(19)	Europäisches Patentamt European Patent Office Office européen des brevets		(11) <b>E</b>	EP 2 311 841 A1
(12)	EUROPEAN PA published in accord	TENT A	APPLICATION h Art. 153(4) EPC	
<ul> <li>(43) Date of publication: 20.04.2011 Bulletin 2011/16</li> <li>(21) Application number: 09742185.3</li> </ul>		(51)	Int Cl.: C07D 493/08 <sup>(2006.01)</sup> C07D 307/20 <sup>(2006.01)</sup> C07C 69/60 <sup>(2006.01)</sup>	C07D 493/10 <sup>(2006.01)</sup> C07D 317/30 <sup>(2006.01)</sup> C07D 303/40 <sup>(2006.01)</sup>
(22) Date of filing: <b>05.05.2009</b>		(86) (87)	International application number: PCT/ES2009/070140 International publication number: WO 2009/135979 (12.11.2009 Gazette 2009/46)	
<ul> <li>(84) Designated Contracting States:</li> <li>AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HR HU IE IS IT LI LT LU LV MC MK MT NL NO PL PT RO SE SI SK TR Designated Extension States:</li> <li>AL BA RS</li> </ul>		R (72)	<ul> <li>(72) Inventors:</li> <li>NOHEDA MARÍN, Pedro E-28006 Madrid (ES)</li> <li>LOZANO GORDILLO, Luis Miguel E-28050 Madrid (ES)</li> </ul>	
<ul> <li>(30) Priority: 06.05.2008 ES 200801303</li> <li>(71) Applicant: Consejo Superior De Investigaciones Científicas 28006 Madrid (ES)</li> </ul>		s	Avenida de Burgos 16 Edificio Euromor 28036 Madrid (ES)	SD

# (54) METHOD FOR OBTAINING ZARAGOZIC ACID AND DERIVATIVES THEREOF

(57) The present invention is aimed at a process for obtaining zaragozic acid and derivatives thereof, to the intermediate compounds of this synthesis and to the use

of these intermediate compounds in the preparation of zaragozic acid.

Printed by Jouve, 75001 PARIS (FR)

#### Description

#### Field of the Invention

<sup>5</sup> **[0001]** The present invention relates to a process for the synthesis of zaragozic acid and derivatives thereof, and to intermediates of said synthesis. It also relates to the use of said intermediates in the synthesis of zaragozic acid.

#### Background of the Invention

10 [0002] Zaragozic acids [Nadin, A.; Nicolaou, K. C. Angew. Chem. Int. Ed. Engl. 1996, 35, 1622-1656] are a family of natural products, of which the first to be isolated was zaragozic acid A (IA, Scheme 1). It was simultaneously isolated between the years 1991 and 1992 by three independent groups: the Merck Group, which called it zaragozic acid A, whereas the Glaxo and Tokyo Noko University/Mitsubishi Kasei Corporation groups called it squalestatin S 1.



20



Scheme 1: Zaragozic acid A (IA)

25

30

**[0003]** WO 93/16066 and WO 93/17557 describe the isolation of derivatives of zaragozic acids from different fungi cultures. Likewise, the chemical modification of the compounds obtained and their usefulness as cholesterol level-reducing agents are shown. WO 94/04144 also discloses various analogs of zaragozic acids, as well as their squalene synthase enzyme inhibitory activity.

**[0004]** Zaragozic acids have in their structure a common bicyclic grouping, 2,8-dioxabicyclo[3.2.1]octane (rings **A** and **B**, Scheme 1), which has 6 consecutive stereocenters (carbons C3, C4, C5, C6, C7 and C1), three of which are quaternary (carbons C1, C4 and C5). The structural differences between the members of the family of zaragozic acids lies in the different  $R_y$  and  $R_x$  substituents (see Scheme 2) that they have in the C6 and C1 positions, respectively, of the common bicyclic backbone.

35 bicyclic backbon

40

45



50

55

[0005] Due to their significant biological activity, as agents useful in cholesterol reduction, and to their high structural complexity, zaragozic acids have caught the attention of a number of research groups. Three total syntheses of zaragozic acid A (IA) have been described in the literature to date. One was developed by Dr. Nicolaou [a) Nicolaou, K. C.; Yue, E. W.; Naniwa, Y.; De Riccardis, F.; Nadin, A.; Leresche, J. E.; La Greca, S.; Yang, Z. Angew. Chem. Int. Ed. Engl. 1994, 33, 2184-2187; b) Nicolaou, K. C.; Nadin, A.; Leresche, J. E.; La Greca, S.; Tsuri, T.; Yue, E. W.; Yang, Z. Angew. Chem. Int. Ed. Engl. 1994, 33, 2187-2190; c) Nicolaou, K. C.; Nadin, A.; Leresche, J. E.; Yue, E. W.; La Greca, S. Angew.

Scheme 2

Chem. Int. Ed. Engl. 1994, 33, 2190-2191; d) Nicolaou, K. C.; Yue, E. W.; La Greca, S.; Nadin, A.; Yang, Z.; Leresche, J. E.; Tsuri, T.; Naniwa, Y. De Riccardis, F. Chem. Eur. J. 1995, 1, 467-494; e) Nicolaou, K. C.; Sorensen, E. J. Classics in Total Synthesis; VCH Publishers: New York, 1996. pp.: 673-709.], another one by Dr. Heathcock [a) Stoermer, D.; Caron, S.; Heathcock, C. H. J. Org. Chem. 1996, 61, 9115-9125; b) Caron, S.; Stoermer, D.; Mapp, A. K.; Heathcock,

- 5 C. H. J. Org. Chem. 1996, 61, 9126-9134]; and one by Dr. Tomooka [Tomooka, K.; Kikuchi, M.; Igawa, K.; Suzuki, M.; Keong, P. -H.; Nakai, T. Angew. Chem. Int. Ed. 2000, 39, 4502-4505]. Five total syntheses have been developed for zaragozic acid C (IC), by the groups of Dr. Carreira, [a) Carreira, E. M.; Du Bois, J. J. Am. Chem. Soc. 1994, 116, 10825-10826; b) Carreira, E. M.; Du Bois, J. J. Am. Chem. Soc. 1995, 117, 8106-8125], Dr. Evans, [Evans, D. A.; Barrow, J. C.; Leighton, J. L.; Robichaud, A. J.; Sefkow, M. J. Am. Chem. Soc. 1994, 116, 12111-12112], Dr. Armstrong [a)
- 10 Armstrong, A.; Jones, L. H.; Barsanti, P. A. Tetrahedron Lett. 1998, 39, 3337-3340; b) Armstrong, A.; Barsanti, P. A.; Jones, L. H.; Ahmed, G. J. Org. Chem. 2000, 65, 7020-7032] and two by Dr. Hashimoto [a) Kataoka, O.; Kitagaki, S.; Watanabe, N.; Kobayashi, J.; Nakamura, S.; Shiro, M.; Hashimoto, S. Tetrahedron Lett. 1998, 39, 2371-2374; b) Nakamura, S.; Hirata, Y.; Kurosaki, T.; Anada, M.; Kataoka, O.; Kitagaki, S.; Hashimoto, S. Angew. Chem. Int. Ed. 2003, 42, 5351-5355; a) Sato, H.; Nakamura, S.; Watanabe, N.; Hashimoto, S. Synlett 1997, 451-454; b) Nakamura, S.; Sato,
- 15 H.; Hirata, Y.; Watanabe, N.; Hashimoto, S. Tetrahedron 2005, 61, 11078-11106; c) Nakamura, S. Chem. Pharm. Bull. 2005, 53, 1-10]).

[0006] On the other hand, the Johnson group prepared the common backbone of zaragozic acids from cycloheptatriene [Xu, Y.; Johnson, C. R. Tetrahedron Lett. 1997, 38, 1117-1120].

[0007] All of them have drawbacks when they are applied due to the large number of steps that they require, the low 20 yield and the large number of protecting groups that they require. It is therefore necessary to provide a process for obtaining zaragozic acid and derivatives thereof, which preferably meets the following characteristics:

- using a small number of protecting groups;
- feasible and inexpensive commercial starting substrates;
- a reasonable number of steps;
  - non-sophisticated experimental processes; or
  - a good final yield.

#### Summary of the Invention

#### 30

35

25

[0008] It has now been found that by following the synthetic sequence of the invention it is possible to obtain compounds of formula (I) from compounds of formula (II) or (III) in a reduced number of steps and with a high yield. Said compounds of formula (I) are versatile intermediates in the preparation of zaragozic acid and derivatives thereof of formula (XXVI), because they contain the bicyclic structure 2,8-dioxabicyclo[3.2.1]octane, common to all zaragozic acids, which allows, through this common structure, obtaining the different derivatives of known zaragozic acid.

[0009] Therefore, a first aspect of the present invention is aimed at a process for obtaining a compound of formula (I), its enantiomers or mixtures thereof, characterized in that it comprises reacting in acidic medium a compound of formula (II), its enantiomers or mixtures thereof, or a compound of formula (III), its enantiomers or mixtures thereof, or a mixture of compounds of formula (II) and (III).

40 [0010] Additional aspects of the invention are aimed at compounds of formula (I), (II), (II), (IV), (V), (VI), (VII), (VII), (VIIb), (VIII), (VIIa), (VIIb), (IX), (IXa), (IXb) (X), (XI), (XII), (XIV) and (XV), their stereoisomers, especially enantiomers, or mixtures thereof, as well as to processes for obtaining them.

[0011] An additional aspect of the present invention is aimed at a process for preparing zaragozic acid and derivatives thereof of formula (XXVI), their stereoisomers, especially enantiomers, or mixtures thereof, characterized in that it

45 comprises the following steps

> (i) reacting in acidic medium a compound of formula (II), its enantiomers or mixtures thereof, or a compound of formula (III), its enantiomers or mixtures thereof, or a mixture of compounds of formula (II) and (III), to obtain a compound of formula (I), its enantiomers or mixtures thereof;

- 50 (ii) hydrolyzing in basic medium the ester groups of said compound of formula (I), its enantiomers or mixtures thereof, to provide a compound of formula (XXV), its enantiomers or mixtures thereof; and (iii) reacting said compound of formula (XXV), its enantiomers or mixtures thereof, with a compound of formula (XXII) in the presence of a base, to obtain a compound of formula (XXVI), its enantiomers or mixtures thereof.
- 55 [0012] An additional aspect of the present invention is aimed at the use of a compound of formula (I), (II), (II), (IV), (V), (VI), (VII), (VIIa), (VIIb), (VIII), (VIIa), (VIIb), (IX), (IXa), (IXb), (X), (XI), (XII), (XIV), (XV) and/or (XVI), its stereoisomers, especially enantiomers, or mixtures thereof, for the synthesis of zaragozic acid and derivatives thereof, of formula (XXVI), their stereoisomers, especially enantiomers, or mixtures thereof.

#### Detailed Description of the Invention

[0013] A first aspect of the present invention is aimed at a process for obtaining a compound of formula (I), its enantiomers or mixtures thereof

5

10



15

#### wherein

 $R^2$  is selected from the group consisting of  $C_1$ - $C_{20}$  alkyl and  $C_1$ - $C_{20}$  alkenyl, which are unsubstituted or substituted in any position with at least one group which is selected from the group consisting of C1-C4 alkyl, C1-C3 alkylidene, C1-C3 alkylcarboxyhydroxyl, hydroxyl and protected hydroxyl; and/or substituted with a group in the end position of the chain

20 which is selected from the group consisting of C6-C10 aryl, mono- or bicyclic heteroaryl with 5- or 6-members in each ring, which can be unsubstituted or substituted with at least one group which is selected from the group formed by C1-C3 alkyl, or halogen; and

 $R^3$ ,  $R^4$  and  $R^5$  are independently selected from the group of  $C_1$ - $C_3$  alkyls; characterized in that it comprises reacting in acidic medium a compound of formula (II), its enantiomers or mixtures thereof, or a compound of formula (III), its enantiomers or mixtures thereof, or a mixture of compounds of formula (II) and (III)

25



35

30

wherein

R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as defined above; and  $R^6$  is a  $C_1$ - $C_3$  alkyl group.

- [0014] According to a preferred embodiment, R<sup>3</sup> and R<sup>4</sup> are identical, more preferably R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are identical, 40 preferably methyl. The compounds of formula (II) and (III) already contain all the stereocenters of zaragozic acid and its derivatives. Without wishing to be bound by theory, it seems that first the OR<sup>6</sup> group is lost with the simultaneous or subsequent formation of an oxonium ion and subsequent rearrangement to form a compound of formula (I). Therefore, the formation of the compounds of formula (I) is independent of the stereochemistry in the acetalic position of the
- 45 tetrahydrofuran ring of the compounds of formula (II) and (III). [0015] According to a preferred embodiment, the acidic medium comprises the addition of an inorganic protic acid, for example, HCl, H<sub>2</sub>SO<sub>4</sub> or HNO<sub>3</sub>. The acidic medium is preferably a diluted acidic medium, preferably at a concentration in volume with respect to the total volume of the reaction comprised between 0.1 and 20%, more preferably between 0.5 and 10%, more preferably between 1 and 5%. According to another preferred embodiment, the solvent is an alcohol
- 50 of formula R<sup>6</sup>OH. According to another preferred embodiment, the reaction is performed at a temperature comprised between 0 and 100°C, more preferably between 25 and 90°C, more preferably between 50 and 80°C. The reaction is preferably performed in a closed vessel (for example, a Kimble ®). [0016] Additional aspects of the present invention are the compounds of formula (II) and of formula (III), their enanti
  - omers or mixtures thereof. [0017] An additional aspect of the present invention is aimed at a process for the synthesis of a compound of formula
- 55 (II), its enantiomers or mixtures thereof, or of a compound of formula (III), its enantiomers or mixtures thereof, characterized in that it comprises the dihydroxylation of a compound of formula (IV), its enantiomers or mixtures thereof



5

10

wherein  $R^2$ ,  $R^3$ ,  $R^4$  and  $R^5$  are as defined above; and  $R^6$  is a  $C_1$ - $C_3$  alkyl group.

- [0018] Therefore, in two synthetic steps from a compound of formula (IV), the creation of the nucleus of zaragozic acid and its derivatives, by means of simple reactions (dihydroxylation and acidic medium) and a high or acceptable yield, and with the additional advantage of not needing additional protecting groups, has been achieved. The dihydroxylation of the compounds of formula (IV) proceeds, followed in the same reaction medium by a rearrangement without needing to form the diol in a previous step, its subsequent protection, and subsequent deprotection prior to the cyclization. [0019] The dihydroxylation reaction is a reaction widely used in the synthesis of organic molecules and can be per-
- formed under conditions known by the skilled person, as described in Smith, M. B.; March, J. March's Advanced Organic Chemistry; John Wiley & Sons: New York, 2001. pp.: 1048-1051. According to a preferred embodiment, the dihydroxylation is performed in the presence of osmium tetroxide/N methylmorpholine-N-oxide or potassium permanganate. More preferably, the hydroxylation is performed in the presence of RuCl<sub>3</sub>/NalO<sub>4</sub> (for conditions useful for performing this transformation, see a) Shing, T. K. M.; Tai, V. W.-F.; Tam, E. K. W. Angew. Chem. Int. Ed. Engl. 1994, 33, 2312-2313;
   b) Shing, T. K. M.; Tai, V. W.-F.; Tam, F. K. W.; Chung, I. H. F.; Jiang, Q. Chem, Fur, J. 1996, 2, 50-57; or c) Plietker.

b) Shing, T. K. M.; Tai, V. W.-F.; Tam, E. K. W.; Chung, I. H. F.; Jiang, Q. Chem. Eur. J. 1996, 2, 50-57; or c) Plietker, B.; Niggemann, M. Org. Lett. 2003, 5, 3353-3356).
[0020] According to a preferred embodiment, the process comprises dihydroxylating in the presence of RuCl<sub>3</sub> a compound of formula (IVa), its enantiomers or mixtures thereof, to yield a compound of formula (II), its enantiomers or

30

mixtures thereof

35

40

wherein  $R^2$ ,  $R^3$ ,  $R^4$  and  $R^5$  are as defined above; and  $R^6$  is a  $C_1$ - $C_3$  alkyl group.

[0021] According to another preferred embodiment, the process comprises dihydroxylating in the presence of RuCl<sub>3</sub>
 a compound of formula (IVb), its enantiomers or mixtures thereof, to yield a compound of formula (III), its enantiomers or mixtures thereof

(IVa)

50



55

wherein

 $\mathsf{R}^2,\,\mathsf{R}^3,\,\mathsf{R}^4$  and  $\mathsf{R}^5$  are as defined above; and

 $R^6$  is a  $C_1$ - $C_3$  alkyl group.

