

Ammonium regeneration: its contribution to phytoplankton nitrogen requirements in a eutrophic environment

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Abstract

Ammonium regeneration, nutrient uptake, bacterial activity and primary production were measured from March to August 1980 in Bedford Basin, Nova Scotia, Canada, a eutrophic environment. Rates of regeneration and nutrient uptake were determined using ¹⁵N isotope dilution and tracer methodology. Although primary production, nutrient uptake and ammonium regeneration were significantly intercorrelated, no relationship was detected between these parameters and heterotrophic activity. The average contribution of ammonium to total nitrogen (ammonium + nitrate) uptake was similar in the spring and in the summer (approximately 60%). On a seasonal average basis, 36% of the phytoplankton ammonium uptake could be supplied by rapid remineralization processes. In spite of the high average contribution of NH₄ regeneration to phytoplankton ammonia uptake, there is indirect evidence suggesting that other NH₄ sources may occasionally be important.

Introduction

Nutrient regeneration is the dominant process controlling primary production in oligotrophic oceans where nutrients limit phytoplankton growth (Dugdale and Goering, 1967; Eppley *et al.*, 1973).

It is now becoming evident that nutrient regeneration also contributes to a significant fraction of the primary production in coastal areas (Eppley and Peterson, 1979; Eppley *et al.*, 1979; Harrison, 1980). The ammonia rapidly recycled in the euphotic zone is an important source of regenerated nitrogen and ammonia uptake by phytoplankton is often taken as a measure of regenerated production (Olson, 1980). In fact, a close coupling between direct measurements of NH_4 regeneration and NH_4 uptake rates has been documented in several field studies (Harrison, 1978; Axler, 1981; Glibert, 1982).

High rates of respiration and nutrient excretion by zooplankton (Conover and Mayzaud, 1975), as well as excretion rates associated with high bacterial biomass and heterotrophic activity (Palumbo and Ferguson, 1978; Billen et al., 1980; Fuhrman et al., 1980), probably account for most of the water column nutrient regeneration in coastal areas. However, the partitioning of regeneration among different groups of organisms is not well documented. Harris' (1959) early experiments in coastal waters indicated that zooplankton excretion could provide 77% of the nutrients required for primary production. More recent studies have suggested much lower contributions. ranging from 5 to 30% (Seki et al., 1974; Smith, 1978; Whiteledge, 1978). Laboratory experiments (Johannes, 1965, 1968) and field studies using size fractionation techniques (Harrison, 1978; Caperon et al., 1979; Glibert, 1982) suggested that microplankton (< $100 \,\mu m$) may provide most of the regenerated nutrients.

Bedford Basin, Nova Scotia, is a eutrophic coastal environment where nutrients are abundant year round (i.e. summer average concentrations of NO_3 and NH_4 are around 0.70 and 1.10 $\mu mol \ l^{-1}$ respectively) (Krauel, 1969; Taguchi et al., 1975; Coté, 1979). The large daily fluctuations (3 fold) in primary production observed throughout the summer in Bedford Basin are not correlated with the variations in ambient nutrient concentrations (Coté, 1979; Harrison and Platt, 1980). The present study investigates the relationship between primary production and nitrogen fluxes. The following hypotheses were tested: (1) Rapid in-situ NH4 regeneration changes proportionally with NO3 uptake (new production), NH₄ uptake (regenerated production), and CO₂ uptake (primary production); (2) Bacterial metabolic activity, measured by glutamic acid uptake, contributes significantly to nutrient remineralization and is therefore correlated with NH4 regeneration rates.

Materials and methods

Sampling

Samples were collected weekly between 09.00 and 10.30 hrs in Bedford Basin from February 29 to May 15 (spring period) and June 26 to August 18, 1980 (summer period). Salinity and temperature profiles were recorded at 2-m intervals to a depth of 20 m using a model Beckman RS-5 conductivity-temperature probe. Water samples for nutrient analysis were collected from 0, 5, 10 and 15 m using a submersible pump, and were stored in opaque plastic bottles for later nutrient analyses. An additional 20 l of water were collected at the 5-m depth using a Niskin bottle, and stored in a dark plastic carboy for later use in tracer experiments.

