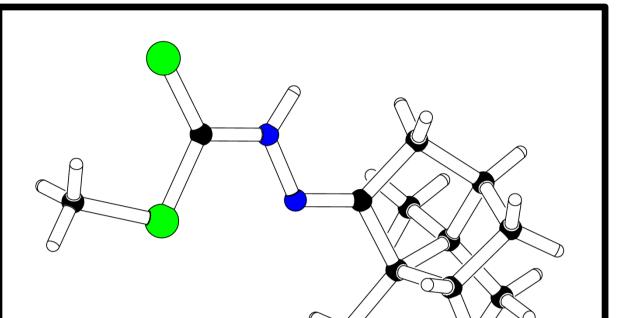


OF STRUCTURAL DIVERSITY AND CRYSTAL PACKING: A CASE STUDY

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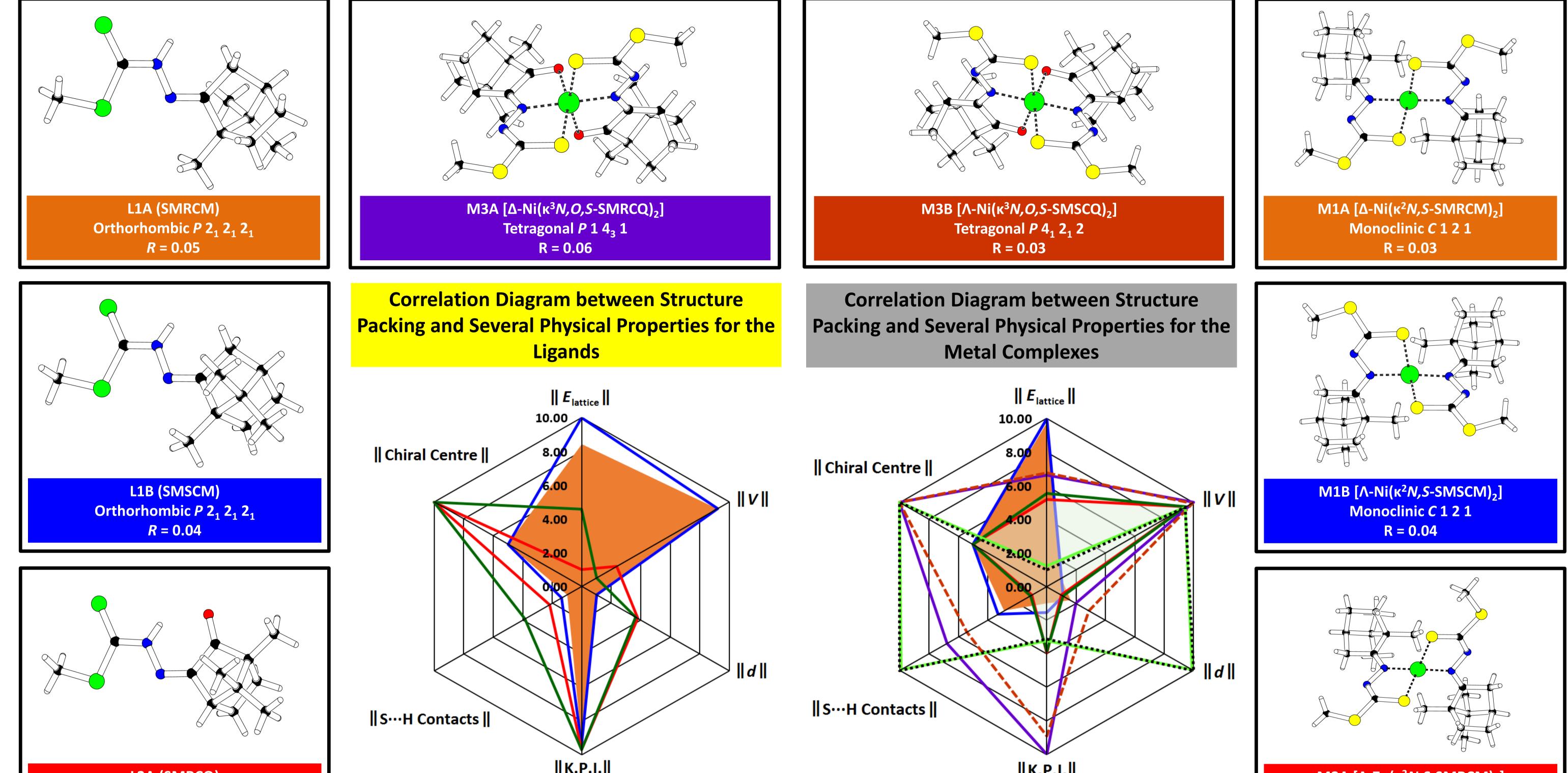
L2A

