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<https://doi.org/10.18297/etd/2781>

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MECHANISM INVESTIGATION OF PSEUDOURIDINE SYNTHASES TruB AND  
RluA WITH RNA CONTAINING 5-FLUOROURIDINE AND 4-THIOURIDINE

By

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M.S., Chosun University, 2010  
B.S., Vietnam National University, 2008

A Thesis  
Submitted to the Faculty of the  
College of Arts and Sciences of the University of Louisville  
in Partial Fulfillment of the Requirements  
for the Degree of

Master of Science in Chemistry

Department of Chemistry  
University of Louisville  
Louisville, Kentucky

August, 2017



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A Thesis Approved on

July 3, 2017

By the Following Thesis Committee:

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Thesis Director  
Professor Eugene G. Mueller

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Professor Ying Li

---

Professor Muriel C. Maurer

---

Professor David J. Schultz

## DEDICATION

This thesis is dedicated to

My parents Duong Hong Hai and Nghiem Thi Thai,  
Brother Duong Hong Quan, sister Duong Thi Kieu Oanh,  
beloved husband Tu Quang Nguyen, and my lovely  
daughters Sophia Nguyen and Ruby Nguyen

## ACKNOWLEDGEMENTS

Foremost, I would like to thank my advisor Professor Eugene Mueller for giving me an opportunity to be a part of his research group. His insightful guidance allowed me to think critically and motivated me to develop throughout 2.5 years at University of Louisville. I extremely appreciate his strong support and endless encouragement. I also had the great opportunity of taking his course in Biochemistry during my first year at the graduate school. His class provided me valuable knowledge in the field of enzymology that I hope to apply through my career. He is the best mentor and teacher I have ever known. Without his helpful guidance, I would never have completed this thesis.

I would like to acknowledge Professor Ying Li, Professor Muriel C. Maurer, Professor David J. Schultz for serving as my committee members. I appreciate their feedback and suggestions that have greatly improved this thesis.

I thank my previous lab mates, Dr. Govardhan reddy Veerareddygari, who was always willing to teach me everything when I first joined the lab, Dr. Kelly Sullivan for her help with troubleshooting and discussions. I would like to thank my current lab Austin Gibbs for sharing his knowledge in Biochemistry and for his friendly behavior.

I would like to thank my friends who was willing to talked and discussed every time I needed. I was not alone since I was shared by Paige Monsen, Faye Carvajal, Ram Hona, Abdullah Mamun, Nour Jamhawi, and Biyun Shi.

Last but not least, I am grateful to my parents, my brother, my sister and my husband for always being with me and loving me. I also owe great deal to my kids Sophia and Ruby for being with me, my life cannot be more beautiful than seeing you every day.

## ABSTRACT

### MECHANISM INVESTIGATION OF PSEUDOURIDINE SYNTHASES TruB AND RluA WITH RNA CONTAINING 5-FLUOROURIDINE AND 4-THIOURIDINE

Uyen T Duong

July 3, 2017

Pseudouridine synthases ( $\Psi$  synthases) are the enzymes that catalyze the isomerization of uridine (U) to pseudouridine ( $\Psi$ ), which is the most prevalent post-transcriptional modification of RNA. The  $\Psi$  synthases fall into six different families that share no significant global sequence similarity; however, they all involve a conserved aspartic acid residue which is absolutely essential for activity.

Tyrosine is a conserved residue in the active site in five of the six families of  $\Psi$  synthases (phenylalanine in the TruD family) and was hypothesized as the general base for the isomerization reaction. To confirm the function of Tyr-96, Y96F RluA was assayed with both ASL and [F<sup>5</sup>U]ASL. U is converted to  $\Psi$  and F<sup>5</sup>U to F<sup>5</sup>U products. These results argue against the role of Tyr serving as general base. However, the slow rates of reactions and higher concentration of Y96F RluA needed for any reaction indicates that Tyr-96 does facilitate at least one step of the reaction.

The major product of F<sup>5</sup>U from the action of  $\Psi$  synthases is a *ribo* isomer whereas the minor product is *arabino*, and its generation requires epimerization at C2'. The



deprotonation at C2' can be achieved by the conserved Asp or O<sup>2</sup>. To test if O<sup>2</sup> is the general base, the isomerized U was replaced by 4-thiouridine (s<sup>4</sup>U). As an essential first step, RNA containing s<sup>4</sup>U needs to be verified as a good substrate for Ψ synthases, so RluA and TruB were incubated with [s<sup>4</sup>U]RNA. Intact [s<sup>4</sup>U]RNA shifted to later and shorter retention times after incubation with RluA and TruB, respectively. Traces of the digestion products of [s<sup>4</sup>U]RNA after incubation with the two enzymes also showed the new peaks that absorbed more strongly at 330 nm than 260 nm. These results indicate that [s<sup>4</sup>U]RNA can be handled as a substrate.

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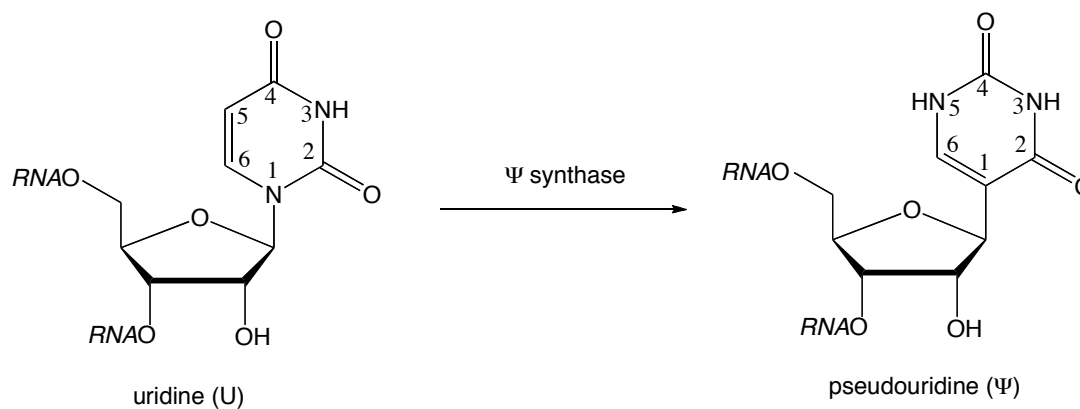
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## CHAPTER I

### PSEUDOURIDINE AND THE PSEUDOURIDINE SYNTHASES

Pseudouridine ( $\Psi$ ), the C-glycoside isomer of uridine (U) is the most common post-transcriptional modification of RNA [1, 2].  $\Psi$  was discovered in 1951 and is also known as the ‘fifth nucleotide’ in RNA [1].  $\Psi$  has been found in every species examined and in all classes of RNA [3, 4]. Pseudouridine synthases ( $\Psi$  synthases) are the enzymes responsible for this site-specific isomerization of U to  $\Psi$ .



**Figure 1.1:** The isomerization of U to  $\Psi$

## 1.1 Physiological significance of pseudouridine

The conversion of U to  $\Psi$  occurs in all three domains of life and is the most prevalent post-transcriptional modification [1]; however, the reason for this isomerization and the importance of  $\Psi$  at many positions in diverse RNAs have remained poorly understood. A speculative role for  $\Psi$  was based on its distinctive physical and chemical properties in comparison with U. The C glycosidic bond in  $\Psi$  enhances rotational freedom as compared to the N glycosidic bond in U; therefore,  $\Psi$  was anticipated to exhibit greater conformational flexibility than U [5]. Also, the N1-H in  $\Psi$  can act as an additional hydrogen bond donor, suggesting that  $\Psi$  might impart structural rigidity in RNA. This conclusion has been confirmed by nuclear magnetic resonance [6], CD spectroscopy [7], and molecular dynamics simulations [8]. Furthermore,  $\Psi$  enhances local RNA stacking in both single and duplex-stranded, which has been proposed to be the most important contribution of  $\Psi$  to the stabilization of RNA structure [3, 6, 7, 9].

The ubiquity of  $\Psi$  in all forms of RNA emphasizes its physiological importance. Lack of  $\Psi$  at position 1911, 1915, and 1917 of 23S rRNA in *Escherichia coli* (*E. coli*) causes severe growth retardation [10-12]. In eukaryotes,  $\Psi$  is localized in the region of interaction between U2 snRNA and pre-mRNA. Absence of  $\Psi$  in this position affects the production of mature transcripts by impairing the splicing activity of pre-mRNA [13].

The X-linked form of the human disease dyskeratosis congenita results from a mutation or deletion in the gene *dyskerin*, an ortholog of the  $\Psi$  synthase-encoding Cbf5 gene. Patients with this disease have defects in highly regenerative tissues, such as skin and bone marrow, and chromosomal instability [14-16]. Cells of individuals with this disorder show reduced amounts of telomerase RNA and thus decreased telomerase

activity with resulting difficulty in maintaining telomere length [14-16]. The lack of  $\Psi$  residues and the reduced telomerase activity may be responsible for dyskeratosis congenita.

## 1.2 Pseudouridine synthases

The pseudouridine synthases catalyze the isomerization of U to  $\Psi$  at specific positions in certain RNAs. Based on sequence data and structural analysis, the  $\Psi$  synthases fall into six different families that share no significant global sequence similarity [17-20] and are named after the first cloned member: TruA [21], TruB [22], RluA [23], RsuA [24], TruD [19], and Pus10 [25]; the first five are from *E. coli*, and Pus10 is found only in archaea and eukaryotes. Crystal structures from representative members of each family show that they share a common fold with a core  $\beta$ -sheet along with several conserved active site amino acid residues, so they likely share a common mechanism. The active site of all six families shows an absolutely conserved aspartic acid residue Asp, which was shown to be critical for enzyme activity and proposed to act as a nucleophile in the isomerization reaction [11, 12, 26, 27]. In cocrystal structures of several  $\Psi$  synthases with RNA, the conserved Asp is located in the same structural position [28].

The work in this thesis focuses on the *E. coli*  $\Psi$  synthases RluA and TruB. RluA (ribosomal large subunit pseudouridine synthases) is responsible for the isomerization of U746 in 23S rRNA and U32 in the anticodon stem-loop (ASL) of fives tRNA [19, 21-24]. TruB is responsible for the pseudouridylation of U55 in the T-arm stem loop (TSL) in tRNAs [22]. Mueller and co-workers have reported the kinetic parameters of RluA with ASL and full-length tRNA. The value of  $k_{\text{cat}}$  with ASL is only 1.5-fold lower than



the value with full-length tRNA, and the  $K_m$  value is 2.9-fold higher than with tRNA, indicating that ASL is only a very slightly poorer substrate than full-length tRNA [29]. Similar results were obtained by Santi and co-workers for TruB in comparison of full-length tRNA and the 17-mer corresponding to yeast TSL [30].

### 1.3 Proposed mechanism for synthases

Two general mechanisms have been proposed for the  $\Psi$  synthases. The first was the “Michael mechanism” (Figure 1.3). In this mechanism, the conserved Asp acting as a nucleophile attacks C6 of the pyrimidine ring to form a covalent adduct (a Michael adduct). Following cleavage of the N-glycosidic bond and rotation 180° around the new bond to the conserved Asp, reattachment at C5 forms the new C-glycosidic bond. The conserved Asp departs as leaving group, and deprotonation at C5 generates  $\Psi$ .

The second mechanism is the “acylal mechanism” (Figure 1.4), which involves nucleophilic attack by the conserved Asp on C1' of the ribose ring either through a concerted or step-wise process to form an acylal intermediate. After N-glycosidic bond breakage, the uracilate ion is free to rotate 180°, moving C5 close to C1' for C-glycosidic bond formation with the conserved Asp as leaving group. The subsequent deprotonation of C5 by the conserved Asp yields  $\Psi$ .

**5-Fluorouridine ( $F^5U$ ).** 5-Fluorouracil ( $F^5Ura$ ) has been known as a potent anti-cancer drug for over 60 years. It is widely used in the treatment of various type of cancers, particularly for colorectal and testicular cancers [31]. However, the mechanisms by which  $F^5Ura$  causes cell death and tumours become resistant to  $F^5Ura$  remains unclear, although the action of  $F^5Ura$  involves its conversion to 5-fluorouridine ( $F^5U$ ). Samuelsson demonstrated that  $F^5U$  inhibits the post-transcriptional conversion of uridine

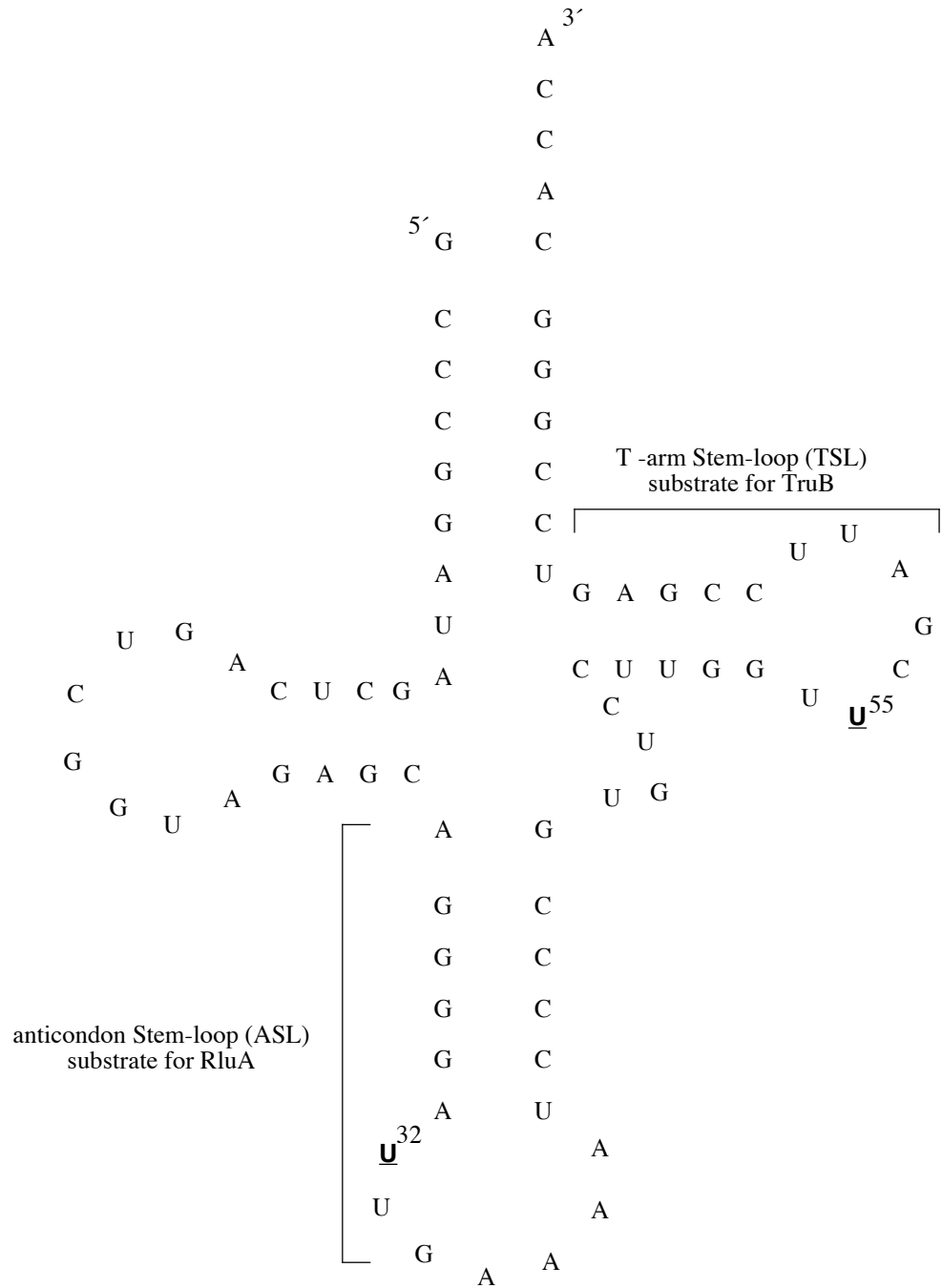
to pseudouridine in RNA species and that RNAs containing F<sup>5</sup>U form stable complexes with Ψ synthases [32]. F<sup>5</sup>U is also converted to several intracellular active metabolites such as 5-fluorodeoxyuridine monophosphate (F<sup>5</sup>dUMP), which acts as an inhibitor of thymidylate synthase and thus blocks the sole *de novo* source of dTTP, which is necessary for DNA replication and repair [31, 33]. To examine the mechanism of the Ψ synthases, RNA containing F<sup>5</sup>U ([F<sup>5</sup>U]RNA) was used as a mechanistic probe (Figure 1.5B).

