

# **UNIVERSITI PUTRA MALAYSIA**

SYNTHESIS OF DIFFERENT TYPES OF CARBON NANOSTRUCTURE ON CARBON FIBER AND THEIR APPLICATION AS FILLERS IN POLYPROPYLENE COMPOSITE

**FERIAL GHAEMI** 

ITMA 2015 3



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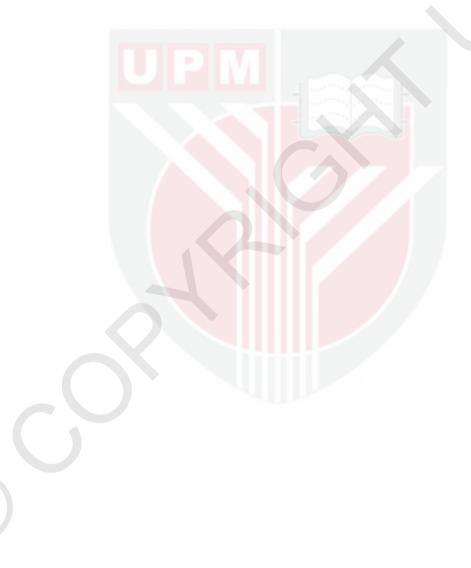
Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia, in fulfilment of the requirements for the Degree of Doctor of Philosophy

July 2015

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# DEDICATIONS

To Prophet Mohammad (*salavato Allah alayh*) and his respectable family



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfillment of the requirement for the degree of Doctor of Philosophy

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### FERIAL GHAEMI

**July 2015** 

Chair: Professor Robiah Yunus, Ph.D. Faculty: Institute of Advanced Technology

The potential application of carbon nanoparticles such as carbon nanotubes (CNT), carbon nanofibers (CNF) and graphene (G) flakes grown on carbon fiber (CF) surface as fillers in polypropylene composite is discussed. Carbon fiber surface must be modified before it can be used as fillers in composites. A one-step process using the chemical vapor deposition (CVD) method has been used to synthesize CNT, CNF and G and also G-CNF and G-CNT on the carbon fiber to modify its surfaces. In this study, CFs (Toho Tenax Co. Ltd.) was utilized as a substrate to grow carbon nanostructures and also as a filler in polypropylene pellets (PP 600G) polymer matrix. To the best of our knowledge, so far nobody has reported any work being widely carried out on synthesizing G layers on the CNF and CNT grown on CF by using a bimetallic catalyst (Ni/Cu) in a one-step CVD method in order to increase the CF surface area as well as to improve its properties. The synthesis of nanostructures on CF was accomplished using high purity acetylene (C<sub>2</sub>H<sub>2</sub>) as a carbon source, and nitrogen (Air Product, 99.9995) and hydrogen as carrier gases. Two types of catalysts namely copper nitrate trihydrate ( $Cu(NO_3)_2.3H_2O$ ) and nickel nitrate hexahydrate ( $Ni(NO_3)_2.6H_2O$ ) were utilized as bimetallic catalyst in the synthesis.

All the operating parameters of CVD process for growing the carbon nanostructures were optimized in order to obtain uniform and high quality carbon nanostructures. These parameters include catalyst concentration (from 50 mM to 150 mM), reaction temperature (different for each kind of carbon nanomaterial), reaction time (from 10 to 50 min) and carbon source flow rate (from 25 sccm to 100 sccm). Based on the SEM, TEM, TGA, BET surface area and Raman spectroscopy results, it was concluded that the optimum conditions are at 100mM catalyst concentration at 50sccm acetylene flow



rate for 30 min at 550°C, 800 °C and 1050 °C for CNF, CNT and G growth, respectively.

