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Production and characterisation of glycoside hydrolases from GH3, GH5, GH10, GH11 and GH61 for chemo-enzymatic synthesis of xylo- and mannooligosaccharides

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Production and Characterisation of Glycoside Hydrolases from GH3, GH5, GH10, GH11 and GH61 for Chemo-Enzymatic Synthesis of Xylo- and Mannooligosaccharides



Adiphol Dilokpimol Ph.D. Thesis June 2010

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Enzyme and Protein Chemistry Department of Systems Biology

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Adiphol Dilokpimol Ph.D. Thesis June 2010 Enzyme and Protein Chemistry Department of Systems Biology Technical University of Denmark

DTU

Preface

The present Ph.D. thesis entitled "Production and characterisation of glycoside hydrolases from GH3, GH5, GH10, GH11 and GH61 for chemo-enzymatic synthesis of xylo- and mannooligosaccharides" summarises my work carried out under the supervision of Prof. Birte Svensson and Assoc. Prof. Maher Abou Hachem at Enzyme and Protein Chemistry (EPC), Department of Systems Biology, Technical University of Denmark (DTU) for the period of June 2007 to June 2010. This study was supported by the Danish Strategic Research Council's Committee on Food and Health (FøSu, to the project 'Biological Production of Dietary Fibres and Prebiotics' no. 2101-06-0067).

This work has resulted in the following manuscripts:

Dilokpimol, A., Nakai, H., Gotfredsen, C. H., Appeldoorn, M., Baumann, M. J., Nakai, N., Schols, H. A., Abou Hachem, M., Svensson, B. Enzymatic synthesis of β -xylosyloligosaccharides by transxylosylation using two β -xylosidases of glycoside hydrolase family 3 from *Aspergillus nidulans* FGSC A4, submitted to *Carbohydrate Research* (see 10.2.3.1)

Dilokpimol, A., Nakai, H., Gotfredsen, C. H., Baumann, M. J., Nakai, N., Abou Hachem, M., Svensson, B. GH5 endo-β-1,4-mannanases from *Aspergillus nidulans* FGSC A4 producing hetero manno-oligosaccharides by transglycosylation, to be submitted to *Biochimica et Biophysica Acta, Proteins and Proteomics* (see 10.2.3.2)

In addition, I am the coauthor of several publications and manuscripts (listed in 10.2) on results obtained in the same project.

Kgs. Lyngby, 3rd June 2010

Adiphol Dilokpimol

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Summary

Plant cell wall hydrolysate by-products are resources for oligosaccharides which potentially can act as prebiotics stimulating the growth of probiotic bacteria and thus provide a number of health benefits to the host. The objective of the present research project is to produce novel prebiotics or biologically active oligosaccharides by enzyme catalysed transglycosylation of starting materials that are related to or derived from industrial plant by-products. Five recombinant glycoside hydrolases, *i.e.* one GH3 β xylosidase, two GH5 mannanases, one GH10 xylanase, and one GH11 xylanase from *Aspergillus nidulans* FGSC A4 were investigated for their transglycosylation activity aiming at production of β -xylo- and mannooligosaccharides. The project includes characterisation of the produced enzymes.

The GH3 β-xylosidase (BxlA) was produced as a His-tag fusion protein secreted by *Pichia pastoris* X-33 in a yield of 16 mg/L culture. BxIA displayed a K_m and k_{cat} towards *para*-nitrophenyl β -D-xylopyranoside of 1.3 mM and 112 s⁻¹, respectively, and also hydrolysed *para*-nitrophenyl α -L-arabinofuranoside albiet with two orders of magnitude efficiency $(k_{\text{cat}}/K_{\text{m}})$. xylooligosaccharides lower catalytic Among (degree of polymerisation 2–6), BxlA preferably hydrolysed xylobiose, while the catalytic efficiency decreased slightly with increasing chain length. Transglycosylation reactions using 10 mono- and six disaccharides, two sugar alcohols, and two amino acids revealed that BxIA possesses broad acceptor specificity, while mannose, lyxose, and talose are preferred acceptors for BxlA with transglycosylation yields of 25%, 23%, and 22%, respectively. Moreover, four di- and two trisaccharides/glycosides produced with lyxose, L-fucose, talose, cysteine, sucrose, and turanose as acceptors are novel compounds and the structures of β -D-Xylp-(1 \rightarrow 4)-D-Lyxp, β -D-Xylp-(1 \rightarrow 4)- α -D-Glcp-(1 \rightarrow 3)- β -D-Fruf, β -D- $Xylp-(1\rightarrow 4)-\alpha$ -D-Glcp-(1 $\rightarrow 2$)- β -D-Fruf, and β -D-Xylp-(1 \rightarrow 6)- β -D-Fruf-(2 \rightarrow 1)- α -D-Glcp, have been assigned by NMR and their molecular masses have been determined by ESI-MS. Glycosynthase mutants of BxlA (D307G, D307A, D307S, D307C) showed low glycosynthase activity using α -xylosyl fluoride as donor with different acceptors and were not suitable for synthesis of xylosyl-oligosaccharides. The roles of Cys308 and Asn313 at subsite +1 of BxlA in transglycosylation were evaluated using subsite +1 mutants (C308W and N313R) created by site-directed mutagenesis. C308W essentially lost activity indicating that Cys308 is crucial for the activity of BxlA, whereas the transglycosylation yields catalysed by N313R were increased by 50–150% using hexoses and maltose as acceptors.

The GH5 mannanases (ManA and ManC, sharing 39% sequence identity) were produced recombinantly as His-tag fusion proteins by *P. pastoris* X-33 in a yield of 120 and 145 mg/L culture, respectively. ManA and ManC hydrolysed mannooligosaccharides (degree of polymerisation 4–6) with highest k_{cat} of 193 and 134 s⁻¹ towards mannohexaose with K_m of 1.8 and 0.6 mM, respectively, but showed weak affinity towards mannotetraose. ManC had higher k_{cat}/K_m than ManA, while mannobiose was the main hydrolysis products from both ManA and ManC. In addition, ManC showed 30–80% higher activity towards konjac glucomannan, guar galactomannan, and locust bean gum galactomannan and possessed 8-fold higher transglycosylation activity than ManA. The maximum transglycosylation yield of mannopentaose and mannohexaose using ManC with 24 mM mannotetraose as donor and 100 mM mannotriose as acceptor was 7.13 mg/mL. Furthermore, in order to investigate the potential role of Trp283 at subsite +1 of ManC, which is equivalent to Ser289 in ManA, the subsite +1 mutants ManCW283S and ManAS289W were made. The subsite +1 mutants have lower hydrolytic activity than the wild-type enzymes. ManAS289W showed lower K_m and the transglycosylation activity increased 50% compared to ManA wild-type, whereas ManCW283S showed higher K_m and the transglycosylation activity decreased 30–45% compared to ManC wild-type, indicating that Trp283 in ManC has some impact on the activity of mannanases, but is not a crucial residue in controlling transglycosylation and hydrolysis. Transglycosylation reactions using 12 mono-, 15 di-, and 9 trisaccharides revealed that ManA and ManC have relatively strict acceptor specificity. Apart from mannotriose, ManC can use isomaltotriose and melezitose as acceptors, but ManA accepts only mannotriose. Two penta- and two hexasaccharides were novel compounds produced by ManC using isomaltotriose and melezitose (α -D-Glc*p*-(1 \rightarrow 3)- β -D-Fur*f*-(2 \rightarrow 1)- α -D-Glc*p*) as acceptors.

The GH10 (XlnC) and GH11 (XlnA) xylanases from *A. nidulans* FGSC A4 were recloned from *P. pastoris* expression host acquired from Fungal Genetics Stock Center (Kansas City, MO) and produced in *Escherichia coli* BL21(DE3) as His-tagged fusion proteins in a yield of 95 and 90 mg/L culture, respectively. XlnA showed higher activity than XlnC towards birchwood xylan, oat spelt arabinoxylan and wheat arabinoxylan with specific activity towards birchwood xylan of 330 and 280 U/mg, respectively. Transglycosylation using six disaccharides and two sugar alcohols revealed that XlnA has very strict acceptor specificity and can use only xylobiose as acceptor. XlnC has also rather narrow acceptor specificity, but can in addition to xylobiose use cellobiose, xylitol, and sorbitol as acceptors with transglycosylation yields of 3%, 22%, and 16%, respectively. The glycosynthase activity using α -xylobiosyl fluoride as donor and the oligosaccharide products, moreover, were degraded by the trace hydrolytic activity of the E244A mutant.

The last part of the study was to perform enzyme discovery by applying a proteomics strategy. Previously, extracellular proteins from culture supernatant of A. nidulans FGSC A4 grown on 10 polysaccharides were separated by two-dimensional gel electrophoresis and identified after trypsin digestion by matrix-assisted laser desorption/ionization mass spectrometry (unpublished work by Kenji Maeda and Hiroyuki Nakai, Enzyme and Protein Chemistry, DTU Systems Biology). Among several potential novel carbohydrate active enzymes detected by this strategy, a GH61 putative endo- β -glucanase (EglF) was chosen for further investigation. The *eglF* gene was obtained by mRNA isolation and cDNA cloning from a cell culture of A. nidulans FGSC A4 using barley β-glucan as substrate and cloned into *P. pastoris* X-33. EglF was produced recombinantly as His-tagged protein in a yield of 66 mg/L culture. EglF showed very low hydrolytic activity towards barley β -glucan and oat spelt arabinoxylan with specific activity of 180 and 145 μ U/mg, respectively, and no activity and binding were found by screening using carbohydrate microarrays (collaboration with Henriette Lodberg Pedersen and William G. T. Willats, University of Copenhagen). Recently, other GH61 proteins were proposed to be cellulase-enhancing factors, and the prepared EglF could be useful for further investigations on how GH61 proteins enhance cellulase activity.

Dansk resumé

Biprodukter fra hydrolyse af plantecellevægge er kilder til oligosakkarider, som potentielt kan fungere som prebiotika ved at stimulere væksten af probiotiske bakterier, og dermed medføre en række sundhedsmæssige fordele for værten. Formålet med dette forskningsprojektet er, at producere nye prebiotika eller biologisk aktive oligosakkarider ved enzymkatalyseret transglykosylering af materialer, der er relateret til, eller stammer fra industrielle biprodukter. Fem forskellige rekombinante glykosidhydrolaser, dvs. en GH3 β -xylosidase, to GH5 mannanaser, en GH10 xylanase, og en GH11 xylanase fra *Aspergillus nidulans* FGSC A4, blev undersøgt for deres transglykosyleringsaktivitet med det formål at producere β -xylo- og mannooligosakkarider. Projektet omfatter karakterisering af de producerede enzymer.

GH3 β-xylosidase (BxlA) blev fremstillet som et His-tag-fusionsprotein, der udskilles af Pichia pastoris X-33, med et udbytte på 16 mg/L kultur. Km og kcat for paranitrofenyl β -D-xylopyranosid var henholdsvis 1,3 mM og 112 s⁻¹, og BxlA hydrolyserede også para-nitrofenyl α-L-arabinofuranosid med to gange lavere katalytisk effektivitet (k_{cat}/K_m) . Blandt xylooligosakkarider (polymeringsgrad 2–6), hydrolyserede BxlA hovedsageligt xylobiose, mens den katalytiske effektivitet faldt en smule ved længere xylooligosakkarider. Transglykosyleringsreaktioner med 10 mono- og sex disakkarider, to sukkeralkoholer, og to aminosyrer viste, at BxlA har bred acceptor-specificitet, mens mannose, lyxose, og talose er foretrukne acceptorer for BxlA med transglykosyleringsudbytter på henholdsvis 25%, 23%, og 22%. Desuden, er fire di- og to trisakkarider produceret med lyxose, L-fukose, talose, cystein, sakkarose, og turanose som acceptorer, nye forbindelser, og strukturerne af β -D-Xylp-(1 \rightarrow 4)-D-Lyxp, β -D-Xylp- $(1\rightarrow 4)-\alpha$ -D-Glcp- $(1\rightarrow 3)-\beta$ -D-Fruf, β -D-Xylp- $(1\rightarrow 4)-\alpha$ -D-Glcp- $(1\rightarrow 2)-\beta$ -D-Fruf, og β -D- $Xvlp-(1\rightarrow 6)-\beta$ -D-Fruf-(2 \rightarrow 1)- α -D-Glcp, er blevet bestemt med NMR og deres molekylvægt er blevet bestemt med ESI-MS. Glycosynthase mutanter af BxIA (D307G, D307A, D307S, D307C) viste lav glycosynthase aktivitet ved α -xylosylfluorid som donor med forskellige acceptorer, og var ikke egnede til syntese af xylosyloligosakkarider. Rollerne for Cys308 og Asn313 på subsite +1 af BxlA for transglykosylering blev evalueret ved brug af subsite +1 mutanter (C308W og N313R) lavet ved site-directed mutagenese. C308W mistede næsten al aktivitet, hvilket indikerer, at Cys308 er afgørende for BxIA, hvorimod udbyttet af transglykosyleringen katalyseret af N313R blev øget med 50–150% ved brug af hexose og maltose som acceptorer.

GH5 mannanaserne (ManA og ManC, med 39% identisk sekvensdata) er blevet fremstillet rekombinant som His-tag-fusionsproteiner af Pichia pastoris X-33 med et udbytte på henholdsvis 120 og 145 mg/L kultur. ManA og ManC hydrolyserede mannooligosakkarider (polymerisationsgrad 4–6) med højeste k_{cat} på henholdsvis 193 og 134 s⁻¹ for mannohexaose med en $K_{\rm m}$ værdi på henholdsvis 1,8 og 0,6 mM, men viste svag affinitet overfor mannotetraose. ManC havde højere katalytiskeffektivitet (k_{cat}/K_m) end ManA, mens mannobiose var det primære hydrolyseprodukt fra både ManA og ManC. Desuden viste ManC 30-80% højere aktivitet overfor konjac glucomannan, guar galactomannan, og johannesbrødkernemel galaktomannan og havde samtidig 8 gange transglykosyleringsaktivitet end maksimale høiere ManA. den og transglykosyleringsproduktion var 7,13 mg/ml mannopentaose og mannohexaose ved 24 mM mannotetraose som donor, og 100 mM mannotriose som acceptor. Med henblik på at undersøge hvilken indfyldelse Trp283 ved subsite +1 af ManC, som svarer til Ser289 i ManA, blev subsite +1 mutanterne, ManCW283S og ManAS289W, fremstillet. Subsite +1 mutanterne har lavere hydrolytisk aktivitet end vildtype enzymet. ManAS289W viste lavere K_m og transglykosyleringsaktiviteten steg 50% i forhold til vildtype ManA, mens ManCW283S viste højere K_m og transglykosyleringsaktiviteten faldt 30–45% i forhold til vildtype ManC. Det indikerer, at Trp283 i ManC har en indvirkning på aktiviteten af mannanaser, men er ikke en afgørende aminosyre for kontrollen af transglykosylering og hydrolyse. Transglykosyleringsreaktioner ved anvendelse af 12 mono-, 15 di-og 9 trisakkarider viste, at ManA og ManC har streng acceptor specificitet. ManA accepterer kun mannotriose, mens ManC kan bruge både mannotriose, isomaltotriose og melezitose som acceptorer. To nye penta- og to hexasakkarid-forbindelser blev produceret af ManC ved hjælp af isomaltotriose og melezitose (α -D-Glcp-(1 \rightarrow 3)- β -D-Fruf-(2 \rightarrow 1)- α -D-Glcp) som acceptorer.

GH10 (XlnC) og GH11 (XlnA) xylanaserne blev genklonet fra *P. pastoris* kloner leveret af Fungal Genetics Stock Center (Kansas City, MO) og produceret i *Escherichia coli* BL21(DE3) som His-tag-fusionsproteiner med et udbytte på henholdsvis 95 og 90 mg/L kultur. XlnA viste højere aktivitet end XlnC mod birketræs-xylan, havre-spelt arabinoxylan og hvede arabinoxylan med specifik aktivitet på henholdsvis 330 og 280 U/mg for birketræs-xylan. Transglykosylering ved brug af sex disakkarider og to sukkeralkoholer viste, at XlnA har meget streng acceptorspecificitet, og kun kan bruge xylobiose som acceptor. XlnC har også ret streng acceptorspecificitet, men kan ud over xylobiose også bruge cellobiose, xylitol og sorbitol som acceptorer, med transglycosylationsudbytter på henholdsvis 3%, 22% og 16%. XlnC glycosynthase mutanten (E244A) var ikke velfungerende, da den viste lav glycosynthase-aktivitet ved brug af α -xylobiosyl-fluorid som donor, og det producerede oligosakkarid var blevet nedbrudt af E244A mutantens sporadiske hydrolytiske aktivitet.

Den sidste del af undersøgelsen var at udføre enzym-detektion ved hjælp af en proteomics-strategi. Tidligere er ekstracellulære proteiner fra kultur-supernatant af A. nidulans FGSC A4, dvrket på 10 polysaccharider, blevet adskilt af 2-dimensional gel elektroforese, og identificeret efter trypsinfordøjelse af matrix-assisteret laser desorption/ioniseringsmassespektrometri (ikke-udgivet værk af Kenji Maeda og Hirovuki Nakai, Enzyme and Protein Chemistry, DTU Systembiologi). Blandt flere potentielle nye kulhydrataktive enzymer opdaget ved brug af denne strategi, var en formodet GH61 endo-β-glucanase (EglF), som blev udvalgt til nærmere undersøgelse. EglF-genet blev fremskaffet ved mRNA isolering og cDNA kloning fra en cellekultur af A. nidulans FGSC A4 ved hiælp af byg β-glucan som substrat, og klonet ind i P. pastoris X-33. EglF blev produceret rekombinant som His-tag-protein med et udbytte på 66 mg/L kultur. EglF viste meget lav hydrolytisk aktivitet mod byg ß-glucan og havre-spelt arabinoxylaner med en specifik aktivitet på henholdsvis 180 og 145 µU/mg, og ingen aktivitet eller bindinger blev fundet ved screening med kulhydratmicroarrays (samarbejde med Henriette Lodberg Pedersen og William G. T. Willats, Københavns Universitet). For nyligt blev der foreslået andre GH61 proteiner som cellulase-styrkende faktorer, og den preparerede EglF kunne være brugbar for yderligere undersøgelser af, hvordan GH61 proteiner øger cellulase-aktivitet.

Abbreviations

2DE:	Two-dimensional gel electrophoresis
2D NMR:	Two-dimensional nuclear magnetic resonance spectroscopy
3D:	Three-dimensional
ACN:	Acetonitrile
Araf:	α-L-Arabinofuranoside
Arap:	α -L-Arabinopyranoside
BMGY:	Buffered glycerol-complex medium
BMM:	Buffered minimal methanol medium
BMMY:	Buffered methanol-complex medium
CAZy:	Carbohydrate-Active EnZymes Database
CBM:	Carbohydrate binding module family (families)
CCNFSDU:	Codex Commission on Nutrition and Foods for Special Dietary
	Uses
CE:	Combinatorial extension
Cys:	L-cysteine
DP:	Degree of polymerisation
ESI-MS:	Electrospray ionization mass spectrometry
ESI-Q-TOF MS:	Electrospray quadrupole time-of-flight mass spectrometry
FAO:	Food and Agriculture Organization of the United Nations
Gal:	α -D-Galactose / galactopyranoside
GalMan:	Galactomannan
GH:	Glycoside hydrolase family (families)
Glc:	β-D-Glucose / glucopyranoside
GlcA:	α -D-Glucuronic acid / glucuronopyranoside
GlcMan [.]	Glucomannan
GT [.]	Glycosyltransferase family (families)
HEPES.	4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid
HPAEC-PAD	High performance anion exchange chromatography with pulsed
	amperometric detection
HPLC:	High performance liquid chromatography
IEF:	Isoelectric focusing
IMAC:	Immobilized metal ion affinity chromatography
IOM:	Institute of Medicine of the National Academies
Iso3:	Isomaltotriose
IUBMB:	International Union of Biochemistry and Molecular Biology
LB:	Luria Bertani (medium)
LBG:	Locust bean gum
M2-M6:	Mannobiose-mannohexaose
mAb:	Monoclonal antibody
MALDI-TOF MS:	Matrix assisted laser desorption/ionisation time-of-flight mass
Mon	R D Mannaga / mannany rangoida
Iviall. MaClaA:	p-D-mannost / mannopyranoside
MedicA:	4-O-ivieinyi-α-D-giucuronopyranoside
MES:	Weiezitose
MD.	2-(<i>iv</i> -morpholino) enanesuitonic acid
	Low fat milk powder
NB1/BCIP:	initio blue tetrazolium and 5-bromo-4-chloro-3-indolyl phosphate

Nuclear magnetic resonance spectroscopy
Nominal molecular weight limit
Optical density
Phosphate buffered saline
Protein Data Bank
para-Nitrophenol
<i>para</i> -Nitrophenyl α-L-arabinofuranoside
<i>para</i> -Nitrophenyl α-L-arabinopyranoside
$pNP \beta$ -D-cellobioside
para-Nitrophenyl β-D-xylopyranoside
pixels per inch
Sodium dodecyl sulfate polyacrylamide gel electrophoresis
L-serine
Triosephosphate isomerase
United States Department of Agriculture
Xylobiose-xylohexaose
β-D-Xylose / xylopyranoside
α-Xylobiosyl fluoride
α-Xylosyl fluoride

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1. The overall objective of the thesis project

To produce novel prebiotic candidates or biologically active oligosaccharides by enzymes catalysed transglycosylation of starting materials that are related to or derived from industrial plant by-products.

2. Introduction

2.1 Cell wall polysaccharides as source of dietary fibres and prebiotics

2.1.1 Plant cell wall composition

The most abundant and renewable source of biopolymers in nature is the plant cell wall composed mainly of polysaccharides, *i.e.* cellulose, hemicelluloses, and pectins, which together with a phenolic polymer lignin and proteins form a complex and rigid structure (Fig. 1) (Sticklen 2008; Taiz and Zeiger, 1991; Waldron and Faulds, 2007). Cellulose, the most abundant polysaccharide in nature, consists of β -1,4 linked D-glucose units forming a linear polymeric chain that forms a flat ribbon, held together by hydrogen bonds between glucose units to form a crystalline structure, which is very stable and insoluble providing the rigidity to the plant cell wall (Taiz and Zeiger, 1991).

Hemicelluloses are heteropolysaccharides containing a variety of sugar units. Hemicelluloses are classified according to the main sugar residues: xylose, mannose, glucose, galactose, and arabinose, forming the backbone of the polymer. The major hemicelluloses are usually divided into four general groups of structurally different polysaccharide backbones: xylans, mannans, xyloglucans, and mixed-linkage β-glucans (Ebringerová, 2006). Xylans (β-1,4 linked D-xylose) are abundant in secondary cell wall of cereals, grasses, and hardwood¹. Mannans, in the forms of galactomannan (β -1.4 linked D-mannose) and glucomannan (β-1,4 linked D-glucose and D-mannose), are found mainly in seed endosperm of legumes and secondary cell walls of softwood, respectively. The structure and composition of xylans and mannans will be discussed in detail later in this chapter. Xyloglucans (β -1,4 linked D-glucose substituted with xylose) are the major building material of primary cell walls in higher plants. They play an important role in interlacing the cellulose microfibrils and have been strongly implicated in the regulation of cell wall extension (involving xyloglucan endotransglycosylase) (Vincken et al., 1997; Waldron and Faulds, 2007). The last main hemicelluoses are mixed-linkage β -(1,3-1,4)glucans or cereal β-glucans, found in cereals and grasses and located in the subaleurone and endospermic cell wall, where they associate with cellulose microfibrils during cell growth. The less abundant hemicelluloses including callose (β -1,3-glucans), galactan (β -1,3 linked D-galactose), and arabinan (α -1,5 linked L-arabinose), are also present in

2. Vascular plants

b. Seed plants (Spermatophytes):

- ii. Hardwood (Angiospermes or flowering plants):
 - Grasses and cereal (Monocotyledons: wheat, barley, rice, and corn)
 - Fruits and vegetables (Dicotyledons: pea and apple)

¹ Simple plant classification (adapted from Graham, et al., 2006) **Plant Kingdom:**

^{1.} Non-vascular plants (Bryophytes: mosses, hornworts)

a. Seedless plants (Pteridophytes: ferns)

i. Softwood (Gymnosperms: pines and cycases)

particular tissues or plant species. The main chain sugars of hemicelluloses are modified by various side groups such as methylglucuronic acid, arabinose, galactose, and acetyl residues, making hemicelluloses branched and variable in structures (Aro, et al., 2005; Ebringerová, 2006; Waldron and Faulds, 2007).

Pectins are a family of complex polysaccharides comprised of polymers containing galacturonic acid and rhamnose as backbone with substitution of additional polymer chains rich in arabinose and galactose as well as several other sugars in less abundant quantities. Pectins are divided into two different types of regions: the smooth region, homogalacturonan that comprises a linear chain of α -1,4 D-galacturonic acid residues, which can be methylated or acetylated, and the hairy region that consists of two different structures, D-xylose substituted galacturonan and rhamnogalacturonan to which long arabinan and galactan chains are linked *via* the rhamnose residues (for reviews of pectin structures see de Vries and Visser, 2003; Waldron and Faulds, 2007; Wong, 2008). Pectins and hemicelluloses are present in dicot primary walls in approximately equal amount, whereas hemicelluloses are more abundant in grasses and cereals (Table 1) (O'Neill and York, 2003).

Apart from the three major polysaccharides, the plant cell walls also contain lignin, a highly insoluble complex branched noncarbohydrate polymer of substituted phenylpropane units joined together by carbon-carbon and ether linkages forming an extensive cross-linked network within the cell wall and found in all vascular plants. It binds to cellulose fibers and hardens as well as strengthens the plant cell walls (Carpita and Gibeaut. 1993). The secondary walls of woody tissues and grasses are composed predominantly of cellulose, hemicellulose, and lignin (lignocelluloses). The cellulose fibrils are embedded in a network of hemicellulose and lignin (Fig. 1B). Cross-linking of these networks is believed to result in the elimination of water from the wall and the formation of a hydrophobic composite that limits accessibility of hydrolytic enzymes and is a major contributor to the structural characteristics of secondary walls (Carpita and Gibeaut. 1993). The function of plant cell wall is not only to protect the plant cells and maintain the cell shape, but also to play major roles in plant physiology involving with cell division, tissue extension, and support through ripening and senescence. The difference in molecular composition and structures of the plant cell wall depends on the cell, tissue, plant species, and the botanical origin. (Taiz and Zeiger, 1991; Waldron and Faulds, 2007).



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Figure 1 Plant plasma membrane and cell-wall structure. A) Cell wall containing cellulose microfibrils, hemicellulose, pectin, lignin and soluble proteins. B) Cellulose synthase enzymes are in the form of rosette complexes, which float in the plasma membrane. C) Lignification occurs in the S1, S2, and S3 layers of the cell wall (O'Neill and York, 2003)

Composition (% dry we			(% dry weig	ht) ¹		Side chain		
Polymer	Primary cell wall		Secondary cell wall		Backbone			
	Grass	Dicot	Grass	Dicot		Carbohydrate	Non-carbohydrate	
Cellulose	20-30	15-30	35-45	45-50	1,4 β-D-Glc	none	none	
Hemicellulose								
Xyloglucan	1-5	20-25	Minor	Minor	1,4 β-D-Glc	α-D-Xyl, β-D-Gal, α-L-Fuc, β-L-Araf	acetyl esters	
Mannan and Glucomannan	Minor	5-10	Minor	3-5	1,4 β-D-Man,	α-D-Gal	acetyl	
					1,4 β-D-Man-1,4 β-D-Glc			
Xylan	20-40	5	40-50	20-30	1,4 β-D-Xyl	α-L-Araf, β-L-GlcA, 4-O-Me β-D-GlcA	acetyl, feruloyl and	
							coumaryl esters	
Mixed-linkage Glucan	10-30	Absent	Minor	Absent	1,3, 1,4 β-D-Glc		none detected	
Pectins ²	5	20-35	0.1	0.1	1,4 α-D-GalA,	Not detected ^{4,5,6}	methyl and acetyl esters	
					1,4 α -D-GalA-1,2 α -L-Rha ³	β-D-Xyl, β-D-Apif ⁷	and borate diester	
Phenolic esters	1-5	Minor	0.5-1.5	Minor	feruloyl and coumaroyl esters (linked to xylans and pectins)			
Lignin	Minor	Minor	20	7-10	cross-linked racemic macromolecule (of the phenylpropanoids <i>p</i> -hydroxyphenyl, guaiacl, and syringal)			

Table 1 Polymeric components of typical dicot and grass primary and secondary cell walls

Adapted from O'Neill and York, 2003; Vogel, 2008

¹ Approximate composition, may vary depending on the plant and tissue types

² Details of pectins see de Vries and Visser, 2003; Waldron and Faulds, 2007

³ As rhamnogalacturonan I

⁴ As rhomogalacturonan

⁵ Side chain for rhamnogalacturonan I side chain: α-L-Araf, β-D-GalA, α-L-Fuc, β-D-GlcA, 4-O-Me β-D-GlcA
 ⁶ Side chain for hamnogalacturonan II side chain: α-L-Rha, β-L-Rha, β-L-Araf, α-L-Ara, β-D-Gal, α-L-AcefA, β-D-Apif, α-L-Fuc, α-D-GalA, β-D-GalA, β-D-GlcA, 2-O-Me α-L-Fuc, β-D-Kdo, 2-O-Me α-D-Xyl, and β-D-Dha

⁷ As xylogalacturonan and apiogalacturonan, respectively

2.1.2 Dietary fibres and prebiotics

As the most abundant biomass in nature, large amount of plant cell wall polysaccharide by-products or waste are generated through forestry and agricultural practices, pulp and paper industries, timber industries, and agro-industries as well as many food industries, *e.g.* sugar production, beer production, milling processing, and fruit and vegetable processing. The various biomasses are often disposed by burning or by landfill. However, the processed plant polysaccharide by-products can also be converted into different value-added bio-products including biofuel/bioethanol, organic chemicals, cheap energy sources for fermentation and improved animal feed and human nutrients (Howard et al., 2003). Among various opportunities, conversion of plant hydrolysate by-products into additives for functional foods gets increasing attention in recent years. Not only because this highly increases the by-product value, but also because these carbohydrates have functions involved with food physico-chemical properties (*e.g.* texture in canned vegetables, consistency in purees, and turbidity in fruit juice) (Fennema, 1996) and can improve human health, especially as dietary fibres (Kendall et al., 2010) and prebiotics (Mussatto and Mancilha, 2007; Swennen et al., 2006).

Codex Commision on Nutrition and Foods for Special Dietary Uses (CCNFSDU; 2008) defined dietary fibre as carbohydrate polymers with ten or more monomeric units, which are not hydrolysed by endogenous enzymes in the small intestine of human beings and belong to the following three categories: 1) edible carbohydrate polymers naturally occurring in food as consumed, 2) carbohydrate polymers, which have been obtained from food raw material by physical, enzymatic, or chemical means and which have been shown to have physiological benefit to health, and 3) synthetic carbohydrate polymers, which have been shown to exert physiological benefit to health (Cummings et al., 2009). However, United States Department of Agriculture defined dietary fiber² consisting of nondigestible carbohydrates and lignin intrinsic in plants (Institute of Medicine, 2005). Hence, plant cell wall polysaccharides and lignin are generally considered as dietary fibres, similarly to chitin and chitosan, inulin, fructans, and resistant starch, whereas nondigestible sugars and sugar alcohols are not considered as dietary fibres. According to the new definition by CCNFSDU, nondigestible oligosaccharides (degree of polymerization; DP 3-9) are no longer considered as dietary fibres. Dietary fibres have health beneficial effects including laxation (fecal bulking and softening), blood cholesterol attenuation, blood glucose attenuation, and metabolic responses to glycaemic carbohydrates (for more details in health benefit of dietary fibres see Kendall et al., 2010).

Regarding human health, certain components of plant cell wall polysaccharides also able to act as prebiotics. Prebiotics were first defined as a nondigestible food ingredient that beneficially affects the host by selectively stimulating the growth, activity, or both of one or a limited number of bacterial species (*i.e.* probiotics³) already resident in the colon (Gibson and Roberfroid, 1995). In 2007, Food and Agriculture Organization of the United Nations (FAO) defined prebiotic as a non-viable food component that confers a health benefit on the host associated with modulation of the microbiota. A food ingredient to be classified, as a prebiotic must fulfill the three following criteria: 1) neither be hydrolyzed nor absorbed in the upper part of the gastrointestinal tract; 2) be a selective substrate for one or a limited number of potentially beneficially bacteria in the

² Dietary fibres: naturally occurred in food, Functional fibres: added into food, including polydextrose and resistant dextrins, Total fibres: sum of both fibres (IOM, 2005)

³ Probiotics are live microorganisms which when administered in adequate amount confer a health benefit on the host (FAO, 2001)

colon; and 3) alter the colonic microflora to a healthier composition (Collins and Gibson, 1999; Gibson and Roberfroid, 1995; for details on requirement see FAO, 2007). The combination of pro- and prebiotics (synbiotics) improves the survival of the probiotics and several health benefits are reported in the gastrointestinal tract and the immune system, such as prevention of diarrhea, inhibition of pathogen colonization, stimulation of the immune system, and prevention of colon cancer (for more details about pro-, pre- and synbiotics see Collins and Gibson, 1999; Davis and Milner, 2009; Steed et al., 2008; Swennen et al., 2006; Ziemer and Gibson 1998).

2.1.3 Oligosaccharides as prebiotics

The first generation of functional foods⁴ was based on deliberate supplementation with vitamins and minerals. However, the last few years, the concept has moved towards and centered more on a positive effect on the gut microbiota in the form of pro- and prebiotics (Isolauri, et al., 2002; Swennen et al., 2006). Even though, the nondigestible oligosaccharides⁵ (DP 3–9) are no longer considered as dietary fibres, according to the new definition by CCNFSDU, they still possess important food physico-chemical properties and some also act as prebiotics (Table 2) (for more details on the oligosaccharide functions as food additives see Nakakuki, 2002). The majority of reported prebiotic oligo- and polysaccharides focused on fructooligosaccharides, galactooligosaccharides, and inulin, which now have the status generally regarded as safe. The rest of the oligosaccharides have been studied to varying degrees *in vitro*, in animal feeding studies, but rarely in human feeding studies (Barreteau et al., 2006; FAO, 2007).

Plant hydrolysate by-products can be considered as a resource for development of novel prebiotic oligosaccharides, which may have better functionality than those currently found on the market. Enhanced persistence of the prebiotic effect along the colon, antipathogen effects, and more targeted prebiotics, might all be possible starting from plant cell wall polysaccharides. Traditionally, the nondigestible oligosaccharides can be obtained by direct extraction from natural sources, produced by chemical processes hydrolyzing polysaccharides, by enzymatic- or by a chemoenzymatic synthesis. At present advanced knowledge on plant cell wall polysaccharides and their degrading enzymes allows development of novel prebiotics, using two strategies i) controlled enzymatic hydrolysis of polysaccharides and ii) enzymatic synthesis of oligosaccharides (Barreteau et al., 2006; Rastall and Hotchkiss, 2003). The enzymatic depolymerization and modification of plant cell walls is a complex system requiring several enzymes for different type of polymers. Several of these enzymes, mainly from fungal sources, are industrially interesting and widely used in many biotechnological applications, particularly for production of second generation biofuel (for reviews on these enzymes see Aro et al., 2005; de Vries and Visser, 2001; Howard et al., 2003; Merino and Cherry, 2007; Shallom and Shoham, 2003). The following part will introduce the detailed structure and composition of xylans and mannans, the target substrates of the present study, as well as cellulose, a related substrate. The enzymes involved will be discussed further in chapter 4.

⁴ Functional food is satisfactorily demonstrated to affect beneficially one or more target functions in the body, beyond adequate nutritional effects in a way relevant to either the state of well being and health and/or to reduction of the risk of a disease (Isolauri et al., 2002).

⁵ Nondigestible oligosaccharides can be longer up to a degree of polymerization 60 (Swennen et al., 2006)

Component	Origin/Manufacturing procedure	Backbone structure	Main linkage	Reference
Oligosaccharides				
Fructooligosaccharide	Jerusalem artichokes, chicories, onions, garlic,	(Fru) _n -Glc	β-(1,2)	Sabater-Molina et al., 2009
	leeks, asparaguses/Transglycosylation of sucrose			
Galactooligosaccharide	Human milk/Enzymatic modification of lactose	(Gal) _n -Glc	β-(1,4), β-(1,6)	Sako et al., 1999
Isomaltooligosaccharide	Honey/Enzymatic modification of starch	(Glc) _n	α-(1,6)	Kuriki et al., 1993
Lactulose, lactitol, lactosucrose	Lactose derivatives	Gal-Fru, Gal-Glc-Fru	β-(1,4)	Sako et al., 1999
Mannooligosaccharide	Guar gum, locust bean gum, konjac/Mannan-	(Man) _n , (Man-Glc) _n	β-(1,4)	Asano et al., 2004
	containing hydrolysate			
Soybean oligosaccharide	Soybean/Extract from soybean whey	(Gal) _n -Glc-Fru	α-(1,6)	Espinosa-Martos and
				Ruperez, 2006
Xylooligosaccharide	Wheat bran, bamboo shoots/	(Xyl) _n	β-(1,4)	Vázquez et al., 2000, Moure
	Xylan-containing hydrolysate			et al., 2006
Polysaccharides				
β-glucan	Bran of cereal grains, yeasts, mushrooms	(Glc) _n	β-(1,4)	Snart et al., 2006
Inulin	As in fructooligosaccharides	(Fru) _n -Glc	β-(1,2)	Van Loo et al., 1999
Pectin	Fruits and vegetables, legumes, potatoes	(GalA) _n , (GalA-Rha) _n	α-(1,4), α-(1,2)	Manderson et al., 2005
Resistant starch	Raw potatoes, green bananas, legumes/	(Glc) _n	α-(1,4)	Topping et al., 2003
	Cooked and cooled starch (retrograded)			

 Table 2 Typical oligosaccharides and polysaccharides with prebiotic effect

Adapted from Mussatto and Mancilha, 2007; Sako et al., 1999; Van Loo et al., 1999.

2.1.4 Xylans and xylooligosaccharides

2.1.4.1 Structure and composition

Xylans are the predominant component of hemicelluloses and the second most abundant polysaccharide after cellulose, amounting to one third of all renewable organic compounds in nature (Collins et al., 2005). Xylans are found in high amounts in hardwoods, cereals, and grasses (up to 50% of the dry tissue in some cereals) and show large heterogeneity depending on the botanical source. Xylans have a backbone of 1,4linked β -D-xylopyranosyl (Xyl) residues with different side chain substitutions such as α -D-glucuronopyranosyl (GlcA), 4-O-methyl- α -D-glucuronopyranosyl (MeGlcA), α -Larabinofuranosyl (Araf), O-acetyl, feruloyl, or coumaroyl residues, and they can be grouped according to their side chains (de Vries and Visser, 2001; Ebringerová, 2006).

Cereal arabinoxylans (L-arabino-D-xylans) are found mostly in starchy endosperm (flour) and outer layer (bran) of cereals, grasses, and bamboo shoot. The side chain of cereal arabinoxylans comprises Araf substituted at position O2 and/or O3 (mono- or disubstitution) of xylose residues. A particular feature of arabinoxylan is the presence of phenolic acids (mainly ferulic and p-coumaric acids) esterified at position O5 of arabinose residues (Fig. 2A) (Ebringerová, 2006). Depending on the tissue, arabinoxylan can be extracted by water (water-extractable) or by alkaline solvent (waterunextractable). This behavior depends on the differences in chemical and/or physical interactions between arabinoxylans and other cell wall components and not on the content and distribution of arabinose side chains. Water-unextractable arabinoxylans are tightly embedded in the cell wall network by interactions with neighbour arabinoxylans through ferulic acid side chain cross-links, or other components such as protein, cellulose and lignin, while water extractable arabinoxylans are only loosely bound at the cell wall surface (Frederix et al., 2004 and Iiyama et al., 1994). These arabinoxylans have different physico-chemical and functional properties. Water-unextractable arabinoxylans are highly viscous and can be oxidatively cross-linked through ester-linked ferulic acids. Water-extractable arabinoxylans have a high water holding capacity. These properties strongly dictate the functional effects of arabinoxylans in biotechnological processes. In lignified tissues of grasses and cereals (straw, stems, stalks, outer pericarp of grains), the arabinoxylans are found in the form of (D-glucurono)-L-arabino-D-xylans (MeGlcA:Xyl:Araf 3–9:10:1–10) with partial O-acetylation (Ebringerová, 2006).

Glucuronoxylans (D-glucurono-D-xylans) are found in the secondary cell wall of dicots and hardwoods, containing single side chains of GlcA and/or its 4-methyl derivative (MeGlcA) attached at O2 position of xylose residues in the backbone (Fig. 2B) of a DP of 150–200. The xylan backbone can also be acetylated at O3 position and less at O2, the presence of these acetyl groups is responsible for partial solubility of xylan in water. In softwood, xylans, next to mannans constitute a minor hemicellulose in the form of arabinoglucuronoxylan ((L-arabino)-D-glucurono-D-xylans) (Fig. 2C), which resembles hardwood glucuronoxylans with addition of Araf residues attached at position O3 of the xylan backbone instead of acetylation. The degree of MeGlcA substitution in softwood is higher than in hardwood with less DP (70–130) of the backbone (Beg et al., 2001; Ebringerová, 2006). Apart from the three major xylans, homoxylans (β -1,3 and mixed β -1,3, β -1,4 linked xyloses) are present in some marine algae and seaweed, whereas the heteroxylans (β -1,4 linked xyloses heavily substituted by a variety of single and oligosaccharide side chains) are present in cereal bran, seeds, gum exudates, and various mucilages. These heteroxylans can form highly viscous solutions behaving similarly to plant exudate gums (Beg et al., 2001; Ebringerová, 2006).

The oligosaccharides of xylan, *i.e.* xylooligosaccharides, are produced from xylan containing lignocellulosic materials by chemical methods (such as, autohydrolysis with water, steam, or diluted mineral acids), enzymatic hydrolysis, or a combination of chemical and enzymatic treatments (Vazquez et al., 2000). The sources of xylans from plant cell wall by-products include barley hulls, brewery spent grains, corncobs, corn stover, rice hulls, wheat straw, and bamboo (Moure et al., 2006) (for reviews on manufacture of xylooligosaccharides see Moure et al., 2006; Vazquez et al., 2000).



Figure 2 Structure of xylans. A) Arabinoxylan from cereals, arabinose residue can be esterified with ferulic and *p*-coumaric acid at position O5. B) Glucuronoxylan from hardwood. C) Arabinoglucuronoxylan from softwood (Ebringerová, 2006).

2.1.4.2 Application of xylans and xylooligosaccharides

In the past decades, the main use of xylan-containing plant hydrolysate by-products for animal feed, fuel, compost, and soil conditioner (Howard et al., 2003). However, with the significant functions related to human and animal health benefits, more technical developments are focusing on production and application of xylooligosaccharides. The health-effects of xylooligosaccharides are mainly related to the gastrointestinal microflora proven a prebiotic effect. Moreover, xylans have ability to prevent gastrointestinal infections, to reduce duration of diarrhea, to maintain the fecal water content, and to enhance cecal epithelial cell proliferation. As food additive, xylooligosaccharides have acceptable odour, are low-caloric, and non-carcinogenic. They also show advantages over inulin and fructooligosaccharides in term of resistance to acids and heat allowing use in low-pH juices and carbonated drinks (Vazquez et al., 2000). In the last few years, xylooligosaccharides have shown a remarkable potential for practical utilization in several areas, including pharmaceuticals, feed formulations, and agricultural application, although the primary use is still in food-related applications.

2.1.5 Mannans and mannooligosaccharides

2.1.5.1 Structure and composition

Whereas xylans are major constituents of hemicelluloses in grasses, cereals and hardwoods, the mannans (glucomannans or galactoglucomannans) are more abundant in softwoods. The mannans are classified into four groups: unsubstituted mannans, galactomannans, glucomannans, and galactoglucomannans, based on their backbone and galactose side chains (Moreira and Filho, 2008; Schröder et al., 2009).

Galactomannans (D-galacto-D-mannans) and unsubstituted mannans (D-mannans) consist of water-soluble 1,4 linked β -D-mannopyranosyl (Man) residues with side chains of single 1,6 linked α -D-galactopyranosyl (Gal) groups attached along the chain of galactomannans (Fig. 3A). They are found predominantly in seed endosperms of legumes, such as guars, taras, and carobs (locust beans). In coffee seed arabinosyl residues are also present as side chains of mannans, attached at O6 position, whereas glucosyl residues are observed at reducing termini (Moreira and Filho, 2008). The galactose side chains are hydrophilic and increased degree of substitution result in higher solubility in water. The water retention makes galactomannans widely used as gelling agent or gum (Man:Gal of guar gum, tara gum, locust bean gum is approximately 2:1, 3:1, and 4:1, respectively). Pure mannans are defined to consist of at least 90% of linear mannan backbone with up to 10% substitution with single galactose residues.

Glucomannans (D-gluco-D-mannans) and galactoglucomannans ((D-galacto)-D-gluco-D-mannans) (Fig. 3B) are the major hemicelluloses in secondary cell walls of softwood. The backbone of glucomannans contain randomly arranged 1,4 β -Man and 1,4 linked β -D-glucopyranosyl (Glc) residues with a DP greater than 200 in softwood. Like galactomannans, galactose side chains are present in glucomannans substituting at O6 of the mannan backbone with Man:Gal:Glc ratio of 3:1:0.1. Galactoglucomannans are glucomannans with higher substitution and Man:Gal:Glc ratio (3:1:1) and *O*-acetyl groups can substitute at O2 or O3 of the mannan backbone. Galactoglucomannan solubility in water is due to the content of galactose side chains that prevent the intermolecular interaction between backbone molecules and also act as flexible groups that can provide noncovalent connecting bridges with water and other cell wall polysaccharides (Dhawan and Kaur, 2007; Moreira and Filho, 2008; Waldron and Faulds, 2007).



Figure 3 Structure of mannans. A) Galactomannan from seed endosperm of legumes. B) Galactoglucomannan from secondary cell wall of softwood (Ebringerová, 2006).

2.1.5.2 Application of mannans and mannooligosaccharides

Both galactomannans and glucomannans are widely used in the food industry. As mentioned above, galactomannans are used as gelling agent or gum. Guar and locust bean gums and their derivatives cause very high viscosity in aqueous solution. As food additives, these gums are widely used in gummy desserts, cream cheese, and salad dressing, as well as ice creams to improve texture and reduce melting. Galactomannans are, moreover, used also as thickening agent in the textile industry in dyebaths in printing and dyeing of fibers, fabrics, and carpets, particularly for sharp and bright patterns. Glucomannans are water-soluble dietary fibres, which are used widely in food industry as emulsifier and thickener. Konjac, as a rich source of glucomannans, that has been used in Asia for centuries in traditional foods, such as noodles, tofu and heat stable gelled food products. Glucomannans have been developed for biodegradable or soluble edible films as well as carrier material for drug delivery (Beneke et al., 2009; Mikkonen et al., 2010). Mannooligosaccharides are mainly produced by steam treatment at high pressure of natural sources such as coffee beans (Asano et al., 2003). They are used in food and feed industries to promote gastrointestinal health and performance (as both prebiotics and nondigestible oligosaccharides) with the extra effect to increase the amount of excreted fat and reduce the blood pressure (Asano et al., 2003; Hoshino-Takao et al., 2008; Kumao et al., 2006).

2.1.6 Cellulose and cellooligosaccharides

2.1.6.1 Structure, composition, and applications

Cellulose, the major component of plant cell wall (40-45% of wood dry weight), consists of β -1,4 linked Glc units forming linear polymers. Each glucose unit rotates 180° from its neighbour and the chain contains 2000–8000 glucose units. In contrast to α -1,4 glucosidically linked starch and glycogen, cellulose is a straight chain polymer with no coiling or branching. Approximately 150 (up to 250) glucan chains are condensed by

hydrogen bonds into a crystalline structure (Taiz and Zeiger, 1991; Waldron and Faulds, 2007). These crystalline structures are impermeable to water and their high strength provides strong resistance to chemical and enzymatic attack. In addition to this crystalline structure, cellulose contains noncrystalline (amorphous) regions within the microfibrils. The relative amount of crystalline and amorphous cellulose varies depending on the origin (de Vries and Visser, 2001; Waldron and Faulds, 2007). Cellooligosaccharides are short-chain cellulose with β -1,4 linked Glc units. These oligosaccharides are usually produced by controlled hydrolysis (partial hydrolysis) of cellulosic materials, which produces a mixture of cellooligosaccharides of different DP, that can be purified from the reaction mixture. Enzymatic hydrolysis of cellulose is an alternative option for cellooligosaccharide production (Akpinar and Penner, 2007). Cellooligosaccharides are commonly used as models for cellulose and in cellulase studies. Cellooligosaccharides are not hydrolyzed or absorbed by the human gastrointestinal tract (nondigestible oligosaccharides) and may have potential as prebiotics and/or non-caloric bulking agents in food products (Akpinar and Penner, 2007).

The main usage of cellulose is for pulp and paper as well as in textiles such as cotton, linen, and other plant fibres. Many products were generated from cellulose including cellophane, rayon, nitrocellulose, methylcellulose, and carboxymethyl cellulose. As a major source of plant cell wall polysaccharide, there are enormous developments focusing on bioconversion of industrial waste containing cellulose involving a combination of physico-chemical and enzymatic reactions (for reviews on bioethanol production see Duff and Murray; 1996, Howard et al., 2003; Merino and Cherry, 2007).

In recent years, production of oligosaccharides by chemical treatment has declined; instead, the manufacture is concentrated on enzymatic reactions, which are more suitable for preparation of food additives. Degradation and modification of plant cell wall polysaccharides require a number of different enzymes, which will be discussed further in the fourth chapter.

3. Enzyme discovery

Enzymes are very specific catalysts of which only small quantities are required to perform the desired processes. The search for enzymes, which work optimally at specific conditions or discovery for novel enzymes, which provide valuable function, is challenging for many industries. Usually the enzyme screening and selection strategies are based on knowledge of the application and the physico-chemical conditions under which the enzyme must operate (Jones, 2007). The classical enzyme screening method is by screening natural isolates, based on culturing a broad diversity of plants, animals, or microorganisms, which in the latter case is the major source for industrial enzymes. The diversity should be reflected in taxonomy and phylogeny as well as physiological, biochemical and ecological variation. This classical screening has been successful including the discovery of extremophiles. However, random selection of microorganisms and screening of very large number of samples, which may also be redundant are the major problems of this approach.

By the aid of bioinformatics, molecular biology, and protein chemistry, several integrated approaches are developed involving molecular screening, genomic screening, environmental gene screening, and proteomics screening (Jones, 2007). Molecular screening, searching for homologous enzymes, takes the advantage of similarity of the genetic sequence, which can be used for prediction of protein function of the encoded enzyme. By using this information, a nucleotide primer probe can be designed and the

recombinant enzyme can be cloned and produced by using an appropriate expression system. The target homologous enzyme may have significantly different properties, enabling screening and cloning as an approach for improvement of pH or temperature stability. Several techniques, such as random shotgun (Staden, 1979) and nanopore technology (Kasianowicz et al., 1996), have been applied for the full genome sequencing. When the genome sequence is available, protein coding genes can be identified computationally either by *ab initio* gene prediction or by similarity search (Korf, 2004). The latter method is more reliable but depends on the availability of homologous genes. Even though molecular screening and genomic screening are very powerful techniques, they appear to be limited by the ability to grow target organisms and extract suitable DNA. Environmental gene screening applied several techniques to extract the DNA or RNA directly from microbial communities from the environment (Lorenz and Schleper, 2002). Although, extraction of the whole chromosomal DNA from the natural samples rarely gives a full-length genome, particularly in eukaryotes, isolation of total mRNA from a eukaryotic cell and cDNA library (Sambrook and Russell, 2001) can be done. The clones can then be screened for new enzyme-coding genes by techniques similar to those applied for classical screening without prior knowledge of homologous sequences. Proteomic screening approaches on the direct identification of protein function. Several separation and analysis techniques have been developed for the large-scale characterization of protein expression. Protein separation by two-dimensional gel electrophoresis (2DE) followed by identification by mass spectrometry is routinely utilized for this approach. An alternative to 2DE uses liquid chromatography after differential labeling of two protein mixtures with related isotope-coded affinity tags (Lai et al., 2004; Pohl, 2004; Uttamchandani et al., 2009). This approach can also make feasible to screen expression libraries of natural diversity (Jones, 2007).

4. Hydrolysis, phosphorolysis, and transglycosylation mechanisms

The enzymes catalyzing hydrolysis of glycosidic linkages are called glycoside hydrolases (glycosidases) and act as exo- (cleaving from an end, usually the non-reducing end) or endo- (cleaving interior bond in the substrate chain) glycosidases. These enzymes are involved in a series of important biological events from energy metabolism by degradation of carbohydrates to expression regulation as well as microbial defense mechanisms. Plant cell wall polysaccharides are chemically and physico-chemically highly complex and enzymatic biodegradation of plant cell walls requires a mixture of several endo- and exo-glycoside hydrolases as well as esterases, which are found in many microorganisms particularly in filamentous fungi (Collins et al., 2005; de Vries and Visser, 2001; Polizeli et al., 2005). Glycoside hydrolases are classified into EC 3.2.1.x⁶ as enzymes acting on O- or S-glycosides, according to the International Union of Biochemistry and Molecular Biology (IUBMB) (Webb, 1992). The IUBMB enzyme nomenclature is based on the specificity of the enzyme and does not reflect its structural and mechanistic features. Different enzymes catalyzing the same reaction have the same EC number and certain enzymes catalyzing several reactions belong to more than one class. For structural and mechanistic enzymology, classification is made in the Carbohydrate-Active EnZymes Database⁷ (CAZy; http://www.cazy.org; Cantarel et al.,

⁶ The Enzyme Commission number (EC number) is a numerical classification scheme for enzymes (Webb 1992).

