In situ Raman spectroscopic analysis of the lithiation and sodiation of antimony microparticles

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Table A1. Table of CHN results for sol-gel synthesised and commercial Sb.

	Sol-gel synthesised Sb (wt.%)	Commercial Sb (wt. %)
С	23.3	0.5
н	0.7	-
Ν	3.6	-

Table A2. Table of discharge, charge and irreversible capacity of sol-gel synthesised and commercial antimony vs. lithium.

	Sol-gel synthesised Sb capacities / mAhg ⁻¹			
Cycle number	Discharge	Charge	Irreversible	
1	1082	744	338	
2	685	640	45	
10	603	596	8	
60	566	564	2	
120	550	549	2	
	Commercial Sb capacities / mAhg ⁻¹			
Cycle number	Discharge	Charge	Irreversible	
1	853	622	231	
2	650	588	62	
10	597	563	35	
60	451	444	7	
120	265	262	3	

Table A3. Table of discharge, charge and irreversible capacity of sol-gel synthesised and commercial antimony vs. sodium.

	Sol-gel synthesised Sb capacities / mAhg ⁻¹			
Cycle number	Discharge	Cycle number	Irreversible	
1	722	600	122	
2	546	511	35	
10	513	504	10	
60	484	478	5	
120	470	465	5	
	Commercial Sb capacities / mAhg ⁻¹			
Cycle number	Discharge	Cycle number	Irreversible	
1	930	625	305	
2	619	575	45	
10	544	528	16	
60	118	114	4	
120	93	90	3	



Figure A1. *In situ* Raman spectroscopy data for commercial antimony vs. lithium over 1 charge/discharge cycle at $60\mu A g^{-1}$.



Figure A2. *In situ* Raman spectroscopy data for commercial antimony vs. sodium over 1 charge/discharge cycle at $60\mu A g^{-1}$.



Figure A3. *In situ* Raman spectroscopy data for sol-gel synthesised antimony vs. A) lithium and B) sodium, including carbon bands.