The changes in the surface characteristics of CFs was studied with scanning electron microscopy (SEM), transmission electron microscope (TEM), Raman spectroscopy and BET surface area analyzer. By analyzing SEM and TEM images, the morphology, structure, size and diameter of the resulting carbon nanostructures were obtained. In Raman spectra, the  $I_D/I_G$  ratio of the samples decreases when graphene flakes are present. When the ratio of CF/catalyst was at the maximum value, the  $I_D/I_G$  ratio ( $\approx 1.13$ ) coated with CNF, CNT and G, decreased to 0.94, 0.88, and 0.47 respectively. The  $I_D/I_G$  ratio of CF-CNF-G (0.85) and CF-CNT-G (0.81) indicates the effect of graphene growth on the crystallinity of the substrate.

Based on the results obtained from the surface modification of CF with various nanostructures, it is concluded that the carbon fiber coated with CNT-G (CF-CNT-G) with 80% yield and 46 m<sup>2</sup>/g BET surface area is the best method for surface modification of CF. Other fillers such as CF-CNF produced 24% yield and 2.31 m<sup>2</sup>/g surface area, CF-CNT with 46% yield and 5.22 m<sup>2</sup>/g surface area, CF-CNF-G with 56% yield and 21 m<sup>2</sup>/g surface area and, CF-G with 54% yield and 10.21 m<sup>2</sup>/g surface area.

Polypropylene (PP) composites with different carbon-based fillers such as G on CF (G-CF), CNF on CF (CNF-CF), CNT on CF (CNT-CF) and also G-CNF-CF and G-CNT-CF were prepared by the melt mixed method and the effects of these nanoparticles on the mechanical and thermal behavior of the composites were analyzed. The mechanical behavior and thermal resistance of the produced composites were evaluated using the tensile test and thermal gravimetric analysis (TGA), respectively. The Raman images were then used to explain the observed mechanical behavior of the different types of fillers/PP composites. The tensile stress and young's modulus of neat PP are 28MPa and 1400MPa. The values increased when various nanostructures were grown on the CF, to about 8.9% and 14.5% for CF/PP, 21% and 30.5% for CF-CNF/PP, 30.7% and 50% for CF-CNF-G/PP, 58.9% and 58% for CF-CNT/PP, 98.2% and 114.2% for CF-CNT-G and finally 82.8% and 97% for CF-G. Based on the results, the PP composite reinforced with CF-CNT-G showed the highest improvement in tensile stress at 55.5 MPa, young's modulus at 2998.9 MPa and enhancement in thermal stability to 130°C.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia Sebagai memenuhi keperluan untuk ijazah Doktor Falsafah

### SINTESIS PELBAGAI JENIS KARBON BERSTRUKTUR NANO PADA SERAT KARBON DAN APLIKASI SEBAGAI PENGISI DALAM KOMPOSIT POLIPROPILENA

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Potensi partikel nano karbon dalam aplikasi tiub nano karbon (CNT), serat nano karbon (CNF) dan serpihan grafin (G) yang dibina dan diperkembangkan di atas permukaan gentian karbon (CF), sebagai bahan pengukuh di dalam komposit polipropilena telah dibangunkan. Permukaan gentian karbon perlu diubah suai sebelum ia boleh digunakan sebagai bahan pengukuh di dalam bahan komposit. Proses satu langkah yang menggunakan kaedah pemendapan wap kimia (CVD) telah digunakan untuk mensintesis CNT, CNF dan G, berserta G-CNF dan G-CNT pada serat karbon untuk mengubah suai permukaannya. Dalam kajian ini, CF (Toho Tenax Co Ltd) telah digunakan sebagai substrat untuk menghasilkan struktur karbon nano dan juga sebagai bahan pengukuh di dalam matriks polimer bagi pelet polipropilena (PP 600g). Pada ketika ini, berdasarkan kajian yang telah dilakukan, masih tiada pihak yang melaporkan mengenai kerja-kerja mensintesis lapisan G di atas CNF dan CNT pada CF dengan menggunakan pemangkin dwi-logam (Ni/Cu) dengan hanya kaedah CVD satu langkah, bagi meningkatkan luas permukaan CF selain daripada menambah baik ciri-cirinya. Sintesis struktur nano pada CF telah berjaya dilaksanakan dengan menggunakan asetilena berketulenan tinggi (C<sub>2</sub>H<sub>2</sub>) sebagai sumber karbon, nitrogen (Air Product, 99,9995) dan hidrogen sebagai gas pembawa. Dua jenis pemangkin iaitu kuprum nitrat trihidrat (Cu (NO<sub>3</sub>)<sub>2</sub>.3H<sub>2</sub>O) dan nikel nitrat heksahidrat (Ni (NO<sub>3</sub>)<sub>2</sub>.6H<sub>2</sub>O) telah digunakan dalam sintesis tersebut.