⁷ CAZy database describes the families of structurally-related catalytic and carbohydrate-binding modules (or functional domains) of enzymes that degrade, modify, or create glycosidic bonds, which covers four of

2009). CAZy classification is based on amino acid sequence similarities. The sequencebased classification allows predicting the mechanism (retaining/inverting, see below), the active site residues and possible substrates as well as the function for newly sequenced enzymes for which function has not been biochemically demonstrated. CAZy classifies enzymes into glycoside hydrolase families (GH) clustered into clans reflecting conserved folding (Cantarel et al., 2009; Henrissat, 1991; Henrissat et al., 1995).

The hydrolytic mechanisms of glycoside hydrolases mostly follow general acidbase catalysis performed by two catalytic residues, typically glutamic acid or aspartic acid: an acid/base proton donor and a nucleophile (Jakeman and Withers, 2002). Depending on the spatial position of the catalytic residues, there are two reaction mechanisms resulting in either retention or inversion of the substrate anomeric configuration in the product. Retaining glycoside hydrolases, hydrolyze with net retention of configuration by a double-displacement or two-step mechanism involving a covalent glycosyl-enzyme intermediate with an oxocarbenium ion-like transition state (Fig. 4A). The two catalytic carboxylates positioned on opposite sides of the sugar ring plane catalyse this reaction by first (glycosylation step) protonating the glycosidic oxygen and, at the same time, the second carboxylic residue acts as nucleophile attacking the anomeric centre to cleave the glycosidic bond and form a glycosyl-enzyme intermediate (Fig. 4A). In the second step (deglycosylation), the glycosyl-enzyme intermediate is hydrolyzed by water. The first catalytic residue acts as a base to deprotonate the incoming nucleophile (a water molecule) as it attacks the intermediate resulting in release of product from the enzyme. Different from the retaining mechanism is found a one-step or singledisplacement reaction with attack of water at the same time occurs in the inverting mechanism, which involves two catalytic carboxylates acting as acid and base catalysts resulting in product with opposite anomeric configuration of the substrate (Fig. 4B). The distance between the two carboxylates is approximately 5.5 Å for retaining and 6–11Å for inverting glycoside hydrolases, since the inverting mechanism requires accommodating a water molecule next to the sugar rings to perform the hydrolysis (Davies and Henrissat, 1995; Henrissat and Davies, 1997).

Phosphorolysis is a reaction that cleaves a glycosidic bond by attack of a phosphate group, analogous to hydrolysis, however, phosphorolysis is reversible in contrast to hydrolysis (Berg et al., 2002). The enzymes catalyzing phosphorolysis are called phosphorylases. Several phosphorylases are reported to act on glycogen, sucrose, maltose, cellobiose, cellodextrin, trehalose, kojibiose, and starch (Cantarel et al., 2009). Phosphorylases are classified into EC 2.4.1.x as glycosyltransferases, which transfer hexose sugar to a phosphate acceptor, whereas they belong to four GH (GH13, GH65, GH94, and GH112) and three glycosyltransferase families (GT; GT3, GT4, and GT35).

The transglycosylation (transfer reaction) applies the same mechanism as in hydrolysis. In the glycosylation step, the glycosyl residue, which attach to the enzyme forming a glycosyl-enzyme intermediate is referred to as donor (or glycosyl donor). However, in the deglycosylation step, instead of a water molecule, other molecules, such as carbohydrates, alcohols, or peptides can attack the intermediate resulting in release of a product with a new glycosidic linkage. The molecule, which accepts the glycoside donor, is referred to as acceptor. In other words, hydrolysis by glycoside hydrolases can be considered as transglycosylation reaction, in which the acceptor is the water molecule (Fig. 5) (Shaikh and Withers, 2008).

enzymes catagories (glycoside hydrolases, glycosyltransferases, polysaccharide lyases, and carbohydrate esterases) and the carbohydrate-binding modules (Cantarel et al., 2009).



Figure 4 Hydrolytic mechanisms of glycoside hydrolases. A) Retaining mechanism of β -glycosidase. B) Inverting mechanism of β -glycosidase (Davies and Henrissat, 1995). AB, catalytic acid/base residue; Nuc, catalytic nucleophile; R and R₁, aglycone side of the substrate; R₂, hydrogen in water (hydrolysis) or other molecule with hydroxyl group (transglycosylation).



Figure 5 Hydrolysis, transglycosylation, and glycosynthase mechanisms of glycoside hydrolase (Shaikh and Withers, 2008). Nuc, catalytic nucleophile; A/B, catalytic acid base; F, fluoride; R, aglycone side of the substrate; acceptor molecule is in red; glycosyl fluoride is in magenta.

4.1 Enzyme catalysed transglycosylation

In general, the glycosylation is considered to be an important method for structural modification of carbohydrate-containing compounds with useful biological activities. The glycosidic linkage, joining carbohydrate to another carbohydrate or other molecule, is one of the most important chemical characteristics of a saccharide. The synthesis of a glycosidic linkage can be performed by either chemical or enzymatic reactions. Chemical synthesis of the glycosidic linkage usually utilises a sugar unit as precursor of an intermediate oxacarbenium ion that has significant positive charge at the anomeric carbon atom. Sugar units have more than one possible site of reaction and each of these sites often shows very similar reactivity, resulting in several positions where a linkage can be introduced. Thus, masking by protecting groups is essential to direct coupling to the right position. For this reason, protection and deprotection of functional groups are used extensively for the chemical synthesis of carbohydrates. In addition, a difficulty in synthesis of oligosaccharides is to ensure the correct stereochemistry of the glycosidic linkage formed (Stick, 2001). The enzymatic glycosylation has several advantages over the chemical synthesis since it is usually more stereo- and regioselective and can be performed under mild reaction conditions without additional steps for protection and deprotection. Moreover, enzymatic glycosylation is more suitable for the modification of biologically active substances particularly in the food or cosmetic fields, where in general harsh conditions or use of toxic catalysts are undesirable (Murata and Usui, 2006).

The enzymes responsible for the synthesis of glycosidic linkage have been recognised as glycosyltransferases, which transfer sugar moieties from activated donor molecules to specific acceptors, forming a specific glycosidic bond *in vivo* (Qian et al., 2002; Sinnott, 1990). Together with glycosidases, they form the major catalytic machinery for synthesis and degradation of glycosidic bonds. The main advantage of using glycosyltransferases is the high efficiency and high regio- and stereoselectivity exhibited by the enzymes; however, the disadvantage is the inaccessibility of the transferases and the high-cost and instability of the activated nucleoside phosphate donors. Even though recombinant glycosyltransferases recently became available and the concept of using nucleotide sugar regeneration has been pursued, the use of these enzymes is limited (Murata and Usui, 2006).

Glycoside hydrolases. in contrast, hydrolyse glycosidic linkages and tranglycosylation is commonly observed as a side reaction, which sometimes provides a possibility to achieve efficient synthesis of carbohydrates. Although, the regioselectivity of glycosidases is rarely absolute and their yields are lower than those of glycosyltransferases, their glycosyl donor substrates are relatively inexpensive, chemically simple, and readily available. Both endo- and exo-glycosidases can perform transglycosylation. Most glycoside hydrolases used for oligosaccharide synthesis are exoglycosidases, which usually transfer a non-reducing terminal monosaccharide unit from a donor to a variety of acceptors, while endo-glycosidases are possibly capable of transferring longer oligosaccharides or a whole carbohydrate block to a new acceptor. By this approach, new bioactive saccharides can be created, including neoglycoproteins and neoglycolipids (Crout and Vic, 1998; Murata and Usui, 2006). Glycoside hydrolase equilibrium is shifted towards hydrolysis of the substrates; this, however, obviously represents a source of degradation of the transglycosylation products. Hence, to achieve oligosaccharide synthesis, the transglycosylation reaction must be faster than hydrolysis and the rate of hydrolysis of the transglycosylation products should be slow. Furthermore, to improve the transglycosylation ability of glycoside hydrolases, several techniques have been developed. The three major strategies in enzymatic synthesis of oligosaccharides are
briefly described below.

4.1.1 Reverse hydrolysis

Even though reactions catalysed by GHs are by far in the direction of hydrolysis for majority of the enzymes, the reverse reaction is possible depending on individual enzyme characteristics in an equilibrium-controlled condition to concur the thermodynamic barrier of the enzymes. Instead of breaking the glycosidic linkage by reaction with water, the reverse hydrolysis links two saccharide molecules by transfer of donor, after departure of the leaving group, to the acceptor molecule and release of a water molecule (Fig. 6). A shift of the thermodynamic equilibrium towards synthesis can be achieved by increasing the concentration of sugar donor substrate and acceptor (typically alcohols, or monosaccharides) together with reaction at medium high temperature (usually above 50°C) and high concentration of co-solvents (80–90%), if the acceptor is not alcohol, to reduce water activity (Plou et al., 2007). The reverse hydrolysis reaction is simple from a mechanistic point of view and moderate yield can be obtained particularly in synthesis of alkyl-saccharides. However, this same level of efficiency is not typical for oligosaccharide synthesis, because of a number of shortcomings, including enzyme instability at high temperature and solubility of saccharides in organic solvents, as well as separation of products from excessive amount of substrate (Crout and Vic, 1998; Murata and Usui, 2006; Stick, 2001; for examples of reverse hydrolysis see Ajisaka et al., 1987; Drouet et al., 1994; Singh et al., 2000).



Figure 6 Schematic reaction mechanism of reverse hydrolysis and transglycosylation (adapted from Shaikh and Withers, 2008). Nuc, catalytic nucleophile; A/B, catalytic acid base; R_1 , aglycone side of the substrate; R_2 , other saccharide with hydroxyl group marked in red.

4.1.2 Reverse phosphorolysis

In a reversible reaction of phosphorolysis, the substrate of the phosphorylase can be synthesied, this is referred to as reverse phosphorolysis. This approach was demonstrated by using maltose phosphorylase from *Lactobacillus brevis* (Tsumuraya et al., 1990). Maltose phosphorylase promotes the cleavage of maltose and releases β -glucose 1-phosphate (maltose + phosphate = D-glucose + β -D-glucose 1-phosphate) (Wood and Rainbow, 1961). By reverse phosphorolysis, β -D-glucosyl fluoride was used instead of β -D-glucose 1-phosphate as donor and D-glucose was used as acceptor, when incubating with maltose phosphorylase, maltose and HF were detected at a rapid rate as products of reverse phosphorolysis (β -D-glucosyl fluoride + D-glucose = maltose + HF) (Tsumuraya

et al., 1990). Furthermore, this approach was developed for production of novel oligosaccharides using cellobiose phosphorylase from *Cellvibrio gilvus* (Percy et al., 1998a, 1998b), cellodextrin phosphorylase from *Clostridium* thermocellum YM4 (Shintate et al., 2003), and maltose phosphorylase from *L. acidophilus* NCFM (Nakai et al., 2009). The obvious advantage with this approach is a very efficient reaction and high regioselectivity similar to the glycosyl transferase reactions (Nakai et al., 2003). Major limitations of the reverse phosphorolysis are the limited number of specificities found for phosphorylases and the amounts and price of 1-phosphate substrate required.



Figure 7 Schematic reaction mechanism of reversible phosphorolysis (Nakai et al., 2009). R, a carbohydrate residue.

4.1.3 Glycosynthase reaction

A very efficient approach for chemoenzymatic glycoside synthesis came with the aid of protein engineering, and is referred to as the glycosynthase reaction. Glycosynthases are glycoside hydrolase mutants in which the catalytic nucleophile was changed to an amino acid residue unable to act as a catalytic group by site-directed mutagenesis. These mutants essentially lost the hydrolytic activity and, instead catalyse the formation of glycosidic linkages when presented with a glycosyl fluoride with the opposite anomeric configuration of the normal substrate (Fig. 5). The first glycosynthase work was made with a retaining exo-glycosidase, β -glucosidase from Agrobacterium sp., in which the catalytic nucleophile glutamic acid was mutated to alanine that catalysed the formation of β -1,4 linked glycosides of glucose and galactose from α -glucosyl and α galactosyl fluorides as donors (Mackenzie et al., 1998; Withers et al., 1992). The glycosynthase has been developed for several enzymes and both exo- and endo-retaining glycosidases have produced different oligosaccharides containing glucose (Faijes et al., 2003; Mackenzie et al., 1998), galactose (Jakeman and Withers, 2002; Mackenzie et al., 1998), mannose (Jahn et al., 2003a), xylose (Ben-David et al., 2007; Kim et al., 2006; Sugimura et al., 2006), fucose (Wang, 2009), and glucuronic acid (Wilkinson et al., 2008) by mutating the catalytic nucleophile (aspartic or glutamic acid) to a small nonnucleophilic amino acid (such as alanine, glycine, cysteine, or serine). A more recent approach uses directed evolution to screen for amino acid substitutions that enhance glycosynthase activity (Mayer et al., 2000, 2001; Shaikh and Withers, 2008). Furthermore, the first inverting glycosynthase was pursued using aspartic acid to cysteine mutant of reducing-end xylose-releasing exo-oligoxylanase from Bacillus halodurans C-125 (Honda and Kitaoka, 2006). Glycosynthases are not only useful for oligosaccharide synthesis, but also for thioglycoside (S-glycosidic linkage) synthesis. This has been demonstrated by the use of retaining glycosidases that lack the catalytic acid/base amino acid residue (thioglycoligases), which requires strong nucleophiles as acceptors that do not need general base catalysis. The two first thioglycoligases were glutamic acid to alanine mutants of β -glucosidase from *Agrobacterium* sp. and β -mannosidase from *Cellulomonas fimi* (Jahn et al., 2003b). Glycosynthases have been useful for the preparation of oligosaccharides; however, their use has certain limitations, for instance, the success rate to convert a given glycosidase into a glycosynthase is rather low and the availability and stability of specific glycosyl fluoride donors is for some cases problematical (Williams and Withers, 2000).

4.2 Xylanases and xylosidases

4.2.1 Enzymatic hydrolysis of xylans

The structures of xylans are heterogeneous and complex (see 2.1.4.1), hence a variety of glycoside hydrolases and esterases are required for their breakdown. These enzymes or enzyme systems are called xylan degrading or xylanolytic enzymes (Beg et al., 2001; Pastor et al., 2007). Xylanolytic enzymes are produced by a variety of microorganisms, mainly filamentous fungi as well as certain rumen microbiota, and they are widely used in industries, particularly pulp and paper as well as in animal feed, manufacture of bread, food and beverage, textiles, ethanol, and xylitol production (Beg et al., 2001). Since several xylan containing plant hydrolysate by-products are applied for second generation biofuels, xylanolytic enzymes are developed for complete hydrolysis of such materials.

Xylanolytic enzymes can be classified into two main groups: the xylan backbone degrading enzymes, *i.e.* endo- β -1,4-xylanases and exo- β -1,4-xylosidases, and the side chain degrading enzymes, *i.e.* α -L-arabinofuranosidases, α -D-glucuronosidases, acetyl xylan esterases, ferulic acid esterases, and *p*-coumaric acid esterases (Fig. 8 and table 3). Xylanases are responsible for hydrolysis of the xylan backbone and release shorter polysaccarides or xylooligosaccharides, while β -xylosidases are responsible for hydrolysis of unbranched xylooligosaccharides and thus contribute to prevent product inhibition of xylanases. The synergy between xylanases and xylosidases as well as the side chain degrading enzymes and esterases are required for complete xylan degradation (Beg et al., 2001; de Vries and Visser, 2001; Pastor et al., 2007).

Table 3 Hemicellulolytic (and certain cellulolytic) enzymes, their classification, and current crystallographic data (adapted and updated from Shallom and Shoham, 2003)

			CA	Zy	Cat	alytic		
Enzymes	Substrates	EC	family	Clan	mech	anism ¹	Fold	3D structure ²
Endo-β-1,4-xylanase	β-1,4-xylan	3.2.1.8	GH5	А	Re:	E/E	$(\beta/\alpha)_8$	3GTN, 1NOF
			GH8	М	In:	D/E	$(\alpha/\alpha)_6$	1H12
			GH10	А	Re:	E/E	$(\beta/\alpha)_8$	1TA3, 1FXM, 1EXP
			GH11	С	Re:	E/E	β-jelly roll	1XYN, 1ENX, 1TE1
			GH43	F	In:	D/E,D	5-fold β-propeller	1UX7, 1W0N
Exo-β-1,4-xylosidase	β-1,4-xylooligomers,	3.2.1.37	GH3	-	Re:	D/E	-	(1EX1), (3F93), (2X40)
	xylobiose		GH30	А	Re:	E/E	$(\beta/\alpha)_8$	(10GS)
			GH39	А	Re:	E/E	$(\beta/\alpha)_8$	1W91, 1PX8
			GH43	F	In:	D/E,D	5-fold β-propeller	1YRZ, 2EXH, 3C2U ³
			GH52	-	Re:	E/D	-	Cryst.
α -L-Arabinofuranosidase	α -Arabinofuranosyl (1,2)	3.2.1.55	GH3	_	Re:	D/E	-	(1EX1), (3F93), (2X40)
	or (1,3) xylooligomers,		GH43	F	In:	D/E,D	5-fold β-propeller	3C7E, 3C2U ³
			GH51	А	Re:	E/E	$(\beta/\alpha)_8$	2C7F, 1PZ2, 2VRK
			GH54	_	Re:	-	-	1WD3
			GH62	F	_	-	-	Cryst.
Endo-α-1,5-arabinanase	α -1,5-arabinan	3.2.1.99	GH43	F	In:	D/E,D	5-fold β-propeller	1UV4, 3CU9, 1GYD
α -Glucuronidase	(4-O-methyl)-α-glucuronic	3.2.1.131	GH67	_	In:	—/E	$(\beta/\alpha)_8$	1K9D, 1K9E, 1GQI
	acid (1,2) xylooligomers		GH115	_	_	-	-	_
Endo-β-1,4-mannanase	β-1,4-mannan	3.2.1.78	GH5	А	Re:	E/E	$(\beta/\alpha)_8$	2MAN, 1VJZ, 1QNO
			GH26	А	Re:	E/E	$(\beta/\alpha)_8$	2WHK, 2QHA, 1GVY
			GH113	А	Re:	E/E	$(\beta/\alpha)_8$	3CIV
Exo-β-1,4-mannosidase	β-1,4-mannooligomers,	3.2.1.25	GH1	А	Re:	E/E	$(\beta/\alpha)_8$	(1QOX)
	mannobiose		GH2	А	Re:	E/E	$(\beta/\alpha)_8$	2JE8
			GH5	А	Re:	E/E	$(\beta/\alpha)_8$	1UUQ
α -galactosidase	α -galactopyranosyl (1,6)	3.2.1.22	GH4	_	Re:	-	-	1S6Y
	mannooligomers		GH27	D	Re:	D/D	$(\beta/\alpha)_8$	1R46, 1SZN, 3A5V
			GH36	D	Re:	D/D	$(\beta/\alpha)_8$	1ZY9
			GH57	_	Re:	Е/-	$(\beta/\alpha)_7$	(2B5D)
			GH97	_	In:	E/E	$(\beta/\alpha)_8$	3A24

Table 3 ((continued)

			CA	Zy	Cat	alytic		
Enzymes	Substrates	EC	family	Clan	mech	anism ¹	Fold	3D structure ²
α -galactosidase (cont.)			GH97	-	Re:	D/E	$(\beta/\alpha)_8$	
			GH110	-	In:	-	_	_
β-glucosidase	β-glucopyranosyl (1,4)	3.2.1.21	GH1	А	Re:	E/E	$(\beta/\alpha)_8$	1BGA, 1GNX, 1QOX
	mannopyranose		GH3	-	Re:	D/E	_	2X40
			GH9	_	In:	D/E	$(\alpha/\alpha)_6$	(1G87)
			GH30	А	Re:	E/E	$(\beta/\alpha)_8$	(10GS)
Endo-galactanase	β -1,4-galactan	3.2.1.89	GH53	А	Re:	E/E	$(\beta/\alpha)_8$	1R8L, 1FHL, 1HJQ
Acetyl xylan esterase	2- or 3-O-acetyl xylan	3.1.1.72	CE1	_	_	_	$(\alpha/\beta/\alpha)$ sandwich	(1JJF)
			CE2	_	_	-	$(\alpha/\beta/\alpha)$ sandwich	2WAA, 2W9X, 2WAB
			CE3	-	_	-	$(\alpha/\beta/\alpha)$ sandwich	2VPT
			CE4	_	_	-	$(\beta/\alpha)_7$ barrel	1HEH, 2C71, 2CC0
			CE5	_	_	-	$(\alpha/\beta/\alpha)$ sandwich	1QOZ, 1G66
			CE6	-	_	-	$(\alpha/\beta/\alpha)$ sandwich	1ZMB, 2AEA
			CE7	-	_	-	$(\alpha/\beta/\alpha)$ sandwich	3FVR, 1L7A, 3FCY
			CE12	_	_	-	$(\alpha/\beta/\alpha)$ sandwich	(2W1W)
Acetyl mannan esterase	2- or 3-O-acetyl mannan	3.1.1.6	CE16	-	_	-	_	_
Feruloyl & <i>p</i> -coumaroyl	5-O-feruloyl or	3.1.1.73	CE1	-	_	-	$(\alpha/\beta/\alpha)$ sandwich	1JJf, 1GKK
esterases	<i>p</i> -coumaroyl arabinoxylan							
Endo-β-1,4-glucanase	Cellulose	3.2.1.4	GH5, 6, 7, 8, 9, 12, 44, 45, 48, 51, 61, and 74					
Cellobiohydrolase	Cellulose (release cellobiose)	3.2.1.91	GH5, 6, 9	9, 48				
Exo-β-1,4-glucanase	β-1,4-cellooligomers	3.2.1.74	GH1, 3					

 Exo-p-1,4-glucanase
 p-1,4-cellooligomers
 3.2.1./4
 GH1, 3

 ⁻¹ Re: retaining, In: inverting, X/X: nucleophile/(acid/base) residues for retaining enzyme or general base/general acid for inverting enzyme, D: aspartic acid, E: glutamic acid.

 ² PDB code are reported, Cryst.: crystallization reported, ...: more structures have been reported, The PDB entries in parenthesis are from the relevant family but with different substrate specificity.

 ³ Bifunctional enzyme



Figure 8 Xylanolytic (upper) and mannan degrading enzymes (lower) with their sites of attack on xylan and mannan substrates (Shallom and Shoham, 2003).

4.2.2 GH10 and GH11 xylanases

Xylanases (endo- β -1,4-xylanases, EC 3.2.1.8) catalyse the hydrolysis of the xylan backbone, yielding shorter polymer or xylooligosaccharides. In addition, they are active on xylooligosaccharides of DP greater than two, showing increase in affinity for xylooligomers of increasing length (Pastor et al., 2007). Xylanases of fungal origin are effective in the acidic pH range 4-6, while the xylanases from actinomycetes and bacteria are effective in a broader pH range 5–9. The optimum temperature for xylanases action ranges between 35-60°C. Several extremophilic xylanases are discovered which are still active up to 100°C or pH 2-11 (Beg et al., 2001). Xylanases can be classified roughly into two groups: low molecular weight (<30 kDa) with basic pI and high molecular weight (>30 kDa) with acidic pI (Wong et al., 1988). Xylanases are classified into five glycoside hydrolase families: 5, 7, 10, 11, and 43 according to the CAZy classification. GH5, GH7, GH10, and GH11 hydrolyse by a retaining mechanism with glutamic acids as both catalytic nucleophile and acid/base, whereas GH43 catalyses hydrolysis by an inverting mechanism with aspartic acid as general base and glutamic or aspartic acid as general acid (table 3) depend on the enzymes. In the present study, we have worked with GH10 and GH11 xylanases. These two families include xylanases of high molecular weight/low pI and low molecular weight/high pI, respectively (Pastor et al., 2007).

Kinetic and structural analysis by GH10 xylanases revealed that their active sites have four to five subsites (-3 or -2 to +2) (Biely et al., 1997; Charnock et al., 1998; Pell et al., 2004; Schmidt et al., 1999). The glycone (or donor) binding subsites are numbered negatively (-1, -2, etc) from the scissile bond towards the non-reducing end of the bound

substrate. The aglycone (or acceptor) binding subsites, heading away from the scissile bond towards the reducing end, are denoted by positive numbers (+1, +2, etc). The cleavage by definition takes place between subsites -1 and +1 (Biely et al., 1981; Davies et al., 1997). GH11 xylanases have up to three (-) subsites and three (+) subsites (Payan et al., 2004; Vardakou et al., 2008). On the hydrolytic activity, GH11 xylanases prefer linear xylans and unsubstituted regions of arabinoxylans to the decorated ones, whereas GH10 xylanases can cleave within decorated region, with the presence of MeGlcA, acetyl, or α -Araf substitution. In general, GH10 xylanases exhibit wider catalytic versatility or lower substrate specificity than GH11 xylanases and can also exhibit low activity towards cellulosic substrates (Biely et al., 1997; Kolenová et al., 2006; Pastor et al., 2007). This reflects the shape of the active sites. GH10 xylanases are $(\beta/\alpha)_8$ barrel or salad bowl-like shaped catalytic domains, with the larger radius on the top of the bowl where a shallow groove active site is situated. The structures of GH11 assume a β-jellyroll fold in which two large β -pleated sheets and one α -helix form a partially closed right hand-like shape, with the active sites (or the palm) formed by the cleft between the fingers and the thumb, which make obscure the binding of decorated substrate into the active sites (Fig. 9) (Berrin and Juge, 2008; Collins et al., 2005; Törrönen and Rouvinen, 1997). The variation of properties of arabinoxylans, such as water-extractable and waterunextractable arabinoxylans, affects the mode of action of xylanases. The relative activity of xylanases towards these substrates referred to as substrate selectivity illustrates the impact of enzyme functionality in cereal based biotechnology processes (Berrin and Juge, 2008; Frederix et al., 2004b; Moers et al., 2003; Moers et al., 2005).

4.2.3 GH3 xylosidases

 β -Xylosidases (exo- β -1,4-xylosidases, EC 3.2.1.37) catalyse hydrolysis of xylobiose and of xylooligosaccharides, generated by the action of xylanases by release of xylose from the non-reducing end. The affinity of β -xylosidases for xylooligosaccharides decreases with increasing of DP of the substrates (Beily, 2003; Pastor et al., 2007). Several β -xylosidases are reported to attack polymeric xylan in an exo-fashion (Beily, 2003). β -Xylosidases act synergistically with other xylan degrading enzymes and are required to accomplish complete hydrolysis of xylan, which is industrially important for the production of second generation biofuels from lignocellulosic biomass (Beg et al., 2001). The pH and temperature optimum of β-xylosidases are in the range of pH 3–5 and 50–70°C, respectively (Herrmann et al., 1996; Kurakake et al., 1997; Kumar and Ramon, 1996; Rasmussen et al., 2006). β -xylosidases are classified into five glycoside hydrolase families: 3, 30, 39, 43, and 52, (54)⁸ according to the CAZy classification. GH3, GH30, GH39, and GH52 hydrolyse by a retaining mechanism, whereas GH43 hydrolyses according to an inverting mechanism (Table 3).

In the present study, we are interested in GH3 β -xylosidases, which belong to a family that includes β -glucosidases (EC 3.2.1.21), β -*N*-acetylhexosaminidases (EC 3.2.1.52), glucan 1,3- β -glucosidases (EC 3.2.1.58), glucan 1,4- β -glucosidases (EC 3.2.1.74), exo-1,3-1,4-glucanases (EC 3.2.1.-) and α -L-arabinofuranosidases (EC 3.2.1.55). The knowledge of structure-function relationships insight of GH3 β -xylosidases is quite limited since there is no structure reported so far, while there are two exo-1,3/1,4- β -glucanase structures from *Hordeum vulgare* and *Pseudoalteromonas* sp. BB1 (PDB ID: 1X38, Hrmova et al., 2005; 1EX1, Varghese et al., 1999; 3F93, Nakatani

⁸ Bifunctional enzyme with main activity as α -L-arabinofuranosidase

et al., to be published) and one β -glucosidase structure from *Thermotoga neapolitana* DSM 4359 (PDB ID: 2X40, Pozzo et al., 2010), which have been reported.



Figure 9 Cartoon representation of 3D structures of GH10, GH11 xylanases, and GH3 xylosidase (upper: top view, lower: side view). A) *A. nidulans* GH10 xylanase (PDB ID: 1TA3) docking with X5 from xylanase from *Penicillium simplicissimum* (PDB ID: 1B3Z). B) 3D homology model of *A. nidulans* GH11 xylanase (using xylanase from *P. funiculosum*, PDB ID: 1TE1, as template; 67% sequence identity) docking with X2 and X3 from xylanase from *Escherichia coli* (PDB ID: 2VGD). C) 3D homology model of *A. nidulans* GH3 xylosidase (β -glucosidase from *Thermotoga neapolitana*, PDB ID: 2WT3, as template; 27% sequence identities) docking with thiocellobiose from exo-1,3/1,4- β -glucanase from *Hordeum vulgare* (PDB ID: 1IEX), three colours representing three domains within the structure (for details on homologous modeling see 5.2, page 34), the enzyme structures were visualized by PyMOL v0.99.

4.2.4 Transglycosylation by xylanases and xylosidases

It is generally accepted that transglycosylation naturally occurs as catalysed by glycoside hydrolases resulting in varying extent of transglycosylation. This also includes xylanases. During the characterization of GH10 and GH11 xylanases, transglycosylation is usually observed as a side activity when high concentration of oligosaccharide substrates is used. The enzymes can transfer xylooligosaccharide donors to create longer xylooligosaccharides (Berrin et al., 2006; Biely et al., 1981; Christakopoulos; 1996; Jiang et al., 2004; Kolenová et al., 2006; Yan et al., 2009). Moreover, the substrate donor is not only limited to short xylooligosaccharides, several transglycosylation reactions utilize xylans as substrate mixed directly with acceptor molecules, particularly different

alcohols. In the reactions using xylans as donor, xylanases can transfer mainly xylobiosyl These have been identified by synthesis of acceptor molecules. alkvl to such xylooligosaccharides hexylxylobioside (Kadi and Crouzet, 2008). as octylxylobioside (Matsumura et al., 1999), and benzylxylobioside (Kadi et al., 2002; Kadi and Crouzet, 2008) as well as alcoholic aroma xylooligosaccharide conjugates synthesized in the same fashion (Kadi et al., 2002). It is rare for xylanases to utilize other saccharides than xylooligosaccharides as acceptor due to the strict recognition, however, in one case; xylosylpsicoses⁹ were synthesized by tranglycosylation using xylanase from A. sojae (Oshima et al., 2006).

Moreover, GH3 β-xylosidases are capable of catalysing transglycosylation at high substrate concentration, in which xylose from β -xylosyl donors, such as *para*-nitrophenyl β -D-xyloside or short xylo-oligosaccharides, are transferred to another xylosyl substrate forming *para*-nitrophenyl β-1,4 xylo-oligosaccharides or longer β-xylooligosaccharides with small amount of 1,3 glycosidic linkage (Eneyskaya et al., 2003; Rodionova et al., 1983; Win et al., 1988). Recently, transglycosylation using A. niger or Aurebasidium pullulans β -xylosidase was applied to produce para-nitrophenyl β -1,4-D-xylobioside, which is used as substrate in xylanase assay (Puchart and Biely 2007). Many ßxylosidases show broad acceptor specificity, including different mono-, disaccharides, sugar alcohols (Kurakake et al., 1997, 2005), alcohols (Shinoyama et al., 1988; Drouet et al., 1994), and hydroquinones (Sulistyo et al., 1994), which allows the formation of novel xylooligosaccharides, alkyl β-xylosides, and hydroquinone β-xylosides, respectively, under appropriate conditions. The heterogeneous transglycosylation using different nonβ-xylo-oligosaccharide acceptors has been tested by A. awamori K4 GH3 β-xylosidase using xylobiose as donor with acceptors including three monosaccharides (arabinose, mannose, fructose), four disaccharides (trehalose, isomaltose, palatinose, maltose), and nine sugar alcohols (erythritol, xylitol, sorbitol, mannitol, dulicitol, maltitol, lactitol, palatinitol, myo-inositol) in yields of 44-69%, 0.8-50%, and 10-68%, respectively (Kurakake et al., 1997; 2005).

Being glycoside hydrolases, xylanases and xylosidases tend to hydrolyse the transglycosylation products. To circumvent this, glycosynthases can be used. So far, for xylanases, only glycosynthases from GH10 have been successfully obtained (Kim et al., 2006; Sugimura et al., 2006) and for the xylosidases, only GH52 xylosidase was reported as a glycosynthase (Ben-David et al., 2007).

4.2.5 Objective for investigation on GH3, GH10, and GH11

Objective for GH3 xylosidases:

- To characterise hydrolytic and transglycosylation activity of recombinant GH3 xylosidases from *Aspergillus nidulans* FGSC A4 (BxlA).
- To synthesise and characterise novel xylooligosaccharides *via* transglycosylation by this GH3 xylosidase.
- To evaluate the effect of subsite +1 of BxlA in transglycosylation.
- To characterise the glycosynthases of BxlA

For GH3 xylosidase, we have reported hydrolytic activity of the enzyme towards different xylooligosaccharides (X2–X6), *para*-nitrophenyl β -D-xyloside (*pNPX*) and *para*-nitrophenyl α -L-arabinofuranoside (*pNPA*), as well as pH and temperature optima

⁹ Psicose (D-ribo-2-hexulose): C-3 epimer of D-fructose.

and stability, pI, and molecular mass of this enzyme in solution. The transglycosylation activity of this enzyme will also be described. The results will be compared with those for another β -xylosidase (BxlB) from the same origin, investigated by Dr. Hiroyuki Nakai (EPC, DTU). Furthermore, we are producing several novel xylooligosaccharides by transglycosylation, some of the structures were determined by nuclear magnetic resonance spectroscopy (NMR) and electrospray ionization mass spectrometry (ESI-MS).

Even though, the three-dimensional (3D) structure of GH3 xylosidase is not available, subsite +1 of BxlA can be tentatively identified by the aid of bioinformatics. In the present study, we evaluated the function of subsite +1 by the site-directed mutagenesis of Cys308 to Trp (C308W) and Asn313 to Arg (N313R) (the details of sequence comparison and homology modeling of this enzyme will be presented, 6.1.6, page 59). Since there is no glycosynthase reported from GH3, four glycosynthase mutants from BxlA were made by site-directed mutagensis of Asp307, the catalytic nucleophile to Gly, Ala, Cys, and Ser.

Objective for GH10 and GH11 xylanases:

- To evaluate the potential transglycosylation activity of recombinant GH10 and GH11 xylanases from *Aspergillus nidulans* FGSC A4 (XlnC and XlnA, respectively) for production of novel xylooligosaccharides
- To evaluate the glycosynthase of XlnC

GH10 and GH11 are well-characterized families and many 3D structures have been reported (Table 3). In the xylanases study, we have investigated the possibility to use other disaccharides than xylobiose as well as sugar alcohols as acceptors in transglycosylation. In the preliminary acceptor screening, the XlnC showed possibility to transfer xylobiosyl to other acceptors than xylooligosaccharides; hence the glycosynthase of XlnC was made, by site-directed mutagenesis of Glu244 to Ala (E244A).

4.3 Mannanases

4.3.1 Enzymatic hydrolysis of mannans

Mannan degrading enzymes comprise four glycoside hydrolases and one esterase, *i.e.* endo- β -1,4-mannanases, exo- β -1,4-mannosidases, α -galactosidases, β -glucosidases, and acetyl mannan esterases (Fig. 8 and Table 3) (Dhawan and Kaur, 2007 and Moreira and Filho, 2008). Mannanases are responsible for the cleavage of β -1,4-linked internal linkages of the mannan backbone in mannans including (galacto)glucomannans, while β mannosidases cleave β -1,4-linked Man, releasing mannose from the non-reducing end of mannans and mannooligosaccharides. B-Glucosidases are required to hydrolyse B-1,4linked Glc at the terminal end of glucomannooligosaccharides released by the hydrolysis of glucomannans and galactoglucomannans by mannanases. α -Galactosidases, as debranching enzymes, catalyse the hydrolysis of α -1,6-linked Gal side chain of galactomannans and galactoglucomannans, while acetyl mannan esterases catalyse the release of acetyl groups from galactoglucomannans (de Vries and Visser, 2001; Dhawan and Kaur, 2007; Moreira and Filho, 2008). Hydrolysis of mannans is affected by the extent and pattern of substitution by the Gal side chain in galactomannans and galactoglucomannans and the pattern of distribution of the Glc in glucomannans and galactoglucomannans backbone. Like the β -xylosidases in xylan degradation or β glucosidases in cellulose degradation, β -mannosidases act synergistically with

mannanases and are essential to obtain complete hydrolysis of mannans (Dhawan and Kaur, 2007).

4.3.2 GH5 mannanases

Mannanases (endo-B-1,4-mannanase, EC 3.2.1.78) classified into three glycoside hydrolase families: 5, 26, and 113 according to CAZy classification. All of these belong to clan GH-A and hydrolyses by a retaining mechanism with glutamic acids as both catalytic nucleophile and acid/base (Table 3; Moreira and Filho, 2008; Zhang et al., 2008). The degradation by mannanases is greatly affected by the extent and pattern of Gal substitution and the pattern of distribution of Glc residues on the main chain. The main products by complete hydrolysis of mannans by mannanases are mannobiose and mannotriose (Dhawan and Kaur, 2007; Moreira and Filho, 2008). In general, mannanases have pH and temperature optima at 3.0-7.5 and 45-70°C, respectively (Dhawan and Kaur, 2007). The mannanase from Thermotoga neapolitana 5068, an extremophilic bacterium, has highest activity at 92°C (Duffaud et al., 1997). Mannanases are mainly used in food, feed, and pulp and paper industries. Since mannans are the major hemicelluloses in softwood, mannanases are applied for aiding in enzymatic bleaching of softwood pulps. In food industries, mannananases are used for reducing the viscosity of coffee extracts as well as for modifying mannan containing materials (e.g. gums), which also have other technical application such as in oil drilling (Dhawan and Kaur, 2007).

In the present study we are interested in GH5, which is divided into 9 subfamilies¹⁰ on the basis of sequence similarity (Béguin, 1990; Lo Leggio et al., 1997; Hilge et al., 1998; Larsson et al., 2006). Mannanases are classified into subfamilies 7 (eukaryotic mannanases), 8 (prokaryotic mannanases), and 10 (Hilge et al., 1998; Larsson, et al, 2006). Several crystal structures of GH5 mannanases have been reported. *Hypocrea jecorina* (anamorph: *Trichoderma reesei*; Sabini et al., 1999), *Thermomonospora fusca* (Hilge et al., 1998), and *Mytilus edulis* (Larsson et al., 2006) mannanases represent subfamilies 7, 8, and 10, respectively. GH26 and GH113 mannanases are exclusively of prokaryotic origin (Hogg et al., 2003). GH5 mannanases have five substrate binding subsites (–3 to +2) similar to GH26 and GH113 (Anderson, et al., 2008; Larsson et al., 2006; Sabini et al., 2000; Tailford et al., 2009). Binding to at least four subsites is required for efficient hydrolysis and the substitution of the substrate restricts hydrolysis, most likely by preventing substrate binding (Dhawan and Kaur, 2007).

4.3.3 Transglycosylation by mannanases

In the same manner as xylanases, mannanases can perform transglycosylation, which was observed during characterisation of the enzymes. Both microbial GH5 and GH113 are capable to perform transglycosylation (Gübitz et al., 1996; Harjunpää et al., 1995; Harjunpää et al., 1999; Hrmova et al., 2006; Zhang et al., 2008), whereas GH26 showed poor transglycosylation ability, which may stem from different subsite affinities (Anderson et al., 2008; Gilbert et al, 2008; Le Nours et al., 2005). Substrate donors for transglycosylation are usuallv mannooligosaccharides (DP 4-6)and the transglycosylation products are longer mannooligosaccharides. In plant, GH5 mannanase may play a significant role in modifying the structure of plant cell walls during cell germination, and fruit ripening, similarly to expansion. seed xvloglucan endotransglucosylase/hydrolase (Schröder et al., 2006, 2009). Currently, there are limited

¹⁰ Subfamilies 5 and 6 are referred to as subfamily 5/6

details reported on reactions using different acceptors, such as alcohols or other saccharides (Hekmet, to be published) and no report on using mannans as donors.

4.3.4 Objective for investigation on GH5

- To characterise hydrolytic and transglycosylation activity of recombinant mannanases of GH5 subfamily 7 from *Aspergillus nidulans* FGSC A4 (ManA and ManC) (in collaboration with Dr. Hiroyuki Nakai)
- To synthesise and characterise novel mannooligosaccharides *via* transglycosylation catalysed by GH5 mannanases.
- To evaluate the effect of subsite +1 on transglycosylation of both GH5 mannanases.

For GH5 mannanases, we present hydrolytic activity towards different mannans, kinetics towards mannooligosaccharides (M4–M6), hydrolysis patterns of these enzymes. The transglycosylation activity of these enzymes will be discussed. Furthermore, we produced novel mannooligosaccharides by transglycosylation catalysed by ManC with isomaltriose and melezitose¹¹ as acceptors. The structures of these products were determined by NMR and matrix assisted laser desorption ionisation time-of-flight mass spectrometry (MALDI-TOF MS).

The preliminary characterization of the two mannanases (by Dr. Hiroyuki Nakai) showed distinctly different transglycosylation activity. One of the reasons for this could be a structural difference at subsite +1. Since there are several structures of GH5 mannanases reported, subsite +1 of these mannanases can be identified by amino acid alignment and protein structure homologous modeling. Subsite +1 of ManA and ManC are well conserved except for Trp283 in ManC, corresponding to Ser289 in ManA. We are thus evaluating the effect of aromatic residue substitution at subsite +1 by the site-directed mutagenesis of both enzymes: Ser289 to Trp of ManA (ManAS289W) and Trp283 to Ser of ManC (ManCW283S; sequence comparison and homology modeling details are presented, 6.2.5, page 68).

4.4 Endo-β-glucanase

4.4.1 Enzymatic hydrolysis of cellulose

Cellulose is a polysaccharide consisting of a linear chain of β -1,4 linked Glc, which has an extremely rigid and compact structure. Cellulose degradation requires three types of cellulolytic enzymes, including cellobiohydrolases, endoglucanases, and β glucosidases, as well as an accessory protein (swollenin) (Brotman et al., 2008; de Vries and Visser, 2001; Merino and Cherry, 2007). The endoglucanases, mainly referred to as cellulases, catalyse randomly the hydrolysis of β -1,4 linkages in cellulose and act mainly on the amorphous regions of cellulose, which also results in exposure of more chain termini for cellobiohydrolases. Cellobiohydrolases catalyse hydrolysis of cellulose both from the non-reducing and the reducing end, depending on the type of enzyme, and release mainly cellobiose. These enzymes are needed for hydrolysis of crystalline cellulose. With a tunnel-shaped active site, they may help open up the cellulose chain, under certain circumstances, and allow significant endoglucanase action (Kubicek, 1992; Xu et al., 2007). β -glucosidases degrade primarily cellobiose (and cellooligosaccharides), which reduces product inhibition by cellobiose on cellobiohydrolases and endoglucanases

¹¹ Melezitose (α -D-Glc*p*-(1 \rightarrow 3)- β -D-Fur*f*-(2 \rightarrow 1)- α -D-Glc*p*)

and are essential for complete breakdown of cellulose (Kubicek, 1992; Xu et al., 2007). Swollenin, an expansin-like protein, helps making cellulose fibres more accessible by disrupting the crystalline structure of the cellulose, which supports the enzymatic degradation, particularly by endo-glucanases (Brotman et al., 2008; Saloheimo et al., 2002).

4.4.2 GH61 putative endo-β-glucanase

The functionally uncharacterised family GH61 has 116 members, which are solely from eukaryotes and mainly from filamentous fungi. The activity, function, and mechanism of this family are not clear. These putative enzymes remarkable have a carbohydrate binding module family (CBM)¹² 1, found in several members such as from Agaricus bisporus (CEL1; Armesilla et al., 1994), A. kawachii (Cel61A; Hara et al., 2003 or AkCel61A: Koseki et al., 2008) and T. reesei (EG IV; Saloheimo et al., 1997, renamed to Cel61A; Karlsson et al., 2001). GH61 was presumed to have some function in degradation of lignocellulose, possibly increasing the cellulase activity (Brown et al., 2005; Karlsson et al., 2001; Koseki et al., 2008; Liu et al., 2006). The proteins have very low levels of endo-glucanase activity, more than 10,000 times lower in activity than usual endo-glucanases, towards cellulose derivative substrates such as carboxymethyl cellulose and hydroxyethyl cellulose, as well as β -glucan and lichenan (Karlsson et al., 2001). Currently, there are two reported 3D structures from two different fungi: Cel61B from Hypocrea jecorina QM6A (PDB ID: 2VTC; Karkehabadi et al., 2008) and Cel61E from Thielavia terrestris NRRL 8126 (PDB ID: 3EII and 3EJA; Harris et al., 2010). Both protein structures contain mainly B-strands forming twisted B-sandwich with metalbinding sites, as nickel-binding for Cel61B and zinc-binding for Cel61E, without CBM. Even though two 3D structures from this family have been reported, there is still no clear identifiable carbohydrate binding cleft or catalytic pocket to guide to unravel the function of this putative endo-glucanase. The function for this protein is still under quest.

4.4.3 Objective for investigation on GH61

- To demonstrate enzyme discovery by using a proteomics strategy.
- To produce and characterize the recombinant GH61 putative endo-β-glucanase from *Aspergillus nidulans* FGSC A4 (EglF).

In this part of the project, a proteomics strategy (Dr. Kenji Maeda, Dr. Hiroyuki Nakai, and PhD student Anders Dysted Jørgensen, EPC, DTU Systems Biology), was applied to analyse the culture supernatant of *A. nidulans* FGSC A4 growing on 10 different polysaccharides: potato starch, tamarind xyloglucan, birchwood xylan, oat spelt xylan, wheat arabinoxylan, sugar beet arabinan, larchwood arabinogalactan, guar galactomannan, carbo galactomannan, and konjac glucomannan (with glucose and without carbohydrate source as controls). The extracellular proteins were separated by 2DE and identified after trypsin digestion by MALDI-TOF MS. As a result, several novel and putative carbohydrate active enzymes were determined by the aid of genomic sequence of the fungus. Among several other novel carbohydrate active enzymes, a GH61 putative endo- β -glucanase was discovered in the culture broth-containing oat spelt xylan.

¹² A carbohydrate-binding module (CBM) is defined as a contiguous amino acid sequence within a carbohydrate-active enzyme with a discreet fold having carbohydrate-binding activity (Tomme et al., 1995)

The actual function of this enzyme was unrevealed and the protein might be useful for synthesis of β -glucooligosaccharides by transglycosylation; hence this protein was selected for further investigation. The cDNA cloning of the GH61 (EglF), the production of the corresponding protein as well as its characterisation including carbohydrate microarrays screening have been performed.

5. Materials and methods

5.1 Materials

5.1.1 Pichia pastoris transformants and Aspergillus strain

Pichia pastoris X-33 transformants harbouring full-length β -xylosidase, mannanase, and xylanase genes (Table 4) from *Aspergillus nidulans* FGSC A4 (Bauer *et al.*, 2006) and *Aspergillus nidulans* FGSC A4 (for cDNA cloning of GH61 putative endo- β -glucanase) were purchased from Fungal Genetics Stock Center, School of Biological Sciences, University of Missouri, Kansas City, MO.

				FGSC database	
Name ^a	GH	Enzyme	Locus tag	accession no.	GenBank ID
BxlA	3	β-xylosidase	AN2359.2	10077	EAA64470.1
BxlB	3	β-xylosidase	AN8401.2	10125	EAA67023.1
ManA	5	Mannanase	AN3358.2	10088	EAA63326.1
ManC	5	Mannanase	AN6427.2	10106	EAA58449.1
XlnC	10	Xylanase	AN1818.2	10073	EAA64983.1
XlnA	11	Xylanase	AN3613.2	10092	EAA59821.1
EglF	61	Endo-β-glucanase ^b	AN3860.2	_	EAA59125.1

Table 4 Enzymes identity and accession number

^a The enzyme names are according to Coutinho et al., 2009

^b putative endo β-glucanase or cellulase-enhancing factor (Harris et al., 2010)

5.1.2 Primers

The oligonucleotide primers (Table 5), constructed based on the *A. nidulans* genomic sequence (Galagan et al., 2005) were purchased from Eurofins MWG Operon (Ebersberg, Germany).

Table 5 Primers for construction of the expression plasmids of BxlA (GH3), BxlB (GH3), ManA (GH5), ManC (GH5), XlnC (GH10), XlnA (GH11), EglF (GH61), and their mutants

Primer	Primer sequence ^a	Restriction site
<u>bxlA</u>		
Sense	ATT <u>CTCGAG</u> AAAAGAGAGGGCTGAAGCTGCGAACACCAGCTACAC	XhoI
Antisense	ACTCTAGATTAATGATGATGATGATGATGAGCAATAACATCCTGC	XbaI
D307G sense	CGTCTCCGGT <u>GGT</u> TGCGGTGCTGTC	-
D307G antisense	GACAGCACCGCA <u>ACC</u> ACCGGAGACG	-
D307A sense	CGTCTCCGGT <u>GCT</u> TGCGGTGCTGTC	_
D307A antisense	GACAGCACCGCA <u>AGC</u> ACCGGAGACG	_
D307C sense	CGTCTCCGGT <u>TGT</u> TGCGGTGCTGTC	-
D307C antisense	GACAGCACCGCA <u>ACA</u> ACCGGAGACG	-
D307S sense	CGTCTCCGGT <u>TCT</u> TGCGGTGCTGTC	-
D307S antisense	GACAGCACCGCA <u>AGA</u> ACCGGAGACG	-
C308W sense	CGTCTCCGGTGAC <u>TGG</u> GGTGCTGTC	-
C308W antisense	GACAGCACC <u>CCA</u> GTCACCGGAGACG	_
N313R sense	GCGGTGCTGTCTAC <u>AGA</u> GTGTGGAACC	_
N313R antisense	GGTTCCACAC <u>TCT</u> GTAGACAGCACCGC	-
bxlA-sequencing 1	AACATCAACACTTTCCGACAC	_
bxlA-sequencing 2	AGACATCGACTGTGGAACTTC	_
bxlA-sequencing 3	AATTGACAACACAATCGAAG	-
<u>bxlB</u>		
Sense	GGG <u>GAATTC</u> AACTACCCGGACTGCACAACGGGCCCTC	EcoRI
Antisense	CCC <u>GGCGGCCG</u> CATCACTGTCGTTACCTGACAACGG	NotI
T1374C-sense	GGACAGAGGTGAA <u>C</u> TCGACAAGCACAGAC	_
T1374C-antisense	GTCTGTGCTTGTCGA <u>G</u> TTCACCTCTGTCC	_
bxlB-sequencing 1	CGCTATGTCAAAGAGTTTGTG	_
bxlB-sequencing 2	CGTTGACGAGACTATATACGAG	_
bxlB-sequencing 3	AGGAGGAACTCATTCTGGAAC	-
manA		
Sense	GGG <u>GAATTC</u> CTCCCTCACGCGTCAACACCTGTGTACAC	EcoRI
Antisense	CCC <u>GCGGCCGC</u> CTACTTGGCACTCCTCTCAATCGTCTCG	NotI
S289W sense	CTCTACCCGGAT <u>TGG</u> TGGGGGCAC	-
S289W antisense	GTGCCCCA <u>CCA</u> ATCCGGGTAGAG	-
<u>manC</u>		
Sense	GGG <u>GAATTC</u> CGCAAGGGCTTTGTGACCACCAAAGGCG	EcoRI
Antisense	CCC <u>GCGGCCGC</u> CTACCGTCTCCGGTTCAACTTGTTCACC	NotI
W283S sense	CTATCCGGA <u>AGT</u> TGGAGCAAGACC	-
W283S antisense	GGTCTTGCTCCA <u>ACT</u> ATCCGGATAG	—
<u>xlnC</u>		
Sense	TACATG <u>CCATGG</u> CCCAAAGCGCCAGCCTAAATGATCTT	NcoI
Antisense	AAGGAAAAAA <u>GCGGCCGC</u> AGACAGAGCGTTGACGATTGACG	NotI
E244A sense	GCCATCACC <u>GCG</u> CTTGATATTGCGGG	-
E244A antisense	CCCGCAATATCAAG <u>CGC</u> GGTGATGGC	-
<u>xlnA</u>		
Sense	TACATG <u>CCATGG</u> CCACGCCCGTCGGGTCTGAAGACC	NcoI
Antisense	AAGGAAAAAA <u>GCGGCCGC</u> GTAAACAGTAATAGAAGCCGACCCAC	NotI
<u>eglF</u>		
Sense	ATT <u>CTCGAG</u> AAAAGAGAGGGCTGAAGCTGGCCACGGATACGTGACG	XhoI
Antisense	AC <u>TCTAGA</u> TTAATGATGATGATGATGATGACCACTGTACAGCTCAGG	XbaI

^a Restriction sites are underlined, mutation positions are doubly underlined

5.1.3 Natural carbohydrate substrates (with acceptors for transglycosylation)

For GH3 β -xylosidases arabinose, lyxose, L-fucose, L-rhamnose, galactose, mannose, talose, fructose, maltose, sucrose, lactose, cellobiose, turanose, L-serine (Ser), and L-cysteine (Cys) were purchased from Sigma Aldrich (St. Louis, MO). Xylose (Xyl) was from Carl Roth (Karlsruhe, Germany). Glucose was from Merck (Darmstadt, Germany). Xylobiose (X2), xylotriose (X3), xylotetraose (X4), xylopentaose (X5), and xylohexaose (X6) were purchased from Megazyme (Wicklow, Ireland).

