

The quaternisation of phosphines:

A blank canvas reaction for the generation of smart materials for electrochemical devices.

Andrew Kavanagh

University College Dublin
8th August 2012.

Talk Outline

Introduction

- NCSR
- CLARITY
- Adaptive Sensors Group

Andrew Kavanagh

- Ionic Liquids
- Ionogels
- Ionic Liquid Synthesis

Electrochemical Devices

- **Device Complexity**

Electrochromic Displays

- **Leaching of electrolyte**

Polymeric Ionic Liquids

- **Device Sensitivity**

2-photon lithography

Conclusions & Acknowledgements

National Centre for Sensor Research



- Housed in the campus of Dublin City University.
- Multidisciplinary research institute invested in the advancement of sensor science.
- Over 260 full time researchers and support staff.
- Investments and income since 1999 now approaching € 100 million.



CLARITY

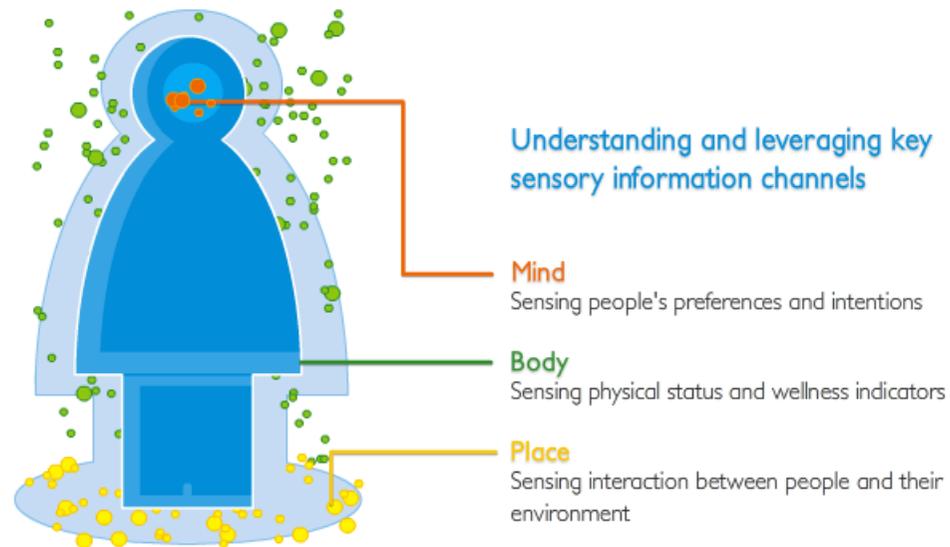
The Centre for Sensor Web Technologies



CLARITY – SFI CSET



Vision: Sensing Mind, Body & Place



- 5-year, €16.4 million research program to develop next generation Sensor Web Technologies with significant environmental focus.
- Brings together fundamental materials science, functional polymers, device prototyping, energy management, adaptive middleware, wearable sensors, distributed environmental monitoring.

CLARITY Centre & Ecosystem

INDUSTRY COLLABORATORS

SOCIAL/AGENCY COLLABORATORS

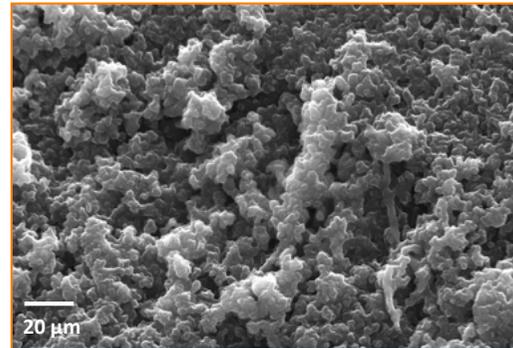


The Adaptive Sensors Group

- Chemical Sensor element of CLARITY.
- Group Leader: Professor Dermot Diamond.
- Contains >20 researchers engaged in:
 - Microfluidics
 - Electronic Engineering
 - Platform Testing (Analytical Chemistry)
 - Novel Materials Synthesis



Prof. Dermot Diamond



www.adaptivesensors.com

The Adaptive Sensors Group

Modus Operandi

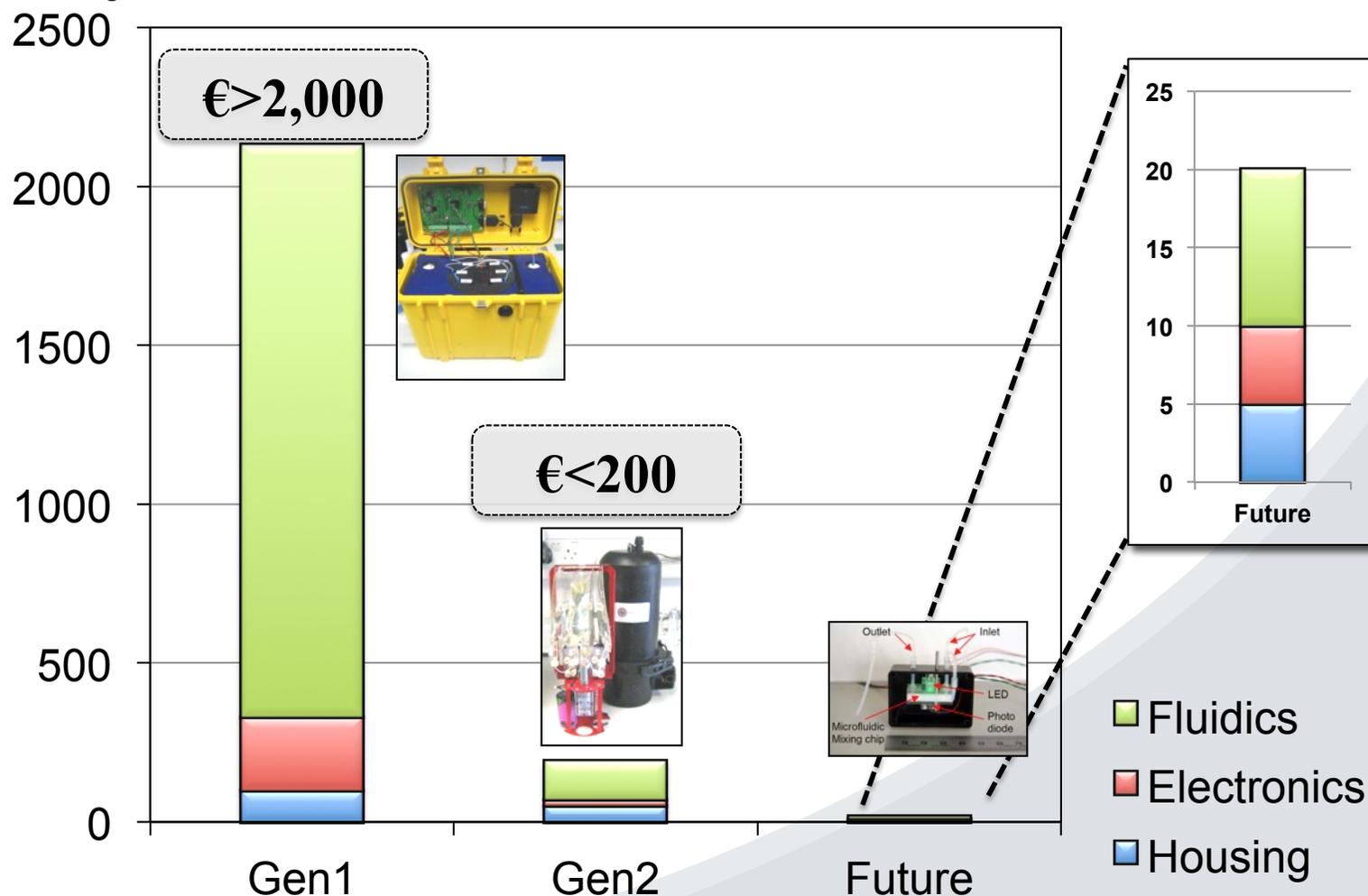
A two-pronged approach to improve the quality of autonomous environmental chemical analysers ¹.

