CYTOBIOLOGIE Zeitschrift für experimentelle Zellforschung

Organ der Deutschen Gesellschaft für Elektronenmikroskopie e.V.

Band/Vol. $11\cdot 1975$



WISSENSCHAFTLICHE VERLAGSGESELLSCHAFT MBH STUTTGART

Arnold, J. M.	
An effect of calcium in cytokinesis as demonstrated with ionophore A 23187 Calciumeffekt in der Cytokinese dargestellt mit Ionophor A 23187	1
Behnke, O.	
Studies on isolated microtubules. Evidence for a clear space component Untersuchungen an isolierten Mikrotubuli. Nachweise für eine äußere Komponente	366
BLAKERS, M., siehe MENZEL, R.	279
Bland, Ch. E., C. Z. Lunney	
Mitotic apparatus of Pilobolus crystallinus Mitoseapparat bei Pilobolus crystallinus	382
BLECHER, ST. R. (Short Communication)	
Actin-like filaments associated with spread chromosomes Actin-ähnliche Filamente verbunden mit gespreiteten Chromosomen	190
Boublik, M., F. Jenkins, H. R. Kaback	
Use of critical point drying in the preparation of Escherichia coli membrane vesicles for electron microscopy	304
Anwendung des critical point-Trocknens bei der elektronenmikroskopischen Präparation von Escherichia coli Membranvesikeln	
Braatz-Schade, K., M. Haberey	
Bioelectrical potentials and motile activity in Amoeba proteus Bioelektrische Potentiale und Bewegungsaktivität bei Amoeba proteus	87
CAMPBELL, G. R., siehe CHAMLEY, J. H.	358
Cassens, R. G., siehe Yamaguchi, M.	335
Chamley, J. H., G. R. Campbell	
Isolated ureteral smooth muscle cells in culture. Including their interaction with intrinsic and extrinsic nerves	358
Glatte Muskelzellen des Urethers in der Gewebekultur und ihre Innervierung	
Cooke, P.	
Filamentous aggregates of purified myosin from smooth muscle Filamentöse Aggregate aus gereinigtem Myosin des glatten Muskels	346
CRANG, R. E., siehe SICKO-GOAD, L. M.	430
Csaba, G., T. Lantos	
Specificity of hormone receptors in Tetrahymena. Experiments with serotonin and histamine antagonists Spezifität der Hormonrezeptoren von Tetrahymena. Versuche mit Antagonisten	44
von Serotonin und Histamin	
	225

DAHL, D., siehe YAMAGUCHI, M.

335

Denker, HW., E. S. E. Hafez	
Proteases and implantation in the rabbit: role of trophoblast vs. uterine secretion	101
Zur Bedeutung von Proteasen für die Implantation beim Kaninchen: Abhängigkeit vom Trophoblasten oder vom Uterussekret?	
Dexheimer, J.	
Etude ultrastructurale de quelques étapes de la différenciation des cellules à reserves du gamétophyte femelle du Ginkgo biloba	264
Ultrastructural study of some differentiation stages of reserve cells in the female gametophyte of Ginkgo biloba	
Dustmann, J. H.	
Die Pigmentgranula im Komplexauge der Honigbiene Apis mellifica bei Wildtyp und verschiedenen Augenfarbmutanten	133
The pigment granules in the compound eye of the honeybee Apis mellifica in wildtype and different eye-color mutants	
Eckert, W. A., W. W. Franke	
Changes in fine structure and composition of macronuclei of Tetrahymena pyriformis induced by drugs interfering with RNA synthesis and processing	392
Feinstruktur und chemische Zusammensetzung von Makronuklei des Ciliaten Tetrahymena pyriformis nach Behandlung mit Inhibitoren der RNA-Synthese	
Egger, G.	
The electron microscopic localisation of succinic dehydrogenase within the membranes of the mitochondrial cristae	110
Die elektronenmikroskopische Lokalisierung der Succinat-Dehydrogenase innerhalb der Membranen der Cristae mitochondriales	
Elbers, P. F., siehe Spies, F.	50
Falk, H., siehe Zentgraf, H.	10
FRANKE, W. W., siehe Eckert, W. A.	392
Franke, W. W., siehe Zentgraf, H.	10
Genbačev, O., V. Šulović	
Localization of human chorionic gonadotropin (HCG) by immunohistochemical method in normal and pathological human placenta	95
Immunhistochemische Lokalisation von menschlichem Choriongonadotropin (HCG) in der Plazenta bei normaler und pathologischer Schwangerschaft	
Goode, D.	
Mitosis of embryonic heart muscle cells in vitro. An immunofluorescence and ultrastructural study	203
Mitose embryonaler Herzmuskelzellen in Gewebekultur: Immunfluoreszenz- und Feinstruktur-Untersuchungen	
Gratzl, M., W. Nastainczyk, D. Schwab	
The spatial arrangement of esterases in the microsomal membrane Die Anordnung der Esterasen in der mikrosomalen Membran	123

