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Original research paper

Synthesis and preliminary screening of benzothiazol-2-yl thiadiazole derivatives for anticonvulsant activity

NADEEM SIDDIQUI^{1*} ARPANA RANA¹ SUROOR A. KHAN¹ S. EHTAISHAMUL HAQUE² M. FAIZ ARSHAD¹ SHARIQUE AHMED³ WAQUAR AHSAN¹

¹Department of Pharmaceutical Chemistry, Faculty of Pharmacy Hamdard University New Delhi-110062, India

²Department of Pharmacology, Faculty of Pharmacy, Hamdard University New Delhi-110062, India

³Department of Biochemistry, Faculty of Medicine, 7th October University Misurata, Libya Various *N*-(5-chloro-6-substituted-benzothiazol-2-yl)-*N*'-(substituted phenyl)-[1,3,4]thiadiazole-2,5-diamines (**5a-t**) were designed and synthesized starting from substituted acetophenones. Structures of all the compounds were confirmed on the basis of spectral and elemental analyses. All the newly synthesized compounds were screened for their anticonvulsant activity and were compared with the standard drug phenytoin sodium. Interestingly, all the compounds showed protections against seizures in the range 50–100 % indicative of the promising nature of the compounds against seizure spread. Compounds **5b** and **5c** showed complete protection against MES induced seizures.

Keywords: benzothiazole, thiadiazole, anticonvulsant, neurotoxicity, hepatotoxicity

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The term epilepsy is a collective term that includes disorders of the brain function characterized by the periodic and unpredictable occurrence of seizures. Epilepsies are common and frequently devastating and affect around 1-2 % of the world population (1).

Current drug therapy for epilepsy suffers from a number of disadvantages, including the fact that the convulsions of approximately 25 % of epileptics are inadequately controlled by medication (2). In recent years, the field of antiepileptic drug development has become quite dynamic, affording many promising research opportunities.

In our previous study (3–5), we designed aryl-substituted semicarbazones with benzothiazole moiety, in which we made some modifications in the structure of semicarbazones. The lipophilic aryl ring was replaced with a versatile heterocyclic molecule ben-

^{*} Correspondence; e-mail: nadeems_03@yahoo.co.in, nadeems_03@rediffmail.com

zothiazole, which possesses significant anticonvulsant properties. On the other hand, thiadiazole moiety was also found to possess significant anticonvulsant activity (6, 7). In continuation of our previous research on benzothiazoles, we report here the synthesis and anticonvulsant activity of several *N*-(5-chloro-6-substituted-benzothiazol-2-yl)-*N*'-(substituted phenyl)-[1,3,4]thiadiazole-2,5-diamines. Incorporation of 1,3,4-thiadiazole moiety into the second position of the benzothiazole ring may result in compounds having better anticonvulsant activity. The neurotoxicity as well as liver toxicity of the compounds has also been evaluated.

EXPERIMENTAL

The chemicals and solvents used for the experimental work were commercially procured from E. Merck, CDH, S. D. Fine Chem. and Qualigens, all from India. The silica gel G used for analytical chromatography (TLC) was obtained from E. Merck, India. Melting points were determined in an open glass capillary using a Kjeldahl flask containing paraffin and are uncorrected. The proton magnetic resonance spectra (¹H NMR) were recorded on a Bruker 300 MHz instrument (Bruker, Germany) in DMSO-*d*₆ using tetramethylsilane [(CH₃)₄Si] as internal standard. Chemical shifts (δ) are expressed in ppm. The infrared spectra of compounds (in cm⁻¹) were recorded in KBr on a Bio-Rad FTIR (Browser Morner, USA) spectrometer. Elemental analyses were performed on a 240c analyzer (Perkin Elmer, USA).

Syntheses

5-Chloro-6-substituted-benzothiazol-2-ylamines (1a,b). – A mixture of substituted aniline (0.01 mol) and potassium thiocyanate (0.01 mol) in glacial acetic acid was cooled and stirred. To this solution, bromine (0.01 mol) was added dropwise at such a rate to keep the reaction temperature below 10 °C throughout the addition. Stirring was continued for additional 3 h and the separated hydrochloride salt was filtered, washed with acetic acid and dried. It was dissolved in hot water and neutralized with aqueous ammonia solution (25 %). The precipitate obtained was filtered, washed with water, dried and recrystallized to afford the 6-substituted-1,3-benzothiazol-2-amine (1a,b). Yield: 64 %; m.p. 248 °C; IR (KBr) cm⁻¹: 3239 (NH₂), 3016 (CH-Ar), 1562 (C=N), 804 (C-Cl); ¹H NMR (DMSO-d₆) δ ppm: 7.89–8.16 (m, 2H, ArH), 9.76 (s, 2H, NH₂, D₂O exchangeable).

