

## **Supplementary information**

# **Electrochemical multi-tagging of cysteinyl peptides during microspray mass spectrometry: numerical simulation of consecutive reactions in a microchannel**

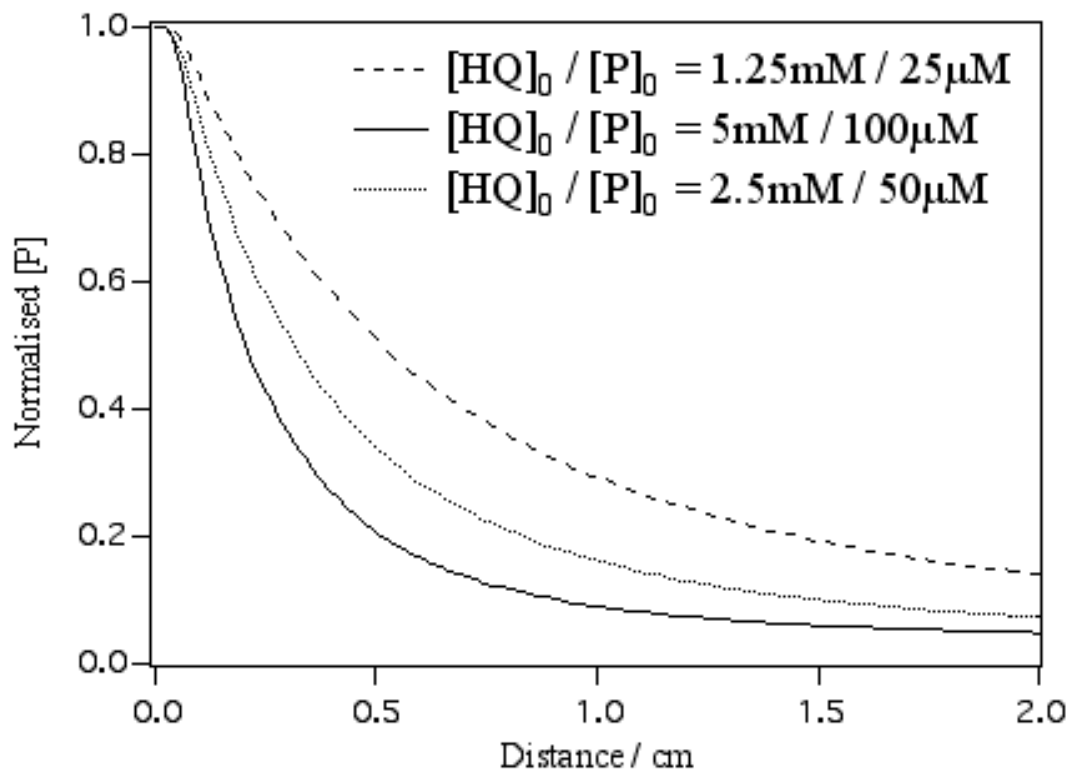
