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Vibrational spectroscopy of *trans* and *cis* deuterated formic acid (HCOOD): Anharmonic calculations and experiments in argon and neon matrices

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ABSTRACT

The absorption spectra of *trans* and *cis* conformers of deuterated formic acid (HCOOD) isolated in argon and neon matrices are analyzed in the mid-infrared and near-infrared spectral regions (7900–450 cm⁻¹). Vibrational excitation by narrow-band IR radiation is used to convert the lower-energy *trans* conformer to the higher-energy *cis* form. A large number of overtone and combination bands are identified. The results of anharmonic vibrational calculations (CC-VSCF) for both conformers are reported and compared to the experimental spectra.

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1. Introduction

Formic acid exists in two planar forms differing by the O–H bond orientation with respect to the C–O bond with the 0° and 180° H–C–O–H dihedral angles (see Fig. 1). First unequivocal identification of the gas-phase absorption of the higher-energy *cis* form was made by analysis of rotational spectra which demonstrated that the *trans* conformer is lower in energy by ~1365 cm⁻¹ compared to the *cis* form [1]. The vibrational spectrum in the mid-IR region of *cis*-HCOOH was first obtained by Pettersson et al. in an argon matrix [2] and the detailed vibrational data on both forms of HCOOH in solid argon were reported later [3]. The preparation of *cis*-HCOOH was achieved by selective vibrational excitation which has been proved to be a powerful tool for promotion and investigation of conformational changes in cryogenic matrices [4].

The IR spectrum of *trans*-HCOOH is well known in matrices and in the gas phase. High-resolution measurements of all fundamental, many overtones and combination bands are available (see Ref. [5] and references therein) and an accurate assignment of the rotation-vibration spectral structure has been made with Fermi resonance taken into account [6]. Most of the fundamental bands of *trans*-HCOOD molecule were also recorded and analyzed in the gas phase [7] and in the matrices although the data for the overtone and combination transitions is sparse. The vibrational spectrum of *trans*-HCOOD measured in solid neon is close to that in the gas phase [8,9]. To the best of our knowledge, only one vibrational band of *cis*-HCOOH [COH torsion, 493.4205092(73) cm⁻¹] was measured in the gas phase [10]. Since the *cis* conformer is about 800 times less abundant at room temperature, the use of a multi-pass cell with total path length of 38.2 m was necessary to accumulate reliable absorption level [10].

The vibrational spectrum of *cis*-HCOOD is not studies in detail. The geometry and rotational constants of the ground state have been reported from the microwave studies in the gas phase for H¹²COOD and H¹³COOD [11,12]. The fundamental transitions for both conformers of HCOOD in different matrices are tabulated in Ref. [13], and a few combinations and overtones are assigned in Ref. [3]. The *cis*-HCOOH force field [14] might provide information about the transitions in *cis*-HCOOD because the potential energy function and the configuration of a molecule are practically unchanged upon isotopic substitution [15]. However, the work by Demaison et al. is mostly focused on *trans*-HCOOH and does not contain the force field for *cis*-HCOOH. It reports only the equilibrium geometry and rotational constants of the ground state of the *cis* conformer.

The present work supplies more information about *cis* and *trans*-HCOOD vibrational transitions, which complements the earlier studies [3,8,9,13]. The IR absorption spectra of *cis* and *trans*-HCOOD are measured in argon and neon matrices at 4.2 K. The anharmonic vibrational frequencies for the *trans* and *cis* conformers, derived from the correlation-corrected vibrational self-consistent-field calculations (CC-VSCF), are used for vibrational assignment of the overtone and combination bands.

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Fig. 1. trans and cis forms of formic acid. The notations follow Refs. [1-3].

