

Content

1		
2	1. Introduction of nanoparticle and fractal geometry.....	2
3	2. Fractal model for thermal conductivity of nanofluids	6
4	3. Fractal and Monte Carlo simulation on convective heat transfer of nanofluids	12
5	3.1 Formulation of Convective Heat Transfer model	12
6	3.2 Methodology for the fractal-Monte Carlo Technique	14
7	3.3 FMCHT model tests	16
8	4. Fractal modeling for critical heat flux of nanofluids	17
9	4.1 Fractal Model	17
10	4.2 FACHF model tests.....	19
11	5. Fractal Model for subcooled pool boiling of nanofluids	19
12	6. Fractal Aggregation of nanoparticles	21
13	7. Fractal analysis on yield stress property of nanoparticle aggregation	25
14	8. Discussion and Future Work	26
15	9. Conclusions.....	29
16	Appendix.....	30
17	References.....	34

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20 1. Introduction of nanoparticle and fractal geometry

21 Nanoparticles are ultrafine particles, in which the particle is thought as a small
22 object that behaves as a whole unit of its transport and properties. [Since the unique](#)
23 [properties of the nanoscale, nanoparticles have been extensively used in a variety of](#)
24 [applications, such as medicine, optics, electronics, manufacturing, materials, solar cells](#)
25 [and catalysts \[1-6\].](#) Fluids with suspended nanoparticles are termed as nanofluids by
26 [Choi and Eastman \[7\].](#) There is a growing interest in experimental, theoretical and
27 numerical investigations on yield stress property of nanoparticle aggregate and the
28 thermal conductivity of nanofluids from theoretical perspectives to engineering applied
29 science[8-11].

1 by scattering technique. A complete particle distribution function is essential for
2 evaluating the enhancement of thermal conductivity that produced by the coherent heat
3 transport between particles separated by thin liquid layers [49-52].

4 As claimed by some researchers, the uniform distribution assumption is not
5 capability to characterize most behaviours of nanoparticle aggregation [53, 54]. In
6 reality, particles in nanofluids normally have different sizes, and some particles may
7 contact each other to form clusters of different sizes [55, 56]. The theory of fractals aids
8 in analysing the response of practical systems besides examining the complex
9 geometries of nature. The nanoparticles in nanofluids have been proven to be fractal
10 objects and the distribution of nanoparticles in suspension exhibits the fractal property
11 [28, 55, 57-59]. The fractal dimensions of clustering CuO nanoparticles in water with
12 different volume concentration are calculated based on Eq. (1) is 1.73 and 1.76 [28].
13 Havlin and Ben-Avraham [58] indicated that the size distributions of nanoparticles and
14 nanoparticles in suspension were shown certain degree of self-similarity. This means
15 that the fractal theory may be used to predict transport property of nanofluids. Wang et
16 al. [28] experimentally found that the size of nanoparticles and its clusters follows
17 fractal distribution, and developed a fractal model for predicting the effective thermal
18 conductivity of liquid with nanoparticle inclusion. By means of fractal geometry, Xu et
19 al. [57] proposed a model for heat convection due to the Brownian motion of
20 nanoparticles, Xiao et al. [59] further improved Xu's model [57] and obtained a novel
21 form of thermal conductivity of nanofluids with Brownian motion effect.

22 The fractal theory is developed in the 1980's [60] and is widely applied to many
23 fields [61-67]. The mass of a fractal object (M) is related to measured scale ε through
24 fractal dimension (D):

$$25 \quad M(\varepsilon) \sim \varepsilon^D \quad (1)$$

1 Besides the exact self-similar fractals, many objects in nature are statistically self-
 2 similar fractals. The former exhibit the self-similarity over an infinite range of length
 3 scales, however, the latter usually exhibit the self-similarity in some average sense and
 4 over a certain local range of length scales (called prefractals).

5 The size distribution of nanoparticles in nanofluids, analogous to pores in fractal
 6 porous media, follows the fractal power law [28]. Thus, the number of particles whose
 7 sizes are within the infinitesimal range from λ to $\lambda + d\lambda$ is [68, 69]:

$$8 \quad dN = -D \frac{\lambda_{\max}^D}{\lambda^{D+1}} d\lambda \quad (2)$$

9 and the probability density function $f(\lambda)$ for particles is given by [69]

$$10 \quad f(\lambda) = D \frac{\lambda_{\min}^D}{\lambda^{D+1}} \quad \lambda_{\min} \leq \lambda \leq \lambda_{\max} \quad (3)$$

11 where λ is the particle diameter, λ_{\min} and λ_{\max} are the minimum and maximum
 12 diameters of nanoparticles, respectively. The fractal dimension D can be measured
 13 by Box-counting method or be determined by [70]:

$$14 \quad D = d_E - \frac{\ln \phi_p}{\ln \xi} \quad (4)$$

15 where $d_E=2$ in two dimensions, ϕ_p is the volume fraction of (primary) nanoparticles,
 16 ratio $\xi = \lambda_{\min}/\lambda_{\max}$ is assigned for simplification of formula writing. However, the
 17 limitation of fractal theory for nanofluids and nanoparticle aggregation is that the fractal
 18 theory must satisfy $\xi \leq 10^{-2}$. So, the fractal theory can be used to analyze the
 19 characters of nanofluids and nanoparticle aggregation.

20 The average diameter λ_{av} of nanoparticles can be decided based on the
 21 statistical property of fractal media, i.e:

$$22 \quad \lambda_{av} = \int_{\lambda_{\min}}^{\lambda_{\max}} \lambda f(\lambda) d\lambda = \frac{D \lambda_{\max}}{D-1} \xi \quad (5)$$

1 Rewriting Eq.(5), the minimum and maximum nanoparticle diameter can be
2 respectively obtained. From Eqs. (4) and (5), the ratio ξ is an important parameter
3 for analysing the properties of nanoparticles by means of fractal geometry. Once ξ ,
4 ϕ_p and λ_{av} are measured, the D can be calculated by Eq. (4). Feng et al. [71]
5 compared the values of D calculated by Eq. (4) with the experimental data by several
6 investigators and found that the ratio of $\xi = 10^{-3}$ best fits the experimental data for
7 different porous media. Similar conclusions can be also found in other applications of
8 fractal theory [66, 72-74]. Considering that the pores in porous media are analogous to
9 the particles in nanofluids, Eq. (4) was used to describe the particles in nanofluids and
10 the ratio of $\xi = 10^{-3}$ was applied to analyse the heat conduction of nanofluids [57, 59,
11 75].

12

13 **2. Fractal model for thermal conductivity of nanofluids**

14 Two portions were normally considered as the main contribution to the heat
15 transfer of nanofluids; there are heat transferred by stationary nanoparticles and the heat
16 convection caused by moving nanoparticles, and the total effective thermal conductivity
17 k_{eff} is hence expressed as:

$$18 \qquad k_{eff} = k_s + k_c \qquad (6)$$

19 where k_c is the thermal conductivity by heat convection caused by Brownian motion of
20 nanoparticles and k_s is the thermal conductivity by stationary nanoparticles in the liquid,
21 which is simulated by the Maxwell-Garnett (MG) model. For the spherical-particle
22 suspensions, the MG model gives [76]:

$$23 \qquad \frac{k_s}{k_f} = \frac{k_p + 2k_f - 2\phi_p(k_f - k_p)}{k_p + 2k_f + \phi_p(k_f - k_p)} \qquad (7)$$

1 where k_p and k_f are the thermal conductivities of particles and fluid (host medium),
2 respectively. The MG model is applicable to liquid with suspension of low
3 concentration particle, and can be fit well with experimental results for randomly
4 distributed and dilute components included in homogeneous host media, the suspended
5 particles are thought as isolated in liquid and no **interaction exists among them** [28].

6 Recent **experimental studies have** shown that molecules **those close to a solid**
7 **surface in normal liquids organize into layered structures as similar to a solid** [77].
8 Furthermore, **there is an evidence** that such an organized solid-like structure of a liquid
9 at the surface is a governing factor in heat conduction from a solid wall to an adjacent
10 liquid [78]. Therefore, **Choi and coworkers** [79] **postulated** that this organized
11 solid/liquid interfacial shell makes the transport of energy across the interface
12 **effectively**. Because the **interfacial shells exist between** the nanoparticles and the liquid
13 matrix, both the interfacial shell and the nanoparticle can be regarded as a “complex
14 nanoparticle”. Therefore, the nanofluid system should be regarded as the complex
15 nanoparticles dispersed in the fluid.

16 Xue [80] **proposed a formula** of calculating the thermal conductivity component
17 along an axis of a complex rotational elliptical particle [48]. He assumed that the
18 complex rotational nanoparticle was composed of a rotational elliptical nanoparticle of
19 thermal conductivity k_p with half radii of $(a, b, b)^1$ and an elliptical shell of thermal
20 conductivity k_{int} with **a thickness of l** . The thermal conductivity component along a and
21 b axes of the complex rotational elliptical particle is expressed as:

$$22 \quad k_{c,a} = k_{int} \frac{(1-B)k_{int} + Bk_p + \beta(1-B)(k_p - k_{int})}{(1-B)k_{int} + Bk_p - B\beta(k_p - k_{int})} \quad (8)$$

¹ Three axis of elliptical nanoparticle is (x, y, z) , here the half radii $y=z=b$ is assumed for simplification

$$k_{c,b} = k_{int} \frac{(1+B)k_{int} + (1-B)k_p + (1+B)\lambda'(k_p - k_{int})}{(1+B)k_{int} + (1-B)k_p - (1-B)\lambda'(k_p - k_{int})} \quad (9)$$

where $\beta = \frac{ab^2}{(a+l)(b+l)^2}$, B is the depolarization factor component of the rotational

elliptical particle along the long axis (a axis), $B = \frac{ab^2}{2} \int_0^\infty \frac{du}{(u+a^2)(u+b^2)\sqrt{u+a^2}}$.