INHALT BAND 11 · CONTENTS VOLUME 11	V
Haberey, M., siehe Braatz-Schade, K.	87
HAFEZ, E. S. E., siehe Denker, HW.	101
HAX, W. M. A., siehe Spies, F.	65
Heinrich, G.	
Über die Lokalisation verschiedener Phosphatasen im Nektarium von Aloe On the localization of different phosphatases in the nectary of Aloe	247
JENKINS, F., siehe BOUBLIK, M.	304
Jensen, Th. E., siehe Sicko-Goad, L. M.	430
Jeserich, G., H. Rahmann (Kurzmitteilung)	
Elektronenmikroskopisch-histochemischer Nachweis saurer Polysaccharid- verbindungen im ZNS von Teleosteern Electronmicroscopic-histochemical demonstration of acid polysaccharides in the CNS of teleosts	483
Jobst, K., siehe Kellermayer, M.	240
KABACK, H. R., siehe BOUBLIK, M.	304
Kalinina, L., siehe Sikora, J.	480
KELLERMAYER, M., M. SOMFAI, K. JOBST Determination of saline extractable material of HeLa cell nuclei Bestimmung des in Salzlösungen extrahierbaren Materials von HeLa Zellkernen	240
Kiss, A., siehe Kovács, J.	309
Kitajima, E. W.	
A peculiar type of glycocalyx on the microvilli of the midgut epithelial cells of the thrips Frankliniella sp. (Thysanoptera, Thripidae) Ein besonderer Glycocalyx-Typ auf den Mikrovilli der Mitteldarmepithelzellen von Frankliniella spec. (Thysanoptera, Thripidae)	299
Komnick, H., M. Schmitz, W. Wichard	
Cytologische, elektrolyt-histochemische und funktionelle Untersuchungen der analen Chloridepithelien aquatischer Brachycerenlarven (Insecta, Diptera) Cytological, electrolyte-histochemical and functional studies on the anal chloride epithelia of aquatic Brachycera larvae (Insecta, Diptera)	448
Koshino, Y., siehe Yasuzumi, G.	30
Kovács, J., G. Réz, A. Kiss (Short Communication)	
Vinblastine-induced autophagocytosis and its prevention by cycloheximide and emetine in mouse pancreatic acinar cells in vivo Vinblastin-induzierte Autophagocytose und seine Verhütung durch Cycloheximid und Emetin in der Acinuszelle des Mäusepankreas in vivo	309

