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Current status of direct pulp-capping materials for permanent teeth

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Direct pulp-capping is a method for treating exposed vital pulp with dental material to facilitate the formation of reparative dentin and to maintain vital pulp. Two types of pulp-capping materials, calcium hydroxide and mineral trioxide aggregate, have been most commonly used in clinics, and an adhesive resin has been considered a promising capping material. However, until now, there has been no comprehensive review of these materials. Therefore, in this paper, the composition, working mechanisms and clinical outcome of these types of pulp-capping materials are reviewed.

Keywords: Pulp-capping, Calcium hydroxide, Mineral trioxide aggregate (MTA), Methyl methacrylate-tributylborane (MMA-TBB) resin, Artificial dentin bridge

INTRODUCTION

There are three causes of vital pulp exposure: caries, mechanical sources and trauma. If pulp exposure occurs before caries is completely removed, it is considered caries exposure. If pulp exposure occurs during the preparation of a cavity without caries, it is called mechanical exposure. Mechanical exposures are typically due to a misadventure during tooth preparation. Traumatic pulp exposure may result from a sports injury when the coronal part of the tooth is chipped. In the event of exposure in vital pulp, direct pulp-capping, pulpotomy or pulpectomy could be the treatment choices.

Direct pulp-capping is a treatment for exposed vital pulp involving the placement of a dental material over the exposed area to facilitate both the formation of protective barrier¹⁻³⁾ and the maintenance of vital pulp^{4,5)}. From a more precise clinical perspective, direct pulp-capping is a clinical technique that lies between indirect pulp-capping and pulpotomy. Indirect pulp-capping is a procedure in which a material is placed on a thin partition of remaining dentin where no vital pulp exposure occurs. Pulpotomy differs from pulp-capping only in that a portion of the remaining pulp is removed before the capping material is applied. Accordingly, direct pulp-capping has been used as an alternative approach to the maintenance of vital pulp, thereby avoiding as many as 22 million annual definitive root canal treatments in the United States⁶⁾. Of these cases, several million fail due to the recurrence of symptoms or through the detection of periradicular disease^{7,8)}. Stanley⁹⁾ and Bender¹⁰⁾ hypothesized that many tooth extractions and root canal treatments could have been avoided through the conservative approach of direct pulp-capping.

Clinical pulp conditions related to patient symptoms are to be considered before the direct pulp-capping

material placement. For evaluating clinical pulp conditions, the most important test is pulp vitality. If the pulp vitality test is negative, pulp necrosis is diagnosed. If the pulp vitality test is positive, then we call the pulp vital pulp. A vital pulp can be divided into three different categories depending on the clinical symptoms: normal pulp, reversible pulpitis, and irreversible pulpitis. Normal pulp has no clinical symptoms. Reversible pulpitis usually has a short-lived thermal sensitivity, which will disappear immediately once the thermal stimulation is removed. Irreversible pulpitis usually has spontaneous and/or lingering pain and it could also have referred pain. Pulp-capping could be performed on tooth with normal pulp or reversible pulpitis. Percussion, palpation, and periodontal probing test results should be within normal limits. The radiograph should show normal apical tissue. The pulp exposure site should be less than 1 mm in diameter and stopping pulpal hemorrhage should be prerequisite before direct pulp-capping material placement. If these requirements cannot be satisfied, the pulp-capping procedure is not recommended.

This review summarizes the current status of direct pulp-capping materials.

BRIEF HISTORY OF DIRECT PULP-CAPPING MATERIALS

The first documented pulp-capping treatment was conducted in 1756 by Pfaff, using gold foil¹⁾. Since then, many agents have been recommended for direct pulp-capping^{11,12)}. However, due to insufficient or inappropriate pre-treatment diagnoses, necrotic pulp was historically capped even though it was contraindicated¹¹⁾.

In 1930, Hermann^{13,14)} discovered that calcium hydroxide is effective in repairing an exposure site. Since then, calcium hydroxide in the form of powder,

paste and cement has been used with clinical success for facilitating the formation of reparative dentin along with the maintenance of vital pulp, the induction of mineralization and the inhibition of bacterial growth^{15,16}. Calcium-hydroxide-based cement was patented in 1962¹⁷, and the first clinical study of Dycal (Dentsply Caulk, Milford, DE, USA) was reported in 1963, with a success rate of 85% compared with that of 80% for the control calcium hydroxide mixed with saline¹⁸.

Glass and his colleagues⁴ introduced zinc oxide eugenol for direct pulp-capping. However, chronic inflammation and a lack of pulp healing were observed, with no dentin bridge formation. It was later reported that eugenol is highly toxic, and zinc oxide eugenol resulted in high interfacial leakage^{19,22}.

In the 1970s, glucocorticoids combined with antibiotics were frequently used in an attempt to control pulpal pain and suppress pulpal inflammation^{19,20,23}. Reports of poor wound-healing and even pulpal necrosis emerged, so steroids are no longer used for direct pulp-capping.

For direct pulp-capping, the use of biological molecules, such as growth factors and extracellular matrices, is considered²⁴. For example, animal studies showed that growth factors such as bone morphogenetic proteins (BMP) and transforming growth factors (TGF) induced reparative dentin formation^{21,24-27}. However, these growth factors are not adequately therapeutic, since they produce a porous osteodentin with tunnel defects²⁴. Extracellular matrix (ECM) dentin molecules, such as bone sialoprotein (BSP)²⁸, matrix extracellular phosphoglycoprotein (MEPE)²⁹, amelogenin²⁴ and dentin phosphophoryn³⁰, have been shown to induce reparative dentin. Capping with ECM molecules is extremely promising, producing a reparative mineralized tissue with structural properties better than those produced in the presence of calcium hydroxide²⁴. Among these, amelogenin is suggested to be most promising as a direct capping material. Implantation of two spliced forms of amelogenin with agarose beads as carriers induced the formation of a homogeneous dentinal bridge or massive pulp mineralization²⁴.

