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9-Aminoacridin-10-ium 4-aminobenzoate dihydrate

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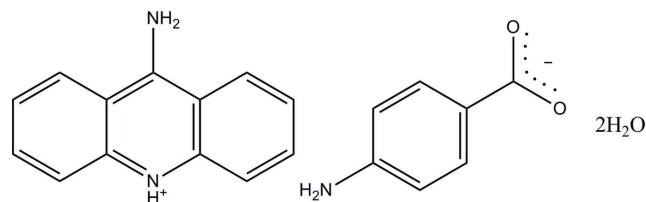
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.052; wR factor = 0.128; data-to-parameter ratio = 18.1.

The asymmetric unit of the title hydrated salt, $\text{C}_{13}\text{H}_{11}\text{N}_2^+ \cdot \text{C}_7\text{H}_6\text{NO}_2^- \cdot 2\text{H}_2\text{O}$, consists of two independent 9-aminoacridinium cations, two 4-aminobenzoate anions and four water molecules. Both 9-aminoacridinium cations are essentially planar, with maximum deviations of 0.034 (1) and 0.025 (2) Å, and are protonated at the pyridine N atoms. The 4-aminobenzoate anions are approximately planar, with dihedral angles of 9.16 (19) and 5.4 (2)° between the benzene ring and the carboxylate group. In the crystal, the two independent anions are connected by $\text{N}-\text{H} \cdots \text{O}$ hydrogen bonds, forming a layer parallel to (100). The layers are connected through the cations by $\text{N}-\text{H} \cdots \text{N}$ and $\text{N}-\text{H} \cdots \text{O}$ hydrogen bonds. The water molecules, which form $\text{O}-\text{H} \cdots \text{O}$ hydrogen-bonded chains along the b -axis direction, connect the anions and the cations by $\text{O}-\text{H} \cdots \text{O}$, $\text{N}-\text{H} \cdots \text{O}$ and $\text{C}-\text{H} \cdots \text{O}$ hydrogen bonds. The crystal structure also features $\pi-\pi$ interactions [centroid-centroid distances = 3.6343 (9)–3.8366 (10) Å] and a $\text{C}-\text{H} \cdots \pi$ interaction.

Related literature

For background to and the biological activity of acridine derivatives, see: Shubber *et al.* (1986); Sondhi *et al.* (2006); Salamanca & Khalil (2005). For related structures, see: Aghabozorg *et al.* (2010); Mei & Wolf (2004). For bond-length data, see: Allen *et al.* (1987). For stability of the temperature controller used for data collection, see: Cosier & Glazer (1986).



Experimental

Crystal data

 $\text{C}_{13}\text{H}_{11}\text{N}_2^+ \cdot \text{C}_7\text{H}_6\text{NO}_2^- \cdot 2\text{H}_2\text{O}$
 $M_r = 367.40$
Monoclinic, $P2_1/c$
 $a = 25.6891$ (9) Å
 $b = 7.2800$ (2) Å
 $c = 21.6485$ (6) Å
 $\beta = 114.865$ (1)° $V = 3673.32$ (19) Å³
 $Z = 8$
Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 100$ K
 $0.64 \times 0.14 \times 0.11$ mm

Data collection

Bruker SMART APEXII CCD
area-detector diffractometer
Absorption correction: multi-scan
(*SADABS*; Bruker, 2009)
 $T_{\min} = 0.942$, $T_{\max} = 0.990$ 39382 measured reflections
10114 independent reflections
7214 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.047$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.052$
 $wR(F^2) = 0.128$
 $S = 1.04$
10114 reflections
559 parameters
1 restraintH atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.53$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.25$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg13 is the centroid of the C1A–C6A ring.

$D-\text{H} \cdots A$	$D-\text{H}$	$\text{H} \cdots A$	$D \cdots A$	$D-\text{H} \cdots A$
O1W–H1W1···O3W	0.84 (3)	1.92 (3)	2.758 (2)	179 (5)
O1W–H2W1···O1B	0.86 (3)	1.90 (3)	2.767 (2)	177 (2)
O2W–H1W2···O2A ⁱ	0.88 (3)	1.92 (3)	2.8066 (19)	176 (3)
O3W–H2W3···O1W ⁱ	0.96 (3)	1.72 (3)	2.676 (2)	174 (2)
O3W–H1W3···O2B ⁱⁱ	0.90 (3)	1.79 (3)	2.674 (2)	166 (3)
O4W–H2W4···O1A	0.87 (3)	1.84 (3)	2.688 (2)	164 (3)
N1A–H1NA···O2B ⁱ	0.92 (2)	2.09 (2)	2.946 (2)	156.0 (19)
N1A–H2NA···O1B ⁱⁱⁱ	0.92 (3)	2.12 (3)	2.983 (2)	158 (2)
N2A–H3NA···O2A ⁱ	0.90 (2)	1.98 (2)	2.882 (2)	173 (2)
N2B–H3NB···O1B ^{iv}	0.98 (2)	1.93 (2)	2.8695 (18)	160 (2)
N3A–H4NA···O4W ^v	0.93 (2)	1.93 (2)	2.8227 (19)	161.6 (18)
N3A–H5NA···N1B ^{vi}	0.91 (2)	2.20 (2)	3.031 (2)	151.4 (18)
N3B–H5NB···O3W	0.90 (2)	1.94 (2)	2.819 (2)	165 (2)
N3B–H4NB···N1A ⁱ	1.00 (3)	2.16 (2)	3.067 (2)	151.3 (19)
C12A–H12A···O4W ^v	0.95	2.55	3.4756 (19)	164
C16B–H16B···O3W	0.95	2.51	3.438 (2)	164
C11A–H11A···Cg13 ^{iv}	0.95	2.95	3.6659 (19)	133

Symmetry codes: (i) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x, y - 1, z$; (iii) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (iv) $-x + 1, -y + 1, -z + 1$; (v) $x - 1, -y + \frac{1}{2}, z + \frac{1}{2}$; (vi) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supporting information for this paper is available from the IUCr electronic archives (Reference: IS5353).

