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Acharya, Hitarth H., 2004, “*Discovering The New Heterocycles of Therapeutic Interest*”, thesis PhD, Saurashtra University

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**DISCOVERING THE NEW
HETEROCYCLES OF
THERAPEUTIC INTEREST**

A THESIS
SUBMITTED TO THE
SAURASHTRA UNIVERSITY
FOR THE DEGREE OF

Doctor of Philosophy
IN
THE FACULTY OF SCIENCE (CHEMISTRY)

BY
Acharya Hitarth H.

UNDER THE GUIDANCE
OF
Dr. H. H. Parekh

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The work included in the thesis is my own work under the supervision of **Dr. H. H. Parekh** and leads to some contribution in chemistry subsidised by a number of references.

Date : - 02- 2004

(Acharya Hitarth H.)

Place : Rajkot

This is to certify that the present work submitted for the Ph. D. Degree of Saurashtra University by **Acharya Hitarth H.** is his own work and leads to the advancement in the knowledge of chemistry. The thesis has been prepared under my supervision.

Date : - 02- 2004

Place : Rajkot

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ACKNOWLEDGEMENTS

*It is with great pleasure and proud privilege that I wish to express from the deepest core of my heart, the feelings of reverence and indebtedness to my learned Guide and Esteemed teacher **Dr.H.H.Parekh**, Prof. and Head, Department of Chemistry, Saurashtra University, for her faith in me, her conviction, inspiration, diligence and ever vigilant guidance without which the aims and objectives of the present work could not have been achieved.*

*I owe a great deal to **Dr. A.R.Parikh**, Rtd. Professor and Head, Department of Chemistry, Saurashtra University, who always showed deep concern and was always approachable in time to show the silver lining in every dark cloud that came across and his contribution continues to be the high spot of excellence.*

*I wish to thank **Dr. N.A.Chauhan**, for his constant guidance and moral support during the course of my research work.*

*Above all, needless to say “Thanks” is insufficient gratitude for my mother, **Devangnaben** for her unsurpassable devotion to the family, my father, **Harshendubhai** for his uncompromising principles which are guiding my life and my Grandmother **Lt.Jivantikaben**, who backed me all through the course of my study with her blessings. I would be remiss if I failed to express my special gratitude to my younger brother, **Yatin** whose unstopping flow of love helped me to reach the goal.*

*As with the completion of this task, I find myself in difficult position on attempting to express my deep indebtedness to **Dr. Ranjan**, **Dr. Neela** and **Dr. Fatema**. I wish to thank **Ashish** for his most willing co-operation and comprehensive exchange of ideas during the course of my research work.*

*I feel lucky and very proud to have intimate friends like **Bharat**, **Mukesh**, **Rajendra**, **Parag**, **Mayur** and **Milan** who have been always participating with my problems and disappointments and rebuilt my confidence at appropriate stages.*

I offer my heartfull gratitude to Prafull, Dinesh, Snehal, Siddharth, Ashok, Kanji, Harshad, Ravi and all my seniors for their support and constructive criticism at various stages. I also like to thank my juniors who proved at all times a valiant comrade-at-arms.

I am very grateful to Mr. Devendra Goswami, who shepherded this thesis from manuscripts to finished pages.

I am thankful to the authorities of CDRI Lucknow, CIL Chandigarh, Dr. V. K. Kansal, Vice President, R & D Department, Alembic Ltd., Baroda for allowing us to use his library facility which is highly benifited and to the authoroties of Saurashtra University for providing all the necessary facilities and Junior Research Felloship during my research work.

My heart felt thanks are due to the esteemed organisation Tuberculosis Antimicrobial Acquisition and Co-ordinating Facilities, Alabama U.S.A. for kind cooperation extended by them for biological screening.

Finally, each individual creature on this beautiful planet is created by God to fulfil a particular role. Whatever I have achieved in life is through His help, and an expression of His will. He showered His grace on me through some outstanding teachers and colleagues and when I pay my tributes to these fine persons, I am merely praising His glory. All this work is His work through a small person called Hitarth.

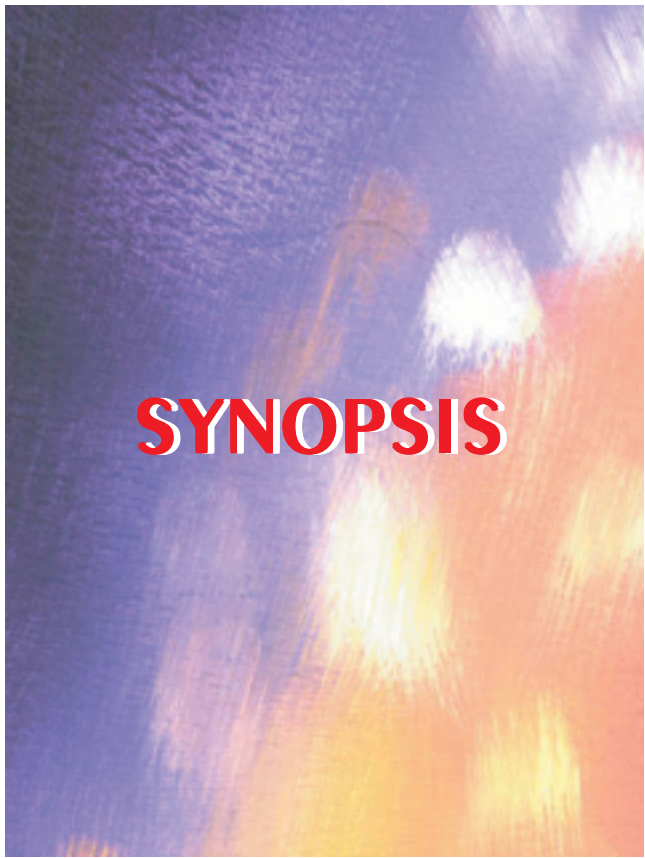
Acharya Hitarth H.

CONTENTS

	Page No.
SYNOPSIS	01
DISCOVERING THE NEW HETEROCYCLES OF THERAPEUTIC INTEREST	
Introduction	10
STUDIES ON QUINOLINE DERIVATIVES	
Introduction	13
PART - I : STUDIES ON PYRAZOLINES	
Introduction	20
Section - I : Synthesis and therapeutic evaluation of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones	
Introduction and Spectral studies.. .. .	27
Experimental	34
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>	38
Primary Antitubercular Assay Results	39
Section - II : Synthesis and therapeutic evaluation of 1-Acetyl-3-aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines	
Introduction and Spectral studies.. .. .	40
Experimental	45
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>	47
Primary Antitubercular Assay Results	48
Section - III : Synthesis and therapeutic evaluation of 1H-3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines	
Introduction and Spectral studies.. .. .	49
Experimental	54
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>	56
PART - II : STUDIES ON CYANOPYRIDINES	
Introduction	57
Section - I : Synthesis and therapeutic evaluation of 2-amino-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl pyridines	
Introduction and Spectral studies.. .. .	61
Experimental	66
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>	68
Primary Antitubercular Assay Results	69

PART - III	:	STUDIES ON CYANOPYRIDONES		
Introduction	70
Section - I	:	Synthesis and therapeutic evaluation of 1,2-Dihydro-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl-2-pyridones		
Introduction and Spectral studies..	74
Experimental	79
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>			..	81
PART - IV	:	STUDIES ON BARBITONES		
Introduction	82
Section - I	:	Synthesis and therapeutic evaluation of [1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propenylidene]-5-barbituric acids		
Introduction and Spectral studies..	85
Experimental	90
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>			..	92
PART - V	:	STUDIES ON ISOXAZOLES		
Introduction	93
Section - I	:	Synthesis and therapeutic evaluation of 3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-isoxazoles		
Introduction and Spectral studies..	97
Experimental	101
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>			..	103
PART - VI	:	STUDIES ON IMIDAZOLINONES		
Introduction	104
Section - I	:	Synthesis and therapeutic evaluation of 1N-Aryl-2-phenyl-4-(2',7'-dichloroquinolin-3'-methinyl)-5-imidazolinones		
Introduction and Spectral studies..	109
Experimental	114
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>			..	116
PART - VII	:	STUDIES ON 5-ARYLIDENE-4-THIAZOLIDINONES		
Introduction	117
Section - I	:	Synthesis and therapeutic evaluation of 2-Arylamino-3N-aryl-5-(2',7'-dichloroquinolin-3'-methinyl)-4-thiazolidinones		
Introduction and Spectral studies..	122
Experimental	126
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>			..	129
PART - VIII	:	STUDIES ON THIAZOLIDINOPYRIMIDINONES		
Introduction	130

Section - I : Synthesis and therapeutic evaluation of 7-(2',7'-dichloroquinolin-3'-yl)-				
2-arylimino-3N-aryl-thiazolidino [4,5 -d] pyrimidine-4,5,6,7-tetrahydro-5-ones				
Introduction and Spectral studies..	135
Experimental	140
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>			..	142
 PART - IX : STUDIES ON α-ARYLAMINONITRILES				
Introduction	143
Section - I : Synthesis and therapeutic evaluation of α -Arylamino- α -				
dichloroquinolin-3-yl-acetonitriles				
Introduction and Spectral studies..	147
Experimental	152
<i>Graphical data of In Vitro Evaluation of Antimicrobial screening</i>			..	154
 REFERENCES		155
 LIST OF NEW COMPOUNDS	177



SYNOPSIS



Synopsis

DISCOVERING THE NEW
HETEROCYCLES OF
THERAPEUTIC INTEREST

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September - 2003

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SYNOPSIS of the Thesis to be submitted to the Saurashtra University for the degree of **Doctor of Philosophy** in Chemistry.

Faculty : Science

Subject : Chemistry

Title : **"DISCOVERING THE NEW HETEROCYCLES OF THERAPEUTIC INTEREST"**

Name of the Candidate : **Acharya Hitarth H.**

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(GUJARAT STATE - INDIA)

A brief summary of the work to be presented in the thesis entitled "DISCOVERING THE NEW HETEROCYCLES OF THERAPEUTIC INTEREST" has been described as under.

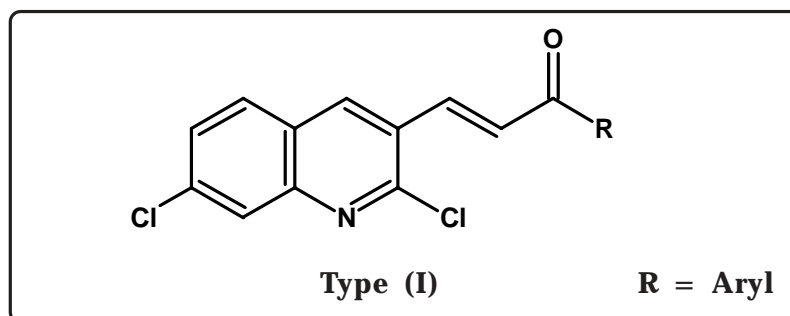
[A] STUDIES ON QUINOLINE DERIVATIVES

The group of the compounds containing quinoline moiety are a prominent structural feature in a variety of natural products like echiopsine, quinine and cinchonine as well as in other compounds of medicinal interest and have attracted attention for their biological activities viz. CNS depressant, hypnotic, sedatives, antimalarial, antiinflammatory, fungicidal, analgesic, antitumor, antiulcer, anticonvulsant, antidiabetic, antiarrhythmic, antitumor, antagonist and antidepressant. Taking in view of the applicability of heterocyclic compounds, we have undertaken the synthesis of heterocycles like pyrazolines, isoxazoles, imidazolines, cyanopyridines, cyanopyridones, pyrimidinones, arylidines, barbitones and nitriles bearing 2,7-dichloroquinoline-3-carboxaldehyde nucleus. Our efforts are focused on the introduction of chemical diversity in the molecular frame work in order to synthesise pharmacologically interesting compounds of widely different composition.

PART - I : STUDIES ON PYRAZOLINES

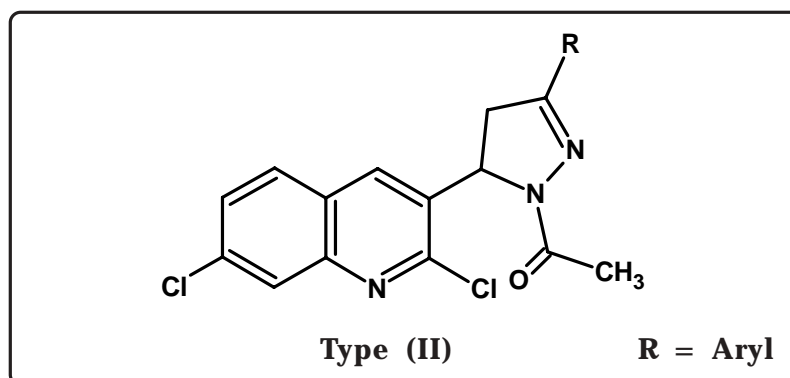
Pyrazolines have been found to possess bactericidal, fungicidal, antiviral and other pharmacological properties like tuberculostatic, local anaesthetic, CNS depressant, anticonvulsant, antiinflammatory, analgesic, antipyretic, antidepressant and antifertile activities. Hence it was thought worthwhile to synthesise pyrazoline derivatives for better drug potential, which have been described in following sections.

SECTION - I : Synthesis, characterisation and therapeutic evaluation of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones



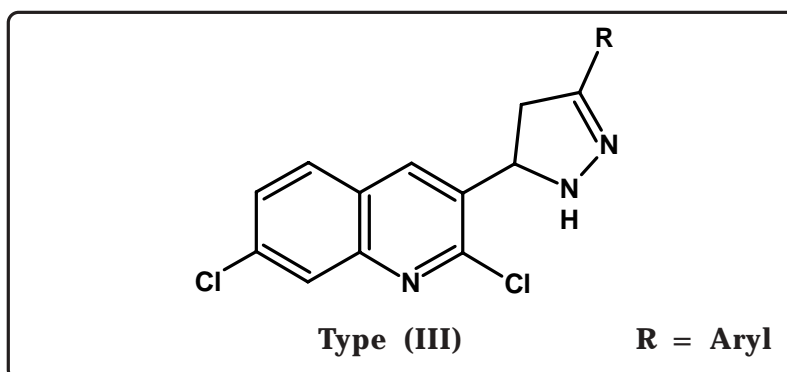
The chalcones of type (I) have been synthesised by the condensation of 2,7-dichloroquinoline-3-carboxaldehyde with various aromatic ketones.

SECTION - II : Synthesis, characterisation and therapeutic evaluation of 1-Acetyl-3-aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines



Pyrazolines of type (II) have been prepared by the condensation of the chalcones of type (I) with hydrazine hydrate and acetic acid.

SECTION - III : Synthesis, characterisation and therapeutic evaluation of 1H-3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines

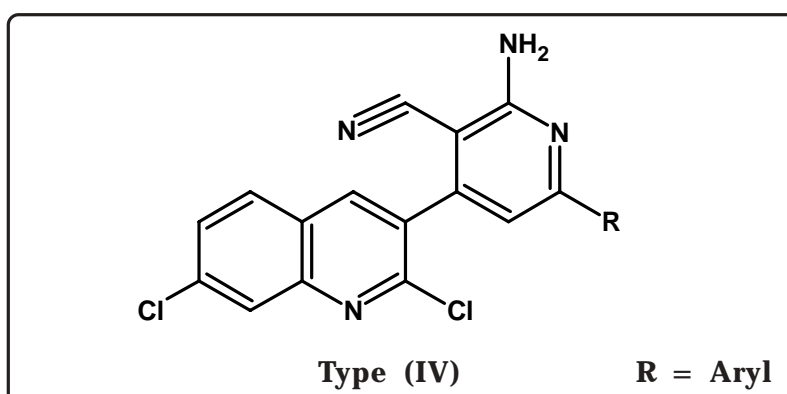


Pyrazoline derivatives of type (III) have been prepared by the condensation of the chalcones of type (I) with hydrazine hydrate.

PART - II : STUDIES ON CYANOPYRIDINES

Like other heterocyclic compounds, pyridine with different functional groups exhibit wide range of applications in the field of pharmaceutical and agriculture. Some cyanopyridine derivatives have been found to be associated with pharmacological activities such as antifungal, antidiabetic, anticholestemic and antihypertensive. They have been also used for dyes for cotton and polyester fabrics. On the basis of these results, we have synthesised new derivatives which have been described as under.

SECTION - I : Synthesis and therapeutic evaluation of 2-Amino-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl pyridines

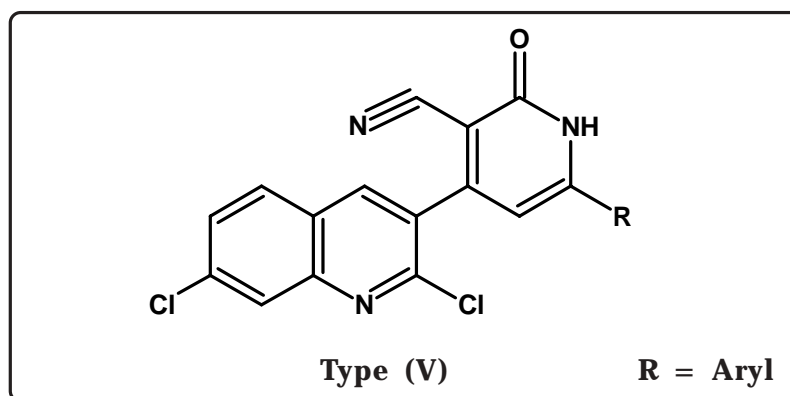


2-Amino-3-cyanopyridine derivatives of type (IV) have been synthesised by the condensation of chalcones of type (I) with ammonium acetate and malononitrile.

PART - III : STUDIES ON CYANOPYRIDONES

Recently substituted cyanopyridone derivatives have drawn considerable attention due to their good pharmacological activities like anticonvulsant, antibacterial, antiinfective and antimalarial. These observations prompted us to combine this nucleus into well known pharmaceutical properties of 2,7-dichloroquinoline-3-carboxaldehyde moiety, so as to enhance the overall activities of resulting moiety, which have been described as under.

SECTION - I : Synthesis and therapeutic evaluation of 1,2-Dihydro-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl-2-pyridones

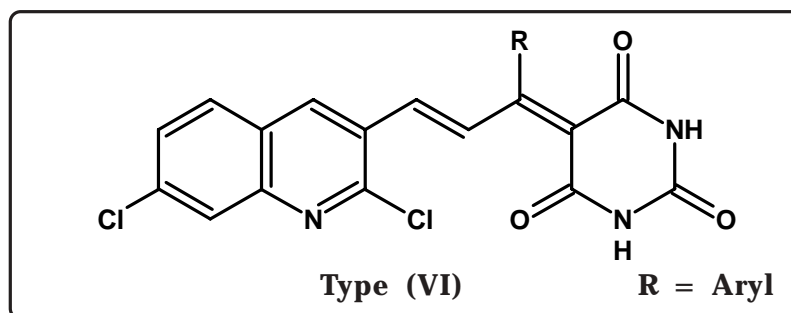


Cyanopyridone derivatives of type (V) have been synthesised by the condensation of chalcones of type (I) with ammonium acetate and ethyl cyano acetate.

PART - IV : STUDIES ON BARBITONES

The group of compounds containing the barbituric acid ring system have a prominent feature in medicinal chemistry and possess biological activities such as hypnotic, herbicidal and anticonvulsant. In view of these valid observations it was contemplated to synthesise some new barbitones possessing higher biological activity which have been described as under.

SECTION - I : Synthesis and therapeutic evaluation of [1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propenylidene]-5-barbituric acids

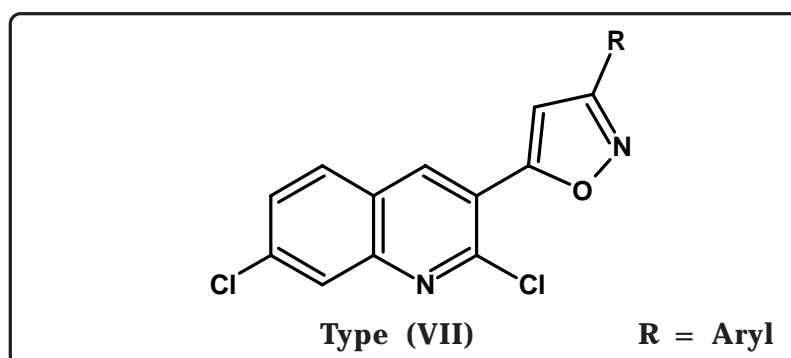


Barbitone derivatives of type (VI) have been synthesised by the condensation of chalcones with barbituric acid.

PART - V : STUDIES ON ISOXAZOLES

Isoxazole derivatives represent one of the modest class of compounds possessing wide range of therapeutic activities such as antidepressant, skeleton muscle relaxant, analgesic and sedative. Keeping this in view, it was considered of interest to synthesise some novel isoxazole derivatives as under.

SECTION - I : Synthesis and therapeutic evaluation of 3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-isoxazoles

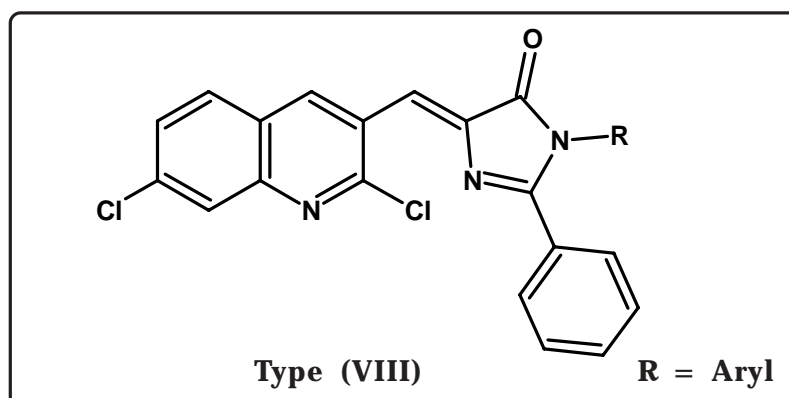


Isoxazole derivatives of type (VII) have been synthesised by the condensation of the chalcones of type (I) with hydroxylamine hydrochloride.

PART - VI : STUDIES ON IMIDAZOLINES

5-Oxo imidazoline derivatives have been reported to be active as antiinflammatory, potent CNS depressant, sedatives, hypnotics, anticancer agent and mono amino oxidase (MAO) inhibitor. Thus significant biological properties associated with imidazolinone derivatives have aroused considerable interest to design the compounds in which 2,7-dichloroquinoline-3-carboxaldehyde nucleus is incorporated with a view to getting compounds with better drug potential, which have been described as under.

SECTION - I : Synthesis and therapeutic evaluation of 1N-Aryl-2-phenyl-4-(2',7'-dichloroquinolin-3'-methinyl)-5-imidazolinones

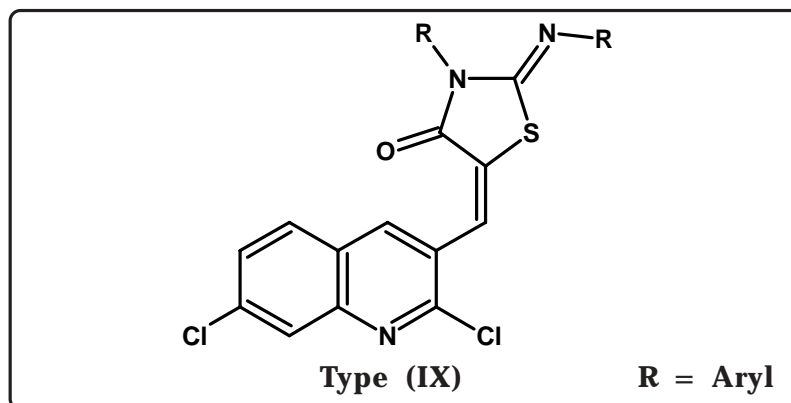


The compounds of type (VIII) have been prepared by the condensation of different amines with azalactones of 2,7-dichloroquinolin-3-carboxaldehyde, which in turn were prepared by well known Erlenmeyer azalactone synthesis.

PART - VII : STUDIES ON 5-ARYLIDENE-4-THIAZOLIDINONES

The interesting biological activities of a novel heterocycle like thiazolidinone have stimulated considerable research work in recent years leading to the synthetic utility of the derivatives of this ring system. In order to achieve compounds having better drug potential, we have combine 2,7-dichloroquinoline-3-carboxaldehyde with different substituted thiazolidinones. The pharmacological profile of the synthesised compounds have been studied and described as under.

SECTION - I : Synthesis and therapeutic evaluation of 2-Arylimino-3N-aryl-5-(2',7'-dichloroquinolin-3'-methinyl)-4-thiazolidinones

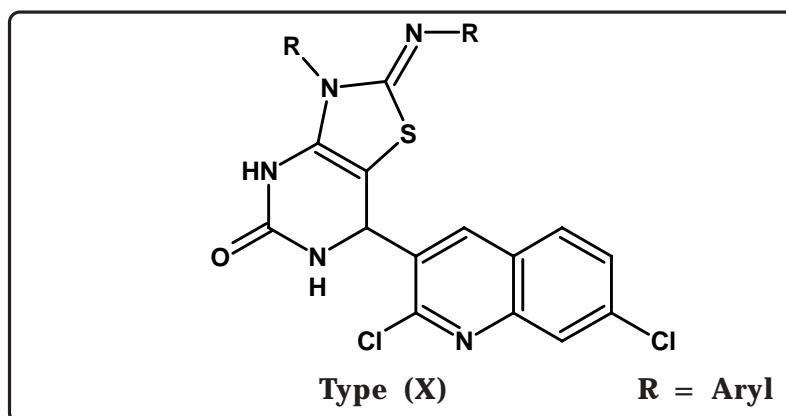


The compounds of type (IX) have been prepared by the condensation of 2,7-dichloroquinoline-3-carboxaldehyde with different 4-thiazolidinone derivatives.

PART - VIII : STUDIES ON THIAZOLIDINOPYRIMIDINONES

Thiazolidinopyrimidinone derivatives represent one of the most active class of compounds possessing a wide spectrum of biological activity such as significant *invitro* activity against unrelated DNA and RNA viruses, antimalarial, diuretic, antibacterial and spermicidal. Led by these considerations, some new thiazolidinopyrimidinone derivatives were prepared, which have been described as under.

SECTION - I : Synthesis and therapeutic evaluation of 7-(2',7'-Dichloroquinolin-3'-yl)-2-arylimino-3-arylthiazolidino[4,5-d]pyrimidine-4,5,6,7-tetrahydro-5-ones

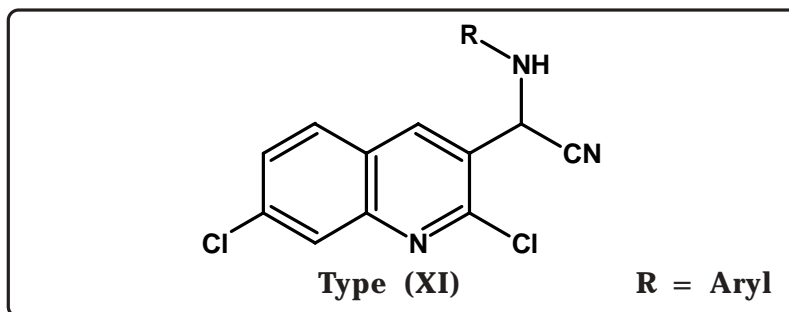


The compounds of type (X) have been prepared by the reaction of 2,7-dichloroquinoline-3-carboxaldehyde with urea, conc HCl and different 4-thiazolidinones.

PART - IX : STUDIES ON α -ARYLAMINONITRILES

The growing potent literature of recent years demonstrates that the nitrile derivatives are used as better herbicidal, antifungal and antibacterial agents. Prompted by above facts some newer nitriles have been synthesised which have been described as under.

SECTION - I : Synthesis and therapeutic evaluation of α -Arylamino- α -2,7-dichloroquinolin-3-yl-acetonitriles.

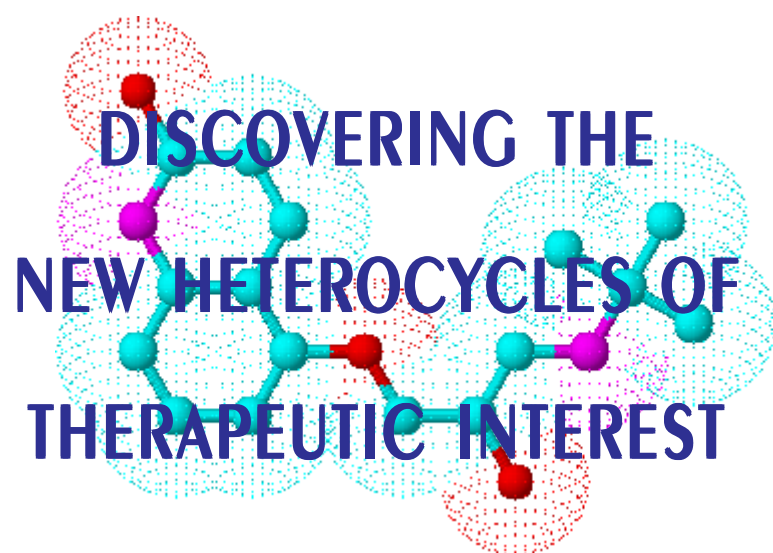


The compounds of type (XI) have been synthesised by the condensation of 2,7-dichloroquinolin-3-carboxaldehyde with potassium cyanide and various aromatic amines.

The constitution of the newly synthesised products have been characterised using elemental analysis, Infrared and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry. Purity of all the compounds has been checked by thin layer chromatography.

In vitro study on multiple biological activities:

- (i) All the compounds have been evaluated for their antibacterial activity towards Gram positive and Gram negative bacterial strain and antifungal activity towards *Aspergillus niger* at a concentration of 40 µg/ml. The biological activity of the synthesised compounds have been compared with standard drug.
- (ii) Selected compounds have been evaluated for their *invitro* biological assay like antitubercular activity towards a strain of *Mycobacterium tuberculosis H 37 Rv* at a concentration of 6.25 µg/ml using Rifampin as a standard drug, which have been tested by Tuberculosis Antimicrobial Acquisition Co-ordinating Facility (TAACF), Alabama U.S.A.



INTRODUCTION

The current interest in the creation of large, searchable libraries of organic compounds has captured the imagination of organic chemists and the drug discovery community. Efforts in numerous laboratories focused on the introduction of chemical diversity have been recently reviewed and pharmacologically interesting compounds have been identified from libraries of widely different composition.

Today our world increasingly is conceived of as being molecular. An ever widening range of phenomena are described logically in terms of molecular properties and molecular interactions. The majority of known molecules are heterocyclic and heterocycles dominate the fields of biochemistry, medicinal chemistry, dye stuffs, photographic science and are of increasing importance in many others, including polymers, adhesive and molecular engineering.

Heterocycles, containing nitrogen are most abundant in nature than those containing oxygen or sulfur. Considerable diversity in the ring system is possible. The number of atoms in the heterocyclic ring can range from three to many. The smallest ring possible is three membered, i.e. ethylene oxide, but very large rings are possible in the case of crown ethers.

Most of the alkaloids, which are nitrogenous bases occurring in plants and many antibiotics including penicillin and streptomycin also contain heterocyclic ring system. Many natural pigments such as indigo, hemoglobin and anthocyanin are heterocycles. Most of the sugars, their derivatives including vitamin C for instance, exist largely in the form of five membered or six membered ring containing nitrogen. Life saving drugs, poisons and medicines (both natural and synthetic) such as sulfathiazole, pyrethrin, rotenone, strychnine, reserpine, the ergot alkaloids caffeine, cocaine, barbiturates etc. are heterocyclic.

The process of drug design is extensively driven by instincts, intuition and experience of pharmaceutical research scientists. It is often instructive to attempt to "capture" these experiences by analyzing the historical research that is successful drug design projects of past. From this analysis, the interference is drawn to play an important role in shaping our own current and future projects. Towards this region, we would like to analyse the structure of a large number of drugs, the ultimate product of a successful drug design effort. Our goal for this is to deconvolute this information in order to apply it to design new drugs.

A drug is a substance having abnormal effect on certain body functions, eg. Strichnine stimulates the action of heart and Aspirin still its action. Since both of them effects abnormally, both are known as drugs. Chemical sciences contributed extensively new discoveries leading to useful drugs since 1930. The modern concept of drug discovery started in 1933 by Gerhand Dogmak with his finding of "Protonsil Red" a compound responsible for the antibacterial activity. The advent of sulfonamides drew attention to the different activity of various chemicals for bacterial and human cells, this important factor prompted the Flory and Chain in 1939 to investigate penicillin which was discovered ten years earlier by Alexander Fleming. The spectacular chemotherapeutical properties of penicillin and its dramatic war time development for the treatment of wounds made penicillin, a most commonly used inexpensive drug.

The "Golden Period" of new drug discovery was 1930 to 1960, a very large number of important drugs had been introduced during that period, such as:

NAME OF DRUGS	YEAR	USAGES
Sulfa drugs	1933	First antibacterial drug
Penicillin	1940	Antibiotic
Chloroquine	1945	Antimalarial
Methyl Dopa	1950	Antihypertensive

Chlorthiazide	1957	Diuretic
Adrenergic beta blockers	1958	Coronary Vasodilatory
Semi synthetic penicillin	1960	Antibacterial
Trimethoprim	1965	Antimicrobial
Disodium chromoglycoate	1970	Antiallergic

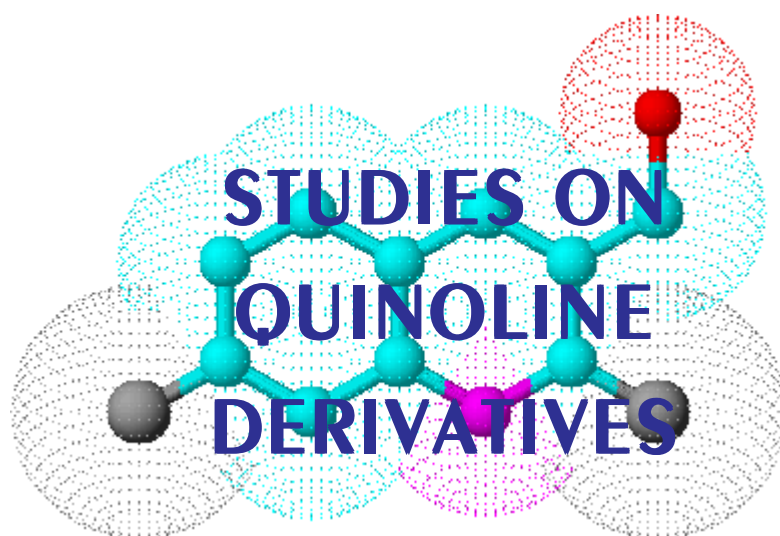
AIMS AND OBJECTIVES

In the pharmaceutical field, these have always been and will continue to be a need for new and novel chemical inhibitors of biological function. Our efforts are focused on the introduction of chemical diversity in the molecular frame work in order to synthesizing pharmacologically interesting compounds of widely different composition.

During the course of research work, looking to the application of heterocyclic compounds, several entities of heterocyclic compounds have been designed, generated and characterised using spectral studies. The details are as under.

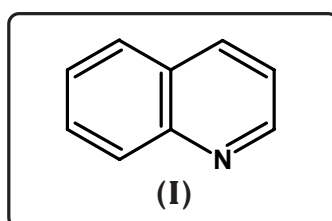
- (i) To generate several derivatives like chalcones, pyrazolines, isoxazoles, cyanopyridones, pyrimidinones, imidazolones bearing quinoline ring system.
- (ii) To synthesise arylidenes, cyanopyridines, barbitones, nitriles bearing quinoline moiety.
- (iii) To characterise these product for structure elucidation using spectroscopic techniques like IR, PMR and mass spectral studies.
- (iv) To check the purity of all compounds using thin layer chromatography.
- (v) To evaluate these new products for better drug potential against different strains of bacteria and fungi.

Taking in view of the applicability of heterocyclic compounds, the construction of new heterocycles of therapeutic importance, bearing quinoline nucleus have been investigated in following parts.



INTRODUCTION

The quinoline contains a phenyl ring fused to pyridine ring, and also known as benzopyridine (I).

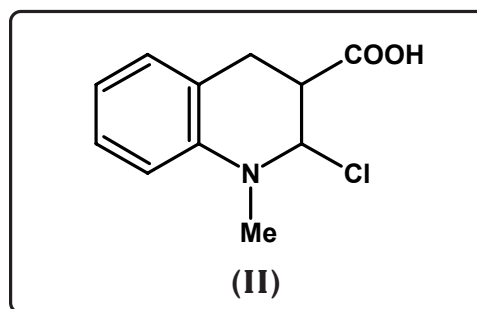


Quinoline is a very important heterocyclic compound as it is featured in a variety of natural products, as well as in other compounds of medicinal interest and have attracted attention for their biological activities. The chemistry of quinoline and its derivatives, has been the centre of attention for long time. R. H. Manske¹ and F. W. Bergstrom² has done excellent work on quinoline and its derivatives.

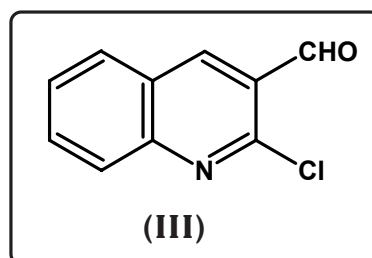
SYNTHETIC ASPECT

Different methods for the synthesis of quinoline derivatives are as follows:

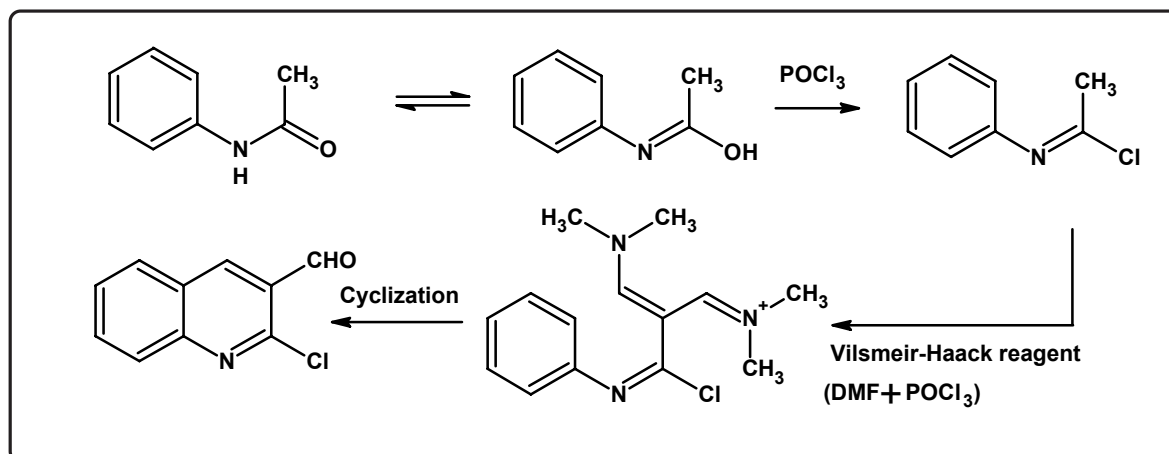
- (1) J. A. Moore et al. have synthesised 6-hydroxyquinoline by the oxidation of hydroxyquinolines by means of ferricyanide³.
- (2) T. Kametani et al.⁴ have synthesised quinoline derivatives by deoxygenation with the help of triphenylphosphite.
- (3) o-Nitrocinnamaldehyde was reduced to quinoline by potassiumtetra carbonylhydrido ferrate was demonstrated by C. K. Bradsher and T. G. Wallis.⁵
- (4) M. Lancaster and D. J. H. Smith⁶ have prepared tetrahydroquinoline derivatives of type (II).



- (5) Meth-Cohn and co-workers⁷ have shown that treatment of acetanilides with the Vilsmeier Haack reagent with POCl_3 as solvent allows the preparation of quinoline derivative of type (III).



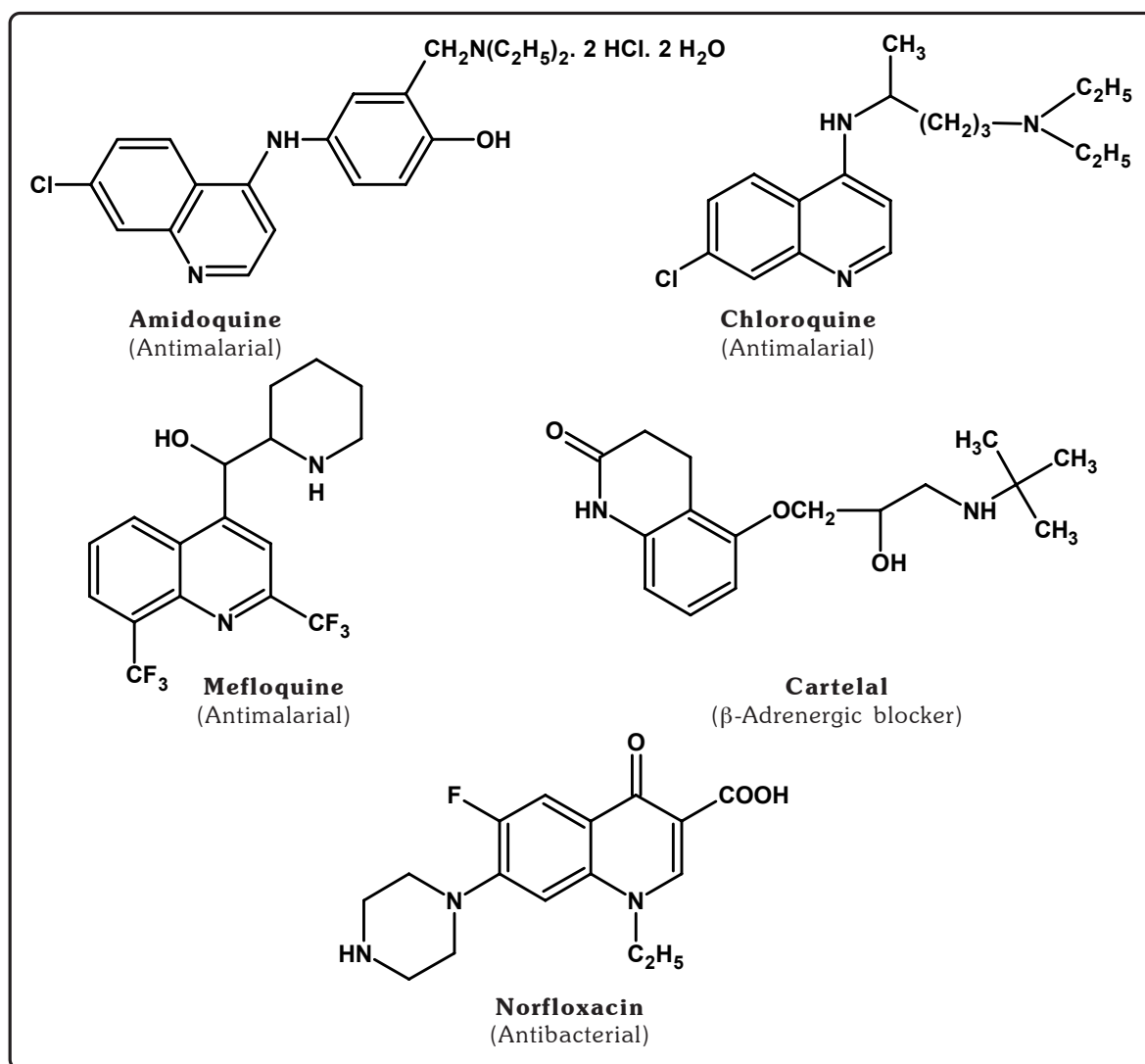
MECHANISM



In the first step hydroxy group is replaced by chlorine, which is possible by lone pair of electron of halogen atom. In the second step Vilsmeier - Haack reagent attacks on nucleophile and forms quaternary ammonium ion as an intermediate. This intermediate on treatment by POCl_3 leads to the quinoline molecule.

THERAPEUTIC IMPORTANCE

Following are some quinoline derivatives reported to possess pharmacological activity :

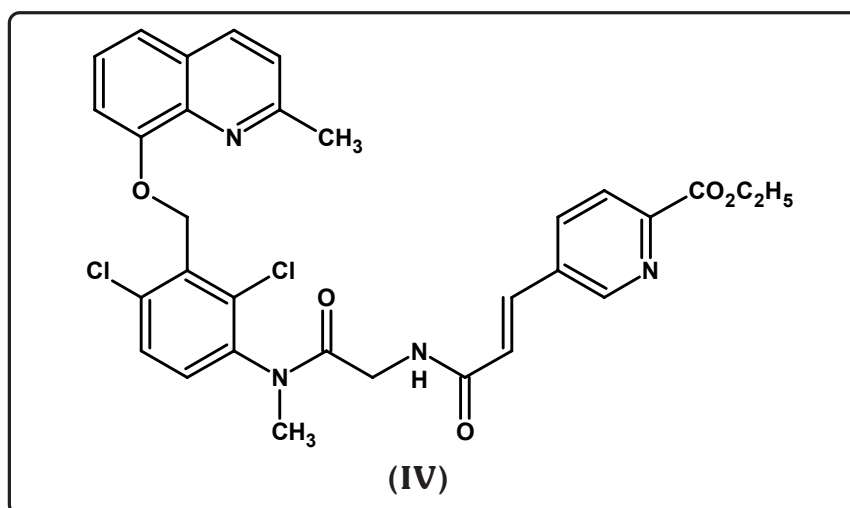


Quinoline and its annelated substrates occupy a unique place in the field of medicinal chemistry due to their wide spectrum of therapeutic activities which can be listed as under.

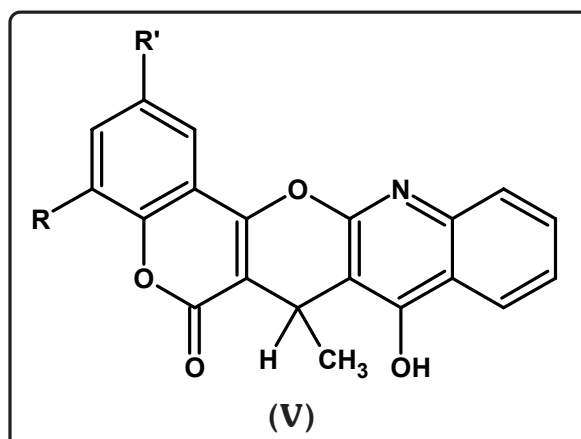
1. Antiinflammatory⁸⁻¹²
2. Antidepressant¹³
3. Antiallergic^{14,15}
4. Antihypertensive¹⁶
5. Antiviral^{17,18}
6. Antitumor^{19,20}
7. Antiulcer²¹
8. Anticonvulsant²²

9. Analgesic^{23,24}
10. Antimicrobial²⁵⁻²⁷
11. Antiarrhythmic²⁸
12. Antitubercular^{29,30}
13. Antithrombotic³¹
14. Antiplatelet³²⁻³⁴
15. Antimalarial³⁵
16. Antithyroid³⁶

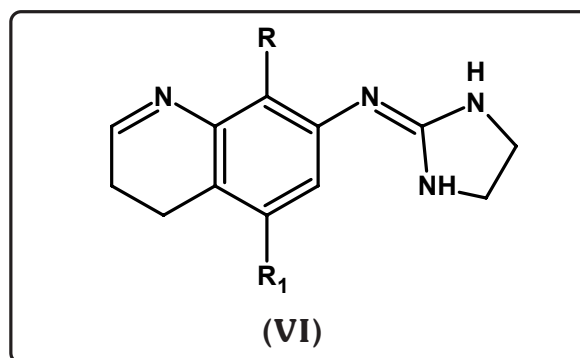
O. Teruo et al.³⁷ have prepared some new quinoline derivatives (IV) and documented them as bradykinin antagonists.



Desai P. K. and co-workers³⁸ reported some new quinoline derivatives as antitubercular agent. Balasubramanian C. and co-workers³⁹ synthesised some novel benzopyranopyranoquinoline derivatives (V) and tested their antibacterial activity.

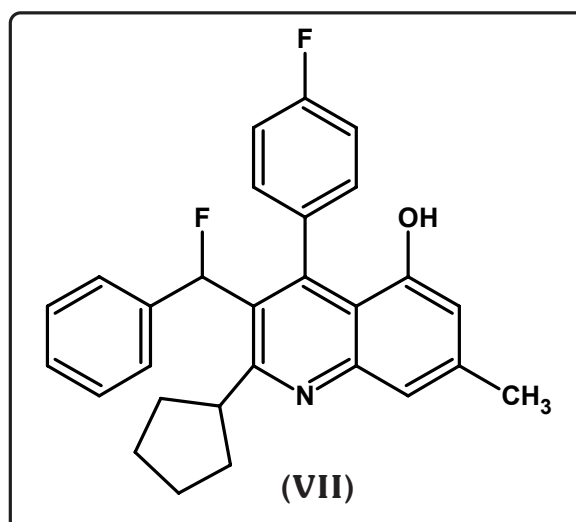


K. Seiichiro et al.⁴⁰ synthesised quinoline sulfide derivatives as selective antibacterial agents. D. H. Joan and co-workers⁴¹ prepared quinoline sulfonamides and described them as tumor necrosis factor (TNF) inhibitors. C. T. Lee et al.⁴² synthesised some new imidazolyl quinolines (VI) and documented them as alpha-2-adrenoceptor agonists.



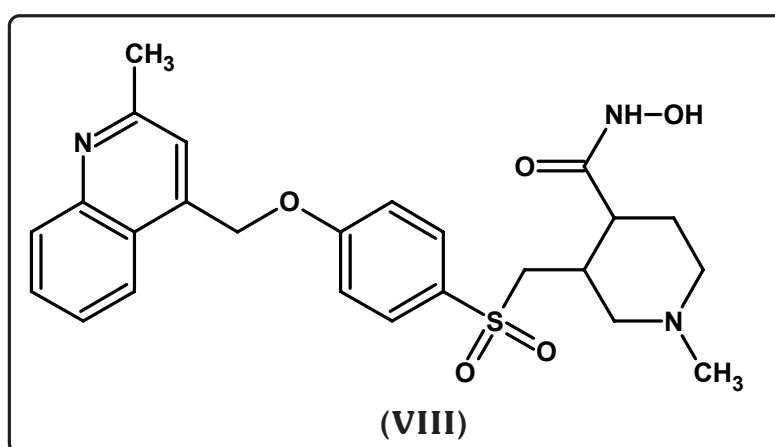
Klein E. S. et al.⁴³ prepared some new quinoline derivatives and reported them as retinoid receptor antagonists. C. S. Hoon and co-workers⁴⁴ reported antitumor activity of some new 2-quinolone derivatives. Multidrug resistance modulating activity of a new series of tetrahydroquinolines was documented by H. Romana and co-workers.⁴⁵

S. Yoshida and co-workers⁴⁶ prepared some new quinolone derivatives and reported their antiulcer activity. 4-Oxoquinoline derivatives are reported as antibacterial agents by Sarvanan J. et al.⁴⁷ Some quinolines of type-(VIII) have been prepared by M. G. Matthias and co-workers⁴⁸ and reported as antiarteriosclerotics.

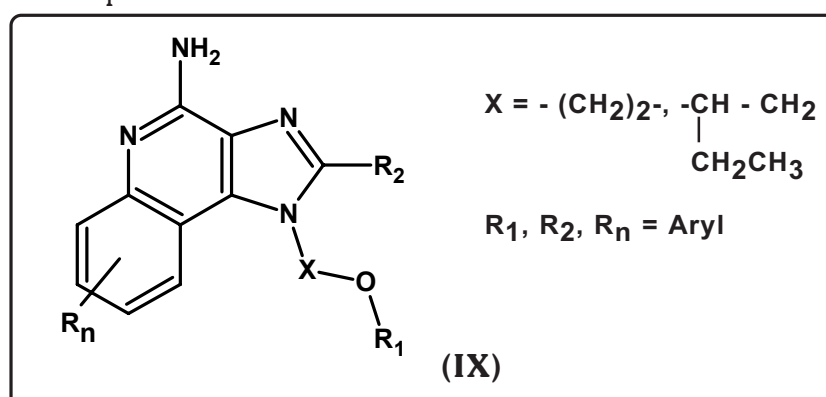


Antimalarial activity of some new 4-aminoquinolines was reported by Raynes K. J. and co-workers.⁴⁹ B. T. Richard et al.⁵⁰ synthesised some new piperazinyl methyl quinolines and documented them as antipsychotics. Anticancer activity of 8-hydroxyquinolines was reported by Shen and co-workers.⁵¹

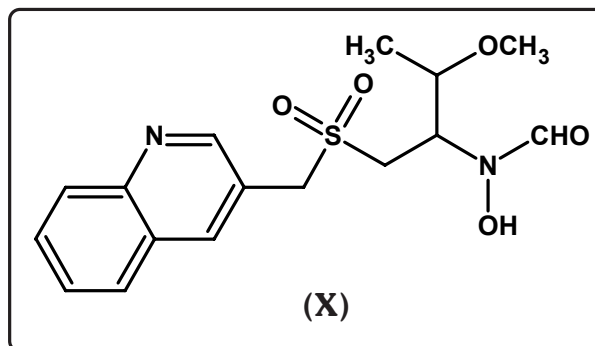
Quinoline derivatives of type (VIII) have been described by Xue C. B. and co-workers⁵² and documented them as TNF inhibitors.



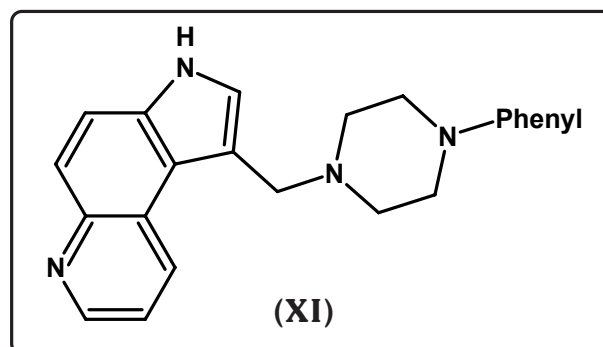
G. Qingchun et al.⁵³ described some new fluoroquinoline derivatives and tested their antibacterial activity. S. Sudahira and co-workers⁵⁴ synthesised some new quinoline carboxylic acid derivatives and reported them as bactericidal. Wackernagel and co-workers⁵⁵ studied some new quinoline derivatives and reported them as insecticides. M. Jigna⁵⁶ synthesised some new arylquinolinyl pyrimidinones and screened for their antibacterial activity. Crooks S. L. et al.⁵⁷ prepared some new amidoether substituted quinolines (IX) as immune response modifiers.



M. Jansen and co-workers⁵⁸ reported some new quinolones and documented them as antagonists. Antibacterial activity of some new quinoline-2-one heterocycles was reported by Suchetak et al.⁵⁹ Use of quinolinylmethylsulfonyl butyl formamides (X) in the treatment of autoimmune disease and allergies were shown by B. D. John and co-workers.⁶⁰



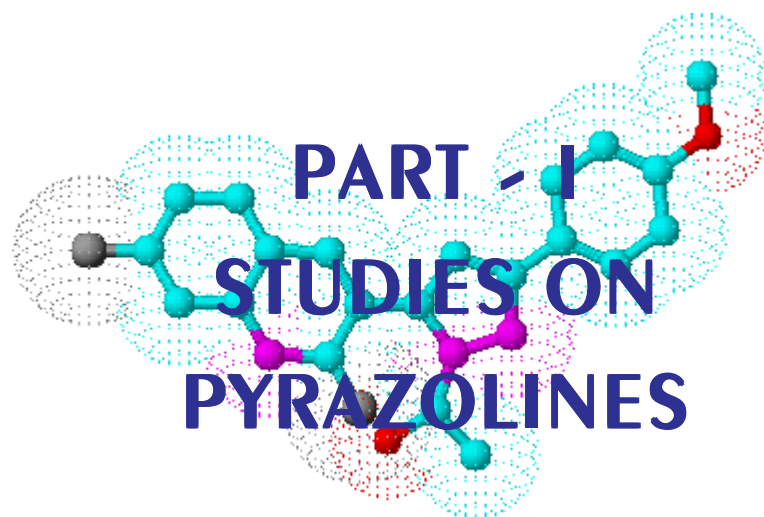
P. Uwe and co-workers⁶¹ have described some new pyrrolidinyl quinolones and reported them as antibacterial agents and food additives. Desouza N. J. et al.⁶² investigated some new quinolone carboxaldehyde derivatives and reported their effect against multidrug resistance bacteria. F. M. Grazia and co-workers⁶³ studied some pyrazoloquinolines (XI) and documented their vasorelaxing effect.



The universal template approach to drug design foresees that quinolines can be modified in such a way to afford better therapeutic agents, we have designed chalcones, pyrazolines, cyanopyridines, cyanopyridones, barbitones, isoxazoles, imidazolines arylidienes, pyrimidinones and nitriles bearing 2,7-dichloroquinoline-3-carboxaldehyde moiety, which is described in the following parts.

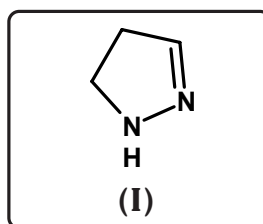
STUDIES ON QUINOLINE DERIVATIVES

- PART-I : STUDIES ON PYRAZOLINES**
- PART-II : STUDIES ON CYANOPYRIDINES**
- PART-III : STUDIES ON CYANOPYRIDONES**
- PART-IV : STUDIES ON BARBITONES**
- PART-V : STUDIES ON ISOXAZOLES**
- PART-VI : STUDIES ON IMIDAZOLINONES**
- PART-VII : STUDIES ON 5-ARYLIDENE-4-THIAZOLIDINONES**
- PART-VIII : STUDIES ON THIAZOLIDINOPYRIMIDINONES**
- PART-IX : STUDIES ON α -ARYLMINONITRILES**



INTRODUCTION

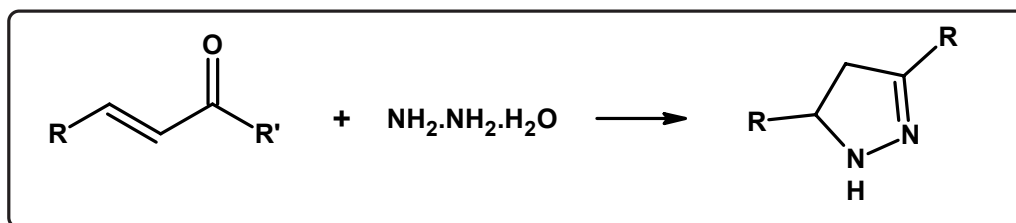
Pyrazolines are known to be among those heterocycles which find their utility in various fields of life. The significance of this moiety has increased due to its biocompatibility. The chemistry of pyrazoline was reviewed by Jarboe in 1967, which have been studied extensively for their biodynamic behaviour⁶⁴, and industrial applications⁶⁵.



SYNTHETIC ASPECT

Different methods are available in literatures for the preparation of 2-pyrazolines. The popular methods are:

1. The most common procedure for the synthesis of 2-pyrazolines is the cyclocondensation of chalcones with hydrazine hydrate⁶⁶.

