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COLLISIONAL EXCITATION ENERGY TRANSFER
IN THE MIXTURE OF MERCURY AND THALLIUM ATOMS
AND THE CALCULATION OF ATOMIC OSCILLATOR STRENGTHS,
TRANSITION PROBABILITIES, AND LIFETIMES IN THALLIUM, AND CESIUM,

by

Murray K. Wade

A Thesis

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the Department of Physics in Partial Fulfillment
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Windsor, Ontario

1977

ABSTRACT

The method of sensitized fluorescence was used to determine the cross sections for excitation transfer from excited $\text{Hg}(6^3\text{P}_1)$ atoms to close-lying states in thallium. A vapour mixture of mercury and thallium was irradiated with $\text{Hg } 2537\text{\AA}$ resonance radiation and the intensities of the sensitized thallium spectral components resulting from the decay of the collisionally-populated levels were measured. No thallium spectral components originating from levels above the $\text{Hg}(6^3\text{P}_1)$ state were observed in the fluorescence. The mercury vapour pressure was kept low (about 10^{-5} torr) to avoid imprisonment of resonance radiation. The observed cross sections were corrected for cascade transitions where necessary. The calculated cross sections ranged from 0.05\AA^2 to 3.0\AA^2 corresponding to a range in the energy defect ΔE , between the energy of the $\text{Hg}(6^3\text{P}_1)$ state and the excited thallium state, of -1.58eV to -0.083eV . A pronounced resonance was observed between the $\text{Hg}(6^3\text{P}_1)$ state and the close-lying thallium states. The results obtained in this investigation for Hg^*-Tl collisions compare favourable with those obtained previously for Hg^*-Na , Hg^*-Cd , Hg^*-Zn , Rb^*-Cs , and K^*-Rb systems.

The behavior of the sensitized fluorescent spectrum of thallium, produced by irradiating a Hg-Tl-N_2 vapour-gas system with $\text{Hg } 2537\text{\AA}$ resonance radiation, was studied in relation to N_2 pressure. In this tertiary system, the thallium atoms become excited by collisional transfer from $\text{Hg}(6^3\text{P}_1)$ atoms and also from $\text{Hg}(6^3\text{P}_0)$ atoms which are formed in $\text{Hg}(6^3\text{P}_1)-\text{N}_2$ collisions. The spontaneous decay of the collisionally-populated thallium states gives rise to the fluorescent spectrum and their radiationless decay (quenching), caused by collisions with N_2 molecules, manifests itself in a

decrease of the fluorescent intensities as N_2 pressure is increased. The quenching cross sections, obtained by analyzing the variation of the fluorescent intensities with N_2 pressure, appear to exhibit a resonance property with respect to upward vibrational transitions in N_2 .

The relativistic oscillator strengths, transition probabilities, and lifetimes involving the first 25 Thallium₁ excited states and the first 64 excited states of Cesium₁ were calculated and compared to the corresponding non-relativistic Bates-Damgaard and Hydrogenic values commonly employed in the experimental determinations of energy transfer and quenching cross sections. The wave functions that were used were calculated by employing a relativistic Hartree-Fock-Slater method in which the core orbitals were allowed to readjust their radial distributions for each excited state configuration. The resulting non-orthogonality amongst states with the same total angular momentum was removed by applying a Gram-Schmidt orthogonalization process to the radial wave functions. Very good agreement with the most reliable experimental and theoretical data was achieved in the case of Cesium₁. In the case of Thallium₁, the results were not as good although considerable improvement should be expected upon improving the representation of the exchange potential.

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INTRODUCTION

When a pure metal vapour is irradiated with light of the appropriate wavelength, ground state atoms become excited to the resonance state from which they may undergo spontaneous decay back to the ground state, emitting resonance fluorescence.

In many binary metal vapour mixtures, consisting of atoms of two different species, A and B, in which atoms of species A have been excited to the resonance state by radiation of the appropriate frequency, the fluorescent spectrum may include not only the resonance radiation from atoms of species A but also several spectral components characteristic of the atoms B. Mercury and thallium form such a system. When this metal-vapour mixture is irradiated with Hg 2537\AA resonance radiation, the resulting fluorescence includes the mercury resonance frequency and several components arising from the decay of various excited states of thallium. The observed thallium emissions, which are indicated in figure (1) and are usually referred to as sensitized fluorescence, arise from collisions of the second kind, in which the optically excited Hg(6^3P_1) atoms transfer their excitation energy to the Tl($6^2P_{1/2}$) ground state atoms.

In the presence of foreign gas molecules, such as nitrogen, it is observed that the intensity of the resonance radiation is reduced in proportion to the density of the foreign gas molecules. This is due to collisions of the second kind between the excited metallic atoms and the foreign gas molecules in the $v=0$ vibrational state. Such a collision process, in which the excited atomic level is depopulated by a radiationless energy transfer to the foreign gas molecule, is called a quenching

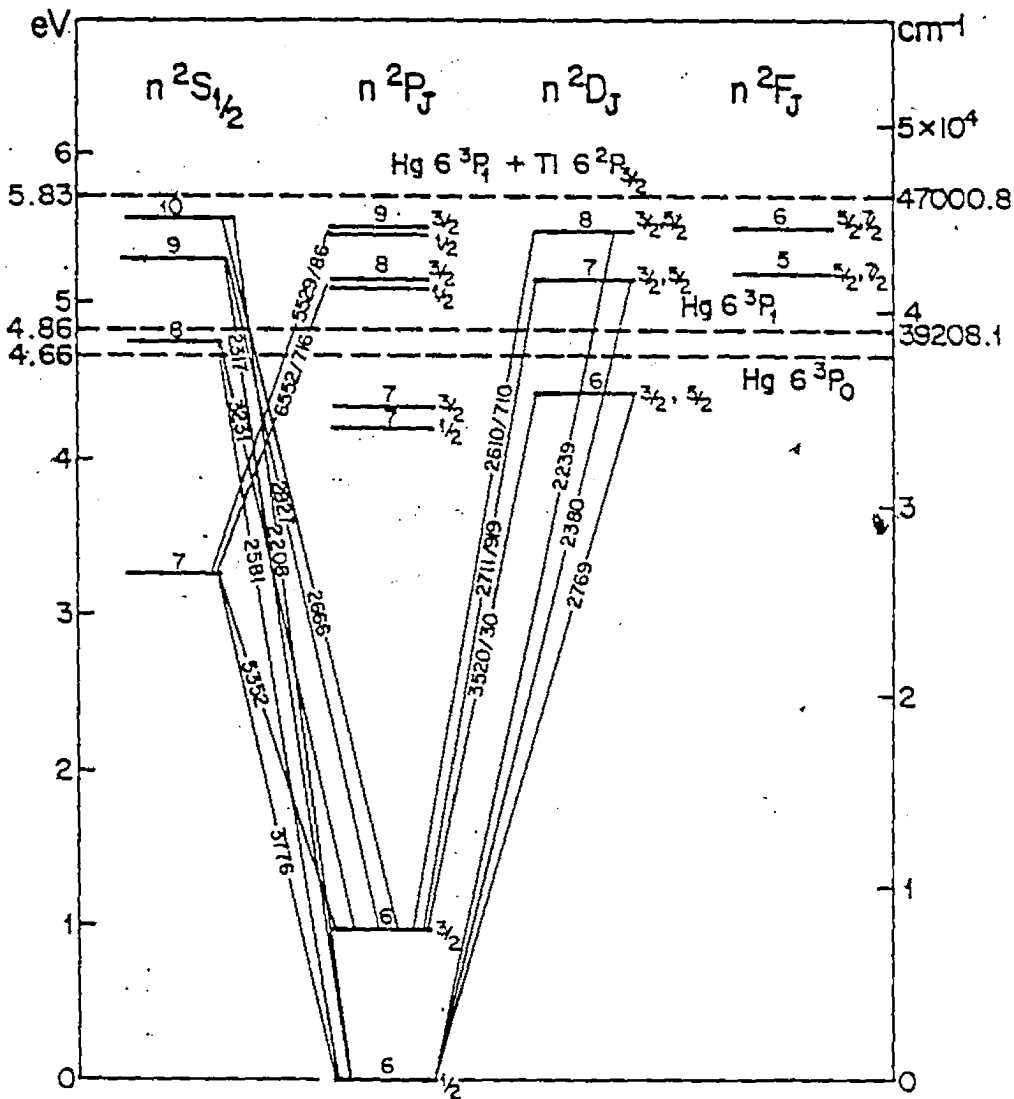


Fig. 1: Energy levels of thallium

collision. An excited atomic state is quenched most efficiently by molecular species, such as nitrogen, but generally atomic gases are almost totally ineffective as quenching agents since quenching collisions between atomic species require the conversion of internal energy into kinetic energy. This represents a very inefficient process in thermal energy collisions. An exception to this general scheme has been found in the case of boron which has been shown to be effectively quenched by argon¹.

The theory of collisions of the second kind was first postulated by Klein and Rosseland² on the basis of the principle of microscopic reversibility. This principle states that, in order to preserve thermodynamic equilibrium, each microscopic process must be accompanied by a reverse process. In light of the results of the Franck-Hertz experiments of 1913, where it was found that a collision of a fast electron with a slow atom resulted in the formation of an excited atom and a slow electron, Klein and Rosseland concluded that excited atoms could collide with slow electrons and produce unexcited atoms and fast electrons, the process being unaccompanied by the emission of a photon.

Franck³ later extended this postulate to a system of excited atoms of one species and unexcited atoms of a second species. He suggested that mercury atoms excited by the Hg 2537Å resonance radiation could, in a collision of the second kind, transfer their excitation energy to unexcited atoms of another species which did not absorb the mercury resonance radiation. The atoms of the second type, if they had energy states close to that of Hg(6^3P_1) at 4.86eV, could become excited and subsequently emit a portion of their characteristic spectrum. This is the essential meaning of the term sensitized fluorescence. Franck and Cario⁴ performed such an experiment using a mercury-thallium mixture in

which the mercury vapour was excited by mercury resonance radiation and the Tl($6^2P_{1/2}$) ground state atoms served as the second element. In the fluorescent spectrum, they observed several components arising from thallium states with excitation energies above the 4.86eV level of the Hg(6^3P_1) resonance state, and suggested that the additional energy required was converted from the available thermal kinetic energy of the colliding partners.

Franck⁵ later formulated the empirical rule that, for a most efficient collisional excitation transfer, the energy defect ΔE between the two atomic states of the two colliding partners should be as small as possible. The formulation of this rule hinges on the principle that collisional excitation transfer occurs with the highest probability when the least amount of excitation energy is transformed into translational energy. On the basis of this empirical rule one would expect that cross sections for atomic excitation transfer decrease with increasing energy defects, ΔE . In general, the term "cross section" denotes a measure of the probability for a particular process to take place. Hence, the cross section for atomic excitation energy transfer is a measure of the probability that, during a collision of the second kind, an atom will transfer some, or all, of its excitation energy to the collision partner.

During the past two decades much effort has been devoted to the study of this resonance effect in inelastic collision processes (Krause^{6, 7}; Czajkowski et.al.⁸; Hrycyshyn and Krause⁹; Stacy and Zare¹⁰; Dashevskaya et.al.¹¹; Frish and Kraulinia¹²; Kraulinia^{13, 14, 15, 16}; Hudson and Curnutte¹⁷; Czajkowski and Krause^{18, 19}). Various atomic systems have been examined although the range of the energy defects, ΔE ,

has been quite narrow (less than 1.0eV). The cross section for energy transfer in metal-metal collisions are also known to depend on the relative velocity of the colliding partners. Recently, beam experiments have been performed to investigate the dependence of the energy transfer cross sections on the relative velocity of the colliding partners in the Na-Ar and Na-Xe systems²⁰ and in the K-He system²¹. Similar experiments have been performed on the K-Rare Gas and some K-Diatomic molecule systems as well²². Since experimental data for the velocity dependence of the energy transfer cross sections in the mercury-thallium system is unavailable and since this dependence cannot be calculated satisfactorily, the energy transfer cross sections are treated as a function of the energy defect ΔE with the velocity dependence being treated as an average over the thermal velocity distribution.

The purpose of this investigation was to study the interactions between excited $\text{Hg}(6^3P_1)$ atoms and ground state thallium atoms, and to supply additional quantitative data pertinent to the study of the resonance effect in excitation transfer. The quenching of thallium by collisions with N_2 molecules was also investigated.

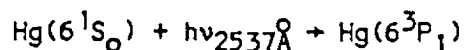
The evaluation of the various cross sections requires the knowledge of thallium lifetimes and transition probabilities for the various states. There is a general lack of experimentally determined values for these parameters and, although several past investigations^{17, 23} have used the Bates-Damgaard approximation²⁴ to generate the required information, such an approach may be entirely inadequate for the higher thallium states. It was felt that more rigorous calculations should be carried out to improve the estimates of the cross sections. A relativistic Hartree-Fock-Slater approach was used to determine the radial wave functions required to calculate the Thallium₁ transition probabilities and lifetimes.

Anticipating future work of a similar nature on the binary Cd*-Cs system, calculations of the Cesium₁ transition probabilities and lifetimes were also performed.

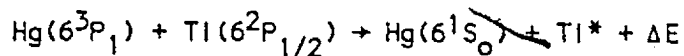
THEORY

Part A: Sensitized Fluorescence

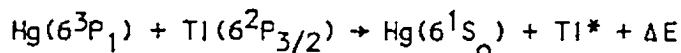
If a mercury-thallium mixture is continuously irradiated with the Hg 2537Å resonance radiation then, as the first step,



As a consequence of inelastic collisions between excited mercury atoms and ground state thallium atoms in the mixture, part of the excitation energy is transferred from mercury to the ground state thallium atoms, causing them to be raised to the excited state. This process is governed by the mechanism



A second possible process involves the excitation energy transfer from excited mercury atoms to the metastable thallium atoms, which have been created by the decay of excited thallium atoms formed during collisions between the Hg(6³P₁) and Tl(6²P_{1/2}) atoms. This second process involving the metastable Tl(6²P_{3/2}) atoms may be expressed by the mechanism



At temperatures used in this investigation (of the order of 500°C) the available thermal energy amounts to only about 0.1eV which is considerably less than the 0.97eV necessary to populate the metastable Tl(6²P_{3/2}) level directly. Consequently, the energy transfer mechanism involving the thallium metastable state may be considered as a second

order effect. Kraulinya¹⁶ attributes the observed sensitized fluorescent spectra originating from the $Tl(8^2D_{3/2,5/2})$, $Tl(9^2S_{1/2})$ and $Tl(7^2D_{3/2,5/2})$ states to this second order process. She further points out that at thallium pressures corresponding to a temperature of $900^\circ C$ the metastable state has a population density three orders of magnitude lower than that of the $Tl(6^3P_{1/2})$ ground state.

The probability of depopulating some upper thallium level, j , by a radiative transition to some lower state, i , is given by

$$P_{em} = A_{ij} + B_{ij} \rho(\nu_{ij}) \quad (1)$$

where A_{ij} is the spontaneous emission probability; B_{ij} is the induced emission coefficient for the particular transition of frequency ν_{ij} between levels j and i ; and $\rho(\nu_{ij})$ is the energy density per unit frequency range of the radiation. The absolute intensity* of the spectral component corresponding to the transition $j \rightarrow i$ is

$$I_{ij} = N_j A_{ij} h\nu_{ij} + N_j B_{ij} h\nu_{ij} \rho(\nu_{ij}) \quad (2)$$

where N_j is the population density of the upper state, j .

The energy density $\rho(\nu_{ij})$ is small and the contribution to the intensity from induced emission may be neglected; equation (2) therefore reduces to

$$I_{ij} = N_j A_{ij} h\nu_{ij} \quad (3)$$

For inelastic collisions the number of exciting collisions per unit time per unit volume between mercury and thallium is governed by the equation

$$\Delta N(Tl^*) = N_o(Tl) N(Hg^*) Q_{oj} \langle \nu_{rel} \rangle \quad (4)$$

* more precisely this is the total power emitted per unit volume of the fluorescing vapour. However, since the geometry of the experiment remains fixed, the volume factor will cancel out in formulae for the energy transfer cross sections which follow.

where $N(Tl^*)$ is the density of excited thallium atoms; $N_0(Tl)$ is the density of ground state thallium atoms and $N(Hg^*)$ is the density of mercury atoms in the $Hg(6^3P_1)$ state; Q_{0j} is the effective inelastic collision cross section for the excitation of a thallium atom from the ground state to the j^{th} excited state; $\langle v_{rel} \rangle$ is the relative velocity of the colliding partners in the vapour mixture assumed to be in thermal equilibrium, and is given by

$$\langle v_{rel} \rangle = \sqrt{\frac{8kT}{\pi\mu}} \quad (5)$$

where k is the Boltzmann constant, T is the temperature in degrees Kelvin and μ is the reduced mass of the colliding partners. Since the vapour mixture will have a Maxwellian velocity distribution, the use of $\langle v_{rel} \rangle$ is justified.

A general approach to the situation may be developed using the j^{th} thallium level as an example. On collision with a mercury atom in the $Hg(6^3P_1)$ state, some thallium states above the j^{th} state may also become excited. These higher states then become sources for the population of the j^{th} state. The depopulation of the j^{th} state is determined by direct transition to the lowest allowed energy level; by cascade transitions; and by collisions of the second kind with $Hg(6^1S_0)$ and $Tl(6^2P_{1/2})$ atoms. At equilibrium the population rate of the j^{th} thallium level must be exactly balanced by its rate of depopulation. Equivalently, one can express the equilibrium condition for the j^{th} thallium state as

$$N_0(Tl) N(Hg^*) Q_{0j} \langle v_{rel} \rangle + \sum_{k=j+1}^{\infty} N_k(Tl^*) A_{jk} = N_0(Hg) N_j(Tl^*) Q_{j0} \langle v_{rel} \rangle + N_j(Tl^*) \sum_{l=j-1}^0 A_{lj} \quad (6)$$

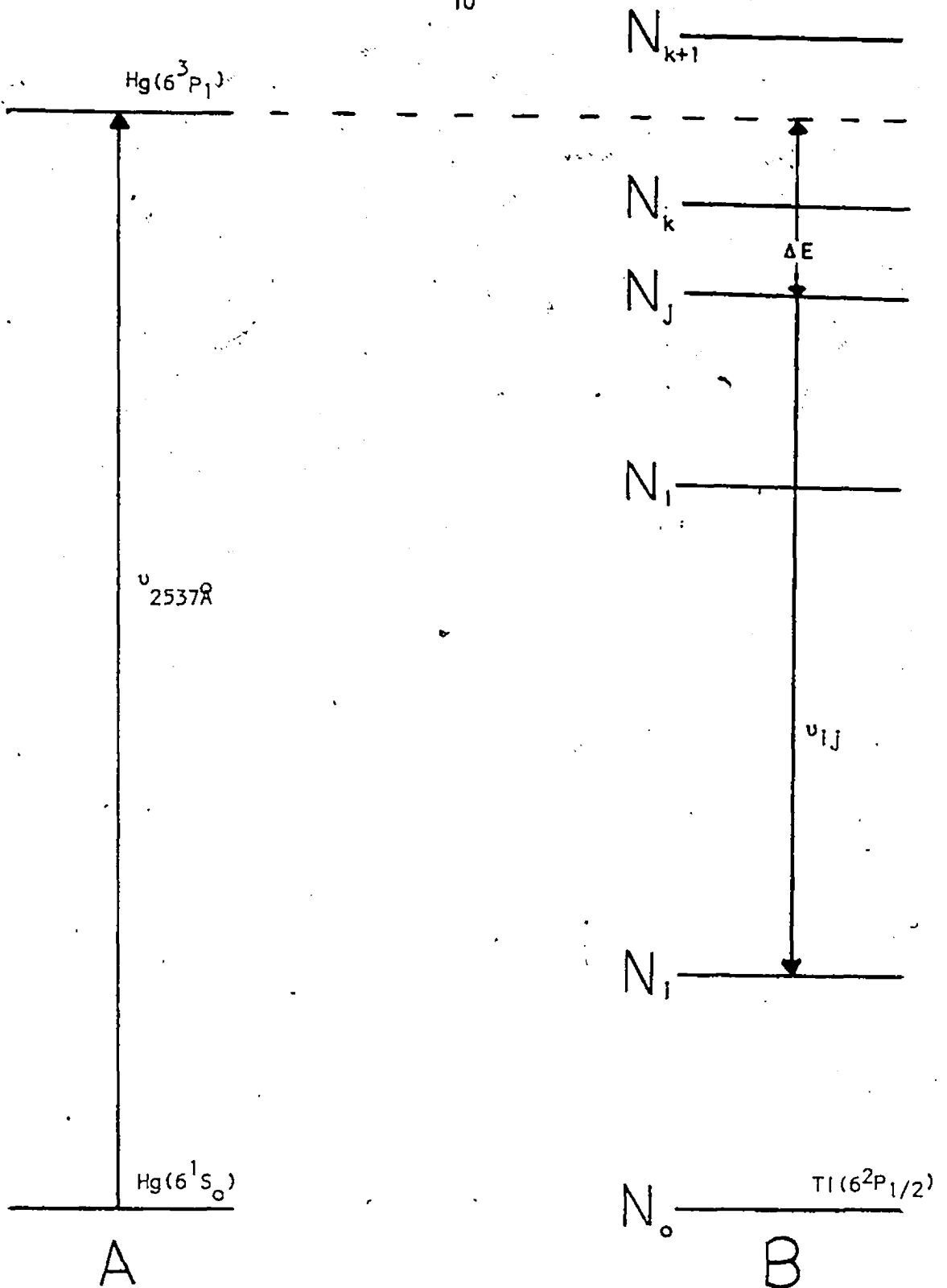


Fig. 2 : Atomic energy level labelling scheme for excitation energy transfer between excited mercury atoms (A) and ground state thallium atoms(B)

where $N_0(\text{Tl})$ is the density of ground state thallium; $\sum_{k=j+1}^{\infty} N_k(\text{Tl}^*) A_{jk}$ is the population rate of the j^{th} state by transitions from all excited states, k , above the j^{th} state; $N_0(\text{Hg})$ is the density of ground state mercury atoms; $N_j(\text{Tl}^*)$ is the density of thallium atoms in the j^{th} excited state; $N_j(\text{Tl}^*) \sum_{i=j-1}^0 A_{ij}$ is the depopulation rate of the j^{th} excited state by transitions to all thallium levels below the j^{th} state; the A_{ij} are the thallium transition probabilities for the transition $j \rightarrow i$; Q_{0j} is the effective cross section for the inelastic collision with $\text{Hg}(6^3\text{P}_1)$ atoms; Q_{j0} is the effective cross section for the reverse inelastic collision process. The principle of detailed balancing²⁵ relates Q_{0j} and Q_{j0} by the relation

$$g_0 p^2 Q_{0j} = g_j p'^2 Q_{j0} \quad (7)$$

where g_0 and g_j are the statistical weights and p and p' are the momenta of mercury atoms in the $\text{Hg}(6^3\text{P}_1)$ state and thallium atoms in the j^{th} excited state respectively.

Substitution of equation (7) into equation (6) gives

$$Q_{0j} = \frac{N_j(\text{Tl}^*) \sum_{i=j-1}^0 A_{ij} - \sum_{k=j+1}^{\infty} N_k(\text{Tl}^*) A_{jk}}{N_0(\text{Tl}) N(\text{Hg}^*) \langle v_{\text{rel}} \rangle - N_0(\text{Hg}) N_j(\text{Tl}^*) \frac{g_0 p^2 \langle v_{\text{rel}} \rangle}{g_j p'^2}} \quad (8)$$

When the pressure of mercury vapour is low, as was the case in this investigation, the time between mercury-thallium collisions is much longer than the lifetimes of the excited thallium states. Thus, the thallium atoms will have lost their excitation energy long before they collide with ground state mercury atoms. Also, the density of excited thallium atoms in the j^{th} state is much smaller than both the density of excited

mercury atoms in the $\text{Hg}(6^3\text{P}_1)$ state and ground state thallium atoms.

Therefore, the second term in the denominator of equation (8) may be neglected with respect to the first term. The effective energy transfer cross section then reduces to the form

$$Q_{0j} = \frac{N_j(\text{Tl}^*) \sum_{l=j-1}^0 A_{lj} - \sum_{k=j-1}^{\infty} N_k(\text{Tl}^*) A_{jk}}{N_0(\text{Tl})N(\text{Hg}^*)\langle v_{\text{rel}} \rangle} \quad (9)$$

To put equation (9) into a form that is experimentally realizable, we make use of equation (12) which relates the intensity per unit volume, I_{ij} , to the density of the j^{th} excited state. From equation (9) the density, $N_j(\text{Tl}^*)$, may be written as

$$N_j(\text{Tl}^*) = \frac{I_{ij}}{A_{ij}h\nu_{ij}} \quad (10)$$

where ν_{ij} is the frequency of the spectral component.

Substitution of equation (10) into equation (9) gives

$$Q_{0j} = \left[\frac{I_{ij}}{A_{ij}h\nu_{ij}} \sum_{l=j-1}^0 A_{lj} - \sum_{k=j+1}^{\infty} N_k(\text{Tl}^*) A_{jk} \right] \frac{1}{N_0(\text{Tl})N(\text{Hg}^*)\langle v_{\text{rel}} \rangle} \quad (11)$$

In the term $1/(N_0(\text{Tl})N(\text{Hg}^*)\langle v_{\text{rel}} \rangle)$, the density of the ground state thallium atoms is a constant parameter and is completely controlled by the temperature at the thallium side arm. The density of mercury atoms in the $\text{Hg}(6^3\text{P}_1)$ state is controlled by the intensity of the $\text{Hg } 2537\text{\AA}$ radiation emitted by the r-f source and may be considered constant since the spectral intensity of the source was essentially constant throughout the investigation and the density of mercury in the reaction region was kept low to avoid the effects of radiation trapping, scattering of incident

light, and formation of the $\text{Hg}(6^3\text{P}_0)$ metastable atoms. The relative velocity depends only on the temperature at the reaction site in the cell and, by careful control of the oven temperature, this parameter could be maintained constant. Hence, under controlled temperature conditions the term $1/(N_0(T)N(\text{Hg}^*)\langle v_{\text{rel}} \rangle)$ remains constant.

The first term in the square brackets of equation (11) encompasses all stepwise transitions and direct transitions from the j^{th} thallium state to all lower levels. The remaining term allows for cascade transitions which populate the j^{th} energy level. In the temperature range used in this investigation, the number of significant cascade transitions from thallium states above the $\text{Hg}(6^3\text{P}_1)$ level would be expected to be very small. The infinite sum accounts for the fact that, in any collision process where energy is exchanged between the colliding partners, there exists a finite probability that the recipient atom may be excited to any higher excited state and even to ionization. In this case, with a thermal energy contribution only of the order of 0.1eV, the probability of exciting thallium atoms to states with energies greater than 5.0eV is very small.

Equation (11) applies to energy transfer from the $\text{Hg}(6^3\text{P}_1)$ state to thallium states lying below the 4.86eV energy level. For excited thallium states above the $\text{Hg}(6^3\text{P}_1)$ level, a portion of the translational energy of the colliding partners is converted into excitation energy. Assuming a Maxwell-Boltzmann energy distribution in the mixture, only a fraction, δ , of the inelastic collisions for which the relative energy of the colliding partners is greater than the energy defect, ΔE , can contribute to the energy exchange. This fraction of inelastic collisions populating thallium levels above the energy of the $\text{Hg}(6^3\text{P}_1)$ state is given by²⁶

$$\delta = \frac{2}{\sqrt{\pi} (kT)^{3/2}} \int_{\Delta E}^{\infty} \exp\left[-\frac{E}{kT}\right] \sqrt{E} dE \quad (12)$$

$$= \exp(-E/kT) \left[\frac{E}{kT} + 1 \right]$$

where ΔE is the energy defect between the $\text{Hg}(6^3P_1)$ level and the particular thallium level. Thus one can write equation (11) in a form applicable to all thallium excited states as

$$Q_{0j} = \frac{1}{\delta N_0(Tl) N(Hg^*) \langle v_{rel} \rangle} \left[\frac{I_{1j}}{A_{1j} h\nu_{1j}} \sum_{l=j-1}^0 A_{lj} - \sum_{k=j+1}^{\infty} N_k(Tl^*) A_{jk} \right] \quad (13)$$

where

$$\delta \begin{cases} = 1 & \text{if state } j \text{ lies below the } \text{Hg}(6^3P_1) \text{ level} \\ = \exp(-E/kT) \left[\frac{E}{kT} + 1 \right] & \text{if state } j \text{ lies above the } \text{Hg}(6^3P_1) \text{ level} \end{cases} \quad (13a)$$

Part B: Quenching

When the effects of radiation diffusion and collision broadening are negligible, the quenching cross section for a particular state may be determined by using the Stern-Volmer formula²⁷.

Consider a resonance cell in which there are n_1 excited atoms and n_2 foreign gas molecules per cm^3 of the emitting layer. If the lifetime of the excited atomic state is τ , then the number of quenching collisions per second per unit volume is

$$Z = 2n_1 n_2 Q \left[\frac{2kT}{\pi} \left(\frac{1}{m_1} + \frac{1}{m_2} \right) \right]^{1/2} \quad (14)$$

where Q is the quenching cross section in \AA^2 , and m_1 and m_2 are the masses of the atom and of the foreign gas molecule, respectively. The

number of quenching collisions per second per unit volume of the emitting layer per atom is

$$Z_Q = Z/n_1 \quad (15)$$

At equilibrium, the rate at which the excited state is depopulated equals the rate at which it is repopulated. Define the rate of formation of the excited state as s_1 . In the absence of foreign gas, the rate of depopulation by spontaneous decay is n_1/τ . The rate of depopulation by quenching collisions with foreign gas molecules is Z_Q . Thus, the rate of depopulation of the excited state is a constant fraction, a , of the total energy emitted by the n_1 excited atoms. At equilibrium,

$$n_1 = \frac{s_1}{1/\tau + Z_Q} \quad (16)$$

The intensity of the emitted radiation in the absence of foreign gas is

$$a \frac{n_1}{\tau} hv = a s_1 hv \quad (17)$$

and the intensity of the emitted radiation in the presence of the foreign gas is

$$a \frac{n_1}{\tau} hv = \frac{a s_1 hv}{1 + \tau Z_Q} \quad (18)$$

The quenching is defined as

$$\frac{I}{I_0} = \frac{\text{Intensity of emitted radiation with foreign gas present}}{\text{Intensity of emitted radiation in the absence of foreign gas}} \quad (19)$$

Thus,

$$\frac{I_0}{I} = \frac{1}{1 + \tau Z_0} \quad (20)$$

A plot of I_0/I against the number density, n_2 , of the foreign gas molecules yields the quenching cross section, Q , for the state:

$$\begin{aligned} \frac{I_0}{I} &= 1 + \tau Q n_2 \left[\frac{8kT}{\pi} \left(\frac{1}{m_1} + \frac{1}{m_2} \right) \right]^{1/2} \\ &= 1 + \tau Q n_2 \langle v_{rel} \rangle \end{aligned} \quad (21)$$

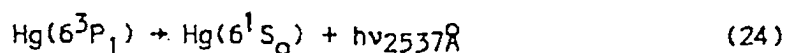
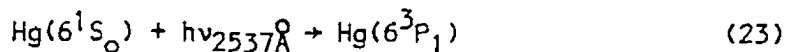
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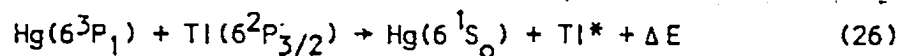
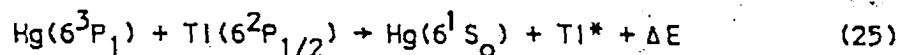
$$Q = \frac{I_0/I - 1}{\tau n_2 \langle v_{rel} \rangle} \quad (22)$$

A plot of I_0/I against the number density of the foreign gas molecules is a straight line passing through (0,1) as origin, with a slope given by $Q \tau \langle v_{rel} \rangle$. From this slope, knowing the lifetime, τ , of the excited state, and calculating the relative velocity, $\langle v_{rel} \rangle$, of the colliding pair, it is possible to determine the quenching cross section, Q , for the excited atomic state.

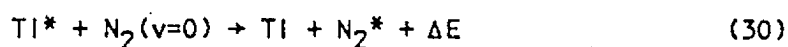
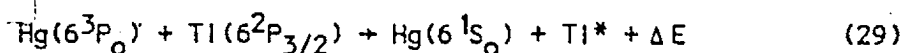
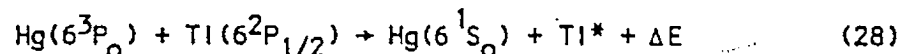
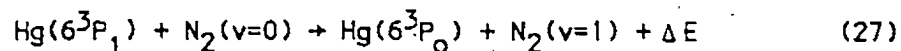
Part C: Radiative and Collisional Excitation Energy Transfer Mechanisms

Several of the most important radiative and collisional processes that may take place in the mercury-thallium vapour mixture irradiated with Hg 2537Å resonance radiation can be described by the following equations:





In the presence of nitrogen molecules the following additional processes may also occur:



Equations (23) and (24) represent the optical excitation of the mercury atom and its decay resulting in the emission of the Hg 2537 \AA resonance fluorescence. Equation (25) indicates the collisional excitation transfer between Hg(6^3P_1) and ground state thallium atoms. The process described by equation (26) is unlikely in the absence of a buffer gas, since the lifetime of the thallium metastable state is effectively limited by the transit time between the fluorescing region in the vapour and the cell wall, which is about 10^{-5} seconds at thallium pressures of the order of 10^{-4} torr. Equation (27) represents the formation of the mercury metastable state by collisions with nitrogen molecules in the ground vibrational state. Equations (28) and (29) indicate the collisional excitation transfer between the metastable mercury atoms and thallium atoms in the ground and metastable states, respectively. In the presence of buffer gases, such as nitrogen, equation (29) may play a significant role. Addition of nitrogen to the metal vapour system has the effect of increasing

the effective lifetime of the metastable $Tl(6^2P_{3/2})$ state, by reducing the number of quenching collisions with the cell walls. Also, the $Tl(6^2P_{3/2})$ state lies 0.97eV above the $Tl(6^2P_{1/2})$ ground state and consequently, it becomes possible to excite the thallium states in the vicinity of the 5.83eV level, (indicated in figure(1)), in collisions with metastable $Hg(6^3P_0)$ atoms. The intensities of the spectral components arising from these higher excited states and the subsequent influence of the latter on the cross sections for excitation transfer to lower thallium levels (through cascade effects), will depend strongly on the population density of the $Tl(6^2P_{3/2})$ metastable state. Finally, equation (30) expresses the quenching action of nitrogen molecules on the excited thallium states.⁷

It should be noted that no provision has been made for the formation of mercury molecules and their subsequent interactions with thallium atoms. At low mercury vapour and buffer gas pressures, the probability of forming Hg_2 molecules is vanishingly small²⁸. This was further verified by Wlnas et.al.²⁹, who worked with a mercury-thallium mixture, and found that the mercury molecular band structures at 3250Å and 4350Å appeared only above 100 torr of mercury pressure and at very high thallium pressures. These observations indicate that one may safely omit the Hg_2 molecules from consideration under conditions employed in this present investigation.

PART D: Calculation of Relativistic Atomic Oscillator Strengths, Transition Probabilities and Lifetimes for Thallium, and Cesium

For the sake of completeness, the discussion of this section will begin with a review of the non-relativistic Hartree-Fock theory and its simplification by the use of the averaged exchange potential introduced by Slater³⁰. The extension to the relativistic case is then presented along with the theoretical calculations for Thallium, and Cesium, using the relativistic Hartree-Fock-Slater method. Unless otherwise specified we work in atomic units (a.u.) where the electronic charge, e , and \hbar , Planck's constant divided by 2π , are set equal to unity. The velocity of light is then equal to α^{-1} where α is the fine structure constant. The speed of light has the numerical value of 137.03602 a.u..

(1.0) Non-Relativistic Hartree-Fock Theory

The foundation of the Hartree-Fock (HF) theory rests on the Self-Consistent-Field (SCF) method devised by Hartree³¹ in which it is assumed that each electron moves in a central or spherically symmetric force field produced by the nucleus and the other $N-1$ electrons of the N electron atomic system. In the SCF procedure the instantaneous action of each electron, j ($=1, 2, \dots, i-1, i+1, \dots, N$), on the i^{th} electron is replaced by the averaged charge distribution of each j^{th} electron, averaged by taking the quantity $\psi_j^* \psi_j$ and summing over all electrons in the atom. This resultant charge distribution is very nearly spherically symmetric. The first approximation made is to assume such a spherically averaged charge distribution so that each electron in the atom can be assumed to move in a central potential field which is written in the form

$$V_i(\vec{r}) = \frac{1}{4\pi} \int \left[-\frac{Z}{r} + \sum_{j \neq i} \int d^3\vec{r}_j \frac{|\psi_j(\vec{r}_j)|^2}{|\vec{r} - \vec{r}_j|} \right] d\Omega \quad (31)$$

Here, $V_i(\vec{r})$ is the potential seen by the i^{th} electron at position \vec{r} . The first term in equation (31) describes the nuclear potential seen by the i^{th} electron at a distance r from the nucleus taken as origin. The second term is the potential due to all other electrons at positions \vec{r}_j . Integration over angles depicts the spherical averaging of the potential. The use of the central potential, $V_i(\vec{r})$, guarantees that the Hamiltonian is spherically symmetric and hence that the solution, $\psi_i(\vec{r})$, of the Schrodinger equation for the i^{th} electron

$$\left[-\frac{1}{2} \nabla^2 + V_i(\vec{r}) \right] \psi_i(\vec{r}) = \epsilon_i \psi_i(\vec{r}) \quad , \quad (32)$$

can be written in the form

$$\begin{aligned} \psi_i(\vec{r}) &= \psi_{nlm}(r, \theta, \phi) \\ &= \frac{1}{r} R_{nl}(r) Y_l^m(\theta, \phi) \end{aligned} \quad . \quad (33)$$

Substitution of this one-electron orbital into the Schrodinger equation gives

$$\frac{1}{2} \frac{d^2}{dr^2} R_{nl} + \left[\epsilon_{nl} - V_{nl} - \frac{l(l+1)}{2r^2} \right] R_{nl} = 0 \quad , \quad (34)$$

where the R_{nl} are the radial wave functions for the one-electron orbitals and are subject to the usual boundary conditions that

$$R_{nl}(r) \rightarrow 0 \text{ as } r \rightarrow \infty \quad (34a)$$

and

$$R_{nl}(r) = 0 \text{ at } r = 0 \quad (34b)$$

The method used in solving the SCF problem is based on that of successive approximations. A trial radial wave function, such as a Thomas-Fermi or Hydrogenic radial function, is chosen for each (nl) orbital of the atom. From this trial function, charge densities and potentials are determined and Schrodinger's equation is solved yielding a final radial function. Using this radial function as a starting function, new charge densities and potentials are recalculated and Schrodinger's equation is again solved. Several such iterations are performed until the condition of self-consistency, which demands that the final potential agree with the initial potential used to solve the Schrodinger equation, is satisfied to within some desired tolerance.

The Hartree SCF method yields one electron orbital wave functions which display the proper $n-l-1$ nodes, but these functions suffer from two important defects. First, they lack the proper antisymmetry. Second, the R_{nl} functions are not orthogonal since for each (nl) shell there is a different central potential, V_{nl} , and hence the Schrodinger equations are not the same for all the orbitals.

These problems are overcome in the HF method in a straightforward manner. The wave function of a many-electron system must change sign when the spatial and spin coordinates of any two electrons are interchanged; that is, the wave function defining a many-electron system must be antisymmetric under the exchange of coordinates between any two electrons in the system. It is intuitively obvious that a total wave function of the form $\Psi(\vec{r}_1, \dots, \vec{r}_n) = \psi_1(\vec{r}_1)\psi_2(\vec{r}_2)\dots\psi_n(\vec{r}_n)$, where the $\psi_i(\vec{r}_i)$ are one-electron orbital wave functions, as assumed by the

Hartree method, does not satisfy the required antisymmetry property. However, a total wave function in the form of a determinantal function, or a linear combination of determinantal functions, does satisfy this requirement. Each of the one-electron wave functions, ψ_i , are written as a product of an orbital function, $\phi_i(\vec{r}_k)$, and a spin function, $\chi_i(\sigma_k)$, where $\sigma_k = \pm 1$ such that for an electron with spin up ($m_s = +1/2$), $\chi_i(1) = 1$ and $\chi_i(-1) = 0$. Similarly, for electron of opposite spin ($m_s = -1/2$), $\chi_i(1) = 0$ and $\chi_i(-1) = 1$. To simplify the calculations one demands that all spin-orbitals, $\chi_i(\vec{r}_k, \sigma_k)$, be orthogonal. The total wave function for the N electron system is

$$\Psi(\vec{r}_1\sigma_1, \vec{r}_2\sigma_2, \dots, \vec{r}_N\sigma_N) = \frac{1}{\sqrt{N!}} \begin{vmatrix} \psi_1(\vec{r}_1\sigma_1) & \dots & \psi_1(\vec{r}_N\sigma_N) \\ \vdots & & \vdots \\ \psi_N(\vec{r}_1\sigma_1) & \dots & \psi_N(\vec{r}_N\sigma_N) \end{vmatrix} \quad (35)$$

which is normalized, provided the $\psi_i(\vec{r}_k\sigma_k)$ are normalized, and is automatically antisymmetric under the interchange of electron coordinates, thereby satisfying the Pauli exclusion principle.

The non-relativistic Hamiltonian for the N electron system is

$$H = - \sum_{i=1}^N \left(\frac{1}{2} \nabla_i^2 + \frac{Z}{r_i} \right) + \sum_{i>j} \frac{1}{r_{ij}} \quad (36)$$

where r_i is the distance of the i^{th} electron from the nucleus taken as origin; r_{ij} is the distance between the i^{th} and j^{th} electrons and ∇_i^2 refers to differentiation with respect to the coordinates of the i^{th} electron. The terms in the first sum of the Hamiltonian involve only single electron operators; those in the second sum operate on pairs of electrons. Defining the one electron operator by

$$f_i = -\frac{1}{2} \nabla_i^2 - \frac{Z}{r_i}$$

and the two electron operator by

$$g_{ij} = \frac{1}{r_{ij}}$$

allows the Hamiltonian in equation (36) to be written in the more compact form

$$H = \sum_i f_i + \sum_{i>j} g_{ij} \quad (37)$$

In the process of calculating atomic structure properties one is concerned with the evaluation of matrix elements involving these one and two electron operators. A detailed analysis of the matrix components is discussed by Slater³² and by Bethe and Jackiw³³.

The Hartree-Fock equations for the spin-orbitals are derived using a variational procedure in which we demand that the energy remain stationary with respect to variations in each of the spin-orbitals. This derivation has been established in detail in many textbooks^{33, 34} hence, only the final form is reproduced here. The spin-orbitals are written as the product of an orbital function and a spin function, namely

$$\psi_i(\vec{r}_j \sigma_j) = \psi_i(\vec{r}_j) \chi_i(\sigma_j)$$

The non-relativistic HF equation is then

$$\begin{aligned}
& -\frac{1}{2} \nabla_1^2 \psi_i(\vec{r}_1) - \frac{Z}{r_1} \psi_i(\vec{r}_1) + \left[\sum_{j=1}^N \int d\tau_2 \frac{1}{r_{12}} |\psi_j(\vec{r}_2)|^2 \right] \psi_i(\vec{r}_1) \\
& - \sum_{j=1}^N \delta_{m_{s_1}, m_{s_j}} \left[\int d\tau_2 \frac{1}{r_{12}} \psi_j^*(\vec{r}_2) \psi_i(\vec{r}_2) \right] \psi_j(\vec{r}_1) = \epsilon_i \psi_i(\vec{r}_1) \quad (38)
\end{aligned}$$

where the sums over j run over all occupied spin-orbitals and integrations over $d\tau_2$ are only over the spatial coordinates. Implicit use of the result that $\sum_{\sigma_j} \chi_i^*(\sigma_j) \chi_j(\sigma_j) = \delta_{m_{s_1}, m_{s_j}}$ has been used in equation (38). By Koopman's theorem³⁵ ϵ_i represents the energy required to remove the i^{th} electron from the atom under the assumption that the remaining core electron orbitals remain unchanged. The HF equations given by equation (38) have N solutions representing the N occupied spin-orbitals comprising the total determinantal wave function representing the state of the N -electron system. With the spin-orbitals normalized, it can be shown that the N solutions are orthonormal³⁶. In addition to the N solutions representing the occupied spin-orbitals there also exist an infinite number of other orthonormal spin-orbitals, $\psi_k(\vec{r}_j, \sigma_k)$, which are solutions of the HF equation and correspond to the unoccupied, or excited orbitals that are found in the central field problem³⁶.

The HF equation differs from that of Hartree's SCF equation (32) only in the terms given by

$$\begin{aligned}
& \int d\tau_2 \frac{1}{r_{12}} |\psi_j(\vec{r}_2)|^2 \psi_j(\vec{r}_1) - \sum_j \delta_{m_{s_1}, m_{s_j}} \left[\int d\tau_2 \frac{1}{r_{12}} \psi_j(\vec{r}_2) \psi_i(\vec{r}_2) \right] \psi_j(\vec{r}_1) \\
& = - \sum_{j \neq i} \delta_{m_{s_1}, m_{s_j}} \left[\int d\tau_2 \frac{1}{r_{12}} \psi_j^*(\vec{r}_2) \psi_i(\vec{r}_2) \right] \psi_j(\vec{r}_1) \quad (39)
\end{aligned}$$

The second term on the left-hand side of equation (39) is the exchange integral which corrects for the fact that the electron does not act on itself. That part of the HF equation in agreement with the SCF equation

represents an electron moving in an equivalent, local potential generated by the nucleus and the other electrons in the system. The exchange term, represented by the right-hand side of equation (39), is the result of a non-local potential. The HF equation (38) can be written in the form

$$\frac{1}{2} \nabla^2 \psi_i(\vec{r}_1) + [\epsilon_i - V(\vec{r}_1)] \psi_i(\vec{r}_1) - \int U(\vec{r}_1, \vec{r}_2) \psi_i(\vec{r}_2) d\tau_2 = 0 \quad (40)$$

where $V(\vec{r}_1)$ is the local potential defined by

$$V(\vec{r}) = -\frac{Z}{r_1} + \sum_{j=1}^N \int |\psi_j(\vec{r}_2)|^2 \frac{1}{r_{12}} d\tau_2 \quad (41)$$

and $U(\vec{r}_1, \vec{r}_2)$ is the non-local potential defined by

$$U(\vec{r}_1, \vec{r}_2) = -\frac{1}{r_{12}} \sum_{j=1}^N \psi_j^*(\vec{r}_2) \psi_j(\vec{r}_1) \delta_{m_{s_1}, m_{s_j}} \quad (42)$$

with the condition that $U(\vec{r}_1, \vec{r}_2)$ be hermitian. This guarantees that the eigenvalues, ϵ_i , of the HF equation (40) are real³⁶. Using standard techniques and the hermitian property of the non-local potential $U(\vec{r}_1, \vec{r}_2)$, it can further be shown that the orbital functions, $\psi_i(\vec{r})$, are orthogonal³⁶. Since the orbitals, $\psi_i(\vec{r})$, are normalized the general result may be written as

$$\int \psi_k^*(\vec{r}_1) \psi_l(\vec{r}_1) d\tau_1 = \delta_{l,k} \quad (43)$$

(1.1) Discussion of the Exchange Term

The exchange term given by the second term on the left-hand side of equation (39) may be regarded as representing the potential energy, at position \vec{r}_1 for the i^{th} electron, of a charge distribution at position \vec{r}_2 of magnitude

$$\sum_j \delta_{m_{s_1}, m_{s_j}} \psi_j^*(\vec{r}_2) \psi_i(\vec{r}_2) \quad (44)$$

Multiplying the top and bottom by $\psi_i^*(\vec{r}_1) \psi_i(\vec{r}_1)$ gives

$$\frac{\sum_j \delta_{m_{s_1}, m_{s_j}} \psi_i^*(\vec{r}_1) \psi_j^*(\vec{r}_2) \psi_j(\vec{r}_1) \psi_i(\vec{r}_1)}{\psi_i^*(\vec{r}_1) \psi_i(\vec{r}_1)} \quad (45)$$

This term is called the exchange charge density and has a total magnitude equal to one electronic charge ($e = 1$ in a.u.) provided ψ_i is one of the occupied orbitals, but is zero if ψ_i represents an unoccupied orbital. To prove this, integrate equation (45) over $d\tau_2$. By the orthogonality of the ψ_i 's and ψ_j 's all terms in the summation go out on integration except for the term $j = i$ which integrates to unity, provided ψ_i is an occupied orbital. Since the summation is only over the N occupied orbitals, no term with $j = i$ exists for any ψ_i which is an excited orbital. The inclusion of the Kronecker delta in equation (45) indicates that the electronic charge density is comprised of charges having the same spin as the spin-orbital ψ_i being considered. When the position \vec{r}_2 is identical to position \vec{r}_1 then equation (45) reduces to

$$\sum_j \delta_{m_{s_1}, m_{s_j}} \psi_j^*(\vec{r}_1) \psi_j(\vec{r}_1) \quad (46)$$

which is the total density of all electrons of the same spin as the i^{th} electron at the position \vec{r}_1 .

The complete potential energy field in which the electron moves, as given by the HF theory, is composed of the nuclear interaction, the interaction with all electrons of opposite spin to that of the i^{th} electron and an interaction with a charge distribution of electrons of

the same spin as the i^{th} electron. The charge distribution of electrons with the same spin adds up to one less than the total number of electrons in this spin state if an occupied-orbital is considered, or in the case of an unoccupied orbital, where the net amount of charge is zero, the charge distribution adds to the total number of electrons with spin the same as that of the i^{th} electron being considered. The net charge density of electrons with the same spin as the spin-orbital, ψ_i , is zero in the case where $\vec{r}_2 = \vec{r}_1$ since here the exchange charge density just cancels the total density of all electrons of this spin.

Slater³⁰ interprets the exchange effects in terms of the Fermi hole theory in which the electron being considered is thought of as carrying with it a hole charge such that for an occupied orbital there is electronic charge with a total magnitude of one unit removed from the immediate vicinity of the electron's position. This Fermi hole is a result of the Pauli Exclusion principle which keeps electrons of the same spin separated. Hence, the effect of the exchange term is to remove electronic charge from the immediate neighbourhood of the i^{th} electron. This tends to reduce the electron-electron Coulomb repulsion and therefore lowers the one-electron energies ϵ_i relative to the Hartree SCF values. Slater³⁷ further points out that the exchange corrections are greatest in the region of large charge density near the nucleus.

The overall effect of exchange is therefore to cause the atom to shrink, with an increased electron density near the nucleus.

(1.2) Simplification of the Exchange Term: The Slater Approximation

Although solutions of the HF equation may be obtained in a straightforward manner similar to the procedure followed in the SCF method, the

occurrence of all the spin-orbitals in each of the equations yields a very complicated system of simultaneous integro-differential equations whose solutions require much effort. Slater³⁰ has proposed an approximation which involves the formation of an average potential field in which all of the electrons are considered to move. The potential field is found by forming a weighted mean of the exchange charges, weighted and averaged over the various electronic wave functions at a given point in space. This average exchange charge is then replaced by the corresponding value given by a free-electron gas whose local density is equal to the density of actual charge at the given spatial point.

Slater's exchange correction as described above is obtained by writing the HF exchange operator in the form of an equivalent potential acting on the i^{th} wave function

$$V_{\text{exch}}(\vec{r}_1) = - \frac{\sum_{j=1}^N \int \psi_i^*(\vec{r}_1) \psi_j^*(\vec{r}_2) \frac{1}{r_{12}} \psi_j(\vec{r}_1) \psi_i(\vec{r}_2) d\tau_2}{\psi_i^*(\vec{r}_1) \psi_i(\vec{r}_1)} \quad (47)$$

where it is understood that integration over $d\tau_2$ is only over the spatial coordinates and where summation over the spin functions has been carried out implicitly.

The atomic electrons are regarded as a degenerate electron gas with the same electron charge density as that of the exchange density. The electrons are assumed to experience a constant potential and hence the corresponding orbital functions can be written as plane waves

$$\psi_j(\vec{r}_1) = \frac{1}{\sqrt{V/2}} \exp(i\vec{k}_j \cdot \vec{r}_1) \quad (48)$$

where \vec{k}_j is the wave vector of the j^{th} wave; \vec{r}_1 is the radial vector from the origin at the nucleus to the j^{th} electron; V is the volume of the electron gas over which the normalization is carried out. Using the plane wave expression of equation (48) for the orbital wave functions, the numerator of equation (47) becomes

$$\psi_i^*(\vec{r}_1)\psi_j^*(\vec{r}_2) \frac{1}{r_{12}} \psi_j(\vec{r}_1)\psi_i(\vec{r}_2) = \frac{1}{V^2} \exp [i(\vec{k}_j - \vec{k}_i) \cdot (\vec{r}_1 - \vec{r}_2)] \frac{1}{r_{12}} \quad (49)$$

and the denominator becomes

$$\psi_i^*(\vec{r}_1)\psi_i(\vec{r}_1) = \frac{1}{V} \quad (50)$$

The exchange potential energy of the interaction of the electron at \vec{r}_1 and the electron at \vec{r}_2 in the states i and j is found by first computing the integral

$$\int \psi_i^*(\vec{r}_1)\psi_j^*(\vec{r}_2) \frac{1}{r_{12}} \psi_j(\vec{r}_1)\psi_i(\vec{r}_2) d\tau_2 = \frac{1}{V^2} \int \exp [i(\vec{k}_j - \vec{k}_i) \cdot (\vec{r}_1 - \vec{r}_2)] \frac{1}{r_{12}} d\tau_2 \quad (51)$$

The result of the integration is³⁶

$$\frac{4\pi}{V^2 |\vec{k}_j - \vec{k}_i|} \quad (52)$$

This result is next summed over the \vec{k}_i which is shown to be equivalent to integration over spherical coordinates in the \vec{k}_i space³⁶. The HF exchange potential then takes on the form

$$V_{\text{exch}} = -4 \left(\frac{3}{8\pi} \frac{N}{V} \right)^{1/3} F(\eta) \quad (53)$$

where N is the number of electrons in the volume V of the Fermi gas and η is defined by

$$\eta = \frac{|\vec{k}_i|}{k_{\text{max}}}$$

where \vec{k}_i is the momentum of the electron (in a.u.) in the orbital ψ_i for which the exchange potential is being evaluated and k_{max} is the Fermi momentum. The value of k_{max} is found by equating the equivalent expressions for the Fermi energy, namely

$$E_f = \frac{\hbar^2}{2m} \left(\frac{3}{8\pi} \frac{N}{V} \right)^{2/3} = \frac{k_{\text{max}}^2 \hbar^2}{2m} \quad (\text{in cgs units})$$

or

$$E_f = 2\pi^2 \left(\frac{3}{8\pi} \frac{N}{V} \right)^{2/3} = \frac{k_{\text{max}}^2}{2} \quad (\text{in a.u.})$$

from which the Fermi momentum is determined to be

$$k_{\text{max}} = \left(3\pi^2 \frac{N}{V} \right)^{1/3} \quad (54)$$

The function $F(\eta)$ is given by³⁰

$$F(\eta) = \frac{1}{2} + \frac{1-\eta^2}{4\eta} \ln \left(\frac{1+\eta}{1-\eta} \right) \quad (55)$$

which has a range of values in the interval $[1/2, 1]$ where the extrema occur at $\eta = 0$ for an electron of zero kinetic energy and $\eta = 1$ for an electron with kinetic energy corresponding to that of an electron at the top of the Fermi distribution, respectively.

Equation (53) expresses the HF exchange energy in an electron gas for an electron whose momentum is the fraction, η , of the maximum Fermi energy. Slater's average exchange potential is found by averaging equation (53) over occupied states. The number of occupied states between η and $\eta + d\eta$ is proportional to $\eta^2 d\eta$; hence, the average value of $F(\eta)$ is

$$\begin{aligned}\bar{F}(\eta) &= \frac{\int \eta^2 F(\eta) d\eta}{\int \eta^2 d\eta} \\ &= \frac{3}{4}\end{aligned}\quad (56)$$

The averaged exchange potential is thus

$$\bar{V}_{\text{exch}} = -\frac{3}{2} \left(\frac{3}{\pi} \frac{N}{V} \right)^{1/3} \quad (57)$$

With this approximate exchange potential the Hartree-Fock equation becomes

$$\begin{aligned}-\frac{1}{2} \nabla_1^2 \psi_1(\vec{r}_1) - \frac{Z}{r_1} \psi_1(\vec{r}_1) + \left[\sum_{j=1}^N \int d\tau_2 \frac{1}{r_{12}} |\psi_j(\vec{r}_2)|^2 \right] \psi_1(\vec{r}_1) \\ - \frac{3}{2} \left[\frac{3}{\pi} \sum_{j=1}^N \psi_j^*(\vec{r}_1) \psi_j(\vec{r}_1) \delta_{m_{s_1}, m_{s_j}} \right]^{1/3} \psi_1(\vec{r}_1) = \epsilon_1 \psi_1(\vec{r}_1)\end{aligned}\quad (58)$$

where the term $\frac{N}{V}$ in equation (59) has been interpreted as the local

density of electrons given by

$$\frac{N}{V} = \sum_{j=1}^N \delta_{m_{s_1}, m_{s_j}} \psi_j^*(\vec{r}_1) \psi_j(\vec{r}_1) \quad (59)$$

which is the exchange charge density of equation (45) for position $\vec{r}_2 = \vec{r}_1$. This Slater exchange approximation is used in the subsequent work.

(2.0) Extension to the Relativistic Case

In this section we are concerned with the generation of the relativistic formalism of the Hartree-Fock-Slater (HFS) approximation established for the non-relativistic case in the above discussion.

The relativistic single-particle wave function for a central-field potential can be written as

$$\Psi_1 = \begin{pmatrix} (G_1/r) \chi_{\kappa}^{\mu} \\ i(F_1/r) \chi_{-\kappa}^{\mu} \end{pmatrix} \quad (60)$$

where the radial functions multiplied by r , $F(\vec{r})$ and $G(\vec{r})$, are real with $G(\vec{r})$ being the large component. The non-zero integer, κ , specifies both the total angular momentum

$$j = |\kappa| - 1/2 \quad (61)$$

and the parity of the state, $(-1)^l$, where

$$l = j + \frac{\kappa}{2|\kappa|} \quad (62)$$

The projection of the total angular momentum, j , on the axis of quantization is designated by μ . The spin-angular function, χ_{κ}^{μ} , is defined by

$$\chi_{\kappa}^{\mu} = \sum_m C(1/2 j; \mu-m, m) x_{1/2}^m Y_l^{\mu-m} \quad (63)$$

with $x_{1/2}^{1/2} = \begin{pmatrix} 1 \\ 0 \end{pmatrix}$ and $x_{1/2}^{-1/2} = \begin{pmatrix} 0 \\ 1 \end{pmatrix}$. In defining the Clebsch-Gordon

coefficient, $C(11/2j; \mu-m, m)$, we use the notation of Rose³⁸.

We still seek solutions of the eigenvalue problem

$$H\Psi = E\Psi$$

as in the non-relativistic case, but for the relativistic problem we must replace the non-relativistic Hamiltonian by the Dirac Hamiltonian, namely

$$H \rightarrow H_D \equiv \vec{\alpha} \cdot \vec{p} + \beta + V(r) \quad (64)$$

Here we define the operators

$$\vec{\alpha} = \begin{pmatrix} 0 & \vec{\sigma} \\ \vec{\sigma} & 0 \end{pmatrix} \text{ and } \beta = \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}$$

where 1 is the 2x2 unit matrix and $\vec{\sigma}$ is the 2x2 Pauli matrix.

The eigenvalue problem being considered for the relativistic case is

$$H_D \Psi_i = \epsilon_i \Psi_i$$

with $E = mc^2 + \epsilon_i$ and ϵ_i is the ionization potential of the i^{th} atomic shell. It can be shown³⁹ that for Ψ_i to be a solution, the two radial functions $G(\vec{r})$ and $F(\vec{r})$ must satisfy the coupled differential equations

$$\frac{d}{dr_1} G_i(\vec{r}_1) + \frac{\kappa}{r_1} G_i(\vec{r}_1) - \left(2c + \frac{V(\vec{r}_1)}{c} - \frac{\epsilon_i}{c} \right) F_i(\vec{r}_1) = 0$$

and

$$\frac{d}{dr_1} F_i(\vec{r}_1) - \frac{\kappa}{r_1} F_i(\vec{r}_1) + \left(\frac{V(\vec{r}_1)}{c} - \frac{\epsilon_i}{c} \right) G_i(\vec{r}_1) = 0 \quad (65)$$

in atomic units. $V(r_1)$ is the effective potential producing the field which affects the optical electron. This potential can be written as

$$\begin{aligned} V(r_1) &= -\frac{Z}{r_1} + \sum_j N_j \int d\tau_2 \frac{1}{r_{12}} |\psi_j(\vec{r}_2)|^2 + V_{\text{exch}}(\vec{r}_1) \\ &= -\frac{Z}{r_1} + \int d\tau_2 \frac{1}{r_{12}} \rho(\vec{r}_2) + V_{\text{exch}}(\vec{r}_1) \end{aligned} \quad (66)$$

where the charge density has the form

$$\rho(\vec{r}_2) = \sum_j N_j \left[G_j^2(\vec{r}_2) + F_j^2(\vec{r}_2) \right] \quad (67)$$

and N_j is the occupation number of the j^{th} shell. The expansion of $\frac{1}{r_{12}}$ in terms of Legendre polynomials gives⁴⁰

$$\frac{1}{r_{12}} = \frac{1}{|\vec{r}_2 - \vec{r}_1|} = \sum_{l=0}^{\infty} \frac{r_{<}^l}{r_{>}^{l+1}} P_l(\cos\theta_{12}) \quad (68)$$

where $r_{<}$ is the smaller of r_1 and r_2 . Under the central-field approximation where an averaging over angles is performed, only the $l=0$ term survives. Using equation (68) in equation (66) under the assumption of a central-field, the potential assumes the form

$$V(\vec{r}_1) = -\frac{Z}{r_1} + \frac{1}{r_1} \int_0^{r_1} r_2^2 \rho(\vec{r}_2) d\tau_2 + \int_{r_1}^{\infty} r_2 \rho(\vec{r}_2) d\tau_2 + V_{\text{exch}}(\vec{r}_1) \quad (69)$$

The solutions of equations (65) must satisfy the boundary conditions

$$\lim_{r \rightarrow 0} G(\vec{r}) = \lim_{r \rightarrow 0} F(\vec{r}) = \lim_{r \rightarrow \infty} G(\vec{r}) = \lim_{r \rightarrow \infty} F(\vec{r}) = 0 \quad (70)$$

Since, in general, the experimental ionization energy of the i^{th} shell is not the true eigenvalue of equations (65), there is no solution of this system of equations that can simultaneously satisfy the boundary conditions of equation (70) at the origin and at infinity. The procedure to solve equations (65) such that the boundary conditions are satisfied, as outlined by Rose⁴¹, is to establish solutions that are regular at $r = 0$. Such solutions satisfying this condition are given as truncated power series expansions in the form

$$G(r) = r^{\gamma} \sum_{m=0}^{m_{\max}} \alpha_m r^m$$

and

$$F(r) = r^{\gamma} \sum_{m=0}^{m_{\max}} \beta_m r^m \quad (71)$$

where $\gamma = \pm(\kappa^2 - (Z/c)^2)^{1/2}$. These expansions are valid only for small r . In the asymptotic region, the solutions are of the form

$$G(r) = a e^{-\mu r}$$

and

$$F(r) = b e^{-\mu r} \quad (72)$$

where $\mu = [2\epsilon - (\epsilon/c)^2]^{1/2}$ and $a/b = [(2c^2 - \epsilon)/\epsilon]^{1/2}$. The numerical solution of equations (65) is begun from the region of small r values using the power series form of equations (71); a similar numerical solution is begun from the region of large r values using the asymptotic form of equations (72) appropriate for the region. By adjusting the coefficients in equations (71) and (72) it is possible to match the solutions for the two regions at some point r_1 to within any desired degree of accuracy. If the solutions in the two regions match at the predetermined point, r_1 , (usually taken in the neighbourhood of the

classical turning point) then they coincide at all values of r and the solution obtained is an eigensolution⁴¹.

A relativistic HFS program, based on the general procedure outlined above, was used to generate the one-electron wave functions required for the calculations of this section.

As an initial estimate of the eigenvalues for each (nl) shell the Dirac hydrogenic energies

$$E_{(nl)} = \frac{-Z^2}{N^2 + N(n - |\kappa| + \gamma)} \quad (73)$$

were used. N is the apparent principal quantum number for the particular (nl) shell defined by

$$N = n^2 - 2(n - |\kappa|)(|\kappa| - \gamma) \quad ; \quad (74)$$

$\gamma = (|\kappa|^2 - (\alpha Z)^2)^{1/2}$ and n is the principal quantum number.

In principle, the wave function describing a given atomic (nl) shell has an infinite radial extent. The numerical representation of these wave functions make it impossible to depict this feature since the radial wave function must be terminated at some sufficiently large radial displacement from the nucleus. The error introduced by neglecting the tail of the wave function can be made negligible by choosing a sufficiently large termination point. The choice of such a suitable termination point for each (nl) shell can be made by considering the asymptotic radial function⁴²

$$R(r) = e^{-\lambda r} f(r) \quad (75)$$

where $f(r)$ is a function which varies slowly at large r and λ is related

to the approximate eigenenergy, $\epsilon_{(nl)}$, being used at a given stage in the solution through the relation

$$\lambda = + \sqrt{-\epsilon_{(nl)}} \quad (76)$$

In our program, the radial wave function was cut off at the point r_{\max} determined by the requirement that the asymptotic form of the radial function in equation (75), with $\epsilon_{(nl)}$ being the approximate eigenenergy, agree with the asymptotic form in which $\epsilon_{(nl)}$ is replaced with the Coulomb energy for that shell to within one part in 10^9 in the region of large radial displacements. This condition is satisfied at some sufficiently large radial displacement, r_{\max} , determined by the relation

$$\exp \left[\left(-\epsilon_{(nl)} + \frac{Z}{r_{\max}} \right)^{1/2} r_{\max} \right] \leq 10^{-9} \quad (77)$$

The numerical solution of equations (65) was begun in the region of large r values for which the asymptotic form of equations (72) is valid. The inward solution began at r_{\max} and was continued up to the classical turning point. Here, the inward solution was terminated and replaced by the outward solution beginning at the origin. The asymptotic form given by equations (71) was used in this region with the solution being carried out to the classical turning point. The ratio formed by comparing the large component values at the position of the classical turning point for the outward and inward integration was used to force the inward solution for the large and small components to match those for the outward solution. The number of nodes appearing in the large component, $G(r)$, of this approximate wave function were counted and

compared to the correct number of nodes, n_r , given by⁴³

$$\begin{aligned} n_r &= (n - |k|) \\ &= (n - l - 1) \end{aligned} \quad (78)$$

where n is the principal quantum number for the (nl) shell. There will be more than n_r nodes in $G(r)$ if the approximate eigenvalue, $\epsilon_{(nl)}$, is too small and fewer than n_r nodes if $\epsilon_{(nl)}$ is too large. This technique provides a useful method for bounding the correct eigenvalue. The energy parameter, $\epsilon_{(nl)}$, so bounded can then be adjusted and the process repeated until the correct number of nodes appear.

Upon securing the proper nodal structure, the initial approximation to the wave function was normalized according to the requirement

$$\int (G(r)^2 + F(r)^2) dr = 1 \quad (79)$$

New charge densities and potentials, which include the exchange effects through the Slater potential, were calculated from this initial approximate wave function. A new approximate eigenenergy was calculated from these new charge densities and potentials and the entire process repeated until the inward and outward integrations matched to within at least one part in 10^7 . This procedure was performed for each occupied (nl) shell of the atomic configuration under study. The final radial wave functions for the configuration were separately normalized and the predicted eigenvalue for each shell⁶ was calculated. Note that since the core is allowed to readjust for each excited state configuration, no guarantee is made that the outer shell radial wave functions between different excited states with the same total angular momentum are orthogonal even though each outer shell is orthogonal to all core shells in the configuration. More discussion of this point will appear later.

The numerical solutions of the first order differential equations (65) were accomplished using the fifth order Adams predictor-corrector method⁴⁴. Equations (65) were solved at points r_i determined by the relation

$$r_i = r_0 \left[\exp(h(i-1)) - 1 \right] \quad (80)$$

where i had a maximum value of 384 and the scaling factor, r_0 , was chosen in such a manner that the wave function cut off point determined by equation (77) was less than, but close to the radial displacement given by r_{384} from equation (80). In general, r_0 was on the order of 10^{-3} a.u. when the step size h was taken to be $1/32$.

(3.0) Calculations of Transition Probabilities

The transition probabilities and oscillator strengths may be obtained from the formulae rigorously derived by Bhalla⁴⁵ using the relativistic single-configuration wave functions from the HFS calculations.

The spontaneous emission probability per unit time ($\tau_0 = \frac{4}{3} \frac{1}{\omega^3} \text{me}^4 = 2.4189 \times 10^{-17} \text{sec}$) for the EL transition $\beta \rightarrow \alpha$ is given by Grant⁴⁶ to be

$$A_{\beta \rightarrow \alpha} = \left[J_{\beta} \right]^{-1} \sum_{m_{\beta}} \sum_{m_{\alpha}} 2\pi |M_{\alpha\beta}^e|^2 \quad (81)$$

with $M_{\alpha\beta}^e$ defined by the following relation involving "3-J" symbols:

$$M_{\alpha\beta}^e = \left(\frac{\omega}{\pi c} \right)^{1/2} (-1)^{J_{\alpha} - m_{\alpha}} \begin{pmatrix} J_{\alpha} & L & J_{\beta} \\ -m_{\alpha} & M & m_{\beta} \end{pmatrix} \\ \times (-1)^{J_{\alpha} - 1/2} \left[J_{\alpha}, J_{\beta} \right]^{1/2} \begin{pmatrix} J_{\alpha} & L & J_{\beta} \\ 1/2 & 0 & -1/2 \end{pmatrix} \bar{M}_{\alpha\beta}^e \quad (82)$$

where $\bar{M}_{\alpha\beta}^e$ is given as

$$\begin{aligned} \bar{M}_{\alpha\beta}^e = i^L & \left[\left(\frac{L}{L+1} \right)^{1/2} \left[(k_\alpha - k_\beta) i_{L+1}^+ + (L+1) \bar{i}_{L+1} \right] \right. \\ & \left. - \left(\frac{L+1}{L} \right)^{1/2} \left[(k_\alpha - k_\beta) i_{L-1}^+ - L \bar{i}_{L-1} \right] \right] \end{aligned} \quad (83)$$

In the velocity form and by

$$\begin{aligned} \bar{M}_{\alpha\beta}^e = i^L & \left[(2L+1) J^{(L)} \left(\frac{\omega r}{c} \right) + (k_\alpha - k_\beta) (i_{L+1}^+ + i_{L-1}^+) \right. \\ & \left. - L \bar{i}_{L-1} + (L+1) \bar{i}_{L+1} \right] \end{aligned} \quad (84)$$

In the length form. In equations (82) and (84) the transition frequency, ω , is the energy difference between the upper and lower states expressed in a.u. and L is the order of the electric multipole which in our present case for dipole transitions has the value $L=1$. Also, the notation⁴⁷

$$[J_\alpha, J_\beta, \dots] = (2J_\alpha + 1)(2J_\beta + 1) \dots \quad (85)$$

is used. The values of k for the upper and lower states are as defined in equation (62). The radial integrals i_λ^+ , \bar{i}_λ and $J^{(L)} \left(\frac{\omega r}{c} \right)$ are given by

Bhalla⁴⁵ to be

$$i_\lambda^+ \equiv \int J_\lambda \left(\frac{\omega r}{c} \right) (G_\alpha F_\beta + F_\alpha G_\beta) dr \quad (86a)$$

$$\bar{i}_\lambda \equiv \int J_\lambda \left(\frac{\omega r}{c} \right) (G_\alpha F_\beta - F_\alpha G_\beta) dr \quad (86b)$$

and

$$J^{(L)} \left(\frac{\omega r}{c} \right) \equiv \int J_L \left(\frac{\omega r}{c} \right) (G_\alpha G_\beta + F_\alpha F_\beta) dr \quad (86c)$$

where $J_\lambda \left(\frac{\omega r}{c} \right)$ is the spherical Bessel function of order λ and $G_{\alpha(\beta)} F_{\alpha(\beta)}$

are the large and small components of the relativistic HFS wave function for the lower (upper) state.

Performing the summation over initial states and averaging over final states in the expression for the spontaneous emission transition probability per unit time reduces equation (81) to the form

$$A_{\beta \rightarrow \alpha} = 2\alpha\omega \begin{matrix} [J_{\alpha}] \\ [L] \end{matrix} \begin{pmatrix} J_{\beta} & L & J_{\alpha} \\ 1/2 & 0 & -1/2 \end{pmatrix}^2 |\bar{M}_{\alpha\beta}^e|^2 \quad (87)$$

with $\bar{M}_{\alpha\beta}^e$ being of the appropriate form for either the velocity or length matrix element given by equations (83) and (84) respectively. Grant⁴⁶ proves that for all pairs of states α and β which are exact eigensolutions of the same Dirac Hamiltonian, the length and velocity forms yield identical results. Hence, the comparison of matrix element calculations using the two formulations provides a check on the accuracy of the subsequent calculations of the atomic transition parameters and indirectly yields an estimate of the quality of the approximation method used to generate the wave functions for the particular system under study.

The absorption oscillator strength is defined in terms of the spontaneous emission transition probability per unit time for the transition $\beta \rightarrow \alpha$ by

$$f_{\alpha\beta} = \frac{[J_{\beta}]}{[J_{\alpha}]} \frac{A_{\beta \rightarrow \alpha}}{2\alpha^3\omega^2} \quad (88)$$

The numerical values of ω are computed from the C.E. Moore tables⁴⁸. (A misprint occurs in Grants equation (5.2) where α appears in the denominator instead of the correct α^3 .)

Considerable controversy exists concerning which form of the matrix

element, $\bar{M}_{\alpha\beta}^e$, should be used to obtain the most accurate representation of the atomic oscillator strengths. Starace⁴⁹ argues that the length form should be preferred over the velocity form if the wave functions are exact eigenfunctions of an approximate non-local Hamiltonian such as is the case in the HF approximation. On the other hand, Grant⁴⁶ considers the gauge invariance of the transition matrices and infers that for electric dipole transitions the dipole velocity form should be given priority. In this investigation all calculations are presented in both forms for the sake of comparison.

(3.1) The Bates-Damgaard Approximation

Several past investigations^{17, 23} related to sensitized fluorescence studies in mercury-thallium vapour mixtures have relied on the results of the Bates-Damgaard²⁴ (BD) approximation for determining the necessary transition probabilities required in the calculations of various cross sections. In the evaluation of such experimental data it becomes necessary to distinguish between trends that are of a physical nature and those introduced through the theoretical calculations of the transition probabilities and lifetimes. As we have mentioned previously, only a few reliable experimental values of transition probabilities and lifetimes exist in the literature, particularly in the case of thallium, and hence the necessity for the theoretical calculation of such parameters. However, because of the inherent dependence of the cross sections on such calculations it is important to choose a theoretical approximation method that not only yields results that agree closely with those found by experiment, but also displays consistent behavior in regions not investigated experimentally. The Bates-Damgaard approximation scheme is introduced here in order to facilitate the comparison of our

experimental cross sections, in which the theoretical relativistic transition probabilities and lifetimes are used, with those of the other authors who have employed the BD results.

Assuming coulombic behavior of the potential outside the core region, Bates and Damgaard proposed to use a solution of the hydrogenic Schrodinger equation,

$$\frac{d^2 R_{nl}(r)}{dr^2} + \left(\frac{2 Z_{\text{eff}}}{r} - \frac{l(l+1)}{r^2} - \epsilon_{nl} \right) R_{nl}(r) = 0 ; \quad (89)$$

where Z_{eff} , the effective charge, is the excess charge on the nucleus when the active electron is removed and the energy parameter, ϵ_{nl} , is taken to be the experimental energy for the particular (nl) orbital. Solutions of equation (89) are subject to the boundary condition

$$R_{nl}(r) \rightarrow 0 \text{ as } r \rightarrow \infty \quad (89a)$$

and are expressed in the form

$$R_{nl}(r) = \left[\frac{n^{*2} \Gamma(n^*+l+1) \Gamma(n^*-l)}{Z_{\text{eff}}} \right]^{-1/2} \times \left(\frac{2rZ_{\text{eff}}}{n^*} \right)^{n^*} \exp\left(\frac{-rZ_{\text{eff}}}{n^*} \right) F\left(-n^*+l+1, 2l+2; \frac{2rZ_{\text{eff}}}{n^*} \right)$$

where n^* is the effective principal quantum number given by

$$n^* = Z_{\text{eff}} / \epsilon_{nl}^{1/2} \quad (90a)$$

and $F\left(-n^*+l+1, 2l+2; \frac{2rZ_{\text{eff}}}{n^*} \right)$ is the confluent hypergeometric function.

The asymptotic expansion of $F(a,b;x)$ is

$$F(a,b;x) = x^{-a} \left[\sum_{n=0}^{R-1} \frac{(a)_n (1+a-b)_n (-x)^{-n}}{n!} + O(x^{-R}) \right] \quad (91)$$

Thus equation (90) can be written in the more compact form

$$R_{nl}(r) = \left[\frac{n^{*2} \Gamma(n^{*}+1) \Gamma(n^{*}-1)}{Z_{\text{eff}}} \right]^{-1/2} \times \left(\frac{2rZ_{\text{eff}}}{n^{*}} \right)^{n^{*}} \exp \left(\frac{-rZ_{\text{eff}}}{n^{*}} \right) \left[1 + \sum_{v=1}^{v_{\text{max}}} \frac{a_v}{r^v} \right] \quad (92)$$

where

$$a_1 = \frac{n^{*}}{2Z_{\text{eff}}} \left[1(1+1) - n^{*}(n^{*}-1) \right] \quad (92a)$$

and

$$a_v = a_{v-1} \left[\frac{n^{*}}{2vZ_{\text{eff}}} \left[1(1+1) - (n^{*}-v)(n^{*}-v+1) \right] \right] \quad (92b)$$

The asymptotic series in equation (92) diverges for any r . In practice the series is terminated at some value of v_{max} in such a way that the dipole transition integrals are convergent.

The validity of the Coulomb approximation depends critically on the assumption that the true potential can be adequately represented by the $\frac{2Z_{\text{eff}}}{r}$ Coulomb potential. For complex systems, where the outer electrons overlap significantly with the core electrons, the Coulomb approximation would be expected to be less reliable. Generally, the BD approximation gives the best results for the highly excited states in these complex systems. In particular, good results are expected in the calculation of oscillator strengths when the principal maxima of both the upper and lower states occur in the outer region where the Coulomb potential is a good approximation.

Choosing v_{max} in equation (92) such that the transition integrals

converge does not avoid divergence of the wave function as $r \rightarrow 0$, nor does it guarantee the proper $(n-l-1)$ number of nodes. In certain types of calculations, such as in the evaluation of the electric dipole transition matrix elements, these latter two deficiencies in themselves may not introduce serious error. This is because the main contribution to such calculations comes from the region of large radial displacements and the improper behavior of the radial function near the origin will often not seriously affect the matrix elements.

The absorption oscillator strength for E1 transitions between the states β and α is given by⁵⁰

$$f_{\alpha\beta} = \frac{2}{3} |\Delta E| (2J_{\beta} + 1) W^2(l_{\alpha}, l_{\beta}; \frac{1}{2}, \frac{1}{2}) l_{\max} |R_{\alpha\beta}|^2 \quad (93)$$

where ΔE is the energy difference between the states, W is the Racah coefficient, $l_{\max} = \max(l_{\alpha}, l_{\beta})$ and the radial integral for the transition $\beta \rightarrow \alpha$ is

$$R_{\alpha\beta} = \int R_{\alpha} R_{\beta} r \, dr \quad (94)$$

where R_{α} and R_{β} are of the form given by equation (92) appropriate for the lower and upper state respectively. The transition probabilities (in units of sec^{-1}) are determined according to the relation⁵¹

$$A_{\beta \rightarrow \alpha} = 0.6669 |\Delta E|^2 f_{\alpha\beta} \quad (95)$$

where ΔE is in cm^{-1} .

(4.0) Results and Discussion

Numerous regularities and systematic trends among oscillator strengths have been studied in recent years. Such regularities are of great practical importance since they may be used to obtain additional

oscillator strengths by simple extrapolation procedures as well as to evaluate the reliability of existing data by the degree of fit with respect to established systematic trends. Regular variations in spectral line intensities were noted by Trumphy⁵² as early as 1925 in the principal series (3s-np) of sodium. It was observed that the oscillator strengths decrease with increasing principal quantum number, n , in a regular fashion which could be approximated by the general expression

$$\text{constant } \times (n^*)^{-3} \quad (96)$$

where n^* is the effective principal quantum number. Similar trends have been observed in numerous other elements.

Appendix (2) indicates the behavior of the relativistic HFS oscillator strengths, taken from Appendix (1), as a function of the effective principal quantum number of the upper state for several of the Thallium₁ series. Also indicated are the curves generated by the combined experimental data of Norton and Gallagher⁵³ and Penkin and Shabanova⁵⁴. Similar plots for the corresponding series in Cesium₁, formed from the data taken from Appendix (3), along with curves resulting from the experimental values of Agnew⁵⁵ are shown in Appendix (4).

From the plots in Appendices (2) and (4), the oscillator strengths can be seen to exhibit the well known hydrogenlike "constant $\times (n^*)^{-3}$ " behavior in the region of large n^* values. However, for the $7^2S_{1/2} - n^2P_{1/2}$ series in thallium and all series shown in the case of cesium with the exception of the $6^2P_J - n^2S_{1/2}$ series, marked departures from the hydrogenic behavior in the form of pronounced minima are indicated in the theoretical curves at intermediate or low n^* values. These minima are absent from the corresponding experimental curves. The

appearance of the "Cooper minima"^{56,57} are associated with sign reversals in the transition matrix elements along a given series and reflect the non-hydrogenic character of the atomic potentials $V(r)$. These minima, which commonly occur in the continuum, are related to the structure of the lower state wave function. If the wave function for the lower state in the transition series has a node then the transition integrals for a spectral series involving this lower level will change sign somewhere along the series and the oscillator strengths, which are related to the square of the transition matrix elements, will pass through a minimum. There is no general way to fix the n^* value for which this minimum will occur in a given series. In other elements, where the Cooper minima occur in the discrete spectra, such as in the case of Li_I ⁵⁸ and Mg_{II} ⁵⁹, one observes the minima at $n^* = 3$ and $n^* = 4$ respectively.

Comparison of the relativistic HFS curves with those generated from the experimental results show a general lack of agreement in both thallium and cesium with the exception of the first members of each series. This divergence between the theoretical and experimental oscillator strengths with increasing energy separation between the upper and lower states involved in the transitions is further manifested over the entire range of oscillator strengths shown in Appendices (1) and (3) when a comparison is made between the radial matrix elements \bar{M}_{IJ}^e , in the length and velocity forms. In this case there is a definite correlation between the relative percentage difference in the two forms of the transition matrix elements and the energy separation of the states involved in the transition. The relative percentage difference, η , between the two formulations is calculated according to the relation

$$\eta = \left| \frac{|\bar{M}_{IJ}^e| \text{velocity} - |\bar{M}_{IJ}^e| \text{length}}{|\bar{M}_{IJ}^e| \text{velocity} + |\bar{M}_{IJ}^e| \text{length}} \right| \times 100\% \quad (97)$$

In order to distinguish between accidentally small energy differences between low lying states and systematically small energy differences occurring between pairs of higher excited states, we introduce a scaling factor, \bar{n}^2 , defined by

$$\bar{n}^2 = \begin{bmatrix} 1 & -1 \\ n_1^2 & n_2^2 \end{bmatrix}$$

for $n_1 - n_1'$ transitions. It is found for both Thallium, and Cesium, that the relative percentage difference, η , tends to decrease as the scaled energy separation, $\bar{n}^2 \Delta E$, between the participating levels becomes smaller. Such a feature is completely consistent with the divergence between the theoretical and experimental plots in Appendices (2) and (4).

Careful testing of the various computer programmes and subroutines used to generate both the wave functions and oscillator strengths did not indicate the presence of any accumulating error that might explain the observed behavior. It was noticed however, that although the wave functions for the excited states were normalized to at least 1 part in 10^9 , states with the same total angular momentum were generally orthogonal to only a few parts in 10^6 . States with different total angular momentum are necessarily orthogonal through the spin-angular portion of the wave functions.

The entire problem of orthogonality amongst the simplified Hartree-Fock solutions stems from the replacement of the HF exchange potential by the averaged approximate potential. What has been done up to this

point is to solve the simplified HF equations for each excited state configuration in such a manner that the core orbitals are allowed to readjust their radial distributions from that of the ground state configuration. Due to this readjustment of the core, and hence the potential, each configuration is defined by a slightly different Hamiltonian. Therefore, radial wave functions for excited states with the same total angular momentum can not be expected to remain orthogonal as they do in the case of full Hartree-Fock calculations for which all configurations are solved using a consistent Hamiltonian.

What is commonly done to avoid complications arising from the non-orthogonality amongst simplified HF wave functions is to make the physically unrealistic approximation that the core wave functions remain fixed in the ground state configuration and that only the outer most orbital changes when solving for excited levels. This Frozen-Core approximation allows the ground and excited state configurations to be defined by the same Hamiltonian and hence guarantees the orthogonality of all radial wave functions amongst states with the same total angular momentum. Noticable increases in the binding energies of the inner most orbitals occur as one proceeds from the ground state configuration to higher excited state configurations. Within a given configuration the relative change in the orbital binding energy decreases as one proceeds from inner to outer core orbitals with the inner core orbitals of thallium being less affected than those of cesium. The variation in binding energy of the core orbitals is linked to the radial redistribution of the core wave functions with changing configurations. This leads one to suspect that the Frozen Core approximation should be more valid for systems with higher atomic number.

In an attempt to remove the non-orthogonality amongst the relativistic HFS radial wave functions, we initiated a Gram-Schmidt orthogonalization process on each series of states with the same total angular momentum. A priori, one would expect wave functions with the lowest symmetry in each such series to be more accurately represented by the HFS approximation than those with a higher degree of symmetry. On this basis we chose such states to begin the inductive orthogonalization process for each series. The resulting radial wave functions, orthogonal to at least 1 part in 10^{15} , were then used to re-evaluate the oscillator strengths.

(4. b) Cesium, Oscillator Strengths, Transition Probabilities and Lifetimes

We compare the oscillator strengths calculated using the orthogonalized Cesium wave functions with the experimental data of Agnew and the Model Potential calculations of Stone⁶⁰ in Appendix (5). Immediately obvious from this comparison is the generally good agreement between our calculated values and those of experiment over the entire range of data. The behavior of these oscillator strengths as a function of the effective principal quantum number n^* , as indicated by the plots of Appendix (6), show that the deviation of the calculated values from the experimental data with increasing n^* values has been almost totally eliminated by the orthogonalization process. The theoretical curves follow closely the shape of the experimental curves generated by Agnew's data with the exception of the $6^2S_{1/2} - n^2P_{1/2}$ series for which the theoretical data predicts the appearance of a large Cooper minimum in the vicinity of $n^* = 8$ which is absent from Agnew's data. Further, the experimental results predict a crossing of the $6^2P_{1/2} - n^2D_{3/2}$ and $6^2P_{3/2} - n^2D_{5/2}$ curves near $n^* = 7$ while the

theoretical data does not show this feature. The relatively large scatter of the experimental values in this region makes it difficult to form the curves accurately and hence the crossing of the two curves may be due to this uncertainty.

In Stone's Model Potential calculations, the one-electron solutions of the non-relativistic Schrodinger equation, including the spin-orbit interaction, were used as radial wave functions. His solutions are obtained using a central symmetric potential that includes the nucleus and the closed shell electrons and is chosen so that the binding energies of the lowest 40 levels agree closely with experimental results. The agreement between these Model Potential calculations and those obtained here is generally good although in many cases our results, in the velocity formulation, give somewhat closer agreement with experiment. In general, it appears that the dipole velocity form yields results superior to those found in the length form when compared to experimental data and theoretical calculations done using other approximation schemes.

Lifetimes calculated using the length and velocity forms of the transition probabilities determined from the orthogonalized relativistic HFS wave functions are shown in Appendix (7) and are compared with those determined by using the non-orthogonal wave functions and the Bates-Damgaard approximation. Also shown in Appendix (7) are the available experimental lifetimes.

In the case where the orthogonalized wave functions have been used, the agreement between the velocity form and available experimental lifetimes for the $n^2S_{1/2}$ states is within about 10 percent. The calculated lifetimes of the n^2P_J levels agree with experiment to within 20 percent while those for the n^2D_J states lie within 30 percent of the

corresponding experimental values.

The Bates-Damgaard approximation is generally considered to give good results for the alkali metals. However, comparison between these lifetimes and those found from experiment indicates that the BD approximation yields results that are generally too high and that tend to diverge from the accepted values for the higher excited states.