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9-Aminoacridin-10-ium 4-aminobenzoate dihydrate

Nallathambi Dhanabalan, Kaliyaperumal Thanigaimani, Suhana Arshad, Ibrahim Abdul Razak and K. Joseph Santhanaraj

S1. Comment

9-Aminoacridine (9-AA) and its derivatives present a very interesting object of research. This group of compounds exhibits a wide spectrum of biological activities: antibacterial, mutagenic, antitumor and antiinflammatory (Shubber *et al.*, 1986; Sondhi *et al.*, 2006; Salamanca & Khalil, 2005) In the viewpoint of crystal engineering, acridine and its 9-amino derivative are very interesting because of their capability for hydrogen bonding *via* N atom of the ring and π - π stacking since they possess three rings (Aghabozorg *et al.*, 2010; Mei & Wolf, 2004). In order to study potential hydrogen bonding interactions the crystal structure determination of the title compound (I) was carried out.

The asymmetric unit of the title salt contains two 9-aminoacridinium cations (*A* and *B*), two 4-aminobenzoate anions (*A* and *B*) and four water molecules (Fig. 1). Each 9-aminoacridinium cation is essentially planar, with a maximum deviation of 0.034 (1) Å for atom C9A (molecule *A*) and 0.025 (2) Å for atom C11B (molecule *B*). The protonation of atoms N2A and N2B lead to a slight increase in C8A—N2A—C20A [122.22 (14)°] and C8B—N2B—C20B [121.86 (14)°] angles. The bond lengths (Allen *et al.*, 1987) and angles are normal.

In the crystal packing (Fig. 2), the cations and anions are linked by N1A—H1NA \cdots O2Bⁱ, N1A—H2NA \cdots O1Bⁱⁱ, N3A—H5NA \cdots N1B^{iv}, N3B—H4NB \cdots N1Aⁱ, N2A—H3NA \cdots O2Aⁱ and N2B—H3NB \cdots O1B^v hydrogen bonds (symmetry code in Table 1). In addition, the water molecules are connected *via* O1W—H1W1 \cdots O3W and O3W—H2W3 \cdots O1Wⁱ hydrogen bonds to form a one-dimensional supramolecular chain along the *b*-axis. Furthermore, the chains formed by water molecules and the ions [cations (*A* & *B*) and anions (*A* & *B*)] are connected *via* O1W—H2W1 \cdots O1B, O4W—H2W4 \cdots O1A, N3B—H5NB \cdots O3W, N3A—H4NA \cdots O4Wⁱⁱⁱ, O2W—H1W2 \cdots O2Aⁱ, O3W—H1W3 \cdots O2B^{vi}, C16A—H16B \cdots O3W and C12A—H12A \cdots O4Wⁱⁱⁱ hydrogen bonds (symmetry code in Table 1), forming a three-dimensional supramolecular network. Further, the cations (*A* & *B*) can establish several π - π interactions, with centroid \cdots centroid distances in the range from 3.6343 (9) to 3.8366 (10) Å (Cg1—Cg3 = 3.6343 (9) Å, Cg1=N2A/C8A/C20A/C13A—C15A and Cg3 = C15A—C20A; Cg1—Cg1 = 3.7520 (9) Å; Cg2—Cg3 = 3.7822 (10) Å, Cg2 = C8A—C13A; Cg7—Cg7 = 3.7441 (9) Å, Cg7 = N2B/C8B/C20B/C13B—C15B; Cg7—Cg9 = 3.7434 (10) Å, Cg9 = C15B—C20B; Cg8—Cg9 = 3.8193 (10) Å, Cg8 = C8B—C13B; Cg7—Cg8 = 3.8366 (10) Å) and C—H \cdots π interactions (Table 1) involving the C1A—C6A (centroid Cg13) ring.

S2. Experimental

Hot methanol solutions (20 ml) of 9-aminoacridine (48 mg, Aldrich) and 4-aminobenzoic acid (34 mg, Loba) was warmed for a half an hour over a water bath. The resulting solution was allowed to cool slowly at room temperature and crystals of the title compound (I) appeared after a few days.

9-Aminoacridin-10-ium 4-aminobenzoate dihydrate

Crystal data

 $C_{13}H_{11}N_2^+ \cdot C_7H_6NO_2^- \cdot 2H_2O$ $M_r = 367.40$ Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

 $a = 25.6891$ (9) Å $b = 7.2800$ (2) Å $c = 21.6485$ (6) Å $\beta = 114.865$ (1)° $V = 3673.32$ (19) Å³ $Z = 8$ $F(000) = 1552$ $D_x = 1.329$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 9979 reflections

 $\theta = 2.6$ – 30.0 ° $\mu = 0.09$ mm⁻¹ $T = 100$ K

Needle, yellow

 $0.64 \times 0.14 \times 0.11$ mm

Data collection

Bruker SMART APEXII CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 φ and ω scansAbsorption correction: multi-scan
(SADABS; Bruker, 2009) $T_{\min} = 0.942$, $T_{\max} = 0.990$

39382 measured reflections

10114 independent reflections

7214 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.047$ $\theta_{\max} = 29.5$ °, $\theta_{\min} = 2.1$ ° $h = -34 \rightarrow 35$ $k = -10 \rightarrow 10$ $l = -29 \rightarrow 28$

Refinement

Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.052$ $wR(F^2) = 0.128$ $S = 1.04$

10114 reflections

559 parameters

1 restraint

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sitesH atoms treated by a mixture of independent
and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0477P)^2 + 1.9401P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.53$ e Å⁻³ $\Delta\rho_{\min} = -0.25$ e Å⁻³

Special details

Experimental. The crystal was placed in the cold stream of an Oxford Cryosystems Cobra open-flow nitrogen cryostat (Cosier & Glazer, 1986) operating at 100.0 (1) K.**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.**Refinement.** Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1A	0.86360 (5)	0.58428 (18)	0.11715 (7)	0.0268 (3)
O2A	0.87227 (5)	0.88589 (17)	0.10751 (6)	0.0222 (3)