Semua parameter yang terlibat dalam proses CVD untuk pembinaan struktur nano karbon dioptimumkan bagi mendapatkan struktur nano karbon yang berkualiti tinggi dan seragam. Parameter ini termasuk kepekatan pemangkin (dari 50mm hingga 150mm), suhu tindak balas (berbeza bagi setiap jenis bahan nano karbon), tempoh masa tindak balas (dari 10 hingga 50min) dan kadar aliran sumber karbon (dari 25sccm

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hingga 100sccm). Berdasarkan maklumat SEM, TEM, TGA, luas permukaan BET dan keputusan spektroskopi Raman, kesimpulan bagi keadaan optimum proses ini adalah pada kepekatan pemangkin 100mM, kadar aliran asetilena 50sccm selama 30 minit dengan suhu masing-masing 550°C, 800°C dan 1050°C untuk pertumbuhan CNF, CNT dan G.

Perubahan pada ciri-ciri permukaan CF telah dikaji dengan menggunakan mikroskopi imbasan elektron (SEM), mikroskopi transmisi elektron (TEM), spektroskopi Raman dan analisis luas permukaan BET. Melalui dua analisa imej SEM dan TEM, morfologi, struktur, saiz dan diameter struktur karbon nano telah berjaya diperolehi. Spektrum Raman menunjukkan nisbah  $I_D/I_G$  sampel berkurangan dengan kehadiran serpihan grafin. Apabila nisbah CF/pemangkin berada pada nilai maksimum, nisbah  $I_D/I_G$  ( $\approx 1.13$ ) yang disaluti dengan CNF, CNT dan G masing-masing menyusut kepada 0.94, 0.88, dan 0.4. Nisbah  $I_G/I_D$  CF-CNF-G (0.85) dan CF-CNT-G (0.81) menunjukkan kesan pertumbuhan grafin pada hablur substrat.

Berdasarkan keputusan yang diperolehi daripada pengubahsuaian permukaan CF terhadap pelbagai struktur nano, kajian mendapati bahawa 80% serat karbon yang disaluti dengan CNT-G (CF-CNT-G) yang telah terhasil dengan luas permukaan BET 46 m<sup>2</sup>/g adalah kaedah terbaik bagi pengubahsuaian permukaan CF. Bahan pengukuh lain seperti CF-CNF memberikan 24% hasil bahan dengan luas permukaan 2.31 m<sup>2</sup>/g, CF-CNT dengan 46% hasil dan luas permukaan 5.22 m<sup>2</sup>/g, CF-CNF-G dengan 56% hasil dan luas permukaan 21 m<sup>2</sup>/g, dan akhir sekali, CF-G dengan 54% hasil dan luas permukaan 10.21 m<sup>2</sup>/g.