When the Ψ synthase TruA was incubated with [F<sup>5</sup>U]tRNA, TruA was irreversibly inhibited and formed a covalent adduct between protein and RNA (Figure 1.5C), as judged by denaturing gels [30]. After heat disruption of the adduct, a hydrated product of F<sup>5</sup>U was observed, which was reasonably ascribed to ester hydrolysis of the Michael adduct and thus taken to support the Michael mechanism (Figure 1.3) [34]. However, a noncovalent complex was observed in cocrystals of TruB and [F<sup>5</sup>U]TSL, but the product of F<sup>5</sup>U was hydrated and also rearranged to the C-glycoside isomer [35]. Based on these observations, Hoang and Ferré-D'Amaré concluded that TruB follows the Michael mechanism and ascribed the absence of a covalent adduct to the slow ester hydrolysis of the Michael adduct during crystal growth and data acquisition [35]. A similar set of experiments with the Ψ synthase RluA revealed that it is also irreversibly inhibited and forms an adduct with [F<sup>5</sup>U]ASL, similarly to TruA when incubated with [F<sup>5</sup>U]tRNA [36].

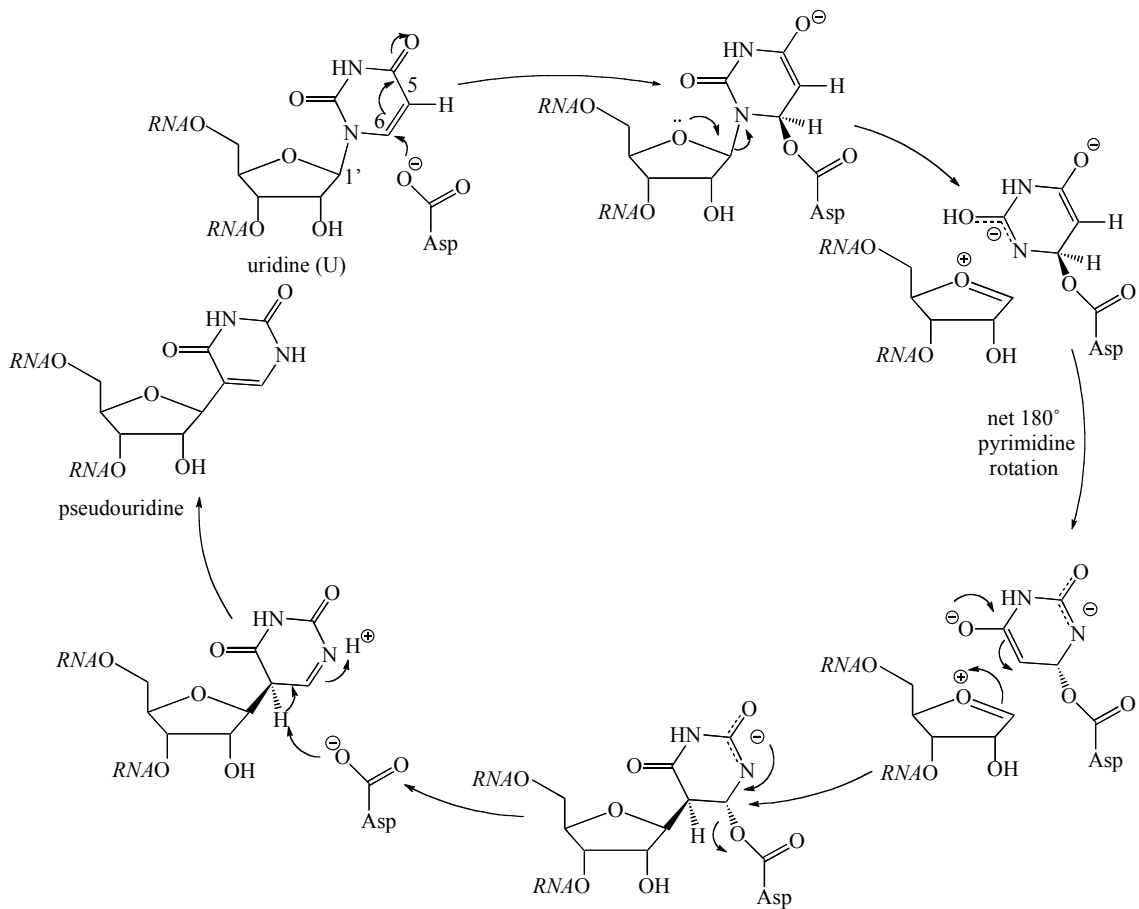
To determine if the products form by ester hydrolysis or direct hydration (Figure 1.6), reactions of [F<sup>5</sup>U]RNA and TruB [37], RluA [29], and TruA [38] were run in buffer containing 50% [<sup>18</sup>O]water. In all cases, <sup>18</sup>O label ends up in RNA rather than in

the conserved Asp. These results eliminated ester hydrolysis of the Michael adduct; instead, a water molecule adds directly to the pyrimidine ring. Based on these observations, Mueller and co-workers proposed a scheme to account for the handling of F<sup>5</sup>U, in which all Ψ syntheses proceeds through either the Michael or acylal mechanisms to rearranged F<sup>5</sup>U (Figure 1.7). With TruB, the conserved Asp does not form a stable covalent adduct, and the rearranged F<sup>5</sup>U is spontaneously hydrated when released into solution. In other Ψ syntheses (such as RluA and TruA), the conserved Asp can reach C6 to form a covalent adduct between the conserved Asp and [F<sup>5</sup>U]RNA, which undergoes elimination (rather than ester hydrolysis) to generate rearranged F<sup>5</sup>U upon heating, followed by spontaneous hydration in solution [29, 38].

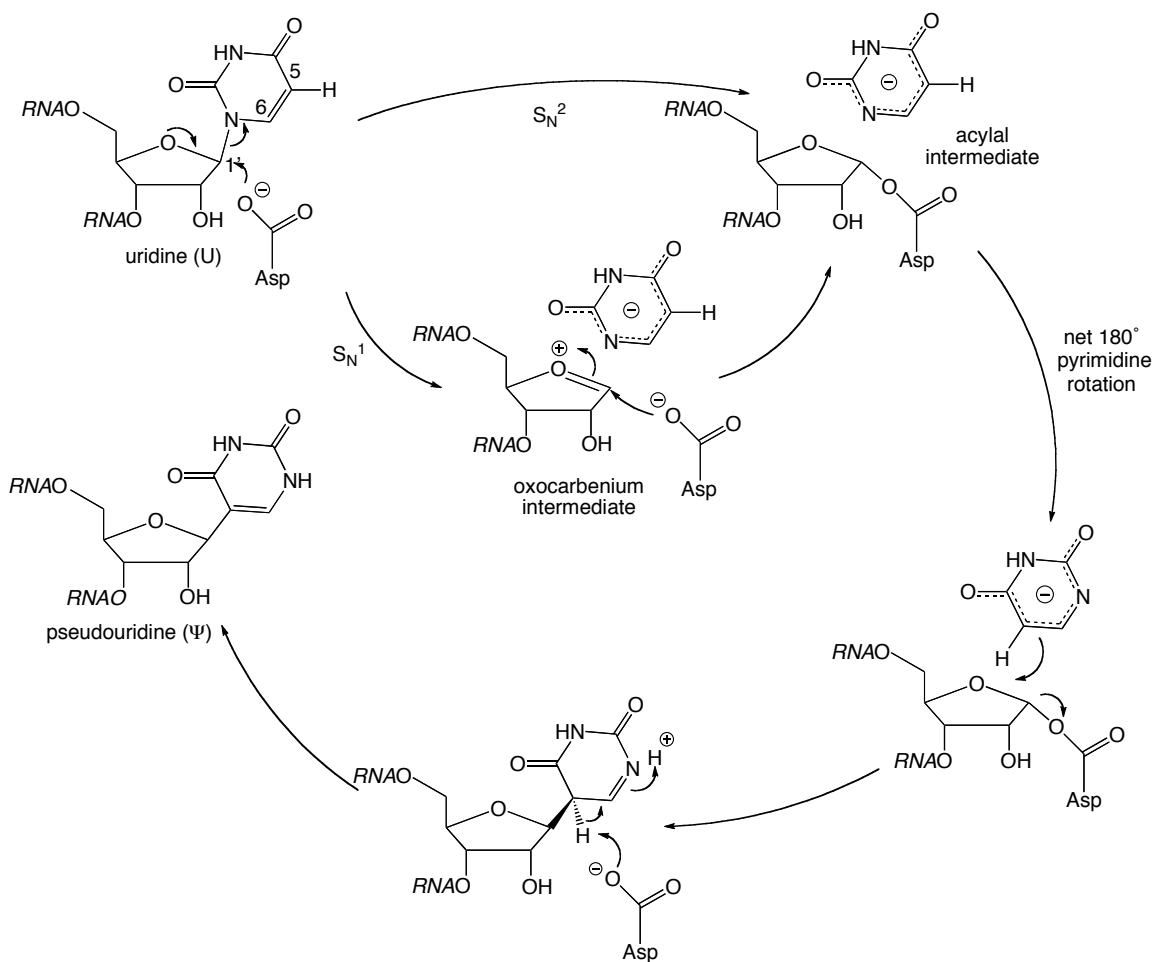
Studies by Spedalieri and Mueller proved that [F<sup>5</sup>U]TSL neither inhibits nor forms a stable adduct with TruB; in fact, TruB converts F<sup>5</sup>U into its rearranged and hydrated C-glycoside isomer in a time frame similar to the natural conversion of U to Ψ [36]. Furthermore, HPLC analysis of the nucleotides resulting from treatment with S1 nuclease and alkaline phosphatase of reacted [F<sup>5</sup>U]TSL showed two more polar products of F<sup>5</sup>U in a ratio of ~3:1 [36]. MALDI-MS analysis confirmed that F<sup>5</sup>U became hydrated after incubation with TruB, consistent with products seen in the cocrystal. NMR experiments revealed that both products of F<sup>5</sup>U were dinucleotides with the cytidine residue that follows F<sup>5</sup>U in [F<sup>5</sup>U]TSL, and an identical result was obtained from the action of TruA on [F<sup>5</sup>U]RNA [38]. However, HPLC analysis of the RluA products showed a single peak with different retention time than either TruB product [29].



**Figure 1.2:** The ‘cloverleaf’ representation of *E. coli* tRNA<sup>Phe</sup>. The anticodon stem loop (ASL) is the substrate for RluA, and T-arm stem loop (TSL) is the substrate for TruB, which isomerizes U32 and U55 (underlined), respectively.

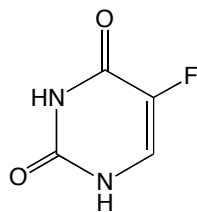


**Figure 1.3:** The proposed “Michael mechanism” for  $\Psi$  synthases. The conserved Asp nucleophilically attacks C6 of the pyrimidine ring to form a covalent adduct (a Michael addition). N-glycosidic bond cleavage occurs, and the pyrimidine ring rotates 180° around the bond to the conserved Asp. Reattachment at C5 forms the C-glycosidic bond, followed by the elimination of Asp and deprotonation to complete the isomerization.

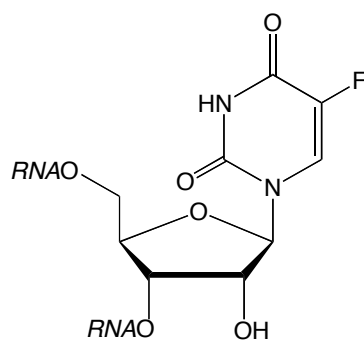


**Figure 1.4:** The proposed “acyl mechanism” for  $\Psi$  synthases. The nucleophilic attack by the conserved Asp on C1' of the ribose ring proceeds through either a concerted or step-wise process to form an acylal intermediate. After N-glycosidic bond breakage, the uracilate ion is free to rotate 180°, bringing C5 close to C1' to form the C-glycosidic bond. Deprotonation by the conserved Asp yields  $\Psi$ .

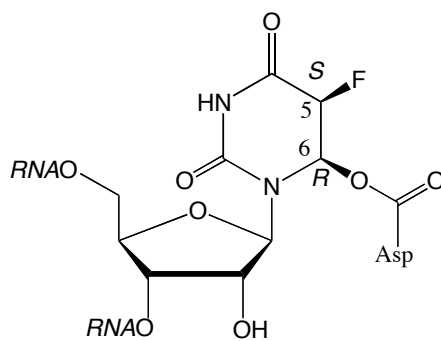
**A**



**B**

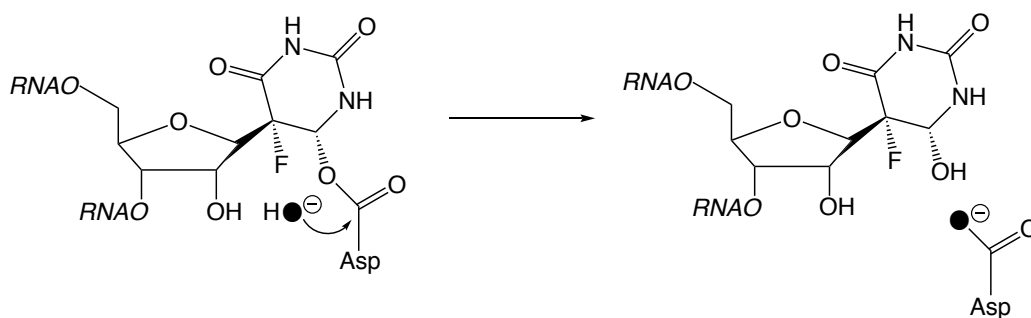


**C**

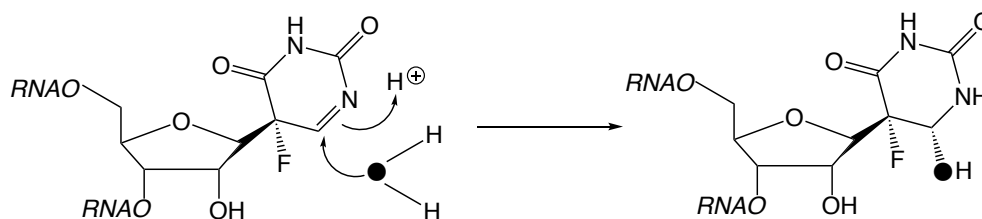


**Figure 1.5:** A, 5-fluorouracil. B, 5-fluorouridine. C, The putative Michael adduct between TruA and [F<sup>5</sup>U]RNA proposed by Santi [30].