N1A	0.67465 (6)	0.8427 (2)	0.21685 (8)	0.0242 (3)
N2A	0.05332 (5)	0.2784 (2)	0.45535 (7)	0.0180 (3)
N3A	-0.04990 (6)	0.2128 (2)	0.55740 (7)	0.0199 (3)
C1A	0.72048 (6)	0.8241 (2)	0.19910 (8)	0.0199 (3)
C2A	0.74005 (7)	0.9733 (3)	0.17409 (9)	0.0235 (4)
H2AA	0.7243	1.0919	0.1732	0.028*
C3A	0.78230 (7)	0.9496 (3)	0.15049 (9)	0.0222 (4)
H3AA	0.7954	1.0529	0.1342	0.027*
C4A	0.80575 (6)	0.7776 (2)	0.15040 (8)	0.0182 (3)
C5A	0.78674 (6)	0.6297 (2)	0.17669 (8)	0.0193 (3)
H5AA	0.8025	0.5112	0.1776	0.023*
C6A	0.74549 (6)	0.6517 (2)	0.20143 (8)	0.0195 (3)
H6AA	0.7341	0.5495	0.2201	0.023*
C7A	0.85012 (6)	0.7471 (2)	0.12315 (8)	0.0193 (3)
C8A	0.07687 (6)	0.2119 (2)	0.52023 (8)	0.0169 (3)
C9A	0.13552 (7)	0.1634 (2)	0.55063 (9)	0.0214 (3)
H9AA	0.1579	0.1734	0.5253	0.026*
C10A	0.16021 (7)	0.1021 (3)	0.61654 (9)	0.0239 (4)
H10A	0.1998	0.0717	0.6368	0.029*
C11A	0.12751 (7)	0.0835 (3)	0.65452 (9)	0.0226 (4)
H11A	0.1452	0.0425	0.7004	0.027*
C12A	0.07030 (7)	0.1245 (2)	0.62534 (8)	0.0193 (3)
H12A	0.0483	0.1091	0.6509	0.023*
C13A	0.04320 (6)	0.1901 (2)	0.55725 (8)	0.0162 (3)
C14A	-0.01692 (6)	0.2343 (2)	0.52463 (8)	0.0166 (3)
C15A	-0.04016 (6)	0.3026 (2)	0.45590 (8)	0.0167 (3)
C16A	-0.09877 (6)	0.3509 (2)	0.41950 (8)	0.0195 (3)
H16A	-0.1242	0.3381	0.4409	0.023*
C17A	-0.11937 (7)	0.4155 (2)	0.35427 (8)	0.0217 (3)
H17A	-0.1587	0.4480	0.3308	0.026*
C18A	-0.08208 (7)	0.4338 (3)	0.32192 (8)	0.0229 (4)
H18A	-0.0966	0.4777	0.2764	0.028*
C19A	-0.02525 (7)	0.3891 (2)	0.35517 (8)	0.0208 (3)
H19A	-0.0005	0.4028	0.3329	0.025*
C20A	-0.00344 (6)	0.3226 (2)	0.42261 (8)	0.0171 (3)
O1B	0.37845 (5)	0.68571 (17)	0.26050 (6)	0.0230 (3)
O2B	0.36397 (5)	0.98772 (18)	0.25207 (6)	0.0261 (3)
N1B	0.17296 (6)	0.7568 (2)	-0.03882 (8)	0.0229 (3)
N2B	0.54393 (6)	0.2571 (2)	0.59970 (7)	0.0202 (3)
N3B	0.44631 (6)	0.2602 (2)	0.39185 (7)	0.0220 (3)
C1B	0.21918 (6)	0.7727 (2)	0.02383 (8)	0.0194 (3)
C2B	0.23209 (7)	0.9435 (3)	0.05693 (9)	0.0245 (4)
H2BA	0.2112	1.0494	0.0343	0.029*
C3B	0.27492 (7)	0.9592 (3)	0.12218 (9)	0.0239 (4)
H3BA	0.2830	1.0761	0.1437	0.029*
C4B	0.30656 (6)	0.8068 (2)	0.15708 (8)	0.0190 (3)
C5B	0.29449 (6)	0.6376 (2)	0.12345 (8)	0.0190 (3)
H5BA	0.3158	0.5321	0.1460	0.023*