Direct pulp-capping with resin-modified glass ionomer has been successfully reported in animal studies in monkeys^{31,32} and in dogs³³. However, it was also examined in humans³⁴, and no dentin bridge formation was observed in 10 months.

In the 1990s, Torabinejad and White³⁵ introduced, mineral trioxide aggregate (MTA), which is basically a hydraulic Portland cement or calcium silicate and releases calcium hydroxide slowly while setting. MTA has been used clinically with success rates similar to those achieved with calcium hydroxide³⁶. In 2006 and thereafter, MTA-like materials were launched, composed of artificial synthetic calcium silicates instead of Portland cement.

CURRENT DIRECT PULP-CAPPING MATERIALS

Calcium hydroxide

Calcium hydroxide has been the gold standard for pulp-capping. The effect of calcium hydroxide is regarded as the result of the chemical injury caused by the hydroxyl ions. The initial effect of calcium hydroxide applied to exposed pulp is the development of a superficial necrosis. Firm necrosis causes slight irritation and stimulates the pulp to defend and repair to form a reparative dentin bridge through cellular differentiation, extracellular matrix secretion and subsequent mineralization³⁷. While the formation of a dentin bridge has been believed to be the key for the clinical success of direct pulp-capping, it has been reported that 89% of dentin bridges formed by calcium hydroxide cement in monkeys contained tunnel defects³⁸. These tunnel defects that form in the heterogeneous dentin bridge not only fail to provide a permanent barrier, but also fail to provide a long-term biological seal against bacterial infection. Another disadvantage of calcium hydroxide is dissolution³⁹. This may lead to the formation of a dead space⁴⁰ and microleakage³⁹.

1. Aqueous calcium hydroxide

Historically, calcium hydroxide powder was applied directly onto the exposed pulp surface. The powder comes into contact with pulpal fluid and forms a paste^{41,42}. This technique is not widely used at the present time. In a study in dogs, Eleazer *et al.*⁴³ reported that calcium hydroxide powder in contact with the pulp caused an inflammatory response. Pereira *et al.*⁴⁴, also in a dog pulp study, reported no differences in pulpal responses to direct pulp-capping achieved with either paste or powder forms in a 120-day period. Aqueous calcium hydroxide paste is used for direct pulp-capping⁴⁵⁻⁴⁷. This paste is generally prepared by the mixing of calcium hydroxide powder and water or saline at the time of application in the clinic. Premixed types of the paste indicated for direct pulp-capping are commercially available, such as UltraCal XS (Ultradent Products, South Jordan, UT, USA) and Calcicur (Voco, Cuxhaven, Germany), which also contain barium sulfate for radiopacity and other ingredients to enhance the properties of the material.

The success rates of direct pulp-capping with calcium hydroxide decrease as follow-up periods increase. Rates are more than 90% after 1 to 2 years and drop from 82% to 37% after 2 to 5 years⁴⁸, or to 80%, 68% and 59% after 1, 5 and 9 years, respectively⁴⁹.

Although aqueous calcium hydroxide has been well-accepted clinically, it has drawbacks, including a lack of setting properties and gradual resorption after placement. Another disadvantage is porosities in the newly formed dentin, known as tunnel defects, which can result in microleakage and lead to the loss of tooth vitality and calcification³⁶.

2. Calcium-hydroxide-based cement

Because of the disadvantages of aqueous calcium hydroxide described above, a cement type of calcium

hydroxide with setting characteristics was developed and has been widely used in clinical practice since the 1960s.

The most popular commercial cement is Dycal, which consists of catalyst and base mixed at a 1:1 ratio. The catalyst contains calcium hydroxide, N-ethyl-o/p-toluene sulfonamide, zinc oxide, titanium dioxide and zinc stearate, and the base contains 1,3-butylene glycol disalicylate, zinc oxide, calcium phosphate and calcium tungstate. Another product is Life (Kerr, Orange, CA, USA), whose setting reaction mechanism between salicylic acid ester and zinc oxide is similar to that of Dycal, but whose ingredients are different. The base contains calcium hydroxide, zinc oxide and butyl benzene sulfonamide, and the catalyst barium sulfate, titanium dioxide and methyl salicylate.

There are two clinical studies that applied calcium hydroxide cement for direct pulp-capping. Al-Hiyasat *et al.*⁵⁰⁾, using only radiography, evaluated the 3-year treatment outcome of pulp-capping in teeth in terms of both mechanical and caries exposure. The success rate for mechanical exposure was 92% compared with 33% for the caries-exposure cases. In another study, Barthel *et al.*⁵¹⁾, using both radiography and pulp vitality testing, examined the 5- and 10-year treatment outcome of caries exposure in pulp-capped teeth. In this case, the success rates for 5 and 10 years were 37% and 13%, respectively. The majority of the failures were asymptomatic; the pulp tended to become necrotic or slowly calcify. Therefore, direct pulp-capping is considered controversial by many clinicians, and pulpectomy is still the standard procedure for treating caries-exposed inflamed vital pulp with a closed apex. The success rate of pulpectomy is reported to be about 95%^{52,53)}.

