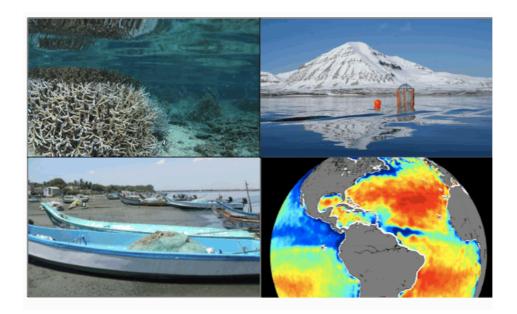
1	Salinity from space unlocks satellite-based
2	assessment of ocean acidification
3	Peter E. Land ^{1*} , Jamie D. Shutler ² , Helen S. Findlay ¹ , Fanny Girard-Ardhuin ³ ,
4	Roberto Sabia ^{4,} Nicolas Reul ³ , Jean-Francois Piolle ³ , Bertrand Chapron ³ , Yves
5	$Quilfen^3$, Joseph Salisbury ⁵ , Douglas Vandemark ⁵ , Richard Bellerby ⁶ , and Punyasloke
6	Bhadury ⁷
7	¹ Plymouth Marine Laboratory, Prospect Place, The Hoe, Plymouth PL1 3DH, UK
8	² University of Exeter, Penryn Campus, Cornwall. TR10 9FE, UK
9	³ Institut Francais Recherche Pour L'Exploitation de la Mer, Pointe du Diable, 29280
10	Plouzané, France
11	⁴ Telespazio-Vega UK for European Space Agency (ESA), ESTEC, Noordwijk, the
12	Netherlands
13	⁵ Ocean Processes Analysis Laboratory, University of New Hampshire, Durham, NH
14	3824, USA
15	⁶ Norwegian Institute for Water Research, Thormøhlensgate 53 D, N-5006 Bergen,
16	Norway
17	⁷ Department of Biological Sciences, Indian Institute of Science Education and
18	Research-Kolkata, Mohanpur - 741 246, West Bengal, India



Abstract artwork

Note to editor (top left to bottom right): Tropical coral; Svalbard in the Barents Sea; Beach in India on the coast of the Bay of Bengal; Salinity from space (SMOS) showing the Amazon plume. All images taken by PML staff and used with permission.

20

21 Approximately a quarter of the carbon dioxide (CO_2) that we emit into the atmosphere 22 is absorbed by the ocean. This oceanic uptake of CO₂ leads to a change in marine 23 carbonate chemistry resulting in a decrease of seawater pH and carbonate ion 24 concentration, a process commonly called 'Ocean Acidification'. Salinity data are key 25 for assessing the marine carbonate system, and new space-based salinity 26 measurements will enable the development of novel space-based ocean acidification 27 assessment. Recent studies have highlighted the need to develop new in situ 28 technology for monitoring ocean acidification, but the potential capabilities of space-29 based measurements remain largely untapped. Routine measurements from space can 30 provide quasi-synoptic, reproducible data for investigating processes on global scales;
31 they may also be the most efficient way to monitor the ocean surface. As the carbon
32 cycle is dominantly controlled by the balance between the biological and solubility
33 carbon pumps, innovative methods to exploit existing satellite sea surface temperature
34 and ocean color, and new satellite sea surface salinity measurements, are needed and
35 will enable frequent assessment of ocean acidification parameters over large spatial
36 scales.

37 **1. Introduction**

38

39 Each year global emissions of carbon dioxide (CO₂) into our atmosphere continue to 40 rise. These increasing atmospheric concentrations cause a net influx of CO₂ into the oceans. Of the roughly 36 billion metric tons of CO2 that is emitted into our 41 42 atmosphere each year, approximately a quarter transfers into the oceans ¹. This CO₂ 43 addition has caused a shift in the seawater carbonate system, termed Ocean 44 Acidification (OA), resulting in a 26% increase in acidity and a 16% decrease in 45 carbonate ion concentration since the industrial revolution². Recently there has been recognition that this acidification is not occurring uniformly across the global oceans, 46 with some regions acidifying faster than others ^{3,4}. However, the overall cause of OA 47 48 remains consistent: the addition of CO_2 into the oceans, and as such, it remains a 49 global issue. Continual emissions of CO_2 into the atmosphere over the next century 50 will decrease average surface ocean pH to levels which will be deleterious to many 51 marine ecosystems and the services they provide 5.