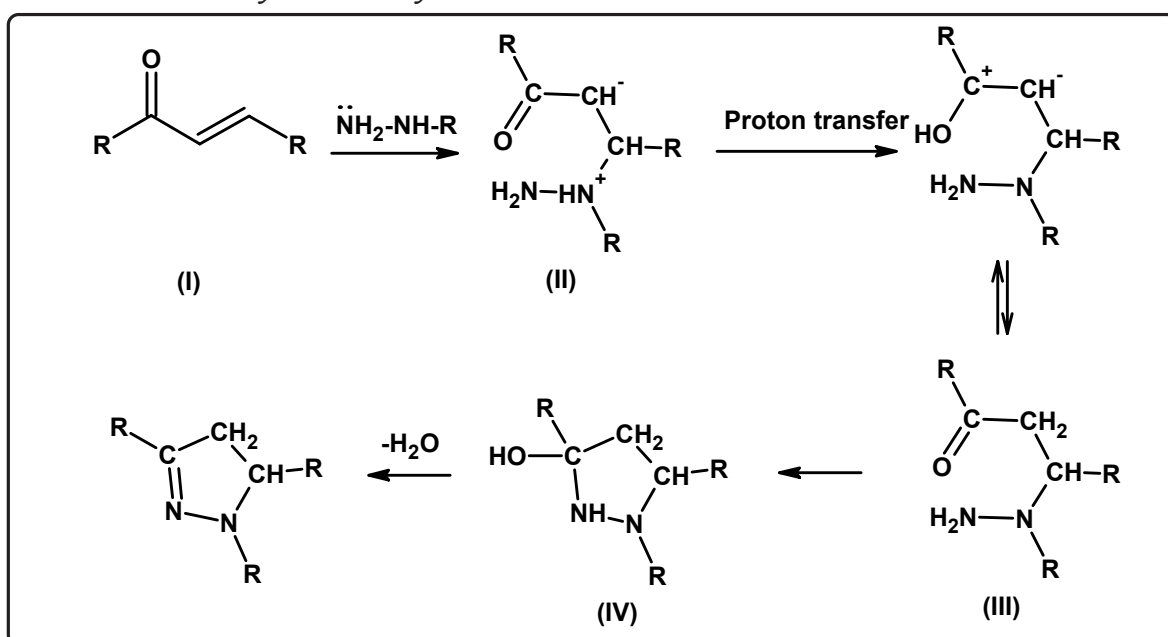


2. 2-Pyrazolines can be also prepared by the condensation of di-bromo derivative of chalcone with hydrazine hydrates⁶⁷.
3. 2-Pyrazolines can be constructed by the cycloaddition of diazomethane to substituted chalcones⁶⁸.

4. A number of diarylidene cycloalkanones on reaction with hydrazine hydrate produce pyrazolines⁶⁹.
5. Epoxidation of chalcones gave epoxy ketones which reacted with hydrazine hydrate and phenyl hydrazine to give pyrazolines⁷⁰.

REACTION MECHANISM

The following mechanism seems to be operable for the condensation of chalcones with hydrazine hydrate⁷¹.



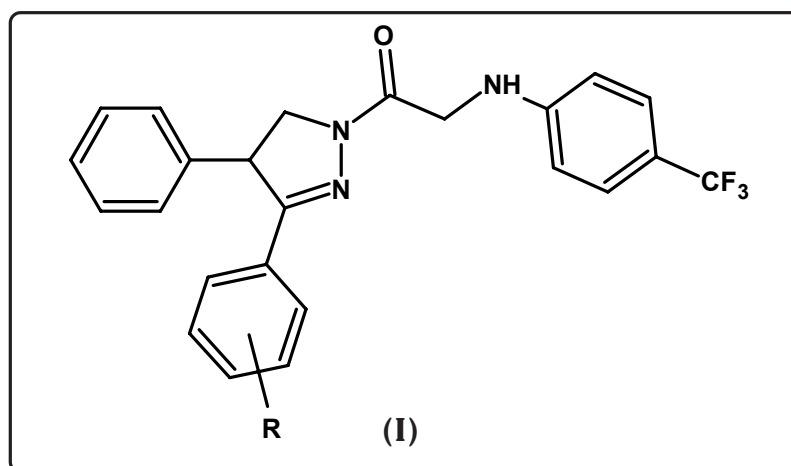
Nucleophilic attack by hydrazine at the β -carbon of the α,β -unsaturated carbonyl system forms species (II), in which the negative charge is mainly accommodated by the electronegative oxygen atom. Proton transfer from the nitrogen to oxygen produces an intermediate end which simultaneously ketonises to ketoamine (III). Another intramolecular nucleophilic attack by the primary amino group of ketoamine on its carbonyl carbon followed by proton transfer from nitrogen to oxygen leads ultimately to hydroxyl amine (IV). The later with a hydroxy group and amine group on the same carbon lose water easily to yield the pyrazolines.

THERAPEUTIC IMPORTANCE

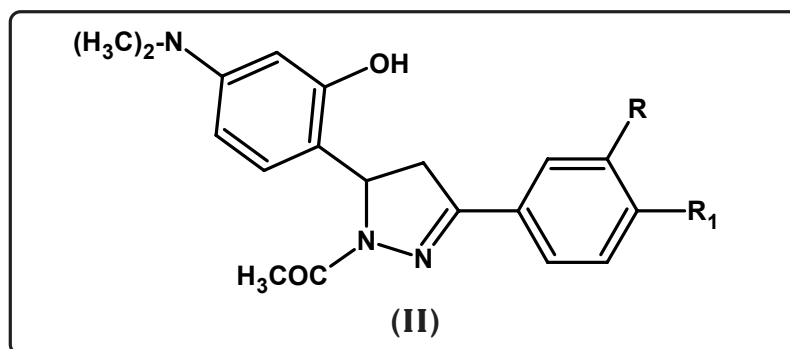
From the literature survey, it was revealed that 2-pyrazoline derivatives are better therapeutic agents.

1. Insecticidal⁷²
2. Diuretic⁷³
3. Fungicidal⁷⁴
4. Herbicidal⁷⁵
5. Hypoglycemic⁷⁶
6. Antiallergic⁷⁷
7. Anticonvulsant⁷⁸
8. Antidiabetic⁷⁹
9. Antiinflammatory⁸⁰
10. Antitumor⁸¹
11. Cardiovascular⁸²
12. Analgesic⁸³
13. Antineoplastic⁸⁴
14. Bactericidal⁸⁵
15. Tranquilizing⁸⁶

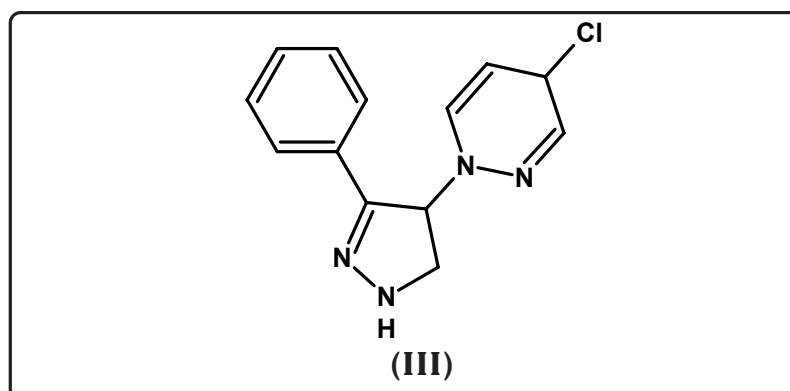
T. S. Katshaki et al.⁸⁷ have synthesised some new (phenylcarbonyl) pyrazoline (I) as an insecticide which at 40% concentration shows 100% mortality of *spodopeira litura* larvae after seven days.



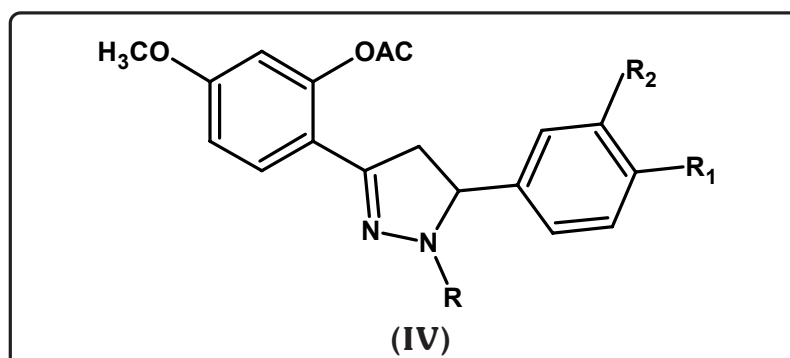
F. Manna and co-workers⁸⁸ have described 1-acetyl-5-(2'-bromophenyl)-4,5-dihydro-3-(2'-hydroxyphenyl)-1H-pyrazolines (II) and its derivatives which acts as potent antiinflammatory, analgesic and antipyretic agents.



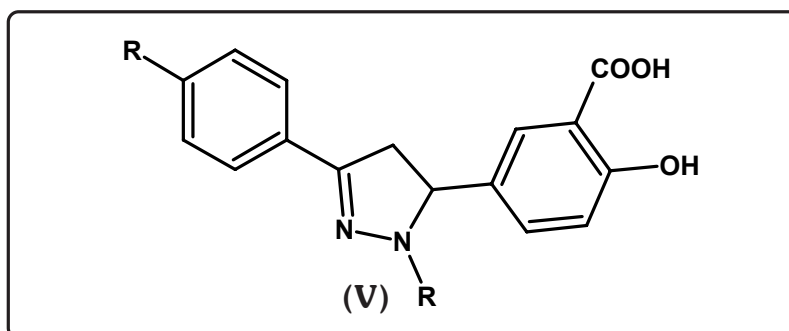
Fuche Rainer et al.⁸⁹ have prepared some new 1H-pyrazoline derivatives (III) and reported them as pesticides.



Moreover Ismil et al.⁹⁰ have prepared pyrazoline derivatives bearing sulfonamide moiety and tested their antimicrobial activity. Sarhan and co-workers⁹¹ have synthesised pyrazolines derivatives and reported their antibacterial, analgesic and antiinflammatory activities. S S. Sonarc et al.⁹² have studied 3-(2-acetoxy-4-methoxyphenyl)-5-(substituted phenyl)-pyrazolines (IV) for their antimicrobial activity.



G. N. Mishirika et al.⁹³ have also prepared 2-pyrazolines derivatives of salicylic acid (V) possessing antimicrobial properties.

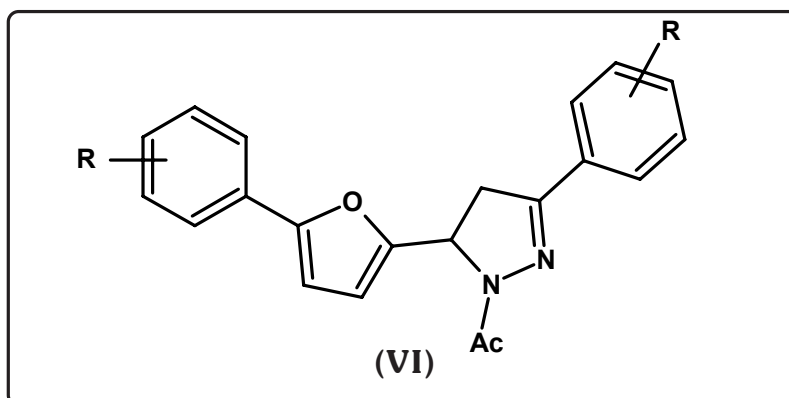


Further more, T.M. Stavenson et al.⁹⁴ have also investigated N-substituted pyrazoline type insecticides. Tanka Katsohori⁹⁵ have patented pyrazoline derivatives as herbicides.

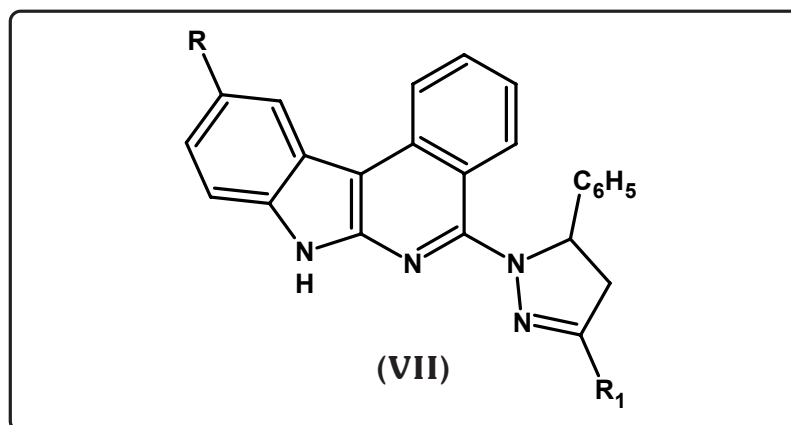
CONTRIBUTION FROM OUR LABORATORY

Jatin Upadhyay et al.⁹⁶ have prepared 1-acetyl-4,5-dihydro-5-(4-hydroxy-3-methoxyphenyl)-3-(4-phenyl-sulfonylaminophenyl)-1H-pyrazole and other derivatives which possess antimicrobial activity. Synthesis of some novel pyrazolines as biologically potent agents bearing quinazolone nucleus has been reported by Oza H. et al.⁹⁷ Parikh et al.⁹⁸ have synthesised m,m'-bis(5-aryl-1-phenyl)-pyrazolin-3-yl-p-phenyl-amino sulfonyl benzophenones and screened for their antimicrobial activity. Antimicrobial activity of pyrazolines have been documented by H. H. Parekh and co-workers.⁹⁹

Shivananda M.K. and co-workers¹⁰⁰ have prepared pyrazolines and reported their antibacterial activity. B. Shivarama Holla et al.¹⁰¹ have synthesised pyrazolines (VI) and reported their antibacterial activity.



E. Palaska et al.¹⁰² have prepared 3,5-diphenyl-2-pyrazoline derivatives and studied their antidepressant activity. S.P. Hiremath and co-workers¹⁰³ have reported analgesic, antiinflammatory and antimicrobial activity of 3,5-disubstituted pyrazolines (VII).



Thus interesting biological activities of a novel heterocycles like pyrazolines have stimulated considerable research work in recent years leading to the synthetic utility of the derivatives of this ring system. In our search for new potential antimicrobial compounds, the reaction series of chalcones with hydrazine hydrate under different conditions has been investigated and the pharmacological profile of the compounds have been studied and described as under.

SECTION I : SYNTHESIS AND THERAPEUTIC EVALUATION OF 1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENE-1-ONES

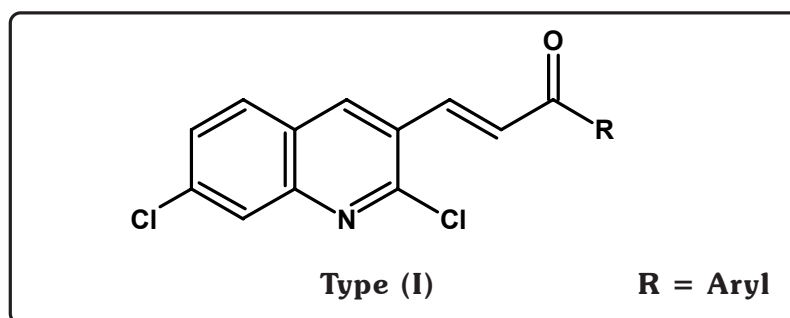
SECTION II : SYNTHESIS AND THERAPEUTIC EVALUATION OF 1-ACETYL-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES

SECTION III : SYNTHESIS AND THERAPEUTIC EVALUATION OF 1H-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL) PYRAZOLINES

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENE-1-ONES

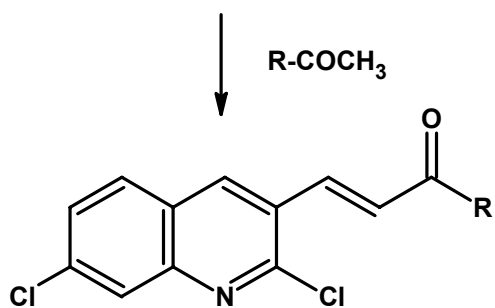
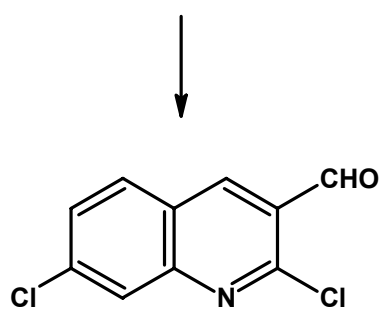
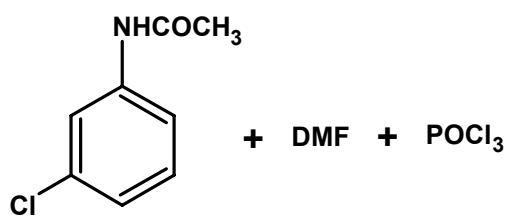
During the past years, considerable evidence has been accumulated to demonstrate the efficiency of chalcones in including variety of therapeutic activities. To further assess the potential of such a class of compounds as antibacterial and antifungal agents, a series of chalcones of type (I) have been synthesised by the condensation of 2,7-dichloroquinoline-3-carboxaldehyde with various aromatic ketones.



The constitution of the synthesised products has been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass-spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I Section-I (D). The antimycobacterial activity was carried out towards *Mycobacterium tuberculosis* at 6.25 μg concentration using Rifampin as a standard drug.

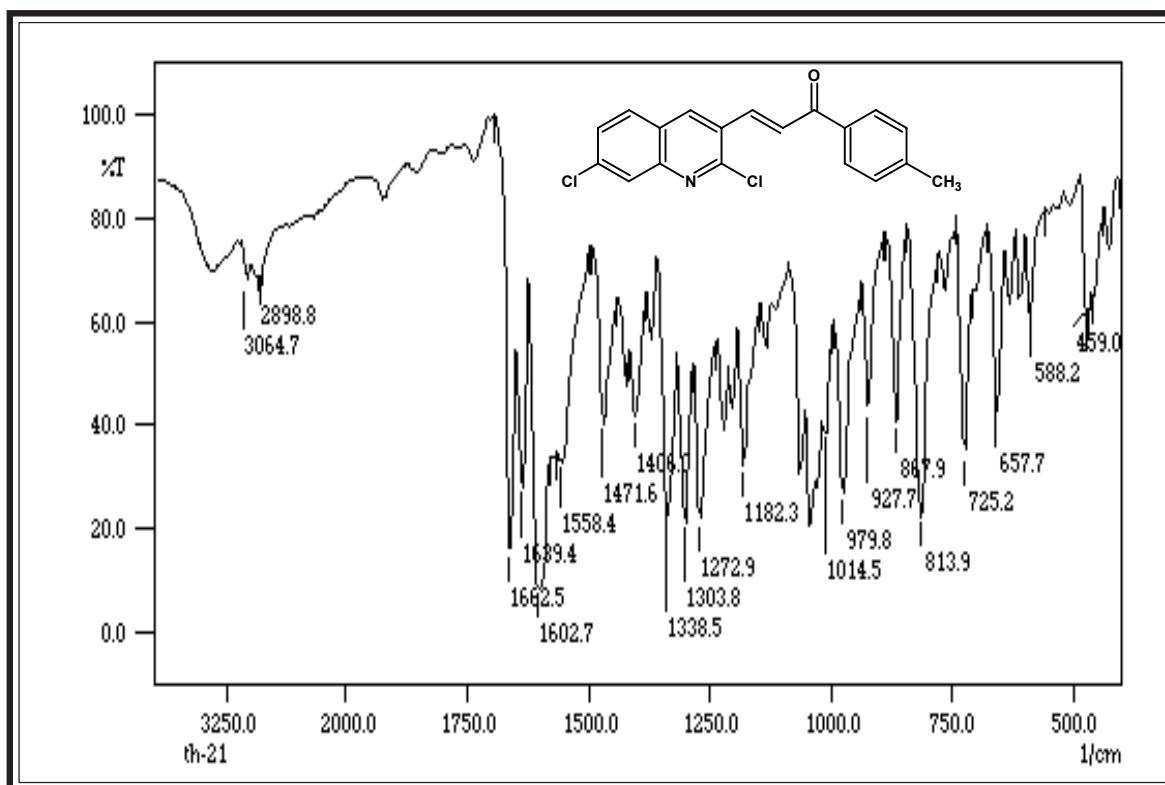
REACTION



Type (I)

R = Aryl

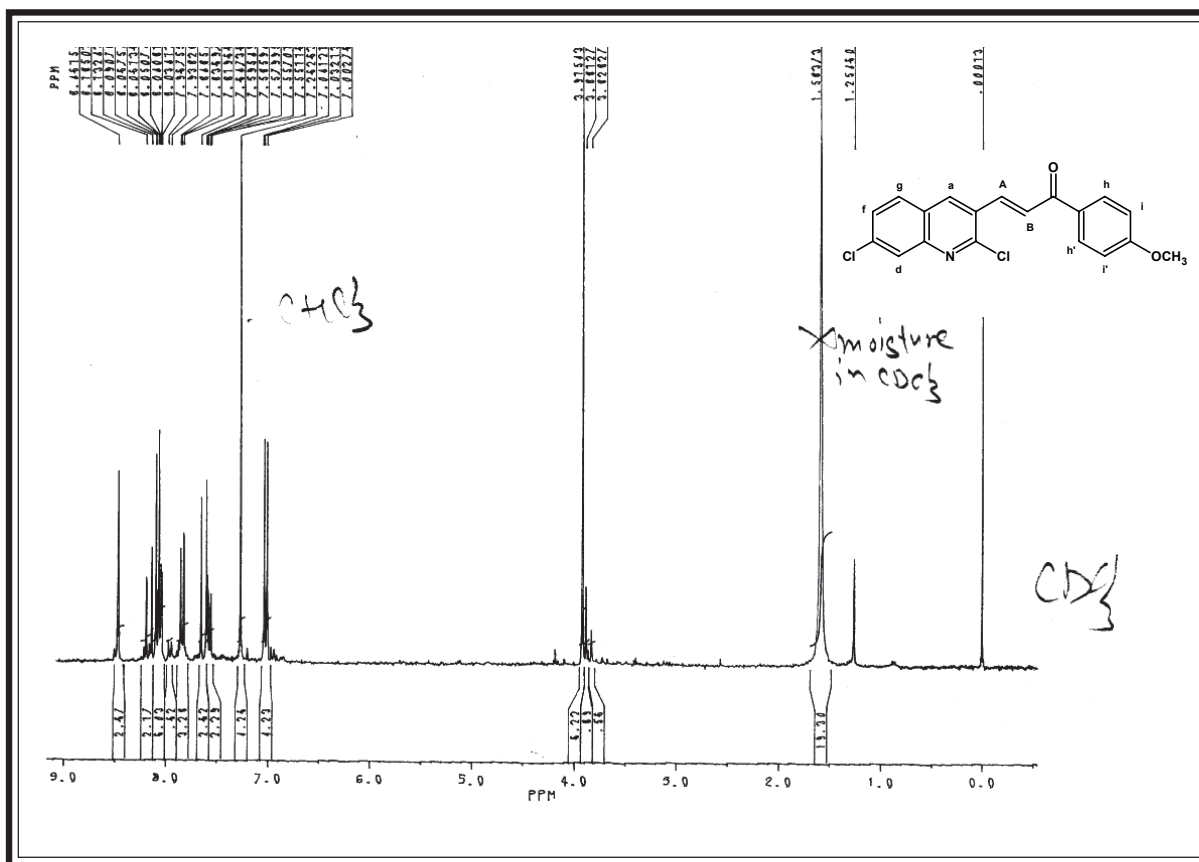
IR SPECTRAL STUDY OF 1-p-TOLYL-3-(2'-7'-DICHLORO-QUINOLIN-3'-YL)-2-PROPENE-1-ONE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane	C - H str.	2998	2975-2850	419
	C - H def.	1471	1470-1435	"
Aromatic	C- H str.	3064	3090-3030	"
	C = C str.	1471	1520-1480	"
	C - H def.	1014	1070-1000	420
		813	835-810	"
Quinoline moiety	C = N str.	1602	1610-1590	419
	C = C str.	1602	1610-1590	"
	C - Cl str.	725.2	750-700	"
Chalcone	C = O str.	1662	1690-1665	"
Vinyl	CH = CH	3064	3050-3000	"

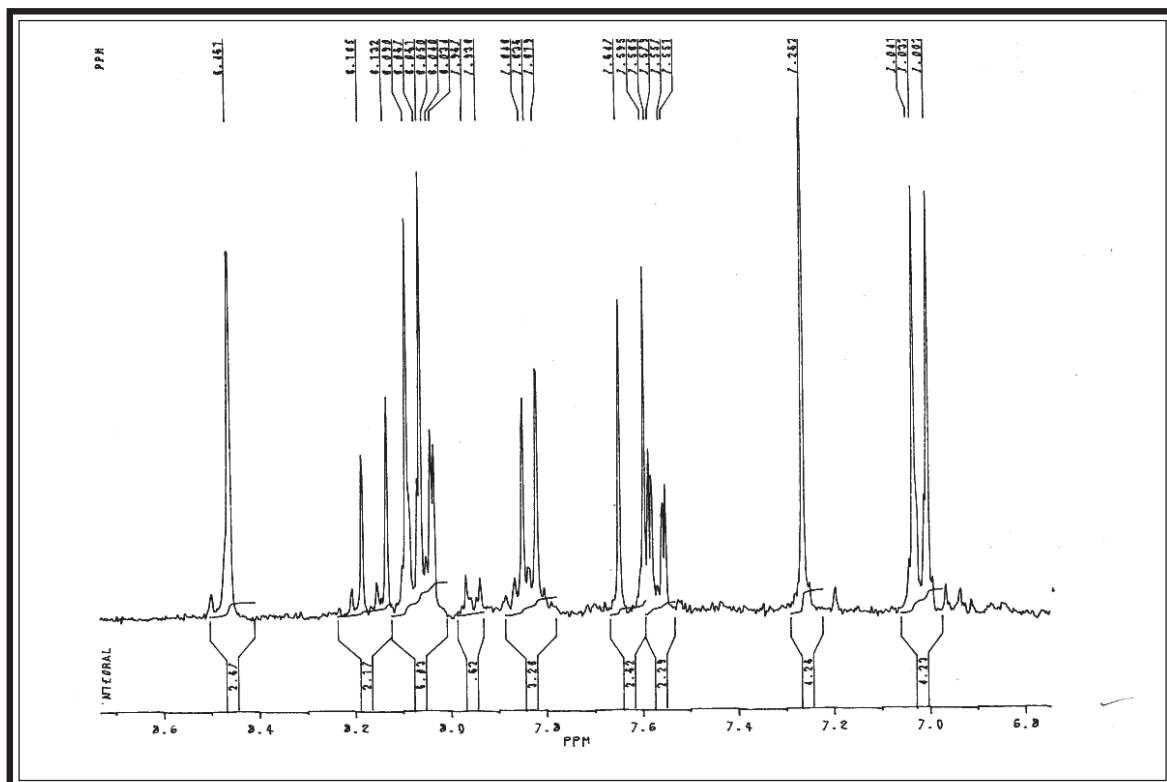
PMR SPECTRAL STUDY OF 1-(p-ANISYL)-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENE-1-ONES



Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1	3.97	3H	singlet	-	Ar-OCH ₃
2.	7.00-7.03	2H	doublet	Jih 8.81	Ar-Hi,i'
3.	7.55-7.59	1H	double doublet	Jfg 8.67 Jfd 1.8	Ar-Hf
4.	7.60-7.65	1H	doublet	JBA 15.56	=CH _B
5.	7.82-7.85	1H	doublet	Jgf 8.64	Ar-Hg
6.	8.03-8.04	1H	doublet	Jdf 2.00	Ar-Hd
7.	8.06-8.09	2H	doublet	Jhi 8.83	Ar-Hh,h'
8.	8.13-8.19	1H	doublet	JAB 15.86	=CH _A
9.	8.46	1H	singlet	-	Ar-Ha

EXPANDED AROMATIC REGION

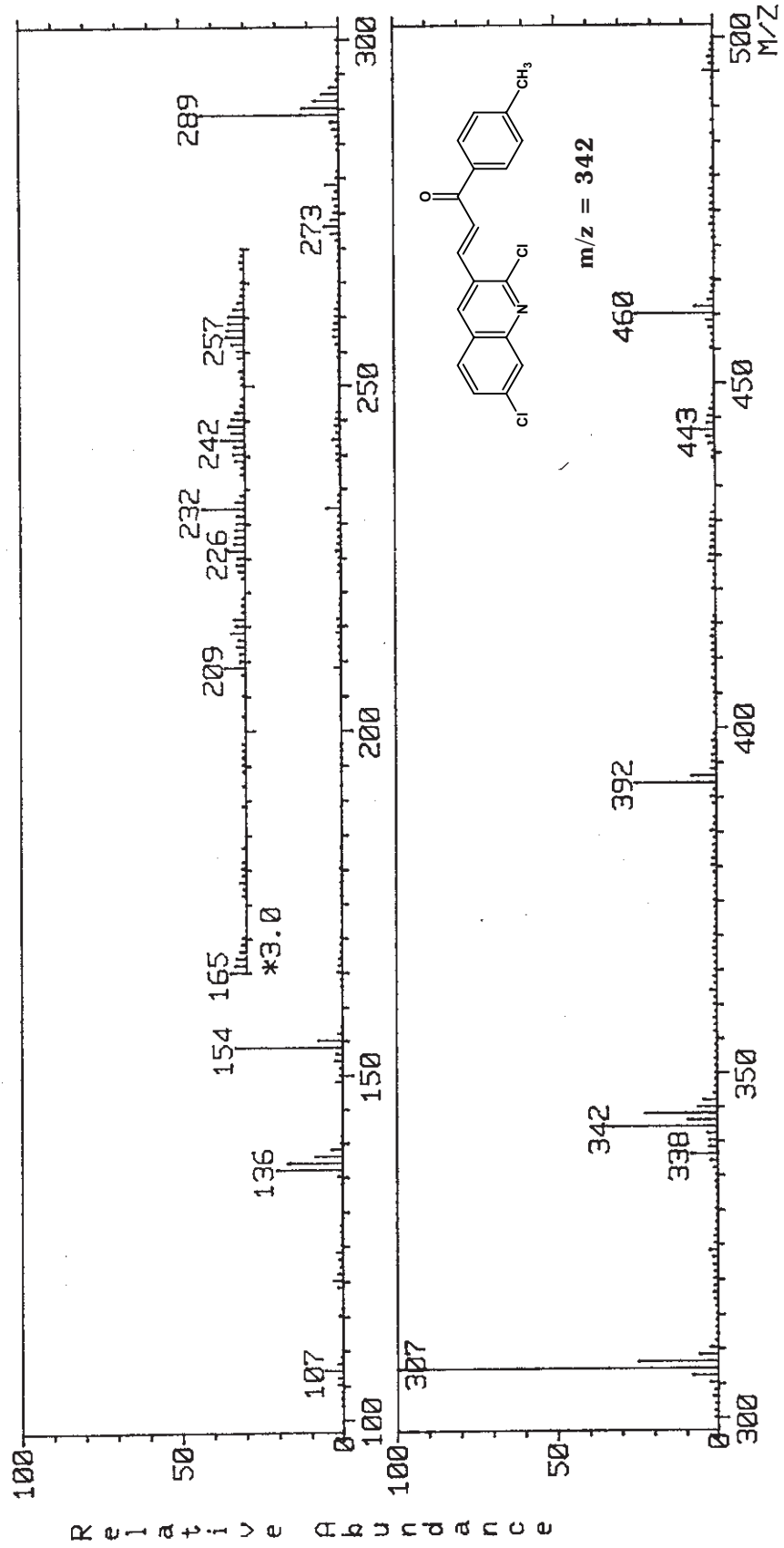


IR SPECTRAL DATA OF 1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENE-1-ONES

Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	C=O str. Chalcone
1a	C_6H_5	1658
1b	3- NH_2 - C_6H_4	1660
1c	4- NH_2 - C_6H_4	1662
1d	4-Br- C_6H_4	1658
1e	4-Cl- C_6H_4	1656
1f	2,6-(OH) $_2$ - C_6H_3	1660
1g	4-F- C_6H_4	1662
1h	$\text{C}_4\text{H}_3\text{O}$	1658
1i	2-OH- C_6H_4	1632
1j	4-OH- C_6H_4	1660
1k	4-OCH $_3$ - C_6H_4	1658
1l	4-CH $_3$ - C_6H_4	1662
1m	3-NO $_2$ - C_6H_4	1660
1n	4-NO $_2$ - C_6H_4	1660

MASS SPECTRUM Data File: 3EJL25M
 Sample: APS-7 DR NA CHAUHAN, RAJKOT #6213
 RT 0.12" FAB(Pos.) GC 1.4c BP: m/z 307.0000 Int. 36.2350 Lv 0.00
 Scan# (2 to 3)



ANTIMICROBIAL ACTIVITY

Method	:	Cup Plate
Gram positive bacteria	:	<i>B. megaterium.</i>, <i>S.aureus</i>
Gram negative bacteria	:	<i>E. coli.</i>, <i>P. vulgaris</i>
Fungi	:	<i>Aspergillus niger</i>
Concentration	:	40 µg.
Solvent	:	Dimethyl formamide
Standard drug	:	ampicillin, amoxicillin, ciprofloxacin, norfloxacin

The antimicrobial activity was compared with standard drugs viz. ampicillin, amoxicillin, ciprofloxacin, erythromycin and antifungal activity was compared with viz. greseofulvin zones of inhibition have been measured in mm.

Antitubercular activity

The antitubercular evaluation of the compounds was carried out at Tuberculous Antimicrobial Acquisition and Co-ordinating Facility (TAACF) U.S.A.

Method	:	BACTEC 460 Radiometric System
Bacteria	:	<i>Mycobacterium tuberculosis H₃₇ Rv</i>
Concentration	:	6.25 µg/ml
Standard drug	:	Rifampin.

EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENE 1-ONES

[A] Synthesis of 3-Chloroacetanilide

m-Chloroaniline (12.7 ml) was cooled to 15° to 20°C in 50 ml water. Acetic anhydride (12.7 ml) was added with stirring and the reaction mixture was heated under reflux for 2 and half hrs. The reaction mixture was poured into ice-water. The product was isolated and crystallised from methanol. yield 95%; m.p. 78°C.

[B] Synthesis of 2,7-Dichloroquinolin-3-carboxaldehyde.

Dimethyl formamide (9.13 g, 9.6 ml, 0.125 mol) was cooled to 0°C in a flask equipped with drying tube and phosphorous oxychloride (53.7 g, 32.2 ml, 0.35 mol) was added dropwise with stirring. To this solution was added m-chloroacetanilide (8.5 g, 0.05 mol) and after 5 min. the solution was heated under refluxed for 6 and half hrs. The reaction mixture was poured into ice-water and stirred for 30 min. at 0-10°C. The product was isolated and crystallised from ethyl acetate yield 54%; m.p. 120°C. Anal calcd. for C₁₀H₅NCl₂O required : C, 53.13%; H, 2.23%; N, 6.20%; found: C, 5.9%; H, 1.85%; N, 6.30%

[C] Synthesis of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones

A solution of 4-tolyl acetophenone (2.64 ml, 0.02 mol) in minimum quantity of ethanol (5 ml) was added to a mixture of 2,7-dichloro quinoline-3-carboxaldehyde (4.52 g, 0.02 mol) in ethanol (20 ml) and 40% NaOH was added to make it slightly alkaline. The reaction mixture was then stirred for 24 hrs. at room temperature. The product was isolated and crystallised from ethanol. yield 63%, m.p. 160°C Anal. calcd. for C₁₉H₁₃NCl₂O required; C, 66.68%; H, 3.83%; N, 4.10% found C, 66.34%; H, 3.68%; N, 4.09%

TLC solvent system: Acetone: Benzene (2.5:7.5) visualizing agent : Iodine

Similarly other 1-Aryl-3-(2'-7'-dichloroquinolin-3'-yl)-2-propene-1-ones were prepared. The physical constants along with infra red spectral data are recorded in Table No.1a

[D] Therapeutic evaluation of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones

All the products have been evaluated for their antimicrobial activity as described under.

(a) Antimicrobial activity¹⁰⁴

It was carried out using cup-plate agar diffusion method which has been described as under.

(i) Antibacterial activity

The purified products were screened for their antibacterial activity. The nutrient agar broth prepared by usual method was inoculated aseptically with 0.5 ml of 24 hrs. old subculture of *B. megaterium*, *S. aureus*, *E. coli* and *P. vulgaris* in separate conical flask at 40-50°C and mixed well by gentle shaking. About 25 ml of the content of the flask were poured and evenly spreaded in a petridish (13 cm in diameter) and allowed to set for two hrs. The cups (10 mm in diameter) were formed by the help of borer in agar medium and filled with 0.04 ml (40 µg) solution of sample in DMF.

The plates were incubated at 37°C for 24 hrs and the control was also maintained with 0.04 ml of DMF in similar manner and the zones of inhibition of the bacterial growth were measured in millimeters and are recorded in Graphical Chart No.1

(ii) Antifungal activity

A. niger was employed for testing antifungal activity using cup-plate method. The culture was maintained on Sabouraud's agar slants. Sterilised Sabouraud's agar medium was inoculated with 72 hrs old 0.5 ml suspension of fungal spores in a separate flask. About 25 ml of the inoculated medium was evenly spreaded in a petridish and allowed to set for two hrs. The cups (10 mm in diameter) were punched in petridish and loaded with 0.04 ml (40 µg) of solution of sample in DMF. The plates were incubated at 30°C for 48 hrs. After the completion of incubation period, the zones of inhibition of growth in the form of diameter in mm was measured. Along the test solution in each petridish one cup was filled up with solvent which act as control. The zones of inhibition are recorded in Graphical Chart No. I.

(iii) Antitubercular activity

The antitubercular evaluation of the compounds was carried out at Tuberculosis Antimicrobial Acquisition and Coordination Facility (TAACF), U.S.A. Primary screening of the compounds for antitubercular activity have been conducted at 6.25 mg/ml towards *Mycobacterium Tuberculosis* H₃₇ Rv in BACTEC 12B medium using the BACTEC 460 radiometric system. The compounds demonstrating atleast > 90% inhibition in the primary screen have been retested at lower concentration towards *Mycobacterium Tuberculosis* H₃₇Rv to determine the actual minimum inhibitory concentration (MIC) in the BACTEC 460.

The antitubercular activity data have been compared with standard drug Rifampin at 6.25 µg/ml concentration and it showed 98% inhibition the data for % inhibition are recorded in Table No. 1b.

TABLE NO. 1a : PHYSICAL CONSTANTS OF 1-ARYL-3-(2',7'-ARYL-3-(2',7'-DICHOROQUINOLIN-3'-YL)-2-PROPENE-1-ONES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
1a	C ₆ H ₅ -	C ₁₈ H ₁₁ NCl ₂ O	328	160	0.589	66	4.23	4.27
1b	3-NH ₂ -C ₆ H ₄ -	C ₁₈ H ₁₂ N ₂ Cl ₂ O	343	196	0.604	72	8.15	8.16
1c	4-NH ₂ -C ₆ H ₄ -	C ₁₈ H ₁₂ N ₂ Cl ₂ O	343	146	0.595	61	8.19	8.16
1d	4-Br-C ₆ H ₄ -	C ₁₈ H ₁₀ NBrCl ₂ O	407	160	0.499	68	3.47	3.44
1e	4-Cl-C ₆ H ₄ -	C ₁₈ H ₁₀ NCl ₃ O	361.5	120	0.569	56	3.85	3.86
1f	2,6-(OH) ₂ -C ₆ H ₃ -	C ₁₈ H ₁₁ NCl ₂ O ₃	360	144	0.575	49	3.92	3.89
1g	4-F-C ₆ H ₄ -	C ₁₈ H ₁₀ NCl ₂ FO	346	148	0.583	62	4.01	4.05
1h	C ₄ H ₃ O-	C ₁₆ H ₉ NCl ₂ O ₂	318	186	0.541	65	4.42	4.40
1i	2-OH-C ₆ H ₄ -	C ₁₈ H ₁₁ NCl ₂ O ₂	344	158	0.589	56	4.08	4.07
1j	4-OH-C ₆ H ₄ -	C ₁₈ H ₁₁ NCl ₂ O ₂	344	142	0.607	60	4.04	4.07
1k	4-OCH ₃ -C ₆ H ₄ -	C ₁₉ H ₁₃ NCl ₂ O ₂	358	270	0.501	75	3.88	3.91
1l	4-CH ₃ -C ₆ H ₄ -	C ₁₉ H ₁₃ NCl ₂ O	342	160	0.484	63	4.10	4.09
1m	3-NO ₂ -C ₆ H ₄ -	C ₁₈ H ₁₀ N ₂ Cl ₂ O ₃	373	140	0.551	50	7.55	7.51
1n	4-NO ₂ -C ₆ H ₄ -	C ₁₈ H ₁₀ N ₂ Cl ₂ O ₃	373	210	0.493	58	7.49	7.51

TLC Solvent System : Acetone : Benzene (2.5 : 7.5).

GRAPHICAL CHART NO. 1 : 1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENE-1-ONES

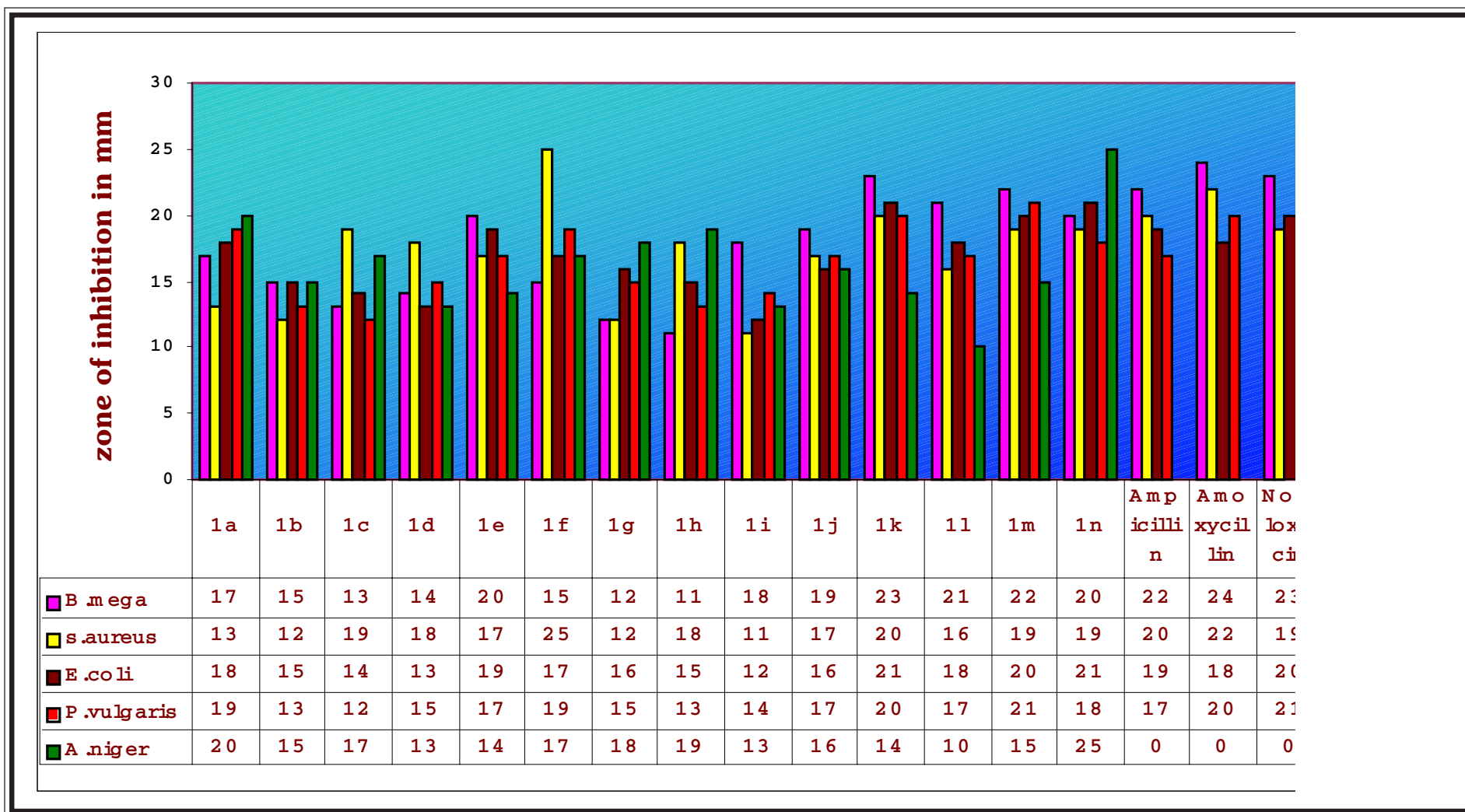
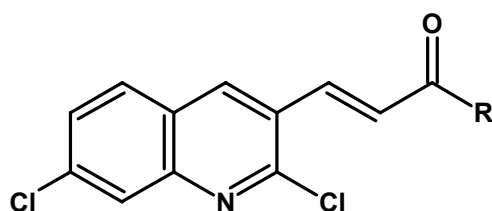


TABLE NO. 1b : PRIMARY ASSAY OF ANTITUBERCULAR ACTIVITY



TAACF, Southern Research Institute
Primary Assay Summary Report

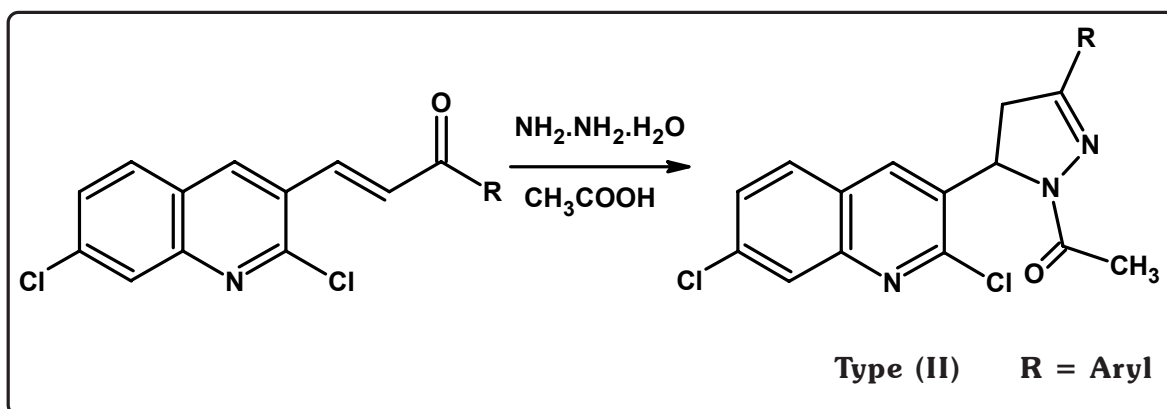
Dr. A. R. Parikh
Saurashtra University

Sample ID	Corp ID	Where, R =	Supplier	Assay	MTb Strain	MIC	% Inhib	Comment
162062	AP-10	4-NH ₂ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	97	MIC Rifampin=0.25µg/ml @98% inhibition
162058	AP-6	4-OH-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	69	MIC Rifampin=0.25µg/ml @98% inhibition
162056	AP-4	4-Cl-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	64	MIC Rifampin=0.25µg/ml @98% inhibition
162060	AP-8	4-F-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	64	MIC Rifampin=0.25µg/ml @98% inhibition
162053	AP-1	C ₆ H ₅	Saurashtra University	Alamar	H37Rv	>6.25	59	MIC Rifampin=0.25µg/ml @98% inhibition
162055	AP-3	4-OCH ₃ -C ₆ H ₅	Saurashtra University	Alamar	H37Rv	>6.25	53	MIC Rifampin=0.25µg/ml @98% inhibition
162057	AP-5	4-Br-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	49	MIC Rifampin=0.25µg/ml @98% inhibition
162054	AP-2	4-CH ₃ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	46	MIC Rifampin=0.25µg/ml @98% inhibition
162061	AP-9	C ₄ H ₃ O	Saurashtra University	Alamar	H37Rv	>6.25	31	MIC Rifampin=0.25µg/ml @98% inhibition
162059	AP-7	4-NO ₂ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	23	MIC Rifampin=0.25µg/ml @98% inhibition

SECTION - II

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1-ACETYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-3-ARYL-PYRAZOLINES

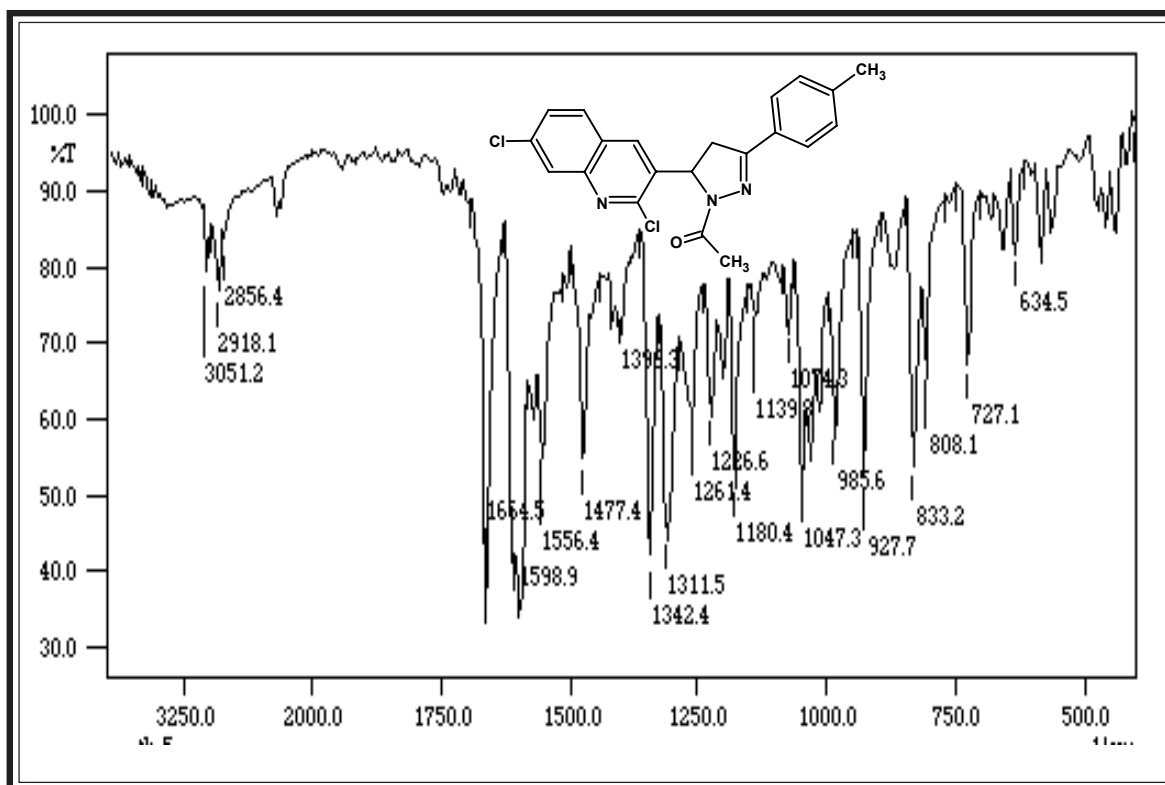
1-Acetyl pyrazolines have been found to be associated with diverse biological activities. These finding prompted us to prepare some new pyrazoline derivatives of type (II) possessing better biological activity, which have been prepared by the condensation of chalcones of type (I) with hydrazine hydrate in acetic acid.



The constitution of the synthesised product has been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section- I (D), antitubercular activity was carried out against *Mycobacterium Tuberculosis* H₃₇Rv at a concentration of 6.25 μg using Rifampin as a standard drug.

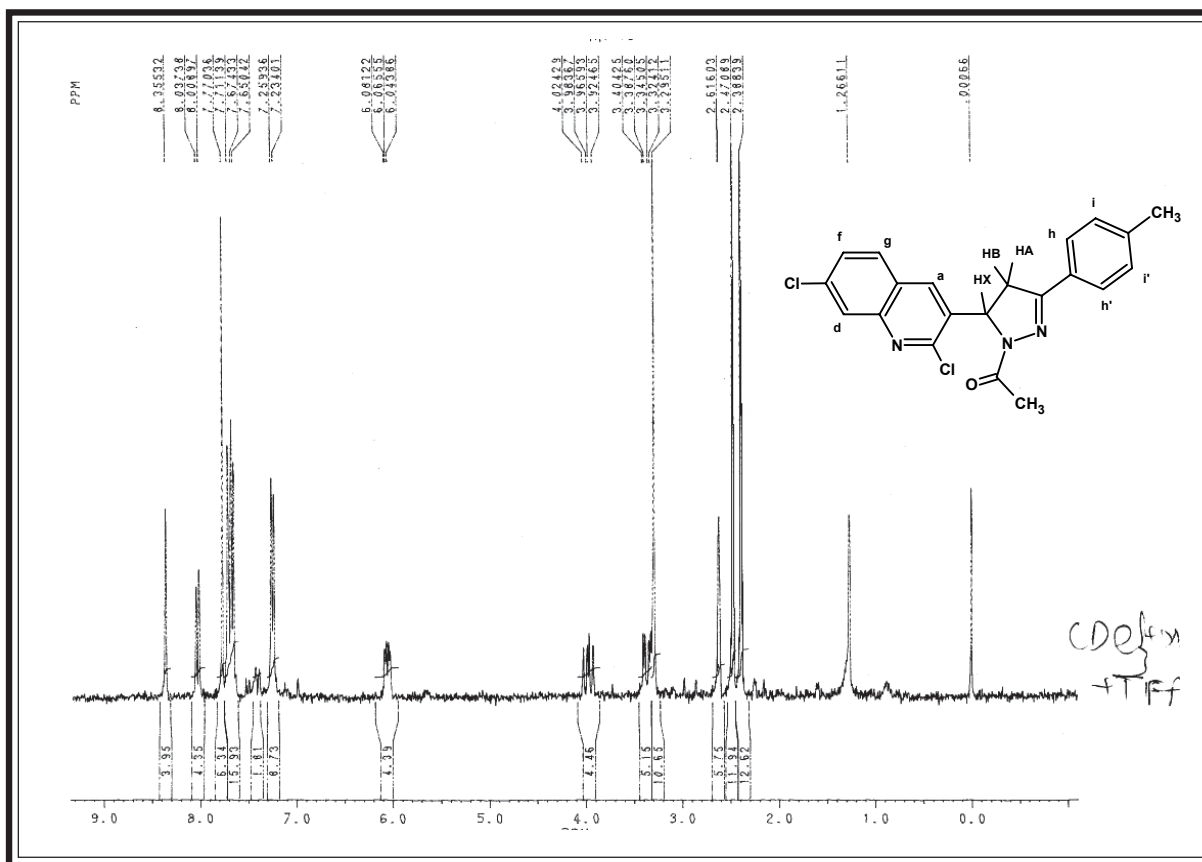
IR SPECTRAL STUDY OF 1-ACETYL-3-*p*-TOLYL-5-(2'-7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane	C – H str.	2918	2975-2850	419
	C – H def.	1477	1470-1435	"
Aromatic	C– H str.	3051	3080-3030	"
	C = C str.	1596	1585-1570	"
		1477	1520-1480	"
	C – H def.	833	835-810	420
Quinoline moiety	C = N str.	1598	1612-1593	419
	C – Cl str.	727	750-700	"
Pyrazoline	C = N str.	1598	1627-1580	418
	N–C–CH ₃ - str. O	1664	1660	"

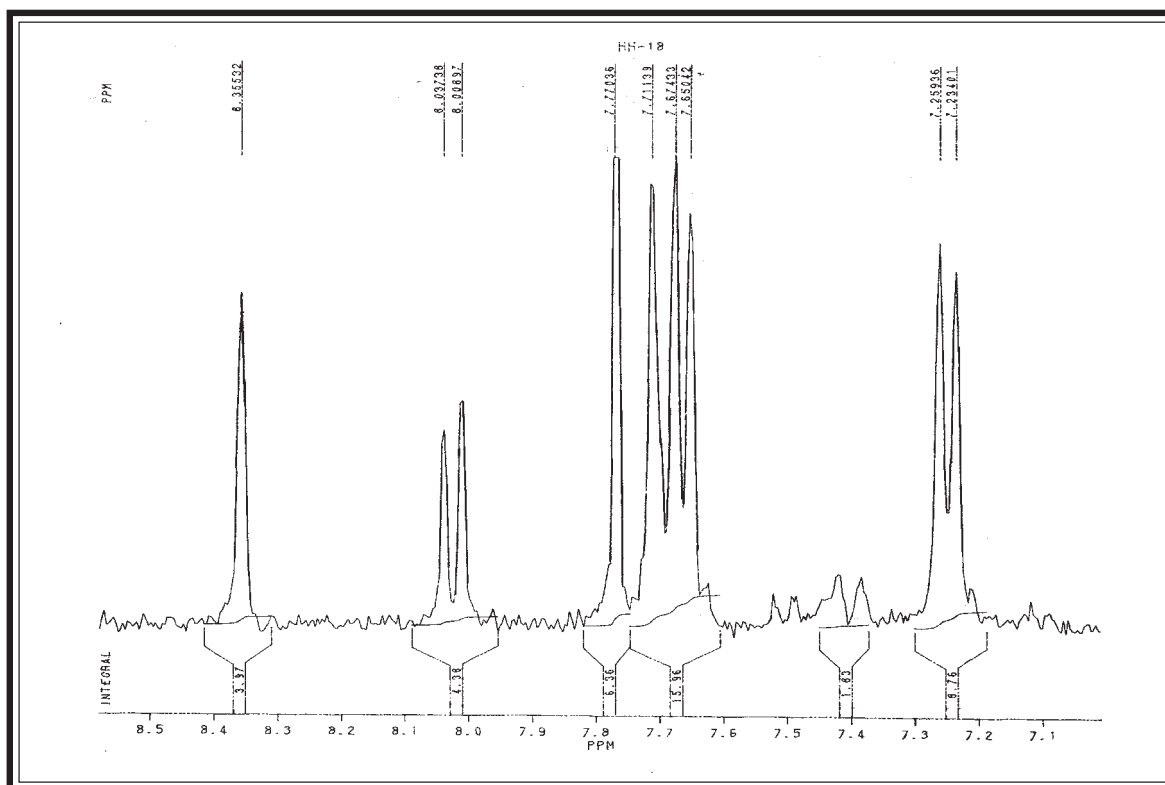
PMR SPECTRAL STUDY OF 1-ACETYL-3-(p-TOLYL)-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PYRAZOLINES



Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	2.39	3H	singlet	-	-COCH ₃
2.	2.47	3H	singlet	-	-Ar-CH ₃
3.	3.32-3.40	1H	double doublet	J _{AX} =5.00 J _{AB} =17.76	-CH _A
4.	3.92-4.02	1H	double doublet	J _{BX} =12.19 J _{BA} =17.71	-CH _B
5.	6.04-6.08	1H	double doublet	J _{XB} =11.21 J _{XA} =4.70	-CH _X
6.	7.23-7.26	2H	doublet	J _{hi} 7.61	Ar-H'hh'
7.	7.65-7.71	4H	multiplate	-	Ar-H'ii Ar-Hf Ar-Hd
8.	8.01-8.04	1H	doublet	J _{gf} 8.52	Ar-Hg
9.	8.36	1H	singlet	-	Ar-Ha

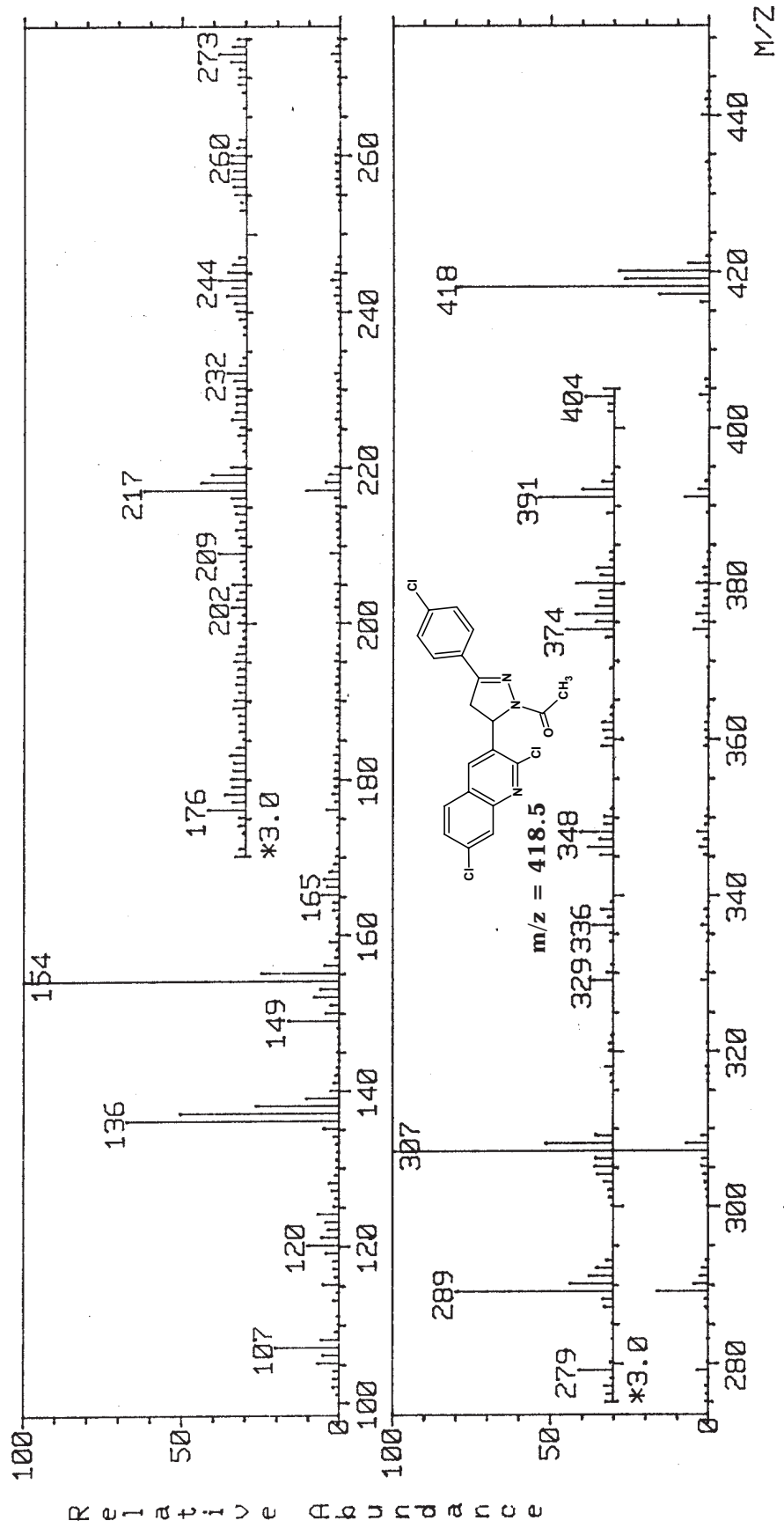
EXPANDED AROMATIC REGION


IR SPECTRAL DATA OF 1-ACETYL-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES

 Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	C=O str.
2a	C_6H_5	1656
2b	$4\text{-NH}_2\text{-C}_6\text{H}_4$	1658
2c	$4\text{-Br-C}_6\text{H}_4$	1658
2d	$4\text{-Cl-C}_6\text{H}_4$	1656
2e	$4\text{-F-C}_6\text{H}_4$	1658
2f	$\text{C}_4\text{H}_3\text{O}$	1658
2g	$2\text{-OH-C}_6\text{H}_4$	1659
2h	$4\text{-OH-C}_6\text{H}_4$	1660
2i	$4\text{-OCH}_3\text{-C}_6\text{H}_4$	1656
2j	$4\text{-CH}_3\text{-C}_6\text{H}_4$	1660
2k	$3\text{-NO}_2\text{-C}_6\text{H}_4$	1660
2l	$4\text{-NO}_2\text{-C}_6\text{H}_4$	1658

MASS SPECTRUM Data File: 3EJL25T 25-JUL- 3 12:11
 Sample: HHA-5 DR N A CHAUHAN,RAJKOT #6204
 RT 0.12" FAB(Pos.) GC 1.4c BP: m/z 154.0000 Int. 26.0172 Lv 0.00
 Scan# (1 to 3)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1-ACETYL-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES

[A] Synthesis of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones

See part - I, Section-I (C)

[B] Synthesis of 1-Acetyl-3-aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines

A mixture of 1-*p*-tolyl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-one (3.43 g, 0.01 mol) in 20 ml acetic acid and hydrazine hydrate (0.59, 0.01 mol) was refluxed for 10 hrs. The contents were poured onto ice & product was isolated, crystallised from ethanol yield 56%, m.p. 204°C. Anal. calcd. for C₂₁H₁₇N₃Cl₂O required: C, 63.33%; H, 4.3%; N, 10.55%; found: C, 63.20%; H, 4.15%; N, 10.43%.