2. Experimental

The gaseous samples were prepared by mixing deuterated formic acid HCOOD (IT Isotop, 95–98%), with argon or neon (AGA, 99.9999%), typically in 1:700 (Ar) and 1:1000 (Ne) proportions in a glass bulb. The matrices were deposited onto a CsI substrate at 18 K for Ar and at 4.2 K for Ne matrices in a closed cycle helium cryostat (Sumitomo Heavy Industries). The IR absorption spectra (7900–450 cm⁻¹) were measured with a Nicolet SX-60 FTIR spectrometer at 4.2 K. The spectral measurements in mid-IR region were carried out with the resolution of 0.25 cm⁻¹ using a cooled MCT detector and KBr beamsplitter, and the near-IR spectra were recorded using a cooled InSb detector and quartz beamsplitter with spectral resolution of 0.5 cm⁻¹. Typically 500 interferrograms were co-added.

trans-HCOOD was converted to the *cis* form by optical excitation of the first OD stretching overtone at 5174.5 cm⁻¹ in a neon matrix and at 5149.7 cm⁻¹ in an argon matrix with pulse energy density of ~1 mJ/cm². For argon matrices, excitation at 4382.0 cm⁻¹ (vOD + vC=O combination) was also used. An optical parametric oscillator (OPO Sunlite with IR extension, Continuum) was used as a source of pulsed IR radiation. A Burleigh WA-4500 wavemeter measured the OPO signal wavelength providing an absolute accuracy better than 1 cm⁻¹ in the IR region.

3. Computational method

The computational method was essentially the same as described in our previous work [3]. The equilibrium structures of *trans*- and *cis*-HCOOD were calculated using the second-order Møller–Plesset perturbation (MP2) theory with the 6-311++G(2d,2p) basis set. This basis set reproduces the experimental structural and vibrational properties of *trans* formic acid with an acceptable accuracy [16,17]. The anharmonic vibrational properties of *trans*- and *cis*-HCOOD were studied by combining the electronic ab initio code GAMESS [18] with the vibrational self-consistent field (VSCF) method and its extension by corrections via second-order perturbation theory [19–23].

4. Assignment and discussion

In this section, we analyze the mid- and near-IR (450–7900 cm⁻¹) spectra of the *trans* and *cis* conformers of deuterated formic acid in argon and neon matrices and compare them with the computational and, when possible, with gas-phase spectra. The bands of *trans*-HCOOD dominate in the spectrum after deposition. An efficient *trans* to *cis* conversion (>90%) was achieved by excitation of the OD stretching overtone or the vOD + vC=O combination (see Fig. 2). *cis*-HCOOD has a larger lifetime compared to *cis*-HCOOH because the tunneling to the *trans* form is slowed down by deuteration [13,24,25], and their increased stability allows accurate data collection of *cis*-HCOOD.



Fig. 2. (a) The OD stretching bands and (b) the OD stretching overtones of HCOOD in Ar and Ne matrices at 4.2 K. The as-deposited matrices mainly contain the *trans* conformer as shown for an argon matrix (upper traces). Vibrational excitation converts *trans*-HCOOD to *cis*-HCOOD as presented by the difference spectra (middle and lower traces).

In a neon matrix, the bands are mostly single absorption peaks, and their frequencies are close to the gas-phase values [8]. In an argon matrix, the band splitting due to two matrix trapping sites is observed similarly to HCOOH [3]. Table 1 presents the observed absorption bands for the *trans* and *cis* conformers of HCOOD in argon and neon matrices and the anharmonic computational frequencies together with the available gas-phase data.

4.1. Fundamental transitions

The literature data [7,8,13,26,27] allows a straightforward assignment of the fundamental bands of *trans*-HCOOD. The fundamental transitions of the *cis* form are relatively easy to assign as well (see Ref. [13]). In Fig. 2(a), the HCOOD spectra in the OD

Table 1

Observed experimental vibrational frequencies (cm⁻¹) for *trans*- and *cis*-HCOOD in Ar and Ne matrices at 4.2 K, CC-VSCF calculated anharmonic vibrational frequencies (intensities in parentheses in km/mol) and gas-phase data.