The effects of the orthogonalization procedure are very evident in the comparison of the lifetimes obtained from the orthogonalized and non-orthogonalized wave functions. The orthogonalization of the wave functions reduces the overall magnitudes of the lifetimes and severely decreases the rate of divergence between theoretical and experimental values.

(4.2) Thallium, Oscillator Strengths, Transition Probabilities and Lifetimes

Oscillator strengths calculated using the orthogonalized Thallium₁ wave functions are compared with the combined experimental data of Norton and Gallagher⁵³ and Penkin and Sabanova⁵⁴ and with the theoretical calculations performed by Migdalek⁶¹ in Appendix (8). Corresponding plots of our theoretical oscillator strengths as a function of the effective principal quantum number are shown in Appendix (9) along with those generated from the experimental data. In the case of the $6^2P_{1/2} - n^2S_{1/2}$ and $6^2P_{3/2} - n^2S_{1/2}$ series, the experimental data indicates minima occurring around $n^* = 4.7$ and $n^* = 4.3$ respectively. These minima are absent in our theoretical curves. However, the proximity of the $6s6p^2 4P_{1/2}$ levels to the $6s^2(1s)10s^2S_{1/2}$ level could account for the appearance of the anomaly in the experimental data through configuration interactions. Perturbations due to the presence of this singly core excited level would be expected to affect most strongly the $10^2S_{1/2}$ levels. Since the HFS program used here to generate the relativistic wave

functions does not take into account the existence of such configuration interactions, one might not expect the minima to manifest themselves in the theoretical curves. The overall agreement between the relativistic HFS oscillator strengths and those obtained experimentally is a bit difficult to assess since the number of data points available for comparison is small, but generally the agreement is not as good as in the case of Cesium, with the exception of the $6^2P_{1/2} - n^2S_{1/2}$ series in Cesium.

Migdalek has calculated oscillator strengths for Thallium, using the "long wave" approximation with radial wave functions determined by a relativistic semi-empirical method which employs an adjustable central symmetric exchange potential of Slater type. The use of this "long wave" approximation can be justified in the case of Thallium, by investigating the structure of the dipole length form of the transition matrix elements. The dominant term in the matrix element of equation (84) for E1 transitions is

$$3J^{(1)}\left(\frac{\omega r}{c}\right) = 3 \int (G_{\alpha}G_{\beta} + F_{\alpha}F_{\beta}) J_1\left(\frac{\omega r}{c}\right) dr \quad (98)$$

It follows from the power series expansion of the spherical Bessel function⁴⁰ that in the limit of small arguments

$$J_1\left(\frac{\omega r}{c}\right) = \frac{1}{3}\left(\frac{\omega r}{c}\right) + O\left[\left(\frac{\omega r}{c}\right)^3\right] \quad (99)$$

When only this lowest order term in $\omega r/c$ is retained, then equation (88) reduces to the "long wave" approximation for the absorption oscillator strength for E1 transitions given by

$$f_{\alpha\beta} = \frac{2}{3} |\Delta E| (2J_{\beta} + 1) W^2 (L_{\alpha} J_{\alpha} l_{\beta} J_{\beta}; 1/2 \ 1) |_{\max} |R_{\alpha\beta}|^2 \quad (100)$$

where the radial integral for the transition $\beta \rightarrow \alpha$ is

$$R_{\alpha\beta} = \int (G_{\alpha} G_{\beta} + F_{\alpha} F_{\beta}) r dr \quad (101)$$

Since the $J^{(1)}$ term in the dipole length matrix element dominates the contributions from the terms involving the l_2^+ and l_2^- radial integrals by at least 5 orders of magnitude and since $(\omega r/c)^2 \leq 10^{-5}$ for r values up to an atomic radius, the use of the "long wave" approximation yields oscillator strengths that agree with those obtained using the length formulation of the relativistic HFS method to within about 1 part in 10^5 . However, since we find that the velocity formulation appears to give results more consistent with the experimental data, the velocity form of the "long wave" approximation should be used. In Migdalek's calculations, use of the adjustable parameter in the exchange potential reduces the dependence of the oscillator strengths on the length or velocity formulation of the transition matrix elements.

Lifetimes calculated using the length and velocity forms of the transition probabilities determined from the orthogonalized relativistic HFS wave functions are shown in Appendix (10) and are compared with those found by using the non-orthogonal wave functions and the Bates-Damgaard approximation. Also listed in Appendix (10) are the few existing lifetimes measured experimentally.

A general comparison between the theoretical results and those of experiment is difficult because of the lack of experimental results. The velocity form of the orthogonalized HFS lifetimes agree well with the experimental data for the $8^2S_{1/2}$ and $6^2D_{3/2,5/2}$ states. However,

the calculated resonance state lifetime is about 40 percent too high and the remaining n^2D_J states have predicted lifetimes that are between 60 and 75 percent too low.

Lifetimes calculated using the BD approximation are about a factor of 2 too large for the $7^2S_{1/2}$ and $8^2S_{1/2}$ states while for the 6^2D_J and 7^2D_J states values are obtained which lie within the experimental range of uncertainty. The behavior of the BD lifetimes with respect to the experimental values indicates that, for the $n^2S_{1/2}$ and n^2P_J states, the approximation of the potential by a coulombic potential is generally not valid in complex systems like Thallium, since in these cases the optically active electron is expected to interact strongly with the core. However, in the cases of n^2D_J and n^2F_J states core penetration would be essentially absent. This allows the inner core electrons to be treated as a point charge to a first approximation and such a situation may then be treated adequately by the BD approximation. Caution must be exercised in using the BD treatment for complex systems for which correlation effects and configuration interactions may be of significance. This is indicated by the poor agreement between the BD and experimental lifetime for the $8^2D_{5/2}$ state.

(4.3) Suggestions for Improving the Orthogonalized Free-Core Simplified Hartree-Fock Procedure

Examination of the theoretical data obtained by using the orthogonalized HFS wave functions still indicates the diverging trends with increasing scaled energy separations between the states involved in the transitions. However, this divergence has been considerably reduced by the orthogonalization process. The theoretical data supplied by Migdalek in the "long wave" approximation for Thallium, indicates a similar trend when compared with the existing experimental data although the magnitude of the deviations are slightly less than found in our

data. The difference between these two sets of theoretical data is partially accounted for by the different forms of the Slater exchange potential used in the two sets of theoretical results. Kohn and Sham⁶² have indicated that the strict Slater exchange potential used in our case generally leads to an over estimation of the exchange by at least a factor of 3/2. Migdalek found that his adjustable multiplying parameter always assumed values less than 2/3. Such a semi-empirical potential as used by Migdalek, although still not physically correct in form for reasons to be discussed below, might be expected to yield a slightly better approximation to the correct exchange potential which may account for the better fit of Migdalek's data to that of experiment. We found, by using the Kohn-Sham exchange potential in which the Slater exchange potential is multiplied by a factor of 2/3, that corrections to the oscillator strengths generally amount to at most a few parts in 10².

We reconsider the Hartree-Fock exchange potential for a electron in the i^{th} spin-orbital given by equation (47). The value of this potential may vary by as much as a factor of 2 for various spin-orbitals, ψ_i , at a fixed value of r . This has been quantitatively shown for the Cu^+ ion by Hartree⁶³. Hence, the replacement of the exchange potential in equation (47) by an averaged exchange potential is a very crude approximation which may lead to serious error. For a given spin-orbital, the averaged Slater exchange potential yields results that are too large at some r values and too small at others. Another serious deficiency of the averaged Slater potential is its incorrect behavior at large r values. In the full HF method the exchange potential of equation (47) can be seen to have the property that

$$rV_{\text{exch}}(r) \rightarrow -1 \text{ as } r \rightarrow \infty \quad (102)$$

To see this, write the exchange potential as

$$V_{\text{exch}}(r) = V_{\text{self}}(r) + V_{\text{ex}}(r) \quad (103)$$

where the self-interaction potential, $V_{\text{self}}(r)$, is written as

$$V_{\text{self}}(r) = - \int \psi_1^*(\vec{r}_2) \psi_1(\vec{r}_2) \frac{1}{r_{12}} d\tau_2 \quad (104)$$

and the true exchange potential, $V_{\text{ex}}(r)$, has the form

$$V_{\text{ex}}(r) = - \frac{\sum_{j \neq 1} \int \psi_i^*(\vec{r}_1) \psi_j^*(\vec{r}_2) \frac{1}{r_{12}} \psi_j(\vec{r}_1) \psi_i(\vec{r}_2) d\tau_2}{\psi_1^*(\vec{r}_1) \psi_1(\vec{r}_1)} \quad (105)$$

Then, in the limit as $r \rightarrow \infty$,

$$rV_{\text{self}}(r) \rightarrow -1 \quad (106)$$

and

$$rV_{\text{ex}}(r) \rightarrow 0 \quad (107)$$

However, the averaged Slater exchange potential multiplied by r tends to zero as r tends to infinity and thus the Slater exchange potential fails to correct for the electron self-interaction at large values of r .

What is often done to compensate for this defect is to apply the empirical correction due to Latter⁶⁴ to the total potential; that is, one makes the replacement

$$rV(r) = -(Z-N+1) \text{ for } r \geq r_c \quad (108)$$

where r_c is determined by the condition

$$r_c V(r_c) = -(Z-N+1) \quad (109)$$

and where N is the total number of electrons in the system. Coulson and Sharma⁶⁵ have pointed out several defects in such an empirical correction. First, the Latter potential yields a discontinuous slope at r_c and consequently gives rise to the non-physical result of a non-vanishing surface charge density on the sphere separating the two regions $r \geq r_c$ and $r < r_c$. Wilson, Wood and Slater⁶⁶ further point out that the Latter potential is not variationally derivable from the expression for the total energy of the system. Gopinathan⁶⁷ demonstrates that the HFS potential underestimates the self-interaction at both large and intermediate values of r . Hence, the potential is generally smaller than the correct potential and the Latter correction to the potential leads to a non-physical shrinkage of the atom.

To remove the discrepancies between the theoretical calculations and the experimental data, it is necessary to use a better approximation of the exchange potential. Recently, Gopinathan⁶⁷ has developed an approximate exchange potential which may prove to be extremely useful in atomic structure calculations. His method of treating the exchange potential is based on the separation of the potential into an electron self-interaction term and a true exchange term. The self-interaction term is evaluated exactly while the true exchange potential is approximated in a manner similar to the $\rho^{1/3}$ approximation of Dirac and Slater. In the Gopinathan scheme, electrons in different spin-orbitals have different exchange potentials thus avoiding the major defects that arise from the use of a single averaged potential for all electrons as is employed in the Slater approximation. In Gopinathan's approach, the

self-interaction is accounted for correctly and the exchange potential is shown to have the proper asymptotic behavior at large r values. One additional feature of this treatment is that the simplicity of the computations featured by the HFS method is retained and computations of exchange potentials, one-electron eigenvalues and spin density distributions, as determined in the cases of Cu^+ and Mn^{+2} , yields results that compare favourably with those obtained by the more elaborate and time consuming full Hartree-Fock procedure.

Using the diagonalization procedure presented in this investigation, together with a more accurate exchange potential such as the one derived by Gopinathan, one might expect to perform atomic structure calculations with an accuracy approaching that of a full Hartree-Fock calculation. Such a procedure would allow the rapid and inexpensive computation of atomic structure data.

EXPERIMENTAL

PART A: Description of the Apparatus

The arrangement of the apparatus is shown in figure (3). Mercury resonance radiation emitted by a radio-frequency mercury vapour lamp was passed through an interference filter with a peak transmission of 10 per cent at 2537\AA , and was focussed in the fluorescence cell, which contained the mercury-thallium vapour mixture and which was located in an oven whose temperature could be accurately controlled. The resulting fluorescence, observed at right angles to the direction of excitation, was resolved with a grating spectrometer, and was brought to a focus at the photocathode of the photomultiplier tube. The signal produced by the photomultiplier tube was registered with a Keithley model 417 picoammeter and a Hewlett-Packard strip-chart recorder.

The fluorescence cell was connected to a vacuum system and could be filled with controlled amounts of nitrogen. The cell was fitted with two sidearms, one for thallium and one for mercury. Both sidearms could be heated independently to produce the required vapour pressures of thallium and mercury.

(1.0) Light Sources

The Hg 2537\AA exciting radiation was produced by an electrodeless discharge in mercury vapour, which was contained in a quartz bulb. The discharge was driven by a radio-frequency oscillator, powered by a Fluke model 407DR power supply. The r-f oscillator is drawn schematically in figure (4). This type of lamp, which has been used previously in this laboratory, provides an intense source of Hg 2537\AA resonance radiation. The resonance line is narrow and relatively free of self-reversal.

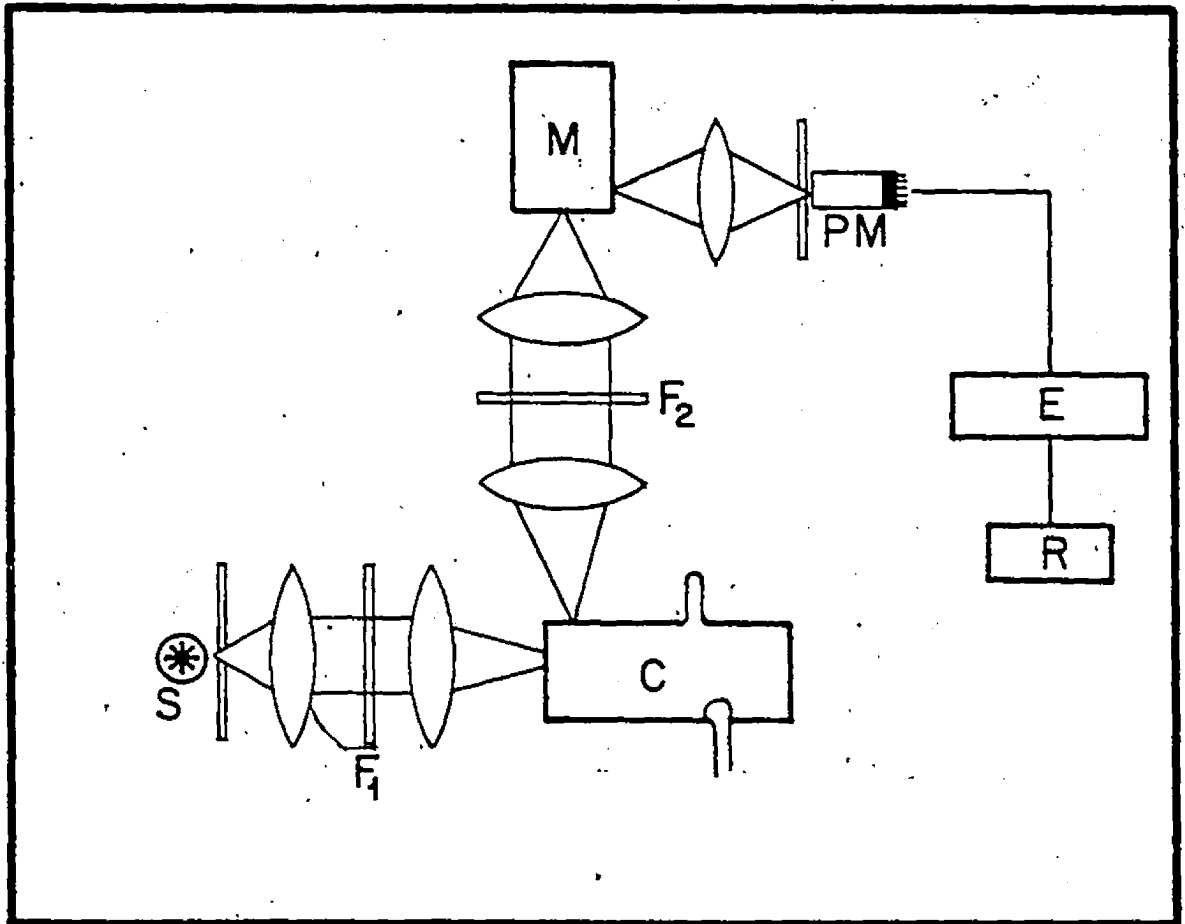


Fig. 3: Arrangement of the apparatus

- S - source
- C - fluorescence cell
- M - monochromator
- PM - photomultiplier tube
- E - Keithley picoammeter
- R - strip-chart recorder
- F₁ - filter
- F₂ - filter

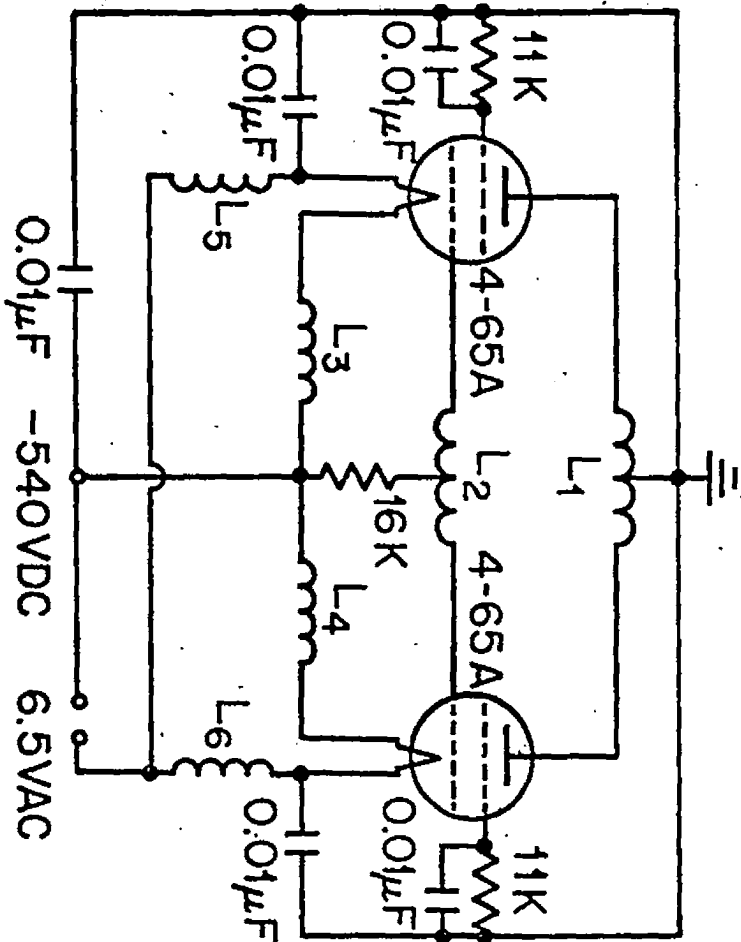


Fig. 4 : R-F oscillator circuit

L₁ - RF coil, 5½ turns with Hg bulb enclosed

L₂ - RF coil, 4½ turns

L₃, L₄, L₅, L₆ - RF coils, 20 turns

The bulbs were constructed of high-quality non-fluorescing quartz tubing with an outside diameter of 1.4cm and a wall thickness of 2mm. Several ampoules were connected to a distillation manifold and small amounts of mercury (previously distilled several times) were distilled under vacuum into each bulb. Argon* at various pressures in the range 0.2 - 1.8 torr was admitted into the ampoules which were then sealed and removed from the manifold. Subsequent tests of the mercury lamps indicated that argon pressures in the range 0.8 - 1.1 torr gave greatest stability and highest output of the Hg 2537 \AA resonance component. It was also found that slight cooling, by an even flow of air about the bulb, resulted in greater stability and a moderate increase in the intensity of the Hg 2537 \AA radiation emitted from the lamps.

When studying the quenching of the $\text{Tl}(7^2\text{S}_{1/2})$ state in the Tl-N_2 mixture, the thallium atoms were excited directly with the Tl 3776 \AA resonance radiation emitted from an air-cooled Osram lamp. The lamp operated at a current of 0.8 amperes which was supplied by an Ealing Universal Spectral Lamp power supply. It was found that at a current of 0.9 amperes, (as recommended by the manufacture), the intensity of the resonance radiation emitted from the lamp fluctuated rather noticeably. By reducing the current to 0.8 amperes the intensity fluctuations could be rendered insignificant.

(2.0) The Fluorescence Cell and Oven

The whole cell, which is depicted in figure (5), was constructed of high optical quality-non-fluorescing quartz[†]. Ordinary (optical quality) quartz, when irradiated with ultraviolet light, fluoresces

* research grade, supplied by the Linde Corporation

† Suprasil W1 supplied by the Amersil Corporation

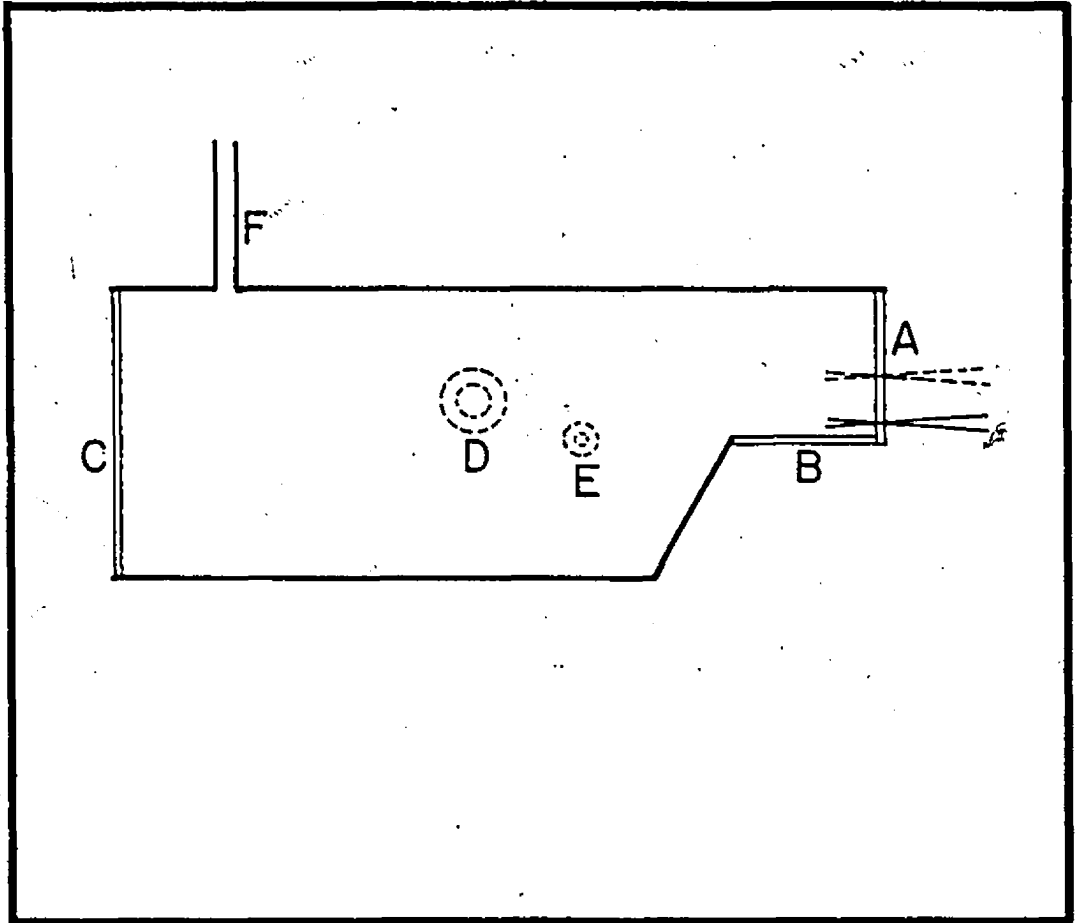


Fig. 5: The fluorescence cell

- A - entrance window
- B - observation window
- C - rear observation window
- D - thallium sidearm
- E - mercury sidearm
- F - capillary to vacuum system

strongly in the spectral region from 3500\AA to 4800\AA . This quartz-fluorescence would completely mask all atomic fluorescence emitted from the metal vapour mixture in the cell. The cell had an overall length of 10cm, an outside diameter of 35mm, and a wall thickness of 3mm. The entrance and exit windows were cut from 1.5mm quartz plates and fused to the "D" shaped portion of the cell. The observation window at the rear of the cell was 3mm thick. This design of the fluorescence cell permitted the exciting beam of the Hg 2537\AA light to be focused just inside the rectangular corner formed by the entrance and observation windows, keeping the optical paths through the cell to less than 1mm and thus reducing the reabsorption of the mercury resonance radiation.

The cell was enclosed in a series of stainless steel shields to reduce the intensity of black body radiation reaching the spectrometer from the heating elements. Aquadag, commonly employed to reduce the stray light, could not be used since, at high temperatures, it vapourizes from the cell surface. The steel shield covering the observation window had in it a rectangular aperture in the form of a slit to provide a well-defined geometry required for fluorescent intensity measurements.

The window at the rear of the cell was provided to facilitate the optical alignment of the cell and to monitor the strong corrosive action which took place on prolonged exposure of the quartz to thallium vapour at temperatures in the range 500°C - 600°C . At these temperatures the corrosive action of thallium on the inner surfaces of the windows and cell body was strong enough to cause sufficient erosion of the window surfaces so that the transmission of the fluorescent light was severely reduced. After each experimental run the windows had to be replaced or repolished in order to maintain optimal transmission.

The cell was fitted with two sidearms one of which, containing metallic thallium, was approximately 9cm long and had an inside bore of 15mm to allow efficient transport of thallium vapour to the cell. The second sidearm, which contained mercury, extended below the cell and protruded out of the oven, and terminated in a reservoir which could be heated separately by means of a "Chromel A" heater wrapped along the whole of the sidearm. Current to this heater was delivered through a Variac autotransformer which allowed the temperature of the sidearm to be accurately controlled. The mercury reservoir was connected to the cell by a capillary approximately 50cm in length and having a bore of 2mm. The reservoir heater was necessary to produce and maintain the proper density of mercury vapour in the cell. Without it, a very long time was required for the mercury to migrate through the capillary against the high thermal gradient, and the mercury density in the cell was difficult to stabilize.

The fluorescence cell was mounted inside a stainless steel oven which was heated by means of nine G.E. strip-heaters fastened to the exterior surfaces of the box in series-parallel combinations. Current to these heaters was supplied by a Variac rated at 1800 watts. Using these heaters only, a cell temperature of about 350°C could be produced. To reach the required temperatures, several coils of "Chromel A" resistance wire were wound and mounted on 6mm thick asbestos fibre boards. These additional heater units were connected in series-parallel configurations and attached to the inner surfaces of the oven. The resistance of each component in the heater circuit was established by a trial and error method so as to provide a thermal gradient along the cell. The front windows were kept at about 10°C above the temperature of the

thallium sidearm, to prevent condensation of thallium metal on the window surfaces. Current to the inner heater elements was supplied by a 2800 watt Variac. With this arrangement it was possible to maintain the oven temperature within $\pm 2^{\circ}\text{C}$, in the range $450^{\circ}\text{C} - 580^{\circ}\text{C}$. The temperatures were measured by chromel-alumel thermocouples placed at various positions on the cell sidearms and connected to a Hewlett-Packard digital multimeter.

The oven was placed in a transite box covered with a water-cooled copper jacket. The space between the two boxes was filled with mica chips to reduce heat loss from the oven.

Non-fluorescent quartz plate windows covered the appropriately placed apertures in the inner and outer boxes. These windows served to reduce the heat lost through the apertures.

(3.0) The Vacuum System

The fluorescence cell was connected to an external vacuum system by means of a capillary tube approximately 40cm in length and 2mm inside diameter. Although the capillary severely reduced the pumping efficiency, it did effectively prevent the migration of atomic vapours from the cell into the vacuum system. The system was evacuated by an Edwards E02 diffusion pump, backed by an Edwards rotary pump which had a pumping rate of 50 litres per minute. The pressure in the vacuum system was of the order of 10^{-8} torr, and was monitored by a CVC type GIC-110B ionization gauge.

(4.0) The Monochromator and Photomultiplier Tube

The fluorescent spectra emitted from the fluorescence cell was resolved by a Jarrel-Ash model 82-000 spectrometer equipped with an 1800 grooves/mm diffraction grating blazed at 3000\AA in first order. The monochromator had an aperture of $f/8.6$ and a focal length of 0.5 meter with a reciprocal dispersion of $16\text{\AA}/\text{mm}$ in first order.

The resolved fluorescence from the monochromator was focused on the photocathode of a 13 dynode EMI D260/9865QAM photomultiplier tube. The tube, mounted in a cryostat cooled to liquid nitrogen temperatures, had a tri-alkali photocathode with an S-20 spectral response. The nominal cathode diameter was 10mm. The tube was operated at a potential of -1000V which produced a gain of 10^6 . The cathode potential was maintained by a Fluke model 412B power supply. The dark current at liquid nitrogen temperature was measured to be 1.5×10^{-11} amperes. For comparative purposes, an EMR type 514N-03-14-03900 photomultiplier tube, equipped with a tri-alkali photocathode having an S-20 spectral response and a 14 stage venetian blind dynode chain, was used. This tube was operated at -1660V which produced a gain of 10^6 . The dark current, at room temperature, was found to be 6×10^{-11} amperes, and was depressed to 5×10^{-13} amperes at liquid nitrogen temperature.

(5.0) Calibration of the Monochromator and Photomultiplier Tube

Although the quenching cross sections are independent of the spectral response characteristics of the optical support equipment, the determination of the energy transfer cross sections, depending on the intensities of the spectral components, demand complete knowledge of the spectral response features of the monochromator and photomultiplier tube, and of the absorption characteristics of the quartz optics.

The spectral response of the monochromator was obtained using the experimental arrangement shown in figure (6). The first Jarrel-Ash spectrometer served to isolate a particular line from the Osram lamp source. This line was focused onto the entrance slit, (with a 0.5mm slit-width), of the second Jarrel-Ash monochromator which was to be calibrated. A vacuum photodiode (ITT F4000), with an effective photocathode 4cm in diameter, was placed behind the entrance slit and detected

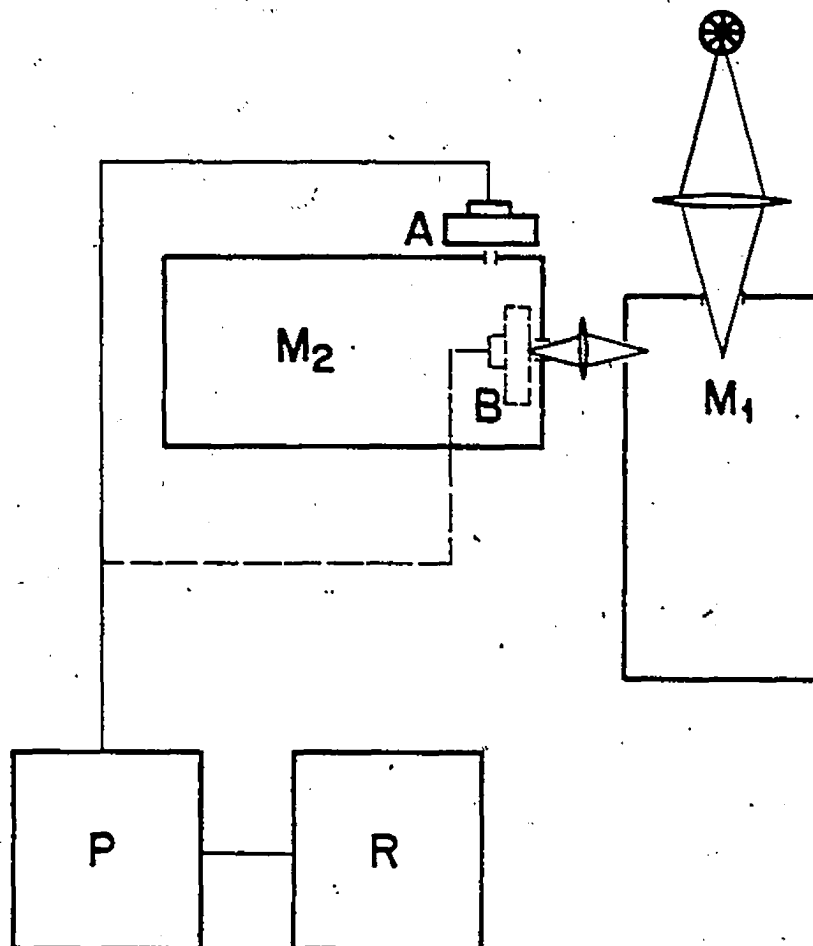


Fig. 6: Experimental arrangement used to calibrate the Jarrel-Ash monochromator

- A - photodiode set to receive light emerging from monochromator M_2
- B - photodiode set to receive light incident on monochromator M_2
- P - Keithley picoammeter
- R - strip-chart recorder
- M_1 - Jarrel-Ash monochromator
- M_2 - Jarrel-Ash monochromator to be calibrated

all the light entering the monochromator. The exit slit was also set at 0.5mm and the emergent light from the monochromator was detected by the same photodiode placed against the outside of the exit slit so as to receive all the emergent light. Several light sources, consisting of Hg, K, Cs, Na, and Zn Osram lamps, were used to produce the spectral response curve (shown in figure(7)) for the Jarrel-Ash monochromator. Several complete runs were made to determine the reproducibility of the curve.

Spectral response characteristics at -40°C had been supplied by the manufacturer for the EMI photomultiplier tube, but previous experience with other photomultipliers indicated that the spectral response may be altered somewhat by depressing the operating temperature to that of liquid nitrogen.

Calibration of the EMI photomultiplier at liquid nitrogen temperature was accomplished by allowing light of the same energy flux to be incident on the photocathode of the photomultiplier as was incident on the window of a calibrated RBL-500 thermopile (supplied by the C.M.Reader company). The thermopile sensitivity was quoted to be $12.27 \frac{\mu\text{V}}{\mu\text{W}}$. Several spectral components emitted by Hg, Zn, and K Osram lamps were used as light sources. The thermopile, which was equipped with a quartz window having rectangular dimensions of 3mm by 0.5mm, was mounted to the holder shown in figure (8). The holder, which could be placed in either of two positions, had a light baffle with an aperture of the same dimensions as the thermopile window. The light baffle aperture and the thermopile window coexisted in the same optical plane situated perpendicular to the light path from the monochromator. In the first position, the thermopile was allowed to detect the light flux emerging from the monochromator. The

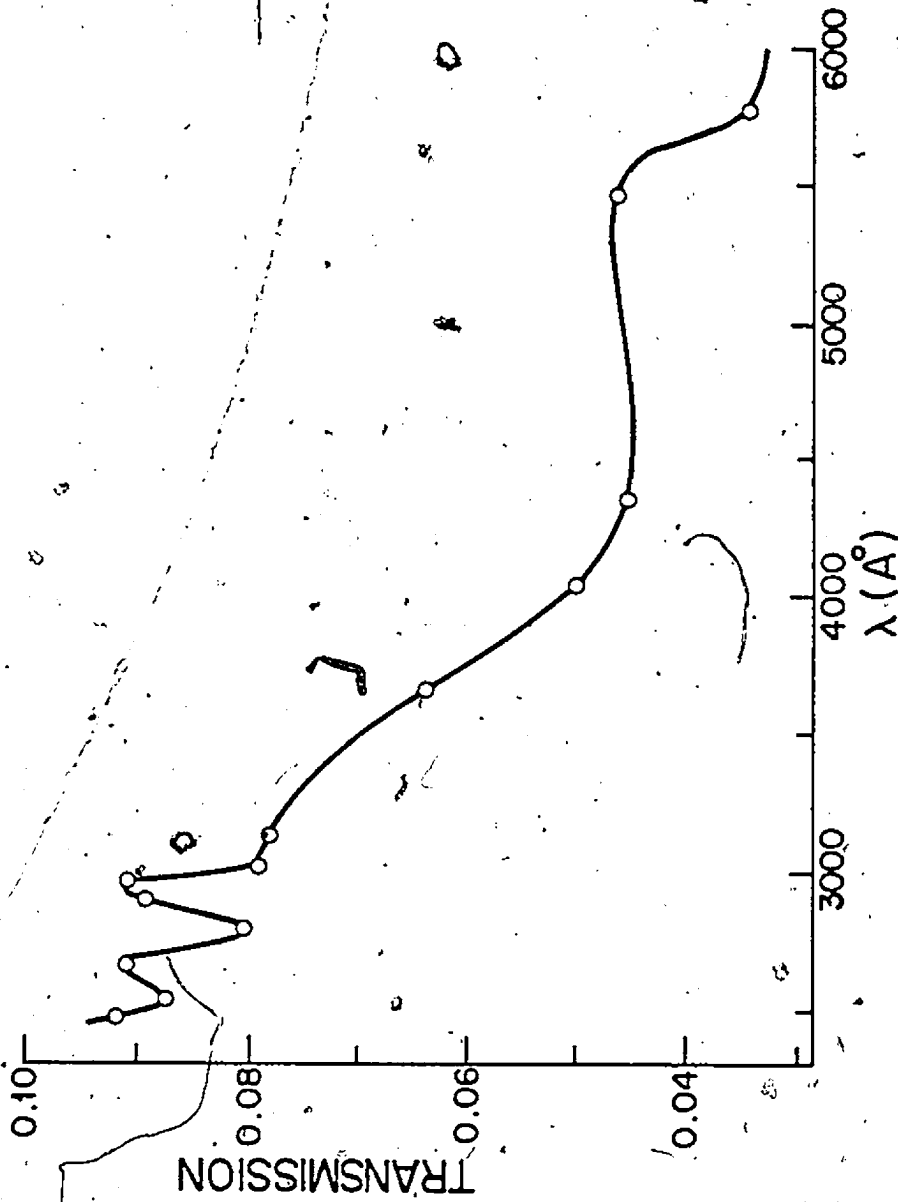


Fig. 7: Transmission of Jarrei-Ash monochromator.

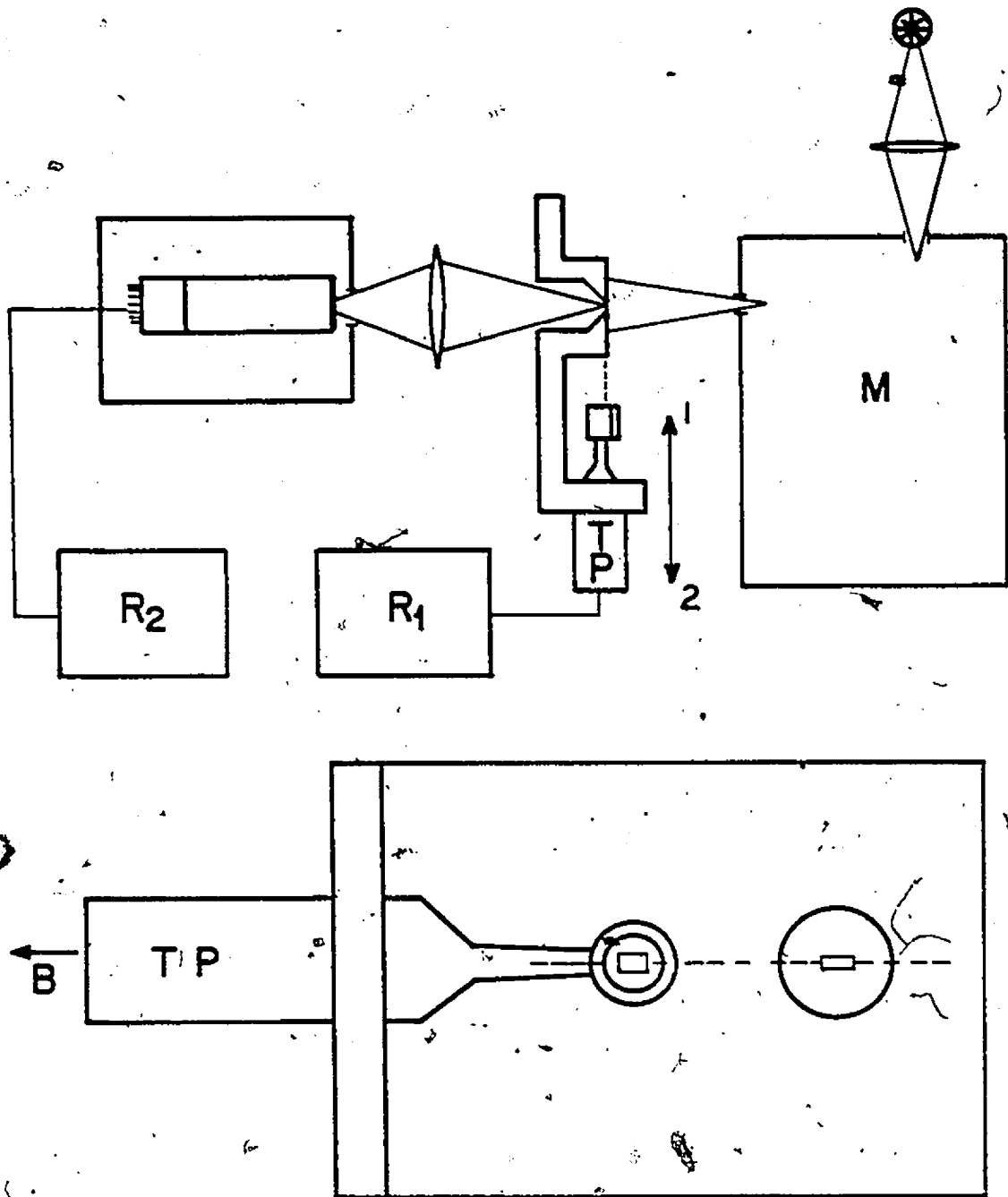


Fig. 8: Apparatus for measuring spectral response of photomultiplier showing enlarged front view of slide-holder.

M - monochromator

TP - thermopile

R₁ - milli-microvoltmeter

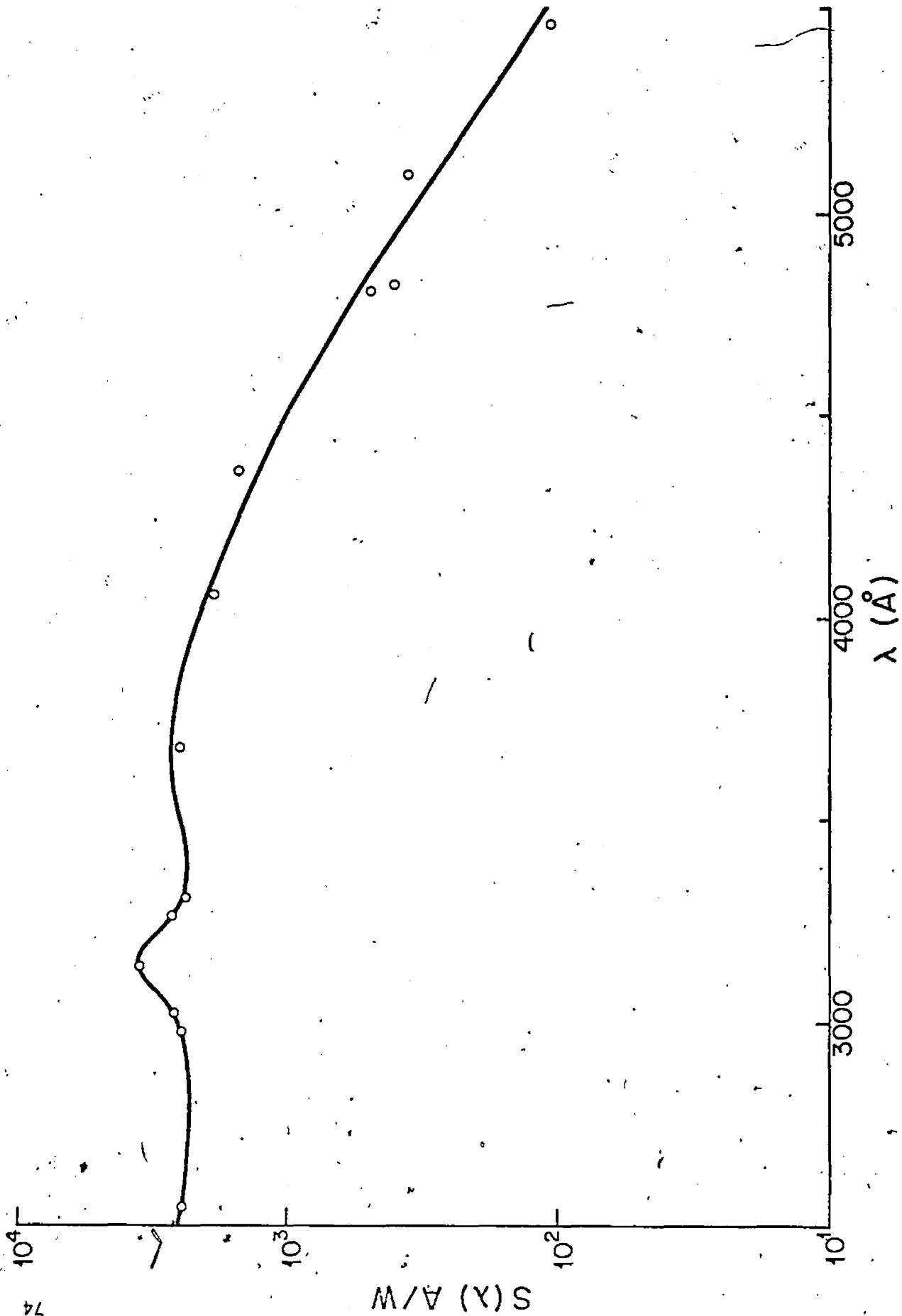
R₂ - Keithley picoammeter

B - milli-microvoltmeter

resulting response, in microvolts, was registered by a Keithley model 149 milli-microvoltmeter. In the second position, the same light flux was transmitted through the aperture and was brought to focus at the photocathode of the photomultiplier tube. The photomultiplier response, corresponding to this energy flux, was measured by the Keithley model 417 electrometer-amplifier. The results of this calibration are shown in figure (9).

The EMR photomultiplier, which was to be used to check the intensity measurements obtained by the EMI photomultiplier tube, was calibrated by the manufacturer.

Fig. 9 : Spectral response of EMI photomultiplier tube



EXPERIMENTAL PROCEDURE

The fluorescence cell was thoroughly cleaned with an acid solution composed of four parts HCl and one part HNO₃. The cell was flushed several times with distilled water and finally rinsed with alcohol. The thallium metal (about 2gm) was sealed into the appropriate sidearm of the cell. Mercury was prepared using the distillation manifold shown in figure (10). The manifold was attached to the vacuum system and baked under vacuum at 180°C for several days. A charge of triply distilled mercury* was introduced into the lowest ampoule through the fill tube which was then sealed. The mercury was distilled under vacuum from ampoule to ampoule, each time leaving behind a small amount of residue which was sealed off and removed from the manifold before continuing the distillation process. The final ampoule containing the clean mercury was transferred to the mercury reservoir at the base of the second sidearm. The mercury was prevented from entering the cell, until required, by a break-seal at the head of the reservoir, which could be opened under vacuum. The cell was then inserted into the oven, aligned, and connected to the vacuum system. To remove water vapour and other contaminants, the oven temperature was raised to about 250°C and the cell was baked under vacuum for several days until a final pressure of the order of 10⁻⁸ torr was attained. Thermocouples were secured to the cell body close to the front windows and near the rear window. Additional thermocouples were placed along the length of the thallium sidearm.

The temperature of the oven was brought up slowly and allowed to

* supplied by A.D. MacKay Co.

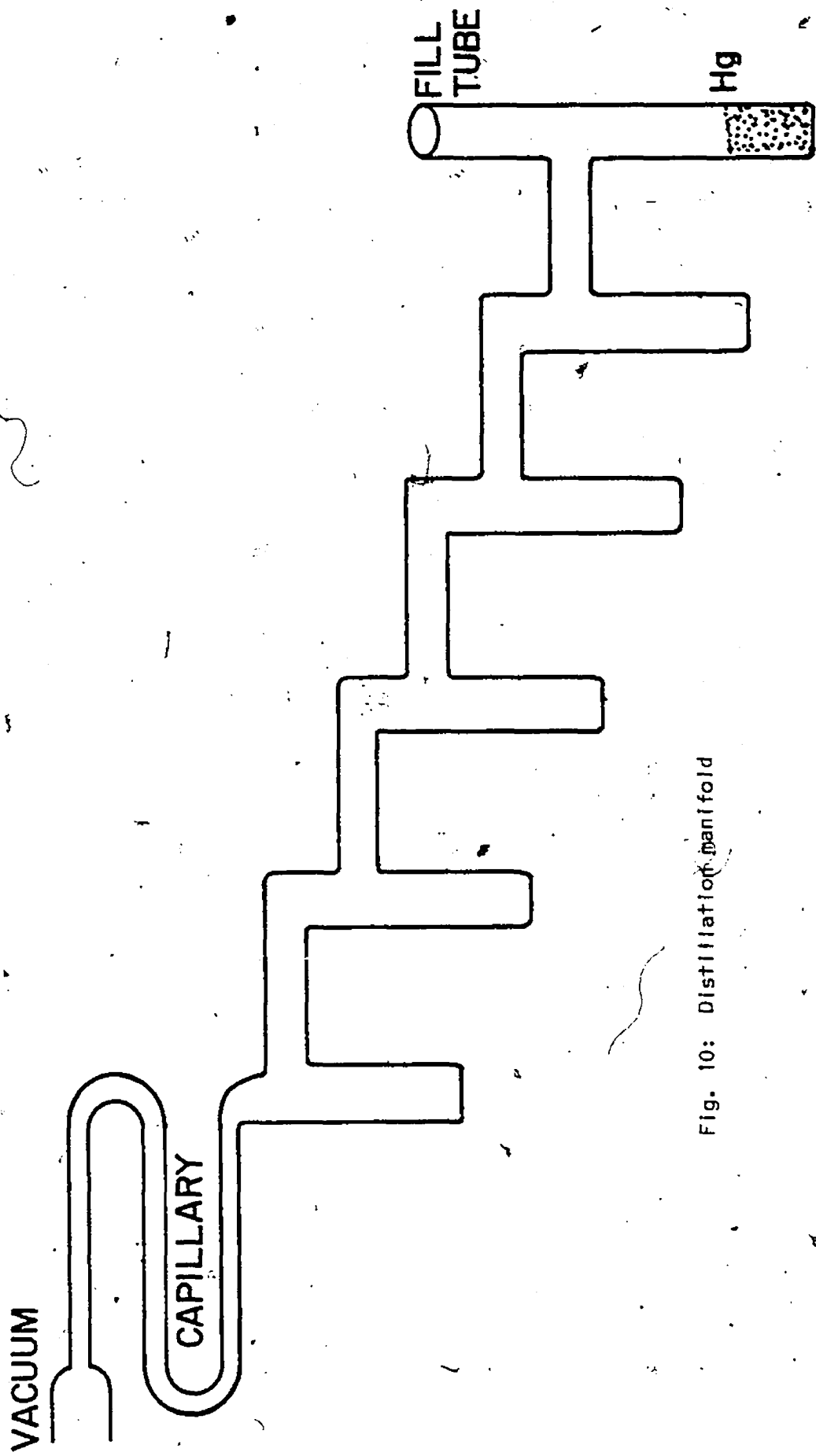


Fig. 10: Distillation manifold

stabilize with the thallium sidearm temperature in the range 430°C - 580°C . The corresponding pressure of the thallium vapour in the cell was determined by this sidearm temperature and was calculated from the formula given by Nesmeyanov⁶⁸. The exciting beam from the mercury r-f electrodeless discharge lamp was first passed through a mercury interference filter with a peak transmission of 10 percent at 2537\AA , and then brought to a focus just inside the entrance window of the cell at a point about 1mm from the observation window. The mercury reservoir was opened and a small amount of mercury vapour was admitted into the cell by heating the reservoir to about 40°C - 50°C . The intensity of the Hg 2537\AA resonance radiation emitted from the cell was continuously monitored during the admission of the mercury vapour. The density of mercury in the cell was stabilized when the intensity of the Hg 2537\AA fluorescence reached a maximum.

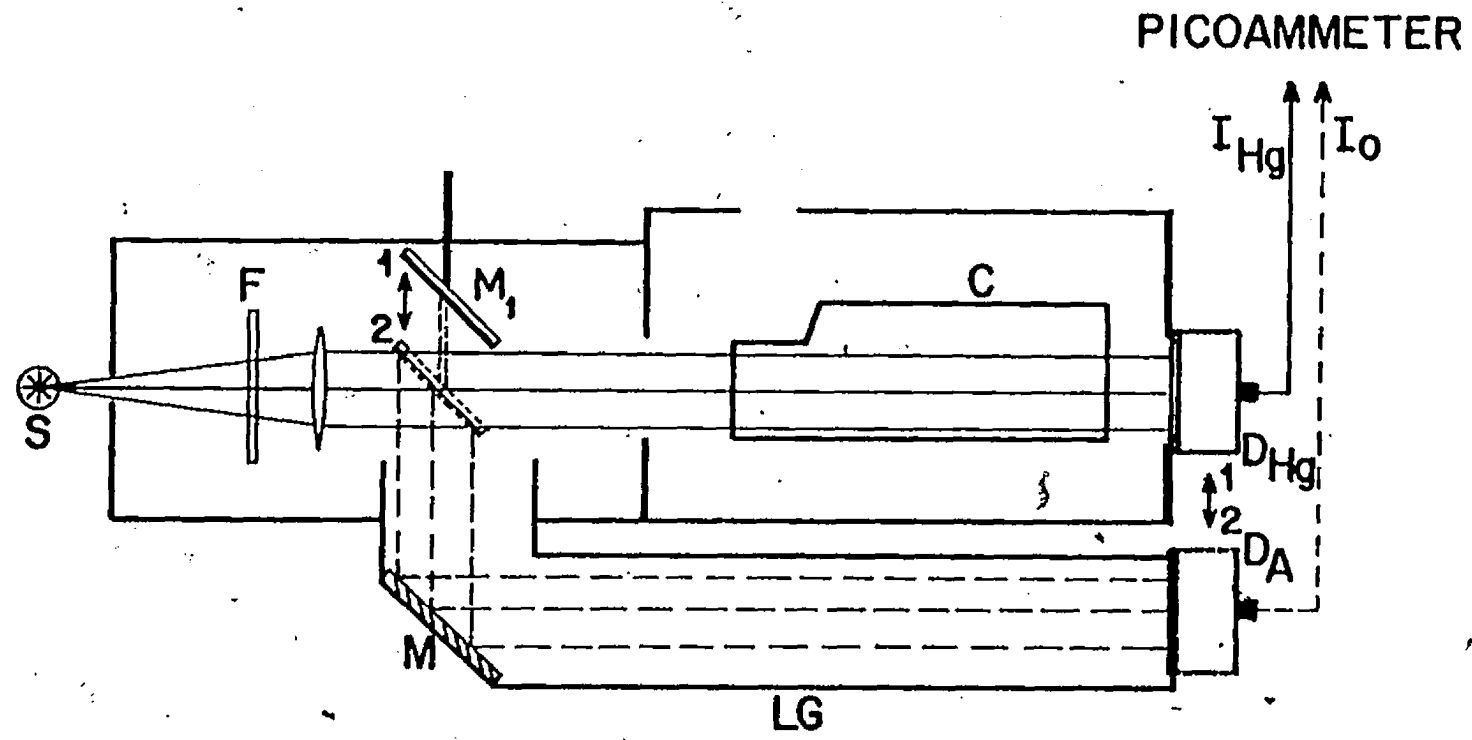
The fluorescent radiation emitted from the cell was focused onto the entrance slit of the monochromator which automatically scanned the thallium fluorescent spectrum at a rate of $10\text{\AA}/\text{min}$. A typical run lasted for several hours, and although the intensity of the mercury resonance radiation remained relatively stable throughout each run, it was monitored frequently to avoid the introduction of experimental uncertainty into the evaluation of the energy transfer cross sections. All intensities were detected by the photomultiplier (cooled to liquid nitrogen temperature) and were recorded by the Hewlett-Packard strip-chart recorder.

Since the mercury admitted to the fluorescence cell had to pass through a long capillary, and since there was a large thermal gradient between the reservoir and the cell, the pressure-temperature formula given by Nesmeyanov could not be used to determine the pressure of

mercury in the cell. An auxiliary experiment was performed which allowed the pressure of mercury in the fluorescence cell to be determined. The experimental arrangement is shown in figure (11). A two position sliding mirror assembly was mounted between the lens and the cell. In the first position the Hg 2537Å radiation was allowed to enter the cell which was void of any mercury vapour in the initial case. In the second position the beam was reflected down the length of a light guide. The intensities (and their ratio) corresponding to these two positions were recorded by the same photodiode (ITT F4000) attached to a sliding assembly at the rear of the oven. Mercury was next placed in the sidearm normally used for the thallium charge, and was frozen to liquid nitrogen temperature. At this temperature the vapour pressure of mercury was essentially zero. The cell was pumped continuously until all mercury vapour that had escaped into the cell body during the admission of the mercury into the sidearm was removed, as indicated by the ratio of light intensities through the cell and through the light guide. By stabilizing the temperature of the mercury sidearm at various temperatures in the range $-20^{\circ}\text{C} - +60^{\circ}\text{C}$, various mercury pressures in the cell could be obtained. To obtain temperatures in the range $-20^{\circ}\text{C} - 0^{\circ}\text{C}$ several anhydrous salts were used in an eutectic mixture with ice (see Appendix (14)). Since the sidearm bore was large (14mm) and the distance from the reservoir to the cell body was short (8mm), the flow of mercury to the cell was unrestricted and the pressure-temperature formula of Nesmeyanov could be used. For each sidearm temperature employed, the intensity, I_{Hg} , of light transmitted through the cell and the intensity, I_0 , of light through the light guide was determined. A plot of the ratio I_0/I_{Hg} as a function of the mercury density is shown in figure (12). The curve was formed over several runs and was found to be reproducible within 10 percent. During the sensitized fluorescence experiments,

Fig. 11: Experimental arrangement used to determine mercury density in the fluorescence cell

- S - source of Hg 2537Å resonance radiation
- F - 2537Å filter
- M - mirror
- C - fluorescence cell
- LG - light guide
- M₁ - sliding mirror
- D_{Hg} - photodiode detecting light through cell
- D_A - photodiode detecting light through light guide



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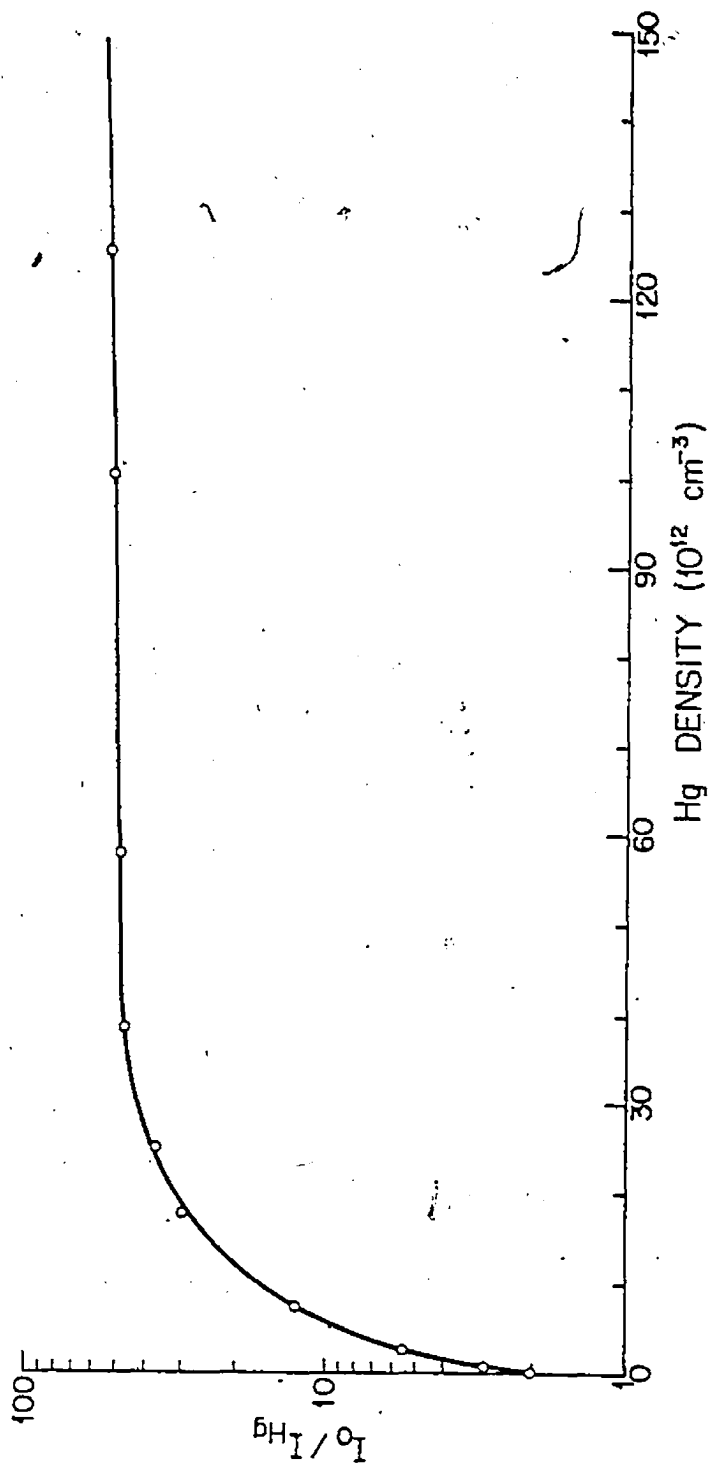


Fig. 12: Intensity variation of mercury 2537 Å resonance radiation transmitted through the cell as a function of the mercury vapour density

the density of mercury in the cell at any given time could be determined by comparing the intensity of the Hg 2537 \AA light being transmitted through the vapour to that being transmitted down the light guide. This calibration curve should remain valid when thallium vapour is also present, since there was no indication that amalgams of mercury and thallium were formed, under the experimental conditions employed, which could alter the shape of the curve.

The reabsorption of the mercury and thallium resonance fluorescence by the vapour mixture must be taken into account when determining the collisional excitation energy transfer cross sections. It has been found previously⁶⁹ that, for partial pressures of mercury of the order of 10^{-5} torr and the experimental arrangement employed here, the reabsorption of mercury resonance radiation can be disregarded. It remained to find the extent of reabsorption for the thallium resonance fluorescence. Pure thallium vapour, contained in the cell and stabilized at a pressure corresponding to that employed in the actual experiment (of the order of 10^{-5} torr), was radiated with Tl 3776 \AA resonance radiation emitted from the filtered Osram lamp source. The exciting beam was passed through a narrow slit (0.1mm wide) and was focused by means of an adjustable lens assembly just inside the cell at a distance d_0 , (a distance of about 0.5mm), from the observation window. The intensity, I_0 , of the resonance radiation was measured for focusing at this point. At such a focal point the optical path through the vapour was restricted to about 1mm. By means of the calibrated adjustment on the lens assembly, the distance from the observation window could be stepwise increased. The intensity, I , of the resonance radiation emitted from the vapour was measured for each such increment. The ratio I/I_0 was determined and plotted as a function of the distance

from the observation window. This plot is depicted in figure (13). By extrapolating this curve to the $1/I_0$ axis one can determine the intensity of the resonance line in the absence of reabsorption. Comparing this value to that of the intensity found at a distance of 0.5mm and 1mm, which represent the distances used in the actual experiment, it is evident that the reabsorption of the thallium resonance fluorescence amounts to only about 7 percent and 13 percent for the two positions, respectively.

In studying the quenching effects of N_2 molecules on the thallium sensitized fluorescence the same experimental arrangement was used. The exciting mercury resonance radiation was made incident on the tertiary system Hg-Tl- N_2 , at nitrogen pressures between 0.2 torr and 8 torr. The intensity of each thallium fluorescent component being emitted from the cell was first measured in the absence of nitrogen. Controlled quantities of nitrogen were admitted into the cell and the system was allowed to return to equilibrium. Several scans of the thallium fluorescence were made and the intensity of each spectral component was recorded on the strip-chart recorder for each nitrogen pressure used. Several complete experimental runs, using different cells, were carried out for a range of thallium densities between about 10^{11}cm^{-3} and 10^{13}cm^{-3} . The mercury density was kept constant at about 10^{11}cm^{-3} . The stability of the mercury density was frequently checked by monitoring the intensity of the Hg 2537 \AA resonance fluorescence emitted from the cell.

The partial pressure of nitrogen being admitted into the cell was found by first measuring the pressure in mm of #704 diffusion pump oil contained in the manometer attached to the vacuum station. This reading was then converted to a pressure in torr by means of the hydrostatic equation

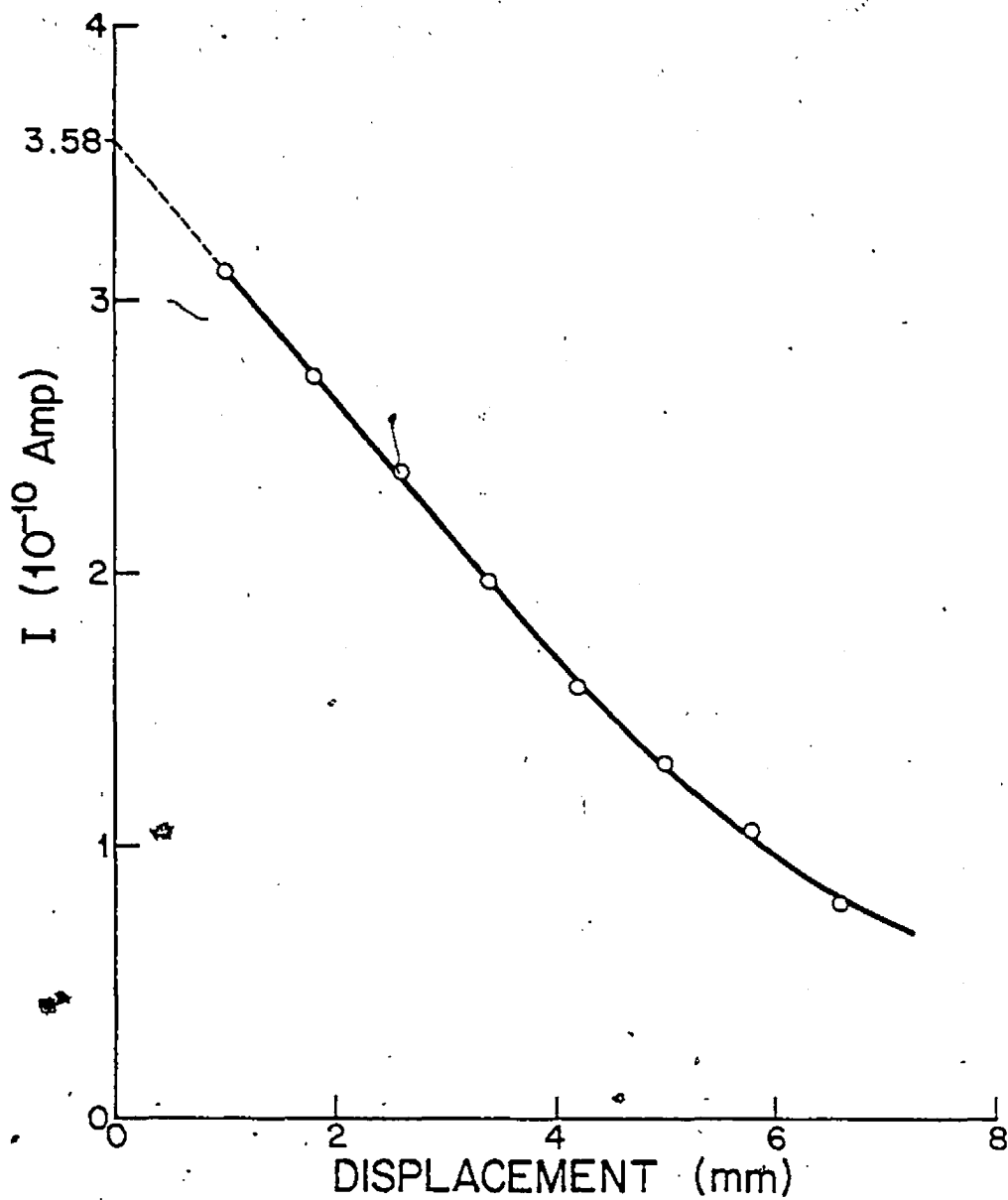


Fig. 13: Intensity of the Tl 3776 Å resonance fluorescence as a function of the displacement of the slit image from the observation window

$$P_{\text{Hg}} = \frac{d_{\text{oil}}}{d_{\text{Hg}}} P_{\text{oil}}$$

where d_{oil} is the density of #704 pump oil (1.05gm/cm^3) and d_{Hg} is the density of mercury (13.54gm/cm^3). P_{oil} is the height of the #704 pump oil in the manometer and P_{Hg} is the corresponding height of an equivalent mercury column at room temperature. The nitrogen partial pressure must be corrected for thermal transpiration effects between the cell and the vacuum system since the thermal gradient between these two regions is very large (about 500°C) and the two regions are connected by a long capillary.

The pressure in the cell is related to the pressure in the manometer, (at room temperature), by the equation

$$P_{\text{cell}} = \sqrt{\frac{T_{\text{cell}}}{T_{\text{room}}}} P_{\text{man}}$$

where T_{cell} is the temperature of the cell in degrees Kelvin and T_{room} is the room temperature in degrees Kelvin (see Appendix 15).

In order to verify that the observed quenching cross sections for the thallium states are due completely to the presence of the nitrogen molecules, and that no other apparent quenching processes related to the metal-metal collisional excitation energy transfer phenomenon were influencing the cross sections, the effect of N_2 molecules on the fluorescence emitted from the Tl-N_2 vapour was investigated. We investigated the quenching action of N_2 molecules on the $\text{Tl}(7^2\text{S}_{1/2})$ resonance state which could be directly populated by irradiating the Tl-N_2 mixture with the $\text{Tl } 3776\text{\AA}$ resonance radiation. Light emitted from the thallium Osram lamp source was first passed through an interference filter with a peak transmission of 21 percent at 3776\AA and, as before, was

focused inside a new clean cell containing only pure thallium in its sidearm. The oven temperature was brought up slowly and allowed to stabilize with a sidearm temperature of about 530°C , which corresponds to a thallium density of about $6 \times 10^{12}\text{cm}^{-3}$. The resonance fluorescence emitted by the Tl- N_2 mixture was focused onto the entrance slit of the monochromator which alternately scanned the two resonance components at 3776\AA and 5352\AA . The intensities of these spectral components were first measured in the absence of N_2 molecules. Successive amounts of nitrogen, in the range 0.2 torr - 8 torr, were admitted into the cell and the intensities of the resonance fluorescent components were recorded for each nitrogen pressure used. The quenching cross section for the $\text{Tl}(7^2\text{S}_{1/2})$ resonance state obtained in this manner could be directly compared with the cross section obtained for the resonance state populated by the collisional excitation energy transfer from excited mercury atoms.

EXPERIMENTAL RESULTS AND DISCUSSION

(1.0) Energy Transfer Cross Sections

Before proceeding with a detailed discussion of the results, we first point out one major source of disagreement between our findings and those of several other authors working with the binary metal vapour system comprised of excited mercury atoms and ground state thallium atoms. Kraulinya and Lezdin¹⁶ and Anderson²³, working at temperatures between 650°C and 850°C, report the appearance of sensitized fluorescent components originating from the $Tl(8^2D_{3/2,5/2}, 7^2D_{3/2,5/2}, 9^2S_{1/2})$ states. In this investigation, where the sensitivity of the detection equipment was superior to that employed by Anderson, we were unable to detect radiative transitions from these levels which lie above the $Hg(6^3P_1)$ state. If these components were present in the thallium sensitized fluorescence, their intensities were well below those which our instruments were able to detect. Using the current suppression device on the Keithley pico-ammeter, it was possible to detect intensity ratios down to about 10^{-7} . We were able to detect the spectral components originating from the $Tl(7^2S_{1/2}, 8^2S_{1/2}, 6^2D_{3/2,5/2})$ states although it was not possible to completely resolve the $Tl\ 3520\text{\AA}$ and $Tl\ 3530\text{\AA}$ spectral components resulting from the $6^2P_{3/2} - 6^2D_{3/2}$ and $6^2P_{3/2} - 6^2D_{5/2}$ transitions, respectively. The spectral components corresponding to transitions from the $Tl(6^2D_{3/2,5/2})$ states to the $Tl(7^2P_{1/2,3/2})$ states and from the $Tl(7^2P_{1/2,3/2})$ states to the $Tl(7^2S_{1/2})$ state could not be detected by our equipment since these spectral components are in the infrared region which is beyond the spectral response range of the photomultipliers used in this investigation, both having S-20 photocathodes. The sensitized fluorescent components observed here are in agreement with those detected by Hudson and Curnutte¹⁷, who also were unable to detect

radiative emissions from thallium states above the $\text{Hg}(6^3\text{P}_1)$ level at thallium densities comparable to those used by the other authors.

Absolute intensities of the thallium fluorescent spectral components and the mercury resonance radiation at 2537\AA were used to obtain the excitation energy transfer cross sections shown in table (1) which are compared to those values obtained by Kraulinya and Lezdin, and by Hudson and Curnutte. In determining these cross sections, corrections to the absolute intensity measurements were made to account for the reabsorption of the thallium resonance fluorescence, (which was determined to be less than 10 percent), and for variations in the spectral response characteristics of the monochromator and photomultiplier tubes at the different wave lengths encountered in the fluorescent spectrum. Further corrections were made for the absorption characteristics of the quartz optical system at the various wave lengths. In this latter correction we have also accounted for variations in the quartz absorption features that occur at higher temperatures as found by Eckstein and Selvar⁷⁰. Both photomultiplier tubes were used to determine the cross sections. The results from both tubes were completely consistent, indicating that the calibration of the tubes with respect to absolute intensity measurements was reliable.

In general, the energy transfer cross section Q_{Oj} associated with the j^{th} excited thallium state is related to the intensity of radiative emissions, I_{ij} , from the thallium state j to some lower state i , and to the population density, $N(\text{Hg}^*)$, of the $\text{Hg}(6^3\text{P}_1)$ state as indicated by equation (13). However, in a fashion similar to equation (10), the density of excited mercury atoms may be expressed in terms of the observed absolute intensity of the $\text{Hg } 2537\text{\AA}$ resonance radiation re-emitted from the cell as

TABLE 1: Cross Sections for Hg(6^3P_1) \rightarrow Tl Excitation Energy Transfer

Collision Process	ΔE (eV) (cm $^{-1}$)		COLLISION CROSS SECTION					
			This Investigation		Kraulinya and Lezdin		Hudson and Curnutte	
			Q(\AA^2)	T($^{\circ}$ C)	Q(\AA^2)	T($^{\circ}$ C)	Q(\AA^2)	T($^{\circ}$ C)
Hg(6^3P_1) \rightarrow Tl($8^2S_{1/2}$)	-0.083	- 666	3.0 \pm 0.6 3.0 \pm 0.6	437 542	33 1.5	660 800	2.2 1.3	800 900
Hg(6^3P_1) \rightarrow Tl($6^2D_{5/2}$)	-0.40	- 3212	0.3 \pm 0.06 0.3 \pm 0.06	437 542	55 33	660 800	9.4 8.5	800 900
Hg(6^3P_1) \rightarrow Tl($7^2P_{3/2}$)	-0.50	- 4251					} 17.6 23.3	800 900
Hg(6^3P_1) \rightarrow Tl($7^2P_{1/2}$)	-0.63	- 5352						
Hg(6^3P_1) \rightarrow Tl($7^2S_{1/2}$)	-1.58	-12935	0.05 \pm 0.03	542	155 150	660 800		
Hg(6^3P_1) \rightarrow Tl($7^2D_{3/2}$)	+0.32	+ 2599			6.8 3.3	660 800		
Hg(6^3P_1) \rightarrow Tl($7^2D_{5/2}$)	+0.33	+ 2637			1.5 2.3	660 800		

TABLE 1: Cross Sections for $\text{Hg}(6^3P_1) \rightarrow \text{Tl}$ Excitation Energy Transfer

Collision Process	ΔE (eV) (cm ⁻¹)	COLLISION CROSS SECTION					
		This Investigation		Kraulnaya and Lezdin		Hudson and Curnutte	
		$Q(\text{\AA}^2)$	T(°C)	$Q(\text{\AA}^2)$	T(°C)	$Q(\text{\AA}^2)$	T(°C)
$\text{Hg}(6^3P_1) \text{ Tl}(9^2S_{1/2})$	+0.45 + 3754	9	660	0.27	800		
$\text{Hg}(6^3P_1) \text{ Tl}(8^2D_{3/2,5/2})$	+0.66 + 5280	44	660	1.7	800		

$$N(\text{Hg}^*) = \frac{I_{2537\text{\AA}}^{\circ}}{A_{\text{Hg}} h\nu_{2537\text{\AA}}^{\circ}}$$

where $I_{2537\text{\AA}}^{\circ}$ is the absolute intensity of the mercury resonance line, A_{Hg} is the transition probability for the mercury transition $6^1S_0 - 6^3P_1$, and $\nu_{2537\text{\AA}}^{\circ}$ is the corresponding transition frequency. Hence, the general quenching cross section of the j^{th} excited thallium state may be rewritten in terms of the observable fluorescent intensities as

$$Q_{0j} = \frac{1}{\delta} \left[\frac{I_{1j} A_{\text{Hg}} h\nu_{2537\text{\AA}}^{\circ}}{I_{2537\text{\AA}}^{\circ} A_{1j} h\nu_{1j}} \sum_{l=j-1}^{\infty} A_{1l} - \sum_{k=j+1}^{\infty} \frac{N_k(\text{Tl}^*) A_{jk} A_{\text{Hg}} h\nu_{2537\text{\AA}}^{\circ}}{I_{2537\text{\AA}}^{\circ}} \right] \times \frac{1}{N_0(\text{Tl}) \langle v_{\text{rel}} \rangle} \quad (110)$$

where δ is as defined in equation (13a). The thallium transition probabilities are derived from the orthogonalized relativistic HFS wave functions in the velocity formulation as quoted in Appendix (8). The transition probability, A_{Hg} , for the mercury resonance transition is taken to be $8.55 \times 10^6 \text{sec}^{-1}$, as determined from the mean radiative lifetime of the $\text{Hg}(6^3P_1)$ state calculated by Deech and Baylis⁷¹.

In our present investigation, equation (110) may be greatly simplified. Since we are unable to observe any spectral components originating from thallium states above the $\text{Hg}(6^3P_1)$ level, we may assume that the population of the $\text{Tl}(8^2S_{1/2})$ and $\text{Tl}(6^2D_{3/2,5/2})$ states by cascade transitions from higher thallium levels may be neglected, and the second term in the square brackets of equation (110) may be ignored for these states. The $\text{Tl}(7^2S_{1/2})$ state will be affected by cascading from the $\text{Tl}(7^2P_{1/2,3/2})$ states which, in turn, are pumped by cascade transitions from the $\text{Tl}(8^2S_{1/2})$ and $\text{Tl}(6^2D_{3/2,5/2})$ states. Since all these transitions are in the infrared region, no direct measurement of their spectral intensities could be made to correct

the energy transfer cross section of the $Tl(7^2S_{1/2})$ state for these cascade effects. However, since the intensity ratio between any two spectral components is directly related to the ratio of the corresponding transition probabilities, it becomes possible to obtain a good estimate of the contribution to the energy transfer cross section of the $Tl(7^2S_{1/2})$ state due to the cascade transitions from the $Tl(7^2P_{1/2,3/2})$ states.

In order to obtain the cross section for the $Tl(7^2S_{1/2})$ state it is necessary to first determine the total population density of the $Tl(7^2P_{1/2,3/2})$ states. These latter states are populated both by direct collisional transfer from the $Hg(6^3P_1)$ atoms and by cascade transitions from the $Tl(8^2S_{1/2})$ and $Tl(6^2D_{3/2,5/2})$ states. According to Franck's empirical rule, which is well supported by recent studies in numerous binary metal vapour systems,^{8,9,10,18,19,7} the probability that, during an inelastic collision between an excited and unexcited atom, the second collisional partner will become excited to some higher excited state decreases as the energy defect ΔE between the excited states of the collision partners increases. Hence, on the basis of these results and since the transition probabilities linking the $Tl(8^2S_{1/2})$ and $Tl(6^2D_{3/2,5/2})$ states with the $Tl(7^2P_{1/2,3/2})$ states are large, we assume that the contribution to the total population density of the $Tl(7^2P_{1/2,3/2})$ states, (which lie relatively far below the $Hg(6^3P_1)$ level), due to the direct collisional energy transfer will be negligible with respect to the pumping of these states by cascading from the $Tl(8^2S_{1/2})$ and $Tl(6^2D_{3/2,5/2})$ states. To determine the contributions to the total population density of the $Tl(7^2P_{1/2,3/2})$ states, we must know the intensities of the transitions from the higher excited states. If we know the intensities corresponding to other observable transitions originating from these higher excited states we can determine the unobservable intensities of the infrared spectral components.

The intensity I' of the infrared component resulting from the transition from one of the upper excited states to the $Tl(7^2P_{1/2,3/2})$ state is related to the observable intensity I , corresponding to a transition from the same upper state, according to the relation

$$I' = I \frac{\nu' A'}{\nu A} \quad (111)$$

where use has been made of equation (10) and where ν' is the transition frequency of the infrared component and A' is its corresponding transition probability, ν is the frequency of the observable transition of intensity I and A is the associated transition probability. The transition probabilities are taken from Appendix (8). Using these intensities I' , the population densities of the $Tl(7^2P_{1/2,3/2})$ states can be determined and finally the cross section of the $Tl(7^2S_{1/2})$ state can be calculated, using equation (110).

It is evident from table (1) that the experimental results of Kraulinya and Lezdin give no indication of a resonance effect between the energy transfer cross sections and the corresponding energy defects ΔE between the $Hg(6^3P_1)$ state and the excited thallium levels. For $\Delta E = -0.083eV$, (corresponding to the $Tl(8^2S_{1/2})$ state), our cross section compares favourably with their value of $1.5 \times 10^{-16} cm^2$ obtained at a temperature of $800^\circ C$. However, their value of $33 \times 10^{-16} cm^2$ obtained at a temperature of $660^\circ C$ is in total disagreement with our value of $3.0 \times 10^{-16} cm^2$ obtained at temperatures of $542^\circ C$ and $437^\circ C$. The cross section for the $Tl(6^2D_{5/2})$ state obtained in this investigation, with $\Delta E = -0.40eV$ is 2 orders of magnitude lower than that obtained by Kraulinya and Lezdin. Their cross section for the $Tl(6^2D_{5/2})$ state is considerably larger than their cross section for the $Tl(8^2S_{1/2})$ state, a result which is in violation of Franck's empirical rule.

In order to explain the appearance of the sensitized fluorescent components originating from thallium states above the $Hg(6^3P_1)$ level, Kraulinya

and Lezdin anticipate the participation of the $Tl(6^2P_{3/2})$ metastable atoms in a process governed by equation (26). They further indicate the participation of some unspecified molecular formation, (most probably taken to mean the mercury-thallium excimer), in populating the excited thallium states which lie above the $Hg(6^3P_1)$ level.

Recently, mercury-thallium excimers have been observed by Drummond and Schlie⁷² with molecular bands at 4600\AA and 6500\AA , but this was at Hg-Tl amalgam densities on the order of $1.5 \times 10^{19}\text{cm}^{-3}$ which exceeds the densities used in this investigation by 7 - 8 orders of magnitude. No such molecular bands could be detected in our investigation, and it seems likely that the participation of such a molecular species, as suggested by Kraulinya, is improbable.

Pickett and Anderson⁷³ have reported lifetimes and self-quenching cross sections for the $Tl(6^2P_{3/2})$ metastable state. Their measurements, performed at 550°C , indicate a lifetime of the order of 10^{-4} seconds. However, such a lifetime gives rise to a large diffusion cross section (of the order of 10^{-14}cm^2) for quenching collisions with the cell walls. A crude order of magnitude estimate based on the value of $\langle v_{rel} \rangle$ characteristic of our experimental conditions gives rise to a mean time of about 10^{-6} seconds required for the $Tl(6^2P_{3/2})$ atoms to migrate across the 1mm distance between the fluorescing region and the cell walls. This time is considerably less than the lifetime of the thallium metastable state and hence the participation of these metastables in the energy transfer process, as defined by equation (26), may be neglected.

The results of Hudson and Curnutte are difficult to assess in relation to our cross sections since they have failed to define the meaning of ΔE . In their paper they determine cross sections for a total energy transfer

from combined $\text{Hg}(6^3P_1)$ and $\text{Hg}(6^3P_0)$ states to the thallium atom.

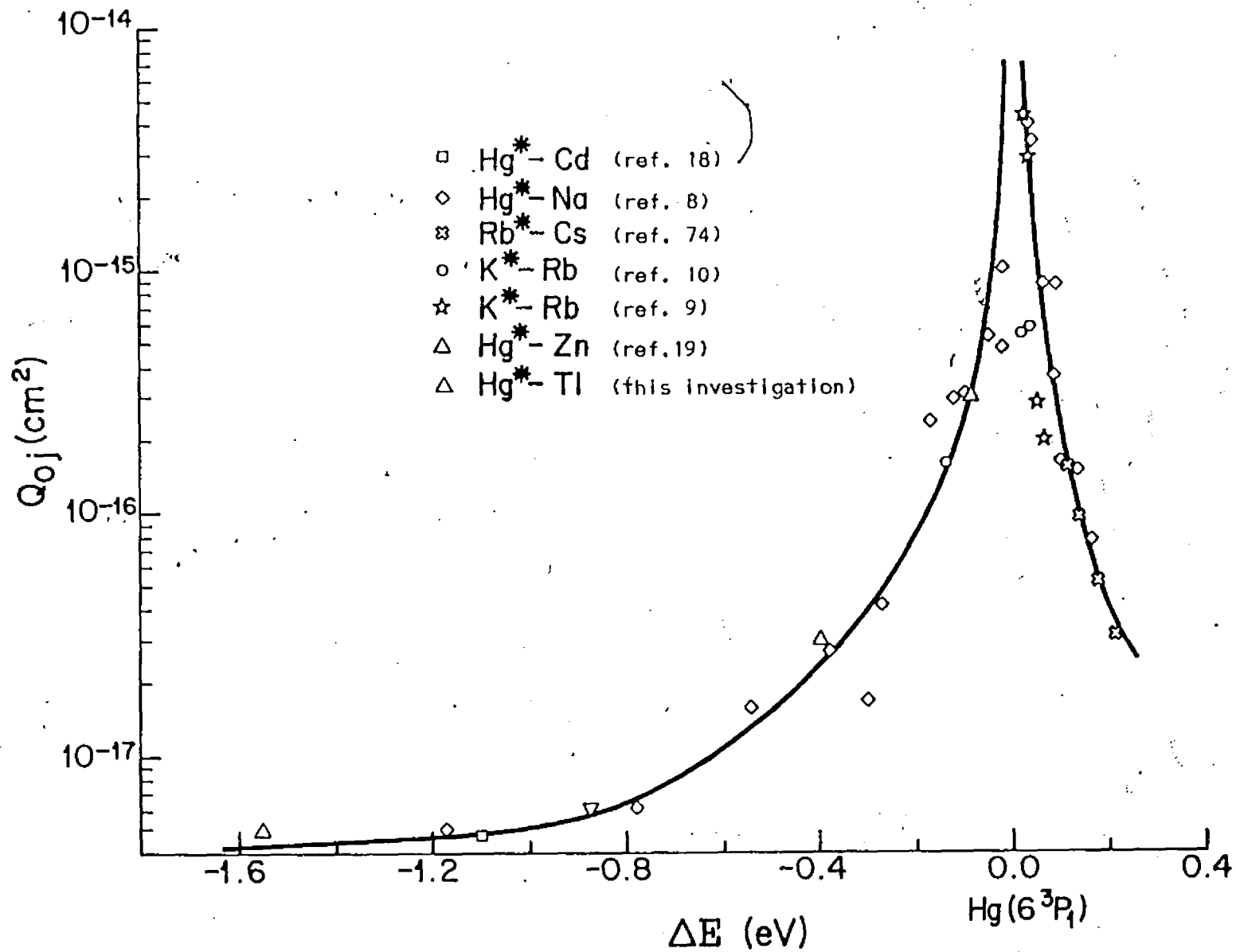
The energy transfer cross sections, as determined by the methods of this investigation, are plotted on the general resonance curve (shown in figure (14)) that has been generated from cross sections measured in several experiments using various metal-metal systems at this University^{8, 9, 18, 19, 74}. It can be seen that the thallium values fit well to the established trends. Since the energy transfer cross sections are also dependent on the relative velocity of the colliding partners, one might question the validity of comparing the cross sections from the different binary metal vapour systems on the same resonance curve. However, the mean relative thermal velocities for all experimental data are of the same order of magnitude, hence a comparison of the respective cross sections on a common resonance curve is not unjustified.

The fact that all cross sections for the various systems studied exhibit a common dependence of ΔE may be regarded as a significant indication that excitation energy transfer in binary metal vapour systems proceeds by a single mechanism. In particular, the results of this investigation indicate that the energy transfer from excited mercury atoms to ground state thallium atoms proceeds by way of pure binary collisions as defined by equation (25), and does not require the participation of HgTl excimers, mercury molecular formations, or the metastable $\text{Tl}(6^2P_{3/2})$ atoms at the temperatures used here. However, it may be possible that at higher mercury and thallium densities some of these other mechanisms leading to the population of higher quantum states in thallium may become significantly prevalent so as to influence the now apparent trends.