Calcium-silicate-based materials

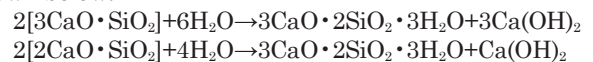
1. Hydraulic cements

1) Mineral trioxide aggregate (MTA)

The original MTA, ProRoot MTA Gray (Dentsply Tulsa Dental Specialties, Johnson City, TN, USA), was marketed in 1998 and was composed of 75% Type I Portland cement, 20% bismuth oxide and 5% calcium sulfate dihydrate. The Portland cement is composed of approximately 55 wt% tricalcium silicate ($3\text{CaO}\cdot\text{SiO}_2$), 19 wt% dicalcium silicate ($2\text{CaO}\cdot\text{SiO}_2$), 10 wt% tricalcium aluminate ($3\text{CaO}\cdot\text{Al}_2\text{O}_3$), 7 wt% tetracalcium aluminoferrite ($4\text{CaO}\cdot\text{Al}_2\text{O}_3\cdot\text{Fe}_2\text{O}_3$), 2.8 wt% magnesium oxide, 2.9 wt% sulfate and 1.0 wt% free calcium oxide. Bismuth oxide and calcium sulfate are the radiopacifier and setting modifier, respectively. ProRoot MTA White was introduced in 2002 and differs from its predecessor in composition, *i.e.*, the elimination of tetracalcium aluminoferrite and an increase of calcium silicates. The gray type of MTA, containing tetracalcium aluminoferrite, and with composition similar to that of the original type, is less popular for esthetic reasons, but several products are available, including: ProRoot MTA Gray, MTA Angelus (Angelus, Londrina, Brazil), Grey MTA Plus (Avalon Biomed, Bradenton, FL, USA), EndoCem MTA (Maruchi, Gangwon-do, Korea) and

Ortho MTA (BioMTA, Daejeon, Korea). MTA without tetracalcium aluminoferrite is more popular, and many products are marketed worldwide: ProRoot MTA White, MTA Angelus White, White MTA Plus (Prevest Denpro, Jammu, India), MM-MTA (Micro Mega SA, Besançon, France), MTA Caps (Acteon, Merignac, France), Tech BioSeal MTA (Isasan S.R.L., Rovello Porro, Italy), Aureose 1 M.T.A. (Ogna Laboratori Farmaceutici, Muggiò, Italy), MTA+product (Cerkamed PPH, Wojciech Pawlowski, Nisko, Poland), Trioxident (VladMiVa, Belgorod, Russia), NEX MTA (GC, Tokyo, Japan) and Endo-Eze MTA (Ultradent Products), among others.

The mechanism of action of MTA is similar to that of calcium hydroxide. The calcium hydroxide produced as a by-product of hydration of MTA is leached out and causes necrosis when in contact with the pulp. When MTA powder is mixed with water at the time of application, calcium silicates in the powder hydrate to produce a calcium silicate hydrate gel and calcium hydroxide, as shown below:



Thus, MTA can be described as a calcium-hydroxide-releasing material and, therefore, is expected to present various properties similar to those described above for calcium hydroxide.

The advantages of MTA are believed to be its sealing ability, biocompatibility, bioactivity and capacity to promote mineralized tissue formation⁵⁴⁻⁵⁷⁾. Also, MTA is suggested to be superior to calcium hydroxide due to its more uniform and thicker dentin bridge formation, less inflammatory response and less necrosis of pulpal tissue^{55,56,58-63)}.

An antibacterial effect of MTA is controversial as reviewed by Parirokh and Torabinejad⁶⁴⁾. MTA showed an antibacterial effect on some of the facultative bacteria but no effect on any of the strictly anaerobic bacteria⁶⁵⁾. MTA demonstrated, in some cases, albeit inferior to calcium hydroxide or zinc oxide/eugenol paste⁶⁶⁻⁷⁰⁾. Taken together, the antimicrobial activity of MTA may not be as strong as those of traditional calcium hydroxide-based cements and sealers⁷⁰⁾.

In spite of its many positive properties, some disadvantages of MTA include long setting times, poor handling⁷¹⁻⁷³⁾, and coronal tooth discoloration⁷⁴⁻⁷⁷⁾. Reported setting times have shown variations: 50 min⁷³⁾, less than 4 h⁷⁸⁾ and 70 and 175 min for the initial and final setting times⁷⁹⁾, respectively. Setting time of MTA Gray (165 min) is shorter than that of MTA White⁶⁴⁾. A long setting time may be inconvenient to both dentist and patient, because it requires direct pulp-capping with MTA in two visits: application of MTA in the first visit and seating of the permanent restoration over the sufficiently hardened MTA in the second visit. Moreover, it may increase the risk of bacterial contamination. Short setting times will make it possible for treatment to be performed in one visit.

As for the handling characteristics, the “sandy”-feeling mixture produced by the coarse particles of ProRoot and water is difficult to be delivered to the

required site and hard to condense adequately. Setting time and handling properties can be affected by the particle size and distribution as well as by the shape of the MTA powder. The particle sizes of MTA are reported to be from 1 to 10 μm ⁸⁰. Angelus White and Gray have median particle size below 10 μm , but contain many particles more coarse than 40 μm , up to 100 μm . Comparisons of particle size and shape were reported among ProRoot MTA, MTA Angelus and ordinary Portland cement. The Angelus particles had relatively low circularity and wide size distribution and were less homogeneous than ProRoot MTA, and ProRoot MTA Gray had many similarities to Portland cement in particle size and distribution. Some particles of MTA were as small as 1.5 μm ⁸¹. Size distribution in ProRoot Gray is greater than in the White^{82,83}.