Kristen, U.	
Feinstrukturveränderungen in den submersen Laubblattdrüsen von Nomaphila stricta Nees während der Sekretion Alterations in fing structurg of the submerged leaf glands	438
of Nomaphila stricta Nees during secretion	
Lantos, T., siehe Csaba, G.	44
LEUNISSEN, J. L. M., siehe SPIES, F.	50
LINNEMANS, W. A. M., siehe Spies, F.	50
LINNEMANS, W. A. M., siehe Spies, F.	65
Lunney, C. Z., siehe Bland, Ch. E.	382
Menzel, R., M. Blakers	
Functional organization of an insect ommatidium with fused rhabdom Die funktionelle Organisation eines Insektenommatidiums mit fusioniertem Rhabdom	279
Nakai, Y., siehe Yasuzumi, G.	30
NASTAINCZYK, W., siehe GRATZL, M.	123
Natarajan, A. T., siehe Raposa, T.	230
Nilshammar-Holmvall, M.	
The effects of calcium deficiency on cell wall formation and autospore release of the green alga Scenedesmus	419
Wirkungen des Calcium-Mangels auf Zellwandbildung und Entleerung der Autosporen der Grünalge Scenedesmus	
Petzoldt, U. (Short Communication)	
Amino acid incorporation into embryonic proteins during rabbit	490
Aminosäure-Einbau in embryonalen Proteinfraktionen während der Frühentwicklung des Kaninchens	170
Rahmann, H., siehe Jeserich, G.	483
Raposa, T., A. T. Natarajan	
The use of the fluorochrome bis-benzimidazol derivative (Hoechst 33258) in the study of spontaneous and induced chromosome aberrations	230
Die Anwendung des Fluorochromes Bis-Benzimidazolderivat (Hoechst 33258) bei der Untersuchung spontaner und induzierter Chromosomenaberrationen	
Réz, G., siehe Kovács, J.	309
Rubio-Huertos, M., siehe Vega, J.	186
Ruyter de Wildt, Th. M. de, siehe Spies, F.	65

SCHMIDT, K. Das Johnstonsche Organ der primär flügellosen Ectognatha (Lepisma, Zygentoma; Machilis, Archaeognatha) Johnston's organ of the primary apterous Ectognatha (Lepisma, Zygentoma; Machilis, Archaeognatha)	153
Schmitz, M., siehe Komnick, H.	448
Schwab, D., siehe Gratzl, M.	123
Shirai, T., siehe Yasuzumi, G.	30
SICKO-GOAD, L. M., R. E. CRANG, TH. E. JENSEN Phosphate metabolism in blue-green algae. IV. In situ analysis of polyphosphate bodies by x-ray energy dispersive analysis Phosphatstoffwechsel bei Blaualgen. IV. Untersuchung der Polyphosphat-Granula mit Hilfe der energiedispersiven Röntgenanalyse	430
SIKORA, J., L. KALININA (Short Communication) Substrate detachment as a test of antigenic diversity of Amoeba proteus strains Substrat-Ablösung als Test der antigenen Verschiedenheit von Stämmen von Amoeba proteus	480
Somfai, M., siehe Kellermayer, M.	240
SPIES, F., W. A. M. LINNEMANS, P. H. J. TH. VERVERGAERT, J. L. M. LEUNISSEN, P. F. ELBERS Encystment of Acanthamoeba castellanii (Neff). A combined freeze etch – thin sectioning analysis of the cell surface Encystierung von Acanthamoeba castellanii (Neff). Kombinierte Gefrierätz- und Dünnschnittanalyse der Zelloberfläche	50
 SPIES, F., W. A. M. LINNEMANS, TH. M. DE RUYTER DE WILDT, W. M. A. HAX Growth phase dependent Concanavalin A agglutinability of Acanthamoeba castellanii (Neff strain) Die Abhängigkeit der Agglutinierbarkeit von Acanthamoeba castellanii durch Concanavalin A von der Wachstumsphase 	65
Šulović, V., siehe Genbačev, O.	95
TRAUT, W. Die Transkriptionsaktivität der Chromosomen in den Oocyten von Ephestia (Lepidoptera) Transcriptional activity of the chromosomes in oocytes of Ephestia (Lepidoptera)	172
VEGA, J., M. RUBIO-HUERTOS (Short Communication) Atypical cristae in wheat coleoptile mitochondria Ungewöhnliche Cristae-Strukturen in Mitochondrien von Weizenkoleoptilen	186
VERVERGAERT, P. H. J. TH., siehe Spies, F.	50
WALZ, B. (Short Communication) Modified ciliary structures in receptor cells of Macrobiotus hufelandi (Tardigrada) Modifizierte Cilienstrukturen in Rezeptorzellen von Macrobiotus hufelandi (Tardigrada)	181

