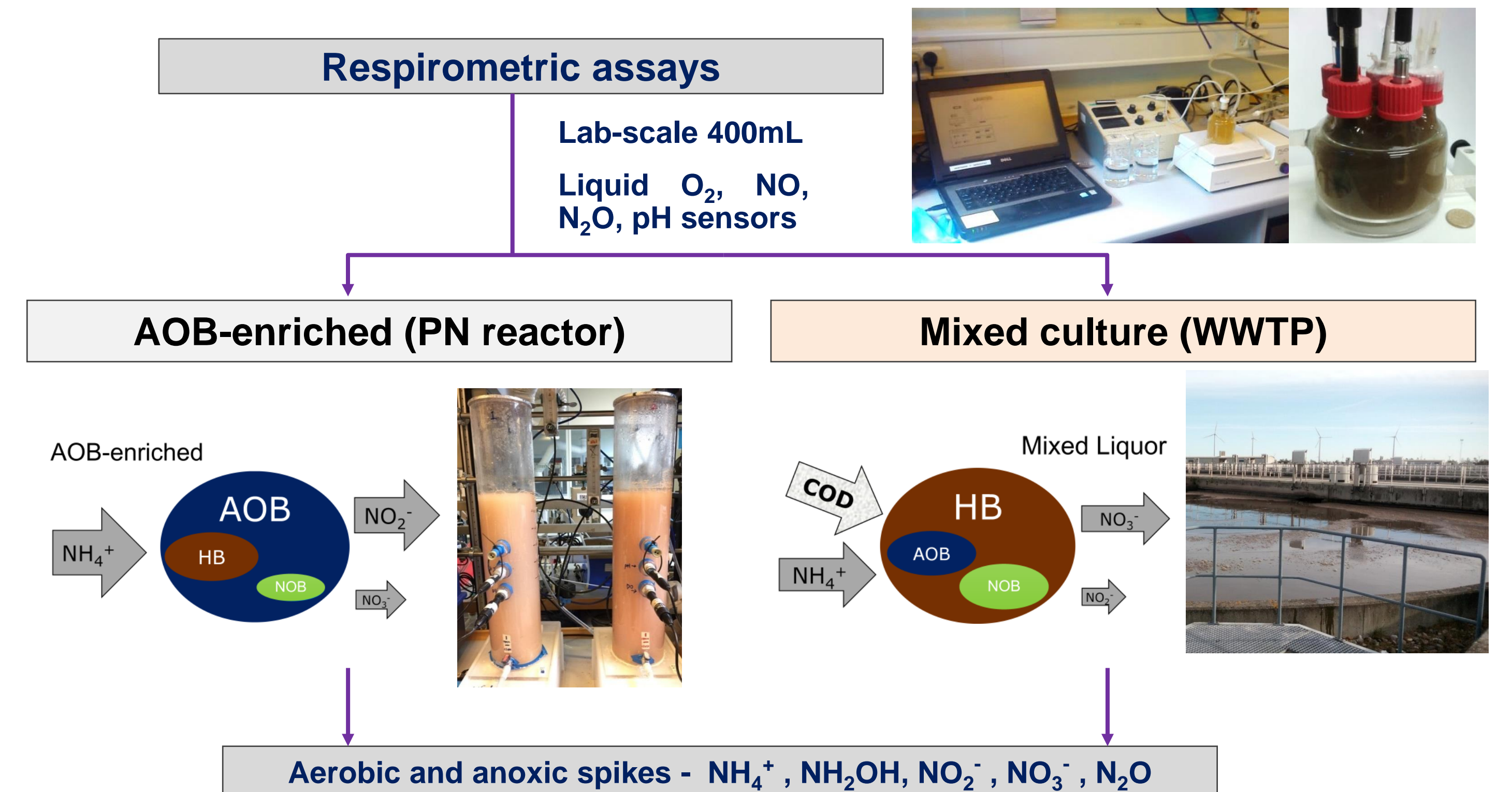
Mixed culture

The mixed culture biomass showed higher specific denitrification rates, N₂O consumption and NOB activity compared to the AOB-enriched.