Tooth discoloration has been reported with the use of gray MTA in direct pulp capping⁷⁴ and therefore the use of white MTA has generally been recommended in the esthetic zone. However, tooth discoloration associated with white MTA was also described in case reports in endodontic treatments^{75,76}. Tooth color change was reported to be induced by both gray and white MTA *in vitro*⁷⁷. Several factors were reported to contribute to tooth discoloration by white MTA: contamination with blood^{84,85}, contact with sodium hypochlorite⁸⁶, the presence of light and oxygen^{87,88}. The possible involvement of the radiopacifier bismuth oxide in the discoloration is postulated⁸⁷. The reason and mechanism of tooth discoloration are not fully understood and remain to be investigated.

MTA cannot bond to dentin. Therefore, there is a risk of bacterial leakage, which could lead to failure of endodontically treated teeth. The sealing ability of MTA as a root-end filling material was evaluated and compared with that of other materials *in vitro* by several methods. In dye penetration tests, the ascending order of leakage was reported as: MTA<glass-ionomer cement<Super EBA (Bosworth Company, Skokie, IL, USA) [reinforced zinc oxide/eugenol (32%)/*o*-ethoxybenzoic acid (68%) cement]<amalgam<IRM (Intermediate Restorative Material, Dentsply Caulk, Milford, DE, USA) (reinforced zinc oxide-eugenol cement)⁸⁹⁻⁹¹. A bacterial microleakage model study showed: composite resin (Prisma TPH, Dentsply Caulk)<amalgam+bonding agent (Probond Primer & Adhesive, Dentsply Caulk)<EBA, amalgam and MTA⁹². Another bacterial leakage study indicated that MTA and EBA leaked, and that leakage increased with time⁹³. The results of a fluid transport model study were: AH26 (Dentsply Maillefer, Ballaigues, Switzerland) (negative control)<<MTA<glass-ionomer cement<EBA<amalgam⁹⁴. The results of marginal adaptation measurement indicated that MTA was better than IRM and Super EBA⁹⁵. A dye leakage study in the orifice of a root canal system indicated: composite resin (Tetric, Ivoclar Vivadent, Schaan, Liechtenstein)<MTA<Cavit (3M ESPE, St. Paul, MN, USA) [zinc oxide/ethylene bis(oxyethylene)diacetate cement]⁹⁶.

When all these results are summarized, it may

be concluded that MTA is better than glass-ionomer cement, EBA cement, amalgam and IRM, but inferior to composite resin and AH26, which is a root canal sealer consisting of epoxy resin, methenamine (hexamethylenetetramine) and bismuth oxide. Thus, it is suggested that the seal provided by MTA should probably be more leakage-proof⁹⁷.

Animal direct pulp-capping studies comparing MTA with calcium hydroxide generally indicate better pulp-healing with MTA than with calcium hydroxide^{59,60,62,98-102}. These studies consistently demonstrated more hard-tissue bridge formation and less inflammation in the MTA group compared with the calcium hydroxide group. Also in human studies, many reports agreed that MTA is better than Dycal, with better hard-tissue formation and less pulp tissue inflammation^{61,97,103,104}. The biological mechanism by which MTA induces dentin bridge formation is currently unknown. The predictable formation of a quality hard-tissue barrier subjacent to MTA is likely to be multifactorial, involving its sealing ability, biocompatibility and the production of an alkaline pulpal environment⁹⁷.

As for clinical outcome of direct pulp-capping with MTA, Miles and colleagues¹⁰⁵ reported on caries-exposed permanent teeth with closed apex. Overall, the one-year pulp survival was 68%, while the two-year survival rate was 56%. A recent study⁸⁷ showed that MTA and calcium hydroxide had a successful outcome of 78% and 60%, respectively. Teeth that were permanently restored within 2 days after being capped had a better prognosis, and there was no difference between mechanical and caries pulp exposure¹⁰⁶. Regarding caries-exposed pulp with an open apex in young immature tooth, MTA showed high clinical success in both primary¹⁰⁷ and permanent teeth^{74,108} in periods ranging from six months to four years.

2) Modified MTAs and MTA-like materials

Some modified MTAs overcoming the drawbacks of the original MTA are available, and most of them aimed to shorten setting time by modifying the composition or particle size of the powder. In Angelus White MTA (setting time, 15 min), calcium sulfate was removed and calcium oxide was added to tricalcium silicate, dicalcium silicate, tricalcium aluminate and bismuth oxide¹⁰⁹. In MM-MTA, calcium carbonate was added; in Tech BioSeal MTA, calcium chloride and montmorillonite were added; and the powder in MTA Plus was more finely ground.

MTA-like materials were marketed after 2006. They are not composed of Portland cement, which is manufactured from minerals of natural origin, but consist of synthetic calcium silicates as the main components and are aluminum-free. The difference in origin of calcium silicates is clearly demonstrated in the release of metal ions from the set materials¹¹⁰. In Angelus MTA and MM-MTA, which are based on Portland cement, a large amount of aluminum and trace amounts of arsenic, beryllium, cadmium and chromium were detected, but in DiaRoot Bioaggregate (DiaDent Group International, Cheongju-si, Korea), based on

synthetic calcium silicates, no metals were detected except a trace amount of aluminum.