**A**

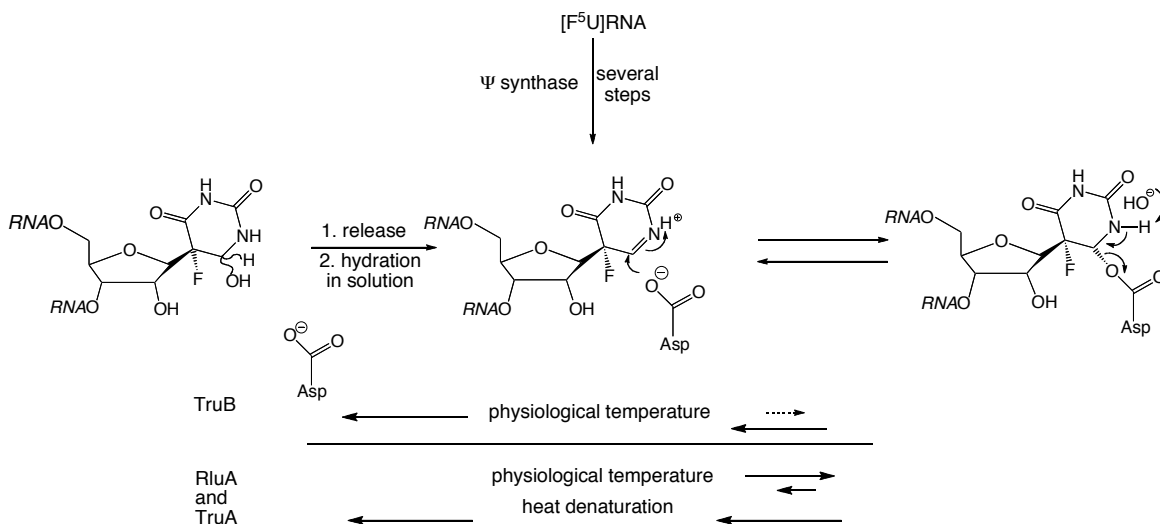


**B**



**Figure 1.6:**  $^{18}\text{O}$  labeling scheme. **A**, Ester hydrolysis would result in  $^{18}\text{O}$  incorporation into the conserved Asp. **B**, Direct hydration of the rearranged product of  $\text{F}^5\text{U}$ .  $^{18}\text{O}$  is abbreviated with a filled O.



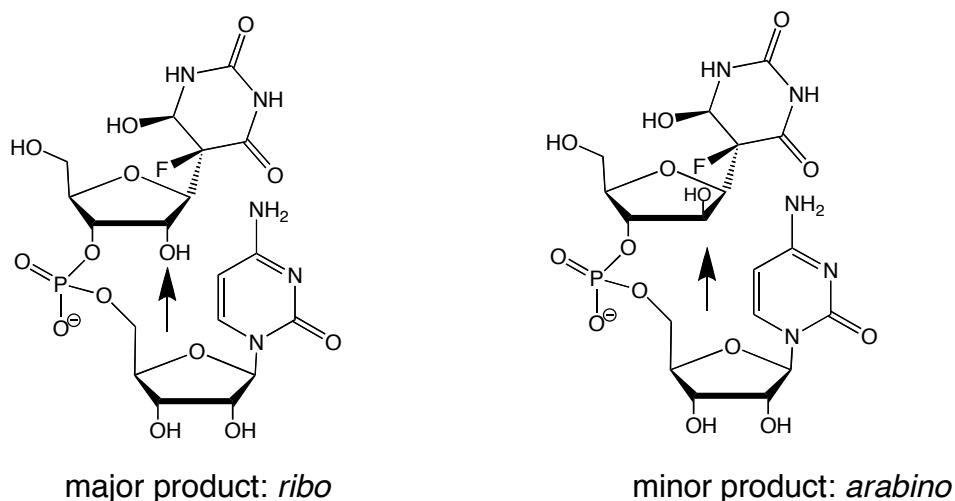


**Figure 1.7:** A consistent scheme for the handling of  $F^5U$  in RNA by different  $\Psi$  synthase [38].

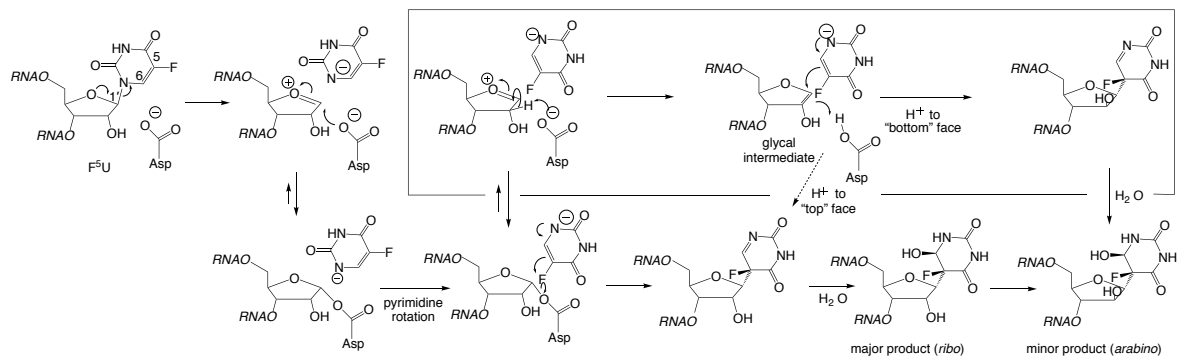
Since uridine rather than cytidine follows  $F^5U$  in  $[F^5U]ASL$ , the dinucleotide products of  $F^5U$  explain the different HPLC behavior between TruB and RluA products. Subsequent NMR and MS analysis confirmed the formation of two RluA products that are both dinucleotides with uridine and comigrate under HPLC conditions [38].

To investigate the mechanism further, the full structural elucidation of the two products of TruB was undertaken by Mueller and Miracco [39]. NMR results showed that the difference between the two products was at  $C2'$  of its pentose ring, which means the minor product is the *arabino* isomer whereas the major product was the *ribo* isomer (Figure 1.8). For the formation of the *arabino* product, epimerization at  $C2'$  must occur, which requires deprotonation from the “top” face to form a glycal intermediate followed by reprotonation from the “bottom” face. To determine whether the protonation at  $C2'$  is

directly from solution or an active site acid, reactions of RNA containing either U or F<sup>5</sup>U with TruB and RluA were run in buffer made with deuterated water, and the products were examined for the incorporation of deuterium into a C2' position. No “wash-in” was observed with either TruB or RluA, indicating that no direct protonation from solution occurred; therefore, the essential Asp must transfer the proton removed from U to the “wrong” face of the glycal intermediate to generate an *arabino* isomer [40]. This result strongly disfavors the Michael mechanism because the Asp would be tied up in an ester bond, making it unable to protonate C2'. The acylal mechanism can accommodate the *arabino* product of F<sup>5</sup>U because the decreased nucleophilicity of the anion of 5-fluorouridine versus uracil provides a longer lifetime of the acylal intermediate, which is in equilibrium with an oxocarbenium species and the free conserved Asp. The acidity of C2' in the oxocarbenium species would facilitate deprotonation by the conserved Asp, and reprotonation from the opposite face would result in an *arabino* product (Figure 1.9).



**Figure 1.8:** The major and minor products of F<sup>5</sup>U\* differ in stereo configuration at C2' [39]

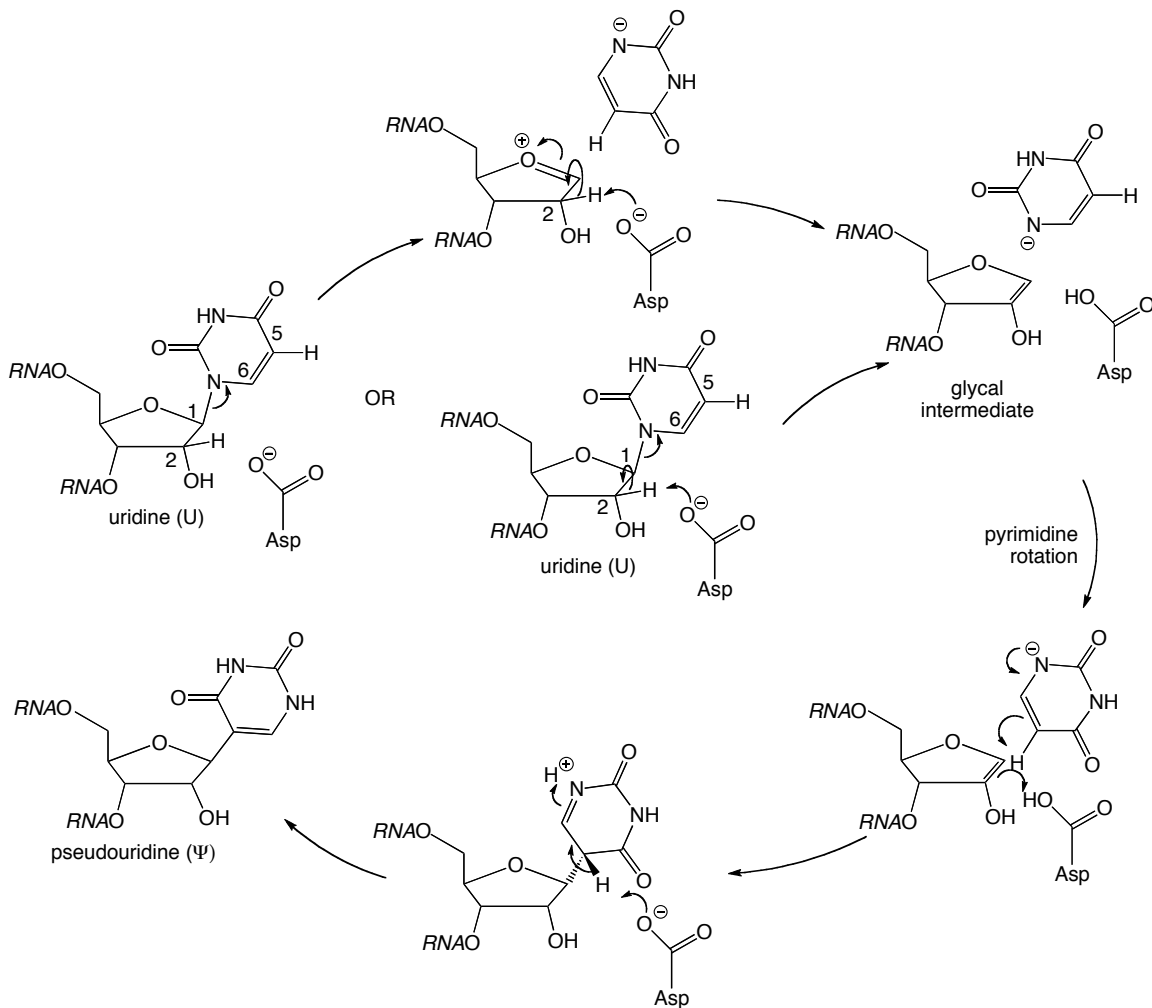


**Figure 1.9:** The acyl mechanism recast to account for the formation of the *arabino* product of. The boxed manifold exists only for F<sup>5</sup>U because of the increased life time of the acyl intermediate, which is equilibrium with the oxocarbenium ion. [39]

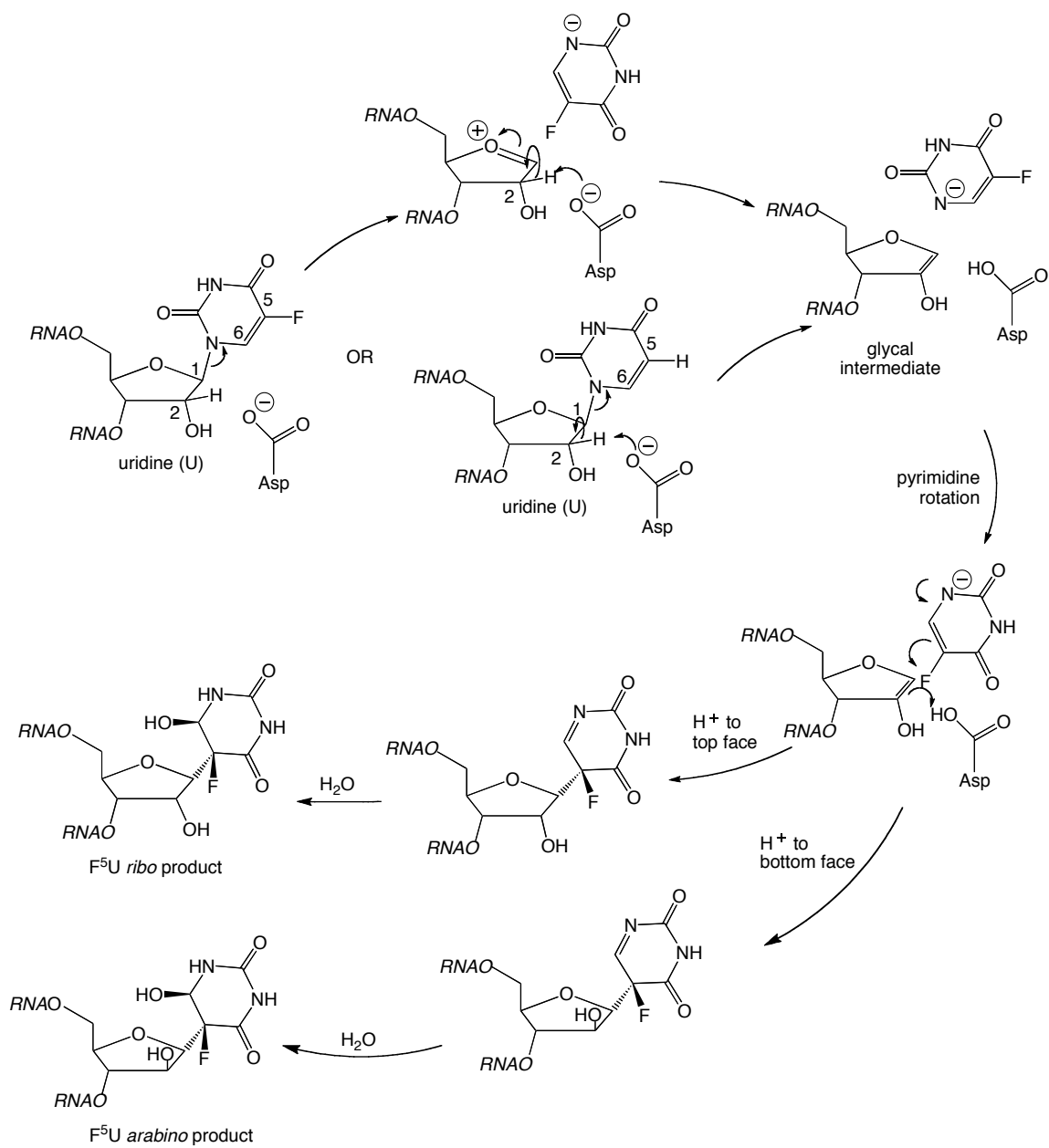
The existence of the glycal intermediate opens a new mechanistic possibility: the “glycal mechanism” (Figure 1.10), which can easily explain both the major and minor products of F<sup>5</sup>U (Figure 1.11). With normal substrate (RNA containing U), the mechanism begins with the deprotonation at C2' to eliminate pyrimidine ring and form the glycal intermediate, followed by rotation of the detached pyrimidine ring. Reattachment at C5 forms the new C-glycosidic bond with reprotonation of C2'. Deprotonation at C5 yield  $\Psi$  and completes the isomerization.

To test if the  $\Psi$  synthesis proceeds through the glycal mechanism, the 2'-deuterated stem-loop substrates for TruB and RluA were prepared, and their reaction rates were compared with unlabeled substrates. The observed kinetic isotope effects with both TruB and RluA clearly indicate that deprotonation or reprotonation of C2' is

partially rate-limiting, which provides direct evidence for the operation of glycal mechanism during the conversion of U to  $\Psi$  [40].