(2.0) Quenching Cross Sections

As we have mentioned previously, it is necessary to verify that

Fig. 14: General "resonance curve" showing dependence of excitation transfer cross sections between dissimilar atoms on ΔE



measured quenching cross sections obtained under sensitized fluorescence remain independent of the metal-metal collisional excitation transfer process. In confirming this independence we compare the quenching cross section of the resonance state found under the conditions of sensitized fluorescence with that determined from the direct optical excitation of the $Tl(7^2S_{1/2})$ state in the $Tl-N_2$ mixture.

Direct optical excitation of the $Tl(7^2S_{1/2})$ resonance state was accomplished by radiating the $Tl-N_2$ mixture with the $Tl\ 3776\text{\AA}$ resonance radiation emitted from the filtered Osram lamp source. The subsequent depopulation of this resonance state via the transitions $6^2P_{1/2} - 7^2S_{1/2}$ and $6^2P_{3/2} - 7^2S_{1/2}$, which give rise to the emission of the $Tl\ 3776\text{\AA}$ and $Tl\ 5352\text{\AA}$ resonance fluorescence, could be monitored as a function of increasing nitrogen density. The results of the quenching action of nitrogen molecules on the thallium resonance state are shown in tables (2) and (3) as obtained by monitoring the $Tl\ 5352\text{\AA}$ and $Tl\ 3776\text{\AA}$ emissions, respectively. In these and subsequent tables, we use I_0 to denote the fluorescent intensity at zero nitrogen density and I_{N_2} to denote the fluorescent intensities at higher nitrogen densities. Corresponding Stern-Volmer plots are depicted in figures (15) and (16). The slope of such curves is given as $(Q \tau_{7^2S_{1/2}} \langle v_{rel} \rangle)$, where Q is the quenching cross section to be determined, $\langle v_{rel} \rangle$ is the relative thermal velocity of the colliding thallium atoms and nitrogen molecules, and $\tau_{7^2S_{1/2}}$ is the lifetime of the resonance state. As before we use our theoretical values for this lifetime. We calculate the quenching cross section of the thallium resonance state from the slopes of the Stern-Volmer plots in figures (15) and (16) to be 11.4\AA^2 and 10.3\AA^2 , respectively.

In the binary metal vapour system comprised of excited mercury atoms

TABLE 2: Quenching of the $\text{Tl}(7^2\text{S}_{1/2})$ state as determined by monitoring the $\text{Tl } 5352\text{\AA}$ resonance radiation emitted from the $\text{Tl}-\text{N}_2$ mixture.

$$\langle v_{\text{rel}} \rangle = 8.45 \times 10^4 \text{ cm/sec}$$

Pressure (torr)	Density (10^{16} cm^{-3})	I_0/I_{N_2}
0.00	0.00	1.000
0.24	0.48	1.015
0.64	1.27	1.029
0.72	1.42	1.032
0.88	1.74	1.038
0.96	1.90	1.046
1.20	2.30	1.016
1.24	2.37	1.017
1.36	2.69	1.026
1.52	2.91	1.032
1.60	3.16	1.051
1.68	3.32	1.052
1.88	3.60	1.033
1.92	3.80	1.052
2.12	4.06	1.039
2.40	4.75	1.061
2.56	4.90	1.046
3.12	5.97	1.060
3.32	6.57	1.078
5.00	9.89	1.110

Fig. 15: Stern-Volmer plot for data obtained by monitoring the TI 5352A resonance radiation emitted from the TI-N₂ mixture

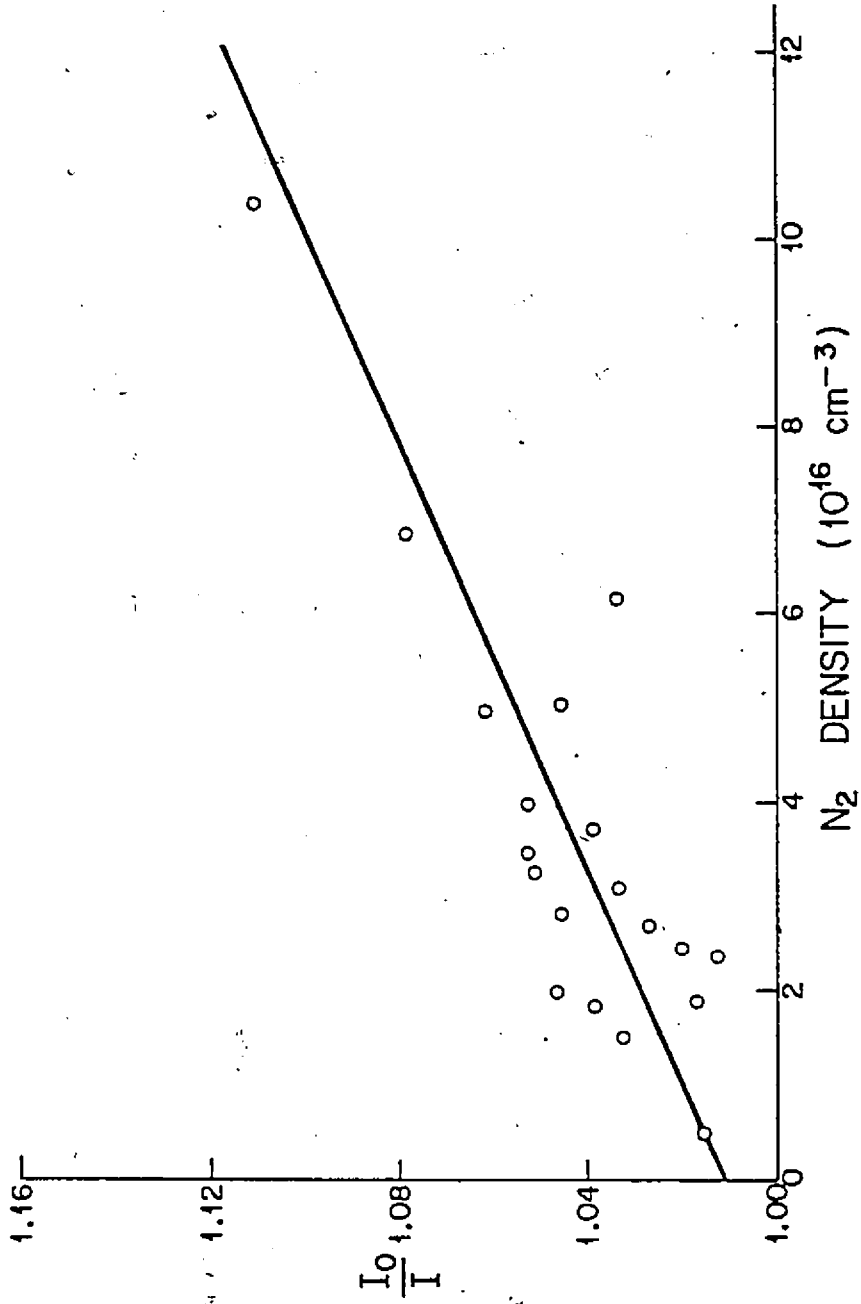
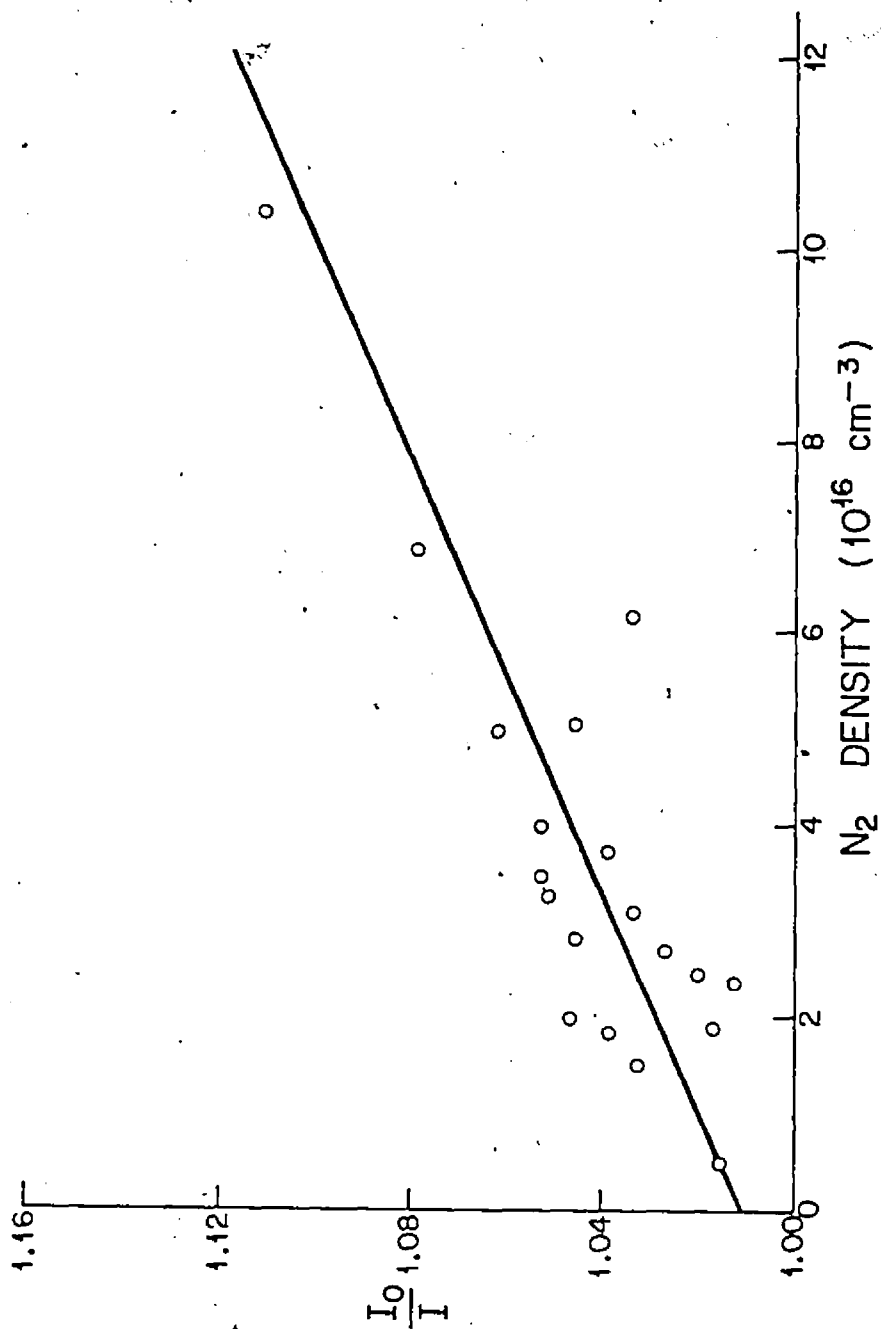


TABLE 3: Quenching of the $Tl(7^2S_{1/2})$ state as determined by monitoring the $Tl3776\text{\AA}$ resonance radiation emitted from the $Tl-N_2$ mixture.

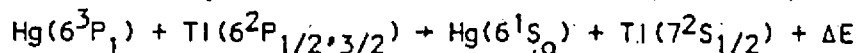
$$\langle v_{rel} \rangle = 8.28 \times 10^4 \text{ cm/sec}$$

N_2		I_0/I_{N_2}
Pressure (torr)	Density (10^{16} cm^{-3})	
0.0	0.0	1.000
0.24	0.48	1.025
0.64	1.27	1.035
0.80	1.59	1.038
1.28	2.55	1.028
1.32	2.63	1.031
1.36	2.69	1.039
1.52	3.01	1.040
1.60	3.17	1.048
1.76	3.51	1.035
1.92	3.83	1.042
2.12	4.20	1.056
2.16	4.28	1.058
2.80	5.55	1.060
2.84	5.66	1.060
3.28	6.54	1.061
4.28	8.48	1.109
4.36	8.69	1.079
5.68	11.30	1.111

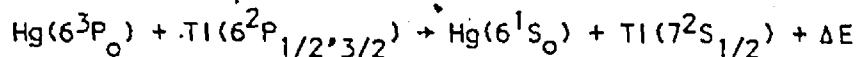
Fig. 16: Stern-Volmer plot for data obtained by monitoring the TI 3776Å resonance radiation emitted from the TI-N₂ mixture



and ground state thallium atoms the thallium resonance state is populated by the transfer of the mercury excitation energy to the ground state thallium atoms. As indicated previously, this process may be defined by the equation



and in the presence of nitrogen molecules the additional process



is also of significance. As in the case of direct optical excitation, the quenching cross section of the thallium resonance state may be determined by monitoring the variations in the intensity of either of the two resonance components as a function of nitrogen density in the tertiary Hg-Tl-N₂ system. Data indicating the quenching action of nitrogen molecules on the thallium resonance state, under the conditions of sensitized fluorescence, is displayed in tables (4) and (5) as obtained by monitoring the Tl 5352^oÅ and Tl 3776^oÅ resonance lines, respectively. The corresponding Stern-Volmer plots are shown in figures (17) and (18). In forming these plots from tables (4) and (5) we are concerned only with that portion of the data which reflects the quenching process. Hence, ratios $I_{N_2}^o / I_{N_2}$ are formed, where $I_{N_2}^o$

is the intensity of the thallium fluorescence corresponding to that for which, at a certain nitrogen density, the onset of quenching appears. As before, I_{N_2} is the intensity of the fluorescence at higher nitrogen pressures.

We determine the quenching cross section of the thallium resonance state from the slopes of these two plots to be 10.2^oÅ² and 11.1^oÅ², respectively.

Since the cross section for the Tl(7²S_{1/2}) resonance state obtained under direct optical excitation of the Tl-N₂ mixture agrees closely with

TABLE 4: Quenching of the Tl($7^2S_{1/2}$) state as determined by monitoring the Tl 5352\AA resonance radiation emitted from the Hg-Tl- N_2 mixture.

$$\langle v_{rel} \rangle = 8.57 \times 10^4 \text{ cm/sec}$$

N_2			
Pressure (torr)	Density (10^{16} cm^{-3})	I_{N_2}/I_0	I_{N_2}/I_{N_2}
0.0	0.0	1.000	
0.80	1.54	2.000	
1.52	2.92	2.427	
2.64	5.07	2.571	1.000
3.48	6.68	2.538	1.014
4.16	7.99	2.500	1.029
5.60	10.80	2.445	1.053
6.64	12.80	2.392	1.075
7.60	14.60	2.358	1.091

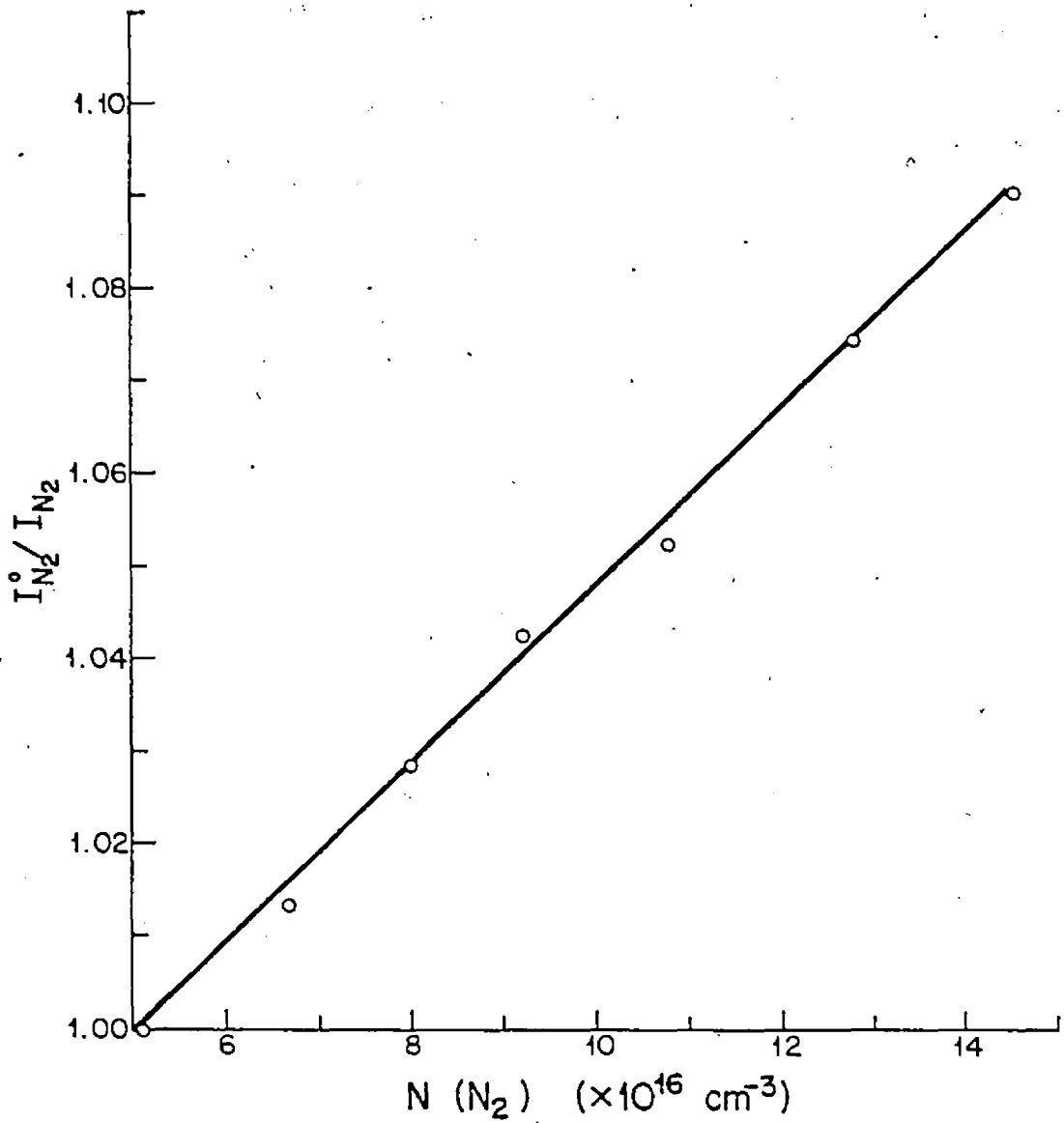


Fig. 17: Stern-Volmer plot for data obtained by monitoring the $\text{TI } 5352\text{\AA}$ resonance radiation emitted from the Hg-Tl-N_2 mixture.

TABLE 5: Quenching of the $\text{Tl}(7^2\text{S}_{1/2})$ state as determined by monitoring the $\text{Tl } 3776\text{\AA}$ resonance radiation emitted from the Hg-Tl-N_2 mixture.

$$\langle v_{\text{rel}} \rangle = 8.57 \times 10^4 \text{ cm/sec}$$

Pressure (torr)	N_2 Density (10^{16} cm^{-3})	I_{N_2}/I_0	$I_{\text{N}_2}^0/I_{\text{N}_2}$
0.0	0.0	1.00	
0.80	1.54	3.571	
1.52	2.92	5.556	
2.64	5.07	6.667	1.000
3.48	6.68	6.579	1.014
4.16	7.99	6.536	1.022
4.80	9.22	6.410	1.037
5.60	10.80	6.329	1.057
6.64	12.80	6.173	1.077
7.60	14.60	6.061	1.098

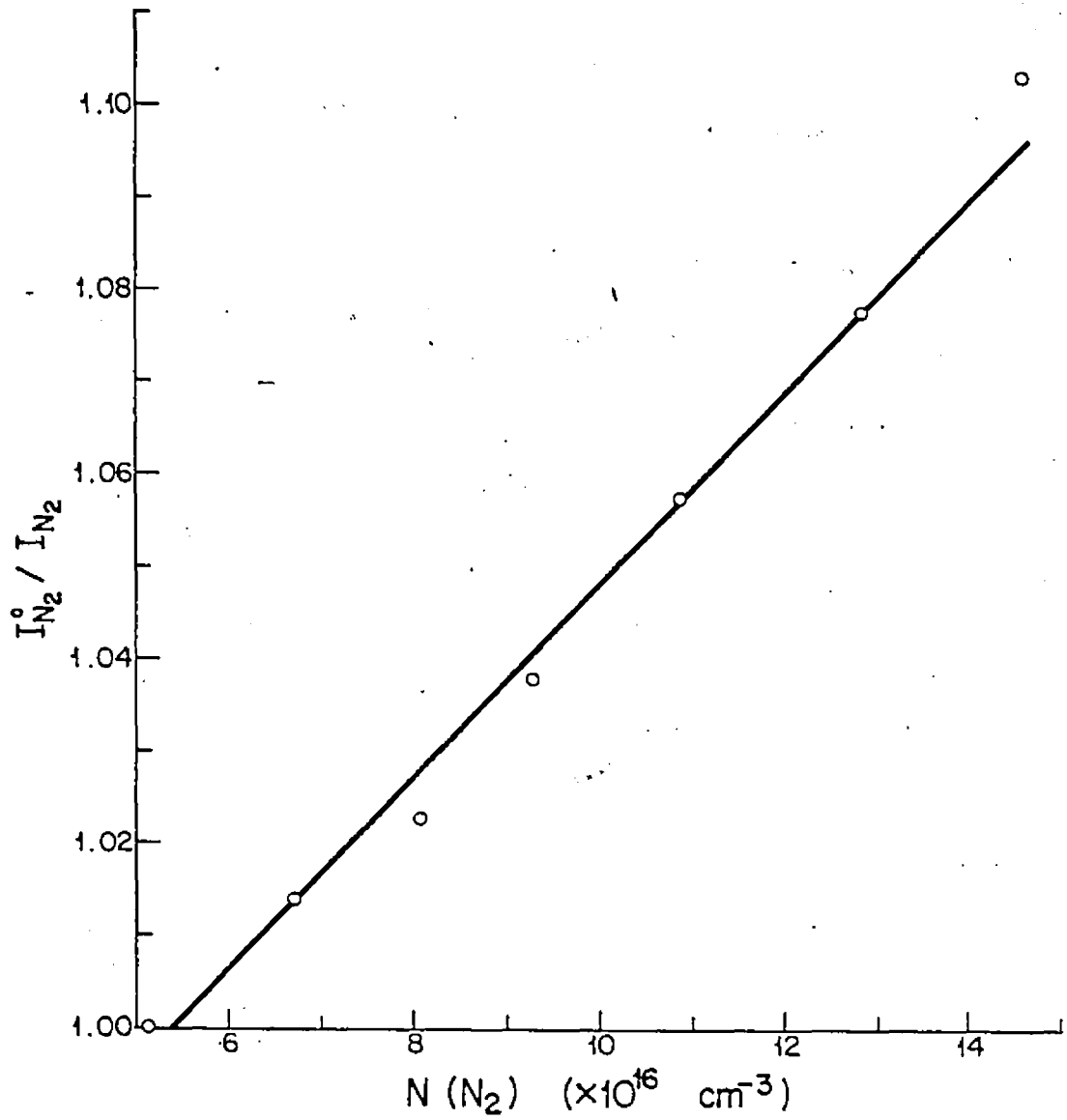


Fig. 18: Stern-Volmer plot for data obtained by monitoring the TI 3776Å resonance radiation emitted from the Hg-Tl-N₂ mixture

that found for the Hg-Tl-N₂ system under sensitized fluorescence, one may be assured that the quenching cross section measurements obtained under sensitized fluorescence are not coupled to the inelastic collision process occurring between the excited mercury atoms and the unexcited thallium atoms. In light of this agreement, we may safely proceed to determine the quenching cross sections for other excited thallium states contributing to the sensitized fluorescent spectrum with confidence that the observed reductions in the emitted intensities of the thallium spectral components with increasing nitrogen pressures (up to about 8 torr) are due solely to the pure quenching action of nitrogen molecules present in the mixture.

Figure (19) shows plots of the experimentally determined ratios of sensitized fluorescent intensities I_{N_2}/I_0 as a function of the nitrogen density for the Tl 3776Å, Tl 5352Å, Tl 3231Å and Tl 3520Å fluorescent components arising from the transitions $6^2P_{1/2} - 7^2S_{1/2}$, $6^2P_{3/2} - 7^2S_{1/2}$, $6^2P_{3/2} - 8^2S_{1/2}$, $6^2P_{3/2} - 6^2D_{5/2}$, respectively. The relevant data for the intensity ratios I_{N_2}/I_0 has been extracted from tables (4), (5), (6), and (7).

Under the experimental conditions employed in this investigation, the atomic densities of both thallium and mercury remain constant during a given experimental run as does their relative thermal velocity. Hence, the variations in the intensity ratios I_{N_2}/I_0 with N₂ density provide direct information about the variations in the population densities of the excited thallium states resulting both directly and indirectly from inelastic collisions with nitrogen molecules.

From equation (27) it is evident that the effect of Hg(6^3P_1) - N₂ collisions is to populate the Hg(6^3P_0) metastable state from which, in turn, this excitation energy can be transferred collisionally to Tl($6^2P_{1/2}$)

Fig. 19: Variations in the intensities of the sensitized fluorescent spectral components of thallium as a function of N_2 density. Also shown is the absorption curve obtained by Gatzke for the Hg 4047Å spectral component indicating the dependence of the population density of the Hg(6^3P_0) metastable atoms as a function of nitrogen density. A-Tl 3776Å; B-Tl 5352 Å; C-Tl 3520Å; D-Tl 3231Å

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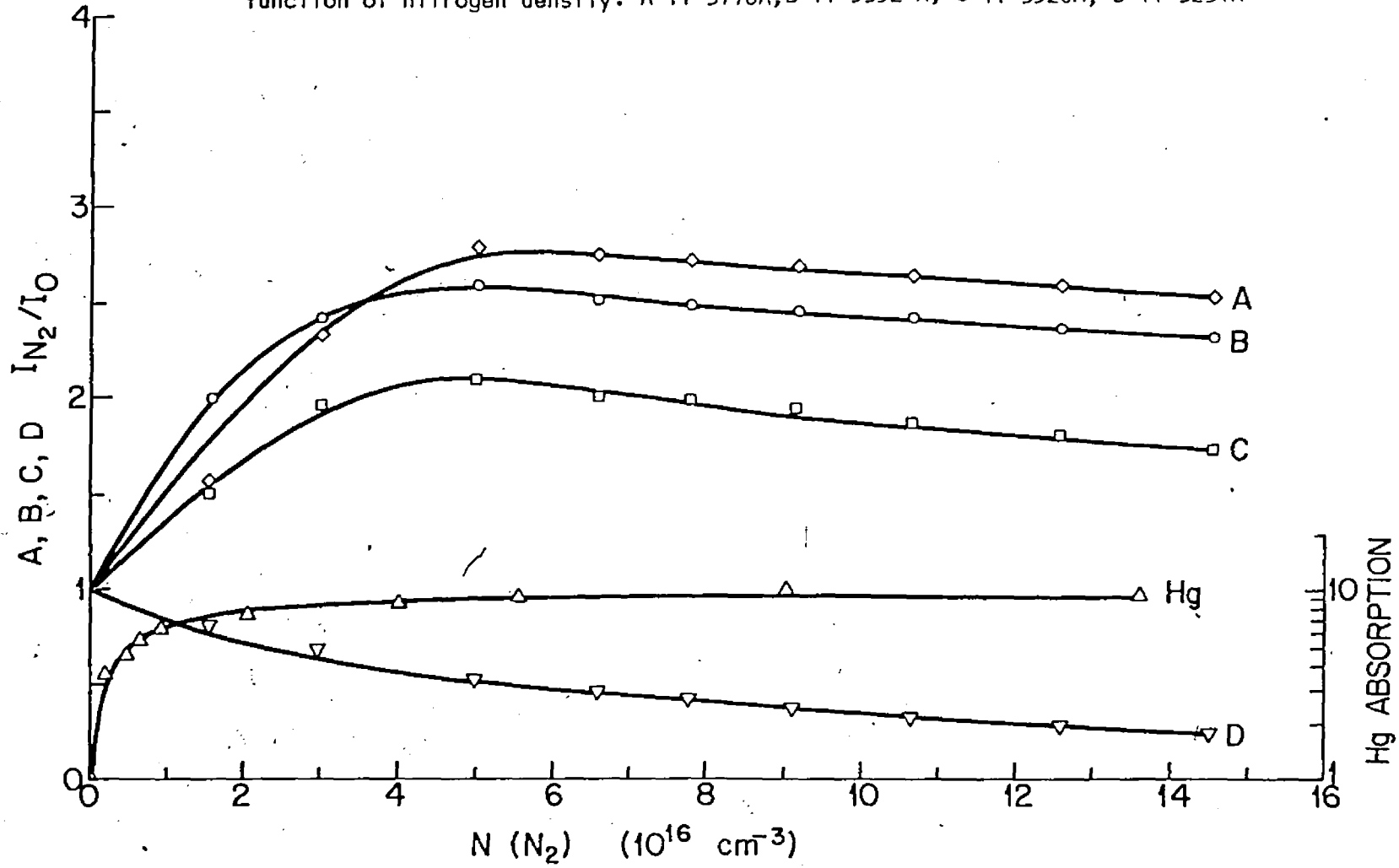


TABLE 6: Quenching of the $Tl(8^2S_{1/2})$ state as determined by monitoring the $Tl\ 3231\text{\AA}$ radiation emitted from the Hg-Tl- N_2 mixture.

$$\langle v_{rel} \rangle = 8.57 \times 10^4 \text{ cm/sec}$$

N_2 Pressure (torr)	Density (10^{16} cm^{-3})	I_{N_2}/I_0	$I_{N_2}^0/I_{N_2}$
0.0	0.0	1.000	1.000
0.80	1.54	0.800	1.250
1.52	2.92	0.680	1.471
2.64	5.07	0.504	1.984
3.48	6.68	0.468	2.137
4.16	7.99	0.440	2.273
4.80	9.22	0.360	2.778
5.60	10.80	0.320	3.125
6.64	12.80	0.280	3.571
7.60	14.60	0.240	4.167



TABLE 7: Quenching of the $Tl(6^2D_{5/2})$ state as determined by monitoring the $Tl\ 3520\text{\AA}$ radiation emitted from the Hg-Tl- N_2 mixture.

$$\langle v_{rel} \rangle = 8.97 \times 10^4 \text{ cm/sec}$$

N_2		I_{N_2}/I_0	$I_0_{N_2}/I_{N_2}$
Pressure (torr)	Density (10^{16} cm^{-3})		
0.0	0.0	1.000	
0.80	1.54	1.522	
1.52	2.92	1.957	
2.64	5.07	2.101	1.000
3.48	6.68	2.045	1.028
4.16	7.99	1.992	1.055
4.80	9.22	1.957	1.074
5.60	10.80	1.883	1.115
6.64	12.80	1.812	1.160
7.60	14.60	1.739	1.208

ground state atoms and $Tl(6^2P_{3/2})$ metastable state atoms resulting in the population of higher excited thallium states as indicated in equations (28) and (29). The population densities of such excited thallium states depend critically on the concentration of $Hg(6^3P_0)$ atoms and hence indirectly depend on the density of N_2 molecules as well as on the magnitude of the resonance defect ΔE between the particular thallium state and the mercury metastable state. In the case of close energy resonance, (that is, ΔE is small), one would expect that in the presence of N_2 molecules the intensity ratio I_{N_2}/I_0 should show an initial enhancement with increasing nitrogen pressures. The competing quenching process defined by equation (30), which results from a direct interaction between the excited thallium atom and a ground-state nitrogen molecule, causes radiationless transitions from the higher excited states to the ground state directly, or indirectly in a series of cascade transitions, with the simultaneous excitation of the nitrogen molecule to higher vibrational states from its ground electronic configuration. Such a process would manifest itself by reducing the ratio I_{N_2}/I_0 with increasing nitrogen pressures.

The presence of these two competing processes together are responsible for the variations in I_{N_2}/I_0 with N_2 pressure as indicated by the curves in figure (19). Also shown in figure (19) is the absorption curve obtained by Gatzke⁷⁶ for the $Hg\ 4047\text{\AA}$ spectral component arising from the mercury transition $6^3P_0 + 7^3S_1$ which indicates the dependence of the concentration of metastable $Hg(6^3P_0)$ atoms on the nitrogen density at mercury vapour pressures similar to those employed in this experiment. It is of significance that the I_{N_2}/I_0 curves corresponding to the $Tl(7^2S_{1/2})$ and the $Tl(6^2D_{5/2})$ states, reach their maxima in the near vicinity of the saturation point of the absorption curve which occurs at about 1 torr. At pressures above this saturation point the

density of $\text{Hg}(6^3\text{P}_0)$ metastable atoms remains constant and the only factor affecting the shapes of the I_{N_2}/I_0 curves is the quenching process. In the case of the $\text{Tl}(8^2\text{S}_{1/2})$ state the quenching effect is much stronger as indicated by the immediate decrease in the I_{N_2}/I_0 curve with increasing nitrogen density.

A basis for rationalizing these results is afforded by considering the relative levels of the various excited thallium states connected with the I_{N_2}/I_0 curves with respect to the $\text{Hg}(6^3\text{P}_0)$ state which becomes populated with the introduction of N_2 molecules into the Hg-Tl system. The $\text{Tl}(7^2\text{S}_{1/2})$ and $\text{Tl}(6^2\text{D}_{5/2})$ states are brought into a closer resonance with the $\text{Hg}(6^3\text{P}_0)$ level ($\Delta E = -1.38\text{eV}$ and $\Delta E = -0.20\text{eV}$, respectively) than exists for the $\text{Hg}(6^3\text{P}_1)$ level and the thallium states ($\Delta E = -1.58\text{eV}$ and $\Delta E = -0.40\text{eV}$, respectively). One would expect that this closer resonance, combined with the relatively long lifetime of the $\text{Hg}(6^3\text{P}_0)$ state, would lead to an efficient energy transfer from the $\text{Hg}(6^3\text{P}_0)$ atoms to the $\text{Tl}(7^2\text{S}_{1/2})$ and $\text{Tl}(6^2\text{D}_{5/2})$ atoms. Since the population density of the $\text{Hg}(6^3\text{P}_0)$ state increases with N_2 density the I_{N_2}/I_0 curves would be expected to rise with increasing N_2 density up to about $5 \times 10^{16}\text{cm}^{-3}$. With further increases in the N_2 density, the density of the $\text{Hg}(6^3\text{P}_0)$ atoms remains constant and the only remaining process is that of the quenching of the excited $\text{Tl}(7^2\text{S}_{1/2})$ and $\text{Tl}(6^2\text{D}_{5/2})$ states which is reflected by the decreasing I_{N_2}/I_0 ratios. In the case of the $\text{Tl}(8^2\text{S}_{1/2})$ state, which lies above the $\text{Hg}(6^3\text{P}_0)$ level, the energy defect between the $\text{Hg}(6^3\text{P}_0)$ level and the $\text{Tl}(8^2\text{S}_{1/2})$ level ($\Delta E = +0.12\text{eV}$) is larger than that for the $\text{Hg}(6^3\text{P}_1)$ state and the $\text{Tl}(8^2\text{S}_{1/2})$ state ($\Delta E = -0.083\text{eV}$). In addition, a Boltzmann factor must be introduced²⁶ when considering the energy transfer from the $\text{Hg}(6^3\text{P}_0)$ state to the $\text{Tl}(8^2\text{S}_{1/2})$ state. Taking these factors into account, we estimate that the $\text{Hg}(6^3\text{P}_0)$ states populate the $\text{Tl}(8^2\text{S}_{1/2})$ state to less than one-tenth of the level to which the $\text{Hg}(6^3\text{P}_1)$

atoms populated this thallium state during inelastic collisions. Hence, one would expect that the dominant process involving the N_2 molecules in the case of the $Tl(8^2S_{1/2})$ state would be that of quenching and indeed this is reflected in the I_{N_2}/I_0 curve which begins to decrease immediately upon the introduction of nitrogen gas into the Hg-Tl vapour mixture.

The actual quenching cross sections may be obtained from the intensity measurements at various nitrogen pressures in the range 0 torr - 8 torr, as found in tables (4), (5), (6) and (7), by using the Stern-Volmer relation

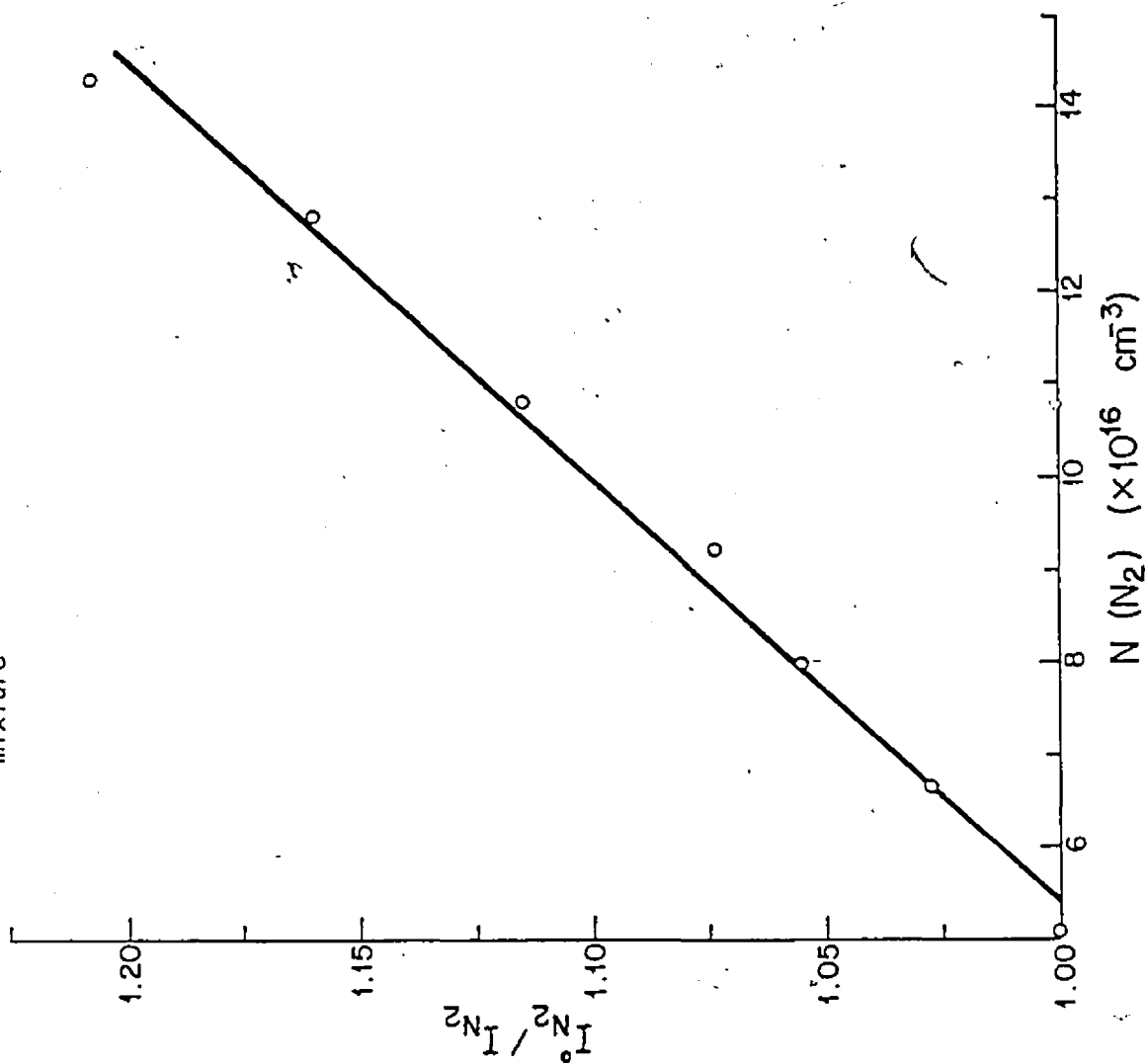
$$I_{N_2} = I_{N_2}^0 / (1 + \tau Z)$$

where $I_{N_2}^0$ is the relative intensity of the sensitized fluorescent component at some fixed N_2 pressure beyond which the quenching process is dominant and I_{N_2} is the relative intensity of the same spectral component at higher N_2 pressures. The lifetime of the appropriate thallium state, τ , is taken from our theoretical calculations which have been done in the velocity formulation using the orthogonalized HFS wave functions. The quenching cross sections are determined from the Stern-Volmer relation by first substituting for the frequency of quenching collisions, Z , in terms of the quenching cross section, Q , according to the relation

$$Z = n_2 \langle v_{rel} \rangle Q$$

where n_2 is the density of N_2 molecules and $\langle v_{rel} \rangle$ is the mean relative thermal velocity of the colliding partners. The quenching cross section Q is given by equation (9) where, in this case, the intensity ratio $I_{N_2}/I_{N_2}^0$ is to be used. The Stern-Volmer plots of $I_{N_2}/I_{N_2}^0$ as a function nitrogen density for the sensitized fluorescent components arising from the thallium transitions $6^2P_{1/2} - 7^2S_{1/2}$, $6^2P_{3/2} - 7^2S_{1/2}$, $6^2P_{3/2} - 6^2D_{5/2}$ and $6^2P_{3/2} - 8^2S_{1/2}$ are

Fig. 20: Stern-Volmer plot for data obtained by monitoring the $\text{Tl } 3520\text{\AA}$ radiation emitted from the Hg-Tl-N₂ mixture



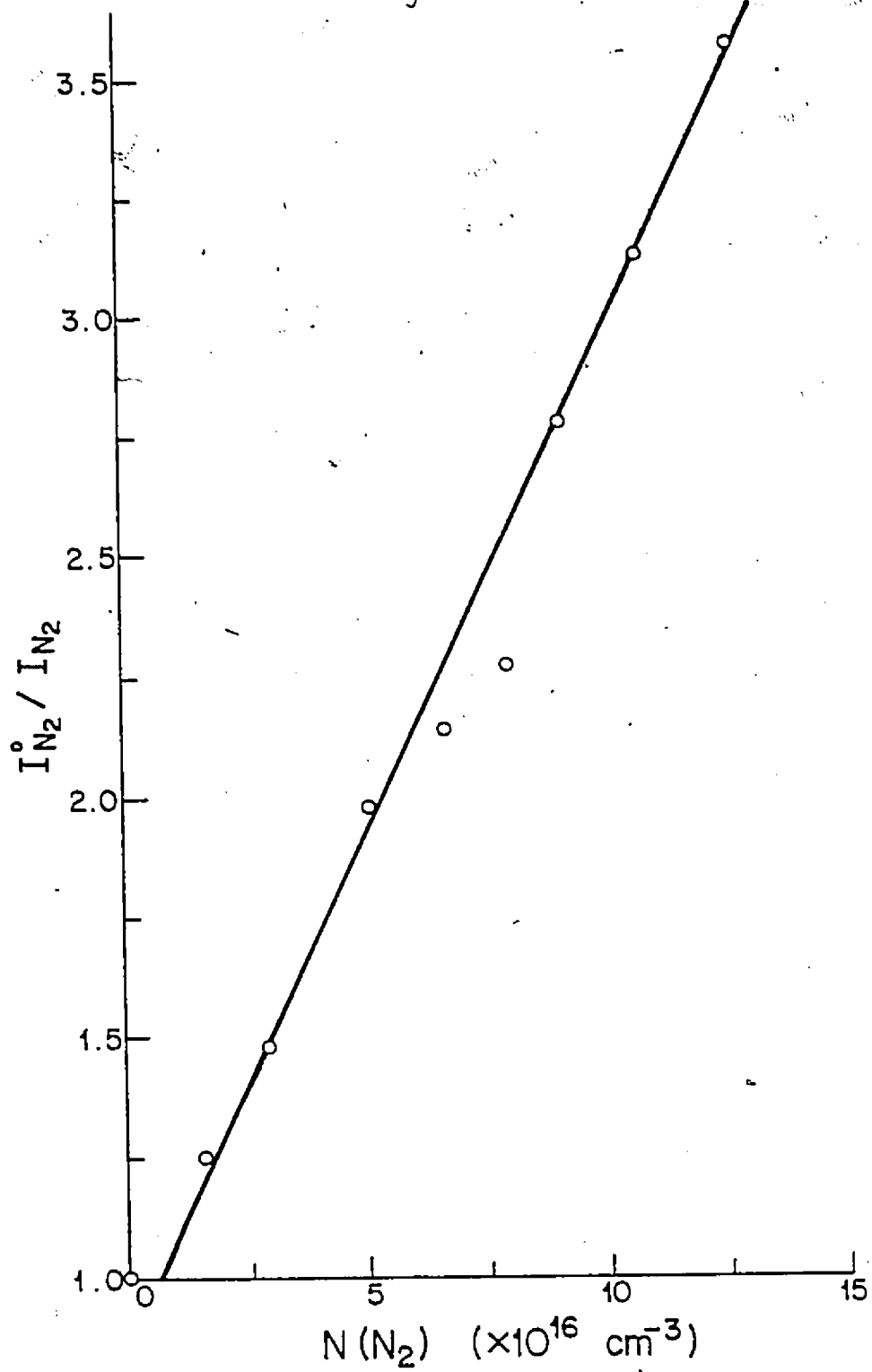


Fig. 21: Stern-Volmer plot for data obtained by monitoring the $\text{TI}^{3231\text{\AA}}$ radiation emitted from the Hg-Tl-N_2 mixture

TABLE 8: Cross sections for Quenching of thallium states by collisions with N_2 molecules.

Observed Sensitized Fluorescence		Description of the Quenched States			Cross Section
Transition	$\lambda(\text{\AA})$	Tl* State	$\tau^{(a)}$ (10^{-9}sec)	$\Delta E \text{ Hg}(6^3P_0) - \text{Tl}^*$ (eV)	$Q(\text{\AA}^2) \pm 20\%$
$6^2P_{3/2} - 8^2S_{1/2}$	3231	$8^2S_{1/2}$	22.4	+0.12	110
$6^2P_{3/2} - 6^2D_{5/2}$	3520	$6^2D_{5/2}$	5.1	-0.20	40
$6^2P_{3/2} - 7^2S_{1/2}$	5352	$7^2S_{1/2}$	11	-1.38	10.2
$6^2P_{1/2} - 7^2S_{1/2}$	3776	$7^2S_{1/2}$	11	-1.38	11.1

(a) Taken from theoretical values in the velocity formulation (listed in Appendix (10)) as determined from the orthogonalized HFS wave functions.

shown in figures (17), (18), (20) and (21) respectively. The quenching cross sections, Q , are listed in table (8) along with other pertinent data describing the quenching states.

Quenching of excited atoms by collisions with diatomic molecules is generally agreed to be accomplished by means of the transfer of the atomic excitation energy to specific vibrational states in the molecule⁷⁵. Pitre, Hammond and Krause⁷⁶ have shown that such is the case when the $\text{Hg}(6^3\text{P}_1)$ state is quenched to the $\text{Hg}(6^3\text{P}_0)$ metastable state by nitrogen molecules which, as a result, become vibrationally excited from the $v=0$ to the $v=1$ state. Such a description of the quenching mechanism has been supported by results of quenching experiments with CO , N_2 , H_2 and other diatomic species⁷⁵. It has been generally observed in experiments of this nature that an atomic state is most efficiently quenched by diatomic molecules when the depopulation of the atomic excited state proceeds by way of a radiationless transition to some lower state in such a manner that the corresponding atomic transition frequency lies in close resonance to that of some upward vibrational transition in the quenching molecule. Quantitative investigation of this resonance property has been performed previously in this laboratory⁸ in the quenching study of the mercury-sensitized fluorescence in sodium. It was shown explicitly that excited sodium states were quenched most efficiently by N_2 molecules when the energy defect, $\Delta E'$, between the energy of a downward radiationless transition in sodium and the energy of the closest upward vibrational transition in the ground electronic state of the N_2 molecule was small. This resonance phenomenon also appears to persist here in the case of thallium.

Quenching of the excited thallium states may be accomplished by radiationless transitions to the ground state, or to intermediate states if allowed by the dipole selection rules. In table (9) we list all possible transitions from each of the excited thallium states observed to be populated by the

TABLE 9: Possible radiationless transitions in thallium, resulting from the quenching action of N_2 molecules, with the most probable corresponding upward vibrational transitions in N_2 from the ground vibration state.

Quenching Transition in Tl	Vibrational Transition in N_2	$\Delta E'$ (eV)
$6^2P_{3/2} - 8^2S_{1/2}$	$v=0 \rightarrow v=15$	-0.10
$6^2P_{1/2} - 8^2S_{1/2}$	$v=0 \rightarrow v=19$	-0.05
$7^2P_{3/2} - 8^2S_{1/2}$	$v=0 \rightarrow v=2$	-0.12
$7^2P_{1/2} - 8^2S_{1/2}$	$v=0 \rightarrow v=2$	+0.01
$6^2P_{3/2} - 6^2D_{5/2}$	$v=0 \rightarrow v=13$	+0.04
$7^2P_{3/2} - 6^2D_{5/2}$	$v=0 \rightarrow v=1$	-0.17
$6^2P_{3/2} - 7^2S_{1/2}$	$v=0 \rightarrow v=8^1$	+0.11
$6^2P_{1/2} - 7^2S_{1/2}$	$v=0 \rightarrow v=12$	+0.06

transfer of the mercury excitation energy to the ground state thallium atoms. Also shown there is the most probable corresponding vibrational transition in the nitrogen molecule. We have chosen these possible transitions so as to obtain the closest resonance between the downward radiationless transition in the thallium atom and the accompanying upward vibrational transition in the N_2 molecule. The vibrational transition energies in N_2 are taken from Gilmore⁷⁸.

The resonance condition will be best satisfied when the atomic transition frequency closely matches a vibrational transition frequency in the N_2 molecule. In the case of the $Tl(7^2S_{1/2})$ state a quenching collision would only result in a radiationless transition to the $Tl(6^2P_{1/2,3/2})$ states. The $Tl(8^2S_{1/2})$ level would be most likely quenched to the $Tl(7^2P_{1/2,3/2})$ states while the $Tl(6^2D_{5/2})$ state would be quenched to the thallium ground state. With the exception of the $Tl(8^2S_{1/2})$ level, the quenching mechanism requires vibrational transitions in nitrogen of up to 13 vibrational quanta. The appearance of such large quantum jumps in quenching collisions are not uncommon. In the investigation of the quenching of the mercury-sensitized fluorescence of sodium by nitrogen molecules, jumps of up to 10 vibrational quanta were predicted⁸ and in similar studies on the quenching of the $Hg(6^3P_1)$ state by collisions with CO molecules⁷⁹ jumps of up to 11 vibrational quanta were observed.

The resonance behavior of the quenching cross sections is depicted graphically in figure (22). It should be stressed that the relatively small sampling of data obtained here for thallium does not, in itself, provide sufficient evidence for the resonance effect in quenching; however, combining our findings with those found in the case of sodium⁸ strongly suggests that the quenching mechanism, as defined by equation (30), proceeds most efficiently when the downward radiationless atomic transition lies in close resonance to an accompanying upward vibrational transition in the quenching gas molecule.

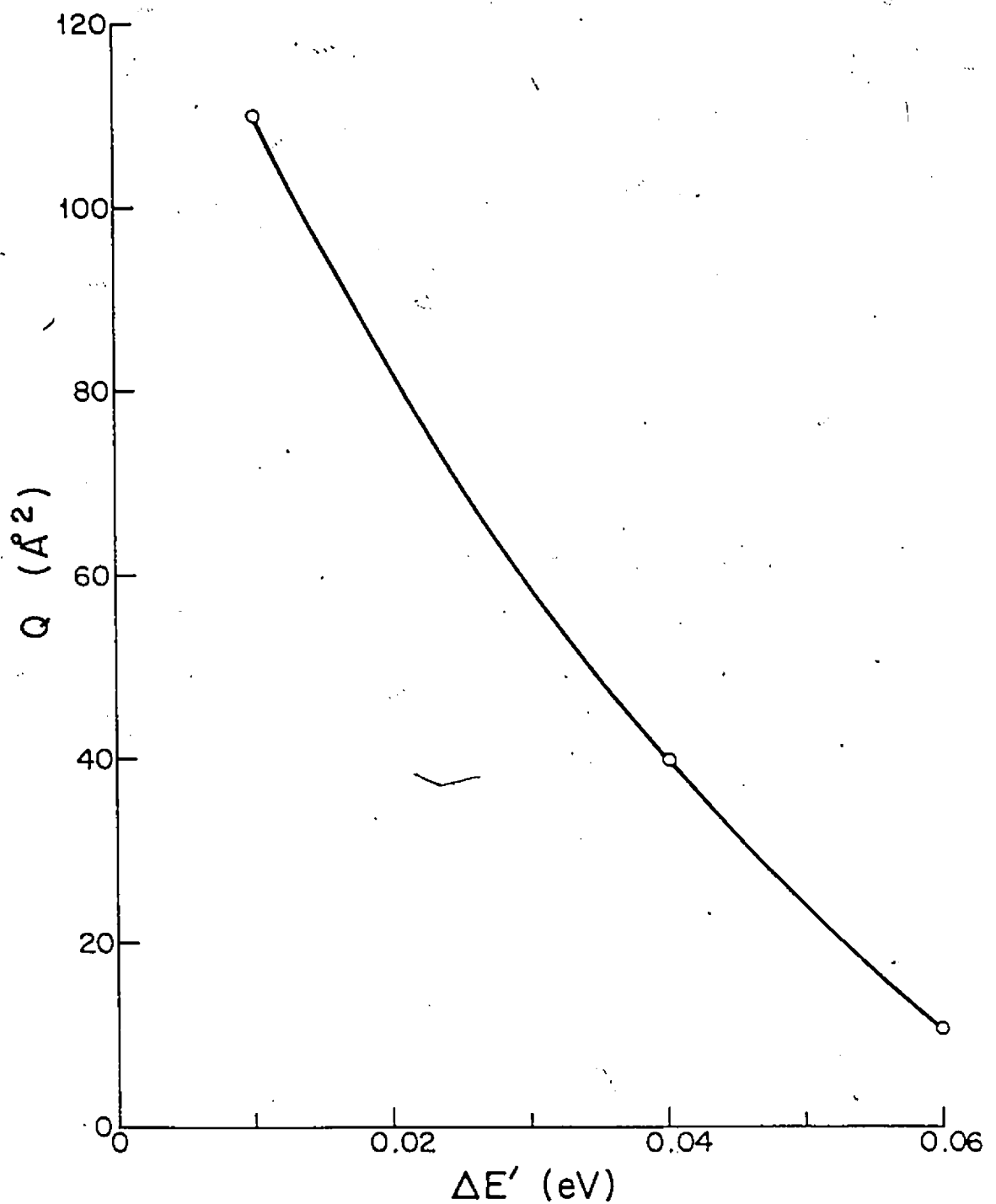


Fig. 22: Plot of quenching cross sections for the thallium states against $\Delta E'$, the energy defect between the energy of the downward radiationless transition in thallium and the accompanying upward vibrational transition in N_2

APPENDIX 1: Thallium Oscillator Strengths Calculated From Non-Orthogonal Wave Functions

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P - ms^2S_{1/2}$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR THALLIUM. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NORTON AND GALLAGHER AND PENKIN AND SHADANOVA (NG-PS) ARE ALSO SHOWN.

N	M	TRANSITION	fp_{ij}		f_{ij}		f_{ij}		$A_{ij} (\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	MIGDALEK	NG-PS	VELOCITY	LENGTH
6	7	1/2-1/2	-0.226D-02	0.323D-02	0.610D-01	0.102D 00	0.131D 00	0.136D 00	0.412D 03	0.852D 08
		3/2-1/2	-0.209D-02	0.315D-02	0.106D 00	0.243D 00	0.176D 00	0.154D 00	0.495D 03	0.113D 09
6	9	1/2-1/2	-0.594D-03	0.111D-02	0.117D-01	0.146D-01	0.166D-01	0.179D-01	0.117E 08	0.146D 08
		3/2-1/2	-0.761D-03	0.753D-03	0.857D-02	0.439D-02	0.159D-01	0.139D-01	0.109E 08	0.107D 08
6	10	1/2-1/2	-0.603D-03	0.532D-03	0.386D-02	0.300D-02	0.579D-02	0.630D-02	0.480D 07	0.373D 07
		3/2-1/2	-0.452D-03	0.285D-03	0.265D-02	0.105D-02	0.521D-02	0.490D-02	0.442D 07	0.175D 07
6	11	1/2-1/2	-0.411D-03	0.303D-03	0.171D-02	0.272D-03	0.285D-02	0.310D-02	0.234D 07	0.127D 07
		3/2-1/2	-0.307D-03	0.126D-03	0.115D-02	0.193D-03	0.258D-02	0.220D-02	0.216D 07	0.363D 06
7	8	1/2-1/2	-0.302D-03	0.194D-03	0.900D-03	0.471D-03	0.156D-02	0.220D-02	0.130D 07	0.534D 06
		3/2-1/2	-0.225D-03	0.605D-04	0.601D-03	0.435D-04	0.136D-02	0.220D-02	0.120D 07	0.868D 05
7	9	1/2-1/2	-0.151D-02	0.147D-02	0.259D 00	0.216D 00	0.269D 00	0.303D 00	0.363D 07	0.303D 07
		3/2-1/2	-0.158D-02	0.137D-02	0.318D 00	0.241D 00	0.303D 00	0.303D 00	0.545D 07	0.413D 07
7	10	1/2-1/2	-0.593D-03	0.583D-03	0.173D-01	0.176D-01	0.226D-01	0.226D-01	0.335D 06	0.952D 06
		3/2-1/2	-0.496D-03	0.463D-03	0.115D-01	0.123D-01	0.164D-01	0.164D-01	0.115D 07	0.195D 07
7	11	1/2-1/2	-0.225D-03	0.347D-03	0.435D-02	0.495D-02	0.769D-02	0.769D-02	0.360D 06	0.410D 06
		3/2-1/2	-0.258D-03	0.255D-03	0.301D-02	0.293D-02	0.546D-02	0.546D-02	0.412D 05	0.402D 06
8	9	1/2-1/2	-0.214D-03	0.237D-03	0.170D-02	0.210D-02	0.352D-02	0.352D-02	0.172D 06	0.212D 06
		3/2-1/2	-0.165D-03	0.167D-03	0.110D-02	0.113D-02	0.237D-02	0.237D-02	0.167D 06	0.192D 06
8	10	1/2-1/2	-0.127D-02	0.117D-02	0.612D 00	0.450D 00	0.740D-02	0.740D-02	0.889D 06	0.755D 06
		3/2-1/2	-0.120D-02	0.108D-02	0.459D 00	0.377D 00	0.590D-02	0.590D-02	0.124D 07	0.102D 07
8	11	1/2-1/2	-0.465D-03	0.451D-03	0.250D-01	0.237D-01	0.155D-01	0.145D-01	0.257D 06	0.244D 06
		3/2-1/2	-0.347D-03	0.335D-03	0.135D-01	0.145D-01	0.155D-01	0.145D-01	0.262D 06	0.244D 06
9	10	1/2-1/2	-0.274D-03	0.274D-03	0.674D-02	0.473D-02	0.574D-02	0.574D-02	0.116D 06	0.116D 06
		3/2-1/2	-0.193D-03	0.193D-03	0.317D-02	0.330D-02	0.537D-02	0.537D-02	0.106D 06	0.109D 06
9	11	1/2-1/2	-0.117D-02	0.053D-03	0.568D 00	0.459D 00	0.568D 00	0.568D 00	0.318D 06	0.257D 06
		3/2-1/2	-0.093D-03	0.095D-03	0.615D 00	0.300D 00	0.615D 00	0.300D 00	0.443D 06	0.360D 06
9	11	1/2-1/2	-0.414D-03	0.382D-03	0.378D-01	0.422D-01	0.227D-01	0.227D-01	0.109D 06	0.926D 05
		3/2-1/2	-0.316D-03	0.283D-03	0.227D-01	0.194D-01	0.227D-01	0.194D-01	0.109D 06	0.926D 05
10	11	1/2-1/2	-0.909D-03	0.794D-03	0.731D 00	0.557D 00	0.731D 00	0.557D 00	0.131D 06	0.996D 05
		3/2-1/2	-0.843D-03	0.735D-03	0.736D 00	0.501D 00	0.736D 00	0.501D 00	0.181D 06	0.137D 06

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2s^1/2 - mp2p^1j^1$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR THALLIUM. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MILGDALEK AND PENKIN AND SHARANOVA (NG-PS) ARE ALSO SHOWN. BY NORTON AND GALLAGHER AND PENKIN AND SHARANOVA (NG-PS) ARE ALSO SHOWN.

N	M	TRANSITION J-J'	\bar{W}_{IJ}^2		f_{IJ}		f_{IJ}		MIGDALEK	f_{IJ}	NG-PS	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH					
7	7	1/2-1/2 1/2-3/2	0.2410-02 0.2690-02	-0.2840-02 -0.3120-02	0.3460 00 0.7610 00	0.4810 00 0.1030 01	0.4630 00 0.9370 00	0.1360 08 0.1910 09	0.1890 08 0.2580 08				
7	9	1/2-1/2 1/2-3/2	0.1660-03 0.5240-03	-0.7960-04 -0.5110-03	0.8470-03 0.1650-01	0.3050-03 0.1570-01	0.1180-01 0.5470-01	0.1250 06 0.1280 07	0.4510 05 0.1220 07				
7	9	1/2-1/2 1/2-3/2	-0.3420-04 0.2150-03	0.1670-03 -0.1230-03	0.3000-04 0.2350-02	0.7120-03 0.7000-03	0.2550-02 0.1540-01	0.6400 04 0.2560 06	0.1520 06 0.8290 05				
7	10	1/2-1/2 1/2-3/2	-0.6870-04 0.1110-03	0.2040-03 -0.1620-05	0.1110-03 0.5770-03	0.9750-03 0.1220-06	0.9400-03 0.6740-02	0.2800 05 0.7370 05	0.2400 06 0.1560 02				
8	9	1/2-1/2 1/2-3/2	0.1770-02 0.1920-02	-0.1810-02 -0.1950-02	0.5440 00 0.1120 01	0.5710 00 0.1160 01		0.2500 07 0.3360 07	0.2620 07 0.3480 07				
8	9	1/2-1/2 1/2-3/2	0.2650-03 0.5290-03	-0.2180-03 -0.4860-03	0.5720-02 0.4410-01	0.3880-02 0.3720-01		0.1210 06 0.4980 06	0.8220 05 0.4190 06				
8	10	1/2-1/2 1/2-3/2	0.6960-04 0.2600-03	-0.2020-04 -0.2120-03	0.3090-03 0.6460-02	0.2590-04 0.5660-02		0.1060 05 0.1500 05	0.8940 03 0.1000 06				
9	9	1/2-1/2 1/2-3/2	0.1370-02 0.1480-02	-0.1350-02 -0.1440-02	0.7080 00 0.1440 01	0.6830 00 0.1360 01		0.6560 06 0.9730 06	0.6780 06 0.8830 06				
9	10	1/2-1/2 1/2-3/2	0.2820-03 0.5780-03	-0.2540-03 -0.4410-03	0.1320-01 0.7260-01	0.1060-01 0.6200-01		0.6750 05 0.2010 06	0.5450 05 0.1710 06				
10	10	1/2-1/2 1/2-3/2	0.1110-02 0.1190-02	-0.1060-02 -0.1130-02	0.8720 00 0.1740 01	0.9000 00 0.1560 01		0.2400 05 0.3230 06	0.2200 06 0.2910 06				

$nd^2_{0j} - mp^2_{jj}$

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR THALLIUM. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGOALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY MORTON AND GALLAGHER AND PENKIN AND SHAJANOVA (NG-PS) ARE ALSO SHOWN.

N	M	TRANSITION	mp_{ij}		f_{ij}		MIGDALEK	f_{ij}	NG-PS	$A_{ij} (\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH				VELOCITY	LENGTH
6	3	3/2-1/2	0.862D-03	-0.111D-02	0.649D-01	0.107D 00				0.238D 07	0.393D 07
		3/2-3/2	0.775D-03	-0.950D-03	0.588D-02	0.147D-01				0.208D 06	0.310D 06
		5/2-3/2	0.796D-03	-0.578D-03	0.628D-01	0.748D-01				0.153D 07	0.291D 07
6	7	3/2-1/2	0.338D-03	-0.452D-03	0.635D-02	0.113D-01				0.578D 06	0.103D 07
		3/2-3/2	0.343D-03	-0.426D-03	0.127D-02	0.176D-02				0.605D 05	0.934D 05
		5/2-3/2	0.348D-03	-0.433D-03	0.795D-02	0.123D-01				0.556D 06	0.862D 06
6	10	3/2-1/2	0.196D-03	-0.280D-03	0.162D-02	0.366D-02				0.208D 06	0.470D 06
		3/2-3/2	0.175D-03	-0.274D-03	0.366D-03	0.693D-03				0.241D 05	0.455D 05
		5/2-3/2	0.202D-03	-0.279D-03	0.228D-02	0.430D-02				0.221D 06	0.417D 06
7	9	3/2-1/2	0.729D-04	-0.764D-03	0.133D 00	0.113D 00				0.769D 06	0.845D 06
		3/2-3/2	0.589D-03	-0.607D-03	0.125D-01	0.132D-01				0.540D 05	0.574D 05
		5/2-3/2	0.627D-03	-0.559D-03	0.860D-01	0.749D-01				0.543D 06	0.600D 06
7	10	3/2-1/2	0.357D-03	-0.382D-03	0.148D-01	0.170D-01				0.305D 06	0.349D 06
		3/2-3/2	0.303D-03	-0.327D-03	0.218D-02	0.243D-02				0.236D 05	0.253D 05
		5/2-3/2	0.325D-03	-0.347D-03	0.145D-01	0.165D-01				0.232D 06	0.264D 06
8	10	3/2-1/2	0.592D-03	-0.545D-03	0.127D 00	0.128D 00				0.271D 06	0.230D 06
		3/2-3/2	0.449D-03	-0.405D-03	0.130D-01	0.110D-01				0.162D 05	0.138D 05
		5/2-3/2	0.434D-03	-0.415D-03	0.951D-01	0.342D-01				0.174D 06	0.154D 06

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_J - md^2D_{J1}$ CALCULATED USING THE RELATIVISTIC H-F-5 WAVE FUNCTIONS FOR THALLIUM. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY HORTON AND GALLAGHER AND PENKIN AND SHARANOVA (NG-PS) ARE ALSO SHOWN.

N	M	TRANSITION	f_{IJ}^{sp}		f_{IJ}		MIGDALEK	f_{IJ}^{NG-PS}	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH			VELOCITY	LENGTH
6	6	1/2-3/2	0.3890-02	-0.4260-02	0.3830 00	0.4200 00	0.4100 00	0.2900 00	0.1670 09	0.2000 09
		3/2-3/2	0.3950-02	-0.4590-02	0.5050-01	0.4500-01	0.5500-01	0.4000-01	0.2700 09	0.3510 09
		3/2-5/2	0.2830-02	-0.3447-02	0.4380 00	0.3700 00	0.4700 00	0.3500 00	0.1570 09	0.2050 09
6	7	1/2-3/2	0.2760-02	-0.2500-02	0.1670 00	0.1360 00	0.1900 00	0.7400-01	0.9210 08	0.8020 08
		3/2-3/2	0.2450-02	-0.2220-02	0.1610 00	0.1320 00	0.1800-01	0.3100-02	0.1260 08	0.1010 08
		3/2-5/2	0.2450-02	-0.2260-02	0.1450 00	0.1240 00	0.1190 00	0.8100-01	0.7560 07	0.6420 08
5	9	1/2-3/2	0.1750-02	-0.1310-02	0.6260-01	0.5200-01	0.4900-01	0.2900-01	0.4160 08	0.2350 08
		3/2-3/2	0.1430-02	-0.0980-02	0.5080-02	0.2420-02	0.5200-02	0.4000-02	0.4610 07	0.2190 07
		3/2-5/2	0.1480-02	-0.1070-02	0.4880-02	0.2560-02	0.4500-02	0.2800-02	0.2950 08	0.1550 08
6	9	1/2-3/2	0.1160-02	-0.0790-02	0.2690-01	0.1140-01	0.2600-01	0.1400-01	0.1910 09	0.8090 07
		3/2-3/2	0.9070-03	-0.4900-03	0.1970-02	0.0740-02	0.2600-02	0.2000-02	0.1920 07	0.5620 06
		3/2-5/2	0.7590-03	-0.5640-03	0.1980-01	0.0350-02	0.2300-01	0.1500-01	0.1290 08	0.4470 07
7	6	1/2-3/2	-0.1550-02	0.0750-02	0.1130 01	0.3580 00	0.1440 07	0.7660 05	0.1440 07	0.4580 06
		3/2-3/2	-0.1140-02	0.4390-03	0.1250 00	0.1840-01	0.1840 07	0.1120 05	0.7660 05	0.1120 05
		3/2-5/2	-0.1190-02	0.4610-03	0.1130 01	0.1930 00	0.5430 05	0.8800 05	0.5430 05	0.8800 05
7	7	1/2-3/2	0.1530-02	-0.1420-02	0.2710 00	0.2370 00	0.5590 07	0.4870 07	0.5590 07	0.4870 07
		3/2-3/2	0.1840-02	-0.1660-02	0.4530 00	0.3600-01	0.1420 07	0.1150 07	0.1420 07	0.1150 07
		3/2-5/2	0.1750-02	-0.1570-02	0.3650 00	0.2940 00	0.7700 07	0.6200 07	0.7700 07	0.6200 07
7	9	1/2-3/2	0.1160-02	-0.1120-02	0.1180 00	0.1090 00	0.4340 07	0.4030 07	0.4340 07	0.4030 07
		3/2-3/2	0.1220-02	-0.1140-02	0.1450-01	0.1240-01	0.8780 06	0.7490 06	0.8780 06	0.7490 06
		3/2-5/2	0.1210-02	-0.1120-02	0.1270 00	0.1080 00	0.5120 07	0.4380 07	0.5120 07	0.4380 07
7	9	1/2-3/2	0.8160-03	-0.7880-03	0.5100-01	0.4760-01	0.2430 07	0.2260 07	0.2430 07	0.2260 07
		3/2-3/2	0.9170-03	-0.7510-03	0.5580-02	0.4720-02	0.4460 06	0.3770 06	0.4460 06	0.3770 06
		3/2-5/2	0.8220-03	-0.7530-03	0.5080-01	0.4330-01	0.2710 07	0.2110 07	0.2710 07	0.2110 07
8	7	1/2-3/2	-0.7480-03	0.5570-03	0.7970 00	0.4420 00	0.1100 06	0.9320 03	0.1100 06	0.6110 05
		3/2-3/2	-0.4190-03	0.2380-03	0.5530-01	0.1910-01	0.2890 04	0.2890 04	0.2890 04	0.2890 04
		3/2-5/2	-0.5020-03	0.2740-03	0.6740 00	0.2010 00	0.2850 05	0.8480 04	0.2850 05	0.8480 04
8	8	1/2-3/2	0.1170-02	-0.1120-02	0.3770 00	0.3490 00	0.1370 07	0.1270 07	0.1370 07	0.1270 07
		3/2-3/2	0.1300-02	-0.1220-02	0.6030-01	0.3480-01	0.3460 06	0.3140 06	0.3460 06	0.3140 06
		3/2-5/2	0.1310-02	-0.1240-02	0.4330 00	0.4310 00	0.1670 07	0.1670 07	0.1670 07	0.1670 07
8	9	1/2-3/2	0.3200-03	-0.2700-03	0.1300 00	0.1210 00	0.7710 06	0.9010 06	0.7710 06	0.9010 06
		3/2-3/2	0.5740-03	-0.4710-03	0.1610-01	0.1450-01	0.2030 06	0.1840 06	0.2030 06	0.1840 06
		3/2-5/2	0.4570-03	-0.3130-03	0.1370 00	0.1250 00	0.1180 07	0.1060 07	0.1180 07	0.1060 07
9	8	1/2-3/2	-0.4050-03	0.4140-03	0.5140 00	0.3380 00	0.1460 05	0.1530 05	0.1460 05	0.1530 05
		3/2-3/2	-0.1140-03	0.1560-03	0.1080-01	0.2030-01	0.9730 02	0.1340 05	0.9730 02	0.1340 05
		3/2-5/2	-0.1940-03	0.1890-03	0.2370 00	0.2230 00	0.1790 04	0.1680 04	0.1790 04	0.1680 04
9	9	1/2-3/2	0.3060-03	-0.2920-03	0.4000 00	0.3310 00	0.3530 06	0.3520 06	0.3530 06	0.3520 06
		3/2-3/2	0.1070-03	-0.1110-02	0.6700-01	0.1200-01	0.1060 06	0.9640 05	0.1060 06	0.9640 05
		3/2-5/2	0.1010-02	-0.2460-03	0.5390 00	0.4770 00	0.5740 06	0.5080 06	0.5740 06	0.5080 06
10	9	1/2-3/2	-0.2740-03	0.3380-03	0.4450 00	0.5550 00	0.3760 04	0.5540 04	0.3760 04	0.5540 04
		3/2-3/2	-0.2060-04	0.1100-03	0.1450-02	0.2340-01	0.2920 01	0.4700 02	0.2920 01	0.4700 02
		3/2-5/2	-0.5770-04	0.11430-03	0.1180 00	0.2530 00	0.2330 03	0.5000 03	0.2330 03	0.5000 03

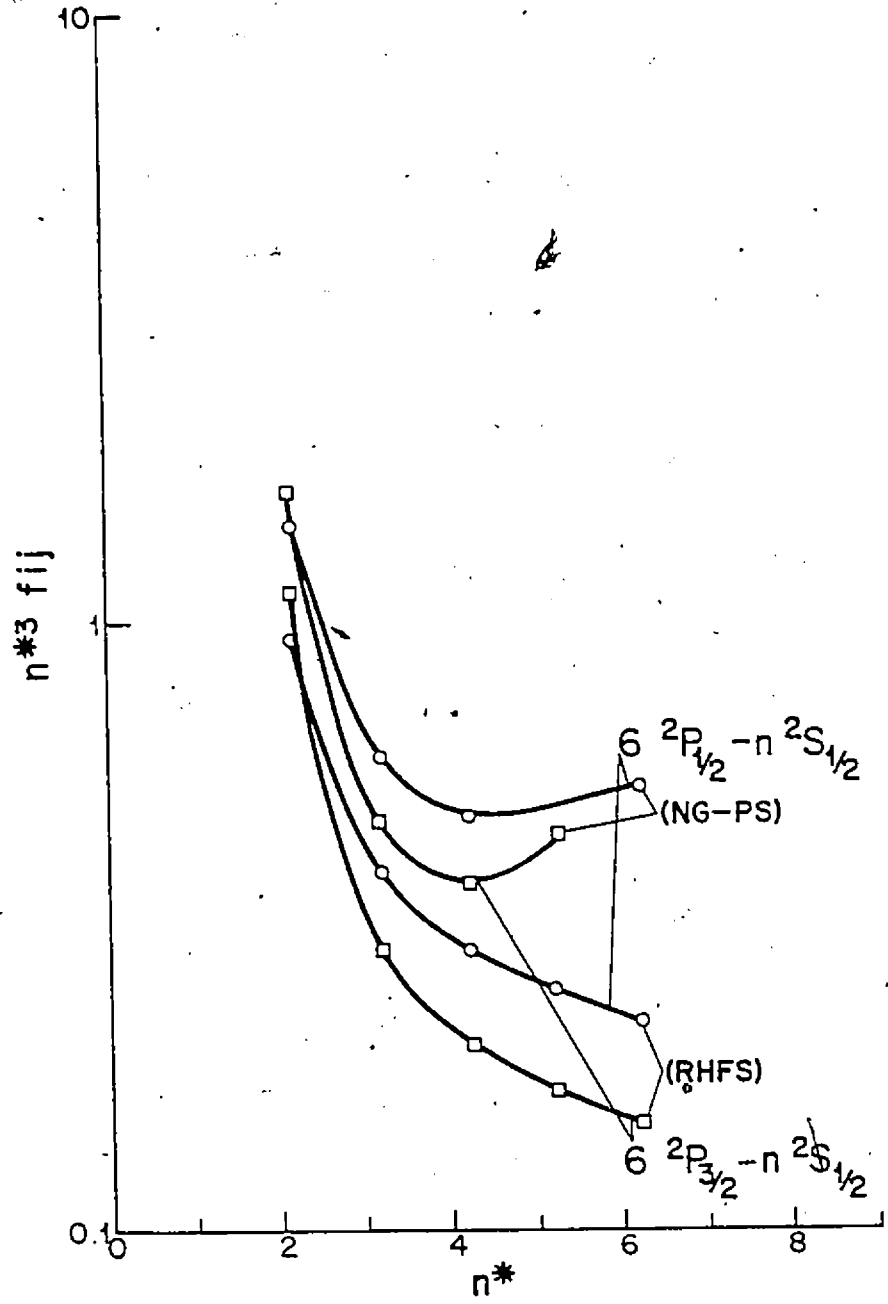
THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_j - mf^2F_j$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR THALLIUM. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NORTON AND GALLAGHER AND PENKIN AND SHABANJVA (NG-PS) ARE ALSO SHOWN.

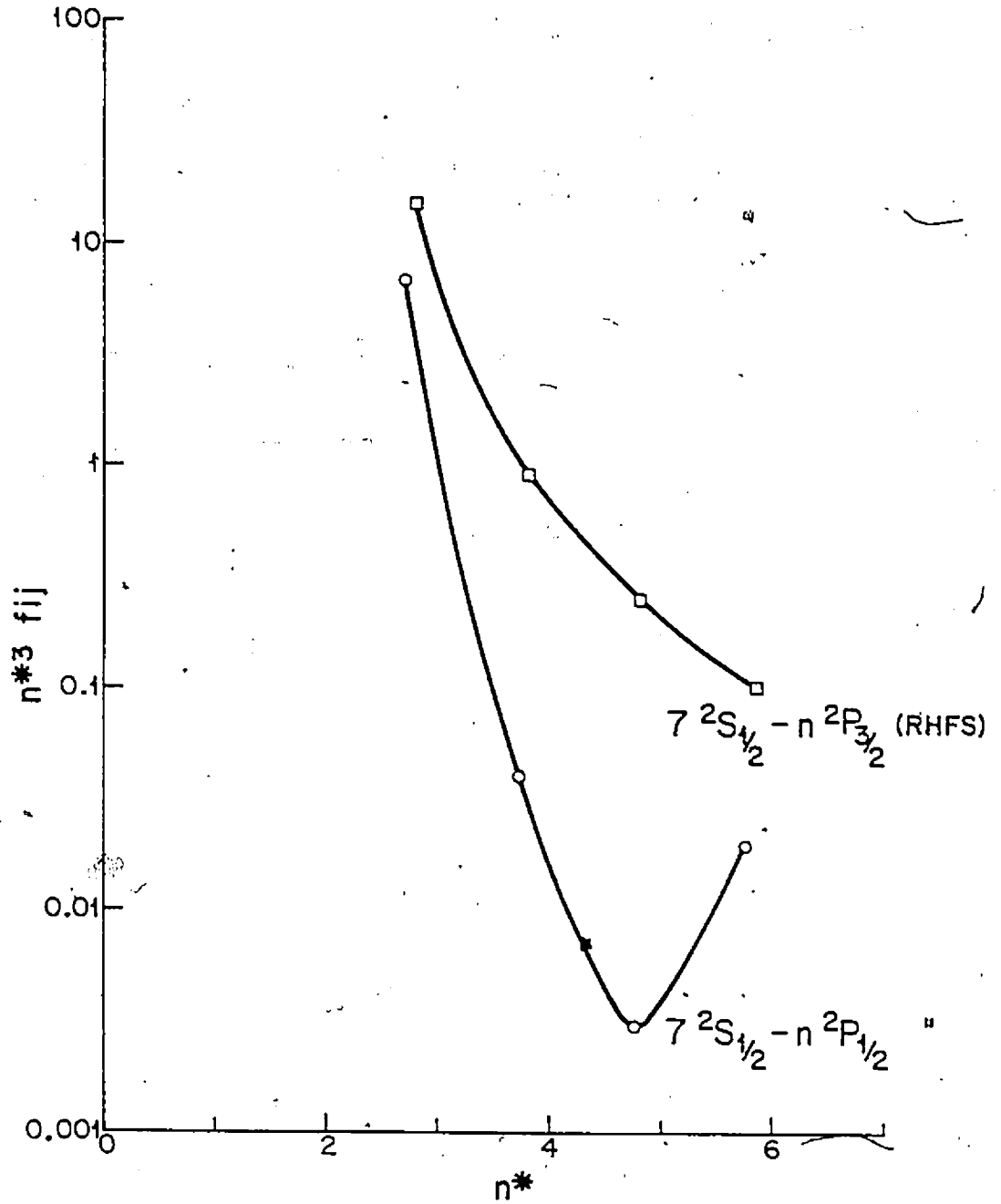
N	M	TRANSITION	f_{ij}^{MP}			f_{ij}			f_{ij}			$A_{ij} \text{ (sec}^{-1}\text{)}$		
			VELOCITY	LENGTH	NG-PS	VELOCITY	LENGTH	MIGDALEK	NG-PS	VELOCITY	LENGTH	VELOCITY	LENGTH	
6	5	3/2-5/2 5/2-5/2 5/2-7/2	-0.2570-02 -0.2570-02 -0.2570-02	0.2740-02 0.2750-02 0.2750-02		0.8750 00 0.4240-01 0.8470 00	0.7970 00 0.4350-01 0.9700 00			0.1500 08 0.1060 07 0.1590 08	0.1700 08 0.1210 07 0.1820 08			
6	6	3/2-5/2 5/2-5/2 5/2-7/2	-0.1320-02 -0.1300-02 -0.1300-02	0.1360-02 0.1350-02 0.1350-02		0.1640 00 0.7700-02 0.1540 00	0.1750 00 0.3240-02 0.1650 00			0.5530 07 0.3820 06 0.5730 07	0.5890 07 0.4090 06 0.6130 07			
7	5	3/2-5/2 5/2-5/2 5/2-7/2	0.5610-03 0.4960-03 0.4960-03	-0.2330-03 -0.2330-03 -0.2330-03		0.8450 00 0.7330-01 0.7360 00	0.1530 00 0.3330-02 0.1290 00			0.3540 05 0.1710 04 0.2560 05	0.6420 04 0.3090 03 0.4530 04			
7	6	3/2-5/2 5/2-5/2 5/2-7/2	-0.1300-02 -0.1350-02 -0.1350-02	0.1140-02 0.1210-02 0.1210-02		0.9560 00 0.2570-01 0.5180 00	0.3830 00 0.2060-01 0.4130 00			0.1740 07 0.1330 05 0.1990 07	0.1350 07 0.1060 06 0.1590 07			
8	6	3/2-5/2 5/2-5/2 5/2-7/2	0.7760-03 0.7540-03 0.7040-03	-0.2140-03 -0.1860-03 -0.1860-03		0.3290 01 0.1450 00 0.2780 01	0.2500 00 0.1040-01 0.2390 00			0.3330 05 0.1700 04 0.2550 05	0.2530 04 0.1190 03 0.1780 04			

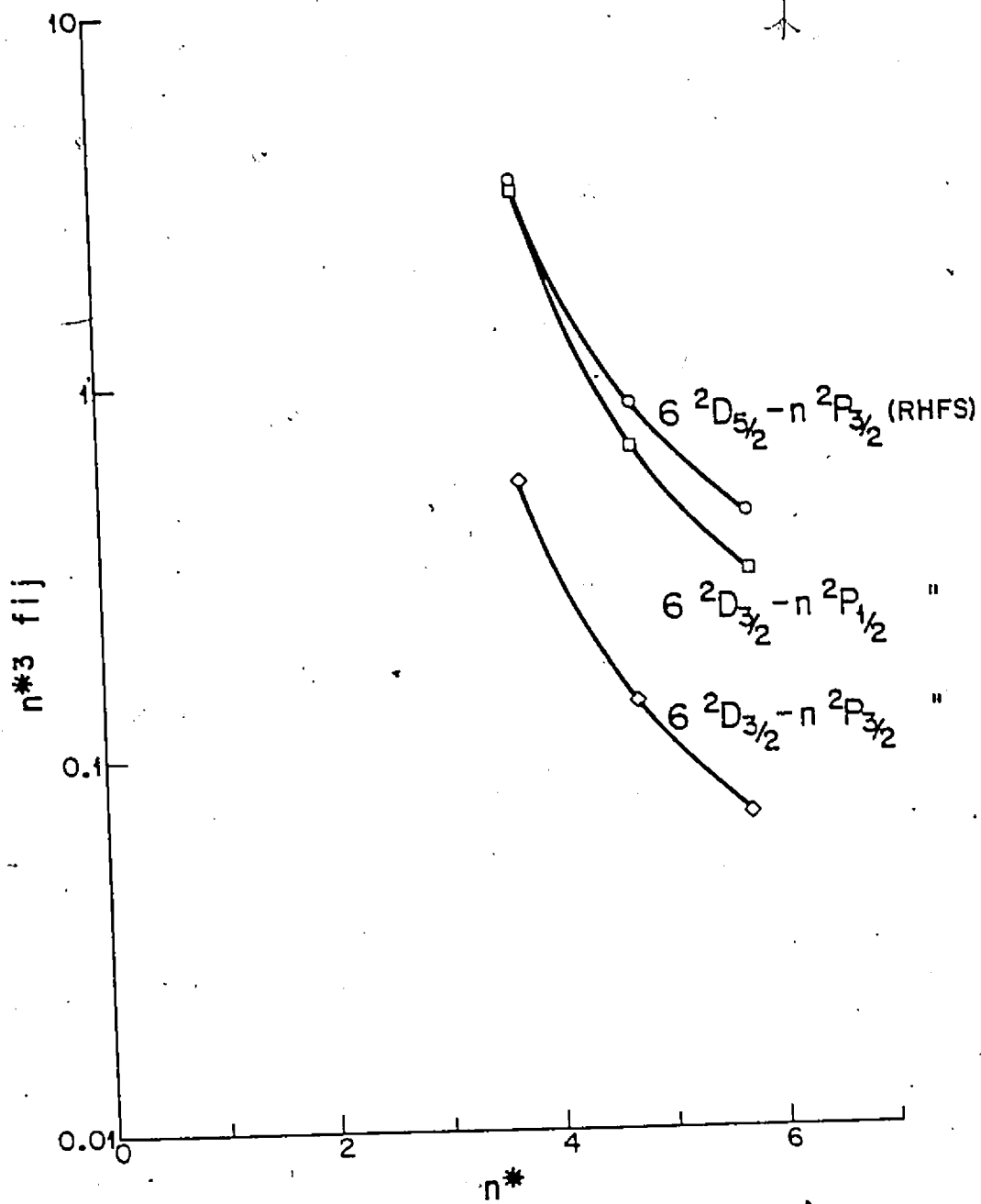
THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2 f_j - md^2 j_1$, CALCULATED USING THE RELATIVISTIC H-F-3 WAVE FUNCTIONS FOR THALLIUM. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NORTON AND GALLAGHER AND FENKIN AND SHABANOVA (NG-PS) ARE ALSO SHOWN.

