

YIELDS OF CYCLOTRON PRODUCED MEDICAL ISOTOPES: A COMPARISON OF
THEORETICAL POTENTIAL AND EXPERIMENTAL RESULTS *

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Abstract

Experimentally obtained yields of most of the medical radioisotopes, produced with cyclotrons through different nuclear reactions at various bombarding energies in laboratories around the world, are presented. These yields are compared with those calculated using experimentally measured cross sections (where available) at similar bombarding conditions. Where experimental cross sections are unavailable, empirically constructed excitation functions have been used.

The information provided in this paper would be a valuable aid in selecting the most suitable nuclear reaction and bombarding conditions for producing a particular radioisotope and in assessing various losses of the isotope during chemical processing of the irradiated target.

Introduction

Production of radioisotopes for diagnostic studies is by far the most well known medical application of cyclotrons, and most medical cyclotrons are occupied in the production of various isotopes for research and routine work in nuclear medicine and nuclear biology. Generally neutron-deficient, carrier free, and shorter-lived isotopes, which cannot be produced in a reactor, are produced with cyclotrons. However, at the same time, neutron enriched isotopes can also be produced with cyclotrons, if required, through reactions of the type (d,p), (^3He ,p), (α ,p) etc.

Method and Results

Cyclotron production of most of the isotopes in current use has been summarized in Table I.

Thick-target yields at saturation of most isotopes through different nuclear reactions have been calculated¹ using experimentally measured (for ^{11}C , ^{13}N , ^{15}O , ^{18}F and ^{52}Fe) or empirically constructed excitation functions² and range-energy data³. In the calculations, the isotopic abundance of the particular isotope contributing to the nuclear reaction has been taken into consideration. However, the matrix which slows down the incoming particle beam in the target is assumed to consist of only the element taking part in the nuclear reaction, which is true when elemental rather than compound targets are being used. The calculated yields are designed as guide-lines only to optimize actual production and may be in error by as much as a factor of two in the case of empirically constructed excitation functions². For comparison between the actually measured and calculated yields, the saturation yields would have to be converted into yields at time 't' where 't' is the time for actual bombardment, by using the factor $(1 - e^{-\lambda t})$.

In compiling the Table, an attempt has been made to include most of the published data regarding the production of various isotopes, in biomedical use, using different nuclear reactions. However, it is possible that some particular isotope publications might have been inaccessible or inadvertently omitted. Only the first or the first significant published paper for radioisotope production under any particular bombarding conditions, such as energy, target material etc., has been included.

The operating costs of a small cyclotron in a developed country are estimated to be around US\$60-90 per hour of useful machine time. Keeping this figure in mind, one can calculate the expected cost per millicurie of any particular isotope. However, the man-hours required for the chemical processing of the irradiated target must also be included in the overall cost estimate.

TABLE I

Isotope	Reaction	Energy (MeV)	Target	Production yield $\mu\text{Ci}/\mu\text{Ah}$, unless otherwise indicated	Ref.	Calculated yield at saturation ¹	
						$\text{mCi}/\mu\text{A}$	Energy (MeV)
^{11}C	$^{11}\text{B}(\text{p},\text{n})$			-----	---	405	11.5
	$^{10}\text{B}(\text{d},\text{n})$	14	B_2O_3	5 mCi/min $\text{CO} - 100 \text{ mCi}$ per litre of H_2 carrier gas - 70 mCi per 35 ml of He $\text{CO}_2 - 50 \text{ mCi}$ in 35 ml of He	4	80	14
	$^{11}\text{B}(\text{d},2\text{n})$						
	$^{12}\text{C}(\text{He}^3,\alpha)$	15-18	CaC_2	2500	5	87	15
	$^{14}\text{N}(\text{p},\alpha)$	15	LiNH_2	18 mCi/18 min of H^{11}C N	6	115	18
	"	18	5% H_2 in N_2	90 mCi/ μAh	7	---	---
^{13}N	$^{12}\text{C}(\text{d},\text{n})$	14	Graphite	30 mCi/ml in gas form 100-300 μCi per ml in solution	8	300	14
	$^{14}\text{N}(\text{He}^3,\alpha)$	30	N_2	15 mCi/ μAh	9	38	30