Analytical methods

Particulate matter collected on Reeve-Angel 984-H glass fiber filters was analysed for chlorophyll *a* by fluorometry (Strickland and Parsons, 1972) and for particulate organic matter (PON and POC) with a Hewlett-Packard 185 B CHN analyser (Sharp, 1974). Ammonium was determined colorimetrically (Solorzano, 1969) approximately one hour after collection and other nutrients were determined with a Technicon autoanalyser a few weeks later (Strickland and Parsons, 1972). Fig. 1 B shows the relationship between NH₄ concentrations determined from fresh and frozen water samples. The contamination of frozen NH₄ samples observed by Carpenter *et al.* (1972) was not found in my data. Acridine orange direct cell counts (Hobbie *et al.*, 1977) were used to enumerate bacteria.

Tracer methods

Uptake experiments began within one hour of water collection. All the incubations were carried out for 4 h under simulated *in-situ* conditions at 50% light attenuation except for the heterotrophic activity incubations, which were conducted in the dark. Samples were not prefiltered.

Photosynthetic rate was determined by $[^{14}C]$ -CO₂ tracer uptake (Strickland and Parsons, 1972). Five μ Ci of ^{14}C -NaHCO₃ were added to a 250-ml water sample and incubated in glass bottles. At the end of the incubations the particulate matter was collected by filtration onto Reeve-Angel 984-H glass fiber filters. The incorporation of ^{14}C was measured by liquid scintillation spectrometry. An estimate of heterotrophic activity was obtained by measuring ^{14}C -glutamic acid uptake (Griffiths *et al.*, 1977).

Nutrient uptake and regeneration experiments were measured using the techniques described by Harrison (in press). Nutrient uptake was measured by inoculating duplicate 1-1 water samples with 0.1 μ mol of either 95 at% ¹⁵N-NO₃ or 99 at% ¹⁵N-NH₄. NH₄ regeneration was determined from the time-dependent dilution of the dis-

solved NH₄ pool initially enriched with $1 \mu mol$ of 99 at% ¹⁵N-NH₄. Calculations were based on time zero and time final differences of ¹⁵N-NH₄ isotopic ratio. Samples were run in triplicate.

15N analysis

The filters for nutrient uptake were dried at 80 °C for 48 h and ground in a mortar with a small amount of CuO. The samples were then stored in vacutainer tubes and kept in a desiccator until they were analysed. Ammonium present in the regeneration samples was concentrated by distillation (Harrison, 1978). NH₄ recovery averaged 60 to 80%. A microdiffusion technique (Blackburn, 1979) was used to transfer the NH₄ samples into small capillary tubes. The capillaries were then placed in discharge tubes containing CaO and CuO. The tubes were attached to a vacuum line and sealed with a torch after evacuation to approximately 2×10^{-4} torrs. The sealed particulate or dissolved NH₄ samples were converted to dinitrogen gas by Dumas combustion at 550 °C for 2 h and analysed by emission spectrometry (Harrison, in press). The precision of isotope analyses was good in general. (The coefficient of variation was 4% for particulate nitrogen and 6 to 10% for NH_4 distillates).

Calculation of uptake and regeneration rates

Nutrient uptake rates were calculated according to Dugdale and Goering (1967). Their method assumes linear substrate uptake and constant specific activity during the incubation period. A linear differential equation model (Blackburn, 1979; Caperon *et al.*, 1979) was used to calculate regeneration rates. This model assumes that regeneration and uptake rates of NH_4 are constant during the incubation period and that ¹⁵N-NH₄ is not recycled.