For GH5 mannanases mannose, melezitose¹³ (Mel3), and locust bean gum galactomannan (LBG GalMan) were purchased from Sigma Aldrich. Guar galactomannan (guar GalMan) was from Carl Roth. Mannobiose (M2), mannotriose (M3), mannotetraose (M4), mannopentaose (M5), mannohexaose (M6) and konjac glucomannan (konjac GlcMan) were from Megazyme. Isomaltotriose (Iso3) was a gift from Assoc. Prof. Haruhide Mori and Prof. Atsuo Kimura, Faculty of Agriculture, Hokkaido University, Japan.

For GH10 and GH11 xylanases xylotriose, mannobiose, and wheat arabinoxylan (low viscosity) were purchased from Megazyme. Birchwood and oat spelt xylans were from Carl Roth. Cellobiose, isomaltose, and maltose were from Sigma Aldrich.

For GH61 putative endo-glucanases, oat spelt xylan was from Carl Roth and barley β -glucan was from laboratory preparation.

5.1.4 Synthetic carbohydrate substrates (with acceptors for transglycosylation)

Acetylated α -xylosyl and α -xylobiosyl fluorides were kind gifts from Prof. Stephen G. Withers (University of British Columbia, Canada). *para*-Nitrophenol (*p*NP), *p*NP β -D-xylopyranoside (*p*NPX), *p*NP α -L-arabinofuranoside (*p*NPA), α -L-arabinopyranoside (*p*NPA*p*), *p*NP β -D-cellobioside (*p*NPC2), xylitol, and sorbitol were purchased from Sigma Aldrich.

5.2 Sequence comparison and homology modeling

for homology search of protein sequences BLAST was used (http://blast.ncbi.nlm.nih.gov/Blast.cgi; Altschul et al., 1997). The sequence alignment performed using T-Coffee (http://tcoffee.vital-it.ch/cgiwas bin/Tcoffee/tcoffee cgi/index.cgi; Notredame et al., 2000) and visualised using Easy Sequencing in Postscript (http://espript.ibcp.fr/ESPript/ESPript; Gouet et al., 1999). Homology models of BxlA, BxlB, ManA, and ManC were generated using HHpred-Homology server (http://toolkit.tuebingen.mpg.de/hhpred; Söding et al., 2005). For BxlA, β-glucosidase from Thermotoga neapolitana DSM 4359 (PDB ID: 2WT3; Pozzo et al., 2010) and β-glucanase from *Hordeum vulgare* (PDB ID: 1IEX; Hrmova et al., 2005) sharing 27% and 26% sequence identity, respectively served as templates. For GH5 mannanases, β-mannanase from Hypocrea jecorina RUTC-30 (PDB ID: 1QNR; Sabini et al., 2000) sharing 56% and 37% identity with ManA and ManC, respectively served as template. For putative GH61 endo-B-glucanase, H. jecorina QM6A (PDB ID: 2VTC; Karkehabadi et al., 2008) sharing 44% sequence

¹³ Melezitose (α -D-Glc*p*-(1 \rightarrow 3)- β -D-Fur*f*-(2 \rightarrow 1)- α -D-Glc*p*)

identity served as template. The models were validated by the Combinatorial Extension (CE) Method (http://cl.sdsc.edu/ce.html; Shindyalov and Bourne, 1998) and ProQ-Protein Quality Predictor (http://www.sbc.su.se/~bjornw/ ProQ/ProQ.html; Wallner and Elofsson, 2003).

For GH3 β-xylosidase the root-mean-square deviations (rmsd) was 0.8 Å for 641 and 529 equivalent C_{α} positions (out of 784 in BxlA) compared to 2WT3 and 1IEX with 7.7 and 7.5 Z-score. LGScore/MaxSub were 4.520/0.383 and 4.577/0.256, respectively, indicated that the models are of reasonable to good quality. Disulfide bonds of BxlA were predicted using DiANNA-Disulfide bond connectivity prediction (http://clavius.bc.edu/~clotelab/DiANNA/; Ferrèl and Clote, 2005). For GH5 mannanases rmsd were 0.3 Å for 343 equivalent C_{α} positions (out of 372 in ManA and 381 in ManC) compared to 1QNR with 7.9 and 7.7 Z-score for ManA and ManC, respectively, indicated that the models are of good to very good quality. For GH61 putative endo-β-glucanase rmsd was 0.7 Å for 224 equivalent C_{α} positions (out of 230 in EgIF) compared to 2VTC with 7.4 Z-score. LGScore/MaxSub was 3.200/0.319 indicated that the models are reasonable to good quality. All models were rendered and analysed using Pymol v0.99 (DeLano Scientific LLC, San Carlos, CA).

5.3 Cloning and mutations

5.3.1 GH3 xylosidases

P. pastoris X-33 transformants (bxlA – formerly xlnD and bxlB, Table 4), encode native full-length enzymes including signal peptides (Met1-Ser18 and Met1-Ser23 for bxlA and bxlB. respectively, predicted by SignalP 3.0 (http://www.cbs.dtu.dk/services/SignalP/; Bendtsen et al., 2004). The genes were therefore recloned to encode mature enzymes lacking signal peptides since the highly hydrophobic and the positively charged nature of signal peptides is likely to have significant effects on the properties of the gene product. Expand High Fidelity DNA polymerase (Roche Diagnostics GmbH, Mannheim, Germany) was used for DNA amplification using oligonucleotide primers (Table 5). The recloning of bxlB was performed by Dr. Hiroyuki Nakai. The silent mutation (1374T>C) of the EcoRI site in bxlB was introduced by overlap PCR (Sambrook et al., 1989) using the primer pair in Table 5. PCR products were digested by XhoI and XbaI for bxlA or EcoRI and NotI for bxlB (New England BioLabs, Ipswich, MA), and cloned into the pPICZ α A vector (Invitrogen, Carlsbad, CA). The plasmids (pPIC α A-BxlA and pPIC α A-BxlB) were transformed into Escherichia coli DH5a (Invitrogen) and transformants were selected on low salt Luria-Bertani (LB; 1% tryptone, 0.5% yeast extract, 0.5% NaCl, pH 7.0) supplemented with 25 µg/mL Zeocin. Purified plasmids (QIAprep Spin Miniprep; QIAGEN, Hilden, Germany) were fully sequenced (Eurofins MWG Operon, Ebersberg, Germany). The pPIC α A-BxlA was linearised by PmeI (New England BioLabs) and transformed into P. pastoris X-33 (Invitrogen) by electroporation (Micropulser; Bio-Rad Laboratories Inc, Hercules, CA). Transformants were selected on yeast peptone dextrose (1% yeast extract, 2% peptone, 2% dextrose) plates containing 1 M sorbitol and 100 µg/mL Zeocin at 30°C for 3 days.

The glycosyntase and subsite mutants of BxlA were obtained by site-directed mutagenesis (QuikChange Lightning Site-Directed Mutagenesis kit; Stratagene, La Jolla, CA) using D307G, D307A, D307C, and D307S primer pairs for glycosyntase

mutants and C308W and N313R primer pairs for subsite +1 mutants (Table 5) with pPIC α A-BxlA as template. The mutated plasmids were transformed into *E. coli* DH5 α , verified by fully sequencing, and transformed into *P. pastoris* X-33 as described above.

5.3.2 GH5 mannanases

P. pastoris X-33 transformants harbouring full-length β -mannanase genes *manA* and *manC* (Table 4) encoded native full-length enzymes including signal peptides (Met1–Ala18 for *manA* and *manC* predicted by SignalP 3.0; Bendtsen et al., 2004), therefore the genes were recloned to encode the mature enzymes lacking signal peptides. The recloning of β -mannanase genes was conducted by Dr. Hiroyuki Nakai using Expand High Fidelity DNA polymerase (Roche Diagnostics GmbH) for DNA amplification using oligonucleotide primers (Table 5). PCR products were digested by EcoRI and NotI (New England BioLabs) for both *manA* and *manC*, and cloned into the pPICZ α A vector (Invitrogen). The plasmids (pPICZ α A-ManA and pPICZ α A-ManC) were transformed into *Escherichia coli* DH5 α (Invitrogen). Transformants were selected on low salt LB supplemented with 25 µg/mL Zeocin and purified plasmids were fully sequenced (Eurofins MWG Operon). The plasmids were linearised by PmeI and transformed into *P. pastoris* X-33 as described in 5.3.1.

Substitution of Ser289 to Trp in ManA (ManAS289W) was obtained by sitedirected mutagenesis (QuikChange Lightning Site-Directed Mutagenesis kit; Stratagene) using S289W primer pair (Table 5) with pPICZ α A-ManA as template. Substitution of Trp283 to Ser in ManC (ManCW283S) was prepared using W283S primer pair with pPICZ α A-ManC as template. Mutated plasmids were transformed into *E. coli* DH5 α , verified by full sequencing, and transformed into *P. pastoris* X-33 as described in 5.3.1.

5.3.3 GH10 and GH11 xylananases

P. pastoris X-33 transformants harbouring full-length xylanase genes *xlnA* and *xlnC* (Table 4) encoded native full-length enzymes including signal peptides (Met1–Ala19 for *xlnA* and *xlnC*, predicted by SignalP 3.0; Bendtsen et al., 2004), therefore the genes were reconstructed to encode the mature enzymes without signal peptides. Expand High Fidelity DNA polymerase (Roche Diagnostics GmbH) was used for DNA amplification using oligonucleotide primers (Table 5). PCR products were digested by NcoI and NotI (New England BioLabs) for both *xlnA* and *xlnC*, and cloned into the pET28-a(+) vector (Novagen, Merck KGaA, Darmstadt, Germany). The plasmids (pET28-a(+)-XlnA and pET28-a(+)-XlnC) were transformed into *E. coli* BL21(DE3) (New England BioLabs) by heatshock at 42°C. Transformants were selected on LB (1% tryptone, 0.5% yeast extract, 1% NaCl, pH 7.0) supplemented with 30 µg/mL kanamycin and purified plasmids were confirmed by full-length sequencing (Eurofins MWG Operon).

The glycosyntase mutant of XlnC was prepared by site-directed mutagenesis (QuikChange Lightning Site-Directed Mutagenesis kit; Stratagene) using E244A primers pair (Table 5) with pET28-a(+)-XlnC as template. The mutated plasmids were transformed into *E. coli* BL21(DE3) and verified by full-length sequencing as described above.

5.4 Recombinant protein production and purification

5.4.1 GH3 xylosidases and mutants

P. pastoris X-33 harbouring BxIA wild type or mutants was grown in 3×1 L buffered glycerol-complex medium (BMGY; 1% yeast extract, 2% peptone, 100 mM potassium phosphate pH 6.0, 1.34% yeast nitrogen base with ammonium sulfate, $4 \times$ 10^{-5} % biotin, 1% glycerol) in 3 × 2.5 L plastic shake flask at 30°C for 24 h to OD₆₀₀ of 25–32. Cells were harvested (3000g, 5 min, 4°C) and resuspended in 1 L buffered minimal methanol medium (BMM; 100 mM potassium phosphate pH 6.0, 1.34% yeast nitrogen base with ammonium sulfate, 4×10^{-5} % biotin, 1% methanol, 1% casamino acid) in 2.5 L plastic shake flask. The induction was continued at 22°C for 72 h. Methanol was supplemented to a final concentration of 1% (v/v) every 24 h until harvest by centrifugation (13,500g, 4°C, 1 h). Then, the culture supernatant was concentrated and buffer-exchanged to 10 mM MES pH 6.0 (Pellicon tangential flow filtration systems, nominal molecular weight limit (NMWL) 10 kDa; Millipore, Billerica, MA), and applied to a Resource Q 6 mL column (GE Healthcare, Uppsala, Sweden) equilibrated with the same buffer. After washing with 37.5 mM NaCl in the same buffer, protein was eluted by a 37.5-325 mM NaCl linear gradient (flow rate: 2 mL/min, 144 mL). Fractions containing activity were pooled, concentrated (Amicon Ultra-15, NMWL 30 kDa, Millipore), and applied to a Hiload 26/60 Superdex G200 column (GE Healthcare) equilibrated with 20 mM MES, 0.25 M NaCl, pH 6.0 (flow rate: 1 mL/min). Fractions containing BxIA, verified by activity assay (5.6.1.1) and/or SDS-PAGE (5.5.1)¹⁴, were pooled, concentrated, and buffer-exchanged to 10 mM MES pH 6.0. All purification steps were performed using an ÄKTAexplorer chromatograph (GE Healthcare) at 4°C. The anion exchange chromatography was used instead of immobilized metal ion affinity chromatography (IMAC) due to the instability of BxlA at pH 7.5 and the very poor yield (<20% activity recovery) by IMAC at lower pH.

5.4.2 GH5 mannanases and mutants

ManA and ManC were produced and purified by Dr. Hiroyuki Nakai. *P. pastoris* X-33 harbouring ManAS289W and ManCW283S were grown in 3×1 L of BMGY at 30°C in 3×2.5 L plastic shake flask for 24 h to OD₆₀₀ of 34 (ManAS289W) and 28 (ManCW283S). The cells were harvested (3000g, 5 min, 4°C) and resuspended in 1 L BMM (containing 0.5% methanol) in 2.5 L plastic shake flask. The induction was continued at 22°C for 72 h and methanol was supplemented to the culture to a final concentration of 0.5% (v/v) every 24 h. The culture supernatants were harvested (13,500g, 4°C, 1 h), adjusted pH to 7.5 with 1 M K₂HPO₄, filtered (0.22 µm; TPP, Trasadingen, Switzerland), and applied to a 5 mL HisTrap HP column (GE Healthcare) equilibrated with 20 mM HEPES, 0.5 M NaCl, 10 mM imidazole, pH 7.5. After washing with 20 mM HEPES, 0.5 M NaCl, 22 mM imidazole, pH 7.5, followed by elution using a 22–400 mM imidazole linear gradient (flow rate: 1.0 mL/min, 25 mL) fractions containing enzymes, verified by SDS-PAGE, were pooled, concentrated (Centriprep YM30, Millipore), and applied to a Hiload 26/60 Superdex G75 column (GE Healthcare) equilibrated with 20 mM MES,

¹⁴ For glycosynthase mutants, the fraction were verified only by SDS-PAGE

0.15 M NaCl, pH pH 7.0, flow rate 0.5 mL/min. Fractions containing enzymes were pooled, concentrated, and buffer-exchanged to 20 mM HEPES pH 7.0. All purification steps were performed at 4°C.

5.4.3 GH10, GH11 xylanases, and GH10 glycosynthase mutant

E. coli BL21(DE3) harbouring XlnA, XlnC, and XlnCE244A were grown in 1 L of LB containing 30 µg/mL kanamycin in 2.5 L plastic shake flask at 37°C to OD_{600} of 0.6. Expression was induced by 0.1 mM isopropyl-1-thio- β galactopyranoside and continued at 12°C for 24 h. Cells were harvested (10,000g, 10 min, 4°C) and resuspended in 30 mL BugBuster Protein Extraction Reagent (Novagen) containing 1 µL Benzonase Nuclease (Sigma Aldrich). Following 30 min incubation at 4°C with rotating mixing, the supernatants were collected (16,000g, 4°C, 20 min), filtered (45 µm; GE Infrastructure Water & Process Technologies Life Science Microseparations, Trevose, PA), and applied to a 5 mL HisTrap HP column (GE Healthcare) equilibrated with 20 mM HEPES, 0.5 M NaCl, 10 mM imidazole, pH 7.5. After washing with 20 mM HEPES, 0.5 M NaCl, 22 mM imidazole, pH 7.5, followed by elution using a 22-400 mM imidazole linear gradient (flow rate: 1.0 mL/min 25 mL) fractions containing enzymes were pooled, concentrated (Centriprep YM10, Millipore), and applied to a Hiload 26/60 Superdex G75 column (GE Healthcare) equilibrated with 20 mM HEPES, 0.15 M NaCl, pH 7.0 (flow rate: 0.5 mL/min). Fractions containing enzymes, verified by activity assay (5.6.3.1) and/or SDS-PAGE (5.5.1)¹⁵ were pooled, concentrated, and buffer-exchanged to 20 mM HEPES pH 7.0. All purification steps were performed at 4°C.

5.5 Protein characterisation

5.5.1 Protein concentration determination

The Protein concentration was determined spectrophotometrically at 280 nm using extinction coefficients $E^{0.1\%} = 1.79$ (BxlA, D307G, D307A, D307C, D307S, N313R), 1.82 (C308W), 2.26 (ManA), 2.57 (ManC), 2.33 (ManAS289W), 2.51 (ManCW283S), 1.86 (EglF) as determined by amino acid analysis, conducted by Anne Blicher (EPC, DTU Systems Biology), or theoretical extinction coefficients $E^{0.1\%} = 2.78$ (XlnA) and 2.08 (XlnC, E244A) (http://www.expasy.ch/tools/protparam. html; Gasteiger et al., 2005).

Protein purity was assessed by sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE) using NuPAGE Novex Bis-Tris 4–12% gels (Invitrogen) in 1% MES buffered polyacrylamide NuPAGE minigel Novex system with reducing agent added in the sample preparation and antioxidant in the buffer system according to the manufacturer's recommendation followed by Coomassie Brilliant Blue G-250 staining (Candiano et al., 2004).

Isoelectric focusing (IEF) was carried out using 400–800 ng enzyme and Pharmacia PhastSystem (GE Healthcare) with 4 cm polyacrylamide PhastGel IEF, pH

¹⁵ For glycosynthase E244A, the fractions were verified only by SDS-PAGE

range 4.0–6.5 using the low range pI (pH 2.5–6.5) calibration kit (GE Healthcare) as marker and stained using PlusOne Silver Stain Kit, Protein (GE Healthcare).

5.5.2 Molecular mass determination

The molecular mass of the enzymes was estimated by SDS-PAGE (as described in 5.5.1) and by analytical gel filtration performed using precalibrated Hiload 16/60 Superdex G75 or G200 columns (flow rate: 1 mL/min; Gel Filtration Calibration Kits; GE Healthcare).

5.5.3 Glycosylation analysis

N-glycosylation was predicted using NetNGlyc 1.0 Server (http://www.cbs.dtu.dk/services/NetNGlyc/; Gupta et al., 2004). Deglycosylation of purified enzymes (5–7 μ g of enzymes) was carried out using Endoglycosidase H (New England BioLabs). The carbohydrate content was quantified by the phenol-sulfuric acid carbohydrate assay (Dubois et al., 1956) using glucose and mannose as standards.

5.6 Enzyme activity and kinetics

5.6.1 GH3 xylosidases

5.6.1.1 *para*-Nitrophenyl β -D-xylopyranoside and *para*-nitrophenyl α -L-arabinofuranoside

β-Xylosidase activity was assayed toward 2 mM *p*NPX at 40°C for 10 min in 50 mM sodium acetate pH 5.0 for BxlA or pH 5.2 for BxlB (50 µL reaction mixture). The reaction was terminated by addition of 1 M Na₂CO₃ (100 µL) and the absorbance of released *p*NP was measured at 410 nm, with a standard curve of 0.05–0.5 mM *p*NP, by a microtiter plate reader (Bio-Tek Instrument Inc., Winooski, VT). One unit was defined as the amount of enzyme liberating 1 µmol of *p*NP per min under the assay condition. Activity and kinetics of BxlB were determined by Dr. Hiroyuki Nakai. The effect of metal ions on activity of BxlA was determined by addition to a final concentration of 5 mM of CaCl₂, CuCl₂, MgCl₂, AgNO₃, or ZnCl₂ in the assay buffer.

The pH optimum of 71 nM BxlA and 79 nM BxlB was determined using 2 mM *p*NPX in 40 mM Britton-Robinson buffer (Britton and Robinson, 1931) in the pH range 2.0–10.0. The temperature optimum was determined in the range 25–80°C in 50 mM sodium acetate pH 5.0 (BxlA) or pH 5.2 (BxlB) using the same enzyme concentrations. The dependence of stability on pH and temperature was deduced from residual activity measured using β -xylosidase activity assay for 355 nM BxlA and 395 nM BxlB. The pH stability was analysed at 40°C for 5 h in 40 mM Britton-Robinson buffer pH 2.0–10.0, and the temperature stability was determined in 50 mM sodium acetate pH 5.0 at 25–80°C, 5 h.

The initial rates of hydrolysis towards 8 different concentrations of pNPX (0.1–8.0 mM) and pNPA (0.2–2.0 mM) were measured using 40 nM (for pNPX) and 465 nM (for pNPA) of BxlA, and towards 11 different concentrations of pNPX (0.1–8.0 mM) and pNPA (0.1–8.0 mM) using 66 nM (for pNPX) and 328 nM (for

*p*NPA) of BxlB. *p*NP was quantified spectrophotometrically at 410 nm using a standard curve of 0.05–0.5 mM *p*NP. Kinetic parameters, K_m and k_{cat} , were determined by fitting the Michaelis-Menten equation to the initial rate data using SigmaPlot v. 9.0.1 (Systat Software Inc. San Jose, CA). For *p*NPA having higher K_m than the highest substrate concentration used, the k_{cat}/K_m was calculated as slope of the plot of velocity per enzyme concentration (v/[E]) as a function of substrate concentration.

5.6.1.2 Xylooligosaccharides

The initial rates of hydrolysis toward 8 different concentrations of xylooligosaccharides (X2–X6, 0.4–6.0 mM) were measured using 32 nM (for X2–X5), and 35 nM (for X6) BxlA. Eleven different concentrations of X2 (1.0–8.0 mM), X3–X4 (0.8–8.0 mM), and X5–X6 (1.0–8.0 mM) were analysed using 66 nM (for *p*NPX and X2–X6) and 328 nM (for *p*NPA) BxlB. The hydrolytic activity towards xylooligosaccharides was assayed in 50 mM sodium acetate pH 5.0 for BxlA or pH 5.2 for BxlB (100 μ L reaction mixture). The reaction was terminated by addition of 2% *p*-bromoaniline (w/v) in glacial acetic acid with 4% thiourea (w/v) (500 μ L) and the mixture was incubated in the dark at 70°C for 10 min followed by incubating at 37°C for 60 min. The absorbance at 520 nm, using standard curve of 0.25–5.0 mM xylose, was measured on a microtiter plate reader (Bio-Tek Instrument Inc.), as the formation of furfural from pentoses in acetic acid containing thiourea with *p*bromoaniline acetate (Deschatelets and Yu, 1986; Roe and Rice, 1948). Kinetic parameters, *K*_m and *k*_{cat}, were determined by fitting the Michaelis-Menten equation to the initial rate data using SigmaPlot v. 9.0.1 (Systat Software Inc. San Jose, CA).

5.6.2 GH5 mannanases

5.6.2.1 Galactomannan and glucomannan

β-Mannanase activity was assayed toward 0.5% (w/v) of three mannans: konjac glucomannan, guar galactomannan, and locust bean gum galactomannan. All substrates (1 g) were dissolved in 6 mL 95% ethanol and stirred in 90 mL deionized water with heating until boiling, and the stirring was continued until the mixture was cooled. The volume of the solution was adjusted to 100 mL with deionized water. The enzyme (40 µL) was incubated with mannan solution (180 µL) and 200 mM sodium acetate pH 5.5 with 0.01% BSA (180 µL) at 37°C for 5 min. The reducing sugars liberated in the reaction mixture were determined by adding 3,5-dinitrosalicylic acid solution (600 µL), incubating at 95°C for 15 min, cooling on ice for 10 min, and measuring the absorbance at 540 nm (Miller, 1959; Mohun and Cook, 1962) in a 96 well microtiter plate (Bio-Tek Instrument Inc.). One unit was defined as the amount of enzyme liberating 1 µmol of mannose per min under the assay condition, using mannose as a standard.

5.6.2.2 Mannooligosaccharides

Initial rates of hydrolysis toward M4–M6 (0.2–6.0 mM) were measured using 60 nM (for M4), 30 nM (for M5), and 10 nM (for M6) of ManA, 75 nM (for M4), 15 nM (for M5), and 12 nM (for M6) of ManC, 50 nM (for M4), 28 nM (for M5), and 12

nM (for M6) of ManAS289W, and 120 nM (for M4), 32 nM (for M5), and 20 nM (for M6) of ManCW283S in 100 mM sodium acetate, 0.005% BSA, pH 5.5. The released reducing sugar was quantified by using copper bicinchoninate and measuring the absorbance at 540 nm using mannose as standard (McFeeters, 1980; Mori et al., 2001). Kinetic parameters, $K_{\rm m}$ and $k_{\rm cat}$, were determined by fitting to the Michaelis-Menten equation to the initial rate data using SigmaPlot v. 9.0.1 (Systat Software Inc).

5.6.2.3 Hydrolysis action patterns

The hydrolysis patterns towards 0.5 mM M4–M6 were investigated using the same enzyme concentration for ManA, ManAS289W, ManC, and ManCW283S as in the kinetic assay in 100 mM sodium acetate pH 5.5 at 37°C. Aliquots (6 μ L) were removed at appropriate intervals and added into 100 mM NaOH (294 μ L) to stop reaction. The reaction mixtures were quantified based on peak areas of high performance anion exchange chromatography with pulsed amperometric detection (HPAEC-PAD; ICS3000 system; Dionex Corporation, Sunnyvale, CA) equipped with CarboPac PA200 anion exchange column (3 × 250 mm with 3 × 50 mm guard column; Dionex) using isocratic elution of 37.5 mM NaOH for 35 min at 25°C, flow rate: 0.35 mL/min, calibrated with mannose and M2–M6.

5.6.3 GH10 xylananase

5.6.3.1 Xylans

β-Xylanase activity was assayed toward 1% (w/v) birchwood glucuronoxylan and 0.5% (w/v) wheat (low viscosity) and oat spelt arabinoxylans. All substrates (1 or 0.5 g) were dissolved in 80 mL of 50 mM sodium acetate pH 5.0 and heating until clear by microwaving and stirred until the mixture was cooled overnight. The volume of the solution was adjusted to 100 mL with acetate buffer. Enzyme (40 µL) was incubated with xylan solution (360 µL) at 40°C for 5 min. The released reducing sugar was determined by adding 3,5-dinitrosalicylic acid solution (600 µL), incubating at 95°C for 15 min, cooling, and measuring the absorbance at 540 nm in a 96 well microtiter plate (Miller, 1959 and Mohun and Cook, 1962). One unit was defined as the amount of enzyme liberating 1 µmol of xylose per min under the assay condition.

5.7 Transglycosylation screening and optimization

5.7.1 GH3 xylosidases

The transglycosylation reactions (1 mL) containing 200 nM BxIA, 30 mM pNPX and 200 mM acceptor candidate (xylose, arabinose, lyxose, L-fucose, L-rhamnose, glucose, galactose, mannose, talose, fructose, xylitol, sorbitol, maltose, sucrose, lactose, cellobiose, turanose, lactulose, L-cysteine, or L-serine) in 50 mM sodium acetate (10 mM for L-cysteine and L-serine reactions) pH 5.0, were incubated at 40°C for 5 h. The products were quantified based on peak areas of HPAEC-PAD (ICS-3000 IC system; Dionex Corporation) equipped with CarboPac PA200 anion exchange column (3 × 250 mm with 3 × 50 mm guard column; Dionex) using a linear 0–75 mM sodium acetate gradient in 100 mM NaOH for 35 min at 25°C, flow rate

0.35 mL/min) using xylobiose and xylotriose as standards for di- and trisaccharide products, respectively. The transglycosylation yield was estimated based on the *p*NPX concentration. Aliquots of 5 μ L were removed at appropriate intervals during five h and transferred to 100 mM NaOH (195 μ L) stop solution.

To evaluate the transglycosylation ability of xylosidase subsite +1 mutants, the time course was followed of the transglycosylation reaction (1 mL) of 200 nM C308W or 200 nM N313R, 30 mM *p*NPX and 200 mM acceptor (xylose, galactose, mannose, fructose, or maltose) in 50 mM sodium acetate pH 5.0, at 40°C for 16 h. The products were quantified as described above. Aliquots of 5 μ L were removed at appropriate intervals during 16 h and transferred to 100 mM NaOH (195 μ L) stop solution.

5.7.2 GH5 mannanases

To evaluate the transglycosylation ability of mannanase subsite +1 mutants, the time course of transglycosylation (1 mL) of 330 nM (ManA and ManAS289W) or 145 nM (ManC and ManCW283S) with 30 mM M4 (without acceptor) or 24 mM M4 with 100 mM M3 in 100 mM sodium acetate pH 5.5, were performed at 37°C for 150–300 min. The reaction mixtures were quantified using HPAEC-PAD (Dionex) as described above using isocratic elution of 50 mM NaOH for 35 min at 25°C (flow rate: 0.35 mL/min) using mannopentaose and mannohexaose as standards for penta-and hexasaccharide products, respectively. The transglycosylation yield was estimated based on the M4 concentration. Aliquots (5 μ L) were removed at 0, 2.5, 5, 10, 15, 20, 25, 30, 35, 40, 45, 60, 80, 100, 120, and 150 min for ManA and ManC; and at 0, 2.5, 5, 10, 20, 30, 45, 60, 90, 120, 150, 180, 210, 240, 270, and 300 min for ManAS289W and ManCW283S, and added to 100 mM NaOH (245 μ L) to stop the reaction.

To optimize the novel mannooligosaccharide production by ManC, time course transglycosylation reactions (1 mL) of 145 nM ManC containing 24 mM M4 with 200 mM isomaltotriose or melezitose in 100 mM sodium acetate pH 5.5, were performed at 37°C for 180 min. Aliquots (5 μ L) were removed at 0, 2.5, 5, 10, 15, 20, 25, 30, 35, 40, 45, 60, 80, 100, 120, 140, 160, and 180 min and added to 100 mM NaOH (245 μ L) to stop the reaction.

5.7.3 GH10 and GH11 xylananases

The transglycosylation reaction mixtures (1 mL), contained 1 μ M XlnA or XlnC, 20 mM X3 and 200 mM acceptor (xylobiose, cellobiose, maltose, isomaltose, lactose, mannobiose, xylitol, or sorbitol) in 50 mM sodium acetate pH 5.0, at 37°C for 5 h. The products were quantified based on peak areas of HPAEC-PAD (ICS-3000 IC system; Dionex Corporation) equipped with CarboPac PA200 anion exchange column (3 × 250 mm with 3 × 50 mm guard column; Dionex) using 0–75 mM sodium acetate linear gradient in 100 mM NaOH for 35 min at 25°C, flow rate: 0.35 mL/min) using xylotriose, xylotetraose, xylopentaose and xylohexaose as standards for tri-, tetra-, penta-, and hexasaccharide products, respectively. The transglycosylation yield was estimated based on the X3 concentration. Aliquots of 5 μ L were removed at 0, 5, 15, 30, 45, 60, 120, 180, 240, and 300 min and transferred to 100 mM NaOH (195 μ L) stop solution.

5.7.4 Glycosynthase reactions

Acetylated α -xylosyl and α -xylobiosyl fluorides received from Prof. Stephen G. Withers (University of British Columbia, Canada) were deacetylated in collaboration with Dr. Karin Mannerstedt, Prof. Monica Palcic, and Prof. Ole Hindsgaul (Carlsberg Laboratory) for glycosynthase reaction using the catalytic nucleophile mutants of GH3 xylosidase and GH10 xylanase, respectively. Tri-Oacetyl α -xylosyl fluoride (99.6 mg) and hepta-O-acetyl α -xylobiosyl fluoride (51.8 mg) was dissolved in dry methanol (10 and 15 mL, respectively) at 0°C with bubbling of NH₃ gas for 5 min. The reaction was stirred at 0°C without NH₃ gas until all the starting material was deacetylated (1 and 3 h, respectively), as monitored by thin layer chromatography (TLC; aluminium-coated silica gel 60 F₂₅₄, Merck) developed by CHCl₃/methanol (6:1, v/v for xylosyl fluoride) and ethylacetate/methanol/water (17:2:1, v/v for xylobiosyl fluoride), respectively, soaked with 8% sulfuric acid followed by tarring at 120°C. The deacetylated substrates were rotary evaporated without heating and dissolved in 5 mM ammonium bicarbonate pH 7.0 (to 100 mM final concentration). The α -xylosyl and α -xylobiosyl fluoride solutions were immediately frozen and stored at -80° C. The α -xylosyl and α -xylobiosyl fluorides are stable at -80° C. The purity of the fluoride substrates was analysed using electospray ionization mass spectra (ESI-MS; Bruker Esquire 3000-Plus Ion Trap instrument, Bruker Daltonik, Bremen, Germany).

GH3 β -xylosidase glycosynthase mutants D288A and D288C (50 μ M) were incubated with 2 mM α -xylosyl fluoride (α XF) as donor and 2 mM acceptor (*p*NP α -L-arabinopyranoside (*p*NPA*p*), lyxose, galactose, or cellobiose) in 100 mM sodium acetate (pH 5.0, pH 5.5, pH 6.0) or 100 mM phosphate buffer (pH 6.5, pH 7.0) at 30°C for 40 h. The reaction products were monitored by thin layer chromatography (TLC) developed by 100% acetonitrile (for reaction with *p*NPA*p*) or acetonitrile/water (85:15, v/v; for reaction with sugars), sprayed with 2% orcinol in sulphuric acid/water/ethanol (10:10:80, v/v/v) followed by tarring at 120°C.

GH10 xylanase glycosynthase mutant E244A 87 μ M was incubated with 10 mM α -xylobiosyl fluoride (α X2F) as donor and 5 mM X2 or X3 as acceptors in 100 mM sodium acetate pH 5.0 at 30°C for 24 h. Control reactions were performed by incubating the E244A mutant with either 10 mM α X2F donor or 5 mM of each acceptor to monitor hydrolytic activity of E244A and incubating 10 mM α X2F without E244A mutant to monitor stability of fluoride substrate during the reaction. The reaction products were monitored by TLC developed by acetonitrile/water (80:20, v/v; twice), sprayed with 2% orcinol solution followed by tarring at 120°C.

5.8 Oligosaccharides production by transglycosylation for structural analysis

5.8.1 GH3 xylosidases

Xylooligosaccharide products were obtained in 1 mL reaction mixtures containing 200 nM BxlA and 30 mM *p*NPX with 200 mM acceptor (xylose, arabinose, lyxose, L-fucose, glucose, mannose, talose, sucrose, or turanose) incubated at 40°C in 50 mM sodium acetate pH 5.0. The reaction time for each reaction was according to maximum transglycosylation yield obtained from 5.7.1. The reaction was terminated by heat inactivation (95°C, 10 min), centrifuged (18,000*g*, 4°C, 5 min), desalted (Amberlite MB-20; Fluka, Sigma Aldrich), and filtered (0.45 μ m nylon filter;

Frisenette Aps, Knebel, Denmark). The xylooligosaccharide products were purified using high performance liquid chromatography (HPLC; UltiMate 3000 Standard LC system; Dionex) equipped with a TSKgel Amide-80 column (5 μ m, 4.6 × 250 mm with 4.6 × 10 mm guard column; TOSOH, Tokyo, Japan) and refractive index detector (RI-101; Showa Denko, Kanagawa, Japan) at constant flow rate 1 mL/min of mobile phase (acetonitrile/water water 80:20 or 85:15, v/v) at 70°C. The purity was confirmed by TLC (aluminium-coated silica gel 60 F₂₅₄, Merck) developed by acetonitrile/water (80:20, v/v), sprayed with 2% orcinol solution followed by tarring at 120°C.

5.8.2 GH5 mannanases

Mannooligosaccharide products were obtained in 1 mL reaction mixtures containing 145 nM ManC and 24 mM M4 with acceptors; 100 mM mannotriose; 200 mM isomaltotriose or 200 mM melezitose incubated at 37°C (60 min for mannotriose, 120 min for isomaltotriose, 180 min for melezitose) in 100 mM sodium acetate, pH 5.5. The reaction was terminated by heat inactivation at 95°C for 10 min, centrifuged (18,000g, 4°C, 5 min), desalted (Amberlite MB-20; Fluka, Sigma Aldrich), and filtered (0.45 μ m nylon filter; Frisenette Aps). The mannooligosaccharide products were purified using HPLC (UltiMate 3000 Standard LC system; Dionex) equipped with a TSKgel Amide-80 column (5 μ m, 4.6 × 250 mm with 4.6 × 10 mm guard column; TOSOH) and refractive index detector (RI-101; Showa Denko) under a constant flow rate (1 mL/min) of mobile phase (acetonitrile/water 70:30, v/v) at 70°C. The purity was confirmed by TLC (aluminium-coated silica gel 60 F₂₅₄, Merck) developed by acetonitrile/water (70:30, v/v) twice, sprayed with 2% orcinol solution followed by tarring at 120°C.

5.9 Oligosaccharide characterisation

5.9.1 Nuclear magnetic resonance spectroscopy

500 µL of 1–2 mM purified xylooligosaccharide products were vacuum dried (Savant SVC100 SpeedVac Concentrator; Savant Instruments, Inc. Farmingdale, NY) or freeze dried (CoolSafe -55°C; ScanLaf A/S, Lynge, Denmark) and dissolved in deuterium oxide. Nuclear magnetic resonance spectroscopy (NMR) was performed in collaboration with Assoc. Prof. Charlotte H. Gotfredsen (DTU Kemi). The structure of transglycosylation products was determined by ¹H- and ¹³C-NMR analysis. Spectra were recorded on a Varian Unity Inova 500 MHz NMR spectrometer and a few on a Bruker Avance 800 MHz spectrometer (at the Danish Instrument Center for NMR Spectroscopy of Biological Macromolecules). A series of two-dimensional homo- and heteronuclear correlated spectra: Double-Quantum Filtered COrrelation SpectroscopY (DQF-COSY), Nuclear Overhauser Enhancement Spectroscopy (NOESY), gradient Heteronuclear Single Quantum Coherence (gHSQC), and gradient Heteronuclear Multiple Bond Correlation (gHMBC), were acquired using standard pulse sequences. The NMR data used for the structural assignment were acquired in D_2O . The chemical shifts were referenced to the solvent δ_H 4.78 ppm and the carbon shifts were references according to C5 of xylose at 66.1 ppm.

5.9.2 Electrospray ionization mass spectrometry

Molecular mass of xylooligosaccharide products as determined in collaboration with Ph.D. student Yvonne Westphal, Dr. Maaike Appeldoorn, and Prof. Henk A. Schols (Laboratory of Food Chemistry, Wageningen University, The Netherlands) by electrospray ionization mass spectrometry (ESI-MS) using an LTQ XL ion trap MS (Thermo Scientific, San Jose, CA). Samples were vacuum dried (Savant SVC100 SpeedVac Concentrator; BxlA) or freeze dried (CoolSafe -55°C; BxlB) and dissolved in milliQ water (100 μ g/mL) prior to introduction through a Thermo Accela UHPLC system equipped with a Hypercarb (100 × 2.1 mm, 3 μ m) column (Thermo Scientific) eluted with a gradient of deionized water/acetonitrile and 0.2% trifluoroacetic acid (0.4 mL·min⁻¹; 70°C). MS detection was performed in the positive mode using a spray voltage of 4.5 kV and a capillary temperature of 260°C and auto-tuned on glucohexaose (Westphal et al., 2010).

Electrospray quadrupole time-of-flight mass spectrometry (ESI-Q-TOF MS) was used for detecting the products from the reaction of L-cysteine and L-serine. The ESI-Q-TOF MS was performed in collaboration with Dr. Martin Zehl and Prof. Peter Roepstorff (Department of Biochemistry and Molecular Biology, University of Southern Denmark, Odense, Denmark). The sample mixtures (200 nM BxlA, 30 mM pNPX and 200 mM L-cysteine or L-serine in 10 mM sodium acetate pH 5.0, 40°C reacted for 60 min) were diluted 100 fold in isopropanol/water (1:1, v/v) and introduced into an orthogonal ESI ion source at a flow rate of 5 μ L/min. Measurements were performed using a Waters SYNAPT-HDMS System (Waters Corporation, Milford, MA) in positive ion V mode. CID fragment ions of Glu-Fibrinopeptide were used for calibration.

5.9.3 Matrix-assisted laser desorption/ionisation-time of flight mass spectrometry

The molecular masses of mannooligosaccharides were determined by Matrixassisted laser desorption/ionisation-time of flight mass spectrometry (MALDI-TOF MS) using an Ultraflex II TOF/TOF instrument (Bruker Daltonics GmbH, Bremen, Germany) equipped with all solid-state laser system. The mass spectrometer was selected for positive ions and analysed at 25040 V and detected using reflector mode. High (75%) laser power was required to obtain good spectra and at least 100 spectra were collected. The mass spectrometer was calibrated using mannopentaose, mannohexaose, and trypsin-digested β -lactoglobulin (mass range *m/z* 851–2313, as Na⁺ adducts for M5 and M6). The matrix solution was prepared by dissolving 10 mg 2,5-dihydroxylbenzoic acid in 1 mL of acetonitrile/water (30:70, v/v) containing 10 mM NaCl. The 0.2 mM (for M5 and manno-isomaltotrioses) or 0.1 mM (for M6 and manno-melezitoses) of samples (1 μ L) and matrix (1 μ L) were pipetted onto a MALDI-TOF-target plate (Bruker Daltonics) and the mixtures were air-dried prior to analysis.

5.10 GH61 putative endo-β-glucanase

5.10.1 Cultures of Aspergillus nidulans FGSC A4 on polymeric substrates

To demonstrate enzyme discovery by using the proteomic strategy and to investigate the function of GH61 putative endo- β -glucanase from *A. nidulans* FGSC A4 (see 4.4.3), cDNA cloning and recombinant production of this putative enzyme was performed. *A. nidulans* FGSC A4 was grown on minimal agar medium (0.6% NaNO₃, 0.052% KCl, 0.052% MgSO₄·7H₂O, 1 mL trace element, 12 mM phosphate buffer pH 6.8, 1.25% agar; Barratt et al., 1965) containing 1% barley β -glucan or oat spelt xylan at 30°C for 5 days. The spores were suspended in 0.02% TWEEN-80 and grown in minimal medium without agar at 30°C for 1 day with shaking at 130 rpm. The cells were harvested by decanting the culture supernatant.

5.10.2 cDNA cloning

Two g of fungal cells was placed in a liquid nitrogen-cooled mortar with addition of liquid nitrogen and ground with a pestle until the cells became a fine powder. The mRNA isolation from the crushed fungal cells was performed using FastTrack 2.0 Kit for isolation of mRNA (Invitrogen) and the cDNA library was prepared using SuperScript III First Strand Synthesis System for reverse transcriptase polymerase chain reaction (Invitrogen). The *eglF* gene was obtained by amplification of target cDNA using Expand High Fidelity DNA polymerase (Roche Diagnostics GmbH) and oligonucleotide primers (Table 5, page 32), constructed based on genomic sequence (Galangan et al., 2005). The original signal peptide was Met1-Ala20 predicted by SignalP 3.0 (http://www.cbs.dtu.dk/services/ SignalP/; Bendtsen et al., 2004). The PCR products were digested by XhoI and XbaI (New England BioLabs) and cloned into the pPICZaA vector (Invitrogen). The plasmids were transformed into Escherichia coli DH5a (Invitrogen) and transformants were selected on low salt LB supplemented with 25 µg/mL Zeocin. The purified plasmids (OIAgen plasmid midi kit; QIAgen, Hilden, Germany) were confirmed by full-length sequencing (Eurofins MWG Operon). The plasmids were linearised by PmeI and transformed into P. pastoris strain X-33 as described in 5.3.1.

5.10.3 Production and purification of GH61 putative endo-β-glucanase

P. pastoris X-33 transformant harbouring EgIF was grown in 3×1 L of BMGY at 30°C in 3×2.5 L plastic shake flask for 24 h to OD₆₀₀ of 32. Cells were harvested (3000g, 5 min, 4°C) and resuspended in 1 L BMM in 2.5 L plastic shake flask. The induction was continued at 22°C for 72 h. Methanol was supplemented to a final concentration of 1% (v/v) every 24 h until harvest by centrifugation (13,500g, 4°C, 1 h). The putative endo- β -glucanase was purified by anion exchange chromatography followed by gel filtration chromatography (as described in 5.4.1). Since this protein showed very low activity and had a low pI (4.0), anion exchange chromatography was used to avoid the effect of high pH as observed in case of BxIA (see 5.4.1) or metal inhibition, if any.

Endo- β -glucanase activity was assayed toward 1% (w/v) barley β -glucan and 0.5% (w/v) oat spelt arabinoxylans in 50 mM sodium acetate pH 4.0–6.0 using 3,5-dinitrosalicylic acid solution as described in 5.6.3.1.

5.10.4 Carbohydrate microarray analysis

Carbohydrate microarray analysis was performed in collaboration with Dr. Henriette Lodberg Pedersen and Prof. William G. T. Willats (Department of Plant Biology and Biotechnology, Faculty of Life Sciences, University of Copenhagen, Denmark). The carbohydrate microarrays contain different mono-, oligo-, and polysaccharides extracted from plant cell walls (Fig. 10). The monoclonal antibodies against specific plant cell wall carbohydrate epitopes were probed to detect any epitope deletion as a result of enzyme hydrolysis (5.10.4.1; Moller et al., 2007; Øbro et al., 2009) or anti-His antibody were probed to detect possible binding of the His-tagged protein to carbohydrates in the arrays (5.10.4.2). The experimental scheme is shown in Fig. 11.

5.10.4.1 Screening for EglF activity on carbohydrate microarrays *via* postprint epitope deletion

Arrays were initially blocked for 1 h at ambient temperature in 5% low fat milk powder (MP) in phosphate buffered saline (PBS: 140 mM NaCl, 2.7 mM KCl, 10 mM Na₂HPO₄, 1.7 mM KH₂PO₄, pH. 7.5). After blocking, the MP-PBS solution was discarded and the arrays were divided into two batches; one batch was probed with a panel of plant cell wall directed monoclonal antibodies (from PlantProbes, University of Leeds) in MP-PBS for 2 h at ambient temperature. The arrays were washed and probed with alkaline phosphatase conjugated anti-rat monoclonal antibody in MP-PBS for 2 h at ambient temperature followed by a second round of washing $(3 \times 10 \text{ min in PBS}, 1 \times 2 \text{ min H}_2\text{O})$. The arrays were finally submerged for 5-10 min in AP buffer (100 mM Tris-base, 100 mM NaCl, 5 mM MgCl₂, pH. 9.5) containing nitro blue tetrazolium (NBT) and 5-bromo-4-chloro-3-indolyl phosphate (BCIP). Binding of the antibodies was visualized as black spots, were binding strength correlates to spot intensity (pixels per inch, ppi). The second batch of arrays was incubated with EglF for 24 h at 25–28°C before the arrays were probed with the same panel of plant cell wall directed monoclonal antibodies as used for the first batch of arrays. Epitope deletion would be observed as a change in spot intensity between two arrays probed with the same antibody, but where one of the arrays initially were incubated with enzyme prior to antibody labeling (Moller et al., 2007; Øbro et al., 2007; Øbro et al., 2009).

The experiment was executed 4 times:

- 1. The second batch of arrays was incubated with EglF (10 μ g/mL) in MES buffer (20 mM MES, 0.5 mM MgCl₂, pH 6.0)
- 2. The second batch of arrays was incubated with EglF (10 μg/mL) in HEPES buffer (20 mM HEPES, 0.5 mM MgCl₂, pH 7.0)
- 3. The second batch of arrays was incubated with EglF (100 μg/mL) in HEPES buffer (20 mM HEPES, 0.5 mM MgCl₂, pH 7.0)
- 4. The second batch of arrays was incubated with EglF (100 μg/mL) in acetate buffer (100 mM sodium acetate, 0.5 mM MgCl₂, pH 4.5)

The plant cell wall directed mAbs used in the experiments were selected such that they collectively recognized the glycan structures presented on the carbohydrate microarrays.

5.10.4.2 Screening for EglF binding to carbohydrate microarrays

Two arrays were blocked in 5% MP in HEPES buffer (20 mM HEPES, 0.5 mM MgCl₂, pH 7.0) or MES buffer (20 mM MES, 0.5 mM MgCl₂, pH 6.0) for 1 h at ambient temperature. The blocking solution was discarded and the arrays were probed with EglF (20 μ g/ml) in MP-buffer solution (the buffer corresponding to the buffer used for blocking) at ambient temperature. After 2 h, the arrays were washed (3 × 10 min in HEPES or MES buffer) and probed with alkaline phosphatase conjugated anti-His antibody in MP-buffer for 2 h at ambient temperature. Finally, the arrays were washed (3 × 10 min buffer, 1 × 2 min H₂O) and developed in MP-buffer (pH 9.5) containing NBT and PBS for 20 min. Control arrays, where MP-buffer was applied instead of EglF solution, were included to reveal possible binding of the anti-His antibody to the epitopes displayed on the arrays.



1

Oligosaccharides - BSA conjugated

AI	α -(I \rightarrow 4)-D-heptagalacturonate (DE 0%)
BI	α -(1 \rightarrow 4)-D-pentagalacturonate (DE 0%)
CI	β -D-galactose
DI	β -(I \rightarrow 4)-D-galactotetraose
A2	α-L-arabinose
B2	α -(1 \rightarrow 5)-L-arabinobiose
C2	α -(1 \rightarrow 5)-L-arabinotriose
D2	α -(1 \rightarrow 5)-L-arabinotetraose
A3	α -(1 \rightarrow 5)-L-arabinopentaose
B3	α -(1 \rightarrow 5)-L-arabinohexaose
C3	α -(1 \rightarrow 5)-L-arabinoheptaose
D3	β -(I \rightarrow 4)-D-glucopentaose
A4	β-D-mannose
B4	β -(1 \rightarrow 4)-D-mannotriose
C4	β -(I \rightarrow 4)-D-mannotetraose
D4	β -(I \rightarrow 4)-D-mannopentaose
A5	β-D-xylose
B5	β -(I \rightarrow 4)-D-xylotriose
C5	β -(I \rightarrow 4)-D-xylotetraose
D5	β -(I \rightarrow 4)-D-xylopentaose
A6	Xyloglucan heptamer XXX(G)-
B6	Xyloglucan octamer XXL(G)-
C6	Xyloglucan nonamer XLL(G)-
D6	6^{I} - α -D-galactosyl- β -(I \rightarrow 4)-D-mannobiose
A7	$(1\rightarrow 3),(1\rightarrow 4)-\beta$ -D-glucotriose [G4G3G(G)-]
B7	$(1\rightarrow 3), (1\rightarrow 4)-\beta$ -D-glucotetraose [G3G4G4G(G)-]
C7	$(1\rightarrow 3), (1\rightarrow 4)-\beta$ -D-glucopentaose [G3G4G3G4G(G)-]
D7	6 ³ ,6 ⁴ -di galactosyl-β-(1→4)-D-mannotetraose
A8	α -(1 \rightarrow 4)-D-glucobiose
B8	α -(I \rightarrow 4)-D-glucopentaose
C8	α -(1 \rightarrow 6)-D-glycosyl- α -(1 \rightarrow 4)-D-maltotriose G6G4G
D8	α -(1 \rightarrow 6)-D-glycosyl- α -(1 \rightarrow 4)-D-maltosyl-maltose
	G6G4G4G6G4G
A9	β-D-glucose
B9	β -(I \rightarrow 3)-D-glucotriose
C9	β -(I \rightarrow 3)-D-glucotetraose
D9	β -(1 \rightarrow 3)-D-glucopentaose

* DE; degree of methyl esterification

Polysaccharides

EI	Lime pectin, DE 11% (Danisco #F11)
FI	Lime pectin, DE 43% (Danisco #F43)
GI	Lime pectin DE 0% (Danisco #B00)
HI	Lime pectin, DE 16% (Danisco #P16)
П	Sugar beet pectin (Danisco #SBP6232)
E2	Pectic galactan #1 (lupin, INRA/Nantes)
F2	Pectic galactan #2 (potato, INRA/Nantes)
G2	Pectic galactan #3 (lupin, Megazyme)
H2	Pectic galactan #4 (tomato, J. Paul Knox)
12	RGII enriched pectin (red wine)
E3	RGI #1 (soy bean, Megazyme)
F3	RGI #2 (carrot, INRA/Nantes)
G3	RGI #3 (sugar beet, INRA/Nantes)
H3	RGI #4 (Arabidopsis, INRA/Nantes)
13	Seed mucilage (Arabidopsis, INRA/Nantes)
E4	Arabinan (sugar beet, Megazyme)
F4	Arabinan, de-branched (sugar beet, Megazyme)
G4	$(I \rightarrow 5)$ -alpha-L-arabinan (sugar beet, Megazyme)
H4	RGI #5 (potato, Megazyme)
14	Xylogalacturonan, XGA (apple, INRA/Nantes)
E5	Xylan (birch wood, Sigma-Aldrich)
F5	Arabinoylan (wheat, Megazyme)
G5	4-Methoxy-glucoronoarabinoxylan (birch, Sigma-Aldrich)
H5	Xyloglucan, non fucosylated (tamarind seed, Megazyme)
15	Xyloglucan, fucosylated (pea, INRA/Nantes)
E6	Carboxymethyl-cellulose, CM-cellulose (Avicel)
F6	Hydroxymethyl-cellulose, MeO-cellulose (Avicel)
G6	Hydroxyethyl-cellulose, EtO-cellulose (Avicel)
H6	$(1\rightarrow 6), (1\rightarrow 3)-\beta$ -D-glucan (laminarin, Sigma-Aldrich)
16	$(1 \rightarrow 3)$ - β -D-glucan (pachyman)
E7	β-glucan #I (lichenan, icelandic moss, Megazyme)
F7	β-glucan #2 (barley, Megazyme)
G7	β-glucan #3 (oat, Megazyme)
H7	β-glucan #4 (yeast)
17	$(1 \rightarrow 6), (1 \rightarrow 4) - \alpha - D$ -glucan (pullulan)
E8	Glucomannan
F8	Galactomannan
G8	Gum guar
H8	Gum Arabic (Sigma-Aldrich)
18	Locust bean gum (Megazyme)
E9	Xanthan gum (Danisco #R80)
F9	Carrageenan #1 (Danisco #100)
G9	Carrageenan #2 (Danisco #201)
H9	Alginate, sodium salt (Danisco #FD-155)
	- ` '

¹⁹ Sugar beet arabinan (Megazyme)

Figure 10 Layout of carbohydrate microarrays (adapted from Øbro et al., 2009) figure courtesy of Dr. Henriette Lodberg Pedersen.