52

53 While the seawater carbonate system is relatively complex, two parameters have been 54 suggested as pertinent to the monitoring and assessment of OA through time and 55 space. These are pH (the measure of acidity) and calcium carbonate (CaCO₃) mineral 56 saturation state, with aragonite generally considered to be an important CaCO₃ 57 mineral to be monitored because of its relevance to marine organisms (e.g. corals) and 58 its relative solubility. Thermodynamically, CaCO₃ is stable when the saturation state 59 (an index of the concentrations of calcium and carbonate ions) is greater than one and 60 becomes unstable when seawater becomes undersaturated with these ions (saturation 61 < 1). While there is significant variability between types of organism, there is ample 62 experimental evidence that many calcifying organisms are sensitive to OA⁶, and that 63 thresholds exist below which some organisms become stressed and their well-being 64 and existence becomes threatened ⁷. Increasingly evidence suggests that the physiology and behaviour of calcifying and non-calcifying organisms can be impacted 65 66 by increasing OA⁸, with cascading effects on the food chain and protein supply for humans³, and alterations to the functioning of ecosystems and feedbacks to our 67 climate ⁹. 68

69

70 In 2012 the Global Ocean Acidification Observing Network (GOA-ON, www.goa-71 on.org) was formed in an attempt to bring together expertise, datasets and resources to 72 improve OA monitoring. At present, OA monitoring efforts are dominated by in situ 73 observations from moorings, ships and associated platforms. Whilst key to any 74 monitoring campaign, in situ data tend to be spatially sparse, especially in 75 inhospitable regions, and so on their own are unlikely to provide a comprehensive, 76 robust and cost effective solution to global OA monitoring. The need to monitor and 77 study large areas of the Earth has driven the development of satellite-based sensors.

Increasingly, as *in situ* data accumulate, attempts are being made to use *in situ* hydrographic data ¹⁰⁻¹³ and/or remotely-sensed data ^{14, 15} to provide proxies and indicators for the condition of the carbonate system, enabling data gaps to be filled in both space and time. The increased availability of *in situ* data creates a substantial dataset to develop and test the capabilities of satellite-derived products, and we suggest that the recent availability of satellite-based salinity measurements provides new key insights for studying and assessing OA from space.

86 **2.** The complexities of the carbonate system

87

e complexities of the carbonate system

The oceanic carbonate system can be understood and probed through four key parameters: total alkalinity (TA), dissolved inorganic carbon (DIC), pH and fugacity of CO_2 (f_{CO2}). The latter may be replaced with the related partial pressure of CO_2 , p_{CO2} , from which f_{CO2} can be calculated, and the two are often used interchangeably. In principle, knowledge of any two of these four is sufficient to solve the carbonate system equations. However, over-determination, the process of measuring at least three parameters, is advantageous.

95

96 The relationships between the different carbonate system parameters are 97 fundamentally driven by thermodynamics, hence influenced by temperature and 98 pressure, and knowing these is fundamental for calculating the carbonate system as a 99 whole ¹⁶. Water temperature is the major controller of the solubility of CO_2 ¹⁷, so 100 seasonal changes in sea temperature can, depending on the region, be significant for 101 driving changes in f_{CO2} (and consequently DIC and pH). Salinity affects the 102 coefficients of the carbonate system equations. Hence to solve the equations, it is 103 necessary to estimate temperature, salinity and pressure along with carbonate104 parameters.

105

106 The ratio between ions (the constituents of salinity) will tend to remain constant 107 anywhere in the global oceans, resulting in a strong relationship between TA and salinity ¹⁸. Unfortunately, a universal relationship between TA and salinity does not 108 109 apply in certain regions, for instance in areas influenced by freshwater outflows from rivers⁷, or areas where calcification and/or CaCO₃ dissolution occurs, such as where 110 calcifying plankton are prevalent ¹⁹. In these regions, it is therefore critical to gain 111 112 additional local knowledge. For example, different rivers will have different ionic 113 concentrations (and therefore different TA concentrations) depending on the 114 surrounding geology and hydrology.

115

For DIC, f_{CO2} (or p_{CO2}) and pH, the other important process is biological activity ¹⁹. Removal or addition of CO₂ by plankton photosynthesis or respiration can be a significant component of the seasonal signal ²⁰. Biological activity, in turn, is driven by factors such as nutrient dynamics and light conditions, which again are regionally specific. Measurements of chlorophyll (a proxy for biomass) and/or oxygen concentration can be useful for interpreting the biological component of the carbon signal.