(5-*Chloro-6-substituted-benzothiazol-2-yl)-ureas* (**2a,b**). – To a solution of sodium cyanate (0.5 g) in minimum quantity of water, glacial acetic acid (5 mL) was added. This solution was heated with the respective 2-amino-6-substituted benzothiazoles (1.7 g, 0.01 mol) in alcohol till the mixture contents became turbid and the volume was half of the original volume. The contents were added to ice cool water. The solid obtained was filtered off and dried. Yield: 73 %; m.p. 262 °C; IR (KBr) cm⁻¹: 3218 (NH), 3165 (NH₂), 2937 (CH-Ar), 1678 (C=O), 1545 (C=N), 816 (C-Cl); ¹H NMR (DMSO-*d*₆) δ ppm: 7.82–8.20 (m, 2H, ArH), 9.14 (s, 2H, NH₂, D₂O exchangeable), 9.86 (s, 1H, NH, D₂O exchangeable).

N-(5-Chloro-6-substituted-benzothiazol-2-yl)hydrazine carboxamides (3a,b). – To a solution of substituted benzothiazole urea (2a,b, 0.01 mol) in ethanol, hydrazine hydrate (0.01

mol) and NaOH (0.04 g) were added. The resulting mixture was refluxed for 16 h. It was then concentrated, cooled and poured over crushed ice to get the precipitate, which was filtered, washed with water and recrystallized from ethanol to get hydrazine carboxamides. Yield: 71 %; m. p. 224 °C; IR (KBr) cm⁻¹: 3265, 3218 (NH), 3112 (NH₂), 3017 (CH-Ar), 1683 (C=O), 1513 (C=N), 845 (C-Cl); ¹H NMR (DMSO-*d*₆) δ ppm: 7.74–8.12 (m, 2H, ArH), 8.98 (s, 2H, NH₂, D₂O exchangeable), 9.41 (s, 1H, NH, D₂O exchangeable), 9.86 (s, 1H, NH, D₂O exchangeable).

N-(5-*Chloro-6-substituted-benzothiazol-2-yl)-2-[(4-substituted phenyl)carbamothioyl] hydrazinecarboxamides* (4a-t). – A solution of carboxamides in glacial acetic acid (5 mL) and ethanol (10 mL) was heated to boiling and refluxed with aromatic ketones (1 g, 0.122 mol) for 5 h. Refluxed solution was cooled to room temperature and kept overnight. The solid was collected out, washed with methanol, dried and recrystallized from ethanol to get the target compound. Yield: 58 %; m.p. 269 °C; IR (KBr) cm⁻¹: 3285, 3245, 3212, 3176 (NH), 3006 (CH-Ar), 1673 (C=O), 1534 (C=N), 1055 (C=S), 832 (C-Cl); ¹H NMR (DMSO-*-d*₆) δ ppm: 6.76–8.32 (m, 7H, ArH), 8.52 (s, 1H, NH, D₂O exchangeable), 8.75 (s, 1H, NH, D₂O exchangeable), 9.18 (s, 1H, NH, D₂O exchangeable), 9.49 (s, 1H, NH, D₂O exchangeable).

N-(5-Chloro-6-substituted-benzothiazol-2-yl)-N'-(substituted phenyl)-[1,3,4]thiadiazole-2,5-diamines (5a-t). – To cold concentrated sulphuric acid (25 mL), compounds (4a-t, 0.01 mol) were added gradually under constant stirring. After the addition was completed, the solution was kept at room temperature for 2 h and poured onto crushed ice. The solid thus separated was washed with water and treated with a dilute solution of so-dium bicarbonate. The product obtained was washed with water, dried and recrystallized from ethanol to afford the titled compounds (5a-t).