According to the average theory, the equation for the effective thermal conductivity of the complex rotational elliptical particles dispersed nanofluid is proposed as:

$$9 \left(1 - \frac{\phi_p}{\lambda'}\right) \frac{k_{eff} - k_f}{2k_{eff} + k_f} + \frac{\phi_p}{\beta} \left[\frac{k_{eff} - k_{c,a}}{k_{eff} + B(k_{c,a} - k_{eff})} + \frac{4(k_{eff} - k_{c,b})}{2k_{eff} + (1-B)(k_{c,b} - k_{eff})} \right] = 0 \quad (10)$$

When the dispersed particles are spheres, Eq. (10) reduces to [81]

$$9 \left(1 - \frac{\phi_p}{\beta}\right) \frac{k_{eff} - k_f}{2k_{eff} + k_f} + \frac{\phi_p}{\beta} \frac{k_{eff} - k_c}{2k_{eff} + k_c} = 0 \quad (11)$$

where k_c is the thermal conductivity of a complex sphere, which is composed of a spherical nanoparticle of thermal conductivity k_p with radius of r and an shell of thermal conductivity k_{int} with a thickness of l ,

$$k_c = k_{int} \frac{2k_{int} + k_p + 2\beta(k_p - k_{int})}{2k_{int} + k_p - \beta(k_p - k_{int})} \quad (12)$$

where $\beta = \frac{r^3}{(r+l)^3}$. It is demonstrated that the theoretical results on the effective

thermal conductivity of carbon oil-nanotube nanofluid and water- Al_2O_3 nanofluid are in good agreement with the experimental data. Furthermore, they interpreted the anomalous enhancement of the effective thermal conductivity of carbon oil-nanotube nanofluid and its nonlinearity with carbon nanotube loadings using Eq. (10).

It is found that the Brownian motion of nanoparticles at the molecular and

1 nanoscale level is a key mechanism that governs the thermal behavior of nanofluids
 2 [41]. The following context presents the second term k_c in Eq. (6) based on fractal
 3 theory [57, 82]. The heat transfer coefficient is defined as:

$$4 \quad h = \frac{Nu \cdot k_f}{\lambda} \quad (13)$$

5 where Nu is Nusselt number for liquid flowing around a sphere, and
 6 $Nu = 2.0 + 0.5 \text{Re} \cdot \text{Pr} + O(\text{Re}^2 \cdot \text{Pr}^2)$ [83], in which Pr is the Prandtl number of fluids and
 7 Re is the Reynolds number. Nu is related to thermal diffusion coefficient, kinematical
 8 viscosity, absolute viscosity, density and specific heat of fluid, as well as the diameter
 9 and velocity of particle. Thus, the heat transferred by convection q_λ for a single
 10 nanoparticle moving in liquids is calculated by:

$$11 \quad q_\lambda = h a_\lambda (T_p - T_f) \quad (14)$$

12 where T_p and T_f are the temperatures of particle and liquid, respectively, a_λ is the
 13 surface area of a particle with diameter is λ .

14 Considering the fractal distribution of nanoparticles in nanofluids and the
 15 assumption of local thermal equilibrium, the heat transferred by convection of all
 16 particles q_c is given as [57]

$$17 \quad q_c = - \int_{\lambda_{\min}}^{\lambda_{\max}} q_\lambda dN = - \int_{\lambda_{\min}}^{\lambda_{\max}} a_\lambda h \delta_T \frac{\Delta T}{\delta_T} dN \quad (15)$$

18 where δ_T is the thickness of the thermal boundary layer of heat convection caused by
 19 the nanoparticles' Brownian motion and $\Delta T = T_p - T_f$. By postulating that the nanolayer
 20 of the ordered liquid molecules act as a hydrodynamic boundary layer, and that three
 21 layers exist at the interface, the hydrodynamic boundary layer δ is given as $\delta = 3d_f$
 22 and $\delta_T = \delta / \text{Pr}$ (d_f is the diameter of the base liquid molecule) [41, 77].

23 The equivalent thermal conductivity contributed by heat convection can be

1 calculated by [57]:

$$2 \quad k_c = -\frac{q_c}{A} \frac{\delta_T}{\Delta T} = \frac{1}{A} \int_{\lambda_{\min}}^{\lambda_{\max}} a_\lambda h \delta_T dN \quad (16)$$

3 where $A = \int_{\lambda_{\min}}^{\lambda_{\max}} \pi \lambda^2 dN$ is the total area of nanoparticles. The dimensionless expression
4 of k_c is

$$5 \quad k_c^+ = \frac{k_c}{k_f} = \frac{1}{k_f A} \int_{\lambda_{\min}}^{\lambda_{\max}} h a_\lambda \delta_T dN \quad (17)$$

6 Combing Eqs. (14) and (2) with Eq. (17), we can obtain the fractal model for
7 the dimensionless thermal conductivity caused by heat convection due to Brownian
8 motion as [57]:

$$9 \quad k_c^+ = c \frac{Nu \cdot d_f}{Pr} \cdot \frac{2-D}{1-D} \cdot \frac{\lambda_{\max}^{1-D} - \lambda_{\min}^{1-D}}{\lambda_{\max}^{2-D} - \lambda_{\min}^{2-D}} \quad (18)$$

10 where c is an empirical constant, which is related to the thickness of the thermal
11 boundary layer ($\delta_T = \delta/Pr$). In other word, c is primarily related to the property of the
12 host liquid, and is independent of particles [57]. The Nusselt number is taken to be 2.0
13 [57]. Further considering the fractals relationship of particle size distribution to fractal
14 dimension (Eqs. (5)), Eq. (18) can be rewritten as [57]:

$$15 \quad k_c^+ = c \frac{Nu \cdot d_f}{Pr} \frac{D(2-D)}{(1-D)^2 \lambda_{av}} \frac{[\xi^{D-1} - 1]^2}{\xi^{D-2} - 1} \quad (19)$$

16 From Eq. (19), thermal conductivity caused by heat convection due to Brownian
17 motion is a function of fluid property, temperature, average nanoparticle size, minimum
18 and maximum nanoparticle sizes and fractal dimension, and it especially shows that it
19 is inversely proportional to λ_{av} .

20 Thus, the total dimensionless effective thermal conductivity of nanofluids is
21 obtained [57]

$$\frac{k_{eff}}{k_f} = k_s^+ + k_c^+ = \frac{k_p + 2k_f - 2\phi_p(k_f - k_p)}{k_p + 2k_f + \phi_p(k_f - k_p)} + c \frac{N_u \cdot d_f}{Pr} \frac{D(2-D)}{(1-D)^2 \lambda_{av}} \frac{[\xi^{D-1} - 1]^2}{\xi^{D-2} - 1} \quad (20)$$

where $k_s^+ = k_s/k_f$. From Eq. (20), the total dimensionless effective thermal conductivity varies with the Prandtl number, the molecule diameter of fluids, the volume fraction of nanoparticles, fractal dimension and sizes of nanoparticles. Since $\xi = 10^{-3}$ is assigned from above analysis, once the ϕ_p is measured, the D can be calculated by Eq. (4), the thermal conductivity is then predicted from Eq. (20) as long as the parameters Pr , d_f , k_f and k_p are known. From Eq. (20), it can be seen that the analytical model for effective thermal conductivity of nanofluids is derived while taking into account the effect of heat convection caused by Brownian motion of nanoparticle based on the fractal geometry theory. The fractal model contains less empirical constants, which are normally required in classical approach, and every parameter in Eq. (20) has clear physical meaning. Besides the analytical method, the present fractal technique might have the potential in the analysis of other transport properties such as optical and electrical properties of nanofluids. So, the proposed fractal technique may provide us with a new approach in addition to the analytical and other numerical methods.

As shown in Fig. 1, the predictions made by the effective thermal conductivity formulation, Eq. (20), estimate the experimental data of nanofluids well [57] (water and ethylene glycol (EG) are host liquids respectively with Al_2O_3 and CuO suspensions [13, 24-26, 84]). The effects of particle volume fraction and particle diameter on thermal conductivity are also demonstrated in Fig. 1. The thermal conductivity increases with the increasing of particle volume fraction and temperature, and decreases with particle diameter. Figure 2 presents the contribution weight of convection to the total thermal conductivity versus volume fraction and diameter of nanoparticles [57].

1 Clearly, there is a critical particle volume fraction ($\phi_p = 0.126$) for the maximum
2 contribution from convection for different particle size. However in other works, the
3 critical volume fraction is independent of particle diameter. This is very important for
4 designing and preparing industrial applications liquids of high thermal conductivity.

5 Particles in nanofluids may randomly aggregate with each other to form various
6 shape structures, and this has been confirmed experimentally [85]. Because chain
7 structures allow more heat to transport than other shape structures along the direction
8 of heat flux, the heat conduction could be enhanced due to the particle aggregation of
9 chain shape [53]. Recently, Wei et al.[86] discussed the influence of random
10 aggregation shape of nanoparticles for the effective thermal conductivity and modified
11 the empirical shape factor F (proposed by Hamilton and Crosser [85] to improve
12 Maxwell equation [87]) based on fractal theory, the F is expressed as[86]:

$$13 \quad F = \frac{3D}{D-1} \frac{3-D}{2-D} \frac{\phi_p}{1-\phi_p} \left(\frac{\phi_p^{D-2}}{\phi_p^{3-D}} - 1 \right) \quad (21)$$

14 In Hamilton and Crosser's model [85], F is constant for same shape particles ($F=6$
15 for ellipse and $F=3$ for sphere). However, it is observed (Eq. (21)) that F is a function
16 of fractal dimension and concentration, and F increases with the increasing of particle
17 concentration. Wei et al.[86] claimed that the shape of aggregation gradually grows to
18 chain with the increasing concentration when considering the fractal distribution of
19 nanoparticle aggregation. However, most aggregation shapes are circles when $F < 6$.