Nutrient uptake rates may be overestimated when the amount of added substrate is large compared to ambient substrate concentrations (Eppley et al., 1973, 1977). In my study, the ammonia concentrations were usually well above the detection level and the ratio of isotope added to ambient nutrient concentration was less than 0.2 in 90% of the nutrient uptake experiments. However, for the NH₄ regeneration experiments, the amount of isotope added represented a 100% NH₄ enrichment on average. A 24-h time-course experiment was performed to verify the linearity of substrate uptake and regeneration with time and measure the enhancement of uptake rate due to NH₄ enrichment. The time-course for ¹⁵N assimilation in the particulate matter and the linear decrease in ammonia concentration observed during the regeneration experiment are shown in Fig. 1A and 1B, respectively. The time courses of nitrogen (NO₃ and NH_4) uptake were linear for 5 h, after which essentially no ¹⁵N was incorporated, thus suggesting nutrient depletion (Fig. 1C). CO₂ uptake and heterotrophic activity were linear for at least 5 h and un-







Fig. 2. (a) Net CO₂ uptake (light-dark bottles) for natural water samples (dark circles) and samples enriched with 1 μ mol of NH₄ (open circles). Error bars represent the range of duplicate measurements. The shaded area on the abscissa delimits the dark period. (b) Time-course of ¹⁴C-glutamic acid uptake with natural water samples (dark circles) and samples enriched with 1 μ mol of NH₄ (open circles). Error bars represent the range of triplicate measurements

affected by enrichment with $1 \mu mol$ of NH₄ over the same time period (Fig. 2 A and B).

Assuming that ¹⁵N labelled NH₄ is not recycled during short-term experiments (Blackburn, 1979; Caperon *et al.*, 1979), Glibert *et al.* (1982) have shown that isotope dilution due to high regeneration rates can result in a significant underestimate of uptake rates. In my experiments, a 15% difference was observed between uptake rates calculated in the conventional way (Dugdale and Goering, 1967) and uptake rates corrected for isotope dilution effects (Glibert *et al.*, 1982). The time-course experiment suggested that a 4-h incubation should minimize many of the technical problems. It reduced the chances of substrate

Results

Figure 3 illustrates the variations in several of the biological parameters measured during the sampling period. Simple correlation analysis indicated a strong correlation between temperature (Fig. 3 A) and assimilation number (mgC mgChl a^{-1} h⁻¹; Fig. 31), glutamic acid uptake (Fig. 3 L) and bacterial abundance (Fig. 3 K). The high nitrate concentration characteristic of the winter months was rapidly depleted at the onset of the spring bloom (Fig. 3 B). After the bloom, NO₃ and NH₄ concentrations as well as NO₃ and NH₄ uptake rates covaried until the end of the sampling period (Fig. 4). Similarly, rate measurements of NO₃ and NH₄ uptake, NH₄ regeneration and CO₂ uptake were all significantly intercorrelated after the

as problems related to isotope dilution.

initial NO₃ depletion at the 5 m depth. However, no relationship could be detected between any of these parameters and glutamic acid uptake (Table 1). NH₄ uptake was most highly correlated with NH₄ regeneration, however not in a 1:1 relationship (Fig. 5). Although glutamic acid uptake was not correlated with the other rate measurements, NH₄ regeneration seemed to be most important when phytoplankton and bacterial biomass were high (Fig. 3F, K). The high regeneration rate observed on July 31 is noteworthy because it corresponds to a peak in bacterial numbers and in primary production and to a minimum in nutrient concentrations. Microscopic observation of the preserved sample showed that on that date, most of the bacteria were associated with particulate matter. However, large variations in replicate bacterial cell counts complicated the statistical interpretation of this parameter.

