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Microstructure, microhardness and thermal expansion of CNT/Al composites prepared by flake powder metallurgy

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ABSTRACT

Carbon nanotube/aluminum (CNT/Al) composites are prepared by a combination of flake powder metallurgy and hot-isostatic-pressing. The specimens are investigated by several techniques including Raman spectroscopy, optical microscopy, scanning- and transmission electron microscopy. The composites show a layered-microstructure with a stacking of CNT/Al flakes with a CNT-rich layer between two flakes. The individual Al grains forming the flakes are about 500 nm in size. The CNTs are well dispersed within a flake and they bridge the micro-cracks. The results reveal that the coefficient of thermal expansion (CTE) decreases markedly upon the increase in carbon content, reaching 15.4×10^{-6} K⁻¹ for the specimen with a carbon content of 2.0 wt% (2.9 vol%), i.e. a 30% decrease compared to the CTE of pure Al. This could arise from the layered-microstructure resulting from the utilization of Al flakes as opposed to rounded particles.

Keywords: A. Carbon nanotubes A. Metal matrix composite B. Thermal properties E. Powder processing

1. Introduction

Silicon, the material at the basis of the semiconductor devices, shows a low coefficient of thermal expansion (CTE) (5×10^{-6}) K^{-1}) in comparison with the common packaging materials like copper $(17 \times 10^{-6} \text{ K}^{-1})$ and aluminum $(21-26 \times 10^{-6} \text{ K}^{-1})$ [1,2]. The CTE difference will cause distortions at the interface upon repeated changes in temperature, which ultimately will lead to failure of the device. Aluminum-matrix composites with a lower CTE than pure Al have been prepared by adding some low CTE materials such as Si, SiC, AlN and diamond [2,3]. Carbon nanotubes (CNTs) have been become an attractive additive material for reducing the CTE of the aluminum-matrix composites [4-13] because of a very low or even negative CTE, in the range -2×10^{-5} – 0.5 \times 10^{-5} K⁻¹ depending on the CNT characteristics [14]. The end results however depend on many materials- or processparameters, including the precise nature of the matrix (pure or alloyed Al), of its grain size (nano-, micrometric), of the carbon content and of the kind of CNTs (single-wall, double-walled, multi-walled) and of the consolidation route (hot-pressing, hot extrusion...) and atmosphere (vacuum, N₂, Ar) (Table 1). CTE as

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low as $10 \times 10^{-6} \text{ K}^{-1}$ have been reported when using nanometric-sized Al with either single-wall CNTs [7] or multiwalled CNTs (MWCNTs) [12]. Many techniques have been developed to prepare CNT/Al composite powders with a uniform dispersion of the CNTs [4,5,13,15,16]. However, high-energy ball-milling [13] tends to damage the CNTs whereas molecular-level mixing [15] and *in situ* synthesis of CNTs in metallic powders [16] may lead to oxide impurities. Besides these techniques, the so-called polyester binder-assisted (PBA) mixing technique has been reported for dispersing CNTs in a metallic powder, without damaging them, with the support of polyesters such as polyvinyl alcohol (PVA), natural rubber and ethylene glycol [17-23]. Moreover, it seems that using Al in the form of flakes, as opposed to isotropic grains, to prepare CNT/Al composites, is beneficial, at least to increase tensile strength without losing too much plasticity [19,20,24–26]. The aim of this study is to investigate the CTE of MWCNT/Al composites prepared by a combination of flake powder metallurgy, PBA mixing technique and hot-isostatic-pressing.

2. Experimental

2.1. Composite preparation

A commercial Al powder (Hunan Jinhao Aluminum Industrial Co., Ltd., 99.5%, average diameter 24 $\mu m)$ was selected for the

