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# Interaction of peptides and small molecules with biological targets 

Coordinatore<br>Chiar.ma Prof.ssa<br>MARIA VALERIA D'AURIA

Supervisore
Chiar.mo Prof.
ALFONSO CAROTENUTO
Candidato
DIEGO BRANCACCIO

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#### Abstract

During my PhD course, I focused my attention on the conformational analysis of peptides that interact with GPCR and ligand-receptor interactions from NMR spectroscopy with possible application to cytotoxic agents binding to DNA.

In particular, I investigated the conformational behaviour of peptide analogues of Urotensin, Melanocortin and Somatostatin in water solution and membrane mimetic environment (SDS and DPC micelles).

Another research field was the NMR-based screening as potent technique for the identification of small molecules that interact with macromolecule targets. Several methods based on the ligand observation have been proposed in the literature, among these WaterLOGSY (water-ligand observed via gradient spectroscopy), and STD-NMR (saturation transfer difference) experiments. I applied these NMR techniques for the evaluation of the DNA interactions of a new series of thiophen-naphthoquinones with interesting cytotoxic activity.


[^0]$h$-UTR, human Urotensin II receptor;
IL, intracellular loop;
LC, liquid chromatography;
MD, molecular dynamic;
MSH, melanocyte stimulating hormones;
MW, molecular weight;
Nal, Naphtylalanine;
NMM, N-methyl morpholine;
NMR, nuclear magnetic resonance;
NOE, nuclear Overhauser effect;
NOESY, nuclear Overhauser enhancement spectroscopy;
Orn, Ornitine;
PE COSY, primitive exclusive correlated spectroscopy;
Pen, penicillamine;
POMC, proopiomelanocortin;
RCM, ring closing metathesis;
RMSD, root mean square deviation;
ROESY, rotating-frame Overhauser effect spectroscopy;
RP-HPLC, reverse phase-high performance liquid chromatography; SPE, Solid
Phase Extraction;
$r$-UTR, rat Urotensin II receptor;
SAR, Structure activity relationship;
SD, standard deviation;
SDS, sodium dodecylsulphate;
SPPS, solid phase peptide synthesis;
SRIF, somatostatin; sst, somatostatin receptor;
STD, saturation transfer difference;
TES, triethylsilane;
TFA, trifluoroacetic acid;
THF, tetrahydrofuran;
Tic, tetrahydroisoquinoline;
TLC, thin-layer chromatography;
TM, trans-membrane domain;
TOCSY, total correlated spectroscopy;
TSP, 3-(trimethylsilanyl)propionic acid;
U-II, Urotensin-II peptide;
WaterLOGSY, water-ligand observed via gradient spectroscopy.

# Chapter 1 - CONFORMATIONAL ANALYSIS OF PEPTIDES THAT INTERACT WITH GPCR 

### 1.1 New Insight into the Binding Mode of Peptide Ligands at Urotensin-II Receptor: Structure-Activity Relationships Study on P5U and Urantide

Urotensin II (U-II) is a disulfide bridged peptide-hormone identified as the ligand of a G protein-coupled receptor. Human U-II (H-Glu-Thr-Pro-Asp-c[Cys-Phe-Trp-Lys-Tyr-Cys]-Val-OH) has been described as the most potent vasoconstrictor compound identified to date.

We have recently identified both a superagonist of $h \mathrm{U}$-II termed P5U (H-Asp-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH) and the compound termed Urantide (H-Asp-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH), which is the most potent UT receptor peptide antagonist described to date.

In the present study, we have synthesized several analogues of P5U and Urantide in which the $\mathrm{Asp}^{4}$ residue in N -terminus position was replaced with coded and non-coded amino acids. The replacement of the $\mathrm{Asp}^{4}$ residue by Tic led to an analogue, compound $\mathbf{1 4}$, more potent as antagonist $\left(\mathrm{pK}_{\mathrm{B}}=8.94\right)$ compared to Urantide. Furthermore, a different SAR was observed for the P5U compared to the Urantide analogues. NMR and docking studies revealed a different binding
mode for the agonist and antagonist ligands which could explain the observed SAR.

### 1.1.1. Introduction

Urotensin-II (U-II) is a cyclic peptide originally isolated from goby fish urophysis.[1] Subsequently, it has been found that U-II is also present in tetrapods and that its gene is expressed in the CNS.[2] The U-II precursor has now been cloned in various vertebrate species including frog, rat and mouse, pig, monkey, and human.[3-6] U-II was identified as the natural ligand of an orphan G-protein-coupled receptor [7] now referred to as UT receptor.

Recently, an analogue of U-II, called urotensin-related peptide (URP), has been identified in mammals.[8] In all U-II and URP isoforms known so far, the sequence of the cyclic C-terminal hexapeptide has been fully conserved across species.[9] The U-II and URP genes are primarily expressed in motoneurons located in discrete brainstem nuclei and in the ventral horn of the spinal cord.[10-13] UII and URP mRNAs have also been detected, although at a much lower level, in various peripheral tissues including the pituitary, heart, spleen, thymus, pancreas, kidney, small intestine, adrenal, and prostate.[3, 8, 14]

The U-II/UT receptor system seems to play an important role in cardiovascular functions; in fact, $h \mathrm{U}-\mathrm{II}$ has been shown to be 1-2 orders of magnitude more potent than endothelin-1 in producing vasoconstriction in mammals and thus is one of the most effective
vasoconstrictor compounds identified to date.[7, 15, 16] On the basis of its spectrum of activities, $h \mathrm{U}-\mathrm{II}$ has been postulated to contribute as modulator to cardiovascular homeostasis and possibly to be involved in certain cardiovascular pathologies.[15, 17] It has been recently demonstrated that U-II is involved in inhibition of insulin release [18] in the perfused rat pancreas and may play an important role in pulmonary hypertension.[19] Central nervous effects of U-II have also been described.[20] Hence, the $h$ U-II ligands could be of therapeutic value in a number of pathological disorders. It has been demonstrated that the C-terminal octapeptide of U-II retains full biological activity and binding properties.[21-26]

The (patho)physiological role(s) of the U-II/UT receptor system and, most importantly, the potential interest of UT receptor ligands as drug candidates prompted the development of low molecular weight compounds as non peptide UT receptor agonists and antagonists (Figure 1).[27]
a)


b)



Palosuran

Figure 1. Some representative structures of non-peptide UTR agonists (a), and antagonists (b).

Our research group has been involved for a long time in the development of UTR peptide ligands. The optimization of a peptide as a lead structure is important to improve its pharmacokinetic properties, and in identifying the pharmacophore elements, that is, to determine the key amino acid residues that are involved in the
biological activity.[28] Interestingly, some common features are observable (two aryl moieties and a protonable nitrogen atom) in organic and peptide UTR ligands.[29] Hence, the structural information obtained by the peptide investigation might be useful for the design of both small-molecules and peptide ligands.

In previous studies, we have identified both a superagonist named P5U (H-Asp-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH) [30] and an antagonist, Urantide (H-Asp-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-ValOH ) [31] of $h \mathrm{U}-\mathrm{II}$. The latter is the most potent peptide antagonist at UT receptor described to date. Actually, Urantide behaves as a pure antagonist in the rat aorta bioassay,[31] and as a full agonist in a calcium mobilization assay performed in CHO cells expressing the $h$ UTR.[32] This point has been widely discussed elsewhere.[33] For sake of simplicity, we will refer to Urantide as an antagonist throughout the manuscript.

Recently, we performed extensive NMR and computational studies on both P5U and Urantide that allowed us to formulate a hypothesis about the structural changes that determine the switching from agonist to antagonist activity.[33, 34]

To aim to identify new leads for the development of both agonists and antagonists at UT receptor, we have studied the structure-activity relationships of a series of novel P5U and Urantide analogues based
on the chemical substitution of the Asp ${ }^{4}$ residue, with several other amino acid residues with different physicochemical properties (Figure 2 and Figure S1). The most interesting analogues were then analysed by NMR and their structures fitted within $h$-UTR models to gain insight into the agonist and antagonist binding modes.

### 1.1.2 Results

Chemistry. Peptides were synthesized according to the solid phase approach using standard Fmoc methodology in a manual reaction vessel (Experimental Section). [35]

The purification was achieved using a semi-preparative RP-HPLC C18 bonded silica column (Vydac 218TP1010). The purified peptide was $98 \%$ pure as determined by analytical RP-HPLC. The correct molecular weight of the peptide was confirmed by mass spectrometry and amino acid analysis (Supporting Information).


Urantide


R A Ala, Phe, Cpa, Nal, $\left(\mathrm{pNO}_{2}\right)$ Phe, Tic, Lys

Figure 2. New synthesized compounds

Biological Data. Receptor affinity at $h$-UTR and biological activity (rat aorta bioassay) of the synthesized compounds are reported
in Table 1. Substitution of the native Asp ${ }^{4}$ residue in P5U by an Ala residue (compound 1), which generated an URP analogue, slightly reduced the contractile potency of the peptide $\left(\mathrm{pEC}_{50}=8.04\right)$. Similar modification in Urantide sequence produced compound 2 with antagonist activity but slightly less potent than Urantide $\left(\mathrm{pK}_{\mathrm{B}} 7.84\right)$. Subsequently, to evaluate the role of an aromatic residue in position 4 we replaced $\mathrm{Asp}^{4}$ with a Phe residue in both sequence of P5U and Urantide. Compound 3 showed to be a superagonist as P5U $\left(\mathrm{pEC}_{50}=9.18\right)$ while the same substitution in Urantide sequence generated compound 4 with a reduced binding affinity but with an increased antagonist activity $\left(\mathrm{pK}_{\mathrm{i}} 7.71\right.$ and $\left.\mathrm{pK}_{\mathrm{B}} 8.68\right)$. Then, the $\mathrm{Asp}^{4}$ residue was replaced with some uncoded aromatic amino acids (Figure 2). Compound 5, in which $\mathrm{Asp}^{4}$ was replaced with a Cpa residue resulted to be less potent as agonist compared to P5U (pEC ${ }_{50}$ 8.86). Similar trend was observed in compound 6 with a reduced antagonist potency $\left(\mathrm{pK}_{\mathrm{B}} 7.85\right)$. Analogue 7 , containing in position 4 a $\mathrm{Nal}(1)$ residue, showed a sensible reduction both in binding $\left(\mathrm{pK}_{\mathrm{i}}\right.$ 7.58) and functional activity $\left(\mathrm{pEC}_{50} 6.99\right)$, while the same substitution in Urantide sequence (compound 8) resulted in a conserved antagonist activity $\left(\mathrm{pK}_{\mathrm{B}}\right.$ 8.50). Interestingly, $\mathrm{Nal}(2)$ derivative of P5U (compound 9) regained high agonist activity $\left(\mathrm{pEC}_{50} 8.28\right)$. On the
other hand, compound $\mathbf{1 0}$ resulted to be slightly less potent compared to compound $\mathbf{8}$ and Urantide ( $\mathrm{pK} \mathrm{K}_{\mathrm{B}} 7.89$ ).

Table 1. Receptor Affinity and Biological Activity of P5U and Urantide Analogues of General Formula:

## R-c[Pen ${ }^{a}$-Phe-Xaa-Yaa-Tyr-Cys]-Val-OH

| Peptide | Xaa | Yaa | R | $\mathrm{pK}_{\mathrm{i}}{ }^{\text {b }}$ | $\mathrm{pEC}_{50}{ }^{\mathrm{c}}$ | $\mathrm{pK}_{\mathrm{B}}{ }^{\text {d }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $h \mathrm{U}$-II | Trp | Lys | * | $9.10 \pm 0.08$ | $8.30 \pm 0.06$ | - |
| $\begin{aligned} & h \mathrm{U}-\mathrm{II}(4- \\ & 11) \end{aligned}$ | Trp | Lys | Asp | $9.60 \pm 0.07$ | $8.60 \pm 0.04$ | - |
| P5U | Trp | Lys | Asp | $9.70 \pm 0.07$ | $9.60 \pm 0.07$ | - |
| Urantide | DTrp | Orn | Asp | $8.30 \pm 0.04$ | Inactive | 8.30 |
| 1 | Trp | Lys | Ala | $9.10 \pm 0.08$ | $8.04 \pm 002$ | - |
| 2 | DTrp | Orn | Ala | $8.78 \pm 0.08$ | - | 7.84 |
| 3 | Trp | Lys | Phe | $9.55 \pm 0.05$ | $9.18 \pm 0.17$ | - |
| 4 | DTrp | Orn | Phe | $7.71 \pm 0.10$ | - | 8.68 |
| 5 | Trp | Lys | Cpa | $9.05 \pm 0.04$ | $8.86 \pm 0.05$ | - |
| 6 | DTrp | Orn | Cpa | $8.02 \pm 0.06$ | - | 7.85 |
| 7 | Trp | Lys | $\mathrm{Nal}(1)$ | $7.58 \pm 0.06$ | $6.99 \pm 0.13$ | - |
| 8 | DTrp | Orn | Nal(1) | $8.41 \pm 0.01$ | - | 8.50 |
| 9 | Trp | Lys | $\mathrm{Nal}(2)$ | $8.19 \pm 0.10$ | $8.28 \pm 0.10$ | - |
| 10 | DTrp | Orn | Nal(2) | $7.93 \pm 0.01$ | - | 7.89 |
| 11 | Trp | Lys | $\left(\mathrm{pNO}_{2}\right)$ Phe | $7.87 \pm 0.08$ | $7.14 \pm 0.09$ | - |
| 12 | DTrp | Orn | $\left(\mathrm{pNO}_{2}\right)$ Phe | $7.80 \pm 0.10$ | - | 7.90 |
| 13 | Trp | Lys | Tic | $8.58 \pm 0.03$ | $8.87 \pm 0.18$ | - |
| 14 | DTrp | Orn | Tic | $8.03 \pm 0.07$ | - | 8.94 |
| 15 | Trp | Lys | Lys | $8.03 \pm 0.11$ | $8.22 \pm 0.24$ | - |
| 16 | DTrp | Orn | Lys | $6.66 \pm 0.01$ | - | 7.49 |
| ${ }^{\text {a }}$ Cys in $h \mathrm{U}-\mathrm{II}$ and $h \mathrm{U}-\mathrm{II}(4-11) .{ }^{\mathrm{b}} \mathrm{pKi}:-\log \mathrm{Ki}^{\mathrm{c}} \mathrm{pEC}_{50}:-\log \mathrm{EC}_{50} .{ }^{\mathrm{d}} \mathrm{pK}_{\mathrm{B}}\left(-\log \mathrm{K}_{\mathrm{B}}\right)$ values are from experiments in the rat thoracic aorta. Each value in the table is mean $\pm$ s.e.m. of at least 4 determinations. * H-Glu-Thr-Pro-Asp- |  |  |  |  |  |  |

Replacing the $\mathrm{Asp}^{4}$ residue with the amino acid $\mathrm{pNO}_{2} \mathrm{Phe}$ in both parent peptides, led to compounds with reduction in activity. In fact, compound 11 resulted to have a reduced binding affinity at UT receptor (pKi 7.87) and a more considerable reduction in functional activity $\left(\mathrm{pEC}_{50} 7.14\right)$. Compound $\mathbf{1 2}$, resulted to be slightly less potent respect to Urantide showing a $\mathrm{pK}_{\mathrm{B}}$ of 7.90 . Analogue 13, in which Asp ${ }^{4}$ residue was replaced with a Tic residue, showed a slightly reduced activity ( $\mathrm{pEC}_{50}$ 8.87). Surprisingly, the same substitution in Urantide sequence produced analogue 14 with increased antagonist potency showing a $\mathrm{p} \mathrm{K}_{\mathrm{B}}$ value of 8.94 . This compound represents a new potent antagonist discovered by this study. Finally, the replacement of Asp ${ }^{4}$ with a Lys residue in P5U (analogue 15) resulted in a reduced activity $\left(\mathrm{pEC}_{50} 8.22\right)$. Worthy of note, the same modification in Urantide sequence produced an analogue (compound 16) showing a dramatic reduction in binding affinity and antagonist activity ( $\mathrm{pK}_{\mathrm{i}} \quad 6.66$ and $\mathrm{p} \mathrm{K}_{\mathrm{B}}$ 7.49), being by far the weakest ligand among the synthesized compounds.

NMR Analysis. A whole set of 1D and 2D NMR spectra in 200 mM aqueous solution of SDS were collected for compounds $\mathbf{1 4}$, and 16. These peptides were chosen since $\mathbf{1 4}$ is the most potent antagonist of the series while $\mathbf{1 6}$ has very low binding affinity and antagonist potency (Table 1). Micelle solution was employed since we have
recently reported the NMR structure of UT agonists (among which P5U) [34] and antagonist (among which Urantide) [33] in this medium. Complete ${ }^{1} \mathrm{H}$ NMR chemical shift assignments were effectively achieved for the two peptides according to the Wüthrich procedure[36] via the usual systematic application of DQF-COSY,[37, 38] TOCSY,[39] and NOESY [40] experiments with the support of the XEASY software package (Supporting Information).[41] Peptides 14 and 16 differs from Urantide only for the N -terminal residue substitution and show diagnostic NMR parameters $\left(\mathrm{H}_{\alpha}\right.$ proton chemical shifts, NOE contacts, ${ }^{3} J_{\mathrm{NH}-\mathrm{H} \alpha}$ and ${ }^{3} J_{\mathrm{H} \alpha-\mathrm{H} \beta}$ coupling constants, NH exchange rates and temperature coefficients) all similar to those observed in the parent peptide (Supporting Information). In particular, NOE contacts between $\mathrm{H}_{\alpha}-\mathrm{NH}_{i+2}$ of $\mathrm{D}-\mathrm{Trp}^{7}$ and $\mathrm{Tyr}^{9}$ and between NH$\mathrm{NH}_{i+1}$ of $\mathrm{Orn}^{8}$ and $\mathrm{Tyr}^{9}$ indicated the presence of a $\beta$-turn. This result was supported by the observation of slowly exchanging NH resonance of residue 9 , and low value of the temperature coefficient for this proton $(-\Delta \delta / \Delta T<3.0 \mathrm{ppb} / \mathrm{K})$. A short stretch of antiparallel $\beta$-sheet involving residues 5-6 and 10-11 is inferred from a number of longrange NOEs including $\mathrm{H}_{\alpha}-\mathrm{NH}$ connectivities between residues 5, 11 and 10, 6 and a NH-NH connectivity between residues 6 and 9. All the
data indicated the preservation, in 14 and 16 , of the $\beta$-hairpin structure.


Figure 3. Superposition of the 10 lowest energy conformers of $\mathbf{1 4}$ (a), $\mathbf{1 6}$ (b). Structures were superimposed using the backbone heavy atoms of residues 5-10. Heavy atoms are shown with different colours (carbon, green; nitrogen, blue; oxygen, red; sulfur, yellow). Hydrogen atoms are not shown for clarity.

NMR-derived constraints obtained for the analyzed peptides (Supporting Information) were used as the input data for a simulated annealing structure calculation. For each peptide, 20 calculated structures satisfying the NMR-derived constraints (violations smaller than $0.40 \AA$ ) were chosen (Figure $3 \mathrm{a}-\mathrm{b}$ ). As shown, both the peptides 14, and 16 show a well defined type II' $\beta$-hairpin structure encompassing residue $5-10$ (backbone rmsd values are 0.41 and 0.37 $\AA$, respectively). In contrast, the N - and C-terminal residues were
highly flexible. Considering the side chains orientation, $\mathrm{Phe}^{6}, \mathrm{Orn}^{8}$, and $\mathrm{Tyr}^{9}$ side chains showed a large preference for trans, $g^{-}$, and $g^{-}$ rotamers, respectively, while $\mathrm{D}-\mathrm{Trp}^{7}$ side chain is found both in trans and $g^{+}$conformation.

Docking Studies of Urantide and its Analogues. The theoretical structure of the $h$-UT receptor (Figure 4) was generated by homology modeling based on the crystal structure of bovine rhodopsin (PDB code 1F88),[42] as described previously.[43] The resulting structure represents an inactive form of the $h$-UT receptor $\left(\mathrm{h}-\mathrm{UTR}_{\mathrm{i}}\right)$ with an overall conformation very similar to that of bovine rhodopsin (1.22 $\AA$ rmsd between the backbone atoms of the transmembrane domains).


Figure 4. Serpentine model of the $h$-UTR sequence. The black lines represent the boundaries of the membrane. Filled circles indicate the residues highly conserved among the GPCRs superfamily. The TM helices are denoted by roman numerals. The arabic numbers indicate the position of the residues inside the TM domain. The glycosilation sites on the N -terminal region are also shown.

Since the currently available docking programs may not work very well for peptide compounds, manual docking was conducted for Urantide. The NMR-derived Urantide structure [33] was placed in between the trans-membrane domains of the $h-\mathrm{UTR}_{\mathrm{i}}$, employing the following criteria to achieve meaningful docking modes: (i) The positively charged amino group of $\mathrm{Orn}^{8}$ had to be close to and pointing in the direction of the carboxylate group of Asp130, which is conserved in many GPCRs and positioned in the TM-III region; (ii) N-terminal residues should point towards extracellular loops as experimentally determined [44]; (iii) No steric clashes should occur between any atom. To assess the stability of the Urantide $/ h-\mathrm{UTR}_{\mathrm{i}}$ complex and to analyze the potential ligand/receptor interactions, energy minimization and MD simulations of 2 ns at a constant temperature of 300 K were run. During the MD simulation, the ligand, the EL's, and all the receptor side chains were allowed to relax, while the TM's and IL's backbone atoms were held frozen. The distances between the peptide and the key receptor residues were monitored along the complete 2 ns MD trajectory (Supporting Information).

To inspect the variations in the ligand conformation, rmsd with respect to the starting structure was calculated. Interestingly, the rmsd of Urantide backbone atoms turned out to be remarkably stable
throughout all the MD simulations $(0<\operatorname{rmsd}<0.6)$, indicating that the peptide settles into the receptor-binding site in a stable $\beta$-hairpin conformation. Also the side chain orientations are those described by NMR. Interestingly, $\mathrm{D}-\mathrm{Trp}^{7}$ prefers a trans orientation about $\chi_{1}$ angle $\left(\chi_{1} \approx 180^{\circ}, \chi_{2} \approx-70^{\circ}\right)$. As shown in Figure 5 a , the hypothetical binding site of Urantide is located among TM-III $\div$ TM-VII, and EL-II. The $\beta$-hairpin is oriented along the receptor helical axis, with the N and C-terminal residues pointing towards the extracellular side. The binding mode of the peptide is determined mainly by the interactions showed in Figure 5b and Table 2.

Table 2. Urantide $/ h-\mathrm{UTR}_{\mathrm{i}}$ Interactions

| Residue $^{*}$ | Surrounding residue |
| :--- | :--- |
| Asp $^{4}$ | Ala187 (EL-II), Met188 (EL-II), Cys199 (EL-II), Arg206 <br> (EL-II), Ala207 (EL-II) |
| Pen $^{5}$ | Gln278 (TM-VI), Pro287 (EL-III) |
| Phe $^{6}$ | Cys123 (EL-I), Val184 (TM-IV), Met188 (EL-II) <br> D-Trp $^{7}$ |
|  | Phe131 (TM-III), Met134 (TM-III), His135 (TM-III), <br>  <br>  <br>  <br>  <br> Leu212 (TM-V), Leu215 (TM-V), Phe216 (TM-V), Ile220 <br> (TM-V), Trp275 (TM-VI), Gln278 (TM-VI) |
| Orn $^{8}$ | Asp130 (TM-III), Thr301 (TM-VII), Thr304 (TM-VII) |
| Tyr $^{9}$ | Phe127 (TM-III), Phe274 (TM-VI), Asn297 (TM-VII), <br>  <br> Cys $^{10}$ <br> Val $^{11}$ |
|  | Thr301 (TM-VII) |

* For sake of clarity, the residue numbers of the ligands are reported as apex while those of the receptor are not.

In particular, (i) a tight charge-reinforced hydrogen-bonding network involving the carboxylate group of Asp130 and the protonated $\delta$-amino group of $\mathrm{Orn}^{8}$ of Urantide is established. Such an interaction, which we assume to be an anchoring point of the ligand to $h$-UTR, remained stable during the whole production run (Supporting Information, Figure S2). (ii) Three hydrophobic pockets, delimited by residues listed in Table 2, host the aromatic side chains of $\mathrm{Phe}^{6}$, D$\operatorname{Trp}^{7}$, and Tyr ${ }^{9}$ of Urantide. Particularly, the indole system of D-Trp ${ }^{7}$ appears to be optimally oriented for a $\pi$-stacking interaction with the aromatic indole system of Trp275. Furthermore, the phenolic OH of $\mathrm{Tyr}^{9}$ is at hydrogen-bonding distance with the side chain CO of Asn297, and OH of Thr301. (iii) $\mathrm{Asp}^{4}$ in Urantide is involved in a hydrogen-bonding network. Particularly, the oxygen atoms of the carboxylate form two charge-reinforced hydrogen bonds with Arg206 guanidinium group. In addition, the protonated N-terminal group of Asp ${ }^{4}$ engages additional hydrogen bonds with the backbone CO of Ala187, Cys 199 and Met188. (iv) Finally, the negatively charged Cterminal group establishes two hydrogen bonds with backbone HN of Cys123 and Cys 199, and a salt bridge with the protonated guanidinium moiety of Arg189 (EL-II). All the aforementioned interactions resulted to be quite stable during the whole MD production run (see Figure S2-S11 for details). The mean structure of
the last 1 ns of MD was extensively minimized and used for subsequent analysis.

Since the NMR results indicate that the 3D structure of the Urantide analogues $\mathbf{1 4}$ and $\mathbf{1 6}$ did not change after the replacement of the N terminal residue, we used the energy-minimized structure of the Urantide $/ h-\mathrm{UTR}_{\mathrm{i}}$ complex as starting point for the docking procedure of these derivatives. After replacing Asp ${ }^{4}$ of Urantide with $\mathrm{Tic}^{4}$ to give compound 14 and with $\mathrm{Lys}^{4}$ to give 16, the complexes were minimized and then subjected to a 200 ps MD simulation. The mean structures of the last 100 ps of the MD trajectory were then minimized and used for subsequent analysis.



Figure 5. (a) Stereoview of $h-\mathrm{UTR}_{\mathrm{i}}$ model complexed with Urantide. Urantide heavy atoms are colour coded as in Figure 2. Receptor backbones are represented in azure and labeled. (b) Stereoview of Urantide within the binding pocket of $h$ $\mathrm{UTR}_{\mathrm{i}}$. Hydrogen bonds are represented with dashed lines.

While the same interactions with $h-\mathrm{UTR}_{\mathrm{i}}$ were recorded for the unchanged residues, in the $14 / \mathrm{UTR}_{\mathrm{i}}$ complex, $\mathrm{Tic}^{4}$ interacts with Val184 (TM-IV), Ala187 (EL-II), Leu200 (EL-II), Pro201 (EL-II), and Tyr211 (TM-V); while in the $\mathbf{1 6} / \mathrm{UTR}$ complex Lys ${ }^{4}$ residue takes
contact with Leu200 (EL-II), and Tyr211 (TM-V). In Table 3 ligand/receptor $\Delta \mathrm{G}_{b i n d}$ values are reported as calculated employing the AutoDock4 program native scoring function.[45-47] Interestingly, there is a clear, although qualitative, correlation between the predicted $\Delta \mathrm{G}_{\text {bind }}$ values and the experimental binding constants (Table 1).

Docking of P5U and its Analogues. The three-dimensional model of the $h$-UTR, in the active state $\left(h-\mathrm{UTR}_{\mathrm{a}}\right)$, was constructed from the model structure of the bovine rhodopsin, proposed by Mosberg,[48] and was generated by homology modeling following the same steps described for the inactive model.[43]

Table 3. Binding free energies $\left(\Delta \mathrm{G}_{\mathrm{AD}}\right)$ calculated for the energy minimized averaged complexes deriving from the MD simulations.

| Receptor $^{\boldsymbol{r}}$ | Ligand | $\Delta \mathbf{G}_{\text {bind }}{ }^{\boldsymbol{a}}$ | Electr $^{\boldsymbol{b}}$ | H-Bond $^{\boldsymbol{b}}$ | VdW $^{\boldsymbol{b}}$ | Desolv $^{\boldsymbol{b}}$ | Tors $^{\boldsymbol{b}}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $h-$ UTR $_{\mathrm{i}}{ }^{c}$ | Urantide | -24.33 | -4.99 | -5.90 | -26.50 | 7.09 | 5.97 |
| $h-$ UTR $_{\mathrm{i}}$ | $\mathbf{1 4}$ | -23.01 | -3.21 | -3.83 | -26.98 | 5.94 | 5.07 |
| $h-$ UTR $_{\mathrm{i}}$ | $\mathbf{1 6}$ | -21.10 | -3.31 | -5.77 | -25.16 | 6.28 | 6.86 |
| $h-$ PTR $_{\mathrm{a}}$ | P5U | -24.53 | -4.99 | -6.11 | -25.89 | 6.69 | 5.76 |
| $h-$ UTR $_{\mathrm{a}}$ | $\mathbf{1 3}$ | -23.53 | -3.35 | -4.19 | -27.40 | 6.03 | 5.37 |
| $h-$ UTR $_{\mathrm{a}}$ | $\mathbf{1 5}$ | -23.01 | -4.11 | -6.77 | -25.40 | 6.31 | 6.96 |
| $h-$ UTR $_{\mathrm{a}}$ | Urantide | -20.65 | -5.92 | -6.42 | -21.39 | 7.11 | 5.97 |
| $h-$ UTR $_{\mathrm{i}}$ | P5U | -18.68 | -3.60 | -3.47 | -24.67 | 6.80 | 6.26 |

${ }^{a} \Delta \mathrm{G}_{\text {bind }}$ : free energy of binding. ${ }^{b}$ Energy terms contributing to the AutoDock4 scoring function. Electr: electrostatic; H-Bond: H-Bonding; VdW: Van der Waals; Desolv: desolvation; Tors: torsional entropy. All terms are given in $\mathrm{kcal} / \mathrm{mol} .{ }^{c} h$ $\mathrm{UTR}_{\mathrm{i}}$ : receptor in the inactive state. $h-\mathrm{UTR}_{\mathrm{a}}$ : receptor in the active state.

A comparison of models for the active and inactive states of $h$ UTR reveals the structural changes that accompany activation.

Overall, the rmsd between these models is $2.3 \AA$ calculated for the backbone atoms of all the TM's, but decreases to $1.7 \AA$ after excluding TM-VI, which experiences a rearrangements upon receptor activation. Indeed, TM-VI shifts outward and rotates counterclockwise (viewed from the extracellular side) during activation, moving its intracellular end away from TM-III and toward TM-V. As a result of this and other changes, the receptor structure tightens near its extracellular surface but opens up at the cytoplasmic side, providing a cavity for binding of the Gas subunit.

The NMR-derived P5U structure [34] was placed in between the trans-membrane domains of the $h-\mathrm{UTR}_{\mathrm{a}}$ model, following the same criteria used for Urantide (see above) to achieve meaningful binding poses. Energy minimization and MD simulations ( 2 ns ) were run to assess the stability of the $\mathrm{P} 5 \mathrm{U} / h-\mathrm{UTR}_{\mathrm{a}}$ complex and to analyze the potential ligand/receptor interactions.

To inspect the variations in the ligand conformation, rmsd with respect to the starting structure was calculated. Interestingly, the rmsd of P5U backbone atoms turned out to be really stable throughout all the MD simulations $(0<\operatorname{rmsd}<0.5)$, indicating that the peptide settles into the receptor-binding site in a stable $\beta$-hairpin conformation. Also the side chain orientations are those described by NMR.[34]

As shown in Figure 6a, the hypothetical binding site of P5U is located among TM-III $\div \mathrm{TM}-\mathrm{VII}$, EL-II and EL-III. The $\beta$-hairpin is oriented along the receptor helical axis, with the N - and C-terminal residues pointing towards the extracellular side. The binding mode of P5U is determined mainly by the interactions showed in Figure 6b and Table 4.

Table 4. $\mathrm{P} 5 \mathrm{U} / h$ - $\mathrm{UTR}_{\mathrm{a}}$ Interactions

| Residue | Surrounding residue |
| :--- | :--- |
| Asp $^{4}$ | Pro201 (EL-II), Gln285 (EL-III) |
| Pen $^{5}$ | His208 (EL-II), Trp277 (TM-VI), Ala281 (TM-VI), Ala286 <br> (EL-III) |
| Phe $^{6}$ | Val184 (TM-IV), Met188 (EL-II), Leu212 (TM-V) |
| Trp $^{7}$ | Phe131 (TM-III), Met134 (TM-III), Phe274 (TM-VI), <br>  <br> Trp275 (TM-VI), Gln278 (TM-VI) $^{8}$ <br> Tyr $^{9}$ <br>  <br>  <br> Asp130 (TM-III), Tyr305 (TM-VII) <br> Trp116(TM-II), Cys123 (EL-I), Leu126 (TM-III), Phe127 <br> (TM-III), Cys199 (EL-II) <br> Val $^{11}$ |

As for Urantide, a stable (Figure S12) charge-reinforced hydrogen-bonding network involved the carboxylate group of Asp130 and the protonated $\varepsilon$-amino group of Lys ${ }^{8}$ of P5U is observed. Three hydrophobic pockets, delimited by residues listed in Table 4, host the aromatic side chains of $\mathrm{Phe}^{6}, \mathrm{Trp}^{7}$, and $\mathrm{Tyr}^{9}$. These hydrophobic pockets only partially overlap with those of Urantide. For instance, $\mathrm{Tyr}^{9} \mathrm{OH}$ group is not engaged in any hydrogen bond. Again, the
negatively charged C-terminal group of $\mathrm{Val}^{11}$ establishes a hydrogen bond with Cys 199 backbone NH , and a salt bridge with the protonated guanidinium moiety of $\operatorname{Arg} 189$.

Differently from Urantide, $\mathrm{Asp}^{4}$ in P5U is involved in a hydrogen-bond with the Gln 285 (EL-III) $\mathrm{NH}_{2}$ group. This H -bond is not stable during the MD trajectory (Figure S13). The mean structure of the last 1 ns of MD was extensively minimized and used for subsequent analysis.

Replacing the $\mathrm{Asp}^{4}$ residue of P5U with Tic or Lys residue (obtaining the derivatives $\mathbf{1 3}$ and 15, respectively) in the P5U/h-UTR model complex, and following the same optimization steps used for the complexes of Urantide analogues (see above), we obtained the two models: $13 / h-\mathrm{UTR}_{\mathrm{a}}$ and $\mathbf{1 5} / h-\mathrm{UTR}_{\mathrm{a}}$, showing similar binding energy (Table 3) in accordance with the experimental binding data (Table 1).

Switching the ligands. To assess the predictive value of the receptor models the ligands were switched, i.e. urantide was docked within $h$-UTR $_{\mathrm{a}}$ and P5U within $h$ - $\mathrm{UTR}_{\mathrm{i}}$ model (Figure S14). For the docking of urantide, we started from the optimized P5U/UTR ${ }_{a}$ complex and superposed the NMR derived urantide structure with that of P5U (backbone atoms of residues 5-10). Then, we removed the P5U structure and optimized the urantide/UTR ${ }_{a}$ complex. Analogous
steps were taken for the $\mathrm{P} 5 \mathrm{U} / \mathrm{UTR}_{\mathrm{i}}$ complex. In Table 3, the binding energies of the two complexes are reported.


Figure 6. (a) Stereoview of h-UTRa model complexed with P5U. P5U heavy atoms are colour coded as in Figure 2. Receptor backbones are represented in azure and labeled. (b) Stereoview of P5U within the binding pocket of h-UTRa. Hydrogen bonds are represented with dashed lines.

### 1.1.3. Discussion

Previous studies have demonstrated that the C-terminal octapeptide of $h \mathrm{U}-\mathrm{II}[\mathrm{U}-\mathrm{II}(4-11)$, Table 1] mimicked the effects of U II on intracellular calcium concentration in UTR-transfected cells and contraction of rat aortic rings.[22, 24, 25] Recently, Coy et coll. have examined the role of the N -terminal Asp residue in UII(4-11) since this acidic amino acid embodies one of the main structural differences between the UII(4-11) and somatostatin octapeptides which results in little somatostatin affinity for the UT receptor.[26] They found that the N -terminal amino acid does not require a negatively charged side chain, merely one which has a hydrogen bond acceptor CO group. The side chain can be constrained into a trans-olefinic configuration and can also contain an aromatic ring substituted with polar groups such as OH and $\mathrm{NO}_{2}$. Afterwards, Salvadori et coll. examined the same position of $h \mathrm{U}-\mathrm{II}(4-11)$ using a number of aromatic residues.[49] They found that all of the new analogues behaved as full agonists, and that aromaticity is well tolerated; size, length and chirality of the side chain are not important, while substituents with a nitrogen atom are preferred. On bases of these considerations and to further investigate the contribution of the N-terminal Asp residue in the biological activity, we synthesized 16 analogues of P5U and Urantide substituted at this position with amino acids bearing different physicochemical
properties (Table 1). In particular, Urantide was used as lead compound to investigate the N-terminal position in analogues with potential antagonist activity. All synthesized compounds were tested for their binding affinity on $h$-UTR-transfected CHO cells and for their contractile activity on de-endothelialized rat aortic rings (Table 1).[31]

Overall, the biological data indicate that in the "agonist series" (i.e. derived from P5U) the N-terminal substitutions of $\mathrm{Asp}^{4}$ with uncharged, aromatic or positively charged residues are generally well tolerated. The consistent reduction in binding and activity is probably due to the lost of a hydrogen bond acceptor/donor group, in accordance with previous results.[26, 49] Concerning the "antagonist series" (i.e. derived from Urantide), while a positively charged amino acid (Lys) strongly reduces the binding and the activity (compound 16), an aromatic residue is well tolerated and can increase the potency. In particular, compound $\mathbf{1 4}$, in which a Tic residue replaces the Asp ${ }^{4}$ of Urantide, showed the highest antagonist potency in the functional rat aorta bioassay ( $\mathrm{pK}_{\mathrm{B}} 8.94$ ). Since the binding constant of $\mathbf{1 4}$ to h UTR is slightly reduced compared to Urantide, the enhanced functional potency should derive from improved tissue penetration of the more hydrophobic Tic amino acid in 14 replacing an Asp residue in Urantide. Species differences between $h$-UTR and $r$-UTR could
also be invoked. To check the last hypothesis, the sequences of $h$-UTR and $r$-UTR were compared (Figure S 15 ). Since only minimal residue differences were observed near the bound ligand and, in particular, near to the Tic residue (EL2 is unchanged in the two receptors), the hypothesis was rejected.

To determine whether the different biological activities of Urantide analogues were driven by different conformational properties of the peptides or by the different chemical functionalities at the N terminus, we performed an NMR study on the interesting analogues 14 and 16 in SDS micelles solution. The use of SDS micelles to study the conformational properties of $h \mathrm{U}$-II analogues is motivated on the basis of their interaction with a membrane receptor. For peptides acting as ligands of membrane receptors (such as GPCR), the use of membrane mimetic media, such as SDS or DPC, is suggested hypothesizing a membrane-assisted mechanism of interactions between the peptides and their receptors.[50] According to this model, the membrane surface plays a key role in facilitating the transition of the peptide from a random coil conformation adopted in the extracellular environment to a conformation that is recognized by the receptor. The increase of the local concentration of the peptide and the reduction of the rotational and translational freedom of the neuropeptide are membrane-mediated events acting as determinant
steps for the conformational transition of the peptide.[51, 52] NMR has proven useful to examine the structures of bioactive peptides that cross membrane barriers.[53-55] Actually, we succeeded in correlating the SDS-bound conformation of $h \mathrm{U}-\mathrm{II}$ analogues with their biological activity.[33, 34]

We showed that $h \mathrm{U}$-II analogues, which retain high affinity for UT receptor, all possess a type II' $\beta$-hairpin backbone conformation regardless their agonist or antagonist activity, indicating that such backbone conformation is necessary for the UT recognition.[33, 34] The main conformational difference observed in the structures of the antagonists and the agonists was established in a different orientation of the $(\mathrm{D} / \mathrm{L})-\operatorname{Trp}^{7}$ side chain. In particular, while in the agonists the (D/L)-Trp ${ }^{7}$ indole moiety is close to the $\mathrm{Lys}^{8}$ side chain, in the antagonists $(\mathrm{D} / \mathrm{L})-\operatorname{Trp}^{7}$ side chain is more flexible and further from the ornitine side chain. The structural features of the "antagonist series" were found also for the analogues 14, and 16 (Figure 3) indicating that the different affinity-activity of the two compounds do not depend on a different spatial disposition of the "pharmacophoric" residues (i.e. (D/L)- $\operatorname{Trp}^{7}$, Lys $/ \mathrm{Orn}^{8}, \mathrm{Tyr}^{9}$ ) [22, 23] but must depend on different interaction of the N -terminal residue with the receptor.


Figure 7. Stereoview of $h$-UTR models in the inactive (azure) and active (sienna) conformations complexed with Urantide (red) and P5U (gold), respectively. The $h$-UTR models are superimposed using the backbone heavy atoms of TM residues apart from TM-VI. Asp ${ }^{4}$ residue is evidenced by an arrow.

To gain insight into this interaction mode we first undertook a docking study between the parent Urantide and $h$-UT receptor model. It is worth noting that, while docking studies regarding peptide agonist have been performed, $[23,43,44,56]$ the docking of peptide antagonist at UT receptor is unprecedented. Since the crystal structure of a GPCR in the active conformation is not yet disposable, we used the "active state" rhodopsin model developed by Mosberg et al. as template to build an $h-\mathrm{UTR}_{\mathrm{a}}$ model.[48] Hence, the rhodopsin receptor template was also chosen for the inactive state model $\left(h-\mathrm{UTR}_{\mathrm{i}}\right)$ to allow a direct comparison of the two models. The structures of other
mammalian GPCR's in inactive state have been solved.[57-59] Interestingly, our $h$ - $\mathrm{UTR}_{\mathrm{i}}$ model and the $\beta_{2}$-adrenergic receptor ( $\beta_{2} \mathrm{AR}, \mathrm{PDB}$ code 2 RH 1 ) are quite similar around the urantide binding site showing an rmsd of the backbone heavy atoms of $1.5 \AA$ (helices $\mathrm{II} \div \mathrm{VII}$, Figure S 16 ).

Urantide $/ h-\mathrm{UTR}_{\mathrm{i}}$ complex (Figure 5) and the MD simulations indicated that: (i) the $\beta$-hairpin structure adequately fits the binding site and is stable during the MD trajectory; (ii) the binding site, situated in the entrance of the TM bundle on the extracellular side, is formed by TM-III $\div$ TM-VII, and EL-II; (iii) particularly important for the present study, the N -terminal $\mathrm{Asp}^{4}$ residue interacts with EL-II, mostly by stable electrostatic interactions with the Arg206. Replacement of $\mathrm{Asp}^{4}$ with a Lys residue (analogue 16), in the model complex, increase the binding energy (Table 3) since the favorable interactions are lost and, in contrast, electrostatic repulsions between $\mathrm{N}^{\varepsilon}$ of Lys ${ }^{4}$ and the guanidinium group of five arginine and the $\mathrm{N}^{\varepsilon}$ of one lysine residues located on the EL-II can occur. In contrast, the loss of favorable electrostatic interaction, upon the replacement of the Asp ${ }^{4}$ of Urantide with a Tic residue (analogue 14), is partially compensated by van der Waals interactions of the phenyl ring of Tic and by a reduced desolvation energy.

Docking study between P5U and $h-\mathrm{UTR}_{\mathrm{a}}$ was also performed. The obtained complex (Figure 6) and the MD simulations indicated that: (i) the $\beta$-hairpin structure adequately fits the binding site and is stable during the MD trajectory; (ii) the binding site, situated in the entrance of the TM bundle on the extracellular side, is formed by TMIII $\div$ TM-VII, EL-II; and EL-III; (iii) the N-terminal Asp ${ }^{4}$ residue lies between EL-II and EL-III. We found similarities, but also some differences, with previous reports describing the docking of peptide agonists (hU-II, and P5U) into an UTR model.[23, 43, 44, 56] In regards to our previous work,[43] the different docking results obtained for the $\mathrm{P} 5 \mathrm{U} / h$-UTR complex is ascribable to the different conformation of both the receptor and the ligand. In fact, in the present study the $h$-UTR structure is based on an active model of rhodopsin,[48] while in the previous work the receptor was constructed starting from the X-ray inactivated form of rhodopsin.[42] Moreover, herein the presented P5U 3D structure is obtained from a NMR study in SDS micelle solution,[34] while the one used in 2005 was derived from a NMR study in DMSO solution.[30]

To assess the predictive value of the models the ligands were switched, i.e. urantide was docked within $h-\mathrm{UTR}_{\mathrm{a}}$ model and P5U within $h-\mathrm{UTR}_{\mathrm{i}}$ (Figure S 14 ). Both urantide/ $\mathrm{UTR}_{\mathrm{a}}$ and $\mathrm{P} 5 \mathrm{U} / \mathrm{UTR}_{\mathrm{i}}$ complexes show negative binding energies (Table 3), but these are
significantly lower (absolute value) than the ones of urantide/ $\mathrm{UTR}_{\mathrm{i}}$ and $\mathrm{P} 5 \mathrm{U} / \mathrm{UTR}_{\mathrm{a}}$ complexes, respectively. These results are not surprising. In fact, urantide still retains agonist activity being a full agonist in a calcium mobilization assay.[32] Interestingly, D-Trp ${ }^{7}$ aromatic moiety of urantide within $\mathrm{UTR}_{\mathrm{a}}$ binding site is close to the $\mathrm{Orn}^{8}$ side chain in a conformation which characterizes the agonist peptide ligands (Figure S14).[33] As concern P5U/UTR ${ }_{i}$ complex, the negative value of the binding energy can be explained admitting that, in a first step, even the agonists bind the receptor in its inactive (ground) state. Then, the system moves to a minimum of free energy which is reached with the receptor activation.