1. Evolutionary: Drive down the cost of analysers through clever engineering.

2. Revolutionary: Develop disruptive technologies based on breakthroughs in fundamental materials science.

1. R. Byrne, C. Ventura, F. B. Lopez, A. Walther, A. Heise and D. Diamond, *Biosensors & Bioelectronics*, 2010, **26**, 1392-1398.

Platform Evolution: Autonomous Analyser for Phosphates in River Water



Current Industry Standard Instrument for Phosphate analysis costs in the region of € 15,000.

Mé Féin: Part of the Revolutionary Team!

Research Interests:

- Thermal Actuators
- Lithography
- Optical Sensing
- Materials Synthesis
- Photodynamic Surface/Liquid control

**Stimuli
Responsive
Materials**

Electrochemical Devices

**Ionic
Liquids**

**Polymer
Gels**

Ionic Liquids

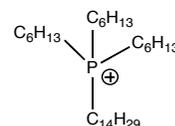
• Organic liquids composed solely of ions.²

• Important liquid properties such as density, viscosity and melting point can be altered by variation of the ion pair.³

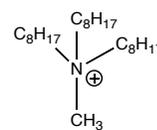
• Ionic Liquids exhibit good ionic conductivity and stable electrochemical windows.⁴

2. P. Wasserscheid and T. Welton, *Ionic Liquids in Synthesis*, WILEY-VCH, Weinheim, 2003.
3. R. E. Del Sesto, C. Corley, A. Robertson and J. S. Wilkes, *Journal of Organometallic Chemistry*, 2005, **690**, 2536-2542.
4. C. Zhao, G. Burrell, A. A. J. Torriero, F. Separovic, N. F. Dunlop, D. R. MacFarlane and A. M. Bond, *Journal of Physical Chemistry B*, 2008, **112**, 6923-6936.

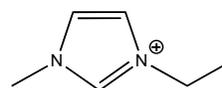
Cations



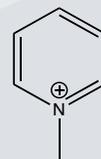
(i) [P_{6,6,6,14}]⁺



(ii) [N_{1,8,8,8}]⁺

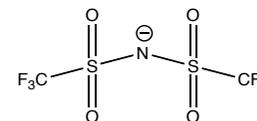


(iii) [C₂mIm]⁺

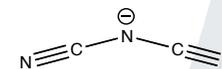


(iv) [mpyr]⁺

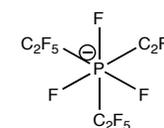
Anions



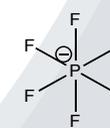
(v) [NTf₂]⁻



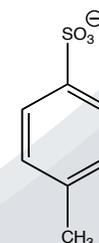
(vi) [DCA]⁻



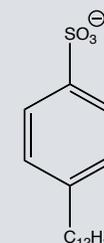
(vii) [FAP]⁻



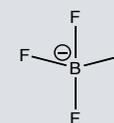
(viii) [PF₆]⁻



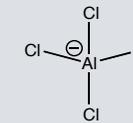
(ix) [TOS]⁻



(x) [DBSA]⁻

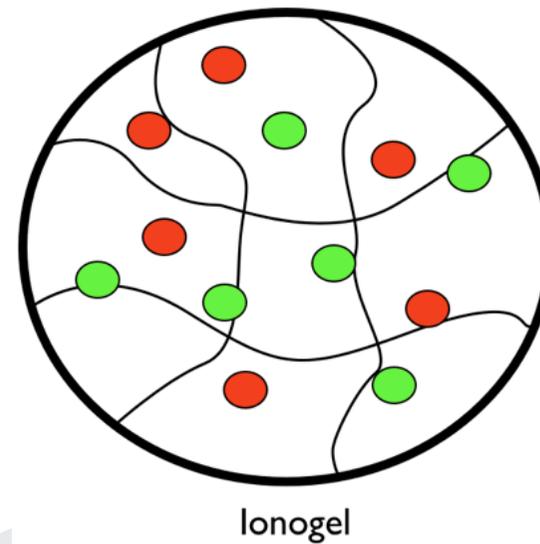
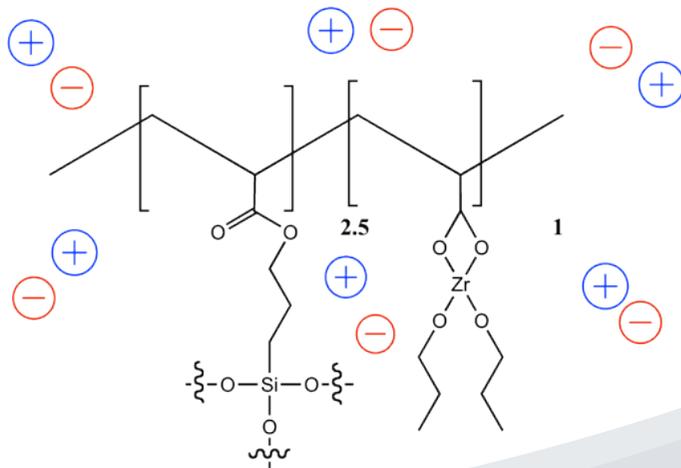
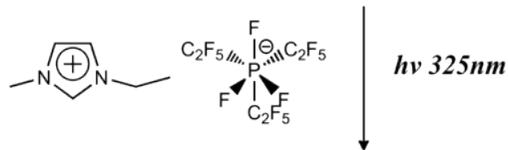
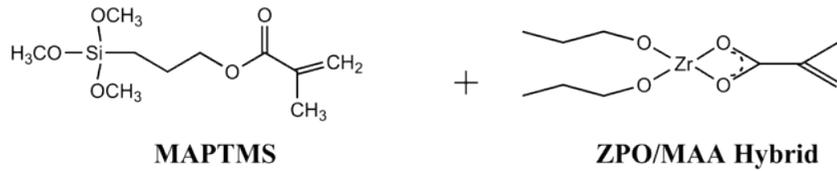


(xi) [BF₄]⁻



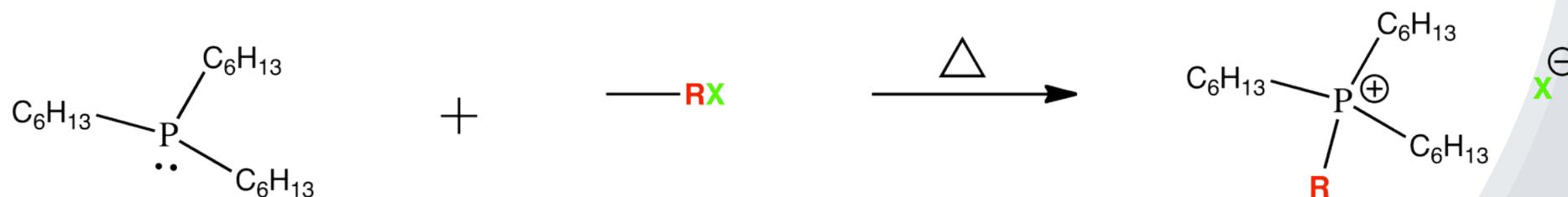
(xii) [AlCl₄]⁻

IL encapsulated within a polymer network: *Ionogel*



- Cation
- Anion
- Polymer matrix

The blank canvas!



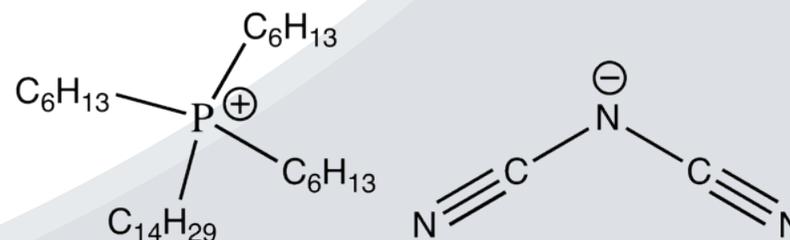
- Direct nucleophilic addition of a trialkyl phosphine with an alkyl halide to form a phosphonium salt.