TLC solvent system: Acetone : Benzene (2:8) visualizing agent : Iodine

Similarly other chalcones were condensed with hydrazine hydrate. The physical data along with infra red spectral data are recorded in Table Ic.

[C] Therapeutic evaluation of 1-Acetyl-3-aryl-5-(2',7',-dichloroquinolin-3'-yl)-pyrazolines

Antimicrobial testing was carried out as described in Part-I, Section-1 (D). The zone of inhibition of the test solution are recorded in Graphical Chart No. 2. The antitubercular activity data have been compared with standard drug rifampin at 6.25 µg/ml concentration and the data for % inhibition are recorded in Table No. 1d.

TABLE NO. 1c: PHYSICAL CONSTANTS OF 1-ACETYL-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
2a	C ₆ H ₅ -	C ₂₀ H ₁₃ N ₃ Cl ₂ O	384	90	0.496	56	10.94	10.80
2b	4-NH ₂ -C ₆ H ₄ -	C ₂₀ H ₁₆ N ₄ Cl ₂ O	399	212	0.550	68	14.03	14.30
2c	4-Br-C ₆ H ₄ -	C ₂₀ H ₁₄ N ₃ BrCl ₂ O	463	288	0.568	72	9.07	9.29
2d	4-Cl-C ₆ H ₄ -	C ₂₀ H ₁₃ N ₃ Cl ₃ O	418.5	160	0.801	70	10.04	10.40
2e	4-F-C ₆ H ₄ -	C ₂₀ H ₁₄ N ₃ Cl ₂ FO	402	218	0.471	64	10.45	10.30
2f	C ₄ H ₃ O-	C ₁₈ H ₁₃ N ₃ Cl ₂ O ₂	374	98	0.425	69	11.23	11.62
2g	2-OH-C ₆ H ₄ -	C ₂₀ H ₁₅ N ₃ Cl ₂ O ₂	400	158	0.569	59	10.50	10.24
2h	4-OH-C ₆ H ₄ -	C ₂₀ H ₁₅ N ₃ Cl ₂ O ₂	400	116	0.511	71	10.50	10.27
2i	4-OCH ₃ -C ₆ H ₄ -	C ₂₁ H ₁₇ N ₃ Cl ₂ O ₂	414	222	0.623	70	10.14	10.32
2j	4-CH ₃ -C ₆ H ₄ -	C ₂₁ H ₁₇ N ₃ Cl ₂ O	398	204	0.571	56	10.55	10.43
2k	3-NO ₂ -C ₆ H ₄ -	C ₂₀ H ₁₄ N ₄ Cl ₂ O ₃	429	192	0.614	64	13.05	13.29
2l	4-NO ₂ -C ₆ H ₄ -	C ₂₀ H ₁₄ N ₄ Cl ₂ O ₃	429	144	0.509	63	13.05	13.38

TLC Solvent System : Acetone : Benzene (2 : 8).

GRAPHICAL CHART NO. 2 : 1-ACETYL-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES

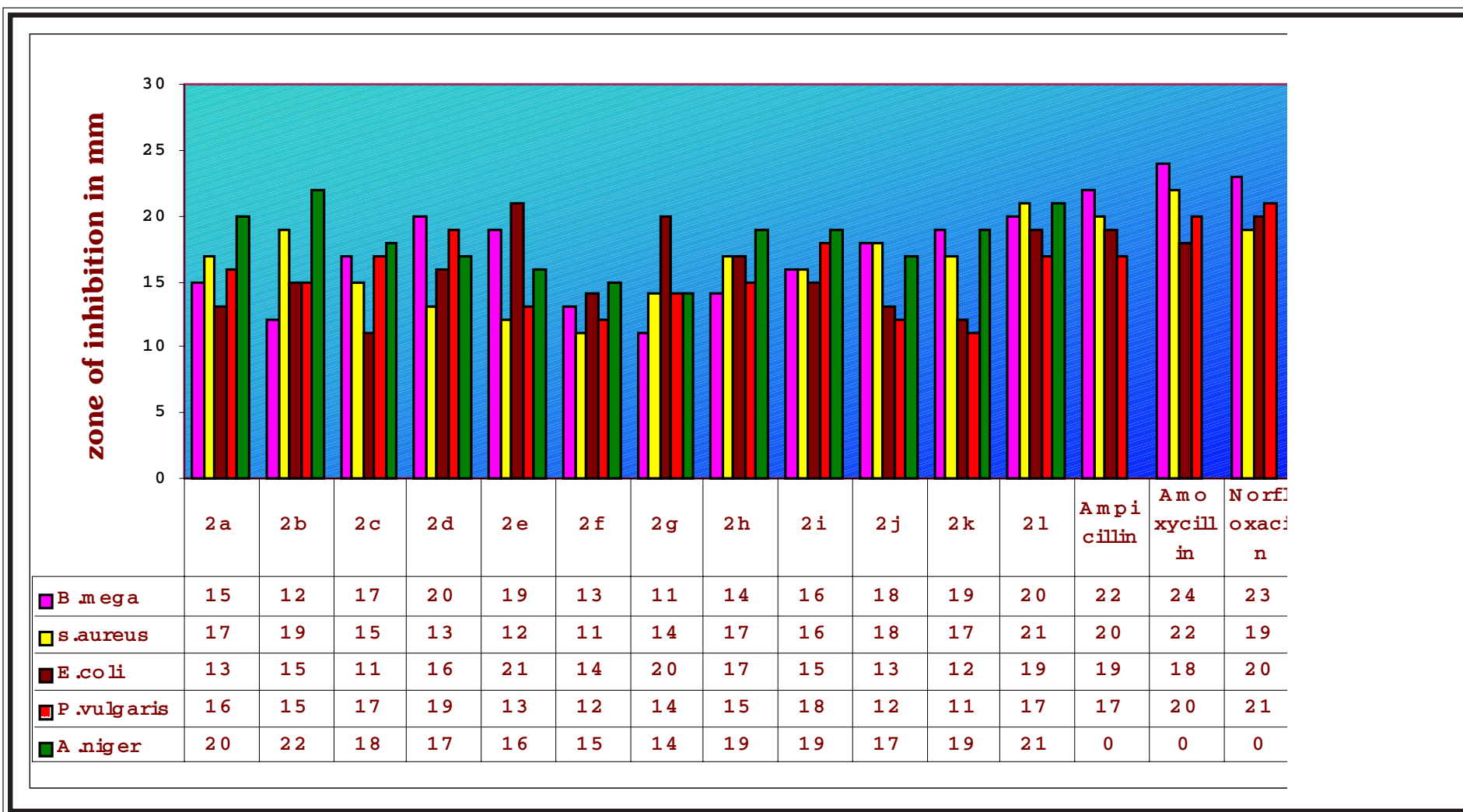
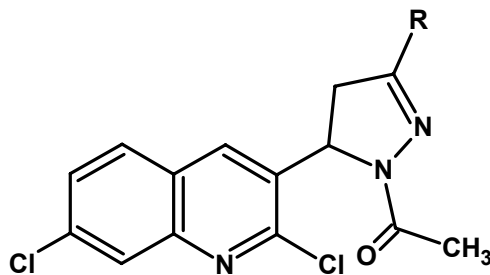


TABLE NO. 1c : PRIMARY ASSAY OF ANTITUBERCULAR ACTIVITY



TAACF, Southern Research Institute
Primary Assay Summary Report

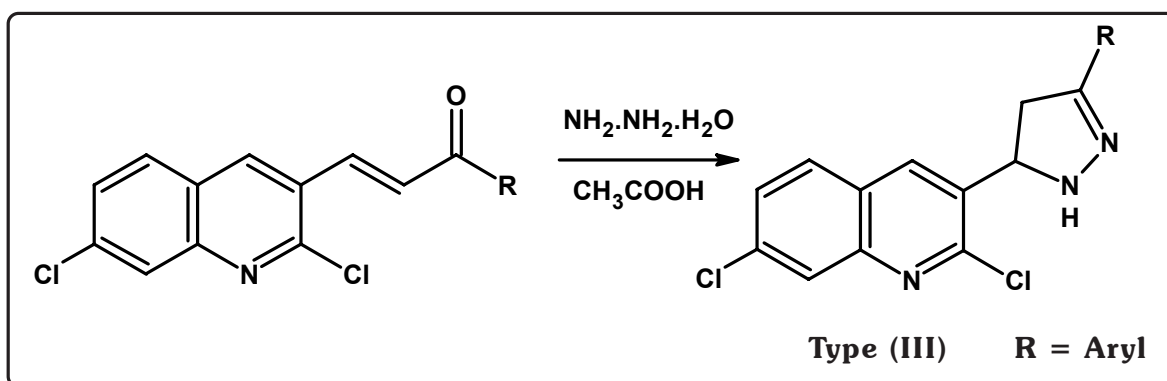
Dr. A. R. Parikh
Saurashtra University

Sample ID	Corp ID	Where, R =	Supplier	Assay	MTb Strain	MIC	% Inhib	Comment
162090	AP-38	4-F-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	54	MIC Rifampin=0.25µg/ml @98% inhibition
162083	AP-31	C ₆ H ₅	Saurashtra University	Alamar	H37Rv	>6.25	39	MIC Rifampin=0.25µg/ml @98% inhibition
16092	AP-40	4-NH ₂ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	24	MIC Rifampin=0.25µg/ml @98% inhibition
162084	AP-32	4-CH ₃ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	23	MIC Rifampin=0.25µg/ml @98% inhibition
162089	AP-37	4-NO ₂ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	21	MIC Rifampin=0.25µg/ml @98% inhibition
162086	AP-34	4-Cl-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	20	MIC Rifampin=0.25µg/ml @98% inhibition
162085	AP-33	4-OCH ₃ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	14	MIC Rifampin=0.25µg/ml @98% inhibition
162088	AP-36	4-OH-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	10	MIC Rifampin=0.25µg/ml @98% inhibition
162091	AP-39	C ₄ H ₃ O	Saurashtra University	Alamar	H37Rv	>6.25	5	MIC Rifampin=0.25µg/ml @98% inhibition
162093	AP-41	2-OH-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	0	MIC Rifampin=0.25µg/ml @98% inhibition

SECTION-III

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1H-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3-YL)-PYRAZOLINES

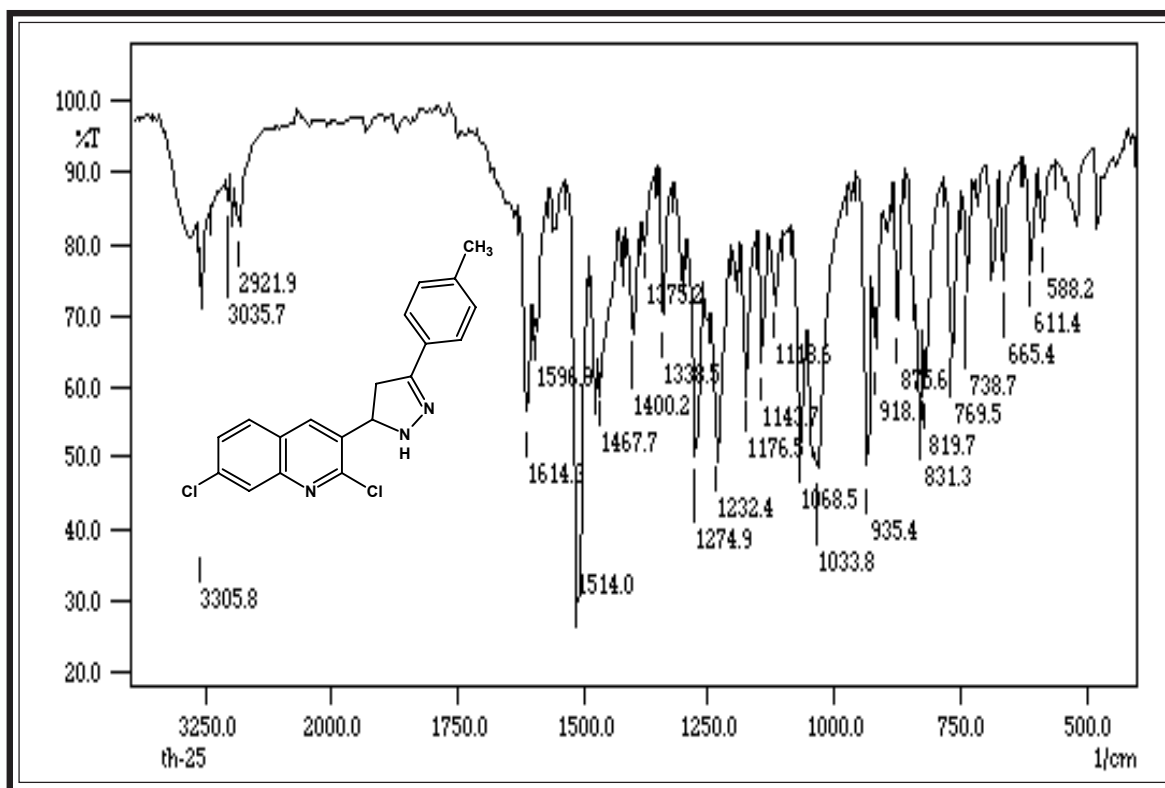
Pyrazoline derivatives are associated with diversified biological properties. Hence, it was thought of interest that a pyrazoline ring, if coupled to a 2,7-dichloroquinolin-3-carboxaldehyde nucleus, the resulting compounds may possess significant biological potency. Hence the synthesis of 1H-3-aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines of type (III) have been undertaken by the cyclo-condensation of chalcones of the type (I) with hydrazine hydrate.



The constitution of the synthesised products have been characterised by using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

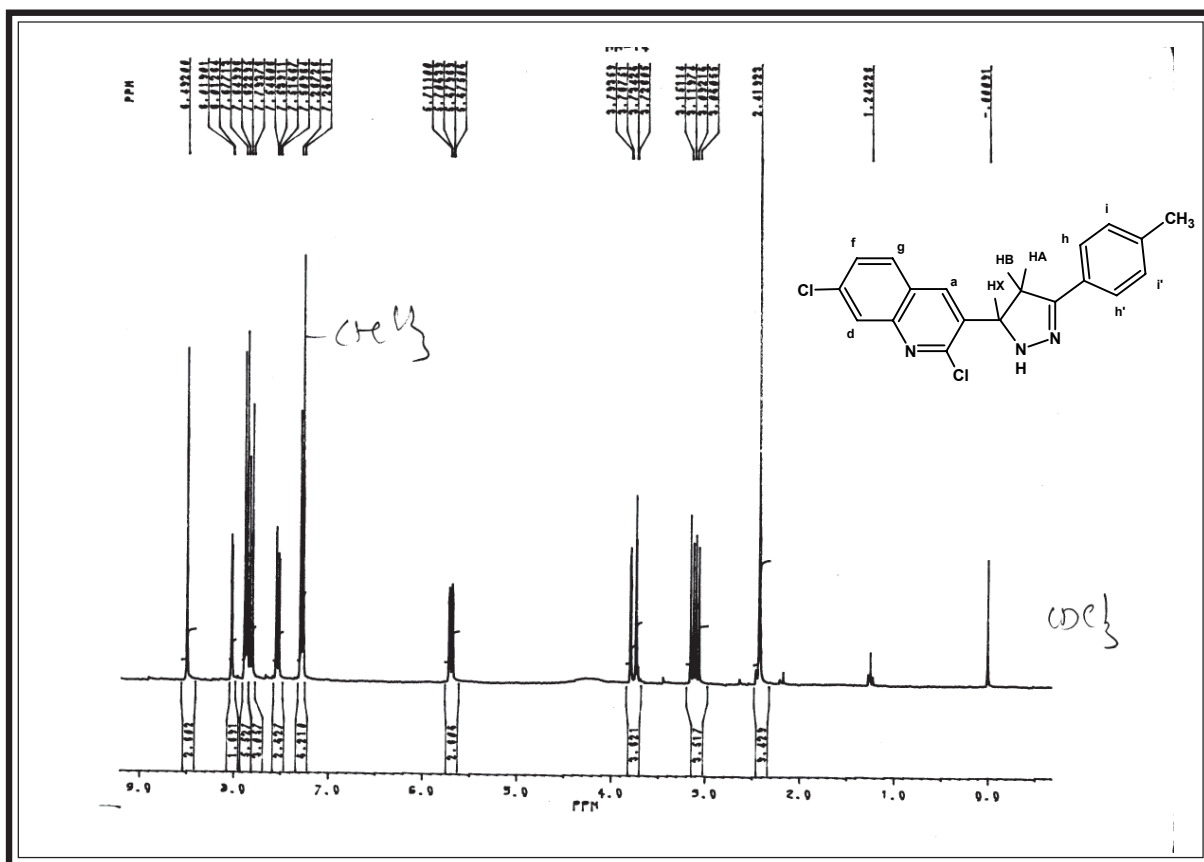
IR SPECTRAL STUDY OF 1H-3-p-TOLYL-5-(2'-7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane	C – H str.	2921	2975-2850	419
	C – H def.	1467	1470-1435	"
Aromatic	C– H str.	3035	3080-3030	"
	C = C str.	1596	1585-1570	"
		1514	1520-1480	"
	C – H def.	1068	1070-1000	420
		831	835-810	"
Quinoline moiety	C = N str.	1596	1612-1593	419
	C – Cl str.	738	750-700	"
Pyrazoline	C = N str.	1614	1627-1580	"
	N–H str.	3305	3200-3400	"

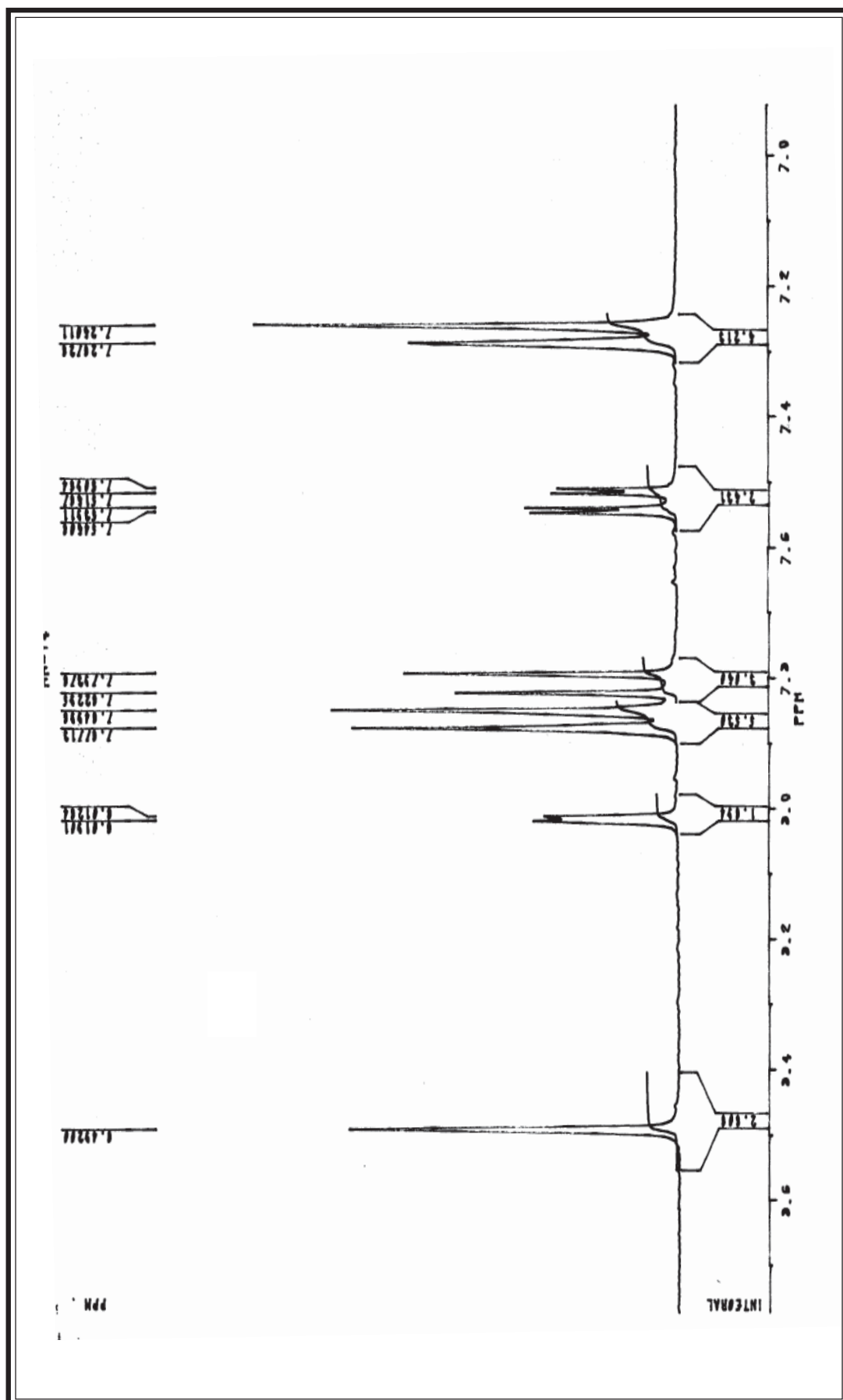
PMR SPECTRAL STUDY OF 1H-3-(p-TOLYL)-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES



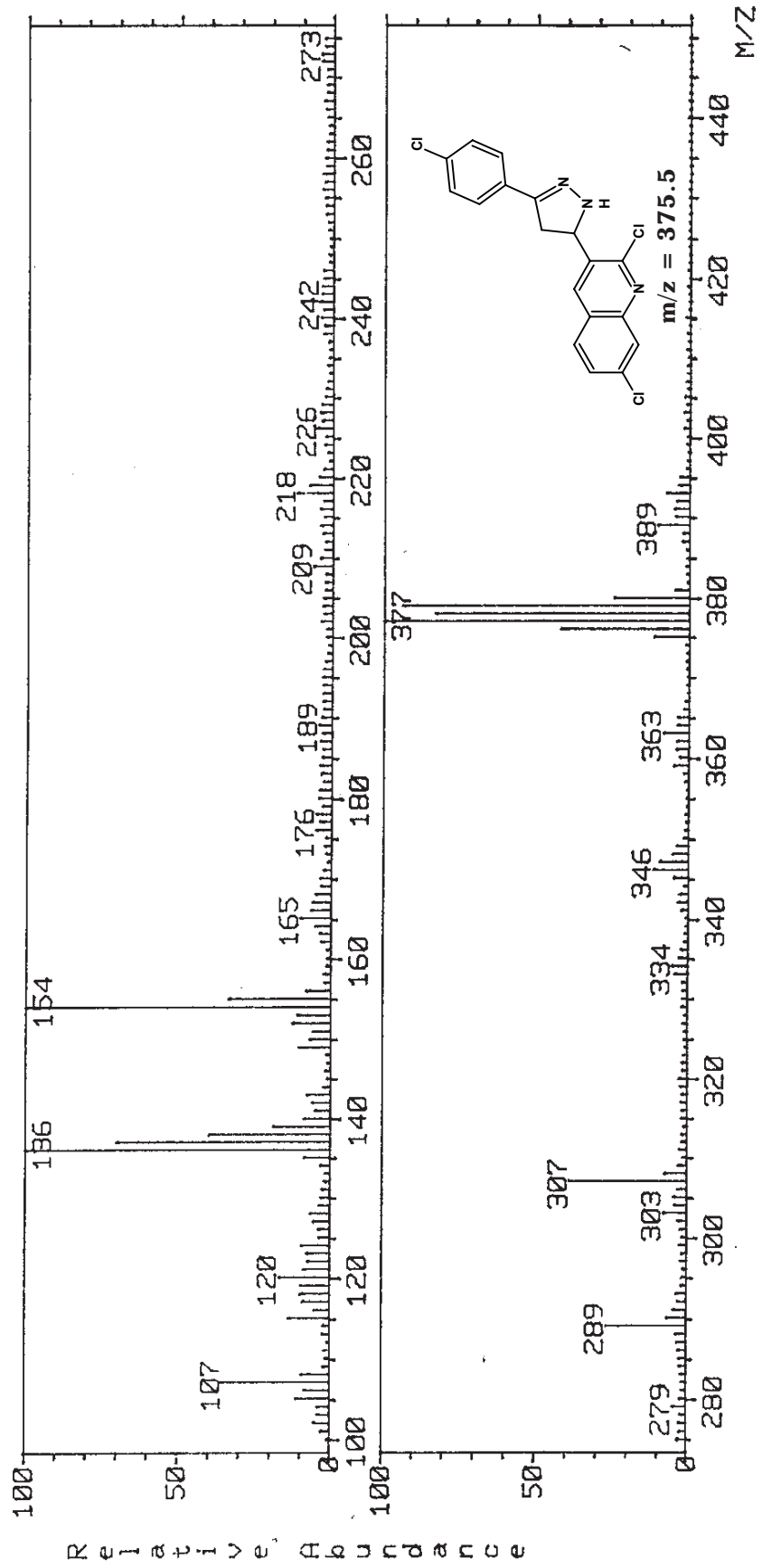
Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	2.41	3H	singlet	-	Ar- CH_3
2.	3.06-3.15	1H	double doublet	$J_{AB}=17.76$ $J_{AX}=9.42$	$-\text{CH}_A$
3.	3.72-3.79	1H	double doublet	$J_{BA}=17.80$ $J_{BX}=1.82$	$-\text{CH}_B$
4.	5.68-5.71	1H	double doublet	$J_{XA}=9.40$ $J_{XB}=1.45$	$-\text{CH}_X$
5.	7.26-7.55	2H	doublet	J _{hi} 8.15	Ar-H h'h'
6.	7.51-7.55	1H	double doublet	J _{fg} 8.76 J _{fd} 2.09	Ar-Hf
7.	7.79-7.82	1H	doublet	J _{gf} 8.75	Ar-Hg
8.	7.85-7.88	2H	doublet	J _{hi} 8.14	Ar-H j'j
9.	8.01-8.02	1H	doublet	J _{df} 2.09	Ar-Hd
10.	8.49	1H	singlet	-	Ar-Ha

EXPANDED AROMATIC REGION



MASS SPECTRUM Data File: 3EJL255 25-JUL- 3 12:06
 Sample: HHA-4 DR N A CHAUHAN,RAJKOT #6204
 RT 0.48" FAB(Pos.) GC 1.4c BP: m/z 136.0000 Int. 89.4714 Lv 0.00
 Scan# (5)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1H-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES

[A] Synthesis of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones

See Part-I, Section I (C).

[B] Synthesis of 1H-3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines

A mixture of 1-tolyl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-one (3.43 g, 0.01 mol) in 25 ml ethanol, hydrazine hydrate (0.5 g, 0.01 mol) and 2-3 drops of acetic acid was refluxed for 12 hrs. The resulting mixture was poured into ice and neutralised with ammonia solution. The product was isolated and crystallised from ethanol. Yield 64%, m.p. 176^oC Anal. Calcd. for C₁₉H₁₅N₃Cl₂ required: C, 64.06%; H, 4.24%; N, 11.80%; found: C, 63.86%; H, 3.99%; N, 11.57%.

TLC solvent system: Acetone: Benzene (1.5 : 8.5) Visualizing agent : Iodine

Similarly other 1H-3-Aryl-5-(2',7'-dichloro quinolin-3-yl)-pyrazolines were prepared. The physical data along with infra red spectral data are recorded in Table Ie.

[C] Therapeutic evaluation of 1H-3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-pyrazolines

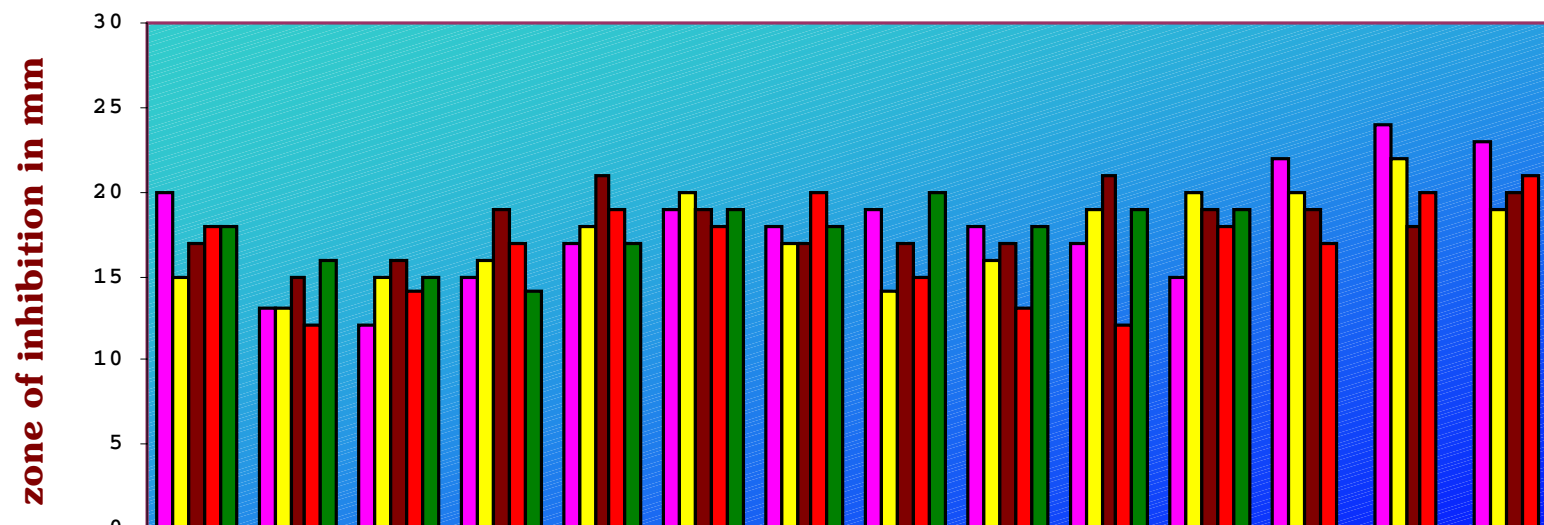
Antimicrobial testing was carried out as described in Part-I, Section-I (D). The zones of inhibition of the test solution are recorded in Graphical Chart No. 3.

TABLE NO. 1e: PHYSICAL CONSTANTS OF 1-H-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES

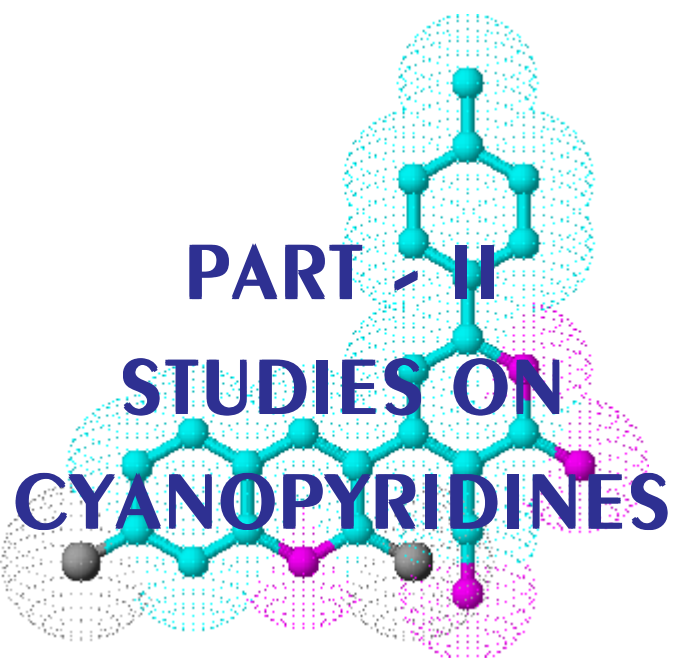
Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
3a	C ₆ H ₅ -	C ₁₈ H ₁₃ N ₃ Cl ₂	342	186	0.485	53	12.28	12.08
3b	4-NH ₂ -C ₆ H ₄ -	C ₁₈ H ₁₄ N ₄ Cl ₂	357	210	0.450	59	15.68	15.46
3c	4-Br-C ₆ H ₄ -	C ₁₈ H ₁₂ N ₃ BrCl ₂	421	>300	0.568	58	9.98	9.65
3d	4-Cl-C ₆ H ₄ -	C ₁₈ H ₁₂ N ₃ Cl ₃	375.5	240	0.612	60	11.16	11.55
3e	4-F-C ₆ H ₄ -	C ₁₈ H ₁₂ N ₃ Cl ₂ F	360	172	0.520	62	11.67	10.27
3f	2-OH-C ₆ H ₄ -	C ₁₈ H ₁₃ N ₃ Cl ₂ O	358	236	0.741	57	11.73	11.99
3g	4-OH-C ₆ H ₄ -	C ₁₈ H ₁₃ N ₃ Cl ₂ O	358	160	0.639	66	11.73	11.58
3h	4-OCH ₃ -C ₆ H ₄ -	C ₁₉ H ₁₅ N ₃ Cl ₂ O	372	184	0.580	69	11.29	11.65
3i	4-CH ₃ -C ₆ H ₄ -	C ₁₉ H ₁₅ N ₃ Cl ₂	356	176	0.565	64	11.80	11.57
3j	3-NO ₂ -C ₆ H ₄ -	C ₁₈ H ₁₂ N ₄ Cl ₂ O ₂	387	136	0.419	70	14.47	14.95
3k	4-NO ₂ -C ₆ H ₄ -	C ₁₈ H ₁₂ N ₄ Cl ₂ O ₂	387	216	0.678	55	14.47	14.05

TLC Solvent System : Acetone : Benzene (1.5 : 8.5).

GRAPHICAL CHART NO. 3 : 1H-3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-PYRAZOLINES



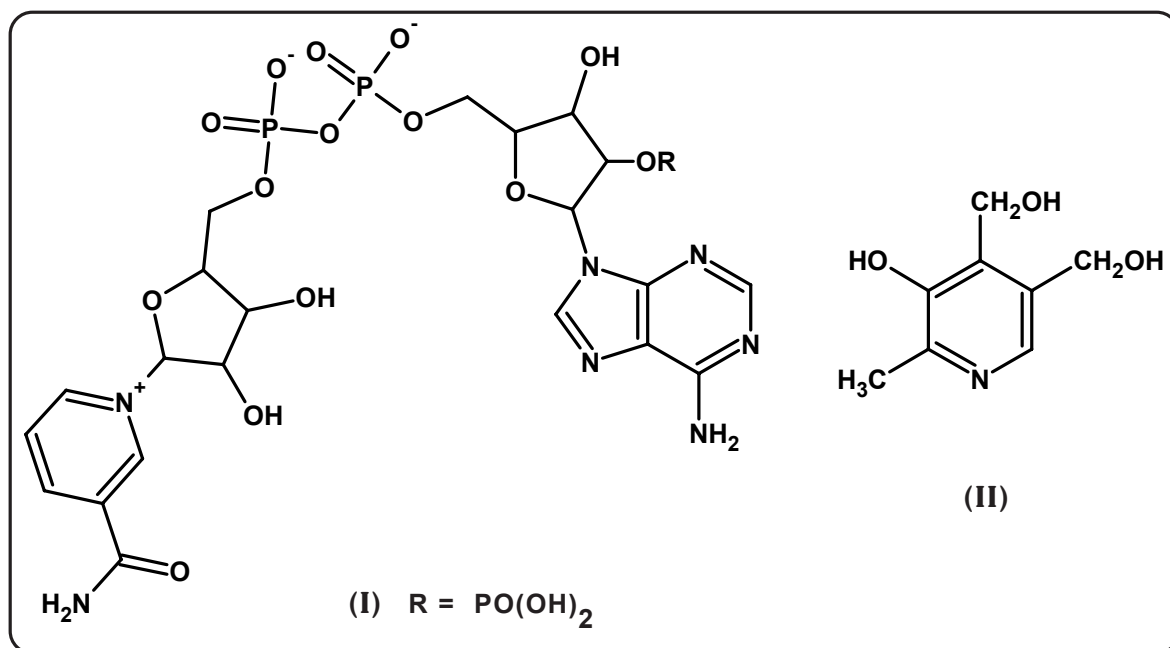
	3 a	3 b	3 c	3 d	3 e	3 f	3 g	3 h	3 i	3 j	3 k	Ampicillin	Amoxicillin	Norfl oxacin
B .mega	20	13	12	15	17	19	18	19	18	17	15	22	24	23
S .aureus	15	13	15	16	18	20	17	14	16	19	20	20	22	19
E .coli	17	15	16	19	21	19	17	17	17	21	19	19	18	20
P .vulgaris	18	12	14	17	19	18	20	15	13	12	18	17	20	21
A .niger	18	16	15	14	17	19	18	20	18	19	19	0	0	0



PART - II
STUDIES ON
CYANOPYRIDINES

INTRODUCTION

Historically, a wide range of biological activities have been attributed to pyridine derivatives. Pyridine-3-carboxamide occurs as a component of the structure of the important coenzymes NADP + (I), one of the B₂ complex of vitamins, occurs in red blood corpuscles and participates in biochemical redox reaction. Pyridoxal (Vitamin B6) (II), occurs in yeast and wheatgerm is an important food additive.

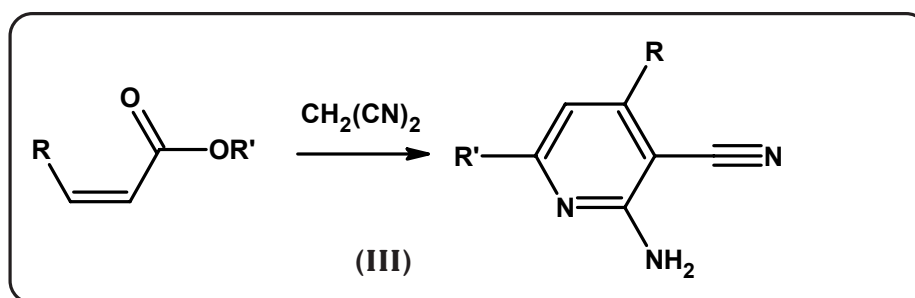


The availability of 3-cyanopyridine, nicotinamide and nicotinic acid make possible their use as synthetic intermediates.

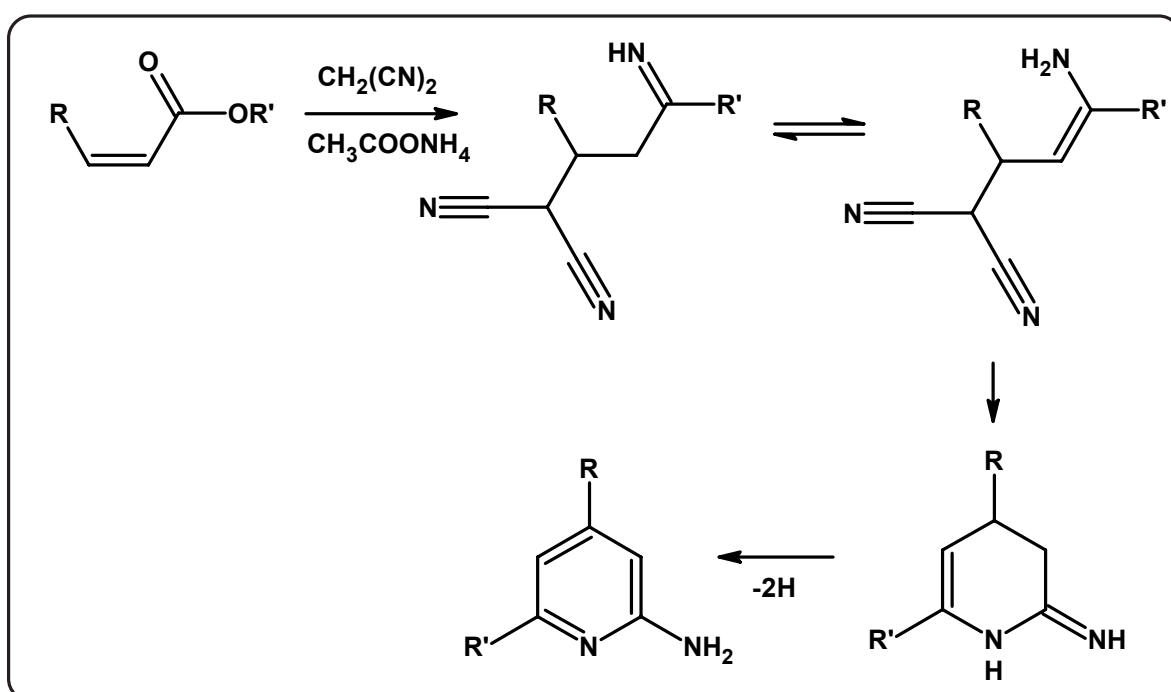
SYNTHETIC ASPECT

Different methods for the preparation of 3-cyanopyridines are available in literature¹⁰⁵⁻¹¹¹. The well known method is:

Sakurai and Midorikwa^{112,113} have reported that malononitrile reacts with α,β unsaturated ketones to give 2-amino-3-cyano-4,6-disubstituted pyridines (III).

**MECHANISM:**

The reaction proceeds through conjugated addition of active methylene compounds to the α,β -unsaturated system as shown below.

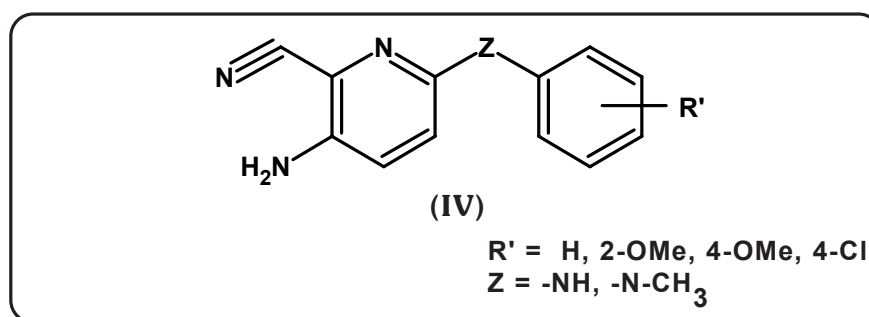
**THERAPEUTIC IMPORTANCE**

The cyanopyridine derivatives are extensively used in medicine, due to its antidiabetic, antihypertensive, anticholestemic, antifungal and antibacterial properties. Various cyanopyridines are known to exhibit a broad spectrum of biological activities such as,

1. Anti HIV¹¹⁴
2. Antitubercular¹¹⁵
3. Analgesic¹¹⁶
4. Insecticidal¹¹⁷
5. Antisoriasis¹¹⁸

6. Antihypertensive¹¹⁹
7. Antifungal¹²⁰
8. Antiepileptic¹²¹
9. Anticonvulsant¹²²
10. Antibacterial^{123,124}
11. Antiinflammatory¹²⁵

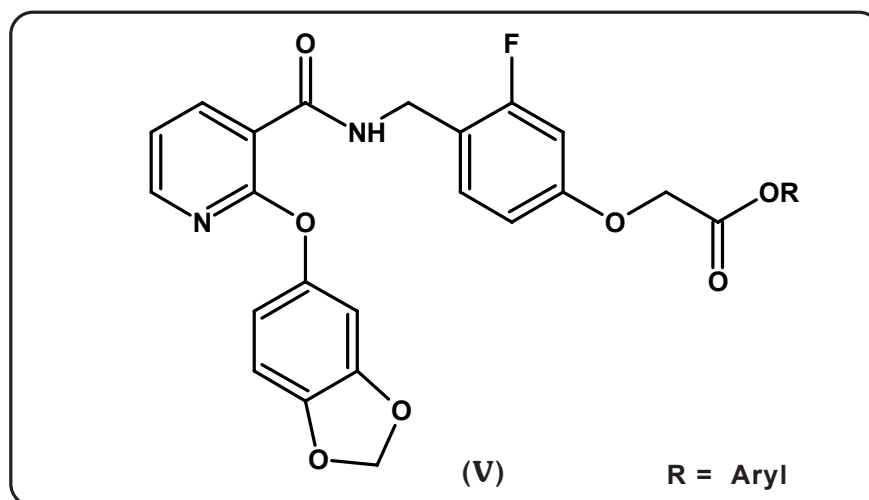
Gangjee et al.¹²⁶ have synthesised some novel cyanopyridine derivatives of type-(IV) as potent antitumor agents.



The insecticidal activity of cyanopyridines has been screened by Y. Sasaki et al.¹²⁷ Umed Ten et al.¹²⁸ have prepared cyanopyridines as agrochemical fungicides. The oxide activator bleaching activity of cyanopyridine has been proved by Rees M.¹²⁹ Oshida Mario¹³⁰ prepared cyanopyridine derivatives which inhibit cerebral edema and delayed neuron death. Hence, they are useful as cerebral edema inhibitors as cerebrovascular disorder remedies.

Several co-workers have prepared some novel cyanopyridine derivatives and reported their cholinesterase inhibitors¹³¹, antihistaminic and antiallergic¹³², adernegic¹³³, herbicidal¹³⁴, antiinflammatory¹³⁵, and insecticidal¹³⁶ activities.

Chambers et al.¹³⁷ have synthesised new carbonyl substituted pyridine derivatives (V) and proved that they are inhibitors of phosphodiesterase (IV) isozymes.



Some new 3-cyanopyridine derivatives have been prepared by Hammama A. and co-workers¹³⁸ showing anticancer and anti HIV-I activity.

CONTRIBUTION FROM OUR LABORATORY

H. Parekh et al.¹³⁹⁻¹⁴¹ have synthesised the series of cyanopyridines and postulated them as antimicrobial agents. A. R. Parikh et al.¹⁴² have prepared some new cyanopyridines and studied their antimicrobial activity. H. H. Parekh et al.¹⁴³ have synthesised 3-cyanopyridines bearing quinoline nucleus and tested their antimicrobial and antitubercular activity.

Abdallah N. et al.¹⁴⁴ have prepared cyanopyridine derivatives which showed analgesic and antiinflammatory activity. Ladouceur Getan H. et al.¹⁴⁵ have synthesised some new pyridine derivatives behaving as glucagon antagonist. Caroline Charlie and co-worker¹⁴⁶ have prepared pyridine derivatives having antiinflammatory activity. Antimicrobial activity of 3-cyanopyridine derivatives have been studied by Mona Komel et al.¹⁴⁷

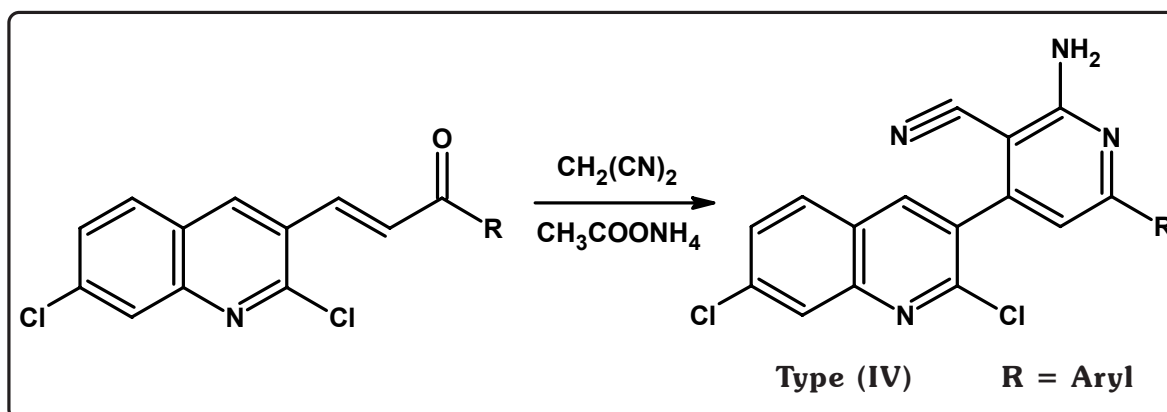
In view of the above observations, the synthesis of new 3-cyanopyridine derivatives bearing 2,7-dichloroquinoline moiety, were aimed at investigating biological activities of these compounds.

SECTION - I : SYNTHESIS AND THERAPEUTIC EVALUATION OF 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL PYRIDINES

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL PYRIDINES

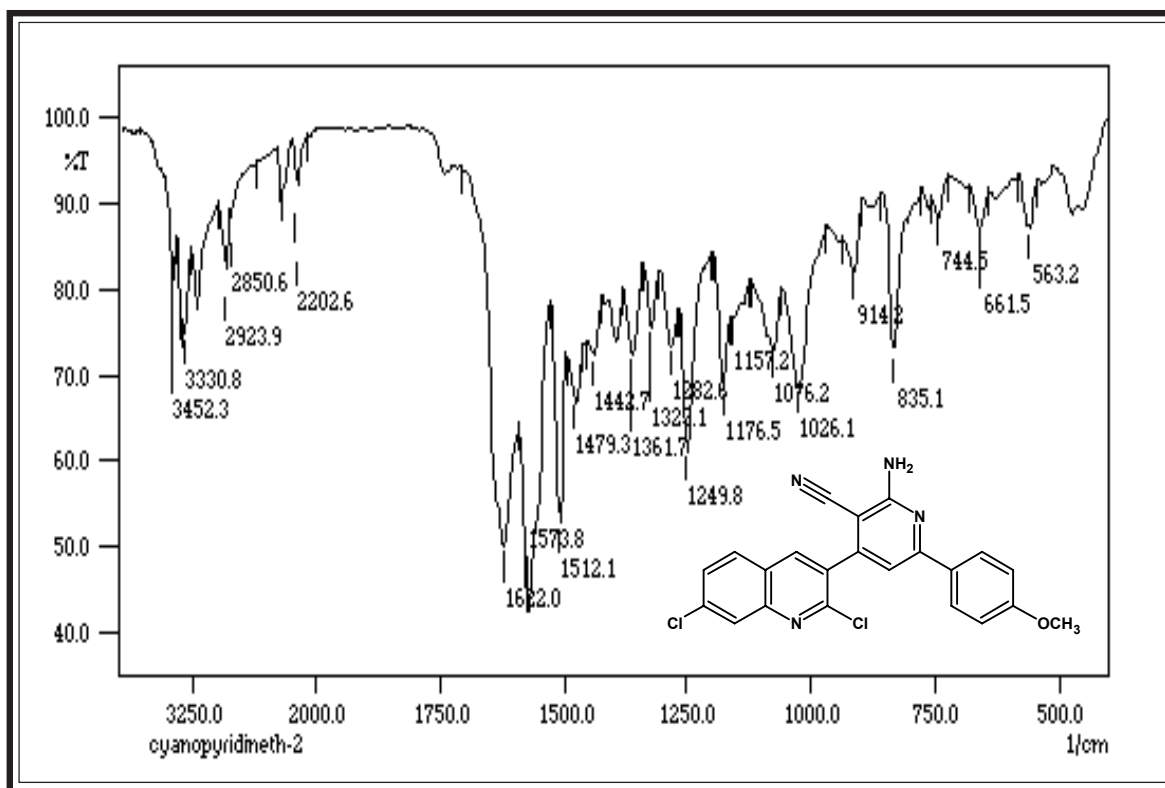
A perusal of the literature has revealed manifold implications of cyanopyridines, viz. antibacterial, antifungal, and antiallergic etc. The cyanopyridine derivatives are endowed with variety of pharmacological activities. In view of these findings, it was contemplated to synthesise some new cyanopyridine derivatives bearing 2,7-dichloroquinolin-3-carboxaldehyde moiety and assess their biopotential. The synthesis was carried out by the condensation of chalcones of type (I) with malononitrile and ammonium acetate.



