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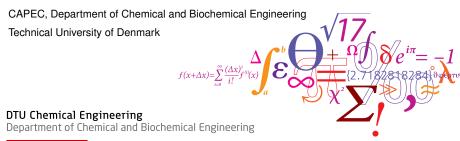
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Dynamic Effects of Diabatization in Distillation Columns

Paper 24, ACD 2012, Session no. 5, Thursday, 8 November 2012

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Presentation Outline

Motivation Introduction Methodology Dynamic Modeling Simulation Study Results Conclusion Future Work

Motivation

- Increasing focus on environmental issues and resource management
- Distillation is a widely used separation process in the industry, but it is highly energy intensive. Some key numbers for USA in 1988:
 - Distillation accounted for 4% of the total energy consumption ¹
 - Roughly 40% of the processing energy used in refining and continuous chemical processes was used for distillation
- Conventional distillation columns operate at low efficiencies ²
- Diabatic distillation columns can provide:
 - Higher efficiencies 3
 - Less utilities and thus less operation cost ⁴

¹Humphrey, J.L. et al., Separation process technology, McGraw-Hill, 1997

²de Koeijer, G. et al., 2000, Int. J. Applied Thermodynamics 3(3):105-110

³Nakaiwa, M. et al., 1998, Appl. Therm. Eng., 18(11), 1077–1087

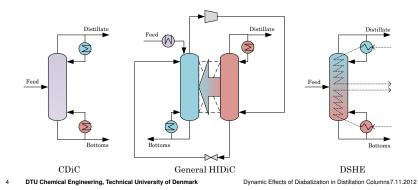
⁴Olujic, Z. et al., 2003, J. Chem. Technol. Biotechnol. 78(2-3):241-248

³ DTU Chemical Engineering, Technical University of Denmark

(1/3)

Introduction

- Adiabatic operation: Heat is added at the highest temperature (reboiler) and heat is removed at the lowest temperature (condenser), thereby preventing integration (e.g. conventional,...)
- Diabatic operation: Heat required to perform the separation is added and/or removed throughout the column (e.g. HIDiC, DSHE, ...)



Introduction

(2/3)

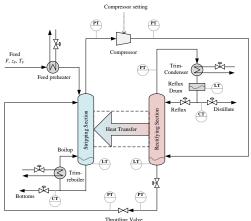
 Research example: Verification of high heat transfer performance and enhanced mass transfer performance in a pilot scale (0,8 m diameter) test unit at TU Delft ⁵



⁵http://www.ecn.nl

Introduction

The Heat-Integrated Distillation Column (HIDiC)



CT: Composition Transmitter

LT: Level Transmitter

PT: Pressure Transmitter

(3/3)

Methodology

- Formulate generic, dynamic first-principle model describing:
 - Adiabatic distillation columns (Conventional)
 - Diabatic distillation columns (HIDiC)
- Implement model in simulator (Matlab[®])
- Formulate realistic case-studies and perform numerical simulations
- Dynamic controllability analysis and compare distillation control configurations on decentralized 2×2 system (remaining perfectly controlled)
- Perform closed-loop simulations based on single-loop control strategies

Dynamic Modeling

- Model features:
 - Binary separation of benzene-toluene with **equally sized** rectifying and stripping section
 - Mass balances
 - Energy balances
 - Liquid tray hydraulics (variable liquid tray holdups)
 - One mathematical term distinguishes configuration (adiabatic, diabatic)
- Major assumptions:
 - Equilibrium stage model
 - Lewis/Randall ideal mixture
 - Constant pressure in each section
 - Negligible vapor holdup compared to liquid holdup
 - Negligible changes in sensible heat
 - Homogeneous stages (ideally mixed)

Simulation Study

- Considered distillation column configurations:
 - Conventional distillation column (CDiC)
 - Heat-Integrated Distillation Column (HIDiC)

Selected model parameters:

Parameter	Value
Top purity	99% benzene
Bottom purity	95% toluene
Number of stages	20
Feed stage location	11
Feed flow	100 kmol
Feed benzene content	0.50
Stripping section pres-	$0.1013 \mathrm{MPa}$
sure	
Hydraulic time con-	0.0042 h
stant	
Heat of vaporization	30,001.1 kmol
Relative volatility	2.317
Heat transfer rate	9,803 kJ
	11 - 15



(1/2)

Results - Open-loop

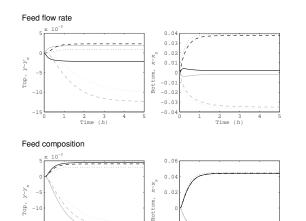
Dynamic responses to step-changes in disturbance variables.

Black curves: +5% Gray curves: -5%

Legends: Dashed lines: HIDiC

Dotted lines: CDiC

(Solid lines: HIDiC without condenser and reboiler)



-0.02

-0.04

4

Time (h)

-15

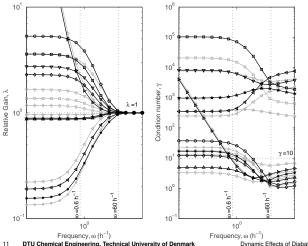
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Time (h)



Results - Open-loop

(2/2)



Controllability Analysis Dynamic RGA Analysis and SVA applied on a decentralized 2×2 : **HIDIC (black lines)** and a CDIC (gray lines).

Legends: (◦): *LV* (∗): *DV* (□): (*L*/*D*)*V*

- $(\triangle): LB \leftarrow (\bigtriangledown): L(V/B)$ $(\triangleright): DB$ $(\triangleleft): D(V/B)$
- (*): (L/D)B
- (\diamond): (L/D)(V/B)

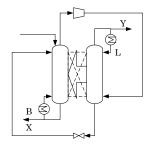
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(1/2)

Results - Closed-loop

 SISO controller based on an approximated First-order-Plus-Time-Delay model



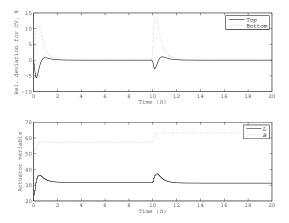
• Transfer function:

$$\begin{bmatrix} dY\\ dX \end{bmatrix} = \begin{bmatrix} \frac{-0.04972}{0.1336s+1} & \frac{-0.2566}{0.5070s+1}e^{-0.0545s} \\ \frac{-0.009427}{0.055465s+1}e^{-0.0897s} & \frac{0.3144}{0.4357s+1} \end{bmatrix} \begin{bmatrix} dL\\ dB \end{bmatrix}$$



Results - Closed-loop

(2/2)



Closed-loop responses for the HIDiC to: t = 0h: +10% step-change in feed flow rate t = 15h: -10% step-change in feed composition

Tuning parameters (From S-IMC): $dL \rightarrow dY_D$: $K_c = -0.0790, \tau_I = 0.5158h$

 $\label{eq:dB} \begin{array}{l} dB \rightarrow X_B \text{:} \\ K_c = 8.3147, \tau_I = 0.4357 \text{h} \end{array}$

Conclusion

- A flexible and generic model have been developed to describe various diabatic distillation column configurations
- Various feasible pairings exist for the HIDiC
- Diabatization causes increased coupling between input-output variables
- In this study, among others an $LB\mbox{-structure}$ seemed to be a promising control structure for a HIDiC

Future Work

- PhD Project initiated with the title **Operation and Design of Diabatic Distillation Processes**
- Develop detailed model framework for adiabatic and diabatic distillation processes
- Investigations, including:
 - Thermodynamics: Impact on complex thermodynamic modeling on e.g. dynamics
 - Operability: Abnormal operation, e.g. steady-state multiplicity
 - Control: Development of regulatory and multivariable control strategies



End

Thank you for your attention