- When: **R** = C₁₄H₂₉ and **X** = DCA⁻....

Liquid at room temperature salt

$\eta = 256 \text{ mPa}\cdot\text{s}$

$T_{\text{deg}} = 395 \text{ }^{\circ}\text{C}$



5. K. J. Fraser and D. R. MacFarlane, *Australian Journal of Chemistry*, 2009, **62**, 309-321.

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Electrochemical Devices

- **Device Complexity**

Electrochromic Displays

- **Leaching of electrolyte**

Polymeric Ionic Liquids

- **Device Sensitivity**

2-photon lithography

Conclusions & Acknowledgements

Electrochemical Devices (ECD's)

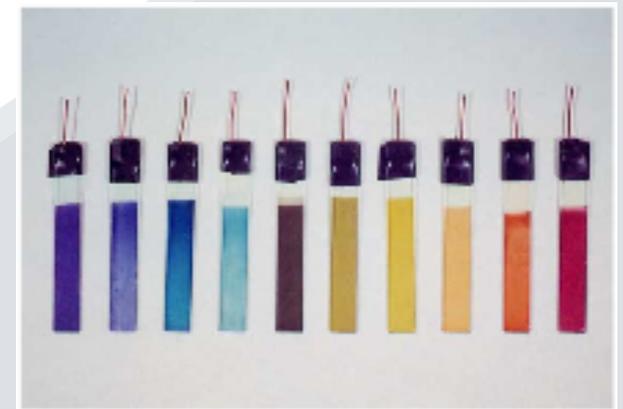
3 Key Issues To Address:

- 1. Device complexity:** Some electrochemical devices can contain multiple components, each with their own defined role relating to device performance.
- 2. Leaching of electrolyte:** Loss of electroactive components dramatically reduces device lifetime!
- 3. Device Sensitivity:** For electrochemical sensors in particular, increased sensitivity of device components can dramatically improve sensor performance.

Ionic Liquids have the answer!!

Device Complexity: Electrochromic Devices

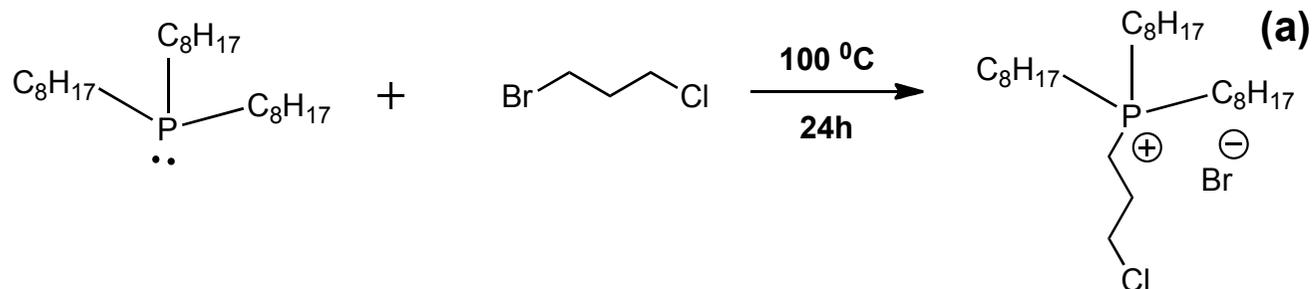
- **Electrodes** Source of charge, site of electrochemical reaction
- **Electrolyte** Facilitates charge transfer between the electrodes
- **Electrochrome** Redox active material which undergoes a change in optical properties in response to the applied voltage
- **Polymer** Inert matrix to house active sensing components
- **Plasticizer** Lowers T_g of polymer to produce solid film



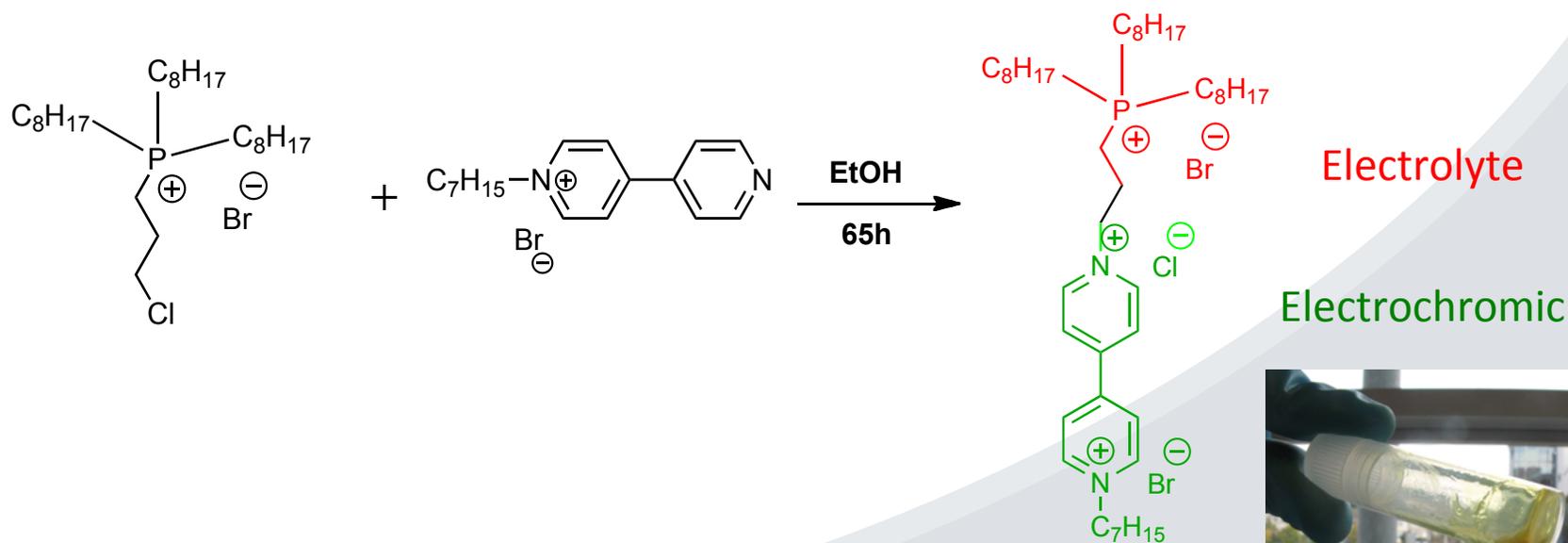
*Rich variation in optical properties of PEDOT based ECD's.*⁶