Werz, G., siehe Zerban, H.	314
Weygoldt, P., siehe Zissler, D.	466
Wichard, W., siehe Komnick, H.	448
YAMAGUCHI, M., R. G. CASSENS, D. DAHL Congenital rod disease: some biochemical aspects of nemaline rods Strukturelle und biochemische Untersuchungen der Kristallstäbchen bei einer Form angeborener Myopathie	335
YASUZUMI, G., T. SHIRAI, Y. NAKAI, Y. KOSHINO Fine structure of nuclei as revealed by electron microscopy. VIII. Possible origin and function of nuclear bodies appearing in precancerous and degenerating cell nuclei Die Feinstruktur des Kerns im elektronenmikroskopischen Bild. VIII. Wahrscheinliche Herkunft und Funktion der in präkanzerösen und degenerierenden Zellkernen auftretenden Nuklear-Körper	30
ZENTGRAF, H., H. FALK, W. W. FRANKE Nuclear membranes and plasma membranes from hen erythrocytes. IV. Characterization of nuclear membrane attached DNA <i>Kernmembranen und Plasmamembranen aus Hühnererythrocyten.</i> IV. Charakterisierung der kernmembranständigen DNA	10
 ZERBAN, H., G. WERZ (Short Communication) Ultrastructure of flagellar microtubules in the green algae Acetabularia mediterranea and Dunaliella salina as revealed in freeze-etch preparations Feinstruktur von Geißel-Mikrotubuli der Grünalgen Acetabularia mediterranea und Dunaliella salina nach Gefrierätzung 	314
ZISSLER, D., P. WEYGOLDT Feinstruktur der embryonalen Lateralorgane der Geißelspinne Tarantula marginemaculata C. L. Koch (Amblypygi, Arachnida) Fine structure of the lateral organs of the embryo of the Whip spider Tarantula marginemaculata C. L. Koch (Amblypygi, Arachnida)	466
Bericht Ultrastrukturelle Probleme normaler und pathologisch veränderter Haut Ultrastructural problems of normal and abnormal skin	321
BUCHBESPRECHUNGEN	
Ashworth, J. M.: Zelldifferenzierung	494
BARRELL, B. G., B. F. C. CLARKE: Handbook of Nucleic Acid Sequences	201
DAVIES, M.: Funktionen biologischer Membranen	494
GARROD, D. R.: Zellentwicklung	494
GRELL, K. G.: Protozoology	202
JACOBI, G. (Hrsg.): Biochemische Cytologie der Pflanzenzelle	201

Jieres, et (Ineg.), sternenbene e) teregre der Inansensene	
LASH, J., J. R. WHITTAKER (Hrsg.): Concepts of Development	333
WOODS, R. A.: Biochemische Genetik	494
Symposium: Mechanoreception	496

The spatial arrangement of esterases in the microsomal membrane

Die Anordnung der Esterasen in der mikrosomalen Membran

MANFRED GRATZL¹), WOLFGANG NASTAINCZYK, and DIETER SCHWAB Fachbereich Theoretische Medizin der Universität des Saarlandes, Homburg/Saar, Germany

Received February 7, 1975

Abstract

Rat liver – microsomal esterases – asymmetric distribution

The location of esterases is studied with microsomal vesicles from rat liver using membrane impermeable charged substrates and inhibitors. By comparing esterase activity against charged and uncharged substrates it is shown that microsomal esterases are not latent. Furthermore charged inhibitors were unable to block both esterase – and amidase – activity differently in disrupted and in intact vesicles. From these biochemical studies it was concluded that the major part of the microsomal esterases/amidases is attached to the cytoplasmic side of the microsomes, which was confirmed by electron microscopic studies with enzyme specific staining showing the electron dense reaction products of the microsomal esterases at the cytoplasmic side exclusively.