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Table 1

Ref.	Consolidation method	Sample	C _n (wt%)	C _v (vol%)	ρ(%)	H (HV)	CTE ($\times~10^{-6}~\text{K}^{-1}$
[39]	SPS	Al	0	-	-	45	-
		MWCNT/Al	0.5	-	-	50	-
[40]	SPS	MWCNT/Al	2	-	99	88	-
[34]	SPS	MWCNT/Al	1	-	-	44	-
			3	-	-	55	-
			5		-	54.5	-
[18]	SPS	MWCNT/Al	2		98	52	-
	+ Hot extrusion						
[32,35]	Vacuum sintering	Al	0		-	34	-
	-	MWCNT/Al	0.75		-	50	
[36]	Hot-extrusion	Al	0		99	39.4	-
		MWCNT/Al	2.5		99	84.5	-
			5		99	95.2	-
[37]	Hot-extrusion	MWCNT/Al	6		97	151	-
[6]	Hot-pressing	Al	0		-	-	26.1
		MWCNT/Al	4		-	-	23.2
		SWCNT/Al	3		-	-	20.4
[7]	Vacuum sintering	n-Al		0	-	-	26.2
	C	SWCNT/n-Al		10	-	-	14.8
		,		15	-	-	9.8
[8]	Hot-extrusion	2024 Al	0		-	_	26.3
		MWCNT/2024Al	1				22.5
[9]	Hot-pressing	2009A1		0	-	_	23.6
		MWCNT/2009Al		1.5	-	_	21.3
		MWCNT/2009Al		4.5	-	-	17.5
[11]	Hot-pressing + Vacuum sintering	MWCNT/2024Al		3	-	-	19.2
		MWCNT/2024Al		5	-	-	17.9
[12]	Sintering in N ₂	n-Al	0	-	90	-	80.1
		MWCNT/n-Al	1		92	-	54.4
		MWCNT/n-Al	3		94	-	10.5
		MWCNT/n-Al	5		92	-	23.2
[13]	Sintering in Air + Hot-extrusion	Al	0		-	64	26.0
		MWCNT/Al	1.5		-	81	24.8
		MWCNT/Al	2.5		-	95	24.0
		MWCNT/Al	3.5		-	115	22.5
		MWCNT/Al	4.5		-	130	21.5
This work	HIP	SO	0	0	98	44	22.0
		S0.5	0.5	0.7	96	55	19.5
		S1	1.0	1.5	96	69	18.0
		S1.5	1.5	2.2	95	83	16.4
		S2	2.0	2.9	94	62	15.4

study. Commercial carboxyl-functionalized MWCNTs (Chengdu Organic Chemicals Co. Ltd.) were used. The details of the functionalization process are not known to the authors. The carbon content in the MWCNT specimen is equal to 95 wt%, the balance probably corresponding to some residual metal catalysts. The key characteristics of the MWCNTs (number of walls, outer and inner diameters, length and presence of defects) are determined later in the paper. MWCNT/Al composite powders with a carbon content (C_n) equal to 0, 0.5, 1, 1.5 and 2 wt% were prepared by a PBA mixing route. They will be noted as PO, P0.5, P1, P1.5 and P2 hereafter. First, the Al powder was ball-milled (200 rpm, 2 h, ball-to-powder weight ratio of 10:1, N₂ atmosphere). The so-obtained Al flakes were slowly added to ethylene glycol under magnetic stirring (400 rpm, 2 h), forming the Al slurry. The appropriate amount of MWCNTs was dispersed in ethanol as reported earlier [27,28] and the soobtained suspension was mixed with the Al slurry under magnetic stirring (400 rpm, 2-3 h) at 80 °C in order to evaporate ethanol. Finally, the resulting slurry was ball-milled (150 rpm, 2 h, ballto-powder weight ratio of 10:1, N₂ atmosphere) and heated at 220 °C for 24 h in vacuum (residual pressure 200 MPa) to remove ethylene glycol. The MWCNT/Al powders were consolidated by cold uniaxial compaction (200 MPa, 5 s) followed by capsule-free hot-isostatic-pressing (HIP, AIP6-30H, Isostatic Press Inc's, US). The specimens were heated (10 °C/min) up to 620 °C, applying a 1 h dwell at this temperature for the pressing process (100 MPa). A natural cooling down to room temperature was performed. The sintered specimens, in the form of pellets 10 mm in diameter and about 5 mm thick, were polished down to 1 μ m using diamond slurries. The sintered specimens are noted S0, S0.5, S1, S1.5 and S2 hereafter.