6. S. A. Sapp, G. A. Sotzing and J. R. Reynolds, *Chemistry of Materials*, 1998, **10**, 2101-2108.

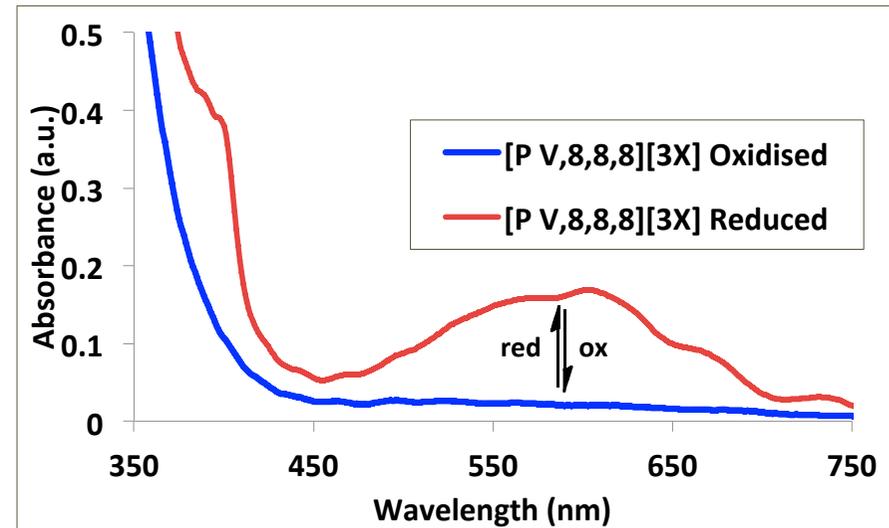
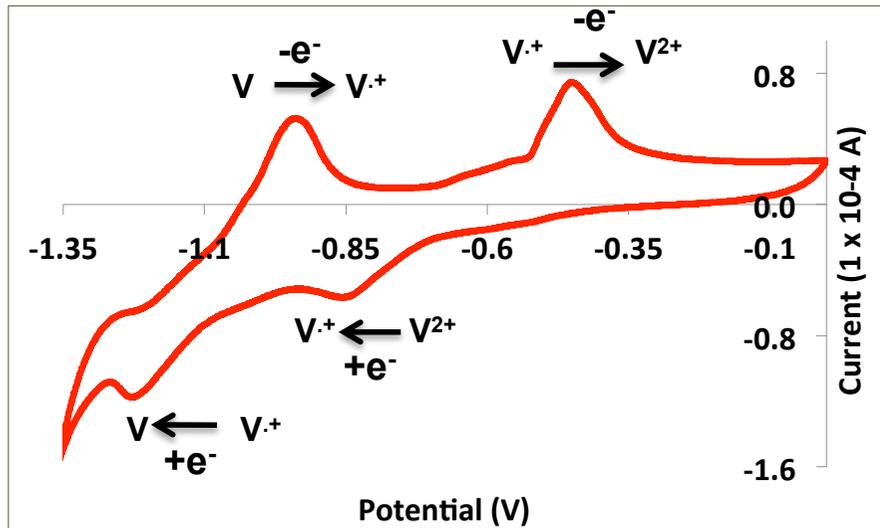
Phosphonium based, electrochromic ionic liquid



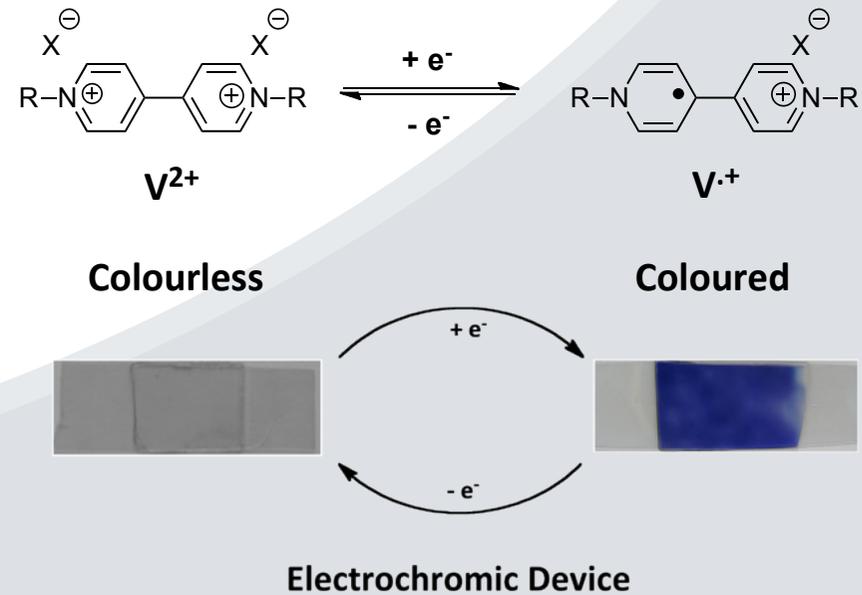
(a): $[\text{P}_{3\text{Cl},8,8,8}][\text{Br}]$



Spectroelectrochemistry

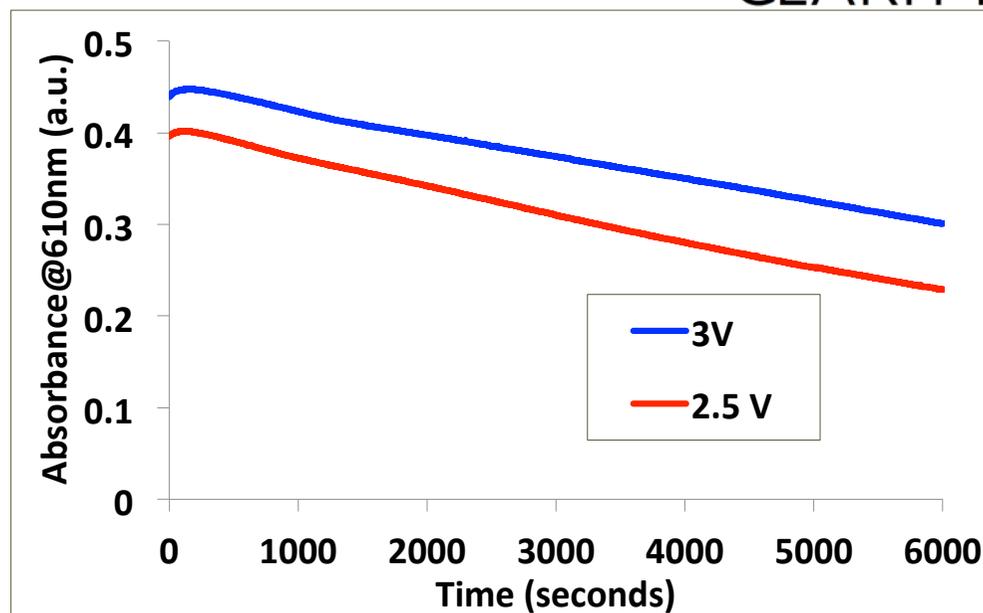


- Electrochromic device fabricated from an ionogel containing the Electrochromic IL.

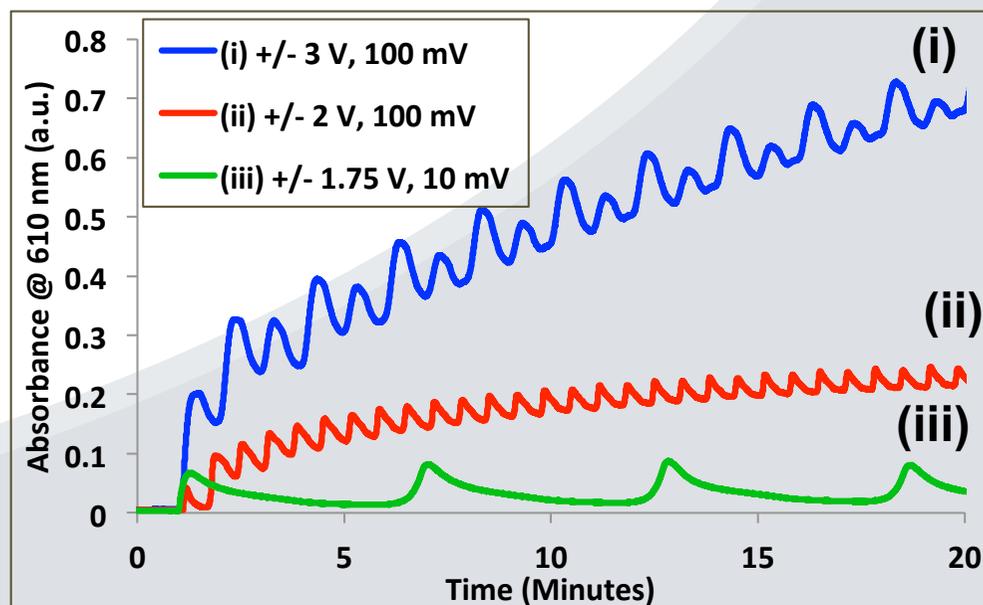


Device Output

- **DC Current:** ~ 60 % of device coloration is maintained after 100 minutes under open circuit.



- **AC Current:** Device can switch between transparent and coloured states under the right cycling conditions.