Introduction

Microsomes isolated by the usual methods from rat liver are closed vesicles, whose inner surface represents the intracisternal side of the endoplasmic reticulum. The outer surface bearing ribosomes in rough microsomes [17, 18], is exposed to the cytoplasm in the intact cell. Very recently it was concluded from iodination studies with lactoperoxidase that most of the microsomal proteins are faced to the cytoplasmic side of the vesicles [15]. Some particular microsomal proteins, however, seem to be located on the inner surface of the microsomal membrane.

There is no doubt about the orientation of microsomal nucleoside diphosphatase [16]: This enzyme shows a high degree of latency (that means the enzymatic activity being greatly stimulated by treatments which affect the lipoprotein structure of the microsomal vesicles). It is also protected from proteolytic attack as well as from inhibition by a specific antibody in intact microsomes but is susceptible to hydrolytic

¹) Dr. M. GRATZL, Fachbereich Theoretische Medizin der Universität des Saarlandes, 665 Homburg/Saar, Germany.

enzymes and antiserum after various treatments which affect the integrity of the membrane structure. Especially some phosphohydrolase and phosphotransferase reactions catalysed by microsomal glucose-6-phosphatase are in a latent state [2, 31]. Electronmicroscope [17, 18] and centrifugation studies [19] concerning the trapping of phosphate by Pb²⁺ also favour an intramicrosomal location of this enzyme. Enzyme specific staining of the acyltransferases involved in the acylation of α -glycerophosphate provides evidence for the attachment of this enzyme at the inner surface of the microsomal membrane [13]. UDP-glucuronyl transferase shows also latency and may be another enzyme which is located at the intracisternal side of the microsomal membrane [8].

Digestion of microsomes with proteases neither solubilizes nor inactivates microsomal amidase activity [1] and an antibody to the acetanilide-hydrolyzing esterase fails to react with the membrane bound enzyme. From these observations it was concluded that microsomal esterase may be anchored to the inner surface of the vesicular membrane. If this assumption is correct this enzyme should show latency at least with charged substrates and inhibitors since microsomal vesicles are regarded to be impermeable to anionic compounds [22].

In the present paper we investigated therefore the hydrolysis of especially designed charged substrates in comparison with the uncharged parent compounds in intact microsomes and in preparations previously treated with agents affecting the integrity of the membrane structure. To reveal, whether esterase activity is latent or not, also studies with a charged inhibitor were performed. In addition enzyme specific staining methods were employed to detect, on which side of the microsomal membrane the esterase may be located.

Materials and methods

Microsomes were isolated from a 20 % homogenate of rat liver in 0.25 M sucrose/0.01 M sodium phosphate buffer, pH 7.4 by gel filtration [26]. Male Sprague-Dawley rats of body weight 150 to 200 g were used.

Protein was determined according to LowRY et al. [21], using a calibration curve with crystalline bovine serum albumin (Serva, Heidelberg, Germany). Esterase activity was determined by the following methods: Spectrophotometric determination of liberated 4-nitrophenol or 4-carboxy-2-nitrophenol from 4-nitrophenyl acetate (= nitrophenyl acetate) and 4-carboxy-2-nitrophenyl acetate (= carboxynitrophenyl acetate) at 400 nm. Carboxynitrophenyl acetate was synthesized by acetylation of 4-hydroxy-3-nitrobenzoic acid [7]. The hydrolysis of two esters could be followed very sensitively by fluorometric estimation of 7-hydroxycoumarin [28] and 4-carboxy-7-hydroxycoumarin formed from umbelliferone acetate and 4-carboxyumbelliferone acetate (carboxyumbelliferone acetate), respectively. As the excitation and emission spectra of 7-hydroxycoumarin was only slightly modified by introduction of the carboxy group the same primary and secondary filters could be used for both assays [28]. Carboxyumbelliferone acetate [30], acetylation of 4-carboxyumbelliferone yielded umbelliferone acetate [27]. The amidase activity of the esterase was determined by measuring the formation of aniline as described [1].