**[0022]** Therefore, another aspect of the present invention is aimed at compounds of formula (IV), (IVa) or (IVb) their stereoisomers, especially their enantiomers, or mixtures thereof as defined above, which are intermediates which allow obtaining the compounds of formula (I), their enantiomers or mixtures thereof, and therefore, also zaragozic acid and

#### its derivatives of formula (XXVI).

**[0023]** An additional aspect of the present invention is aimed at a process for the synthesis of a compound of formula (IV), its stereoisomers, especially its enantiomers, or mixtures thereof, comprising the treatment in acidic medium of a compound of formula (V), its stereoisomers, especially its enantiomers, or mixtures thereof

10

5

15



wherein

 $\mathsf{R}^2,\,\mathsf{R}^3,\,\mathsf{R}^4$  and  $\mathsf{R}^5$  are as defined above; and

- <sup>25</sup> each of  $\mathbb{R}^7$  and  $\mathbb{R}^8$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_4$  alkyl and  $C_6$ - $C_{10}$  aryl; or  $\mathbb{R}^7$ and  $\mathbb{R}^8$ , together with the carbon atom to which they are attached, form a  $C_2$ - $C_7$  alkylidene group. **[0024]** The conditions under which this process is performed are the typical ones for the elimination of acetals, which in this case is surprisingly accompanied by the concomitant cyclization to form the acetalic tetrahydrofuran ring charac-
- teristic of the compounds of formula (IV). Conditions useful for obtaining this transformation can be found in a) Lu, W.;
   Zheng, G.; Cai, J. Tetrahedron 1999, 55, 4649-4654; or b) Greene, T. W.; Wuts, P. G. M. Greene's Protective Groups in Organic Synthesis; John Wiley & Sons: Hoboken, 2007. pp.: 306-321, which are incorporated by reference. According to a preferred embodiment, the reaction is performed in the presence of para-toluenesulfonic acid (p-TsOH). According to another preferred embodiment, the solvent is an alcohol of formula R<sup>6</sup>OH more preferably also of formula R<sup>4</sup>OH.
- [0025] Therefore, another aspect of the present invention is aimed at a compound of formula (V) its stereoisomers, especially its enantiomers, or mixtures thereof as defined above, intermediates useful for the synthesis of zaragozic acid and its derivatives of formula (XXVI).

**[0026]** An additional aspect of the present invention relates to a process for the synthesis of a compound of formula (V), its stereoisomers, especially its enantiomers, or mixtures thereof, **characterized in that** it comprises

- 40 (i) the removal of the trialkylsilyl group of a compound of formula (VII), its stereoisomers, especially its enantiomers, or mixtures thereof, to yield a compound of formula (VI), its stereoisomers, especially its enantiomers, or mixtures thereof
- 45





55 wherein

 $R^2$ ,  $R^3$ ,  $R^5$ ,  $R^4$ ,  $R^7$  and  $R^8$  are as defined above; and

R<sup>9</sup> is a trialkylsilyl group; and

(ii) the oxidation of the hydroxyl group of a compound of formula (VI), its stereoisomers, especially its enantiomers,



#### or mixtures thereof.

**[0027]** As can be observed, step (i) comprises the removal of the trialkylsilyl group, i.e., the transformation of R<sup>9</sup> into a hydrogen. Said transformation is normally understood as a deprotection and can be performed under different conditions (for example, see Kocienski, P. J. Protecting Groups; Thieme: Stuttgart, 2000. pp.: 188-230). According to a preferred

<sup>5</sup> (for example, see Kocienski, P. J. Protecting Groups; Thieme: Stuttgart, 2000. pp.: 188-230). According to a preferred embodiment, the trialkylsilyl group is removed from a compound of formula (VII) to give rise to a compound of formula (VI) in diluted acidic medium, such as 1% HCl for example.
(VI) in diluted acidic medium, such as 1% HCl for example.

**[0028]** According to a preferred embodiment, step (ii) is performed in the presence of PCC or IBX, preferably IBX. Suitable conditions are described in a) Frigerio, M.; Santagostino, M.; Sputore, S.; Palmisano, G. J. Org. Chem. 1995,

- 60, 7272-7276; b) Frigerio, M.; Santagostino, M. Tetrahedron Lett. 1994, 35, 8019-8022; c) Corey, E. J.; Palani, A. Tetrahedron Lett. 1995, 36, 3485-3488; d) Wirth, T. Angew. Chem. Int. Ed. 2001, 40, 2812-2814. Preferably, ethyl acetate is used as a solvent and in the workup of the reaction the excess reagent, as well as the derivative byproducts thereof, can be removed by means of filtration once the reaction has concluded. (More, J. D.; Finney, N. S. Org. Lett. 2002, 4, 3001-3003).
- <sup>15</sup> **[0029]** Therefore, another aspect of the present invention is aimed at compounds of formula (VI) or (VII), their stereoisomers, especially their enantiomers, or mixtures thereof as defined above, which are intermediates which allow obtaining zaragozic acid and its derivatives of formula (XXVI).

**[0030]** An additional aspect of the present invention is aimed at a process for the synthesis of a compound of formula (VII), its stereoisomers, especially its enantiomers, or mixtures thereof, which comprises reacting a compound of formula (VIII), its enantiomers, or mixtures thereof, with a compound of formula (XX)

20 (VIII), its enantiomers, or mixtures thereof, with a compound of formula (XX)





30

35

### wherein

 $R^2$ ,  $R^3$ ,  $R^5$ ,  $R^4$ ,  $R^7$ ,  $R^8$  and  $R^9$  are as defined above; and

each of the Ar groups is independently selected from among C<sub>6</sub>-C<sub>10</sub> aryl groups.

**[0031]** The preparation of the ylide of formula (XX) can be performed according to conditions known in the state of the art (Villa, M. J.; Warren, S. J. Chem. Soc. P. T 1 1994, 12, 1569-1572) or be commercially purchased. According to a preferred embodiment, said ylide is [(methoxycarbonyl)methylene]triphenylphosphorane.

- [0032] Both the compounds of formula (VII) and the compounds of formula (VIII) are useful for obtaining zaragozic acid independently of the configuration of carbon C7. The configuration of said tertiary hydroxyl can be inverted, for example, according to the conditions described in Shi, Y. -J.; Hughes, D.L.; McNamara, J.M. Tetrahedron Lett. 2003, 44, 3609-3611; or Mukaiyama, T.; Shintou, T.; Fukumto, K. J. Am. Chem. Soc. 2003, 125, 10538-10539). Therefore, according to a preferred embodiment, the compounds of formula (VII), their stereoisomers, especially their enantiomers, or mixtures thereof, are compounds of formula (VIIa) or (VIIb), their stereoisomers, especially their enantiomers, or
- 45 mixtures thereof

50



**[0033]** According to another preferred embodiment, the compounds of formula (VIII), their stereoisomers, especially their enantiomers, or mixtures thereof, are compounds of formula (VIIIa) or (VIIIb), their stereoisomers, especially their enantiomers, or mixtures thereof



- 20 [0034] According to a preferred embodiment, a compound of formula (VIIb), its stereoisomers, especially its enantiomers, or mixtures thereof, is transformed into a compound of formula (VIIa), its stereoisomers, especially its enantiomers, or mixtures thereof, by means of a Mitsunobu configuration inversion reaction. Said transformation is preferably performed under the conditions described in Mukaiyama, T.; Shintou, T.; Fukumto, K. J. Am. Chem. Soc. 2003, 125, 10538-10539, which is incorporated in its entirety by reference, more preferably in the presence of another chlorodiphenylphosphine and a carboxylic acid. According to another preferred embodiment, said transformation is performed under the conditions described in Shi, Y. -J.; Hughes, D.L.; McNamara, J.M. Tetrahedron Lett. 2003, 44, 3609-3611, which is incorporated in its entirety by reference, more preferably in the presence of diethyl azodicarboxylate (DEAD)/triphenylphosphine (TPP).
  - **[0035]** Preferably, said compound of formula (VIII), its enantiomers, or mixtures thereof, is obtained by oxidizing, preferably in the presence of IBX, a compound of formula (IX), its enantiomers, or mixtures thereof

30

35



wherein

 $\mathsf{R}^2,\,\mathsf{R}^3,\,\mathsf{R}^4,\,\mathsf{R}^5,\,\mathsf{R}^7,\,\mathsf{R}^8$  and  $\mathsf{R}^9$  are as defined above.

[0036] The oxidation with IBX, is preferably performed in the presence of ethyl acetate, which allows directly subjecting the compounds of formula (VIII) obtained to the following reaction step, without needing to purify them. Therefore, according to a preferred embodiment, the transformation of a compound of formula (IX), its enantiomers, or mixtures thereof, into a compound of formula (VII), its stereoisomers, especially its enantiomers, or mixtures thereof, is performed without isolating said compound of formula (VIII).

**[0037]** Therefore, another aspect of the present invention is aimed at compounds of formula (VIII) or (IX), their enantiomers, or mixtures thereof as defined above, which are intermediates which allow obtaining zaragozic acid and its derivatives of formula (XXVI).

**[0038]** According to a preferred embodiment, the compound of formula (IX), its enantiomers or mixtures thereof, is a compound of formula (IXa) or (IXb), its enantiomers or mixtures thereof:

55





10

5

wherein

R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup>, R<sup>5</sup>, R<sup>7</sup>, R<sup>8</sup> and R<sup>9</sup> are as defined above.

[0039] An additional aspect of the present invention is aimed at a process for the synthesis of a compound of formula (IX), its enantiomers, or mixtures thereof, characterized in that it comprises the dihydroxylation, preferably in the presence of OsO<sub>4</sub>, of a compound of formula (X), its enantiomers, or mixtures thereof







15





#### wherein

R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup>, R<sup>7</sup>, R<sup>8</sup> and R<sup>9</sup> are as defined above.

[0040] The dihydroxylation of the compounds of formula (X) can proceed by attack on the alpha or beta side, giving 30 rise to two possible compounds of formula (IX), their enantiomers or mixtures thereof, specifically compounds of formula (IXa) or (IXb), their enantiomers or mixtures thereof, mentioned above.

[0041] Both compounds are useful for the purposes of the present invention since, as has been seen above, it will then be possible to invert the configuration of carbon C7 in compounds of formula (VII) or (VII) by means of a Mitsunobu reaction.

- 35 [0042] Alternatively, said hydroxylation can be performed in an enantioselective manner. In this case, if the starting material is a racemate, one of the possible diastereoisomers would preferably be generated. It would therefore be a kinetic resolution of the starting racemate. See: Kolb, H. C.; Van Nievwenhze, M. S.; Sharpless, K. B. Chem. Rev. 1994, 94, 2483. Therefore, another aspect of the present invention is aimed at a compound of formula (X) its enantiomers, or mixtures thereof as defined above, which are intermediates which allow obtaining zaragozic acid and its derivatives of
- 40 formula (XXVI).

[0043] An additional aspect of the present invention is aimed at a process for the synthesis of a compound of formula (X), its enantiomers, or mixtures thereof, characterized in that it comprises the acetalization or hemiacetalization of a compound of formula (XI), its enantiomers, or mixtures thereof

45



50

55

in the presence of a compound of formula (R7)(R8)C=O or hydrates or aryl or alkyl acetals or hemiacetals thereof wherein

[0044]  $R^2$ ,  $R^3$ ,  $R^4$ ,  $R^7$ ,  $R^8$  and  $R^9$  are as defined above.

The protection of the 1,2-diol group of a compound of formula (XI) as an acetal or hemiacetal can be performed [0045] by following methods known in the state of the art and allows maintaining this group stable throughout the synthesis.



For conditions for the protection of 1,2-diols useful in the present invention see a) Konno, H.; Makabe, H.; Tanaka, A.; Oritani, T. Tetrahedron 1996, 52, 9399-9408; or b) Greene, T. W.; Wuts, P. G. M. Greene's Protective Groups in Organic Synthesis; John Wiley & Sons: Hoboken, 2007. pp.: 306-321, which are incorporated by reference. According to a preferred embodiment,  $R_7$  and  $R_8$  are methyl, or hydrogen or phenyl; or together with the carbon to which they are

- <sup>5</sup> attached form a cyclohexane or cyclopentane ring. According to another preferred embodiment,  $R_7$  is methyl and  $R_8$  is phenyl. The compounds of formula ( $R_7$ )( $R_8$ )C=O can be used in ketone form or in acetal or hemiacetal form. According to another preferred embodiment, a compound of formula (XI), its enantiomers, or mixtures thereof, is reacted with 2,2dimethoxypropane in the presence of catalytic amounts of acid, preferably para-toluenesulfonic acid.
- [0046] Therefore, another aspect of the present invention is aimed at a compound of formula (XI) its enantiomers, or mixtures thereof as defined above, which are intermediates which allow obtaining zaragozic acid and its derivatives of formula (XXVI).

**[0047]** An additional aspect of the present invention is aimed at a process for the synthesis of a compound of formula (XI), its enantiomers, or mixtures thereof, **characterized in that** it comprises the following steps

(i) reacting a compound of formula (XXI) in the presence of a compound of formula PY<sub>3</sub>, and subsequently adding a compound of formula (XVI), its stereoisomers or mixtures thereof, to yield a compound of formula (XV), its stereoisomers or mixtures thereof



(ii) epoxidizing, preferably with meta-chloroperbenzoic acid (m-CPBA), said compound of formula (XV), to obtain a compound of formula (XIV), its stereoisomers, especially its enantiomers, or mixtures thereof,



40

45



<sup>50</sup> (iv) introducing a trialkylsilyl group into said compound of formula (XII) to obtain a compound of formula (XII), its stereoisomers, especially its enantiomers, or mixtures thereof



- 5
- 10 and
  - (v) dihydroxylating said compound of formula (XII) wherein  $R^2$ ,  $R^3$ ,  $R^4$  and  $R^9$  are as defined above; and each of the Y groups is independently selected from among  $C_6$ - $C_{10}$  aryl groups or  $C_1$ - $C_6$  alkyl groups.
- <sup>15</sup> [0048] The preparation of compounds of formula (XV) (step (i)) can be performed according to processes described in the state of the art, and it has been performed at a multigram scale (Maryanoff, B. E.; Reitz, A. B. Chem. Rev. 1989, 89, 863-927; Trost, B. M.; Melvin, L. S. Jr. J. Am. Chem. Soc. 1976, 98, 1204-1212.). According to a preferred embodiment, the compound of formula PY<sub>3</sub> is n-Bu<sub>3</sub>P. The compounds (XXI) and (XVI) can be commercially purchased or repaired according to established processes. The compounds of formula (XXI) are fumarate esters, preferably dimethyl fumarate.
- It is possible to purchase different fumarates, for example dimethyl or diisobutyl fumarates, among others.
   [0049] Various compounds of formula (XVI) are also easily obtainable. Other compounds of formula (XVI) not commercially obtainable can be prepared according to methods similar to those described in, for example, a) Evans, D. A.; Barrow, J. C.; Leighton, J. L.; Robichaud, A. J.; Sefkow, M. J. Am. Chem. Soc. 1994, 116, 12111-12112 (compound 4); b) Tomooka, K.; Kikuchi, M.; Igawa, K.; Suzuki, M.; Keong, P. H.; Nakai, T. Angew. Chem. Int. Ed. 2000, 39, 4502-4505
- (compounds 14, 15 or 16); b) Nicolaou, K. C.; Yue, E. W.; Naniwa, Y.; De Riccardis, F.; Nadin, A.; Leresche, J. E.; La Greca, S.; Yang, Z. Angew. Chem. Int. Ed. Engl. 1994, 33, 2184-2187 (compound 6); or c) Armstrong, A.; Jones, L. H.; Barsanti, P. A. Tetrahedron Lett. 1998, 39, 3337-3340 (compound 6); or d) Evans, et al Tetrahedron Lett. 1993, 34, 8403 (see reference 4b in Carreira, E. M.; Du Bois, J. J. Am. Chem. Soc. 1994, 116, 10825-10826 for the preparation of compound 9), all incorporated herein by reference.
- <sup>30</sup> **[0050]** The epoxidation of the compounds of formula (XV) (step (ii)) can be performed in the presence of epoxidizing agents such as meta-chloroperbenzoic acid, resulting in a compound of formula (XIV) in racemic form. Alternatively, the epoxidation can be performed by means of chiral reagents, giving rise to enantiomerically pure or enantiomerically enriched compounds of formula (XIV), which gives rise to the subsequent intermediates of formula (XIII) to (I) and (XXV) and (XXVI) (defined below) also being obtained in an enantiomerically pure or enantiomerically enriched manner. There-
- <sup>35</sup> fore, the use at this point of a chiral epoxidizing agent allows obtaining zaragozic acid and its derivatives of formula (XXVI) in an enantiomerically pure or enantiomerically enriched manner. Some conditions useful for performing the epoxidation enantiomerically can be found in Jacobsen-Katsuki (see: Katsuki, T. Adv. Synth. Catal. 2002, 344, 131-147) 2); or Shi (see: Wang, Z. X.; Tu, Y.; Frohn, M.; Zhang, J. R; Shi, Y. J. Am. Chem. Soc. 1997, 119, 11224-11235). **IODE 11** The incomprised of the compounde of formula (XIV) (step (iii)) allows appoint the enarging the ena
- [0051] The isomerization of the compounds of formula (XIV) (step (iii)) allows opening the epoxide and isomerizing the double bond to provide a compound of formula (XIII). The base used is preferably DBU (1,8-diazabicyclo[5.4.0] undec-7-ene).