BioAggregate, marketed in 2006, consists of tricalcium silicate, dicalcium silicate, tantalum pentoxide (radiopacifier), calcium phosphate monobasic (calcium dihydrogen phosphate) and amorphous silicon oxide. Calcium phosphate reacts with part of the calcium hydroxide produced from setting (hydrating) calcium silicates, and during the reaction, hydroxyapatite and water are formed. The water thus produced contributes to the hydration reaction speed. Silicon oxide also reacts with calcium hydroxide by the so-called pozzolanic reaction and thus contributes to setting time. The setting time is within 4 h at a normal optimal powder/liquid ratio (1 g/0.38 mL water). The antibacterial effect was reported to be similar to Dycal and inferior to zinc oxide/eugenol cement¹¹¹.

Biodentine (Septodont, Lancaster, PA, USA), launched in 2009, contains tricalcium silicate, calcium carbonate and oxide and zirconium oxide (radiopacifier) in the powder, which is mixed with calcium chloride solution containing modified polycarboxylate instead of water. Both substances in the liquid contribute to shortened setting times (from 10 to 12 min). Calcium chloride accelerates the hydration reaction, and polycarboxylate reduces the amount of water required for mixing by providing proper consistency, which also contributes to easy handling of the mixture. Calcium carbonate in the powder is expected to act as a nucleation site in the hydrating mass, enhancing the hydration and leading to faster setting. Finer particles in the powder with larger specific surface areas can also contribute to short setting time: the specific surface area of Biodentine was reported to be about 2.8-fold compared with that of MTA Angelus White¹⁰⁹. Biodentine was reported to have efficacy similar to that of MTA in direct capping over mechanically exposed molar pulps. Complete dentinal bridge formation, an absence of inflammatory pulp response and layers of well-arranged odontoblasts and odontoblast-like cells were observed after 6 weeks¹¹².

EndoSequence BC RRM (Brasseler USA, Savannah, GA, USA), introduced in 2009, includes tricalcium silicate, dicalcium silicate, tantalum pentoxide, zirconium oxide, calcium dihydrogen phosphate, calcium hydroxide and thickening agent and is used as a premixed-syringeable paste or putty without being mixed with water. Recently, in 2014, BC RRM-Fast Set Putty has been launched, which is made with a fast-set formula and equipped with a syringe delivery system. In this material, water required to hydrate calcium silicates depends on the presence of a natural source in dentin. The calcium silicates in the powder hydrate to produce a calcium silicate hydrate gel and calcium hydroxide. The calcium hydroxide reacts with the phosphate ions to precipitate hydroxyapatite and water. The water continues to react with the calcium silicates to precipitate additional gel-like calcium silicate hydrate. The water supplied through this reaction is an important factor in controlling the hydration rate and the setting time. Setting time is ~2 h for RRM and 20 min for BC RRM-Fast

Set Putty, according to the manufacturer. However, this appears questionable because the experiment studying the effect of the addition of water to the material showed a tendency for the initial setting time to increase (from about 75 h to 110 h) and the final setting time to decrease (from about 240 h to 170 h) when increasing amounts of water were added (from 1% to 9%)¹¹³. The largest particle size of the powder was 0.35 μm , with approximately 50% of the particles being nano (1×10^{-3} μm) in size¹¹⁴. This material was shown to have cytotoxicity levels similar to those of ProRoot MTA and MTA Angelus¹¹⁴. BC RRM putty has similar *in vitro* biocompatibility to MTA¹¹⁵. It had similar results compared with MTA when used as pulp-capping agents¹¹⁶ and induced the proliferation of dental pulp cells and the formation of reparative dentin bridge¹¹⁷.

2. Resin-modified MTA cement

TheraCal LC (Bisco, Schaumburg, IL, USA) is a light-curing, resin-modified calcium-silicate-filled single paste, containing calcium oxide, calcium silicate particles (type III Portland cement), strontium glass, fumed silica, barium sulphate, barium zirconate and resin consisting of Bis-GMA and polyethylene glycol dimethacrylate¹¹⁸. The formation and leaching of calcium hydroxide were shown to be negligible, and little or no hydration was exhibited¹¹⁸. The inflammatory response was more intense than with MTA Angelus, and this material did not stimulate mineralization¹¹⁹.

Resin-based cements

1. Composite and MMA-based cements

As described above, at present, the sealing ability of the materials based on inorganic compounds used as the clinical standard needs further improvement, and adhesive resins should be helpful in this regard. The effectiveness of adhesives has been demonstrated *in vitro* and *in vivo*. A bonded coronal seal of either core paste (composite resin build-up material) and Tenure adhesives, or amalgam and Panavia, was reported to result in virtually no penetration of the India ink in which the teeth had been immersed for 10 days¹²⁰.

Cox *et al.*¹²¹ investigated the effect on pulp of calcium hydroxide capping and restoration with amalgam for cavities with exposed pulps in monkeys. Half of those pulps showed complete healing, and the remainder presented pulp inflammation of severity varying from localized low-grade accumulations of mononuclear leukocytes to extensive breakdown of the pulp tissue with abscess formation and necrosis after 1 and 2 years. This study demonstrated that recurring pulp inflammation was associated with bacterial contamination and implied the need for effective sealing of the exposure site to prevent marginal leakage of bacteria. Although a dentin bridge was formed by calcium hydroxide, it was reported that the bridge was heterogeneous and contained tunnel defects³⁸. MTA was also reported to form tunnel defects^{59,122}.