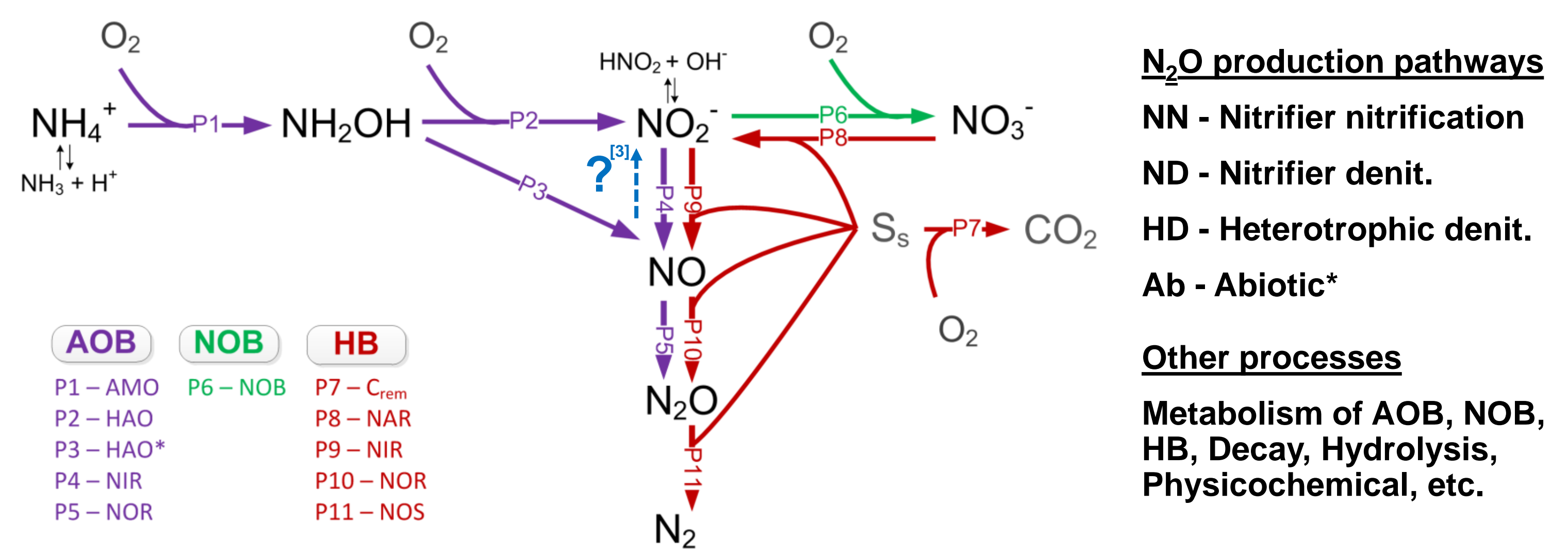


Top – Heterotrophic activity, Denitrification and aerobic carbon removal (left). Maximum N₂O consumption rate at different pH (middle). NO and N₂O dynamics after NO₂⁻ spike under electron donor limiting conditions (right). Middle – Aerobic NH₄⁺ oxidation. DO consumption (left), NO production (middle), N₂O production (right). Bottom – Model evaluation at varying DO and NO₂⁻ concentrations. From left to right: Pathway contributions to total N₂O pool NN, ND, HD; N₂O emission factor.

Experimental procedure and Model structure



The NDHA model^[1] comprehensively describes N₂O and NO producing pathways by both autotrophic ammonium oxidizing and heterotrophic bacteria:



- N₂O production pathways**
- NN - Nitrifier nitrification
 - ND - Nitrifier denit.
 - HD - Heterotrophic denit.
 - Ab - Abiotic*
- Other processes**
- Metabolism of AOB, NOB, HB, Decay, Hydrolysis, Physicochemical, etc.