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Isotope	Reaction	Energy (MeV)	Target	Production yield μCi/μAh, unless otherwise indicated	Ref.	Calculated yield at saturation ¹ mCi/μA	Energy (MeV)
¹³ N	¹⁶ O(p,α)	18	H ₂ O	24 mCi/μAh as NH ₃	10	---	---
¹⁵ O	¹⁴ N(d,n)	5	4% O ₂ in N ₂	120 mCi of O ₂ per litre of carrier gas, 40 mCi/ml of H ₂ O	4	12.5	5
	¹⁴ N(³ He,d)	---	---	-----	---	26	17
	¹⁶ O(p,pn)	27	O ₂	1.5mCi of ¹⁵ O per 15 ml of CO ₂ - 6mCi/min of O-15 labelled CO ₂ at 15μA	11	---	---
¹⁸ F	¹⁶ O(α,d)	30	H ₂ O	1100	12	5.4	30
	(α,pn)	40	"	11000	13	27	40
	(α,2n)	40	O ₂	14300	13	---	---
		65	H ₂ O	19000	14	---	---
	¹⁶ O(³ He,p)	22	H ₂ O	6000	15	12	9.5
	(³ He,n)	30	H ₂ O	8500	13	32	30
		---	---	-----	---	43	35
	¹⁹ F(³ He,α)	---	---	-----	---	8	30
	²⁰ Ne(d,α)	8	Neon Gas	10000	16	52	8
		---	---	-----	---	100	16
²⁸ Mg	²⁷ Al(α,3p)	45	Al	6.2	17	---	---
		140	Al	35	18	---	---
⁴³ K	⁴⁰ Ar(α,p)	17	Argon	57	19	9	17
⁵² Fe	⁵⁰ Cr(α,2n)	30	natural Cr	3.3	20	.007	30
	⁵² Cr(³ He,3n)	45.5	"	50	21	9	45
	⁵⁵ Mn(p,4n)	65	MnO ₂	0.7	22	0.18	23
				160	23	90	65
⁶² Zn	⁶³ Cu(p,2n)	38 → 18	Cu-foils 1.6g/cm ²	6000	24	---	---
	⁶⁰ Ni(α,2n)	30	natural Ni	100	25	6	30
⁶⁷ Ga	⁶⁵ Cu(α,2n)	30	natural Cu	160	26	8	30
	Zn(p,xn)	22	natural Zn	430	27	95	22
	Zn(d,xn)	8	"	100	28	2	8
	"	"	"	30	22	—	—
	"	16	"	340	29	—	—
	"	16	enriched ⁶⁶ Zn (90%)	946	29	—	—
⁷⁷ Br	⁷⁵ As(α,2n)	28	As ₂ O ₅	160	30	25	28
	"	28	As ₂ O ₃	290	31	—	—
	"	28	As metal	170	31	—	—
	^{79,81} Br(α,6n)	100	NaBr	322	32	---	---
	(α,pxn)						
	(d,pxn)	55 → 20	"	550	33	---	---
	(d,xn)	"	"	640	33	---	---
^{81m} Kr			Decay product of ⁸¹ Rb				
^{85m} Kr	⁸⁴ Kr(d,p)	15	Kr	790	34	60	15
⁸¹ Rb	⁷⁹ Br(α,2n)	30	NaBr	2000	35	18	30
		50	"	2900	36		
	⁸¹ Br(³ He,n)	22	NaBr	30	37	0.13	22
^{82m} Rb	⁸¹ Br(³ He,2n)	22	NaBr	80	37	1.2	22
⁸³ Rb	⁸³ Kr(p,n)	22	Natural Kr-gas	7	38	60	22
⁸⁵ Sr	⁸⁵ Rb(d,2n)	13	RbCl	15	39	70	13
⁸⁷ Y	⁸⁵ Rb(α,2n)	32	RbCl	174	40	30	32