**[0052]** Non-illustrative examples of conditions in which the hydroxyl group of a compound of formula (XIII) (step (iv)) can be protected to obtain a compound of formula (XII) can be found in, for example, Dalla, V.; Catteau, J. P. Tetrahedron 1999, 55, 6497-6510, and the trialkylsilyl groups which can be used in this reaction, as well as reagents suitable for their

<sup>45</sup> introduction and removal, are known for the person skilled in the art (for example, see Greene, T. W.; Wuts, P. G. M. Greene's Protective Groups in Organic Synthesis; John Wiley & Sons: Hoboken, 2007).
 [0053] According to a particular embodiment, the base used is imidazole and the silylating agent is TBDMSCI (tert-butyldimethylsilyl chloride). According to another preferred embodiment, the silylating agent is TBDMSOTf (tert-butyld-imethylsilyl trifluoromethanesulfonate).

- <sup>50</sup> **[0054]** As has been mentioned above, the dihydroxylation (step (v)) can be performed under conditions known by the skilled person, as described in Smith, M. B.; March, J. March's Advanced Organic Chemistry; John Wiley & Sons: New York, 2001. pp.: 1048-1051. According to a preferred embodiment, the dihydroxylation is performed in the presence of osmium tetroxide/*N*-methylmorpholine-*N*-oxide or potassium permanganate.
- [0055] Therefore, additional aspects of the present invention are aimed at compounds of formula (XII), (XII), (XIV) or (XV), their enantiomers, or mixtures thereof as defined above, which are intermediates which allow obtaining zaragozic acid and its derivatives of formula (XXVI).

**[0056]** According to an additional aspect, the present invention relates to a process for preparing zaragozic acid and derivatives thereof of formula (XXVI), their stereoisomers, especially enantiomers, or mixtures thereof, **characterized** 

#### in that it comprises the following steps

(i) reacting in acidic medium a compound of formula (II), its enantiomers or mixtures thereof, or a compound of formula (III), its enantiomers or mixtures thereof, or a mixture of compounds of formula (III) and (III), as defined above; to obtain a compound of formula (I), its enantiomers or mixtures thereof, as defined above;

(ii) hydrolyzing in basic medium the ester groups of said compound of formula (I), its enantiomers or mixtures thereof, to provide a compound of formula (XXV), its enantiomers or mixtures thereof

HO

HO<sub>2</sub>( HO<sub>2</sub>(

OH



5

10

wherein

R<sup>2</sup> is as defined above;

and

(iii) reacting said compound of formula (XXV), its enantiomers or mixtures thereof, with a compound of formula (XXII) in the presence of a base, to obtain a compound of formula (XXVI), its enantiomers or mixtures thereof

(XXV)



20

30













35

40

#### wherein

 $R^1$  is selected from the group consisting of  $C_1$ - $C_{20}$  alkyl or  $C_1$ - $C_{20}$  alkenyl, which are unsubstituted or substituted with at least one group which is selected from the group consisting of C1-C4 alkyl; and/or a group in the end position which is selected from the group consisting of C<sub>6</sub>-C<sub>10</sub> aryl;

Z is selected from the group consisting of hydroxyl and alkoxyl; and R<sup>2</sup> is as defined above.

[0057] The hydrolysis of ester groups to provide the corresponding acid groups is known by the person skilled in the art (see, for example, Kocienski, P.J. Protecting Groups; Thieme: Stuttgart, 2000. pp.: 393-425). Non-limiting conditions for the purposes of the present invention are, preferably, those in which the hydrolysis of the compounds of formula

45 (XXV) is performed in the presence of an alkali or alkaline earth metal hydroxide, for example in the presence of LiOH, NaOH, Ba(OH)<sub>2</sub>, or in the presence of  $Na_2S$ .

[0058] The coupling of the compound of formula (XXII) involves the esterification of the compound of formula (XXV) to provide the compound of formula (XXVI). Suitable conditions for this transformation are known in the state of the art.

- 50 For example, it can be performed according to the conditions described in Carreira, E. M.; Du Bois, J. J. Am. Chem. Soc. 1994, 116, 10825-10826, wherein the compound of formula (XXII) is an acid chloride (Z=CI), described in Stoermer, D.; Caron, S.; Heathcock, C. H. J. Org. Chem. 1996, 61, 9115-9125, or in Evans, D. A.; Barrow, J. C.; Leighton, J. L.; Robichaud, A. J.; Sefkow, M. J. Am. Chem. Soc. 1994, 116, 12111-12112, wherein the compound of formula (XXII) is an acid (Z=OH) and the reaction proceeds in the presence of DMAP.
- 55 [0059] As can be seen, the sequence described in the present invention allows obtaining zaragozic acid and derivatives thereof of formula (XXVI), in a few steps, using reagents which are usual in the synthesis of organic compounds. Said synthesis can be performed by carrying from the start (compounds of formula (XVI)) the complete chain R<sup>2</sup>. Alternatively, it is possible to start the synthesis with a compound of formula (XVI) which does not comprise a complete chain, but

rather a precursor moiety thereof. The sequence can be performed as has been described above, and the complete chain can be constructed at the most convenient time, either by means of a single synthetic step or by means of consecutive or non-consecutive successive steps.

[0060] Therefore, according to a preferred embodiment  $R^2$  is a  $C_1$ - $C_5$  alkyl or alkenyl substituted, preferably in the

- 5 end position, with a hydroxyl group or a protected hydroxyl. At the time of constructing the chain, it will be possible to functionalize said hydroxyl such that it is activated for an alkylation reaction. For example, it is possible to oxidize the alcohol to aldehyde (for example, in the presence of PCC) and then react the resulting compound in the presence of an ylide (or vice versa) or then react with an alkynyl in the presence of a base. For example, see the formation of the compound 30 in Nakamura, S.; Hirata, Y.; Kurosaki, T.; Anada, M.; Kataoka, O.; Kitagaki, S.; Hashimoto, S. Angew.
- 10 Chem. Int. Ed. 2003, 42, 5351-5355; or the formation of the compounds 57 and 58 in Kataoka, O.; Kitagaki, S.; Watanabe, N.; Kobayashi, J.; Nakamura, S.; Shiro, M.; Hashimoto, S. Tetrahedron Lett. 1998, 39, 2371-2374. [0061] For example, different compounds of formula (XVI) can be prepared, wherein  $R^2$  is a  $C_1-C_5$  alkyl or alkenyl substituted in the end position with a hydroxyl group or a protected hydroxyl (alpha,beta-unsaturated aldehydes) from the corresponding hydroxyaldehyde and a suitable vlide (see Scheme 3 - R=hydrogen or protecting group).





#### Scheme 3

The compound of formula (XXXI) ((formylmethyl)triphenylphosphonium

35

40

45

50

chloride) can be obtained, for example, according to methods described in Phytochemistry 1995, 38, 1169-1173, which is incorporated in its entirety by reference. The conditions for reacting a compound of formula (XXXI) and a compound of formula (XXXII) can be found in Tetrahedron 1993, 49, 10643-10654. Said document also teaches how to prepare a compound of formula (XXXII) from a diol (1,4-butanediol).

[0062] As has been indicated above, the compound of formula (XVI) can be constructed from the start with all the functionalities of the final compound of formula (XXVI). For example, reacting the compound of formula (XXX) with the compound 32 described in Carreira, et al., J. Am. Chem. Soc. 1995, 117, 8106-8125 following a process similar to the one shown in Scheme 3 would provide the compound of formula (XVI) necessary for obtaining, for example, zaragozic acid C (see Scheme 4).

OAc



compound of formula (XVI) necessary for obtaning zaragozic acid C



#### Definitions

[0063] For the purpose of facilitating the understanding of the present invention, the meanings of several terms and expressions as they are used in the context of the invention are included herein.

5 [0064] "Alkyl" refers to a radical with a linear or branched hydrocarbon chain which consists of carbon and hydrogen atoms, which does not contain unsaturations and which is attached to the rest of the molecule by means of a single bond, for example, methyl, ethyl, propyl, isopropyl or n-butyl.

[0065] "Alkenyl" refers to a radical with a linear or branched hydrocarbon chain which consists of carbon and hydrogen atoms, which contains at least one unsaturation, and which is attached to the rest of the molecule by means of a single bond, for example, ethenyl, n-propenyl, i-propenyl, n-butenyl, n-pentenyl, etc.

- [0066] "Alkylidene" refers to a radical with a linear hydrocarbon chain which consists of carbon and hydrogen atoms, and which is attached to the rest of the molecule from the two ends by means of single bonds to the same carbon atom, and therefore form a cycle, for example, ethylene (-CH2-CH2-), n-propylene (-CH2-CH2-CH2-), n-butylene (-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>the = $CH_2$  group.
- 15

10

25

45

50

"Halide" or "halogen" means-F, -Cl, -Br or -I; [0067]

[0068] A "stereoisomer" in the present application refers to compounds formed by the same atoms attached by the same sequence of bonds but having different three-dimensional structures which are not interchangeable.

[0069] "Enantiomer" is understood as the mirror image of a stereoisomerically pure compound. For the purposes of 20 the invention, an enantiomer can be considered as a mixture of two enantiomers having an enantiomeric excess greater than 95%, preferably greater than 98%, more preferably greater than 99%, more preferably greater than 99,5%.

[0070] "Heteroaryl" preferably means a fraction of monocyclic or bicyclic hydrocarbon comprising 1 or 2 aromatic nuclei, said nuclei being attached with, and/or covalently linked to one another, al least one of such nuclei containing 1, 2, 3 or 4 heteroatoms independently selected from the group consisting of N, O and S, such as -pirrolyl, - furyl, -thienyl,

-pyridyl, -quinolyl, -isoquinolyl, -indolyl, -oxazolyl, -isoxazolyl, -diazinyl, and the like.

[0071] "Aryl" refers to an aromatic hydrocarbon radical such as phenyl, or naphthyl.

[0072] "Arylalkyl" refers to an aryl group attached to the rest of the molecule through an alkyl group, for example, benzyl ("-(CH<sub>2</sub>)-phenyl" or "Bn").

[0073] "Alkoxyl" refers to a radical of formula -O-R<sup>10</sup>, wherein R<sup>10</sup> represents a group selected from the group consisting 30 of substituted or unsubstituted alkyl, substituted or unsubstituted alkenyl, substituted or unsubstituted aryl, and substituted or unsubstituted arylalkyl.

[0074] "Alkylcaboxyhydroxyl" refers to a radical of formula R<sup>11</sup>(C=O)O-, wherein R<sup>11</sup> is selected from the group consisting of  $C_1$ - $C_6$  alkyl,  $C_1$ - $C_6$  alkenyl,  $C_6$ - $C_{10}$  aryl and  $C_7$ - $C_{15}$  arylalkyl.

- [0075] When the present application refers to a "protected hydroxyl", it indicates a hydroxyl group blocked such that 35 it is inert to determined reactions and which can later be removed under controlled conditions. Said groups are known by the person skilled in the art and the most suitable ones can be selected according to the reactions to which the hydroxyl group is to be inert and/or the conditions under which said protecting group is to be removed, i.e., the conditions under which the hydroxyl group is to be taken off. Examples of suitable hydroxyl protecting groups and the conditions for their removal can be found in reference texts such as for example, Greene and Wuts' "Protective Groups in Organic 40 Synthesis", John Wiley & Sons, Inc., New York, 4th Ed., 2007.
  - **[0076]** Preferred protecting groups for the purposes of the present invention are:
    - silyl derivatives of formula -Si(R<sup>12</sup>)3 (commonly referred to as trialkylsilyls), such as trimethylsilyl ("TMS"), triethylsilyl, tert-butyldimethylsilyl ("TBDMS"), tert-butyldiphenylsilyl, tri-isopropylsilyl, diethylisopropylsilyl, texyldimethylsilyl ether, triphenylsilyl, di-tert-butylmethylsilyl;
    - ethers of formula -R<sup>12</sup>, such as methyl ether, tert-butyl ether, benzyl ether, p-methoxybenzyl ether, 3,4-dimethoxybenzyl ether, trityl ether; allyl ether;
  - alkoxymethyl ethers of formula -CH<sub>2</sub>-O-R<sup>12</sup>, such as methoxymethyl ether, 2-methoxyethoxymethyl ether, benzyloxymethyl ether, p-methoxybenzyloxymethyl ether, 2-(trimethylsilyl)ethoxymethyl ether. The oxygen atom can be replaced by a sulfur atom to form an alkylthiomethyl ether of formula -CH<sub>2</sub>-S-R<sup>12</sup>, such as methylthiomethyl ether. Tetrahydropyranyl ethers and derivatives are also commonly used hydroxyl protecting groups;
    - esters of formula -C(=O)R<sup>12</sup>, acetate, benzoate; pivalate; methoxiacetate; chloroacetate; levulinate;
    - carbonates of formula -C(=O)-O-R<sup>12</sup>, such as benzyl carbonate, p-nitrobenzyl carbonate, tert-butyl carbonate, 2,2,2trichloroethyl carbonate, 2-(trimethylsilyl)ethyl carbonate, or allyl carbonate; or
- sulfates of formula SO<sub>3</sub>-O-R<sup>12</sup> or salts thereof, such as SO<sub>3</sub>·pyridine. 55

[0077] In all the previous formulas, R<sup>12</sup> represents a group selected from the group consisting of substituted or unsubstituted  $C_1$ - $C_{12}$  alkyl, substituted or unsubstituted  $C_1$ - $C_{12}$  alkenyl, substituted or unsubstituted  $C_6$ - $C_{10}$  aryl, and substituted or unsubstituted C7-C15 arylalkyl.

**[0078]** The references of the present document to substituted groups in the compounds of the present invention refer to the specified moiety which can be substituted in one, two or three available positions with one, two, three suitable groups, which are independently selected from the group consisting of cyano; alkanoyl, such as a  $C_1$ - $C_6$  alkanoyl group,

- <sup>5</sup> such as acyl and the like; carboxamido (-(C=O)NH<sub>2</sub>); trialkylsilyl; carbocyclic aryl having 6 or more carbons, particularly phenyl or naphthyl and (C<sub>1</sub>-C<sub>3</sub>)alkylaryl such as tolyl. As a non-limiting example, "substituted alkyl" includes groups such as cyanoethyl, acetylmethyl, carboxamidomethyl (-CH<sub>2</sub>CONH<sub>2</sub>), 2-trimethylsilylethyl, benzyl, diphenylmethyl.
   [0079] In each case, when the number of carbon atoms of the corresponding "Cx-Cy" group is specified, it indicates that the group comprises between "x" and "y" carbon atoms. For example, when "C<sub>1</sub>-C<sub>3</sub> alkyl" is indicated, it refers to
- an alkyl group of one, two or three carbon atoms, i.e., methyl, ethyl, propyl, or isopropyl. For example, when "C<sub>10</sub>-C<sub>15</sub> alkyl" is indicated, it refers to an alkyl group of ten, eleven, twelve, thirteen, fourteen or fifteen carbon atoms, such as decyl, undecyl, dodecyl, tridecyl, tetradecyl or pentadecyl.
   [0080] Unless otherwise indicated, the compounds of the invention also refer to those including compounds which

differ only in the presence of one or more isotopically enriched atoms. For example, the compounds having the present structures, with the exception of the substitution of a hydrogen with a deuterium or with tritium, or the substitution of a carbon with a <sup>13</sup>C- or <sup>14</sup>C-enriched carbon, are within the scope of this invention.

**[0081]** The following examples illustrate different embodiments of the invention and must not be considered as limiting the scope thereof.