20

21 **3. Fractal and Monte Carlo simulation on convective heat transfer of nanofluids**

22 **3.1 Formulation of Convective Heat Transfer model**

23 Figure 3 shows Al_2O_3 nanoparticles dispersed in distilled water, in which the

1 particles are spheres with an average diameter of 47 nm in a range from 10 nm to 100
 2 nm, as evaluated from Transmission Electron Microscopy (TEM) images [88]. The
 3 number of nanoparticles is very large in nanofluids, and their sizes are different.

4 It is demonstrated that the existence of nanoparticles enhances the single-phase
 5 convection heat transfer since the nanoparticles moving in fluids carry energy and the
 6 heat exchange may occur between hot and cold regions [75, 89]. Recently, Xiao et al.
 7 [90] developed a model for the total heat flux q_t from the convective heat transfer of
 8 nanofluids (CHTNs)

$$9 \quad q_t = \left[\frac{k_f}{\sum_{i=1}^J \lambda_i^2} \sum_{i=1}^J \lambda_i \left(2 + \frac{3}{2\alpha} \sqrt{\frac{2k_B T}{\pi \rho_p \lambda_i}} \right) + h_c \right] \Delta T \quad (22)$$

10 where $k_B = 1.38 \times 10^{-23} J/K$ is the Boltzmann constant, ρ_p is the density of
 11 nanoparticle, and λ_i is the diameter of nanoparticle, and h_c is the average heat
 12 transfer coefficient by natural convection from the base fluids. The first term of the
 13 right side of Eq. (22) indicates the heat transferred by heat convection caused by
 14 Brownian motion of nanoparticles, and the second term indicates the contribution by
 15 natural convection from the base fluids. The h_c can be given by [91]

$$16 \quad h_c = \begin{cases} 0.14 \rho_f c_p \left[\frac{\gamma_1 g (T_w - T_f) \alpha^2}{\nu} \right]^{1/3}, & 2 \times 10^7 < Ra < 3 \times 10^{10} \\ 0.54 \rho_f c_p \left[\frac{\gamma_1 g (T_w - T_f) \alpha^3}{\sqrt{A_h} \nu} \right]^{1/4}, & 10^5 < Ra < 2 \times 10^7 \end{cases} \quad (23)$$

17 where A_h is the area of heating surface, α is thermal diffusivity of fluid, ν is
 18 kinematic viscosity, Ra is Rayleigh number, ρ_f is the base fluids density, c_p is
 19 specific heat at constant pressure, γ_1 is volumetric thermal expansion coefficient of

1 liquid, g is the gravity acceleration, T_w is wall temperature.

2 Equation (22) takes account of the effect of convection caused by the Brownian
3 motion, and relates the total heat flux from CHTNs to the parameters of nanofluid, such
4 as the nanoparticles sizes, the thermal conductivities of base fluids, and the wall
5 superheat as well as fluid properties. In Eq. (22) ΔT is a variable. When a pure liquid
6 was boiled over heating surfaces, there is a significant effect of surface characteristics
7 on boiling performance and mechanisms [92, 93]. Actually, the transport properties of
8 the heater affect the extent of the thermal interaction among the cavities, causing
9 activation and deactivation of individual cavities.

10 **3.2 Methodology for the *fractal-Monte Carlo Technique***

11 The cumulative probability R (0~1) in the range of $\lambda_{\min} \sim \lambda_i$ can be obtained
12 from **integrating the** probability density function (rewriting Eq. (3) as
13 $f(\lambda_i) = D\lambda_{\min}^D \lambda_i^{-(D+1)}$) [94]

$$14 \quad R_i(\lambda_i) = \int_{\lambda_{\min}}^{\lambda_i} f(\lambda_i) d\lambda_i = \int_{\lambda_{\min}}^{\lambda_i} D\lambda_{\min}^D \lambda_i^{-(D+1)} d\lambda_i = 1 - \left(\frac{\lambda_{\min}}{\lambda_{\max}} \right)^D \quad (24)$$

15 Eq. (24) indicates that $R=0$ as $\lambda \rightarrow \lambda_{\min}$ and $R \approx 1$ as $\lambda \rightarrow \lambda_{\max}$. For the i^{th}
16 nanoparticles chosen randomly, the diameter λ_i is expressed as[94]

$$17 \quad \lambda_i = \frac{\lambda_{\min}}{(1-R_i)^{1/D}} = \frac{\lambda_{\min}}{\lambda_{\max}} \frac{\lambda_{\max}}{(1-R_i)^{1/D}} \quad (25)$$

18 where $i = 1, 2, 3, \dots, J$ (J is the total number of Monte Carlo simulations in one run for a
19 given concentration). Equation (25) presents an explicit probability model for
20 nanoparticles size distribution in the present simulation, also denotes that since R_i is a
21 random number of 0 ~ 1 produced by computer, the nanoparticle diameter λ_i is
22 determined randomly, which simulates the randomness and fractal distribution of

1 nanoparticles size.

2 The average diameter λ'_{av} of the all nanoparticles calculated in the presented
3 Monte Carlo simulations can be written as [90]

$$4 \quad \lambda'_{av} = \frac{1}{J} \sum_{i=1}^J \lambda_i \quad (26)$$

5 The algorithm for determination of the CHTNs is summarized as below:

6 i). Given a ϕ_p and λ_{av} .

7 ii). Find D , λ_{min} , and λ_{max} from Eq. (4), respectively.

8 iii). Produce a random number R_i of $0 \sim 1$ by computer.

9 iv). Calculate λ_i from Eq. (25).

10 iv). If $\lambda_i > \lambda_{max}$ or $\lambda_i < \lambda_{min}$, return to procedure iii, otherwise continue to the next
11 procedure.

12 vi). Find λ'_{av} by Eq. (26).

13 vii). Find the total heat flux from CHTNs q_t from Eq. (22).

14 Procedures iv-vii are repeated for calculation of total heat flux from CHTNs until
15 a converged value is obtained at a given concentration. The convergence criterion is
16 that when the following condition is satisfied, i.e.: [90]

$$17 \quad \lambda'_{av} = \frac{1}{J} \sum_{i=1}^J \lambda_i \geq \lambda_{av} \quad (27)$$

18 and then stop the simulation and record the final q_t and the total number (J) in one run
19 for a given concentration. In Eq. (27), λ_{av} is calculated from Eq. (5). If the
20 converged heat flux from CHTNs is obtained in one run, set the heat flux from CHTNs
21 as $q_t^{(n)}$ ($n= 1, 2, 3, \dots, Y$). Then, the average q_t is calculated by [90]

$$\langle q_t \rangle = \frac{1}{Y} \sum_{n=1}^Y q_t^{(n)} \quad (28)$$

where Y is the total number of runs for a given volumetric nanoparticle concentration.

The variance ν is defined as [90, 94]

$$\nu = \sqrt{\langle q_t^2 \rangle - \langle q_t \rangle^2}, \text{ where } \langle q_t^2 \rangle = \frac{1}{Y} \sum_{n=1}^Y q_t^{2(n)} \quad (29)$$

Above formulas presented a simple algorithm by combining the fractal geometry and Monte Carlo technique for the total heat flux from CHTNs, hereafter referred to as the FMCHT model. This model has characters of both analytical and numerical solutions, in which the characterizations of randomness and fractal distribution of nanoparticle sizes are included.

10

11 **3.3 FMCHT model tests**

12 The comparisons between the experimental results [95, 96] and FMCHT model
 13 predictions for heat flux from convective heat transfer of water-Al₂O₃ nanofluids are
 14 plotted in Fig.4a ($\lambda_{av}=47\text{nm}$ and $\phi_p=1\%$) and in Fig. 4b ($\lambda_{av}=38\text{nm}$ and $\phi_p=0.1\%$).

15 It should be also noted from Fig 4 [97] that the heat flux from convective heat transfer
 16 increases with ΔT , which may be interpreted that the higher temperature may cause
 17 the stronger Brownian motion, thus may produce more contribution to the heat transfer
 18 from convection. Figure 5 shows the heat flux from convective heat transfer of CuO
 19 nanofluids versus the average diameter of nanoparticles at $\phi_p=0.2\%$. The heat flux at
 20 natural convection stage decreases when the nanoparticles average size increases. This
 21 can be explained by the theory of Brownian motion, smaller average size of
 22 nanoparticles in the fluids can result in higher velocity of nanoparticles' Brownian
 23 motion, and thus the heat transferred by heat convection is improved, which is

1 consistent with the practical physical phenomena.

2 The Monte Carlo technique combined with fractal geometry theory is successfully
3 applied to predict the CHTNs, in which the convection caused by the Brownian motion
4 and the fractal distribution of nanoparticle sizes are taken into account. The CHTNs is
5 negatively correlated with the average size of nanoparticles but positively correlated
6 with the wall superheat. Besides the analytical and numerical methods, the above
7 referred techniques also have the potential in the analysis of the transport properties
8 such as magnetic and electrical properties of nanofluids.