Significant differences in several environmental and biological variables were observed between the spring and summer sampling periods (Table 2). Total nitrogen con-



Fig. 3. Variations in some of the chemical and biological variables measured during the sampling period. All variables were measured at 5 m. Error bars represent the range of replicate measurements. Variations in (A) temperature, (B) NO₃ concentration, (C) NH₄ concentration, (D) NO₅ uptake, (E) NH₄ uptake, (F) NH₄ regeneration, (G) chlorophyll *a*, (H) CO₂ uptake, (I) assimilation number, (J) particulate organic nitrogen, (K) bacterial abundance, (L) ¹⁴C-glutamic acid uptake

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Table 1. Correlation matrix of uptake and regeneration rates for data after the spring bloom. The few experiments where more than 80% of the substrate uptake occurred were omitted. $NO_3 = NO_3$ uptake (μ mol 1⁻¹ h⁻¹), $NH_40.1 = NH_4$ uptake (μ mol 1⁻¹ h⁻¹) with the addition of 0.1 μ mol of labelled substrate, $NH_41.0 = NH_4$ uptake (μ mol 1⁻¹ h⁻¹) with the addition of 1.0 μ mol 1⁻¹ of substrate, $NH_4R = NH_4$ regeneration (μ mol 1⁻¹ h⁻¹), $CO_2 = CO_2$ uptake (μ g C 1⁻¹ h⁻¹), GLU = Glutamic acid uptake ($\times 10^{-1} \mu$ g C 1⁻¹ h⁻¹). All the correlations are statistically significant (P < 0.05, d.f. = 17) except for the correlations with glutamic acid uptake

	NO_3	NH₄0.1	NH ₄ 1.0	TOTALN	NH₄R	CO_2	GLU
 NO ₃	1						
NH₄0.1	0.92	1				491	
NH41.0	0.87	0.90	1				
TOTALN	0.97	0.98	0.91	1			
NH₄R	0.68	0.74	0.66	0.73	1		
CO ₂	0.77	0.86	0.91	0.84	0.68	1	
GLŪ	0.23	0.20	0.41	0.22	0.08	0.20	1







Fig. 4. Relationship between (A) NH_4 and NO_3 concentrations and (B) NH_4 and NO_3 uptake rates for the data collected after the onset of water column stratification. The data from experiments where more than 80% of the substrate was taken up was omitted. The solid line represents the 1:1 relationship between the two parameters



Fig. 5. Relationship between NH_4 uptake and regeneration for the data collected after the onset of the water column stratification. The data, where more than 80% of the substrate added was assimilated, was omitted. The solid line represents the 1:1 relationship

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Table 2. Comparison of summer and spring mean values of selected variables. P represents the level of significance of a one tail *t*-test. Temperature is in °C, light is in langley h⁻¹, PON is in μ g N l⁻¹, NO₃, NH₄, [NO₃ + NH₄] are in μ mol l⁻¹ h⁻¹, NO₃, NH₄, regeneration rate and total nitrogen uptake are in μ mol l⁻¹ h⁻¹, Chl *a* is in μ g C ll *a* l⁻¹, C: Chl *a* is in μ g C ll *a*, CO₂ uptake is in μ g C l⁻¹ h⁻¹, A.N. is in μ g C (μ g Chl *a*⁻¹) h⁻¹, C: N uptake is mol:mol ratio, bacterial counts are in cell ml⁻¹, heterotrophic activity is in μ g X 10⁻¹ l⁻¹ h⁻¹

Variables	Spring		Summer	Р	
	Mean	SD	Mean	SD	
Temperature	2.77	1.48	10.85	1.68	0.00
Light	1.52	0.83	2.35	0.60	0.01
PŎN	93.9	33	115.5	23.9	0.08
NO ₃	2.26	2.76	0.72	0.48	0.08
NH₄	1.43	1.17	1.13	0.66	0.23
$[NH_4 + NO_3]$	3.68	2.72	1.85	1.11	0.05
NO ₃ uptake	0.085	0.077	0.105	0.089	0.50
NH₄ uptake	0.115	0.071	0.182	0.123	0.06
Total nitrogen					
uptake	0.20	0.11	0.29	0.21	0.12
Chlorophyll a	7.8	5.9	2.87	1.21	0.01
C: Chl a	36.6		115.0		0.01
CO ₂ uptake	25.24	17.24	33.23	13.07	0.20
Assimilation					
number	4.01	2.05	11.73	1.82	0.00
C/N uptake	10.99	5.32	13.67	9.53	0.21
Bacterial counts	1.67	0.70	6.23	2.69	0.00
Heterotrophic					
activity	0.75	0.54	1.22	0.76	0.03
Regenerate rate	0.041	0.08	0.066	0.12	0.28