For effective tooth sealing, adhesive resins should be helpful. In animal studies¹²³⁻¹²⁶, successful pulp

healing and dentin bridge formations have been reported when adhesive resins were used for direct pulp-capping. However, in the literature, including several histology-based reports of direct pulp-capping for mechanically exposed human teeth, the situation is different. Bonding agents such as All Bond 2 (Bisco)¹²⁷, Clearfil Liner Bond 2 (Kuraray, Tokyo, Japan)¹²⁸, Scotch Bond Multi-Purpose (3M ESPE)¹²⁹, Single Bond (3M ESPE)¹³⁰ and composite resin (Z100) (3M ESPE) were applied to the pulp for periods ranging from 2 to 10 months. Calcium-hydroxide-based cement (Dycal) was used as a control in all cases. These reports concluded that Dycal is better than the resin systems, possibly due to the effect of residual monomer on the vital pulp.

Monomer cytotoxicity can affect vital pulp in the short term, before setting, or in the long term, after setting^{131–133}. Before setting, the cytotoxicity of the monomer itself will affect the pulp, while after setting, the effect of residual monomer contained in the set resin will depend on the amount and elution kinetics of the monomer.

Cytotoxicity testing of 39 monomers used in dental materials revealed that methyl methacrylate (MMA) was least cytotoxic¹³⁴. Other reports indicate similarly the lowest cytotoxicity for MMA^{135,136}. Because of the low toxicity of MMA, MMA-based resin has long been successfully, and widely used for bone cements to anchor artificial joints in orthopedic surgery.

The inferior outcome of the resin systems used for direct pulp-capping compared with Dycal may be due to the high cytotoxicity of the monomers used in those systems. In terms of monomer cytotoxicity, the MMA-based resin should be the best option for a pulp-capping resin. The minimal effect of MMA on pulp tissue has been reported. Pulp tissues removed from rabbit incisors were immersed in MMA for 1 min. The MMA-immersed and the untreated control pulp tissues were autotransplanted beneath the kidney capsule. The MMA-immersed pulp and the untreated control pulp tissue were positive for osteocalcin and presented osteodentin formation at 7 days. This suggested that MMA did not inhibit the osteogenic activity of pulp tissue¹³⁷.

An *in vivo* comparison of MMA-based resins with composite resin has been published by Tronstad and Spångberg¹³⁸. The pulp responses to Bis-GMA-based composite resin (Concise) (3M, St. Paul, MN, USA) and MMA-based resins initiated by sulfinic acid (Sevriton, de Trey, Zürich, Switzerland) or tributylborane (TBB) (Polycap, Ivoclar Vivadent) in deep Class V cavities in monkeys were compared. After 8 days, the degrees and percentages of responses were slight, moderate and severe in this order: 30%, 50% and 20% for Concise; 30%, 20% and 50% for Sevriton; and 75%, 25% and 0% for Polycap, respectively. No severe response was seen in Polycap. Overall, the severity of the pulp response was lowest for Polycap compared with that of Concise and Sevriton. The remarkable difference in the responses between Sevriton and Polycap is noteworthy because both resins are composed basically of MMA monomer,

but the polymerization initiator is different. This suggests that the pulp response to resins is significantly affected by not only the type of resin monomer but also by the polymerization initiator (catalyst). In the paper cited above¹³⁸, no material information on the resins was provided. From the literature, the catalyst of Sevriton is known to be sulfinic acid¹³⁹, but no information on Polycap is available. However, the statement in the discussion citing two papers^{140,141} clearly suggests that Polycap was actually the same material as Palakav (Firma Kulzer, Hanau, Germany) and its premarketing material F1, which were MMA-based resins initiated by TBB (MMA-TBB resin). It was reported that Polycap is a joint product of Vivadent and Kulzer¹⁴².

2. MMA-TBB resin cement

A successful clinical trial of the MMA-TBB resin was reported in 1968, indicating a minimal pulp damage histologically after 9–12 months in unlined cavities of vital teeth by filling F1 (MMA-TBB resin)^{143,144}. A similar favorable clinical result was reported in 1976. No pulpal necrosis or partial pulpitis could be observed after 2–35 months by filling Polycap (MMA-TBB resin) in unlined cavities of vital teeth¹⁴². Thus, MMA-TBB resin seems promising as a pulp-capping resin. Christensen has referred to 4-META bonding agent, which is the MMA-TBB resin containing 4-[2-(methacryloyloxy)-ethoxycarbonyl]phthalic anhydride (4-META), as a clinically successful bonding agent for pulp-capping¹⁴⁵. In Japan, it has been used for direct capping with clinical success by practitioners experienced with the resin. Although clinical reports are scarce^{146–150}, several *in vitro* and *in vivo* studies supporting clinical success have been published.

Cellular activities of rat dental pulp cells cultured on 4-META/MMA-TBB resin (4-META resin for short) (SuperBond, Sun Medical, Moriyama, Japan) were comparable to control plastic plates, suggesting that the resin did not induce cytotoxic responses¹⁵¹. Cell viability of 4-META resin was reported to increase from 66% for fresh material to 100% for the set resin. This suggests that cytotoxicity of the resin was significantly reduced during the course of setting¹⁵².

Evident dentin bridge formation was reported on the surgically exposed dental pulp in germ-free rats after application of 4-META resin¹⁵³. Favorable periapical tissue healing in the rat molar after retrofilling with 4-META resin was reported that the resin produced the least severe inflammatory reaction and the greatest amount of new bone, and thus fostered the natural regeneration of the periapical tissue¹⁵⁴. Investigation of nerve regeneration and proliferative activity in amputated pulp tissue of dogs after the application of 4-META resin or calcium hydroxide revealed that wound-healing of the exposed pulp surface occurred for the resin in a manner similar to that with calcium hydroxide¹⁵⁵.