9

10 **4. Fractal modeling for critical heat flux of nanofluids**

11 **4.1 Fractal Model**

12 The critical heat flux (CHF) of nucleate pool boiling heat transfer in Al_2O_3
13 nanofluids is pictured in Fig.6 [98]. For the base fluids, it is generally recognized that
14 the main mechanism contributing to nucleate boiling heat transfer is the bubble
15 generation and departure from the active cavity on the superheated surface in CHF
16 region. Thus there are two main mechanisms contributing to nucleate pool boiling heat
17 transfer of nanofluids in the CHF region: one is the heat $q_{t,c}$ transferred by the heat
18 convection caused by the Brownian motion of nanoparticles, and the other is the heat
19 $q_{b,c}$ transferred by the bubbles generation and departure from the base fluids in the
20 CHF region. In the CHF region, the total heat flux q_t of nucleate pool boiling heat
21 transfer of nanofluids can be expressed as [97]

$$22 \quad q_t = q_{t,c} + q_{b,c} \quad (30)$$

23 in which the $q_{t,c}$ is calculated by

$$q_{i,c} = \frac{D^{0.25}(4-2D)k_f[k_p(1+2\phi_p)+2k_f(1-\phi_p)](\xi^{1-D}-1)\Delta T}{(D-1)(4-D)^{0.25}[k_p(1-\phi_p)+k_f(2+\phi_p)](1-\xi^{2-D})\lambda_{av}} \quad (31)$$

The small diameter of nanoparticle may cause the increase of the velocity of nanoparticle, leading to more heat transferred by nanoparticles moving in nanofluids, as expressed in Eq. (31) that $q_{i,c} \propto 1/\lambda_{av}$.

The distribution of available cavities on the heater surface and the liquid-solid contact angle determines which cavities could potentially be active. At the same time, the transport properties of the heater affect the extent of the thermal interaction among the cavities, causing activation and deactivation of individual cavities. Concluding, the surface characteristics affect the pool boiling performance and mechanisms when a pure liquid is boiled over heating surfaces. The density of active sites on the heater surface is affected by the interaction among several parameters on the heater and the liquid sides, as well as the liquid-solid contact angle [92, 93, 99-107].

The $q_{b,c}$ in Eq. (30) can be expressed as[93]

$$q_{b,c} = \frac{c_q D_a}{D_a + 2} \frac{4\pi\alpha}{3d_{c,max}} \left(\frac{\Delta T}{T_w - T_f} \right)^2 \left(\frac{d_{c,max}}{d_{c,min}} \right)^{D_a+2} \quad (32)$$

where $c_q = \pi h_{fg} \rho_g \lambda_b^3 / 6$ is the heat flux removed by a single bubble, h_{fg} is the latent heat of vaporization, ρ_g is the vapor density, $\Delta T = T_w - T_s$, T_s is the saturation temperature of liquids, λ_b is the bubble departure diameter, $d_{c,max}$ and $d_{c,min}$ are respectively the maximum and the minimum diameters of active cavity, and D_a is the fractal dimension of active cavity on the heated surface.

Equation (23) is the fractal analytical expressions of CHF for pool boiling heat transfer in nanofluids, hereafter referred to as the FACHF model, and it indicates that the CHF of pool boiling heat transfer in nanofluids is explicitly related to the average diameter of nanoparticles, the volumetric nanoparticle concentration, the thermal

1 conductivity of nanoparticles, the fractal [dimensions of nanoparticles and active](#) cavity
2 on the heated surface, the temperature, and the properties of fluids.

3 **4.2 FACHF model tests**

4 [The required parameters in Eq. \(30\)](#) can be found from **Appendix**. The fractal
5 dimension for active cavities on the heated surfaces is in the range of $1 < D_a < 2$ in
6 two dimensions, and increases with wall superheat. The CHF of pool boiling heat
7 transfer predicted by FACHF model are compared to experiments with different
8 nanofluids versus ΔT as shown in Fig. 7 [108, 109].

9 [For the thermal heated disk](#) heater under saturated temperature and atmospheric
10 pressure, Kim et al. [108] studied the CHF characteristics of pool boiling for TiO₂
11 nanofluids with $\lambda_{av}=45$ nm, $\phi_p=0.1\%$ and Al₂O₃ nanofluids with $\lambda_{av}=47$ nm, ϕ_p
12 $=0.1\%$. [Besides that](#), the CHF of SiO₂ nanofluids in pool boiling was investigated at
13 $\lambda_{av}=35$ nm and $\phi_p=0.5\%$ under atmospheric pressure [109]. The calculated CHF of
14 nanofluids [using the introduced](#) fractal methods was [shown to a good agreement](#) with
15 the available experimental results reported in the literature.

16 By considering of nanoparticles moving in liquids, analytical expressions for pool
17 boiling heat transfer of nanofluids in the CHF region based on the fractal geometry can
18 be derived, which can reveal the mechanism of pool boiling heat transfer on CHF in
19 nanofluids.

20

21 **5. Fractal Model for subcooled pool boiling of nanofluids**

22 In general, there are two main [mechanisms contribute](#) to subcooled pool boiling
23 heat transfer of nanofluids: the heat flux ($q_{t,c}$) from all nanoparticles moving in liquid

1 and the other ($q_{s,b}$) from subcooled pool boiling of the **base fluids, respectively**. Xiao
 2 et al [110] derived a fractal analytical heat flux model for subcooled pool boiling of
 3 nanofluids as

$$4 \quad q_t = q_{t,c} + q_{s,b} \quad (33)$$

5 where

$$7 \quad q_{t,c} = \left\{ \frac{3}{\alpha} \sqrt{\frac{2k_B T}{\pi \rho_p \lambda_{av}}} \frac{\xi^{\gamma-1.5} - 1}{3-2\gamma} \left[\frac{\xi(2-\gamma)(1-\xi^{1-\gamma})}{1-\gamma} \right]^{3/2} + \frac{2\xi(2-\gamma)(\xi^{\gamma-1} + \xi^{1-\gamma} - 2)}{(1-\gamma)^2} \right\} \frac{k_f \gamma \cdot \Delta T}{(1-\xi^\gamma) \lambda_{av}}$$

8 (34)

$$9 \quad q_{s,b} = \frac{\pi^2 \alpha c_p \rho_f \rho_g h_{fg}}{12g(\rho_f - \rho_g)T_s} \frac{D_a \gamma^{-d_{fc}}}{D_a + 1} \frac{(\Delta T)^3}{\Delta T + \Delta T_{sub}} \quad (35)$$

10 In Eq. (34), $\gamma = \log_{\xi}^{\phi_p}$ is used for simplification. Eq. (35) indicates that the heat flux
 11 of subcooled pool boiling heat transfer in nanofluids is explicitly related to the
 12 nanoparticle concentration (ϕ_p), the average diameter of nanoparticles (λ_{av}), the
 13 fractal dimension (D_a) of active cavity on the heated surfaces, the wall superheat (ΔT)
 14 and the subcooling of fluids (ΔT_{sub}).

15 Zhou [111] investigated experimentally heat transfer characteristics of CaCO₃ and
 16 Cu nanofluids with and without acoustic cavitation, and discussed the effects of such
 17 factors as acoustical parameters, nanoparticle concentration and fluids subcooling on
 18 heat transfer enhancement around a heated horizontal copper tube. Their experimental
 19 results are used to test the fractal analytical **model (Eq.(33))**. **As shown in Fig.8, there**
 20 **are obvious deviations** between theoretical and experimental data spotted at large ΔT .
 21 **This circumstance probably be resulted from experiment error and/or** the uncertainty of
 22 the parameters that has been used in the theoretical calculation of heat transfer.

1

2 **6. Fractal Aggregation of nanoparticles**

3 Nanoparticles aggregation is a time dependent dynamic process [19, 112-114].
4 The structure of aggregation changes continuously because of the Brownian motion.
5 **Initially** (time $t=0$), the particles is dispersed, and then particles agglomerate so that
6 form multiple aggregates. These individual aggregates could be treated as a new
7 *particles* with an effective radius R_a and can thus enhance the thermal conductivity of
8 nanofluids. **Due to the aggregations**, there is a maximum thermal conductivity for well-
9 dispersed aggregates at somewhere between the two **extremes**, no aggregation ($t =0$)
10 and complete aggregation ($t \rightarrow \infty$) [115].

11 The cluster structures formed by the aggregation of gold colloids, silica-colloid,
12 coagulated aerosols or soot exhibits scale-invariance and which can be well described
13 as fractals [60, 116-121]. Weitz and Oliveria [117] utilized transmission-electron
14 micrographs to study the structure formed by the irreversible kinetic aggregation of
15 uniformly sized aqueous gold colloids, and found that the structures were highly
16 ramified and exhibited a scale invariance with fractal dimension 1.75 (see Fig. 9), which
17 is in good consistent with simulated value of diffusion-limited aggregation when the
18 clusters themselves are allowed to aggregate. Gharagozloo and Goodson [122] utilized
19 static light scattering to measure the fractal dimension of aggregates formed in
20 nanofluids over time at various temperatures and concentrations, and found that
21 aggregates formed more quickly at higher concentrations and temperatures.

22 The number of particles in an aggregate N is related to the gyration aggregate
23 radius R_a and single particle radius r_p by [123]

$$N = \left(\frac{R_a}{r_p} \right)^{D_c} = 1 + \frac{t}{t_p} \quad (36)$$

where t_p is the aggregation time constant, D_c is the fractal dimension of the aggregate, $1 \leq D_c \leq 3$ ($D_c=3$ is the limit of a completely compact spherical aggregate). Available studies indicate that the D_c ranges from 1.75 to 2.5 [115]. The reaction limited particle-cluster or diffusion limited cluster-cluster aggregation (DLCCA) mode can be distinguished by the fractal dimension D_c . Irreversible particle-cluster aggregation leads to a denser aggregate than cluster-cluster aggregation with fractal dimensions of 2.5 and 1.8, respectively [123]. Waite et al.[124] found the D_c ranged from 1.8 to 2.3 for aggregation of Al_2O_3 . Wang et al. [28] found that aggregation is DLCCA (D_c close to 1.8) in nanofluid. Gharagozloo and Goodson [122] found that the permanent aggregates in the nanofluid have a fractal dimension of 2.4 and the aggregate formations that grow over time are found to have a fractal dimension of 1.8, which is consistent with diffusion limited aggregation. $D_c=1.8$ is assumed in model calculations by Prasher et al.[115].