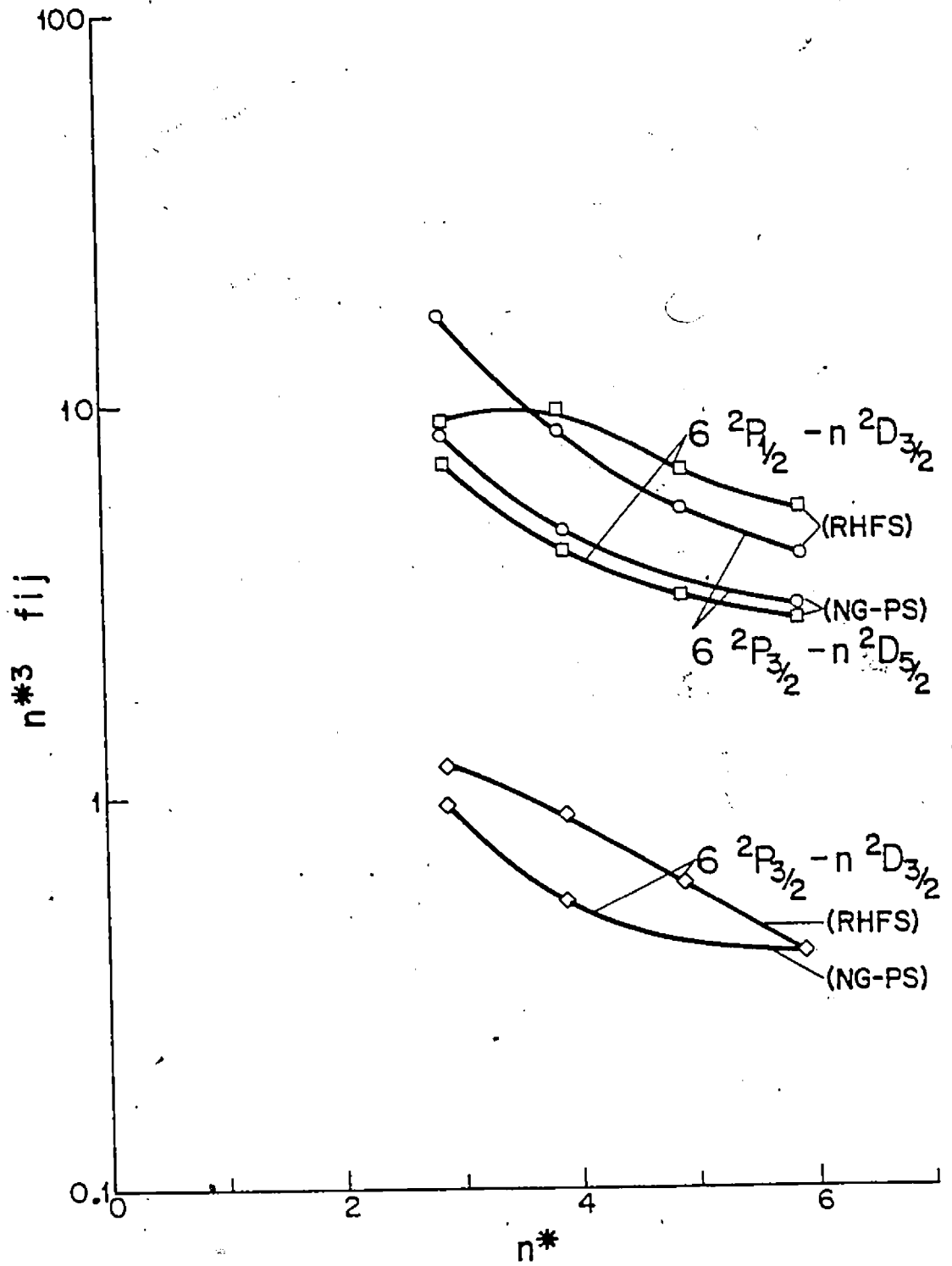
N	M	TRANSITION	P_{ij}^0		f_{ij}		MIGDALEK	f_{ij}	NG-PS	$A_{ij}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH				VELOCITY	LENGTH
5	B	5/2-3/2	-0.536D-03	0.649D-03	0.671D-01	0.375D-01				0.372D 06	0.540D 06
		5/2-5/2	-0.513D-03	0.6C5D-03	0.436D-02	0.505D-02				0.164D 05	0.228D 05
		7/2-5/2	-0.513D-03	0.6C6D-03	0.653D-01	0.707D-01				0.327D 06	0.456D 06
5	9	5/2-3/2	-0.253D-03	0.277D-03	0.101D-01	0.112D-01				0.144D 05	0.159D 06
		5/2-5/2	-0.255D-03	0.273D-03	0.695D-03	0.756D-03				0.667D 04	0.725D 04
6	9	7/2-5/2	-0.259D-03	0.270D-03	0.104D-01	0.113D-01				0.133D 06	0.145D 06
		5/2-3/2	-0.609D-03	0.780D-03	0.160D 00	0.263D 00				0.260D 06	0.427D 06
		5/2-5/2	-0.520D-03	0.718D-03	0.1C6D-01	0.166D-01				0.117D 05	0.193D 05
		7/2-5/2	-0.590D-03	0.749D-03	0.159D 00	0.247D 00				0.234D 06	0.367D 06

APPENDIX 2 : Plots of $n^3 f_{ij}$ versus n^* indicating the behavior of the relativistic Hartree-Fock-Slater oscillator strengths as a function of the effective principal quantum number for several of the Thallium series. The oscillator strengths are taken from Appendix (1) and correspond to those derived in the velocity formulation. The theoretical plots are denoted by (RHFS) and the corresponding plots formed from the combined experimental data of Norton and Gallagher, and Penkin and Shabanova are designated by (NG-PS).









APPENDIX 3 : Cesium Oscillator Strengths Calculated From Non-Orthogonal HFS Wave Functions

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_J - ns^2S_{1/2}$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

TRANSITION			f_{IJ}^0		f_{IJ}		f_{IJ}		$A_{IJ}(\text{sec}^{-1})$	
N	M	J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH
6	7	1/2-1/2	-0.162D-02	0.207D-02	0.163D 00	0.268D 00	0.171D 00		0.587D 07	0.967D 07
		3/2-1/2	-0.160D-02	0.205D-02	0.172D 00	0.284D 00	0.208D 00		0.106D 08	0.175D 08
6	8	1/2-1/2	-0.449D-03	0.609D-03	0.702D-02	0.129D-01	0.202D-01		0.808D 06	0.149D 07
		3/2-1/2	-0.405D-03	0.543D-03	0.596D-02	0.107D-01	0.204D-01		0.126D 07	0.226D 07
6	9	1/2-1/2	-0.231D-03	0.341D-03	0.155D-02	0.338D-02	0.702D-02	0.122D-01	0.256D 06	0.558D 06
		3/2-1/2	-0.202D-03	0.296D-03	0.123D-02	0.264D-02	0.687D-02		0.378D 06	0.810D 06
6	10	1/2-1/2	-0.148D-03	0.201D-03	0.589D-03	0.108D-02	0.326D-02	0.433D-02	0.115D 06	0.212D 06
		3/2-1/2	-0.128D-03	0.168D-03	0.453D-03	0.779D-03	0.299D-02	0.381D-02	0.166D 06	0.285D 06
6	11	1/2-1/2	-0.107D-03	0.135D-03	0.293D-03	0.463D-03	0.193D-02	0.243D-02	0.630D 05	0.995D 05
		3/2-1/2	-0.918D-04	0.109D-03	0.222D-03	0.313D-03	0.166D-02	0.261D-02	0.896D 05	0.126D 06
6	12	1/2-1/2	-0.826D-04	0.984D-04	0.169D-03	0.240D-03	0.122D-01	0.151D-02	0.365D 05	0.547D 05
		3/2-1/2	-0.705D-04	0.778D-04	0.127D-03	0.155D-03	0.117D-02	0.140D-02	0.544D 05	0.664D 05
6	13	1/2-1/2	-0.664D-04	0.761D-04	0.108D-03	0.141D-03	0.820D-03	0.121D-02	0.253D 05	0.332D 05
		3/2-1/2	-0.566D-04	0.592D-04	0.804D-04	0.879D-04	0.780D-03	0.104D-02	0.357D 05	0.390D 05
6	14	1/2-1/2	-0.552D-04	0.617D-04	0.731D-04	0.916D-04	0.590D-03	0.700D-03	0.177D 05	0.222D 05
		3/2-1/2	-0.469D-04	0.475D-04	0.544D-04	0.557D-04	0.560D-03	0.690D-03	0.249D 05	0.255D 05
6	15	1/2-1/2	-0.469D-04	0.518D-04	0.523D-04	0.637D-04		0.380D-03	0.130D 05	0.158D 05
		3/2-1/2	-0.398D-04	0.398D-04	0.388D-04	0.382D-04			0.131D 05	0.176D 05
7	8	1/2-1/2	-0.137D-02	0.138D-02	0.335D 00	0.340D 00	0.297D 00		0.145D 07	0.148D 07
		3/2-1/2	-0.133D-02	0.139D-02	0.343D 00	0.371D 00	0.333D 00		0.257D 07	0.278D 07
7	9	1/2-1/2	-0.499D-03	0.501D-03	0.222D-01	0.223D-01	0.305D-01		0.392D 06	0.394D 06
		3/2-1/2	-0.444D-03	0.451D-03	0.182D-01	0.188D-01	0.256D-01		0.598D 06	0.616D 06
7	10	1/2-1/2	-0.289D-03	0.294D-03	0.585D-02	0.606D-02	0.964D-02		0.167D 06	0.172D 06
		3/2-1/2	-0.250D-03	0.255D-03	0.449D-02	0.468D-02	0.842D-02		0.242D 06	0.252D 06
7	11	1/2-1/2	-0.198D-03	0.202D-03	0.243D-02	0.255D-02	0.501D-02		0.879D 05	0.922D 05
		3/2-1/2	-0.168D-03	0.172D-03	0.181D-02	0.188D-02	0.455D-02		0.124D 06	0.129D 06
7	12	1/2-1/2	-0.148D-03	0.152D-03	0.127D-02	0.134D-02	0.287D-02		0.527D 05	0.557D 05
		3/2-1/2	-0.125D-03	0.127D-03	0.925D-03	0.959D-03	0.259D-02		0.735D 05	0.762D 05
7	13	1/2-1/2	-0.117D-03	0.119D-03	0.761D-03	0.795D-03	0.183D-02		0.341D 05	0.357D 05
		3/2-1/2	-0.981D-04	0.990D-04	0.550D-03	0.560D-03	0.164D-02		0.471D 05	0.480D 05
7	14	1/2-1/2	-0.558D-04	0.983D-04	0.495D-03	0.522D-03	0.126D-02		0.238D 05	0.250D 05
		3/2-1/2	-0.802D-04	0.812D-04	0.355D-03	0.363D-03	0.113D-02		0.326D 05	0.334D 05
7	15	1/2-1/2	-0.808D-04	0.832D-04	0.344D-03	0.365D-03			0.173D 05	0.184D 05
		3/2-1/2	-0.676D-04	0.684D-04	0.246D-03	0.252D-03			0.237D 05	0.243D 05
8	9	1/2-1/2	-0.114D-02	0.104D-02	0.494D 00	0.415D 00			0.476D 06	0.400D 06
		3/2-1/2	-0.112D-02	0.105D-02	0.511D 00	0.455D 00			0.853D 06	0.759D 06
8	10	1/2-1/2	-0.449D-03	0.421D-03	0.356D-01	0.313D-01			0.159D 06	0.140D 06
		3/2-1/2	-0.403D-03	0.382D-03	0.297D-01	0.267D-01			0.249D 06	0.224D 06
8	11	1/2-1/2	-0.277D-03	0.263D-03	0.103D-01	0.927D-02			0.803D 05	0.724D 05
		3/2-1/2	-0.244D-03	0.233D-03	0.815D-02	0.742D-02			0.121D 06	0.110D 06

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_J - ms^2S_{1/2}$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M^0_{1J}		f_{1J}		f_{1J}		AGNEW	$A_{1J} (\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	LENGTH		VELOCITY	LENGTH
8	12	1/2-1/2 3/2-1/2	-0.198D-03 -0.172D-03	0.190D-03 0.166D-03	0.454D-02 0.351D-02	0.418D-02 0.324D-02				0.474D 05 0.702D 05	0.436D 05 0.649D 05
8	13	1/2-1/2 3/2-1/2	-0.152D-03 -0.132D-03	0.145D-03 0.125D-03	0.250D-02 0.191D-02	0.226D-02 0.172D-02				0.302D 05 0.443D 05	0.273D 05 0.400D 05
8	14	1/2-1/2 3/2-1/2	-0.123D-03 -0.106D-03	0.118D-03 0.102D-03	0.153D-02 0.116D-02	0.141D-02 0.106D-02				0.210D 05 0.306D 05	0.193D 05 0.281D 05
6	15	1/2-1/2 3/2-1/2	-0.103D-03 -0.083D-04	0.092D-04 0.051D-04	0.102D-02 0.767D-03	0.0951D-03 0.712D-03				0.153D 05 0.222D 05	0.143D 05 0.206D 05
9	10	1/2-1/2 3/2-1/2	-0.055D-03 -0.042D-03	0.074D-03 0.080D-03	0.636D 00 0.657D 00	0.528D 00 0.574D 00				0.186D 06 0.355D 06	0.155D 06 0.292D 06
9	11	1/2-1/2 3/2-1/2	-0.039D-03 -0.0354D-03	0.0363D-03 0.0331D-03	0.470D-01 0.396D-01	0.404D-01 0.347D-01				0.699D 05 0.111D 06	0.601D 05 0.969D 05
9	12	1/2-1/2 3/2-1/2	-0.025D-03 -0.0221D-03	0.023D-03 0.0208D-03	0.141D-01 0.113D-01	0.123D-01 0.097D-02				0.387D 05 0.594D 05	0.337D 05 0.524D 05
9	13	1/2-1/2 3/2-1/2	-0.0183D-03 -0.0160D-03	0.0168D-03 0.0148D-03	0.656D-02 0.517D-02	0.555D-02 0.439D-02				0.237D 05 0.359D 05	0.200D 05 0.305D 05
9	14	1/2-1/2 3/2-1/2	-0.0143D-03 -0.0125D-03	0.0134D-03 0.0117D-03	0.360D-02 0.280D-02	0.314D-02 0.245D-02				0.163D 05 0.245D 05	0.142D 05 0.215D 05
9	15	1/2-1/2 3/2-1/2	-0.0117D-03 -0.0102D-03	0.0111D-03 0.0064D-04	0.224D-02 0.173D-02	0.199D-02 0.154D-02				0.118D 05 0.177D 05	0.105D 05 0.157D 05
10	11	1/2-1/2 3/2-1/2	-0.0827D-03 -0.0812D-03	0.0753D-03 0.0756D-03	0.778D 00 0.803D 00	0.644D 00 0.095D 00				0.842D 05 0.131D 06	0.698D 05 0.131D 06
10	12	1/2-1/2 3/2-1/2	-0.0344D-03 -0.0312D-03	0.0319D-03 0.0291D-03	0.577D-01 0.487D-01	0.496D-01 0.425D-01				0.339D 05 0.541D 05	0.291D 05 0.472D 05
10	13	1/2-1/2 3/2-1/2	-0.0224D-03 -0.0199D-03	0.0196D-03 0.0175D-03	0.186D-01 0.150D-01	0.142D-01 0.116D-01				0.190D 05 0.253D 05	0.146D 05 0.227D 05
10	14	1/2-1/2 3/2-1/2	-0.0166D-03 -0.0147D-03	0.0152D-03 0.0135D-03	0.829D-02 0.659D-02	0.697D-02 0.556D-02				0.127D 05 0.196D 05	0.107D 05 0.165D 05
10	15	1/2-1/2 3/2-1/2	-0.0132D-03 -0.0116D-03	0.0123D-03 0.0108D-03	0.459D-02 0.362D-02	0.402D-02 0.316D-02				0.911D 04 0.139D 05	0.797D 04 0.121D 05
11	12	1/2-1/2 3/2-1/2	-0.0727D-03 -0.0713D-03	0.0660D-03 0.0661D-03	0.922D 00 0.951D 00	0.761D 00 0.817D 00				0.423D 05 0.760D 05	0.349D 05 0.654D 05
11	13	1/2-1/2 3/2-1/2	-0.0205D-03 -0.0277D-03	0.0254D-03 0.0231D-03	0.760D-01 0.646D-01	0.527D-01 0.448D-01				0.160D 05 0.255D 05	0.111D 05 0.177D 05
11	14	1/2-1/2 3/2-1/2	-0.0202D-03 -0.0179D-03	0.0173D-03 0.0157D-03	0.222D-01 0.179D-01	0.164D-01 0.136D-01				0.105D 05 0.162D 05	0.774D 04 0.124D 05
11	15	1/2-1/2 3/2-1/2	-0.0151D-03 -0.0134D-03	0.0138D-03 0.0122D-03	0.993D-02 0.794D-02	0.935D-02 0.658D-02				0.727D 04 0.113D 05	0.612D 04 0.933D 04
12	13	1/2-1/2 3/2-1/2	-0.0648D-03 -0.0355D-03	0.0364D-03 0.0349D-03	0.172D 01 0.185D 01	0.543D 00 0.558D 00				0.144D 05 0.247D 05	0.454D 04 0.746D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_j - ms^2s_{1/2}$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION J-J'	f_{IJ}^0		f_{IJ}		f_{IJ}		f_{IJ}		$\lambda_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH
12	14	1/2-1/2 3/2-1/2	-0.274D-03 -0.249D-03	0.230D-03 0.209D-03	0.874D-01 0.745D-01	0.615D-01 0.525D-01					0.899D 04 0.144D 05	0.632D 04 0.102D 05
12	15	1/2-1/2 3/2-1/2	-0.184D-03 -0.163D-03	0.153D-03 0.137D-03	0.259D-01 0.207D-01	0.179D-01 0.146D-01					0.620D 04 0.950D 04	0.428D 04 0.674D 04
13	14	1/2-1/2 3/2-1/2	-0.100D-03 -0.359D-03	0.110D-03 0.109D-03	0.582D-01 0.831D 00	0.695D-01 0.764D-01					0.245D 03 0.559D 04	0.293D 03 0.514D 03
13	15	1/2-1/2 3/2-1/2	-0.684D-04 -0.225D-03	0.344D-03 0.357D-03	0.749D-02 0.836D-01	0.190D 00 0.210D 00					0.408D 03 0.858D 04	0.103D 05 0.215D 05
14	15	1/2-1/2 3/2-1/2	-0.120D-03 -0.485D-03	0.225D-04 0.327D-04	0.112D 00 0.206D 01	0.395D-02 0.935D-02					0.257D 03 0.751D 04	0.907D 01 0.342D 02

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2 \text{ } ^1S_{1/2} - mp^2 \text{ } ^1P_1$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M_{IJ}^0		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	VELOCITY		LENGTH	
6	6	1/2-1/2 1/2-3/2	0.289D-02 0.302D-02	-0.317D-02 -0.331D-02	0.342D 00 0.712D 00	0.411D 00 0.854D 00	0.394D 00 0.814D 00		0.285D 08 0.327D 08	0.342D 08 0.392D 08	
6	7	1/2-1/2 1/2-3/2	0.799D-04 0.210D-03	0.277D-03 0.488D-06	0.134D-03 0.401D-02	0.162D-02 0.993D-08	0.284D-02 0.174D-01	0.320D-02 0.133D-01	0.424D-05 0.644D 06	0.511D 06 0.160D 01	
6	8	1/2-1/2 1/2-3/2	-0.140D-03 0.778D-05	0.5C7D-03 0.337D-03	0.351D-03 0.215D-05	0.459D-02 0.404D-02	0.317D-03 0.349D-02	0.385D-03 0.294D-02	0.155D 06 0.477D 03	0.202D 07 0.897D 06	
6	9	1/2-1/2 1/2-3/2	-0.151D-03 -0.502D-04	0.456D-03 0.346D-03	0.376D-03 0.833D-04	0.345D-02 0.397D-02	0.725D-04 0.125D-02	0.830D-04 0.910D-03	0.191D 06 0.213D 05	0.176D 07 0.101D 07	
6	10	1/2-1/2 1/2-3/2	-0.132D-03 -0.594D-04	0.382D-03 0.305D-03	0.279D-03 0.112D-03	0.233D-02 0.297D-02	0.289D-04 0.620D-03	0.190D-04 0.417D-03	0.153D 06 0.310D 05	0.128D 07 0.818D 06	
6	11	1/2-1/2 1/2-3/2	-0.113D-03 -0.567D-04	0.319D-03 0.262D-03	0.187D-03 0.100D-03	0.159D-02 0.213D-02	0.124D-04 0.356D-03	0.800D-05 0.223D-03	0.114D 06 0.289D 05	0.915D 06 0.616D 06	
6	12	1/2-1/2 1/2-3/2	-0.558D-04 -0.513D-04	0.269D-03 0.225D-03	0.141D-03 0.808D-04	0.111D-02 0.155D-02	0.620D-05 0.208D-03	0.430D-05 0.126D-03	0.838D 05 0.240D 05	0.661D 06 0.460D 06	
6	13	1/2-1/2 1/2-3/2	-0.767D-04 -0.404D-04	0.208D-03 0.173D-03	0.892D-04 0.495D-04	0.654D-03 0.913D-03	0.516D-02 0.256D-02	0.240D-05 0.790D-04	0.541D 05 0.150D 05	0.397D 06 0.277D 06	
6	14	1/2-1/2 1/2-3/2	-0.663D-04 -0.357D-04	0.178D-03 0.150D-03	0.662D-04 0.383D-04	0.479D-03 0.678D-03	0.479D-03 0.678D-03	0.160D-05 0.580D-04	0.408D 05 0.118D 05	0.295D 06 0.209D 06	
6	15	1/2-1/2 1/2-3/2	-0.580D-04 -0.318D-04	0.156D-03 0.132D-03	0.505D-04 0.302D-04	0.362D-03 0.519D-03	0.556D 00 0.112D 01		0.314D 05 0.942D 04	0.226D 06 0.162D 06	
7	7	1/2-1/2 1/2-3/2	0.203D-02 0.212D-02	-0.181D-02 -0.186D-02	0.583D 00 0.121D 01	0.462D 00 0.929D 00			0.406D 07 0.470D 07	0.362D 07 0.360D 07	
7	8	1/2-1/2 1/2-3/2	0.255D-03 0.413D-03	-0.175D-03 -0.321D-03	0.416D-02 0.215D-01	0.195D-02 0.130D-01	0.516D-02 0.256D-02		0.143D 06 0.378D 06	0.668D 05 0.229D 06	
7	9	1/2-1/2 1/2-3/2	0.359D-04 0.143D-03	0.278D-04 -0.731D-04	0.648D-04 0.206D-02	0.388D-04 0.335D-03	0.620D-03 0.502D-02		0.358D 04 0.575D 05	0.215D 04 0.149D 05	
7	10	1/2-1/2 1/2-3/2	-0.175D-04 0.599D-04	0.692D-04 -0.320D-05	0.138D-04 0.322D-03	0.215D-03 0.920D-06	0.170D-03 0.187D-02		0.956D 03 0.112D 05	0.149D 05 0.320D 02	
7	11	1/2-1/2 1/2-3/2	-0.321D-04 0.268D-04	0.749D-04 0.200D-04	0.435D-04 0.602D-04	0.236D-03 0.337D-04	0.620D-04 0.911D-03		0.343D 04 0.238D 04	0.186D 05 0.133D 04	
7	12	1/2-1/2 1/2-3/2	-0.350D-04 0.117D-04	0.710D-04 0.276D-04	0.495D-04 0.111D-04	0.204D-03 0.616D-04	0.270D-04 0.486D-03		0.423D 04 0.474D 03	0.174D 05 0.264D 04	
7	13	1/2-1/2 1/2-3/2	-0.369D-04 -0.114D-05	0.934D-04 0.592D-04	0.526D-04 0.102D-06	0.344D-03 0.276D-03	0.620D-04 0.911D-03		0.538D 04 0.463D 01	0.310D 05 0.125D 05	
7	14	1/2-1/2 1/2-3/2	-0.358D-04 -0.419D-05	0.839D-04 0.554D-04	0.496D-04 0.135D-05	0.272D-03 0.237D-03	0.496D-04 0.135D-05		0.65D 04 0.635D 02	0.255D 05 0.111D 05	
7	15	1/2-1/2 1/2-3/2	-0.327D-04 -0.558D-05	0.752D-04 0.509D-04	0.406D-04 0.237D-05	0.215D-03 0.197D-03	0.406D-04 0.237D-05		0.352D 04 0.114D 03	0.207D 05 0.952D 04	
8	8	1/2-1/2 1/2-3/2	0.147D-02 0.155D-02	-0.134D-02 -0.137D-02	0.715D 00 0.149D 01	0.588D 00 0.117D 01			0.924D 06 0.108D 07	0.760D 06 0.849D 06	

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n_2 S_{1/2} - n_1 p^2 P_{j_1}$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION J-J'	M_{IJ}^0		f_{IJ}		f_{IJ}		f_{IJ}		$A_{IJ} (\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH		
8	9	1/2-1/2 1/2-3/2	0.254D-03 0.365D-03	-0.220D-03 -0.321D-03	0.890D-02 0.362D-01	0.659D-02 0.280D-01			0.654D 05 0.137D 06		0.492D 05 0.106D 06	
8	10	1/2-1/2 1/2-3/2	0.858D-04 0.164D-03	-0.656D-04 -0.137D-03	0.764D-03 0.553D-02	0.447D-03 0.388D-02			0.951D 04 0.363D 05		0.580D 04 0.254D 05	
8	11	1/2-1/2 1/2-3/2	0.337D-04 0.916D-04	-0.191D-04 -0.724D-04	0.102D-03 0.150D-02	0.328D-04 0.941D-03			0.176D 04 0.131D 05		0.566D 03 0.818D 04	
8	12	1/2-1/2 1/2-3/2	0.128D-04 0.581D-04	-0.119D-05 -0.430D-04	0.136D-04 0.557D-03	0.118D-06 0.305D-03			0.277D 03 0.571D 04		0.241D 01 0.313D 04	
8	13	1/2-1/2 1/2-3/2	-0.106D-04 0.261D-04	0.320D-04 -0.244D-05	0.871D-05 0.107D-03	0.800D-04 0.929D-06			0.199D 03 0.122D 04		0.183D 04 0.106D 02	
8	14	1/2-1/2 1/2-3/2	-0.133D-04 0.172D-04	0.323D-04 0.366D-05	0.133D-04 0.445D-04	0.784D-04 0.202D-05			0.327D 03 0.549D 03		0.193D 04 0.249D 02	
8	15	1/2-1/2 1/2-3/2	-0.140D-04 0.120D-04	0.307D-04 0.644D-05	0.143D-04 0.210D-04	0.693D-04 0.607D-05			0.372D 03 0.273D 03		0.180D 04 0.791D 02	
9	9	1/2-1/2 1/2-3/2	0.115D-02 0.121D-02	-0.108D-02 -0.111D-02	0.831D 00 0.173D 01	0.733D 00 0.146D 01			0.293D 06 0.343D 06		0.258D 06 0.289D 06	
9	10	1/2-1/2 1/2-3/2	0.226D-03 0.311D-03	-0.206D-03 -0.284D-03	0.129D-01 0.481D-01	0.107D-01 0.399D-01			0.284D 05 0.545D 05		0.236D 05 0.452D 05	
9	11	1/2-1/2 1/2-3/2	0.945D-04 0.155D-03	-0.834D-04 -0.139D-03	0.164D-02 0.878D-02	0.128D-02 0.707D-02			0.680D 04 0.185D 05		0.529D 04 0.149D 05	
9	12	1/2-1/2 1/2-3/2	0.494D-04 0.953D-04	-0.427D-04 -0.852D-04	0.380D-03 0.282D-02	0.203D-03 0.225D-02			0.220D 04 0.820D 04		0.164D 04 0.655D 04	
9	13	1/2-1/2 1/2-3/2	0.108D-04 0.478D-04	0.706D-06 -0.344D-04	0.163D-04 0.640D-03	0.701D-07 0.333D-03			0.115D 03 0.227D 04		0.495D 00 0.118D 04	
9	14	1/2-1/2 1/2-3/2	0.273D-05 0.331D-04	-0.736D-05 -0.215D-04	0.981D-06 0.287D-03	0.713D-05 0.121D-03			0.794D 01 0.117D 04		0.577D 02 0.492D 03	
9	15	1/2-1/2 1/2-3/2	-0.118D-05 0.244D-04	0.101D-04 -0.143D-04	0.175D-06 0.149D-03	0.128D-04 0.509D-04			0.155D 01 0.667D 03		0.114D 03 0.227D 03	
10	10	1/2-1/2 1/2-3/2	0.558D-03 0.100D-02	-0.893D-03 -0.915D-03	0.984D 00 0.203D 01	0.856D 00 0.169D 01			0.120D 06 0.139D 06		0.104D 06 0.116D 06	
10	11	1/2-1/2 1/2-3/2	0.223D-03 0.289D-03	-0.207D-03 -0.267D-03	0.206D-01 0.684D-01	0.178D-01 0.583D-01			0.167D 05 0.287D 05		0.144D 05 0.244D 05	
10	12	1/2-1/2 1/2-3/2	0.168D-03 0.157D-03	-0.100D-03 -0.145D-03	0.346D-02 0.145D-01	0.295D-02 0.122D-01			0.556D 04 0.118D 05		0.475D 04 0.999D 04	
10	13	1/2-1/2 1/2-3/2	0.440D-04 0.829D-04	-0.370D-04 -0.735D-04	0.476D-03 0.336D-02	0.336D-03 0.264D-02			0.110D 04 0.353D 04		0.781D 03 0.309D 04	
10	14	1/2-1/2 1/2-3/2	0.264D-04 0.576D-04	-0.206D-04 -0.501D-04	0.152D-03 0.145D-02	0.931D-04 0.109D-02			0.445D 03 0.213D 04		0.272D 03 0.161D 04	
10	15	1/2-1/2 1/2-3/2	0.171D-04 0.431D-04	-0.121D-04 -0.366D-04	0.590D-04 0.752D-03	0.295D-04 0.542D-03			0.202D 03 0.129D 04		0.101D 03 0.929D 03	

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES. FOR THE SERIES $ns^2s_{1/2} - np^2p_{1/2}$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION J-J'	P_{IJ}^0		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	VELOCITY		LENGTH	
11	11	1/2-1/2 1/2-3/2	0.8190-03 0.8560-03	-0.7700-03 -0.7890-03	0.1120 01 0.2300 01	0.9920 00 0.1960 01				0.5610 05 0.6510 05	0.4960 05 0.5530 05
11	12	1/2-1/2 1/2-3/2	0.2090-03 0.2630-03	-0.1960-03 -0.2450-03	0.2750-01 0.8640-01	0.2430-01 0.7510-01				0.9600 04 0.1550 05	0.8480 04 0.1350 05
11	13	1/2-1/2 1/2-3/2	0.8350-04 0.1270-03	-0.7460-04 -0.1150-03	0.3080-02 0.1410-01	0.2460-02 0.1150-01				0.2210 04 0.5110 04	0.1760 04 0.4190 04
11	14	1/2-1/2 1/2-3/2	0.4970-04 0.8270-04	-0.4560-04 -0.7670-04	0.6970-03 0.4930-02	0.7540-03 0.4240-02				0.9540 03 0.2650 04	0.8020 03 0.2280 04
11	15	1/2-1/2 1/2-3/2	0.3310-04 0.6000-04	-0.2990-04 -0.5530-04	0.3500-03 0.2290-02	0.2850-03 0.1950-02				0.4790 03 0.1580 04	0.3900 03 0.1340 04
12	12	1/2-1/2 1/2-3/2	0.7150-03 0.7460-03	-0.6810-03 -0.6970-03	0.1250 01 0.2560 01	0.1140 01 0.2240 01				0.2520 05 0.3370 05	0.2650 05 0.2950 05
12	13	1/2-1/2 1/2-3/2	0.1610-03 0.2110-03	-0.1550-03 -0.2010-03	0.2380-01 0.7990-01	0.2190-01 0.7300-01				0.3560 04 0.6890 04	0.3650 04 0.6290 04
12	14	1/2-1/2 1/2-3/2	0.8050-04 0.1190-03	-0.6870-04 -0.1030-03	0.4080-02 0.1760-01	0.2970-02 0.1330-01				0.1440 04 0.3150 04	0.1050 04 0.2390 04
12	15	1/2-1/2 1/2-3/2	0.5140-04 0.7960-04	-0.4960-04 -0.7650-04	0.1350-02 0.6430-02	0.1260-02 0.5950-02				0.7230 03 0.1740 04	0.6740 03 0.1610 04
13	13	1/2-1/2 1/2-3/2	0.2470-04 0.4750-04	-0.3620-03 -0.3820-03	0.1390-02 0.1000-01	0.2980 00 0.6370 00				0.3760 02 0.1470 03	0.8030 04 0.9340 04
13	14	1/2-1/2 1/2-3/2	-0.1380-03 -0.2090-04	-0.5190-03 -0.5520-03	0.2030-01 0.9230-03	0.2880 00 0.6430 00				0.2490 04 0.5820 02	0.3530 05 0.4050 05
13	15	1/2-1/2 1/2-3/2	-0.2070-03 -0.5380-04	-0.5160-03 -0.5610-03	0.3290-01 0.4390-02	0.2040 00 0.4770 00				0.7860 04 0.5330 03	0.4870 05 0.5790 05
14	14	1/2-1/2 1/2-3/2	0.4550-03 0.2030-03	-0.2990-03 -0.2890-03	0.6540 00 0.2450 00	0.2770 00 0.4960 00				0.9500 04 0.1540 04	0.4020 04 0.3920 04
14	15	1/2-1/2 1/2-3/2	0.2810-03 0.1360-03	-0.5230-03 -0.5090-03	0.1140 00 0.5230-01	0.3950 00 0.7360 00				0.7660 04 0.1810 04	0.2660 05 0.2550 05
15	15	1/2-1/2 1/2-3/2	0.5960-03 0.2340-03	-0.1260-03 -0.5760-04	0.1460 01 0.4310 00	0.6560-01 0.7510-01				0.1210 05 0.1940 04	0.5430 03 0.3380 03

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_J - mp^2P_J'$ CALCULATED USING THE RELATIVISTIC M-F-S WAVE FUNCTIONS FOR CESIUM. THE OSCILLATOR STRENGTHS FOR CESIUM, AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M^0_{IJ}		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	VELOCITY		LENGTH	
5	7	3/2-1/2	0.429D-03	-0.828D-03	0.116D-01	0.432D-01	0.251D 00	0.816D 06	0.305D 07	0.816D 06	0.305D 07
		3/2-3/2	0.410D-03	-0.752D-03	0.207D-02	0.696D-02	0.211D-01	0.696D 05	0.258D 06	0.696D 05	0.258D 06
		5/2-3/2	0.437D-03	-0.792D-03	0.143D-01	0.469D-01	0.204D 00	0.469D 01	0.253D 07	0.469D 01	0.253D 07
5	8	3/2-1/2	0.620D-04	-0.366D-03	0.157D-03	0.547D-02	0.652D 00	0.263D 05	0.917D 06	0.263D 05	0.917D 06
		3/2-3/2	0.724D-04	-0.351D-03	0.425D-04	0.100D-02	0.208D 00	0.362D 04	0.851D 05	0.362D 04	0.851D 05
		5/2-3/2	0.816D-04	-0.361D-03	0.327D-03	0.641D-02	0.153D 01	0.641D 02	0.804D 06	0.641D 02	0.804D 06
5	9	3/2-1/2	-0.1C7D-04	-0.234D-03	0.401D-05	0.190D-02	0.915D-01	0.922D 03	0.438D 06	0.922D 03	0.438D 06
		3/2-3/2	-0.991D-06	-0.228D-03	0.682D-08	0.363D-03	0.299D-01	0.363D 00	0.420D 05	0.420D 00	0.420D 05
		5/2-3/2	0.411D-05	-0.233D-03	0.709D-06	0.228D-02	0.219D 00	0.228D 02	0.390D 06	0.228D 02	0.390D 06
5	10	3/2-1/2	-0.284D-04	-0.170D-03	0.260D-06	0.929D-03	0.109D-01	0.701D 04	0.251D 06	0.701D 04	0.251D 06
		3/2-3/2	-0.209D-04	-0.168D-03	0.281D-05	0.181D-03	0.100D-01	0.381D 03	0.245D 05	0.381D 03	0.245D 05
		5/2-3/2	-0.176D-04	-0.170D-03	0.120D-04	0.113D-02	0.733D-01	0.241D 04	0.226D 06	0.241D 04	0.226D 06
5	11	3/2-1/2	-0.316D-04	-0.132D-03	0.306D-04	0.535D-03	0.143D-01	0.907D 04	0.159D 06	0.907D 04	0.159D 06
		3/2-3/2	-0.258D-04	-0.131D-03	0.409D-05	0.105D-03	0.450D-02	0.607D 03	0.157D 04	0.607D 03	0.157D 04
		5/2-3/2	-0.234D-04	-0.133D-03	0.203D-04	0.653D-03	0.335D-01	0.446D 04	0.144D 06	0.446D 04	0.144D 06
5	12	3/2-1/2	-0.304D-04	-0.107D-03	0.276D-04	0.340D-03	0.798D-02	0.867D 04	0.107D 06	0.867D 04	0.107D 06
		3/2-3/2	-0.259D-04	-0.107D-03	0.599D-05	0.676D-04	0.248D-02	0.628D 03	0.107D 05	0.628D 03	0.107D 05
		5/2-3/2	-0.240D-04	-0.108D-03	0.207D-05	0.418D-03	0.184D-01	0.484D 04	0.174D 05	0.484D 04	0.174D 05
5	13	3/2-1/2	-0.288D-04	-0.693D-04	0.242D-04	0.140D-03	0.792D 04	0.792D 04	0.460D 05	0.792D 04	0.460D 05
		3/2-3/2	-0.252D-04	-0.701D-04	0.372D-05	0.287D-04	0.610D 03	0.471D 04	0.426D 05	0.610D 03	0.471D 04
		5/2-3/2	-0.237D-04	-0.705D-04	0.197D-04	0.176D-03	0.479D 04	0.479D 04	0.426D 05	0.479D 04	0.426D 05
5	14	3/2-1/2	-0.261D-04	-0.577D-04	0.196D-04	0.961D-04	0.661D 04	0.324D 05	0.324D 05	0.661D 04	0.324D 05
		3/2-3/2	-0.232D-04	-0.585D-04	0.309D-05	0.197D-04	0.521D 03	0.335D 04	0.335D 04	0.521D 03	0.335D 04
		5/2-3/2	-0.218D-04	-0.588D-04	0.166D-04	0.120D-03	0.414D 04	0.300D 05	0.300D 05	0.414D 04	0.300D 05
5	15	3/2-1/2	-0.236D-04	-0.493D-04	0.158D-04	0.693D-04	0.545D 04	0.239D 05	0.239D 05	0.545D 04	0.239D 05
		3/2-3/2	-0.211D-04	-0.501D-04	0.254D-05	0.143D-04	0.437D 03	0.247D 04	0.222D 05	0.437D 03	0.247D 04
		5/2-3/2	-0.200D-04	-0.503D-04	0.137D-04	0.871D-04	0.350D 04	0.350D 04	0.222D 05	0.350D 04	0.222D 05
6	8	3/2-1/2	0.361D-03	-0.269D-03	0.191D-01	0.106D-01	0.248D 06	0.138D 06	0.138D 06	0.248D 06	0.138D 06
		3/2-3/2	0.273D-03	-0.203D-03	0.213D-02	0.117D-02	0.145D 05	0.805D 04	0.805D 04	0.145D 05	0.805D 04
		5/2-3/2	0.290D-03	-0.217D-03	0.146D-01	0.818D-02	0.146D 06	0.817D 05	0.817D 05	0.146D 06	0.817D 05
6	9	3/2-1/2	0.218D-03	-0.180D-03	0.430D-02	0.294D-02	0.146D 06	0.100D 06	0.100D 06	0.430D 02	0.100D 06
		3/2-3/2	0.174D-03	-0.143D-03	0.547D-03	0.370D-03	0.946D 04	0.640D 04	0.640D 04	0.946D 04	0.640D 04
		5/2-3/2	0.182D-03	-0.152D-03	0.362D-02	0.251D-02	0.923D 05	0.639D 05	0.639D 05	0.923D 05	0.639D 05
6	10	3/2-1/2	0.153D-03	-0.131D-03	0.175D-02	0.128D-02	0.880D 05	0.645D 05	0.645D 05	0.880D 05	0.645D 05
		3/2-3/2	0.126D-03	-0.107D-03	0.236D-03	0.171D-03	0.599D 04	0.432D 04	0.432D 04	0.599D 04	0.432D 04
		5/2-3/2	0.131D-03	-0.113D-03	0.154D-02	0.114D-02	0.576D 05	0.429D 05	0.429D 05	0.576D 05	0.429D 05
6	11	3/2-1/2	0.117D-03	-0.102D-03	0.912D-03	0.694D-03	0.565D 05	0.430D 05	0.430D 05	0.565D 05	0.430D 05
		3/2-3/2	0.574D-04	-0.842D-04	0.127D-03	0.950D-04	0.296D 04	0.296D 04	0.296D 04	0.296D 04	0.296D 04
		5/2-3/2	0.101D-03	-0.884D-04	0.820D-03	0.633D-03	0.378D 05	0.292D 05	0.292D 05	0.378D 05	0.292D 05
6	12	3/2-1/2	0.928D-04	-0.819D-04	0.543D-03	0.423D-03	0.382D 05	0.268D 05	0.268D 05	0.382D 05	0.268D 05
		3/2-3/2	0.785D-04	-0.665D-04	0.775D-04	0.590D-04	0.274D 04	0.208D 04	0.208D 04	0.274D 04	0.208D 04
		5/2-3/2	0.809D-04	-0.718D-04	0.497D-03	0.392D-03	0.260D 05	0.205D 05	0.205D 05	0.260D 05	0.205D 05
6	13	3/2-1/2	0.667D-04	-0.802D-04	0.269D-03	0.389D-03	0.206D 05	0.296D 05	0.296D 05	0.206D 05	0.296D 05
		3/2-3/2	0.563D-04	-0.706D-04	0.383D-04	0.602D-04	0.147D 04	0.231D 04	0.231D 04	0.147D 04	0.231D 04
		5/2-3/2	0.580D-04	-0.730D-04	0.245D-03	0.389D-03	0.140D 05	0.221D 05	0.221D 05	0.140D 05	0.221D 05

$nd^2 J - \text{mp}^2 J'$

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES. FOR THE SERIES CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M^0_{IJ}			f_{IJ}			STONE	f_{IJ}	AGNEW	$A_{IJ}(\text{sec}^{-1})$		
			VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH				VELOCITY	LENGTH	
6	14	3/2-1/2 3/2-3/2 5/2-3/2	0.555D-04 0.470D-04 0.484D-04	-0.687D-04 -0.608D-04 -0.628D-04	0.181D-03 0.259D-04 0.166D-03	0.277D-03 0.433D-04 0.279D-03	0.147D 05 0.106D 04 0.100D 05				0.225D 05 0.176D 04 0.168D 05			
6	15	3/2-1/2 3/2-3/2 5/2-3/2	0.473D-04 0.402D-04 0.414D-04	-0.596D-04 -0.530D-04 -0.547D-04	0.129D-03 0.186D-04 0.119D-03	0.204D-03 0.322D-04 0.207D-03	0.109D 05 0.137D 05 0.747D 04				0.173D 05 0.130D 05			
7	9	3/2-1/2 3/2-3/2 5/2-3/2	0.417D-03 0.350D-03 0.350D-03	-0.319D-03 -0.254D-03 -0.265D-03	0.500D-01 0.363D-02 0.416D-01	0.293D-01 0.239D-02 0.239D-01	0.169D 06 0.112D 05 0.108D 06				0.646D 04 0.622D 05			
7	10	3/2-1/2 3/2-3/2 5/2-3/2	0.249D-03 0.208D-03 0.215D-03	-0.203D-03 -0.168D-03 -0.175D-03	0.106D-01 0.147D-02 0.950D-02	0.705D-02 0.961D-03 0.624D-02	0.101D 06 0.717D 04 0.685D 05				0.675D 05 0.450D 05			
7	11	3/2-1/2 3/2-3/2 5/2-3/2	0.176D-03 0.150D-03 0.155D-03	-0.146D-03 -0.123D-03 -0.128D-03	0.424D-02 0.612D-03 0.393D-02	0.293D-02 0.413D-03 0.268D-02	0.636D 05 0.465D 04 0.441D 05				0.439D 05 0.314D 04 0.301D 05			
7	12	3/2-1/2 3/2-3/2 5/2-3/2	0.135D-03 0.116D-03 0.120D-03	-0.114D-03 -0.965D-04 -0.998D-04	0.220D-02 0.325D-03 0.207D-02	0.155D-02 0.224D-03 0.144D-02	0.300D 05 0.217D 04 0.208D 05				0.300D 05 0.264D 05 0.205D 04 0.193D 05			
7	13	3/2-1/2 3/2-3/2 5/2-3/2	0.107D-03 0.940D-04 0.963D-04	-0.102D-03 -0.901D-04 -0.924D-04	0.127D-02 0.190D-03 0.124D-02	0.117D-02 0.180D-03 0.114D-02	0.288D 05 0.223D 04 0.209D 05				0.196D 05 0.153D 04 0.144D 05			
7	14	3/2-1/2 3/2-3/2 5/2-3/2	0.894D-04 0.779D-04 0.798D-04	-0.859D-04 -0.759D-04 -0.778D-04	0.824D-03 0.128D-03 0.809D-03	0.778D-03 0.121D-03 0.758D-03	0.208D 05 0.162D 04 0.152D 05				0.196D 05 0.153D 04 0.144D 05			
7	15	3/2-1/2 3/2-3/2 5/2-3/2	0.749D-04 0.663D-04 0.678D-04	-0.735D-04 -0.652D-04 -0.668D-04	0.569D-03 0.890D-04 0.562D-03	0.548D-03 0.861D-04 0.544D-03	0.155D 05 0.121D 04 0.114D 05				0.149D 05 0.117D 04 0.110D 05			
8	10	3/2-1/2 3/2-3/2 5/2-3/2	0.377D-03 0.304D-03 0.318D-03	-0.289D-03 -0.232D-03 -0.243D-03	0.710D-01 0.89D-02 0.59D-01	0.419D-01 0.525D-02 0.349D-01	0.754D 05 0.533D 04 0.519D 05				0.468D 05 0.311D 04 0.302D 05			
8	11	3/2-1/2 3/2-3/2 5/2-3/2	0.228D-03 0.190D-03 0.198D-03	-0.187D-03 -0.155D-03 -0.161D-03	0.150D-01 0.206D-02 0.135D-01	0.100D-01 0.137D-02 0.895D-02	0.507D 05 0.357D 04 0.345D 05				0.339D 05 0.236D 04 0.229D 05			
8	12	3/2-1/2 3/2-3/2 5/2-3/2	0.164D-03 0.139D-03 0.144D-03	-0.135D-03 -0.114D-03 -0.119D-03	0.601D-02 0.856D-03 0.555D-02	0.412D-02 0.578D-03 0.379D-02	0.334D 05 0.241D 04 0.231D 05				0.229D 05 0.162D 04 0.158D 05			
8	13	3/2-1/2 3/2-3/2 5/2-3/2	0.131D-03 0.114D-03 0.118D-03	-0.118D-03 -0.102D-03 -0.105D-03	0.335D-02 0.503D-03 0.323D-02	0.270D-02 0.405D-03 0.260D-02	0.248D 05 0.187D 04 0.179D 05				0.199D 05 0.151D 04 0.144D 05			
8	14	3/2-1/2 3/2-3/2 5/2-3/2	0.106D-03 0.925D-04 0.954D-04	-0.566D-04 -0.843D-04 -0.868D-04	0.200D-02 0.303D-03 0.194D-02	0.166D-02 0.252D-03 0.161D-02	0.177D 05 0.135D 04 0.129D 05				0.147D 05 0.112D 04 0.107D 05			
8	15	3/2-1/2 3/2-3/2 5/2-3/2	0.885D-04 0.775D-04 0.799D-04	-0.813D-04 -0.713D-04 -0.733D-04	0.130D-02 0.200D-03 0.128D-02	0.110D-02 0.169D-03 0.108D-02	0.132D 05 0.101D 04 0.964D 04				0.111D 05 0.856D 03 0.811D 04			

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D - mp^2P'$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION J-J'	M_{IJ}^0		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	LENGTH		VELOCITY	LENGTH
9	11	3/2-1/2	0.325D-03	-0.252D-03	0.861D-01	0.505D-01	0.861D-01	0.505D-01		0.379D 05	0.223D 05
		3/2-3/2	0.264D-03	-0.202D-03	0.108D-01	0.262D-02	0.108D-01	0.262D-02		0.253D 04	0.147D 04
		5/2-3/2	0.277D-03	-0.211D-03	0.721D-01	0.419D-01	0.721D-01	0.419D-01		0.247D 05	0.144D 05
9	12	3/2-1/2	0.203D-03	-0.167D-03	0.184D-01	0.124D-01	0.184D-01	0.124D-01		0.257D 05	0.174D 05
		3/2-3/2	0.168D-03	-0.138D-03	0.249D-02	0.168D-02	0.249D-02	0.168D-02		0.178D 04	0.120D 04
		5/2-3/2	0.175D-03	-0.143D-03	0.164D-01	0.110D-01	0.164D-01	0.110D-01		0.174D 05	0.116D 05
9	13	3/2-1/2	0.157D-03	-0.135D-03	0.844D-02	0.623D-02	0.844D-02	0.623D-02		0.201D 05	0.149D 05
		3/2-3/2	0.139D-03	-0.115D-03	0.124D-02	0.905D-03	0.124D-02	0.905D-03		0.150D 04	0.109D 04
		5/2-3/2	0.139D-03	-0.120D-03	0.793D-02	0.592D-02	0.793D-02	0.592D-02		0.142D 05	0.106D 05
9	14	3/2-1/2	0.122D-03	-0.108D-03	0.437D-02	0.340D-02	0.437D-02	0.340D-02		0.143D 05	0.111D 05
		3/2-3/2	0.105D-03	-0.927D-04	0.648D-03	0.501D-03	0.648D-03	0.501D-03		0.117D 04	0.825D 03
		5/2-3/2	0.109D-03	-0.960D-04	0.417D-02	0.324D-02	0.417D-02	0.324D-02		0.102D 05	0.792D 04
9	15	3/2-1/2	0.995D-04	-0.893D-04	0.261D-02	0.210D-02	0.261D-02	0.210D-02		0.105D 05	0.844D 04
		3/2-3/2	0.861D-04	-0.773D-04	0.390D-03	0.315D-03	0.390D-03	0.315D-03		0.787D 03	0.635D 03
		5/2-3/2	0.891D-04	-0.796D-04	0.252D-02	0.201D-02	0.252D-02	0.201D-02		0.756D 04	0.603D 04
10	12	3/2-1/2	0.295D-03	-0.228D-03	0.104D 00	0.619D-01	0.104D 00	0.619D-01		0.205D 05	0.122D 05
		3/2-3/2	0.238D-03	-0.183D-03	0.131D-01	0.775D-02	0.131D-01	0.775D-02		0.137D 04	0.811D 03
		5/2-3/2	0.250D-03	-0.192D-03	0.876D-01	0.517D-01	0.876D-01	0.517D-01		0.134D 05	0.793D 04
10	13	3/2-1/2	0.200D-03	-0.174D-03	0.263D-01	0.198D-01	0.263D-01	0.198D-01		0.171D 05	0.128D 05
		3/2-3/2	0.170D-03	-0.148D-03	0.373D-02	0.283D-02	0.373D-02	0.283D-02		0.124D 04	0.940D 03
		5/2-3/2	0.177D-03	-0.152D-03	0.245D-01	0.181D-01	0.245D-01	0.181D-01		0.120D 05	0.890D 04
10	14	3/2-1/2	0.146D-03	-0.124D-03	0.106D-01	0.762D-02	0.106D-01	0.762D-02		0.121D 05	0.869D 04
		3/2-3/2	0.126D-03	-0.106D-03	0.157D-02	0.110D-02	0.157D-02	0.110D-02		0.908D 03	0.638D 03
		5/2-3/2	0.130D-03	-0.111D-03	0.995D-02	0.730D-02	0.995D-02	0.730D-02		0.854D 04	0.627D 04
10	15	3/2-1/2	0.115D-03	-0.100D-03	0.549D-02	0.419D-02	0.549D-02	0.419D-02		0.878D 04	0.669D 04
		3/2-3/2	0.991D-04	-0.860D-04	0.819D-03	0.616D-03	0.819D-03	0.616D-03		0.660D 03	0.496D 03
		5/2-3/2	0.102D-03	-0.895D-04	0.524D-02	0.403D-02	0.524D-02	0.403D-02		0.628D 04	0.483D 04
11	13	3/2-1/2	0.301D-03	-0.242D-03	0.154D 00	0.998D-01	0.154D 00	0.998D-01		0.149D 05	0.966D 04
		3/2-3/2	0.250D-03	-0.200D-03	0.207D-01	0.132D-01	0.207D-01	0.132D-01		0.106D 04	0.682D 03
		5/2-3/2	0.260D-03	-0.209D-03	0.136D 00	0.872D-01	0.136D 00	0.872D-01		0.102D 05	0.658D 04
11	14	3/2-1/2	0.186D-03	-0.165D-03	0.319D-01	0.252D-01	0.319D-01	0.252D-01		0.105D 05	0.830D 04
		3/2-3/2	0.158D-03	-0.142D-03	0.453D-02	0.366D-02	0.453D-02	0.366D-02		0.764D 03	0.618D 03
		5/2-3/2	0.165D-03	-0.145D-03	0.300D-01	0.231D-01	0.300D-01	0.231D-01		0.748D 04	0.577D 04
11	15	3/2-1/2	0.137D-03	-0.114D-03	0.130D-01	0.886D-02	0.130D-01	0.886D-02		0.769D 04	0.525D 04
		3/2-3/2	0.120D-03	-0.971D-04	0.197D-02	0.129D-02	0.197D-02	0.129D-02		0.591D 03	0.386D 03
		5/2-3/2	0.122D-03	-0.103D-03	0.122D-01	0.866D-02	0.122D-01	0.866D-02		0.544D 04	0.387D 04
12	14	3/2-1/2	0.297D-03	0.971D-04	0.205D 00	0.220D-01	0.205D 00	0.220D-01		0.105D 05	0.113D 04
		3/2-3/2	-0.146D-03	0.135D-03	0.970D-02	0.821D-02	0.970D-02	0.821D-02		0.265D 03	0.224D 03
		5/2-3/2	0.575D-03	0.129D-03	0.907D 00	0.459D-01	0.907D 00	0.459D-01		0.363D 05	0.184D 04
12	15	3/2-1/2	0.265D-03	0.680D-04	0.878D-01	0.578D-02	0.878D-01	0.578D-02		0.157D 05	0.103D 04
		3/2-3/2	-0.132D-03	0.122D-03	0.430D-02	0.369D-02	0.430D-02	0.369D-02		0.335D 03	0.339D 03
		5/2-3/2	0.522D-03	0.115D-03	0.406D 00	0.197D-01	0.406D 00	0.197D-01		0.551D 05	0.267D 04
13	15	3/2-1/2	0.148D-03	0.236D-03	0.684D-01	0.173D 00	0.684D-01	0.173D 00		0.158D 04	0.500D 04
		3/2-3/2	-0.671D-04	0.258D-03	0.272D-02	0.400D-01	0.272D-02	0.400D-01		0.418D 02	0.616D 03
		5/2-3/2	0.258D-03	0.254D-03	0.244D 00	0.236D 00	0.244D 00	0.236D 00		0.550D 04	0.533D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_j - md^2D_j$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M^0_{ij}		f_{ij}		f_{ij}		f_{ij}		$A_{ij} (\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH		
6	5	1/2-3/2 3/2-3/2 3/2-5/2	-0.560D-03 -0.829D-03 -0.900D-03	0.961D-03 0.805D-03 0.848D-03	0.254D 00 0.227D 01 0.233D 00	0.254D 00 0.214D 01 0.207D 00	0.251D 00		0.935D 06 0.116D 06 0.850D 06	0.936D 06 0.109D 06 0.755D 06		
6	6	1/2-3/2 3/2-3/2 3/2-5/2	0.299D-02 0.299D-02 0.298D-02	-0.364D-02 -0.366D-02 -0.363D-02	0.717D 00 0.756D 01 0.670D 00	0.106D 01 0.113D 00 0.999D 00	0.298D 00 0.397D-01 0.332D 00		0.311D 08 0.594D 07 0.354D 08	0.461D 08 0.888D 07 0.527D 08		
6	7	1/2-3/2 3/2-3/2 3/2-5/2	0.132D-02 0.125D-02 0.126D-02	-0.142D-02 -0.135D-02 -0.137D-02	0.107D 00 0.101D 01 0.920D-01	0.125D 00 0.117D 01 0.107D 00	0.927D-01 0.110D-01 0.951D-01		0.786D 07 0.138D 07 0.840D 07	0.918D 07 0.160D 07 0.980D 07		
6	8	1/2-3/2 3/2-3/2 3/2-5/2	0.800D-03 0.745D-03 0.756D-03	-0.780D-03 -0.718D-03 -0.735D-03	0.353D-02 0.317D-02 0.293D-01	0.335D-01 0.294D-02 0.277D-01	0.419D-01 0.561D-02 0.439D-01		0.325D 07 0.546D 06 0.337D 07	0.309D 07 0.507D 06 0.319D 07		
6	9	1/2-3/2 3/2-3/2 3/2-5/2	0.548D-03 0.502D-03 0.511D-03	-0.480D-03 -0.430D-03 -0.444D-03	0.156D-01 0.135D-02 0.126D-01	0.120D-01 0.991D-02 0.950D-02	0.228D-01 0.250D-02 0.223D-01		0.162D 07 0.263D 06 0.164D 07	0.124D 07 0.193D 06 0.124D 07		
6	10	1/2-3/2 3/2-3/2 3/2-5/2	0.406D-03 0.369D-03 0.377D-03	-0.331D-03 -0.291D-03 -0.302D-03	0.827D-02 0.703D-03 0.659D-02	0.550D-02 0.436D-03 0.423D-02	0.139D-01 0.150D-02 0.135D-01		0.923D 06 0.147D 06 0.922D 06	0.614D 06 0.915D 05 0.592D 06		
6	11	1/2-3/2 3/2-3/2 3/2-5/2	0.319D-03 0.288D-03 0.294D-03	-0.247D-03 -0.214D-03 -0.223D-03	0.497D-02 0.417D-03 0.393D-02	0.299D-02 0.230D-03 0.226D-02	0.920D-02 0.100D-02 0.860D-02		0.581D 06 0.918D 05 0.576D 06	0.350D 06 0.506D 05 0.331D 06		
6	12	1/2-3/2 3/2-3/2 3/2-5/2	0.260D-03 0.234D-03 0.239D-03	-0.195D-03 -0.166D-03 -0.174D-03	0.326D-02 0.270D-03 0.255D-02	0.183D-02 0.137D-03 0.136D-02	0.869D-02 0.837D-02 0.837D-02		0.353D 06 0.615D 05 0.387D 06	0.220D 06 0.312D 05 0.208D 06		
6	13	1/2-3/2 3/2-3/2 3/2-5/2	0.218D-03 0.195D-03 0.200D-03	-0.159D-03 -0.135D-03 -0.142D-03	0.226D-02 0.187D-03 0.177D-02	0.121D-02 0.895D-04 0.890D-03	0.470D-02 0.530D-02 0.530D-02		0.279D 06 0.485D 05 0.274D 06	0.149D 06 0.208D 05 0.138D 06		
6	14	1/2-3/2 3/2-3/2 3/2-5/2	0.187D-03 0.167D-03 0.171D-03	-0.135D-03 -0.113D-03 -0.119D-03	0.165D-02 0.136D-03 0.128D-02	0.854D-03 0.624D-04 0.623D-03	0.380D-02 0.396D-02 0.396D-02		0.207D 06 0.321D 05 0.203D 06	0.107D 06 0.148D 05 0.985D 05		
7	6	1/2-3/2 3/2-3/2 3/2-5/2	0.140D-03 0.251D-03 0.259D-03	0.490D-03 0.376D-03 0.406D-03	0.217D-01 0.120D-01 0.807D-01	0.267D 00 0.202D-01 0.198D 00	0.327D 00 0.320D-01 0.309D 00		0.491D 04 0.331D 04 0.168D 05	0.604D 05 0.556D 04 0.414D 05		
7	7	1/2-3/2 3/2-3/2 3/2-5/2	0.181D-02 0.188D-02 0.186D-02	-0.189D-02 -0.200D-02 -0.198D-02	0.701D 00 0.788D 01 0.693D 00	0.761D 00 0.893D-01 0.785D 00	0.237D 00 0.340D-01 0.282D 00		0.429D 07 0.843D 06 0.523D 07	0.465D 07 0.105D 07 0.593D 07		
7	8	1/2-3/2 3/2-3/2 3/2-5/2	0.103D-02 0.101D-02 0.101D-02	-0.990D-03 -0.984D-03 -0.986D-03	0.159D 00 0.159D 01 0.143D 00	0.148D 00 0.151D-01 0.136D 00	0.822D-01 0.105D-01 0.890D-01		0.154D 07 0.366D 06 0.220D 07	0.181D 07 0.347D 06 0.210D 07		
7	9	1/2-3/2 3/2-3/2 3/2-5/2	0.690D-03 0.664D-03 0.667D-03	-0.644D-03 -0.621D-03 -0.626D-03	0.617D-01 0.586D-02 0.533D-01	0.537D-01 0.513D-02 0.469D-01	0.391D-01 0.470D-02 0.410D-01		0.103D 07 0.165D 06 0.112D 07	0.894D 06 0.142D 06 0.900D 06		
7	10	1/2-3/2 3/2-3/2 3/2-5/2	0.507D-03 0.482D-03 0.486D-03	-0.464D-03 -0.440D-03 -0.445D-03	0.306D-01 0.283D-02 0.258D-01	0.256D-01 0.235D-02 0.217D-01	0.221D-01 0.260D-02 0.228D-01		0.606D 06 0.107D 06 0.651D 06	0.586D 06 0.888D 05 0.546D 06		

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2P_{1/2} \rightarrow m^2D_{3/2}$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M_{if}^0		f_{ij}		STONE	f_{ij}	AGNEW	$A_{ij}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH				VELOCITY	LENGTH
7	11	1/2-3/2 3/2-3/2 3/2-5/2	0.396D-03 0.373D-03 0.376D-03	-0.357D-03 -0.335D-03 -0.340D-03	0.176D-01 0.160D-02 0.147D-01	0.144D-01 0.129D-02 0.120D-01	0.139D-01 0.160D-02 0.142D-01			0.389D 06 0.674D 05 0.413D 06	0.317D 06 0.544D 05 0.336D 06
7	12	1/2-3/2 3/2-3/2 3/2-5/2	0.321D-03 0.301D-03 0.304D-03	-0.287D-03 -0.267D-03 -0.272D-03	0.112D-01 0.101D-02 0.924D-02	0.897D-02 0.793D-03 0.738D-02				0.266D 06 0.456D 05 0.280D 06	0.213D 06 0.360D 05 0.223D 06
7	13	1/2-3/2 3/2-3/2 3/2-5/2	0.268D-03 0.251D-03 0.254D-03	-0.239D-03 -0.221D-03 -0.225D-03	0.762D-02 0.679D-03 0.625D-02	0.603D-02 0.492D-02				0.190D 06 0.325D 05 0.200D 06	0.150D 06 0.252D 05 0.157D 06
7	14	1/2-3/2 3/2-3/2 3/2-5/2	0.229D-03 0.214D-03 0.216D-03	-0.203D-03 -0.187D-03 -0.191D-03	0.547D-02 0.485D-03 0.447D-02	0.429D-02 0.372D-03 0.347D-02				0.142D 06 0.241D 05 0.148D 06	0.111D 06 0.185D 05 0.115D 06
8	7	1/2-3/2 3/2-3/2 3/2-5/2	-0.255D-04 0.115D-03 0.909D-04	0.363D-03 0.287D-03 0.312D-03	0.175D-02 0.471D-02 0.246D-01	0.397D 00 0.294D-01 0.290D 00				0.670D 06 0.294D-01 0.246D-01	0.152D 05 0.129D 04 0.990D 04
8	8	1/2-3/2 3/2-3/2 3/2-5/2	0.124D-02 0.132D-02 0.130D-02	-0.124D-02 -0.135D-02 -0.133D-02	0.670D 00 0.789D-01 0.687D 00	0.672D 00 0.825D-01 0.718D 00				0.987D 06 0.215D 06 0.126D 07	0.991D 06 0.224D 06 0.132D 07
8	9	1/2-3/2 3/2-3/2 3/2-5/2	0.767D-03 0.777D-03 0.774D-03	-0.730D-03 -0.746D-03	0.173D 00 0.162D 00	0.156D 00 0.150D 00				0.561D 06 0.112D 06 0.669D 06	0.507D 06 0.103D 06 0.617D 06
8	10	1/2-3/2 3/2-3/2 3/2-5/2	0.544D-03 0.540D-03 0.540D-03	-0.508D-03 -0.507D-03 -0.507D-03	0.721D-01 0.727D-02 0.654D-01	0.628D-01 0.639D-02 0.577D-01				0.340D 06 0.655D 05 0.334D 06	0.296D 06 0.376D 05 0.347D 06
8	11	1/2-3/2 3/2-3/2 3/2-5/2	0.416D-03 0.408D-03 0.409D-03	-0.385D-03 -0.379D-03 -0.380D-03	0.378D-01 0.372D-02 0.336D-01	0.324D-01 0.320D-02 0.290D-01				0.231D 06 0.416D 05 0.252D 06	0.189D 06 0.360D 05 0.218D 06
8	12	1/2-3/2 3/2-3/2 3/2-5/2	0.333D-03 0.325D-03 0.326D-03	-0.307D-03 -0.300D-03 -0.301D-03	0.227D-01 0.220D-02 0.199D-01	0.192D-01 0.187D-02 0.170D-01				0.152D 06 0.284D 05 0.172D 06	0.129D 06 0.241D 05 0.147D 06
8	13	1/2-3/2 3/2-3/2 3/2-5/2	0.276D-03 0.268D-03 0.269D-03	-0.253D-03 -0.246D-03 -0.248D-03	0.148D-01 0.142D-02 0.129D-01	0.125D-01 0.120D-02 0.109D-01				0.110D 06 0.203D 05 0.123D 06	0.924D 05 0.171D 05 0.104D 06
8	14	1/2-3/2 3/2-3/2 3/2-5/2	0.235D-03 0.227D-03 0.228D-03	-0.215D-03 -0.208D-03 -0.209D-03	0.103D-01 0.987D-03 0.897D-02	0.868D-02 0.835D-03 0.754D-02				0.819D 05 0.151D 05 0.915D 05	0.687D 05 0.126D 05 0.769D 05
9	8	1/2-3/2 3/2-3/2 3/2-5/2	-0.217D-04 0.574D-04 0.745D-04	0.310D-03 0.229D-03 0.251D-03	0.249D-02 0.672D-02 0.324D-01	0.566D 00 0.370D-01 0.367D 00				0.251D 02 0.749D 02 0.286D 03	0.510D 04 0.413D 03 0.325D 04
9	9	1/2-3/2 3/2-3/2 3/2-5/2	0.943D-03 0.101D-02 0.599D-03	-0.950D-03 -0.104D-02 -0.102D-02	0.683D 00 0.622D-01 0.713D 00	0.594D 00 0.855D-01 0.750D 00				0.323D 06 0.721D 05 0.422D 06	0.328D 06 0.759D 05 0.444D 06
9	10	1/2-3/2 3/2-3/2 3/2-5/2	0.558D-03 0.617D-03 0.613D-03	-0.573D-03 -0.591D-03	0.179D 00 0.195D 00 0.173D 00	0.164D 00 0.181D-01 0.161D 00				0.200D 06 0.415D 05 0.246D 06	0.184D 06 0.386D 05 0.230D 06

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2p - md^2d$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M_{IJ}^2			f_{IJ}			f_{IJ}			AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	VELOCITY	LENGTH	STONE	VELOCITY	LENGTH				
9	11	1/2-3/2 3/2-3/2 3/2-5/2	0.436D-03 0.441D-03 0.440D-03	-0.410D-03 -0.417D-03 -0.416D-03	0.771D-01 0.805D-02 0.720D-01	0.682D-01 0.719D-02 0.694D-01			0.131D 06 0.263D 05 0.158D 06	0.116D 06 0.235D 05 0.141D 06				
9	12	1/2-3/2 3/2-3/2 3/2-5/2	0.341D-03 0.341D-03 0.341D-03	-0.319D-03 -0.320D-03 -0.320D-03	0.415D-01 0.424D-02 0.381D-01	0.363D-01 0.374D-02 0.336D-01			0.907D 05 0.179D 05 0.107D 06	0.794D 05 0.158D 05 0.947D 05				
9	13	1/2-3/2 3/2-3/2 3/2-5/2	0.278D-03 0.277D-03 0.277D-03	-0.259D-03 -0.259D-03 -0.259D-03	0.254D-01 0.256D-02 0.230D-01	0.221D-01 0.224D-02 0.202D-01			0.654D 05 0.128D 05 0.767D 05	0.570D 05 0.112D 05 0.672D 05				
9	14	1/2-3/2 3/2-3/2 3/2-5/2	0.233D-03 0.231D-03 0.232D-03	-0.217D-03 -0.216D-03 -0.216D-03	0.169D-01 0.169D-02 0.152D-01	0.147D-01 0.147D-02 0.133D-01			0.489D 05 0.948D 04 0.570D 05	0.425D 05 0.824D 04 0.497D 05				
10	9	1/2-3/2 3/2-3/2 3/2-5/2	0.417D-05 0.105D-03 0.845D-04	0.261D-03 0.191D-03 0.210D-03	0.156D-03 0.134D-01 0.723D-01	0.613D 00 0.444D-01 0.443D 00			0.541D 00 0.501D 02 0.217D 03	0.212D 04 0.167D 03 0.133D 04				
10	10	1/2-3/2 3/2-3/2 3/2-5/2	0.755D-03 0.818D-03 0.805D-03	-0.769D-03 -0.847D-03 -0.832D-03	0.704D 00 0.658D-01 0.743D 00	0.730D 00 0.718D-01 0.794D 00			0.129D 06 0.292D 05 0.171D 06	0.134D 05 0.313D 05 0.183D 06				
10	11	1/2-3/2 3/2-3/2 3/2-5/2	0.487D-03 0.507D-03 0.503D-03	-0.471D-03 -0.494D-03 -0.490D-03	0.186D 00 0.206D-01 0.182D 00	0.174D 00 0.195D-01 0.173D 00			0.847D 05 0.179D 05 0.106D 06	0.793D 05 0.170D 05 0.101D 06				
10	12	1/2-3/2 3/2-3/2 3/2-5/2	0.361D-03 0.369D-03 0.368D-03	-0.341D-03 -0.351D-03 -0.349D-03	0.811D-01 0.865D-02 0.771D-01	0.723D-01 0.781D-02 0.696D-01			0.584D 05 0.120D 05 0.717D 05	0.521D 05 0.109D 05 0.646D 05				
10	13	1/2-3/2 3/2-3/2 3/2-5/2	0.286D-03 0.290D-03 0.289D-03	-0.270D-03 -0.275D-03 -0.274D-03	0.443D-01 0.464D-02 0.414D-01	0.394D-01 0.415D-02 0.371D-01			0.421D 05 0.854D 04 0.509D 05	0.375D 05 0.765D 04 0.457D 05				
10	14	1/2-3/2 3/2-3/2 3/2-5/2	0.236D-03 0.238D-03 0.237D-03	-0.223D-03 -0.225D-03 -0.225D-03	0.275D-01 0.284D-02 0.254D-01	0.245D-01 0.234D-02 0.227D-01			0.315D 05 0.632D 04 0.377D 05	0.280D 05 0.564D 04 0.338D 05				
11	10	1/2-3/2 3/2-3/2 3/2-5/2	0.131D-04 0.993D-04 0.820D-04	0.227D-03 0.165D-03 0.182D-03	0.242D-02 0.191D-01 0.106D 00	0.725D 00 0.524D-01 0.521D 00			0.340D 01 0.128D 03 0.128D 03	0.102D 04 0.786D 02 0.629D 03				
11	11	1/2-3/2 3/2-3/2 3/2-5/2	0.628D-03 0.683D-03 0.672D-03	-0.644D-03 -0.712D-03 -0.700D-03	0.732D 00 0.899D-01 0.776D 00	0.771D 00 0.978D-01 0.843D 00			0.593D 05 0.147D 05 0.791D 05	0.625D 05 0.147D 05 0.859D 05				
11	12	1/2-3/2 3/2-3/2 3/2-5/2	0.405D-03 0.429D-03 0.425D-03	-0.401D-03 -0.418D-03 -0.418D-03	0.193D 00 0.217D-01 0.191D 00	0.186D 00 0.210D-01 0.185D 00			0.406D 05 0.873D 04 0.514D 05	0.390D 05 0.842D 04 0.497D 05				
11	13	1/2-3/2 3/2-3/2 3/2-5/2	0.306D-03 0.316D-03 0.315D-03	-0.289D-03 -0.302D-03 -0.300D-03	0.849D-01 0.918D-02 0.819D-01	0.756D-01 0.839D-02 0.745D-01			0.290D 05 0.606D 04 0.362D 05	0.258D 05 0.554D 04 0.329D 05				
11	14	1/2-3/2 3/2-3/2 3/2-5/2	0.245D-03 0.252D-03 0.250D-03	-0.232D-03 -0.238D-03 -0.237D-03	0.469D-01 0.500D-02 0.443D-01	0.420D-01 0.447D-02 0.398D-01			0.216D 05 0.440D 04 0.265D 05	0.194D 05 0.401D 04 0.238D 05				

2

$np^2_j - md^2_{0j}$

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2_j - md^2_{0j}$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION J-J'	M^0_{IJ}		f_{IJ}		STONE	AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH			VELOCITY	LENGTH
12	11	1/2-3/2 3/2-3/2 3/2-5/2	0.157D-04 0.914D-04 0.755D-04	0.200D-03 0.146D-03 0.160D-03	0.514D-02 0.240D-01 0.134D 00	0.837D 00 0.613D-01 0.600D 00			0.329D 01 0.163D 02 0.735D 02	0.535D 03 0.416D 02 0.330D 03
12	12	1/2-3/2 3/2-3/2 3/2-5/2	0.536D-03 0.585D-03 0.575D-03	-0.553D-03 -0.613D-03 -0.602D-03	0.764D 00 0.944D-01 0.815D 00	0.814D 00 0.104D 00 0.893D 00			0.302D 05 0.695D 04 0.406D 05	0.322D 05 0.764D 04 0.444D 05
12	13	1/2-3/2 3/2-3/2 3/2-5/2	0.352D-03 0.372D-03 0.366D-03	-0.351D-03 -0.371D-03 -0.367D-03	0.202D 00 0.230D-01 0.200D 00	0.200D 00 0.229D-01 0.201D 00			0.213D 05 0.466D 04 0.272D 05	0.212D 05 0.464D 04 0.274D 05
12	14	1/2-3/2 3/2-3/2 3/2-5/2	0.266D-03 0.275D-03 0.275D-03	-0.248D-03 -0.260D-03 -0.258D-03	0.890D-01 0.966D-02 0.869D-01	0.776D-01 0.866D-02 0.766D-01			0.157D 05 0.330D 04 0.199D 05	0.137D 05 0.296D 04 0.175D 03
13	12	1/2-3/2 3/2-3/2 3/2-5/2	0.510D-03 0.557D-03 0.266D-03	0.158D-04 0.7C5D-05 0.883D-05	0.769D 01 0.373D 01 0.235D 01	0.734D-02 0.202D-03 0.259D-02			0.247D 04 0.126D 04 0.646D 03	0.235D 01 0.683D-01 0.709D 00
13	13	1/2-3/2 3/2-3/2 3/2-5/2	0.508D-03 0.923D-03 0.259D-03	-0.339D-04 -0.714D-04 -0.610D-04	0.946D 00 0.323D 00 0.228D 00	0.421D-02 0.193D-02 0.126D-01			0.197D 05 0.126D 05 0.599D 04	0.878D 02 0.752D 02 0.332D 03
13	14	1/2-3/2 3/2-3/2 3/2-5/2	0.460D-03 0.819D-03 0.232D-03	-0.148D-03 -0.197D-03 -0.184D-03	0.468D 00 0.151D 00 0.109D 00	0.487D-01 0.879D-02 0.684D-01			0.268D 05 0.166D 05 0.802D 04	0.279D 04 0.966D 03 0.504D 04
14	13	1/2-3/2 3/2-3/2 3/2-5/2	0.106D-03 0.272D-03 0.700D-04	0.524D-04 0.362D-04 0.399D-04	0.446D 00 0.409D 00 0.221D 00	0.110D 00 0.723D-02 0.717D-01			0.780D 02 0.748D 02 0.328D 02	0.192D 02 0.132D 01 0.106D 02
14	14	1/2-3/2 3/2-3/2 3/2-5/2	0.221D-03 0.457D-03 0.122D-03	0.238D-03 0.204D-03 0.213D-03	0.239D 00 0.106D 00 0.675D-01	0.277D 00 0.211D-01 0.206D 00			0.280D 04 0.230D 04 0.998D 03	0.324D 04 0.461D 03 0.304D 04
15	14	1/2-3/2 3/2-3/2 3/2-5/2	-0.246D-03 -0.365D-03 -0.105D-03	0.558D-04 0.411D-04 0.456D-04	0.321D 01 0.977D 00 0.664D 00	0.166D 00 0.124D-01 0.124D 00			0.317D 03 0.101D 03 0.563D 02	0.164D 02 0.129D 01 0.105D 02

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2F_j - md^2D_j$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION	M_{ij}^0			f_{ij}			f_{ij}			$A_{ij}(\text{sec}^{-1})$		
			VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW
4	7	5/2-3/2	-0.487D-03	0.953D-03		0.828D-01	0.317D 00		0.206D 06	0.786D 06		0.206D 06	0.786D 06	
		5/2-5/2	-0.497D-03	0.958D-03		0.607D-02	0.226D-01		0.103D 05	0.383D 05		0.103D 05	0.383D 05	
		7/2-5/2	-0.496D-03	0.962D-03		0.906D-01	0.341D 00		0.205D 06	0.774D 06		0.205D 06	0.774D 06	
4	8	5/2-3/2	-0.580D-04	0.806D-04		0.554D-03	0.107D-02		0.618D 04	0.119D 05		0.618D 04	0.119D 05	
		5/2-5/2	-0.667D-04	0.899D-04		0.521D-04	0.946D-04		0.390D 03	0.708D 03		0.390D 03	0.708D 03	
		7/2-5/2	-0.653D-04	0.886D-04		0.750D-03	0.138D-02		0.749D 04	0.138D 05		0.749D 04	0.138D 05	
4	9	5/2-3/2	-0.115D-04	0.319D-04		0.165D-04	0.128D-03		0.314D 03	0.243D 04		0.314D 03	0.243D 04	
		5/2-5/2	-0.167D-04	0.372D-04		0.231D-05	0.1125D-04		0.319D 03	0.158D 03		0.319D 03	0.158D 03	
		7/2-5/2	-0.159D-04	0.365D-04		0.341D-04	0.180D-03		0.577D 02	0.305D 04		0.577D 02	0.305D 04	
4	10	5/2-3/2	0.862D-06	0.184D-04		0.819D-07	0.372D-04		0.205D 01	0.930D 03		0.205D 01	0.930D 03	
		5/2-5/2	-0.275D-05	0.221D-04		0.592D-07	0.382D-05		0.987D 00	0.638D 02		0.987D 00	0.638D 02	
		7/2-5/2	-0.223D-05	0.215D-04		0.584D-06	0.542D-04		0.130D 02	0.121D 04		0.130D 02	0.121D 04	
4	11	5/2-3/2	0.457D-05	0.126D-04		0.212D-05	0.160D-04		0.623D 02	0.470D 03		0.623D 02	0.470D 03	
		5/2-5/2	0.179D-05	0.153D-04		0.230D-07	0.170D-05		0.453D 00	0.34D 02		0.453D 00	0.34D 02	
		7/2-5/2	0.220D-05	0.149D-04		0.524D-06	0.124D-04		0.137D 02	0.632D 03		0.137D 02	0.632D 03	
4	12	5/2-3/2	0.553D-05	0.947D-05		0.293D-05	0.861D-05		0.962D 02	0.282D 03		0.962D 02	0.282D 03	
		5/2-5/2	0.331D-05	0.117D-04		0.753D-07	0.936D-06		0.165D 01	0.205D 02		0.165D 01	0.205D 02	
		7/2-5/2	0.365D-05	0.114D-04		0.137D-05	0.133D-04		0.400D 02	0.389D 03		0.400D 02	0.389D 03	
4	13	5/2-3/2	0.554D-05	0.760D-05		0.284D-05	0.534D-05		0.100D 03	0.189D 03		0.100D 03	0.189D 03	
		5/2-5/2	0.368D-05	0.944D-05		0.895D-07	0.588D-06		0.211D 01	0.139D 02		0.211D 01	0.139D 02	
		7/2-5/2	0.396D-05	0.921D-05		0.156D-05	0.840D-05		0.489D 02	0.264D 03		0.489D 02	0.264D 03	
4	14	5/2-3/2	0.519D-05	0.636D-05		0.243D-05	0.364D-05		0.906D 02	0.136D 03		0.906D 02	0.136D 03	
		5/2-5/2	0.360D-05	0.792D-05		0.832D-07	0.403D-06		0.207D 01	0.100D 02		0.207D 01	0.100D 02	
		7/2-5/2	0.384D-05	0.773D-05		0.142D-05	0.576D-05		0.471D 02	0.191D 03		0.471D 02	0.191D 03	
5	8	5/2-3/2	-0.518D-03	0.905D-03		0.176D 00	0.537D 00		0.124D 06	0.378D 06		0.124D 06	0.378D 06	
		5/2-5/2	-0.528D-03	0.910D-03		0.129D-01	0.382D-01		0.622D 04	0.185D 05		0.622D 04	0.185D 05	
		7/2-5/2	-0.524D-03	0.923D-03		0.190D 00	0.589D 00		0.122D 06	0.380D 06		0.122D 06	0.380D 06	
5	9	5/2-3/2	-0.823D-04	0.110D-03		0.200D-02	0.358D-02		0.691D 04	0.123D 05		0.691D 04	0.123D 05	
		5/2-5/2	-0.910D-04	0.119D-03		0.174D-03	0.299D-03		0.404D 03	0.693D 03		0.404D 03	0.693D 03	
		7/2-5/2	-0.857D-04	0.114D-03		0.232D-02	0.412D-02		0.716D 04	0.127D 05		0.716D 04	0.127D 05	
5	10	5/2-3/2	-0.367D-04	0.554D-04		0.297D-03	0.675D-03		0.185D 04	0.421D 04		0.185D 04	0.421D 04	
		5/2-5/2	-0.423D-04	0.609D-04		0.281D-04	0.581D-04		0.117D 03	0.243D 04		0.117D 03	0.243D 04	
		7/2-5/2	-0.388D-04	0.579D-04		0.355D-03	0.787D-03		0.158D 04	0.439D 04		0.158D 04	0.439D 04	
5	11	5/2-3/2	-0.210D-04	0.371D-04		0.830D-04	0.258D-03		0.710D 03	0.221D 04		0.710D 03	0.221D 04	
		5/2-5/2	-0.250D-04	0.410D-04		0.838D-05	0.225D-04		0.479D 02	0.129D 03		0.479D 02	0.129D 03	
		7/2-5/2	-0.226D-04	0.385D-04		0.103D-03	0.299D-03		0.786D 03	0.228D 04		0.786D 03	0.228D 04	
5	12	5/2-3/2	-0.138D-04	0.275D-04		0.327D-04	0.129D-03		0.340D 03	0.134D 04		0.340D 03	0.134D 04	
		5/2-5/2	-0.170D-04	0.365D-04		0.351D-05	0.113D-04		0.244D 02	0.785D 02		0.244D 02	0.785D 02	
		7/2-5/2	-0.151D-04	0.287D-04		0.416D-04	0.150D-03		0.385D 03	0.139D 04		0.385D 03	0.139D 04	
5	13	5/2-3/2	-0.100D-04	0.217D-04		0.159D-04	0.754D-04		0.189D 03	0.895D 03		0.189D 03	0.895D 03	
		5/2-5/2	-0.126D-04	0.242D-04		0.181D-05	0.665D-05		0.143D 02	0.527D 02		0.143D 02	0.527D 02	
		7/2-5/2	-0.110D-04	0.228D-04		0.208D-04	0.891D-04		0.220D 03	0.941D 03		0.220D 03	0.941D 03	
5	14	5/2-3/2	-0.773D-05	0.180D-04		0.910D-05	0.492D-04		0.119D 03	0.641D 03		0.119D 03	0.641D 03	
		5/2-5/2	-0.996D-05	0.200D-04		0.108D-05	0.435D-05		0.937D 01	0.378D 02		0.937D 01	0.378D 02	
		7/2-5/2	-0.862D-05	0.189D-04		0.121D-04	0.585D-04		0.141D 03	0.678D 03		0.141D 03	0.678D 03	

$\pi^2 f_j - m^2 D_j$

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2 f_j - m^2 D_j$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	H	TRANSITION	R_{ij}^0			f_{ij}			f_{ij}			$A_{ij}(\text{sec}^{-1})$		
			VELOCITY	LENGTH		VELOCITY	LENGTH		STONE	AGNEW	VELOCITY	LENGTH		
6	9	5/2-3/2 5/2-5/2 7/2-5/2	-0.431D-03 -0.443D-03 -0.476D-03	0.936D-03 0.943D-03 0.902D-03		0.204D 00 0.152D-01 0.264D 00	0.365D 00 0.699D-01 0.746D 00			0.510D 05 0.260D 04 0.601D 05	0.241D 06 0.118D 03 0.216D 06			
6	10	5/2-3/2 5/2-5/2 7/2-5/2	-0.342D-04 -0.436D-04 -0.733D-04	0.545D-04 0.654D-04 0.975D-04		0.565D-03 0.653D-04 0.277D-02	0.143D-02 0.147D-03 0.489D-02			0.734D 03 0.570D 02 0.322D 04	0.186D 04 0.128D 03 0.569D 04			
6	11	5/2-3/2 5/2-5/2 7/2-5/2	-0.131D-04 -0.192D-04 -0.376D-04	0.253D-04 0.316D-04 0.513D-04		0.602D-04 0.924D-05 0.529D-03	0.224D-03 0.249D-04 0.988D-03			0.148D 03 0.152D 02 0.116D 04	0.550D 03 0.409D 02 0.217D 04			
6	12	5/2-3/2 5/2-5/2 7/2-5/2	0.723D-04 -0.124D-03 -0.249D-04	-0.630D-03 -0.628D-03 0.359D-04		0.154D-02 0.321D-03 0.190D-03	0.117D 00 0.827D-02 0.405D-03			0.537D 04 0.748D 03 0.590D 03	0.407D 06 0.193D 05 0.126D 04			
6	13	5/2-3/2 5/2-5/2 7/2-5/2	0.467D-04 -0.631D-04 -0.180D-04	-0.570D-03 -0.569D-03 0.273D-04		0.574D-03 0.130D-03 0.914D-04	0.856D-01 0.609D-02 0.210D-03			0.250D 04 0.378D 03 0.354D 03	0.373D 06 0.177D 05 0.816D 03			
6	14	5/2-3/2 5/2-5/2 7/2-5/2	0.314D-04 -0.586D-04 -0.142D-04	-0.513D-03 -0.512D-03 0.221D-04		0.241D-03 0.599D-04 0.525D-04	0.642D-01 0.458D-02 0.127D-03			0.122D 04 0.202D 03 0.237D 03	0.326D 06 0.155D 05 0.575D 03			
7	10	5/2-3/2 5/2-5/2 7/2-5/2	-0.563D-04 -0.859D-04 -0.450D-03	0.328D-04 0.291D-04 0.824D-03		0.159D-01 0.891D-03 0.367D 00	0.185D-02 0.102D-03 0.123D 01			0.163D 04 0.628D 02 0.345D 05	0.190D 03 0.719D 01 0.115D 06			
7	11	5/2-3/2 5/2-5/2 7/2-5/2	-0.645D-04 -0.582D-04 -0.796D-04	0.320D-04 0.286D-04 0.100D-03		0.305D-02 0.177D-03 0.497D-02	0.753D-03 0.427D-04 0.787D-02			0.171D 04 0.667D 02 0.250D 04	0.422D 03 0.161D 02 0.396D 04			
7	12	5/2-3/2 5/2-5/2 7/2-5/2	-0.479D-04 -0.439D-04 -0.448D-04	0.248D-04 0.221D-04 0.548D-04		0.120D-02 0.720D-04 0.112D-02	0.323D-03 0.182D-04 0.168D-02			0.132D 04 0.531D 02 0.110D 04	0.355D 03 0.134D 02 0.166D 04			
7	13	5/2-3/2 5/2-5/2 7/2-5/2	-0.375D-04 -0.343D-04 -0.313D-04	0.205D-04 0.183D-04 0.401D-04		0.611D-03 0.364D-04 0.456D-03	0.183D-03 0.103D-04 0.745D-03			0.982D 03 0.351D 02 0.653D 03	0.294D 03 0.111D 02 0.107D 04			
7	14	5/2-3/2 5/2-5/2 7/2-5/2	-0.306D-04 -0.278D-04 -0.241D-04	0.174D-04 0.155D-04 0.310D-04		0.360D-03 0.212D-04 0.239D-03	0.116D-03 0.650D-05 0.395D-03			0.738D 03 0.290D 02 0.438D 03	0.238D 03 0.897D 01 0.722D 03			
8	11	5/2-3/2 5/2-5/2 7/2-5/2	-0.126D-04 -0.303D-05 -0.210D-04	0.375D-05 0.313D-06 0.690D-05		0.400D-03 0.163D-05 0.118D-02	0.356D-04 0.174D-07 0.127D-03			0.190D 02 0.531D-01 0.512D 02	0.169D 01 0.566D-03 0.551D 01			
8	12	5/2-3/2 5/2-5/2 7/2-5/2	-0.117D-04 -0.523D-05 -0.173D-04	0.656D-05 0.332D-06 0.878D-05		0.146D-03 0.206D-05 0.340D-03	0.456D-04 0.330D-06 0.872D-04			0.353D 02 0.372D 00 0.620D 02	0.122D 02 0.150D 00 0.210D 02			
8	13	5/2-3/2 5/2-5/2 7/2-5/2	-0.563D-05 -0.609D-05 -0.132D-04	0.282D-05 0.150D-06 0.594D-05		0.691D-04 0.197D-05 0.138D-03	0.592D-05 0.120D-08 0.281D-04			0.376D 02 0.717D 00 0.670D 02	0.322D 01 0.436D-03 0.136D 02			
8	14	5/2-3/2 5/2-5/2 7/2-5/2	-0.748D-05 -0.424D-05 -0.103D-04	0.256D-05 0.310D-06 0.456D-05		0.341D-04 0.781D-06 0.696D-04	0.401D-05 0.418D-08 0.160D-04			0.277D 02 0.425D 00 0.505D 02	0.326D 01 0.227D-02 0.116D 02			