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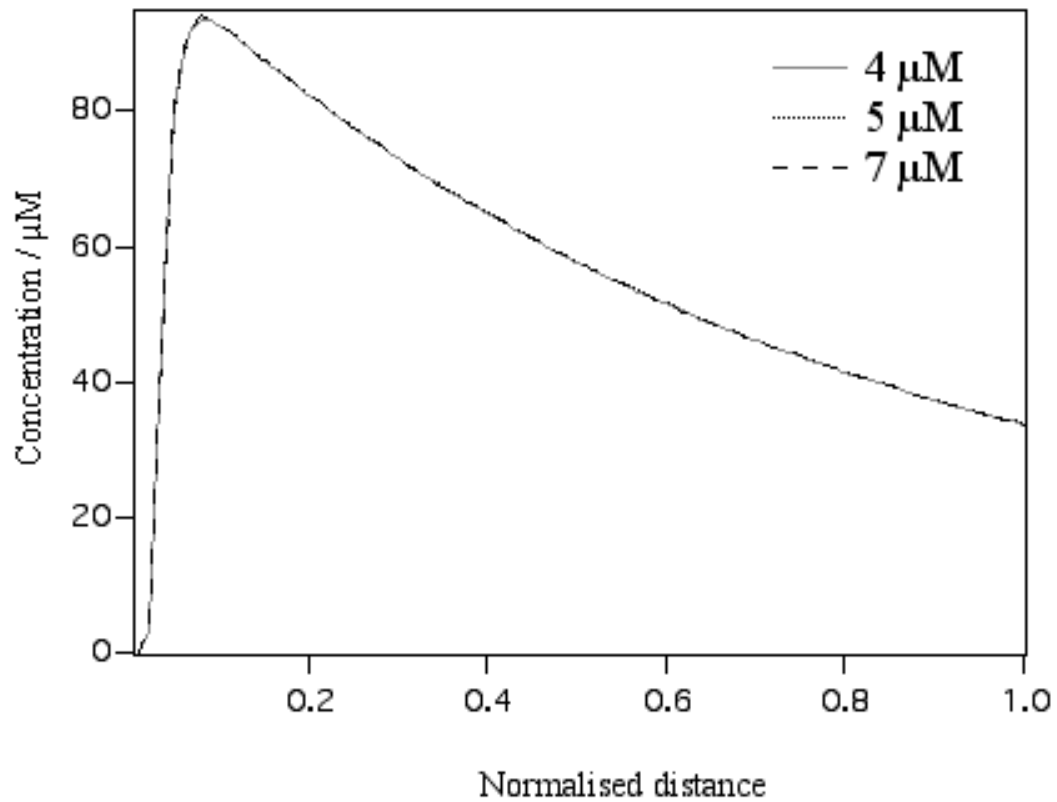
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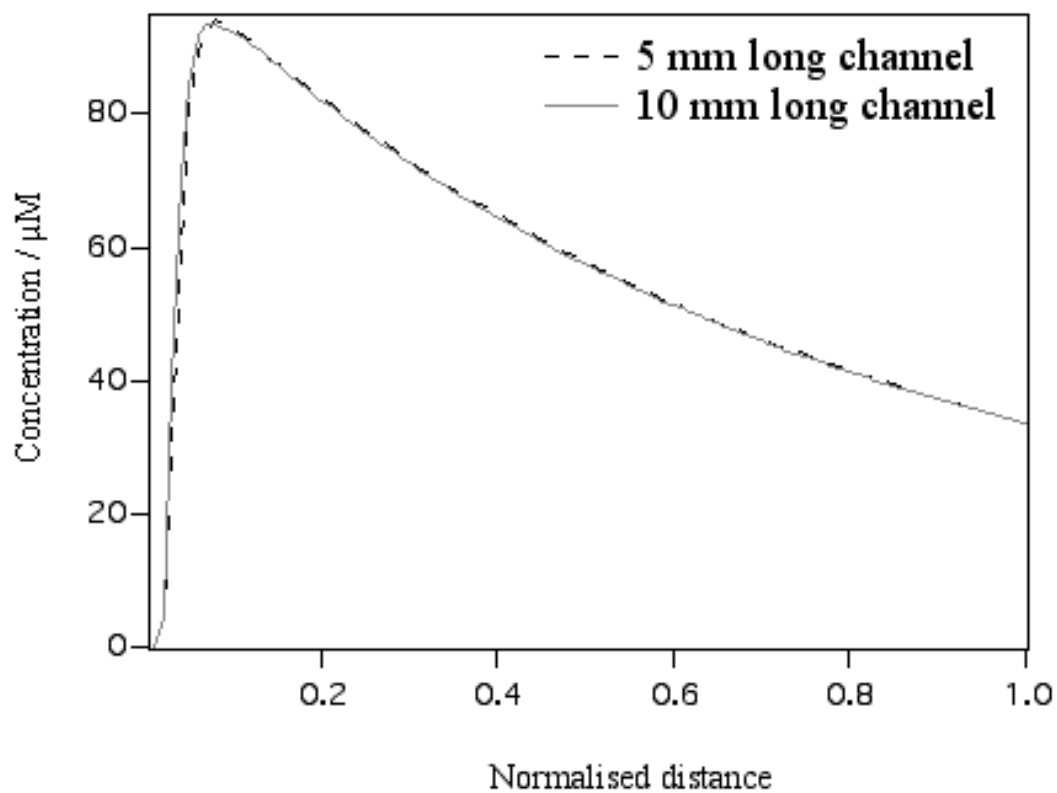
E-mail: [hubert.girault@epfl.ch](mailto:hubert.girault@epfl.ch)