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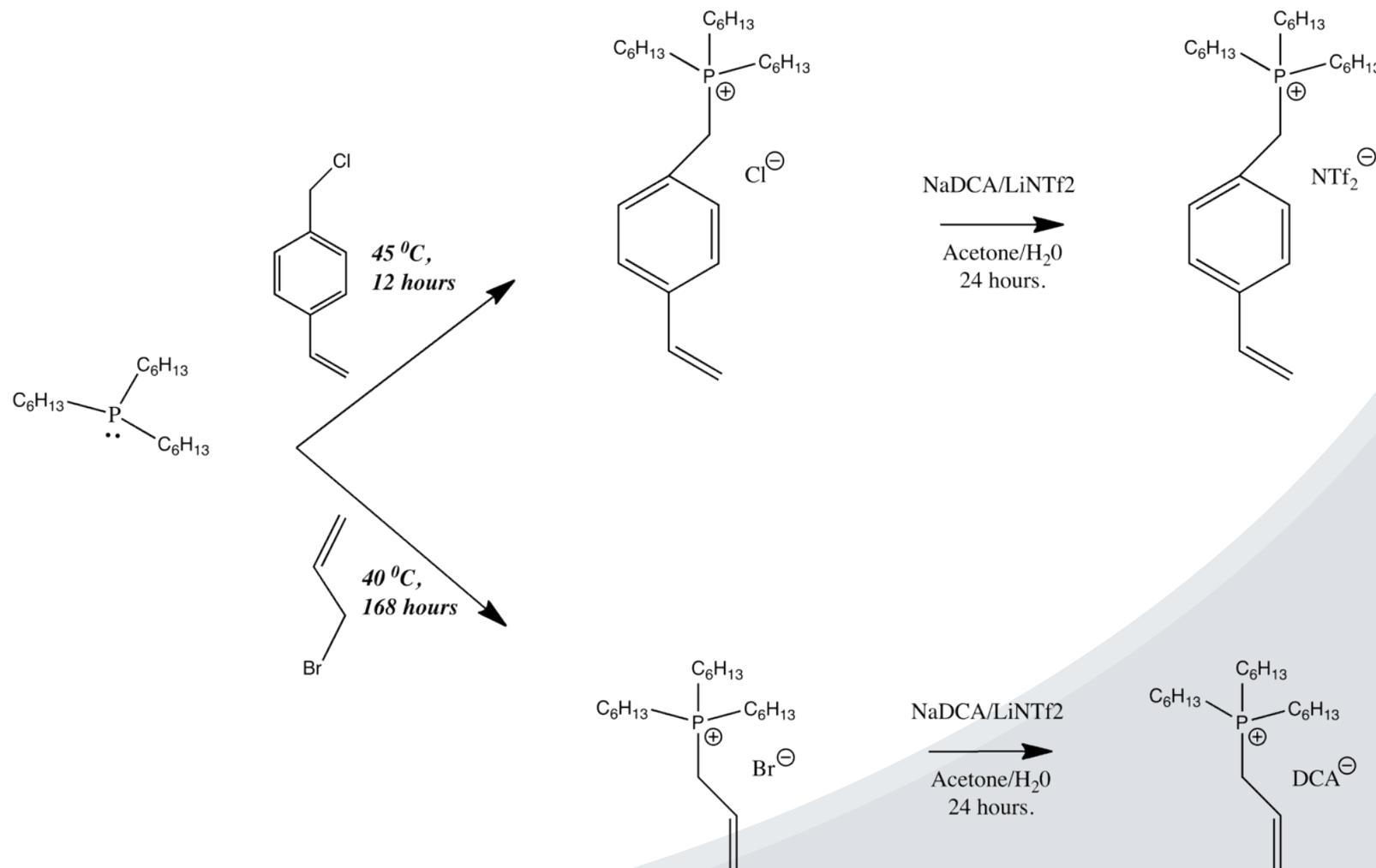
Polymeric Ionic Liquids

- **Device Sensitivity**

2-photon lithography

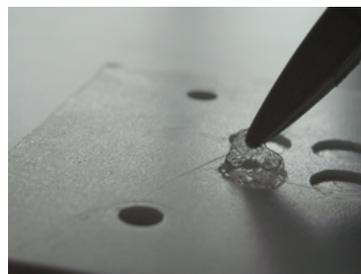
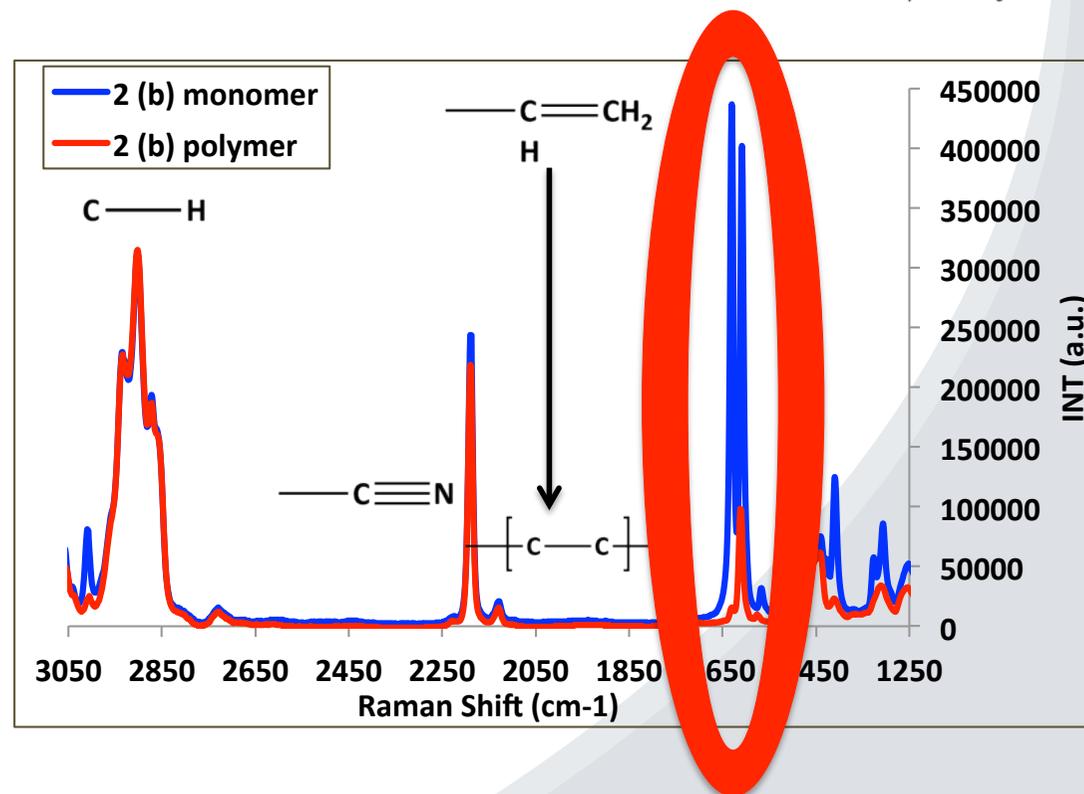
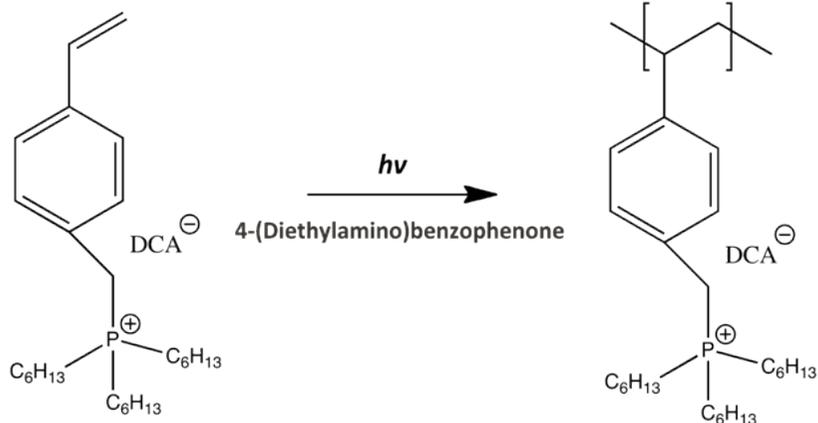
Conclusions & Acknowledgements