centrations (NO₃ + NH₄) and chlorophyll *a* were lower in the summer while C:CHl *a*, assimilation number, PON, bacterial numbers and heterotrophic activity were higher. Note, on the other hand, that no significant differences were observed in the means of CO₂ uptake, NO₃ uptake, NH₄ uptake and regeneration and C/N uptake ratios.

Discussion

There exists a large record of the seasonal variations in physical, chemical and biological parameters for Bedford Basin. My data agrees very well with the recent work on yearly cycles (Harrison and Platt, 1981) and summer daily variations (Coté, 1979) of biological parameters related to phytoplankton production.

However, heterotrophic activity, NH_4 regeneration and nutrient uptake had not been measured simultaneously in this study area. My initial hypotheses were tested using data from the experiments performed after the water column became stratified (April 11). The correlation matrix presented in Table 1 supports the hypothesis that NO_3 uptake, NH_4 uptake, as well as NH_4 regeneration, changed proportionally with total primary production. However, the suggestion that bacterial activity as measured by glutamic acid uptake is correlated with NH_4 regeneration rates is not supported. There are now several field studies suggesting a tight coupling between NH_4 regeneration and NH_4 assimilation by phytoplankton. While Caperon *et al.* (1979) and Paasche and Kristiansen (1982) measured low regeneration to assimilation ratios in some coastal embayments, Harrison (1978), Axler (1981) and Glibert (1982) observed that regeneration rates were generally equal to or greater than NH_4 uptake rates. Harrison (1978) found in two coastal areas (Southern California Bight, USA and Saanich Inlet, British Columbia, Canada) that NH_4 production and NH_4 assimilation were often in a one to one balance. More recently, Glibert (1982) also found that, even on time scales of a few minutes, all of the NH_4 assimilated by phytoplankton could be supplied by rapid NH_4 regeneration. Caperon *et al.* (1979) observed in

Table 3. Estimated contribution of bacterial activity and microplankton regeneration to the nitrogen requirements of phytoplankton using mean values for spring (February 29 – May 15), summer (June 26 – August 18) and post-bloom (average) data set. A C/N composition ratio of 6.6 was used to convert the mean CO₂ uptake rate to nitrogen uptake rate. All the rates are expressed in μ mol l⁻¹ h⁻¹ and ratios as percentages. (A) Contribution of (NO₃ + NH₄) uptake to the total nitrogen requirements and contribution of NH₄ uptake to (NO₃ + NH₄) uptake, (B) Contribution of NH₄ regeneration to NH₄ uptake. (C) Contribution of bacterial activity to NH₄ regeneration

	Spring	Summer	Average
(A) Nitrogen demand set by CO ₂ assimilation	0.319	0.420	0.370
Nitrogen uptake measured by NO_3 and NH_4 assimilation	0.200	0.287	0.240
Nitrogen assimilation/ nitrogen demand (%)	62.7%	68.3%	65.5%
NH_4 assimilation/ ($NO_3 + NH_4$) assimilation	58.0%	63.0%	60.5%
 (B) NH₄ assimilation NH₄ regeneration NH₄ regeneration/ NH₄ assimilation NH₄ regenerated during 4 h/ ambient [NH₄] 	0.115 0.041 36.0% 11.5%	0.182 0.066 36.0% 23.0%	0.147 0.056 36.0% 17.5%
% of NH ₄ regenerated assimilated by phytoplankton	38.0%	39.5%	38.7%
(C) Bacterial nitrogen excretion	0.0057	0.0135	0.0092
Bacterial nitrogen excretion/ NH₄ regeneration (%)	13.9%	18.5%	16.4%
Maximum bacterial nitrogen excretion*	0.0106	0.0191	0.0132
Max bacterial N excretion/ NH₄ regeneration	25.9%	26.2%	26.0%