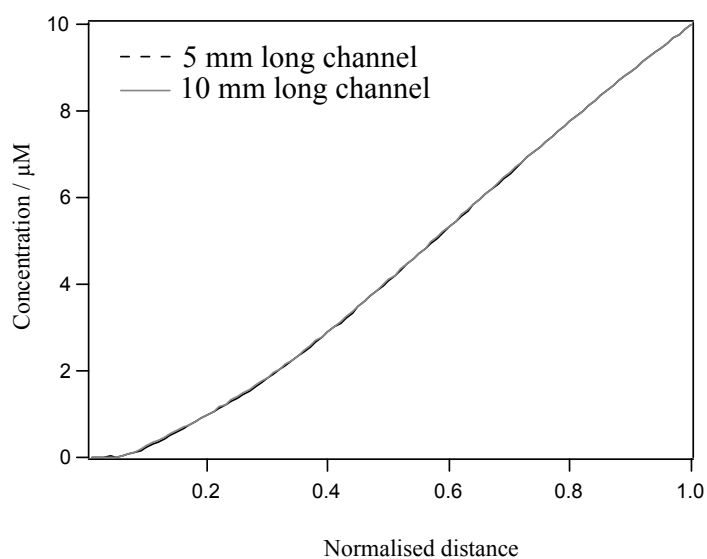
**Figure S1.** Kinetics of the tagging (evolution of  $[P] / [P]_0$ ) for several initial concentrations of HQ and P in the case of a three-cysteine-containing biomolecule. The ratio  $[HQ]_0 / [P]_0$  is kept constant.



**Figure S2.** Meshing validation by comparison of BQ concentration along the channel with main mesh size of  $4 \mu\text{m}$ ,  $5 \mu\text{m}$  and  $7 \mu\text{m}$ .



**Figure S3.** Scaling validation by comparison of BQ concentrations along the channel for a scaling by 2 (channel length of 10 mm) and by 4 (channel length of 5 mm). The flow rate of the fluid was adapted to conserve the flux of species.



**Figure S4.** Scaling validation by comparison of  $PQ_3$  concentrations (three-cysteine-containing target simulation) along the channel for a scaling by 2 (channel length of 10 mm) and by 4 (channel length of 5 mm). The flow rate of the fluid was adapted to conserve the flux of species.

HQ	BQ	P	PQ <sub>1</sub>	PQ <sub>2</sub>	PQ <sub>3</sub>	PQ <sub>4</sub>	PQ <sub>5</sub>
$D_Q \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$ $+\alpha k_{ox} \beta$	$-\alpha k_{red} \beta$						
$-\alpha k_{ox} \beta$	$D_Q \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$ $+\alpha k_{red} \beta$ $+\alpha k_1 c_P \beta$ $+\alpha k_2 c_{PQ_1} \beta$ $+\alpha k_3 c_{PQ_2} \beta$ $+\alpha k_4 c_{PQ_3} \beta$ $+\alpha k_5 c_{PQ_4} \beta$						
		$D_P \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$ $+\alpha k_1 c_{BQ} \beta$					
		$-\alpha k_1 c_{BQ} \beta$	$D_P \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$ $+\alpha k_2 c_{BQ} \beta$				
			$-\alpha k_2 c_{BQ} \beta$	$D_P \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$ $+\alpha k_3 c_{BQ} \beta$			
				$-\alpha k_3 c_{BQ} \beta$	$D_P \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$ $+\alpha k_4 c_{BQ} \beta$		
					$-\alpha k_4 c_{BQ} \beta$	$D_P \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$ $+\alpha k_5 c_{BQ} \beta$	
						$-\alpha k_5 c_{BQ} \beta$	$D_P \nabla \alpha \nabla \beta$ $+\alpha \nu \nabla \beta$

**Figure S5.** The  $M_{ij}$  steady state matrix is described. All the gradients are written in a nabla form  $\nabla$ . The function  $\beta$  is the interpolation function of the unknown. The position in the matrix of the reaction terms is chosen to maximize the weight of the diagonal.