Urantide $/ h-\mathrm{UTR}_{\mathrm{i}}$ and $\mathrm{P} 5 \mathrm{U} / h-\mathrm{UTR}_{\mathrm{a}}$ interactions found in our models (Table 2, Table 4 and Figure 7) are different. In particular, Urantide plunges more deeply into the TM's bundle compared to P5U, probably due to the ornitine side chain length reduction, and to the D $\operatorname{Trp}^{7}$ higher flexibility. As a consequence, the exocyclic carboxylate group of Asp ${ }^{4}$ of P5U, lying at the interface between EL-II and EL-III, is more external compared to the corresponding residue in Urantide, and establishes only non-stable hydrogen bond with the receptor. In accordance with SAR data obtained by us and others,[26, 49] the presence of both aromatic (13) or positively charged (15) residues at
position 4 of P5U leads to compounds with similar binding energy (Table 3).

Recently published experimental results, reporting that the agonists and antagonists (partial agonists) interact differently with the UT receptor, are in accordance with our models.[44, 60] Boivin et al. measured the interactions of $h \mathrm{U}-\mathrm{II}, \mathrm{URP}$ and Urantide with separately synthesized $h$-UT receptor EL's.[60, 61] They observed that agonist $h$ U-II and URP bind EL-II and EL-III while the binding of Urantide was observed only with EL-II. None of these ligands were able to interact with EL-I. These results are fully consistent with our models. Leduc et al. found various interactions between photoreactive $h \mathrm{U}$-II and Urantide analogues and $r$-UTR.[44, 62] Also, these interactions are compatible with our models.

The proposed binding modes are also in qualitative agreement to the observed SAR at the core -Phe-Trp-Lys-Tyr- sequence. In fact, pharmacophoric residues $\operatorname{Trp}^{7}, \operatorname{Lys}(\mathrm{Orn})^{8}$, and $\mathrm{Tyr}^{9}$, whose substitution with Ala significantly reduces or abolishes the binding affinity of U-II analogues, show a high number of receptor interactions. In contrast, Phe $^{6}$ shows only a few interactions in accordance with SAR indicating that its substitution with Ala results in a still full agonist peptide. Furthermore, substitution of the hydroxyl group of Tyr ${ }^{9}$ of U-II with methoxy, nitro, amino, methyl,
fluoro, or a hydrogen atom does not affect the potency and the efficacy of the U-II analogues in the rat aorta bioassay.[63] These observations agree with our model since the phenolic OH is not involved in receptor binding in the $\mathrm{P} 5 \mathrm{U} / \mathrm{UTR}_{\mathrm{a}}$ model. Substitution of the Tyr residue by bulky aromatic amino acids such as (2-naphthyl)-Lalanine, biphenylalanine,[23] or 3-iodo-tyrosine [25] may even increase the binding affinity and the biological activity. Consistently, the tyrosine-binding pocket of our model can accommodate a bulkier side chain with an enhancement of the hydrophobic interactions. SAR data suggest that the presence of an aliphatic amine at position 9 is mandatory for U-II activity.[63] The position of the $\mathrm{NH}_{2}{ }^{\varepsilon}$ from the peptide backbone has been investigated using ornithine, 2,4diaminobutyric acid (Dab), and 2,3-diaminopropionic acid (Dap), i.e. with distances of 3, 2 and 1 carbon atoms, respectively. Reduction of the distance between the primary aliphatic amine and the peptide backbone of 3 and 2 methylene groups gradually reduces the potency and efficacy of the analogs and switch the activity towards antagonism. Further shortening of the amino acid side-chain increases potency and restores efficacy. Interestingly, the $\mathrm{Dab}^{8}$-urantide analogue UFP-803 behaves as a pure antagonist ( $\mathrm{pA}_{2} 7.46$ ).[64] Our model can explain these results. In fact, a distance of 3 methylene groups is suitable for both $\mathrm{UTR}_{\mathrm{i}}$ and $\mathrm{UTR}_{\mathrm{a}}$ ligands, such as urantide
$\left(\Delta \Delta \mathrm{G}_{\text {bind }}=-3.88 \mathrm{Kcal} / \mathrm{mol}\right.$, Table 3). A distance of 2 methylene groups is also suitable for the two receptor states but with a much preferred antagonist mode (for UFP-803, $\Delta \Delta \mathrm{G}_{\mathrm{bind}}=-5.41 \mathrm{Kcal} / \mathrm{mol}$; data not shown). Little attention has been paid to the $\operatorname{Trp}^{9}$ residue in the SAR studies of U-II apart from the Ala- and D-scan approaches. Replacement of the $\operatorname{Trp}$ residue with $2-\mathrm{Nal},[23]$ or 4-benzoyl-Lphenylalanine (Bpa) [62] significantly decreased agonist binding affinity and potency. This would suggest that the indole NH function may establish a hydrogen bond with some UTR residue. We don't observe this postulated H -bond and believe that the indole electron rich system is more suitable for a cation $-\pi$ interaction with the Lys ${ }^{8}$ side chain observed in the peptide agonist ligands.[33]

Based on the binding mode of UTR peptide agonists and antagonists, we derived new 3D pharmacophore models illustrated in Figure 8. The distances between the pharmacophoric residues (i.e. mean distances observed during the 2 ns MD simulations) are in good accordance with those previously reported both for peptide agonists and antagonists.[33] These pharmacophore models might be useful for the next design cycle and, in particular, for the design of smallmolecule ligands.
a)

b)


Figure 8. Stereoview of the pharmacophore model for peptide antagonists (a) and agonists (b). The distances between the aryl ring centroids of (D)Trp7 and Tyr9, and the $\mathrm{N} \varepsilon / \delta$ of $\operatorname{Lys}(\mathrm{Orn}) 8$, are displayed. Distances and standard deviations are obtained from one hundred structures saved every 20 ps of the MD simulations.

### 1.1.4 Conclusions

In conclusion, we observed a different SAR at the N -terminus for P5U compared to Urantide analogues. P5U shows a high degree of tolerance upon N-terminal substitutions. In Urantide analogues, an aromatic residue is well tolerated and can increase the potency. In fact, replacement of the Asp ${ }^{4}$ residue by Tic led to an analogue, compound 14, more potent as antagonist $\left(\mathrm{pK}_{\mathrm{B}}=8.94\right)$ compared to Urantide. Conversely, a positively charged amino acid (Lys) drastically reduces the binding and the activity. The results could be explained on the basis of the different receptor binding mode of the agonist P5U vs the antagonist Urantide. Understanding of the impact of amino acid substitutions in position 4, combined with information regarding the interactions between UT receptor and its ligands, is crucial to increase the knowledge of structure-function relationships focused to the design of new potent UT receptor ligands.

### 1.1.5. Experimental Section

Synthesis. $N^{\alpha}$-Fmoc-protected amino acids, HBTU and HOBt were purchased from Inbios (Naples, Italy). Wang resin was purchased from Advanced ChemTech (Louisville, KY). Protected Pen was purchased from Bachem (Basel, Switzerland). Peptide synthesis solvents, reagents, as well as $\mathrm{CH}_{3} \mathrm{CN}$ for HPLC were reagent grade and were acquired from commercial sources and used without further purification unless otherwise noted. The synthesis of $h \mathrm{U}$-II analogues was performed in a stepwise fashion via the solid-phase method. $\mathrm{N}^{\alpha}$ -Fmoc-Val-OH was coupled to Wang resin $\left(0.5 \mathrm{~g}, 0.7 \mathrm{mmol} \mathrm{NH}_{2} / \mathrm{g}\right)$. The following protected amino acids were then added stepwise $\mathrm{N}^{\alpha}$ -Fmoc-Cys(Trt)-OH, $\quad \mathrm{N}^{\alpha}-\mathrm{Fmoc}-\mathrm{Tyr}(\mathrm{OtBu})-\mathrm{OH}, \quad \mathrm{N}^{\alpha}$-Fmoc-Yaa( $\mathrm{N}^{\varepsilon}$ -Boc)-OH (Yaa: Lys, Orn), $\mathrm{N}^{\alpha}-\mathrm{Fmoc}-\mathrm{Xaa}\left(\mathrm{N}^{\text {in }}-\mathrm{Boc}\right)-\mathrm{OH}$ (Xaa: Trp, DTrp), $\mathrm{N}^{\alpha}$-Fmoc-Phe-OH, $\mathrm{N}^{\alpha}$-Fmoc-Pen(Trt)-OH and $\mathrm{N}^{\alpha}$ - - - $-\mathrm{Omoc}-\mathrm{R}-$ OH ( $\mathrm{R}=$ Phe, Cpa, Ala, $\left(\mathrm{pNO}_{2}\right)$ Phe, Tic, $\mathrm{Nal}(1), \mathrm{Nal}(2)$, Lys $)$. Each coupling reaction was accomplished using a 3-fold excess of amino acid with HBTU and HOBt in the presence of DIEA.

The $\mathrm{N}^{\alpha}$-Fmoc protecting groups were removed by treating the protected peptide resin with a $25 \%$ solution of piperidine in DMF, ( $1 \times 5 \mathrm{~min}$ and $1 \times 20 \mathrm{~min}$ ). The peptide resin was washed three times with DMF and the next coupling step was initiated in a stepwise manner. All reactions were performed under an Ar atmosphere. The
peptide resin was washed with DCM (3x), DMF (3x) and DCM (4x), and the deprotection protocol was repeated after each coupling step. The N-terminal Fmoc group was removed as described above and the peptide was released from the resin with $\mathrm{TFA} / \mathrm{Et}_{3} \mathrm{SiH} / \mathrm{H}_{2} \mathrm{O}$ (90:5:5) for 3 h . The resin was removed by filtration and the crude peptide was recovered by precipitation with cold anhydrous ethyl ether to give a white powder which was purified by RP-HPLC on a semi-preparative C18-bonded silica column (Vydac 218TP1010, $1.0 \times 25 \mathrm{~cm}$ ) using a gradient of $\mathrm{CH}_{3} \mathrm{CN}$ in $0.1 \%$ aqueous TFA (from 10 to $90 \%$ in 45 min ) at a flow rate of $1.0 \mathrm{~mL} / \mathrm{min}$. The product was obtained by lyophilization of the appropriate fractions after removal of the $\mathrm{CH}_{3} \mathrm{CN}$ by rotary evaporation. Analytical RP-HPLC indicated a purity $>98 \%$ and molecular weights were confirmed by FAB-MS (Fisons mod. Prospec) or HR-MS (Kratos Analytical mod. Kompact) (Supporting Information).

General Method of Oxidation and Cyclization. The peptides were oxidized by the syringe pump method previously reported.[65] The linear peptide $(300-500 \mathrm{mg})$ was dissolved in 40 ml of $50 \% \mathrm{H}_{2} \mathrm{O} / 25 \%$ acetonitrile $/ 25 \%$ methanol, and nitrogen gas was passed through the solution for 20 min . Five milliliters of saturated ammonium acetate solution were added, and the pH was taken to 8,5 with $\mathrm{NH}_{4} \mathrm{OH}$. The peptide solution was then added at room
temperature via syringe pump to a stirred oxidant solution. The oxidant solution was prepared as follows: 2 equiv of potassium ferricyanide were dissolved in 400 ml of $\mathrm{H}_{2} \mathrm{O} / 200 \mathrm{ml}$ of acetonitrile $/ 200 \mathrm{ml}$ of methanol. To this solution was added 100 ml of saturated ammonium acetate, and the pH was then taken to 8,5 with $\mathrm{NH}_{4} \mathrm{OH}$. The peptide solution was added at such a rate that approximately 10 mg of peptide were delivered per hour per liter of the oxidant. After the addition of peptide was complete, the reaction mixture was stirred for an additional 5-6h and then taken to pH 3.5 with glacial acetic acid. Amberlite IRA-68 (Cl - form) was added to remove the iron ions and the solution stirred for 20 min and then filtered. The solution was concentrated using a rotary evaporator at $30^{\circ} \mathrm{C}$ and then lyophilized. The material thus obtained was dissolved in glacial acetic acid, filtered to remove inorganic salts, and relyophilized. The crude cyclic peptides were purified by preparative HPLC on the system described above, using a gradient of $100 \%$ buffer for 20 min , then $0-20 \%$ acetonitrile in 5 min , followed by 20-60\% acetonitrile in 40 min , all at $40 \mathrm{ml} / \mathrm{min}$. Again the peptides eluted near $50 \%$ organic $/ 50 \%$ buffer. The purity of the cyclic peptides was checked by analytical HPLC (C18 column, Vydac 218TP104, 4,6mm X 25cm), using a Shimadzu SPD 10A vp with detection at 230 and 254 nm and by TLC in four solvent systems in silica gel with detection by UV light, iodine
vapours, and ninhydrin. The analytical data of the compounds synthesized in this work are given in the Supporting Information.

Organ Bath Experiments. The experimental procedures employed in this study were approved by Institutional Animal Care and Use Committee and carried out in accordance with the legislation of Italian authorities (D.L. 116 27/01/1992), which complies with European Community guidelines (CEE Directive 86/609) for the care and use of experimental animals.

Male albino rats (Wistar strain, 275-350 g) were euthanized by cervical dislocation, under ether anaesthesia. The thoracic aorta was cleared of surrounding tissue and excised from the aortic arch to the diaphragm. From each vessel, a helically cut strip was prepared, and then it was cut into two parallel strips. The endothelium was removed by gently rubbing the vessel intimal surface with a cotton-tip applicator; the effectiveness of this manoeuvre was assessed by the loss of relaxation response to acetylcholine $(1 \mu \mathrm{M})$ in noradrenaline (1 $\mu \mathrm{M})$ precontracted preparations. All preparations were placed in 5 ml organ baths filled with normal Krebs solution warmed at $37^{\circ} \mathrm{C}$ and oxygenated with $95 \% \mathrm{O}_{2}, 5 \% \mathrm{CO}_{2}$. The tissues were connected to isotonic force transducers (Ugo Basile, VA, Italy) under a constant load of 5 mN and motor activity was digitally recorded by an Octal Bridge Amplifier connected to PowerLab/8sp hardware system and
analyzed using the Chart 4.2 software (AD Instruments, Australia). After 60 min equilibration, tissue responsiveness was assessed by the addition of $1 \mu \mathrm{M}$ noradrenaline followed by a further equilibration of 60 min .

To assess the agonist activity cumulative concentration-response curves to $h \mathrm{U}$-II and to the agonist peptide under examination were constructed in paired aortic strips and responses obtained were normalized towards the control $h \mathrm{U}-\mathrm{II}$ maximal contractile effect $\left(\mathrm{E}_{\max }\right)$.

To assess the antagonist activity concentration-response curves to $h \mathrm{U}$-II were constructed cumulatively in paired aortic strips. One strip was pretreated with vehicle ( $\mathrm{DMSO} ; 1-3 \mu \mathrm{l} / \mathrm{ml}$ ) and used as a control, while the other strip was pretreated with the antagonist peptide under examination and, after a $30-\mathrm{min}$ incubation period, $h \mathrm{U}$ II was administered cumulatively to both preparations.

In each preparation only one cumulative concentration-response curve to $h \mathrm{U}$-II was carried out and only one concentration of antagonist was tested. Concentration-response curves were analyzed by sigmoidal nonlinear regression fit using the GraphPad Prism 4.0 program (San Diego, CA, U.S.A.) to determine the molar concentration of the agonist producing the $50 \%\left(\mathrm{EC}_{50}\right)$ of its maximal effect. Agonist activity of all compounds was expressed as $\mathrm{pEC}_{50}$
$\left(-\log \mathrm{EC}_{50}\right)$. The antagonist potency was expressed as apparent $\mathrm{pK}_{\mathrm{B}}$ $\left(-\log \mathrm{K}_{\mathrm{B}}\right)$ calculated from the equation: $\mathrm{pK}_{\mathrm{B}}=-(\log [\mathrm{CR}-1]-\log$ [antagonist concentration]) where the concentration-ratio (CR) is the ratio of equieffective concentrations $\left(\mathrm{EC}_{50}\right)$ of $\mathrm{hU}-\mathrm{II}$ in the presence and absence of antagonist.[66,67] The nature of the antagonism was checked by means of Schild analysis.

Binding experiments. All experiments were performed on membranes obtained from stable CHO-K1 cells expressing the recombinant human UT receptor (Euroscreen ES-440-M, Bruxelles, Belgium). Assay conditions were: TRIS-buffer (20mM, pH 7.4 at $37^{\circ}$ C) added with $\mathrm{MgCl}_{2}(5 \mathrm{mM})$ and $0.5 \%$ BSA. Final assay volume was 0.1 ml , containing $1 \mu \mathrm{~g}$ membrane proteins. The radioligand used for competition experiments was $\left[{ }^{125}\right.$ I]Urotensin II (specific activity 2000 $\mathrm{Ci} / \mathrm{mmol}$; Amersham Biosciences, Buckinghamshire, U.K.) in the range $0.07-1.4 \mathrm{nM}$ (corresponding to $1 / 10-1 / 5$ of its KD ). Nonspecific binding was determined in the presence of $1 \mu \mathrm{M}$ of unlabelled $h \mathrm{U}-\mathrm{II}$, and ranged between $10-20 \%$ of total binding. Competing ligands were tested in a wide range of concentrations $(1 \mathrm{pM}-10 \mu \mathrm{M})$. The incubation period ( 120 min at $37^{\circ} \mathrm{C}$ ) was terminated by rapid filtration through UniFilter-96 plates (Packard Instrument Company), pre-soaked for at least 2 h in BSA $0.5 \%$, and using a MicroMate 96 Cell Harvester (Packard Instrument Company). The filters were then
washed 4 times with 0.2 ml aliquots of Tris- HCl buffer ( $20 \mathrm{mM}, \mathrm{pH}$ 7.4, $\left.4^{\circ} \mathrm{C}\right)$. Filters were dried and soaked in Microscint $40(50 \mu \mathrm{l}$ in each well, Packard Instrument Company), and bound radioactivity was counted by a TopCount Microplate Scintillation Counter (Packard Instrument Company). Determinations were performed in duplicate. All binding data were fitted by using GraphPad Prism 4.0 in order to determine the equilibrium dissociation constant $\left(\mathrm{K}_{\mathrm{d}}\right)$ from homologous competition experiments, the ligand concentration inhibiting the radioligand binding of the $50 \%\left(\mathrm{IC}_{50}\right)$ from heterologous competition experiments. $\mathrm{K}_{\mathrm{i}}$ values were calculated from $\mathrm{IC}_{50}$ using the ChengPrusoff equation $\left(\mathrm{K}_{\mathrm{i}}=\mathrm{IC}_{50} /\left(1+[\right.\right.$ radioligand $\left.] / \mathrm{K}_{\mathrm{d}}\right)$ according to the concentration and $\mathrm{K}_{\mathrm{d}}$ of the radioligand.[56]

NMR Sample Preparation. $99.9 \%{ }^{2} \mathrm{H}_{2} \mathrm{O}$ were obtained from Aldrich (Milwaukee, USA), $98 \%$ SDS- $\mathrm{d}_{25}$ was obtained from Cambridge Isotope Laboratories, Inc. (Andover, USA), [(2,2,3,3-tetradeuterio-3-(trimethylsilanyl)]propionic acid (TSP) from MSD Isotopes (Montreal, Canada).

NMR Spectroscopy. The samples for NMR spectroscopy were prepared by dissolving the appropriate amount of peptide in 0.45 ml of ${ }^{1} \mathrm{H}_{2} \mathrm{O}(\mathrm{pH} 5.5), 0.05 \mathrm{ml}$ of ${ }^{2} \mathrm{H}_{2} \mathrm{O}$ to obtain a concentration $1-2 \mathrm{mM}$ of peptides and 200 mM of $\mathrm{SDS}_{\mathrm{d}}^{25}$. NH exchange studies were performed dissolving peptides in 0.50 ml of ${ }^{2} \mathrm{H}_{2} \mathrm{O}$ and 200 mM of

SDS- $d_{25}$. NMR spectra were recorded on a Varian INOVA 700 MHz spectrometer equipped with a z-gradient 5 mm triple-resonance probe head. All the spectra were recorded at a temperature of $25^{\circ} \mathrm{C}$. The spectra were calibrated relative to TSP ( 0.00 ppm ) as internal standard. One-dimensional (1D) NMR spectra were recorded in the Fourier mode with quadrature detection. The water signal was suppressed by gradient echo.[68] 2D DQF-COSY,[37, 38] TOCSY [39], NOESY [40] and PE-COSY [69] spectra were recorded in the phase-sensitive mode using the method from States.[70] Data block sizes were 2048 addresses in $t_{2}$ and 512 equidistant $t_{1}$ values. Before Fourier transformation, the time domain data matrices were multiplied by shifted $\sin ^{2}$ functions in both dimensions. A mixing time of 70 ms was used for the TOCSY experiments. NOESY experiments were run with mixing times in the range of $150-300 \mathrm{~ms}$. The qualitative and quantitative analyses of DQF-COSY, TOCSY, and NOESY spectra, were obtained using the interactive program package XEASY.[41] ${ }^{3} J_{\mathrm{HN}-\mathrm{H} \alpha}$ coupling constants were obtained from $1 \mathrm{D}{ }^{1} \mathrm{H}$ NMR and 2D DQF-COSY spectra. ${ }^{3} J_{\mathrm{H} \alpha-\mathrm{H} \beta}$ coupling constants were obtained from 1D ${ }^{1} \mathrm{H}$ NMR and 2D PE-COSY spectra, the last performed with a $\beta$ flip angle of $35^{\circ}$. The temperature coefficients of the amide proton chemical shifts were calculated from $1 \mathrm{D}{ }^{1} \mathrm{H}$ NMR and 2D TOCSY
experiments performed at different temperatures in the range $25^{\circ}-40$ ${ }^{\circ} \mathrm{C}$ by means of linear regression.

Structural Determinations. The NOE-based distance restraints were obtained from NOESY spectra collected with a mixing time of 200 ms . The NOE cross peaks were integrated with the XEASY program and were converted into upper distance bounds using the CALIBA program incorporated into the program package DYANA.[71] Cross peaks which were overlapped more than $50 \%$ were treated as weak restraints in the DYANA calculation. In a first step only NOE derived constraints (Supporting Information) were considered in the annealing procedures. Overall, 76 meaningful NOEderived restraints (9 NOEs per residue; that is: 32 intraresidue, 32 sequential, 11 medium-range, and 1 long-range) for peptide 14, and 73 (9 NOEs per residue; that is: 34 intraresidue, 29 sequential, 9 mediumrange, and 1 long-range) for peptide 16, were used as input for the calculation. For each examined peptide, an ensemble of 200 structures was generated with the simulated annealing of the program DYANA. An error-tolerant target function (tf-type=3) was used to account for the peptide intrinsic flexibility. Non standard Pen, D-Trp, Orn, and Tic residues were added to DYANA residue library using MOLMOL.[72] From these structures we could univocally determine the hydrogen bond atom acceptors corresponding to the slowly exchanging NH's
previously determined for each peptide. In a second DYANA run these hydrogen bonds were explicitly added as upper and lower limit constraints (NH of $\mathrm{Phe}^{6}$ with CO of $\mathrm{Tyr}^{9}$, and NH of $\mathrm{Tyr}^{9}$ with CO of Phe ${ }^{6}$ ), together with the NOE derived upper limit constraints (Supporting Information). The second annealing procedure produced 200 conformations from which 50 structures were chosen, whose interprotonic distances best fitted NOE derived distances, and then refined through successive steps of restrained and unrestrained EM calculations using the Discover algorithm (Accelrys, San Diego, CA) and the consistent valence force field (CVFF) [73] as previously described.[34] Coupling constants were not used in the constrained simulated annealing calculation, however, backbone and side chain conformations are in accordance with the experimental ${ }^{3} J_{\mathrm{HN}-\mathrm{H} \alpha}$ and ${ }^{3} J_{\mathrm{H} \alpha-\mathrm{H} \beta}$ coupling constants, respectively. The final structures were analyzed using the InsightII program (Accelrys, San Diego, CA). Graphical representation were carried out with the InsightII program (Accelrys, San Diego, CA). RMS deviation analysis between energy minimized structures were carried out with the program MOLMOL.[72]
$\boldsymbol{h}$-UTR Models and Docking. The theoretical structure of the $h$ UT receptor, in the inactive state, was generated by homology modeling based on the crystal structure of bovine rhodopsin (PDB
code 1F88),[42] as previously described.[43] The three-dimensional model of the the $h$-UTR, in the active state, was constructed from the model structure of the bovine rhodopsin, proposed by Mosberg,[48] and was generated by homology modeling following the same steps described for the inactive model.[43] To validate the reliability of the calculated models, the program PROCHECK,[74, 75] which automatically checks the stereochemical accuracy, packing quality, and folding reliability, was employed. All amino acids in the $\alpha$-helices were located in the favored region of the right-handed $\alpha$-helix in the Ramachandran plot. From calculated $\omega$ angles, there were no cis peptide bonds in the calculated $h$-UTR model. All $\mathrm{C} \alpha$ atoms except Cys displayed $S$-chirality. For the packing quality, there were no bump regions in the calculated $h$-UTR models.

The peptides Urantide and P5U were manually docked in the suspected binding site of the $h-\mathrm{UTR}_{\mathrm{i}}$ and $h-\mathrm{UTR}_{\mathrm{a}}$, respectively. Employing the criteria described in the Results section, we generated 10 poses for both urantide $/ h-\mathrm{UTR}_{\mathrm{i}}$ and $\mathrm{P} 5 \mathrm{U} / h-\mathrm{UTR}_{\mathrm{a}}$ complexes. Refinement of each pose was achieved by in vacuo energy minimization with the Discover algorithm (50 000 steps; $\varepsilon=1$ ). The backbone atoms of the TM and IL domains of the $h$-UTR were held in their position; the ligand and EL's were free to relax. Minimization was followed by a brief MD simulation period (200 ps). After this
period, many poses ( 7 and 8 out of the 10 poses for urantide and P 5 U , respectively) were discarded since the ligand was driven away from its starting position and lost the salt-bridge with the conserved Asp residue. The other poses ( 3 for urantide and 2 for P 5 U ) converged to a very similar conformation (rmsd of the backbone atoms $<1 \AA$ ) and the lowest energy complex for each ligand was chosen as starting point for subsequent 2 ns MD simulations (time step $=1 \mathrm{fs}, \mathrm{T}=300 \mathrm{~K}$ ). The backbone coordinates of the TM helices were fixed during the MD simulations because, without environmental constraints (i.e. lipid bilayer and water solution), they can move away from each other and can lose their helical structure. Fixing TM helices should still allows for sufficient spatial/conformational sampling of the docked complexes since the ligand, in the discarded poses (see above), significantly changed both the initial position and conformation, after the MD simulations. An average structure was calculated from the last 1 ns trajectory and energy-minimized using the steepest descent and conjugate gradient methods until a rmsd of $0.05 \mathrm{Kcal} / \mathrm{mol}$ per $\AA$ was reached. Starting from these energy minimized structures, the model complexes of the Urantide and P5U analogues 13-16 were obtained. The Asp ${ }^{4}$ was replaced with a Lys or a Tic residue and the complex was minimized first relaxing only the replaced residue (10 000 steps); then relaxing all the ligand (40 000 steps). Whereupon, a 200 ps MD
simulations was performed. The average structure of the last 100 ps was re-minimized until a rmsd of $0.05 \mathrm{Kcal} / \mathrm{mol}$ per $\AA$ was reached. For the docking of urantide within $\mathrm{UTR}_{\mathrm{a}}$ (switching of the ligands), we started from the optimized $\mathrm{P} 5 \mathrm{U} / \mathrm{UTR}_{\mathrm{a}}$ complex and superposed the NMR derived urantide structure with that of P5U (backbone atoms of residues 5-10). Then, we removed the P5U structure. The complex was minimized relaxing the ligand (40 000 steps). Whereupon, a 200 ps MD simulations was performed. The average structure of the last 100 ps was re-minimized until a rmsd of $0.05 \mathrm{Kcal} / \mathrm{mol}$ per $\AA$ was reached. Analogous steps were taken for the $\mathrm{P} 5 \mathrm{U} / \mathrm{UTR}_{\mathrm{i}}$ complex. All the MD trajectories were analyzed by means of the Analysis module of InsightII package. Molecular graphics images of the complexes were produced using the UCSF Chimera package.[76] Rescoring of the ligand/receptor models according to the AutoDock4 (AD4) [4547] scoring function was attained using a script provided within the MGLTools software package (http://mgltools.scripps.edu/).

### 1.2 Conformational Study on Cyclic Melanocortin Ligands and New Insight into their Binding Mode at the MC4 Receptor

The melanocortin receptors are involved in many physiological functions, including pigmentation, sexual function, feeding behavior, and energy homeostasis, making them potential targets to treat obesity, sexual dysfunction, etc. Understanding the basis of the ligandreceptor interactions is crucial for the design of potent and selective ligands for these receptors.

The conformational preferences of the cyclic melanocortin ligands MTII (Ac-Nle ${ }^{4}-\mathrm{c}\left[\mathrm{Asp}^{5}-\mathrm{His}^{6}-D\right.$ Phe $^{7}-$ Arg $^{8}-$ Trp $\left.\left.^{9}-\mathrm{Lys}^{10}\right]-\mathrm{NH}_{2}\right)$ and SHU9119 (Ac-Nle ${ }^{4}-\mathrm{c}\left[\mathrm{Asp}^{5}-\mathrm{His}^{6}-\mathrm{DNal}\left(2^{\prime}\right)^{7}-\mathrm{Arg}^{8}-\mathrm{Trp}^{9}-\mathrm{Lys}^{10}\right]-$ $\mathrm{NH}_{2}$ ), which show agonist and antagonist activity at the $h$-MC4R, respectively, were comprehensively investigated by solution NMR spectroscopy in different environments. In particular, water and water/DMSO (8:2) solutions were used as isotropic solutions and an aqueous solution of DPC (dodecylphosphocholine) micelles was used as a membrane mimetic environment. NMR derived conformations of these two ligands were docked within $h$-MC4R models. NMR and docking studies revealed intriguing differences which can help explain the different activities of these two ligands.

### 1.2.1 Introduction

The melanocortin family contains five human receptors ( $h$ -MC1R-h-MC5R) cloned to date and stimulates the cAMP second messenger and other signal transduction pathways.[77-84] Melanocortin receptors belong to the class A superfamily of rhodopsin-like G-protein-coupled receptors (GPCRs), characterized by having seven transmembrane $\alpha$-helices (TM1-TM7) linked by three extracellular and three intracellular loops.[78, 81, 82] The endogeneous agonists of the MCRs, the melanocortins, are a family of peptides comprised of $\alpha, \beta$-, and $\gamma$-melanocyte stimulating hormones (MSH) and adreno-corticotropic hormone (ACTH). They are derived from post-translational modification of a common precursor, proopiomelanocortin (POMC).[85] The natural melanocortins are all agonist for $h$-MCRs with exception of the $h$-MC2R, for which only ACTH is a full agonist.[86] Also, synthetic melanocortins have different pharmacological profiles for the five $h$-MCRs. For example, NDP- $\alpha$-MSH [87] and MTII [88] are agonist for all $h$-MCRs except the $h$-MC2R.[89] There are also the endogeneous protein antagonists known as agouti and agouti-related protein (AGRP).[90, 91] Interaction of these effectors with MCRs results in the modulation of numerous biological functions which include among others regulation of skin pigmentation (MC1R), steroid production (MC2R), the
immune response, thermoregulation, food intake, sexual function (MC3R and MC4R), and stress-induced anxiety and depression.[9297] The MC4R subtype is regarded as a potential drug target, because it is involved in feeding and sexual behaviour.[90, 97-100] Mammals with a defective MC4R gene, which is expressed in the brain, are characterized by obese phenotype and increased food intake.[101-103] Pharmacological studies indicate that activation of the MC4R in rodents [100] and humans [93] modulates erectile function. Consequently, research efforts have been focused on the development of potent and MC4R-selective agonists as potential antiobesity drugs or as treatments for sexual dysfunction. [104] On the other hand, a MC4R antagonist that blocks the satiety-inducing effect of $\alpha$-MSH could be helpful for treatment of anorexia or cancer cachexia.[105]

A molecular understanding of MTII and SHU9119 activity at the $h$-MC4R may have important implications in the design of drugs. In addition, the identification of the essential amino acid residues of the $h$-MC4R responsible for MTII agonism and SHU9119 antagonism should be important for understanding the signalling events that regulate the melanocortin system under physiologic conditions.[106, 107]

Hence, we first studied the conformational preferences of the cyclic melanocortin ligands MTII, and SHU9119 (Table 5), agonist
and antagonist at $h$-MC4R, respectively. Conformational analysis was carried out by NMR spectroscopy in water, water/DMSO solutions, and 200 mM aqueous solution of DPC as membrane mimetic environment. Then, NMR derived structures of MTII and SHU9119 were docked within the $h$-MC4 receptor model, in the active and inactive state, respectively.

Table 5. Ligand Sequences

| Peptide | Sequence |
| :---: | :---: |
| MTII | $\mathrm{Ac}^{2}-\mathrm{Nle}^{4}-\mathrm{c}\left[\mathrm{Asp}^{5}-\mathrm{His}^{6}-\right.$ DPhe $\left.^{7}-\mathrm{Arg}^{8}-\mathrm{Trp}^{9}-\mathrm{Lys}^{10}\right]-\mathrm{NH}_{2}$ |
| SHU9119 | $\mathrm{Ac}_{2}-\mathrm{Nle}^{4}-\mathrm{c}\left[\mathrm{Asp}^{5}-\mathrm{His}^{6}-\mathrm{DNal}\left(\mathbf{2}^{9}\right)^{7}-\mathrm{Arg}^{8}-\mathrm{Trp}^{9}-\mathrm{Lys}^{10}\right]-\mathrm{NH}_{2}$ |
| NDP-MSH | $\mathrm{Ser}^{1}-\mathrm{Tyr}^{2}-\mathrm{Ser}^{3}-\mathrm{Nle}^{4}-\mathrm{Clu}^{5}-\mathrm{His}^{6}-\mathrm{DPhe}^{7}-\mathrm{Arg}^{8}-\mathrm{Trp}^{9}-\mathrm{Gly}^{10}-\mathrm{Lys}^{11} \mathrm{Pro}^{12} \mathrm{Val}^{13-} \mathrm{Gly}^{14}$ |

### 1.2.2 Results

Chemistry. Peptides were synthesized using the solid phase approach and standard Fmoc methodology in a manual reaction vessel (Experimental Section).[35]

The purification was achieved using a semi-preparative RP-HPLC C18 bonded silica column (Vydac 218TP1010). The purified peptide was $98 \%$ pure as determined by analytical RP-HPLC. The correct molecular weight and composition of the peptide was confirmed by mass spectrometry and amino acid analysis (Table S5, Supporting Information).

NMR Analysis. Complete ${ }^{1} \mathrm{H}$ NMR chemical shift assignments were achieved for MTII and SHU9119 according to the Wüthrich procedure [36] via the usual systematic application of DQF-COSY [37, 38], TOCSY [39], and NOESY [40] experiments (Tables S6-S13, Supporting Information) with the support of the XEASY software package.[41] The conformational preferences for the two peptides were investigated by solution NMR spectroscopy in different solvent environments. In particular, water and water/DMSO (8:2) solutions were used as isotropic solutions and 200 mM aqueous solution of DPC (dodecylphosphocholine micelles) was used as a membrane mimicking environment.

Water solution. NMR analysis was performed in water at pH 5
and a peptide concentration of about 2 mM . First, we analyzed the peptides at $25{ }^{\circ} \mathrm{C}$. Both peptides showed similar NMR parameters (Tables S6-S7) as previously reported.[108, 109] Almost all NMR parameters indicate structural flexibility: a) temperature effects $|\Delta \delta / \Delta \mathrm{T}|>6 \mathrm{ppb} / \mathrm{K} ; \mathrm{b}){ }^{3} J_{\mathrm{HN}-\mathrm{H} \alpha}$ coupling constants are all within the range $6-8 \mathrm{~Hz} ; \mathrm{c}$ ) No standard $\alpha$-helix or $\beta$-sheet structure from $\mathrm{H} \alpha$ CSI (chemical shift index) values [110] (Figure S17, Supporting Information); d) No unambiguous medium- or long-range backbone NOE connectivities were found in the ROESY or NOESY. Strong $d_{\alpha \mathrm{N}}(i, i+1)$ NOEs, which are generally observed in extended structures, appeared along almost the entire length of the peptides.

To reduce the conformational flexibility, we acquired the spectra also at $5^{\circ} \mathrm{C}$. At this temperature, most of the NMR parameters did not change significantly (Tables S8-S9), though improvements of the signal quality in the NOESY spectra were observed. Weak $d_{\mathrm{\alpha N}}(i, i+2)$ NOEs between $\mathrm{Nle}^{4}$ and $\mathrm{His}^{6}$ and between $\mathrm{His}^{6}$ and $\mathrm{Arg}^{8}$ could be observed. Also the methyl protons of the N-terminal acetyl group show a weak NOE contact with the H $\alpha$ of $\operatorname{Asp}^{5}$. Medium $d_{\mathrm{NN}}(i, i+1)$ NOE between $D \operatorname{Phe}^{7}\left(D \mathrm{Nal}\left(2^{\prime}\right)^{7}\right)$ and $\mathrm{Arg}^{8}$ was observed. Among the possible observable contacts, the $d_{\alpha N}(i, i+2)$ NOE between $\mathrm{Asp}^{5}$ and $D$ Phe $^{7}$ could not be observed due to spectral overlap.

Cryoscopic solution. To further reduce the peptide - 60 -
conformational flexibility, NMR spectra were acquired at $-10^{\circ} \mathrm{C}$ in a cryomixture solution of water/DMSO 8:2. Such cryomixtures have been shown to produce physico-chemical conditions compatible with those of biological fluids.[111] Again, most of the NMR parameters did not change significantly (Tables S10-S11). Apart from the two $d_{\alpha \mathrm{N}}(i, i+2)$ NOEs observed also in water $\left(5^{\circ} \mathrm{C}\right)$ the NOESY spectra showed $d_{\alpha N}(i, i+2)$ NOEs between $\operatorname{Asp}^{5}$ and $D \operatorname{Phe}^{7}\left(D \operatorname{Nal}\left(2^{\prime}\right)^{7}\right)$. Furthermore, $d_{\beta \mathrm{N}}(i, i+3)$ NOEs between $\mathrm{Asp}^{5}$ and $\operatorname{Arg}^{8}$ and $d_{\mathrm{NN}}(i, i+2)$ NOEs between $\mathrm{His}^{6}$ and $\mathrm{Arg}^{8}$ also were observed.

DPC micelles. Several NMR parameters indicate that MTII and SHU9119 are highly structured in DPC solution. In particular, ${ }^{3} J_{\mathrm{HN}-\mathrm{H} \alpha}$ coupling constants (Tables S12-S13) and $\mathrm{H} \alpha$ CSI values (Figure S17) and many NOE signals (Tables S14-S15, Supporting Information) clearly point to a folded structure encompassing the N -terminal residues (4-7) and extended conformation of residues 8-9. Non-trivial medium range NOE interactions, among which $d_{\alpha \mathrm{N}}(i, i+2) 4-6,5-7,6-$ $8, d_{\mathrm{NN}}(i, i+2) 6-8$, and $d_{\beta \mathrm{N}}(i, i+3) 5-8$, are observed indicating that the membrane mimetic environment stabilizes intrinsic conformational tendencies of the peptide. Additional medium and long range NOE contacts were observed. In particular, various NOEs connected the $\mathrm{Nle}^{4}$ side chain with both $D \mathrm{Phe}^{7}\left(D \mathrm{Nal}\left(2^{\prime}\right)^{7}\right)$ and $\operatorname{Trp}^{9}$ aromatic moieties indicating spatial proximity of these side chains. Other -61-
interesting contacts were observed between the methyl protons of the N-terminal acetyl group and the HN of $\mathrm{Asp}^{5}$, and between amide protons of the C-terminal $\mathrm{NH}_{2}$ group and the $\mathrm{H} \delta 1$ and $\mathrm{H} \varepsilon 1$ of $\mathrm{Trp}^{9}$.

Structure Determination. NOE distance restraints obtained for MTII and SHU9119 in DPC micelles were used as the input data for a simulated annealing structure calculation using the program DYANA.[71] The annealing procedure produced 100 conformations from which 20 structures were chosen, whose interprotonic distances best fitted the NOE derived distances, and then refined through successive steps of restrained and unrestrained EM calculations using the Insight/Discover package (Accelrys Inc, San Diego, CA).

Structure Description. Superposition of the 10 lowest energy conformers of MTII and SHU9119 are shown in Figure 9. Since a $\beta$ turn may be defined as four consecutive non-helical residues that have a $\mathrm{C} \alpha(\mathrm{i})-\mathrm{C} \alpha(\mathrm{i}+3)$ distance $<7 \AA$, two $\beta$-turns that involve $\mathrm{Nle}^{4}$ to $D$ Phe $^{7}\left(\mathrm{DNal}\left(2^{\prime}\right)^{7}\right)$ and $\mathrm{Asp}^{5}$ to $\mathrm{Arg}^{8}$, can be identified. Examination of the backbone dihedral angles at the central $(i+1$, and $i+2)$ residues of the turns showed that these turn structures most closely resembled type I (residue 4 to 7 ) and type II (residue 5 to 8 ) $\beta$-turns, although deviations from the standard dihedral angles of these two types of $\beta$ turn occurred (Table S16, Supporting Information). Residues 8 to 10 are in extended conformations. The side chain $\chi_{1}$-angles of $\mathrm{Asp}^{5}$,
$\mathrm{Arg}^{8}, \mathrm{Trp}^{9}$ and Lys ${ }^{10}$ are also well defined, preferring trans, gauche-, trans and gauche- orientations, respectively. Side chains of $\mathrm{Nle}^{4}$ and $\mathrm{His}^{6}$ are more flexible. The $D \mathrm{Phe}^{7}$ orientation in MTII is also well defined as trans. The $D \operatorname{Nal}\left(2^{\prime}\right)^{7}$ orientation in SHU9119 is less defined showing an equilibrium between trans and gauche + rotamers. These results are consistent with the measured ${ }^{3} J_{\mathrm{H} \alpha \mathrm{H} \beta}$ coupling constants (Tables S12-S13).[112, 113]

The peptide surface has amphipathic nature. In fact, considering the $p$ seudo-plane defined by the backbone atoms (green ribbon, Figure 9) the hydrophobic residues $\mathrm{Nle}^{4}, D \mathrm{Phe}^{7}\left(D \mathrm{Nal}\left(2^{2}\right)^{7}\right)$ and $\operatorname{Trp}^{9}$ lie on one side (right in Figure 9) while the positively charged residues His ${ }^{6}$ and $\mathrm{Arg}^{8}$ lie on the other side.


Figure 9. Stereoviews of the 10 lowest energy conformers of MTII (a), and SHU9119 (b). Structures were superimposed using the backbone heavy atoms of residues 5-10. Heavy atoms are shown with different colours (carbon, green; nitrogen, blue; oxygen, red). Hydrogen atoms are not shown for clarity.

Docking Studies. NMR derived structures of MTII and SHU9119 were docked within the $h$-MC4R models proposed by Mosberg.[114, 115] In particular, the agonist MTII was docked within an "active state" model ( $h$ - $\left.\mathrm{MC}_{\mathrm{a}}^{\mathrm{a}}\right)$, while the antagonist SHU9119 was docked within an "inactive state" model $\left(h-\mathrm{MC}_{\mathrm{i}}^{\mathrm{i}}\right)$. Figure 10 shows a snake-like diagram of the $h$-MC4R sequence.


Figure 10. Snake-like diagram of the h-MC4R sequence. This plot was generated with the RbDe software.[116] Black residues indicate that mutation data are available. The ' . . .' indicates hidden residues (see reference [114] for the complete sequence).