* Maximum bacterial excretion was calculated from the upper value of a 95% confidence interval of the mean respiration rate of glutamic acid (see "Discussion" for more details)

Kaneohe Bay, Hawaii, USA an enhancement of NH₄ production at night that could balance the uptake of NH₄ on a daily basis. Additionally, Axler's (1981) study of a subalpine lake suggested that a large proportion of regenerated nutrients was available to the phytoplankton during the night. Glibert (1982) observed diel variations in regeneration to uptake ratios in the Sargasso sea. On the other hand, she also measured NH₄ regeneration rates equal to NH₄ uptake rates in short-term experiments performed in daylight hours. Results from remineralization studies are difficult to compare when experimental conditions (i.e. incubation time, amount of isotope added, time of study, etc.) differ considerably (Harrison, in press). For example, the Saanich Inlet measurements were made from columns of seawater contained in polyethylene cylindrical enclosures. In this type of experimental system where seawater is confined, physical processes replenishing nutrients are decreased and NH₄ regeneration may become a relatively more important mechanism of nutrient renewal. This can help to explain the high regeneration: assimilation (R:A) ratio measured in this coastal embayment.

The average seasonal importance of NH₄ and NO₃ uptake, NH₄ regeneration and bacterial activity to the nitrogen requirements of phytoplankton were estimated and are presented in Table 3. Phytoplankton nitrogen requirements were estimated from photosynthetic measurements, assuming an C:N assimilation ratio of 6.6 (Table 3 A; Redfield et al., 1963). Individual POC: PON ratios or the C:N ratios derived from the slopes of the linear regressions of POC and PON on Chla supported this assumption. Physical conditions were very different during the spring and the summer sampling periods. The increase in POC/Chl a possibly indicates a change in the physiological state of the phytoplankton (Healy and Hendzel, 1980) or a shift in species composition (Malone, 1977). However, the average estimates of the relative contribution of NH₄ uptake and regeneration to the phytoplankton nitrogen requirements were similar for both periods. NH4 assimilation rates accounted for 60% of the phytoplankton total nitrogen $(NH_4 + NO_3)$ uptake on the average. Except in regions of strong upwelling (Whiteledge, 1978), NH₄ is often more important than NO₃ as a source of nitrogen for phytoplankton (see Harrison, 1980). The results of my study are no exception to this more general observation.

The seasonal average of NH_4 regeneration to assimilation observed in my study was relatively high (36%; Table 3 B). However, the contribution of regeneration to NH_4 uptake was very small on several days during the sampling period (Fig. 6). On these dates, phytoplankton probably utilised the ammonia that had accumulated in the euphotic zone as a result of processes other than rapid *in-situ* NH_4 regeneration. Although alternative mechanisms of NH_4 supply cannot be identified from my data, one can speculate on the basis of indirect evidence that the deeper water of Bedford Basin may be a reservoir of NH_4 , supplying nutrients to the euphotic zone. Fig. 7 shows the strong correlation between nitrate and NH_4 concentrations observed at 4 depths. The ammonia concentrations mea-



sured at 10 and 15 m (i.e. below the euphotic zone) are in general higher than nitrate concentrations found at the same depths. The high ammonia concentrations found at depth could result from benthic regeneration processes (Sorensen, 1978; Blackburn, 1979) or may simply result from an increase in R:A below the euphotic zone (Harrison, 1978). Although my data suggest that sources of NH₄ other than *in-situ* regeneration are important, further work is needed in order to identify the mechanisms involved. The variability of the R:A suggested that in Bedford Basin, new and regenerated production as defined by Eppley and Peterson (1979) cannot be identified only on the basis of ammonia and nitrate uptake rates.

