# Determination of Structure of Precursor of *Ibuprofen*: Purification and Analysis

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## Abstract

(S)-2-(p-isobutylphenyl)propanoic acid, so called ibuprofen, is one of the most popular anti-inflammatory agents. An efficient cross-coupling reaction of organozinc compound of ethyl 2-bromoacrylate, prepared by the reaction of highly reactive zinc was carried out with aryl iodide to synthesize 2-(4-isobutylphenyl)-propenoate, a precursor of ibuprofen. Highly reactive zinc metal (EGZn/Naph) was readily prepared by electrolysis of a DMF solution containing naphthalene and a supporting electrolyte in a one-compartment cell fitted with a platinum cathode and a zinc anode. The synthesized precursor of ibuprofen was purified by thin layer chromatography on silica gel and its structure was confirmed by elemental analysis, mass, IR, <sup>1</sup>H NMR and <sup>13</sup>C NMR spectroscopic analysis.

Keywords:, reactive zinc, organozinc compound, ibuprofen, thin layer chromatography

#### Introduction

(S)-Ibuprofen is one of the most popular anti-inflammatory agents and one of the ingredients of mostly used medicine for a cold due to its high potency. Therefore, an asymmetric synthesis of (S)-ibuprofen is an extremely important study. The author applied the efficient cross-coupling reaction of organozine compound of ethyl 2-bromoacrylate, prepared by the reaction with EGZn/Naph, with aryl iodides in the presence of palladium catalyst [1] to a synthesis of ethyl 2-(4-isobutylphenyl)propenoate (4), a precursor of ibuprofen. Zhang et al. have already reported an asymmetric hydrogenation of 2-(p-isobutylphenyl)propenoic acid with Ru/(S)-BINAP gave (S)-ibuprofen in a high enantioselectivity (Scheme 1) [2].

$$CO_2H$$
 Ru/(S)-BINAP catalyst  $H_2$   $(S)$ -(+)-Ibuprofen 97 %, 97 %ee

Scheme 1

## Result and Discussion

Highly reactive electrogenerated zinc metal (EGZn/Naph) was readily prepared by electrolysis of a DMF solution containing naphthalene and a supporting electrolyte in a one-compartment cell fitted with a platinum cathode and a zinc anode (Scheme 2) [2]. The EGZn/Naph thus prepared can

Scheme 2

readily convert ethyl 2-bromoacrylate (1) to the corresponding organozine compound (2) in an almost quantitative yield. Subsequent cross-coupling reaction of 2

with 1-iodo-4-isobutylbenzene (3) in the presence of 5 mol%  $Pd(P(o-Tol)_3)_2Cl_2$  catalyst gave the crude product of ethyl 2-(4-isobutylphenyl)- propenoate (4) (Scheme 3).

EGZn/Naph 
$$\frac{Br + CO_2Et}{DMF, -20^{\circ}C, 1 \text{ h}} \begin{bmatrix} BrZn + 2CO_2Et \end{bmatrix} \xrightarrow{DMF, 70^{\circ}C, 3 \text{ h}} \begin{bmatrix} CO_2Et + 2CO_2Et \end{bmatrix} \xrightarrow{DMF, 70^{\circ}C, 3 \text{ h}} \begin{bmatrix} CO_2Et + 2CO_2Et + 2CO_2Et + 2CO_2Et \end{bmatrix}$$

## Scheme 3

The synthesized crude product was purified by thin-layer chromatography on a Merck Kieselgel 60 PF<sub>254</sub> with ethyl acetate-hexane (1:5) to give 4, the precursor of ibuprofen in 93% yield. The structure of this ibuprofen was confirmed by IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR, mass spectroscopic analysis and elemental analysis. Spectral data of 4 is shown below.

Colorless oil. IR (neat, cm<sup>-1</sup>) v: 1720, 1615, 1512, 1466, 1367, 1199, 1181, 1089, 850, 806. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 7.34 (d, 2H, J=8.3 Hz), 7.13 (d, 2H, J=8.3 Hz), 6.28 (d, 1H, J=1.3 Hz), 5.87 (d, 1H, J=1.3 Hz), 4.29 (q, 2H, J=7.3 Hz), 2.48 (d, 2H, J=6.9 Hz), 1.87 (m, 1H), 1.33 (t, 3H, J=7.3 Hz), 0.91 (d, 6H, J=6.9 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ : 163.99, 142.41, 141.37, 130.91, 128.75, 125.45, 118.98, 60.90, 44.94, 30.01, 22.28, 13.95. EIMS m/z (relative intensity) 232(63), 189(100), 161(16), 145(15), 115(28). HRMS Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>2</sub> m/z 232.1463. Found m/z 232.1472. Anal. Calcd for C<sub>15</sub>H<sub>20</sub>O<sub>2</sub>: C, 77.55; H, 8.68. Found: C, 77.68; H, 8.83.

## References

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