Since the currently available docking programs may not work very well for peptide compounds (more than eight rotatable bonds) [117], manual docking was conducted. The NMR-derived MTII structure was placed in between the trans-membrane domains of the $h$ $M C 4 R_{a}$. The following criteria were employed to achieve meaningful docking modes: (i) The positively charged side of the amphipathic surface of the peptides should be close to the carboxylate groups of Asp122, Asp126, and Glu100, as suggested by several mutagenesis studies [114, 115, 118-125]; (ii) $D \mathrm{Phe}^{7}$ residue should point towards Leu133 residue as suggested by a mutagenesis study [118]; (iii) No steric clashes should occur between any atom. To assess the stability
of the $\mathrm{MTII} / h-\mathrm{MC}_{\mathrm{a}} \mathrm{R}_{\mathrm{a}}$ complex we analyzed the potential ligand/receptor interactions, energy minimization and MD simulations for 1 ns at a constant temperature of $300^{\circ} \mathrm{K}$. During the MD simulation, the ligand, the EL's, and all the receptor side chains were allowed to relax, while the TM's and intracellular loops (ILS) backbone atoms were held frozen. The distances between the peptide and the key receptor residues were monitored along the complete 1 ns MD trajectory (Supporting Information). The mean structure of the last 0.5 ns of MD was energy minimized and used for subsequent analysis.

Table 6. $\mathrm{MTII} / h-\mathrm{MC} 4 \mathrm{R}_{\mathrm{a}}$ Interactions

| Residue $^{\mathrm{a}}$ | Surrounding residue |
| :--- | :--- |
| Nle $^{4}$ | Val193(TM5), His264 <br> (TM6) (TM6), Leu265 (TM6), Tyr268 |
| Asp $^{5}$ | Asp122 (TM3), Asn123(TM3), Asp126 (TM3) |
| His $^{6}$ | Asp126 (TM3), Ile129 (TM3), Leu133 (TM3), Phe184 <br> DPhe |
|  | (TM4), Phe261 (TM6), Phe284 (TM7), Leu288 (TM7) |
| Arg $^{8}$ | Glu100 (TM2), Asp122 (TM3), Ile125 (TM3), Asp126 |
| Trp $^{9}$ | His264 (TM6), Phe267 (TM6), Pro272 (EL3), Val278 <br> (EL3), Phe280 (EL3), Phe284 (TM7) |
| Lys $^{10}$ | Tyr276 (EL3), Val278 (EL3), Met281 (EL3) |

${ }^{\text {a }}$ For sake of clarity, the residue numbers of the ligands are reported as superscript while those of the receptor are not. ${ }^{\mathrm{b}}$ Receptor residues involved in mutagenesis studies are shown in bold.

To inspect the variations in the ligand conformation, the rmsd with the respect to the starting structure was calculated. Interestingly, the rmsd of the MTII backbone atoms turned out to be stable throughout all of the MD simulations $(0<\operatorname{rmsd}<0.8 \AA)$, indicating that the peptide settles into the receptor-binding site in a stable conformation. Also the side chain orientations are those described by NMR. In particular, the $D \mathrm{Phe}^{7}$ and $\operatorname{Trp}^{9}$ side chain prefer a trans orientation about $\chi_{1}$ angle $\left(\chi_{1} \approx 160^{\circ}\right.$ for $D$ Phe $^{7}, \chi_{1} \approx-163^{\circ}$ for $\left.\operatorname{Trp}^{9}\right)$. As shown in Figure 11a, the hypothetical binding site of MTII is located among TM2-TM7, and EL3. C- and N-terminal residues point towards the extracellular side. The binding mode of the peptide is determined mainly by the interactions shown in Figure 11 b and summarized in Table 6.

b)



${ }^{L 133}>$


Figure 11. (a) Stereoview of $h-\mathrm{MC}_{\mathrm{H}} \mathrm{R}_{\mathrm{a}}$ model complexed with MTII. MTII heavy atoms are shown with different colours (carbon, green; nitrogen, blue; oxygen, red). Hydrogen atoms are not shown for clarity. Receptor backbones are represented in gray and labeled. (b) Stereoview of MTII within the binding pocket of $h-\mathrm{MC}_{\mathrm{a}}$. Hydrogen bonds are represented with dashed lines.

In particular, (i) a tight charge-reinforced hydrogen-bonding network involving the carboxylate groups of Glu100 and Asp126 with the protonated guanidinium group of $\mathrm{Arg}^{8}$ of MTII are established. The guanidinium group of $\mathrm{Arg}^{8}$ is also involved in an electrostatic interaction with the carboxylate group of Asp122. Such interactions, which we assume to be anchoring points of the ligand to the $h$ $M C 4 R_{a}$, remained stable during the entire MD simulation (Figure $S 18$, Supporting Information). The oxygen atoms of the carboxylate of Asp126 form a charge-reinforced hydrogen bond with the protonated imidazole group of $\mathrm{His}^{6}$, which was not stable during the MD simulation (Figure S18). (ii) Two hydrophobic pockets, delineated by residues listed in Table 6, host the side chains of $D \mathrm{Phe}^{7}$, and $\operatorname{Trp}^{9}$ of MTII. Particularly, the side chain of $D \mathrm{Phe}^{7}$ occupies the hydrophobic pocket involving residues Ile129, Leu133, Phe184, Phe261, Phe284, while the indole system of $\operatorname{Trp}^{9}$ is surrounded by Phe280, Pro272, Phe267, Phe284 and appears to be optimally oriented for a $\pi$-stacking interaction with the imidazolic system of His264. iii) Terminal groups also contribute to the complex stabilization. The $\mathrm{Nle}^{4}$ side chain is close to Val193 and Tyr268. The acetyl group CO of $\mathrm{Nle}^{4}(\mathrm{~N}-$ terminal) engages hydrogen bonds with imidazole $\mathrm{NH} \varepsilon$ of His264. Amide group $\mathrm{NH}_{2}$ of Lys ${ }^{10}$ (C-terminal) established a hydrogen bond with the phenolic OH of Tyr276. These H-bonds are not stable during
the MD production run (data not shown).
The NMR-derived SHU9119 structure was placed within the trans-membrane domains of the $h-\mathrm{MC}_{\mathrm{C}}^{\mathrm{i}}$ model, following the same criteria used for MTII (see above) to achieve meaningful binding interactions. Energy minimization and MD simulations (1 ns) were run to assess the stability of the SHU9119/h-MC4R $\mathrm{R}_{\mathrm{i}}$ complex and to analyze the potential ligand/receptor interactions. The mean structure of the last 0.5 ns of MD was extensively minimized and used for subsequent analysis. SHU9119 backbone atoms turned out to be stable throughout all of the MD simulations $(0<\operatorname{rmsd}<0.9 \AA)$, indicating that the peptide settles into the receptor-binding site in a stable conformation. Also the side chain orientations are those described by NMR. In particular, $\operatorname{Trp}^{9}$ prefers a trans orientation about the $\chi_{1}$ angle $\left(\chi_{1} \approx 175^{\circ}\right)$. Finally, the $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ side chain adopts a gauche + conformation $\left(\chi_{1} \approx 80^{\circ}\right)$.

b)



Figure 12. (a) Stereoview of $h-\mathrm{MC}_{\mathrm{C}}^{\mathrm{i}}$ model complexed with SHU9119. SHU9119 heavy atoms are shown with different colours (carbon, orange; nitrogen, blue; oxygen, red). Hydrogen atoms are not shown for clarity. Receptor backbones are represented in cyan and labeled. (b) Stereoview of SHU9119 within the binding pocket of $h-\mathrm{MC}_{\mathrm{i}}^{\mathrm{i}}$. Hydrogen bonds are represented with dashed lines.

Table 7. SHU9119/h-MC4R $\mathrm{R}_{\mathrm{i}}$ Interactions

| Residue ${ }^{\text {a }}$ | Surrounding residue |
| :---: | :---: |
| $\mathrm{Nle}^{4}$ | His264 (TM6), Leu265 (TM6), Tyr268 (TM6), <br> Tyr276(EL3)      |
| Asp ${ }^{5}$ | Phe184 (TM4) |
| His ${ }^{6}$ | Thr1 18 (EL1), Asp122 (TM3), Asn123 (TM3), Asp126 (TM3), Phe184 (TM4) |
| $D \mathrm{Nal}^{7}$ | Leu133 (TM3), Phe 184 (TM4), Cys196 (TM5), Leu197 (TM5), Met200 (TM5), Phe261 (TM6), Phe262 (TM6), Leu265 (TM6) |
| Arg ${ }^{8}$ | Glu100 (TM2), Asp 122 (TM3), Ile125 (TM3), Asp 126 (TM3), |
| Trp ${ }^{9}$ | His264 (TM6), Leu265 (TM6), Tyr268 (TM6), Phe267 (TM6), Phe280 (EL3), Met281 (EL3), Phe284 (TM7) |
| Lys ${ }^{10}$ | Thr112 (EL1), Asp113 (EL1), Ser116 (EL1), Thr118 (EL1), Tyr268 (TM6), Tyr276 (EL3), Val278 (EL3), Met281 (EL3) |

${ }^{\text {a }}$ For sake of clarity, the residue numbers of the ligands are reported as superscript while those of the receptor are not. ${ }^{\mathrm{b}}$ Receptor residues involved in mutagenesis studies are evidenced in bold.

As shown in Figure 12a, the hypothetical binding site of SHU9119 is located among TM2-TM7, and EL3. C- and N-terminal residues point towards the extracellular side. The binding mode of the peptide is determined mainly by the interactions showed in Figure 12b and Table 7. In particular, (i) considering $\mathrm{Arg}^{8}$ and $\mathrm{His}^{6}$, the same interactions observed in the MTII/h-MC4Ra complex are observed also for SHU9119. Again, Hbonds involving the $\mathrm{Arg}^{8}$ guanidinium group remained stable during the whole MD production run (Figure S19, Supporting Information), while those of $\mathrm{His}^{6}$ were not. (ii) Two hydrophobic pockets, involving the residues listed in Table 7, host the aromatic side chains of $D \operatorname{Nal}\left(2^{\prime}\right)^{7}$, and $\operatorname{Trp}^{9}$ of SHU9119. These
pockets only partially overlap with those hosting the aromatic side chains of MTII. Particularly, the side chain of $D \operatorname{Nal}\left(2^{\prime}\right)^{7}$ occupy the hydrophobic pockets involving residues Leu133, Phe184, Cys196, Leu197, Met200, Phe261, Phe262, Leu265, while the indole group of Trp ${ }^{9}$ is surrounded by His264, Phe267, Met281, Phe284 and appears to be optimally oriented for a $\pi$-stacking interaction with the aromatic group of Tyr268. This is different from MTII, because $\mathrm{Trp}^{9}$ in MTII makes $\pi$-stacking with His 264 . The $\mathrm{Nle}^{4}$ side chain is close to Leu265, Tyr268, and Tyr276. The acetyl group CO of $\mathrm{Nle}^{4}$ (N-terminal) in SHU9119 isn't involved in hydrogen-bonding. Finally, the amide group $\mathrm{NH}_{2}$ and the oxygen atom of the terminal carboxamide group of Lys ${ }^{10}$ (C-terminal) established two hydrogen bonds: with the OH of Ser116 and with the phenolic OH of Tyr268. These Hbonds are not stable during the MD production run (data not shown).

### 1.2.3 Discussion

We investigated the conformational preferences of the cyclic melanocortin ligands MTII and SHU9119 by solution NMR spectroscopy in different environmental situations: water, water/DMSO (8:2) and an aqueous solution of DPC (dodecylphosphocholine). In water and water/DMSO cryoscopic mixture, NMR parameters were very similar for both peptides and indicate structural flexibility. A few NOEs, however, point to a tendency of the peptides to form a turn-helical conformation at the N terminus (residue 4-8). The data could be indicative of a nascent helix in solution.[126] The nascent helix consists of a population of different conformations, in which a significant proportion contains backbone conformations in the $\alpha$-region of $(\phi, \psi)$ space in the Ramachandran plot, rather than of any single defined solution conformation.

In DPC micelle solution, the peptides exhibited a higher conformational stability. The use of micelles to study the conformational properties of peptides has been described in section 1.1.3.[50-55]

Two consecutive $\beta$-turns that involved $\mathrm{Nle}^{4}$ to $D \mathrm{Phe}^{7} / D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ (distorted type I) and $\mathrm{Asp}^{5}$ to $\mathrm{Arg}^{8}$ (distorted type II) and a short extended segment along residues $\operatorname{Trp}^{9}$ and Lys ${ }^{10}$ were observed in the
calculated structures of MTII and SHU9119 (Figure 9 and Table S16). It is noteworthy that an amphiphilic molecular surface was obtained for the message sequence residues in both peptides. The main conformational difference observed in the structures of the two ligands was established in a different orientation of the $D \mathrm{Phe}^{7}$ and $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ side chains. $D \mathrm{Phe}^{7}$ of MTII preferred the trans rotamer, while the $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ side chain of SHU9119 was more flexible.

A type II $\beta$-turn structure encompassing residues $5-8$ was already found by NMR analysis of MTII and SHU9119 in water solution.[108, 109] This $\beta$-turn led to stacking between the aromatic rings of $\mathrm{His}^{6}$ and $D$ Phe $^{7}$ in MTII while no aromatic stacking between $\operatorname{His}^{6}$ and $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ was found in SHU9119. This stacking was not observed in the structures obtained in DPC micelles. Considering the $\beta$-turn encompassing residues 4-7, it has never been observed in the structure of MTII or SHU9119. Interestingly, the presence of this turn is in accordance with the results of N-methylation of MTII backbone amide bonds. In fact, N-methylation of $D \mathrm{Phe}^{7}$, which should destabilize this $\beta$-turn, caused a total loss of binding as well as adenylate cyclase activity at the $h$-MC4R ( $h$-MC1R, $h$-MC3R and $h$-MC5R).[127] NMethylation of $\mathrm{Arg}^{8}$ caused a dramatic reduction of the binding (about 500-fold at the $h$-MC4R) but yielded a compound that retained full agonist activity toward all subtypes of melanocortin receptors.

To gain insight into the interaction mode of these ligands with the $h$-MC4R, we first undertook a docking study between MTII and $h$ MC4R model. Since the crystal structure of a GPCR in the active conformation has not yet been obtained, we used a $h$-MC4R model in the "active state" proposed by Mosberg et al. (h-MC4R ${ }_{\mathrm{a}}$ ).[114] According to these authors, upon activation, the receptor experiences a rearrangement which involves mainly the TM6 helix. The TM6 helix shifts outward and rotates counterclockwise (viewed from the extracellular side) during activation, moving its intracellular end away from TM3 and toward TM5. As a result of this and other changes, the receptor structure tightens near its extracellular surface but opens up at the cytoplasmic side, providing a cavity for binding of the $\mathrm{G}_{\alpha \mathrm{s}}$ subunit. In the active state model, several side chains change their orientation among which Trp258, in accordance with earlier spectroscopic results.[128] Similar conformational changes upon activation of the MC4R were subsequently proposed also by Hogan et al.[121] During the manuscript preparation another model of the $h-\mathrm{MC}_{\mathrm{a}} \mathrm{R}_{\mathrm{a}}$ has been published.[129] This model was based on recent crystal structures of the GPCR opsin in the ligand-free and in the G-protein-interacting conformations.[130, 131] Interestingly, our $h-\mathrm{MC}_{\mathrm{a}}$ model and that built by Chapman et al. are quite similar showing an rmsd of the TM's backbone heavy atoms of $2.0 \AA$ (Figure S20, Supporting Information).

For the $\mathrm{MTII} / h-\mathrm{MC}_{\mathrm{a}} \mathrm{R}_{\mathrm{a}}$ complex, docking and the MD simulations (Figure 11) indicated that: (i) the structure adequately fits the binding site and is stable during the MD trajectory; (ii) the binding site, situated in the entrance of the TM bundle on the extracellular side, is formed by TM2-TM7, and EL3 (Figure 11a); (iii) the pharmacophore residues $D \mathrm{Phe}^{7}$, the $\mathrm{Arg}^{8}$ and $\mathrm{Trp}^{9}$ side chains establish the highest number of interactions with the receptor. In particular, $\mathrm{Arg}^{8}$ residue is involved in a charge-reinforced hydrogen bonding network with carboxylate groups of Glu100, Asp122, and Asp126 which was stable during the MD simulations (Figure S18). In contrast, the His ${ }^{6}$ imidazole group participates only in an unstable hydrogen bond with Asp126 (Figure S18). Two wide hydrophobic pockets host the side chains of $D \mathrm{Phe}^{7}$, and $\mathrm{Trp}^{9}$ of MTII (Figure 11b). The N - and C-terminal groups point towards the extracellular side and are involved only in limited interactions with the receptor consistent with the observation that these termini can be substituted with retention of potent binding affinity.

The proposed binding mode is in qualitative accordance with the known structure-activity relationships of MTII. In fact, substitution of $D$ Phe $^{7}$ or $\operatorname{Trp}^{9}$, which show a large number of receptor interactions, with alanine resulted in compounds with very low affinities for $h$ MC4R ( $h$-MC3R and $h$-MC5R).[132] The $\mathrm{Arg}^{8}$ involved in stable
interactions with the receptor, its replacement with the neutral residue, alanine, led to an active analogue but with more than a 1000 -fold reduced affinity at $h$-MC4R compared to the parent compound, in accordance with the stable interactions exhibited by this residue. In contrast, the substitution of $\operatorname{His}^{6}$ with alanine yielded a peptide with activation and binding affinity similar to MTII towards the $h$-MC4R ( $h$-MC3R and $h$-MC5R). Therefore, the imidazole group was shown not to be essential to binding of MTII with the $h$-MC4R ( $h$-MC3R and $h$-MC5R). A similar result was reported for the 'core' peptide AcHis $^{6}-D$ Phe $^{7}-$ Arg $^{8}-\operatorname{Trp}^{9}$-amide in which the omission of histidine resulted in the tripeptide that was only 2 -fold less potent at $h$-MC4R than the tetrapeptide.[124] Considering N-terminal acetyl group, an analogue of MTII without the acetyl group was as potent as MTII at the $h$-MC4R ( $h$-MC3R and $h$-MC5R).[133] Replacement of Ac-Nle ${ }^{4}$ with Ala or Ac-Ala yielded compounds with agonist potencies at $h$ MC4R similar to that of MTII. The analogue without both acetyl group and norleucine was 200 -fold less active at $h$-MC4R. Also replacement of MTII residues with proline (Pro-scan) gave similar results. Proline replacement was acceptable only at $\mathrm{Nle}^{4}$ and $\mathrm{His}^{6}$ positions yielding compounds with agonist potencies at the $h$-MC4R similar to that of MTII.[133]

Interestingly, many residues of the receptor involved in the interaction with MTII were identified as molecular determinants of ligand binding by mutagenesis studies (Table 6).[107, 115, 118-124] In particular, His264 has been demonstrated to be essential for melanocortin peptide activation of the MC4R.[134] $\pi$-stacking interaction of imidazolic nucleus of His264 and indole system of $\operatorname{Trp}{ }^{9}$ can trigger the MC4R activation. Interestingly, this $\pi$-stacking interaction is not observed in the $\mathrm{SHU} 9119 / h-\mathrm{MC}_{\mathrm{i}} \mathrm{R}_{\mathrm{i}}$ complex described below.

A docking study between SHU9119 and $h$ - $\mathrm{MC}_{\mathrm{C}} \mathrm{R}_{\mathrm{i}}$ also was performed. The $h$ - $\mathrm{MC} 4 \mathrm{R}_{\mathrm{i}}$ model, built by Mosberg et al. and based on the rhodopsine crystal structure, was used.[115] The main differences between inactive and active models of $h$-MC4R were discussed above. The obtained complex (Figure 12) and the MD simulations indicated that SHU9119 positioning within the $h-\mathrm{MC}_{\mathrm{i}}$ is similar to that observed for $\mathrm{MTII} / h-\mathrm{MC}_{\mathrm{a}}$ (Figure 12 and Figure 13). In particular, backbone atoms of MTII and SHU9119 are almost superimposable lying inside the TM2-TM7 bundle. Also the side chains of the positively charged residues $\mathrm{His}^{6}$ and the $\mathrm{Arg}^{8}$ show the same orientation and the $\mathrm{Arg}^{8}$ residue is involved in a charge-reinforced hydrogen bonding network with carboxylate groups of Glu100,

Asp122, and Asp126 which was stable during the MD simulations period (Figure S19).


Figure 13. Stereoview of $h$-MC4R models in the active (gray) and inactive (cyan) conformations complexed with MTII (carbon, green; nitrogen, blue; oxygen, red) and SHU9119 (carbon, orange; nitrogen, blue; oxygen, red), respectively. The $h$ MC4R models are superimposed using the backbone heavy atoms of TM residues apart from TM6. Hydrogen atoms are not shown for clarity.

In contrast, the $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ and $\operatorname{Trp}^{9}$ binding pockets are quite different compared to those of the corresponding residues of MTII. These differences depend on the different orientations of the $D \operatorname{Nal}\left(2^{\prime}\right)^{7}$ and $\operatorname{Trp}^{9}$ side chains and on the movement of TM6 during activation.[115] In the $\mathrm{SHU} 9119 / h-\mathrm{MC}_{\mathrm{L}} \mathrm{R}_{\mathrm{i}}$ complex the $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ prefers a gauche+ orientation due to steric interaction with Leu133, while in the $\mathrm{MTII} / h-\mathrm{MC}_{4} \mathrm{R}_{\mathrm{a}}$ complex the $D \mathrm{Phe}^{7}$ side chain could adopt a trans orientation. Furthermore, the $\chi_{2}$ torsion angle of $\operatorname{Trp}^{9}$ rotates from $9,7^{\circ}$ in the $\mathrm{MTII} / h-\mathrm{MC}_{\mathrm{a}} \mathrm{R}_{\mathrm{a}}$ to $-56,0^{\circ}$ in SHU9119/h$M C 4 R_{i}$. Interestingly, the 2 '-naphthalene and indole moieties of SHU9119 show many van der Waals interactions with hydrophobic residues of the TM6 helix which could stabilize the inactive state of the $h$-MC4R (Table 7). As a matter of fact, different groups have proposed that large aromatic side chain substitutions at the $\mathrm{Phe}^{7}$ position of $\alpha-\mathrm{MSH}$ analogues can interfere with MC4R activation by interacting with receptor residues within TM6, physically hindering the conformational changes necessary to elicit full efficacy.[120, 125] Our model is also supported by the observation that the $D \operatorname{Nal}\left(2^{\prime}\right)^{7}$ naphthalene external ring fills the same cleft as the Phe ${ }^{113}$ benzene ring of AGRP (Agouti related protein, an endogenous antagonist) in a model of AGRP/h-MC4R $\mathrm{i}_{\mathrm{i}}$ complex (Figure 14).[115] It can be observed in the same Figure 14 that also the $\mathrm{Arg}^{8}$ guanidinium group
of SHU9119 is perfectly overlapped with the same groups of $\mathrm{Arg}^{111}$ of AGRP.


Figure 14. Stereoview of SHU9119/h-MC4R $\mathrm{R}_{\mathrm{i}}$ (orange-cyan) and AGRP/h-MC4R $\mathrm{i}_{\mathrm{i}}$ (gold-purple) models in the inactive conformations. On the left side, AGRP's labels are shown; on the right side, SHU9119's labels are shown. The $h$-MC4R models are superimposed using the backbone heavy atoms of TM residues. Hydrogen atoms are not shown for clarity.

Similar conformation and positioning of MTII and SHU9119 within the MC4R are not surprising since it was shown that the single substitution of Leu133 with a methionine residue in the receptor converted SHU9119 from an antagonist into an agonist at the $h$ MC4R.[118] Probably, according to our model, when Leu133 was replaced with methionine, which is more flexible than leucine, the hindering amino acid was removed and the $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ bulky aromatic
side chain could be accommodated in the same cleft occupied by $D$ Phe $^{7}$ of the agonist MTII. An analogous point mutation in the $h$ MC3R had the same effect on SHU9119 activity.[135] Interestingly, SHU9119 behaves as an agonist at the $h$-MC1R and $h$-MC5R where a methionine or a (smaller) valine residue, respectively, occupies the position corresponding to Leu133 according to the sequence alignment reported in the reference.[115]

Other groups have suggested modeled docked conformations of melanotropin peptides with the MC4R for both agonist and antagonist ligands. In particular concerning the agonists, a few models of the NDP-MSH,[114, 129, 136] a model of $\alpha$-MSH-ND (the open analogue of MTII),[137] and a model of the tetrapeptide His- $D$ Phe-Arg-Trp [121] complexed with $h$-MC4R have been proposed. It is noteworthy that different ligand conformations were employed for the peptide agonist/MC4R models proposed (Table 8). As a consequence of the lack of an accepted melanocortin peptide active conformation, all the models proposed in literature, included the MTII $/ h-\mathrm{MC} 4 \mathrm{R}$ presented here, were only partially superposable.

## Table 8.

| Peptide | Conformation | Central Residues | Reference |
| :---: | :---: | :---: | :---: |
| NDP-MSH | type II (hairpin) | 6-7 | [112] |
| NDP-MSH | type I' | 6-7 | [135] |
| NDP-MSH | type II' | 7-8 | [127] |
| $\alpha$-MSH-ND | type I | 6-7 | [136] |
| Core ${ }^{\text {a }}$ | type II' | 7-8 | [118] |

${ }^{\mathrm{a}}$ His- $D$ Phe-Arg-Trp tetrapeptide core sequence.

For example, considering the NDP-MSH/h-MC4R complex proposed by Mosberg et al.,[114] it is quite different from our MTII/ $h$-MC4R model described above (Figure 15). NDP-MSH [87] and MTII [88] differ by cyclization at Asp $^{5}$-Lys ${ }^{10}$ residues (in MTII), but share the same pharmacophoric sequence $\operatorname{His}^{6}-D$ Phe $^{7}-\operatorname{Arg}^{8}-\mathrm{Trp}^{9}$.

b)



Figure 15. (a) Stereoview of MTII $/ h-\mathrm{MC}_{\mathrm{C}} \mathrm{R}_{\mathrm{a}}$ (green-gray) and NDP-MSH $/ h$ MC4R $\mathrm{a}_{\mathrm{a}}$ (gold-purple) models in the active conformations. (b) Bottom stereoview of MTII and NDP-MSH within the binding pocket of $h-\mathrm{MC}_{4}$. Only pharmacophoric side chains of ligands (His6-Arg9) are showed for clarity. Important residues of receptors are represented. The $h$-MC4R models are superimposed using the backbone heavy atoms of TM residues. Hydrogen atoms are not shown for clarity.

The receptor coordinates of the two complexes are very similar; indeed, we started from the Mosberg's $h-\mathrm{MC}_{\mathrm{a}} \mathrm{R}_{\mathrm{a}}$ coordinates, and
only marginal changes of the extracellular loops could be observed after the $\mathrm{MTII} / h-\mathrm{MC}_{\mathrm{a}}$ complex optimization (Figure 15). In contrast, ligand conformations are different considering the common tetrapeptide fragment. A $\beta$-hairpin-like structure with a distorted type II $\beta$-turn spanning His $^{6}$ - $D$ Phe $^{7}$ was proposed for Mosberg's NDPMSH, while our NMR-derived MTII structure shows two consecutive $\beta$-turns spanning residues $\mathrm{Asp}^{5}-\mathrm{His}^{6}$ and $\mathrm{His}^{6}-\mathrm{DPhe}^{7}$ (see above). Also side chain orientation of $\operatorname{Trp}^{9}$ was different in the two peptides being trans in MTII and gauche- in NDP-MSH. Hence, even if both peptides are located within the TM2-TM7 bundle at the extracellular side, their interactions with $h-\mathrm{MC}_{\mathrm{a}} \mathrm{R}_{\mathrm{a}}$ appear to be different. In particular, in the $\mathrm{MTII} / h-\mathrm{MC} 4 \mathrm{R}_{\mathrm{a}}$ model $\mathrm{His}^{6}$ and $\mathrm{Arg}^{8}$ are swapped compared to NDP-MSH/h-MC4R ${ }_{\mathrm{a}}$ in Mosberg's model. In Mosberg's model, $\mathrm{His}^{6}$ forms the most stable interactions with the Glu100, Asp122, and Asp126 negatively charged side chains, while $\mathrm{Arg}^{8}$ is more solvent exposed. Furthermore, in the NDP-MSH $/ h-\mathrm{MC}_{\mathrm{C}} \mathrm{R}_{\mathrm{a}}$ model, the indole group of $\operatorname{Trp}^{9}$ roughly occupies the same position as $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ or Phe ${ }^{113}$ of the antagonists SHU9119 and AGRP, respectively. Finally, the $D$ Phe $^{7}$ residues are located in similar positions within the receptor.

Some differences in the SAR data of NPD-MSH and MTII were observed both in terms of binding affinity: alanine substitutions were
generally better tolerated in linear NDP-MSH than in cyclic MTII $[119,125,132]$; and in terms of efficacy: the $D-(4-\mathrm{Cl}) \mathrm{Phe}^{7}$ substitution in NDP-MSH substantially reduces $\mathrm{E}_{\text {max }}$ but does not appreciably affect MC4R activation by the cyclic MTII.[106, 125] Different SARs could indicate different orientations of MTII and NDP-MSH (or other linear peptides) within the binding pocket thus justifying the different interactions found in the complex models. Finally, Mosberg et al. also proposed a MTII $/ h-\mathrm{MC}_{\mathrm{a}}$ a complex model.[138] In this complex, the receptor model was the same as previously developed by the authors while the MTII structure was modeled from that of NDP-MSH. The lack of details about the interactions within this complex model does not allow any comparison with our model.

Considering the peptide antagonists, a SHU9119/h-MC4R complex model has been very recently proposed.[139] Apart from $\operatorname{Arg}^{8}$ which was close to Glu100, Asp122, and Asp126 also in this model, other side chain interactions were different from those observed in our model. Again, a different backbone conformation of the bound peptide, a type-I $\beta$-turn in that case, can explain these differences. A few AGRP derived peptides were also docked within MC4R models. The triplet peptide Arg-Phe-Phe, the smallest conserved motif of AGRP which mediates the key interactions with

MC4R, was docked into the $h$-MC4R.[140] A bicyclic $h$ AGRP derivative was docked into the mouse MC4R (m-MC4R).[141] The refined averaged NMR structure of $h \operatorname{AGRP}(87-132)$ was docked both into a $h$-MC4R [115] and a $m$-MC4R [142] model. When considered the $\mathrm{Arg}^{111}$ - Phe ${ }^{112}$ - Phe ${ }^{113}$ triad, the docked structures of AGRP derivatives all maintain similar putative ligand-receptor locations, which are illustrated in Figure 14.

### 1.2.4 Conclusions

In conclusion, NMR-derived MTII and SHU9119 structures show two consecutive $\beta$-turns spanning residues Asp $^{5}-\mathrm{His}^{6}$ and $\mathrm{His}^{6}$ $D$ Phe $^{7}$ (or $D \mathrm{Nal}\left(2^{\prime}\right)^{7}$ ) with some differences in the $\mathrm{Phe} / \mathrm{Nal}{ }^{7}$ side chain orientation. Computational docking experiments of these structures, using three-dimensional homology molecular model of the $h$-MC4R, identified the main interactions between MC 4 receptor and its peptide ligands. These findings may be crucial to increase our knowledge of structure-function relationships focused on the design of new potent MC4 receptor ligands.

### 1.2.5 Experimental Section

Synthesis. $N^{\alpha}$-Fmoc-protected amino acids, HBTU and HOBt were purchased from Inbios (Naples, Italy). Wang resin was purchased from Advanced ChemTech (Louisville, KY). Synthesis of MT-II and SHU-9119 were performed by standard FMOC Strategy.[35]

NMR Sample Preparation. $99.9 \%{ }^{2} \mathrm{H}_{2} \mathrm{O}$ were obtained from Aldrich (Milwaukee, USA), $98 \%$ SDS- $_{25}$ was obtained from Cambridge Isotope Laboratories, Inc. (Andover, USA), [(2,2,3,3-tetradeuterio-3-(trimethylsilanyl)]propionic acid (TSP) from MSD Isotopes (Montreal, Canada).

NMR Spectroscopy. The samples for NMR spectroscopy were prepared by dissolving the appropriate amount of peptide to obtain a concentration $1-2 \mathrm{mM}$ in 0.55 ml of ${ }^{1} \mathrm{H}_{2} \mathrm{O}(\mathrm{pH} 5.5), 0.05 \mathrm{ml}$ of ${ }^{2} \mathrm{H}_{2} \mathrm{O}$ for water samples, 0.48 mL of ${ }^{1} \mathrm{H}_{2} \mathrm{O}(\mathrm{pH} 5.5), 0.12 \mathrm{~mL}$ of $\mathrm{DMSO}_{\mathrm{d} 6}$ for cryoscopic solution, 200 mM of $\mathrm{SDS}^{-\mathrm{d}_{25}}$ or DPC- $_{38}$ for micelle samples. NMR spectra were recorded on a Varian INOVA 700 MHz spectrometer equipped with a z-gradient 5 mm triple-resonance probe head. All the spectra were recorded at a temperature of $25^{\circ} \mathrm{C}$. The spectra were calibrated relative to TSP ( 0.00 ppm ) as internal standard. One-dimensional (1D) NMR spectra were recorded in the Fourier mode with quadrature detection. Water suppression was
achieved by using the double-pulsed field gradient spin-echo (DPFGSE) scheme [68]. 2D DQF-COSY [37, 38], TOCSY [39], NOESY [40], and PE-COSY [69] spectra were recorded in the phasesensitive mode using the method of States.[70] Data block sizes were 2048 addresses in $t_{2}$ and 512 equidistant $t_{1}$ values. Before Fourier transformation, the time domain data matrices were multiplied by shifted $\sin ^{2}$ functions in both dimensions. A mixing time of 70 ms was used for the TOCSY experiments. NOESY experiments were run with mixing times in the range of $150-300 \mathrm{~ms}$. The qualitative and quantitative analyses of DQF-COSY, TOCSY, and NOESY spectra, were obtained using the interactive program package XEASY.[41]. ${ }^{3} J_{\mathrm{HN}-\mathrm{H} \alpha}$ coupling constants were obtained from $1 \mathrm{D}{ }^{1} \mathrm{H}$ NMR and 2D DQF-COSY spectra. The temperature coefficients of the amide proton chemical shifts were calculated from $1 \mathrm{D}{ }^{1} \mathrm{H}$ NMR and 2D TOCSY experiments performed at different temperatures by means of linear regression.

Structural Determinations. The NOE-based distance restraints were obtained from NOESY spectra collected with a mixing time of 200 ms . The NOE cross peaks were integrated with the XEASY program and were converted into upper distance bounds using the CALIBA program incorporated into the program package DYANA.[71] Cross peaks which overlapped more than $50 \%$ were
treated as weak restraints in the DYANA calculation. For each examined peptide, an ensemble of 200 structures was generated with the simulated annealing of the program DYANA. An error-tolerant target function (tf-type=3) was used to account for the peptide intrinsic flexibility of the peptide. The annealing procedure produced 200 conformations from which 50 structures were chosen, whose interprotonic distances best fitted NOE derived distances, and then refined through successive steps of restrained and unrestrained EM calculations using the Discover algorithm (Accelrys, San Diego, CA) and the consistent valence force field (CVFF) [73] as previously described. Coupling constants were not used in the constrained simulated annealing calculation, however, backbone and side chain conformations are in accordance with the experimental ${ }^{3} J_{\mathrm{HN}-\mathrm{H} \alpha}$ and ${ }^{3} J_{\mathrm{H} \alpha-\mathrm{H} \beta}$ coupling constants, respectively. The PROMOTIF program, was used to extract details on the location and types of structural secondary motifs.[143] Graphical representation were carried out with the InsightII program (Accelrys, San Diego, CA). RMS deviation analysis between energy minimized structures were carried out with the program MOLMOL.[72]

Docking Procedures. The peptides MTII and SHU91119 were manually docked in the proposed binding site of the $h-\mathrm{MC}_{4} \mathrm{R}_{\mathrm{a}}$ and $h$ $M C 4 R_{i}$, respectively. Employing the criteria described in the Results
section, we generated 10 structures for both $\mathrm{MTII} / h-\mathrm{MC}_{\mathrm{a}}$ and SHU9119/h-MC4R $\mathrm{i}_{\mathrm{i}}$ complexes. Refinement of each structure was achieved by in vacuo energy minimization with the Discover algorithm (50 000 steps; $\varepsilon=1$ ). The backbone atoms of the TM and IL domains of the $h$-MC4R were held in their position; the ligand and EL's were free to relax. Minimization was followed by a brief MD simulation period (200 ps). After this period, many poses (7 and 8 out of the 10 poses for MTII and SHU9119, respectively) were discarded since the ligand was driven away from its starting position and lost the salt-bridge with the conserved Asp residues. The other structures (3 for MTII and 2 for SHU9119) converged to a very similar conformation (rmsd of the backbone atoms $<1 \AA$ ) and the lowest energy complex for each ligand was chosen as the starting point for subsequent 1 ns MD simulations (time step $=1 \mathrm{fs}, \mathrm{T}=300 \mathrm{~K}$ ). The backbone coordinates of the TM helices were fixed during the MD simulations because, without environmental constraints (i.e. lipid bilayer and water solution), they can move away from each other and can lose their helical structure. Fixing TM helices should still allows for sufficient spatial/conformational sampling of the docked complexes since the ligand, in the discarded poses (see above), significantly changed both the initial position and conformation, after the MD simulations. An average structure was calculated from the last
0.5 ns trajectory and energy-minimized using the steepest descent and conjugate gradient methods until a rmsd of $0.05 \mathrm{Kcal} / \mathrm{mol}$ per $\AA$ was reached. All the MD trajectories were analyzed by means of the Analysis module of the InsightII package. Molecular graphics images of the complexes were produced using the UCSF Chimera package.[76] Rescoring of the ligand/receptor models according to the AutoDock4 (AD4) [45-47] scoring function was attained using a script provided within the MGLTools software package (http://mgltools.scripps.edu/).

### 1.3 Novel Octreotide Dicarba-Analogues with High Affinity and Different Selectivity for Somatostatin Receptors.

A limited set of novel octreotide dicarba-analogues with nonnative aromatic side-chains in positions 7 and/or 10 were synthesized. Their affinity towards the $\operatorname{ssts}_{1-5}$ was determined. Derivative 4 exhibited a pan-somatostatin activity, except sst $_{4}$, and derivative 8 exhibited high affinity and selectivity towards $\mathrm{sst}_{5}$. Actually, compound 8 has similar $\mathrm{sst}_{5}$ affinity $\left(\mathrm{IC}_{50} 4.9 \mathrm{nM}\right)$ to SRIF-28 and octreotide. Structure-activity relationships suggest that the $Z$ geometry of the double bond bridge is that preferred by the receptors. The NMR study on the conformations of these compounds in $\mathrm{SDS}_{-\mathrm{d} 25}$ micelles solution shows that all these analogues have the pharmacophore $\beta$ turn spanning $\mathrm{Xaa}^{7}-\mathrm{d}-\mathrm{Trp}^{8}-\mathrm{Lys}^{9}-\mathrm{Yaa}^{10}$ residues. Notably, the correlation between conformation families and affinity data strongly indicates that the $\mathrm{sst}_{5}$ selectivity is favored by a helical conformation involving the C-terminus triad, while a pan-SRIF mimic activity is based mainly on a conformational equilibrium between extended and folded conformational states.

### 1.3.1 Introduction

The cyclic tetradecapeptide somatostatin ( $\mathrm{H}-\mathrm{Ala}^{1}-\mathrm{Gly}^{2}-c\left[\mathrm{Cys}^{3}-\right.$ $\left.\mathrm{Lys}^{4}-\mathrm{Asn}^{5}-\mathrm{Phe}^{6}-\mathrm{Phe}^{7}-\mathrm{Trp}^{8}-\mathrm{Lys}^{9}-\mathrm{Thr}^{10}-\mathrm{Phe}^{11}-\mathrm{Thr}^{12}-\mathrm{Ser}^{13}-\mathrm{Cys}^{14}\right]-$ OH, SRIF-14) was first isolated from mammalian hypothalamus. [144] This hormone is widely distributed in the human body and is found in the gut, pancreas, nervous system and in some exocrine and endocrine glands. By interactions with a family of five SRIF receptors (ssts), the native peptide exerts a great number of regulatory effects, especially those related to GH release. Different receptor subtypes mediate various functions but only sst $_{2}$ and sst $_{5}$ activities have been precisely related to specific physiological activities.[145] SRIF receptors are strongly expressed in various types of malignant cells, particularly in some neuroendocrine or neuroendocrine-like tumors. Over the last three decades, this has prompted researchers to prepare a huge number of new cyclic and acyclic analogues, which are more stable than SRIF in physiological conditions. Amongst these, a large number of reduced-size cyclic analogues, with or without the disulfide bridge, were synthesized and tested for their affinity towards the ssts. Furthermore, their pharmacological behaviour was studied and several NMR investigations on their affinity/conformations relationships were carried out. J. E. Rivier's group, at the Salk Institute of La Jolla, carried out a careful structure/affinity study on SRIF analogues,
introducing non natural aminoacids in the sequence and preparing variably sized S-S bridged cyclopeptides. These authors related the structure/conformation of the cyclopeptides to the $\mathrm{sst}_{1-4}$ specific affinity by means of NMR studies.[146-149]

Octreotide [150] (compound 1, Figure 16), a cyclic octapeptide analogue of somatostatin, containing a disulfide tether, and showing high affinity and selectivity for $\mathrm{sst}_{2}$, was the first analogue to be used in clinical protocols. Following the enormous growth in preparation and application of radiolabelled peptides for tumor imaging and therapy, the somatostatin analogues thus far obtained were designed mainly for the targeting of malignant cells with $\gamma$ - or $\beta$-emitting radionuclides.[147, 151, 152] As a matter of fact, octreotide derivatives [ ${ }^{111}$ In-DTPA] octreotide (OctreoScan) and $\left[{ }^{90} \mathrm{Y}\right.$-DOTA$\mathrm{Tyr}^{3}$ ]octreotide (OctreoTher) are both quite successfully used in the clinical diagnosis and therapy of neuroendocrine tumors, respectively.[153] Nevertheless, the vulnerability of the S-S bridge to endogenous and exogenous oxidating and reducing agents, such as those employed in the experimental conditions of labelling with the radioisotopes ${ }^{99 \mathrm{~m}} \mathrm{Tc}$ or ${ }^{188} \mathrm{Re}$,[154] prompted us to synthesize dicarbaanalogues of similar ring size, by the RCM reaction on two allylglycines, substituting the relevant $\mathrm{Cys}^{3,14}$ residues in the linear peptide. Compound 2, reported in Figure 16 as an example, is the first
octreotide dicarba-tethered analogue synthesised by us and has the same aminoacid sequence of the corresponding, $\mathrm{S}-\mathrm{S}$ bridged, molecule. [155, 156]


1


2

Figure 16. Structure of Octreotide (SMS201-995) (1) and of the first dicarba SRIF mimetic (2).[155] (Note: numbering of the residues follows that of the native SRIF).

The resulting unsaturated dicarba bridge proved to be insensitive to the conditions used for ${ }^{99 \mathrm{~m}} \mathrm{Tc}$ or ${ }^{188} \mathrm{Re}$ labelling (unpublished results) and the molecules obtained were very stable in human serum.[155, 156] Recently, the stability of these compounds was exploited in the successful conjugation of cytotoxic dichloroplatinum complexes to analogue 2 as well as to the double bond hydrogenated derivative.[157] The same reaction, attempted with the octreotide molecule, failed. When the affinities of these analogues towards the five ssts were determined, we ascertained that some of them showed unexpected specific affinity for the $\mathrm{sst}_{5}$ subtype, which led us to define a novel pharmacophore model for this receptor.[156]

This study reports the synthesis of new cyclooctapeptide dicarbaanalogues, that have structures similar to those depicted in Figure 16, but are designed to carry different aromatic residues in positions 7 and/or 10 (Table 9). In the following, ssts subtypes affinities found for the new compounds $4-8$ are correlated with the $\mathrm{C}=\mathrm{C}$ bridge geometry and with the conformational behaviour in SDS $_{-\mathrm{d} 25}$ micelles solution, investigated by NMR experiments. Characteristic structure/affinity relationships of this class of somatostatin analogues are widely discussed.