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass-spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 $\mu\text{g}/\text{ml}$. The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

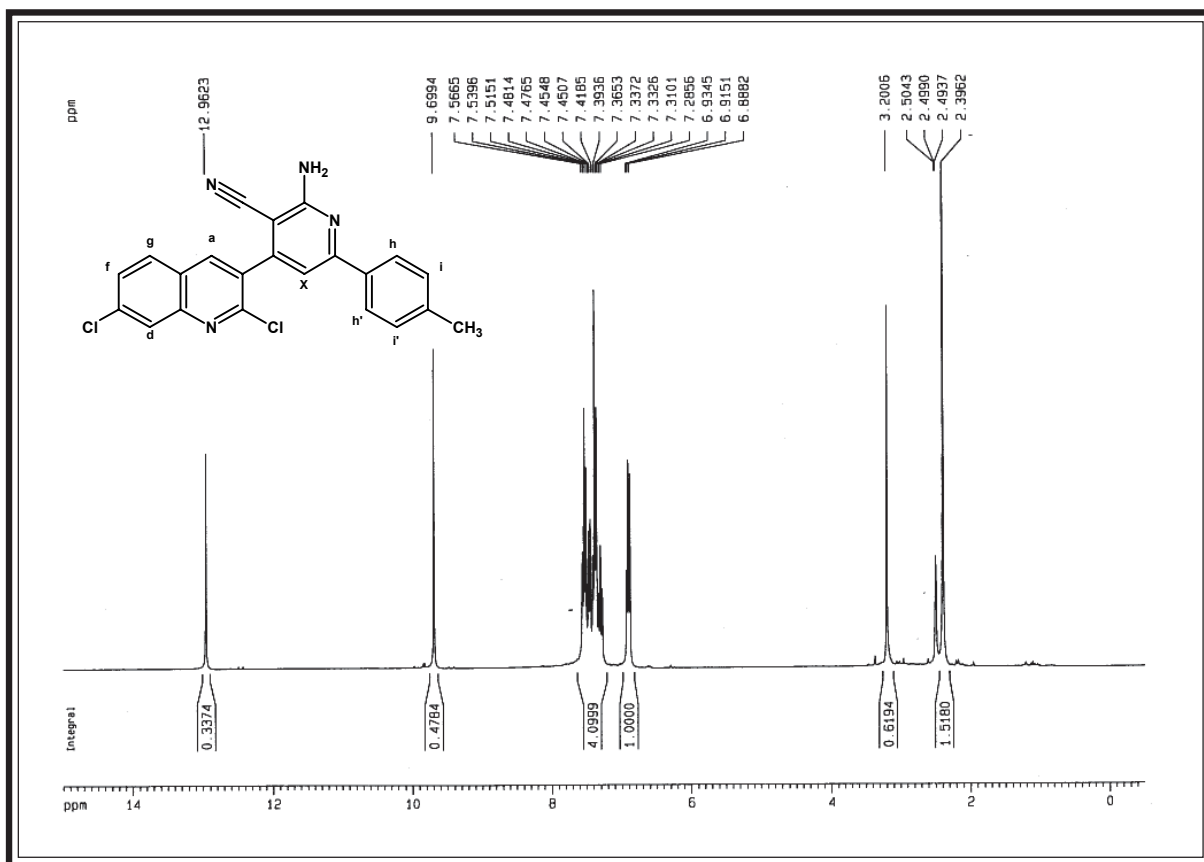
IR SPECTRAL STUDY OF 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-p-ANISYL PYRIDINE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane -CH ₃	C – H str. (asym)	2923	2975-2950	419
	C – H str. (sym)	2850	2880-2860	"
	C – H def. (asym)	1442	1470-1435	"
	C – H def. (sym)	1361	1385-1370	"
Aromatic	C– H str.	3088	3080-3030	"
	C = C str.	1576	1585-1570	"
		1512	1520-1480	"
	C – H i.p.def.	1076	1125-1090	420
		1026	1070-1000	"
Quinoline moiety	C – H. o.o.p. def.	835	835-810	"
	C = N str.	1622	1612-1593	419
Pyridine	C – Cl str.	744	750-700	"
	C = H str.	3088	3080-3030	"
	C = C str.	1622	1650-1520	"
	C = N str.	(overlaped) 1573	1580-1550	"
Nitrile Ether	(overlaped)			
	C \equiv N str.	2202	2240-2220	"
	C – O – C str.	1249	1275-1200	418
	N-H str.	3330	3400-3250	419

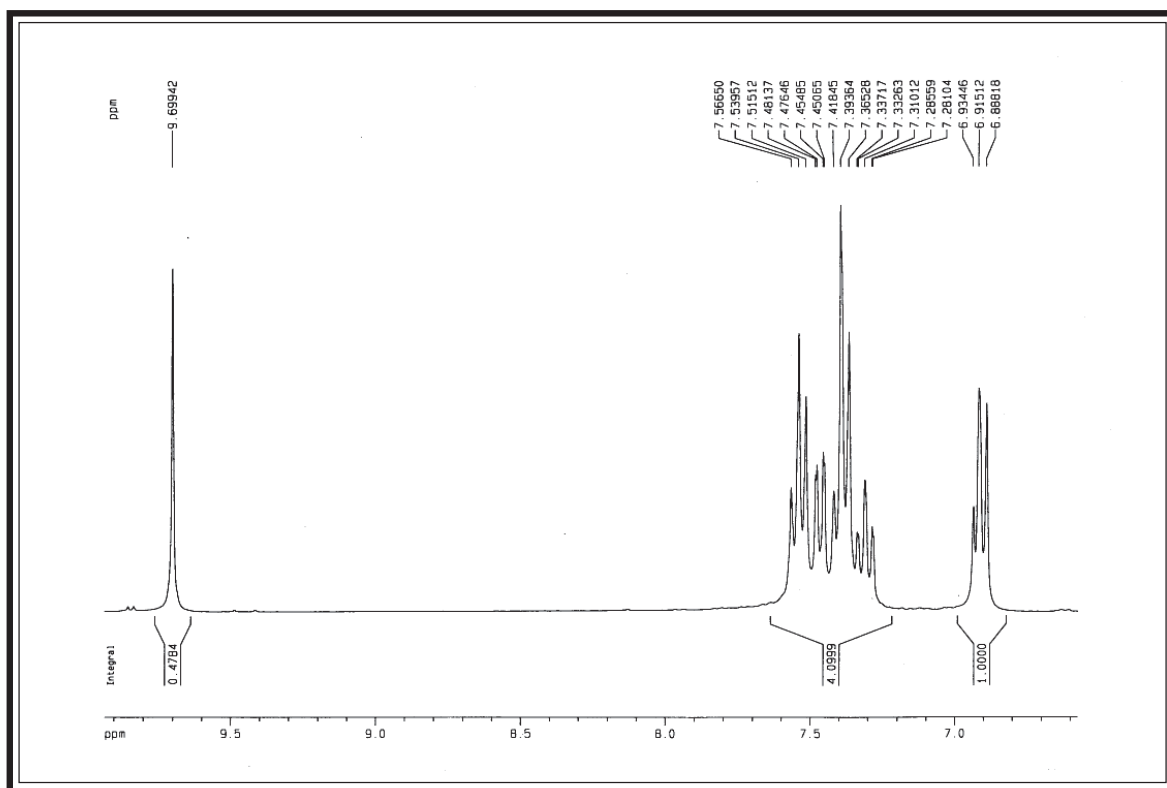
PMR SPECTRAL STUDY OF 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-(p-TOLYL)-PYRIDINES



Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	2.4	3H	singlet	-	Ar- CH_3
2.	6.89-6.92	2H	doublet	Jih 8.08	Ar-H i,i'
3.	7.28-7.29	1H	doublet	Jdf 1.37	Ar-Hd
4.	7.31	1H	singlet	-	Ar-HX
5.	7.39-7.42	2H	doublet	Jhi 8.51	Ar-H h,h'
6.	7.45-7.46	1H	double doublet	Jfg 7.96 Jfd 1.47	Ar-Hf
7.	7.51-7.54	1H	doublet	Jgf 7.34	Ar-Hg
8.	9.70	1H	singlet	-	Ar-Ha

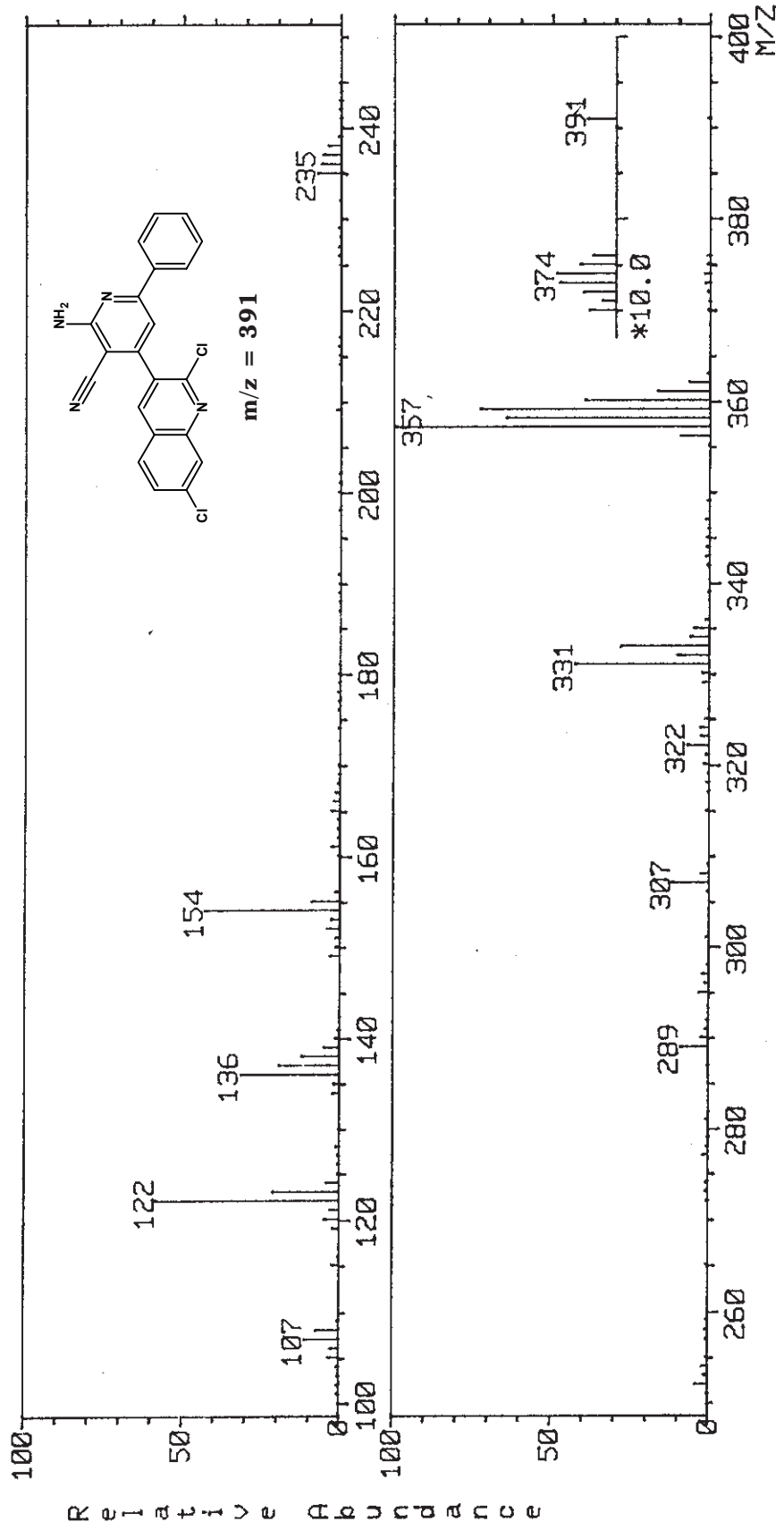
EXPANDED AROMATIC REGION


IR SPECTRAL DATA OF 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL PYRIDINES

 Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	C≡N str.
4a	C_6H_5	2200
4b	$4\text{-NH}_2\text{-C}_6\text{H}_4$	2198
4c	$4\text{-Br-C}_6\text{H}_4$	2202
4d	$4\text{-Cl-C}_6\text{H}_4$	2210
4e	$4\text{-F-C}_6\text{H}_4$	2198
4f	$\text{C}_4\text{H}_3\text{O}$	2196
4g	$2\text{-OH-C}_6\text{H}_4$	2198
4h	$4\text{-OH-C}_6\text{H}_4$	2200
4i	$4\text{-OCH}_3\text{-C}_6\text{H}_4$	2202
4j	$4\text{-CH}_3\text{-C}_6\text{H}_4$	2202
4k	$3\text{-NO}_2\text{-C}_6\text{H}_4$	2204
4l	$4\text{-NO}_2\text{-C}_6\text{H}_4$	2202

MASS SPECTRUM Data File: 3EJL25X 25-JUL- 3 12:33
Sample: HHA-9 DR NA CHAUHAN, RAJKOT #6204
RT 0.12" FAB(Pos.) GC 1.4c BP: m/z 357.0000 Int. 16.3777 Lv 0.00
Scan# (1 to 3)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-PYRIDINES

[A] Synthesis of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones.

See part (I) Section - I (C).

[B] Synthesis of 2-Amino-3-cyano-4-(2',7'-dichloroquinoline-3'-yl)-6-aryl-pyridines

A mixture of 1-p-tolyl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-one (342 g, 0.01 mol), malononitrile (1.29 g, 0.01 mol) and ammonium acetate (9.24 g, 0.12 mol) dissolved in absolute alcohol (50 ml) were heated under reflux for 12 hrs., cooled and poured into crushed ice. The product separated was filtered and recrystallised from ethanol. Yield 52%, m.p 162^oC; Anal. Calcd. for C₂₂H₁₄N₄Cl₂ required : C, 65.20%; H, 3.48%; N, 13.82%; found : C, 65.25%; H, 3.45%; N, 13.58%.

TLC Solvent System : Acetone : Benzene (10:0), Visualizing Agent : Iodine.

Similarly, other 2-amino-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl-pyridines were prepared. The physical data along with infra red spectral data are recorded in Tabel No. 2a.

[C] Therapeutic evaluation of 2-Amino-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl-pyridines

Antimicrobial testing was carried out as described in Part-I, Section-I (D). The zone of inhibition of the test solution are recorded in Graphical Chart No. 4. The antitubercular activity data have been compared with standard drug rifampin at 6.25 µg/ml concentration and the data for % inhibition are recorded in Table No. 2b.

TABLE NO. 2a : PHYSICAL CONSTANTS OF 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-PYRIMIDINES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
4a	C ₆ H ₅ -	C ₂₁ H ₁₂ N ₄ Cl ₂	391	192	0.413	63	14.32	14.67
4b	4-NH ₂ -C ₆ H ₄ -	C ₂₁ H ₁₅ N ₅ Cl ₂	406	182	0.477	53	17.24	17.64
4c	4-Br-C ₆ H ₄ -	C ₂₁ H ₁₁ N ₄ BrCl ₂	470	198	0.550	54	11.92	11.65
4d	4-Cl-C ₆ H ₄ -	C ₂₁ H ₁₁ N ₄ Cl ₃	425.5	104	0.509	58	13.16	13.50
4e	4-F-C ₆ H ₄ -	C ₂₁ H ₁₁ N ₄ Cl ₂ F	409	162	0.635	56	13.69	13.93
4f	C ₄ H ₃ O-	C ₁₉ H ₁₀ N ₄ Cl ₂ O	381	202	0.521	59	13.70	14.88
4g	2-OH-C ₆ H ₄ -	C ₂₁ H ₁₂ N ₄ Cl ₂ O	407	184	0.487	61	13.76	13.99
4h	4-OH-C ₆ H ₄ -	C ₂₁ H ₁₂ N ₄ Cl ₂ O	407	166	0.620	63	13.76	13.51
4i	4-OCH ₃ -C ₆ H ₄ -	C ₂₂ H ₁₄ N ₄ Cl ₂ O	421	122	0.537	60	13.30	13.64
4j	4-CH ₃ -C ₆ H ₄ -	C ₂₂ H ₁₄ N ₄ Cl ₂	405	162	0.566	52	13.82	13.58
4k	3-NO ₂ -C ₆ H ₄ -	C ₂₁ H ₁₁ N ₅ Cl ₂ O ₂	436	152	0.498	51	16.50	16.77
4l	4-NO ₂ -C ₆ H ₄ -	C ₂₁ H ₁₁ N ₅ Cl ₂ O ₂	436	>250	0.538	57	16.50	16.83

TLC Solvent System : Acetone : Benzene (10 : 0).

GRAPHICAL CHART NO. 4 : 2-AMINO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL PYRIDINES

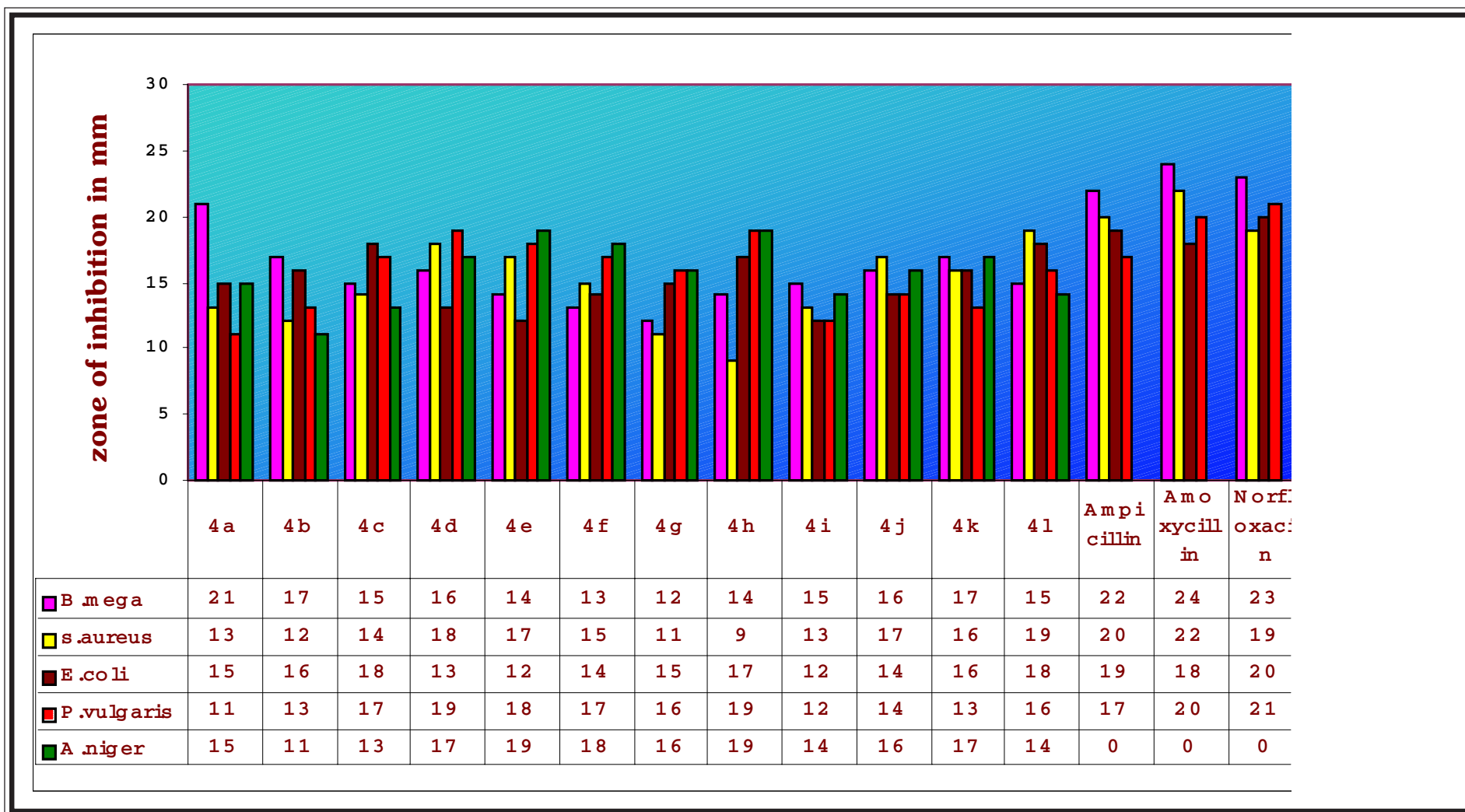
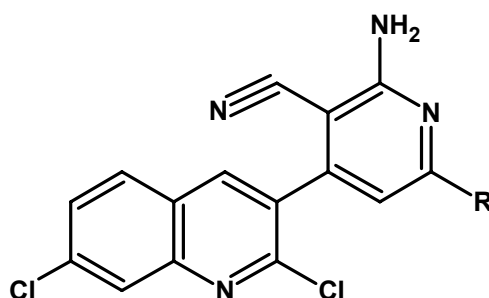


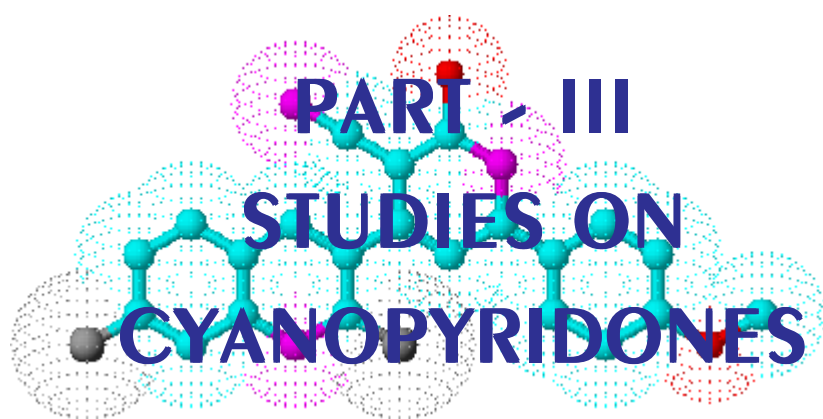
TABLE NO. 2a : PRIMARY ASSAY OF ANTITUBERCULAR ACTIVITY



TAACF, Southern Research Institute
Primary Assay Summary Report

Dr. A. R. Parikh
Saurashtra University

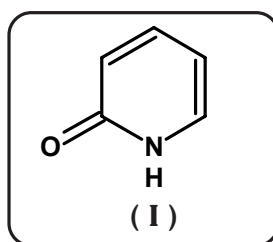
Sample ID	Corp ID	Where, R =	Supplier	Assay	MTb Strain	MIC	% Inhib	Comment
162100	AP-48	4-F-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	93	MIC Rifampin=0.25µg/ml @98% inhibition
162095	AP-43	4-CH ₃ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	64	MIC Rifampin=0.25µg/ml @98% inhibition
162096	AP-44	4-OCH ₃ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	57	MIC Rifampin=0.25µg/ml @98% inhibition
162098	AP-46	4-OH-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	44	MIC Rifampin=0.25µg/ml @98% inhibition
162099	AP-47	4-NO ₂ -C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	32	MIC Rifampin=0.25µg/ml @98% inhibition
162097	AP-45	4-Br-C ₆ H ₄	Saurashtra University	Alamar	H37Rv	>6.25	30	MIC Rifampin=0.25µg/ml @98% inhibition
162101	AP-49	C ₄ H ₃ O	Saurashtra University	Alamar	H37Rv	>6.25	25	MIC Rifampin=0.25µg/ml @98% inhibition



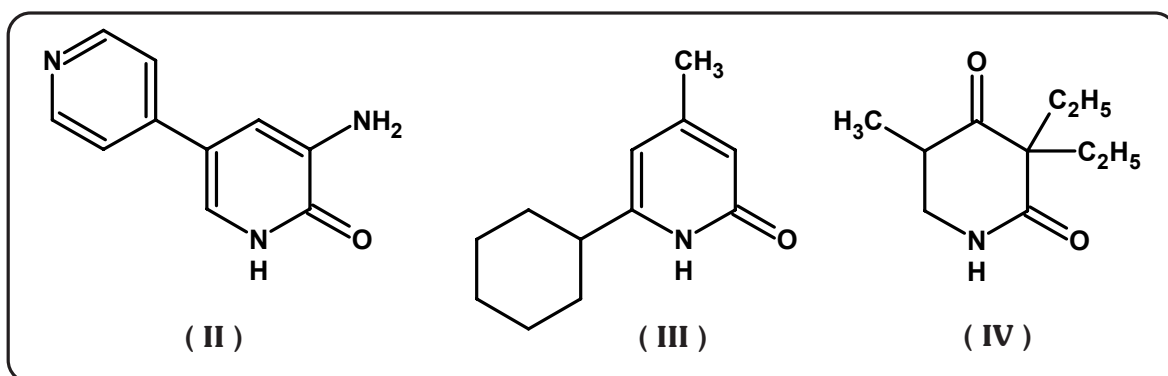
PART - III
STUDIES ON
CYANOPYRIDONES

INTRODUCTION

Biosignificant pyridones form a component in a number of useful drugs and are associated with many biological, pharmaceutical and therapeutic activities. Pyridones, with a carbonyl group at position 2(I) have been subject of extensive study in the recent years. Numerous reports have appeared in the literature which highlight their chemistry and use.



2-Pyridones are derivatives of pyridine with carbonyl group at 2-position(I), some 2-pyridones, which are pharmacologically important are as under: eg. amrinone(II), ciclopirox(III) and methylphylone (IV).

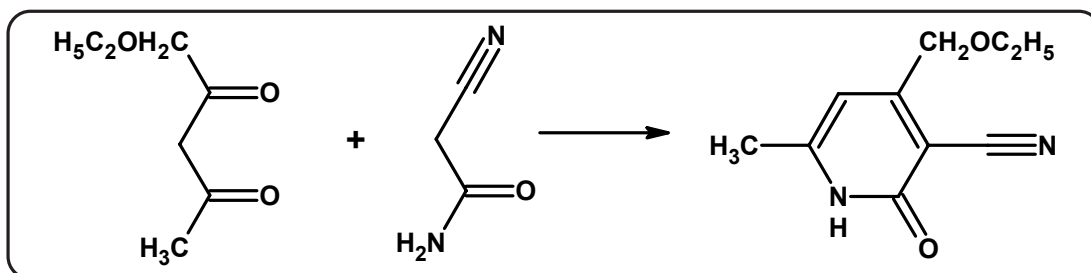


Synthetic pyridone derivatives contribute much to the searchable literature of pyridone derivatives in huge libraries owing to their wide applicability in different fields.

SYNTHETIC ASPECT :

Different methods for the preparation of 2-pyridones are as under.

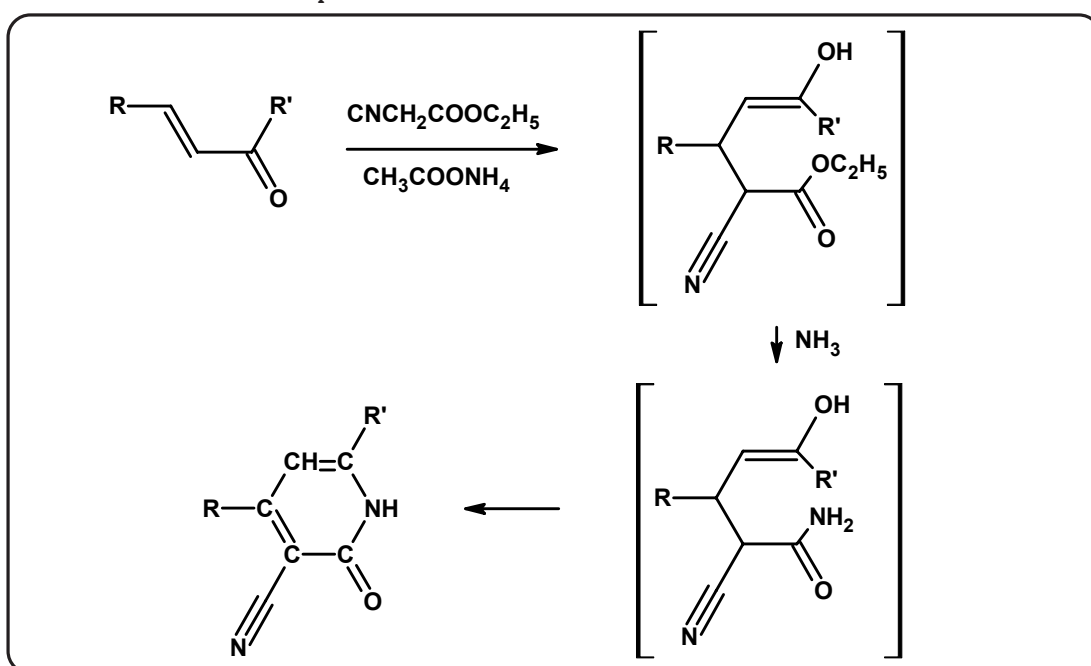
1. S. A. Harris and K. Falkers¹⁴⁸ have reported 3-cyano-2-pyridone by the condensation of cyanoacetamide with 1,3-diketone or β -ketoester.



2. M. A. Slwyter et al.¹⁴⁹ have prepared fused 2-pyridones.
3. G. Simchen and G. Entemann¹⁵⁰ have synthesised 2-pyridone in which the ring nitrogen comes from a nitrile group in acyclic precursor. Addition of HCl to the nitrile produces an imidoyl chloride which can cyclize.

MECHANISM

The addition reaction between ethylcyanoacetate and α,β -unsaturated ketone give cyanopyridone via Michael addition. Here, α,β -unsaturated compound is known as acceptor and active methylene group containing compound known as addender. It involves nucleophilic addition of carbanion to the C=C of the acceptor.

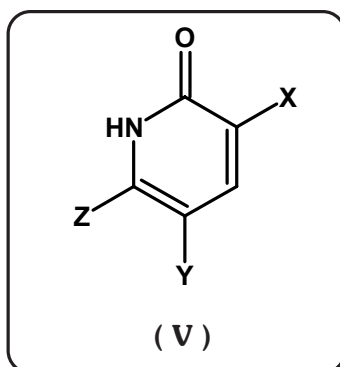


THERAPEUTIC IMPORTANCE

Literature survey reveals that various pyridones having potential of drugs are known to exhibit a broad spectrum of biological activities such as:

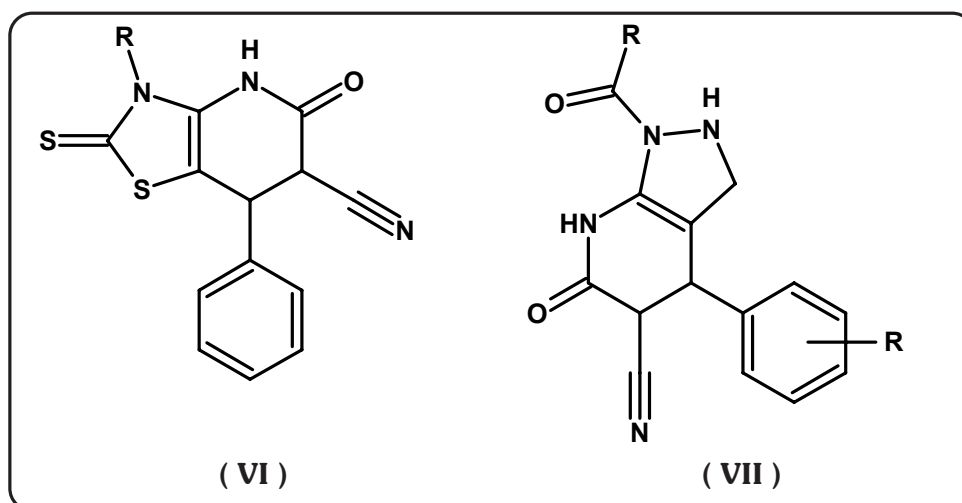
1. Anticancer¹⁵¹
2. Antimicrobial¹⁵²
3. Pesticidal¹⁵³
4. Angitensin II antagonistic¹⁵⁴
5. Herbicidal¹⁵⁵
6. Antiviral¹⁵⁶
7. Antibacterial¹⁵⁷

Peter et al.¹⁵⁸ have prepared pyridinylmethyl substituted pyridines and pyridones as angiotensin II antagonists. H. Poner¹⁵⁹ reported 2-pyridones as physiologically active compounds. Collins and co-workers¹⁶⁰ prepared heteroaryl pyridones as GABA α_2/α_3 ligands (V).

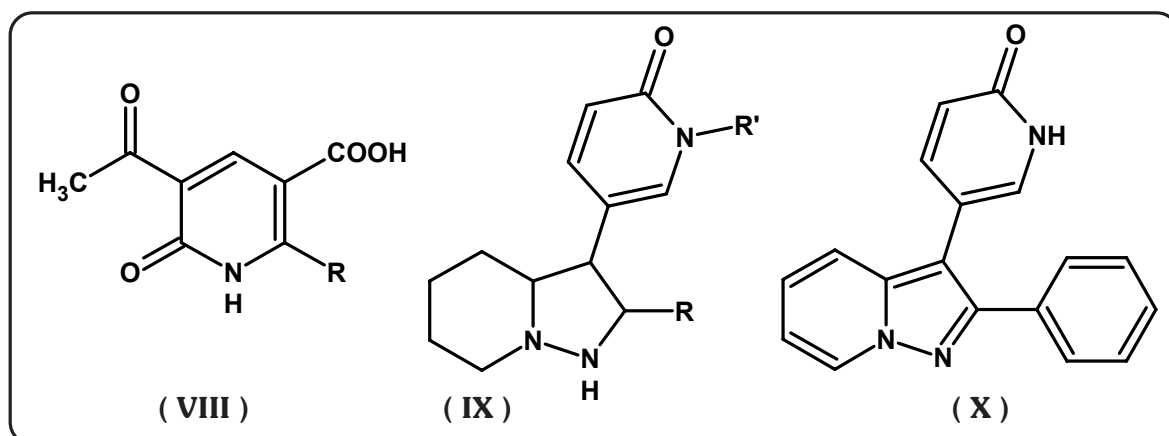


More over, several co-workers have pre-reported 2-pyridones as S_3 site of thrombin inhibitor¹⁶¹, herbicidal¹⁶², SH_2 domain inhibitor¹⁶³, antimicrobial¹⁶⁴, GABA-A recceptor¹⁶⁵ and antiinflammatory¹⁶⁶.

Mukhta Hussain Khan and co-workers^{167,168} have synthesised 2-pyridone derivatives (VI) and (VII) which possess insecticidal and pesticidal activity.



E. Amer¹⁶⁹ reported 3-cyano-2-pyridone derivatives, displaying high antimicrobial activity. Abou El-Fotooh et al.¹⁷⁰ have demonstrated pyridones as anticancer agent. F. Paala¹⁷¹ synthesised 2-pyridone derivatives (VIII), showing good cardiotoxic activity. Gulcan Ozturk et al.¹⁷² described 1,2,5-substituted-4-pyridone derivatives having analgesic and antiinflammatory activity. Tanaka A. and co-workers¹⁷³ reported 2-pyridone derivatives (IX and X) which showed adenosine antagonist activity and anticatalepsy activity.



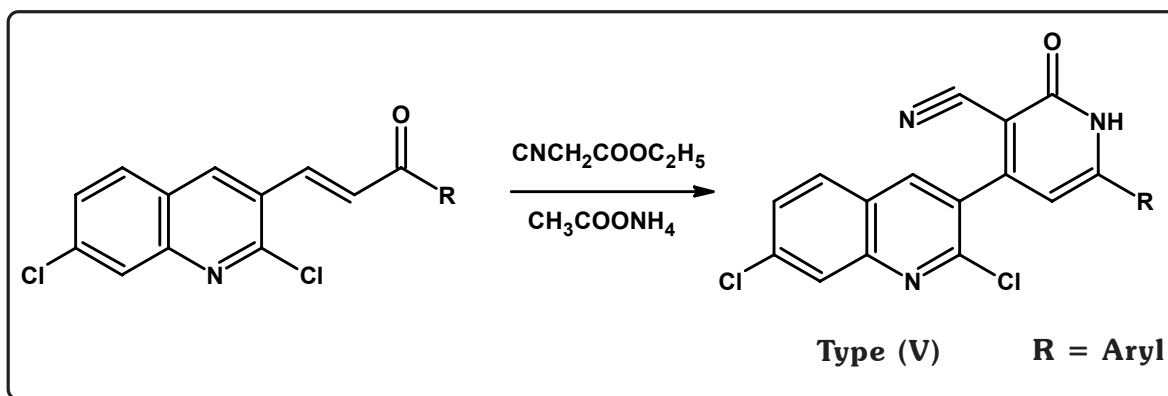
Thus, significant biological properties associated with pyridones, have aroused considerable interest to design the compounds with a view to get better drug potential, to study their pharmacological profile, which have been described as under.

SECTION-I : SYNTHESIS AND THERAPEUTIC EVALUATION OF 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-2-PYRIDONES

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-2-PYRIDONES

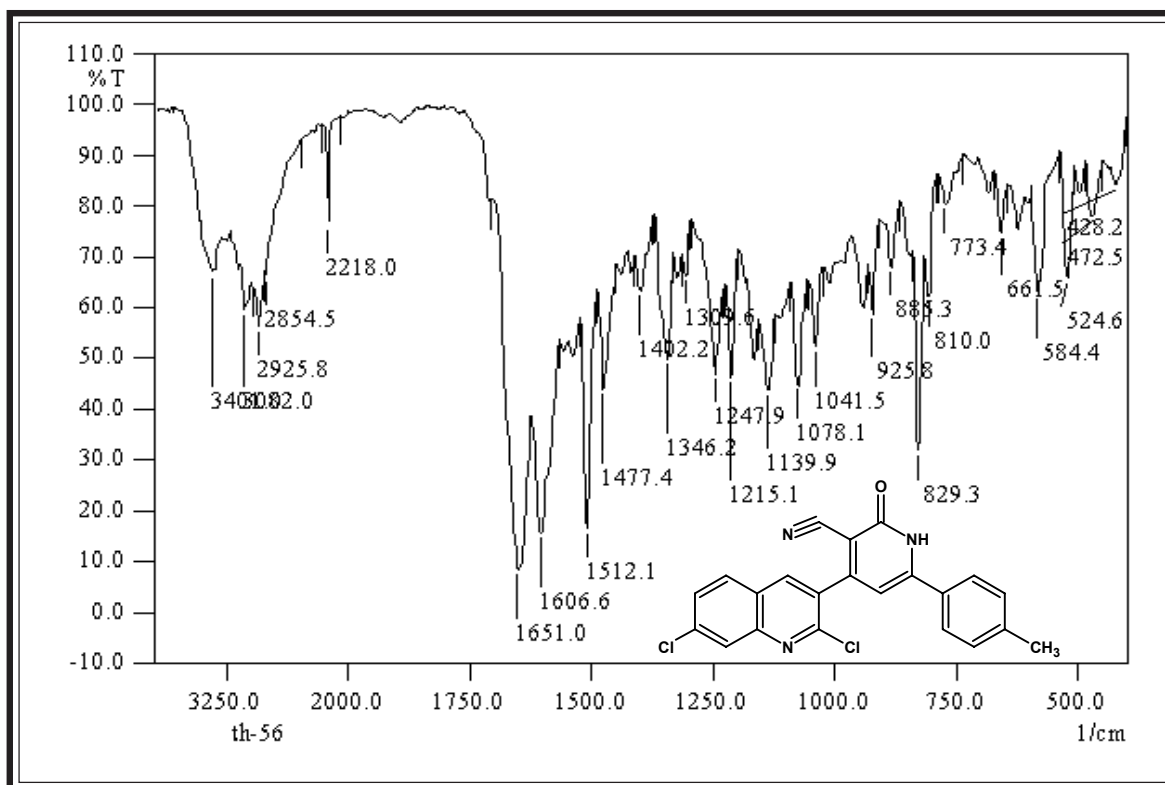
During the last few years, interest in the use of cyanopyridone derivatives in medicine and agriculture has greatly increased. To further assess the potential of such a class of compounds as antimicrobial agents, a series of cyanopyridone of type (V) have been synthesised by the condensation of chalcones of type(I) with ethyl cyanoacetate in presence of ammonium acetate as under.



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass-spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

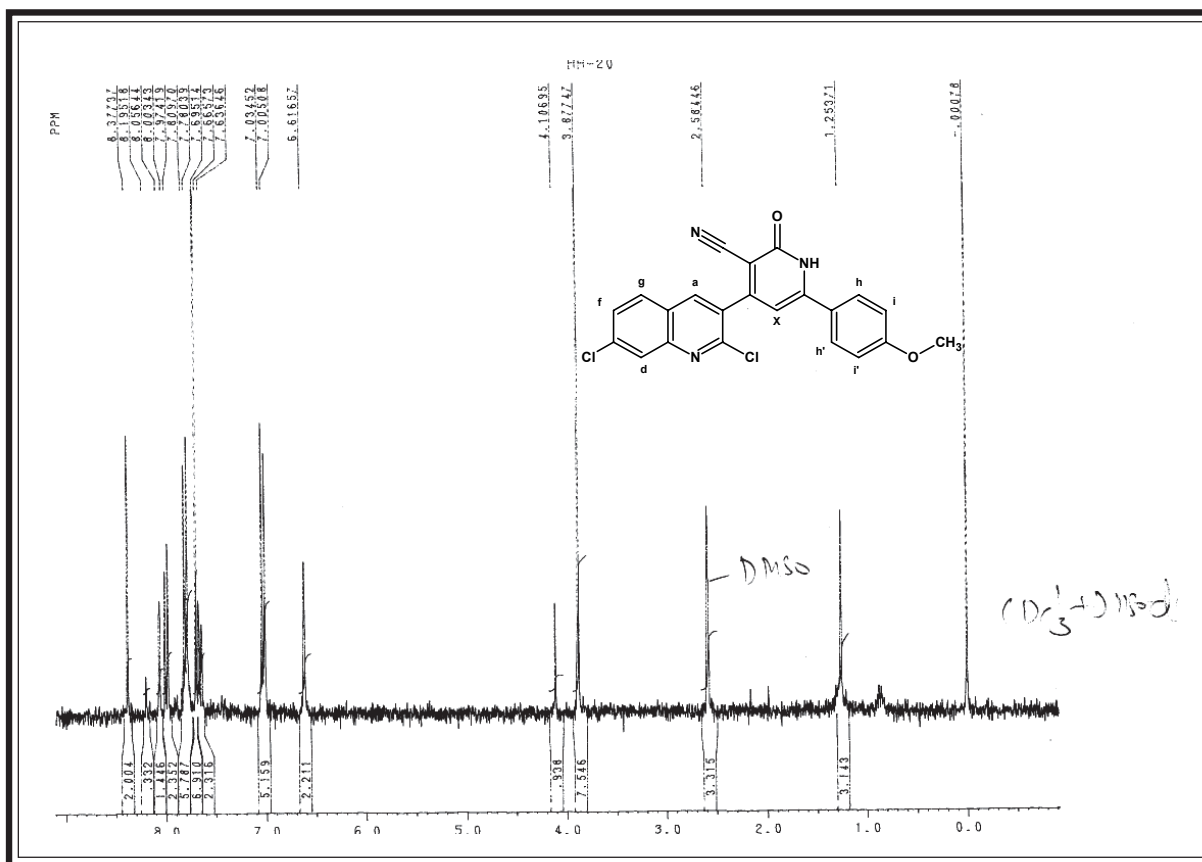
IR SPECTRAL STUDY OF 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-p-TOLYL-2-PYRIDONE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane -CH ₃	C - H str.	2925	2975-2950	419
	C - H def. (asym)	1477	1470-1435	"
Aromatic	C- H str.	3082	3090-3030	"
	C = C str.	1512	1550-1480	"
	C - H.i.p. def.	1139	1125-1090	420
	C - H. o.o.p. def.	829	835-810	"
Quinoline moiety	C = N str.	1606	1612-1593	419
	C - Cl str.	773	750-700	"
Pyridone ring	N - H str.	3401	3400-3250	"
	C \equiv N str.	2218	2250-2120	"
	C = O str	1651	1780-1655	"

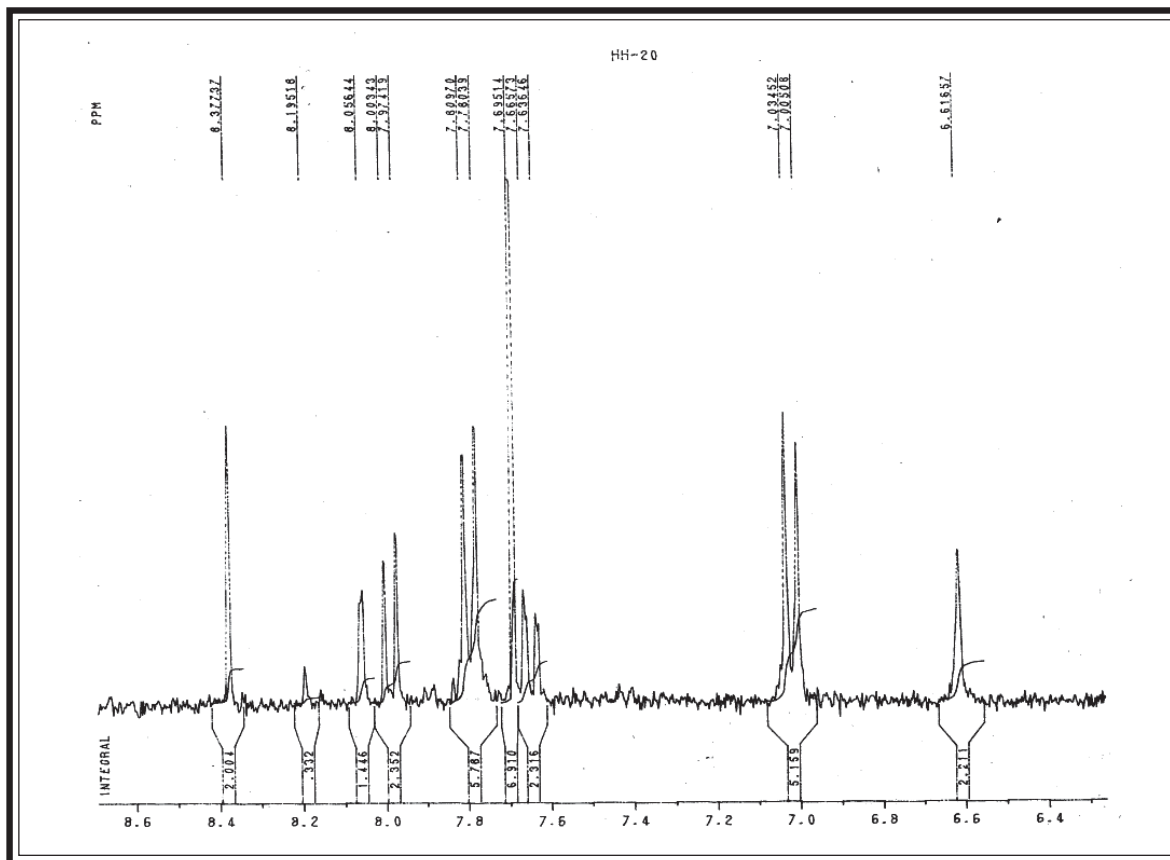
PMR SPECTRAL STUDY OF 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-(p-METHOXYPHENYL)-2-PYRIDONES



Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	3.88	3H	singlet	-	Ar-OCH ₃
2.	6.62	1H	singlet	-	Ar-HX
3.	7.00-7.03	2H	doublet	J _{in} 8.83	Ar-Hi,i'
4.	7.64-7.67	1H	doublet doublet	J _{fg} 8.78	Ar-Hf
5.	7.78-7.81	2H	doublet	J _{hi} 8.79	Ar-Hh,h'
6.	7.97-8.00	1H	doublet	J _{gf} 8.77	Ar-Hg
7.	8.05	1H	singlet	-	Ar-Hd
8.	8.37	1H	singlet	-	Ar-Ha

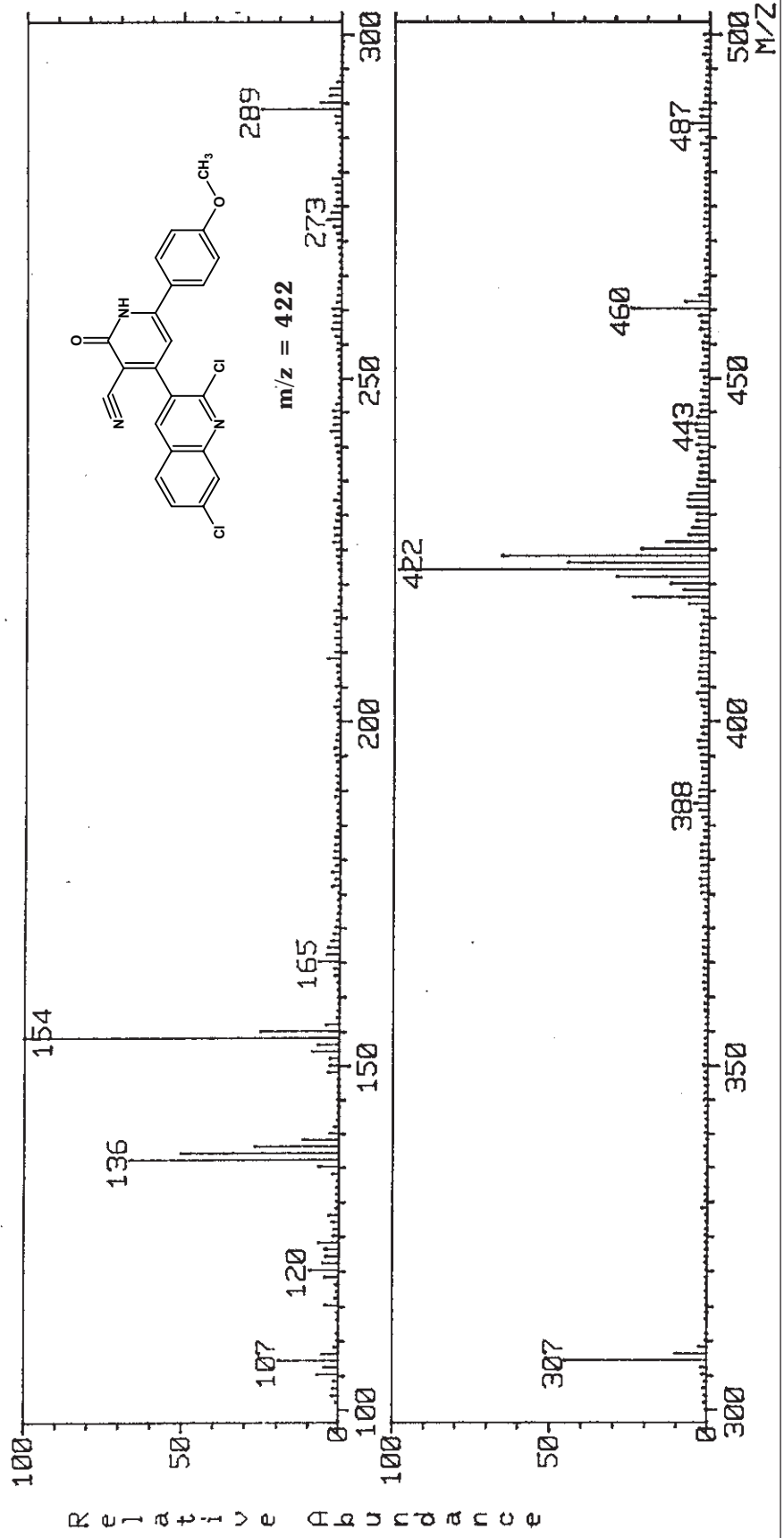
EXPANDED AROMATIC REGION


IR SPECTRAL DATA OF 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-2-PYRIDONES

 Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	-C≡N	C=O
5a	C_6H_5	2218	1649
5b	$3\text{-NH}_2\text{-C}_6\text{H}_4$	2218	1651
5c	$4\text{-NH}_2\text{-C}_6\text{H}_4$	2210	1649
5d	$4\text{-Br-C}_6\text{H}_4$	2216	1649
5e	$4\text{-Cl-C}_6\text{H}_4$	2220	1649
5f	$2,6\text{-(OH)}_2\text{-C}_6\text{H}_3$	2218	1651
5g	$4\text{-F-C}_6\text{H}_4$	2218	1651
5h	$2\text{-OH-C}_6\text{H}_4$	2216	1651
5i	$4\text{-OH-C}_6\text{H}_4$	2218	1649
5j	$4\text{-OCH}_3\text{-C}_6\text{H}_4$	2210	1649
5k	$3\text{-CH}_3\text{-C}_6\text{H}_4$	2218	1651
5l	$3\text{-NO}_2\text{-C}_6\text{H}_4$	2210	1649
5m	$4\text{-NO}_2\text{-C}_6\text{H}_4$	2215	1652

MASS SPECTRUM Data File: 3EJL25W 25-JUL- 3 12:29
Sample: HHA-8 DR NA CHAUHAN, RAJKOT #6204
RT 0.24" FAB(Pos.) GC 1.4c BP: m/z 154.0000 Int. 38.1446 Lv. 0.00
Scan# (3 to 4)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-2-PYRIDONES

[A] Synthesis of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones

See Part (I), Section-I(C).

[B] Synthesis of 1,2-Dihydro-3-cyno-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl-2-pyridones

1-p-Tolyl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-one (3.42g 0.01mol) was dissolved in ethanol (50 ml). Add ethylcyanoacetate (1.13g. 0.01mol) and ammonium acetate (9.24g. 0.12 mol) to it. The mixture was then heated under reflux for 8 hrs. The product separated was filtered and recrystallised from methanol. Yield. 57%, m.p. 224°C; Anal. calcd for C₂₂H₁₃N₃Cl₂O : C, 65.04%; H, 3.23%; N, 10.34%; Found : C, 65.07%; H, 3.25%; N, 9.93%.

TLC solvent system: Acetone: Benzene (0.5: 9.5) Visualizing Agent : Iodine.

Similarly, other 1,2-Dihydro-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl-2-pyridones were prepared. The physical data along with infra red spectral data are recorded in Table No. 3

[C] Therapeutic evaluation of 1,2-Dihydro-3-cyano-4-(2',7'-dichloroquinolin-3'-yl)-6-aryl-2-pyridones

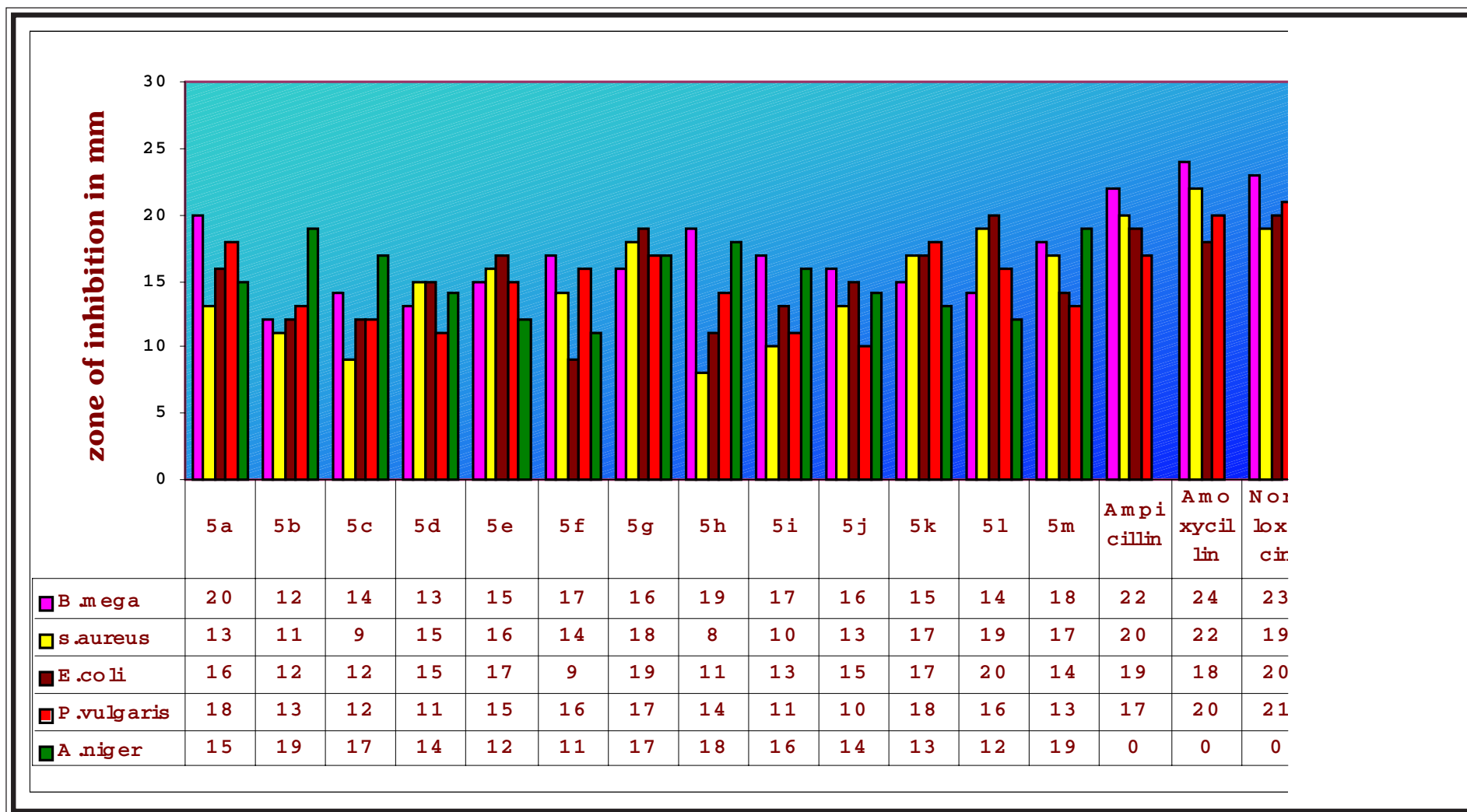
Antimicrobial testing was carried out as described in Part-I, Section-I(D). The zones of inhibition of the felt solution are recorded in Graphical Chart No. 5.

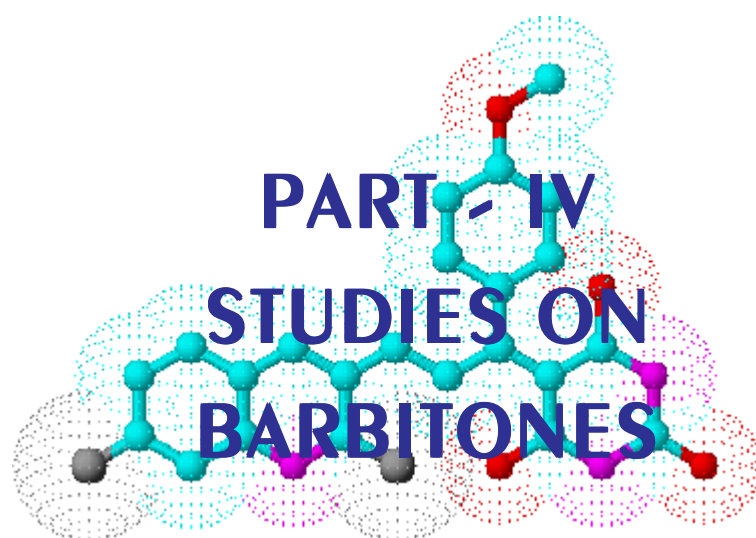
TABLE NO. 3 : PHYSICAL CONSTANTS OF 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-2-PYRIDONES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
5a	C ₆ H ₅ -	C ₂₁ H ₁₁ N ₃ Cl ₂ O	392	124	0.534	63	10.71	10.29
5b	3-NH ₂ -C ₆ H ₄ -	C ₂₁ H ₁₂ N ₄ Cl ₂ O	407	158	0.405	61	13.76	13.35
5c	4-NH ₂ -C ₆ H ₄ -	C ₂₁ H ₁₂ N ₄ Cl ₂ O	407	204	0.435	59	13.76	13.99
5d	4-Br-C ₆ H ₄ -	C ₂₁ H ₁₀ N ₃ BrCl ₂ O	471	172	0.585	58	8.92	8.54
5e	4-Cl-C ₆ H ₄ -	C ₂₁ H ₁₀ N ₃ Cl ₃ O	426.5	170	0.609	59	9.85	9.57
5f	2,6-(OH) ₂ -C ₆ H ₃ -	C ₂₁ H ₁₁ N ₃ Cl ₂ O ₃	424	168	0.659	60	9.90	9.63
5g	4-F-C ₆ H ₄ -	C ₂₁ H ₁₀ N ₃ Cl ₂ FO	410	148	0.714	55	10.24	10.51
5h	2-OH-C ₆ H ₄ -	C ₂₁ H ₁₁ N ₃ Cl ₂ O ₂	408	202	0.650	54	10.29	10.59
5i	4-OH-C ₆ H ₄ -	C ₂₂ H ₁₁ N ₃ Cl ₂ O ₂	408	150	0.473	56	10.29	10.05
5j	4-OCH ₃ -C ₆ H ₄ -	C ₂₂ H ₁₃ N ₃ Cl ₂ O ₂	422	176	0.512	58	9.95	9.69
5k	4-CH ₃ -C ₆ H ₄ -	C ₂₂ H ₁₃ N ₃ Cl ₂ O	406	224	0.591	57	10.34	9.93
5l	3-NO ₂ -C ₆ H ₄ -	C ₂₁ H ₁₀ N ₄ Cl ₂ O ₃	437	184	0.481	60	12.81	12.69
5m	4-NO ₂ -C ₆ H ₄ -	C ₂₁ H ₁₀ N ₄ Cl ₂ O ₃	437	196	0.541	62	12.81	12.97

TLC Solvent System : Acetone : Benzene (0.5 : 9.5).