Figure 11 Overview of the experimental approach for carbohydrate microarray analysis. A) Screening for EglF activity on carbohydrate microarrays *via* post-print epitope deletion, B) screening for EglF binding to carbohydrate microarrays. 1° mAbs, plant cell wall directed monoclonal antibodies (from Plant Probes, Leeds University); 2° mAb, alkaline phosphatase conjugated anti-rat monoclonal antibody; anti-His Ab, alkaline phosphatase conjugated anti-His antibody.

6. **Results**

6.1 GH3 xylosidases

6.1.1 Production and purification

The β -xylosidase BxlA from *A. nidulans* FGSC A4 was produced recombinantly as His-tag fusion protein in *P. pastoris* X-33 secreated as directed by the α -factor secretion signal peptide. BxlA was purified by anion exchange chromatography (Fig. 12A,B) followed by gel filtration chromatography (Fig. 12C,D) in a yield of 16 mg/L culture.



Figure 12 Purification of BxlA. A) anion exchange chromatogram. B) SDS-PAGE of BxlA purified using a Resource Q 6 mL column; 15 μ L samples loaded. C) gel filtration chromatogram of pooled BxlA fractions from (A). D) SDS-PAGE of BxlA purified by Hiload 26/60 Superdex G200 column; 20 μ L samples loaded; marker, Mark12 unstained protein standard marker (Invitrogen). The red bars in (A) and (C) show the fractions which are analysed by SDS-PAGE in (B) and (D) (see 5.4.1).

6.1.2 Protein characterisation

BxlA migrated as a single band in SDS-PAGE with an estimated size of 110 kDa (Fig. 13A). This value is larger than the theoretical mass of $M_r = 86045.2$ and Endoglycosidase H treatment reduced the molecular mass to 86 kDa (Fig. 13A). BxlA contains 10 possible *N*-glycosylation sites and the carbohydrate content was estimated to 29% and 17% by phenol-sulphuric method using glucose and mannose as standards, respectively. These contents are in excellent agreement with the apparent molecular mass found by SDS-PAGE. Gel filtration gave a value of 220 kDa (Fig. 13B) indicating that BxlA is a dimer in solution. The isoelectric point (pI) of BxlA was 3.0 (Fig. 13C) resembling native BxlA (called XlnD) β -xylosidases from *A. nidulans* having values of 3.4 (Kumar and Ramon, 1996).



Figure 13 Molecular mass determination using SDS-PAGE and gel filtration, deglycosylation, and pI of BxIA. A) SDS-PAGE of purified and deglycosylated BxIA; left lane, marker, Mark12 unstained protein standard marker (Invitrogen); lane (A) 7 μ g BxIA; lane (B) 7 μ g deglycosylated BxIA, endoglycosidase H presented as a band at 29 kDa. B) calibration of superdex G200 16/60 column (open circle), BxIA has molecular weight of 220 kDa (filled square). C) isoelectric focusing of 400 μ g purified BxIA using polyacrylamide PhastGel IEF (pH 4–6.5) with low range pI (pH 2.5–6.5) calibration standard (GE Healthcare).

6.1.3 Enzymatic specificity and kinetics

BxlA had a specific activity of 32 ± 1.6 U/mg and 2 ± 0.2 U/mg towards 2 mM *p*NPX and *p*NPA, respectively. Among five tested metal ions, BxlA was inhibited importantly by Ag²⁺ and Cu²⁺ (Table 6). The pH optima for hydrolysis of *p*NPX by BxlA and BxlB (sharing 45% sequence identity, see 4.2.5 and Table 4) were 5.0 and 5.2, respectively (Fig. 14A). The enzymes were stable at pH 4.0–6.0 and 4.0–8.0 (>95% residual activity), respectively (Fig. 14B). BxlA and BxlB showed maximum activity at 50 and 60°C (Fig. 14C) and retained activity up to 40 and 55°C after 5 h incubation (>95% residual activity), respectively (Fig. 14D).

Table 6 Effect of metal ions on the hydrolytic activity towards pNPX of BxlA.

5 mM	Relative activity (%)				
Metal salt	BxlA	XlnD ^a			
CaCl ₂	100	111			
CuCl ₂	35	9			
MgCl ₂	100	114			
AgNO ₃	1	7			
ZnCl ₂	100	50			

^a From Kumar and Ramón, 1996



Figure 14 Effect of pH and temperature on activity and stability of purified BxlA (black circle) and BxlB (white square). A) pH-dependence for hydrolysis of *p*NPX by 71 nM BxlA and 79 nM BxlB in 40 mM Britton-Robinson buffer; B) pH-stability of 355 nM BxlA and 395 nM BxlB in 40 mM Britton-Robinson buffer and 5 h incubation; C) temperature-activity dependence for 71 nM BxlA and 79 nM BxlB at $25-80^{\circ}$ C with 10 min reaction; D) stability of 355 nM BxlA and 395 nM BxlB in the temperature range $25-80^{\circ}$ C for 5 h. Each experiment was made in triplicate. Standard deviations are shown as error bars

Kinetics hydrolysis parameters of **BxlA** and **BxlB** towards xylooligosaccharides (X2–X6), pNPX, and pNPA are shown in Table 7. Among the xylooligosaccharides, both enzymes have highest catalytic efficiency (k_{cat}/K_m) towards xylobiose, whereas the efficiency decreased slightly with longer oligosaccharides due to small decreases and increases in k_{cat} and K_m , respectively. The catalytic efficiency towards pNPX by BxlA was slightly higher than by BxlB for xylobiose, while BxlB showed the same catalytic efficiency towards these substrates. Both enzymes hydrolysed pNPA albeit with two and one orders of magnitude lower efficiency, respectively, compared to pNPX. BxlA and BxlB have poor affinity for pNPA and only k_{cat}/K_m values were determined.
		BxlA			BxlB	
	$k_{\rm cat}$	K _m	$k_{\rm cat}/K_{\rm m}$	$k_{\rm cat}$	K _m	$k_{\rm cat}/K_{\rm m}$
Substrate	s^{-1}	mM	$s^{-1} mM^{-1}$	s ⁻¹	mM	$s^{-1} mM^{-1}$
X2	139 ± 7.8	2.0 ± 0.07	70	36 ± 0.7	2.5 ± 0.40	14
X3	122 ± 7.8	2.1 ± 0.14	58	28 ± 0.1	3.0 ± 0.06	9
X4	104 ± 1.1	2.5 ± 0.11	42	25 ± 0.2	3.2 ± 0.15	8
X5	93 ± 4.7	2.8 ± 0.15	33	23 ± 0.2	4.3 ± 0.15	5
X6	65 ± 1.5	3.0 ± 0.14	22	22 ± 0.5	5.2 ± 0.58	4
pNPX	112 ± 1.3	1.3 ± 0.16	90	18 ± 0.4	1.3 ± 0.11	14
pNPA	_	_	0.9	_	_	1.1

Table 7 Kinetic parameters of BxlA and BxlB for hydrolysis of xylooligosaccharides, *p*NPX, and *p*NPA.

Standard deviations were calculated from triplicate experiments (see 5.6.1.1 for experimental details)

6.1.4 Transglycosylation

The acceptor specificity in transglycosylation for BxlA was investigated using *p*NPX as donor with 10 monosaccharides, six disaccharides, two sugar alcohols, and two amino acids as candidate acceptors. Progress of transglycosylation using 30 mM *p*NPX donor with 200 mM acceptor was monitored by HPAEC-PAD during 5 h (Fig. 15). In many transglycosylation reactions, more than one transglycosylation products were formed as detected by HPAEC-PAD (Fig. 16). The mixed products may be due to the formation of products with different linkages, different forms (α , β , furanosyl, pyranosyl), or longer chains (in case the enzyme uses a transglycosylation product as acceptor in a second transglycosylation reaction), therefore the total product yields (products/donor, mM/mM) for each acceptor was estimated by the maximum formation of products during the observation period (Table 8).

All tested monosaccharides except L-rhamnose were accommodated as acceptors by BxIA, which showed preference, however, for mannose, talose, lyxose, and xylitol, as gave similar yield to xylose and was less efficient in accepting glucose, galactose and fructose. BxIA poorly used disaccharides as acceptors, even though it showed preference to use turanose and sucrose to lactulose, maltose, and cellobiose, and could not use lactose as acceptor



Figure 15 Progress of transglycosylation with different acceptors catalysed by BxIA. Reactions containing 200 nM BxIA, 30 mM *p*NPX, and 200 mM acceptor; A) acceptor: xylose (\blacktriangle), arabinose, (Δ), lyxose (\blacklozenge), xylitol (\bullet), sorbitol (\circ); B) acceptor: glucose (\blacksquare), galactose (\square), mannose (\blacktriangledown), talose (∇), fructose (\blacklozenge), L-fucose (\diamondsuit); C) acceptor: maltose (\blacksquare), sucrose (\blacklozenge), cellobiose (\square), turanose (\blacklozenge), lactulose (\circ).



Figure 16 HPAEC-PAD chromatograms monitoring transglycosylation with selected acceptors to illustrate different product formations; A) xylose; B) glucose; C) mannose; D) sucrose; black line, reaction time: 0 min; blue line, reaction time: 10 min; magenta line; reaction time: 60 min for glucose, mannose, sucrose, and 120 min for xylose.

		Production	No. of major
Acceptor	Yield (%)	(mg/mL)	TG peaks
Monosaccharides			
D-xylose	21	1.81	1
D-arabinose	17	1.41	2
D-lyxose	23	1.92	3
L-fucose	14	1.27	3
L-rhamnose	n.d.	n.d.	n.d.
D-glucose	10	0.92	1
D-galactose	9	0.81	1
D-mannose	25	2.34	3
D-talose	22	2.04	3
D-fructose	8	0.79	2
Sugar alcohols			
Xylitol	36	3.05	2
Sorbitol	20	1.85	2
Disaccharides			
Maltose	2	0.27	2
Sucrose	6	0.81	2
Lactose	n.d.	n.d.	n.d.
Cellobiose	2	0.33	1
Turanose	7	0.95	2
Lactulose	3	0.44	2

Table 8 Maximum transglycosylation yields with different acceptors.

Products were quantified by HPAEC-PAD using xylobiose and xylotriose as standards for di- and trisaccharides, respectively. The total yield was estimated as percentage of product/donor concentration (mM/mM); n.d., not detected. Products in bold were purified for structural analysis.

Two amino acids, L-cysteine and L-serine, were also tested as acceptors; the formation of Xyl-Cys and Xyl-Ser could not be clearly monitored either by TLC or by HPAEC-PAD, but have been evaluated by ESI-Q-TOF MS (Fig. 17). The peak of cysteine from HPAEC-PAD chromatogram was broadened over time and overlapped with the transglycosylation products (Fig. 17A). Xyl-Cys could be detected in the form of Xyl-Cys+H (m/z 254), Xyl-Cys+Na⁺ (m/z 276), and Xyl-Cys-H+2Na⁺ (m/z298), whereas the formation of Xyl-Ser as Xyl-Ser+Na⁺ (m/z 260) could not be confirmed by MS/MS. The thio-glycoside linkage of Xyl-Cys was additionally supported by the lack of reaction with iodoacetamide, which would form Scarbamidomethylcysteine with а free thiol. ESI-MS showed no *S*carbamidomethylation of Xyl-Cys, while S-carbamidomethyl cysteine formed with the L-cysteine itself (data not shown).



Figure 17 HPAEC-PAD chromatogram and ESI-Q-TOF mass spectra (m/z 50-320) of Xyl-Cys transglycosylation reaction mixture containing 200 nM BxlA, 30 mM *p*NPX, and 200 mM L-cysteine in 10 mM sodium acetate pH 5.0 at 40°C. A) HPAEC-PAD chromatogram. Black, reaction time: 0 min; blue, reaction time: 30 min; magenta; reaction time: 60 min; brown, reaction time: 120 min; B) reaction with L-cysteine at 0 min; C) reaction with L-cysteine at 60 min reaction time; D) ESI-Q-TOF MS/MS spectrum of Xyl-Cys at m/z 276; E) ESI-Q-TOF MS/MS spectrum of Xyl-Cys at m/z 276; E) ESI-Q-TOF MS/MS spectrum of Xyl-Cys at m/z 298 (see Appendix Figure 1 for details in Fig. 17B-E, page 113–114).

6.1.5 Characterisation of xylosyl-oligosaccharide products

Among the tested transglycosylations, nine products (from xylose, arabinose, lyxose, L-fucose, glucose, mannose, talose, sucrose, and turanose) formed by BxlA were purified by HPLC and subjected to structure-determination by two-dimensional nuclear magnetic resonance spectroscopy (2D NMR) and ESI-MS. The molecular masses for all products gave signals corresponding to the calculated molecular masses of Li^+ adducts of the acceptor conjugate of xylose. Chemical shifts for NMR linkage analysis were assigned based on 2D NMR spectra and chemical shifts of selected

products are presented in Table 9. The xylobiose formation using xylose as well as the transglyocsylation products from arabinose and lyxose acceptors were linked mainly with β -1,4 linkages as identified by long range proton-carbon bond correlations between the non-reducing anomeric proton and C-4 of the substituted position and as confirmed by inter NOE correlation. In contrast the transglycosylation products from mannose and glucose acceptors linked mainly with β -1,6 linkages as identified by the downfield shift of the C6 or by inter NOE correlation across the glycosidic bond. The presence of β -1-6 linkages might be because the 6-OH as primary alcohol is less steric hindered and the enzymes having a loose acceptor binding site. The preference for a primary alcohol is in agreement with the β -xylosidase from *A. awamori* (Kurakake et al., 2005). In case of disaccharide acceptors, the xylosyl residues could form β -1,4 or β -1,6 linked to either reducing or non-reducing ends of the disaccharides, which may depend on the accessibility or less steric hindrance configuration of the disaccharide acceptors.

The structures of four novel transglycosylation products including: β -D-xylopyranosyl-(1 \rightarrow 4)-D-lyxopyranoside, β -D-xylopyranosyl-(1 \rightarrow 4)- α -D-glucopyranosyl-(1 \rightarrow 3)- β -D-fructofuranoside, β -D-xylopyranosyl-(1 \rightarrow 4)- α -D-glucopyranosyl-(1 \rightarrow 2)- β -D-fructofuranoside, and β -D-xylopyranosyl-(1 \rightarrow 6)- β -D-fructofuranosyl-(2 \rightarrow 1)- α -D-glucopyranoside have been assigned by 2D NMR (Table 9). The structures of transglycosylation products from L-fucose and talose as acceptors, which are also novel xylosyl-oligosaccharides cannot be solved by NMR due to the presence of mixtures of compounds, but ESI-MS confirmed that they are disaccharides with molecular masses of m/z 303 and 319, respectively.

Compound	Chemical shifts (δ, p.p.m.)					
	H-1	H-2	H-3	H-4	H-5a/H-5b	H-6a/H-6b
(molecular mass) ¹	C-1	C-2	C-3	C-4	C-5	C-6
β-Xyl <i>p</i> -(1,4)-	4.43	3.23	3.4	3.6	3.95/3.28	-
	102.7	73.6	76.4	70.1	66.1 ²	-
α-Xyl <i>p</i>	5.16	3.52	n.d.	n.d.	3.80/3.73	-
	92.9	72.2	72	74.3	59.7	-
β- Χ γΙ <i>ρ</i>	4.56	3.22	3.52	3.76	3.35/4.04	-
(<i>m</i> / <i>z</i> 289)	97.3	~74.9	~74.9	77.3	63.8	_
β-Xyl <i>p</i> -(1,4)-	4.43	3.32	3.42	3.62	3.26/3.93	-
	102.3	73.8	76.6	70.2	66.1 ²	-
α-Arap	5.22	3.79	3.91	4.1	3.78/3.96	-
	93.4	n.d.	n.d.	77.2	61	-
β-Ara <i>p</i>	4.53	3.48	3.68	4.05	3.60/4.00	-
(<i>m</i> / <i>z</i> 289)	97.3	73	72.5	76.8	64.6	-
β-Xyl <i>p</i> -(1,4)-	4.47	3.26	3.42	3.6	3.30/3.94	-
	103.6	73.7	76.6	70.2	66.1 ²	-
α-Lyx <i>p</i>	4.95	3.74	4.03	3.91	3.80/3.88	-
	94.9	70.8	~70.2	77	62.4	-
β-Lyx <i>p</i>	4.88	3.92	n.d.	n.d.	n.d.	-
(<i>m</i> / <i>z</i> 289)	95.2	n.d.	n.d.	n.d.	n.d.	-
β-Xyl <i>p</i> -(1,6)-	4.43	3.28	3.42	3.6	3.31/3.94	-
	104.4	73.9	76.4	70.2	66.1 ²	-
α-Man <i>p</i>	5.14	3.91	3.81	3.69	3.93	3.85/4.09
	95	~72.1	~71	~67.6	~72.2	~69.9
β-Man <i>p</i>	4.87	3.92	3.63	3.59	3.49	3.82/4.13
(<i>m</i> / <i>z</i> 319)	94.6	71.8	~73.8	70.1	~76	~69.9
β-Xyl <i>p</i> -(1,6)-	4.42	3.27	3.43	3.6	3.29/3.93	-
	104.2	73.9	76.6	70	66.1 ²	-
α-Glc <i>p</i>	5.19	3.5	3.68	3.4	3.93	3.84/4.10
	92.9	72.3	73.3	70.5	71.1	~69.8
β-Glc <i>p</i>	4.61	3.22	3.45	3.94	3.58	3.80/4.12
(<i>m</i> / <i>z</i> 319)	96.9	75	76.5	71.3	76	~69.9
β-Xyl <i>p</i> -(1,4)-	4.43	3.33	3.42	3.63	3.93/3.26	-
	102.1	73.6	70.2	70.1	66.1 ²	-
α-Glc <i>p</i> -(1,3)-	5.31	3.56	3.72	3.4	3.71	3.83/3.77
	101.5	73	73.7	76.5	73.7	61.4
β-Fru <i>f</i>	3.48/3.73	_	3.97	4.16	4.12	3.84/3.97
(<i>m</i> / <i>z</i> 481)	64.7	n.d.	77	70.2	70.2	61.8
β-Xyl <i>p</i> -(1,4)-	4.41	3.27	3.41	3.61	3.96/3.28	-
	104.2	74	76.6	70.1	66.1 ²	-
α-Glc <i>p</i> -(1,2)-	5.37	3.55	3.81	3.62	3.93	3.82/3.88
	92.9	71.9	71.9	78.8	72.1	60.3
β-Fru <i>f</i>	3.66/3.59	-	4.18	3.99	3.85	3.75/3.80
(<i>m</i> / <i>z</i> 481)	62.1	104.6	77.3	74.9	82.2	63.4
β-Xyl <i>p</i> -(1,6)-	4.47	3.27	3.42	3.59	3.94/3.29	-
	104.3	74	76.5	70.1	66.1 ²	-
β-Fru <i>f</i> -(2,1)-	n.d.	-	4.18	4.03	4.02	3.95
	n.d.	104.9	77	75.4	81	n.d.
α-GICp	5.37	3.52	3.73	3.44	3.81	3.78/3.82
(<i>m</i> / <i>z</i> 481)	93.2	72	n.d.	n.d.	n.d.	n.d.

Table 9 ¹H and ¹³C NMR data assignments and molecular masses for β -xylosyloligosaccharides produced by enzymatic transglycosylation of BxlA.

^a molecular mass as Li⁺ adduct (estimated mass + m/z 7) determined by ESI-MS ^b reference carbon shifts according to C5

n.d., not detected.

6.1.6 Engineering of transglycosylation activity by glycosynthase and subsite +1 mutants

The first reported GH3 structure, exo-1,3/1,4-β-glucanase from Hordeum vulgare (ExoI; Fig 18A; Varghese et al., 1997) has two domains: a TIM barrel (domain 1) and a six-stranded β -sandwich (domain 2), whereas the structure of β glucosidase from Thermotoga neapolitana DSM 4359 (Bgl3B; Fig 18B; Pozzo et al., 2010) has three domains: a TIM barrel (domain 1), a five-stranded α/β sandwich (domain 2), and a fibronectin type III domain (domain 3). No structure is reported of GH3 B-xylosidases, however, small-angle X-ray scattering data of a GH3 Bxylosidase of Trichoderma reesei, indicates three domains (Seidle et al., 2005). The 3D models of BxlA have two (Fig. 18C) and three domains (Fig. 18D) corresponding to the template structures of ExoI and Bgl3B. The quality of the protein models is reasonable to good, as verified by CE method and ProQ (see 5.2, page 33). In general, domain 1 is well conserved among GH3 enzymes and the catalytic nucleophile of all query sequences align together in this domain. Domains 2 and 3 have no highly conserved sequence motifs, which make it difficult to unambiguously identify the catalytic acid/base residue; therefore these were identified according to the templates of each homology modelling.

The subsite +1 mutants C308W and N313R on domain 1 in BxlA were designed based on the alignment of BxlA with related GH3 enzymes (Fig. 18 and Appendix Fig. 2, page 115). Cys308 of BxlA is situated next to its catalytic nucleophile Asp307, whereas in case of GH3 β-glucanase and β-glucosidase, the corresponding residue was Trp (Trp 286 for ExoI and Trp243 for Bgl3B) that formed a stacking interaction to the sugar ring at acceptor binding site (subsite +1) as shown for β-glucanase from barley (Hordeum vulgare) ExoI (PDB ID: 1IEX; Hrmova et al., 2005). However, Cvs308 was predicted to form a disulfide with Cvs340 (see 5.2). The substitution of Cys308 to Trp in BxlA (C308W) could increase transglycosylation activity of BxIA by providing a stacking interaction at subsite +1 or it could cause the loss of a disulfide in BxIA and affect its structural integrity. Asn313 of BxIA is equivalent to Arg291 of ExoI, which can form a hydrogen bond to 6-OH of the sugar ring at subsite +1, however it is equivalent to Asn248 of Bgl3B, which located far (>15Å) from the active site and may not be involved in acceptor interaction. The substitution of Asn313 by Arg in BxlA (N313R) was made to investigate the role of this residue and in an attempt to increase the transglycosylation activity of BxlA towards hexose acceptors.

The production and purification of BxlA glycosynthase and subsite +1 mutants were done essentially as for BxlA (see 5.4.1) with yields of 9, 13, 11, 8, 11, and 12 mg/L culture for D307G, D307A, D307C, D307S, C308W, and N313R BxlA, respectively. The specific activities of the mutants are shown in Table 10. The activity of the glycosynthase mutants and C308W decresed by three orders of magnitude, whereas that of N313R was decreased three fold. The severe reduction in activity of C308W indicated that Cys308 is essential for BxlA. This could be because Cys308 forms disulfide bond with Cys340 (3.3 Å from Cys308) as found by using the DiANNA-Disulfide bond connectivity prediction.



Figure 18 Cartoons representing the active site in 3D structures and homology models of: A) exo-1,3/1,4- β -glucanase from *H. vulgare* (ExoI; PDB ID: 1IEX), B) β -glucosidase from *T. neapolitana* DSM 4359 (Bgl3B; PDB ID: 2WT3), C) model structure of BxlA (using 1IEX as template), D) model structure of BxlA (using 2WT3 as template; see 5.2). The thiocellobiose in grey from the complex with ExoI was superimposed on the modelled structures of BxlA as well as on the 3D structure of Bgl3B; green sticks show catalytic nucleophile (Asp) and acid/base (Glu) residues.

Enzyme	Specific ac	tivity (U/mg)	Ratio ¹
BxlA	32	± 1.64	1
D307G	0.027	± 0.0032	8×10^{-4}
D307A	0.028	± 0.0023	9×10^{-4}
D307C	0.034	± 0.0027	11×10^{-4}
D307S	0.029	± 0.0061	9×10^{-4}
C308W	0.027	± 0.0014	9×10^{-4}
N313R	11	± 0.80	0.3
1 normaliz	ad with activ	ity of DylA	

Table 10 Specific activity of subsite +1 mutants of BxIA towards 2 mM *p*NPX.

¹ normalized with activity of BxlA Standard deviations were calculated from 4

experiments

The glycosynthase reactions of GH3 β -xylosidase D307A and D307C catalytic nucleophile mutants were done by first screen for pH suitable for the reaction in the pH range 5.0–7.0 (Fig. 19; data not shown for pH 5.0). *pNPAp* was used as acceptor for the screening since aryl sugars bind well to the subsite of glycoside hydrolases. After 16 h incubation, the product could be visualized at pH 7 by TLC and it became clearly visible after 40 h. The glycosynthase reactions were slow and the mutants showed quite low activity. When replacing *pNPAp* by lyxose, galactose, or cellobiose, the glycosynthase products could not be deteted neither by TLC nor HPAEC-PAD (data not shown), confirming the weak glycosynthase activity by these mutants.



Figure 19 Glycosynthase reactions using 50 μ M D307A or D307C with 2 mM α XF as donor and 2 mM *p*NPA*p* as acceptor in 100 mM sodium acetate (pH 5.5, pH 6.0) or 100 mM phosphate buffer (pH 6.5, 7.0) at 30°C for: 16 h (A) and 40 h (B). Lane 1, control reaction without enzyme; lane 2, 50 μ M D307A; lane 3, 50 μ M D307C. The TLC was developed with 100% acetonitrile (see 5.7.4 for TLC method).

C308W showed three orders of magnitude diminished hydrolytic activity as seen also for the glycosynthase mutants compared to BxlA and no transglycosylation activity was detected after incubation for 24 h using 200 nM C308W with 30 mM pNPX and 200 mM acceptor (xylose, galactose, mannose, fructose, or maltose) confirming the loss of its activity. N313R showed three folds decrase in hydrolytic activity compared to BxlA wild-type. The transglycosylation activity of N313R toward hexoses increased 1.5–2 fold and up to three folds towards maltose, but slightly decreased towards xylose (Fig. 20 and Table 11), indicating that this residue may be involved in binding to 6-OH of hexose acceptors as predicted by 3D model of BxlA using 1IEX as template (Fig. 18C).



Figure 20 Progress of transglycosylation with different acceptors by N313R. Reactions containing 200 nM N313R with 30 mM *p*NPX and 200 mM acceptor; xylose (\blacktriangle), galactose (\square), mannose (Δ), fructose (\blacklozenge), or maltose (\blacksquare).

Table 11 Maximum transglycosylation yields with different acceptors by N313R in comparison with BxlA.

	N313R]		
Accontor	Yield	Production	Yield	Production	Patio ¹
Acceptor	(%)	(mg/mL)	(%)	(mg/mL)	Katio
Monosaccharides					
D-xylose	18	1.55	21	1.81	0.8
D-galactose	14	1.28	9	0.81	1.6
D-mannose	37	3.50	25	2.34	1.5
D-fructose	16	1.54	8	0.79	1.9
Disaccharides					
Maltose	5	0.78	2	0.27	2.7
1					

¹ Normalized by transglycosylation yield of BxlA

6.2 GH5 mannanases

6.2.1 Production and purification

Two subsite +1 mutants ManAS289W and ManCW283S of ManA and ManC, respectively, from *A. nidulans* FGSC A4 were produced recombinantly as His-tag fusion proteins in *P. pastoris* X-33 (Fig. 21 and 22). The enzymes were purified by nickel chelating chromatography followed by gel filtration in a yield of 12 and 24 mg/L culture for ManAS289W and ManCW283S, respectively. The ManA and ManC wild-type enzymes were produced by Dr. Hiroyuki Nakai by induction with buffered methanol-complex medium (BMGY with 0.5% methanol without glycerol) for 5 days and purified in the same protocol as their subsite +1 mutants in a yield of 120 and 145 mg/L culture, respectively (see manuscript 2).



Figure 21 Purification of ManAS289W. A) immobilized metal ion affinity chromatogram. B) SDS-PAGE of purified ManAS289W by HisTrap 5 mL column; 25 μ L samples loaded. C) gel filtration chromatogram of pooled ManAS289W fraction from (A). D) SDS-PAGE of ManAS289W purified by Hiload 26/60 Superdex G75 column; 25 μ L samples loaded; marker, Mark12 unstained protein standard marker (Invitrogen). The red bars in (A) and (C) show the fractions which are analysed by SDS-PAGE in (B) and (D) (see 5.4.2, page 36).



Figure 22 Purification of ManCW283S. A) immobilized metal ion affinity chromatogram. B) SDS-PAGE of purified ManCW283S by HisTrap 5 mL column; 20 μ L samples loaded. C) gel filtration chromatogram of pooled ManCW283S fraction from (A). D) SDS-PAGE of ManCW283S purified by Hiload 26/60 Superdex G75 column; 20 μ L samples loaded; marker, Mark12 unstained protein standard marker (Invitrogen). The red bars in (A) and (C) show the fractions which are analysed by SDS-PAGE in (B) and (D) (see 5.4.2, page 36).

6.2.2 Protein characterisation

The recombinant ManCW283S migrated in SDS-PAGE as a single band similarly to ManC with an estimated size of 56 kDa (Fig. 23, Lane A), whereas ManAS289W migrated as a smear similarly to ManA with lowest band of 56 kDa and the smearing up to 90 kDa (Fig. 23, Lane C). The theoretical masses were 43946.2 and 46544.6 Da, for ManAS289W and ManCW283S, respectively. Deglycosylation by Endoglycosidase H reduced the molecular mass of ManCW283S by 5 kDa (Fig. 23, Lane B), and of ManAW283S by 10 kDa (Fig. 23, Lane D), while the smear bands became a sharp band at 45 kDa. *N*-glycosylation was predicted to occur and the enzymes have 4 and 2 possible sites for ManAW283S and ManCW283S, respectively, which correspond well with the reduced molecular masses after deglycosylation. This indicated slight glycosylation is in agreement with what was found for ManA and ManC (Fig. 23E–H).



Figure 23 SDS-PAGE of purified and deglycosylated ManA, ManC, ManAS289W, and ManCW283S. Lane A, 7 μ g ManCW283S; lane B, 7 μ g deglycosylated ManCW283S; lane C, 5 μ g ManAS289W; lane D, 5 μ g deglycosylated ManAS289W; lane E, 5 μ g ManC; lane F, 5 μ g deglycosylated ManC; lane G, 5 μ g ManA; lane H, 5 μ g deglycosylated ManA; lane M, Mark12 unstained protein standard marker (Invitrogen); endoglycosidase H is seen as a 29 kDa band in lane B, D, E, and H.

6.2.3 Enzymatic specificity, kinetics, and hydrolysis action patterns

The activity towards konjac glucomannan (konjac GlcMan), guar galactomannan (guar GalMan), and locust bean gum galactomannan (LBG GalMan) is shown in table 12. ManC showed 30–80% higher activity towards polymeric mannans than ManA and also showed preference to LBG GalMan with 20% and 40% higher activity towards guar GalMan and konjac GlcMan, respectively. In contrast, ManA showed no different in specific activity on LBG and guar GalMans, but showed 15% lower specific activity on konjac GlcMan than on both types of GalMans. LBG and guar GalMans have different degree of Gal substitution (Gal:Man, 1:4 and 1:2, respectively; see 2.1.5.1, page 10). Thus, ManA seems to be less discriminate regarding different substitution in GalMans as well as the backbone composition in case of konjac GlcMan, whereas ManC appears to have a preference towards less substitution mannans.

The subsite +1 mutants showed decreased activity towards polymeric mannans. The specific activity of ManAS289W towards LBG GalMan was 15% and 20% higher than that of guar GalMan and konjac GlcMan, respectively, whereas the specific activity of ManCW283S on LBG GalMan showed only 5% and 15% higher specific activity towards guar GalMan and konjac GlcMan. Thus, substitution of Trp283 in ManC reduced the impact of Gal substitution and *vice versa* in case of ManA mutation. This could be because the bulky Trp obsecure the side chain of mannan to bind to the substrate binding site.

		Specific activity
Enzyme	Substrate (0.5%)	U/mg
ManA	Konjac GlcMan	340 ± 2
	LBG GalMan	$400 \ \pm 4$
	Guar GalMan	390 ± 6
ManAS289W	Konjac GlcMan	220 ± 1
	LBG GalMan	300 ± 1
	Guar GalMan	250 ± 1
ManC	Konjac GlcMan	$430 \ \pm 4$
	LBG GalMan	730 ± 2
	Guar GalMan	600 ± 2
ManCW283S	Konjac GlcMan	270 ± 1
	LBG GalMan	320 ± 4
	Guar GalMan	300 ± 1

Table 12 Specific activity of ManA, ManAS289W, ManC, and ManCW283S towards mannan polymers

Standard deviations were calculated from triplicate experiments (see 5.6.2.1 for experimental details).

The kinetic parameters towards mannooligosaccharides (M4–M6) are shown in Table 13. Both mannanase wild-type and mutant enzymes have highest k_{cat} towards M6 and their catalytic efficiency (k_{cat}/K_m) increased with increasing length of mannooligosaccharides, which is typical behavior for endo-acting enzymes having 5–6 substrate binding subsites. All mannanases showed weak affinity towards M4, therefore only k_{cat}/K_m values were reported. ManC showed higher catalytic efficiency than ManA. Both mannanase subsite +1 mutants showed slightly lower k_{cat} than the corresponding wild-type enzymes but the effect was even smaller than for the specific activity towards mannans (Table 12). The catalytic efficiency of ManCW283S decreased approximately 50%, but K_m was slightly increased compared with ManC. The K_m value of ManAS289W was slightly decreased, whereas the catalytic efficiency was quite similar compared with ManA. These could be the effect of the Trp substitution, which affected the substrate binding affinity of both mutants.

To gain further understanding of the hydrolysis action patterns of ManA and ManC, 0.5 mM M4–M6 were hydrolysed by ManA, ManAS289W, ManC, and ManCW283S, and the time course of the hydrolysis was monitored by HPAEC-PAD (Table 14 and Fig. 24). Both ManA and ManC form mannobiose as major hydrolysis

product from M4–M6, indicating that ManA and ManC may have strong substrate binding affinity at subsite –2, +2 or both. M1 and M3 were also produced importantly from M4 in case of ManA, which may be due to the weak affinity as shown in the kinetics analysis towards M4 giving some flexibility in the productive binding positioning at the active site. From the hydrolysis action patterns of M5 and M6, subsite –3 and/or +3 of the ManA could play role also in substrate binding and hence the trisaccharide was produced. In case of the subsite +1 mutants, ManAS289W and ManCW283S also formed mannobiose as main hydrolysis product from M4–M6 and their hydrolysis action pattern showed no obvious different from their wild-type enzymes (Table 14 and Fig. 24 D–F, J–L).

		$k_{\rm cat}$	K _m	$k_{\text{cat}}/K_{\text{m}}$
Enzyme	Substrate	s^{-1}	mM	$s^{-1}mM^{-1}$
ManA	M4	_	_	6
	M5	67 ± 2.8	2.9 ± 0.03	23
	M6	193 ± 5.8	1.8 ± 0.08	109
ManAS289W	M4	_	_	10
	M5	42 ± 2.5	1.6 ± 0.16	27
	M6	148 ± 1.7	1.3 ± 0.05	115
ManC	M4	_	_	7
	M5	112 ± 8.1	1.8 ± 0.13	61
	M6	134 ± 5.2	0.6 ± 0.04	215
ManCW283S	M4	_	_	3
	M5	98 ± 3.1	2.8 ± 0.13	36
	M6	117 ± 3.3	1.0 ± 0.03	118

Table 13 Hydrolysis kinetic parameters of ManA, ManAS289W, ManC, and ManCW283S towards mannooligosaccharides (M4–M6).

Standard deviations were calculated from triplicate experiments (see 5.6.2.2 for experimental details)

Table 14Hydrolysis product formation from 0.5 mM mannooligosaccharideshydrolysed by ManA, ManAS289W, ManC or ManCW283S for 20 min.

Substrate	ManA	ManA S289W	ManC	ManC W283S
M4	64%	68%	91%	88%
	36%	32%	9%	12%
M5] 100%	100%	96%	100%
	} -	-	4%	-
	88%	88%	92%	94%
000000	12%	12%	8%	6%



Figure 24 Hydrolysis action patterns of ManA (A–C), ManAS289W (D–F), ManC (G–I), and ManCW283S (J–L) towards 0.5 mM M4–M6 in 100 mM sodium acetate pH 5.5 at 37°C. A), 60 nM ManA with M4; B), 30 nM ManA with M5; C), 10 nM ManA with M6; D), 50 nM ManAS289W with M4; E), 28 nM ManAS289W with M5; F), 12 nM ManAS289W with 0.5 mM M6; G), 75 nM ManC with M4; H), 15 nM ManC with M5; I), 12 nM ManC with M6; J), 120 nM ManCW283S with M4; K), 32 nM ManCW283S with M5; L), 20 nM ManCW283S with M6; M1, (Δ); M2, (\Box); M3, (\circ); M4, (\diamond); M5, (\blacktriangle); M6, (\blacksquare).

6.2.4 Transglycosylation catalysed by ManA and ManC

The time course of transglycosylation catalysed by ManA and ManC was studied using either 30 mM M4 without addition of acceptor or 24 mM M4 with 100 mM M3 as acceptor as monitored during 150 min at 37°C by HPAEC-PAD. When using 30 mM M4, M5–M7 were produced by ManA (Fig. 25A), whereas ManC produced only M5 and M6 (Fig. 25C). The total product yields (M5+M6) were 2% and 16% for ManA and ManC, respectively. M7 was predicted from HPAEC-PAD

profile thus not included when estimating the total transglycosylation yield because there is no M7 standard available and the low yields hampered producing enough M7 for use as standard. The concentration of M7, however, was estimated by using standard curve of M6. Addition of M3 helped increasing the transglycosylation ability of both ManA and ManC by approximately 50%. The total product yields (M5+M6) were 4% and 25% for ManA and ManC, respectively (Fig, 25E-25H). In both cases, ManC possessed higher transglycosylation ability than ManA by almost 10-fold.



Figure 25 Time course of transglycosylation products formation of ManA and ManC using 30 mM M4 or 24 mM M4 with 100 mM M3. A) transglycosylation products of 330 nM ManA with 30 mM M4; B) hydrolysis products and substrate of 330 nM ManA with 30 mM M4; C) transglycosylation products of 145 nM ManC with 30 mM M4; D) hydrolysis products and substrate of 145 nM ManC with 30 mM M4; E) transglycosylation products of 330 nM ManA with 24 mM M4 and 100 mM M3; F) hydrolysis products and substrate of 330 nM ManA with 24 mM M4 and 100 mM M3; G) transglycosylation products of 145 nM ManC with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM ManC with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM ManC with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM ManC with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM ManC with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM ManC with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM ManC with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM ManC with 24 mM M4 and 100 mM M3; M1, (Δ); M2, (\Box); M3, (\circ); M4, (\diamond); M5, (\blacktriangle); M6, (\blacksquare); M7, (\bullet); see also Fig. 39, page 88.

6.2.5 Transglycosylation catalysed by subsite +1 mutants

3D homology modelling of ManA and ManC was performed using GH5 mannanase from *H. jecorina* RUTC-30 complexed with mannobiose at subsite +1 and +2 (PDB ID: 1QNR) as template and superimposed with mannobiose (subsite -2 to -3) from *Thermobifida fusca* KW3 (PDB ID: 2MAN; Hilge et al., 1998) to investigate possible substrate binding site interactions of ManA and ManC (Fig. 26 and Appendix Fig. 3, page 119). ManA and ManC models showed quite similar substrate binding sites to *H. jecorina* mannanase. The mannobiose from structure of *T. fusca* mannanase was used for illustrating possible substrate binding at the donor binding subsites, although it belongs to subfamily 8, which is less conserved to ManA and ManC (subfamily 7). Both ManA and ManC shared 39% sequence identity with a quite conserved possible subsite +1 and +2 except for a Trp residue (Trp283) at subsite +1 of ManC, which is equivalent to Ser289 of ManA. To investigate the effect

of this Trp283 of ManC, ManCW283S and ManAS289W mutants were made to mimic Ser289 of ManA and Trp283 of ManC, respectively.



Figure 26 Cartoons representing the active site as sticks and surface in 3D structure of mannanase from *H. jecorina* RUTC-30 and homology models of ManA and ManC. A), Model structure of ManA; B), model structure of ManC; C), structure of *H. jecorina* RUTC-30 in complex with mannobiose along subsite +1 and +2 (PDB ID: 1QNR, Sabini et al., 2000); mannobiose along subsite -2 to -3 was from *T. fusca* KW3 mannanase (PDB ID: 2MAN; Hilge et al., 1998); Nuc, catalytic nucleophile; a/b, acid/base catalyst.

The transglycosylation catalysed by subsite +1 mutants were investigated by using 30 mM M4 without addition of acceptor or 24 mM M4 with 100 mM M3 in the same fashion as ManA and ManC. ManAS289W and ManCW283S showed similar transglycosylation pattern as the wild-type enzymes. Using 30 mM M4, ManAS289W produced M5–M7 (Fig. 27A), whereas ManCW283S produced only M5 and M6 (Fig. 27C). ManAS289W increased the total product yield (M5+M6) up to 55% and the total product yield (M5+M6) was 5%, using 30 mM M4, whereas ManCW283S gave 45% yield decreased and the total product yield (M5+M6) was found to be 9%. Addition of M3 helps increasing the transglycosylation ability of ManAS289W and ManCW283S by approximately 50% as observed for ManA and ManC. The total product yields (M5+M6) were 8% and 24% for ManAS289W and ManCW283S (Fig. 27E–27H), respectively. The transglycosylation yield (M5+M6) thus increased 50% in case of ManAS289W and decreased 31% in case of ManCW283S compared to

ManA and ManC, respectively. The changes in transglycosylation yield in both subsite +1 mutants showed that the Trp283 of ManC helped to increase transglycosylation yield possibly by increasing acceptor binding affinity at subsite +1 as indicated by decreasing of K_m in case of ManAS289W and increasing of K_m in case of ManCW283S from kinetics analysis.



Figure 27 Time course of transglycosylation products formation of ManAS289W and ManCW283S using 30 mM M4 or 24 mM M4 with 100 mM M3. A) transglycosylation products of 330 nM ManAS289W with 30 mM M4; B) hydrolysis products and substrate of 330 nM ManAS289W with 30 mM M4; C) transglycosylation products of 145 nM ManCW283S with 30 mM M4; D) hydrolysis products and substrate of 145 nM ManCW283S with 30 mM M4; E) transglycosylation products of 330 nM ManAS289W with 24 mM M4 and 100 mM M3; F) hydrolysis products and substrate of 330 nM ManAS289W with 24 mM M4 and 100 mM M3; G) transglycosylation products of 145 nM ManCW283S with 24 mM M4 and 100 mM M3; H) hydrolysis products and substrate of 145 nM M3; M1, (Δ); M2, (\Box); M3, (\circ); M4, (\diamond); M5, (\blacktriangle); M6, (\blacksquare); M7, (\bullet).

6.2.6 Characterisation of mannooligosaccharide products

ManA and ManC wild-type enzymes were screened for suitable acceptors using 14 mM M2 as donor and 400 mM of different mono- to trisaccharides as acceptors (performed by Dr. Hiroyuki Nakai, EPC, DTU Systems Biology; see manuscript 2). ManC possesses higher transglycosylation activity than ManA as shown by its 8-fold higher production of M5 within 30 min screening reaction, using M3 as acceptor. ManC can also use isomaltotriose and melezitose as acceptors, whereas ManA strictly accepts only M3.

Even though ManAS289W increased the transglycosylation ability of ManA by 50%, ManC wild-type produced higher concentration of transglycosylation products. Therefore, ManC was used for the production of transglycosylation products for structural analysis. Time course of the reactions using 145 nM ManC with 24 mM M4 and 200 mM isomaltotriose or melezitose is shown as total product formation in Fig. 28. The concentration of acceptors was reduced from 400 mM to

200 mM here because of the low solubility of acceptors in acetonitrile used in the HPLC purification (see 5.8.2) and high acceptor concentrations present practical challenges in purification of the transglycosylation products due to broad acceptor peaks masking the products. The transglycosylation reactions using isomaltotriose and melezitose as acceptors resulted in two transglycosylation products from each acceptor with ratio of about 5:1 based on peak area. The mixed products occurred mainly because ManC transferred mannobiosyl and mannotriosyl to the acceptors. The maximum production of manno-isomaltotrioses and manno-melezitoses were 2.51 and 0.64 mM at 120 and 180 min corresponding to 10% and 3% total product yields, respectively, compared with the M4 donor concentration (mM/mM).



Figure 28 Time course of total transglycosylation products formation from 145 nM ManC acting on 200 mM isomaltotriose (A) and 200 mM melezitose (B).

The molecular masses of M5 and M6 produced by transglycosylation of ManC using 24 mM M4 and 100 mM M3 as well as the two forms of manno-isomaltotriose and manno-melezitose were m/z 851 and 1013 determined by MALDI-TOF MS (Fig. 29), indicating formation of a pentasaccharide as predominant and a hexasaccharide as a minor product, respectively (analysed as Na⁺ adducts). The structures of major transglycosylation products from manno-isomaltotriose and manno-melezitose as well as M5 produced by transglycosylation of ManC were assigned by 2D NMR (Table 15). The mannobiosyl isomaltotriose and mannobiosyl melezitose showed β -1,4 linked between mannobiosyl donor and trisaccharide acceptor, as identified by the downfield shift of the C6, similarly to structure of M5 confirming the regioselectivity as β -1,4 glycosidic linkage formation by ManC. Noticeably, mannobiosyl melezitose also contained minor compounds which were identified as mannobiosyl β -1,6 linked to fructose residue of melezitose and two mannobiosyl linked to one melezitose acceptor, which in later case also visualised as a small peak at m/z 1175.3 in MALDI-TOF MS chromatogram (Fig. 29C).



Figure 29 Molecular masses of transglycosylation products as Na⁺ adducts (mass range m/z 300–1500). A) Manno-isomaltotriose (1): m/z 851.239; B) manno-isomaltotriose (2): m/z 1013.255; C) manno-melezitose (1): m/z 851.213 with trace of heptasaccharide at m/z 1175.291; D) manno-melezitose (2): m/z 1013.236 (see 5.9.3 for MALDI-TOF MS method).

Table 15 ¹H and ¹³C NMR data assignments for mannooligosaccharides produced by enzymatic transglycosylation of ManC.

Compound	Chemical shifts (δ, p.p.m.)					
	H-1	H-2	H-3	H-4	H-5	H-6a/H-6b
(molecular mass) ¹	C-1	C-2	C-3	C-4	C-5	C-6
β-Man <i>p</i> -(1,4)-	4.70	4.01	3.6	3.51	3.39	3.89/3.68
	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
β-Man <i>p</i> -(1,4)-	4.7	4.08	3.77	3.77	3.51	3.87/3.71
	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
β-Man <i>p</i> -(1,4)-	4.7	4.08	3.77	3.77	3.51	3.87/3.71
	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
β-Man <i>p</i> -(1,4)-	4.7	4.08	3.77	3.77	3.51	3.87/3.71
	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
β-Man <i>p</i>	5.13	3.94	3.85	n.d.	n.d.	n.d.
	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.
α-Man <i>p</i>	4.86	n.d.	n.d.	n.d.	n.d.	n.d.
(<i>m</i> / <i>z</i> 851)	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.

¹ molecular mass as Na⁺ adduct (estimated mass + *m*/*z* 23) determined by MALDI-TOF MS

Compound			Chemica	al shifts (δ, p.j	o.m.)	
	H-1	H-2	H-3	H-4	H-5	H-6a/H-6b
(molecular mass) ¹	C-1	C-2	C-3	C-4	C-5	C-6
β-Man <i>p</i> -(1,4)-	4.79	4.14	3.82	n.d.	n.d.	n.d.
	101.2	71	72.4	n.d.	n.d.	n.d.
β-Man <i>p</i> -(1,4)-	4.75	4.08	3.68	3.58	n.d.	n.d.
, , , , ,	101.3	71.5	n.d.	76.1	n.d.	n.d.
α-Glc <i>p</i> -(1,6)-	4.98	3.62	3.91	3.71	3.85	3.78/3.86
	98.8	72.1	72.7	79.7	n.d.	61.2
α-Glcp-(1.6)-	4.98	3.6	3.75	3.52	3.93	3.79/3.98
	98.8	72.3	n.d.	n.d.	71.2	66.7
α-Glc <i>p</i>	5.26	3.56	n.d.	n.d.	n.d.	n.d.
	93.2	n.d.	n.d.	n.d.	n.d.	n.d.
β-Glc <i>p</i>	4.69	3.28	3.49	n.d.	n.d.	n.d.
(<i>m</i> / <i>z</i> 851)	97.2	75	77.1	n.d.	n.d.	n.d.
β-Man <i>p</i> -(1,4)-	4.75	4.12	3.81	3.55	3.55	3.97/n.d.
	100.1	69.7	71.4	66.7	n.d.	n.d.
β-Man <i>p</i> -(1,4)-	4.72	4.05	3.65	3.56	3.81	n.d.
	100.1	n.d.	n.d.	74.9	n.d.	n.d.
α-Glc <i>p</i> -(1.3)-	5.18	3.57	3.72	3.43	3.89	3.75/3.89
	100.3	71.3	72.7	76.3	72.4	60.3
β-Fru <i>f</i> -(2.1)-	n.d.	_	4.31	4.32	4.06	3.99/4.12
$r = \chi / f$	n.d.	103.3	82.9	73.5	79.7	70.3
α-Glcp	5.42	3.53	3.64	3.41	3.92	3.75/3.91
(m/z 851)	917	74 8	72 7	69 4	72	60.3

Table 15 (continued)

¹ molecular mass as Na⁺ adduct (estimated mass + m/z 23) determined by MALDI-TOF MS

6.3 GH10 and GH11 xylanases

6.3.1 Production and purification

GH10 (XlnC) and GH11 (XlnA) xylanases from *A. nidulans* FGSC A4 were produced recombinantly as His-tag fusion proteins in *E. coli* BL21(DE3). The enzymes were first tested for expression in 100 mL scale with 10–100 μ M IPTG induction (data not shown) and purified by immobilized metal ion affinity chromatography using a HisTrap 1 mL column (Fig. 30). The highest enzyme production was obtained with 100 μ M IPTG in yields of 9 and 8 mg/100 mL culture for XlnA and XlnC, respectively. Afterward, XlnC was produced in 1 L scale and purified by nickel chelating chromatography followed by gel filtration in a yield of 95 mg/L culture (Fig. 31). Both enzymes migrated as a single band in SDS-PAGE and the molecular masses were in agreement with the theoretical masses of M_r = 23976.8 and 33501.3 Da for XlnA and XlnC, respectively (Fig. 30).



Figure 30 Preliminary productions of XlnA and XlnC; Left, SDS-PAGE of supernatant from BugBuster treated cell, the corresponding bands to XlnC (GH10) and XlnA (GH11) are marked with light yellow arrows. Center, immobilized metal ion affinity chromatograms. Right, SDS-PAGE of XlnA and XlnC, 20 μ L samples loaded. The red bars in show the fractions which are analysed by SDS-PAGE (see 5.4.3).



Figure 31 Purification of XlnC. A) immobilized metal ion affinity chromatogram. B) SDS-PAGE of purified XlnC by HisTrap 5 mL column; 15 μ L samples loaded. C) gel filtration chromatogram of pooled XlnC fractions from (A). D) SDS-PAGE of purified XlnC by Hiload 26/60 Superdex G75 column; 25 μ L samples loaded; Marker, SeeBlue pre-stained protein standard marker (Invitrogen). The red bars in (A) and (C) show the fractions which are analysed by SDS-PAGE in (B) and (D) (see 5.4.3).

6.3.2 Enzyme activity towards xylans

The activities towards birchwood glucuronoxylan, oat spelt arabinoxylan, and wheat arabinoxylan by XlnA and XlnC are shown in Table 16. XlnA and XlnC showed quite similar activity towards the tested xylans, even though on a molar basis XlnC had slightly higher activity than XlnA due to the lower molecular mass of XlnA.

		Specifi	c activity
Enzyme	Substrate	U/mg	U/mol (×10 ⁶)
XlnA	1.0% Birchwood xylan	330 ± 13	7,900
	0.5% Oat spelt arabinoxylan	$290 \hspace{0.1in} \pm \hspace{0.1in} 12$	7,000
	0.5% Wheat arabinoxylan	$350\ \pm 10$	8,400
XlnC	1.0% Birchwood xylan	$280~\pm~9$	9,500
	0.5% Oat spelt arabinoxylan	250 ± 9	8,500
	0.5% Wheat arabinoxylan	$300\ \pm 10$	10,200

Table 16 Specific activity of purified XlnA and XlnC towards xylans.

Standard deviations were calculated from 4 experiments (see 5.6.3.1 for experimental details)

6.3.3 Transglycosylation

The preliminary screening of transglycosylation by XlnA and XlnC was done using six disaccharides (xylobiose, cellobiose, maltose, isomaltose, lactose, mannobiose) and two sugar alcohols (xylitol and sorbitol) as acceptor candidates and the reaction was monitored as product formation by TLC and MALDI-TOF MS (data not shown). XlnC can use xylobiose, cellobiose, xylitol, and sorbitol as acceptors, whereas XlnA can use only xylobiose. Therefore, XlnC was used for further investigation in an attempt to make novel oligosaccharides. The progess of transglycosylation by 1 μ M XlnC with 20 mM X3 and 200 mM acceptor (xylobiose, cellobiose, xylitol, or sorbitol) is shown in Fig. 32 and the total product yields with each acceptor is given in Table 17.



Figure 32 Progress of transglycosylation with different acceptors catalysed by XlnC. Reaction mixtures contained 1 μ M XlnC, 20 mM X3 and 200 mM acceptor; X2 (\blacktriangle), cellobiose (Δ), xylitol (\bullet), and sorbitol (\circ).

	Yield	Production	No. of major
Acceptor	(%)	(mg/mL)	TG product peaks
Disaccharides			
Xylobiose	5	0.58	2^{1}
Cellobiose	3	0.33	2
Sugar alcohols			
Xylitol	22	1.80	3
Sorbitol	16	1.40	3

 Table 17 Maximum transglycosylation yields with different acceptors by XlnC.