C6B	0.25205 (7)	0.6210 (2)	0.05779 (9)	0.0203 (3)
H6BA	0.2452	0.5051	0.0356	0.024*
C7B	0.35265 (6)	0.8283 (2)	0.22785 (8)	0.0197 (3)
C8B	0.48841 (7)	0.3151 (2)	0.57499 (8)	0.0198 (3)
C9B	0.46617 (7)	0.3731 (3)	0.62140 (9)	0.0239 (4)
H9BA	0.4899	0.3729	0.6689	0.029*
C10B	0.41040 (7)	0.4296 (3)	0.59768 (9)	0.0260 (4)
H10B	0.3956	0.4675	0.6291	0.031*
C11B	0.37436 (7)	0.4325 (3)	0.52712 (9)	0.0227 (4)
H11B	0.3356	0.4720	0.5113	0.027*
C12B	0.39578 (7)	0.3779 (2)	0.48192 (9)	0.0225 (4)
H12B	0.3716	0.3808	0.4345	0.027*
C13B	0.45352 (6)	0.3169 (2)	0.50437 (8)	0.0193 (3)
C14B	0.47737 (6)	0.2568 (2)	0.45815 (8)	0.0173 (3)
C15B	0.53592 (6)	0.1946 (2)	0.48670 (8)	0.0161 (3)
C16B	0.56309 (7)	0.1301 (2)	0.44572 (8)	0.0204 (3)
H16B	0.5422	0.1278	0.3976	0.025*
C17B	0.61859 (7)	0.0714 (3)	0.47412 (9)	0.0238 (4)
H17B	0.6360	0.0285	0.4459	0.029*
C18B	0.64998 (7)	0.0744 (3)	0.54509 (9)	0.0246 (4)
H18B	0.6888	0.0344	0.5646	0.030*
C19B	0.62536 (7)	0.1343 (2)	0.58664 (9)	0.0225 (4)
H19B	0.6468	0.1340	0.6347	0.027*
C20B	0.56781 (6)	0.1966 (2)	0.55777 (8)	0.0176 (3)
O1W	0.43871 (6)	0.4726 (2)	0.20699 (7)	0.0283 (3)
O2W	0.06211 (6)	0.5855 (2)	0.27425 (7)	0.0285 (3)
O3W	0.46498 (6)	0.14630 (19)	0.27833 (6)	0.0279 (3)
O4W	0.96560 (6)	0.42092 (19)	0.18638 (7)	0.0246 (3)
H1NA	0.6681 (9)	0.745 (3)	0.2392 (11)	0.034 (6)*
H2NA	0.6685 (9)	0.956 (4)	0.2310 (11)	0.043 (7)*
H3NA	0.0766 (9)	0.302 (3)	0.4345 (10)	0.033 (5)*
H4NA	-0.0364 (9)	0.170 (3)	0.6019 (11)	0.037 (6)*
H5NA	-0.0885 (9)	0.234 (3)	0.5365 (10)	0.030 (5)*
H1NB	0.1604 (9)	0.851 (3)	-0.0650 (11)	0.035 (6)*
H2NB	0.1705 (9)	0.659 (3)	-0.0644 (11)	0.038 (6)*
H3NB	0.5628 (10)	0.266 (3)	0.6495 (12)	0.051 (7)*
H4NB	0.4051 (10)	0.296 (3)	0.3709 (11)	0.044 (6)*
H5NB	0.4588 (9)	0.224 (3)	0.3608 (9)	0.041 (6)*
H1W1	0.4462 (11)	0.373 (4)	0.2282 (13)	0.053 (8)*
H2W1	0.4187 (11)	0.536 (4)	0.2230 (13)	0.055 (8)*
H1W2	0.0841 (11)	0.525 (4)	0.3115 (14)	0.060 (8)*
H2W2	0.0538 (11)	0.692 (4)	0.2861 (13)	0.055 (8)*
H1W3	0.4345 (12)	0.077 (4)	0.2731 (14)	0.073 (9)*
H2W3	0.4977 (11)	0.079 (4)	0.2807 (12)	0.052 (7)*
H1W4	0.9960 (10)	0.486 (4)	0.2157 (12)	0.048 (7)*
H2W4	0.9360 (11)	0.492 (4)	0.1677 (12)	0.050 (7)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1A	0.0245 (6)	0.0221 (7)	0.0370 (7)	0.0003 (5)	0.0161 (5)	-0.0081 (6)
O2A	0.0229 (5)	0.0225 (7)	0.0252 (6)	-0.0014 (5)	0.0140 (5)	-0.0016 (5)
N1A	0.0249 (7)	0.0214 (9)	0.0320 (8)	-0.0004 (6)	0.0176 (6)	0.0004 (7)
N2A	0.0198 (6)	0.0182 (8)	0.0200 (7)	0.0004 (5)	0.0124 (6)	-0.0002 (5)
N3A	0.0181 (6)	0.0240 (8)	0.0192 (7)	-0.0002 (5)	0.0093 (6)	0.0018 (6)
C1A	0.0162 (7)	0.0235 (9)	0.0186 (8)	0.0002 (6)	0.0059 (6)	0.0001 (7)
C2A	0.0243 (8)	0.0178 (9)	0.0309 (9)	0.0042 (7)	0.0141 (7)	0.0020 (7)
C3A	0.0233 (8)	0.0199 (9)	0.0260 (9)	0.0005 (7)	0.0128 (7)	0.0033 (7)
C4A	0.0158 (7)	0.0203 (9)	0.0177 (8)	0.0000 (6)	0.0064 (6)	-0.0024 (6)
C5A	0.0163 (7)	0.0168 (9)	0.0207 (8)	0.0011 (6)	0.0036 (6)	-0.0022 (6)
C6A	0.0177 (7)	0.0176 (9)	0.0215 (8)	-0.0022 (6)	0.0064 (6)	0.0005 (7)
C7A	0.0157 (7)	0.0239 (10)	0.0160 (8)	0.0005 (6)	0.0045 (6)	-0.0035 (7)
C8A	0.0192 (7)	0.0134 (8)	0.0191 (8)	-0.0001 (6)	0.0091 (6)	-0.0016 (6)
C9A	0.0195 (7)	0.0208 (9)	0.0278 (9)	0.0012 (6)	0.0139 (7)	-0.0009 (7)
C10A	0.0184 (7)	0.0256 (10)	0.0270 (9)	0.0041 (7)	0.0089 (7)	-0.0003 (7)
C11A	0.0249 (8)	0.0223 (10)	0.0191 (8)	0.0031 (7)	0.0078 (7)	0.0012 (7)
C12A	0.0229 (7)	0.0180 (9)	0.0196 (8)	0.0001 (6)	0.0115 (6)	-0.0002 (6)
C13A	0.0188 (7)	0.0126 (8)	0.0188 (8)	-0.0004 (6)	0.0094 (6)	-0.0024 (6)
C14A	0.0200 (7)	0.0117 (8)	0.0204 (8)	-0.0020 (6)	0.0107 (6)	-0.0034 (6)
C15A	0.0199 (7)	0.0122 (8)	0.0189 (8)	-0.0007 (6)	0.0090 (6)	-0.0019 (6)
C16A	0.0191 (7)	0.0182 (9)	0.0230 (8)	-0.0012 (6)	0.0107 (6)	-0.0021 (7)
C17A	0.0208 (7)	0.0217 (10)	0.0207 (8)	0.0013 (6)	0.0069 (6)	-0.0012 (7)
C18A	0.0284 (8)	0.0227 (10)	0.0168 (8)	0.0008 (7)	0.0086 (7)	-0.0007 (7)
C19A	0.0258 (8)	0.0190 (9)	0.0211 (8)	-0.0009 (6)	0.0133 (7)	-0.0013 (7)
C20A	0.0203 (7)	0.0135 (8)	0.0193 (8)	-0.0016 (6)	0.0101 (6)	-0.0022 (6)
O1B	0.0241 (6)	0.0221 (7)	0.0224 (6)	0.0018 (5)	0.0094 (5)	0.0003 (5)
O2B	0.0246 (6)	0.0209 (7)	0.0302 (7)	-0.0014 (5)	0.0089 (5)	-0.0067 (5)
N1B	0.0218 (7)	0.0222 (9)	0.0235 (8)	-0.0007 (6)	0.0084 (6)	-0.0014 (7)
N2B	0.0231 (7)	0.0204 (8)	0.0160 (7)	0.0007 (6)	0.0072 (6)	0.0002 (6)
N3B	0.0188 (6)	0.0276 (9)	0.0182 (7)	0.0006 (6)	0.0063 (6)	-0.0009 (6)
C1B	0.0154 (7)	0.0233 (9)	0.0221 (8)	-0.0014 (6)	0.0105 (6)	-0.0002 (7)
C2B	0.0245 (8)	0.0179 (9)	0.0296 (9)	0.0019 (7)	0.0098 (7)	0.0005 (7)
C3B	0.0242 (8)	0.0178 (9)	0.0288 (9)	-0.0012 (7)	0.0103 (7)	-0.0046 (7)
C4B	0.0171 (7)	0.0194 (9)	0.0233 (8)	-0.0024 (6)	0.0112 (6)	-0.0019 (7)
C5B	0.0181 (7)	0.0180 (9)	0.0232 (8)	-0.0005 (6)	0.0107 (6)	0.0008 (7)
C6B	0.0208 (7)	0.0171 (9)	0.0281 (9)	-0.0035 (6)	0.0152 (7)	-0.0044 (7)
C7B	0.0175 (7)	0.0218 (9)	0.0238 (8)	-0.0009 (6)	0.0127 (6)	-0.0021 (7)
C8B	0.0241 (8)	0.0141 (9)	0.0229 (8)	-0.0028 (6)	0.0115 (7)	-0.0003 (6)
C9B	0.0275 (8)	0.0246 (10)	0.0199 (8)	-0.0012 (7)	0.0103 (7)	-0.0004 (7)
C10B	0.0274 (8)	0.0259 (10)	0.0285 (9)	0.0002 (7)	0.0154 (7)	-0.0017 (8)
C11B	0.0199 (7)	0.0226 (10)	0.0262 (9)	0.0025 (6)	0.0104 (7)	0.0013 (7)
C12B	0.0208 (7)	0.0214 (10)	0.0227 (8)	-0.0012 (6)	0.0067 (7)	0.0012 (7)
C13B	0.0169 (7)	0.0149 (9)	0.0258 (9)	-0.0012 (6)	0.0087 (6)	0.0009 (7)
C14B	0.0183 (7)	0.0131 (8)	0.0175 (8)	-0.0036 (6)	0.0045 (6)	0.0018 (6)
C15B	0.0168 (7)	0.0133 (8)	0.0173 (7)	-0.0008 (6)	0.0063 (6)	0.0022 (6)

