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## Hydrogen dynamics in bulk sodium alanates

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Since the discovery of Ti-doped NaAlH<sub>4</sub> as a possible candidate for reversible solid state hydrogen storage [1], a lot of effort has been made to investigate and understand the decomposition processes and kinetics of NaAlH<sub>4</sub> and Na<sub>3</sub>AlH<sub>6</sub>, an intermediate product in the desorption processes. Of particular interest is the effect of the Ti-doping on cyclic hydrogen discharging and recharging [2], being essential for mobile hydrogen storage applications.

In this contribution, we present results of density-functional calculations [3] for bulk NaAlH<sub>4</sub> and Na<sub>3</sub>AlH<sub>6</sub>. We have determined activation energies for hydrogen diffusion and rotation of AlH<sub>4</sub><sup>-</sup> and AlH<sub>6</sub><sup>3-</sup> anions, respectively. We furthermore provide results for bulk substitutions of Ti at Al-sites.

[1] Bogdanović, Schwickardi, J. Alloys Compd. 253 (1997)

[2] Bogdanović *et al.*, J. Alloys Compd. 350 (2003)

[3] Dacapo pseudopotential code. URL <http://www.camp.dtu.dk/campos>  
Hammer, Hansen, Nørskov, PRB 59, 7413 (1999)