¹ as xylotetraose and xylohexaose

The yield was estimated as percentage of product/donor concentration (mM/mM)

6.3.4 GH10 glycosynthase

The GH10 glycosynthase mutant E244A from XlnC was produced recombinantly as His-tag fusion proteins in *E. coli* BL21(DE3). E224A was purified by nickel chelating chromatography followed by gel filtration in the same fashion as for XlnC and was obtained in a yield of 28 mg/L culture (Fig. 33). E244A had

specific activity of 0.26 U/mg towards 1% birchwood xylan (see 5.6.3.1) which was two orders of magnitude lower than for XlnC wild-type.



Figure 33 Purification of E244A using immobilized metal ion affinity chromatography followed by gel filtration. A) immobilized metal ion affinity chromatogram, 15 μ L samples loaded; B) SDS-PAGE of purified E244A by HisTrap 5 mL column; C) gel filtration chromatogram of pooled E244A fractions from (A); D) SDS-PAGE of purified E244A by Hiload 26/60 Superdex G75 column; 20 μ L samples loaded; marker, SeeBlue marker (Invitrogen). The red bars in (A) and (C) show the fractions which are analysed by SDS-PAGE in (B) and (D) (see 5.4.3).

The glycosynthase reactions of 87 μ M E244A using α -xylobiosyl fluoride (α X2F) as donor and X2 or X3 as acceptor were monitored by TLC during 48 h at 30°C (Fig. 34). The reactions with donor and acceptor without addition of E244A were done in parallel with the glycosynthase reactions as controls. $\alpha X2F$ was degraded during the incubation period seen as X2 in the control reaction with X3 as acceptor, but more than 50% aX2F still remained after 48 h. No severe changes of pH of the reactions occurred during the incubation period as monitored by pH paper (MACHEREY-NAGEL GmbH & Co. KG, Düren, Germany). The glycosynthase products X4 and X5 from X2 and X3 as acceptors, respectively, were detected after 1 h incubation and increased until 5 h. However, the glycosynthase products were hydrolysed during the incubation because of the hydrolytic activity of E244A, therefore the glycosynthase products were not accumulated. In addition, E244A can carry out transglycosylation involving hydrolysis of acceptor e.g. X2, the enzyme hydrolyses X2 and transfer xylosyl residue to another X2 molecule resulting in X3, which can be visualised by the increased in X1 and X3 during the reaction. In case of X3 as acceptor, the transglycosylation product X4 was formed. The presence of xylose in control reactions could be because of the degradation of αXF , which is present as minor contaminant in $\alpha X2F$.



Figure 34 Progress of glycosynthase reactions using 87 μ M E244A with 10 mM α X2F as donor and 5 mM X2 or X3 as acceptor in 100 mM sodium acetate pH 5.0 at 30°C during 48 h, monitored by TLC. Lane 1, reaction with X2 as acceptor without E244A; lane 2, reaction with X2 as acceptor with E244A; lane 3, reaction with X3 as acceptor without E244A; lane 4, reaction with X3 as acceptor with E244A; developed by acetonitrile/water (80:20, v/v) twice and visualized by 0.2% orcinol solution (see 5.7.4); blue arrows indicate glycosynthase products (very faint); green arrows indicate transglycosylation products (very faint).

6.4 GH61 putative endo-β-glucanase

6.4.1 cDNA cloning

The putative GH61 endo- β -glucanase (*eglF*) gene from *A. nidulans* FGSC A4 was obtained by cDNA cloning, cloned into pPICZ α A vector (pPICZ α A-EglF), and transformed into *P. pastoris* (see 5.10, page 44). The pPICZ α A-EglF was fully sequenced using 5'AOX as sequencing primer (Fig. 35) confirmed the successful cloning of this gene. Sequence alignment of *eglF* with the AN3860 from *A. nidulans* FGSC A4 genome (Galagan et al., 2005) is shown in (Fig. 35) with the intron between 327 and 378 bases.

AN 3860 _____ eglf_5AOX TTTTACGACACTTGAGAAGATCAAAAAACAACTAATTATTCGAAACGATGAGATTTCCTT 60 AN 3860 _____ eglf 5AOX CAATTTTTACTGCTGTTTTATTCGCAGCATCCTCCGCATTAGCTGCTCCAGTCAACACTA 120 AN 3860 _____ eglf_5AOX CAACAGAAGATGAAACGGCACAAATTCCGGCTGAAGCTGTCATCGGTTACTCAGATTTAG 180 AN 3860 _____ eglf_5AOX AAGGGGATTTCGATGTTGCTGTTTTGCCATTTTCCAACAGCACAAATAACGGGTTATTGT 240 AN 3860 -----ATGGCCATGTCGAAGATTGCTACCCTCGC----GGGTCT-TCTCGCCTCTG 46 eglf_5AOX TTATAAATACTACTATTGCCAGCATTGCTGCTAAAGAAGAGGGGGTATCTCTCGAGAAAA 300 * ** * * **** * * ***** AN 3860 CTGGCCTCGTTGCAGGCCACGGATACGTGACGAAGATGACGATGGAGAAGAATATG 106 * GTGGATGGCTCGCTGATAGCTATTACTACATGGACTCGCCGCCGGATAACTACGGATGGA 166 AN 3860 eglf_5AOX GTGGATGGCTCGCTGATAGCTATTACTACATGGACTCGCCGGCGGATAACTACGGATGGA 420 AN 3860 GTACGACAGTGACCGACAATGGCTTCGTCTCCCCCGATGCCTTTGGTACCGACGACATCA 226 eglF_5AOX GTACGACAGTGACCGACAATGGCTTCGTCTCCCCCGATGCCTTTGGTACCGACGACATCA 480 AN 3860 CCTGCCACAGGGGCGCCACTCCTGGTGCTCTCTCTCCCCGGTGACAGCCGGCAGCAAGA 286 eglf_5AOX CCTGCCACAGGGGCGCCACTCCTGGTGCTCTCTCGCCCCGGTGACAGCCGGCAGCAAGA 540 AN 3860 eq1F_5AOX TCGATATCACTTGGAACACCTGGCCCGAATCTCATAAGGG------580 AN 3860 CTTTTGCTTGTCCTACTAACAGATAATCAAGGCCCCATCATCAACTACCTCGCCAAATGC 406 eglf 5AOX -----CCCCATCATCAACTACCTCGCCAAATGC 608 AN 3860 AACGGCGACTGCTCTTCAGCCGACAAGACCAGCCTCGAATTCGTGAAGATCCAGGCCGAA 466 eglf_5AOX AACGGCGACTGCTCTTCAGCCGACAAGACCAGCCTCGAATTCGTGAAGATCCAGGCCGAA 668 AN 3860 GCGATTGTCGACGCTTCCACCACCACCTGGGTTACCGATGAACTCATTGAGAACTCCTTC 526 eglf_5AOX GCGATTGTCGACGCTTCCACCAACACCTGGGTTACCGATGAACTCATTGAGAACTCCTTC 728 AN 3860 ACTACTTCCGTTACGATCCCTGCCTCCATCGCCCCAGGTAACTACGTCCTCCGACACGAG 586 eq1F 5AOX ACTACTTCCGTTACGATCCCTGCCTCCATCGCCCCAGGTAACTACGTCCTCCGACACGAG 788 AN 3860 ATCATCGCCCTCCACTCTGCAGGCCAGCAGCAGCAGGTGCCCAAGCATACCCTCAGTGCTTG 646 eglf_5AOX ATCATCGCCCTCCACTCTGCAGGCCAGCAGCAGCAGGTGCCCAAGCATACCCTCAGTGCTTG 848 AACCTGGTTGTCAGCGGCAGCGGTACTGACAACCCATCTGGTACTCCTGGTACTCAGCTG 706 AN 3860 eglf_5AOX AACCTGGTTGTCAGCGGCAGCGGTACTGACAACCCATCTGGTACTCCTGGTACTCAGCTG 908 TACTCTGCTAACGATGAGGGCATCGTCTTCGATATCTACTCGAACCCGACCTCGTACCCT 766 AN 3860 eglF_5AOX TACTCTGCTAACGATGAGGGCATCGTCTTCGATATCTACTCGAACCCGACCTCGTACCCT 968 AN 3860 ATGCCTGGTCCTGAGCTGTACAGTGGTTAG----- 796 eglf_5AOX ATGCCTGGTCCTGAGCTGTACAGTGGTCATCATCATCATCATCATCATCTAGAACAAAA 1028 AN 3860 eglF_5AOX ACTCATCTCAGAGA 1042

Figure 35 Sequence alignment of GH61 putative endo- β -glucanase (*eglF*) gene (using 5'AOX as sequencing primer) and the corresponding gene from genomic sequence (AN3860); the nucleic acids translated to mature protein is in red and the intron is in blue.

6.4.2 Production and purification

The *eglF* gene was first cloned into pET28-a(+) using *E. coli* BL21(DE3) as expression host, however *E. coli* was not able to produced this protein. Therefore, *eglF* was cloned into pPICZ α A vector to give the putative GH61 endo- β -glucanase and expressed using *P. pastoris* X-33 as host. The putative GH61 endo- β -glucanase (EglF) was produced recombinantly as a His-tag fusion protein and purified by anion exchange chromatography (Fig. 36A,B) followed by gel filtration chromatography (Fig. 36C,D) in a yield of 66 mg/L culture.



Figure 36 Purification of GH61 EglF. A) anion exchance chromatogram; B) SDS-PAGE of purified EglF by Resource Q 6 mL column, 15 μ L samples loaded; C) gel filtration chromatogram of pooled EglF fraction from (A); D) SDS-PAGE of purified EglF by Hiload 26/60 Superdex G200 column, 15 μ L samples loaded; marker, SeeBlue marker (Invitrogen). The red bars in (A) and (C) show the fractions which are analysed by SDS-PAGE in (B) and (D) (see 5.4.1).

6.4.3 Protein characterisation and screening for activity

EglF migrated as a single band in SDS-PAGE with estimated size of 30 kDa (Fig. 37A). The theoretical mass (M_r) of EglF was 25004.3 Da and gel filtration gave a value of 26 kDa (Fig. 37B), indicating that EglF is a monomer in solution. EglF contains two possible *N*-glycosylation sites, however, endoglycosidase H treatment showed no different between untreated and treated protein forms. The isoelectric point of EglF was 4.0 (Fig. 37C). EglF showed very low specific activity towards barley β -glucan and oat spelt arabinoxylan with values of 180±9.0 and 145±2.8

 μ U/mg (see 5.10.3), when incubated for 16 h, indicating that this protein was either not a typical β -glucanase or it was produced as an inactive enzyme.



Figure 37 Molecular mass of EglF determined by SDS-PAGE (without and with endoglycosidase H treatment) and gel filtration, and pI of EglF. A) SDS-PAGE of purified and deglycosylated EglF; Left lane, marker, Mark12 unstained protein standard marker (Invitrogen); lane A, 7 μ g EglF; lane B, 7 μ g deglycosylated EglF, endoglycosidase H is seen as a band at 29 kDa; B) calibration curve of superdex G75 16/60 column (open circle) and EglF (black square), EglF has molecular weight of 26 kDa indicated by a black square; C) isoelectric focusing of 800 μ g purified EglF using polyacrylamide PhastGel IEF (pH 4–6.5) with low range pI (pH 2.5–6.5) calibration standard (GE Healthcare).

6.4.4 Binding to carbohydrate microarrays

The putative GH61 endo- β -glucanase EglF showed no binding towards the tested carbohydrate microarrays at pH 6 and 7 (see 5.10.4), indicating that the protein may not function as a binding protein, has very low affinity towards the tested carbohydrates, or has the specificity towards other carbohydrate than the tested ones.

7. Discussion

7.1 GH3 xylosidases

7.1.1 Production of BxlA

The GH3 β -xylosidase from *A. nidulans* FGSC A4 (BxlA) was secreted in recombinant form from *P. pastoris* as directed by the α -factor secretion signal sequence from *S. cerevisiae*. The production of BxlA was much more efficient than the production by *A. nidulans* CECT2544 resulting in 0.5 mg/L (Kumar and Ramón, 1988) and was in the same range as found for the previously reported recombinant BxlA (as XlnD) in *A. nidulans* G191 host (Pérez-González et al., 1998). BxlA was glycosylated in *P. pastoris* (see Fig. 13, page 51) and the carbohydrate content corresponded well with its apparent molecular mass. However, the glycosylation was not present in native BxlA secreted from *A. nidulans* (Kumar and Ramón, 1988) hence this high glycosylation is a result of production by *P. pastoris*.

7.1.2 Comparision of hydrolytic activity of BxIA and BxIB

In comparison with BxlB sharing 45% sequence identity with BxlA, Both BxlA and BxlB have quite similar pH optima of 5.0 and 5.2, respectively, but different temperature optima (50°C for BxlA and 60°C for BxlB). Additionally, BxlA is more sensitive to pH and temperature than BxlB (see Fig. 14, page 52). Most fungal GH3 β-xylosidases, as well as BxlA and BxlB, have pH and temperature optima in the pH range 4–6 and 50–70°C (Herrmann et al., 1997; Knob et al., 2010; Kumar and Ramón, 1996; Kurakake et al., 1997). Additionally, BxlB has a wider pH stability range (pH 4–8) compared with most GH3 β-xylosidases having pH 4–6 (Herrmann et al., 1997; Knob et al., 2010; Kumar and Ramón, 1996; Kurakake et al., 1997). Also the temperature stability up till 50-60°C of BxlB resembled that of other fungal GH3 β-xylosidases (Herrmann et al., 1997; Knob et al., 2010; Kumar and Ramón, 1996; Kurakake et al., 1997; Rasmussen et al., 2006), whereas BxIA was sensitive to the temperatures >45°C. The incubation time of this experiment was 5 h, which was used to investigate the stability of the enzymes during transglycosylation reactions. In addition, Ag^{2+} and Cu^{2+} have strong effect on BxIA (see Table 6, page 51) as well as the native A. nidulans GH3 B-xylosidases (XlnD; Kumar and Ramón, 1996). XlnD was found to be inhibited also by Zn^{2+} , whereas no inhibition was observed in BxlA. The metal inhibition may be due to the metal ion complex particularly with sulfhydryl groups (Whitaker, 1994). BxIA and BxIB have the capability to hydrolyse also α -Larabinofuranosidase (pNPA; see Table 7, page 53) with two and one orders of magnitude lower catalytic efficiency, respectively, compared to activity for pNPX. The low specificity towards pNPA has also been observed in some GH3 βxvlosidases (Bauer et al., 2006; Herrmann et al., 1997; Lee et al., 2003). Structurally, β-D-xylopyranose, preferably forming a pyranose ring with chair conformation projecting the hydroxyl groups in equatorial positions away from the ring, is related to the α -L-arabinofuranose, which forms a pucker conformation that resembles equatorial 2-OH and 3-OH of β -D-xylopyranose, and the hydroxyl configuration at the anomeric carbon of α -L-arabinofuranose also resembles that of β -D-xylopyranose (Fig. 38). These conformations are energetically more favourable than other configurations (Carpita and McCann, 2000) and may be the reason for β -xylosidase to have the side activity of α -L-arabinofuranose. The K_m of 1.3 mM towards *p*NPX of BxlA and BxlB produced in this study is quite similar to that of the native BxlA from *A. nidulans* (Kumar and Ramón, 1996), but the k_{cat} of BxlA is three fold higher. The kinetic parameters of GH3 β -xylosidases towards *p*NPX from different sources are shown in Table 18. BxlA has quite high hydrolytic activity towards *p*NPX, whereas BxlB gives values quite similar to other β -xylosidases. In addition, BxlA showed higher k_{cat} than BxlB towards all tested xylooligosaccharides (see Table 7, page 53). The kinetic parameters towards xylooligosaccharides (X2–X6) showed that both BxlA and BxlB prefer to hydrolyse xylobiose, while the catalytic efficiency decreased slightly towards longer xylooligosaccharides, as is the typical behaviour of exo-type enzymes and and suggesting that BxlA and BxlB possibly possess with two substrate binding subsites (Biely, 2003; Herrmann et al., 1997; Matsuo and Yasui, 1984; Rasmussen et al., 2006).



 β -D-Xylopyranose

 α -L-arabinofuranose

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	k _{cat}	K _m	$k_{\rm cat}/K_{\rm m}$	
Organism	s^{-1}	mМ	$s^{-1} mM^{-1}$	Reference
A. nidulans (BxlA)	112	1.3	90	This study
A. nidulans (BxlB)	18	1.3	14	This study
A. nidulans (XlnD ^a)	37 ^b	1.1	34	Kumar and Ramón, 1996
A. awamori	11 ^c	4.1	2.7	Eneyskaya et al., 2003
A. awamori	17.5	0.25	70	Eneyskaya et al., 2007
T. reesei	176	0.42	419	Hermann et al., 1997
Hordeum vulgare	15.5	1.7	9.1	Lee et al., 2003

Table 18 Kinetic parameters of GH3 β -xylosidases towards *p*NPX.

^a BxlA was referred to as XlnD

^b calculated from 25.6 µmol/min/mg protein; MW: 85000 Da

^c calculated from 5.3 nmol/min/µg protein; MW: 125000 Da

7.1.3 Transglycosylation and acceptor specificity of BxlA

BxlA displays broad acceptor specificity in transglycosylation and most of the transglycosylation reactions result in more than one product observed in HPAEC-PAD chromatograms (see Fig. 16 and Table 8, page 54–55). The minor products, possibly with different linkages or products synthesised from different forms of one acceptor (α , β , pyranosyl, furanosyl forms), complicates identification of their

structures. Thus, the heterogeneity of such samples hampers the assignment of NMR spectra and precludes assignment for some of the transglycosylation products. β -Xylosidases of GH3 from *A. niger* (Puchart and Biely, 2007), *A. awamori* (Eneyskaya et al., 2003), and *Penicillium wortmanni* (Win et al., 1988) were reported to possess transglycosylation activity, demonstrated by the synthesis of mainly β -1,4 linked xylooligosaccharides, whereas β -1,6 were formed predominantly when using hexose-containing acceptors (Kurakake et al., 1997; 2005). This is similar to the case of BxlA having mainly β -1,4 regioselectivity when using pentose acceptors (xylose, arabinose, lyxose) and mainly β -1,6 xylosidic-linkage when using hexose-contining acceptors (glucose, mannose; see Table 9, page 58). In case of disaccharide acceptors, β -1,4 and β -1,6 xylosyl linkage was found reflecting that BxlA has loose acceptor recognition.

BxlA displayed broad acceptor specificity in transglycosylation and had the highest transglycosylation yield using xylitol as acceptor and, among 10 tested monosaccharides, can transglycosylate mannose, lyxose, and talose in similar maximum total yields as xylose (>20% yield), while it was less efficiency in transglycosylate arabinose, glucose, galactose, and fructose and could not use Lrhamnose as acceptor (see Table 8, page 55). In general, BxIA as well as exo β glucanase from *H. vulgare* (ExoI) and β -glucosidase from *T. neapolitana* (Bgl3B) seem to have loose acceptor recognition at subsite +1, which could be the reason for the broad acceptor specificity of BxlA. The loose acceptor recognition could allow the acceptor molecule to have some freedom to bind with different conformation resulting in different linkage formation as shown by the transglycosylation products from hexose and disaccharide acceptors (see Table 9, page 58). This may be affected by the accessibility or less steric hindrance configuration of the acceptors. The subsite +1 of BxlA is not quite clear on its 3D-models due to the not well conserved domain 2, however Tyr446 could be the key interaction with suitable acceptor molecules, which is equivalent to Trp434 in exo β -glucanase from *H. vulgare* and Ser370 in β glucosidase from T. neapolitana (see Fig 18, page 60). According to the conformation of Tyr446 in 3D models, the hydroxyl group of this tyrosine could provide a hydrogen bond to 2-OH or 3-OH of a sugar acceptor at subsite +1 or it could form stacking interaction to the sugar ring by its aromatic residue. The thiocellobiose in the active site of the model structures was created by superimposition with ExoI to help illustrating structural details of subsite -1 and +1. In addition, disaccharides are generally poor acceptors for BxlA (2-7%) as compared to their monosaccharide counterparts (21-25%). The low transglycosylation yield using maltose in case of BxlA (2%) is in agreement with 0.8% yield of the GH3 β -xylosidase from A. awamori (Kurakake et al., 2005) which tested four disaccharides: trehalose, palatinose, isomaltose, and maltose. This observation also supports the lack of recognition at subsites beyond +1 and is in agreement with the kinetics on hydrolysis of xylooligosaccharides. Noticebly, with the extended choices of mono- and disaccharide as acceptor candidates in transglycosylation screening, BxIA were able to produce 14 xylosyl oligosaccharides including five, which were not previously reported using lyxose, L-fucose, talose, sucrose, and turanose as acceptors.

In preliminary transglycosylation screening for BxlA, increasing acceptor/donor ratio (from 200/30 to 1000/40 mM; data not shown) resulted in a higher tranglycosylation potency of BxlA. However, high acceptor concentrations present practical challenges in purification of the transglycosylation products due to broad acceptor peaks masking the products and decrease solubility of acceptors in acetonitrile used in the HPLC purification.

7.1.4 Transglycosylation using amino acids as acceptors

Interestingly, this is the first report revealing the possibility to use amino acids as acceptors for transglycosylation catalysed by a wild-type glycoside hydrolase, which could be an alternative way to produce glycopeptides or thio-sugars. With Xyl-Cys (see Fig 17, page 56), we demonstrated that cysteine can be used as acceptor and forms a thio-glycosidic linkage. Such formation of thio-glycosidic linkage by protein engineered glycoside hydrolase was previously demonstrated by production of thiosugar or thio-linked saccharides catalysed by thioglycosynthase of B-glucosidase from Agrobacterium sp. using α -glucosyl fluoride as donor and pNP 4-deoxy-4-thio- β -Dglucopyranodise as acceptor (Jahn and Withers, 2003; Müllegger et al., 2005). In case of L-serine as acceptor, very low amount of Xyl-Ser was observed, but this cannot be confirmed by MS/MS. This could be because L-serine is not a suitable acceptor for BxlA or the reaction conditions were not suitable. The use of low pH (pH 5.0) may not be suitable for Ser as well as Cys to be deprotonated (Whitaker, 1994). To investigate this reaction further, a suitable detection method is required to improve monitoring of the progress of the reaction. Tyrosine has hydroxyl group in the side chain, which makes it eligible as acceptor for transglycosylation by BxlA, but it was not tested because of its low solubility in aquous solution.

7.1.5 Glycosynthase and subsite mutants of BxIA

In case of glycosynthase and subsite +1 mutants of BxIA, the mutation has some effect on the recombinant protein production (see 6.1.6, page 59), and the yields were 50% lower as compared with BxlA wild-type enzyme. These mutant enzymes could be purified in soluble form which indicated that the overall structures are intact, although one cannot exclude major changes at the active site. The glycosynthase reactions are usually conducted at pH 6-7 (Failes et al., 2003; Kim et al., 2006; Sugimura et al., 2006) to adapt to the stability of fluoride substrates. Even though BxlA glycosynthase mutants showed some activity (see Fig. 19, page 61), reactions by these glycosynthase mutants were considered as unsuccessful since the reaction consumed very high amount of enzymes and the enzymes cannot accept natural sugars (lyxose, galactose, and cellobiose; see 6.1.6). Surprisingly, the highest activity was observed at pH 7.0, even though the BxlA wild-type enzyme was not stable at pH higher that 6. This may indicate that the enzyme still has intact structure at pH 7.0 and catalytic acid/base residue (Glu) can act as general base to perform glycosynthase reaction through the deglycosylation step (see Fig. 4 and 5, page 15). In case of subsite +1 mutants, the substitution of Cys308 to Trp caused severe reduction in hydrolytic activity similarly to nucleophile mutants and the enzyme could almost not hydrolyse *p*NPX as observed in transglycosylation reactions, indicating that Cys308 is essential for BxlA, and may form a disulfide bond as explained earlier in 6.1.6 (see also Fig. 18, page 60). In case of Asn313, N313R showed 50-150% increase in transglycosylation activity using hexoses and maltose, but a slight decrease using xylose as acceptors, which could show the effect of Arg residue on accepting the hexose ring by forming a hydrogen bond to 6-OH at subsite +1 as guided by the 3D model of BxlA using exo β -glucanase from *H. vulgare* as template (see Fig. 18C, page 60).

7.2 GH5 mannanases

7.2.1 Production of ManAS289W and ManCW283S

Two GH5 mannanases (ManA and ManC), sharing 38% sequence identity, from A. nidulans FGSC A4 and subsite +1 mutants (ManAS289W and ManCW283S) were secreted in recombinant forms as His-tag fusion proteins in P. pastoris X-33 as directed by the α -factor secretion signal sequence. The production of subsite +1 mannanase mutants (ManAS289W and ManCW283S) was lower than that of ManA and ManC wild-type enzymes. This is because the difference in induction media (complex medium (BMMY) for wild-type and minimal medium (BMM) for mutant enzymes) and duration of induction (5 days for wild-type and 3 days for mutant enzymes). The use of BMM instead of BMMY is to overcome a technical problem during the production of the enzymes. Noticeably, in both cases, the production of the recombinant enzymes was more efficient than from the production of native enzymes from A. niger (Ademark et al., 1998) or H. jecorina (Arisan-Atac, et al., 1993; Stålbrand, 1993) cultures or from the recombinant production in S. cerevisiae (Evodia Setati et al., 2001). Both ManAS289W and ManCW283S were slightly glycosylated similarly to their corresponding wild-type enzymes (see Fig. 23, page 64). The glycosylation has also been observed in native *H. jecorina* mannanase (Arisan-Atac, et al., 1993).

7.2.2 Enzymatic properties of GH5 mannanases

ManC showed higher activity towards mannans than ManA (see Table 12, page 65) and had slightly higher activity than *H. jecorina* and Aspergilli mannanases (Ademark et al., 1998; Evodia Setati et al., 2001; Hägglund et al., 2003; Kurakake and Komaki, 2001), which have specific activity in the range of 230-630 U/mg towards LBG GalMan. The kinetic parameters towards mannooligosaccharides (M4-M6) of ManA and ManC are compared to H. jecorina and M. edulis GH5 mannanases (Table 19). The k_{cat}/K_m values of ManA and ManC were similar to those of M. edulis GH5 mannanase (Larsson et al., 2006), but lower than that from H. jecorina (Anderson, 2006) and showed increasing in k_{cat}/K_m with increasing chain length of mannooligosaccharides, which is typical behavior of endo-acting enzymes and suggesting that ManA and ManC possibly possess with 5-6 substrate binding subsites. Both ManA and ManC wild-type enzymes have mannobiose as main hydrolysis products from M4-M6 (Table 14, page 66), which are also in agreement with the result from transglycosylation reactions. The possible formation of transglycosylation products is illustrated in Fig. 39, which was according to all transglycosylation products (Fig. 25, page 68). M5-M7 are the possible transglycosylation products using M4 as donor and acceptor, which were present in the reaction with 30 mM M4, while M6 was produced as the major product. M4–M6 are the possible products using M4 as donor and M3 as acceptor, which were present in the reaction using 24 mM M4 with 100 mM M3, while M5 was produced as the major product. Even though M4 as transglycosylation product cannot be monitored because of the use of M4 as donor (and M3 as acceptor), both cases confirmed that the mannobiosyl accumulated at the donor binding sites ("minus subsites") during the first step of the hydrolysis and interact with an incoming M4 or M3 acceptor. This is in agreement with the hydrolysis action pattern analysis, which showed that M2 is the main product from M4 and M6 hydrolysis. This clearly showed that the subsite -2 dominates binding at the glycone part of the active site cleft for both ManA and ManC. ManA produced more M1 and M3 from M4 than ManC in hydrolysis action pattern analysis, which may reflect either a higher affinity at subsite(s) -3 and/or +3 or weaker substrate affinity at subsite(s) -2 and/or +2 as compared to ManC. The hydrolysis action pattern on M6, however, does not support a difference in affinity of subsite(s) -3 and/or +3 for ManA and ManC, since they produce essentially the same amount of M3. The more likely explanation is that there is a difference in the -2 and/or +2 subsite(s) between these two enzymes.

		k _{cat}	Km	$k_{\rm cat}/K_{\rm m}$	
Organism	Substrate	s^{-1}	mМ	$s^{-1} mM^{-1}$	Reference
	M4	_	_	6	
A. nidulans (ManA)	M5	67	2.9	23	This study
	M6	193	1.8	109	
	M4	_	_	7	
A. nidulans (ManC)	M5	112	1.8	61	This study
	M6	134	0.6	215	
	M4	6.3	0.31	21	
H. jecorina (Man5A)	M5	13	0.08	163	Anderson, 2006
	M6	20	0.04	500	
	M4	0.70	2.49	0.28	
M. edulis (Man5A)	M5	45	1.61	28	Larsson et al., 2006
	M6	99	0.5	198	

Table 19 Kinetic parameters of GH5 β -mannanases towards mannooligosaccharides (M4-M6).



Figure 39 Illustration of transglycosylation products formation using A) M4 as donor and acceptor and B) M4 as donor and M3 as acceptor.

7.2.3 Effect of subsites of GH5 mannanases

According to 3D homology models of ManA and ManC (see Fig. 26, page 69), Asn319 of mannanase from H. jecorina (PDB ID: 1QNR) could form a hydrogen bond to 6-OH of the mannose residue at subsite -3, whereas the equivalent amino acid in ManA, Asp362 was oriented to bind with 3-OH of the mannose residue at subsite -2 (see Fig. 26A). In case of ManC, His368 equivalent to Asn319 of H. jecorina mannanase and Asp362 of ManA are exposed on the protein surface and may obscure substrate binding at subsite -3 (see Fig. 26B). As a consequence both ManA and ManC produced predominantly mannobiose as hydrolysis product as well as mannobiosyl bound to acceptor in transglycosylation reactions. The possible subsites +1 and +2 of ManA and ManC are well conserved, compared to H. jecorina mannanase, with one amino acid difference (Ser289 of ManA and Trp283 of ManC). Substitution of these residues (as ManAS289W and ManCW283S) showed that Trp283 of ManC had some impact on mannanases, but was not the critical residue controlling hydrolytic and transglycosylation ability of the enzymes. The changes in transglycosylation activity and substrate binding affinity (as $K_{\rm m}$) of ManAS289W and ManCW283S could be due to the effect of Trp238 of ManC. However, ManAS289W showed slightly lower $K_{\rm m}$ compared with ManA, whereas ManCW283S showed slightly higher K_m compared with ManC (see Table 13, page 66) indicating that this Trp has impact on the substrate binding ability and the changes in transglycosylation activity could be affected of the Trp substitution. More interestingly, the Trp283 of ManC seemed to involve in modulating the enzyme activity towards branched polymeric mannans possibly by defining a more restricted or open active site cleft at the proximal subsite -1 and +1. It is tempting to speculate that the more open active site cleft in ManA helps accommodating more branched substrates possibly aided by the space provided by the Trp to Ser substitution results in a looser affinity at the proximal subsites in ManA as compared to ManC (Fig. 26, page 69). Additionally, the homology models of ManA and ManC suggested that Tyr259 and Trp253, respectively, could act as subsite +3 and have a possible role in accepting the trisaccharides. These aromatic residues are strictly conserved among GH5 mannanases from fungi. Furthermore, GH5 mannanase from H. jecorina can accommodate a glucose residue at the acceptor recognition site ("plus subsites") which will advance the hydrolysis of glucomannan (Harjunpää et al., 1999), and which could be one reason for ManC to accept glucose residue from isomaltotriose and melezitose. In addition, these acceptors could have suitable conformation to bind at the plus subsites of ManC.

7.3 GH10 and GH11 xylanases

7.3.1 Production and activity of XlnC and XlnA

GH10 (XlnC) and GH11 (XlnA) xylanases were first reported together with another xylanase (XlnB) from *A. nidulans* distinguished by their molecular masses: 22 kDa for XlnA, 24 kDa for XlnB, and 34 kDa for XlnC (Fernández-Espinar et al., 1992). XlnC was further produced and purified from *A. nidulans* culture (Fernández-Espinar et al., 1994) and recloned into an *A. nidulans* expression host (MacCabe et al., 1996; MacCabe and Ramón, 2001) as well as into *S. cerevisiae* (Ganga et al., 1998).
Together with XlnB, XlnA was cloned and produced in *S. cerevisiae* (Ganga et al., 1996; Pérez-González et al., 1996). In this study, we recloned *xlnA* and *xlnC* genes from the *P. pastoris* transformants harbouring these xylanase genes from *A. nidulans* FGSC A4 (Bauer et al., 2006) into *E. coli*, aiming to investigate the possibility of XlnA and XlnC to produce novel xylooligosaccharides *via* transglycosylation reactions. The recombinant production of XlnA and XlnC by *E. coli* (see 6.3.1, page 74) was more efficient than from the native host (5 mg/L for XlnC; Fernández-Espinar et al., 1994) or the recombinant production in *S. cerevisiae* (1.1 mg/L for XlnA and 1.0 mg/L for XlnC; Ganga et al., 1996). The activity towards birchwood xylan of XlnC (280 U/mg, Table 16, page 75) was 25% lower than that of native XlnC (379 U/mg; Fernández-Espinar et al., 1994), whereas the activity of XlnA (330 U/mg, Table 16, page 75) was in comparable with the reported GH11 xylanases (691 U/mg from *A. niger* BRFM281, Levasseur et al., 2005; 638 U/mg from *Streptomyces matensis* DW67, Yan et al., 2009; 420 and 101 U/mg from *T. reesei*; Tenkanen et al., 1992).

7.3.2 Transglycosylation by XlnC and XlnA

Xylanases usually possess transglycosylation as a side activity when high concentration of oligosaccharide substrates is used (see 4.2.4, page 24). Using six disaccharides (see 6.3.3, page 76) as acceptors, XlnA strictly accepted only xylobiose, whereas XlnC can use cellobiose, xylitol, and sorbitol as acceptors (see 6.3.3). Both XlnA and XlnC cannot use maltose, isomaltose, lactose, and mannobiose as acceptors. XlnC and GH10 xylanases have a shallow groove active site (see 4.2.2, page 22) with loose acceptor binding site at subsites +1 and +2 (Fig. 40A,B). The cellobiose can be situated in the acceptor binding site ("plus subsites") and His93 and Arg145 can possibily form hydrogen bonds with 6-OH of the glucose residue at subsite +1. XlnA and GH11 xylanases have a deep and narrow cleft for active site, which restricts xylose molecule to be accommodated in the active site and the extra 6-OH of a hexose residue cannot fit in (Fig. 40C,D). The sugar alcohols gave higher yield than disaccharide acceptors in case of XlnC, which could be due to their more flexible conformations allowing multi-hydroxyl groups to attack the glycosyl-enzyme intermediate or they could behave as alcohol, which requires no interaction with the acceptor binding site. However, GH11 cannot use xylitol and sorbitol as acceptors indicating that the sugar alcohols need somehow to interact with acceptor binding site and the linear molecule makes it difficult for a sugar alcohol to enter the acceptor binding site. The transglycosylation products from XlnC were not purified further since these products are not novel. Xylobiosyl-cellobiose had been synthesised by reverse phosphololysis using recombinant cellodextrin phosphorylase from *Clostridium thermocellum* YM4 with α -D-xylose 1-phosphate as donor and cellobiose as acceptor (Shintate et al., 2003). Xylotritol was synthesised together with xylitol and xylobitol in the production of reduced xylooligosaccharides as sweeteners (Shimizu et al., 1987). Xylobiosyl-sorbitol (xylobiosyl-D-glucitol) was synthesised as by-product from a chemical reaction for production of glycosyl-alditols (Sharkov et al., 1961) and as by-product from the synthesis of N-glycosyl derivatives of sodium sulfanilate (Kahl and Kuszlik-Jochym, 1964). The SciFinder search engine was used for evaluation of novelty of transglycosylation products (https://scifinder.cas.org; American Chemical Society, Columbus, Ohio).

7.3.3 Glycosynthase of XlnC

The GH10 glycosynthase mutant (E224A) was produced in three fold lower amounts than the XlnC wild-type enzyme and showed three orders of magnitude lower activity than wild-type (see 6.3.4, page 76). The glycosynthase activity of E244A was found to be rather low as judged by TLC (see Fig. 34, page 78) and, when using high concentration of the enzyme, remaining hydrolytic activity degrades the glycosynthase products (Fig. 34), therefore this glycosynthase is considered unsuccessful. The glycosynthase of GH10 xylanase has been reported by substituting the nucleophilic glutamic acid with Gly, Ala, and Ser (Kim et al., 2006; Sugimura et al., 2006). The glycine mutants exhibited highest glycosynthase activity and the alanine mutants showed good glycosynthase activity, whereas the serine mutants showed poor activity (Sugimura et al., 2006). The low activity of E244A could be due to the inappropriate substitution or the nature of this enzyme as in the study using glycosynthase mutants from different GH10 xylanases exhibited different activity (Sugimura et al., 2006).



Figure 40 Cartoons representing active sites in the 3D structure of GH10 (XlnC) and a homology model of GH11 (XlnA) xylanases. A) 3D structure of XlnC (PDB ID: 1TA3; Payan et al., 2004) superimposed with xylobiose (subsite -1 and -2, pale yellow•) and cellobiose (subsite +1 and +2, lime green•) from *P. simplicissimum* GH10 xylanase (PDB ID: 1B3Z; Schmidt et al., 1999), cellobiose was created by superimposition two β -glucoses on xylobiose at the acceptor binding sites; B) structure of XlnC similar to (A) with surface representation; C) model structure of XlnA using *P. funiculosum* xylanase as template (PDB: 1TE1; 67% sequence identity; Payan et al., 2004) and superimposed with xylobiose (subsites -1 and -2, blue•) and cellobiose (subsites +1 and +2, yellow•) from *Neocallimastix patriciarum* xylanase (PDB ID: 2VGD; Vardakou, et al., 2008), cellobiose was created by superimposition two β -glucoses on xylobiose on the acceptor binding sites; D) side view of the model structure of XlnA similar to (C).

7.4 GH61 putative endo-β-glucanase

The objective for this part of the study was to perform enzyme discovery by applying a proteomics strategy. The proteomics strategy involved analysis of culture supernatants of A. nidulans FGSC A4 growing on 10 different polysaccharides (see 4.4.3, page 29) and was conducted by Dr. Kenji Maeda, Dr. Hiroyuki Nakai, and PhD student Anders Dysted Jørgensen, EPC, DTU Systems Biology. The GH61 putative endo-β-glucanase from A. nidulans FGSC A4 (EglF) was successfully cloned into P. pastoris by the cDNA cloning technique (see 5.10.2, page 45) and secreted in recombinant form as directed by the α -factor secretion signal sequence. The production of EglF (66 mg/L) was lower than the previously reported either from culture broth of H. jecorina (120 mg/L; Karlsson et al., 2001) or recombinant form of H. jecorina secreted by P. pastoris (280 mg/L; Liu et al., 2006) as Cel61A containing carbohydrate binding module family 1 (CBM1). Several GH61 proteins contain CBM1 at the C-terminal but EglF has no CBM (Fig 41). EglF showed extremely low activity towards barley β -glucan and oat spelt arabinoxylan and no binding ability was observed using carbohydrate microarrays. It is possible that the tested substrates or the reaction conditions have not been suitable for ElgF showing very low activity towards barley β -glucan and oat arabinoxylan, or ElgF has other functions. However, such extremely low activity was also observed in the other reports on GH61 proteins either from A. kawachii (Koseki et al., 2008) or H. jecorina (Harris et al., 2010; Karlsson et al., 2001; Saloheimo et al., 1997).

Two structures of GH61 proteins from H. jecorina (PDB ID: 2VTC) and Thielavia terrestris NRRL 8126 (PDB ID: 3EII) (see 4.4.2, page 29) showed no evidence of large surface clefts, crevices, or holes that would be possible binding pocket for soluble polysaccharides (Harris et al., 2010). There are two highly conserved carboxylic residues (Glu178 and Asp223 of EglF; Fig. 41 and 42) which may play a catalytic role. Glu178 is a buried residue and forms a salt bridge with a highly conserved Arg176, this has been proved to be important by mutagenesis of Glu137 of T. terrestris (3EII) resulting in lack of protein production (Harris et al., 2010). Even though Asp223 is surface accessible, it cannot act as catalytic residue, according to classical mode of action of glycoside hydrolases (Fig. 4, page 15), since it has no other carboxylic side chain close by (Karkehabadi et al., 2008). It is possible that GH61 may not follow a classical acid/base catalytic mechanism, but their activity is very low towards polysaccharides for being glycoside hydrolases. Several reports suggested that GH61 proteins may not act as glycoside hydrolases but have other functions (Brown et al., 2005; Harris et al., 2010; Karkehabadi et al., 2008; Liu et al., 2006). For example, a few reports suggested possible role of GH61 proteins as cellulase-enhancing factors which significantly improve the degradation of lignocellulosic biomass by cellulases (Brown et al., 2005; Harris et al., 2010; Liu et al., 2006). However several GH61 proteins from the same origin (different isoforms) are completely inactive as enhancers and some can act synergistically with cellulases from different sources making it ambiguous what is the true function of GH61 proteins (Harris et al., 2010). Therefore, the quest for understanding GH61 proteins still remains.



Figure 41 Alignment of putative GH61 endo- β -glucanase (EglF) with reported structures from *Thielavia terrestris* NRRL 8126 (Cel61E_THITE; PDB ID: 3EII; Harris et al., 2010) and *H. jecorina* QM6A (Cel61B_TRIRE; PDB ID: 2VTC; Karkehabadi et al., 2008) and two Cel61A containing CBM1 from *H. jecorina* RUTC-30 (Cel61A_TRIRE; GenBank ID: CAA71999.1) and *A. kawachii* (Cel61A_ASPKA; GenBank ID: BAB62318.1). Light blue arrows: two conserved Glu and Asp; green box: CBM1.



Figure 42 Cartoons representing 3D structures and homology model of GH61 proteins. Pink, 3D model of EglF using structure from *H. jecorina* QM6A (PDB ID: 2VTC; Karkehabadi et al., 2008) as template; green, 3D structure of GH61 from *H. jecorina* QM6A (PDB ID: 2VTC; sharing 44% sequence identity); purple, 3D structure of *Thielavia terrestris* NRRL 8126 (PDB ID: 3EII; sharing 33% sequence identity; Harris et al., 2010). A) superimposition of three cartoons with grey surface from EglF model; B) metal ion-binding site; dark green circle, metal ion (nickel for 2VTC and zinc for 3EII); C) conserved acidic residues (Glu and Asp) with grey surface from EglF model.

7.5 Novel oligosaccharides

Several novel oligosaccharides were produced by the present study *i.e.* six novel xylosyl- and four novel mannooligosaccharides, using β -xylosidase (BxlA) and β -mannanase (ManC), respectively (Table 20). As mentioned in 7.1, several transglycosylation products were produced in more than one form (Table 8, page 55) possibly with different linkages or products synthesised from different forms of an acceptor (α , β , pyranosyl, furanosyl forms), all complicate identification of the structures of the obtained products.

Both xylo- and mannooligosaccharides act as prebiotics and are used as food additives in different food products due to these properties (see 2.1.4.2 and 2.1.5.2, page 9,11). Xylooligosaccharides stimulate the growth of several intestinal Bifidobacteria (Grootaert et al., 2007; Moure et al., 2006; Vázquez et al., 2000; Zeng et a., 2007) and mannooligosaccharides can stimulate both Lactobacilli and Bifidobacteria (Asano et al., 2003), and both oligosaccharide types cannot be used by pathogenic bacteria *e.g. Staphylococcus, Clostridium* spp. and *Escherichia coli*

(Asano et al., 2003; Vázquez et al., 2000). Novel oligosaccharides produced via transglycosylation may exhibit the health benefits from both donor and acceptor and have better prebiotic effect than oligosaccharides present in nature by enhancing persistence of the prebiotic effect along the colon, antipathogen effects and more specific targeted prebiotics.

Enzyme	Donor	Acceptor ¹	Yield	Production	Mass ²	NMR ³
			(%)	(mg/mL)	(m/z)	
200 nM BxlA	30 mM <i>p</i> NPX	D-lyxose	23	1.92	289	\checkmark
		L-fucose	14	1.27	303	×
		D-talose	22	2.04	319	×
		Sucrose ⁴	6	0.81	481	\checkmark
		Turanose	7	0.95	481	\checkmark
		Cysteine	n.d.	n.d.	276, 298 ⁵	×
145 nM ManC	24 mM M4	Isomaltotriose	8.8	1.75	851	\checkmark
			1.4	0.33	1013	×
		Malazitosa	3.0	0.62	851	\checkmark
		wiciczitose	0.5	0.13	1013	×

Table 20 Novel oligosaccharide production by transglycosylation

¹ acceptor concentration = 200 mM.

² xylosyloligosaccharides were determined by ESI-MS as Li⁺ adduct,

mannooligosaccharides were determined by MALDI-TOF MS as Na⁺ adduct.

³ see Table 9 and 15 for NMR assignment of xylosyl- and mannooligosaccharides, respectively.

⁴ Sucrose forms two transglycosylation products, *i.e.* β -xylopyranosyl- $(1\rightarrow 4)$ - α -glucopyranosyl-

 $(1\rightarrow 2)$ - β -fructofuranoside and β -xylopyranosyl- $(1\rightarrow 6)$ - β -fructofuranosyl- $(2\rightarrow 1)$ - α -glucopyranoside. ⁵ determined by ESI-Q-TOF as Na⁺ adduct and [Xyl-Cys-H+2Na]⁺.

n.d., not determined; \checkmark , the structures have been assigned by NMR; \star , the structures cannot be assigned by NMR.

8. Conclusion and future perspectives

The present PhD project has demonstrated production of novel oligosaccharides by enzymes catalysed transglycosylation. Five glycoside hydrolases: one GH3 β -xylosidase (BxIA), two GH5 mannanases (ManA and ManC), one GH10 xylanase (XlnC), and one GH11 xylanase (XlnA); from *Aspergillus nidulans* FGSC A4 were chosen for investigation of their transglycosylation ability. The use of different acceptors to investigate the acceptor specificity in transglycosylation of these enzymes also provides a route to produce novel oligosaccharides that might be useful as novel prebiotics or food ingredients. Six novel xylosyl- and four novel mannooligosaccharides were achieved by this study, using BxIA and ManC, respectively. Three other xylobiosyl-oligosaccharides achieved by transglycosylation catalysed by XlnC have been reported earlier by others.

The exo-acting enzyme BxlA showed broad acceptor specificity, which is useful as tool for synthesis of novel xylosyl oligosaccharides. para-Nitrophenyl xyloside was used as a donor containing a good leaving group which helps improving the transglycosylation yield. Optionally, using xylobiose, a natural substrate of β xylosidase will decrease the transglycosylation yield, but will be more suitable for large-scale production of transglycosylation products for their usage as food ingredients. The endo-acting enzymes ManC and XlnC, in contrast, showed narrow acceptor specificity resulting in formation of a few different transglycosylation products in low yield (3-10% yield). Using high acceptor concentration can improve the transglycosylation yield, but is problematic for purification by HPLC to separate the products from their acceptors. In large-scale production, this problem could be solved by addition of purification step e.g. gel filtration to remove some of the acceptor prior further purification by HPLC. The cost of acceptors and donors for transglycosylation is also necessary to consider prior to large scale production. The substitution at subsite +1 in case of β -xylosidase and mannanase mutants showed impact on transglycosylation ability of these enzymes, but was not crucial. Even though it is possible to use the reported GH3 β -glucosidase and β -glucanase as templates for homology models of GH3 β -xylosidase, there is still some conflict between the models with different templates and missing information on the active sites, therefore the 3D structure of this enzyme is needed to gain more knowledge on the structural functional relationship of GH3 β -xylosidase. Even though, glycosynthase is one of the powerful tools for synthesis of oligosaccharides, the glycosynthase reactions using GH3 β-xylosidase and GH10 xylanase mutants in this study were not successful (see 6.1.6 and 6.3.4, respectively). No other attemps of using GH3 as glycosynthase have been reported.

In the last part of this PhD study, the GH61 putative endo- β -glucanase from *A*. *nidulans* FGSC A4 (EglF) was chosen to apply enzyme discovery using a proteomics strategy (see 4.4.3). EglF was successfully cloned and produced recombinantly, but showed very low hydrolytic activity towards barley β -glucan and oat arabinoxylan as well as no detectable activity and binding ability by carbohydrate microarray analysis. The actual function of this putative enzyme has not been revealed, but recent publications proposed the possible role of GH61 protein as cellulase-enhancing factor (Brown et al., 2005; Harris et al., 2010; Liu et al., 2006). The mechanism behind this function remains unknown. The present work has explored obvious enzymes to be used in chemoenzymatic oligosaccharide synthesis and provides a basis for further developments either through protein engineering to increase yields or make new compounds, *e.g.* including glycosylated amino acids. It, moreover, adds to the understanding of the enigmatic GH61, and further work with the present prepared protein may contribute to shed light on the less known facts of cellulose degradation.

9. References

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10. Appendix

10.1 Supplementary Figures



Supplementary Fig. 1 ESI-Q-TOF mass spectra (m/z 50-320) of Xyl-Cys transglycosylation reaction mixture containing 200 nM BxlA, 30 mM pNPX, and 200 mM L-cysteine in 10 mM sodium acetate pH 5.0 at 40°C. B) reaction with L-cysteine at 0 min. C) reaction with L-cysteine at 60 min reaction time (from Figure 17).



Supplementary Fig. 1 (cont.) ESI-Q-TOF mass spectra of Xyl-Cys transglycosylation reaction mixture. D) ESI-Q-TOF MS/MS spectrum of Xyl-Cys at m/z 276. E) ESI-Q-TOF MS/MS spectrum of Xyl-Cys at m/z 298 (from Figure 17).