Pharmacology

Anticonvulsant activity. – Anticonvulsant evaluation was undertaken using a reported procedure (8, 9). Albino mice (Swiss, 25–30 g) were used in groups of six each as experimental animals. The test compounds and standard drug were suspended in Tween 80 (1%) or in a 0.5% methyl cellulose-water mixture and administered intraperitoneally. The animals were maintained on an adequate diet and allowed free access to food and water except during the short time they were removed from cages for testing. The animals were maintained at room temperature (25 ± 2 °C). All the experimental protocols were carried out with the permission of the Institutional Animal Ethics Committee (IAEC). Animals were obtained from the Central Animal House Facility, Jamia Hamdard University, New Delhi, India.

Maximal electroshock seizure test (MES). – Maximal electroshock seizure was elicited with a 60 cycle alternating current of 50 mA intensity delivered for 0.25 s *via* ear clip electrodes. Maximal seizure typically consists of a short period of tonic extension of the hind limbs and a final clonic episode. Abolition of the hind limb tonic extensor component of the seizure is defined as protection and the results are expressed as % protection.

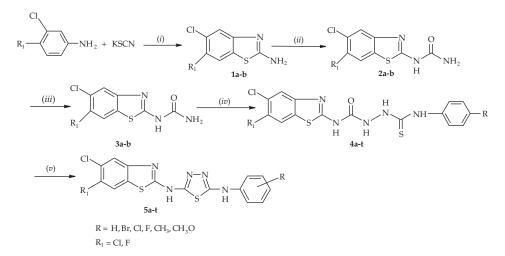
Neurotoxicity (NT). – The rotarod test was used to evaluate neurotoxicity (10). The animal was placed on a 3.2-cm diameter knurled rod rotating at 6 rpm. Normal mice can remain indefinitely on a rod rotating at this speed. Neurological toxicity is defined as the failure of the animal to remain on the rod for 1 min. Results are expressed as the number of animals exhibiting toxicity/number of animals tested.

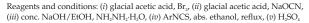
Assessment of liver function. – Biochemical parameters such as serum glutamate oxaloacetate transaminase (SGOT) and serum glutamate pyruvate transaminase (SGPT) were estimated by the Reitman and Frankel method (11) and alkaline phosphatase was measured using the King and Armstrong method (12). Total protein and total albumin were also measured according to the biuret method (13).

Statistical analyses. – All the statistical analyses were performed using the SigmaStat version 4.0 software and ANOVA followed by Dunnett's multiple comparison test.

RESULTS AND DISCUSSION

N-(5-chloro-6-substituted-benzothiazol-2-yl)-*N*'-(substituted phenyl)-[1,3,4]thiadiazole-2,5-diamines (**5a-t**) were synthesized by the sequence described in Scheme 1. Initial compounds, 5-chloro-6-substituted-1,3-benzothiazol-2-amines (**1a,b**), were synthesized by treating arylamines with potassium thiocyanate. The 5-chloro-6-substituted-benzothiazol-2-yl-ureas (**2a,b**) were prepared by treating appropriate benzothiazole with sodium cyanate in the presence of glacial acetic acid. Further *N*-(5-chloro-6-substituted-1,3benzothiazol-2-yl) hydrazine carboxamides (**3a, b**) were obtained by refluxing them with





Scheme 1

hydrazine hydrate to yield carboxamides. The final compounds (**5a-t**) were obtained by refluxing carboxamides with different substituted phenyl isothiocyanates in ethanol and then cyclizing them in the presence of cold concentrated sulphuric acid. Physicochemical parameters of the synthesized compounds are presented in Table I.

The structures and purity of the final compounds were confirmed on the basis of spectral and elemental analyses and the data were within \pm 0.4 % of theoretical values