20 Examples

15

### General methods and materials

- **[0082]** All the reactions were performed under an argon atmosphere, except those indicated in each case. The solvents used were distilled and dried under an argon atmosphere. The reagents and solvents used are from the companies Aldrich, Fluka, Merck, Sigma, Acros, Lancaster, SDS or Scharlau, and were purified by usual processes when necessary. The purification of the reaction products was performed by column chromatography under pressure (flash chromatography), using 60 Merck silica gel (with a 230-400 mesh particle size) as a stationary phase.
- [0083] The (fully decoupled) <sup>1</sup>H and <sup>13</sup>C nuclear magnetic resonance spectra were performed at room temperature in the solvent indicated in each case (CDCl<sub>3</sub> and CD<sub>3</sub>OD) using the following apparatuses: Varian Gemini-200 (200 MHz), Varian INOVA-300 (300 MHz), Bruker Avance-300 (300 MHz) and Varian INOVA-400 (400 MHz). The values of the chemical shifts are expressed in parts per million (δ, ppm), using as an internal reference the residual signal of the solvent: CDCl<sub>3</sub>, 7.26 ppm (<sup>1</sup>H-NMR) and 77.0 ppm (<sup>13</sup>C-NMR); CD<sub>3</sub>OD, 3.31 ppm (<sup>1</sup>H-NMR) and 49.0 ppm (<sup>13</sup>C-NMR). The <sup>1</sup>H-NMR spectra are described indicating the number of protons and the apparent multiplicity of each signal. The
- <sup>35</sup> coupling constants (J) are the apparent ones and are expressed in Hz. The following abbreviations have been used: s (singlet), d (doublet), t (triplet), c (quadruplet), q (quintuplet) and m (multiplet).
   [0084] The melting points (m.p.) were measured in a Reichert brand Kofler microscope. The infrared (IR) spectra were recorded in the Perkin-Elmer spectrophotometer models 681 and FT-IR Spectrum One. The low resolution mass spectra (LRMS) were recorded: (1) by direct injection of the sample into a Hewlett Packard 5973 MSD spectrophotometer using
- 40 the electron impact (EI) ionization technique; or (2) in a Hewlett Packard LCMS 1100 MSD spectrophotometer (an HPLC-coupled quadrupole analyzer) using the electrospray chemical ionization technique (API-ES) in positive or negative modes. The elemental analyses (E.A.) were performed with the Perkin-Elmer 240C and Heraus CHN-O-Rapid analyzers. [0085] Unless otherwise indicated, all the products shown in the examples are racemic (*rac*).

#### 45 Example 1: Preparation of methyl (3*E*,5*E*)-3-(methoxycarbonyl)-3,5-tridecadienoate (58)

[0086]

50

55



n-Bu<sub>3</sub>P (7.12 g, 35.2 mmoles) was added to a solution of dimethyl fumarate (3.97 g, 27.6 mmoles) and (E)-2-decenal

(3.89 g, 25.2 mmoles) in THF (44 ml). The mixture was stirred at room temperature for 48 hours. After that time, AcOEt (30 ml) and H<sub>2</sub>O (30 ml) were added. The phases were separated, and the aqueous phase was extracted with AcOEt ( $3 \times 25$  ml). The organic phase was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 20:1), obtaining (5.03 g, yield 71%) methyl (3E,5E)-3-(methoxycarbonyl)-3,5-tridecadienoate (**58**), as a transparent oil.

- <sup>5</sup> methyl (3E,5E)-3-(methoxycarbonyl)-3,5-tridecadienoate (58), as a transparent oil.
  IR(NaCl): v 2949, 2928, 2856, 1744, 1714, 1641, 1436, 1324, 1258, 1198, 1170, 1085, 975 cm<sup>-1</sup>.
  <sup>1</sup>H-NMR (200 MHz, CDCl<sub>3</sub>). δ 7.33 (1H, d, *J* = 10.4 Hz, H-4), 6.22 (1H, dt, *J* = 7.0, 15.1 Hz, H-6), 6.20 (1H, dd, *J* = 10.4, 15.1 Hz, H-5), 3.75 (3H, s, -OCH<sub>3</sub>), 3.68 (3H, s, -OCH<sub>3</sub>), 3.43 (2H, s, H-2), 2.18 (2H, m, H-7), 1.40-1.20 (10H, m, -CH<sub>2</sub>-), 0.87 (3H, m, -CH<sub>3</sub>).
- <sup>13</sup>C-NMR (50 MHz,CDCl<sub>3</sub>). δ 171.0, 167.6, 145.8, 141.5, 124.9, 121.5, 51.7, 51.6, 33.1, 32.0, 31.5, 28.9, 28.8, 28.5, 22.4, 13.9.

**LRMS(EI):** *m*/*z* 282(M<sup>+</sup>, 28), 251(16), 222(12), 190(8), 183(100), 166(12), 137(43). **E.A.** (C<sub>16</sub>H<sub>26</sub>O<sub>4</sub>): Found: C, 68.20, H, 9.30; Calculated: C, 68.06, H, 9.28.

#### <sup>15</sup> Example 2: Preparation of methyl *rac*-(*E*,5*S*,6*S*)-5,6-epoxy-3-(methoxycarbonyl)-3-tridecenoate (60)

[0087]

20

25



*m*-CPBA (3.66 g, 21.2 mmoles) was added to a solution of methyl (3E,5E)-3-(methoxycarbonyl)-3,5-tridecadienoate (58) (3.0 g, 10.62 mmoles) in CCl<sub>4</sub>, (120 ml). The mixture was stirred at room temperature for 24 hours. After that time, the solvent was removed under reduced pressure. The crude reaction product was dissolved in AcOEt (50 ml) and washed with sat. NaHCO<sub>3</sub> (10  $\times$  10 ml), dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 7:1), obtaining (1.93 g, yield 61%) methyl *rac*-(*E*,5*S*,6*S*)-5,6-epoxy-3-(methoxycarbonyl)-3-tridecenoate (60), as a transparent oil.

- IR (NaCl): v 3468, 2949, 2857, 1738, 1721, 1655, 1575, 1437, 1315, 1265, 1202, 1173, 1080, 1012, 932, 865, 776 cm<sup>-1</sup>. <sup>35</sup> <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>).  $\delta$  6.55 (1H, d, J = 8.3 Hz, H-4), 3.75 (3H, s, -OCH<sub>3</sub>), 3.70 (3H, s, -OCH<sub>3</sub>), 3.57 (1H, d, J = 16.6 Hz, H-2), 3.44 (1H, d, J = 16.6 Hz, H-2), 3.28 (1H dd, J = 2.1, 8.3 Hz, H-5), 2.97 (1H, dt, J = 2.1, 5.3 Hz, H-6), 1.59 (2H, m, -CH<sub>2</sub>-), 1.44 (2H, m, -CH<sub>2</sub>-), 1.26 (8H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.87 (3H, t, J = 5.1 Hz, -CH<sub>3</sub>). <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>).  $\delta$  170.7, 166.4, 141.8, 131.2, 60.4, 54.3, 52.2, 52.2, 32.3, 31.8, 31.6, 29.2, 29.1, 25.7, 22.5, 14.0.
- 40 LRMS(EI): m/z 298(M<sup>+</sup>, 0), 282(0), 266(1), 237(2), 206(2), 179(3), 170(100), 139(10), 111(80).
   E.A. (C<sub>16</sub>H<sub>26</sub>0<sub>5</sub>): Found: C, 64.25, H, 8.95; Calculated: C, 64.41, H, 8.78.

Example 3: Reaction of methyl rac-(E,5S,6S)-5,6-epoxy-3-(methoxycarbonyl)-3-tridecenoate (60) with DBU

45 **[0088]** 



**[0089]** DBU (4.31 g, 28.35 mmoles) was added to a solution of methyl *rac*-(E,5S,6S)-5,6-epoxy-3-(methoxycarbonyl)-3-tridecenoate **(60)** (5.64 g, 18.90 mmoles) in Et<sub>2</sub>O (190 ml). The mixture was stirred at room temperature for 20 minutes. After that time, Celite was added and the solvent was removed under reduced pressure. The product was purified by a

chromatographic column (hexane/AcOEt, 4:1), obtaining (3.46 g, yield 61%) methyl *rac*-(2*Z*,4*E*,*S*)-6-hydroxy-3-(methoxycarbonyl)-2,4-tridecadienoate **(61-***cis***)** and methyl *rac*-(2*E*,4*E*,*S*)-6-hydroxy-3-(methoxycarbonyl)-2,4-tridecadienoate of **(61-***trans***)** in a 2:1 ratio, respectively, both as a colorless oil.

<sup>5</sup> Methyl *rac*-(2*Z*,4*E*,*S*)-6-hydroxy-3-(methoxycarbonyl)-2,4-tridecadienoate (61-*cis*)

#### [0090]

10

15

20



 $IR(NaCl): v \, 3431, 2949, 2928, 2856, 1738, 1722, 1634, 1613, 1436, 1375, 1275, 1203, 1170, 1152, 1018, 968, 843 \, \rm cm^{-1}.$ 

<sup>1</sup>**H-NMR** (300 MHz, CDCl<sub>3</sub>).  $\delta$  6.32 (1H, d, J = 15.7 Hz, H-4), 5.98 (1H, dd, J = 5.3, 15.7 Hz, H-5), 5.84 (1H, s, H-2), 4.24 (1H, m, H-6), 3.89 (3H, s, -OCH<sub>3</sub>), 3.72 (3H, s, -OCH<sub>3</sub>), 1.84-1.52 (4H, m, -CH<sub>2</sub>-), 1.25 (8H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.86 (3H, t, J = 6.5 Hz, - CH<sub>3</sub>).

<sup>13</sup>**C-NMR** (75 MHz, CDCl<sub>3</sub>). δ 167.7, 165.4, 147.2, 143.2, 125.1, 118.7, 71.7, 52.5, 51.8, 36.8, 31.7, 29.3, 29.1, 25.2, 22.5, 14.0.

LRMS(EI): m/z 298(M<sup>+</sup>, 0), 280(0), 266(6), 234(24), 167(41), 153(10), 139(100), 127(58).
 E.A. (C<sub>16</sub>H<sub>26</sub>O<sub>5</sub>): Found: C, 64.30, H, 8.90; Calculated: C, 64.41, H, 8.78.

Methyl rac-(2E,4E,S)-6-hydroxy-3-(methoxycarbonyl)-2,4-tridecadienoate (61-trans)

30 [0091]

35



#### 40

**IR** (NaCl): v 3431,2928,2856,1723,1634,1600,1435,1206,1125, 1019, 980, 883 cm<sup>-1</sup>. **<sup>1</sup>H-NMR** (300 MHz, CDCl<sub>3</sub>).  $\delta$  7.38 (1H, d, J = 16.1 Hz, H-4), 6.46 (1H, dd, J = 6.1, 16.1 Hz, H-5), 6.33 (1H, s, H-2), 4.25 (1H, m, H-6), 3.82 (3H, s, -OCH<sub>3</sub>), 3.75 (3H, s, - OCH<sub>3</sub>),1.85-1.56 (4H, m, -CH<sub>2</sub>-), 1.26 (8H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.86 (3H, m, -CH<sub>3</sub>).

<sup>45</sup> <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>). δ 166.7, 165.8, 144.3, 143.0, 122.5, 121.2,72.7,52.5, 51.7, 36.9, 31.7, 29.4, 29.1, 25.2, 22.6, 14.1.

 $\label{eq:LRMS(EI): $m/z$ 298(M^+, 0), 281(0), 267(7), 234(18), 207(7), 179(4), 169(90), 139(100), 127(41).$$$ 

50 Example 4: Preparation of methyl *rac*-(2*Z*,4*E*,*S*)-6-(tert-biqldimethylsilyloxy)-3-(methoxycarbonyl)-2.4-tridecadienoate (62-*cis*)

[0092]



- 10 TBDMSOTf (0.403 g, 1.52 mmoles) was added to a solution of methyl rac-(2Z,4E,S)-6-hydroxy-3-(methoxycarbonyl)-2,4-tridecadienoate (61-cis) (0.380 g, 1.27 mmoles) and Et<sub>3</sub>N (0.167 g, 1.65 mmoles) in CH<sub>2</sub>Cl<sub>2</sub> (13.5 ml) at 0°C. The mixture was stirred at room temperature for 18 hours. After that time, AcOEt (10 ml) and Celite were added, and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 10:1), obtaining (0.430 g, yield 82%) methyl rac-(2Z,4E,S)-6-(tert-butyldimethylsilyloxy)-3-(methoxycarbonyl)-2,4-tridec-
- 15 adienoate (62-cis), as a colorless oil. IR (NaCl): v 3426, 2949, 2929, 2857, 1744, 1723, 1634, 1614, 1461, 1438, 1375, 1271, 1253, 1202, 1169, 1152, 1094, 967, 837, 807, 777 cm<sup>-1</sup>. <sup>1</sup>**H-NMR** (300 MHz, CDCl<sub>3</sub>). δ 6.27 (1H, d, J = 15.8 Hz, H-4), 5.97 (1H, dd, J = 5.1, 15.8 Hz, H-5), 5.80 (1H, s, H-2), 4.22 (1H, m, H-6), 3.89 (3H, s, -OCH<sub>3</sub>), 3.73 (3H, s, - OCH<sub>3</sub>), 1.57 (2H, m, -CH<sub>2</sub>-), 1.25 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.89 (9H, 20 s, tert-BuSi), 0.87 (3H, t, J = 6.5 Hz, -CH<sub>3</sub>), 0.04 (3H, s, MeSi), 0.01 (3H, s, MeSi).

<sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>). δ 167.8, 165.6, 147.6, 144.4, 124.5, 118.0, 72.2, 52.4, 51.8, 37.6, 31.7, 29.5, 29.1, 25.8, 25.3, 24.9, 22.6, 18.2, 14.1, 14.0, -4.5, -4.8.

LRMS(EI): m/z 412(M<sup>+</sup>, 0), 397(2), 380(5), 355(100), 323(58), 313(60),191(39). E.A. (C<sub>22</sub>H<sub>40</sub>O<sub>5</sub>Si): Found: C, 64.10, H, 9.85; Calculated: C, 64.04, H, 9.77.

25

30

5



[0093]



- 40 [0094] OsO<sub>4</sub> (2.5% in tert-BuOH, 0.015 g, 0.062 mmoles) was added to a solution of methyl rac-(2Z, 4E, S)-6-(tert-butyldimethylsilyloxy)-3-(methoxycarbonyl)-2,4-tridecadienoate (62-cis) (0.430 g, 1.042 mmoles) and NMO (0.134 g, 1.14 mmoles) in a 5:1 acetone/H<sub>2</sub>O mixture (4.8 ml). The mixture was stirred at room temperature for 24 hours. After that time, AcOEt (10 ml) and Celite were added, and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 5:1), obtaining (0.270 g, yield 65%) a mixture in a 4:1 ratio
- 45 of methyl rac-(Z,4S,5R,6S)-6-(tert-butyldimethylsilyloxy)-4,5-dihydroxy-3-(methoxycarbonyl)-2-tridecenoate (63a) and methyl rac-(Z,4R,5S,6S)-6-(tert-butyldimethylsilyloxy)-4,5-dihydroxy-3-(methoxycarbonyl)-2-tridecenoate (63b), respectively, as a colorless oil.

IR(NaCl): v 3463, 2949, 2929, 2857, 1730, 1654, 1436, 1360, 1258, 1201, 1170, 1089, 837, 776 cm<sup>-1</sup>.

- <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>). δ 6.27 (1H, d, *J* = 1.9 Hz, H-2 **63***a*), 6.17 (1H, d, *J* = 1.4 Hz, H-2 **63***b*), 4.76 (1H, s<sub>broad</sub>, H-50 4 63a), 4.35 (1H, ddd, J = 1.4,3.6,5.1 Hz, H-4 63b), 3.94 (1H, dt, J = 3.9, 6.6 Hz, H-6 63a), 3.91 (1H, d, J = 2.9 Hz, -OH<sub>4</sub> 63a), 3.82 (3H, s, -OCH<sub>3</sub> 63b), 3.81 (3H, s, -OCH<sub>3</sub> 63a), 3.78 (1H, m, H-6 63b), 3.73 (3H, s, -OCH<sub>3</sub> 63a, 63b), 3.64 (1H, m, H-5 63b), 3.58 (1H, dd, J = 3.9, 7.8 Hz, H-5 63a), 3.27 (1H, d, J = 4.8 Hz, -OH 63b), 2.71 (1H, d, J = 6.8 Hz, -OH 63b), 2.55 (1H, d, J = 7.8 Hz, -OH<sub>5</sub> 63a), 1.57 (2H, m, H-7 63a, 63b), 1.26 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.89 (9H, s, *tert*-BuSi), 0.88 (3H, m, -CH<sub>3</sub>), 0.12 (3H, s, MeSi 63a), 0.11 (3H, s, MeSi 63b), 0.10 (3H, s, MeSi 63a), 0.09 (3H, s, MeSi 63b).
- 55 <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>). δ 167.7, 167.5, 165.7, 165.2, 148.6, 148.1, 121.9, 121.7, 76.0, 73.1, 72.2, 71.4, 70.4, 52.5, 52.5, 52.0, 51.9, 33.9, 31.7, 29.6, 29.1, 25.8, 25.7, 24.9, 22.5, 21.0, 17.9, 14.1, 14.0, -4.5, -4.7.

[0095] LRMS(EI): m/z 445(M<sup>+</sup>-1, 0), 415(3), 389(2), 357(16), 297(5), 243(92), 215(100), 142(34).