Table 9. Peptide Sequences; General Formula: d-Phe ${ }^{2}-c\left[d h D S A-N^{3}-\right.$ Xaa ${ }^{7}-\mathrm{d}-\mathrm{Trp}^{8}-\mathrm{Lys}^{9}-\mathrm{Yaa}^{10}-\mathrm{dhDSA}-\mathrm{C}^{14} \mathrm{~J}-\mathrm{Thr}(\mathrm{ol})^{15}-\mathrm{OH}$.

| Peptide | $\mathrm{Xaa}^{7}$ | $Y a a^{10}$ | Double bond geometry |
| :---: | :---: | :---: | :---: |
| $2^{\text {a }}$ | Phe | Thr | Z |
| $3^{\text {a }}$ | Phe | Tyr(Bzl) | E |
| 4 | Phe | Tyr(Bzl) | Z |
| 5 | 1-Nal | Thr | Z |
| 6 | Phe | Tyr | E |
| 7 | Phe | Tyr | Z |
| 8 | $1-\mathrm{Na}$ | Tyr(Bzl) | Z |

${ }^{\text {a }}$ These compounds were previously reported.[156]

### 1.3.2 Results

Peptide Synthesis and Purification. The synthesis of dicarbaanalogues followed the procedure described in our previous articles. $[155,156]$ Starting from $\mathrm{H}-\mathrm{L}-\mathrm{Thr}(\mathrm{tBu})-\mathrm{ol}-2$-chlorotrityl resin ( 0.5 $\mathrm{mmol} / \mathrm{g})$ already containing $\left[\mathrm{Thr}(\mathrm{ol})^{15}\right]$, the elongation of the peptide sequence was stopped after the coupling of $\mathrm{Hag}^{3}$ residue, with the aim of removing any possible interference of the aromatic ring of D-Phe in the correct orientation of the allylglycine side chains. After the Fmoc$\mathrm{Hag}^{3}$ coupling, the resin loading ( $0.5 \mathrm{mmol} / \mathrm{g}$ ) already met the requirements of the pseudo-dilution effect, minimizing the risk of the formation of intermolecular bonds. The linear heptapeptides were then converted by RCM by the $2^{\text {nd }}$ generation Grubbs catalyst (9) (Figure 17) to the corresponding cyclic analogues.


9
Figure 17. ${ }^{\text {2nd }}$ Generation Grubbs Catalyst.

The D -Phe ${ }^{2}$ terminal residue was added only after ring-closing, thus facilitating the cyclization step. Cleavage of the crude peptides
from the resin was obtained using the standard cleavage mixture TFA/ $\mathrm{H}_{2} \mathrm{O} / \mathrm{EDT} /$ Phenol (94:2:2:2, 3 h ) for compounds 5, 6 and 7 and with the new percentage mixture (70:26:2:2, $2,30 \mathrm{~h}$ ) for compounds 4 and $\mathbf{8}$, in order to overcome the loss of the benzyl group of the $\operatorname{Tyr}(\mathrm{Bzl})$ residue by hydrolysis, as described in our previous article.[156] All compounds obtained by RCM with 9 were prepurified by SPE. The concentrated compound adsorbed on the SPE was eluted with an increased percentage of $\mathrm{CH}_{3} \mathrm{CN}$ in $\mathrm{H}_{2} \mathrm{O}$ (from $0 \%$ to $100 \%$ ). The fractions enriched with each desired compound were then purified by semi-preparative RP-HPLC and characterized by ESIMS. For each peptide, with the exception of $\mathbf{5}$ and $\mathbf{8}$, the HPLC chromatogram showed two peaks with the same MW, corresponding to the geometric isomers $(Z / E$ ratio $\approx 90: 10)$. In particular, the $E$ structure of the $\mathrm{C}-\mathrm{C}=\mathrm{C}-\mathrm{C}$ tether of the sample eluted at lower $\mathrm{R}_{t}$ and the $Z$ structure one of the second, more intense, peak, was ascertained by ${ }^{1} \mathrm{H}$ NMR inspection. The HPLC purity of each compound studied was $>97 \%$ and the isolated compounds showed unique $E$ or $Z$ configuration, confirmed by NMR analysis. No oligomer by-products were observed.

Binding Affinity to sst $_{1-5}$ Receptors. All compounds were tested for their ability to bind to the five human $\operatorname{sst}_{1-5}$ receptors subtypes in complete displacement experiments using the universal somatostatin
radioligand $\left[{ }^{125} \mathrm{I}\right]-\left[\mathrm{Leu}^{8}, \mathrm{D}-\operatorname{Trp}^{22}, \mathrm{Tyr}^{25}\right]$-somatostatin-28. SRIF-28 was run in parallel as control. $\mathrm{IC}_{50}$ values were calculated after quantification of the data using a computer-assisted image processing system. Binding data indicate that all compounds show sub- $\mu \mathrm{M}$ binding affinities towards the $\operatorname{sst}_{5}$ (Table 10). Compounds 2 and 3 have already been described [156] and are reported for comparison. While peptide 3 was a potent and selective sst $_{5}$ ligand, its $Z$-isomer, peptide 4 , exhibited a pan-somatostatin affinity, apart from sst $_{4}$. In fact, the analogue 4 doubled the affinity toward sst $_{5}$ but completely lost the selectivity of $\mathbf{3}$.

Peptide 5 is the $1-\mathrm{Nal}^{7}$ analogue of 2 (Table 9). This peptide exhibited a low nanomolar $\mathrm{sst}_{2,5}$ affinity. Actually, it is the most potent $\mathrm{sst}_{2}$ ligand among the dicarba analogues prepared to date. The doublebond isomer analogues $6(E)$ and $7(Z)$, in which the phenolic group of $\mathrm{Tyr}^{10}$ replaces the $\mathrm{Tyr}(\mathrm{Bzl})$ residue of 3 and 4, respectively, did not show any significant affinity toward sst $_{1-5}$ subtypes apart from a slight affinity of 7 to sst $_{2}$.

Finally, analogue 8 shared $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ residue with peptide 4 and $1-\mathrm{Nal}^{7}$ residue with peptide 5 . Like compound 4, it showed affinity for all the receptor subtypes except sst $_{4}$. However, the significant enhancement of the sst ${ }_{5}$ affinity (nearly 3-fold compared to compound 4) and the simultaneous reduction of affinity towards $\operatorname{sst}_{1-4}$ make
compound $\mathbf{8}$ a strong and selective sst ${ }_{5}$ ligand. Indeed, compound $\mathbf{8}$ is the most potent $\mathrm{sst}_{5}$ dicarba-analogue synthesized so far, showing an affinity close to the value found for the reference compound SRIF-28 (Table 10).

Table 10. Receptor affinities of the somatostatin analogues.

| $\mathrm{IC}_{50}(\mathrm{nM})^{\mathrm{a}}$ |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| No. | $\mathrm{sst}_{1}$ | $\mathrm{sst}_{2}$ | $\mathrm{sst}_{3}$ | $\mathrm{sst}_{4}$ | $\mathrm{sst}_{5}$ |
| SRIF-28 | $2.3 \pm 0.4(7)$ | $3.0 \pm 0.2(7)$ | $3.6 \pm 0.5(7)$ | $1.6 \pm 0.3(7)$ | $2.4 \pm 0.2(6)$ |
| $\mathbf{2}^{\mathrm{b}}$ | $>1000(2)$ | $44 \pm 1(2)$ | $>1000(2)$ | $412 \pm 68(2)$ | $28 \pm 2(2)$ |
| $\mathbf{3}^{\mathrm{b}}$ | $>1000(2)$ | $>1000(2)$ | $892 \pm 245(2)$ | $>1000(2)$ | $29 \pm 1(2)$ |
| $\mathbf{4}$ | $25 \pm 1(3)$ | $46 \pm 3(3)$ | $25 \pm 4(3)$ | $346 \pm 23(3)$ | $12.3 \pm 0.3(3)$ |
| $\mathbf{5}$ | $>1000(3)$ | $9.6 \pm 0.9(3)$ | $>1000(3)$ | $249 \pm 51(3)$ | $16.5 \pm 4.5(3)$ |
| $\mathbf{6}$ | $1000(3)$ | $355.5 \pm 45.5(3)$ | $1000(3)$ | $1000(3)$ | $418 \pm 56(3)$ |
| $\mathbf{7}$ | $>1000(3)$ | $87 \pm 18(3)$ | $>1000(3)$ | $>1000(3)$ | $161 \pm 27(3)$ |
| $\mathbf{8}$ | $57.5 \pm 12.5(3)$ | $101 \pm 9(3)$ | $92.5 \pm 0.5(3)$ | $>1000(3)$ | $4.9 \pm 1.0(4)$ |

${ }^{\text {a }}$ The number of independent repetitions to obtain the mean values $\pm$ SEM are indicated between brackets. SRIF-28 is used as internal control. ${ }^{\text {b }}$ Corresponds to data published previously.[155]

NMR Analysis. NMR analysis of the analogues 3-8 was performed by means of 1 D and 2D proton homonuclear experiments. NMR experiments were recorded on a Varian Inova-Unity 700 MHz spectrometer. Spectra were collected in $\operatorname{SDS}_{-\mathrm{d} 25}(200 \mathrm{mM})$ micelles solution. All samples (about 2 mM ) were kept at 308 K and at $\mathrm{pH} \cong 5$.

Complete ${ }^{1} \mathrm{H}$ NMR chemical shift assignments were effectively achieved for all the analyzed molecules according to the Wüthrich procedure,[36] via the usual systematic application of TOCSY [39] and NOESY [40] experiments with the support of the XEASY software package (Tables S19-S24, Supporting Information).[41] NMR-derived constraints obtained for all compounds were used as the input data for a simulated annealing structure calculation, as implemented within the standard protocol of the DYANA program.[71]

Compound 3. The analogue 3 bears $\operatorname{Tyr}(\mathrm{Bzl})$ in position 10. We have already analyzed this peptide in our previous work in water/DMSO- $\mathrm{d}_{6}$ solution.[156] The geometry of the double bond was confirmed as trans $(E)$ from the coupling constant $\left({ }^{3} J_{\mathrm{CH}=\mathrm{CH}}=15.1 \mathrm{~Hz}\right)$ between the two olefinic protons of the bridge and NOE contacts between the same olefinic and the $\mathrm{H}_{\beta} \mathrm{S}$ of residue 14 (3). A qualitative analysis of short- and medium-range NOEs, ${ }^{3} J_{\mathrm{NH}-\mathrm{H} \alpha}$ coupling constants, and temperature coefficients for exchanging NH was used to characterize the secondary structure of $\mathbf{3}$. Spectra analysis pointed to the presence of a $\beta$-turn about residues 7-10. Interestingly, the upfield shift observed for $\mathrm{H}_{\gamma} \mathrm{s}$ of $\operatorname{Lys}^{9}(\delta=0.52,0.43 \mathrm{ppm})$ has been used for decades as diagnostic for biological activity.[158] NOEderived constraints obtained for 3 were used as the input data for a
simulated annealing structure calculation (Table S25, Supporting Information). The backbone arrangement of $\mathbf{3}$ was well-defined, possessing an average root mean square deviation (rmsd) of the heavy atoms equal to $0.15 \AA$. No violation higher than $0.1 \AA$ was observed again indicating conformational stability (Table S25). Main backbone features were a type II' $\beta$-turn spanning residues D - $\mathrm{Trp}^{8}$ - $\mathrm{Lys}^{9}$, followed by a short $3_{10}$-helix along residues $\operatorname{Tyr}(\mathrm{Bzl})^{10}-\mathrm{dhDsa}-\mathrm{C}^{14}-$ Thr $(\mathrm{ol})^{15}$ (Figure 18) The turn structure is stabilized by hydrogen bonds between $\mathrm{Phe}^{7}-\mathrm{CO}$ and $\operatorname{Tyr}(\mathrm{Bzl})^{10}-\mathrm{NH}$. The helical structure is stabilized by H-bonds between $\mathrm{D}-\mathrm{Trp}^{8}-\mathrm{CO}$ and dhDsa- $\mathrm{C}^{14}-\mathrm{NH}$ and between $\mathrm{Lys}^{9}-\mathrm{CO}$ and $\operatorname{Thr}(\mathrm{ol})^{15}-\mathrm{NH}$. These bonds are typical of $3_{10^{-}}$ helix structure $(i, i+3)$. The side chains of dhDsa- $\mathrm{N}^{3}, \mathrm{D}-\mathrm{Trp}^{8}$, Lys $^{9}$, $\operatorname{Tyr}(\mathrm{Bzl})^{10}$, and dhDsa-C ${ }^{14}$ showed well-defined $\chi 1$ values (i.e., trans, trans, gauche ${ }^{-}$, gauche ${ }^{-}$, and gauche ${ }^{+}$orientations, respectively). These orientations allowed a close spatial proximity between D-Trp ${ }^{8} / \mathrm{Lys}^{9}$ side chains; moreover, the tyrosyl group of the residue 10 points toward the Lys ${ }^{9}$ side chain. In contrast, D-Phe ${ }^{2}$ and $\mathrm{Phe}^{7}$ side chain showed almost free rotation about the $\chi 1$ torsion angle. Also, the Bzl group of residue 10 was highly flexible.


Figure 18. Stereoview of the lowest energy conformer of compound 3. Backbone is evidenced as a ribbon. Side chains of the ten lowest energy conformers are also shown as mesh surface. Surfaces are distinguished with different colors. N-Term, N-terminus; C-Term, C-terminus.

Compound 4. The analogue $\mathbf{4}$ is the geometric $(Z)$ isomer of $\mathbf{3}$ as established by the coupling constant ( ${ }^{3} \mathrm{~J}_{\mathrm{CH}=\mathrm{CH}}=8.1 \mathrm{~Hz}$ ) between the two olefinic protons of the bridge and the relative strong NOE between the same olefinic $\mathrm{H}_{\gamma} \mathrm{s}$. This analogue shows spectral features similar to those found in $\mathbf{3}$ but with a greater tendency to conformational heterogeneity. In fact, NOESY spectra of 4 showed, simultaneously, both diagnostic connectivities consistent with folded structures: $\mathrm{d}_{\alpha \mathrm{N}}(\mathrm{i}, \mathrm{i}+2)$ between $\mathrm{H}_{\alpha}-8 / \mathrm{NH}-10, \mathrm{H}_{\alpha}-9 / \mathrm{NH}-14, \mathrm{H}_{\alpha}-10 / \mathrm{NH}-$ 15 and $\mathrm{d}_{\alpha \mathrm{N}}(\mathrm{i}, \mathrm{i}+3)$ between $\mathrm{H}_{\alpha}-9 / \mathrm{NH}-15$; and NOE contacts characteristic of extended regions: strong $\mathrm{d}_{\alpha \mathrm{N}}(\mathrm{i}, \mathrm{i}+1)$ between $\mathrm{H}_{\alpha}$ -9/NH-10, $\mathrm{H}_{\alpha}-10 / \mathrm{NH}-14$, and $\mathrm{H}_{\alpha}-14 / \mathrm{NH}-15$ (Table S26, Supporting

Information). The apparently contradictory NOEs are indicative of the presence of at least two conformations in solution. The impossibility of resuming all the data in a single structure prompted us to consider incompatible NOEs separately in different calculation cycles (Experimental Section). Hence, we obtained two families of conformations. The first calculation cycle gave an ensemble of structures (family I) showing a similar conformation to compound 3, with a type II' $\beta$-turn spanning residues D - $\mathrm{Trp}^{8}-\mathrm{Lys}^{9}$, followed by a short $3_{10}$-helix along residues $\operatorname{Tyr}(\mathrm{Bzl})^{10}-\mathrm{dhDsa}-\mathrm{C}^{14}-\mathrm{Thr}(\mathrm{ol})^{15}$ (Figure 19a). Moreover, side chain orientations were the same as those described for 3. The main difference was a better definition of the $\mathrm{Phe}^{7}$ side chain which preferred the trans rotamer. For this set of structures, a number of consistent violations were observed (Table S26). In a second MD cycle, the violated upper limit constraints were upweighted for the contribution to the target function. Thus, a second conformational family (family II) was obtained which differed from the first mainly in that C-terminal residues were in extended conformations (Figure 19b). Furthermore, the side chain orientation of $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ was trans. Hence, the tyrosyl nucleus was further from the Lys ${ }^{9}$ side chain. This is in accordance with the down-field shifts of the $\mathrm{H}_{\gamma}$ resonances of $\mathrm{Lys}^{9}$ compared to the corresponding shifts of compound 3. Interestingly, the complete ensemble of structures (helix
and extended) fulfilled the NOE restraints, with no violations exceeding $0.5 \AA$ (Table S26).


Figure 19. Stereoview of the lowest energy conformer of compound 4: family I (a), family II (b). Backbone is evidenced as a ribbon. Side chains of the ten lowest energy conformers are also shown as mesh surface. Surfaces are distinguished with different colors. N-Term, N-terminus; C-Term, C-terminus.

Compound 5. Compound 5 maintains the same octreotide scaffold, but position 7, which bears 1-Nal. Apart from the dicarba bridge, it has the same peptide sequence of the analogue NOC,
formerly prepared and studied as DOTA-conjugate by Maecke, Reubi and co-workers.[159] The coupling constant $\left({ }^{3} J_{\mathrm{CH}=\mathrm{CH}}=8.1 \mathrm{~Hz}\right)$ between the two olefinic protons of the bridge and the relative strong NOE between the same olefinic $\mathrm{H}_{\gamma} \mathrm{s}$ established $(Z)$ configuration for compound 5. Only one isomer is obtained from RCM.

NMR-based structure calculations gave two conformational families, like compound 4. Family I, obtained by a first run of MD calculation, showed a type II' $\beta$-turn spanning residues $\mathrm{D}-\mathrm{Trp}^{8}-\mathrm{Lys}^{9}$, followed by a short $3_{10}$-helix along residues $\mathrm{Thr}^{10}-\mathrm{dhDsa}-\mathrm{C}^{14}-\operatorname{Thr}(\mathrm{ol})^{15}$ (Figure 20a). As found with compound 4, a number of consistent violations were observed (Table S27, Supporting Information). In a second MD run, we obtained a second conformational family (II), which differed from the first mainly because the C-terminal residues were in extended conformations (Figure 20b). In both the families, residue 7 showed a defined trans orientation which forces $1-\mathrm{Nal}^{7}$ naphthyl moiety close to $\mathrm{D}-\mathrm{Trp}^{8}$ residue. This orientation is in accordance with the intense up-field shift observed for many D-Trp ${ }^{8}$ proton resonances.


Figure 20. Stereoview of the lowest energy conformer of compound 5: family I (a), family II (b). Backbone is evidenced as a ribbon. Side chains of the ten lowest energy conformers are also shown as mesh surface. Surfaces are distinguished with different colors. N -Term, N -terminus; C-Term, C-terminus.

Compounds 6 and 7. The analogues 6 and 7 differ from compounds 3 and 4 in that the $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ residue was replaced by a Tyr (i.e. without the Bzl group). Following the same arguments given for $\mathbf{3}$ and 4, an $E$ configuration was assigned to compound 6 and a $Z$ configuration to the compound 7 at the double bond.

Many potential diagnostic NOEs could not be observed in the NOESY spectra of these analogues due to signal overlapping and this precluded structure calculations. For instance, $\mathrm{H}_{\alpha}$ protons of Lys ${ }^{9}$ and Thr-ol ${ }^{15}$ resonated at the same chemical shift for both peptides. Actually, the NMR parameters of $6\left(\mathrm{H}_{\alpha}\right.$ shifts, coupling constants, and temperature coefficients) are very similar to those of compound $\mathbf{3}$, and this was also true for 4 and 7. Therefore, it could be hypothesized that 3D structures should be similar too.


Figure 21. Stereoview of the lowest energy conformer of compound 8: family I (a), family II (b). Backbone is evidenced as a ribbon. Side chains of the ten lowest energy conformers are also shown as mesh surface. Surfaces are distinguished with different colors. N-Term, N-terminus; C-Term, C-terminus.

Compound 8. The analogue $\mathbf{8}$ structure was rationalized starting from the peptide sequence of the previous compounds $\mathbf{4}$ and 5 . In fact, it bears both $1-\mathrm{Nal}^{7}$ and $\operatorname{Tyr}(\mathrm{Bzl})^{10}$. For compound $\mathbf{8}$, a $Z$ configuration was established from the NOEs and coupling constant $\left({ }^{3} J_{\mathrm{CH}=\mathrm{CH}}=8.1\right.$ Hz ) between the two olefinic protons. NMR-based structure calculation (Table S28, Supporting Information), gave two - 112 -
conformational families, as it did for compounds 4 and 5, the first (family I) showing a short $3_{10}$-helix along the $\mathrm{Thr}^{10}-\mathrm{dhDsa}-\mathrm{C}^{14}-$ $\operatorname{Thr}(\mathrm{ol})^{15}$ residues (Figure 21a), and the second (family II) an extended conformation along the same residues (Figure 21b). In both families, the $\mathrm{D}-\operatorname{Trp}^{8}$, $\mathrm{Lys}^{9}$ and $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ side chains were spatially closed in accordance with the increased up-field shift of the $H_{\gamma}$ and $H_{\beta}$ resonances of Lys ${ }^{9}$. Differently from compound 5, $1-\mathrm{Nal}^{7}$ residue could not adopt a trans conformation, probably due to steric hindrance with $\operatorname{Tyr}(\mathrm{Bzl})^{10}$. In fact, $1-\mathrm{Nal}^{7}$ side chain was preferentially in a gauche conformation.

### 1.3.3 Discussion

In our ongoing efforts to develop new somatostatin ligands with improved stability and affinity towards sst receptors, we have rationally designed and analyzed a limited set of peptides (Table 9). In these peptides the labile disulfide bridge was replaced by a dicarbabridge, through the RCM reaction. As can be seen from Table 10, variation of the residues 7 and 10 results in analogues having a low sub- $\mu$ molar potency and a range of sst receptor subtype selectivities.

Recently, we have investigated some octreotide analogues, including compound $\mathbf{3}$, in a water/DMSO- $\mathrm{d}_{6} 8: 2$ solution.[156] Here, an NMR study was performed on the developed analogues of octreotide in SDS micelles solution (see section 1.1.3).

Some apparently contradictory NOEs were indicative of the presence of at least two conformations in solution for analogues $\mathbf{4 , 5}$, and 8. To deal with this incongruence we used a practical approach. Incompatible NOEs were considered separately in different calculation cycles. Hence, two families of conformations were obtained which differed mainly in that C-terminal residues were in $3_{10}$-helix (family I) or extended (family II) conformation. Eventually, the experimental restraints were fulfilled over the entire ensemble. It is noteworthy that the NMR data of the cognate molecule octreotide, using a single average conformation reveal several important
inconsistencies, including severe violations of mutually exclusive backbone-to-backbone NOEs.[160]

On the basis of the NMR results, some general conformationaffinity relationships concerning the binding to the sst receptors can be outlined. Similar to most of the bioactive analogues of SRIF reported so far,[161] the structures of the peptidomimetics presented here have a $\beta$-turn of type $\mathrm{II}^{\prime}$ spanning residues $\mathrm{D}-\mathrm{Trp}^{8}$ and $\mathrm{Lys}{ }^{9}$. The side chain of D-Trp ${ }^{8}$ is in the trans conformer, and the side chain of Lys ${ }^{9}$ is in the gauche conformer, bringing the two side chains adjacent to each other in close proximity. Analogues with the $Z$ configuration at the double bond can adopt both helical and extended structures at the C-terminus, showing a conformational equilibrium (Figures 19-21). As a consequence of this conformational behavior, $Z$ analogues show greater potency compared to the corresponding $E$ isomers ( $\mathbf{4} v s \mathbf{3}$ and 7 vs 6) although it can go to the detriment of the selectivity, as in the case of compound 4 compared to 3 . It could be argued, from the data of Table 10 , that $Z$-geometry of the double bond is a better mimic of the S-S bridge which, in turn, was hypothesized to be directly involved in the interaction with the sst receptors.[162, 163]

Analogues 3 and 4 bear a $\operatorname{Tyr}(\mathrm{Bzl})$ residue in position 10. The side chain of $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ was designed to replace Phe $^{6}$, Phe $^{7}$, and Phe ${ }^{11}$ of SRIF14.[164] Compound 3 ( $E$-isomer) selectively binds sst $_{5}$ while
its Z-isomer, 4, showed a pan-SRIF-activity, apart from sst $_{4}$. In compound $\mathbf{3}$, the $\beta$-turn motif is followed by a short $3_{10}$-helix along residues $\operatorname{Tyr}(\mathrm{Bzl})^{10}-\mathrm{dhDsa}-\mathrm{C}^{14}-\mathrm{Thr}(\mathrm{ol})^{15}$. The side chain of $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ is in the gauche conformer and is located in close proximity to the D -$\mathrm{Trp}^{8}-\mathrm{Lys}^{9}$ pair (Figure 18). This is in accordance with our recent results which correlates $\mathrm{sst}_{5}$ selectivity to conformationally restricted helical structure at the C-terminus.[156] The conformational properties of $\mathbf{3}$ in SDS micelles are similar to those observed in water/DMSO solution (data not shown). Only N-terminal residue DPhe ${ }^{2}$ is more flexible in the SDS solution compared to DMSO.

In addition to high-affinity binding to $\mathrm{sst}_{2,3,5}$ like octreotide (1), compound 4 also exhibited a low nanomolar binding to sst $_{1}$, hence its affinity pattern resembles that of the hexa-cyclic peptide SOM230 (pasireotide) which also bears a $\operatorname{Tyr}(\mathrm{bzl})^{10}$ residue.[164] Since compound 4 fits $4 / 5$ receptor binding sites, it was expected to display a high degree of flexibility. In fact, a dynamical equilibrium between extended and helical conformations was observed. Moreover, $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ side chain orientation was different in the two conformations (Figure 19). Notably, NMR [160] and X-ray crystallography analyses have already suggested an equilibrium between extended and folded conformational states for the parent peptide 1.[165] Furthermore, SOM230 exhibited similar backbone
conformational equilibrium in a theoretical MD study; the side chain of $\operatorname{Tyr}(\mathrm{Bzl})$ of SOM 230 underwent great flexibility which was associated with low selectivity.[166] Although sst $_{2}$ is probably the most abundantly expressed SRIF receptor in human cancer,[167] recent literature data indicates that also $\operatorname{sst}_{1}$ and sst $_{3-5}$ may also be present in some human tumors.[168] Hence, peptides with an improved receptor binding profile are desirable in order to extend the spectrum of tumors accessible to diagnosis and internal radiotherapy. As a matter of fact, SOM230 is being investigated in clinical trials as a potential treatment for acromegaly, neuroendocrine tumors and Cushing's disease.[169, 170]

Among the tested compounds, peptide 5 showed the highest affinity towards sst $_{2}$ and also a good affinity towards sst $_{5}$. The sequence of this cyclopeptide is the same as the $\mathrm{S}-\mathrm{S}$ bridged NOC,[159] whose DOTA derivative, DOTA-NOC, exhibited high affinity towards $\mathrm{sst}_{2,3,5}$. The loss of $\mathrm{sst}_{3}$ affinity in 5 is probably due to the absence of the D-Phe ${ }^{2}$-bonded DOTA chelating group. Insertion of different arms at the N -terminus may, in fact have, a dramatic effect particularly on $\mathrm{sst}_{3}$ affinity.[171] Compound 5 is also closely related to the previously described compound $\mathbf{2}$, sharing its configuration at the double bond and the amino acid sequence but with $\mathrm{Nal}^{7}$ in replacing $\mathrm{Phe}^{7}$.[155] The activity profile of analogues 2 and 5 is
similar, showing an increase in the sst $_{2}\left(\sim 4\right.$-fold) and sst ${ }_{5}$ affinity ( $\sim 2-$ fold) of 5 compared to 2 (Table 10). Since 2 showed similar conformational properties as 5 (data not shown), the improved affinity towards $\mathrm{sst}_{2}$ and $\mathrm{sst}_{5}$ is probably attributable to the $1-\mathrm{Nal}^{7}$ aromatic side chain which, oriented in a trans conformation, adequately fits the binding pocket of both receptors.

Compounds 6 and 7 are analogues to $\mathbf{3}$ and 4, respectively, with the $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ residue replaced by a tyrosine. This change renders compounds 6 and 7 strictly related to U-II. Analogue 6 showed a marked reduction of affinity towards $\mathrm{sst}_{5}$ compared to the correlated compound 3. Analogously, compound 7 showed a marked reduction of affinity towards $\mathrm{sst}_{1,3,5}$ and a 2-fold reduction towards $\mathrm{sst}_{2}$ compared to 4. NMR data of the analogue couples pointed to similar conformational behavior, hence it can be argued that the Tyr ${ }^{10}$ phenol group is detrimental for binding to the sst receptors. This is in accordance with the low affinity of U-II to the $\operatorname{sst}_{2 \mathrm{~A}}$ receptor.[172] On the other hand, residual affinity of compound 7 towards sst $_{2}$ and sst $_{5}$ (Table 10) parallels the capability of U-II to activate these two receptors at high doses.[173]

By combining $1-\mathrm{Nal}^{7}$ and $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ residue replacements, we obtained compound $\mathbf{8}$ as a pure $Z$-isomer. Compound $\mathbf{8}$ showed the highest affinity towards sst $_{5}$ (Table 10) with at least a 10 -fold
selectivity compared to the other ssts. Compound $\mathbf{8}$ is closely related to compound $\mathbf{4}$, sharing its configuration at the double bond and the amino acid sequence but with $1-\mathrm{Nal}^{7}$ replacing $\mathrm{Phe}^{7}$. Actually, the activity profile of the two analogues is similar (pan-SRIF-activity, apart from $\mathrm{sst}_{4}$ ) with a decrease of the $\mathrm{sst}_{1-3}$ affinity (2- to 4-fold) and increase of the $\mathrm{sst}_{5}$ activity ( $\sim 3$-fold) of $\mathbf{8}$ compared to 4 . The conformational behavior of $\mathbf{8}$ also resembles that of $\mathbf{4}$, in accordance to the activity similarity (Figure 19 and Figure 21). Since the helicalextended conformational equilibrium is also observable in the case of analogue 8 , the affinity changes could be tentatively attributed to the orientation of the $1-\mathrm{Nal}^{7}$ side chain which was differently oriented in the two peptides. In particular, it passed from a trans conformation observed in 4, to a gauche conformation in $\mathbf{8}$. Such gauche orientation of the naphthyl group is likely still suitable (or preferred) for $\mathrm{sst}_{5}$ but not for $\mathrm{sst}_{1-3}$ binding.

Based on the results reported above, we updated the previously proposed pharmacophore model for $\mathrm{sst}_{5}$-selective analogues.[156] The model involves the classical four side chains of the $\operatorname{sst}_{2 / 3 / 5}$ pharmacophore,[162] namely, those of residues D-Phe ${ }^{2}, \mathrm{Phe}^{7}\left(\mathrm{Nal}^{7}\right)$, D- $\operatorname{Trp}^{8}$ and $\mathrm{Lys}^{9}$, plus the $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ side chain. The distances between the $\mathrm{C}_{\gamma}$ atoms of these side chains, observed in the potent sst ${ }_{5}$ ligands 3-5, 8 are reported in Table 11. The $\mathrm{C}_{\gamma}-\mathrm{C}_{\gamma}$ distances found by

Melacini et al. for the $\operatorname{sst}_{2 / 3 / 5}$-selective SRIF analogues are also reported in the same table.[162] It can be observed that the distances found in our derivatives agree with the $\mathrm{sst}_{2 / 3 / 5}$ pharmacophore.

Table 11. $\mathrm{C} \gamma-\mathrm{C} \gamma$ distances $(\AA)$ between putative pharmacophoric residues. ${ }^{\text {a }}$

| Compd | 3 | 4 | 5 | 8 | SSt $_{2 / 3 / 5}{ }^{\text {b }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Ar}^{2}-\mathrm{Ar}^{7}$ | $8.5 \pm 0.8^{\text {c }}$ | $9.0 \pm 1.2$ | $9.5 \pm 0.5$ | $8.4 \pm 1.0$ | 5-11 |
| $\mathrm{Ar}^{2}-\mathrm{Ar}^{8}$ | $14.3 \pm 0.6$ | $14.0 \pm 0.5$ | $14.3 \pm 0.6$ | $13.4 \pm 0.5$ | 11-15 |
| $\mathrm{Ar}^{2}-\mathrm{Lys}{ }^{9}$ | $14.8 \pm 0.9$ | $14.4 \pm 1.0$ | $13.9 \pm 1.0$ | $14.5 \pm 1.0$ | 12-15 |
| $\mathrm{Ar}^{2}-\mathrm{Ar}^{10}$ | $8.1 \pm 1.1$ | $7.8 \pm 1.2$ | - | $13.4 \pm 1.0$ |  |
| $\mathrm{Ar}^{7}-\mathrm{Ar}^{8}$ | $7.8 \pm 0.8$ | $7.6 \pm 0.3$ | $6.8 \pm 0.2$ | $8.2 \pm 0.5$ | 7-9 |
| $\mathrm{Ar}^{7}$-Lys ${ }^{9}$ | $10.9 \pm 0.6$ | $9.8 \pm 0.7$ | $9.7 \pm 0.3$ | $11.0 \pm 0.6$ | 9-11 |
| $\mathrm{Ar}^{8}$-Lys ${ }^{9}$ | $5.5 \pm 0.2$ | $5.6 \pm 0.2$ | $4.7 \pm 0.3$ | $5.2 \pm 0.4$ | 5 |
| $\mathrm{Ar}^{8}-\mathrm{Ar}^{10}$ | $8.8 \pm 0.2$ | $8.9 \pm 0.2$ | - | $8.1 \pm 0.1$ | - |
| Lys ${ }^{9}-\mathrm{Ar}^{10}$ | $7.2 \pm 0.2$ | $7.2 \pm 0.1$ | - | $5.9 \pm 0.2$ | - |
| ${ }^{\text {a }}$ Only the family I of peptides $\mathbf{4 , 5 , 8}$ were considered. ${ }^{b}$ Pharmacophore for the sst $_{2}$, sst $_{3}$, sst ${ }_{5}$ selective SRIF analogues [162] ${ }^{\mathrm{c}}$ Average distance and standard deviation calculated from the ensemble of ten structures. |  |  |  |  |  |

### 1.3.4 Conclusions

A limited set of compounds of biostable SRIF analogues with dicarba bridge replacing the disulfide bridge of the parent octreotide (1) were prepared. Compounds were obtained by on-resin RCM by second generation Grubbs catalyst. All the analogues were tested for their affinity toward the sst $_{1-5}$ receptor subtypes. Among the synthesized compounds, derivative 4 exhibited a pan-somatostatin activity (except sst $_{4}$ ) and derivative $\mathbf{8}$ exhibited high affinity and selectivity towards sst $_{5}$. Actually, compound $\mathbf{8}$ had a similar sst $_{5}$ affinity ( $\mathrm{IC}_{50} 4.9 \mathrm{nM}$ ) to SRIF-28 and octreotide. Conformationaffinity relationships confirmed that helical propensity correlates with the peptide sst $_{5}$-affinity while a pan-SRIF activity is obtained by conformational equilibria. Both pan- and selective-SRIF analogues are potentially useful for the diagnosis and internal radiotherapy of tumors.

### 1.3.5 Experimental Section

General Procedures. Fmoc protected amino acids were purchased from Calbiochem-Novabiochem (Laufelfingen. Switzerland). $2^{\text {nd }}$ generation Grubbs catalyst was obtained from Aldrich. Fmoc-Hag, Fmoc-O-benzyl-L-tyrosine and $\mathrm{H}-1-\mathrm{Thr}(\mathrm{tBu})-$ ol-2-chlorotrityl resin were purchased from Iris Biotech (Marktredwitz, Germany). HATU was obtained from Chempep (Miami, USA). Peptide grade DMF was from Scharlau (Barcelona, Spain). All the other solvents and reagents used for SPPS were of analytical quality and used without further purification. Analytical RP-HPLCs were performed on a Waters instrument equipped with a UV detector on a Phenomenex Juppiter C18 column ( $5 \mu \mathrm{~m}, 250$ x 4.6 mm ) using a flow rate of $1 \mathrm{ml} / \mathrm{min}$, with the following solvent system: $0.1 \%$ TFA in $\mathrm{H}_{2} \mathrm{O}$ (A), $0.1 \%$ TFA in $\mathrm{MeCN}(\mathrm{B})$ ). Semi-preparative RP-HPLC analyses were performed on the same instrument using a flow rate of $4 \mathrm{ml} / \mathrm{min}$ with the same solvent system, on a Phenomenex Juppiter C18 column (10 $\mu \mathrm{m}, 250 \mathrm{x} 10 \mathrm{~mm}$ ). Mass spectra were registered on an ESI LCQ Advantage mass spectrometer (ThermoFinnigan). LC-ESI-MS analyses were performed on a Phenomenex Juppiter C18 column ( $5 \mu \mathrm{~m}, 150 \times 2.0 \mathrm{~mm}$ ) using a flow rate of 500 $\mu \mathrm{L} / \mathrm{min}$ on a ThermoFinnigan Surveyor HPLC system coupled to ESIMS, using the solvent system: $\mathrm{H}_{2} \mathrm{O}(\mathrm{A}), \mathrm{MeCN}(\mathrm{B}), 1 \%$ TFA in $\mathrm{H}_{2} \mathrm{O}$
(C). Routine NMR spectra were acquired on a Varian Inova 700 apparatus. TSP was purchased from MSD Isotopes (Montreal, Canada). ${ }^{2} \mathrm{H}_{2} \mathrm{O}$ was obtained from Aldrich. SDS- $d_{25}$ was obtained from Cambridge Isotope Laboratories, Inc. (Andover, MA). SPPS was performed in Teflon reactor on a manual synthesizer PLS 4 x 4 (AdvancedChemTech). Receptor autoradiography was performed on $20-\mu \mathrm{m}$ thick cryostat (Microm HM 500, Walldorf, Germany).

Synthesis and Purification of Compounds 3-8. Peptides were synthesized following the method reported in the preceding work.[156] Briefly, the peptides were prepared using the general Fmoc-SPPS strategy on pre-swelled $\mathrm{H}-\mathrm{L}-\mathrm{Thr}(t \mathrm{Bu})-\mathrm{ol}-2-$ chlorotrityl resin. Couplings were performed by adding two equivalents of protected amino acid activated by HATU and four equivalents of NMM in DMF. Each coupling was monitored by the qualitative ninhydrin (Kaiser) test.[35] At the end of the linear peptides synthesis, a microscale cleavage was performed. RP-HPLC analysis of the crude products revealed the presence of the linear peptides in approximately $95 \%$ purity, without traces of isomers due to amino acid racemization. The cyclization was performed on-resin by $2^{\text {nd }}$ generation Grubbs catalyst ( 0.5 mole equiv. calculated on the basis of $0.5 \mathrm{mmol} / \mathrm{g}$ of peptide). After swelling, $\mathrm{NH}_{2}$ terminal Fmoc--Hag was deprotected and coupled with Fmoc-D-Phe affording the on-resin peptides $\mathbf{4 - 8}$
which were deprotected and cleaved $[5,6$ and 7 with $\mathrm{TFA} / \mathrm{H}_{2} \mathrm{O} / \mathrm{EDT} /$ phenol $(94: 2: 2: 2,3 \mathrm{~h})$ while 4 and $\mathbf{8}$ with TFA/ $\mathrm{H}_{2} \mathrm{O} / \mathrm{EDT} /$ phenol (70:26:2:2, 2.30 h )]. The aqueous solutions of the peptides $\mathbf{4 - 8}$ were pre-purified by SPE, and after subjected to the purification by semi-preparative RP-HPLC and subsequently characterized by ESI-MS. Analytical RP-HPLC and ESI-MS analysis of the crude compounds revealed two chromatographic peaks with the same MW for compounds 4, -7, corresponding to the geometric isomers $(Z / E$ ratio $\approx 90: 10)$. Compounds were then purified by semipreparative RP-HPLC and the most abundant chromatographic peaks were collected. For all the products HPLC purity was $\geq 97 \%$.. Further experimental data are reported in the Supporting Information.

NMR Spectroscopy. The samples for NMR spectroscopy were prepared by dissolving the appropriate amount of peptide in 0.55 ml of ${ }^{1} \mathrm{H}_{2} \mathrm{O}(\mathrm{pH} 5), 0.05 \mathrm{ml}$ of ${ }^{2} \mathrm{H}_{2} \mathrm{O}$ to obtain a concentration $1-2 \mathrm{mM}$ of peptides and 200 mM of $\operatorname{SDS}-\mathrm{d}_{25}$. TSP was used as internal chemical shift standard. The water signal was suppressed by gradient echo.[68] NMR experiments were recorded on a Varian Inova-Unity 700 MHz at 308.1 K. Complete ${ }^{1} \mathrm{H}$ NMR chemical shift assignments were effectively achieved for all the analyzed peptides (Supporting Information, Tables S19-S24) according to the Wüthrich procedure [36] via the usual systematic application of TOCSY [39] and NOESY
[40] experiments recorded in the phase-sensitive mode using the method from States.[70]

Typical data block sizes were 2048 addresses in $t_{2}$ and 512 equidistant $t_{1}$ values. Before Fourier transformation, the time domain data matrices were multiplied by shifted $\sin ^{2}$ functions in both dimensions. A mixing time of 70 ms were used for the TOCSY experiments. NOESY experiments were run with mixing times of 100 and 200 ms . The qualitative and quantitative analyses of TOCSY and NOESY spectra were obtained with the support of the XEASY software package.[41]

Structural Determinations and Computational Modeling. The NOE-based distance restraints were obtained from NOESY spectra collected with the mixing time of 100 ms . The NOE cross peaks were integrated with the XEASY program and were converted into upper distance bounds using the CALIBA program incorporated into the program package DYANA.[71] Only NOE derived constraints (Supporting Information, Tables S25-S28) were considered in the annealing procedures. In a first calculation run, all the upper distance bounds were used, generating an ensemble of 100 structures with the simulated annealing standard protocol of the program DYANA. For peptides $\mathbf{4}, \mathbf{5}$, and $\mathbf{8}$, a number of consistent (i.e. in all calculated structures) violated upper limit constraints ( $>0.1 \AA$ ) were observed
(Supporting Information, Tables S25-S28). These violations were discarded in a subsequent MD run. This step was repeated till no violation was observed (two runs were enough for all peptides). Thus, we obtained a first family of structures (family I). In a second MD cycle, the violated upper limit constraints of the first cycle were upweighted (10-fold) for the contribution to the target energy function of DYANA. Hence, we obtained a new set of violated constraints which were discarded in the subsequent MD runs. After two MD runs, no violations were observed. In the final calculation run, we applied the same weight to the undiscarded constraints and obtained a second family of structures (family II). Since, the two sets of violations had no common member we did not repeat further the described procedure.

Finally, 20 structures for peptide 3, and 20 structures for each family of peptides $\mathbf{4}, 5$ and $\mathbf{8}$ were chosen, whose interprotonic distances best fitted NOE derived distances, and then refined through successive steps of restrained and unrestrained energy minimization calculations using the Discover algorithm (Accelrys, San Diego, CA) and the consistent valence force field (CVFF).[73]

The minimization lowered the total energy of the structures. The final structures were analyzed using the InsightII program (Accelrys, San Diego, CA). Graphical representation were carried out with the

UCSF Chimera package.[76] The root-mean-squared-deviation analysis between energy-minimized structures were carried out with the program MOLMOL.[72]

Determination of Somatostatin Receptor Affinity Profiles. Cell membrane pellets were prepared from human sst $_{1}$-expressing CHO cells, sst $_{2}{ }^{-}$, sst $_{3}{ }^{-}$, sst $_{4}$-expressing CCL39 cells and sst $_{5}$-expressing HEK293 cells and stored at $-80^{\circ} \mathrm{C}$. Receptor autoradiography was performed on $20-\mu \mathrm{m}$ thick cryostat (Microm HM 500, Walldorf, Germany) sections of the membrane pellets, mounted on microscope slides, and then stored at $-20^{\circ} \mathrm{C}$ as previously described.[174, 175] For each of the tested compounds, complete displacement experiments with the universal SRIF radioligand [Leu ${ }^{8}$, D-Trp $\left.{ }^{22},{ }^{125} \mathrm{I}-\mathrm{Tyr}^{25}\right]$-SRIF$28\left({ }^{125}\right.$ I-[LTT]-SRIF-28) (2,000 $\mathrm{Ci} / \mathrm{mmol} ; ~ A n a w a, ~ W a n g e n$, Switzerland) using $15,000 \mathrm{cpm} / 100 \mu \mathrm{~L}$ and increasing concentrations of the unlabelled peptide ranging from $0.1-1000 \mathrm{nM}$ were performed. As control, unlabelled SRIF-28 was run in parallel using the same increasing concentrations. The sections were incubated with ${ }^{125}$ I-[LTT]-SRIF-28 for 2 hours at room temperature in $170 \mathrm{mmol} / \mathrm{L}$ Tris- HCl buffer ( pH 8.2 ), containing $1 \% \mathrm{BSA}, 40 \mathrm{mg} / \mathrm{L}$ bacitracin, and $10 \mathrm{mmol} / \mathrm{L} \mathrm{MgCl}_{2}$ to inhibit endogenous proteases. The incubated sections were washed twice for 5 min in cold $170 \mathrm{mmol} / \mathrm{L}$ Tris- HCl ( pH 8.2 ) containing $0.25 \%$ BSA. After a brief dip in $170 \mathrm{mmol} / \mathrm{L}$ Tris-
$\mathrm{HCl}(\mathrm{pH} 8.2)$, the sections were dried quickly and exposed for 1 week to Kodak BioMax MR film. $\mathrm{IC}_{50}$ values were calculated after quantification of the data using a computer-assisted image processing system as described previously.[175] Tissue standards (Autoradiographic $\left[{ }^{125} \mathrm{I}\right]$ and/or $\left[{ }^{[14} \mathrm{C}\right]$ microscales, GE Healthcare, Little Chalfont, UK) that contain known amounts of isotope, crosscalibrated to tissue-equivalent ligand concentrations were used for quantification.[146]

## Chapter 2 -LIGAND-RECEPTOR INTERACTIONS FROM NMR SPECTROSCOPY. APPLICATION TO CYTOTOXIC AGENTS BINDING TO DNA

# 2.1 Design, Synthesis, and Cytotoxic Evaluation of Acyl Derivatives of 3-Aminonaphtho[2,3-b]thiophene-4,9-dione, a Quinone-Based System 

A series of 3-acyl derivatives of dihydronaphtho[2,3-b]thiophen-4,9-dione system were studied with respect to cytotoxicity and topoisomerase II inhibitory activity. These analogues were designed as electron-deficient anthraquinone analogues with potential intercalation ability. Derivatives 3-(diethylamino)-N-(4,9-dioxo-4,9-dihydronaphtho[2,3-b]thiophen-3-yl)propanamide (11m) and 3-(2-(dimethylamino)ethylamino)-N-(4,9-dioxo-4,9-dihydronaphtho[2,3-b]thiophen-3-yl) propanamide (11p) showed a high efficacy in cell lines that were highly resistant to treatment with doxorubicin, such as MDA-MB435 (melanoma), IGROV (ovarian), SF-295 (glioblastoma) human cell lines. Both compounds inhibit topoisomerase II mediated relaxation of DNA, while only 11p incites arrest at $S$ phase in Caco-2 cells, inducing a delay of cell cycle progression and an increase of cell differentiation. The ability of these derivatives of modulate small heat shock proteins and the cardiotoxicy effects were also explored. In addition, DNA-binding properties of these compounds were investigated and discussed.