The total mass (m_a) of particles in a single aggregate is expressed as[115]:

$$m_a = m_p \left(1 + \frac{t}{t_p} \right) \quad (37)$$

where m_p is the particle mass for a well-dispersed system.

The aggregation time constant t_p is calculated by [123]

$$t_p = \frac{\pi \mu r_p^3 W}{k_B T \phi_p} \quad (38)$$

where $W (\geq 1)$ is the stability ratio, ϕ_p is the volume fraction of the primary particles.

From Eq. (38), t_p increases rapidly with the increasing of r_p , which means rapider

1 aggregation can take place for smaller particles. $t_p \rightarrow \infty$ means the system is stable
 2 and nanoparticles are well dispersed. When repulsive force and hydrodynamic
 3 interactions between the nanoparticles are absented, $W=1$, otherwise, $W > 1$ [115].

4 The thermal conductivity of nanofluids can be significantly enhanced by the
 5 aggregation of nanoparticles into clusters [125]. Considering that the conductivity of
 6 aggregates is based on the Bruggeman model [28], the conductivity of an aggregate (k_a)
 7 is [115]:

$$8 \quad (1 - \phi_{in}) \frac{(k_f - k_a)}{k_f + 2k_a} + \phi_{in} \frac{k_p - k_a}{k_f + 2k_a} = 0 \quad (39)$$

9 where ϕ_{in} is the volume fraction of particles in aggregates, it is calculated by [126]:

$$10 \quad \phi_{in} = \left(\frac{R_a}{r_p} \right)^{D_c - 3} = \left(1 + \frac{t}{t_p} \right)^{\frac{D_c - 3}{D_c}} \quad (40)$$

11 in Eq. (40), the maximum value $\phi_{in} = 1$ and the minimum value $\phi_{in} = \phi_p$ (see Fig.
 12 10) [115]. The contribution due to conduction for the aggregated system can be
 13 calculated by MG model, thus rewriting Eq. (7) by instead of ϕ_p by ϕ_a , yeilds [115,
 14 127] :

$$15 \quad \frac{k_s}{k_f} = \frac{k_a + 2k_f - 2\phi_a(k_f - k_p)}{k_a + 2k_f + \phi_a(k_f - k_p)} \quad (41)$$

16 where $\phi_p = \phi_{in}\phi_a$. Eq. (41) is a fractal thermal conductivity model of nanofluids which
 17 combines the micro-convective effect due to Brownian motion with the change of
 18 conduction caused by particles aggregation. And it has been valided experimently using
 19 data of nanofluids made from different sizes of nanoparticles. In developing Eq. (41),
 20 nanoparticles are assumed to be spherical and of uniform size, and effects of thermal

1 boundary resistance between particles and fluid are neglected. For well dispersed
2 system, $\phi_{in} = 1$ and $\phi_p = \phi_a$, Eq. (41) reduces to the MG model (see Eq. (7)) [115].

3 Figure 10 shows the comparison between the model predictions for aggregated
4 nanoparticles (Eq. (41)) and well dispersed nanoparticles (Eq. (7)). As shown in this
5 figure, the enhancement due to particle aggregation is well demonstrated compared
6 with that for a well-dispersed system. For the percolation effects in the agglomerate,
7 the limiting value ($\phi_{in} = \phi_p$) is slightly higher than that in the MG model. Obviously,
8 particle aggregation enhances the conduction contribution when the aggregates are well
9 dispersed and none large aggregate is formed [115].

10 Following above mentioned works on the effects of aggregation and its kinetics
11 on thermal conductivity [28, 115], Prasher et al [125] further developed a three-level
12 homogenization theory to evaluate the effective thermal conductivity of colloids
13 containing fractal clusters. In other aspect, Gaganpreet and Srivastava [128]
14 theoretically studied the viscosity of oxide nanoparticle dispersions based of fractals of
15 irregular structure of aggregation, and they used prolate ellipsoid aggregation to study
16 the viscosity of nanofluids.

17 Nanoparticle aggregate also shows multifractal [129, 130]. However, for the
18 theatrical determining of the fractal dimension of nanoparticle aggregates, the available
19 methods usually under the limitation of the finite scale/range of self-similarity of
20 physical objects and the resolution of scanning electron microscope (SEM) and TEM
21 methods [131]. Recently, Wozniak et al. [132] also found that multi-scale analysis of a
22 large sample is not suitable to derive morphological parameter of multi-fractal samples
23 of particle aggregates. They further introduced the modified Box-Counting (MBC)
24 algorithm to estimate the fractal dimension of each aggregate from its own self-

1 similarity properties. MBC validity was tested successfully on synthetic aggregates
2 whose fractal dimension was independent of or correlated with aggregate size.

3

4 **7. Fractal analysis on yield stress property of nanoparticle** 5 **aggregation**

6 Besides the attentions on enhancement thermal conductivity of nanoparticles and
7 its cluster in nanofluids, other physical behaviours of the nanoparticle aggregation
8 system also received many attentions, e.g., its yield stress property. It is well accepted
9 that the expression of yield stress μ is a power function of the solid volume fraction
10 [118, 121, 133]:

$$11 \quad \mu = \mu_0 \phi_a^m \quad (42)$$

12 where μ_0 is the referenced parameter ($\mu = \mu_0$ at $\phi_a = 1$), ϕ_a is the solid volume
13 fraction, m is a constant, which is set as different values by different researchers through
14 capturing the role of the aggregate size and the solid volume fraction on yield stress,
15 leaving the number of fitting parameters to a minimum [133]. Combining the fractal
16 model for the aggregate backbones and the aggregate volume, Xi et al. [133] developed
17 a fractal model for the yield stress of aggregates by taking the solid volume fraction and
18 the aggregate diameter into consideration. In Xi et al.'s model [133], the constant m is
19 expressed as:

$$20 \quad m = \frac{2X - 3D_c}{3(D_c - 3)} \quad (43)$$

21 where X is the backbone fractal dimension, which is less than the fractal dimension of
22 the aggregate D_c and is larger than unity to provide a connected path. In Eq. (43), D_c
23 can be determined using small angle X-ray scattering (SAXS), but X is not clearly stated

1 in literatures.

2 Eq. (43) is more generalized compared to other available models. If $X = D_c$, Eq.
3 (43) reduces to the model by Xu et al.[134]:

$$4 \quad m = \frac{D_c}{3(3 - D_c)} \quad (44)$$

5 and if $X = 5D_c/2 - 3$, the exponent $m = 2/3$ from Eq. (43), which is the same as the
6 result of Son and Hsu[135]. Based on Mandelbrot's rules of thumb [60] and analysis of
7 experimental results, Xi et al[133] argued that the relation $X = D_c - 1$ can be as a
8 simple method to evaluate the backbone fractal dimension.

9 By introducing the novel express of constant, the model, Eq. (42), can fit
10 experimental data for polymer system [136] and silica aerogel system well [137], as
11 shown in Fig. 12 [133]. The fractal dimension $D_c = 2.4$ is respectively measured by
12 SAXS method. The SAXS experiments are briefly presented here. A beam of light is
13 directed onto the sample and the scattered light intensity $I(Q)$ is measured as a function
14 of the magnitude of the scattering vector Q , the scaling law between them can be
15 expressed as [138]:

$$16 \quad I(Q) \sim Q^{-D_c} \quad (45)$$

17 From Eq. (45), the fractal dimension D_c can be determined by the value of the slope
18 of a linear fit through data on a logarithmic plot of $I(Q)$ versus Q in the range of
19 $1/R_a < Q < 1/r_p$.

20

21 **8. Discussion and Future Work**

22 Although fractal-based approaches have been proposed to study the heat transfer
23 of nanofluids and the aggregation process of nanoparticles as well as the yield stress

1 property of nanoparticles aggregation, a gap still exists between the expected fruits and
2 presented status. Generally, comparing with other mathematical models, fractal
3 methods expressed the thermal conductivity analytically. What's more less empirical
4 constants are included in the fractal models, which are normally required in other
5 mathematical models. In addition to these advantages, the fractal technique might be
6 potentially applied in analyzing of other transport properties such as optical and
7 electrical properties of nanofluids. So, the fractal technique provides us a new approach
8 besides other numerical methods. However, the shortcoming of fractal theory for
9 application in nanofluids and nanoparticle aggregation is that it should be noticed that
10 it only works fair when $\zeta \leq 10^{-2}$. For nanofluid and nanoparticle aggregation $\zeta \leq 10^{-2}$,
11 so the fractal theory can be used to analyze the characters of nanofluids and nanoparticle
12 aggregation.

13 Further research directions and subjects concerning the transport and other
14 properties of nanoparticle system may be anticipated in:

15 (a) The nanoparticle aggregation is a kinetic process, which is verified to be
16 characterized well by fractal model [115] and be consistent with the diffusion
17 limited aggregation [122]. Thus, the fractal and multi-fractal theories [139, 140]
18 combined with diffusion limited aggregation model can be used to simulate the
19 nanoparticle aggregation and its influence on the heat transfer of nanofluids.

