ELEMENT GEOCHEMISTRY OF CHEROKEE GROUP COALS (MIDDLE PENNSYLVANIAN) FROM SOUTH-CENTRAL AND SOUTHEASTERN IOWA

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ABSTRACT

Middle Pennsylvanian Cherokee Group coals from southcentral and southeastern Iowa are typical highsulfur, high-ash coals. These coals have an arithmetic mean sulfur content of 5.8 percent and a mean ash content of 15.9 percent. Apparent rank for most samples is high-volatile C bituminous coal. The relatively high contents of sulfur and 23 other elements in Iowa Cherokee Group coals are related to near neutral pH conditions (6-8) in the depositional and early diagenetic environments, and to postdepositional epigenetic sphalerite/calcite/pyrite/ kaolinite/barite mineralization. Changes from an aluminosilicate- to a sulfide-element association for U, Mo, Cr, and V, and an increase in element content for U, Mo, Cr, V, Na, Mg, and K in stratigraphically higher coals are thought to be related to differences in depositional environments of the coal-associated rocks, which change from predominantly terrestrial in the Lower Cherokee Group, to predominantly marine in the upper part of the Upper Cherokee Group. Coals overlain by marine, phosphatic, black shale lithologies have the highest content of U. Mo. Aq. Sb. Se, and V.

INTRODUCTION

Pennsylvanian coal-bearing rocks underlie all or parts of 44 counties in southern and western Iowa, an area of 20,000 approximately square miles (Landis and Van Eck, 1965). Identified coal resources, in beds at least 14 inches thick, total about 6.5 billion tons (Landis and Van Eck, 1965). Estimated hypothetical coal resources are an additional 14 billion tons for a total of 20.5 billion tons of identified and hypothetical coal resources (Averitt, 1975).

An essential part of any complete coalresource evaluation is a chemical characterization of the coal by proximate. ultimate, and minor and trace element analyses. In order to chemically characterize Iowa's coal resources, a cooperative program was initiated in 1973 between the Iowa Geological Survey and the U.S. Geological Survey. From 1973 to 1978, 106 coal samples were collected by the Iowa Geological Survey surface from drill cores and and underground coal mines in 11 southcentral and southeastern Iowa counties (figure 1), which incorporates the area of present-day mining. One hundred and five of the 106 coal samples represent beds from the Cherokee Group, (Landis and Van Eck, 1965), (figure 2). One sample (D192373) is from the Marmaton Group. Sample numbers, locations, and descriptive information for the 106 coal samples are listed in table 1. This report lists, statistically summarizes, and briefly discusses the proximate and ultimate analyses, heatof-combustion. forms-of-sulfur. and ash-fusion-temperature determinations. and minor- and trace-element composition of these 106 Iowa coal samples.

Analyses of samples D176169-D176200 have been published in Swanson and others (1976, tables 19A-19E). Ash percent and zinc and cadmium contents in both ash and whole coal for 68 samples (D166027-D166043, D176169-D176200, and D179838-D179856) from this report are also listed and discussed in Hatch, Avcin, and others (1976). Other reports listing elemental analyses of Iowa coals include Zubovic and others (1967, 23 bench samples, 2 locations) and Abernethy and others (1969a and 1969b, 3 samples, 3 locations).

ANALYTICAL METHODS AND RESULTS

Proximate and ultimate analyses, heatof-combustion, air-dried-loss, formsfree-swelling-index, of-sulfur, and determinations ash-fusion-temperature for 90 coal samples were provided by the U.S. Bureau of Mines (now a part of the U.S. Department of Energy), Pittsburgh. Pennsylvania. Analytical procedures for these analyses are described in U.S. Office of Coal Research Analyses of 106 Iowa coal (1967). samples for ash content, 34 to 36 major and minor oxides and trace elements in the laboratory ash (ashed at 525°C). and 7 trace elements in whole coal were provided by the U.S. Geological Survey, Denver, Colorado. Analytical procedures used by the U.S. Geological Survey are described in Swanson and Huffman (1976). Chemical analyses and physical tests performed by the two laboratories are listed in table 2. The sequence of sample preparation and chemical analyses is shown in figure 3. Table 3 lists the results of the U.S. Bureau of Mines analyses; table 4 the results of the U.S. Geological Survey analyses of coal ash; and table 5 the results of the U.S. Geological Survey analyses of 42 elements in whole coal. Whole-coal data in table 5 for all elements except As, F, Hg, Sb, Se, Th, and U were calculated from analyses of coal ash.

Analytical results from the six-step emission spectrographic technique are identified with geometric brackets whose boundaries are (in ppm) 12, 8.3,



Figure 1. Index map of south-central and southeastern Iowa showing coal sample collection sites.



Figure 2. Stratigraphic nomenclature, after Landis and Van Eck (1965).

Table 1. Identification numbers, locations, and brief descriptions of 106 Middle Pennsylvanian coal samples from south-central and southeastern Iowa

[Sample D192373 is from the Marmaton Group; all other samples are from the Cherokee Group. Coal-zone designations are from unpublished Iowa Geological Survey data. -- = not applicable. One foot = 0.305 meters]

U.S. Geological Survey laboratory number	Iowa Geological Survey sample number	U.S. Bureau of Mines laboratory number	Location	Index map key	Sample type	lowa Geological Survey core-hole number	Depth interval or bed thickness sampled (feet)	Coa 1 zone	Notes
D166027	1-A-73	**	T.74N., R.16W., Mahaska County	A	channe]		4	4	
D166028	[-8-73		do	A	do		4	4	
D166029	1-D-73	K34084	dodo	Δ	run-of- mine		4	4	
D166030	2-A-73		T.74N., R.17W., Mahaska County	B	channel		5	4	
D166031	2-B-73	K34085	do	B	do		5	4	
D166032	3-A-73		do	С	do		4	4	
D166033	3-B-73		dodo	C	do		4	4	
D166034	3-D-73	K34086	dodo	C	run-of- mine		4	4	
D166035	4-1-73		T.73N., R.18W., Monroe County	D	channe1		4	4	
D166036	4-B-73	K34087	do	D	do		4	4	
D166037	5-A-73	K34088	T.72N., R.19W., Monroe County	ε	run-of- mine		4.5	4	
D166038	5-B-73		d0d0	E	do		4.5	4	
D166039	6-A-73	K34089	T.73N., R.20W., Lucas County	F	do		4	4	
D166040	6-B-73		do	F	do		4	4	
D166041	7-A-73		T.73N., R.15W., Wapello County	G	channel		3	3	

U.S. Geological Survey laboratory number	lowa Geological Survey sample number	U.S. Bureau of Mines laboratory number	Location	Index map key	Sample type	Iowa Reological Survey core-hole number	Depth interval or bed thickness sampled (feet)	Coal zone	Notes
D166042	7-8-73		do	• G	do	••	3	3	
D166043	7-D-73	K34090	dodo	G	run-of- mine		3	3	
D176169	CP7-5	K57858	SE,SE,SE, sec. 36, T.71N., R.14W., Wapello County	7	core	CP-7	109.1-110.2	7	
D176170	CP7-15	K57859	do	. 7	-do-	-do-	149.4-150.1	5	
D176171	CP7-22	K57860	do	. 7	-do-	-do-	165.3-165.9	4	
D176172	CP7-26	K57861	dodo	. 7	-do-	-do-	193.2-194.5	4	
D176173	CP7-30	K57862	do	- 7	-do-	-do-	206.2-207.7	4	
D176174	CP7-35	K57863	dodo	- 7	-do-	-do-	229.1-234.2	4	
D176175	CP7-37	K57864	dodo	. 7	-do-	-do-	252.2-253.8	3	
D176176	CP7-46	K57865	do	- 7	-do-	-do-	288.1-289.5	2	
D176177	CP7-50	K57866	dodo	. 7	-do-	-do-	323.8-324.7	1	
D176178	CP10-26	K61200	SE,SW,SW,SW, sec. 6, T.68N., R.17W., Appanoose County	, 10	-do-	CP-10	240.1-241.0	9	0.25-foot-thick parting 0.5 foot from top
D176179	CP10-28	K61201	do	- 10	-do-	do-	253.1-254.3	8	
D176180	CP10-31	K61202	do	- 10	-do-	do-	279.0-280.2	7	
D176181	CP10-34	K61203	dodo	- 10	-do-	do-	306.9-308.7	6	
D176182	CP10-40	K61204	do	- 10	-do-	do-	361.5-362.3	4	
D176183	CP-10-46	K61205	do	- 10	-do-	do-	387.7-389.1	4	

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Table 1. Identification numbers, locations, and brief descriptions of 106 Middle Pennsylvanian coal samples from south-central and southeastern Iowa--continued

U.S. Geological Survey laboratory number	Iowa Geological Survey sample number	U.S. Bureau of Mines laboratory number	Location	Index map key	Sample type	Iowa Geological Survey core-hole number	Depth interval or bed thickness sampled (feet)	Coal zone	Notes
D176184	CP10-50	K61206	do	10	-do-	do-	406.7-407.4	4	
D176185	CP10-58	K61207	dodo	10	-do-	do-	425.2-426.0	3	
0176186	CP-10-67	K61208	do	10	-do-	do-	465.9-466.8	1	
0176187	CP17-9	K61209	NE,NE,NW,NW, sec. 34, T.74N. R.25W. Mahaska County	17	-do-	CP-17	138.1-142.8	2	
D176188	CP20-13	K57867	SE,SE,NW, sec. 18, T.72N., R.14W., Wapello County	20	-do-	CP-20	128.6-133.6	2	
D176189	CP21-2	K58854	NW,NW,NW, sec. 18, T.71N., R.14W., Wapello County	21	-do-	CP-21	76.7-79.6	8	0.5-foot-thick parting, 0.3 foot from base; clay dike in upper por- tion
D176190	CP21-8	K58855	dodo	21	-do-	do-	131.9-133.3	7	0.5-foot-thick parting 0.25 foot from base
D176191	CP21-14	K58856	dodo	21	-do-	do-	167.1-167.9	6	
D176192	CP21-18	K58857	do	21	-do-	do-	183.8-186.4	5	
D176193	CP21-20	K58858	dodo	21	-do-	do-	191.7-195.3	5	0.5-foot-thick parting, 0.9 foot from top; boney below parting
D176194	CP21-22	K58859	dodo	21	-do-	do-	197.9-199.0	5	0.2-foot-thick parting 0.3 foot from base
D176195	CP21-26	K58860	do	21	-do-	do-	215.7-217.1	4	
D176196	CP21-29	K58861	d0	21	-do-	do-	243.8-246.2	4	
D176197	CP21-31	K58862	dodo	21	-do-	do-	263.0-264.2	4	
D176198	CP21-34 `	K58863	d0	21	-do-	do-	316.0-319.5	3	boney in lower two-thirds

Table 1. Identification numbers, locations, and brief descriptions of 106 Middle Pennsylvanian coal samples from south-central and southeastern Iowa-continued

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U.S. Geological Survey laboratory `number	Iowa Geological Survey sample number	U.S. Bureau of Mines laboratory number	Location	Index map key	Sample type	Iowa Geological Survey çore-hole number	Nepth interval or bed thickness sampled (feet)	Coal zone	Notes
D176199	CP21-36	K58864	do	21	-do-	do-	322.0-326.1	2	
0176200	CP21-40		do	21	-do-	do-	357.9-358.4	1	
D179838	CP24-3	Z-598	NW,NE,SE, sec. 1, T.67N., R.14W., Davis County	24	-do-	CP-24	174.1-176.2	8	
D179839	CP24-5U		do	24	-do-	do-	179.2-179.6	8	
D179840	CP24-5L	Z-599	do	24	-do-	do-	179.9-181.0	8	
D179841	CP24-17	Z-600		24	-do-	do-	238.6-241.3	6	
D179842	CP24-9		dodo	24	-do-	do-	210.4-211.0	7	severely disturbed by clay dike
D179843	CP30-7	Z-601	NE,NE,NW, sec. 17, T.72N., R.15W., Wapello County	30	-do-	CP-30	65.7-66.6	4	
D179844	CP30-9	Z-602	do	30	-do-	do-	78.7-79.5	4	
D179845	CP30-17		do	30	-do-	do-	111.9-112.5	3	
D179846	CP30-20	Z-603	do	30	-do-	do-	123.7-124.9	2	
D179847	CP32-2	Z-604	SW,SW,SW, sec. 18, T.72N., R.15W., Wapello County	32	-do-	CP-32	24.6-25.7	7	core loss at base, bed may be thicker.
D179848	CP32-6	Z-605	dodo	32	-do-	do-	59.2-60.2	6	
D179849	CP32-20		dodo	32	-do-	do-	117.0-117.4	4	boney
D179850	CP32-28	Z-606	do	32	-do-	do-	177.5-180.7	2	
D179851	CP28-4	Z-607	NW,SE,NE, sec. 36, T.72N., R.15W., Wapello County	28	-do-	CP-28	55.8-56.9	7	

Table 1. Identification numbers, locations, and brief descriptions of 106 Hiddle Pennsylvanian coal samples from south-central and southeastern lowa--continued

U.S. Geological Survey laboratory number	Iowa Geological Survey sample number	U.S. Bureau of Mines laboratory number	Location	Index map key	Sample type	Iowa Geological Survey core-hole number	Depth interval or bed thickness sampled (feet)	Coa) zone	Notes
D179852	CP28-11		do	29	-do-	do-	96.7-97.5	6	
D179853	CP28-24	Z-608	do	28	-do-	do-	153.7-154.7	4	
D179854	CP28-31	Z-609	do	28	-do-	do-	181.2-184.4	4	two partings 0.1 and 0.3 feet thick in lower half of bed
D179855	CP28-35	Z-610	do	29	-do-	do-	211.4-217.7	3	
D179856	CP28-38	Z-611	do	28	-do-	do-	229.4-231.2	2	0.4-foot-thick parting 0.4 foot from top
D185601	C-AN	K-69871	sec. 24, T.74N., R.17W., Mahaska County	н	channel		4.0	4	
D185602	C-AA	K-69872	do	н	do		4.0	4	
D185603	H-B1	K-69873	sec. 19, T.75N., R.16W., Mahaska County	t	do		2.7	4	
D185604	H-B2	K-69874	do	I	do		2.7	4	
D185605	H-B3	K-69875	do	I	do		2.7	4	
D185606	E-A	K-69876	sec. 21, T.77N., R.19W., Marion County	J	do		2.2	4	
D185609	E1-A	K-69877	do	J	do		4.0	4	
D185610	E1-8	K-69878	do	J	do		4.0	4	
D185611	D-A	K-69879	sec. 7, T.77N., R.20W., Marion County	К	do		2.0	4	
D185612	D-8	K-69880	do	ĸ	do		3.5	4	

Table 1. Identification numbers, locations, and brief descriptions of 106 Middle Pennsylvanian coal samples from south-central and southeastern Iowa--continued

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U.S. Geological Survey laboratory number	Iowa Geological Survey sample number	U.S. Bureau of Mines laboratory number	Location	Index map key	Sample type	lowa Geological Survey core-hole number	Depth interval or bed thickness sampled (feet)	Coal zone	Notes
D185613	0-8B	K-69881	dodo	ĸ	do		3,5	4	
D186062	I - B	K-69959	sec. 21, T.73N., R.20W., Lucas County	F	do		4.5	4	
D186063	1- C	K-69960	do	F	do		4.5	4	
D186064	1-D	K-69961	dodo	F	do		4.5	4	
D186065	ST-1	K-69962	sec. 10, T.74N., R.17W., Mahaska County	8	do		3.7	4	
D186066	EM-1	K-69963	sec. 11, T.74N., R.17W., Mahaska County	L	do		3.2	3	
D186067	CP3-18	K-69964	SW,SW,SW, sec. 30, T.71N., R.11W., Jefferson County	3	core	CP-3	145.5-147.5	3a	rider coal associated with but not equivalent to coal zone 3
D186068	CP3-21	K-69965	do	3	-do-	-do-	162.0-165.0	2	
D186069	CP6-18	K-69966	NW,SW,NW, sec. 8, T.7ON., R.12W., Davis County	6	-do-	CP-6	97.5-101.0	4	lower 0.3 foot boney
D186070	CP11-7	K-69967	NW,NW,NE,NE sec. 1, T.68N., R.16W., Appanoose County	11	-do-	CP-11	347.2-351.7	2	1.5-foot-thick coal ball 1.3 feet from top
0186071	CP13-5	K-69968	SE,NE,SE, sec. 1, T.7ON., R.16W., Appanoose County	13	-do-	CP-13	264.0-266.7	3	
D186072	CP15-7	K-69969	NE,SE,SE, sec. 30, T.72N., R.15W., Wapello County	15	-do-	CP-15	142.8-146.3	2	upper 0.7 foot boney
D186073	CP20-11	K-69970	SW,SE,NW, sec. 18, T.72N., R.14W., Wapello County	20	-do-	CP-20	115.1-117.3	3	
D186074	CP38-2	K-69971	SW,NW,NE, sec. 5, T.74N., R.15W., Mahaska County	38	-do-	CP-38	82.6-85.0	3	minor sphalerite in cleats

Table 1. Identification numbers, locations, and brief descriptions of 106 Middle Pennsylvanian coal samples from south-central and southeastern Iowa--continued

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U.S. Geological Survey laboratory number	Iowa Geological Survey sample number	U.S. Bureau of Mines laboratory number	Location	Index map key	Sample type	lowa Geological Survey core-hole number	Depth interval or bed thickness sampled (feet)	Coal zone	Notes
D186075	L2-1	K-69972	sec. 8, T.72N., R.19W., Monroe County	E	channel		6.2	- 4	
D186076	L2-3	K-69973	dodo	Ε	do		5.8	4	
D192368	CP23-8	K-81517	SE,NE,SE, sec. 36, T.68N., R.15W., Davis County	23	core	CP-23	252.3-254.6	5	contains 0.2 foot thick calcareous pyrite concretion
D192369	CP41-48	K-81518	NE,NE,SE,SW, sec. 36, T.75N. R.20W., Marion County	41	-do-	CP-41	277.5-281.7	3	
0192370	CP41-51	K-81519		41	-do-	do-	297.0-299.2	3	
0192371	CP42-19	K-81520	SE,SE,SE, sec. 1, T.75N., R.21W., Marion County	42	-do-	CP-42	105.1-107.2	4	minor sphalerite in cleats
D192372	CP43-18	K-81521	NE,SW,NE, sec. 6, T.76N., R.21W., Marion County	43	-do-	CP-43	203.8-206.8	2	
D192373	CP22-7	K-81522	SE,S¥,SE, sec. 36, T.70N., R.19W., Appanoose County	2 2	-do-	CP-22	138.6-141.0	-	no zone assigned; Mystic Coal, Marmaton Group
D192374	CP22-48	K-81523	do	22	-do-	do-	419.8-422.3	2	large pyrite blebs
D192375	CP37-57	K-81524	NE,SE,NE, sec. 2, T.72N., R.26W., Clarke County	37	-do-	CP-37	436.8-439.8	4	0.3-foot-thick clay parting 1.5 feet below top
D192376	CP37-66	K-81525	do	37	-do-	do-	483.4-486.1	2	minor sphalerite in cleats
D192377	CP45-2	K-81526	NE,NE,NW, sec. 32, T.78N., R.23W., Polk County	45	-do-	CP-45	57.4-59.8	3	0.8-foot-thick boney zone near middle
D192378	CP47-38	K-81527	SE,NE,NE, sec. 4, T.79N., R.25W., Polk County	47	-do-	CP-47	265.5-269.3	2	sphalerite in cleats near bottom
D192379	CP40-8	K-81528	SW,NW,NE, sec. 6, T.74N., R.17W., Mahaska County	40	-do-	CP-40	78.3-81.6	2	lower 0.4 foot boney

Table 1. Identification numbers, locations, and brief descriptions of 106 Middle Pennsylvanian coal samples from south-central and southeastern Iowa--continued