C16B	0.0230 (7)	0.0202 (9)	0.0193 (8)	-0.0020 (6)	0.0099 (6)	0.0007 (7)
C17B	0.0228 (8)	0.0239 (10)	0.0294 (9)	0.0007 (7)	0.0156 (7)	0.0019 (7)
C18B	0.0154 (7)	0.0244 (10)	0.0322 (10)	0.0019 (6)	0.0082 (7)	0.0042 (8)
C19B	0.0197 (7)	0.0226 (10)	0.0189 (8)	-0.0009 (6)	0.0021 (6)	0.0027 (7)
C20B	0.0185 (7)	0.0147 (8)	0.0184 (8)	-0.0024 (6)	0.0065 (6)	0.0007 (6)
O1W	0.0313 (7)	0.0254 (8)	0.0297 (7)	0.0064 (6)	0.0143 (6)	0.0006 (6)
O2W	0.0336 (7)	0.0227 (8)	0.0252 (7)	0.0030 (6)	0.0085 (6)	0.0018 (6)
O3W	0.0266 (6)	0.0280 (8)	0.0281 (7)	-0.0028 (6)	0.0104 (5)	-0.0018 (6)
O4W	0.0265 (6)	0.0227 (7)	0.0247 (7)	0.0024 (5)	0.0110 (5)	-0.0016 (5)

Geometric parameters (Å, °)