20 (b) From the theoretical equation for predicting fractal dimension (Eq. (4)), the
21 necessary values for the volume fraction and the minimum and maximum size of
22 nanoparticle aggregation needed to be measured by other experimental methods
23 (Such as SEM and TEM). However, in the process of aggregation of nanoparticles,
24 fractal dimension of aggregates is a time dependent variable. How would the fractal
25 dimension change over time at various temperatures need to be further analysed.

1 Static light scattering has been used to obtain the average fractal dimension of
2 aggregation of nanoparticles over time at different length-scales and temperatures
3 [122]. By utilizing Static light scattering technology, the effect of dynamic
4 aggregation process of nanoparticle on heat transfer of nanofluids can be analysed.

5 (c) Generally, the particles are not spherical and smooth, the shape and surface
6 roughness influence the contact area, and further influence the heat transfer between
7 particles and particles, also between the particles and host liquid [141]. The fractal
8 theory can be used to characterize the surface roughness [142, 143] and analyse its
9 influence on the heat transfer of nanofluids and the yield stress property of
10 nanoparticle aggregation.

11 (d) Nanoparticle aggregation in nanofluids is a bi-dispersed porous medium, multiscale
12 phenomenon and its effect also needed to be further analysed. Besides the pore mass
13 fractal model, the solid mass fractal model as well as the pore-solid fractal model
14 [144-146] may also be potential approach to nanofluids and nanoparticle
15 aggregation.

16 (e) Nanoparticle aggregates in the gas phase is demonstrated to be multifractal [129].
17 We argue that this conclusion also apply to nanoparticle in liquid. Thus, how the
18 strength and conductivity properties of these multifractal aggregates influenced by
19 the interaction between the different scales needs to be attended.

20 (f) For the particle aggregates influence the properties of nanofluids, is it possible to
21 design arithmetic with prospective fractal structures to optimize the heat transfer of
22 nanofluids, which is an interesting direction and a challenging issue in nanoscience.

23 (g) Nanoparticle clusters is mixed dynamic behaviour, in which fractal is one of key
24 characterizations. In other word, whole phenomenon of kinetics aggregation can't
25 be fully explained only by one approach. It is necessity to combine other theory or
26 method, such as effective medium theory, percolation theory [147], which would

1 be applied to analyse properties of nanofluids. Remarkable, fractal theory has been
2 one of the basic methods for kinetics aggregation of nanofluids.

3 (h) Besides the combining of fractal theory and Monte Carlo technique [94, 148], the
4 fractal-based method may also be incorporated in other numerical simulation
5 technique in future, such as molecular dynamics simulation, lattice Boltzmann
6 methods, and other computational fluid dynamics methods.

7 (i) The dye diffusion in nanofluids is analogous to heat transfer in nanofluids and has
8 been taken as a strong evidence for the role of micro convection by Brownian
9 motion of particles, the enhanced mass transport visualized could be due to the
10 stabilizer effect as the introduced surfactant could significantly reduce viscosity
11 [149]. Therefore, whether the thermal conductivity enhancement is caused by
12 Brownian motion particles or the reduced viscosity due to the surfactant still need
13 to be further discussed.

14 **9. Conclusions**

15 Nanofluids, consisting of **suspended nanoparticles and base liquids**, usually have
16 much higher thermal conductivity **than the pure base liquids even at very small volume**
17 **fractions of nanoparticles**. Nanoparticles aggregation is a time dependent phenomenon,
18 and can form continuously complex structure system because of the Brownian motion.
19 It has been shown experimentally and numerically that nanoparticles and its
20 aggregation can be well described by fractal theory.

21 This review briefly reviewed the advances of nanoparticles researches and
22 introduced the fractal theory. Then, presented the fractal model of thermal conductivity
23 of nanofluids by taking into account the fractal distribution of nanoparticle sizes and
24 heat convection between nanoparticles and liquids due to the Brownian motion of
25 nanoparticles in fluids, in which the nanoparticles is assumed to be dispersed.

1 With the consideration of nanoparticles moving fluids, three novel fractal models
 2 for heat transfer of nanofluids including convective heat transfer, critical heat flux and
 3 subcooled pool boiling heat transfer were introduced. Besides, three formulas of
 4 predicting the heat flux of boiling heat transfer was summarized, in which the discussed
 5 fractal models were in terms of the average diameter of nanoparticles, the volumetric
 6 nanoparticle concentration, the thermal conductivity of nanoparticle, the fractal
 7 [dimensions of nanoparticles and active cavity](#) on the heated surface, the temperature,
 8 the wall superheat, the subcooling of fluids, and the properties of fluids. An excellent
 9 agreement between the fractal model predictions and experimental data was found.

10 By considering the fractal property of particle aggregate, we also further analyzed
 11 the contribution thermal conductivity due to conduction for the aggregated system
 12 developed from MG model. At last, the yield stress property of nanoparticle
 13 aggregation was fractal [summarized](#).

14 **Appendix**

15 The bubble departure diameter λ_b can be obtained as [150]

$$16 \quad \lambda_b = c_0 \left[\frac{\sigma}{g(\rho_f - \rho_g)} \right]^{1/2} Ja^{*5/4} \quad (A1)$$

17 with $c_0 = 1.5 \times 10^{-4}$ for water, and $c_0 = 4.65 \times 10^{-4}$ for the other liquid, σ is the
 18 surface tension of liquid, Ja^* is the Jakob number which is given by

$$19 \quad Ja^* = \frac{\rho_l c_{pl} T_s}{\rho_g h_{fg}} \quad (A2)$$

20 [The minimum and the maximum active cavity diameter \(\$d_{c,\min}\$ and \$d_{c,\max}\$ \) can be](#)
 21 [predicted](#) by the model as [151]

$$d_{c,\min} = \frac{2\delta}{C_1} \left[1 - \frac{\theta_s}{\theta_w} - \sqrt{\left(1 - \frac{\theta_s}{\theta_w}\right)^2 - \frac{4\zeta_1 C_2}{\delta\theta_w}} \right] \quad (\text{A3})$$

$$d_{c,\max} = \frac{2\delta}{C_1} \left[1 - \frac{\theta_s}{\theta_w} + \sqrt{\left(1 - \frac{\theta_s}{\theta_w}\right)^2 - \frac{4\zeta_1 C_2}{\delta\theta_w}} \right] \quad (\text{A4})$$

where $\zeta_1 = 2\sigma T_s / (\rho_g h_{fg})$; $C_1 = (1 + \cos\theta) / \sin\theta$ and $C_2 = 1 + \cos\theta$, with θ being the contact angle of the fluid and the heater material; $\theta_s = T_s - T_f$; $\theta_w = T_w - T_f$; and δ is the thermal boundary layer thickness in nanofluid which can be expressed as

$$\delta = \frac{k_{eff}}{h} \quad (\text{A5})$$

In nucleate pool boiling of the base fluids, the fractal dimension of active cavity D_a on the heated surface is given by [99]

$$D_a = 2 \frac{\ln \sqrt{2} d_{c,\min} - \ln \bar{d}_{c,\max}}{\ln \gamma} \quad (\text{A6})$$

where $\gamma = d_{c,\min} / d_{c,\max}$. Here $\bar{d}_{c,\max}$ is the averaged value over all the maximum active cavities

$$\bar{d}_{c,\max} = \frac{1}{T_w - T_s} \int_{T_s}^{T_w} d_{c,\max} T_w dT_w = \frac{1}{\Delta T} \sum_{j=1}^N d_{c,\max} T_{w_j} \delta T_w = \frac{1}{N} \sum_{j=1}^N d_{c,\max} T_{w_j} \quad (\text{A7})$$

where $N = \Delta T / \delta T_w$, and δT_w is assumed to be a constant. In the above equation, $T_{w_j} = T_s + j(\delta T_w)$ with $j=1, 2, \dots, N$. For example, if we choose $\delta T_w = 0.2^\circ\text{C}$ then $N=5$ for $\Delta T = 1^\circ\text{C}$, and $N=50$ for $\Delta T = 10^\circ\text{C}$.

16 Nomenclature

17 A total area of nanoparticles

18 c empirical constant in Eq. (18)

19 c_p specific heat at constant pressure

- 1 D fractal dimension of nanoparticle
- 2 D_a fractal dimension of active cavity
- 3 D_c fractal dimension of the aggregate
- 4 $d_{c,\max}$ maximum diameters of active cavity
- 5 $d_{c,\min}$ minimum diameters of active cavity
- 6 d_E Euclidean dimension
- 7 d_f diameter of base liquid molecule
- 8 g gravity acceleration,
- 9 h heat transfer coefficient
- 10 I scattered light intensity
- 11 Ja^* Jakob number
- 12 k thermal conductivity
- 13 k_B Boltzmann constant
- 14 M mass of a fractal object
- 15 m constant in Eq. (41)
- 16 m_a total mass of particles aggregate
- 17 m_p dispersed particle mass
- 18 N number of particle
- 19 Nu Nusselt number
- 20 Pr Prandtl number
- 21 q heat transfer
- 22 Q scattering vector
- 23 R cumulative probability
- 24 R_a gyration aggregate radius

- 1 Ra Rayleigh number,
- 2 Re Reynolds number.
- 3 r_p single particle radius
- 4 T temperatures
- 5 t_p aggregation time constant
- 6 T_s saturation temperature of liquids
- 7 W stability ratio
- 8 X backbone fractal dimension
- 9 Subscripts
- 10 a aggregate
- 11 av average
- 12 c convection
- 13 eff effective
- 14 f fluid
- 15 h heating
- 16 min minimum
- 17 max maximum
- 18 p particle
- 19 s stationary
- 20 t total
- 21 w wall
- 22 Greek letters
- 23 α thermal diffusivity
- 24 γ_1 volumetric thermal expansion coefficient
- 25 δ_T thermal boundary layer thickness