Example 6: Preparation of methyl *rac-(Z,*4*S*,5*R*,6*S*)-6-(*tert*-butyldimethylsilyloxy)-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)-2-tridecenoate (64)

[0096]

5

10



- <sup>15</sup> **[0097]** TsOH (0.020 g, cat.) was added to a solution of methyl *rac*-(*Z*,4*S*,5*R*,6*S*)-6-(*tert*-butyldimethylsilyloxy)-4,5dihydroxy-3-(methoxycarbonyl)-2-tridecenoate (**63***a*) (1.36 g, 3.04 mmoles) and 2,2-dimethoxypropane (0.951 g, 9.13 mmoles) in acetone (6 ml) at 0°C. The mixture was stirred at room temperature for 5 hours. After that time, Et<sub>2</sub>O(20 ml) was added, and the mixture was washed with sat. NaHCO<sub>3</sub> (2 × 10 ml), dried with anhydrous MgSO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt,
- 5:1), obtaining (1.34 g, yield 91%) methyl *rac*-(*Z*,4*S*,5*R*,6*S*)-6-(*tert*-butyldimethylsilyloxy)-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)-2-tridecenoate (64), as a colorless oil.
   IR(NaCl): v 3444, 2985, 2949, 2930, 2857, 1734, 1655, 1461, 1436, 1381, 1256, 1167, 1068, 836, 776 cm<sup>-1</sup>.
   <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>). δ 6.08 (1H, s, H-2), 4.68 (1H, d, *J* = 7.5 Hz, H-4), 4.23 (1H, dd, *J* = 3.1, 7.5 Hz, H-5), 3.86 (1H, m, H-6), 3.84 (3H, s, -OCH<sub>3</sub>), 3.74 (3H, s, -OCH<sub>3</sub>), 1.42 (3H, s, -CH<sub>3</sub>), 1.32 (3H, s, -CH<sub>3</sub>), 1.24 (12H, s<sub>broad</sub>, -CH<sub>2</sub>-),
- <sup>25</sup> 0.89 (9H, s, *tert*-BuSi), 0.88 (3H, m, -CH<sub>3</sub>), 0.07 (6H, s, Me<sub>2</sub>Si).
   <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>). δ 167.0, 164.7, 149.0, 121.3, 109.7, 81.3, 76.5, 71.5, 52.3, 52.0, 33.9, 31.7, 29.1, 29.1, 27.1, 26.3, 25.9, 25.1, 22.6, 18.1, 14.0, -4.4, -4.4.
   LRMS(EI): *m/z* 486(M<sup>+</sup>, 0), 471(4), 455(0), 429(44), 411(6), 397(1), 371(7), 339(34), 321(3), 311(7), 279(7), 243(100), 156(36).
- <sup>30</sup> **E.A.** (C<sub>25</sub>H<sub>46</sub>O<sub>7</sub>Si): Found: C, 61.60, H, 9.60; Calculated: C, 61.69, H, 9.53.

Example 7: Reaction of methyl *rac-*(Z,4S,5R,6S)-6-(*tert*-Butyldimethylsilyloxy)-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)-2-tridecenoate (64) with OsO<sub>4</sub>

*35* **[0098]** 

40



45

50

**[0099]** OsO<sub>4</sub> (2.5% in *tert*-BuOH, 0.030 g, 0.12 mmoles) was added to a solution of methyl *rac*-(*Z*,4*S*,5*R*,6*S*)-6-(*tert*-butyldimethylsilyloxy)-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)-2-tridecenoate (**64**) (0.980 g, 2.01 mmoles) and NMO (0.518 g, 4.43 mmoles) in a 5:1 acetone/H<sub>2</sub>O mixture (9.2 ml). The mixture was stirred at room temperature for 5 days. After that time, an aqueous solution of 5% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (0.5 ml), AcOEt (20 ml) and Celite were added, and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/ AcOEt, 10:1), obtaining (0.320 g, yield 31%) methyl *rac*-(2*S*,3*S*,4*R*,5*R*,6*S*)-6-(*tert*-butyldimethylsilyloxy)-2,3-dihydroxy-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)tridecanoate (**65a**) and (0.322 g, yield 31%) methyl *rac*-(2*R*,3*R*,4*R*,5*R*,6*S*)-6-(*tert*-butyldimethylsilyloxy)-2,3-dihydroxy-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)tridecanoate (65b), both as a colorless oil.

Methyl *rac*-(2*S*,3*S*,4*R*,5*R*,6*S*)-6-(*tert*-butyldimethylsilyloxy)-2,3-dihydroxy-4,5-(dimethylmethylenedioxy)-3-(methoxy-carbonyl)tridecanoate (65*a*)

[0100]

5

10



15

- **IR**(NaCl): v 3490, 2949, 2930, 2857, 1748, 1461, 1439, 1379, 1367, 1255, 1213, 1109, 1088, 1000, 837, 775 cm<sup>-1</sup>. **<sup>1</sup>H-NMR** (300 MHz, CDCl<sub>3</sub>).  $\delta$  4.75 (1H, d, J = 5.8 Hz, H-4), 4.43 (1H, s, H-2), 4.41 (1H, dd, J = 5.8, 9.7 Hz, H-5), 3.85 (3H, s, -OCH<sub>3</sub>), 3.80 (1H, m, H-6), 3.76 (3H, s, - OCH<sub>3</sub>), 1.73 (1H, m), 1.59 (1H, m), 1.36 (3H, s, -CH<sub>3</sub>), 1.35 (3H, s, -CH<sub>3</sub>), 1.26 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.92 (9H, s, *tert*-BuSi), 0.91 (3H, m, -CH<sub>3</sub>), 0.16 (3H, s, MeSi), 0.14 (3H, s, MeSi).
- <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>). δ 172.0, 170.7, 110.1, 80.8, 79.6, 78.0, 74.7, 73.2, 52.8, 52.5, 33.3, 31.6, 29.6, 29.0, 27.7, 27.0, 26.9, 25.7, 22.8, 22.4, 18.0, 13.8, 0.8, -4.3, -4.6. LRMS(EI): m/z 520(M<sup>+</sup>, 0), 463(0), 431(1), 405(4), 369(13), 327 (5), 295(5), 243(100), 187(18), 73(60).

**E.A.** (C<sub>25</sub>H<sub>48</sub>O<sub>9</sub>Si): Found: C, 57.60, H, 9.35; Calculated: C, 57.66, H, 9.29.

25 Methyl rac-(2R,3R,4R,5R,6S)-6-(tert-butyldimethylsilyloxy)-2,3-dihydroxy-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)tridecanoate (65b)

[0101]

30



IR (NaCl): v 3489, 2949, 2929, 2857, 1748, 1461, 1438, 1380, 1367, 1252, 1216, 1101, 1050, 1003, 836, 775 cm<sup>-1</sup>.
 <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>). δ 4.65 (1H, m, H-4), 4.36 (1H, s, H-2), 4.34 (1H, m, H-5), 3.80 (3H, s, -OCH<sub>3</sub>), 3.79 (3H, s, -OCH<sub>3</sub>), 3.76 (1H, m, H-6), 1.61 (2H, m), 1.41 (3H, s, -CH<sub>3</sub>), 1.39 (3H, s, -CH<sub>3</sub>), 1.26 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.90 (9H, s, *tert*-BuSi), 0.89 (3H, m, -CH<sub>3</sub>), 0.09 (3H, s, MeSi), 0.08 (3H, s, MeSi).
 <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>). δ 171.1, 170.9, 110.2, 78.9, 78.8, 76.9, 72.7, 71.1, 52.8, 31.6, 29.4, 29.0, 27.6, 26.5, 25.7,

<sup>45</sup> 24.8, 22.4, 18.0, 13.8, -4.4, -4.6.
 LRMS(EI): m/z 520(M<sup>+</sup>, 0), 505(2), 463(0), 431(1), 387(9), 327(4), 299(6), 243(100), 187(19), 73(69).
 E.A. (C<sub>25</sub>H<sub>48</sub>O<sub>9</sub>Si): Found: C, 57.58, H, 9.37; Calculated: C, 57.66, H, 9.29.

Example 8: Preparation of methyl *rac-(Z,4S,5R,6R,7S)-7-(tert-*butyldimethylsilyloxy)-5,6-(dimethylmethylenedioxy) 4-hydroxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (76*a*)

[0102]

55



**ŌTBDMS** 



10

5

IBX (0.474 g, 1.69 mmoles) was added to a solution of methyl *rac-*(2S,3S,4R,5R,6S)-6-(*tert*-butyldimethylsilyloxy)-2,3-dihydroxy-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)tridecanoate (**65***a*) (0.294 g, 0.564 mmoles) in AcOEt (5 ml). The mixture was heated at 80°C for 8 hours. After that time, the mixture was filtered under vacuum over Celite and the solvent was removed under reduced pressure. The residue was dissolved in  $CH_2Cl_2$  (10 ml), and [(methoxycarbonyl)

- <sup>15</sup> methylene]triphenylphosphorane (0.451 g, 1.34 mmoles) was added. The mixture was stirred at room temperature for 24 hours. After that time AcOEt (10 ml) and Celite were added, and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 5:1), obtaining methyl *rac-(Z,4S,5R,6R,7S)-7-(tert*-butyldimethylsilyloxy)- 5,6-(dimethylmethylenedioxy)- 4- hydroxy- 3,4- bis (methoxycarbonyl)- 2- tetradecenoate (**76a**) (0.220 g, yield 71%), as a colorless oil.
- IR (NaCl): v 3477, 2949, 2929, 2854, 1743, 1642, 1461, 1435, 1367, 1252, 1213, 1166, 1095, 1062, 836, 774 cm<sup>-1</sup>.
  <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>). δ 6.49 (1H, s, H-2), 4.45 (1H, d, *J* = 7.3 Hz, H-5), 4.28 (1H, dd, *J* = 2.6, 7.3 Hz, H-6), 4.00 (1H, s, -OH), 3.85 (3H, s, -OCH<sub>3</sub>), 3.81 (3H, s, OCH<sub>3</sub>), 3.77 (1H, m, H-7), 3.73 (3H, s, -OCH<sub>3</sub>), 1.63-1.42 (2H, m, H-8), 1.38 (6H, s, CH<sub>3</sub>), 1.28 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.89 (3H, m, -CH<sub>3</sub>), 0.88 (9H, s, *tert*-BuSi), 0.02 (3H, s, MeSi), 0.02 (3H, s, MeSi).
- <sup>25</sup> <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>). δ 171.0, 166.3, 164.5, 147.1, 122.5, 110.1, 79.9, 79.8, 78.4, 72.7, 53.9, 52.5, 52.0, 31.8, 31.2, 29.8, 29.3, 27.7, 26.3, 25.9, 25.7, 25.4, 22.6, 18.2, 14.0, -4.4. LRMS(EI): m/z 574(M<sup>+</sup>, 0), 517(3), 485(2), 409(5), 335(14), 257(13), 243(100), 215(31), 171(16). E.A. (C<sub>28</sub>H<sub>50</sub>O<sub>10</sub>Si): Found: C, 58.60, H, 8.85; Calculated: C, 58.51, H, 8.77.
- 30 Example 9: Preparation of methyl *rac-(Z,*4*R*,5*R*,6*R*,7*S*)-7-(*tert*-butyldimethylsilyloxy)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate

[0103] (76*b*)



40



- IBX (0.519 g, 1.85 mmoles) was added to a solution of methyl *rac-(2R,3R,4R,5R,6S)-6-(tert-butyldimethylsi-lyloxy)-2,3-dihydroxy-4,5-(dimethylmethylenedioxy)-3-(methoxycarbonyl)tridecanoate* (65b) (0.322 g, 0.618 mmoles) in AcOEt (6 ml). The mixture was heated at 80°C for 8 hours. After that time, the mixture was filtered under vacuum over Celite, and the solvent was removed under reduced pressure. The residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (10 ml), and [(methoxycarbonyl)methylene]triphenylphosphorane (0.483 g, 1.44 mmoles) was added. The mixture was stirred at room
- 50 temperature for 24 hours. After that time, AcOEt (10 ml) and Celite were added, and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 5:1), obtaining methyl *rac-(Z*, 4*R*,5*R*,6*R*,7*S*)-7-(*tert*-butyldimethylsilyloxy)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbonyl)-2-tet-radecenoate (**76b**) (0.270 g, yield 76%), as a colorless oil.

IR (NaCl): v 3471, 2949, 2930, 2857, 1735, 1645, 1461, 1435, 1368, 1254, 1167, 1086, 836, 775 cm<sup>-1</sup>.

<sup>55</sup> <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>). δ 6.51 (1H, s, H-2), 4.60 (1H, d, J = 6.6 Hz, H-5), 4.50 (1H, s, -OH), 4.04 (1H, dd, J = 6.2, 6.6 Hz, H-6), 3.81 (3H, s, -OCH<sub>3</sub>), 3.78 (3H, s, - OCH<sub>3</sub>), 3.74 (3H, s, -OCH<sub>3</sub>), 3.68 (1H, dt, J = 6.2, 10.9 Hz, H-7), 1.56 (2H, m, H-8), 1.39 (6H, s, -CH<sub>3</sub>), 1.28 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.90 (12H, m, *tert*-BuSi, -CH<sub>3</sub>), 0.10 (6H, m, -CH<sub>3</sub>). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>). δ 170.2, 166.1, 165.0, 145.9, 124.0, 110.6, 79.5, 78.7, 76.3, 73.4, 53.3, 52.1, 51.8, 33.1,

31.6, 29.5, 29.0, 27.0, 26.8, 26.1, 25.6, 23.5, 17.9, 13.8, -4.4, -4.7. LRMS(EI): m/z 574(M<sup>+</sup>, 0), 559(3), 517(2), 485(2), 409(15), 335(26), 243(100), 215(34). E.A. (C<sub>28</sub>H<sub>50</sub>O<sub>10</sub>Si): Found: C, 58.58, H, 8.87; Calculated: C, 58.51, H, 8.77.

#### 5 Example 10: Preparation of methyl *rac-(Z,4S,5R,6S)-4,5-epoxy-6-hydroxy-3-(methoxycarbonyl)-2-tridecenoate (82c)*

#### [0105]

10



15

*m*-CPBA (0.101 g, 0.59 mmoles) was added to a solution of methyl *rac*-(2Z,4*E*,*S*)-6-hydroxy-3-(methoxycarbonyl)-2,4-tridecadienoate (**3***a*) (0.080 g, 0.26 mmoles) in CCl<sub>4</sub> (2.5 ml). The mixture was stirred at room temperature for 3 days. After that time, the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 4:1), obtaining (0.080 g, yield 95%) methyl *rac*-(Z,4*S*,5*R*,6*S*)-4,5-epoxy-6-hydroxy-3-(methoxycarbo-

- 20 (hexane/AcOEt, 4:1), obtaining (0.080 g, yield 95%) methyl *rac*-(*Z*,4*S*,5*R*,6*S*)-4,5-epoxy-6-hydroxy-3-(methoxycarbo-nyl)-2-tridecenoate (82c), as a colorless oil.
   IR (NaCl): v 3473, 2929, 2857, 1730, 1652, 1574, 1437, 1372, 1276, 1204, 1171, 1020, 870 cm<sup>-1</sup>.
   <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>). δ 6.15 (1H, s, H-2), 3.86 (1H, m), 3.81 (3H, s, -OCH<sub>3</sub>), 3.73 (3H, s, -OCH<sub>3</sub>), 3.71 (3H, s, -OCH<sub>3</sub>), 3.63 (3H, s, -OCH<sub>3</sub>), 3.58 (1H, m), 3.05 (1H,
- <sup>25</sup> [0106] m), 1.63 (2H, m, -CH<sub>2</sub>-), 1.25 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.85 (3H, m, -CH<sub>3</sub>).
   <sup>13</sup>C-NMR (75 MHz, CDCl<sub>3</sub>). δ 165.6, 165.6, 165.0, 164.9, 144.5, 144.1, 122.0, 121.8, 70.1, 67.8, 63.9, 63.4, 53.8, 52.6, 52.2, 52.1, 34.3, 33.0, 31.6, 29.4, 29.3, 29.0, 25.1, 24.9, 22.5, 14.0.
   LRMS(EI): m/z 314(M<sup>+</sup>, 0), 282(5), 237(1), 223(1), 173(10), 156(100), 141(82), 127(10).
   E.A. (C<sub>16</sub>H<sub>26</sub>O<sub>6</sub>): Found: C, 61.00, H, 8.41; Calculated: C, 61.13, H, 8.34.