Device Leaching



9 Monomeric ILs synthesised in total.

Poly IL characterisation



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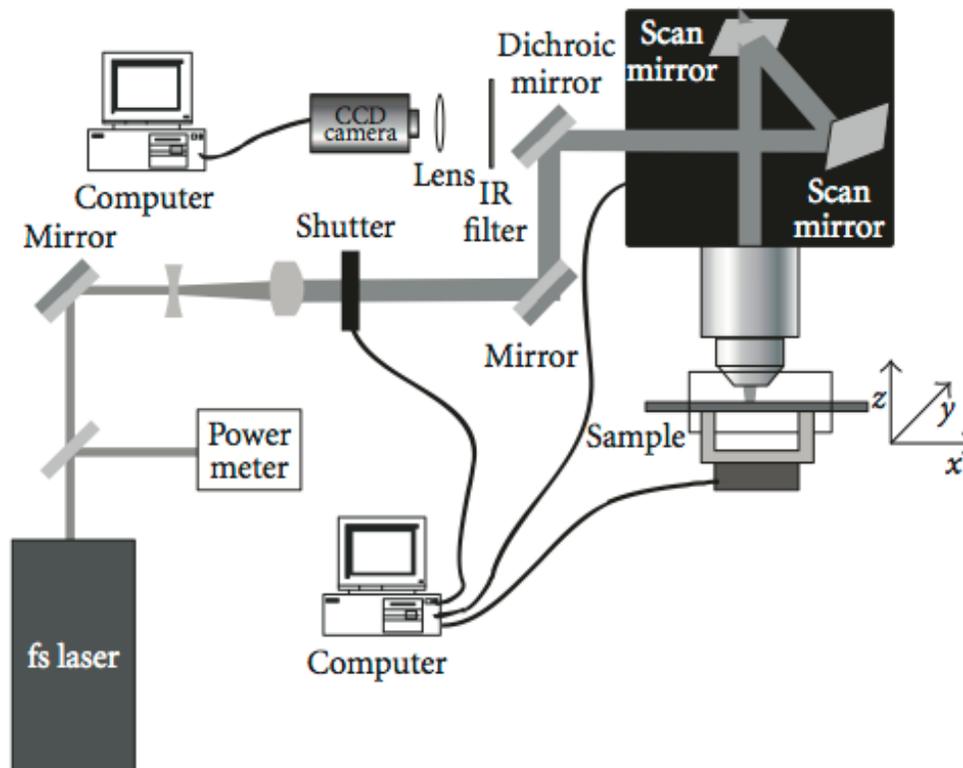
- **Device Sensitivity**

2-photon lithography

Conclusions & Acknowledgements

Material sensitivity

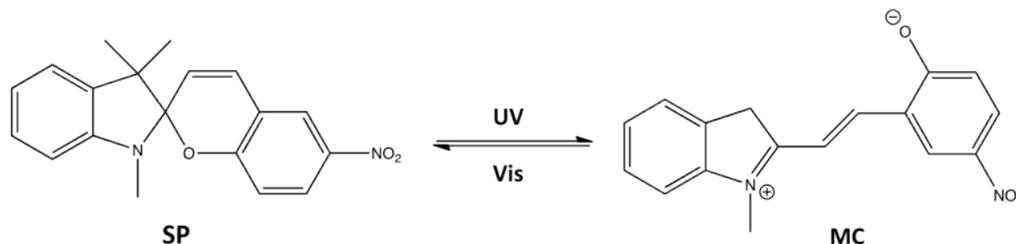
2-photon polymerisation: a direct laser lithographic technique.



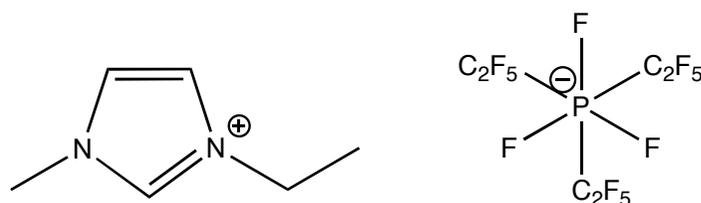
- High frequency laser impulses are passed through high aperture lensing.
- Results in the two photon effect, i.e. Initiator in liquid resin absorbs two photons instead of one.
- Piezoelectric motor controls the movement of the resin in three dimensions. ⁷

7. A. Ovsianikov, A. Gaidukeviciute, B. N. Chichkov, M. Oubaha, B. D. MacCraith, I. Sakellari, A. Giakoumaki, D. Gray, M. Vamvakaki, M. Farsari, C. Fotakis, *Laser Chemistry*, 2008.

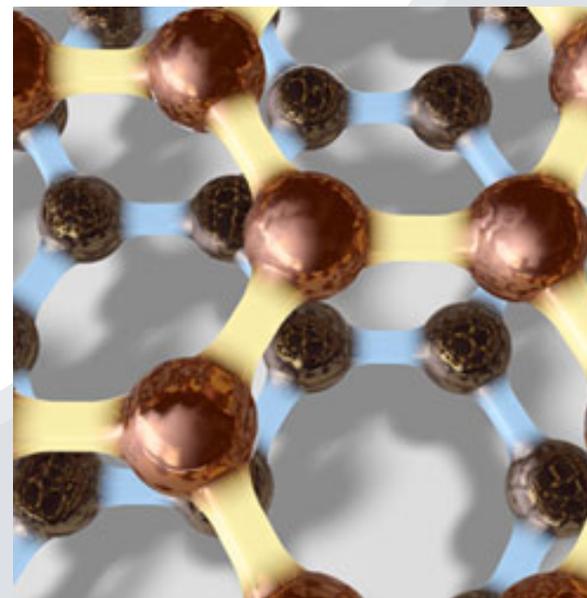
Direct, 3-D laser printing of stimuli responsive materials