### 2.1.1 Introduction

Anthracyclines are among the most effective and useful anticancer agents developed, and they are used to treat more types of cancer than any other chemotherapy agent.[176, 177] Their clinical importance has stimulated wide research [178-181] directed to the development of new structurally related compounds with the goal of bypassing significant problems that limit their utility, such as their failure in resistant tumors expressing the ABCB1 (MDR1) gene [182184] and the emergence of severe short- and long-term side effects associated with bone marrow and myocardial cell toxicity.[185, 186] With this aim, our research group has developed different series of quinone-based compounds containing the 3-amino-3-(ethoxycarbonyl)-2,3-dihydrothieno[2,3-b]naphtho-4,9-dione system (4, DTNQ) as chromophore (Figure 22).[187] The effected modifications on this template and the analysis of the structureactivity relationship (SAR) on the different synthesized series showed that the incorporation of a distal protonated alkyl amine linked to chromophore DTNQ system through a five- or six-membered heterocycle or the presence of a cycloalkyl as the fifth ring were effective approaches to identify new compounds endowed with potent cytotoxic activity, and able to overcome multidrug resistance of tumor cells. Thus, the 3-glycyl-amino-3-(ethoxycarbonyl)-2,3-
dihydrothieno[2,3-b]naphtho-4,9-dione (5),[188] the spirohydantoin derivatives 3 -[2-( $N, N$-dimethylamino)ethyl or propyl]-spiro[(dihydroimidazo-2,4-dione)-5,3'-(2',3'-dihydrothieno[2,3$b]$ naphtho $-4^{\prime}, 9^{\prime}$-dione)] (6a,b) [189] as well as the spirodiketopiperazine derivatives $4-[(2-N, N-$ dimethyl)amino]ethylspiro[(dihydropirazin-2,5-dione)-6,3'-(2', $3^{\prime}-$ dihydrothieno[2,3-b]naphtho-4',9'-dione) (7) [190] and spiro [(hexahydropyrrolo[1,2-a]pyrazine-1,4-dione)-6,3'-(2', $3^{\prime}-$ dihydrothieno[2,3-b]naphtho-4',9'-dione)] (8) [191] showed remarkable cytotoxic activity against several solid tumors and doxorubicin- and cis-platinum-resistant human cell lines.


Figure 22. Structure of some DTNQ derivatives and the new TNQ system.

In addition, STD-NMR spectroscopy investigation performed on compounds $\mathbf{7}$ and $\mathbf{8}$ demonstrated that these derivatives interact with DNA with a dual binding mode: intercalative for the dihydrothieno[2,3-b]naphtho-4,9-dionetricyclic core and external considering the side-chain moiety.[190, 191] However, even though these derivatives had many of the structural characteristics of classical quinone-based DNA intercalating agents, they were not able to inhibit topoisomerase II (topo II) at equicytotoxic concentrations, indicating that other factors such as differences in cellular uptake, distribution within the cell, and additional targets within the cell might also affect the cytotoxicity of these derivatives.[192]

Now we have considered the possibility of using a new DTNQ derivative, the 3-aminonaphtho[2,3-b]thiophene-4,9-dione (9, TNQ) recently synthesized in our laboratories,[193] as a more planar chromophore. This quinone-based amine system showed interesting cytotoxic activity toward the MCF-7 human breast carcinoma $\left(\mathrm{IC}_{50}=\right.$ $3.2 \mu \mathrm{M})$ and SW 620 human colon carcinoma cell lines $\left(\mathrm{IC}_{50}=4.0\right.$ $\mu \mathrm{M})$ indicating its potential as a template in the development of efficient cytotoxic agents. The new system presents a more "planar core" compared to initial DTNQ structure and an amine group able to be functionalized with appropriate side chain in a defined orientation with respect to the chromophore, thus guaranteeing two of the main
structural requisites for the antineoplastic activity of intercalating agents. According to literature data, among heterocyclic quinones endowed with cytotoxic activity, those containing a thiophene nucleus fused to a quinone system have received little attention, despite the antitumoral activity of thiophene analogues of daunomycin and mitoxantrone described by the work groups of Kita [194] and Krapcho,[195] respectively.

Thus, we developed a series of 3-substituted-aminonaphtho[2,3-b]thiophene-4,9-dione derivatives in which the amine group of the planar chromophore (TNQ) was linked to several amino acids (Gly, Ala, Phe, Lys, Pro, $\beta$-Ala), substituted-alkylcarbonyl chains (hydroxyacetyl, hydroxypropionyl, ( $N, N$-diethyl)aminoacetyl, ( $N, N$ diethyl)aminopropionyl, 2-morpholinacetyl, 3-morpholinpropionyl, ( $N$ ', $N$ '-methyl)( $N$-aminoethyl)-aminopropionyl, thioacetyl, thiopropionyl) and carbamoyl chains (propyl, aminoethyl), which represent the side chain functionalities of the more active compounds of the precedent series. The objectives of this investigation are: a) validation of TNQ system as template in the development of new quinone-based antitumoral agents exploring so new chemical spaces; b) identification of the structural parameters which are important for the cytotoxic activity, through a comparative study of the structure activity relationships (SARs) of TNQ derivatives; and c) exploration
of the basic biochemical events correlated to cytotoxic activity of new derivatives. The present work deals with the preliminary studies concerning the synthesis of novel TNQ derivatives, the cytotoxic activity, the interaction with topo II and DNA, and their influence on cell cycle progression.

### 2.1.2 Results and Discussion

Chemistry. The synthetic approach to new 3-substitutedaminonaphtho $[2,3-b]$ thiophene-4,9-dione derivatives was based on the capacity of DTNQ system and its 3-N-acyl derivatives to undergo oxidative decarboxylation in hydrolytic basic way, as we recently described.[193] Condensation of 3-amino-3-ethoxycarbonyl-2,3-dihydrothieno[2,3-b]naphtho-4,9-dione (4, DTNQ) with different Bocamino acids $(\mathbf{a}=$ Gly, $\mathbf{b}=$ Ala, $\mathbf{c}=$ Phe, $\mathbf{d}=\operatorname{Lys}, \mathbf{e}=\operatorname{Pro}, \mathbf{f}=\beta$-Ala,$)$ using HBTU, HOBt, and DIPEA in DMF afforded, with high yields (50-65\%), the appropriate pseudodipeptide intermediates $\mathbf{1 0} \mathbf{\prime} \mathbf{a - f}$, as shown in Figure 23. Treatment with DBU in $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ medium gave directly the corresponding decarboxylated intermediates 11' a-f in $76-82 \%$ yields. Finally, after removal of the Boc protecting group using 20\% TFA in dichloromethane and triethylsilane as scavanger, the final compounds 11a-f were obtained as trifluoroacetate salts in 40-48 \% overall yields.


Figure 23. Reagents and conditions: i) Boc-Aaa-OH HBTU, HOBt, DIPEA in DMF, room temperature ii) DBU in $\mathrm{MeOH} / \mathrm{H} 20$, room temperature, iii) TFA/DCM, TES.

Two homologue series of compounds containing a linear substituted-alkyl chain were synthesized from 3-(2'-chloro)acetamide-3-ethoxycarbonyl-2,3-dihydrothieno[2,3-b]naphtho-4,9-dione (12) and 3-(acrylamido)-3-ethoxycarbonyl-2,3-dihydrothieno[2,3-b]naphtho-4,9-dione (13) respectively, followed a similar methodology (Figure 24).









h, $\mathbf{m}: \mathbf{R}=\mathrm{NH}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{2}$
i,n: $\mathrm{R}=\mathrm{STrt} \xrightarrow{\mathrm{V}} \mathrm{SH}$
k, $\mathbf{p}: \mathrm{R}=\mathrm{HN}$


Figure 24. Reagents and conditions: i) chloroacetyl chloride, TEA in THF; ii) bromopropionyl chloride, TEA in THF, iii) DBU in $\mathrm{MeOH} / \mathrm{H} 2 \mathrm{O}$, room temperature; iv) Nucleophilic reagents in THF, TEA, reflux temperature; v) then, for $\mathbf{1 1 i}$ and $\mathbf{1 1 n} \mathbf{2 0 \%}$ TFA in dichloromethane; vi) then, for $\mathbf{1 1 j}$ and $\mathbf{1 1 k}, \mathbf{1 1 m}$, 110, and $\mathbf{1 1 p} \mathrm{HCl}(\mathrm{g}) /$ diethyl ether solution.

Condensation of 4 with chloroacethyl chloride in THF, using triethylamine as base, afforded the ( $2^{\prime}$-chloro) acetamide derivative $\mathbf{1 2}$ with $92 \%$ yield. Under these conditions, the reaction of 4 with bromopropionyl chloride gave the 3-bromopropionamide intermediate
( $90 \%$ yield), which partially evolved to $\beta$-elimination product 3-(acrylamido)-3-ethoxycarbonyl-2,3-dihydrothieno[2,3-b] naphtho-4,9dione (13), during work-up of reaction. Decarboxylation performed on 12 and 13 intermediates gave directly the 2-hydroxyacetamide ( $\mathbf{1 1 g} \mathbf{g})$ and 3-hydroxypropyonamide (111) as final compound, respectively. Nucleophilic displacement of the chlorine atom (12) or Michael-type addiction to acrylamido moiety (13) using diethylamine, triphenylmethanethiol, morpholine or $N, N$-diethylethylendiamine, in THF and triethylamine at reflux, readily provided the corresponding acetamide ( $\mathbf{1 2}^{\mathbf{\prime}} \mathbf{h - k}$ ) or propionamide ( $\mathbf{1 3}^{\prime} \mathbf{m} \mathbf{- p}$ ) analogues. Basic hydrolysis of these derivatives afforded the corresponding decarboxylated compounds ( $\mathbf{1 1 \mathbf { h }} \mathbf{- j}$ and $\mathbf{1 1 m} \mathbf{- p}$ ), except in the case of $\mathbf{1 2}^{\prime} \mathbf{k}\left(\mathrm{R}=\mathrm{HNCH}_{2} \mathrm{CH}_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right)$. In fact, under the cited conditions, this intermediate gave the cyclic derivative $4-[(2-N, N-$ dimethyl)amino]ethylspiro[(dihydropirazin-2,5-dione)-6,3'-(2', $3^{\prime}-$ dihydrothieno[2,3-b]naphtho-4',9'-dione)]
(7) previously described.[190] Then, final compounds presenting an amine functionality $\mathbf{1 1 j}, \mathbf{1 1 k}, \mathbf{1 1 m}, \mathbf{1 1 0}$, and $\mathbf{1 1 p}$, were treated with a solution of gaseous hydrochloric acid in diethyl ether to provide corresponding hydrochloride salts. This was found to both aid purification, and provide an improved solubility profile for the biological assays. The final thioacetamide 11i and thiopropyonamide

11n derivatives were obtained after S-Trt deprotection using 20\% TFA in dichloromethane in quantitative yields.

For the synthesis of compound $\mathbf{1 1 k}$ and the urea-based derivatives $\mathbf{1 1 q}$ and $\mathbf{1 1 r}$ we chose an alternative route which implied the use of 3-amino-naptho[2,3-b]thiophene-4.9-dione (TNQ, 9) as starting material (Figure 25). The condensation of 9, obtained after deprotection of corresponding N-Boc TNQ (14) using 50\% TFA in dichloromethane,[193] with chloroacethyl chloride afforded the ( $2^{\prime}-$ chloro)acetamide intermediate 15 ( $88 \%$ yield). Reaction of 15 with diethylamine in THF and triethylamine at reflux, afforded the final derivative $\mathbf{1 1 k}$. Compounds $\mathbf{1 1 q}$ and $\mathbf{1 1 r}$ were obtained by treatment of 9 with triphosgene and TEA in THF followed by addiction of propylamine or $N, N$-dimethylethylendiamine. Also in this case, the use of 9 as starting material was necessary since the corresponding N carbamoyl derivatives of DTNQ (compounds 6') evolved rapidly to spirohydantoin derivatives 6 under hydrolitic conditions.[189]



Figure 25. Reagents and conditions: i) (Boc)2O, ii) DBU in $\mathrm{MeOH} / \mathrm{H} 2 \mathrm{O}$, room temperature; iii) $50 \%$ TFA/dichloromethane; iv) chloroacetyl chloride, TEA in THF; v) (N,N-dimethyl) ethylenediamine in THF, TEA, reflux temperature; vi) triphosgen, TEA, THF, room temperature, 10 min , then $\mathrm{R}-\mathrm{NH}_{2}$.

In Vitro Cytotoxicity. TNQ derivatives were first examined for antiproliferative activity against the MDA231 human breast carcinoma, SW 620 human colon carcinoma, and U937 human leukemic monocyte lymphoma cell lines, and the obtained $\mathrm{IC}_{50}$ values are summarized in Table 12. For comparative purposes, the template 9 and doxorubicin were also included in the assay.

Results in Table 12 confirmed the compound 9 as potential scaffold of new antitumoral agents with a cytotoxic activity into the micromolar range on three cell lines used in the assay. The improved antitumor activity and spectra of some of the newly synthesized
compounds, compared to $\mathbf{9}$, demonstrated that chemical modification at C-3 was an effective approach to optimize the activity profiles of TNQ moiety. The wide activity range observed for compounds 11a$11 \mathbf{r}\left(\mathrm{IC}_{50}\right.$ from 0.6 to $\left.>40 \mu \mathrm{M}\right)$ indicated that the nature of substituents on amine group at C-3 position markedly affects the activity profile of these compounds. Incorporation through the 3-amino group of different amino acids was well tolerated in the case of linear amino acids such as glycine (11a). The presence of amino acids containing alkyl (Ala, 11b) or benzyl (Phe, 11c) side chain, relatively more rigid and more electron rich when compared to non substituted side chain, led to significant loss of activity, especially in the MDA231 cell line. This negative effect was more noteworthy with the introduction of an alkyl amino side chain (Lys, 11d). The incorporation of Pro gave the derivative 11e, which turned out to be the most active in the leukemic cell line $\left(\mathrm{IC}_{50}=0.9 \mu \mathrm{M}\right)$.

Table 12. Cytotoxic activities of 3-(amino)naphtho[2,3-b]thiophene-4,9-dione (9) and 3-[(acyl)amino]naphtho[2,3-b]thiophene-4,9-dione derivatives (11 a-f).


9


11

| compd | R | $\mathrm{IC}_{50}(\mathrm{M}) \pm \mathrm{SD}^{\text {a }}$ |  |  | TopoII-activity ${ }^{\text {e }}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | MDA231 ${ }^{\text {b }}$ | SW620 ${ }^{\text {c }}$ | U937 ${ }^{\text {d }}$ | $5 \mu \mathrm{M}$ | $10 \mu \mathrm{M}$ |
| 9 |  | $11.3 \pm 0.4$ | $4.0 \pm 0.3$ | $10.1 \pm 0.4$ |  |  |
| 11a | $\mathrm{CH}_{2} \mathrm{NH}_{2}{ }^{\text {f }}$ | $6.2 \pm 4.6$ | $2.3 \pm 0.4$ | $7.0 \pm 0.07$ |  |  |
| 11b | $\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{NH}_{2}{ }^{\text {f }}$ | $>40$ | $12.4 \pm 1.5$ | $9.1 \pm 0.2$ |  |  |
| 11c | $\mathrm{CH}\left[\mathrm{CH}_{2}\left(\mathrm{C}_{6} \mathrm{H}_{5}\right) \mathrm{NH}_{2}{ }^{\text {f }}\right.$ | $>40$ | $30.50 \pm 6.4$ | >40 | + | + |
| 11d | $\mathrm{CH}\left[\left(\mathrm{CH}_{2}\right)_{4} \mathrm{NH}_{2}\right] \mathrm{NH}_{2}{ }^{\text {f }}$ | $>40$ | >40 | $20 \pm 0.01$ |  |  |
| 11e | 2-pyrrolidiny ${ }^{\text {f }}$ | $6.7 \pm 2.5$ | $5.4 \pm 0.1$ | $0.9 \pm 0.06$ |  |  |
| 11f | $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{NH}_{2}{ }^{\text {f }}$ | $3.7 \pm 0.9$ | $0.8 \pm 0.27$ | $1.7 \pm 0.01$ |  |  |
| 11g | $\mathrm{CH}_{2} \mathrm{OH}$ | $10.1 \pm 0.2$ | $18.5 \pm 0.7$ | $15.1 \pm 0.06$ |  |  |
| 11h | $\mathrm{CH}_{2} \mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{2}{ }^{\text {g }}$ | $8.5 \pm 0.12$ | $4.0 \pm 0.14$ | $5.1 \pm 0.07$ |  |  |
| 11i | $\mathrm{CH}_{2} \mathrm{SH}$ | $13.6 \pm 0.15$ | $20.9 \pm 0.16$ | $30.1 \pm 0.04$ |  |  |
| 11j | $\mathrm{CH}_{2}$-morpholine ${ }^{\text {g }}$ | $7.1 \pm 0.2$ | $10.8 \pm 0.12$ | $4.3 \pm 0.02$ |  |  |
| 11k | $\mathrm{CH}_{2} \mathrm{NH}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}{ }^{\text {g }}$ | $4.9 \pm 0.4$ | $2.1 \pm 0.3$ | $4.0 \pm 0.03$ |  |  |
| 111 | $\left(\mathrm{CH}_{2}\right)_{2} \mathrm{OH}$ | $9.2 \pm 0.6$ | $20.3 \pm 0.8$ | $15 \pm 0.06$ |  |  |
| 11m | $\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{2}{ }^{\mathrm{g}}$ | $2.5 \pm 0.1$ | $1.5 \pm 0.2$ | $1.1 \pm 0.01$ | + | +++ |
| 11n | $\left(\mathrm{CH}_{2}\right)_{2} \mathrm{SH}$ | $15.2 \pm 0.1$ | $20.7 \pm 0.8$ | $23.9 \pm 0.34$ |  |  |
| 110 | $\left(\mathrm{CH}_{2}\right)_{2}$-morpholine ${ }^{\mathrm{g}}$ | $10.1 \pm 1.3$ | $20.1 \pm 0.3$ | $7.2 \pm 0.07$ |  |  |
| 11p | $\left(\mathrm{CH}_{2}\right)_{2} \mathrm{NH}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}{ }^{\text {g }}$ | $2.0 \pm 0.1$ | $0.6 \pm 0.08$ | $1.3 \pm 0.03$ | ++ | +++ |
| 11q | $\mathrm{NH}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{3}{ }^{\text {g }}$ | $9.5 \pm 0.52$ | $6.5 \pm 1.20$ | $10.1 \pm 0.50$ |  |  |
| 11r | $\mathrm{NH}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}{ }^{\mathrm{g}}$ | $8.7 \pm 0.30$ | $5.9 \pm 0.20$ | $9.8 \pm 0.20$ |  |  |
|  | Doxorubicin | $1.13 \pm 0.01$ | $0.12 \pm 0.01$ | $0.93 \pm 0.01$ | 0 | 0 |

${ }^{a}$ Data represent mean values (SD) for three independent determinations. ${ }^{b}$ Human melanoma cell line. ${ }^{\text {c }}$ Human colon carcinoma cell line. ${ }^{\text {d }}$ Human leukemic monocyte lymphoma cell line. ${ }^{\text {e }}$ The semiquantitative evaluation of TopoIImediated DNA relaxation activity was as follows: +++ , high; ++, intermediate; +, low; 0, absent. All the rest compounds were not tested. ${ }^{\text {f }}$ Evaluated as TFA salts. ${ }^{\text {g }}$ Evaluated as HCl salts.

Other interesting results were obtained with the incorporation of a primary or tertiary amine to the end of the ethyl side chains. Compounds 11f, 11m, and 11p retained cytotoxic levels similar to those of doxorubicin on the SW 620 cell line, with $\mathrm{IC}_{50}$ values of 0.8 , 1.5 , and $0.6 \mu \mathrm{M}$ respectively, and maintained the activity on the

MDA231 and U937 cell lines within the micromolar range (2.0-3.7 $\mu \mathrm{M}$ and 1.1-1.7 $\mu \mathrm{M}$, respectively). These derivatives were 2-5 fold more potent than their methylene homologues (11a, 11h, and $\mathbf{1 1 k}$, respectively) on all the cell lines. Congeners with a hydroxyl (compounds 11 g and 111), thiol (compounds 11 i and 11n) or morpholin (compounds 11j and 110) groups were remarkably less potent compared to their primary and tertiary amine analogues.

Finally, the incorporation of an alkyl or alkylamino side chain through an ureide group led to a decrease of the activity in the resultant analogues $\mathbf{1 1 q}$ and $\mathbf{1 1 r}$, respectively. These results imply a minor tolerance to structural modifications in this series compared to precedent series.

To further determine the antitumor spectra, the most potent compounds 11f, 11m, and 11p were selected and screened against a panel of human tumor cell lines, including MDA-MB435 and SKMEL 28 (melanoma), IGROV (ovarian), SF-295 and SNB-19 (glioblastoma), and Colo205, HT-29, and undifferentiated Caco-2 (colon). Differentiated Caco-2, a well accepted model of normal cell line due to its ability to acquire the phenotype of mature smallintestinal cell,[196, 197] was utilized to characterize a safety profile of the compounds at least in terms of "cell-selectivity".[198, 199]

As observed in Table 13, selected compounds were more potent than doxorubicin on the melanoma, colon and CNS human tumor cell lines, with $\mathrm{IC}_{50}$ values in the range $0.1-1.0 \mu \mathrm{M}$. Compounds $\mathbf{1 1 m}$ and 11p turned out to be the most active derivatives against SK-MEL 28 human melanoma cell line $\left(\mathrm{IC}_{50}=0.6\right.$ and $0.3 \mu \mathrm{M}$, respectively $)$ and were equipotent to doxorubicin $\left(\mathrm{IC}_{50}=0.4 \mu \mathrm{M}\right)$. Analogously to that observed in the previously described series,[189-191] these compounds showed a remarkable activity against tumoral cell lines generally highly resistant to treatment with doxorubicin. Compounds 11m and 11p presented a cytotoxic activity in the micromolar range against undifferentiated Caco-2 tumoral colon cell lines $\left(\mathrm{IC}_{50}=0.8\right.$ $1.0 \mu \mathrm{M})$, while showed to be 4 -fold less active $\left(\mathrm{IC}_{50}=3.8-4.1 \mu \mathrm{M}\right)$ on differentiated Caco-2 cell line. These data indicated a good profile of cell-selectivity for our derivatives (Selectivity Index (SI) $\sim 0.22$ ) especially if they are compared with the high toxicity data obtained with doxorubicin $(\mathrm{SI}=11.1)$.

Table 13. Inhibition of multiple human tumor cell lines by selected compound.

| Origin of tumor | Cell line | $\mathrm{IC}_{50}(\mu \mathrm{M}) \pm \mathrm{SD}^{\mathrm{a}}$ |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
|  |  | $\mathbf{1 1 f}$ | $\mathbf{1 1 m}$ | $\mathbf{1 1 p}$ | Doxorubicin |
| Melanoma | MDA-MB435 | $0.4 \pm 0.10$ | $0.5 \pm 0.08$ | $0.5 \pm 0.09$ | $1.3 \pm 0.21$ |
|  | SK-MEL 28 | $1.5 \pm 0.08$ | $0.6 \pm 0.08$ | $0.3 \pm 0.07$ | $0.6 \pm 0.09$ |
| Ovarian | IGROV | $1.2 \pm 0.30$ | $2.5 \pm 0.10$ | $2.0 \pm 0.20$ | $1.3 \pm 0.30$ |
| Glioblastoma | SF-295 | $2.8 \pm 0.20$ | $0.6 \pm 0.06$ | $0.6 \pm 0.09$ | $4.4 \pm 0.50$ |
|  | SNB-19 | $1.6 \pm 0.60$ | $0.7 \pm 0.04$ | $0.9 \pm 0.10$ | $0.8 \pm 0.05$ |
| Colon | Colo205 | $0.4 \pm 0.04$ | $0.9 \pm 0.05$ | $1.1 \pm 0.05$ | $1.5 \pm 0.30$ |
|  | HT-29 | $0.6 \pm 0.08$ | $0.8 \pm 0.05$ | $0.5 \pm 0.10$ | $1.1 \pm 0.20$ |
|  | Caco-2 $^{\mathrm{b}}$ | $2.6 \pm 0.2$ | $1.0 \pm 0.6$ | $0.8 \pm 0.03$ | $6.7 \pm 0.80$ |
|  | Caco-2 $^{\mathrm{c}}$ | $6.1 \pm 0.32$ | $4.1 \pm 0.10$ | $3.8 \pm 0.09$ | $0.6 \pm 0.05$ |
|  | SI $^{\mathrm{d}}$ | 0.43 | 0.25 | 0.21 | 11.1 |

${ }^{a}$ Data represent mean values (SD) for three independent determinations. ${ }^{b}$ Pre confluent Caco-2 cell line. ${ }^{\text {c }}$ Post confluent Caco-2 cell line. ${ }^{\text {d }}$ SI $=$ selectivity index ( $\mathrm{IC}_{50}$ on undifferentiated Caco- 2 cell line/ $\mathrm{IC}_{50}$ on differentiated Caco-2 cell line ratio)

Subcellular distribution of TNQ derivatives in MCF-7 cell line.
Distribution of the labelled forms of our derivatives within the cell was investigated by confocal microscopy in MCF-7 cell line, using 50 nM of $\mathbf{1 1 m}\left(\mathrm{IC}_{50}=0.5 \mu \mathrm{M}\right)$ and $\mathbf{1 1 p}\left(\mathrm{IC}_{50}=0.6 \mu \mathrm{M}\right)$. As showed in Figure 26, these TNQ derivatives are clearly localized in the nuclei indicating a site of cytotoxic action similar to classic quinone-based intercalators.[200]


Figure 26. Distribution of labeled 11m and 11p in MCF-7 cells by confocal microscopy.

Topoisomerase inhibition. A number of quinone antitumor drugs are thought to be cytotoxic by virtue of their ability to stabilize a covalent topo II-DNA intermediate, the cleavable complex.[201] Topo II is an essential enzyme that plays an important role in DNA replication, repair, transcription, and chromosome segregation.[202] Topo II alters the topological state of nucleic acids by passing an intact DNA helix through a transient break which generates a separate DNA helix.[203, 204] We analyzed the possibility that compounds 11m and 11p could inhibit the activity of topo II. The effect of cytotoxic compounds 11 m and $\mathbf{1 1} \mathbf{p}$ and of the inactive compound 11c on the strand passage activity of topo II was determined by the enzyme-mediated negatively supercoiled pBR322 relaxation.[205]


Figure 27. Effects of compounds $\mathbf{1 1 p}, 11 \mathrm{~m}$ and $\mathbf{1 1} \mathrm{c}$ on the topo II-mediated DNA cleavage. Supercoiled plasmid pBR $322(0.5 \mathrm{pmol})$ was incubated with 1 unit of purified human topo II in the presence or absence of the tested agents: (lane 1), supercoiled DNA; (lane 2) relaxated DNA enzyme control; (lanes 3 and 4) 5 and $10 \mu \mathrm{M}$ of compound $\mathbf{1 1 p}$; (lanes 5 and 6) 5 and $10 \mu \mathrm{M}$ of compound 11m; (lanes 7 and 8) 5 and $10 \mu \mathrm{M}$ of compound 11 c .

As indicated in Figure 27, compounds 11m and 11p displayed significant inhibition of topo II mediated relaxation in a concentrationdependent mode, while 11c does not inhibit this activity at the concentrations tested. These results, showed also in Table 12 as semiquantitative form, parallel the cytotoxicity data enumerated in the same table, thus suggesting a behavior similar to classical intercalators. Moreover, at the assay concentrations, the doxorubicin showed a lack of activity (see Supporting Information) which agrees with the results described in different studies.[206, 207] These works show as the doxorubicin inhibits topo II only at the concentration range of 0.04 to $0.92 \mu \mathrm{M}$ while at higher concentration the inhibition is either diminished or totally abolished.

DNA Binding Properties by NMR. Representative compounds,
$\mathbf{1 1 \mathbf { c }}, \mathbf{1 1 m}$ and $\mathbf{1 1 p}$, were tested to see if they interact with DNA, using both saturation transfer difference (STD) [208] and water-ligand observed via gradient spectroscopy (WaterLOGSY) NMR techniques.[209] STD NMR and WaterLOGSY are techniques that can be used to characterize and identify binding. These techniques have become increasingly important as a tool in the investigation of biomolecular recognition phenomena.[210] In the STD NMR, resonances of the macromolecule are selectively saturated, and in a binding ligand, enhancements are observed in the difference (STD NMR) spectrum resulting from subtraction of this spectrum from a reference spectrum in which the macromolecule is not saturated. All the proton resonances of $\mathbf{1 1 m}$ and $\mathbf{1 1} \mathbf{p}$ were observed in the STD spectra acquired in the presence of poly(dG-dC) $\operatorname{poly}(\mathrm{dG}-\mathrm{dC})$ copolymer as DNA target (Figure 28), demonstrating that 11m/ and 11p/DNA interactions did occur. In contrast, the absence of the proton resonances of 11c in its STD spectra (Figure 28) demonstrates that 11c does not interact with DNA. The same results were obtained using the WaterLOGSY experiment. In this experiment, the large bulk water magnetization is partially transferred via the macromolecule-ligand complex to the free ligand. Due to the very different tumbling times of the free ligand and of the macromolecule-ligand complex, LOGSY
signals are typically negative for free ligands in solution, and relatively less negative or positive for binders in the presence of the macromolecule. Figure 29 shows the WaterLOGSY spectra of 11c, 11m and 11p with and without the poly(dG-dC) $\operatorname{poly}(d G-d C)$ copolymer. As observed, $\mathbf{1 1 m}$ and 11p signals became positive in the presence of DNA while 11c signals remain negative demonstrating that $\mathbf{1 1 m}$ and $\mathbf{1 1 p}$ but not $\mathbf{1 1 c}$ interact with the DNA polymer.

Furthermore, we applied the so-called DF-STD (differential frequency STD) spectroscopy,[211] to study the binding modes of 11 m and 11 p with the DNA. The method allows the discrimination of base-pair intercalators, minor-groove, and external binders. The approach is based on the comparison of two parallel sets of STD experiments performed under the same experimental conditions, in which saturation is centered either in the aromatic or in the low-field aliphatic spectral regions.


Figure 28. 1D proton spectra ( $\mathrm{a}, \mathrm{d}, \mathrm{g}$ ) and the corresponding STD NMR spectra recorded upon saturation at $10 \mathrm{ppm}(\mathrm{b}, \mathrm{e}, \mathrm{h})$ and $-1 \mathrm{ppm}(\mathrm{c}, \mathrm{f}, \mathrm{i})$ of $\mathbf{1 1 c} /, \mathbf{1 1 m} /$, and $\mathbf{1 1 p} /$ DNA complexes, respectively. The STD NMR spectra were plotted with the same noise level.


Figure 29. Water LOGSY spectra of $\mathbf{1 1 c}, \mathbf{1 1 m}$, and $\mathbf{1 1 p}$ in the absence (b, d, e) or in the presence ( $\mathrm{a}, \mathrm{c}, \mathrm{e}$ ) of poly(dG-dC)$\cdot \operatorname{poly}(\mathrm{dG}-\mathrm{dC})$ copolymer.* DMSO residual signal.

A ligand making proximate contacts with aromatic base protons, such as an intercalator sandwiched by consecutive base pairs, would receive more saturation upon irradiation of DNA aromatic protons rather than irradiation of deoxyribose protons. The converse would be true for an external ligand. The "binding mode index" (BMI), a numerical parameter that expresses the relative sensitivity of ligand protons to the perturbation arising from base versus sugar/backbone saturation was used.[211] Three BMI ranges were defined in the original contribution:[211] $0<\mathrm{BMI}<0.50$ for external (nonspecific) electrostatic backbone binding; $0.90<\mathrm{BMI}<1.10$ for minor groove binding; and $1.20(0.90)<\mathrm{BMI}<1.50$ for base-pair intercalation. DFSTD analysis of compound $\mathbf{1 1 m}$ gave different BMI values: $\mathrm{BMI}=$ 0.86 , for the aliphatic signals; and $\mathrm{BMI}=1.35$ for the aromatic signals. This result can be explained assuming two different DNA binding modes for $\mathbf{1 1 m}$. An intercalative mode of binding is sustained by its tricyclic planar core, and an external backbone binding can be attributed to its side chain. This is similar to that observed for doxorubicin [211] and for compounds 7 and $\mathbf{8}$ in our previous works.[190, 191] Considering 11p, BMI $=1.01$ was measured for the aromatic protons and $\mathrm{BMI}=0.90$ for the aliphatics. These BMI values are compatible with both intercalative and minor groove binders.

Cell Cycle Effects. To investigate the cytotoxic effects of these derivatives in more detail, we examined the effects on cell cycle progression in CaCo-2 cell line. The percentage of these cells in G1, S, and G2/M phases was analyzed after 48 h of treatment with $1 \mu \mathrm{M}$ of $\mathbf{1 1 m} \mathbf{m} \mathbf{1 1} \mathbf{p}$, and 11c (Figure 30). Under these conditions, the control cells were in the G1 phase $42 \%$, G2/M phase $21 \%$, and S phase $36 \%$. The treatment with 11p resulted in a significant accumulation of cells in the S phase while concomitantly the G1 populations decreased. About $53 \%$ of the $\mathrm{CaCo}-2$ cells treated with this compound were arrested at the S phase. Under the same conditions, treatment with 11m induces an weak increased of cell in both $G 2$ and $S$ phases and with compound 11c tempts a less significant response.


Figure 30. Effects of 11m, 11p, and 11c on the distribution of Caco-2 cell populations data represent the percentage of cells in each cell cellular cycle phases. For 11m: G1, $29 \%$; G2/M, $28 \%$; S, $41 \%$; 11p: G1, $20 \%$; G2/M, $26 \%$, S, $53 \%$; 11c: G1, $37 \%$; G2/M, 18\%, S, $44 \%$.

Accordingly, treatment of Caco-2 cells with $1 \mu \mathrm{M}$ of our derivatives for 48 h induced an increase of cyclin A expression [212] only in the case of $\mathbf{1 1 p}$ ( $43 \%$, see Supporting Information) indicating that the cell cycle progression of cells in the $S$ phase was prompted. The expression of cyclin A was not upset in treated Caco-2 cells with 11 m and 11 c .

Since cell division arrest is one of the prerequisites for cell differentiation,[213] we determined the effect of our molecules on Caco-2 differentiation. In Figure 31 we report alkaline phosphatase (ALP) activity, a marker of enterocytic differentiation correlated to post-confluent phase.[214]


Figure 31. Differentiation of Caco-2 cells assessed by measurement of alkaline phosphatase activity after 48 h of culture in the presence of $0,1,5$, and $10 \mu \mathrm{M}$ of $\mathbf{1 1 p}, 11 \mathrm{~m}$ and 11 c .

Treatment of pre-confluent Caco-2 with $1 \mu \mathrm{M}$ 11p increased ALP activity of $35 \%$ ( p value $<0.005$ ). A more significant increase of the
differentiation, with ALP augment of $>180 \%$, was only obtained by treatment of Caco- 2 cells with $5 \mu \mathrm{M} \mathrm{11p}$ or $10 \mu \mathrm{M} \mathbf{1 1 m}$ for 48 h . All these preliminary results suggested that, for this series, the cell growth inhibition was not related to cell cycle perturbation.

Modulation of heat shock protein (hsp) expression. Small heat shock proteins are involved in a variety of cellular processes including cell growth and differentiation.[215, 216] We previously reported the ability of a DTNQ analogue, compound $\mathbf{8}$, to modulate the heat shock protein expression on Caco-2 cells.[192] In order to evaluate the behavior of the new synthesized derivatives, we carried out a study preliminary of the effect of $\mathbf{1 1 m}$ and $\mathbf{1 1 p}$ at 1 and $5 \mu \mathrm{M}$ on Hsp27 expression in Caco-2 cells for 48 h . Hsp27 is weakly expressed in Caco-2 (Figure 32), and treatment of this cell line with 11p led to significant dose-dependent increase of its expression. 11m produced a weak enhanced of hsp27 expression at $1 \mu \mathrm{M}$, which was not observed at $5 \mu \mathrm{M}$.


Figure 32. Effects of Caco-2 cells treatment with 1 and $5 \mu \mathrm{M}$ of compounds 11p and 11 m on hsp27 expression.

Cardiomyocyte cell viability. It is well known that the clinical use of anthracyclines, specially doxorubicin, in the treatment of many neoplastic diseases is limited by cumulative cardiotoxicty.[186] One of the cause of this effect has been attributed to the redox process involving the quinone system which results in the formation of reactive oxygen species and ultimately in myocyte death. In order to evaluate the potential toxicity of our quinone ring we examined the cell viability in cardiac derived H 9 C 2 myocytes exposed to $1 \mu \mathrm{M} \mathbf{1 1 m}$ 11p and doxorubicin for $24,48,72,96$ and 120 h . Previous studies reported in the literature used this cell line as a model system to evaluate the cardiotoxicity caused by doxorubicin.[217, 218] As show in Figure 33, treatment with doxorubicin induced cardiotoxicity in a time-dependent manner [186] while compounds 11m and 11p
maintained an good cell viability after 120 h (74 and 76\%, respectively)


Figure 33. Results of cell viability assay of compounds $11 \mathrm{~m}, 11 \mathrm{p}$, and doxorubicin on H 9 C 2 cells at concentration of $1 \mu \mathrm{M}$.

The possible correlation [219] between these data and the preliminary results obtained with our products on the modulation of hsp27, will be object of more in-depth studies.

### 2.1.3 Conclusions

We report the synthesis and biological evaluation of a series of quinone-based derivatives, designed as conjugated structures linking a planar naphtho[2,3-b]thiophenedione core with different acylsubstituted groups. Among the designed molecules, compounds containing an 3-(diethylamino)propanamide (11m) or 3-(2(dimethylamino)ethylamino)propanamide (11p) protonable side chain, showed a greater cytotoxic potency than doxorubicin against cell lines which were highly resistant to treatment with this drug, such as the melanoma (MDA-MB435), glioblastoma (SF-295) and colon (SW 620, Col205, and HT-29) human tumor cell lines,

Preliminary results about the mechanism of action indicate that these derivatives had a significant effect on topoisomerase II activity targeting the nuclear DNA, which is generally considered as an attractive target for anticancer therapy. The NMR results suggested that DNA interactions do occur for highly active compounds $\mathbf{1 1 m}$ and 11p but not for inactive compound 11c. Experimental data indicate that $\mathbf{1 1 m}$ and $\mathbf{1 1 p}$ intercalate the DNA through their aromatic portion. Furthermore, a non intercalative mode of binding to DNA can also hold for 11p. These data revealed significant similarities in the cytotoxic behavior and the site of action of these compounds compared to classical intercalators. However, tested compounds
showed a minor influence on the regulation of the cellular cycle and only derivative $11 \mathbf{p}$ prolonged the S phase of the Caco- -2 cell cycle inducing both delay of cell cycle progression in responsive cells and moderate cellular differentiation. This last compound showed also a high ability of increase hsp27 expression. Finally, the compounds under study affect the viability of H9C2 cells after chronic treatment at less extend compared to doxorubicin. Further development and more in-depth studies on mechanism of action of this series are in progress.

### 2.1.4 Experimental Section

General: Reagents, starting materials, and solvents were purchased from commercial suppliers and used as received. Analytical TLC was performed on plates coated with a 0.25 mm layer of silica gel 60 F254 Merck and preparative TLC on $20 \times 20 \mathrm{~cm}$ glass plates coated with a 0.5 mm layer of silica gel PF254 Merck. Flash and gravity chromatographic purification were performed using 230-400 mesh silica gel unless otherwise noted. Melting points were taken on a Kofler apparatus and are uncorrected. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded with a Varian-400 spectrometer, operating at 400 and 100 MHz , respectively. Chemical shifts are reported in $\delta$ values (ppm) relative to internal $\mathrm{Me}_{4} \mathrm{Si}$, and $J$ values are reported in Hertz (Hz). ESI-MS experiments were performed on an Applied Biosystem API 2000 triple-quadrupole spectrometer. Combustion microanalyses were performed on a Carlo Erba CNH 1106 analyzer and all reported values are within $0.4 \%$ of calculated values. These elemental analyses confirmed $\geq 95 \%$ purity.

General procedure for the synthesis of 3-[(Acyl)amino]naphtho[2,3-b]thiophene-4,9-dione trifluoroacetate salts (11 a-f). The 3-amino-3-ethoxycarbonyl-2,3-dihydrothieno[2,3$b]$ naphtho-4,9-dione system (DTNQ) (1), the 3-(N-tert-butyloxyaminoacyl)amino-3-ethoxycarbonyl-2,3-dihydrothieno[2,3-
$b]$ naphtho-4,9-dione (10, a-f), and the 3-(N-tertbutyloxyaminoacyl)aminonaphtho [2,3-b]thiophene-4,9-dione (11' af) derivatives were synthesized according to the references 12,13 , and 17 respectively. Then, TFA was added to a solution of decarboxylated Boc-protected derivatives ( $\mathbf{1 1}^{\prime} \mathbf{a - f}$ ) ( 0.1 mmol ) in DCM ( 10 mL ), using triethylsilane as scavenger. Stirring was continued for 3-4 hat room temperature, the reaction mixture was concentrated to half volume and ether was added. The title compounds as the trifluoroacetate salt, were collected by filtration as yellow solids.

## 3-[(Glycyl)amino]naphtho[2,3-b]thiophene-4,9-dione

trifluoroacetate (11a). $45 \%$, mp 207-208 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 4.10\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right.$ ) ; 7.85-7.87 (m, 2H, H-6 and H-7); 8.208.23 (m, 2H, H-5 and H-8); $8.50(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-2) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 45.7\left(\mathrm{CH}_{2}\right), 119.2(\mathrm{C}-2) ; 127.5(\mathrm{C}-6$ and $\mathrm{C}-7) ; 129.4(\mathrm{C}-3) ;$ 132.9 (C-8a); 134.2 (C-4a); 134.7 (C-5 and C-8); 135.9 (C-3a); 147.1 (C-9a); 171.8, 180.7 and 182.7 (C=O). ESI-MS $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}, 400.03$; found, 400.11.

3-[(L-phenylyl)amino]naphtho[2,3-b]thiophene-4,9-dione trifluoroacetate (11c). $41 \%$, mp 195-196 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta$ 2.96-3.07 ( $2 \mathrm{H}, \mathrm{m}, \beta \mathrm{CH}_{2}$ ), 4.44-4.47 ( $1 \mathrm{H}, \mathrm{m}, \alpha \mathrm{CH}$ ), 7.12$7.22(5 \mathrm{H}, \mathrm{m}$, aryl), $7.87-7.89(2 \mathrm{H}, \mathrm{m}, \mathrm{H}-6$ and H 7), 8.22-8.25 ( 2 H , m, H-5 and H 8), 8.47 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{H}-2$ ). ${ }^{13} \mathrm{C}$ NMR ( 100 MHz ,
$\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 37.9\left(\beta \mathrm{CH}_{2}\right), 50.6(\alpha \mathrm{CH}), 118.8(\mathrm{C}-2) ; 127.9(\mathrm{C}-6$ and $\mathrm{C}-$ 7); 125.9, 127.6, 128.3, 128.9 and 137.9 (aryl),131.4 (C-3); 133.6 (C8a); 134.2 (C-4a); 134.9 (C-5 and C-8); 139.0 (C-3a); 142.5 (C-9a); 172.7179 .8 and $181.9(\mathrm{C}=\mathrm{O})$. ESI-MS $m / z$ calcd for $\mathrm{C}_{23} \mathrm{H}_{17} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$, 490.08; found, 490.01.