123

The combination of these processes means that it is extremely challenging to produce a global relationship between any component of the carbonate system and its drivers. To enable us to understand these dynamics, extrapolation from collected data points to the global ocean is needed, and along with model predictions, empirical

relationships and datasets are important and need to be studied and developed. OA needs to be assessed using these relationships on a global scale, but regional complexities, particularly where riverine and coastal processes dominate ^{21, 22}, cause significant challenges for global empirical relationships.

132

133 **3. Current** *in situ* approaches and challenges

134

Laboratory measurements are the gold standard for assessing the carbonate system in 135 seawater, with accuracy far in excess of that achievable from satellites.²³⁻²⁵ However, 136 137 research vessel time is expensive and limited in coverage, so autonomous in situ instruments are also deployed, e.g. on buoys, with less accuracy 26 . A notable example 138 139 is the Argo network of over 3000 drifters, which measure temperature and salinity 140 throughout the deep global ocean. Interpolation of Argo data is much less challenging 141 than for most *in situ* measurements. Argo is the closest *in situ* data have come to the 142 global, synoptic measurements possible with satellites, but shallow or enclosed seas 143 are not represented (there are as yet no Argo instruments in the open Arctic Ocean). 144 Table 1 lists more examples. Of the four key parameters, only f_{CO2} (or p_{CO2}) and pH 145 are routinely monitored *in situ*. As yet there are limited capabilities to measure DIC 146 and TA autonomously, hence these parameters must be measured either in a ship-147 based laboratory or on land.

Dataset name and reference	Temporal period	Geographic location	Variables	No. of data points
SOCAT v2.0 ²⁷	1968-2011	Global*	fCO ₂ , SSS, SST	6,000,000+
LDEO v2012 ²⁸	1980-present	Global*	pCO ₂ , SSS, SST	6,000,000+
GLODAP ²⁹	1970-2000	Global	TA, DIC, SSS, SST, Nitrate	10,000+

CARINA AMS v1.2 ³⁰	1980-2006	Arctic	TA, DIC, SSS, SST	1500+
CARINA ATL v1.0 ³¹		Atlantic		
CARINA SO v1.1 ³²		Southern Ocean		
AMT ³³	1995-present	Atlantic	pCO _{2W} , SSS, SST, Chl, pH	1000+
NIVA Ferrybox ³⁴	2008-present	Arctic	pCO _{2W} , TA, DIC, SSS, SST	1000+
OWS Mike ³⁵	1948-2009	Arctic	TA, DIC, SSS, SST, Chl	1000+
RAMA Moored buoy array ³⁶	2007-present	Bay of Bengal	SSS, SST	1000+
ARGO buoys ³⁷	2003-present	Global	SSS, SST	1,000,000+
OOI ³⁸	2014 onwards	Global (6 sites)	pCO ₂ , SSS, SST, nitrate	New program
SOCCOM ³⁹	2014 onwards	Southern Ocean	SSS, SST, pH, nitrate	New program

150 Table 1. In situ datasets and programs than can be used for the development and

151 validation of OA remote sensing algorithms.

152 4. Potential of space based observations153

154 4.1 Advantages and disadvantages

155

156 While it has proven difficult to use remote sensing to directly monitor and detect

157 changes in seawater pH and their impact on marine organisms ²², satellites can

158 measure sea surface temperature and salinity (SST and SSS) and surface chlorophyll-

a, from which carbonate system parameters can be estimated using empirical

160 relationships derived from *in situ* data. Although surface measurements may not be

161 representative of important biological processes, e.g. fish or shellfish, observations at

- 162 the surface are particularly important for OA because the change in carbonate
- 163 chemistry due to atmospheric CO₂ occurs in the surface first. Thus satellites have

164 great potential as a tool for assessing changes in carbonate chemistry.