Model calibration: AOB-enriched vs. Mixed culture

AOB-enriched biomass			Mixed culture biomass		
DO, NH ₄ ⁺ , NO ₂ ⁻ , NO ₃ ⁻ dynamics			DO, NH ₄ ⁺ , NO ₂ ⁻ , NO ₃ ⁻ dynamics		
Parameter	Unit	Value	Parameter	Unit	Value
μ _{AOB,AMO}	Maximum AMO-mediated reaction rate	d ⁻¹ 0.49 ± 0.01	μ _{AOB,AMO}	Maximum AMO-mediated reaction rate	d ⁻¹ 0.0031 ± 0.0001
μ _{NOB}	Maximum NOB growth	d ⁻¹ 0.67 ± 0.07	μ _{NOB}	Maximum NOB growth	d ⁻¹ 1.04 ± 0.05
K _H	Hydrolysis rate	d ⁻¹ 2.01 ± 0.02	μ _{HB}	Maximum HB growth rate	d ⁻¹ 5.15 ± 0.11
K _{AOB,NH3}	NH ₃ affinity for AOB	mgN/L 0.12 ± 0.005	K _{AOB,NH3}	NH ₃ affinity for AOB	mgN/L 0.007 ± 0.0012
K _{AOB,O2,AMO}	O ₂ AMO-mediated affinity constant	mgO ₂ /L 0.23 ± 0.02	K _{NOB,HNO2}	HNO ₂ affinity for NOB	μgN/L 0.027 ± 0.006
NO and N ₂ O dynamics			NO and N ₂ O dynamics		
Parameter	Unit	Value	Parameter	Unit	Value
ε _{AOB}	Reduction factor HAO-mediated reaction rate	(-) 0.48 ± 0.005 (x10 ⁻³)	ε _{AOB}	Reduction factor HAO-mediated reaction rate	(-) 0.0031 ± 0.0001
Γ _{NOR}	Reduction factor for NO reduction	(-) 0.16 ± 0.005	Γ _{NOR}	Reduction factor for NO reduction	(-) 0.36 ± 0.02
K _{AOB,NH2OH,NH2OH}	NH ₂ OH affinity for AOB during NO reduction	mgN/L 0.25 ± 0.005	Γ _{NIR}	Anoxic reduction factor for HNO ₂ reduction	(-) 0.22 ± 0.01
K _{AOB,HNO2}	HNO ₂ affinity for AOB	μgN/L 0.67 ± 0.03	pH _{opt,nosZ}	Optimum pH for N ₂ O-reduction	(-) 7.9 ± 0.1
			W _{nosZ}	Sinusoidal parameter for N ₂ O-reduction	(-) 2.2 ± 0.2
			K _{HB,N2O}	N ₂ O affinity constant for HB	mgN/L 0.078 ± 0.020

The calibrated model describes N₂O production from AOB-enriched^[2] and mixed culture biomass; a total of 10 and 17 parameters were accurately estimated respectively. Parameter sets for each biomass (maximum rates, substrate affinities) highlighted differences in microbial community composition:

- The estimated NH₃ affinity differed, probably due to the different NH₄⁺ and pH levels at which the biomasses operated: NH₄⁺_AOB-enriched >> NH₄⁺_Mixed-culture → K_{AOB,NH3} = 0.12 vs. 0.007 mgN/L respectively.
- The fractions of NH₄⁺ oxidized (NN pathway) and NO₂⁻ reduced to N₂O by AOB (ND pathway) also varied between systems → ε_{AOB} = 0.003 vs. 0.0005 (-).

A pH-dependent function to describe N₂O consumption is proposed (max pH = 8).

The NDHA model structure is in agreement with the newly proposed AOB metabolism of aerobic/anoxic NH₂OH oxidation^[3].

Conclusions

- A novel experimental design to calibrate N₂O models through extant respirometry is proposed that combines DO, N₂O and NO measurements.
- N₂O and NO production from mixed liquor biomass increased during NH₄⁺ oxidation at low DO concentrations and in the presence of NO₂⁻.
- The NDHA model response was validated and described N₂O production at varying DO, NH₄⁺ and NO₂⁻ concentrations.
- During NH₄⁺ oxidation the NN pathway showed the largest contribution at high DO levels, while the ND and HD pathways increased and dominated the total N₂O production at low DO and high NO₂⁻ concentrations.