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BAE65591 ASPOR		P.LSELPICDTSLSPL	.SRAKSLVSALTLEEKINNTGHE
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AAK38482 HORVU		A.PSSAAFCDRRLPIE	.QRAADLVSKLTLEEKISQLGDE
CAW52613 ZEAMA ABT29899 THENE Boll3B	PISTNGKNYTKVCDPARF	/ALGLDMSRFRYCDASLPYA	. DRVRDLVGRLALEEKVRNLGDQ
CAA33665_CLOTH			DIKKIIKQMTLEEKAGLCSGLD
BAA10968 ASPAC		P.WANGQGEWAEAY	. QRAVAIVSQMTLDEKVNLTTGTGWEL
AAD23382_HORVU_ExoI		D.YVLYKDATKPV	EDRVADLLGRMTLAEKIGQMTQIERLV
AAA64351 BACSU YbbD		SASKR	AIDANQIVNRMSLDEKLGQMLMPDFRNWQK
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Supplementary Fig. 2 Amino acid sequence alignment of BxlA (GenBank ID: EAA64470) and BxlB (GenBank ID: EAA67026) from *A. nidulans* FGSC A4 with related of functionally characterized GH3 enzymes (see detail in page 119)

EAA64470_ASPNI_BxlA		as 20.2020202020 180 190 200 210
EAA64470_ASPNI_BxlA ABA40420_ASPFU_Xld EED57542_ASPFL_XylA EAA67023_ASPNI_BxlB BAE65591_ASPOR EED45224_ASPFL CAA93248_HYPJE_Bxl1 CAB9357_ARATH_AtBX3 AAK38482_HORVU CAM52613_ZEAMA AB129899_THENE_Bgl3B CAA33665_CLOTH BAA10968_ASPAC ABC88234_PSBLB_ExOP AAD23382_HORVU_ExOI AA64351_BACSU_YbbD AAF93857_VIBCH_NagZ AAQ05801_CELFI ABU76375_ENTSA	. GGVD. VYGDNIN YGLD. VYGDNIN YGLD. VYGDNIN AGID. YWGDNVN AGID. YWGDNVN AGLD. YWGDNN YGLD. VYGNVN YGLD. VYGNVN AGLT. FWSDNVN AGLT. FWSDNVN AGLT. FWSDNIN . AGLT. FWSDNIN . GVDVLLA AAM . GVDVLLA AAM . GIDVQLGAAAG . GIEWSFAATVA . GIQYAFABCIA . GIQYAFABCIA . GINVAFGVVD . DYDLSFAVLD . GINWAFGVVD . DGLNMWADMVD	T. FRHPVWG.RGQETPGEDAF.LTSVVGYLYITALQGV. D. PETLK A. FRSAMMG.RGQETPGEDAYCLASIYAYEYITALQGV. D. PETLK A. FRSAMMG.RGQETPGEDAYCLASIYAYEYITGIQGGV. D. PEHLK F. FRHPVWG.RGQETPGEDAYCLASIYAYEYITGIQGGV. D. ANPLK P. FRDPRWG.RGQETPGEDPLHCSRVVKEFVGGLQGD. D. PEKPK P. FRDPRWG.RGQETPGEDPLHLSRVVKHLVDGLQDGI. G. PERPK G. FRSPLWG.RGQETPGEDPLHSRVVYHLVDGLQDGI. G. PERPK V. FRDPRWG.RGQETPGEDPLHSRVVYHLVDGLQGGV. D. PEHLK V. FRDPRWG.RGQETPGEDPT.LSSXVAVAVVKGLQETD. GGD.PNRLK V. FRDPRWG.RGQETPGEDPT.LSSXVAVAVVKGLQETD. GGD.PNRLK V. FRDPRWG.RGQETPGEDPT.VGRVAAVFVKGVQSQ. AIN.SDLE V. VRDPRWG.RASETFGEDPT.VGRVAAVFVKGVQSQ. G I. KRSPLCG.RNFEYPPEDPV.LSSELAASHIKGVQSQ. G PL.GRSPDGG.RNWEGFSPDPA.LTGVLFAETIKGUDA. GV V. VRDDRWG.RTYESYSEDPA.LTGVLFAETIKGUDA. GV V. VRDDRWG.RTYESYSEDPA.LTGVLFAETIKGUDA. GV V. VRDDRWG.RTYESYSEDPA.VGRVAGEMVTGIQGDVGADFL.KGSN V. CRDPRWG.RGVRSFSNELTSRLGLYMVTGLQGDVPKDFTSGMPFVAGKNK U. NNNPDNFVIGVSSEDPQ.VILKHSSAFLRGMKA. VG DILNFRNPI. TNTRTFGADAA TVAAMGAEYVBAIQAQ. G V. TRDPRWG.RGSEFFEDVYLSSKLGLYMAKGLQR. VG
		α6 α7 β1
EAA64470_ASPNI_BxlA	220 23	240 250 260 270 240 250 260 270
EAA64470 ASPNI ExlA ADA40420 ASPFU X1d EED57542 ASPFL Xy1A EAA67023 ASPNI ExlB BAE65591 ASPOR EED45224 ASPFL CAA93248 HYPJE Bxl1 CAB9357 ARATH ALEX3 AAK38482 HORVU CAM52613_ZEAMA ABI29899 THENE Bg13B CAA33665 CLOTH BAA10968 ASPAC ABC88234 PSEBE ExOP AAA23382 HORVU ExoI AAA64351 BACSU YDbD AAF93857 VIECH MagZ AAQ05801 CELFI ABU76375_ENTSA	I IATAK HYAGYDI LVATAK HYAGYDI LIATAK HYAGYDI VVATCK HFAAYDL VVATCK HFAAYDL VVATCK HFAAYDL VAATVK HFAGYDL VAAC K HFTAYDI ASACCK HFTAYDI ASACCK HFTAYDU VSSCCK HFTAYDU VGACLK HFTANNQ VYATAK HFTANNQ VVATAK HFTANNQ VVATAK HFTANNQ VVATAK HFTANN VGACLK HFAANQ VAACAK HFDGGG IASALK HFDGGG VAACAK HFDGGG VAACAK HFDGGG VAACAK HFDGGG VATSVK HFDGGV VMTSVK HFAAYGA	SWNNHSR. LGNDM.OITQCELSEYYTPPFIVASRDAKV.RSVM.CS SMNDGHSR. LGNDM.NITQCELSEYYTPOFIVARADAKV.RSVM.CS SNWDNHSR. LGNDM.NITQCELSEYYTPOFIVARADAKV.HSVM.CS SNWDNHSR. LGNDM.QITQCDLAEYYTPOFIVARADAKV.HSVM.CS SNWDSHSR. LGNDM.VSPQDLSEYYDPFKTCAVDASV.GARM.CS SNWEGIER. YAFDA.VVSPQDLSEYYDPFKTCAVDASV.GARM.CS SNWNNSR. LGNDA.VVSPQDLSEYYDPFKTCATDAKV.DAVC.CS SNWNNNSR. LGFDA.IITQQDLSEYYTPOFILAARVAKS.SS.C.CA SNWNKGVR. LFFDA.VVSPQDLSEYYDPFKSCVVDGGA.SGIM.CS SNWKGVTR. FFDA.VVSPQDLADTFOPFKSCVVDGGA.SGIM.CS SNWKQVR. LFFDA.VVSPQDLADTFOPFKSCVVDGGA.SGIM.CS SNWKGVTR. FFDA.VVSPQLADTYNQFLAAVKSKRS.WS.SA ST. SAFDA.KVTEQDLADTYNQFELAVKSKKP.WSVM.CS SNWKGVTR. GFTTLREIYTSSFENAVKKSKP.WSVM.CS ST. SAFDA.KVTEQDLADTYNPFFKSCVVDGGA.SGIM.CS ST. SAFDA.KVTEQDLADTYNGFELAVKSKFP.WSVM.CS ST. SAFDA.KUTEGULADTYNGFELAVKSKFP.WSVM.CS ST. SAFDA.KUTEGULADTYNGFELAVKSKFP.WSVM.CS ST. SAFDA.KUTEGULADTYNFFFERFFEMCIRDGCA.SCVM.CS ST. SAFDA.KUTEGULADTYNFFYSFENAUKKSKFP.WSVM.CS <t< th=""></t<>
EAA64470_ASPNI_BxlA		$\begin{array}{cccccccccccccccccccccccccccccccccccc$
EAA64470 ASPNI EXLA ABA40420 ASPNI EXLA EED57542 ASPFL XyLA EED57542 ASPFI XyLA EAA65023 ASPNI EXLB EAA53244 ASPFL CAA93248 HYPJE EXL1 CAB9357 ARATH AtBX3 AAX38482 HORVU CAW52613 ZEAMA ABI29899 THENE Bg13B CAA33665 CLOTH BAA10968 ASPAC ABC88234 PSEBB EXOP AD23822 HORVU EXOI AAA64351 BACSU YbbD AAF93857 VIECH NagZ AAQ05801 CELFI ABU76375 ENTSA	YMAV YMAV YMAV YMAL YMSL YMSV YMSV YMV YMV YMV YMRL YMRL YMRL YMRL YMRL YMRL YMRL YMRL YMRL YMRL YMRL YMR YMRL	N
EAA64470_ASPNI_BxlA	<u>α9</u> <u>00000000</u> 330	α10 α11 <u>202000000000000000000000000000000000</u>
EAA64470_ASPNI_BxlA ABA40420_ASPFU_Xld EED57542_ASPFL_XylA EAA67023_ASPNI_BxlB BA865591_ASPOR EED45224_ASPFL CAA93246_HYPJE_Bxl1 CAB9357_ARATH_AtBX3 AAX38482_HORVU CAW52613_ZEAMA ABI29899_THENE_Bgl3B CAA33665_CLOTH_ BAA10966_ASPAC ABC88234_PSEBB_EXOP AD23382_HORVU ExoI	NEAAASADSILAG NITGAADSIRAG NESSAAADSIRAG GGPEAAAAALNAG DGAHAAAAALNAG DGAHAAAAALNAG DGAHAAAAALNAG NQSSAAASSIRAG TPEEAVAKSILAG APEDAVADVIKAG TGVEATAAAMKAG RVSGIDAG RVSGIDAG CDLEQCAQAINAG	TD TDCGT

Supplementary Fig. 2 (cont.)

EAA64470_ASPNI_BxlA			<u>α12</u> <u>000000000000</u>	$\beta_{\alpha} = -\frac{\beta_{3}}{2}$
EAA64470 ASPNI ExlA ABA40420 ASPFU_Xld EED57542 ASPFI_XylA EAA67023 ASPNI ExlB BA655591 ASPOR EED45224 ASPFL CAA93248 HYPJE Bxl1 CAB9357 AARTH AtBX3 AX38482 HORVU CAW52613 ZEAMA AB129899 THENE Bgl3B CAA33665 CLOTH BAA10966 ASPAC ABC68234 FSEBE ExoP AAD23382 HORVU ExoI AA64351 EACSU FXDD AAF93857 VIBCH NagZ AAQ05801 CELFI ABU76375_ENTSA	Jov YFDGE.D.A YFDGK.T.S YFDDA.G.G YFDPAE.G.Q YFDPAD.D.Q YFDPAD.D.Q YFDPAD.D.Q YFDPAN.D.Q UFDGNPKY.N FFDGDPKK.Q LFDGNPKY.N FFDGMPE.S SFKNYRYS.N FFNKKKEN.A RLYQPPNFSSWTRDEYGFKYFYQ.E SFKNYRYS.N LFENPYAD. LFENPYAD.S LHRGT.NL LFNDPYSHLG	SYRDIT.W.DDV VYRDLT.W.NDV PYRNIT.W.SDV PLRSLG.W.DDV PYRSIG.W.DV QYRSLG.W.KDV LYGGLG.P.KDV RYGNIG.A.DQV .FESLG.A.SNV KP	LISTDAWN IAYEAAV VTTDAWN ISYEAAV VTTDAWN ISYEAAV ATSEAEELAKUVAI FTPAAEELAKKATV VKTDAWN ISYEAAV VKTDAWN ISYEAAV CTADNQELARDGAR CSKEHQDLALQAR CSKEHQDLALQAR CDHKELAADAAR EQDAHHRLARQAAA VQRNHSEVIRKLGA GAAEHRLAREAVR GGXQHLKAEKKLAE GSKQHLKAEKKLAE SCHLKAEKKLAE	EGIVLLKND EGIVLLKND QSIVLLKND GSIVLKND EGIVMLKND EGIVMLKND EGIVLKND GIVLLKND GIVLLKND GIVLLKND GSIVLLKND EGVVLLKNE ESVVLKNE SSIVLLKNE SSIVLKNE ESVLLKNE ESIVLKNE ESITLVKEE ESITLVKEE ESITLVKEE
	β4			
EAA64470_ASPNI_BxlA	420 430 440			450
EAA64470 ASPNI BxlA ABA40420 ASPFU X1d EED57542 ASPFU X1d EAA67023 ASPNI BxlB BA655591 ASPOR EED45224 ASPFL CAA93248 HYPJE Bxl1 CAB9357 ARATH AtBX3 AK33482 HORVU CAW52613 ZEAMA ABI29899 THENE Bgl3B CAA33665 CLOTH BAA10968 ASPAC ABC88234 PSEB EXOP AAD2382 HORVU EXOI AAA64351 BACSU YDDD AAAF93857 VIBCH MagZ	ETLPLSKD IKSVAVIGPW ANVTEEL GTLPLAKS VRSVALIGPW ANVTTEU GILPLTSSSTKTIALIGPW MNTTTQU WTLPLKANGTLALIGPFNFTTEL GTLPLKSNGTVALIGPFNATTQU GTLPLSKK.VRSIALIGPFNATTQL GTLPLSKK.VRSIALIGPFNATTQU GLPLSKK.VRSIALIGPMANATTQM BALPLSKK.VSLAVIGPNGNNASLL RRLPLDPNK.INSVSLVGLLEHINATDVM EALPLSENSKIALFGTGQIETIK. NALPLKSSGTIALIGEDASSNSYG SILPIKSSGTIALIGEDASSNSYG SILPIKSSKIALFGTGQIETIK. NALPLKSSKIALIGED SILPIKSSKIALIGED GSILPIKSSKIALIGED SILPIKSSKIALGED SILPIKSSKIALGED SILPIKSSKIALGED SILPIKSSKILVAGG ADNLGYQ HTLPFKPKK.GSRILIVAPYEEQTASI	Q	GNYF GNYF SNYA GNYE GNYE GNYE GNYH GNYH GDYR GDYR GDYR SGS SH TLAMAWGSGT GGWSVSWQ GGWTIEWQ QTIHDLIK	GP.AP GP.AP GP.AP GP.AP GP.AK GP.PK GP.PK GP.PK GP.PK GP.PK GP.PK GP.PK GP.PK GP.PK GP.PK GP.PC GK.PC TH.PR AE.FP G.TDNTNSDFP G.DTGR.TT RKKIKPVSLSKMNFA
AAQ05801_CELFI ABU76375_ENTSA	GVLPITRERYPRVLVYDLQNGGSP. ETLPLKKSGTIAVVGPLADSKRDM		IGQGARA	GAVE
EAA64470_ASPNI_Bx1A	<u>460</u>		470	
EAA64470_ASPNI_BxlA ABA40420_ASPFU_Xld EED57542_ASPFL_XylA EAA67023_ASPNI_BxlB BA65551_ASPOR EED45224_ASPFL CA8935246_HYPJE_Bxl1 CA893577_RARTH_AEBX3 AAK38482_HORVU CAW52613_EAMA AB129899_THENE_Bgl3B CA33665_CLOTH BAA10966_ASPAC ABC88234_PSEBB_ExOP AADC3382_HORVU_EXOI AAA64351_BACSU_YbDD AAF93857_VIECH_NagZ AAQ05801_CELFI ABU76375_ENTSA	YLISPLTGFRDSGLDVHY. YLISPLQAFQDSGFDVNY. YIISPLQAFQDSEYKITY. HIPTMIEAAERLG. YNVLT YIRTLIWAAVHNGYKVKF YIRTLIWAAVHNGYKVKF YLISPLEAAKKAGYHVNF. KYTTPLQGLAETVSSTY. ISVTPLQALQGYVKDARF RIVTPYNAIRMMVNATY YAISILEGIKERGLNFDEELAKTYE RLDDIYEEIKKAG.ADKVNLVY. YLVTPEQAIQAEVLK. NATSIFSGLQSQVTKAGGKITL. VGTTILEAVKAAVD.PSTVVVF. SQVFKTEHEKQVKE.ADVII. QFVDALV. QSVTLLQGMKNVA.GHKAKILY.	DYIKKMRETEEY	ALGTNLTSH AFGTNISSH TIGTNTTD APGTEVNST SQGTDINSN ELGTEIAGN QLGCNVA.C VQGCNAA.V VQGCNAA.V KPRDSWGTIIKPK SEGYRLEND HKGSVYAI SES.GEY. AENPDAEFV AKGANVTDD	L G. KGIVDFLNLYEPAVV
FAA64470 ASONT Byla	α13	β5	Ĩ	0.0.0.0
EARCHY // ASPNI_EXIA EARCHY // ASPNI_EXIA ABA40420_ASPNI_EXIA EED57542_ASPFI_XylA EARC57023_ASPNI_BxlB BAC65591_ASPOR EED45224_ASPFI_CAA93246_HYPJE_Bxl1 CAA93246_HYPJE_Bxl1 CAA93248_HYPJE_Bxl1 CAA9357_ARATH_ALBX3 AAX38482_HORVU CAA9365_CLOTH_ BAA10966_ASPAC ABC88234_PSEBB_EXOP AAD2382_HORVU_EXOI AAA54351_BACSU_YbbD AAF93857_VIECH_NagZ AA008801_CELFI	480 480 490 STS.GF. EEALTAAKQADAI STD.GF. SEALSAAKKSDVI PDST.SQ. STALTTAKEADLI STD.GF. DDALAIAAEADAL SSA.GF. AEAISAAKEADTV SSA.GF. AEAISAAKEADTV ST.GF. AKAIAAKKEDAI VD.A.DI. GSAVDLAASADAV CNVS.NI. GEAVHAAGSADYV NTAE.GM. GRASGTAKIADAT PENFLSE. KEIHKLAKKDVA IDEE.LI. NEAKKAASSSDVA TDNW.AL. SQVETLAKQASVS TDNW.AL. SQVETLAKQASVS KK. SGGFSYA	TPAGGID TPAGGID TFAGGID TFFGGID TFFGGID TYAGGID TYAGGID TYAGGID TYLOGID VLVVGAD VLVVGAD VLVVGAD VLVVGAD VLVVGAD TVIAGLP LVFVNSDAGEGY TVVIGEE TVAVGEH VNDGVID	NTIEAEAMDRE NTLETEAQDRS NTVEEESLDRT NTIEKESQDRT NTIEKESQDRT QSIEREGHDRV QSUEREGHDRV RISGEGYDRKPVK DEYESEGFDRT ISVDGNEGDRN PYAEWF.GDIE DYISDS.S.KW	520 NITW. PGNQLQL NITW PGNQLQL NITW PSNQLSL RIDW. PGNQLDL TIVW. PGNQLDL DIAW. PGNQLDL DLYL. PGKQQEL ELGL. PGMQESL DLLL. PWNQSSW GDFYL. SDDETDL HMSI. PENQNRL NLTL. WKNGDNL LLEFQHETKHALAL NLTI. PEPGLST ATVF. PR. AV
ABU76375_ENTSA	VDKRTPKA.MIDEAVNVAKQSDVV	WAVVGEA	QGMAHEASSRT	.DLTIPQSQRDL

Supplementary Fig. 2 (cont.)

EAA64470 ASPNI BxlA	α14 0000000	- <u>β6</u> -	β	7 ►-	0000
	530	54 Q	550 560	57	o <u>580</u>
EAA64470 ASPNI EXLA BEAA0420 ASPFU X1d EED57542 ASPFL Xy1A EAA657023 ASPNI EXLB BAE65591 ASPOR EED45224 ASPFL CAA93246 HYPJE Bx11 CAB93577 ARATH AtBX3 AAX38482 HORVU CAW52613 ZEAMA AB129899 THENE Bg13B CAA33665 CLOTH BAA10966 ASPAC ABC88234 PSEBE EXOP AAD64351 BACSU YADD AAF93857 VIECH NagZ AAQ05801 CELPI ABU76375 ENTSA	ISKLSEL. IDQLSQL. ITKLALL. IEQLSDL. IEQLSDL. IEQLSDL. IKQLSEV. VTRVAMAS INAVAMAS IKTVSREFHEQ. IEAVAEV. IKAAANN. LKQLKA.D VQAVCG.G MKAALQ.H YLANL ISALKAT.	GKPLVVLQMGGGQVD GKPLIVLQMGGQQVD GKPLIVVQFGGQQVD BKPLIVVQFGGQQVD GKPLVVLQMGGGQVD GKPLVVLQMGGGQVD GKPVULVIMSGGGFD FXPVULVIMSGGVD GKVIVLNIGGPVE QSNIVVVLNGSPVE QSNIVVVLNGSPVE NIPVVTVFLSGRPLW VR.CATVLISGRPVU GKPLVLVLMNGRPLA	SSSLKSNENNANS SSLKSNKNUNSL SSLKSNKNUNSL DSSLLANAGUGAL DSSLLANAGUGAL SSSLKSNKKVNSL ITFAKNPEKITSI VTFAKNPEKIGAI VSFAHNTKIGAI VSFAHNTKIGAI VSFAHNTKIGAI VSFAHNTKIGAI NPWI.DKVKSV VDEWYDHPNVTAI VNKELNASDAF VQPLLAASDAF AANFEEAKALI STRENQTVVRI LVKEDQQADAI	IWGGYPGQS IWGGYPGQS IWGGYPSQA UWAGYPSQA UWAGYPSQA UWAGYPGQA UWAGYPGEE LLVWQAGYPGEE LLVWQAGQE LLAVGGQE UMAGLPGQE VAAWLPGSE VAAWLPGSE VAAWLPGSE UAAWLPGSE UAAWLPGSE UAAWLPGSE UAAWLPGSE UAAWLPGSE UAAWLPGSE	GGHALADIITGK. GGQALLDIITGK. GGQALADIITGK. GGQALADIITGK. GGQAVFDILTGK. GGAAVFDILTGK. GGLAIADVIFGR. GGTAIADVIFGR. GGTAIADVLFGK. TGRIVADVLTGR. LGGR.WRMCYSV. SGNSLADVLTGR. GGEVADVLLTNKG. GGQVTDALFGD. NGRYLQPNIPAGVMAIFGQ.
EAA64470_ASPNI_BxlA	5 9	0 600	6	10	620
EAA64470 ASPNI BxlA ABA40420 ASPFU XId EED57542 ASPFU XylA EAA67023 ASPNI BxlB BA655591 ASPOR EED45224 ASPFI CAA93248 HYPJE Bxl1 CAB9357 ARATH ALBX3 AX38482 HORVU CAW52613 ZEAMA ABI29899 THENE Bgl3B CAA33665 CLOTH BAA10968 ASPAC ABC88234 PSEBB ExOP AAD23382 HORVU EXOI AAA64351 BACSU YhbD AAF93857 VIBCH NagZ AAQ05801 CELFI ABU76375 ENT3A	RAPAGRLV RAPAGRLV SAPAGRLP SAPAGRLP RAPAGRLP RAPAGRLP RAPAGRLV RAPAGRLV RAPAGRLV NPSGRLP VNPGGRLP VNPGGRLP KSIVGKLA VNPGAKSP GKTQFDFTGKLS FGFTGRLP AKPKGTLP	TTQYP. AEYATQFP VTQYP. AEYATQFP TTQYP. AEYATQFP TTQYP. AEYVDEVP VTQYP. ASYVDEVP VTQYP. ASYVDEVP TTQYP. AEYVHQFP MTWYP. QSYVEKVP TTQYP. AEYVHQFP TTQF. AEYVHQFP TT.FP. REFT.AVP LTWFK. NEYVNQIP TT.FP. VKLS.HNP FTWGKTREAYGDYL. FSWP. K.YDDQFT RTWF. K.SVDQFT RTWF. K.SVDQLP VDIP. S.V M	AI.DMNL AT.DMSL AI.DMNL MT.DMNL MT.DMTL QN.DMNL QN.DMNL QN.DMNL QN.DMNL QN.DMNL SW.TFP.G SYLNFP.G SYLNFP.G SYLNFP.G SYLNFP.G SYLNFP.G VR.ELN.NGNG UN.LND MN.VGD QA.EA VY.FVSFEN. VY.YSHLNTGRPY	. PPNET. SGN RPHG. NN RPNG. SN RPGS. NN RPDG. KSN RPDG. KSN RPDG. KSN RPDAA. LGY EPK. DN EPK. DN APQDPSEG 	PGQ TYMWY T PGQ TYMWY T PGQ TYMWY T PGQ TYMWY T PGR TYRWY T PGR TYRFY T PGR TYRFY T PQKVVYEEDIYVGYRYYDTFC VFI AI VFI DYRGFDKRY AI
EAA64470_ASPNI_BxlA					
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EAA64470 ASPNI BxlA ABA40420 ASPFU Xld EED57542 ASPFL XylA EAA67023 ASPNI BxlB BAE65591 ASPOR EED45224 ASPFL CAA93244 HYPJE Bxl1 CAB9357 ARATH AtEX3 AAX8482 HORVU CAW52613 ZEAMA ABI29899 THENE Bg13B CAA33665 CLOTH BAA10968 ASPAC ABC88234 PSEBB ExOP AD23382 HORVU ExOI AAA64351 BACSU YDbD AAF93857 VIBCH NagZ AAQ05801 CELFI ABU76375 ENTSA	630 GTP. VYEFGHGL GTP. VYEFGHGL 	FYTTFE ESTETTD. FYTTFE ESTETTD. FYTTFH.ASLPGTG. FYTTFN.VSWAKKA. HYTTFN.VSWAKKA. HYTTFN.VSWNHAE. HYTTFN.VSWNHAE. FYTTFK.ETLASHP. TYTKFD.HQLIKAP. SYTKFS.YASGTTG. SYTKFS.YASGTTG. SYTKFE.YSDISVD. SYTTFR.YSGLHIQ. TYQDNINVPVLSEK. TT. KQQF.S. .YHLFD.VPRVRT. SYTTFS.VSDVKLS.	AGSFNI KD. KTSFNI. TKN. RTSFNI. FGFYDA. YGFYNT. YGFYNT. .K. SLKFNT. R. LVSLS. .TK. PPSNSGI A. TVTIHIGA. F. .VL. NASSNAQVA. TSPKKTVNS.	WEHC. TETGAAPTF	GQVGNASDYVYPEGLTRISK
EAA64470_ASPNI_BxlA ABA40420_ASPFU_Xld EED57542_ASPFL_XylA EAA67023_ASPNI_BxlB BAE65591_ASPOR EED45224_ASPFL CAA32244_HYPJE_Bxl1 CAB69357_ARATH_AtBX3 AAK38462_HORVU CAW52613_ZEAMA ABI29899_THENE_Bg13B CAA33665_CLOTH BAA10968_ASPAC ABC88234_PSEBB_ExOP AD23382_HORVU_ExOI AAA64351_BACSU_YDbD AAF93857_VIBCH_NagZ AAQ05801_CELFI ABU76375_ENTSA EAA644470_ASPNI_BxlA	630 GTP. VYEFGHGL GTP. VYEFGHGL 	FYTTFE ESTETTD. FYTTFE ESTETTD. FYTTFH.ASLPGTG. FYTTFN.VSWAKKA. HYTTFN.VSWAKKA. HYTTFN.VSWNHAE. HYTTFN.VSWNHAE. FYTTFK.ETLASHP. TYTKFD.HQLIKAP. SYTTFS.YASGIG. SYTTFS.YASGIG. SYTTFE.YSDISVD. SYTTFE.YSDISVD. SYTTFN.YSGLHIQ. TYQDNINVPVLSEK. TT. KQQF.S. YHTFD.VPRVRT. SYTTFS.VSDVKLS.	AGSFNI. .KD.KTSFNI. TKN.RTSFNI. FGPYDA. YGPYNT. K.SLKFNT. R.LVSLS. .TK.PPSMSGI. A.TVTIHIGA. F. KK.D. VL.NASSNAQVA TSPKKTVNS.	WEHC. TETGAAPTF	GQVGNASDYVYPEGLTRISK
EAA64470 ASPNI BxlA ABA40420 ASPFU Xld EED57542 ASPFL XylA EAA57023 ASPNI BxlB BAE5591 ASPOR EED45224 ASPFL CAA93246 HYPJE Bxl1 CAB9357 ARATH ALEX3 AK38482 HORVU CAW52613 ZEAMA BB129899 THEME Bg13B CAA33665 CLOTH BAA10968 ASPAC ABC88234 PSEBB ExOP AAC4351 BACSU YbDD AAF93857 VIBCH NagZ AAQ05801 CELFI ABU76375 ENTSA EAA64470 ASPNI BxlA BAB0420 ASPFU Xld EED57542 ASPFL XylA EAA64470 ASPNI BxlA ABA64470 ASPNI BxlA ABA0420 ASPFU Xld EED57542 ASPFL XylA EAA67423 ASPNI BxlB BAE65591 ASPOR EED45224 ASPFL CAA93246 HYPJE Bxl1 CAB9357 TRATH ALEX3 AAX38482 HORVU CAW52613 ZEAMA AB129899 THEME Bg13B CAA33665 CLOTH BAA10966 ASPAC ABC88234 PSEBB ExOP AAD23382 HORVU ExOI AA64351 BACSU YbD	630 GTP. VYEFGHGL GTP. VYEFGHGL GTP. VYEFGHGL DA. VLPFGFGL KA. VLPFGFGL KA. VLPFGFGL GET. VYAFADAL GET. VYAFADAL GET. VYAFGYGL UEPAVLYPFGHGL IEP. LFPFGHGL VEPAYEFGYGL VEPAYEFGYGL GET. LYPFGHGL VDP. LFRLGYGL GNT. LYPLGYGL LK	640 FWTTFE.E.STETTD. FYTTFE.ASLPGTG.FYTNFT.ASSAGSG FYTNFT.ASSAGSG FYTTFN.VSWNHAE.FYTTFN.VSWNHAE.FYTTFN.SSNHAE.FYTFF.STLASHP.TYTKFD.HQLIKAP.SYTFF.YSDLNVS.SYTFF.YSDLNVS.SYTFF.YSDLNVS.SYTFF.YSDLNVS.SYTFF.YSDLNVS.SYTFF.S.VSDVKLS. 660		WEHC. TETGAAPTF GPH. EM. PEGATDG ASGAS	670 GQVGNASDYVYPEGLTRISK GYEHAQQKT GFANVEQMI GYAVVEQMI

Supplementary Fig. 2 (cont.)

	β8	β9			
EAA64470_ASPNI_BxlA				720	
	690	7	71	729	
ABA40420 ASPNI BX1A	L. LNFTATVKNTGER	SDYTALVYVN	TT AGPAPYPKK	WVVGFDRLGG.LE	• • • • • • • • • • • • • • • •
EED57542 ASPFL XVIA	L LNFTVDVKNTGDR	VSDYTAMAFVN	TT. AGPAPHPNK	WLVGFDRLSA.VE	
EAA67023 ASPNI Bx1B	VDTFSLAVTNTGDV	ASDYVALVFAS.	APELGAOPAPIK	TLVGYSRASL.IK	
BAE65591_ASPOR	F DTFSITVTNTGNV	ASDYIALLFLT.	ADGVGPEPYPIK	TLVGYSRAKG.IE	
EED45224_ASPFL	F DTFSITVTNTGNV	ASDYIALLFLT.	ADRVGPEPYPIK	CTLVGYSRAKG.IE	• • • • • • • • • • • • • • • •
CAA93248_HYPJE_Bx11	VFTFEANIKNSGKT	ESPYTAMLFVR	TSNAGPAPYPNK	WLVGFDRLAD.IK	
CABS9357 ARATH AtBX3	D FEVHLNVKNTGDR	AGSHTVFLFTT	SP.Q.VHGSPIK	QLLGFEKIRLG	
CAW52613 ZEAMA	V VSFSLRVANTGGV	GGDHVVPVVTA	DP PEVGDAPLE	OLVAFRRVF VP	
ABI29899 THENE Bol3B	T. LRVOYRIENTGGR	AGKEVSOVYIK.	AP.KGKIDKPFC	DELKAFHKTRL.LN	
CAA33665 CLOTH	I INVSVKVKNVGKM	AGKEIVQLYVK	DV.KSSVRRPEK	ELKGFEKVFLN	
BAA10968 ASPAC	L IRVSVTVKNTGRV	AGDAVPQLYVS	LGGPNEPKV	VVLRKFDRLTLK	
ABC88234 PSEBB_ExoP	LLMQSVNLSYQEDGR.	GFNWRAQAALS	LSYLEPTPLDSK	CFSTGYLELKMRIDKAP	EQGANLQVMCSESNC
AAD23382_HORVU_ExoI					
AAA64351 BACSU YEED				VORT V DI RDWO	· · · · · · · · · · · · · · · · · · ·
AAF93857_VIBCH_Nag2			V GSSPVVLE	TILAA	QASANM
ABU76375 ENTSA	K. VTASVTVTNTGKR	EGATAVOMYLO	DV. TASMSRPVK	OLRGFEKITLK	
		Dour for the former of the for		Enter Direct Control of Control	
		B10		811	B12
EAA64470 ASPNT Byla		-, o	.000	- >	
ADFMI_DAIA		730	740	750 76	0 770
EAA64470 ASPNT Byla	PGD	S.OTLTVPVTV	. ESVARTDROG	N. RVLYPGSVELAT.NN	ER. SVVVKFELKGEF
ABA40420 ASPFU X1d	PGD	S.QTMTIPVTI	.DSVARTDEAG.	N.RVLYPGKYELALNN	ER.SVVLOFVLTGRE
EED57542 ASPFL Xy1A	PGS	A.KTMVIPVTV	.DSLARTDEEG.	N.RVLYPGRYEVALNN	ER.EVVLGFTLTGEK
EAA67023 ASPNI Bx1B	PGE	T.RKVDVEVTV	.APLTRATEDG.	R.VVLYPGEYTLLVDV	NDEYPTAKFEIKGDV
BAE65591_ASPOR	PGQ	S.QQVKLDVSV	.GSVARTAENG.	D.LVLYPGSYKLEVDV	GQDFPTATFTVSGKE
BED45224_ASPFL	PGQ	S.QQVKLDVSV	.GSVARTAENG.	D.LVLYPGSYKLEVDV	GQDFPTATFTVSGKE
CAA93248 HYPJE Bx11	PGH	S.SKLSIPIPV	.SALARVDSHG.	N.RIVYPGKYELALNT	DE.SVKLEFELVGEE
AAK38482 HOPVII		A. AHVEFEVED	CKULSVVDETG.	R. KVIDOGSHFVRVCD	
CAW52613 ZEAMA		A. VDVPFALNV	CKTFAIVEETA	Y. TVVPSGVSTVVVGD	DALVLSF
ABI29899 THENE Bg13B	PGE	S.EEVVLEIPV	.RDLASFNGE.	E.WVVEAGEYEVRVGA	SSRNIKLKGTFSV
CAA33665_CLOTH	PGE	E.KTVTFTLDK	.RAFAYYNTQIK	D.WHVESGEFLILIGR	SSRDIVLKESVRVNS
BAA10968 ASPAC	PSE	E.TVWTTTLTR	.RDLSNWDVAA.	QDWVITSYPKKVHVGS	SSRQLP
ABC88234_PSEBB_ExoP	LRDIDFSSFSQLMADK	SWHTLAIPLHC	.DDSDQAEQPI.	T.DALRITS	QNLSLAIAD
AAD23382 HORVU ExoI				• • • • • • • • • • • • • • • • • • • •	
AAR93857 VIBCH Nac7					
AA005801 CELFI			LOG	KAPFAGSSPV	DA
ABU76375 ENTSA	PGE	S.QTVSFPIDI	. DALKFWNQRM.	K.YDAEPGKFNVFIGL	DSARVK
EAA64470 ASPNI BxlA					
				7	80
EAA64470 ASPNI Bx1A	ΔV			TLS	WPED
ABA40420 ASPFU X1d	AV			VFK	WPVE
EED57542 ASPFL Xy1A	AV			LFK	WPKE
EAA67023_ASPNI_Bx1B	QV			LEK	FPLS
BAE65591_ASPOR	KV			LDE	FPEP
EED45224_ASPFL	KV			LDE	FPEP
CAR93248 HIPJE BXII	V1				WPLE
AAK38482 HORVU					
CAW52613 ZEAMA					. PV
ABI29899 THENE Bg13B					
CAA33665_CLOTH	TVKIRKRFTVNSAVED	VMSDSSAAAVL	GPVLKEITDALQ	IDMDNAHDMMAANIKN	MPLRSLVGYSQGRLS
BAA10968 ASPAC					
ABC88234_PSEBB_ExoP	VA				KPSD
AAD23382 HORVU EXOI					• • • • • • • • • • • • • • • • •
AAF93857 VIBCH Nacz					
AAQ05801 CELFI					
ABU76375 ENTSA					
EAA64470 ASPNI Bx1A					
	790	800			
EAA64470 ASPNT Bx14	.T. TSDEVSSIDGALD	RKODVIA			
ABA40420 ASPFU X1d		QQQISSA			
EED57542 ASPFL Xy1A		EQLIAPQ			
EAA67023 ASPNI Bx1B		GNDSD			
BAE65591_ASPOR	.Q.QNATS	AVTRRGR			
EED45224_ASPFL	.Q.QNATS	AVTRWGR			
CAA93248 HYPJE Bx11	.в.001	CLNICY			
AAK38482 HOPVII		RISEMA			
CAW52613 ZEAMA		.TINLAV			
ABI29899 THENE Bol3B		EERRFKP			
CAA33665_CLOTH	EEMLEELV	DKINNVE			
BAA10968 ASPAC	LH	AALPKVQ			
ABC88234_PSEBB_ExoP	DS	ISLTCAK			
AAD23382_HORVU_ExoI		. NATKKY			
AAA64351 BACSU YbbD		LIFORGE			
AAC05801 CELET	QR	GOWDTHL			

Supplementary Fig. 2 Amino acid sequence alignment of BxIA (GenBank ID: EAA64470) and BxIB (GenBank ID: EAA67026) from *A. nidulans* FGSC A4 with related of functionally characterized GH3 enzymes; β -xylosidases: *A. fumigatus* Af293 (GenBank ID: ABA40420), *A. flavus* NRRL 3357 (GenBank ID: EED57542), *A. oryzae* RIB40 (GenBank ID: BAE65591), *A. flavus* NRRL 3357 (GenBank ID: EED45224), *H. jecorina* RUTC-30 (GenBank ID: CAA93248), *Hordeum* vulgare (GenBank ID: AAK38482), Zae mays BZ3 (GenBank ID: CAW52613); β -glucosidases: *T. neapolitana* DSM 4359 (GenBank ID: ABI29899), *Clostridium thermocellum* DSM 1237 (GenBank ID: CAA33665), *A. aculeatus* F-50 (GenBank ID: BAA10968); exo-1,3/1,4-glucanases: *Pseudoalteromonas* sp. BB1 (GenBank ID: ABC88234), *Hordeum* vulgare (GenBank ID: AAD23382); β -N-acetylglucosaminidases: *Bacillus subtilis* subsp. subtilis str. 168 (GenBank ID: AAA64351), *Vibrio cholerae* O1 biovar El Tor str. N16961 (GenBank ID: AAF93857), *Cellulomonas fimi* ATCC 484 (GenBank ID: AAQ05801), *Enterobacter sakazakii* ATCC BAA-894 (GenBank ID: ABU76375); Nuc, catalytic nucleophile; A/B, catalytic acib/base; \blacktriangledown , mutation position for C308W; \blacktriangledown , mutation position for N313R



Supplementary Fig. 3 Amino acid sequence alignment of ManC and ManA from *A. nidulans* FGSC A4 with reported crystal structures of GH5 mannanases (see detail on next page).

ManC ManA HjMan5A_1qnr SPLYGQCGGSGYTGPTCCAQGTCIYSNYWYSQCLNT SlMan4_1rh9 CmMan5A_1uuq MeMan5A_2c0h TfMan_1bqc

ManC

Supplementary Fig. 3 Amino acid sequence alignment of ManC and ManA from *A. nidulans* FGSC A4 with reported crystal structures of GH5 mannanases: *Trichoderma reesei* (Teleomorph: *H. jecorina*; UniProt ID: Q99036), *Solanum lycopersicum* (UniProt ID: Q8L5J1), *Mytilus edulis* (UniProt ID: Q8WPJ2), *Thermobifida fusca* KW3 (UniProt ID: Q9ZF13), and *Bacillus* sp. JAMB-602 (UniProt ID: Q4W8M3). ▼), possible subsite -2; ▼), possible subsite +1 and +2; ▼), possible subsite +3; ▼), mutation position (subsite +1).

10.2 List of publications

10.2.1 Oral presentations

20 Nov 2007. Xylooligosaccharides, xylanases, and xylosidases as tools for novel prebiotics production. Protein and Enzyme Engineering Workshop, EPC, BioCentrum-DTU.

07 Jan 2007. Chemo-enzymatic synthesis of prebiotics: xylooligosaccharides, xylanases, and xylosidases: as tools for novel prebiotics production. 07-annual meeting, Center for Biological Produciton of Dietary Fibres and Prebiotics, Department of Chemical Engineering, DTU.

07 Apr 2007. Chemo-enzymatic synthesis of prebiotics: xylooligosaccharides, xylanases, and xylosidases: as tools for novel prebiotics production, an update status. Seminar at Danisco, Langebrogade, Copenhagen.

14 May 2008. The mutation plan for xylosidase (GH3) and xylanase (GH10). Protein and Enzyme Engineering Workshop, EPC, Department of Systems Biology, DTU.

22 Jul 2008. Production of xylooligosaccharides by transglycosylation of β -xylosidase and xylanase. Protein and Enzyme Engineering Workshop, EPC, Department of Systems Biology, DTU.

29 Oct 2008. Production of xylooligosaccharides by transglycosylation of β -xylosidase and xylanase. Center for Biological Produciton of Dietary Fibres and Prebiotics, Department of Chemical Engineering, DTU.

23 Jun 2009. GH3 β -xylosidases characterization and transglycosylation & update status of GH61 putative endoglucanase. Center for Biological Produciton of Dietary Fibres and Prebiotics Workshop, Comwell Holte Hotel, Holte.

As well as, nine EPC group meetings, EPC, Department of Systems Biology, DTU.

10.2.2 Poster presentations

Dilokpimol A., H. Nakai., M.J. Baumann, M. Abou Hachem & B. Svensson: The production of xylanases and xylosidase from *Aspergillus nidulans*. 27818 Enzyme Technology, Lignocellulose, and Non-Starch Polysaccharides, Lyngby, Denmark, May 2008.

Dilokpimol A., H. Nakai., M.J. Baumann, M. Abou Hachem & B. Svensson: Recombinant production of xylanases and xylosidase from *Aspergillus nidulans*. Protein Science Meeting, Tisvildeleje, Denmark, Nov 2008.

Posters and conference talks (given by others):

Nakai, H., A. Majumder, **A. Dilokpimol**, M. Ejby, M. Abou Hachem, S. Lahtinen, S. Jacobsen & B. Svensson: The relationship between probiotics and prebiotic carbohydrates. Chemoenzymatic synthesis and differential proteome analysis. 4th Conference on Biocatalysis in the Food and Drink Industries, Toulouse, France, April 2010.

Seo, E.-S., M.M. Nielsen, J.M. Andersen, M.B. Vester-Christensen, J.M. Jensen, C. Christiansen, A. **Dilokpimol**, M. Abou Hachem, P. Hägglund, K. Maeda, C. Finnie, A. Blennow & B. Svensson: α -amylases. Interaction with polysaccharide substrates, proteinaceous inhibitors and regulatory proteins. Agricultural Biotechnology Symposium, Seoul, Korea, September 2008.

Svensson, B., M.M. Nielsen, E.-S. Seo, J. Andersen, A. Dilokpimol, C. Christiansen, S. Bozonnet, M. Abou Hachem, A. Blennow, R. Haser & N. Aghajari: The many facets of the molecular interactions and mechanisms involved in carbohydrate recognition and processing by barley α -amylase. AACC International Annual Meeting, Honolulu, Hawaii, September 2008.

Svensson, B., E.-S. Seo, M.M. Nielsen, L. Kandra, J. Andersen, **A. Dilokpimol**, M. Abou Hachem, G. Gyémánt, R. Haser, N. Aghajari, S. Bozonnet & A. Blennow: Polysaccharide multi-site interactions in α -amylase: From surface site-active site cross-talk to starch granule recognition. Plant Polysaccharide Workshop, Sigtuna, Sweden, August 2008.

Svensson, B., M.M. Nielsen, E.-S. Seo, J.M. Andersen, **A. Dilokpimol**, S. Bozonnet, M. Abou Hachem, A. Blennow, R. Haser & N. Aghajari: Roles of individual carbohydrate binding surface sites in barley α -amylase. 24th International Carbohydrate Symposium, Oslo, Norway, July 2008.

Andersen, J., M.M. Nielsen, **A. Dilokpimol**, S. Bozonnet, M. Abou Hachem, A. Blennow & B. Svensson: Structure-function relationship investigations and protein engineering of carbohydrate binding surface sites of barley α -amylase 1. Plant Biotech Denmark Annual Meeting, Frederiksberg, Denmark, January 2008.

Nielsen, M.M., E.-S. Seo, J. Andersen, **A. Dilokpimol**, S. Bozonnet, M. Abou Hachem, H. Naested, A. Blennow & B. Svensson: Mutational analysis of individual roles of carbohydrate binding surface sites in barley α -amylase 1. ALAMY-3, Smolenice, Slovakia, September 2007.

Seo, E.-S., M. Abou Hachem, M.M. Nielsen, **A. Dilokpimol**, K. Fukuda & B. Svensson: Mutational analysis of sugar tongs in barley α -amylase 2. ALAMY-3, Smolenice, Slovakia, September 2007.

Abou Hachem, M., W.-F. Xue, **A. Dilokpimol**, B.W. Sigurskjold & B. Svensson: Calcium binding to barley α -amylase isozymes and its implications on stability and activity. ALAMY-3, Smolenice, Slovakia, September 2007.

Nielsen, M.M., E. S. Seo, **A. Dilokpimol**, M. Abou Hachem, S. Bozonnet, H. Naested & B. Svensson: Mutational analysis of the function of surface binding sites in barley α -amylase. VIIth European Symposium of the Protein Society, Stockholm/Uppsala, Sweden, May 2007.

Nielsen, M. M., E.-S. Seo, J. Andersen, **A. Dilokpimol**, S. Bozonnet, M. Abou Hachem, H. Naested, A. Blennow & B. Svenson: Site-directed mutagenesis of secondary carbohydrate binding sites in barley α -amylase 1. 7th Carbohydrate Bioengineering Meeting, Braunschweig, Germany, April 2007.

Svensson, B., M.M. Nielsen, E.-S. Seo, **A. Dilokpimol**, J. Andersen, M. Abou Hachem, H. Naested, M., Willemoës, S. Bozonnet, L. Kandra, G. Gyémánt, N. Aghajari & R. Haser: Roles of multiple surface sites, long substrate binding clefts, and carbohydrate binding modules in the action of amylolytic enzymes on polysaccharide substrates. 7th Carbohydrate Bioengineering Meeting, Braunschweig, Germany, April 2007.

Nielsen, M.M., E.-S. Seo, M. Abou Hachem, **A. Dilokpimol** & B.Svensson: Molecular enzymology of starch degradition: Workshop on Food and Feed – Nutrition, Safety and Improved Use of Raw Material, Hyderabad, India, January 2007.

Nielsen, M.M., E.-S. Seo, J. Andersen, A. Dilokpimol, S. Bozonnet, M. Abou Hachem, H. Naested, V. Barkholt & B. Svensson: Site-directed mutagenesis of secondary sugar binding site in barley α -amylase 1. 23rd International Carbohydrate Symposium, Whistler, Canada, July 2006.

Nielsen, M.M., E.-S. Seo, J. Andersen, **A. Dilokpimol**, S. Bozonnet, B. Kramhøft, M. Abou Hachem, H. Naested, V. Barkholt & B. Svensson: Site-directed mutagenesis of secondary sugar binding site in barley α -amylase 1. INPEC Annual Meeting, Elsinore, Denmark, 2006.

10.2.3 Manuscripts and publications

Dilokpimol, A., Nakai, H., Gotfredsen, C. H., Appeldoorn, M., Baumann, M. J., Nakai, N., Schols, H. A., Abou Hachem, M., Svensson, B. Enzymatic synthesis of β -xylosyl-oligosaccharides by transxylosylation using two β -xylosidases of glycoside hydrolase family 3 from *Aspergillus nidulans* FGSC A4, submitted to *Carbohydrate Research* (see 10.2.3.1)

Dilokpimol, A., Nakai, H., Gotfredsen, C. H., Baumann, M. J., Nakai, N., Abou Hachem, M., Svensson, B. GH5 endo- β -1,4-mannanases from *Aspergillus nidulans* FGSC A4 producing hetero manno-oligosaccharides by transglycosylation, to be submitted to *Biochimica et Biophysica Acta, Proteins and Proteomics* (see 10.2.3.2)

Nakai, H., **Dilokpimol, A.**, Abou Hachem, M., Svensson, B. Efficient one-pot enzymatic synthesis of α -(1 \rightarrow 4)-glucosidic disaccharides through a coupled reaction catalysed by Lactobacillus acidophilus NCFM maltose phosphorylase. *Carbohyd. Res.* 2010. 345: 1061-1064.

Nakai, H., Baumann, M.J., Petersen, B.O., Westphal, Y., Schols, H.A., **Dilokpimol, A.**, Abou Hachem, M., Lahtinen, S.J., Duus, J.Ø., Svensson, B. The maltodextrin transport system and metabolism in *Lactobacillus acidophilus* NCFM and production of novel α -glucosides through reverse phosphorolysis by maltose phosphorylase. *FEBS J.* 2009. 276: 7353-7365.

Nakai, H., Nakai, N., Petersen, B.O., Westphal, Y., **Dilokpimol, A.**, Abou Hachem, M., Lahtinen, S.J., Duus, J.Ø., Schols, H.A., Svensson, B. Substrate recognition by two members of glycoside hydrolase family 32 involved in fructo-oligosaccharide metabolism in *Lactobacillus acidophilus* NCFM. *In preparation*.

Nakai, H., Baumann, M.J., Petersen, B.O., Westphal, Y., Abou Hachem, M., **Dilokpimol, A.**, Duus, J.Ø., Schols, H.A., Svensson, B. *Aspergillus nidulans* α -galactosidase of glycoside hydrolase family 36 catalyses formation of α -galacto-oligosaccharides by transglycosylation. *FEBS J. Submitted*.

Nakai, H., Abou Hachem, M., Petersen, B.O., Westphal, Y., Mannerstedt, K., Baumann, M.J., **Dilokpimol, A.**, Schols, H.A., Duus, J.Ø., Svensson, B. Efficient chemoenzymatic oligosaccharide synthesis by reverse phosphorolysis using cellobiose phosphorylase and cellodextrin phosphorylase from *Clostridium thermocellum. Biochimie. Submitted.*

Nakai, H., Petersen, B.O., Westphal, Y., **Dilokpimol, A.**, Abou Hachem, M., Duus, J.Ø., Schols, H.A., Svensson, B. Rational engineering of *Lactobacillus acidophilus* NCFM maltose phosphorylase into either trehalose or kojibiose dual specificity phosphorylase. *Protein Eng. Des. Sel. In preparation.*

Publications (with other projects)

Nielsen, M.M., Bozonnet, S., Seo, E.-S., Motyan, J.A., Andersen, J.M., **Dilokpimol, A.**, Abou Hachem, M., Gyemant, G., Næsted, H., Kandra, L., Sigurskjold, B.W., Svensson, B. Two secondary

carbohydrate binding sites on the surface of barley α -amylase 1 have distinct functions and display synergy in hydrolysis of starch granules. *Biochemistry*. 2009. 48: 7686-7697.

Nielsen, M.M., Seo, E.-S., **Dilokpimol, A.**, Andersen, J.M., Abou Hachem, M., Naested, H., Willemoës, M., Bozonnet, S., Kandra, L., Gyémánt, G., Haser, R., Aghajari N., Svensson, B. Roles of multiple surface sites, long substrate binding clefts, and carbohydrate binding modules in the action of amylolytic enzymes on polysaccharide substrates. *Biocatal. Biotransform.* 2008. 26: 59-67.

Seo, E.-S., Nielsen, M.M., Andersen, J.M., Vester-Christensen, M.B., Jensen, J.M., Christiansen, C., **Dilokpimol, A.**, Abou Hachem, M., Hägglund, P., Maeda, K., Finnie, C., Blennow, A., Svensson, B. α -amylases. Interaction with polysaccharide substrates, proteinaceous inhibitors and regulatory proteins, in: H.-K. Park (Ed), *Carbohydrate-active enzymes: Structure, function and applications*, Woodhead Publishing Ltd., Cambridge, United Kingdom, 2008, pp 20-36.
Manuscript I

Enzymatic synthesis of β -xylosyl-oligosaccharides by transxylosylation using

two β -xylosidases of glycoside hydrolase family 3

from Aspergillus nidulans FGSC A4

Adiphol Dilokpimol^a, Hiroyuki Nakai^a, Charlotte H. Gotfredsen^b, Maaike Appeldoorn^c, Martin J.

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Graphical abstract

Enzymatic synthesis of novel β -xylosyl-oligosaccharides catalysed by transglycosylation of two recombinant *Aspergillus nidulans* FGSC A4 β -xylosidases of glycoside hydrolase family 3.



R, monosaccharides, disaccharides and sugar alcohols

Enzymatic synthesis of β -xylosyl-oligosaccharides by transxylosylation using

two β-xylosidases of glycoside hydrolase family 3

from Aspergillus nidulans FGSC A4

Adiphol Dilokpimol^a, Hiroyuki Nakai^a, Charlotte H. Gotfredsen^b, Maaike Appeldoorn^c, Martin J. Baumann^a, Natsuko Nakai^a, Henk A. Schols^c, Maher Abou Hachem^a, Birte Svensson^a*

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Abstract

Two β-xylosidases of glycoside hydrolase family 3 (GH 3) from Aspergillus nidulans FGSC A4, BxlA and BxlB were produced recombinantly in Pichia pastoris and secreted to the culture supernatants in yields of 16 and 118 mg/L, respectively. BxIA showed about 6 fold higher catalytic efficiency (k_{cat}/K_m) than BxlB towards *para*-nitrophenyl β -D-xylopyranoside (pNPX) and β -1,4xylo-oligosaccharides (degree of polymerisation 2-6). For both enzymes k_{cat}/K_m decreased with increasing β -1,4-xylo-oligosaccharide chain length. Using *p*NPX as donor with 9 monosaccharides, 7 disaccharides, and two sugar alcohols as acceptors 18 different β-xylosyl-oligosaccharides were synthesised in 2-36% (BxlA) and 6-66% (BxlB) yields by transxylosylation. BxlA utilised the monosaccharides D-mannose, D-lyxose, D-talose, D-xylose, D-arabinose, L-fucose, D-glucose, Dgalactose, and D-fructose as acceptors, whereas BxlB used the same except for D-lyxose, Darabinose, and L-fucose. BxlB transxylosylated the disaccharides xylobiose, lactulose, sucrose, lactose, and turanose in up to 35% yield, while BxIA gave inferior yields. The regioselectivity was acceptor dependent and primarily involved β -1,4 or 1,6 product linkage formation although minor products with different linkages were also obtained. Five of the 18 transxylosylation products obtained from D-lyxose, D-galactose, turanose and sucrose (two products) as acceptors were novel xylosyl-oligosaccharides, β -D-Xylp-(1 \rightarrow 4)-D-Lyxp, β -D-Xylp-(1 \rightarrow 6)-D-Galp, β -D-Xylp-(1 \rightarrow 4)- α -D-Glcp-(1 \rightarrow 3)- β -D-Fruf, β -D-Xylp-(1 \rightarrow 4)- α -D-Glcp-(1 \rightarrow 2)- β -D-Fruf, and β -D-Xylp-(1 \rightarrow 6)- β -D-Fruf-($2\rightarrow 1$)- α -D-Glcp, as structure-determined by 2D NMR, indicating that GH3 β -xylosidases are able to transxylosylate a larger variety of carbohydrate acceptors than earlier reported. Furthermore, transxylosylation of certain acceptors resulted in mixtures of products of which the individual structures have not been determined. Some of these products are also novel.

Keywords: Glycoside hydrolase family 3; Transglycosylation; β-Xylosyl-oligosaccharides; β-Xylosidase; Regioselectivity; NMR.

Abbreviations: BxlA and BxlB, *Aspergillus nidulans* FGSC A4 GH3 β -xylosidases, encoded by *bxlA* (GenBank EAA64470.1) and *bxlB* (GenBank EAA67023.1); DP, degree of polymerisation; ESI-MS, electrospray ionization mass spectrometry; GH, glycoside hydrolase family; HPAEC-PAD, high performance anion exchange chromatography with pulsed amperometric detection; HPLC, high performance liquid chromatography; IMAC, immobilized metal ion affinity chromatography; NMR, nuclear magnetic resonance spectroscopy; NMWL, nominal molecular weight limit; NOE, nuclear Overhauser effect; *pNP*, *para*-nitrophenol; *pNPA*, *para*-nitrophenyl α -L-arabinofuranoside; *pNPX*, *para*-nitrophenyl β -D-xylopyranoside; X2, xylobiose; X3, xylotriose; X4, xylotetraose; X5, xylopentaose; X6, xylohexaose; XOS, β -xylo-oligosaccharides.