General procedure for the synthesis of $N$-(4,9-dioxo-4,9-dihydronaphtho[2,3-b]thiophen-3-yl)-3-(substituted)propanamide (11 m-p). To a solution of $13(0.1-0.3 \mathrm{mmol})$ in THF $(20 \mathrm{~mL})$ were added $\mathrm{N}, \mathrm{N}$-diethylamine, or triphenylmethanethiol, or morpholine, or $\mathrm{N}, \mathrm{N}$-dimethylethylendiamine (1.1 equiv) and DIPEA (2 equiv). After stirring at reflux temperature for $12-24 \mathrm{~h}$, the solvent was evaporated. Then, the residues ( $13{ }^{\prime} \mathrm{m}-\mathrm{p}$ ) were dissolved into methanol-water (9:1, 20 mL ) and DBU (5 equiv) was added dropwise to these solutions. The reaction mixtures were stirred for $0.5-1 \mathrm{~h}$, then the solvents were evaporated and the reaction residues were dissolved in chloroform and washed with water and dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The corresponding free bases of compounds $\mathbf{1 1 m}, 11 \mathrm{n}$, and 11 p were first purified by FC using DCM/Methanol 9/1 as eluent system. Then, the treatment with a $\mathrm{HCl}(\mathrm{g}) /$ diethyl ether solution give the final compounds as hydrochloride saltsand yellow solids. Compound protected 110 was purified by FC using n-hexane/ethylacetate $3 / 2$ as eluent. Then, the
final compound was obtained after Trt removal with a 50\% TFA/DCM solution.

3-(diethylamino)-N-(4,9-dioxo-4,9-dihydronaphtho[2,3-b]thiophen-3-yl)propanamide hydrochloride (11m). 43\%, mp 201$202{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR (400 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 1.36-1.40\left(\mathrm{t}, 6 \mathrm{H}, \mathrm{CH}_{3}\right) ; 3.10-$ 3.13 (m, 2H, CH2); 3.30-3.34 (q, 4H, CH2); 3.56-3.59 (m, 2H, CH2); 7.85-7.87 (m, 2H, H-6 and H-7); 8.20-8.24 (m, 2H, H-5 and H-8); $8.48(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-2) .{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 15.8\left(\mathrm{CH}_{3}\right), 35.8$ $\left(\alpha \mathrm{CH}_{2}\right), 47.9\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 51.5\left(\mathrm{CCH}_{2}\right), 118.6(\mathrm{C}-2) ; 126.9(\mathrm{C}-6$ and $\mathrm{C}-$ 7); 129.2 (C-3); 132.7 (C-8a); 133.1 (C-4a); 134.8 (C-5 and C-8); 138.9 (C-3a); 145.0 (C-9a); 172.5, 178.9 and 182.6 (C=O). ESI-MS $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{ClN}_{2} \mathrm{O}_{3} \mathrm{~S}, 392.10$; found, 390.17.

3-(2-(dimethylamino)ethylamino)-N-(4,9-dioxo-4,9-
dihydronaphtho[2,3-b]thiophen-3-yl) propanamide dihydrochloride (11p). $45 \%, \operatorname{mp} 227-228{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 2.98(\mathrm{~s}$, $\left.6 \mathrm{H}, \mathrm{CH}_{3}\right), 3.09-3.12\left(\mathrm{t}, 2 \mathrm{H}, \alpha \mathrm{CH}_{2}\right) ; 3.29-3.31\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{~N}(\mathrm{Me})_{2}\right) ;$ $3.47-3.50\left(\mathrm{~m}, 2 \mathrm{H}, \beta \mathrm{CH}_{2}\right) 3.50-3.53\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{NHCH}_{2}\right), 7.85-7.87(\mathrm{~m}$, 2H, H-6 and H-7); 8.20-8.25 (m, 2H, H-5 and H-8); 8.49 (s, 1H, H-2). ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 32.1\left(\alpha \mathrm{H}_{2}\right), 42.4\left(\mathrm{NHCH}_{2}\right), 42.8$ $\left(\mathrm{CH}_{3}\right), 43.9\left(\mathrm{\beta CH}_{2}\right) 53.1\left(\mathrm{CH}_{2} \mathrm{~N}(\mathrm{Me})_{2}\right), 119.5(\mathrm{C}-2) ; 126.8$ and 127.0 (C-6 and C-7); 129.0 (C-3); 132.9 (C-8a); 133.5 (C-4a); 134.2 and
134.3 (C-5 and C-8); 137.5 (C-3a); 144.0 (C-9a); 171.9, 178.9 and 182.5 (C=O). ESI-MS $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}, 443.08$; found, 443.18 .

Biology. Dulbecco's modified Eagle's medium (DMEM), fetal bovine serum (FBS), trypsin-EDTA solution (1 x) penicillin and streptomycin, phosphate-buffered saline (PBS) were from Cambrex Biosciences. 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT), propidium iodide (PI), Triton X-100, sodium citrate, formamide, mouse monoclonal anti-tubulin were purchased from Sigma (Milan, Italy). Rabbit polyclonal anti-cyclin A primary antibody were from Cell Signaling Technology (Celbio; Milan, Italy). ECL reagent was obtained from Amersham Pharmacia Biotech, UK.

Cell culture. Human breast MDA231, human colon carcinoma SW620, Colo205, HT-29, and Caco-2, human monocytic leukemia U937, human melanoma MDA-MB435 and SK-MEL28 human ovarian cancer IGROV, and human glioblastoma SF-295 and SNB-19 cell lines, were grown at $37{ }^{\circ} \mathrm{C}$ in Dulbecco's modified Eagle's medium containing 10 mM glucose (DMEM-HG) supplemented with $10 \%$ fetal calf serum and 100 units $/ \mathrm{ml}$ each of penicillin and streptomycin and $2 \mathrm{mmol} / \mathrm{L}$ glutamine. In each experiment, cells were placed in fresh medium, cultured in the presence of synthesized compounds (from 0.1 to 25 mM ) and followed for further analyses.

Cell Viability Assay. Cell viability for all cell lines was determined using the 3-[4,5-demethylthiazol-2,5-diphenyl-2Htetrazolium bromide (MTT) colorimetric assay. The test is based on the ability of mitochondrial dehydrogenase to convert, in viable cells, the yellow MTT reagent (Sigma Chemical Co., St Louis, MO.) into a soluble blue formazan dye. Cells were seeded into 96-well plates to a density of 105 cells $/ 100 \mu \mathrm{~L}$ well. After 24 h of growth to allow attachment to the wells, compounds were added at various concentrations (from 0.1 to 25 mM ). After 24 or 48 h of growth and after removal of the culture medium, $100 \mu \mathrm{~L} /$ well of medium containing $1 \mathrm{mg} / \mathrm{mL}$ of MTT was added. Cell cultures were further incubated at $37{ }^{\circ} \mathrm{C}$ for 2 hrs in the dark. The solution was then gently aspirated from each well, and the formazan crystals within the cells were dissolved with $100 \mu \mathrm{~L}$ of DMSO. Optical densities were read at 550 nm using a Multiskan Spectrum Thermo Electron Corporation reader. Results were expressed as percentage relative to vehicletreated control ( $0.5 \%$ DMSO was added to untreated cells ) $\mathrm{IC}_{50}$ (concentration eliciting 50\% inhibition) value were determined by linear and polynomial regression. Experiments were performed in triplicate.

Topo II-mediated supercoiled pBR322 relaxation. DNA relaxation assays were based on the procedure of Osheroff et al.[204]

Reaction buffer contained 10 mM Tris. HCl (pH 7.9), $50 \mathrm{mM} \mathrm{KCl}, 50$ $\mathrm{mM} \mathrm{NaCl}, 5 \mathrm{mM} \mathrm{MgCl} 2,0.1 \mathrm{mM}$ EDTA, and $15 \mu \mathrm{~g} / \mathrm{mL}$ of bovine serum albumin (BSA), $0.15 \mu \mathrm{~g}$ supercoiled pBR 322 , 4 units of topo II in a total of $20 \mu \mathrm{~L}$. Relaxation was employed at $37{ }^{\circ} \mathrm{C}$ for 6 min and stopped by the addition of $3 \mu \mathrm{~L}$ of stop solution ( 100 mM EDTA, $0.5 \%$ SDS, $50 \%$ glycerol, $0.05 \%$ bromophenol blue). Electrophoresis was carried out in a $1 \%$ agarose gel in $0.5 \times \mathrm{TBE}(89 \mathrm{mM}$ Tris base, 89 mM boric acid and 2 mM EDTA) at $4 \mathrm{~V} / \mathrm{cm}$ for 1 h . DNA bands were stained with $0.5 \mu \mathrm{~g} / \mathrm{mL}$ of ethidium bromide (E.B.) solution and photographed through a Gel Document System GDS8000 (UVP). The amount of DNA bands was quantified by Gel 1D Intermediate software.

Confocal microscopy. For immunocytochemistry, cells were fixed in $0.04 \mathrm{~g} /$ liter paraformaldehyde for 30 min at $4^{\circ} \mathrm{C}$ and permeabilized with $0.01 \mathrm{~g} /$ liter Triton $\mathrm{X}-100$ for 30 min at 4 C . Cells were then washed and stained with Hoechst 33342 (Vector, Burlingame, CA). Images were acquired with a LSM510 inverted confocal microscope (Zeiss, Oberkochen, Germany) using 63X oil objective and processed using LSM software (Zeiss).

Flow Cytometry. Analysis of Cell Cycle. CaCo-2 cells were seeded in six multiwell plates at the density of $25 \times 105$ cells/plate. After 48 h of incubation with $\mathbf{1 1 m}$, and $\mathbf{1 1 p}$ derivatives and
doxorubicin in DMEM without serum at $37^{\circ} \mathrm{C}$, cells were washed in PBS, pelleted in centrifuged, and directly stained in a propidium iodide (PI) solution (50 mg PI in $0.1 \%$ sodium citrate, $0.1 \% \mathrm{NP} 40, \mathrm{pH}$ 7.4) for 30 min at $4^{\circ} \mathrm{C}$ in the dark. Flow cytometric analysis was performed using a FACScan flow cytometer (Becton Dickinson, San Jose, CA). To evaluate cell cycle PI fluorescence was collected as FL2 (linear scale) by the ModFIT software (Becton Dickinson). For the evaluation of intracellular DNA content, at least 20000 events for each point were analyzed in at least three different experiments giving a s.d. less than $5 \%$.

Western Blot Assay. The effects of 11m, 11p and 11c on expression of Ciclyn $A$ and of 11m and 11p on HSP27, were determined by Western blots. Compounds stimulated and unstimulated (control) cell lysates were prepared using an ice cold lysis buffer ( $50 \mathrm{mMTris}, 150 \mathrm{mM} \mathrm{NaCl}, 10 \mathrm{mM}$ EDTA, $1 \%$ Triton) supplemented with a mixture of protease inhibitors containing antipain, bestatin, chymostatin, leupeptin, pepstatin, phosphoramidon, Pefabloc, EDTA, and aprotinin (Boehringer, Mannheim, Germany). Equivalent protein samples were resolved on $8-12 \%$ sodium dodecyl sulfate (SDS)-polyacrylamide gels and transferred to nitrocellulose membranes (Bio-Rad, Germany). For immunodetection, membranes were incubated overnight with specific antibody at the concentrations
indicated in manufacter's protocol (Santa Cruz Biotechnology). The two antibodies were diluted in Tris-buffered saline/Tween 20-1\% milk powder. This step was followed by incubation with the corresponding horseradish peroxidase conjugated antibody (antirabbit-IgG 1:6000; Biosource, Germany). Bands were read by enhanced chemiluminescence (ECL-kit, Amersham, Germany).

Alkaline phosphatase activity. Alkaline phosphatase (ALP) activity was used as marker of the degree of cells differentiation. Attached and floating cells were washed and lysed with $0.25 \%$ sodium deoxycholate, essentially as described by Herz et al.[220] ALP activity was determined using Sigma Diagnostics ALP reagent (No. 245). Total cellular protein content of the samples was determined in a microassay procedure as described by Bradford [221] using the Coomassie Protein Assay Reagent Kit (Pierce). ALP activity was calculated as units of activity per mg of protein.

H9C2 cell viability. Cardiomyoblasts H9C2 were cultured in Dulbecco's minimal essential medium (DMEM, GIBCO) supplemented with $0.1 \mathrm{~g} / \mathrm{L}$ fetal bovine serum (FBS, GIBCO) 200 $\mathrm{mg} / \mathrm{mL}$ L-glutamine, 100 units $/ \mathrm{mL}$ penicillin, and $10 \mathrm{mg} / \mathrm{mL}$ streptomycin (Sigma-Aldrich), at $37^{\circ} \mathrm{C}$ in $0.95 \mathrm{~g} / \mathrm{L}$ air- $0.05 \mathrm{~g} / \mathrm{L} \mathrm{CO} 2$. The H9C2 were studied between passages 4 and 10. The MTTcolorimetric assay (Invitrogen, San Diego, CA), was used to evaluate
cell proliferation in presence or absence of inhibitors. Briefly, H9C2 cells were plated into 96 multiwell at a density of 2000 cells/well in quadruplicate. Inhibitors (11m, 11p, and Doxorubicin) were added to each well at a concentration of $1 \mu \mathrm{M}$ for the indicated time points.. Then $10 \mu l$ of MTT reagent were added to each well, the plate was returned to cell culture incubator for 2 hours. The absorbance in each well, including the blanks, was measured at 570 nm in a microplate plate reader.

Statistical analysis. Data were expressed as mean $\pm$ standard deviation (SD). Statistical significance was assessed by Student-t test. $P$ value adjustment for multiple comparisons was done by the Holm (sequential Bonferonni correction method). $\mathrm{P}<0.05$ was considered statistically significant.

STD-NMR and WaterLOGSY Spectroscopy. STD-NMR [208] and WaterLOGSY [209] experiments were performed on a Varian Inova 700 MHz spectrometer at $25^{\circ} \mathrm{C}$. NMR samples were prepared by dissolving the ligand and the poly(dGdC) poly(dG-dC) copolymer (Pharmacia Biochemicals) in $\mathrm{H}_{2} \mathrm{O} / \mathrm{D}_{2} \mathrm{O} 9: 1$ (final volume $600 \mu \mathrm{~L} ; \mathrm{D}_{2} \mathrm{O}$ 99.996\%, CIL Laboratories) containing phosphate-buffered saline $(100 \mathrm{mM})$ at pH 7.1. A high ligand-receptor molar excess (20:1) was used. In particular, the concentration of $\mathbf{1 1 c}, \mathbf{1 1 m}$, and $\mathbf{1 1 p}$ was 1.0 mM , whereas that of the DNA was $50 \mu \mathrm{M}$, expressed as molarity of
phosphate groups. Water suppression was achieved by using the double-pulsed field gradient spin-echo (DPFGSE) scheme.[68] The STD effects of the individual protons were calculated for each compound relative to a reference spectrum with off-resonance saturation at $\delta=-16 \mathrm{ppm}$. Typically, 512 scans were recorded for each DF-STD spectrum (saturation time $=2 \mathrm{~s}$ ). The relative STD effect was calculated for each signal as the difference between the intensity (expressed as $\mathrm{S} / \mathrm{N}$ ratio) of one signal in the on-resonance STD spectrum and that of the same signal in the off-resonance NMR spectrum divided by the intensity of the same signal in the offresonance spectrum. BMI values were obtained as ratio of the relative STD effects upon irradiation at 10.0 and -1.0 ppm .[210] The absence of STD effects in samples in which the DNA was not added ensured a selective macromolecule saturation. WaterLOGSY NMR experiments employed a 20 ms selective Gaussian $180^{\circ}$ pulse at the water signal frequency and an NOE mixing time of 1.5 s .

## Chapter 3 - SUPPORTING INFORMATION



Cpa


Orn

$\mathrm{NaI}(1)$



Tic


Dab


Dap

$\mathrm{NaI}(2)$

Figure S1. Chemical structures of the non-coded amino acids cited throughout the manuscript.

Table S1. Analytical Data for the P5U and Urantide Analogues

| Peptide | Structure | HPLC ${ }^{\text {a }}$ | MS (M+H) |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  | $k^{\prime}$ | Found | Calcd |
| 1 | H-Ala-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH | 8.22 | 1045.80 | 1045.30 |
| 2 | H-Ala-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH | 8.16 | 1031.90 | 1031.27 |
| 3 | H-Phe-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH | 8.31 | 1121.90 | 1121.40 |
| 4 | H-Phe-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH | 8.26 | 1108.10 | 1107.37 |
| 5 | H-Cpa-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH | 8.33 | 1156.44 | 1155.85 |
| 6 | H-Cpa-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH | 8.30 | 1142.32 | 1141.82 |
| 7 | H-Nal(1)-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH | 8.35 | 1171.98 | 1171.46 |
| 8 | H-Nal(1)-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH | 8.32 | 1158.10 | 1157.43 |
| 9 | H-Nal(2)-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH | 8.35 | 1171.96 | 1171.46 |
| 10 | H-Nal(2)-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH | 8.31 | 1158.15 | 1157.43 |
| 11 | $\begin{aligned} & \mathrm{H}-\left(\mathrm{pNO}_{2}\right) \text { Phe-c }[\text { Pen-Phe-Trp-Lys-Tyr-Cys }] \text {-Val- } \\ & \mathrm{OH} \end{aligned}$ | 8.31 | 1167.01 | 1166.40 |
| 12 | $\mathrm{H}-\left(\mathrm{pNO}_{2}\right) \text { Phe-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val- }$ $\mathrm{OH}$ | 8.27 | 1152.97 | 1152.38 |
| 13 | H-Tic-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH | 8.34 | 1134.03 | 1133.41 |
| 14 | H-Tic-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH | 8.31 | 1120.10 | 1119.38 |
| 15 | H-Lys-c[Pen-Phe-Trp-Lys-Tyr-Cys]-Val-OH | 8.12 | 1103.12 | 1102.39 |
| 16 | H-Lys-c[Pen-Phe-DTrp-Orn-Tyr-Cys]-Val-OH | 8.09 | 1089.11 | 1088.37 |

${ }^{\text {a }} k^{\prime}=[($ peptide retention time - solvent retention time $) /$ solvent retention time $]$.

Table S2. Amino Acid Analysis of the P5U and Urantide Analogues ${ }^{\text {a }}$

## R-c[Pen-Phe-Xaa-Yaa-Tyr-Cys]-Val-OH

| Peptide | $\mathbf{R}$ | Pen | Phe $^{\text {Xaa }}{ }^{\mathbf{b}}$ | Yaa $^{\mathbf{c}}$ | Tyr | Cys | Val |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1}$ | 0.97 | N.D. | 1.0 | - | 0.91 | 0.92 | 1.02 | 0.98 |
| $\mathbf{2}$ | 0.98 | 0.98 | 0.97 | - | 0.92 | 0.98 | 1.00 | 0.99 |
| $\mathbf{3}$ | 0.89 | N.D. | 0.98 | - | 1.00 | 1.00 | 0.98 | 0.97 |
| $\mathbf{4}$ | 0.92 | 0.98 | 0.93 | - | 0.92 | 0.93 | 0.97 | 1.02 |
| $\mathbf{5}$ | 0.96 | N.D. | 0.96 | - | 0.88 | 0.97 | 0.99 | 0.93 |
| $\mathbf{6}$ | 0.98 | 0.91 | 0.91 | - | 0.89 | 0.91 | 0.94 | 0.92 |
| $\mathbf{7}$ | 0.87 | N.D. | 0.96 | - | 0.97 | 0.95 | 0.91 | 0.89 |
| $\mathbf{8}$ | 0.91 | N.D. | 0.90 | - | 0.89 | 0.91 | 1.00 | 0.93 |
| $\mathbf{9}$ | 0.89 | 0.91 | 0.98 | - | 0.91 | 0.96 | 0.97 | 0.96 |
| $\mathbf{1 0}$ | 0.91 | 0.89 | 0.97 | - | 0.90 | 0.97 | 0.89 | 0.99 |
| $\mathbf{1 1}$ | 0.98 | 0.93 | 0.95 | - | 0.90 | 0.93 | 0.95 | 0.92 |
| $\mathbf{1 2}$ | N.D. | 0.92 | 0.93 | - | 0.94 | 0.91 | 0.93 | 0.94 |
| $\mathbf{1 3}$ | 0.93 | N.D. | 0.99 | - | 0.93 | 0.97 | 0.95 | 0.96 |
| $\mathbf{1 4}$ | 0.96 | 0.90 | 1.01 | - | 0.99 | 0.98 | 0.96 | 0.93 |
| $\mathbf{1 5}$ | 0.98 | 0.95 | 0.97 | - | 0.98 | 0.95 | 0.99 | 0.91 |
| $\mathbf{1 6}$ | 0.99 | 0.94 | 0.97 | - | 0.98 | 0.99 | 1.00 | 0.93 |

${ }^{\text {a }}$ The analyses were performed using an Applied Biosystems Model 420A amino acid analyzer with automatic hydrolysis (Vapor Phase at $160{ }^{\circ} \mathrm{C}$ for 1 h 40 min using 6 N HCl ) and a precolumn phenylthiocarbamyl-amino acid (PTC-AA) analysis. No correction is made for amino acid decomposition. ${ }^{\mathrm{b}}$ Trp was not well determined due to decomposition under these conditions. Other notations: "ND" (not determined) refers to the amino acid that could not be estimated due to unavailability of a standard sample. ${ }^{\text {c }}$ Yaa $=$ Lys ( $\mathbf{1 , 3 , 5 , 7 , 9 , 1 1 , 1 3 , 1 5 ) \text { , Orn }}$ (2,4,6,8,10,12,14,16);

Table S3. NMR Resonance Assignments ${ }^{a}$ of Peptide 14 in SDS- $\mathrm{d}_{25}$ 200mM Solution.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}}, \text { exc, }-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}\left({ }^{3} J_{\alpha \beta}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :--- | :--- | :--- | :--- | :--- |
| Tic 4 |  | $4.57(5.5,9.5)$ | $3.22,2.93$ | $6.97(\delta) ; 7.23(\varepsilon) ;$ |
| Pen 5 | $8.50(8.8, \mathrm{f}, 6.9)$ | 5.14 |  | $1.32,1.13(\gamma)$ |
| Phe 6 | $9.05(8.5, \mathrm{~ms}, 3.2)$ | 4.71 (overl.) | 2.98 | $7.08(\delta) ;$ |
| D-Trp 7 | $8.18(4.5, \mathrm{f}, 6.3)$ | $4.58(9.0,7.3)$ | $3.19,3.00$ | $7.14(\delta) ; \quad 9.92, \quad 7.67(\varepsilon) ;$ |
| Orn 8 | $7.47(5.9, \mathrm{f}, 4.7)$ | $4.09(3.2,7.3)$ | $1.44,1.13$ | $0.97,0.65(\gamma) ; 2.71,2.67(\delta) ;$ |
| Tyr 9 | $8.07(8.4, \mathrm{~s}, 2.7)$ | $4.69(5.3,10.0)$ | $2.99,2.90$ | $7.11(\delta) ; 6.76(\varepsilon)$ |
| Cys 10 | $8.30(8.7, \mathrm{f}, 6.8)$ | $5.38($ overl.) | 2.89 |  |
| Val 11 | $8.14(8.6$, ms, 3.4) | 4.40 | 2.05 | $0.88,0.80(\gamma)$ |

${ }^{a}$ Obtained at $25^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$, and ${ }^{3} J_{\alpha \beta}$ coupling constants in Hz . exc $=\mathrm{NH}$ exchange rate (f, fast; ms, moderately slow; s , slow;). $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ).

Table S4. NMR Resonance Assignments ${ }^{a}$ of Peptide 16 in SDS- $\mathrm{d}_{25}$ 200 mM Solution.

| residue | NH ( ${ }^{3} J_{\alpha N}$, exc, $\left.-\Delta \delta / \Delta T\right)^{\text {b }}$ | $\mathrm{C}^{\alpha} \mathrm{H}\left({ }^{3} J_{\alpha \beta}\right)^{\text {b }}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| Lys 4 |  | 4.05 (8.7, 7.3) | 1.71, 1.61 | $\begin{aligned} & 1.40, \quad 1.26(\gamma) ; \quad 1.68(\delta) ; \\ & 2.96,2.90(\varepsilon) ; 7.48(\zeta) \end{aligned}$ |
| Pen 5 | 8.26 (8.8, f, 6.7) | 5.06 |  | 1.26, 1.06( $\gamma$ ) |
| Phe 6 | 9.21 (8.5, ms, 3.2) | 4.69 (9.0, 6.0) | 3.05, 2.90 | 7.13(8) |
| D-Trp 7 | 8.55 (4.5, f, 6.3) | 4.62 (9.1, 7.2) | 3.24, 3.07 | $\begin{aligned} & 7.16(\delta) ; \quad 9.99, \quad 7.66(\varepsilon) ; \\ & 7.51,7.10(\zeta) ; 7.18(\eta) \end{aligned}$ |
| Orn 8 | 7.30 (5.9, f, 4.8) | 4.11 (3.4, 7.1) | 1.42, 1.01 | 0.87, $0.50(\gamma) ; 2.69,2.63$ ( $\delta$ ) |
| Tyr 9 | 8.09 (8.4, s, 2.7) | 4.73 (5.5, 9.9) | 3.05, 2.97 | 7.13(8); $6.75(\varepsilon)$ |
| Cys 10 | 8.50 (8.7, f, 6.8) | 5.44 (overl.) | 2.87 |  |
| Val 11 | 8.39 (8.6, ms, 3.1) | 4.46 | 2.27 | 1.01 ( $\gamma$ ) |

${ }^{a}$ Obtained at $25^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ and ${ }^{3} J_{\alpha \beta}$ coupling constants in Hz. exc $=\mathrm{NH}$ exchange rate (f, fast; ms, moderately slow; s, slow;). $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ).

NOE derived Upper Limit Constraints.

| 4 | TIC | HA | 5 | PEN | HN | 2.83 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 5 | PEN | HN | 5 | PEN | QG1 | 3.71 |
| 5 | PEN | HN | 10 | CYSS | HA | 5.50 |
| 5 | PEN | HA | 6 | PHE | HN | 2.71 |
| 5 | PEN | HA | 10 | CYSS | HA | 2.65 |
| 5 | PEN | HA | 10 | CYSS | QB | 5.35 |
| 5 | PEN | HA | 11 | VAL | HN | 3.55 |
| 5 | PEN | QG1 | 10 | CYSS | HA | 6.53 |
| 5 | PEN | QG1 | 10 | CYSS | QB | 6.51 |
| 5 | PEN | QG2 | 6 | PHE | HN | 5.85 |
| 5 | PEN | QG2 | 7 | DTRP | HN | 6.50 |
| 5 | PEN | QG2 | 10 | CYSS | HA | 6.40 |
| 5 | PEN | QG2 | 10 | CYSS | QB | 7.03 |
| 6 | PHE | HN | 9 | TYR | HN | 4.26 |
| 6 | PHE | HN | 10 | CYSS | HA | 4.11 |
| 6 | PHE | HA | 7 | DTRP | HN | 2.71 |
| 7 | DTRP | HN | 7 | DTRP | HB2 | 2.86 |
| 7 | DTRP | HN | 7 | DTRP | HB3 | 3.02 |
| 7 | DTRP | HN | 7 | DTRP | HD1 | 5.31 |
| 7 | DTRP | HA | 7 | DTRP | HB2 | 2.83 |
| 7 | DTRP | HA | 7 | DTRP | HB3 | 2.80 |
| 7 | DTRP | HA | 7 | DTRP | HD1 | 4.01 |
| 7 | DTRP | HA | 7 | DTRP | HE3 | 4.24 |
| 7 | DTRP | HA | 8 | ORN | HN | 2.59 |
| 7 | DTRP | HA | 9 | TYR | HN | 3.42 |
| 7 | DTRP | HB2 | 7 | DTRP | HD1 | 3.14 |
| 7 | DTRP | HB2 | 7 | DTRP | HE3 | 4.07 |
| 7 | DTRP | HB2 | 8 | ORN | HN | 3.76 |
| 7 | DTRP | HB3 | 7 | DTRP | HD1 | 3.70 |
| 7 | DTRP | HB3 | 7 | DTRP | HE3 | 3.86 |
| 7 | DTRP | HB3 | 8 | ORN | HN | 4.07 |
| 8 | ORN | HN | 8 | ORN | HB2 | 3.52 |
| 8 | ORN | HN | 8 | ORN | HB3 | 3.52 |
| 8 | ORN | HN | 8 | ORN | QB | 3.28 |
| 8 | ORN | HN | 8 | ORN | HG2 | 4.35 |
| 8 | ORN | HN | 8 | ORN | HG3 | 4.35 |
| 8 | ORN | HN | 8 | ORN | QG | 4.18 |
| 8 | ORN | HN | 8 | ORN | HD2 | 5.50 |
| 8 | ORN | HN | 8 | ORN | HD3 | 5.50 |
| 8 | ORN | HN | 9 | TYR | HN | 2.86 |
| 8 | ORN | HA | 8 | ORN | HB2 | 3.05 |
| 8 | ORN | HA | 8 | ORN | HB3 | 3.05 |
| 8 | ORN | HA | 8 | ORN | QG | 4.05 |
| 8 | ORN | HA | 9 | TYR | HN | 3.36 |
| 8 | ORN | HA | 9 | TYR | QD | 7.64 |
| 8 | ORN | HB2 | 8 | ORN | HE1 | 5.50 |
| 8 | ORN | HB2 | 9 | TYR | HN | 4.51 |
| 8 | ORN | HB2 | 9 | TYR | QD | 7.64 |
| 8 | ORN | HB2 | 9 | TYR | QE | 7.63 |
| 8 | ORN | HB3 | 8 | ORN | HE1 | 5.50 |
| 8 | ORN | HB3 | 9 | TYR | HN | 4.51 |
| 8 | ORN | HB3 | 9 | TYR | QD | 7.64 |
| 8 | ORN | HB3 | 9 | TYR | QE | 7.63 |
| 8 | ORN | QB | 9 | TYR | HN | 3.91 |
| 8 | ORN | HG2 | 9 | TYR | HN | 5.50 |
| 8 | ORN | HG2 | 9 | TYR | QD | 7.64 |
| 8 | ORN | HG2 | 9 | TYR | QE | 7.63 |


| 8 | ORN | HG3 | 9 | TYR | HN | 5.50 |
| ---: | :--- | :--- | :--- | :--- | :--- | :--- |
| 8 | ORN | HG3 | 9 | TYR | QD | 7.64 |
| 8 | ORN | HG3 | 9 | TYR | QE | 7.63 |
| 8 | ORN | HD2 | 9 | TYR | QE | 7.63 |
| 8 | ORN | HD3 | 9 | TYR | QE | 7.63 |
| 9 | TYR | HN | 9 | TYR | HB2 | 2.90 |
| 9 | TYR | HN | 9 | TYR | HB3 | 2.90 |
| 9 | TYR | HN | 9 | TYR | QE | 7.63 |
| 9 | TYR | HA | 9 | TYR | HB2 | 3.08 |
| 9 | TYR | HA | 9 | TYR | HB3 | 3.08 |
| 9 | TYR | HA | 10 | CYSS | HN | 3.11 |
| 10 | CYSS | HA | 11 | VAL | HN | 2.74 |
| 10 | CYSS | $H A$ | 11 | VAL | QQG | 8.09 |
| 10 | CYSS | QB | 11 | VAL | HN | 4.08 |
| 10 | CYSS | QB | 11 | VAL | QG1 | 7.40 |
| 10 | CYSS | QB | 11 | VAL | QG2 | 7.40 |
| 11 | VAL | HN | 11 | VAL | HB | 3.95 |
| 11 | VAL | HA | 11 | VAL | HB | 2.86 |


| Peptide 16 |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 4 | LYS | HA | 4 | LYS | HG2 | 3.70 |
| 4 | LYS | HA | 4 | LYS | HG3 | 3.70 |
| 4 | LYS | HA | 4 | LYS | QG | 3.52 |
| 4 | LYS | HA | 5 | PEN | HN | 2.80 |
| 4 | LYS | HB2 | 5 | PEN | HN | 3.58 |
| 4 | LYS | HB3 | 5 | PEN | HN | 3.58 |
| 4 | LYS | QB | 5 | PEN | HN | 3.39 |
| 4 | LYS | HG2 | 4 | LYS | HE2 | 4.29 |
| 4 | LYS | HG2 | 4 | LYS | HE3 | 4.29 |
| 4 | LYS | HG2 | 4 | LYS | QZ | 6.44 |
| 4 | LYS | HG3 | 4 | LYS | HE2 | 4.29 |
| 4 | LYS | HG3 | 4 | LYS | HE3 | 4.29 |
| 4 | LYS | HG3 | 4 | LYS | QZ | 6.44 |
| 4 | LYS | QG | 5 | PEN | HN | 6.38 |
| 5 | PEN | HN | 5 | PEN | QG1 | 3.90 |
| 5 | PEN | HA | 6 | PHE | HN | 3.58 |
| 5 | PEN | HA | 10 | CYSS | HA | 3.02 |
| 5 | PEN | HA | 10 | CYSS | QB | 5.54 |
| 5 | PEN | HA | 11 | VAL | HN | 3.70 |
| 5 | PEN | QG2 | 6 | PHE | HN | 4.73 |
| 5 | PEN | QG2 | 7 | DTRP | HN | 5.69 |
| 5 | PEN | QG2 | 9 | TYR | HN | 5.16 |
| 5 | PEN | QG2 | 10 | CYSS | HN | 6.53 |
| 5 | PEN | QG2 | 10 | CYSS | HA | 5.16 |
| 6 | PHE | HN | 6 | PHE | HB2 | 3.73 |
| 6 | PHE | HN | 6 | PHE | HB3 | 3.73 |
| 6 | PHE | HN | 6 | PHE | QB | 3.50 |
| 6 | PHE | HN | 9 | TYR | HN | 5.50 |
| 6 | PHE | HN | 10 | CYSS | HA | 3.89 |
| 6 | PHE | HA | 6 | PHE | HB2 | 2.96 |
| 6 | PHE | HA | 6 | PHE | HB3 | 2.96 |
| 6 | PHE | HA | 6 | PHE | QB | 2.76 |
| 6 | PHE | HA | 7 | DTRP | HN | 2.99 |
| 7 | DTRP | HN | 7 | DTRP | HB2 | 3.08 |
| 7 | DTRP | HN | 7 | DTRP | HB3 | 3.48 |
| 7 | DTRP | HN | 7 | DTRP | HE3 | 5.50 |
| 7 | DTRP | HA | 7 | DTRP | HB2 | 3.02 |
| 7 | DTRP | HA | 7 | DTRP | HE3 | 3.39 |
| 7 | DTRP | HA | 8 | ORN | HN | 2.90 |
|  | DTRP | HA | 9 | TYR | HN | 3.55 |


| 7 | DTRP | HB2 | 7 | DTRP | HD1 | 3.17 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 7 | DTRP | HB2 | 8 | ORN | HN | 3.89 |
| 7 | DTRP | HB3 | 7 | DTRP | HD1 | 3.42 |
| 7 | DTRP | HB3 | 8 | ORN | HN | 4.32 |
| 7 | DTRP | HE3 | 8 | ORN | HN | 4.57 |
| 8 | ORN | HN | 8 | ORN | HB2 | 3.55 |
| 8 | ORN | HN | 8 | ORN | HB3 | 3.55 |
| 8 | ORN | HN | 8 | ORN | QB | 3.30 |
| 8 | ORN | HN | 8 | ORN | HG2 | 5.04 |
| 8 | ORN | HN | 8 | ORN | HG3 | 5.04 |
| 8 | ORN | HN | 8 | ORN | QG | 4.73 |
| 8 | ORN | HN | 9 | TYR | HN | 3.21 |
| 8 | ORN | HA | 8 | ORN | QG | 3.97 |
| 8 | ORN | HA | 9 | TYR | HN | 3.58 |
| 8 | ORN | HB2 | 9 | TYR | QE | 7.63 |
| 8 | ORN | HB3 | 9 | TYR | QE | 7.63 |
| 8 | ORN | QB | 9 | TYR | QE | 7.30 |
| 8 | ORN | HG2 | 9 | TYR | HN | 5.50 |
| 8 | ORN | HG2 | 9 | TYR | QE | 7.63 |
| 8 | ORN | HG3 | 9 | TYR | HN | 5.50 |
| 8 | ORN | HG3 | 9 | TYR | QE | 7.63 |
| 8 | ORN | QG | 9 | TYR | QD | 8.52 |
| 8 | ORN | HD2 | 9 | TYR | QE | 7.63 |
| 8 | ORN | HD3 | 9 | TYR | QE | 7.63 |
| 9 | TYR | HN | 9 | TYR | HB2 | 3.08 |
| 9 | TYR | HN | 9 | TYR | HB3 | 3.08 |
| 9 | TYR | HN | 9 | TYR | QE | 7.63 |
| 9 | TYR | HN | 10 | CYSS | HN | 5.28 |
| 9 | TYR | HA | 10 | CYSS | HN | 3.08 |
| 10 | CYSS | HA | 11 | VAL | HN | 3.08 |
| 10 | CYSS | QB | 11 | VAL | HN | 4.30 |
| 10 | CYSS | QB | 11 | VAL | HA | 5.20 |
| 11 | VAL | HA | 11 | VAL | HB | 2.86 |



Figure S2. Results of MD simulations of $h$-UTR complexed with Urantide. Graphic shows plot of the monitored distance, in the complex, between the protonated $\mathrm{N}^{\varepsilon}$ of $\mathrm{Orn}^{8}$ in Urantide and the $\mathrm{O}^{\delta}$ of Asp130 in $h$-UTR.


Figure S3. Distance between the OH oxygen of $\mathrm{Tyr}^{9}$ in Urantide and the side chain CO oxygen of Asn297 in $h$-UTR.


Figure S4. Distance between the OH oxygen of Tyr ${ }^{9}$ in Urantide and the OH hydrogen of Thr301 in $h$-UTR.


Figure S5. Distance between the the $\mathrm{O}^{\delta}$ of $\mathrm{Asp}^{4}$ in Urantide and the guanidinium group $\mathrm{N}^{\eta}$ of Arg206 in $h$-UTR.


Figure S6. Distance between the protonated backbone $\mathrm{NH}_{3}$ nitrogen of $\mathrm{Asp}^{4}$ in Urantide and the backbone CO oxygen of Ala187 in $h$-UTR.


Figure S7. Distance between the protonated backbone $\mathrm{NH}_{3}$ nitrogen of $\mathrm{Asp}^{4}$ in Urantide and the backbone CO oxygen of Cys199 in $h$-UTR.


Figure S8. Distance between the protonated backbone $\mathrm{NH}_{3}$ nitrogen of $\mathrm{Asp}^{4}$ in Urantide and the backbone CO oxygen of Met188 in $h$-UTR.


Figure S9. Distance between the negatively charged carboxyl group of Val ${ }^{11}$ in Urantide and the backbone NH hydrogen of Cys 123 in $h$-UTR.


Figure S10. Distance between the negatively charged carboxyl group of Val ${ }^{11}$ in Urantide and the backbone NH hydrogen of Cys199 in $h$-UTR.


Figure S11. Distance between the negatively charged carboxyl group of Val ${ }^{11}$ in Urantide and the backbone NH hydrogen of Arg 189 in $h$-UTR.


Figure S12. Distance between the protonated $\mathrm{N}^{\zeta}$ of $\mathrm{Lys}^{8}$ in P5U and the $\mathrm{O}^{\delta}$ of Asp130 in $h$-UTR.


Figure S13. Distance between the the $\mathrm{O}^{\delta}$ of Asp ${ }^{4}$ in P5U and the side chain $\mathrm{NH}_{2}{ }^{\text {® }}$ hydrogen of Gln285 in $h$-UTR.


Figure S14. (a) Stereoview of Urantide $/ h-$ UTR $_{\mathrm{a}}$ (a) and P5U/ $h-\mathrm{UTR}_{\mathrm{i}}$ (b) model complex. Urantide and P5U heavy atoms are colour coded (carbon, green; nitrogen, blue; oxygen, red; sulfur, yellow). Receptor backbones are represented in azure.

| $n$-UTR | MALTPESPSSFPGLAATGSSVPEPPGGPNATLNSSWASPTEPSSLEDLVATGTIGTLLSA 60 |
| :---: | :---: |
| $r$-UTR | MALSLESTTSFHMLTVSGSTVTELPGDSNVSLNSSWSGPTDPSSLKDLVATGVIGAVLSA 60 ***: **.:** *:.:**:*.* **...*.:*****:.**:****:******.**::*** |
| $n$-UTR | MGVVGVVGNAYTLVVTCRSLRAVASMYVYVVNLALADLLYLLSIPFIVATYVTKEWHFGD 120 |
| $r$-UTR | MGVVGMVGNVYTLVVMCRFLRASASMYVYVVNLALADLLYLLSIPFIIATYVTKDWHFGD 120 |
| $n$-UTR | VGCRVLFGLDFLTMHASIFTLTVMSSERYAAVLRPLDTVQRPKGYRKLLALGTWLLALLL 180 |
| $r$-UTR | VGCRVLFSLDFLTMHASIFTLTIMSSERYAAVLRPLDTVQRSKGYRKLLVLGTWLLALLL 180 <br> *******.**************:******************.*******.********** |
|  | EL-II |
| $n$-UTR | TLPVMLAMRLVRRGPKSLCLPAWGPRAHRAYLTLLFATSIAGPGLLIGLLYARLARAYRR 240 |
| $r$-UTR | TLPMMLAIQLVRRGSKSLCLPAWGPRAHRTYLTLLFGTSIVGPGLVIGLLYVRLARAYWL 240 <br> ***: ***: : *****.**************:******.***.****:*****.****** |
| $n$-UTR | SQRASFKRARR-PGARALRLVLGIVLLFWACFLPFWLWQLLAQYHQA-PLAPRTARIVNY 298 |
| $r$-UTR | SQQASFKQTRRLPNPRVLYLILGIVLLFWACFLPFWLWQLLAQYHEAMPLTPETARIVNY 300 <br> **: ****:: ** *...*.* *: ************************:* **:*.******* |
| $n$-UTR | LTTCLTYGNSCANPFLYTLLTRNYRDHLRGRVRGPGSGGGRGPVPS--LQPRARFQRCSG 356 |
| $r$-UTR | LTTCLTYGNSCINPFLYTLLTKNYREYLRGRQRSLGSSCHSPGSPGSFLPSRVHLQQDSG 360 $\star \star \star * * * * * * * * * * * * * * * *: * * *:: * * * * * * * . ~$ |
| $n$-UTR | RSLSSCSPQPTDSLVLAPAAPARPAPEGPRAPA 389 |
| $r$-UTR | RSLSSSSQQATETLMLSPVPRNGALL------- 386 |
|  | *****.* *.*::*:*:*.. |

Figure S15. Pairwise alignment of h-UTR and r-UTR. The conserved key residues used to align the sequences are shown in bold. In all sequence alignment figures, an asterisk (*) indicates an identical amino acid; punctuations indicate a "conserved" amino acid, which meets the criteria for either highly conservative substitutions (:) or semiconservative substitutions (.), as defined by CLUSTALW.


Figure S16. Stereoview of the TM domains of the $h-\mathrm{UTR}_{\mathrm{i}}$ model (azure) superimposed to $\beta_{2} \mathrm{AR}$ crystal structure (gold, PDB code 2RH1). Backbone heavy atoms of TM-II $\div$ TM-VII were used for the superimposition. Bound urantide is shown as green surface.

## CSI MTII



Figure S17. Chemical shift deviations from the random coil shift values (reference [110] of the manuscript) for $\mathrm{H} \alpha$ resonances at $25^{\circ} \mathrm{C}$.


Figure S18. Results of MD simulations of $h-\mathrm{MC}_{\mathrm{a}} \mathrm{R}_{\mathrm{a}}$ complexed with MTII. Graphic shows plot of the monitored distances, in the complex, between the protonated guanidinium $\mathrm{N}^{\varepsilon}$ of $\mathrm{Arg}^{8}$ in MTII and the $\mathrm{O}^{\varepsilon}$ of Glu100 (a), $\mathrm{O}^{\delta}$ of Asp122 (b), Asp126 (c), and between $\mathrm{N}^{\varepsilon}$ of $\mathrm{His}^{6}$ and the $\mathrm{O}^{\varepsilon}$ of Glu100 (d).


Figure S19. Results of MD simulations of $h$ - MC4R $_{\mathrm{i}}$ complexed with SHU9119. Graphic shows plot of the monitored distances, in the complex, between the protonated guanidinium $\mathrm{N}^{\varepsilon}$ of $\mathrm{Arg}^{8}$ in SHU9119 and the $\mathrm{O}^{\varepsilon}$ of Glu100 (a), $\mathrm{O}^{\delta}$ of Asp122 (b), Asp126 (c), and between $\mathrm{N}^{\varepsilon}$ of $\mathrm{His}^{6}$ and the $\mathrm{O}^{\varepsilon}$ of Glu100 (d).