All enzyme assays were performed at pH 7.4 in 0.01 M sodium phosphate buffer containing 0.25 M sucrose at a temperature of 20° C. To avoid uncertainties from incomplete ionisation of phenols or quenching of fluorescence a known amount of product was added for calibration to the cuvettes after each assay. Initial reaction velocities were determined using an external recorder.

For electron microscopic investigations microsomes were prepared from fasted rats by gel filtration in 0.01 M cacodylate buffer, pH 7.0, containing 0.14 M NaCl. One ml of microsomal

Location of microsomal esterases

suspension (3 mg of protein/ml) was spun down several times in an Eppendorf centrifuge model 3200 to form a pellet. The supernatant was removed and the pellet was fixed in 2,5 % glutaraldehyde in 0,03 M cacodylate buffer pH 7.0 for 60 min. at room temperature. After postfixation in 1% osmium tetroxide in the same buffer the pellets were dehydrated with ethanol and embedded in Epon 812. Thin sections were cut with a Reichert Ultramicrotome, stained with uranyl acetate and lead citrate in the conventional manner and examined with a Siemens Elmiskop 101. Thiophenol acetate, the substrate used for electron microscopic observation of esterase activity, was prepared from thiophenol and acetyl chloride [10]. Enzyme specific staining [29] was achieved by preincubation of the unfixed microsomal pellets in 5 mM gold sodium thiosulphate (Sanocrysin[®], a gift from Ferrosan, Copenhagen, Denmark) dissolved in 0.01 M cacodylate buffer pH 7/0.14 M NaCl for five minutes at the temperature of melting ice. After addition of 10 ul thiophenyl acetate (destilled just before use to remove traces of thiophenol from this unstable ester and diluted 1/1000 by sonification in incubation buffer), the pellets were shaked gently for ten minutes at 30° C, stopped with 2,5 % glutaraldehyde and processed as described above. For controls microsomes were incubated 15 min. before centrifugation with 1 mM bis(4-nitrophenyl)phosphate.

All reagents not specified in detail were the best grade available and were used without further purification.

Results

In order to find a method to disrupt microsomal membranes avoiding an inactivation of microsomal esterases we undertook preliminary experiments. Many treatments commonly used for this purpose including taurocholate, deoxycholate, acetone and alkali treatment, or sonication were accompanied by considerable inactivation of the enzyme activity. Among the compounds tested only Triton X 100 caused no inactivation of the microsomal esterase. Figure 1 shows the insensitivity of the enzyme to various concentrations of Triton X 100.

Since the detergent did not inactivate the microsomal esterase, it was expected, that the activity to the uncharged substrate (nitrophenyl acetate) was unaffected. Surprising, however, was the finding that even the hydrolytic activity against the charged substrate carboxynitrophenyl acetate was unchanged upon addition of different concentrations



Fig. 1. Effect of Triton X 100 on the esterase activity of microsomes. – Open circles: Hydrolytic activity against carboxynitrophenylacetate (10^{-4} M) . – Closed circles: Hydrolytic activity against nitrophenylacetate (10^{-4} M) .

of the detergent. To compare the influence of detergent upon the hydrolysis of uncharged and charged substrates by microsomal esterases the initial reaction velocities were determined at different substrate concentrations. Reciprocal plots of the data obtained are presented in Figures 2 and 3.



Fig. 2. Plot of reciprocal initial reaction velocity against reciprocal millimolar concentration of nitrophenylacetate in the absence (o) and in the presence of $0.1 \, ^{0}/_{0}$ (\bigcirc) or $0.5 \, ^{0}/_{0}$ (\bigtriangledown) Triton X 100.

Fig. 3. Plot of reciprocal initial reaction velocity against reciprocal millimolar concentration of carboxynitrophenylacetate in the absence (\odot) and in the presence (\bigcirc) of 0.1 % Triton X 100.