1. Introduction

β-Xylo-oligosaccharides (often referred to as XOS) are oligomers of β -1,4 linked xylose which act as prebiotics and have important applications in the food industry as functional ingredients promoting growth of probiotic bacteria in the human gut.¹⁻³ XOS occur naturally in fruits and vegetables and are produced commercially by enzymatic and/or chemical hydrolysis of xylan for example from barley hulls, brewery spent grains, corn cobs, corn stover, rice hulls, wheat straw, and bamboo.^{2,3} Most heteropolysaccharides xylans are containing different substituents such as (4-O-methyl)-a-Dglucuronopyranosyl, α -L-arabinofuranosyl, acetyl, feruloyl and coumaroyl residues.^{2,3} Complete degradation of xylans therefore requires a battery of glycoside hydrolases and esterases including endo- β -1,4-xylanases (EC 3.2.1.8), exo- β -1,4-xylosidases (EC 3.2.1.37), α-glucuronosidases (EC 3.2.1.139), α -L-arabinofuranosidases (EC 3.2.1.55), acetylxylan esterases (EC 3.1.1.72), ferulic acid esterases (EC 3.1.1.73), and p-coumaric acid esterases (EC 3.1.1.-).⁴⁻⁸ Synergistic action of such xylanolytic enzymes accomplishes the degradation of xylan.^{5,8–10} which is central e.g. in production of second generation biofuels from lignocellulosic biomass.^{8,9,11}

The $exo-\beta-1,4$ -xylosidases (β -xylosidases) catalyse hydrolysis of \beta-xylo-oligosaccharides to release xylose from the non-reducing end and are classified into five glycoside hydrolase families GH3. 30, 39, 43 and 52, based on sequence similarity (http://www.cazy.org).8,9,12 Except for GH43, these families use the double-displacement (retaining) mechanism where the substrate glycone part forms a covalent glycosyl-enzyme intermediate that is attacked by water in case of hydrolysis and by a carbohydrate leading to formation of a new glycosidic linkage in the case of transglycosylation (Fig. 1).¹³⁻¹⁵ Retaining GH3 β-xylosidases have been reported to use *para*-nitrophenyl β-Dxylopyranoside (pNPX) or β -xylo-oligosaccharides of degree of polymerisation (DP) 2 and 3 in transxylosylations resulting in $pNP \beta$ -xylo-

oligosaccharides or longer β-xylo-oligosaccharides with mainly β -1,4 linkages.^{16–19} β -1,6 linkages, however, were formed predominantly when using hexose-containing acceptors.^{20,21} There are a few addressing studies GH3 β-xylosidase transxylosylation with acceptors other than xylose or β -xylo-oligosaccharide, different i.e., disaccharides,^{20,21} monosaccharides,²⁰ sugar alcohols,^{20,21} aliphatic or aromatic alcohols,^{21–23} and hydroquinones.²⁴ Although this seems to indicate a broad acceptor specificity, the transglycosylation capacity of GH3 \beta-xylosidases was in fact previously only tested with a few carbohvdrate (three mono- and five disaccharide) acceptors.^{20,21} Indeed insight lags behind as in the diversity of acceptor preference of GH3 β-xylosidases and this motivated the present screening of acceptor specificity.

GH3 β-xylosidases are found in prokaryotes and eukaryotes, but have been purified and characterised mostly from fungi such as Aspergillus spp.^{13,20,25} and Hypocrea jecorina (anamorph Trichoderma reesei);^{13,18,26} some recombinant GH3 β-xylosidases been have obtained previously.^{10,11,27-29} Two GH3 A. nidulans FGSC A4 β -xylosidases, BxlA and BxlB, has of 45% sequence identity activity of which hydrolytic and biochemical properties of native BxlA (previously called XlnD; from A. nidulans CECT2544) were subjected to limited characterisation,²⁵ whereas BxlB was not isolated. BxlA and BxlB were recombinantly, produced previously but characterised very superficially.^{27,30} However, in the present work, BxlA and BxlB were efficiently produced in *Pichia pastoris* and using a large variety of acceptors, enzymatic synthesis was achieved of 18 hetero β -xylosyl-oligosaccharides having a β xvlosvl residue at the non-reducing end, which may be prebiotic candidates. The present comparative transxylosylation reveals that isozymes can vary considerably in transxylosylation capacity and acceptor preference, which has a direct impact on their potential as synthetic tools.



Figure 1. Schematic diagram of hydrolysis and transxylosylation reactions catalysed by GH3 β -xylosidases. A/B, acid/base catalyst; Nuc, catalytic nucleophile; R₁, R₂, carbohydrate residues.

2. Results and discussion

2.1. Production and purification of recombinant BxlA and BxlB

A. nidulans FGSC A4 BxlA and BxlB were obtained as His-tag fusion proteins by secretory expression in *P. pastoris* in yields of 16 and 118 mg/L culture supernatant, respectively. BxlB was purified by immobilized metal ion affinity chromatography (IMAC) followed by gel filtration. BxlA, however, had to be purified by anion exchange chromatography and gel filtration due to its instability at pH 7.5 (see 2.2, Fig. 2B) used for IMAC purification. IMAC purification was tried at pH 5.5, but gave poor purification yield. In comparison approximately 0.5 mg/L native BxlA (called XlnD) was purified from an *A. nidulans* CECT2544 culture.²⁵

The isoelectric point (pI) of BxlA and BxlB was 3.0 (Supplementary Fig. S1A), similar to pI =3.4 of native BxlA purified from A. nidulans CECT2544.²⁵ BxlA and BxlB gave single bands in SDS-PAGE with apparent molecular masses of 110 and 106 kDa, respectively (Supplementary Fig. S1B). These are higher than the theoretical values of 86045.2 and 92402.0 Da, respectively. Endoglycosidase H treatment of BxIA and BxIB reduced the size to 86 and 92 kDa, respectively (Supplementary Fig. S1B) suggesting some of the 10 and 8 putative sites predicted in BxlA and BxlB

to be N-glycosylated, respectively. The total carbohydrate content was estimated to 18% and 29% (w/w) for BxlA and 10% and 17% for BxlB with mannose and glucose as standards, respectively. Such native BxlA secreted from A. nidulans CECT2544²⁵ was not glycosylated, the glycosylation of recombinant BxlA was expression host-related.³¹ Gel filtration indicated that BxIA is a dimer of 220 kDa, similarly to native BxlA from A. nidulans CECT2544²⁵ and A. awamori GH3 Bxylosidase,¹⁹ and that BxlB is a monomer of 135 kDa, as previously found for a GH3 β-xylosidase from A. *japonica*.³²

2.2. Biochemical properties and substrate specificity

BxlA and BxlB hydrolysed *p*NPX with pH optimum of 5.0 and 5.2 (Fig. 2A) and were stable at pH 4.0–6.0 and 4.0–8.0, respectively (Fig. 2B). Maximum activity of BxlA was at 50°C and of BxlB was at 60°C (Fig. 2C) and both enzymes retained >95% activity after 5 h at up to 40°C and 55°C, respectively (Fig. 2D). Other GH3 β -xylosidases show pH and temperature optima in the ranges of pH 4–6 and 50–70°C.^{13,18,20,25,26} BxlB displayed broader pH and temperature stabilities than BxlA. pH and temperature stabilities of GH3 β -xylosidases from *A. nidulans* CECT2544²⁵ and *A. awamori*²⁰ were similar to BxlA, thus glycosylation of BxlA by *P. pastoris* did not confer extra stability.

pNPX and xylobiose were the best substrates and **BxlB** for **BxlA** (Table 1). The k_{cat}/K_m values gradually decreased with the size of β -1,4-xylo-oligosaccharides (DP 3–6). Activity para-nitrophenyl towards α-L- $(k_{\rm cat}/K_{\rm m})$ arabinofuranoside (pNPA) was essentially the same for BxlA and BxlB and was 100 and 13 fold lower than the activity of these two enzymes towards pNPX. H. jecorina¹⁸ and Hordeum vulgare³⁴ GH3 β-xylosidases also showed low activity towards pNPA. β -D-Xylopyranose preferably adopts a pyranose ring conformation having equatorial hydroxyl groups, which resemble the conformation of α -L-arabinofuranose (Fig. 3).³⁵ The structural energetically favourable similarity of the the conformations explain may α-Larabinofuranosidase side-activity displayed by Bxylosidases.

BxlA produced by *P. pastoris* had three fold higher k_{cat} (112 s⁻¹) towards *p*NPX (Table 1) but the same K_m (1.3 mM) as native BxlA purified from *A. nidulans* CECT2544,²⁵ while BxlB had the same K_m , but 6 fold lower k_{cat} (18 s⁻¹; Table 1). These values are comparable to those of GH3 β -xylosidases from *A. awamori*,^{19,33} *H. jecorina*¹⁸ and *H. vulgare*,³⁴ which have K_m and k_{cat} for *p*NPX ranging from 0.23 to 4.1 mM and from 11 to 176 s⁻¹, respectively. BxlA showed 5–7 fold higher k_{cat}/K_m towards β xylo-oligosaccharides than BxlB (Table 1). The shared preference for shorter xylooligosaccharides of BxlA and BxlB is in accordance with other GH3 β -xylosidases.^{9,11,18}

2.3. Transxylosylation catalysed by BxlA and BxlB

The transxylosylation ability was investigated using *p*NPX as donor with 10 monosaccharides, six disaccharides, and two sugar alcohols as acceptor candidates for BxIA, and 12 monosaccharides and 13 disaccharides in case of BxIB (Table 2). The progress of transxylosylation was monitored over 5 h (Fig. 4). Mostly more than one product was detected by HPAEC-PAD (not shown). Total product yields (products/donor mM/mM) were calculated for each acceptor based on the sum of all oligosaccharide products. As monitored here, the reaction yields in most cases showed a maximum after which they decreased. The obtained maximum yields are listed in Table 2. Noticeably, the extent of product decrease differed for the individual acceptors (Fig. 4A-F), reflecting variation in hydrolysis of transxylosylation products or their participation acceptors subsequent as in transxylosylations or both. From these two possibilities, hydrolysis is the most probable, as activities are lower on longer substrates as judged from catalytic efficiencies on xylooligosaccharides (Table 1), and transxylosylation yields are lower on disaccharides compared to monosaccharides (Table 2). Noticeably, transxylosylation products of Dgalactose (for both BxlA and BxlB; Fig. 4B,E), Dfructose (for BxlB; Fig. 4E), sorbitol (for BxlA; Fig. 4A) and lactose (for BxlB; Fig. 4F) remained at the maximumlevel without subsequent decline suggesting a much slower hydrolysis rate than of products formed with other acceptors.

BxlA and BxlB both possess broad acceptor specificity (Table 2). For BxlA the highest yield was with xylitol (36%), while D-mannose, D-lyxose and D-talose were transxylosylated with a similar maximum yield compared to as D-xylose (>20%). Transxylosylation of D-arabinose, L-fucose, Dglucose, D-galactose and D-fructose was less efficient and L-rhamnose seemed not to act as acceptor (Table 2). Among 12 monosaccharides, BxlB preferred D-mannose, D-xylose, D-glucose and D-talose as acceptors as judged by high product yields (43-66%), while yields with D-galactose and D-fructose were only 16%. Transxylosylation yields for BxlB were in general higher than for BxlA. Noticeably except for D-xylose, BxlB did not accommodate pentoses (D-arabinose, L-arabinose, Dlyxose, L-lyxose). Also the deoxy sugars L-fucose and L-rhamnose were not acceptors for BxlB. BxlA in contrast used arabinose, lyxose and L-fucose as acceptors. **BxlB** efficiently transxylosylated lactulose, sucrose and turanose, and gave lower yields with lactose and xylobiose (Table 2), whereas disaccharides were poor acceptors for BxlA (Table 2, Fig. 4C).



Figure 2. Effect of pH and temperature on activity and stability of BxlA and BxlB. A) pH-dependence for activity towards 2 mM pNPX by 71 nM BxlA (•) and 179 nM BxlB (□) in 40 mM Britton-Robinson buffer. B) pH-stability of 355 nM BxlA (•) and 895 nM BxlB (□) in 40 mM Britton-Robinson buffer after 5 h incubation at 40°C. C) Temperature-activity dependence for 71 nM BxlA (•) and 179 nM BxlB (□) at 25–80°C with 10 min reaction time. D) Thermal stability of 355 nM BxlA (•) and 895 nM BxlB (□) incubated 5 h at 25–80°C in 50 mM sodium acetate pH 5.0. Each experiment was made in triplicate. Standard deviations are shown as error bars.

		BxlA		_		BxlB	
	$k_{\rm cat}$	$K_{ m m}$	$k_{\rm cat}/K_{\rm m}$		$k_{\rm cat}$	K _m	$k_{\rm cat}/K_{\rm m}$
Substrate	s^{-1}	mM	s ⁻¹ ·mM ⁻¹	-	s^{-1}	mM	s ⁻¹ ·mM ⁻¹
Xylobiose	139 ± 7.8	2.0 ± 0.07	70		36 ± 0.7	2.5 ± 0.40	14
Xylotriose	122 ± 7.8	2.1 ± 0.14	58		28 ± 0.1	3.0 ± 0.06	9
Xylotetraose	104 ± 1.1	2.5 ± 0.11	42		25 ± 0.2	3.2 ± 0.15	8
Xylopentaose	93 ± 4.7	2.8 ± 0.15	33		23 ± 0.2	4.3 ± 0.15	5
Xylohexaose	65 ± 1.5	3.0 ± 0.14	22	_	22 ± 0.5	5.2 ± 0.58	4
<i>p</i> NPX	112 ± 1.3	1.3 ± 0.16	90		18 ± 0.4	1.3 ± 0.11	14
<i>p</i> NPA	_	_	0.9		_	_	1.1

Table 1. Kinetic parameters of BxlA and BxlB for hydrolysis of β -xylo-oligosaccharides, *p*NPX and *p*NPA.

^a BxlA secreted from A. *nidulans* CECT2544 gave $k_{cat} = 37 \text{ s}^{-1}$ and $K_m = 1.1 \text{ mM}^{25}$

^b Low affinity precludes determination of $K_{\rm m}$ and $k_{\rm cat}$

Standard deviations are calculated from triplicate experiments.







α-L-Arabinofuranose

Figure 3. Preferred conformational structures of β -D-xylopyranose and α -L-arabinofuranose.

Accontor	Total y	ield (%)	Accontor	Total yield (%)	
Acceptor	BxlA	BxlB		BxlA	BxlB
Monosaccharides			Disaccharides		
D-Xylose	21 ^b	64 ^b	Xylobiose	_	6
D-Arabinose	17 ^b	n.d.	Maltose	2	n.d.
L-Arabinose	_	n.d.	Sucrose	6 ^c	33 ^c
D-Lyxose	23 ^c	n.d.	Lactose	n.d.	9 ^d
L-Lyxose	_	n.d.	Cellobiose	2	n.d.
L-Fucose	14 ^d	n.d.	Turanose	7 ^c	30 ^c
L-Rhamnose	n.d.	n.d.	Lactulose	3	35 ^d
D-Glucose	10 ^b	50 ^b	Isomaltose	_	n.d.
D-Galactose	9	16 ^c	Trehalose	_	n.d.
D-Mannose	25 ^b	66 ^b	Palatinose	_	n.d.
D-Talose	22^{d}	43 ^d	Melibiose	_	n.d.
D-Fructose	8	16	Sorphorose	_	n.d.
Xylitol ^a	36	_	Gentibiose	_	n.d.
Sorbitol ^a	20	-	_		

Table 2. Maximum transxylosylation yields with different acceptors

^a sugar alcohol; ^b known compound (structure-confirmed, see Supplementary Table S1); ^c novel compound (structure-determined, see Table 3 and Fig. 5); ^d novel compound (not structure-determined, due to complex reaction mixture). Products were quantified by HPAEC-PAD using xylobiose and xylotriose as standards for di- and trisaccharides, respectively. The total yield was estimated including all products and calculated as a percentage of products/donor concentration; n.d., not detected; –, product was not tested. Products in bold were purified for structural analysis and showed no difference in structure from BxlA and BxlB.

In total BxlA and BxlB formed 14 and 11 βxylosyl oligosaccharides from 18 and 25 acceptor candidates, respectively. The new oligosaccharides were prepared at the maximum production conditions (Fig. 4) for structure determination by NMR. Five such structure-determined oligosaccharides, which were obtained with Dlyxose, D-galactose, sucrose (two different products) and turanose as acceptors, were not reported before. BxlA and BxlB displayed different tranglycosylation capacity with respect to both yield and preferred acceptors. In comparison, a homologous GH3 βxylosidase from A. $awamori^{20,21}$ (having 65%) sequence identity with BxlA and 46% with BxlB) showed different acceptor preference and transglycosylation yield, except for that all three enzymes transxylosylate D-mannose with highest yield. The variations among related enzymes emphasise the importance of screening many

different acceptor candidates to evaluate the transxylosylation capacity.

2.4. Structure determination of novel β-transxylosylation oligosaccharide products

Generally BxIA and BxIB formed more than one tranxylosylation product with a given acceptor enzymatic as reported also for other transxylosylations.^{20,21,36,37} This product heterogeneity, as observed on the NMR chemical shifts (data not shown), reflects variation in regioselectivity due to several acceptor binding modes and possibly reaction with different forms (e.g. α , β , pyranosyl and furanosyl forms) of an Such product mixtures complicate acceptor. analysis and hamper NMR based structural assignments, thus precluding identification of some of the obtained products.



Figure 4. Progress of di- and trisaccharide formation by transxylosylation with *p*NPX as donor and different acceptors as catalysed by BxlA (A–C) and BxlB (D–F). A) D-Xylose (\blacktriangle), D-arabinose, (\triangle), D-lyxose (\diamondsuit), xylitol (\bullet) and sorbitol (\circ). B) D-Glucose (\blacksquare), D-galactose (\square), D-mannose (\blacktriangledown), D-talose (∇), D-fructose (\blacklozenge) and L-fucose (\diamondsuit). C) Maltose (\blacksquare), sucrose (\blacklozenge), cellobiose (\square), turanose (\bullet) and lactulose (\circ). D) D-Xylose (\blacktriangle). E) D-Glucose (\blacksquare), D-galactose (\square), D-mannose (\heartsuit). D) D-Xylose (\bigstar). E) D-Glucose (\blacksquare), D-galactose (\square), D-mannose (\heartsuit). D) D-Xylose (\bigstar). E) D-Glucose (\blacksquare), D-galactose (\square), D-mannose (\heartsuit), D-talose (\heartsuit), normanose (\heartsuit), D-talose (\heartsuit) and lactulose (\circ). B) D-Clucose (\blacksquare), D-galactose (\square), D-mannose (\heartsuit), D-talose (\heartsuit) and D-fructose (\diamondsuit). F) Xylobiose (\blacktriangle), sucrose (\diamondsuit), lactose (\diamondsuit), turanose (\bullet) and lactulose (\circ). Product concentrations were calculated from peak areas of HPAEC-PAD chromatograms using xylobiose and xylotriose as standards for di- and trisaccharides, respectively (see 3.6 for details).

Oligosaccharides were purified by HPLC from BxlA- (9) and BxlB- (11) reaction mixtures (Table 2) and subjected to structure determination by 2D NMR and ESI-MS. The molecular mass of each product was in agreement with the calculated molecular mass of the Li⁺ adduct of the corresponding acceptor conjugate of D-xylose (Table 3) and also supported the presence of product mixtures. Linkage analysis by NMR was based on chemical shifts of the products using 2D NMR (Table 3 and Supplementary Table S1) and showed that both BxlA and BxlB produced mainly β-1,4 linked products with D-xylose, D-lyxose and Darabinose as acceptors as identified by long range proton-carbon bond correlations between the nonreducing anomeric proton and C-4 of the substituted position and confirmed by inter-residue nuclear Overhauser effect (NOE) correlations. This

mainly β -1,4 corresponds to regioselective formation of pNP β -xylo-oligosaccharides and β xylo-oligosaccharides reported for GH3 βxylosidases from A. awamori,¹⁹ A. niger,³⁶ H. *iecorina*¹⁸ and *Penicillium wortmanni*,³⁷ although small amounts of products with different linkages were also found. In contrast transxylosylation of Dmannose, D-glucose and D-galactose occurred with predominant β -1,6 regioselectivity, as identified by the downfield shift of the C6 or by inter NOE correlation across the glycosidic bond. Obviously these monosaccharides are accommodated at subsite +1 of the active site in a different way possibly due specific protein-carbohydrate interactions to rendering the primary alcohol group (6-OH) best positioned to perform a nucleophilic attack on the covalent xylosyl-enzyme intermediate. Preference

for the primary alcohol group in hexose-containing acceptors was also reported for β -xylosidase from A. awamori.^{20,21} In case of disaccharide acceptors, β-1.4 or β -1.6 xylosyl linkage was formed with either of the disaccharide moieties reflecting variation in binding mode indicative of rather loose acceptor recognition at subsite +1 in BxlA and BxlB. five novel oligosaccharides were Noticeably generated with D-lyxose, D-galactose, turanose and sucrose (two products) as acceptors: β-Dxylopyranosyl- $(1\rightarrow 4)$ -D-lyxopyranoside; β-Dxylopyranosyl- $(1\rightarrow 6)$ -D-galactopyranoside; β-Dxylopyranosyl- $(1\rightarrow 4)$ - α -D-glucopyranosyl- $(1\rightarrow 3)$ β-D-fructofuranoside; β-D-xylopyranosyl- $(1 \rightarrow 4)$ -α-D-glucopyranosyl- $(1\rightarrow 2)$ - β -D-fructofuranoside; and β -D-xylopyranosyl-(1 \rightarrow 6)- β -D-fructofuranosyl- $(2\rightarrow 1)$ - α -D-glucopyranoside as identified by 2D NMR (Table 3; Fig. 5). In contrast, structures of the transxylosylation products obtained with the other novel acceptors L-fucose, D-talose, lactose and lactulose were not solved due to the formation of product mixtures, which could not be separated and purified in sufficient amounts enabling an NMR analysis. ESI-MS, however, identified that these

analysis. ESI-MS, however, identified that these products, as expected, were Li-adducts of di- and trisaccharides with molecular masses of m/z 303, 319, 481 and 481, respectively.



Figure 5 Structures determined by 2D-NMR of five novel β -xylosyl-oligosaccharides produced by transxylosylation using BxlA or BxlB. A) β -D-Xylp-(1 \rightarrow 4)-D-Lyxp; B) β -D-Xylp-(1 \rightarrow 6)-D-Galp; C) β -D-Xylp-(1 \rightarrow 4)- α -D-Glcp-(1 \rightarrow 3)- α -D-Fruf; D) β -D-Xylp-(1 \rightarrow 4)- α -D-Glcp-(1 \rightarrow 2)- β -D-Fruf; E) β -D-Xylp-(1 \rightarrow 6)- β -D-Fruf-(2 \rightarrow 1)- α -D-Glcp.

2.5. Conclusion

GH3 β -xylosidases BxlA and BxlB from A. nidulans FGSC A4 showed remarkably different hydrolytic and transglycosylation activities. Whereas BxlA had about 6 fold higher catalytic efficiency than BxlB in hydrolysis of pNPX and β -1,4-xylo-oligosaccharides, BxlB was more efficient in transxylosylation, the highest yield being 66% obtained with mannose. BxlA and BxlB showed broad acceptor specificity, however the results cannot be interpreted on the basis of a 3D structural model, since no GH3 β-xylosidase structure has been solved and the other solved GH3 structures are too taxonomically distant to allow reliable modelling. All together BxlA or BxlB were found to form β-xylosyl-oligosaccharides with 18 different acceptors and five not previously described oligosaccharides were identified. Although some of the transxylosylation reactions resulted in several reflecting multiple products acceptor accommodation modes, some of these products are also novel and together with the five novel β -xylosyl oligosaccharides represent new candidates for evaluation as prebiotics.

3. Experimental

3.1. Materials

pNP, pNPX, pNPA, D-arabinose, L-arabinose, D-lyxose, L-lyxose, L-fucose, L-rhamnose, Dgalactose, D-mannose, D-talose, D-fructose, xylitol, sorbitol, maltose, sucrose, lactose, cellobiose, turanose, lactulose, isomaltose, trehalose, palatinose, melibiose, sophorose. and gentibiose were purchased from Sigma Aldrich (St. Louis, MO). D-Xylose was from Carl Roth (Karlsruhe, Germany). D-Glucose was from Merck (Darmstadt, Germany). Xylobiose, xylotriose, xylotetraose, xylopentaose, and xylohexaose were purchased from Megazyme (Wicklow, Ireland).

Compound	Chemical shifts (δ, p.p.m.)					
	H-1	Н-2	H-3	H-4	H-5a/H-5b	H-6a/H-6b
(molecular mass) ^a	C-1	C-2	C-3	C-4	C-5	C-6
β-D-Xyl <i>p</i> -(1,4)-	4.47	3.26	3.42	3.60	3.30/3.94	
	103.6	73.7	76.6	70.2	66.1 ^b	
α-D-Lyx <i>p</i>	4.95	3.74	4.03	3.91	3.80/3.88	
	94.9	70.8	~70.2	77.0	62.4	
β-D-Lyx <i>p</i>	4.88	3.92	n.d.	n.d.	n.d.	
(<i>m</i> / <i>z</i> 289)	95.2	n.d.	n.d.	n.d.	n.d.	
β-D-Xyl <i>p-</i> (1,6)-	4.42	3.25	3.40	3.58	3.28/3.92	
	104.4	74.1	76.6	n.d.	66.1 ^b	
α-D-Galp	5.22	3.76	3.81	3.95	n.d.	n.d.
	93.3	n.d.	n.d.	n.d.	n.d.	n.d.
β-D-Galp	4.54	3.44	3.60	3.88	3.85	3.78/3.95
(<i>m</i> / <i>z</i> 319)	97.4	n.d.	n.d.	n.d.	n.d.	70.3
β-D-Xyl <i>p</i> -(1,4)-	4.43	3.33	3.42	3.63	3.93/3.26	
	102.1	73.6	70.2	70.1	66.1 ^b	
α-D-Glc <i>p</i> -(1,3)-	5.31	3.56	3.72	3.40	3.71	3.83/3.77
2 • • • •	101.5	73.0	73.7	76.5	73.7	61.4
β-D-Fru <i>f</i>	3.48/3.73	3 —	3.97	4.16	4.12	3.84/3.97
(<i>m</i> / <i>z</i> 481)	64.7	n.d.	77.0	70.2	70.2	61.8
β-D-Xyl <i>p</i> -(1,4)-	4.41	3.27	3.41	3.61	3.96/3.28	
	104.2	74.0	76.6	70.1	66.1 ^b	
α-D-Glc <i>p</i> -(1,2)-	5.37	3.55	3.81	3.62	3.93	3.82/3.88
/	92.9	71.9	71.9	78.8	72.1	60.3
β-D-Fru <i>f</i>	3.66/3.59) _	4.18	3.99	3.85	3.75/3.80
$(m/z \ 289)$	62.1	104.6	77.3	74.9	82.2	63.4
β-D-Xyl <i>p</i> -(1,6)-	4.47	3.27	3.42	3.59	3.94/3.29	
	104.3	74.0	76.5	70.1	66.1 ^b	
β-D-Fru <i>f</i> -(2,1)-	n.d.	_	4.18	4.03	4.02	3.95
	n.d.	104.9	77.0	75.4	81.0	n.d.
α-D-Glcp	5.37	3.52	3.73	3.44	3.81	3.78/3.82
(m/z 481)	93.2	72.0	n.d.	n.d.	n.d.	n.d.

Table 3. ¹H and ¹³C NMR spectra assignment and ESI-MS m/z values of novel β -xylosyloigosaccharides produced by transxylosylation with *p*NPX as donor and suitable acceptors.

^a Molecular mass as Li^+ adduct (estimated mass + m/z 7) determined by ESI-MS shown in parenthesis; ^b Reference carbon shifts according to C5 ; n.d., not detected

3.2. Cloning of *bxlA* and *bxlB* encoding GH3 β -xylosidases

P. pastoris X-33 (FGSC database accession no. 10077 and 10125; www.fgsc.net) harbouring A. nidulans FGSC A4 full-length bxlA (GenBank ID: EAA64470.1) (GenBank and bxlB ID: EAA67023.1)³⁰ were purchased from Fungal Genetics Stock Center, School of Biological Sciences, University of Missouri, (Kansas City, MO). The two genes were recloned to encode mature polypeptides lacking Met1-Ser18 (BxlA) and Met1-Ser23 (BxlB) (http://www.cbs.dtu.dk/ services/SignalP/).³⁸ Expand High Fidelity DNA polymerase (Roche Diagnostics GmbH, Mannheim, Germany) was used for DNA amplification with oligonucleotide primers (Supplementary Table S1) based on the genomic sequence.³⁹ A silent mutation $(1374T \rightarrow C)$ in the *bxlB* EcoRI site was introduced by overlap PCR⁴⁰ using appropriate primers (Supplementary Table S2). PCR products were digested for *bxlA* by XhoI and XbaI and for *bxlB* by EcoRI and NotI (New England BioLabs, Ipswich, MA) and cloned in frame with the α -mating factor secretion signal in pPICZaA (Invitrogen, Carlsbad, The plasmids were transformed into CA). Escherichia coli DH5a (Invitrogen) and selected on low salt Luria Bertani medium supplemented with 25 µg/mL Zeocin. Fully sequenced plasmids (Eurofins MWG Operon, Ebersberg, Germany) were linearised by PmeI (New England BioLabs), transformed into P. pastoris X-33 (Invitrogen) by electroporation (Micropulser; Bio-Rad Laboratories Inc, Hercules, CA), and selected at 30°C for 3 days on yeast peptone dextrose plates (1% yeast extract, 2% peptone, 2% dextrose) containing 1 M sorbitol and 100 µg/mL Zeocin.

3.3. Production and purification of BxlA and BxlB

P. pastoris transformants were grown in 3 L (BxlA) or 4 L (BxlB) buffered glycerol-complex medium (1% yeast extract, 2% peptone, 100 mM potassium phosphate pH 6.0, 1.34% yeast nitrogen

base with ammonium sulfate, 4×10^{-5} % biotin, and 1% glycerol) at 30°C for 24 h to OD_{600} of 25 (BxlA) and 52 (BxlB). Cells were harvested (3000g, 5 min, 4°C) and resuspended for induction in 1 L buffered minimal methanol medium (BxlA; 100 mM potassium phosphate pH 6.0, 1.34% yeast nitrogen base with ammonium sulfate, 4×10^{-5} % biotin, 1% methanol, 1% casamino acid) or in 1 L of buffered methanol-complex medium (BxlB; 0.5% methanol, 1% yeast extract, 2% peptone, 100 mM potassium phosphate pH 6.0, 1.34% yeast nitrogen base with ammonium sulfate, 4×10^{-5} % biotin) at 22°C for 72 (BxlA) and 120 h (BxlB). Methanol was supplemented every 24 h to a final concentration (v/v) of 1% (BxlA) and 0.5% (BxlB), until harvest (13,500g, 4°C, 1 h).

The BxlA culture supernatant was concentrated with buffer-exchange to 10 mM MES pH 6.0 (Pellicon tangential flow filtration systems, nominal molecular weight limit (NMWL) 10 kDa; Millipore, Billerica, MA) and applied to a 6 mL Resource Q column (GE Healthcare, Uppsala, Sweden) equilibrated in the same buffer. After wash with 37.5 mM NaCl in the same buffer, protein was eluted by a linear 37.5-325 mM NaCl gradient (flow rate: 2 mL/min). Fractions containing BxlA were pooled, concentrated (Amicon Ultra-15, NMWL 30 kDa, Millipore), and applied to a Hiload 26/60 Superdex 200 column (GE Healthcare) equilibrated with 20 mM MES, 0.25 M NaCl pH 6.0 (flow rate: 1 mL/min). BxlA-containing fractions were pooled, concentrated, and buffer-exchanged to 10 mM MES pH 6.0. The BxlB culture supernatant was adjusted to pH 7.5 by 1 M K₂HPO₄, filtered (0.22 µm; TPP, Trasadingen, Switzerland) and applied to 5 mL HisTrap HP column (GE Healthcare) equilibrated with 20 mM HEPES, 0.5 M NaCl, 10 mM imidazole pH 7.5. Following wash with the above buffer containing 22 mM imidazole, BxlB was eluted by a linear 22-400 mM imidazole gradient (flow rate: 1.0 mL/min). BxlB-containing fractions were pooled, concentrated (Centriprep YM30, Millipore), and applied to a Hiload 26/60 Superdex 200 column (GE Healthcare) equilibrated with 20 mM MES, 0.15 M NaCl pH 6.8 (flow rate: 0.5 mL/min). Fractions containing BxlB were

pooled, concentrated, and buffer exchanged to 20 mM HEPES pH 7.0. All purification steps were performed using an ÄKTAexplorer chromatograph (GE Healthcare) at 4°C.

3.4. Properties of BxlA and BxlB

Protein concentration was determined spectrophotometrically at 280 nm using $E^{0.1\%} = 1.79$ and 0.97 for BxlA and BxlB, respectively, determined by aid of amino acid analysis. Molecular masses were estimated by SDS-PAGE (NuPAGE Novex Bis-Tris gels; Invitrogen) and gel filtration (precalibrated Hiload 16/60 Superdex 200 column; flow rate: 1 mL/min; Gel Filtration Calibration Kits; GE Healthcare). Theoretical molecular masses were calculated using ExPASy-ProtParam tool (http://www.expasy.ch/tools/protparam.html).⁴¹ Nglycosylation was predicted using NetNGlyc 1.0 (http://www.cbs.dtu.dk/services/NetN Server Glvc/).⁴² England Endoglycosidase H (New BioLabs) deglycosylation of BxlA (7 µg) and BxlB performed according was to the (5 μg) manufacturer's protocol. Carbohydrate contents were determined on 400 µg enzyme using the phenol-sulfuric acid method⁴³ with either glucose or mannose as standard. Isoelectric focusing of 600 ng BxlA and 400 ng BxlB was carried out (Pharmacia PhastSystem; GE Healthcare) at pH 4.0-6.5 (4 cm polyacrylamide PhastGel IEF) using the low range pI (pH 2.5-6.5) calibration kit and PlusOne Silver Stain Kit, Protein (GE Healthcare).

3.5. Enzyme activity assays of BxIA and BxIB

*p*NPX (2 mM) was hydrolysed at 40°C for 10 min by BxlA (20–150 nM; 50 mM sodium acetate pH 5.0) or BxlB (25–250 nM; 50 mM sodium acetate pH 5.2) in 50 μ L reaction mixtures. The reaction was stopped by 1 M Na₂CO₃ (100 μ L) and *p*NP was quantified spectrophotometrically at 410 nm using 0.05–0.5 mM *p*NP as standard. One unit was defined as the amount of enzyme liberating 1 μ mol of *p*NP per min under the assay conditions. pH activity optima of BxlA (71 nM) and BxlB (179 nM) were determined for the pH range 2.0–10.0 towards *p*NPX (2 mM) in 40 mM Britton-Robinson buffer.⁴⁴ Activity was determined in the temperature range of 25–80°C using the above BxlA and BxlB concentrations in 50 mM sodium acetate pH 5.0 and 5.2, respectively. The stability at 40°C in the range pH 2.0–10.0 (40 mM Britton-Robinson buffers) and at 25–80°C was deduced from residual activity towards *p*NPX of 355 nM BxlA and 895 nM BxlB after 5 h incubation.

Initial rates of hydrolysis at 8 concentrations of *p*NPX (0.1–8.0 mM), *p*NPA (0.2–2 mM), and βxylo-oligosaccharides (X2-X6, 0.4-6.0 mM) were measured using 40 nM (for pNPX), 465 nM (for pNPA), 32 nM (for X2-X6) BxlA, and at 11 different concentrations of pNPX (0.1-8.0 mM), pNPA (0.1-8.0 mM), X2-X6 (1.0-8.0 mM) using 66 nM (for pNPX and X2-X6) and 328 nM (for pNPA) BxlB. Hydrolysis of X2-X6 was assayed in 50 mM sodium acetate pH 5.0 for BxlA and pH 5.2 for BxlB (100 µL reaction mixture) and stopped by addition of 2% p-bromoaniline and 4% thiourea in glacial acetic acid (500 µL) and incubated in the dark at 70°C for 10 min, followed by 60 min at 37°C and quantified from the absorbance at 520 nm using 0.25-5.0 mM xylose as standard.^{45,46} Kinetic parameters, $K_{\rm m}$ and $k_{\rm cat}$, were determined by fitting the Michaelis-Menten equation to initial rates using SigmaPlot v. 9.0.1 (Systat Software Inc. San Jose, CA). Due to poor affinity ($K_{\rm m} > 8$ mM), $k_{\rm cat}/K_{\rm m}$ of pNPA was obtained from the slope of the plot of initial rate (v/[E]) vs substrate concentration.

3.6. Transxylosylation by BxIA and BxIB

Transxylosylation with 1) 200 nM BxlA, 30 mM pNPX and 200 mM acceptor (D-xylose, D-arabinose, D-lyxose, L-fucose, L-rhamnose, D-glucose, D-galactose, D-mannose, D-talose, D-fructose, xylitol, sorbitol, maltose, sucrose, lactose, cellobiose, turanose, or lactulose) in 50 mM sodium acetate pH 5.0 (1 mL) or 2) 1.7 μ M BxlB, 25 mM pNPX and 400 mM acceptor as above except for

sugar alcohols and including also (L-arabinose, Llyxose, xylobiose, isomaltose, trehalose, palatinose, melibiose, sophorose, or gentibiose) in 40 mM sodium acetate pH 5.2 (1 mL) occurred at 40°C during 5 h. Aliquots (5 µL for BxlA; 10 µL for BxlB) were removed at time intervals and added to 100 mM NaOH (195 µL for BxlA) and 500 mM NaOH (90 µL for BxlB). The BxlB samples were diluted to a suitable concentration level before analysis. Products were quantified based on peak areas in HPAEC-PAD (ICS-3000 IC system; Dionex Corporation, Sunnyvale, CA) equipped with CarboPac PA200 anion exchange column (3×250 mm and 3×50 mm guard column; Dionex) using a linear 0-75 mM sodium acetate gradient in 100 mM NaOH (35 min; 25°C; flow rate: 0.35 mL/min) calibrated with xylobiose and xylotriose for di- and trisaccharide products, respectively. Transxylosylation yields were calculated based on the *p*NPX concentration.

3.7. Xylosyl-oligosaccharide purification

For structural analysis of tranxylosylation products reactions in 50 mM sodium acetate pH 5.0 (1 mL) were performed as described in 3.6 with 1) BxlA and D-xylose, D-arabinose, D-lyxose, L-fucose, D-glucose, D-mannose, D-talose, sucrose, or turanose and 2) BxlB and D-xylose, D-glucose, D-D-mannose, D-talose, D-fructose, galactose, xylobiose, sucrose, lactose, turanose, or lactulose at 40°C for optimal reaction times (Fig. 4). Reactions were terminated by heat inactivation (95°C, 10 min) followed by centrifugation (18,000g, 4°C, 5 min), desalting (Amberlite MB-20; Fluka, Sigma Aldrich), and filtration (0.45 µm nylon filter; Frienette Aps, Kenbel, Denmark). Oligosaccharides were purified by HPLC (UltiMate 3000 Standard LC system; Dionex) equipped with a TSKgel Amide-80 column (5 μ m, 4.6 \times 250 mm and 4.6 \times 10 mm guard column; TOSOH, Tokyo, Japan) and refractive index detector (RI-101; Showa Denko, Kanagawa, Japan) at 1 mL/min (for mobile phase composition and column temperature see Supplementary Table S3). The purity was assessed by thin laver chromatography (TLC; aluminium-coated silica gel 60 F_{254} , Merck) developed by acetonitrile/water (80:20, v/v), sprayed with orcinol/sulphuric acid/water/ethanol, (2:10:10:80, w/v/v/v) or with α -naphthol/sulphuric acid/methanol (0.03:15:85, w/v/v) and tarred at 120°C.

3.8. Nuclear Magnetic Resonance spectroscopy

Purified oligosaccharide products (500 μ L,1-2 mM) were vacuum dried (from BxlA; Savant SVC100 SpeedVac Concentrator; Savant Instruments, Inc. Farmingdale, NY) or freeze dried (from BxlB; CoolSafe -55°C; ScanLaf A/S, Lynge, Denmark) and dissolved in deuterium oxide. Structures were determined by ¹H- and ¹³C-NMR analysis. The majority of the spectra were recorded on a Varian Unity Inova 500 MHz NMR spectrometer and a few on a Bruker Avance 800 MHz spectrometer. A series of 2D homo- and heteronuclear correlated DQF-COSY, NOESY, gHSQC, and gHMBC spectra were obtained using standard pulse sequences. The proton chemical shifts were referenced to the solvent $\delta_{\rm H}$ 4.78 ppm and the carbon shifts were referenced according to C5 of xylose at 66.1 ppm.

3.9. Electrospray ionization mass spectrometry

ESI-MS was performed (LTQ XL ion trap MS; Thermo Scientific, San Jose, CA) on vacuum dried or lyophilised samples (see 3.8). Samples were dissolved in milliQ water (100 µg/mL) and applied to a Thermo Accela UHPLC system equipped with a Hypercarb (100 \times 2.1 mm, 3 µm) column (Thermo Scientific). Elution was performed with a gradient water/acetonitrile of deionized and 0.2% trifluoroacetic acid (0.4 mL/min; 70°C). MS detection was performed in the positive mode using a spray voltage of 4.5 kV and a capillary temperature of 260°C and auto-tuned on glucohexaose.47

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Supplementary Documents



Supplementary Figure S1. Isoelectric focusing (A) and endoglycosidase H treatment monitored by SDS-PAGE (B) of purified BxlA and BxlB. A) Lane 1 and 4, low range pI (pH 2.5–6.5) calibration standard (GE Healthcare); lane 2, 600 ng BxlA; lane 3, 400 ng BxlB. B) Lane 1 and 6, Mark12 unstained protein standard marker (Invitrogen). Lane 2, 7 µg BxlA; lane 3, 7 µg deglycosylated BxlA; lane 4, 5 µg BxlB; lane 5, 5 µg deglycosylated BxlB. Endoglycosidase H is seen as the 29 kDa band in lanes 3 and 5.

Supplementary Table S1 ¹H and ¹³C NMR spectra assignment and ESI-MS m/z values of known β -xylosyl-oligosaccharides produced by transxylosylation with *p*NPX as donor and suitable acceptors.

Compound			Chemica	al shifts ((ð, p.p.m.)	
	H-1	H-2	H-3	H-4	H-5a/H-5b	H-6a/H-6b
(molecular mass) ^a	C-1	C-2	C-3	C-4	C-5	C-6
β-D-Xyl <i>p</i> -(1,4)-	4.43	3.23	3.40	3.60	3.95/3.28	
	102.7	73.6	76.4	70.1	66.1 ^b	
α-D-Xylp	5.16	3.52	n.d.	n.d.	3.80/3.73	
	92.9	72.2	72.0	74.3	59.7	
β-D-Xylp	4.56	3.22	3.52	3.76	3.35/4.04	
(<i>m</i> / <i>z</i> 289)	97.3	~74.9	~74.9	77.3	63.8	
β-D-Xyl <i>p</i> -(1,4)-	4.43	3.32	3.42	3.62	3.26/3.93	
	102.3	73.8	76.6	70.2	66.1 ^b	
α-D-Arap	5.22	3.79	3.91	4.10	3.78/3.96	
	93.4	n.d.	n.d.	77.2	61.0	
β-D-Arap	4.53	3.48	3.68	4.05	3.60/4.00	
(<i>m</i> / <i>z</i> 289)	97.3	73.0	72.5	76.8	64.6	
β-D-Xyl <i>p</i> -(1,6)-	4.43	3.28	3.42	3.60	3.31/3.94	
	104.4	73.9	76.4	70.2	66.1 ^b	
α-D-Manp	5.14	3.91	3.81	3.69	3.93	3.85/4.09
	95.0	~72.1	~71	~67.6	~72.2	~69.9
β-D-Man <i>p</i>	4.87	3.92	3.63	3.59	3.49	3.82/4.13
(<i>m</i> / <i>z</i> 319)	94.6	71.8	~73.8	70.1	~76	~69.9
β-D-Xyl <i>p</i> -(1,6)-	4.42	3.27	3.43	3.60	3.29/3.93	
	104.2	73.9	76.6	70.0	66.1 ^b	
α-D-Glcp	5.19	3.50	3.68	3.4	3.93	3.84/4.10
	92.9	72.3	73.3	70.5	71.1	~69.8
β-D-Glcp	4.61	3.22	3.45	3.94	3.58	3.80/4.12
(<i>m</i> / <i>z</i> 319)	96.9	75.0	76.5	71.3	76.0	~69.9

^a Molecular mass as Li⁺ adduct (estimated mass + m/z 7) determined by ESI-MS shown in parenthesis; ^b Reference carbon shifts according to C5 ; n.d., not detected

Primer	Primer sequence	Restriction site
<u>bxlA</u>		
Sense	ATT <u>CTCGAG</u> AAAAGAGAGGCTGAAGCTGCGAACACCAGCTACAC ^a	XhoI
Antisense	$AC\underline{TCTAGA}TTAATGATGATGATGATGATGAGCAATAACATCCTGC^a$	XbaI
<u>bxlB</u>		
Sense	GGG <u>GAATTC</u> AACTACCCGGACTGCACAACGGGCCCTC ^a	EcoRI
Antisense	CCCG <u>GCGGCCGC</u> ATCACTGTCGTTACCTGACAACGG ^a	NotI
1374T→C-sense	GGACAGAGGT <u>GAACTC</u> GACAAGCACAGAC ^b	
$1374T \rightarrow C$ -antisense	GTCTGTGCTTGTC <u>GAGTTC</u> ACCTCTGTCC ^b	

Supplementary Table S2 Primers for construction of expression plasmids encoding *bxlA* and *bxlB*

^a Restriction sites are underlined

^b Silent mutation position is double underlined

	BxlA	4	BxlB		
Acceptor	Mobile phase	Column temp.	Mobile phase	Column temp.	
	(ACN ^a /water, v/v)	(°C)	(ACN ^a /water, v/v)	(°C)	
Monosaccharides					
D-Xylose	85/15	70	80/20	70	
D-Lyxose	85/15	70	_	_	
D-Arabinose	80/20	70	_	_	
L-Fucose	80/20	70	_	_	
D-Glucose	80/20	70	80/20	65	
D-Galactose	_	—	80/20	50	
D-Mannose	85/15	70	80/20	70	
D-Talose	85/15	70	80/20	55	
D-Fructose	_	—	80/20	55	
Disaccharides					
Xylobiose	_	—	80/20	65	
Turanose	85/15	70	80/20	65	
Lactose	_	_	85/15	70	
Lactulose	_	—	85/15	70	
Sucrose	85/15	70	85/15	70	

Supplementary Table S3 HPLC purification conditions of β -xylosyl-oligosaccharides

^a ACN: Acetonitrile

-, not tested

Manuscript II

GH5 endo-β-1,4-mannanases from *Aspergillus nidulans* FGSC A4 producing

hetero manno-oligosaccharides by transglycosylation

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Abstract

Two Aspergillus nidulans FGSC A4 endo-β-1,4-mannanases, ManA and ManC, of glycoside hydrolase family 5 were secretory expressed by Pichia pastoris in high yields. 120 and 145 mg/L were purified from culture supernatant, respectively, by His-tag affinity and gel filtration chromatography. Both enzymes hydrolysed β -1,4 linked manno-oligosaccharides with increase in catalytic efficiency (k_{cat}/K_{M}) from degree of polymerisation (DP) 4 through 6. Both enzymes hydrolysed konjac glucomannan, guar gum and locust bean gum galactomannans. ManC showed two fold higher k_{cat}/K_{M} towards mannooligosaccharides of DP 5 and 6, and 30-80% higher activity towards the mannans than ManA. By transglycosylation ManC formed β-1,4 manno-oligosaccharides in eight fold higher yield than ManA and used mannotriose, isomaltotriose and melezitose as acceptors, whereas ManA only used mannotriose. Two novel penta- and two novel hexasaccharides were obtained with isomaltotriose and melezitose acceptors. ManA and ManC share 39% sequence identity and homology modelling indicated very similar substrate interactions at subsites +1 and +2 except that Trp283 in ManC (subsite +1) corresponds to Ser289 in ManA. The ManAS289W mutant decreased 30-45% K_M towards manno-oligosaccharides of DP 5 and 6 and increased transglycosylation yield by 50% as compared with wild-type ManA. ManCW283S in contrast increased 40% $K_{\rm M}$ for manno-oligosaccharides and decreased transglycosylation yields by 30–45%. This mutation also lowered the higher activity found for ManC towards different mannans to the level found for ManA. This first mutational analysis at subsite +1 of GH5 endo-B-1,4-mannanases identifies Trp283 as a residue that discriminates between mannans with different degree of branching and has impact on transglycosylation and binding affinity.

Keywords: Endo- β -1,4-mannanase, Glycoside hydrolase family 5, Transglycosylation, Mannooligosaccharide action pattern, Mannans, Subsite +1 mutagenesis

Abbreviations: DP, degree of polymerisation; GH, glycoside hydrolase family; *Hj*Man5A, endo- β -1,4mannanase from *Hycocrea jecorina* RUTC-30 (GenBank AAA34208.1); HPAEC-PAD, high performance anion exchange chromatography with pulsed amperometric detection; HPLC, high performance liquid chromatography; IMAC, immobilized metal ion affinity chromatography; M1, mannose; M2, mannobiose; M3, mannotriose; M4, mannotetraose; M5, mannopentaose; M6, mannohexaose; M7, mannoheptaose; MALDI-TOF MS, matrix-assisted laser desorption/ionisation time-of-flight mass spectrometry; ManA and ManC, GH5 endo- β -1,4-mannanase from *A. nidulans* FGSC A4 (GenBank EAA63326.1 and EAA58449.1, respectively); *Tf*Man, GH5 endo- β -1,4-mannanase from *Thermobifida fusca* KW3 (GenBank CAA06924.1).

1. Introduction

Mannans are found as structural constituents in plant cell walls being the dominant hemicelluloses in softwood and are also major storage polysaccharides [1]. Based on the backbone composition and galactose side chain substitution mannans are classified into four groups; unsubstituted mannans; galactomannans; glucomannans; and galactoglucomannans, [1,2]. Unsubstituted mannans and galactomannans (with α -1,6 D-galactosyl substituents) have 1,4 linked β -Dmannopyranosyl backbones and are mainly storage seed endosperm. polysaccharides in legume Glucomannans and galactoglucomannans contain distributed 1.4 randomly linked β-Dmannopyranosyl and β-D-glucopyranosyl residues and are the major hemicelluloses in secondary cell walls of softwood. Galactose substitution occurs mainly at the O6 position while O-acetylation is found at the O2 or O3 position of main chain mannosyl residues [1-4]. Mannans are used in the food industry as gelling agents and thickeners and are also reported to function as dietary fibres [1]. βmanno-oligosaccharides are mostly produced by steam treatment at high pressure from natural sources such as coffee beans [5]. They are used in food and feed industries as prebiotics promoting gastrointestinal health as well as increasing the amount of excreted fat and reducing blood pressure [5-7].

Depending on the mannan type a consortium of endo-β-1,4-mannanases (EC 3.2.1.78), exo-β-1,4mannosidases (EC 3.2.1.25), α-galactosidases (EC 3.2.1.22), β -glucosidases (EC 3.2.1.21) and acetyl mannan esterases (EC 3.1.1.6) is needed for complete depolymerisation [1,2,8]. Hydrolysis of backbone linkages by endo-β-1,4-mannanases is influenced by the extent and pattern of galactose substitution and the distribution of glucose in the polysaccharide main chain [1,2,8]. Endo-β-1,4mannanases are classified into three glycoside hydrolase families (GH5, 26 and 113) based on amino acid similarities sequence (http://www.cazy.org) [9]. GH5 contains bacterial

and eukaryotic endo- β -1,4-mannanases, whereas GH26 and GH113 endo-β-1,4-mannanases are mainly from prokaryotes [10]. Endo- β -1,4mannanases hydrolyse mannans to mainly mannobiose and mannotriose [11-14]. The active sites in GH5 endo- β -1,4-mannanases are reported to possess at least five subsites able to accommodate substrate backbone carbohydrate residues [15–18]. Hydrolysis seems to require binding involving at least four subsites (mostly -2 through +2); hence degradation of mannotriose is extremely slow [2,10,13,19].

The three endo- β -1,4-mannanase-containing GHs belong to clan GH-A and sharing a common fold and the general acid/base catalysis proceeding via a double displacement mechanism resulting in retention of the anomeric configuration [9]. The acid/base and nucleophile catalysts are two glutamic acid residues [1,20]. First the catalytic nucleophile attacks the anomeric carbon center and forms a covalent glycosyl-enzyme intermediate aided by the acid catalyst protonation of the glycosidic bond to be cleaved. Subsequently water (in hydrolysis) or a carbohydrate hydroxyl group (in transglycosylation) is activated by the base catalyst and performs a nucleophilic attack at the anomeric centre of the covalent intermediate to release products with retention of the anomeric configuration of the substrate [21–22]. In transglycosylation, the substrate glycone that forms the glycosyl-enzyme intermediate and the incoming carbohydrate are referred to as donor and acceptor, respectively. Previously manno-oligosaccharides (DP 4-6) were used as donors for transglycosylation to produce longer manno-oligosaccharides catalysed by GH5 and GH113 endo-β-1,4-mannanases [19,20,23-25], whereas GH26 had poor transglycosylation activity [15,26,27]. Previously no heterogeneous transglycosylation, however, using non-mannooligosaccharide acceptors has been reported and insight moreover is lagging behind on how structural features of the active sites in different endo-β-1,4enzymatic mannanases influence properties, including the transglycosylation ability. Two 39% sequence identical GH5 mannan degrading enzymes,

ManA and ManC, were identified in the genome of the cell-wall degrading fungus Aspergillus nidulans FGSC A4 [28] and showed mannanolytic activity after heterologous expression in Pichia pastoris of the full-length enzymes including signal peptides [29]. In the present study, high amounts of ManA and ManC were obtained by secretion of P. pastoris after recloning without signal peptides. The substrate specificity as well as transglycosylation potential were characterised and homology modelling suggested that an important difference was Trp283 at subsite +1 of ManC, which corresponded to Ser289 in ManA of the otherwise very similar substrate binding subsites +1 and +2. This motivated site-directed mutagenesis to discern the contribution of this position to hydrolytic and transglycosylation activities, representing the first subsite +1 engineering in GH5 endo- β -1,4mannanases.