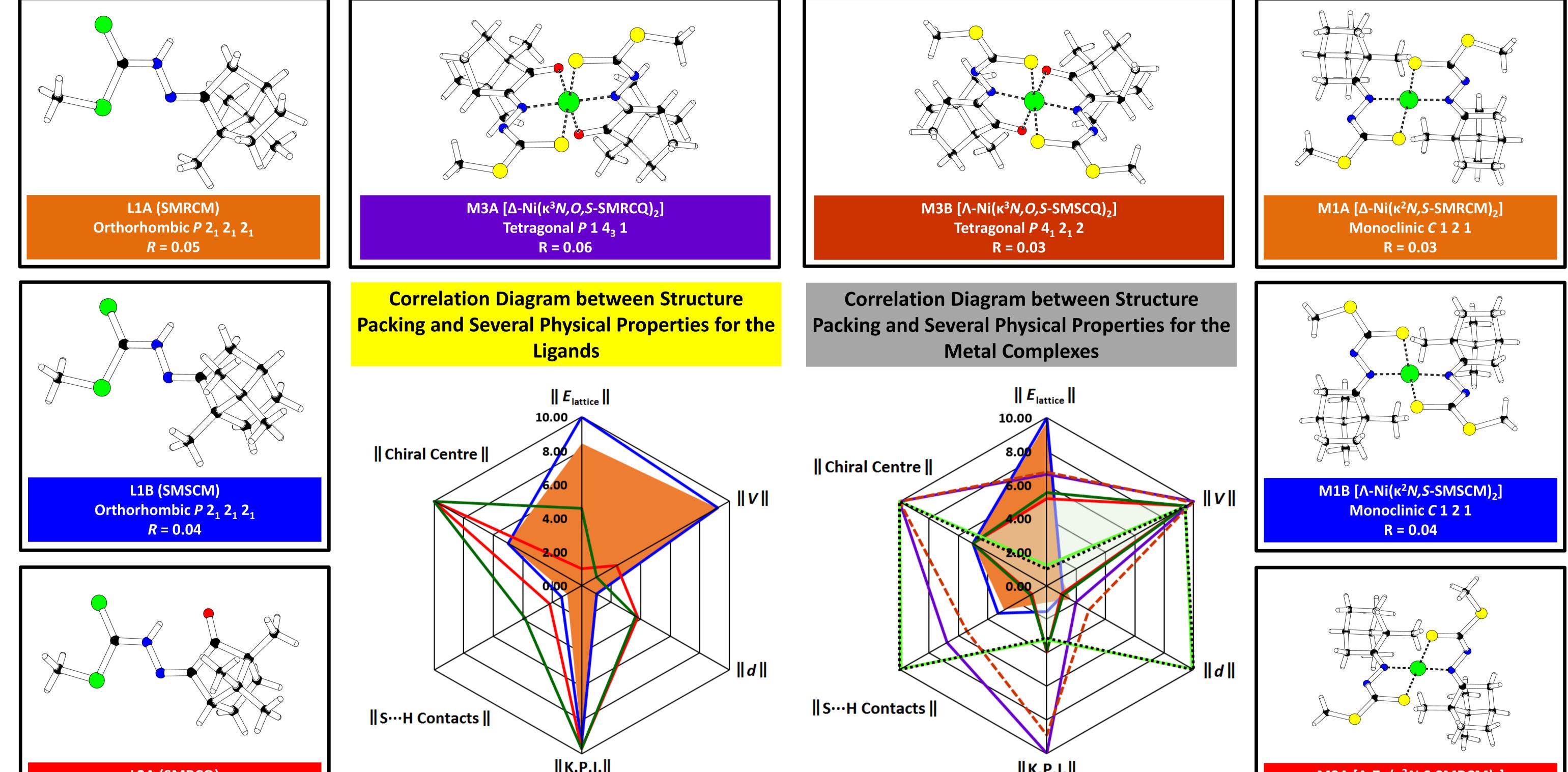
1R,4S

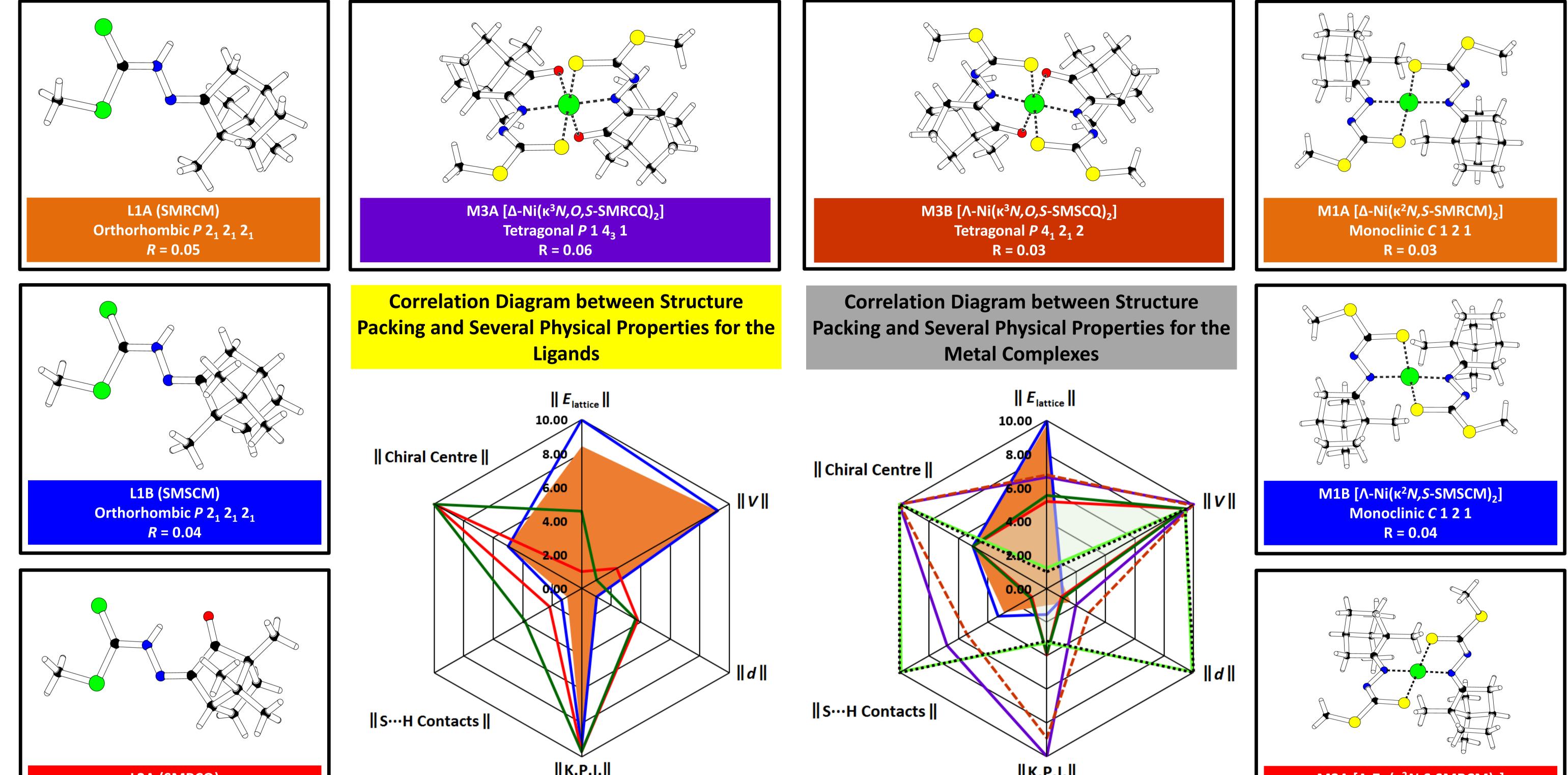
What factors

determine crystal

packing?







L2A (SMRCQ) Triclinic P 1 R = 0.03	K.P.I. L1A (Orthorhombic) L1B (Orthorhombic) L2A (Triclinic) L2B (Monoclinic)	IK.P.I.IIM1A (Monoclinic)M1B (Monoclinic)M2A (Monoclinic)M2B (Monoclinic)M3A (Tetragonol)M3B (Tetragonal)M4A (Orthorhombic)M4B (Orthorhombic)	$\begin{tabular}{lllllllllllllllllllllllllllllllllll$		
L2B (SMSCQ) Monoclinic P 1 2, 1 R = 0.03	$\begin{tabular}{lllllllllllllllllllllllllllllllllll$	M4B [Δ-Zn(κ2N,S-SMSCQ)2]Orthorhombic P 21 21 21R = 0.04			
Code Chiral Z Total Cell Der Centre Z E _{lattice} Volume	nsity Molecular Volume Area K.P.I. Relative Composition of Contacts H···H S···H N···H Other	Summary			
(kJ mol ⁻¹) (Åm³) (g c	m ⁻¹) (Å ³) (Å ²) (%) (%) (%) (%)	For ligands, their crystal packing is found to be determined by the relative composition of			