O1A—C7A	1.257 (2)	N1B—H2NB	0.89 (2)
O2A—C7A	1.273 (2)	N2B—C8B	1.363 (2)
N1A—C1A	1.390 (2)	N2B—C20B	1.365 (2)
N1A—H1NA	0.91 (2)	N2B—H3NB	0.98 (2)
N1A—H2NA	0.92 (3)	N3B—C14B	1.317 (2)
N2A—C8A	1.363 (2)	N3B—H4NB	0.99 (2)
N2A—C20A	1.365 (2)	N3B—H5NB	0.898 (9)
N2A—H3NA	0.90 (2)	C1B—C6B	1.396 (2)
N3A—C14A	1.3234 (19)	C1B—C2B	1.404 (2)
N3A—H4NA	0.93 (2)	C2B—C3B	1.383 (2)
N3A—H5NA	0.91 (2)	C2B—H2BA	0.9500
C1A—C2A	1.398 (2)	C3B—C4B	1.395 (2)
C1A—C6A	1.401 (2)	C3B—H3BA	0.9500
C2A—C3A	1.391 (2)	C4B—C5B	1.398 (2)
C2A—H2AA	0.9500	C4B—C7B	1.500 (2)
C3A—C4A	1.390 (2)	C5B—C6B	1.386 (2)
C3A—H3AA	0.9500	C5B—H5BA	0.9500
C4A—C5A	1.398 (2)	C6B—H6BA	0.9500
C4A—C7A	1.504 (2)	C8B—C13B	1.410 (2)
C5A—C6A	1.383 (2)	C8B—C9B	1.413 (2)
C5A—H5AA	0.9500	C9B—C10B	1.366 (2)
C6A—H6AA	0.9500	C9B—H9BA	0.9500
C8A—C9A	1.412 (2)	C10B—C11B	1.415 (2)
C8A—C13A	1.414 (2)	C10B—H10B	0.9500
C9A—C10A	1.370 (2)	C11B—C12B	1.368 (2)
C9A—H9AA	0.9500	C11B—H11B	0.9500
C10A—C11A	1.407 (2)	C12B—C13B	1.424 (2)
C10A—H10A	0.9500	C12B—H12B	0.9500
C11A—C12A	1.367 (2)	C13B—C14B	1.443 (2)
C11A—H11A	0.9500	C14B—C15B	1.438 (2)
C12A—C13A	1.422 (2)	C15B—C20B	1.407 (2)
C12A—H12A	0.9500	C15B—C16B	1.420 (2)
C13A—C14A	1.439 (2)	C16B—C17B	1.362 (2)
C14A—C15A	1.439 (2)	C16B—H16B	0.9500
C15A—C20A	1.415 (2)	C17B—C18B	1.403 (2)
C15A—C16A	1.419 (2)	C17B—H17B	0.9500

C16A—C17A	1.366 (2)	C18B—C19B	1.370 (2)
C16A—H16A	0.9500	C18B—H18B	0.9500
C17A—C18A	1.411 (2)	C19B—C20B	1.416 (2)
C17A—H17A	0.9500	C19B—H19B	0.9500
C18A—C19A	1.368 (2)	O1W—H1W1	0.84 (3)
C18A—H18A	0.9500	O1W—H2W1	0.87 (3)
C19A—C20A	1.411 (2)	O2W—H1W2	0.88 (3)
C19A—H19A	0.9500	O2W—H2W2	0.87 (3)
O1B—C7B	1.273 (2)	O3W—H1W3	0.90 (3)
O2B—C7B	1.256 (2)	O3W—H2W3	0.96 (3)
N1B—C1B	1.381 (2)	O4W—H1W4	0.91 (3)
N1B—H1NB	0.86 (2)	O4W—H2W4	0.87 (3)
C1A—N1A—H1NA	115.8 (13)	H1NB—N1B—H2NB	109 (2)
C1A—N1A—H2NA	117.7 (14)	C8B—N2B—C20B	121.86 (14)
H1NA—N1A—H2NA	115.3 (19)	C8B—N2B—H3NB	111.3 (13)
C8A—N2A—C20A	122.22 (13)	C20B—N2B—H3NB	126.9 (14)
C8A—N2A—H3NA	118.9 (13)	C14B—N3B—H4NB	122.9 (13)
C20A—N2A—H3NA	118.7 (13)	C14B—N3B—H5NB	124.5 (14)
C14A—N3A—H4NA	123.4 (13)	H4NB—N3B—H5NB	112.5 (19)
C14A—N3A—H5NA	121.4 (13)	N1B—C1B—C6B	122.06 (16)
H4NA—N3A—H5NA	115.1 (18)	N1B—C1B—C2B	119.75 (16)
N1A—C1A—C2A	121.01 (16)	C6B—C1B—C2B	118.08 (15)
N1A—C1A—C6A	120.43 (16)	C3B—C2B—C1B	120.71 (17)
C2A—C1A—C6A	118.41 (14)	C3B—C2B—H2BA	119.6
C3A—C2A—C1A	120.60 (16)	C1B—C2B—H2BA	119.6
C3A—C2A—H2AA	119.7	C2B—C3B—C4B	121.29 (17)
C1A—C2A—H2AA	119.7	C2B—C3B—H3BA	119.4
C4A—C3A—C2A	121.16 (16)	C4B—C3B—H3BA	119.4
C4A—C3A—H3AA	119.4	C3B—C4B—C5B	117.89 (15)
C2A—C3A—H3AA	119.4	C3B—C4B—C7B	120.02 (16)
C3A—C4A—C5A	117.87 (14)	C5B—C4B—C7B	122.08 (15)
C3A—C4A—C7A	122.25 (15)	C6B—C5B—C4B	121.15 (16)
C5A—C4A—C7A	119.88 (15)	C6B—C5B—H5BA	119.4
C6A—C5A—C4A	121.61 (16)	C4B—C5B—H5BA	119.4
C6A—C5A—H5AA	119.2	C5B—C6B—C1B	120.81 (16)
C4A—C5A—H5AA	119.2	C5B—C6B—H6BA	119.6
C5A—C6A—C1A	120.28 (16)	C1B—C6B—H6BA	119.6
C5A—C6A—H6AA	119.9	O2B—C7B—O1B	123.11 (15)
C1A—C6A—H6AA	119.9	O2B—C7B—C4B	117.87 (15)
O1A—C7A—O2A	123.20 (14)	O1B—C7B—C4B	119.02 (15)
O1A—C7A—C4A	117.83 (15)	N2B—C8B—C13B	120.76 (14)
O2A—C7A—C4A	118.97 (15)	N2B—C8B—C9B	118.87 (15)
N2A—C8A—C9A	119.55 (14)	C13B—C8B—C9B	120.36 (15)
N2A—C8A—C13A	120.78 (13)	C10B—C9B—C8B	119.75 (16)
C9A—C8A—C13A	119.67 (14)	C10B—C9B—H9BA	120.1
C10A—C9A—C8A	120.19 (14)	C8B—C9B—H9BA	120.1
C10A—C9A—H9AA	119.9	C9B—C10B—C11B	121.14 (16)

