Technical University of Denmark



Overcoming kinetic limitations in biocatalysis

Al-Haque, Naweed; Tufvesson, Pär; Gani, Rafiqul; Woodley, John

Publication date: 2012

Document Version Publisher's PDF, also known as Version of record

Link back to DTU Orbit

Citation (APA): Al-Haque, N., Tufvesson, P., Gani, R., & Woodley, J. (2012). Overcoming kinetic limitations in biocatalysis. Abstract from biocat 2012, Hamburg, Germany.

DTU Library

Technical Information Center of Denmark

General rights

Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

• Users may download and print one copy of any publication from the public portal for the purpose of private study or research.

- You may not further distribute the material or use it for any profit-making activity or commercial gain
- You may freely distribute the URL identifying the publication in the public portal

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Overcoming kinetic limitations in biocatalysis

Naweed Al-Haque, Pär Tufvesson, Rafiqul Gani and John M. Woodley

Department of Chemical and Biochemical Engineering, Technical University of Denmark (DTU), DK-2800, Kgs. Lyngby, Denmark; E-mail: nalh@kt.dtu.dk

The primary rationale for implementing biocatalysis is that it usually gives highly selective conversions resulting in reduced waste. However, for effective introduction of biocatalysis high product concentrations are required [1]. Usually at an industrial scale, the concentrations of the components are such that the reactions are challenged by substrate and product inhibition to the enzyme and therefore different strategies have to be applied to overcome this problem. One option is to introduce an auxiliary phase in the form of a solid resin. The resin can act as a carrier which supplies the inhibitory substrate within the critical concentration. Biocatalytic consumption of substrate from the aqueous phase causes additional substrate to diffuse from the carrier phase to accomplish *in-situ* substrate supply (ISSS) at the rate of biocatalytic demand. Similarly, as the product is formed, it diffuses back into the carrier phase to accomplish *in-situ* product into the carrier phase to accomplish *in-situ* substrate supply (ISSS) at the rate of biocatalytic demand. Similarly, as the product is formed, it diffuses back into the carrier phase to accomplish *in-situ* product removal (ISPR) [2].

In order to test this hypothesis, an ω -transaminase catalyzed reaction for the asymmetric synthesis of (S)-1-phenylethylamine was selected for experimental validation. The specific challenges that have to be addressed in this system include substrate and product inhibition and low aqueous solubility of the substrate [3]. In this study, a systematic methodology to select a resin as the auxiliary phase is presented. A broad range of resins comprising various surface areas, pore volumes and functional groups was used in experiments to quantify the partition coefficient and overall capacity of the resin for the substrate and product. With the selection of an appropriate resin, it was loaded with substrate and released into the reaction media. To make a comparison, the same reaction was carried out by adding the substrate directly to the reaction media. From the experimental findings, it can be stated that the addition of resin had a significant positive impact with respect to observed biocatalytic activity.

^{1.} Pollard D.J.; Woodley J.M. (2006). Trends Biotechnol., 25:66-73.

^{2.} Kim P-Y.; Pollard D. J.; Woodley J.M. (2007). Biotechnol. Prog., 23:74-82.

Tufvesson P.; Lima-ramos J.; Jensen SJ.; Al-Haque N.; Neto W.; Woodley JM. (2011). Biotech. Bioeng., 108: 1479 – 1493