Figure S20. Stereoview of $h$-MC4R models in the "active" state proposed by Chapman et al. (violet ribbon; reference [129] of the manuscript) and that proposed by Mosberg et al. (grey ribbon; reference [114] of the manuscript). The $h$-MC4R models were superimposed using the backbone heavy atoms of TM residues.

Table S5. Analytical data of synthesized peptides MTII and SHU9119.

| Code | M.W. | MS | HPLC $^{\text {a,b }}$ |
| :---: | :---: | :---: | :---: |
| MT-II | 1024.22 | 1024.87 | 4.15 |
| SHU9119 | 1074.28 | 1075.12 | 4.31 |

${ }^{a}$ HPLC column, Vydac 218TP1010, $1.0 \times 25 \mathrm{~cm}$, using a gradient of $\mathrm{CH}_{3} \mathrm{CN}$ in $0.1 \%$ aqueous TFA (from 10 to $90 \%$ in 30 min ) at a flow rate of $1.0 \mathrm{~mL} / \mathrm{min}$..
${ }^{b} k^{\prime}=[($ peptide retention time - solvent retention time $) /$ solvent retention time $]$.

Table S6. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide MTII in $\mathrm{H}_{2} \mathrm{O}$ Solution at $25^{\circ} \mathrm{C}$.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha N}, \text { exc, }-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| Nleu ${ }^{4}$ | 8.22 (6.6, f, 9.1) | 4.20 | 1.65 | $1.26(\gamma) ; 0.85(\varepsilon)$ |
| Asp ${ }^{5}$ | 8.52 (7.5, f, 8.3) | 4.63 | 2.89, 2.68 |  |
| His ${ }^{6}$ | 8.46 (7.0, f, 6.5) | 4.40 | 3.16, 3.01 | 7.03(8); $8.33(\varepsilon)$ |
| $D P h e^{7}$ | 8.39 (7.1, f, 8.3) | 4.57 | 3.15, 2.88 | 7.22(8); $7.35(\varepsilon)$ |
| Arg ${ }^{8}$ | 7.89 (7.2, f, 6.3) | 4.27 | 1.61, 1.57 | 1.32( $\gamma$ ); 3.08( $\delta$ ); 7.10( $\varepsilon$ ) |
| Trp ${ }^{9}$ | 8.44 (6.7, f, 8.7) | 4.65 | 3.31 | 7.27(ס); 10.15, 7.69(ع); 7.50, 7.33(弓); |
|  |  |  |  | 7.18( 7 ) |
| Lys ${ }^{10}$ | 8.06 (8.1, f, 7.3) | 4.22 | 1.75, 1.60 | 1.31, 1.23( $\gamma$ ); 1.51, 1.41( $\delta$ ); 3.22, |
|  |  |  |  | 3.17(8); $7.91(\zeta)$ |

[^1]Table S7. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide SHU9119 in $\mathrm{H}_{2} \mathrm{O}$ Solution at $25^{\circ} \mathrm{C}$.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}}, \mathrm{exc},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
| $\mathrm{Nleu}^{4}$ | $8.18(5.7, \mathrm{f}, 8.3)$ | 4.15 | 1.60 | $1.22(\gamma) ; 0.83(\varepsilon)$ |
| $\mathrm{Asp}^{5}$ | $8.46(7.3, \mathrm{f}, 8.1)$ | 4.61 | $2.82,2.63$ |  |
| $\mathrm{His}^{6}$ | $8.41(7.2, \mathrm{f}, 6.0)$ | 4.45 | $3.14,2.96$ | $6.95(\delta) ; 8.04(\varepsilon)$ |
| $\mathrm{DNal}^{7}$ | $8.53(7.0, \mathrm{f}, 8.3)$ | 4.70 | $3.29,3.06$ | $7.65,7.38(\delta) ; 7.87(\varepsilon)$ |
| $\mathrm{Arg}^{8}$ | $7.93(7.1, \mathrm{f}, 6.8)$ | 4.18 | $1.48,1.39$ | $1.05,1.01(\gamma) ; 2.83(\delta) ; 6.90(\varepsilon)$ |
| $\mathrm{Trp}^{9}$ | $8.39(6.1, \mathrm{f}, 9.1)$ | 4.64 | $3.31,3.27$ | $7.26(\delta) ; 10.13,7.26(\varepsilon) ; 7.49,7.33(\zeta) ; 7.19(\eta)$ |
| $\mathrm{Lys}^{10}$ | $7.99(8.8, \mathrm{f}, 6.3)$ | 4.21 | $1.74,1.59$ | $1.49(\gamma) ; 1.29,1.19(\delta) ; 3.18(\varepsilon) ; 7.91(\zeta)$ |

[^2]Table S8. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide MTII in $\mathrm{H}_{2} \mathrm{O}$ Solution at $5^{\circ} \mathrm{C}$.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}}, \operatorname{exc},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| Nleu ${ }^{4}$ | 8.41 (6.0, f, 9.0) | 4.18 | 1.64 | 1.23( $\gamma$ ); 0.83(8) |
| Asp ${ }^{5}$ | 8.70 (7.5, f, 8.4) | 4.63 | 2.90, 2.68 |  |
| His ${ }^{6}$ | 8.57 (6.4, f, 6.2) | 4.38 | 3.16, 3.04 | 7.02(8); 8.32(ع) |
| $D \mathrm{Phe}^{7}$ | 8.58 (6.8, f, 6.3) | 4.13 | 3.16, 2.87 | 7.21(8); 7.34(8) |
| Arg ${ }^{8}$ | 8.00 (7.3, f, 6.2) | 4.28 | 1.62, 1.56 | 1.29(\%); 3.07( $\delta$ ); 7.17( $\varepsilon$ ) |
| Trp ${ }^{9}$ | 8.63 (6.4, f, 8.8) | 4.65 | 3.30 | 7.27(ס); 10.24, 7.67(ع); 7.49, 7.33 (弓); |
|  |  |  |  | 7.17( $\dagger$ ) |
| Lys ${ }^{10}$ | 8.17 (8.1, f, 7.4) | 4.19 | 1.72, 1.58 | 1.31, 1.21( $\gamma$ ); 1.51, 1.39( $\delta$; 3.23, |
|  |  |  |  | 3.15(8); 8.10(弓) |

${ }^{a}$ Obtained at $\mathrm{pH}=5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz . exc $=\mathrm{NH}$ exchange rate ( f , fast; ms, moderately slow; s, slow; ). $-\Delta \delta / \Delta T=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ) calculated in the range $5-15^{\circ} \mathrm{C}$. Further signals: $\mathrm{C}_{3} \underline{\mathrm{CO}}, 1.99 ; \mathrm{CONH}_{2}, 6.52,6.94$.

Table S9. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide SHU9119 in $\mathrm{H}_{2} \mathrm{O}$ Solution at $5^{\circ} \mathrm{C}$.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}}, \mathrm{exc},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
| $\mathrm{Nleu}^{4}$ | $8.34(5.7, \mathrm{f}, 8.3)$ | 4.14 | 1.59 | $1.23(\gamma) ; 0.82(\varepsilon) ;$ |
| $\mathrm{Asp}^{5}$ | $8.61(7.2, \mathrm{f}, 8.1)$ | 4.60 | $2.83,2.63$ |  |
| $\mathrm{His}^{6}$ | $8.53(6.5, \mathrm{f}, 6.0)$ | 4.44 | $3.14,2.95$ | $6.93(\delta) ; 8.04(\varepsilon)$ |
| $\mathrm{DNal}^{7}$ | $8.73(6.7, \mathrm{f}, 6.5)$ | 4.70 | $3.29,3.04$ | $7.66,7.40(\delta) ; 7.85(\varepsilon)$ |
| $\mathrm{Arg}^{8}$ | $8.10(7.1, \mathrm{f}, 6.7)$ | 4.16 | $1.48,1.39$ | $1.05,0.97(\gamma) ; 2.82(\delta) ; 6.95(\varepsilon)$ |
| $\mathrm{Trp}^{9}$ | $8.55(6.2, \mathrm{f}, 9.1)$ | 4.63 | $3.29,3.27$ | $7.25(\delta) ; 10.21(\varepsilon) ; 7.47,7.33(\zeta) ; 7.16(\eta)$ |
| $\mathrm{Lys}^{10}$ | $8.09(8.2, \mathrm{f}, 6.3)$ | 4.19 | $1.74,1.59$ | $1.49(\gamma) ; 1.29,1.19(\delta) ; 3.15(\varepsilon) ; 8.09(\zeta)$ |

${ }^{\text {a }}$ Obtained at $\mathrm{pH}=5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz. exc $=\mathrm{NH}$ exchange rate (f, fast; ms, moderately slow; s , slow;). $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ) calculated in the range $5-15{ }^{\circ} \mathrm{C}$. Further signals: $\mathrm{CH}_{3} \mathrm{CO}, 1.96 ; \mathrm{CONH}_{2}, 6.73,6.96$.

Table S10. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide MTII in $\mathrm{H}_{2} \mathrm{O} /$ DMSO Solution at $-10^{\circ} \mathrm{C}$.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}}, \text { exc, }-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
| $\mathrm{Nleu}^{4}$ | $8.43(6.4, \mathrm{f}, 9.0)$ | 4.18 | 1.60 | $1.21(\gamma) ; 0.83(\varepsilon)$ |
| Asp $^{5}$ | $8.71(6.8, \mathrm{f}, 8.2)$ | 4.60 | $2.90,2.66$ |  |
| $\mathrm{His}^{6}$ | $8.59(6.4, \mathrm{f}, 6.2)$ | 4.33 | $3.17,3.01$ | $7.01(\delta) ; 8.38(\varepsilon)$ |
| $D P R e^{7}$ | $8.56(7.0, \mathrm{f}, 6.3)$ | 4.52 | $3.16,2.86$ | $7.22(\delta) ; 7.33(\varepsilon)$ |
| $\mathrm{Arg}^{8}$ | $7.99(7.0, \mathrm{f}, 6.6)$ | 4.30 | $1.64,1.58$ | $1.33(\gamma) ; 3.08(\delta) ; 7.25(\varepsilon)$ |
| $\mathrm{Trp}^{9}$ | $8.64(6.0, \mathrm{f}, 8.7)$ | 4.63 | $3.30,3.26$ | $7.27(\delta) ; 10.35,7.68(\varepsilon) ; 7.47,7.16(\zeta) ; 7.23(\eta)$ |
| $\mathrm{Lys}^{10}$ | $8.25(8.1, \mathrm{f}, 7.3)$ | 4.18 | $1.74,1.58$ | $1.32,1.22(\gamma) ; 1.49,1.37(\delta) ; 3.19(\varepsilon) ; 8.11(\zeta)$ |

${ }^{\text {a }}$ Obtained at $-10{ }^{\circ} \mathrm{C}$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz . exc $=\mathrm{NH}$ exchange rate ( f , fast; ms, moderately slow; s, slow; ). $-\Delta \delta / \Delta T=$ temperature coefficients $(\mathrm{ppb} / \mathrm{K})$ calculated in the range -10 to $0{ }^{\circ} \mathrm{C}$. Further signals: $\mathrm{CH}_{3} \mathrm{CO}, 1.97 ; \mathrm{CONH}_{2}, 6.59,7.01$.

Table S11. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide SHU9119 in $\mathrm{H}_{2} \mathrm{O} /$ DMSO Solution.

| residue | NH ( ${ }^{3} J_{\alpha \mathrm{N}}$, exc, $\left.-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| Nleu ${ }^{4}$ | 8.38 (5.8, f, 8.4) | 4.14 | 1.56 | 1.18( $\gamma$ ); 0.79( ) $^{\text {; }}$ |
| $\mathrm{Asp}^{5}$ | 8.65 (6.7, f, 8.1) | 4.58 | 2.85, 2.62 |  |
| His ${ }^{6}$ | 8.57 (6.4, f, 6.0) | 4.36 | 3.15, 2.96 | 6.93( $\delta$ ); 8.07(8) |
| $D \mathrm{Nal}^{7}$ | 8.70 (6.7, f, 6.5) | 4.67 | 3.30, 3.04 | 7.66, 7.39(8); 7.86(8) |
| Arg ${ }^{8}$ | 8.07 (7.0, f, 6.7) | 4.21 | 1.53, 1.48 | 1.11, 1.08( $\gamma$ ); 2.90( $\delta$ ); 7.09(8) |
| Trp ${ }^{9}$ | 8.60 (6.1, f, 9.1) | 4.63 | 3.29, 3.24 | 7.23(8); 10.33, 7.26(ع); 7.46, 7.15(弓); 7.23(ך) |
| Lys ${ }^{10}$ | 8.17 (8.3, f, 6.3) | 4.19 | 1.73, 1.58 | 1.24( $\gamma$ ); 1.48, 1.37( $\delta$ ); 3.17( $\varepsilon$ ); 8.10( ) $^{(1)}$ |

${ }^{\text {a }}$ Obtained at $-10{ }^{\circ} \mathrm{C}$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz . exc $=\mathrm{NH}$ exchange rate (f, fast; ms, moderately slow; s, slow;). $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ) calculated in the range -10 to $0{ }^{\circ} \mathrm{C}$. Further signals: $\mathrm{C}_{\underline{3}} \mathrm{CO}, 1.94 ; \mathrm{CONH}_{2}, 6.66,7.02$.

Table S12. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide MTII in DPC 200 mM Solution.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}}, \operatorname{exc},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}\left(\mathrm{J}_{\alpha \beta(\mathrm{l})} \mathrm{J}_{\alpha \beta(\mathrm{h})}\right)$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| Nleu ${ }^{4}$ | 8.23 (5.8, f, 7.6) | 4.22 (8.6, 7.3) | 1.69, 1.63 | 1.23( $\gamma$ ) ${ }^{\text {1.30( }}$ ( $) ; 0.82(\varepsilon) ;$ |
| Asp ${ }^{5}$ | 8.70 (5.5, f, 5.0) | 4.40 (6.1, 8.5) | 3.13, 2.47 |  |
| His ${ }^{6}$ | 8.84 (6.0, f, 4.7) | 4.14 (6.3, 7.2) | 3.29, 3.15 | 7.12(8); 8.51( $\varepsilon$ ) |
| $D$ Phe $^{7}$ | 8.21 (7.0, ms, 3.2) | 4.46 (9.5, 5.2) | 3.16, 2.92 | 7.19(8); 7.22(8); 7.16(弓) |
| Arg ${ }^{8}$ | 7.49 (8.2, s, 1.4) | 4.54 (ov.) | 1.73, 1.70 | 1.52, $\quad 1.47(\gamma) ; \quad 3.15(\delta) ;$ |
|  |  |  |  | 7.23(8) |
| Trp ${ }^{9}$ | 8.72 (7.0, f, 8.9) | 4.75 (9.0, 5.8) | 3.33, 3.14 | 7.21(8); 10.63, 7.61(ع); 7.46, |
|  |  |  |  | 6.99(弓); 7.07( $\eta$ ) |
| Lys ${ }^{10}$ | 8.19 (8.1, f, 5.3) | 4.16 (5.2, 9.1) | 1.73, 1.53 | 1.35, 1.28( $\gamma$ ); 1.42( $\delta$ ); 3.20, |
|  |  |  |  | 3.06(ع); $7.95(\zeta)$ |

${ }^{\text {a }}$ Obtained at $25^{\circ} \mathrm{C}, \mathrm{pH}=5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\text {b }}{ }^{3} J_{\alpha \mathrm{N}}$ and ${ }^{3} J_{\alpha \beta}$ coupling constants in Hz. exc $=\mathrm{NH}$ exchange rate ( f , fast; ms, moderately slow; s, slow; ). $-\Delta \delta / \Delta T=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ) calculated in the range $25-40^{\circ} \mathrm{C}$. The subscripts (1) and (h) denote the coupling constant of the low- and high-field $\mathrm{H}_{\beta}$ signal, respectively. Further signals: $\mathrm{CH}_{3} \mathrm{CO}, 1.93$; $\mathrm{CONH}_{2}, 6.37,7.08$.

Table S13. NMR Resonance Assignments ${ }^{\text {a }}$ of Peptide SHU9119 in DPC 200 mM Solution

| residue | NH ( ${ }^{3} J_{\alpha \mathrm{N}}$, exc, $\left.-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}\left(\mathrm{J}_{\alpha \beta(\mathrm{l})} \mathrm{J}_{\alpha \beta(\mathrm{h})}\right)$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| Nleu ${ }^{4}$ | 8.18 (5.6, f, 7.8) | 4.26 (8.6, 7.4) | 1.69, 1.63 | 1.26( $\gamma$ ); 1.18(8); 0.78(ع); |
| Asp ${ }^{5}$ | 8.73 (5.4, f, 5.7) | 4.42 (6.0, 8.5) | 3.13, 2.49 |  |
| His ${ }^{6}$ | 8.87 (6.1, f, 4.5) | 4.16 (6.5, 7.3) | 3.30, 3.16 | 7.12(8); 8.39(E) |
| $D \mathrm{Nal}^{7}$ | 8.38 (6.9, ms, 3.2) | 4.57 (8.5, 7.2) | 3.31, 3.11 | $\begin{aligned} & 7.65,7.35(\delta) ; 7.75(\varepsilon) ; 7.80(\zeta) \\ & 7.80,7.79(\eta) ; 7.43(\theta) \end{aligned}$ |
| Arg ${ }^{8}$ | 7.53 (8.2, s, 1.5) | 4.55 (6.8, 8.0) | 1.76, 1.69 | 1.52, $1.47(\gamma) ; 3.12(\delta) ; 7.23(\varepsilon)$ |
| Trp ${ }^{9}$ | 8.74 (7.1, f, 8.5) | 4.77 (9.0, 5.8) | 3.35, 3.15 | $\begin{aligned} & 7.23(\delta) ; 10.65, \quad 7.63(\varepsilon) ; 7.47 \\ & 7.01(\zeta) ; 7.08(\eta) \end{aligned}$ |
| Lys ${ }^{10}$ | 8.22 (8.1, f, 5.8) | 4.18 (5.2, 9.0) | 1.74, 1.55 | $\begin{aligned} & 1.31(\gamma) ; \quad 1.44, \quad 1.37(\delta) ; \quad 3.22 \\ & 3.07(\varepsilon) ; 7.97(\zeta) \end{aligned}$ |

${ }^{\text {a }}$ Obtained at $25^{\circ} \mathrm{C}, \mathrm{pH}=5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ and ${ }^{3} J_{\alpha \beta}$ coupling constants in Hz. exc $=\mathrm{NH}$ exchange rate (f, fast; ms, moderately slow; s, slow;). $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ) calculated in the range $25-40^{\circ} \mathrm{C}$. The subscripts (l) and (h) denote the coupling constant of the low- and high-field $\mathrm{H}_{\beta}$ signal, respectively. Further signals: $\mathrm{C}_{3} \underline{3}_{3} \mathrm{CO}, 1.90$; $\mathrm{CONH}_{2}, 6.40,7.10$.

Table S14. NOE Derived Upper Limit Constraints of MTII in DPC solution

| 3 | ACE | QH | 5 | ASP | HN | 6.19 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 4 | NLE | HN | 4 | NLE | HA | 2.93 |
| 4 | NLE | HN | 4 | NLE | HB2 | 3.92 |
| 4 | NLE | HN | 4 | NLE | HB3 | 3.92 |
| 4 | NLE | HN | 4 | NLE | QB | 2.84 |
| 4 | NLE | HN | 4 | NLE | QG | 5.27 |
| 4 | NLE | HN | 4 | NLE | QD | 6.38 |
| 4 | NLE | HN | 5 | ASP | HN | 2.99 |
| 4 | NLE | HA | 4 | NLE | HB2 | 2.96 |
| 4 | NLE | HA | 4 | NLE | HB3 | 2.96 |
| 4 | NLE | HA | 4 | NLE | QD | 5.02 |
| 4 | NLE | HA | 4 | NLE | QE | 6.53 |
| 4 | NLE | HA | 5 | ASP | HN | 3.50 |
| 4 | NLE | HA | 6 | HIS | HN | 4.76 |
| 4 | NLE | HB2 | 4 | NLE | QE | 6.53 |
| 4 | NLE | HB2 | 5 | ASP | HN | 3.48 |
| 4 | NLE | HB2 | 6 | HIS | HN | 4.72 |
| 4 | NLE | HB3 | 4 | NLE | QE | 6.53 |
| 4 | NLE | HB3 | 5 | ASP | HN | 3.48 |
| 4 | NLE | HB3 | 6 | HIS | HN | 4.72 |
| 4 | NLE | QB | 4 | NLE | QE | 6.41 |
| 4 | NLE | QB | 5 | ASP | HN | 3.26 |
| 4 | NLE | QB | 6 | HIS | HN | 4.50 |
| 4 | NLE | QG | 5 | ASP | HN | 6.32 |
| 4 | NLE | QG | 6 | HIS | HN | 6.38 |
| 4 | NLE | QG | 7 | DPHE | QD | 8.50 |
| 4 | NLE | QG | 9 | TRP | HE3 | 6.38 |
| 4 | NLE | QG | 9 | TRP | HZ3 | 6.38 |
| 4 | NLE | QD | 5 | ASP | HN | 6.38 |
| 4 | NLE | QD | 9 | TRP | HE3 | 6.38 |
| 4 | NLE | QE | 7 | DPHE | QD | 8.65 |
| 4 | NLE | QE | 9 | TRP | HE3 | 6.53 |
| 5 | ASP | HN | 5 | ASP | HA | 2.55 |
| 5 | ASP | HN | 6 | HIS | HN | 3.17 |
| 5 | ASP | HN | 10 | LYS | HZ1 | 5.50 |
| 5 | ASP | HA | 5 | ASP | HB2 | 2.99 |
| 5 | ASP | HA | 5 | ASP | HB3 | 2.99 |
| 5 | ASP | HA | 6 | HIS | HN | 3.60 |
| 5 | ASP | HA | 7 | DPHE | HN | 5.00 |
| 5 | ASP | HA | 10 | LYS | HZ1 | 5.13 |
| 5 | ASP | HB2 | 10 | LYS | HE2 | 6.85 |
| 5 | ASP | HB2 | 10 | LYS | HE3 | 6.85 |
| 5 | ASP | HB2 | 10 | LYS | HZ1 | 3.11 |
| 5 | ASP | HB3 | 10 | LYS | HE2 | 6.85 |
| 5 | ASP | HB3 | 10 | LYS | HE3 | 6.85 |
| 5 | ASP | HB3 | 10 | LYS | HZ1 | 3.11 |
| 5 | ASP | QB | 8 | ARG | HN | 5.50 |
| 5 | ASP | QB | 10 | LYS | QB | 7.25 |
| 5 | ASP | QB | 10 | LYS | QE | 5.90 |
| 5 | ASP | QB | 10 | LYS | HZ1 | 2.88 |
| 6 | HIS | HN | 6 | HIS | HA | 2.68 |
| 6 | HIS | HN | 6 | HIS | HB2 | 3.42 |
| 6 | HIS | HN | 6 | HIS | HB3 | 3.42 |
| 6 | HIS | HN | 6 | HIS | HD2 | 5.25 |
| 6 | HIS | HN | 7 | DPHE | HN | 3.21 |
| 6 | HIS | HN | 8 | ARG | HN | 4.63 |
| 6 | HIS | HA | 6 | HIS | HB2 | 2.77 |
| 6 | HIS | HA | 6 | HIS | HB3 | 2.77 |
| 6 | HIS | HA | 6 | HIS | HD2 | 4.60 |


| 6 | HIS | HA | 7 | DPHE | HN | 2.86 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 6 | HIS | HA | 7 | DPHE | QD | 7.62 |
| 6 | HIS | HA | 8 | ARG | HN | 4.88 |
| 6 | HIS | QB | 7 | DPHE | HN | 4.86 |
| 7 | DPHE | HN | 7 | DPHE | HB2 | 2.83 |
| 7 | DPHE | HN | 7 | DPHE | HB3 | 2.83 |
| 7 | DPHE | HN | 7 | DPHE | QB | 2.63 |
| 7 | DPHE | HN | 8 | ARG | HN | 3.50 |
| 7 | DPHE | HA | 8 | ARG | HN | 3.14 |
| 8 | ARG | HN | 9 | TRP | HN | 4.58 |
| 8 | ARG | HA | 8 | ARG | QB | 2.76 |
| 8 | ARG | HA | 8 | ARG | HG2 | 4.04 |
| 8 | ARG | HA | 8 | ARG | HG3 | 4.04 |
| 8 | ARG | HA | 9 | TRP | HN | 2.55 |
| 8 | ARG | HB2 | 9 | TRP | HN | 3.24 |
| 8 | ARG | HB3 | 9 | TRP | HN | 3.24 |
| 8 | ARG | HG2 | 9 | TRP | HN | 5.25 |
| 8 | ARG | HG3 | 9 | TRP | HN | 5.25 |
| 8 | ARG | QG | 9 | TRP | HN | 5.10 |
| 9 | TRP | HN | 9 | TRP | HB2 | 2.90 |
| 9 | TRP | HN | 9 | TRP | HB3 | 2.90 |
| 9 | TRP | HN | 9 | TRP | QB | 2.68 |
| 9 | TRP | HN | 9 | TRP | HD1 | 5.20 |
| 9 | TRP | HN | 10 | LYS | HN | 4.61 |
| 9 | TRP | HA | 9 | TRP | HD1 | 5.07 |
| 9 | TRP | HA | 9 | TRP | HE3 | 4.67 |
| 9 | TRP | HA | 10 | LYS | HN | 2.93 |
| 9 | TRP | HB2 | 9 | TRP | HE3 | 3.64 |
| 9 | TRP | HB2 | 10 | LYS | HN | 3.83 |
| 9 | TRP | HB3 | 9 | TRP | HE3 | 3.64 |
| 9 | TRP | HB3 | 10 | LYS | HN | 3.83 |
| 9 | TRP | QB | 9 | TRP | HD1 | 3.49 |
| 9 | TRP | QB | 9 | TRP | HE3 | 3.42 |
| 9 | TRP | QB | 10 | LYS | HN | 3.63 |
| 9 | TRP | HD1 | 10 | LYS | HN | 4.72 |
| 9 | TRP | HD1 | 10 | LYS | HA | 5.50 |
| 9 | TRP | HD1 | 11 | CNH2 | HN1 | 5.16 |
| 9 | TRP | HE1 | 11 | CNH2 | HN1 | 5.50 |
| 10 | LYS | HN | 10 | LYS | HA | 2.90 |
| 10 | LYS | HN | 10 | LYS | HB2 | 3.08 |
| 10 | LYS | HN | 10 | LYS | HB3 | 3.08 |
| 10 | LYS | HN | 11 | CNH2 | HN1 | 4.35 |
| 10 | LYS | HA | 10 | LYS | HB2 | 2.83 |
| 10 | LYS | HA | 10 | LYS | HB3 | 2.83 |
| 10 | LYS | HA | 10 | LYS | QB | 2.63 |
| 10 | LYS | HA | 10 | LYS | HG2 | 3.30 |
| 10 | LYS | HA | 10 | LYS | HG3 | 3.30 |
| 10 | LYS | HA | 10 | LYS | QG | 2.95 |
| 10 | LYS | HB2 | 11 | CNH2 | HN1 | 5.50 |
| 10 | LYS | HB3 | 11 | CNH2 | HN1 | 5.50 |
| 10 | LYS | HG2 | 10 | LYS | HE2 | 4.04 |
| 10 | LYS | HG2 | 10 | LYS | HE3 | 4.04 |
| 10 | LYS | HG2 | 10 | LYS | HZ1 | 3.76 |
| 10 | LYS | HG3 | 10 | LYS | HE2 | 4.04 |
| 10 | LYS | HG3 | 10 | LYS | HE3 | 4.04 |
| 10 | LYS | HG3 | 10 | LYS | HZ1 | 3.76 |
| 10 | LYS | QG | 10 | LYS | QE | 3.23 |
| 10 | LYS | QG | 10 | LYS | HZ1 | 3.41 |

ACE $Q H$ is the methyl group of the $N$-terminal acetyl funtion. CNH2 HN1 is one of the amide protons of the C-terminal carboxamide function.

Table S15. NOE Derived Upper Limit Constraints of SHU9119 in DPC solution

| 3 | ACE | QH | 5 | ASP | HN | 6.09 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 4 | NLE | HN | 4 | NLE | HA | 2.86 |
| 4 | NLE | HN | 4 | NLE | HB2 | 3.90 |
| 4 | NLE | HN | 4 | NLE | HB3 | 3.90 |
| 4 | NLE | HN | 4 | NLE | QB | 2.80 |
| 4 | NLE | HN | 4 | NLE | QG | 5.39 |
| 4 | NLE | HN | 4 | NLE | QD | 5.92 |
| 4 | NLE | HN | 5 | ASP | HN | 2.97 |
| 4 | NLE | HA | 4 | NLE | HB2 | 2.96 |
| 4 | NLE | HA | 4 | NLE | HB3 | 2.96 |
| 4 | NLE | HA | 4 | NLE | QD | 4.62 |
| 4 | NLE | HA | 5 | ASP | HN | 3.52 |
| 4 | NLE | HA | 6 | HIS | HN | 4.88 |
| 4 | NLE | HB2 | 4 | NLE | QE | 6.25 |
| 4 | NLE | HB2 | 5 | ASP | HN | 3.52 |
| 4 | NLE | HB3 | 4 | NLE | QE | 6.25 |
| 4 | NLE | HB3 | 5 | ASP | HN | 3.52 |
| 4 | NLE | QB | 4 | NLE | QE | 6.03 |
| 4 | NLE | QB | 5 | ASP | HN | 3.22 |
| 4 | NLE | QB | 6 | HIS | HN | 5.82 |
| 4 | NLE | QB | 7 | DNAL | HN | 5.67 |
| 4 | NLE | QB | 7 | DNAL | QB | 5.50 |
| 4 | NLE | QG | 5 | ASP | HN | 6.38 |
| 4 | NLE | QG | 9 | TRP | HE3 | 6.38 |
| 4 | NLE | QD | 5 | ASP | HN | 6.38 |
| 4 | NLE | QD | 9 | TRP | HE3 | 6.38 |
| 4 | NLE | QE | 5 | ASP | HN | 6.53 |
| 4 | NLE | QE | 9 | TRP | HE3 | 6.53 |
| 5 | ASP | HN | 5 | ASP | HA | 2.55 |
| 5 | ASP | HN | 6 | HIS | HN | 3.08 |
| 5 | ASP | HN | 7 | DNAL | HN | 4.30 |
| 5 | ASP | HN | 10 | LYS | HZ1 | 5.50 |
| 5 | ASP | HA | 5 | ASP | HB2 | 2.90 |
| 5 | ASP | HA | 5 | ASP | HB3 | 2.90 |
| 5 | ASP | HA | 5 | ASP | QB | 2.68 |
| 5 | ASP | HA | 6 | HIS | HN | 3.64 |
| 5 | ASP | HA | 7 | DNAL | HN | 4.98 |
| 5 | ASP | HA | 10 | LYS | HZ1 | 5.07 |
| 5 | ASP | HB2 | 10 | LYS | HZ1 | 3.05 |
| 5 | ASP | HB3 | 10 | LYS | HZ1 | 3.05 |
| 5 | ASP | QB | 8 | ARG | HN | 5.50 |
| 5 | ASP | QB | 10 | LYS | HZ1 | 2.85 |
| 6 | HIS | HN | 6 | HIS | HA | 2.71 |
| 6 | HIS | HN | 6 | HIS | HB2 | 3.48 |
| 6 | HIS | HN | 6 | HIS | HB3 | 3.48 |
| 6 | HIS | HN | 6 | HIS | QB | 3.20 |
| 6 | HIS | HN | 7 | DNAL | HN | 3.33 |
| 6 | HIS | HN | 8 | ARG | HN | 4.66 |
| 6 | HIS | HA | 6 | HIS | HB2 | 2.72 |
| 6 | HIS | HA | 6 | HIS | HB3 | 2.72 |
| 6 | HIS | HA | 6 | HIS | HD2 | 4.51 |
| 6 | HIS | HA | 7 | DNAL | HN | 3.02 |
| 6 | HIS | HA | 8 | ARG | HN | 4.83 |
| 6 | HIS | QB | 7 | DNAL | HN | 4.80 |
| 7 | DNAL | HN | 7 | DNAL | HB2 | 2.90 |
| 7 | DNAL | HN | 7 | DNAL | HB3 | 2.90 |
| 7 | DNAL | HN | 7 | DNAL | QB | 2.66 |
| 7 | DNAL | HN | 8 | ARG | HN | 3.52 |
| 7 | DNAL | HA | 8 | ARG | HN | 3.24 |


| 8 | ARG | HN | 9 | TRP | HN | 4.52 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 8 | ARG | HA | 8 | ARG | QB | 2.78 |
| 8 | ARG | HA | 9 | TRP | HN | 2.59 |
| 8 | ARG | HB2 | 9 | TRP | HN | 3.24 |
| 8 | ARG | HB3 | 9 | TRP | HN | 3.24 |
| 9 | TRP | HN | 9 | TRP | HB2 | 2.93 |
| 9 | TRP | HN | 9 | TRP | HB3 | 2.93 |
| 9 | TRP | HN | 9 | TRP | QB | 2.69 |
| 9 | TRP | HN | 9 | TRP | HD1 | 5.22 |
| 9 | TRP | HN | 10 | LYS | HN | 4.64 |
| 9 | TRP | HA | 9 | TRP | HD1 | 5.04 |
| 9 | TRP | HA | 9 | TRP | HE3 | 4.74 |
| 9 | TRP | HA | 10 | LYS | HN | 2.68 |
| 9 | TRP | HB2 | 9 | TRP | HE3 | 3.79 |
| 9 | TRP | HB2 | 10 | LYS | HN | 3.84 |
| 9 | TRP | HB3 | 9 | TRP | HE3 | 3.79 |
| 9 | TRP | HB3 | 10 | LYS | HN | 3.84 |
| 9 | TRP | QB | 9 | TRP | HD1 | 3.49 |
| 9 | TRP | QB | 9 | TRP | HE3 | 3.53 |
| 9 | TRP | QB | 10 | LYS | HN | 3.63 |
| 9 | TRP | HD1 | 10 | LYS | HN | 5.04 |
| 9 | TRP | HD1 | 10 | LYS | HA | 5.50 |
| 9 | TRP | HD1 | 11 | CNH2 | HN1 | 5.18 |
| 9 | TRP | HE1 | 11 | CNH2 | HN1 | 5.46 |
| 10 | LYS | HN | 10 | LYS | HA | 2.96 |
| 10 | LYS | HN | 10 | LYS | HB2 | 3.14 |
| 10 | LYS | HN | 10 | LYS | HB3 | 3.14 |
| 10 | LYS | HN | 11 | CNH2 | HN1 | 4.36 |
| 10 | LYS | HA | 10 | LYS | HB2 | 2.86 |
| 10 | LYS | HA | 10 | LYS | HB3 | 2.86 |
| 10 | LYS | HA | 10 | LYS | QB | 2.64 |
| 10 | LYS | HB2 | 11 | CNH2 | HN1 | 5.50 |
| 10 | LYS | HB3 | 11 | CNH2 | HN1 | 5.50 |
| 10 | LYS | HG2 | 10 | LYS | HE2 | 3.98 |
| 10 | LYS | HG2 | 10 | LYS | HE3 | 3.98 |
| 10 | LYS | HG2 | 10 | LYS | HZ1 | 3.73 |
| 10 | LYS | HG3 | 10 | LYS | HE2 | 4.08 |
| 10 | LYS | HG3 | 10 | LYS | HE3 | 4.08 |
| 10 | LYS | HG3 | 10 | LYS | HZ1 | 3.76 |
| 10 | LYS | QG | 10 | LYS | HZ1 | 4.64 |

ACE QH is the methyl group of the $N$-terminal acetyl funtion. CNH2 HN1 is one of the amide protons of the C-terminal carboxamide function.

Table S16. Dihedral angles of the NMR structures of MTII and SHU9119

MTII

| 4 | NLE | OMEGA | $-178.8+/-$ | 6.4 |
| :--- | :--- | :--- | ---: | ---: |
| 4 | NLE | PHI | $-90.2+/-$ | 22.8 |
| 4 | NLE | CHI1 | $-125.6+/-$ | 44.7 |
| 4 | NLE | CHI2 | $63.0+/-$ | 52.3 |
| 4 | NLE | CHI3 | $54.1+/-$ | 68.6 |
| 4 | NLE | PSI | $-72.4+/-$ | 29.2 |
| 5 | ASP | OMEGA | $-175.6+/-$ | 5.5 |
| 5 | ASP | PHI | $-118.0+/-$ | 25.6 |
| 5 | ASP | CHI1 | $-167.9+/-$ | 23.9 |
| 5 | ASP | CHI2 | $-0.1+/-$ | 24.3 |
| 5 | ASP | PSI | $-51.6+/-$ | 10.3 |
| 6 | HIS | OMEGA | $173.0+/-$ | 5.6 |
| 6 | HIS | PHI | $-164.6+/-$ | 4.0 |
| 6 | HIS | CHI1 | $-128.3+/-$ | 48.3 |
| 6 | HIS | CHI2 | $-77.7+/-$ | 64.1 |
| 6 | HIS | PSI | $42.1+/-$ | 13.1 |
| 7 | DPHE | OMEGA | $-174.8+/-$ | 6.8 |
| 7 | DPHE PHI | $87.7+/-$ | 24.4 |  |
| 7 | DPHE | CHI1 | $157.3+/-$ | 4.4 |
| 7 | DPHE | CHI2 | $-118.5+/-$ | 84.2 |
| 7 | DPHE | PSI | $-55.4+/-$ | 23.0 |
| 8 | ARG | OMEGA | $-179.2+/-$ | 6.5 |
| 8 | ARG | PHI | $-79.4+/-$ | 6.0 |
| 8 | ARG | CHI1 | $-74.3+/-$ | 12.6 |
| 8 | ARG | CHI2 | $-123.0+/-$ | 38.8 |
| 8 | ARG | CHI3 | $-112.5+/-$ | 113.2 |
| 8 | ARG | CHI4 | $-168.3+/-$ | 45.0 |
| 8 | ARG | PSI | $159.6+/-$ | 7.1 |
| 9 | TRP | OMEGA | $-175.4+/-$ | 7.7 |
| 9 | TRP | PHI | $-90.6+/-$ | 18.4 |
| 9 | TRP | CHI1 | $179.4+/-$ | 4.6 |
| 9 | TRP | CHI2 | $-99.9+/-$ | 23.7 |
| 9 | TRP | PSI | $155.0+/-$ | 3.9 |
| 10 | LYS | OMEGA | $-172.8+/-$ | 4.1 |
| 10 | LYS | PHI | $-92.1+/-$ | 11.5 |
| 10 | LYS | CHI1 | $-73.6+/-$ | 5.5 |
| 10 | LYS | CHI2 | $173.0+/-$ | 43.8 |
| 10 | LYS | CHI3 | $172.6+/-$ | 4.2 |
| 10 | LYS | CHI4 | $44.8+/-$ | 44.3 |
| 10 | LYS | PSI | $88.3+/-$ | 26.1 |
|  |  |  |  |  |

SHU9119

| 4 | NLE | OMEGA | $169.1+/-$ | 2.4 |
| :--- | :--- | :--- | ---: | ---: |
| 4 | NLE | PHI | $-78.6+/-$ | 24.4 |
| 4 | NLE | CHI1 | $-134.5+/-$ | 32.7 |
| 4 | NLE | CHI2 | $119.8+/-$ | 54.9 |
| 4 | NLE | CHI3 | $98.8+/-$ | 111.5 |
| 4 | NLE | PSI | $-57.9+/-$ | 27.6 |
| 5 | ASP | OMEGA | $176.7+/-$ | 4.6 |
| 5 | ASP | PHI | $-135.4+/-$ | 24.5 |
| 5 | ASP | CHI1 | $-170.6+/-$ | 15.3 |
| 5 | ASP | CHI2 | $10.1+/-$ | 29.5 |
| 5 | ASP | PSI | $-63.8+/-$ | 7.9 |
| 6 | HIS | OMEGA | $-178.1+/-$ | 2.9 |
| 6 | HIS | PHI | $-143.8+/-$ | 7.9 |
| 6 | HIS | CHI1 | $-67.6+/-$ | 83.8 |
| 6 | HIS | CHI2 | $-85.0+/-$ | 78.3 |
| 6 | HIS | PSI | $53.4+/-$ | 14.7 |
| 7 | DNAL | OMEGA | $-173.7+/-$ | 2.5 |
| 7 | DNAL | PHI | $64.0+/-$ | 18.3 |
| 7 | DNAL | CHI1 | $112.0+/-$ | 48.0 |
| 7 | DNAL | CHI2 | $101.6+/-$ | 36.4 |
| 7 | DNAL | PSI | $5.0+/-$ | 21.1 |
| 8 | ARG | OMEGA | $-178.9+/-$ | 2.0 |
| 8 | ARG | PHI | $-132.9+/-$ | 18.1 |
| 8 | ARG | CHI1 | $-71.6+/-$ | 13.1 |
| 8 | ARG | CHI2 | $-166.3+/-$ | 33.6 |
| 8 | ARG | CHI3 | $174.0+/-$ | 56.6 |
| 8 | ARG | CHI4 | $172.5+/-$ | 49.7 |
| 8 | ARG | PSI | $157.7+/-$ | 4.3 |
| 9 | TRP | OMEGA | $179.9+/-$ | 2.7 |
| 9 | TRP | PHI | $-68.2+/-$ | 25.0 |
| 9 | TRP | CHI1 | $-174.7+/-$ | 3.1 |
| 9 | TRP | CHI2 | $-104.7+/-$ | 11.7 |
| 9 | TRP | PSI | $149.1+/-$ | 3.6 |
| 10 | LYS | OMEGA | $-168.8+/-$ | 1.3 |
| 10 | LYS | PHI | $-92.8+/-$ | 14.0 |
| 10 | LYS | CHI1 | $-84.1+/-$ | 10.0 |
| 10 | LYS | CHI2 | $166.0+/-$ | 27.5 |
| 10 | LYS | CHI3 | $-175.0+/-$ | 6.9 |
| 10 | LYS | CHI4 | $59.7+/-$ | 68.3 |
| 10 | LYS | PSI | $97.2+/-$ | 8.8 |
| 7 |  |  |  |  |



Scheme 1. Synthesis of linear peptides on $\mathrm{H}-1-\mathrm{Thr}(\mathrm{t}-\mathrm{Bu})-\mathrm{ol}-2$-chlorotrityl resin. a) (i) Fmoc- L-Hag, HATU/NMM, 40 min r.t.; (ii) $20 \%$ piperidine in DMF (2 x 15 min); (iii) coupling with the amino acids.

The resin aliquots containing the linear peptides were swollen for 2 h in anhydrous DCM. After two hours, the vessels were heated to $45^{\circ} \mathrm{C}$ and a DCM solution of catalyst 9 ( 0.5 mole equiv. calculated on the basis of $0.5 \mathrm{mmol} / \mathrm{g}$ of peptide) was added. The suspension was then stirred for 48 h at $45^{\circ} \mathrm{C}$. The resin aliquots were washed with DCM, DMF, and MeOH , then swelled for 45 min at room temperature in DMF. Fmoc-Hag was deprotected ( 2.5 mL of $20 \%$ piperidine in DMF for $5 \mathrm{~min}, 4$ time repeated) and coupled with Fmoc-D-Phe affording the on-resin peptides $\mathbf{4 - 8}$ which were deprotected and cleaved as previously described (Scheme 2).



|  | $\mathrm{Y}=\mathrm{L}-\mathrm{Phe}$ | Z | Z-isome |
| :---: | :---: | :---: | :---: |
| 5) | $\mathrm{Y}=\mathrm{L}-1-\mathrm{Na}$ | $\mathrm{Z}=\mathrm{L}-\mathrm{Th} \mathrm{r}$ |  |
| 6) |  |  |  |
| 7) | $\mathrm{Y}=\mathrm{L}-\mathrm{Phe}$ |  |  |
|  | $\mathrm{Y}=\mathrm{L}-1-\mathrm{N}$ | $\mathrm{Z}=\mathrm{L}-\mathrm{Tyr}(\mathrm{B}$ |  |

Scheme 2. Synthesis of ciclic octapeptides. b) (i) Catalyst, $45^{\circ} \mathrm{C}, 48 \mathrm{~h}$; (ii) $20 \%$ piperidine in DMF, Fmoc-D-Phe/HATU/NMM 40 min, r.t.
c) i) $20 \%$ piperidine in DMF; ii) Cleavage of 5-7 by TFA/DCM/EDT/Phenol (94:2:2:2); $\mathbf{4}$ and $\mathbf{8}$ by TFA/DCM/EDT/Phenol (70:26:2:2).