Major and minor elements (percent)1Proximate analysis (percent)Major and minor elements (percent)1Proximate analysis (percent)Silicon (Si)MoistureAluminum (AI)Volatile matterCalcium (Ca)Fixed carbonMagnesium (Mg)AshSodium (Na)Ultimate analysis (percent)Prostassium (X)Ultimate analysis (percent)Iron (Fe)Hydrogen (H)Titanium (Ti)Hydrogen (H)Phosphorous (P)Carbon (C)Sulfur (S)Nitrogen (N)Oxygen (0)-by differenceSulfur (S)Antimony (Sb)AshArsenic (As)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCobalt (Co)OrganicCobalt (Co)GorteningCadmium (Cd)Initial deformationGermanium (Ge)Initial deformationLanthanum (La)FluidLathanum (La)Free-swelling indexMaganese (Mn)Mercury (Hg)Molybdenum (Md)Scandium (Sc)Scandium (Sc)Selenium (Sc)Silver (Ag)Strontium (Sr)Thorium (Th)Thorium (Th)Vanadum (V)Ytterbium (Yb)Ytterium (Y)Zinc (Zn)Zirconium (Zr)Yinc (Zn)	U.S. Geological Survey	U.S. Bureau of Mines and U.S. Department of Energy
Silicon (Si)MoistureAluminum (A1)Volatile matterCalcium (Ca)Fixed carbonMagnesium (Mg)AshSodium (Na)Ultimate analysis (percent)Potassium (X)Ultimate analysis (percent)Iron (Fe)Hydrogen (H)Phosphorous (P)Carbon (C)Sulfur (S)Nitrogen (N)Oxygen (0)-by differenceSulfur (S)Antimony (Sb)AshArsenic (As)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)SulfateBoron (B)Carbon (C)Cadmium (Cd)SulfateCadmium (Ca)OrganicCobalt (Co)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationSofteningLanthanum (La)Lead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Nicoburum (Sc)Selenium (Sc)Selenium (Sc)Selenium (Sc)Selenium (Sc)Selenium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Zinconium (Zr)	Major and minor elements (percent) ¹	Proximate analysis (percent)
Aluminum (Ai)Volatile matter Fixed carbonMagnesium (Mg)AshSodium (Na)Ultimate analysis (percent)Potassium (K)Ultimate analysis (percent)Iron (Fe)Hydrogen (H)Phosphorous (P)Carbon (C)Sulfur (S)Nitrogen (N)Trace elements (ppm) ² Sulfur (S)Antimony (Sb)AshArsenic (As)Heat of combusion (Btu/lb;cal/kg)Beryllium (Be)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCopper (Cu)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexMagnesse (Mn)Mercury (Hg)Mercury (Hg)Molydenum (Mo)Mitckel (N1)Nickel (N1)Nickel (N1)Yiterbium (Sc)Selenium (Sc)Silver (Ag)Strontium (Y)Yiterbium (Yb)Yiterbium (Yb)Yiterbium (Yb)Yiterbium (Yb)Yiterbium (Yc)Yiterbium (Yc)	Silicon (Si)	Moisture
Calcium (Ca)'Fixed carbon AshNodium (Na)AshPotassium (Na)Ultimate analysis (percent)Iron (Fe)Hydrogen (H)Titanium (Ti)Hydrogen (N)Phosphorous (P)Carbon (C)Sulfur (S)Nitrogen (N)Trace elements (ppm) ² AshAntimony (Sb)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCerium (Ce)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluide (Pb)Initial deformationLanthanum (La)FluidLathanum (Mo)Nickel (N1)Mitogen (Mo)Nickel (N1)Nickel (N1)Nickel (N1)Nickel (N1)Nickel (N1)Nickel (N1)Nickel (N1)Nickel (N1)Yiterbium (N2)Scandium (V)Yiterbium (Y)Yiterbium (Y)Yiterbium (Y)Yiterbium (Y)Yiterbium (Y)Yiterbium (Y)Yiterbium (Y)Yiterbium (Y)Yiterbium (Zr)	Aluminum (Al)	Volatile matter
Magnesium (Mg)AshSodium (Na)Ultimate analysis (percent)Potassium (K)Ultimate analysis (percent)Iron (Fe)Hydrogen (H)Phosphorous (P)Carbon (C)Sulfur (S)Witrogen (N)Trace elements (ppm) ² Sulfur (S)Antimony (Sb)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCerium (Ce)PyriticChoronium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Mercury (Hg)Molydenum (Mo)Mitckel (N1)Noblum (No)Mitckel (N1)Nitolum (Sc)Selenium (Sc)Silver (Ag)Strontium (Tr)Yiterbium (Yb)Yiterbium (Yb)Yiterbium (Yb)Yiterbium (Yc)Yiterbium (Zr)	Calcium (Ca)	Fixed carbon
Sodium (Na)Ultimate analysis (percent)Potassium (K)Ultimate analysis (percent)Iron (Fe)Hydrogen (H)Phosphorous (P)Carbon (C)Sulfur (S)Nitrogen (N)Trace elements (ppm) 2Sulfur (S)Antimony (Sb)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCardmium (Cd)SulfateCromum (Ce)PyrtitcChromium (Cr)OrganicCobalt (Co)SolfatingCopper (Cu)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)FluidLanthanum (La)FluidLathanum (Mo)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nitcket (Ni)Nitcket (Ni)Nitobium (Nb)Neodynium (Nd)Scandium (Sc)Selenium (Sc)Selenium (Sc)Selenium (Th)Ytterbium (Yb)Ytterium (Y)Ytterium (Yb)Ytterium (Y)Yitrium (Yb)Ytterium (Yb)Yitrium (Yb)Ytterium (Zr)Yitroinm (Zr)	Magnesium (Mg)	Ash
Potassium (K)Ultimate analysis (percent)Iron (Fe)Hydrogen (H)Titanium (Ti)Hydrogen (H)Phosphorous (P)Carbon (C)Sulfur (S)Nitrogen (N)Trace elements (ppm) 2Sulfur (S)Antimony (Sb)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCerium (Ce)PyriticChronium (Cr)OrganicCoopper (Cu)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationSofteningFluidLanthanum (La)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Md)Scandium (Sc)Scandium (Sc)Selenium (Sc)Storutium (Sr)Thorium (Th)Varanium (U)Varadium (V)Ytterbium (Y)Yttrium (Y)Zinc (Zn)Zinconium (Zr)	Sodium (Na)	
Iron (Fe) Titanium (Ti) Phosphorous (P) Sulfur (S) Trace elements (ppm) ² Antimony (Sb) Arsenic (As) Barium (Ba) Beryllium (Be) Boron (B) Carbon (C) Nitrogen (N) Oxygen (O)-by difference Sulfur (S) Ash Heat of combusion (Btu/lb;cal/kg) Barium (Cd) Cambun (Cd) Cabalt (Co) Copper (Cu) Fluorine (F) Gallium (Ga) Gallium (Ga) Lathanum (La) Lithium (Li) Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodynium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Vb) Ytterium (Zr)	Potassium (K)	Ultimate analysis (percent)
Titanium (Ti)Hydrogen (H)Phosphorous (P)Carbon (C)Sulfur (S)Nitrogen (N)Trace elements (ppm)2AshAntimony (Sb)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCarbinum (Cc)OrganicCobalt (Co)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ba)Free-swelling indexMangaese (Mn)Mercury (Hg)Molybdenum (No)Nickel (Ni)Nicolum (No)Scandium (Sc)Selenium (Se)Silver (Ag)Stortium (Y)Ytterbium (Y)Ytterbium (Y)Ytterbium (Y)Ytterbium (Y)Ytterbium (Y)Ytterbium (Y)Ytterbium (Y)Ytterbium (Zr)Ytterbium (Zr)	Iron (Fe)	•
Phosphorous (P) Sulfur (S)Carbon (C) Nitrogen (N) Oxygen (O)-by difference Sulfur (S) AshTrace elements (ppm) 2Sulfur (S) AshAntimony (Sb) Arsenic (As) Barium (Ba) Beryllium (Be) Boron (B) Cadmium (Cd)Heat of combusion (Btu/lb;cal/kg) Barium (Ba) Boron (B) Cadmium (Cd)Beryllium (Be) Boron (B) Cadmium (Cd)Forms of sulfur (percent) Organic Cobalt (Co) Copper (Cu)Cobalt (Co) Copper (Cu)Fusibility of ash (temperature °C) Fluorine (F) Gallium (Ga)Initial deformation Germanium (Ce) Lithium (Li)Free-swelling indexManganese (Mn) Mercury (Hg) Molydenum (No) Nickel (Ni) Nickel (Ni) Niobium (Nb) Needymium (Nd) Scandium (Sc) Selenium (Sc) Selenium (Sc) Silver (Ag) Strontium (Y) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yc) Zinc (Zn) Zinconium (Zr)	Titanium (Ti)	Hydrogen (H)
Sulfur (S)Nitrogen (N) Oxygen (O)-by difference Sulfur (S) AshTrace elements (ppm) 2Sulfur (S) AshAntimony (Sb)Heat of combusion (Btu/lb;cal/kg) Barium (Ba) Beryllium (Be) Boron (B) Cadmium (Cd)Boron (B) Cadmium (Cd)Forms of sulfur (percent) Boron (B) Cadmium (Cc)Cobalt (Co) Copper (Cu)Sulfate Pyritic Organic Cobalt (Co)Gallium (Ga) Lanthanum (La) Lithium (Li)Initial deformation Softening FluidLanthanum (La) Lead (Pb) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Sc) Selenium (Sc) Stontium (Y) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yc)	Phosphorous (P)	Carbon (C)
Oxygen(0)-by differenceTrace elements (ppm)2AsiAntimony (Sb) Arsenic (As) Barium (Ba) Beryllium (Be) Cadmium (Cd)Heat of combusion (Btu/lb;cal/kg)Barium (Ba) Boron (B) Cadmium (Cd)Forms of sulfur (percent)Boron (B) Cadmium (Cd)Sulfate Pyritic OrganicCobalt (Co) Copper (Cu)Fusibility of ash (temperature °C)Fluorine (F) Gallium (Ga) Lanthanum (La) Lithium (L1)Initial deformation Softening Fluid Lead (Pb) Lithium (L1)Manganese (Mn) Mecdymium (Nd) Scandium (Sc) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbum (YD) Ytterbum (YD) Ytterbum (YD) Ytrium (Y) Zinc (Zn) Zirconium (Zr)	Sulfur (S)	Nitrogen (N)
Trace elements (ppm) 2Sulfur (S) AshAntimony (Sb) Arsenic (As)Heat of combusion (Btu/lb;cal/kg)Barium (Ba) Beryllium (Be)Forms of sulfur (percent)Boron (B) Cadmium (Cd)Sulfate PyriticCartium (Ce)Pyritic OrganicCobalt (Co) Copper (Cu)Fusibility of ash (temperature °C)Fluorine (F) Gallium (Ga)Initial deformation Softening Fluorine (Fb)Lanthanum (La) Lead (Pb)Free-swelling indexLithium (L1)Free-swelling indexManganese (Mn) Mecoury (Hg) Molybdenum (Mo) Nickel (N1)Free-stalling indexNickel (N1) Vanadium (V) Vtterbium (Yb) Vtterbium (Yb) Vtterbium (Yb)Trace (Zn) Zinc (Zn) Zinc (Zn)		Oxygen (0)-by difference
AshAntimony (Sb)Arsenic (As)Barium (Ba)Beryllium (Be)Boron (B)Cadmium (Cd)Cadmium (Cd)Cobalt (Co)Cobalt (Co)Cobalt (Co)Copper (Cu)Fluorine (F)Gallium (Ga)Initial deformationGermanium (Ca)Lanthanum (La)Lanthanum (La)Lithium (Li)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Nickel (Ni)Nickel (Ni)Uranium (Sc)Silver (Ag)Strontium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Ytterbium (Yb)Ytterbium (Yb)Zinc (Zn)Zirconium (Zr)	Trace elements (ppm) ²	Sulfur (S)
Antimony (Sb)Arsenic (As)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCerium (Ce)PyriticChromium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Nickel (Ni)Softenium (Se)Silver (Ag)Strontium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Ytterbium (Yb)Ytterbium (Zr)		Ash
Arsenic (As)Heat of combusion (Btu/lb;cal/kg)Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCerium (Ce)PyriticChromium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Nickel (Ni)Scandium (Sc)Selenium (Se)Silver (Ag)Silver (Ag)Strontium (Sr)Thorium (V)Ytterbium (Vb)Ytterbium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Sandium (Zr)	Antimony (Sb)	
Barium (Ba)Forms of sulfur (percent)Boron (B)SulfateCerium (Ce)PyriticChromium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Niobium (Nb)Neodymium (Sc)Selenium (Sc)Selenium (Sc)Silver (Ag)Strontium (Th)Uranium (U)Vanadium (V)Ytterbium (Y)Zinc (Zn)Zirconium (Zr)Zirconium (Zr)	Arsenic (As)	Heat of combusion (Btu/lb;cal/kg)
Beryllium (Be)Forms of sulfur (percent)Boron (B)SulfateCadmium (Cd)SulfateCerium (Ce)PyriticChromium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Nickel (Ni)Scandium (Sc)Selenium (Sc)Selenium (Sr)Strontium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Zinc (Zn)	Barium (Ba)	
Boron (B) Cadmium (Cd) Cadmium (Cd) Corium (Cr) Cobalt (Co) Copper (Cu) Fluorine (F) Gallium (Ga) Germanium (Ge) Lanthanum (La) Lead (Pb) Lithium (Li) Maganese (Mn) Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Nickel (Ni) Nickel (Ni) Nickel (Sc) Selenium (Sc) Selenium (Sc) Silver (Ag) Strontium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yb) Ytterbium (Yb)	Beryllium (Be)	Forms of sulfur (percent)
Cadmium (Cd)SulfateCerium (Ce)PyriticChromium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickei (Ni)Niobium (Nb)Neodymium (Nd)Scandium (Sc)Selenium (Sc)Selenium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)	Boron (B)	
Cerium (Ce)PyriticChromium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Free-swelling indexMercury (Hg)Molybdenum (Mo)Nickel (Ni)Niobium (Nb)Neodymium (Nd)Scandium (Sc)Selenium (Sc)Selenium (Sr)Thorium (Th)Urandium (V)Ytterbium (V)Ytterbium (V)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Zinc (Zn)	Cadmium (Cd)	Sulfate
Chromium (Cr)OrganicCobalt (Co)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGermanium (Ga)Initial deformationGermanium (La)FluidLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Nickel (Ni)Nickel (Ni)Nickel (Ni)Scandium (Sc)Selenium (Sc)Selenium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Zirconium (Zr)	Cerium (Ce)	Pyritic
Cobalt (Co)Fusibility of ash (temperature °C)Copper (Cu)Fusibility of ash (temperature °C)Gallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Niobium (Nb)Neodymium (Nd)Scandium (Sc)Selenium (Sc)Silver (Ag)Strontium (Sr)Thorium (U)Vanadium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zinconium (Zr)	Chromium (Cr)	Organic
Copper (Cu)Fusibility of ash (temperature °C)Fluorine (F)Initial deformationGallium (Ga)Initial deformationGermanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Nickel (Ni)Nickel (Ni)Niobium (Xb)Scandium (Sc)Selenium (Se)Silver (Ag)Silver (Ag)Strontium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Standard (Zr)	Cobalt (Co)	
Fluorine (F) Gallium (Ga) Germanium (Ge) Lanthanum (La) Lanthanum (La) Lithium (Li) Manganese (Mn) Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Copper (Cu)	Fusibility of ash (temperature °C)
Gallium (Ga)Initial deformation Softening Lanthanum (La)Lanthanum (La)Softening FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Niobium (Nb)Neodynium (Nd)Scandium (Sc)Selenium (Sc)Silver (Ag)Strontium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Strontium (Zr)	Fluorine (F)	
Germanium (Ge)SofteningLanthanum (La)FluidLead (Pb)Free-swelling indexManganese (Mn)Mercury (Hg)Molybdenum (Mo)Nickel (Ni)Niobium (Nb)Neodymium (Nd)Scandium (Sc)Selenium (Se)Silver (Ag)Strontium (Sr)Thorium (Th)Uranium (U)Vanadium (V)Ytterbium (Yb)Yttrium (Y)Zinc (Zn)Zirconium (Zr)Softenium (Zr)	Gallium (Ga)	Initial deformation
Lanthanum (La) Fluid Lead (Pb) Lithium (Li) Free-swelling index Manganese (Mn) Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Germanium (Ge)	Softening
Lead (Pb) Lithium (Li) Free-swelling index Manganese (Mn) Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Sc) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Lanthanum (La)	Fluid
Lithium (Li) Free-swelling index Manganese (Mn) Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Lead (Pb)	
Manganese (Mn) Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Sc) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Lithium (Li)	Free-swelling index
Mercury (Hg) Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Manganese (Mn)	
Molybdenum (Mo) Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Yb) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Mercury (Hg)	
Nickel (Ni) Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Y) Ytterbium (Y) Zinc (Zn) Zirconium (Zr)	Molybdenum (Mo)	
Niobium (Nb) Neodymium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Y) Ytterbium (Y) Zinc (Zn) Zirconium (Zr)	Nickei (Ni)	
Neodymium (Nd) Scandium (Sc) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Y) Ytterbium (Y) Zinc (Zn) Zirconium (Zr)	Niobium (Nb)	
Scandium (SC) Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (V) Ytterbium (Y) Zinc (Zn) Zirconium (Zr)	Neodym1um (Nd)	
Selenium (Se) Silver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Y) Ytterbium (Y) Zinc (Zn) Zirconium (Zr)	Scandium (SC)	
Stiver (Ag) Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Y) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Selenium (Se)	
Strontium (Sr) Thorium (Th) Uranium (U) Vanadium (V) Ytterbium (Y) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Sliver (Ag)	
Uranium (U) Vanadium (V) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Strontium (Sr) Thenium (Th)	
Vanadium (V) Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	licanium (II)	
Ytterbium (Yb) Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Vanadium (V)	
Yttrium (Y) Zinc (Zn) Zirconium (Zr)	Vttorhium (V)	
Zinc (Zn) Zirconium (Zr)	Yttrium (Y)	
Zirconium (Zr)	7inc (7n)	
L'I CONTAIN (41)	Zinc (Zn) Zircopium (Zr)	

Table 2. Analyses and physical tests performed by various laboratories

Reported as oxides in 525°C laboratory ash as well as on a whole-coal basis. Reported as parts per million in 525°C laboratory ash and (or) on a whole-coal basis.



Figure 3. Flow chart showing sequence of sample preparation and chemical analysis. Modified from Swanson and Huttman (1976, figure 1).

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5.6, 3.8, 2.6, 1.8, 1.2, etc., but are reported arbitrarily as midpoints of these brackets, i.e. 10, 7, 5, 3, 2, 1.5, 1, etc. The precision of a reported value is approximately plus or minus one bracket at 68 percent confidence, or two brackets at 95 percent confidence.

additional elements not Twenty-two listed in tables 4 and 5 were not found during emission spectrographic analysis. These elements and their lower limits of determination in ppm are: palladium (5); bismuth, indium, and tin (20); gold, holmium, and thulium (50); (70); dysprosium, erbium, lutetium gadolinium, platinum, rhenium, and (100); europium, hafnium, thallium praseodymium, samarium, and tungsten (200); terbium (700); tantalum (1,000); and tellurium (5,000).

Changes in the analytical procedures outlined in figure 3 for phosphorus content in ash, and thorium and arsenic contents in whole coal, result in variable lower-detection limits. Modifications in the analytical technique for determining phosphorus content in ash, as determined by X-ray fluorescence spectroscopy, resulted in a lowerdetection limit of 0.04 percent phosphorus $(0.1 \text{ percent } P_2 O_5)$ for samples D166027-D166043 and 0.4° percent phosphorus (1.0 percent P₂0₅) for all other Thorium contents of samples samples. D176169-D176200, D166027-D166043. D179838-D179841, D179844-D179847 D179849, D179850, and D179852-D179856 determined by were delayed-neutron activation analysis with a lower-detection limit of 3.0 ppm; the remaining 42 samples were analyzed by instrumental neutron-activation analysis with a lower-detection limit 0.1 ppm. of Arsenic contents of samples listed in this report were determined by three different analytical methods: samples D166027-D166043 and D176169-D176200 were analyzed spectro-photometrically (lower-detection limit of 1.0 ppm); samples D179838-D179856 were analyzed by the graphite-furnace atomic absorption method (lower-detection limit of 0.5 ppm); and the other 38 samples were analyzed by instrumental neutron activation analysis (lower-detection limit of 0.1 ppm).

Summary Tables

To aid in the statistical comparison of the data, the coal samples were divided into five groups based upon nine palynologically-determined stratigraphically-positioned coal zones (Ravn, 1980, personal communication: Ravn. in press). The first group of samples (15 samples) is from coal-zone 2; the second group (15 samples) is from coalzone 3; the third group (49 samples) is from coal-zone 4; the fourth group (5 samples) is from coal-zone 5; and the fifth group (16 samples) is from coalzones 6, 7, 8, and 9. Only five coal zones are equivalent with currently recognized named memebers. Coal-zones 1, 2, and 3 occur stratigraphically below the Laddsdale coal. Coal-zone 4 is equivalent to most historical references to Laddsdale Coal the (Howes, 1981, personal communication). Coal-zone 5 occurs above the Laddsdale Coal and below the Wiley Coal. Coalzones 6, 7, 8, and 9 are equivalent, respectively, to the currently named coal beds: the Wiley Coal, the Whitebreast Coal, the Wheeler Coal, and the Bevier Coal (Ravn, 1980, personal communication). Statistical summaries were not made for the Mystic Coal, in the Marmaton Group, the coal associated with coal-zone 3, sample D186067, and coal-zone 1 because of the limited number of samples (one, one, and three respectively) available. Sample D179842 (zone 7) was not included in the summaries because it contained 68.8 percent ash and is a coaly shale.

Unweighted statistical summaries for coal-zone-2 samples are listed in tables 6 and 7; coal-zone-3 samples, tables 8 and 9; coal-zone-4 samples, tables 10 and 11; coal-zone-5 samples, tables 12 and 13; coal-zones 6, 7, 8, and 9, tables 14 and 15; and for 105 Cherokee Group coal samples from Iowa, tables 16 and 17. The number of U.S. Bureau of Mines analyses summarized for each coal zone is less than the number of U.S. Geological Survey analyses, because not every sample was sent to the U.S Bureau of Mines and because of sampling problems discussed later in the section on Apparent Rank.

For comparison with the Iowa Cherokee Group coals, statistical summaries of analyses for 114 Pennsylvanian coal samples from the Eastern Interior coal region (Illinois Basin) from Gluskoter and others (1977), and summaries of analyses of 63 Upper Cretaceous Williams Fork Formation coal samples from the Yampa field, Routt and Moffat Counties, Colorado from Hildebrand and others (1981) are listed in tables 18and 20-21 respectively. 19 These analyses are from coals of about the same rank as the Iowa coals. Similar rank coals were selected for comparison because element composition of coal. particularly for lower rank coals. varies with rank (Hildebrand and Hatch. 1977). Illinois Basin coal data summarized in tables 18 and 19 primarily represent samples of the Harrisburg (No. 5) coal (32 samples) and Herrin (No. 6) coal (49 samples) that were collected almost exclusively from operating coal mines (105 of 114 samples). Consequently, the statistics for the Illinois Basin samples are the biased toward thicker. better quality (lower ash and lower sulfur contents) coals of the Illinois Basin and are not strictly comparable to the statistics for the Cherokee Group sample set. A more serious source of disparity between the Iowa Cherokee Group and Illinois Basin sample sets results from the exclusion of all mineral bands, partings, and nodules more than one centimeter (3/8 inch) thick during collection of the Illinois Basin samples. This sampling procedure results in an underestimation of ash and sulfur contents and an overestimation of the heat of combustion (Btu/lb) of the samples relative to the in-place coal.

Statistical Methods

In tables 6-21, the geometric mean (GM) is used as the estimate of the most probable element content (mode); the geometric mean is calculated by taking the logarithm of each analytical value, summing the logarithms, dividing the sum by the total number of values, and obtaining the antilogarithm of the result. The measure of scatter about the mode used in this report is the geometric deviation (GD), which is the antilog of the standard deviation of the logarithms of the analytical values. These statistics are used because the quantities of trace elements in natural materials commonly exhibit positively skewed frequency distributions; such distributions are normalized by analyzing and summarizing trace-element data on a logarithmic basis.

If the frequency distributions are lognormal, the geometric mean is the best estimate of the mode, and the estimated range of the central two-thirds of the observed distribution has a lower limit equal to GM/GD and an upper limit equal to GMxGD. The estimated range of the central 95 percent of the observed distribution has a lower limit equal to $GM/(GD)^2$ and an upper limit equal to $GMx(GD)^2$ (Connor and others, 1976).

Although the geometric mean is, in general, an adequate estimate of the most common analytical value, it is, nevertheless, a biased estimate of the arithmetic mean. The estimates of the arithmetic means as listed in the summary tables are Sichel's \underline{t} statistic (Miesch, 1967).

A common problem in statistical summaries of trace-element data arises when the element content of one or more of the samples is below the limit of analytical determination. This circumstance, which occurs for 14 elements listed in the summary tables, is called a "censored" distribution. Procedures developed by Cohen (1959) were used to compute unbiased estimates of the geometric mean, geometric deviation, and arithmetic mean when the data are "censored."

To be consistent with the precision of the semiquantitative emission spectrographic technique, arithmetic and geometric means of elements determined by this method were reported as the midpoint of the enclosing six-step brackets.

Data summaries for phosphorus and cerium contents were not included in any of the summary tables because these elements were detected in too few samples to calculate meaningful statistics. For the same reason, summaries of silver and niobium contents in tables 7, 9, 11, and 13, neodymium content in tables 9, 11, 13, 15, and 17, lanthanum content in tables 11, 13, and 15, and ytterbium content in tables 7, 9, 11, 13, and 17 were also omitted.

DISCUSSION OF RESULTS

Apparent Rank

The apparent rank of each of the 90 coal samples from south-central and southeastern Iowa was calculated using the data in table 3, and the approximation to the Parr formula and classifications in ASTM designation D-388-77 (American Society for Testing and Materials, 1978). The ASTM classification scheme is reproduced in table 22. The apparent ranks range from subbituminous B coal (1 sample) through subbituminous A coal (5 samples), highvolatile C bituminous coal (68 samples), high-volatile B bituminous coal (15 samples), to high-volatile Α bituminous coal (1 sample).

The single sample of subbituminous B rank (D185606) was from a shallow surface mine; the coal at that location has probably been slightly weathered. At the other end of the distribution. high-volatile-A-bituminous-coal the sample (D186069), 12 of 14 highvolatile-B-bituminous-coal samples, and 11 samples from the higher end of highvolatile-C-bituminous-coal range are from core holes CP3, CP6, CP7, CP10, CP11, CP13, and CP15. These 24 samples were collected early in the project and were apparently allowed to dry out before bagging, resulting in low moiserroneously high ture contents and apparent ranks. This conclusion is based on data from Neelv H. Bostick (written communication, 1982) who found no significant difference in average vitrinite reflectance (R_0) , a second measure of coal rank, in samples from these seven core holes when compared to samples collected later in the study. relationship moist. The between mineral-matter-free (mmmf) Btu/lb and apparent rank for the remaining 65 Iowa coal samples is shown in figure 4.

Average Btu/lb (mmmf) for coal-zones 2, 3, 4, 5, and 6-9 are very similar (table 23), ranging from 11,820 Btu/lb for coal zone 5 to 12,110 Btu/lb for coal-zone 4. Average heat of combustion for 65 Iowa Cherokee Group coal samples is 12.040 Btu/lb (mmmf), and is similar to the heat of combustion of samples from the Yampa Field (11,910 Btu/lb, mmmf). Both sets have an average apparent rank of high-volatile C bituminous coal. Average heat of combustion for Illinois Basin the samples is higher (12,990 Btu/1b) reflecting a slightly higher average apparent rank (high-volatile C to highvolatile B bituminous coal).

Proximate, Ultimate, and Related Analyses

Arithmetic means and ranges (asreceived basis) of the proximate and ultimate analyses, heat-of-combustion.



Moist, mineral-matter-free





Figure 5. Arithmetic means (*) and ranges [---] of proximate and ultimate analyses and forms of sulfur (as-received basis) for Iowa Cherokee Group coal samples from coal-zones 2, 3, 4, 5, and 6-9.



kcal/kg



Figure 6. Arithmetic means (*) and ranges [---] of ash-fusion temperature and heats of combustion (as-received basis) for Iowa Cherokee Group coal samples from coal-zones 2, 3, 4, 5, and 6-9.

forms-of-sulfur, and ash-fusion-temperature data for samples from coal-zones 2, 3, 4, 5, and 6-9 are shown in figures 5 and 6. Statistical comparisons (Student's t test, 95 percent confidence level) of the means from the different Cherokee Group coal zones show two significant differences: a) zone 5 coals have the highest ash and organic-sulfur contents and the lowest heats of combustion, and b) ash-fusion significantly temperatures decrease from zone 2 through zones 6-9. For the other analyses, average composition of coals from one zone may be significantly higher or lower than those from a second zone, but may be similar to the analyses of a third zone. For example, mean sulfur content of zones 6-9 is significantly lower than mean sulfur contents of zones 2 and 5, but is statistically similar to the mean sulfur contents of zones 3 and 4.

Arithmetic means and ranges of the proximate and ultimate analyses, heatof-combustion, and forms-of-sulfur data for the Iowa Cherokee Group, Illinois Basin, and Yampa field sample sets are shown in figures 7 and 8. A strict statistical comparison of these three data sets is not possible because of a lack of standard deviations for the Iowa Cherokee Group and Yampa field



Figure 7. Arithmetic means (*) and ranges [---] of proximate and ultimate analyses and forms of sulfur (as-received basis) for 65 Iowa Cherokee Group coal samples, 114 Illinois Basin coal samples, and 44 Yampa field, Colorado coal samples. [Illinois Basin data are from Gluskoter and others (1977, table 8). Yampa Field data are from Hildebrand and others (1981, table 7a).]



------ lowa somples ----- Illinois Basin samples ------ Yampa field, Colorado, samples

Figure 8. Arithmetic means (*) and ranges [---] of heats of combustion (asreceived basis) for 65 Iowa Cherokee Group coal samples, 114 Illinois Basin coal samples, and 44 Yampa field, Colorado, coal samples. [Illinois Basin data are from Gluskoter and others (1977, table 8); Yampa field data are from Hildebrand and others (1981, table 7a).]

sampling biases previously sets and discussed. However, the information listed in tables 16, 18, and 20 and illustrated in figures 7 and 8 shows that Iowa Cherokee Group coals probably have lower nitrogen, fixed-carbon, carbon, and hydrogen contents; higher ash, and total-, sulfate-, and pyriticsulfur contents, and a lower heat of combustion. Illinois Basin coals have lower moisture contents, higher volatile matter and carbon contents, and a higher heat of combustion. The Yampa field samples have higher oxygen and oxygen/carbon mole ratios (moisturefree basis, table 23) and much lower sulfate-, total-. pyritic-, and organic-sulfur contents. Hydrogen/carbon mole ratios (moisture-free basis, table 23) for the Iowa Cherokee Group, Illinois Basin, and Yampa field sample sets are similar.

The lower oxygen/carbon mole ratios for the high-sulfur coals probably resulted from greater bacterial activity in the peat swamps that produced the high-sul-Because bacteria utilize fur coals. oxygen-rich organic components (for example, cellulose or lignin) more easily than more hydrogen-rich components (for example, cuticles, spore and resins) pollen exines, waxes and (Waksman and Stevens, 1928), increased bacterial activity would result in a depletion of oxygen-rich organic matter and decreased oxygen/carbon mole ratios.

Differences in ash and total-. sulfate. and pyritic-sulfur contents between Iowa Cherokee Group and Illinois Basin coals result in part from the sampling hiases discussed earlier (core and mine samples of Iowa coal versus mine samples for Illinois coal). between Differences Iowa Cherokee Group, Illinois, and Yampa field coals in moisture and carbon contents and heat of combustion are probably due to differences the minor in thermal maturity.

Element Analyses

Geometric means and ranges for the contents of 35 elements in coal samples from coal-zones 2, 3, 4, 5, and 6-9 are shown in figure 9. Statistical comparisons (Student's t test, 95 percent confidence) of the summary data from the different coal zones show few significant differences. One significant difference, however, is that the contents of nine elements (Na, Mg, K, As, Mn, Mo, Sb, U, and V) increase from coal-zone 2 through coal-zones 6-9.

Geometric means and ranges for the contents of 34 elements in the Iowa, Cherokee Group, Illinois Basin, and Yampa field sample sets are shown in



Figure 9. Geometric means (*) and ranges [---] for contents of 40 elements (air-dried, whole-coal basis) in Iowa Cherokee Group coal samples from coal-zones 2, 3, 4, 5, and 6-9. Wavy lines to the left of the range brackets for Cd, F, La, Mo, Nd, Sb, Se, Th, and U indicate data that are less than the lower-detection limit.



Figure 9. Geometric means (*) and ranges [---] for contents of 40 elements (air-dried, whole-coal basis) in Iowa Cherokee Group coal samples from coal-zones 2, 3, 4, 5, and 6-9--continued.



Figure 9. Geometric means (*) and ranges [---] for contents of 40 elements (air-dried, whole-coal basis) in Iowa Cherokee Group coal samples from coal-zones 2, 3, 4, 5, and 6-9--continued.



Figure 10. Geometric means (*) and ranges [---] for contents of 38 elements (whole-coal basis) in 105 Iowa Cherokee Group coal samples, 114 Illinois Basin coal samples, and 63 Yampa field coal samples. Iowa and Yampa field data are on an air-dried basis, Illinois Basin data are on a moisture-free basis. Illinois Basin data are from Gluskoter and others (1977, table 8); Yampa field data are from Hildebrand and others (1981, table 9a). Wavy lines to the left of the range brackets for K, Ag, Cd, Co, Cr, F, La, Mo, Nb, Pb, Sb, Sc, Se, Sr, Th, and U indicate data that are less than the lower-detection limit. Illinois Basin data summaries for La, Li, Nb, and Y are not available; Yampa field data summaries for Ag, Ge, and La are not available.

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Figure 10. Geometric means (*) and ranges [---] for contents of 38 elements (whole-coal basis) in 105 Iowa Cherokee Group coal samples, 114 Illinois Basin coal samples, and 63 Yampa field coal samples-- continued.

figure 10. The information listed in tables 17, 19, and 21 and illustrated in figure 10 shows that the contents of at least 16 elements are directly related to sulfur content of the coals. Of the three sample sets, Iowa Cherokee Group coals have the highest mean sulfur content (5.8 percent), Illinois Basin coals an intermediate content (3.1 percent), and Yampa field coals the lowest sulfur content (0.6 per-Iowa Cherokee Group coal has cent). the highest content of Ca, Fe, Ag, As, Be, Co, Cu, Ge, Mn, Ni, Pb, Sc, Se, U, Y, and Zn; Illinos coal has an intermediate content of these elements; and Yampa field coal has the lowest content of these elements. The low-sulfur Yampa field coal also has significantly lower mean contents of K, Cd, Cr, Hg, Mo, Sb, and V than the relatively highsulfur Iowa Cherokee Group and Illinois Basin coals.

Element distributions in coals are controlled by many factors, including provenance, geochemical conditions (pH, Eh. salinity) of the depositional and early diagenetic environments, thermal maturity (rank), groundwater composition, and nature and intensity of any epigenetic mineralization (Hatch, A, in press). According to Cecil and others (1982) the most important geochemical parameter during deposition and early diagenesis is the pH of waters in the peat swamp. Under low-pH conditions (3-4.5), solution of most metal ions is favored and the activity of sulfatereducing bacteria is minimal; the resulting peat has low sulfur and metal contents. The activity of sulfatereducing bacteria reaches a maximum when pH conditions are near neutral (6-8) (Baas Becking and others, 1960).

A strong relationship exists between sulfur content in coals and CaCO₃ content of associated rocks (Cecil and others, 1982). Lack of carbonate rocks would indicate minimal carbonate buffering of depositional and early diagenetic connate waters, resulting in relatively low-pH conditions (3-4.5); presence of carbonates and calcareous shales would indicate relatively highpH conditions (6-8). Iowa Cherokee Group and Illinois Basin coals are associated with carbonates and have high sulfur contents. In contrast, Yampa field coals have low sulfur contents and are associated with noncalcareous rocks.