2.2. Characterization

The MWCNTs were observed by high-resolution transmission electron microscopy (HRTEM, JEOL JEM 2100F operated at 200 kV). Their length was evaluated from field-emission-gun scanning electron microscopy images (FESEM, Hitachi S-4800 operated at 5 kV). The Raman spectra of the CNTs, powders and sintered specimens were recorded with a confocal RAMAN Microscope (Labram HR 800 Jobin Yvon) using 632 nm laser excitation. For each specimen, the spectra were averaged from five areas. The pellet density was measured by Archimedes method. X-ray diffraction (XRD) patterns of powders and sintered specimens were recorded using a Rigaku Rint Ultima diffractometer with Cu K_{α} radiation. The Al grain size in sintered specimens was determined by optical microscopy (3D KEYENCE VHX-1000) on surfaces chemically etched by a weak reagent (mixture of $0.25 \text{ mol } L^{-1} \text{ KMnO}_4$ and $0.25 \text{ mol } L^{-1}$ NaOH) at room temperature for 6 s. The Al powders and flakes as well as the sintered specimens were observed by FESEM (JEOL JSM 6700F operated at 5 kV and Hitachi S-4800 operated at 5

kV). For the sintered specimens, a preliminary step of chemical etching (mixture of 0.5 mol L⁻¹ NaOH and 0.4 mol L⁻¹ Na₂CO₃, 30 s) was performed [27]. Indentation tests (50 g for 10 s in air at room temperature) on polished surfaces were performed with a Vickers indenter (Shimadzu HMV 2000). The corresponding diagonals of the indentation were measured using an optical microscope attached to the indenter. The calculated microhardness values are the average of five measurements. The CTE (DIL 402 PC dilatometer) was measured along the length of parallelepipedic specimens (6 mm \times 4 mm \times 5 mm). The measurements were performed from 50 to 250 °C (heating-rate 5 °C min⁻¹).

3. Results and discussion

3.1. Microstructure

FESEM images of the as-received Al powder show that it is made up of rounded particles with an average size around 24 μ m (Fig. 1a, b). The Al flakes obtained after ball-milling have an average length equal to about 60 μ m with a thickness of about 1.5 μ m (Fig. 1c, d). FESEM images of the MWCNTs reveal that their length is in the range 1–3 μ m (Fig. 2). There are some structural defects along the length, such as etched positions (Fig. 3a) and

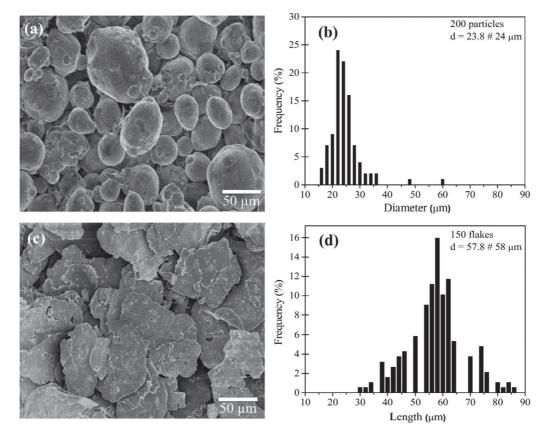


Fig. 1. FESEM images and grain size distributions of (a, b) the as-received Al powder and (c, d) after ball-milling resulting in Al flakes formation.

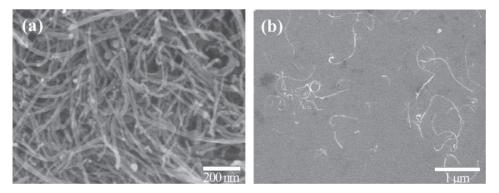


Fig. 2. FESEM images of MWCNTs (a) as-received and (b) dispersed in water then dropletted on a Si substrate.

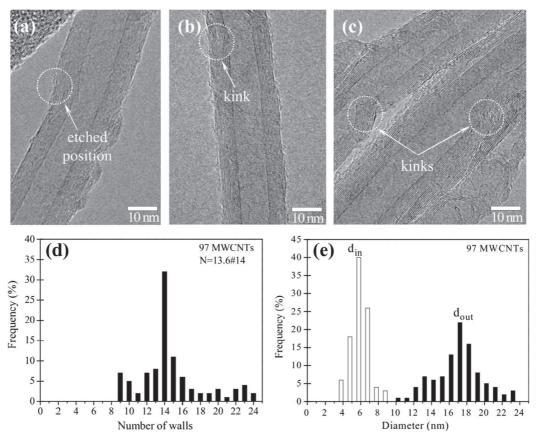


Fig. 3. (a-c) HRTEM images of the MWCNTs and distributions of (d) the number of walls and (e) the inner and outer diameters.

kinks (Fig. 3b, c), which are fairly typical of as-grown MWCNTs but could partly result from the functionalization process. The number of walls (Fig. 3d) and the outer and inner diameters (Fig. 3e) were measured for about 100 CNTs on HRTEM images. The average number of walls (N) is equal to 13.6, which is rounded to 14. The average outer and inner diameters are equal to 17.4 nm and 5.8 nm. These values are in excellent agreement with the empirical law giving the correlation between diameter and N for MWCNTs prepared by catalytic chemical vapor deposition [29]. The theoretical density of the CNTs (1.85 g cm⁻³) was calculated from these data using the CNT density chart [30].