#### 30

Example 11: Preparation of methyl *rac-(Z,4S,5R,6R)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbo-nyl)-7-oxo-2-tetradecenoate (110a)* 

[0107]



40

35



76a



45

**[0108]** Et<sub>3</sub>N·(HF)<sub>3</sub> (0.492 g, 3.05 mmoles) was added to a solution of methyl *rac-*(*Z*,4*S*,5*R*,6*R*,7*S*)-7-(*tert*-butyldimethylioxy)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (**76a**) (0.220 g, 0.382 mmoles) in MeOH (5 ml). The mixture was stirred at room temperature for 6 days. After that time, AcOEt (10 ml) was added and the mixture washed with  $H_2O$  (2 × 5 ml), dried with anhydrous  $Na_2SO_4$ , filtered and the solvent was removed upder reduced processor. The residue was discelled in AcOEt (2 6 ml) and IBX (0.224 g, 0.801 mmoles, 2 cg.) was

<sup>50</sup> under reduced pressure. The residue was dissolved in AcOEt (3.6 ml), and IBX (0.224 g, 0.801 mmoles, 3 eq.) was added. The mixture was heated at 80°C for 7 hours. After that time, the mixture was filtered under vacuum over Celite and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 4:1), obtaining (0.110 g, yield 63%) methyl *rac-(Z,4S,5R,6R)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis* (methoxycarbonyl)-7-oxo-2-tetradecenoate (**110***a*), as a colorless oil. **IR** (NaCI): v 3477, 2949, 2931, 2854, 1733, 1645, 1455, 1436, 1373, 1201, 1162, 1083, 985, 876 cm<sup>-1</sup>.

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>). δ 6.45 (1H, s, H-2), 5.02 (1H, d, J = 4.6 Hz, H-6), 4.51 (1H, d, J = 4.6 Hz, H-5), 3.88 (3H, s, -OCH<sub>3</sub>), 3.80 (3H, s, -OCH<sub>3</sub>), 3.73 (3H, s, - OCH<sub>3</sub>), 2.64 (2H, m, H-8), 1.57 (2H, m, H-9), 1.45 (3H, s, -CH<sub>3</sub>), 1.25 (13H, s<sub>broad</sub>, - CH<sub>2</sub>-, -CH<sub>3</sub>), 0.86 (3H, m, -CH<sub>3</sub>).

<sup>13</sup>**C-NMR** (100 MHz, CDCl<sub>3</sub>). δ 209.2, 170.7, 166.0, 164.4, 146.2, 122.9, 112.3, 80.6, 79.7, 77.5, 54.2, 52.6, 52.1, 39.2, 31.6, 29.0, 29.0, 26.5, 26.5, 23.1, 22.5, 14.0. **LRMS(EI)**: *m*/*z* 458(M<sup>+</sup>, 0), 443(1), 427(0), 399(0), 369(12), 331(7), 313 (14), 273(92), 227(32), 127(100).

E.A. (C<sub>22</sub>H<sub>34</sub>O<sub>10</sub>): Found: C, 57.70, H, 7.50; Calculated: C, 57.63, H, 7.47.

5

Example 12: Preparation of methyl *rac*-(*Z*,4*R*,5*R*,6*R*)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbo-nyl)-7-oxo-2-tetradecenoate (110*b*)

[0109]

10

15



20

25

**[0110]** Et<sub>3</sub>N·(HF)<sub>3</sub> (0.605 g, 3.75 mmoles) was added to a solution of methyl *rac-(Z,4R,5R,6R,7S)*-7-(*tert*-butyldimethylsilyloxy)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate **(76b)** (0.270 g, 0.469 mmoles) in MeOH (6 ml). The mixture was stirred at room temperature for 6 days. After that time, AcOEt (10 ml) was added and the mixture washed with  $H_2O$  (2 × 5 ml), dried with anhydrous  $Na_2SO_4$ , filtered and the solvent was removed

- under reduced pressure. The residue was dissolved in AcOEt (4 ml), and IBX (0.324 g, 1.15 mmoles) was added. The mixture was heated at 80°C for 7 hours. After that time, the mixture was filtered under vacuum over Celite, and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 4:1), obtaining (0.160 g, yield 74%) methyl *rac*-(2*Z*,4*R*,5*R*,6*R*)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(meth-
- <sup>30</sup> oxycarbonyl)-7-oxo-2-tetradecenoate (**110***b*), as a colorless oil. **IR** (NaCl): v 3470, 2985, 2949, 2930, 2854, 1732, 1648, 1436, 1373, 1352, 1255, 1166, 1090, 882 cm<sup>-1</sup>.
  <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>). δ 6.45 (1H, s, H-2), 4.70 (1H, d, *J* = 6.6 Hz, H-6), 4.37 (1H, d, *J* = 6.6 Hz, H-5), 3.96 (1H, s, -OH), 3.83 (3H, s, -OCH<sub>3</sub>), 3.81 (3H, s, -OCH<sub>3</sub>), 3.74 (3H, s, -OCH<sub>3</sub>), 2.66-2.61 (2H, m, H-8), 1.55 (2H, m, H-9), 1.47 (3H, s, -CH<sub>3</sub>), 1.38 (3H, s, -CH<sub>3</sub>), 1.27 (8H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.87 (3H, m, -CH<sub>3</sub>).
- <sup>35</sup> <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>). δ 210.6, 169.9, 166.2, 164.7, 146.1, 123.4, 111.7, 80.4, 79.1, 76.3, 53.7, 52.3, 51.9, 38.6, 31.4, 28.8, 28.8, 26.2, 26.08, 22.6, 22.3, 13.8.
   LRMS(EI): m/z 459(M<sup>+</sup>+1, 0), 443(2), 399(1), 369(4), 331(10), 313(11), 299(7), 273(50), 255(32), 127(100).
   E.A. (C<sub>22</sub>H<sub>34</sub>O<sub>10</sub>): Found: C, 57.72, H, 7.55; Calculated: C, 57.63, H, 7.47.
- 40 Example 13: Reaction of methyl *rac-(Z,*4*S,*5*R,*6*R*)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbonyl)-7-oxo-2-tetradecenoate (110*a*) with *p*-TsOH and MeOH

[0111]

45

50



*p*-TsOH (0.008 g, cat.) was added to a solution of methyl *rac*-(*Z*,4*S*,5*R*,6*R*)-5,6-(dimethylmethylenedioxy)-4-hydroxy-3,4-bis(methoxycarbonyl)-7-oxo-2-tetradecenoate (**110***a*) (0.045 g, 0.098 mmoles) in MeOH (1.3 ml). The mixture was stirred at room temperature for 6 days. After that time, NaHCO<sub>3</sub> (0.03 g) was added and the mixture was stirred for 10 minutes. Then, the mixture was filtered under vacuum over Celite and the solvent was removed under reduced pressure.

The product was purified by a chromatographic column (hexane/AcOEt, 3:1), obtaining (0.030 g, yield 71%) methyl rac-(Z,4S,5R,6R,7S)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (**111***a*) and (0.011 g, yield 26%) methyl rac-(Z,4S,5R,6R,7R)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (**112***a*), both as a colorless oil.

5

Methyl rac-(Z,4S,5R,6R,7S)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (111a)

#### [0112]

10

15



20 **[0113] IR** (NaCl): v 3466, 2950, 2926, 2852, 1758, 1732, 1642, 1435, 1258, 1169, 1080 cm<sup>-1</sup>. <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>). δ 6.47 (1H, s, H-2), 4.47 (1H, dd, J = 4.1, 8.6 Hz, H-5), 4.21 (1H, dd, J = 8.6, 10.7 Hz, H-6), 3.81 (3H, s, -OCH<sub>3</sub>), 3.74 (3H, s, -OCH<sub>3</sub>), 3.73 (3H, s, -OCH<sub>3</sub>), 3.64 (1H, d, J = 4.1 Hz, -OH), 3.32 (3H, s, -OCH<sub>3</sub>), 2.42 (1H, d, J = 10.7 Hz, -OH), 1.99 (1H, m, H-8), 1.83 (1H, m, H-8), 1.29 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.88 (3H, m, -CH<sub>3</sub>). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>). δ 169.5, 167.6, 165.6, 144.0, 122.8, 105.6, 85.0, 82.5, 78.0, 52.9, 52.9, 52.2, 49.1, 32.9, 31.7, 29.7, 29.0, 23.0, 22.6, 14.0

31.7, 29.7, 29.0, 23.0, 22.6, 14.0. LRMS(EI): *m*/*z* 432(M<sup>+</sup>, 0), 401(0), 383(0), 369(1), 333(6), 323(9), 255(2), 213(73), 181(100), 143 (16), 99(3). E.A. (C<sub>20</sub>H<sub>32</sub>O<sub>10</sub>): Found: C, 55.60, H, 7.55; Calculated: C, 55.55, H, 7.46.

Methyl rac-(Z,4S,5R,6R,7R)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (112a)

30

[0114]

35





IR (NaCl): v 3491, 2954, 2855, 1732, 1650, 1436, 1351, 1268, 1173, 1076, 1030, 783 cm<sup>-1</sup>.

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>). δ 6.41 (1H, s, H-2), 4.43 (1H, d, J = 11.7 Hz, H-5), 4.04 (1H, d, J = 8.5 Hz, H-6), 3.95 (1H, d, J = 11.7 Hz, -OH), 3.86 (3H, s, -OCH<sub>3</sub>), 3.81 (3H, s, -OCH<sub>3</sub>), 3.73 (3H, s, -OCH<sub>3</sub>), 3.43 (3H, s, -OCH<sub>3</sub>), 3.26 (1H, d, J = 8.5 Hz, -OH), 1.96 (1H, m, H-8), 1.70 (1H, m, H-8), 1.30 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.88 (3H, m, -CH<sub>3</sub>). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>). δ 168.2, 168.2, 164.5, 148.3, 122.5, 114.9, 91.5, 86.2, 78.9, 53.1, 53.1, 52.1, 49.2, 31.7, 29.5, 29.1, 28.1, 23.7, 22.6, 14.0.

LRMS(EI): *m*/*z* 432(M<sup>+</sup>, 0), 383(0), 333(4), 301(1), 255(1), 213(72), 181(100), 173(39), 99(9).

<sup>50</sup> **E.A.** (C<sub>20</sub>H<sub>32</sub>O<sub>10</sub>): Found: C, 55.62, H, 7.53; Calculated: C, 55.55, H, 7.46.

Example 14: Reaction of methyl *rac*-(*Z*,4*S*,5*R*,6*R*,7*S*)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (111*a*) with OsO<sub>4</sub>

55 **[0115]** 



10

15

5

**[0116]**  $OsO_4$  (2.5% in *tert*-BuOH, 0.001 g, 0.004 mmoles) was added to a solution of methyl *rac*-(*Z*,4*S*,5*R*,6*R*,7*S*)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (**111***a*) (0.030 g, 0.069 mmoles) and NMO (0.017 g, 0.15 mmoles) in a 5:1 acetone/H<sub>2</sub>O mixture (0.6 ml). The mixture was stirred at room temperature for 6 days. After that time, an aqueous solution of 5% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (0.5 ml), AcOEt (10 ml) and Celite were added, and the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 1:1),

obtaining (0.005 g, yield 17%) a mixture of methyl *rac*-(2*S*,3*S*,4*R*,5*R*,6*R*,7*S*)-4,7-epoxy-7-methoxy-3-(methoxycarbonyl)-3,5,6-trihydroxytetradecanoate 4,2-carbolactone (120) and methyl *rac*-(2*R*,3*R*,4*R*,5*R*,6*R*,7*S*)-4,7-epoxy-7-methoxy-3-(methoxycarbonyl)-3,5,6-trihydroxytetradecanoate 4,2-carbolactone (121) in a ratio of 1:11, respectively, as a colorless oil.

Methyl *rac*-(2*S*,3*S*,4*R*,5*R*,6*R*,7*S*)-4,7-epoxy-7-methoxy-3-(methoxycarbonyl)-3,5,6-trihydroxytetradecanoate 4,2-carbolactone (120)

25

30



<sup>35</sup> **IR** (KBr): v 3426, 2922, 2853, 1739, 1636, 1437, 1375, 1077 cm<sup>-1</sup>. <sup>1</sup>**H-NMR** (400 MHz, CDCl<sub>3</sub>). δ 4.98 (1H, s, H-2), 4.77 (1H, d, J = 3.0 Hz, -OH), 4.65 (1H, dd, J = 3.0, 7.9 Hz, H-5), 4.19 (1H, dd, J = 7.9, 10.5 Hz, H-6), 3.91 (3H, s, -OCH<sub>3</sub>), 3.86 (3H, s, -OCH<sub>3</sub>), 3.82 (1H, s, -OH), 3.19 (3H, s, -OCH<sub>3</sub>), 2.36 (1H, d, J = 10.5 Hz, -OH), 1.80 (2H, m, H-8), 1.25 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.87 (3H, t, J = 5.1 Hz, - CH<sub>3</sub>). <sup>13</sup>**C-NMR** (100 MHz, CDCl<sub>3</sub>). δ 169.9, 169.4, 166.5, 105.7, 85.8, 78.8, 78.5, 75.8, 67.1, 54.3, 53.6, 49.0, 32.1, 31.7,

<sup>40</sup> 29.7, 29.0, 23.1, 22.6, 14.0.
 LRMS(EI): m/z 434(M<sup>+</sup>, 0), 335(48), 315(16), 290(2), 275(10), 247(52), 229(78), 201(23), 159(100), 101(53).
 E.A. (C<sub>19</sub>H<sub>30</sub>O<sub>11</sub>): Found: C, 52.60, H, 7.00; Calculated: C, 52.53, H, 6.96.

methyl *rac*-(2*R*,3*R*,4*R*,5*R*,6*R*,7*S*)-4,7-epoxy-7-methoxy-3-(methoxycarbonyl)-3,5,6-trihydroxytetradecanoate 4,2-carbolactone (121)

[0117]

50



IR (KBr): v 3433, 2953, 2922, 2847, 1804, 1739, 1630, 1437, 1148, 1083, 1055, 800, 499 cm<sup>-1</sup>.

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>). δ 5.43 (1H, s, H-2), 4.28 (2H, m, H-5, H-6), 3.94 (3H, s, - OCH<sub>3</sub>), 3.83 (3H, s, -OCH<sub>3</sub>), 3.62 (1H, s<sub>broad</sub>, -OH), 3.26 (3H, s, -OCH<sub>3</sub>), 2.36 (1H, d, J = 9.0 Hz, -OH), 1.89 (2H, m, H-8), 1.23 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.89 (3H, m, -CH<sub>3</sub>). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>). δ 169.0, 167.5, 165.6, 106.1, 94.9, 80.9, 78.4, 77.7, 77.5, 53.1, 52.9, 49.2, 32.0, 31.6, 29.8, 29.0, 23.2, 22.5, 14.1.

**LRMS(EI):** *m*/*z* 385(1), 355(1), 335(11), 303(1), 290(2), 278(4), 247(15), 217(4), 159(40), 101(15), 83(100). E.A. (C<sub>19</sub>H<sub>30</sub>O<sub>11</sub>): Found: C, 52.45, H, 6.90; Calculated: C, 52.53, H, 6.96.

Example 14: Reaction of methyl *rac-(Z,*4*S*,5*R*,6*R*,7*S*)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (111*a*) with RuCl<sub>3</sub>/NalO<sub>4</sub>

[0118]

15

20

5

10



**[0119]** A solution of  $\operatorname{RuCl}_3 \cdot 3H_2O$  (0.003 g, 0.014 mmoles) and  $\operatorname{NalO}_4$  (0.018 g, 0.086 mmoles) in  $H_2O$  (0.2 ml) was added to a solution of methyl *rac-(Z,4S,5R,6R,7S)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tet-radecenoate* (**111***a*)

(0.025 g, 0.057 mmoles) in a 1:1 AcOEt/MeCN mixture (1 ml) at 0°C. The resulting mixture was stirred at 0°C for 5 minutes. After that time, sat. NaHCO<sub>3</sub> (1 ml) and AcOEt (2 ml) were added. The phases were separated, and the aqueous phase was extracted with AcOEt (2  $\times$  1 ml). The organic phase was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and the

solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 1:1), obtaining (0.008 g, yield 32%) methyl *rac*-(2*S*,3*S*,4*R*,5*R*,6*R*,7*S*)-4,7-epoxy-7-methoxy-3-(methoxycarbonyl)-3,5,6-trihydroxytetradecanoate 4,2-carbolactone (120), as a colorless oil.
 IR (KBr): v 3426, 2922, 2853, 1739, 1636, 1437, 1375, 1077 cm<sup>-1</sup>.

<sup>1</sup>**H-NMR** (400 MHz, CDCl<sub>3</sub>). δ 4.98 (1H, s, H-2), 4.77 (1H, d, J = 3.0 Hz, -OH), 4.65 (1H, dd, J = 3.0, 7.9 Hz, H-5), 4.19 (1H, dd, J = 7.9, 10.5 Hz, H-6), 3.91 (3H, s, -OCH<sub>3</sub>), 3.86 (3H, s, -OCH<sub>3</sub>), 3.82 (1H, s, -OH), 3.19 (3H, s, -OCH<sub>3</sub>), 2.36 (1H, d, J = 10.5 Hz, -OH), 1.80 (2H, m, H-8), 1.25 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.87 (3H, t, J = 5.1 Hz, -CH<sub>3</sub>).