Elements whose contents are significantly higher in the high-sulfur sample sets may be fixed by a variety of processes: (1) they form highly insoluble sulfides (Fe, Ag, As, Cd, Co, Cu, Hg, Ni, Sb, and Zn); (2) they are included in minerals that form at (or are less readily leached at) near-neutral pH's (Ca and Mn carbonates, Sc and Th phosphates, and K and Ca in illite or smectite clays); or (3) they have multiple valence states (Fe, S, U, Se, Mo, Ge, Cr, V, and Be), and may be fixed in the coal during the peat stage or subsequent stages of coalification through reduction of the element by reaction with H₂S or other reactive sulfur species, and may subsequently be incorporated into stable organic or mineral phases. Except for chromium, the elements listed in (3) are also the same elements enriched in roll-type uranium deposits (Harshman, 1974), suggesting that similar geochemical processes are operating in both environments (Hatch, B, in press).

Correlation Analyses

Correlation coefficients relating the variability of each parameter with the variability of every other parameter were calculated from the element data for coal-zones 2, 3, 4, 5, and 6-9. Geochemical associations of some elements are apparent from these correlation coefficients.

(1) In all five coal groups there are strong positive correlation coefficients (\geq 0.89) between zinc and

cadmium. In Iowa coals, zinc occurs as sphalerite (ZnS) that is found along cleats and fractures in the coal, and is associated with pyrite, calcite, kaolinite, and barite (Hatch, Avcin, and others, 1976). In similar occurrences in Illinois Basin coals, cadmium is found in solid solution with zinc in the sphalerite (Gluskoter and Lindahl, 1973, and Hatch, Gluskoter and Lindahl, 1976). The strong positive correlation coefficients for Iowa Cherokee Group coals suggest a similar relationship. Similar high zinc:cadmium correlation coefficients were found for Missouri coals by Wedge and Hatch (1980). These occurrences of sphalerite and associated minerals have been interpreted by Hatch, Gluskoter, and Lindahl (1976) to represent post-depositional, epigenetic mineralization of the coals.

The zinc and cadmium contents of Iowa Cherokee Group coal samples are as much as 18,000 and 100 ppm, respectively, and have arithmetic means of 1,100 and 18 ppm, respectively. Most zinc and cadmium could be removed and recovered from the coals by conventional washing techniques (Hatch, Gluskoter, and Lindahl, 1976).

An apparent zoning of zinc/cadmium molecular ratios was noted in five cores (CP-7, CP-10, CP-21, CP-28, and CP-32) from Wapello and Appanoose Counties by Hatch, Avcin, and others (1976). The data listed in table 24 and illustrated in figure 11 show that cadmium is enriched relative to zinc in the stratigraphically higher coals. indicating a chemical differentiation of the solutions from which the sphalerite and associated minerals were precipitated.

(2) In all five coal groups, calcium and manganese have high positive correlation coefficients (0.6 to 0.9); manganese does not correlate well with any other element. Because manganese commonly substitutes for calcium in calcite (CaCO₃), it is presumed to occur in calcite in Iowa Cherokee Group coals. Similar associations were found in Illinois Basin coals by Gluskoter and others (1977) and in Missouri coals by Wedge and Hatch (1980). Abundant calcite occurs in cleats and fractures in Iowa Cherokee Group coals and is thought to have been deposited during the sphalerite mineralization process.

(3) In all five coal groups, elements that generally occur in sedimentary rocks as aluminosilicate minerals (Al, Si, Mg, K, Li, and Zr) or as resistant oxides or phosphates (Ti, Th, Sc, and La) have mutually positive correlation coefficients (0.6 to 0.9) suggesting a detrital origin (water or wind transported). In coal-zones 2, 3, and 4, Cr, Mo, U, and V are positively correlated (0.5 to 0.7) to this element assemblage as are Na and Y (correlation coefficients 0.7 to 0.9) in coal-zones 5 and 6-9.

(4) In all five coal groups, Fe, As, Hg, and Sb have mutually high positive correlation coefficients (0.6 to 0.9). Most of the iron is probably present in the coals as pyrite (FeS₂). The other three elements are commonly found in nature as sulfides and are presumably associated with the pyrites. In coalzones 2, 3, and 4, Cu, Pb, Co, and Ni are positively correlated (0.5 to 0.9) with the Fe, As, Hg, and Sb assemblage of elements; in coal-zones 5 and 6-9, U, Se, Mo, Cr, and V are positively correlated (0.6 to 0.9) with this assemblage.

The changes in element association of U, Mo, Cr, and V from an aluminosilicate assemblage in the lower part of the section (coal-zones 2, 3, and 4) to a sulfide assemblage in the upper part of the section (coal-zones 5 and 6-9), are related to the increases in mean element content with higher stratigraphic position (illustrated in figure 9). With higher stratigraphic position Na, Mg, and K also have higher mean contents. Avcin and Koch (1979)



Figure 11. Relationship of the zinc/cadmium mole ratio in coal to depth in five core holes from Wapello and Appanoose Counties, Iowa.

indicated that depositional environments for the Cherokee and Marmaton Groups changed from predominantly terrestrial in the lower part of the Cherokee Group to predominantly marine higher in the stratigraphic section.

Presumably Na, Mg, K, Cr, V, U, Mo, Sb, Se, and Ag were more readily available in the more marine-influenced environments in which upper Cherokee and Marmaton rocks were deposited. Changes in contents of Na, Mg, and K apparently affect the ash-fusion temperatures which drop significantly from coal-zone 2 to coal-zones 6-9 (figure 6).

All six coal samples from coal-zone 7 and three samples from coal-zone 4 (D176171, D176183, and D179843) have much higher contents of U, Mo, Sb, Se, V, and Ag. The coals at these sample sites are overlain by black, sometimes phosphatic, shale; shales that were deposited under anoxic marine conditions (Heckel, 1977).

SUMMARY

- Middle Pennsylvanian Cherokee Group coals from south-central and southeastern Iowa are typical highsulfur, high-ash coals, with a mean total sulfur content of 5.8 percent; and a mean ash content of 15.9 percent. Mean, as-received heat of combustion is 9,640 Btu/lb. Mean, moist, mineral-matter-free heat of combustion is 12,040 Btu/ lb, which is equivalent to an apparent rank of high-volatile C bituminous coal.
- 2. In a comparison with equivalentrank Illinois Basin and Yampa field coals, Iowa coals have the highest mean contents of 16 elements (S, Ca, Fe, Ag, As, Be, Co, Cu, Ge, Mn, Ni, Pb, Sc, Se, U, and Zn).

- 3. Low sulfur Yampa field coals have significantly lower mean contents of K, Cd, Cr, Hg, Mo, Sb, and V than high-sulfur, Iowa Cherokee Group and Illinois Basin coals.
- 4. Iowa Cherokee Group coals have been subject to post-depositional, epigenetic sphalerite/calcite/pyrite/ kaolinite/barite mineralization. Zinc and cadmium contents of Iowa coal samples are as much as 18,000 ppm and 100 ppm respectively. Most zinc and cadmium can be removed and recovered from the coals by conventional washing techniques.
- 5. Iowa coal samples also have as much as 300 ppm Co, 150 ppm Cr, 70 ppm Ge, 150 ppm Mo, 700 ppm Ni, and 300 ppm V. As conventional world supplies of these metals and zinc and cadmium are depleted, Iowa Cherokee Group coal should be considered as a possible source.
- 6. Element associations of U, Mo, Cr, and V change from an aluminosilicate assemblage in the lower part of the section (coal-zones 2, 3, and 4) to a sulfide assemblage higher in the section (coal-zones 5 and 6-9). This change is related to increased element content with higher stratigraphic position. These increases are thought to be related to differences in depositional environments of the coalassociated rocks which change from predominantly terrestrial lower in the stratigraphic section to pre-dominantly marine higher in the section. The decrease in ashfusion temperatures with higher stratigraphic position is probably related to increased contents of Na, Mg, and K.
- Coals, in particular, samples from coal-zone 7, which are overlain by marine, phosphatic, black-shale
lithologies, have the highest contents of U, Mo, Ag, Sb, Se, and V.

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(All analyses except heat of combustion, free-swelling index, and ash-fusion temperatures in percent. For each sample, the analyses are reported three ways: first, as received; second, moisture free; and third, moisture and ash free. Kcal/kg = 0.556 (Btu/lb); °F = (°C x 1.8) + 32; L, less than the value shown; B not determined }

		Proximate an	alysis			Ulti	mate analysis	;		
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D166029	14.5	32.7	40.2	12.6	5.4	56.0	1.0	19.7	5.3	B
		38.2	47.0	14.7	4.4	65.5	1.2	8.0	6.2	
		44.9	55.1		5.2	76.8	1.4	9.3	7.3	
D166031	11.9	33.9	37.5	16.7	5.1	54.3	.9	15.4	7.6	В
0100001		38.5	42.6	19.0	4.3	61.6	1.0	5.5	8.6	
		47.5	52.5		5.3	76.1	1.3	6.8	10.6	
D166034	11.5	35.1	34.7	18.7	4.8	52.1	.8	11.9	11.7	В
010000.		39.7	39.2	21.1	4.0	58.9	.9	1.9	13.2	
		50.3	49.7		5.0	74.6	1.1	2.4	16.8	
D166035	13.1	37.6	35.8	13.5	5.4	56.4	1.1	19.8	3.8	В
		43.3	41.2	15.5	4.5	64.9	1.3	9.4	4.4	
		51.2	48.8		5.4	76.8	1.5	11.1	5.2	
D166037	11.9	37.0	37.6	. 13.5	5.5	57.5	1.2	18.7	3.6	8
010000.		42.0	42.7	15.3	4.7	65.3	1.4	9.2	4.1	
		49.6	50.4		5.6	77.1	1.6	10.9	4.8	
D166039	12.1	33.4	38.5	16.0	5.2	56.0	1.2	19.3	2.3	В
0100000		38.0	43.8	18.2	4.4	63.7	1.4	9.7	2.6	
		46.5	53.5		5.4	77.9	1.7	11.9	3.2	
D166043	9.8	38.7	41.2	10.3	5.5	62.7	1.0	14.8	5.7	B
2200010		42.9	45.7	11.4	4.9	69.5	1.1	6.8	6.3	
		48.4	51.6		5.5	78.5	1.3	7.6	7.1	

	Heat of co	ombustion	F	Forms of sulfu	ır		Ash fusion temperature, C°			
Sample number	Kcal/kg	Btu/lb	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D166029	5,650 6,610 7,750	10,170 11,890 13,950	0.01L .01L .01L	3.24 3.79 4.44	2.05 2.40 2.81	В	В	8	В	
D166031	5,560 6,310 7,780	10,000 11,350 14,010	.06 .07 .08	5.25 5.96 7.35	2.29 2.60 3.21	B	В	В	B	
D166034	5,320 6,010 7,620	9,570 10,810 13,710	.14 .16 .20	9.35 10.56 13.40	2.16 2.44 3.09	В	8	В	B	
D166035	5,550 6,390 7,560	9,990 11,500 13,610	.04 .05 .05	2.78 3.20 3.79	1.03 1.19 1.40	В	В	В	В	
D166037	5,790 6,570 7,760	10,420 11,830 13,970	.01L .01L .01L	2.82 3.20 3.78	.77 .87 1.03	B	В	B	В	
D166039	5,510 6,270 7,660	9,920 11,290 13,800	.06 .07 .08	1.64 1.87 2.28	.61 .69 .85	В	B	B	В	
D166043	6,360 7,050 7,960	11,450 12,690 14,330	.01 .01 .01	4.34 4.81 5.43	1.31 1.45 1.64	B	B	В	В	

		Proximate an	alysis			Ulti	mate analysis	;		
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D176169	5.7	40.4	41.7	12.2	4.9	64.5	1.0	13.7	3.7	1.9
		42.8 49.2	44.2 50.8	12.9	4.5 5.2	68.4 78.6	1.1 1.2	9.2 10.5	3.9 4.5	
D176170	6.3	36.4	40.1	17.2	4.4	56.1	1.0	13.4	7.9	1.8
		38.8 47.6	42.8 52.4	18.4	3.9 4.8	59.9 73.3	1.1 1.3	8.3 10.2	8.4 10.3	~ *
D176171		20.2	40.0	07 A	2.7	£1 7	1.0	11 2		0
01/01/1	4.3	20.3	40.0	2/.4	3./ 2.A	51.7	1.0	11.3	4.9	•0
		41.4	58.6	20.0	4.7	75.7	1.5	10.9	7.2	
D176172	3.4	37.1	33.7	25.8	3.5	49.7	.9	6.6	13.5	.8
		38.4	34.9	26.7	3.2	51.4	.9	3.7	14.0	
		52.4	47.6		4.4	70.2	1.3	5.1	19.1	~
D176173	4.7	38.8	34.2	22.3	4.3	55.1	1.0	12.2	5.1	2.0
		40.7	35.9	23.4	4.0	57.8	1.0	8.4	5.4	
		53.2	46.8		5.2	75.5	1.4	11.0	7.0	
D176174	4.3	36.5	40.5	18.7	4.7	60.4	1.1	11.7	3.4	1.2
		38.1	42.3	19.5	4.4	63.1	1.1	8.2	3.6	
		47.4	52.6	a = 4	5.5	78.4	1.4	10.2	4.4	
D176175	5.1	40.7	38.8	15.4	5.0	62.0	1.1	12.9	3.6	1.7
		42.9	40.9	16.2	4.7	65.3	1.2	8.8	3.8	
		51.2	48.8		5.6	78.0	1.4	10.5	4.5	
D176176	4.5	35.0	37.7	22.8	4.2	52.5	.9	7.7	11.9	.6
		36.6	39.5	23.9	3.9	55.0	.9	3.9	12.5	
		48.1	51.9		5.1	72.2	1.2	5.1	16.4	

	Heat of co	ombustion	F	orms of sulfu	ir .		Ash fusion	temperature,	, C°
Sample number	Kcal/kg	Btu/lb	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid
D176169	6,230 6,610 7,590	11,220 11,900 13,670	0.25 .27 .30	2.00 2.12 2.44	1.46 1.55 1.78	1.0	1,140	1,165	1,200
D176170	5,640 6,020 7,380	10,160 10,840 13,280	1.20 1.28 1.57	4.51 4.81 5.90	2.17 2.32 2.84	1.0	1,110	1,140	1,170
0176171	4,970 5,200 7,280	8,950 9,350 13,100	.90 .94 1.32	1.86 1.94 2.72	2.14 2.24 3.13	.5	1,110	1,140	1,200
0176172	5,160 5,340 7,280	9,280 9,610 13,110	.80 .83 1.13	11.85 12.27 16.74	.80 .83 1.13	1.0	1,145	1,170	1,200
D176173	5,420 5,680 7,420	9,750 10,230 13,360	.23 .24 .32	3.15 3.31 4.32	1.76 1.85 2.41	1.5	1,145	1,170	1,200
D176174	5,990 6,260 7,780	10,790 11,270 14,010	.21 .22 .27	1.38 1.44 1.79	1.83 1.91 2.38	1.5	1,290	1,315	1,345
0176175	6,150 6,480 7,740	11,070 11,660 13,920	.15 .16 .19	1.43 1.51 1.80	2.00 2.11 2.52	3.5	1,080	1,110	1,145
0176176	5,260 5,510 7,240	9,470 9,920 13,030	1.60 1.68 2.20	8.90 9.32 12.24	1.38 1.45 1.90	.5	1,260	1,290	1,320

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		Proximate an	alysis							
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D176177	4.0	41.5	38.3	16.2	4.7	57.9	1.0	10.3	9.9	0.8
		43.2	39.9	16.9	4.4	60.3	1.0	7.0	10.3	
		42.0	48.0		5.3	72.6	1.3	8.5	12.4	
D176178	4.7	30.7	35.3	29.3	3.8	46.0	.9	11.7	8.3	.7
		32.2	37.0	30.7	3.4	48.3	.9	7.9	8.7	
		46.5	53.5		5.0	69.7	1.4	11.4	12.6	
D176179	4.1	36.6	41.4	17.9	4.2	57.5	.6	12.4	7.4	1.0
		38.2	43.2	18.7	3.9	60.0	.6	9.1	7.7	
		46.9	53.1		4.8	73.7	.8	11.2	9.5	
D176180	4.9	41.2	45.0	8.9	5.1	67.2	1.2	14.0	3.6	1.8
		43.3	47.3	9.4	4.8	70.7	1.3	10.1	3.8	
		47.8	52.2		5.3	78.0	1.4	11.2	4.2	
D176181	5.4	35.5	44.3	14.8	4.6	59.1	1.1	15.6	4.8	1.4
		37.5	46.8	15.6	4.2	62.5	1.2	11.4	5.1	
		44.5	55.5		5.0	74.1	1.4	13.5	6.0	
D176182	3.4	33.4	31.4	31.8	3.3	41.7	-8	7.0	15.4	1.4
		34.6	32.5	32.9	3.0	43.2	.8	4.1	15.9	
		51.5	48.5		4.5	64.4	1.2	6.1	23.8	
D176183	4.4	34.5	35.9	25.2	4.0	49.5	.9	9.9	10.5	1.2
		36.1	37.6	26.4	3.7	51.8	.9	6.3	11.0	
		49.0	51.0		5.0	70.3	1.3	8.5	14.9	
D176184	5.7	33.9	40.6	19.8	4.3	53.5	1.2	12.4	8.8	2.0
		35.9	43.1	21.0	3.9	56.7	1.3	7.8	9.3	£ 8 17
		45.5	54.5		4.9	71.8	1.6	9.8	11.8	

	Heat of combustion		F	Forms of sulfur			Ash fusion	temperature,	C°
Sample number	Kcal/kg	Btu/lb	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid
0176177	6,010 6,260 7,530	10,810 11,260 13,550	0.96 1.00 1.20	6.38 6.65 7.99	2.60 2.71 3.26	2.5	1,180	1,210	1,240
0176178	4,710 4,940 7,130	8,470 8,890 12,830	1.05 1.10 1.59	5.94 6.23 9.00	1.31 1.37 1.98	.5	1,110	1,140	1,170
)176179	5,730 5,970 7,340	10,310 10,750 13,220	1.09 1.14 1.40	4.33 4.52 5.55	1.95 2.03 2.50	1.5	1,040	1,060	1,080
176180	6,680 7,030 7,750	12,030 12,650 13,960	.21 .22 .24	1.18 1.24 1.37	2.27 2.39 2.63	3.0	1,040	1,060	1,080
176181	5,910 6,240 7,400	10,630 11,240 13,320	.78 .82 .98	1.83 1.93 2.29	2.18 2.30 2.73	1.5	1,040	1,060	1,080
176182	4,430 4,580 6,830	7,970 8,250 12,300	.70 .72 1.08	13.60 14.08 20.99	1.05 1.09 1.62	1.5	1,225	1,260	1,290
176183	5,090 5,320 7,230	9,160 9,580 13,010	.90 .94 1.28	9.08 9.50 12.90	.48 .50 .68	1.0	1,170	1,215	1,260
176184	5,320 5,640 7,140	9,570 10,150 12,850	1.54 1.63 2.07	5.85 6.20 7.85	1.45 1.54 1.85	.5	1,050	1,075	1,200

Table 3.	Proximate and ultimate analyses,	heat-of-combustion, forms-of-sulfur,	free-swelling index, and ash-fusion-tempera-
	ture determinations for 90 Iowa	coal samplescontinued	

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		Proximate an	alysis							
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D176185	2.7	31.5 32.4 47.2	35.2 36.2 52.8	30.6 31.4	3.3 3.1 4.5	45.3 46.6 67.9	0.8 .8 1.2	2.7 .3 .4	17.3 17.8 25.9	1.0
D176186	2.3	30.4 31.1 49.0	31.7 32.4 51.0	35.6 36.4	2.9 2.7 4.3	37.4 38.3 60.2	.7 .7 1.1	2.5 .5 .7	20.9 21.4 33.7	.8
D176187	11.1	38.3 43.1 48.4	40.9 46.0 51.6	9.7 10.9	5.9 5.2 5.9	60.2 67.7 76.0	1.0 1.1 1.3	17.6 8.7 9.8	5.6 6.3 7.1	8.4
D176188	9.9	32.4 36.0 47.7	35.5 39.4 52.3	22.2 24.6	4.7 4.0 5.3	50.4 55.9 74.2	.8 .9 1.2	13.8 5.5 7.4	8.1 9.0 11.9	6.3
D176189	14.0	27.3 31.7 42.9	36.3 42.2 57.1	22.4 26.0	4.7 3.7 4.9	45.7 53.1 71.9	.9 1.0 1.4	17.4 5.8 7.8	8.9 10.3 14.0	10.2
D176190	9.6 	28.2 31.2 52.5	25.5 28.2 47.5	36.7 40.6	3.9 3.1 5.3	38.4 42.5 71.5	.6 .7 1.1	15.7 7.9 13.3	4.7 5.2 8.8	7.3
D176191	15.8	35.0 41.6 47.0	39.5 46.9 53.0	9.7 11.5	5.7 4.7 5.3	57.9 68.8 77.7	1.0 1.2 1.3	22.0 9.4 10.7	3.7 4.4 5.0	13.8
D176192	12.3	34.1 38.9 46.6	39.0 44.5 53.4	14.6	5.3 4.5 5.4	55.8 63.6 76.3	1.0 1.1 1.4	18.2 8.3 9.9	5.1 5.8 7.0	8.7

	Heat of co	ombustion	F	orms of sulfu	ır		Ash fusion temperature, C°			
Sample number	Kcal/kg	Btu/1b		Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D176185	4,770 4,900 7,150	8,590 8,830 12,880	0.70 .72 1.05	16.16 16.61 24.23	0.42 .43 .63	1.5	1,155	1,175	1,200	
D176186	4,330 4,430 6,970	7,790 7,970 12,540	.79 .81 1.27	18.75 19.19 30.19	1.35 1.38 2.17	1.0	1,065	1,095	1,120	
D176187	6,060 6,810 7,650	10,900 12,260 13,760	.87 .98 1.10	2.52 2.82 3.18	2.24 2.52 2.83	1.5	1,110	1,140	1,170	
D176188	5,130 5,700 7,560	9,240 10,260 13,610	.49 .54 .72	5.37 5.96 7.91	2.24 2.49 3.30	.5	1,145	1,170	1,205	
D176189	4,540 5,280 7,140	8,170 9,500 12,850	.99 1.15 1.56	6.40 7.44 10.06	1.52 1.77 2.39	.0	1,080	1,110	1,140	
D176190	3,710 4,100 6,900	6,670 7,380 12,420	.72 .80 1.34	2.77 3.06 5.16	1.22 1.35 2.27	.0	1,080	1,110	1,140	
D176191	5,700 6,770 7,650	10,260 12,190 13,770	.01 .01 .01	1.33 1.58 1.79	2.41 2.86 3.23	5.0	1,080	1,110	1,140	
D176192	5,580 6,360 7,630	10,040 11,450 13,730	.27 .31 .37	1.82 2.08 2.49	2.99 3.41 4.09	.5	1,095	1,120	1,150	

		Proximate and	alysis			Ulti	mate analysis	;		
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
 D176193	8.9	29.3	28.1	33.7	4.1	39.6	0.7	13.3	8.6	6.4
		32.2	30.8	37.0	3.4	43.5	.8	5.9	9.4	
		51.0	49.0		5.4	69.0	1.2	9.4	15.0	
D176194	7.7	31.1	31.2	30.0	4.0	41.0	.7	13.5	10.8	2.7
		33.7	33.8	32.5	3.4	44.4	.8	7.2	11.7	
		49.9	50.1		5.0	65.8	1.1	10.7	17.3	
D176195	13.3	36.3	39.8	10.6	5.7	56.9	1.0	19.5	6.3	10.4
		41.9	45.9	12.2	4.9	65.6	1.2	8.9	7.3	
		47.7	52.3		5.5	74.8	1.3	10.1	8.3	
D176196	13.2	37.1	31.8	17.9	5.2	52.3	.9	20.0	3.7	10.8
	***	42.7	36.6	20.6	4.3	60.3	1.0	9.5	4.3	
		53.8	46.2		5.4	75.9	1.3	12.0	5.4	
D176197	11.6	37.5	39.6	11.3	5.7	58.6	1.0	16.8	6.6	8.9
		42.4	44.8	12.8	5.0	66.3	1.1	7.3	7.5	
		48.6	51.4		5.7	76.0	1.3	8.4	8.6	
D176198	12.7	30.4	33.0	23.9	4.9	46.2	.8	18.5	5.7	9.3
		34.8	37.8	27.4	4.0	52.9	.9	8.3	6.5	
		47.9	52.1		5.5	72.9	1.3	11.4	9.0	.
D176199	12.0	40.6	35.5	11.9	5.6	58.3	.9	18.1	5.2	10.1
		46.1	40.3	13.5	4.8	66.2	1.0	8.4	5.9	
		53.4	46.6		5.6	76.6	1.2	9.8	6.8	
D179838	12.2	34.0	43.2	10.6	5.4	58.8	-7	20.0	4.5	2.3
		38,7	49.2	12.1	4.6	67.0	.8	10.4	5.1	
		44.0	56.0		5.2	76.2	.9	11.9	5.8	

	Heat of co	ombustion	F	forms of sulfu	ır		Ash fusion temperature, C°			
Sample number	Kcal/kg	Btu/1b	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D176193	3,900 4,280 6,790	7,020 7,710 12,230	1.15 1.26 2.00	5.16 5.66 8.00	2.27 2.49 3.95	0.5	1,130	1,160	1,280	
D176194	4,210 4,560 6,760	7,580 8,210 12,170	1.89 2.05 3.03	6.03 6.53 9.68	2.84 3.08 4.56	.5	1,080	1,110	1,140	
D176195	5,800 6,690 7,620	10,440 12,040 13,720	.65 .75 .85	3.49 4.03 4.59	2.11 2.43 2.77	1.0	1,105	1,130	1,160	
D176196	5,070 5,840 7,360	9,130 10,520 13,250	.37 .43 .54	1.66 1.91 2.41	1.68 1.94 2.44	1.0	1,140	1,170	1,200	
D176197	5,930 6,710 7,690	10,670 12,070 13,840	.71 .80 .92	3.97 4.49 5.15	1.93 2.18 2.50	.5	1,275	1,305	1,330	
D176198	4,690 5,380 7,400	8,450 9,680 13,330	.62 .71 .98	3.12 3.57 4.92	1.94 2.22 3.06	.5	1,175	1,205	1,230	
D176199	5,910 6,710 7,760	10,630 12,080 13,970	.31 .35 .41	2.27 2.58 2.98	2.61 2.97 3.43	4.0	1,230	1,260	1,290	
D179838	5,910 6,730 7,650	10,630 12,110 13,770	.82 .93 1.06	2.25 2.56 2.91	1.40 1.59 1.82	1.0	1,095	1,155	1,180	

		Proximate an	alysis				·			
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D179840	13.5	31.6 36.5	45.2 52.3	9.7 11.2	5.9 5.1	60.3 69.7	0.8	20.1 9.4	3.2 3.7	4.7
		41.1	58.9		5.7	78.5	1.0	10.5	4.2	
D179841	7.9	35.6 38.7 47.5	39.4 42.8 52.5	17.1 18.6	4.8 4.3 5.2	58.2 63.2 77.6	1.0 1.1 1.3	12.5 5.9 7.3	5.4 6.9 8.5	2.0
D179843	9.1 	28.2 31.0 48.5	29.9 32.9 51.5	32.8 36.1	3.5 2.7 4.3	34.4 37.8 59.2	•5 •6 •9	9.6 1.7 2.6	19.2 21.1 33.0	3.1
D179844	13.1	29.5 33.9 49.4	30.2 34.8 50.6	27.2 31.3	4.6 3.6 5.3	42.3 48.7 70.9	.8 .9 1.3	15.9 4.9 7.1	9.2 10.7 15.4	5.0
D179846	9.4	27.9 30.8 46.9	31.6 34.9 53.1	31.1 34.3	3.6 2.8 4.3	36.6 40.4 61.5	-4 -4 -7	10.0 1.8 2.8	18.3 20.2 30.8	3.7
D179847	13.6	37.9 43.9 49.2	39.2 45.4 50.8	9.3 10.8	5.8 5.0 5.6	61.0 70.6 79.1	.9 1.0 1.2	19.4 8.5 9.5	3.6 4.2 4.7	7.0
D179848	18.7	32.1 39.5 44.5	40.0 49.2 55.5	9.2 11.3	6.0 4.8 5.4	56.9 70.0 78.9	.7 .9 1.0	24.8 10.1 11.3	2.4 3.0 3.3	12.1
D179850	13.0	31.7 36.4 45.4	38.1 43.8 54.6	17.2 19.8	5.2 4.3 5.4	51.9 59.7 74.4	.7 .8 1.0	19.5 9.1 11.4	5.5 6.3 7.9	3.4