Table S17. RP-HPLC data of the purified analogues.

| Compound | HPLC method $^{\text {a }}$ | Retention times (min.) |
| :---: | :---: | :---: |
| $\mathbf{4}$ | $45 \%-55 \%$ B in 20 min. | 9.64 |
| $\mathbf{5}$ | $20 \%-60 \%$ B in 20 min. | 13.26 |
| $\mathbf{6}$ | $32 \%$ B in 20 min. | 13.03 |
| $\mathbf{7}$ | $30 \%-40 \%$ B in 20 min. | 15.08 |
| $\mathbf{8}$ | $50 \%-60 \%$ B in 10 min. | 7.02 |

${ }^{a} \mathrm{~A}: \mathrm{H}_{2} \mathrm{O} 0.1 \%$ TFA;B: $\mathrm{CH}_{3} \mathrm{CN} 0.1 \%$ TFA. ${ }^{b} \mathrm{R}_{t}$ of the pure compounds.

Table S18. Mass Spectral data of the purified analogues.

| Compound | $[\mathrm{M}]^{+}$calcd. | $[\mathrm{M}+\mathrm{H}]^{+}$found | $[\mathrm{M}+2 \mathrm{H}]^{2+}$ | $[\mathrm{M}+\mathrm{Na}]^{+}$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{4}$ | 1132.57 | 1133.53 | 567.6 | 1156.66 |
| $\mathbf{5}$ | 1030.53 | 1032.0 | 514.46 | 1054.98 |
| $\mathbf{6}$ | 1042.53 | $1043.55(20 \%)$ | $523,53(100 \%)$ | 1066.67 |
| $\mathbf{7}$ | 1042.53 | $1043.63(20 \%)$ | $522,35(100 \%)$ | 1066.67 |
| $\mathbf{8}$ | 1182.59 | $1184,04(10 \%)$ | $592,51(100 \%)$ | 1206.94 |



Figure S21. RP-HPLC trace of pure (4)


Figure S22. RP-HPLC trace of pure (5)


Figure S23. RP-HPLC trace of pure (6)- $E$ isomer


Figure S24. RP-HPLC trace of pure (7) Z-isomer


Figure S25. RP-HPLC trace of pure (8)

Table S19. NMR Resonance Assignments ${ }^{a}$ of Peptide 3 in SDS-d ${ }_{25}$ 200 mM Solution.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| D-Phe ${ }^{2}$ |  | 4.23 | 3.23, 3.10 | 7.26 (8); 7.31(8); |
| dhDsa- ${ }^{\text {c }}$ | 8.02 (6.8, 6.7) | 3.81 | 2.22 | 5.16( $\gamma$ ) |
| Phe ${ }^{7}$ | 7.00 (8.2, 6.2) | 4.74 | 2.98 | 7.03(8); 7.16(8) |
| D-Trp ${ }^{8}$ | 8.09 (5.5, 6.5) | 4.38 | 3.21, 3.09 | $\begin{aligned} & 7.21(\delta) ; 10.05, \\ & 7.68(\varepsilon) ; 7.49,7.14(\zeta) \\ & 7.14(\eta) \end{aligned}$ |
| Lys ${ }^{9}$ | 7.36 (5.9, 3.7) | 4.09 | 1.44, 1.25 | $\begin{aligned} & 0.52,0.43(\gamma) ; 1.39(\delta) ; \\ & 2.76(\varepsilon) \end{aligned}$ |
| $\operatorname{Tyr}(\mathrm{Bzl})^{10}$ | 7.67 (5.4, 2.6) | 4.25 | 3.13, 2.87 | 7.09( $\delta$ ); 6.89( $)$; ${ }^{\text {d }}$ |
| dhDsa-C ${ }^{\text {c }}$ | 7.15 (6.7, 3.0) | 4.39 | 2.52, 2.23 | 5.27( $\gamma$ ) |
| $\operatorname{Thr}(\mathrm{ol})^{15}$ | 7.28 (6.9, 3.4) | 3.80 | 3.95 | 1.15( $\gamma$ ); 3.68, 3.61 ( $\omega$ ) |

${ }^{a}$ Obtained at $35{ }^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm} .{ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in $\mathrm{Hz} . \quad-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients $(\mathrm{ppb} / \mathrm{K}) .{ }^{\mathrm{c}}$ dh-DSA-N: N-terminal portion of dehydrodiaminosuberic acid. dh-DSA-C: C-terminal portion of dehydrodiaminosuberic acid. ${ }^{\mathrm{d}}$ Other signals: $\mathrm{CH}_{2}: 5.09,5.04$, Arom: 7.44, 7.40, 7.30 .

Table S20. NMR Resonance Assignments ${ }^{a}$ of Peptide 4 in SDS- $\mathrm{d}_{25}$ 200 mM Solution.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha N},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
| D-Phe $^{2}$ |  | 4.21 | $3.08 ; 3.29$ | $7.25(\delta) ; 7.32(\varepsilon)$ |
| dhDsa-N $^{\mathrm{c}}$ | $8.12(8.2,6.9)$ | 3.79 | $2.32,2.13$ | $5.11(\gamma)$ |
| Phe $^{7}$ | $6.94(8.5,5.3)$ | 4.73 | $2.95,2.90$ | $6.84(\delta) ; 7.02(\varepsilon)$ |
| D-Trp $^{8}$ | $8.24(4.7,6.3)$ | 4.61 | $3.34,3.04$ | $7.28(\delta) ; 10.03,7.89(\varepsilon) ;$ |
| Lys $^{9}$ | $7.51(5.9,4.7)$ | 4.07 | $1.51,1.37$ | $0.68,0.63(\gamma) ; 1.45(\delta) ;$ |
| Tyr(Bzl) ${ }^{10}$ | $7.92(6.5,2.7)$ | 4.41 | $3.09,2.96$ | $7.14(\delta) ; 6.89(\varepsilon) ;{ }^{\text {d }}$ |
| dhDsa-C |  |  |  |  |
| Thr(ol $)^{15}$ | $7.61(7.1,5.8)$ | 4.27 | $2.44,2.12$ | $5.20(\gamma)$ |

${ }^{a}$ Obtained at $35{ }^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm} .{ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in $\mathrm{Hz} . \quad-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients $(\mathrm{ppb} / \mathrm{K}) .^{\mathrm{c}}$ dh-DSA-N: N-terminal portion of dehydrodiaminosuberic acid. dh-DSA-C: C-terminal portion of dehydrodiaminosuberic acid. ${ }^{\text {d }}$ Other signals: $\mathrm{CH}_{2}$ : 5.03; Arom: 7.43, 7.39, 7.32.

Table S21. NMR Resonance Assignments ${ }^{a}$ of Peptide 5 in SDS- $\mathrm{d}_{25}$ 200 mM Solution.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha N},-\Delta \delta / \Delta \mathrm{T}\right)^{\text {b }}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| D-Phe ${ }^{2}$ |  | 4.24 | 3.24, 3.14 | 7.31( 8 ) |
| dhDsa- | 7.96 (8.1, 7.9 ) | 4.23 | 2.45, 2.29 | $5.29(\gamma)$ |
| $1-\mathrm{Nal}^{7}$ | 8.13 (7.5, 6.3) | 4.73 | 3.54, 3.42 | $\begin{aligned} & 7.69(\zeta) ; 8.18,7.42(\delta) ; \\ & 7.45(\varepsilon) \end{aligned}$ |
| D-Trp ${ }^{8}$ | 7.31 (4.8, 6.3) | 4.04 | 2.59, 2.28 | $\begin{aligned} & \text { 6.88(ס);9.86, 7.07(ع);7.43, } \\ & 6.98(\zeta) ; 7.11(\eta) \end{aligned}$ |
| Lys ${ }^{9}$ | 7.90 (6.2, 4.6) | 3.88 | 1.50, 1.25 | $\begin{aligned} & 0.58,0.32(\gamma) ; 1.33(\delta) ; 2.69 \\ & 2.65(\varepsilon) ; 7.24(\zeta) \end{aligned}$ |
| Thr ${ }^{10}$ | 8.04 (6.1, 2.7) | 4.04 | 4.22 | 1.25( $\gamma$ ) |
| dhDsa- | 7.76 (7.5, 6.0) | 4.36 | 2.41, 2.22 | $5.39(\gamma)$ |
| $\operatorname{Thr}(\mathrm{ol})^{15}$ | 7.26 (8.0, 6.4) | 3.77 | 3.83 | 1.05( $\gamma$ ); 3.63, 3.53 ( $\omega$ ) |

[^3]Table S22. NMR Resonance Assignments ${ }^{a}$ of Peptide 6 in SDS- $\mathrm{d}_{25}$ 200 mM Solution.

| residue | NH ( $\left.{ }^{3} J_{\alpha N},-\Delta \delta / \Delta \mathrm{T}\right)^{\text {b }}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :---: | :---: | :---: | :---: | :---: |
| D-Phe ${ }^{2}$ |  | 4.21 | 3.24, 3.07 | 7.26(8); 7.31 (8) |
| dhDsa- | 7.87 (6.7, 7.7) | 3.96 | 2.18 | 5.21 ( $\gamma$ ) |
| Phe ${ }^{7}$ | 7.27 (8.3, 6.7) | 4.67 | 2.92 | 6.98(8); 7.16(8) |
| D-Trp ${ }^{8}$ | 7.81 (5.3, 6.6) | 4.43 | 3.07, 2.97 | $\begin{aligned} & \text { 7.11( }(\delta) ; 10.01,7.58(\varepsilon) ; \\ & 7.44,7.08(\zeta) ; 7.15(\eta) \end{aligned}$ |
| Lys ${ }^{9}$ | 7.66 (6.0, 3.8) | 3.95 | 1.31, 1.24 | $\begin{aligned} & 0.54,0.36(\gamma) ; 1.34(\delta) ; \\ & 2.71(\varepsilon) ; 7.31(\zeta) \end{aligned}$ |
| Tyr ${ }^{10}$ | 7.98 (5.3, 2.8) | 4.22 | 3.04, 2.86 | 6.77(8); $7.03(\gamma)$ |
| dhDsa- | 7.27 (6.8, 3.1) | 4.31 | 2.49, 2.30 | 5.33( $\gamma$ ) |
| $\mathrm{Thr}(\mathrm{ol})^{15}$ | 7.31 (6.9, 3.8) | 3.78 | 3.94 | 1.14( $\gamma$ ); 3.66, 3.59 ( $\omega$ ) |

${ }^{a}$ Obtained at $35{ }^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm} .{ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz. $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ). ${ }^{\mathrm{c}}$ dh-DSA-N: N-terminal portion of dehydrodiaminosuberic acid. dh-DSA-C: C-terminal portion of dehydrodiaminosuberic acid.

Table S23. NMR Resonance Assignments ${ }^{a}$ of Peptide 7 in SDS-d ${ }_{25}$ 200 mM Solution.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha N},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
| D-Phe $^{2}$ |  | 4.23 | $3.29,3.10$ | $7.28(\delta) ; 7.34(\varepsilon)$ |
| dhDsa- | $8.12(8.2,6.9)$ | 3.80 | $2.42,2.11$ | $5.29(\gamma)$ |
| Phe $^{7}$ | $6.97(8.1,6.6)$ | 4.73 | $2.89,2.76$ | $6.67(\delta) ; 6.95(\varepsilon) ; 7.03(\zeta)$ |
| D-Trp $^{8}$ | $8.16(4.8,6.8)$ | 4.74 | $3.34,3.00$ | $7.29(\delta) ; 9.99,7.89(\varepsilon) ; 7.45$, |
|  |  |  |  | $7.18(\zeta) ; 7.21(\eta)$ |
| Lys $^{9}$ | $7.83(6.9,4.3)$ | 3.94 | 1.41 | $0.74,0.61(\gamma) ; 1.46(\delta) ;$ |
|  |  |  |  | $2.81(\varepsilon) ; 7.33(\zeta)$ |
| Tyr $^{10}$ | $8.36(6.4,2.6)$ | 4.41 | $3.14,2.88$ | $7.12(\delta) ; 6.79(\varepsilon)$ |
| dhDsa- | $7.53(7.2,5.9)$ | 4.28 | $2.49,1.93$ | $5.43(\gamma)$ |
| Thr $(01)^{15}$ | $7.44(7.5,5.5)$ | 3.83 | 3.96 | $1.15(\gamma) ; 3.70,3.61(\omega)$ |

${ }^{a}$ Obtained at $35{ }^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02$ ppm. ${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz. $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ). ${ }^{\mathrm{c}}$ dh-DSA-N: N-terminal portion of dehydrodiaminosuberic acid. dh-DSA-C: C-terminal portion of dehydrodiaminosuberic acid.

Table S24. NMR Resonance Assignments ${ }^{a}$ of Peptide 8 in SDS-d ${ }_{25}$ 200 mM Solution.

| residue | $\mathrm{NH}\left({ }^{3} J_{\alpha \mathrm{N}},-\Delta \delta / \Delta \mathrm{T}\right)^{\mathrm{b}}$ | $\mathrm{C}^{\alpha} \mathrm{H}$ | $\mathrm{C}^{\beta} \mathrm{H}$ | Others |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
| D-Phe $^{2}$ |  | 4.26 | $3.30,3.17$ | $7.31(\delta) ; 7.37(\varepsilon)$ |
| dhDsa-N $^{\mathrm{c}}$ | $8.12(8.1,6.8)$ | 4.20 | 2.34 | $5.19(\gamma)$ |
| 1-Nal $^{7}$ | $7.60(8.4,5.4)$ | 4.93 | $3.64,3.53$ | $7.67(\zeta) ; 8.22,7.31(\delta) ; 7.80(\varepsilon)$ |
| D-Trp $^{8}$ | $8.22(4.8,6.3)$ | 4.10 | $3.03,2.72$ | $7.12(\delta) ; 10.00,7.46(\varepsilon) ; 7.51$, |
| Lys $^{9}$ | $6.64(6.9,4.6)$ | 4.00 | $1.29,0.93$ | $0.22,0.08(\gamma) ; 1.27(\delta) ; 2.69(\varepsilon)$ |
| Tyr(Bzl) $^{10}$ | $7.83(6.5,2.6)$ | 4.43 | $3.06,2.91$ | $7.09(\delta) ; 6.87(\varepsilon) ;{ }^{\text {d }}$ |
| dhDsa-C | $7.95(7.1,5.7)$ | 4.44 | $2.45,2.35$ | $5.37(\gamma)$ |
| Thr(ol $)^{15}$ | $7.51(7.6,5.9)$ | 3.85 | 3.97 | $1.15(\gamma) ; 3.71,3.62(\omega)$ |

${ }^{a}$ Obtained at $35{ }^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$. ${ }^{\mathrm{b}}{ }^{3} J_{\alpha N}$ coupling constants in Hz . $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients $(\mathrm{ppb} / \mathrm{K})$. ${ }^{\mathrm{c}}$ dh-DSA-N: N-terminal portion of dehydrodiaminosuberic acid. dh-DSA-C: C-terminal portion of dehydrodiaminosuberic acid. ${ }^{\mathrm{d}}$ Other signals: $\mathrm{CH}_{2}: 5.05,5.02$, Arom: 7.47, 7.42, 7.35 .

Table S25. NOE Derived Upper Limit Constraints of Compound 3

| Atom1 |  | Atom2 | Upper Limit | Violation |
| :---: | :---: | :---: | :---: | :---: |
| 2 | DPHE HA | 3 DHS HN | 2.52 |  |
| 2 | DPHE HB2 | 3 DHS HN | 5.31 |  |
| 2 | DPHE HB3 | 3 DHS HN | 5.31 |  |
| 2 | DPHE QB | 3 DHS HN | 4.49 |  |
| 3 | DHS HN | 3 DHS QB | 3.84 |  |
| 3 | DHS HA | 14 DHS HG | 4.60 |  |
| 3 | DHS HA | 7 PHE HN | 2.40 |  |
| 3 | DHS HG | 14 DHS HA | 3.58 |  |
| 3 | DHS HG | 14 DHS QB | 3.21 |  |
| 7 | PHE HN | 10 TBZ HB2 | 2.99 |  |
| 7 | PHE HA | 8 DTRP HN | 2.83 |  |
| 7 | PHE QB | 8 DTRP HN | 3.93 |  |
| 7 | PHE QD | 8 DTRP HN | 6.69 |  |
| 8 | DTRP HN | 8 DTRP HB2 | 2.86 |  |
| 8 | DTRP HN | 8 DTRP HB3 | 2.86 |  |
| 8 | DTRP HN | 8 DTRP QB | 2.64 |  |
| 8 | DTRP HN | 8 DTRP HD1 | 5.50 |  |
| 8 | DTRP HA | 8 DTRP HD1 | 4.97 |  |
| 8 | DTRP HA | 8 DTRP HE3 | 2.70 |  |
| 8 | DTRP HA | 9 LYS HN | 2.40 |  |
| 8 | DTRP HB2 | 8 DTRP HD1 | 3.70 |  |
| 8 | DTRP HB2 | 8 DTRP HE3 | 4.19 |  |
| 8 | DTRP HB2 | 9 LYS HN | 4.60 |  |
| 8 | DTRP HB3 | 8 DTRP HD1 | 3.70 |  |
| 8 | DTRP HB3 | 8 DTRP HE3 | 4.19 |  |
| 8 | DTRP HB3 | 9 LYS HN | 4.60 |  |
| 8 | DTRP QB | 8 DTRP HD1 | 3.13 |  |
| 8 | DTRP QB | 8 DTRP HE3 | 3.70 |  |
| 8 | DTRP QB | 9 LYS HN | 4.18 |  |
| 8 | DTRP HD1 | 9 LYS QG | 6.38 |  |
| 8 | DTRP HE1 | 9 LYS QG | 6.38 |  |
| 8 | DTRP HE1 | 9 LYS QD | 6.38 |  |
| 8 | DTRP HE1 | 9 LYS QE | 6.38 |  |
| 8 | DTRP HE1 | 15 THO HA | 5.50 |  |
| 8 | DTRP HZ2 | 9 LYS QE | 6.38 |  |
| 9 | LYS HN | 9 LYS HA | 2.91 |  |
| 9 | LYS HN | 9 LYS HB2 | 2.60 |  |
| 9 | LYS HN | 9 LYS HB3 | 3.74 |  |
| 9 | LYS HN | 9 LYS HG2 | 3.45 |  |
| 9 | LYS HN | 9 LYS HG3 | 3.45 |  |
| 9 | LYS HN | 9 LYS QG | 3.21 |  |
| 9 | LYS HN | 10 TBZ HN | 3.03 |  |
| 9 | LYS HA | 9 LYS HG2 | 3.76 |  |
| 9 | LYS HA | 9 LYS HG3 | 3.76 |  |
| 9 | LYS HA | 9 LYS QG | 3.35 |  |
|  | LYS HA | 9 LYS QD | 6.38 |  |


| 9 | LYS HA | 10 | TBZ HN | 3.56 |
| :--- | :--- | :--- | :--- | :--- |
| 9 | LYS HA | 15 | THO HN | 4.80 |
| 9 | LYS HA | 15 | THO QG2 | 6.53 |
| 9 | LYS HB2 | 10 | TBZ HN | 3.54 |
| 9 | LYS HB2 | 10 | TBZ QD | 7.64 |
| 9 | LYS HB3 | 10 | TBZ HN | 3.69 |
| 9 | LYS HB3 | 10 | TBZ QD | 7.64 |
| 9 | LYS HG2 | 10 | TBZ HN | 5.50 |
| 9 | LYS HG3 | 10 | TBZ HN | 5.50 |
| 9 | LYS QG | 10 | TBZ QE | 8.51 |
| 10 | TBZ HN | 10 | TBZ HB2 | 2.63 |
| 10 | TBZ HN | 10 | TBZ HB3 | 3.60 |
| 10 | TBZ HN | 10 | TBZ QE | 6.83 |
| 10 | TBZ HN | 14 | DHS HN | 2.60 |
| 10 | TBZ HN | 15 | THO HN | 4.83 |
| 10 | TBZ HA | 14 | DHS HN | 3.35 |
| 14 | DHS HN | 14 | DHS HA | 2.91 |
| 14 | DHS HN | 14 | DHS HB2 | 3.02 |
| 14 | DHS HN | 14 | DHS HB3 | 3.90 |
| 14 | DHS HN | 14 | DHS HG | 5.07 |
| 14 | DHS HA | 14 | DHS HG | 4.04 |
| 14 | DHS HA | 15 | THO HN | 3.50 |
| 14 | DHS HB2 | 15 | THO HN | 3.72 |
| 14 | DHS HB3 | 15 | THO HN | 4.49 |
| 15 | THO HN | 15 | THO HA | 2.85 |
| 15 | THO HN | 15 | THO HB | 3.05 |
| 15 | THO HA | 15 | THO HB | 2.40 |
| 3 | DHS N- | 10 |  |  |

3 DHS: N-terminal portion of dehydrodiaminosuberic acid. 14 DHS: C-terminal portion of dehydrodiaminosuberic acid.
TBZ: $\operatorname{Tyr}(\mathrm{Bzl})$. THO: Threoninol. ${ }^{\mathrm{b}}$ Violations ( $\AA$ ) observed for the mean structure of the ensemble.

Table S26. NOE derived Upper Limit Constraints. Compound 4

|  | tom1 ${ }^{\text {a }}$ | Atom2 | Upper Limit | $\begin{gathered} \text { Vi } \\ \text { Helix } \end{gathered}$ | olation ${ }^{\text {b }}$ Extend Ensemble |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | DPHE HA | 3 DHS HN | 2.59 |  |  |
| 2 | DPHE QD | 3 DHS HA | 7.62 |  |  |
| 3 | DHS HN | 3 DHS HB2 | 3.24 |  |  |
| 3 | DHS HN | 3 DHS HB3 | 3.24 |  |  |
| 3 | DHS HN | 3 DHS QB | 3.02 |  |  |
| 3 | DHS HA | 3 DHS HB2 | 2.74 |  |  |
| 3 | DHS HA | 3 DHS HB3 | 2.74 |  |  |
| 3 | DHS HA | 3 DHS QB | 2.48 |  |  |
| 3 | DHS HA | 3 DHS HG | 3.75 |  | 0.40 |
| 3 | DHS HA | 14 DHS HG | 5.10 |  |  |
| 3 | DHS HA | 7 PHE HN | 2.40 |  |  |
| 3 | DHS HA | 7 PHE QD | 7.62 |  |  |
| 3 | DHS HA | 14 DHS HA | 3.30 | 1.20 | 0.28 |
| 3 | DHS QB | 3 DHS HG | 2.75 |  |  |
| 3 | DHS HG | 14 DHS HA | 4.72 |  |  |
| 7 | PHE HN | 7 PHE HB2 | 3.73 |  |  |
| 7 | PHE HN | 7 PHE HB 3 | 3.73 |  |  |
| 7 | PHE HN | 7 PHE QB | 3.35 |  |  |
| 7 | PHE HN | 14 DHS HA | 4.01 | 1.38 | 0.39 |
| 7 | PHE HA | 8 DTRP HN | 2.49 |  |  |
| 7 | PHE QB | 8 DTRP HN | 3.93 |  |  |
| 7 | PHE QD | 8 DTRP HN | 7.28 |  |  |
| 8 | DTRP HN | 8 DTRP HB2 | 2.68 |  |  |
| 8 | DTRP HN | 8 DTRP HB3 | 2.68 |  |  |
| 8 | DTRP HN | 8 DTRP QB | 2.42 |  |  |
| 8 | DTRP HN | 8 DTRP HD1 | 5.28 |  |  |
| 8 | DTRP HA | 8 DTRP HE3 | 2.83 |  |  |
| 8 | DTRP HA | 9 LYS HN | 2.40 |  |  |
| 8 | DTRP HA | 10 TBZ HN | 3.83 |  |  |
| 8 | DTRP HB2 | 8 DTRP HD1 | 3.70 |  |  |
| 8 | DTRP HB2 | 8 DTRP HE3 | 4.22 |  |  |
| 8 | DTRP HB2 | 9 LYS HN | 4.40 |  |  |
| 8 | DTRP HB3 | 8 DTRP HD1 | 3.70 |  |  |
| 8 | DTRP HB3 | 8 DTRP HE3 | 4.22 |  |  |
| 8 | DTRP HB3 | 9 LYS HN | 4.40 |  |  |
| 8 | DTRP QB | 8 DTRP HD1 | 3.16 |  |  |
| 8 | DTRP QB | 9 LYS HN | 4.20 |  |  |
| 8 | DTRP QB | 8 DTRP HE3 | 3.60 |  |  |
| 8 | DTRP HD1 | 9 LYS QG | 6.38 |  |  |
| 8 | DTRP HE3 | 9 LYS HN | 4.01 |  |  |
| 9 | LYS HN | 9 LYS HB2 | 2.50 |  |  |
| 9 | LYS HN | 9 LYS HB3 | 3.70 |  |  |
| 9 | LYS HN | 9 LYS QG | 3.49 |  |  |
| 9 | LYS HN | 10 TBZ HN | 3.00 |  |  |
| 9 | LYS HA | 9 LYS HG2 | 3.83 |  |  |
| 9 | LYS HA | 9 LYS HG3 | 3.83 |  |  |


${ }^{\text {a }} 3$ DHS: N-terminal portion of dehydrodiaminosuberic acid. 14 DHS: C-terminal portion of dehydrodiaminosuberic acid. TBZ: $\operatorname{Tyr}(\mathrm{Bzl})$. THO: Threoninol.
${ }^{5}$ Violations ( $\AA$ ) observed for the mean structure of the family I (helix), family II (extended) and an ensemble of 20 structures of both families (ensemble).

Table S27. NOE Derived Upper Limit Constraints of Compound 5

| Atom1 ${ }^{\text {a }}$ |  | Atom2 | Upper Limit | ViolationHelix Extend Ensemble |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | DPHE HA | 3 DHS HN | 2.40 |  |  |  |
| 2 | DPHE HB2 | 3 DHS HN | 3.86 |  |  |  |
| 2 | DPHE HB3 | 3 DHS HN | 3.86 |  |  |  |
| 2 | DPHE QB | 3 DHS HN | 3.62 |  |  |  |
| 2 | DPHE QD | 3 DHS HN | 7.62 |  |  |  |
| 3 | DHS HN | 3 DHS HB2 | 3.24 |  |  |  |
| 3 | DHS HN | 3 DHS HB3 | 3.24 |  |  |  |
| 3 | DHS HN | 3 DHS QB | 3.24 |  |  |  |
| 3 | DHS HN | 3 DHS HG | 4.35 |  |  |  |
| 3 | DHS HN | 7 NAL HN | 3.79 |  |  |  |
| 3 | 3 DHS HA | 3 DHS HG | 3.48 |  | 0.66 |  |
| 3 | 3 DHS HA | 14 DHS HG | 4.38 |  |  |  |
| 3 | 3 DHS HA | 7 NAL HN | 2.43 |  |  |  |
| 3 | 3 DHS HA | 14 DHS HA | 3.25 | 1.25 |  | 0.32 |
| 3 | 3 DHS QB | 14 DHS HN | 5.77 |  |  |  |
| 3 | 3 DHS HG | 7 NAL HN | 5.50 |  |  |  |
| 3 | 3 DHS HG | 14 DHS HA | 4.88 |  |  |  |
| 3 | 3 DHS HG | 14 DHS QB | 4.31 |  |  |  |
| 7 | 7 NAL HN | 7 NAL HB2 | 2.71 |  |  |  |
| 7 | 7 NAL HN | 7 NAL HB3 | 2.74 |  |  |  |
| 7 | 7 NAL HN | 7 NAL HD1 | 5.50 |  |  |  |
| 7 | 7 NAL HN | 8 DTRP HN | 4.48 |  |  |  |
| 7 | 7 NAL HN | 14 DHS HA | 3.82 | 1.57 |  | 0.46 |
| 7 | NAL HN | 14 DHS QB | 5.00 |  |  |  |
| 7 | 7 NAL HA | 8 DTRP HN | 2.40 |  |  |  |
| 7 | 7 NAL HB2 | 7 NAL HD1 | 2.74 |  |  |  |
| 7 | 7 NAL HB2 | 7 NAL HD3 | 2.49 |  |  |  |
| 7 | 7 NAL HB2 | 8 DTRP HN | 4.11 |  |  |  |
| 7 | 7 NAL HB3 | 7 NAL HD1 | 2.86 |  |  |  |
| 7 | 7 NAL HB3 | 7 NAL HD3 | 2.52 |  |  |  |
| 7 | 7 NAL HB3 | 8 DTRP HN | 3.92 |  |  |  |
| 7 | 7 NAL HZ | 8 DTRP HE3 | 5.50 |  |  |  |
| 7 | 7 NAL HE3 | 8 DTRP HE3 | 5.50 |  |  |  |
| 7 | 7 NAL HD3 | 8 DTRP HN | 4.82 |  |  |  |
| 8 | DTRP HN | 8 DTRP HB2 | 2.83 |  |  |  |
| 8 | DTRP HN | 8 DTRP HB3 | 2.83 |  |  |  |
| 8 | 8 DTRP HN | 8 DTRP QB | 2.53 |  |  |  |
| 8 | 8 DTRP HN | 8 DTRP HD1 | 5.50 |  |  |  |
| 8 | 8 DTRP HA | 8 DTRP HD1 | 4.82 |  |  |  |
| 8 | 8 DTRP HA | 8 DTRP HE3 | 3.02 |  |  |  |
| 8 | 8 DTRP HA | 9 LYS HN | 2.43 |  |  |  |
| 8 | 8 DTRP HB2 | 9 LYS HN | 4.32 |  |  |  |
| 8 | 8 DTRP HB3 | 9 LYS HN | 4.32 |  |  |  |
| 8 | 8 DTRP QB | 8 DTRP HE3 | 4.38 |  |  |  |
| 8 | 8 DTRP QB | 9 LYS HN | 3.80 |  |  |  |
| 8 | 8 DTRP HD1 | 9 LYS HN | 5.50 |  |  |  |


| 8 | DTRP HE3 | 9 LYS | HN | 5.16 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 8 | DTRP HE3 | 9 LYS | HA | 5.37 |  |  |  |
| 8 | DTRP HE1 | 9 LYS | QG | 6.38 |  |  |  |
| 8 | DTRP HE1 | 9 LYS | QD | 6.38 |  |  |  |
| 8 | DTRP HZ2 | 9 LYS | QD | 6.38 |  |  |  |
| 9 | LYS HN | 9 LYS | HB2 | 2.60 |  |  |  |
| 9 | LYS HN | 9 LYS | HB3 | 3.71 |  |  |  |
| 9 | LYS HN | 9 LYS | HG2 | 4.14 |  |  |  |
| 9 | LYS HN | 9 LYS | HG3 | 4.14 |  |  |  |
| 9 | LYS HN | 9 LYS | QG | 3.71 |  |  |  |
| 9 | LYS HN | 10 THR | HN | 2.77 |  |  |  |
| 9 | LYS HA | 9 LYS | QG | 3.91 |  |  |  |
| 9 | LYS HA | 9 LYS | QD | 6.38 |  |  |  |
| 9 | LYS HA | 10 THR |  | 3.33 |  |  |  |
| 9 | LYS HA | 14 DHS | HN | 5.50 |  |  |  |
| 9 | LYS HA | 15 THO | HN | 5.35 |  | 2.90 | 0.41 |
| 9 | LYS QB | 10 THR |  | 5.51 |  |  |  |
| 10 | THR HN | 10 THR |  | 2.90 |  |  |  |
| 10 | THR HN | 10 THR |  | 3.21 |  |  |  |
| 10 | THR HN | 14 DHS | HN | 2.89 |  | 0.42 |  |
| 10 | THR HA | 10 THR | HB | 2.83 |  |  |  |
| 10 | THR HA | 14 DHS | HN | 2.57 | 0.88 |  | 0.31 |
| 10 | THR HA | 15 THO |  | 5.34 |  |  |  |
| 10 | THR HB | 14 DHS | HN | 3.61 |  |  |  |
| 10 | THR QG2 | 14 DHS | HN | 6.53 |  |  |  |
| 14 | 4 DHS HN | 14 DHS | HB2 | 2.86 |  |  |  |
| 14 | 4 DHS HN | 14 DHS | HB3 | 3.39 | 0.33 |  |  |
| 14 | 4 DHS HN | 14 DHS | HG | 4.85 |  |  |  |
| 14 | 4 DHS HN | 15 THO | HN | 3.22 |  | 1.20 | 0.14 |
| 14 | 4 DHS HA | 14 DHS | HG | 3.86 |  |  |  |
| 14 | 4 DHS HA | 15 THO | HN | 2.55 | 0.68 |  |  |
| 14 | 4 DHS HB2 | 15 THO | HN | 3.80 |  | 0.80 |  |
| 14 | 4 DHS HB3 | 15 THO | HN | 4.39 |  |  |  |
| 15 | 5 THO HN | 15 THO | HB | 3.58 |  |  |  |
|  | 5 THO HA | 15 THO | HB | 2.40 |  |  |  |

${ }^{\text {a }} 3$ DHS: N-terminal portion of dehydrodiaminosuberic acid. 14 DHS: C-terminal portion of dehydrodiaminosuberic acid. NAL: 1-naphtylalanine. THO:
Threoninol. ${ }^{\text {b }}$ Violations $(\AA)$ observed for the mean structure of the family I (helix), family II (extended) and an ensemble of 20 structures of both families (ensemble).

Table S28. NOE Derived Upper Limit Constraints of Compound $\mathbf{8}$


| 9 | LYS HN | 10 TBZ | QD | 7.64 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 9 | LYS HA | 9 LYS | QG | 3.88 |  |  |  |
| 9 | LYS HA | 9 LYS | QD | 5.45 |  |  |  |
| 9 | LYS HA | 10 TBZ | HN | 3.45 |  |  |  |
| 9 | LYS HA | 14 DHS | HN | 5.00 |  |  |  |
| 9 | LYS HA | 15 THO | HN | 5.50 |  | 2.76 | 0.48 |
| 9 | LYS HA | 15 THO | QG2 | 6.51 |  | 2.10 | 0.35 |
| 9 | LYS HB2 | 10 TBZ | HN | 3.66 | 0.68 |  |  |
| 9 | LYS HB3 | 10 TBZ | HN | 4.16 | 0.20 |  |  |
| 9 | LYS QB | 10 TBZ | QD | 8.52 |  |  |  |
| 9 | LYS QG | 10 TBZ | HN | 5.50 |  |  |  |
| 9 | LYS QG | 10 TBZ | QE | 7.63 |  |  |  |
| 10 | TBZ HN | 10 TBZ | HB2 | 2.90 |  |  |  |
| 10 | TBZ HN | 10 TBZ | HB3 | 3.57 |  |  |  |
| 10 | TBZ HN | 10 TBZ | QE | 7.63 |  |  |  |
| 10 | TBZ HN | 14 DHS | HN | 2.98 |  | 0.34 |  |
| 10 | TBZ HA | 10 TBZ | HB2 | 2.93 |  |  |  |
| 10 | TBZ HA | 10 TBZ | HB3 | 2.74 |  |  |  |
| 10 | TBZ HA | 14 DHS | HN | 2.64 | 0.82 |  | 0.23 |
| 10 | ) TBZ HB2 | 14 DHS | HN | 3.95 |  | 0.46 |  |
| 10 | TBZ HB3 | 14 DHS | HN | 3.82 |  | 0.63 |  |
| 14 | 4 DHS HN | 14 DHS | HB2 | 2.96 |  |  |  |
| 14 | 4 DHS HN | 14 DHS | HB3 | 3.39 | 0.33 |  |  |
| 14 | 4 DHS HN | 14 DHS | HG | 4.51 |  |  |  |
| 14 | 4 DHS HN | 15 THO | HN | 3.21 |  | 1.21 | 0.13 |
| 14 | 4 DHS HA | 14 DHS | HB2 | 2.93 |  |  |  |
| 14 | 4 DHS HA | 14 DHS | HB3 | 2.83 |  |  |  |
| 14 | 4 DHS HA | 14 DHS | HG | 3.81 |  |  |  |
| 14 | 4 DHS HA | 15 THO | HN | 2.66 | 0.58 |  |  |
| 14 | 4 DHS HB2 | 15 THO | HN | 3.86 |  | 0.75 |  |
| 14 | 4 DHS HB3 | 15 THO | HN | 4.43 |  |  |  |
| 15 | 5 THO HN | 15 THO | HB | 3.58 |  |  |  |
| 15 | 5 THO HN | 15 THO | QG2 | 4.88 |  |  |  |
|  | 5 THO HA | 15 THO |  | 2.40 |  |  |  |

${ }^{\text {a }} 3$ DHS: N-terminal portion of dehydrodiaminosuberic acid. 14 DHS: C-terminal portion of dehydrodiaminosuberic acid. NAL: 1-naphtylalanine. TBZ: $\operatorname{Tyr}(\mathrm{Bzz})$. THO: Threoninol. ${ }^{\text {b }}$ Violations $(\AA)$ observed for the mean structure of the family I (helix), family II (extended) and an ensemble of 20 structures of both families (ensemble).

Table S29. Microanalysis data for the all final products

|  |  | Found (\%) |  |  |  |  | Calculated (\%) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Comp. | $\mathrm{Formula}^{2}$ | C | H | N | S | C | H | N | S |
| $\mathbf{1 1 a}$ | $\mathrm{C}_{16} \mathrm{H}_{11} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ | 48.00 | 2.77 | 7.00 | 8.01 | 48.15 | 2.89 | 7.05 | 8.08 |
| $\mathbf{1 1 b}$ | $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ | 49.28 | 3.16 | 6.76 | 7.74 | 49.39 | 3.03 | 6.66 | 7.63 |
| $\mathbf{1 1 \mathbf { 1 }}$ | $\mathrm{C}_{23} \mathrm{H}_{17} \mathrm{H}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ | 56.32 | 3.49 | 5.71 | 6.54 | 56.24 | 3.59 | 5.74 | 6.63 |
| $\mathbf{1 1 d}$ | $\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{~F}_{6} \mathrm{~N}_{3} \mathrm{O}_{7} \mathrm{~S}$ | 45.13 | 3.62 | 7.18 | 5.48 | 45.01 | 3.70 | 7.21 | 5.51 |
| $\mathbf{1 1 e}$ | $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ | 51.82 | 3.43 | 6.36 | 7.28 | 51.68 | 3.47 | 6.27 | 7.41 |
| $\mathbf{1 1 f}$ | $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ | 49.28 | 3.16 | 6.76 | 7.74 | 49.13 | 3.12 | 6.84 | 7.87 |
| $\mathbf{1 1 g}$ | $\mathrm{C}_{14} \mathrm{H}_{4} \mathrm{NO}_{4} \mathrm{~S}$ | 58.53 | 3.16 | 4.88 | 11.16 | 58.42 | 3.25 | 4.97 | 10.98 |
| $\mathbf{1 1 h}$ | $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{ClN}_{2} \mathrm{O}_{3} \mathrm{~S}$ | 57.06 | 5.05 | 7.39 | 8.46 | 56.96 | 5.14 | 7.30 | 8.59 |
| $\mathbf{1 1 i}$ | $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{ClN}_{2} \mathrm{O}_{4} \mathrm{~S}$ | 55.03 | 4.36 | 7.13 | 8.16 | 55.00 | 4.40 | 7.16 | 8.24 |
| $\mathbf{1 1 j}$ | $\mathrm{C}_{14} \mathrm{H}_{9} \mathrm{NO}_{3} \mathrm{~S}_{2}$ | 55.43 | 2.99 | 4.62 | 21.14 | 55.35 | 2.95 | 4.69 | 21.26 |
| $\mathbf{1 1 k}$ | $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ | 50.24 | 4.92 | 9.76 | 7.45 | 50.12 | 5.01 | 9.69 | 7.58 |
| $\mathbf{1 1 1}$ | $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{NO}_{4} \mathrm{~S}$ | 59.79 | 3.68 | 4.65 | 10.64 | 59.68 | 3.63 | 4.72 | 10.71 |
| $\mathbf{1 1 m}$ | $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{ClN}_{2} \mathrm{O}_{3} \mathrm{~S}$ | 58.08 | 5.39 | 7.13 | 8.16 | 57.98 | 5.44 | 7.02 | 8.29 |
| $\mathbf{1 1 n}$ | $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{ClN}_{2} \mathrm{O}_{4} \mathrm{~S}$ | 56.09 | 4.71 | 6.88 | 7.88 | 56.00 | 4.76 | 6.78 | 8.01 |
| $\mathbf{1 1 0}$ | $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{NO}_{3} \mathrm{~S}_{2}$ | 56.76 | 3.49 | 4.41 | 20.21 | 56.64 | 3.55 | 4.30 | 20.34 |
| $\mathbf{1 1 p}$ | $\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{Cl}_{2} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S}$ | 51.35 | 5.22 | 9.46 | 7.22 | 51.42 | 5.24 | 9.39 | 7.36 |
| $\mathbf{1 1 q}$ | $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ | 61.13 | 4.49 | 8.91 | 10.20 | 61.18 | 4.53 | 8.82 | 10.31 |
| $\mathbf{1 1 r}$ | $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{ClN}_{3} \mathrm{O}_{3} \mathrm{~S}$ | 53.75 | 4.78 | 11.06 | 8.44 | 53.83 | 4.68 | 11.11 | 8.38 |



Figure S26. Expression of cyclin A in control cells and upon treatment with 11p, 11 m , and 11c ( $1 \mu \mathrm{M}$ ) for 48 h


Figure S27. Effects of doxorubicin on the topo II-mediated DNA cleavage

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[^0]:    Abbreviations
    Abbreviations used for amino acids and designation of peptides follow the rules of the IUPAC-IUB Commission of Biochemical Nomenclature in J. Biol. Chem. 1972, 247, 977-983. Amino acid symbols denote L-configuration unless indicated otherwise. The following additional abbreviations are used:

    1D, 2D and 3D, one-, two- and three-dimensional;
    ACTH, adreno-corticotropic hormone;
    AGRP, agouti-related protein;
    Boc, tert-butyloxycarbonyl;
    Bzl, benzyl;
    cAMP, Cyclic adenosine monophosphate;
    Cpa, p-chloro-phenylalanine;
    DBU, 1,8-Diazabicyclo(5.4.9)undec-7-ene;
    DCM, dichloromethane;
    DF, differential frequency;
    dh-DSA- N , dehydrodiaminosuberic acid, N -terminus;
    dh-DSA-C, dehydrodiaminosuberic acid, C-terminus;
    DIPEA, $N, N$-diisopropylethyl-amine;
    DMF, $N, N$ dimethylformamide;
    DMSO, dimethylsulfoxide;
    DOTA, 1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid; EDT, 1,2ethanedithiol;
    DPC, dodecyl phosphocholine;
    DQF-COSY, double quantum filtered correlated spectroscopy;
    EL, extracellular loop;
    EM, energy minimization;
    ESI-MS, electrospray ionization-mass spectrometry;
    Fmoc, 9-fluorenylmethoxycarbonyl;
    GH, growth hormone; Hag, L-2-allyl-Gly;
    GPCR, G-protein-coupled receptor;
    HATU, exafluorophosphate salt of the O-(7-Azabenzotriazol-yl)-tetramethyl uranium cation (this acronym does not longer correspond to the true structure);
    HBTU,
    2-(1 H -benzotriazole-1-yl)-1,1,3,3-tetramethyluronium
    hexafluorophosphate;
    $h$-MCR, human Melanocortin Receptor,
    HOBt, Nhydroxy-benzotriazole;
    $h \mathrm{U}-\mathrm{II}$, human Urotensin-II peptide;

[^1]:    ${ }^{\text {a }}$ Obtained at $\mathrm{pH}=5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
    ${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz . exc $=\mathrm{NH}$ exchange rate (f, fast; ms, moderately slow; s, slow;). $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ) calculated in the range $25-40^{\circ} \mathrm{C}$. Further signals: $\mathrm{C}_{3} \underline{\mathrm{CO}}, 2.01 ; \mathrm{CONH}_{2}, 6.65,6.84$.

[^2]:    ${ }^{\text {a }}$ Obtained at $\mathrm{pH}=5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm}$.
    ${ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in Hz . exc $=\mathrm{N} \underline{H}$ exchange rate (f, fast; ms, moderately slow; s, slow;). $-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ) calculated in the range $25-40^{\circ} \mathrm{C}$. Further signals: $\mathrm{C}_{3} \mathrm{CO}, 1.96 ; \mathrm{CONH}_{2}, 6.65,6.84$.

[^3]:    ${ }^{a}$ Obtained at $35{ }^{\circ} \mathrm{C}, \mathrm{pH}=5.5$, with TSP ( $\delta 0.00 \mathrm{ppm}$ ) as reference shift. Chemical shifts are accurate to $\pm 0.02 \mathrm{ppm} .{ }^{\mathrm{b}}{ }^{3} J_{\alpha \mathrm{N}}$ coupling constants in $\mathrm{Hz} . \quad-\Delta \delta / \Delta \mathrm{T}=$ temperature coefficients ( $\mathrm{ppb} / \mathrm{K}$ ). ${ }^{\mathrm{c}}$ dh-DSA-N: N-terminal portion of dehydrodiaminosuberic acid. dh-DSA-C: C-terminal portion of dehydrodiaminosuberic acid.