Compd.			Mol. formula	Found/calcd. (%)		Yield	M.p.	
No.	R	R ₁	$M_{ m r}{}^{ m a}$	С	Н	Ν	(%)	(°C) ^b
5a	4-H	Cl	C ₁₅ H ₉ Cl ₂ N ₅ S ₂ (394.29)	45.65 45.69	2.32 2.30	17.77 17.76	55	259–261
5b	4-Br	Cl	C ₁₅ H ₈ BrCl ₂ N ₅ S ₂ (473.19)	38.12 38.07	1.67 1.70	14.78 14.80	45	279–281
5c	3-Cl	Cl	$C_{15}H_8Cl_3N_5S_2$ (428.74)	42.03 42.02	1.82 1.88	16.37 16.33	50	257–159
5d	4-Cl	Cl	$C_{15}H_8Cl_3N_5S_2$ (428.74)	42.00 42.02	1.84 1.88	16.37 16.33	54	261–263
5e	4-F	Cl	$C_{15}H_8Cl_2FN_5S_2$ (412.28)	43.72 43.70	1.97 1.96	16.96 16.99	60	265–267
5f	3-CH ₃	Cl	$\begin{array}{c} C_{16}H_{11}Cl_2N_5S_2\\ (408.\ 32) \end{array}$	47.04 47.06	2.74 2.72	17.12 17.15	50	267–269
5g	4-CH ₃	Cl	$\begin{array}{c} C_{16}H_{11}Cl_2N_5S_2\\ (408.\ 32) \end{array}$	46.03 46.06	2.75 2.72	17.19 17.15	54	249–251
5h	2-OCH ₃	Cl	C ₁₆ H ₁₁ Cl ₂ N ₅ OS ₂ (424.32)	45.26 45.29	2.59 2.61	15.53 16.50	48	239–241
5i	3-OCH ₃	Cl	C ₁₆ H ₁₁ Cl ₂ N ₅ OS ₂ (424.32)	47.26 47.29	2.64 2.61	16.51 16.50	45	243–245
5j	4-OCH ₃	Cl	C ₁₆ H ₁₁ Cl ₂ N ₅ OS ₂ (424.32)	45.31 45.29	2.64 2.61	16.54 16.50	45	269–271
5k	4-H	F	C ₁₅ H ₉ ClFN ₅ S ₂ (377.84)	47.71 47.68	2.42 2.40	18.58 18.54	65	255–257
51	4-Br	F	C ₁₅ H ₈ BrClFN ₅ S ₂ (456.73)	39.41 39.45	1.81 1.77	15.30 15.33	50	253–255
5m	3-Cl	F	$C_{15}H_8Cl_2FN_5S_2$ (412.28)	43.73 43.70	1.92 1.96	16.96 16.99	64	247–249
5n	4-Cl	F	C ₁₅ H ₈ Cl ₂ FN ₅ S ₂ (412.28)	43.67 43.70	1.99 1.96	16.96 16.99	64	229–231

Table I. Physicochemical data of synthesized compounds 5a-t

50	4-F	F	C ₁₅ H ₈ ClF ₂ N ₅ S ₂ (395.83)	45.50 45.52	1.99 2.04	17.67 17.69	60	227–229
5p	3-CH ₃	F	C ₁₆ H ₁₁ ClFN ₅ S ₂ (391.86)	49.00 49.04	2.86 2.83	17.90 17.87	66	225–227
5q	4-CH ₃	F	C ₁₆ H ₁₁ ClFN ₅ S ₂ (391.86)	49.01 49.04	2.86 2.83	17.91 17.87	62	219–221
5r	2-0CH ₃	F	C ₁₆ H ₁₁ ClFN ₅ OS ₂ (407.86)	47.14 47.12	2.69 2.72	17.21 17.17	52	221–223
5s	3-OCH ₃	F	C ₁₆ H ₁₁ ClFN ₅ OS ₂ (407.86)	47.17 47.12	2.70 2.72	17.13 17.17	50	235–237
5t	4-OCH ₃	F	C ₁₆ H ₁₁ ClFN ₅ OS ₂ (407.86)	46.15 46.12	2.70 2.72	17.21 17.17	62	245–247

^a Solvent of crystallization: methanol.

^b Melting point of the compounds at decomposition.

(Table II). The FTIR spectra showed bands at 3370–3153 cm⁻¹ for the NH and 1674–1633 and 720–689 cm⁻¹ for C=O and C-S-C, respectively. The ¹H NMR spectra showed two singlets at δ 8.85–9.30 and 11.14–11.78 ppm for the two NH protons.