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nf^2F_j - md^2D_j$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION		M_{ij}^e		f_{ij}		f_{ij}		f_{ij}		$A_{ij}(\text{sec}^{-1})$	
		J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH	
9	12	5/2-3/2	-0.397D-03	-0.138D-04	0.560D 00	0.690D-03	0.890D-02	0.134D 05	0.162D 02			0.147D 04	0.139D 01
		5/2-5/2	0.599D-03	-0.184D-04	0.897D-01	0.347D-04	0.190D-02	0.516D 05	0.183D 02			0.226D 05	0.890D 02
		7/2-5/2	-0.793D-03	-0.149D-04	0.236D 01	0.838D-03	0.336D-02	0.862D 05	0.246D 04	0.156D 01			0.246D 04
9	13	5/2-3/2	-0.332D-03	0.208D-04	0.162D 00	0.638D-03	0.109D 04	0.233D 05	0.109D 04			0.226D 05	0.890D 02
		5/2-5/2	0.500D-03	0.126D-04	0.262D-01	0.166D-04	0.166D-04	0.862D 05	0.156D 01			0.246D 04	0.156D 01
		7/2-5/2	-0.662D-03	0.198D-04	0.688D 00	0.615D-03	0.336D-02	0.862D 05	0.246D 04	0.156D 01			0.246D 04
9	14	5/2-3/2	-0.281D-03	0.607D-04	0.806D-01	0.376D-02	0.233D 05	0.233D 05	0.109D 04			0.253D 04	0.368D 02
		5/2-5/2	0.423D-03	0.511D-04	0.130D-01	0.190D-03	0.190D-03	0.862D 05	0.109D 04			0.253D 04	0.368D 02
		7/2-5/2	-0.560D-03	0.602D-04	0.343D 00	0.336D-02	0.336D-02	0.862D 05	0.109D 04			0.253D 04	0.368D 02
10	13	5/2-3/2	-0.261D-03	-0.113D-03	0.329D 00	0.620D-01	0.425D 04	0.425D 04	0.801D 03			0.456D 05	0.417D 02
		5/2-5/2	0.388D-03	-0.117D-03	0.513D-01	0.469D-02	0.469D-02	0.163D 05	0.815D 03			0.456D 05	0.417D 02
		7/2-5/2	-0.520D-03	-0.116D-03	0.138D 01	0.689D-01	0.689D-01	0.163D 05	0.815D 03			0.456D 05	0.417D 02
10	14	5/2-3/2	-0.246D-03	-0.182D-03	0.119D 00	0.653D-01	0.924D 04	0.924D 04	0.505D 04			0.924D 04	0.505D 04
		5/2-5/2	0.366D-03	-0.188D-03	0.189D-01	0.499D-02	0.499D-02	0.352D 05	0.502D 04			0.924D 04	0.505D 04
		7/2-5/2	-0.490D-03	-0.185D-03	0.506D 00	0.723D-01	0.723D-01	0.352D 05	0.502D 04			0.924D 04	0.505D 04
11	14	5/2-3/2	-0.675D-04	-0.146D-03	0.292D-01	0.136D 00	0.215D 03	0.215D 03	0.100D 04			0.215D 03	0.100D 04
		5/2-5/2	0.977D-04	-0.148D-03	0.430D-02	0.983D-02	0.983D-02	0.215D 03	0.100D 04			0.215D 03	0.100D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2 \rightarrow mf^2$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION J-J'	M_{IJ}^0			f_{IJ}			f_{IJ}			f_{IJ}		
			VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW	STONE	AGNEW	VELOCITY	LENGTH	AGNEW	
5	4	3/2-5/2	-0.219D-02	0.214D-02	0.379D 00	0.395D 00	0.379D 00	0.302D 00	0.175D 08	0.167D 09		0.175D 08	0.167D 09	
		5/2-5/2	-0.221D-02	0.219D-02	0.194D-01	0.194D-01	0.191D-01	0.324D 00	0.126D 07	0.124D 07		0.126D 07	0.124D 07	
		5/2-7/2	-0.219D-02	0.218D-02	0.379D 00	0.382D 00	0.379D 00	0.186D 08	0.186D 08		0.186D 08	0.185D 08		
5	5	3/2-5/2	-0.185D-02	0.157D-02	0.162D 00	0.227D 00	0.162D 00	0.122D 00	0.157D 08	0.112D 08		0.157D 08	0.112D 08	
		5/2-5/2	-0.184D-02	0.156D-02	0.107D-01	0.107D-01	0.792D-02	0.110D 07	0.110D 07		0.110D 07	0.809D 06		
		5/2-7/2	-0.181D-02	0.157D-02	0.208D 00	0.208D 00	0.156D 00	0.127D 00	0.160D 08		0.160D 08	0.120D 06		
5	6	3/2-5/2	-0.121D-02	0.109D-02	0.876D-01	0.876D-01	0.707D-01	0.627D-01	0.745D 07	0.601D 07		0.745D 07	0.601D 07	
		5/2-5/2	-0.110D-02	0.110D-02	0.416D-02	0.416D-02	0.344D-02	0.650D-01	0.524D 06	0.433D 06		0.524D 06	0.433D 06	
		5/2-7/2	-0.133D-02	0.114D-02	0.102D 00	0.102D 00	0.739D-01	0.373D-01	0.957D 07	0.637D 07		0.957D 07	0.637D 07	
5	7	3/2-5/2	0.608D-03	-0.860D-03	0.208D-01	0.208D-01	0.416D-01	0.373D-01	0.198D 07	0.397D 07		0.198D 07	0.397D 07	
		5/2-5/2	0.629D-03	-0.879D-03	0.107D-02	0.107D-02	0.209D-02	0.383D-01	0.151D 06	0.294D 06		0.151D 06	0.294D 06	
		5/2-7/2	-0.109D-02	0.890D-03	0.642D-01	0.642D-01	0.427D-01	0.680D 07	0.452D 07		0.680D 07	0.452D 07		
5	8	3/2-5/2	0.589D-03	-0.672D-03	0.188D-01	0.188D-01	0.245D-02	0.235D-01	0.193D 07	0.251D 07		0.193D 07	0.251D 07	
		5/2-5/2	0.599D-03	-0.682D-03	0.934D-03	0.934D-03	0.121D-02	0.241D-01	0.142D 06	0.183D 06		0.142D 06	0.183D 06	
		5/2-7/2	0.548D-03	-0.663D-03	0.157D-01	0.157D-01	0.229D-01	0.178D 07	0.260D 07		0.178D 07	0.260D 07		
5	9	3/2-5/2	0.502D-03	-0.528D-03	0.134D-01	0.134D-01	0.148D-01	0.160D-01	0.144D 07	0.159D 07		0.144D 07	0.159D 07	
		5/2-5/2	0.509D-03	-0.534D-03	0.657D-03	0.657D-03	0.724D-03	0.164D-01	0.105D 06	0.115D 06		0.105D 06	0.115D 06	
		5/2-7/2	0.480D-03	-0.524D-03	0.117D-01	0.117D-01	0.140D-01	0.140D 07	0.167D 07		0.140D 07	0.167D 07		
5	10	3/2-5/2	0.427D-03	-0.429D-03	0.952D-02	0.952D-02	0.960D-02	0.114D-01	0.106D 07	0.107D 07		0.106D 07	0.107D 07	
		5/2-5/2	0.432D-03	-0.433D-03	0.466D-03	0.466D-03	0.848D-03	0.766D 05	0.770D 05		0.766D 05	0.770D 05		
		5/2-7/2	0.412D-03	-0.427D-03	0.847D-02	0.847D-02	0.910D-02	0.104D 07	0.112D 07		0.104D 07	0.112D 07		
5	11	3/2-5/2	0.368D-03	-0.359D-03	0.696D-02	0.696D-02	0.664D-02	0.792D 06	0.752D 06		0.792D 06	0.752D 06		
		5/2-5/2	0.371D-03	-0.362D-03	0.340D-03	0.340D-03	0.323D-03	0.572D 05	0.544D 05		0.572D 05	0.544D 05		
		5/2-7/2	0.226D-02	-0.121D-02	0.223D 01	0.223D 01	0.637D 00	0.352D 07	0.101D 07		0.352D 07	0.101D 07		
6	4	3/2-5/2	0.224D-02	-0.119D-02	0.107D 00	0.107D 00	0.301D-01	0.115D-01	0.241D 06	0.680D 05		0.241D 06	0.680D 05	
		5/2-5/2	0.224D-02	-0.119D-02	0.215D 01	0.215D 01	0.603D 00	0.364D 07	0.102D 07		0.364D 07	0.102D 07		
		5/2-7/2	0.293D-03	-0.233D-03	0.161D-01	0.161D-01	0.102D-01	0.138D 06	0.869D 05		0.138D 06	0.869D 05		
6	5	3/2-5/2	0.257D-03	-0.207D-03	0.598D-03	0.598D-03	0.389D-03	0.750D 04	0.890D 04		0.750D 04	0.890D 04		
		5/2-5/2	0.280D-03	-0.224D-03	0.142D-01	0.142D-01	0.908D-02	0.134D 06	0.855D 05		0.134D 06	0.855D 05		
		5/2-7/2	0.125D-03	-0.103D-03	0.225D-02	0.225D-02	0.151D-02	0.329D 05	0.222D 05		0.329D 05	0.222D 05		
6	6	3/2-5/2	0.595D-04	-0.833D-04	0.682D-04	0.682D-04	0.478D-04	0.104D 04	0.104D 04		0.104D 04	0.104D 04		
		5/2-5/2	0.123D-04	-0.210D-04	0.207D-04	0.207D-04	0.607D-04	0.336D 03	0.986D 03		0.336D 03	0.986D 03		
		5/2-7/2	-0.435D-03	0.365D-03	0.238D-01	0.238D-01	0.168D-01	0.455D 06	0.321D 06		0.455D 06	0.321D 06		
6	7	3/2-5/2	-0.415D-03	0.367D-03	0.104D-02	0.104D-02	0.727D-03	0.254D 05	0.264D 05		0.254D 05	0.264D 05		
		5/2-5/2	-0.594D-04	0.687D-04	0.119D-02	0.119D-02	0.569D-03	0.253D 05	0.121D 05		0.253D 05	0.121D 05		
		5/2-7/2	-0.174D-03	0.907D-04	0.352D-02	0.352D-02	0.956D-03	0.788D 05	0.214D 05		0.788D 05	0.214D 05		
6	8	3/2-5/2	-0.156D-03	0.758D-04	0.136D-04	0.136D-04	0.320D-04	0.450D 04	0.106D 04		0.450D 04	0.106D 04		
		5/2-5/2	-0.187D-03	0.102D-03	0.389D-02	0.389D-02	0.117D-02	0.966D 05	0.289D 05		0.966D 05	0.289D 05		
		5/2-7/2	-0.817D-04	0.972D-05	0.738D-03	0.738D-03	0.105D-04	0.182D 05	0.258D 03		0.182D 05	0.258D 03		
6	9	3/2-5/2	-0.665D-04	-0.208D-05	0.237D-04	0.237D-04	0.229D-07	0.665D 03	0.840D 00		0.665D 03	0.840D 00		
		5/2-5/2	-0.650D-04	-0.125D-04	0.765D-03	0.765D-03	0.165D-04	0.210D 05	0.454D 03		0.210D 05	0.454D 03		
		5/2-7/2	-0.405D-04	-0.197D-04	0.176D-03	0.176D-03	0.413D-04	0.465D 04	0.109D 04		0.465D 04	0.109D 04		
6	10	3/2-5/2	-0.283D-04	-0.259D-04	0.409D-05	0.409D-05	0.438D-05	0.461D 03	0.172D 03		0.461D 03	0.172D 03		
		5/2-5/2	-0.410D-04	-0.195D-04	0.172D-03	0.172D-03	0.389D-04	0.506D 04	0.114D 04		0.506D 04	0.114D 04		
		5/2-7/2												

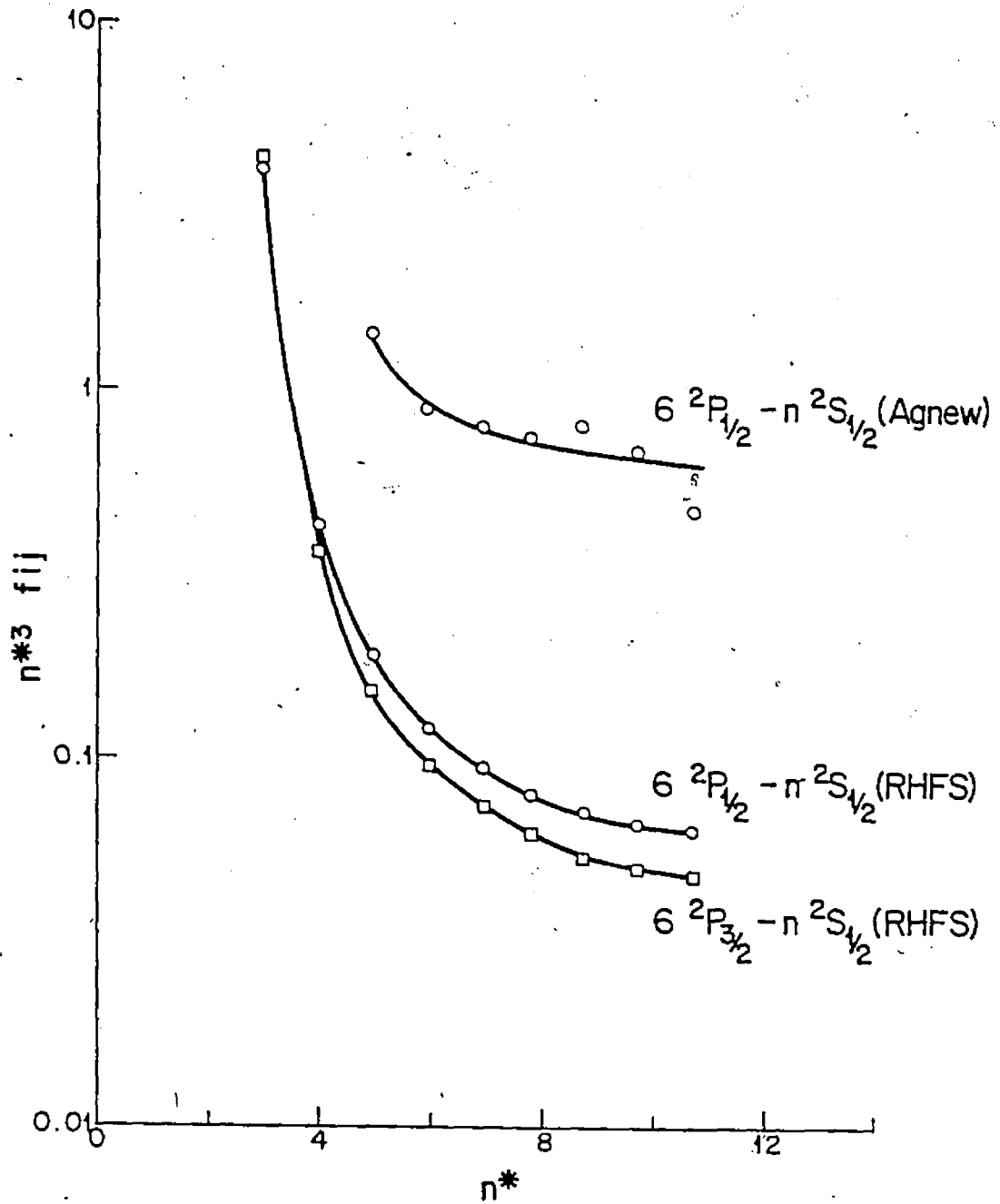
THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_j - mf^2F_{j1}$ CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

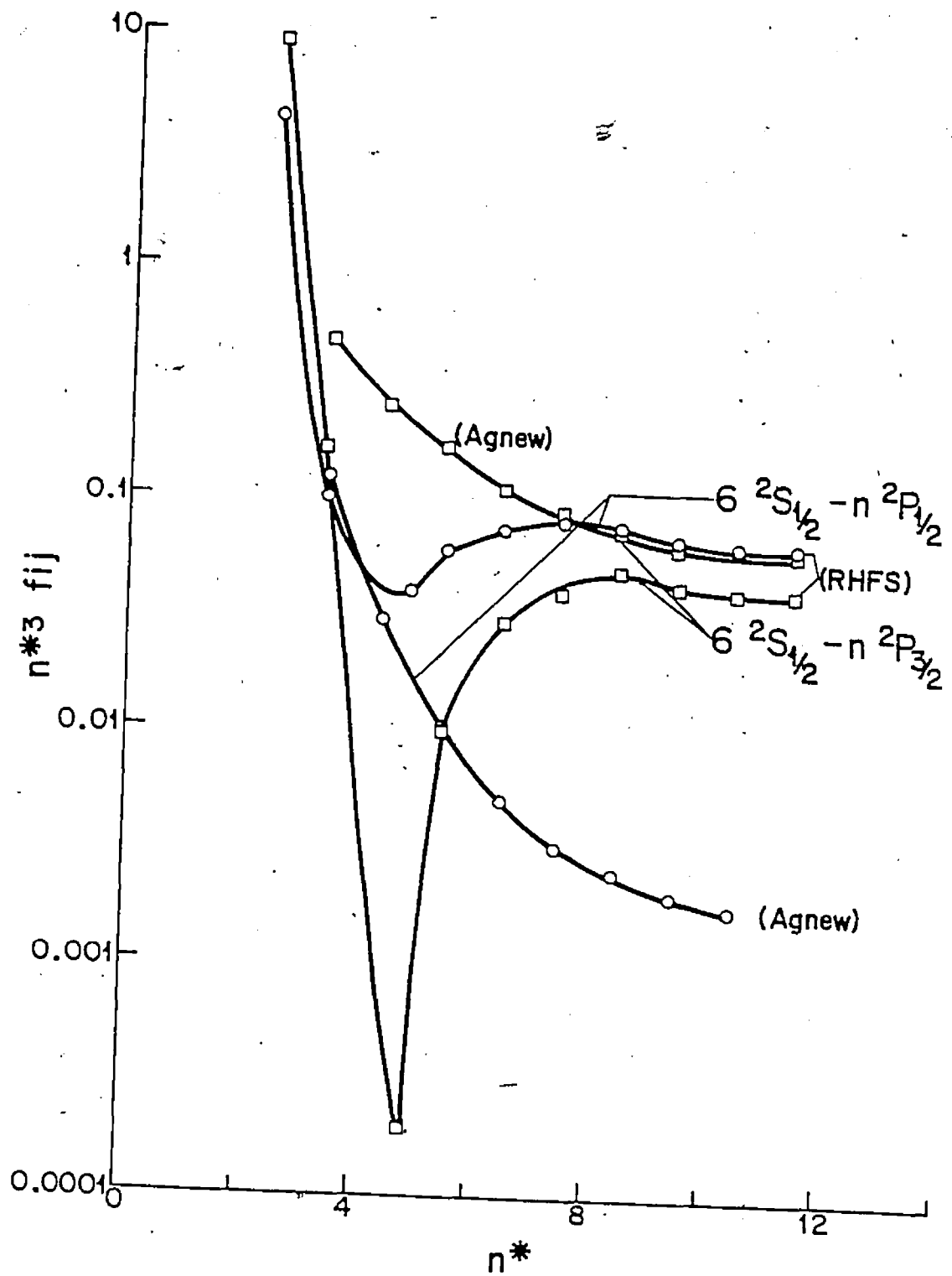
N	M	TRANSITION J-J'	f_{ij}^0		f_{ij}		STONE	f_{ij}	AGNEW	f_{ij}^1	
			VELOCITY	LENGTH	VELOCITY	LENGTH				VELOCITY	LENGTH
6	11	3/2-5/2 5/2-5/2	-0.204D-04 -0.598D-05	-0.302D-04 -0.382D-04	0.433D-04 0.497D-06	0.952D-04 0.729D-05				0.120D 04 0.205D 02	0.265D 04 0.301D 03
7	5	3/2-5/2 5/2-5/2 5/2-7/2	0.164D-02 0.163D-02 0.164D-02	-0.952D-03 -0.938D-03 -0.939D-03	0.240D 01 0.116D 00 0.234D 01	0.809D 00 0.383D-01 0.767D 00				0.910D 06 0.628D 05 0.955D 06	0.307D 06 0.208D 05 0.312D 06
7	6	3/2-5/2 5/2-5/2 5/2-7/2	0.568D-03 0.551D-03 0.484D-03	-0.427D-03 -0.414D-03 -0.367D-03	0.116D 00 0.528D-02 0.814D-01	0.658D-01 0.297D-02 0.467D-01				0.270D 06 0.180D 05 0.208D 06	0.152D 06 0.101D 05 0.119D 05
7	7	3/2-5/2 5/2-5/2 5/2-7/2	-0.761D-03 -0.750D-03 0.189D-03	0.750D-03 0.739D-03 -0.151D-03	0.154D 00 0.718D-02 0.911D-02	0.149D 00 0.695D-02 0.582D-02				0.659D 06 0.454D 05 0.432D 05	0.639D 06 0.440D 05 0.276D 05
7	8	3/2-5/2 5/2-5/2 5/2-7/2	-0.385D-03 -0.372D-03 -0.395D-03	0.329D-03 0.318D-03 0.340D-03	0.336D-01 0.151D-02 0.339D-01	0.246D-01 0.110D-02 0.251D-01				0.157D 06 0.131D 05 0.221D 06	0.144D 06 0.953D 04 0.164D 06
7	9	3/2-5/2 5/2-5/2 5/2-7/2	-0.229D-03 -0.219D-03 -0.233D-03	0.175D-03 0.166D-03 0.179D-03	0.109D-01 0.473D-03 0.107D-01	0.633D-02 0.271D-03 0.632D-02				0.770D 05 0.499D 04 0.847D 05	0.449D 05 0.285D 04 0.499D 05
7	10	3/2-5/2 5/2-5/2 5/2-7/2	-0.153D-03 -0.144D-03 -0.154D-03	0.106D-03 0.982D-04 0.107D-03	0.453D-02 0.193D-03 0.441D-02	0.218D-02 0.834D-04 0.214D-02				0.365D 05 0.230D 04 0.355D 05	0.175D 05 0.107D 04 0.192D 05
7	11	3/2-5/2 5/2-5/2	-0.111D-03 -0.103D-03	0.707D-04 0.641D-04	0.227D-02 0.944D-04	0.926D-03 0.365D-04				0.199D 05 0.123D 04	0.814D 04 0.476D 03
8	6	3/2-5/2 5/2-5/2 5/2-7/2	0.134D-02 0.134D-02 0.133D-02	-0.744D-03 -0.737D-03 -0.754D-03	0.287D 01 0.139D 00 0.274D 01	0.881D 00 0.420D-01 0.881D 00				0.343D 06 0.239D 05 0.351D 06	0.105D 06 0.720D 04 0.113D 06
8	7	3/2-5/2 5/2-5/2 5/2-7/2	-0.130D-02 -0.130D-02 0.470D-03	0.165D-02 0.167D-02 -0.365D-03	0.104D 01 0.504D-01 0.131D 00	0.169D 01 0.920D-01 0.791D-01				0.828D 06 0.590D 05 0.115D 06	0.134D 07 0.963D 05 0.695D 05
8	8	3/2-5/2 5/2-5/2 5/2-7/2	-0.618D-03 -0.608D-03 -0.623D-03	0.630D-03 0.621D-03 0.633D-03	0.168D 00 0.763D-02 0.164D 00	0.175D 00 0.814D-02 0.170D 00				0.261D 06 0.180D 05 0.283D 06	0.272D 06 0.187D 05 0.292D 06
8	9	3/2-5/2 5/2-5/2 5/2-7/2	-0.364D-03 -0.354D-03 -0.366D-03	0.335D-03 0.326D-03 0.337D-03	0.489D-01 0.222D-02 0.473D-01	0.914D-01 0.188D-02 0.401D-01				0.108D 06 0.730D 04 0.117D 06	0.915D 05 0.617D 04 0.987D 05
8	10	3/2-5/2 5/2-5/2 5/2-7/2	-0.245D-03 -0.237D-03 -0.245D-03	0.211D-03 0.205D-03 0.213D-03	0.190D-01 0.888D-03 0.190D-01	0.151D-01 0.668D-03 0.144D-01				0.548D 05 0.364D 04 0.585D 05	0.416D 05 0.274D 04 0.443D 05
8	11	3/2-5/2 5/2-5/2	-0.180D-03 -0.173D-03	0.151D-03 0.144D-03	0.994D-02 0.439D-03	0.701D-02 0.306D-03				0.318D 05 0.209D 04	0.224D 05 0.145D 04
9	7	3/2-5/2 5/2-5/2 5/2-7/2	-0.253D-03 -0.268D-03 0.110D-02	-0.680D-03 -0.662D-03 -0.627D-03	0.165D 00 0.902D-02 0.303D 01	0.119D 01 0.551D-01 0.389D 00				0.747D 04 0.586D 03 0.147D 06	0.541D 05 0.358D 04 0.482D 05

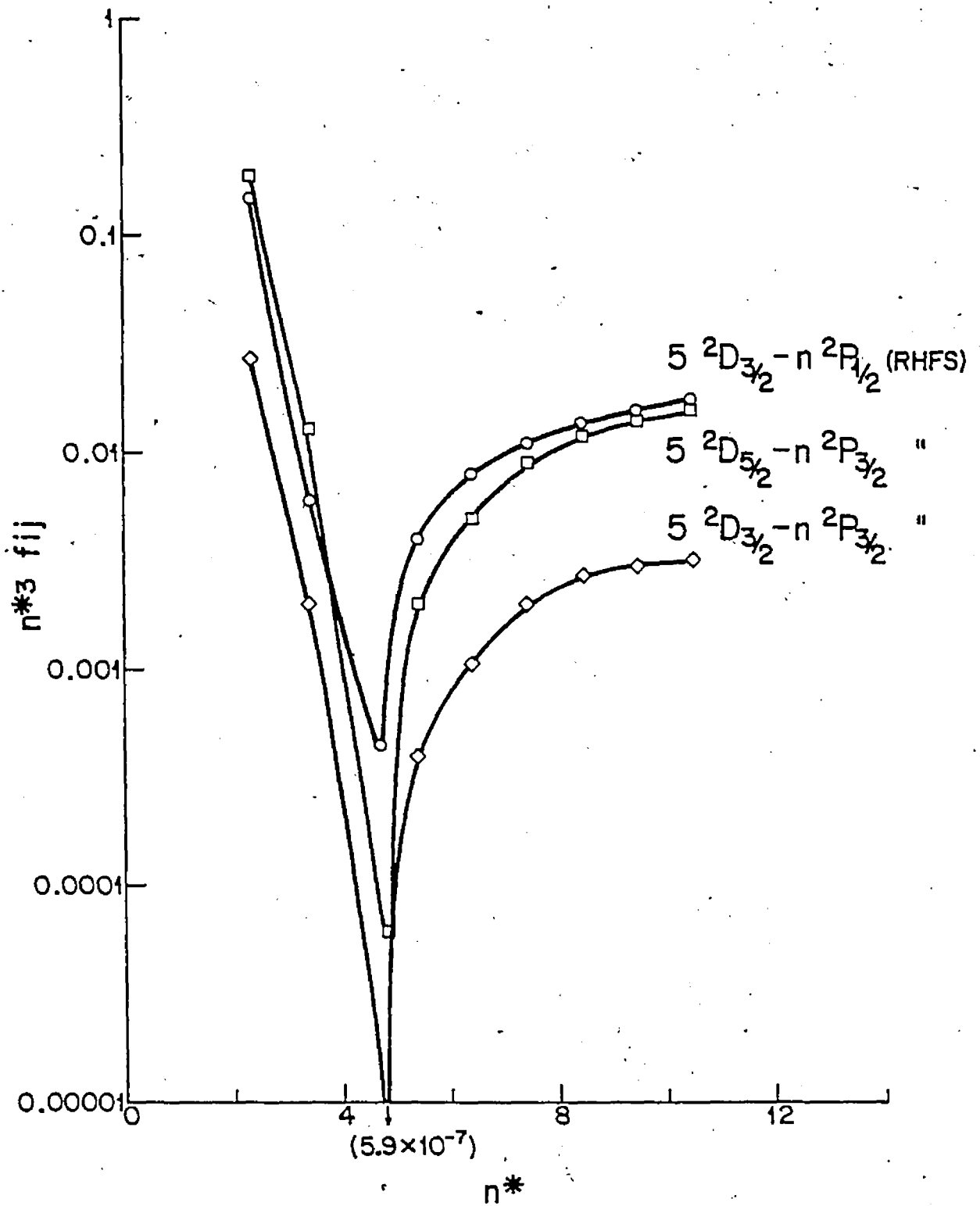
THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n\sigma^2D_J - m^2F_J'$, CALCULATED USING THE RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

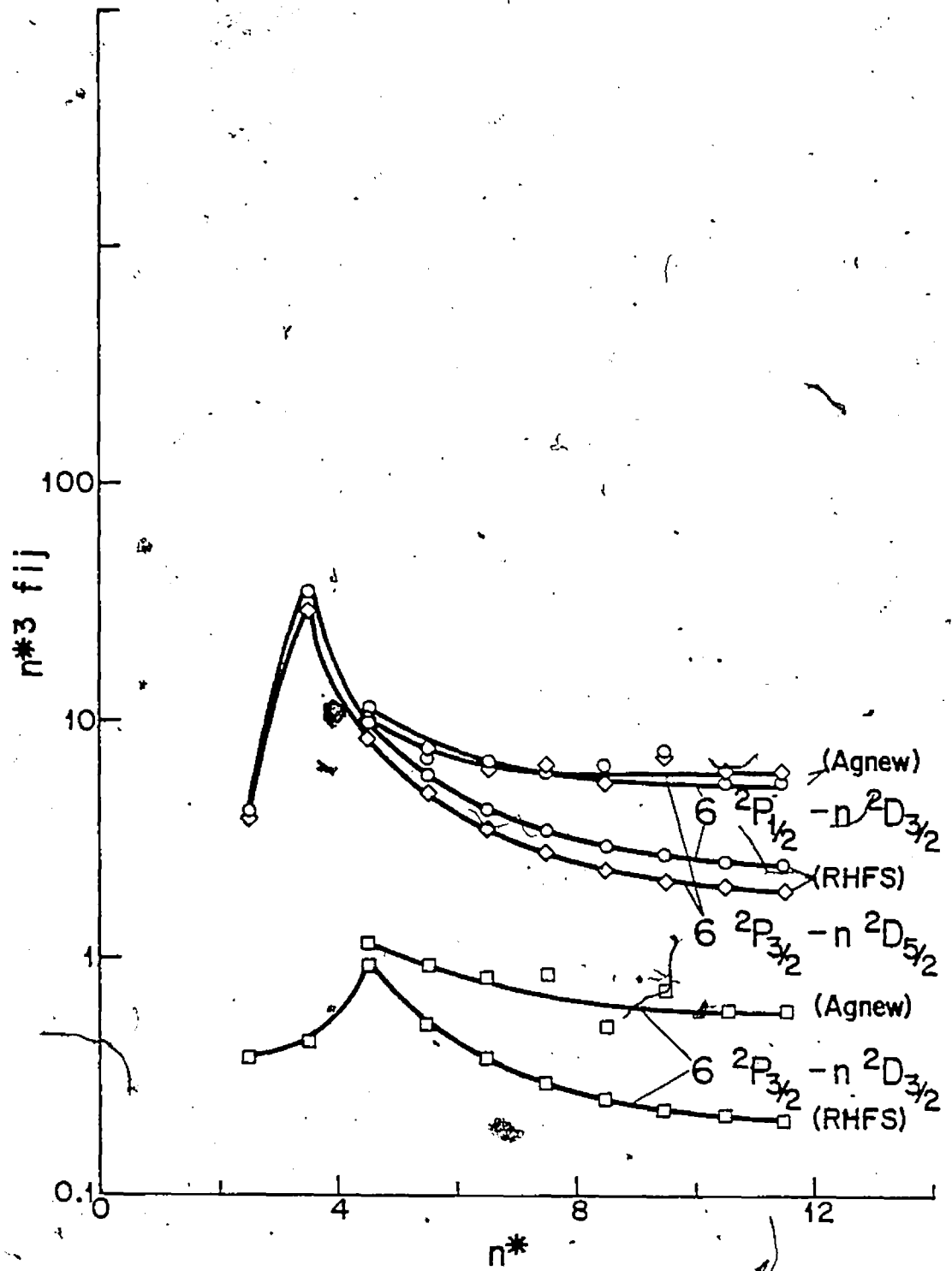
N	M	TRANSITION	H_{IJ}^0			f_{IJ}			f_{IJ}			f_{IJ}		
			VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW
9	8	3/2-5/2 5/2-7/2	-0.1100-02 -0.1100-02 -0.1090-02	0.1510-02 0.1520-02 0.1490-02		0.1170 01 0.5650-01 0.1110 01	0.2220 01 0.1080 00 0.2060 01		0.3760 06 0.2680 05 0.3950 06		0.7130 06 0.5120 05 0.7310 06			
9	9	3/2-5/2 5/2-7/2	-0.5330-03 -0.5270-03 -0.5340-03	0.5650-03 0.5580-03 0.5630-03		0.1930 00 0.9020-02 0.1850 00	0.2160 00 0.1010-01 0.2070 00		0.1260 06 0.8750 04 0.1350 06		0.1420 06 0.9820 04 0.1500 06			
9	10	3/2-5/2 5/2-5/2 5/2-7/2	-0.3370-03 -0.3300-03 -0.3370-03	0.3240-03 0.3180-03 0.3240-03		0.6370-01 0.2920-02 0.6090-01	0.5870-01 0.2710-02 0.5630-01		0.6150 05 0.4190 04 0.6550 05		0.5660 05 0.3880 04 0.6050 05			
9	11	3/2-5/2	-0.2410-03 -0.2350-03	0.2230-03 0.2170-03		0.2880-01 0.1310-02	0.2460-01 0.1110-02		0.3550 05 0.2400 04		0.3030 05 0.2040 04			
10	8	3/2-5/2 5/2-5/2 5/2-7/2	-0.3410-03 -0.3530-03 -0.2290-03	0.5990-03 0.5820-03 0.5910-03		0.4550 00 0.2380-01 0.4130 00	0.1410 01 0.6470-01 0.1330 01		0.8950 04 0.6710 03 0.8740 04		0.2770 05 0.1830 04 0.2820 05			
10	9	3/2-5/2 5/2-5/2 5/2-7/2	-0.9320-03 -0.5340-03 -0.9290-03	0.1340-02 0.1350-02 0.1330-02		0.1250 01 0.6020-01 0.1190 01	0.2590 00 0.1260 00 0.2430 01		0.1830 06 0.1300 05 0.1930 06		0.3790 06 0.2730 05 0.3950 06			
10	10	3/2-5/2 5/2-5/2 5/2-7/2	-0.4650-03 -0.4610-03 -0.4650-03	0.5100-03 0.5040-03 0.5070-03		0.2140 00 0.1000-01 0.2050 00	0.2570 00 0.1200-01 0.2430 00		0.6620 05 0.4610 04 0.7030 05		0.7930 05 0.5510 04 0.8370 05			
10	11	3/2-5/2 5/2-5/2	-0.3060-03 -0.3010-03	0.2010-03 0.2970-03		0.7540-01 0.3470-02	0.7260-01 0.3390-02		0.3530 05 0.2410 04		0.3400 05 0.2360 04			
11	9	3/2-5/2 5/2-5/2 5/2-7/2	-0.3390-03 -0.3420-03 -0.3350-03	0.5480-03 0.5330-03 0.5380-03		0.6470 00 0.3360-01 0.6180 00	0.1700 01 0.7810-01 0.1600 01		0.6130 04 0.4560 03 0.6280 04		0.1610 05 0.1060 04 0.1620 05			
11	10	3/2-5/2 5/2-5/2 5/2-7/2	-0.8050-03 -0.8080-03 -0.8040-03	0.1200-02 0.1210-02 0.1190-02		0.1320 01 0.6360-01 0.1260 01	0.2920 01 0.1420 00 0.2760 01		0.9630 05 0.6870 04 0.1020 06		0.2130 06 0.1540 05 0.2240 06			
11	11	3/2-5/2 5/2-5/2	-0.4120-03 -0.4060-03	0.4640-03 0.4590-03		0.2330 00 0.1100-01	0.2970 00 0.1390-01		0.3710 05 0.2600 04		0.4720 05 0.3280 04			
12	10	3/2-5/2 5/2-5/2 5/2-7/2	-0.7230-03 -0.1620-02 -0.3600-03	0.1980-03 0.1970-03 0.1910-03		0.4090 01 0.1900 01 0.9870 00	0.3070 00 0.1480-01 0.2790 00		0.2020 05 0.7110 04 0.5240 04		0.1510 04 0.1050 03 0.1480 04			
12	11	3/2-5/2 5/2-5/2	-0.5230-03 -0.1160-02	0.5230-03 0.5240-03		0.7570 00 0.1800 00	0.7570 00 0.3650-01		0.2980 05 0.1050 05		0.2980 05 0.2120 04			
13	11	3/2-5/2 5/2-5/2	-0.5430-03 -0.1240-02	0.2590-05 0.6170-05		0.3090 01 0.7800 00	0.7020-04 0.1940-04		0.8480 04 0.3070 04		0.11920 00 0.7650-01			

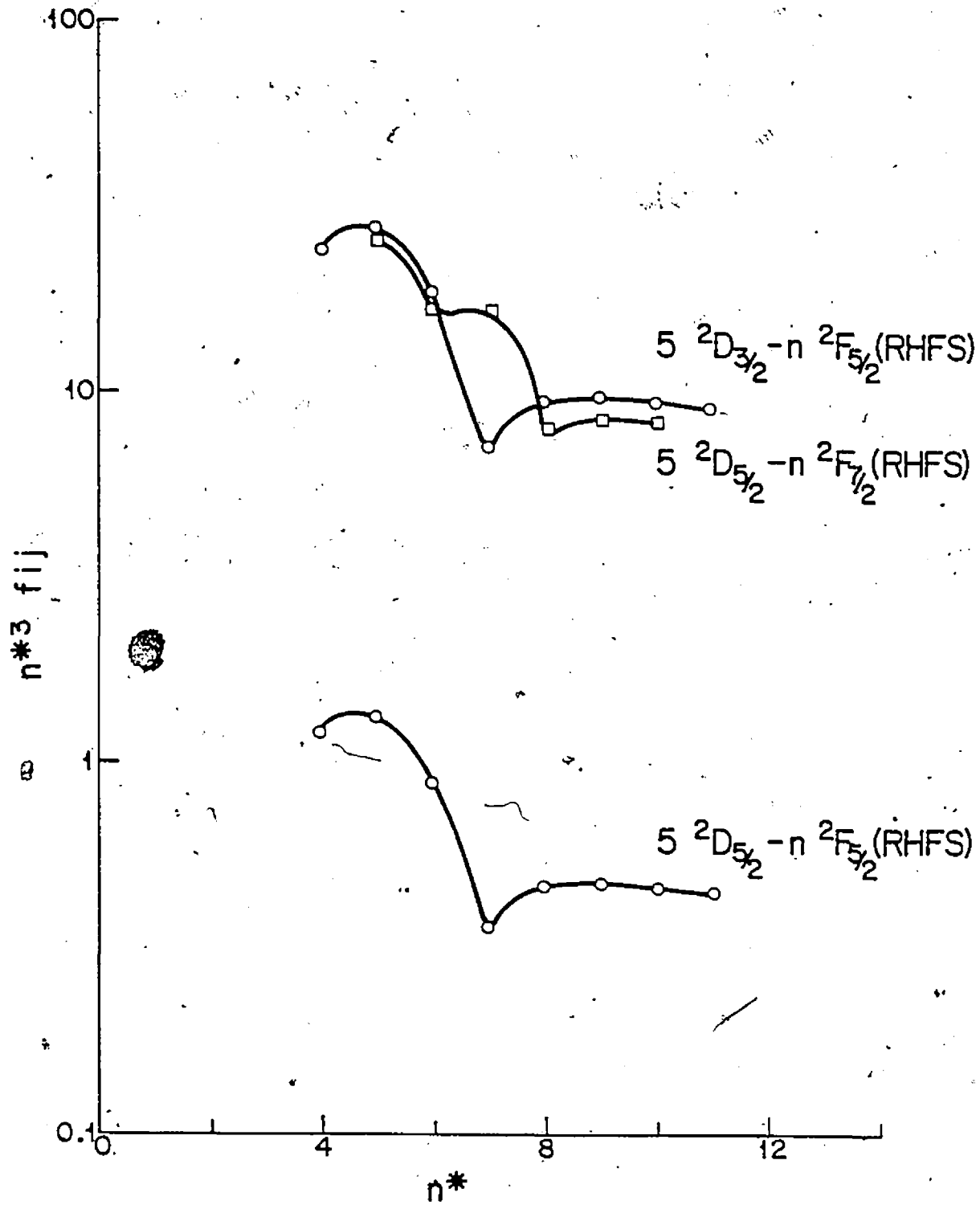
APPENDIX 4 : Plots of $n^3 f_{ij}$ versus n^* indicating the behavior of the relativistic oscillator strengths as a function of the effective principal quantum number for several Cesium series. The oscillator strengths are taken from Appendix (3) and correspond to those derived in the velocity formulation. The theoretical curves are denoted by (RHFS) and the corresponding plots formed from Agnew's experimental data are designated by (AGNEW).

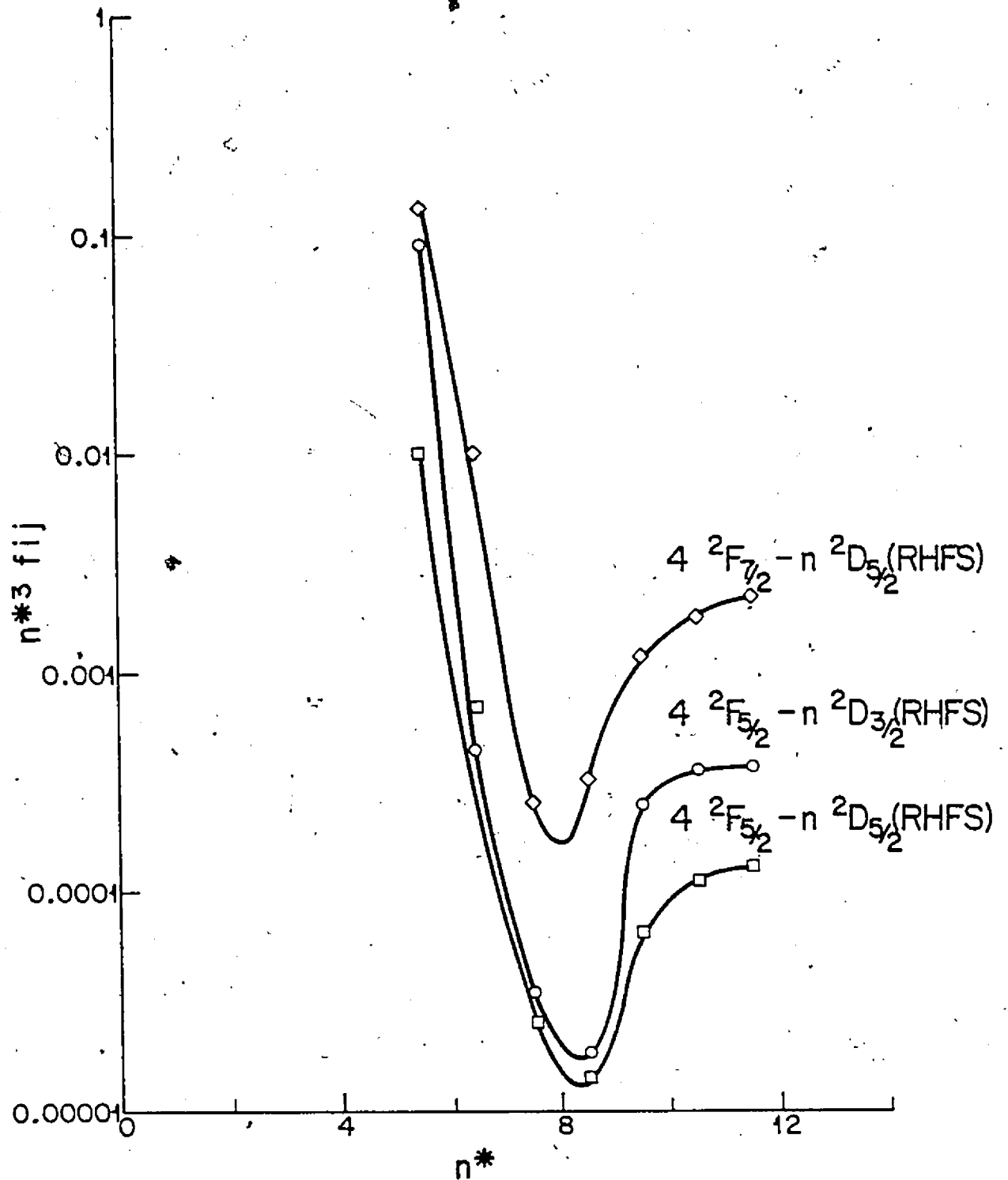












APPENDIX 5 : Cesium Oscillator Strengths

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_j - ms^2S_j$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION		M^2_{IJ}		f_{IJ}		f_{IJ}		f_{IJ}		$\Lambda_{IJ}(\text{sec}^{-1})$	
		J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH			
6	7	1/2 - 1/2	-0.202D-02	0.182D-02	0.253D 00	0.235D 00	0.171D 00		0.914D 07	0.740D 07			
		3/2 - 1/2	-0.201D-02	0.182D-02	0.273D 00	0.222D 00	0.208D 00		0.168D 06	0.137D 06			
6	8	1/2 - 1/2	-0.832D-03	0.511D-03	0.241D-01	0.911D-02	0.202D-01		0.278D 07	0.1C5D 07			
		3/2 - 1/2	-0.795D-03	0.467D-03	0.232D-01	0.735D-02	0.204D-01		0.492D 07	0.168D 07			
6	9	1/2 - 1/2	-0.520D-03	0.276D-03	0.786D-02	0.221D-02	0.702D-02	0.122D-01	0.130D 07	0.365D 06			
		3/2 - 1/2	-0.497D-03	0.247D-03	0.746D-02	0.184D-02	0.687D-02		0.229D 07	0.566D 06			
6	10	1/2 - 1/2	-0.385D-03	0.187D-03	0.397D-02	0.940D-03	0.326D-02	0.433D-02	0.777D 06	0.184D 06			
		3/2 - 1/2	-0.368D-03	0.166D-03	0.375D-02	0.765D-03	0.299D-02	0.381D-02	0.137D 07	0.290D 06			
6	11	1/2 - 1/2	-0.298D-03	0.138D-03	0.227D-02	0.486D-03	0.193D-02	0.243D-02	0.488D 06	0.104D 06			
		3/2 - 1/2	-0.285D-03	0.122D-03	0.214D-02	0.390D-03	0.186D-02	0.261D-02	0.864D 06	0.158D 06			
6	12	1/2 - 1/2	-0.239D-03	0.107D-03	0.142D-02	0.285D-03	0.122D-01	0.151D-02	0.324D 06	0.649D 05			
		3/2 - 1/2	-0.229D-03	0.943D-04	0.134D-02	0.227D-03	0.127D-02	0.140D-02	0.575D 06	0.972D 05			
6	13	1/2 - 1/2	-0.198D-03	0.862D-04	0.951D-03	0.181D-03	0.820D-03	0.121D-02	0.224D 06	0.426D 05			
		3/2 - 1/2	-0.185D-03	0.756D-04	0.896D-03	0.144D-03	0.780D-03	0.104D-02	0.357D 06	0.637D 05			
6	14	1/2 - 1/2	-0.167D-03	0.717D-04	0.667D-03	0.123D-03	0.590D-03	0.706D-03	0.162D 06	0.299D 05			
		3/2 - 1/2	-0.155D-03	0.628D-04	0.628D-03	0.976D-04	0.560D-03	0.690D-03	0.287D 06	0.446D 05			
6	15	1/2 - 1/2	-0.143D-03	0.6C9D-04	0.485D-03	0.880D-04	0.465D-03		0.120D 06	0.218D 05			
		3/2 - 1/2	-0.137D-03	0.533D-04	0.457D-03	0.694D-04	0.457D-03	0.380D-03	0.214D 06	0.325D 05			
7	8	1/2 - 1/2	-0.161D-02	0.131D-02	0.466D 00	0.308D 00	0.297D 00		0.202D 07	0.134D 07			
		3/2 - 1/2	-0.159D-02	0.132D-02	0.489D 00	0.337D 00	0.333D 00		0.777D 06	0.343D 06			
7	9	1/2 - 1/2	-0.7C3D-03	0.467D-03	0.440D-01	0.194D-01	0.305D-01		0.131D 07	0.542D 06			
		3/2 - 1/2	-0.656D-03	0.422D-03	0.400D-01	0.165D-01	0.256D-01		0.455D 06	0.167D 06			
7	10	1/2 - 1/2	-0.478D-03	0.289D-03	0.160D-01	0.585D-02	0.964D-02		0.769D 06	0.253D 06			
		3/2 - 1/2	-0.445D-03	0.255D-03	0.143D-01	0.470D-02	0.842D-02		0.286D 06	0.943D 05			
7	11	1/2 - 1/2	-0.357D-03	0.204D-03	0.790D-02	0.260D-02	0.5C1D-02		0.4E4D 06	0.140D 06			
		3/2 - 1/2	-0.331D-03	0.179D-03	0.700D-02	0.204D-02	0.455D-02		0.150D 06	0.583D 05			
7	12	1/2 - 1/2	-0.286D-03	0.156D-03	0.456D-02	0.140D-02	0.287D-02		0.320D 06	0.864D 05			
		3/2 - 1/2	-0.261D-03	0.135D-03	0.403D-02	0.178D-02	0.259D-02		0.707D 05	0.196D 05			
7	13	1/2 - 1/2	-0.228D-03	0.123D-03	0.292D-02	0.46D-03	0.193D-02		0.119D 06	0.286D 05			
		3/2 - 1/2	-0.212D-03	0.107D-03	0.258D-02	0.649D-03	0.164D-02		0.608D 06	0.387D 06			
7	14	1/2 - 1/2	-0.191D-03	0.102D-03	0.198D-02	0.560D-03	0.126D-02		0.111D 07	0.732D 05			
		3/2 - 1/2	-0.178D-03	0.879D-04	0.174D-02	0.427D-03	0.113D-02		0.2E6D 06	0.145D 06			
7	15	1/2 - 1/2	-0.163D-03	0.861D-04	0.140D-02	0.391D-03	0.113D-02		0.4R2D 06	0.236D 06			
		3/2 - 1/2	-0.152D-03	0.742D-04	0.124D-02	0.297D-03	0.113D-02		0.173D 06	0.769D 05			
8	9	1/2 - 1/2	-0.129D-02	0.103D-02	0.632D 00	0.401D 00	0.401D 00		0.254D 06	0.122D 06			
		3/2 - 1/2	-0.127D-02	0.1C4D-02	0.665D 00	0.439D 00	0.439D 00						
8	10	1/2 - 1/2	-0.6C2D-03	0.428D-03	0.640D-01	0.324D-01	0.281D-01						
		3/2 - 1/2	-0.564D-03	0.393D-03	0.581D-01	0.281D-01	0.281D-01						
8	11	1/2 - 1/2	-0.4C7D-03	0.272D-03	0.222D-01	0.137D-02	0.137D-02						
		3/2 - 1/2	-0.375D-03	0.244D-03	0.197D-01	0.018D-02	0.018D-02						

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES. FOR THE SERIES $np^2P - ms^2S_1$ CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUE, OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION J-J'	f_{ij}^e		f_{ij}		f_{ij}		AGNEW	$A_{ij}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	LENGTH		VELOCITY	LENGTH
8	12	1/2 - 1/2 3/2 - 1/2	-0.306D-03 -0.285D-03	0.197D-03 0.175D-03	0.109D-01 0.961D-02	0.449D-02 0.364D-02				0.113D 06 0.152D 06	0.467D 05 0.728D 05
8	13	1/2 - 1/2 3/2 - 1/2	-0.244D-03 -0.227D-03	0.151D-03 0.133D-03	0.639D-02 0.564D-02	0.244D-02 0.195D-02				0.773D 05 0.131D 06	0.295D 05 0.455D 05
8	14	1/2 - 1/2 3/2 - 1/2	-0.201D-03 -0.187D-03	0.123D-03 0.108D-03	0.407D-02 0.359D-02	0.152D-02 0.121D-02				0.558D 05 0.951D 05	0.209D 05 0.320D 05
8	15	1/2 - 1/2 3/2 - 1/2	-0.170D-03 -0.158D-03	0.103D-03 0.907D-04	0.278D-02 0.245D-02	0.102D-02 0.808D-03				0.418D 05 0.707D 05	0.153D 05 0.234D 05
9	10	1/2 - 1/2 3/2 - 1/2	-0.107D-02 -0.106D-02	0.854D-03 0.861D-03	0.798D 00 0.839D 00	0.504D 00 0.549D 00				0.234D 06 0.430D 06	0.148D 06 0.290D 06
9	11	1/2 - 1/2 3/2 - 1/2	-0.495D-03 -0.463D-03	0.353D-03 0.323D-03	0.751D-01 0.679D-01	0.382D-01 0.329D-01				0.112D 06 0.150D 06	0.566D 05 0.922D 05
9	12	1/2 - 1/2 3/2 - 1/2	-0.336D-03 -0.313D-03	0.225D-03 0.202D-03	0.255D-01 0.226D-01	0.115D-01 0.940D-02				0.703D 05 0.119D 06	0.315D 05 0.492D 05
9	13	1/2 - 1/2 3/2 - 1/2	-0.255D-03 -0.237D-03	0.161D-03 0.142D-03	0.128D-01 0.112D-01	0.508D-02 0.406D-02				0.463D 05 0.781D 05	0.194D 05 0.282D 05
9	14	1/2 - 1/2 3/2 - 1/2	-0.205D-03 -0.190D-03	0.127D-03 0.112D-03	0.736D-02 0.646D-02	0.285D-02 0.224D-02				0.324D 05 0.566D 05	0.129D 05 0.157D 05
9	15	1/2 - 1/2 3/2 - 1/2	-0.170D-03 -0.158D-03	0.104D-03 0.916D-04	0.471D-02 0.412D-02	0.178D-02 0.139D-02				0.249D 05 0.422D 05	0.938D 04 0.142D 05
10	11	1/2 - 1/2 3/2 - 1/2	-0.503D-03 -0.492D-03	0.743D-03 0.746D-03	0.926D 00 0.572D 00	0.927D 00 0.678D 00				0.100D 05 0.164D 05	0.678D 05 0.128D 06
10	12	1/2 - 1/2 3/2 - 1/2	-0.411D-03 -0.383D-03	0.313D-03 0.286D-03	0.822D-01 0.737D-01	0.479D-01 0.412D-01				0.484D 05 0.819D 05	0.282D 05 0.459D 05
10	13	1/2 - 1/2 3/2 - 1/2	-0.280D-03 -0.255D-03	0.192D-03 0.172D-03	0.290D-01 0.254D-01	0.137D-01 0.112D-01				0.251D 05 0.456D 05	0.140D 05 0.220D 05
10	14	1/2 - 1/2 3/2 - 1/2	-0.214D-03 -0.198D-03	0.148D-03 0.131D-03	0.138D-01 0.120D-01	0.960D-02 0.927D-02				0.212D 05 0.356D 05	0.101D 05 0.156D 05
10	15	1/2 - 1/2 3/2 - 1/2	-0.172D-03 -0.159D-03	0.119D-03 0.105D-03	0.789D-02 0.686D-02	0.374D-02 0.295D-02				0.157D 05 0.264D 05	0.744D 04 0.113D 05
11	12	1/2 - 1/2 3/2 - 1/2	-0.777D-03 -0.768D-03	0.655D-03 0.656D-03	0.105D 01 0.110D 01	0.749D 00 0.805D 00				0.464D 05 0.881D 05	0.344D 05 0.645D 05
11	13	1/2 - 1/2 3/2 - 1/2	-0.350D-03 -0.325D-03	0.251D-03 0.228D-03	0.698D-01 0.891D-01	0.514D-01 0.437D-01				0.210D 05 0.351D 05	0.108D 05 0.172D 05
11	14	1/2 - 1/2 3/2 - 1/2	-0.240D-03 -0.221D-03	0.173D-03 0.157D-03	0.314D-01 0.271D-01	0.164D-01 0.137D-01				0.148D 05 0.246D 05	0.773D 04 0.124D 05
11	15	1/2 - 1/2 3/2 - 1/2	-0.183D-03 -0.165D-03	0.134D-03 0.119D-03	0.146D-01 0.126D-01	0.790D-02 0.519D-02				0.107D 05 0.179D 05	0.579D 04 0.876D 04
12	13	1/2 - 1/2 3/2 - 1/2	-0.683D-03 -0.674D-03	0.361D-03 0.348D-03	0.191D 01 0.208D 01	0.534D 00 0.553D 00				0.160D 05 0.277D 05	0.446D 04 0.740D 04

ms^{-2}
 ms^{-2}
 ms^{-2}

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P - ms^2s^1$
 CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM.
 OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW
 ARE ALSO SHOWN

N	M	TRANSITION		P^0_{IJ}		f_{IJ}		f_{IJ}		f_{IJ}		$A_{IJ}(\text{sec}^{-1})$	
		J-J'		VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH		
12	14	1/2 - 1/2		-0.3050-03	0.2260-03	0.1680 00	0.5960-01			0.1120 05	0.6120 04		
		3/2 - 1/2		-0.2830-03	0.2060-03	0.9630-01	0.5090-01			0.1260 05	0.9840 04		
12	15	1/2 - 1/2		-0.1920-03	0.1560-03	0.3390-01	0.1850-01			0.8140 04	0.4420 04		
		3/2 - 1/2		-0.1920-03	0.1400-03	0.2870-01	0.1520-01			0.1320 05	0.6990 04		
13	14	1/2 - 1/2		-0.6330-03	0.3080-03	0.2310 01	0.3450 00			0.9720 04	0.2300 04		
		3/2 - 1/2		-0.6250-03	0.2990-03	0.2520 01	0.5760 00			0.1690 05	0.3870 04		
13	15	1/2 - 1/2		-0.2570-03	0.2160-03	0.1410 00	0.7510-01			0.7690 04	0.4090 04		
		3/2 - 1/2		-0.2780-03	0.2000-03	0.1280 00	0.8610-01			0.1310 05	0.6780 04		
14	15	1/2 - 1/2		-0.5680-03	0.2740-03	0.2520 01	0.5860 00			0.5750 04	0.1340 04		
		3/2 - 1/2		-0.5590-03	0.2690-03	0.2740 01	0.6310 00			0.1000 05	0.2310 04		

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2s\ 1/2 - mp^2p\ 1/2$ CALCULATED USING THE CRITICALLY EVALUATED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION		f_{ij}^0			f_{ij}			f_{ij}			$A_{ij}(\text{sec}^{-1})$		
		$J - J'$		VELOCITY	LENGTH		VELOCITY	LENGTH		STONE	AGNEW	VELOCITY	LENGTH		
6	6	1/2 - 1/2	1/2	0.285D-02	-0.317D-02	0.342D 00	0.411D 00	0.394D 00	0.285D 08		0.285D 08	0.342D 08			
		1/2 - 3/2	3/2	0.302D-02	-0.331D-02	0.712D 00	0.854D 00	0.814D 00	0.327D 08		0.327D 08	0.392D 08			
6	7	1/2 - 1/2	1/2	0.377D-03	-0.355D-03	0.299D-02	0.265D-02	0.284D-02	0.747D 06	0.320D-02	0.221D 07	0.839D 06			
		1/2 - 3/2	3/2	0.587D-03	-0.563D-03	0.144D-01	0.132D-01	0.174D-01	0.221D 07	0.133D-01	0.221D 07	0.213D 07			
6	8	1/2 - 1/2	1/2	0.106D-03	-0.755D-04	0.201D-03	0.102D-03	0.317D-03	0.865D 05	0.385D-03	0.865D 05	0.446D 05			
		1/2 - 3/2	3/2	0.260D-03	-0.218D-03	0.240D-02	0.168D-02	0.349D-02	0.533D 05	0.294D-02	0.533D 05	0.173D 05			
6	9	1/2 - 1/2	1/2	0.365D-04	-0.123D-04	0.221D-04	0.249D-05	0.725D-04	0.112D 05	0.830D-04	0.112D 05	0.127D 04			
		1/2 - 3/2	3/2	0.150D-03	-0.112D-03	0.743D-03	0.418D-03	0.125D-02	0.150D 06	0.910D-03	0.150D 06	0.177D 06			
6	10	1/2 - 1/2	1/2	0.129D-04	-0.588D-05	0.267D-05	0.550D-06	0.289D-04	0.147D 04	0.190D-04	0.147D 04	0.303D 03			
		1/2 - 3/2	3/2	0.559D-04	-0.689D-04	0.318D-03	0.151D-03	0.620D-03	0.876D 05	0.417D-03	0.876D 05	0.418D 05			
6	11	1/2 - 1/2	1/2	0.361D-05	-0.114D-04	0.202D-06	0.201D-05	0.124D-04	0.117D 03	0.800D-05	0.117D 03	0.116D 04			
		1/2 - 3/2	3/2	0.728D-04	-0.471D-04	0.165D-03	0.692D-04	0.356D-03	0.475D 05	0.223D-03	0.475D 05	0.200D 05			
6	12	1/2 - 1/2	1/2	-0.456D-06	-0.127D-04	0.319D-08	0.247D-05	0.620D-05	0.150D 01	0.430D-05	0.150D 01	0.147D 04			
		1/2 - 3/2	3/2	0.562D-04	-0.347D-04	0.569D-04	0.370D-04	0.208D-03	0.288D 05	0.126D-03	0.288D 05	0.110D 05			
6	13	1/2 - 1/2	1/2	-0.250D-05	-0.146D-04	0.186D-06	0.324D-05	0.186D-06	0.112D 03	0.240D-05	0.112D 03	0.197D 04			
		1/2 - 3/2	3/2	0.486D-04	-0.285D-04	0.723D-04	0.247D-04	0.723D-04	0.220D 05	0.790D-04	0.220D 05	0.748D 04			
6	14	1/2 - 1/2	1/2	-0.429D-05	-0.136D-04	0.277D-06	0.278D-05	0.475D-04	0.146D 05	0.160D-05	0.146D 05	0.467D 04			
		1/2 - 3/2	3/2	0.397D-04	-0.225D-04	0.475D-04	0.152D-04	0.333D-04	0.151D 03	0.580D-04	0.151D 03	0.145D 04			
6	15	1/2 - 1/2	1/2	-0.452D-05	-0.124D-04	0.306D-06	0.232D-05	0.306D-06	0.104D 05	0.430D-04	0.104D 05	0.316D 04			
		1/2 - 3/2	3/2	0.334D-04	-0.184D-04	0.333D-04	0.101D-04	0.333D-04	0.556D 00	0.430D-04	0.556D 00	0.300D 07			
7	7	1/2 - 1/2	1/2	0.227D-02	-0.174D-02	0.734D 00	0.431D 00	0.112D 01	0.513D 07		0.513D 07	0.300D 07			
		1/2 - 3/2	3/2	0.238D-02	-0.180D-02	0.152D 01	0.871D 00	0.112D 01	0.567D 07		0.567D 07	0.338D 07			
7	8	1/2 - 1/2	1/2	0.555D-03	-0.297D-03	0.197D-01	0.564D-02	0.516D-02	0.674D 06		0.674D 06	0.193D 06			
		1/2 - 3/2	3/2	0.721D-03	-0.434D-03	0.655E-01	0.237D-01	0.256D-02	0.115D 07	0.790D-04	0.115D 07	0.419D 06			
7	9	1/2 - 1/2	1/2	0.371D-03	-0.944D-04	0.369D-02	0.449D-03	0.620D-03	0.240D 06		0.240D 06	0.248D 05			
		1/2 - 3/2	3/2	0.392D-03	-0.191D-03	0.154D-01	0.63D-02	0.502D-02	0.430D 06		0.430D 06	0.101D 06			
7	10	1/2 - 1/2	1/2	0.167D-03	-0.360D-04	0.125D-02	0.582D-04	0.170D-03	0.864D 05		0.864D 05	0.403D 04			
		1/2 - 3/2	3/2	0.259D-03	-0.107D-03	0.601D-02	0.102D-02	0.187D-02	0.250D 05		0.250D 05	0.356D 05			
7	11	1/2 - 1/2	1/2	0.116D-03	-0.140D-04	0.568D-03	0.427D-05	0.620D-04	0.46D 05		0.46D 05	0.653D 03			
		1/2 - 3/2	3/2	0.195D-03	-0.687D-04	0.390D-02	0.337D-03	0.911D-03	0.119D 06		0.119D 06	0.157D 05			
7	12	1/2 - 1/2	1/2	0.671D-04	-0.444D-05	0.377D-03	0.799D-06	0.270D-04	0.262D 05		0.262D 05	0.692D 02			
		1/2 - 3/2	3/2	0.146D-03	-0.444D-04	0.173D-02	0.139D-03	0.486D-03	0.740D 05		0.740D 05	0.810D 04			
7	13	1/2 - 1/2	1/2	0.736D-04	-0.147D-05	0.215D-03	0.853D-07	0.113D-03	0.154D 05		0.154D 05	0.769D 01			
		1/2 - 3/2	3/2	0.128D-03	-0.380D-04	0.129D-02	0.113D-03	0.113D-03	0.563D 05		0.563D 05	0.513D 04			
7	14	1/2 - 1/2	1/2	0.551D-04	-0.176D-05	0.135D-03	0.545D-06	0.135D-03	0.127D 05		0.127D 05	0.513D 02			
		1/2 - 3/2	3/2	0.105D-03	-0.290D-04	0.844D-03	0.051D-04	0.844D-03	0.356D 05		0.356D 05	0.306D 04			
7	15	1/2 - 1/2	1/2	0.485D-04	-0.476D-05	0.912D-04	0.362D-06	0.912D-04	0.881D 04		0.881D 04	0.931D 02			
		1/2 - 3/2	3/2	0.682D-04	-0.232D-04	0.592D-03	0.410D-04	0.592D-03	0.286D 05		0.286D 05	0.158D 04			
8	8	1/2 - 1/2	1/2	0.167D-02	-0.130D-02	0.520D 00	0.553D 00	0.520D 00	0.119D 07		0.119D 07	0.715D 06			
		1/2 - 3/2	3/2	0.176D-02	-0.133D-02	0.191D 01	0.110D 01	0.191D 01	0.139D 07		0.139D 07	0.798D 06			

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2S_{1/2} - mp^2P_{1/2}$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

TRANSITION		f_{1j}^0		f_{1j}		f_{1j}		f_{1j}		$A_{1j}(\text{sec}^{-1})$	
N	M	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH	VELOCITY	LENGTH
8	9	1/2 - 1/2	0.443D-03	-0.246D-03	0.271D-01	0.835D-02		0.159D 06	0.612D 05	0.327D 06	0.119D 06
		1/2 - 3/2	0.564D-03	-0.341D-03	0.866D-01	0.317D-01		0.744D 05	0.125D 05	0.143D-06	0.368D 05
8	10	1/2 - 1/2	0.236D-03	-0.963D-04	0.576D-02	0.963D-03		0.369D 05	0.353D 04	0.773D 05	0.154D 05
		1/2 - 3/2	0.225D-03	-0.165D-03	0.218D-01	0.361D-02		0.212D 05	0.120D 04	0.471D 05	0.773D 04
8	11	1/2 - 1/2	0.154D-03	-0.477D-04	0.214D-02	0.204D-03		0.154D 05	0.451D 03	0.212D 05	0.120D 04
		1/2 - 3/2	0.223D-03	-0.555D-04	0.892D-02	0.178D-02		0.369D 05	0.353D 04	0.773D 05	0.154D 05
8	12	1/2 - 1/2	0.112D-03	-0.267D-04	0.104D-02	0.588D-04		0.212D 05	0.120D 04	0.471D 05	0.773D 04
		1/2 - 3/2	0.167D-03	-0.676D-04	0.491D-02	0.755D-03		0.154D 05	0.451D 03	0.212D 05	0.120D 04
8	13	1/2 - 1/2	0.529D-04	-0.159D-04	0.675D-03	0.197D-04		0.369D 05	0.353D 04	0.773D 05	0.154D 05
		1/2 - 3/2	0.143D-03	-0.518D-04	0.318D-02	0.420D-03		0.100D 05	0.165D 03	0.245D 05	0.283D 04
8	14	1/2 - 1/2	0.734D-04	-0.944D-05	0.406D-03	0.672D-05		0.650D 04	0.637D 02	0.175D 05	0.182D 04
		1/2 - 3/2	0.115D-03	-0.390D-04	0.198D-02	0.229D-03		0.360D 06	0.248D 06	0.422D 06	0.277D 06
8	15	1/2 - 1/2	0.602D-04	-0.578D-05	0.266D-03	0.245D-05		0.649D 05	0.253D 05	0.166D 05	0.168D 05
		1/2 - 3/2	0.556D-04	-0.369D-04	0.135D-02	0.140D-03		0.270D 05	0.666D 04	0.504D 05	0.168D 05
9	9	1/2 - 1/2	0.127D-02	-0.106D-02	0.102D 01	0.705D 00		0.144D 05	0.248D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.134D-02	-0.108D-02	0.213D 01	0.140D 01		0.291D 05	0.406D 04	0.291D 05	0.406D 04
9	10	1/2 - 1/2	0.342D-03	-0.213D-03	0.294D-01	0.115D-01		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.434D-03	-0.288D-03	0.934D-01	0.412D-01		0.291D 05	0.406D 04	0.291D 05	0.406D 04
9	11	1/2 - 1/2	0.188D-03	-0.936D-04	0.650D-02	0.161D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.256D-03	-0.148D-03	0.240D-01	0.800D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
9	12	1/2 - 1/2	0.127C-03	-0.525D-04	0.249D-02	0.430D-03		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.180D-03	-0.944D-04	0.100D-01	0.276D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
9	13	1/2 - 1/2	0.100D-03	-0.340D-04	0.141D-02	0.162D-03		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.147D-03	-0.696D-04	0.603D-02	0.136D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
9	14	1/2 - 1/2	0.776D-04	-0.223D-04	0.779D-03	0.655D-04		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.115D-03	-0.512D-04	0.349D-02	0.688D-03		0.291D 05	0.406D 04	0.291D 05	0.406D 04
9	15	1/2 - 1/2	0.626D-04	-0.156D-04	0.481D-03	0.305D-04		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.547D-04	-0.399D-04	0.225D-02	0.398D-03		0.291D 05	0.406D 04	0.291D 05	0.406D 04
10	10	1/2 - 1/2	0.106D-02	-0.904D-03	0.117D 01	0.336D 00		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.110D-02	-0.904D-03	0.243D 01	0.165D 01		0.291D 05	0.406D 04	0.291D 05	0.406D 04
10	11	1/2 - 1/2	0.306D-03	-0.209D-03	0.373D-01	0.181D-01		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.373D-03	-0.268D-03	0.114D 00	0.587C-01		0.291D 05	0.406D 04	0.291D 05	0.406D 04
10	12	1/2 - 1/2	0.172D-03	-0.104D-03	0.874D-02	0.320D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.228D-03	-0.148D-03	0.303D-01	0.129D-01		0.291D 05	0.406D 04	0.291D 05	0.406D 04
10	13	1/2 - 1/2	0.126D-03	-0.670D-04	0.391D-02	0.110D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.173D-03	-0.103D-03	0.146D-01	0.522D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
10	14	1/2 - 1/2	0.525D-04	-0.451D-04	0.191D-02	0.446D-03		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.131D-03	-0.741D-04	0.753D-02	0.240D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04
10	15	1/2 - 1/2	0.736D-04	-0.329D-04	0.110D-02	0.219D-03		0.291D 05	0.406D 04	0.291D 05	0.406D 04
		1/2 - 3/2	0.106D-03	-0.569D-04	0.450D-02	0.131D-02		0.291D 05	0.406D 04	0.291D 05	0.406D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2 1/2 - mp^2 j^1$ CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION J-J'	$H_{IJ}^{(0)}$		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW		VELOCITY	LENGTH
11	11	1/2 - 1/2 1/2 - 3/2	0.6750-03 0.5210-03	-C.7640-03 -0.7820-03	0.1290 01 0.2670 01	0.9770 00 0.1920 01				0.6450 05 0.7520 05	0.4880 05 0.5460 05
11	12	1/2 - 1/2 1/2 - 3/2	0.2610-03 0.2210-03	-C.1960-03 -0.2450-03	0.4310-01 0.1280 00	0.2430-01 0.7490-01				0.1500 05 0.2310 05	0.8470 04 0.1350 05
11	13	1/2 - 1/2 1/2 - 3/2	0.1610-03 0.2100-03	-0.1070-03 -C.1450-03	0.1150-01 0.3870-01	0.5060-02 0.1040-01				0.8230 04 0.1410 05	0.3620 04 0.6700 04
11	14	1/2 - 1/2 1/2 - 3/2	0.1110-03 0.1490-03	-C.6690-04 -0.5740-04	0.4440-02 0.1600-01	0.1620-02 0.6840-02				0.4710 04 0.8600 04	0.1720 04 0.3680 04
11	15	1/2 - 1/2 1/2 - 3/2	0.6350-04 0.1150-03	-C.4700-04 -0.7130-04	0.2230-02 0.8430-02	0.7050-03 0.3230-02				0.3060 04 0.5830 04	0.5670 03 0.2230 04
12	12	1/2 - 1/2 1/2 - 3/2	0.7560-03 0.7510-03	-C.6770-03 -0.6930-03	0.1400 01 0.2880 01	0.1120 01 0.2210 01				0.3270 05 0.3800 05	0.2620 05 0.2910 05
12	13	1/2 - 1/2 1/2 - 3/2	0.2380-03 0.2910-03	-0.1860-03 -C.2290-03	0.5180-01 0.1520 00	0.3160-01 0.3460-01				0.8640 04 0.1310 05	0.5250 04 0.9140 04
12	14	1/2 - 1/2 1/2 - 3/2	0.1380-03 0.1790-03	-C.5280-04 -C.1240-03	0.1190-01 0.4000-01	0.3420-02 0.1920-01				0.4220 04 0.7190 04	0.1910 04 0.3440 04
12	15	1/2 - 1/2 1/2 - 3/2	0.5640-04 0.1280-03	-C.6170-04 -0.8740-04	0.4740-02 0.1660-01	0.1940-02 0.7760-02				0.2550 04 0.4510 04	0.1040 04 0.2110 04
13	13	1/2 - 1/2 1/2 - 3/2	0.6740-03 0.7060-03	-0.9230-03 -C.9260-03	0.1030 01 0.2180 01	0.1940 01 0.3740 01				0.2790 05 0.3200 05	0.5250 05 0.5500 05
13	14	1/2 - 1/2 1/2 - 3/2	0.1920-03 0.2390-03	-C.1500-03 -C.1970-03	0.4000-01 0.1200 00	0.2390-01 0.8140-01				0.4520 04 0.7570 04	0.2930 04 0.5130 04
13	15	1/2 - 1/2 1/2 - 3/2	0.1140-03 0.1510-03	-C.6430-04 -C.5750-04	0.9920-02 0.3440-01	0.3170-02 0.1440-01				0.2370 04 0.4180 04	0.7570 03 0.1750 04
14	14	1/2 - 1/2 1/2 - 3/2	0.5840-03 0.6130-03	-C.8500-03 -C.8550-03	0.1060 01 0.2230 01	0.2240 01 0.4340 01				0.1540 05 0.1770 05	0.3260 05 0.3440 05
14	15	1/2 - 1/2 1/2 - 3/2	0.1630-03 0.2040-03	-0.1480-03 -0.1960-03	0.3850-01 0.1190 00	0.3170-01 0.1100 00				0.2590 04 0.4110 04	0.2130 04 0.3800 04
15	15	1/2 - 1/2 1/2 - 3/2	0.5210-03 0.5490-03	-C.7770-03 -C.7790-03	0.1120 01 0.2380 01	0.2480 01 0.4780 01				0.9260 04 0.1070 05	0.2050 05 0.2150 05

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_j - mp^2P_j$ CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION		M_{ij}^0		f_{ij}		f_{ij}		AGNEW	$A_{ij}(\text{sec}^{-1})$	
		J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	VELOCITY	LENGTH			
5	7	3/2 - 1/2	0.530D-C3	-C.617D-03	0.177D-01	0.240D-01	0.124D 07	0.169D 07			0.124D 07	0.169D 07
		3/2 - 3/2	0.487D-03	-C.558D-03	0.292D-02	0.383D-02	0.118D 06	0.142D 06			0.118D 06	0.142D 06
		5/2 - 3/2	0.521D-03	-C.597D-03	0.203D-01	0.266D-01	0.110D 07	0.144D 07			0.110D 07	0.144D 07
5	8	3/2 - 1/2	0.173D-03	-C.188D-03	0.122D-02	0.144D-02	0.205D 06	0.241D 06			0.205D 06	0.241D 06
		3/2 - 3/2	0.164D-03	-C.178D-03	0.218D-03	0.258D-03	0.165D 05	0.220D 05			0.165D 05	0.220D 05
		5/2 - 3/2	0.180D-03	-C.191D-03	0.160D-02	0.178D-02	0.200D 06	0.224D 06			0.200D 06	0.224D 06
5	9	3/2 - 1/2	0.781D-04	-C.858D-04	0.213D-03	0.257D-03	0.428D 04	0.591D 04			0.428D 04	0.591D 04
		3/2 - 3/2	0.728D-04	-C.825D-04	0.390D-04	0.473D-04	0.438D 04	0.550D 04			0.438D 04	0.550D 04
		5/2 - 3/2	0.850D-04	-C.892D-04	0.304D-03	0.334D-03	0.521D 05	0.575D 05			0.521D 05	0.575D 05
5	10	3/2 - 1/2	0.418D-04	-C.479D-04	0.561D-04	0.740D-04	0.152D 05	0.200D 05			0.152D 05	0.200D 05
		3/2 - 3/2	0.385D-04	-C.420D-04	0.973D-05	0.137D-04	0.132D 04	0.186D 04			0.132D 04	0.186D 04
		5/2 - 3/2	0.473D-04	-C.506D-04	0.868D-04	0.996D-04	0.174D 05	0.200D 05			0.174D 05	0.200D 05
5	11	3/2 - 1/2	0.250D-04	-C.304D-04	0.192D-04	0.283D-04	0.571D 04	0.839D 04			0.571D 04	0.839D 04
		3/2 - 3/2	0.228D-04	-C.292D-04	0.319D-05	0.523D-05	0.475D 03	0.777D 03			0.475D 03	0.777D 03
		5/2 - 3/2	0.294D-04	-C.325D-04	0.320D-04	0.371D-04	0.703D 04	0.860D 04			0.703D 04	0.860D 04
5	12	3/2 - 1/2	0.162D-04	-C.210D-04	0.785D-05	0.131D-04	0.247D 04	0.413D 04			0.247D 04	0.413D 04
		3/2 - 3/2	0.144D-04	-C.201D-04	0.240D-05	0.240D-05	0.155D 03	0.378D 03			0.155D 03	0.378D 03
		5/2 - 3/2	0.198D-04	-C.226D-04	0.141D-04	0.184D-04	0.328D 04	0.430D 04			0.328D 04	0.430D 04
5	13	3/2 - 1/2	0.114D-04	-C.159D-04	0.378D-05	0.740D-05	0.124D 04	0.242D 04			0.124D 04	0.242D 04
		3/2 - 3/2	0.575D-05	-C.151D-04	0.134D-06	0.134D-05	0.910D 02	0.219D 03			0.910D 02	0.219D 03
		5/2 - 3/2	0.147D-04	-C.174D-04	0.757D-05	0.106D-04	0.164D 04	0.258D 04			0.164D 04	0.258D 04
5	14	3/2 - 1/2	0.785D-05	-C.119D-04	0.177D-05	0.436D-05	0.559D 03	0.137D 04			0.559D 03	0.137D 04
		3/2 - 3/2	0.647D-05	-C.112D-04	0.241D-06	0.725D-06	0.470D 02	0.122D 03			0.470D 02	0.122D 03
		5/2 - 3/2	0.106D-04	-C.131D-04	0.390D-05	0.593D-05	0.976D 03	0.148D 04			0.976D 03	0.148D 04
5	15	3/2 - 1/2	0.566D-05	-C.926D-05	0.914D-06	0.244D-05	0.315D 03	0.492D 03			0.315D 03	0.492D 03
		3/2 - 3/2	0.449D-05	-C.873D-05	0.115D-06	0.435D-06	0.188D 02	0.748D 02			0.188D 02	0.748D 02
		5/2 - 3/2	0.803D-05	-C.103D-04	0.222D-05	0.364D-05	0.566D 03	0.910D 03			0.566D 03	0.910D 03
6	8	3/2 - 1/2	0.189D-03	-C.265D-03	0.523D-02	0.103D-01	0.678D 05	0.134D 06			0.678D 05	0.134D 06
		3/2 - 3/2	0.557D-04	-C.197D-03	0.284D-03	0.111D-02	0.154D 04	0.757D 04			0.154D 04	0.757D 04
		5/2 - 3/2	0.130D-03	-C.212D-03	0.295D-02	0.784D-02	0.255D 05	0.791D 05			0.255D 05	0.791D 05
6	9	3/2 - 1/2	0.673D-04	-C.166D-03	0.411D-03	0.249D-02	0.140D 05	0.477D 05			0.140D 05	0.477D 05
		3/2 - 3/2	0.200D-04	-C.128D-03	0.720D-05	0.233D-03	0.124D 03	0.568D 04			0.124D 03	0.568D 04
		5/2 - 3/2	0.376D-04	-C.157D-03	0.154D-03	0.203D-02	0.352D 04	0.517D 05			0.352D 04	0.517D 05
6	10	3/2 - 1/2	0.282D-04	-C.115D-03	0.557D-04	0.282D-03	0.300D 04	0.492D 05			0.300D 04	0.492D 05
		3/2 - 3/2	0.364D-05	-C.855D-04	0.197D-06	0.119D-03	0.500D 01	0.301D 04			0.500D 01	0.301D 04
		5/2 - 3/2	0.649D-05	-C.954D-04	0.646D-05	0.316D-03	0.242D 03	0.306D 05			0.242D 03	0.306D 05
6	11	3/2 - 1/2	0.125D-04	-C.858D-04	0.106D-04	0.494D-03	0.653D 03	0.106D 05			0.653D 03	0.106D 05
		3/2 - 3/2	0.113D-04	-C.672D-04	0.173D-05	0.606D-04	0.537D 02	0.189D 04			0.537D 02	0.189D 04
		5/2 - 3/2	0.221D-05	-C.715D-04	0.356D-06	0.414D-03	0.183D 02	0.191D 05			0.183D 02	0.191D 05
6	12	3/2 - 1/2	0.529D-05	-C.675D-04	0.177D-05	0.237D-03	0.124D 04	0.202D 05			0.124D 04	0.202D 05
		3/2 - 3/2	0.126D-04	-C.530D-04	0.233D-05	0.354D-04	0.823D 02	0.125D 04			0.823D 02	0.125D 04
		5/2 - 3/2	0.437D-05	-C.563D-04	0.308D-05	0.241D-03	0.161D 03	0.126D 05			0.161D 03	0.126D 05
6	13	3/2 - 1/2	0.713D-06	-C.592D-04	0.307D-07	0.212D-03	0.235D 01	0.162D 05			0.235D 01	0.162D 05
		3/2 - 3/2	0.161D-04	-C.465D-04	0.313D-05	0.291D-04	0.120D 03	0.100D 04			0.120D 03	0.100D 04
		5/2 - 3/2	0.503D-05	-C.493D-04	0.675D-05	0.177D-03	0.177D 03	0.101D 05			0.177D 03	0.101D 05

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES FOR THE SERIES $nd^2D_j - mp^2P_j$ CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	TRANSITION		M^0_{IJ}		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ}(\text{sec}^{-1})$	
	M	J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH		VELOCITY	LENGTH
6	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.131D-05 -0.152D-04 -0.528D-05	-0.4863-04 -0.382D-04 -0.405D-04	0.101D-06 0.272D-05 0.690D-05	0.139D-03 0.171D-04 0.116D-03	0.823D 01 0.111D 03 0.418D 04	0.112D 06 0.699D 03 0.703D 04			
6	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.231D-05 -0.142D-04 -0.562D-05	-0.410D-04 -0.324D-04 -0.343D-04	0.308D-06 0.230D-05 0.640D-05	0.955D-04 0.120D-04 0.816D-04	0.260D 02 0.950D 02 0.403D 03	0.819D 04 0.513D 03 0.513D 04			
7	9	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.0258D-03 0.174D-03 0.194D-03	-0.308D-03 -0.245D-03 -0.255D-03	0.192D-01 0.170D-02 0.128D-01	0.274D-01 0.335D-02 0.222D-01	0.649D 05 0.303D 04 0.333D 05	0.922D 05 0.595D 04 0.579D 05			
7	10	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.125D-03 0.807D-04 0.521D-04	-0.194D-03 -0.159D-03 -0.166D-03	0.268D-02 0.220D-03 0.173D-02	0.544D-02 0.861D-03 0.562D-02	0.257D 05 0.117D 04 0.125D 05	0.616D 05 0.422D 04 0.406D 04			
7	11	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.764D-04 0.462D-04 0.542D-04	-0.139D-03 -0.115D-03 -0.120D-03	0.797D-03 0.580D-04 0.481D-03	0.262D-02 0.361D-03 0.235D-02	0.120D 05 0.478D 03 0.542D 04	0.394D 05 0.274D 04 0.265D 05			
7	12	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.525D-04 0.258D-04 0.355D-04	-0.107D-03 -0.894D-04 -0.927D-04	0.332D-03 0.213D-04 0.187D-04	0.137D-02 0.192D-03 0.124D-02	0.641D 04 0.207D 03 0.269D 04	0.265D 05 0.186D 04 0.179D 05			
7	13	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.411D-04 0.214D-04 0.268D-04	-0.916D-04 -0.770D-04 -0.757D-04	0.187D-03 0.102D-04 0.961D-04	0.734D-03 0.132D-03 0.351D-03	0.426D 04 0.150D 04 0.162D 04	0.211D 05 0.150D 04 0.143D 05			
7	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.312D-04 0.153D-04 0.197D-04	-0.746D-04 -0.629D-04 -0.651D-04	0.103D-03 0.494D-05 0.433D-04	0.586D-03 0.320D-04 0.337D-03	0.259D 04 0.624D 02 0.926D 03	0.148D 05 0.105D 04 0.101D 05			
7	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.248D-04 0.116D-04 0.153D-04	-0.625D-04 -0.530D-04 -0.548D-04	0.625D-04 0.271D-05 0.285D-04	0.360D-03 0.569D-04 0.367D-03	0.170D 04 0.370D 02 0.575D 03	0.107D 05 0.770D 03 0.744D 04			
8	10	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.268D-03 0.192D-03 0.209D-03	-0.285D-03 -0.228D-03 -0.239D-03	0.360D-01 0.358D-02 0.259D-01	0.407D-01 0.506D-02 0.337D-01	0.402D 05 0.212D 04 0.224D 05	0.455D 05 0.100D 04 0.292D 05			
8	11	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.142D-03 0.100D-03 0.110D-03	-0.184D-03 -0.152D-03 -0.158D-03	0.578D-02 0.570D-03 0.419D-02	0.710D-02 0.131D-02 0.857D-02	0.196D 05 0.984D 03 0.107D 05	0.328D 05 0.226D 04 0.219D 05			
8	12	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.524D-04 0.37D-04 0.706D-04	-0.133D-03 -0.111D-03 -0.116D-03	0.191D-02 0.181D-03 0.135D-02	0.399D-02 0.391D-03 0.362D-02	0.107D 05 0.508D 03 0.562D 04	0.222D 05 0.155D 04 0.151D 05			
8	13	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.705D-04 0.467D-04 0.525D-04	-0.110D-03 -0.922D-04 -0.958D-04	0.566D-03 0.866D-04 0.654D-03	0.235D-02 0.129D-03 0.214D-02	0.715D 04 0.315D 03 0.362D 04	0.174D 05 0.123D 04 0.119D 05			
8	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.525D-04 0.342D-04 0.392D-04	-0.876D-04 -0.738D-04 -0.765D-04	0.497D-03 0.413D-04 0.328D-03	0.136D-02 0.193D-03 0.125D-02	0.442D 04 0.150D 03 0.218D 04	0.121D 05 0.860D 03 0.831D 04			
8	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.415D-04 0.265D-04 0.307D-04	-0.724D-04 -0.614D-04 -0.635D-04	0.291D-03 0.234D-04 0.189D-03	0.471D-03 0.125D-03 0.307D-03	0.254D 04 0.118D 03 0.143D 04	0.981D 04 0.633D 03 0.608D 04			