Assignment	trans-HCOOD			cis-HCOOD			
	Calculated	Gas ^a	Ar	Ne	Calculated	Ar	Ne
τCOD	497.5 (103.1)	507.5	505.2s 506.7s	506.9s	361.4 (29.3)	-	-
OCO–CODdef.	555.7 (52.2)	560.2	558.7m 559.7s	557.9m	621.5 (17.9)	633.0m 633.6m	629.4m
δCOD	972.7 (86.2)	972.8	970.0m 970.6s	971.1s	897.9 (7.0)	910.5m 911.2w	906.8m
2τCOD	960.9 (4.2)	1011.7	1008.3m 1009.4m	1009.5m	688.3 (0.16)	740.4vw	746.2vw
ωCH	1033.2 (0.1)	-	1035.1vw 1036.4vw	1037.3vw	1014.4 (1.0)	1041.2vw	1039.3vw
vC—O	1163.7 (206.1)	1177.1	1181.7vs	1176.9vs	1140.9 (348.2)	1164.0vs	1159.0vs
vCH	1381 1 (7 2)	1368	1371 1vw	1367 9w	1403.6(2.7)	1391 Ovw	1388 Ovw
, en	100111 (712)	1500	1374 3vw	10071011	110010 (217)	1395 4vw	190010111
$\tau COD + \delta COD$	1441.0 (0.0)	_	1427.5vw	1433.0vw	1207.8 (0.0)	_	_
$OCO-CODdef + \delta COD$	1533 5 (0.1)	_	_	1526 3vw	15249(01)	1543 2vw	1534 2vw
OCO-CODdef + vC-O	17156(37)	1732.1	1735 7w	1730.0w	17607(0.8)	_	-
v(=0	1751 3 (460.8)	1773.6	1765 6vs	1771 Ovs	1789 5 (434 2)	1788 9m ^b	1781 2w ^b
	1751.5 (100.0)	1775.0	1767 2vs	1771.045	1703.5 (131.2)	1799 3vs ^b	1798 9vs ^b
			1707.245			1821.0m ^b	1818 2vs ^b
2δCOD	1938.0 (0.3)		1949.7vw	1950.6vw	1779.9 (0.1)	-	-
$\omega CH + \delta COD$	2004.8 (0.0)	_	2008 4vw	2008 6vw	1912.3 (0.0)	_	_
$v = 0 + \delta C O D$	21296(37)	_	2144 1w	2139.8w	20301(21)	2065 1w	2055 2w
			2181.0w	2176.8w			
$vC = 0 + \omega CH$	21937(01)	_	2194 7vw	2192 9vw	21518(01)	_	_
2vC-0	23110(09)	_	2349 4vw	2341 1vw	22647 (45)	2314 7vw	2301 5vw
vOD	2625.6 (67.4)	2631.4	2618.0s	2631.55	2677 4 (128 6)	2668.75	2685 55
100	2023.0 (07.1)	2031.1	2619.55	2031.33	2077.1 (120.0)	2670.0m	2005.55
$vC=0 + \delta COD$	2720.2 (2.5)	_	2723.1vw	2716.6vw	2684.7 (2.1)	2688.7vw	2674.4vw
$vC=0 + \omega CH$	2776.4 (0.2)	_	2732.1vw	2737.9w	2795.8 (0.2)	2752.0w	2742.8w
	,		2728.7w			2758.2vw	2724.9vw
vC=0 + vC=0	2914.8 (1.9)	_	2940.7w	2935.5m	2929.8 (2.4)	_	2939w
vCH	2952.8 (29.7)	2942	2961.2m	2951.9m	2861.4 (47.4)	2895.3m	2872.7m
						2897.9m	
$vOD + \tau COD$	3104.4 (10.0)	-	3123.3vw	3136.7w	3008.1 (7.6)	3050.6vw	3059.2vw
			3125.8vw	3138.1vw			
$vC = O + \gamma CH$	3126.5 (0.4)		3134.8vw		3186.9 (0.1)		
			3139.5vw				
vOD + OCO–CODdef	3139.3 (0.5)	-	3171.2vw	3183.0vw	3240.4 (0.3)	-	-
			3173.9vw				
2vC=0	3483.9 (4.5)	-	3512.9vw	3524.1w	3553.7 (4.6)	3571.1vw	3582.2w
			3522.0vw	3529.1vw		3577.3vw	3583.2vw
							3598.8w
$vOD + \omega CH$	3628.6 (0.0)	-	-	3601.3vw	3647.4 (0.0)	-	3629.4vw
vOD + vC - O	3754.7 (0.0)	-	3736.1vw	3728.7vw	3760.4 (3.9)	3741.5vw	3740.2vw
vCH + vC—O	4114.4 (1.3)	4125	4140.1vw	4123.8vw	4001.5 (1.4)	4052.4vw	4025.9vw
						4058.3vw	
vOD + vC = 0	4345.8 (0.5)	4404	4382.0vw	4399.5vw	4418.2 (0.3)	-	
vCH + vC = 0	4704.3 (1.8)	4719	4720.7vw	4714.2vw	4653.6 (2.1)	4687.9vw	4666.7vw
			4723.8vw			4707.9vw	4686.8vw
2vOD	5163.6 (1.1)	5181	5146.3vw	5174.5vw	5284.7 (1.1)	5249.5w	5280.4vw
			5149.7w			5252.4vw	
			5152.9vw			5253.7vw	
2vCH	5791.3 (0.5)	-	5801.7vw	5769.3vw	5690.9 (0.5)	5592.9vw	5548.6vw
						5594.4vw	