The best support for clinical success of the resin is provided by Inoue *et al.*, who conducted comprehensive *in vitro* and *in vivo* studies of the resin. Their works have

also been summarized as review articles^{156,157}. Important points from their studies will be referred to below.

Cytotoxicity testing with L-929 cells by the Millipore filter method was carried out by the placement of 4-META resin at 1, 5 and 10 min, and 1 and 24 h after the resin components were mixed on the filter. Bis-GMA resin (Panavia) (Kuraray) was also tested as a control. 4-META resin showed only slight toxic effects up to 10 min, at which time the resin was fully cured. However, the Bis-GMA resin showed moderate toxic effects up to 60 min¹⁵⁸. A cell proliferation test was performed on completely polymerized 4-META and Bis-GMA resins for up to 4 days. The results showed that the proliferation rate of the cells on 4-META resin was slower than that on the control (culture dish), but the cells remained viable over the test periods. The polymerized 4-META resin appeared to cause almost no cytotoxic damage to the cells. No cells were found on the Bis-GMA resin at any experimental period¹⁵⁸.

The 4-META resin was polymerized directly onto human pulp. The tooth with the polymerized resin was treated with 36% hydrochloric acid to remove the pulp tissue underlying the resin and then treated with acetone to remove the polymerized resin. After these treatments, a thin layer of film was obtained, which the authors called the residual soft-tissue hybrid layer (STHL). SEM and TEM observation studies showed the presence of collagen fibers, cells and capillaries in the STHL. Moreover, XPS analysis showed the presence of nitrogen and sulfur on the STHL, indicating that this layer contained pulp tissue components¹⁵⁶. These results clearly demonstrated graft polymerization of MMA to pulp tissue, which means that poly(methyl methacrylate) bonds chemically to pulp tissue. STHL is undoubtedly a graft polymer, although the authors did not refer to the formation of graft polymer. Generally, graft polymer is composed of a main backbone polymer to which different type of the branch polymers is chemically connected through covalent bonds.

In vivo testing in humans was performed in premolars. After preparation of an occlusal cavity, the pulp was exposed and the cavity was filled with 4-META resin. The teeth were extracted at 7 to 294 days and observed histologically. Patients complained of neither pain nor hypersensitivity. The *in vivo* studies showed that: (1) only slight inflammatory cell infiltration was found in some cases in the early stage; (2) dentin bridge formation occurred in half of the experiment cases; and (3) macrophages appeared in some cases in the later stage. Based on these results, the authors concluded that 4-META resin could be used to conserve pulp because the resin had no cytotoxic effects on pulp and maintained a biological seal, which was achieved by the STHL described above. Thus, they stated that STHL might have a protective function on exposed pulp, much like that of a dentin bridge, and therefore might be called an “artificial dentin bridge”^{156,159}. Figure 1 illustrates the formation of a soft-tissue hybrid layer (STHL)¹⁵⁶ or “artificial dentin bridge”¹⁵⁶ in direct pulp-capping with MMA-TBB resin by graft polymerization and interfacial

initiation of a polymerization mechanism¹⁶⁰.

There are four probable reasons for the success of MMA-TBB resin: (1) MMA is least cytotoxic among the monomers used in dentistry¹³⁴⁻¹³⁶; (2) TBB initiator reduces the residual MMA after setting with time^{160,161}; (3) TBB has the capacity to induce interfacial polymerization of MMA at the dentin interface^{160,162}; and (4) TBB causes graft polymerization of MMA onto dentin collagen to produce a graft polymer composed of collagen and MMA polymer^{140,160}.

Reasons (1) and (2) appear to correlate with the results of the studies by Inoue *et al.*^{158,159}, whose results showed that (a) 4-META resin demonstrated cytotoxicity *in vitro* only at the early period after the start of setting of the resin, and (b) the resin caused slight inflammatory cell infiltration in the early stage *in vivo*. It was reported that residual MMA monomer decreased from 8.15% to 1.96%, 0.84% and 0.48% at 30 min, 24 h, 1 and 4 weeks after the start of setting of MMA-TBB resin¹⁶¹. This significant decrease of the residual monomer with time can be correlated with cell responses *in vitro* and *in vivo* as described above. Moreover, dentin bridge formation in the human study was not always observed in all experimental cases, but occurred in half of the cases^{156,159}. The absence of dentin bridge formation in half of the cases may be correlated to a weak inflammatory response caused by 4-META resin. In pulp repair, it is suggested that the initial mild inflammatory reaction as caused by calcium hydroxide application is a prerequisite for tissue repair, which would not occur if this essential step was omitted²⁴. Therefore, an inflammatory reaction caused by the resin may sometimes fall below the level of mild reaction.