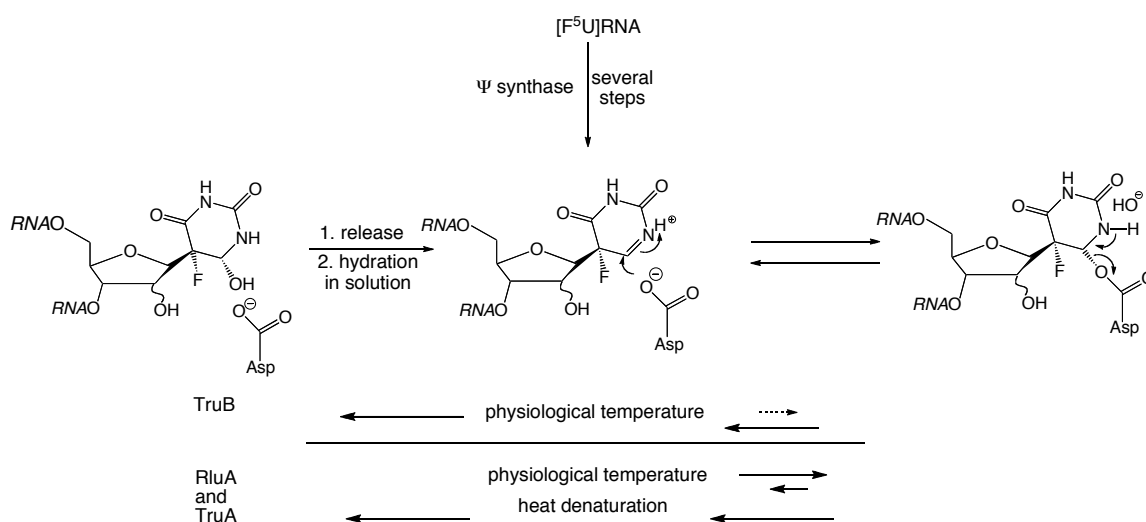


**Figure 1.10:** The “glycal mechanism” for  $\Psi$  formation. The elimination of uracil proceeds through either a concerted (lower path) or step-wise (upper path) to form a glycal intermediate. After the C-glycoside rearrangement, C2' is reprotonated by the conserved Asp, followed by deprotonation to generates  $\Psi$ . [39]



**Figure 1.11:** The “glycol mechanism” with  $F^5U$ . Reprotonation from the “bottom” face of the glycal ring yields the *arabino* product of  $F^5U$ .

The Mueller group proposed a mildly revised scheme to account for the *arabino* product of F<sup>5</sup>U (Figure 1.12). The formation of products of F<sup>5</sup>U proceed through either the glycol mechanism or the acylal mechanism (recast for the observed *arabino* product) to rearranged F<sup>5</sup>U.

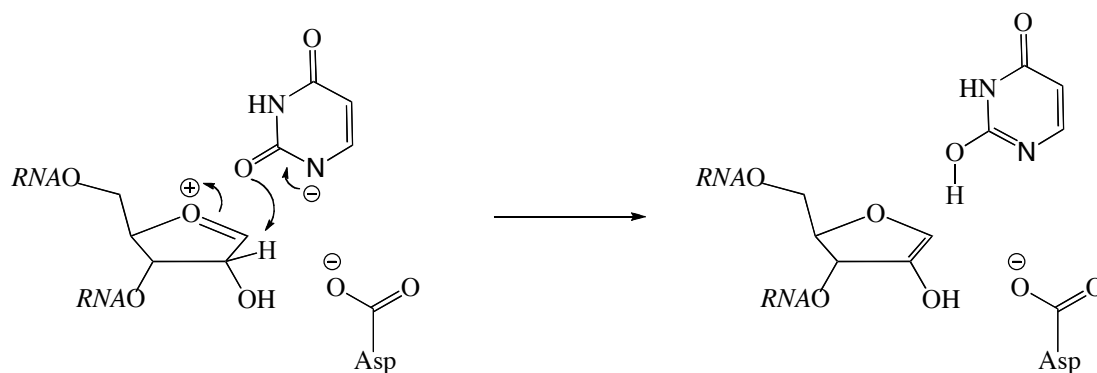


**Figure 1.12:** A scheme proposed by the Mueller group consistent with all observations, including the *arabino* product of F<sup>5</sup>U.

**4-Thiouridine as a mechanistic probe.** Three different mechanisms have been proposed for the  $\Psi$  synthases: the Michael (Figure 1.3), acylal (Figure 1.4), and glycal (Figure 1.10) mechanisms. To explain both the *ribo* and *arabino* products of F<sup>5</sup>U, the Mueller group proposed the glycal mechanism for the  $\Psi$  synthases, which is formulated with deprotonation at C2' by the conserved Asp to form the glycal intermediate. O<sup>2</sup> of uracil could also deprotonate at C2' (Figure 1.13); however, in this case, the proton must migrate to the conserved Asp in order to reprotonate at C2' from the opposite face to yield the *arabino* product since the bulky pyrimidine ring may not allow O<sup>2</sup> access the “bottom face” of the glycal intermediate. In the crystal structures of the glycal complexes, O<sup>2</sup> of fluorouracil is positioned directly above C2' in a good position to act as the base for glycal formation [41].

To test if O<sup>2</sup> is the general base, the isomerized U will be replaced with 4-thiouridine (s<sup>4</sup>U). Some bacteria naturally have s<sup>4</sup>U at position 8 of several tRNA, and this s<sup>4</sup>U not only stabilizes the fold of the tRNA [42, 43] but also plays an important role as a photosensor for near-UV light [44]. Replacing oxygen with sulfur at position 4 of the pyrimidine ring results in a significantly lower pK<sub>a</sub> for s<sup>4</sup>U (8.2) than for U (9.3) [42], making the anion of 4-thiouracil more stable and a better leaving group than the anion of uracil itself and hence increasing the rate of reaction. However, the lower pK<sub>a</sub> of s<sup>4</sup>U also means that 4-thiouracil(ate) is a weaker base than uracil(ate); therefore deprotonation of C2' should slow, thus decreasing the rate of reaction. The rate of the overall reaction depends on two processes: leaving group departure and deprotonation of C2', and s<sup>4</sup>U is expected to have opposite effects on the speed of each. A slower rate of overall reaction will imply that O<sup>2</sup> is the general base since slower deprotonation is the expected effect of

$s^4U$  substitution. A faster rate is less informative because  $O^2$  could still be the base that deprotonates  $C2'$ , but the effect of being a better leaving group dominates. To ensure  $s^4U$  is handled as a substrate and to identify the products, assays of  $s^4U$  with TruB and RluA were performed and analyzed by HPLC. This work with  $s^4U$  in a RNA substrate is described in Chapter 3 of this thesis.



**Figure 1.13:** Deprotonation at  $C2'$  by  $O^2$ . The proton then migrates to  $O^4$  or the conserved Asp to reprotonate at  $C2'$ . Since neither  $O^2$  nor  $O^4$  seem likely to be able to access the “bottom” face of the glycol intermediate, the *arabino* product of  $F^5U$  requires migration of the removed proton to the conserved Asp.



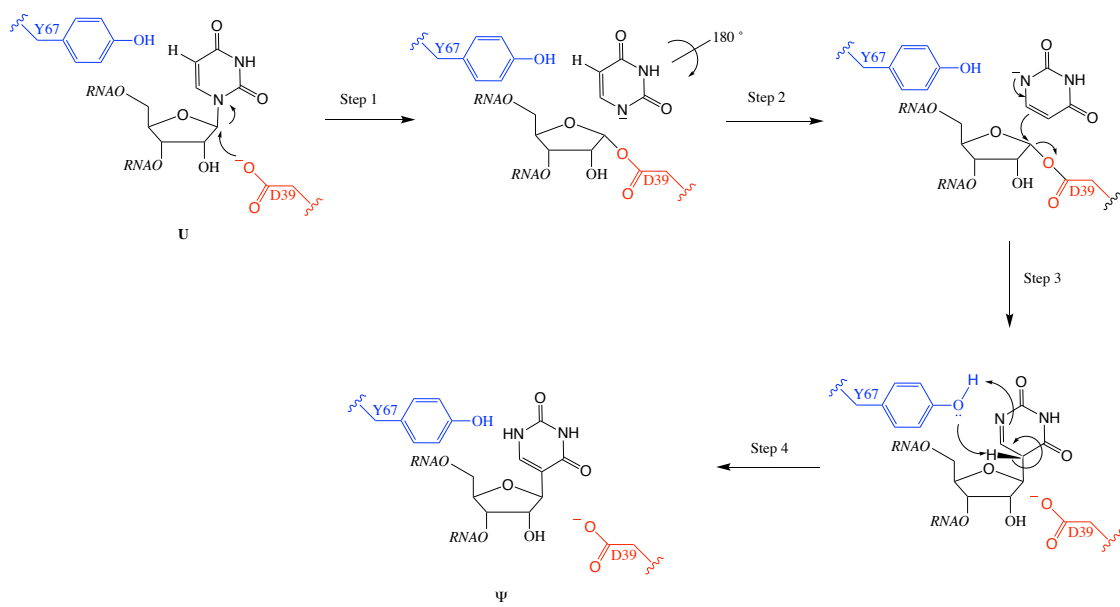
## CHAPTER II

### MECHANISTIC INVESTIGATION OF Y96F RluA USING ASL AND ASL CONTAINING 5-FLUOROURIDINE

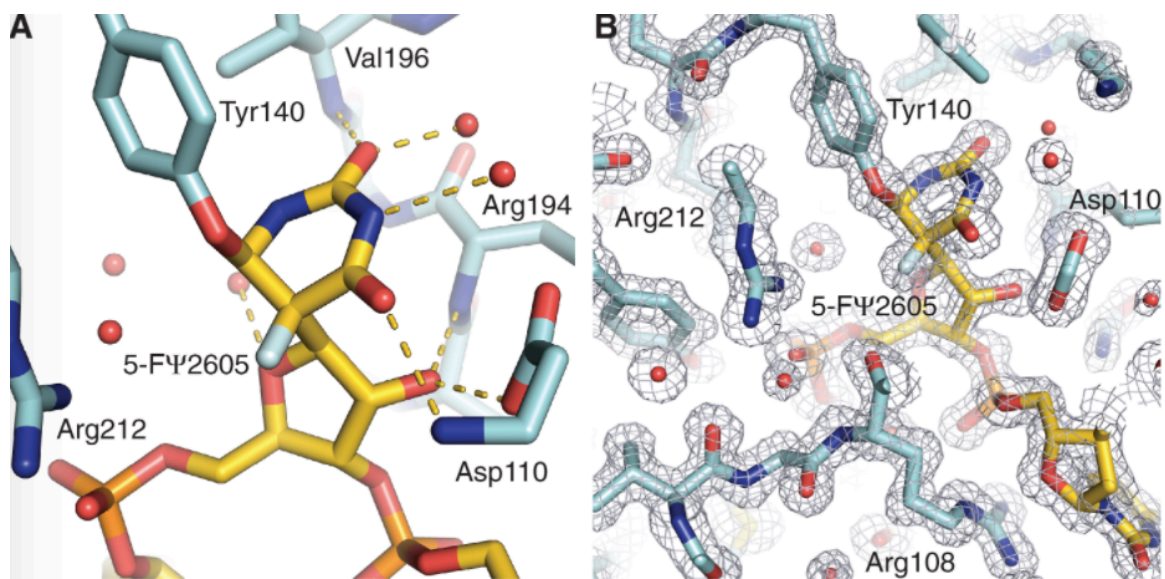
#### 2.1 Introduction

In chapter 1, three  $\Psi$  synthase mechanisms were presented, and they all involve an active site aspartic acid residue (Asp) acting as a nucleophile or a base [34, 39]. In the “Michael mechanism”, the conserved Asp acts as a nucleophile to attack C6 of the pyrimidine ring to form a Michael adduct (Figure 1.3). The alternative “acylal mechanism” instead proposes a nucleophilic attack by the conserved Asp on C1' of the ribose ring (Figure 1.4). The third mechanism is the “glycal mechanism”, which involves the conserved Asp acting as base to deprotonate at C2' to form a glycal intermediate (Figure 1.10). Huang and co-workers noted a conserved tyrosine (Tyr) residue in the active site in four of the five families of  $\Psi$  synthases (Phe in the TruD family) and conducted site-directed mutagenesis on this Tyr [45]. When Y76F TruB was incubated with natural RNA substrate, no conversion of U to  $\Psi$  was observed. However, when F<sup>5</sup>U replaced the isomerized U in the stem-loop, Y76F TruB catalyzes the reaction of [F<sup>5</sup>U]RNA, making a rearranged and hydrated product just like wild-type TruB [45]. Based on these observations, Huang concluded that the hydroxyl group in the side chain of Tyr-76 must play a critical role in the final step of the reaction, which is deprotonation

at C5 to form the product of  $\Psi$  (Figure 2.1). This step does not happen with  $[F^5U]RNA$  since a base cannot remove the fluorine as it does a proton. The rationale seems to be sound; however, there is evidence arguing against the role of Tyr-76 serving as a general base. First, if Tyr-76 is a general base and only conducts the final step of the reaction (removes the proton from C5 of rearranged U), then the other steps should still occur, allowing the rearrangement of uridine to the C-glycoside. Since the intermediate is more polar than U, it would elute at an earlier retention time than U from a reverse-phase HPLC column, but no new peaks were detected [45]. Second, the pH profile studies on TruB from the Mueller group shows a descending limb with a  $pK_a$  of 9, suggesting that the deprotonation of Tyr could slow the reaction by disrupting the electrostatic environment of the active site [46]. Therefore, Mueller and co-workers proposed that Tyr-76 donates either a proton or hydrogen bond to the pyrimidine ring (the leaving group) to facilitate N-glycosidic bond cleavage. Third, the conserved Tyr seen in the first four families of the  $\Psi$  synthase was not found in TruD family; instead, Phe residues occupy in this position. Finally, the crystal structure of wild-type RluB with  $[F^5U]RNA$  shows a covalent bond of the hydroxyl group of Tyr-140 and the pyrimidine ring of  $F^5U$  (Figure 2.2) [47]. To confirm the function of Tyr-76, additional experiments are required. As described, in this chapter, Y96F RluA was overexpressed and incubated with both ASL and  $[F^5U]ASL$  in order to illuminate the function of the conserved Tyr in the formation of  $\Psi$  and to test whether results with Y76F TruB are common to another family of  $\Psi$  synthases. Reverse-phase HPLC analysis of both intact and digested RNA from each reaction were analyzed.



**Figure 2.1:** Mechanism proposed by Phannachet *et al* for the Ψ synthases [45]



**Figure 2.2.** The crystal structural shows a covalent bond between Tyr-140 and the pyrimidine ring. From Ref [47]

## 2.2 Experimental

### 2.2.1 Materials

Competent BLR(DE3) pLysS *E. coli* cells were purchased from Novagen (Madison, WI). A CLYPEUS C<sub>8</sub> 5- $\mu$ m column (250  $\times$  4.6 mm) was purchased from Higgins Analytical, Inc. (Mountain View, CA), and a Zorbax analytical SB-C<sub>18</sub> 5- $\mu$ m column (50  $\times$  4.6 mm) was purchased from Agilent (Santa Clara, CA). Ni-NTA superflow resin and QIAprep Spin Mini prep kits were purchased from Qiagen (Chatsworth, CA). S1 nuclease and calf intestinal alkaline phosphatase were purchased from Promega (Madison, WI). Oligonucleotide ASL (5'-GGGGAUUGAAAAUCCCC-3') and [<sup>5</sup>U]ASL (5'-GGGGAF<sup>5</sup>UUGAAAAUCCCC-3') were purchased from Dharmacon (Lafayette, CO). Ultra-pure deionized water was obtained using a Millipore<sup>®</sup> Milli-Q integral system (Billerica, MA) equipped with a 0.2  $\mu$ m filter. Laemmli buffer (2 $\times$ ) was 120 mM Tris•HCl buffer, pH 6.8, containing SDS (4%, w/v), glycerol (20%, v/v), and bromophenol blue (0.02%, w/v). RNA loading dye (2 $\times$ ) was 95% aqueous formamide containing bromophenol blue (0.025%, w/v), xylene cyanol (0.025%, w/v), and EDTA (0.5 mM).

### 2.2.2 Overexpression and Purification of Y96F RluA

**Overexpression.** An overnight culture of BLR(DE3) pLysS/pBH222 was used to inoculate LB broth (1 L) in a baffled flask and shaken vigorously at 37 °C. When OD<sub>600</sub> reached 0.4–0.6, isopropyl- $\beta$ -D-thiogalactopyranoside (4 mL, 100 mM) was added (0.4 mM final concentration). The cells were harvested 3 h after induction (OD<sub>600</sub> = 1.1–1.2). The cells were centrifuged at 6,000g for 30 min at 4 °C and resuspended in lysis buffer (10 mL), which is 50 mM sodium phosphate buffer, pH 8.0, containing imidazole

(10 mM) and NaCl (300 mM). The suspensions were quick-frozen and stored at  $-80^{\circ}\text{C}$ .