Komposit polipropilena (PP) dengan bahan pengukuh berasaskan karbon yang berbeza seperti G pada CF(G-CF), CNF pada CF (CNF-CF), CNT pada CF (CNT-CF) juga G-CNF-CF dan G-CNT -CF telah dihasilkan melalui kaedah peleburan campuran, seterusnya kesan partikel nano ini terhadap ciri-ciri mekanikal dan terma bagi komposit telah dianalisa. Ciri-ciri mekanikal dan rintangan haba bagi komposit yang dihasilkan, masing-masing telah dinilai dengan menggunakan ujian tegangan dan analisis gravimetri haba (TGA). Imej-imej Raman kemudiannya digunakan untuk menerangkan tingkah laku mekanikal yang diperhatikan daripada jenis pengukuh/komposit PP. Kekuatan tegangan dan modulus Young untuk bahan asas PP adalah 28MPa dan 1400MPa. Nilai tersebut meningkat apabila pelbagai struktur nano telah dibina pada kombinasi bahan yang berbeza atas di atas CF, kepada 8.9% dan 14.5% bagi CF/PP, 21% dan 30.5% bagi CF-CNF/PP, 30.7% dan 50% bagi CF-CNF-G/PP, 58.9% dan 58% bagi CF-CNT/PP, 98.2% dan 114.2% bagi CF-CNT-G dan juga 82.8% dan 97% bagi CF-G. Keputusan telah membuktikan bahawa komposit PP yang diperkukuhkan dengan CF-CNT-G menunjukkan peningkatan tertinggi dalam kekuatan tegangan pada 55.5 MPa, modulus Young pada 2998.9 MPa dan peningkatan dalam kestabilan terma sehingga mencapai 130°C.

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I certify that a Thesis Examination Committee has met on 6 July 2015 to conduct the final examination of Ferial Ghaemi on his thesis entitled "Synthesis of Different Types of Carbon Nanostructure on Carbon Fiber and Their Application as Fillers in Polypropylene Composite" in accordance with the Universities and University Colleges Act 1971 and the Constitution of the Universiti Putra Malaysia [P.U.(A) 106] 15 March 1998. The Committee recommends that the student be awarded the Doctor of Philosophy.

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### LIST OF ABBREVIATIONS

- BET: Brunauer-Emmett-Teller
- CCCs: Carbon/Carbon composites

CCVD: Catalyst Chemical Vapor Deposition

CDE: Carbon Deposition Efficiency

CF: Carbon Fiber

CM: Carbon fiber Mass

CNF: Carbon Nanofiber

CNP: Carbon Nanoparticle

CNT: Carbon Nanotube

CVD: Chemical Vapor Deposition

EDX: Energy Dispersion X-ray

ESD: Electrostatic Discharge

FESEM: Field Emission Scanning Electron Microscopy

G: Graphene

IFSS: Interfacial Shear Strength

MM: Metal Mass

MWCNC: Multi-walled carbon nanocoils

MWCNT: Multi-walled carbon Nanotube

PECVD: Plasma Enhanced Chemical Vapor Deposition

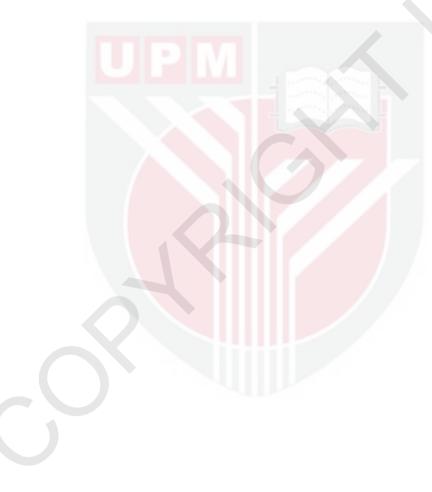
PP: Polypropylene

PM: Product Mass

**RBM:** Radial Breathing Mode

SA: Surface Area

Sccm: Standard Cubic Centimeter per minute SEM: Scanning Electron Microscopy SWCNT: Single-walled Carbon Nanotube TEM: Transmission Electron Microscopy TGA: Thermogravimetric Analysis



### **CHAPTER 1**

### **INTRODUCTION**

#### 1.1 General background

### 1.1.1 Polypropylene

Polypropylene (PP) is one of the most widely used polymers in automobiles, housewares, packaging, and electronics because of several useful properties like high heat distortion temperature, transparency, flame resistance and dimensional stability (An et al., 2012<sub>a</sub>). Since, PP has been used in many applications, it has been mixed with various nanofillers in order to improve its mechanical properties (Seo and Park, 2004).