C8A—C9A—H9AA	119.9	C9B—C10B—H10B	119.4
C9A—C10A—C11A	120.74 (15)	C11B—C10B—H10B	119.4
C9A—C10A—H10A	119.6	C12B—C11B—C10B	119.38 (15)
C11A—C10A—H10A	119.6	C12B—C11B—H11B	120.3
C12A—C11A—C10A	119.94 (15)	C10B—C11B—H11B	120.3
C12A—C11A—H11A	120.0	C11B—C12B—C13B	121.41 (16)
C10A—C11A—H11A	120.0	C11B—C12B—H12B	119.3
C11A—C12A—C13A	121.01 (14)	C13B—C12B—H12B	119.3
C11A—C12A—H12A	119.5	C8B—C13B—C12B	117.95 (15)
C13A—C12A—H12A	119.5	C8B—C13B—C14B	119.16 (14)
C8A—C13A—C12A	118.41 (14)	C12B—C13B—C14B	122.88 (15)
C8A—C13A—C14A	118.87 (14)	N3B—C14B—C15B	121.28 (15)
C12A—C13A—C14A	122.72 (13)	N3B—C14B—C13B	120.70 (14)
N3A—C14A—C15A	120.96 (14)	C15B—C14B—C13B	118.02 (14)
N3A—C14A—C13A	120.49 (14)	C20B—C15B—C16B	118.28 (14)
C15A—C14A—C13A	118.55 (13)	C20B—C15B—C14B	119.32 (14)
C20A—C15A—C16A	118.08 (14)	C16B—C15B—C14B	122.40 (14)
C20A—C15A—C14A	119.11 (13)	C17B—C16B—C15B	121.18 (15)
C16A—C15A—C14A	122.81 (14)	C17B—C16B—H16B	119.4
C17A—C16A—C15A	121.24 (14)	C15B—C16B—H16B	119.4
C17A—C16A—H16A	119.4	C16B—C17B—C18B	119.99 (15)
C15A—C16A—H16A	119.4	C16B—C17B—H17B	120.0
C16A—C17A—C18A	119.80 (15)	C18B—C17B—H17B	120.0
C16A—C17A—H17A	120.1	C19B—C18B—C17B	120.81 (15)
C18A—C17A—H17A	120.1	C19B—C18B—H18B	119.6
C19A—C18A—C17A	120.89 (15)	C17B—C18B—H18B	119.6
C19A—C18A—H18A	119.6	C18B—C19B—C20B	119.78 (15)
C17A—C18A—H18A	119.6	C18B—C19B—H19B	120.1
C18A—C19A—C20A	119.83 (15)	C20B—C19B—H19B	120.1
C18A—C19A—H19A	120.1	N2B—C20B—C15B	120.86 (14)
C20A—C19A—H19A	120.1	N2B—C20B—C19B	119.18 (14)
N2A—C20A—C19A	119.40 (13)	C15B—C20B—C19B	119.96 (14)
N2A—C20A—C15A	120.45 (14)	H1W1—O1W—H2W1	107 (2)
C19A—C20A—C15A	120.15 (14)	H1W2—O2W—H2W2	108 (2)
C1B—N1B—H1NB	120.6 (15)	H1W3—O3W—H2W3	114 (2)
C1B—N1B—H2NB	119.1 (14)	H1W4—O4W—H2W4	110 (2)
N1A—C1A—C2A—C3A	-173.95 (16)	N1B—C1B—C2B—C3B	174.26 (15)
C6A—C1A—C2A—C3A	1.6 (2)	C6B—C1B—C2B—C3B	-2.1 (2)
C1A—C2A—C3A—C4A	0.7 (3)	C1B—C2B—C3B—C4B	0.0 (3)
C2A—C3A—C4A—C5A	-1.9 (2)	C2B—C3B—C4B—C5B	1.4 (2)
C2A—C3A—C4A—C7A	178.03 (15)	C2B—C3B—C4B—C7B	-179.87 (15)
C3A—C4A—C5A—C6A	0.7 (2)	C3B—C4B—C5B—C6B	-0.7 (2)
C7A—C4A—C5A—C6A	-179.19 (14)	C7B—C4B—C5B—C6B	-179.40 (14)
C4A—C5A—C6A—C1A	1.6 (2)	C4B—C5B—C6B—C1B	-1.4 (2)
N1A—C1A—C6A—C5A	172.84 (15)	N1B—C1B—C6B—C5B	-173.47 (14)
C2A—C1A—C6A—C5A	-2.8 (2)	C2B—C1B—C6B—C5B	2.8 (2)
C3A—C4A—C7A—O1A	-171.22 (15)	C3B—C4B—C7B—O2B	-4.6 (2)