166 SST has been measured from space with infrared radiometry since the 1960s, but the data are only globally of sufficient quality for climate studies since 1991⁴⁰. Satellite 167 168 measurements of chlorophyll-a in the visible are more recent, starting in 1986 and delivering high quality global data since 1997⁴¹. Both measurements are made 169 170 globally at high spatial and temporal resolution, but with data gaps due to effects such 171 as cloud, which can greatly affect data availability in cloudy regions. SST is measured 172 in the top few microns, and chlorophyll-a is generally measured to depths around 1-173 100m, depending on water clarity. Data quality can be affected by many issues, e.g. 174 adjacent land or ice may affect both SST and chlorophyll-a retrievals, and suspended 175 sediment may affect chlorophyll-a retrievals.

176

177 Only since 2009 has a satellite-based capability for measuring SSS existed. Increasing 178 salinity decreases the emissivity of seawater and so changes the microwave radiation 179 emitted at the water surface. ESA Soil Moisture and Ocean Salinity (SMOS) and 180 NASA-CONAE Aquarius (launched in 2009 and 2011 respectively, both currently in 181 operation), are L-band microwave sensors designed to detect variations in microwave 182 radiation and thus estimate ocean salinity in the top centimeter. The instruments are 183 novel and the measurement is very challenging, and research is ongoing to improve data quality⁴². The instruments can measure every few days at a spatial resolution of 184 185 35-100km, but single measurements are very noisy, so the instantaneous swath data 186 are generally spatially and temporally averaged over 10 days or a month, with an 187 intended accuracy around 0.1 - 0.2 g/kg for monthly 200 km data. A particular issue 188 close to urban areas is radio frequency interference from illegal broadcasts, which are 189 gradually being eliminated but still result in large data gaps, particularly for SMOS.

190 The signal can be affected by nearby land or sea ice, and the sensitivity to SSS 191 decreases for cold water, by about 50% from 20°C to $0^{\circ}C^{43}$.

192

193 With these challenges, a central question is whether satellite SSS can bring new 194 complementary information to in situ SSS measurements such as Argo for assessing OA. Direct comparisons^{44, 45} indicate differences of 0.15-0.5 g/kg in a 1°x1° region 195 over 10-30 days. The two are difficult to compare directly however, as Argo measures 196 197 5m or more from the surface, so some differences are expected even in the absence of 198 errors, especially where the water column is stratified. A better strategy might be to compare their effectiveness in estimating OA. How the uncertainties propagate 199 200 through the carbonate system calculations is the subject of ongoing research.

201

202 Despite biases and uncertainties, satellite measurements of SSS in the top centimeter contain geophysical information not detected by Argo^{46,47}. In addition, Argo coverage 203 204 can be much poorer than satellite SSS in several regions such as the major western 205 boundary or equatorial currents and across strong oceanic fronts. The use of 206 interpolated Argo products presents an additional source of uncertainty due to the interpolation scheme.⁴⁸ Satellite SSS can also resolve mesoscale spatial structures not 207 resolved by Argo measurements⁴⁹, and unlike Argo, satellites provide a synoptic 208 209 'snapshot' of a region at a given time.

210

Regular mapping of the SSS field with unprecedented temporal and spatial resolution at global scale is now possible from satellites. The impact of using satellite SSS for carbonate system algorithms can now be tested, where previously there was a reliance on climatology, *in situ* or model data. For example, this provides the means to study

the impact that freshwater influences (sea ice melt, riverine inputs and rain) can have on the marine carbonate system. The use of satellite SSS data will also allow evaluation of the impact on the carbonate system of the inter- and intra-annual variations in SSS.

219

220 Recent advances in radar altimetry (e.g. Cryosat-2 and Sentinel 1 satellites and 221 sensors) are already enabling significant improvements in satellite sea-ice thickness measurements⁵⁰. Thin sea ice thickness can now also be determined from SMOS, 222 complementing altimeter estimates mostly valid for thick sea ice⁵¹. Sea ice thickness 223 224 is important for OA research as it indicates whether ice is seasonal or multi-year, 225 supporting the interpretation of carbonate parameters. Altimetry is also used to 226 measure wind speeds and increases the coverage of scatterometer estimates in polar 227 regions. It provides higher-resolution (along track) estimates of surface wind stress, 228 which can potentially be used to indicate regions of upwelling. Wind-driven 229 upwelling causes dense cooler water (with higher concentrations of CO₂ and thus more acidic) to be drawn up from depth to the ocean surface. This upwelling can have 230 significant impacts on local OA and ecosystems ^{4, 52}, especially at eastern oceanic 231 boundaries 53, 54. 232

233

It is important to emphasise that the use of Earth observation data to derive carbonate parameters should not be seen as a replacement for *in situ* measurement campaigns, especially due to the current reliance on empirical and regional algorithms. Earth observation algorithms need calibration and validation with *in situ* data such as those taken by GOA-ON, and if the carbonate system response changes over time, empirical and regional algorithms tuned to previous conditions may become less reliable.