Low-magnification FESEM images of the MWCNT/Al powders (left panel in Fig. 4) show the Al flakes and do not reveal MWCNT agglomerate or cluster, let alone individual MWCNTs. Higher-magnification images (right panel in Fig. 4) show individual MWCNTs well distributed on the Al flakes for the P0.5, P1 and P1.5 powders (Fig. 4b, d, f). For powder P2 (Fig. 4h), a few MWCNT clusters are observed laying on the Al flake surfaces, along with individual MWCNTs. The microstructure of the composite powders is similar to that reported for milling times lower than 3 h [31].

The Raman spectra (Fig. 5) of the MWCNTs, powder P1 and sintered specimen S1 reveal that I_D/I_G , the ratio between the intensities of the D band (*ca.* 1320 cm⁻¹) and the G band (*ca.* 1580 cm⁻¹) is equal to 0.88, 0.89 and 0.89, respectively. An increased I_D/I_G corresponds to an increased proportion of disordered carbon, which is generally attributed to the presence of more structural defects along the CNT walls. The almost identical values observed here show that the mixing process and the HIP process did not dramatically damage the MWCNTs.

Analysis of the XRD patterns (Fig. 6) of the S specimens reveals only the Al diffraction peaks. Aluminum oxides and carbides are not detected. By contrast, Al₄C₃ was detected by HRTEM in specimens prepared using spark plasma sintering and vacuum sintering techniques at temperatures higher than 600 °C [18,32]. Al₄C₃ formation results from the interfacial reaction between the Al matrix and the carbon atoms from amorphous carbon or the defective region of CNTs [18,33]. Moreover, partially damaged CNTs may also act as carbon source for the formation of Al_4C_3 [32]. Liu et al. [33] showed using XRD and HRTEM that Al₄C₃ in CNT/Al composites is observed when the powders are ball-milled for over 4 h and is not detected for shorter milling times. Therefore, we propose that our milling and HIP experimental conditions are mild enough in order that no detectable Al_4C_3 is formed. It is possible that it is below the detection limit in our XRD experimental conditions. Thus, local formation is not ruled out and future characterization by HRTEM could shed light on that matter. The relative density (ρ - Table 1), calculated using 1.85 and 2.7 g cm⁻³ for the MWCNTs and Al, respectively, decreases upon the increase in carbon content, from 98% for S0 to 94% for S2, which reflects that the presence of the MWCNT inhibits composite densification in agreement with several works [18,32-37].

The specimens have been investigated by optical microscopy, considering three concentric areas (noted as Areas 1, 2 and 3) from center to periphery, respectively (Fig. 7). Optical microscopy images of the surface (Area 1) of polished and chemically etched specimens are presented in Fig. 8. For S0 (Fig. 8a), the Al grain size, or more precisely the flake diameter distribution, is large (10–100 μ m) and the average flake diameter is equal to 22 μ m (Fig. 8b). This

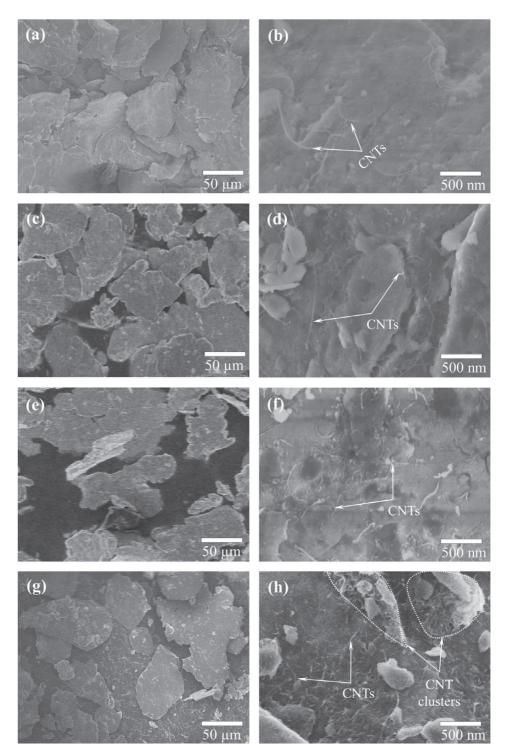


Fig. 4. FESEM images of the MWCNT/Al powders (a-b) P0.5, (c-d) P1, (e-f) P1.5 and (g-h) P2.