GRAPHICAL CHART NO. 5 : 1,2-DIHYDRO-3-CYANO-4-(2',7'-DICHLOROQUINOLIN-3'-YL)-6-ARYL-2-PYRIDONES

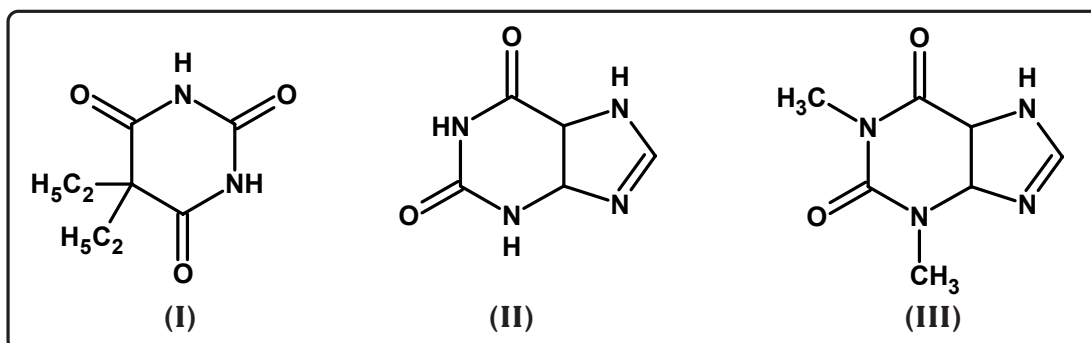




INTRODUCTION

Barbituric acid derivatives have gained prominence because of their potential pharmaceutical values. Many barbituric acid derivatives play vital role in many physiological action. Most important is the effect of barbiturates on the central nervous system. There are more than 40 synthetic drugs bearing barbituric acid, in use recently. They possess diverse type of biological properties including hypnotic, sedatives, anticonvulsant, cardiovascular etc. The first member of hypnotic drugs series was barbital (I).

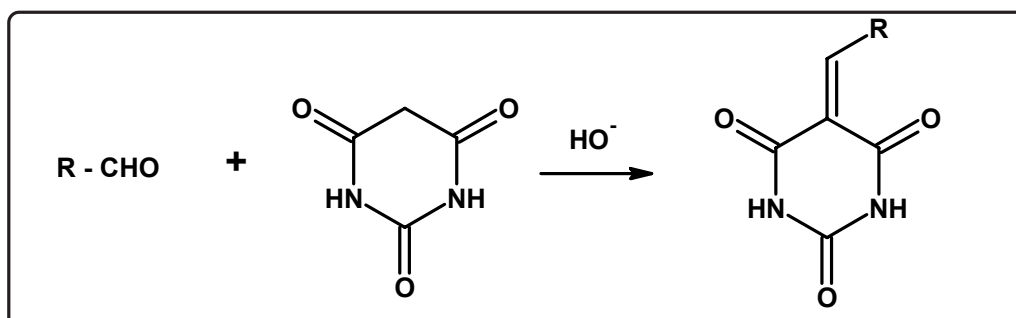
Barbituric acid ring system has been found in many natural products like alkaloids, which includes Xanthine (II) and Theophylline (III) are constituents of tea leaves. Theobromine is found in cocoa beans.



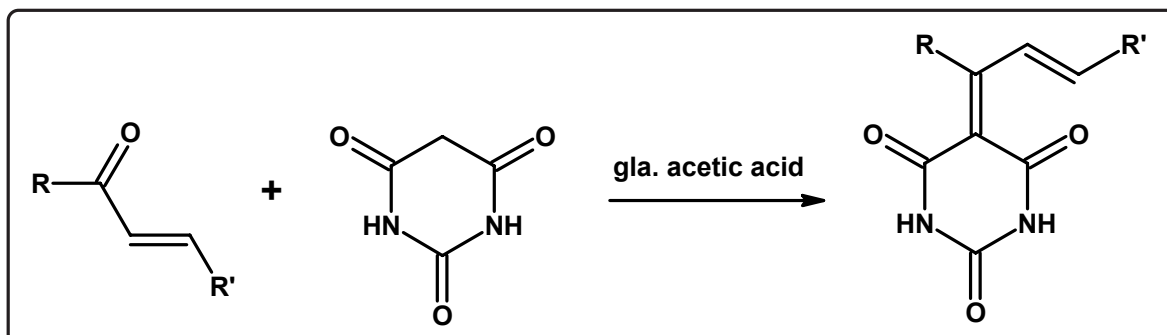
SYNTHETIC ASPECT

Different methods are used for the preparation of barbitones in literature^{174,175}.

1. Cao-Yun Weu et al.¹⁷⁶ have prepared barbituric acid derivatives by the reaction of different aldehydes with barbituric acid in basic media.



2. M. R. Mahmoud et al.¹⁷⁷ have synthesised barbituric acid derivatives from chalcone.

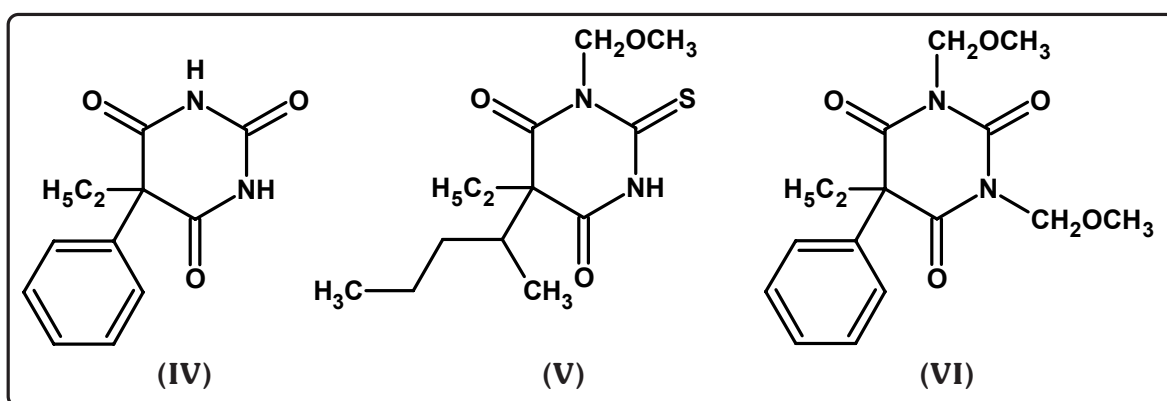


3. Ogus Funda et al.¹⁷⁸ have synthesised barbiturates by the reaction between acetone and barbituric acid.

THERAPEUTIC IMPORTANCE

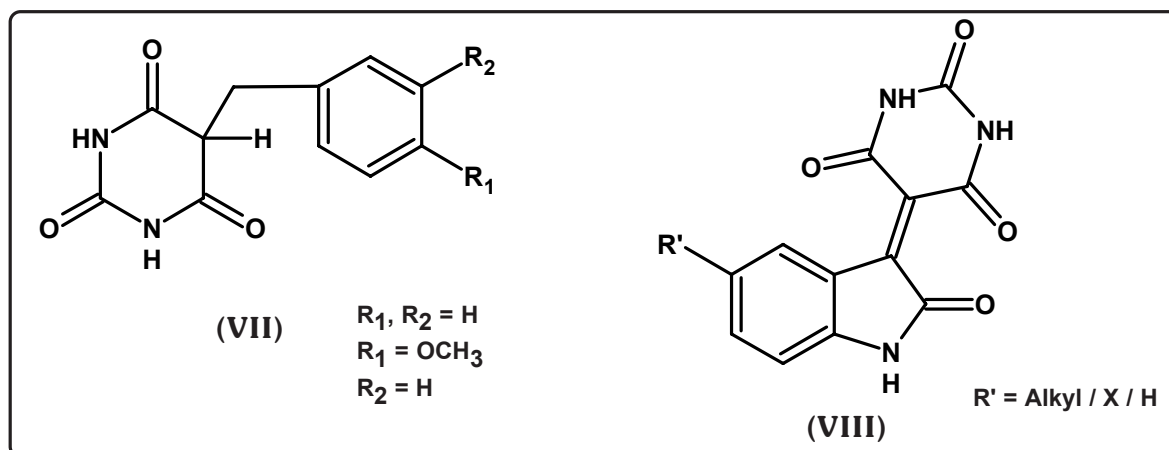
Barbituric acid derivatives demonstrate a very broad spectrum of biological activity, because of their structural relationship with nucleic acids, viz. uracil, thymine and cytosine. Perhaps barbituric acid derivatives are most widely used pyrimidines in medicinal chemistry.

Phenobarbitone (IV) possess sedative and hypnotic activities, thiopentone (V) is a useful local anaesthetic while Eterobarb (VI) is an anticonvulsant drug.



Carbubarb¹⁷⁹ is a barbituric acid derivative, which is used as a veterinary anaesthetics. Some isoxazole pyrimidine derivatives have been studied because of their potential as as pesticidal^{180,181} activity. Some barbiturates showing cardiovascular¹⁸²⁻¹⁸⁴ and analgesic and antiinflammatory activities¹⁸⁵ have been reported.

Ulf Wellmar et al.¹⁸⁶ have synthesised some uracil derivatives and screened for antiviral activity^{187,188}. Raymond et al.¹⁸⁹ investigated some barbiturates (VII), showing anticancer activity while Mahmoud et al.¹⁹⁰ reported their antimicrobial activity.



Some isoxazolo pyrimidine derivatives have been extensively studied and reported as antagonist¹⁹¹ and antitumor¹⁹² agents. Agricultural activity of 5-(3-benzylthiazolidine-2-ylidene)-1,3-dimethyl hexahydro pyrimidine-2,4,6-trione was reported by Wolf-Gang et al.¹⁹³ Harbicial and insecticidal activity of barbiturates was documented by Andre Roland and co-workers¹⁹⁴. Omar M. T.¹⁹⁵ have showed barbitone derivatives demonstrating antimicrobial activity. Sakai and co-workers¹⁹⁶ has synthesised some new barbitones which are assessed for bone and cartilage diseases. R. T. Pardasani and co-workers¹⁹⁷ have described some new isatylidene barbitones (VIII) and tested their antibacterial activity.

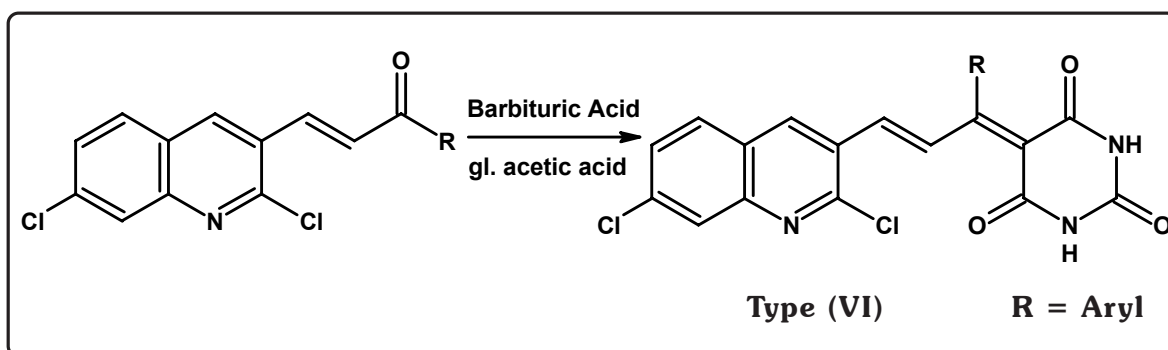
Vital contribution of barbituric acid ring system to the medicinal chemistry as an active constituent of hypnotics and sedatives made chemists to explore for its other derivatives as therapeutic agents. Accordingly several derivatives of barbituric acid have been designed as under.

SECTION - I : SYNTHESIS AND THERAPEUTIC EVALUATION OF [1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENYLIDENE]-5-BARBITURIC ACIDS

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF [1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENYLIDENE]-5-BARBITURIC ACIDS

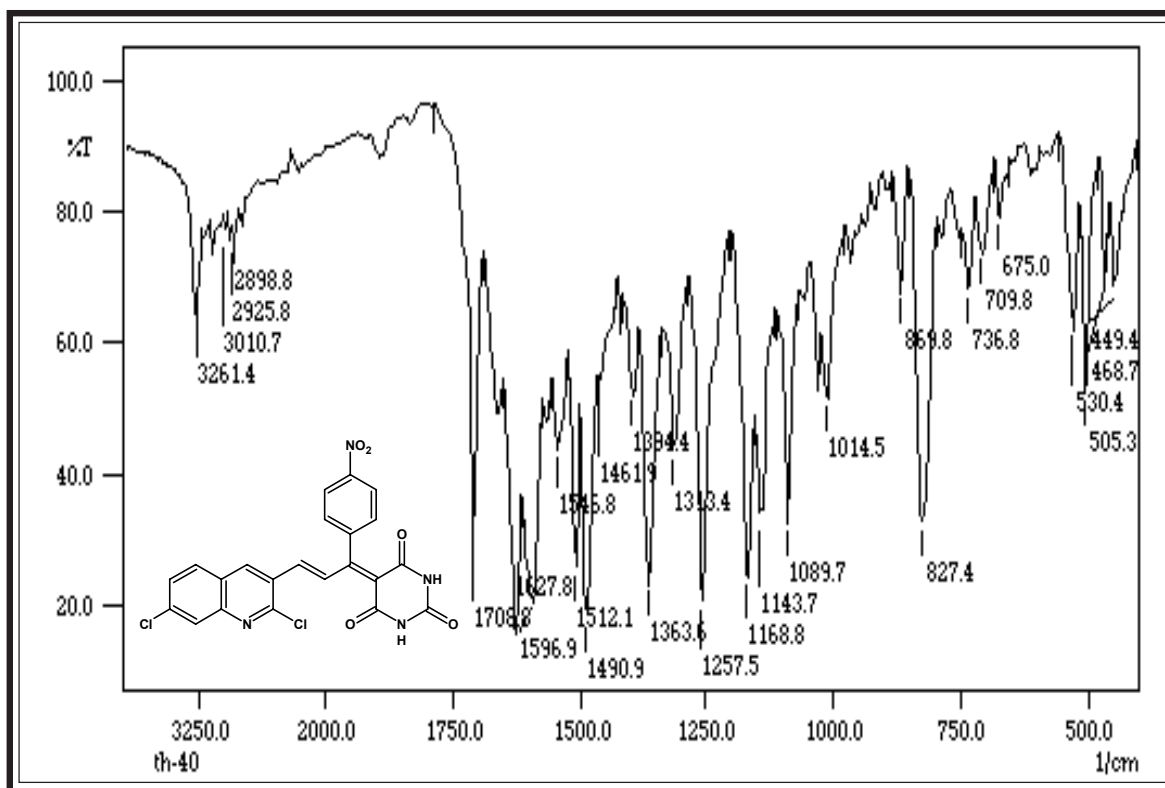
Barbiturates are known to play an important role in medicinal chemistry because of their effect on central nervous system. The first member of hypnotic drugs series was barbital. Considering this background, some new barbituric acid derivatives of type (VI) were prepared by the condensation of compounds of type (I) with barbituric acid in glacial acetic acid.



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

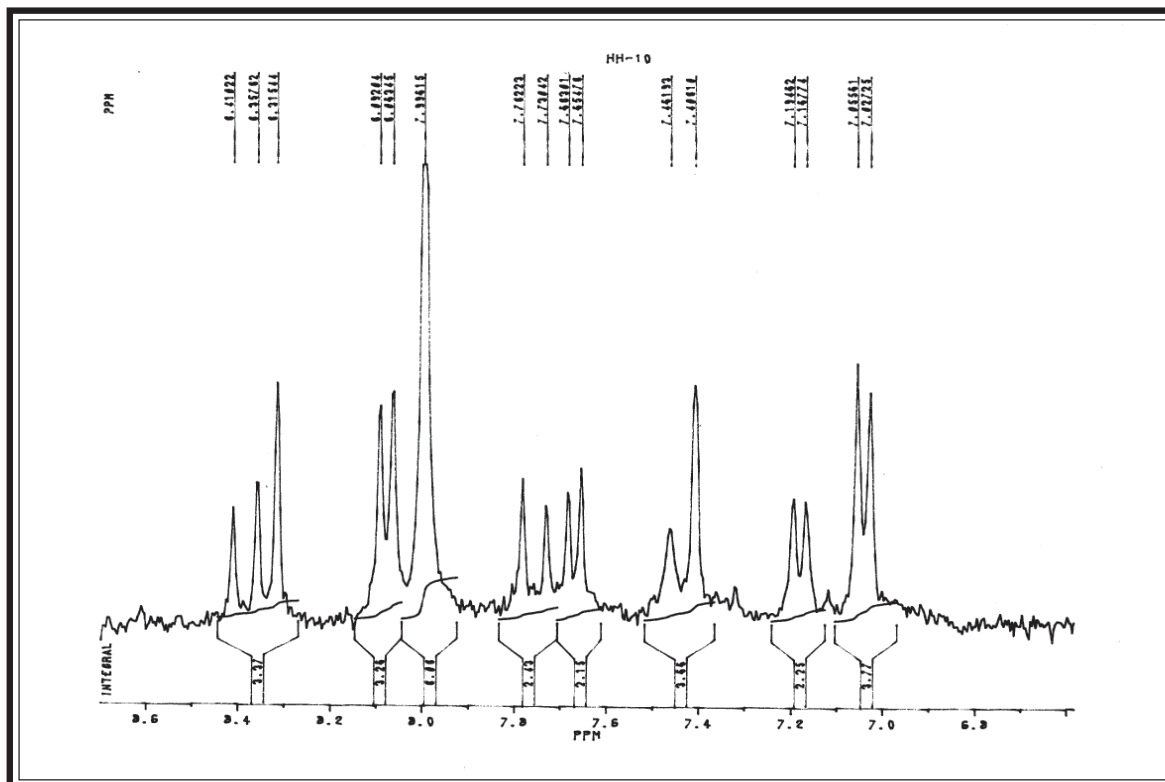
IR SPECTRAL STUDY OF [1-p-NITROPHENYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENYLIDENE]-5-BARBITURIC ACID



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm⁻¹ (KBr disc.)

Type	Vibration mode	Frequency in cm ⁻¹		Ref.
		Observed	Reported	
Alkane -CH ₃	C - H str. (asym)	2925	2975-2950	419
	C - H def. (sym)	1490	1470-1435	"
Aromatic	C- H str.	3010	3080-3030	"
	C = C str.	1512	1520-1480	"
	C - H.i.p. def.	1143	1150-1090	420
	C - H. o.o.p. def.	827	835-810	"
Quinoline moiety	C = N str.	1627	1650-1590	419
	C - Cl str.	736	750-700	"
Barbitone	N - H str.	3261	3350-3140	422
	C = O str.	1708	1750-1700	"
Vinyl	HC= CH str.	3010	3050-3000	"
	N - O str.	827	850-810	"

EXPANDED AROMATIC REGION

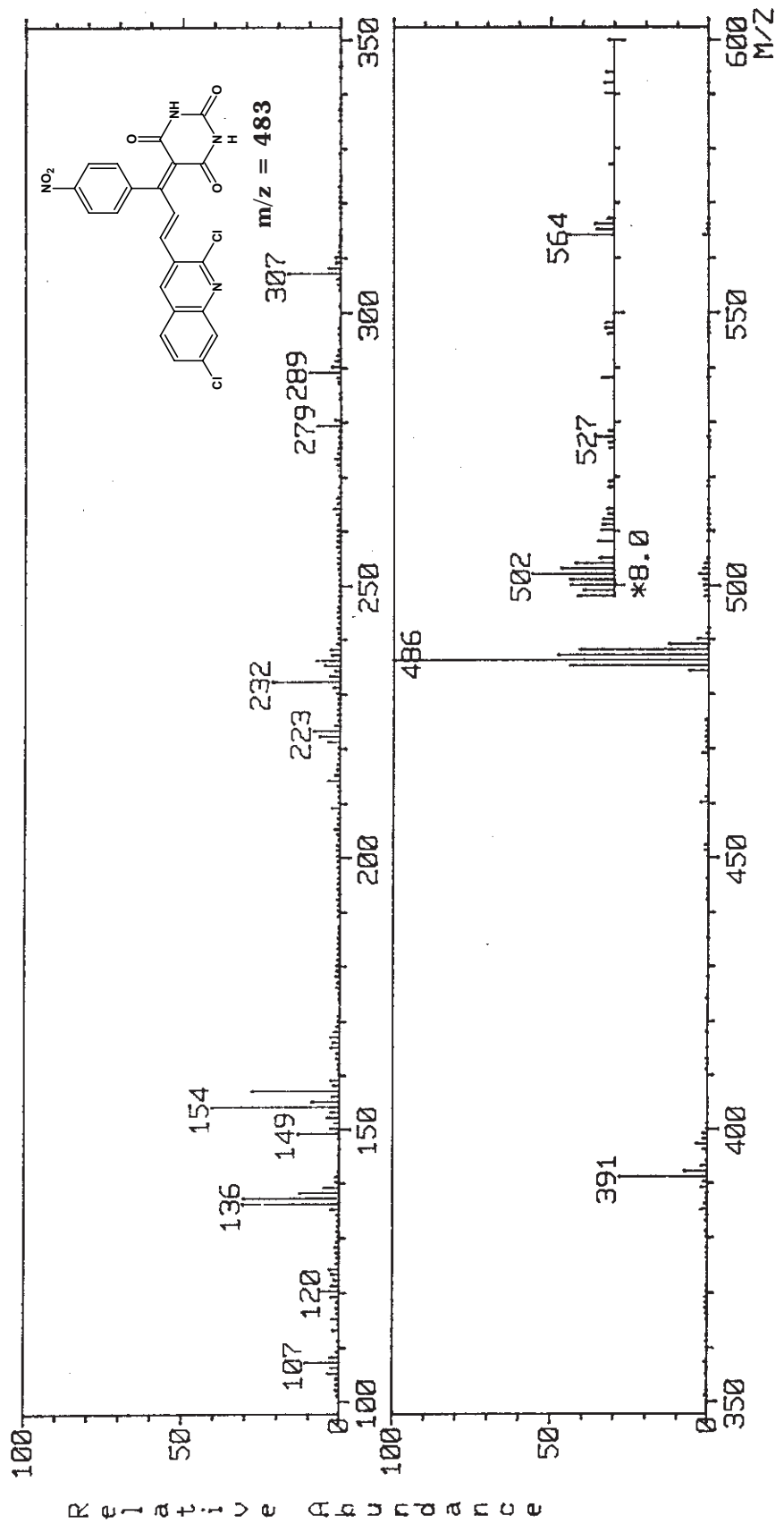


IR SPECTRAL DATA OF [1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENYLIDENE]-5-BARBITURIC ACIDS

Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	C=O str.
6a	C_6H_5	1706
6b	$3\text{-NH}_2\text{-C}_6\text{H}_4$	1708
6c	$4\text{-NH}_2\text{-C}_6\text{H}_4$	1710
6d	$4\text{-Br-C}_6\text{H}_4$	1706
6e	$4\text{-Cl-C}_6\text{H}_4$	1710
6f	$2,6\text{-(OH)}_2\text{-C}_6\text{H}_3$	1700
6g	$4\text{-F-C}_6\text{H}_4$	1710
6h	$2\text{-OH-C}_6\text{H}_4$	1708
6i	$4\text{-OH-C}_6\text{H}_4$	1706
6j	$4\text{-OCH}_3\text{-C}_6\text{H}_4$	1710
6k	$4\text{-CH}_3\text{-C}_6\text{H}_4$	1708
6l	$4\text{-NO}_2\text{-C}_6\text{H}_4$	1710
6m	$4\text{-NO}_2\text{-C}_6\text{H}_4$	1708

MASS SPECTRUM Data File: 3EJL25Y 25-JUL- 3 12:37
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 RT 0.24" FAB(Pos.) GC 1.4c BP: m/z 486.0000 Int. 43.0831 Lv 0.00
 Scan# (3 to 4)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF [1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENYLIDENE]-5-BARBITURIC ACIDS

[A] Synthesis of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones

See Part - (I), Section - I (C).

[B] Synthesis of [1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propenylidene]-5-barbituric acids

A mixture of 1-p-tolyl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones (3.42g. 0.01 mol) and barbituric acid (1.28 g., 0.01 mol) in glacial acetic acid was refluxed for 12 hrs. The contents were poured into ice and product was isolated, crystallised from ethanol. Yield 54%, m.p. 270°C, Anal. Calcd for $C_{23}H_{15}N_3Cl_2O_3$ required : C, 60.81%; H, 3.77%; N, 9.25%; found : C, 60.85%; H, 3.75%; N, 9.53%.

TLC Solvent system : Acetone : Benzene (1:9) visulising Agent : Iodine.

Similarly, other [1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propenylidene]-5-barbituric acids were prepared. The physical data along with infra red spectral data are recorded in Table No. 4.

[C] Therapeutic evaluation of [1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propenylidene]-5-barbituric acids

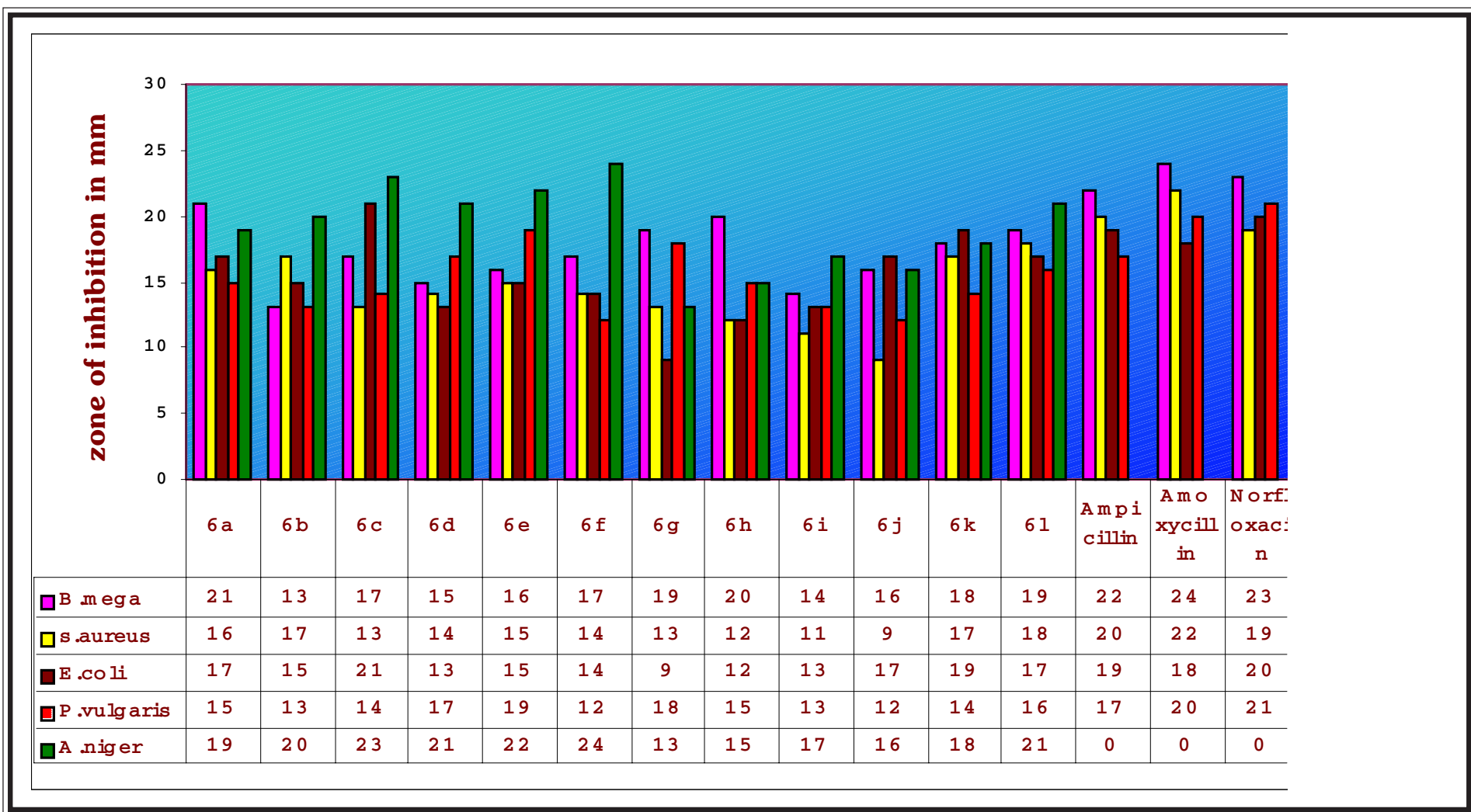
Antimicrobial testing was carried out as described in Part-I, Section-I (D). The zones of inhibition of the test solutions are recorded in Graphical Chart No. 6.

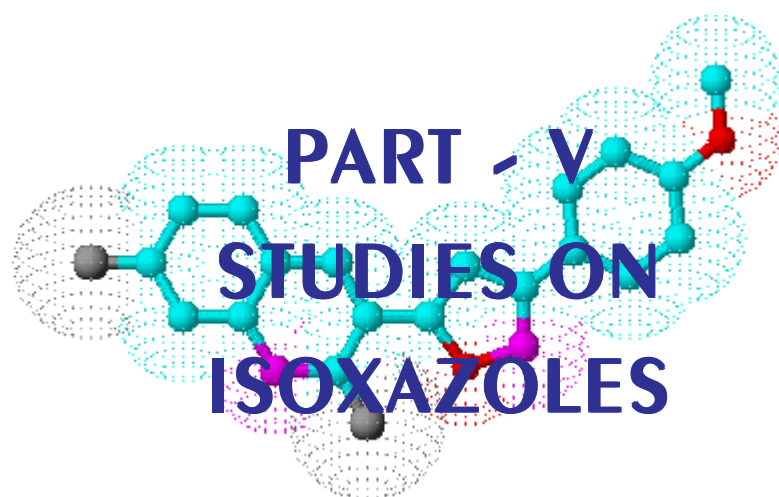
TABLE NO. 4 : PHYSICAL CONSTANTS OF [1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENYLIDENE]-5-BARBITURIC ACIDS

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
6a	C ₆ H ₅ -	C ₂₂ H ₁₃ N ₃ Cl ₂ O ₃	438	212	0.686	53	9.54	9.19
6b	3-NH ₂ -C ₆ H ₄ -	C ₂₂ H ₁₄ N ₄ Cl ₂ O ₃	453	234	0.652	51	12.31	12.59
6c	4-NH ₂ -C ₆ H ₄ -	C ₂₂ H ₁₄ N ₄ Cl ₂ O ₃	453	214	0.565	54	12.31	11.98
6d	4-Br-C ₆ H ₄ -	C ₂₂ H ₁₂ N ₃ BrCl ₂ O ₃	517	298	0.489	56	8.09	8.23
6e	4-Cl-C ₆ H ₄ -	C ₂₂ H ₁₂ N ₃ Cl ₃ O ₃	472.5	286	0.580	57	8.85	8.57
6f	2,6-(OH) ₂ -C ₆ H ₃ -	C ₂₂ H ₁₃ N ₃ Cl ₂ O ₅	470	280	0.691	56	8.90	8.64
6g	4-F-C ₆ H ₄ -	C ₂₂ H ₁₂ N ₃ Cl ₂ FO ₃	456	276	0.666	55	9.17	9.43
6f	2-OH-C ₆ H ₄ -	C ₂₂ H ₁₃ N ₃ Cl ₂ O ₄	454	274	0.613	59	9.21	9.39
6h	4-OH-C ₆ H ₄ -	C ₂₂ H ₁₃ N ₃ Cl ₂ O ₄	454	266	0.602	57	9.21	9.51
6j	4-OCH ₃ -C ₆ H ₄ -	C ₂₃ H ₁₅ N ₃ Cl ₂ O ₄	468	288	0.549	58	8.93	8.69
6k	4-CH ₃ -C ₆ H ₄ -	C ₂₃ H ₁₅ N ₃ Cl ₂ O ₃	452	270	0.586	54	9.25	9.53
6l	3-NO ₂ -C ₆ H ₄ -	C ₂₂ H ₁₂ N ₄ Cl ₂ O ₅	483	296	0.494	53	11.55	11.80
6m	4-NO ₂ -C ₆ H ₄ -	C ₂₂ H ₁₂ N ₄ Cl ₂ O ₅	483	290	0.439	52	11.55	11.83

TLC Solvent System : Acetone : Benzene (1 : 9).

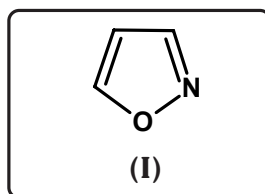
GRAPHICAL CHART NO. 6 : [1-ARYL-3-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-PROPENYLIDENE]-5-BARBITURIC ACIDS





INTRODUCTION

Issoxazoles offer a wide range of applications in medicinal field, as they are associated with a broad spectrum of biological activity. The investigation of isoxazole was made in 1888, by Claisen, who named isoxazole (I) for a product obtained by the reaction of 1,3-diketone with hydroxylamine¹⁹⁸. There after, Claisen and his students form a strong foundation for the chemistry of isoxazoles. Isoxazole possess typical properties of an aromatic system but under certain reaction conditions, particularly in reducing or basic media, it becomes highly labile.

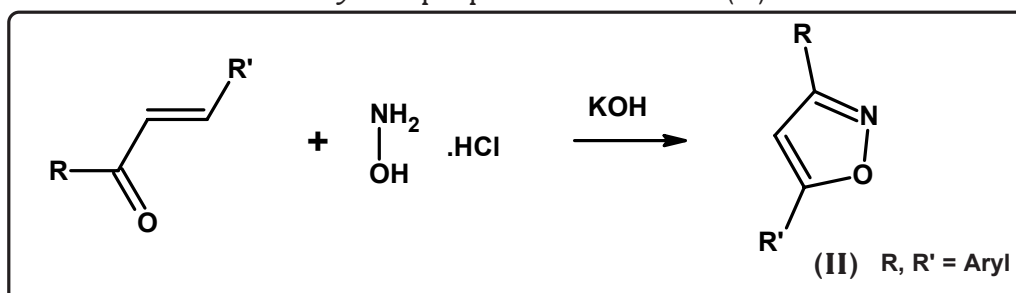


A huge role in the chemistry of isoxazoles was played by Quelico in 1946, when he began to study the formation of isoxazoles from nitrile N-oxides and unsaturated compounds¹⁹⁹.

SYNTHETIC ASPECT

The construction of isoxazole ring was carried out by the reaction of hydroxylamine with α,β -unsaturated carbonyl compounds, which results in an oxime, which later on undergo cyclisation.

(1) Fanshawe and Crawley²⁰⁰ prepared isoxazoles (II) from chalcones.



- (2) Substituted isoxazoles were prepared by the reaction between hydroxylamine hydrochloride and chalcone dibromides²⁰¹.
- (3) J. F. Hansen and S. A. Strong²⁰² isolated isoxazoles from α,β -unsaturated ketones and N-bromosuccinamide.
- (4) D'Alcontres and G. Ae Gamco²⁰³ synthesised isoxazoles from benzonitrile N-oxide and unsaturated aldehydes.

THERAPEUTIC IMPORTANCE

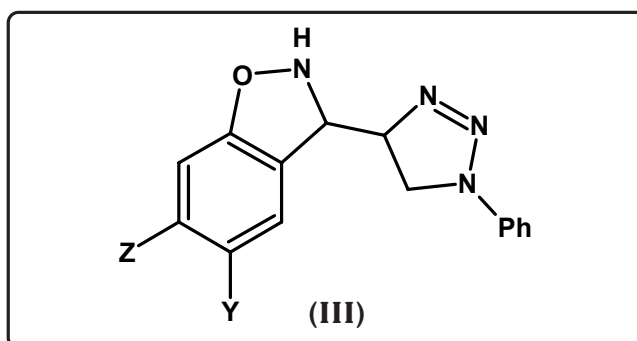
Isoxazoles have various medicinal applications such as:

- (1) Antiinflammatory²⁰⁴⁻²⁰⁷
- (2) Anticonvulsant^{208,209}
- (3) Muscle Relaxant^{210,211}
- (4) Antipyretic²¹²
- (5) Anticholesteremic²¹³
- (6) Antibacterial²¹⁴⁻²¹⁶
- (7) Antidiabetic²¹⁷
- (8) Nematocidal²¹⁸
- (9) Fungicidal^{219,220}
- (10) Antiviral²²¹
- (11) Herbicidal^{222, 223}
- (12) Anthelmintics²²⁴
- (13) Antileukemic²²⁵
- (14) Antitumor²²⁶
- (15) Hypoglycemic²²⁷
- (16) Analgesic²²⁸

Isoxazoles bearing quinazolone moiety were prepared by Maggio et al.²²⁹ and tested for their analgesic and antiinflammatory activities. Some of them had a very low ulcerogenic effect. C. B. Xue and co-worker²³⁰ reported the replacement of the benzamide in XUO57 (potent inhibitor) with an isoxazole carboxamide resulted in significant improvement *in vitro* potency. Isoxazoles

having pesticidal activity have been synthesised by Masui et al.²³¹ Some excellent herbicidal results are obtained by Reddy and co-workers²³². Nyitrai et al.²³³ have reported remarkable anxiolytic and antihypertensive effect of some new isoxazole derivatives. Mishra and co-workers²³⁴ have synthesised and reported isoxazoles as useful agents for analgesic and antiinflammatory activities.

Sezer Ozkan et al.²³⁵ have prepared 3-(1-phenyl-1,2,3-triazol-4-yl)-benzisoxazoles (III) and studied their insecticidal activity.

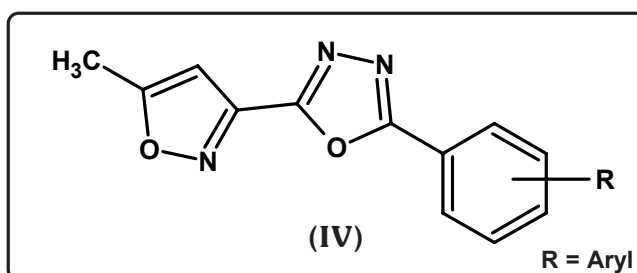


Some isoxazoles are found to possess herbicidal^{236,237}, potential antiinflammatory^{238,239} antimicrobial,^{240,241} estrogen receptor modulators²⁴² and inhibitor of p38 MAP kinase activities²⁴³. R. Ulrich et al.²⁴⁴ have synthesised some new isoxazole derivatives and reported their adrenergic antagonist activity.

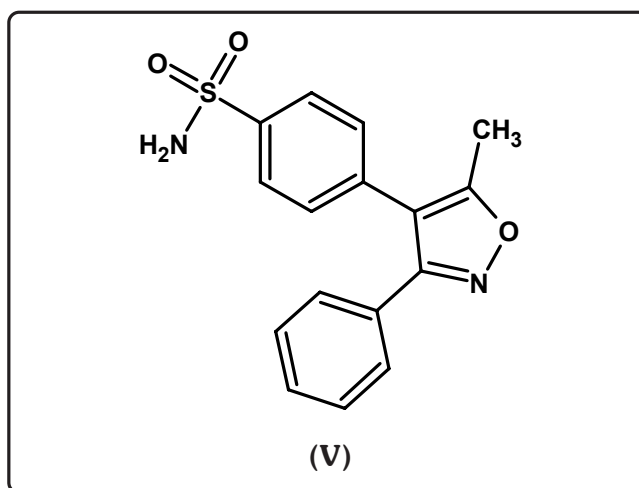
CONTRIBUTION FROM OUR LABORATORY

A. R. Parikh et al.²⁴⁵ have prepared some new isoxazoles and documented them as antimicrobial agent. H. Parekh and co-workers²⁴⁶ have synthesised new isoxazole derivatives and reported their antitubercular activity.

Xin-Ping Hui et al.²⁴⁷ prepared isoxazole derivatives of type (IV) and tested their antibacterial activity.



W. Chengde and co-workers²⁴⁸ have synthesised isoxazole derivatives and reported them as endothelin modulators. Herbicidal activity of isoxazole derivatives have been documented by Q. Chuanmin et al.²⁴⁹ F. Gallemi and co-workers²⁵⁰ described isoxazoles as antitumor agent. Caroline C. et al.²⁵¹ prepared isoxazoles of type (V), which have been evaluated for clinical trials for asthma.



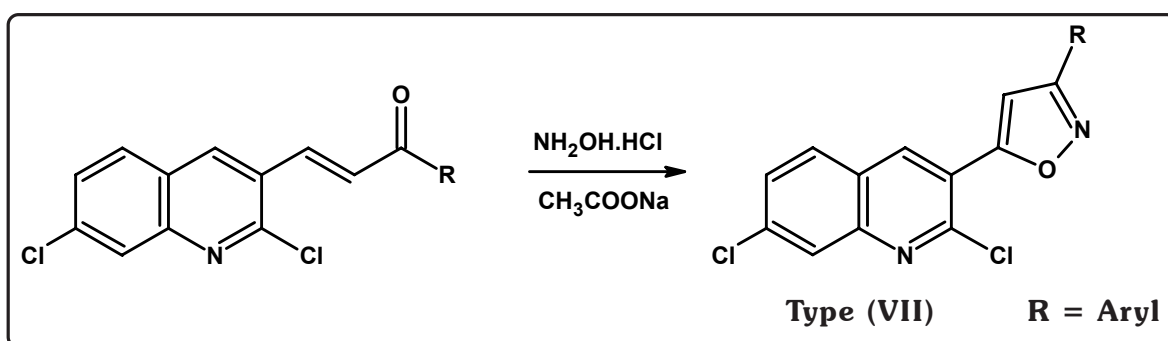
With a view to getting better therapeutic value, it was contemplated to synthesise isoxazole derivatives incorporating quinoline as parent molecule, to enhance the overall activity of resulting compounds which have been described as under.

SECTION - I : SYNTHESIS AND THERAPEUTIC EVALUATION OF 3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-ISOXAZOLES

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF 3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-ISOXAZOLES.

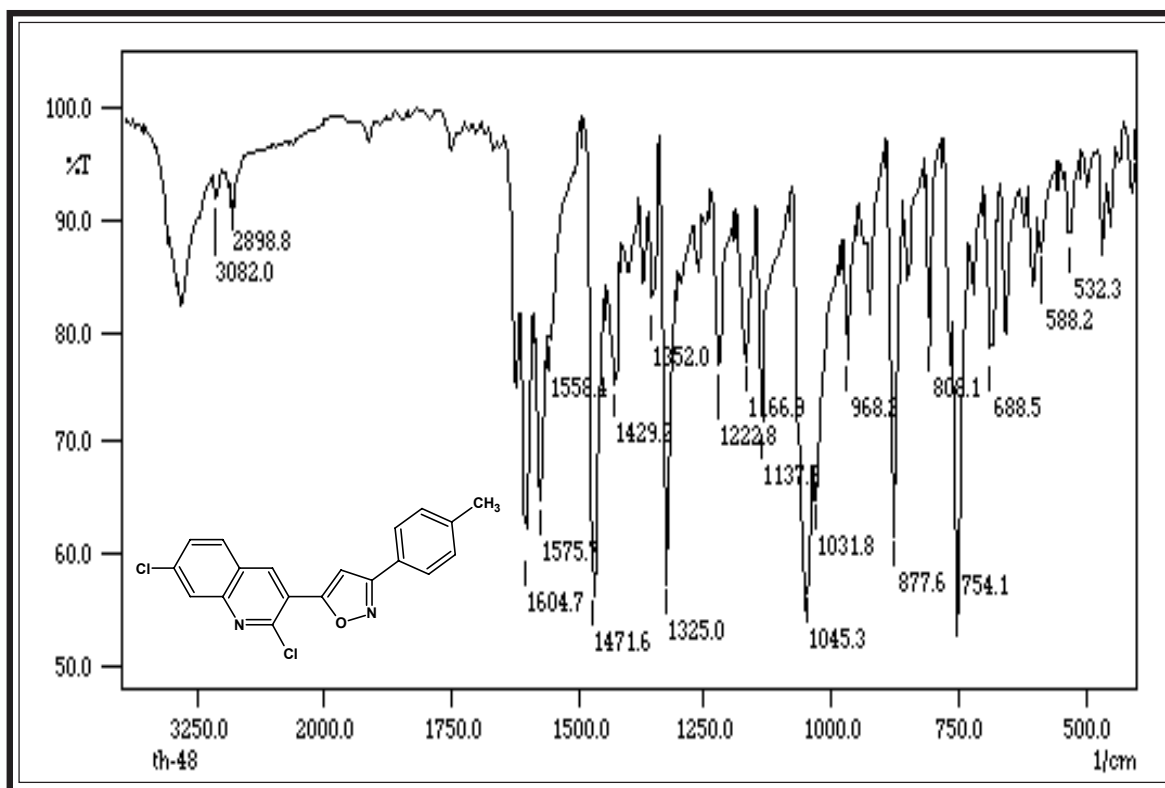
Isoxazole derivatives are well known for their valid pharmacological activities. Therefore, it was planned to synthesise some new isoxazoles of type (VII), bearing 2,7-dichloroquinoline-3-carboxaldehyde nucleus which have been described as under.



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

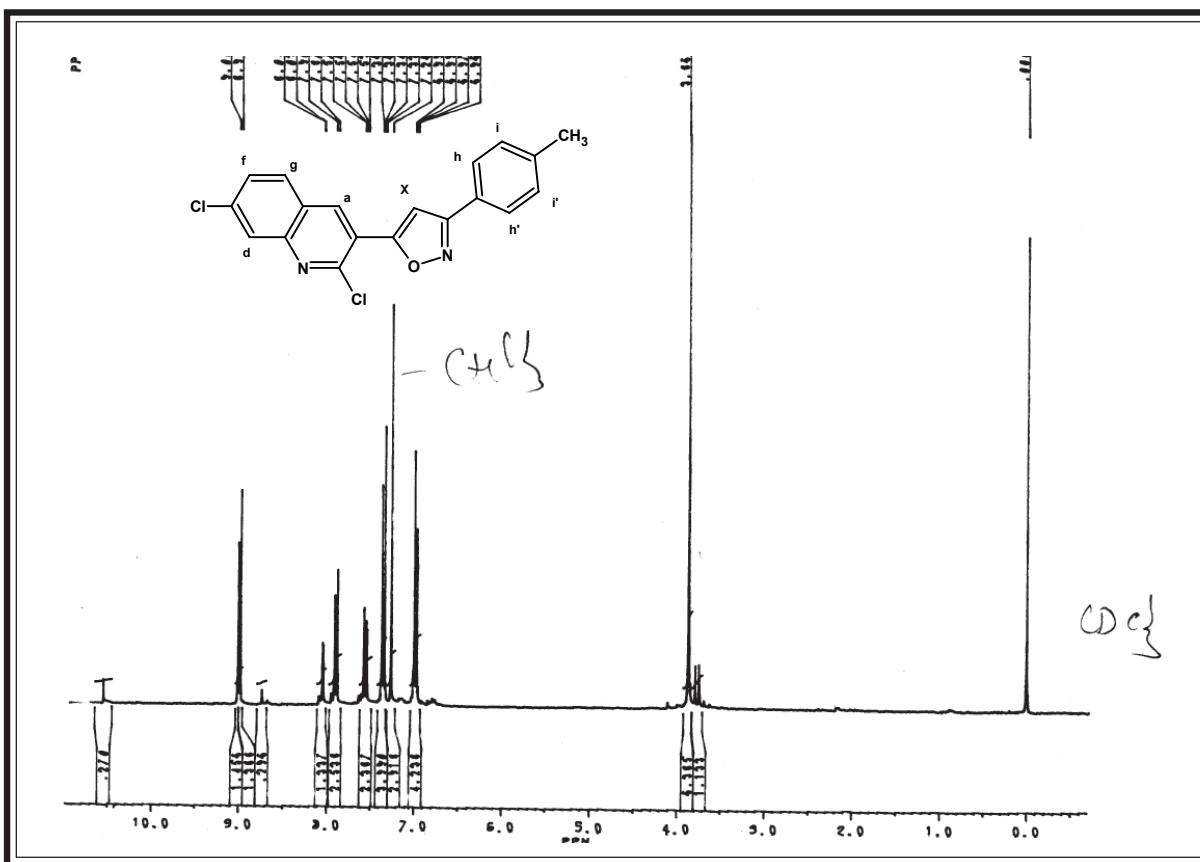
IR SPECTRAL STUDY OF 3-TOLYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-ISOXAZOLE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane	C – H str.	2898	2975-2950	419
	C – H. def.	1471	1470-1435	"
Aromatic	C– H str.	3082	3080-3030	"
	C = C str.	1575	1585-1570	"
	C – H. def.	1137	1125-1090	420
		808	835-810	"
Quinoline moiety	C = N str.	1604	1612-1693	419
	C – Cl str.	754	750-700	"
Isoxazole	C = C str.	1575	1580-1550	"
	C = N str.	1471	1470-1460	"
	N – O str.	808	850-810	"
	(overlaped)			

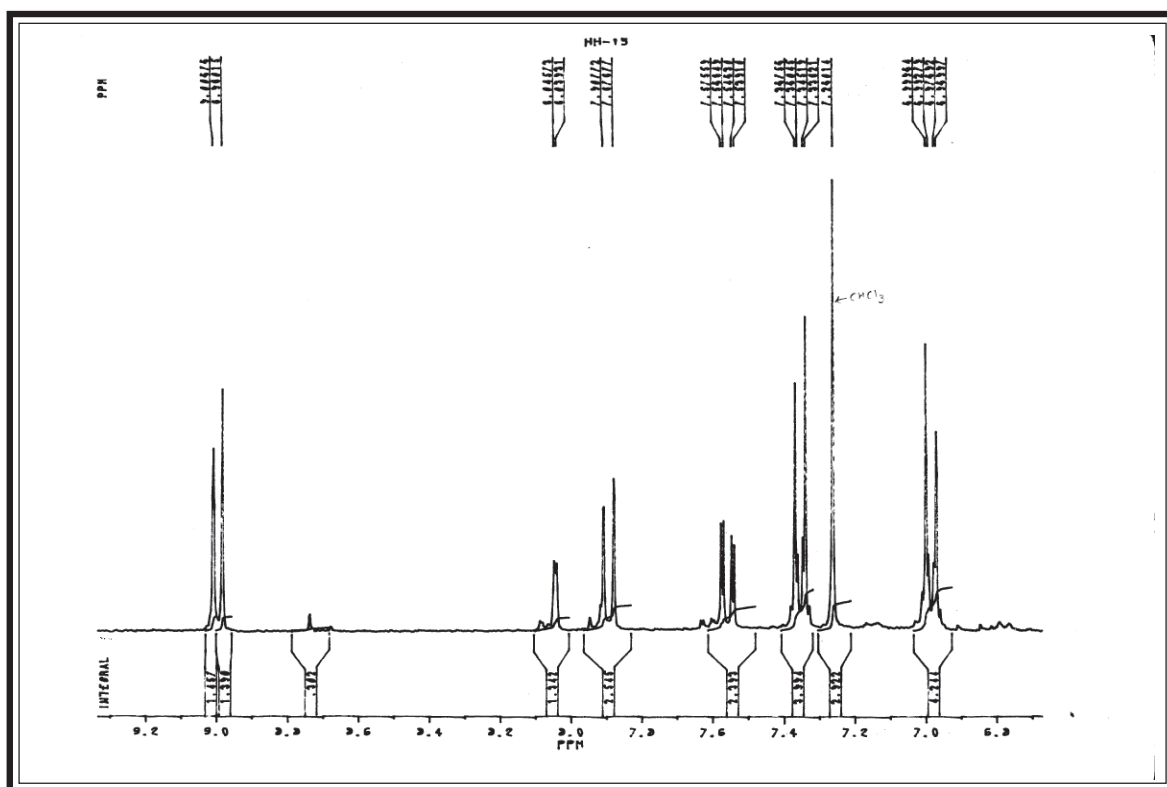
PMR SPECTRAL STUDY OF 3-(p-ANISYL)-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-ISOXAZOLE



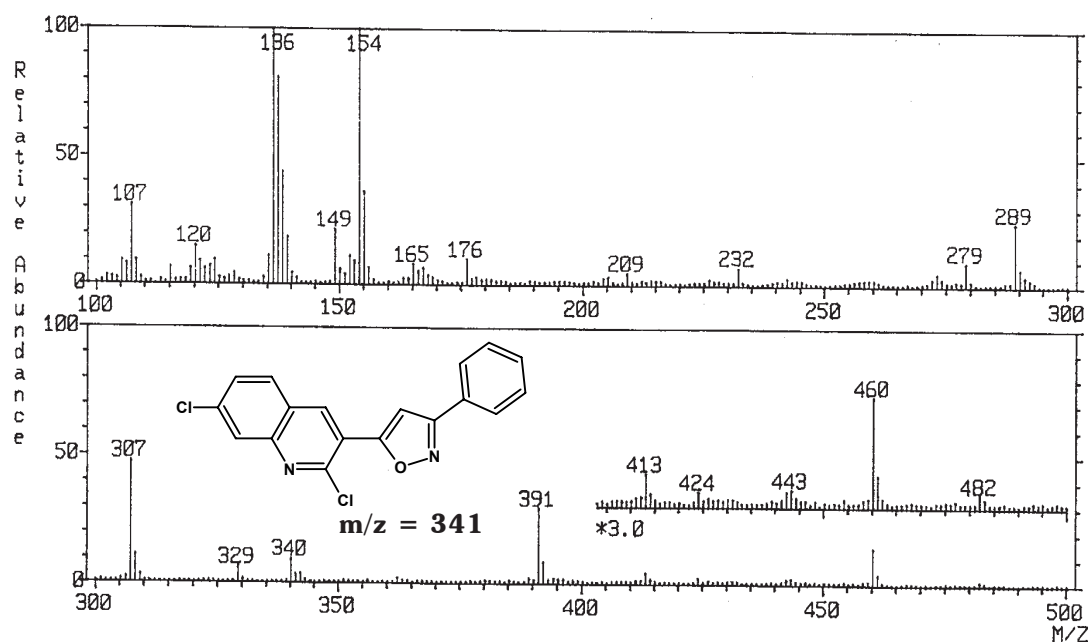
Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	3.87	3H	singlet	-	Ar-OCH ₃
2.	6.97-7.00	2H	doublet	J _{ih} 8.9	Ar-H i,i'
3.	7.34-7.37	2H	doublet	J _{hi} 8.81	Ar-H h,h'
4.	7.54-7.58	1H	double doublet	J _{fg} 8.79 J _{fd} 2.13	Ar-Hf
5.	7.88-7.91	1H	doublet	J _{gf} 8.68	Ar-Hg
6.	8.04-8.05	1H	doublet	J _{df} 1.93	Ar-Hd
7.	8.98	1H	singlet	-	Ar-Ha
8.	9.0	1H	singlet	-	Ar-HX

EXPANDED AROMATIC REGION



MASS SPECTRUM Data File: 3EJL25Z 25-JUL- 3 12:41
 Sample: HHA-11 DR NA CHAUHAN, RAJKOT #6204
 RT 0.24' FAB(Pos.) GC 1.4c BP: m/z 136.0000 Int. 69.3725 Lv 0.00
 Scan# (2 to 4)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-ISOXAZOLES.

[A] Synthesis of 1-Aryl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-ones

See Part (I), Section - 1 (C).

[B] Synthesis of 3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-isoxazoles

Anhydrous sodium acetate (1.46g, 0.02 mol) dissolved in a minimum amount of hot acetic acid was added to a solution of hydroxylamine hydrochloride (1.4 g, 0.02 mol) in ethanol (20 ml). This solution was added to a solution of 1-p-tolyl-3-(2',7'-dichloroquinolin-3'-yl)-2-propene-1-one (3.42 g, 0.01 mol) in ethanol (25 ml). The mixture was heated under reflux on waterbath for 12 hrs. The product was isolated and recrystallised from ethanol. Yield 64%, m.p. 174°C. Anal. Calcd. for $C_{19}H_{12}N_2Cl_2O$ required : C, 64.24%; H, 3.41%; N, 7.89%; Found : C, 64.20%; H, 3.44%; N, 7.69%.

TLC Solvent system : Acetone : Benzene : (3 : 7) Visulising Agent : Iodine

Similarly other 3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-isoxazoles were prepared. The physical data along with infra red spectral data are recorded in Table No. 5.

[C] Therapeutic evaluation 3-Aryl-5-(2',7'-dichloroquinolin-3'-yl)-isoxazoles

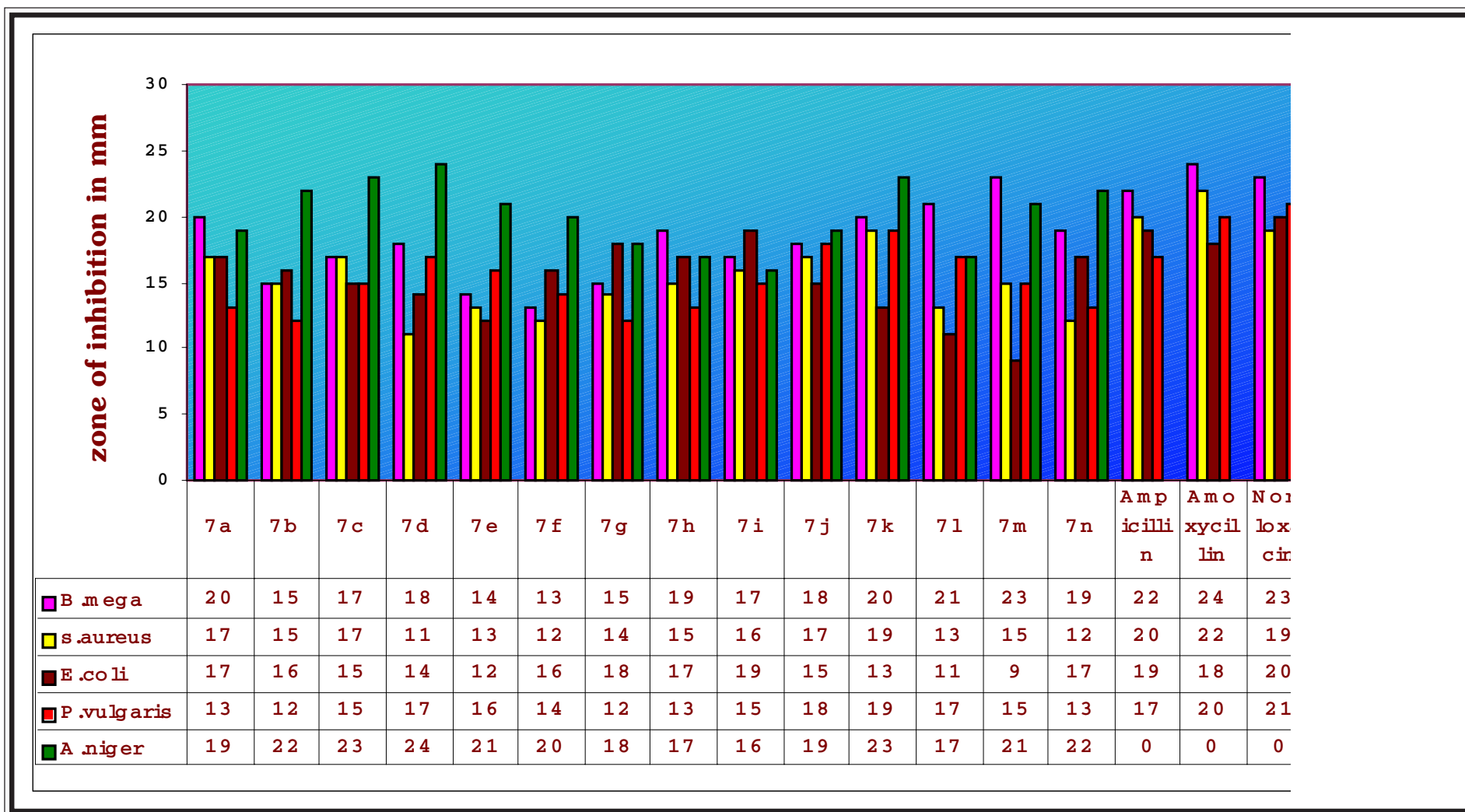
Antimicrobial testing was carried out as described in Part-I, Section-I (D). The zones of inhibition of the test solutions are recorded Graphical Chart No. 7.