1	ε	measured scale
2	θ	contact angle
3	λ	particle size
4	μ	yield stress
5	ξ	dimensionless coefficient
6	ρ_p	nanoparticle density
7	ρ_g	vapor density,
8	σ	surface tension of liquid
9	ν	kinematic viscosity
10	ϕ	volume fraction
11	ϕ_m	volume fraction of particles in aggregates

12

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19

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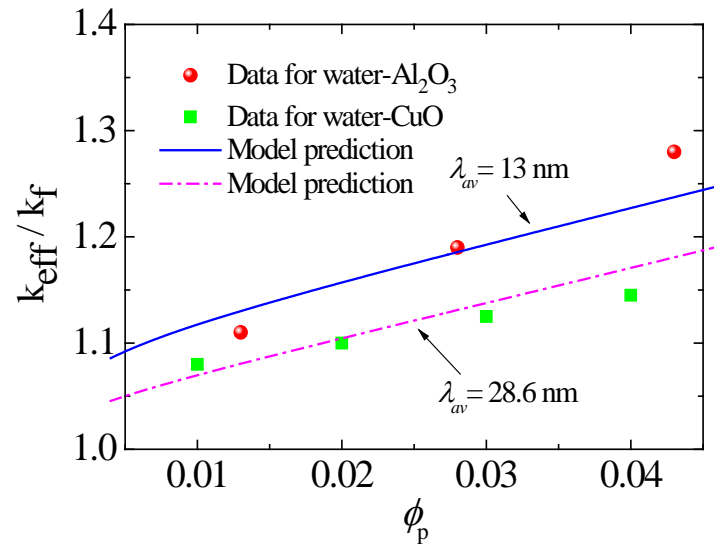
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1 Table 1. The summaries on the effective thermal conductivity property of nanofluids

Year	Ref.	Nanoparticle			Host media	η (%)	Note
		Type	λ_{av} (nm)	ϕ_p (vol. %)			
1993	[24]	Al ₂ O ₃	13	4.3	water	30	the thermal conductivity of water-SiO ₂ system almost never increased
1999	[25]	Al ₂ O ₃	33	4.3	water	15	transient hot-wire method
	[25]	CuO	35	4	EG	20	transient hot-wire method
2001	[13]	Cu	10	0.3	EG	40	transient hot-wire method
2002	[26]	Al ₂ O ₃		5	water	20	transient hot-wire method
2002	[27]	SiC	26	4.2	water	15.8	transient hot-wire method, the effect of particle size and shape is considered
			600	4	water	22.9	
2003	[28]	CuO	50	0.4	water	17	Effective medium theory and fractal theory
2003	[20]	Au	3-4	0.00026	water	21	compared to the conductivity of the basic solution at 30 °C
		Ag	10-20	0.001	toluene	16.5	with respect to the conductivity of the basic toluene at 30 °C
2005	[29]	TiO ₂	-	5	water	33	TiO ₂ nanoparticles is rod-shapes of 10×40 nm (diameter by length)
		TiO ₂	15	5	water	30	TiO ₂ nanoparticles is in spherical shapes of 15 nm
2010	[30]	Al ₂ O ₃	43	3	water	9.7	at room temperature
2011	[31]	Al ₂ O ₃	43	0.086	EG	19	Find the Maxwell method over predicts their experimental values
2012	[32]	Al ₂ O ₃	40-50	0.5	methanol	10.74	transient hot-wire method
		SiO ₂	10-20	0.5	methanol	14.29	
2013	[33]	Fe ₃ O ₄	25	0.1	paraffin	20	transient hot-wire method
2014	[34]	Al ₂ O ₃	36	1.5	EG/water	32.26	base fluids is 20:80% by weight of EG and water mixtures at 60 °C
2016	[35]	TiO ₂	5	1	water	6.55	heat transfer in microchannel heat sinks
2016	[36]	Cu	50-100	0.2	EG	25	transient hot wire method
		Cu	50-100	0.2	water	35	

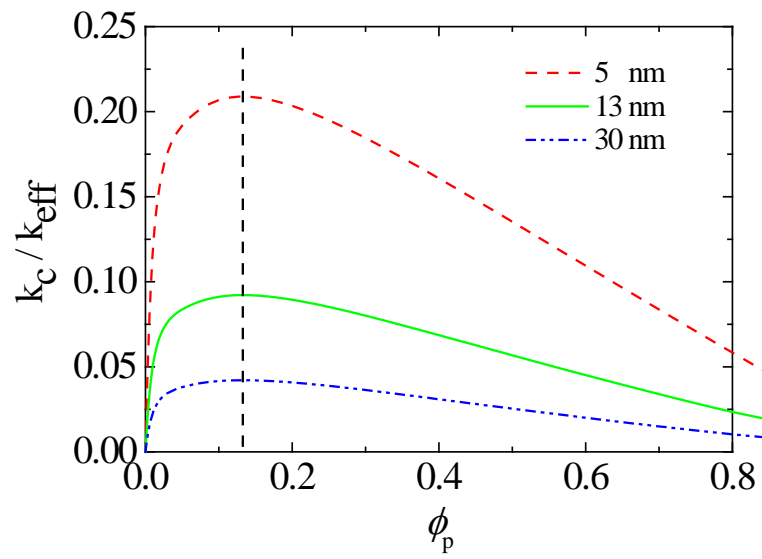
1 EG: ethylene glycol; λ_{av} : average diameter of nanoparticles; ϕ_p : Volume fraction;
 2 η : Increase ratio for thermal conductivity of nanofluids compared to base fluid.

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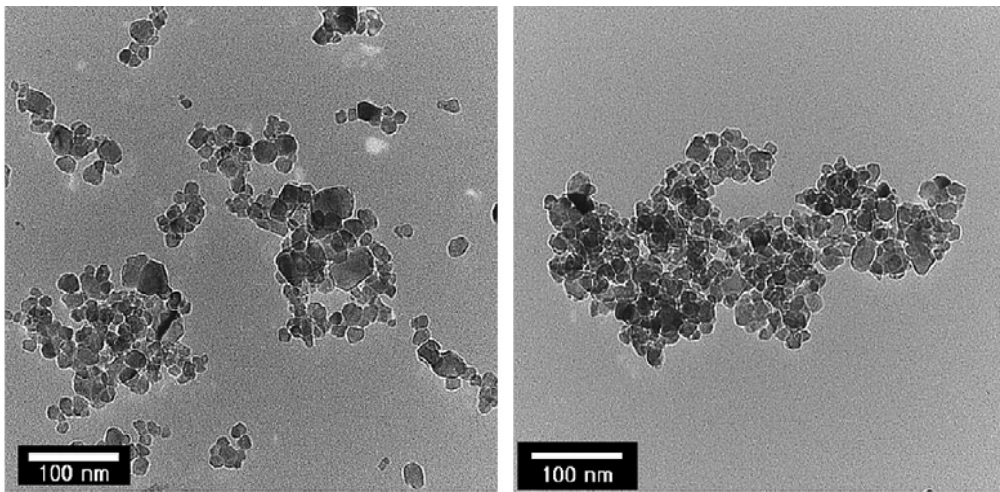
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 7 Figure 1 Test of fractal model (Eq. (20)) for effective thermal conductivity of
 8 nanofluids by experiment data (water-Al₂O₃ [24] and water-CuO [84]). The used
 9 parameters: $Pr = 6.0$, $k_f = 0.610 \text{ W mK}^{-1}$, $d_f = 4.5 \times 10^{-10} \text{ m}$, $c=85$, $k_p = 46.0 \text{ W mK}^{-1}$
 10 and 69.0 W mK^{-1} respectively for Al₂ O₃ and CuO nanoparticles [57].
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7 Figure 2 The effects of particle volume fraction and particle size on effective thermal
8 conductivity of nanofluids [57].
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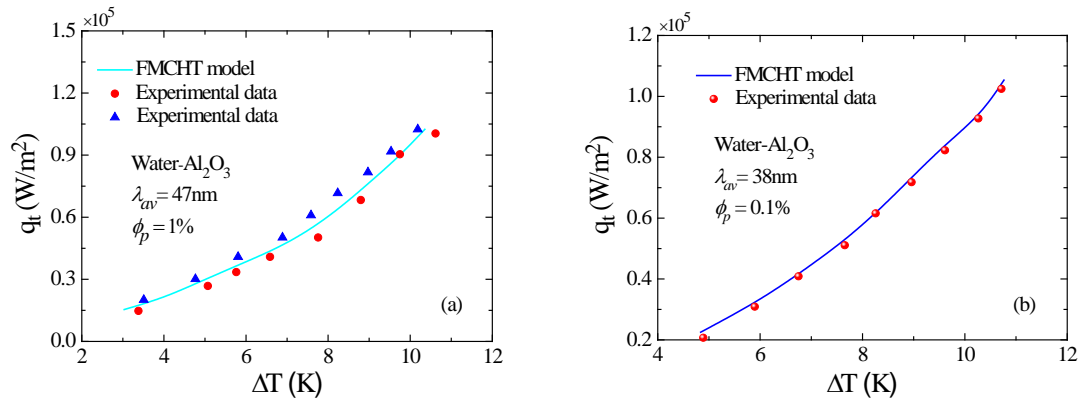
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Figure 3 TEM images of dispersed (Al_2O_3) in distilled water [88]. Copyright 2007, Elsevier.