is about half the value found for the flakes in the powder (Fig. 1d), revealing that many were broken under the applied pressure during cold compaction and/or HIP. Upon the increase in carbon content, there is no significant evolution for both the flake diameter distribution and average value (Fig. 8f, h) but some pores and/or CNT agglomerates (clusters) are observed, as black dots about 5 and 8 μ m in size for S1.5 (Fig. 8e) and S2 (Fig. 8g), respectively. For specimen S1, the three areas were investigated into more details. A top-view image showing all three areas is presented in

Fig. 9a. A few isolated pores are observed in all of them. Highermagnification images reveal that the Al flakes are isotropic in Area 1 (Fig. 9b) and Area 2 (Fig. 9c) whereas they appear to be elongated in Area 3 (Fig. 9d). This could reflect the effect of non-uniform forces caused by a gradient in specimen densification. Indeed, densification begins on the outside of the pressed specimen then progressively affects the inner parts of the compact. The outer shell formed during the first steps of sintering may screen the inner parts from high pressure and therefore the Al flakes in Areas 2

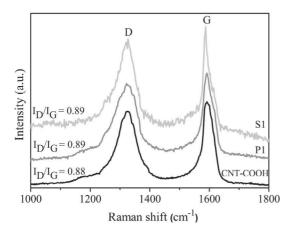


Fig. 5. Raman spectra of the MWCNTs, powder P1 and sintered specimen S1.

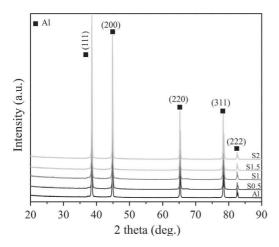


Fig. 6. XRD patterns of sintered specimen S0 (Al), S0.5, S1, S1.5 and S2.

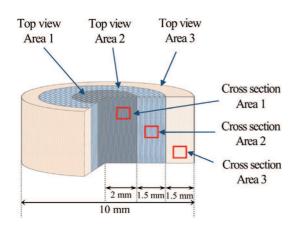


Fig. 7. Microstructure map of the sintered specimens, showing three different Areas for further investigation. See text for details. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

and 1 are subject to lower pressures than those in Area 3. A crosssection image (Fig. 10a) reveals the essentially layered microstructure of the sample and that Area 1 (center) is more porous than Area 2 and, more significantly, Area 3 (periphery). Note that all black parts in Areas 1 and 2 are not pores, but could reflect that some sample parts were removed because of the stronger influence of the etching in those areas. Higher-magnification images show that the multi-layered microstructure is much distorted at the center (Area 1, Fig. 10b), the flakes (1–2 μ m thick) being much less aligned and ordered than at the periphery (Area 3, Fig. 10d).

For Area 1, top-view FESEM observations were performed on the polished and chemically etched specimens (Fig. 11). The individual Al grains that form the flakes appear to be about 500 nm in size. Flake/flake boundaries and intra-flake micro-cracks, all less than 250 nm wide, are observed in all specimens. MWCNTs tend to bridge the micro-cracks (inset in Fig. 11f), as observed elsewhere [38]. The MWCNTs spatial distribution is satisfactory for S0.5 (Fig. 11b), S1 (Fig. 11d) and S1.5 (Fig. 11f) whereas some MWCNT clusters are observed for S2 (Fig. 11h) in agreement with the optical microscopy results presented above (Fig. 8). Cross-section FESEM images for different Areas of specimen S1 (Fig. 12) clearly confirm the layered microstructure with flakes stacked one upon the other. Interestingly, the higher magnification images (right panel in Fig. 12) show the surface of the MWCNT/Al flakes. The flake/flake interface is therefore quite rich in MWCNTs, which would decrease the sintering ability.