**Purification.** The cell pellets containing overexpressed Y96F RluA were thawed and sonicated, and then the lysate was centrifuged (13,500g for 30 min) to pellet cell debris. A slurry of Ni-NTA superflow resin (3 mL settled resin volume) was added to the supernatant and nutated for 1 h to bind the His-tagged Y96F RluA to the resin. The resin was pelleted (7500g for 30 min), resuspended in lysis buffer (3 mL), and packed into a column. The column was washed ( $3 \times 5$  mL) with 50 mM sodium phosphate buffer, pH 8.0, containing imidazole (20 mM) and NaCl (300 mM). The bound Y96F RluA was eluted ( $2 \times 5$  mL) with 50 mM sodium phosphate buffer, pH 8.0, containing imidazole (250 mM) and NaCl (300 mM). The elutions were dialyzed in 20 mM HEPES buffer, pH 7.5, EDTA (0.5 mM),  $\beta$ -mercaptoethanol (1 mM), and glycerol (5% v/v) for  $2 \times 2$  h. The Y96F RluA was then further purified by POROS HS anion exchange chromatography, eluting with a linear gradient of KCl (0–1 M) in 20 mM HEPES buffer, pH 7.5, containing EDTA (0.5 mM),  $\beta$ -mercaptoethanol (1 mM), and glycerol (5% v/v). Each protein-bearing fraction (1 mL) was analyzed for purity by SDS-PAGE.

**RNase Activity Assays.** The fractions containing Y96F RluA were tested for RNase contamination. The RluA-bearing fractions from POROS HS chromatography were assayed at a constant protein concentration (100 nM). Each RluA-bearing fraction (100 nM) was incubated with RNA markers at  $37^{\circ}\text{C}$  for 1 h, which were then analyzed by urea-PAGE. Fractions that were pure by SDS-PAGE and RNase-free were combined and dialyzed ( $2 \times 2$  h) against 23.5 mM HEPES buffer, pH 7.5, containing KCl (176.5 mM), EDTA (0.58 mM), DTT (0.58 mM), and glycerol (5% v/v). Y96F RluA concentrations were determined using  $A_{280}$  and  $\epsilon_{280} = 30700 \text{ M}^{-1}\text{cm}^{-1}$ .

### 2.2.3 Deprotection and Purification of 2'-ACE RNA Oligonucleotides

Synthetic oligonucleotide (17-mer) stemloop (ASL or [F<sup>5</sup>U]ASL) was deprotected according to the manufacturer's protocol. Lyophilized ASL or [F<sup>5</sup>U]ASL (340 nmol) was dissolved in 100 mM acetate buffer, pH 3.8, made by the addition of tetramethylethylenediamine (TEMED) to a solution of acetic acid, and incubated at 60 °C for 2 h. The deprotected ASL or [F<sup>5</sup>U]ASL was lyophilized using a Speed Vac concentrator and then dissolved in water (100 µl). ASL and [F<sup>5</sup>U]ASL were further purified by HPLC over a C<sub>8</sub> preparatory column (Clipeus C<sub>8</sub> 5µm, 250 × 10 mm, Higgins Analytical, Mountain View, CA), eluting with a gradient of acetonitrile (0–40%) in 5 mM ammonium acetate buffer, pH 6.0. The ASL or [F<sup>5</sup>U]ASL was lyophilized, redissolved in water, and stored at –20 °C. The final ASL or [F<sup>5</sup>U]ASL concentration was determined using A<sub>260</sub> and an extinction coefficient (170500 M<sup>-1</sup>cm<sup>-1</sup>) provided by Dharmacon.

### 2.2.4 Assay of Y96F RluA Activity

**Reaction of ASL.** The reaction mixture (800 µL) was 50 mM HEPES buffer, pH 7.5, containing sodium chloride (175 mM), DTT (5 mM), EDTA (1 mM), RNase inhibitor (30 units), and ASL (50 µM). After pre-equilibration at 37 °C, reaction was initiated by the addition of Y96F RluA (to 5 µM). Aliquots (100 µL) were removed at various times and immediately heated at 97 °C for 10 min. The protein precipitate was pelleted by centrifugation, and the supernatant was transferred to a fresh tube. Residual protein was extracted from the supernatant with an equal volume of phenol/chloroform (1:1), and the aqueous layer was removed and transferred to a fresh tube. One tenth volume of 0.3 M sodium acetate buffer pH 6.0, was added and well-mixed followed by cold absolute ethanol (330 µL), and the mixture placed in –80 °C for 2 h. The supernatant

was removed, and the pellet was rinsed with cold 70% aqueous ethanol (100  $\mu$ L), air-dried, and resuspended in 50 mM HEPES buffer (100  $\mu$ L), pH 7.5, containing sodium chloride (175 mM) and EDTA (1mM) then passed through a 0.2  $\mu$ m filter before being analyzed by reverse phase HPLC (Cliepus C<sub>18</sub> 5  $\mu$ m column, 150  $\times$  4.6 mm; Higgins Analytical, Mountain View, CA). RNA was eluted with an acetonitrile gradient by the following program (the first number is the percentage of aqueous acetonitrile, 40% v/v, in 25 mM ammonium acetate buffer, pH 6.0; the second number is the elapsed time in minutes): 0, 0; 0, 3; 5, 8; 15, 18; 15, 23; 30, 26; 50, 27.5; 50, 29; 100, 30; 100, 31; 0, 32; 0, 36.

***Digestion of ASL.*** To confirm the formation of product, ASL was digested after incubation with Y97F RluA and ethanol precipitation. The protocol described above was followed through the phenol/chloroform extraction. The air-dried pellet was resuspended in 50 mM sodium acetate buffer (100  $\mu$ L), pH 4.5, containing sodium chloride (280 mM) and zinc chloride (4.5 mM). S1 nuclease (100 units) was added; after 1 h at 37 °C, the digestion mixture was heated to 100 °C for 5 min and then cooled on ice. Additional S1 nuclease (100 units) and alkaline phosphatase (5 units) were added, and the digestion mixture was incubated for an additional 3 h at 37 °C, then passed through a 0.2  $\mu$ m filter and analyzed by reverse-phase HPLC as described for intact ASL.

***Reaction of [<sup>5</sup>U]ASL.*** The reaction mixture (1.00 mL) was 50 mM HEPES buffer, pH 7.5, containing sodium chloride (175 mM), DTT (5 mM), EDTA (1 mM), RNase inhibitor (30 units), and [<sup>5</sup>U]ASL (2  $\mu$ M). After pre-equilibration at 37 °C, the reaction was initiated by the addition of Y96F RluA (to 1  $\mu$ M). Aliquots (500  $\mu$ L) were removed at 3 h and immediately incubated at 97 °C for 10 min. The protein precipitate

was pelleted by centrifugation, and the supernatant was transferred to a fresh tube. Residual protein was extracted from the supernatant with an equal volume of phenol/chloroform (1:1), and the aqueous layer was removed and transferred to a fresh tube. One tenth volume of 0.3 M sodium acetate buffer, pH 6.0, was added and well-mixed followed by cold absolute ethanol (1.65 mL), and the mixture placed at  $-80\text{ }^{\circ}\text{C}$  for 2 h. The supernatant was removed, and the pellet was rinsed with cold 70% aqueous ethanol (100  $\mu\text{L}$ ), air-dried, and resuspended in 50 mM HEPES buffer (100  $\mu\text{L}$ ), pH 7.5, containing sodium chloride (175 mM) and EDTA (1mM) then passed through a 0.2  $\mu\text{m}$  filter and analyzed by reverse-phase HPLC as described for intact ASL.

***Digestion of [ $\text{F}^5\text{U}$ ]ASL.*** The digestion to nucleosides of [ $\text{F}^5\text{U}$ ]ASL before or after incubation with Y96F RluA was performed by the method described above for the digestion of ASL.

#### *2.2.5 Formation of the Adduct between Y96F RluA and [ $\text{F}^5\text{U}$ ]ASL*

The adduct between Y96F RluA and [ $\text{F}^5\text{U}$ ]ASL was formed by incubating Y96F RluA (1  $\mu\text{M}$ ) and [ $\text{F}^5\text{U}$ ]ASL (2  $\mu\text{M}$ ) in the reaction buffer for 3 h at  $37\text{ }^{\circ}\text{C}$ . The sample (60  $\mu\text{L}$ ) was then diluted with 2 $\times$  Laemmli buffer (60  $\mu\text{L}$ ) and split into two samples of equal volume; one was heated at  $97\text{ }^{\circ}\text{C}$  for 10 min, and the other was kept on ice. Aliquots of each sample were analyzed by SDS-PAGE (10% gel) to verify the presence of adduct, which was indicated by a band that shifted to a higher apparent molecular weight.

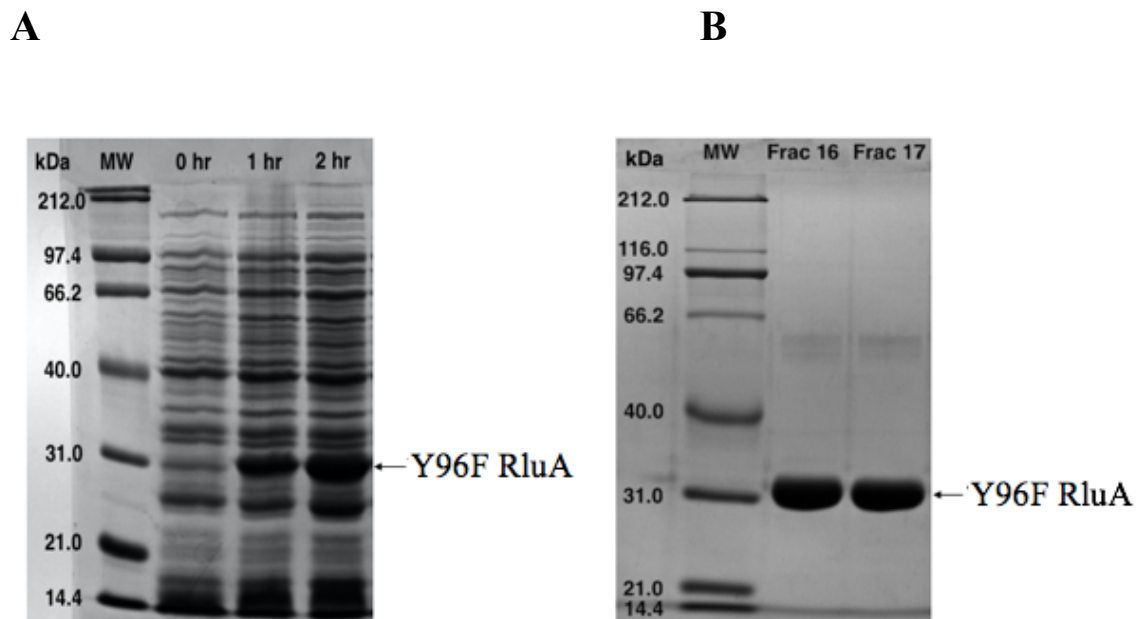
## **2.3 Results**

### *2.3.1 Overexpression and Purification of Y96F RluA*

The expression vector pBH222 was used to encode Y96F RluA. Protein was



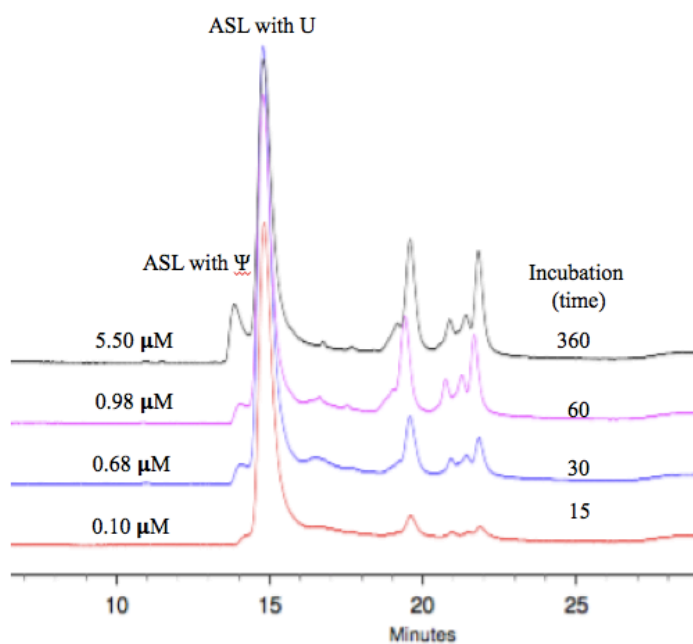
overexpressed and purified by nickel affinity chromatography followed by cation exchange chromatography. For long term stability, pure fractions of Y96F RluA were combined, concentrated to 2–4 mg/ml, and made up to 20% (v/v) with added glycerol, and stored at  $-20^{\circ}\text{C}$ .



**Figure 2.3:** SDS-PAGE analysis of the overexpression and purification of Y96F RluA (28 kDa). A, Overexpression of Y96F RluA. B, purified fractions of Y96F RluA. MW, molecular weight markers, wide range (Fisher Scientific; Waltham, MA).

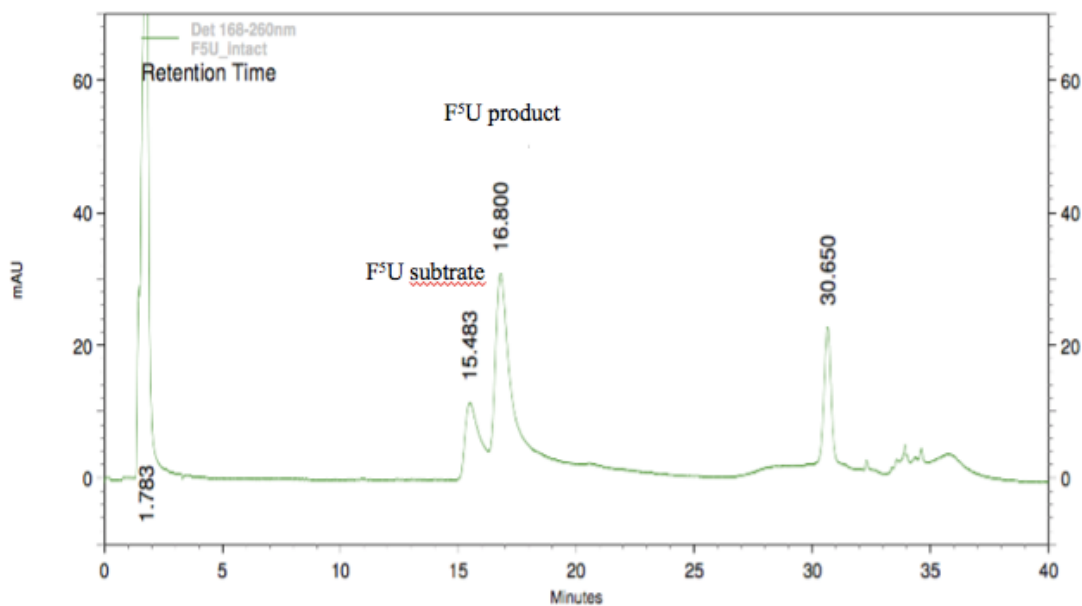
### 2.3.2 Y96F RluA isomerizes uridine and F<sup>5</sup>U in stem-loop substrate

To verify that Y96F RluA isomerizes the U in ASL and F<sup>5</sup>U in [F<sup>5</sup>U]ASL, both ASL (50  $\mu$ M) and [F<sup>5</sup>U]ASL (2  $\mu$ M) were incubated with Y96F RluA (5  $\mu$ M and 1  $\mu$ M, respectively). Intact ASL and [F<sup>5</sup>U]ASL were both analyzed by reverse-phase HPLC. The ASL peak shifted to a shorter retention time after incubation with Y96F RluA, which is similar to the behavior of wild-type RluA when it incubated with ASL. However, the HPLC analysis showed less than 11% conversion of substrate ASL (with U) to product ASL (with  $\Psi$ ) after 6 h, which is consistent with only a single turnover.