### 1.1.2 Carbon Fiber

Carbon fiber is a material consisting of extremely thin fibers about 0.005–0.010 mm in diameter and composed mostly of carbon atoms. Carbon fiber (CF) has been widely used in various industry fields because of its high strength and low weight, and also its ability to be used as filler in polymer matrix to reinforce a composite (Zhang et al., 2009). A low portion of this filler in the polymer composite has revealed remarkable improvement of the thermal and mechanical properties (Rezaei et al., 2008; Aziz et al., 2014).

### 1.1.3 Polymer Composite

The area of hybrid fibre-reinforced polymeric composites has received considerable interest by the engineering community because of its unique structure and mechanical properties (Fu et al., 2009). CF reinforced polymeric composites have a wide range of unexplored potential applications in various technological areas such as aerospace, automobile, electronic and process industries due to their outstanding properties, such as high specific strength and stiffness, lower weight and flexible tailoring (Shazed et al., 2014).

### 1.1.4 Nanostructures and Polymer Nanocomposites

In addition, studies showed that when carbon nanostructure are grown on the carbon fiber surface, the mechanical interlocking between polymer matrix and carbon fiber was improved (Miranda et al., 2011). Recently, CF reinforced polymer composites incorporating carbon nanoparticles have attracted significant interest due to their extensive applications that conventional CF reinforced composites cannot offer (Yamamoto et al., 2009). Advanced polymer-based nanocomposites have been produced with improved properties such as electrical conductivity, mechanical, and thermal stability (Jia et al., 2011).

The properties of composite materials depend not only on the reinforcing fillers and polymer matrix but also on the interfacial adhesion between them. High interfacial adhesion provides the strong structure of composites with an effective load transfer from the polymer matrix to the fiber. Carbon nanomaterials, such as G and CNF, on CF have been shown to provide a robust network with a polymer matrix to enhance the interfacial properties of composites (Liang et al., 2013a). Today's, various types of carbon nanomaterials such as fullerenes, carbon nanofibers, carbon nanofibers, carbon nanofibers, carbon nanospirals, carbon onions (multilayer fullerenes) and graphene have been applied as reinforcing materials. Polymeric composites based on carbon nanomaterials, such as carbon nanofiber (CNF), carbon nanotube (CNT) and graphene (G) have attracted tremendous attention due to their excellent physical and mechanical properties (Kattab et al., 2012; Coleman et al., 2006; Ghaemi et al., 2014).

The CNFs can act as rod-shaped fillers and enhance the polymer properties in the polymer composite (Novais et al., 2012). There is broad range of different carbon nanofibre types, depending on the size and orientation of the graphene layers within their structure. The graphene plates are grown at an angle to the fiber axis in the herringbone shape to form carbon nanofiber, and tubular graphene walls are parallel to the fiber axis in the carbon nanotube (Teo et al., 2003). However all types of nanofibres are used in scientific studies and commercially, with applications including fillers in composites.

CNTs are allotropes of carbon with exceptional mechanical, electrical and thermal properties. They are made by graphene sheets, where carbon atoms are arranged in hexagonal patterns. Since their discovery by Iijima in 1991 (Iijima, 1991), CNTs have become attractive candidates for fundamental engineering applications. Graphene with a two-dimensional structure and honeycomb lattice is the most stable carbon format standard conditions, which was discovered by Novoselov et al. (2004). This nanomaterial has the potential to be applied in both scientific research and industrial applications because of its remarkable characteristics in terms of the mechanical, thermal and electrical properties (Zhang et al., 2013b). Additionally, G is also known as outstanding reinforcing filler in a composite with good dispersion (Allen et al., 2009; Cai and Song, 2010; Kim et al., 2010; Kuilla et al., 2010; Sengupta et al., 2011;).