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_j - mp^2P_j$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. THE EXPERIMENTAL VALUES OBTAINED BY AGNEW OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION		Ω_{ij}^2		f_{ij}		f_{ij}		f_{ij}		$\Lambda_{ij}(\text{sec}^{-1})$	
		J-J'		VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH
9	11	3/2 - 1/2	-C.251D-03	0.257D-03	-C.251D-03	0.526D-01	0.502D-01	0.526D-01	0.502D-01			0.232D 05	0.221D C5
		3/2 - 3/2	-0.201D-03	0.185D-03	-0.201D-03	0.551D-02	0.523D-02	0.551D-02	0.523D-02			0.146D 04	0.142D 05
		5/2 - 3/2	-0.210D-03	0.204D-03	-0.210D-03	0.390D-01	0.415D-01	0.390D-01	0.415D-01			0.134D 05	0.142D 05
9	12	3/2 - 1/2	-0.166D-03	0.144D-03	-0.166D-03	0.925D-02	0.124D-01	0.925D-02	0.124D-01			0.129D 05	0.173D C5
		3/2 - 3/2	-C.137D-03	0.165D-03	-C.137D-03	0.982D-03	0.166D-02	0.982D-03	0.166D-02			0.703D 03	0.119D 04
		5/2 - 3/2	-C.142D-03	0.115D-03	-C.142D-03	0.701D-02	0.108D-01	0.701D-02	0.108D-01			0.740D 04	0.115D 05
9	13	3/2 - 1/2	-C.128D-03	0.103D-03	-C.128D-03	0.362D-02	0.557D-02	0.362D-02	0.557D-02			0.864D 04	0.133D 05
		3/2 - 3/2	-0.106D-03	0.741D-04	-0.106D-03	0.374D-03	0.759D-03	0.374D-03	0.759D-03			0.451D 03	0.918D 03
		5/2 - 3/2	-0.111D-03	0.8C5D-04	-0.111D-03	0.266D-02	0.503D-02	0.266D-02	0.503D-02			0.475D 04	0.931D 04
9	14	3/2 - 1/2	-0.983D-04	0.742D-04	-0.983D-04	0.161D-02	0.283D-02	0.161D-02	0.283D-02			0.525D 04	0.922D 04
		3/2 - 3/2	-0.821D-04	0.524D-04	-0.821D-04	0.160D-03	0.193D-03	0.160D-03	0.193D-03			0.263D 03	0.645D 03
		5/2 - 3/2	-C.856D-04	0.577D-04	-C.856D-04	0.117D-02	0.258D-02	0.117D-02	0.258D-02			0.288D 04	0.628D 04
9	15	3/2 - 1/2	-0.796D-04	0.572D-04	-0.796D-04	0.864D-03	0.167D-02	0.864D-03	0.167D-02			0.347D 04	0.670D 04
		3/2 - 3/2	-0.671D-04	0.397D-04	-0.671D-04	0.832D-04	0.237D-03	0.832D-04	0.237D-03			0.168D 03	0.480D 03
		5/2 - 3/2	-C.655D-04	0.443D-04	-C.655D-04	0.623D-03	0.153D-02	0.623D-03	0.153D-02			0.167D 04	0.459D C4
10	12	3/2 - 1/2	-0.228D-03	0.246D-03	-0.228D-03	0.723D-01	0.619D-01	0.723D-01	0.619D-01			0.142D 05	0.122D 05
		3/2 - 3/2	-0.183D-03	0.186D-03	-0.183D-03	0.800D-02	0.773D-02	0.800D-02	0.773D-02			0.839D 03	0.910D 03
		5/2 - 3/2	-0.192D-03	0.195D-03	-0.192D-03	0.555D-01	0.515D-01	0.555D-01	0.515D-01			0.852D 04	0.790D 04
10	13	3/2 - 1/2	-0.159D-03	0.150D-03	-0.159D-03	0.147D-01	0.165D-01	0.147D-01	0.165D-01			0.955D 04	0.107D 05
		3/2 - 3/2	-C.132D-03	0.113D-03	-C.132D-03	0.165D-02	0.225D-02	0.165D-02	0.225D-02			0.550D 03	0.748D 03
		5/2 - 3/2	-0.136D-03	0.122D-03	-0.136D-03	0.116D-01	0.146D-01	0.116D-01	0.146D-01			0.571D 04	0.715D 04
10	14	3/2 - 1/2	-0.113D-03	0.102D-03	-0.113D-03	0.516D-02	0.629D-02	0.516D-02	0.629D-02			0.567D 04	0.715D 04
		3/2 - 3/2	-0.929D-04	0.774D-04	-0.929D-04	0.590D-03	0.849D-03	0.590D-03	0.849D-03			0.341D 03	0.492D 03
		5/2 - 3/2	-0.981D-04	0.821D-04	-0.981D-04	0.400D-02	0.571D-02	0.400D-02	0.571D-02			0.343D 04	0.492D 04
10	15	3/2 - 1/2	-0.878D-04	0.755D-04	-0.878D-04	0.238D-02	0.323D-02	0.238D-02	0.323D-02			0.381D 04	0.517D 04
		3/2 - 3/2	-0.740D-04	0.565D-04	-0.740D-04	0.268D-03	0.456D-03	0.268D-03	0.456D-03			0.215D 03	0.368D 03
		5/2 - 3/2	-0.776D-04	0.668D-04	-0.776D-04	0.186D-02	0.302D-02	0.186D-02	0.302D-02			0.223D 04	0.363D 04
11	13	3/2 - 1/2	-0.214D-03	0.244D-03	-0.214D-03	0.101D 00	0.779D-01	0.101D 00	0.779D-01			0.976D 04	0.752D 04
		3/2 - 3/2	-0.173D-03	0.188D-03	-0.173D-03	0.117D-01	0.984D-02	0.117D-01	0.984D-02			0.559D 03	0.508D 03
		5/2 - 3/2	-C.181D-03	0.200D-03	-C.181D-03	0.798D-01	0.556D-01	0.798D-01	0.556D-01			0.664D 04	0.496D 04
11	14	3/2 - 1/2	-0.138D-03	0.141D-03	-0.138D-03	0.182D-01	0.176D-01	0.182D-01	0.176D-01			0.590D 04	0.579D 04
		3/2 - 3/2	-C.116D-03	0.168D-03	-C.116D-03	0.213D-02	0.245D-02	0.213D-02	0.245D-02			0.360D 03	0.413D 03
		5/2 - 3/2	-C.119D-03	0.117D-03	-C.119D-03	0.150D-01	0.156D-01	0.150D-01	0.156D-01			0.375D 04	0.389D 04
11	15	3/2 - 1/2	-C.972D-04	0.553D-04	-C.972D-04	0.678D-02	0.094D-02	0.678D-02	0.094D-02			0.462D 04	0.384D 04
		3/2 - 3/2	-C.813D-04	0.751D-04	-C.813D-04	0.854D-03	0.102D-03	0.854D-03	0.102D-03			0.257D 04	0.271D 03
		5/2 - 3/2	-0.868D-04	0.818D-04	-0.868D-04	0.550D-02	0.620D-02	0.550D-02	0.620D-02			0.246D 04	0.277D 04
12	14	3/2 - 1/2	-0.231D-03	0.234D-03	-0.231D-03	0.128D 00	0.124D 00	0.128D 00	0.124D 00			0.657D 04	0.637D 04
		3/2 - 3/2	-C.193D-03	0.186D-03	-C.193D-03	0.157D-01	0.159D-01	0.157D-01	0.159D-01			0.470D 03	0.463D 03
		5/2 - 3/2	-0.201D-03	0.196D-03	-0.201D-03	0.105D 00	0.111D 00	0.105D 00	0.111D 00			0.422D 04	0.442D 04
12	15	3/2 - 1/2	-0.158D-03	0.136D-03	-0.158D-03	0.231D-01	0.311D-01	0.231D-01	0.311D-01			0.413D 04	0.554D 04
		3/2 - 3/2	-0.138D-03	0.107D-03	-0.138D-03	0.285D-02	0.472D-02	0.285D-02	0.472D-02			0.262D 03	0.434D 03
		5/2 - 3/2	-C.140D-03	0.117D-03	-C.140D-03	0.203D-01	0.291D-01	0.203D-01	0.291D-01			0.276D 04	0.396D 04
13	15	3/2 - 1/2	-C.217D-03	0.230D-03	-C.217D-03	0.165D 00	0.147D 00	0.165D 00	0.147D 00			0.475D 04	0.426D 04
		3/2 - 3/2	-0.179D-03	0.197D-03	-0.179D-03	0.212D-01	0.194D-01	0.212D-01	0.194D-01			0.326D 03	0.258D 03
		5/2 - 3/2	-0.187D-03	0.195D-03	-0.187D-03	0.139D 00	0.128D 00	0.139D 00	0.128D 00			0.315D 04	0.289D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2p_j - md^2D_j$
 CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM.
 OSCILLATOR STRENGTHS FROM THE MODEL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW
 ARE ALSO SHOWN

N	M	TRANSITION		P_{IJ}^0		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ}(\text{sec}^{-1})$	
		J-J'	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	VELOCITY		LENGTH	
6	5	1/2 - 3/2	0.560D-03	0.254D 00	0.254D 00	0.254D 00	0.251D 00	0.924D 06	0.934D 06	0.116D 06	0.116D 06	0.852D 06
		3/2 - 3/2	-0.829D-03	0.227D-01	0.227D-01	0.227D-01	0.214D-01	0.811D 06	0.811D 06	0.852D 06	0.852D 06	0.757D 06
		3/2 - 5/2	-0.900D-03	0.233D 00	0.233D 00	0.233D 00	0.207D 00	0.204D-01	0.204D-01	0.204D-01	0.204D-01	0.204D-01
6	6	1/2 - 3/2	0.280D-02	0.649D 00	0.649D 00	0.649D 00	0.510D 00	0.258D 00	0.258D 00	0.258D 00	0.258D 00	0.222D 0A
		3/2 - 3/2	0.251D-02	0.712D-01	0.712D-01	0.712D-01	0.574D-01	0.597D-01	0.597D-01	0.597D-01	0.597D-01	0.451D C7
		3/2 - 5/2	0.286D-02	0.617D 00	0.617D 00	0.617D 00	0.490D 00	0.322D 00	0.322D 00	0.322D 00	0.322D 00	0.259D C8
6	7	1/2 - 3/2	0.149D-02	0.137D 00	0.137D 00	0.137D 00	0.743D-01	0.927D-01	0.927D-01	0.123D 00	0.123D 00	0.550D C7
		3/2 - 3/2	0.146D-02	0.136D-01	0.136D-01	0.136D-01	0.727D-02	0.110D-01	0.110D-01	0.130D-01	0.130D-01	0.992D 06
		3/2 - 5/2	0.146D-02	0.122D 00	0.122D 00	0.122D 00	0.653D-01	0.951D-01	0.951D-01	0.111D 00	0.111D 00	0.595D 07
6	8	1/2 - 3/2	0.103D-02	0.580D-01	0.580D-01	0.580D-01	0.265D-01	0.419D-01	0.419D-01	0.419D-01	0.419D-01	0.245D 07
		3/2 - 3/2	0.585D-03	0.553D-02	0.553D-02	0.553D-02	0.246D-02	0.480D-02	0.480D-02	0.561D-02	0.561D-02	0.422D 06
		3/2 - 5/2	0.588D-03	0.500D-01	0.500D-01	0.500D-01	0.223D-01	0.418D-01	0.418D-01	0.439D-01	0.439D-01	0.256D 07
6	9	1/2 - 3/2	0.777D-03	0.313D-01	0.313D-01	0.313D-01	0.129D-01	0.228D-01	0.228D-01	0.251D-01	0.251D-01	0.134D 07
		3/2 - 3/2	0.735D-03	0.293D-02	0.293D-02	0.293D-02	0.116D-02	0.250D-02	0.250D-02	0.300D-02	0.300D-02	0.226D 06
		3/2 - 5/2	0.743D-03	0.266D-01	0.266D-01	0.266D-01	0.106D-01	0.223D-01	0.223D-01	0.237D-01	0.237D-01	0.138D 07
6	10	1/2 - 3/2	0.667D-03	0.184D-01	0.184D-01	0.184D-01	0.711D-02	0.139D-01	0.139D-01	0.152D-01	0.152D-01	0.794D 06
		3/2 - 3/2	0.574D-03	0.170D-02	0.170D-02	0.170D-02	0.627D-03	0.150D-02	0.150D-02	0.200D-02	0.200D-02	0.131D 06
		3/2 - 5/2	0.578D-03	0.155D-01	0.155D-01	0.155D-01	0.577D-02	0.135D-01	0.135D-01	0.153D-01	0.153D-01	0.806D 06
6	11	1/2 - 3/2	0.485D-03	0.117D-01	0.117D-01	0.117D-01	0.434D-02	0.920D-02	0.920D-02	0.108D-01	0.108D-01	0.508D 06
		3/2 - 3/2	0.461D-03	0.107D-02	0.107D-02	0.107D-02	0.378D-03	0.100D-02	0.100D-02	0.810D-03	0.810D-03	0.831D 05
		3/2 - 5/2	0.465D-03	0.981D-02	0.981D-02	0.981D-02	0.349D-02	0.880D-02	0.880D-02	0.893D-02	0.893D-02	0.513D 06
6	12	1/2 - 3/2	0.405D-03	0.790D-02	0.790D-02	0.790D-02	0.285D-02	0.250D-02	0.250D-02	0.869D-02	0.869D-02	0.344D 06
		3/2 - 3/2	0.381D-03	0.720D-03	0.720D-03	0.720D-03	0.246D-03	0.246D-03	0.246D-03	0.837D-02	0.837D-02	0.558D 05
		3/2 - 5/2	0.384D-03	0.659D-02	0.659D-02	0.659D-02	0.227D-02	0.227D-02	0.227D-02	0.837D-02	0.837D-02	0.345D 06
6	13	1/2 - 3/2	0.342D-03	0.558D-02	0.558D-02	0.558D-02	0.197D-02	0.470D-02	0.470D-02	0.470D-02	0.470D-02	0.243D 06
		3/2 - 3/2	0.321D-03	0.507D-03	0.507D-03	0.507D-03	0.170D-03	0.530D-02	0.530D-02	0.530D-02	0.530D-02	0.355D C5
		3/2 - 5/2	0.324D-03	0.464D-02	0.464D-02	0.464D-02	0.157D-02	0.464D-02	0.464D-02	0.530D-02	0.530D-02	0.243D 06
6	14	1/2 - 3/2	0.295D-03	0.410D-02	0.410D-02	0.410D-02	0.143D-02	0.380D-02	0.380D-02	0.380D-02	0.380D-02	0.179D 06
		3/2 - 3/2	0.277D-03	0.372D-03	0.372D-03	0.372D-03	0.122D-03	0.396D-02	0.396D-02	0.396D-02	0.396D-02	0.290D 05
		3/2 - 5/2	0.279D-03	0.341D-02	0.341D-02	0.341D-02	0.113D-02	0.396D-02	0.396D-02	0.396D-02	0.396D-02	0.179D 06
7	6	1/2 - 3/2	0.256D-03	0.728D-01	0.728D-01	0.728D-01	0.254D 00	0.327D 00	0.327D 00	0.470D-02	0.470D-02	0.165D 05
		3/2 - 3/2	0.403D-03	0.231D-01	0.231D-01	0.231D-01	0.195D-01	0.320D-01	0.320D-01	0.530D-02	0.530D-02	0.118D 06
		3/2 - 5/2	0.341D-03	0.148D 00	0.148D 00	0.148D 00	0.191D 00	0.309D 00	0.309D 00	0.530D-02	0.530D-02	0.243D 06
7	7	1/2 - 3/2	0.153D-02	0.800D 00	0.800D 00	0.800D 00	0.667D 00	0.237D 00	0.237D 00	0.380D-02	0.380D-02	0.165D 05
		3/2 - 3/2	0.200D-02	0.855D-01	0.855D-01	0.855D-01	0.737D-01	0.340D-01	0.340D-01	0.869D-02	0.869D-02	0.537D 06
		3/2 - 5/2	0.198D-02	0.781D 00	0.781D 00	0.781D 00	0.588D 00	0.282D 00	0.282D 00	0.396D-02	0.396D-02	0.179D 06
7	8	1/2 - 3/2	0.121D-02	0.222D 00	0.222D 00	0.222D 00	0.141D 00	0.822D-01	0.822D-01	0.470D-02	0.470D-02	0.165D 05
		3/2 - 3/2	0.120D-02	0.225D-01	0.225D-01	0.225D-01	0.146D-01	0.105D-01	0.105D-01	0.530D-02	0.530D-02	0.118D 06
		3/2 - 5/2	0.119D-02	0.199D 00	0.199D 00	0.199D 00	0.130D 00	0.890D-01	0.890D-01	0.530D-02	0.530D-02	0.200D 05
7	9	1/2 - 3/2	0.884D-03	0.101D 00	0.101D 00	0.101D 00	0.361D-01	0.391D-01	0.391D-01	0.169D 07	0.169D 07	0.934D 06
		3/2 - 3/2	0.860D-03	0.985D-02	0.985D-02	0.985D-02	0.553D-02	0.470D-02	0.470D-02	0.169D 07	0.169D 07	0.175D 06
		3/2 - 5/2	0.859D-03	0.862D-01	0.862D-01	0.862D-01	0.499D-01	0.410D-01	0.410D-01	0.169D 07	0.169D 07	0.105D 07
7	10	1/2 - 3/2	0.678D-03	0.547D-01	0.547D-01	0.547D-01	0.280D-01	0.221D-01	0.221D-01	0.169D 07	0.169D 07	0.554D 06
		3/2 - 3/2	0.654D-03	0.521D-02	0.521D-02	0.521D-02	0.269D-02	0.260D-02	0.260D-02	0.169D 07	0.169D 07	0.102D 06
		3/2 - 5/2	0.654D-03	0.468D-01	0.468D-01	0.468D-01	0.244D-01	0.228D-01	0.228D-01	0.118D 06	0.118D 06	0.616D 06

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2 p_j - nd^2 d_j$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M ₁	TRANSITION	M _{1J} ⁰		f _{1J}		f _{1J}		f _{1J}	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH
7	11	1/2 - 3/2	0.541D-03	-C.379D-03	0.330D-01	0.162D-01	0.139D-01	0.728D 06	0.356D 06	
		3/2 - 3/2	0.515D-03	-C.364D-03	0.310D-02	0.153D-02	0.160D-02	0.131D 06	0.645D 05	
		3/2 - 5/2	0.520D-03	-C.366D-03	0.280D-01	0.139D-01	0.142D-01	0.785D 06	0.391D 06	
7	12	1/2 - 3/2	0.445D-03	-C.367D-03	0.216D-01	0.102D-01	0.102D-01	0.513D 06	0.243D 06	
		3/2 - 3/2	0.426D-03	-C.293D-03	0.201D-02	0.956D-03	0.434D 05	0.914D 05	0.434D 05	
		3/2 - 5/2	0.426D-03	-C.296D-03	0.182D-01	0.373D-02	0.550D 06	0.264D 06	0.264D 06	
7	13	1/2 - 3/2	0.375D-03	-C.256D-03	0.149D-01	0.693D-02	0.693D-02	0.372D 06	0.173D 06	
		3/2 - 3/2	0.357D-03	-C.244D-03	0.138D-02	0.642D-03	0.642D-03	0.661D 05	0.337D 05	
		3/2 - 5/2	0.258D-03	-C.246D-03	0.125D-01	0.587D-02	0.197D 06	0.359D 06	0.197D 06	
7	14	1/2 - 3/2	0.322D-03	-C.218D-03	0.108D-01	0.494D-02	0.494D-02	0.279D 06	0.128D 06	
		3/2 - 3/2	0.306D-03	-C.217D-03	0.995D-03	0.455D-03	0.226D 05	0.456D 05	0.226D 05	
		3/2 - 5/2	0.307D-03	-C.209D-03	0.900D-02	0.416D-02	0.416D-02	0.258D 06	0.138D 06	
8	7	1/2 - 3/2	0.165D-03	C.375D-03	0.769D-01	0.380D 00	0.380D 00	0.254D 04	0.146D 05	
		3/2 - 3/2	0.307D-03	C.261D-03	0.336D-01	0.282D-01	0.147D 04	0.124D 04	0.124D 04	
		3/2 - 5/2	0.276D-03	C.306D-03	0.226D 00	0.278D 00	0.765D 04	0.951D 04	0.951D 04	
8	8	1/2 - 3/2	0.138D-02	-C.117D-02	0.825D 00	0.536D 00	0.536D 00	0.122D 07	0.876D 06	
		3/2 - 3/2	0.146D-02	-C.128D-02	0.966D-01	0.738D-01	0.201D 06	0.243D 06	0.201D 06	
		3/2 - 5/2	0.148D-02	-C.126D-02	0.837D 00	0.642D 00	0.154D-07	0.118D 07	0.118D 07	
8	9	1/2 - 3/2	0.917D-03	-C.690D-03	0.247D 00	0.140D 00	0.140D 00	0.802D 06	0.455D 06	
		3/2 - 3/2	0.531D-03	-C.711D-03	0.262D-01	0.153D-01	0.153D-01	0.101D 06	0.938D 05	
		3/2 - 5/2	0.524D-03	-C.709D-03	0.231D 00	0.136D 00	0.136D 00	0.951D 06	0.562D 06	
8	10	1/2 - 3/2	0.677D-03	-C.481D-03	0.112D 00	0.563D-01	0.563D-01	0.525D 06	0.265D 06	
		3/2 - 3/2	0.676D-03	-C.485D-03	0.114D-01	0.506D-02	0.506D-02	0.103D 06	0.529D 05	
		3/2 - 5/2	0.673D-03	-C.485D-03	0.102D 00	0.527D-01	0.527D-01	0.612D 06	0.316D 06	
8	11	1/2 - 3/2	0.530D-03	-C.364D-03	0.614D-01	0.290D-01	0.290D-01	0.359D 05	0.169D 05	
		3/2 - 3/2	0.525D-03	-C.343D-03	0.614D-02	0.294D-02	0.294D-02	0.330D 05	0.330D 05	
		3/2 - 5/2	0.523D-03	-C.364D-03	0.549D-01	0.266D-01	0.266D-01	0.412D 06	0.200D 06	
8	12	1/2 - 3/2	0.421D-03	-C.290D-03	0.379D-01	0.172D-01	0.172D-01	0.254D 06	0.115D 06	
		3/2 - 3/2	0.424D-03	-C.287D-03	0.374D-02	0.171D-02	0.171D-02	0.464D 05	0.222D 05	
		3/2 - 5/2	0.423D-03	-C.288D-03	0.335D-01	0.155D-01	0.155D-01	0.250D 05	0.134D 06	
8	13	1/2 - 3/2	0.360D-03	-C.239D-03	0.252D-01	0.111D-01	0.111D-01	0.166D 06	0.923D 05	
		3/2 - 3/2	0.353D-03	-C.235D-03	0.247D-02	0.110D-02	0.110D-02	0.352D 05	0.156D 05	
		3/2 - 5/2	0.353D-03	-C.236D-03	0.222D-01	0.936D-02	0.936D-02	0.211D 06	0.947D 05	
8	14	1/2 - 3/2	0.307D-03	-C.202D-03	0.177D-01	0.768D-02	0.768D-02	0.141D 06	0.608D 05	
		3/2 - 3/2	0.304D-03	-C.198D-03	0.173D-02	0.751D-03	0.751D-03	0.225D 05	0.115D 05	
		3/2 - 5/2	0.306D-03	-C.199D-03	0.155D-01	0.684D-02	0.684D-02	0.158D 06	0.699D 05	
9	8	1/2 - 3/2	0.111D-03	C.305D-03	0.643D-01	0.490D 00	0.490D 00	0.649D 03	0.456D 04	
		3/2 - 3/2	0.223D-03	C.380D-01	0.380D-01	0.158D-01	0.158D-01	0.426D 03	0.399D 03	
		3/2 - 5/2	0.202D-03	C.247D-03	0.245D 00	0.156D 00	0.217D 04	0.315D 04	0.315D 04	
9	9	1/2 - 3/2	0.104D-02	-C.902D-03	0.832D 00	0.26D 00	0.26D 00	0.354D 06	0.256D 06	
		3/2 - 3/2	0.112D-02	-C.953D-03	0.595D-01	0.738D-01	0.738D-01	0.872D 05	0.690D 05	
		3/2 - 5/2	0.110D-02	-C.977D-03	0.861D 00	0.6d1D 00	0.6d1D 00	0.587D 06	0.493D 06	
9	10	1/2 - 3/2	0.653D-03	-C.542D-03	C.240D 00	0.147D 00	0.147D 00	0.268D 06	0.165D 06	
		3/2 - 3/2	0.715D-03	-C.566D-03	0.262D-01	0.164D-01	0.164D-01	0.558D 05	0.349D 05	
		3/2 - 5/2	0.718D-03	-C.562D-03	0.231D 00	0.145D 00	0.145D 00	0.329D 06	0.207D 06	

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_J - md^2D_J$ CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVEFUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	K	TRANSITION	M_{IJ}^0		f_{IJ}		f_{IJ}		f_{IJ}		AGNEW	$A_{IJ} \text{ (sec}^{-1}\text{)}$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW		VELOCITY	LENGTH
9	11	1/2 - 3/2	0.5150-03	-0.3870-03	0.1050 00	0.6070-01	0.1660 06	0.1030 06				0.1660 06	0.1030 06
		3/2 - 5/2	0.5270-03	-0.3940-03	0.1150 01	0.5780-02	0.3760 05	0.2120 05				0.3760 05	0.2120 05
		3/2 - 5/2	0.5240-03	-0.3940-03	0.1020 00	0.5780-02	0.2230 06	0.2170 06				0.2230 06	0.2170 06
9	12	1/2 - 3/2	0.4120-03	-0.3000-03	0.0680 01	0.3210-01	0.1330 06	0.7030 05				0.1330 06	0.7030 05
		3/2 - 3/2	0.4150-03	-0.3030-03	0.1340 02	0.3340 02	0.2650 05	0.1410 05				0.2650 05	0.1410 05
		3/2 - 5/2	0.4130-03	-0.3020-03	0.0590 01	0.2990-01	0.1590 06	0.8430 05				0.1590 06	0.8430 05
9	13	1/2 - 3/2	0.2390-03	-0.2430-03	0.3800 01	0.1950-01	0.9760 05	0.5000 05				0.9760 05	0.5000 05
		3/2 - 3/2	0.2400-03	-0.2440-03	0.3880 02	0.1990-02	0.1930 05	0.9920 04				0.1930 05	0.9920 04
		3/2 - 5/2	0.2390-03	-0.2440-03	0.3460 01	0.1790-01	0.1150 06	0.5550 05				0.1150 06	0.5550 05
9	14	1/2 - 3/2	0.2870-03	-0.2030-03	0.2560 01	0.1290-01	0.7400 05	0.3710 05				0.7400 05	0.3710 05
		3/2 - 3/2	0.2870-03	-0.2030-03	0.2600 02	0.1300 02	0.1460 05	0.7280 04				0.1460 05	0.7280 04
		3/2 - 5/2	0.2860-03	-0.2030-03	0.2320 01	0.1170-01	0.8680 05	0.4380 05				0.8680 05	0.4380 05
10	9	1/2 - 3/2	0.8850-04	0.2580-03	0.7040 01	0.5990 00	0.2430 03	0.2670 04				0.2430 03	0.2670 04
		3/2 - 3/2	0.1930-03	0.1880-03	0.4540 01	0.4330 01	0.1700 03	0.1620 03				0.1700 03	0.1620 03
		3/2 - 5/2	0.1710-03	0.2080-03	0.2920 00	0.4330 00	0.8760 03	0.1300 04				0.8760 03	0.1300 04
10	10	1/2 - 3/2	0.8200-03	-0.7480-03	0.8310 00	0.6920 00	0.1530 06	0.1270 06				0.1530 06	0.1270 06
		3/2 - 3/2	0.8370-03	-0.8260-03	0.1010 00	0.8740 01	0.3440 05	0.1900 05				0.3440 05	0.1900 05
		3/2 - 5/2	0.8720-03	-0.8120-03	0.8710 00	0.7550 00	0.2610 06	0.1460 05				0.2610 06	0.1460 05
10	11	1/2 - 3/2	0.5480-03	-0.4570-03	0.2350 00	0.1640 00	0.1070 05	0.7480 05				0.1070 05	0.7480 05
		3/2 - 3/2	0.5710-03	-0.4800-03	0.2610 01	0.1350 01	0.1170 05	0.7490 04				0.1170 05	0.7490 04
		3/2 - 5/2	0.5650-03	-0.4760-03	0.2300 00	0.1630 00	0.6500 05	0.4300 05				0.6500 05	0.4300 05
10	12	1/2 - 3/2	0.4150-03	-0.3320-03	0.1070 00	0.6860-01	0.7730 05	0.4920 05				0.7730 05	0.4920 05
		3/2 - 3/2	0.4260-03	-0.3420-03	0.1150 01	0.7440-02	0.1600 05	0.1030 05				0.1600 05	0.1030 05
		3/2 - 5/2	0.4230-03	-0.3400-03	0.1920 00	0.6600-01	0.9470 05	0.6120 05				0.9470 05	0.6120 05
10	13	1/2 - 3/2	0.3320-03	-0.2610-03	0.5990 01	0.3690-01	0.5710 05	0.3510 05				0.5710 05	0.3510 05
		3/2 - 3/2	0.3390-03	-0.2660-03	0.6330 02	0.3900-02	0.1170 05	0.7490 04				0.1170 05	0.7490 04
		3/2 - 5/2	0.3370-03	-0.2650-03	0.5620 01	0.3490-01	0.6500 05	0.4300 05				0.6500 05	0.4300 05
10	14	1/2 - 3/2	0.2770-03	-0.2150-03	0.3780 01	0.2280-01	0.4340 05	0.2610 05				0.4340 05	0.2610 05
		3/2 - 3/2	0.2810-03	-0.2170-03	0.3950 02	0.2370-02	0.8760 04	0.5290 04				0.8760 04	0.5290 04
		3/2 - 5/2	0.2790-03	-0.2170-03	0.3510 01	0.2130-01	0.5210 05	0.3160 05				0.5210 05	0.3160 05
11	10	1/2 - 3/2	0.6580-04	0.2250-03	0.6880 01	0.7130 00	0.9630 02	0.1000 04				0.9630 02	0.1000 04
		3/2 - 3/2	0.1660-03	0.1640-03	0.4920 01	0.5170-01	0.7400 02	0.7770 02				0.7400 02	0.7770 02
		3/2 - 5/2	0.1490-03	0.1800-03	0.3120 00	0.5120 00	0.6500 05	0.4300 05				0.6500 05	0.4300 05
11	11	1/2 - 3/2	0.6720-03	-0.6250-03	0.8410 00	0.7490 00	0.6620 05	0.6080 05				0.6620 05	0.6080 05
		3/2 - 3/2	0.7320-03	-0.7030-03	0.1030 00	0.9510-01	0.1550 05	0.1430 05				0.1550 05	0.1430 05
		3/2 - 5/2	0.7180-03	-0.6900-03	0.6890 00	0.3200 00	0.9050 05	0.8350 05				0.9050 05	0.8350 05
11	12	1/2 - 3/2	0.4510-03	-0.3930-03	0.2340 00	0.1780 00	0.4920 05	0.3750 05				0.4920 05	0.3750 05
		3/2 - 3/2	0.4730-03	-0.4140-03	0.2640 01	0.2020-01	0.1060 05	0.8100 04				0.1060 05	0.8100 04
		3/2 - 5/2	0.4680-03	-0.4100-03	0.2320 00	0.1780 00	0.6240 05	0.4880 05				0.6240 05	0.4880 05
11	13	1/2 - 3/2	0.3430-03	-0.2870-03	0.1070 00	0.7430-01	0.3650 05	0.2540 05				0.3650 05	0.2540 05
		3/2 - 3/2	0.3550-03	-0.3000-03	0.1160 01	0.8270-02	0.7550 04	0.5860 04				0.7550 04	0.5860 04
		3/2 - 5/2	0.3530-03	-0.2970-03	0.1030 00	0.7290-01	0.4550 05	0.3220 05				0.4550 05	0.3220 05
11	14	1/2 - 3/2	0.2760-03	-0.2280-03	0.6000 01	0.4040-01	0.2770 05	0.1860 05				0.2770 05	0.1860 05
		3/2 - 3/2	0.2860-03	-0.2330-03	0.6440 02	0.4300-02	0.5790 04	0.3860 04				0.5790 04	0.3860 04
		3/2 - 5/2	0.2830-03	-0.2320-03	0.5680 01	0.3330-01	0.3400 05	0.2300 05				0.3400 05	0.2300 05

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P \rightarrow nd^2D$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. THE EXPERIMENTAL VALUES OBTAINED BY AGNEW OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION J-J'	M^0_{IJ}		f_{IJ}		STONE	f_{IJ}	AGNEW	$A_{IJ}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH				VELOCITY	LENGTH
12	11	1/2 - 3/2	0.554D-04	0.199D-03	0.642D-01	0.829D 00				0.411D 02	0.529D 03
		3/2 - 3/2	0.135D-03	0.148D-03	0.522D-01	0.628D-01				0.354D 02	0.426D 02
		3/2 - 5/2	0.117D-03	0.159D-03	0.320D 00	0.392D 00				0.176D 03	0.325D 03
12	12	1/2 - 3/2	0.568D-03	-0.549D-03	0.859D 00	0.802D 00				0.339D 05	0.317D 05
		3/2 - 3/2	0.620D-03	-0.609D-03	0.106D 00	0.102D 00				0.781D 04	0.752D 04
		3/2 - 5/2	0.605D-03	-0.597D-03	0.913D 00	0.879D 00				0.455D 05	0.438D 05
12	13	1/2 - 3/2	0.381D-03	-0.344D-03	0.237D 00	0.193D 00				0.250D 05	0.204D 05
		3/2 - 3/2	0.403D-03	-0.364D-03	0.270D-01	0.221D-01				0.446D 04	0.446D 04
		3/2 - 5/2	0.397D-03	-0.362D-03	0.235D 00	0.195D 00				0.320D 05	0.265D 05
12	14	1/2 - 3/2	0.293D-03	-0.251D-03	0.108D 00	0.791D-01				0.151D 05	0.140D 05
		3/2 - 3/2	0.303D-03	-0.263D-03	0.118D-01	0.685D-02				0.402D 04	0.302D 04
		3/2 - 5/2	0.263D-03	-0.259D-03	0.105D 00	0.772D-01				0.241D 05	0.177D 05
13	12	1/2 - 3/2	0.330D-04	0.179D-03	0.322D-01	0.742D 00				0.103D 02	0.302D 03
		3/2 - 3/2	0.108D-03	0.138D-03	0.474D-01	0.780D-01				0.160D 02	0.264D 02
		3/2 - 5/2	0.891D-04	0.142D-03	0.264D 00	0.668D 00				0.723D 02	0.183D 03
13	13	1/2 - 3/2	0.484D-03	-0.442D-03	0.858D 00	0.715D 00				0.179D 05	0.149D 05
		3/2 - 3/2	0.533D-03	-0.497D-03	0.108D 00	0.936D-01				0.422D 04	0.364D 04
		3/2 - 5/2	0.525D-03	-0.486D-03	0.927D 00	0.799D 00				0.244D 05	0.210D 05
13	14	1/2 - 3/2	0.336D-03	-0.294D-03	0.250D 00	0.190D 00				0.143D 05	0.109D 05
		3/2 - 3/2	0.360D-03	-0.315D-03	0.293D-01	0.224D-01				0.322D 04	0.246D 04
		3/2 - 5/2	0.352D-03	-0.313D-03	0.251D 00	0.198D 00				0.165D 05	0.146D 05
14	13	1/2 - 3/2	0.115D-04	0.166D-03	0.526D-02	0.110D 01				0.918D 00	0.193D 03
		3/2 - 3/2	0.784D-04	0.129D-03	0.340D-01	0.923D-01				0.620D 01	0.169D 02
		3/2 - 5/2	0.614D-04	0.132D-03	0.174D 00	0.784D 00				0.252D 02	0.117D 03
14	14	1/2 - 3/2	0.416D-03	-0.381D-03	0.645D 00	0.711D 00				0.988D 04	0.831D 04
		3/2 - 3/2	0.459D-03	-0.437D-03	0.107D 00	0.965D-01				0.234D 04	0.211D 04
		3/2 - 5/2	0.451D-03	-0.426D-03	0.919D 00	0.821D 00				0.136D 05	0.121D 05
15	14	1/2 - 3/2	0.173D-05	0.153D-05	0.159D-03	0.125D 01				0.157D-01	0.123D 03
		3/2 - 3/2	0.620D-04	0.176D-03	0.282D-01	0.885D-01				0.293D 01	0.102D 02
		3/2 - 5/2	0.475D-04	0.124D-03	0.135D 00	0.910D 00				0.115D 02	0.773D 02

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THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nf^2 F_j - md^2 D_j$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STIGKE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION	f_{ij}^{No}			f_{ij}			f_{ij}			f_{ij}		
			VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW	VELOCITY	LENGTH	AGNEW
4	7	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.301D-03 -0.566D-03 -0.309D-03	0.924D-03 0.528D-03 0.931D-03		0.316D-01 0.230D-02 0.352D-01	0.278D 00 0.212D-01 0.320D 00		0.785D 05 0.352D 04 0.723D 06		0.785D 05 0.352D 04 0.723D 06			
4	8	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.120D-04 0.764D-05 0.564D-05	0.112D-03 0.120D-03 0.119D-03		0.236D-04 0.581D-06 0.559D-05	0.207D-02 0.168D-03 0.247D-02		0.266D 03 0.434D 03 0.558D 02		0.266D 03 0.434D 03 0.558D 02			
4	9	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.183D-04 0.154D-04 0.142D-04	0.654D-04 0.699D-04 0.691D-04		0.421D-04 0.213D-05 0.272D-04	0.540D-03 0.439D-04 0.645D-03		0.802D 03 0.271D 02 0.463D 03		0.802D 03 0.271D 02 0.463D 03			
4	10	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.163D-04 0.144D-04 0.134D-04	0.463D-04 0.494D-04 0.487D-04		0.293D-04 0.211D-04 0.211D-04	0.235D-03 0.192D-04 0.279D-03		0.732D 03 0.271D 02 0.467D 03		0.732D 03 0.271D 02 0.467D 03			
4	11	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.139D-04 0.124D-04 0.116D-04	0.354D-04 0.377D-04 0.372D-04		0.195D-04 0.111D-05 0.146D-04	0.127D-03 0.103D-04 0.150D-03		0.575D 03 0.219D 02 0.363D 03		0.575D 03 0.219D 02 0.363D 03			
4	12	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.118D-04 0.106D-04 0.958D-05	0.284D-04 0.302D-04 0.299D-04		0.134D-04 0.777D-06 0.102D-04	0.772D-04 0.626D-05 0.917D-04		0.438D 03 0.170D 02 0.259D 03		0.438D 03 0.170D 02 0.259D 03			
4	13	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.102D-04 0.963D-05	0.235D-04 0.250D-04		0.953D-05 0.558D-06 0.738D-05	0.510D-04 0.413D-05 0.607D-04		0.337D 03 0.131D 02 0.232D 03		0.337D 03 0.131D 02 0.232D 03			
4	14	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.84D-05 0.63D-05 0.754D-05	0.199D-04 0.212D-04 0.210D-04		0.703D-05 0.414D-06 0.548D-05	0.358D-04 0.290D-05 0.425D-04		0.263D 03 0.103D 02 0.182D 03		0.263D 03 0.103D 02 0.182D 03			
5	8	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.392D-03 -0.358D-03 -0.400D-03	0.872D-03 0.876D-03 0.889D-03		0.100D 00 0.731D-02 0.111D 00	0.497D 00 0.354D-01 0.546D 00		0.707D 05 0.363D 04 0.715D 05		0.707D 05 0.363D 04 0.715D 05			
5	9	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.222D-04 -0.289D-04 -0.285D-04	0.114D-03 0.124D-03 0.119D-03		0.146D-03 0.176D-04 0.256D-03	0.337D-02 0.322D-03 0.449D-02		0.504D 03 0.408D 02 0.750D 03		0.504D 03 0.408D 02 0.750D 03			
5	10	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.224D-05 -0.127D-05 -0.102D-05	0.590D-04 0.620D-04 0.647D-04		0.229D-05 0.378D-06 0.162D-07	0.766D-03 0.904D-03 0.656D-04		0.143D 02 0.210D 01 0.678D-01		0.143D 02 0.210D 01 0.678D-01			
5	11	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.677D-05 0.581D-05 0.521D-05	0.391D-04 0.432D-04 0.410D-04		0.145D-04 0.453D-06 0.546D-05	0.287D-03 0.251D-04 0.338D-03		0.124D 03 0.259D 01 0.418D 02		0.124D 03 0.259D 01 0.418D 02			
5	12	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.583D-05 0.750D-05 0.656D-05	0.286D-04 0.318D-04 0.302D-04		0.165D-04 0.684D-06 0.683D-05	0.149D-03 0.123D-04 0.166D-03		0.172D 03 0.475D 01 0.819D 02		0.172D 03 0.475D 01 0.819D 02			
5	13	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.561D-05 0.770D-05 0.723D-05	0.229D-04 0.249D-04 0.236D-04		0.147D-04 0.675D-06 0.892D-05	0.774D-04 0.703D-05 0.955D-04		0.535D 03 0.943D 02 0.101D 04		0.535D 03 0.943D 02 0.101D 04			
5	14	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.502D-05 0.741D-05 0.658D-05	0.181D-04 0.202D-04 0.193D-04		0.124D-04 0.597D-06 0.795D-05	0.500D-04 0.445D-05 0.606D-04		0.162D 03 0.521D 01 0.922D 02		0.162D 03 0.521D 01 0.922D 02			

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2f_j - md^2j$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. THE EXPERIMENTAL VALUES OBTAINED BY AGNEW OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN.

N	M	TRANSITION		f_{ij}^{pe}		f_{ij}		f_{ij}		f_{ij}		$\Lambda_{ij}(\text{sec}^{-1})$	
		J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH	VELOCITY	LENGTH	
6	9	5/2 - 3/2	-0.466D-03	0.009D-03	0.181D 00	0.439D 00			0.451D 05	0.227D 06			
		5/2 - 5/2	-0.414D-03	0.013D-03	0.133D-01	0.648D-01			0.228D 04	0.111D 05			
		7/2 - 5/2	-0.425D-03	0.067D-03	0.210D 00	0.097D 00			0.460D 05	0.202D 06			
6	10	5/2 - 3/2	-0.266D-04	0.478D-04	0.346D-03	0.110D-02			0.496D 03	0.143D 04			
		5/2 - 5/2	-0.462D-04	0.894D-04	0.110D-02	0.411D-02			0.128D 04	0.480D 04			
		7/2 - 5/2	-0.342D-04	0.580D-04	0.402D-04	0.116D-03			0.351D 02	0.101D 03			
6	11	5/2 - 3/2	-0.666D-05	0.193D-04	0.263D-04	0.131D-03			0.645D 02	0.322D 03			
		5/2 - 5/2	-0.133D-04	0.243D-04	0.445D-04	0.155D-04			0.712D 01	0.255D 02			
		7/2 - 5/2	-0.176D-04	0.442D-04	0.116D-03	0.733D-03			0.255D 03	0.161D 04			
6	12	5/2 - 3/2	-0.352D-05	0.118D-04	0.366D-05	0.410D-04			0.128D 02	0.143D 03			
		5/2 - 5/2	-0.642D-05	0.161D-04	0.850D-06	0.542D-05			0.202D 01	0.127D 02			
		7/2 - 5/2	-0.666D-05	0.288D-04	0.236D-04	0.261D-03			0.736D 02	0.810D 03			
6	13	5/2 - 3/2	-0.168D-05	0.790D-05	0.740D-06	0.164D-04			0.322D 01	0.715D 02			
		5/2 - 5/2	-0.400D-05	0.110D-04	0.301D-06	0.228D-05			0.876D 00	0.666D 01			
		7/2 - 5/2	-0.476D-05	0.209D-04	0.639D-05	0.124D-03			0.248D 02	0.480D 03			
6	14	5/2 - 3/2	-0.811D-06	0.559D-05	0.161D-06	0.703D-05			0.814D 00	0.387D 02			
		5/2 - 5/2	-0.280D-05	0.797D-05	0.137D-06	0.111D-05			0.463D 00	0.374D 01			
		7/2 - 5/2	-0.276D-05	0.163D-04	0.200D-05	0.921D-04			0.961D 01	0.312D 03			
7	10	5/2 - 3/2	-0.212D-03	0.160D-03	0.778D-01	0.441D-01			0.758D 04	0.451D 04			
		5/2 - 5/2	-0.411D-03	0.808D-03	0.305D 00	0.118D 01			0.287D 05	0.111D 06			
		7/2 - 5/2	-0.211D-03	0.158D-03	0.540D-02	0.302D-02			0.380D 03	0.213D 03			
7	11	5/2 - 3/2	-0.825D-04	0.146D-03	0.499D-02	0.157D-01			0.280D 04	0.881D 04			
		5/2 - 5/2	-0.637D-04	0.145D-03	0.365D-03	0.109D-02			0.138D 03	0.411D 03			
		7/2 - 5/2	-0.640D-04	0.949D-04	0.228D-02	0.706D-02			0.115C 04	0.355D 04			
7	12	5/2 - 3/2	-0.370D-04	0.117D-03	0.727D-03	0.712D-02			0.758D 03	0.795D 04			
		5/2 - 5/2	-0.396D-04	0.115D-03	0.569D-04	0.437D-03			0.418D 02	0.366D 03			
		7/2 - 5/2	-0.241D-04	0.509D-04	0.330D-03	0.145D-02			0.325D 03	0.143D 04			
7	13	5/2 - 3/2	-0.171D-04	0.950D-04	0.127D-03	0.391D-02			0.204D 03	0.628D 04			
		5/2 - 5/2	-0.187D-04	0.943D-04	0.108D-04	0.275D-03			0.116D 02	0.256D 03			
		7/2 - 5/2	-0.141D-04	0.348D-04	0.924D-04	0.560D-03			0.112D 03	0.802D 03			
7	14	5/2 - 3/2	-0.705D-05	0.794D-04	0.193D-04	0.442D-02			0.396D 02	0.496D 04			
		5/2 - 5/2	-0.653D-05	0.700D-04	0.199D-05	0.171D-03			0.213D 01	0.234D 03			
		7/2 - 5/2	-0.536D-05	0.264D-04	0.359D-04	0.235D-03			0.657D 02	0.521D 03			
8	11	5/2 - 3/2	-0.154D-04	0.216D-04	0.596D-03	0.118D-02			0.282D 02	0.558D 02			
		5/2 - 5/2	-0.780D-05	0.108D-04	0.108D-04	0.374D-04			0.351D 00	0.187D 01			
		7/2 - 5/2	-0.153D-03	0.127D-03	0.627D-01	0.433D-01			0.272D 04	0.188D 04			
8	12	5/2 - 3/2	-0.203D-05	0.305D-04	0.416D-05	0.136D-03			0.117D 01	0.265D 03			
		5/2 - 5/2	-0.261D-05	0.273D-04	0.513D-06	0.564D-04			0.926D-01	0.102D 02			
		7/2 - 5/2	-0.544D-04	0.126D-03	0.334D-02	0.180D-01			0.806D 03	0.434D 04			
8	13	5/2 - 3/2	0.396D-05	0.278D-04	0.119D-04	0.375D-03			0.645D 01	0.313D 03			
		5/2 - 5/2	0.581D-05	0.246D-04	0.179D-05	0.321D-04			0.653D 00	0.117D 02			
		7/2 - 5/2	-0.164D-04	0.103D-03	0.271D-03	0.851D-02			0.131D 03	0.413D 04			
8	14	5/2 - 3/2	0.728D-05	0.260D-04	0.324D-04	0.382D-03			0.263D 02	0.311D 03			
		5/2 - 5/2	0.893D-05	0.250D-04	0.347D-05	0.220D-04			0.149D 01	0.120D 02			
		7/2 - 5/2	-0.261D-05	0.861D-04	0.443D-05	0.483D-02			0.321D 01	0.350D 04			

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2F_j - m^2D_j$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. THE EXPERIMENTAL VALUES OBTAINED BY AGNEW OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	M	TRANSITION	M_{ij}^2		f_{ij}		f_{ij}		f_{ij}		$A_{ij}(\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH	VELOCITY	LENGTH
9	12	5/2 - 3/2	0.158D-C4	0.167D-05	0.883D-03	0.396D-05			0.211D 02	0.238D 00		
		5/2 - 5/2	0.230D-04	-0.189D-05	0.132D-03	0.393D-06			0.217D 01	0.146D-01		
		7/2 - 5/2	0.236D-05	0.123D-04	0.209D-04	0.571D-03			0.455D 00	0.125D 02		
9	13	5/2 - 3/2	0.124D-04	0.843D-05	0.226D-03	0.104D-03			0.316D 02	0.146D 02		
		5/2 - 5/2	0.187D-04	0.579D-05	0.366D-04	0.351D-05			0.344D 01	0.330D 00		
		7/2 - 5/2	0.704D-05	0.195D-04	0.778D-C4	0.596D-03			0.976D 01	0.744D 02		
9	14	5/2 - 3/2	0.104D-04	0.626D-05	0.110D-03	0.401D-04			0.319D 02	0.116D 02		
		5/2 - 5/2	0.119D-04	0.293D-05	0.103D-04	0.623D-06			0.159D 01	0.121D 00		
		7/2 - 5/2	0.103D-04	0.197D-04	0.116D-03	0.424D-03			0.361D 02	0.110D 03		
10	13	5/2 - 3/2	0.156D-04	0.259D-05	0.185D-02	0.325D-04			0.240D 02	0.418D 00		
		5/2 - 5/2	0.263D-04	-0.595D-05	0.236D-03	0.121D-04			0.210D 01	0.107D 00		
		7/2 - 5/2	0.185D-04	-0.566D-06	0.175D-02	0.164D-05			0.268D 02	0.193D-01		
10	14	5/2 - 3/2	0.133D-04	0.146D-05	0.350D-03	0.419D-05			0.271D 02	0.324D 00		
		5/2 - 5/2	0.203D-04	-0.440D-06	0.576D-04	0.273D-07			0.300D 01	0.142D-02		
		7/2 - 5/2	0.127D-04	0.363D-05	0.340D-03	0.278D-04			0.236D 02	0.193D 01		
11	14	5/2 - 3/2	0.162D-04	-0.526D-05	0.169D-02	0.177D-03			0.125D 02	0.131D 01		
		5/2 - 5/2	0.265D-04	-0.876D-05	0.190D-03	0.345D-04			0.963D 00	0.175D 00		

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_J - mf^2F_1$, CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY SICNE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

N	TRANSITION		P_{IJ}		f_{IJ}		f_{IJ}		f_{IJ}		$\Lambda_{IJ} (\text{sec}^{-1})$	
	M	J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH
5	4	3/2 - 5/2	-0.215D-02	C.214D-02	0.355D 00	0.379D 00	0.302D 00	0.174D 08	0.174D 08	0.174D 08	0.167D 08	
		5/2 - 5/2	-0.221D-02	0.219D-02	0.194D-01	0.191D-01	C.324D 00	0.126D 07	0.126D 07	0.126D 07	0.124D 07	
		5/2 - 7/2	-0.219D-02	0.218D-02	0.342D 00	0.379D 00		0.186D 08	0.186D 08	0.186D 08	0.185D 08	
5	5	3/2 - 5/2	-0.190D-02	C.162D-02	0.238D 00	0.174D 00	0.122D 00	0.165D 08	0.165D 08	0.165D 08	0.120D 08	
		5/2 - 5/2	-0.189D-02	C.160D-02	0.113D-01	0.431D-02	0.127D 00	0.115D 07	0.115D 07	0.115D 07	0.868D 06	
		5/2 - 7/2	-0.186D-02	0.163D-02	0.220D 00	0.168D 00		0.168D 00	0.168D 00	0.168D 00	0.129D 08	
5	6	3/2 - 5/2	-0.110D-02	C.983D-03	0.725D-01	0.576D-01	0.627D-01	0.712D-01	0.712D-01	0.712D-01	0.488D 07	
		5/2 - 5/2	-0.110D-02	C.790D-03	0.345D-02	0.240D-02	0.650D-01	0.434D 06	0.434D 06	0.434D 06	0.352D 06	
		5/2 - 7/2	-0.129D-02	C.110D-02	0.945D-01	0.637D-01		0.453D 07	0.453D 07	0.453D 07	0.649D 07	
5	7	3/2 - 5/2	-0.231D-02	C.130D-02	0.301D 00	0.455D-01	0.373D-01	0.418D-01	0.418D-01	0.418D-01	0.901D 07	
		5/2 - 5/2	-0.222D-02	C.127D-02	0.133D-01	0.432D-02	0.363D-01	0.409D-01	0.409D-01	0.409D-01	0.612D 06	
		5/2 - 7/2	-0.166D-02	C.860D-03	0.603D-01	0.399D-01		0.637D 07	0.637D 07	0.637D 07	0.422D 07	
5	8	3/2 - 5/2	-0.224D-03	-0.153D-03	0.571D-02	0.127D-02	0.253D-01	0.252D-01	0.252D-01	0.252D-01	0.130D 06	
		5/2 - 5/2	-0.277D-03	-0.176D-03	0.200D-03	0.304D-04	0.241D-01	0.303D 05	0.303D 05	0.303D 05	0.122D 05	
		5/2 - 7/2	-0.208D-02	C.105D-02	0.226D 00	0.576D-01		0.258D 09	0.258D 09	0.258D 09	0.657D 07	
5	9	3/2 - 5/2	0.475D-04	-0.340D-03	0.406D-03	0.614D-02	0.160D-01	0.173D-01	0.173D-01	0.173D-01	0.661D 06	
		5/2 - 5/2	0.115D-03	-0.355D-03	0.337D-04	0.320D-03	0.164D-01	0.537D 04	0.537D 04	0.537D 04	0.508D 05	
		5/2 - 7/2	-0.273D-03	-C.139D-03	0.378D-02	0.989D-03		0.451D 06	0.451D 06	0.451D 06	0.118D 06	
5	10	3/2 - 5/2	0.185D-03	-C.344D-03	0.179D-02	0.617D-02	0.114D-01	0.118D-01	0.118D-01	0.118D-01	0.686D 05	
		5/2 - 5/2	0.204D-03	-0.355D-03	0.104D-03	0.314D-03	0.115D-01	0.123D-01	0.123D-01	0.123D-01	0.517D 05	
		5/2 - 7/2	0.517D-04	-0.275D-03	0.344D-03	0.378D-02		0.165D 05	0.165D 05	0.165D 05	0.467D 06	
5	11	3/2 - 5/2	0.205D-03	-C.215D-03	0.217D-02	0.512D-02	0.821D-02	0.821D-02	0.821D-02	0.821D-02	0.593D 06	
		5/2 - 5/2	0.219D-03	-0.323D-03	0.119D-03	0.238D-03	0.820D-02	0.200D 05	0.200D 05	0.200D 05	0.434D 05	
		5/2 - 7/2	0.151D-02	-C.114D-02	0.954D 00	0.569D 00		0.157D 07	0.157D 07	0.157D 07	0.897D 06	
6	4	3/2 - 5/2	0.147D-02	-C.112D-02	0.460D-01	0.247D-01	0.157D 07	0.157D 07	0.157D 07	0.157D 07	0.604D 05	
		5/2 - 5/2	0.148D-02	-0.112D-02	0.937D 00	0.535D 00		0.159D 07	0.159D 07	0.159D 07	0.905D 06	
		5/2 - 7/2	-0.464D-03	-C.679D-04	0.405D-01	0.367D-03		0.346D 06	0.346D 06	0.346D 06	0.740D 04	
6	5	3/2 - 5/2	-0.502D-03	0.372D-04	0.228D-02	0.125D-04	0.624D-03	0.367D 06	0.367D 06	0.367D 06	0.157D 03	
		5/2 - 5/2	-0.464D-03	-0.587D-04	0.389D-01	0.624D-03		0.367D 06	0.367D 06	0.367D 06	0.587D 04	
		5/2 - 7/2	-0.375D-03	C.100D-03	0.202D-01	0.144D-02		0.256D 06	0.256D 06	0.256D 06	0.212D 05	
6	6	3/2 - 5/2	-0.402D-03	0.122D-03	0.111D-02	0.162D-03	0.241D 05	0.241D 05	0.241D 05	0.241D 05	0.221D 04	
		5/2 - 5/2	-0.553D-03	0.168D-03	0.422D-01	0.390D-02		0.686D 06	0.686D 06	0.686D 06	0.433D 05	
		5/2 - 7/2	-0.203D-02	C.871D-03	0.513D 00	0.754D-01		0.980D 07	0.980D 07	0.980D 07	0.192D 07	
6	7	3/2 - 5/2	-0.201D-02	C.984D-03	0.243D-01	0.471D-02	0.650D 06	0.650D 06	0.650D 06	0.650D 06	0.134D 06	
		5/2 - 5/2	-0.574D-03	0.236D-03	0.397D-01	0.672D-02		0.843D 06	0.843D 06	0.843D 06	0.143D 06	
		5/2 - 7/2	-0.749D-03	C.360D-03	0.653D-01	0.151D-01		0.146D 07	0.146D 07	0.146D 07	0.337D 06	
6	8	3/2 - 5/2	-0.715D-03	C.347D-03	0.288D-02	0.670D-03	0.955D 05	0.955D 05	0.955D 05	0.955D 05	0.222D 05	
		5/2 - 5/2	-0.206D-02	0.914D-03	0.472D 00	0.340D-01		0.117D 08	0.117D 08	0.117D 08	0.231D 07	
		5/2 - 7/2	-0.216D-03	0.157D-03	0.111D-01	0.273D-02		0.273D 06	0.273D 06	0.273D 06	0.674D 05	
6	9	3/2 - 5/2	-C.296D-03	0.143D-03	0.446D-03	0.109D-03	0.109D-03	0.109D-03	0.109D-03	0.109D-03	0.399D 04	
		5/2 - 5/2	-0.636D-03	0.288D-03	0.428D-01	0.376D-02		0.117D 07	0.117D 07	0.117D 07	0.241D 06	
		5/2 - 7/2	-0.155D-03	C.780D-04	0.258D-02	0.650D-03		0.682D 05	0.682D 05	0.682D 05	0.172D 05	
6	10	3/2 - 5/2	-0.134D-03	0.660D-04	0.916D-04	0.223D-04	0.223D-04	0.223D-04	0.223D-04	0.223D-04	0.876D 03	
		5/2 - 5/2	-0.273D-03	C.121D-03	0.760D-02	0.149D-02		0.224D 06	0.224D 06	0.224D 06	0.438D 05	
		5/2 - 7/2	-0.273D-03	C.121D-03	0.760D-02	0.149D-02		0.224D 06	0.224D 06	0.224D 06	0.438D 05	

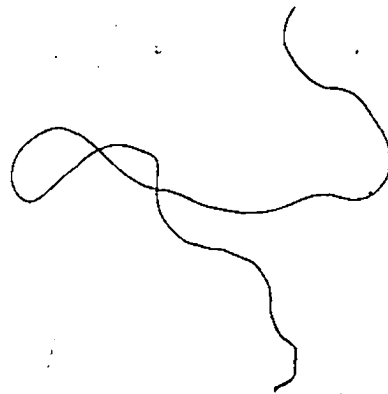
THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_j - m^2F_j$, CALCULATED USING THE CRYPGONALIZED RELATIVISTIC P-F-S WAVE FUNCTIONS FOR CESIUM. OSCILLATOR STRENGTHS FROM THE MODEL POTENTIAL CALCULATIONS PERFORMED BY STONE AND THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

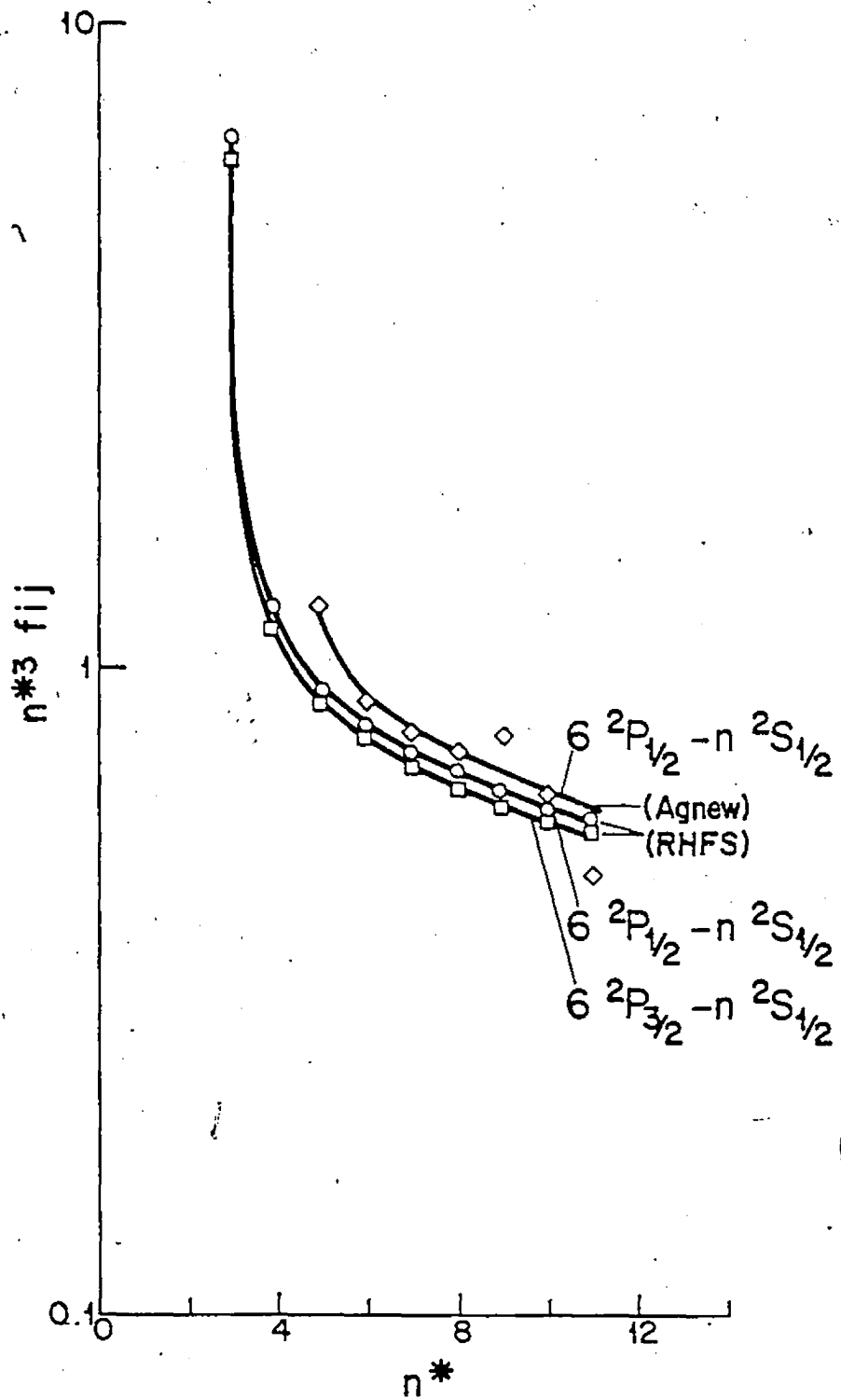
N	M	J-J'	\bar{P}_{IJ}^2		f_{IJ}		STONE	AGNEW	$A_{IJ} (\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH			VELOCITY	LENGTH
6	11	3/2 - 5/2	-0.812D-04	C.4C9D-04	0.690D-03	0.174D-03			0.152D 05	0.484D 04
		5/2 - 5/2	-0.633D-04	0.306D-04	0.200D-04	0.466D-05			0.823D 03	0.192D 03
7	5	3/2 - 5/2	0.122D-02	-C.940D-03	0.156D 01	0.789D 00			0.591D 06	0.299D 06
		5/2 - 5/2	0.131D-02	-C.926D-03	0.743D-01	0.373D-01			0.404D 05	0.203D 05
		5/2 - 7/2	0.132D-02	-0.927D-03	0.153D 01	0.748D 00			0.624D 06	0.305D 06
7	6	3/2 - 5/2	0.263D-03	-0.276D-03	0.250D-01	0.275D-01			0.579D 05	0.637D 05
		5/2 - 5/2	0.243D-03	-0.261D-02	0.103D-02	0.119D-02			0.351D 04	0.404D 04
		5/2 - 7/2	0.170D-03	-C.252D-03	0.598D-02	0.221D-01			0.255D 05	0.562D 05
7	7	3/2 - 5/2	-0.113D-02	C.485D-03	0.337D 00	0.625D-01			0.144D 07	0.267D 06
		5/2 - 5/2	-0.113D-02	0.52D-03	0.164D-01	0.321D-02			0.104D 06	0.203D 05
		5/2 - 7/2	-0.757D-04	-C.334D-04	0.162D-02	0.235D-03			0.769D 04	0.135D 04
7	8	3/2 - 5/2	-0.724D-03	C.462D-03	0.119D 00	0.485D-01			0.659D 06	0.284D 06
		5/2 - 5/2	-0.710D-03	0.457D-03	0.547D-02	0.227D-02			0.475D 05	0.197D 05
		5/2 - 7/2	-0.129D-02	C.664D-03	0.363D 00	0.958D-01			0.236D 07	0.624D 06
7	9	3/2 - 5/2	-0.413D-03	C.283D-03	0.352D-01	0.165D-01			0.250D 06	0.117D 06
		5/2 - 5/2	-0.398D-03	0.275D-03	0.157D-02	0.748D-03			0.165D 05	0.790D 04
		5/2 - 7/2	-0.596D-03	0.370D-03	0.703D-01	0.270D-01			0.554D 06	0.214D 06
7	10	3/2 - 5/2	-0.268D-03	C.190D-03	0.139D-01	0.899D-02			0.112D 06	0.562D 05
		5/2 - 5/2	-0.255D-03	0.183D-03	0.602D-03	0.310D-03			0.719D 04	0.371D 04
		5/2 - 7/2	-0.337D-03	0.222D-03	0.211D-01	0.914D-02			0.189D 06	0.819D 05
7	11	3/2 - 5/2	-0.185D-03	0.137D-03	0.661D-02	0.350D-02			0.563D 05	0.308D 05
		5/2 - 5/2	-0.178D-03	0.131D-03	0.280D-03	0.153D-03			0.366D 04	0.200D 04
8	6	3/2 - 5/2	0.121D-02	-0.784D-03	0.234D 01	0.376D 00			0.279D 06	0.117D 06
		5/2 - 5/2	0.121D-02	-0.774D-03	0.113D 00	0.465D-01			0.193D 05	0.798D 04
		5/2 - 7/2	0.117D-02	-0.778D-03	0.213D 01	0.337D 00			0.273D 06	0.120D 06
8	7	3/2 - 5/2	-0.424D-03	-C.401D-03	0.111D 00	0.989D-01			0.881D 05	0.785D 05
		5/2 - 5/2	-0.437D-03	-C.385D-03	0.566D-02	0.440D-02			0.666D 04	0.517D 04
		5/2 - 7/2	0.212D-03	-0.384D-03	0.576D-01	0.875D-01			0.504D 05	0.769D 05
8	8	3/2 - 5/2	-0.758D-03	0.413D-03	0.248D 00	0.773D-01			0.365D 06	0.120D 06
		5/2 - 5/2	-0.741D-03	0.414D-03	0.116D-01	0.363D-02			0.267D 05	0.835D 04
		5/2 - 7/2	-0.793D-03	-C.422D-04	0.266D 00	0.752D-03			0.459D 06	0.129D 04
8	9	3/2 - 5/2	-0.476D-03	C.279D-03	0.836D-01	0.237D-01			0.125D 06	0.637D 05
		5/2 - 5/2	-0.464D-03	C.272D-03	0.381D-02	0.131D-02			0.165D 05	0.430D 04
		5/2 - 7/2	-0.575D-03	C.250D-03	0.117D 00	0.222D-01			0.268D 06	0.546D 05
8	10	3/2 - 5/2	-0.325D-03	C.191D-03	0.350D-01	0.121D-01			0.967D 05	0.335D 05
		5/2 - 5/2	-0.315D-03	C.185D-03	0.157D-02	0.342D-03			0.645D 04	0.222D 04
		5/2 - 7/2	-0.367D-03	C.185D-03	0.426D-01	0.109D-01			0.131D 06	0.334D 05
8	11	3/2 - 5/2	-0.239D-03	C.140D-03	0.175D-01	0.606D-02			0.562D 05	0.194D 05
		5/2 - 5/2	-0.230D-03	0.135D-03	0.775D-03	0.267D-03			0.369D 04	0.127D 04
9	7	3/2 - 5/2	0.239D-03	-0.374D-03	0.147D 00	0.362D 00			0.670D 04	0.164D 05
		5/2 - 5/2	0.230D-03	-0.368D-03	0.663D-02	0.171D-01			0.430D 03	0.111D 04
		5/2 - 7/2	C.101D-02	-0.635D-03	0.258D 01	0.101D 01			0.126D 06	0.492D 05

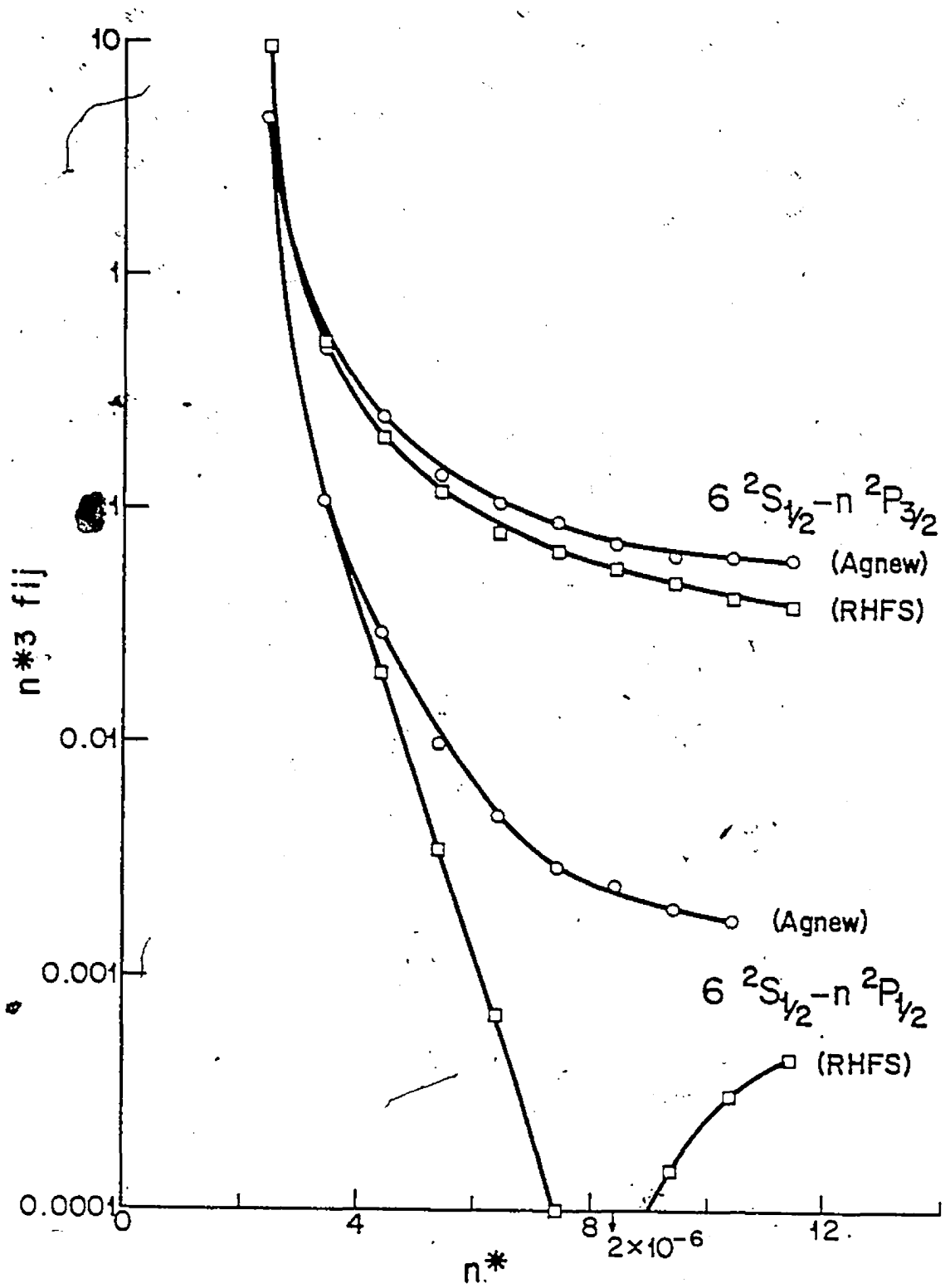
THEORETICAL RADIAL ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSMISSION PROBABILITIES, FOR THE SERIES $nd^2D_J - m^2F_J$ CALCULATED USING THE ORTHOGONALIZED RELATIVISTIC H-F-S WAVE FUNCTIONS FOR CESIUM. THE EXPERIMENTAL VALUES OBTAINED BY AGNEW ARE ALSO SHOWN

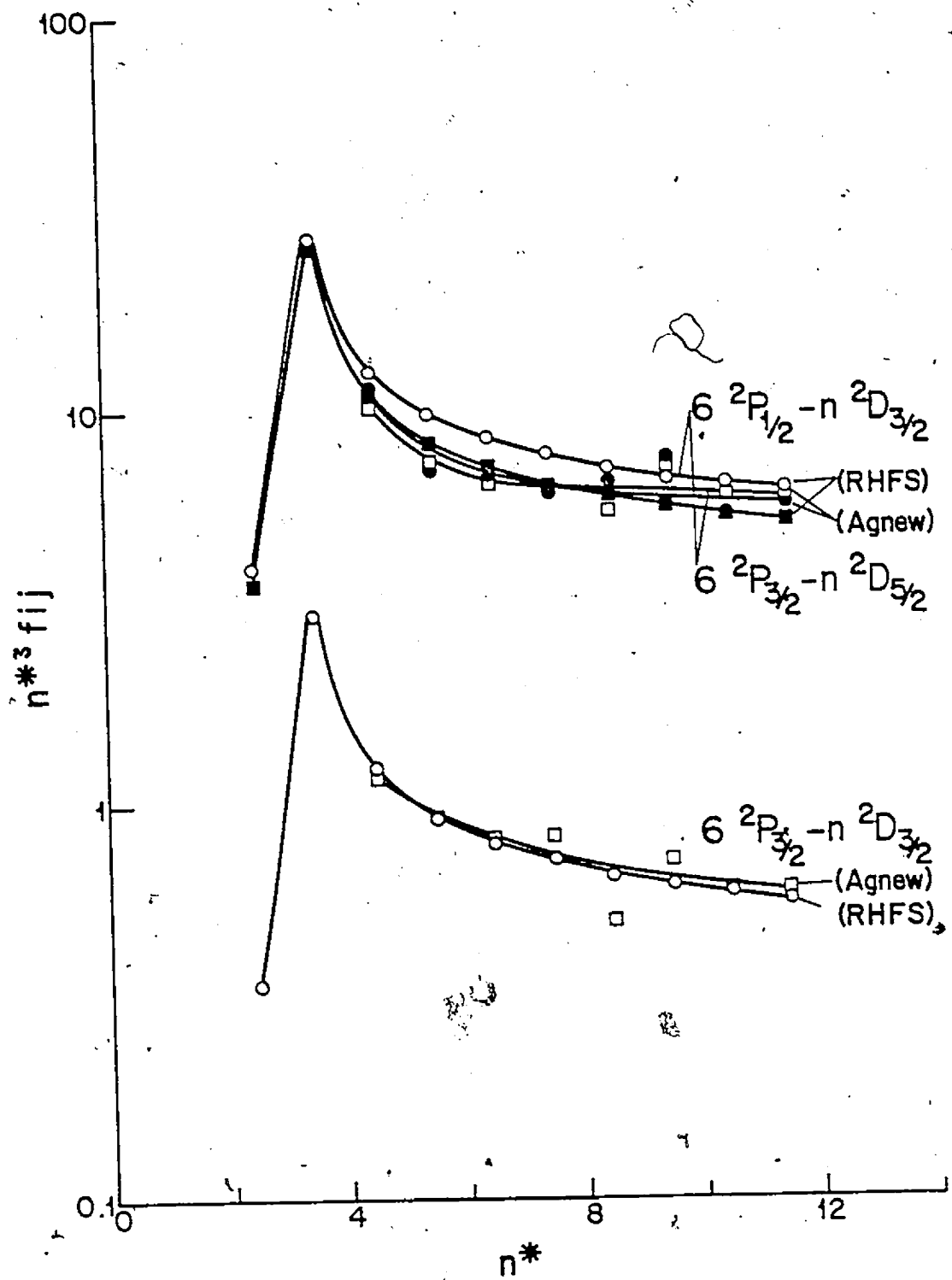
N	M	TRANSITION $J-J'$	P_{IJ}		f_{IJ}		f_{IJ}		f_{IJ}	
			VELOCITY	LENGTH	VELOCITY	LENGTH	STONE	AGNEW	VELOCITY	LENGTH
9	6	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.108D-02 -0.108D-02 -0.388D-03	0.130D-02 0.131D-02 -0.182D-03	0.113D 01 0.548D-01 0.140D 00	0.104D 01 0.903D-01 0.908D-01			0.365D 06 0.260D 05 0.500D 05	0.529D 06 0.361D 05 0.110D 05
9	9	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.555D-03 -0.587D-03 -0.617D-03	0.518D-03 0.512D-03 0.442D-03	0.241D 00 0.112D-01 0.247D 00	0.182D 00 0.852D-02 0.105D 00			0.158D 06 0.109D 05 0.180D 06	0.119D 06 0.827D 04 0.769D 05
9	10	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.391D-03 -0.383D-03 -0.405D-03	0.300D-03 0.255D-03 0.267D-03	0.855D-01 0.392D-02 0.897D-01	0.503D-01 0.233D-02 0.383D-01			0.827D 05 0.562D 04 0.963D 05	0.488D 05 0.334D 04 0.411D 05
9	11	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.283D-03 -0.275D-03 -0.314D-03	0.204D-03 0.199D-03 0.592D-03	0.395D-01 0.179D-02 0.385D 00	0.207D-01 0.935D-03 0.137D 01			0.488D 05 0.329D 04 0.761D 04	0.255D 05 0.172D 04 0.270D 05
10	8	5/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.325D-03 0.124D-03 -0.548D-03	-0.576D-03 -0.328D-03 0.127D-02	0.201D 01 0.588D-01 0.129D 01	0.633D-01 0.411D 00 0.233D 01			0.566D 03 0.124D 04 0.189D 06	0.179D 04 0.872D 04 0.341D 06
10	9	5/2 - 5/2 5/2 - 7/2	-0.550D-03 -0.922D-03	0.128D-02 0.120D-02	0.622D-01 0.117D 01	0.114D 00 0.198D 01			0.134D 05 0.150D 06	0.246D 05 0.321D 06
10	10	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.499D-03 -0.493D-03 -0.506D-03	0.468D-03 0.463D-03 0.480D-03	0.246D 00 0.115D-01 0.242D 00	0.236D 00 0.110D-01 0.218D 00			0.761D 05 0.529D 04 0.835D 05	0.728D 05 0.508D 04 0.748D 05
10	11	3/2 - 5/2 5/2 - 5/2	-0.336D-03 -0.329D-03	0.292D-03 0.289D-03	0.905D-01 0.416D-02	0.685D-01 0.321D-02			0.422D 05 0.289D 04	0.321D 05 0.223D 04
11	9	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.245D-03 -0.255D-03 -0.317D-03	-0.532D-03 -0.535D-03 -0.535D-03	0.673D 00 0.347D-01 0.553D 00	0.169D 01 0.779D-01 0.157D 01			0.627D 04 0.471D 03 0.562D 04	0.160D 05 0.106D 04 0.160D 05
11	10	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.819D-03 -0.821D-03 -0.817D-03	0.117D-02 0.118D-02 0.145D-02	0.136D 01 0.657D-01 0.130D 01	0.277D 01 0.135D 00 0.256D 01			0.956D 05 0.711D 04 0.105D 06	0.203D 06 0.146D 05 0.208D 06
11	11	3/2 - 5/2 5/2 - 5/2	-0.431D-03 -0.427D-03	0.450D-03 0.445D-03	0.256D 00 0.120D-01	0.280D 00 0.131D-01			0.407D 05 0.284D 04	0.446D 05 0.309D 04
12	10	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.329D-03 -0.328D-03 -0.326D-03	-0.507D-03 -0.493D-03 -0.458D-03	0.846D 00 0.435D-01 0.807D 00	0.201D 01 0.266D-01 0.189D 01			0.418D 04 0.308D 03 0.430D 04	0.992D 04 0.657D 03 0.100D 05
12	11	3/2 - 5/2 5/2 - 5/2	-0.717D-03 -0.719D-03	0.106D-02 0.107D-02	0.142D 01 0.687D-01	0.314D 01 0.153D 00			0.562D 05 0.399D 04	0.124D 06 0.893D 04
13	11	3/2 - 5/2 5/2 - 5/2	-0.306D-03 -0.315D-03	-0.472D-03 -0.459D-03	0.986D 00 0.508D-01	0.234D 01 0.108D 00			0.270D 04 0.200D 03	0.641D 04 0.422D 03

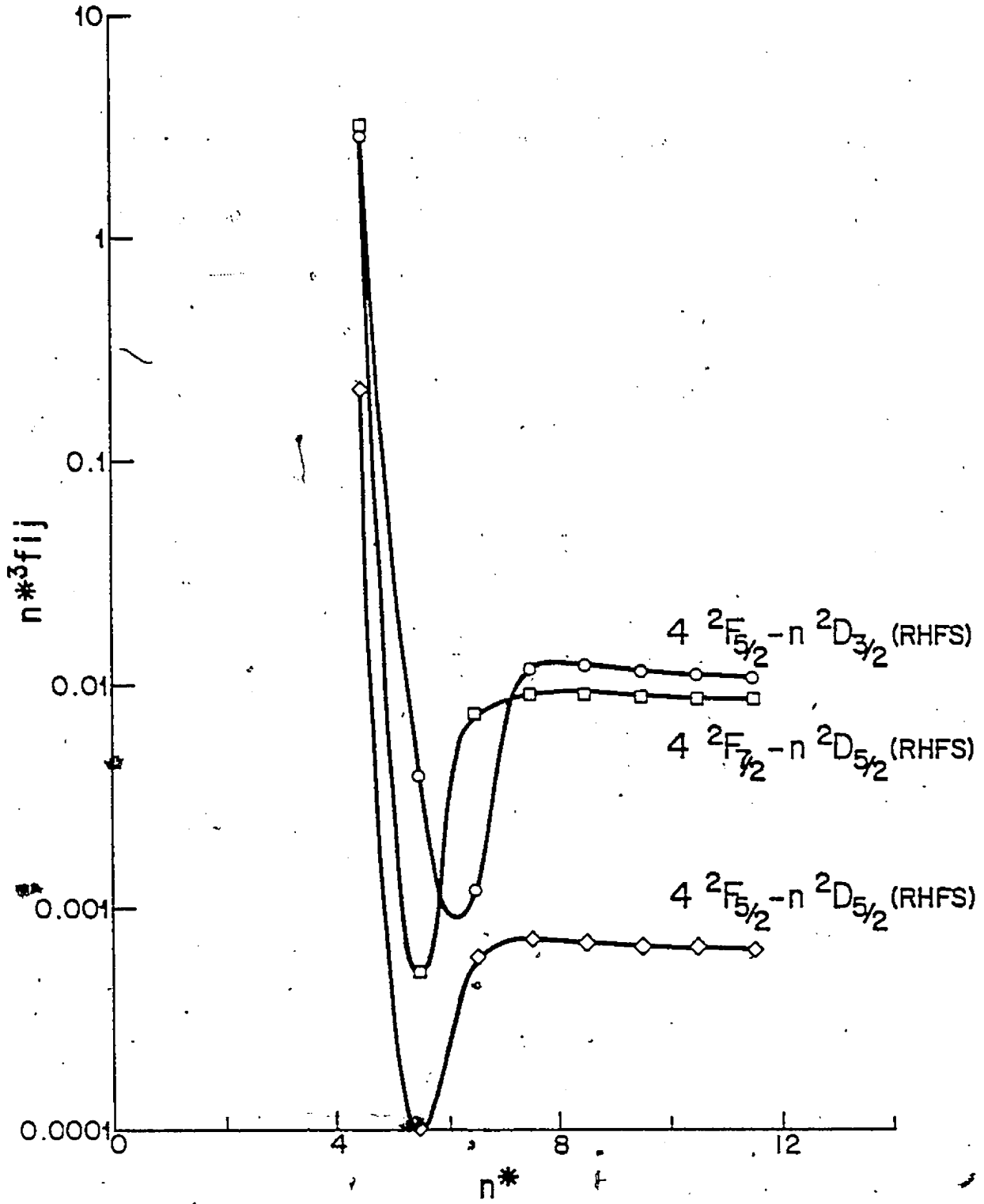
APPENDIX 6 : Plots of $n^3 f_{ij}$ versus n^* indicating the behavior of the relativistic oscillator strengths as a function of the effective principal quantum number for several Cesium series. The oscillator strengths were calculated from the orthogonalized free-core Hartree-Fock-Slater wave functions and are taken from Appendix (5). The oscillator strengths correspond to those derived in the velocity formulation. The theoretical curves are denoted by (RHFS) and the corresponding plots formed from Agnew's experimental data are designated by (AGNEW).











APPENDIX 7: CESIUM_I LIFETIMES

Atomic lifetimes for Cs_I are calculated from both the length and velocity forms of the transition probabilities determined from the HFS wave functions and from the orthogonalized HFS wave functions. These lifetimes are compared to those found from the Bates-Damgaard (B-D) transition probabilities shown in Appendix (13), and to existing experimental values.

STATE	NON-ORTHOGONAL WAVE FUNCTIONS			ORTHOGONAL WAVE FUNCTIONS				$\tau_{EXPT}(nsec)$
	$\tau_{VEL}(nsec)$	$\tau_{LNG}(nsec)$	$\tau_{AVG}(nsec)$	$\tau_{VEL}(nsec)$	$\tau_{LNG}(nsec)$	$\tau_{AVG}(nsec)$	$\tau_{B-D}(nsec)$	
7 ² S _{1/2}	60.7	36.8	48.8	38.6	47.4	43	56.2	
8 ² S _{1/2}	164	125	144	74.7	152	113		87 ⁺ ₉ (b); 96 ⁺ ₁₄ (c)
9 ² S _{1/2}	339	283	311	135	341	238	188	147 ⁺ ₁₅ (b); 231 ⁺ ₃₅ (e)
10 ² S _{1/2}	618	577	598	208	591	400	313	260 ⁺ ₁₂ (a); 270 ⁺ ₅ (f)
11 ² S _{1/2}	1020	1010	1015	315	962	639	492	343 ⁺ ₂₂ (a); 411 ⁺ ₈ (f)
12 ² S _{1/2}	1570	1600	1585	461	1471	966	738	545 ⁺ ₃₀ (a); 517 ⁺ ₁₅ (f)
13 ² S _{1/2}	2470	2840	2655	673	2515	1594	5340	754 ⁺ ₃₅ (f)

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ORTHOGONAL WAVE FUNCTIONS

NON-ORTHOGONAL WAVE FUNCTIONS

STATE	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{B-D}}(\text{nsec})$	$\tau_{\text{EXPT}}(\text{nsec})$
$14^2S_{1/2}$	3580	3940	3760	911	3402	2157	7959	$959^{+50}(f)$
$15^2S_{1/2}$	4690	4640	4665	1203	4464	2232	11177	
$6^2P_{1/2}$	35.1	29.2	32.2	35.1	29.2	32.2	34.7	
$6^2P_{3/2}$	30.6	25.5	28.1	30.6	25.5	28.1	30.6	
$7^2P_{1/2}$	203	147	175	137	181	159	143	$158^{+5}(d)$; $158^{+3}(f)$
$7^2P_{3/2}$	162	157	159	107	141	124	116	$136^{+4}(d)$; $135^{+3}(f)$
$8^2P_{1/2}$	668	256	462	449	753	601	456	
$8^2P_{3/2}$	601	32.6	317	301	521	411	371	$274(e)$
$9^2P_{1/2}$	1150	3700	2425	1109	1750	2859	656	
$9^2P_{3/2}$	1280	502	891	682	1270	976	563	$502(e)$
$10^2P_{1/2}$	1700	2220	1960	2204	3121	2663	1044	
$10^2P_{3/2}$	2120	720	1420	1299	2461	1880	906	
$11^2P_{1/2}$	2770	7660	5215	3780	4789	4284	1594	o
$11^2P_{3/2}$	3150	1010	2080	2190	4056	3123	1359	

STATE	NON-ORTHOGONAL WAVE FUNCTIONS				ORTHOGONAL WAVE FUNCTIONS			
	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{B-D}}(\text{nsec})$	$\tau_{\text{EXPT}}(\text{nsec})$
$12^2P_{1/2}$	4020	1060	2540	5868	6777	6322	1723	
$12^2P_{3/2}$	4140	1390	2765	3390	6085	4738	1529	
$13^2P_{1/2}$	4340	1660	3000	7176	6404	6790	1754	
$13^2P_{3/2}$	6980	2220	4600	4156	6085	5121	1609	
$14^2P_{1/2}$	5960	2070	4015	10496	9917	10206	1837	
$14^2P_{3/2}$	7360	2670	5015	6038	8719	7378	1722	
$15^2P_{1/2}$	6480	2390	4435	14507	12186	13346	1979	
$15^2P_{3/2}$	12300	2900	7600	8298	11956	10127	4359	
$5^2D_{3/2}$	951	957	954	952	958	955	845	
$5^2D_{5/2}$	1180	1320	1250	1174	1321	1248	1202	
$6^2D_{3/2}$	27	18.2	22.6	29.6	37.4	33.5	72.8	
$6^2D_{5/2}$	28.2	19	23.6	30.6	38.6	34.6	71.1	
$7^2D_{3/2}$	68.4	58	63.2	55.8	75.3	65.6	106	$98^{+10}(b)$
$7^2D_{5/2}$	72.2	60.4	66.3	58.1	83.8	71	104	$88^{+9}(b)$
$8^2D_{3/2}$	137	143	140	90.4	157	124	167	$152^{+3}(f)$
$8^2D_{5/2}$	144	73.3	109	95.9	163	129	164	

ORTHOGONAL WAVE FUNCTIONS

NON-ORTHOGONAL WAVE FUNCTIONS

STATE	$\tau_{VEL}(nsec)$	$\tau_{LNG}(nsec)$	$\tau_{AVG}(nsec)$	$\tau_{VEL}(nsec)$	$\tau_{LNG}(nsec)$	$\tau_{AVG}(nsec)$	$\tau_{BLD}(nsec)$	$\tau_{EXPT}(nsec)$
$9^2D_{3/2}$	237	266	251	137	260	198	116	$218 \pm 4 (r)$
$9^2D_{5/2}$	255	283	269	146	275	210	252	
$10^2D_{3/2}$	386	389	388	207	440	324	373	$311 \pm 6 (r)$
$10^2D_{5/2}$	413	494	453	221	445	333	369	
$11^2D_{3/2}$	579	761	670	300	658	479	526	$428 \pm 12 (r)$
$11^2D_{5/2}$	629	818	723	324	603	464	520	
$12^2D_{3/2}$	843	748	795	421	939	680	719	$561 \pm 18 (r)$
$12^2D_{5/2}$	857	1170	1014	454	987	720	713	
$13^2D_{3/2}$	1080	1000	1040	571	1298	934	958	$741 \pm 22 (r)$
$13^2D_{5/2}$	1110	1680	1395	619	1397	1008	957	
$14^2D_{3/2}$	1370	1250	1310	753	1732	1242	1104	$980 \pm 30 (r)$
$14^2D_{5/2}$	1400	750	1075	818	1863	1340	1084	
$4^2F_{5/2}$	44.4	52.6	48.5	49.2	52.9	51.1	46.7	
$4^2F_{7/2}$	45	51.2	48.1	49.5	51.5	50.5	46.8	
$5^2F_{5/2}$	55.8	80.5	68.1	53.6	75.8	64.7	80.3	
$5^2F_{7/2}$	58.5	80.7	69.6	56.2	75.7	66	79.9	

NON-ORTHOGONAL WAVE FUNCTIONS

ORTHOGONAL WAVE FUNCTIONS

STATE	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{B-D}}(\text{nsec})$	$\tau_{\text{EXPT}}(\text{nsec})$
$6^2F_{5/2}$	115	131	123	137	184	160	130	
$6^2F_{7/2}$	98.7	139	119	101	76.1	88.6	129	
$7^2F_{5/2}$	237	147	192	234	83.5	158	198	
$7^2F_{7/2}$	140	214	177	135	222	178	198	
$8^2F_{5/2}$	327	252	290	268	654	461	288	
$8^2F_{7/2}$	359	260	310	24.8	105	64.9	288	
$9^2F_{5/2}$	476	359	417	847	673	760	404	
$9^2F_{7/2}$	511	420	466	352	960	656	403	
$10^2F_{5/2}$	664	630	647	1265	826	1046	547	
$10^2F_{7/2}$	721	543	682	1177	1042	1110	548	
$11^2F_{5/2}$	932	1000	968	1635	1071	1353	724	

a) Lundberg and Svanberg⁸⁰ d) Pace and Atkinson⁸³

b) Marek⁸¹ e) Bulos, Gupta and Happer⁸⁴

c) Marek and Niemax⁸² f) Deech, Laypaert, Pordrill and Series⁸⁵

APPENDIX 8 : Thallium Oscillator Strengths

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_j - ms^2S_{1/2}$ CALCULATED USING THE CRITICALIZED M-F-S WAVE FUNCTIONS. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY MORTON AND GALLAGHER AND PENKIN AND SHABANOVA (MG-PS) ARE ALSO SHOWN.