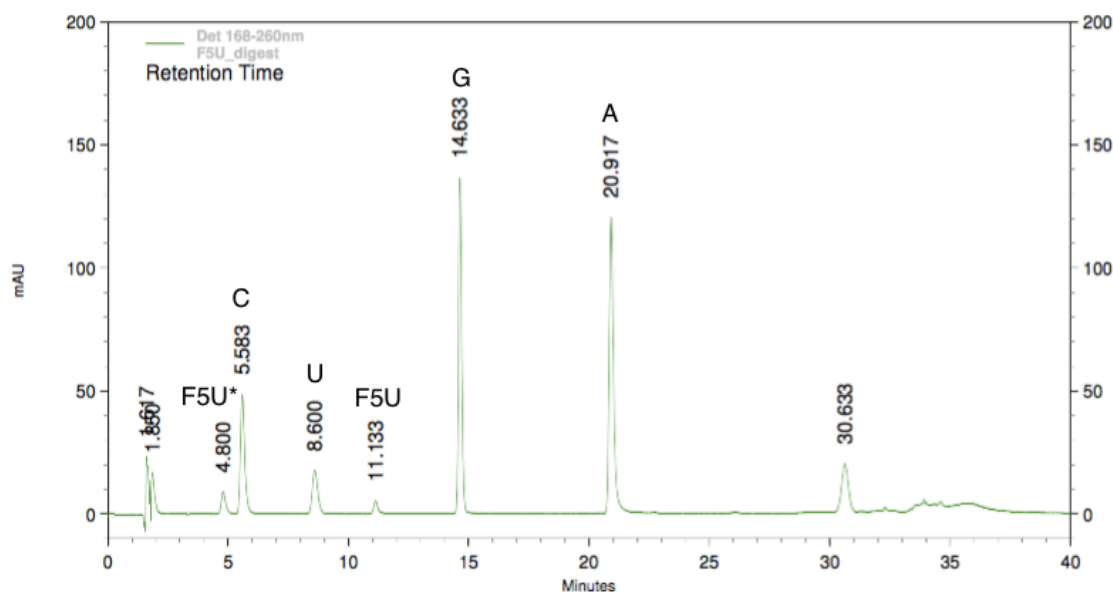


**Figure 2.4:** Time course for the conversion of ASL (50  $\mu$ M) upon incubation with Y96F RluA (5  $\mu$ M). HPLC traces showing the increase in  $\Psi$  with time. The major peaks are substrate ASL (with U) at 15 min and product (with  $\Psi$ ) at 14 min.

The Y96F RluA also modified [F<sup>5</sup>U]ASL and shifted it to a greater retention time, which is the behavior of wild-type RluA. Since the Y96F RluA reaction was conducted with excess substrate, unreacted [F<sup>5</sup>U]ASL remained. The peak integration indicated 75% of [F<sup>5</sup>U]ASL was converted to product after 4 h, which is 1.5 turnovers per Y96F RluA present. To ensure that the shift was due to formation of the products of [F<sup>5</sup>U]ASL (hereafter, the product of F<sup>5</sup>U = F<sup>5</sup>U\*), a complete reaction of [F<sup>5</sup>U]ASL after incubation with Y96F RluA was digested to free nucleosides and analyzed by reverse-phase HPLC. Before incubation of [F<sup>5</sup>U]ASL with Y96F RluA, peaks corresponding to C, U, F<sup>5</sup>U, G, and A were observed. After incubation, the C, U, G, and A peaks remained, but the F<sup>5</sup>U peak was reduced, and a new peak appeared at 4.8 min. The new of peak was observed at 4.8 min when [F<sup>5</sup>U]ASL incubated with Y96F RluA, indicating the formation of F<sup>5</sup>U\*.



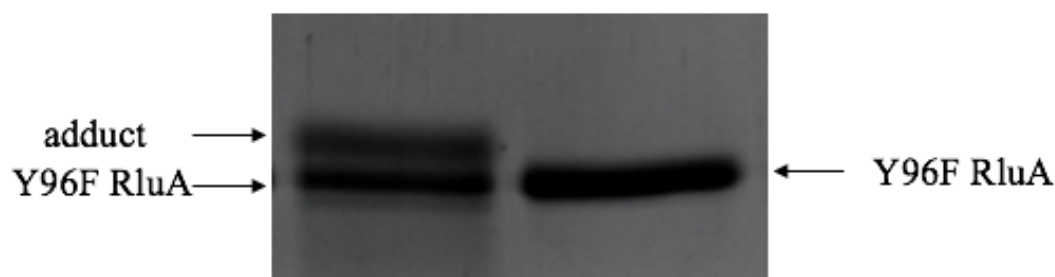
**Figure 2.5:** HPLC analysis of [F<sup>5</sup>U]ASL after incubation with Y96F RluA. The peak at 15.5 min is substrate and peak at 16.8 min is product.



**Figure 2.6:** Digestion of [F<sup>5</sup>U]ASL after incubated with Y96F RluA.

### 2.3.3 Adduct formation between Y96F RluA and [F<sup>5</sup>U]ASL

To probe adduct formation, [F<sup>5</sup>U]ASL (2  $\mu$ M) was incubated with Y96F RluA (1  $\mu$ M) for 3 h at 37 °C. After reaction, an aliquot (60  $\mu$ L) was diluted into Laemmli buffer (60  $\mu$ L) and split into two samples of equal volume; one of those was heated at 97 °C for 10 min, and the other was kept on ice. Aliquots of each sample were analyzed by SDS-PAGE (10% gel) to verify the presence of adduct. The gel analysis revealed a new, slower moving band, which indicated that the adduction of [F<sup>5</sup>U]ASL and Y96F RluA causes a gel shift to a higher apparent molecular weight due to retardation of the protein by the added mass of the RNA. When heated, the normal migration of Y96F RluA is observed, which means that heating disrupts the adduct with release of the RNA from the protein. SDS-PAGE characteristic showed ~40% adduction of Y96F RluA by the 2-fold excess of [F<sup>5</sup>U]ASL.



**Figure 2.7:** SDS-PAGE (10%) analysis of the formation of the adduct between Y96F RluA and [F<sup>5</sup>U]ASL. The gel shifted of Y96F RluA upon incubation with [F<sup>5</sup>U]ASL indicates adduct formation (lane 1). Heating disrupts the adduct (lane 2).

## 2.4 Discussion

Phannachet *et al.* have proposed that the active site Tyr acts as the general base for the isomerization reaction of U to  $\Psi$  [45]. If this result is common to another family of  $\Psi$  synthases, then the absence of a modified U product is expected when Y96F RluA is incubated with natural substrate (ASL). Instead, U is converted to  $\Psi$  and F<sup>5</sup>U to F<sup>5</sup>U\* when Y96F RluA is incubated with ASL and [F<sup>5</sup>U]ASL, respectively. The products from the action of Y96F RluA have identical elution behavior on reverse-phase HPLC as those from the action of wild-type RluA, indicating the formation of the same products. The appearance of  $\Psi$  upon incubation of Y96F RluA with ASL casts doubt on the conclusion of Phannachet *et al.* about the active site of Tyr serving as general base. At the least, this Tyr is not essential in the final step of the reaction, which is deprotonation at C5 to form the product of  $\Psi$ .

To determine whether Y96F RluA converts U to  $\Psi$  in a time frame similar to wild-type RluA, Y96F RluA (5  $\mu$ M) was incubated with an excess of ASL (50  $\mu$ M), and

the product was monitored over time by reverse-phase HPLC. The results showed that roughly 10% conversion of U to  $\Psi$ , which means only a single turnover occurred after 6 h incubation. The slow rate of reaction indicated longer incubation times are required for Y96F RluA to produce detectable  $\Psi$ , and thus the Tyr might not be essential for activity but does facilitate at least one step of the reaction.

Incubation of Y96F RluA with a two-fold excess of  $[F^5U]ASL$  results in the formation of an adduct as observed on denaturing SDS-PAGE. However, only ~40% of the enzyme formed an adduct. The HPLC analysis performed in parallel to the gel analysis revealed 75% of  $[F^5U]ASL$  is converted into its hydrated product, which is more than a single turnover. These results showed that Y96F RluA is not completely driven into the adduct as judged by SDS-PAGE. This behavior of Y96F RluA could arise for two possible reasons. First, Y96F RluA may adopt a specific conformation to bind RNA, and only a part of the protein exists in such a conformation at any given time. Second, an adduct may form between  $F^5U$  and Tyr-96 instead of the conserved Asp, just like the covalent bond between  $F^5U$  and Tyr-140 of RluB [47].

Based on the observations reported here, it seems that Tyr-96 plays a very important but not essential role in enzyme activity period. The slow rates of reactions and higher concentration of Y96F RluA needed for any reaction indicates that Tyr-96 may provide binding stability or facilitates glycosidic bond cleavage by donating a proton or a hydrogen bond to the uracil(ate). Since the anion of 5-fluorouracil is a better leaving group than the anion of uracil, the glycosidic bond of  $F^5U$  can undergo cleavage without the assistance provided by Tyr-96, thus explaining the slower reaction of ASL than  $[F^5U]ASL$  when incubated with Y96F RluA.

CHAPTER III  
MECHANISTIC INVESTIGATION OF RluA AND TruB  
USING RNA CONTAINING 4-THIOURIDINE

**3.1 Introduction**

Pseudouridine synthases are the enzymes that catalyze an isomerization reaction of specific U residues to  $\Psi$  in RNAs [1, 2].  $\Psi$  synthases are classified into six families that share minimal sequence similarity with each other. However, crystal structures of all six families reveal that they share a common fold with a core  $\beta$ -sheet along with several conserved active site amino acid residues and therefore they likely share a common mechanism [48].

Researchers have used RNA containing 5-fluorouridine as a mechanistic probe. Studies of the interaction of TruB and RluA with  $[F^5U]RNA$  revealed that TruB does not form a covalent adduct with  $[F^5U]RNA$  but RluA does [36]. However, both enzymes convert  $F^5U$  to rearranged and hydrated products. Two peaks in the HPLC analysis showed that TruB generates two products of  $F^5U$ , but RluA generated only one product peak from  $F^5U$ , which differed from either of the TruB product peaks [29, 36]. Mass spectrometric analysis and the observation of two resonances in the  $^{31}P$  NMR spectrum confirmed that both TruB products were isolated as dinucleotides containing the cytidine that follow  $F^5U$  in  $[F^5U]TSL$  [38]. The dinucleotide products can explain the difference

in HPLC retention times between the RluA and the TruB products peaks since uridine rather than cytidine follows F<sup>5</sup>U in [F<sup>5</sup>U]ASL. Subsequent NMR and MS experiments from the Mueller group revealed that RluA also converts F<sup>5</sup>U into two products. They are both dinucleotides with uridine and are very similar to the TruB products, confirming that both enzymes generate the same two isomeric products from F<sup>5</sup>U [38]. NMR experiments by Miracco and Mueller showed that both products of F<sup>5</sup>U from the action of TruB and RluA differ at C2' of the pentose ring rather than in the pyrimidine ring: the major product of F<sup>5</sup>U is the *ribo* isomer, and the minor product is the *arabino* isomer [39]. The change in configuration at C2' requires deprotonation to form a glycal intermediate, followed by reprotonation from the “top” and “bottom” face to afford the *ribo* and *arabino* products. The deprotonation at C2' can be achieved by the conserved Asp or O<sup>2</sup> of the pentose ring; however, since O<sup>2</sup> seem to be unable to access the “bottom” face of the glycal intermediate, the proton from O<sup>2</sup> then must migrate to the conserved Asp to reprotonate the glycal intermediate from the “bottom” face to generate the *arabino* product. Cocrystal structures show that the O<sup>2</sup> is located right above C2' of the glycal intermediate where it can possibly act as a base to deprotonate C2' [41].

In this chapter, the isomerized U is replaced by 4-thiouridine (s<sup>4</sup>U) in order to test if O<sup>2</sup> is a general base of the glycal mechanism. Some bacteria have s<sup>4</sup>U at position 8 in tRNA, and it serves as a sensor for near-UV light [49]. The lower pK<sub>a</sub> for s<sup>4</sup>U (8.2) relative to U (9.2) reflects the greater stability of the anion of 4-thiouracil than the anion of uracil; hence the leaving group is better for s<sup>4</sup>U than U, which increase the rates of reaction. However, the lower pK<sub>a</sub> for s<sup>4</sup>U also makes the 4-thiouracil(ate) a weaker base than uracil(ate), thus making it harder to deprotonate C2' and slowing the rate of reaction.



Because the overall rate of the reaction depends on leaving group departure and deprotonation of C2', a slower rate of reaction will indicate O<sup>2</sup> is a general base since leaving group departure should be faster with s<sup>4</sup>U. As an essential first step, RNA containing s<sup>4</sup>U needs to be verified as a good substrate for Ψ synthase. Therefore, assays of s<sup>4</sup>U with TruB and RluA were performed. Both intact and digested of [s<sup>4</sup>U]TSL and [s<sup>4</sup>U]ASL were analyzed by reverse-phase HPLC after incubation with TruB and RluA, respectively.

## 3.2 Experimental

### 3.2.1 Materials

Competent BLR(DE3) pLysS *E. coli* cells were purchased from Novagen (Madison, WI). A CLYPEUS C<sub>8</sub> 5-μm column (250 × 4.6 mm) was purchased from Higgins Analytical, Inc. (Mountain View, CA), and a Zorbax analytical SB-C<sub>18</sub> 5-μm column (50 × 4.6 mm) was purchased from Agilent (Santa Clara, CA). Ni-NTA superflow resin and QIAprep Spin Mini prep kits were purchased from Qiagen (Chatsworth, CA). S1 nuclease and calf intestinal alkaline phosphatase were purchased from Promega (Madison, WI). Oligonucleotide [s<sup>4</sup>U]ASL (5'-GGGGAs<sup>4</sup>UUGAAAAU-CCCC-3') and [s<sup>4</sup>U]TSL (5'-CUGUGUs<sup>4</sup>UCGAUCCACAG-3') were purchased from Thermofisher Dharmacon (Lafayette, CO). Ultra-pure deionized water was obtained using a Millipore<sup>®</sup> Milli-Q integral system (Billerica, MA) equipped with a 0.2 μm filter. Laemmli buffer (2×) was 120 mM Tris•HCl buffer, pH 6.8, containing SDS (4%, w/v), glycerol (20%, v/v), and bromophenol blue (0.02%, w/v). RNA loading dye (2×) was 95% aqueous formamide containing bromophenol blue (0.025%, w/v), xylene cyanol (0.025%, w/v), and EDTA (0.5 mM).

### 3.2.2 Overexpression and purification of wild-type RluA and TruB

**Overexpression and purification of wild-type RluA.** An overnight culture of BLR(DE3) pLysS/pΨ746 was grown and put through the protocol previously described for overexpression and purification of Y96F RluA (Chapter 2.2.2).

**Overexpression and purification of wild-type TruB.** Purified, RNase-free TruB was generously provided by A. Gibbs (University of Louisville).

### 3.2.3 Deprotection and Purification of 2'-ACE RNA Oligonucleotides

ASL containing 4-thiouridine ([<sup>4</sup>S]ASL) (5'-GGGGAs<sup>4</sup>UUGAAAAUCCCC-3') and TSL containing 4-thiouridine ([<sup>4</sup>S]TSL) (5'-CUGUGUs<sup>4</sup>UCGAUCCACAG-3') were purchased from Thermofisher Dharmacon (Lafayette, CO). The [<sup>4</sup>S]ASL and [<sup>4</sup>S]TSL were deprotected and purified as described above (Chapter 2.2.3). The final [<sup>4</sup>S]ASL and [<sup>4</sup>S]TSL concentration were determined using A<sub>260</sub> and extinction coefficients (170,500 M<sup>-1</sup>cm<sup>-1</sup> and 162,800 M<sup>-1</sup>cm<sup>-1</sup>, respectively) provided by Dharmacon.