3.2. Microhardness and coefficient of thermal expansion

The microhardness (H - Table 1) increases upon the increase in carbon content from 44 HV (S0) to 83 HV (S1.5) but decreases to 62 HV for S2, probably reflecting both the lower densification and the higher agglomeration of MWCNTs for the latter specimen. These results are in line with earlier reports [18,32–35,39,40]. Bradbury et al. [37] reported a very high value (151 HV) for a MWCNT/Al composite (6 wt% carbon), but note that it is macro-hardness (20 kg load). Yang et al. [13] reported micro-hardness up to 130 HV, which could result from a number of factors although it is not precisely determined. Further studies concerning to the effect of CNTrich layer on the other mechanical properties such as strength and ductility will be presented in future works. Note that other researchers have demonstrated that the mechanical behavior of CNT/Al composites prepared by flake powder metallurgy can indeed be much improved. For example, Jiang et al. [19,20,26] reported that CNT/Al composites showed tensile strength of 375 MPa and a higher (by 12%) plasticity. Similarly, Xu et al. [24] reported the balanced strength and ductility in CNT/Al composites.

The average CTE of the composites in the temperature range 50-250 °C is shown in Fig. 13a. It increases from 50 to 150 °C and reaches a plateau for this temperature. Interestingly the CTE plateau value (i.e. the 150-250 °C value) decreases regularly upon the increase in carbon content by about 12, 18, 25 and 30%, for S0.5, S1, S1.5 and S2, respectively, compared to the value measured for S0. For S2, the 150–250 °C CTE is equal to $15.4\times 10^{-6}\,K^{-1}$ which is significantly lower than results reported earlier for CNT/ Al composites with about the same carbon content and for which the authors confirmed that the CNTs are uniformly dispersed on the surface of the Al grains in the powders as well as inside the Al matrix (Table 1). In fact, to the best of our knowledge, the only lower values are reported for composites using nanometric-sized Al and with higher carbon contents: a 65% decrease for 15 vol% SWCNT/Al samples (CTE $\approx 9.8 \times 10^{-6} \text{ K}^{-1}$) [7] and a 70% decrease for 5 wt% CNT/Al composites [12]. To evaluate the influence of

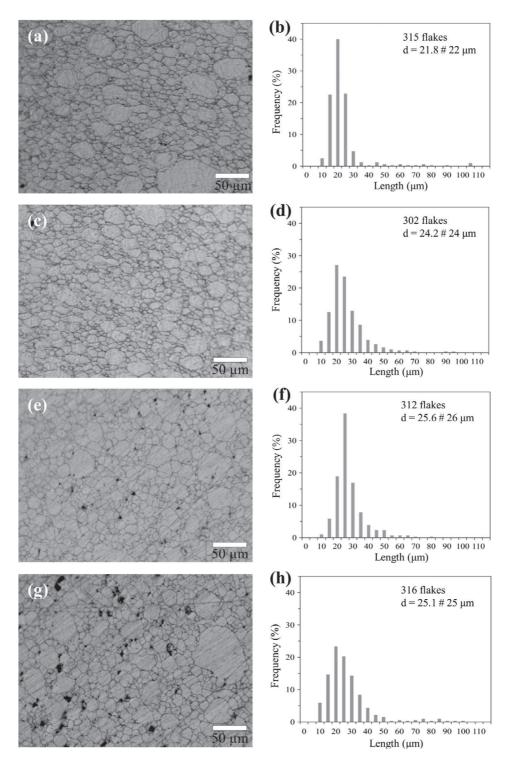


Fig. 8. Top-view optical images and flake size distribution of the sintered specimens (a-b) S0, (c-d) S0.5, (e-f) S1.5 and (g-h) S2.

the MWCNTs on the CTE of the composite, the rule of mixtures (ROM) (Eq. (1)) and Turner's model referring to interconnected or continuous reinforcement ([41,42], Eq. (2)), were used in order to estimate the CTE of composites [7,9,10,14]:

$$\varepsilon_c = \varepsilon_{CNT} V_{CNT} + \varepsilon_{AI} (100 - V_{CNT}) \tag{1}$$

$$\varepsilon_c = \frac{E_{CNT}\varepsilon_{CNT}V_{CNT} + E_{Al}\varepsilon_{Al}(100 - V_{CNT})}{E_{CNT}V_{CNT} + E_{Al}(100 - V_{CNT})}$$
(2)

where ε_c , ε_{CNT} (≈ 0 for MWCNTs with N = 14 walls [14]), ε_{AI} ($\approx 22.1 \times 10^{-6} \text{ K}^{-1}$) [1] are the CTE of composites, CNTs and Al, respectively; E_{CNT} (≈ 950 GPa) [43], E_{AI} (≈ 69 GPa) are the elastic modulus of CNTs and Al, respectively; and V_{CNT} is the carbon content (vol%). Note that Turner's model will be the same than the lower bound of Schapery's model [14] using elastic energy principles to derive bounds for effective CTEs of anisotropic composites made from isotropic constituents. The CTE values calculated using the ROM model (open circles in Fig. 13b) are not in agreement with the experimen-