- 241 **4.2** Algorithms for estimating carbonate parameters
- 242

The four key OA parameters (pCO₂, DIC, TA, pH) are largely driven by temperature, 243 244 salinity and biological activity, allowing empirical relationships to be developed using 245 in situ measurements of OA parameters. Table 2 shows a range of published 246 algorithms based on such relationships, while Figure 1 shows their geographical 247 coverage. Both illustrate that most of the literature has focused on the northern basins 248 of the Pacific and Atlantic and the Arctic, especially the Barents Sea, with all other regions only attracting algorithms for a single parameter or none at all.⁵⁵ 249

250

Parameter	Dependencies	Region and references
	SST	Global ⁵⁶ , Barents Sea ⁵⁷
	SST, SSS	Barents Sea ⁵⁸ , Caribbean ¹⁴
pCO ₂	SST, Chl	N Pacific ⁵⁹
p = 0 2	SSS, Chl	North Sea ⁶⁰
	SST, SSS, Chl	N Pacific ⁶¹
	SST, Chl, MLD	Barents Sea ⁶²
	SSS	Barents Sea ⁵⁷
ТА	SST, SSS	Global ^{18, 63} , Arctic ¹⁵
	SSS, nitrate	Global ⁵⁵
DIC	SST, SSS	Equatorial pacific ⁶⁴
	SST, SSS, Chl	Arctic ¹⁵
рН	SST, Chl	N Pacific ¹⁰

252 Table 2. Example regional algorithms for each carbonate parameter illustrating the 253 variable dependencies. Chl is chlorophyll-a and MLD is mixed layer depth.

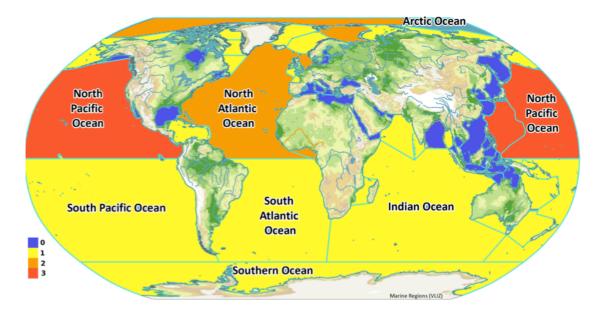


Figure 1. The number of key carbonate parameters (f_{CO2} or p_{CO2} , TA, DIC, pH) for which regional algorithms exist in the literature that can be implemented using just satellite Earth observation data. Regions are indicative of open ocean areas, as implementation of algorithms in coastal areas may be problematic.

NOAA's experimental Ocean Acidification Product Suite (OAPS) is a regional 256 257 example of using empirical algorithms with a combination of climatological SSS and 258 satellite SST to provide synoptic estimates of sea surface carbonate chemistry in the Greater Caribbean Region ¹⁴. p_{CO2} and TA were derived from climatological SSS and 259 260 satellite SST, then used to calculate monthly estimates of the remaining carbonate 261 parameters, including aragonite saturation state and carbonate ion concentration. In general the derived data were in good agreement with in situ measured data (e.g. 262 mean derived TA = $2375 \pm 36 \text{ }\mu\text{mol }\text{kg}^{-1}$ compared to a mean ship-measured TA = 263 $2366 \pm 77 \text{ }\mu\text{mol kg}^{-1}$). OAPS works well in areas where chlorophyll-a is low, 264 265 however in regions of high chlorophyll-a, where net productivity is likely to perturb the carbonate system, and in areas where there are river inputs, the approach tends to underestimate aragonite saturation state, for example 21 .

268

A quite different approach is the assimilation of satellite data into ocean circulation models ⁶⁵. The model output carbonate parameters can then be used directly. This allows satellite-observed effects to be extended below the water surface, albeit with the uncertainties inherent in model data. Here we seek to assess the direct use of satellite data through empirical algorithms to improve OA estimates.