TABLE NO. 5 : PHYSICAL CONSTANTS OF 3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-ISOXAZOLES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
7a	C ₆ H ₅ -	C ₁₈ H ₁₀ N ₂ Cl ₂ O	341	238	0.682	64	8.21	8.49
7b	3-NH ₂ -C ₆ H ₄ -	C ₁₈ H ₁₁ N ₃ Cl ₂ O	356	248	0.520	66	11.80	11.55
7c	4-NH ₂ -C ₆ H ₄ -	C ₁₈ H ₁₁ N ₃ Cl ₂ O	356	210	0.469	68	11.80	12.05
7d	4-Br-C ₆ H ₄ -	C ₁₈ H ₉ N ₂ BrCl ₂ O	420	260	0.652	69	6.67	6.95
7e	4-Cl-C ₆ H ₄ -	C ₁₈ H ₉ N ₂ Cl ₃ O	375.5	186	0.534	70	7.46	7.69
7f	2,6-(OH) ₂ -C ₆ H ₃ -	C ₁₈ H ₁₀ N ₂ Cl ₂ O ₃	373	196	0.554	71	7.51	7.57
7g	4-F-C ₆ H ₄ -	C ₁₈ H ₉ N ₂ Cl ₂ FO	359	210	0.480	75	7.80	7.23
7h	C ₄ H ₃ O-	C ₁₆ H ₈ N ₂ Cl ₂ O ₂	331	224	0.587	67	8.46	8.63
7i	2-OH-C ₆ H ₄ -	C ₁₈ H ₁₀ N ₂ Cl ₂ O ₂	357	164	0.643	68	7.84	7.59
7j	4-OH-C ₆ H ₄ -	C ₁₈ H ₁₀ N ₂ Cl ₂ O ₂	357	216	0.687	69	7.84	7.22
7k	4-OCH ₃ -C ₆ H ₄ -	C ₁₉ H ₁₂ N ₂ Cl ₂ O ₂	371	212	0.667	74	7.55	7.55
7l	4-CH ₃ -C ₆ H ₄ -	C ₁₉ H ₁₂ N ₂ Cl ₂ O	355	174	0.713	64	7.89	7.69
7m	3-NO ₂ -C ₆ H ₄ -	C ₁₈ H ₉ N ₃ Cl ₂ O ₃	386	252	0.728	62	10.88	10.62
7n	4-NO ₂ -C ₆ H ₄ -	C ₁₈ H ₉ N ₃ Cl ₂ O ₃	386	128	0.678	69	10.88	10.82

TLC Solvent System : Acetone : Benzene (3 : 7).

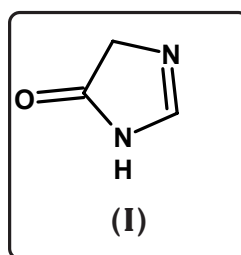
GRAPHICAL CHART NO. 7 : 3-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-YL)-ISOXAZOLES





INTRODUCTION

Imidazolinone (I) is a five membered ring containing 2-nitrogen atoms at the 1- and 3-positions and a carbonyl group at 5- position.



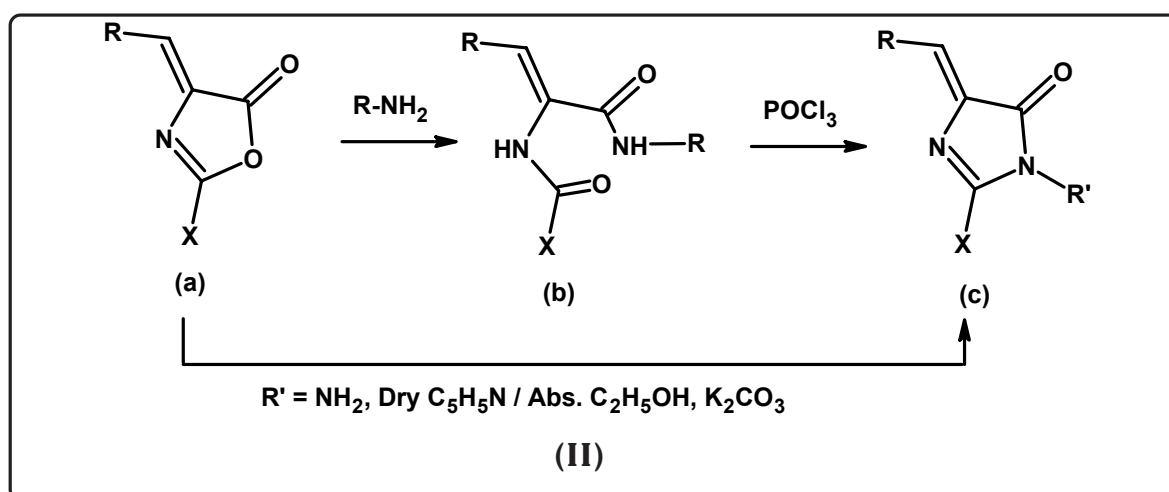
2-Substituted-5-oxoimidazoline was discovered a century back in 1888 by A. W. Hoffmann²⁵², when he prepared 2-methyl-2-imidazoline (lysidine) by heating N'-diacetyleneethylenediamine in a steam of dry hydrogen chloride. A. Ladenburg²⁵³ prepared the same compound by fusing two equivalents of sodium acetate with one equivalent of ethylenediamine dihydrochloride.

SYNTHETIC ASPECT

Different methods have been documented for the synthesis of imidazolines in literature²⁵⁴⁻²⁵⁷. Accordingly, imidazolones can be prepared by the condensation of substituted azalactones and primary amine under anhydrous condition. Formation of imidazolones by the aminolysis of oxazalone with amine has been mentioned in the literature²⁵⁸.

MECHANISM

Azalactone (a) forms amides (b) of α -acylamino acrylic acid by reaction with different amines. The amide on ring closure reaction is converted into imidazolinone derivative (c).



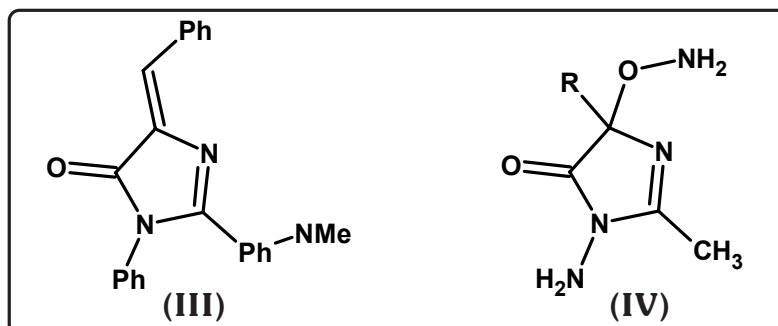
The ring closure can be afforded under a variety of conditions, substituted anilides have been converted into imidazolinone derivatives by the action of phosphorous oxy-chloride. Scheme (II).

THERAPEUTIC IMPORTANCE

Historically, a wide range of biological activities have been attributed to imidazolinone derivatives. A number of imidazolinones have resulted in many potential drugs and are known to possess a broad spectrum of biological activity such as:

1. Antihistamine²⁵⁹
2. Antiinflammatory²⁶⁰
3. Potent CNS depressant²⁶¹
4. Antipyretics and analgesic²⁶²
5. Anthelmintic²⁶³
6. Hypertensive²⁶⁴
7. Local anaesthetic²⁶⁵
8. Monoamine oxidase (MAO) inhibitor²⁶⁶
9. Antiparkinsonian activity²⁶⁷
10. Bactericidal²⁶⁸
11. Anticonvulsant²⁶⁹

Ding M. et al.²⁷⁰ (III) and Pilkington B. et al.²⁷¹ have reported a new series of biologically active analogues of 5-oxo-imidazolines.

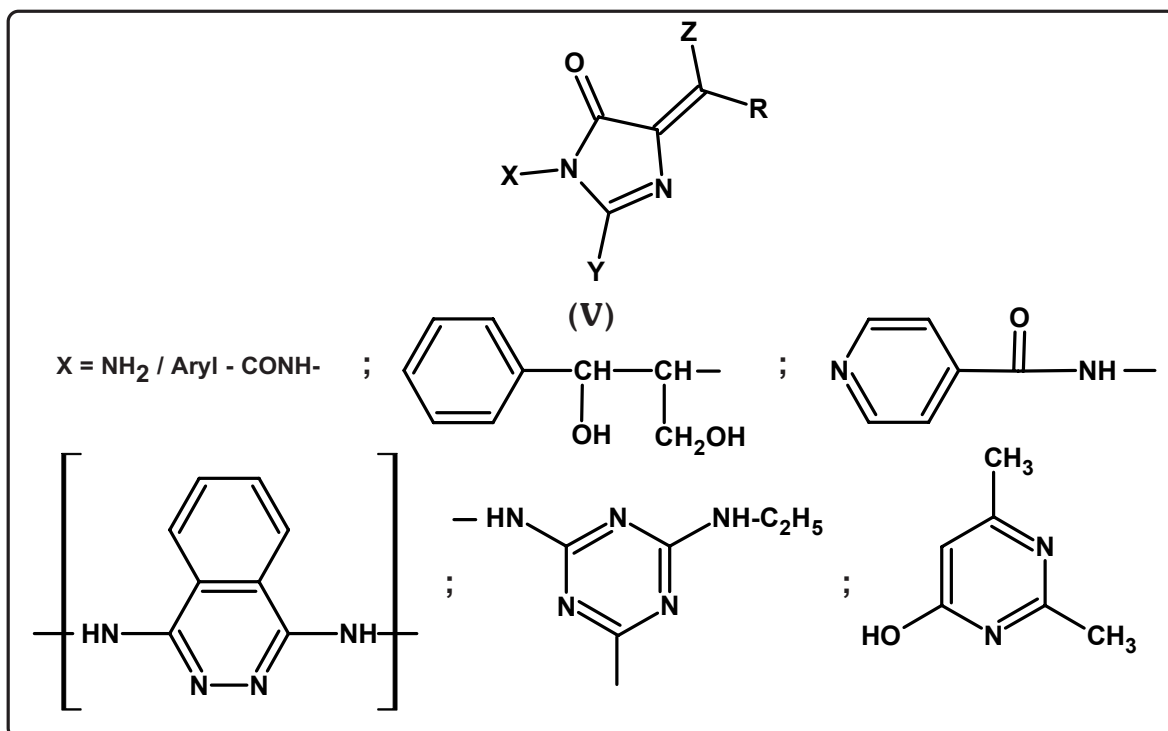


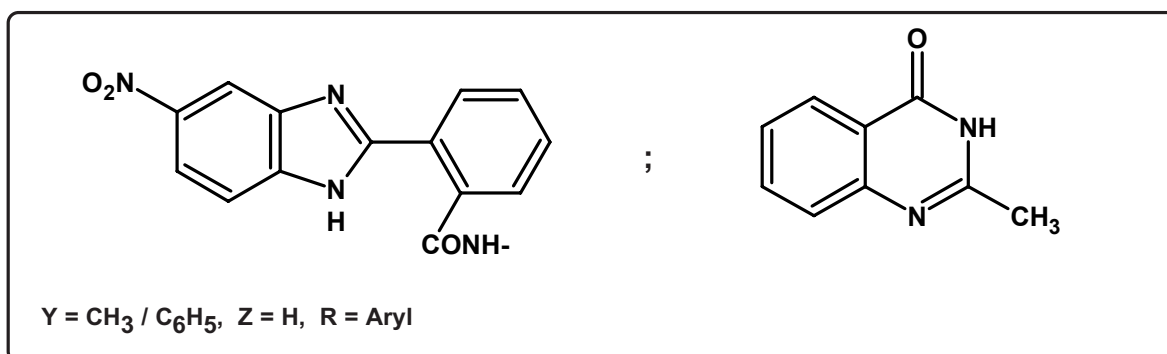
Yokohama Shuichi et al.²⁷² prepared some new imidazolinone derivatives having calmodulin inhibitors activity. Jagham Samir et al.²⁷³ have documented imidazolinone derivatives (IV) as histaminergic H₃ receptors.

CONTRIBUTION FROM OUR LABORATORY

H. H. Parekh and co-workers have assessed imidazolinones containing isoniazide²⁷⁴, and s-triazine²⁷⁵ moieties which were demonstrated as potent antimicrobial agents. A. R. Parikh et al. have synthesised imidazolinones bearing phthalazine²⁷⁶ and chloramphenicol²⁷⁷ moiety at one position which were evaluated for their antimicrobial activity.

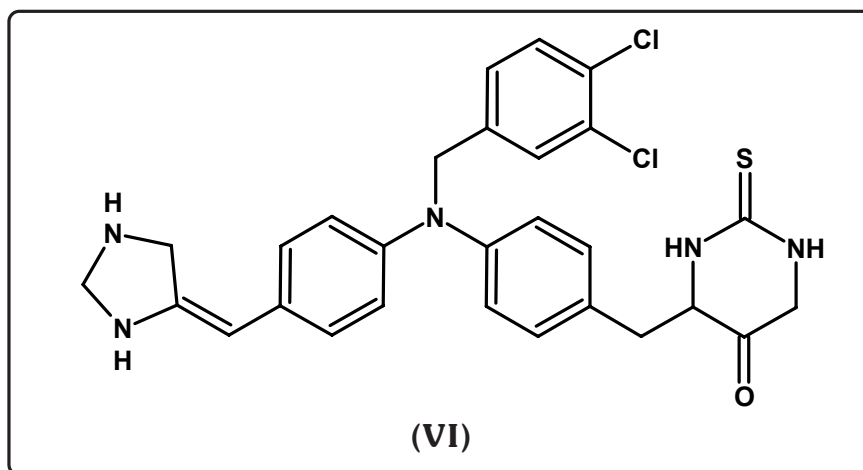
General structure for above references are as under.



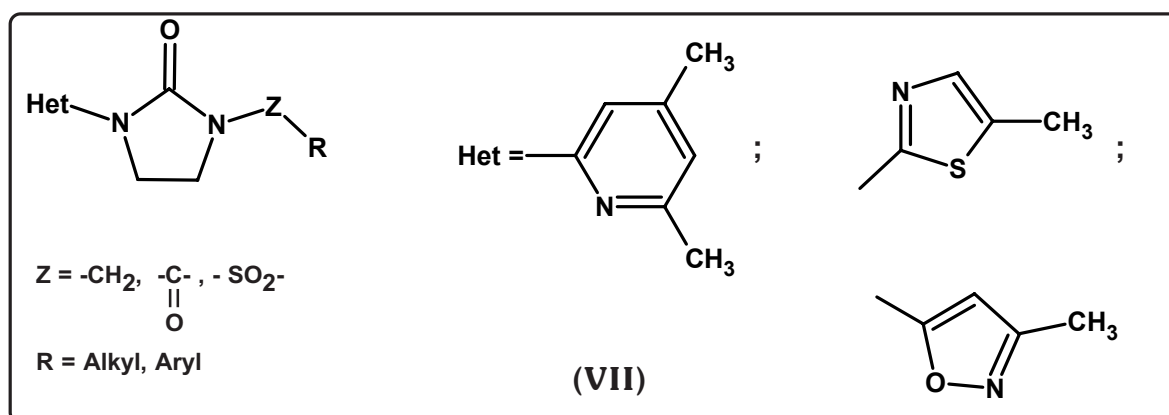


Dharti Joshi et al.²⁷⁸ have reported several novel imidazolinones bearing benzimidazole moiety which were evaluated for their ability to potentiate antimicrobial activity. Akhil Bhatt et al.²⁷⁹ have reported 4-(4'-arylidene-2'-phenyl-5'-oxo-imidazoline-1'-yl)-benzophenones which were screened for their antimicrobial activity. K. Parikh and co-workers²⁸⁰ have synthesised imidazolinone derivatives and described them as antimicrobial agents. H. H. Parekh et al.²⁸¹ have prepared imidazolinone derivatives and reported their antimicrobial activity.

Stefan A. L. et al.²⁸² have prepared imidazolinones and tested for the treatment of cytokine release. Battistina and co-workers²⁸³ have demonstrated some novel imidazolinones to possess insecticidal activity. Anti-HIV activity of imidazolinone derivatives have been reported by Arthur and co-workers²⁸⁴. M. Daisuke and co-workers²⁸⁵ prepared 4-oxo-2-thioxoimidazolidine derivatives (III), which behave as telomerase inhibitors and antitumor agents.



Irene and co-workers²⁸⁶ synthesised some new imidazole derivatives showing antiretroviral activity. Jin-Michel H. et al.²⁸⁷ constructed 2-oxo-tetrahydro imidazole derivatives (VII) possessing antileishmanial activity. Biological activity of some new imidazolinone derivatives have been reported by C. Hamdouchi et al.²⁸⁸



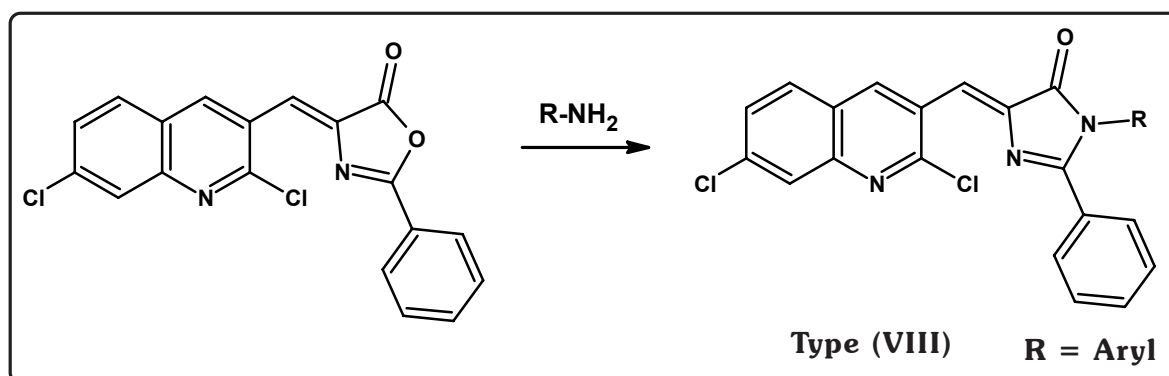
In pursuing the work on the systems incorporating pharmaceutically active molecules, 4-imidazolinone derivatives bearing 2,7-dichloroquinoline nucleus have been synthesised which have been described as under:

SECTION - I : SYNTHESIS AND THERAPEUTIC EVALUATION OF 1N-ARYL-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-5-IMIDAZOLINONES

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1N-ARYL-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-5-IMIDAZOLINONES

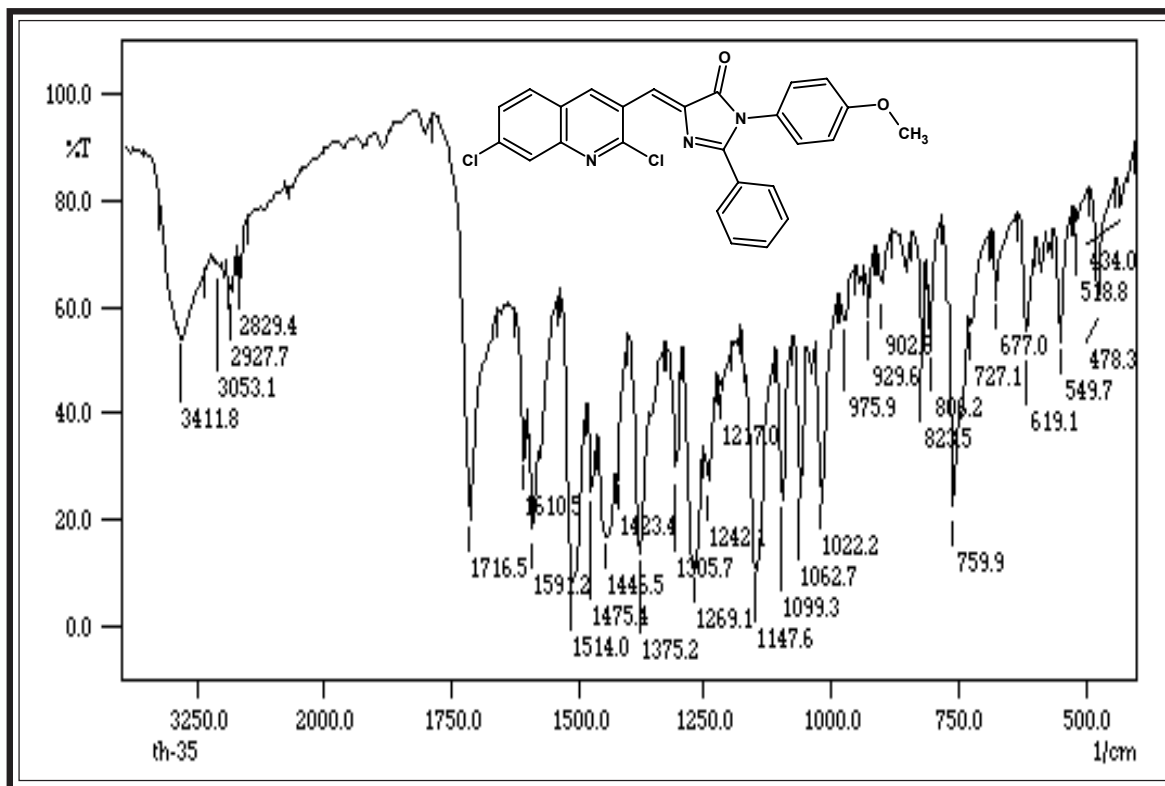
Imidazolinone derivatives possess varieties of biological and pharmacological properties. Here, we report the synthesis of some new 5-Imidazolinones of type (VIII). The strategy employed for the synthesis of desired compounds involved the reaction of 2-phenyl-4-(2',7'-dichloroquinolin-3'-methinyl)-5-oxazolones with different amines.



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

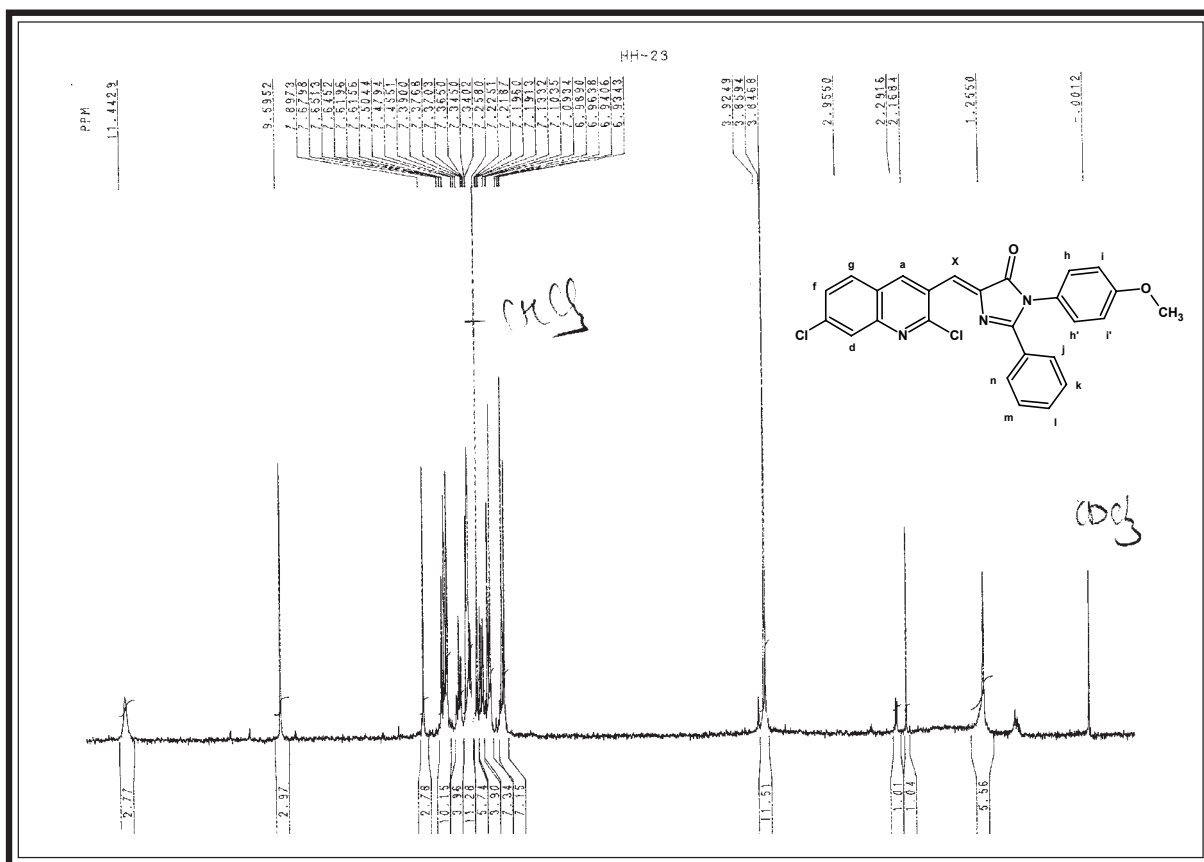
IR SPECTRAL STUDY OF 1N-ANISYL-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-3-IMIDAZOLINONE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane -CH ₃	C - H str. (asym)	2927	2975-2950	419
	C - H str. (sym)	2829	2880-2860	"
	C - H def. (asym)	1446	1470-1435	"
	C - H def. (sym)	1375	1385-1370	"
Aromatic	C- H str.	3053	3080-3030	"
		1591	1585-1570	"
	C - H. i.p. def.	1099	1125-1090	420
	C - H. o.o.p. def.	823	835-810	"
Quinoline moiety	C = N str.	1610	1612-1693	419
	C - Cl str.	759	750-700	"
Imidazole	C = O str.	1716	1760-1665	"
	C - N - C str.	1147	1146-1132	"
Ether	C - O - C	1242	1275-1200	418

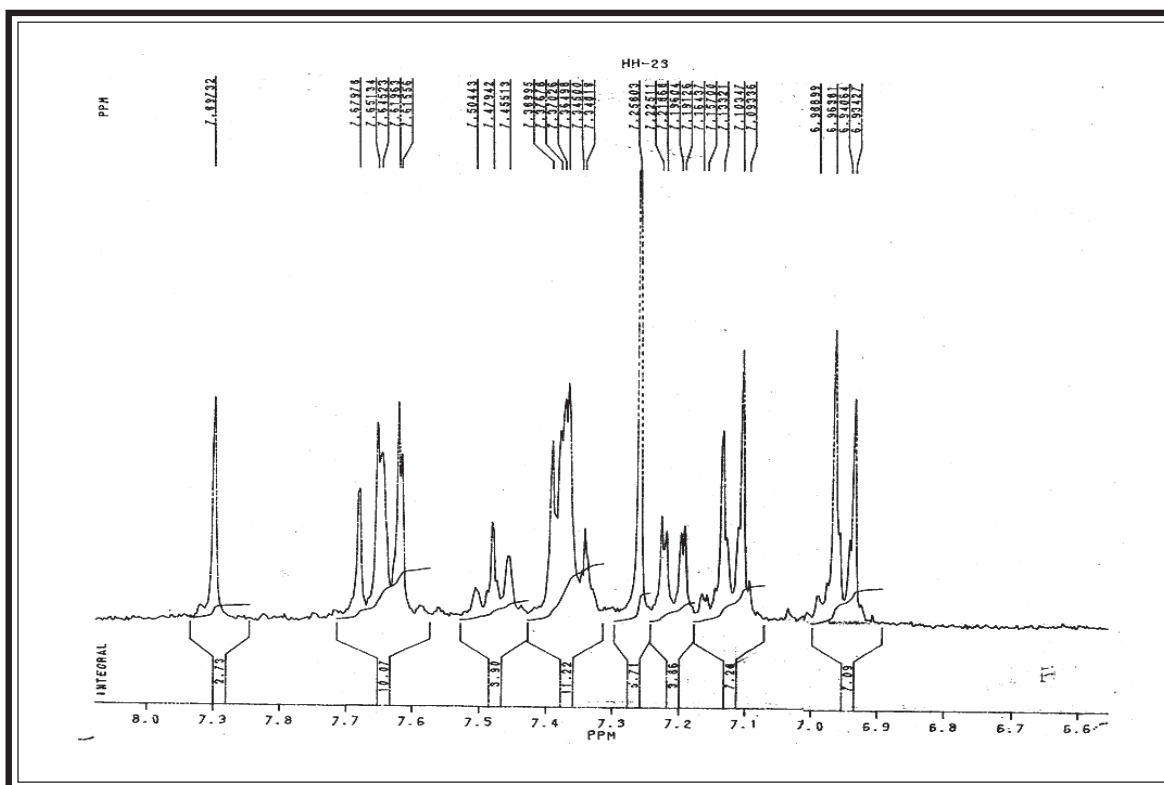
PMR SPECTRAL STUDY OF 1N-(p-ANISYL)-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-5-IMIDAZOLINONE



Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	3.85	3H	singlet	-	Ar-OCH ₃
2.	6.93-6.96	2H	doublet	J _{ih} 8.86	Ar-Hi,i'
3.	7.10-7.13	2H	doublet	J _{hi} 8.92	Ar-Hh,h'
4.	7.19-7.23	1H	double doublet	J _{fg} 8.72 J _{fd} 1.93	Ar-Hf
5.	7.34-7.40	3H	multiplate	-	Ar-Hg,j,n
6.	7.46-7.50	1H	triplet	-	Ar-Hi
7.	7.62-7.68	3H	multiplate	-	Ar-Hd,k,m
8.	7.89	1H	singlet	-	Ar-Ha
9.	9.60	1H	singlet	-	Ar-HX

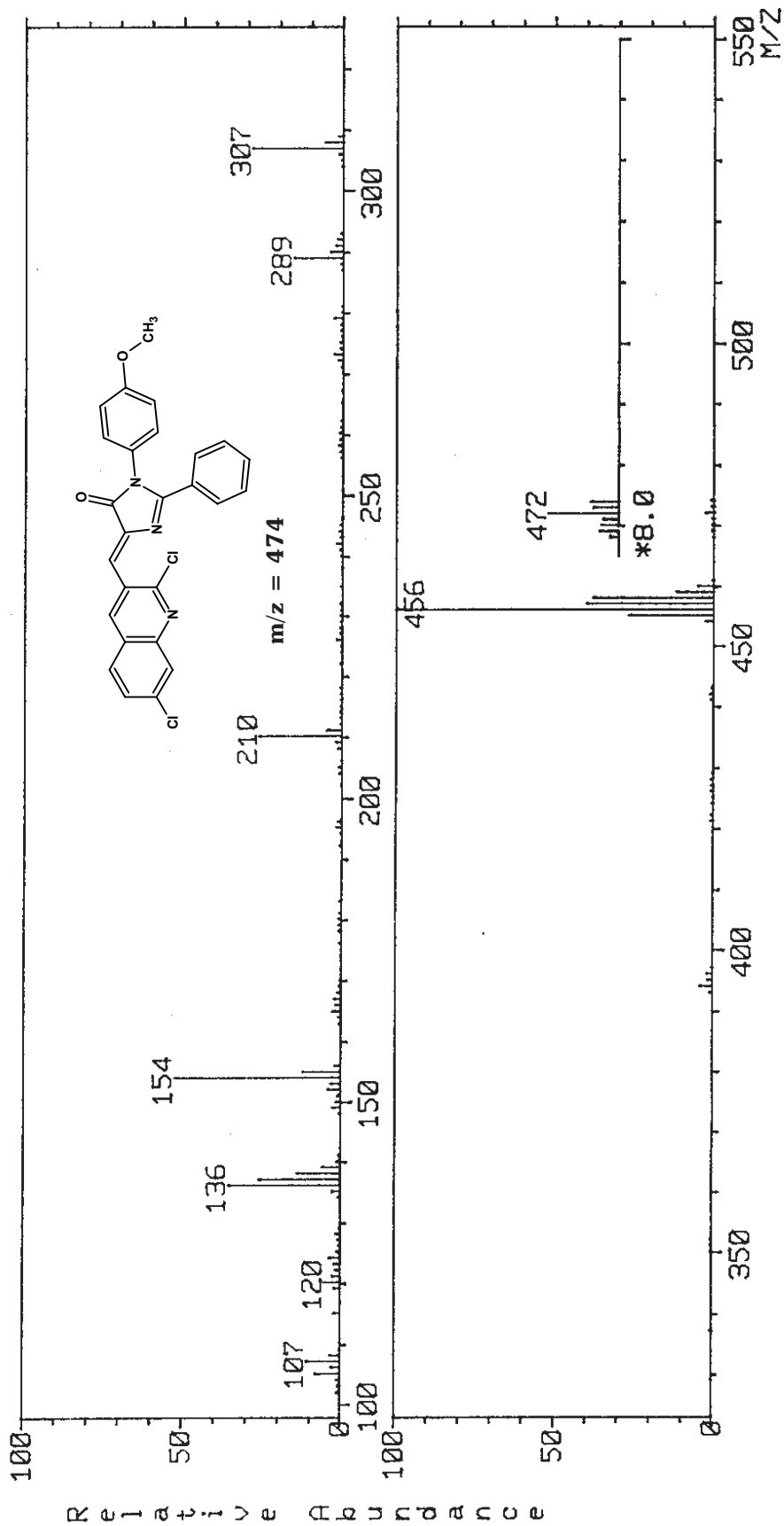
EXPANDED AROMATIC REGION


IR SPECTRAL DATA OF 1N-ARYL-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-5-IMIDAZOLINONES

 Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : $4000-400\text{ cm}^{-1}$ (KBr disc)

Sr. No.	R	C=O str.
8a	C ₆ H ₅	1716
8b	2-Cl-C ₆ H ₄	1716
8c	3-Cl-C ₆ H ₄	1726
8d	4-Cl-C ₆ H ₄	1716
8e	2,5-(Cl) ₂ -C ₆ H ₃	1720
8f	3,4-(Cl) ₂ -C ₆ H ₃	1710
8g	3-Cl,4-F-C ₆ H ₃	1714
8h	2-OCH ₃ -C ₆ H ₄	1714
8i	4-OCH ₃ -C ₆ H ₄	1716
8j	2-CH ₃ -C ₆ H ₄	1716
8k	4-CH ₃ -C ₆ H ₄	1712
8l	2-NO ₂ -C ₆ H ₄	1718
8m	3-NO ₂ -C ₆ H ₄	1710
8n	4-NO ₂ -C ₆ H ₄	1712

MASS SPECTRUM Data File: 3EJL25AA 25-JUL- 3 12:46
 Sample: HHA-12 DR NA CHAUHAN, RAJKOT #6204
 RT 0.24" FAB(Pos.) GC 1.4c BP: m/z 456.0000 Int. 17.1925 Lv 0.00
 Scan# (2 to 4)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 1N-ARYL-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-5-IMIDAZOLINONES

[A] Synthesis of 2-Phenyl-4-(2',7'-dichloroquinolin-3'-methinyl)-5-oxazolone

This was prepared by the condensation of 2,7-dichloroquinolin-3-carboxaldehyde with benzoyl glycine in presence of sodium acetate and acetic anhydride as described by Vogel.²⁸⁹

[B] Synthesis of 1N-Aryl-2-phenyl-4-(2',7'-dichloroquinolin-3'-methinyl)-5-imidazolinones

To a mixture of 2-phenyl-4-(2',7'-dichloroquinolin-3'-methinyl)-5-oxazolone (3.69 g, 0.01 mol) and *p*-anisidine (1.23 g, 0.01 mol), 20 ml of dry pyridine was added and contents were refluxed for 9 hrs. Resulting mass was poured onto crushed ice and neutralised with HCl, filtered and the product was crystallised from ethanol. Yield 57% m.p. 226°C, Anal. Calcd. for $C_{26}H_{17}N_3Cl_2O_2$ required : C, 65.83%; H, 3.61%; N, 8.86%; found : C, 65.85%; H, 3.66%; N, 8.60%. TLC Solvent System : Acetone : Benzene (1.4:8.6) Visulising Agent : Iodine.

Similarly other 1N-Aryl-2-phenyl-4-(2',7'-dichloroquinoline-3'-methiyl)-5-imidazolinones were prepared. The physical data along with infra red spectral data are recorded in Table No. 6.

[C] Therapeutic evaluation of 1N-Aryl-2-phenyl-4-(2',7'-dichloroquinolin-3'-methinyl)-5-imidazolinones

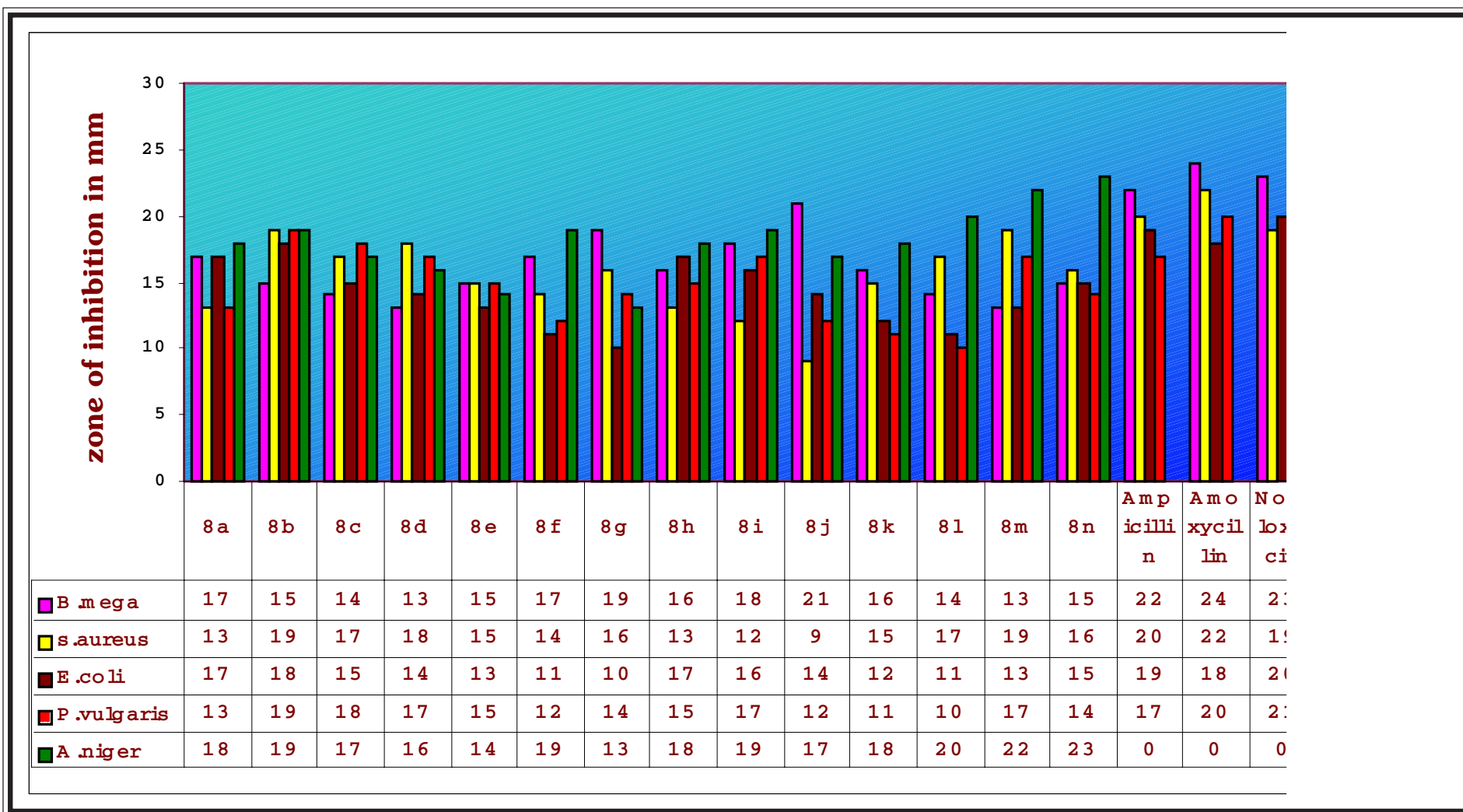
Antimicrobial testing was carried out as described in Part-I, Section-I (D). The zone of inhibition of the test solutions are recorded in Graphical Chart No. 8.

TABLE NO. 6 : PHYSICAL CONSTANTS OF 1N-ARYL-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-5-IMIDAZOLINONES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
8a	C ₆ H ₅ -	C ₂₅ H ₁₅ N ₃ Cl ₂ O	444	216	0.744	57	9.46	9.24
8b	2-Cl-C ₆ H ₄ -	C ₂₅ H ₁₄ N ₃ Cl ₃ O	478.5	>300	0.674	56	8.78	8.59
8c	3-Cl-C ₆ H ₄ -	C ₂₅ H ₁₄ N ₃ Cl ₃ O	478.5	178	0.542	59	8.78	8.25
8d	4-Cl-C ₆ H ₄ -	C ₂₅ H ₁₄ N ₃ Cl ₃ O	478.5	186	0.771	51	8.78	8.58
8e	2,5-(Cl) ₂ -C ₆ H ₃ -	C ₂₅ H ₁₃ N ₃ Cl ₄ O	513	220	0.681	53	8.19	8.39
8f	3,4-(Cl) ₂ -C ₆ H ₃ -	C ₂₅ H ₁₃ N ₃ Cl ₄ O	513	110	0.742	60	8.19	8.44
8g	3-Cl,4-F-C ₆ H ₃ -	C ₂₅ H ₁₃ N ₃ Cl ₃ FO	496.5	182	0.648	65	8.46	8.29
8h	2-OCH ₃ -C ₆ H ₄ -	C ₂₆ H ₁₇ N ₃ Cl ₂ O ₂	474	178	0.694	55	8.86	8.58
8i	4-OCH ₃ -C ₆ H ₄ -	C ₂₆ H ₁₇ N ₃ Cl ₂ O ₂	474	226	0.744	57	8.86	8.60
8j	2-CH ₃ -C ₆ H ₄ -	C ₂₆ H ₁₇ N ₃ Cl ₂ O	458	212	0.548	58	9.17	9.37
8k	4-CH ₃ -C ₆ H ₄ -	C ₂₆ H ₁₇ N ₃ Cl ₂ O	458	172	0.624	54	9.17	9.40
8l	2-NO ₂ -C ₆ H ₄ -	C ₂₅ H ₁₄ N ₄ Cl ₂ O ₃	489	192	0.621	51	11.45	11.23
8m	3-NO ₂ -C ₆ H ₄ -	C ₂₅ H ₁₄ N ₄ Cl ₂ O ₃	489	208	0.634	56	11.45	11.67
8n	4-NO ₂ -C ₆ H ₄ -	C ₂₅ H ₁₄ N ₄ Cl ₂ O ₃	489	222	0.567	59	11.45	11.18

TLC Solvent System : Acetone : Benzene (1.4 : 8.6).

GRAPHICAL CHART NO. 8 : 1N-ARYL-2-PHENYL-4-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-5-IMIDAZOLINONES

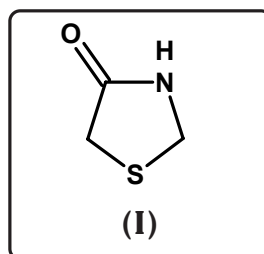




PART - VII
STUDIES ON
5-ARYLIDENE-
4-THIAZOLIDINONES

INTRODUCTION

In the family of heterocyclic compounds, thiazolidinones are presumably important. Thiazolidinones are reported to possess a broad spectrum of biological activity. 4-Thiazolidinones are derivatives of thiazolidine with a carbonyl group at 4-position (I). Numerous reports have been appeared in the literature which highlight their chemistry and use.

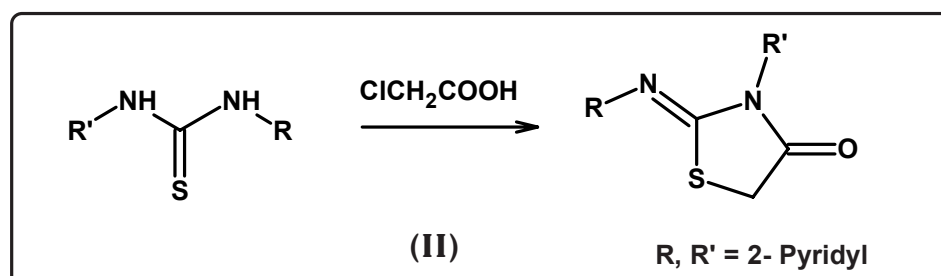


The cyclic structure was assigned after recognition of mercaptoacetic acid as a primary product of hydrolysis of 3-phenyl-2-phenylimino-4-thiazolidinones²⁹⁰.

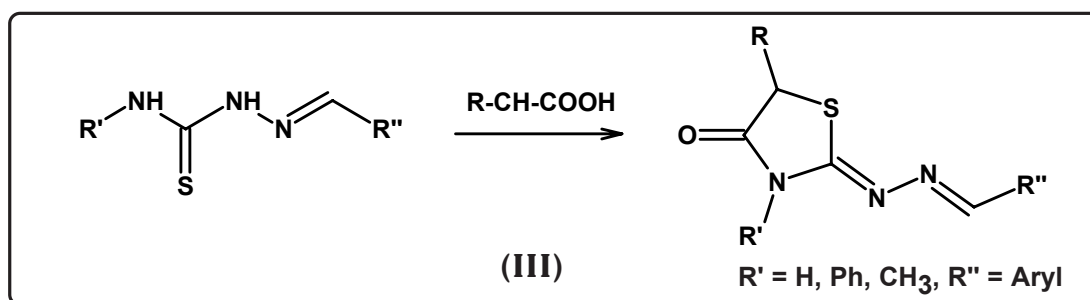
SYNTHETIC ASPECT

Several methods for the preparation of 4-thiazolidinones are narrated in literature²⁹¹⁻²⁹⁹.

1. R. Nath and K. Shanker³⁰⁰ have prepared 4-thiazolidinone by cyclization of N-aryl-N'-(2'-pyridyl)-thiocarbamide with chloroacetic acid.



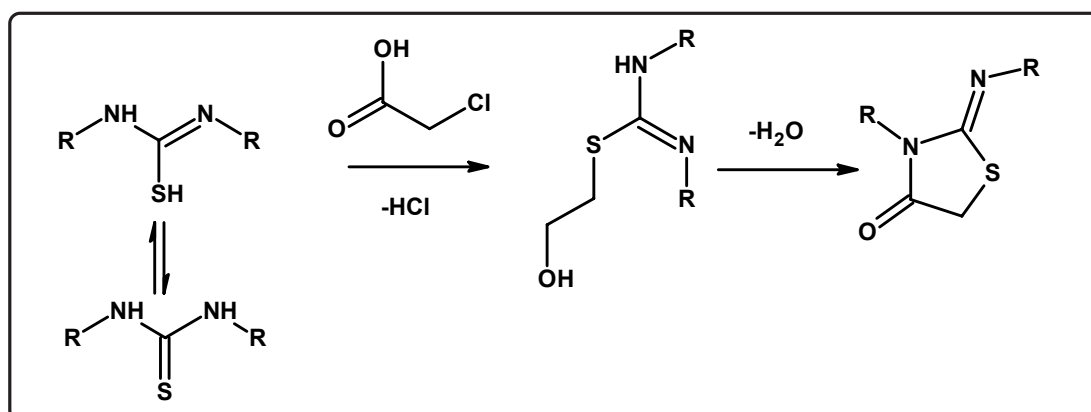
2. I. D. Shah and J. P. Trivedi³⁰¹ synthesised thiazolidinones from the 4-aryl thiosemecarbazones by condensing them with chloroacetic acid, α -bromopropionic acid and α -bromophenyl acetic acid.



3. M. Seada and co-workers³⁰² synthesised some new thiazolidinones from N-thiourea derivatives and chloroacetic acid.

MECHANISM

The reaction proceeds with the tautomerisation of bis-thiourea derivative which on reaction with chloroacetic acid gives thioether derivative. The later on cyclodehydration gives 4-thiazolidinones derivative.



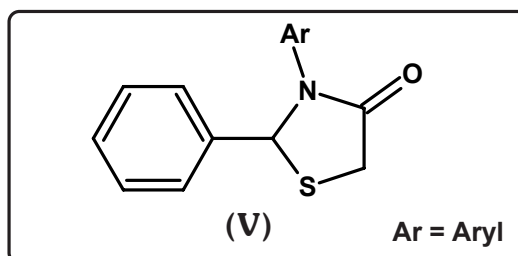
THERAPEUTIC IMPORTANCE

Thiazolidinone, being a biologically active molecule, has attracted many scientist to investigate therapeutic values of thiazolidinone derivatives. 2,3-Disubstituted thiazolidinones are reported to demonstrate a wide spectrum of biological activity.

1. Antimicrobial^{303,304}
2. Thrombin inhibitors³⁰⁵
3. Bactericidal^{306, 307}
4. Antidiabetic³⁰⁸
5. Antifungal^{309,310}
6. Antipsychotic³¹¹

7. CNS effect³¹²
8. Antihypertensive³¹³
9. Antiinflammatory³¹⁴
10. Anthelmintics³¹⁵
11. Anticonvulsant³¹⁶
12. Plant Growth Inhibitor^{317,318}
13. H1-histamine antagonists³¹⁹
14. Antitubercular³²⁰
15. Antioxidant³²¹

R. S. Lodhi and co-workers³²² have synthesised 4-thiazolidinone derivatives, which demonstrate antimicrobial, antiinflammatory and analgesic activities. V. S. Ingle et al.³²³ have prepared some novel 4-thiazolidinones and documented as potent antimicrobial agent. Sharma R. C. and co-workers³²⁴ have synthesised some new thiazolidin-4-one derivatives (V) as possible antimicrobial agents.

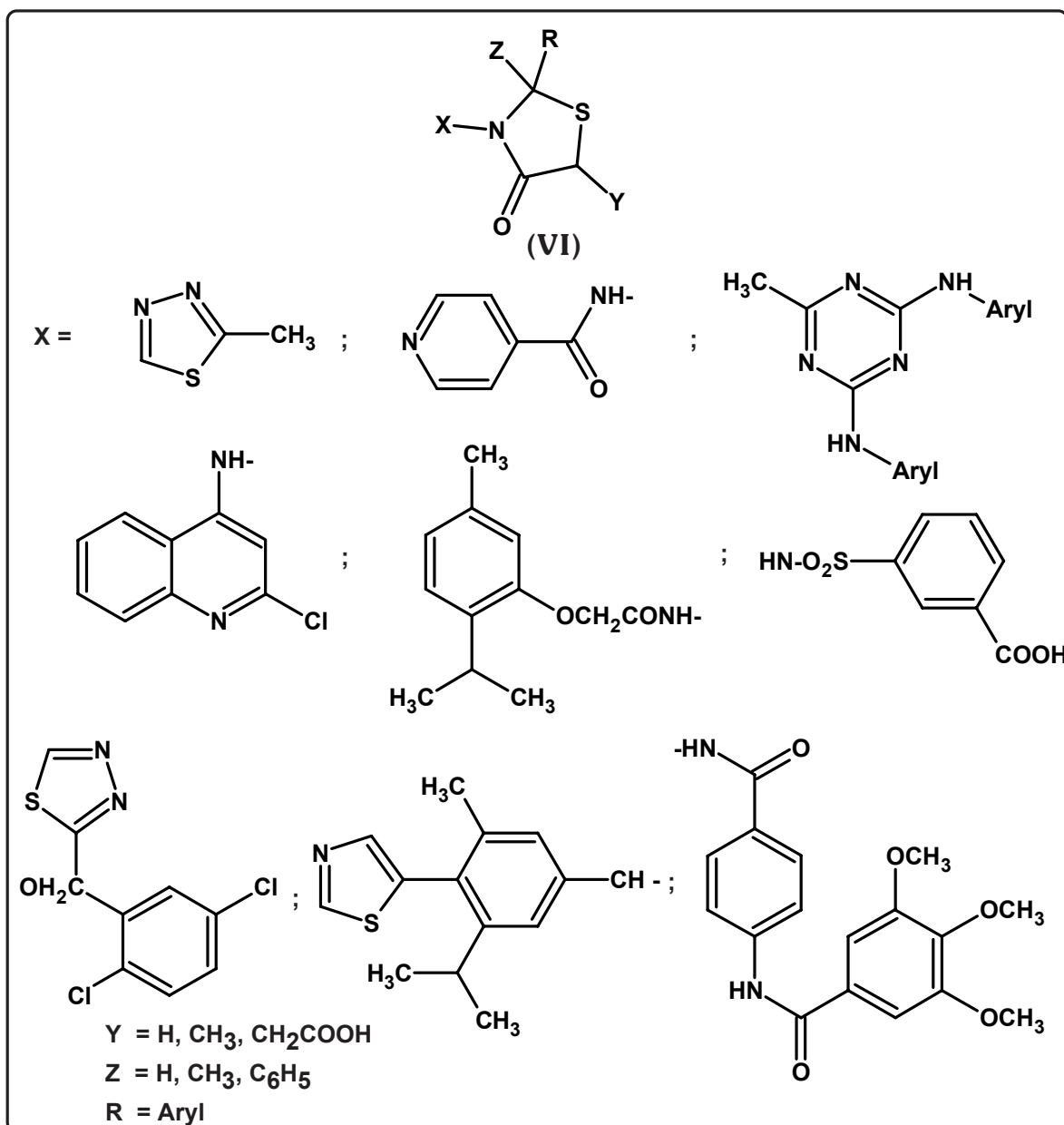


CONTRIBUTION FROM OUR LABORATORY

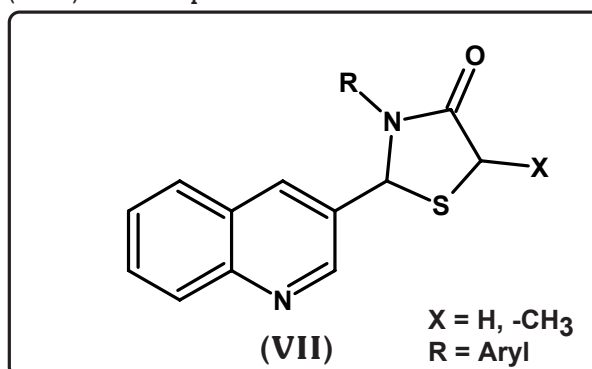
4-Thiazolidinones (VI) bearing s-triazine^{325,326}, acridin-9-yl³²⁷, dapson^{328,329}, 6-hydroxypyrimidine³³⁰, thymoloxo acetamido³³¹ have been synthesised by H. H. Parekh et al. and documented as potent antimicrobial agent.

Parikh et al. have synthesised variety of 4-thiazolidinone derivatives bearing arsenilic acid³³², 2-aryl-1,3,4-thiadiazol³³³, γ -picolinylamino³³⁴, s-triazine³³⁵, benzoylamino acetamido³³⁶, sulphonamido-benzoylamino³³⁷, substituted aryl³³⁸, substituted quinoline³³⁹ moieties and reported as potent antimicrobial agent.

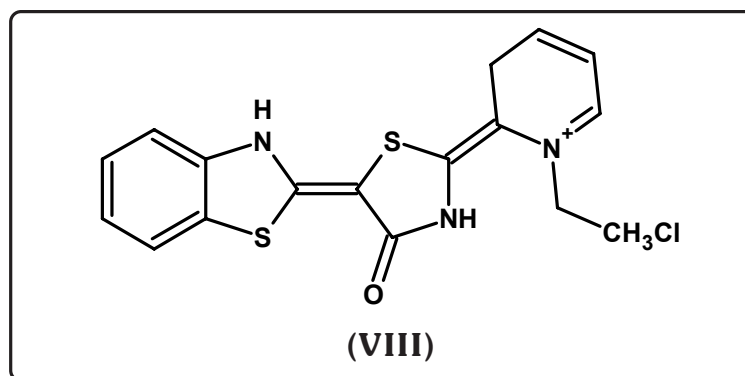
General structure for above references are as under.



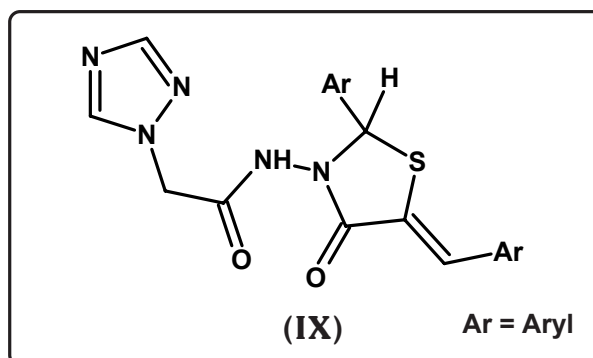
A. R. Parikh and co-workers³⁴⁰ have prepared 4-thiazolidinones bearing quinoline moiety (VIII) and reported their antibacterial activity.



4-Thiazolidinones incorporating benzthiazole nucleus with good antimalarial activity have been synthesised by Takasu et al.³⁴¹ Abdel-Megid and co-worker³⁴² have prepared some new bis-4-thiazolidinone derivatives and reported their antifungal activity. A. Tsutomu and co-workers³⁴³ have synthesised some novel thiazolidine-diones and described them as telomerase inhibitors. S. G. Kucukguzel et al.³⁴⁴ have reported antimicrobial and antitumor activity of some novel 4-thiazolidinones (VIII).



S. K. Srivastava et al.³⁴⁵ have described some new 5-arylidene-4-thiazolidinone derivatives (IX) and documented their antimicrobial, analgesic and diuretic activities.