### 3.2.4 Assay of RluA and ASL containing 4-thiouridine

**Reaction of [<sup>4</sup>S]ASL.** The reaction mixture (200 μL) was 50 mM HEPES buffer, pH 7.5, containing sodium chloride (175 mM), DTT (5 mM), EDTA (1mM), RNase inhibitor (30 units), and [<sup>4</sup>S]ASL (10 μM). After pre-equilibration at 37 °C, the reaction was initiated by the addition of RluA (to 10 μM). Aliquots (75 μL) were removed after 3 h and immediately added to quench solution (25 μL), which was 0.5 M sodium phosphate buffer, pH 7.5, containing sodium chloride (0.5M), then heated at 97 °C for 10 min. The quenched aliquots were passed over G-25 spin columns that had been pre-equilibrated with 125 mM sodium phosphate buffer, pH 7.5, containing sodium chloride

(125 mM), then passed through a 0.2  $\mu\text{m}$  filter before being analyzed by reverse-phase HPLC as described for intact ASL (Chapter 2.2.4).

***Digestion of [ $s^4\text{U}$ ]ASL.*** To confirm the formation of product, [ $s^4\text{U}$ ]ASL was digested after incubation with RluA. Aliquots (100  $\mu\text{L}$ ) were removed after 3 h and immediately heated at 97  $^\circ\text{C}$  for 10 min. The aliquot was spun for 10 min at 14,000g to pellet denatured protein, and the supernatant was transferred to a fresh tube and diluted into 50 mM sodium acetate buffer (100  $\mu\text{L}$ ), pH 4.5, containing sodium chloride (280 mM) and zinc chloride (4.5 mM). S1 nuclease (100 units) was added; after 1 h at 37  $^\circ\text{C}$ , the digestion mixture was incubated at 100  $^\circ\text{C}$  for 5 min and then cooled on ice. Additional S1 nuclease (100 units) and alkaline phosphatase (5 units) were added, and the digestion mixture was incubated for an additional 3 h at 37  $^\circ\text{C}$ , then passed through a 0.2  $\mu\text{m}$  filter and analyzed by reverse phase HPLC as described for intact ASL (Chapter 2.2.4).

### 3.2.5 Assay of TruB and TSL containing 4-thiouridine

***Reaction of [ $s^4\text{U}$ ]TSL.*** The reaction mixture (200  $\mu\text{L}$ ) was 50 mM HEPES buffer, pH 7.5, containing ammonium chloride (100 mM) and EDTA (0.1 mM), RNase inhibitor (30 units), and [ $s^4\text{U}$ ]TSL (10  $\mu\text{M}$ ). After pre-equilibration at 37  $^\circ\text{C}$ , the reaction was initiated by the addition of TruB (to 10  $\mu\text{M}$ ) and incubated for 3 h. The analysis followed the protocol as described for the reaction of ASL (Chapter 2.2.4).

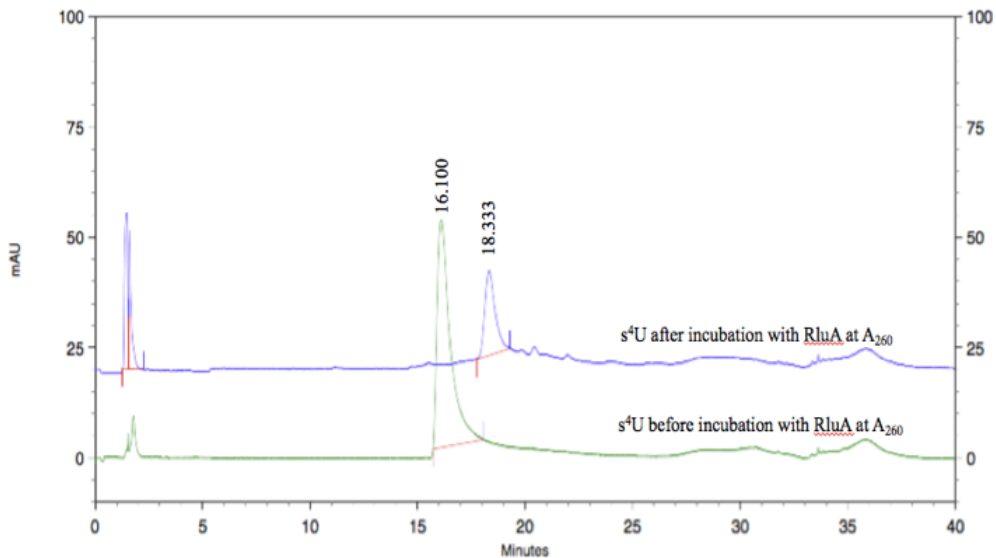
***Digestion of [ $s^4\text{U}$ ]TSL.*** To confirm the formation of product, the digestion of [ $s^4\text{U}$ ]TSL after incubation with TruB was accomplished by the same method described previously for the digestion ASL (Chapter 2.2.4).

### 3.3 Results

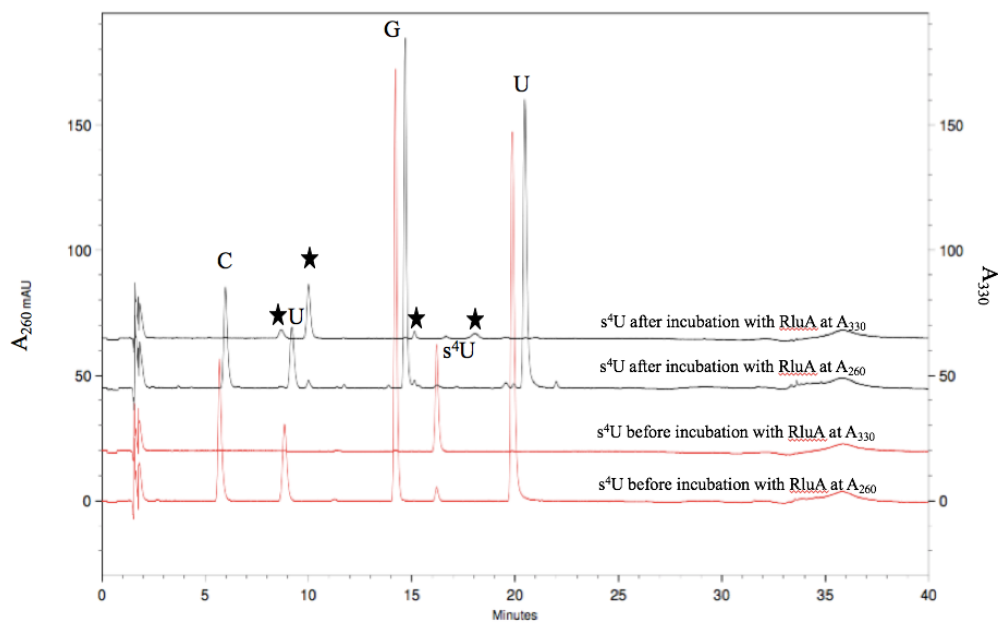
#### 3.3.1 Assay of RluA and ASL containing 4-thiouridine

In order to determine if [ $s^4U$ ]ASL is modified by RluA, [ $s^4U$ ]ASL (10  $\mu$ M) was incubated with RluA (10  $\mu$ M) at 37 °C for 3 h and analyzed by reverse-phase HPLC. The HPLC analysis revealed a shift to later retention time after incubation with RluA (Figure 3.1).

To ensure that the difference in chromatographic behavior of [ $s^4U$ ]ASL after incubation with RluA arose from alteration of the  $s^4U$ , [ $s^4U$ ]ASL was digested to free nucleosides before and after incubation with RluA and the products were analyzed by reverse-phase HPLC. With detection at 260 nm, C, U, G,  $s^4U$ , A were observed with the  $s^4U$  peak greatly diminished, and a new peak (10 min) appeared after reaction. With detection at 330 nm,  $s^4U$  (16.5 min) appeared before incubation with RluA, but the  $s^4U$  was essentially replaced by four sizable peaks, with the largest product peak (10 min) at the same retention time as the new peak detected at 260 nm. These results show that RluA acts on [ $s^4U$ ]ASL, but the identity of the products needs to be investigated further.



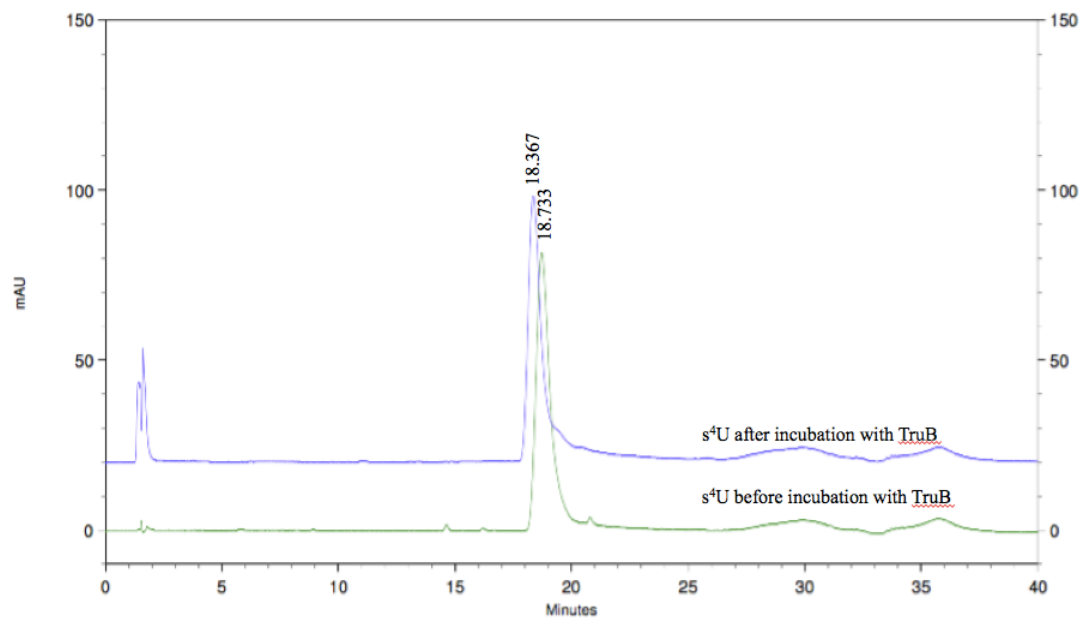
**Figure 3.1:** HPLC analysis of  $[s^4U]ASL$  before (green) and after (blue) incubation with RluA. The peak at 16.1 min and 18.3 min are  $[s^4U]ASL$  before and after incubation with WT RluA, respectively.



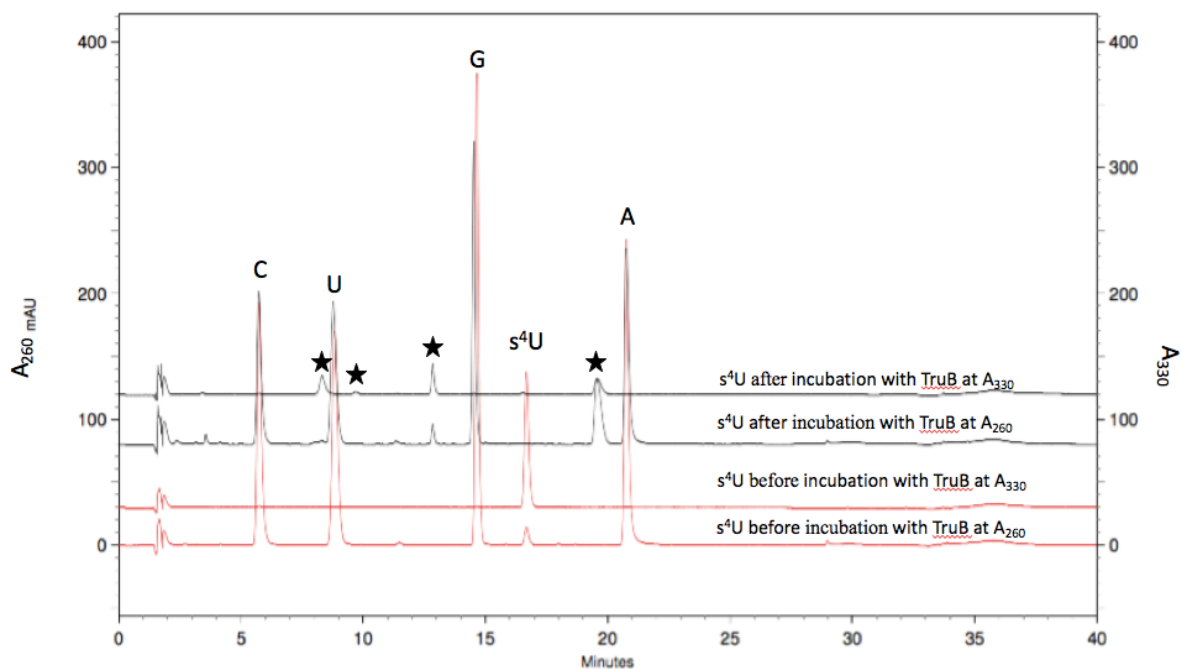
**Figure 3.2:** HPLC analysis of the nucleosides resulting from digestion of  $[s^4U]ASL$  before (red) and after (black) incubation with RluA. The stars indicates the location of the new peaks.

### 3.3.2 Assay of TruB and TSL containing 4-thiouridine

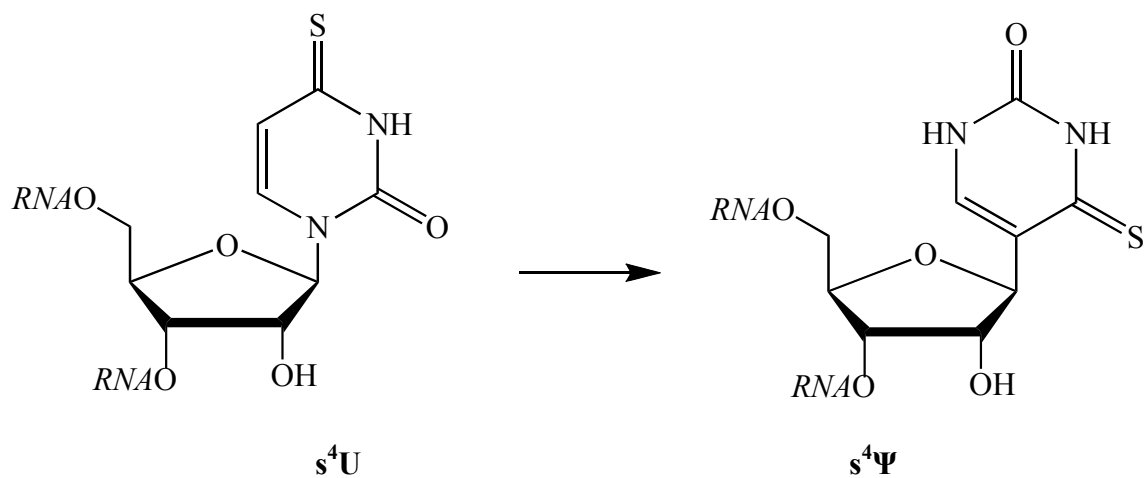
To investigate further whether results with RluA and [<sup>3</sup>S]ASL are common to another family of Ψ synthases, the same set of reactions were performed with *E. coli* TruB and [<sup>3</sup>S]TSL. HPLC analysis showed that incubation with TruB slightly shifts [<sup>3</sup>S]TSL to a shorter retention time (18.3 min) compared to unreacted [<sup>3</sup>S]TSL (18.7 min) (Figure 3.4). To make sure that the change in [<sup>3</sup>S]TSL was due to reaction of <sup>3</sup>S, [<sup>3</sup>S]TSL both before and after incubation with TruB was digested to free nucleosides, which were characterized by HPLC. With detection at 260 nm, C, U, G, <sup>3</sup>S, A were observed after reaction, but the <sup>3</sup>S peak was replaced by two new peaks (13 and 19.5 min). With detection at 330 nm, <sup>3</sup>S (16.5 min) appeared before incubation with TruB but was essentially replaced by four sizable peaks (8, 10, 13, and 19.5 min) after incubation. These results show that TruB acts on [<sup>3</sup>S]TSL, but the identity of the products needs to be further investigated.