<sup>13</sup>**C-NMR** (100 MHz, CDCl<sub>3</sub>). δ 169.9, 169.4, 166.5, 105.7, 85.8, 78.8, 78.5, 75.8, 67.1, 54.3, 53.6, 49.0, 32.1, 31.7, 29.7, 29.0, 23.1, 22.6, 14.0.

LRMS(EI): *m*/*z* 434(M<sup>+</sup>, 0), 335(48), 315(16), 290(2), 275(10), 247(52), 229(78), 201(23), 159(100), 101(53).

<sup>40</sup> **E.A.** (C<sub>19</sub>H<sub>30</sub>O<sub>11</sub>): Found: C, 52.60, H, 7.00; Calculated: C, 52.53, H, 6.96.

Example 15: Reaction of methyl *rac*-(*Z*,4*S*,5*R*,6*R*,7*R*)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetradecenoate (112*a*) with RuCl<sub>3</sub>/NalO<sub>4</sub>

45 **[0120]** 



55

A solution of  $RuCl_3 \cdot 3H_2O(0.002 \text{ g}, 0.010 \text{ mmoles})$  and  $NalO_4(0.013 \text{ g}, 0.062 \text{ mmoles})$  in  $H_2O(0.2 \text{ ml})$  was added to a solution of methyl *rac*-(*Z*,4*S*,5*R*,6*R*,7*R*)-5,6-dihydroxy-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2-tetrade-

cenoate (**112***a*) (0.018 g, 0.041 mmoles) in a 1:1 AcOEt/MeCN mixture (1 ml) at 0°C. The mixture was stirred at 0°C for 5 minutes. After that time, an aqueous solution of 10% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (0.2 ml), H<sub>2</sub>O (2 ml) and AcOEt (2 ml) were added. The phases were separated, and the aqueous phase was extracted with AcOEt (2 × 1 ml). The organic phase was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and the solvent was removed under reduced pressure. The product was purified by a

<sup>5</sup> chromatographic column (hexane/AcOEt, 1:2), obtaining (0.014 g, yield 74%) methyl *rac*-(2*S*,3*S*,4*S*,5*R*,6*R*,7*R*)-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2,3,5,6-tetrahydroxytetradecanoate (122), as a colorless oil.
IR (NaCI): v 3435, 2957, 2926, 2852, 1643, 1438, 1074 cm<sup>-1</sup>.
<sup>1</sup>H-NMR (400 MHz, CDCI<sub>3</sub>). δ 5.25 (1H, d, *J* = 6.7 Hz, H-2), 5.06 (1H, d, *J* = 10.1 Hz, H-5), 4.51 (1H, s, -OH), 4.18 (1H, d, *J* = 10.7 Hz, -OH), 3.92 (1H, d, *J* = 10.7 Hz, H-6), 3.85 (3H, s, -OCH<sub>3</sub>), 3.83 (3H, s, -OCH<sub>3</sub>), 3.75 (3H, s, -OCH<sub>3</sub>), 3.59

- <sup>10</sup> (1H, d, J = 6.7 Hz, -OH), 3.36 (3H, s, -OCH<sub>3</sub>), 3.25 (1H, d, J = 10.1 Hz, -OH), 1.85 (1H, m, H-8), 1.66 (1H, m, H-8), 1.27 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.87 (3H, m, -CH<sub>3</sub>). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>).  $\delta$  170.8, 170.8, 170.7, 113.7, 79.7, 79.4, 78.8, 78.3, 73.8, 53.9, 52.9, 52.4, 49.5, 31.8, 29.7, 29.1, 28.1, 23.4, 22.6, 14.0. LRMS(EI): m/z 466(M<sup>+</sup>, 0), 385(2), 375(3), 367(4), 335(11), 315(6), 275(4), 247(15), 229(26), 185(22), 159(100), 127(49).
- **ERMS(EI):** 772(40), 75(3), 75(3), 75(3), 75(3), 75(3), 75(4), 75(4), 275(4), 247(15), 229(20), 75(22), 75(22), 75(100), 727(49). **E.A.** (C<sub>20</sub>H<sub>34</sub>O<sub>12</sub>): Found: C, 51.65, H, 7.20; Calculated: C, 51.50, H, 7.35.

Example 16: Preparation of *rac*-(1*R*,3*S*,4*S*,5*S*,6*R*,7*R*)-1-heptyl-3,4,5-tris(methoxycarbonyl)-4,6,7-trihydroxy-2,8-dioxabicyclo[3.2.1]octane (3)

20 [0121]



<sup>45</sup> 

**[0122]** A solution of methyl *rac*-(2*S*,3*S*,4*R*,5*R*,6*R*,7*S*)-4,2-carbolactone-4,7-epoxy-7-methoxy-3-(methoxycarbonyl)-3,5,6-trihydroxy-2-tetradecenoate **(120)** (0.008 g, 0.018 mmoles) in HCI (2% in MeOH, 2.37 ml) was heated in a Kimble at 80°C for 30 hours. After that time, the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 1:1), obtaining (0.006 g, yield 75%) *rac*-(1*R*,3*S*,4*S*,5*S*,6*R*,7*R*)-1-heptyl-3,4,5tris(methoxycarbonyl)-4,6,7-trihydroxy-2,8-dioxabicyclo[3.2.1]octane **(3)**, as a white solid.

55

# Method B





10

15

**[0123]** A solution of methyl *rac*-(2*S*,3*S*,4*S*,5*R*,6*R*,7*R*)-4,7-epoxy-7-methoxy-3,4-bis(methoxycarbonyl)-2,3,5,6-tetrahydroxy-2-tetradecanoate **(122)** (0.026 g, 0.055 mmoles) in HCI (2% in MeOH, 2.5 ml) was heated in a Kimble at 80°C for 30 hours.<sup>73</sup> After that time, the solvent was removed under reduced pressure. The product was purified by a chromatographic column (hexane/AcOEt, 1:1), obtaining (0.019 g, yield 73%) *rac*-(1*R*,3*S*,4*S*,5*S*,6*R*,7*R*)-1-heptyl-3,4,5-tris (methoxycarbonyl)-4,6,7-trihydroxy-2,8-dioxabicyclo[3.2.1]octane **(3)**, as a white solid.

rac-(1R,3S,4S,5S,6R,7R)-1-Heptyl-3,4,5-tris(methoxycarbonyl)-4,6,7-trihydroxy-2,8-dioxabicyclo[3.2.1]octane (3)

HO

MeO<sub>2</sub>(

MeO<sub>2</sub>(

#### 20 [0124]

25

#### *m.p.:* > 220°C.

<sup>1</sup>**H-NMR** (400 MHz, CDCl<sub>3</sub>). δ 5.09 (1H, s, H-3), 5.08 (1H, dd, J = 2.4, 5.4 Hz, H-6), 4.14 (1H, dd, J = 2.4, 3.7 Hz, H-7), 3.84 (3H, s, -OCH<sub>3</sub>), 3.74 (3H, s, -OCH<sub>3</sub>), 3.68 (3H, s, -OCH<sub>3</sub>), 3.67 (1H, s, -OH), 2.5 (1H, d, J = 5.4 Hz, -OH), 2.40 (1H, d, J = 3.7 Hz, - OH), 1.89 (2H, m, H-1'), 1.21 (10H, s<sub>broad</sub>, -CH<sub>2</sub>-), 0.81 (3H, m, -CH<sub>3</sub>).

3

<sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>). δ 169.7, 167.0, 166.9, 106.3, 91.4, 82.2, 78.4, 75.4, 74.6, 53.6, 53.0, 52.6, 35.5, 31.7, 29.5, 29.0, 22.7, 22.6, 14.0.

**LRMS(EI):** *m*/*z* 434(M<sup>+</sup>, 0), 336(0), 314(0), 278(0), 243(0), 219(1), 149(3), 83(100). **E.A.** (C<sub>19</sub>H<sub>30</sub>O<sub>11</sub>): Found: C, 52.50, H, 6.90; Calculated: C, 52.53, H, 6.96.

#### 40 Claims

1. A process for obtaining a compound of formula (I), its enantiomers or mixtures thereof

45



50

55

## wherein

 $R^2$  is selected from the group consisting of  $C_1$ - $C_{20}$  alkyl and  $C_1$ - $C_{20}$  alkenyl, which are unsubstituted or substituted in any position with at least one group which is selected from the group consisting of  $C_1$ - $C_4$  alkyl,  $C_1$ - $C_3$  alkylidene,  $C_1$ - $C_3$  alkylcarboxyhydroxyl, hydroxyl and protected hydroxyl; and/or substituted with a group in the end position of the chain which is selected from the group consisting of  $C_6$ - $C_{10}$  aryl, mono- or bicyclic heteroaryl with 5- or 6members in each ring, which can be unsubstituted or substituted with at least one group which is selected from the

group formed by C1-C3 alkyl, or halogen; and

 $R^3$ ,  $R^4$  and  $R^5$  are independently selected from the group of  $C_1$ - $C_3$  alkyls;

characterized in that it comprises reacting in acidic medium a compound of formula (II), its enantiomers or mixtures thereof, or a compound of formula (III), its enantiomers or mixtures thereof, or a mixture of compounds of formula (II) and (III)

(III)

5





20

wherein R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as defined above; and  $R^6$  is a  $C_1$ - $C_3$  alkyl group.

2. The process according to claim 1, wherein R<sup>3</sup> and R<sup>4</sup> are identical.

 $R^{3}O_{2}C$ 

(II)

- The process according to any of the previous claims, wherein R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are identical, preferably methyl. 3.
- 25 A compound of formula (II), its enantiomers or mixtures thereof, as defined in any of claims 1 to 3. 4.
  - 5. The compound of formula (III), its enantiomers or mixtures thereof, as defined in any of claims 1 to 3.
- A process for the synthesis of a compound of formula (II), its enantiomers or mixtures thereof, or of a compound of 6. 30 formula (III), its enantiomers or mixtures thereof, as defined in any of claims 1 to 3, characterized in that it comprises the dihydroxylation of a compound of formula (IV), its enantiomers or mixtures thereof

35





40

45

wherein R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as they have been defined in any of claims 1 to 3; and  $R^6$  is a  $C_1$ - $C_3$  alkyl group.

- 7. The process according to claim 6, wherein the dihydroxylation takes place in the presence of OsO<sub>4</sub> or of RuCl<sub>3</sub>.
- The process according to any of claims 6 and 7, wherein a compound of formula (IVa), its enantiomers or mixtures 8. 50 thereof, is dihydroxylated in the presence of RuCl<sub>3</sub> to yield a compound of formula (II), its enantiomers or mixtures thereof





5

10

wherein

 $R^2,\,R^3,\,R^4$  and  $R^5$  are as they have been defined in any of claims 1 to 3; and  $R^6$  is a  $C_1\text{-}C_3$  alkyl group.

15 9. The process according to any of claims 6 and 7, wherein a compound of formula (IVb), its enantiomers or mixtures thereof, is dihydroxylated in the presence of RuCl<sub>3</sub> to yield a compound of formula (III), its enantiomers or mixtures thereof



#### 30 wherein

 $R^2,\,R^3,\,R^4$  and  $R^5$  are as they have been defined in any of claims 1 to 3; and  $R^6$  is a  $C_1\text{-}C_3$  alkyl group.

- **10.** A compound of formula (IV), (IVa) or (IVb), its stereoisomers, especially its enantiomers, or mixtures thereof as defined in any of claims 6 to 9.
  - 11. A process for the synthesis of a compound of formula (IV), its stereoisomers, especially its enantiomers, or mixtures thereof, comprising the treatment in acidic medium of a compound of formula (V), its stereoisomers, especially its enantiomers, or mixtures thereof

40

35





50

55

wherein

- $R^2$ ,  $R^3$ ,  $R^4$  and  $R^5$  are as they have been defined in any of claims 1 to 3; and each of  $R^7$  and  $R^8$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_4$  alkyl and  $C_6$ - $C_{10}$  aryl; or  $R^7$  and  $R^8$ , together with the carbon atom to which they are attached, form a  $C_2$ - $C_7$  alkylidene group.
- **12.** The process according to claim 11, wherein a compound of formula (V), its stereoisomers, especially its enantiomers, or mixtures thereof, reacts in the presence of p-toluenesulfonic acid.

- 13. A compound of formula (V), its stereoisomers, especially its enantiomers, or mixtures thereof, as defined in claim 11.
- 14. A process for the synthesis of a compound of formula (V), its stereoisomers, especially its enantiomers, or mixtures thereof, characterized in that it comprises

(i) the removal of the trialkylsilyl group from a compound of formula (VII), its stereoisomers, especially its enantiomers, or mixtures thereof, to yield a compound of formula (VI), its stereoisomers, especially its enantiomers, or mixtures thereof

 $R^3O_2$ 

(VI)

(XX)

10

5

15



20

25

(VII) wherein  $R^2$ ,  $R^3$  and  $R^5$  are as they have been defined in any of claims 1 to 3;  $R^6$  is a  $C_1$ - $C_3$  alkyl group;

 $CO_2R$ 

- each of R<sup>7</sup> and R<sup>8</sup> is independently selected from the group consisting of hydrogen, C1-C4 alkyl and C6-C10 aryl; or R<sup>7</sup> and R<sup>8</sup>, together with the carbon atom to which they are attached, form a C<sub>2</sub>-C<sub>7</sub> alkylidene group; and R<sup>9</sup> is a trialkylsilyl group; and

30

(ii) the oxidation of the hydroxyl group of a compound of formula (VI), its stereoisomers, especially its enantiomers, or mixtures thereof.

15. The process according to claim 14, wherein the oxidation is performed in the presence of PCC or IBX.

- 16. A compound of formula (VI), its stereoisomers, especially its enantiomers, or mixtures thereof, as defined in claim 14.
- 35
- 17. A compound of formula (VII), its stereoisomers, especially its enantiomers, or mixtures thereof, as defined in claim 14.
- 18. A process for the synthesis of a compound of formula (VII), as defined in claim 17, its stereoisomers, especially its enantiomers, or mixtures thereof, which comprises reacting a compound of formula (VIII), its enantiomers, or mixtures thereof, with a compound of formula (XX)
- 40

45





wherein

R<sup>2</sup>, R<sup>3</sup>, R<sup>4</sup> and R<sup>5</sup> are as they have been defined in any of claims 1 to 3; and each of R<sup>7</sup> and R<sup>8</sup> is independently selected from the group consisting of hydrogen, C1-C4 alkyl and C6-C10 aryl; or R7 and R8, together with the carbon 55 atom to which they are attached, form a  $C_2$ - $C_7$  alkylidene group;

ÕR<sup>9</sup>

R<sup>9</sup> is a trialkylsilyl group; and

each of the Ar groups is independently selected from among  $C_6$ - $C_{10}$  aryl groups.

(VIII)

19. The process according to claim 18, wherein said compound of formula (VIII), its enantiomers, or mixtures thereof, is obtained by oxidizing, preferably in the presence of IBX, a compound of formula (IX), its enantiomers, or mixtures thereof

5

10



#### 15 wherein

R<sup>2</sup>, R<sup>3</sup> and R<sup>4</sup> are as they have been defined in any of claims 1 to 3; each of  $R^7$  and  $R^8$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_4$  alkyl and  $C_6$ - $C_{10}$  aryl; or  $R^7$  and  $R^8$ , together with the carbon atom to which they are attached, form a  $C_2$ - $C_7$  alkylidene group; and R<sup>9</sup> is a trialkylsilyl group.

20

- 20. The process according to claims 18 and 19, wherein the transformation of a compound of formula (IX), its enantiomers, or mixtures thereof, into a compound of formula (VII), its stereoisomers, especially its enantiomers, or mixtures thereof, is performed without isolating said compound of formula (VIII).
- 25 21. A compound of formula (VIII), its enantiomers, or mixtures thereof, as defined in claim 18.
  - 22. A compound of formula (IX), its enantiomers, or mixtures thereof, as defined in claim 19.
  - 23. A process for the synthesis of a compound of formula (IX), its enantiomers, or mixtures thereof, characterized in that it comprises the dihydroxylation, preferably in the presence of OsO<sub>4</sub>, of a compound of formula (X), its enantiomers, or mixtures thereof

35

30





40



wherein

45  $R^2$ ,  $R^3$  and  $R^4$  are as they have been defined in any of claims 1 to 3; each of  $R^7$  and  $R^8$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_4$  alkyl and  $C_6$ - $C_{10}$  aryl; or  $R^7$  and  $R^8$ , together with the carbon atom to which they are attached, form a  $C_2$ - $C_7$  alkylidene group; and R<sup>9</sup> is a trialkylsilyl group.