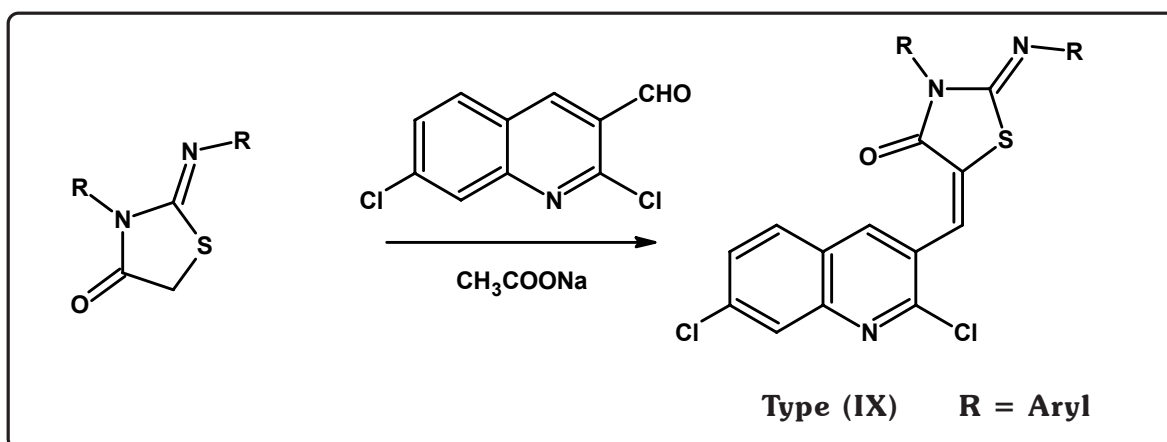
In view of the above observations, the synthesis of novel 4-thiazolidinone derivatives starting from different amines were aimed at investigating therapeutic importance of these compounds, which has been described as under:

**SECTION - I : SYNTHESIS AND THERAPEUTIC EVALUATION OF
2-ARYLIMINO-3N-ARYL-5-(2',7'-DICHLORO-
QUINOLIN-3'-METHINYL)-4-THIAZOLIDINONES**

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF 2-ARYLIMINO-3N-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-4-THIAZOLIDINONES

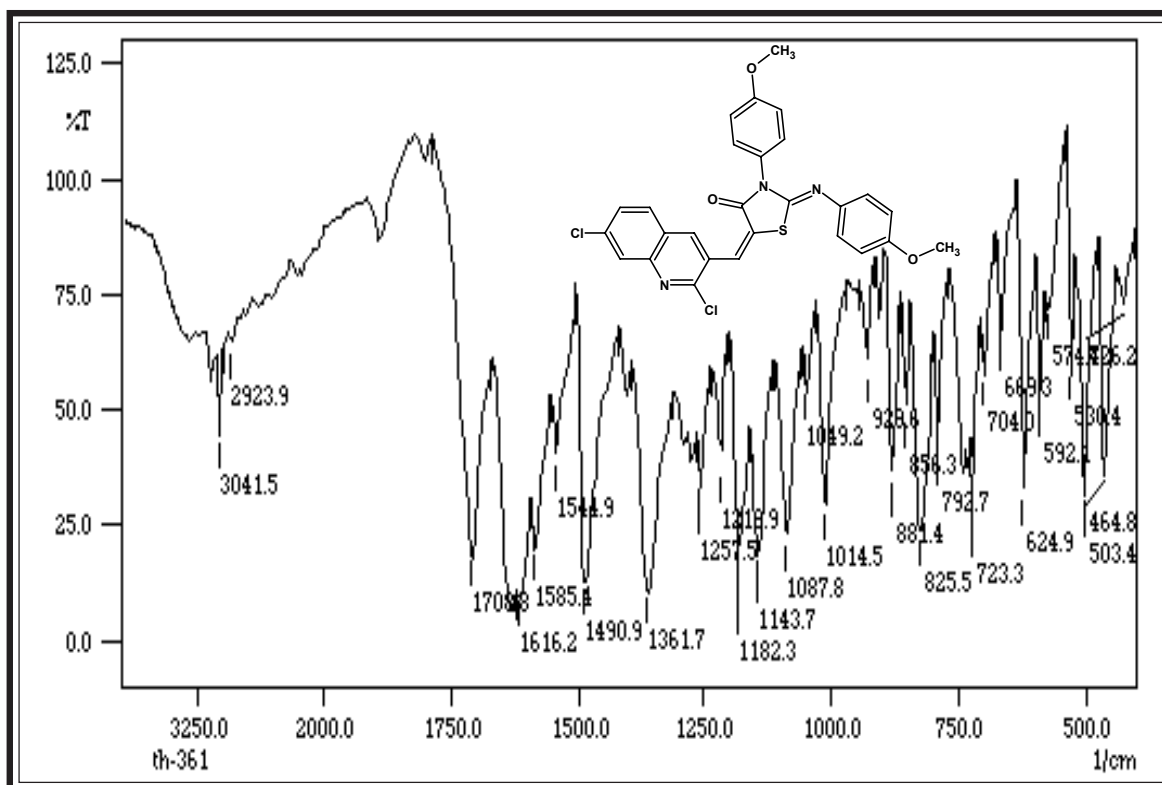
Thiazolidinone derivatives are known to be among those heterocycles which find their utility in various fields of life. The significance of this moiety has increased due to its biocompatibility, which inspired us to synthesis some new thiazolidinones of type (IX).



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

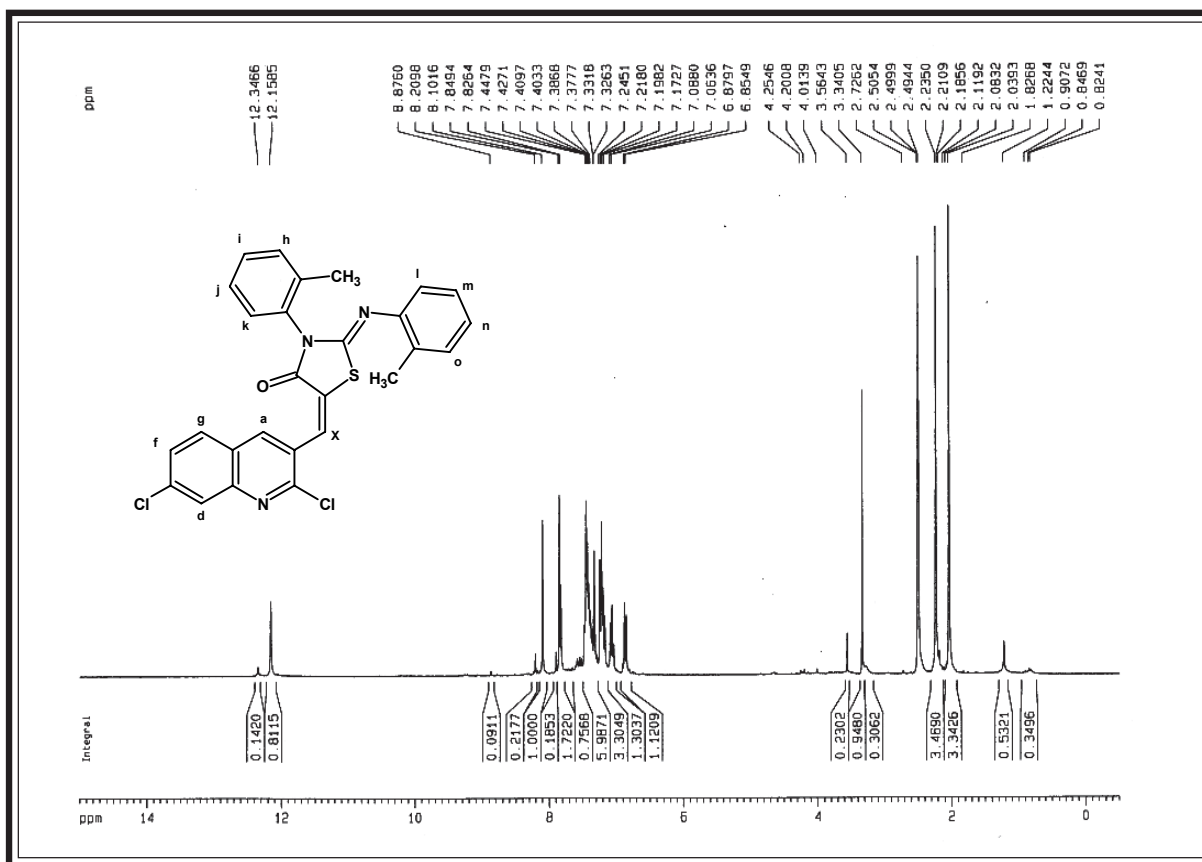
IR SPECTRAL STUDY OF 2-ANISYLIMINO-3N-ANISYL-5-(2'-7'-DICHLORO-QUINOLIN-3'-METHINYL)-4-THIAZOLIDINONE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

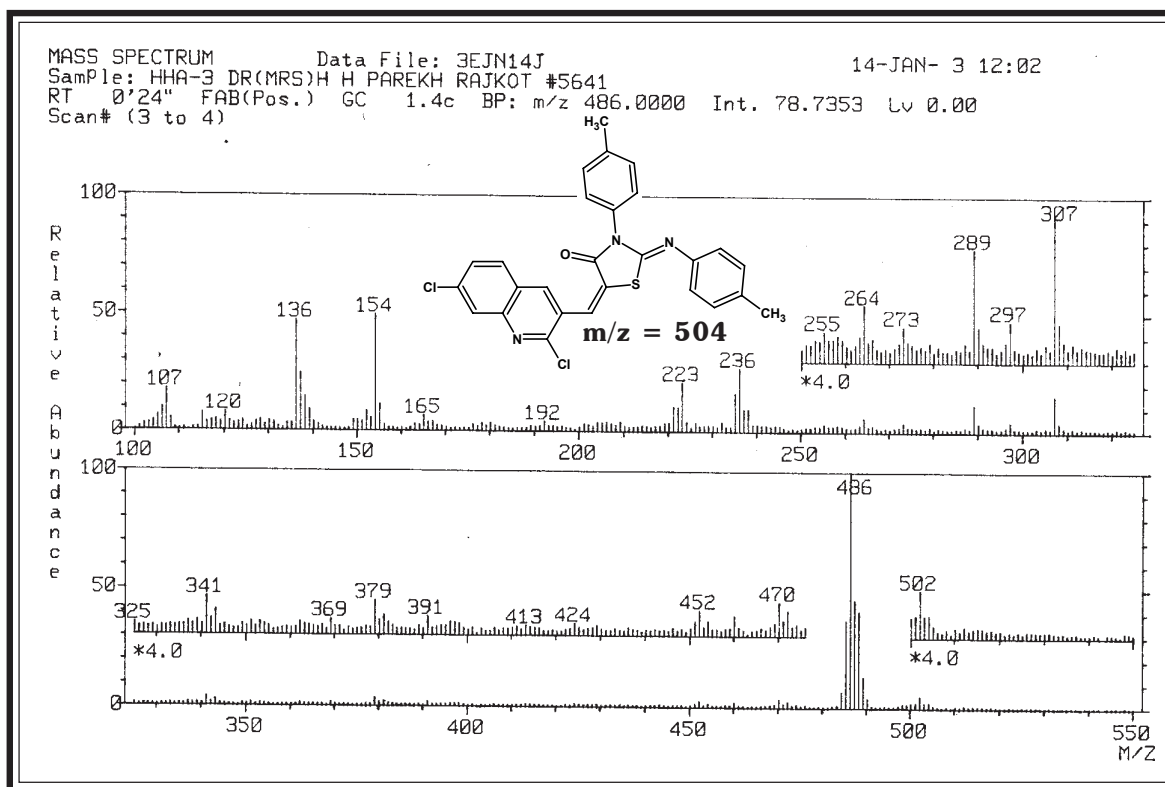
Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane	C – H str.	2923	2975-2850	419
	C – H def.	1361	1385-1370	"
Aromatic	C– H str.	3041	3080-3030	"
	C = C str.	1585	1585-1570	"
	C – H. def.	1087	1125-1090	420
		825	835-810	"
Quinoline moiety	C = N str.	1616	1612-1593	419
	C – Cl str.	723	750-700	"
Ether	C – O – C	1257	1275-1200	"
Thiazolidinone	C = O str.	1708	1760-1655	418
	C – N str.	1218	1220-1020	"
	C – S – C str.	624	700-600	"

PMR SPECTRAL STUDY OF 2-(*o*-TOLYL)-IMINO-3N-(*o*-TOLYL)-5-(2',7'-DICHLORO-QUINOLIN-3'-METHINYL)-4-THIAZOLIDINONE



Internal Standard : TMS; Solvent : CDCl_3 ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	2.03	3H	singlet	-	Ar- CH_3
2.	2.23	3H	singlet	-	Ar- CH_3
3.	6.73-6.76	1H	doublet	Jim 6.62	Ar-Hi
4.	7.06-7.09	1H	doublet	Jhi 7.31	Ar-Hh
5.	7.17-7.25	3H	multiplate	-	Ar-H f,k,o
6.	7.32-7.48	5H	multiplate	-	Ar-H d,i,j,m,n
7.	7.83-7.86	2H	doublet	Jef 6.89	Ar-H g,x
8.	8.1	1H	singlet	-	Ar-Ha



IR SPECTRAL DATA OF 2-ARYLIMINO-3N-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-4-THIAZOLIDIONES

Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	C=O str.
9a	C_6H_5	1703
9b	4-Cl- C_6H_4	1707
9c	3,4-(Cl) $_2$ - C_6H_3	1708
9d	4-F- C_6H_4	1710
9e	2-OCH $_3$ - C_6H_4	1712
9f	4-OCH $_3$ - C_6H_4	1708
9g	2-CH $_3$ - C_6H_4	1703
9h	3-CH $_3$ - C_6H_4	1703
9i	4-CH $_3$ - C_6H_4	1710
9j	2-NO $_2$ - C_6H_4	1714
9k	3-NO $_2$ - C_6H_4	1715
9l	4-NO $_2$ - C_6H_4	1710

EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 2-ARYLIMINO-3N-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-4-THIAZOLIDINONES

[A] Synthesis of Bis-thiourea derivatives

A solution of *p*-anisidine (1.23g, 0.01 mol) in methanol and carbon disulfide was refluxed for 4 hrs., poured into ice water to afford the desired product. Excess of CS₂ was removed by placing the reaction mixture at room temp. for 2 hrs. The crude product was isolated and crystallised in methanol. Yield 80% m.p. 196°C. Anal Calcd. : C, 63.50%; H, 5.55%; N, 9.72%; Found : C, 62.48%; H, 5.53%; N, 9.70%.

Similarly other bithiourea derivatives have been prepared.

[B] Synthesis of 2-Arylimino-3N-aryl-4-thiazolidinone

A solution of 1,3-dianisyl thiourea (2.88 g. 0.01 mol) and chloroacetic acid in glacial acetic acid (20 ml) was refluxed with fused sodium acetate (1.25 g. 0.015 mol.) for 10 hrs. The reaction mixture was poured into ice water and glacial acetic acid was neutralised by sodium bicarbonate and crude product was isolated and crystallised from methanol. Yield 72% m. p. 98°C. Anal Calcd. C, 62.19%; H, 4.87%; N, 8.53%; Found : C, 62.20%; H, 4.90%; N, 8.51%.

Similarly, other 2-arylimino-3N-aryl-4-thiazolidinones were prepared.

[C] Synthesis of 2-Arylimino-3N-aryl-5-(2',7'-dichloroquinolin-3'-methinyl)-4-thiazolidinones

A mixture of 1-anisylimino-3-anisyl-5H-4-thiazolidinone (3.28 g., 0.01 mol), 2,7-dichloroquinolin-3-carboxaldehyde (2.26g. 0.01 mol) and fused sodium acetate (1.25 g. 0.01 mol) were refluxed in glacial acetic acid (25 ml) for 12 hrs. The reaction mixture was then poured into ice water. The crude mass was isolated and treated with sodium bisulfide. The solid thus obtained was filtered, washed,

dried and crystallised from DMF. Yield 72%, m.p. 284°C. Anal Calcd. for $C_{27}H_{19}N_3Cl_2O_3S$: C, 60.45%; H, 3.57%; N, 7.83%; Found : C, 60.43%; H, 3.58%; N, 7.55%.

TLC Solvent System: Acetone : Benzene (4:6), Visualizing Agent : Iodine.

Similarly other 4-thiazolidinone derivatives were prepared. The physical data along with infra red and spectral data are recorded in Table No. 7.

[D] Therapeutic evaluation of 1-Arylimino-3-aryl-5-(2',7'-dichloroquinolin-3'-methinyl)-4-thiazolidinones

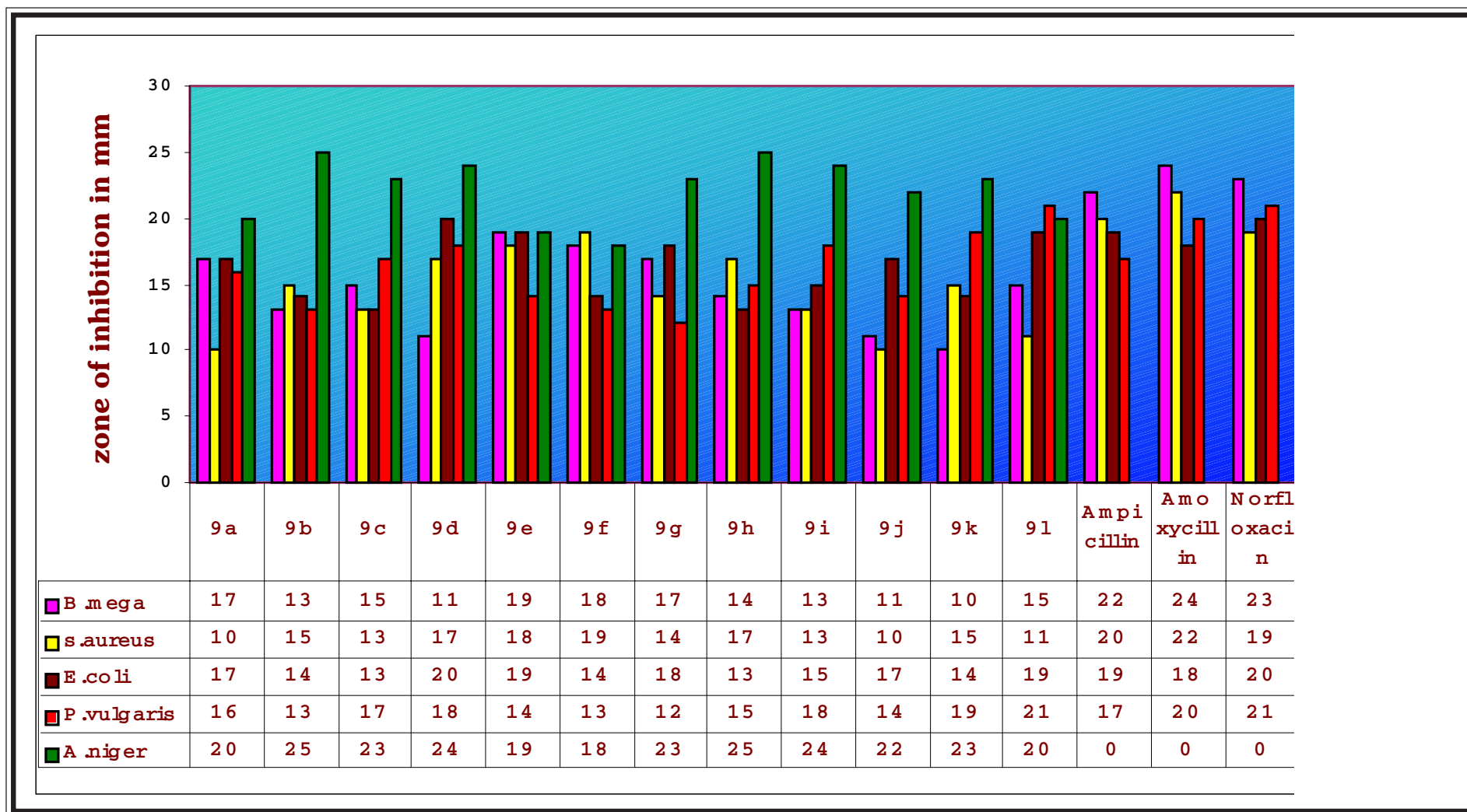
Antimicrobial testing was carried out as described in Part - I, Section-I (D). The zones of inhibition of the test solutions are recorded in Graphical Chart No. 9.

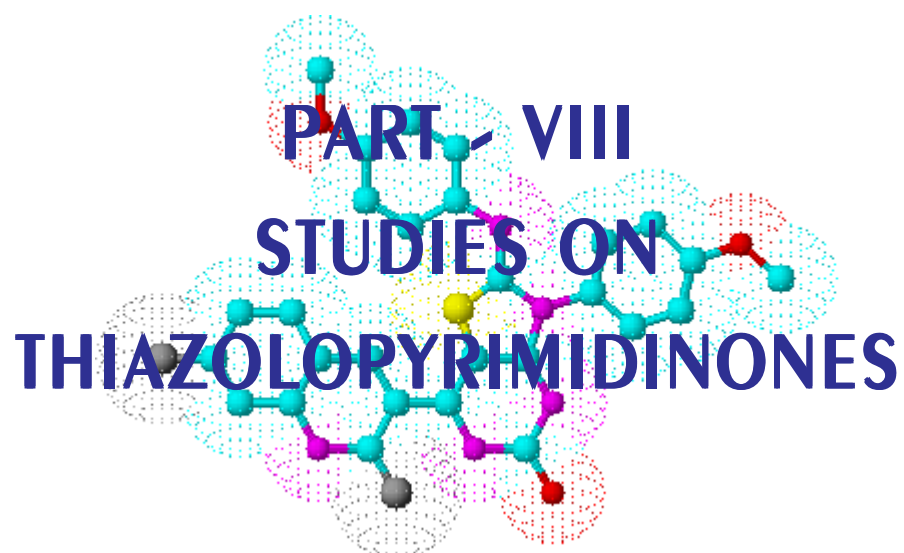
TABLE NO. 7 : PHYSICAL CONSTANTS OF 2-ARYLIMINO-3N-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-4-THIAZOLIDINONES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
9a	C ₆ H ₅ -	C ₂₅ H ₁₅ N ₃ Cl ₂ OS	476	272	0.878	59	8.82	8.55
9b	4-Cl-C ₆ H ₄ -	C ₂₅ H ₁₃ N ₃ Cl ₄ OS	545	264	0.495	63	7.71	7.57
9c	3,4-(Cl) ₂ -C ₆ H ₃ -	C ₂₅ H ₁₁ N ₃ Cl ₆ OS	614	258	0.485	67	6.84	6.59
9d	4-F-C ₆ H ₄ -	C ₂₅ H ₁₃ N ₃ Cl ₂ F ₂ OS	512	254	0.675	68	8.20	8.47
9e	2-OCH ₃ -C ₆ H ₄ -	C ₂₇ H ₁₉ N ₃ Cl ₂ O ₃ S	536	288	0.625	69	7.83	7.60
9f	4-OCH ₃ -C ₆ H ₄ -	C ₂₇ H ₁₉ N ₃ Cl ₂ O ₃ S	536	284	0.523	72	7.83	7.55
9g	2-CH ₃ -C ₆ H ₄ -	C ₂₇ H ₁₉ N ₃ Cl ₂ OS	504	294	0.439	70	8.33	8.65
9h	3-CH ₃ -C ₆ H ₄ -	C ₂₇ H ₁₉ N ₃ Cl ₂ OS	504	242	0.760	66	8.33	8.08
9i	4-CH ₃ -C ₆ H ₄ -	C ₂₇ H ₁₉ N ₃ Cl ₂ OS	504	234	0.563	69	8.33	8.19
9j	2-NO ₂ -C ₆ H ₄ -	C ₂₅ H ₁₃ Cl ₂ N ₅ O ₅ S	566	256	0.614	73	12.37	12.59
9k	3-NO ₂ -C ₆ H ₄ -	C ₂₅ H ₁₃ Cl ₂ N ₅ O ₅ S	566	280	0.512	75	12.37	12.62
9l	4-NO ₂ -C ₆ H ₄ -	C ₂₅ H ₁₃ Cl ₂ N ₅ O ₅ S	566	276	0.496	72	12.37	12.08

TLC Solvent System : Acetone : Benzene (4 : 6).

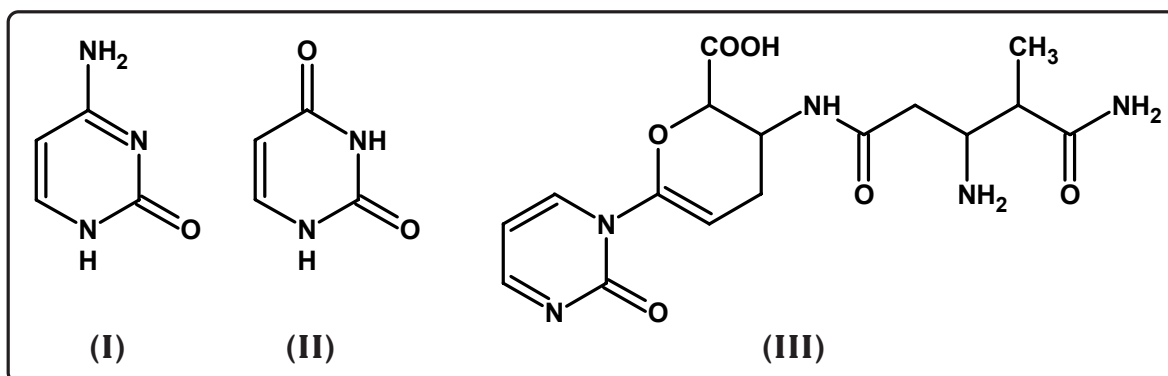
GRAPHICAL CHART NO. 9 : 2-ARYLIMINO-3N-ARYL-5-(2',7'-DICHLOROQUINOLIN-3'-METHINYL)-4-THIAZOLIDINONES





INTRODUCTION

Primidinones is very interesting class of heterocyclic compounds having wide range of applications in medicine and synthetic chemistry. Pyrimidine derivatives like cytosine (I), urecil (II) and blasticidin (III) occur widely in nature, have remarkable pharmaceutical importance because of their diverse physiological action. Several analogs of nucleic acids have been used as compounds that interfere with the synthesis and functioning of nucleic acids, an example is fluorouracil which has been used in cancer treatment.

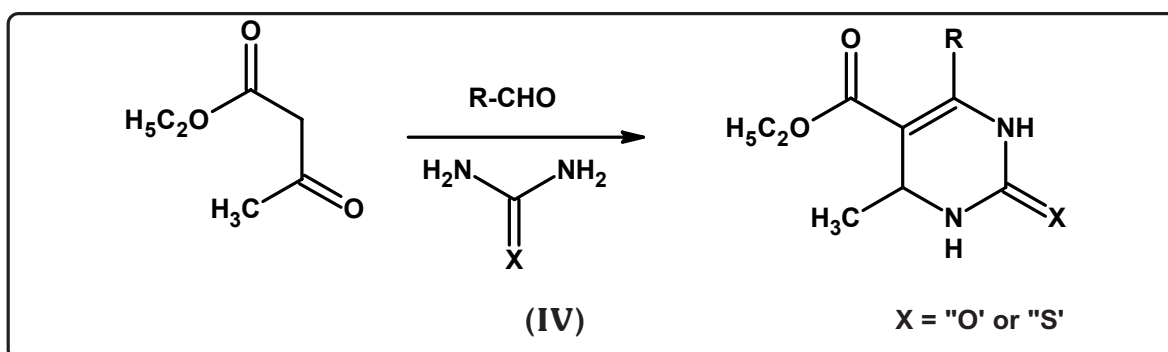


Despite considerable localization of π -electrons at the nitrogen atom of pyrimidinone, the ring system is still sufficiently aromatic to possess substantial stability. This has great advantage in the primary synthesis of pyrimidinones.

SYNTHETIC ASPECT

Various methods for the synthesis of pyrimidinones have been cited in the literature³⁴⁶⁻³⁴⁹.

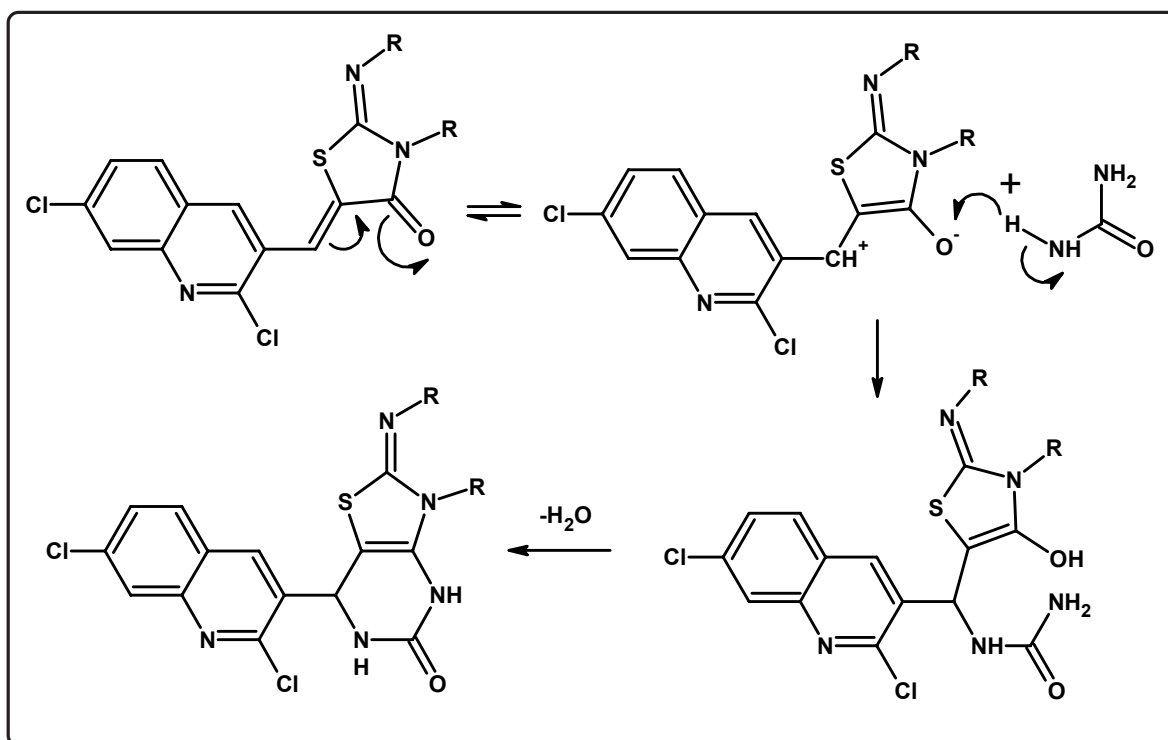
1. Rajiv Gupta and co-workers³⁵⁰ improved synthesis of 4-aryl-6-methyl-1,2,3,4-tetrahydro-pyrimidine-2-one/thione-5-carboxaldehyde by microwave irradiation.
2. Biginelli³⁵¹ investigated the reaction of an aromatic aldehyde with β -ketoester and urea or thiourea which yielded pyrimidine derivative (IV).



MECHANISM

Pyrimidinone derivatives are formed through cycloaddition process. The reaction proceeds through conjugated addition of urea to the α,β -unsaturated system. Here bond formation take place between N-atom of urea and C-atom of arylidene.

In the first step migration of electron takes place due to the more electronegativity of O-atom than C-atom. So carbon has positive charge while nitrogen atom loses proton so it acquires negative charge, with simultaneous removal of water.

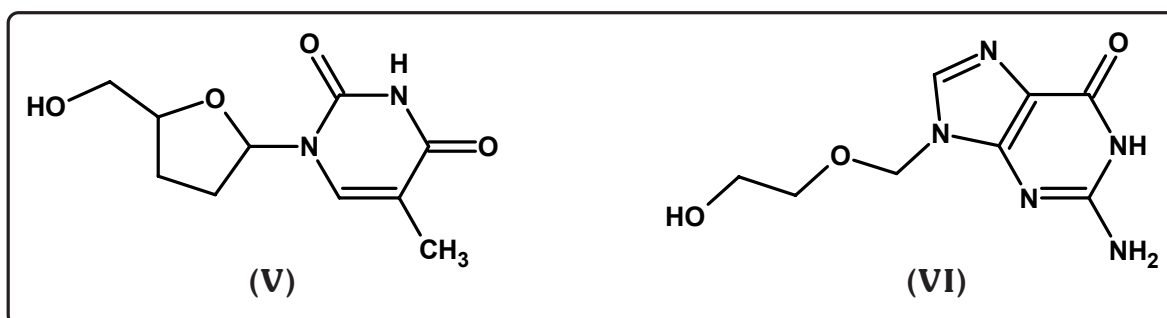


THERAPEUTIC IMPORTANCE

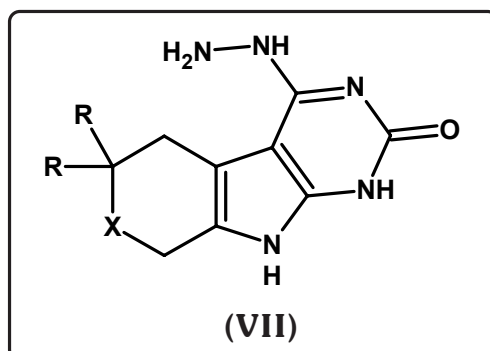
The pyrimidinone derivatives have been proven to be of great importance in exhibiting and enhancing the biological activities. The important biological activities shown by the pyrimidinones are as under:

1. Anticonvulsant³⁵²
2. Anthelmintics³⁵³
3. Cardiovascular^{354,355}
4. Anti HIV^{356,357}
5. Antihypertensive^{358,359}
6. Antiviral^{360,361}
7. Antitumor^{363,363}
8. Antibiotic³⁶⁴
9. Analgesic³⁶⁵
10. Antagonist³⁶⁶⁻³⁶⁸
11. Antiinflammatory³⁶⁹

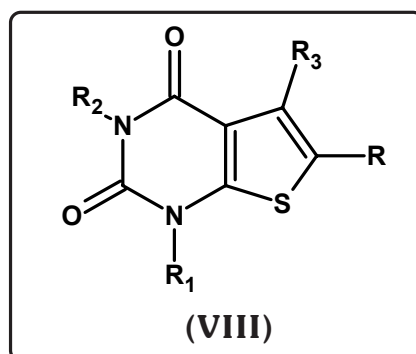
Azaryan and co-workers³⁷⁰ synthesised pyrimidine dione as antitumor agent. Some important drugs of this type have been produced such as zidovudine (AZT) (V) used in the treatment of AIDS is a close analogue of thymidine. A very highly useful drug for herpes virus infection is acyclovir (ACV) (VI) because of its high selectivity.



Anti HIV activity of pyrimidinone molecule has been reported by Watanabe and Harado³⁷¹. Paronikyan E.G. and co-workers³⁷² have prepared some new pyrimidinones of type (VII) and tested their biological activities.

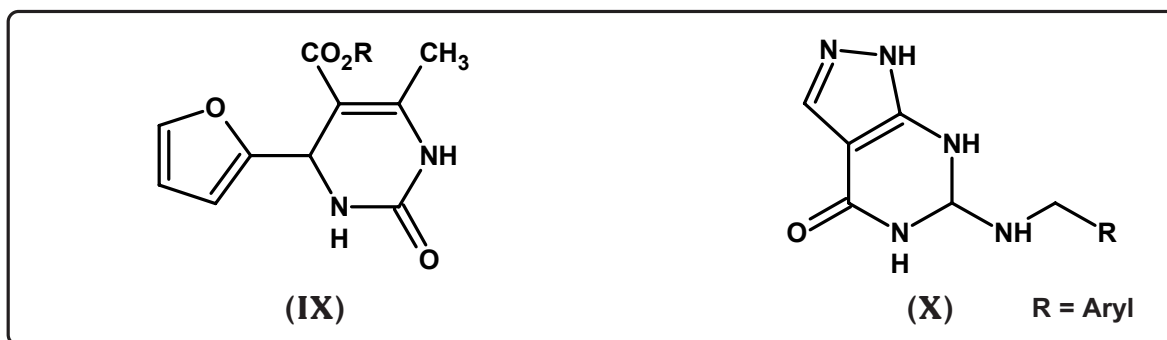


Antiviral, antifungal and insecticidal activity of pyrimidones was documented by M. M. Heravi and co-workers³⁷³. Sidler and Larson³⁷⁴ have reported pyrimidone derivatives useful as an α/A adrenergic receptor antagonists. Antiinflammatory agent bearing parent molecule pyrimidone has been documented by Modica and co-worker³⁷⁵. Cho et al.³⁷⁶ have prepared pyrimidinone derivative and reported their antiviral activity. Bantick J. and co-workers³⁷⁷ synthesised thienopyrimidinediones (VIII) and reported their antiasthmatic activity.



Baldev Kumar and co-workers³⁷⁸ have prepared a new series of substituted tetrahydropyrimidines and reported as calcium channel blockers. Antiinflammatory activity of some fused pyrimidinones have been reported by A. E. Galil et al.³⁷⁹ S. Raffaele and co-workers³⁸⁰ have synthesised some new pyrimidones and documented their anti HIV activity.

Hu Chun et al.³⁸¹ prepared some new tetrahydropyrimidone derivatives (IX) and reported them as calcium antagonist. A. Ali et al.³⁸² have synthesised pyrazolo pyrimidone derivatives (X) and documented their antimicrobial activity.



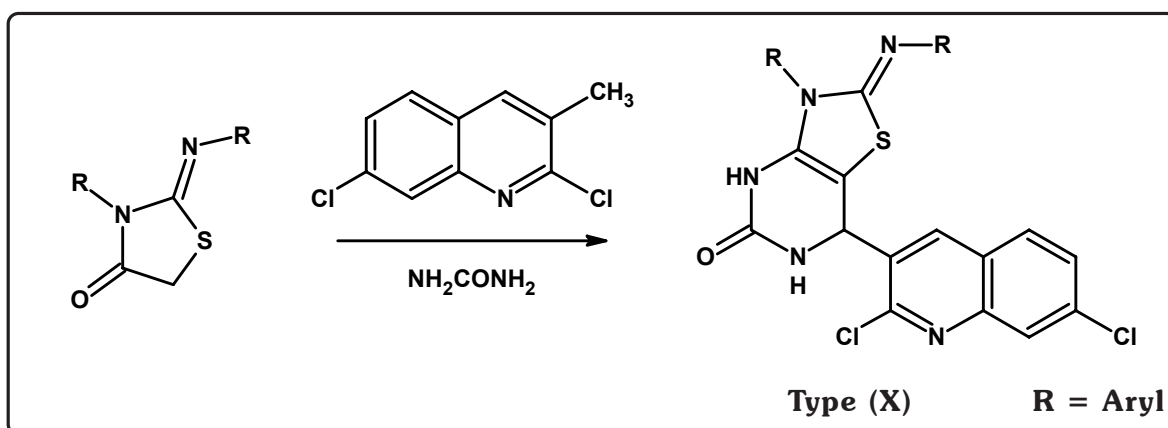
In view of the above observations, it was thought worth while to synthesise some new pyrimidones bearing 2,7-dichloroquinolin-3-carboxaldehyde nucleus, which has been described in the following section.

**SECTION - I : SYNTHESIS AND THERAPEUTIC EVALUATION OF
OF 7-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-
ARYLIMINO-3N-ARYLTHIAZOLIDINO[4,5-d]-
PYRIMIDINE-4,5,6,7-TETRAHYDRO-5-ONES**

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF 7-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-ARYLIMINO-3N-ARYLTHIAZOLIDINO [4,5-d]-PYRIMIDIN-4,5,6,7-TETRAHYDRO-5-ONES

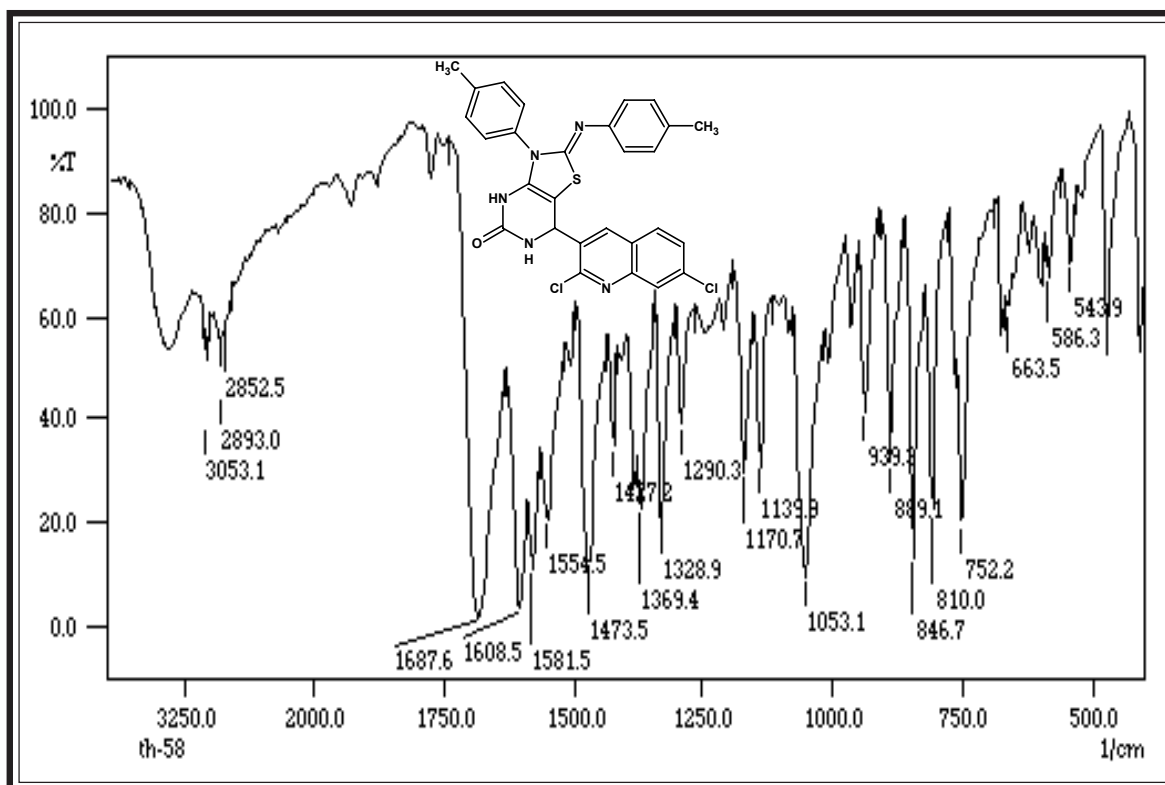
Pyrimidinones are found to have antimicrobial, anti-tumor, CNS depressant, cardiovascular activities. Keeping this in view, we planned to synthesise some new pyrimidinones of type (X). The desired pyrimidinones were synthesised using substituted 4-thiazolidinone with 2,7-dichloroquinolin-3-carboxaldehyde and urea.



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I Section-I (D).

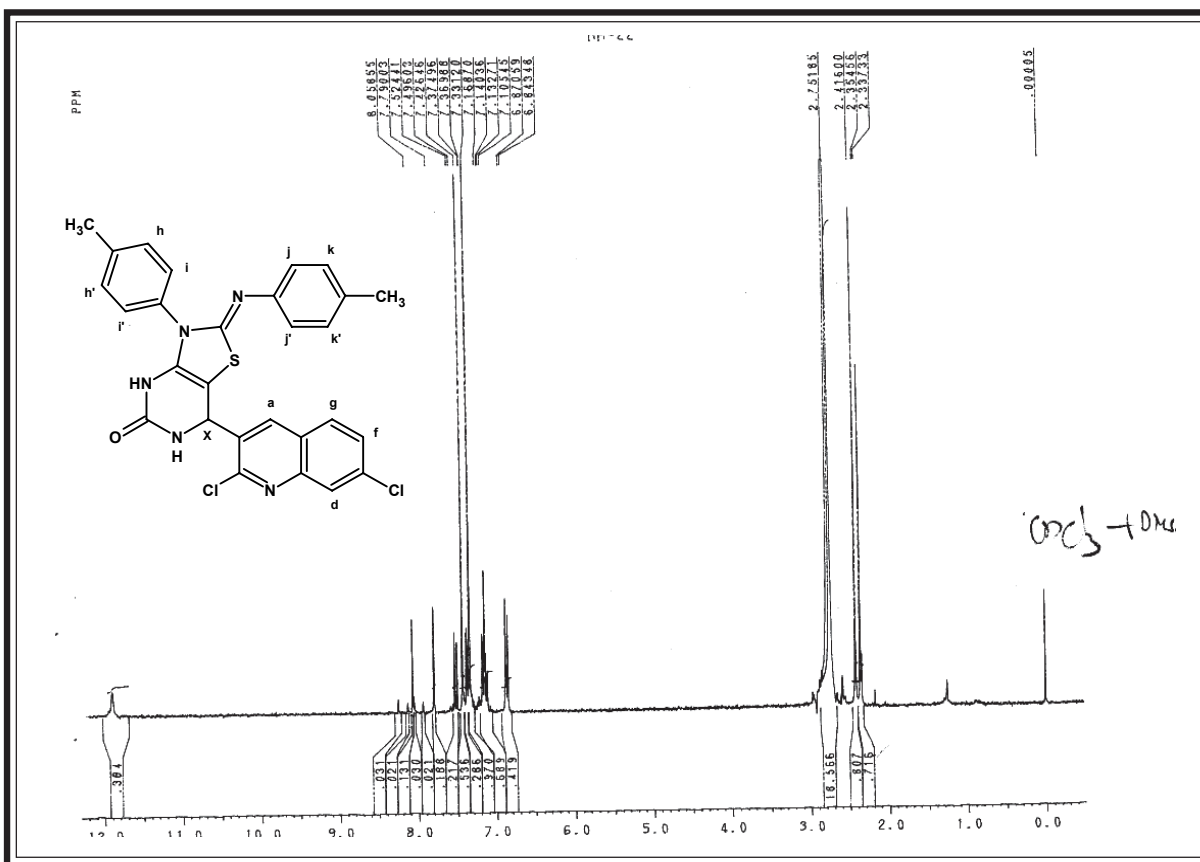
IR SPECTRAL STUDY OF 7-(2'-7'-DICHLOROQUINOLIN-3'-YL)-2-TOLYLIMINO-3N-TOLYLTHIAZOLIDINO-[4,5-d]-PYRIMIDINE-4,5,6,7-TETRAHYDRO-5-ONE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane	C – H str.	2893	2975-2850	419
	C – H def.	1369	1385-1370	"
Aromatic	C– H str.	3053	3080-3030	"
	C = C str.	1581	1585-1570	"
		1473	1520-1480	"
	C – H. def.	1053	1070-1000	420
		810	835-810	"
Quinoline moiety	C = N str.	1608	1612-1593	419
	C – Cl str.	752	750-700	"
Thiazolidine	C – N str.	1170	1220-1020	418
	C – S – C str.	663	700-600	"
Pyrimidinone	C = O str.	1687	1690-1630	"

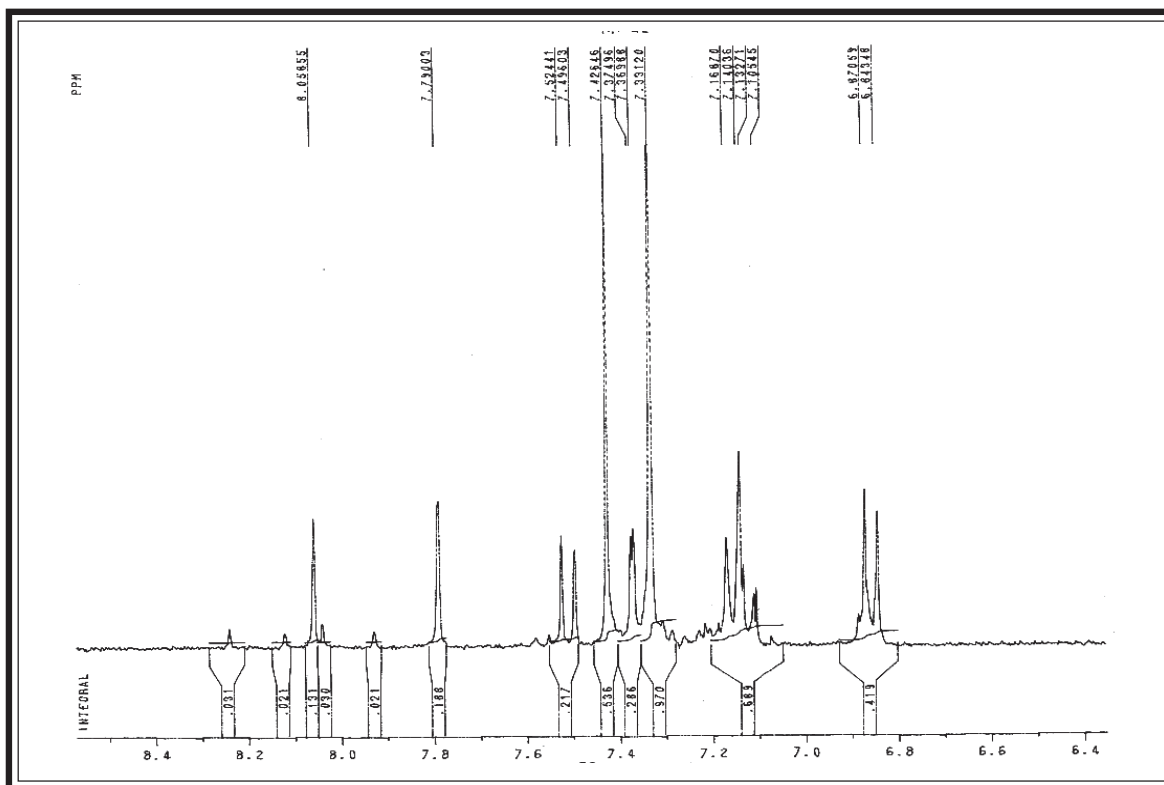
PMR SPECTRAL STUDY OF 7-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-(p-TOLYL)-IMINO-3N-(p-TOLYL)-THIAZOLIDINO-[4,5-d]-PYRIMIDINE-4,5,6,7-TETRAHYDRO-5-ONE



Internal Standard : TMS; Solvent : CDCl₃ ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	2.35	3H	singlet	-	Ar-CH ₃
2.	2.41	3H	singlet	-	Ar-CH ₃
3.	6.84-6.87	2H	singlet	J _{hi} 8.13	Ar-H h, h'
4.	7.11-7.17	3H	multiplate	-	Ar-H i, i', f
5.	7.33	4H	singlet	-	Ar-H j, j', k, k'
6.	7.37-7.38	1H	doublet	J _{df} 1.52	Ar-Hd
7.	7.49-7.52	1H	doublet	J _{gf} 8.51	Ar-Ha
8.	7.79	1H	singlet	-	Ar-Ha
9.	8.06	1H	singlet	-	Ar-HX

EXPANDED AROMATIC REGION

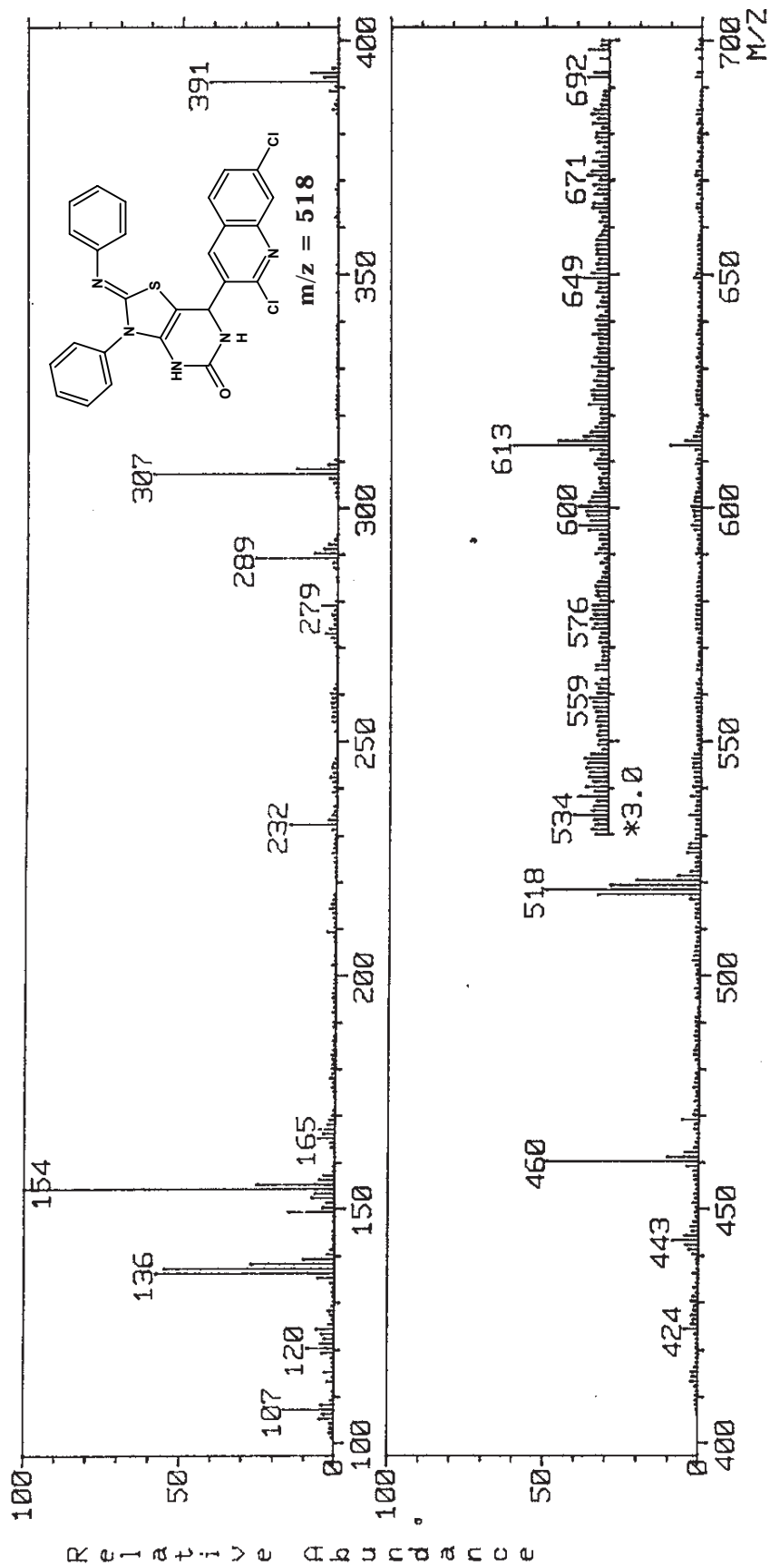


IR SPECTRAL DATA OF 7-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-ARYLIMINO-3N-ARYL THIAZOLIDINO [4,5,d]-PYRIMIDINE-4,5,6,7-TETRAHYDRO-5-ONES

Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	C=O str.
10a	C_6H_5	1687
10b	4-Cl- C_6H_4	1689
10c	3,4-(Cl) $_2$ - C_6H_3	1690
10d	4-F- C_6H_4	1691
10e	2-OCH $_3$ - C_6H_4	1689
10f	4-OCH $_3$ - C_6H_4	1689
10g	2-CH $_3$ - C_6H_4	1689
10h	3-CH $_3$ - C_6H_4	1689
10i	4-CH $_3$ - C_6H_4	1687
10j	2-NO $_2$ - C_6H_4	1691
10k	3-NO $_2$ - C_6H_4	1696
10l	4-NO $_2$ - C_6H_4	1680

MASS SPECTRUM Data File: 3EJL25F
 Sample: APS-10 DR H H PAREKH,RAJKOT #6262
 RT 0.36" FAB(Pos.) GC 1.4c BP: m/z 154.0000 Int. 31.1530 Lv 0.00
 Scan# (4 to 5)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF 7-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-ARYLIMINO-3N-ARYLTHIAZOLIDINO [4,5-d]-PYRIMIDINE-4,5,6,7-TETRAHYDRO-5-ONES

[A] Synthesis of Bis-thiourea derivatives

See Part - VII, Section - I (A)

[B] Synthesis of 2-Arylimino-3-aryl-5H-4-thiazolidinones

See Part - VII, Section I (B)

[C] Synthesis of 7-(2',7'-Dichloroquinolin-3'-yl)-2-arylimino-3N-aryl thiazolidino [4,5 d]-pyrimidin-4,5,6,7-tetrahydro-5-ones

A mixture of 2-anisylimino-3-anisyl-5H-4-thiazolidinone (3.28g, 0.01mol), 2,7-dichloroquinolin-3-carboxaldehyde (2.26g, 0.01 mol) and urea (0.56g, 0.01 mol) in methanol was added a few drops of conc. HCl and heated under reflux for 4 hrs. The product isolated was filtered and crystallised from DMF. Yield 61%, m.p. 274°C Anal Calcd. for $C_{28}H_{21}N_5Cl_2O_3S$ required : C, 56.57%, H, 3.56%, N, 12.11%; found : C, 56.59%, H, 3.59% N, 12.44%.

TLC Solvent System : Acetone : Benzene (1.2 :8.8), Visualizing Agent : Iodine.

Similarly other pyrimidinones were prepared. The physical data along with infra red spectral data are reported in Table No. 8.

[D] Therapeutic evaluation of 7-(2',7'-Dichloroquinolin-3'-yl)-2-arylimino-3N-arylthiazolidino [4,5 d]-pyrimidin-4,5,6,7-tetrahydro-5-ones

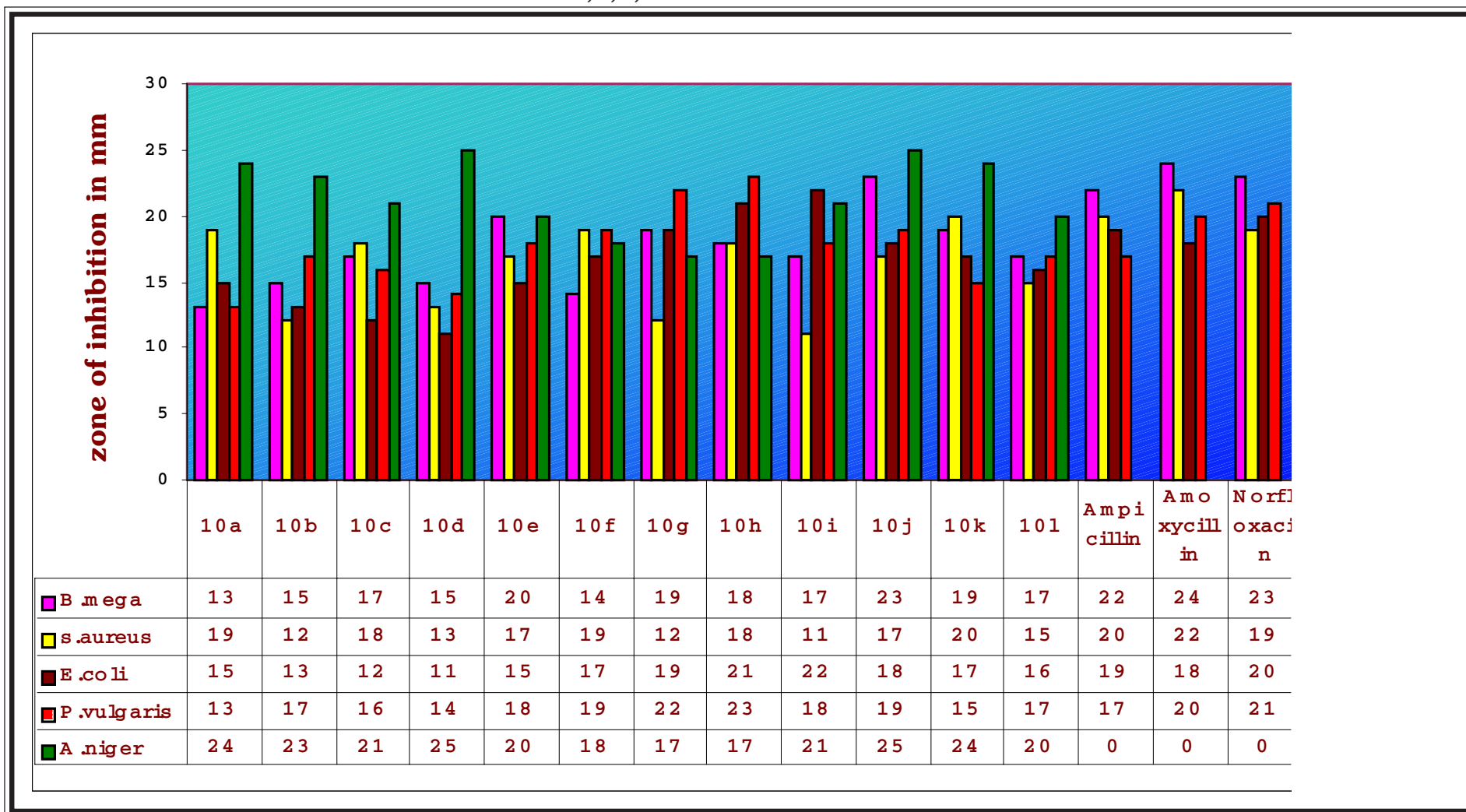
Antimicrobial testing was carried out as described in Part-I, Section-I (D). The zones of inhibition of the test solutions are recorded in Graphical Chart No. 10.