The effects of reasons (3) and (4) enable reliable sealing of the interface to occur. These characteristics of TBB in polymerization will be correlated with favorable dental tissue responses in the filling of MMA-TBB resin in deep unlined cavities in clinical trials^{143,144} and in animal studies, as described above. The comparison of polymerization behavior in a cavity model initiated by conventional benzoyl peroxide (BPO)/amine and TBB initiators is illustrated in Fig. 1. The polymerization of conventional restorative resins starts from the resin side, and the effect of polymerization shrinkage occurs between the dentin and resin. However, according to the report by Imai *et al.*¹⁶², the polymerization of MMA-TBB resin starts from the dentin interface. The dentin interface is thus sealed tightly with the resin, in cooperation with excellent dentin bonding property of the resin through the graft polymer formation mechanism. While the shrinkage of resins creates space in conventional resin, MMA-TBB produces minimum space, which leads to less leakage compared with that of conventional resins. The apical sealing ability of MMA-TBB resin was reported to be significantly better compared with that of gutta-percha/sealer and root canal filling of the resin formed resin tags in dentinal tubules¹⁶³. This excellent seal helps protect against bacterial contamination, the importance of which in pulp-capping was suggested by Cox *et al.*¹²¹

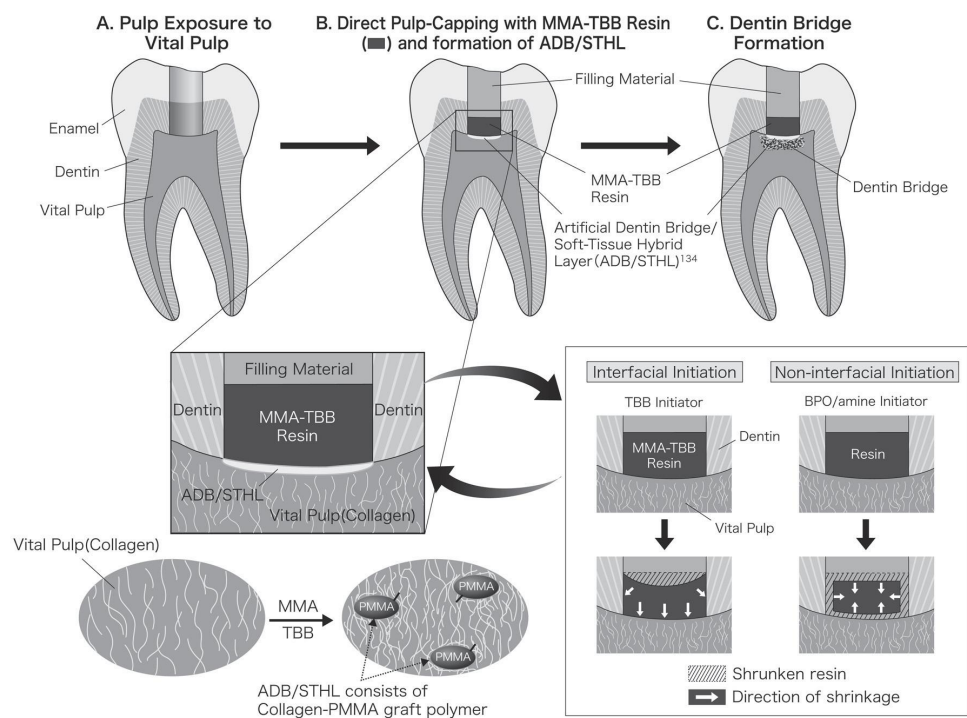


Fig. 1 Formation of Artificial Dentin Bridge (ADB) or Soft-Tissue Hybrid Layer (STHL) in direct pulp-capping with MMA-TBB resin by graft polymerization and interfacial initiation of polymerization mechanism.

A: Exposure of vital pulp. B: Direct pulp-capping with MMA-TBB resin and formation of ADB/STHL, which consists of collagen-PMMA graft polymer. Interfacial initiation of polymerization begins on the dentin side, to which the resin is attracted during polymerization, leading to elimination of gap formation between dentin and the resin. C: Dentin bridge formation, which occurred in half of the experiments^{15,6)}.

Generally pulp capping should be performed in cases with no clinical symptoms and with normal clinical examination. If pulp capping need to be performed in pulp that is suspected of having limited inflammation, such as pulp exposure during or after removing caries, calcium hydroxide/MTA would be the choice instead of MMA-TBB resin. Since MMA-TBB resin does not possess bactericidal action, MMA-TBB resin will be used clinically only for uninfected pulp.

FUTURE CONSIDERATIONS FOR PULP-CAPPING MATERIAL

The success rate of direct pulp-capping is inferior to that of pulpectomy. Current materials lack effectiveness mainly because of leakage at the interface between the dentin and pulp-capping material as well as restorative material. For the development of future pulp-capping materials, the use of ECM may be considered an option, as suggested by Goldberg *et al.*^{24,27)}. The other is the application of high-quality dentin adhesive material with the capability of initiating interfacial polymerization for restorative material over the direct pulp-capping material. The well-sealed restoration will prevent

leakage and enhance long-term success. In this respect, MMA-TBB resin is useful not only for restoration but also for direct pulp-capping in the future. The reason is as follows: While it takes at least one week for calcium hydroxide and MTA to form a natural dentin bridge, one day is enough for MMA-TBB resin to form an “artificial dentin bridge”. Rapid formation of a pulp-protective barrier against bacterial contamination should be desirable to minimize the damage to the pulp. Moreover, the former natural bridges sometimes have tunnel defects, but the latter impermeable “artificial bridge” has no such defect. MTA/MTA related materials and MMA-TBB resin have the advantages of anti-bacterial effects and rapid formation of a pulp-protective barrier, respectively. Therefore, in the future, both materials may be chosen depending on each clinical case for direct pulp-capping. The addition of anti-bacterial substrates to the resin may plus another advantage. It is expected that MMA-TBB resin will be widely accepted worldwide in clinical practice in the future, although only limited numbers of clinicians in Japan are using it. For the resin to be widely used, it must receive official approval as a direct-capping material, because the present commercial resin is approved for restorations, and direct

pulp-capping is not included in the indications for use. Therefore, currently, the resin is used at the dentist's discretion.

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