Fig. 4. Kinetics of umbelliferoneacetate hydrolysis in the presence (\bigcirc) and in the absence (o) of 0.1 % Triton X 100.

The kinetic constants are unchanged by Triton X 100 for charged and uncharged substrates. The affinity of the enzyme for a substrate does not seem to be changed greatly by the introduction of a negative charge. In contrast the apparent maximal velocity is more than doubled with nitrophenyl acetate compared to carboxynitrophenyl acetate. to measure esterase activity against charged and uncharged substrates fluorometrically. Preliminary experiments again showed no change in esterase activity using concentrations of Triton X 100 up to $1^{0/0}$. Lineweaver-Burk plots of the hydrolysis of umbelliferone acetate and carboxyumbelliferone acetate by microsomal esterases in the presence and in the absence of detergent are shown in Figures 4 and 5.

The affinity of the microsomal esterases for these two substrates also was nearly the same. The apparent maximal velocity was even 40 times higher with the uncharged substrate compared with the charged one. But again with this couple of substrates neither K_m nor V_{max} was changed by detergents indicating that the esterase activity of microsomes is not latent.

Negatively charged inhibitors also should have a limited access to the active site of an enzyme located inside a membrane barrier. We expected, that after disruption of the membrane, bis(4-nitrophenyl)phosphate, a well known esterase inhibitor, would be more powerful than in intact microsomes. However, as shown in Figure 6 the time course of inhibition of microsomal esterases was unchanged in Triton X 100 compared to the assays in absence of detergent.

It is known, that microsomal esterases from rat liver are heterogeneous. Some enzymes are able to split esters exclusively and others can also cleave amides [3, 4, 5]. Unfortunately we could not repeat the described experiments with charged and uncharged amides since we did not measure any hydrolytic activety against negatively charged amides. The experiment concerning the inhibition of amidase activity by a charged inhibitor presented in Figure 7 led to the conclusion that the major part of the acetanilide hydrolyzing esterases is also accessible and therefore located on the outer surface of the microsomes.

The experiments described with charged substrates and a charged inhibitor suggest that microsomal esterases are attached to the cytoplasmic side of the microsomal vesicles and are not excluded from the surrounding medium by a membrane which is impermeable for charged substances. This finding could be confirmed by enzyme specific staining of microsomal esterases using thiophenyl acetate as a substrate. In preliminary studies we were able to follow the splitting of this ester by microsomes spectrophoto-



Fig. 5. Plot of reciprocal initial reaction velocity against reciprocal millimolar concentration of substrate (carboxyumbelliferoneacetate). Closed circles represent assays in 0.1 % Triton X 100, open circles control assays without detergent.

metrically based on the production of a yellow color with 5,5'-dithiobis-2-nitrobenzoate [10]. This enzymatic hydrolysis could be completely inhibited by 1 mM bis(4-nitrophenyl)phosphate. As shown in Figures 8 a and b the reaction product (Au⁺-trapped thiophenol) appears outside the microsomal vesicles exclusively. In preparations of microsomes previously treated with the inhibitor nearly no deposits can be seen in the sections (Fig. 8 c).



Fig. 6. Inhibition of microsomal esterase (substrate 10^{-4} M p-nitrophenylacetate by bis(p-nitrophenyl)phosphate. Circles represent control values without detergent (o) and in the presence of 0.1% Triton X 100 (), *squares* represent inhibition of microsomal esterases (0.43 mg/ml) after supplementation of suspension with 1,31 μ mole inhibitor per ml in the absence () and in the presence () of 0.1% Triton X 100.

Fig. 7. Inhibition of microsomal acetanilide-hydrolyzing esterases (substrate 25 mM acetanilide) by bis(p-nitrophenyl)phosphate. *Circles* represent control values without detergent (o) and in the presence of $0.6 \ 0/0$ Triton X 100 (\odot), *squares* show inhibitor of microsomal esterases (0,36 mg/ml) after supplementation of the suspension with 0,3 μ mole inhibitor per ml in the absence (\Box) and in the presence (\blacksquare) of 0,6 0/0 Triton X 100.