Pharmacology

1-(5,6-Disubstituted-1,3-benzothiazol-2-yl)-3-[5-(4-substituted phenyl)-1,3,4-thiadiazol-2-yl urea derivatives (5a-t) obtained from the reactional sequence were injected intraperitoneally into mice and evaluated by the maximal electroshock (MES), neurotoxicity screen using rotarod at the dose of 30 mg kg^{-1} body mass and observations were carried out at two different time intervals of 0.5 and 4 h. The data are presented in Table III. Phenytoin was used as the standard for the comparison at the dose level of 30 mg kg^{-1} . All the compounds showed anti-MES activity indicative of their ability to prevent seizure spread. The compounds that showed 100 % protection against the MES model at 30 mg kg⁻¹ body mass were **5b**, **5c** both after 0.5 and 4 h. Compounds **5k**, **5o**, **5e** and **5f** showed 83 % protection at both time intervals except for compound 50, which showed 66 % protection after 4 h, indicating a rapid onset and shorter duration of action. Compounds 5d, 5j, 5l, 5r and 5t had 66 % protection, whereas compounds 5a, 5g, 5h, 5i, 5m, 5n, 5p, 5q and 5s were 50 % protective in the anti-MES screen. In the neurotoxicity screen, compounds with 83-100 % protection were selected and checked for neurotoxicity at the dose of 30 mg kg⁻¹. None of the compounds displayed neurotoxicity, since they successfully passed the rotarod test without any sign of motor impairment (Table III). Bioevaluation led to correlation of the anticonvulsant screening with the basic structure of the compounds. In general, the disubstituted benzothiazole ring with 5,6-Cl substituents had higher potency than the 5-Cl and 6-F substitution. Disubstitution with Cl at the 5,6 position of the benzothiazole ring and 4-Br, 3-Cl substituents at the distant phenyl ring resulted in highly potent compounds, whereas replacement with 5-Cl, 6-F at the benzothiazole ring and H, 3-Cl, 3-CH₃, 2-OCH₃, 3-OCH₃, 4-CH₃, 4-Cl, 4-F substituents at the

Compd. No.	FT-IR (KBr, cm ⁻¹)	¹ H NMR (DMSO-d ₆ , δ ppm)	¹³ C NMR (CDCl ₃ , TMS, δ ppm)
5a	3304 (NH str.), 3066 (Ar-CH), 1577 (C=N), 809 (C-Cl), 657 (C-S-C)	6.96–7.59 (m, 7H, Ar-H), 9.18 (s, 1H, NH, D ₂ O exchangeable), 9.32 (bs, 1H, NH, D ₂ O exchangeable)	185.63, 164.70, 156.68, 150.43, 136.23, 133.54, 130.75, 125.78, 123.89, 121.56, 118.34