Isotope	Reaction	Energy	Target	Production yield μCi/μAh, unless otherwise indicated	Ref.	Calculated yield at saturation ¹ mCi/μA Energy (MeV)	
¹¹¹ In	¹⁰⁹ Ag(α,2n)	30	natural Ag	200	41	9	30
	Cd(p,xn)	15	natural Cd	140	22	35	15
	¹¹¹ Cd(p,n)	16	enriched ¹¹¹ Cd	515	42	150	16
	¹¹² Cd(p,2n)	22	natural Cd	1035	43	300	22
	¹¹⁰ Cd(d,n)	12	natural Cd	117	43	20	12
	¹²³ I	25	natural Sb	150	44	3	25
¹²³ I	¹²¹ Sb(α,2n)	25-36	enriched ¹²¹ Sb	900	45	6-30	25-36
	¹²¹ Sb(³ He,n)	23	natural Sb	24	25	0.22	23
	¹²² Te(d,n)	6-9	enriched ¹²² Te	100	45	0.3	9
	¹²³ Te(p,n)	15.5	enriched ¹²³ Te	450	46	140	16
	¹²⁴ Te(p,2n)	30	enriched ¹²⁴ Te	40000	47	1300	30
	¹²² Xe	46	enriched ¹²² Te	5000	48	70	46
¹²³ Xe	¹²³ Te(³ He,3n)	30	enriched ¹²³ Te	750	49	20	30
	"	32	enriched (¹²³ Te (88%))	4954	50	25	32
		35	"	6441	50	30	35
	¹²⁴ Te(³ He,4n)	38	enriched ¹²⁴ Te (96%)	1000	50	8.5	38
		52	"	7155	50	33	52
	¹²⁷ I(p,5n)	60-75	NaI powder 1.5g/cm ²	5600	24	---	---
¹²⁷ Cs		52	CH ₂ I ₂ (Circulating liquid)	1000	51	---	---
		53	" "	2200	52	---	---
		72	KI & NaI	11500	53	---	---
	¹²⁷ I(d,6n)	78	NaI	8000	54	---	---
	¹²⁷ Cs	22	NaI	500	55	6	22
	¹²⁷ I(α,2n)	30	NaI	170	56	18	30
¹⁵⁷ Dy		35	"	300	57	30	35
		36	"	700	58	30	36
	¹⁵⁹ Tb(p,3n)	30	Terbium foil TbCl ₃ .6H ₂ O	23000	59	500	30
				2500	60	---	---
	¹⁵⁵ Gd(α,2n)	30	Gd	80	61	3	30
	¹⁹⁷ Hg	12.5	Gold	14	62	2	12.5
^{197m} Hg	¹⁹⁷ Au(p,n)	12.5	Gold	15	62	10	12.5
²⁰³ Pb	²⁰³ Tl(p,n)	15	Tl metal	50	63	1	15
	²⁰³ Tl(d,2n)	16	Tl ₂ O	100	4	6	16
²⁰⁴ Bi	²⁰⁶ Pb(p,3n)	32	Pb	2000	64	180	32
²⁰⁶ Bi	²⁰⁷ Pb(p,2n)	22	Pb	700	65	110	22
	²⁰⁶ Pb(d,2n)	16	Pb	30	4	6	16

Isotope	Reaction	Energy (MeV)	Target	Production yield μCi/μAh, unless otherwise indicated	Ref.	Calculated yield at saturation ¹ mCi/μA	Energy (MeV)
²⁰¹ Tl	²⁰³ Tl(p,3n)	30	natural Tl	700	66	---	---
	Hg(p,n)	14	natural Hg	180	67	---	---
		16	"	350	67	---	---
		20	"	350	67	---	---

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