Aside from examining the possible role of microheterotrophs, my study made no attempt to identify the



Fig. 7. Relationship between NH₄ and NO₃ concentrations for the data collected after the spring bloom. $\bullet = \text{surface}, \circ = 5 \text{ m},$ $+ = 10 \text{ m}, \mathbf{x} = 15 \text{ m}$. The solid line represents the 1:1 relationship

Fig. 6. Variations in the regeneration to assimilation ratio of NH_4 (R:A) with time. The dotted line represents the seasonal average of R:A (36%)

organisms responsible for NH_4 regeneration. However, copepod excretion rates measured previously in the same study area (Conover and Mayzaud, 1975) suggested that these organisms could account for only about 5% of the NH_4 regeneration rates I observed. Size fractionation studies of nutrient regeneration have generally been consistent in suggesting that the microplankton provide most of the remineralized NH_4 (Harrison, 1978; Caperon *et al.*, 1979; Axler, 1981; Glibert, 1982; Paasche and Kristiansen, 1982).

A maximum production of NH_4 by microheterotrophs was estimated from glutamic acid respiration measurements using the following assumptions:

(1) Observed substrate respiration rates were maximal.

(2) All the amino acids present in seawater and utilised by microheterotrophs are assimilated at comparable rates. This assumption is supported by several workers who found that turnover rates of several amino acids were similar (Crawford *et al.*, 1974; Wright, 1974; Hollibaugh *et al.*, 1980).

(3) Respiration and ammonium excretion are equatable. If it is assumed that all the nitrogen in the metabolized amino acids is excreted, then each mole of amino acid will yield on average 1.5 mol NH_4 .

(4) Several studies of amino acid abundance in seawater showed that the composition of the amino acid pool is relatively constant and glutamic acid represents 5 to 13% of the total pool (Siegel and Degens, 1966; Riley and Segar, 1970; Clark, 1972; Daumas, 1976; Williams *et al.*, 1976; Billens *et al.*, 1980; Dawson and Gocke, 1980).

Using these assumptions and estimates of heterotrophic activity, the maximum estimated contribution of bacteria to NH_4 regeneration was 25% with an average 16% for spring and summer (Table 3 C). J. La Roche: Ammonium regeneration in a eutrophic environment

Glutamic acid uptake was related with temperature and bacterial numbers only. Thus my results indicated that this measure of bacterial activity from short-term incubations was not useful in predicting regeneration fluxes in this environment.

Although the observations made in this study contradict the general idea that bacterioplankton is responsible for a large fraction of the nutrient recycling, other conflicting results have been observed in recent works. For example, Hollibaugh et al. (1980) suggested, on the basis of indirect evidence, that nitrogen excretion by microheterotrophs could provide a significant fraction of the total microplankton NH₄ regeneration. However, other workers (Fuhrman et al., 1980) failed to demonstrate a direct relationship between heterotrophic activity and processes related to primary production. They observed that thymidine uptake rates in a coastal area were significantly correlated with Chl a and other measures of phytoplankton standing stock but not with primary production. These conflicts may be resolved by giving more attention to the methodology. A time lag may exist between the assimilation of the organic molecule and the excretion of metabolites (Iturriaga and Zsolnay, 1981). For example, Horrigan and McCarthy (1981) recently observed preferential incorporation of nitrogen over carbon during ¹⁴C labelled urea uptake of phytoplankton cultures. This type of selectivity for a particular element of an organic molecule may also occur in substrate assimilation by heterotrophs. Although there is evidence that most amino acids have similar turnover rates (Billen et al., 1980), it is not clear that the use of a single substrate is appropriate for estimating bacterial nitrogen excretion in coastal eutrophic waters (Fuhrman et al., 1980; Hollibaugh et al., 1980). Furthermore, the relative importance of the bacterial contribution to the total community nutrient regeneration may vary seasonally (Taguchi and Platt, 1977). However, the results obtained from the methodology described in this paper indicated that bacterial NH₄ remineralization was apparently unimportant in the near-surface water during the time of my study.

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