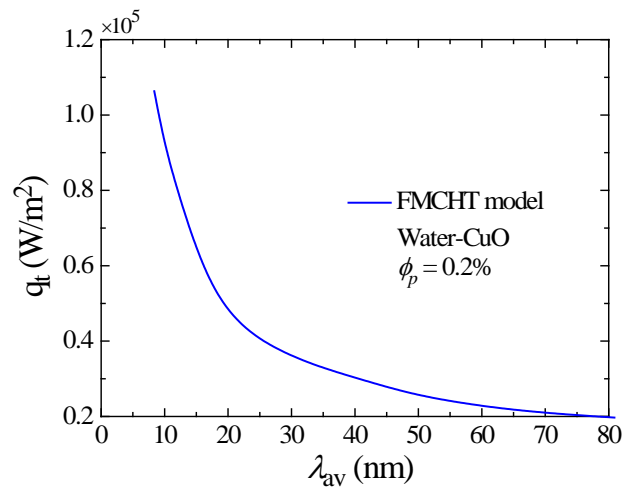
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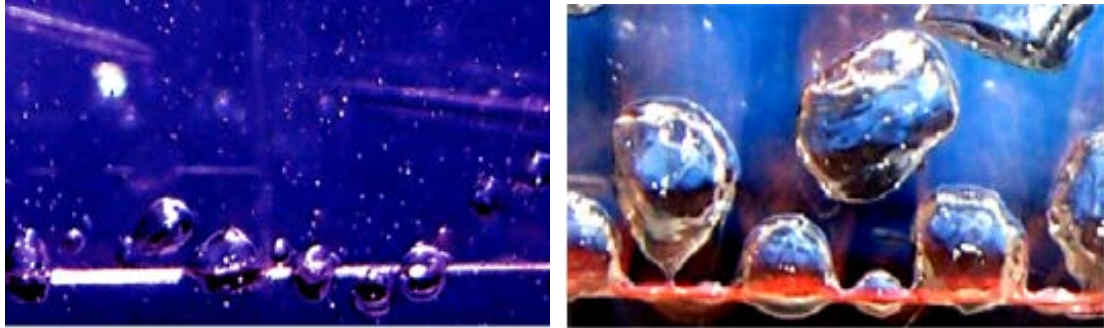
Figure 4 Comparisons between the FMCHT model predictions and the experimental data for Al₂O₃ nanofluids [95, 96]

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7 Figure 5 The heat flux from convective heat transfer of CuO nanofluid versus the
8 average diameter of nanoparticles.
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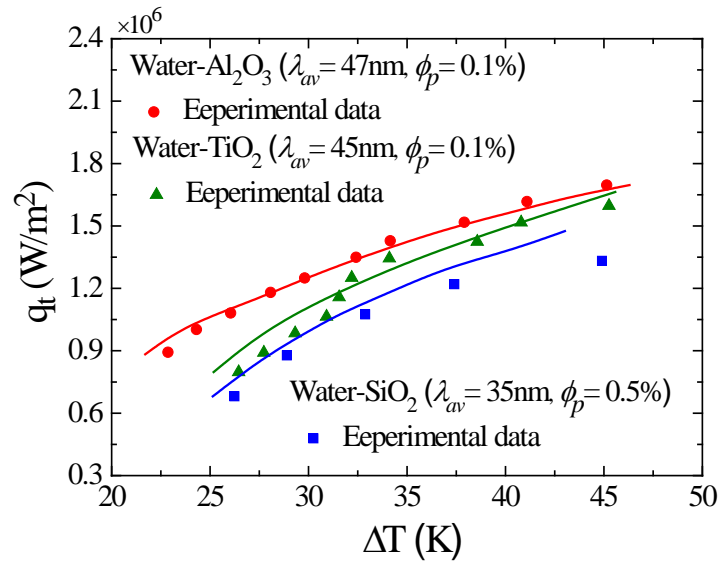
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Figure 6 SEM micrographs [98] of CHF region in Al_2O_3 nanofluids (a) $\text{CHF} = 0.5 \times 10^6 \text{ W} / \text{m}^2$; (b) $\text{CHF} = 1.0 \times 10^6 \text{ W} / \text{m}^2$. Copyright 2007, Elsevier.

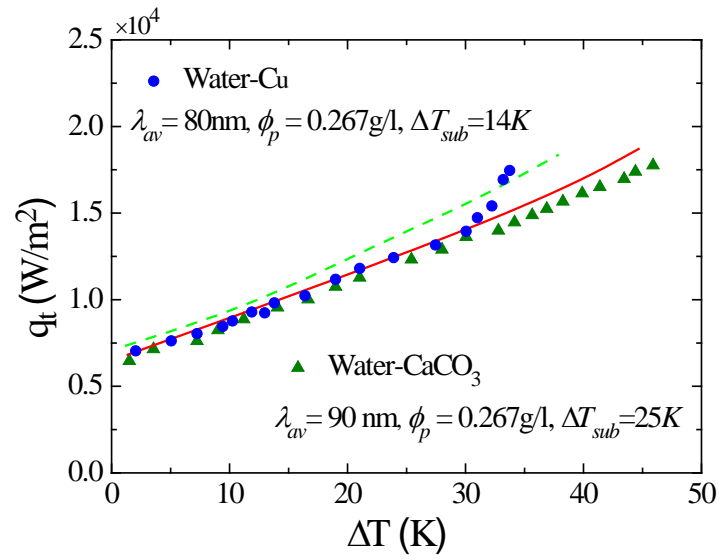
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Figure 7 Comparisons of the present FACHF model predictions and the experimental data[108, 109].

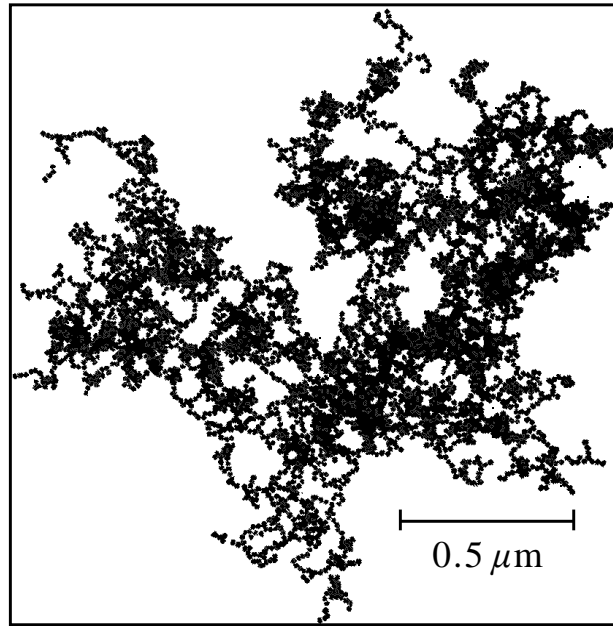
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Figure 8 A comparison between the present model predictions and the experimental data for Cu and CaCO₃ nanofluids[111].

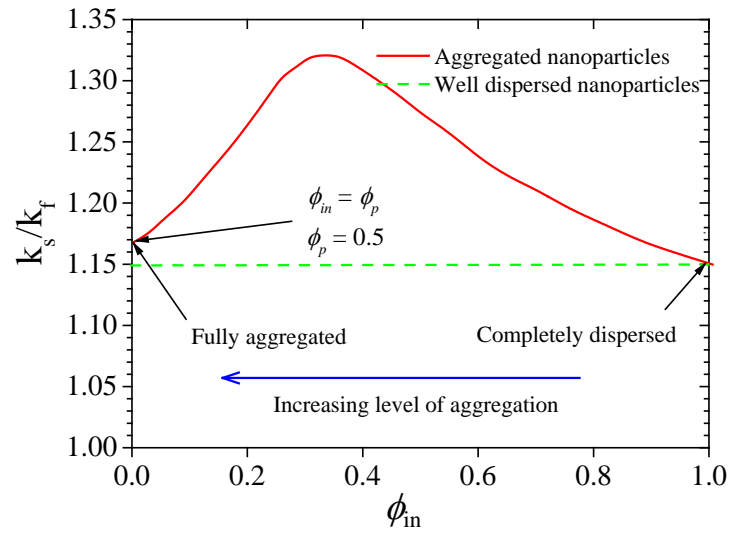
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Figure 9 The fractal dimension (1.75) of typical gold colloid aggregate from its TEM image, in which 4739 gold particles are contained[117]. Copyright 1984, American Physical Society.

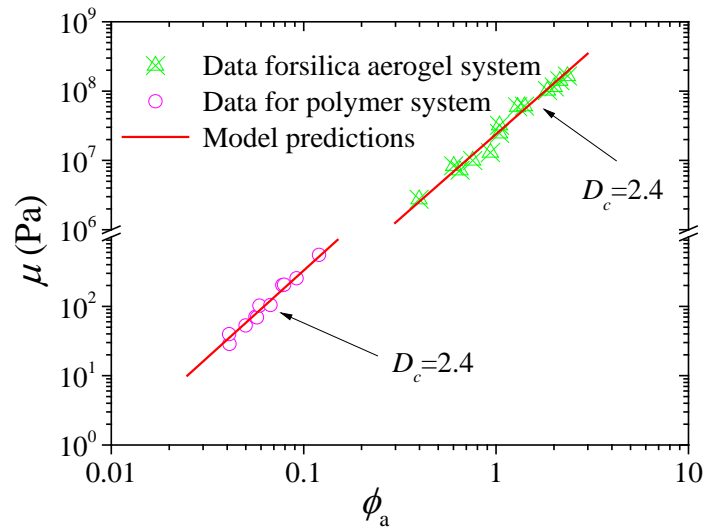
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Figure 10 Effect of aggregation on the conductive contribution to thermal conductivity of nanofluids, compared to that for a well-dispersed system by MG model[115].

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Figure 11 Experimental test on the fractal yield stress model of nanoparticle aggregation versus solid volume fraction[133].