N	M	TRANSITION		\bar{M}_{ij}^0		f_{ij}		f_{ij}		f_{ij}		$A_{ij}(\text{sec}^{-1})$	
		J-J'	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	MIGDALEK	MG-PS	VELOCITY	LENGTH		
6	7	1/2 - 1/2	0.125D-02	-0.226D-02	0.182D 00	0.880D-01	0.131D 00	0.136D 00	0.412D 08	0.952D C9			
		3/2 - 1/2	0.315D-02	-0.208D-02	0.243D 00	0.106D 00	0.176D 00	0.154D 00	0.456D 08	0.113D 09			
6	8	1/2 - 1/2	0.143D-02	-0.115D-02	0.242D-01	0.155D-01	0.166D-01	0.179D-01	0.155D 09	0.242D 08			
		3/2 - 1/2	0.110D-02	-0.091D-03	0.179D-01	0.120D-01	0.199D-01	0.139D-01	0.153D 09	0.229D 09			
6	9	1/2 - 1/2	0.0912D-03	-0.0792D-03	0.382D-02	0.665D-02	0.579D-02	0.630D-02	0.827D 07	0.110D 08			
		3/2 - 1/2	0.682D-03	-0.617D-03	0.002D-02	0.493D-02	0.521D-02	0.490D-02	0.623D 07	0.100D 08			
6	10	1/2 - 1/2	0.0643D-03	-0.0583D-03	0.118D-02	0.344D-02	0.285D-02	0.310D-02	0.471D 07	0.571D 07			
		3/2 - 1/2	0.477D-03	-0.445D-03	0.278D-02	0.253D-02	0.258D-02	0.220D-02	0.475D 07	0.521D 07			
6	11	1/2 - 1/2	0.0484D-03	-0.0450D-03	0.231D-02	0.199D-02	0.156D-02	0.156D-02	0.267D 07	0.312D 07			
		3/2 - 1/2	0.357D-03	-0.352D-03	0.151D-02	0.147D-02	0.136D-02	0.122D-02	0.292D 07	0.302D 07			
7	9	1/2 - 1/2	0.0141D-02	-0.0158D-02	0.198D 00	0.393D 00	0.269D 00	0.303D 00	0.550D 07	0.277D 07			
		3/2 - 1/2	0.134D-02	-0.1196D-02	0.229D 00	0.492D 00	0.303D 00	0.303D 00	0.343D 07	0.363D 07			
7	1	1/2 - 1/2	0.0591D-03	-0.0555D-03	0.178D-01	0.463D-01	0.266D-01	0.266D-01	0.251D 07	0.963D 06			
		3/2 - 1/2	0.502D-03	-0.4662D-03	0.144D-01	0.425D-01	0.164D-01	0.164D-01	0.363D 07	0.123D 07			
7	10	1/2 - 1/2	0.0361D-03	-0.0333D-03	0.337D-02	0.165D-01	0.769D-02	0.769D-02	0.136D 07	0.442D 06			
		3/2 - 1/2	0.299D-03	-0.2567D-03	0.405D-02	0.145D-01	0.546D-02	0.546D-02	0.159D 07	0.554D 06			
7	11	1/2 - 1/2	0.0251D-03	-0.02465D-03	0.180D-02	0.806D-02	0.352D-02	0.352D-02	0.914D 05	0.240D 06			
		3/2 - 1/2	0.203D-03	-0.1616D-03	0.175D-02	0.702D-02	0.237D-02	0.237D-02	0.119D 07	0.297D 06			
8	9	1/2 - 1/2	0.0112D-02	-0.0150D-02	0.321D 00	0.576D 00	0.321D 00	0.321D 00	0.124D 07	0.690D 06			
		3/2 - 1/2	0.105D-02	-0.08145D-02	0.352D 00	0.675D 00	0.352D 00	0.352D 00	0.183D 07	0.955D 06			
8	10	1/2 - 1/2	0.0433D-03	-0.0475D-03	0.213D-01	0.531D-01	0.213D-01	0.213D-01	0.546D 06	0.224D 06			
		3/2 - 1/2	0.332D-03	-0.274D-03	0.142D-01	0.424D-01	0.142D-01	0.142D-01	0.715D 06	0.240D 05			
8	11	1/2 - 1/2	0.0263D-03	-0.0249D-03	0.230D-02	0.181D-01	0.230D-02	0.230D-02	0.313D 05	0.107D 05			
		3/2 - 1/2	0.193D-03	-0.137D-03	0.362D-02	0.138D-01	0.362D-02	0.362D-02	0.410D 05	0.107D 05			
9	10	1/2 - 1/2	0.0935D-03	-0.0812D-02	0.480D 00	0.747D 00	0.480D 00	0.480D 00	0.418D 06	0.245D-06			
		3/2 - 1/2	0.877D-03	-0.716D-03	0.480D 00	0.843D 00	0.480D 00	0.480D 00	0.608D 06	0.345D 06			
9	11	1/2 - 1/2	0.0374D-03	-0.0351D-03	0.308D-01	0.671D-01	0.308D-01	0.308D-01	0.153D 06	0.885D 05			
		3/2 - 1/2	0.281D-03	-0.2454D-03	0.190D-01	0.498D-01	0.190D-01	0.190D-01	0.238D 06	0.910D 05			
10	11	1/2 - 1/2	0.0783D-03	-0.0710D-02	0.907D 00	0.907D 00	0.907D 00	0.907D 00	0.162D 06	0.967D 05			
		3/2 - 1/2	0.727D-03	-0.561D-03	0.102D 01	0.102D 01	0.102D 01	0.102D 01	0.237D 06	0.134D 06			

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2 S_{1/2} - mp^2 P_{1/2}$, CALCULATED USING THE CRITICALIZED P-F-S WAVE FUNCTIONS. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NORTON AND GALLAGHER AND PENKIN AND SHABANOVA (NG-PS) ARE ALSO SHOWN.

N	M	TRANSITION		M_{IJ}^0		f_{IJ}		f_{IJ}		f_{IJ}		$A_{IJ}(\text{sec}^{-1})$	
		$J-J'$	$J-J'$	VELOCITY	LENGTH	VELOCITY	LENGTH	MIGDALEK	NG-PS	VELOCITY	LENGTH	VELOCITY	LENGTH
7	7	1/2 - 1/2	1/2 - 3/2	0.281D-02	-0.272D-02	0.471D 00	0.442D 00	0.463D 00		0.186D 08	0.174D 08	0.260D 08	0.222D 08
7	9	1/2 - 1/2	1/2 - 3/2	0.513D-02	-0.289D-02	0.104D 01	0.384D 00	0.937D 00		0.112D 07	0.212D 06	0.376D 07	0.134D 07
7	9	1/2 - 1/2	1/2 - 3/2	0.895D-03	-0.216D-03	0.761D-02	0.143D-02	0.547D-01		0.309D 06	0.571D 04	0.153D 07	0.343D 06
7	10	1/2 - 1/2	1/2 - 3/2	0.218D-03	-0.249D-03	0.140D-01	0.314D-02	0.154D-01		0.123D 06	0.537D 03	0.754D 05	0.131D 06
8	8	1/2 - 1/2	1/2 - 3/2	0.144D-03	0.951D-03	0.488D-03	0.213D-05	0.940D-03		0.315D 07	0.252D 07	0.426D 07	0.111D 07
8	9	1/2 - 1/2	1/2 - 3/2	0.364D-03	-0.148D-03	0.621D-02	0.103D-02	0.674D-02		0.459D 06	0.175D 06	0.112D 07	0.342D 06
8	10	1/2 - 1/2	1/2 - 3/2	0.138D-02	-0.177D-02	0.688D 00	0.549D 00			0.168D 05	0.358D 05	0.525D 06	0.196D 06
9	9	1/2 - 1/2	1/2 - 3/2	0.517D-03	-0.319D-03	0.218D-01	0.328D-02			0.857D 06	0.645D 06	0.120D 07	0.839D 06
9	10	1/2 - 1/2	1/2 - 3/2	0.752D-03	-0.552D-03	0.397D-01	0.480D-01			0.391D 06	0.180D 06	0.178D 06	0.641D 05
10	10	1/2 - 1/2	1/2 - 3/2	0.276D-03	-0.128D-01	0.487D-02	0.104D-02			0.258D 06	0.212D 06	0.402D 06	0.280D 06
				0.155D-02	-0.132D-02	0.295D-01	0.110D-01						
				0.168D-02	-0.140D-02	0.910D 00	0.056D 00						
				0.667D-03	-0.453D-03	0.184D 01	0.129D 01						
				0.456D-03	-0.275D-03	0.142D 00	0.053D-01						
				0.123D-02	-0.104D-02	0.108D 01	0.772D 00						
				0.133D-02	-0.111D-02	0.216D 01	0.150D 01						

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_j - mp^2P_j'$, CALCULATED USING THE CRYSTAL-FIELD WAVE FUNCTIONS. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NORTON AND GALLAGHER AND PENKIN AND SHARANOVA (NG-PS) ARE ALSO SHOWN.

N	M	TRANSITION		R_{ij}^2			f_{ij}			f_{ij}			f_{ij}			$A_{ij} (\text{sec}^{-1})$		
		J'	J	VELOCITY	LENGTH	LENGTH	VELOCITY	LENGTH	LENGTH	MIGDALEK	NG-PS	VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	
6	8	3/2	1/2	0.590D-03	-C.1C5D-02	0.304D-01	0.358D-01	0.304D-01	0.358D-01	0.112D 07	0.352D C7	0.112D 07	0.352D C7	0.612D 05	0.295D C6	0.612D 05	0.295D C6	
		3/2	3/2	0.422B-03	-C.912D-03	0.290D-02	0.135D-01	0.290D-02	0.135D-01	0.512D 06	0.267D 07	0.512D 06	0.267D 07	0.236D 06	0.889D C6	0.236D 06	0.889D C6	
		5/2	3/2	0.449D-03	-0.937D-03	0.200D-01	0.870D-01	0.200D-01	0.870D-01	0.112D 05	0.831D 05	0.112D 05	0.831D 05	0.119D 06	0.769D 06	0.119D 06	0.769D 06	
6	9	3/2	1/2	0.216D-03	-C.419D-03	0.260D-02	0.974D-02	0.260D-02	0.974D-02	0.852D 05	0.379D C6	0.852D 05	0.379D C6	0.346D 04	0.382D 05	0.346D 04	0.382D 05	
		3/2	3/2	0.147D-03	-C.4C2D-03	0.235D-03	0.175D-02	0.235D-03	0.175D-02	0.442D 06	0.716D 06	0.442D 06	0.716D 06	0.442D 06	0.716D 06	0.442D 06	0.716D 06	
		5/2	3/2	0.161D-03	-0.409D-03	0.170D-02	0.110D-01	0.170D-02	0.110D-01	0.204D 05	0.500D 05	0.204D 05	0.500D 05	0.231D 06	0.525D C6	0.231D 06	0.525D C6	
6	10	3/2	1/2	C.115D-C3	-C.251D-C3	0.663D-03	0.295D-02	0.663D-03	0.295D-02	0.135D 01	0.278D 06	0.135D 01	0.278D 06	0.135D 01	0.278D 06	0.135D 01	0.278D 06	
		3/2	3/2	C.756D-04	-C.251D-C3	0.527D-04	0.582D-03	0.527D-04	0.582D-03	0.533D 04	0.208D C5	0.533D 04	0.208D C5	0.641D 05	0.210D 06	0.641D 05	0.210D 06	
		5/2	3/2	C.850D-04	-0.255D-03	C.404D-03	0.163D-02	C.404D-03	0.163D-02	0.173D 06	0.703D 04	0.173D 06	0.703D 04	0.852D 05	0.146D 06	0.852D 05	0.146D 06	
7	9	3/2	1/2	0.552D-03	-0.714D-03	0.589D-01	0.385D-01	0.589D-01	0.385D-01									
		3/2	3/2	0.362D-03	-0.567D-03	0.471D-02	0.116D-01	0.471D-02	0.116D-01									
		5/2	3/2	C.405D-03	-C.616D-03	0.366D-01	0.129D-01	0.366D-01	0.129D-01									
7	10	3/2	1/2	0.237D-03	-C.340D-03	0.656D-02	0.135D-01	0.656D-02	0.135D-01									
		3/2	3/2	0.147D-03	-C.291D-03	0.490D-C3	0.192D-02	0.490D-C3	0.192D-02									
		5/2	3/2	0.171D-03	-0.3C9D-03	C.402D-02	0.131D-01	C.402D-02	0.131D-01									
8	10	3/2	1/2	0.472D-03	-0.535D-03	C.8C7D-01	0.103D 00	C.8C7D-01	0.103D 00									
		3/2	3/2	C.290D-03	-0.397D-03	0.560D-02	0.105D-01	0.560D-02	0.105D-01									
		5/2	3/2	0.335D-03	-0.444D-03	0.467D-01	0.301D-01	0.467D-01	0.301D-01									

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_j - md^2D_j$, CALCULATED USING THE ORTHOGONALIZED H-F-S WAVE FUNCTIONS. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NORTON AND GALLAGHER AND PENKIN AND SHAMUNOVA (NG-PS) ARE ALSO SHOWN.

N	M	TRANSITION	P_{ij}^{ns}		f_{ij}^{ns}		f_{ij}^{ns}		f_{ij}^{ns}		$A_{ij}^{ns} (\text{sec}^{-1})$	
			VELOCITY	LENGTH	VELOCITY	LENGTH	MIGDALEK	NG-PS	VELOCITY	LENGTH	VELOCITY	LENGTH
6	6	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.395D-02 0.395D-02 0.395D-02	-0.426D-02 -0.450D-02 -0.444D-02	0.383D 00 0.550D-01 0.438D 00	0.460D 00 0.573D 00 0.573D 00	0.410D 00 0.550D-01 0.470D 00	0.290D-00 0.400D-01 0.350D 00	0.167D 09 0.270D 09 0.157D 09	0.200D 09 0.351D 08 0.205D 04	0.167D 09 0.270D 09 0.157D 09	
6	7	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.230D-02 0.299D-02 0.294D-02	-0.317D-02 -0.295D-02 -0.292D-02	0.237D 00 0.240D-01 0.208D 00	0.219D 00 0.213D-01 0.206D 00	0.110D 00 0.130D-01 0.110D 00	0.740D-01 0.910D-02 0.810D-01	0.140D 09 0.182D 08 0.109D 09	0.129D 09 0.137D 09 0.137D 09	0.140D 09 0.182D 08 0.109D 09	
6	8	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.254D-02 0.219D-02 0.219D-02	-0.225D-02 -0.156D-02 -0.198D-02	0.132D 00 0.119D-01 0.107D 00	0.103D 00 0.950D-02 0.973D-01	0.490D-01 0.520D-02 0.450D-01	0.290D-01 0.400D-02 0.290D-01	0.261D 09 0.108D 09 0.645D 04	0.696D 08 0.860D 07 0.529D 08	0.261D 09 0.108D 09 0.645D 04	
6	9	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.189D-02 0.139D-02 0.160D-02	-0.159D-02 -0.134D-02 -0.137D-02	0.709D-01 0.625D-02 0.554D-01	0.304D-01 0.260D-02 0.404D-01	0.260D-01 0.260D-02 0.230D-01	0.140D-01 0.200D-02 0.150D-01	0.504D 08 0.551D 07 0.361D 08	0.357D 08 0.416D 07 0.264D 08	0.504D 08 0.551D 07 0.361D 08	
7	6	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	-0.535D-03 -0.405D-03 -0.472D-03	0.849D-03 0.418D-03 0.458D-03	0.405D 00 0.160D-01 0.177D 00	0.137D 00 0.157D-01 0.167D 00	0.405D 00 0.160D-01 0.177D 00	0.137D 00 0.157D-01 0.167D 00	0.525D 06 0.980D 04 0.852D 05	0.430D 06 0.102D 05 0.798D 05	0.525D 06 0.980D 04 0.852D 05	
7	7	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.185D-02 0.231D-02 0.221D-02	-0.109D-02 -0.101D-02 -0.111D-03	0.418D 00 0.140D-01 0.583D 00	0.138D 00 0.251D-01 0.211D 00	0.418D 00 0.140D-01 0.583D 00	0.138D 00 0.251D-01 0.211D 00	0.860D 07 0.230D 07 0.123D 08	0.293D 07 0.819D 06 0.446D 07	0.860D 07 0.230D 07 0.123D 08	
7	8	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.161D-02 0.177D-02 0.173D-02	-0.909D-03 -0.101D-02 -0.101D-02	0.225D 00 0.301D-01 0.258D 00	0.719D-01 0.101D-01 0.874D-01	0.225D 00 0.301D-01 0.258D 00	0.719D-01 0.101D-01 0.874D-01	0.827D 07 0.612D 07 0.104D 08	0.265D 07 0.353D 07 0.265D 07	0.827D 07 0.612D 07 0.104D 08	
7	9	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.122D-02 0.129D-02 0.128D-02	-0.666D-03 -0.712D-03 -0.711D-03	0.114D 00 0.139D-01 0.123D 00	0.140D-01 0.250D-02 0.181D-01	0.114D 00 0.139D-01 0.123D 00	0.140D-01 0.250D-02 0.181D-01	0.542D 07 0.111D 07 0.653D 07	0.162D 07 0.339D 06 0.203D 07	0.542D 07 0.111D 07 0.653D 07	
8	7	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	-0.501D-03 -0.113D-03 -0.201D-03	0.538D-03 0.230D-03 0.265D-03	0.357D 00 0.431D-02 0.108D 00	0.413D 00 0.179D-01 0.108D 00	0.357D 00 0.431D-02 0.108D 00	0.413D 00 0.179D-01 0.108D 00	0.492D 05 0.210D 03 0.459D 04	0.571D 05 0.872D 03 0.794D 04	0.492D 05 0.210D 03 0.459D 04	
8	8	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.129D-02 0.158D-02 0.150D-02	-0.915D-03 -0.114D-02 -0.106D-02	0.461D 00 0.782E-01 0.627D 00	0.232D 00 0.403D-01 0.314D 00	0.461D 00 0.782E-01 0.627D 00	0.232D 00 0.403D-01 0.314D 00	0.168D 07 0.446D 06 0.243D 07	0.843D 06 0.231D 06 0.122D 07	0.168D 07 0.446D 06 0.243D 07	
8	9	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.582D-03 0.110D-02 0.107D-02	-0.658D-03 -0.729D-03 -0.708D-03	0.187D 00 0.252D-01 0.215D 00	0.432D-01 0.112D-01 0.347D-01	0.187D 00 0.252D-01 0.215D 00	0.432D-01 0.112D-01 0.347D-01	0.140D 07 0.320D 06 0.162D 06	0.624D 06 0.141D 06 0.802D 06	0.140D 07 0.320D 06 0.162D 06	
9	8	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	-0.265D-03 0.614D-04 -0.229D-04	0.401D-03 0.152D-03 0.183D-03	0.221D 00 0.313D-02 0.363D-02	0.266D 00 0.192D-01 0.211D 00	0.221D 00 0.313D-02 0.363D-02	0.266D 00 0.192D-01 0.211D 00	0.628D 04 0.273D 02 0.273D 02	0.143D 05 0.155D 04 0.160D 04	0.628D 04 0.273D 02 0.273D 02	
9	9	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.556D-03 0.119D-02 0.112D-02	-0.742D-03 -0.937D-03 -0.273D-03	0.847D 00 0.846D-01 0.674D 00	0.294D 00 0.524D-01 0.436D 00	0.847D 00 0.846D-01 0.674D 00	0.294D 00 0.524D-01 0.436D 00	0.480D 06 0.133D 06 0.719D 06	0.289D 06 0.923D 05 0.434D 06	0.480D 06 0.133D 06 0.719D 06	
10	9	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	-0.188D-03 0.676D-04 0.136D-04	0.331D-03 0.119D-03 0.140D-03	0.203D 00 0.128D-01 0.234D-02	0.332D 00 0.235D-01 0.243D 00	0.203D 00 0.128D-01 0.234D-02	0.332D 00 0.235D-01 0.243D 00	0.172D 04 0.257D 02 0.463D 01	0.533D 04 0.471D 02 0.460D 03	0.172D 04 0.257D 02 0.463D 01	

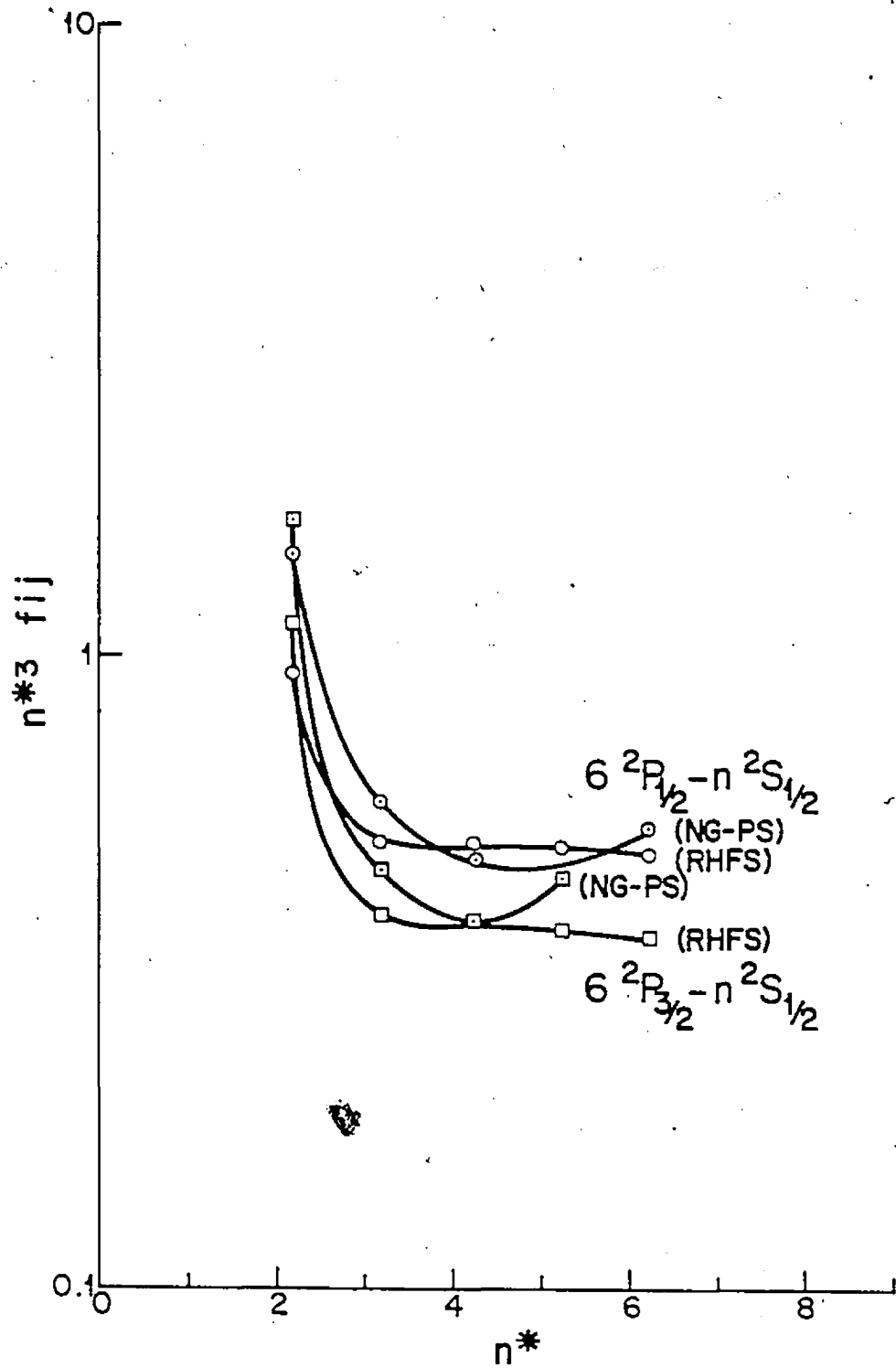
THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES. FOR THE SERIES $nd^2D_j - m^2F_j$ CALCULATED USING THE ORTHOGONALIZED P-F-S WAVE FUNCTIONS. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NOFTEN AND GALLAGHER AND PENKIN AND SHABANOVA (NG-PS) ARE ALSO SHOWN.

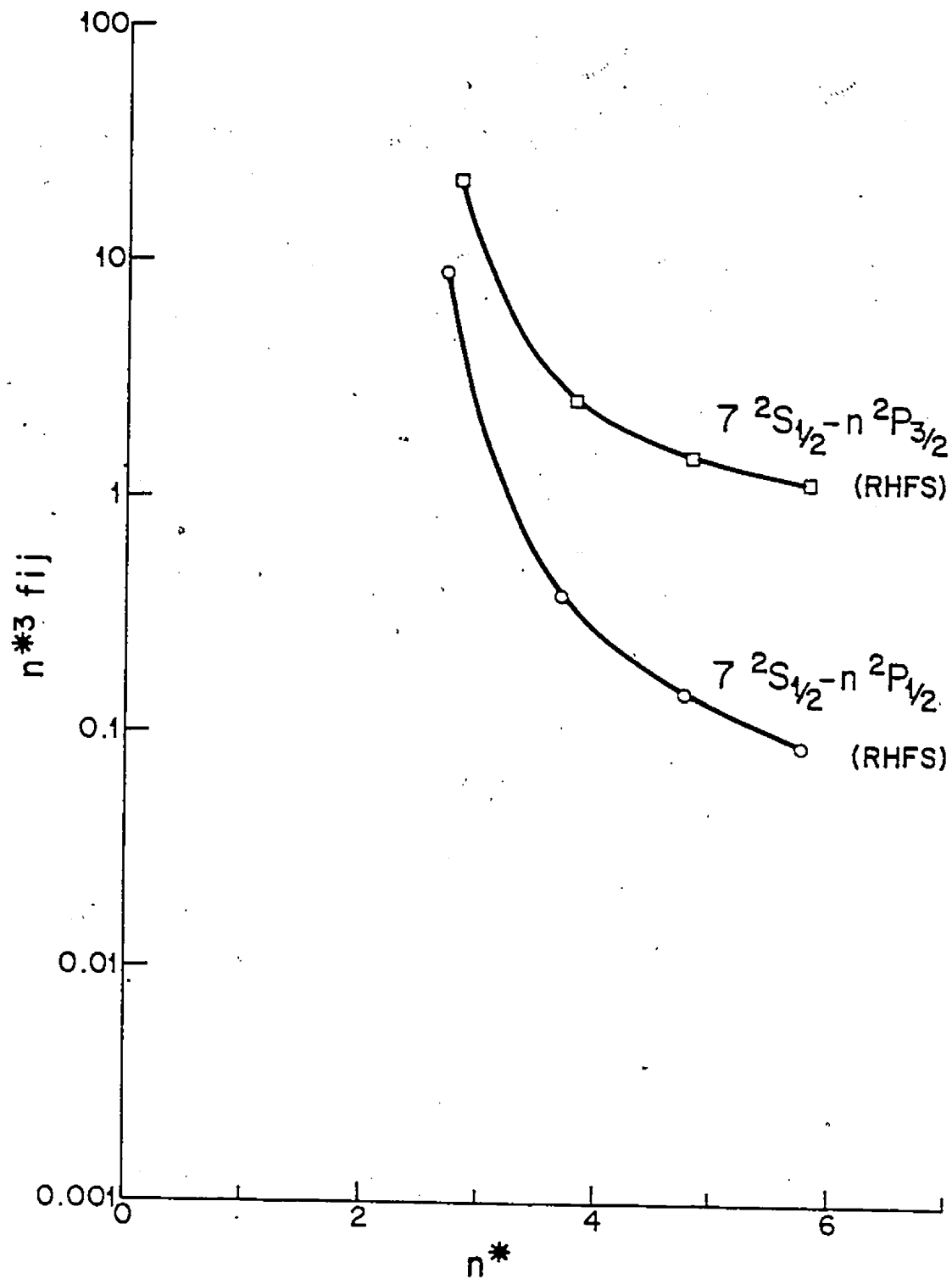
N	M	TRANSITION		M_{IJ}^2		f_{IJ}		f_{IJ}		f_{IJ}		$\Lambda_{IJ} \text{ (sec}^{-1}\text{)}$	
		J'	J	VELOCITY	LENGTH	VELOCITY	LENGTH	MIGDALEK	NG-PS	VELOCITY	LENGTH	VELOCITY	LENGTH
6	5	3/2	5/2	-0.2570-02	0.2740-02	0.8750 00	0.970 00			0.1500 08	0.1700 08	0.1500 08	0.1700 08
		5/2	5/2	-0.2570-02	0.2750-02	0.4240-01	0.4350-01			0.1660 07	0.1210 07	0.1660 07	0.1210 07
		5/2	7/2	-0.2570-02	0.2750-02	0.6470 00	0.9700 00			0.1590 09	0.1620 08	0.1590 09	0.1620 08
6	6	3/2	5/2	-0.1330-02	0.1380-02	0.1670 00	0.1800 00			0.5620 07	0.6040 07	0.5620 07	0.6040 07
		5/2	5/2	-0.1310-02	0.1360-02	0.7840-02	0.4470-02			0.3890 06	0.4220 06	0.3890 06	0.4220 06
		5/2	7/2	-0.1310-02	0.1360-02	0.1570 00	0.1990 00			0.5830 07	0.6280 07	0.5830 07	0.6280 07
7	5	3/2	5/2	0.2300-03	-0.2230-03	0.1420 00	0.1340 00			0.5550 04	0.5820 04	0.5550 04	0.5820 04
		5/2	5/2	0.1860-03	-0.1960-03	0.5020-02	0.5600-02			0.2430 03	0.2710 03	0.2430 03	0.2710 03
		5/2	7/2	0.1860-03	-0.1960-03	0.1000 00	0.1120 00			0.3650 04	0.4070 04	0.3650 04	0.4070 04
7	6	3/2	5/2	-0.1480-02	0.1200-02	0.6450 00	0.4230 00			0.2270 07	0.1480 07	0.2270 07	0.1480 07
		5/2	5/2	-0.1520-02	0.1260-02	0.3260-01	0.240-01			0.1670 06	0.1150 06	0.1670 06	0.1150 06
		5/2	7/2	-0.1520-02	0.1260-02	0.6530 00	0.4480 00			0.2510 07	0.1730 07	0.2510 07	0.1730 07
6	6	3/2	5/2	0.4830-03	-0.2060-03	0.1270 01	0.2330 00			0.1290 05	0.2360 04	0.1290 05	0.2360 04
		5/2	5/2	0.4220-03	-0.1800-03	0.5350-01	0.7090-02			0.6120 03	0.1110 03	0.6120 03	0.1110 03
		5/2	7/2	0.4220-03	-0.1800-03	0.1070 01	0.1940 00			0.9180 04	0.1660 04	0.9180 04	0.1660 04

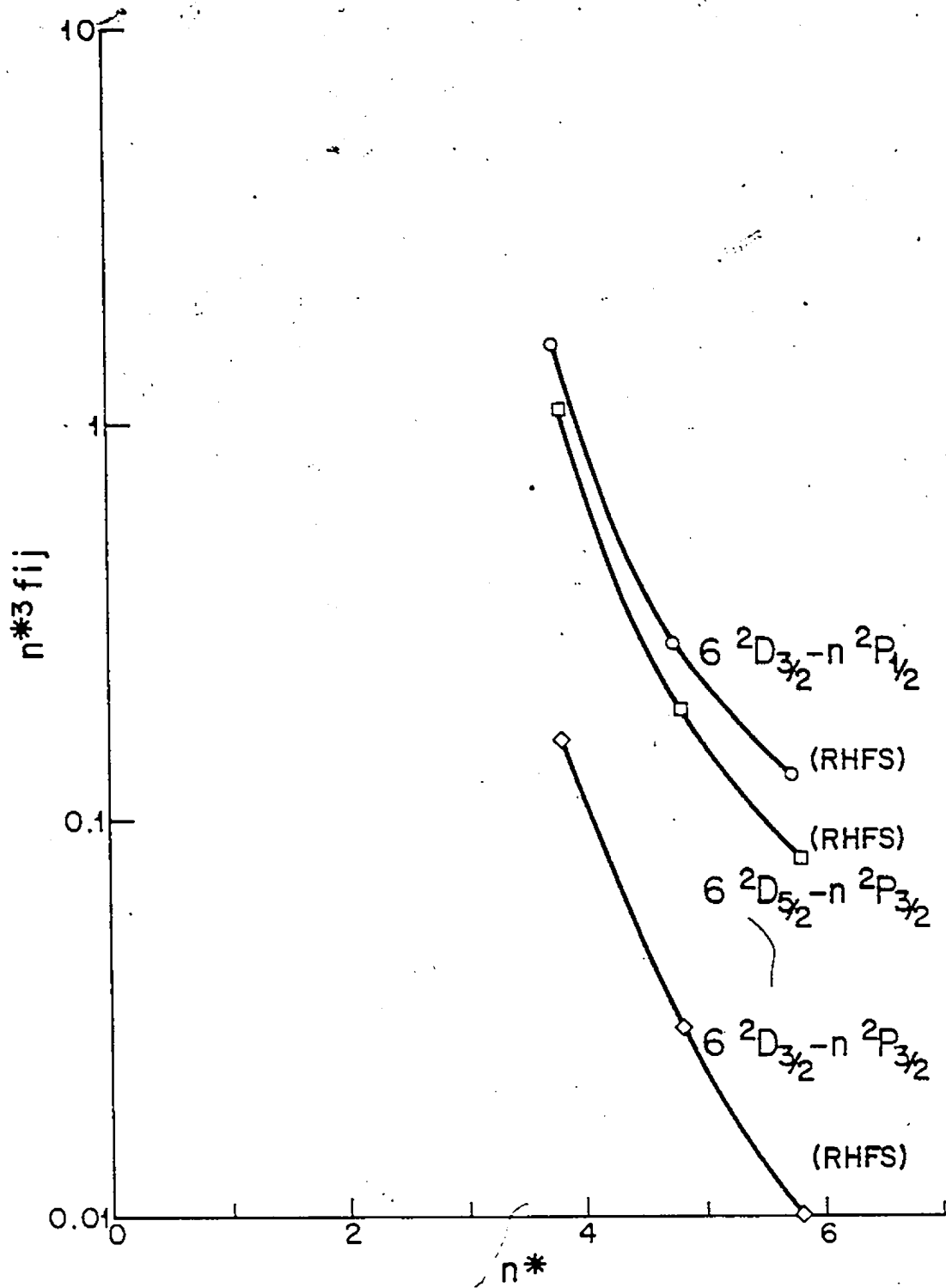
THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2 f_j - md^2 j_i$ CALCULATED USING THE ORTHOGONALIZED P-F-S WAVE FUNCTIONS. OSCILLATOR STRENGTHS FROM THE SEMI-EMPIRICAL CALCULATIONS PERFORMED BY MIGDALEK AND THE COMBINED EXPERIMENTAL DATA OBTAINED BY NORTON AND GALLAGHER AND PENKIN AND SHABANOVA (NG-PS) ARE ALSO SHOWN.

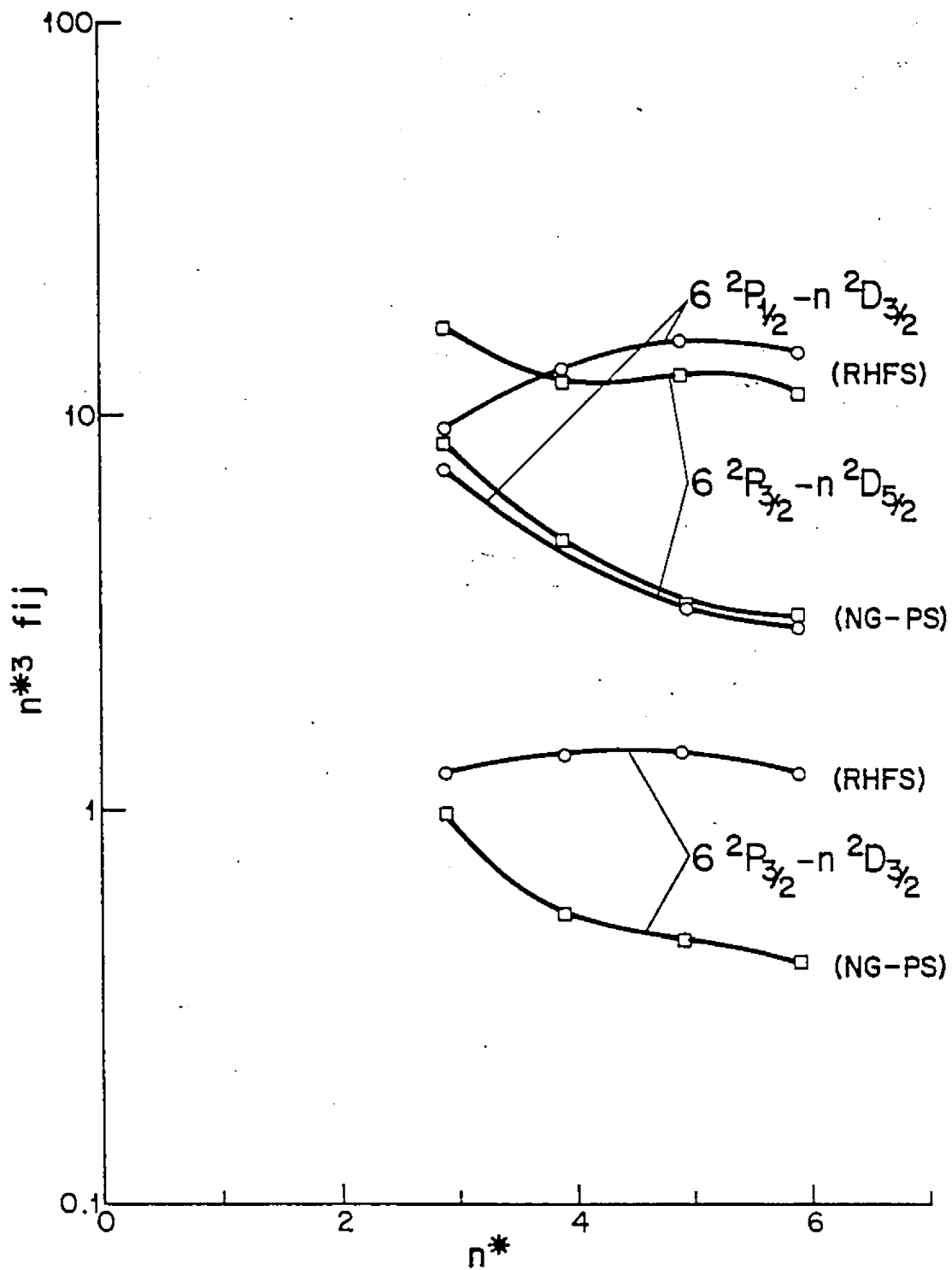
N	TRANSITION		M_{ij}^2		f_{ij}		f_{ij}		f_{ij}		$A_{ij}(\text{sec}^{-1})$	
	M	J-J'	VELOCITY	LENGTH	VELOCITY	LENGTH	VELOCITY	LENGTH	MIGDALEK	NG-PS	VELOCITY	LENGTH
5	8	5/2 - 3/2	-0.2110-03	0.5610-03	0.2250-01	0.7340-01	0.2250-01	0.7340-01			0.1250-06	0.4770-06
		5/2 - 5/2	-0.2940-03	0.5240-03	0.1430-02	0.4540-02	0.1430-02	0.4540-02			0.5370-04	0.1710-05
		7/2 - 5/2	-0.2940-03	0.5240-03	0.2140-01	0.6820-01	0.2140-01	0.6820-01			0.1170-06	0.3420-06
5	9	5/2 - 3/2	-0.1190-03	0.2370-03	0.2070-02	0.8200-02	0.2070-02	0.8200-02			0.2560-05	0.1170-06
		5/2 - 5/2	-0.1160-03	0.2300-03	0.1390-03	0.3300-03	0.1390-03	0.3300-03			0.1330-04	0.5250-04
		7/2 - 5/2	-0.1160-03	0.2300-03	0.2080-02	0.6240-02	0.2080-02	0.6240-02			0.2670-05	0.1050-06
6	9	5/2 - 3/2	-0.4120-03	0.7300-03	0.8030-01	0.2100-00	0.8030-01	0.2100-00			0.1710-05	0.3730-06
		5/2 - 5/2	-0.4120-03	0.6900-03	0.5210-02	0.1450-01	0.5210-02	0.1450-01			0.5750-04	0.1400-05
		7/2 - 5/2	-0.4120-03	0.6900-03	0.7820-01	0.2180-00	0.7820-01	0.2180-00			0.1150-06	0.3200-06

APPENDIX 9 : Plots of $n^{*3}f_{ij}$ versus n^* indicating the behavior of the relativistic oscillator strengths as a function of the effective principal quantum number for several Thallium, series. The oscillator strengths were calculated from the orthogonalized free-core Hartree-Fock-Slater wave functions and are taken from Appendix (8). The oscillator strengths correspond to those derived in the velocity formulation. The theoretical curves are denoted by (RHFS) and the corresponding plots formed from the combined experimental data of Norton and Gallagher, and Penkin and Shabanova are designated by (NG-PS).









APPENDIX 10: THALLIUM_I LIFETIMES

Atomic lifetimes for Tl_I are calculated from both the length and velocity forms of the transition probabilities determined from the HFS wave functions and from the orthogonalized wave functions. The lifetimes are compared to those found from the Bates-Damgaard (B-D) transition probabilities shown in Appendix (12), and to existing experimental values.

STATE	NON-ORTHOGONAL WAVE FUNCTIONS			ORTHOGONAL WAVE FUNCTIONS			$\tau_{B-D}(nsec)$	$\tau_{EXPT}(nsec)$
	$\tau_{VEL}(nsec)$	$\tau_{LNG}(nsec)$	$\tau_{AVG}(nsec)$	$\tau_{VEL}(nsec)$	$\tau_{LNG}(nsec)$	$\tau_{AVG}(nsec)$		
$7^2S_{1/2}$	11	5.1	8.1	11	5.1	8.1	16.9	$7.7^{+0.5(a)}$; $7.65^{(b)}$; $7.55^{+0.08(c)}$ $23^{+9(a)}$
$8^2S_{1/2}$	31.6	30.8	31.2	22.4	18.6	20.5	43	
$9^2S_{1/2}$	74.4	108	91.2	38.9	40.2	39.6	94	
$10^2S_{1/2}$	153	282	217	66.2	78.6	72.4	174	
$11^2S_{1/2}$	277	600	438	107	133	120		
$7^2P_{1/2}$	73.5	52.9	63.2	53.8	57.5	55.7	61.7	
$7^2P_{3/2}$	52.4	38.8	45.6	38.5	45	41.8	48.3	
$8^2P_{1/2}$	200	152	176	186	160	173	183	
$8^2P_{3/2}$	148	126	137	115	131	123	193	
$9^2P_{1/2}$	461	359	410	427	408	418	333	

NON-ORTHOGONAL WAVE FUNCTIONS ORTHOGONAL WAVE FUNCTIONS

STATE	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{VEL}}(\text{nsec})$	$\tau_{\text{LNG}}(\text{nsec})$	$\tau_{\text{AVG}}(\text{nsec})$	$\tau_{\text{B-D}}(\text{nsec})$	$\tau_{\text{EXPT}}(\text{nsec})$
$9^2P_{3/2}$	345	334	339	236	317	278	234	
$10^2P_{1/2}$	885	637	761	728	763	746		
$10^2P_{3/2}$	695	674	685	475	689	582		
$6^2D_{3/2}$	5.1	4.3	4.7	5.1	4.2	4.7	6.7	$6.8^{\pm 0.5}(a)$; $6.9^{\pm 0.4}(b)$;
$6^2D_{5/2}$	6.4	4.9	5.6	6.4	4.9	5.6	7.4	$7.6^{\pm 0.5}(a)$
$7^2D_{3/2}$	8.5	10.4	9.5	5.9	6.6	6.3	14.9	$16^{\pm 4}(a)$
$7^2D_{5/2}$	12	14.2	13.1	8.2	9.0	8.6	16.5	$19^{\pm 4}(a)$
$8^2D_{3/2}$	18.7	30.7	24.7	9.0	12.2	10.6	28	
$8^2D_{5/2}$	27.1	45.4	36.2	12.9	17.2	15.1	31.3	$50^{\pm 10}(a)$
$9^2D_{3/2}$	38.5	74.5	56.5	15.3	23	19.2		
$9^2D_{5/2}$	56.4	113	85	22.1	33.2	27.7		
$5^2F_{5/2}$	62.1	54.9	58.5	62.2	54.9	58.6	58.4	
$5^2F_{7/2}$	62.8	54.9	58.9	62.9	54.9	58.9	58.8	

NON-ORTHOGONAL WAVE FUNCTIONS ORTHOGONAL WAVE FUNCTIONS

STATE	$\tau_{VEL}(nsec)$	$\tau_{LANG}(nsec)$	$\tau_{AVG}(nsec)$	$\tau_{VEL}(nsec)$	$\tau_{LANG}(nsec)$	$\tau_{AVG}(nsec)$	$\tau_{B-D}(nsec)$	$\tau_{EXPT}(nsec)$
$6^2F_{5/2}$	128	121	125	118	124	121	107	
$6^2F_{7/2}$	129	130	130	120	125	123	109	

a) Anderson and Sørensen⁸⁶

b) Cunningham and Link⁸⁷

c) Hsieh⁸⁸

APPENDIX 11: COMPARISON OF OBSERVED AND PREDICTED ENERGIES FOR Tl_I AND Cs_I STATES

TABLE 1: THALLIUM_I

STATE	OBSERVED ENERGY(cm^{-1})	NON-ORTHOGONALIZED RELATIVISTIC HFS ENERGY(cm^{-1})	PERCENTAGE DIFFERENCE
$7^2S_{1/2}$	22786.7	22598.4	0.8
$8^2S_{1/2}$	10518.3	10997.2	4.6
$9^2S_{1/2}$	6098.0	6472.3	6.1
$10^2S_{1/2}$	3967.4	4228.6	6.6
$11^2S_{1/2}$	2807.3	2968.2	5.7
$6^2P_{1/2}$	49264.2	42110.3	14.5
$6^2P_{3/2}$	41471.5	35293.6	14.9
$7^2P_{1/2}$	15104.3	16116.7	6.7
$7^2P_{3/2}$	14103.1	15136.2	7.3
$8^2P_{1/2}$	7896.1	8448.9	7.0
$8^2P_{3/2}$	7523.4	8060.0	7.1
$9^2P_{1/2}$	4883.3	5249.6	7.5
$9^2P_{3/2}$	4701.7	5044.1	7.3
$10^2P_{1/2}$	3324.9	3560.9	7.1
$10^2P_{3/2}$	3220.6	3443.9	6.9
$6^2D_{3/2}$	13146.3	12652.4	3.8
$6^2D_{5/2}$	13064.3	12564.9	3.8
$7^2D_{3/2}$	7252.8	7593.3	4.7
$7^2D_{5/2}$	7215.2	7505.3	4.0
$8^2D_{3/2}$	4591.6	4961.7	8.1

TABLE 1: THALLIUM_I

STATE	OBSERVED ENERGY(cm^{-1})	NON-ORTHOGONALIZED RELATIVISTIC HFS ENERGY(cm^{-1})	PERCENTAGE DIFFERENCE
$8^2D_{5/2}$	4571.5	4907.7	7.4
$9^2D_{3/2}$	3165.7	3428.5	8.3
$9^2D_{5/2}$	3153.9	3397.2	7.7
$5^2F_{5/2}$	6945.8	6876.7	1.0
$5^2F_{7/2}$	6945.8	6876.7	1.0
$6^2F_{5/2}$	4440.7	4415.7	0.6
$6^2F_{7/2}$	4440.7	4415.6	0.6

TABLE 2: CESIUM_I

STATE	OBSERVED ENERGY(cm^{-1})	NON-ORTHOGONALIZED RELATIVISTIC HFS ENERGY(cm^{-1})	PERCENTAGE DIFFERENCE
$6^2S_{1/2}$	31406.71	29814.8	5.1
$7^2S_{1/2}$	12871.20	13797.4	8.2
$8^2S_{1/2}$	7089.54	7669.3	8.2
$9^2S_{1/2}$	4496.03	4826.4	7.4
$10^2S_{1/2}$	3106.43	3322.6	7.0
$11^2S_{1/2}$	2276.71	2419.2	6.3
$12^2S_{1/2}$	1740.71	1837.4	5.6
$13^2S_{1/2}$		1441.9	
$14^2S_{1/2}$		1161.2	
$15^2S_{1/2}$		954.9	
$6^2P_{1/2}$	20228.47	19929.4	1.5
$6^2P_{3/2}$	19674.36	19397.3	1.4
$7^2P_{1/2}$	9641.06	10237.4	6.2
$7^2P_{3/2}$	9460.05	9975.0	5.4
$8^2P_{1/2}$	5697.57	6142.7	7.8
$8^2P_{3/2}$	5614.93	6016.9	7.2
$9^2P_{1/2}$	3769.42	4053.7	7.5
$9^2P_{3/2}$	3724.75	3986.8	7.0
$10^2P_{1/2}$	2679.62	2863.8	6.9
$10^2P_{3/2}$	2652.78	2824.8	6.5
$11^2P_{1/2}$	2003.03	2127.1	6.2

TABLE 2: CESIUM_I

STATE	OBSERVED ENERGY(cm^{-1})	NON-ORTHOGONALIZED RELATIVISTIC HFS ENERGY(cm^{-1})	PERCENTAGE DIFFERENCE
$11^2\text{P}_{3/2}$	1985.61	2102.4	5.9
$12^2\text{P}_{1/2}$	1553.86	1640.9	5.6
$12^2\text{P}_{3/2}$	1541.99	1624.4	5.3
$13^2\text{P}_{1/2}$	1240.71	1311.6	5.7
$13^2\text{P}_{3/2}$	1232.20	1299.7	5.5
$14^2\text{P}_{1/2}$	1013.55	1066.5	5.2
$14^2\text{P}_{3/2}$	1007.22	1057.9	5.0
$15^2\text{P}_{1/2}$	843.44	883.9	4.8
$15^2\text{P}_{3/2}$	838.73	877.5	4.6
$5^2\text{D}_{3/2}$	16907.22	16379.2	3.1
$5^2\text{D}_{5/2}$	16809.63	16198.9	3.6
$6^2\text{D}_{3/2}$	8817.82	10463.6	18.7
$6^2\text{D}_{5/2}$	8774.88	10402.6	18.6
$7^2\text{D}_{3/2}$	5358.85	6121.8	14.2
$7^2\text{D}_{5/2}$	5337.88	6099.7	14.3
$8^2\text{D}_{3/2}$	3595.46	4043.4	12.5
$8^2\text{D}_{5/2}$	3683.77	4030.3	9.4
$9^2\text{D}_{3/2}$	2577.81	2867.2	11.2
$9^2\text{D}_{5/2}$	2570.65	2859.3	11.2
$10^2\text{D}_{3/2}$	1938.17	2131.9	10.0
$10^2\text{D}_{5/2}$	1933.49	2126.9	10.0
$11^2\text{D}_{3/2}$	1510.07	1644.9	8.9

TABLE 2: CESIUM_I

STATE	OBSERVED ENERGY(cm^{-1})	NON-ORTHOGONALIZED RELATIVISTIC. HFS ENERGY(cm^{-1})	PERCENTAGE DIFFERENCE
$11^2D_{5/2}$	1506.82	1641.5	8.9
$12^2D_{3/2}$	1209.69	1306.8	8.0
$12^2D_{5/2}$	1207.36	1304.3	8.0
$13^2D_{3/2}$	990.65	1062.7	7.3
$13^2D_{5/2}$	988.95	1060.9	7.3
$14^2D_{3/2}$	826.25	880.9	6.6
$14^2D_{5/2}$	824.92	879.6	6.6
$4^2F_{5/2}$	6934.42	6924.9	0.1
$4^2F_{7/2}$	6934.25	6919.9	0.2
$5^2F_{5/2}$	4435.30	4533.1	2.2
$5^2F_{7/2}$	4435.15	4522.7	2.0
$6^2F_{5/2}$	3076.95	3103.2	0.9
$6^2F_{7/2}$	3077.05	3133.3	1.8
$7^2F_{5/2}$	2258.49	2989.6	32.4
$7^2F_{7/2}$	2258.55	2308.5	2.2
$8^2F_{5/2}$	1727.73	2251.7	30.3
$8^2F_{7/2}$	1727.78	2241.7	29.7
$9^2F_{5/2}$	1364.17	1734.7	27.2
$9^2F_{7/2}$	1364.20	1730.0	26.8
$10^2F_{5/2}$	1104.32	1373.1	24.3
$10^2F_{7/2}$	1104.33	1370.4	24.1
$11^2F_{5/2}$	912.12	1112.3	22.0

APPENDIX 12 : Non-Relativistic Caesium Oscillator Strengths

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SEKTCS $np^2P_J - ms^2S_{1/2}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
6	7	1/2 - 1/2	0.4800 01	0.1710 00	0.6180 07	0.1620 00	0.1960-03	0.7090 04
		3/2 - 1/2	0.5210 01	0.1870 00	0.1160 03	0.6890 00	0.5450-02	0.3360 06
6	9	1/2 - 1/2	0.6020 00	0.5780-02	0.9540 06	-0.1850 01	0.5470-01	0.9030 07
		3/2 - 1/2	0.6050 00	0.5630-02	0.1730 07	-0.1920 01	0.5640-01	0.1730 08
6	10	1/2 - 1/2	0.3570 00	0.2730-02	0.5330 06	-0.1060 01	0.1960-01	0.3930 07
		3/2 - 1/2	-0.3970 00	0.2640-02	0.9670 06	-0.1180 01	0.2300-01	0.8540 07
6	11	1/2 - 1/2	0.2900 00	0.1520-02	0.3280 06	-0.4640 00	0.3910-02	0.8400 06
		3/2 - 1/2	0.2890 00	0.1470-02	0.5930 06	-0.5870 00	0.6070-02	0.2450 07
6	12	1/2 - 1/2	-0.2250 00	0.9440-03	0.2150 06	-0.6050-01	0.6860-04	0.1560 05
		3/2 - 1/2	-0.2230 00	0.9050-03	0.3880 06	-0.1750 00	0.5530-03	0.2370 06
6	13	1/2 - 1/2	0.1070 00	0.2180-03	0.5130 05	0.2350-01	0.1050-04	0.2480 04
		3/2 - 1/2	0.5180-01	0.1560-03	0.6900 05	-0.7550-01	0.1050-03	0.4660 05
6	14	1/2 - 1/2	-0.8740-01	0.1480-03	0.3580 05	0.2200 00	0.9300-03	0.2260 06
		3/2 - 1/2	-0.7460-01	0.1040-03	0.4770 05	0.1330 00	0.3320-03	0.1520 06
6	15	1/2 - 1/2	0.7360-01	0.1060-03	0.2620 05	0.3440 00	0.2310-02	0.5710 06
		3/2 - 1/2	0.6260-01	0.7430-04	0.3470 05	0.2700 00	0.1390-02	0.6470 06
7	9	1/2 - 1/2	-0.2350 01	0.2870-01	0.5070 06	-0.1960 01	0.2010-01	0.3540 06
		3/2 - 1/2	-0.2310 01	0.2690-01	0.8850 06	-0.1630 01	0.1340-01	0.4410 06
7	10	1/2 - 1/2	0.1190 01	0.9380-02	0.2670 06	-0.2650 01	0.4630-01	0.1320 07
		3/2 - 1/2	0.1160 01	0.8620-02	0.4640 06	-0.2520 01	0.4100-01	0.2210 07
7	11	1/2 - 1/2	-0.7690 00	0.4410-02	0.1600 06	-0.2370 01	0.4190-01	0.1520 07
		3/2 - 1/2	-0.7430 00	0.4020-02	0.2760 06	-0.2350 01	0.4030-01	0.2770 07
7	12	1/2 - 1/2	0.5560 00	0.2470-02	0.1030 06	-0.1900 01	0.2990-01	0.1200 07
		3/2 - 1/2	0.5340 00	0.2230-02	0.1770 06	-0.1940 01	0.2940-01	0.2340 07
7	13	1/2 - 1/2	-0.1880 00	0.2920-03	0.1310 05	-0.1560 01	0.2010-01	0.9020 06
		3/2 - 1/2	-0.1400 00	0.1590-03	0.1370 05	-0.1600 01	0.2070-01	0.1760 07
7	14	1/2 - 1/2	0.1470 00	0.1860-03	0.8900 04	-0.1200 01	0.1210-01	0.5900 06
		3/2 - 1/2	0.1680 00	0.5880-04	0.9080 04	-0.1260 01	0.1330-01	0.1230 07
7	15	1/2 - 1/2	-0.1200 00	0.1280-03	0.6420 04	-0.0880 00	0.6880-02	0.3450 06
		3/2 - 1/2	-0.8630-01	0.6710-04	0.6480 04	-0.9570 00	0.7890-02	0.7610 06
8	9	1/2 - 1/2	0.1950 02	0.4620 00	0.4450 06	0.1070 02	0.1340 00	0.1340 06
		3/2 - 1/2	0.2070 02	0.4860 00	0.8110 06	0.1230 02	0.1700 00	0.2850 06
8	10	1/2 - 1/2	-0.3840 01	0.3860-01	0.1730 06	-0.6060 00	0.9620-03	0.4310 04
		3/2 - 1/2	-0.3730 01	0.3540-01	0.2970 06	-0.1340 00	0.4570-04	0.3840 04
8	11	1/2 - 1/2	0.1890 01	0.1230-01	0.9630 06	-0.2410 01	0.2020-01	0.1570 06
		3/2 - 1/2	0.1810 01	0.1100-01	0.1640 06	-0.2160 01	0.1590-01	0.2380 06
8	12	1/2 - 1/2	-0.1200 01	0.5760-02	0.6020 06	-0.2730 01	0.2990-01	0.3130 06
		3/2 - 1/2	-0.1140 01	0.5080-02	0.1020 06	-0.2600 01	0.2650-01	0.5310 06
8	13	1/2 - 1/2	0.3140 00	0.4250-03	0.5140 04	-0.2300 01	0.2280-01	0.2750 06
		3/2 - 1/2	0.2070 00	0.1820-03	0.4220 04	-0.2200 01	0.2050-01	0.4760 06

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATION STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P - ms^2S, J'$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
8	14	1/2 - 1/2	-0.232D 00	0.246D-03	0.338D 04	-0.219D 01	0.221D-01	0.303D 06
		3/2 - 1/2	-0.149D 00	0.101D-03	0.256D 04	-0.215D 01	0.206D-01	0.551D 06
8	15	1/2 - 1/2	0.182D 00	0.160D-03	0.240D 04	-0.197D 01	0.187D-01	0.281D 06
		3/2 - 1/2	0.116D 00	0.636D-04	0.184D 04	-0.197D 01	0.182D-01	0.529D 06
9	10	1/2 - 1/2	0.299D 02	0.588D 00	0.175D 06	0.178D 02	0.212D 00	0.622D 05
		3/2 - 1/2	0.316D 02	0.626D 00	0.319D 06	0.198D 02	0.245D 00	0.125D 06
9	11	1/2 - 1/2	-0.561D 01	0.476D-01	0.708D 05	0.126D 01	0.242D-02	0.359D 04
		3/2 - 1/2	-0.541D 01	0.430D-01	0.120D 06	0.184D 01	0.499D-02	0.140D 05
9	12	1/2 - 1/2	0.270D 01	0.150D-01	0.410D 05	-0.165D 01	0.559D-02	0.153D 05
		3/2 - 1/2	0.255D 01	0.131D-01	0.687D 05	-0.131D 01	0.343D-02	0.130D 05
9	13	1/2 - 1/2	-0.555D 00	0.725D-03	0.262D 04	-0.156D 01	0.576D-02	0.208D 05
		3/2 - 1/2	-0.329D 00	0.250D-03	0.174D 04	-0.132D 01	0.406D-02	0.282D 05
9	14	1/2 - 1/2	0.370D 00	0.361D-03	0.164D 04	-0.215D 01	0.122D-01	0.556D 05
		3/2 - 1/2	0.269D 00	0.114D-03	0.995D 03	-0.199D 01	0.102D-01	0.898D 05
9	15	1/2 - 1/2	-0.274D 00	0.214D-03	0.113D 04	-0.233D 01	0.155D-01	0.920D 05
		3/2 - 1/2	-0.151D 00	0.640D-04	0.655D 03	-0.222D 01	0.138D-01	0.142D 06
10	11	1/2 - 1/2	0.425D 02	0.736D 00	0.797D 05	0.263D 02	0.232D 00	0.306D 05
		3/2 - 1/2	0.449D 02	0.767D 00	0.145D 06	0.288D 02	0.316D 00	0.596D 05
10	12	1/2 - 1/2	-0.767D 01	0.560D-01	0.329D 05	0.353D 01	0.117D-01	0.697D 04
		3/2 - 1/2	-0.735D 01	0.499D-01	0.553D 05	0.420D 01	0.163D-01	0.131D 05
10	13	1/2 - 1/2	0.116D 01	0.167D-02	0.171D 04	0.135D 01	0.228D-02	0.233D 04
		3/2 - 1/2	0.645D 00	0.510D-03	0.998D 03	0.172D 01	0.365D-02	0.713D 04
10	14	1/2 - 1/2	-0.638D 00	0.626D-03	0.963D 03	-0.735D 00	0.832D-03	0.128D 04
		3/2 - 1/2	-0.322D 00	0.157D-03	0.466D 03	-0.450D 00	0.306D-03	0.909D 03
10	15	1/2 - 1/2	0.427D 00	0.318D-03	0.631D 03	-0.170D 01	0.5C7D-02	0.101D 05
		3/2 - 1/2	0.264D 00	0.716D-04	0.275D 03	-0.149D 01	0.377D-02	0.146D 05
11	12	1/2 - 1/2	0.574D 02	0.874D 00	0.401D 05	0.363D 02	0.350D 00	0.161D 05
		3/2 - 1/2	0.605D 02	0.907D 00	0.726D 05	0.393D 02	0.593D 00	0.307D 05
11	13	1/2 - 1/2	-0.376D 01	0.805D-02	0.169D 04	0.965D 01	0.527D-01	0.111D 05
		3/2 - 1/2	-0.229D 01	0.287D-02	0.113D 04	0.102D 02	0.572D-01	0.226D 05
11	14	1/2 - 1/2	0.133D 01	0.150D-02	0.709D 04	0.292D 01	0.729D-02	0.344D 04
		3/2 - 1/2	0.640D 00	0.342D-03	0.310D 03	0.332D 01	0.922D-02	0.836D 04
11	15	1/2 - 1/2	-0.729D 00	0.564D-03	0.413D 03	0.256D 00	0.696D-04	0.510D 02
		3/2 - 1/2	-0.310D 00	0.100D-03	0.142D 03	0.574D 00	0.344D-03	0.488D 03
12	13	1/2 - 1/2	0.937D 02	0.294D 00	0.831D 04	0.691D 02	0.541D 00	0.452D 04
		3/2 - 1/2	0.566D 02	0.246D 00	0.126D 03	0.727D 02	0.533D 00	0.716D 04
12	14	1/2 - 1/2	-0.438D 01	0.764D-02	0.786D 03	0.128D 02	0.655D-01	0.673D 04
		3/2 - 1/2	-0.246D 01	0.233D-02	0.450D 03	0.134D 02	0.691D-01	0.134D 05
12	15	1/2 - 1/2	0.151D 01	0.138D-02	0.330D 03	0.463D 01	0.130D-01	0.312D 04
		3/2 - 1/2	0.628D 00	0.235D-03	0.108D 03	0.505D 01	0.151D-01	0.696D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SUMMED $np^2 p_j - ms^2 5 1/2$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

N	l	TRANSITION	BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
			R_{lj}	f_{lj}	$A_{lj}(\text{sec}^{-1})$	R_{lj}	f_{lj}	$A_{lj}(\text{sec}^{-1})$
13	14	1/2 - 1/2	0.118D 03	0.112D 01	0.470D 04	0.855D 02	0.584D 00	0.248J 04
		3/2 - 1/2	0.121D 03	0.106D 01	0.711D 04	0.898D 02	0.580D 00	0.390J 04
13	15	1/2 - 1/2	-0.514D 01	0.763D-02	0.416D 03	0.162D 02	0.762D-01	0.415D 04
		3/2 - 1/2	-0.269D 01	0.204D-02	0.209D 03	0.168D 02	0.794D-01	0.815D 04
14	15	1/2 - 1/2	0.148D 03	0.124D 01	0.284D 04	0.103D 03	0.636D 00	0.146J 04
		3/2 - 1/2	0.149D 03	0.117D 01	0.428D 04	0.108D 03	0.624D 00	0.228J 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2s_{1/2} - m^2p_{1/2}$, CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF RATTS AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	$J-J'$	R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$	R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$	
6	6	1/2 - 1/2	0.5520 01	0.3450 9C	0.2880 03	-0.1350 02	0.2070 01	0.1730 09	
		1/2 - 3/2	0.5470 01	0.7120 00	0.3270 08	-0.1300 02	0.3990 01	0.1330 39	
6	7	1/2 - 1/2	-0.5780 00	0.7350-02	0.2320 07	-0.2600 01	0.1490 00	0.4700 28	
		1/2 - 3/2	-0.7150 00	0.2270-01	0.3650 07	-0.2380 01	0.2510 00	0.4040 08	
6	8	1/2 - 1/2	0.2570 00	0.1720-02	0.7600 06	0.4460-01	0.5170-04	0.2280 05	
		1/2 - 3/2	0.3370 00	0.5920-02	0.1310 07	0.1780 00	0.1650-02	0.3660 36	
6	9	1/2 - 1/2	-0.1590 00	0.7070-03	0.3600 06	0.1010 01	0.2850-01	0.1450 08	
		1/2 - 3/2	-0.2130 00	0.2530-02	0.6480 06	0.1080 01	0.6560-01	0.1680 08	
6	10	1/2 - 1/2	0.1130 00	0.3690-03	0.2030 06	0.1300 01	0.4950-01	0.2730 09	
		1/2 - 3/2	0.1520 00	0.1350-02	0.3730 06	0.1340 01	0.1050 00	0.2900 08	
6	11	1/2 - 1/2	-0.8600-01	0.2200-03	0.2300 06	0.1330 01	0.5240-01	0.3020 08	
		1/2 - 3/2	-0.1170 00	0.8180-03	0.2360 06	0.1350 01	0.1080 00	0.3110 08	
6	12	1/2 - 1/2	0.6520-01	0.1450-03	0.8600 05	0.1240 01	0.4650-01	0.2760 09	
		1/2 - 3/2	0.9440-01	0.5390-03	0.1600 06	0.1250 01	0.9420-01	0.2800 08	
6	13	1/2 - 1/2	-0.5710-01	0.9960-04	0.6050 05	0.1120 01	0.3820-01	0.2320 08	
		1/2 - 3/2	-0.7830-01	0.3740-03	0.1140 06	0.1120 01	0.7660-01	0.2330 08	
6	14	1/2 - 1/2	0.4830-01	0.7180-04	0.4420 05	0.9910 00	0.3020-01	0.1860 38	
		1/2 - 3/2	0.6640-01	0.2720-03	0.8370 05	0.9900 00	0.6040-01	0.1860 08	
6	15	1/2 - 1/2	-0.4190-01	0.5440-04	0.3390 05	0.8730 00	0.2360-01	0.1470 08	
		1/2 - 3/2	-0.5730-01	0.2030-03	0.6340 05	0.8700 00	0.4690 01	0.1460 08	
7	7	1/2 - 1/2	0.1250 02	0.5090 00	0.3540 07	-0.2740 02	0.2460 01	0.1710 08	
		1/2 - 3/2	0.1220 02	0.1040 01	0.4020 07	-0.2680 02	0.4960 01	0.1920 08	
7	8	1/2 - 1/2	-0.1210 01	0.1070-01	0.3690 06	-0.7670 01	0.4270 00	0.1470 08	
		1/2 - 3/2	-0.1490 01	0.3250-01	0.5710 06	-0.7560 01	0.8430 00	0.1460 09	
7	9	1/2 - 1/2	0.4850 00	0.2170-02	0.1200 06	-0.3860 01	0.1380 00	0.7600 37	
		1/2 - 3/2	0.6420 00	0.7620-02	0.2130 06	-0.3730 01	0.2540 00	0.7130 37	
7	10	1/2 - 1/2	-0.2800 00	0.8090-03	0.5610 05	-0.1860 01	0.3530-01	0.2480 37	
		1/2 - 3/2	-0.3860 00	0.3080-02	0.1070 06	-0.1740 01	0.6300-01	0.2130 37	
7	11	1/2 - 1/2	0.1900 00	0.3900-03	0.3140 05	-0.6800 00	0.5080-02	0.4000 06	
		1/2 - 3/2	0.2690 00	0.1590-02	0.6280 05	-0.5850 00	0.7540-02	0.2980 36	
7	12	1/2 - 1/2	-0.1420 00	0.2330-03	0.1990 05	0.3620-01	0.1500-04	0.1280 34	
		1/2 - 3/2	-0.2040 00	0.9510-03	0.4070 05	0.1690 00	0.2730-03	0.1170 05	
7	13	1/2 - 1/2	0.1120 00	0.1480-03	0.1330 05	0.4660 00	0.2550-02	0.2310 06	
		1/2 - 3/2	0.1620 00	0.6180-03	0.2790 05	0.5220 00	0.6420-02	0.2900 36	
7	14	1/2 - 1/2	-0.9140-01	0.1000-03	0.9410 04	0.7200 00	0.6220-02	0.5840 06	
		1/2 - 3/2	-0.1340 00	0.4280-03	0.2010 05	0.7630 00	0.1400-01	0.6560 36	
7	15	1/2 - 1/2	0.7750-01	0.7310-04	0.7050 04	0.8640 00	0.9030-02	0.9770 36	
		1/2 - 3/2	0.1130 00	0.3100-03	0.1500 05	0.8960 00	0.1360-01	0.9440 36	
8	8	1/2 - 1/2	-0.1390 00	0.4190-03	0.1280 06	-0.1550 01	0.5230-01	0.1600 38	
		1/2 - 3/2	-0.3170-01	0.4370-04	0.6710 04	-0.1910 01	0.1540 00	0.2640 08	

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2s_{1/2} - mp^2p_{j_1}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		RATES-DAMGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	$J-J'$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	
8	9	1/2 - 1/2 1/2 - 3/2	0.1030 00 0.3360-01	0.2500-03 0.5340-04	0.9060 05 0.9730 04	-0.3030 01 -0.3190 01	0.2100 00 0.4810 00	0.7950 08 0.8760 08	
8	10	1/2 - 1/2 1/2 - 3/2	-0.7870-01 -0.2870-01	0.1530-03 0.4080-04	0.6080 05 0.8120 04	-0.3230 01 -0.3300 01	0.2530 00 0.5390 00	0.1030 09 0.1070 09	
8	11	1/2 - 1/2 1/2 - 3/2	0.6250-01 0.2390-01	0.9920-04 0.2900-04	0.4160 05 0.6090 04	-0.2980 01 -0.3000 01	0.2260 00 0.4580 00	0.9470 08 0.9630 08	
8	12	1/2 - 1/2 1/2 - 3/2	-0.5070-01 -0.1580-01	0.6640-04 0.2020-04	0.2890 05 0.4400 04	-0.2590 01 -0.2590 01	0.1740 00 0.3480 00	0.7570 08 0.7570 08	
8	13	1/2 - 1/2 1/2 - 3/2	0.4250-01 0.1680-01	0.4730-04 0.1480-04	0.2110 05 0.3300 04	-0.2200 01 -0.2190 01	0.1270 00 0.2500 00	0.5640 08 0.5580 08	
8	14	1/2 - 1/2 1/2 - 3/2	-0.3630-01 -0.1440-01	0.3490-04 0.1100-04	0.1580 05 0.2490 04	-0.1840 01 -0.1820 01	0.8950-01 0.1750 00	0.4060 08 0.3990 08	
8	15	1/2 - 1/2 1/2 - 3/2	0.3110-01 0.1260-01	0.2580-04 0.8450-05	0.1180 05 0.1940 04	-0.1530 01 -0.1510 01	0.6230-01 0.1220 00	0.2860 08 0.2800 08	
9	9	1/2 - 1/2 1/2 - 3/2	0.3300 02 0.3220 02	0.8030 00 0.1620 01	0.2830 06 0.3210 05	-0.6240 02 -0.6150 02	0.2860 01 0.5900 01	0.1010 07 0.1170 07	
9	10	1/2 - 1/2 1/2 - 3/2	-0.3400 01 -0.4010 01	0.2130-01 0.6000-01	0.4680 05 0.6800 05	-0.1440 02 -0.1470 02	0.3710 00 0.8120 00	0.8350 06 0.9200 06	
9	11	1/2 - 1/2 1/2 - 3/2	0.1340 01 0.1690 01	0.4530-02 0.1450-01	0.1880 05 0.3040 03	-0.9170 01 -0.9270 01	0.2120 00 0.4370 00	0.8800 06 0.9190 06	
9	12	1/2 - 1/2 1/2 - 3/2	-0.7620 00 -0.9520 00	0.1730-02 0.5890-02	0.9480 04 0.1710 05	-0.6680 01 -0.6680 01	0.1330 00 0.2670 00	0.7670 06 0.7770 06	
9	13	1/2 - 1/2 1/2 - 3/2	0.5050 00 0.6790 00	0.8520-03 0.3050-02	0.6020 04 0.1080 05	-0.4500 01 -0.5000 01	0.8370-01 0.1650 00	0.5910 06 0.5870 06	
9	14	1/2 - 1/2 1/2 - 3/2	-0.3730 00 -0.5080 00	0.4910-03 0.1820-02	0.3970 04 0.7390 04	-0.3840 01 -0.3790 01	0.5210-01 0.1010 00	0.4210 06 0.4120 06	
9	15	1/2 - 1/2 1/2 - 3/2	0.2930 00 0.4000 00	0.3170-03 0.1190-02	0.2820 04 0.5290 04	-0.2940 01 -0.2680 01	0.3190-01 0.6140-01	0.2340 06 0.2740 06	
10	10	1/2 - 1/2 1/2 - 3/2	0.4680 02 0.4550 02	0.9460 00 0.1900 01	0.1500 06 0.1300 05	-0.8400 02 -0.8290 02	0.3050 01 0.6320 01	0.3700 08 0.4340 06	
10	11	1/2 - 1/2 1/2 - 3/2	-0.4900 01 -0.5710 01	0.2680-01 0.7410-01	0.2170 05 0.3100 05	-0.1690 02 -0.1760 02	0.3200 00 0.7730 00	0.2600 06 0.2790 06	
10	12	1/2 - 1/2 1/2 - 3/2	0.1940 01 0.2400 01	0.5710-02 0.1820-01	0.9500 04 0.1490 05	-0.1090 02 -0.1120 02	0.1470 00 0.3930 00	0.3010 06 0.3220 06	
10	13	1/2 - 1/2 1/2 - 3/2	-0.1090 01 -0.1400 01	0.2260-02 0.7450-02	0.5250 04 0.8730 04	-0.8280 01 -0.8370 01	0.1230 00 0.2660 00	0.3000 06 0.3120 06	
10	14	1/2 - 1/2 1/2 - 3/2	0.7290 00 0.9560 00	0.1130-02 0.3890-02	0.3290 04 0.5710 04	-0.6590 01 -0.6620 01	0.9200-01 0.1860 00	0.2670 06 0.2730 06	
10	15	1/2 - 1/2 1/2 - 3/2	-0.5370 00 -0.7110 00	0.6620-03 0.2320-02	0.2260 04 0.3980 04	-0.5350 01 -0.5340 01	0.6580-01 0.1310 00	0.2240 06 0.2250 06	

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $ns^2S_{1/2} - mp^2P_{1/2}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
11	11	1/2 - 1/2	0.6260 02	0.1090 01	0.5430 05	-0.1080 03	0.3250 01	0.1620 06
		1/2 - 3/2	0.6080 02	0.2180 01	0.6150 05	-0.1070 03	0.6760 01	0.1910 06
11	12	1/2 - 1/2	-0.6750 01	0.3340-01	0.1160 05	-0.1930 02	0.2720 00	0.9480 05
		1/2 - 3/2	-0.7790 01	0.9030-01	0.1620 05	-0.2030 02	0.6110 00	0.1100 06
11	13	1/2 - 1/2	0.2680 01	0.7550-02	0.5410 04	-0.1230 02	0.1590 00	0.1140 06
		1/2 - 3/2	0.3270 01	0.2260-01	0.8230 04	-0.1270 02	0.3420 00	0.1250 06
11	14	1/2 - 1/2	-0.1520 01	0.2950-02	0.3130 04	-0.09510 01	0.1100 00	0.1230 06
		1/2 - 3/2	-0.1910 01	0.9370-02	0.5040 04	-0.0790 01	0.2420 00	0.1300 06
11	15	1/2 - 1/2	0.1020 01	0.1500-02	0.2060 04	-0.07790 01	0.6800-01	0.1210 06
		1/2 - 3/2	0.1300 01	0.4910-02	0.3390 04	-0.07860 01	0.1910 00	0.1250 06
12	12	1/2 - 1/2	0.8050 02	0.1230 01	0.2860 03	-0.1350 03	0.2400 01	0.8070 05
		1/2 - 3/2	0.7800 02	0.2450 01	0.3230 05	-0.1340 03	0.7210 01	0.9490 05
12	13	1/2 - 1/2	-0.8540 01	0.4050-01	0.6750 04	-0.2150 02	0.2340 00	0.3900 05
		1/2 - 3/2	-0.1020 02	0.1080 00	0.9270 04	-0.2280 02	0.5370 00	0.4630 05
12	14	1/2 - 1/2	0.3580 01	0.2430-02	0.3330 04	-0.1350 02	0.1340 00	0.4720 05
		1/2 - 3/2	0.4300 01	0.2750-01	0.4940 04	-0.1410 02	0.2940 00	0.5270 05
12	15	1/2 - 1/2	-0.2040 01	0.3780-02	0.2030 04	-0.1050 02	0.1000 00	0.5370 05
		1/2 - 3/2	-0.2510 01	0.1150-01	0.3120 04	-0.1080 02	0.2130 00	0.5770 05
13	13	1/2 - 1/2	0.7260 02	0.1080 01	0.2900 05	-0.1470 03	0.4390 01	0.1190 06
		1/2 - 3/2	0.6860 02	0.2000 01	0.2930 05	-0.1440 03	0.3850 01	0.1300 06
13	14	1/2 - 1/2	-0.1680 02	0.1230 00	0.1500 05	-0.3350 02	0.4800 00	0.5930 05
		1/2 - 3/2	-0.1700 02	0.2540 00	0.1600 05	-0.3440 02	0.1090 01	0.6570 05
13	15	1/2 - 1/2	0.8280 01	0.4160-01	0.9930 04	-0.1930 02	0.2270 00	0.5420 05
		1/2 - 3/2	0.6520 01	0.8970-01	0.1080 05	-0.1980 02	0.4790 00	0.5820 05
14	14	1/2 - 1/2	0.8850 02	0.1170 01	0.1700 03	-0.1770 03	0.4660 01	0.6320 05
		1/2 - 3/2	0.8240 02	0.2170 01	0.1720 05	-0.1740 03	0.9400 01	0.7480 05
14	15	1/2 - 1/2	-0.2050 02	0.1360 00	0.9150 04	-0.3810 02	0.4600 00	0.3140 05
		1/2 - 3/2	-0.2070 02	0.2800 00	0.9700 04	-0.3920 02	0.1030 01	0.3480 05
15	15	1/2 - 1/2	0.1060 03	0.1270 01	0.1050 03	-0.2100 03	0.4950 01	0.4120 05
		1/2 - 3/2	0.1000 03	0.2350 01	0.1060 03	-0.2070 03	0.1000 02	0.4520 05

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_{j_1} - mp^2P_{j_2}$, CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION			BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
N	M	J-J'	R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$	R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$
5	7	3/2 - 1/2	0.1470 01	0.159D-01	0.112D 07	0.143D 01	0.151D-01	0.106D 07
		3/2 - 3/2	0.1290 01	0.250D-02	0.925D 05	0.138D 01	0.288D-02	0.106D 06
		5/2 - 3/2	0.135D 01	0.163D-01	0.881D 06	0.140D 01	0.175D-01	0.948D 06
5	8	3/2 - 1/2	-0.356D 00	0.144D-02	0.241D 06	0.120D 00	0.164D-03	0.275D 05
		3/2 - 3/2	-0.314D 00	0.226D-03	0.192D 05	0.661D-01	0.100D-04	0.850D 03
		5/2 - 3/2	-0.332D 00	0.150D-02	0.188D 06	0.880D-01	0.105D-03	0.132D 05
5	9	3/2 - 1/2	0.154D 00	0.317D-03	0.730D 05	-0.394D 00	0.205D-02	0.473D 06
		3/2 - 3/2	0.134D 00	0.478D-04	0.554D 04	-0.428D 00	0.490D-03	0.567D 05
		5/2 - 3/2	0.143D 00	0.326D-03	0.558D 05	-0.413D 00	0.271D-02	0.464D 06
5	10	3/2 - 1/2	-0.837D-01	0.101D-03	0.272D 05	-0.574D 00	0.475D-02	0.128D 07
		3/2 - 3/2	-0.704D-01	0.143D-04	0.194D 04	-0.595D 00	0.102D-02	0.138D 06
		5/2 - 3/2	-0.766D-01	0.101D-03	0.202D 05	-0.583D-02	0.583D-02	0.118D 07
5	11	3/2 - 1/2	0.511D-01	0.395D-04	0.117D 05	-0.612D 00	0.566D-02	0.168D 07
		3/2 - 3/2	0.415D-01	0.519D-05	0.771D 03	-0.624D 00	0.119D-02	0.175D 06
		5/2 - 3/2	0.460D-01	0.381D-04	0.837D 04	-0.618D 00	0.688D-02	0.151D 07
5	12	3/2 - 1/2	-0.336D-01	0.176D-04	0.553D 04	-0.590D 00	0.540D-02	0.170D 07
		3/2 - 3/2	-0.261D-01	0.212D-05	0.333D 03	-0.596D 00	0.110D-02	0.174D 06
		5/2 - 3/2	-0.296D-01	0.163D-04	0.380D 04	-0.593D 00	0.652D-02	0.152D 07
5	13	3/2 - 1/2	0.234D-01	0.868D-05	0.284D 04	-0.543D 00	0.468D-02	0.153D 07
		3/2 - 3/2	0.172D-01	0.940D-06	0.154D 03	-0.546D 00	0.947D-03	0.155D 06
		5/2 - 3/2	0.201D-01	0.764D-05	0.185D 04	-0.545D 00	0.563D-02	0.137D 07
5	14	3/2 - 1/2	-0.169D-01	0.461D-05	0.155D 04	-0.490D 00	0.387D-02	0.130D 07
		3/2 - 3/2	-0.117D-01	0.440D-06	0.742D 02	-0.491D 00	0.777D-03	0.131D 06
		5/2 - 3/2	-0.141D-01	0.382D-05	0.953D 03	-0.492D 00	0.464D-02	0.116D 07
5	15	3/2 - 1/2	0.125D-01	0.255D-05	0.877D 03	-0.438D 00	0.312D-02	0.107D 07
		3/2 - 3/2	0.811D-02	0.214D-06	0.369D 02	-0.438D 00	0.623D-03	0.107D 06
		5/2 - 3/2	0.102D-01	0.201D-05	0.512D 03	-0.439D 00	0.374D-02	0.955D 06
6	8	3/2 - 1/2	0.411D 01	0.535D-01	0.694D 06	0.265D 01	0.222D-01	0.289D 06
		3/2 - 3/2	0.361D 01	0.847D-02	0.580D 05	0.262D 01	0.445D-02	0.304D 05
		5/2 - 3/2	0.377D 01	0.546D-01	0.546D 06	0.264D 01	0.267D-01	0.267D 06
6	9	3/2 - 1/2	-0.127D 01	0.327D-02	0.281D 05	0.823D 00	0.346D-02	0.118D 06
		3/2 - 3/2	-0.115D 01	0.136D-02	0.235D 05	0.757D 00	0.513D-03	0.132D 05
		5/2 - 3/2	-0.119D 01	0.669D-02	0.222D 06	0.786D 00	0.373D-02	0.968D 05
6	10	3/2 - 1/2	0.675D 00	0.283D-02	0.142D 06	-0.726D-02	0.328D-06	0.165D 02
		3/2 - 3/2	0.611D 00	0.467D-03	0.118D 05	-0.601D-01	0.451D-05	0.114D 03
		5/2 - 3/2	0.633D 00	0.298D-02	0.112D 06	-0.345D-01	0.833D-05	0.331D 03
6	11	3/2 - 1/2	-0.408D 00	0.115D-02	0.712D 05	-0.438D 00	0.133D-02	0.821D 05
		3/2 - 3/2	-0.446D 00	0.276D-03	0.858D 04	-0.417D 00	0.214D-03	0.679D 04
		5/2 - 3/2	-0.426D 00	0.150D-02	0.690D 05	-0.411D 00	0.143D-02	0.643D 05
6	12	3/2 - 1/2	-0.557D 00	0.262D-02	0.105D 06	0.316D 00	0.713D-03	0.516D 05
		3/2 - 3/2	-0.624D 00	0.574D-03	0.203D 05	0.286D 00	0.120D-01	0.425D 04
		5/2 - 3/2	-0.616D 00	0.325D-02	0.170D 06	0.296D 00	0.771D-03	0.433D 05
6	13	3/2 - 1/2	-0.675D 00	0.349D-02	0.267D 05	-0.243D 00	0.453D-03	0.347D 05
		3/2 - 3/2	-0.654D 00	0.739D-03	0.284D 05	-0.219D 00	0.740D-04	0.284D 04
		5/2 - 3/2	-0.682D 00	0.426D-02	0.243D 06	-0.228D 00	0.475D-03	0.270D 05