^a From Refs. [7,11,26,27,29].

^b Components of Fermi resonance.

stretching region are presented. The band at 2631.5 cm⁻¹ is the OD stretching mode (vOD) of *trans*-HCOOD in a neon matrix. It shows good agreement with the anharmonic calculations yielding 2625.6 cm⁻¹. For *cis*-HCOOD, this mode shifts to 2685.5 cm⁻¹ (computationally to 2677.4 cm⁻¹). In an argon matrix, the *trans*-and *cis*-HCOOD *v*OD absorptions are at 2618.0 and 2668.7 cm⁻¹, showing a typical matrix shift from the neon matrix values.

The CC-VSCF calculations reproduce well the experimental frequencies for most of the fundamental modes. The calculated intensities of HCOOD match the experimental values (see Fig. 3). Similarly good agreement between the anharmonic calculations and the experimental spectra was reported for HONO [28]. The high-frequency OD and CH stretching modes for the *trans* conformer, where anharmonicity is significant, are obtained with the 0.3% inaccuracy compared to the gas-phase data. For the torsional mode the anharmonic calculations and the experimental frequencies from the gas-phase experiments agree better for the deuterated molecule than for that of HCOOH (\sim 2% error for *trans*-HCOOD, \sim 7% for *trans*-HCOOH). This is explained by the fact that the vibrational amplitude is smaller for HCOOD and the limitation of large-amplitude vibrational mode description by the CC-VSCF method becomes less important.

Among other fundamental modes, the largest difference between the calculated and experimental neon matrix values is ob-



Fig. 3. Relative calculated and experimental intensities for the fundamental modes of HCOOD. The intensities are normalized by the vC=0 intensity of *trans*-HCOOD. The experimental values are obtained in a neon matrix. The straight line shows 1:1 agreement with the experiment.

served for the C—O stretching and CH-wagging vibrations of the *cis* conformer (\sim 20 cm⁻¹, \sim 2%). For the *v*C—O vibration, this discrepancy possibly originates from the fact that the CC-VSCF method accounts only for pairwise couplings and the multiple interactions are neglected [19–21]. The *v*C—O vibration is a mixed mode and the contribution of the OCO—COD deformational motion is large [8]. The reason for the difference between the calculated and experimental frequencies of the CH-wagging mode remains an open question because this vibrational mode is well-isolated [8].

The assignment of vibrational transitions in the vC=O region is less certain due to Fermi resonances. An accurate analysis of the vC=O vibration of trans-HCOOD in the gas phase yields that the strongest perturbation for the vC=O vibrational state arises from the nearest OCO–CODdef + vC–O combination. The energy difference between these levels is about 40 cm^{-1} [7]. Another combinational state ($vC-O + \tau COD$) perturbs the vC=O level in a considerably weaker extent because the energy gap between the levels is ~90 cm⁻¹. The key role of the vC–O + τ COD state is the perturbation of the OCO–CODdef + vC–O state [7]. In the case of *cis*-HCOOD, one may expect that the influence of the *v*C–O + τ COD combination on the vC=O state gets even smaller due to the increasing energy gap between the interacting levels, but the first overtone of the δ COD bending vibration (computationally at 1779.9 cm⁻¹) might play an important role on the frequency and intensity of the vC=0 band as well as the OCO-CODdef + vC-0and γ CH + τ COD combinations (computationally at 1760.7 and 1761.5 cm⁻¹, respectively). The different "partners" of the Fermi resonance in the *cis* and *trans* molecules might be responsible for the significant different spectra of the two conformers in this region.