Discussion

The experiments described were designed to determine the localization of esterases in the microsomal vesicles. The results obtained demonstrate that esterase activity is not latent in microsomes. However, latency of membrane bound enzymes in microsomes is considered to indicate the presence of a permeability barrier and a location of the latent enzyme inside the vesicle [2, 8, 13, 16, 31]. In agreement with our cytochemical demonstration of esterolytic enzymes it may be concluded therefore that microsomal esterases are attached to the cytoplasmic side of the microsomal vesicles.

Fig. 8 a. Thin section of a microsomal fraction previously treated with thiophenyl acetate, showing besides microsomes (M) the reaction product of the microsomal esterases, the Au⁺-trapped thiophenol, as electron dense particles (*arrows*). – 12 000 \times . – **b.** In higher magnification the reaction product is clearly visible outside the microsomes (M), partly contacting the microsomal membranes (*arrows*). – 50 000 \times . – **c.** Microsomal fraction after inhibition of the enzymatic hydrolysis of thiophenyl acetate by bis-p-nitrophenyl-phosphate. Nearly no esterase reaction products are visible. – 12 000 \times .



This interpretation is at variance with immunological experiments reported by AKAO and OMURA [1] showing that a precipitating antibody prepared to purified acetanilidehydrolyzing esterase did not interact with the enzyme associated with microsomal vesicles [1]. The resistance of microsomal acetanilide-hydrolyzing esterase to proteolysis also suggested that this enzyme is not exposed to the outside medium [1].

However, protease treatment may be no ideal tool for studying the sideness of esterases since soluble and membrane bound acetanilide-hydrolyzing esterase is insensitive to proteolytic attack [1]. Furthermore enzymatic activity against aromatic esters, which are split by all species of microsomal esterases [5], is neither inactivated nor solubilized during digestion of microsomes with trypsin in concentrations up to $0,1^{0}/_{0}^{2}$), although it is known from experiments described in this paper, that main esterase activity is localized on the outer surface of the microsomal vesicle.

The conflicting results obtained by AKAO and OMURA [1] and by us could be due to the heterogeneity of microsomal esterases. From rat liver five esterases could be differentiated during one purification procedure [3, 4, 5]. Other preparations of microsomal esterases from rat liver are described in the literature [1, 11, 12, 20] but the relations to the five mentioned above are not clear [5].

Two of these enzymes variants hydrolyze carboxyl esters only (designated E_1 and E_2), whereas the remaining three exhibit both esterase and amidase activities. The controverse conclusions drawn in this work compared to interpretations arising from immunological and digestion experiments [1] may be explained by different distribution of the esterase variants on both sides of the microsomal membrane. In this model, according to the experiments reported in this publication, the two esterases exhibiting hydrolyzing activity against carboxyl esters (E_1 , E_2) are located on the outer surface of the microsomal vesicles. One of the esterase variants exhibiting amidase activity seems to be linked to the inner surface of the vesicular membrane as supported mainly by the finding that an antibody to the esterase fails to react with the microsome-bound enzyme [1], while others, as concluded above from studied with a membrane impermeable inhibitor of amidase activity, together with NADH-cytochrome b5 reductase, NADPH-cytochrome c reductase, cytochrome b5 [14, 23, 24, 25] and the esterases E1 and E2 are facing the cytoplasmic side of the microsomal membrane.

Acknowledgements. The authors thank DR. V. ULLRICH for many helpful discussions and DR. T. OMURA for his valuable advice during preparation of this manuscript. Parts of this paper have been presented at the Joint Meeting 1974 of the Biochemical Societes of Belgium, the Federal Republic of Germany und the Netherlands. Düsseldorf 2nd to 5th Oktober 1974 [9]. – This work was supported by the Sonderforschungsbereich 38, "Membranforschung".

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