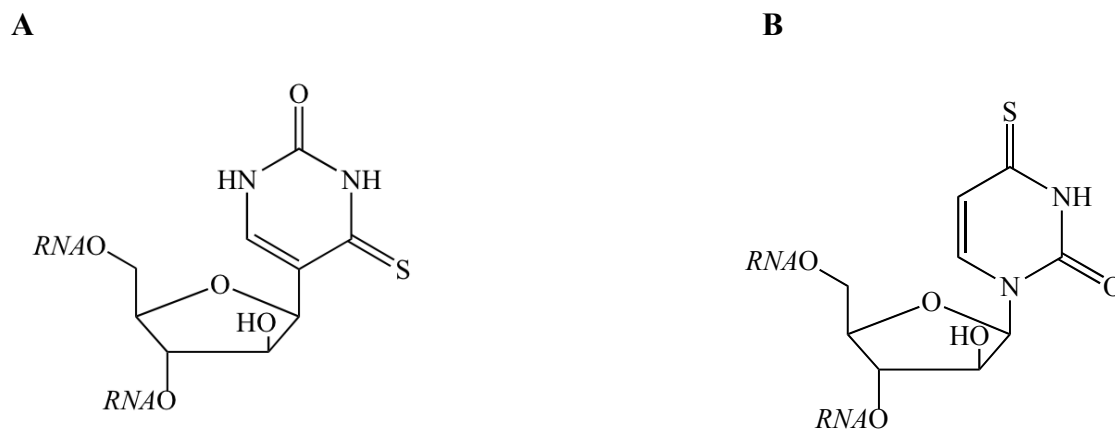
**Figure 3.3:** HPLC analysis of  $[s^4U]TSL$  before (green) and after (blue) incubation with TruB. The peak at 18.7 min is  $[s^4U]TSL$  before incubation with TruB, and the peak at 18.3 min is  $[s^4U]TSL$  after incubation with TruB.



**Figure 3.4:** HPLC analysis of the nucleosides resulting from digestion of  $[s^4U]TSL$  before (red) and after (black) incubation with TruB. The stars indicates the location of the new peaks.

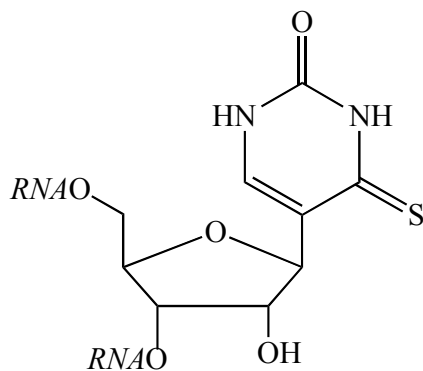
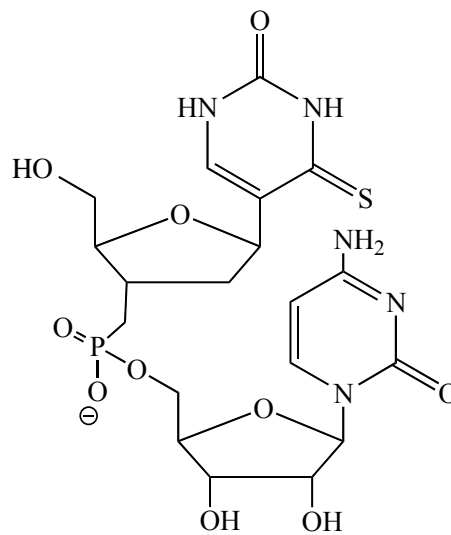


**Figure 3.5:** The isomerization of  $s^4U$  to  $s^4\Psi$ .

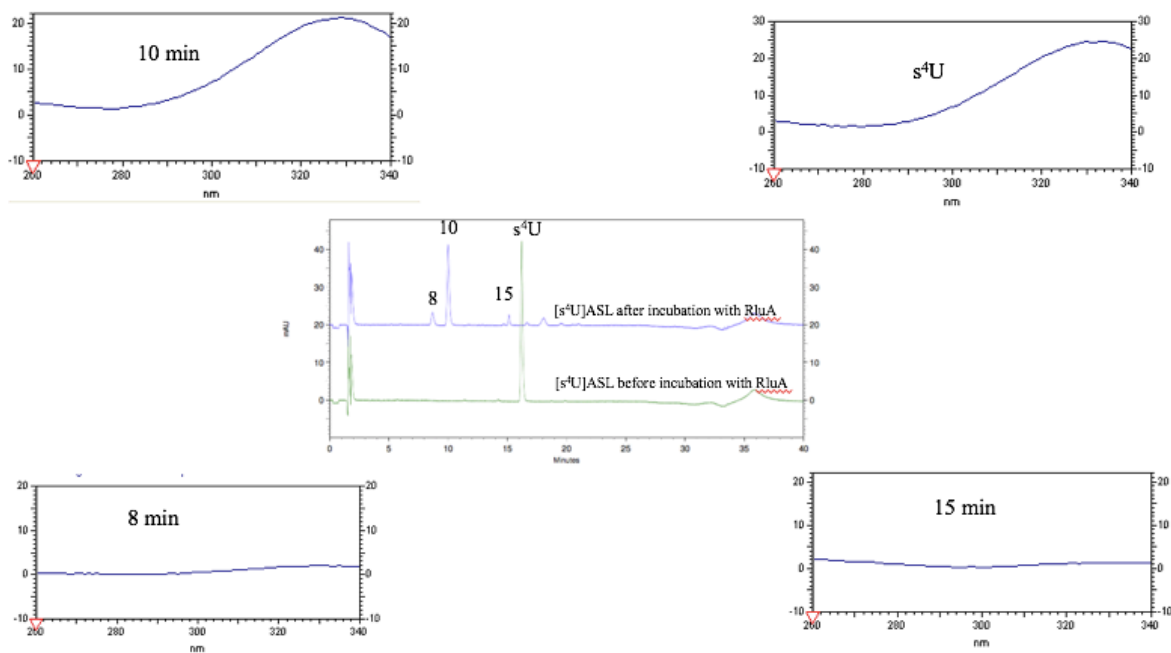


**Figure 3.6:** A, the *arabino*- $s^4\Psi$  product. B, the *arabino*- $s^4U$  product.

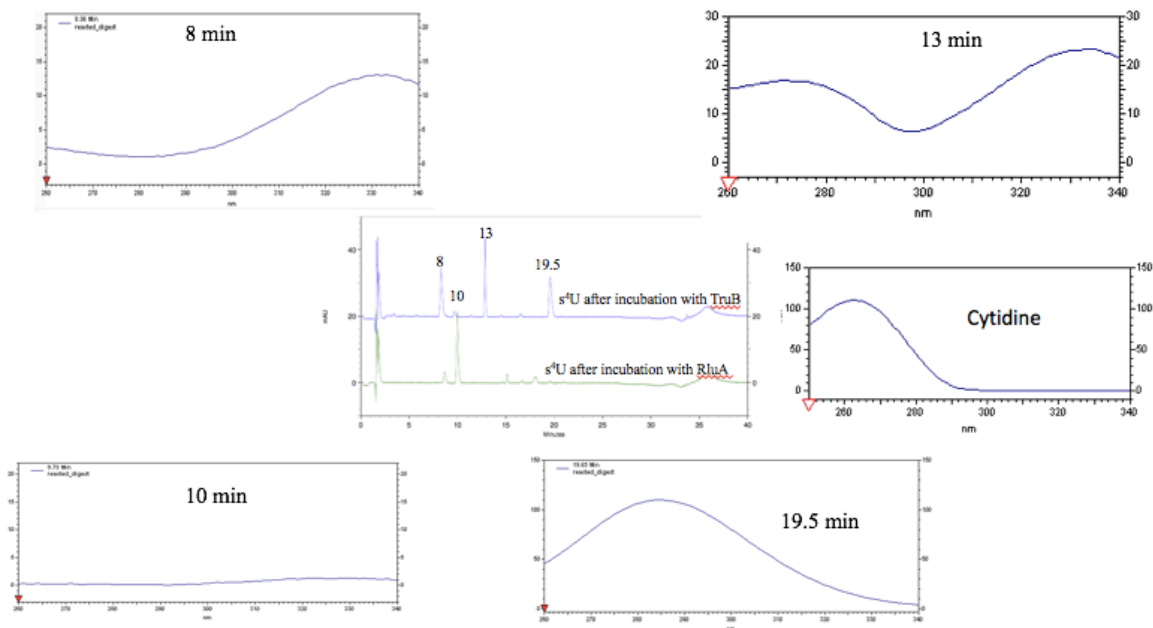


**A****RluA product****B****TruB product**

**Figure 3.7: A, The *ribo*-  $s^4\Psi$  product from the action of RluA. B, Dinucleotide product from the action of TruB.**



**Figure 3.8: The spectra of new peaks from the action of RluA.**



**Figure 3.9:** The spectra of new peaks from the action of TruB.

### 3.4 Discussion

Studies of the interaction of TruB and RluA with  $[F^5U]RNA$  revealed that both enzymes convert  $F^5U$  into two products, and the difference between them is at C2' of the pentose ring: the major product is the *ribo* isomer whereas the minor product is the *arabino* isomer [39]. To explain the *arabino* product of  $F^5U$ , C2' must be deprotonated followed by reprotonation from the opposite face. The deprotonation at C2' can be achieved either by the conserved Asp or  $O^2$  of uracil.  $O^2$  is positioned directly above C2', which is a good position to act as base to generate the glycal intermediate [41]. However,  $O^2$  seems to be unable to access the “bottom face” of the glycal intermediate, so the proton must migrate from  $O^2$  to the conserved Asp in order to reprotonate from the distal face. To determine whether  $O^2$  is the general base, the reaction of RNA containing  $s^4U$

([ $s^4$ U]RNA) in place of U was used. To ensure that [ $s^4$ U]RNA can act as a substrate, assays of TruB and RluA with [ $s^4$ U]RNA were performed and analyzed by reverse-phase HPLC. Intact [ $s^4$ U]RNA shifted to later and shorter retention times after incubation with RluA and TruB, respectively, and peak integration indicated that all of the [ $s^4$ U]RNA reacted. Traces of the digestion products of [ $s^4$ U]RNA after incubation with the two enzymes showed the new peaks that absorbed more strongly at 330 nm than 260 nm. These results indicate that both RluA and TruB act on [ $s^4$ U]RNA, and thus [ $s^4$ U]RNA can be handled as a substrate. The product is expected to be 4-thiopseudouridine ( $s^4\Psi$ ; Figure 3.5).

The action of RluA replaced the  $s^4$ U peak with four sizable peaks (8, 10, 15, and 18 min). Because  $s^4\Psi$  is more polar than  $s^4$ U,  $s^4\Psi$  should elute earlier in reverse phase HPLC, which agrees with the tentative assignment of the largest product peak (10 min; Figure 3.2) as  $s^4\Psi$  (Figure 3.7A). That peak (10 min) also absorbs more strongly at 330 nm than 260 nm, which is consistent with the assignment (Figure 3.8). The identities of the minor products (8 and 15 min; Figure 3.2) are less sure, but based on the precedent of [ $F^5$ U]RNA [26], one may be *arabino*- $s^4\Psi$  (Figure 3.6A) and the other *arabino*- $s^4$ U (Figure 3.6B). The latter is rationalized because sulfur is larger than oxygen, so the rotation of 4-thiouracil(ate) in the active site may be harder than for uracil(ate), leading N1 to attack C1' of the glycol intermediate with reprotonation from “bottom” face.

Similar to the results with RluA, the  $s^4$ U peak was replaced with four sizable peaks (8, 10, 13, and 19.5 min) after incubation with TruB. The peak at 19.5 min has an absorbance maximum at 285 nm, so it does not contain a 4-thiouracil chromophore (Figure 3.9). Among the other three peaks 8, 10, 13 min with ratio of 9:1:10, respectively;

the spectrum of the peak at 13 min shows two peaks with absorbance maxima at 270 nm and 330 nm (Figure 3.9), which are characteristic of cytidine and  $s^4U$  and therefore suggests that the  $s^4U$  product at 13 min formed by the action of TruB is a dinucleotide of an  $s^4U$  product with the cytidine residue that follows  $s^4U$  in  $[s^4U]TSL$  (Figure 3.7B). The peaks at 8 and 10 min seem to elute at the same retention time as peaks at 8 and 10 min on the action of RluA. Also, the spectra of the two peaks show they are not dinucleotides (Figure 3.9), thus the peak at 8 may be *arabino*- $s^4\Psi$  or *arabino*- $s^4U$ , and the peak at 10 min is tentatively assigned as  $s^4\Psi$ . To confirm the different products, liquid chromatography-mass spectroscopy can be used since the dinucleotide product have higher molecular mass than mononucleotide.  $^1H$ -NMR can distinguish between *arabino*- $s^4\Psi$  or *arabino*- $s^4U$  and  $s^4\Psi$  product.

Once the products are identified, the full kinetic characterization of RluA and TruB with  $[s^4U]RNA$  substrate will be performed in order to compare the difference in rates of the reaction catalyzed by RluA and TruB with natural substrate versus  $[s^4U]RNA$ . A slower rate with  $[s^4U]RNA$  will imply  $O^2$  is the general base, but a faster rate will not.

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## CURRICULUM VITAE

Uyen Duong

### EDUCATION

#### **M.S. in Chemistry, Anticipated Graduation Date: August, 2017**

Thesis: Mechanistic investigations of Pseudouridine synthases TruB and RluA with RNA containing 5-Fluorouridine and 4-Thiouridine  
University of Louisville (2015-Present)

Louisville, KY.

#### **M.S. in Polymer Science & Engineering**

Thesis: Synthesis and SAR of thiazolidinedione as a novel class of algicides against harmful algal species

University of Chosun (2008-2010)

Gwangju, Korea

B.S. in Chemistry

Vietnam National University (2004-2008)

Ha Noi, Viet Nam

### WORK EXPERIENCE

#### **Research assistant/Teaching assistant**

Department of Chemistry, University of Louisville (Jan, 2015-present)

- Expertise in the fields of enzyme assay, enzyme kinetics, protein overexpression, purification and characterization using conventional biochemistry methods
- Two years of experience in handling instruments such as IR, NMR, HPLC, FPLC, UV-Vis. Efficient in troubles shooting HPLC, FPLC instrument problems
- Two years teaching experience as teaching assistant for Chem343 (Organic 1), Chem344 (Organic 2)

**Research assistant**

Department of Polymer Science & Engineering, Chosun University (Sep. 2008 - Jul. 2008)

- Two years of experience in organic synthesis, and purification of organic compounds
- Ability to work in various chromatographic techniques such as thin-layer chromatography, gas-chromatography, ion exchange, affinity

**Research assistant**

Organic Synthesis Laboratory, Department of Chemistry, Vietnam National University (Jul. 2006 – Jun. 2008)

**PUBLICATIONS**

- 4) Y.-M. Kim, Y. Wu, **U. T. Duong**, S.-G. Jung, S. W. Kim, H. Cho, E. S. Jin. Algicidal Activity of Thiazolidinedione Derivatives Against Harmful Algal Blooming Species. *Mar Biotechnol* **2012**, 14 (3), 312-322
- 3) Y.-M. Kim, Y. Wu, **U. T. Duong**, G. S. Ghodake, S. W. Kim, E. S. Jin, H. Cho. Thiazolidinediones as a Novel Class of Algicides Against Red Tide Harmful Algal Species. *Appl Biochem Biotechnol* **2010**, 162, 2273–2283
- 2) Y. Wu, **U. T. Duong**, H. Cho. Synthesis and Evaluation of Novel 15-Hydroxyprostaglandin Dehydrogenase Inhibitors. Proceedings of Spring International Convention. May 2009, Daejeon, Korea
- 1) Nguyen Dinh Trieu, **Duong Thi Uyen**, Hoang Thi Ly, Pham Thi Ngoc Ha, Le Thi Minh Hang. Synthesis and mass spectroscopy of some derivatives of 1-Arylterazoles. The 5th national Conference of Science and Organic Chemical Technology, 257-262, **2007**, Hanoi, Vietnam