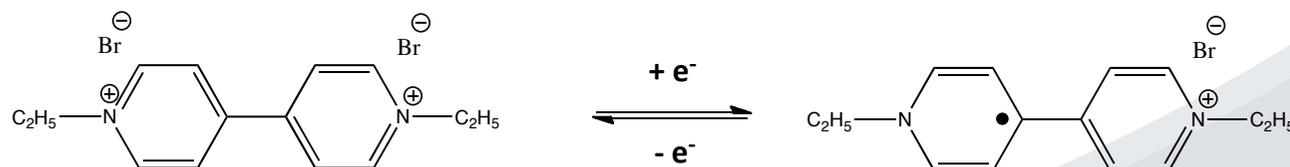
Photochromics - *Spirocyclics*



Ionic Liquids



Dispersed graphene sheets

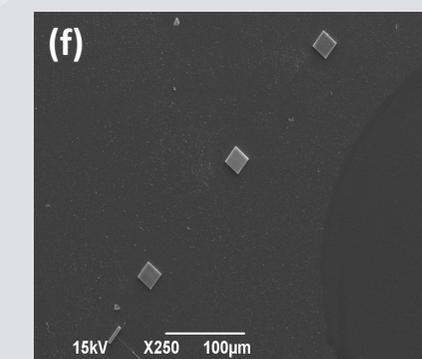
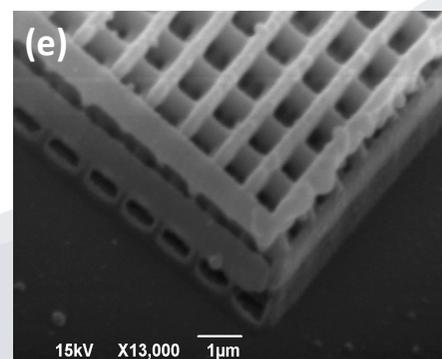
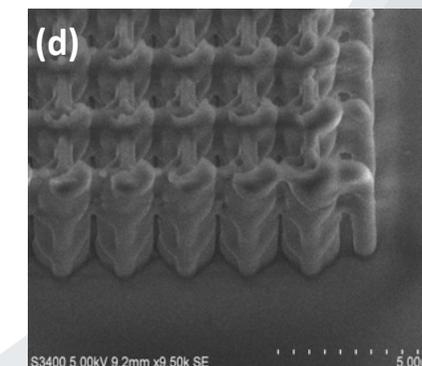
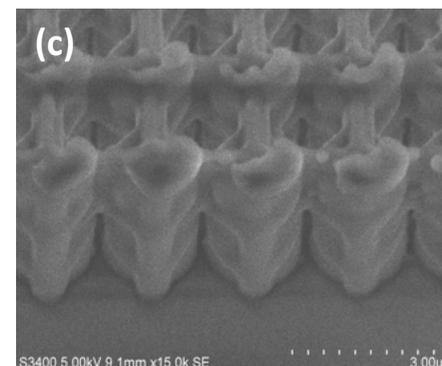
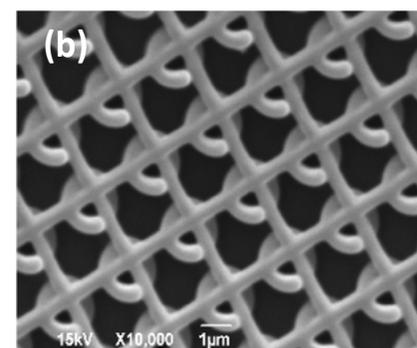
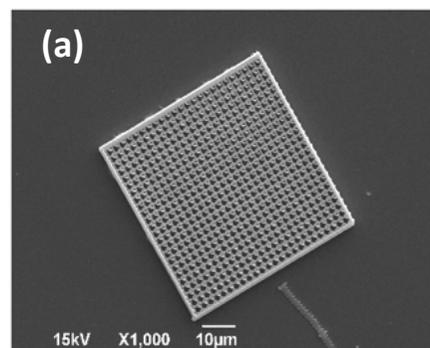
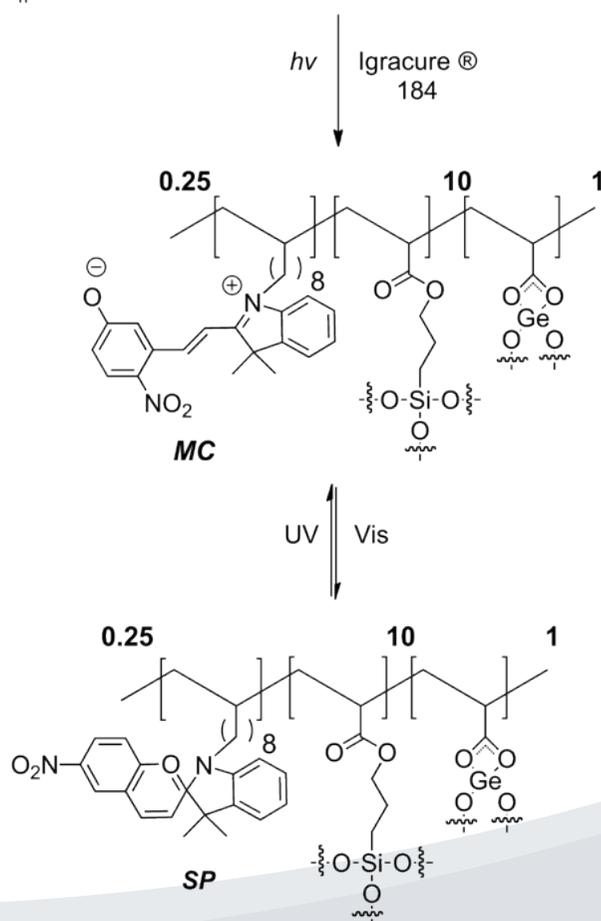
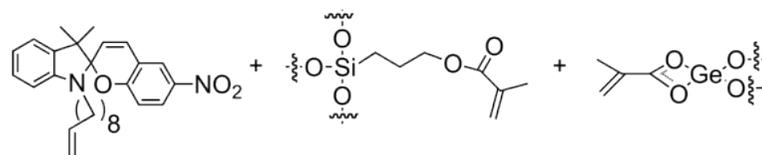


EV 2^+
Colourless

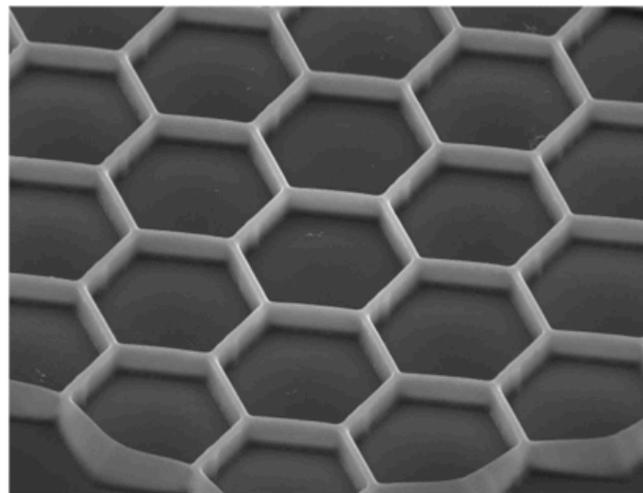
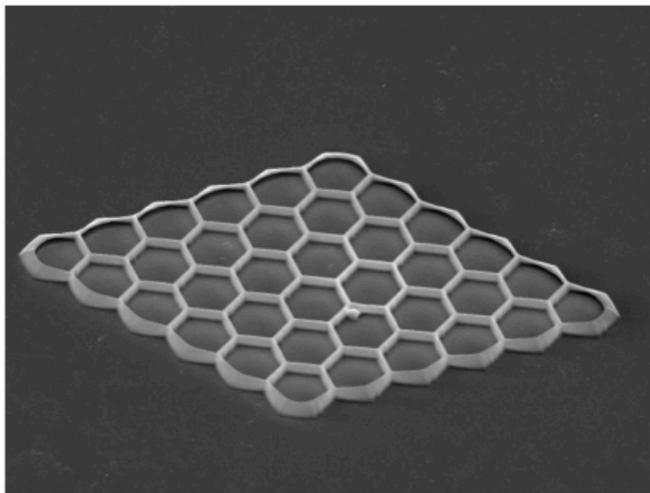
EV \cdot^+
Blue

Electrochromics - *Viologens*

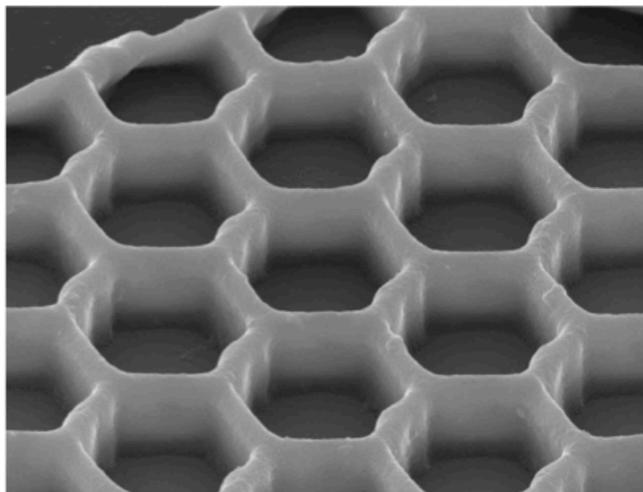
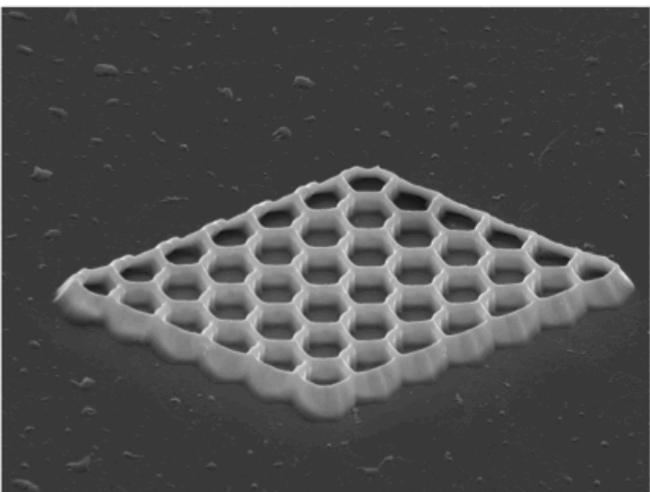
Sub-micron, photoresponsive structures