The vC=O vibration of *trans*-HCOOD in a neon matrix has a single absorption shifted from the gas-phase value by -2.6 cm^{-1} and in an argon it is a doublet due to the matrix site effect which is shifted by -6.4 and -8.2 cm^{-1} from the gas-phase value. Similar doublet was observed for *trans*-HCOOH in an argon matrix [3]. The OCO-CODdef + vC-O combination is observed ~40 cm⁻¹ below (1730.0 cm⁻¹ in a Ne matrix) in agreement with the gas-phase values.

A triplet structure is observed in the vC=0 region of *cis*-HCOOD in various matrices [13]. The intensity distribution is strikingly different for different matrices: the triplet intensity ratio is 1:7:6 in a neon matrix and 3:6:2 in an argon matrix. The presumably inter-

acting combinations are predicted to be \sim 1000 smaller in intensity than the fundamental transition, so that the comparable values of the experimentally observed peaks suggest that the Fermi interaction actually takes place. The precise assignment of the triplet components remains an open question and exceeds the scope of the present work.

The computed intensities of the fundamental transitions agree very well with the experimental values (see Fig. 3). The strongest absorption is the *trans* C=O stretching vibration both computationally and experimentally. This allows us to relay to some extent on the calculated intensities, in addition to the frequency values, for assignment of overtones and combinations.

4.2. Overtone and combination transitions

A number of overtones and combination bands are observed and identified in the mid- and near-IR regions. The assignment of the overtones and combinations is based on the anharmonic CC-VSCF calculations. The experimentally observed and calculated vibrational frequencies are collected in Table 1.

As an example, Fig. 2(b) presents the first OD stretching overtone of *trans*- and *cis*-HCOOD in argon and neon matrices. The 2vOD band of *trans*-HCOOD is observed in a neon matrix at 5174.5 cm⁻¹ and predicted by the calculations at 5163.6 cm⁻¹, which is in good agreement with the gas-phase value (5181 cm⁻¹) [11]. For *cis*-HCOOD, the 2vOD absorption is predicted at 5284.7 cm⁻¹ and observed at 5280.4 cm⁻¹ in a neon matrix. The 2vOD bands are shifted to 5149.7 (*trans*) and 5249.5 cm⁻¹ (*cis*) in an argon matrix.

If the vC=O stretching vibration is involved in a combination transition, the resulting spectrum of the *cis* form might show a characteristic splitting (ca. 20 cm^{-1}) due to the Fermi resonance in the fundamental region. For example, a very weak doublet at 4686.8 and 4666.7 cm⁻¹ in a neon matrix and at 4687.9 and 4707.9 cm⁻¹ in an argon matrix is assigned to the vCH + vC=O combination mode for *cis*-HCOOD. The *trans*-HCOOD absorption band at 4714.2 cm⁻¹ (Ne) and 4720.7 cm⁻¹ (Ar) is assigned to the vCH + vC=O combination. In the gas-phase spectrum, this combination was observed at 4719 cm⁻¹ for *trans*-HCOOD [11].

In the previous work, the doublet at 2753.4 and 2759.9 cm⁻¹ was tentatively assigned to the *v*C=O + 2 τ COH combination of *cis*-HCOOH in an argon matrix [3]. Two bands at 2752.0 and 2758.0 cm⁻¹ appear in the present spectra of *cis*-HCOOD. We reconsider the earlier assignment because the absorptions under question show negligible isotopic shifts whereas the *v*C=O + 2 τ COH combination should be sensitive to the isotopic substitution with a shift of ~300 cm⁻¹. We suggest that the combination transition *v*C=O + ω CH gives rise to this doublet in the spectrum of *cis*-HCOOD in an argon matrix. The corresponding band of *trans* form appear ~20 cm⁻¹ lower in frequency in agreement with the calculations. The calculated values are 2795.8 cm⁻¹ for *cis*- and 2776.4 cm⁻¹ for *trans*-HCOOD.