274

275 **4.3 Regions of interest for Earth observation**

276

277 Arctic Seas

It is increasingly recognised that the Polar Oceans (Arctic and Antarctic) are particularly sensitive to OA ⁶⁶. Lower alkalinity (and thus buffer capacity), enhanced warming, reduced sea-ice cover resulting in changes in the freshwater budget ⁶⁷, and nutrient limitation make it more vulnerable to future OA ⁶⁸. Retreating ice also provides increased open water for air-sea gas exchange and primary production ⁶⁹.

283

The remote nature of the Arctic Ocean provides difficulties for collecting *in situ* datasets, with limited ship, autonomous vehicle and buoy access, and *in situ* data collection during winter months is often impossible. Therefore the use of remote sensing techniques is very attractive, if sufficient *in situ* data can be found to calibrate satellite algorithms, and if the challenges of Arctic remote sensing can be overcome. These waters are very challenging regions for satellite remote sensing. For instance, low water temperatures reduce the sensitivity range of SSS sensors ⁴³, and sea ice can

291 complicate retrievals of SSS and chlorophyll- $a^{70, 71}$. Improvement in the accuracy of

292 high latitude satellite SSS is expected soon by combining observations from SMOS,

293 Aquarius and the upcoming SMAP sensor, all polar-orbiting L-band radiometers.

294

295 The Bay of Bengal

296 This region is clearly a focus of current OA research with unique characteristics due 297 to the large freshwater influence. The flow of fresh water from the Ganges Delta into 298 Bay of Bengal (42,000 m³/sec) represents the second greatest discharge source in the 299 world. Additionally, rainfall along with freshwater inputs exceeds evaporation, 300 resulting in net water gain annually in the Bay of Bengal. Collectively these provide 301 an annual positive water balance that reduces surface salinity by 3-7 g/kg compared to the adjacent Arabian Sea^{72,73}, resulting in distinctly different biogeochemical regimes 302 303 ⁷⁴. Biogeochemically, the Indian Ocean is one of the least studied and most poorly understood ocean basins in the world ⁷⁴. This is particularly true for the Bay of Bengal 304 305 where a relatively small number of hydrographic sections and underway surface 306 observations have been undertaken, despite the notable influence of freshwater on particle dynamics, air-sea carbon flux and surface carbonate chemistry ⁷⁵⁻⁷⁹. North of 307 15° S, TA increases relative to salinity⁸⁰, indicating the presence of an important land 308 309 source that can broadly affect acidification dynamics.

310

To date there is little work on acidification dynamics and air sea exchange of CO₂ in the Bay of Bengal ⁸¹⁻⁸³. In 2013, the Bay of Bengal Ocean Acidification (BOBOA) Mooring was deployed for the first time in Bay of Bengal (15°N, 90°E) by PMEL (NOAA) and the Bay of Bengal Large Marine Ecosystem Program (BOBLME). Data 315 from the buoy will improve our understanding of biogeochemical variations in the 316 open ocean environment of the Bay of Bengal.

317

It is an open question whether SSS can be used to estimate TA in the Bay of Bengal. An important step towards answering this question would be to investigate the spatial variability of the TA to salinity relationship in the region. Use of satellite SSS in the region is also challenged by heavy radio frequency interference.

322

323 The Greater Caribbean and the Amazon plume

324 The reefs in the Greater Caribbean Region are economically important to the US and Caribbean nations with an estimated annual net value of US\$3.1-4.6 billion in 2000⁸⁴. 325 326 At least two thirds of these reefs are threatened from human impacts including OA. 327 The skeleton of a coral is made of aragonite and the growth of their skeletons is reduced by OA⁶, and numerous studies have shown a net decline in coral 328 calcification (growth) rates in accordance with declining CaCO₃ saturation state ⁸⁵. 329 The waters of the Greater Caribbean region are predominantly oligotrophic and 330 similar to the subtropical gyre from which it receives most of its water ¹⁴. Whilst the 331 332 often shallow water environments of coral reefs and the plethora of small islands can 333 make it challenging for Earth observation instruments to collect reliable data, the 334 oligotrophic nature and the similarities in water type across the whole region make it 335 ideal for the development of novel products. This region therefore provides an ideal case study to develop and evaluate algorithms representative of a shallow, 336 337 oligotrophic environment.

338

339 The Amazon plume, south of the Greater Caribbean, is the largest freshwater 340 discharge source in the world (209,000 m³/sec). It can cause SSS decreases of several units many hundreds of kilometers from land, and has an area that seasonally can 341 reach 10⁶ km². These characteristics make it an ideal case study for testing and 342 evaluating remote sensing algorithms, particularly to study the space-time resolution 343 344 tradeoffs using SSS sensors.

