

## organic compounds

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## Amiloride hydrochloride methanol disolvate

Cuong Quoc Ton<sup>a</sup> and Michael Bolte<sup>b\*</sup>

<sup>a</sup>Institut für Organische Chemie, J. W. Goethe-Universität Frankfurt, Max-von-Laue-Strasse 7, 60438 Frankfurt/Main, Germany, and <sup>b</sup>Institut für Anorganische Chemie, J. W. Goethe-Universität Frankfurt, Max-von-Laue-Strasse 7, 60438 Frankfurt/Main, Germany

Correspondence e-mail: [bolte@chemie.uni-frankfurt.de](mailto:bolte@chemie.uni-frankfurt.de)

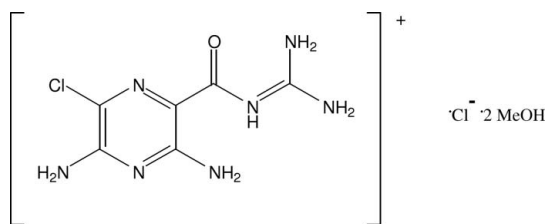
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Key indicators: single-crystal X-ray study;  $T = 173$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.050;  $wR$  factor = 0.080; data-to-parameter ratio = 12.9.

In the crystal of the title compound [systematic name: 2-(3,5-diamino-6-chloropyrazin-2-ylcarbonyl)guanidinium chloride methanol disolvate],  $\text{C}_6\text{H}_9\text{ClN}_7\text{O}^+\cdot\text{Cl}^-\cdot 2\text{CH}_3\text{OH}$ , the components are connected by  $\text{N}-\text{H}\cdots\text{N}$ ,  $\text{N}-\text{H}\cdots\text{Cl}$ ,  $\text{N}-\text{H}\cdots\text{O}$ ,  $\text{O}-\text{H}\cdots\text{Cl}$  and  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds into a three-dimensional network. The dihedral angle between the aromatic ring and the guanidine residue is  $6.0$  ( $2$ )°.

## Related literature

For other salts of amiloride, see: Pretscher *et al.* (2001); Zeslawska *et al.* (2004).



## Experimental

## Crystal data

 $\text{C}_6\text{H}_9\text{ClN}_7\text{O}^+\cdot\text{Cl}^-\cdot 2\text{CH}_3\text{O}$  $M_r = 330.19$ Monoclinic,  $P2_1/n$  $a = 5.9473$  (5) Å $b = 16.7278$  (17) Å $c = 14.7784$  (15) Å $\beta = 90.080$  (8)° $V = 1470.2$  (2) Å<sup>3</sup> $Z = 4$ Mo  $K\alpha$  radiation $\mu = 0.46$  mm<sup>-1</sup> $T = 173$  K $0.30 \times 0.25 \times 0.20$  mm

## Data collection

Stoe IPDS II two-circle diffractometer

Absorption correction: multi-scan (MULABS; Spek, 2009; Blessing, 1995)

 $T_{\min} = 0.874$ ,  $T_{\max} = 0.914$ 

19184 measured reflections

2739 independent reflections

1852 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.071$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.050$  $wR(F^2) = 0.080$  $S = 0.96$ 

2739 reflections

212 parameters

9 restraints

H atoms treated by a mixture of independent and constrained refinement

 $\Delta\rho_{\text{max}} = 0.22$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.25$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N31}-\text{H31A}\cdots\text{N4}^{\text{i}}$	0.88 (1)	2.14 (1)	2.996 (3)	165 (3)
$\text{N31}-\text{H31B}\cdots\text{Cl1}^{\text{ii}}$	0.87 (1)	2.50 (2)	3.281 (3)	150 (3)
$\text{N51}-\text{H51A}\cdots\text{Cl1}^{\text{iii}}$	0.88 (1)	2.54 (1)	3.396 (3)	165 (3)
$\text{N51}-\text{H51B}\cdots\text{O11}$	0.88 (1)	2.11 (2)	2.781 (3)	133 (3)
$\text{N12}-\text{H12}\cdots\text{O2M}^{\text{iv}}$	0.87 (1)	2.14 (2)	2.912 (3)	149 (3)
$\text{N14}-\text{H14A}\cdots\text{Cl1}^{\text{v}}$	0.88 (1)	2.34 (1)	3.188 (3)	162 (3)
$\text{N14}-\text{H14B}\cdots\text{O2M}^{\text{v}}$	0.88 (1)	1.93 (2)	2.783 (3)	162 (3)
$\text{N15}-\text{H15A}\cdots\text{Cl1}^{\text{v}}$	0.88 (1)	2.61 (2)	3.367 (3)	145 (3)
$\text{N15}-\text{H15B}\cdots\text{O11}$	0.88 (1)	2.03 (3)	2.688 (3)	131 (3)
$\text{O1M}-\text{H1M}\cdots\text{Cl1}$	0.84	2.26	3.091 (2)	171
$\text{O2M}-\text{H2M}\cdots\text{O1M}$	0.84	1.91	2.745 (3)	170