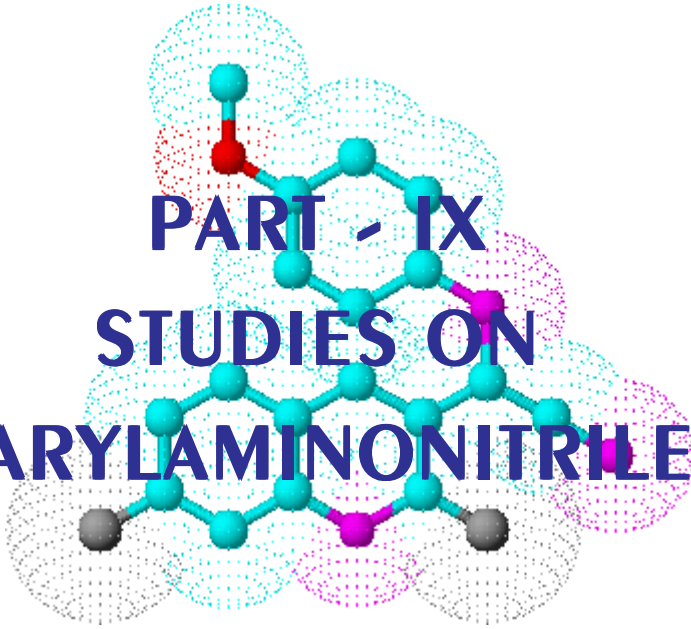
TABLE NO. 8 : PHYSICAL CONSTANTS OF 7-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-ARYLIMINO-3N-ARYL THIAZOLIDINO [4,5-d]-PYRIMIDINE-4,5,6,7-TETRAHYDRO-5-ONES

Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
10a	C ₆ H ₅ -	C ₂₆ H ₁₇ N ₅ Cl ₂ OS	318	176	0.540	68	13.51	13.80
10b	4-Cl-C ₆ H ₄ -	C ₂₆ H ₁₅ N ₅ Cl ₄ OS	587	208	0.629	60	11.92	11.63
10c	3,4-(Cl) ₂ -C ₆ H ₃ -	C ₂₆ H ₁₃ N ₅ Cl ₆ OS	656	240	0.551	63	10.67	10.87
10d	4-F-C ₆ H ₄ -	C ₂₆ H ₁₅ N ₅ Cl ₂ F ₂ OS	554	252	0.721	65	12.63	12.57
10e	2-OCH ₃ -C ₆ H ₄ -	C ₂₈ H ₂₁ N ₅ Cl ₂ O ₃ S	578	268	0.650	62	12.11	11.88
10f	4-OCH ₃ -C ₆ H ₄ -	C ₂₈ H ₂₁ N ₅ Cl ₂ O ₃ S	578	274	0.658	61	12.11	12.44
10g	2-CH ₃ -C ₆ H ₄ -	C ₂₈ H ₂₁ N ₅ Cl ₂ OS	546	256	0.545	60	12.82	12.53
10h	3-CH ₃ -C ₆ H ₄ -	C ₂₈ H ₂₁ N ₅ Cl ₂ OS	546	260	0.591	65	12.82	12.99
10i	4-CH ₃ -C ₆ H ₄ -	C ₂₈ H ₂₁ N ₅ Cl ₂ OS	546	262	0.623	69	12.82	12.67
10j	2-NO ₂ -C ₆ H ₄ -	C ₂₆ H ₁₅ N ₇ Cl ₂ O ₅ S	608	276	0.680	70	16.12	16.33
10k	3-NO ₂ -C ₆ H ₄ -	C ₂₆ H ₁₅ N ₇ Cl ₂ O ₅ S	608	284	0.730	76	16.12	16.49
10l	4-NO ₂ -C ₆ H ₄ -	C ₂₆ H ₁₅ N ₇ Cl ₂ O ₅ S	608	242	0.494	72	16.12	15.87

TLC Solvent System : Acetone : Benzene (1.2 : 8.8).

GRAPHICAL CHART NO. 10 : 7-(2',7'-DICHLOROQUINOLIN-3'-YL)-2-ARYLIMINO-3N-ARYL-THIAZOLIDINO[4,5-d]PYRIMIDINE-4,5,6,7-TETRAHYDRO-5-ONES





PART - IX
STUDIES ON
 α -ARYLAMINONITRILES

INTRODUCTION

The nitrile derivatives possess wide range of therapeutic activities such as antipyretic, analgesic, antiseptic, antimalarial, anticonvulsant and antimicrobial etc. The first synthesis of nitriles has been reported in 1832 by Wohler and Liebig^{383,384} and in 1834 by Pelouze³⁸⁵. Because of their high toxicity, nitriles are much used in agricultural field. The nitriles are very useful intermediate for various products such as acrylonitrile for plastics, synthetic rubber, fibers and phthalonitriles for a dye stuff.

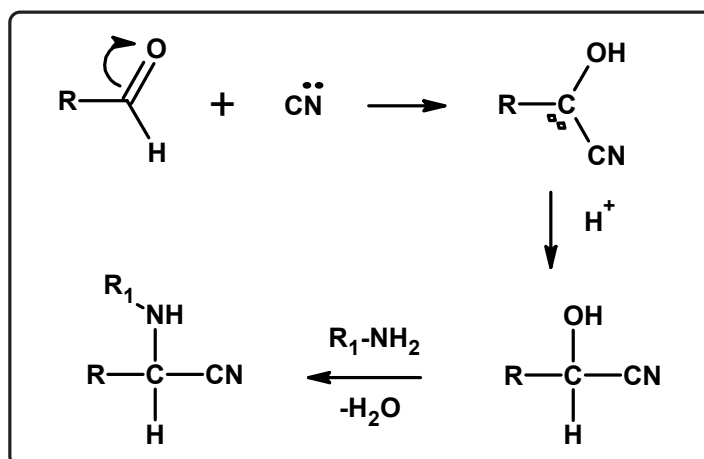
SYNTHETIC ASPECT

A numerous methods for the preparation of nitriles have been reviewed by David Mowry³⁸⁶. Some current methods are as under:

- (i) From halides using NaCN, Al₂O₃³⁸⁷
- (ii) From alkyl halides using KCN, tetraalkylammonium salt and water in trace³⁸⁸
- (iii) Preparation of metathesis³⁸⁹
- (iv) Dehydrating amides using POCl₃³⁹⁰
- (v) The pyrolysis of Schiff's base³⁹¹
- (vi) A practical method for the preparation of nitriles from primary amines under microwave irradiation, has been reported³⁹²⁻³⁹⁴

MECHANISM

The mechanism of nitrile is shown as under.



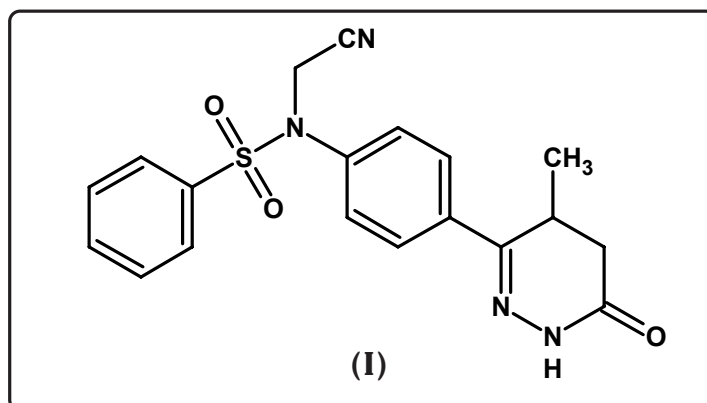
Reaction between $\text{C}\ddot{\text{N}}$ and aldehyde is a type of nucleophilic substitution reaction. From the above reaction, it can be seen that a nucleophile ($\text{C}\ddot{\text{N}}$) attacks on the carbonyl carbon of aldehyde and yields cyanohydrin which reacts with amine to yield nitrile derivatives.

THERAPEUTIC IMPORTANCE

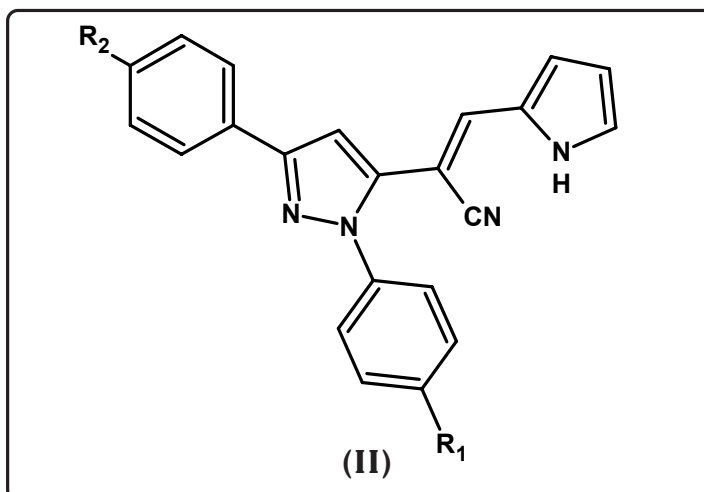
Nitrile shows various biological activities such as

1. Herbicidal and Viricidal³⁹⁵
2. Central nervous stimulants³⁹⁶
3. Antihypotoxic³⁹⁷
4. Antiinflammatory³⁹⁸
5. Antihypertensive³⁹⁹
6. Fungicidal⁴⁰⁰
7. Antimicrobial⁴⁰¹
8. Pesticidal⁴⁰²
9. Antiarrhythmic⁴⁰³

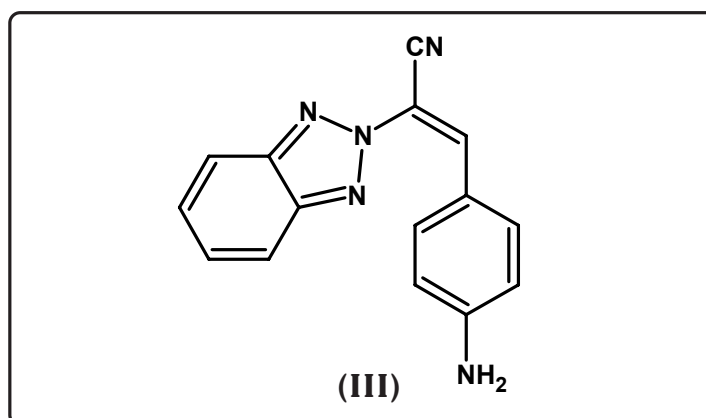
Nitriles with fused pyridine ring were reported as ulcer inhibitor⁴⁰⁴. The 3'-cyanophenyl alkanamines were found to be sympathomimetic drugs⁴⁰⁵. Cardenolide nitriles demonstrated moderate biological activity in rats (Cather Method)⁴⁰⁶. The benzylidene derivatives of maleic acid dinitriles showed 98% effectiveness against tetranychus urticae⁴⁰⁷. Sabani et al.⁴⁰⁸ have reported nitriles as a refrigeration lubricating oil. Nobuyuki and co-workers⁴⁰⁹ have described some new nitriles (I) as tumor necrosis factor(TNF) production inhibitors.



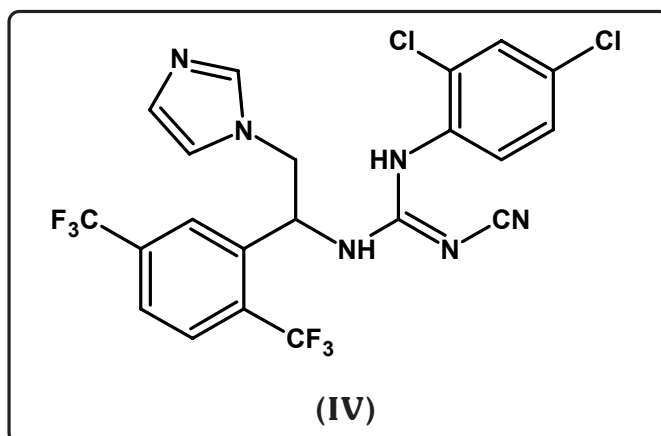
Kobayashi et al.⁴¹⁰ have synthesised new derivatives of nitrile. Peterson I. A. and co-workers⁴¹¹ have studied antiapoptotic activity of some new amino nitriles. A series of nitriles (II) as novel antioxidants was synthesised by Parmar V. S. and co-workers⁴¹².



Parikh et al.⁴¹³ prepared acetonitrile derivatives and documented as bactericidal and fungicidal. Parlo S. and co-workers⁴¹⁴ prepared some new acrylonitrile derivatives (III) and tested their antitubercular activity.



Nosyruva et al.⁴¹⁵ have synthesised nitrile derivatives and tested their antimicrobial activity. Catherine M. and co-workers⁴¹⁶ synthesised some new iminonitriles and reported as thromboxane receptor antagonist. Cyanoguanidine derivative (IV) has been synthesised and reported as inhibitors of mitochondrial F₁FO₁ ATPase by A. Karnails et al.⁴¹⁷



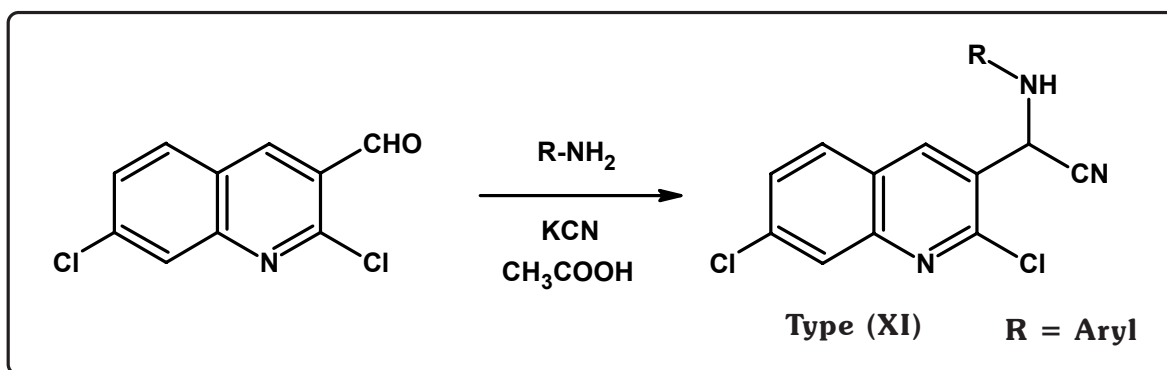
The synthesis of nitriles is of current interest owing to their enormous occurrence in biologically active derivatives. Hence, considerable attention has been focused on the study of pharmaceutically important nitriles bearing quinoline moiety as described below.

SECTION - I : SYNTHESIS AND THERAPEUTIC EVALUATION OF α -ARYLAMINO- α -2,7-DICHLOROQUINOLIN-3-YL-ACETONITRILES

SECTION - I

SYNTHESIS AND THERAPEUTIC EVALUATION OF α -ARYLAMINO- α -2,7-DICHLOROQUINOLIN-3-YL-ACETONITRILES

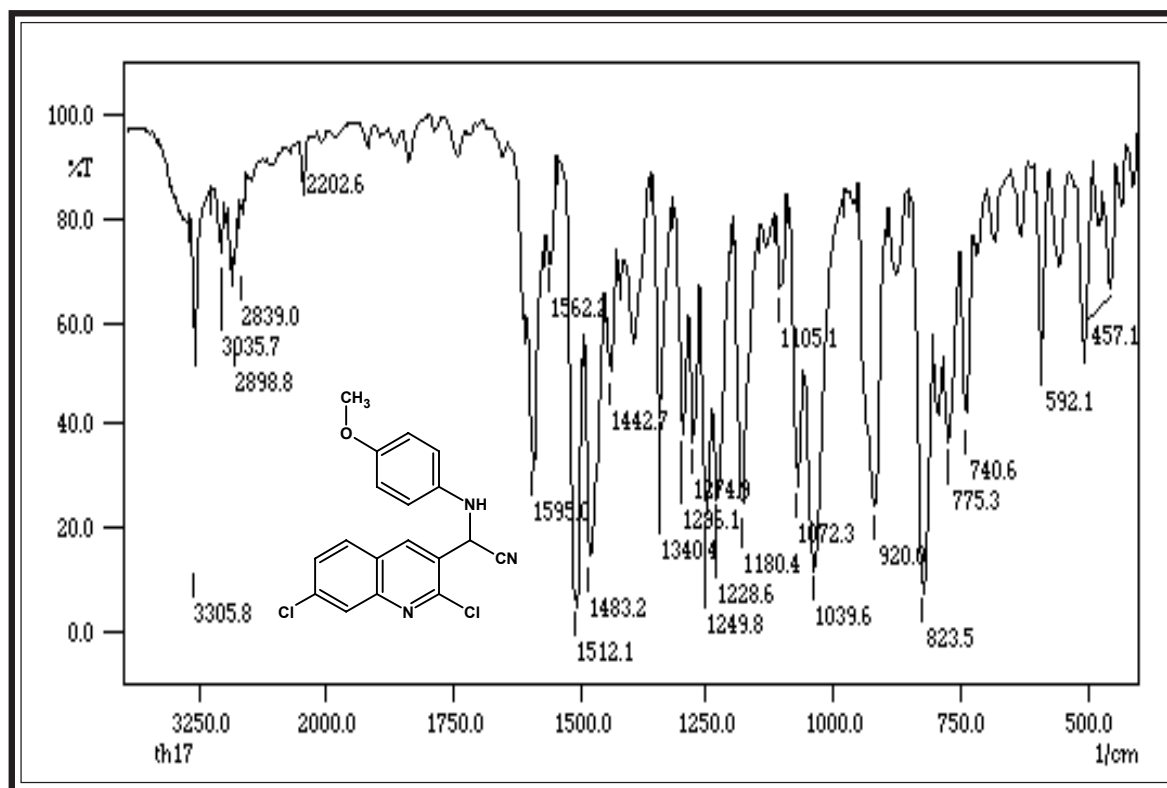
Nitrile derivatives have received considerable attention in recent years due to their wide spectrum of biological activities. Led by these considerations, we have designed some nitriles of type (XI). The technique applied to get the desired compounds involved the condensation of different amines with 2,7-dichloroquinolin-3-carboxaldehyde using KCN in presence of glacial acetic acid.



The constitution of the synthesised products have been characterised using elemental analyses, infra red and ^1H nuclear magnetic resonance spectroscopy and further supported by mass spectrometry.

The products have been screened for their *in vitro* biological assay like antibacterial activity towards Gram positive and Gram negative bacterial strains and antifungal activity towards *Aspergillus niger* at a concentration of 40 μg . The biological activities of the synthesised compounds have been compared with standard drugs. The details have been cited in Part-I, Section-I (D).

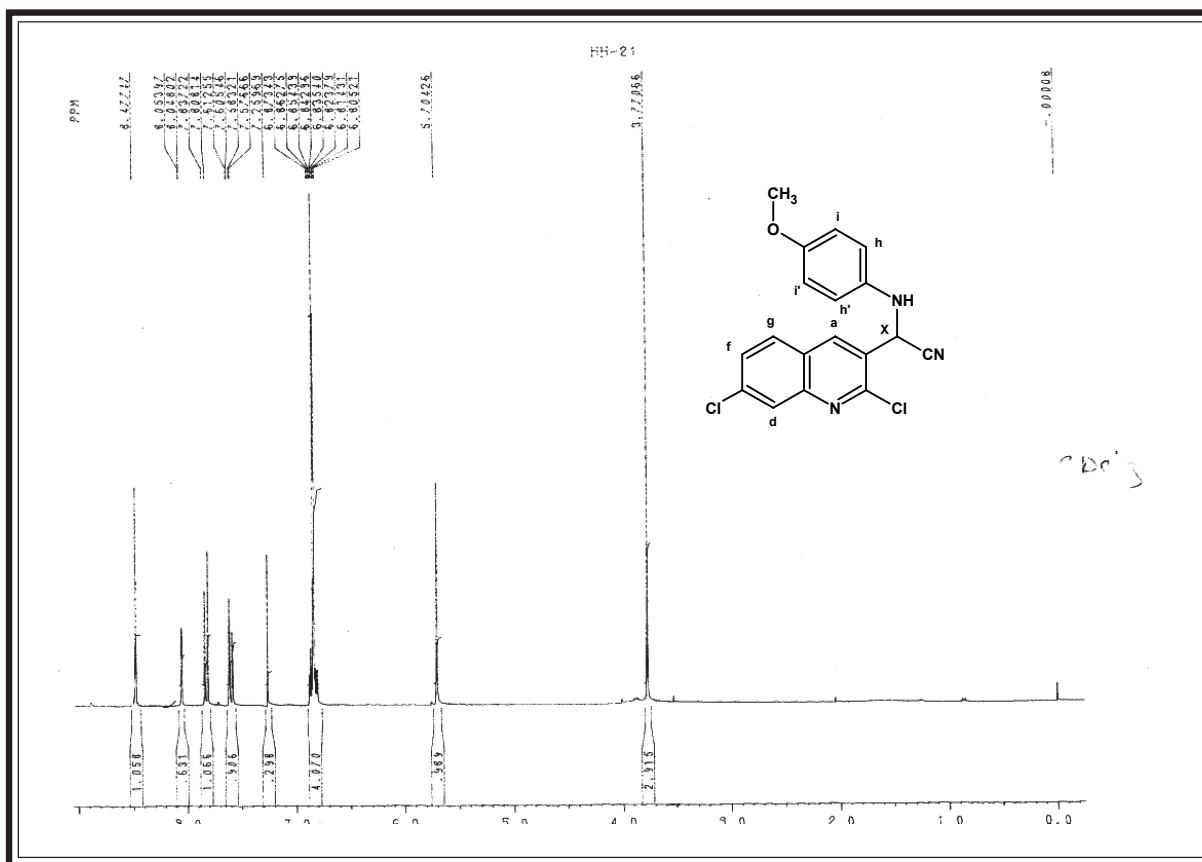
IR SPECTRAL STUDY OF α -p-ANISYLAMINO- α -2'-7'-DICHLOROQUINOLIN-3-YL-ACETONITRILE



Instrument : SHIMADZU-FT-IR 8400-Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc.)

Type	Vibration mode	Frequency in cm^{-1}		Ref.
		Observed	Reported	
Alkane -CH ₃	C - H str. (asym)	2925	2975-2950	419
	C - H str. (sym)	2880	2880-2850	"
	C - H def. (asym)	1442	1475-1450	"
	C - H def. (sym)	1340	1385-1370	"
Aromatic	C- H str.	3010	3080-3030	"
	C = C str.	1562	1585-1570	"
	C - H.i.p. def.	1105	1125-1090	420
	.	1072	1070-1000	"
	C - H. o.o.p def.	823	835-810	"
	N - H str.	3305	3400-3200	"
Quinoline moiety	C = N str.	1595	1612-1593	419
	C - Cl str.	740	750-700	"
Nitrile	C \equiv N str.	2202	2260-2200	421
Ether	C - O - C str.	1228	1275-1200	"

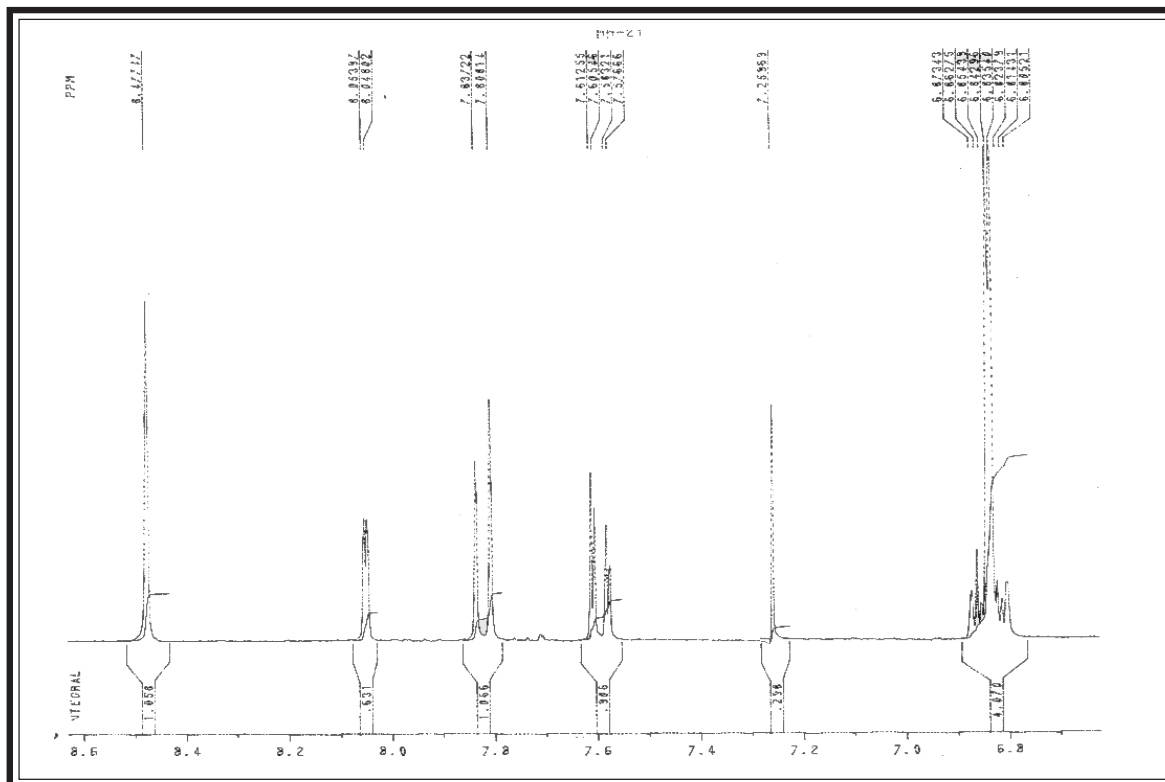
PMR SPECTRAL STUDY OF α -p-ANISYLAMINO- α -2,7-DICHLOROQUINOLIN-3-YL-ACETONITRILE



Internal Standard : TMS; Solvent : CDCl₃ ; Instrument : BRUKER Spectrometer (300 MHz)

Signal No.	Signal Position (δ ppm)	Relative No. of Protons	Multiplicity	J. Value in Hz	Inference
1.	3.77	3H	singlet	-	Ar-OCH ₃
2.	5.70	1H	singlet	-	-CHX
3.	6.81-6.87	4H	multiplate	-	Ar-H h,h',i,i'
4.	7.58-7.61	1H	double doublet	J _{fg} 8.64 J _{fd} 1.97	Ar-Hf
5.	7.81-7.84	1H	doublet	J _{gf} 8.72	Ar-Hg
6.	8.48-7.54	1H	doublet	J _{df} 1.79	Ar-Hd
7.	8.48	1H	singlet	-	Ar-Ha

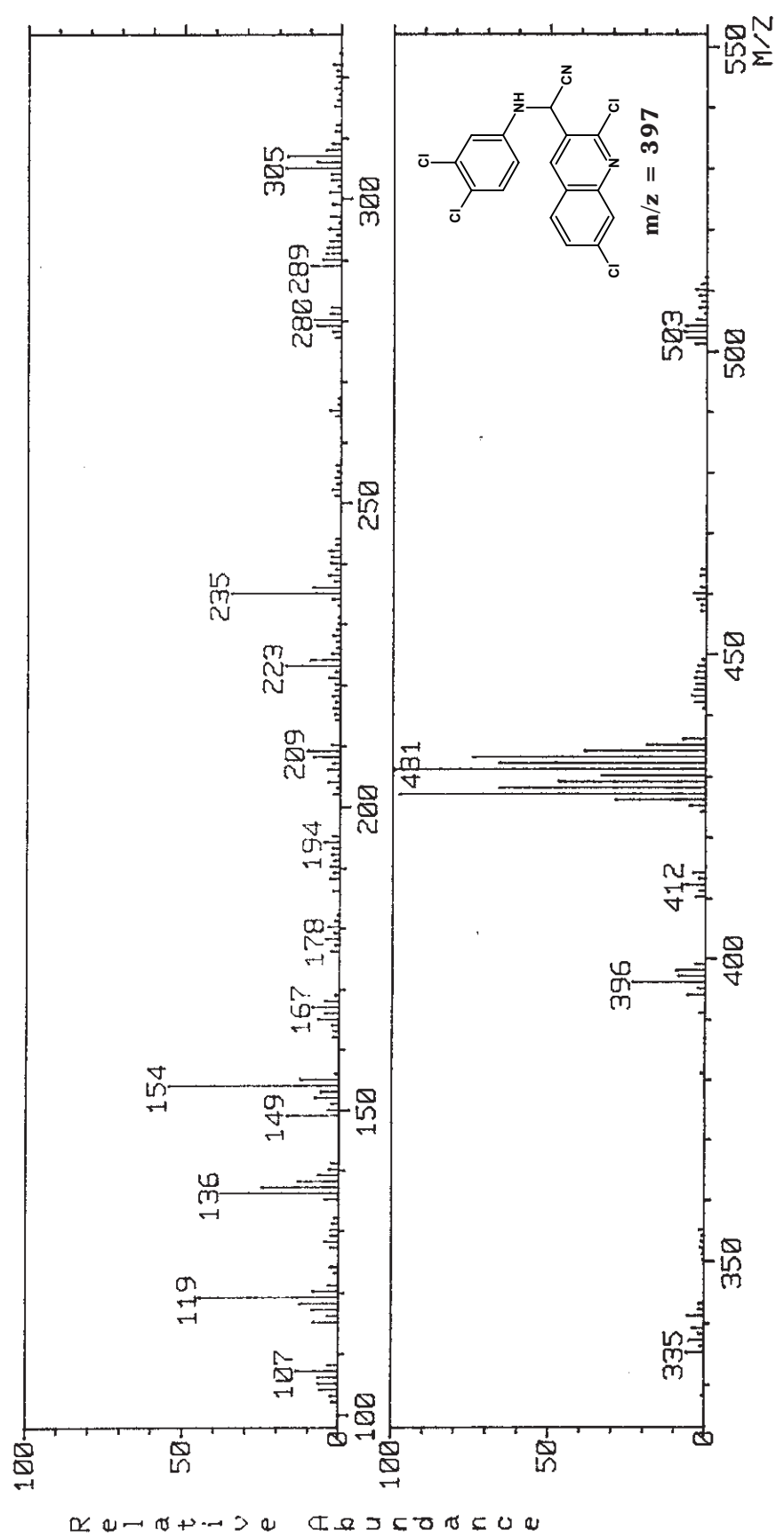
EXPANDED AROMATIC REGION


IR SPECTRAL DATA OF α -ARYLAMINO- α -2,7-DICHLOROQUINOLIN-3-YL-ACETONITRILES

 Instrument : SHIMADZU-FT-IR-8400 Spectrophotometer; Frequency range : 4000-400 cm^{-1} (KBr disc)

Sr. No.	R	C \equiv N str.
11a	C ₆ H ₅	2198
11b	2-Cl-C ₆ H ₄	2196
11c	3-Cl-C ₆ H ₄	2200
11d	4-Cl-C ₆ H ₄	2002
11e	2,6-(Cl) ₂ -C ₆ H ₃	2202
11f	3,4-(Cl) ₂ -C ₆ H ₃	2210
11g	3-Cl,4-F-C ₆ H ₄	2203
11h	4-F-C ₆ H ₄	2200
11i	2-OCH ₃ -C ₆ H ₄	2202
11j	4-OCH ₃ -C ₆ H ₄	2202
11k	2-CH ₃ -C ₆ H ₄	2202
11l	4-CH ₃ -C ₆ H ₄	2200
11m	2-NO ₂ -C ₆ H ₄	2201
11n	3-NO ₂ -C ₆ H ₄	2200
11o	4-NO ₂ -C ₆ H ₄	2200

MASS SPECTRUM Data File: 3EJL25V 25-JUL- 3 12:25
Sample: HHA-7 DR N A CHAUHAN, RAJKOT #6204
RT 0.12" FAB(Pos.) GC 1.4c BP: m/z 431.0000 Int. 6.0541 Lv 0.00
Scan# (1 to 3)



EXPERIMENTAL

SYNTHESIS AND THERAPEUTIC EVALUATION OF α -ARYLAMINO- α -2,7-DICHLOROQUINOLIN-3-YL-ACETONITRILES

[A] Synthesis of 2,7-Dichloroquinolin-3-carboxaldehydes

See, Part - I, Section - I (B).

[B] Synthesis of α -Arylamino- α -2,7-dichloroquinolin-3-yl-acetonitriles

2,7-Dichloroquinolin-3-carboxaldehyde (2.26 g, 0.01 mol) dissolved in ethanol (10 ml) was added to potassium cyanide (6.4 g, 0.1 mol) dissolved in water (8 ml.) followed by glacial acetic acid (12 ml). The contents were then stirred for 5 mins. to form aldehyde cyanohydrine and kept at 0°C. *p*-Anisidine (1.23 g, 0.01 mol) in ethanol (10 ml) was added to the reaction mixture, contents were kept at room temperature for 24 hrs. and poured into ice. The resulting solid was crystallised from ethanol. Yield 50%, m.p. 126°C. Anal Calcd. for $C_{18}H_{13}N_3Cl_2O$ required : C, 60.35%; H, 3.66%; N, 11.73%; found : C, 60.38%; H, 3.63%; N, 11.71%.

TLC Solvent system : Acetone : Benzene (2.8 : 7.2), Visualizing Agent : Iodine.

Similarly other α -arylamino- α -2,7-dichloroquinolin-3-yl-acetonitriles were prepared. The physical data along with infra red spectral data are reported in Table No. 9.

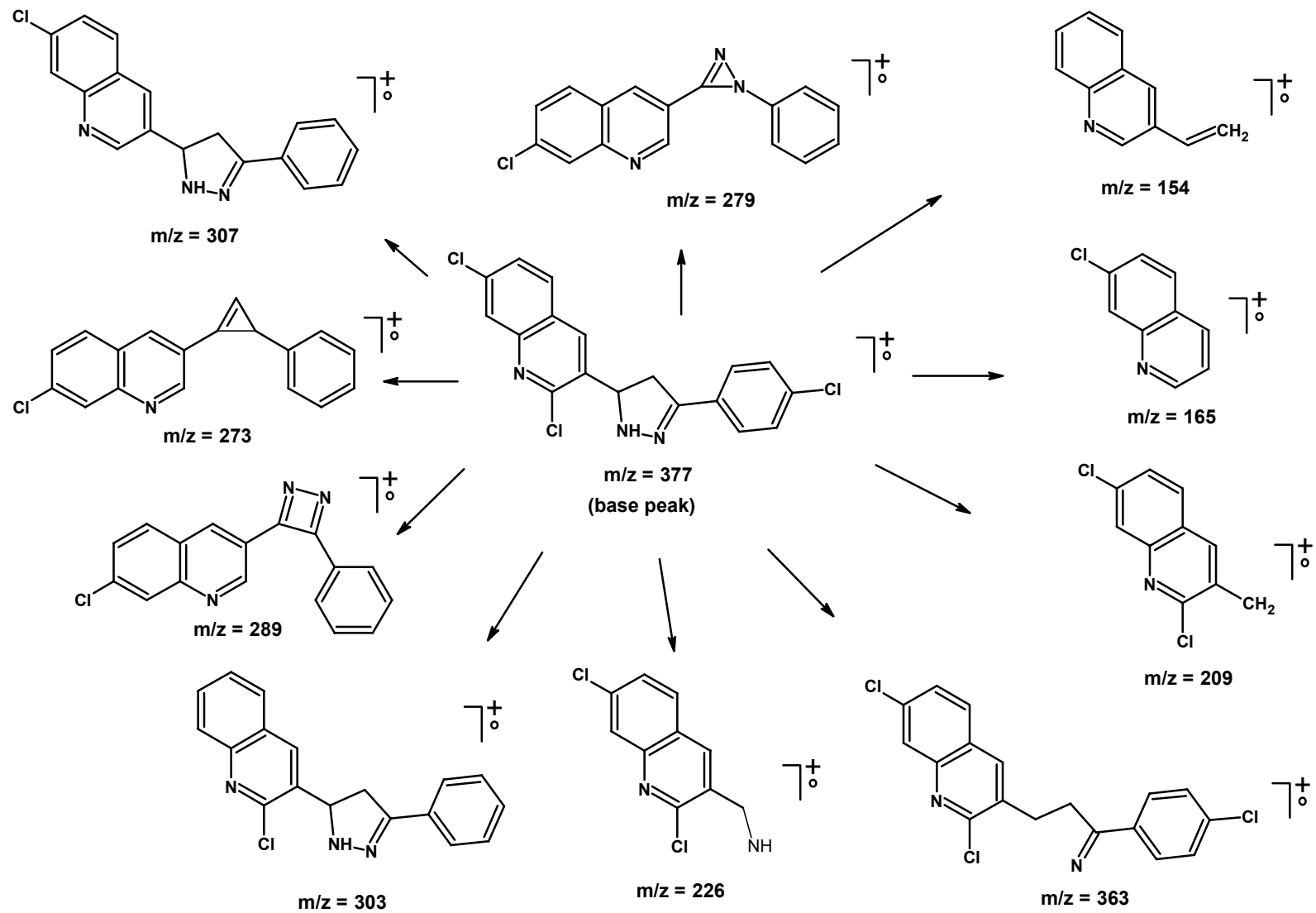
[C] Therapeutic evaluation of α -Arylamino- α -2,7-dichloroquinolin-3-yl-acetonitriles

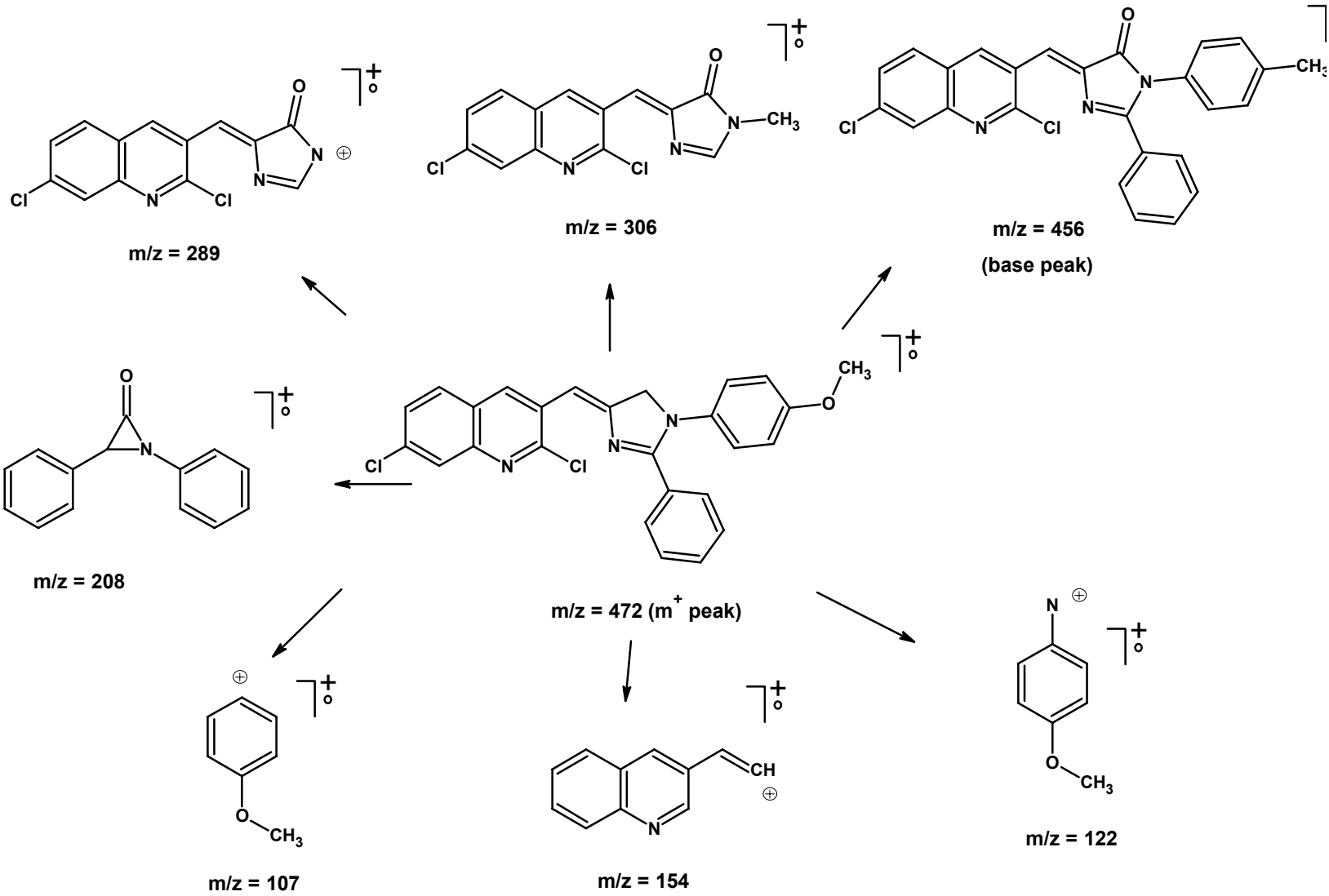
Antimicrobial testing was carried out as described in Part - I, Section - I (D). The zones of inhibition of the test solutions are recorded in Graphical Chart No. 11.

TABLE NO. 9 : PHYSICAL CONSTANTS OF α -ARYLAMINO- α -2,7-DICHLOROQUINOLIN-3-YL-ACETONITRILES

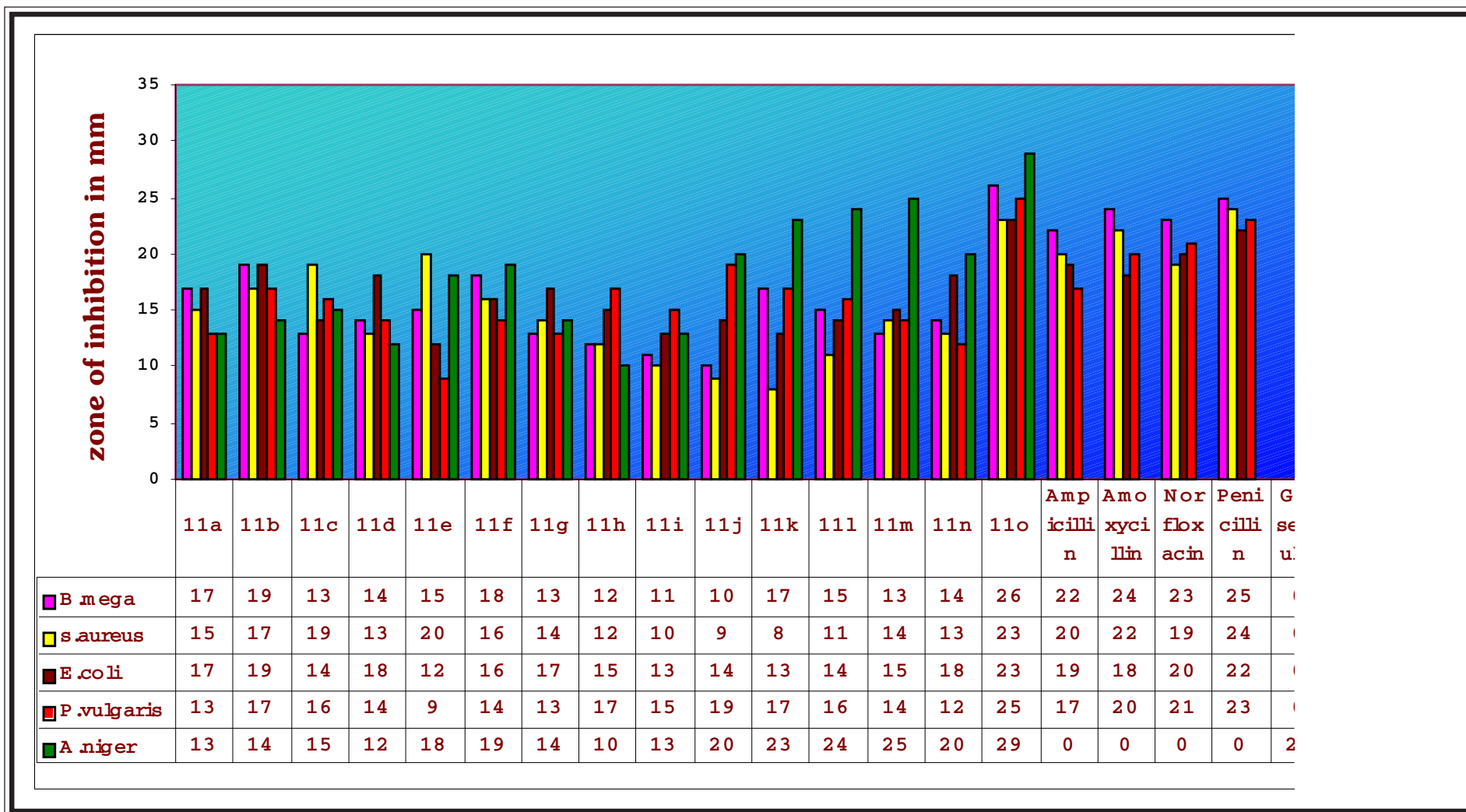
Sr. No. 1	R 2	Molecular Formula 3	Molecular Weight 4	M.P. °C 5	Rf Value 6	Yield % 7	% of Nitrogen	
							Calcd. 8	Found 9
12a	C ₆ H ₅ -	C ₁₇ H ₁₁ N ₃ Cl ₂	328	110	0.562	58	12.28	12.50
12b	2-Cl-C ₆ H ₄ -	C ₁₇ H ₁₀ N ₃ Cl ₃	362.5	162	0.489	64	11.59	11.75
12c	3-Cl-C ₆ H ₄ -	C ₁₇ H ₁₀ N ₃ Cl ₃	362.5	120	0.595	63	11.59	11.33
12d	4-Cl-C ₆ H ₄ -	C ₁₇ H ₁₀ N ₃ Cl ₃	362.5	176	0.632	61	11.59	11.40
12e	2,6-(Cl) ₂ -C ₆ H ₃ -	C ₁₇ H ₉ N ₃ Cl ₄	397	138	0.658	66	10.58	10.77
12f	3,4-(Cl) ₂ -C ₆ H ₃ -	C ₁₇ H ₉ N ₃ Cl ₄	397	140	0.767	72	10.58	10.82
12g	3-Cl,4-F-C ₆ H ₃ -	C ₁₇ H ₉ N ₃ Cl ₃ F	379.5	118	0.684	79	11.07	11.23
12h	4-F-C ₆ H ₄ -	C ₁₇ H ₁₀ N ₃ Cl ₂ F	346	162	0.639	57	12.14	12.45
12i	2-OCH ₃ -C ₆ H ₄ -	C ₁₈ H ₁₃ N ₃ Cl ₂ O	358	152	0.527	55	11.73	11.98
12j	4-OCH ₃ -C ₆ H ₄ -	C ₁₈ H ₁₃ N ₃ Cl ₂ O	358	126	0.510	50	11.73	11.51
12k	2-CH ₃ -C ₆ H ₄ -	C ₁₈ H ₁₃ N ₃ Cl ₂	342	162	0.687	45	12.28	12.49
12l	4-CH ₃ -C ₆ H ₄ -	C ₁₈ H ₁₃ N ₃ Cl ₂	342	160	0.676	43	12.28	12.53
12m	2-NO ₂ -C ₆ H ₄ -	C ₁₇ H ₁₀ N ₄ Cl ₂ O ₂	373	142	0.662	58	15.01	15.24
12n	3-NO ₂ -C ₆ H ₄ -	C ₁₇ H ₁₀ N ₄ Cl ₂ O ₂	373	140	0.537	59	15.01	14.68
12o	4-NO ₂ -C ₆ H ₄ -	C ₁₇ H ₁₀ N ₄ Cl ₂ O ₂	373	158	0.428	64	15.01	14.79

TLC Solvent System : Acetone : Benzene (2.8 : 7.2).





GRAPHICAL CHART NO. 11 : α -ARYLAMINO- α -2,7-DICHLOROQUINOLIN-3-YL-ACETONITRILES





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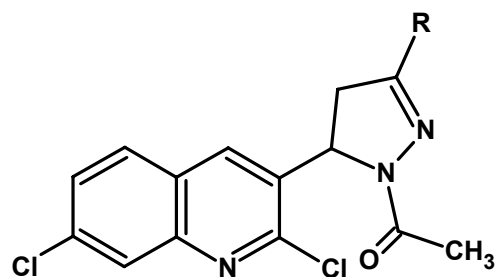
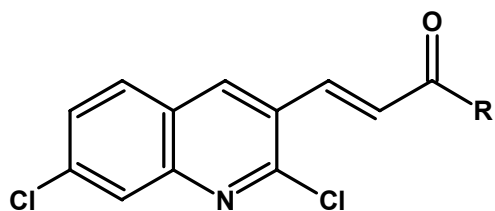
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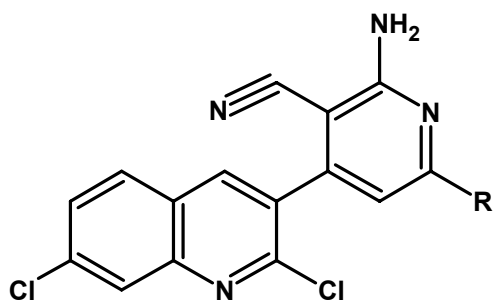
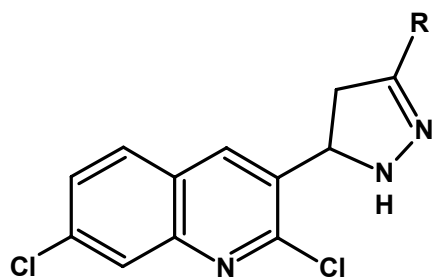


**LIST OF
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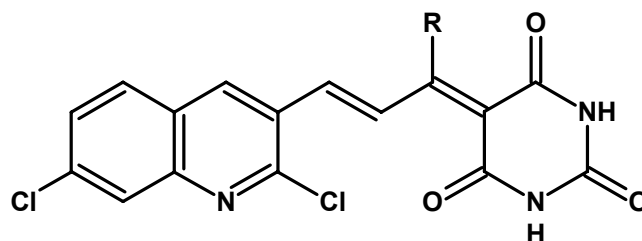
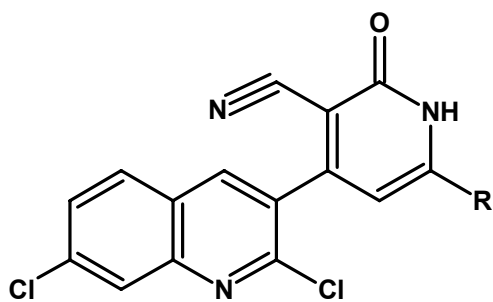




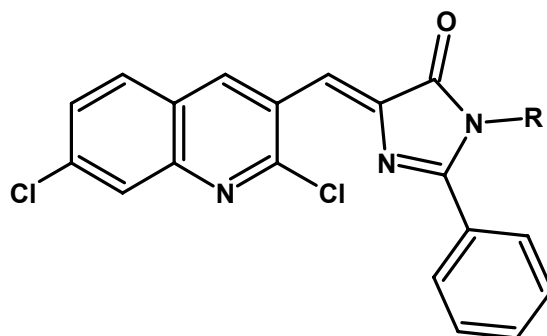
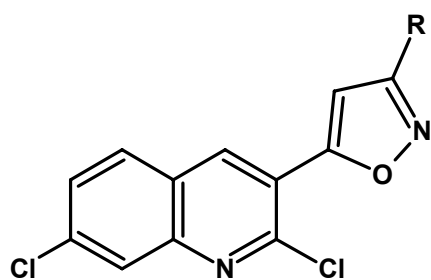
R	R
C_6H_5	C_6H_5
3-NH ₂ -C ₆ H ₄	4-NH ₂ -C ₆ H ₄
4-NH ₂ -C ₆ H ₄	4-Br-C ₆ H ₄
4-Br-C ₆ H ₄	4-Cl-C ₆ H ₄
4-Cl-C ₆ H ₄	4-F-C ₆ H ₄
2,6-(OH) ₂ -C ₆ H ₃	C_4H_3O
4-F-C ₆ H ₄	2-OH-C ₆ H ₄
C_4H_3O	4-OH-C ₆ H ₄
2-OH-C ₆ H ₄	4-OCH ₃ -C ₆ H ₄
4-OH-C ₆ H ₄	4-CH ₃ -C ₆ H ₄
4-OCH ₃ -C ₆ H ₄	3-NO ₂ -C ₆ H ₄
4-CH ₃ -C ₆ H ₄	4-NO ₂ -C ₆ H ₄
3-NO ₂ -C ₆ H ₄	
4-NO ₂ -C ₆ H ₄	



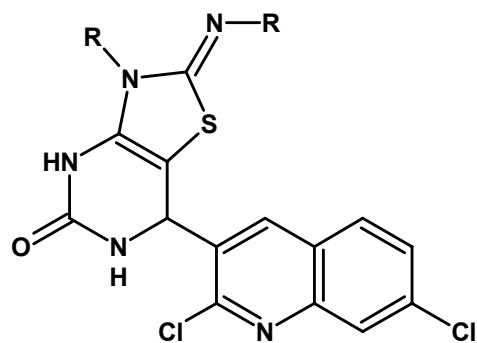
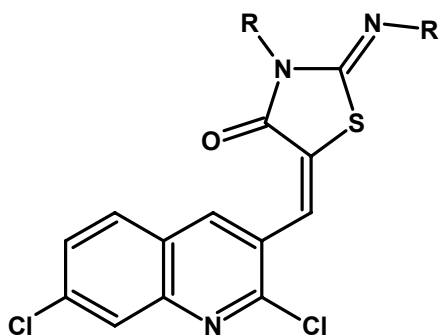
R	R
C_6H_5	C_6H_5
4-NH ₂ -C ₆ H ₄	4-NH ₂ -C ₆ H ₄
4-Br-C ₆ H ₄	4-Br-C ₆ H ₄
4-Cl-C ₆ H ₄	4-Cl-C ₆ H ₄
4-F-C ₆ H ₄	4-F-C ₆ H ₄
2-OH-C ₆ H ₄	C ₄ H ₃ O
4-OH-C ₆ H ₄	2-OH-C ₆ H ₄
4-OCH ₃ -C ₆ H ₄	4-OH-C ₆ H ₄
4-CH ₃ -C ₆ H ₄	4-OCH ₃ -C ₆ H ₄
3-NO ₂ -C ₆ H ₄	4-CH ₃ -C ₆ H ₄
4-NO ₂ -C ₆ H ₄	3-NO ₂ -C ₆ H ₄
	4-NO ₂ -C ₆ H ₄



R	R
C_6H_5	C_6H_5
3-NH ₂ -C ₆ H ₄	3-NH ₂ -C ₆ H ₄
4-NH ₂ -C ₆ H ₄	4-NH ₂ -C ₆ H ₄
4-Br-C ₆ H ₄	4-Br-C ₆ H ₄
4-Cl-C ₆ H ₄	4-Cl-C ₆ H ₄
2,6-(OH) ₂ -C ₆ H ₃	2,6-(OH) ₂ -C ₆ H ₃
4-F-C ₆ H ₄	4-F-C ₆ H ₄
2-OH-C ₆ H ₄	2-OH-C ₆ H ₄
4-OH-C ₆ H ₄	4-OH-C ₆ H ₄
4-OCH ₃ -C ₆ H ₄	4-OCH ₃ -C ₆ H ₄
4-CH ₃ -C ₆ H ₄	4-CH ₃ -C ₆ H ₄
3-NO ₂ -C ₆ H ₄	3-NO ₂ -C ₆ H ₄
4-NO ₂ -C ₆ H ₄	4-NO ₂ -C ₆ H ₄



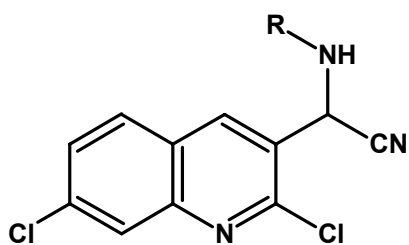
R	R
C_6H_5	C_6H_5
3-NH ₂ -C ₆ H ₄	2-Cl-C ₆ H ₄
4-NH ₂ -C ₆ H ₄	3-Cl-C ₆ H ₄
4-Br-C ₆ H ₄	4-Cl-C ₆ H ₄
4-Cl-C ₆ H ₄	2,5-(Cl) ₂ -C ₆ H ₃
2,6-(OH) ₂ -C ₆ H ₃	3,4-(Cl) ₂ -C ₆ H ₃
4-F-C ₆ H ₄	3-Cl,4-F-C ₆ H ₃
C ₄ H ₃ O	2-OCH ₃ -C ₆ H ₄
2-OH-C ₆ H ₄	4-OCH ₃ -C ₆ H ₄
4-OH-C ₆ H ₄	2-CH ₃ -C ₆ H ₄
4-OCH ₃ -C ₆ H ₄	4-CH ₃ -C ₆ H ₄
4-CH ₃ -C ₆ H ₄	2-NO ₂ -C ₆ H ₄
3-NO ₂ -C ₆ H ₄	3-NO ₂ -C ₆ H ₄
4-NO ₂ -C ₆ H ₄	4-NO ₂ -C ₆ H ₄



R

R

 C_6H_5 C_6H_5 4-Cl- C_6H_4 4-Cl- C_6H_4 3,4-(Cl) $_2$ - C_6H_3 3,4-(Cl) $_2$ - C_6H_3 4-F- C_6H_4 4-F- C_6H_5 2-OCH $_3$ - C_6H_4 2-OCH $_3$ - C_6H_4 4-OCH $_3$ - C_6H_4 4-OCH $_3$ - C_6H_4 2-CH $_3$ - C_6H_4 2-CH $_3$ - C_6H_4 3-CH $_3$ - C_6H_4 3-CH $_3$ - C_6H_4 4-CH $_3$ - C_6H_4 4-CH $_3$ - C_6H_4 2-NO $_2$ - C_6H_4 2-NO $_2$ - C_6H_4 3-NO $_2$ - C_6H_4 3-NO $_2$ - C_6H_4 4-NO $_2$ - C_6H_4 4-NO $_2$ - C_6H_4



R

C_6H_5

2-Cl- C_6H_4

3-Cl- C_6H_4

4-Cl- C_6H_4

2,6-(Cl) $_2$ - C_6H_3

3,4-(Cl) $_2$ - C_6H_3

3-Cl, 4-F- C_6H_3

4-F- C_6H_4

2-OCH $_3$ - C_6H_4

4-OCH $_3$ - C_6H_4

2-CH $_3$ - C_6H_4

4-CH $_3$ - C_6H_4

2-NO $_2$ - C_6H_4

3-NO $_2$ - C_6H_4

4-NO $_2$ - C_6H_4