L1A 1R,4R 8 -118.4 2715.42 1.2	254 332.84 296.24 67.1 57.2 21.0 5.8 3.0	contacts, density as well as number of			
L1B 1S,4S 8 -115.8 2720.00 1.2	252 333.41 296.47 67.1 58.5 21.3 5.9 3.0	stereogenic centre. The <i>E</i> _{lattice} for crystal system is in the order of orthorhombic < monoclinic <			

15.7

7.1

✤ For metal complexes, the crystal packing is influenced by the relative composition of close

triclinic.

L2B	1S,4R	2	-124.7	689.14	1.303	337.73	303.93	67.2	53.8	23.4	7.7	15.0
M1A	1R,4R	2	-205.6	1391.69	1.359	686.88	509.40	65.9	61.9	25.6	4.9	3.4
M1B	1S,4S	2	-201.0	1395.81	1.355	688.89	510.06	65.9	61.9	25.4	4.9	3.5
M2A	1R,4R	4	-278.6	2825.73	1.354	697.57	506.42	66.2	56.8	27.5	4.6	3.1
M2B	1S,4S	4	-273.0	2824.40	1.355	697.23	506.57	66.3	56.7	27.6	4.6	3.1
МЗА	1R,4S	4	-255.4	2914.19	1.362	719.28	515.49	64.3	58.7	28.6	3.9	8.8
МЗВ	1S,4R	4	-253.3	2898.35	1.369	715.35	513.36	64.3	58.3	28.6	4.1	9.0
M4A	1R,4S	4	-342.9	2815.47	1.425	694.92	520.38	67.3	57.8	26.4	3.7	12.0
M4B	1S,4R	4	-346.2	2814.19	1.426	694.59	520.57	67.3	57.8	26.5	3.7	12.1

337.10

303.64

67.2

55.3

22.0

-130.6 1031.22 1.306

contacts and density. The E_{lattice} for crystal system is in the order of monoclinic < tetragonal < orthorhombic.

Overall, it is observed that the relative ** composition of close contacts and crystal density are the most influential factors in determining the packing of a crystal system. Greater cell density and relative composition of close contacts generally lead to greater lattice energy.



Understanding of packing behaviour may help in successful crystal structure prediction!





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