5b	3415 (NH str.), 3003 (Ar-CH), 1545 (C=N), 823 (C-Cl), 617 (C-S-C)	6.87–7.64 (m, 6H, Ar-H), 9.23 (s, 1H, NH, D ₂ O exchangeable), 9.51 (bs, 1H, NH, D ₂ O exchangeable)	184.78, 164.54, 154.65, 151.85, 137.52, 133.08, 130.31, 128.45, 124.65, 121.21, 118.54, 112.85
5c	3334 (NH str.), 3026 (Ar-CH), 1551 (C=N), 904 (C-Cl), 643 (C-S-C)	6.91–7.79 (m, 6H, Ar-H), 9.19 (s, 1H, NH, D ₂ O exchangeable), 9.39 (bs, 1H, NH, D ₂ O exchangeable)	182.24, 166.47, 154.79, 151.65, 138.47, 132.48, 130.86, 126.55, 123.23, 119.32
5d	3357 (NH str.), 3015 (Ar-CH), 1564 (C=N), 879 (C-Cl), 634 (C-S-C)	6.92–7.71 (m, 6H, Ar-H), 9.16 (s, 1H, NH, D ₂ O exchangeable), 9.47 (bs, 1H, NH, D ₂ O exchangeable)	186.65, 164.88, 154.84, 152.67, 134.52, 129.29, 126.24, 123.70, 117.85
5e	3319 (NH str.), 3034 (Ar-CH), 1554 (C=N), 1361 (C-F), 865 (C-Cl), 621 (C-S-C)	6.81–7.59 (m, 6H, Ar-H), 9.16 (s, 1H, NH, D ₂ O exchangeable), 9.45 (bs, 1H, NH, D ₂ O exchangeable)	185.55, 164.22, 154.75, 151.63, 140.66, 136.54, 132.32, 131.65, 126.88, 123.74, 121.55, 116.84, 111.91
5f	3339 (NH str.), 3115 (Ar-CH), 1561 (C=N), 837 (C-Cl), 637 (C-S-C)	3.11 (s, 3H, CH ₃), 6.96–7.79 (m, 6H, Ar-H), 9.09 (s, 1H, NH, D ₂ O exchangeable), 9.37 (bs, 1H, NH, D ₂ O exchangeable)	184.54, 164.74, 154.70, 151.56, 129.54, 126.54, 123.65, 121.54, 118.21, 113.84, 20.54
5g	3416 (NH str.), 3057 (Ar-CH), 1504 (C=N), 869 (C-Cl), 659 (C-S-C)	3.19 (s, 3H, CH ₃), 6.87–7.71 (m, 6H, Ar-H), 9.12 (s, 1H, NH, D ₂ O exchangeable), 9.41 (bs, 1H, NH, D ₂ O exchangeable)	183.52, 164.85, 154.21, 151.65, 130.54, 128.33, 124.44, 117.36, 20.65
5h	3345 (NH str.), 3016 (Ar-CH), 2933 (CH-aliph.), 1528 (C=N), 892 (C-Cl), 687 (C-S-C)	4.25 (s, 3H, OCH ₃), 6.90–7.69 (m, 6H, Ar-H), 9.32 (s, 1H, NH, D_2O exchangeable), 9.56 (bs, 1H, NH, D_2O exchangeable)	182.22, 166.32, 155.74, 151.69, 130.54, 129.54, 126.21, 124.99, 122.55, 120.68, 116.44, 114.54, 57.41
5i	3315 (NH str.), 3009 (Ar-CH), 2945 (CH-aliph.), 1556 (C=N), 837 (C-Cl), 669 (C-S-C)	4.16 (s, 3H, OCH ₃), 6.78–7.54 (m, 6H, Ar-H), 9.16 (s, 1H, NH, D ₂ O exchangeable), 9.43 (bs, 1H, NH, D ₂ O exchangeable)	189.65, 166.52, 159.55, 154.79, 140.22, 133.85, 127.12, 124.11, 109.85, 104.32, 54.23
5j	3308 (NH str.), 3054 (Ar-CH), 3004 (CH-aliph.), 1579 (C=N), 845 (C-Cl), 698 (C-S-C)	4.34 (s, 3H, OCH ₃), 6.84–7.79 (m, 6H, Ar-H), 9.10 (s, 1H, NH, D ₂ O exchangeable), 9.41 (bs, 1H, NH, D ₂ O exchangeable)	181.22, 164.54, 156.12, 154.85, 152.65, 131.55, 130.39, 124.54, 121.69, 118.45, 113.54, 110.69, 56.21