A weak absorption at 2716.6 cm⁻¹ (Ne) and 2723.1 cm⁻¹ (Ar) probably belongs to the $vC=O + \delta COD$ combination mode of the *trans* molecule (computed at 2720.2 cm⁻¹). The corresponding band of the *cis* form is found ~30 cm⁻¹ lower in frequency in agreement with the calculations. The bands of *trans*-HCOOD at 2778.6 cm⁻¹ (Ne) and at 2773.2 and 2782.5 cm⁻¹ (Ar) might belong to the $vC=O + 2\tau COD$ mode although we do not have a theoretical estimate for this frequency.

Several other absorptions also admit multiple assignments. There are two doublets of *trans*-HCOOD at 3123.3, 3125.8 cm⁻¹ and at 3134.8, 3139.5 cm⁻¹ in argon matrices, whose assignment is not certain. Two combination modes of *trans*-HCOOD are expected to absorb in this region: the vOD + τ COD mode (computed at 3104.4 cm⁻¹) and the vC=O + γ CH mode (computed at 3126.5 cm⁻¹). We tentatively assign two peaks at 3123.3 and 3125.8 cm⁻¹ to the vOD + τ COD mode

because they are lower in frequency than the other pair which makes a better agreement with the calculations. The corresponding peaks in a neon matrix are significantly shifted to higher frequencies (3136.7 and 3138.1 cm⁻¹) as it should be for this combination whereas the $vC=O + \gamma CH$ mode should demonstrate only a small shift (if any) due to the opposite matrix-induced shifts of the vC=O and γCH fundamentals. The $vOD + \tau COD$ bands of the *cis* form is found substantially lower in frequency [3050.6 (Ar) and 3059.2 cm⁻¹ (Ne)] in agreement with calculations (3008.1 cm⁻¹). The other doublet (3134.8, 3139.5 cm⁻¹) probably originates from the $vC=O + \gamma CH$ combination of *trans*-HCOOD. The $vC=O + \gamma CH$ band of *cis*-HCOOD is predicted at higher frequencies (3186.9 cm⁻¹), and it is not found due to its weakness.

For *trans*-HCOOD, the bands at 3524.1 and 3529.1 cm⁻¹ in a neon matrix cannot be unambiguously distinguished between the *v*C=O overtone and the *v*OD + δ COD combination. We tend to assign these bands and the corresponding absorptions of *cis*-HCOOD at 3582.2 and 3598.8 cm⁻¹ to the 2*v*C=O mode since they show a characteristic splitting of the *v*C=O fundamental region. In addition, the *v*OD + δ COD absorption of the *cis* form is expected at significantly lower frequencies (calculated at 3492.5 cm⁻¹) and ~10 times smaller in intensity while we observe a blue shift of the *cis* absorptions and comparable intensities.

For very weak bands of the *trans* conformer at 1949.7 and 1950.6 cm⁻¹ in argon and neon matrices two plausible assignments are suggested by the calculations: 2δ COD (1938.3 cm⁻¹) and OCO—CODdef + γ CH (1938.3 cm⁻¹). Nevertheless, several arguments favor the assignment to 2δ COD. For OCO—CODdef + γ CH, the experimental coupling constants have unrealistically large positive values (ca. +20 cm⁻¹) whereas the anharmonicity constants for 2δ COD are +4.2 cm⁻¹ for both matrices which is more reasonable. Moreover, the computed intensity of the 2δ COD mode is higher than for the other candidate. The 2δ COD band of *cis*-HCOOD cannot be unambiguously assigned because it is predicted in the C=O stretching region and probably participates in the Fermi resonance mentioned above.

A few absorptions are observed only for one conformer. The bands at 4399.5 cm^{-1} in neon and 4382.0 cm^{-1} in argon matrices

are assigned to the vOD + vC=O combination of the *trans* form, which is predicted by calculations at 4345.8 cm⁻¹. The vC=O + ω CH combination band of *trans*-HCOOD (calculated at 2193.7 cm⁻¹) is observed at 2192.9 cm⁻¹ in neon and at 2194.7 cm⁻¹ in argon matrices. These modes are not observed for *cis*-HCOOD supposedly due to their weakness.