C5A—C4A—C7A—O1A	8.7 (2)	C5B—C4B—C7B—O2B	174.05 (14)
C3A—C4A—C7A—O2A	9.0 (2)	C3B—C4B—C7B—O1B	175.43 (14)
C5A—C4A—C7A—O2A	-171.12 (14)	C5B—C4B—C7B—O1B	-5.9 (2)
C20A—N2A—C8A—C9A	178.09 (16)	C20B—N2B—C8B—C13B	-0.8 (2)
C20A—N2A—C8A—C13A	-1.6 (2)	C20B—N2B—C8B—C9B	178.96 (16)
N2A—C8A—C9A—C10A	177.90 (16)	N2B—C8B—C9B—C10B	-179.17 (17)
C13A—C8A—C9A—C10A	-2.5 (3)	C13B—C8B—C9B—C10B	0.6 (3)
C8A—C9A—C10A—C11A	1.1 (3)	C8B—C9B—C10B—C11B	-0.5 (3)
C9A—C10A—C11A—C12A	0.9 (3)	C9B—C10B—C11B—C12B	0.0 (3)
C10A—C11A—C12A—C13A	-1.4 (3)	C10B—C11B—C12B—C13B	0.5 (3)
N2A—C8A—C13A—C12A	-178.44 (15)	N2B—C8B—C13B—C12B	179.58 (15)
C9A—C8A—C13A—C12A	1.9 (2)	C9B—C8B—C13B—C12B	-0.2 (2)
N2A—C8A—C13A—C14A	1.7 (2)	N2B—C8B—C13B—C14B	-0.2 (2)
C9A—C8A—C13A—C14A	-177.91 (15)	C9B—C8B—C13B—C14B	-179.97 (16)
C11A—C12A—C13A—C8A	0.0 (2)	C11B—C12B—C13B—C8B	-0.3 (3)
C11A—C12A—C13A—C14A	179.83 (16)	C11B—C12B—C13B—C14B	179.42 (17)
C8A—C13A—C14A—N3A	178.94 (16)	C8B—C13B—C14B—N3B	-178.63 (16)
C12A—C13A—C14A—N3A	-0.9 (2)	C12B—C13B—C14B—N3B	1.6 (3)
C8A—C13A—C14A—C15A	-1.2 (2)	C8B—C13B—C14B—C15B	1.0 (2)
C12A—C13A—C14A—C15A	178.99 (15)	C12B—C13B—C14B—C15B	-178.80 (15)
N3A—C14A—C15A—C20A	-179.66 (16)	N3B—C14B—C15B—C20B	178.81 (16)
C13A—C14A—C15A—C20A	0.5 (2)	C13B—C14B—C15B—C20B	-0.8 (2)
N3A—C14A—C15A—C16A	0.2 (2)	N3B—C14B—C15B—C16B	-1.4 (2)
C13A—C14A—C15A—C16A	-179.69 (15)	C13B—C14B—C15B—C16B	179.00 (15)
C20A—C15A—C16A—C17A	-0.2 (2)	C20B—C15B—C16B—C17B	0.0 (2)
C14A—C15A—C16A—C17A	179.94 (16)	C14B—C15B—C16B—C17B	-179.77 (16)
C15A—C16A—C17A—C18A	0.4 (3)	C15B—C16B—C17B—C18B	-0.1 (3)
C16A—C17A—C18A—C19A	-0.6 (3)	C16B—C17B—C18B—C19B	0.6 (3)
C17A—C18A—C19A—C20A	0.4 (3)	C17B—C18B—C19B—C20B	-0.9 (3)
C8A—N2A—C20A—C19A	-179.39 (15)	C8B—N2B—C20B—C15B	1.0 (2)
C8A—N2A—C20A—C15A	0.8 (2)	C8B—N2B—C20B—C19B	-178.62 (16)
C18A—C19A—C20A—N2A	179.99 (16)	C16B—C15B—C20B—N2B	-179.97 (15)
C18A—C19A—C20A—C15A	-0.2 (2)	C14B—C15B—C20B—N2B	-0.2 (2)
C16A—C15A—C20A—N2A	179.90 (15)	C16B—C15B—C20B—C19B	-0.3 (2)
C14A—C15A—C20A—N2A	-0.3 (2)	C14B—C15B—C20B—C19B	179.44 (15)
C16A—C15A—C20A—C19A	0.1 (2)	C18B—C19B—C20B—N2B	-179.56 (16)
C14A—C15A—C20A—C19A	179.93 (15)	C18B—C19B—C20B—C15B	0.8 (3)

Hydrogen-bond geometry (Å, °)

Cg13 is the centroid of the C1A—C6A ring.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O1 <i>W</i> —H1 <i>W</i> 1...O3 <i>W</i>	0.84 (3)	1.92 (3)	2.758 (2)	179 (5)
O1 <i>W</i> —H2 <i>W</i> 1...O1 <i>B</i>	0.86 (3)	1.90 (3)	2.767 (2)	177 (2)
O2 <i>W</i> —H1 <i>W</i> 2...O2 <i>A</i> ⁱ	0.88 (3)	1.92 (3)	2.8066 (19)	176 (3)
O3 <i>W</i> —H2 <i>W</i> 3...O1 <i>W</i> ⁱⁱ	0.96 (3)	1.72 (3)	2.676 (2)	174 (2)
O3 <i>W</i> —H1 <i>W</i> 3...O2 <i>B</i> ⁱⁱ	0.90 (3)	1.79 (3)	2.674 (2)	166 (3)
O4 <i>W</i> —H2 <i>W</i> 4...O1 <i>A</i>	0.87 (3)	1.84 (3)	2.688 (2)	164 (3)

N1A—H1NA···O2B ⁱ	0.92 (2)	2.09 (2)	2.946 (2)	156.0 (19)
N1A—H2NA···O1B ⁱⁱⁱ	0.92 (3)	2.12 (3)	2.983 (2)	158 (2)
N2A—H3NA···O2A ⁱ	0.90 (2)	1.98 (2)	2.882 (2)	173 (2)
N2B—H3NB···O1B ^{iv}	0.98 (2)	1.93 (2)	2.8695 (18)	160 (2)
N3A—H4NA···O4W ^v	0.93 (2)	1.93 (2)	2.8227 (19)	161.6 (18)
N3A—H5NA···N1B ^{vi}	0.91 (2)	2.20 (2)	3.031 (2)	151.4 (18)
N3B—H5NB···O3W	0.90 (2)	1.94 (2)	2.819 (2)	165 (2)
N3B—H4NB···N1A ⁱ	1.00 (3)	2.16 (2)	3.067 (2)	151.3 (19)
C12A—H12A···O4W ^v	0.95	2.55	3.4756 (19)	164
C16B—H16B···O3W	0.95	2.51	3.438 (2)	164
C11A—H11A···Cg13 ^{iv}	0.95	2.95	3.6659 (19)	133

Symmetry codes: (i) $-x+1, y-1/2, -z+1/2$; (ii) $x, y-1, z$; (iii) $-x+1, y+1/2, -z+1/2$; (iv) $-x+1, -y+1, -z+1$; (v) $x-1, -y+1/2, z+1/2$; (vi) $-x, y-1/2, -z+1/2$.