345

346

5. Future opportunities and focus

347 The Copernicus program is a European flagship initiative, worth more than €7 348 billion, which aims to provide an operational satellite monitoring capability and related services for the environment and security ⁸⁶. The launch of the Sentinel-1A 349 350 satellite in 2014 signaled its start. Of the five Sentinel satellite types, Sentinels 2 and 3 are most appropriate for assessment of the marine carbonate system⁸⁷⁻⁸⁹. These 351 352 satellites will provide chlorophyll-a and SST with unprecedented spatial and temporal 353 coverage. The development of higher spatial resolution geostationary sensors that 354 continually monitor chlorophyll-a and SST over the same area of the Earth also holds 355 much potential for the future of OA assessment and research ⁹⁰. These satellites and 356 sensors are able to provide 10 or more observations per day, allowing the study of the 357 effect of tidal and diurnal cycles on OA. The societal importance of measuring and 358 observing the global carbon cycle was further highlighted with the launch of the 359 NASA Orbiting Carbon Observatory (OCO-2) in 2014. This satellite and its sensors 360 are designed to observe atmospheric CO₂ concentrations, but its potential for marine 361 carbon cycle and OA is likely to be a focus of future research.

363 SMOS and Aquarius have recently passed their nominal lifetimes, with SMOS now 364 extended until 2017. Based on the lifetimes of previous satellite Earth observation 365 sensors, they may well operate until the early 2020s. NASA's SMAP satellite, to be 366 launched in January 2015, should provide short-term continuity. The development of the technology and the clear importance of monitoring ocean salinity are likely to 367 368 support the development of future satellite sensors. Also, historical time series data 369 from alternative microwave sensors hold the potential for a 10+ year time series of satellite based SSS observations ⁹¹, and this sort of measurement record is likely to 370 371 extend into the future as it forms the basis of a global SSS monitoring effort.

372

373 In summary, satellite products developed up to now in the OA context have been 374 regional, empirical or derived with a limited variety of satellite datasets, rendering an 375 effort to systematically exploit remote sensing assets (capitalizing on the recent 376 advent of satellite salinity measurements) absolutely timely. To-date there is only regional application of satellite SST to address the issue of assessing OA⁶², along 377 with two non-peer-reviewed attempts to calculate carbonate system products using 378 satellite SSS data ^{92, 93}. Supported by good *in situ* measurement campaigns, especially 379 380 in places with currently poor in situ coverage such as the Arctic, satellite 381 measurements are likely to become a key element in understanding and assessing OA. 382

383 AUTHOR INFORMATION

384 Corresponding Author

385 *Peter Land, peland@pml.ac.uk

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400 BIOGRAPHICAL STATEMENT

401 Peter Land is a remote sensing scientist at Plymouth Marine Laboratory (PML), 402 specializing in atmosphere-ocean gas exchange and carbonate chemistry. Jamie 403 Shutler is an oceanographer and former European Space Agency (ESA) fellow 404 specializing in atmosphere-ocean gas exchange at the University of Exeter. Helen 405 Findlay is an oceanographer at PML specializing in ocean acidification and carbonate 406 chemistry. Fanny Girard-Ardhuin is a remote sensing scientist specializing in sea ice 407 at l'Institut Français de Recherche pour l'Exploitation de la Mer (Ifremer). Nicolas 408 Reul is a remote sensing scientist at Ifremer and member of the SMOS scientific 409 team. Jean-Francois Piolle is a computer scientist at Ifremer. Bertrand Chapron leads 410 remote sensing research at Ifremer. Yves Quilfen is an altimetry remote sensing 411 scientist at Ifremer. Joseph Salisbury and Douglas Vandemark are oceanographers at

the University of New Hampshire focusing on biogeochemistry and ecology in coastal

413 areas. Richard Bellerby is a chemical oceanographer at the Norwegian Institute for

414 Water Research, a member of the GOA-ON executive committee, and leader of the

415 AMAP and SCAR ocean acidification working groups. Punyasloke Bhadury is a

- 416 coastal ecologist at the Indian Institute of Science Education and Research-Kolkata.
- 417 Roberto Sabia is a specialist in remote sensing of salinity working for ESA.
- 418

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