	Heat of co	ombustion	F	Forms of sulfur			Ash fusion temperature, C°			
Sample number	Kca]/kg	Btu/1b	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D179840	6,010 6,940 7,820	10,810 12,500 14,080	0.34 .39 .44	1.92 2.22 2.50	0.97 1.12 1.26	1.0	1,100	1,140	1,160	
D179841	5,910 6,420 7,880	10,640 11,550 14,190	.36 .39 .48	3.42 3.71 4.57	2.45 2.66 3.27	3.5	1,105	1,140	1,165	
D179843	3,840 4,220 6,610	6,910 7,600 11,890	1.34 1.47 2.31	16.56 18.22 28.50	1.30 1.43 2.24	.0	1,180	1,225	1,235	
D179844	4,410 5,080 7,390	7,940 9,410 13,300	.84 .97 1.41	6.78 7.80 11.36	.64 .74 1.07	.0	1,050	1,105	1,125	
D179846	4,040 4,460 6,800	7,280 8,040 12,240	.72 .79 1.21	15.47 17.08 26.00	2.09 2.31 3.51	.0	1,195	1,240	1,260	
0179847	6,090 7,050 7,900	10,970 12,700 14,230	.22 .25 .29	1.66 1.92 2.15	1.71 1.98 2.22	1.5	1,120	1,155	1,175	
D179848	5,600 6,890 7,770	10,080 12,400 13,980	.04 .05 .06	.33 .41 .46	2.05 2.52 2.84	2.0	1,125	1,175	1,195	
D179850	5,280 6,070 7,560	9,500 10,920 13,610	.62 .71 .89	3.21 3.69 4.60	1.71 1.97 2.45	•0	1,100	1,155	1,185	

Table 3.	Proximate and ultim	ate analy	rses, hea	t-of-combustion,	forms-of-sulfur,	free-swelling index	, and ash-fusion-tempera-
	ture determinations	for 90 I	lowa coal	samplescontin	ued		

		Proximate and	alysis				_			
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D179851	12.5	39.0	40.1	8.4	6.0	62.5	0.8	18.8	3.5	4.9
		44.6 49.3	45.8 50.7	9.6	5.3 5.8	/1.4 79.0	.9 1.0	8.8 9.7	4.0	
D179853	12.0	34.5	22.8	30.7	4.5	44.7	.8	11.9	7.4	4.6
		39.2 60.2	25.9 39.8	34.9	3.6 5.5	50.8 78.0	.9 1.4	1.4 2.2	8.4 12.9	
D179854	9.5	33.4	35.0	22.1	4.8	49.7	.7	16.0	6.7	2.2
		36.9 48.8	38.7 51.2	24.4	4.1 5.5	54.9 72.7	.8 1.0	8.3 11.0	7.4 9.8	
D179855	12.2	34.4 39.2 44.6	42.8 48.7 55.4	10.6 12.1	5.5 4.7 5.4	59.6 67.9 77.2	.7 .8 .9	18.6 8.8 10.0	5.0 5.7 6.5	2.7
D179856	8.8	28.4 31.1 45.4	34.1 37.4 54.6	28.7 31.5	4.0 3.3 4.8	40.1 44.0 64.2	.7 .8 1.1	13.4 6.1 8.9	13.1 14.4 21.0	2.4
D185601	13.9	33.4 38.8	39.3 45.6	13.4 15.6	5.3	53.4 62.0	•8 •9	18.3 6.9	8.9 10.3	10.1
		45.9	54.1		5.2	73.5	1.1	8.2	12.2	
D185602	15.4	33.5 39.6 46.9	38.0 44.9 53.1	13.1 15.5	5.5 4.5 5.3	54.2 64.1 75.8	.8 .9 1.1	19.3 6.6 7.8	7.1 8.4 9.9	12.4
D185603	15.6	35.6 42.2 48.2	38.3 45.4 51.8	10.5	5.7 4.7 5.4	57.7 68.4 78.1	1.0 1.2 1.4	21.3 8.8 10.1	3.8 4.5 5.1	12.8

	Heat of co	ombustion	F	forms of sulfu	Ir		Ash fusion temperature, C°			
Sample number	Kcal/kg	Btu/1b	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D179851	6,270 7,160 7,920	11,280 12,890 14,260	0.20 .23 .25	1.73 1.98 2.19	1.60 1.83 2.02	1.5	1,045	1,070	1,095	
D179853	4,420 5,020 7,710	7,950 9,030 13,870	.44 .50 .77	5.85 6.65 10.21	1.07 1.22 1.87	1.0	1,235	1,270	1,285	
D179854	5,050 5,580 7,380	9,090 10,040 13,290	.66 .73 .96	3.69 4.08 5.39	2.28 2.52 3.33	1.0	1,100	1,150	1,175	
D179855	5,960 6,780 7,710	10,720 12,210 13,890	1.37 1.56 1.77	3.13 3.56 4.05	.51 .58 .66	1.0	1,240	1,290	1,320	
D179856	4,340 4,760 6,940	7,810 8,560 12,500	2.98 3.27 4.77	11.70 12.83 18.72	.69 .76 1.10	.0	1,245	1,290	1,325	
D185601	5,500 6,390 7,560	9,900 11,490 13,610	1.21 1.41 1.66	4.77 5.54 6.56	2.92 3.39 4.02	1.0	1,180	1,235	1,295	
D185602	5,530 6,530 7,730	9,950 11,760 13,910	.96 1.13 1.34	3.37 3.98 4.71	2.76 3.26 3.86	1.0	1,095	1,155	1,220	
D185603	5,780 6,850 7,830	10,410 12,340 14,090	•52 •62 •70	2.30 2.73 3.11	.97 1.15 1.31	1.0	1,165	1,205	1,255	

		Proximate an	alysis				_			
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D185604	14.8	35.6	39.0	10.6	5.6	58.8	1.2	20.9	2.9	11.9
		41.8	45.8	12.6	4.6	69.0	1.4	9.1	3.4	
		47.7	52.3		5.3	78.8	1.6	10.4	3.9	
D185605	13.3	34.2	43.5	9.0	5.6	61.0	1.3	20.3	2.8	10.0
		39.4	50.2	10.4	4.8	70.4	1.5	9.8	3.2	
		44.0	56.0		5.3	78.5	1.7	10.9	3.6	
D185606	21.1	31.6	36.1	11.2	5.7	51.1	.9	27.9	3.2	16.5
		40.1	45.8	14.2	4.3	64.8	1.1	11.6	4.1	
		46.7	53.3		5.0	75.5	1.3	13.5	4.7	
D185609	16.5	34.8	30.3	18.4	5.4	50,7	1.0	19.7	4.8	13.4
		41.7	36.3	22.0	4.3	60.7	1.2	6.0	5.7	
		53.5	46.5		5.5	77.9	1.5	7.7	7.4	
D185610	11.9	34.9	38.6	14.6	5.1	54.2	1.1	17.9	7.1	4.5
		39.6	43.8	16.6	4.3	61.5	1.2	8.3	8.1	
		47.5	52.5		5.1	73.7	1.5	10.0	9.7	
D185611	14.5	35.8	34.5	15.2	5.3	54.1	.9	22.0	2.5	11.9
		41.9	40.4	17.8	4.3	63.3	1.1	10.7	2.9	
		50.9	49.1		5.2	77.0	1.3	13.0	3.6	
D185612	14.9	33.2	41.2	10.7	5.4	55.8	1.0	21.2	5.8	9.8
	***	39.0	48.4	12.6	4.4	65.6	1.2	9.3	6.8	
		44.6	55.4		5.0	75.0	1.3	10.7	7.8	
D185613	16.6	34.3	42.1	7.0	5.9	59.9	1.1	23.0	3.1	12.6
÷	-	41.1	50.5	8.4	4.9	71.8	1.3	9.9	3.7	
		44.9	55.1		5.3	78.4	1.4	10.8	4.1	

	Heat of co	ombustion	F	Forms of sulfu	ır		Ash fusion temperature, C°			
Sample number	Kcal/kg	Btu/1b	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D185604	5,850 6,860 7,840	10,530 12,350 14,110	0.19 .22 .25	1.66 1.95 2.23	1.05 1.23 1.41	1.5	1,040	1,090	1,140	
D185605	6,100 7,030 7,850	10,980 12,660 14,130	.37 .43 .48	1.68 1.94 2.16	.73 .84 .94	1.0	1,055	1,115	1,155	
D185606	5,030 6,370 7,430	9,050 11,470 13,370	.47 .60 .69	.99 1.25 1.46	1.70 2.15 2.51	.0	1,100	1,150	1,215	
D185609	5,090 6,100 7,820	9,160 10,980 14,080	.54 .65 .83	3.43 4.11 5.27	.85 1.02 1.31	1.0	1,215	1,265	1,325	
D185610	5,490 6,230 7,470	9,880 11,220 13,440	.08 .09 .11	4.18 4.74 5.69	2.83 3.21 3.85	.0	1,150	1,205	1,255	
D185611	5,220 6,100 7,420	9,390 10,980 13,360	.76 .89 1.08	.45 .53 .64	1.27 1.49 1.81	1.5	1,485	1,530	1,540	
D185612	5,670 6,670 7,630	10,210 12,000 13,730	.51 .60 .69	3.36 3.95 4.52	1.94 2.28 2.61	1.0	1,050	1,115	1,165	
D185613	5,990 7,180 7,840	10,780 12,920 14,110	•20 •24 •26	1.01 1.21 1.32	1.87 2.24 2.45	1.5	1,120	1,170	1,230	

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		Proximate an	alysis							
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D186062	17.0	32.1 38.7 44.5	40.0 48.2 55.5	10.9 13.1	5.5 4.4 5.0	56.7 68.3 78.6	1.2 1.4 1.7	20.6 6.6 7.6	5.1 6.1 7.1	10.3
D186063	16.8 	33.7 40.5 48.0	36.5 43.9 52.0	13.0 15.6	5.5 4.4 5.2	55.0 66.1 78.3	1.2 1.4 1.7	20.3 6.5 7.6	4.9 5.9 7.0	11.8
D186064	17.7	33.2 40.3 45.2	40.3 49.0 54.8	8.8 10.7	5.8 4.7 5.2	57.9 70.4 78.8	1.3 1.6 1.8	22.4 8.1 9.1	3.8 4.6 5.2	13.2
D186065	13.1	34.6 39.8 48.3	37.1 42.7 51.7	15.2 17.5	5.3 4.4 5.4	53.7 61.8 74.9	.9 1.0 1.3	18.0 7.3 8.9	6.9 7.9 9.6	7.2
D186066	13.0	37.8 43.4 49.6	38.4 44.1 50.4	10.8 12.4	5.7 4.9 5.6	59.8 68.7 78.5	.8 .9 1.0	18.2 7.6 8.7	4.7 5.4 6.2	9.1
D186067	4.3	35.2 36.8 49.0	36.7 38.3 51.0	23.8 24.9 	4.3 4.0 5.3	49.5 51.7 68.8	1.1 1.1 1.5	11.8 8.3 11.1	9.5 9.9 13.2	.7
D186068	3.1	38.5 39.7 46.2	44.9 46.3 53.8	13.5 13.9	4.6 4.4 5.1	65.5 67.6 78.5	1.1 1.1 1.3	11.5 9.0 10.5	3.8 3.9 4.6	.7
D186069	2.7	23.3 23.9 32.6	48.1 49.4 67.4	25.9 26.6	4.5 4.3 5.9	53.6 55.1 75.1	.9 .9 1.3	10.4 8.2 11.2	4.7 4.8 6.6	.3

	Heat of co	mbustion	F	orms of sulfu	ır		Ash fuston temperature, C°			
Sample number	Kcal/kg	Btu/lb	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D186062	5,690 6,850 7,890	10,240 12,330 14,200	0.26 .31 .36	3.52 4.24 4.88	1.30 1.57 1.80	1.0	1,145	1,205	1,260	
D186063 ·	5,530 6,650 7,880	9,960 11,970 14,190	.51 .61 .73	2.85 3.43 4.06	1.57 1.89 2.24	1.0	1,125	1,180	1,245	
D186064	5,820 7,080 7,920	10,480 12,740 14,260	.36 .44 .49	2.39 2.90 3.25	1.02 1.24 1.39	1.0	1,085	1,135	1,185	
D186065	5,460 6,290 7,620	9,830 11,310 13,710	1.38 1.59 1.92	3.34 3.84 4.66	2.14 2.46 2.98	1.0	1,095	1,155	1,215	
D186066	6,080 6,990 7,980	10,940 12,570 14,360	.45 .52 .59	1.81 2.08 2.38	2.44 2.80 3.20	1.0	1,175	1,235	1,290	
D186067	5,150 5,380 7,160	9,260 9,680 12,880	1.85 1.93 2.57	7.21 7.53 10.03	.45 .47 .63	.0	1,050	1,100	1,150	
D186068	6,440 6,650 7,720	11,590 11,960 13,900	.60 .62 .72	2.35 2.43 2.82	.85 .88 1.02	1.0	1,180	1,235	1,290	
D186069	5,570 5,720 7,790	10,020 10,300 14,030	.72 .74 1.01	2.28 2.34 3.19	1.74 1.79 2.44	1.0	1,240	1,300	1,365	

		Proximate and	alysis			Ulti	mate analysis	;		_
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D186070	3.3	38.2	45.5	13.0	4.8	64.4	1.2	11.1	5.4	0.5
		39.5	47.1	13.4	4.6	66.6	1.2	8.4	5.6	
		45.6	54.4		5.3	76.9	1.4	9.8	6.5	
D186071	2.8	38.5	43.2	15.5	4.6	62.0	1.1	8.9	7.9	.4
- · ·		39.6	44.4	15.9	4.4	63.8	1.1	6.6	8.1	
		47.1	52.9		5.2	75.9	1.3	7.8	9.7	••••
D186072	7.0	39.7	30.6	22.7	4.6	53.7	.9	12.8	5.3	2.1
		42.7	32.9	24.4	4.1	57.7	1.0	7.1	5.7	
		56.5	43.5		5.4	76.4	1.3	9.4	7.5	6 10 4
D186073	9.4	35.1	38.2	17.3	4.7	57.4	.9	14.5	5.3	3.1
		38.7	42.2	19.1	4.0	63.4	1.0	6.8	5.8	
		47.9	52.1		5.0	78.3	1.2	8.4	7.2	
D186074	12.2	35.1	35.4	17.3	5.0	53.5	.9	17.2	6.0	7.0
		40.0	40.3	19.7	4.2	60.9	1.0	7.2	6.8	
		49.8	50.2		5.2	75.9	1.3	9.0	8.5	
D186075	16.2	33.8	37.1	12.9	5.4	56.4	1.2	20.8	3.2	11.3
		40.3	44.3	15.4	4.3	67.3	1.4	7.6	3.8	
		47.7	52.3		5.1	79.5	1.7	9.0	4.5	
D186076	15.1	34.1	35.6	15.2	5.5	55.4	1.1	19-6	3.1	9.3
		40.2	41.9	17.9	4.5	65.3	1.3	7.3	3.7	
		48.9	51.1		5.5	79.5	1.6	8.9	4.4	
D192368	5.0	39.0	36.3	19.7	4.3	58.4	1.0	10.9	5.7	2.5
		41.1	38.2	20.7	3.9	61.5	1.1	6.8	6.0	
		51.8	48.2		5.0	77.6	1.3	8.6	7.6	

	Heat of co	ombustion	F	orms of sulfu	ır		Ash fusion	temperature,	, C°
Sample number	Kcal/kg	Btu/lb	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid
D186070	6,390 6,610 7,640	11,510 11,900 13,750	0.73 .75 .87	2.88 2.98 3.44	1.81 1.87 2.16	1.0	1,100	1,155	1,205
D186071	6,350 6,530 7,770	11,430 11,760 13,990	.51 .52 .62	6.02 6.19 7.37	1.36 1.40 1.66	1.0	1,125	1,175	1,240
D186072	5,290 5,680 7,520	9,520 10,230 13,540	.83 .89 1.18	1.97 2.12 2.80	2.51 2.70 3.57	1.0	1,335	1,380	1,430
D176073	5,680 6,270 7,750	10,230 11,290 13,950	.98 1.08 1.34	2.58 2.85 3.52	1.78 1.96 2.43	1.0	1,105	1,175	1,230
D186074	5,410 6,160 7,680	9,740 11,090 13,820	.97 1.10 1.38	3.64 4.15 5.16	1.44 1.54 2.04	1.0	1,095	1,155	1,225
D186075	5,640 6,730 7,950	10,150 12,110 14,310	.24 .29 .34	2.37 2.83 3.34	.63 .75 .89	1.0	1,105	1,155	1,220
D186076	5,480 6,460 7,870	9,870 11,630 14,160	.21 .25 .30	2.05 2.41 2.94	.82 .97 1.18	1.0	1,125	1,180	1,235
D192368	5,790 6,090 7,680	10,410 10,960 13,830	.18 .19 .24	2.49 2.62 3.31	3.01 3.17 4.00	1.5	1,150	1,205	1,265

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		Proximate an	alysis			Ulti	mate analysis	6		_
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D192369	13.8	32.2	43.1	10.9	5.4	58.5	1.3	21.1	2.8	10.5
		37.4	50.0	12.6	4.5	67.9	1.5	10.2	3.2	
		42.8	57.2		5.1	77.7	1.7	11.7	3.7	
D192370	8.1	33.5	36.6	21.8	4.2	44.1	.8	16.4	12.8	3.2
		36.5	39.8	23.7	3.6	48.0	.9	10.0	13.9	
		47.8	52.2		4.7	62.9	1.1	13.1	18.3	
D192371	11.1	29.8	39.2	19.9	4.5	46.7	.8	17.6	10.6	6.7
		33.5	44.1	22.4	3.7	52.5	.9	8.7	11.9	
		44.5	56.8		4.7	67.7	1.2	11.2	15.4	
D192372	10.0	35.0	43.7	11.3	5.3	59.0	1.0	17.9	5.5	6.4
		38.9	48.6	12.6	4.7	65.6	1.1	10.0	6.1	
		44.5	55.5	40 40 40	5.3	75.0	1.3	11.4	7.0	
D192373	8.6	33.1	47.3	11.0	5.0	59.3	1.0	18.9	4.8	4.8
		36.2	51.8	12.0	4.4	64.9	1.1	12.3	5.3	
		41.2	58.8		5.0	73.8	1.2	14.0	6.0	
D192374	5.3	32.7	48.5	13.5	4.7	60.1	1.2	13.6	6.9	2.3
		34.5	51.2	14.3	4.3	63.5	1.3	9.4	7.3	
		40.3	59.7		5.1	74.0	1.5	10.9	8.5	
D192375	7.9	31.7	39.6	20.8	4.4	48.6	•8	14.9	10.5	3.7
		34.4	43.0	22.6	3.8	52.8	.9	8.6	11.4	
		44.5	55.5	**	4.9	68.2	1.1	11.0	14.7	
D192376	9.0	33.6	39.7	17.7	4,9	55.3	1.0	15.4	5.7	5.3
		36.9	43.6	19.5	4.3	60.8	i. ĭ	8.1	6.3	
		45.8	54.2		5.3	75.4	1.4	10.1	7.8	

	Heat of co	mbustion	F	orms of sulfu	ır		Ash fusion temperature, C°			
Sample number	Kcal/kg	Btu/1b	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D192369	5,760 6,690 7,660	10,380 12,040 13,780	0.45 .52 .60	1.41 1.64 1.87	0.90 1.04 1.20	1.0	1,040	1,095	1,150	
D192370	4,540 4,940 6,480	8,170 8,890 11,660	3.57 3.88 5.09	6.30 6.86 8.99	2.89 3.14 4.12	.5	1,180	1,235	1,290	
D192371	4,790 5,380 6,940	8,620 9,690 12,490	1.87 2.10 2.71	5.57 6.27 8.07	3.13 3.52 4.54	.5	1,180	1,230	1,295	
D192372	5,900 6,560 7,500	10,630 11,810 13,500	1.04 1.16 1.32	2.11 2.34 2.68	2.39 2.66 3.04	1.0	1,040	1,100	1,150	
D192373	5,840 6,390 7,270	10,510 11,500 13,080	1.04 1.14 1.29	1.59 1.74 1.98	2.15 2.35 2.67	1.0	985	1,040	1,095	
D192374	6,040 6,380 7,440	10,870 11,480 13,390	1.12 1.18 1.38	3.89 4.11 4.79	1.85 1.95 2.28	1.0	1,040	1,095	1,155	
D192375	4,990 5,420 7,000	8,990 9,760 12,610	1.88 2.04 2.64	5.84 6.34 8.19	2.73 2.96 3.83	1.0	1,015	1,070	1,125	
0192376	5,460 6,000 7,450	9,820 10,800 13,400	1.29 1.42 1.76	2.47 2.71 3.37	1.99 2.19 2.71	1.0	1,235	1,290	1,345	

Table 3. Proximate and ultimate analyses, heat-of-combustion, forms-of-sulfur, free-swelling index, and ash-fusion-temperature determinations for 90 Iowa coal samples--continued

		Proximate and	alysis							
Sample number	Moisture	Volatile matter	Fixed carbon	Ash	Hydrogen	Carbon	Nitrogen	Oxygen	Sulfur	Air-dried loss
D192377	7.7	32.2	41.6	18.4	4.4	49.6	1.0	18.1	8.6	3.4
		34.9	45.1	19.9	3.8	53.7	1.1	12.2	9.3	
		43.6	56.3		4.8	67.1	1.4	15.2	11.6	
D192378	10.6	34.5	39.2	15.7	4.8	53.2	.9	18.4	7.0	6.4
	*	38.6	43.8	17.6	4.1	59.5	1.0	10.0	7.8	
		46.8	53.2		4.9	72.2	1.2	12.2	9.5	
D192379	6.5	34.2	38.5	20.8	4.6	55.7	1.2	14.4	3.2	3.5
		36.6	41.2	22.2	4.1	59.6	1.3	9.2	3.4	
		47.0	53.0		5.3	76.6	1.7	11.9	4.4	

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Sample number	Heat of combustion		F	Forms of sulfur			Ash fusion temperature, C°			
	Kcal/kg	Btu/1b	Sulfate	Pyritic	Organic	Free Swelling	Initial deformation	Softening	Fluid	
D192377	5,110 5,540 6,920	9,210 9,970 12,460	2.74 2.97 3.71	3.41 3.69 4.61	2.42 2.62 3.27	0.5	1,070	1,125	1,180	
D192378	5,280 5,900 7,160	9,500 10,620 12,890	2.07 2.32 2.81	3.03 3.39 4.11	1.87 2.09 2.54	.5	1,180	1,235	1,290	
D192379	5,450 5,830 7,500	9,810 10,490 13,500	.52 .56 .72	2.25 2.41 3.09	.42 .45 .58	1.0	1,095	1,150	1,200	

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Sample number	Ash (percent)	SiO ₂ (percent)	Al ₂ 0 ₃ (percent)	Can (percent)	MgO (percent)	Na ₂ 0 (percent)	K ₂ 0 (percent)	Fe ₂ 03 (percent)	TiO ₂ (percent)
D166027	23.7	16	8.0	16	0.38	0.07	0.61	35	0.20
D166028	25.5	27	14	15	.70	.12	1.4	22	.50
D166029	15.5	21	11	17	.51	.08	.78	28	.50
D166030	15.1	28	13	4.5	.46	.15	.95	37	.50
D166031	20.0	22	12	.84	•56	.12	.85	42	.30
0166032	20.9	21	7.2	19	.46	.11	.56	31	- 20
D166033	17.9	24	9.3	13	.45	.15	.87	32	.50
D166034	21.9	15	5.5	8.1	.27	.09	.35	46	.15
D166035	21.0	14	5.8	28	.38	.09	.36	22	.20
D166036	12.8	16	7.3	19	.40	.11	.43	30	.30
D166037	14.2	28	9.6	18	.46	.30	.69	22	.20
D166038	15.3	29	9.7	15	.46	.30	.73	25	.30
D166039	18.3	44	18	9.3	.88	.26	1.7	15	.50
D166040	29.3	49	21	5.2	2.21	.30	2.5	13	1.0
D166041	17.2	18	10	.71	.22	.11	.34	46	.30
D166042	16.2	31	11	9.3	.33	.11	.72	30	.50
D166043	11.2	23	11	2.4	.25	.12	.40	44	.50
D176169	16.2	12	4.1	30	.51	.14	.57	18	.25
D176170	18.5	28	9.6	4.5	.50	.30	1.0	42	.64
D176171	29.5	48	16	6.2	.38	.24	1.2	15	.90
D176172	25.6	12	9.4	7.1	.17	.15	.21	53	.19
D176173	27.9	24	8.5	19	1.36	.20	.71	22	.44
D176174	20.1	46	23	4.6	.51	.32	1.2	14	1.2
D176175	16.6	31	17	15	.61	.31	1.2	16	.52
D176176	22.2	11	6.5	7.0	.22	.14	.34	56	.30

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{Coal ashed at 525°C. L, less than the values shown; N, not detected; B, not determined. S after element title indicates determinations by semiquantitative emission spectrography }

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples

Sample number	P ₂ 05 (percent)	SO3 (percent)	Ag-S (ppm)	B-S (ppm)	Ba-S (ppm)	Be-S (ppm)	Cd (ppm)	Ce-S (ppm)	Co-S (ppm)	Cr-S (ppm)
D166027	0.13	8.7	N	300	3,000	7	2.0		30	
D166028	.42	6.7	Ň	300	2,000	7	6.0	N	30	70
D166029	.18	8.6	N	700	3,000	15	13.0	N	30	70
D166030	.13	3.2	Ň	1.000	150	15	4.0	N	30	70
D166031	.16	2.6	N	700	100	7	1.0L	N	70	70
D166032	.17	12	N	700	100	7	1.0L	N	15	100
D166033	.070	6.1	N	700	150	7	8.0	N	20	70
D166034	.060	6.0	N	500	70	7	1.0L	N	15	30
D166035	.51	14	N	700	70	7	1.0L	N	15	50
D166036	.72	6.5	N	1,500	100	15	1.0L	N	15	50
D166037	.98	8.4	N	1,500	500	7	2.0	N	30	50
D166038	.87	11	N	1,500	300	10	2.0	N	30	50
D166039	.56	6.0	N	1,000	200	15	1.0L	500L	30	70
D166040	.30	3.4	N	700	300	7	1.0L	500L	30	100
D166041	.060	3.4	N	700	150	15	85.0	N	70	50
D166042	.11	6.1	N	700	150	15	33.0	N	30	70
D166043	.10	2.7	1.5	1,000	50	20	28.0	N	70	70
D176169	1.0L	15	5	700	100	10	1.0L	N	20	100
D176170	1.0L	4.8	N	500	150	15	1.0L	N	30	70
D176171	1.0L	5.7	3	200	200	15	165	N	150	150
D176172	1.0L	6.1	3	300	150	7	58.0	500L	200	30
D176173	1.0L	10	N	300	150	10	1.0	N	30	70
D176174	1.0L	2.7	N	500	500	15	64.0	500L	30	150
D176175	1.0L	5.1	N	1,000	1,000	15	18.0	N	30	100
D176176	1.0L	6.3	N	300	300	10	1.0L	500L	70	50