C

The incorporation of CNTs into polymer matrix was firstly reported by Ajayan and coauthors (1994). Since then, polymer composites of carbon nanoparticles have been studied in various composite research fields, mostly focusing on their mechanical applications (Coleman et al., 2006; Kumar et al., 2002; Shafer and Windle, 1999; Yu et al., 2006). Strong interaction between these carbon nanomaterials and the host polymer is the key for mechanical strength. The bulk mechanical strength and stiffness of such composites is directly dependent on the interface of polymer matrix with carbon

nanoparticles. The elongated cylindrical forms of both CNTs and CNFs result in anomalously large interface area per particle (Yu et al., 2000,). In recent years, polymer/G nanocomposites have drawn more and more attention from both scientists and engineers due to their unique behaviors (Kim et al., 2010).

The advantages of hybrid systems include the combined nanomaterials properties of each component, more design choices, cost effectiveness, and superior performance (Kalaprasad et al., 2004). But the hybrid concept has rarely been considered for nanocomposites (Li et al., 2008a). By varying the reinforcement scale, it may be possible to tailor the mechanical and physical properties of the composites. The hybrids of CNFs and CNTs with graphene may avoid the aggregates of single nanofillers due to the high concentration (Sui et al., 2009).

#### 1.1.5 Methods of carbon nanomaterials production

Although there are several methods to obtain carbon nanomaterials, such as arcdischarge (Iijima, 1991), laser ablation (Yudasaka et al., 1998), chemical vapor deposition (CVD) (Cao et al., 2003; Gu et al., 2010), self assembly (Wu et al., 2007), mechanical exfoliation and cleavage (Novoselov et al., 2004), chemical methods (Choucair et al., 2009) and unzipping CNT (Kosynkin et al., 2009) those based on chemical solution are stressed by its practical approach for scale up the production. To provide large-scale production with low dimension and high purity is an important issue in advanced nanomaterials research, the CVD technique has been investigated (Muñoz and Gómez-Aleixandre, 2013) and applied in the production of carbon nanoparticles.

CVD method is considered as the most effective method and has been applied to grow carbon nanoparticles by many researchers (Liu et al., 2010). Basically, the deposition of reactants on the surface of the catalyst involves a balance between diffusion of reactant molecules to the catalyst and the kinetics of deposition reaction. The deposition reaction, which takes place on the catalyst surface is composed of various steps, such as adsorption of the products from the active sites. From a heat transfer point of view, the particles perform an extra function by carrying the heat, which is very important during carbon nanoparticles synthesis.

It is well known that the CVD reaction conversion depends on the extent of heat and mass transfer occurring within and between the gas-solid phases. Exothermic CVD reaction results in hot spots that frequently lead to catalyst deactivation and adverse changes in conversion and/or selectivity of the process. To achieve different structure and morphology, some critical parameters of CVD such as growth time, growth temperature, flow rate of carbon source gas and catalyst concentration can be varied (Thostenson et al., 2002; Zhu et al., 2003; Ghaemi et al., 2015). Synthesis of each carbon nanoparticle with CVD method needs to have different conditions. So for

growing two 3D carbon nanoparticles (carbon nanofiber and carbon nanotubes and then graphene), different conditions are needed.

### **1.2 Problem statement**

Carbon fiber (CF)-reinforced polymer composites have been extensively applied in the areas of aerospace, aircraft, rocket, sport and military industries due to the superior strength-to-weight, stiffness-to-weight ratio and high service temperature. One problem in these composites is that CF typically shows poor interfacial interactions with polymer matrix. Another critical drawback is the presence of matrix-rich and free-volume regions formed in the gaps between the interlaced fiber bundles. The interfacial and mechanical properties of a fiber-reinforced polymer composite are significantly influenced by interfacial characteristics between the reinforcing fibers and the polymer matrix. For improving the mechanical properties of composite material it is necessary to optimize the interface between the fiber and matrix using certain methods for modification of reinforcing fiber (Dobreva and Nenkova 2006).

Using carbon fibers as reinforcing filler, the surface area is an important factor that plays a contributing role in the interfacial behavior with the polymer matrix. The defective flow of the polymer matrix around neat carbon fiber causes decreased interfacial properties and easily pulls out of carbon fiber from the matrix. Appropriate surface-treatment may modify the fiber surface by increasing the surface area and/or by growing carbon nanoparticles on the CF surface that may provide high adhesion between the fibers and the polymer matrix (Miller et al., 2001). So one of the key points of their properties is the interface between the fibers and matrix, which can be modified by nanoparticles growth.