The computed CC-VSCF frequencies for the bands, which were not detected in the spectra, are collected in Table 3. Most of these modes are computationally very weak (<0.5 km/mol). The torsional mode for *cis*-HCOOD most probably lies below our measurement limit of 450 cm⁻¹. The expected positions of some combination bands or overtones are close to intense fundamentals, which complicate their detection. For example, the 20CO–CODdef band for *trans*-HCOOD should be at ~1110 cm⁻¹, which coincides with the *v*C–O stretching region.

There are small discrepancies with the previously published data (see Table 3 in Ref. [3]). We suppose that the observed frequency deviations of $1-2 \text{ cm}^{-1}$ are due to relatively high inaccuracy for the high-frequency region and differences in sample preparations. A relatively poor signal-to-noise ratio impeded to find two peaks at 4142.6 and 4380.2 cm⁻¹ reported previously for the vCH + vC—O and vOD + vC=O modes in an argon matrix [3].

4.3. Anharmonicity and coupling constants

Table 2 shows coupling and anharmonicity constants of for *trans*- and *cis*-HCOOD which are calculated by the formulas [15]:

$$\boldsymbol{x}_{ij} = \boldsymbol{v}_{i+j} - \boldsymbol{v}_i - \boldsymbol{v}_j \tag{1}$$

$$\mathbf{x}_{ii} = \frac{1}{2} (\mathbf{v}_{2i} - 2\mathbf{v}_i). \tag{2}$$

Here v_i and v_j are the fundamental frequencies, v_{2i} is the overtone frequency, v_{i+j} is the combination frequency, x_{ij} is the coupling constant, and x_{ii} is the anharmonicity constant. The corresponding results for *trans*- and *cis*-HCOOH were previously reported by Maçõas et al. [3].

Table 2

Experimental and CC-VSCF calculated anharmonicity and coupling constants (in cm⁻¹) for trans- and cis-HCOOD in gas phase, argon and neon matrices.

	trans-HCOOD				cis-HCOOD		
	Ar	Ne	Gas ^a	Calc	Ar	Ne	Calc
Combinations	Coupling constants (x_{ii})						
OCO–CODdef. + vOD	-5.5	-6.4		-42.0			-58.5
OCO–CODdef. + vC–O	-5.7	-4.8	-5.2	-3.8			-1.7
δ COD + OCO—CODdef.		-2.7		+5.1	-0.9	-2.0	+5.5
$\delta \text{COD} + \omega \text{CH}$	+1.4	+0.2		-1.1			0.0
$\delta COD + vC - O$	-8.2	-8.2		-6.8	-9.4	-10.6	-8.7
$\delta \text{COD} + \tau \text{COD}$	-47.6	-45.0		-29.2			-51.5
$\delta COD + \nu C = 0$	-21.2	-19.0		-3.8	-21.1	-31.3	-2.7
ν C — O + ω C H	-23.4	-21.3		-3.2			-3.5
vC-O + vOD	-65.1	-79.7		-34.6	-91.2	-104.3	-57.9
vC-O + vCH	-2.8	-5.0	-4.9	-2.1	-9.5	-5.8	-0.8
vC=O + vC-O	-6.6	-12.4		-0.2			-0.6
$vC=O + \omega CH$	-26.7	-29.7		-8.1			-8.1
vC=O + vOD	-1.6	-3.0	-1.0	-31.1			-48.7
vC=O + vCH	-6.1	-8.7	-13.0	+0.2			-2.7
$vC=O + \gamma CH$	-3.5			-5.9			-6.2
$vOD + \tau COD$	-1.4	-1.7		-18.7			-30.7
$vOD + \omega CH$		-67.5		-30.2		-95.4	-44.4
Overtones	Anharmonicity constants (x_{ii})						
2vCH	-60.4	-67.3		-57.2	-98.9	-98.4	-15.9
2vOD	-44.9	-44.3	-40.9	-43.8	-43.9	-45.3	-35.1
2vC=0	-9.35	-9.35		-9.35	-13.75	-7.3 ^b	-12.6 ^b
2vC—0	-7.0	-6.4		-8.2	-5.15	-8.1	-8.5
2∂COD	+4.2	+4.2		-3.7			-7.9
2τCOD	-1.1	-2.1	-1.6	-17.0			-17.3

^a From Refs. [7,11,26,27,29].