Table II. Spectral characterization of synthesized compounds

5k	3356 (NH str.), 3116 (Ar-CH), 2958 (CH-aliph.), 1456 (C=N), 1357 (C-F), 684 (C-S-C)	6.67–7.74 (m, 7H, Ar-H), 9.24 (s, 1H, NH, D ₂ O exchangeable), 9.58 (bs, 1H, NH, D ₂ O exchangeable)	184.65, 164.32, 154.69, 151.01, 149.64, 137.25, 129.81, 124.24, 122.64, 116.69, 112.14, 110.54
51	3343 (NH str.), 3048 (Ar-CH), 1474 (C=N), 1369 (C-F), 656 (C-S-C), 563 (C-Br)	6.54–7.69 (m, 6H, Ar-H), 9.16 (s, 1H, NH, D ₂ O exchangeable), 9.49 (bs, 1H, NH, D ₂ O exchangeable)	183.24, 165.35, 154.87, 151.54, 147.56, 136.56, 133.58, 130.56, 128.21, 124.54, 121.45, 116.24, 112.54, 110.94, 108.37
5m	3323 (NH str.), 3007 (Ar-CH), 1465 (C=N), 1342 (C-F), 852 (C-Cl), 647 (C-S-C)	6.48–7.63 (m, 6H, Ar-H), 9.23 (s, 1H, NH, D ₂ O exchangeable), 9.58 (bs, 1H, NH, D ₂ O exchangeable)	184.62, 166.71, 154.75, 152.62, 149.59, 138.75, 133.94, 131.27, 126.35, 122.45, 120.57, 118.41, 116.28, 112.70
5n	3416 (NH str.), 3084 (Ar-CH), 1478 (C=N), 1356 (C-F), 835 (C-Cl), 667 (C-S-C)	6.51–7.79 (m, 6H, Ar-H), 9.23 (s, 1H, NH, D ₂ O exchangeable), 9.45 (bs, 1H, NH, D ₂ O exchangeable)	186.36, 166.25, 154.74, 153.67, 149.45, 136.12, 130.56, 126.56, 124.69, 119.43, 112.45, 111.63
50	3423 (NH str.), 3091 (Ar-CH), 1504 (C=N), 1345 (C-F), 678 (C-S-C)	6.79–7.85 (m, 6H, Ar-H), 9.18 (s, 1H, NH, D ₂ O exchangeable), 9.59 (bs, 1H, NH, D ₂ O exchangeable)	186.24, 164.22, 153.63, 152.28, 150.27, 147.27, 142.69, 139.57, 135.16, 126.77, 124.20, 122.65, 120.54, 116.27, 111.49
5p	3478 (NH str.), 3167 (Ar-CH), 2967 (CH-aliph.), 1545 (C=N), 1378 (C-F), 667 (C-S-C)	3.21 (s, 3H, CH ₃), 6.81–7.95 (m, 6H, Ar-H), 9.23 (s, 1H, NH, D ₂ O exchangeable), 9.46 (bs, 1H, NH, D ₂ O exchangeable)	183.54, 166.56, 154.21, 141.27, 129.57, 124.74, 120.74, 114.75, 112.74, 20.96
5q	3445 (NH str.), 3145 (Ar-CH), 2956 (CH-aliph.), 1523 (C=N), 1402 (C-F), 679 (C-S-C)	3.24 (s, 3H, CH ₃), 6.87–7.94 (m, 6H, Ar-H), 9.16 (s, 1H, NH, D ₂ O exchangeable), 9.52 (bs, 1H, NH, D ₂ O exchangeable)	185.96, 166.34, 154.27, 150.65, 138.96, 130.24, 125.29, 117.41, 112.63, 21.64
5r	3312 (NH str.), 3006 (Ar-CH), 2919 (CH-aliph.), 1535 (C=N), 1392 (C-F), 697 (C-S-C)	4.21 (s, 3H, OCH ₃), 6.78–7.45 (m, 6H, Ar-H), 9.17 (s, 1H, NH, D ₂ O exchangeable), 9.38 (bs, 1H, NH, D ₂ O exchangeable)	188.24, 166.34, 155.67, 150.91, 147.45, 132.79, 127.94, 125.17, 124.91, 122.92, 120.35, 118.54, 114.14, 111.79, 56.34
5s	3359 (NH str.), 3019 (Ar-CH), 2958 (CH-aliph.), 1504 (C=N), 1378 (C-F), 684 (C-S-C)	4.24 (s, 3H, OCH ₃), 6.72–7.49 (m, 6H, Ar-H), 9.25 (s, 1H, NH, D ₂ O exchangeable), 9.46 (bs, 1H, NH, D ₂ O exchangeable)	185.25, 166.34, 160.97, 155.39, 154.31, 150.63, 146.11, 139.33, 134.96, 125.42, 117.52, 109.64, 105.34, 55.74
5t	3308 (NH str.), 3008 (Ar-CH), 2916 (CH-aliph.), 1515 (C=N), 1354 (C-F), 673 (C-S-C)	4.25 (s, 3H, OCH ₃), 6.69–7.51 (m, 6H, Ar-H), 9.19 (s, 1H, NH, D_2O exchangeable), 9.51 (bs, 1H, NH, D_2O exchangeable)	189.24, 166.27, 157.94, 154.87, 150.61, 134.61, 132.94, 124.67, 121.49, 118.37, 115.41, 111.27, 55.67

distant phenyl ring led to a 50 % decrease in potency. Disubstitution with 5,6-Cl and 5-Cl, 6-F at the benzothiazole ring and 4-F, 3-CH₃ at the distant phenyl ring resulted in compounds with highly significant activity. Compounds with disubstitution such as 5,6-Cl and 5-F, 6-Cl and 4-Br, 4-Cl or 4-OCH₃, 2-OCH₃ at the distant phenyl ring had significant activity.