Recently it was suggested that carbon nanoparticles grafting onto carbon fibers might be a new method of improving the interface and increasing IFSS (Zhao et al., 2008). There are two main routes for adding carbon nano-filler into conventional fiberreinforced polymeric materials. The first one is by dispersing carbon nano-fillers entirely throughout the polymeric matrix, which afterwards is layered with reinforcing fibers. The second route concerns the direct attachment of nano-fillers onto primary reinforcing fibers (Miranda et al., 2011). In this study, G with excellent properties was synthsized to not only modify the CF surface but also use as a filler to reinforce polymer composite. Besides, the potential of CNT, CNF and G flakes on carbon fiber and also the effects of the graphene growth in the CNF-G and CNT-G grown on carbon fiber (CF) surface as fillers in composite materials are explored and compared with together. In order to modify the CF surface, a one-step process using the chemical vapor deposition method, has been used to synthesize CNT, CNF and G and also G-CNF and G-CNT on carbon fibers surface.

The high cost of carbon nanoparticles (CNF, CNT and G) synthesis, which restricts its large-scale production, is the key factor to the development and commercialization of

CNF, CNT and G sheet related industrials. It is prudent to note that the carbon nanoparticles are currently produced with a relatively high quality but in limited quantities. Therefore, future use of this worthwhile material strongly depends on the development of the technology for its large-scale production. Accordingly, large-scale syntheses of CNF, CNT and G have been the subject of intensive researches, and many attempts have been explored to optimize and control the carbon nanoparticles growth.

Chemical vapor deposition (CVD) provides a large contact area among the reactant and particle catalyst. The catalyst solution has a critical role to synthesize the CNT, CNF and the G layers on CF. Besides; to synthesize high quality graphene flakes on CNF or CNT, which were grown on CF, in one step, selecting a proper catalyst is important part of the CVD method. The selected catalyst should have the potential to grow CNF, CNF and G on CF and also G flakes on CF-CNF or CF-CNT in a large-area and a few-layer garphene. Among the different catalysts, Fe, Ni and Co are found suitable to synthesize CNF and CNT by CVD. On the other hand, for graphene growth Ni, Co and Cu have been used as the effective catalysts. Therefore, it is particularly important to develop a scalable synthesis method such as CVD method that could effectively control the size of fibers and tubes of CNF and CNT as well as the size and number of graphene layers, to enable large-scale production of the carbon nanoparticles (Liu et al., 2013). Consequently, to synthesis two layers of the carbon nanoparticles on the CF surface, using proper catalyst (bimetallic) with optimum amount should be found.

### 1.3 Objectives

In this study four objectives have been identified as follows:

- 1. To optimize the condition for CF surface modification by growing different carbon nanomaterials (CNF, CNT and G) in the vertical fixed bed CVD reactor
- 2. To develop one-step method by use of bimetallic catalyst to synthesis two layers of carbon nanomaterials (CNF-G and CNT-G) on CF surface.
- 3. To prepare polymer nanocomposite by mixing different fillers with polypropylene matrix and analysis the mechanical and thermal properties.

#### 1.4 Thesis Scope

In this thesis work, we studied the synthesis and compared different kind of carbon nanoparticles grown on CF in order to increase the surface area of the CF as well as improve its properties. To synthesize high quality graphene, CVD on Cu catalyst is considered as one of the most promising methods because of its fabrication in a largearea and a single-layer graphene. However, Ni is one of the most widely studied catalysts for the synthesis of carbon nanaofibers and also graphene because a strong Ni–C interaction causes a repulsive interaction within the C–C interaction and causes the dissolution at the edge of a graphene. However, limited research has been devoted to the usage of a bimetallic catalyst (Ni/Cu) to synthesize CNF, CNT and G. The Cu/Ni alloy is an excellent bimetallic system to control carbon solubility by tuning the atomic fraction of Ni in Cu in order to synthesis two layers of carbon nanostructures (CNF-G and CNT-G) in one-step production. To grow small size of CNF/CNT and few layer of graphene with high quality, the parameters of CVD method including catalyst concentration, reaction temperature, reaction time and acetylene flow rate should be optimized. These parameters include temperature, run time, catalyst amount and ratio of Ni/Cu, and flow rate of carbon source.