Symmetry codes: (i)  $-x + 2, -y, -z + 1$ ; (ii)  $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (iii)  $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$ ; (iv)  $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (v)  $x - 1, y, z$ .

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG2772).

## References

- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.  
 Pretscher, A., Brisander, M., Bauer-Brandl, A. & Hansen, L. K. (2001). *Acta Cryst.* **C57**, 1217–1219.  
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.  
 Stoe & Cie (2001). *X-AREA*. Stoe & Cie, Darmstadt, Germany.  
 Zeslawska, E., Oleksyn, B. & Stadnicka, K. (2004). *Struct. Chem.* **15**, 567–571.

**supplementary materials**

*Acta Cryst.* (2010). E66, o1364 [ doi:10.1107/S1600536810017484 ]

## Amiloride hydrochloride methanol disolvate

C. Q. Ton and M. Bolte

### Experimental

Crystals of the title structure were obtained by recrystallization of amiloride hydrochloride (100 mg) from a methanol (3 g) solution.

### Refinement

H atoms bonded to O and C were geometrically positioned and refined using a riding model with fixed individual displacement parameters [ $U(\text{H}) = 1.5 U_{\text{eq}}(\text{C}, \text{O})$ ] using a riding model with  $\text{C—H} = 0.98 \text{ \AA}$  and  $\text{O—H} = 0.84 \text{ \AA}$ , respectively. The methyl and hydroxyl groups were allowed to rotate but not to tip. H atoms bonded to N were refined with a distance restraint of  $0.88 (1) \text{ \AA}$  and with  $U(\text{H}) = 1.2 U_{\text{eq}}(\text{N})$ .

### Figures

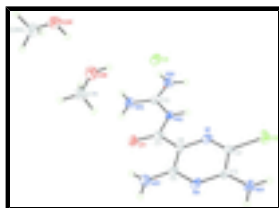


Fig. 1. Perspective view of the asymmetric unit of the title compound. Displacement ellipsoids are at the 50% probability level.

## 2-(3,5-diamino-6-chloropyrazin-2-ylcarbonyl)guanidinium chloride methanol disolvate

### Crystal data

$\text{C}_6\text{H}_9\text{ClN}_7\text{O}^+\cdot\text{Cl}^-\cdot 2\text{CH}_4\text{O}$

$M_r = 330.19$

Monoclinic,  $P2_1/n$

Hall symbol:  $-P 2_1n$

$a = 5.9473 (5) \text{ \AA}$

$b = 16.7278 (17) \text{ \AA}$

$c = 14.7784 (15) \text{ \AA}$

$\beta = 90.080 (8)^\circ$

$V = 1470.2 (2) \text{ \AA}^3$

$Z = 4$

$F(000) = 688$

$D_x = 1.492 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 11166 reflections

$\theta = 3.7\text{--}25.7^\circ$

$\mu = 0.46 \text{ mm}^{-1}$

$T = 173 \text{ K}$

Block, colourless

$0.30 \times 0.25 \times 0.20 \text{ mm}$

### Data collection

Stoe IPDS II two-circle

2739 independent reflections

# supplementary materials

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diffractometer

Radiation source: fine-focus sealed tube

graphite

$\omega$  scans

Absorption correction: multi-scan  
(MULABS; Spek, 2009; Blessing, 1995)

$T_{\min} = 0.874$ ,  $T_{\max} = 0.914$

19184 measured reflections

1852 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.071$

$\theta_{\max} = 25.6^\circ$ ,  $\theta_{\min} = 3.6^\circ$

$h = -7 \rightarrow 7$

$k = -20 \rightarrow 20$

$l = -17 \rightarrow 17$

## Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.050$

$wR(F^2) = 0.080$

$S = 0.96$

2739 reflections

212 parameters

9 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0253P)^2]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.22 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.25 \text{ e } \text{\AA}^{-3}$