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Cu (ppm)	Ga-S (ppm)	Ge-S (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)
	62	B	150	N	40	700	15	201		150
D166028	58	B	100	100L	91	1,000	15	201	Ň	150
D166029	84	B	200	-~ N	43	1,000	30	201	B	150
D166030	160	B	100	100	76	700	15	201	Ň	150
D166031	204	B	50	100L	76	3,000	15	20L	150	200
D166032	98	B	70	N	22	1,000	15	20L	В	70
D166033	98	В	100	N	46	700	15	20L	8	70
D166034	100	В	30	N	17	500	15	20L	В	50
D166035	56	8	100	N	26	2,000	7	20L	В	100
D166036	86	В	150	N	32	1,500	7	20L	В	150
D166037	60	8	70	N	48	1,000	7	20L	B	150
D166038	60	В	70	N	57	700	7	20L	В	150
D166039	94	20	70	100L	91	700	7	20L	150L	150
D166040	80	8	70	100L	114	700	10	20L	150L	200
D166041	148	8	100	N	93	300	30	20L	8	150
D166042	104	В	150	N	138	700	70	20L	B	200
D166043	142	В	150	100L	75	300	30	20L	150L	70
D176169	82	15	150	N	10	2,340	100	20L	В	100
D176170	116	8	200	N	35	585	7	20L	В	150
D176171	162	15	30	N	143	485	150	20L	B	300
D176172	180	В	150	100	53	440	30	20L	150	700
D176173	70	B	70	100L	38	1,530	50	20L	N	150
D176174	120	50	30	100	179	255	15	30	150	100
D176175	120	30	100	100	99	705	20	20L	150	150
D176176	70	8	70	100L	40	350	N	20L	150L	200

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Pb (ppm)	Sc-S (ppm)	Sr-S (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
D166027	400	15	150	70	30	В	1,480	70
D166028	110	30	100	150	50	В	3,000	70
D166029	250	30	100	150	30	В	7,800	70
D166030	70	30	150	150	30	В	1,500	70
D166031	75	20	150	150	70	В	1,080	70
D166032	70	15	70	100	30	B	292	70
D166033	45	20	70	100	30	B	2,400	70
D166034	90	10	50	70	30	B	52	50
D166035	360	15	150	70	30	В	188	70
D166036	450	15	150	70	30	B	364	70
D166037	305	15	200	70	30	В	2,240	50
D166038	345	15	200	70	30	3	1,740	70
D166039	150	20	150	150	30	3	740	100
D166040	120	20	150	150	30	В	260	150
D166041	500	20	30	100	70	B	32,000	70
D166042	270	30	70	150	50	В	11,000	150
D166043	630	15	150	150	70	В	7,800	70
D176169	110	15	100	300	50	5	64	50
D176170	560	30	150	70	50	8	64	100
D176171	200	30	300	300	30	7	14,200	200
D176172	220	30	2.000	150	30	ß	8,840	50
D176173	90	30	100	100	70	8	200	100
D176174	75	50	150	150	70	5	17,500	200
D176175	120	30	300	150	100	7	6,000	70
D176176	300	20	300	70	30	В	60	70

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Ash (percent)	SiO ₂ (percent)	A1 ₂ 03 (percent)	CaO (percent)	MgO (percent)	Na ₂ 0 (percent)	K ₂ 0 (percent)	Fe ₂ 03 (percent)	TiO ₂ (percent)
D176177	16.6	15	9.6	5.7	.30	0.16	.62	52	48
D176178	30.4	31	17	1.3	.76	.40	1.6	35	.73
D176179	18.4	24	8.4	8.0	.58	.45	1.0	39	•58
D176180	10.3	23	7.5	20	.76	.50	1.1	20	.42
D176181	18.1	29	9.2	15	.70	-40	1.1	22	.48
D176182	41.6	5.6	1.9	13	.23	.09	.14	51	.17
D176183	29.4	12	5.7	11	.27	.19	.43	46	.23
D176184	22.5	18	10	3.9	.32	.26	.62	50	.45
D176185	32.0	3.5	1.4	8.1	.13	.11	.070	64	.14
D176186	34.2	8.0	4.3	3.8	.10L	.09L	.11	66	.15
D176187	12.6	13	7.3	12	.32	.18	.51	42	.37
D176188	25.3	24	16	7.2	.55	.18	1.1	32	.63
D176189	24.9	26	10	3.0	.76	.35	1.1	45	.57
D176190	45.9	42	6.6	14	.88	.61	1.2	14	.64
D176191	12.8	25	7.1	19	.42	.36	.60	21	.33
D176192	17.0	35	12	7.5	.53	.32	1.2	23	.52
D176193	37.8	43	18	1.2	.37	.28	.87	28	.91
D176194	31.5	28	12	4.5	.37	.26	.72	36	.50
D176195	10.7	13	5.1	6.9	.25	.40	.36	52	.36
D176196	23.7	20	7.4	24	.56	.26	•57	16	.36
D176197	16.4	6.4	4.8	6.4	.17	.19	.17	60	.16
D176198	29.2	42	21	1.7	.73	.20	2.1	21	.74
D176199	16.2	12	6.1	25	.32	.22	.34	25	.29
D176200	30.1	13	6.0	5.0	.27	.14	.48	50	.73
D179838	11.6	17	8.6	9.2	.43	.18	.67	36	.37
D179389	49.8	52	23	.84	1.29	.39	2.4	8.0	1.0
D379840	11.8	32	15	5.9	.51	.27	1.1	23	.63
D179841	20.6	18	4.9	17	.45	.18	.39	26	.23
D179842	68.8	39	9.4	2.5	1.08	.53	1.9	26	.43
D179843	38.6	11	3.7	8.7	.13	.05	.25	58	.28

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	P205 (percent)	SO ₃ (percent)	Ag-S (ppm)	B-S (ppm)	Ba-S (ppm)	Be-S (ppm)	Cd (ppm)	Ce-S (ppm)	Co-S (ppm)	Cr-S (ppm
D176177	1.0L	2.8	N	700			1.0L	 N	100	70
D176178	1.0L	5.0	2	500	200	15	328	500L	70	100
D176179	1.0L	4.4	N	700	150	15	1.0L	N	30	70
D176180	1.0L	6.9	5	1,500	150	30	22.0	N	30	150
D176181	1.0L	4.6	1	700	150	15	4.0	N	50	70
D176182	1.0L	16	1	150	70	3L	1.0	· N	30	20
D176183	1.OL	11	5	300	100	7	68.0	500L	100	50
D176184	1.OL	3.3	1	500	150	10	1.0L	N	300	70
D176185	1.0L	7.6	N	200	20	7	1.0L	N	50	30
D176186	1.0L	2.8	N	150	50	7	5.0	500L	50	30
D176187	1.0L	8.0	N	500	150	15	6.0	N	50	70
D176188	1.0L	4.9	N	500	300	7	1.0	500L	30	150
D176189	1.0L	3.4	N	500	200	10	1.0L	N	30	70
D176190	1.0L	9.9	3	200	5,000	N	1.0L	N	30	70
0176191	1.OL	7.8	N	100	300	20	73.0	N	15	50
D176192	1.0L	6.3	2	700	200	15	437	N	50	150
D176193	1.0L	2.0	N	200	150	7	35.0	500L	100	150
D176194	1.0L	6.0	N	200	200	7	1.0	500L	300	100
D176195	1.0L	4.1	N	1,500	100	30	1.0L	N	100	70
D176196	1.0L	11	N	500	100	10	1.0L	N	20	70
D176197	1.0L	4.8	3	1.000	50	15	9,0	N	30	70
D176198	1.0L	1.5	N	500	700	7	1.0L	500L	30	150
D176199	1.0L	8.3	Ň	700	700	15	1.0L	500L	30	50
0176200	1.0L	10	N	300	10.000	7	188	N	70	70
0179838	1.0L	6.2	1	700	150	30	1.0L	N	70	150
0179389	1.0L	.74	N	300	700	10	1.01	200	50	300
0379840	1.0L	3.1	2	700	300	30	9.0	200	150	300
0179841	1.0L	14	N	300	100	15	26.0	Ň	15	30
0179842	1.0L	4.5	7	70	500	Ň	1.0	N	30	150
179843	1.0L	9.9	7	150	3,000	15	1.01	N	700	30

Sample number	Cu (ppm)	Ga-S (ppm)	Ge-S (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)
D176177		B	300	100L	70	335	7	20L	150L	200
D176178	520	8	100	100	242	220	20	20L	150L	200
D176179	172	B	100	N	92	1,250	20	20L	8	150
D176180	142	В	300	N	20	2,270	150	20L	8	150
01/0101	150	В	200	N	50	2,040	30	20	В	200
D176182	220	В	100	N	10	1,130	30	20L	В	100
D176183	240	В	150	150	30	785	70	20L	150	700
D176184	170	B	70	N	65	375	30	20L	В	300
D176185	118	B	70	N	11	635	N	20L	В	150
D176186	84	В	70	100	30	240	N	20L	200	150
D176187	72	В	150	150	54	925	15	20L	150	70
D176188	70	30	70	150	192	390	Ň	20L	150	70
D176189	170	8	100	100L	88	590	15	20L	N	150
D176190	62	10	20	N	22	1,710	30	20L	B	70
D176191	120	30	300	N	22	2,580	20	20L	B	150
D176192	128	20	70	N	78	960	100	201	В	300
D176193	120	20	N	150	212	280	20	20L	150	300
D176194	112	30	30	150	78	2,340	15	20L	150	700
D176195	100	· B	300	N	28	1,180	30	20L	В	200
D176196	64	30	70	N	46	2,310	10	20L	8	100
D176197	264	В	300	N	71	670	30	201	B	200
D176198	156	30	30	150	110	275	15	20	150	150
D176199	44	20	100	150	35	1,760	N	20L	150	70
D176200	82	В	200	N	55	395	15	20L	B	150
D179838	228	70	200	N	25	1,130	70	20	В	150
D179389	147	70	20	100	177	270	N	30	200	150
D379840	197	70	200	150	100	895	70	30	200	300
D179841	36	30	100	N	17	1,470	30	N	B	70
D179842	157	50	N	N	26	460	70	20	B	150
D179843	440	20	150	N	11	790	300	20	B	1,500

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Pb (ppm)	Sc-S (ppm)	Sr-S (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
	390		150	150	100	B	60	70
D176178	350	50	150	500	70	B	17,600	150
D176179	630	15	150	100	70	В	97	100
D176180	130	15	150	700	70	7	960	70
D176181	485	30	200	150	70	В	210	150
D176182	530	20	150	50	50	В	476	30
D176183	520	15	700	150	200	В	11,000	70
D176184	320	30	150	200	50	В	182	100
D176185	155	30	30	70	30	В	42	30
D176186	190	15	50	50	150	B	1,360	30
D176187	160	20	300	150	100	В	2,800	70
D176188	130	30	1,000	150	70	В	176	100
D176189	1,020	20	150	100	70	B	58	150
D176190	130	15	150	200	20	3	46	150
D176191	190	30	150	70	70	7	6,120	70
D176192	130	15	3,000	1,000	70	7	30,000	150
D176193	75	30	1,500	150	70	B	3,480	200
D176194	100	30	300	200	100	В	130	150
D176195	335	50	300	150	100	8	95	150
D176196	155	20	150	100	70	3	66	70
D176197	535	30	100	150	50	В	2,720	70
D176198	110	50	200	200	70	7	54	150
D176199	85	15	700	100	100	B	44	50
D176200	285	30	700	150	30	B	60,000	100
D179838	850	30	300	300	100	7	76	70
D179389	60	30	300	700	70	7	11	200
D379840	100	50	300	700	100	15	1,450	150
D179841	300	7	500	70	70	7	4,540	50
D179842	300	15	150	300	30	7	40	70
D179843	1,100	10	150	70	70	7	120	70

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued
Sample number	Ash (percent)	SiO ₂ (percent)	Al ₂ 03 (percent)	Ca() (percent)	MgO (percent)	Na ₂ () (percent)	K ₂ 0 (percent)	Fe ₂ 03 (percent)	TiO ₂ (percent)
D179844	32.9	21	8.7	7.9	0,46	0,11	0,68	33	0,38
D179845	38.7	2.1	2.2	3.6	.03	.03	.030L	79	.10
D179846	35.6	13	8.9	6.4	.18	.11	.35	58	.30
D179847	12.7	15	5.9	19	.60	.11	.77	22	.27
D179848	7.9	39	10	17	.55	.31	.87	4.9	.48
D179849	21.8	11	4.8	17	1.48	.14	.32	30	.23
D179850	13.0	40	21	1.1	.65	.42	1.3	21	.81
D179851	5.7	22	9.0	16	.83	.32	1.2	17	.36
D179852	10.6	27	7.6	18	.51	.32	.76	14	•36
D179853	32.8	9.3	4.3	22	1.58	.12	.27	23	.18
D179854	24.6	32	18	4.8	.53	.32	1.3	23	.63
D179855	13.4	14	13	11	.22	.26	.48	46	.37
D179856	32.0	14	9.8	2.6	.17	.12	.28	49	.48
D185601	13.9	7.7	3.7	13	.31	.10	1.1	60	.15
D185602	14.3	13	5.3	17	.45	.11	1.1	48	.31
D185603	12.3	15	1.6	17	.45	.18	.84	47	.12
D185604	11.8	20	5.0	23	.52	.20	1.0	32	.36
D185605	10.6	18	4.7	18	.55	.22	1.1	42	.29
D185606	13.1	34	11	11	1.37	.16	1.3	24	1.0
D185609	21.8	12	3.0	25	.39	.10	.93	35	.24
D185610	17.0	9.4	2.7	16	.24	.11	1.0	54	. 18 ·
D185611	23.0	15	2.8	43	.41	.10	.49	5.8	.27
D185612	12.1	28	12	2.2	.57	.13	1.6	42	.54
D185613	8.4	32	12	12	.73	.19	1.6	25	.70
D186062	11.6	15	5.0	19	.38	.35	.81	44	.16
D186063	14.9	10	4.1	24	.31	.26	.67	41	.12
D186064	8.9	21	8.1	13	.46	.44	.98	42	.26
D186065	16.6	17	8.1	14	.40	.14	1.1	46	.32
D186066	12.1	31	17	14	.35	.16	1.1	28	.63
D186067	24.7	25	12	6.4	.47	.20	1.6	47	.27

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

	D_0_							<u> </u>	<u> </u>	
number	(percent)	(percent)	Ag-S (ppm)	(bbw) R-2	(ppm)	(ppm)	(ppm)	(ppm)	(ppm)	(ppm)
D179844	1.0L	8.8	1.5	200	500	10	22.0	N	70	150
D179845	1.OL	5.3	1.5	100	70	7	23.0	N	70	30
D179846	1.0L	5.6	1.5	150	70	10	3.0	N	30	70
D179847	1.OL	13	15	700	150	30 1	20	N	70	150
D179848	1.0L	5.3	N	700	700	30	14.0	N	15	100
D179849	1.0L	10	1.5	200	150	15	1.0L	200	150	150
D179850	1.OL	1.2	N	500	300	30	4.0	N	70	300
D179851	1.4	11	10	1,000	200	30	35.0	N	20	500
D179852	1.0L	8.2	1.5	300	2,000	20	93.0	. N	30	150
D179853	1.OL	17	1.5	150	70	7	5.0	N	20	50
D179854	1.0L	2.9	2	300	300	20	1.0L	300	50	200
D179855	1.0L	8.5	N	700	150	30	1.0L	N	30	150
D179856	1.0L	3.4	3	150	3,000	15	4.0	N	150	300
D185601	1.0L	11	N	300	70	15	1.0L	N	50	20
D185602	1.0L	14	N	300	70	15	1.0L	N	50	30
D185603	1.0L	9.2	N	1,000	50	15	1.0	N	50	15
D185604	1.0L	8.3	N	1,500	100	15	1.0	500L	50	30
D185605	1.0L	10	N	1,500	100	15	1.0L	500L	50	30
D185606	1.0L	15	N	700	2,000	15	4.0	700	50	100
D185609	1.0L	11	N	300	70	5	2.0	N	30	20
D185610	1.0L	8.9	N	500	70	7	1.0L	N	70	20
D185611	1.0L	11	N	300	50	5	1.0L	N	10	20
D185612	1.0L	5.4	10	1.000	150	15 5	10	Ň	50	50
D185613	1.0L	7.1	5	500	300	20 3	70	N	30	70
D186062	1.0L	10	1.5	700	70	15	1.0L	N	30	30
D186063	1.0L	12	1.5	700	100	10	1.0	N	30	30
D186064	1.0L	7.7	1.5	1,500	100	20	12.0	N	50	70
D186065	1.0L	12	Ň	500	700	- 7	1.0L	N	15	70
D186066	1.0L	6.4	Ň	700	150	15	1.5	500L	30	100
D186067	1.0L	5.3	1.5	300	150	7	1.0L	N	100	70

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Cu (ppm)	Ga-S (ppm)	Ge-S (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)
D179844	145	70	100	N	44	780	70	20	8	200
D179845	179	30	100	N	10L	230	30	N	В	300
D179846	157	70	100	N	44	380	100	20	B	200
D17984/	146	70	300	N	12	2,780	1,000	20	B	300
D179848	100	/ 0	500	N	63	2,060	50	20	В	/0
D179849	192	30	150	150	33	2,780	30	20	N	300
D179850	112	70	100	N	250	160	30	20	B	200
D179851	149	50	300	N	17	1,710	300	20	В	200
D179852	149	50	300	N	25	2,060	30	20	B	150
D179853	109	50	100	N	17	2,060	50	N	В	150
D179854	137	100	70	200	222	530	50	20	300	300
D179855	69	70	100	150	45	820	15	20	N	200
D179856	171	50	70	N	72	560	15	20	В	500
D185601	29	B	70	N	10	1,160	7	20L	В	50
D185602	34	30	100	N	16	1,240	7	20L	В	50
D185603	99	В	150	N	10	1.840	7	20L	B	200
D185604	77	30	150	100	16	1,660	10	20L	150	150
D185605	96	30	150	100	18	1,880	7	20L	150	150
D185606	125	30	150	200	109	1,600	10	20	300	100
D185609	67	15	30	N	14	1,920	N	20L	B	100
D185610	109	В	50	N	11	1.330	N	201	В	200
D185611	22	15	70	Ň	13	4,260	15	N	B	30
D185612	157	В	150	Ň	97	365	70	20	B	500
D185613	98	20	150	N	68	1,260	70	20	B	300
D186062	157	В	70	N	13	1,490	15	20L	В	300
D186063	150	В	70	N	101	1.800	15	N	B	300
D186064	175	B	150	N	25	970	15	20	B	300
D186065	153	B	70	100	29	915	15	201	300	100
D186066	88	30	150	150	94	610	15	20	150	70
0186067	153	R	70	1001	27	270		201	200	200

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Pb (ppm)	Sc-S (ppm)	Sr-S (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
D179844	400	30	150	300	70	7	7,400	150
D179845	800	15	150	150	30	10	7,350	N
D179846	700	15	150	70	50	7	1,370	70
D179847	500	20	500	1,500	70	10	9,300	70
D179848	75	30	300	150	150	15	1,000	70
D179849	500	30	300	200	100	7	320	70
D179850	170	30	300	300	100	10	1,030	150
D179851	190	15	500	3,000	100	15	2,200	70
D179852	525	30	300	200	100	15	6,250	70
D179853	260	30	300	100	70	7	1,000	50
D179854	190	50	1,500	300	70	10	370	100
D179855	150	30	2,000	200	70	7	970	70
D179856	225	30	150	300	70	10	2,560	150
D185601	360	10L	70	30	50	B	76	30
D185602	400	10L	70	50	50	В	82	50
D185603	200	15	300	30	50	В	83	30
D185604	105	15	700	70	70	В	80	70
D185605	160	15	700	50	70	В	87	70
D185606	230	30	1,000	200	150	B	1,050	150
D185609	170	10L	150	15	50	В	847	30
D185610	245	10L	150	20	50	ß	54	30
D185611	45	10L	200	30	70	7	76	30
D185612	515	15	150	70	70	B	31,600	100
D185613	130	20	150	150	70	В	13,000	100
D186062	650	15	150	70	70	В	501	50
D186063	795	15	150	70	70	8	402	30
D186064	665	20	150	150	70	8	3,120	70
D186065	80	30	150	100	70	В	· 77	70
D186066	75	30	150	150	150	В	1,010	70
D186067	595	15	100	150	70	В	60	70

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Ash (percent)	SiO ₂ (percent)	Al ₂ 0 ₃ (percent)	CaO (percent)	MgO (percent)	Na ₂ 0 (percent)	K ₂ 0 (percent)	Fe ₂ 03 (percent)	TiO2 (percent)
D186068	13.8	14	7.3	27	0,24	0,20	0.62	31	0.22
D186069	26.0	46	23	3.1	.85	.19	2.1	16	.88
D186070	11.1	29	18	10	.47	.23	1.4	28	.63
D186071	14.5	11	7.5	14	.25	.17		51	28
D186072	22.2	15	8.1	32	.44	.19	.82	20	.45
D186073	15.2	22	9.4	18	.44	.18	1.2	32	.54
D186074	19.9	18	6.4	18	.50	.13	1.0	37	.40
D186075	13.2	23	7.5	25	.58	.40	.90	22	.43
D186076	15.1	18	4.7	21	_60	.31	.86	33	.29
D192368	24.2	19	4.5	21	.30	.13	.21	22	.26
D192369	11.4	31	9.6	12	.55	.38	.71	22	.49
D192370	25.9	5.1	1.8	5.9	.09	.13	.030L	59	.080
D192371	20.6	14	5.6	8.6	.31	.26	.29	46	.24
D192372	13.5	24	12	9.5	.36	.31	.47	31	.46
D192373	14.1	29	9.4	11	.51	.35	.77	27	.40
D192374	13.4	15	5.8	8.4	-25	.26	.24	46	.30
D192375	22.9	28	12	3.4	.32	.21	.76	40	.55
D192376	16.4	15	4.9	13	.22	.22	.23	35	.50
D192377	19.9	26	10	3.2	.35	.16	.51	41	44
D192378	18.0	13	8.0	13	.24	.21	.27	38	.27
D192379	20.9	31	12	15	.57	.15	1.1	16	.55

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	P ₂ 05 (percent)	SO3 (percent)	Ag-S (ppm)	B-S (ppm)	Ba-S (ppm)	Be-S (ppm)	Cd (ppm)	Ce-S (ppm)	Co-S (ppm)	Cr-S (ppm)
D186068	1.0L	14	 N	700	200		1.0L	500		70
D186069	1.0L	2.3	Ň	300	150	15	5.0	500L	50	150
D186070	1.3	5.7	N	1.000	500	20	1.0L	500	50	150
D186071	1.0L	9.1	1.5	700	150	15	74.0	500L	70	70
D186072	1.0L	16	N	500	150	10	2.0	N	15	70
D186073	1.0L	12	N	500	100	15	23.0	N	70	150
D186074	1.0L	13	N	500	100	10	62.0	N	70	70
D186075	1.0L	9.9	N	1,000	150	15	1.5	N	50	70
D186076	1.0L	10	N	1,000	150	15	24.0	N	30	50
D192368	1.0L	19	N	300	1,000	7	9.0	N	15	30
D192369	1.0L	9.7	N	1,500	200	15	2.5	N	100	70
D192370	1.0L	13	Ň	300	20	10	1.0L	N	15	30
D192371	1.0L	13	Ň	700	70	15	82.0	N	200	70
D192372	1.0L	7.7	N	700	300	15	5.0	N	50	100
D192373	1.0L	5.1	N	1,500	100	15	1.0L	N	30	70
D192374	1.0L	6.1	N	1,000	70	15	3.0	N	100	70
D192375	1.0L	7.6	Ň	300	150	15	31.0	N	50	50
D192376	1.0L	16	Ň	500	5,000	20	115	N	30	50
D192377	1.0L	6.5	Ň	500	150	15	2.0	Ň	70	70
D192378	1.0L	12	N	700	150	15	17.0	N	20	50
0192379	1.0L	8.3	N	700	200	15	3.0	N	30	70

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Cu (ppm)	Ga-S (ppm)	Ge~S (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)
D186068	55	8	100	150	40	1.570	N	201	150	150
D186069	112	30	70	150	225	240	15	20	N	150
D186070	75	50	150	150	162	430	15	20	300	100
D186071	131	30	150	150	31	575	15	20	150	300
D186072	77	20	100	100L	79	2,740	15	20L	N	50
D186073	77	50	100	N	85	785	50	201	B	150
D186074	112	30	100	100L	35	1.070	15	20	1501	300
D186075	47	30	150	100L	31	1.890	7	20	150	300
D186076	59	30	150	100L	20	3,520	7	201	150	300
D192368	37	20	100	N	17	1,550	7	N	B	150
D192369	103	50	70	N	43	1.180	N	N	В	300
D192370	60	30	50	N	20	600	N	Ň	B	100
D192371	125	70	150	100L	35	1.370	7	Ň	Ň	500
D192372	112	70	100	N	134	1.010	15	Ň	B	150
D192373	89	30	300	N ⁻	21	1,240	30	N	B	100
D192374	129	70	150	100	33	670	N	N	N	300
D192375	57	30	100	N	88	400	Ň	201	B	200
D192376	82	50	150	100	45	750	N	201	150	200
D192377	232	50	150	N	99	800	30	20L	B	300
D192378	73	30	100	100L	69	1,370	15	N	Ň	100
D192379	66	30	70	N	69	1,950	N	N	В	150

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Sample number	Pb (ppm)	Sc-S (ppm)	Sr-S (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
D186068	220	30	1.500	150	150	ß	196	70
0186069	300	30	150	200	70	Ř	1.880	150
D186070	110	30	2,000	300	150	B	488	150
D186071	260	30	1 000	100	70	B	14,000	70
D186072	155	20	700	150	70	B	462	70
D186073	150	30	150	200	70	В	4,820	100
D186074	405	30	150	150	70	В	18,700	70
D186075	170	20	700	100	70	В	1,240	70
D186076	230	15	700	70	70	В	6,790	70
D192368	150	10	150	50	30	В	700	30
D192369	270	30	500	150	70	8	1,500	70
D192370	1,800	20	70	70	30	B	37	30
D192371	440	30	150	150	100	15	12,400	70
D192372	95	30	150	150	70	8	2,250	150
D192373	365	15	150	100	30	8	68	70
D192374	615	15	300	100	30	8	517	50
D192375	750	15	300	70	30	В	3,000	70
D192376	490	20	1.000	70	50	B	25,700	70
D192377	1.000	20	200	150	50	В	58	70
D192378	620	15	1,000	100	30	B	3,000	50
D192379	135	20	300	100	70	В	200	70

Table 4. Major- and minor-oxide and trace element composition of the laboratory ash of 106 Iowa coal samples--continued

Tabi	le	5.	E	lement	COM	oosi	tion	of	106	Iowa	coal	sampl	es

{As, F, Hg, Sb, Se, Th, and U values are from direct determinations on air-dried (32°C) coal; all other values calculated from analyses of coal ash. S means analysis by emission spectrography; L, less than the value shown; N, not detected; B, not determined}

Sample number	Si (percent)	Al (percent)	Ca (percent)	Mg (percent)	Na (percent)	K (percent)	Fe (percent)	Ti (percent)	Ag-S (ppm)	As (ppm)
D166027	1.8	1.0	2.6	0.054	0.012	0.12	5.7	0.028	 N	15
0166028	3.3	1.9	2.7	.11	.023	.29	3.9	.076	N	10
D166029	1.5	.91	1.9	.048	.009	.10	3.0	.046	N	5.0
D166030	2.0	1.0	.49	.042	.017	.12	3.9	.045	N	12
D166031	2.0	1.2	.12	.067	.018	.14	5.9	.036	N	20
D166032	2.0	.80	2.8	.058	.017	.097	4.5	.025	N	5.0
D166033	2.0	.88	1.7	.048	.020	.13	4.0	.054	Ň	8.0
D166034	1.5	.63	1.3	.036	.015	.064	7.0	.020	Ň	10
D166035	1.3	.64	4.2	.048	.014	.063	3.2	.025	N	30
D166036	.98	.50	1.7	.031	.010	.046	2.7	.023	N	30
0166037	1.9	.72	1.9	.039	.032	.082	2.2	.017	N	20
D166038	2.1	.79	1.7	.042	.034	.093	2.6	.027	N	30
D166039	3.8	1.7	1.2	.097	.035	.27	1.9	.055	N	20
D166040	6.6	3.2	1.1	.39	.065	.61	2.7	.18	N	25
D166041	1.5	.95	.087	.023	.014	.049	5.6	.031	N	20
D166042	2.3	.98	1.1	.032	.013	.097	3.4	.049	N	20
D166043	1.2	.64	.19	.017	.010	.037	3.4	.034	.15	15
D176169	.94	.35	3.5	.050	.017	.077	2.1	.024	.7	5.0
D176170	2.4	.94	.59	.056	.041	.16	5.4	.071	N	20
D176171	6.7	2.5	1.3	.067	.052	.29	3.1	.16	1	30
D176172	1.4	1.3	1.3	.026	.028	.045	9.5	.029	.7	160
D176173	3.1	1.2	3.8	.23	.041	.17	4.3	.074	N	10
D176174	4.3	2.5	.66	.062	.048	.20	2.0	.15	N	3.0
D176175	2.4	1.5	1.8	.061	.038	.16	1.9	.052	N	3.0
-D176176	1.2	.76	1.1	.029	.023	.063	8.6	.040	N	25