Enzyme estimation and histopathological studies of the selected compounds **5b** and **5c** showing 100 % protection were also performed to check the magnitude of liver toxicity and the data are presented in Table IV. The concentrations of alkaline phosphatase,

	Intraperitoneal injection in mice ^a					
- Compd.	MES screen	protection (%)		city screen		
No.	Time (h)					
-	0.5	4	0.5	4		
Control	-	_	_	_		
5a	50	50	NT	NT		
5b	100	100	_	_		
5c	100	100	-	-		
5d	66	66	NT	NT		
5e	83	83	-	-		
5f	83	83	-	-		
5g	50	50	NT	NT		
5h	50	50	NT	NT		
5i	50	50	NT	NT		
5j	66	66	NT	NT		
5k	83	83	_	_		
51	66	66	NT	NT		
5m	50	50	NT	NT		
5n	50	50	NT	NT		
50	83	83	_	_		
5p	50	50	NT	NT		
5q	50	50	NT	NT		
5r	66	66	NT	NT		
5s	50	50	NT	NT		
5t	66	66	NT	NT		
Phenytoin ^a	100	100	-	_		

Table III. Anticonvulsant and neurotoxicity screening of compounds 5a-t

Number of animals tested (n = 6).

^a Dose of 30 mg kg⁻¹ was administered *i.p.* The test compounds and standard drug were suspended in 1 % Tween 80 or in 0.5 % methylcellulose-water mixture. The animals were examined 0.5 and 4 h after administration. The (–) indicates the absence of activity and NT denotes not tested.

Treatment	Alkaline phosphatase (units L ⁻¹)	SGOT (units L ⁻¹)	SGPT (units L ⁻¹)	Total protein (g dL ⁻¹)
Control	13.06 ± 0.25	148.67 ± 1.50	27.67 ± 0.840	1.80 ± 0.01
5b	12.82 ± 0.16	124.33 ± 1.94^{b}	25.33 ± 0.49	2.85 ± 0.03^{b}
5c	$15.31 \pm 0.23^{\circ}$	$108.50 \pm 0.72^{\circ}$	$33.17 \pm 0.870^{\circ}$	$2.36 \pm 0.04^{\circ}$

Table IV. Liver toxicity studies of selected compounds^a

^a Mean \pm SEM values (n = 6).

Significantly different from control: $^{b} p < 0.05$, $^{c} p < 0.01$.

SGOT, SGPT and total protein were determined and the values are represented as mean \pm SEM. Compound **5b** showed a significant change in the SGOT and total protein level compared to the control (p < 0.05), whereas compound **5c** was found to increase significantly the concentration of all the liver enzymes estimated as well as the total protein level compared to the control (p < 0.01).

CONCLUSIONS

Thiadiazole incorporated benzothiazole derivatives can be regarded as a new structural class of anticonvulsant agents. Some of the compounds displayed encouraging activities in the MES test with less neurotoxicity and minimal effect on the liver. These compounds have promising prospects of future applications.

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SAŽETAK

Sinteza i preliminarna ispitivanja antikonvulzivnog djelovanja derivata benzotiazol-2-il tiadiazola

NADEEM SIDDIQUI, ARPANA RANA, SUROOR A. KHAN, S. EHTAISHAMUL HAQUE, M. FAIZ ARSHAD, SHARIQUE AHMED I WAQUAR AHSAN

U radu je opisano dizajniranje i sinteza različitih *N*-(5-klor-6-supstituiranih benzotiazol-2-il)-*N*'-(supstituiranih fenil)-[1,3,4]tiadiazol-2,5-diamina (**5a-t**) polazeći od odgovarajućih acetofenona. Strukture spojeva određene su na temelju spektroskopskih podataka i elementarne analize. Ispitano je antikonvulzivno djelovanje svih novosintetiziranih spojeva i uspoređeno s djelovanjem natrijeve soli fenitoina. Spojevi **5b** i **5c** pružaju potpunu zaštitu od konvulzija uzrokovanih MES-om, a svi spojevi štite od konvulzija u rasponu od 50 do 100 %.

Ključne riječi: benzotiazol, tiadiazol, antikonvulziv, neurotoksičnost, hepatotoksičnost

Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Hamdard University, New Delhi-110062, India

Department of Pharmacology, Faculty of Pharmacy, Hamdard University, New Delhi-110062, India

Department of Biochemistry, Faculty of Medicine, 7th October University, Misurata, Libya