## Special details

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.4136 (4)	0.07364 (14)	0.32999 (16)	0.0187 (6)
C2	0.5851 (5)	0.02607 (17)	0.32413 (18)	0.0179 (6)
Cl2	0.62266 (14)	-0.02883 (5)	0.22493 (5)	0.0315 (2)
C3	0.7485 (5)	0.01732 (16)	0.39568 (19)	0.0182 (6)
N31	0.9232 (4)	-0.03176 (16)	0.38692 (17)	0.0241 (6)
H31A	1.006 (5)	-0.0400 (18)	0.4354 (14)	0.029*
H31B	0.939 (5)	-0.0621 (16)	0.3394 (14)	0.029*
N4	0.7256 (4)	0.06106 (13)	0.47133 (16)	0.0179 (5)
C5	0.5473 (5)	0.11111 (16)	0.47856 (19)	0.0173 (6)

N51	0.5303 (5)	0.15414 (15)	0.55477 (18)	0.0256 (6)
H51A	0.624 (4)	0.1497 (18)	0.6007 (15)	0.031*
H51B	0.410 (3)	0.1839 (16)	0.561 (2)	0.031*
C6	0.3874 (5)	0.11725 (17)	0.40695 (19)	0.0168 (6)
C11	0.1930 (5)	0.17089 (18)	0.4093 (2)	0.0202 (7)
O11	0.1433 (4)	0.21444 (12)	0.47391 (14)	0.0249 (5)
N12	0.0642 (4)	0.16832 (14)	0.33015 (15)	0.0173 (5)
H12	0.110 (5)	0.1370 (14)	0.2874 (15)	0.021*
C13	-0.1158 (5)	0.21632 (17)	0.3097 (2)	0.0196 (7)
N14	-0.2054 (4)	0.20867 (16)	0.22820 (18)	0.0250 (6)
H14A	-0.326 (3)	0.2377 (16)	0.218 (2)	0.030*
H14B	-0.134 (5)	0.1739 (15)	0.1935 (18)	0.030*
N15	-0.1985 (4)	0.26602 (16)	0.37008 (18)	0.0250 (6)
H15A	-0.310 (4)	0.2979 (16)	0.355 (2)	0.030*
H15B	-0.141 (5)	0.2664 (19)	0.4246 (11)	0.030*
Cl1	0.36907 (12)	0.32420 (4)	0.23835 (5)	0.02329 (18)
O1M	0.4195 (4)	0.44145 (13)	0.39875 (15)	0.0335 (6)
H1M	0.4001	0.4139	0.3519	0.050*
C1M	0.2507 (6)	0.4216 (2)	0.4641 (2)	0.0355 (8)
H1M1	0.2637	0.3650	0.4802	0.053*
H1M2	0.1015	0.4318	0.4384	0.053*
H1M3	0.2716	0.4545	0.5183	0.053*
O2M	0.4108 (4)	0.59993 (13)	0.35076 (15)	0.0315 (6)
H2M	0.4196	0.5504	0.3592	0.047*
C2MA	0.2302 (6)	0.6318 (2)	0.4037 (2)	0.0370 (9)
H2M1	0.0883	0.6071	0.3848	0.056*
H2M2	0.2214	0.6898	0.3945	0.056*
H2M3	0.2573	0.6205	0.4678	0.056*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0148 (14)	0.0184 (13)	0.0229 (14)	-0.0003 (11)	-0.0025 (10)	0.0018 (11)
C2	0.0171 (15)	0.0209 (15)	0.0157 (14)	-0.0018 (13)	-0.0030 (12)	0.0000 (12)
Cl2	0.0278 (5)	0.0391 (5)	0.0276 (4)	0.0122 (4)	-0.0061 (3)	-0.0114 (4)
C3	0.0142 (15)	0.0149 (15)	0.0254 (16)	-0.0011 (13)	0.0002 (12)	0.0037 (13)
N31	0.0209 (14)	0.0244 (15)	0.0268 (15)	0.0082 (12)	-0.0059 (12)	-0.0050 (12)
N4	0.0133 (13)	0.0146 (12)	0.0258 (14)	0.0018 (10)	-0.0036 (10)	0.0012 (10)
C5	0.0172 (16)	0.0129 (14)	0.0220 (16)	-0.0037 (12)	-0.0017 (12)	0.0012 (12)
N51	0.0250 (16)	0.0269 (15)	0.0250 (15)	0.0067 (12)	-0.0070 (12)	-0.0061 (12)
C6	0.0125 (15)	0.0141 (14)	0.0238 (16)	-0.0012 (12)	-0.0014 (12)	-0.0002 (12)
C11	0.0186 (16)	0.0162 (15)	0.0257 (16)	-0.0046 (13)	-0.0006 (13)	0.0037 (14)
O11	0.0265 (12)	0.0212 (11)	0.0268 (12)	0.0062 (9)	-0.0021 (9)	-0.0042 (9)
N12	0.0174 (13)	0.0166 (13)	0.0180 (13)	0.0043 (11)	-0.0024 (10)	-0.0011 (10)
C13	0.0141 (15)	0.0156 (15)	0.0290 (17)	-0.0004 (12)	0.0001 (13)	0.0027 (13)
N14	0.0197 (15)	0.0272 (15)	0.0280 (16)	0.0083 (11)	-0.0061 (12)	0.0000 (12)
N15	0.0197 (15)	0.0263 (15)	0.0290 (15)	0.0103 (12)	-0.0039 (12)	-0.0018 (12)
Cl1	0.0196 (4)	0.0206 (4)	0.0296 (4)	0.0024 (3)	-0.0022 (3)	0.0006 (3)