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2 \rightarrow nd^1$ - $mp^2 \rightarrow mp^1$, CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

N	M	TRANSITION $J_1 - J_2$	BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
			R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
6	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.6530 00 -0.7060 00 -0.6580 00	0.3800-02 0.7890-03 0.4600-02	0.3080 00 0.3210 00 0.2770 00	0.1950 00 0.1820 00 -0.1610 00	0.3010-03 0.3140-03 0.2190-03	0.2440 J5 0.1900 J5 0.1780 05
6	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.6880 00 -0.1140 00 -0.1510 00	0.3740-02 0.3420-04 0.2200-03	0.3170 00 0.1450 04 0.1390 05	0.1760 00 -0.6890 00 -0.6830 00	0.4900-04 0.7630-03 0.4500-02	0.1970 04 0.3260 05 0.2840 05
7	9	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.7650 01 0.6740 01 0.7020 01	0.9420-01 0.1500-01 0.9670-01	0.3170 00 0.2680 05 0.2520 06	0.4100 01 0.4110 01 0.4120 01	0.2710-01 0.5530-02 0.3330-01	0.9130 J5 0.9360 J4 0.8660 J5
7	10	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.2430 01 -0.2210 01 -0.2280 01	0.1600-01 0.2680-02 0.1700-01	0.1540 05 0.1310 05 0.1230 06	0.1730 01 0.1660 01 0.1690 01	0.8070-02 0.1510-02 0.9310-02	0.7730 J5 0.7370 04 0.6710 05
7	11	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.1310 01 0.1200 01 0.1240 01	0.5830-02 0.9810-03 0.6220-02	0.8760 05 0.7450 04 0.6990 05	0.5960 00 0.5320 00 0.5620 00	0.1210-02 0.1940-03 0.1290-02	0.1810 05 0.1470 04 0.1450 05
7	12	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.8600 00 -0.7890 00 -0.8130 00	0.2850-02 0.4810-03 0.3050-02	0.5500 05 0.4670 04 0.4390 05	-0.1410-01 -0.6550-01 -0.3920-01	0.7670-00 0.3320-05 0.7030-05	0.1480 02 0.3220 02 0.1020 J3
7	13	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.6270 00 0.5750 00 0.5520 00	0.1640-02 0.2760-03 0.1750-02	0.3700 05 0.3140 04 0.2750 05	-0.3560 00 -0.3970 00 -0.3750 00	0.5300-03 0.1320-03 0.7000-03	0.1200 05 0.1500 04 0.1180 05
7	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.4860 00 -0.4460 00 -0.4590 00	0.1040-02 0.1750-03 0.1110-02	0.2620 05 0.2210 05 0.2080 05	-0.5490 00 -0.5810 00 -0.5620 00	0.1330-02 0.2670-03 0.1660-02	0.3380 05 0.3360 J4 0.3120 J5
7	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.3930 00 0.3610 00 0.3720 00	0.7060-03 0.1190-03 0.7560-03	0.1920 05 0.1620 04 0.1530 05	-0.6540 00 -0.6780 00 -0.6620 00	0.1950-02 0.4200-03 0.2400-02	0.5310 J5 0.5730 04 0.4850 05
8	10	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.1220 02 0.1070 02 0.1120 02	0.1370 00 0.2200-01 0.1410 00	0.1530 05 0.1300 05 0.1220 05	0.5720 01 0.5780 01 0.5770 01	0.3030-01 0.6390-02 0.3770-01	0.3390 J5 0.3780 J4 0.3270 05
8	11	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.3880 01 -0.3540 01 -0.3650 01	0.2420-01 0.4090-02 0.2590-01	0.8200 03 0.7060 04 0.6610 05	0.2730 01 0.2680 01 0.2700 01	0.1200-01 0.2340-02 0.1420-01	0.4080 05 0.4030 04 0.3630 J5
8	12	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.2090 01 0.1920 01 0.1980 01	0.9010-02 0.1540-02 0.9700-02	0.5010 05 0.4320 04 0.4040 05	0.1310 01 0.1240 01 0.1270 01	0.3120-02 0.6330-03 0.4900-02	0.1380 05 0.1900 J4 0.1670 J5
8	13	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.1370 01 -0.1270 01 -0.1300 01	0.4490-02 0.7690-03 0.4850-02	0.3320 05 0.2860 04 0.2680 05	0.5020 J0 0.4430 00 0.4420 00	0.6010-03 0.9170-04 0.6160-03	0.4430 J4 0.3870 J3 0.3520 04
8	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.1000 01 0.9250 00 0.9500 00	0.2620-02 0.4490-03 0.2830-02	0.2330 05 0.2000 04 0.1880 05	0.1670-01 -0.3400-01 -0.7510-02	0.7290-06 0.6030-06 0.1770-06	0.5480 J1 0.2700 J1 0.1170 01
8	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.7770 00 -0.7190 00 -0.7390 00	0.1680-02 0.2890-03 0.1820-02	0.1700 05 0.1460 04 0.1370 05	-0.2870 00 -0.3580 00 -0.3050 00	0.2300-03 0.6010-04 0.3100-03	0.2320 04 0.3060 J3 0.2330 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_j - mp^2P_j$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION			BATES-DAMGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$		
9	11	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.1770 02 0.1560 02 0.1620 02	0.1810 00 0.2930 01 0.1880 00	0.7990 03 0.6850 04 0.6420 05	0.7480 01 0.7620 01 0.7590 01	0.7260 01 0.6970 02 0.6100 01	0.1440 05 0.1630 04 0.1400 05		
9	12	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.5620 01 -0.5150 01 -0.5300 01	0.3270 01 0.5560 02 0.3510 01	0.4570 03 0.3970 04 0.3710 05	0.3830 01 0.3790 01 0.3810 01	0.1520 01 0.3010 02 0.1810 01	0.2120 05 0.2150 04 0.1920 05		
9	13	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.3020 01 0.2790 01 0.2870 01	0.1230 01 0.2120 02 0.1330 01	0.2940 05 0.2560 04 0.2390 05	0.2090 01 0.2020 01 0.2050 01	0.5900 02 0.1120 02 0.6860 02	0.1410 05 0.1390 04 0.1230 05		
9	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.1580 01 -0.1840 01 -0.1880 01	0.6210 02 0.1070 02 0.6740 02	0.2030 05 0.1760 04 0.1650 05	0.1090 01 0.1030 01 0.1060 01	0.1900 02 0.3370 03 0.2140 02	0.6190 04 0.5550 03 0.5220 04		
9	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.1440 01 0.1340 01 0.1370 01	0.3640 02 0.6320 03 0.3570 02	0.1460 05 0.1270 04 0.1190 05	0.4710 00 0.4150 00 0.4430 00	0.3840 03 0.6060 04 0.4140 03	0.1560 04 0.1220 03 0.1240 04		
10	12	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.2410 02 0.2140 02 0.2220 02	0.2260 00 0.3670 01 0.2350 00	0.4460 03 0.3840 04 0.3600 05	0.9400 01 0.9630 01 0.9570 01	0.3440 01 0.7430 02 0.4360 01	0.6770 04 0.7740 03 0.6680 04		
10	13	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.7660 01 -0.7030 01 -0.7230 01	0.4140 01 0.7070 02 0.4460 01	0.2690 05 0.2350 04 0.2190 05	0.5000 01 0.4980 01 0.5000 01	0.1770 01 0.3550 02 0.2130 01	0.1150 05 0.1180 04 0.1050 05		
10	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.4100 01 0.3800 01 0.3900 01	0.1570 01 0.2720 02 0.1710 01	0.1790 05 0.1570 04 0.1470 05	0.2930 01 0.2870 01 0.2900 01	0.8030 02 0.1550 02 0.9440 02	0.9150 04 0.8170 03 0.8110 04		
10	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.2680 01 -0.2490 01 -0.2560 01	0.7950 02 0.1390 02 0.8680 02	0.1270 05 0.1120 04 0.1040 05	0.1740 01 0.1670 01 0.1700 01	0.3340 02 0.6240 03 0.3660 02	0.5350 04 0.5030 03 0.4630 04		
11	13	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.3160 02 0.2810 02 0.2910 02	0.2730 00 0.4430 01 0.2830 00	0.2640 05 0.2280 04 0.2140 05	0.1150 02 0.1180 02 0.1170 02	0.3580 01 0.7820 02 0.4570 01	0.3460 04 0.4020 03 0.3450 04		
11	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.1000 02 -0.9200 01 -0.9450 01	0.5030 01 0.8620 02 0.5420 01	0.1650 03 0.1450 04 0.1350 05	0.6250 01 0.6260 01 0.6270 01	0.1960 01 0.3900 02 0.2380 01	0.6460 04 0.6730 03 0.5950 04		
11	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.5330 01 0.4550 01 0.5080 01	0.1910 01 0.3340 02 0.2090 01	0.1130 05 0.1000 04 0.9340 04	0.3820 01 0.3770 01 0.3680 01	0.9950 02 0.1940 02 0.1170 01	0.5340 04 0.5920 03 0.5220 04		
12	14	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.4010 02 0.3560 02 0.3690 02	0.3190 00 0.5190 01 0.3310 00	0.1640 05 0.1420 04 0.1330 05	0.1370 02 0.1410 02 0.1400 02	0.3710 01 0.8170 02 0.4750 01	0.1900 04 0.2240 03 0.1910 04		
12	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	-0.1260 02 -0.1160 02 -0.1190 02	0.5900 01 0.1020 01 0.6400 01	0.1060 05 0.9340 03 0.8690 04	0.7580 01 0.7630 01 0.7620 01	0.2130 01 0.4370 02 0.2600 01	0.3310 04 0.4310 03 0.3540 04		
13	15	3/2 - 1/2 3/2 - 3/2 5/2 - 3/2	0.4950 02 0.4410 02 0.4570 02	0.3650 00 0.5930 01 0.3820 00	0.1060 05 0.9220 04 0.8610 04	0.1600 02 0.1660 02 0.1640 02	0.3840 02 0.8470 02 0.4940 01	0.1110 04 0.1400 03 0.1110 04		

$np^2 p_j - md^2 D_j$

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF RATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	$J-J'$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	
6	5	1/2 - 3/2	0.6560 01	0.2890 00	0.1060 07	0.1360 02	0.1240 01	0.4570 07	
		3/2 - 3/2	0.6550 01	0.2400-01	0.1230 00	0.1320 02	0.9730-01	0.4970 05	
		3/2 - 5/2	0.6610 01	0.2280 00	0.8320 06	0.1330 02	0.9230 00	0.3370 05	
6	6	1/2 - 3/2	0.3350 01	0.2590 00	0.1120 03	-0.7140 01	0.1130 01	0.5120 08	
		3/2 - 3/2	0.3760 01	0.3100-01	0.2400 07	-0.8150 01	0.1400 00	0.1150 08	
		3/2 - 5/2	0.3660 01	0.2660 00	0.1400 03	-0.7990 01	0.1270 01	0.6700 08	
6	7	1/2 - 3/2	-0.1630 01	0.8000-01	0.5900 07	-0.2640 01	0.2100 00	0.1550 08	
		3/2 - 3/2	-0.1760 01	0.8960-02	0.1220 07	-0.3110 01	0.2800-01	0.3630 07	
		3/2 - 5/2	-0.1730 01	0.7830-01	0.7160 07	-0.3050 01	0.2330 00	0.2220 08	
6	8	1/2 - 3/2	0.1020 01	0.3540-01	0.3260 07	-0.1050 01	0.3710-01	0.3420 07	
		3/2 - 3/2	0.1090 01	0.3870-02	0.6670 06	-0.5170-02	0.6170-02	0.1060 07	
		3/2 - 5/2	0.1080 01	0.3410-01	0.3920 07	-0.1340 01	0.5270-01	0.6060 07	
6	9	1/2 - 3/2	-0.7280 00	0.1900-01	0.1370 07	-0.2870 00	0.2930-02	0.3050 06	
		3/2 - 3/2	-0.7650 00	0.2050-02	0.3990 06	-0.9370 00	0.9330-03	0.1940 06	
		3/2 - 5/2	-0.7620 00	0.1810-01	0.2350 07	-0.5120 00	0.8130-02	0.1060 07	
6	10	1/2 - 3/2	0.5550 00	0.1140-01	0.1270 07	0.1190 00	0.5290-03	0.5900 05	
		3/2 - 3/2	0.5840 00	0.1230-02	0.2570 06	-0.7990-01	0.2300-04	0.4820 04	
		3/2 - 5/2	0.5790 00	0.1090-01	0.1520 07	-0.6250-01	0.1260-03	0.1770 05	
6	11	1/2 - 3/2	-0.4430 00	0.7450-02	0.8700 06	0.3440 00	0.4430-02	0.5230 06	
		3/2 - 3/2	-0.4650 00	0.7950-03	0.1750 06	0.1810 00	0.1210-03	0.2660 05	
		3/2 - 5/2	-0.4620 00	0.7050-02	0.1040 07	0.1940 00	0.1250-02	0.1930 06	
6	12	1/2 - 3/2	0.3650 00	0.5140-02	0.6200 06	0.4670 00	0.9330-02	0.1010 07	
		3/2 - 3/2	0.3830 00	0.5480-03	0.1250 06	0.3320 00	0.4130-03	0.9400 05	
		3/2 - 5/2	0.3800 00	0.4860-02	0.7370 06	0.3420 00	0.3940-02	0.5980 06	
6	13	1/2 - 3/2	-0.3080 00	0.3700-02	0.4570 06	0.5320 00	0.1130-01	0.1160 07	
		3/2 - 3/2	-0.3220 00	0.3930-03	0.9160 05	0.4200 00	0.6600-03	0.1550 06	
		3/2 - 5/2	-0.3200 00	0.3500-02	0.5430 06	0.4270 00	0.6220-02	0.9650 06	
6	14	1/2 - 3/2	0.2650 00	0.2760-02	0.3470 06	0.5620 00	0.1240-01	0.1560 07	
		3/2 - 3/2	0.2770 00	0.2930-03	0.6740 05	0.4670 00	0.8320-03	0.1170 06	
		3/2 - 5/2	0.2750 00	0.2600-02	0.4110 06	0.4730 00	0.7680-02	0.1210 07	
7	6	1/2 - 3/2	0.1560 02	0.4060 00	0.9180 05	0.2190 02	0.7730 00	0.1760 06	
		3/2 - 3/2	0.1560 02	0.3180-01	0.8760 04	0.2060 02	0.5730-01	0.1570 05	
		3/2 - 5/2	0.1570 02	0.3090 00	0.6450 05	0.2120 02	0.5630 00	0.1170 06	
7	7	1/2 - 3/2	0.5480 01	0.2610 00	0.1530 07	-0.1830 02	0.2900 01	0.1730 08	
		3/2 - 3/2	0.6230 01	0.3230-01	0.3620 06	-0.1960 02	0.3200 00	0.3590 07	
		3/2 - 5/2	0.6040 01	0.2740 00	0.2070 07	-0.1940 02	0.2330 01	0.2140 09	
7	8	1/2 - 3/2	-0.2650 01	0.8620-01	0.1050 07	-0.7750 01	0.7340 00	0.9950 07	
		3/2 - 3/2	-0.2890 01	0.9910-02	0.2270 06	-0.8200 01	0.7930-01	0.1430 08	
		3/2 - 5/2	-0.2830 01	0.8590-01	0.1320 07	-0.8140 01	0.7340 00	0.1040 08	
7	9	1/2 - 3/2	0.1680 01	0.4020-01	0.6680 05	0.1790 01	0.4430-02	0.1420 06	
		3/2 - 3/2	-0.4500 01	0.2900 00	0.4820 07	-0.4810 01	0.3230-01	0.1320 07	
		3/2 - 5/2	-0.1770 01	0.3320-01	0.8280 05	-0.4770 01	0.2860 00	0.6740 07	
7	10	1/2 - 3/2	-0.1200 01	0.2240-01	0.4430 05	-0.2900 01	0.1310 00	0.2000 07	
		3/2 - 3/2	-0.1270 01	0.2470-02	0.9310 05	-0.3150 01	0.1520-01	0.5720 06	
		3/2 - 5/2	-0.1260 01	0.2170-01	0.5450 06	-0.3130 01	0.1340 00	0.3380 07	

$np^2p_j - md^2O_j$
 THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES
 CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF FAYE AND DAMGAARD AND BY THE NON-RELATIVISTIC
 HYDROGENIC APPROXIMATION.

TRANSITION			BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
7	11	1/2 - 3/2	0.919D 00	0.139D-01	0.307D 06	-0.195D 01	0.627J-01	0.136J J7
		3/2 - 3/2	0.972D 00	0.152D-02	0.641D 05	-0.217D 01	0.758J-02	0.320J 06
		3/2 - 5/2	0.960D 00	0.134D-01	0.376D 06	-0.215D 01	0.669J-01	0.188J 07
7	12	1/2 - 3/2	-0.738D 00	0.931D-02	0.221D 06	-0.134D 01	0.304D-01	0.722J 06
		3/2 - 3/2	-0.778D 00	0.101D-02	0.459D 05	-0.153D 01	0.390J-02	0.177J 06
		3/2 - 5/2	-0.776D 00	0.891D-02	0.270D 06	-0.151D 01	0.343D-01	0.104J J7
7	13	1/2 - 3/2	0.612D 00	0.655D-02	0.163D 06	-0.913D 00	0.146J-01	0.364J 06
		3/2 - 3/2	0.643D 00	0.709D-03	0.339D 05	-0.108D 01	0.202D-02	0.968J 05
		3/2 - 5/2	0.637D 00	0.626D-02	0.200D 06	-0.107D 01	0.177J-01	0.564D J6
7	14	1/2 - 3/2	-0.519D 00	0.481D-02	0.125D 06	-0.615D 00	0.073D-02	0.175J 06
		3/2 - 3/2	-0.545D 00	0.519D-03	0.258D 05	-0.706D 00	0.103D-02	0.513J J5
		3/2 - 5/2	-0.540D 00	0.458D-02	0.152D 06	-0.756D 00	0.899D-02	0.298J J6
8	7	1/2 - 3/2	0.276D 02	0.524D 00	0.200D 05	0.323D 02	0.714D 00	0.273D 05
		3/2 - 3/2	0.277D 02	0.399D-01	0.174D 04	0.313D 02	0.509J-01	0.223D 04
		3/2 - 5/2	0.278D 02	0.591D 00	0.134D 05	0.317D 02	0.506J 00	0.173J 05
8	8	1/2 - 3/2	0.781D 01	0.260D 00	0.383D 06	-0.312D 02	0.414D 01	0.610J J7
		3/2 - 3/2	0.899D 01	0.331D-01	0.899D 05	-0.329D 02	0.442D 00	0.126J 07
		3/2 - 5/2	0.868D 01	0.279D 00	0.512D 06	-0.326D 02	0.392D 01	0.720J 07
8	9	1/2 - 3/2	-0.373D 01	0.879D-01	0.285D 05	-0.128D 02	0.104J 01	0.338J 07
		3/2 - 3/2	-0.410D 01	0.104D-01	0.637D 05	-0.152D 02	0.108D 00	0.661J 06
		3/2 - 5/2	-0.410D 01	0.892D-01	0.368D 06	-0.132D 02	0.962J 00	0.397J 07
8	10	1/2 - 3/2	0.235D 01	0.421D-01	0.198D 03	-0.779D 01	0.461D 00	0.217J 07
		3/2 - 3/2	0.254D 01	0.481D-02	0.434D 05	-0.802D 01	0.479J-01	0.432J 06
		3/2 - 5/2	0.250D 01	0.418D-01	0.252D 06	-0.799D 01	0.429J 00	0.258J J7
8	11	1/2 - 3/2	-0.168D 01	0.240D-01	0.140D 05	-0.541D 01	0.243D 00	0.145D 07
		3/2 - 3/2	-0.180D 01	0.270D-02	0.303D 05	-0.561D 01	0.261D-01	0.294J 06
		3/2 - 5/2	-0.177D 01	0.235D-01	0.177D 06	-0.556D 01	0.233J 00	0.175J 07
8	12	1/2 - 3/2	0.129D 01	0.152D-01	0.102D 06	-0.401D 01	0.146D 00	0.931J 06
		3/2 - 3/2	0.138D 01	0.169D-02	0.219D 05	-0.419D 01	0.156D-01	0.202J 06
		3/2 - 5/2	0.136D 01	0.148D-01	0.128D 06	-0.417D 01	0.143J 00	0.121J J7
8	13	1/2 - 3/2	-0.104D 01	0.103D-01	0.762D 05	-0.308D 01	0.974J-01	0.658J J6
		3/2 - 3/2	-0.110D 01	0.114D-02	0.163D 05	-0.560D 01	0.560J-02	0.141J 06
		3/2 - 5/2	-0.109D 01	0.100D-01	0.953D 05	-0.323D 01	0.374D-01	0.836J 06
8	14	1/2 - 3/2	0.865D 00	0.738D-02	0.584D 05	-0.242D 01	0.574J-01	0.458J 06
		3/2 - 3/2	0.916D 00	0.814D-03	0.124D 05	-0.258D 01	0.643J-02	0.984J 05
		3/2 - 5/2	0.904D 00	0.714D-02	0.728D 05	-0.256D 01	0.573J-01	0.589J 06
9	8	1/2 - 3/2	0.426D 02	0.640D 00	0.645D 04	0.452D 02	0.720J 00	0.727J J4
		3/2 - 3/2	0.428D 02	0.479D-01	0.534J 03	0.440D 02	0.505J-01	0.564J 03
		3/2 - 5/2	0.429D 02	0.473D 00	0.418D 04	0.444D 02	0.505D 00	0.447J 04
9	9	1/2 - 3/2	0.104D 02	0.262D 00	0.124D 05	-0.462D 02	0.514D 01	0.244J 07
		3/2 - 3/2	0.121D 02	0.341D-01	0.299D 05	-0.483D 02	0.541J 00	0.474J J6
		3/2 - 5/2	0.117D 02	0.286D 00	0.169D 06	-0.478D 02	0.481J 01	0.295J 07
9	10	1/2 - 3/2	-0.491D 01	0.896D-01	0.100D 05	-0.181D 02	0.122D 00	0.136J 07
		3/2 - 3/2	-0.545D 01	0.108D-01	0.229D 05	-0.184D 02	0.123D 00	0.251J 06
		3/2 - 5/2	-0.531D 01	0.322D-01	0.131D 05	-0.184D 02	0.113J 01	0.157J J7

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE ISOTOPES $np^2P - md^2D, J'$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$
9	11	1/2 - 3/2	0.368D 01	0.434D-01	0.739D 05	-0.110D 02	0.551D 00	0.938D 06
		3/2 - 3/2	0.336D 01	0.505D-02	0.165D 03	-0.111D 02	0.554D-01	3.141D 06
		3/2 - 5/2	0.529D 01	0.437D-01	0.955D 05	-0.111D 02	0.493D 00	3.109D 07
9	12	1/2 - 3/2	-0.220D 01	0.251D-01	0.548D 05	-0.777D 01	0.313D 00	0.684D 06
		3/2 - 3/2	-0.237D 01	0.287D-02	0.121D 05	-0.789D 01	0.317D-01	0.134D 06
		3/2 - 5/2	-0.223D 01	0.249D-01	0.703D 05	-0.788D 01	0.285D 00	0.803D 05
9	13	1/2 - 3/2	0.169D 01	0.161D-01	0.413D 05	-0.593D 01	0.153D 00	0.510D 06
		3/2 - 3/2	0.181D 01	0.182D-02	0.907D 04	-0.605D 01	0.202D-01	0.101D 06
		3/2 - 5/2	0.178D 01	0.158D-01	0.527D 05	-0.604D 01	0.182D 00	0.605D 06
9	14	1/2 - 3/2	-0.136D 01	0.110D-01	0.319D 05	-0.473D 01	0.133D 00	0.385D 06
		3/2 - 3/2	-0.145D 01	0.124D-02	0.696D 04	-0.484D 01	0.133D-01	0.770D 05
		3/2 - 5/2	-0.143D 01	0.108D-01	0.405D 05	-0.483D 01	0.123D 00	0.461D 06
10	9	1/2 - 3/2	0.665D 02	0.755D 00	0.261D 04	0.604D 02	0.751D 00	0.260D 06
		3/2 - 3/2	0.668D 02	0.561D-01	0.210D 03	0.588D 02	0.524D-01	0.197D 03
		3/2 - 5/2	0.609D 02	0.556D 00	0.167D 04	0.593D 02	0.526D 00	0.158D 04
10	10	1/2 - 3/2	0.134D 02	0.268D 00	0.491D 05	-0.633D 02	0.603D 01	0.110D 07
		3/2 - 3/2	0.156D 02	0.354D-01	0.121D 05	-0.659D 02	0.623D 00	0.214D 06
		3/2 - 5/2	0.150D 02	0.256D 00	0.680D 05	-0.653D 02	0.560D 01	0.129D 07
10	11	1/2 - 3/2	-0.622D 01	0.715D-01	0.417D 05	-0.237D 02	0.133D 01	0.606D 06
		3/2 - 3/2	-0.694D 01	0.112D-01	0.971D 04	-0.238D 02	0.132D 00	0.115D 06
		3/2 - 5/2	-0.675D 01	0.952D-01	0.556D 05	-0.238D 02	0.113D 01	0.693D 06
10	12	1/2 - 3/2	0.388D 01	0.447D-01	0.322D 05	-0.141D 02	0.594D 00	0.428D 06
		3/2 - 3/2	0.425D 01	0.528D-02	0.733D 04	-0.142D 02	0.583D-01	3.813D 05
		3/2 - 5/2	0.415D 01	0.455D-01	0.422D 05	-0.142D 02	0.523D-00	3.491D 06
10	13	1/2 - 3/2	-0.276C 01	0.260D-01	0.247D 05	-0.100D 02	0.343D 00	0.327D 06
		3/2 - 3/2	-0.299D 01	0.301D-02	0.555D 04	-0.101D 02	0.300D 00	0.626D 05
		3/2 - 5/2	-0.253D 01	0.261D-01	0.321D 05	-0.101D 02	0.300D 00	0.377D 06
10	14	1/2 - 3/2	0.211D 01	0.168D-01	0.192D 05	-0.773D 01	0.224D 00	0.257D 06
		3/2 - 3/2	0.228D 01	0.193D-02	0.429D 04	-0.777D 01	0.223D-01	3.496D 05
		3/2 - 5/2	0.224D 01	0.167D-01	0.248D 05	-0.777D 01	0.201D 00	3.293D 06
11	10	1/2 - 3/2	0.814D 02	0.870D 00	0.122D 04	0.777D 02	0.732D 00	0.111D 04
		3/2 - 3/2	0.818D 02	0.643D-01	0.965D 02	0.757D 02	0.551D-01	0.826D 03
		3/2 - 5/2	0.819D 02	0.637D 00	0.770D 05	0.763D 02	0.554D 00	0.609D 03
11	11	1/2 - 3/2	0.166D 02	0.275D 00	0.223D 05	-0.828D 02	0.644D 01	0.534D 06
		3/2 - 3/2	0.156D 02	0.369D-01	0.556D 04	-0.857D 02	0.733D 00	3.197D 06
		3/2 - 5/2	0.188D 02	0.307D 00	0.313D 05	-0.851D 02	0.632D 01	0.644D 06
11	12	1/2 - 3/2	-0.765D 01	0.740D-01	0.197D 05	-0.296D 02	0.140D 01	0.233D 06
		3/2 - 3/2	-0.860D 01	0.116D-01	0.466D 04	-0.296D 02	0.133D 00	0.552D 05
		3/2 - 5/2	-0.834D 01	0.988D-01	0.266D 05	-0.296D 02	0.124D 01	0.335D 06
11	13	1/2 - 3/2	0.473D 01	0.459D-01	0.157D 05	-0.173D 02	0.613D 00	3.203D 06
		3/2 - 3/2	0.522D 01	0.549D-02	0.362D 04	-0.172D 02	0.595D-01	3.330D 05
		3/2 - 5/2	0.509D 01	0.471D-01	0.208D 05	-0.172D 02	0.533D 00	3.238D 06
11	14	1/2 - 3/2	-0.336D 01	0.269D-01	0.124D 05	-0.122D 02	0.354D 00	3.164D 06
		3/2 - 3/2	-0.367D 01	0.316D-02	0.203D 04	-0.121D 02	0.364D-01	3.309D 05
		3/2 - 5/2	-0.359D 01	0.272D-01	0.163D 05	-0.121D 02	0.312D 00	0.187D 06

$\alpha^2 P_{JJ} - m^2 D_{JJ}$

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES
 CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC
 HYDROGENIC APPROXIMATION.

TRANSITION N M J-J'			BATES-DAMGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$		
12	11	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.105D 03 0.116D 03 0.106D 03	0.982D 00 0.723D-01 0.719D 00	0.628D 03 0.492D 02 0.395D 03	0.971D 02 0.948D 02 0.955D 02	0.837D 00 0.583D-01 0.585D 00	0.535D 03 0.394D 02 0.322D 03		
12	12	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.203D 02 0.240D 02 0.229D 02	0.286D 00 0.387D-01 0.321D 00	0.113D 05 0.285D 04 0.160D 03	-0.105D 03 -0.108D 03 -0.107D 03	0.761D 01 0.735D 00 0.701D 01	0.301D 06 0.578D 05 0.549D 06		
12	13	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	-0.921D 01 -0.104D 02 -0.101D 02	0.968D-01 0.121D-01 0.102D 00	0.102D 05 0.245D 04 0.139D 05	-0.358D 02 -0.357D 02 -0.357D 02	0.140D 01 0.142D 00 0.124D 01	0.155D 06 0.288D 05 0.175D 06		
12	14	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	-0.205D 02 0.629D 01 -0.203D 02	0.618D 00 0.573D-02 0.540D 00	0.109D 06 0.196D 04 0.123D 05	0.568D 01 -0.203D 02 0.613D 01	0.475D-01 0.595D-01 0.490D-01	0.859D 04 0.203D 05 0.112D 05		
13	12	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.132D 03 0.135D 03 0.133D 03	0.109D 01 0.803D-01 0.798D 00	0.351D 03 0.271D 02 0.219D 03	0.119D 03 0.116D 03 0.117D 03	0.880D 00 0.612D-01 0.618D 00	0.234D 03 0.207D 02 0.169D 03		
13	13	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.241D 02 0.287D 02 0.274D 02	0.295D 00 0.403D-01 0.334D 00	0.615D 04 0.157D 04 0.879D 04	-0.126D 03 -0.132D 03 -0.132D 03	0.835D 01 0.850D 00 0.767D 01	0.174D 06 0.334D 05 0.202D 06		
13	14	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	-0.109D 02 -0.124D 02 -0.120D 02	0.100D 00 0.126D-01 0.106D 00	0.573D 04 0.138D 04 0.785D 04	-0.424D 02 -0.421D 02 -0.422D 02	0.151D 01 0.145D 00 0.132D 01	0.564D 05 0.160D 05 0.974D 05		
14	13	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.162D 03 0.163D 03 0.163D 03	0.121D 01 0.888D-01 0.882D 00	0.212D 03 0.163D 02 0.131D 03	0.143D 03 0.139D 03 0.140D 03	0.943D 00 0.651D-01 0.655D 00	0.165D 03 0.113D 02 0.973D 02		
14	14	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.284D 02 0.340D 02 0.324D 02	0.307D 00 0.423D-01 0.349D 00	0.359D 04 0.923D 03 0.516D 04	-0.155D 03 -0.159D 03 -0.158D 03	0.907D 01 0.923D 00 0.831D 01	0.106D 06 0.203D 05 0.123D 06		
15	14	1/2 - 3/2 3/2 - 3/2 3/2 - 5/2	0.194D 03 0.196D 03 0.196D 03	0.132D 01 0.965D-01 0.963D 00	0.130D 03 0.100D 02 0.816D 02	0.168D 03 0.165D 03 0.166D 03	0.983D 00 0.683D-01 0.692D 00	0.974D 02 0.711D 01 0.587D 02		

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE $3s^2 3p^5 \text{ } ^2F_{3/2} - \text{ } ^2D_{3/2}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION			BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
4	7	5/2 - 3/2	0.8400 01	0.1350 00	0.3360 03	-0.1390 30	0.3710-04	3.9200 02
		5/2 - 5/2	0.8260 01	0.3450-02	0.1610 03	-0.1370 30	0.2530-05	3.4400 01
		7/2 - 5/2	0.8260 01	0.1420 00	0.3210 03	-0.1370 30	0.3830-04	3.8300 02
4	8	5/2 - 3/2	-0.1620 01	0.1060-01	0.1180 04	0.3320-01	0.4470-05	0.4790 02
		5/2 - 5/2	-0.1620 01	0.7640-03	0.3520-01	0.3610-06	0.3610-06	0.2700 01
		7/2 - 5/2	-0.1620 01	0.1150-01	0.1140 06	0.3530-01	0.5120-05	0.5410 32
4	9	5/2 - 3/2	0.7780 00	0.3200-02	0.6080 05	0.7740-01	0.3170-04	3.6320 33
		5/2 - 5/2	0.7820 00	0.2310-03	0.2940 04	0.7870-01	0.2350-05	0.2980 02
		7/2 - 5/2	0.7820 00	0.3470-02	0.5880 05	0.7870-01	0.3520-04	0.5980 03
4	10	5/2 - 3/2	-0.4880 00	0.1440-02	0.3600 05	0.8720-01	0.4520-04	0.1150 34
		5/2 - 5/2	-0.4510 00	0.1050-03	0.1740 04	0.8610-01	0.3370-05	0.5610 32
		7/2 - 5/2	-0.4510 00	0.1570-02	0.3490 05	0.8810-01	0.5050-04	0.1120 04
4	11	5/2 - 3/2	0.3470 00	0.7920-03	0.2330 05	0.8600-01	0.4870-04	0.1430 04
		5/2 - 5/2	0.3490 00	0.5750-04	0.1130 04	0.8650-01	0.3520-05	0.6920 32
		7/2 - 5/2	0.3490 00	0.8620-03	0.2260 05	0.8650-01	0.5230-04	0.1390 04
4	12	5/2 - 3/2	-0.2650 00	0.4890-03	0.1600 05	0.8090-01	0.4530-04	0.1490 34
		5/2 - 5/2	-0.2670 00	0.3550-04	0.7760 03	0.8120-01	0.3230-05	3.7170 32
		7/2 - 5/2	-0.2670 00	0.5320-03	0.1550 03	0.8120-01	0.4320-04	0.1440 34
4	13	5/2 - 3/2	0.2130 00	0.3260-03	0.1150 05	0.7470-01	0.4030-04	0.1420 04
		5/2 - 5/2	0.2140 00	0.2370-04	0.3590 03	0.7500-01	0.2800-05	0.6830 32
		7/2 - 5/2	0.2140 00	0.3560-03	0.1120 03	0.7500-01	0.4330-04	3.1370 04
4	14	5/2 - 3/2	-0.1760 00	0.2300-03	0.8580 04	0.6850-01	0.3880-04	0.1300 34
		5/2 - 5/2	-0.1780 00	0.1670-04	0.4160 04	0.6870-01	0.2500-05	3.6220 02
		7/2 - 5/2	-0.1780 00	0.2510-03	0.8320 04	0.6870-01	0.3750-04	0.1240 04
5	8	5/2 - 3/2	0.1680 02	0.2860 00	0.2030 06	-0.3360 30	0.1150-03	0.9100 02
		5/2 - 5/2	0.1650 02	0.2020-01	0.9750 04	-0.3310 30	0.8100-05	0.3920 01
		7/2 - 5/2	0.1650 02	0.3020 00	0.1950 06	-0.3310 00	0.1220-03	0.7840 32
5	9	5/2 - 3/2	-0.3230 01	0.2350-01	0.8110 05	0.2550-01	0.1470-05	0.5080 31
		5/2 - 5/2	-0.3230 01	0.1690-02	0.3930 04	0.2960-01	0.1420-06	0.3270 30
		7/2 - 5/2	-0.3230 01	0.2540-01	0.7850 05	0.2970-01	0.2140-05	3.6610 01
5	10	5/2 - 3/2	0.1550 01	0.7240-02	0.4520 05	0.1260 00	0.4900-04	0.2990 33
		5/2 - 5/2	0.1550 01	0.5250-03	0.2190 04	0.1280 00	0.3580-05	0.1490 02
		7/2 - 5/2	0.1550 01	0.7870-02	0.4380 05	0.1280 00	0.5380-04	0.2490 03
5	11	5/2 - 3/2	-0.9670 00	0.3320-02	0.2840 05	0.1540 00	0.8430-04	3.7210 03
		5/2 - 5/2	-0.9740 00	0.2410-03	0.1380 00	0.1560 00	0.5170-05	0.3530 02
		7/2 - 5/2	-0.9740 00	0.3620-02	0.2760 05	0.1560 00	0.9260-04	3.7060 03
5	12	5/2 - 3/2	0.6660 00	0.1850-02	0.1920 05	0.1580 30	0.9730-04	0.1010 34
		5/2 - 5/2	0.6920 00	0.1340-03	0.9320 03	0.1590 00	0.7670-05	3.4910 32
		7/2 - 5/2	0.6920 00	0.2010-02	0.1860 05	0.1590 30	0.1060-03	0.9330 33
5	13	5/2 - 3/2	-0.5250 00	0.1150-02	0.1370 05	0.1520 00	0.9430-04	3.1130 34
		5/2 - 5/2	-0.5290 00	0.8380-04	0.6640 05	0.1530 00	0.7000-05	3.5540 02
		7/2 - 5/2	-0.5290 00	0.1260-02	0.1330 05	0.1530 00	0.1030-03	0.1110 04
5	14	5/2 - 3/2	0.4210 00	0.7760-03	0.1010 05	0.1430 00	0.8990-04	3.1170 34
		5/2 - 5/2	0.4240 00	0.5640-04	0.4910 03	0.1440 00	0.6480-05	0.5630 32
		7/2 - 5/2	0.4240 00	0.8470-03	0.9810 04	0.1440 00	0.9720-04	3.1130 34

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_{j_1} - mf^2F_{j_1}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{ij}	f_{ij}	$A_{ij} (\text{sec}^{-1})$	R_{ij}	f_{ij}	$A_{ij} (\text{sec}^{-1})$
8	10	3/2 - 5/2	-0.821D 00	0.306D-02	0.845D 04	-0.978D 71	0.438D 00	0.120D 77
		5/2 - 5/2	-0.857D 00	0.173D-03	0.710D 03	-0.988D 01	0.210D-01	0.801D 05
		5/2 - 7/2	-0.858D 00	0.347D-02	0.107D 05	-0.988D 01	0.420D 00	0.123D 07
8	11	3/2 - 5/2	0.768D 00	0.245D-02	0.784D 04	-0.782D 71	0.293D 00	3.957D 06
		5/2 - 5/2	0.766D 00	0.136D-03	0.647D 03	-0.787D 01	0.145D-01	3.683D 05
		5/2 - 7/2	0.766D 00	0.272D-02	0.971D 04	-0.787D 01	0.287D 00	0.102D 97
8	12	3/2 - 5/2	-0.602D 00	0.187D-02	0.665D 04	-0.635D 01	0.204D 00	0.741D 06
		5/2 - 5/2	-0.649D 00	0.103D-03	0.545D 03	-0.638D 01	0.925D-02	0.527D 05
		5/2 - 7/2	-0.649D 00	0.206D-02	0.818D 04	-0.638D 01	0.193D 00	0.770D 06
9	7	3/2 - 5/2	0.584D 02	0.199D 01	0.901D 05	-0.190D 02	0.210D 00	0.951D 04
		5/2 - 5/2	0.588D 02	0.938D-01	0.810D 04	-0.187D 02	0.951D-02	0.618D 03
		5/2 - 7/2	0.588D 02	0.188D 01	0.914D 05	-0.187D 02	0.190D 00	0.926D 04
9	8	3/2 - 5/2	-0.181D 01	0.507D-02	0.163D 04	0.831D 02	0.107D 02	0.143D 07
		5/2 - 5/2	-0.139D 01	0.142D-03	0.674D 02	0.831D 02	0.505D 00	0.239D 06
		5/2 - 7/2	-0.139D 01	0.282D-02	0.100D 04	0.831D 02	0.101D 02	0.359D 07
9	9	3/2 - 5/2	-0.394D 00	0.343D-03	0.224D 03	-0.119D 02	0.314D 00	0.206D 06
		5/2 - 5/2	-0.610D 00	0.389D-04	0.378D 02	-0.125D 02	0.163D-01	0.158D 05
		5/2 - 7/2	-0.611D 00	0.783D-03	0.570D 03	-0.125D 02	0.325D 00	0.237D 06
9	10	3/2 - 5/2	0.665D 00	0.112D-02	0.115D 04	-0.151D 02	0.613D 00	0.594D 06
		5/2 - 5/2	0.803D 00	0.621D-04	0.118D 03	-0.154D 02	0.301D-01	0.432D 05
		5/2 - 7/2	0.804D 00	0.164D-02	0.177D 04	-0.154D 02	0.602D 00	0.647D 06
9	11	3/2 - 5/2	-0.657D 00	0.131D-02	0.162D 04	-0.124D 02	0.465D 00	0.575D 06
		5/2 - 5/2	-0.755D 00	0.621D-04	0.151D 03	-0.125D 02	0.223D-01	0.413D 05
		5/2 - 7/2	-0.755D 00	0.164D-02	0.226D 04	-0.125D 02	0.450D 00	0.620D 05
9	12	3/2 - 5/2	0.581D 00	0.112D-02	0.163D 04	-0.386D 01	0.321D 00	0.463D 06
		5/2 - 5/2	0.657D 00	0.675D-04	0.147D 03	-0.992D 01	0.154D-01	0.335D 05
		5/2 - 7/2	0.657D 00	0.135D-02	0.220D 04	-0.992D 01	0.308D 00	0.502D 06
10	8	3/2 - 5/2	0.771D 02	0.228D 01	0.449D 03	-0.260D 02	0.253D 00	0.510D 04
		5/2 - 5/2	0.776D 02	0.108D 00	0.304D 04	-0.256D 02	0.117D-01	0.332D 03
		5/2 - 7/2	0.776D 02	0.215D 01	0.456D 05	-0.256D 02	0.213D 00	3.477D 04
10	9	3/2 - 5/2	-0.352D 01	0.130D-01	0.190D 04	0.107D 03	0.121D 02	0.177D 07
		5/2 - 5/2	-0.301D 01	0.448D-03	0.968D 02	0.107D 03	0.571D 00	3.123D 06
		5/2 - 7/2	-0.301D 01	0.893D-02	0.145D 04	0.107D 03	0.114D 02	0.145D 07
10	10	3/2 - 5/2	0.218D 00	0.721D-04	0.223D 02	-0.146D 02	0.324D 00	3.100D 06
		5/2 - 5/2	-0.454D-01	0.176D-06	0.806D-01	-0.153D 02	0.164D-01	3.769D 04
		5/2 - 7/2	-0.505D-01	0.367D-05	0.126D 01	-0.153D 02	0.336D 00	3.115D 06
10	11	3/2 - 5/2	0.343D 00	0.221D-03	0.103D 03	-0.188D 02	0.653D 00	3.308D 06
		5/2 - 5/2	0.514D 00	0.234D-04	0.163D 02	-0.190D 02	0.322D-01	3.243D 05
		5/2 - 7/2	0.514D 00	0.463D-03	0.244D 03	-0.190D 02	0.643D 00	0.336D 06
10	12	3/2 - 5/2	-0.431D 00	0.397D-03	0.242D 03	-0.153D 02	0.493D 00	3.105D 06
		5/2 - 5/2	-0.553D 00	3.309D-04	0.281D 02	-0.154D 02	0.241D-01	0.219D 05
		5/2 - 7/2	-0.553D 00	0.519D-03	0.422D 03	-0.154D 02	0.432D 00	0.329D 06
11	9	3/2 - 5/2	0.981D 02	0.256D 01	0.242D 05	-0.340D 02	0.307D 00	3.290D 04
		5/2 - 5/2	0.588D 02	0.121D 00	0.164D 04	-0.335D 02	0.134D-01	3.183D 03
		5/2 - 7/2	0.988D 02	0.242D 01	0.246D 05	-0.335D 02	0.273D 00	3.283D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_J - mf^2F_J$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	K	J-J'	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ} (\text{sec}^{-1})$
8	10	3/2 - 5/2	-0.8210 00	0.3060-02	0.8450 04	-0.9780 01	0.4340 00	0.1200 37
		5/2 - 5/2	-0.2570 00	0.1730-03	0.7100 03	-0.9880 01	0.2100-01	0.8210 05
		5/2 - 7/2	-0.8580 00	0.3470-02	0.1070 05	-0.9880 01	0.4200 00	0.1230 07
8	11	3/2 - 5/2	0.7680 00	0.2450-02	0.7840 04	-0.7820 01	0.2940 00	3.9570 36
		5/2 - 5/2	0.7660 00	0.1360-03	0.6470 03	-0.7870 01	0.1440-01	3.6830 05
		5/2 - 7/2	0.7660 00	0.2720-02	0.9710 04	-0.7870 01	0.2870 00	0.1020 07
8	12	3/2 - 5/2	-0.6020 00	0.1870-02	0.6660 04	-0.6350 01	0.2080 00	0.7410 36
		5/2 - 5/2	-0.6490 00	0.1030-03	0.5450 03	-0.6380 01	0.9750-02	0.5270 05
		5/2 - 7/2	-0.6490 00	0.2060-02	0.8180 04	-0.6380 01	0.1940 00	0.7300 06
9	7	3/2 - 5/2	0.5840 02	0.1990 01	0.9010 05	-0.1900 02	0.2100 00	0.9510 04
		5/2 - 5/2	0.5880 02	0.9380-01	0.6100 04	-0.1870 02	0.9510-02	0.6180 03
		5/2 - 7/2	0.5880 02	0.1880 01	0.9140 05	-0.1870 02	0.1900 00	0.9260 04
9	8	3/2 - 5/2	-0.1810 01	0.5070-02	0.1630 04	0.8310 02	0.1070 02	0.3430 07
		5/2 - 5/2	-0.1390 01	0.1420-03	0.6740 02	0.8310 02	0.5050 00	0.2390 06
		5/2 - 7/2	-0.1390 01	0.2820-02	0.1000 04	0.8310 02	0.1010 02	0.3590 37
9	9	3/2 - 5/2	-0.3940 00	0.3430-03	0.2240 03	-0.1190 02	0.3140 00	0.2080 06
		5/2 - 5/2	-0.6110 00	0.3890-04	0.3780 02	-0.1250 02	0.1640-01	0.1380 05
		5/2 - 7/2	-0.6110 00	0.7830-03	0.5700 03	-0.1250 02	0.3250 00	0.2370 06
9	10	3/2 - 5/2	0.6650 00	0.1190-02	0.1150 04	-0.1510 02	0.6150 00	0.5940 06
		5/2 - 5/2	0.8030 00	0.8210-04	0.1180 03	-0.1540 02	0.3010-01	0.4320 35
		5/2 - 7/2	0.8040 00	0.1640-02	0.1770 04	-0.1540 02	0.6020 00	0.6470 36
9	11	3/2 - 5/2	-0.6570 00	0.1310-02	0.1620 04	-0.1240 02	0.4680 00	0.5730 06
		5/2 - 5/2	-0.7550 00	0.8210-04	0.1510 03	-0.1250 02	0.2250-01	0.8130 05
		5/2 - 7/2	-0.7550 00	0.1640-02	0.2260 04	-0.1250 02	0.4500 00	0.6600 33
9	12	3/2 - 5/2	0.5810 00	0.1120-02	0.1630 04	-0.3860 01	0.3210 00	0.4630 06
		5/2 - 5/2	0.6570 00	0.6750-04	0.1470 03	-0.3920 01	0.1540-01	0.3350 05
		5/2 - 7/2	0.6570 00	0.1350-02	0.2200 04	-0.3920 01	0.3080 00	0.5020 06
10	8	3/2 - 5/2	0.7710 02	3.2280 01	0.4490 05	-0.2600 02	0.2540 00	0.5100 34
		5/2 - 5/2	0.7760 02	0.1680 00	0.3040 04	-0.2560 02	0.1170-01	0.3320 33
		5/2 - 7/2	0.7760 02	0.2150 01	0.4560 05	-0.2560 02	0.2350 00	3.4970 34
10	9	3/2 - 5/2	-0.3520 01	0.1300-01	0.1900 04	0.1070 03	0.1210 02	0.1770 37
		5/2 - 5/2	-0.3010 01	0.4480-03	0.9680 02	0.1070 03	0.5710 00	3.1230 36
		5/2 - 7/2	-0.3010 01	0.8930-02	0.1450 04	0.1070 03	0.1140 02	0.1450 37
10	10	3/2 - 5/2	0.2180 00	0.7210-04	0.2230 02	-0.1460 02	0.3240 00	3.1000 34
		5/2 - 5/2	-0.4540-01	0.1760-06	0.8060-01	-0.1530 02	0.1630-01	3.7090 34
		5/2 - 7/2	-0.5050-01	0.3670-05	0.1260 01	-0.1530 02	0.3700 30	3.1150 36
10	11	3/2 - 5/2	0.3430 00	0.2210-03	0.1030 03	-0.1880 02	0.6530 00	0.3980 06
		5/2 - 5/2	0.5140 00	0.2340-04	0.1630 02	-0.1900 02	0.3220-01	3.2830 35
		5/2 - 7/2	0.5140 00	0.4630-03	0.2440 03	-0.1900 02	0.6430 00	0.5360 06
10	12	3/2 - 5/2	-0.4310 00	0.3970-03	0.2420 03	-0.1510 02	0.4940 00	3.3050 36
		5/2 - 5/2	-0.5530 00	3.3090-04	0.2810 02	-0.1540 02	0.2130-01	0.2130 35
		5/2 - 7/2	-0.5530 00	0.3190-03	0.4220 03	-0.1540 02	0.4320 00	0.3290 06
11	9	3/2 - 5/2	0.5810 02	0.2560 01	0.2420 05	-0.3400 02	0.3670 00	3.2300 34
		5/2 - 5/2	0.5880 02	0.1210 00	0.1640 04	-0.3350 02	0.1340-01	0.1940 03
		5/2 - 7/2	0.9880 02	0.2420 01	0.2460 05	-0.3350 02	0.2750 00	3.2830 34

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2D_{j_1} - m^2F_{j_2}$, CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DANGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DANGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
N	M	R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$	R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$
11	10	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.544D 01 -0.482D 01 -0.481D 01	0.219D-01 0.811D-03 0.162D-01	0.160D 04 0.876D 02 0.131D 04	0.135D 03 0.135D 03 0.135D 03	0.981D 06 0.685D 05 3.103D 07
11	11	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	0.923D 00 0.557D 00 0.597D 00	0.528D-03 0.184D-04 0.368D-03	0.147D 03 0.434D 01 0.651D 02	-0.177D 02 -0.185D 02 -0.185D 02	0.540D 05 0.416D 04 3.624D 05
11	12	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.727D-01 0.134D 00 0.134D 00	0.718D-05 0.116D-05 0.232D-04	0.177D 01 0.426D 00 0.638D 01	-0.227D 02 -0.231D 02 -0.231D 02	3.173D 06 0.126D 05 3.188D 06
12	10	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	0.121D 03 0.122D 03 0.122D 03	0.283D 01 0.134D 00 0.268D 01	0.140D 05 0.548D 03 0.142D 05	-0.429D 02 -0.424D 02 -0.424D 02	0.175D 04 0.114D 03 0.171D 04
12	11	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	-0.764D 01 -0.691D 01 -0.691D 01	0.317D-01 0.122D-02 0.244D-01	0.125D 04 0.711D 02 0.107D 04	0.165D 03 0.165D 03 0.165D 03	0.580D 06 0.405D 05 0.608D 06
12	12	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	0.182D 01 0.144D 01 0.144D 01	0.268D-02 0.790D-04 0.158D-02	0.235D 03 0.103D 02 0.154D 03	-0.209D 02 -0.219D 02 -0.219D 02	0.310D 05 0.238D 04 0.358D 05
13	11	3/2 - 5/2 5/2 - 5/2 5/2 - 7/2	0.147D 03 0.148D 03 0.148D 03	0.310D 01 0.147D 00 0.293D 01	0.849D 04 0.577D 03 0.866D 04	-0.529D 02 -0.522D 02 -0.522D 02	0.110D 04 0.716D 02 3.107D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_J - m^2F_J$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF RATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BASES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
6	10	3/2 - 5/2	-0.7150 00	0.7190-02	0.1900 06	-0.3350 01	0.1770 00	0.4680 07
		5/2 - 5/2	-0.7360 00	0.3610-03	0.1410 03	-0.3570 01	0.8470-02	0.3320 06
		5/2 - 7/2	-0.7360 00	0.7210-02	0.2120 06	-0.3570 01	0.1630 00	0.4990 07
6	11	3/2 - 5/2	0.6560 00	0.5290-02	0.1470 06	-0.2950 01	0.1260 00	0.3490 07
		5/2 - 5/2	0.6230 00	0.2650-03	0.1090 03	-0.2360 01	0.6200-02	0.2470 06
		5/2 - 7/2	0.6230 00	0.5290-02	0.1640 06	-0.2960 01	0.1230 00	0.3710 07
6	12	3/2 - 5/2	-0.5190 00	0.3950-02	0.1140 05	-0.2510 01	0.9230-01	0.2660 07
		5/2 - 5/2	-0.5330 00	0.1970-03	0.8450 04	-0.2520 01	0.4400-02	0.1880 06
		5/2 - 7/2	-0.5330 00	0.3950-02	0.1270 06	-0.2520 01	0.8790-01	0.2820 07
7	5	3/2 - 5/2	0.2810 02	0.1330 01	0.5030 06	-0.7240 01	0.1760 00	0.4020 05
		5/2 - 5/2	0.2820 02	0.6230-01	0.3390 03	-0.7830 01	0.4800-02	0.2510 04
		5/2 - 7/2	0.2820 02	0.1250 01	0.5080 06	-0.7830 01	0.9600-01	0.3910 05
7	6	3/2 - 5/2	0.1090 01	0.4580-02	0.1150 05	0.4310 02	0.7720 01	0.1790 08
		5/2 - 5/2	0.1340 01	0.3520-03	0.1200 04	0.4310 02	0.3650 00	0.1240 07
		5/2 - 7/2	0.1340 01	0.7080-02	0.1810 05	0.4310 02	0.7300 01	0.1570 08
7	7	3/2 - 5/2	-0.1410 01	0.1120-01	0.4780 03	-0.7820 01	0.3460 00	0.1480 07
		5/2 - 5/2	-0.1530 01	0.6270-03	0.3960 04	-0.8140 01	0.1770-01	0.1120 06
		5/2 - 7/2	-0.1530 01	0.1260-01	0.5960 03	-0.8150 01	0.3550 00	0.1680 07
7	8	3/2 - 5/2	0.1170 01	0.9020-02	0.5280 05	-0.9160 01	0.5500 00	0.3260 07
		5/2 - 5/2	0.1240 01	0.4850-03	0.4220 04	-0.9110 01	0.2720-01	0.2360 06
		5/2 - 7/2	0.1250 01	0.9720-02	0.6340 05	-0.9120 01	0.5440 00	0.3540 07
7	9	3/2 - 5/2	-0.9550 00	0.6630-02	0.4710 03	-0.7480 01	0.4070 00	0.2890 07
		5/2 - 5/2	-0.1010 01	0.3510-03	0.3700 04	-0.7550 01	0.1970-01	0.2070 05
		5/2 - 7/2	-0.1010 01	0.7040-02	0.5960 03	-0.7550 01	0.3930 00	0.3110 07
7	10	3/2 - 5/2	0.7540 00	0.4890-02	0.3940 03	-0.5980 01	0.2770 00	0.2230 07
		5/2 - 5/2	0.8360 00	0.2570-03	0.3070 04	-0.6020 01	0.1330-01	0.1590 06
		5/2 - 7/2	0.8360 00	0.5140-02	0.4600 05	-0.6020 01	0.2660 00	0.2390 07
7	11	3/2 - 5/2	-0.6720 00	0.3650-02	0.3210 05	-0.4870 01	0.1720 00	0.1670 07
		5/2 - 5/2	-0.7650 00	0.1910-03	0.2490 04	-0.4890 01	0.9200-02	0.1200 06
		5/2 - 7/2	-0.7650 00	0.3820-02	0.3740 05	-0.4890 01	0.1340 00	0.1300 07
7	12	3/2 - 5/2	0.5710 00	0.2730-02	0.2560 03	-0.4060 01	0.1330 00	0.1290 07
		5/2 - 5/2	0.5990 00	0.1420-03	0.1780 04	-0.4070 01	0.6370-02	0.9160 05
		5/2 - 7/2	0.5990 00	0.2840-02	0.2970 05	-0.4070 01	0.1310 00	0.1370 07
8	6	3/2 - 5/2	0.4210 02	0.1680 01	0.2000 06	-0.1300 02	0.1530 00	0.1700 05
		5/2 - 5/2	0.4240 02	0.7900-01	0.1350 03	-0.1280 02	0.7210-02	0.1230 06
		5/2 - 7/2	0.4240 02	0.1580 01	0.2030 03	-0.1280 02	0.1440 00	0.1350 05
8	7	3/2 - 5/2	-0.2890 00	0.2040-03	0.1620 03	0.6160 02	0.9200 01	0.7300 07
		5/2 - 5/2	-0.3670-01	0.1550-06	0.1820 03	0.6170 02	0.5120 00	0.5120 06
		5/2 - 7/2	-0.4680-01	0.3840-05	0.3370 01	0.6170 02	0.8730 01	0.7690 07
8	8	3/2 - 5/2	-0.9280 00	0.2930-02	0.4550 04	-0.9650 01	0.3170 00	0.4310 06
		5/2 - 5/2	-0.1100 01	0.1940-03	0.4450 03	-0.1010 02	0.1530-01	0.3730 05
		5/2 - 7/2	-0.1100 01	0.3890-02	0.6700 04	-0.1010 02	0.3270 00	0.5530 06
8	9	3/2 - 5/2	0.9340 00	0.3550-02	0.7860 04	-0.1190 02	0.5730 00	0.1280 07
		5/2 - 5/2	0.1040 01	0.2090-03	0.6860 03	-0.1210 02	0.2430-01	0.9310 05
		5/2 - 7/2	0.1040 01	0.4180-02	0.1030 05	-0.1210 02	0.5670 00	0.1400 07

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2 - n'l^2f$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	$f - j'$	R_{ij}	f_{ij}	$A_{ij} (\text{sec}^{-1})$	R_{ij}	f_{ij}	$A_{ij} (\text{sec}^{-1})$
5	4	3/2 - 5/2	0.4810 01	0.4210 00	0.1860 03	0.1430 02	0.3720 01	0.1640 39
		5/2 - 5/2	0.4890 01	0.2050 -01	0.1340 07	0.1430 02	0.1750 00	0.1140 08
		5/2 - 7/2	0.4500 01	0.4110 00	0.2000 03	0.1430 02	0.3510 01	0.1710 39
5	5	3/2 - 5/2	-0.2610 01	0.1550 00	0.1070 08	-0.5900 01	0.7920 00	0.5430 03
		5/2 - 5/2	-0.2640 01	0.7480 -02	0.7640 00	-0.6030 01	0.3910 -01	0.3900 07
		5/2 - 7/2	-0.2640 01	0.1500 00	0.1150 03	-0.6030 01	0.7810 00	0.5980 09
5	6	3/2 - 5/2	0.1730 01	0.7590 -01	0.6450 07	-0.5080 01	0.6400 00	0.5520 09
		5/2 - 5/2	0.1750 01	0.3650 -02	0.4590 06	-0.5130 01	0.5130 01	0.3950 07
		5/2 - 7/2	0.1750 01	0.7310 -01	0.6890 07	-0.5130 01	0.6280 00	0.5920 08
5	7	3/2 - 5/2	-0.1280 01	0.4340 -01	0.4140 07	-0.3850 01	0.3570 00	0.3780 38
		5/2 - 5/2	-0.1290 01	0.2090 -02	0.2950 06	-0.3880 01	0.1900 -01	0.2600 07
		5/2 - 7/2	-0.1290 01	0.4170 -01	0.4420 07	-0.3880 01	0.3810 00	0.4030 08
5	8	3/2 - 5/2	0.9550 00	0.2740 -01	0.2800 07	-0.3000 01	0.2480 00	0.2550 09
		5/2 - 5/2	0.1000 01	0.1310 -02	0.1990 06	-0.3010 01	0.1190 -01	0.1900 37
		5/2 - 7/2	0.1000 01	0.2630 -01	0.2990 07	-0.3010 01	0.2380 00	0.2700 08
5	9	3/2 - 5/2	-0.8070 00	0.1840 -01	0.1980 07	-0.2410 01	0.1640 00	0.1760 08
		5/2 - 5/2	-0.8120 00	0.8850 -03	0.1410 06	-0.2420 01	0.7930 -02	0.1250 07
		5/2 - 7/2	-0.8120 00	0.1770 -01	0.2110 07	-0.2420 01	0.1570 00	0.1870 08
5	10	3/2 - 5/2	0.6730 00	0.1310 -01	0.1450 07	-0.1990 01	0.1140 00	0.1260 08
		5/2 - 5/2	0.6780 00	0.6260 -03	0.1030 06	-0.1990 01	0.5420 -02	0.8920 06
		5/2 - 7/2	0.6780 00	0.1250 -01	0.1540 07	-0.1990 01	0.1030 00	0.1340 08
5	11	3/2 - 5/2	-0.5730 00	0.9580 -02	0.1090 07	-0.1680 01	0.8220 -01	0.3350 37
		5/2 - 5/2	-0.5770 00	0.4590 -03	0.7740 05	-0.1680 01	0.5910 -02	0.6590 06
		5/2 - 7/2	-0.5770 00	0.9180 -02	0.1160 07	-0.1680 01	0.7820 -01	0.9880 37
5	12	3/2 - 5/2	0.4550 00	0.7200 -02	0.8340 09	-0.1440 01	0.6130 -01	0.7100 07
		5/2 - 5/2	0.4580 00	0.3650 -03	0.5920 05	-0.1450 01	0.2910 -02	0.5000 36
		5/2 - 7/2	0.4580 00	0.6900 -02	0.8880 06	-0.1450 01	0.5830 -01	0.7510 07
6	4	3/2 - 5/2	0.1600 02	0.8740 00	0.1380 07	-0.3860 01	0.5100 -01	0.8050 35
		5/2 - 5/2	0.1600 02	0.4090 -01	0.9250 05	-0.3800 01	0.2310 -02	0.5210 04
		5/2 - 7/2	0.1600 02	0.8180 00	0.1390 07	-0.3800 01	0.4610 -01	0.7820 05
6	5	3/2 - 5/2	0.2480 01	0.4920 -01	0.4200 06	0.2740 02	0.5990 01	0.5120 38
		5/2 - 5/2	0.2680 01	0.2650 -02	0.3330 05	0.2740 02	0.2830 00	0.3550 07
		5/2 - 7/2	0.2660 01	0.5310 -01	0.5000 05	0.2740 02	0.5660 01	0.5330 08
6	6	3/2 - 5/2	-0.1870 01	0.3660 -01	0.5360 06	-0.6490 01	0.4400 00	0.6450 37
		5/2 - 5/2	-0.1550 01	0.1880 -02	0.4080 05	-0.6710 01	0.2230 -01	0.4820 36
		5/2 - 7/2	-0.1550 01	0.3770 -01	0.6130 06	-0.6710 01	0.4450 00	0.7230 37
6	7	3/2 - 5/2	0.1380 01	0.2280 -01	0.4360 06	-0.6850 01	0.5610 00	0.1070 38
		5/2 - 5/2	0.1430 01	0.1160 -02	0.3200 05	-0.6950 01	0.2740 -01	0.7750 06
		5/2 - 7/2	0.1430 01	0.2320 -01	0.4920 06	-0.6960 01	0.5470 00	0.1160 06
6	8	3/2 - 5/2	-0.1070 01	0.1480 -01	0.3310 05	-0.5480 01	0.3990 00	0.8690 37
		5/2 - 5/2	-0.1100 01	0.7460 -03	0.2470 05	-0.5540 01	0.1630 -01	0.6210 36
		5/2 - 7/2	-0.1110 01	0.1430 -01	0.3710 05	-0.5540 01	0.3750 00	0.9120 37
6	9	3/2 - 5/2	0.8620 00	0.1010 -01	0.2490 06	-0.4360 01	0.2570 00	0.6390 37
		5/2 - 5/2	0.8880 00	0.5070 -03	0.1960 05	-0.4390 01	0.1240 00	0.4550 36
		5/2 - 7/2	0.8850 00	0.1020 -01	0.2790 05	-0.4390 01	0.2430 00	0.6820 37

THEORETICAL RADIAL MATRIX ELEMENTS: OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $n^2F - md^2D_1$, CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION IN n STATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{ij}	f_{ij}	$A_{ij} (\text{sec}^{-1})$	R_{ij}	f_{ij}	$A_{ij} (\text{sec}^{-1})$
9	12	5/2 - 3/2	0.700D 02	0.919D 00	0.219D 05	-0.185D 01	0.644D-03	0.154D J2
		5/2 - 5/2	0.689D 02	0.646D-01	0.106D 04	-0.184D 01	0.461D-04	0.758D J0
		7/2 - 5/2	0.689D 02	0.968D 00	0.212D 05	-0.184D 01	0.692D-03	0.151D J2
9	13	5/2 - 3/2	-0.130D 02	0.762D-01	0.106D 05	-0.291D 00	0.385D-04	0.537D J1
		5/2 - 5/2	-0.130D 02	0.553D-02	0.519D 03	-0.276D 00	0.243D-05	0.234D 00
		7/2 - 5/2	-0.130D 02	0.829D-01	0.104D 05	-0.276D 00	0.373D-04	0.467D J1
9	14	5/2 - 3/2	0.607D 01	0.241D-01	0.696D 04	0.192D 00	0.243D-04	0.636D J1
		5/2 - 5/2	0.613D 01	0.176D-02	0.341D 03	0.203D 00	0.192D-05	0.373D 00
		7/2 - 5/2	0.613D 01	0.264D-01	0.682D 04	0.203D 00	0.287D-04	0.747D J1
10	13	5/2 - 3/2	0.883D 02	0.108D 01	0.139D 05	-0.241D 01	0.801D-03	0.104D J2
		5/2 - 5/2	0.862D 02	0.756D-01	0.671D 03	-0.240D 01	0.575D-04	0.510D 00
		7/2 - 5/2	0.869D 02	0.113D 01	0.134D 05	-0.240D 01	0.862D-03	0.102D 02
10	14	5/2 - 3/2	-0.162D 02	0.688D-01	0.687D 04	-0.439D 00	0.652D-04	0.504D 01
		5/2 - 5/2	-0.163D 02	0.545D-02	0.336D 03	-0.421D 00	0.429D-05	0.224D 00
		7/2 - 5/2	-0.163D 02	0.968D-01	0.672D 04	-0.421D 00	0.644D-04	0.447D J1
11	14	5/2 - 3/2	0.109D 03	0.124D 01	0.911D 04	-0.304D 01	0.962D-03	0.710D J1
		5/2 - 5/2	0.107D 03	0.968D-01	0.440D 03	-0.302D 01	0.691D-04	0.351D 00
		7/2 - 5/2	0.107D 03	0.130D 01	0.881D 04	-0.302D 01	0.194D-02	0.701D J1

$$n^2 f_{IJ} - m^2 D_{IJ}$$

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DANGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION			BATES-DANGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R _{IJ}	f _{IJ}	A _{IJ} (sec ⁻¹)	R _{IJ}	f _{IJ}	A _{IJ} (sec ⁻¹)	A _{IJ} (sec ⁻¹)	
6	9	5/2 - 3/2 5/2 - 5/2	0.2710 02 0.2660 02 0.2660 02	0.4450 00 0.3120 01 0.4680 00	0.1110 00 0.5330 04 0.1070 0a	-0.6060 00 -0.6000 00 -0.6000 00	0.2230-03 0.1560-04 0.2370-03	0.5550 02 0.2700 01 0.5910 02		
6	10	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.5160 01 -0.5180 01 -0.5180 01	0.3690-01 0.2660-02 0.3990-01	0.4780 05 0.2320 04 0.4650 05	-0.1160-01 -0.5170-02 -0.5080-02	0.1950-06 0.2650-08 0.3340-07	0.2400 00 0.2310-02 0.4460-01		
6	11	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.2460 01 0.2480 01 0.2480 01	0.1150-01 0.8360-03 0.1250-01	0.2830 05 0.1380 04 0.2750 05	0.1610 00 0.1660 00 0.1660 00	0.4950-04 0.3740-05 0.5620-04	0.1220 02 0.6150 01 0.1230 03		
6	12	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.1530 01 -0.1550 01 -0.1550 01	0.5330-02 0.3880-03 0.5820-02	0.1860 05 0.9040 03 0.1810 05	0.2160 00 0.2190 00 0.2190 00	0.1060-03 0.7760-05 0.1810-03	0.3680 02 0.1810 02 0.3620 05		
6	13	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.1090 01 0.1100 01 0.1100 01	0.2990-02 0.2180-03 0.3270-02	0.1390 05 0.6330 03 0.1270 05	0.2290 00 0.2310 00 0.2310 00	0.1330-03 0.9670-05 0.1450-03	0.5780 03 0.2810 02 0.5530 03		
6	14	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.8290 00 -0.8370 00 -0.8370 00	0.1880-02 0.1370-03 0.2050-02	0.9520 04 0.4630 03 0.9260 04	0.2260 00 0.2270 00 0.2270 00	0.1390-03 0.1010-04 0.1510-03	0.7050 03 0.3410 02 0.5820 03		
7	10	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.3940 02 0.3870 02 0.3870 02	0.6030 00 0.4230-01 0.6340 00	0.2190 05 0.2980 04 0.5780 03	-0.9500 00 -0.9420 00 -0.9410 00	0.3510-03 0.2500-04 0.3750-03	0.3600 02 0.1760 01 0.3520 02		
7	11	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.7430 01 -0.7460 01 -0.7460 01	0.5020-01 0.3630-02 0.5450-01	0.2810 05 0.1370 04 0.2740 05	-0.7720-01 -0.6810-01 -0.6800-01	0.5420-05 0.3030-06 0.4520-05	0.3030 01 0.1140 00 0.2270 01		
7	12	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.3520 01 0.3550 01 0.3550 01	0.1580-01 0.1150-02 0.1720-01	0.1740 03 0.8460 03 0.1690 05	0.1840 00 0.1900 00 0.1900 00	0.4300-04 0.3300-05 0.4350-04	0.4740 02 0.2430 01 0.4870 02		
7	13	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.2190 01 -0.2210 01 -0.2210 01	0.7360-02 0.5360-03 0.8040-02	0.1180 05 0.5760 03 0.1150 05	0.2720 00 0.2760 00 0.2770 00	0.1140-03 0.8420-05 0.1250-03	0.1930 03 0.9050 01 0.1310 03		
7	14	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.1540 01 0.1560 01 0.1560 01	0.4140-02 0.3020-03 0.4540-02	0.8500 04 0.4140 03 0.8290 04	0.2980 00 0.3020 00 0.3020 00	0.1550-03 0.1130-04 0.1700-03	0.3180 03 0.1550 02 0.3100 03		
8	11	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.5360 02 0.5280 02 0.5280 02	0.7610 00 0.5340-01 0.8010 00	0.3610 05 0.1740 04 0.3480 05	-0.1360 01 -0.1360 01 -0.1360 01	0.4300-03 0.3520-04 0.5280-03	0.2490 02 0.1150 01 0.2290 02		
8	12	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.1000 02 -0.1010 02 -0.1010 02	0.6320-01 0.4590-02 0.6880-01	0.1700 05 0.8280 03 0.1660 05	-0.1710 00 -0.1590 00 -0.1590 00	0.1340-04 0.1140-05 0.1710-04	0.4940 01 0.2060 00 0.4120 01		
8	13	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	0.4720 01 0.4770 01 0.4770 01	0.2000-01 0.1460-02 0.2180-01	0.1030 05 0.5300 01 0.1060 05	0.1540 00 0.2030 00 0.2030 00	0.3330-04 0.2630-05 0.3350-04	0.1940 02 0.1940 00 0.1920 02		
8	14	5/2 - 3/2 5/2 - 5/2 7/2 - 5/2	-0.2920 01 -0.2950 01 -0.2950 01	0.3330-02 0.6820-03 0.1020-01	0.7590 04 0.3710 03 0.7410 04	0.3220 00 0.3280 00 0.3280 00	0.1140-03 0.3430-05 0.1270-03	0.6250 02 0.4590 01 0.9190 02		

APPENDIX 13 : Non-Relativistic Thallium Oscillator Strengths

THEORETICAL RADIAL MATRIX ELEMENTS, OSCILLATOR STRENGTHS AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_{1/2} - ms^2S_{1/2}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DANGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

N	M	TRANSITION J-J'	BATES-DANGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
			R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
6	7	1/2-1/2	0.1270 01	7.4370-01	0.2020 09	-0.2330 01	0.2300 00	0.1070 09
		3/2-1/2	0.2100 01	0.3370-01	0.3930 03	-0.2010 01	0.7610-01	0.3540 03
6	9	1/2-1/2	-0.3600 00	0.5100-02	0.5100 07	-0.3050 00	0.3640-02	0.3640 07
		3/2-1/2	-0.5000 00	0.7920-02	0.1900 03	-0.1330 01	0.5960-01	0.7610 03
6	7	1/2-1/2	0.1950 00	0.1660-02	0.2000 07	0.1760 01	0.4830-01	0.6070 03
		3/2-1/2	0.2530 00	0.2500-02	0.4170 07	0.3350 00	0.4020-02	0.6710 07
6	10	1/2-1/2	-0.1270 00	0.7360-03	0.1010 07	0.1240 01	0.7050-01	0.7650 03
		3/2-1/2	-0.1730 00	0.1140-02	0.2140 07	0.1020 01	0.3930-01	0.7370 03
7	8	1/2-1/2	0.7200 01	0.2410 00	0.3380 07	0.1510 01	0.1050-01	0.1430 03
		3/2-1/2	0.3750 01	0.3780 00	0.4770 07	0.4690 01	0.7930-01	0.1370 07
7	9	1/2-1/2	-0.1430 01	0.2010-01	0.1090 07	-0.3130 01	0.3250-01	0.5010 07
		3/2-1/2	-0.1380 01	0.1540-01	0.1310 07	-0.2630 01	0.5600-01	0.4790 07
7	10	1/2-1/2	0.7710 00	0.6700-02	0.5540 06	-0.2410 01	0.0550-01	0.5420 07
		3/2-1/2	0.7030 00	0.5060-02	0.6940 06	-0.2610 01	0.6930-01	0.7570 07
8	9	1/2-1/2	0.1470 02	0.3330 00	0.8470 03	0.7230 01	0.7640-01	0.2030 03
		3/2-1/2	0.1720 02	0.4290 00	0.1160 07	0.1170 02	0.1970 00	0.5340 03
8	10	1/2-1/2	-0.2740 01	0.2930-01	0.3070 06	-0.2670 01	0.2430-01	0.2910 03
		3/2-1/2	-0.2380 01	0.1050-01	0.3450 06	-0.1460 01	0.7630-02	0.1230 03
9	10	1/2-1/2	0.2330 02	0.3300 00	0.2970 03	0.1370 02	0.1730 00	0.9630 03
		3/2-1/2	0.2770 02	0.5720 00	0.4120 03	0.1930 02	0.2770 00	0.1930 06

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THEORETICAL RADIAL MATRIX ELEMENTS CALCULATED BY THE METHOD OF BATES AND DANGAARD FOR THE TRANSITIONS \$S_{1/2} - P_{1/2}\$
 CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE METHOD OF BATES AND DANGAARD AND BY THE NON-RELATIVISTIC
 HYDROGENIC APPROXIMATION

TRANSITION		BATES-DANGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	J-J'								
7	1/2-1/2	0.7270 01	0.4110 00	0.1620 08	0.2230 01	0.8790 03			
	1/2-3/2	3.6840 01	0.1240 00	0.2070 08	2.3610 01	0.3130 03			
7	1/2-1/2	-0.5840 00	0.1450-01	0.2160 07	0.4610-01	0.1420 03			
	1/2-3/2	-0.1350 01	0.5660-01	0.4400 07	0.7450-01	0.5790 07			
7	1/2-1/2	0.4570 00	0.3780-02	0.8090 06	0.1920 01	0.4030 07			
	1/2-3/2	3.6820 00	0.1710-01	0.1950 07	0.0510-01	0.0290 07			
9	1/2-1/2	0.1460 02	0.5660 00	0.2590 07	0.2870 01	0.1230 08			
	1/2-3/2	0.1350 02	0.1100 01	0.5300 07	0.5370 01	0.1590 03			
8	1/2-1/2	-0.1910 01	0.2040-01	0.4400 06	0.4370 00	0.3210 07			
	1/2-3/2	-0.2570 01	0.7770-01	0.9760 06	0.7410 00	0.3390 07			
9	1/2-1/2	0.2400 02	0.7110 00	0.6790 06	0.3110 01	0.3050 07			
	1/2-3/2	0.2200 02	0.1370 01	0.8380 06	0.8210 01	0.4050 07			

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES FOR THE S_{1/2}, P_{1/2}, P_{3/2}, D_{3/2}, D_{5/2}, AND F_{7/2} STATES OF HYDROGEN. CALCULATED FROM WAVE FUNCTIONS PREPARED BY THE COULOMB APPROXIMATION METHOD AND DAMAGED BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

N	M	TRANSITION J-J'	HATES-DANGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
			R _{IJ}	f _{IJ}	A _{IJ} (sec ⁻¹)	R _{IJ}	f _{IJ}	A _{IJ} (sec ⁻¹)
6	A	3/2-1/2	0.192D 01	0.135D-01	0.720D 00	0.255D 01	0.345D-01	0.127D J7
		3/2-3/2	0.174D 01	0.140D-02	0.730D 00	0.161D 01	0.297D-02	0.620D J5
		5/2-3/2	0.184D 01	0.227D-01	0.658D 00	0.166D 01	0.187D-01	0.573D J0
6	Q	3/2-1/2	-0.797D 00	0.331D-02	0.483D 00	-0.102D 00	0.997D-04	0.908D 04
		3/2-3/2	-0.590D 00	0.597D-03	0.283D 00	-0.355D 00	0.216D-03	0.103D 05
		5/2-1/2	-0.617D 00	0.157D-02	0.270D 00	-0.115D 00	0.101D-02	0.705D 05
7	Q	3/2-1/2	0.566D 01	0.768D-01	0.575D 00	0.346D 01	0.281D-01	0.216D J5
		3/2-3/2	0.389C 01	0.420D-02	0.340D 00	0.323D 01	0.537D-02	0.234D J5
		5/2-3/2	0.411D 01	0.116D-01	0.326D 00	0.328D 01	0.321D-01	0.203D 00



THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $np^2P_1 - md^2D_1$, CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

N	M	J-J'	BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
			R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$
6	6	1/2-3/2	0.199D 01	0.290D 00	0.126D 04	0.363D 01	0.962D 00	0.413D 07
		3/2-3/2	0.275D 01	0.435D-01	0.233D 08	0.157D 01	0.141D-01	0.754D 07
		3/2-5/2	0.265D 01	0.375D 00	0.135D 04	0.167D 01	0.145D 00	0.517D 03
6	7	1/2-1/2	-0.102D 01	0.881D-01	0.518D 08	0.307D 01	0.900D 01	0.471D 07
		3/2-3/2	-0.130D 01	0.116D-01	0.930D 07	0.247D 01	0.422D-01	0.130D 08
		3/2-5/2	-0.120D 01	0.102D 00	0.531D 08	0.251D 01	0.392D 00	0.205D 03
6	8	1/2-3/2	0.660D 00	0.334D-01	0.262D 08	0.236D 01	0.504D 00	0.335D 07
		3/2-3/2	0.614D 00	0.493D-02	0.449D 07	0.235D 01	0.414D-01	0.375D 03
		3/2-5/2	0.604D 00	0.435D-01	0.263D 08	0.237D 01	0.377D 00	0.229D 07
7	6	1/2-3/2	0.515D 01	0.332D 00	0.424D 06	0.190D 02	0.144D 01	0.194D 07
		3/2-3/2	0.515D 01	0.122D-01	0.991D 04	0.171D 02	0.570D-01	0.348D 05
		3/2-5/2	0.925D 01	0.162D 00	0.777D 05	0.174D 02	0.572D 00	0.275D 06
7	7	1/2-3/2	0.394D 01	0.246D 00	0.507D 07	-0.716D 01	0.815D 00	0.168D 08
		3/2-3/2	0.547D 01	0.415D-01	0.130D 07	-0.111D 02	0.167D 00	0.530D 07
		3/2-5/2	0.531D 01	0.354D 00	0.747D 07	-0.109D 02	0.145D 01	0.307D 08
7	8	1/2-3/2	-0.195D 01	0.946D-01	0.312D 07	-0.161D 01	0.554D-01	0.204D 07
		3/2-3/2	-0.246D 01	0.116D-01	0.701D 06	-0.345D 01	0.230D-01	0.138D 07
		3/2-5/2	-0.242D 01	0.101D 00	0.410D 07	-0.334D 01	0.194D 00	0.784D 07
8	7	1/2-3/2	0.195D 02	0.425D 00	0.683D 05	0.290D 02	0.110D 01	0.151D 06
		3/2-3/2	0.156D 02	0.211D-01	0.103D 04	0.264D 02	0.381D-01	0.186D 04
		3/2-5/2	0.157D 02	0.219D 00	0.924D 04	0.267D 02	0.401D 00	0.169D 05
8	8	1/2-3/2	0.592D 01	0.234D 00	0.854D 06	-0.191D 02	0.245D 01	0.892D 07
		3/2-3/2	0.850D 01	0.429D-01	0.246D 06	-0.241D 02	0.345D 00	0.193D 07
		3/2-5/2	0.821D 01	0.363D 00	0.141D 07	-0.237D 02	0.303D 01	0.117D 03
9	8	1/2-3/2	0.326D 02	0.620D 00	0.178D 05	0.413D 02	0.101D 01	0.286D 03
		3/2-3/2	0.329D 02	0.241D-01	0.195D 03	0.378D 02	0.319D-01	0.258D 03
		3/2-5/2	0.330D 02	0.257D 00	0.195D 04	0.383D 02	0.349D 00	0.262D 04

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES, FOR THE SERIES $nd^2D_J - mf^2F_{J'}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

TRANSITION		BATES-DAMGAARD APPROXIMATION				NON-RELATIVISTIC HYDROGENIC APPROXIMATION			
N	M	J-J'	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	R_{IJ}	f_{IJ}	$A_{IJ}(\text{sec}^{-1})$	
6	5	3/2-5/2	0.912D 01	0.939D 00	0.160D 08	0.219D 02	0.544D 01	0.929D 08	
		5/2-5/2	0.924D 01	0.453D-01	0.113D 07	0.219D 02	0.255D 00	0.638D 07	
6	5	5/2-7/2	0.924D 01	0.906D 00	0.170D 08	0.219D 02	0.511D 01	0.956D 08	
		3/2-5/2	-0.343D 01	0.186D 00	0.627D 07	-0.419D 01	0.279D 00	0.939D 07	
6	6	5/2-5/2	-0.342D 01	0.877D-02	0.435D 06	-0.442D 01	0.146D-01	0.725D 06	
		5/2-7/2	-0.342D 01	0.175D 00	0.652D 07	-0.442D 01	0.292D 00	0.109D 08	
7	5	3/2-5/2	0.16CD 02	0.143D 00	0.601D 04	-0.327D 01	0.599D-02	0.251D 03	
		5/2-5/2	0.16CD 02	0.556D-02	0.288D 03	-0.317D 01	0.235D-03	0.114D 02	
7	5	5/2-7/2	0.160D 02	0.119D 00	0.432D 04	-0.317D 01	0.471D-02	0.171D 03	
		3/2-5/2	0.116D 02	0.691D 00	0.243D 07	0.360D 02	0.664D 01	0.233D 09	
7	6	5/2-5/2	0.119D 02	0.341D-01	0.175D 06	0.360D 02	0.312D 00	0.160D 07	
		5/2-7/2	0.119D 02	0.682D 00	0.263D 07	0.360D 02	0.623D 01	0.240D 08	
8	6	3/2-5/2	0.302D 02	0.250D 00	0.253D 04	-0.565D 01	0.877D-02	0.887D 02	
		5/2-5/2	0.301D 02	0.103D-01	0.117D 03	-0.548D 01	0.341D-03	0.389D 01	
8	6	5/2-7/2	0.301D 02	0.206D 00	0.176D 04	-0.548D 01	0.681D-02	0.583D 02	

THEORETICAL RADIAL MATRIX ELEMENTS; OSCILLATOR STRENGTHS; AND TRANSITION PROBABILITIES. FOR THE SERIES $n^2F_{j'} - m^2D_{j''}$ CALCULATED FROM WAVE FUNCTIONS GENERATED BY THE COULOMB APPROXIMATION OF BATES AND DAMGAARD AND BY THE NON-RELATIVISTIC HYDROGENIC APPROXIMATION.

N	M	TRANSITION J-J'	BATES-DAMGAARD APPROXIMATION			NON-RELATIVISTIC HYDROGENIC APPROXIMATION		
			R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$	R_{ij}	f_{ij}	$A_{ij}(\text{sec}^{-1})$
5	3	5/2-3/2	0.274D 01	0.215D-01	0.119D 06	0.189D 00	0.102D-03	0.564D 03
		5/2-5/2	0.266D 01	0.140D-02	0.524D 04	0.201D 00	0.835D-05	0.314D 02
		7/2-5/2	0.260D 01	0.209D-01	0.105D 06	0.291D 00	0.125D-03	0.628D 03

APPENDIX 14: Eutectic Mixtures

The following table lists the percentage of anhydrous materials (by weight) used in a eutectic mixture with crushed ice formed from distilled water. The eutectic temperatures are taken from Lange's Handbook of Chemistry (11th edition) and the values quoted are the lowest temperature that can be obtained from the mixture of the substance with ice. Temperatures between these lowest possible values and 0°C may be obtained by reducing the percentage of anhydrous material added to the mixture.

Anhydrous Material	Percentage	Eutectic Temperature °c
Na Cl	23.3	-21.13
Mg Cl ₂	21.6	-33.6
K Cl	19.75	-11.1

APPENDIX 15: Derivation of the Correction Formula for Thermal Transpiration

Consider two volumes containing a gas at low pressure connected by a capillary. Let the temperature and pressure of the gas in volume V_1 be T_1 and P_1 ; and let the temperature and pressure in volume V_2 be T_2 and P_2 . Assume that the cross sectional area of the capillary at V_1 is the same as at V_2 - let the value of this area be $A \text{ cm}^2$. The number of molecules striking a unit area of the containing vessel per second is

$$N = \frac{1}{4} n \langle v \rangle$$

where n is the density of molecules in the container and v is the average thermal velocity of the molecules. The rate at which the molecules strike the area of the capillary is equal to the rate at which the molecules leave the volume; this rate is

$$q = nA$$

The rate at which molecules leave the volume V_1 is

$$\begin{aligned} q_1 &= \frac{1}{4} n_1 A \langle v_1 \rangle \\ &= \frac{1}{2} n_1 A \left(\frac{2kT_1}{\pi m} \right)^{1/2} \end{aligned}$$

and the rate at which molecules leave the volume V_2 is

$$q_2 = \frac{1}{2} n_2 A \left(\frac{2kT_2}{\pi m} \right)^{1/2}$$

The pressure of the gas in the container is given by

$$P = \frac{1}{3} m n \langle v \rangle^2$$

hence, the pressure of gas in volume V_1 is

$$\begin{aligned} P_1 &= \frac{1}{3} m n_1 \langle v_1 \rangle^2 \\ &= \frac{4}{3} m n_1 \left(\frac{2kT_1}{\pi m} \right) \end{aligned}$$

and the pressure of gas in volume V_2 is

$$P_2 = \frac{4}{3} m n_2 \left(\frac{2kT_2}{\pi m} \right)$$

These can be written in the form

$$P_1 = \frac{8}{3} q_1 \frac{m}{A} \langle v_1 \rangle$$

and

$$P_2 = \frac{8}{3} q_2 \frac{m}{A} \langle v_2 \rangle$$

At equilibrium the rate at which molecules leave V_1 will be identical to the rate at which molecules leave V_2 ; that is,

$$q_1 = q_2$$

The equilibrium condition gives

$$\frac{P_1}{P_2} = \frac{v_1}{v_2} = \sqrt{\frac{T_1}{T_2}}$$

Therefore, the pressure in the cell is related to the pressure measured at room temperature in the manometer by the formula

$$P_{\text{cell}} = \sqrt{\frac{T_{\text{cell}}}{T_{\text{room}}}} P_{\text{man}}$$

where T_{cell} is the temperature of the gas in the cell in degrees Kelvin and T_{room} is the temperature of nitrogen at room temperature in degrees Kelvin; P_{man} is the manometer pressure reading in torr.

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