Sample number	B-S (ppm)	Ba-S (ppm)	Be-S (ppm)	Cd (ppm)	Ce-S (ppm)	Co-S (ppm)	Cr-S (ppm)	Cu (ppm)	F (ppm)	Ga-S (ppm)	Ge-S (ppm)
D166027	70	700	1.5	0.47	N	7	15	15		В	30
D166028	70	500	2	1.5	N	7	20	15	145	• B	20
D166029	100	500	2	2.0	N	5	10	13	75	B	30
D166030	150	20	2	.60	N	5	10	24	85	В	15
D166031	150	20	1.5	.20L	N	15	15	41	150	B	10
D166032	150	. 20	1.5	.21L	N	3	20	20	90	В	15
D166033	150	30	1.5	1.4	Ň	3	15	18	115	B	20
D166034	100	15	1.5	.22L	N	3	7	22	60	B	7
D166035	150	15	1.5	.21L	N	3	10	12	80	B	20
D166036	200	15	2	.13L	N	2	7	11	60	В	20
D166037	200	70	1	.28	N	5	7	8.5	100	В	10
D166038	200	50	1.5	.31	N	5	7	9.2	80	B	10
D166039	200	30	3	.18L	100L	5	15	17	115	3	15
D166040	200	100	2	.29L	150L	10	30	23	185	В	20
D166041	100	20	2	15	Ň	10	10	25	40	B	15
D166042	100	20	2	5.3	N	5	10	17	40	В	20
D166043	100	5	2	3.1	N	7	7	16	30	B	15
D176169	100	15	1.5	.16L	N	3	15	13	40	2	20
D176170	100	30	3	.19L	N	5	15	21	45	B	30
D176171	70	70	5	49	N	50	50	48	85	5	10
D176172	70	50	2	15	150L	50	7	46	40	B	50
D176173	100	50	3	.28	N	10	20	20	70	В	20
D176174	100	100	3	13	100L	7	30	24	65	10	7
D176175	150	150	2	3.0	N	5	15	20	80	5	15
D176176	70	70	2	.22L	100L	15	10	16	25	8	15

Table 5. <u>Element composition of 106 Iowa coal samples</u>--continued

Sample number	Hg (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)	P (ppm)	РЬ (ррт)
D166027	0.12	N	9.5	170	3	 5L	B	30	130	95
D166028	.09	20L	23	250	3	5L	N	30	470	28
D166029	.09	Ň	6.7	150	5	3L	B	20	120	39
D166030	.07	15L	11	110	2	3L	N	20	86	11
D166031	.07	20L	15	600	3	5L	30	50	140	15
D166032	.08	N	4.6	210	3	5L	В	15	160	15
D166033	.08	N	8.2	130	3	3L	B	15	55	8.1
D166034	.10	N	3.7	110	3	5L	В	10	57	20
D166035	.18	N	5.5	420	1.5	5L	8	20	470	76
D166036	.12	N	4.1	190	1	2L	B	20	400	58
D166037	.11	N	6.8	140	1	3L	B	20	610	43
D166038	.16	N	8.7	110	1	3L	B	20	580	53
D166039	.08	20L	17	130	1.5	3L	30L	30	450	27
D166040	.09	30L	33	210	3	7L	50L	70	380	35
D166041	.22	N	16	52	5	3L	В	20	45	86
D166042	.14	N	22	110	10	3L	В	30	78	44
D166043	.19	10L	8.4	34	3	2L	15L	7	49	71
D176169	.14	N	1.6	380	15	3L	8	15	710L	18
D176170	.14	N	6.5	110	1.5	3L	B	30	810L	100
D176171	.27	N	42	140	50	7L	В	100	1,300L	59
D176172	.17	20	14	110	7	5L	50	200	1.100L	56
D176173	.09	30L	11	430	15	5L	N	50	1,200L	25
D176174	.17	20	36	51	3	7	30	20	880L	15
D176175	.11	15	16	120	3	3L	20	20	730L	20
D176176	.11	20L	8.9	78	N	5L	30L	50	970L	67

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Sb (ppm)	Sc-S (ppm)	Se (ppm)	Sr-S (ppm)	Th (ppm)	U (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
D166027	0.5	3	1.1	30	3.01	1.2				350	15
D166028	.6	7	1.3	20	6.2	1.2	30	15	B	760	20
D166029	.4	5	.9	15	3.01	2.1	20	-5	B	1,200	10
D166030	.4	5	1.7	20	3.01	2.8	20	Š	B	230	10
D166031	.8	5	2.0	30	3.0L	4.2	30	15	B	220	15
D166032	.4	3	2.6	15	3.0L	2.7	20	7	B	61	15
D166033	.4	3	1.3	15	3.9	1.4	20	5	B	430	15
D166034	.6	2	1.7	10	3.01	1.4	15	7	B	11	10
D166035	.5	3	1.4	30	3.0L	.5	15	7	B	39	15
D166036	.4	2	1.0	20	3.0L	.7	10	5	B	47	10
D166037	.6	2	.5	30	3.0L	.6	10	5	В	320	7
D166038	•2	2	1.2	30	3.0L	.6	10	5	.5	270	10
D166039	1.8	3	1.0	30	3.0L	1.4	30	5	.5	140	20
D166040	2.2	7	1.3	50	4.2	1.7	50	10	В	76	50
D166041	1.0	3	3.0	5	3.0L	1.6	15	10	B	5,500	10
D166042	.9	5	2.6	10	3.0L	1.8	20	7	В	1.800	20
D166043	.8	1.5	3.1	15	3.8	1.6	15	7	B	870	7
D176169	15.7	2	11	15	В	30	50	7	.7	10	. 7
D176170	1.0	5	3.2	30	3.0L	2.2	15	10	B	12	20
D176171	7.6	10	29	100	В	43	100	10	2	4,200	70
D176172	1.9	7	3.3	500	9.7	4.9	50	7	В	2.300	15
D176173	.7	10	3.4	30	18.0	9.3	30	20	В	56	30
D176174	.2	10	6.5	30	11.0	2.4	30	15	1	3,500	50
D176175	.3	5	4.7	50	8.7	4.6	20	15	1	1,000	10
D176176	.5	5	3.7	70	3.0L	1.6	15 .	7	B	13	15

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Si (percent)	Al (percent)	Ca (percent)	Mg (percent)	Na (percent)	K (percent)	Fe (percent)	Ti (percent)	Ag-S (ppm)	As (ррт)
	1.2	84	68	0,030	0,020	0,086	6.0	0.048	N	12
0176178	4.4	2.7	.28	.14	.090	.41	7.4	.13	.7	50
D176179	2.1	.82	1.1	.064	.061	.15	5.0	.064	N	30
D176180	1.1	.41	1.5	.047	.038	.097	1.5	.026	.5	5.0
D176181	2.4	.88	1.9	.076	.054	.16	2.8	.052	.2	20
D176182	1.1	.41	3.9	.058	.028	.049	15	.042	.5	90
D176183	1.6	.89	2.3	.048	.041	.11	9.4	.041	1.5	60
D176184	1.9	1.2	•63	.043	.043	.12	7.9	.061	.2	60
D176185	.52	.24	1.9	.025	.026	.019	14	.027	N	40
D176186	1.3	.77	.93	.021L	.023L	.031	16	.031	N	25
D176187	.76	.49	1.1	.024	.017	.054	3.7	.028	N	8.0
D176188	2.9	2.2	1.3	.084	.034	.23	5.7	.095	N	4.0
D176189	3.1	1.4	.53	.11	.065	.24	7.8	.085	N	240
D176190	9.0	1.6	4.6	.24	.21	.47	4.6	.18	1.5	40
D176191	1.5	.48	1.7	.032	.034	. 064	1.9	.025	N	15
D176192	2.8	1.0	.91	.054	.040	.16	2.7	.053	.3	8.0
D176193	7.6	3.6	.32	.084	.078	.27	7.4	.21	N	30
D176194	4.1	2.0	1.0	.070	.061	.19	8.0	.094	N	12
D176195	.66	.29	.53	.016	.032	.032	3.9	.023	N	20
D176196	2.2	.92	4.1	.080.	.046	.11	2.6	.051	N	5.0
D176197	.49	.42	.75	.017	.023	.023	6.8	.016	.5	30
D176198	5.8	3.2	.35	.13	.043	.50	4.4	.13	Ň	12
D176199	.90	.52	2.9	.031	.026	.046	2.9	.028	N	5.0
D176200	1.8	.96	1.1	.049	.031	.12	10	.13	N	50
D179838	.91	.53	.76	.030	.015	.065	3.0	.026	.1	18
D179839	12	5.9	.30	.39	.14	1.0	2.8	.31	N	33
D179840	1.8	.97	.50	.036	.024	.11	1.9	.045	.2	22
D179841	1.7	.53	2.4	.056	.027	.067	3.8	.028	Ň	5.5
D179842	12	3.4	1.2	.45	.27	1.1	13	.18	5	250
D179843	2.0	.76	2.4	.030	.014	.080	16	.065	3	230

Table 5. Element composition of 106 Iowa coal samples--continued

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Sample number	B-S (ppm)	Ba-S (ppm)	Be-S (ppm)	Cd (ppm)	Ce-S (ppm)	Co-S (ppm)	Cr-S (ppm)	Cu (ppm)	F (ppm)	Ga-S (ppm)	Ge-S (ppm)
D176177	100	100	2	0.17L	N	15	10	13	70	B	50
D176178	150	70	5	100	150L	20	30	160	100	B	30
D176179	150	30	3	.18L	N	5	15	32	50	B	20
D176180	150	15	3	2.3	N	3	15	15	50	B	30
D176181	150	30	3	.72	N	10	15	27	110	В	30
D176182	70	30	1.5L	.42	N	15	10	92	45	В	50
D176183	100	30	2	20	150L	30	15	71	45	B	50
D176184	100	30	2	.23L	N	70	15	38	65	8	15
D176185	70	7	2	.32L	N	15	10	38	20	B	20
D176186	50	15	2	1.7	150L	15	10	29	25	В	20
D176187	70	20	2	.76	N	7	10	9.1	30	В	20
D176188	150	70	1.5	.25	150L	7	30	18	140	7	15
D176189	150	50	2	.25L	N	7	15	42	50	В	20
D176190	100	2,000	N	.46L	N	15	30	28	140	5	10
D176191	15	50	2	9.3	N	2	7	15	30	5	50
D176192	100	30	2	74	N	10	20	22	120	3	10
D176193	70	70	3	13	200L	30	70	45	155	7	N
D176194	70	70	2	.32	150L	100	30	35	100	10	10
D176195	150	10	3	.11L	N	10	7	11	30	В	30
D176196	100	20	2	.24L	N	5	15	15	55	7	15
0176197	150	7	2	1.5	N	5	10	43	40	В	50
D176198	150	200	Ž	.29L	150L	10	50	46	160	10	10
D176199	100	100	2	.16L	70L	5	7	7.1	110	3	15
0176200	100	3.000	2	57	N	20	20	25	80	В	70
D179838	70	15	3	.12L	N	7	15	26	40	7	20
D179839	150	300	5	.50L	100	20	150	73	350	30	10
D179840	100	30	3	1.1	20	15	30	23	40	10	20
D179841	70	20	3	5.4	N	3	7	7.4	50	7	20
0179842	50	300	• N	.69L	N	20	100	110	370	30	Ň
D179843	70	1.000	7	.39L	N	300	10	170	20	7	70
• • • • •	• •			4 4 7 L	••	000	A-17			•	

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Hg (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)	р (ррт)	Pb (ppm)
D176177	0.07	15L	12	56	1	3L	20L	30	7301	65
0176178	.25	30	74	67	7	7L	50L	70	1.300L	110
D176179	.23	N	17	230	3	3L	В	30	800L	120
D176180	.09	N	2.1	230	15	2L	B	15	450L	13
D176181	.15	N	9.1	370	5	3	B	30	790L	88
D176182	.37	N	4.2	470	15	10L	B	50	1.800L	220
D176183	.44	50	8.8	230	20	7L	50	200	1.300L	150
D176184	.20	N	15	84	7	5L	8	70	980L	72
D176185	.12	N	3.5	200	N	7L	В	50	1,400L	50
D176186	.08	30	10	82	N	7L	70	50	1,500L	65
D176187	.08	20	6.8	120	2	2L	20	10	500L	20
D176188	.10	30	49	99	N	5L	30	15	1.100L	33
D176189	.17	20L	22	150	3	5L	N	30	1.100L	260
D176190	.20	N	10	780	15	10L	В	30	2,000L	60
0176191	.08	N	2.8	330	2	21_	В	20	560L	24
D176192	.14	N	13	160	15	3L	B	50	7401	22
D176193	.20	70	80	110	7	7L	70	100	1.700L	28
D176194	.12	50	25	740	5	7L	50	200	1.400L	32
D176195	.08	N	3.0	130	3	2L	B	20	470L	36
D176196	.08	N	11	550	2	5L	8	20	1,000L	37
D176197	.34	N	12	110	5	3L	8	30	7201	88
D176198	.10	50	32	80	5	7	50	50	1.300L	32
D176199	.09	20	5.7	290	N	3L	20	10	710L	14
D176200	.21	N	17	120	5	7L	В	50	1.300L	86
D179838	.25	N	2.9	130	7	2	B	15	510L	99
D179839	.14	50	88	130	N	15	100	70	2,2001	30
D179840	.07	15	12	110	10	3	20	30	520L	12
D179841	.13	N	3.5	300	7	Ň	B	15	900L	62
D179842	.41	· N	18	320	50	15	B	100	3,000	210
D179843	.49	N	4.2	300	100	7	Ē	700	1.700L	420

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Sb (ppm)	Sc-S (ppm)	Se (ppm)	Sr-S (ppm)	Th (ppm)	U (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
D176177	0.5	10	5.4	20	9.3	1.8	20	15	 B	10	10
D176178	10.0	15	7.1	50	22.0	9.3	150	20	В	5,400	50
D176179	.9	3	3.6	30	3.0L	1.1	20	15	B	18	20
D176180	1.5	1.5	12	15	3.OL	12	70	7	.7	99	7
D176181	2.5	5	3.6	30	6.6	1.7	30	15	В	38	30
D176182	4.0	10	2.4	70	6.8	3.5	20	20	В	200	15
D176183	6.4	5	18	200	3.0L	18	50	70	В	3,200	20
D176184	1.3	7	7.5	30	14.0	4.9	50	10	B	41	20
D176185	.5	10	2.5	10	3.OL	3.7	20	10	В	13	10
D176186	.4	5	3.5	15	3.0L	6.8	15	50	В	470	10
D176187	.3	2	6.4	30	3.0L	1.3	20	15	В	350	10
D176188	.3	7	3.2	200	8.1	2.0	30	15	В	45	20
D176189	10.5	5	4.2	30	3.0L	5.3	20	15	В	14	30
D176190	16.0	7	21	70	В	19	100	10	1.5	21	70
D176191	1.3	5	2.1	20	3.0L	1.6	10	10	1	780	10
D176192	3.9	2	75	500	В	35	150	10	1	5,200	20
D176193	1.7	10	17	700	3.0L	8.4	70	30	B	1,300	70
D176194	1.3	10	4.9	100	10.0	3.3	70	30	В	41	50
D176195	.3	5	1.8	30	3.0L	2.2	15	10	В	10	15
D176196	.3	5	2.3	30	3.OL	3.6	20	15	.7	16	15
0176197	2.6	5	11	15	3.0L	8.3	20	7	8	450	10
D176198	.5	15	8.2	70	17.0	5.3	70	20	2	16	50
D176199	.3	2	2.3	100	3.0L	.9	15	15	B	7.1	7
D176200	.8	10	2.6	200	3.0L	2.7	50	10	B	18,000	30
D179838	.6	3	4.0	30	4.4	2.6	30	10	.7	8.8	7
D179839	1.2	15	3.5	150	11.0	3.8	300	30	3	55	100
D179840	2.1	7	3.0	30	3.0L	.7	100	10	1.5	170	15
D179841	.3	1.5	2.2	100	3.0L	.5	15	15	1.5	940	10
D179842	39.0	10	53	100	4.3	6.0	200	20	5	28	50
D179843	22.0	5	4.0	70	1.0	41	30	30	3	46	30

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Si (percent)	Al (percent)	Ca (percent)	Mg (percent)	Na (percent)	K (percent)	Fe (percent)	Ti (percent)	Ag-S (ppm)	As (ppm)
D179844	3.2	1.5	1.8	0.091	0,027	0.19	7.6	0.075	0.5	28
D179845	.38	.45	.99	.007	.009	.010L	21	.023	.7	49
D179846	2.2	1.7	1.6	.039	.029	.10	14	.064	.5	13
D179847	.86	.40	1.8	.046	.010	.081	1.9	.021	2	28
D179848	1.4	.43	.98	.026	.018	.057	.27	.023	N	4.5
D179849	1.1	.56	2.6	.19	.023	.058	4.6	.030	.3	24
D179850	2.4	1.4	.10	.051	.040	.14	1.9	.063	Ň	11
D179851	.59	.27	.64	.028	.014	.058	.68	.012	.7	10
0179852	1.3	.42	1.4	.033	.025	.067	1.1	.023	.15	16
D179853	1.4	.75	5.1	.31	.029	.074	5.2	.035	•5	26
D179854	3.6	2.4	-84	.078	.058	.26	3.9	.093	.5	7.5
D179855	.88	.92	1.1	.018	.026	.054	4.3	.030	Ň	13
D179856	2.1	1.7	.60	.033	.028	.075	11	.092	1	16
D185601	.50	.27	1.3	.026	.010	.13	5.8	.012	Ň	4.9
D185602	.87	.40	1.7	.039	.012	.13	4.8	.027	N	6.4
D185603	.86	.10	1.5	.033	.016	.086	4.0	.009	N	11
D185604	1.1	.31	1.9	.037	.017	.098	2.6	.025	Ň	6.9
D185605	.89	.26	1.4	.035	.017	.097	3.1	.018	N	9.8
D185606	2.1	.76	1.0	.11	.016	.14	2.2	.078	N	6.6
D185609	1.2	.35	3.9	.051	.016	.17	5.3	.031	N	19
D185610	-75	.24	1.9	.025	.014	.14	6.4	-018	N	20
D185611	1.6	.34	7.1	.057	.017	.094	.93	.037	Ň	4.3
D185612	1.6	.77	.19	.042	.012	.16	3.6	.039	1	13
D185613	1.3	.53	.72	.037	.012	.11	1.5	.035	5	3.7
D186062	.81	.31	1.6	.027	.030	.078	3.6	.011	.15	47
D185063	.70	.32	2.6	.028	.029	-083	4.3	.011	.2	62
D185064	.87	.38	.83	.025	.029	.073	2.6	.014	.15	43
D185065	1.3	.71	1.7	.040	.017	.15	5.3	.032	Ň	3.9
D185066	1.8	1.1	1.2	.025	.014	.11	2.4	.046	Ň	2.0
D185067	2.9	1.6	1.1	.070	.037	.33	8.1	.040	.3	75

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	B-S (ppm)	Ba-S (ppm)	Be~S (ppm)	Cd (ppm)	Ce-S (ppm)	Co-S (ppm)	Cr-S (ppm)	Cu (ppm)	F (ppm)	Ga-S (ppm)	Ge-S (ppm)
D179844	70	150	3	7.2	N	20	50	48	105	20	30
D179845	50	30	3	8.9	N	30	10	69	20L	10	50
D179846	50	20	3	1.1	N	10	20	56	30	20	30
D179847	100	20	3	15	N	10	20	19	30	10	30
D179848	50	50	2	1.1	N	1	7	7.9	40	5	50
D179849	50	30	3	.22L	50	30	30	42	45	7	30
D179850	70	50	5	.52	N	10	50	15	175	10	15
D179851	70	10	1.5	2.0	N	1	30	8.5	135	3	15
D179852	30	200	2	9.9	N	3	15	16	50	5	30
D179853	50	20	2	1.6	N	7	15	36	85	15	30
D179854	70	70	5	. 251	70	15	50	34	100	20	15
D179855	100	20	5	.13L	Ň	5	20	9.2	60	10	15
D179856	50	1.000	5	1.3	N	50	100	55	30	15	20
D185601	50	10	2	_14L	Ň	7	3	4.0	90	B	10
D185602	50	10	2	.14L	Ň	7	5	4.9	100	5	15
D185603	150	7	2	.12	N	7	2	12	85	8	20
D185604	150	10	1.5	.12	70L	7	3	9.1	75	3	15
D185605	150	10	1.5	.11L	50L	5	3	10	75	3	15
D185606	100	300	2	.52	100	7	15	16	95	5	20
D185609	70	15	ī	.44	N	7	5	15	75	3	7
D185610	100	10	1	. 171	N	10	3	45	45	8	10
D185611	70	10	ī	.231	Ň	2	5	70	70	3	15
0185612	100	20	2	62	N	7	7	250	250	8	20
D185613	50	20	1.5	31	Ň	2	7	70	70	1.5	15
D186062	70	7	1.5	.12L	N	3	3	70	70	B	7
D185063	100	15	1.5	.15	N	5	5	22	75	В	10
D185064	150	10	2	1.1	Ň	5	7	16	80	Ē	15
D185065	100	100	ī	17	N	2	10	25	65	B	10
D185066	100	20	2	.18	70L	3	10	11	60	3	20
D185067	70	30	ī.5	.25L	Ň	20	15	38	65	B	15

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Hg (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)	р (ррт)	Pb (ppm)
D179844	0.25	N	14	260	20	7	B	70	1.400L	130
D179845	.42	N	3.9L	89	10	N	В	100	1,700L	310
D179846	.46	N	16	140	30	7	В	70	1,600L	250
D179847	.22	N	1.5	350	150	2	ß	30	550L	64
D179848	.07	N	5.0	160	5	1.5	В	5	350L	5.9
D179849	.40	30	7.2	610	7	5	N	70	950L	110
D179850	.15	N	33	21	5	2	В	20	570L	22
D179851	.30	N	1.0	98	15	1	B	10	340	11
D179852	.12	N	2.7	220	3	2	В	15	460L	56
D179853	.24	N	5.6	680	15	N	8	50	1,400L	85
D179854	.24	50	55	130	15	5	70	70	1,100	47
D179855	.22	20	6.0	110	2	3	N	30	5901	20
D179856	.41	N	23	180	5	7	B	150	1.4001	72
D185601	.13	N	1.4	160	ī	3L	B	7	610L	50
D185602	.14	N	2.3	180	1	3L	В	7	620L	57
D185603	.12	~ N	1.2	230	1	2L	B	20	5401	25
D185604	.08	10	1.9	200	ī	2L	15	15	5201	12
D185605	.08	10	1.9	200	7	2L	15	15	4601	17
D185606	.16	30	14	210	1.5	3	50	15	570L	30
D185609	.09	N	3.1	420	N	5L	B	20	950L	37
D185610	.12	N	1.9	230	N	31.	В	30	740)	42
D185611	.07	N	3.0	980	3	Ň	B	7	1.000	10
D185612	.40	N	12	44	10	2	B	70	530	62
D185613	.14	Ň	5.7	110	7	1.5	B	20	3701	11
D186062	.13	N	1.5	170	1.5	2L	B	30	510L	75
D185063	.18	N	1.5L	270	2	N	B	50	6501	120
D185064	.10	N	2.2	86	ī.5	2	Ē	30	3901	59
D185065	.07	15L	4.8	150	2	3L	50	15	7301	13
D185066	.08	20	11	74	2	2	20	10	530	Q_1
D185067	.13	20L	9.1	91	1.5	51	70	70	1.100	150

Table 5. Element composition of 106 Iowa coal samples--continued

<u> </u>											
Sample number	Sb (ppm)	Sc-S (ppm)	Se (ppm)	Sr-S (ppm)	Th (ppm)	U (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
D179844	2.2	10	5.1	50	3.0L	13	100	20	2	2,400	50
D179845	1.2	7	2.0	70	3.0L	4.3	70	10	5	2,800	N
D179846	1.4	5	3.3	50	3.0L	3.8	20	20	2	490	20
D179847	37.0	2	25	70	.7	26	200	10	1.5	1,200	10
D179848	4.4	2	.9	20	1.4	2.5	10	10	1	79	5
D179849	1.8	7	3.2	70	3.0L	4.6	50	20	1.5	70	15
D179850	.3	5	3.2	50	3.0L	3.0	50	15	1.5	130	20
D179851	29.0	1	15	30	.6	36	150	7	1	130	5
D179852	2.0	3	1.8	30	3.OL	1.9	20	10	1.5	660	7
D179853	1.0	10	2.6	100	3.0L	6.5	30	20	2	410	15
D179854	.6	15	5.5	300	5.8	5.8	70	15	2	91	20
D179855	.3	5	2.8	300	3.0L	.8	30	10	1	130	10
D179856	.6	10	17	50	3.0L	2.7	100	20	3	820	50
D185601	. 1L	1.5L	1.0	10	.5	.2L	5	7	B	11	5
D185602	.1L	1.5L	1.1	10	.9	.6	7	7	B	12	7
D185603	.3	2	1.7	30	.4	.5	3	7	B	10	3
0185604	.2	1.5	1.1	100	.7	.2L	10	10	В	9.4	10
D185605	.1	1.5	1.2	70	.6	.4	5	7	B	9.2	7
D185606	.2	5	4.2	150	3.8	2.5	30	20	B	140	20
D185609	.3	2L	1.4	30	1.0	.5	3	10	В	180	7
D185610	.5	1.5L	1.6	20	.6	.4	3	10	В	9.2	5
D185611	.6	2L	.8	50	.9	3.2	7	15	1.5	17	7
D185612	12.5	2	13	20	.1L	1.8	10	10	B	3,800	10
D185613	11.1	1.5	6.0	15	1.2	2.3	15	7	B	1,100	10
D186062	1.0	1.5	1.6	15	1.1	1.7	7	7	B	58	7
D185063	1.2	2	1.5	20	1.1	1.7	10	10	В	60	5
D185064	1.4	2	1.5	15	1.1	1.9	15	7	B	280	7
D185065	.1	5	1.7	20	1.9	2.0	15	10	В	13	10
D185066	.1	3	3.1	20	1.8	1.7	20	20	B	120	10
D185067	3.2	3	.1L	20	3.1	2.6	30	15	B	15	15
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Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Si (percent)	Al (percent)	Ca (percent)	Mg (percent)	Na (percent)	K (percent)	Fe (percent)	Ti (percent)	Ag-S (ppm)	As (ppm)
D186068	0.90	0.53	2.7	0.020	0,020	0.071	3.0	0.018		13
D186069	5.6	3.2	.58	.13	.037	.45	2.9	.14	Ň	7.4
D186070	1.5	1.1	.79	.031	.019	.13	2.2	.042	Ň	7.0
D186071	.74	.58	1.4	.022	.018	.11	5.2	.024	.2	33
D186072	1.6	.95	5.1	.059	.031	.15	3.1	.060	N	3.2
D186073	1.6	.76	2.0	.040	.020	.15	3.4	.049	N	6.3
D186074	1.7	.67	2.6	.060	.019	.17	5.1	.048	Ň	14
D186075	1.4	.52	2.4	.046	.039	.099	2.0	.034	Ň	12
D186076	1.3	.38	2.3	.055	.035	.11	3.5	.026	Ň	22
D192368	2.1	.58	3.6	.044	.023	.042	3.7	.038	N	15
D192369	1.7	.58	.98	.038	.032	.067	1.8	.033	N	29
D192370	.62	.25	1.1	.014	.025	.006L	11	.012	N	15
D192371	1.3	.61	1.3	.038	.040	.050	6.6	.030	N	19
D192372	1.5	.86	.92	.029	.031	.053	2.9	.037	N	2.8
D192373	1.9	.70	1.1	.043	.037	.090	2.7	.034	N	11
D192374	.94	.41	.80	.020	.026	.027	4.3	.024	N	13
D192375	3.0	1.5	.56	.044	.036	.14	6.4	.075	N	27
D192376	1.1	.43	1.5	.022	.027	.031	4.0	.049	Ň	16
D192377	2.4	1.1	.45	.042	.024	.085	5.7	.052	N	17
D192378	1.1	.76	1.7	.026	.028	.040	4.8	.029	N	4.6
D192379	3.0	1.3	2.2	.072	.023	.19	2.3	.069	N	18