## supplementary materials

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O1M	0.0372 (14)	0.0317 (14)	0.0317 (13)	-0.0091 (11)	0.0034 (11)	-0.0026 (10)
C1M	0.040 (2)	0.0347 (19)	0.0318 (19)	-0.0092 (17)	-0.0015 (16)	0.0054 (16)
O2M	0.0344 (14)	0.0271 (13)	0.0328 (13)	-0.0055 (11)	-0.0006 (11)	0.0064 (11)
C2MA	0.039 (2)	0.033 (2)	0.039 (2)	-0.0027 (17)	0.0032 (17)	0.0001 (16)

### *Geometric parameters (Å, °)*

N1—C2	1.297 (4)	N12—H12	0.866 (10)
N1—C6	1.360 (4)	C13—N15	1.316 (4)
C2—C3	1.443 (4)	C13—N14	1.322 (4)
C2—Cl2	1.745 (3)	N14—H14A	0.880 (10)
C3—N31	1.331 (4)	N14—H14B	0.884 (10)
C3—N4	1.343 (4)	N15—H15A	0.880 (10)
N31—H31A	0.880 (10)	N15—H15B	0.876 (10)
N31—H31B	0.872 (10)	O1M—C1M	1.432 (4)
N4—C5	1.355 (4)	O1M—H1M	0.8400
C5—N51	1.341 (4)	C1M—H1M1	0.9800
C5—C6	1.426 (4)	C1M—H1M2	0.9800
N51—H51A	0.879 (10)	C1M—H1M3	0.9800
N51—H51B	0.878 (10)	O2M—C2MA	1.433 (4)
C6—C11	1.464 (4)	O2M—H2M	0.8400
C11—O11	1.238 (3)	C2MA—H2M1	0.9800
C11—N12	1.398 (4)	C2MA—H2M2	0.9800
N12—C13	1.371 (4)	C2MA—H2M3	0.9800
C2—N1—C6	118.4 (2)	C11—N12—H12	117 (2)
N1—C2—C3	122.9 (3)	N15—C13—N14	121.9 (3)
N1—C2—Cl2	118.7 (2)	N15—C13—N12	120.8 (3)
C3—C2—Cl2	118.4 (2)	N14—C13—N12	117.2 (3)
N31—C3—N4	119.8 (3)	C13—N14—H14A	116 (2)
N31—C3—C2	121.1 (3)	C13—N14—H14B	114 (2)
N4—C3—C2	119.1 (3)	H14A—N14—H14B	131 (3)
C3—N31—H31A	117 (2)	C13—N15—H15A	119 (2)
C3—N31—H31B	122 (2)	C13—N15—H15B	119 (2)
H31A—N31—H31B	120 (3)	H15A—N15—H15B	122 (3)
C3—N4—C5	118.8 (2)	C1M—O1M—H1M	109.5
N51—C5—N4	117.2 (3)	O1M—C1M—H1M1	109.5
N51—C5—C6	122.3 (3)	O1M—C1M—H1M2	109.5
N4—C5—C6	120.5 (3)	H1M1—C1M—H1M2	109.5
C5—N51—H51A	124 (2)	O1M—C1M—H1M3	109.5
C5—N51—H51B	117 (2)	H1M1—C1M—H1M3	109.5
H51A—N51—H51B	119 (3)	H1M2—C1M—H1M3	109.5
N1—C6—C5	120.3 (3)	C2MA—O2M—H2M	109.5
N1—C6—C11	116.1 (2)	O2M—C2MA—H2M1	109.5
C5—C6—C11	123.6 (3)	O2M—C2MA—H2M2	109.5
O11—C11—N12	122.2 (3)	H2M1—C2MA—H2M2	109.5
O11—C11—C6	124.6 (3)	O2M—C2MA—H2M3	109.5
N12—C11—C6	113.2 (3)	H2M1—C2MA—H2M3	109.5
C13—N12—C11	126.4 (2)	H2M2—C2MA—H2M3	109.5
C13—N12—H12	116 (2)		