For growing carbon nanofiber on the CF, the operating parameters varied were as follows: Temperature, 550-650°C, time: 10-60 min, catalyst amount: 50mM-150mM, catalyst ratio of Ni/Cu: 0/100, 30/70. 50/50, 70/30, 100/0 and carbon source flow rate: 25-150sccm. For growing carbon nanotubes the parameters varied were as follows: Temperature: 700-850°C, time: 10-60 min, catalyst amount: 50mM-150mM, catalyst ratio of Ni/Cu: 0/100, 30/70. 50/50, 70/30, 100/0 and carbon source flow rate: 25-150sccm. For growing graphene the operating parameters varied were as follows: Temperature: 900-1050°C, time: 10-60 min, catalyst amount: 50mM-150mM, catalyst ratio of Ni/Cu: 0/100, 30/70. 50/50, 70/30, 100/0 and carbon source flow rate: 25-150sccm. For growing graphene the operating parameters varied were as follows: Temperature: 900-1050°C, time: 10-60 min, catalyst amount: 50mM-150mM, catalyst ratio of Ni/Cu: 0/100, 30/70. 50/50, 70/30, 100/0 and carbon source flow rate: 25-150sccm.

Finally, the produced CNF, CNT, G, CNF-G and CNT-G, on CF surface, were incorporated into a polypropylene to fabricate the different composites. Furthermore, the effects of the carbon nanoparticles as coated phases on the CF surface were investigated in terms of the mechanical and thermal properties of the PP composite. Therefore, a tensile test as well as thermal gravimetric analysis (TGA) was applied. The surface morphology and structural characterization of the samples were analyzed through scanning electron microscopy (SEM), transmission electron microscope (TEM) and Raman spectroscopy.

#### 1.5 Thesis Layout

This thesis is organized into five chapters. Following this chapter, chapter two begins with an extensive literature review. Firstly, it begins with the introduction of the carbon fiber and carbon nanoparticles. This chapter also provides description of some common characterization techniques for different carbon nanoparticles employed in this research. Secondly, the chapter covers the CVD process that is commonly used for CNF, CNT and G synthesis on CF. The growth mechanisms of each carbon nanoparticles are presented, including descriptions of the roles of the all effected parameters namely temperature, time, carbon amount and ratio and flow rate of carbon source gas. Then discussion is shifted towards synthesis of CNF-G and CNT-G on CF including principles and achievements obtained hitherto.



The last part of the literature review section is dedicated to a brief description and explanation of the carbon nanoparticles potential application polymer matrix and evaluate the mechanical and thermal properties. The considered polymer in this research is polypropylene, which was the preferred reinforcing polymer since it is a member of the group of commodity thermoplastics synthesized in large quantities and not very responsive to chemical stress cracking. This part presents related information about PP and its properties, mechanical reinforcement as well as thermal resistance. Moreover, the result of researchers who work with PP and add nanofiller in its matrix was studied.

After a comprehensive literature review with respect to the dissertation topic, in Chapter 2, research methodology and obtained results and discussions are presented in the following Chapters, 3 and 4. Research methodology was designed and conducted according to the objectives of this dissertation. Chapter 4 begins with results from synthesis of Carbon nanofibers on CF and optimize the catalyst concentration, growth time and temperature and also hydrocarbon flow rate. After that, CNT and G synthesis and optimize their growth parameters. Besides, the CNF-G and the CNT-G on CF are produced and analyzed their structures and characterizations. Moreover, the resulting nanoparticles are used as fillers in polymer matrix and analysis the thermal and mechanical behavior of the different composites. Finally, chapter 5 conducts with conclusion and recommendation.

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