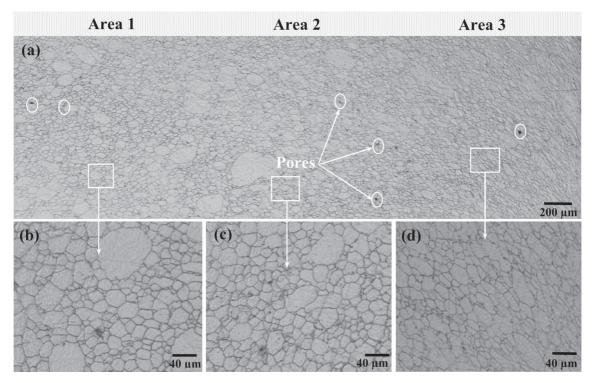


Fig. 9. Top-view optical images of S1 at different positions from center to periphery: (a) global image, (b) Area 1, (c) Area 2 and (d) Area 3.

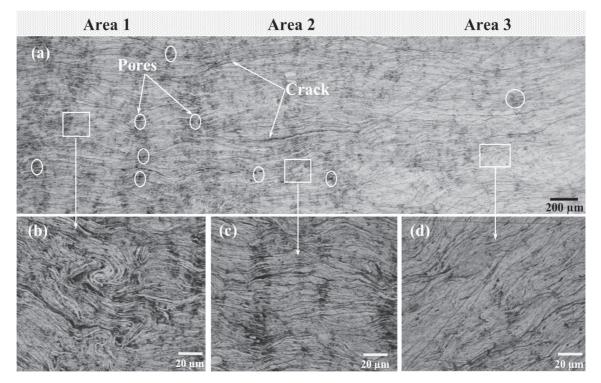


Fig. 10. Cross-section optical images of S1 at different positions from center to periphery: (a) global image, (b) Area 1, (c) Area 2 and (d) Area 3.

tal data. This is mainly because the ROM does not take into account the interfacial thermal stress between the CNTs and Al matrix and the Al matrix restriction influence [7,44]. By contrast, the values calculated using Turner's model are in good agreement with the experimental results, being only slightly higher. The model takes into account the mechanical interaction between the phases in the

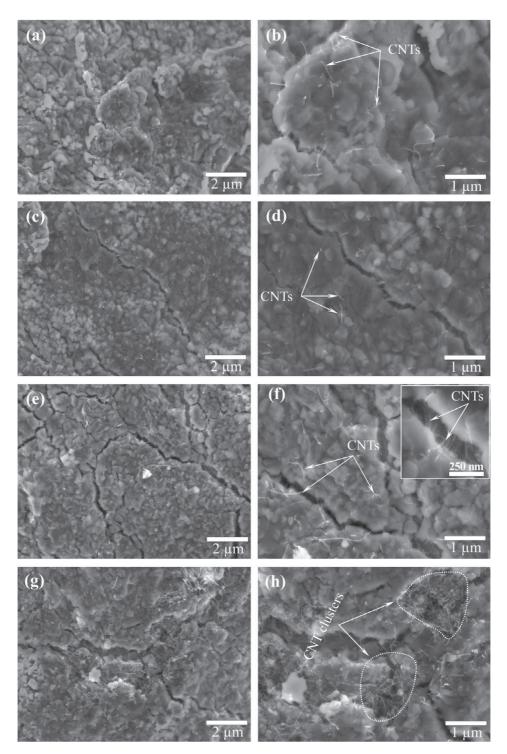


Fig. 11. Top-view FESEM images in Area 1 for (a-b) S 0.5, (c-d) S1, (e-f) S1.5 and (g-h) S2. Inset in (f) shows crack bridging by the MWCNTs.

heterogeneous composite material but disregards any shape and distribution factor of particles within the composite which may hold true for low loading fractions as is the case here. The good fit could reflect, in addition to the input of the elastic constants into the calculation, the uniform dispersion of both the CNTs and CNT-rich layer, the contribution of the MWCNT-Al interface [7,9,10], where within a given flake there is probably a relatively good bonding, or also the influence of the MWCNT-rich flake/flake interfacial regions, which could be viewed as a disruption of the composite