C6—N1—C2—C3	0.1 (4)	N51—C5—C6—N1	178.7 (3)
C6—N1—C2—Cl2	-178.9 (2)	N4—C5—C6—N1	-0.8 (4)
N1—C2—C3—N31	-179.6 (3)	N51—C5—C6—C11	0.6 (4)
Cl2—C2—C3—N31	-0.5 (4)	N4—C5—C6—C11	-178.9 (3)
N1—C2—C3—N4	-1.1 (4)	N1—C6—C11—O11	179.6 (3)
Cl2—C2—C3—N4	178.0 (2)	C5—C6—C11—O11	-2.2 (4)
N31—C3—N4—C5	179.5 (3)	N1—C6—C11—N12	-0.1 (4)
C2—C3—N4—C5	1.1 (4)	C5—C6—C11—N12	178.1 (3)
C3—N4—C5—N51	-179.7 (2)	O11—C11—N12—C13	6.0 (4)
C3—N4—C5—C6	-0.2 (4)	C6—C11—N12—C13	-174.3 (3)
C2—N1—C6—C5	0.8 (4)	C11—N12—C13—N15	-6.9 (4)
C2—N1—C6—C11	179.0 (3)	C11—N12—C13—N14	175.0 (3)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N31—H31A $\cdots$ N4 <sup>i</sup>	0.88 (1)	2.14 (1)	2.996 (3)	165 (3)
N31—H31B $\cdots$ Cl1 <sup>ii</sup>	0.87 (1)	2.50 (2)	3.281 (3)	150 (3)
N51—H51A $\cdots$ Cl1 <sup>iii</sup>	0.88 (1)	2.54 (1)	3.396 (3)	165 (3)
N51—H51B $\cdots$ O11	0.88 (1)	2.11 (2)	2.781 (3)	133 (3)
N12—H12 $\cdots$ O2M <sup>iv</sup>	0.87 (1)	2.14 (2)	2.912 (3)	149 (3)
N14—H14A $\cdots$ Cl1 <sup>v</sup>	0.88 (1)	2.34 (1)	3.188 (3)	162 (3)
N14—H14B $\cdots$ O2M <sup>iv</sup>	0.88 (1)	1.93 (2)	2.783 (3)	162 (3)
N15—H15A $\cdots$ Cl1 <sup>v</sup>	0.88 (1)	2.61 (2)	3.367 (3)	145 (3)
N15—H15B $\cdots$ O11	0.88 (1)	2.03 (3)	2.688 (3)	131 (3)
O1M—H1M $\cdots$ Cl1	0.84	2.26	3.091 (2)	171.
O2M—H2M $\cdots$ O1M	0.84	1.91	2.745 (3)	170.

Symmetry codes: (i)  $-x+2, -y, -z+1$ ; (ii)  $-x+3/2, y-1/2, -z+1/2$ ; (iii)  $x+1/2, -y+1/2, z+1/2$ ; (iv)  $-x+1/2, y-1/2, -z+1/2$ ; (v)  $x-1, y, z$ .

Fig. 1