^b Values for the most intensive component of vC=O band (1798.9 in Ne and 1799.3 cm⁻¹ in Ar).

Table 3

CC-VSCF vibrational frequencies (cm^{-1}) of non-observed absorptions for the *trans*and *cis*-HCOOD. Calculated intensities in km/mol are given in parentheses.

Assignment	trans-HCOOD	cis-HCOOD
OCO–CODdef + τ COD	1038.7 (0.2)	950.0 (0.0)
20C0–CODdef	1107.0 (0.1)	1231.4 (0.0)
ω CH + τ COD	1533.2 (0.1)	1378.9 (0.2)
OCO–CODdef + ω CH	1588.4 (0.0)	1634.8 (0.0)
$vC-O + \tau COD$	1653.7 (0.0)	1479.7 (0.2)
γ CH + τ COD	1877.2 (0.0)	1761.5 (0.0)
OCO–CODdef + γ CH	1938.3 (0.0)	2026.9 (0.0)
$vC=O + \tau COD$	2245.6 (0.1)	2145.6 (0.0)
γ CH + δ COD	2354.1 (0.0)	2303.2 (0.1)
$vC-O + \gamma CH$	2542.8 (0.0)	2540.8 (0.1)
2γCH	2760.9 (0.7)	2980.1 (0.9)
v CH + τ COD	3447.4 (0.0)	3220.0 (0.1)
vCH + OCO–CODdef	3509.6 (0.1)	3481.8 (0.1)
$vOD + \delta COD$	3535.4 (2.0)	3492.5 (0.3)
$vCH + \omega CH$	3923.4 (0.0)	3816.5 (0.1)
v CH + δ COD	3926.9 (0.0)	3760.7 (0.1)
$vOD + \gamma CH$	3975.5 (0.1)	4035.1 (0.0)
νCH + γCH	4275.8 (0.4)	4208.0 (0.1)

It is seen from Table 2 that the CC-VSCF calculations reproduce very well the anharmonic properties of the overtone transitions (with an exception of the 2τ COD mode) although they are somewhat less precise for the *cis* form. The absolute value of the anharmonicity constant for the 2ν OD band is about two times smaller than for 2ν OH [3], and the agreement with the gas-phase data is good.

The calculated coupling constants show poorer agreement with the experimental results. For most of the bands (δ COD + ν C=O, ν C-O + ν OD, etc.) the experimental coupling constants for both matrix hosts and the gas phase (when available) are larger than the predicted value, which demonstrate that the calculations underestimate the anharmonicity effect for HCOOD. The calculated coupling constants appear overestimated only for the OCO-COD-def. + ν OD and ν C=O + ν OD combination modes. The opposite trend was observed for HCOOH, where most of the coupling constants were overestimated [3].

The COD bending mode has unusual properties. The 2δ COD overtone has positive experimental anharmonicity constant (+4.2 cm⁻¹ for both matrices) whereas the calculations predict a negative value of -3.7 cm⁻¹. The δ COD + ω CH combination band also has positive experimental coupling constants, and two combinations (δ COD + OCO—CODdef and ν C=O + γ CH) demonstrate the positive values of calculated coupling constants. We assume that the Fermi resonance of δ COD and the overtone of the τ COD mode in *trans*-HCOOD [27] might be at least partly responsible for the peculiarities of the δ COD mode.

Except for the vC=O + vCH vibration, no significant differences between experimental anharmonicity and coupling constants in matrices and the gas phase are observed which means that the geometries and the electronic properties of the molecules are not significantly perturbed by matrix hosts.

5. Conclusions

The vibrational spectra of *trans*- and *cis*-HCOOD in neon and argon matrices are studied in the 7900–450 cm⁻¹ region. Most of the

relatively strong combination and overtone modes predicted by the CC-VSCF method are detected in matrices. The fundamental frequencies and intensities are well reproduced by the CC-VSCF calculations. The calculated combination and overtones frequencies show sometimes a considerable shift from the experimental values up to ~50 cm⁻¹, nevertheless, the relative error does not exceed ~3%. The near- and mid-IR data for *cis*-HCOOD in a neon matrix is reported here for the first time.

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