Photopatterned, graphene dispersed ionogels

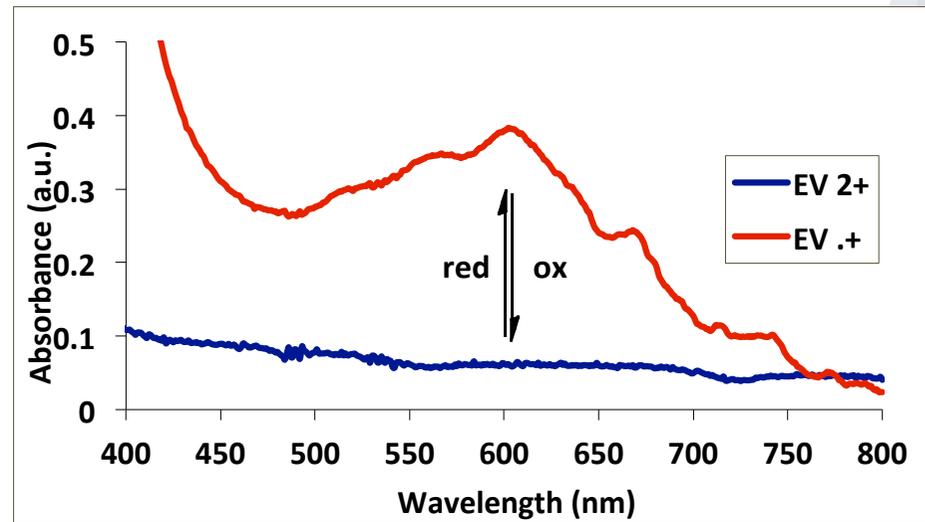
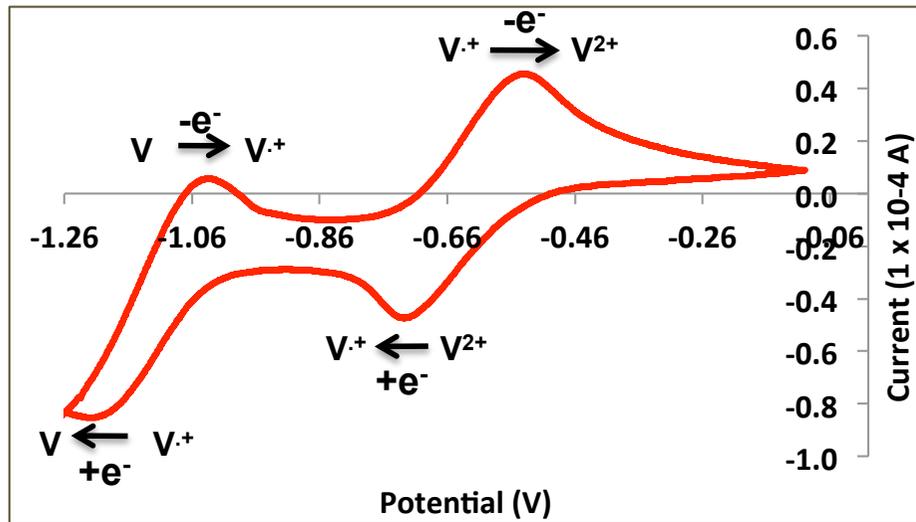


• 8 μm
honeycomb
diameter

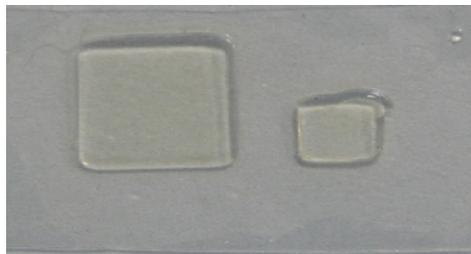


• 2 μm
honeycomb
diameter

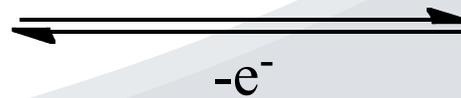
Photopatterned, electrochromic device



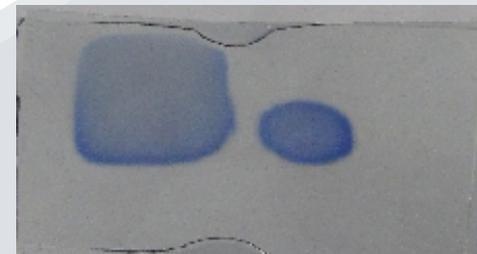
EV²⁺



+e⁻, E_{red}: -1V



EV⁺



λ_{max} 610 nm

Conclusions

- **Device Complexity:** Synthesised Electrochromic IL acts as the electrolyte and the electrochrome when encapsulated between two electrodes.
- **Leaching of Electrolytes:** Synthesised polymeric Ionic Liquids form flexible, conductive films.
- **Material Sensitivity:** 2 - photon lithography is an attractive technique for the generation of high resolution structures *in situ*.

Further Reading

1. **Andrew Kavanagh**, Kevin J. Fraser, Robert Byrne and Dermot Diamond “*An electrochromic ionic liquid: design, synthesis, characterisation and performance in a solid-state platform*”, **Chemistry of Materials**, 2012 (Article Submitted).
2. Mohamed Oubaha, **Andrew Kavanagh**, Arnaud Gorin, Gabija Bickaускаite, Robert Byrne, Maria Farsari, Richard Winfield, Dermot Diamond, Colette McDonagh and Robert Copperwhite, “*Graphene-doped photo-patternable ionogels: tuning of conductivity and mechanical stability of 3D microstructures*”, **Journal of Materials Chemistry**, 2012, **22**, 10552-10559.
3. **Andrew Kavanagh**, Robert Byrne Dermot Diamond and Kevin J. Fraser, “*Stimuli Responsive Ionogels for Sensing Applications - An Overview*” - **Membranes**, 2012, **2**, 16-39. **Review contribution as part of special edition entitled "Stimuli Responsive Polymers"**.
4. **Andrew Kavanagh**, Matthias Hilder, Noel Clark, Aleksandar Radu and Dermot Diamond, “*Wireless Radio Frequency Detection of Greatly Simplified Polymeric Membranes based on a Multifunctional Ionic Liquid*”, **Electrochimica Acta**, 2011, **56**, 8947-8953.
5. Dimitrije Cicmil, Salzitsa Anastasova, **Andrew Kavanagh**, Dermot Diamond, Ulriika Mattinen, Johan Bobacka, Andrzej Lewenstam and Aleksandar Radu, “*Ionic Liquid based, Liquid Junction Free Reference Electrode*”, **Electroanalysis**, 2011, **23(8)**, 1881–1890.
6. **Andrew Kavanagh**, Robert Copperwhite, Mohamed Oubaha, Jessica Owens, Colette McDonagh, Dermot Diamond and Robert Byrne, “*Photopatternable Ionogels for Electrochromic Applications*”, **Journal of Materials Chemistry**, 2011, **21**, 8687-8693.
7. **Andrew Kavanagh**, Robert Byrne, Dermot Diamond and Aleksandar Radu, “*A two component optode membrane based on a multifunctional ionic liquid*”, **Analyst**, 2011, **136**, 348-353.

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Cytec:

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Thanks for listening!!