[42]. We propose that the present layered-microstructure with a MWCNT-rich layer located between two MWCNT/Al flakes, resulting from the use of Al flakes as opposed to rounded particles, is mainly responsible for the low CTE measured. The MWCNT-rich layer located between two Al flakes will block the grain growth of each Al flake in the vertical direction. Al flakes were broken into smaller flakes during HIP but they are still connected together by MWCNT bridges in the horizontal direction (Fig. 11). These MWCNT bridges will act as joint splices to adjust the grain growth of Al

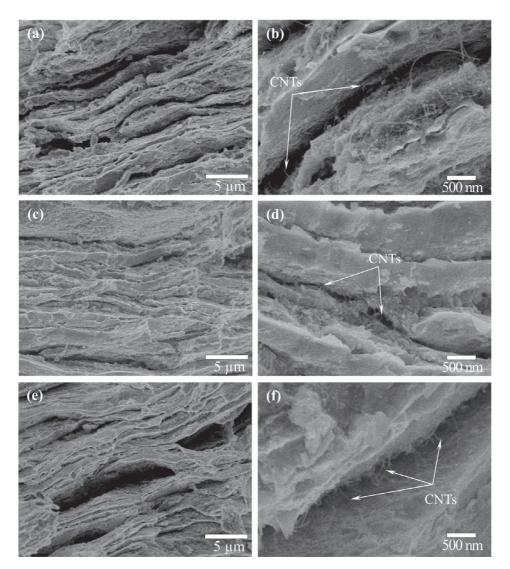


Fig. 12. Cross-section FESEM images of S1 at different positions (e) from center to periphery (a-b) Area 1 (c-d), Area 2 and (e-f) Area 3.

flakes in the horizontal direction. This is the reason why specimen S2, with a less uniform dispersion of MWCNTs, still has a lower CTE compared to other specimens. Thus, the layer-by-layer structure, is an important factor that strongly affects the CTE of MWCNT/Al composites, together with the uniform MWCNT dispersion. It is our opinion that none of the usual CTE models (Kerner, Schapery, Turner and Hashin-Shtrikman) is simply applicable to such complex materials where the load (MWCNTs) is highly anisotropic, the matrix grains are flake-shaped and the billet structure is non-uniform. The influence of the layer-by-layer structure, in particular, will be modeled in a future paper.

4. Conclusions

Layered-microstructure MWCNT/Al composites were prepared by a combination of flake powder metallurgy, PBA mixing technique and hot-isostatic-pressing. The composites show a layeredmicrostructure with a stacking of MWCNT/Al flakes $1-2 \mu m$ thick, with a MWCNT-rich layer between two MWCNT/Al flakes. The flakes however are much less ordered at the center of the specimen than at the periphery. The individual Al grains forming the flakes are about 500 nm in size. Flake/flake boundaries and intra-flake micro-cracks, all less than 250 nm wide, are observed in all specimens. The MWCNTs are well dispersed within a flake and they bridge the micro-cracks. The CTE decreases markedly upon the increase in carbon content, reaching 15.4×10^{-6} K⁻¹ for the specimen with a carbon content of 2.0 wt% (2.9 vol%), i.e. a 30% decrease compared to the CTE of pure Al. Turner's model, referring to interconnected or continuous additive, is a good fit to the experimental data. The low CTE measured could arise from the present layered-microstructure with a MWCNT-rich layer located between two MWCNT/Al flakes, resulting from the use of Al flakes as opposed to rounded particles. The study of a possible anisotropy of the CTE is the subject of future work.

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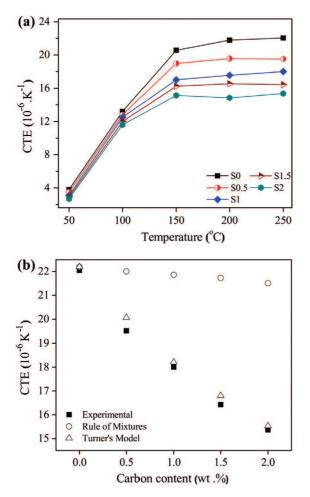


Fig. 13. (a) CTE of CNT/Al composites versus temperature and (b) comparison between calculated and experimental CTE versus carbon content in the temperature range of 200–250 °C. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

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