Table 5. Element composition of 106 Iowa coal samples--continued

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Sample number	B-S (ррт)	Ba-S (ppm)	Be-S (ppm)	Cd (ppm)	Ce-S (ppm)	Co-S (ppm)	Cr-S (ppm)	Cu (ppm)	F (ppm)	Ga-S (ppm)	Ge-S (ppm)
D186068	100	30	2	0.14	70	7	10	7.6	135	B	15
D186069	70	50	5	1.3	150	15	50	29	180	7	20
D186070	100	50	ž		50	15	15	8.3	160	5	15
D186071	100	20	2	11	701	10	10	19	40	5	20
D186072	100	20	2	.44	N	3	15	17	95	5	20
D186073	70	15	2	3.5	N	10	20	12	85	7	15
D186074	100	20	2	12	Ň	15	15	22	75	7	20
D186075	150	20	2	.20	Ň	7	10	6.2	95	5	20
D186076	150	20	2	3.6	N	5	7	8.9	125	5	20
D192368	70	200	1.5	2.2	N	3	7	9.0	35	5	20
D192369	150	20 .	1.5	.29	N	10	7	12	60	7	7
D192370	70	5	2	.26L	N	5	7	16	201	7	15
D192371	150	15	3	17	Ň	50	15	26	25	15	30
D192372	100	50	2	.68	N	7	15	15	35	10	15
D192373	200	15	2	.14L	N	5	10	13	45	5	50
D192374	150	10	2	.40	N	15	10	17	25	10	20
D192375	70	30	3	7.1	N	10	10	13	55	7	20
D192376	70	700	3	19	Ň	5	7	13	30	, ,	20
D192377	100	30	3	.40	N	15	15	46	40	10	30
D192378	150	30	3	3.1	N	3	10	13	60	5	20
D192379	150	50	3	.63	N	7	15	14	90	7	15

Table 5. Element composition of 106 Iowa coal samples--continued

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Sample number	Hg (ppm)	La-S (ppm)	Li (ppm)	Mn (ppm)	Mo-S (ppm)	Nb-S (ppm)	Nd-S (ppm)	Ni-S (ppm)	P (ppm)	Pb (ppm)
D186068	0,09	20	5.5	220	N	31	20	20	6001	30
D186069	.14	50	59	62	5	5	Ň	50	1,100	78
D186070	-08	15	18	48	1.5	2	30	10	630	12
D186071	.17	20	4.5	83	2	3	20	50	6301	38
D186072	.13	20L	18	610	3	5L	Ň	10	970L	34
D186073	.14	N	13	120	7	3L	В	20	660L	23
D186074	.15	20L	7.0	210	3	5	30L	70	870L	81
D186075	.08	15L	4.1	250	1	3	20L	50	580L	22
D186076	.10	15L	3.0	530	ī	3L	20L	50	660L	35
D192368	.14	N	4.1	380	1.5	N	B	30	1,100L	36
0192369	.07	N	4.9	130	N	N	В	30	500L	31
0192370	.70	N	5.2	160	N	N	В	20	1.100L	470
0192371	.17	20L	7.2	280	1.5	N	N	100	900L	91
0192372	.09	N	18	140	2	N	В	20	590L	13
0192373	.12	N	3.0	170	5	N	B	15	620L	51
D192374	.28	15	4.4	90	N	N	N	50	590L	82
D192375	.28	N	20	92	N	5L	В	50	1.000L	170
D192376	.29	15	7.4	120	N	3L	20	30	720L	80
D192377	.25	N	20	160	7	5L	В	70	870L	200
D192378	.23	20L	12	250	3	N	N	20	790L	110
D192379	.10	N	14	410	N	N	В	30	910L	28

Table 5. Element composition of 106 Iowa coal samples--continued

Sample number	Sb (ppm)	Sc-S (ppm)	Se (ppm)	Sr-S (ppm)	Th (ppm)	U (ppm)	V-S (ppm)	Y-S (ppm)	Yb-S (ppm)	Zn (ppm)	Zr-S (ppm)
D186068	0.3	5	3.3	200	1.2	0.5	20	20		27	10
D186069	.3	7	.11	50	6.0	5.6	50	20	B	490	50
D186070	.1	3	2.6	200	1 9	1 1	30	15	Ř	54	15
D186071	1.2	5	3.2	150	1 2		16	10	R	2 000	10
D186072	.1	5	1.9	150	2.2	4.5	30	15	B	100	15
D186073	.4	5	.11	20	1.5	6.5	30	10	B	730	15
D186074	.5	7	.1L	30	1.3	1.8	30	15	B	3,700	15
D186075	.7	3	.10	100	1.1	.6	15	10	B	160	10
D186076	.8	2	B	100	.8	.4	10	10	B	1.000	10
D192368	.8	2	1.7	30	1.0	1.2	10	7	B	170	7
D192369	1.2	3	1.5	70	1.0	1.1	15	7	В	170	7
D192370	.2	5	2.5	20	.9	1.8	20	7	B	9.6	7
D192371	.7	7	1.3	30	1.0	2.5	30	20	3	2.600	15
D192372	.1L	5	2.8	20	1.5	2.4	20	10	B	300	20
D192373	5.4	2	2.5	20	1.1	5.2	15	5	B	9.6	10
D192374	.5	2	3.4	50	.9	1.1	15	5	В	69	7
D192375	.6	3	3.5	70	2.6	2.2	15	7	В	690	15
D192376	.4	3	2.9	150	1.0	1.8	10	7	В	4.200	10
D192377	1.8	5	3.4	50	1.7	6.9	30	10	B	12	15
D192378	.1L	3	2.9	200	1.4	1.0	20	5	B	540	10
D192379	.7	5	1.7	70	2.1	1.2	20	15	В	42	15

Table 5. Element composition of 106 Iowa coal samples--continued

Table 6.	Arithmetic mean, observed range, geometric mean, and geometric deviation
	of proximate and ultimate analyses, heat of combustion, forms of sulfur,
	and ash-fusion temperatures for 11 Middle Pennsylvanian-age coal samples
	from coal-zone 2, Cherokee Group, south-central and southeastern Iowa

[All values are in percent except kcal/kg, Btu/lb, ash-fusion temperatures and geometric deviations and are reported on an as-received basis. Kcal/kg = 0.556 (Btu/lb). °F = (°C x 1.8) + 32]

		Obser	ved range	0	Geometric deviation
	Arithmetic mean	Minimum	Maximum	Geometric mean	
		Proximate and	l ultimate analy	sis	
Moisture	9.6	5.3	13.0	9.3	1.3
matter Fixed	34.4	27.9	40.6	33.9	1.1
carbon	38.6	30.6	48.5	37.9	1.1
Ash	18.0	9.7	31.1	16.8	1.4
Hydrogen	4.9	3.6	5.9	4.8	1.1
Carbon	54.0	36.6	65.5	53.0	1.2
Nitrogen	.9	.4	1.2	q	1.3
Ovvden	15 3	10 0	10 5	15 1	1 3
Sulfur	7 2	2.2	10 2	10+1	1.5
	/•C	J. Z	10.5	0.5	1.7
		Heat of	f combustion		
Kcal/kg Btu/lb	5,360 9,630	4,050 7,280	6,060 10,900	5,315 9,560	1.1 1.1
		Forms	s of sulfur		
Sulfate	1.04	0.31	2.98	0.87	1.8
Pyritic	4.51	1.97	15.47	3.59	1.9
Organic	1.82	.42	2.61	1.57	1.7
		Ash-fusion	temperatures (°C	;)	
Initial gefor-					
mation	1,165	1,040	1,335	1,165	1.1
Softening	1,210	1,095	1,380	1,210	1.1
Fluid	1,255	1,150	1,430	1,255	1.1

Table 7. Arithmetic mean, observed range, geometric mean, and geometric deviation of 37 elements in 15 Middle Pennsylvanian age coal samples from coalzone 2, Cherokee Group, southcentral and southeastern lowa

[All analyses except geometric deviations are in percent or parts per million and are reported on a whole-coal basis. L, less than the value shown]

• •

		Obser	ved range		
Element	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation
		Р	ercent		
Si	1.6	0.76	3.0	1.5	1.6
A1	1.0	.41	2.2	.88	1.7
Ca Ma	1.8	.10	5.1	1.2	2.4
ny	•030	.020	*U0 4	•004	1.0
Na	.027	.017	.040	.026	1.3
K	.096	.027	.23	.077	1.9
Fe	4.9	1.9	14	4.1	1.8
11	.050	.018	•095	•(J44	1.0
		Parts	per million		
As		28	25	8.7	2 0
B	100	50	150	100	1.5
Ba	100	10	1.000	50	3.5
Be	3	1.5	5	2	1.4
Cd	1.4	.2L	19	.5	4.4
Co	10	3	50	7	2.0
Cr	20	7	100	15	2.1
Cu	18	7.1	56	15	1.8
F Ga	80 10	25	1/5 20	60 7	2.1
0-	10	15			
Ge	15	15	3()	15	1.2
ng La	15	151	-40 30	•15 15	1.0
Li	16	4.4	49	13	2.0
Mn	190	21	610	140	2.3
Mo	3	1.51	30	1.5	3.7
Nd	20	20L	30	20	1.3
Ni	30	10	150	20	2.2
Pb	58	12	250	39	2.4
SD	.4	.1L	1.4	.3	2.5
Sc	5	2	10	5	1.6
Se	3.9	1.7	17	3.3	1.7
35 Th	100	20	200	100	2.1
U	2.0	.5	4.5	1.0	2.4 1.8
v	30	10	100	20	17
Ŷ	15	5	20	15	1.6
Zn	470	7.1	4,200	120	5.5
Zr	15	7	50	15	1.7

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Table 8. Arithmetic mean, observed range, geometric mean, and geometric deviation of proximate and ultimate analyses, heat of combustion, forms of sulfur, and ash-fusion temperatures for 9 Middle Pennsylvanian-age coal samples from coal-zone 3, Cherokee Group, south-central and southeastern Iowa

[All values are in percent except kcal/kg, Btu/lb, ash-fusion temperatures and geometric deviations, and are reported on an as-received basis. Kcal/kg = 0.556 (Btu/lb). °F = (°C x 1.8) + 32]

 		Obser	ved range	·····	`
	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation
		Proximate and	i ultimate analy	Ses	
Moisture Volatile	11.1	7.7	13.8	10.9	1.3
matter Fixed	34.5	30.4	40.7	33.9	1.1
carbon	38.4	33.0	43.2	38.2	1.1
Ash	16.7	10.3	30.6	15.6	1.4
Hydrogen	5.0	4.2	5.5	5.0	1 2
Carbon	54.2	44 1	62 7	53 1	1 1
Nitrogen	0	7	1 3	55.1	1 2
Ovygon	• ⁵ 17 1	14 5	21.1	•7	1.2
Culfur	1/.1	14.5		15./	1.3
Sultur	7.0	2.8	17.3	0.1	1./
		Heat of	f combustion		
Kcal/kg Btu/lb	5,410 9,730	4,545 8,170	6,080 10,940	5,380 9,680 ·	1.1 1.1
		Forms	s of sulfur		
Sulfate	1.59	0.01	3.57	0.54	4.6
Pvritic	4.31	1.41	16.2	3.36	2.0
Organic	1.66	.42	2.89	1.36	1.8
		Ash-fusion 1	temperatures (°C)	
Initial defor-					, , , , <u>, , , , , , , , , , , , , ,</u>
mation	1,130	1,040	1,240	1,130	1.1
Softening	1,180	1,095	1,290	1,180	1.1
Fluid	1,225	1,145	1,320	1,225	1.0

Table 9. Arithmetic mean, observed range, geometric mean, and geometric deviation of 36 elements in 15 Middle Pennsylvanian-age coal samples from coalzone 3, Cherokee Group, south-central and southeastern Iowa

[All Analyses except geometric deviations are in percent or parts per million and are reported on a whole-coal basis. L, less than the value shown]

		Obser	ved range			
Element	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation	
		р	ercent			
Si	1.7	0.38	5.8	1.3	2.0	
ĂÌ	.92	.24	3.2	.75	1.9	
Ca	1.3	.087	2.6	.85	2.6	
Mg	.037	.007	.13	.029	2.0	
Na	.022	.009	.043	.020	1.6	
ĸ	.12	.019	.50	.068	2.9	
Fe	6.1	1.8	21	4.8	2.0	
Ti	.043	.012	.13	•037	1./	
		Parts	per million			
A		<u>.</u>	40	14	2 /	
AS R	100	50	150	14	2,4	
Ba	30	5	200	20	2.8	
Be	2	1.5	5	2	1.3	
Cd	9.2	•2L	15	9	10	
Co	10	3	30	10	1.8	
Cr	15	7	50	15	1.7	
Cu	25	9.2	69	21	1.8	
F	56	201	160	45	2.0	
Ga	/	3	10	/	1.5	
Ge	20	7	50	15	1.6	
Hg	.10	•07	.7	.17	1.8	
La	15	15L	50	10	2.1	
1 1	12	3.5	32	9.0	2.1	
1311	120	54	210	100	1.0	
Мо	5	2L	10	3	2.1	
N1 Dh	50	/	100	30	2.1	
70 Sh	90 90	9.1	4/0	54	3.0	
Sc	7	1 5	1.0	•0 5	6.6 1 7	
54	•	1.0	15	5	£+7	
Se	3.0	1L	8.2	2.6	1.7	
36 Th	/U 2 E	5	300	30	3.0	
Ű	3.0	יאר פ	۲۱ ۲۱	1.2	5.4 2 1	
v	30	15	70	20	1.7	
Y	10	7	20	10	1 A	
Zn	2,800	9.6	5.500	290	9.8	
Zr	15	7	50	10	1.7	
<u> </u>						

Table 10.	Arithmetic mean, observed range, geometric mean, and geometric dev	iation
	of proximate and ultimate analyses, heat of combustion, forms of s	ulfur,
	and ash-fusion temperatures for 32 Middle Pennsylvanian-age coal s	amples
	from coal-zone 4, Cherokee Group, south-central and southeastern I	owa

[All values are in percent except kcal/kg, Btu/lb, ash-fusion temperatures and geometric deviations and are reported on an as-received basis. L, less than the value shown. Kcal/kg = 0.556 (Btu/lb). $^{\circ}F = (^{\circ}C \times 1.8) + 32$]

		Obser	ved range			
	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation	
		Proximate and	ultimate analy	Ses		
Moisture Volatile	13.4	7.9	17.7	13.2	1.2	
matter Fixed	33.4	23.3	38.8	32.8	1.1	
carbon	36.7	22.8	48.1	35.8	1.1	
Ash	17.1	7.0	32.8	15.7	1.5	
Hydrogen	5.1	3.5	5.8	5.1	1.2	
Carbon	52.6	34.4	61.0	51.5	1.1	
Nitrogen	1.0	•2	1.3	1.0	1.2	
Oxygen	18.1	9.6	23.0	18.1	1.2	
Sulfur	6.4	2.3	19.2	5.5	1.7	
		Heat of	combustion			
Kcal/kg Btu/lb	5,440 9,780	3,840 6,910	6,365 11,450	5,410 9,730	1.1 1.1	
		Forms	; of sulfur			
Sulfate	0.68	0.01L	1.88	0.43	2.6	
Pyritic	4.26	.45	16.6	3.20	2.1	
Organic	1.58	.48	3.13	1.36	1.7	
		Ash-fusion 1	cemperatures (°C)		
Initial deform				<u></u>		
mation	1,145	1,015	1,485	1,145	1,1	
Softening	1,195	1,070	1,530	1,190	1.1	
Flufd	1,235	1,105	1,540	1,235	1.1	

Table 11.Arithmetic mean, observed range, geometric mean, and geometric deviation
of 35 elements in 49 Middle Pennsylvanian-age coal samples from coal-
zone 4, Cherokee Group, south-central and southeastern lowa

[All analyses except geometric deviations are in percent or parts per million and are reported on a whole-coal basis. L, less than the value shown]

		Obser	ved range			
Element	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation	
		Р	ercent			
Si Al Ca	2.0 .94 2.0	0.48 .10	6.7 3.2 7.1	1.6 .71 1.5	1.8 2.1 2.2	
Mg	.065	.016	.39	.052	2.0	
Na K Fe Ti	.027 .13 4.7 .046	.009 .023 .93 .009	.065 .61 16 .18	.023 .11 4.1 .036	1.7 1.9 1.7 2.0	
		Parts	per million			
As B Ba Be Cd	26 100 70 2 12	3 50 7 1 .12L	230 200 1,000 7 62	17 100 30 2 .24	2.7 1.5 3.5 1.6 19	
Co Cr Cu F Ga	15 15 25 84 7	2 2 4.0 20 1.5	300 50 170 250 20	7 10 19 75 7	2.8 2.2 2.1 1.6 2.0	
Ge Kg Li Mn Mo	20 .16 11 240 7	7 .07 1.2 44 .7	70 .49 58 980 100	15 .14 6.8 190 3	1.7 1.7 2.8 2.0 3.4	
N1 Pb Sb Sc Se	50 63 1.7 5 3.3	.7 8.1 1.5 .1L	700 420 22 15 29	30 43 .7 3 2.1	2.5 2.4 3.6 2.1 2.5	
Sr Th V V Y	50 2.8 4.2 20 10	10 .4 .2L 3 5	500 18 43 100 70	30 .7 2.0 15 10	2.4 6.3 3.3 2.3 1.7	
Zn Zr	850 15	9.2 3	4,200 70	150 15	6.7 1.9	

Table 12.Arithmetic mean, observed range, geometric mean, and geometric deviation
of proximate and ultimate analyses, heat of combustion, forms of sulfur,
and ash-fusion temperatures for four Middle Pennsylvanian-age coal
samples from coal-zone 5, Cherokee Group, south-central and southeastern
Iowa

[All values are in percent except kcal/kg, Btu/lb, ash-fusion temperatures and geometric deviations, and are reported on an as-received basis. Kcal/kg = 0.556 (Btu/lb). °F = (°C x 1.8) + 32]

e.		Obser	ved range		
	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation
		Proximate and	l ultimate analy	ses	
Moisture Volatile	8.5	5.0	12.3	8.1	1.5
matter Fixed	33.9	29.3	39.0	33.7	1.1
carbon	34.9	28.1	40.1	34.5	1.2
Ash	23.2	14.6	33.7	21.8	1.4
Hydrogen	4.4	4.0	5.3	4.4	1.1
Carbon	50.2	39.6	58.4	49.3	1.2
Nitrogen	.9	.7	1.0	.9	1.2
Oxygen	14.2	10.9	18.2	14.0	1.2
Sulfur	7.7	5.1	10.8	7.3	1.4
		Heat of	f combustion		
Kcal/kg Btu/lb	4,870 8,760	3,905 7,020	5,790 10,410	4,800 8,640	1.2 1.2
		Forms	s of sulfur		
Sulfate	1.07	0.18	1.89	0.66	2.8
Pyritic Organic	4.11 2.65	1.82 2.17	6.03 3.01	3.62 2.62	1.7 1.2
		- 			<u></u>
		Ash-fusion 1	temperatures (°C)	
Initial defor-			1 150	1 110	1.0
mation	1,115	1,080	1,150	1,115	1.0
Softening	1,145	1,110	1,205	1,145	1.0
Fluid	1,200	1,140	1,280	1,200	1.1

Table 13.Arithmetic mean, observed range, geometric mean, and geometric deviationof 35 elements in five Middle Pennsylvanian-age coal samples from coal-
zone 5, Cherokee Group, south-central and southeastern Iowa

[All analyses except geometric deviations are in percent or parts per million and are reported on a whole-coal basis]

		Obser	ved range		
Element	Arithmetic mean	Mintmum	Maximum	Geometric mean	Geometric deviation
		P	ercent		
Si	3.9	2.1	7.6	3.4	1.7
. A1	1.7	.58	3.6	1.3	2.1
Ca	1.3	.32	3.6	.92	2.4
Mg	.062	.044	.084	.060	1.3
Na	.050	.023	.078	.045	1.6
ĸ	.18	.042	.27	.14	2.0
Fe	5.6	2.7	8.0	5.1	1.6
Ťi	.095	.038	.21	.077	1.9
		Parts	per million		
	17				1.6
AS	1/	0 70	30	15	1.2
D Ra	70	30	300	70	2 2
Bo	2	1 5	300	2	1 3
0e C4	20	2 T+D	7/	21	17
CU .	23	•3		C • 1	17
Co	30	3	100	15	4.1
Cr	30	7	70	20	2.3
Cu	28	9.0	45	23	1.9
F	96	35	155	78	1.9
Ga	/	3	10	/	1./
Ge	15	10	30	15	1.8
Hg	.15	.12	.20	.15	1.2
LĪ	27	4.1	80	15	3.2
Mn	310	110	740	220	2.3
Мо	7	1.5	15	5	2.7
NI	100	30	200	70	2.3
РЬ	45	22	100	38	1.8
Sb	1.8	.8	3.9	1.5	1.8
Sc	7	2	10	5	2.2
Se	21	1.7	75	8.1	4.5
Sr	300	30	700	150	4.5
Th	3.5	1.0	10	1.6	2.8
U	10	1.2	35	4.8	3.7
v	/0	10	150	50	3.1
r	20	7	30	15	2.0
Zn	2,100	12	5,200	220	12
Zr	30	7	70	20	2.5
				••	

Table 14.Arithmetic mean, observed range, geometric mean, and geometric deviation
of proximte and ultimate analyses, heat of combustion, forms of sulfur,
and ash-fusion temperatures for 9 Middle Pennsylvanian-age coal samples
from coal-zones 6, 7, 8, and 9, Cherokee Group, south-central and
southeastern lowa.

[All values are in percent except kcal/kg, Btu/lb, ash-fusion temperatures, and geometric deviations, and are reported on an as-received basis. Kcal/kg = 0.556 (Btu/lb). $^{\circ}F = (^{\circ}C \times 1.8) + 32$]

···	• • • • • • • •	Obsei	rved range		
	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation
		Proximate and	1 ultimate analy	Ses	
Moisture	13.1	7.9	18.7	12.7	1.3
matter	33.8	27.3	41.2	33.2	1.1
carbon	38.8	25.5	45.2	38.0	1.2
Ash	14.9	8.4	36.7	13.2	1.6
Hydrogen	5.3	3.9	6.0	5.2	1.2
Carbon	55.3	38.4	67.2	54.1	1.2
Nitrogen	.9	.6	1.2	.9	1.2
Oxvaen	18.8	12.5	24.8	16.9	1.3
Sulfur	4.8	2.4	8.9	4.4	1.5
	<u>,, , , , , , , , , , , , , , , ,</u>	Heat o	f combustion		·····
Kcal/kg Btu/lb	5,530 9,940	3,710 6,670	6,270 11,280	5,560 9,820	1.2 1.2
<u></u>		Form	s of sulfur	<u></u>	
Sulfate	0.70	0.01	1.09	0.29	3,8
Pyritic Organic	2.72 1.71	.33 .97	6.40 2.45	2.04 1.62	2.1 1.3
	<u></u>	Ash-fusion	temperatures (°C)	
Initial defor-			<u></u>		
mation	1,085	1,040	1,140	1,085	1.0
Softening	1,120	1,060	1,175	1,115	1.0
Fluid	1,145	1,080	1,200	1,140	1.0

Table 15.	Arithmetic mean, observed range, geometric mean, and geometric deviation
	of 38 elements in 16 Middle Pennsylvanian-age coal samples from coal-
	zones 6, 7, 8 and 9, Cherokee Group, south-central and southeastern Iowa

[All analyses except geometric deviations are in percent or parts per million and are reported on a whole-coal basis]

		Observed range			
Element	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation
		Ρ	ercent		
Si	2.7	0.59	12	1.9	2.3
Al	1.1	.27	5.9	.74	2.3
Ca	1.5	.28	4.9	1.1	2.3
Mg	.083	.026	.39	.061	2.2
Na	.052	.010	.21	.036	2.4
K	.18	.057	1.0	.13	2.4
Fe	3.3	.27	7.8	2.3	2.4
T1	.064	.012	.31	.043	2.4
		Parts	per million		
Ag	0.5	0.1	2	0.2	4.8
As	31	4.5	230	18	2.8
B	100	15	150	100	1.9
Ba	100	10	2,000	50	3.9
Be	3	1.5	5	2	1.5
Cd	8.9	.71	100	1.3	8.5
Co	10	1	20	5	2.7
Cr	20	7	150	20	2.1
Cu	31	7.4	160	22	2.3
F	78	30	350	62	2.0
Ga	7	2	30	7	2.0
Ge	20	10	50	20	1.6
Hg	.17	.07	.30	.14	1.6
Li	15	1.0	88	6.2	3.9
Mn	250	67	780	210	1.9
Mo Nb Ni Pb Sb	15 2 30 69 9.4	2 1 5.9 .3	150 15 70 260 37	7 1 20 40 3.5	3.1 3.5 2.0 2.9 4.3
Sc	5	1	15	3	2.2
Se	7.7	.9	25	4.9	2.6
Sr	50	15	150	30	1.9
Th	8.4	.6	22	.9	2.8
U	11	.5	36	4.3	4.0
V	100	10	300	50	3.0
Y	15	7	30	10	1.5
Yb	1.5	.7	3	1	1.5
Zn	620	8.8	5,400	110	7.0
Zr	20	5	100	15	2.6

Table 16.Arithmetic mean, observed range, geometric mean, and geometric deviation
of proximate and ultimate analyses, heat of combustion, forms of sulfur,
and ash-fusion temperatures for 65 Middle Pennsylvanian-age coal samples
from the Cherokee and Marmaton Groups, south-central and southeastern
lowa

[All values are in percent except kcal/kg, Btu/lb, ash-fusion temperatures and geometric deviations, and are reported on an as-received basis. L, less than the value shown. Kcal/kg = 0.556 (Btu/lb). $^{\circ}F = (^{\circ}C \times 1.8) + 32$]

		Observed range		-	
	Arithmetic mean	Mintmum	Maximum	Geometric mean	Geometric deviation
		Proximte and	ultimate analys	25	<u>, , , , , , , , , , , , , , , , , , , </u>
Moisture Volatile	12.1	5.3	18.7	11.7	1.3
matter Fixed	33.1	23.3	41.5	33.4	1.1
carbon	37.3	22.8	48.5	37.0	1.1
Ash	17.2	7.0	36.7	15.9	1.5
Hydrogen	5.0	2.9	6.0	5.0	1.2
Carbon	52.9	34.4	67.2	52.5	1.1
Nitrogen	.9	.4	1.3	.9	1.2
Oxygen	17.2	2.5	27.9	17 1	1 3
Sulfur	6.6	2.3	20.9	5.8	1.7
<u> </u>	······································	Heat of	combustion		
Kcal/kg Btu/lb	5,400 9,710	3,710 6,670	6,370 11,450	5,360 9,640	1.1 1.1
••••••••••••••••••••••••••••••••••••••		Forms	of sulfur		
Sulfate	0.92	0.01L	3.57	0.50	3.0
Pyritic Organic	4.19 1.70	.33 .42	18.8 3.13	3.20 1.50	2.1 1.7
		Ash-fusion t	cemperatures (°C)	
Initial defor-					
mation	1,130	985	1,485	1,130	1.1
Softening	1,175	1,040	1,530	1,175	1.1
Fluid	1,215	1,080	1,540	1,215	1.1

Table 17.	Arithmetic mean, observed range, geometric mean, and geometric deviation
	of 38 elements in 105 Middle Pennsylvanian-age coal samples from the
	Cherokee and Marmaton Groups, south-central and southeastern lowa