2. Material and methods

2.1 Materials

N-Acetyl-glucosamine, arabinose. Lfructose. galactose, arabinose. L-fucose, glucosamine, glucose, mannose (M1), L-rhamnose, talose, cellobiose, gentibiose, isomaltose, lactose, lactulose. laminaribiose, maltose, melibiose, sophorose, sucrose. trehalose. turanose, laminaritriose, maltotriose, melezitose, panose. raffinose and locust bean gum galactomannan were purchased from Sigma Aldrich (St. Louis, MO); xylose and guar gum galactomannan were from Carl Roth (Karlsruhe, Germany); arabinobiose, arabinotriose, mannobiose (M2), mannotriose (M3), mannotetraose (M4), mannopentaose (M5), mannohexaose (M6), xylobiose, xylotriose and were konjac glucomannan from Megazyme (Wicklow, Ireland); and isomaltotriose was from Tokyo Chemical Industry (Tokyo, Japan).

2.2 Sequence comparison and homology modelling

The BLAST tool was used for homology search of protein sequences (http://blast.ncbi.nlm.

nih.gov/Blast.cgi) [30]. Homology models of ManA and ManC were generated using the HHpred-Homology server (http://toolkit.tuebingen.mpg.de/ hhpred) [31] with the template endo- β -1,4mannanase from Hypocrea jecorina RUTC-30 complex with (*Hi*Man5A) in mannobiose accommodated at subsites +1 and +2 (PDB ID: 1QNR [17]), which has 56% and 37% sequence identity to ManA and ManC, respectively. The models validated homology were by the combinatorial extension method (http://cl.sdsc.edu /ce.html) [32] giving an RMSD of 0.3 Å for 343 equivalent C^{α} positions (of 372 in ManA and 381 in ManC) compared to 1QNR with 7.9 and 7.7 Z-score for ManA and ManC, respectively, whereas ProQ-Protein Quality Predictor (http://www.sbc.su.se/ ~bjornw/ProQ/ProQ.html) [33] gave LGScore/ MaxSub of 5.337/0.627 and 5.468/0.567 for ManA and ManC, respectively, indicating good to very good quality of both models. Models were rendered and analysed using Pymol v0.99 (DeLano Scientific LLC, San Carlos, CA) and superimposed with βmannanase from Thermobifida fusca KW3 (TfMan; PDB ID: 2MAN) [34] having 23% and 37% sequence identity to ManA and ManC, respectively, in complex with mannobiose bound at subsites -3through -2, to illustrate possible carbohydrateprotein interactions at the substrate glycone binding area.

2.3 Cloning of β -mannanase genes manA and manC and site-directed mutagenesis of subsite +1 residues Trp283 and Ser289

P. pastoris strain X-33 transformants (FGSC database accession no. 10088 and 10106; www.fgsc.net) harbouring *A. nidulans* FGSC A4 β-mannanase genes (*manA*; GenBank EAA63326.1 and *manC*; GenBank EAA58449.1) [29] were purchased from Fungal Genetics Stock Center, School of Biological Sciences, University of Missouri, (Kansas City, MO). These clones encoded the full-length enzymes, signal peptides inclusive, predicted as Met1–Ala18 for both genes (SignalP 3.0; http://www.cbs.dtu.dk/services/SignalP) [35]

and were recloned to encode the mature polypeptides with the C-terminal hexa His-tag. Expand High Fidelity DNA polymerase (Roche Diagnostics GmbH, Mannheim, Germany) was used for DNA amplification with oligonucleotide primers (Supplementary Table S1) constructed based on the genomic sequence [28]. The PCR products were digested by EcoRI and NotI for manA and manC (New England BioLabs, Ipswich, MA) and cloned downstream in frame with **Saccharomyces** *cerevisiae* α -mating factor secretion signal in pPICZaA (Invitrogen, Carlsbad, CA). The obtained plasmids pPICZaA-ManA and pPICZaA-ManC were purified from Escherichia coli DH5a (Invitrogen) transformants selected on low salt Luria Bertani medium supplemented with 25 µg/mL zeocin, fully sequenced (Eurofins MWG Operon, Ebersberg, Germany), linearised by PmeI (New England BioLabs), and transformed into P. pastoris strain X-33 according to the manufacturer's recommendation.

Site-directed mutagenesis (QuikChange Lightning Site-Directed Mutagenesis kit; Stratagene, La Jolla, CA) was performed using ManAS289W and ManCW283S primer pairs (Supplementary Table S1) and pPICZ α A-ManA and pPICZ α A-ManC as templates. Mutant plasmids were transformed in *E. coli* DH5 α , verified by sequencing, and used for transformation of *P. pastoris* X-33 (as above).

2.4 Production and purification of recombinant ManA and ManC

P. pastoris transformants were grown in 4 × 1 L (ManA; ManC) and 3 × 1 L (ManAS289W; ManCW283S) buffered glycerol-complex medium (1% yeast extract, 2% peptone, 100 mM potassium phosphate pH 6.0, 1.34% yeast nitrogen base with ammonium sulfate, 4 × 10⁻⁵% biotin, 1% glycerol) in 2.5 L plastic baffled shake flasks at 30°C for 24 h to OD₆₀₀ of 34 (ManA), 39 (ManC), 34 (ManAS289W), and 28 (ManCW283S). The cells were harvested (3000*g*, 5 min, 4°C) and resuspended in 1 L buffered methanol-complex medium for

ManA and ManC (100 mM potassium phosphate pH 6.0, 1% yeast extract, 2% peptone, 1.34% yeast nitrogen base with ammonium sulfate, 4×10^{-5} % biotin, 0.5% methanol) or 1 L buffered minimal methanol medium for ManAS289W and ManCW283S (100 mM potassium phosphate pH 6.0, 1.34% yeast nitrogen base with ammonium sulfate, 4×10^{-5} % biotin, 0.5% methanol, 1% casamino acid). The induction continued at 22°C for 120 h (ManA; ManC) and 72 h (ManAS289W; ManCW28S) with methanol supplemented to the culture to a final concentration of 0.5% (v/v) every 24 h. Culture supernatants were harvested (13,500g, 4°C, 1 h), adjusted to pH 7.5 with 1 M K₂HPO₄, filtered (0.22 µm; TPP, Trasadingen, Switzerland), and applied to a 5 mL HisTrap HP column (GE Healthcare, Uppsala, Sweden) equilibrated with 20 mM HEPES, 0.5 M NaCl, 10 mM imidazole, pH 7.5. After wash with 20 mM HEPES, 0.5 M NaCl, 22 mM imidazole, pH 7.5, protein was eluted by a linear 22-400 mM imidazole gradient in the same buffer (25 mL; flow rate 1.0 mL/min). Fractions containing enzyme were pooled, concentrated (Centriprep YM30, Millipore), and applied to a Hiload 26/60 Superdex G75 column (GE Healthcare) equilibrated with 20 mM MES, 0.15 M NaCl, pH 6.8 for ManA and ManC or pH 7.0 for ManAS289W and ManCW283S (flow rate 0.5 mL/min). Enzyme-containing fractions were pooled, concentrated, and the buffer was exchanged to 20 mM HEPES pH 7.0 (Centriprep YM30, Millipore). All purification steps were performed at 4°C.

2.5 Protein characterisation

Protein concentration was determined spectrophotometrically at 280 nm using $E^{0.1\%} = 2.26$, 2.57, 2.33 and 2.51 for ManA, ManC, ManAS289W and ManCW283S, respectively, as determined by aid of amino acid analysis. Molecular masses were estimated by SDS-PAGE (NuPAGE Novex Bis-Tris gels; Invitrogen) and by analytical gel filtration using a precalibrated Hiload 16/60 Superdex G75 column (flow rate: 1 mL/min; GE Healthcare). Theoretical molecular masses were calculated by ExPASy–ProtParam tool (http://www.expasy.ch/ tools/protparam.html) [36]. *N*-glycosylation was predicted using NetNGlyc 1.0 Server (http://www.cbs.dtu.dk/services/NetNGlyc) [37]. Purified enzymes (7 μ g) were treated with endoglycosidase H (New England BioLabs) as recommended by the manufacturer. Carbohydrate content of purified enzymes (400 μ g) was quantified by the phenol-sulfuric acid procedure [38] using glucose and mannose as standards.

2.6 Enzyme stability and activity

 β -Mannanase activity towards 0.5% (w/v) konjac glucomannan, locust bean gum or guar gum galactomannan was measured at 37°C for 5 min in 100 mM sodium acetate, 0.005% BSA, pH 5.5 (400 μ L). Substrate (1 g) was dissolved in 95% ethanol (6 mL), added Milli-Q water (90 mL) with stirring and heated until boiling. Stirring continued until the mixture cooled down to room temperature and the volume was adjusted to 100 mL with Milli-Q water. Released reducing sugars were determined by adding 3,5-dinitrosalicylic acid solution (600 µL), incubating at 95°C (15 min), cooling on ice (10 min), and measuring absorbance at 540 nm [39,40] using a microtiter plate reader (Bio-Tek Instrument Inc., Winooski, VT). One unit (U) was defined as the amount of enzyme liberating 1 µmol/min reducing sugar equivalents under the assay condition, using mannose as standard.

The activity of 213 nM ManA, 104 nM ManC, 276 nM ManAS289W and 249 nM ManCW283S was measured using 0.5% locust bean gum galactomannan (as above) in 40 mM Britton-Robinson buffers pH 2.0–12.0 [41] and at 20–80°C in 50 mM sodium acetate pH 5.5. The dependence of stability on pH and temperature was deduced from residual activity measured as above after 3 h at 37°C in 40 mM Britton-Robinson buffers pH 2.0–10.0 or at 25–80°C in 50 mM sodium acetate pH 5.5 for 2130 nM ManA, 1040 nM ManC, 2760 nM ManAS289W, and 2490 nM ManCW283S.

2.7 Enzyme kinetics and action pattern towards manno-oligosaccharides

Initial rates of hydrolysis of M4-M6 (0.2-6.0 mM) were determined for ManA (60 nM with M4; 30 nM with M5; 10 nM with M6); ManC (75 nM with M4; 15 nM with M5; 12 nM with M6); ManAS289W (50 nM with M4; 28 nM with M5; 12 nM with M6); and ManCW283S (120 nM with M4; 32 nM with M5; 20 nM with M6) in 100 mM sodium acetate, 0.005% BSA, pH 5.5 (300 µL) from released reducing sugars quantified after 2.5, 5.0, 7.5, 10.0, 12.5 and 15.0 min reaction by copper bicinchoninate measuring the absorbance at 540 nm [42,43] and using mannose as standard. Kinetic parameters, $K_{\rm M}$ and $k_{\rm cat}$, were determined by fitting the Michaelis-Menten equation to initial rate data using SigmaPlot v. 9.0.1 (Systat Software Inc. San Jose, CA).

Action patterns on 0.5 mM M4-M6 were analysed using the same enzyme concentrations as above in 100 mM sodium acetate pH 5.5 at 37°C. Aliquots (6 μ L) were removed at appropriate intervals up to 30 min for M4-M6, and added to 100 mM NaOH (294 µL) to stop the reaction. Mannooligosaccharides were quantified from peak areas in HPAEC-PAD (ICS3000 system; Dionex Corporation, Sunnyvale, CA) equipped with CarboPac PA200 anion exchange column (3×250) mm with 3×50 mm guard column; Dionex) using isocratic elution with 37.5 mM NaOH for 35 min at 25°C; flow rate: 0.35 mL/min) calibrated with M1-M6.

2.8 Transglycosylation

ManA and ManC (91 nM) reacted with 14 mM M4 as donor and 400 mM of 36 potential acceptors (arabinose, L-arabinose, xylose, L-fucose, L-rhamnose, glucose, galactose, mannose, talose, fructose. glucosamine, N-acetylglucosamine, arabinobiose, cellobiose, gentibiose, isomaltose, lactulose, laminaribiose. lactose. maltose. mannobiose, melibiose, sophorose, sucrose, trehalose, turanose, xylobiose, arabinotriose, isomaltotriose, laminaritriose, mannotriose, maltotriose, melezitose, panose, raffinose and xylotriose) in 100 mM sodium acetate pH 5.5 at 37°C for 30 min (1 mL). The reaction products were quantified using HPAEC-PAD (see 2.7) and isocratic elution with 50 mM NaOH, 35 min at 25°C (flow rate: 0.35 mL/min) using M5 and M6 as standards for penta- and hexasaccharide products. Transglycosylation yield was calculated based on the donor concentration.

ManA or ManAS289W (330 nM) and ManC or ManCW283S (145 nM) reacted either with 30 mM M4 (without acceptor) or with 24 mM M4 and 100 mM M3, in 100 mM sodium acetate pH 5.5 at 37°C for 150-300 min (1 mL). For ManA and ManC aliquots (5 µL) removed at 0, 2.5, 5, 10, 15, 20, 25, 30, 35, 40, 45, 60, 80, 100, 120 and 150 min, and for ManAS289W and ManCW283S aliquots removed at 0, 2.5, 5, 10, 20, 30, 45, 60, 90, 120, 150, 180, 210, 240, 270 and 300 min were added to 100 mM NaOH $(245 \ \mu L)$ to stop the reaction. To optimize transglycosylation yields, the time course of 145 nM ManC with 24 mM M4 and 200 mM isomaltotriose or melezitose in 100 mM sodium acetate pH 5.5 (1 mL) at 37°C was determined from aliquots (5 µL) removed at 0, 2.5, 5, 10, 15, 20, 25, 30, 35, 40, 45, 60, 80, 100, 120, 140, 160 and 180 min and stopping the reaction as above. A minor product M7 appeared in HPAEC-PAD and was estimated using M6 as standard as M7 is not available and too small amounts of M7 were produced to establish a standard curve using purified M7.

2.9 Manno-oligosaccharide purification

For structural analysis of transglycosylation products reaction mixtures of ManC (145 nM), 24 mM M4 and 100 mM M3, 200 mM isomaltotriose or 200 mM melezitose were incubated at 37°C (60, 120 and 180 min for M3, isomaltotriose and melezitose, respectively, as guided by time course experiments in 2.8) in 100 mM sodium acetate, pH 5.5 (1 mL) and heated (95°C, 10 min), centrifuged (18,000*g*, 4°C, 5 min), desalted (Amberlite MB-20; Fluka, Sigma Aldrich), filtered (0.45 µm nylon filter; Frisenette Aps, Knebel, Denmark), and purified by HPLC (UltiMate 3000 Standard LC system; Dionex) equipped with a TSKgel Amide-80 column (5 μ m, 4.6 × 250 mm and 4.6 × 10 mm guard column; TOSOH, Tokyo, Japan) and refractive index detector (RI-101; Showa Denko, Kanagawa, Japan) at constant flow rate (1 mL/min) of mobile phase (acetonitrile/water 70:30, v/v) at 70°C. Purity was confirmed by thin layer chromatography (aluminium-coated silica gel 60 F₂₅₄, Merck) developed twice by acetonitrile/water (70:30, v/v), sprayed with 2% orcinol in sulphuric acid/water/ethanol (10:10:80, v/v/v) [44] and tarred at 120°C.

2.10 Nuclear Magnetic Resonance spectroscopy (NMR)

Purified oligosaccharides (500 μ L, 1–2 mM) were freeze dried (CoolSafe –55°C; ScanLaf A/S, Lynge, Denmark) and dissolved in deuterium oxide prior to recording of spectra on a Varian Unity Inova 500 MHz NMR spectrometer or a Bruker Avance 800 MHz spectrometer. A series of 2D homo- and heteronuclear correlated spectra: DQF-COSY, NOESY, gHSQC and gHMBC, were acquired using standard pulse sequences. The NMR data for structural assignment were acquired in D₂O.

2.11 Matrix-assisted laser desorption/ionisationtime of flight mass spectrometry (MALDI-TOF MS)

MALDI-TOF MS was made using an Ultraflex II TOF/TOF instrument (Bruker Daltonics GmbH, Bremen, Germany) selected for positive ions and analysed at 25040 V and detected using reflector mode. High (75%) laser power was required to obtain good spectra and >100 spectra were collected. Calibration was done with M5 and M6 (as Na^+ adducts) and trypsin-digested β -lactoglobulin (mass range m/z 851–2313.3). The matrix solution was prepared by dissolving 10 mg 2.5dihydroxybenzoic acid in 1 mL acetonitrile/water (30:70, v/v) containing 10 mM NaCl. Purified samples of 0.2 mM M5 or manno-isomaltotrioses and 0.1 mM M6 or manno-melezitoses (1 µL) and matrix (1 µL) were pipetted onto a MALDI-TOF-

target plate (Bruker Daltonics) and air-dried prior to analysis [44].

3. Results and discussion

3.1 Production and purification of recombinant ManA, ManC, ManAS289W and ManCW283S

His-tag fusions of ManA and ManC were secreted by P. pastoris and 120 and 145 mg/L, from respectively, were purified culture supernatants. These yields are very high compared to native GH5 endo- β -1,4-mannanases secreted by A. niger (1.6 mg/L) [11] or H. jecorina (0.8 mg/L [13]; 2 mg/L [45]) and to recombinant A. aculeatus endo-β-1,4-mannanase produced GH5 in S. cerevisiae (1.8 mg/L) [12].

ManC migrated in SDS-PAGE as a single band of 56 kDa, whereas ManA gave a smear with the lower band corresponding to 56 kDa (Supplementary Fig. S1); the calculated values are 43847.1 and 46643.8 Da, respectively, and endoglycosidase H treatment reduced the apparent molecular mass of ManA and ManC by 10 and 5 kDa, respectively, and the deglycosylated ManA migrated as sharp band of 45 kDa (Supplementary Fig. S1). The effect of endoglycosidase H confirmed that ManA and ManC were glycosylated in agreement with a content of four and two predicted N-glycosylation sites and carbohydrate contents were estimated to 19% and 15% using glucose, and 13% and 9% using mannose as standard, respectively. Glycosylation is not uncommon for fungal secreted GH5 endo-\beta-1,4-mannanases as found e.g. for H. jecorina mannanase [45] that shares one site with ManA. Gel filtration indicated that both ManA and ManC are monomers of approximately 58 kDa (data not shown).

The crystal structure of *H. jecorina* GH5 endo- β -1,4-mannanase (*Hj*Man5A) with bound mannobiose [17] (see 2.2) showed Trp114, Arg171, Glu205 and Trp247 are involved in substrate binding at subsites +1 and +2 (Fig. 1C). These residues are well conserved in ManA (Trp155, Arg213, Glu247, and Trp290) and ManC (Trp140, Arg198, Glu243, and Trp284) (Fig. 1; Supplementary Fig. S2). According to the ManC homology model, Trp283 is close enough to interact with a mannosyl residues bound at subsite +1, together with Trp140 (Fig. 1B). Trp283 corresponds to Ser289 in ManA (Fig. 1A) and HiMan5A Ser246, both located about 5Å from the sugar ring and thus unlikely to interact with the mannosyl moiety at subsite +1. To investigate the impact of ManC Trp283 in hydrolysis and transglycosylation, ManAS289W and ManCW283S mimicking Trp283 and Ser289 in ManC and ManA, were made by sitedirected mutagenesis and obtained in yields of 12 and 24 mg/L culture, respectively. ManAS289W and ManCW283S were monomers of 59 and 56 kDa as determined by gel filtration and migrated in SDS-PAGE as the parent enzymes (Supplementary Fig. S1).

3.2 Hydrolytic activity of ManA, ManC, ManAS289W and ManCW283S

ManA and ManC showed increasing catalytic efficiency (k_{cat}/K_M) with increasing DP of manno-oligosaccharides (M4-M6) (Table 1) as is characteristic of endo-acting mannanases, suggesting that ManA and ManC possess with 5-6 substrate binding subsites [15–18]. Both enzymes exhibited low affinity for M4 similarly to GH5 endo-β-1,4-mannanases from *Cellvibrio japonicus* [10] and Mytilus edulis [16]. ManC moreover showed about twice as high k_{cat}/K_M towards M5 and M6 as ManA, in part due to a lower $K_{\rm M}$ tentatively attributed to the presence of Trp283 in ManC (see Fig. 1). The k_{cat}/K_{M} of the subsite +1 mutants, ManCW283S and ManAS289W, increased with DP of manno-oligosaccharides as for the wildtype enzymes. While k_{cat}/K_M of ManCW283S decreased bv 50% compared to ManC. ManAS289W and ManA gave similar k_{cat}/K_{M} values, although $K_{\rm M}$ of the mutant decreased and more so for M5 than for M6. Hence, the differences in k_{cat}/K_{M} towards M5 and M6 for subsite +1 mutants as compared to wild-type enzymes, stemmed mainly from changes in $K_{\rm M}$, which decreased for ManAS289W and increased for ManCW283S. These observations are in accordance

with contribution of ManC Trp283 to binding at that the Trp is not critical for activity. subsite +1. The modest effect, however, suggests



Figure 1. 3D cartoons representing the key active site residues of the homology models of ManA (A) and ManC (B), and the crystal structure of endo- β -1,4-mannanase from *H. jecorina* RUTC-30 (C) with mannobiose bound at subsites +1 and +2 (PDB ID: 1QNR) [17]; mannobiose is superimposed at subsites -2 and -3 from the structure of the complex with *T. fusca* KW3 mannanase (PDB ID: 2MAN) [34] (see 2.2). Nuc, catalytic nucleophile; A/B, catalytic acid/base; Nuc and A/B residues of ManA and ManC were predicted according to sequence alignment and 3D-models. The same color coding is residues as used on the stereo view panels (to the left) and the surface presentations (to the right).

(1111 1110)				
		$k_{\rm cat}$	K_{M}	$k_{\rm cat}/K_{\rm M}$
Enzyme	Substrate	s ⁻¹	mM	$s^{-1}mM^{-1}$
ManA	M4	_	_	6
	M5	67 ± 2.8	2.9 ± 0.03	23
	M6	193 ± 5.8	1.8 ± 0.08	109
ManAS289W	M4	_	—	10
	M5	42 ± 2.5	1.6 ± 0.16	27
	M6	148 ± 1.7	1.3 ± 0.05	115
ManC	M4	_	_	7
	M5	112 ± 8.1	1.8 ± 0.13	61
	M6	134 ± 5.2	0.6 ± 0.04	215
ManCW283S	M4	_	_	3
	M5	98 ± 3.1	2.8 ± 0.13	36
	M6	117 ± 3.3	1.0 ± 0.03	118

Table 1. Hydrolysis kinetic parameters of ManA, ManAS289W, ManC and ManCW283S towards manno-oligosaccharides (M4–M6)

Standard deviations were calculated from triplicate experiments

The activity towards locust bean gum galactomannan, guar gum galactomannan, and konjac glucomannan was 30-80% higher for ManC than ManA (Table 2), which has activity towards locust bean gum galactomannan comparable to other fungal GH5 endo- β -1,4-mannanases from *H*. jecorina [46], A. fumigatus [47], A. niger [11], A. culeatus [12] of 230-630 U/mg. ManC has the highest reported activity (730 U/mg) and prefers locust bean gum galactomannan (Table 2), possibly due to its lower content of α -1,6-galactosyl substituents (galactose:mannose, 1:4) compared to guar gum galactomannan (galactose:mannose, 1:2) also higher [1–2]. ManC has activity for galactomannan than glucomannan (Table 2). whereas ManA appears almost insensitive to the side chain substitution and backbone composition as the 340–400 U/mg for all three activity was polysaccharides (Table 2). Currently no structural basis has been identified for the discriminate action of GH5 endo-β-1,4-mannanases towards polysaccharides with varying extent of side chain substitution or backbone composition. A. nidulans FGSC A4 produces at least three homologous GH5 endo- β -1,4-mannanases [47] differing with regard to substrate specificity, and these may confer synergy required for depolymerisation of mannans [2,10]. In fact the subsite +1 ManCW283S mutant in contrast to wild-type ManC had essentially the same activity towards all three polysaccharides, suggesting that Trp283 (Fig. 1B) contributes to the specificity differences between ManC and ManA (Table 2). Interestingly, the mutations seemed to modulate specificity with respect to degree of branching and backbone composition. Hence the ratio of specific activities of guar/locust bean gum galactomannan glucomannan/locust and konjac bean gum galactomannan of ManCW283S resembles those of ManA and vice versa (Table 2). Both ManCW283S and ManAS289W lost 25-60% activity towards all three mannans indicating higher catalytic efficiency of the wild-type enzymes, although ManAS289W has comparable activity to wild-type towards manno-oligosaccharides (Table 1). Interrogating the structure of *H. jecorina* in complex with

mannobiose (Fig. 1C) and the homology models of ManA and ManC (Fig. 1A,B) suggested that galactose substitution at the mannosyl ring at subsite +1 is not tolerated as the 6-OH group points into the active site (Fig. 1C). An O6 galactosyl substituted mannosyl, however, seems more readily accommodated at subsite -1 this can also be affected by the Trp \rightarrow Ser substitution at the neighbouring subsite +1. Lack of structural data prevents detailed insight, but Trp283 in ManC inevitably makes the active site cleft narrower and more restrained than the Ser in ManA (Fig. 1).

ManA and ManC have activity optimum at pH 5.5 (Fig. 2A) and stable (>95 % residual activity) at pH 4.5–7.0 and 4.0–8.0, respectively (Fig. 2B). Both enzymes showed maximum activity at 50°C (Fig. 2C) and retained >95% activity after 3h incubation up to 40°C (Fig. 2D). ManCW283S showed slightly reduced stability in the alkaline pH range compared to the parent enzyme, but otherwise the subsite +1 mutation had little effect on the dependence of activity stability of pH and temperature (Fig. 2). This behaviour resembled that of other fungal GH5 β -1,4-mannanases showed having reported activity optima at pH 3.0–7.5 and 45–70°C [2,11–13,45–47].

To gain further insight into the mode of substrate interaction, the time course of hydrolysis of M4-M6 was monitored by HPAEC-PAD (Fig. 3), and the products were quantified towards the end of the reaction (Fig. 4). ManA and ManC both formed M2 as main hydrolysis product from M4, and M2 together with M3 and M4 from M5 and M6. These action patterns suggest higher-affinity substrate binding at subsite(s) -2 and/or +2, even though the hydrolysis of M4 was not fully complete (>65% hydrolysed). Noticeably, ManA produced more M1 and M3 from M4 than ManC (Fig. 4), suggesting relatively more important affinity at subsite(s) -3and/or +3 or lower affinity at subsite(s) -2 and/or +2for ManA. However, ManA has a relatively lower affinity at the -2 and/or +2 subsites since ManA and ManC produced 12% and 8% M3 from M6, reflecting a similar relative affinity for their subsite(s) -3 and/or +3. The 0.5 mM of mannooligosaccharides was used for hydrolysis action pattern analysis to prevent transglycosylation at higher substrate concentration.

Sequence alignment and homology models of ManA and ManC superimposed onto the mannobiose complex of T. fusca GH5 mannanase (TfMan; see 2.2) identify ManA Tyr68, Asp157, Tyr158, Asp362 and ManC Tyr45, Asp142, Tyr143, His368 to interact with substrate at subsites -2 and -3 (Fig 1A,B). Moreover ManA Trp155, Arg213, Glu247, Trp290 and ManC Trp140, Arg198, Glu243, Trp283, Trp284 appear to be involved in substrate binding at subsites +1 and +2 (see 3.1). These proposed interactions can agree with ManA and ManC producing mostly M2 in hydrolysis of M4-M6. Moreover, ManC His368 is solvent exposed and may obscure substrate binding at subsite -3, possibly resulting in the higher formation of M2 in hydrolysis of M4 than in ManA. In case of subsite +1mutants, ManAS289W the and ManCW283S also formed M2 as main hydrolysis product from M4-M6 and the action patterns showed no obvious difference from their wild-type enzymes (Fig. 3D-F,J-L and Fig. 4). Hence ManC Trp283 has no effect on substrate binding pattern.

3.3 Transglycosylation by ManA, ManC, ManAS289W and ManCW283S

Transglycosylation catalysed by ManA and ManC was monitored using 30 mM M4 (Fig. 5; Supplementary Fig. S3) and M5–M7 were produced by ManA (Fig. 5A), whereas ManC produced only M5 and M6 (Fig. 5C). Maximum product yields (M5+M6) of ManA and ManC obtained after 25 and 100 min, were 2% and 16% (Fig 5A,C), respectively. ManC transglycosylation was thus 8 times more efficient than by ManA. The major transglycosylation product M6 reflected transfer of mannobiosyl to M4 as acceptor in agreement with the action pattern analysis (Fig. 4) showing M2 as the main hydrolysis product from M4, and M2 and M4 as main products from M6. Clearly subsite –2 governs binding of the glycone part of the active site. Using 24 mM M4 as donor and 100 mM M3 as acceptor resulted in approximately 50% increased transglycosylation yields (M5+M6) of 4% and 35% for ManA (at 40 min) and ManC (at 120 min), respectively (Fig, 5E,G). M5 was the major product confirming preferred transfer of mannobiosyl to M3.

The subsite +1 mutant ManAS289W produced M5-M7 (Fig. 5B) while ManCW283S produced only M5 and M6 (Fig. 5D) from 30 mM M4, thus the mutants caused no real difference from the parent ManA and ManC enzymes although the M5+M6 yield of ManAS289W was 5% (at 30 min), which is 2.5 fold higher than for ManA, whereas the vield of ManCW283S was reduced to 9% (at 90 min) compared to ManC (16%). Thus somehow the Trp283 and Ser289 determine the fine-specificity of the enzymes. This is also supported when using 24 mM M4 and 100 mM M3, resulting in yields of 8% (at 90 min) and 24% (at 150 min) for ManAS289W and ManCW283S (Fig. 5F.H). respectively. representing doubling the yield for ManAS289W compared to ManA (4%), whereas yield decreased 1.5 fold with ManCW283S compared to wild-type ManC Trp283 role (35%). has а in transglycosylation possibly increasing acceptor affinity at subsite +1, which is reminiscent of the decrease and increase of K_{M} found for ManAS289W and ManCW283S, respectively (see 3.2).

		Specific activity ^b	
Enzyme	Substrate ^a	U/mg	Ratio ^c
ManA	Locust bean gum galactomannan	$400 \hspace{0.1in} \pm 4$	1
	Guar gum galactomannan	$390 \ \pm 6$	0.98
	Konjac glucomannan	340 ± 2	0.85
ManAS289W	Locust bean gum galactomannan	300 ± 1	1
	Guar gum galactomannan	$250\ \pm 1$	0.83
	Konjac glucomannan	220 ± 1	0.73
ManC	Locust bean gum galactomannan	$730\ \pm 2$	1
	Guar gum galactomannan	$600 \ \pm 2$	0.82
	Konjac glucomannan	$430 \ \pm 4$	0.59
ManCW283S	Locust bean gum galactomannan	$320 \ \pm 4$	1
	Guar gum galactomannan	$300 \ \pm 1$	0.94
	Konjac glucomannan	$270 \ \pm 1$	0.84

Table 2. Specific activities of purified ManA, ManAS289W, ManC, andManCW283S towards mannans

^a 0.5% mannan substrates

^b Standard deviations were calculated from triplicate experiments

^c Normalized by specificity activity towards locust bean gum galactomannan



Figure 2. Effect of pH and temperature on activity and stability of ManA (•), ManC (\blacksquare), ManAW289S (\circ) and ManCS283W (\Box) using 0.5% locust bean gum galactomannan as substrate (broken lines are used for mutants). A) pH-activity dependence for 213 nM ManA, 104 nM ManC, 276 nM ManAS289W and 249 nM ManCW283S in 40 mM Britton-Robinson. B) pH-stability dependence of 2.13 μ M ManA, 1.04 μ M ManC, 2.76 μ M ManAS289W and 2.49 μ M ManCW283S in 40 mM Britton-Robinson buffer after 3 h incubation at 37°C. C) Temperature-activity dependence for the same enzyme concentrations as in B at 20–80°C after 3 h incubation in 50 mM sodium acetate pH 5.5. Each experiment was made in triplicate. Standard deviations are shown as error bars.



Figure 3. Hydrolysis action patterns of ManA (A–C), ManAS289W (D–F), ManC (G–I), and ManCW283S (J–L) towards 0.5 mM M4–M6 in 100 mM sodium acetate pH 5.5 at 37°C. A) 60 nM ManA with M4; B) 30 nM ManA with M5; C) 10 nM ManA with M6; D) 50 nM ManAS289W with M4; E) 28 nM ManAS289W with M5; F) 12 nM ManAS289W with M6; G) 75 nM ManC with M4; H) 15 nM ManC with M5; I) 12 nM ManC with M6; J) 120 nM ManCW283S with M4; K) 32 nM ManCW283S with M5; L) 20 nM ManCW283S with M6. M1 (Δ); M2 (\Box); M3 (\circ); M4 (\diamond); M5 (\blacktriangle); M6 (\blacksquare).

Substrate	ManA	ManA S289W	ManC	ManC W283S
M4	64%	68%	91%	88%
	- 36%	32%	9%	12%
M5	- 100%	100%	96%	100%
		-	4%	
M6	- 88%	88%	92%	94%
	12%	12%	8%	6%



Figure 4. Schematics of hydrolysis products from action pattern analysis (Fig. 3) after 20 min reaction by ManA, ManAS289W, ManC or ManCW283S on 0.5 mM manno-oligosaccharides. ●) Mannose residue; ▲) possible cleavage site.

Figure 5. Transglycosylation product formation catalysed by ManA (330 nM), ManAS289W (330 nM), ManC (145 nM), and ManCW283S (145 nM) using 30 mM M4 (A–D) or 24 mM M4 as donor with 100 mM M3 as acceptor (E–H) in 100 mM sodium acetate pH 5.5 at 37°C. M5 (\blacktriangle); M6 (\blacksquare); M7 (\diamondsuit).

3.4 Identification of carbohydrate acceptors for production of novel manno-oligosaccharides

The specificity of ManA and ManC determined by screening with 12 mono-, 15 di-, and 9 trisaccharide acceptor candidates (400 mM; all listed in 2.8) with M4 (14 mM) as donor. ManC transglycosylated M3, isomaltotriose and melezitose, while none of the monosaccharides and disaccharides served as acceptors during 30 min reaction. ManA only used M3 as acceptor. These findings suggest that ManA and ManC require binding at subsite +3 for functional accommodation of an acceptor in transglycosylation. This behaviour is supported by kinetics analysis on mannooligosaccharides showing poor affinity for M4 while $K_{\rm M}$ was 0.6–2.9 mM for M5 and M6 (Table 1) most probably occupying subsite +3 (Fig. 3 and 4). No crystal structure is reported of GH5 endo-β-1,4mannanase with sugar bound at subsite +3, but the models (Fig. 1) suggest ManA Tyr259 and ManC Trp253 are surface exposed and may stack onto a trisaccharide acceptor at subsite +3. An aromatic residue is conserved at this position in fungal GH5 endo- β -1,4-mannanases suggesting this site has functionally importance in recognition of substrate.

ManC gave highest transglycosylation yields and was used to generate oligosaccharide products from 24 mM M4 with 200 mM isomaltotriose or melezitose as acceptor (Fig. 6) for NMR structure determination. Two products were obtained in a ratio of about 5:1 with each acceptor due to transfer of mannobiose and mannotriose to give highest vield (sum of penta- and hexasaccharide products) of manno-isomaltotriose and manno-melezitose of 2.51 and 0.64 mM after 120 and 180 min reaction (Fig. 6) calculated to 10% and 3% total yields, respectively, based on the donor concentration. MALDI-TOF MS showed molecular mass with Na⁺ adducts of the products with M3 (100 mM), isomaltose and melezitose to be m/z 851 and 1013 (Supplementary Fig. S4) indicating the predominant and minor products to be а pentasaccharide and а hexasaccharide, respectively.

The structures of the major transglycosylation products: mannobiosyl-

isomaltotriose, mannobiosyl-melezitose, and M5 were assigned by 2D NMR (Supplementary Table S2). Similarly to M5, the mannobiosylisomaltotriose $(\beta$ -mannopyranosyl- $(1 \rightarrow 4)$ - β mannopyranosyl- $(1\rightarrow 4)$ - α -glucopyranosyl- $(1\rightarrow 6)$ - α -glucopyranosyl-(1 \rightarrow 6)- α -glucopyranoside; Fig. 7A) mannobiosyl-melezitose and (βmannopyranosyl- $(1\rightarrow 4)$ - β -mannopyranosyl- $(1\rightarrow 4)$ - α -glucopyranosyl-(1 \rightarrow 3)- β -fructofuranosyl-(2 \rightarrow 1)- α -glucopyranoside; Fig. 7B) showed a β -1,4 linkage between the mannobiosyl and trisaccharide acceptor, as identified by the downfield shift of the C4, β-1,4 indicating regioselectivity. Noticeably, mannobiosyl-melezitose also contained minor compounds identified as mannobiosyl 1,6 linked to the fructosyl group, as indicated by the downfield shift of the C6 in fructose residue, and two mannobiosyl groups linked to one melezitose acceptor, which appeared as a small peak at m/z1175.3 in MALDI-TOF MS (Supplementary Fig. S4E). The minor transglycosylation products: mannotriosyl-isomaltotriose and mannotriosylmelezitose, could not be purified in sufficient amounts enabling the 2-D NMR analysis. MALDI-MS, however, identified that these products were Na⁺ adducts of hexasaccharides (Supplementary Fig. S4D,F).



Figure 6. Progress of total transglycosylation products (sum of penta- and hexasaccharide products) formation catalysed by ManC (145 nM) using 30 mM M4 with 200 mM isomaltotriose (A) or 200 mM melezitose (B) in 100 mM sodium acetate pH 5.5 at 37°C during 180 min.



Figure 7. Structure of novel manno-oligosaccharides. A) β -Man*p*-(1 \rightarrow 4)- β -Man*p*-(1 \rightarrow 4)- α -Glc*p*-(1 \rightarrow 6)- α -Glc*p*-(1 \rightarrow 6)- α -Glc*p*; B) β -Man*p*-(1 \rightarrow 4)- β -Man*p*-(1 \rightarrow 4)- α -Glc*p*-(1 \rightarrow 3)- β -Fur*f*-(2 \rightarrow 1)- α -Glc*p*.

4. Conclusion

The A. nidulans FGSC A4 endo-β-1,4mannanase ManC showed 30-80% and 2-3 fold higher activity towards mannans and catalytic efficiency towards manno-oligosaccharides, respectively, compared to ManA and moreover had eight fold higher transglycosylation capacity. While ManA seemed to have strict acceptor specificity in transglycosylation reactions and accepted only manno-oligosaccharides, ManC could also use isomaltotriose and melezitose as acceptors resulting in novel β -mannobiosyl- and β -mannotriosyloligosaccharides, that have interest as potential prebiotics. In ManC, Trp283 at subsite +1 is involved in both manno-oligosaccharide hydrolysis and transglycosylation. Moreover this Trp residue contributed to the polysaccharide substrate specificity by discriminating the mannans with varying degree of galactose side chain substitution as well as glucose backbone composition.

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Supplementary documents



Supplementary Figure S1. SDS-PAGE of purified and deglycosylated ManA, ManC, ManAS289W and ManCW283S; Lane A, 7 μ g ManCW283S; Lane B, 7 μ g deglycosylated ManCW283S; Lane C, 5 μ g ManAS289W; Lane D, 5 μ g deglycosylated ManAS289W; Lane E, 5 μ g ManC; Lane F, 5 μ g deglycosylated ManC; Lane G, 5 μ g ManA; Lane H, 5 μ g deglycosylated ManA; M, Marker: Mark12 unstained protein standard marker (Invitrogen); endoglycosidase H migrates as a 29 kDa band seen in lane B, D, F, and H.

ManC	<u>α0</u>	-β1	^{β2}	β3
ManC ManA HjMan5A_lqnr SlMan4_lrh9 CmMan5A_luuq MeMan5A_2c0h TfMan_lbqc	1 10 M. IFSTLLSLALLA	20 RKGFVTTKG SGSFATTSG S.SFVTISG NNNFVYTDG HEHFVRVNG CQARLSVSG GLHVKNG	30 KFQLDG IQFVIDG IQFNIDG IHFALNG GHFELQG INLNYNG RLYEANG	40 KDFYFACSNA EAGYFPGSNA KVGYFGSNA KSLYINGFNA KPYVITGVNM HHIFLSGANQ OEFIIRGVSH
ManC ManC ManA HjMan5A_lqnr	αl <u>perconnection</u> 50 60 70 	<u>2222</u> 80 NDKNVTYIEI NDVNTQ	90 DGLPQYG PTDG PSPG	100 Gegagttevv Svy Qiw
CmMan5A_luuq MeMan5A_2c0h TfMan_1bqc	WYAAY. LGAPNEVGDRDRLAKELDNLKAIGVNNIRVLAV AWVNYARDFGHNQYSKGKSTFESILSDIQSHGGMSVRVWLH PHNWYPQHTQAFADIKSHGANTVRVVLS	SEKSEINSA IEGEST NG	VKPA PEFD VRWS	
ManC ManA HjMan5A_lqnr SlMan4_lrh9 CmMan5A_luuq MeMan5A_2c0h TfMan_lbqc	110 120 120 130 140 FQWWQNGISTID LEPFDKVVNAAKTGIKLIVTLVNNWAI FQLHQDGKSTINTGKDGLORLDYVVHSIAEKHGIKLIINFVNNWD FQKLSATGSTINTGADGLOTLDYVVQSAECHNLKLIIFFVNNWD VYNCGHCGLDFVISEAKKYGIHLIMSLVNNWD VINCFGNYDETLLGLDYLLVELAKRDMVVVLYFNNFWQ NNGYVTGIDNTLISDMRAYLHAACRHNILIFFTLWN KNGPSDVANVISLCKQNRLICMLE	150 DYGGMDVYT DYGGMNAYMI AFGGKKQYVI WSGGMTQYMJ GA	VNLGG RAYGG NAF.G EWAVQRG AWIEGEP	QKL VQDPNVINEW
ManC ManC HjMan5A_1qnr SlMan4_1rh9	α3 200000000000000000 160 170 180 	. 2 . YRNSPAIM . YINSPAIM . YANSTAIF . XANSTAIF AYKDDPTILS	35 AWELANE AWELANE AWELGNE SWELINE	PRC PRC PRC
CmMan5A_luuq MeMan5A_2cOh TfMan_lbqc ManC	EAFMAKSASFYRSEKAQQEYRKTLEKIITTVVSINGK. VKQSTHYRLNGLMVDTRKLQSYIDHALKPMANA VHDTTGYGEQSGASTLDQAVDYWIE.LKSV	AYVDDATIM: .LKNEKALG(.LQGEEDYV)	SWQLANE SWDIMNE LINIGNE	PRP PEGEIKPGES PYG
ManC ManA HjMan5A_lqnr SlMan4_lrh9 CmMan5A_luuq MeMan5A_2c0h TfMan_lbqc	200 210 220 230 GADGVRNLPASDECTPELLTSWIDEMSTYVKR GCEPSVLHNWIEKTSAFIKG NGCSTDVIVCWATSVSCYVKS 	LDPHHLY LDEKHLY LDSNHLY IDSNHLY LDAHHLY VDPGAMY LRAAGFEHTY	Z40 TTWGGEG VCIGDEG VTLGDEG VTLGDEG VSSGSEG VTVGSWN LVVDAPN	250 .GFNYDSD FGLDTGSD LGLSTG.D FYGNDMR .EMGSVNDMQ MKADTDA WGQDWINTMR
ManC	▼ 260 270 280 ▼	2222222	α5 	0000000 300
ManC ManA HjMan5A_1qnr SlMan4_1rh9 CmMan5A_1uuq MeMan5A_2c0h TfMan_1bqc	DWAYNGSDGOF.EAELKLKNTDFGVF SYPDWWSKT GSYPFQYTEGSDF.AAALTIDTIDFGTFBLYPDSWGTN GAYPYTYGEGTDF.AKNVQIKSLDFGTFBLYPDSWGTN QYNPNSYIFGTNF.ISNNCVQGIDFATTHMYPNQWLEGLTQEAQ VF.IDAHAFPDIDYLIYHWIKNKNWSWFDKTKPA MGFHNLYSDHCLVKAGGKOSGTLSFYQVHTY.DWQ	VEWTNKW NDWGKLW YTWGNGW DKWASQW ETWPSAWEKJ NHFGNES GVYSQJ	IVD IQT IQT IQV AQNYMRA PFK AST.ITS	HARAARRVGK HAAACAAAGK HAAACLAAGK HIDDSKMLKK HIDDSKQLNK HSFSNFRLKK YLEHFVNAGL
ManC	β6			60
ManC ManA HjMan5A_lqnr SlMan4_lrh9 CmMan5A_luuq MeMan5A_2c0h TfMan_lbqc	VVHEEYGWLTPQGRLDNLGTVSI.NITRLEAVGGWQSISLREK PCLFEEYGVTSNHCAIEK.QWQNAALNAT PCVFEEYGAQQNPCTNEA.PWQTTSLTR PLLIABFGKSTKTPGYTVAKRDNYFEKIYGTIFNCAKSGG PLVLEBFGLDRDMGSYAMDSTTEYRDNYFRGVFELMLASLEQGE PMVIGEFNQEHGAGMSSESMFEWAYTK PLIIGEFGHDHSDGNPDEDTIMA.	MSDMFW. GIAADLYW. GMGGDMFW. PCGGGLFW. PSAGYNIW. GYSGAWIWSI LKLGYIG <mark>W</mark> .	QFG QYG QWG QVL AWN RTDVSWN SWS	YSGYSYG DTLSSGP DTFANGA GQGMS GYGRTTRANY NQLRGIQHL. GNGGGVEYLD
ManC	₹ -►00000000000000000000000000000000			
ManC ManA HjMan5A_1qnr SlMan4_1rh9 CmMan5A_1uuq MeMan5A_2c0h TfMan_1bqc	370 380 390 RNHDDGFTIYLDDAEAQELVYKHAKEVNKLNR SPDDGNTFYYGSEFECLVTNHVETIERSAK QSNSDPYTVWYNSSNWQCLVKNHVDAINGGTT SFDDGYQVVLQESFSTSRVIWWQEGDDFMGDPPQEEQGMYGVFDTDTSTIAIMKEFNARFQPK KSRTDHGQVQFGL KSRTD	RR	TTSSRTS	STPPPPGGSC

ManC																									•											
HjMan5A_1qnr	s	P	ŗ	Ŷ	Ġ	ò	ċ	Ġ	Ġ	s	Ġ	Ŷ	Ť	Ġ	P	T	ċ	ċ	À	Q	Ġ	Ť	ċ	i	Ŷ	ŝ	Ņ	Ŷ	w	ż	s.	ò	ċ	ŗ	Ň	Ť
CmMan5A_1uuq	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:			?	•	•	А •	
MeMan5A_2c0h TfMan_1bqc	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:	:

ManC

Supplementary Figure S2. Amino acid sequence alignment of ManA and ManC from *A. nidulans* FGSC A4 and GH5 endo- β -1,4-mannanases with reported crystal structures from *Hypocrea jecorina* (*Hj*Man5A_1qnr; UniProt ID: Q99036), *Solanum lycopersicum* (*Sl*Man4_1rh9; UniProt ID: Q8L5J1), *Cellvibrio mixtus* (*Cm*Man5A_1uuq; UniProt ID: Q6QT42), *Mytilus edulis* (*Me*Man5A_2c0h; UniProt ID: Q8WPJ2), and *Thermobifida fusca* KW3 (*Tf*Man_1bqc; UniProt ID: Q9ZF13). The sequence alignment was performed using T-Coffee (http://tcoffee.vital-it.ch/cgi-bin/Tcoffee/tcoffee_cgi/index.cgi) [49] and visualised using Easy Sequencing in Postscript (http://espript.ibcp.fr/ESPript/ESPript) [50]; \checkmark) possible subsite -2 location; \checkmark) possible subsite +1 and +2 location; \checkmark) possible subsite +3 location; \checkmark) mutation position (subsite +1) in present work.



Supplementary Figure S3. Progress of transglycosylation reactions (substrates and hydrolysis products) of ManA, ManAS289W, ManC and ManCW283S using 30 mM M4 (A–D) or 24 mM M4 with 100 mM M3 (E–H). A) 330 nM ManA; B) 330 nM ManAS289W; C) 145 nM ManC; D) 145 nM ManCW283S; E) 330 nM ManA; F) 330 nM ManAS289W; G) 145 nM ManC; H) 145 nM ManCW283S; M1, (Δ); M2, (\Box); M3, (\circ); M4, (\diamond).



Supplementary Figure S4. Molecular masses of transglycosylation products as Na⁺ adduct (mass range 300–1500 *m/z*). A), M5: 851.3 *m/z*; B), M6: 1013.3 *m/z*; C), mannobiosyl isomaltotriose: 851.2 *m/z*; D), mannotriosyl isomaltotriose: 1013.3 *m/z*; E), mannobiosyl melezitose: 851.2 *m/z*; F), mannotriosyl melezitose: 1013.2 *m/z*.

Supplementary Table S1 Primers for construction of the expression plasmids for ManA, ManC, ManAS289W, and ManCW283S

Primer name	Primer sequence	Restriction site
ManA		
Sense	GGG <u>GAATTC</u> CTCCCTCACGCGTCAACACCTGTGTACAC ^a	EcoRI
Antisense	CCC <u>GCGGCCGC</u> CTACTTGGCACTCCTCTCAATCGTCTCG ^a	NotI
ManC		
Sense	GGG <u>GAATTC</u> CGCAAGGGCTTTGTGACCACCAAAGGCG ^a	EcoRI
Antisense	CCC <u>GCGGCCGC</u> CTACCGTCTCCGGTTCAACTTGTTCACC ^a	NotI
ManAS289W		
Sense	CTCTACCCGGAT <u>TGG</u> TGGGGCAC ^b	
Antisense	GTGCCCCA <u>CCA</u> ATCCGGGTAGAG ^b	
ManCW283S		
Sense	CTATCCGGA <u>AGT</u> TGGAGCAAGACC ^b	
Antisense	GGTCTTGCTCCA <u>ACT</u> ATCCGGATAG ^b	

^a Restriction sites are underlined

^b Mutation positions are double-underlined

Supplementary Table S2 ¹H and ¹³C NMR data assignment and molecular mass of β -manno-oligosaccharides produced by transglycosylation catalysed ManC using M4 as donor with isomaltotriose or melezitose as acceptor.

Compound	Chemical shifts (δ, p.p.m.)											
-	H-1	Н-2	H-3	H-4	H-5	H-6a/H-6b						
(molecular mass) ^a	C-1	C-2	C-3	C-4	C-5	C-6						
β -Man <i>p</i> -(1,4)-	4.70	4.01	3.60	3.51	3.39	3.89/3.68						
	_	_	_	_	_	_						
β -Manp-(1,4)-	4.70	4.08	3.77	3.77	3.51	3.87/3.71						
	_	_	_	_	_	_						
β -Manp-(1.4)-	4.70	4.08	3.77	3.77	3.51	3.87/3.71						
	_	_	_	_	_	_						
β -Manp-(1.4)-	4.70	4.08	3.77	3.77	3.51	3.87/3.71						
r r ())	_	_	_	_	_	_						
β-Man <i>p</i>	5.13	3.94	3.85	n.d.	n.d.	n.d.						
r ··· r	_	_	_	_	_	_						
α-Man <i>p</i>	4.86	n.d.	n.d.	n.d.	n.d.	n.d.						
(m/z, 851)	_	_	_	_	_	_						
β -Man p -(1.4)-	4.79	4.14	3.82	n.d.	n.d.	n.d.						
p 1.1p (1,1)	101.2	71.0	72.4	n.d.	n.d.	n.d.						
β -Man p -(14)-	4.75	4.08	3.68	3.58	n.d.	n.d.						
p 1.1p (1,1)	101.3	71.5	n.d.	76.1	n.d.	n.d.						
α -Glc <i>p</i> -(1.6)-	4.98	3.62	3.91	3.71	3.85	3.78/3.86						
or or p (1,0)	98.8	72.1	72.7	79.7	n.d.	61.2						
α -Glc <i>p</i> -(1.6)-	4.98	3.60	3.75	3.52	3.93	3.79/3.98						
a onep (1,0)	98.8	72.3	n.d.	n.d.	71.2	66.7						
α-Glc <i>p</i>	5.26	3.56	n.d.	n.d.	n.d.	n.d.						
	93.2	n.d.	n.d.	n.d.	n.d.	n.d.						
β-Glc <i>p</i>	4.69	3.28	3.49	n.d.	n.d.	n.d.						
(m/z 851)	97.2	75.0	77.1	n.d.	n.d.	n.d.						
β-Man <i>p</i> -(1.4)-	4.75	4.12	3.81	3.55	3.55	3.97/n.d.						
	100.1	69.7	71.4	66.7	n.d.	n.d.						
β -Manp-(1,4)-	4.72	4.05	3.65	3.56	3.81	n.d.						
	100.1	n.d.	n.d.	74.9	n.d.	n.d.						
α -Glcp-(1.3)-	5.18	3.57	3.72	3.43	3.89	3.75/3.89						
1 1 7-7	100.3	71.3	72.7	76.3	72.4	60.3						
β-Fru <i>f</i> -(2,1)-	?	-	4.31	4.32	4.06	3.99/4.12						
• • • • • •	n.d.	103.3	82.9	73.5	79.7	70.3						
α-Glc <i>p</i>	5.42	3.53	3.64	3.41	3.92	3.75/3.91						
(m/z 851)	91.7	74.8	72.7	69.4	72.0	60.3						

^a molecular mass as Na⁺ adduct (estimated mass + 23 m/z) determined by MALDI-TOF MS n.d., not detected.

Enzyme and Protein Chemistry (EPC) works within protein biochemistry, carbohydrate biochemistry, molecular biology, microbiology and plant biochemistry. The main activities of EPC are related to food and raw materials for food, but the methods and main strategies are relevant to various biotechnological issues. The aim is to explain the molecular mechanisms and interactions relevant for functionality and quality of foodstuffs and raw materials and to identify the biochemical mechanisms connected to health promoting and nutrition related effects of food.

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DTU

Production and Characterisation of Glycoside Hydrolases from GH3, GH5, GH10, GH11 and GH61 for Chemo-Enzymatic Synthesis of Xylo- and Mannooligosaccharides



Enzyme and Protein Chemistry Department of Systems Biology