- 50 24. A compound of formula (X), its enantiomers, or mixtures thereof, as defined in claim 23.
  - 25. A process for the synthesis of a compound of formula (X), its enantiomers, or mixtures thereof, characterized in that it comprises the acetalization or hemiacetalization, of a compound of formula (XI), its enantiomers, or mixtures thereof



- 5
- <sup>10</sup> in the presence of a compound of formula  $(R_7)(R_8)C=O$  or hydrates or aryl or alkyl acetals or hemiacetals thereof wherein

 $R^2$ ,  $R^3$  and  $R^4$  are as they have been defined in any of claims 1 to 3; and each of  $R^7$  and  $R^8$  is independently selected from the group consisting of hydrogen,  $C_1$ - $C_4$  alkyl and  $C_6$ - $C_{10}$  aryl; or  $R^7$  and  $R^8$ , together with the carbon atom to which they are attached, form a  $C_2$ - $C_7$  alkylidene group; and  $R^9$  is a trialkyleiking around

- <sup>15</sup> R<sup>9</sup> is a trialkylsilyl group.
  - **26.** A compound of formula (XI), its enantiomers, or mixtures thereof, as defined in claim 25.
- 27. A process for the synthesis of a compound of formula (XI), its enantiomers, or mixtures thereof, characterized in
   20 that it comprises the following steps (i) reacting a compound of formula (XXI) in the presence of a compound of formula PY<sub>3</sub>, and subsequently adding a compound of formula (XVI), its stereoisomers or mixtures thereof, to yield a compound of formula (XV), its stereoisomers or mixtures thereof



30

35

40

wherein

 $R^2$ ,  $R^3$  and  $R^4$  are as they have been defined in any of claims 1 to 3; each of the groups Y is independently selected from among  $C_6$ - $C_{10}$  aryl groups or  $C_1$ - $C_6$  alkyl groups;

(ii) epoxidizing, preferably with meta-chloroperbenzoic acid (m-CPBA), said compound of formula (XV), to obtain a compound of formula (XIV), its stereoisomers, especially its enantiomers, or mixtures thereof,



45







wherein

 $R^2$ ,  $R^3$  and  $R^4$  are as defined above;

(iv) introducing a trialkylsilyl group into said compound of formula (XIII) to obtain a compound of formula (XII), its stereoisomers, especially its enantiomers, or mixtures thereof

5



10

20



15 wherein

R<sup>2</sup>, R<sup>3</sup> and R<sup>4</sup> are as defined above; and
R<sup>9</sup> is a trialkylsilyl group;
and
(v) dihydroxylating said compound of formula (XII).

28. A compound of formula (XV), as defined in claim 27, its stereoisomers, or mixtures thereof.

29. A compound of formula (XIV), as defined in claim 27, its stereoisomers, especially enantiomers, or mixtures thereof.

**30.** A compound of formula (XIII), as defined in claim 27, its stereoisomers, especially enantiomers, or mixtures thereof.

31. A compound of formula (XII), as defined in claim 27, its stereoisomers, especially enantiomers, or mixtures thereof.

32. A process for preparing zaragozic acid and derivatives thereof of formula (XXVI), their stereoisomers, especially
 30 enantiomers, or mixtures thereof, characterized in that it comprises the following steps

(i) reacting in acidic medium a compound of formula (II), its enantiomers or mixtures thereof, or a compound of formula (III), its enantiomers or mixtures thereof, or a mixture of compounds of formula (II) and (III), as defined in any of claims 1 to 3; to obtain a compound of formula (I), its enantiomers or mixtures thereof, as defined in any of claims 1 to 3;

(ii) hydrolyzing in basic medium the ester groups of said compound of formula (I), its enantiomers or mixtures thereof, to provide a compound of formula (XXV), its enantiomers or mixtures thereof

40

35



45

50

wherein

R<sup>2</sup> is as defined in claim 1;

and

(iii) reacting said compound of formula (XXV), its enantiomers or mixtures thereof, with a compound of formula (XXII) in the presence of a base, to obtain a compound of formula (XXVI), its enantiomers or mixtures thereof



- $R^1$  is selected from the group consisting of  $C_1$ - $C_{20}$  alkyl or  $C_1$ - $C_{20}$  alkenyl, which are unsubstituted or substituted with at least one group which is selected from the group consisting of  $C_1$ - $C_4$  alkyl; and/or a group in the end position which is selected from the group consisting of  $C_6$ - $C_{10}$  aryl; Z is selected from the group consisting of hydroxyl and alkoxyl; and  $R^2$  is as defined in claim 1.
- 33. Use of a compound of formula (I), (II), (II), (IV), (V), (VI), (VII), (VII), (VIII), (VIII), (VIII), (VIII), (IX), (IX),

~	_
2	5
-	~

# INTERNATIONAL SEARCH REPORT

International application No. PCT/ ES 2009/070140

A. CLASSIFICATION OF SUBJECT MATTER					
see extra sheet According to International Patent Classification (IPC) or to both national classification and IPC					
B. FIELDS SEARCH					
Minimum documentation searched (classification system followed by classification symbols) C07D, C07C					
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched					
Electronic data base co	onsulted during the international search (name of data	base and, where practicable, sear	ch terms used)		
INVENES, EPOD	OC, WPI, NPL, XPESP, MEDLINE, BIOS	IS, CAS.			
C. DOCUMENTS CO	NSIDERED TO BE RELEVANT				
Category*	Citation of document, with indication, where appro	priate, of the relevant passages	Relevant to claim No.		
X	TROST, B. M. et al. "Bromomalonates as Reagents. Transfer Alkylations". Journal o American Chemical Society, 1976, Volum Number 5, pages 1204-1212. See page 120 table I.	Synthetic f the e 98, 55,	28		
x	NAKAMURA, S. et al. "Total synthesis of zaragozic acid C by an aldol-based strategy Tetrahedron, 2005, Volume 61, pages 110 11106. See abstract; page 11079, scheme 1 page 11085, scheme 10.	y". 78- ;	33		
A	EVANS, D. A. et al. "Asymmetric Synthes the Squalene Synthase Inhibitor Zaragozic C". Journal of the American Chemical Soc 1994, Volume 116, pages 12111-12112. So page 12112, scheme 2.	1-32			
Further documents	s are listed in the continuation of Box C.	See patent family annex.			
* Special categories "A" document defining to be of particular "E" earlier document b	of cited documents: "T" g the general state of the art which is not considered relevance. ut published on or after the international filing date	later document published after priority date and not in conflict understand the principle or theory	the international filing date or with the application but cited to underlying the invention		
<ul> <li>"L" document which may throw doubts on priority claim(s) or which is "X" cited to establish the publication date of another citation or other special reason (as specified)</li> <li>"O" document referring to an oral disclosure use, exhibition, or other "Y" means</li> <li>"P" document published prior to the international filing date but later than the priority date claimed</li> </ul>		document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other documents, such combination being obvious to a person skilled in the art			
"&"		document member of the same pa	tent family		
15 July 2000 (15.07.2000)		(17/08/2009)	onai search report		
15 July 2009     (15.07.2009)       Name and mailing address of the ISA/		Authorized officer			
O.E.P.M.		N. Mart	ín Laso		
Paseo de la Castellana, 75 28071 Madrid, España. Facsimile No. 34 91 3495304		   Telephone No. +34 91 349 3	2 78		

Form PCT/ISA/210 (second sheet) (July 2008)

# INTERNATIONAL SEARCH REPORT

International application No.

PCT/ES 2009/070140

C (continuation).	DOCUMENTS CONSIDERED TO BE RELEVANT				
Category*	Citation of documents, with indication, where appropriate, of the relevant passages	Relevant to claim No.			
A	TOMOOKA, K. et al."Stereoselective Total Synthesis of Zaragozic Acid A based on an acetal [1,2] Wittig Rearrangement". Angewandte Chemie International Edition in English, 2000, Volume 39, Number 24, pages 4502-4505. See page 4503, scheme 3 and 5.	1-32			
A	MARUYAMA, Y. et al. "Ruthenium-catalysed reductive cleavage of allylic esters with formic acid and triethylamine. Application to short- step synthesis of $\alpha$ -hydroxy acids". Journal of Organometallic Chemistry, 1994, Volume 473, pages 257-264. See page 259, column 2, compound 10.	29			
A	WATANABLE, H. et al. "A facile route to 3,7-cis- disubstituted cycloocta-1,5-diene-1,2,5,6- tetracarboxylates through photochemical [2+2] cycloaddition of 3-substituted cyclobutene-1,2- dicarboxylates and thermal isomerization". Chemical Communications, 1999, pages 1753-1754. See page 1754, scheme 3, compound 5.	31			

Form PCT/ISA/210 (continuation of second sheet) (July 2008)

# INTERNATIONAL SEARCH REPORT

International application No.

PCT/ ES 2009/070140

#### CLASSIFICATION OF SUBJECT MATTER

C07D 493/08 (2006.01) C07D 493/10 (2006.01) C07D 307/20 (2006.01) C07D 317/30 (2006.01) C07C 69/60 (2006.01) C07D 303/40 (2006.01)

Form PCT/ISA/210 (extra sheeet) (July 2008)

#### **REFERENCES CITED IN THE DESCRIPTION**

This list of references cited by the applicant is for the reader's convenience only. It does not form part of the European patent document. Even though great care has been taken in compiling the references, errors or omissions cannot be excluded and the EPO disclaims all liability in this regard.

#### Patent documents cited in the description

- WO 9316066 A [0003]
- WO 9317557 A [0003]

#### Non-patent literature cited in the description

- Nadin, A.; Nicolaou, K. C. Angew. Chem. Int. Ed. Engl., 1996, vol. 35, 1622-1656 [0002]
- Nicolaou, K. C.; Yue, E. W.; Naniwa, Y.; De Riccardis, F.; Nadin, A.; Leresche, J. E.; La Greca, S.; Yang, Z. Angew. Chem. Int. Ed. Engl., 1994, vol. 33, 2184-2187 [0005] [0049]
- Nicolaou, K. C.; Nadin, A.; Leresche, J. E.; La Greca, S.; Tsuri, T.; Yue, E. W.; Yang, Z. Angew. Chem. Int. Ed. Engl., 1994, vol. 33, 2187-2190 [0005]
- Nicolaou, K. C.; Nadin, A.; Leresche, J. E.; Yue,
   E. W.; La Greca, S. Angew. Chem. Int. Ed. Engl., 1994, vol. 33, 2190-2191 [0005]
- Nicolaou, K. C.; Yue, E. W.; La Greca, S.; Nadin, A.; Yang, Z.; Leresche, J. E.; Tsuri, T.; Naniwa, Y.; De Riccardis, F. Chem. Eur. J., 1995, vol. 1, 467-494 [0005]
- Nicolaou, K. C.; Sorensen, E. J. Classics in Total Synthesis. VCH Publishers, 1996, 673-709 [0005]
- Stoermer, D. ; Caron, S. ; Heathcock, C. H. J. Org. Chem., 1996, vol. 61, 9115-9125 [0005] [0058]
- Caron, S.; Stoermer, D.; Mapp, A. K.; Heathcock, C. H. J. Org. Chem., 1996, vol. 61, 9126-9134
   [0005]
- Tomooka, K.; Kikuchi, M.; Igawa, K.; Suzuki, M.; Keong, P. -H.; Nakai, T. Angew. Chem. Int. Ed., 2000, vol. 39, 4502-4505 [0005]
- Carreira, E. M.; Du Bois, J. J. Am. Chem. Soc., 1994, vol. 116, 10825-10826 [0005] [0049] [0058]
- Carreira, E. M.; Du Bois, J. J. Am. Chem. Soc., 1995, vol. 117, 8106-8125 [0005]
- Evans, D. A.; Barrow, J. C.; Leighton, J. L.; Robichaud, A. J.; Sefkow, M. J. Am. Chem. Soc., 1994, vol. 116, 12111-12112 [0005] [0049] [0058]
- Armstrong, A.; Jones, L. H.; Barsanti, P. A. Tetrahedron Lett., 1998, vol. 39, 3337-3340 [0005]
   [0049]
- Armstrong, A.; Barsanti, P. A.; Jones, L. H.; Ahmed, G. J. Org. Chem., 2000, vol. 65, 7020-7032 [0005]
- Kataoka, O.; Kitagaki, S.; Watanabe, N.; Kobayashi, J.; Nakamura, S.; Shiro, M.; Hashimoto, S. *Tetrahedron Lett.*, 1998, vol. 39, 2371-2374 [0005] [0060]

- Nakamura, S.; Hirata, Y.; Kurosaki, T.; Anada, M.; Kataoka, O.; Kitagaki, S.; Hashimoto, S. *Angew. Chem. Int. Ed.*, 2003, vol. 42, 5351-5355 [0005]
   [0060]
- Sato, H. ; Nakamura, S. ; Watanabe, N. ; Hashimoto, S. Synlett, 1997, 451-454 [0005]
- Nakamura, S.; Sato, H.; Hirata, Y.; Watanabe, N.; Hashimoto, S. *Tetrahedron*, 2005, vol. 61, 11078-11106 [0005]
- Nakamura, S. Chem. Pharm. Bull., 2005, vol. 53, 1-10 [0005]
- Xu, Y.; Johnson, C. R. Tetrahedron Lett., 1997, vol. 38, 1117-1120 [0006]
- Smith, M. B.; March, J. March's Advanced Organic Chemistry. John Wiley & Sons, 2001, 1048-1051 [0019] [0054]
- Shing, T. K. M.; Tai, V. W.-F.; Tam, E. K. W. Angew. Chem. Int. Ed. Engl., 1994, vol. 33, 2312-2313
   [0019]
- Shing, T. K. M.; Tai, V. W.-F.; Tam, E. K. W.; Chung, I. H. F.; Jiang, Q. Chem. Eur. J., 1996, vol. 2, 50-57 [0019]
- Plietker, B.; Niggemann, M. Org. Lett., 2003, vol. 5, 3353-3356 [0019]
- Lu, W.; Zheng, G.; Cai, J. Tetrahedron, 1999, vol. 55, 4649-4654 [0024]
- Greene, T. W.; Wuts, P. G. M. Greene's Protective Groups in Organic Synthesis. John Wiley & Sons, 2007, 306-321 [0024] [0045]
- Kocienski, P. J. Protecting Groups. Thieme, 2000, 188-230 [0027]
- Frigerio, M.; Santagostino, M.; Sputore, S.; Palmisano, G. J. Org. Chem., 1995, vol. 60, 7272-7276 [0028]
- Frigerio, M.; Santagostino, M. Tetrahedron Lett., 1994, vol. 35, 8019-8022 [0028]
- Corey, E. J.; Palani, A. Tetrahedron Lett., 1995, vol. 36, 3485-3488 [0028]
- Wirth, T. Angew. Chem. Int. Ed., 2001, vol. 40, 2812-2814 [0028]
- More, J. D.; Finney, N. S. Org. Lett., 2002, vol. 4, 3001-3003 [0028]

WO 9404144 A **[0003]** 

- Villa, M. J.; Warren, S. J. Chem. Soc. P. T 1, 1994, vol. 12, 1569-1572 [0031]
- Shi, Y. -J.; Hughes, D.L.; McNamara, J.M. Tetrahedron Lett., 2003, vol. 44, 3609-3611 [0032] [0034]
- Mukaiyama, T.; Shintou, T.; Fukumto, K. J. Am. Chem. Soc., 2003, vol. 125, 10538-10539 [0032] [0034]
- Kolb, H. C.; Van Nievwenhze, M. S.; Sharpless, K. B. Chem. Rev., 1994, vol. 94, 2483 [0042]
- Konno, H.; Makabe, H.; Tanaka, A.; Oritani, T. *Tetrahedron,* 1996, vol. 52, 9399-9408 [0045]
- Maryanoff, B. E.; Reitz, A. B. Chem. Rev., 1989, vol. 89, 863-927 [0048]
- Trost, B. M.; Melvin, L. S. Jr. J. Am. Chem. Soc., 1976, vol. 98, 1204-1212 [0048]
- Tomooka, K.; Kikuchi, M.; Igawa, K.; Suzuki, M.; Keong, P. - H.; Nakai, T. Angew. Chem. Int. Ed., 2000, vol. 39, 4502-4505 [0049]
- Evans et al. Tetrahedron Lett., 1993, vol. 34, 8403 [0049]

- Katsuki, T. Adv. Synth. Catal., 2002, vol. 344, 131-147 [0050]
- Wang, Z. X.; Tu, Y.; Frohn, M.; Zhang, J. R; Shi, Y. J. Am. Chem. Soc., 1997, vol. 119, 11224-11235 [0050]
- Dalla, V.; Catteau, J. P. Tetrahedron, 1999, vol. 55, 6497-6510 [0052]
- Greene, T. W.; Wuts, P. G. M. Greene's Protective Groups in Organic Synthesis. John Wiley & Sons, 2007 [0052]
- Kocienski, P.J. Protecting Groups. Thieme, 2000, 393-425 [0057]
- Phytochemistry, 1995, vol. 38, 1169-1173 [0061]
- *Tetrahedron,* 1993, vol. 49, 10643-10654 [0061]
- Carreira et al. J. Am. Chem. Soc., 1995, vol. 117, 8106-8125 [0062]
- Greene ; Wuts. Protective Groups in Organic Synthesis. John Wiley & Sons, Inc, 2007 [0075]