Arithmetic mean Minimum Maximum Geometric mean Geometric deviation Si 2.0 0.38 12 1.7 1.9 Si 2.0 0.38 12 1.7 1.9 Si 2.0 0.38 12 1.7 1.9 Al 1.0 5.9 .78 2.0 Ca 1.7 .087 7.1 1.2 2.3 Mg .030 .009 .21 .026 1.8 K .13 .010L 1.0 .10 2.1 Fe 5.0 .27 21 .040 2.0 Ti .051 .009 .31 .040 2.0 Itis .01 15 2.0 1.6 8 100 15 200 100 1.6 Ba 100 5 3.000 7 2.5 Cr 15 2 150 15 2.1 Cu 2		Observed range					
Percent Si 2.0 0.38 12.9 1.7 1.9 A1 1.0 5.9 .78 2.0 Ca 1.7 0.87 7.1 1.2 2.3 Mg .058 .007 .39 .045 2.0 Na .030 .009 .21 .026 1.8 Fe 5.0 .27 21 4.0 2.0 Ti .051 .009 .31 .040 2.0 Parts per million Ag 0.05 8.1 As 24 2.0 240 15 2.5 Ba 100 15 2.00 100 1.6 Ba 100 5 3.000 30 3.6 Ba 100 .4 17 Co 1.5 2.1 Cd 18 .1L 100 .4 17 Co 1.2 Co 15	Element	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$			F	Percent			
A1 1.0 .10 5.9 .78 2.0 Ca 1.7 .087 7.1 1.2 2.3 Mg .058 .007 .39 .045 2.0 Na .030 .009 .21 .026 1.8 K .13 .010L 1.0 .10 2.1 Fe 5.0 .27 21 4.0 2.0 Ti .051 .009 .31 .040 2.0 Parts per million Parts per million Ag 0.3 0.11 3 0.05 8.1 As 24 2.0 240 15 2.5 B 100 15 2.00 30 3.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 2 150 15 2.1 Cd 18 .1L 100 .4 17 Co 15 1 </td <td>Si</td> <td>2.0</td> <td>0.38</td> <td>12</td> <td>1.7</td> <td>1.9</td>	Si	2.0	0.38	12	1.7	1.9	
Ca 1.7 .087 7.1 1.2 2.3 Mg .058 .007 .39 .045 2.0 Na .0300 .009 .21 .026 1.8 K .13 .010L 1.0 .10 2.1 Fe 5.0 .27 21 4.0 2.0 Parts per million Ag 0.3 0.11 3 0.040 2.0 Parts per million Ag 0.3 0.11 3 0.05 8.1 As 24 2.0 240 15 2.5 B 100 15 200 100 1.6 Ba 100 5 3.000 30 3.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 1.8 Ga	A1	1.0	.10	5.9	.78	2.0	
Mg .058 .007 .39 .045 2.0 Na .030 .009 .21 .026 1.8 K .13 .010L 1.0 .10 2.1 Fe 5.0 .27 21 .026 1.8 .051 .009 .31 .040 2.0 Parts per million	Ca	1.7	.087	7.1	1.2	2.3	
Na .030 .009 .21 .026 1.8 Fe .13 .010 1.0 .10 2.1 Fe 5.0 .27 21 4.0 2.0 Parts per million Parts per million Ag 0.3 0.11 3 0.05 8.1 As 2.0 240 15 2.5 5 5 5 2.5 5 2.0 1.6 5 Ba 100 5 3.000 30 3.6 5 2.5 5 7 2 1.5 Cd 18 .1L 100 .4 17 2 2.5 Cr 15 1 300 7 2.5 2.1 2.0 1.5 2.1 Cu 25 4.0 170 19 2.0 1.7 1.5 30 7 1.8 5 4.0 1.7 1.7 1.7 1.7 1.7 </td <td>Mg</td> <td>.058</td> <td>.007</td> <td>.39</td> <td>.045</td> <td>2.0</td>	Mg	.058	.007	.39	.045	2.0	
K .13 .010L 1.0 .10 2.1 Fe 5.0 .27 21 4.0 2.0 Parts per million Parts per million Ag 0.3 0.11 3 0.05 8.1 As 24 2.0 240 15 2.5 B 100 15 200 100 1.6 Ba 100 5 3.000 30 3.6 Be 2 1 7 2 1.5 Cd 18 .11L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Ga 7 1.5 30 7 1.8 Ge 20 7 70 .14 1.7 Hg .17 .07 .70 .14 1.7 Hg .10 <t< td=""><td>Na</td><td>.030</td><td>.009</td><td>.21</td><td>.026</td><td>1.8</td></t<>	Na	.030	.009	.21	.026	1.8	
Fe 5.0 .27 21 4.0 2.0 Ti .051 .009 .31 .040 2.0 Parts per million Parts per million Ag 0.3 0.1L 3 0.05 8.1 As 24 2.0 240 15 2.5 B 100 15 200 100 1.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Cu 25 4.0 170 19 2.0 F 78 20L 350 65 1.7 Ga 7 1.5 30 7 1.8 Ga 70 70 20 1.7 1.7 Li 13 1.0	ĸ	.13	.010L	1.0	.10	2.1	
Ti .051 .009 .31 .040 2.0 Parts per million Ag 0.3 0.1L 3 0.05 8.1 As 24 2.0 240 15 2.5 B 100 15 200 100 1.6 Ba 100 5 3.000 30 3.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Cu 25 4.0 170 19 2.0 F 78 20L 350 65 1.8 Ga 7 1.5 30 7 1.8 Ge 20 7 70 2.1 7.1 Hg .17 .07 .70 .1.7 8	Fe	5.0	.27	21	4.0	2.0	
Ag 0.3 0.1L 3 0.05 8.1 As 24 2.0 240 15 2.5 B 100 15 2.0 100 1.6 Ba 100 5 3.000 30 3.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Cu 25 4.0 170 19 2.0 F 78 20L 350 65 1.8 Ga 7 1.5 30 7 1.8 Ge 20 7 70 20 1.7 Hg .17 .07 .70 3 3.7 Hg .17 .07 .70 3 3.7 Hg .17 .07	Ti	.051	.009	.31	•040	2.0	
Ag 0.3 0.1L 3 0.05 8.1 As 24 2.0 240 15 2.5 B 100 15 200 100 1.6 Ba 100 5 3.000 30 3.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Cu 25 4.0 170 19 2.0 F 78 20L 350 65 1.8 Ga 7 1.5 30 7 1.8 Ga 7 1.5 30 7 1.8 Ga 7 7.07 .14 1.7 Hg .17 .07 .70 .14 1.7 Hg .17 .07 .3 3.7 3 3.7 Mn 210 21 960 170<			Parts	per million			
As242.0240152.5B100152001001.6Ba10053.000303.6Be21721.5Cd18.1L100.417Co15130072.5Cr152150152.1Cu254.0170192.0F7820L350651.8Ga71.53071.8Ge20770201.7Hg.17.07.70.141.7La1010L7054.0Li131.0887.92.8Mn210219801702.0Li15.33.733.7Nb1.51L15.35.8Ni505700302.3Pb675.9470442.5Sb2.1.1L752.82.7Sr705700502.6Th3.6.4L221.04.0U4.5.2L432.43.1V303300202.4V3033001507.5Zr1531001507.5Zr153<	Ag	0.3	0.1L	3	0,05	8.1	
B 100 15 200 100 1.6 Ba 100 5 3,000 30 3.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Cu 25 4.0 170 19 2.0 F 78 20L 350 65 1.8 Ga 7 1.5 30 7 1.8 Ge 20 7 70 20 1.7 Hg .17 .07 .70 .14 1.7 La 10 10L 70 5 4.0 Li 13 1.0 88 7.9 2.8 Mn 210 21 980 170 2.0 Mo 7.5 1L	As	24	2.0	240	15	2.5	
Ba 100 5 3,000 30 3.6 Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Cu 25 4.0 170 19 2.0 F 78 20L 350 65 1.8 Ga 7 1.5 30 7 1.8 Ge 20 7 70 20 1.7 Hg .17 .07 .70 20 1.7 Hg .17 .07 .70 2.8 Mn 210 21 980 170 2.0 Li 15 .1L 150 .3 5.8 Mi 50 5 700 30 2.3 Sb 2.1 .1L 37	В	100	15	200	100	1.6	
Be 2 1 7 2 1.5 Cd 18 .1L 100 .4 17 Co 15 1 300 7 2.5 Cr 15 2 150 15 2.1 Cu 25 4.0 170 19 2.0 F 78 20L 350 65 1.8 Ga 7 1.5 30 7 1.8 Ge 20 7 70 20 1.7 Hg .17 .07 .70 .14 1.7 La 10 10L 70 5 4.0 Li 13 1.0 88 7.9 2.8 Mn 210 21 980 170 2.0 Mo 7 5.1L 15 .3 5.8 Mi 50 5 700 30 2.3 Pb 67 5.9	Ba	100	5	3,000	30	3.6	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Be	2	1	7	2	1.5	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Cd	18	. 1L	100	.4	17	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Co	15	1	300	7	2.5	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Cr	15	2	150	15	2.1	
F7820L350651.8Ga71.53071.8Ge20770201.7Hg.17.07.70.141.7La1010L7054.0Li131.0887.92.8Mn210219801702.0Mo77L15033.7Nb1.51L15.35.8Ni505700302.3Pb675.9470442.5Sb2.1.1L752.82.7Sr705700502.6Th3.6.4L221.04.0U4.5.2L432.43.1V303300202.4Y15570101.7Zn1,1007.118,0001507.5Zr153100152.0	Cu	25	4.0	170	19	2.0	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	F	78	20L	350	65	1.8	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Ga	7	1.5	30	7	1.8	
Hg.17.07.70.141.7La1010L7054.0Li131.0887.92.8Mn210219801702.0Mo77L15033.7Nb1.51L15.35.8Ni505700302.3Pb675.9470442.5Sb2.1.1L1552.1Sc51L1552.1Se4.5.1L752.82.7Sr705700502.6Th3.6.4L221.04.0U4.5.2L432.43.1V303300202.4Y15570101.7Zn1,1007.118,0001507.5Zr153100152.0	Ge	20	7	70	20	1.7	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Hg	.17	.07	.70	.14	1.7	
L1 13 1.0 88 7.9 2.8 Mn 210 21 980 170 2.0 Mo 7 7L 150 3 3.7 Nb 1.5 1L 15 .3 5.8 Ni 50 5 700 30 2.3 Pb 67 5.9 470 44 2.5 Sb 2.1 .1L 37 .8 4.0 Sc 5 1L 15 5 2.1 Se 4.5 .1L 75 2.8 2.7 Sr 70 5 700 50 2.6 Th 3.6 .4L 22 1.0 4.0 U 4.5 .2L 43 2.4 3.1 V 30 3 300 20 2.4 Y 15 5 70 10 1.7 Zn 1.100 7.1 18,000 150 7.5 Zr 15 3 100 15	La	10	10L	70	5	4.0	
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	L1	13	1.0	88	7.9	2.8	
Mo77L15033.7Nb1.51L15.35.8Ni505700302.3Pb675.9470442.5Sb2.1.1L37.84.0Sc51L1552.1Se4.5.1L752.82.7Sr705700502.6Th3.6.4L221.04.0U4.5.2L432.43.1V303300202.4Y15570101.7Zn1,1007.118,0001507.5Zr153100152.0	Mn	210	21	980	170	2.0	
Nb1.51L15.35.8Ni505700302.3Pb675.9470442.5Sb2.1.1L37.84.0Sc51L1552.1Se4.5.1L752.82.7Sr705700502.6Th3.6.4L221.04.0U4.5.2L432.43.1V303300202.4Y15570101.7Zn1.1007.118,0001507.5Zr153100152.0	Mo	7	7L	150	3	3.7	
N1505700302.3Pb675.9470442.5Sb2.1.1L37.84.0Sc51L1552.1Se4.5.1L752.82.7Sr705700502.6Th3.6.4L221.04.0U4.5.2L432.43.1V303300202.4Y15570101.7Zn1,1007.118,0001507.5Zr153100152.0	Nb	1.5	1L	15	.3	5.8	
Pb 67 5.9 470 44 2.5 Sb 2.1 $.1L$ 37 $.8$ 4.0 Sc 5 $1L$ 15 5 2.1 Se 4.5 $.1L$ 75 2.8 2.7 Sr 70 5 700 50 2.6 Th 3.6 $.4L$ 22 1.0 4.0 U 4.5 $.2L$ 43 2.4 3.1 V 30 3 300 20 2.4 Y 15 5 70 10 1.7 Zn $1,100$ 7.1 $18,000$ 150 7.5 Zr 15 3 100 15 2.0	Ni	50	5	700	30	2.3	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	סץ	67	5.9	470	44	2.5	
Sc51L1552.1Se 4.5 .1L75 2.8 2.7 Sr70570050 2.6 Th 3.6 .4L 22 1.0 4.0 U 4.5 .2L 43 2.4 3.1 V30330020 2.4 Y1557010 1.7 Zn $1,100$ 7.1 $18,000$ 150 7.5 Zr15310015 2.0	Sb	2.1	.1L	37	.8	4.0	
Se 4.5 .1L 75 2.8 2.7 Sr 70 5 700 50 2.6 Th 3.6 .4L 22 1.0 4.0 U 4.5 .2L 43 2.4 3.1 V 30 3 300 20 2.4 Y 15 5 70 10 1.7 Zn 1,100 7.1 18,000 150 7.5 Zr 15 3 100 15 2.0	Sc	5	11	15	5	2.1	
Sr705700502.6Th 3.6 .4L22 1.0 4.0 U 4.5 .2L 43 2.4 3.1 V 30 3 300 20 2.4 Y 15 5 70 10 1.7 Zn $1,100$ 7.1 $18,000$ 150 7.5 Zr 15 3 100 15 2.0	Se	4.5	.1L	75	2.8	2.7	
In 3.0 .4L 22 1.0 4.0 U 4.5 .2L 43 2.4 3.1 V 30 3 300 20 2.4 Y 15 5 70 10 1.7 Zn 1,100 7.1 18,000 150 7.5 Zr 15 3 100 15 2.0	35 76	/0	5	700	50	2.6	
U4.5.2L432.43.1V303300202.4Y15570101.7Zn1,1007.118,0001507.5Zr153100152.0	11	3.0	.4L	Z2	1.0	4.0	
V 30 3 300 20 2.4 Y 15 5 70 10 1.7 Zn 1,100 7.1 18,000 150 7.5 Zr 15 3 100 15 2.0	U	4.5	.2L	43	2.4	3.1	
T 15 5 70 10 1.7 Zn 1,100 7.1 18,000 150 7.5 Zr 15 3 100 15 2.0	¥.	30	3	300	20	2.4	
Zr 15 3 100 150 7.5 Zr 15 3 100 15 2.0	1 7e	15	5	70	10	1.7	
L, 10 3 100 15 2.0	211 7 m	1,100	/•1	100	150	7.5	
	41	10	3	100	15	2.0	

[All analyses except geometric deviations are in percent or parts per million and are reported on a whole-coal basis. L, less than the value shown]
Table 18.	Arithmetic mean, observed range, and geometric mean of proximate
	and ultimate analyses, heat of combustion, and forms of sulfur for
	114 Pennsylvanian-age coal samples from the Illinois Basin

[Data are calculated from Gluskoter and others (1977, Table 8). All values are in percent except kcal/kg and Btu/lb and are reported on an as-received basis. Kcal/kg = 0.556 (Btu/lb)]

	• • • • • • • •	Obser	Observed range			
Element	Arithmetic mean	Minimum	Maximum	Geometric mean		
	Proxima	te and ultimate	analyses			
Moisture	9.4	0.5	18	8.1		
Volatile matter	37	27	43	37		
Fixed carbon	45	37	57	45		
Ash	10	4.4	20	10		
Hydrogen	5.6	4.6	_6.4	5.5		
Carbon	64	57	11	64		
Nitrogen	1.2	•//	1.8	1.2		
Uxygen	10	4.0	24	15		
		Heat of combusti	on	J•1		
Kcal/kg Btu/lb	6,410 11,520	5,550 9,985	7,650 13,760	6,490 11,670		
		Forms of sulfur	•	<u></u>		
Sulfate	0.09	0.01	1.0	0.05		
Pyritic	1.8	.26	4.2	1.7		
Organic	1.4	.34	2.8	1.3		

Table 19.Arithmetic mean, observed range, and geometric mean of 35 elementsIn 114 Pennsylvanian-age coal samples from the Illinois Basin

[Data are from Gluskoter and others (1977, table 8). All analyses are in percent or parts per million and are reported on moisture-free basis. L = less than the value shown]

		Observ	ed range	
Element	Arithmetic mean	Minimum	Maximum	Geometric mean
		Percent		
Si	2.4	0.58	4.7	2.3
Al	1.2	.43	3.0	1.2
Ca	.67	.01	2.7	.51
Mg	.05	.01	.17	.05
Na	.05	.004	.2	.03
K	.17	.04		.16
re Ti	.06	•45 •02	4.1	.06
	Р	arts per million	1	
Ag	0.03	0.02	0.08	0.03
As	14	1.0	120	7.4
B	110	12	230	98
Ba	100	5.0	750	75
Be	1.7	.5	4.0	1.6
Cd	2.2	.11	65	.59
Co	7.3	2.0	34	6.0
Cr	18	4	60	16
Cu	14	5	44	13
F	67	29	140	63
Ga	3.2	.8	10	3.0
Ge	6.9	1.0L	43	4.8
Hg	.2	.03	1.6	.16
Mn	53	6.0	210	40
Mo	8.1	.3L	29	6.2
Nî	21	7.6	68	19
Pb	32	.8L	220	15
Sb	1.3	.1	8.9	.81
Sc	2.7	1.2	7.7	2.5
Se	2.2	.4	7.7	2.0
Sr	35	10L	130	30
Th	2.1	.71	5.1	1.9
U	1.5	.31L	4.6	1.3
V	32	11	90	29
Yb	.56	.27	1.5	.53
Zn	250	10	5,300	87
Zr	47	12	130	41

Table 20. Arithmetic mean, observed range, geometric mean, and geometric deviation of proximate and ultimate analyses, and heat of combustion, forms of sulfur, and ash-fusion temperatures of 44 coal samples from the Upper Cretaceous Williams Fork Formation, Yampa coal field, northwestern Colorado

[From Hildebrand and others (1981, table 7a). All values are in percent except kcal/kg, Btu/lb, and ash-fusion temperatures, and are reported on an as-received basis. Kcal/kg = 0.556 x (Btu/lb) $F = (^{\circ}C \times 1.8) + 32$. L, less than the value shown]

		Obser	ved range		
Element	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation
		Proximate and	ultimate analy	5es	<u>, </u>
Moisture Volatile	11.1	5.7	17.1	10.6	1.3
matter Fixed	34.9	26.1	40.3	34.8	1.1
carbon	44.4	33.2	52.3	44.2	1.1
Ash	9.6	3.0	32.7	8.4	1.7
Hydrogen	5.6	3.9	7.1	5.6	1.1
Carbon	60.9	44.0	71.0	60.7	1.1
Nitrogen	1.3	.5	1.9	1.2	1.4
Uxygen	22.0	16.0	27.2	21.8	1.2
Sultur	.6	.3	3.1	•6	1.5
		Heat of	f combustion		
Kcal/kg Btu/lb	5,930 10,670	4,170 7,500	6,920 12,440	5,910 10,630	1.1 1.1
		Forms	of sulfur	·····	· · · · · · · · · · · · · · · · · · ·
Sulfate	0.01	0.01L	0.04	0.01	1.4
Pyritic	.18	.01	2.18	.10	2.9
Organic	.45	.16	2.27	.40	1.6
		Ash-fusion	temperatures (C	°)	
Initial		, <u>, , , , , , , , , , , , , , , , , , </u>			
tion	1,315	1,070	1,600+	1,310	1.1
Softening	1,360	1,095	1,600+	1,350	1.1
Fluid	1,390	1,115	1,600+	1,385	1.1

Table 21. Arithmetic mean, observed range, geometric mean, and geometric deviation of 37 elements in 63 coal samples from the Williams Fork Formation, Yampa coal field, northwestern Colorado

[From Hildebrand and others (1981, table 9a). All analyses are in percent or parts per million and are reported on a whole-coal basis. L, less than the value shown]

1

		Obser	ved range			
Element	Arithmetic mean	Minimum	Maximum	Geometric mean	Geometric deviation	
		Р	ercent			
Si	3.1	0.34	13	2.2	2.3	
Al Ca	1.4 41	08.9 10	9.3	1.2	1.5	
Mg	.079	.011	.31	.066	1.8	
Na	.055	.005	.20	.035	2.6	
K Fo	•11	.003	./2	-Ubl 25	2.9	
Ti	.059	.018	.23	.052	1.6	
		Parts p	er million			
As	1.0	0,2	6.0	0.5	3.3	
В	100	20	300	70	1.7	
Ba	200	30	1,000	150	1.9	
Cd	.12	.15 .04L	•60	.06	3,3	
Co	1.5	.1L	7	1	2.1	
Cr Cu	5	.1L 2 3	30	3	2.4	
F	125	20L	740	95	2.1	
Ga	5	1	20	3	1.8	
Hg	.06	.01	.29	.04	2.3	
L1 Ma	10	1.5	52 210	8.4 17	1.9	
Mo	.7	.2	3	.3	2.9	
Nb	3	.7L	15	2	2.2	
Ni	2	.7	10	1.5	2.0	
P	520	45L	2,000	120	5.6	
PD Sh	0.U 3	1.5	33	4.4	2.2	
Sc	1.5	.7	10	1.5	1.9	
Se	1.0	.1L	6.4	•9	1.6	
36 Th	100	15	500	100	2.0	
Ű	1.2	./	15	1.0	3.n 1.8	
V	10	3	50	7	1.9	
Y Yb	7	1	20	5	2.0	
Zn	11	1.4	د 100	•5 7.3	2.5	
Zr	30	7	200	20	1.9	
<u></u>						

Table 22. Classification of coals by rank¹

[from American Society for Testing Materials, 1978, Table 1]

			Fixed carbon limits, per cent (Dry, mineral-matter- (free basis)		Volatile matter limits, per cent (Dry, mineral-matter- free basis)		Calorific value limits Btu per pound (Moist, ² mineral-matter- free basis)		
Class	Group		Equal or greater than	Less than	Equal or greater than	al or ater Less an than	Equal or greater than	Less than	Agglomerating character
I. Anthracitic	1. 2. 3.	Meta-anthracite	98 92 86	98 92	2 8	? 8 14	••••	••••	nonagglomerating ³
II. Bituminous	1. 2. 3.	Low-volatile bituminous coal Medium-volatile bituminous coal . High-volatile A bituminous coal .	78 69 	86 78 69	14 22 31	22 31	14,000 4	••••	commonly agglomerating ⁵
	4. 5.	High-volatile C bituminous coal .	••••	****	••••	• • • •	13,600 4 11,500 10,500	13,000 11,500	agglomerating
III. Subbituminous	1. 2. 3.	Subbituminous A coal	••••	••••	••••	••••	10,500 9,500 8,300	11,500 10,500 9,500	nonagglomerating
IV. Lignitic	1. 2.	Lignite A	••••	••••	••••		6,300	8,300 6,300	

¹This classification does not include a few coals, principally nonbanded varieties, which have unusual physical and chemical properties and which come within the limits of fixed carbon or calorific value of the high-volatile bituminous and subbituminous ranks. All of these coals either contain less than 48 per cent dry, mineral-matter-free fixed carbon or have more than 15,500 moist, mineral-matter-free British thermal units per pound. Whist refers to coal containing its natural inherent moisture but not including visible water on the surface of the coal. If agglomerating, classify in low-volatile group of the bituminous class.

"Coals having 69 percent or more fixed carbon on the dry, mineral-matter-free basis shall be classified according to fixed carbon, regardless of calorific value.

"It is recognized that nonagglomerating varieties may occur in these groups of the bituminous class, and notable exceptions exist in high-volatile C bituminous group.

	Btu/lb ^l (mmmf)	Carbon ²	Hydrogen ³	Mineral ⁴ matter	0xygen ⁵	Hc/Cc	0c/Cc	S _{pyrite} S _{organic}
Coal zones 6-9	11,930	63.0	4.3	22.6	7.2	0.81	0.086	5.1
Coal zone 5	11,820	54.6	3.6	32.6	5.3	.79	.073	7.4
Coal zone 4	12,110	60.1	4.0	26.3	6.6	.80	.082	6.7
Coal zone 3	11,990	60.6	4.1	27.4	5.1	.82	.063	6.7
Coal zone 2	12,080	59.1	4.1	27,5	6.3	.83	.081	6.6
Iowa Coal	12,040	59.8	4.0	26.5	6.7	.81	.083	6.7
Illinois Coal ⁶ (114 samples)	12,990	70.6	4.9	13.1	8.5	.84	.090	3.5
Yampa Field, Colo. (44 samples)	⁷ 11,910	68.5	4.8	11.8	12.9	.83	.14	.71

Table 23.	Calculated	Btu/1b,	hydrog	gen/carbon	and o	oxygen/	carbon	molecular	ratios	
	(moisture-	free bas	is) for	· Iowa Chei	rokee	Group	coal zo	ones 2, 3,	4, 5,	and
	6-9. Iowa	coal. II	linois	Basin coal	. and	i Yampa	field	Colorado	coal	

Parr formula; moist, mineral-matter-free Btu = [(as-received Btu - 50S)/100 - (1.08 Ash + 0.55S)] x 100.

- 2 Carbon content of organic matter corrected for carbonate carbon content by Ccalculated = Canalytical 12/40 Ca, where (12/40 calcium) replaces the (12/44 CO₂) in the British Standard Formula (see Given and Yarzab, 1978, equation 2). Ca is the amount of calcium assumed present as carbonate above a 0.5% non-carbonate calcium base level.
- 3 Hydrogen content of organic matter corrected for water of clays by H_{calculated} = H_{analytical} -0.014 Ash + 0.02 S_{pyritic} x 0.02 (44/40 Ca) + 0.014 SO₃ (modified from Leighton and Tomlinson, 1960), where 44/40 Ca replaces CO₂; calcium is defined as in 2 , and SO₃ is SO₃ in coal ash reported on a whole-coal basis.
- ⁴ Mineral matter = 1.13 ash + 0.8 (44/40 Ca) + 0.5 Spyritic 2.8 (Sash Sso4 (modified from King, Maries and Crossley (1936)). Calcium is defined as in ²; Sash and S_{SO4} represents sulfur as sulfate in the ash and in the coal respectively.
- ⁵ Oxygen content calculated by difference; $\Omega_{calculated} = 100$ (mineral matter + $C_{calculated} + H_{calculated} + N_{analytical} + S_{organic}$).
- ⁶ Data from Gluskoter and others (1977, table 8).
- ⁷ Data from Hildebrand and others (1981, table 7a).

Table 24. Depths from the surface (in meters) and the zinc/cadmium mole ratios for 40 Cherokee Group coal samples from five core holes in Wapello and Appanoose Counties, Iowa.

[Data from table 6, core hole number CP-10 is in Appanoose County; the other four core holes are in Wapello County. Because of data uncertainties, qualified cadmium values (L) and real cadmium values of 1.0 ppm or less were not used to calculate the Zn/Cd mole ratio. Table modified from Hatch and others (1976, table 5)]

Depth moles Zn Depth moles Zn Hole No. moles Cd Sample No. (meters) Sample No. (meters) Hole No. moles Cd CP-7 D176169 33.3 CP-21 50.9 D176191 144 D176170 45.5 (cont.) D176192 56.0 120 ---D176171 50.4 148 58.3 D176193 170 D176172 58.9 261 60.3 D176194 223 D176173 62.8 345 D176195 65.7 ---D176174 69.8 469 D176196 73.5 ----D176175 76.9 572 80.1 515 D176197 87.8 D176176 D176198 95.2 ___ ---D176177 98.7 96.9 D176199 ------D176200 107.8 548 CP-10 D176178 74.0 93 CP-32 7.5 D176179 79.6 D179847 134 ---85.1 D176180 D179848 123 76 18.0 D176181 93.6 91 35.7 D179849 ---D176182 109.9 54.1 445 D179850 ___ D176183 118.5 278 D176184 123.8 CP-28 D179851 17.0 108 ___ 129.5 D176185 29.5 D179852 115 ___ 142.2 46.9 D176186 468 D179853 430 55.2 D179854 ---CP-21 D176189 23.4 D179855 64.4 -------D176190 40.2 D179856 69.5 1100 _ _ _

Table modified from Hatch and others (1976, table 5)

